

DIRECT MEASUREMENT OF MERCURY REACTIONS IN COAL POWER PLANT PLUMES

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ABSTRACT

This project was awarded under U.S. Department of Energy (DOE) National Energy Technology Laboratory (NETL) Program Solicitation DE-PS26-02NT41422 and specifically addresses Program Area of Interest: #5 – Environmental and Water Resources. The project team includes the Electric Power Research Institute (EPRI) as the contractor and the University of North Dakota Energy & Environmental Research Center (EERC) and Frontier Geosciences as subcontractors. Wisconsin Energies and its Pleasant Prairie Power Plant acted as host for the field-testing portion of the research.

The project is aimed at clarifying the role, rates, and end results of chemical transformations that may occur to mercury that has been emitted from elevated stacks of coal-fired electric power plants. Mercury emitted from power plants emerges in either its elemental, divalent, or particulate-bound form. Deposition of the divalent form is more likely to occur closer to the source than that of the other two forms, due to its solubility in water. Thus, if chemical transformations occur in the stack emissions plume, measurements in the stack may mischaracterize the fate of the material. Initial field and pilot plant measurements have shown significant and rapid chemical reduction of divalent to elemental mercury may occur in these plumes.

Mercury models currently assume that the chemical form of mercury occurring in stacks is the same as that which enters the free atmosphere, with no alteration occurring in the emissions plume. Recent data indicate otherwise, but need to be evaluated at full operating scale under field conditions.

Prestbo and others have demonstrated the likelihood of significant mercury chemical reactions occurring in power plant plumes (Prestbo et al., 1999; MDNR-PPRP, 2000; EERC, 2001). This experiment will thus increase our understanding of mercury atmospheric chemistry, allowing informed decisions regarding source attribution.

The experiment was carried out during the period August 22-September 5, 2003. The experimental site was the Pleasant Prairie Power Plant in Pleasant Prairie, Wisconsin, just west of Kenosha. The experiment involved using an aircraft to capture emissions and document chemistry changes in the plume. While using the airplane for sampling, supplemental fast-response sensors for NO_x, connected to data loggers, were used to gauge entry and exit times and transect intervals through plume emissions material. The Frontier Geosciences Static Plume Dilution Chamber (SPDC) was employed simultaneously adjacent to the stack to correlate its findings with the aircraft sampling, as well as providing evaluation of the SPDC as a rapid, less costly sampler for mercury chemistry. A complementary stack plume method, the Dynamic Plume Dilution (DPD) was used in the latter portion of the experiment to measure mercury speciation to observe any mercury reduction reaction with respect to both the reaction time (5 to 30 seconds) and dilution ratio. In addition, stack sampling using the

“Ontario Hydro” wet chemistry method and continuous mercury monitors (CMM) were used to establish the baseline chemistry in the stack. Comparisons among stack, SPDC, DPD and aircraft measurements allow establishment of whether significant chemical changes to mercury occur in the plume, and of the verisimilitude of the SPDC and DPD methods.

This progress report summarizes activities during a period of results review from the stack/aircraft subcontractor, data analysis and synthesis, and preparation and presentation of preliminary results to technical and oversight meetings.

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DIRECT MEASUREMENT OF MERCURY REACTIONS IN COAL POWER PLANT PLUMES

1) ACTIVITIES DURING THE REPORTING PERIOD

Papers and Presentations

Subcontractor University of North Dakota Energy and Environmental Research Center (UNDEERC) served as primary author and presenter on a summary paper describing the Pleasant Prairie measurements at the 2004 EPRI/U.S. DOE/U.S. EPA Megasyposium, Washington, D.C. (see Appendix A).

A presentation on the work was made by Subcontractor UNDEERC at the 7th International Conference on Mercury as a Global Pollutant, Ljubljana, Slovenia, June-July 2004.

Principal Investigator Leonard Levin presented summary results and interpretation of results for source-receptor relationships to conferences, including:

- University of Heidelberg Institute for Environmental Geochemistry, August 2004
- EPRI Environment Division Advisors, Boston, September 2004
- United Nations Global Environmental Inventory Activity workshop, Paris, France, June 2004
- United Nations Environment Program Regional Mercury workshop, Bangkok, Thailand, April 2004

Subcontractor Reporting

Laboratory analyses and data processing proceeded at the laboratories of Subcontractor UNDEERC

Repeated phone and e-mail communications with Subcontractor Frontier Geosciences (FG) were engaged in by Principal Investigator EPRI. Subcontractor FG indicated data processing was underway while simultaneously preparing summary papers for presentations at the 7th International Conference on Mercury as a Global Pollutant and elsewhere.

2) APPENDIX A

PROJECT PRESENTATION AT 2004 MEGASYMPOSIUM

Evaluation of Mercury Speciation in a Power Plant Plume

Paper #167

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ABSTRACT

There is no known benign tracer material that mimics the complex behavior of mercury (Hg) in the natural environment. For that reason, a combination of source and receptor measurements and computer modeling are required for tracking Hg cycling. Although in-stack Hg speciation measurements are essential for the development of control technologies and to provide source data for atmospheric deposition models, Hg transformations that may occur in the emissions plume will also determine the rate and the form of Hg deposited in waterways. Yet such processes are poorly understood and not currently incorporated in atmospheric models of Hg. Therefore, EPRI and the U.S Department of Energy have funded surface and aircraft studies of stack emissions and plume chemistry at two power plants. The first study was at Plant Bowen, Cartersville, Georgia, operated by Georgia Power, part of the Southern Company. Aircraft studies were conducted by the Tennessee Valley Authority, with surface measurements by the Energy & Environmental Research Center (EERC) and by Frontier Geosciences. The second study was at the Pleasant Prairie Power Plant, Pleasant Prairie, Wisconsin, operated by We Energies; there, surface and air measurements were carried out by the EERC, with additional studies at the stack by Frontier. The primary focus of this paper is on work conducted by the EERC at Pleasant Prairie.

The overall project goal was to gain an understanding of Hg chemistry as a plume is transported downwind from the stack. This was to be carried out by flying an airplane through the plume at several locations (a point nearest the stack, 5 miles from the stack, and 10 miles from the stack) and measuring the speciated Hg composition in the plume at each location. The speciated Hg results obtained at the stack were then compared to the measurements obtained

from the plume sampling, using dilution factors to accommodate turbulent dispersion and allow comparison on a common basis.

In general, using a dilution factor based on the plume and stack NO_x , a reasonable Hg mass balance was obtained when comparing the Hg in the stack to the Hg in the plume. This indicates there was little if any loss of Hg from the plume material between the stack and the 10-mile sample point. However, there appeared to be a lower fraction of reactive gaseous mercury (RGM) in the plume compared to the RGM value in the stack, with a corresponding increase in the proportion of elemental mercury (Hg^0). The percentage of Hg^0 in the stack was 66% compared to values of 84, 89%, and 88% Hg^0 in the plume measured closest to the stack, at 5 miles from the stack, and at 10 miles from the stack, respectively.

INTRODUCTION

Characterization of environmental mercury (Hg) and its atmospheric processes, from emissions to deposition, requires both measurements and model simulations for a full understanding. Although in-stack Hg speciation measurements are essential to the development of control technologies and to provide data for input into the atmospheric deposition models, the determination of speciation and changes in speciation in emission plumes themselves are required for a complete description of atmospheric Hg. In particular, the rapid temperature changes, mixing with ambient air, and presence of other emitted constituents may impact Hg chemistry over short time and spatial scales. While substantial research has been done in the past on Hg transformations with energy production facilities—determining the concentrations of speciated Hg within combustion and exhaust gas systems—little has been done to determine the Hg chemistry, kinetics, and thermodynamics in the stack emissions plume.¹ It is the Hg transformations that occur in the plume that determine the rate and the form of Hg entering the ambient atmosphere after complete mixing of the plume and that partially determines the amount of Hg deposited in lakes and streams downwind. Therefore, a logical step in Hg research is to apply known methods and data from in-facility measurements, and from measurements in the ambient atmosphere, to the near-source region through which the plume passes.

Therefore, EPRI and the U.S. Department of Energy (DOE) have funded surface and aircraft studies of stack emissions and plume chemistry at two coal-fired power plants. The first study was at Plant Bowen, Cartersville, Georgia (operated by Georgia Power, part of the Southern Company). Aircraft studies were conducted by the Tennessee Valley Authority (TVA), with surface measurements by the Energy & Environmental Research Center (EERC) and by Frontier Geosciences. The second study was at the Pleasant Prairie Power Plant, Pleasant Prairie, Wisconsin, operated by We Energies; there, surface and air measurements were carried out by the EERC, with additional studies at the stack by Frontier. The primary focus of this paper is on work conducted by the EERC at Pleasant Prairie. However, this work depends on the earlier source studies conducted by TVA^{2,3} and the static plume dilution chamber work done by Frontier Geosciences.^{1,4} Additionally, ambient surface measurements made by Atmosphere Research & Analysis⁵ at the Yorkville SEARCH site some 25 km from Plant Bowen indicating that there is change in Hg speciation within the plume must be considered as well.

PROJECT OBJECTIVES

The overall project goal is to gain an understanding of Hg chemistry as a plume is transported downwind from the stack. Specific objectives are to:

- Develop sampling techniques to measure speciated Hg in the plume.
- Develop techniques to determine the location of the plume at various points downwind of the stack.
- Determine the speciation of Hg emissions at the stack and compare these results to Hg species obtained from the plume sampling.
- Compute a Hg mass balance using dilution factors and other relevant parameters.

PROJECT DESCRIPTION

Power Plant Description

The Pleasant Prairie Power Plant, owned and operated by We Energies, is located in the village of Pleasant Prairie just west of Kenosha, Wisconsin. The Pleasant Prairie plant consists of two units (Units 1 and 2) identical in operation with the exception of one (Unit 2) having a selective catalytic reduction (SCR) system at the time of the measurements. Each unit has an electrostatic precipitator (ESP) for particulate control. The two Pleasant Prairie units share a common single stack. Specifications of the Pleasant Prairie facility are as follows:

- Fuel type: Powder River Basin (PRB) subbituminous coal
- Boiler capacity: 617 MW (each unit)
- Boiler type: opposed-fired pulverized coal (both units)
- NO_x control: SCR on Unit 2; low-NO_x burners on both units
- SO₂ control: none; combustion of low-sulfur coal (both units)
- Particulate control: ESP (both units)

The coal is fairly typical of PRB in that both the Hg and chlorine levels in the coal are comparatively low; the Hg averaged 0.041 μg/g and the chlorine 10 ppm.

Aircraft and Equipment Used for the Project

The aircraft used for the emissions plume sampling was a turboprop DHC-6-300 Twin Otter deHavilland Vistaliner. Aircraft and crew were supplied by Twin Otter International, Las Vegas, Nevada (the same aircraft was used for the earlier Plant Bowen project). The pilots were very experienced in doing research involving ambient air sampling. The aircraft is shown in Figure 1. This type of aircraft had a number of design and performance characteristics that made it an ideal platform for the project. The twin-engine plane had a relatively large capacity and sufficient on-board electrical supply and could be operated efficiently at low altitudes. These capabilities were accompanied by relatively low fuel consumption at all altitudes. Most importantly for the plume study project, it could be flown at relatively slow speeds (80–160

knots/150–300 km/hr) and in tight formations, to allow consistent plume traverses at fixed heights and downwind distances.

The Hg analyzer used in the aircraft (shown in Figure 2) was a fully automated Tekran[®] Model 2537A Mercury Vapor Analyzer coupled with a Tekran[®] Model 1130 Mercury Speciation Unit and a Tekran[®] Model 1135 Particulate Module. The analyzer portion of the system is based on

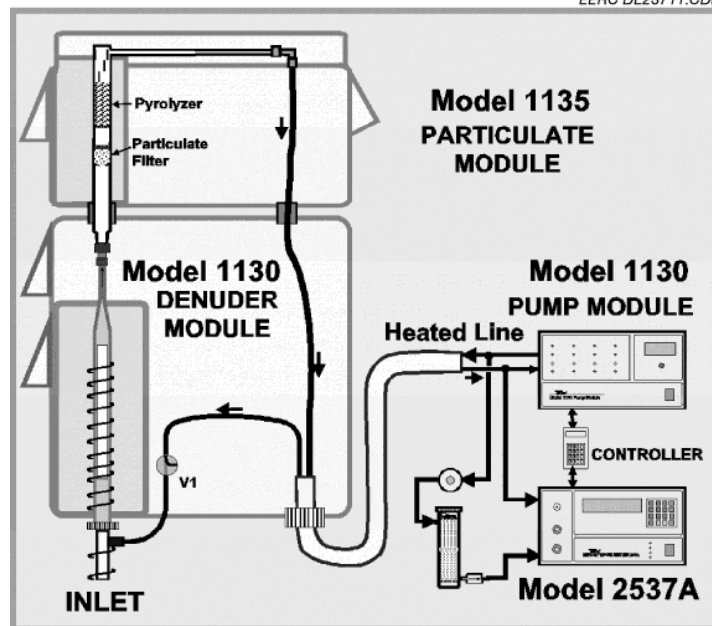
Figure 1. Photograph of the DHC-6-300 Twin Otter deHaviland Vistaliner

EERC DL23709.CDR



Figure 2. Schematic of the Tekran Mercury System

EERC DL23711.CDR



the principle of atomic fluorescence and has detection limits $<1 \text{ pg/m}^3$. With this system we were able to simultaneously measure elemental mercury (Hg^0), reactive gaseous mercury (RGM), and particulate-bound Hg species during the flight. Therefore, it was possible to rapidly evaluate data each day and make sampling decisions for succeeding flights.

To compare the Hg concentrations in the plume to those in the stack, a dilution factor was determined. Based on the earlier tests conducted at Plant Bowen by TVA,² it was decided that

the NO_x concentration in the stack vs. that in the plume at traverse distances be used to determine the dilution factor as shown in the following equation:

$$\text{Dilution Ratio} = \frac{(\text{Stack NO}_x - \text{Background NO}_x)}{(\text{Plume NO}_x - \text{Background NO}_x)}$$

In addition, the NO_x analyzer was used to determine the location of the plume; the rapid response of the NO_x analyzer allowed recording of NO_x concentration changes as the aircraft crossed the edges of plume material. The assumption of neutral buoyancy and equivalent Froude numbers for all plume constituents at and beyond equivalent stack height and distance was used to calculate in-plume portions of the Hg concentrations. The NO_x analyzer that was used for this project was a dual range Model 42C Thermo Electron Ambient Air NO_x Analyzer. This analyzer was able to measure 0–50 ppb NO_x in its low range and up to 500 ppb NO_x in its high range.

EXPERIMENTAL APPROACH

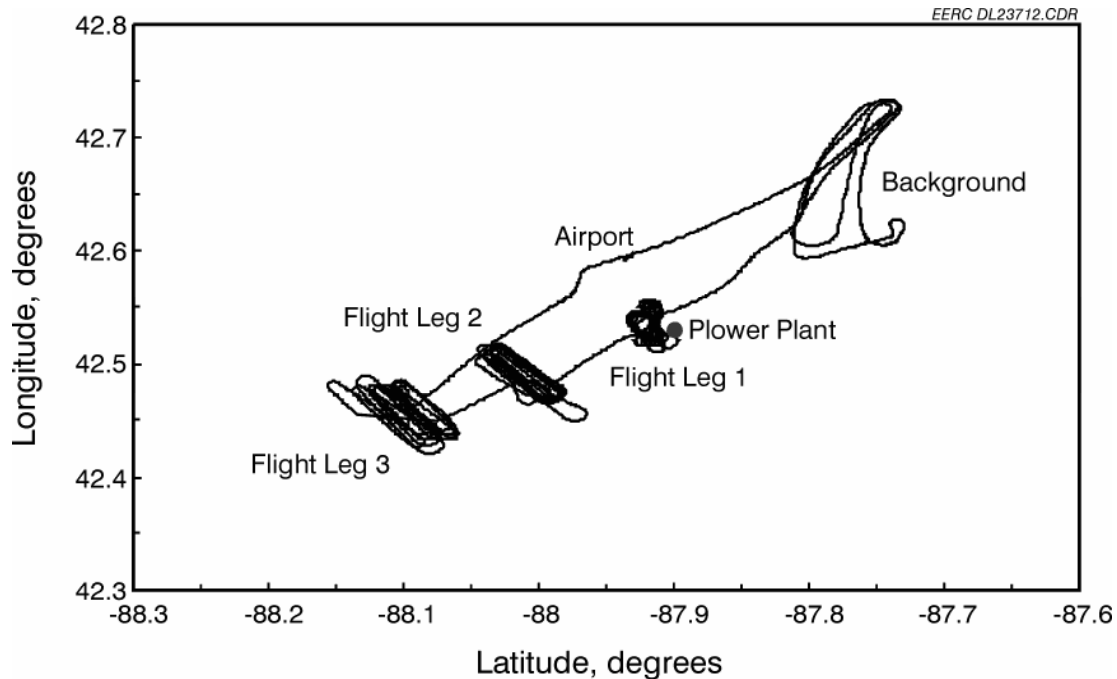
Aircraft Sampling Locations

Following background sampling at a location upwind of the stack, aircraft sampling was done at three locations downwind of the plume. The first location was approximately 1500 ft from the stack (the closest the plane was permitted to approach the stack). The second and third locations were approximately 5 and 10 miles downwind of the stack, respectively. Figure 3 is an example of the flight diagram showing sampling locations and patterns. As can be seen, a number of different patterns were flown at the closest sampling point in an effort to optimize the time in the plume. At the other two locations a race track pattern was flown. Aircraft position vs. time was determined using a satellite-based global positioning system.

Determining the altitude and direction of the plume was done based on a combination of visual inspection (location closest to the stack, at which plume material was evident visually) and the measurements of NO_x concentrations. Except for the closest location, the Tekran Hg analyzer was triggered by a NO_x set point.

Aircraft sampling flights were conducted in daylight under atmospheric conditions that permitted adequate sampling of the stack plume. The weather conditions during the project were fairly consistent with the historical climate record with many fair weather days. In the original test plan the ideal wind speeds were considered to be approximately 2.5–5 m/s; however, the average wind speeds were somewhat above this, ranging from 5.3 to 8.3 m/s. These windspeeds were still low enough that the aircraft was able to find the plume and good sampling was able to be completed.

Figure 3. Example of a Flight Pattern



Stack Sampling

To establish baseline conditions for comparison with the plume samples, in-stack sampling was carried out simultaneously with the flights, providing measurements of the Hg speciation in the stack. Hg sampling at the stack was completed using the Ontario Hydro (OH) Hg speciation sampling method and a continuous mercury monitor (CMM). Prior to the start of the testing, a CMM was placed at the stack outlet and remained there during the entire project. During each flight day, one OH sample was taken at the stack.

In addition to Hg measurements by the EERC, the plant operators continuously measured the NO_x concentration in the stack using a continuous emission monitor. Although the NO_x concentrations in the stack at the Pleasant Prairie Power Plant were somewhat lower than desired for aircraft plume measurements (since half the flue gas was being treated using SCR), it was still possible to detect a difference between the background and plume, even at 10 miles. The average stack NO_x concentration was 144 ppm(v).

RESULTS AND DISCUSSION

Stack Results at Pleasant Prairie

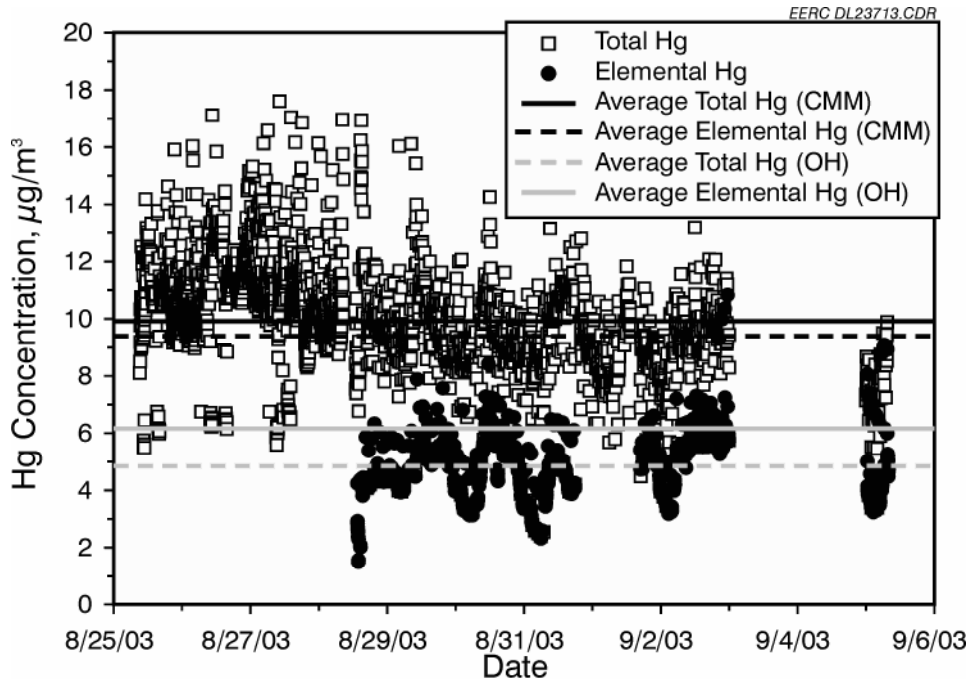
The Hg concentrations in the stack are shown in Table 1 and the OH and CMM sampling in Figure 4. As the results show in Table 1, the Hg concentrations were relatively constant during the entire duration of the project. The RGM averaged 34.3% of the total Hg measured. These results were well-matched by the CMM, as shown in Figure 4.

Table 1. Stack Ontario Hydro mercury speciation results.*

Date	Particulate-Bound Hg, $\mu\text{g}/\text{Nm}^3$	RGM, $\mu\text{g}/\text{Nm}^3$	Hg ⁰ , $\mu\text{g}/\text{Nm}^3$	Hg (Total), $\mu\text{g}/\text{Nm}^3$	Hg ⁰ as a % of Total Hg
08/27/03	0.00	3.3	6.7	10.0	67.0
08/29/03	0.00	3.4	6.1	9.5	64.2
08/30/03	0.00	3.0	6.6	9.6	68.8
08/31/03	0.00	4.0	5.1	9.1	56.0
09/02/03	0.00	2.5	5.7	8.2	69.5
09/04/03	0.00	3.1	6.8	9.9	68.7

*All concentrations (Nm^3) are based on 1 atmosphere pressure and 20°C.

Figure 4. Stack Mercury Emissions at the Pleasant Prairie Power Plant (actual O₂ concentrations)



Plume Results at Pleasant Prairie

A summary of the results from the plume sampling at the Pleasant Prairie plant are presented in Table 2. The plume Hg concentrations have been converted to equivalent stack concentrations and are presented as such. The plume data were reduced by first correcting for the amount of sampling time in the plume. This gives a Hg concentration in the plume. The plume concentrations were then converted to an equivalent stack concentration by applying dilution factors which were based on the plume and stack NO_x concentrations.

A mass balance was calculated using the average total Hg data from each sampling point each day and the corresponding stack data. The results are presented in Table 3. Although many of the

Table 2. Average Mercury speciation results for each location.*

	Hg _p , µg/Nm ³	Hg ⁰ , µg/Nm ³	RGM, µg/Nm ³	Total Hg, µg/Nm ³	% Hg ⁰
Stack					
Average	0.00	3.2	6.2	9.4	65.7
Std. Dev.	0.00	0.5	0.7	0.7	5.1
0 Miles					
Average	0.06	10.1	2.0	12.1	83.6
Std. Dev.	0.04	4.9	0.5	3.4	5.3
5 Miles					
Average	0.10	13.5	1.7	15.2	88.9
Std. Dev.	0.06	10.0	0.6	5.1	3.1
10 Miles					
Average	0.09	12.8	1.6	14.4	88.2
Std. Dev.	0.08	7.5	0.7	5.4	2.2

*All concentrations are based on normal (N) conditions defined as 1 atmosphere pressure, 20°C, and 3% O₂.

Table 3. Total Hg mass balance: plume Hg compared to stack Hg.

0 Miles			5 Miles			10 Miles		
Total Hg in Plume, µg/Nm ³	Total Hg in Stack, µg/Nm ³	Ratio, %	Total Hg in Plume, µg/Nm ³	Total Hg in Stack, µg/Nm ³	Ratio, %	Total Hg in Plume, µg/Nm ³	Total Hg in Stack, µg/Nm ³	Ratio, %
13.5	10.0	135	15.0	10.0	150	16.5	10.0	165
12.1	9.7	125	14.3	9.7	148	17.8	9.7	184
10.7	9.1	118	22.0	9.1	242			
7.8	8.2	95	9.6	8.2	11	7.9	82	96
12.0	9.9	121						
18.2	9.9	184						

mass balance closures are somewhat high, they are quite reasonable considering the variability of the data and the difficult nature of the sampling. The two most likely causes for the high Hg concentrations are underreported NO_x concentration, which does not affect the ratio of Hg⁰ to RGM, or Hg offgassing from the soda lime traps used in the aircraft sampling line when exposed to plume gas. Nonetheless, the ratio of Hg⁰ to RGM does show an evolution as the plume moves downwind from the stack, implying a change in speciation occurring solely within the plume itself between the stack exit and full dispersion into the ambient atmosphere.

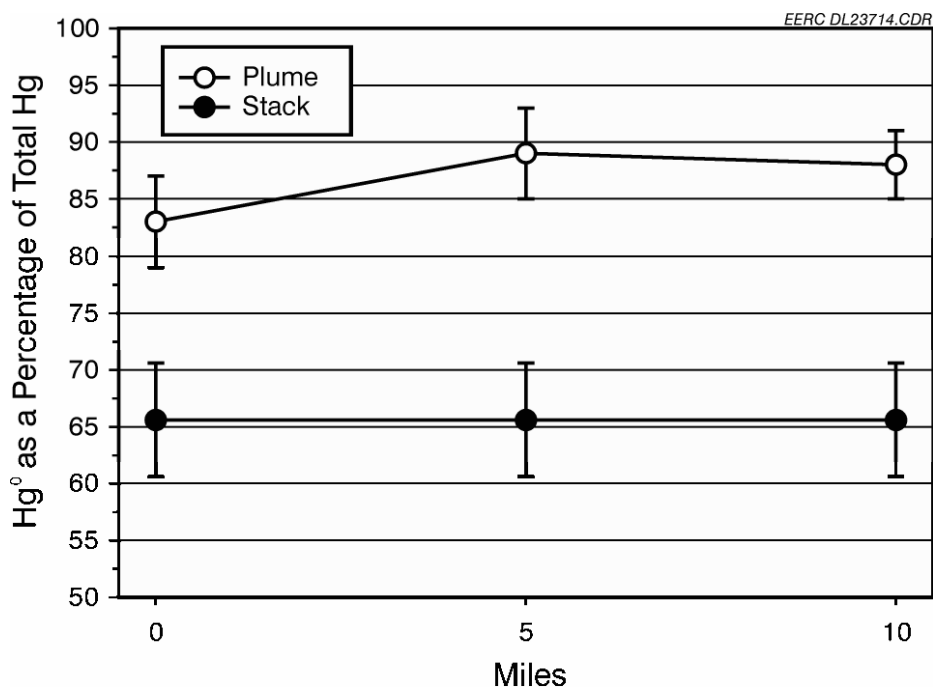
It should be noted there were a number of high readings that corresponded to very high dilution ratios. These results were not included in the average. The high dilution ratios occurred during intermittent aircraft transit of smaller plume eddies separated from the main body of the plume, with relatively high proportions of ambient air sampled in short times, not statistically characteristic of the plume material at that distance from the stack. Even with relatively stable winds, turbulent eddies would result in complex plume structure, causing some passes through the plume to only intersect edges of the material rather than the cross-centerline transects that were sought. These peripheral intersects would capture relatively large proportions of ambient air with markedly lower NO_x values than the main body of the plume.

During these sampling episodes, the NO_x concentration would be just high enough to trigger the zero air inlet valve in the sampling system. Once it was triggered, the valve was set with a delay to stay off for 5 seconds once the NO_x level again fell below the trigger point. This was done to prevent rapid cycling of the valve. Since the NO_x concentration was near ambient levels in these segments of the plume, the resulting dilution ratios were artificially high.

The overall results are shown in Figure 5, in a plot of the average percentage of Hg^0 in the plume as it moves from the stack exit to 10 miles downwind of the stack. There is a substantial increase in the concentration of Hg^0 from the stack to the first sample point, a smaller increase from the first to the second sample point, and then no change from there to the last sample location (10 miles). Overall, the Hg^0 increases from 67% of the total Hg to 89% by the time it reaches the 5-mile sample point. The question arises, is this increase real, or does it represent a high bias in the Hg^0 concentration measurements in the plume? Several factors could cause the Hg^0 concentration (and total Hg) to be reported higher in the plume than in the stack. These include the following:

- Low background Hg readings
- Inefficient trapping of the RGM in the Tekran denuder
- Underreported plume NO_x concentrations
- Problems with the soda lime traps

Figure 5. Changes in Mercury Speciation as a Function of Distance from the Stack



Low Background Mercury Readings

The subtraction of low background concentrations from the measured plume concentrations would result in higher in-plume concentrations when they are converted to equivalent stack concentrations. However, this does not appear to be the case, because the measured background concentrations were fairly consistent and in the expected range. Also, low background RGM concentrations would have caused those values to be higher as well, and this is not the case.

Inefficient Trapping of the RGM in the Annular Denuder

If the RGM were not trapped in the denuder but passed through to the analyzer, it would be reported as Hg^0 . However, if the RGM were to pass through the annular denuder, it would most likely be trapped on the particulate filter and be reported instead as particulate-bound Hg. The data for the particulate-bound Hg concentrations do not suggest this was the case.

Underreported Plume NO_x Concentrations

This would result in larger dilution ratios and higher equivalent stack concentrations, but it would affect both the Hg^0 and the RGM concentrations. However, if the plume NO_x concentrations were underreported, it would affect both the Hg^0 and RGM concentrations in the same sense, and this is clearly not the case.

Problems with the Soda Lime Traps

Modified soda lime traps were used to scrub acid gases from the sample gas stream before it reached the Hg analyzer. At the Bowen plant it was observed that Hg desorbed from these traps when they were exposed to ambient air rather than the scrubbed zero air. There did not appear to be any offgassing of Hg when sampling ambient air at the Pleasant Prairie plant, but a different type of soda lime trap (supplied by Frontier Geosciences) was used.

CONCLUSIONS

The following conclusions can be stated based on the Pleasant Prairie Power Plant stack and plume Hg sampling:

- Hg can be measured by aircraft in plumes with reasonable accuracy and precision. However, great care must be taken to prevent contamination in sampling lines used aboard the aircraft.
- Using a dilution factor based on the plume and stack NO_x , a reasonable Hg mass balance can be obtained to compare the Hg in the stack to the Hg in the plume.

- There appeared to be a decline in the fraction of RGM at Pleasant Prairie when comparing values in the plume to those in the stack (with a corresponding increase in the proportion of Hg⁰). There was a 38% reduction of RGM between the in-stack measurement and the first plume sampling point closest to the stack and a 47% reduction of RGM from the stack to the 5-mile sample point, with no additional reduction evident between the 5-mile and 10-mile locations. This is consistent with earlier measurements by Frontier Geosciences of Hg reduction reactions in its static plume dilution chamber, which indicated rapid reduction reactions after mixing of flue gas emissions with ambient air conditions representing the region just outside the stack exit.

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KEY WORDS

mercury, utilities, plume