

Oceanic CO<sub>2</sub> Uptake and Future Atmospheric CO<sub>2</sub> Concentrations †

CONF-9009164--2

DE90 017762

Tsung-Hung Peng

Environmental Sciences Division  
 Oak Ridge National Laboratory  
 P. O. Box 2008  
 Oak Ridge, Tennessee 37831-6335

## DISCLAIMER

This report was prepared as an account of work sponsored by an agency of the United States Government. Neither the United States Government nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.

"The submitted manuscript has been authored by a contractor of the U.S. Government under contract DE-AC05-84OR21400. Accordingly, the U.S. Government retains a nonexclusive, royalty-free license to publish or reproduce the published form of this contribution, or allow others to do so, for U.S. Government purposes."

† Research sponsored by Carbon Dioxide Research Program, Atmospheric and Climate Research Division, Office of Health and Environmental Research, U.S. Department of Energy, under contract DE-AC05-84OR21400 with Martin Marietta Energy Systems, Inc. Publication No. \_\_\_\_\_. Environmental Sciences Division, Oak Ridge National Laboratory.

MASTER

eB

## **DISCLAIMER**

**This report was prepared as an account of work sponsored by an agency of the United States Government. Neither the United States Government nor any agency thereof, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.**

---

## **DISCLAIMER**

**Portions of this document may be illegible in electronic image products. Images are produced from the best available original document.**

# Oceanic CO<sub>2</sub> Uptake and Future Atmospheric CO<sub>2</sub> Concentrations

Tsung-Hung Peng

**Abstract:** Models of the lateral transport of surface water in the Atlantic, Indian, and Pacific oceans are proposed on the basis of the distribution pattern of bomb-produced <sup>14</sup>C in the ocean as observed during the Geochemical Ocean Sections Study program (GEOSECS 1973-1978). The water column inventories of bomb-produced <sup>14</sup>C and the input function of the time history of bomb-produced <sup>14</sup>C in the atmosphere require that the mean ocean-wide CO<sub>2</sub> exchange flux should be  $22.3 \text{ mol} \cdot \text{m}^{-2} \cdot \text{yr}^{-1}$  in the Atlantic Ocean,  $10.4 \text{ mol} \cdot \text{m}^{-2} \cdot \text{yr}^{-1}$  in the Indian Ocean, and  $19.2 \text{ mol} \cdot \text{m}^{-2} \cdot \text{yr}^{-1}$  in the Pacific Ocean. The global pattern of water column inventories of bomb-produced <sup>14</sup>C suggests that an upwelling of bomb-<sup>14</sup>C-free water takes place in the Antarctic, northern Pacific, and equatorial regions, whereas a downwelling of bomb-<sup>14</sup>C-rich surface waters occurs in the temperate oceans and northern Atlantic. A global model of the closed ocean-atmosphere system is constructed on the basis of these dynamic flows for the purpose of estimating the oceanic uptake of excess CO<sub>2</sub>. To produce a close match between the historical pattern of atmospheric CO<sub>2</sub> predicted by the model and that observed in the Siple ice core and the Mauna Loa record requires that a net release of  $82 \times 10^{15} \text{ g}$  of carbon from the terrestrial ecosystem be posited for the period 1800-1980, in addition to fossil fuel CO<sub>2</sub> input. The model uses four scenarios of future CO<sub>2</sub> emissions from fossil fuel consumption and tropical deforestation to predict atmospheric CO<sub>2</sub> concentrations. Although the resulting patterns of predicted CO<sub>2</sub> levels are similar to those in the report given by Intergovernmental Panel on Climate Change (IPCC), the uptake of CO<sub>2</sub> from fossil fuel in the lateral transport model is more efficient than that in the general box-diffusion models used by IPCC: that is, all predictions from the model are lower than those given by IPCC.

## INTRODUCTION

Since the industrial revolution, the combustion of fossil fuels and deforestation have led to a 26% increase in the atmospheric concentration of CO<sub>2</sub>. There is concern that such an increase may change the climate of the globe by increasing the temperature of the Earth's surface, through so-called greenhouse warming. The amount of the temperature increase is related to the atmospheric concentration of greenhouse gases, of which carbon dioxide is a major constituent. To predict greenhouse warming, we need to know the fate of anthropogenic CO<sub>2</sub> after its release into the atmosphere. As a major carbon reservoir, the ocean will eventually absorb most of these human-made CO<sub>2</sub> gases. Although we know that about half of the emitted fossil fuel CO<sub>2</sub> stays in the atmosphere, we do not know well how much of the remainder is taken up by the world's oceans in the short term. Estimates of oceanic uptake of anthropogenic CO<sub>2</sub> and predictions of future atmospheric CO<sub>2</sub> concentrations need to be made by using models of world ocean circulation. Many ocean models have been constructed, and most of which are based on box-diffusion model of Oeschger et al. (1975). A more recent model, the lateral transport model proposed by Broecker et al. (1985), is calibrated from the distribution of bomb-produced <sup>14</sup>C in the oceans. Recently, a few three-dimensional ocean general circulation models have been adopted to study the carbon cycle in the ocean; however, they are not yet capable of predicting atmospheric CO<sub>2</sub> levels. In this paper, the lateral transport model of the ocean will be modified to compute the uptake of human-made CO<sub>2</sub> and to predict concentrations of atmospheric CO<sub>2</sub>.

## LATERAL TRANSPORT MODEL FOR CO<sub>2</sub> UPTAKE

Radiocarbon produced during the nuclear bomb tests in the 1950s and early 1960s is delivered from the atmosphere to the sea surface through gas exchange processes. It is subsequently transported to the deeper oceans through the dynamics of ocean mixing and

circulation. By studying the estimated delivery and the measured water column inventory of this bomb-produced  $^{14}\text{C}$ , we can calculate the mean  $\text{CO}_2$  gas exchange rates for three major ocean basins. Further more, we can place important constraints on the rates at which water is mixed vertically and transported from region to region.

Based on the history of the atmospheric  $\Delta^{14}\text{C}$  in excess of the prebomb value (as measured directly on  $\text{CO}_2$  extracted from the air) and the history of the surface ocean  $\Delta^{14}\text{C}$  excess, we can calculate the input of bomb  $^{14}\text{C}$  if the  $\text{CO}_2$  invasion rate for the region is known. Because reliable estimates for the regional variations in  $\text{CO}_2$  invasion rates are not available, a single average  $\text{CO}_2$  invasion rate for each ocean is assumed (Broecker et al. 1985). By requiring that the total amount of bomb-produced  $^{14}\text{C}$  in a given ocean must equal to the net bomb  $^{14}\text{C}$  input to that ocean, we can obtain the average  $\text{CO}_2$  invasion rate for that ocean. The inventory of bomb-produced  $^{14}\text{C}$  for each ocean was derived from measurements of  $^{14}\text{C}$  profiles throughout the global oceans during the GEOSECS program (Stuiver and Ostlund 1980, 1983; Ostlund and Stuiver 1980). Net input of bomb  $^{14}\text{C}$  to each ocean is calculated by using the required  $\text{CO}_2$  invasion rates. Computations balancing the net excess  $^{14}\text{C}$  input and the total observed excess  $^{14}\text{C}$  inventory for each ocean give invasion rates as follow:  $22.3 \text{ mol}\cdot\text{m}^{-2}\cdot\text{yr}^{-1}$  for the Atlantic Ocean,  $19.4 \text{ mol}\cdot\text{m}^{-2}\cdot\text{yr}^{-1}$  for the Indian Ocean, and  $19.2 \text{ mol}\cdot\text{m}^{-2}\cdot\text{yr}^{-1}$  for the Pacific Ocean.

The global distribution of water column inventories of bomb  $^{14}\text{C}$ , as analyzed by Broecker et al. (1985) and summarized in Fig. 1, shows distinctive zonal excesses and deficiencies in  $^{14}\text{C}$  inventories with respect to net  $^{14}\text{C}$  input in each ocean. The most prominent feature is the presence of excess  $^{14}\text{C}$  inventories in the temperate regions and deficient  $^{14}\text{C}$  inventories in the equatorial and polar regions. The only exception to the pattern is the high latitude region of the North Atlantic, which shows an excess of  $^{14}\text{C}$  inventories. Based on this pattern of distribution, a lateral transport model (Fig. 2) for each ocean was proposed. In the model, water upwells in the equatorial and polar regions.

bringing waters free of bomb-produced- $^{14}\text{C}$  from the deep to the surface, whereas water in the temperate regions downwells from the surface to the deep, bringing waters rich in bomb-produced- $^{14}\text{C}$  to the deep water column. Because of the lack of other geochemical constraints, the lateral transport of water is assumed to take place in the surface ocean. For the Atlantic Ocean, the surface water is transported mainly from the southern hemisphere toward the northern hemisphere. For the Pacific Ocean, the water upwelled from the equatorial region is mostly transported to the southern hemisphere, and a smaller fraction of it is transported northward.

In addition to the pattern of bomb- $^{14}\text{C}$  inventories, the distribution of excess  $\Delta^{14}\text{C}$  (over the prenuclear values) in surface waters also shows a similar zonal pattern, that is, high values in temperate regions and low values in equatorial and polar regions. This distribution pattern provides additional constraints for vertical mixing rates in each latitudinal region. For example, a low value for excess  $\Delta^{14}\text{C}$  in surface waters in the high latitude region of the North Atlantic, coupled with an excess  $^{14}\text{C}$  inventory in this region, calls for a high rate of eddy diffusion in the water column. The advective water fluxes and the vertical eddy diffusion rates for each latitudinal region in each ocean are thus determined in order to match the observed distribution of bomb- $^{14}\text{C}$ . Table 1 summarizes the model parameters that were derived by Broecker et al. (1985).

To use the lateral transport model for estimating the uptake of human-made  $\text{CO}_2$  and predicting atmospheric  $\text{CO}_2$  concentrations, we need to have a closed ocean-atmosphere system. Accordingly, the following modifications are made for the lateral transport model (Fig. 2). Beneath the surface mixed layer in each ocean, a diffusive layer is assumed to extend to the depth of 2000 m. and beneath this depth, a well-mixed deep water reservoir is assumed to extend to the bottom of the ocean at 4000 m. Because of the sinking of surface water and the apparent high vertical mixing rate in the high latitudes of the North Atlantic region, the diffusive layer is assumed to extend to the bottom of the ocean. Although the

oceans need to be linked to represent a global ocean, the model parameters calibrated in the present fashion will not permit such linkage. However,  $\text{CO}_2$  can be transported between oceans through the atmospheric reservoir which is assumed to be well mixed. The revised lateral transport model, as shown in Fig. 2, could be useful in computing  $\text{CO}_2$  uptake for a time scale of a few centuries (i.e., shorter than the deep ocean ventilation time of about a thousand years).

The surface temperature, alkalinity, and nutrient concentrations for each region in each ocean are taken from GEOSECS data (Takahashi et al. 1981). Before human-made  $\text{CO}_2$  is introduced into this ocean-atmosphere system, a steady-state distribution of  $\Sigma\text{CO}_2$  in each ocean region is computed by fixing the atmospheric  $\text{CO}_2$  concentration at 280 ppm. The size of the atmospheric carbon reservoir is taken to be 600 Gt of carbon. The release of anthropogenic  $\text{CO}_2$  into the atmosphere is assumed to begin in 1800 A.D. The history of fossil fuel  $\text{CO}_2$  release between 1860 and 1949 is taken from Keeling (1973), and that between 1950 and 1990 is taken from Marland (1990). The pre-1860 values are linearly interpolated from zero release in 1800. The computed history of the atmospheric  $\text{CO}_2$  content is compared with a standard curve consisting of measured atmospheric  $\text{CO}_2$  concentrations from air bubbles in the Siple ice core (Neftel et al. 1985) for the period 1800-1958 and from air samples at the atmospheric monitoring station at Mauna Loa Observatory for the period 1959-1988 (Keeling et al. 1989).

The atmospheric  $\text{CO}_2$  (Fig. 3) computed from fossil fuel input only is in general  $\sim 14$  ppmv lower than the observed values. Apparently other  $\text{CO}_2$  sources have to be considered. The most likely extra human-made  $\text{CO}_2$  sources are deforestation and oxidation of organic carbon in the soil by agricultural cultivation. The magnitude and time history of these extra sources can be obtained by demanding that the atmospheric  $\text{CO}_2$  computed by the model completely match the standard curve. The history of the required extra carbon source is shown in Fig. 4, which indicates that a substantial terrestrial carbon release

occurred before 1900, with a maximum release of about 1 Gt per year in 1890. Since 1890, the trend generally declines toward the negative value (i.e., the terrestrial biosphere becomes a sink, notably by 1980). This downward trend was reversed after 1980, and the terrestrial carbon flux reached about 0.85 Gt per year in 1990. The total amount of CO<sub>2</sub> released during the period 1800-1980 is estimated to be about 82 Gt of carbon, an amount which is consistent with the lower end of the range (90-120 Gt) estimated from land use data (Houghton 1989). However, as shown in Fig. 4, the history based on land use data indicates that CO<sub>2</sub> releases have increased steadily since 1800. Furthermore, a drastic increase of carbon flux took place by 1950, and it reached about 2 Gt per year by 1990.

## PREDICTION OF THE FUTURE ATMOSPHERIC CO<sub>2</sub>

One main objective of studying the ocean dynamics and chemistry is to be able to reliably estimate the role of the ocean in taking up human-made CO<sub>2</sub> in the future under a given scenario of CO<sub>2</sub> emission. The lateral transport model for each ocean is essentially a composite of a number of box-diffusion models with advective flows that are linked in the surface layers. Working from the hypothetical fossil fuel emission cases as proposed by the Intergovernmental Panel on Climate Change (IPCC 1990), the lateral transport model computes the possible future atmospheric CO<sub>2</sub> concentrations. Results of these computations are shown in Fig. 5. A steady-state distribution of carbon in the model ocean is assumed before the human perturbation started in 1800. The transient distributions of anthropogenic CO<sub>2</sub> in the atmosphere and the ocean are computed by using the known fossil fuel productions and the deforestation releases deduced from this model. During the prediction computations for the years after 1990, the deforestation emission is assumed to stay equivalent to 1990 releases, which total 0.85 Gt of carbon per year. In addition to an uncontrolled fossil fuel consumption case, four other hypothetical study cases proposed by IPCC (1990) are adopted for computations. In case (a), fossil fuel emissions continue

at constant 1990 levels. In case (b), emissions are reduced to 50% of the 1990 levels and continue at that level. In case (c), emissions are reduced by 2% per year from 1990. In case (d), emissions are allowed to increase 2% per year until 2010 and then are reduced 2% per year after that. Results of prediction indicate that atmospheric CO<sub>2</sub> will remain below 500 ppm for the next century if fossil fuel emissions can be controlled as in these hypothetical cases. However, as in case (e), if the average 20th century increasing trend, with an approximate 1% annual increase in fossil fuel production, is allowed to continue to the next century, a doubling of the preindustrial CO<sub>2</sub> concentration (i.e., about 600 ppm) will occur in 2060, and by 2100 it will reach a level higher than triple the preindustrial CO<sub>2</sub> concentration. These model predictions also indicate that a constant emission level results in an almost linear increase in the atmospheric CO<sub>2</sub> concentration [cases (a) and (b)]. However, in cases of a 2% annual reduction of emissions, the model predicts a stabilizing CO<sub>2</sub> concentration toward the second half of the next century.

The IPCC (1990) has also made predictions of future atmospheric CO<sub>2</sub> with various box-diffusion-types of ocean models using CO<sub>2</sub> emission scenarios (Fig. 6) provided by Working Group III of IPCC. Scenario A is assumed to be an uncontrolled "Business-as-Usual" case, in which the energy supply is coal intensive and the deforestation continues until the tropical forests are depleted. In Scenario B, the energy supply shifts toward lower carbon fuels, notably natural gas, and the deforestation is reversed. In Scenario C the energy supply shifts toward renewable and nuclear energy in the second half of the 21st century. In Scenario D, CO<sub>2</sub> emissions are reduced to 50% of the 1985 levels near the middle of the next century by shifting to renewable and nuclear energy in the first half of the next century. This scenario is designed to stabilize the atmospheric concentration by stringent controls in industrialized countries and moderate growth of emissions in developing countries.

Using these CO<sub>2</sub> emission scenarios and the revised lateral transport model, we can predict concentrations of atmospheric CO<sub>2</sub> for the period 1991-2100. The transient distribution of anthropogenic CO<sub>2</sub> in 1990 is computed in the same way. As shown in Fig. 7, for Scenario A, the atmospheric CO<sub>2</sub> in 2067 is twice the preindustrial concentration. This projection is quite compatible with the IPCC results (e.g., the value for 2060). In fact, the patterns of the atmospheric CO<sub>2</sub> levels predicted by this model are very similar to those given by IPCC for all four scenarios. However, the predicted concentrations are lower than the IPCC results because of the more efficient CO<sub>2</sub> uptake in the lateral transport models. In Scenario D, for example, the lateral transport model gives a stabilized atmospheric CO<sub>2</sub> of 410 ppm by the middle of the next century, whereas IPCC gives 450 ppm. However, concentrations for Scenarios B and C in the model never reach 600 ppm before 2100, levels which are consistent with IPCC results.

## CONCLUSIONS

The Broecker et al. (1985) lateral transport model of the world ocean have been revised to calculate the uptake of anthropogenic CO<sub>2</sub> and to predict atmospheric CO<sub>2</sub> concentrations for the next century. The results of this model show a more efficient uptake of CO<sub>2</sub> than those of box-diffusion models used by IPCC. Although the lateral transport model is calibrated with the distribution of bomb-produced <sup>14</sup>C, the carbon chemistry and nutrient distribution have not been calibrated with the observed data. To make predictions of atmospheric CO<sub>2</sub> concentrations more realistic, nutrient distribution, and hence the biological carbon fluxes, needs to be taken into consideration. In addition, because the global ocean is a single body of water, with the regional major oceans connected by the Antarctic ocean, the individual lateral transport models for each ocean need to be linked into a single global model. To do so, the distribution of bomb-produced <sup>14</sup>C will have to be analyzed again, especially in the Antarctic regions. Calibration of this new model will

require substantially more data about  $^{14}\text{C}$  and nutrients in the Antarctic Ocean than are currently available from the GEOSECS program in 1973-1978.

## ACKNOWLEDGEMENT

I thank W. R. Emanuel and A. W. King for reviewing the manuscript. Research sponsored by the Carbon Dioxide Research Program, Atmospheric and Climate Research Division, Office of Health and Environmental Research, U.S. Department of Energy, under contract DE-AC05-84OR21400 with Martin Marietta Energy Systems, Inc. Publication No. 0000, Environmental Sciences Division, Oak Ridge National Laboratory.

"The submitted manuscript has been authored by a contractor of the U.S. Government under contract DE-AC05-84OR21400. Accordingly, the U.S. Government retains a nonexclusive, royalty-free license to publish or reproduce the published form of this contribution, or allow others to do so, for U.S. Government purposes."

Table 1. Summary of model parameters<sup>a</sup>

Latitude belt	Area ( $10^{12} \text{m}^2$ )	W (m/year)	Flux (sverdrups)	K ( $\text{cm}^2/\text{s}$ )	Z (m)
Atlantic Ocean					
80°N-40°N	18.6	-8.5	-5	9.9	710
40°N-20°N	15.8	-30.0	-15	0.5	470
20°N-20°S	26.7	+21.2	+18	1.0	190
20°S-45°S	18.4	-22.3	-13	1.0	400
45°S-80°S	15.1	+31.2	+15	3.0	270
Indian Ocean					
25°N-15°S	27.0	+15.2	+13	1.0	215
15°S-45°S	29.8	-20.1	-19	1.0	510
45°S-70°S	20.7	+9.1	+6	3.0	410
Pacific Ocean					
65°N-40°N	15.1	+10.4	+5	1.5	260
40°N-15°N	35.0	-10.8	-12	1.0	365
15°N-10°S	50.0	+16.4	+26	1.0	255
10°S-55°S	63.0	-12.5	-25	2.0	410
55°S-80°S	13.8	+13.7	+6	3.0	335

<sup>a</sup>W = Mean advective flow rate (minus represents downwelling); K = Vertical diffusivity; and Z = Mean penetration depth of bomb-produced  $^{14}\text{C}$

## Figure Captions

Fig. 1. Bomb testing  $^{14}\text{C}$  inventories as measured during the GEOSECS, NORPAX, and TTO programs (as summarized by Broecker et al. 1985). The dashed lines show the amount expected at the time of the GEOSECS program under the assumption that there was no lateral transport from one zone of each ocean to another.

Fig. 2. Modified lateral transport models for the Atlantic, Indian, and Pacific oceans. The well-mixed atmospheric box linked these oceans during the  $\text{CO}_2$  uptake calculations. The magnitude of the advective flows and the vertical diffusivity for each water column are summarized in Table 1.

Fig. 3. Model-derived atmospheric  $\text{CO}_2$  concentration (dashed curve) when fossil fuel combustion is assumed to be the sole source of perturbation. The solid curve represents observed concentrations of atmospheric  $\text{CO}_2$  from the Siple ice core and the Mauna Loa atmospheric record.

Fig. 4. Extra  $\text{CO}_2$  sources (solid curve) needed to match the ocean-model-derived atmospheric  $\text{CO}_2$  curve with the observed one. The dashed curve represents the level of  $\text{CO}_2$  emissions that result from deforestation, as derived from land use data (Houghton 1989; values before 1855 are linearly interpolated on the basis of zero emissions in 1770).

Fig. 5. Atmospheric  $\text{CO}_2$  concentrations predicted by the lateral transport model using five hypothetical fossil fuel emissions: (a) constant at the 1990 level, (b) constant at 50% of the 1990 level, (c) reduced by 2% per year from 1990, (d) growing by 2% per year from 1990 until 2010 and then reduced by 2% per year thereafter, and (e) current trend of the uncontrolled growth into the next century.

Fig. 6. CO<sub>2</sub> emission scenarios and predicted CO<sub>2</sub> concentrations published by the Intergovernmental Panel on Climate Change. Details of each scenario are given in the text.

Fig. 7. Atmospheric CO<sub>2</sub> concentrations predicted by the lateral transport model using four CO<sub>2</sub> emission scenarios proposed by Intergovernmental Panel on Climate Change. The CO<sub>2</sub> concentrations in this figure for year 2100 are 784, 549, 478, and 411 ppmv for scenarios A, B, C, and D, respectively.

## REFERENCES

Broecker, W. S., Peng, T.-H., Ostlund, G., and Stuiver, M. 1985. "The distribution of bomb radiocarbon in the ocean," J.Geophys.Res. 90: 6953-6970.

Houghton, R. A. 1989. "The long-term flux of carbon to the atmosphere from changes in land use." Extended abstract from the Third International Conference on Analysis and Evaluation of Atmospheric CO<sub>2</sub> Data, Present and Past. Hinterzarten, 16-20 October, 1989.

IPCC (Intergovernmental Panel on Climate Change), 1990. "Scientific assessment of climate change," Report prepared for IPCC by Working Group I, June 1990.

Keeling, C.D. 1973. "Industrial production of carbon dioxide from fossil fuels and limestone," Tellus 25: 174-198.

Keeling, C.D., Bacastow, R. B., Carter, A.F., Piper, S.C., Whorf, T.P., Heimann, M., Mook, W.G., and Roeloffzen, H. 1989. "A three-dimensional model of atmospheric CO<sub>2</sub> transport based on observed winds: 1. Analysis of observational data." Geophys.Mono. 55, Am. Geophys. Union, pp.165-231.

Marland, G. 1990. "Global CO<sub>2</sub> emissions." IN: Boden, T. A., Kanciruk, P., and Farrell, M.P. (eds.), Trends'90: A Compendium of Data on Global Change, ORNL/CDIAC-36, Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, Oak Ridge, Tenn., p.92.

Neftel, A., Moor, E., Oeschger, H., and Stauffer, B. 1985. "Evidence from polar ice cores for the increase in atmospheric CO<sub>2</sub> in the past two centuries," Nature, 315: 45-47.

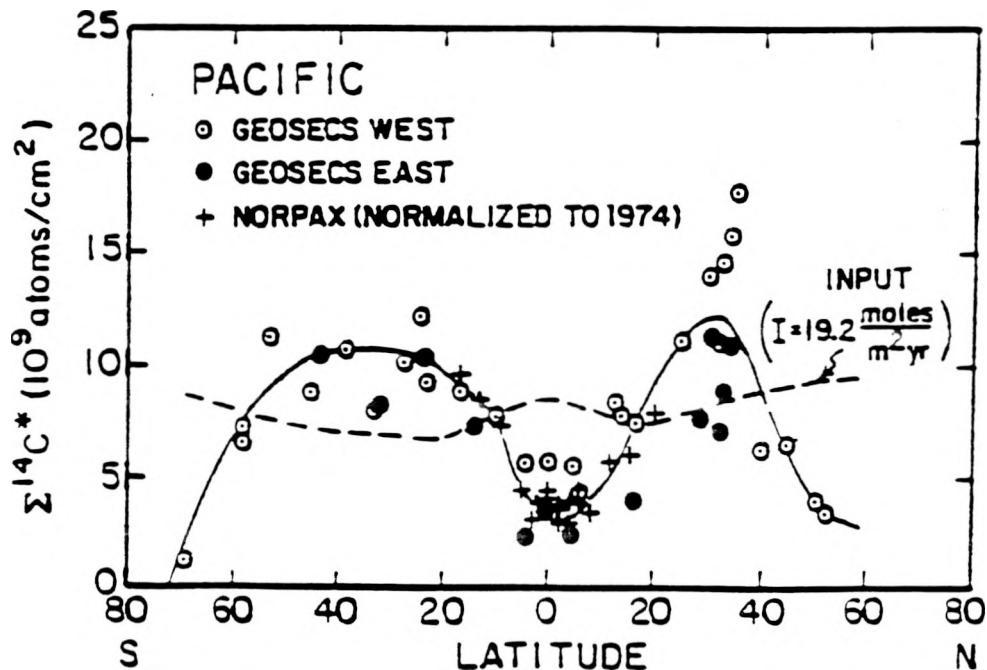
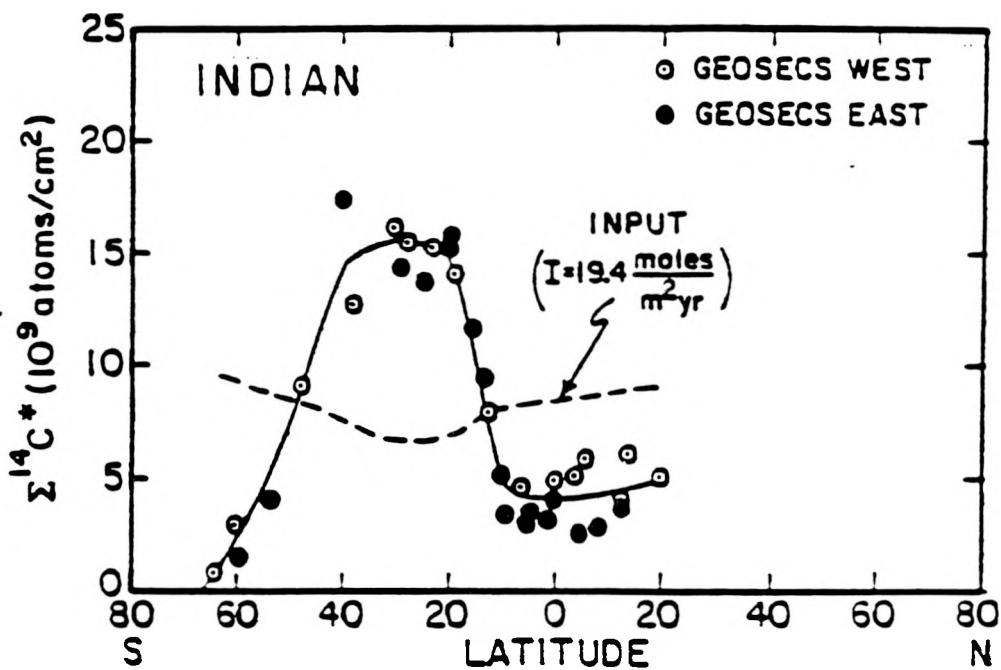
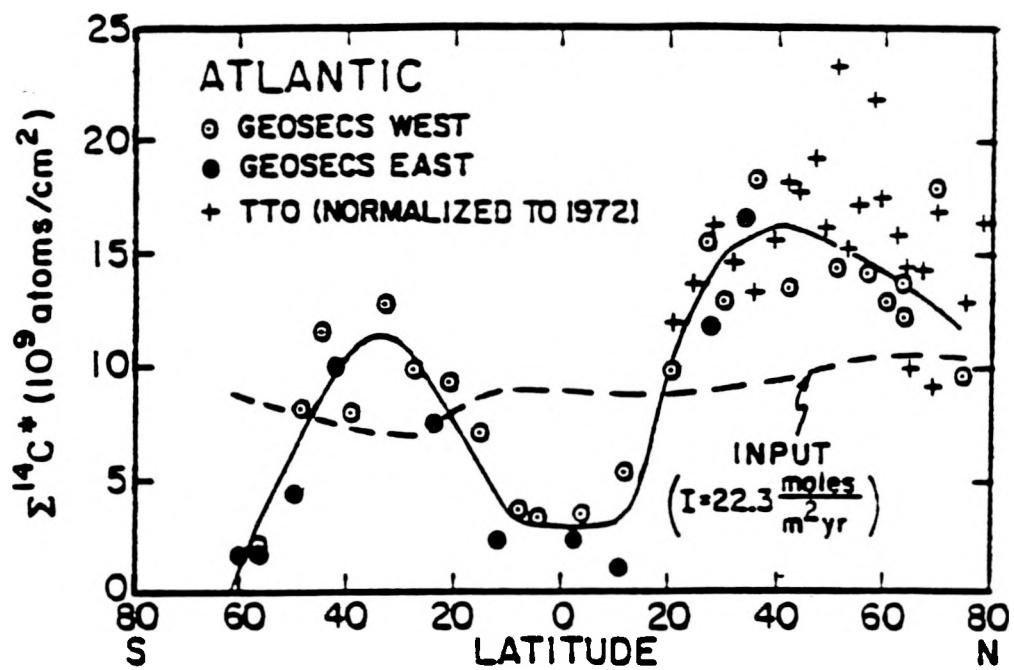
Oeschger, H., Siegenthaler, U., Schotterer, U., and Gugelmann, A. 1975. "A box diffusion model to study the carbon dioxide exchange in nature," *Tellus*, 27: 168-192.

Ostlund, H. G., and Stuiver, M. 1980. "GEOSECS Pacific radiocarbon," *Radiocarbon* 22: 25-53.

Stuiver, M., and Ostlund, H. G. 1980. "GEOSECS Atlantic radiocarbon," *Radiocarbon* 22: 1-24.

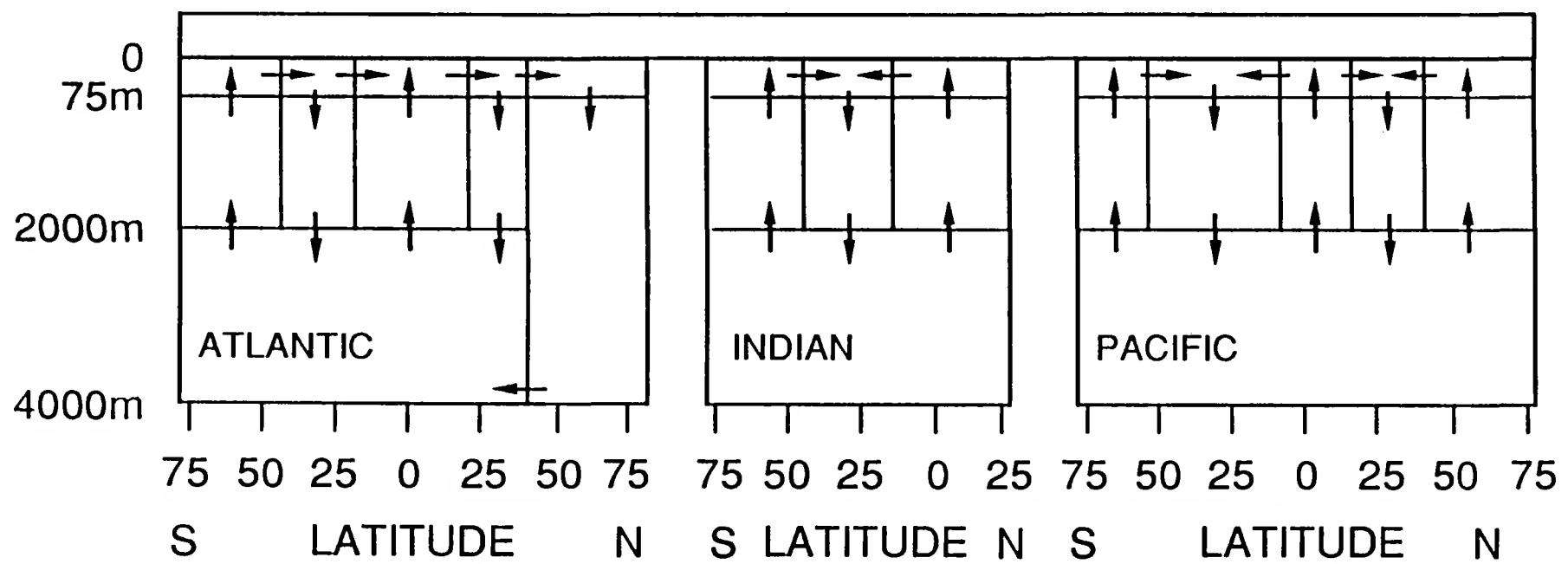
Stuiver, M., and Ostlund, H. G. 1983. "GEOSECS Indian Ocean and Mediterranean radiocarbon," *Radiocarbon* 25: 1-29.

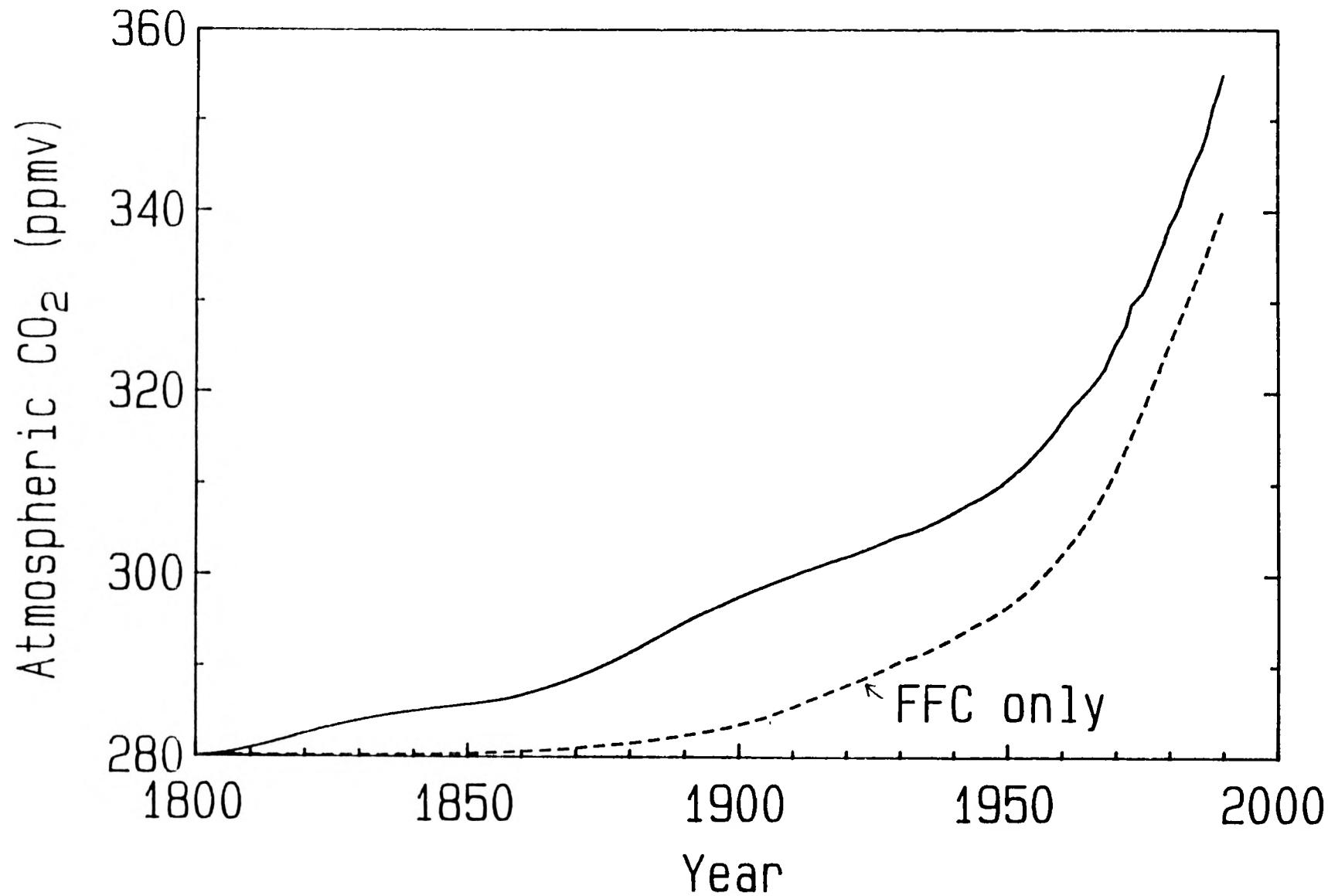
Takahashi, T., Broecker, W. S., and Bainbridge, A. E. 1981. "Supplement to the alkalinity and total carbon dioxide concentration in the world oceans," IN: Bolin, B. (ed.) *SCOPE16: Carbon Cycle Modeling*, John Wiley & Sons, New York, pp.159-200.

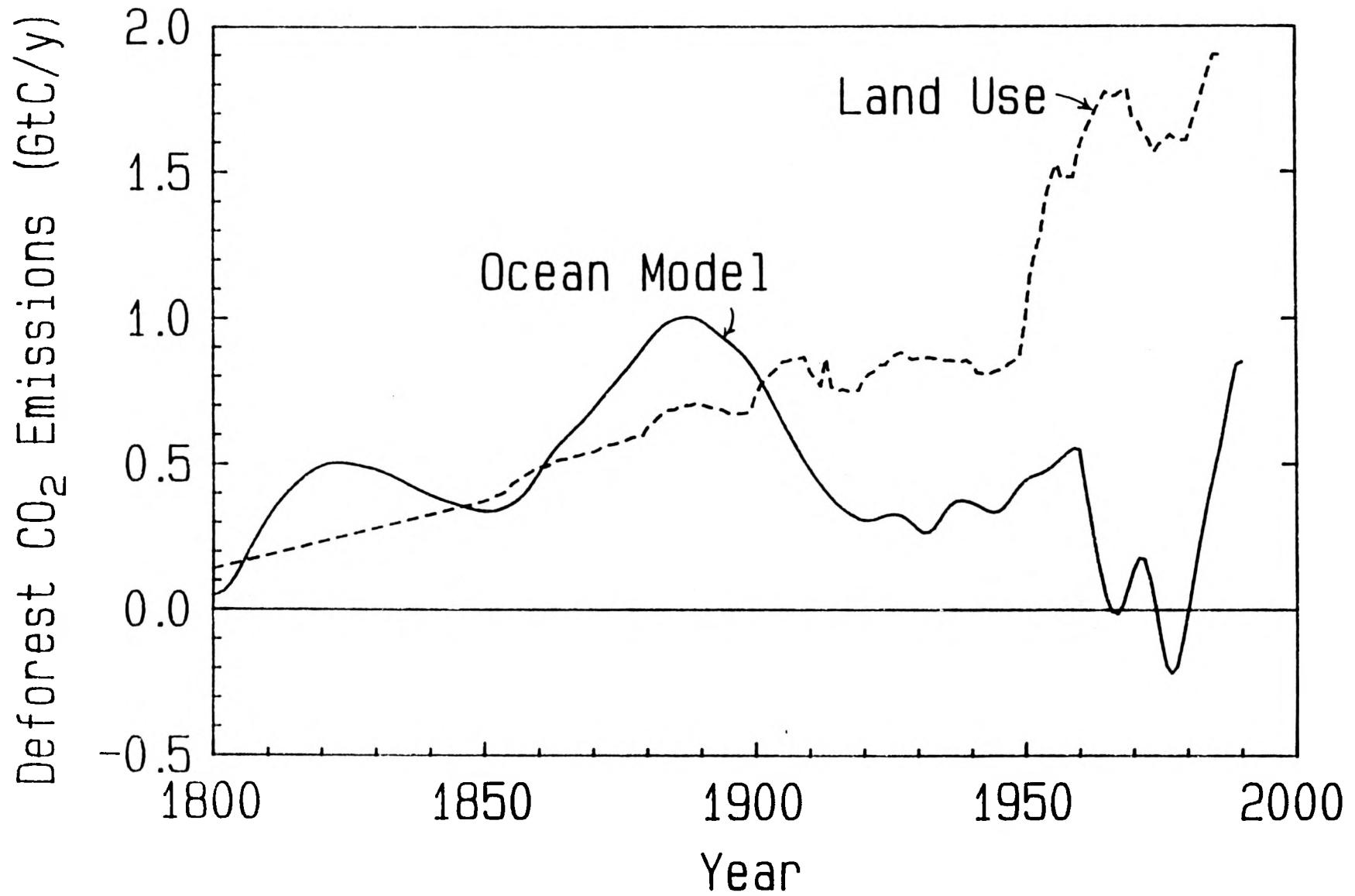


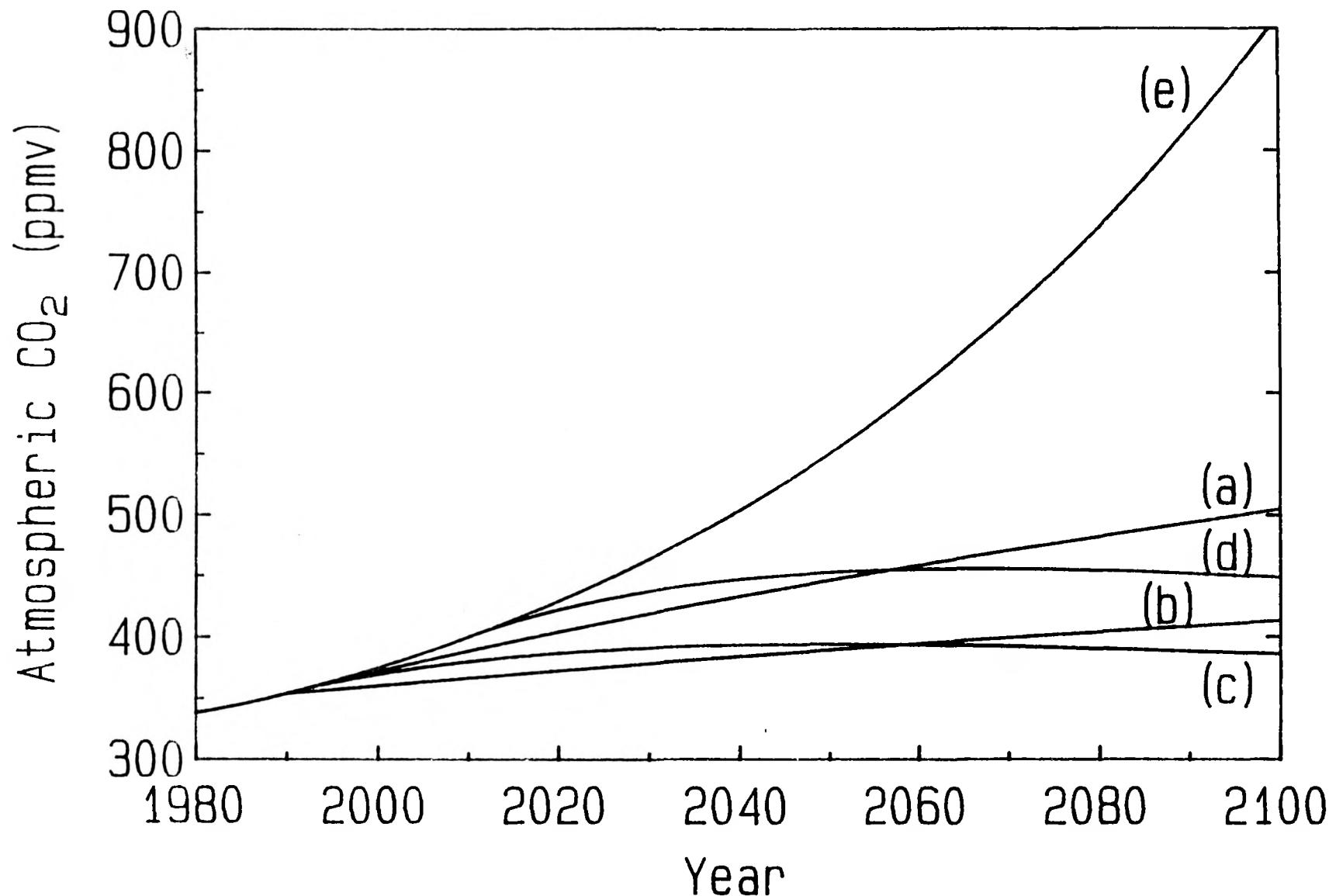
ORNL-DWG 90M-13416

## Atmosphere









## IPCC CO<sub>2</sub> Emission Scenarios

