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Title: CRITICAL MASSES OF URANIUM DILUTED WITH MATRIX MATERIAL

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Critical Masses of Uranium Diluted with Matrix Material

By

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ABSTRACT

Critical masses of square-prisms of highly enriched uranium diluted in various $X/^{235}\text{U}$ ratios with matrix material and polyethylene were measured. The configuration cores were 22.86-cm and 45.72-cm square and were reflected with 8.13-cm and 10.16-cm thick side polyethylene reflectors, respectively. The configurations had 10.16-cm thick top and bottom polyethylene reflectors. For some configurations, the Rossi- α , which is an eigenvalue value characteristic for a particular configuration, was measured to establish a reactivity scale based on the degree of subcriticality. Finally, the critical mass experiments are compared with values calculated with MCNP and ENDF/B-VI cross-sections.

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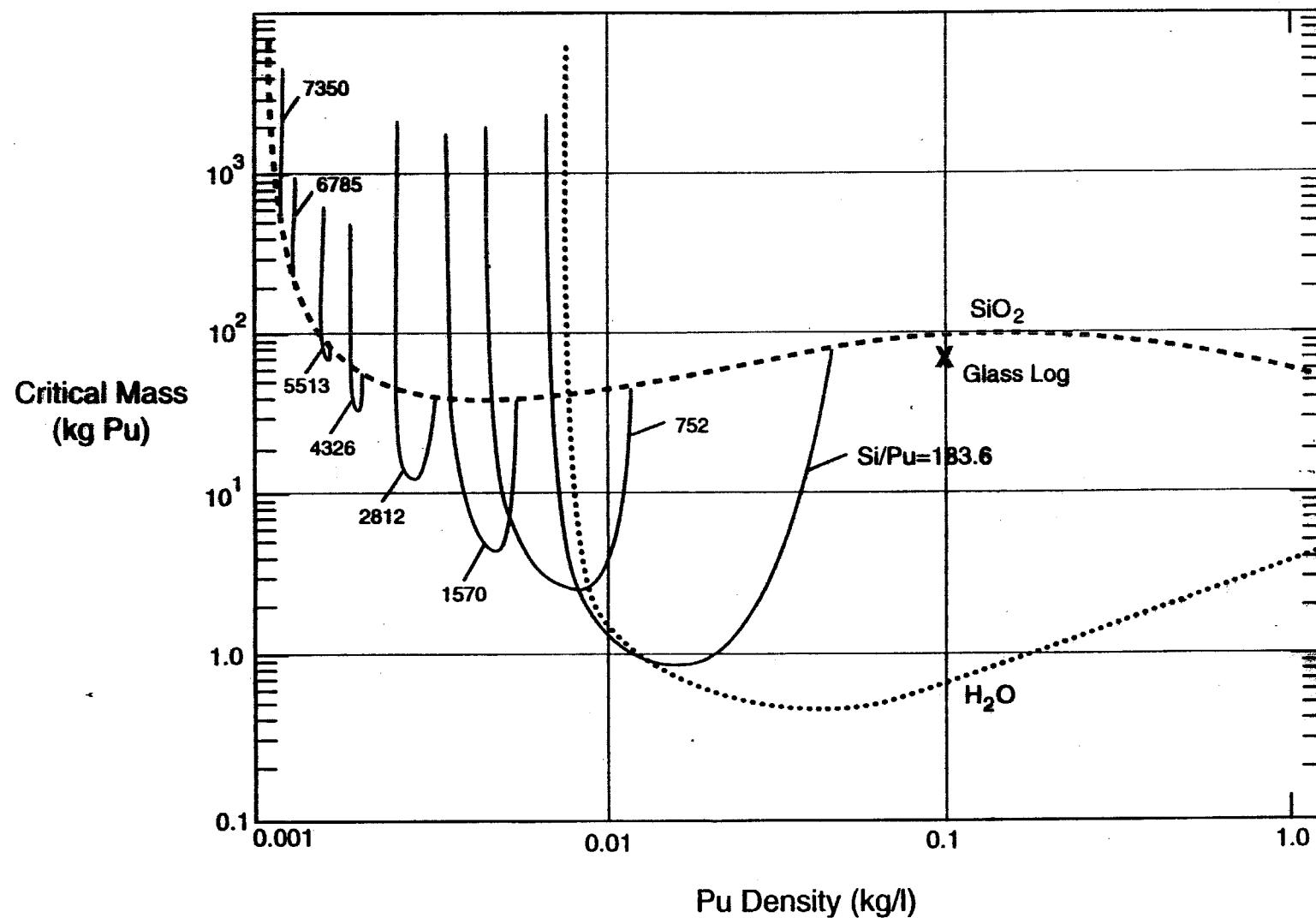
Brief History

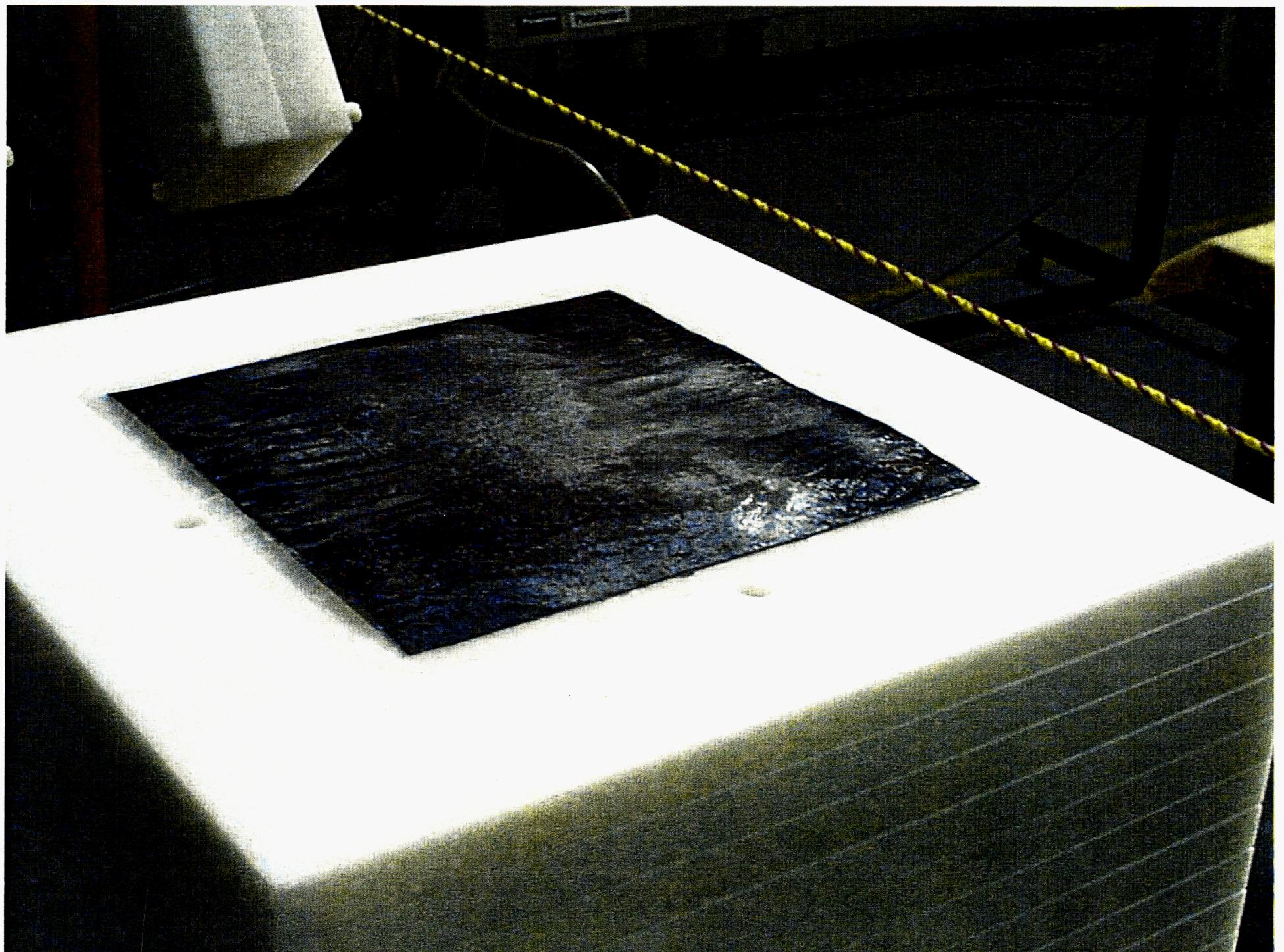
- 1990-1994 B. Briggs and associates performed calculations for TRU waste in storage at the RWMC
- 1995- C. Bowman “Underground Autocatalytic Criticality from Plutonium and Other Fissile Material”
- 1996- 1997 LACEF- Criticality of Mixures of Plutonium, Silicon Dioxide, and Water - Dynamic Behavior

Purpose

- The purpose of these experiments is to test the cross sections
 - » untested materials (SiO_2 , MgO , Al_2O_3 , CaO)
 - » untested regions
- No experimental critical data that can assure the correctness of calculations

Reflected Critical Mass of Plutonium Metal Mixed With SiO_2 and Water vs Plutonium Density





Weight and I. C. of the Foils

Foil #1 68 g

Foil #11 70 g

U-235 93.23 w%

Foil #2 71 g

Foil #14 66 g

U-238 5.37 w%

Foil #3 71 g

Foil #18 69 g

U-236 0.26 w%

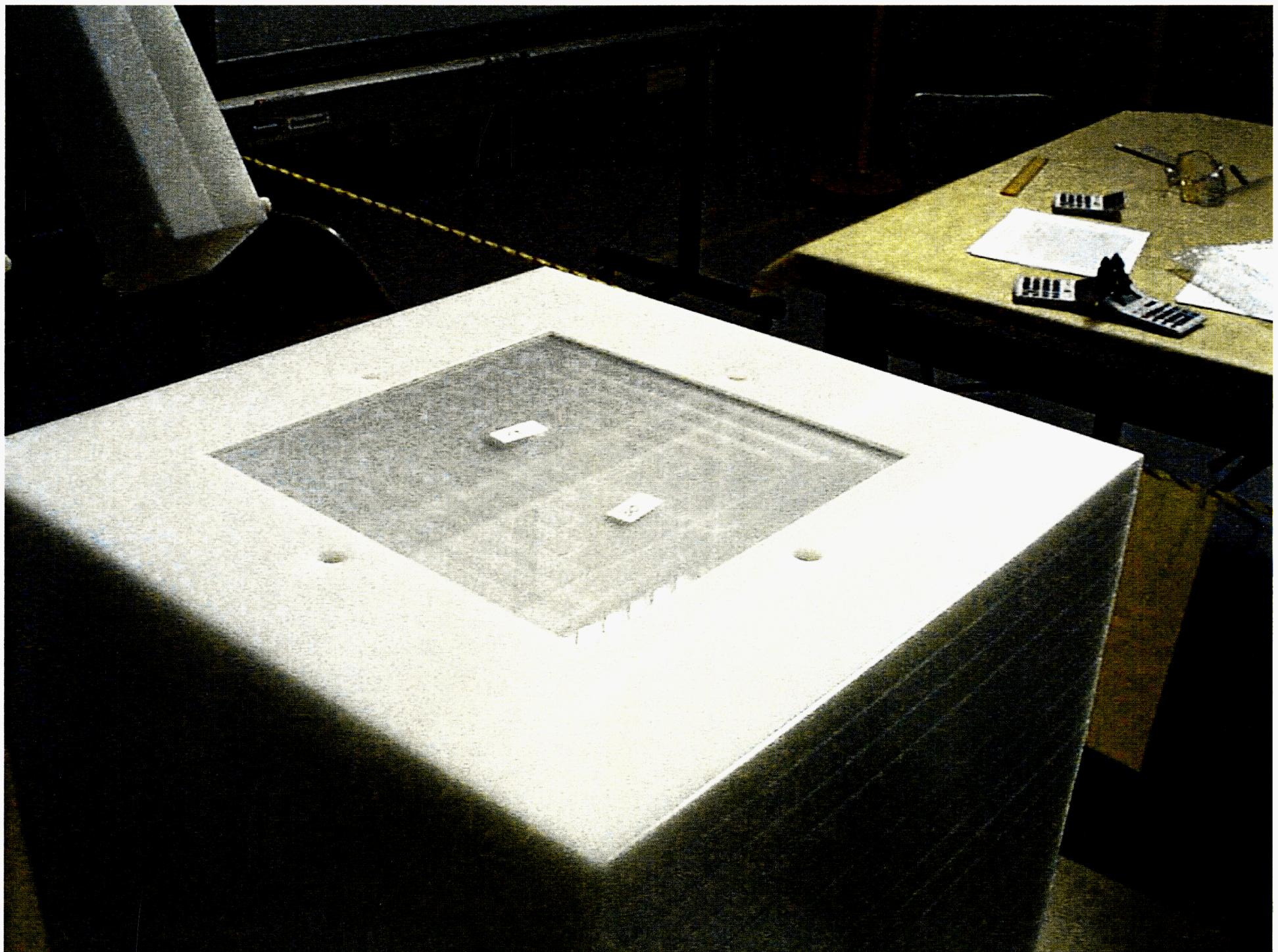
Foil #10 70 g

Foil #25 72 g

U-234 1.13 w%

Foil #15 71 g

Foil #26 69 g



SiO_2 Plates

Low neutron absorbing material

Dimensions: 9"x9"x0.25" Weight: 753.3 g

Density = 2.2 g/cc

Chemical Compositon:

SiO_2 99.997 w%

Al 14 ppm

As <0.002 ppm

B <0.2 ppm

Ca 0.4 ppm

Cd <0.01 ppm

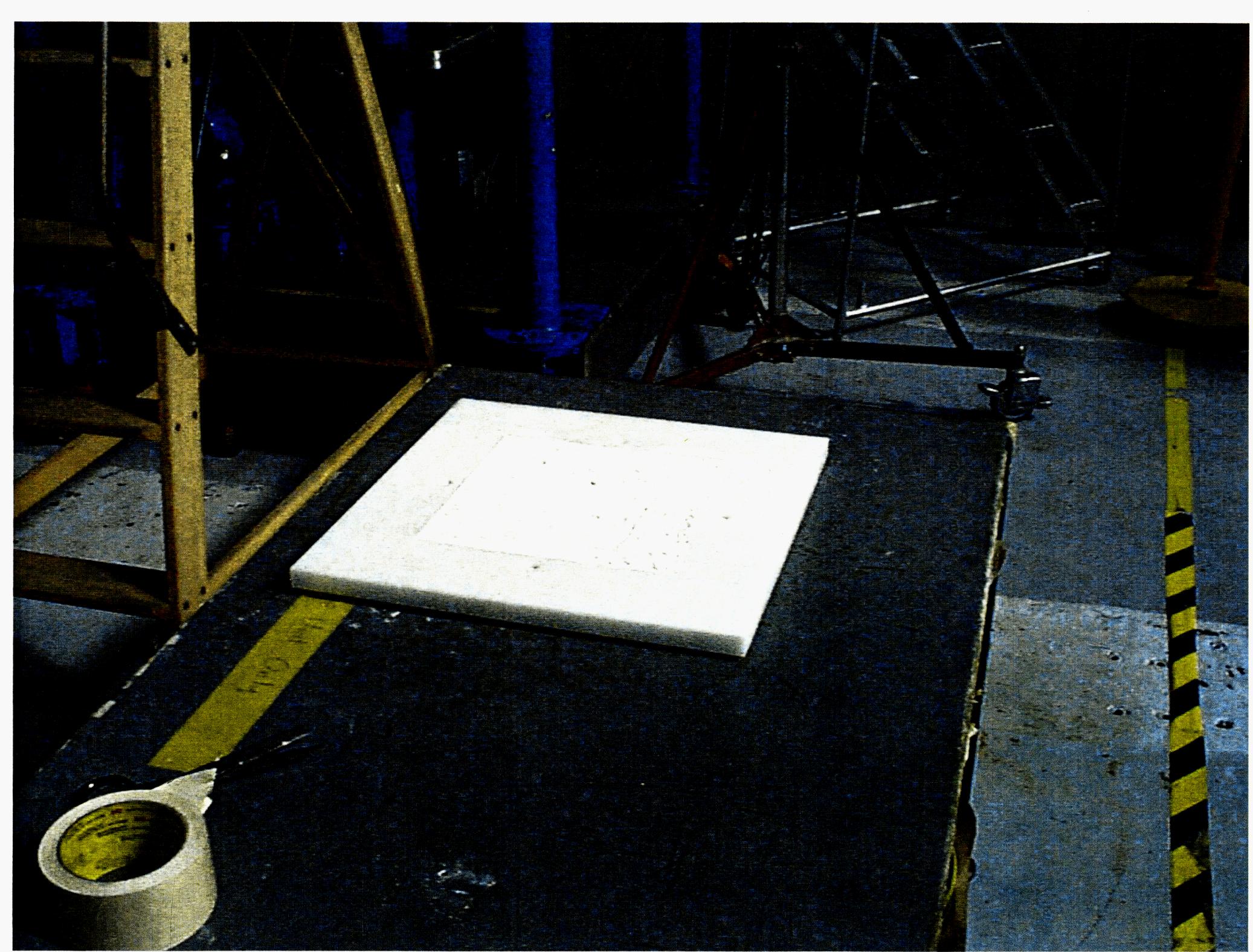
Cr <0.05 ppm

Fe 0.2 ppm

Mg 0.1 ppm

Li 0.6 ppm





MgO Powder

Density = 1.17-1.36 g/cc

Element Composition (in Percent)

MgO 97-98.2

SiO₂ 0.4-0.5

Al₂O₃ 0.1-0.2

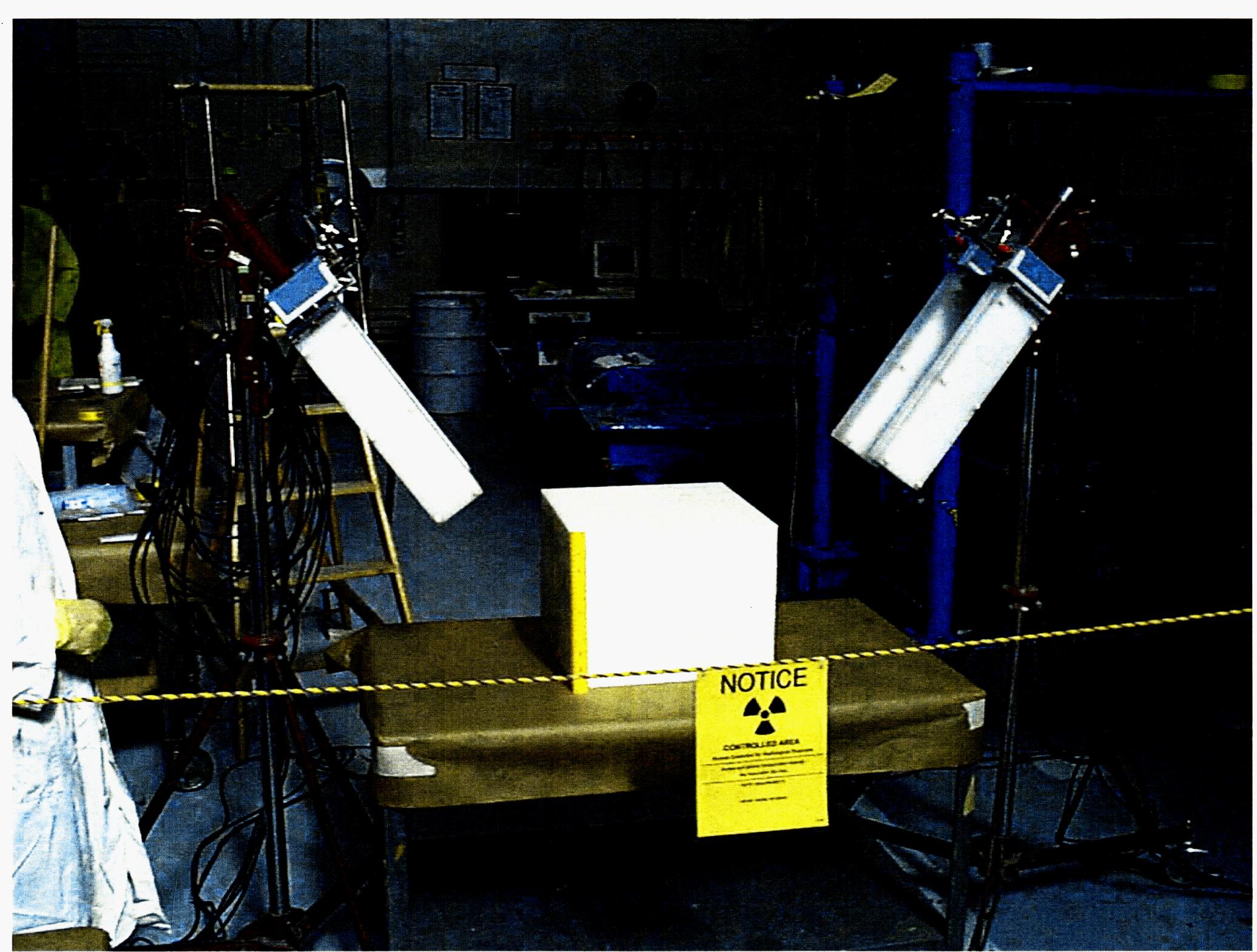
SO₃ 0.01-0.02

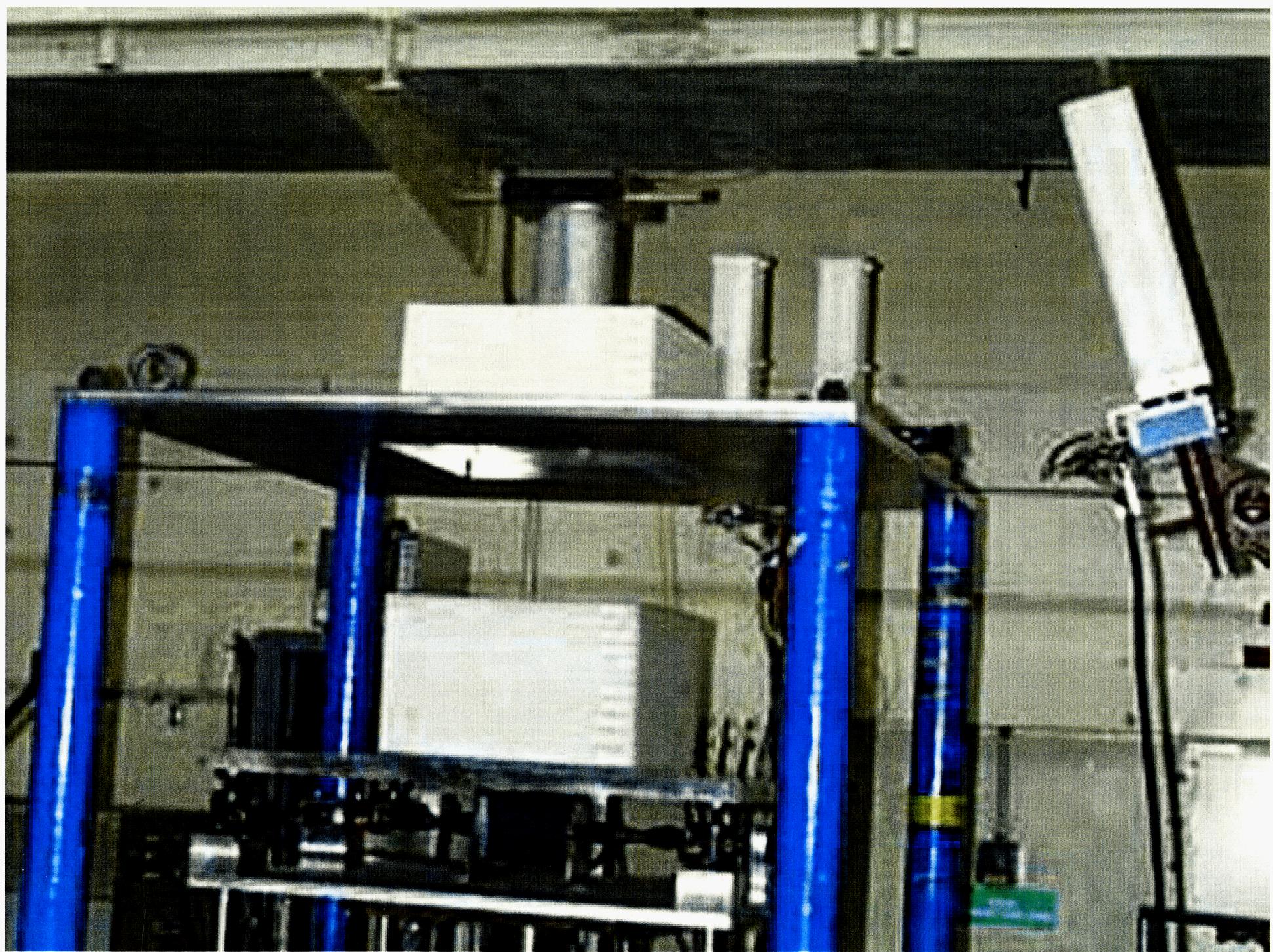
Loss on Ignition 0.25-0.5

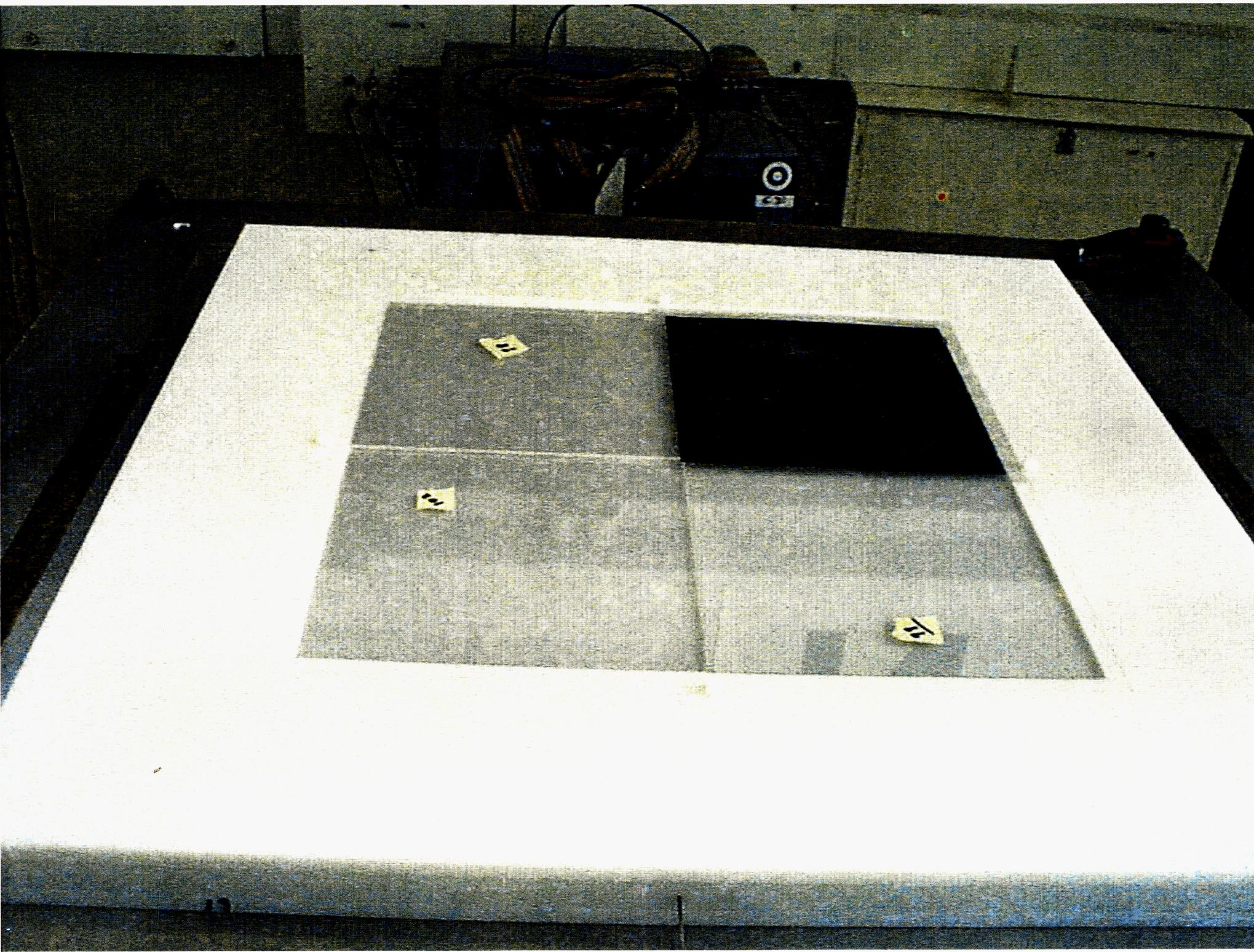
CaO 0.9-1.0

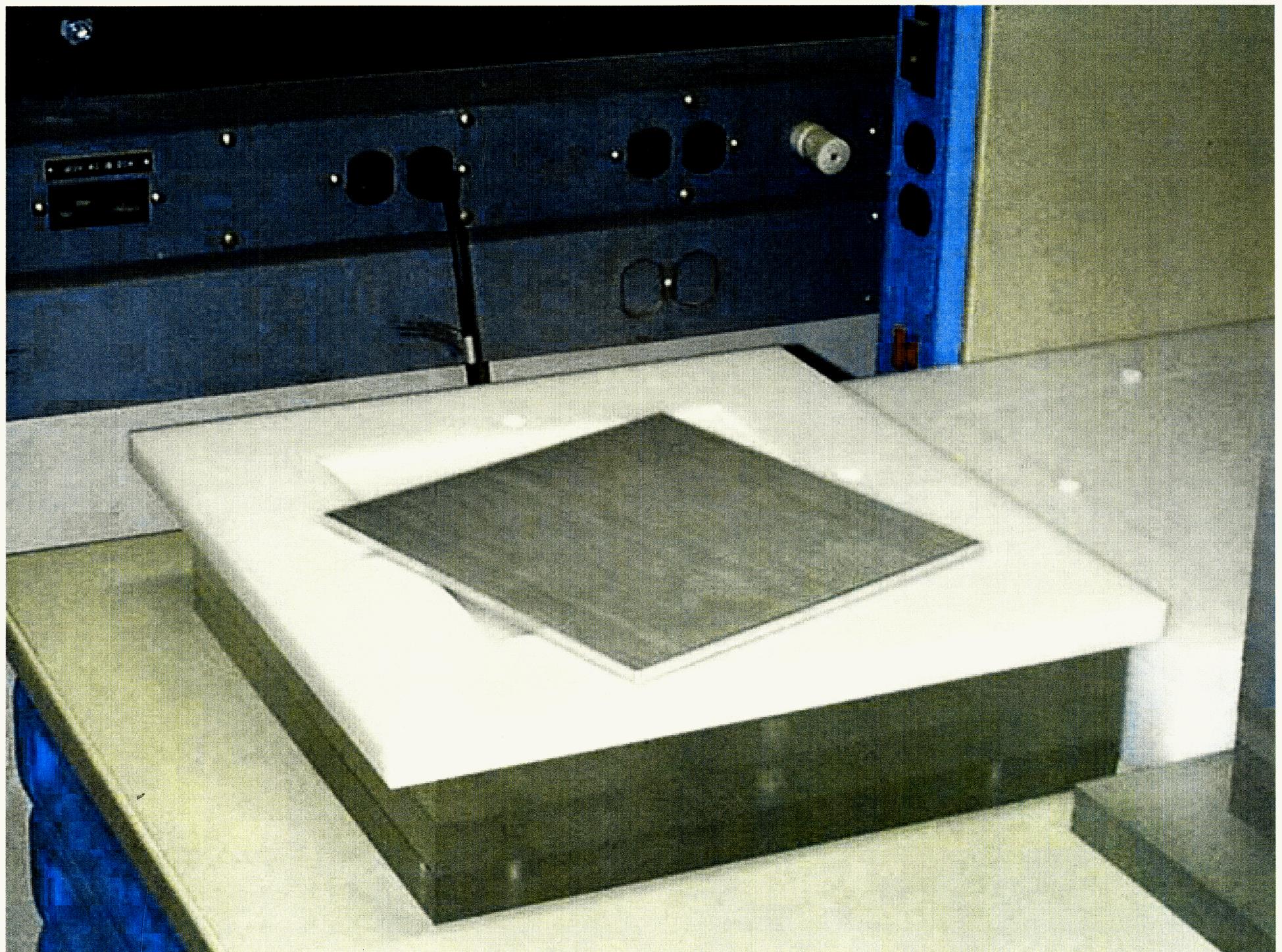
Fe₂O₃ 0.2-0.3

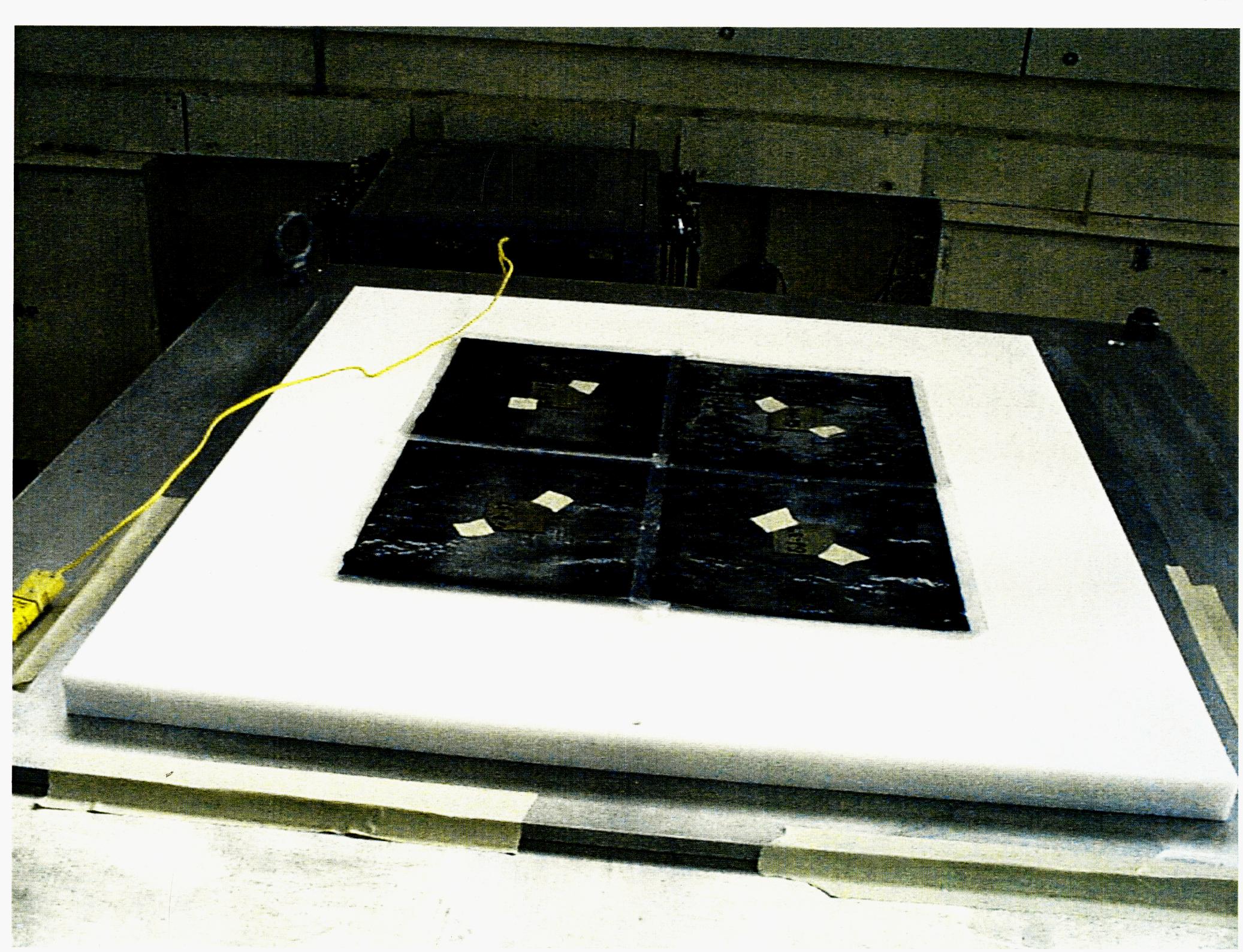
Cl 0.01-0.02











Experimental Results

| Diluent | X/ ^{235}U H/ ^{235}U | Mass of Uranium | Extrapolated Critical Mass |
|----------------------------|----------------------------------------------------|--------------------|-------------------------------|
| SiO_2/CH_2 | Si/ $^{235}\text{U}=42$ H/ $^{235}\text{U}=312$ | 2196 g | 2878 g |
| | Si/ $^{235}\text{U}=21$ H/ $^{235}\text{U}=156$ | 2233 g | 2285 g |

Experimental Results (Cont.)

| Diluent | X/ ^{235}U H/ ^{235}U | Mass of Uranium | Extrapolated Critical Mass |
|----------------------|--------------------------------------------------------|--------------------|-------------------------------|
| Al/CH ₂ | Al/ ^{235}U = 60 H/ ^{235}U = 159 | 2604 g | 2609 g |
| MgO/ CH ₂ | Mg/ ^{235}U = 18 H/ ^{235}U = 160 | 2878 g | 2951 g |

KENO-Va Results

| | Pu-239 concentration | Kinf |
|----------------|----------------------|---------|
| SiO_2 | 0.20 g Pu-239/lb | 1.00844 |
| | 0.25 | 1.11525 |
| MgO | 0.20 | 1.23570 |
| | 0.25 | 1.32796 |
| Al | 0.50 | 0.72271 |
| | 0.75 | 0.86322 |

Table III. Critical masses and parameters of uranium diluted systems.

| Final measured configuration | | | Critical configuration | | | keff |
|---------------------------------|---------------------------------------------------------|---------------------|-----------------------------------|-------------------------------|---------------------------------------------------------------------------|----------------------------------------------------------------|
| Diluent | $X/^{235}\text{U}$, $H/^{235}\text{U}$ | Mass of Uranium (g) | Extrapolated critical mass g of U | Core average density (g/cc) U | Core average density (g/cc) diluent | |
| SiO ₂ / polyethylene | Si/ ²³⁵ U = 42, H/ ²³⁵ U = 312 | 2196 | 2878 | 0.071 | For SiO ₂ density = 0.74 For CH ₂ density = 0.64 | |
| SiO ₂ / polyethylene | Si/ ²³⁵ U = 21, H/ ²³⁵ U = 156 | 2233.1 | 2285.3 | 0.149 | For SiO ₂ density = 0.77 For CH ₂ density = 0.66 | 1.002 Exp 1.0070 Cal ± 0.0015 ENDF/B-V ⁵ |
| Al/ polyethylene | Al/ ²³⁵ U = 60, H/ ²³⁵ U = 159 | 2604 | 2609.1 | 0.145 | For Al density = 0.93 For CH ₂ density = 0.66 | 1.001 Exp 1.0016 Cal ± 0.0004 ENDF/B-VI ⁶ |
| MgO/ Polyethylene | Mg/ ²³⁵ U = 18, H/ ²³⁵ U = 160 | 2878 | 2950.8 | 0.145 | For MgO density = 0.42 For CH ₂ density = 0.64 | 1.0009 Exp 1.0435 Cal ± 0.0004 ENDF/B-VI ⁷ |
| Gd/ Polyethylene | Gd/ ²³⁵ U = 0.09, H/ ²³⁵ U = 230 | 1951.3 | 1958.3 | 0.146 | For Gd density = 0.008 For CH ₂ density = 0.96 | 0.9976 Exp 1.00149 Cal ± 0.0005 ENDF/B-VI |
| Gd/ Polyethylene | Gd/ ²³⁵ U = 0.046, H/ ²³⁵ U = 228 | 1811.0 | 1893.6 | 0.15 | For Gd density = 0.004 For CH ₂ density = 0.96 | 1.0025 Exp 1.00602 Cal ± 0.0005 ENDF/B-VI |
| Gd/ Polyethylene | Gd/ ²³⁵ U = 0.18, H/ ²³⁵ U = 245 | 2801.0 | 2872.8 | 0.141 | For Gd density = 0.016 For CH ₂ density = 0.96 | 0.9905 Exp 1.00354 Cal ± 0.0005 ENDF/B-VI |
| Fe/ Polyethylene | Fe/ ²³⁵ U = 4.51, H/ ²³⁵ U = 224 | 1388 | 1408.9 | 0.154 | For Fe density = 0.15 For CH ₂ density = 0.96 | 0.9973 Exp 1.0096 Cal ± 0.0003 ENDF/B-VI ⁷ |
| Polyethylene | H/ ²³⁵ U = 220 | 1251.0 | 1301.4 | 0.157 | For CH ₂ density = 0.96 | 1.0050 Exp 1.0146 Cal ± 0.0005 ENDF/B-VI |
| Polyethylene | H/ ²³⁵ U = 471 | 1319.0 | 1331.6 | 0.073 | For CH ₂ density = 0.96 | 0.9969 Exp 0.9909 Cal ± 0.0004 ENDF/B-VI |
| Polyethylene | H/ ²³⁵ U = 239 | 2233.1 | 2454.3 | 0.145 | For CH ₂ density = 0.96 | 1.0079 Exp 1.0137 Cal ± 0.0005 ENDF/B-VI |

Conclusions

- MCNP calculations with matrix materials agreed quite well with experimental results.
- MCNP calculations with polyethylene yielded a bit higher k_{eff} .
- Future experiments will include other materials (CaO, concrete, Fe, etc.) in contact with HEU, ^{233}U , ^{239}Pu .