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RECHARGEABLE ALKALINE ZINC/FERRICYANIDE BATTERY

Final Report, September 29, 1978--September 28, 1979

By
George B. Adams
Roger P. Hollandsworth
Bruce D. Webber

Work Performed Under Contract No. EM-78-C-01-S163

Lockheed Missiles & Space Company, Inc.
Lockheed Palo Alto Research Laboratory
Palo Alto, California



U. S. DEPARTMENT OF ENERGY

Division of Energy Storage Systems

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ALKALINE ZINC/FERRICYANIDE BATTERY**

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for the period

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**George B. Adams, Roger P. Hollandsworth
and Bruce D. Webber**

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**U. S. Department of Energy
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Contract EM-78-C-01-5163

**Lockheed Palo Alto Research Laboratory
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ABSTRACT

First-year contract effort was directed primarily to assessing technical and economic feasibility of the alkaline zinc/ferricyanide rechargeable battery for utility load-leveling applications.

This battery meets the requirements for this application with cell voltages of 1.94 V on charge and 1.78 V on discharge. (35 mA/cm^2) at 40°C , with peak power density of 4.5 kW/m^2 . Mean energy efficiency is 84% at 760 and 86% at 1110 4-hour cycles in full-cell, and redox half-cell cycling, respectively.

An updated, upgraded economic analysis suggests a battery selling price of \$32/kWh, installed price of \$230/kW and footprint of 8.7 kWh/ft^2 as realistic goals for a 20 MW 100-MWh system.

A major effort was successfully completed on design, construction, programming, and debugging of a microprocessor-controlled battery cycling system, which is the heart of the cell cycling test facility developed on this project.

Other technical areas addressed include: electrode substrate evaluation, solubilities and conductivities of zinc oxide and sodium ferro- and ferricyanide in sodium hydroxide electrolytes, mechanism of the slow chemical deligation of alkaline ferricyanide and its reversal, optimal compositions of zincate and iron redox electrolytes, effect of electrolyte flow rate and additions on quality of zinc electrodeposits, cell and electrode design and fabrication, and full and half-cell cyclic testing.

In overview, significant technical progress was made in concept configuration and improving performance of the alkaline zinc/ferricyanide battery.

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SUMMARY

This project was directed primarily to assessing the technical and economic feasibility of the alkaline zinc/ferricyanide rechargeable battery for utility load leveling application. During this first year contract (Phase I), many of the most critical issues have been probed on a laboratory scale. Also, our previous preliminary economic analysis has been updated and upgraded. Cyclic cell test results over many hundreds of four-hour charge-discharge cycles indicate that the performance of the alkaline zinc/ferricyanide system exceeds the original proposed goal of 70% for turnaround energy efficiency by over 10%; the economic analysis suggests that a battery selling price of ca \$32/kWh and an installed price of \$230/kW are realistic goals for a 20 MW, 100 MWh system.

Specifically, the following accomplishments are highlighted:

- In full-cell cyclic testing, after 760 four-hour cycles, overall energy efficiency is 84%, with cycling continuing (sec. 4.6.5).

- A ferro-ferricyanide-redox half-cell currently stands at 1110 four-hour cycles (4440 hours), with a mean energy efficiency of 86%, with cycling continuing (sec. 4.5.2.4). With this redox couple, mean coulombic and voltaic efficiencies were determined as functions of electrolyte flow rate and current density. Optimal efficiency values are 98.5% coulombic and 96% voltaic at 35 mA/cm² (sec. 4.5.2).

- Mean coulombic and voltaic efficiencies as functions of current density and electrolyte concentration were determined for the zinc electrode in half-cell testing. Optimal values are 97% coulombic and 92% voltaic at 35 mA/cm² (sec. 4.5.1).

- An updated and upgraded cost analysis, based on a production rate of twenty-five, 20 MW, 100-MWh batteries per year, indicates a battery selling price of \$31.50/kWh. Battery footprint is 8.7 kWh/ft² (sec. 4.7).

- A major effort was completed on design, construction, programming, and debugging of a microprocessor-controlled battery cycling system. This system is the heart of the cell cycling test facility. It controls the charge and discharge cycles of six or more test cells, based upon five different operator-selected cell parameters. It computes, at the end of each cycle, the voltaic, coulombic, and energy efficiencies for each cell (sec. 3.3).

- An instrumented test stand was built to independently service six full or half cells for long-term cyclic testing. Six 60-cm² test cells were designed, fabricated, and checked for hydrodynamic flow characteristics. Operational characteristics were found to be excellent (sec. 3.2).

- The first step in the long-term very slow chemical degradation of the alkaline iron redox couple has been tentatively identified as the abstraction of a cyanide ion from the ferricyanide ion to form a pentacyanoaqua iron(III) specie. The metals, Cu, Ag, and Hg have been found to accelerate this process, while Zn, Cd, and Ni, which are used in cell components, are inert (sec. 4.2.2). The religation of ferricyanide in cyclically aged redox electrolyte has been demonstrated, with subsequent recovery of redox electrode capacity (Sec. 4.2.3).

- Ten materials have been evaluated by cyclic voltammetry for use as substrates for the zinc electrode. Three of these were run in half-cell tests. A cadmium-plated iron substrate was found to be optimal (secs. 4.1, 4.5.1). Eight ferro-ferricyanide redox electrode substrate materials have been evaluated by cyclic voltammetry. Porous nickel has been selected as optimal but other less expensive candidates have been identified (Sec. 4.2.1, 4.5.2.1).

- Solubilities of sodium ferro- and ferricyanides in sodium hydroxide electrolytes and of potassium ferro- and ferricyanides in both sodium and potassium hydroxide electrolytes, and conductivities of these saturated solutions were

measured over the temperature range 22° - 50°C (sec 4.3.2). Solubilities of ZnO in ZnO-saturated sodium hydroxide electrolytes and conductivities of the electrolytes were determined, over the temperature range 28° - 50°C (sec. 4.3.1).

- A limiting factor affecting the redox electrode coulombic efficiency at high charge capacity was identified as concentration polarization in 5N NaOH occurring in the last 15% of the charge cycle. The redox electrolyte concentration was found to be optimal at 2.2 N NaOH. With this electrolyte at 40°C , coulombic efficiency is 99% (sec. 4.5.2.3).

- The optimal zincate electrolyte is saturated in zinc oxide. Coulombic efficiency is temperature independent from 25° to 50°C (secs. 4.5.1.2, 4.5.1.5). Physical characteristics of zinc electrodeposits were found to relate to electrolyte face velocity (sec. 4.5.1.3). Four zincate electrolyte additives have been evaluated. Of these, sodium silicate appears the most promising (sec. 4.5.1.7). The presence of ferrocyanide ion does not seriously affect the energy efficiency of the zinc electrode on long-term cycling (sec. 4.5.1.4). Passivation of the zinc electrode by formation of an insoluble film of zinc ferrocyanide due to cross-diffusion of ferro- or ferricyanide into the anolyte is precluded since zinc ferrocyanide is soluble in 2 - 5N NaOH electrolyte (sec. 4.3.1).

- Five teflon-based battery separator materials have been evaluated (sec. 4.4).

Those technical areas requiring additional effort have been identified, and are the subject of a proposed Phase II follow-on program (sec. 5).

In overview, it is felt that significant technical progress has been made in confirming and improving the performance of the alkaline zinc/ferricyanide battery. None of the results from Phase I experimental work to date would point to a change in the original concept, in which system size is minimized by using saturated solutions of reactants and products and storing them in solid form. In fact, the overall system concept and electrochemical performance have exceeded expectations.

Section 1

INTRODUCTION

This report details technical results achieved on the first 12-month phase of a research and development effort conducted at Lockheed Missiles & Space Co., Inc. Palo Alto Research Laboratory for the Division of Energy Storage Systems of the Department of Energy under Contract EM-78-C-01-5163, on a "Rechargeable Alkaline Zinc/Ferricyanide Hybrid Redox Battery." This battery is a low-cost electrical energy storage system which is adaptable to either solar photovoltaic/wind or distributed energy storage for electric utility load-leveling applications.

Increasing demands for electric power in the face of rising energy costs have created an acute problem for the electric utility industry in meeting the intensive peak power demands of industry. Leveling or peak-shaving these loads and using off-peak stored energy allow more efficient utilization of base-load energy, with consequent significant savings in hydrocarbon fuels as required for base-load power generation. An alternative or supplemental* method to a pumped hydroelectric system for large-scale energy storage is urgently required to meet the pressure of these rapidly growing electric power demands, since placement of pumped hydroelectric systems is severely limited by environmental restrictions. An electrochemical system for bulk electrical energy storage has a number of advantages over pumped hydroelectric storage. These advantages include minimal environmental and siting problems, short construction lead times, instant start-up capability, potentially high operational efficiency, and in common with other energy storage options, batteries confer substantial spinning-reserve and system regulation benefits because of their operating characteristics. The modular construction of battery systems would allow location at utility substations in order to meet the specific needs of a local industrial, commercial, or resi-

*A 1977 Electric Power Research Institute (EPRI) assessment indicates that oil consumption in the United States would be reduced, in the year 2000, by some ten to one-hundred million barrels annually for total installed electrical storage capacities of 18-50 GW.

dential market. Battery cost and performance goals established by EPRI for utility load-leveling application include:

- Installed cost: \$25/kWh + \$75/kW or \$40/kWh overall (1977\$)
- Overall energy efficiency: 70%
- Footprint: 8kWh/ft²
- Useful life: 2500 charge/discharge cycles over a 10-year system service life
- Height: 20 feet, maximum
- Minimum siting restrictions

In addition, EPRI standardized cost estimating guidelines for computing battery installed capital cost, specify an annual production rate of 25 100-MWh batteries per year and the definition of a standard manufacturing facility for battery production.

The objective of Phase I of the present program was to determine, within the constraints of allocated funding, the technical feasibility of the alkaline zinc/ferricyanide rechargeable battery for ultimate application in utility load leveling. Results to date are most promising based on the electrochemical operating characteristics of the system and an upgraded cost estimate for a 100 MWh storage facility using this battery.

This battery has a number of advantages that should be stressed. It is an alkaline aqueous electrolyte battery which operates best at a temperature slightly above ambient. The advantages of an alkaline over a strongly acidic electrolyte become apparent when one considers the number of successful alkaline electrolyte batteries that have been developed compared to successful acid electrolyte batteries (primary and secondary nonreserve type). Thus, on the alkaline side we have zinc/copper oxide; zinc/oxygen; zinc/mercuric oxide; zinc/silver oxide; cadmium/nickel oxide; iron/nickel oxide; cadmium/silver oxide and, of more recent vintage, zinc/nickel oxide and hydrogen/nickel oxide. Commercially only the lead/lead dioxide battery, which is a unique system, stands out as the sole acid electrolyte contender.

When cost is a prime consideration, an alkaline electrolyte battery has decided advantages, which are mainly due to the much less corrosive electrolyte which allows the use of lower cost electrode substrate materials and system auxiliaries.

The alkaline zinc/ferricyanide battery is an ambient temperature, flowing electrolyte system using zinc/zincate and ferro/ferricyanide electrode couples.

A major advantage of this system is the high cell voltage (1.86 V OCV) and electrode reversibility (1.78 V at 35 mA/cm² or 60 mW/cm²). Peak power density is over 4.5 kW/m² with present cell design; mean energy efficiency is 84% over 698 4-hour cycles to date, with cyclic testing continuing. At 690 cycles, energy efficiency has increased to 87%.

The basic chemical components of this battery are zinc, iron, carbon, nitrogen, and sodium, all of which are abundant and low in cost. The alkaline electrolyte allows the use of relatively inexpensive electrode substrate materials and systems auxiliaries, which contribute to a lower capital cost per kilowatt-hour of energy stored. A preliminary analysis of capital costs for this system, based on a projected design for a utility load-leveling battery sized at 20 MW, 100 MWh, and a production rate of 25 batteries per year, suggests a battery selling price of \$32/kWh.

Other advantages of this battery include the use of relatively low cost separator materials; flat charge-discharge curves due to circulation of saturated solutions of reactants; minimal storage volume required for reactants because of external storage as solids; deep depth of discharge; low toxicity components, and long cycle life inherent to redox-type electrodes.

Technical achievements to date under the current contract (Phase I) have confirmed concept viability. In fact, the predicted performance has exceeded the original proposed goals of 70% for energy efficiency by some 10% in long-term cell cycling studies.

Section 2

TECHNICAL BACKGROUND

2.1 INTRODUCTION

The principal criteria to be met in the selection of a new secondary battery for utility load-leveling applications are those of maximum allowed cost per unit of electrical energy output and the ability to provide this energy efficiently over thousands of deep charge-discharge cycles. Battery cost and performance goals for electric utility load leveling application include round-trip energy efficiency of 70 percent, initial battery cost of \$30/kWh, and cycle life of 2500 cycles (Ref. 1).

A redox battery system has the intrinsic advantages of extensive deep-cycling life potential due to charge transfer from an inert conducting substrate to dissolved reactant/products, low-temperature operation, and low-cost storage of reactants/products externally to the battery itself. Disadvantages of those redox systems having sufficient electrode reversibility to be practical usually involve low-cell voltages and corrosive electrolytes. The low-cell voltage problem is eliminated in metal-halogen systems, such as zinc-chlorine or zinc-bromine, but serious corrosion and halogen storage problems remain that require costly solutions in terms of higher cost materials of construction.

The hybrid redox zinc/ferricyanide battery is an alkaline electrolyte system that is subject to only minimal corrosion problems and has a high reversible cell voltage. The two electrochemical couples involved are thermodynamically reversible and operate under mass transfer control.

Single-cell tests with this system have demonstrated high coulombic and turn-around energy efficiencies over hundreds of charge/deep discharge cycles. The system has been analyzed on a preliminary basis for cost, and has been found to meet those capital cost requirements generally accepted for commercial applicability of utility load-leveling.

The following paragraphs provide a description of the overall system concept, as applied to utility load-leveling, a description of a novel approach for storing cell reactants externally as solids, and technical background information on the zinc and ferro-ferricyanide redox electrodes.

2.2 ZINC/FERRICYANIDE BULK ELECTRICAL ENERGY STORAGE SYSTEM

The complete bulk electrical energy storage system based on the alkaline zinc/ferro-ferricyanide battery is depicted in Fig. 2.2.1.

In the charged state, metallic zinc is stored in the redox converter, and sodium ferricyanide monohydrate is stored externally as a solid in a storage tank C. On discharge, zinc dissolves as the zincate ion and is stored in an external Tank B, as solid zinc oxide. Solid sodium ferricyanide monohydrate in Tank C is converted to solid sodium ferrocyanide decahydrate. The cell is operated at some temperature above ambient so that the reactant solids will crystallize out of the supersaturated solutions in the storage tanks, at ambient temperature.

This approach has two important advantages: It minimizes storage tank volume requirements and also maintains discharge voltage more nearly constant over the complete charge and discharge cycles, since the ferrocyanide and ferricyanide ion concentrations are automatically maintained constant at the ambient temperature saturation values. Minimal amounts of electrolyte are circulated through the system.

After a prolonged period of operation (perhaps every 1000 cycles), sufficient sodium ferrocyanide decahydrate solid will accumulate in Tank B necessitating recovery. This recovery is accomplished as follows. The cells would be charged fully to dissolve all zinc oxide solid in the anolyte. Tank B would then be drained of free electrolyte by pumping into Tank A. The crystalline sodium ferrocyanide slurry would then be pumped back into Tank C. Tank B would be refilled with anolyte from Tank A, and the system would be ready for

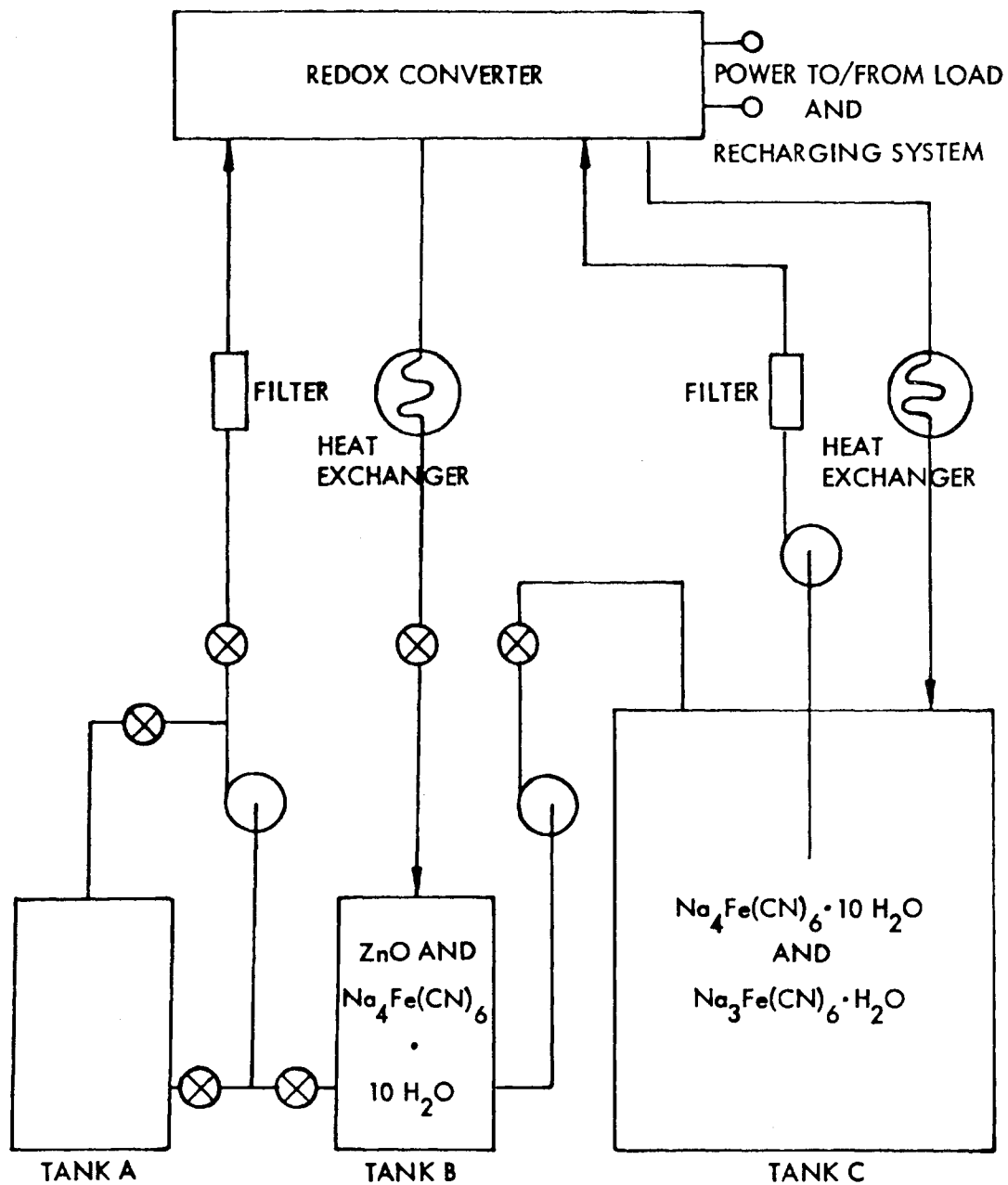


Fig. 2.2.1 Electrically Rechargeable Hybrid Redox Bulk Energy Storage System

discharge. It should be noted that Tank C capacity is about 40 times that of A or B. A mechanical stirrer could be incorporated in Tank B to facilitate removal of the crystalline slurry. If necessary, liquor from Tank C would be used to facilitate the transfer.

The zinc/ferricyanide storage system has the following features:

- This recycling concept should permit the use of relatively low-cost membrane separator material with significant reduction in overall capital cost.
- The intermixing of reactants does not cause irreversible voltage degradation, as discussed in Section 4.5.1.
- Any zinc lost from the electrode by mechanical dislodgement will be slowly redissolved by reaction with ferricyanide ion diffusing in from the catholyte compartment, and reenter the system as zincate ion.

2.3 ZINC ELECTRODE

2.3.1 Electrode Substrate

In the operation of the battery, zinc is electrodeposited at the negative electrode on charge and anodically stripped from this current-conducting substrate upon discharge. The nature of the substrate used for the electrode affects the adhesion characteristics of the deposit (Ref.2).

Certain criteria can be used in selecting substrate materials for the zinc electrode. These criteria relate to hydrogen overvoltage, adherency of the metal deposit, and corrosion inhibition and can be listed as follows:

- The substrate should have a high hydrogen overvoltage (Ref.3)
(Pb > Hg > Cd > Ag > Cu > Fe > Ni > Pt)
- The substrate should exhibit properties of good zinc adherency (Ref.4 ,5)
(Zn(Hg) > Zn > Cd > Ag > Cu > Pb > Sn)
- Zinc corrosion inhibitors are beneficial (Cd^{++} , Al^{+++} , Pb^{++}) (Ref.6)
- Zinc corrosion accelerators are to be avoided (Refs.6 , 7, 8)
(Cu^{++} , Fe^{++} , Sb^{+++} , As^{+++} , Ag^{+})

2.3.2 Zinc Electrodeposition

The electrochemistry of zinc in static secondary alkaline batteries has been studied for many years. The formation of voluminous, porous and/or dendritic growths rather than compact, low porosity crystalline deposits in zinc electrodeposition is a problem that has been investigated, mainly in quiescent solutions. Studies in vigorously stirred electrolytes are not as well documented. It has been reported that smooth deposits are formed at low overpotentials in stirred electrolytes, whereas in quiescent solutions mossy deposits are formed at low overpotentials (<75 mV) and dendritic deposits occur at overpotentials above 75 mV (Refs. 2, 9). Dendritic deposits are formed when the rate of the electrode reaction is controlled by mass transport of the zincate species to the electrode surface, i.e., when the surface

concentration of the electroactive species is less than the bulk concentration (Ref. 9). Studies of morphology of alkaline deposited zinc as a function of current density and face velocity under both turbulent and laminar flow show that flow conditions greatly modify the deposit morphologies. Flat crystalline deposits are obtained at 50 mA/cm^2 for a Reynold's number of 1500 and at 300 mA/cm^2 for a Reynold's number of 14,000 (Ref. 10).

Rotating disc investigations have shown that there is a very narrow current density range where smooth zinc electrodeposits are obtained, which is intermediate to those for moss and dendrite formation (Refs. 11, 12). It is well known that small amounts of Pb^{++} inhibit mossy zinc formation (Ref. 13, 14); lead additions in the range of 5×10^{-5} to $3 \times 10^{-4} \text{ M}$ drastically alter the electrode double-layer capacity and the impedance diagram (Ref. 15). Additions of other metal ions such as Sn^{++} , Hg^{++} , and Cd^{++} and various organics such as ethylene glycol polymers and derivatives have all been reported to induce crystalline deposits in static zinc batteries (Refs. 11, 12, 16-23).

A current German patent (Ref. 24) specifies the use of sodium silicate as an additive to zincate baths for improving the quality and grain structure of the electroplate over a broad range of current densities.

It would appear that most of the cycle-life problems reported with alkaline zinc batteries are related to concentration gradients of zincate ion building up in the static electrolyte, and to sizable changes of zincate ion concentration occurring in the boundary layer during charge and discharge. When a circulating electrolyte is used, these problems largely disappear due to thinning of the boundary layer and elimination of concentration gradients in the cell. This has been observed in practice with a fluidized bed zinc electrode which performs well at charge-discharge rates of 1 A/cm^2 (Ref. 25). Continuous electrolyte circulation has also proven out in the zinc-air system (Ref. 26). Circulation produces adherent zinc deposits because of the reduced thickness of the diffusion layer. Zinc oxide settles out in the storage tank from the supersaturated zincate solution produced in the battery.

The cycle-life problem with zinc should be diminished greatly by completely stripping the zinc from the electrode surface on each discharge. It is believed that the use of a zinc electrolyte composition optimized for zinc adherence, combined with the strip-discharging of zinc on each cycle, will provide the most successful approach to achieving extended cycle life on the zinc electrode.

2.4 FERRO- FERRICYANIDE ELECTRODE

2.4.1 Reversible Potential

The ferro-ferricyanide ion couple reaches a rapid, reversible equilibrium, and many concordant values have been obtained for cells employing this couple in the form of the potassium salts. The potential, however, is a function of the potassium ion concentration, and the correction to an accurate standard EMF value becomes very large. With equal concentrations of the two anions, the observed potential is 0.48 V, whereas the standard potential is 0.36 V (Ref. 27). In a recent study (Ref. 28) of this cation effect on the equilibrium potential of the couple (using a supporting electrolyte of LiNO_3 , over a concentration range of 0.2 to 10 M), the equilibrium potential ranged from 0.43 to 0.57 V. The explanation for this surprisingly large effect is that the $\text{Fe}(\text{CN})_6^{-4}$ ion with a higher negative charge forms ion pairs with the cations of the supporting electrolyte more strongly than the $\text{Fe}(\text{CN})_6^{-3}$ ion. As a result, the activity of the reduced species is lowered to a greater extent than that of the oxidized species, with increasing supporting electrolyte concentration and with the resultant rise in EMF.

In our cell tests, open circuit potentials of 1.8 to 1.9 V have been observed in the charged state. The standard reversible potential is only 1.58 V. The difference of 0.27 V is quite large and of course beneficial. It can be approximately accounted for by taking 0.48 V as the equilibrium potential of the ferro-ferricyanide couple, uncorrected for cationic effects, to compute the EMF of the zinc/ferricyanide cell, charged to a ferricyanide-ferrocyanide ion ratio of 1.0 in an electrolyte of 5 M NaOH. This estimated EMF is 1.79 V; the 70-mV

difference is no doubt largely due to the additional cationic coupling effect of 5M Na^+ compared to about 1M K^+ . This work suggests that it may actually be possible to further increase the reversible potential of the ferro-ferricyanide couple by addition of inert salts to the alkaline electrolyte.

2.4.2 Reversibility of the Couple

A recent kinetic study on the ferro-ferricyanide ion couple in acidic, neutral, and alkaline solutions (Ref. 29), using a lithiated nickel-oxide electrode, shows that, in alkaline solution, the electrode reaction is sufficiently fast that both oxidation and reduction are under almost pure diffusion control.

In another study (Ref. 30), the effect of flow velocity in the range 1 to 10 m/s on the electrochemical reduction of potassium ferricyanide to ferrocyanide was measured. No activation polarization was found; only concentration polarization is observed. Linear increases in current densities with increasing velocities were obtained at all polarization potentials.

In other work (Ref. 31), the redox reaction was done on a graphite rotating disk electrode in 0.5 M KCl over the concentration range 10^{-2} to 10^{-4} M. The presence of a dimer is postulated in the absorption layer from which transfer occurs, which consists of one molecule of oxidized to one of reduced form. The dimer is more rapidly converted electrochemically than the monomer. The system was found to be reversible, although somewhat more complex than previous work had indicated.

Current-voltage curves were obtained for electrolytes 10^{-4}M in potassium ferrocyanide, potassium ferricyanide, and also for mixed ferro-ferricyanide ions, all in 1 M KCl at 21°C , using a tubular graphite electrode at flow rates of 6 to 14 ml/min (Ref. 32). It was found that the oxidation-reduction of ferro-ferricyanide ion is reversible under either quiescent or dynamic conditions.

In recent research (Ref. 28), the influence of alkali metal cations on the rate of the ferro-ferricyanide redox reaction was determined, using a current impulse

technique on a spherical gold microelectrode. In concentrated electrolytes, the reaction rate shows a first-order dependence on the electrolyte cation concentration due to an activated species which contains one or more cations. The catalytic influence of alkali metal cations on rate increases in the order $\text{Li}^+ < \text{Na}^+ < \text{K}^+ < \text{Cs}^+$.

From these studies, it is evident that the ferro- ferricyanide couple is thermodynamically reversible, with no activation polarization involved in the electrode reaction. This electrode operates solely under mass transfer control.

2.5.3 Chemical Stability

The chemistry of saturated ferro- and ferricyanide in highly alkaline solutions has not been thoroughly studied; however, between pH 3 and 10, ferro- and ferricyanide are both inert with respect to exchange of bound cyanides for free radio-labeled cyanide (Ref. 33). Loss of cyanide from these complexes is, therefore, not spontaneous but must be promoted for decomposition to occur. Past work on the decomposition of hexacyano iron species has been conducted principally in acid solutions, where decomposition occurs by successive protonation of cyanide and loss of HCN, followed by aquation (Ref. 34). Loss of cyanide with formation of the aquopentacyano iron specie can occur in the presence of acid (Ref. 8, 33, 34), UV light, (Ref. 36), and certain metal ions such as Hg^{2+} and Ag^+ (Ref. 37). The first intermediate in these decompositions, $\text{Fe}(\text{CN})_5\text{OH}_2^{2,3-}$, is isolatable, and many pentacyano iron(III) complexes have been characterized (Ref. 38). Aquopentacyano iron (II) may also be prepared by reduction of nitroprusside (Ref. 35) and by solvolysis of aminopentacyano iron (II) (Ref. 54). In alkaline solution, aquopentacyano iron(II) and (III) behave as monoprotic acids (Ref. 39) and show appreciable stability. More highly aquated iron cyanides have been prepared (Ref. 40), but they are much less stable toward basic decomposition (Ref. 41).

Electrochemical (Ref. 42), substitution kinetic (Ref. 43), and electronic spectral (Ref. 44, 45) evidence has been presented in support of an equilibrium between monomeric and dimeric forms of aquopentacyano iron(II) and (III) in solution. Formation of dimers is favored by high ionic strength, low temperature, and high

$\text{Fe}(\text{CN})_5\text{OH}_2^{2,3-}$ concentration; addition of lithium salts, urea, and high pH favor the monomer (Ref. 46, 47). Equilibrium between monomer and dimer is established slowly enough that they may be separated chromatographically (Ref. 47, 48).

Other binuclear species formed by the association of penta- and hexacyano iron complexes have been prepared and characterized (Ref. 49, 50). Bridging $\text{Fe}-\text{C}\equiv\text{N}-\text{Fe}$ units are proposed for these complexes. In such a complex, the bridging CN would be expected to display an infrared stretching absorption shifted to higher frequency, in line with similar complexes of Co and Ni (Ref. 51, 52).

The photochemical decomposition of $\text{Fe}(\text{CN})_6^{4-}$ has also been studied. Excitation with light of 360 nm or less causes the loss of CN^- producing $\text{Fe}(\text{CN})_5^{3-}$, which aquates to $\text{Fe}(\text{CN})_5\text{OH}_2^{3-}$. If irradiation is stopped after a short time, $\text{Fe}(\text{CN})_5\text{OH}_2^{3-}$ and CN^- recombine to give ferrocyanide. Prolonged irradiation gives $\text{Fe}^{2+}(\text{aq})$ and HCN or, if oxygen is present, prussian blue (Ref. 53).

The involvement of pentacyano iron species in the decomposition of alkaline ferro- ferricyanide electrolyte has been verified experimentally by the development of a distinctive violet color ($\lambda_{\text{max}} = 430 \text{ nm}$) when reduced samples of old electrolyte are treated with nitrosobenzene in buffered solution. A spectrophotometric method based upon this color reaction (Ref. 8) allows the decomposition of the electrolyte to be followed during long-term cycling of operating cells.

Section 3

TECHNICAL APPROACH

3.1 CYCLIC VOLTAMMETRY

Initial substrate evaluations were conducted, using cyclic voltammetry, in the test setup shown in figure 3.1.1. The 5N NaOH electrolyte, saturated in ZnO or an electrochemically formed mixture of $\text{Na}_4\text{Fe}(\text{CN})_6$ and $\text{Na}_3\text{Fe}(\text{CN})_6$ was circulated at a nominal flow of 2.3 lpm and maintained at a temperature of 50°C. A platinum screen having a surface area of 100 cm² was used as the counterelectrode, while the working electrode, the candidate substrate as typically shown in Figure 3.1.2, was installed perpendicular to the electrolyte flow path. The substrate, having a surface area reduced to approximately 0.5 cm² by Microflex stop-off lacquer was electrically monitored using a Hg/HgO reference electrode (0.098 V vs SHE) via a Luggin capillary placed 1-2 mm from its surface.

Cyclic voltammetric studies were carried out using the instrumentation shown in figure 3.1.3 which consists of a Princeton Applied Research (PAR) model 173 Potentiostat/Galvanostat equipped with a PAR model 176 Current Follower and controlled by a PAR model 175 Universal Programmer operating at 10 mV/sec scan speed. The current and voltage were monitored using a Houston Instruments model 2000 Omnigraphic X-Y recorder.

Typical methodology for the Zinc investigation consisted of starting the scan at OCV for the substrate, scanning negative to -1.30 V vs SHE, reversal of the scan direction, followed by positive scanning to -1.00 V vs SHE. Multiple scans were performed in order to determine any surface effects caused by the substrate.

Similarly, the methodology for the Redox electrode involved multiple cyclic scanning from OCV in a positive direction to 0.8 V vs SHE, reversal of the scan direction, followed by negative scanning to -0.4 V vs SHE.

3.2 TEST STAND AND CELL CONFIGURATION

The test stand, shown in figure 3.2.1, consists of six cell test setups with a common thermostated sandbox for electrolyte thermoconditioning and a water bath for general thermal control via a YSI model 74 Temperature Controller using

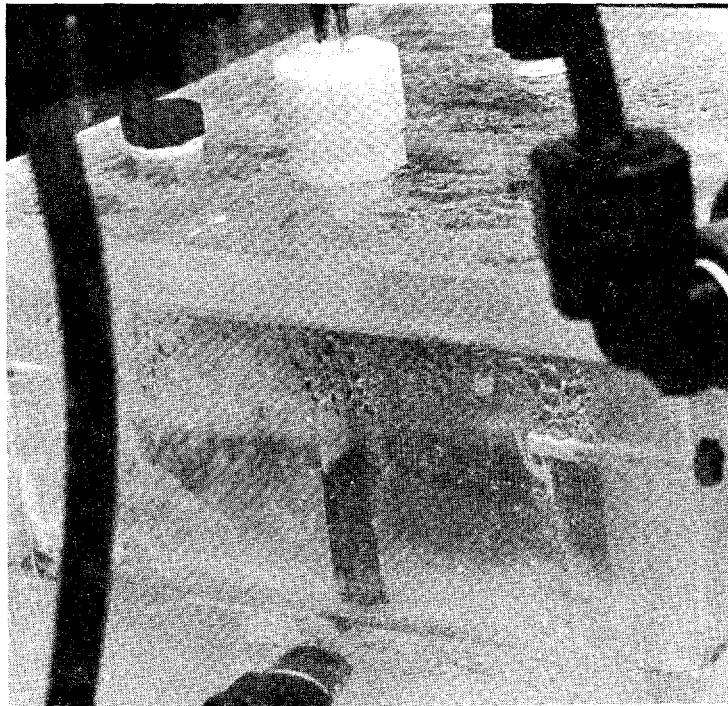


Figure 3.1.1 Cyclic Voltammetry Test Cell

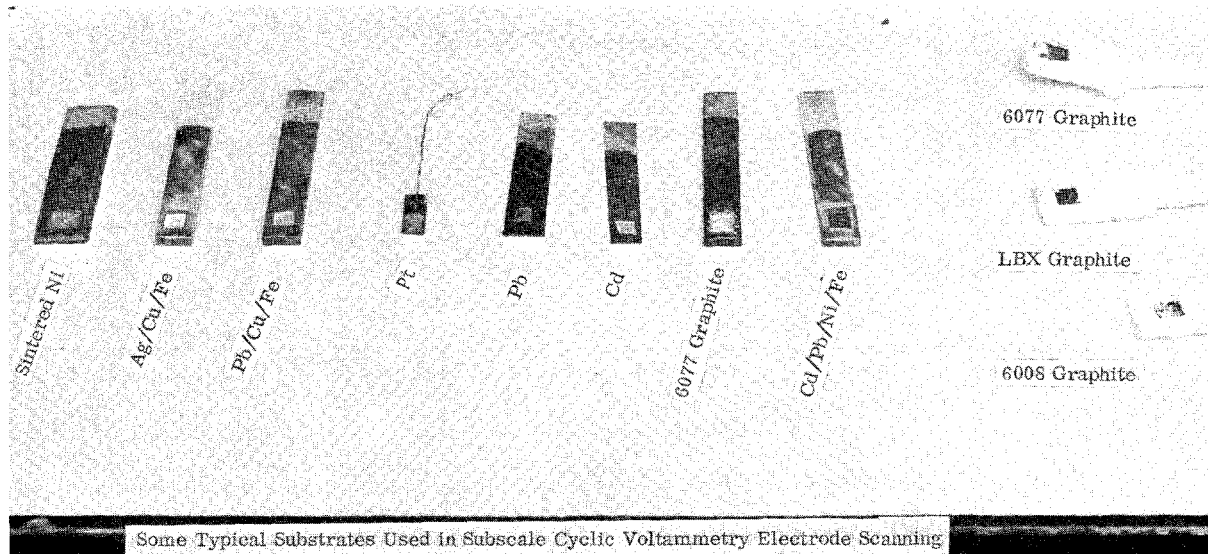


Figure 3.1.2 Typical Cyclic Voltammetry Substrates

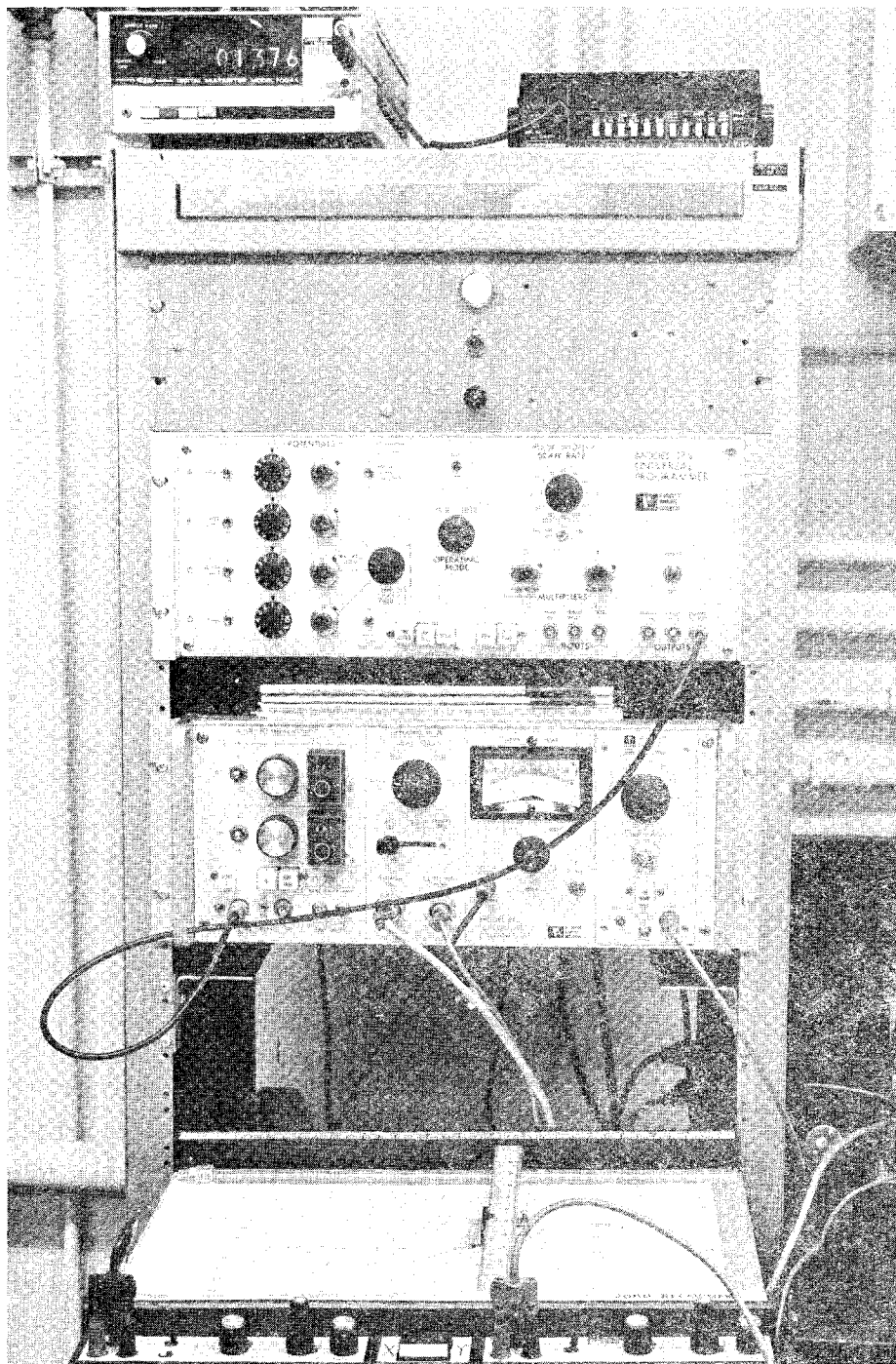


Figure 3.1.3 Cyclic Voltammetry Instrumentation

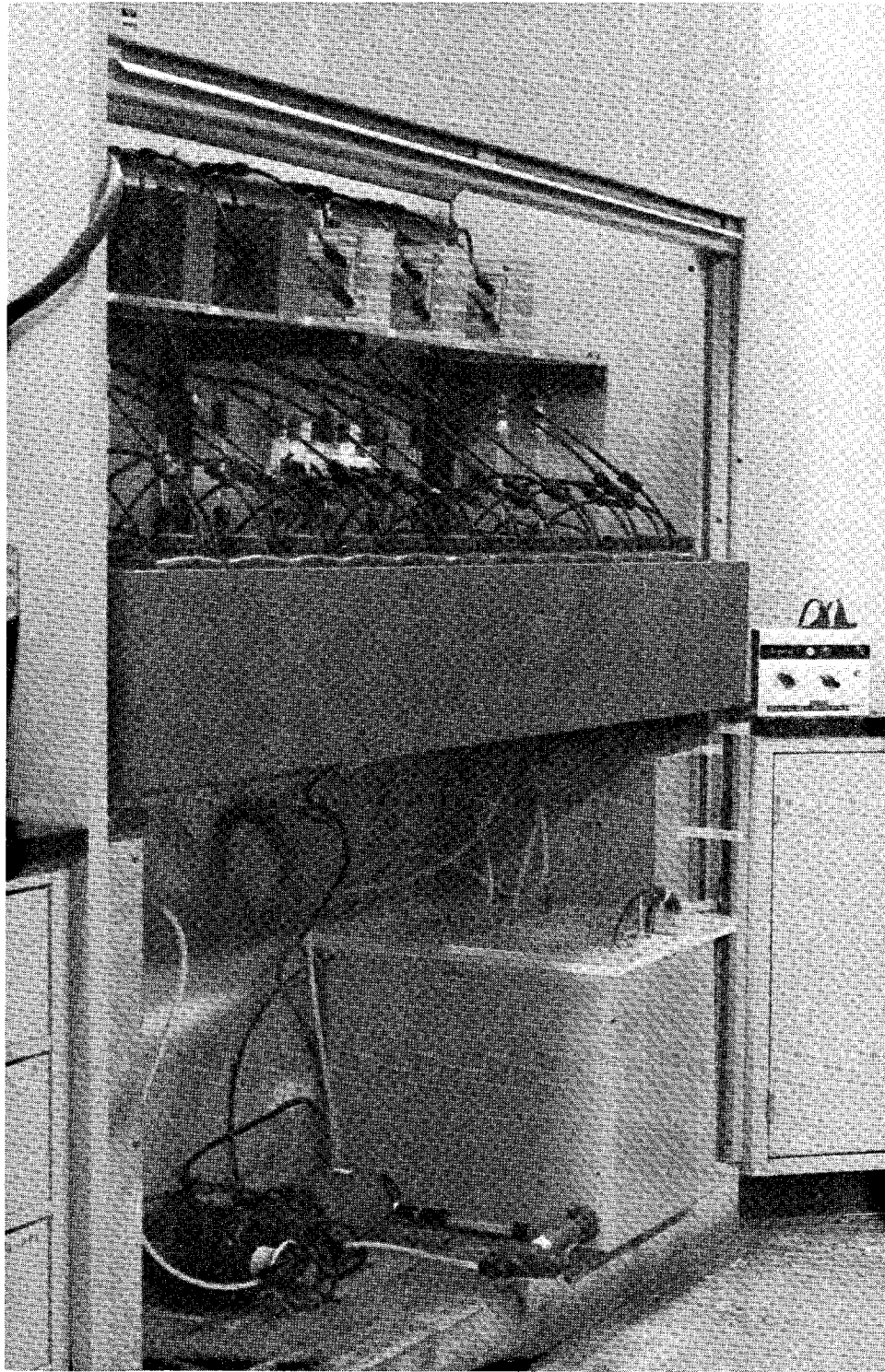


Figure 3.2.1 Test Stand

a YSI model 416 thermister probe embedded centrally in the sandbox. The five-gallon water bath is equipped with constant level control, a 1000-watt Glo-Quartz Type L1000 quartz heater on 100% duty cycle, a second 1000 watt quartz heater whose duty cycle is controlled by the temperature controller which also controls cooling via a Skinner model V52DA2100 solenoid valve actuating cold water flow through a stainless steel cooling coil, and an Eastern model D-6 pump to circulate the water through 3/8" copper tubing within the sandbox and then returning through the cell's acrylic thermo-jacketing.

Each electrolyte loop consists of an electrolyte reservoir and filter housing (Figure 3.2.2) embedded in sand, a pump, a valve to control flow rate and a flowmeter. The one-liter opaque polyethylene bottle reservoir, plastic filter housing containing a five-inch-long medium-density (25-50 micron) polypropylene web cartridge, March model MDX-MT3 Ryton pump, 1/2" PVC globe valve, and Gilmont size 5 flowmeter are all connected within the electrolyte loop with 3/8" O.D. opaque polyethylene tubing using opaque polypropylene Parker fittings.

Figure 3.2.3 shows a close-up of a 60 cm² disassembled cell detailing the sintered nickel electrode on the left, the Hg/HgO reference electrode and Luggin capillary, the cadmium-plated iron substrate on the right, and two of the separators used in this study on top of the cell. The cells are configured with internal thermo-jacketed acrylic endplates for maintaining the cells at desired set operating temperatures and are coated with opaque lacquer to eliminate photochemical decomposition of the redox electrolyte.

The sintered nickel electrodes for the flowing electrolyte ferro-ferricyanide redox couple, obtained from the Clevite Corp., are 3.175-mm thick sheets formed by pressing nickel carbonyl powder onto a nickel screen and sintering in a hydrogen atmosphere. The sheets are cut to 6.35 x 10.15 cm and an electrode stud strip spot-welded to it for electrical contact. The electrode stud strip is comprised of a 10 x 1 x 0.0635 cm nickel sheet to which two 4.76 mm diameter steel studs, 5.33 cm long, are silver-soldered 7.36 cm apart and are threaded to receive #10-32 nuts for the last 1.27 cm. The area around the silver-solder joint is coated with Trucor chemical milling mask to prevent contact with the electrolyte.

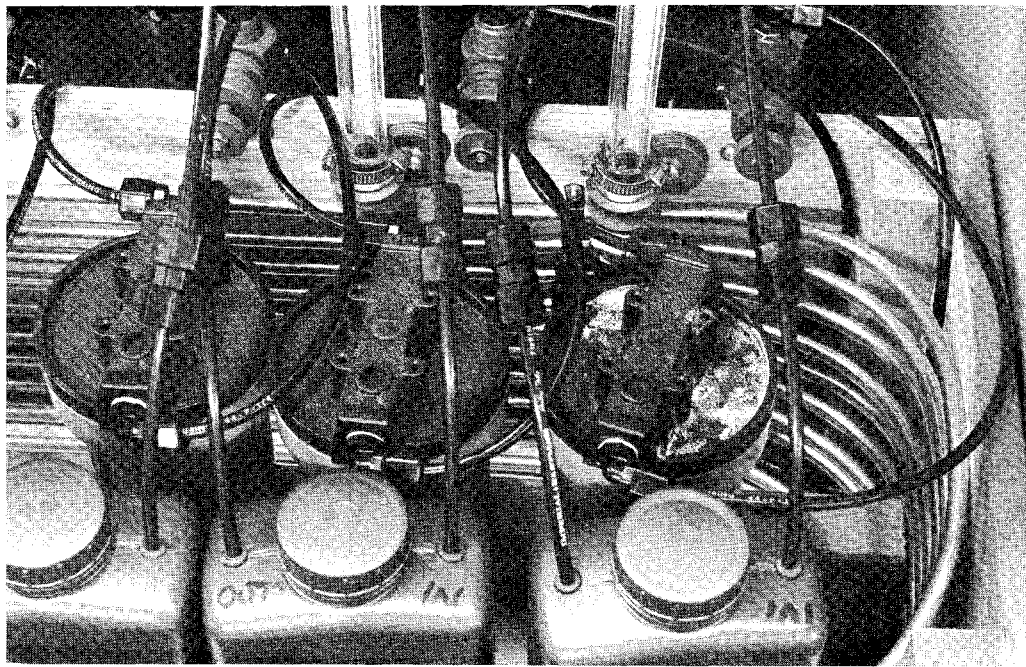


Figure 3.2.2 Section of Sandbox showing electrolyte reservoir, filter housing and copper cooling coils

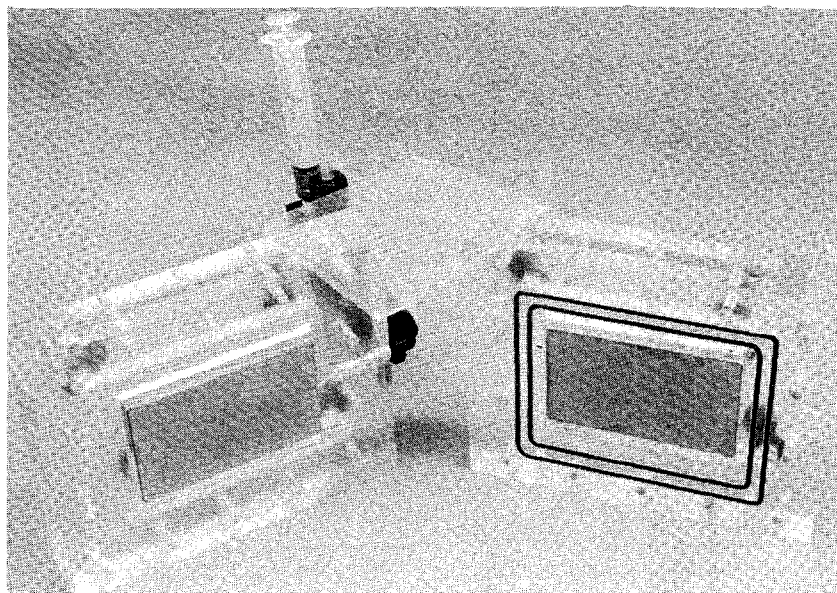


Figure 3.2.3 Disassembled Test Cell (60 cm^2)

The zinc electrode is constructed from 2.54-mm thick iron which is cut to 5.97 x 9.98 cm. A 6.35-mm hole is drilled at the geometric center and a 6.35-mm diameter stud, 5.33-cm long, is silver-soldered in place while the outer 1.27 cm of the stud is threaded for a #1/4-28 nut. The entire backside, the stud side, is coated with Trucor chemical milling mask for electrical insulation. The front is sandblasted and 25 microns of cadmium is electrodeposited from a cyanide plating solution.

The reference electrode, a standard Hg/HgO (0.098 vs SHE), is encased in a separate acrylic holder, shown in figure 3.2.4, and is comprised of a platinum wire contacting a pool of mercury which is in contact with a paste formed by mixing mercury and mercuric oxide with 5N NaOH. The sodium hydroxide acts as the supporting electrolyte in contact with the redox electrolyte via the Luggin capillary. The syringe, on top of the reference electrode compartment, is for clearing air blockages from the capillary.

3.3 MICROCOMPUTER SYSTEM AND INSTRUMENTATION

A major effort of this project was the design, construction, and implementation of a microcomputer system for complete control of half- and full-cell long-term cyclic testing including preliminary data reduction. The system is composed of two consoles having six basic modules. These are the microcomputer; the keypad and display panel; the data output and storage devices; the current sources panel; the current source power supplies; and the cell parameter selector panel for analog recorder output. Figure 3.3.1 shows the microcomputer console with microcomputer, keypad/display panel, teletype, and digital tape cassette recorder. Figure 3.3.2 shows the power source console with the cell parameter panel, the current source panel, and power supplies for the constant current sources.

The National Semiconductor Model RMC 80/11 microcomputer chassis contains the single board microcomputer (μ C) board, a Burr-Brown model MP8632 8-bit, 32 differential channel analog to digital (A/D) converter board, and two Electronic Solutions, Inc. model RAM-8 8K-byte Random Access Memory (RAM) boards. The

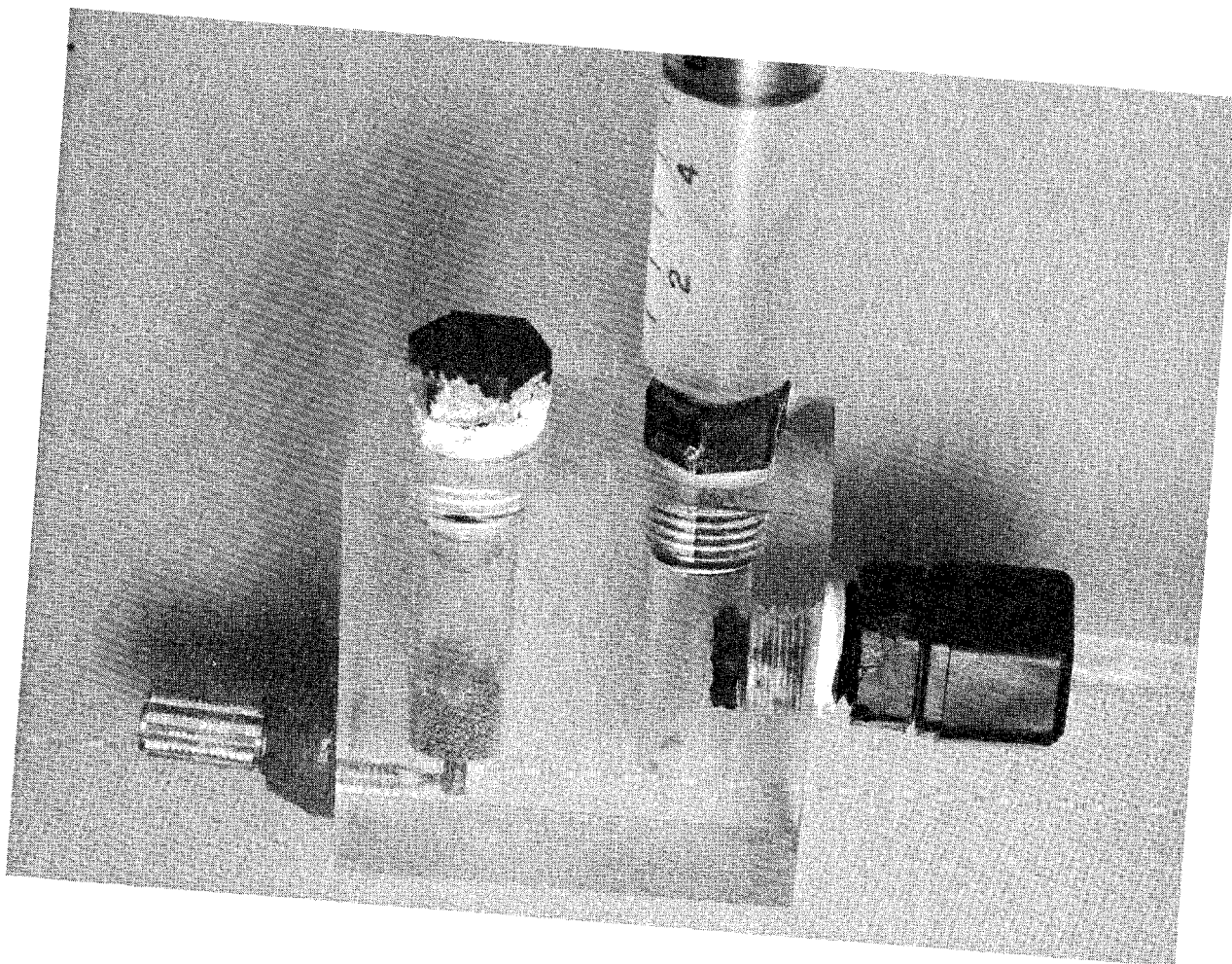


Figure 3.2.4 Acrylic Hg/HgO reference electrode holder



Figure 3.3.1 Microcomputer Console

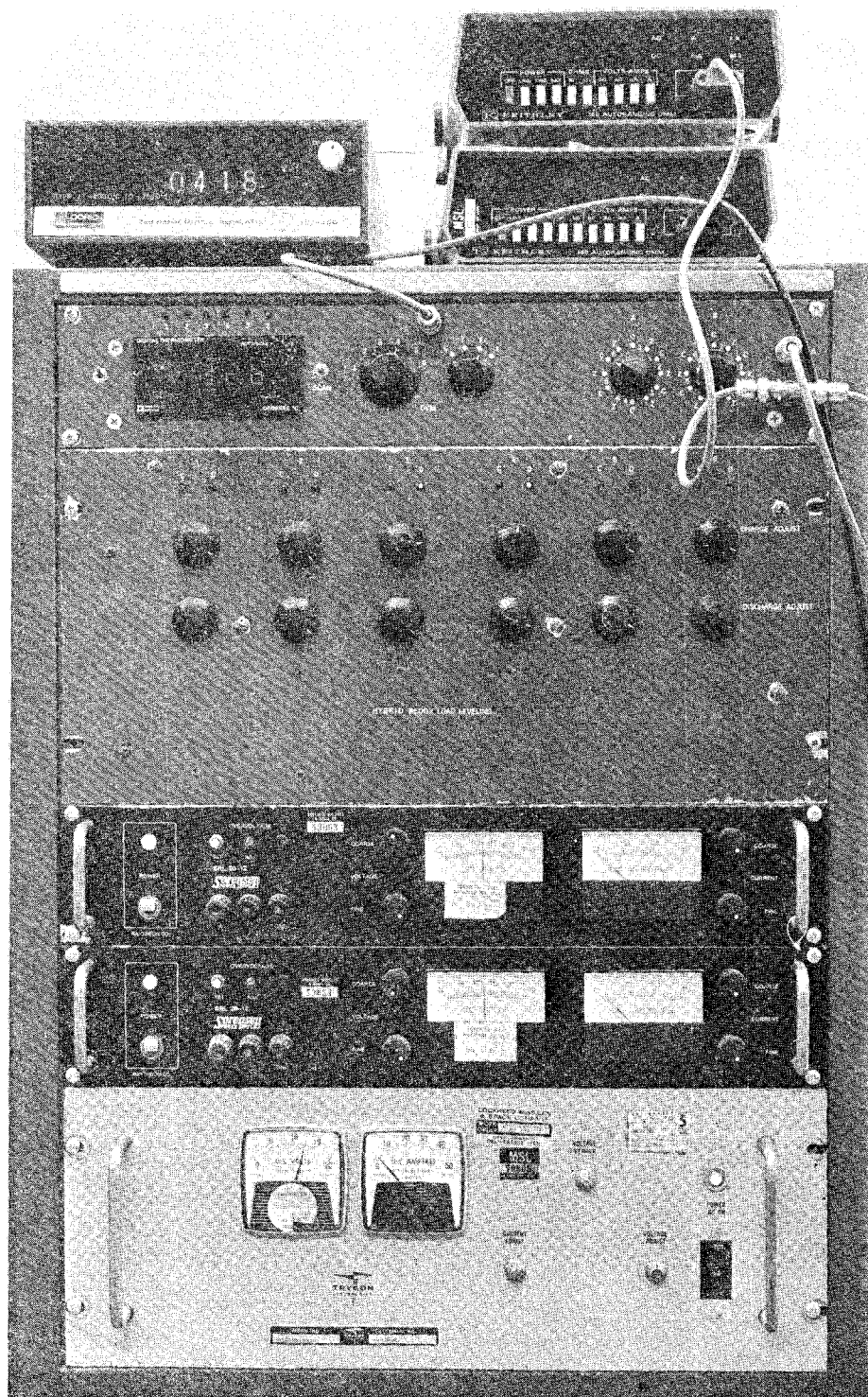


Figure 3.3.2 Power Source Console

present total memory is 8 Kbyte of UV-Erasable Programmable Read Only Memory (E-PROM) located on the μ C board and 17 Kbyte of RAM with 1K located on the μ C board.

The Keypad/display panel contains the keypad (circuitry shown in Appendix A) for operator control of the cells and for setting operational parameters; a 12-digit alphanumeric display (circuitry detailed in Appendix A) enabling the operator to monitor cell operation, setpoints, and cell parameters; and a real-time clock (circuit shown in Appendix A) for continuous update of time while the microcomputer makes various time-consuming excursions for data input and output. The microcomputer controls the cell operation via optically isolated relays which turn on constant current sources to charge or discharge the cells. The current source (circuitry shown in Appendix A) is a 3-amp adjustable voltage regulator used in the constant current mode with current adjusted by a variable potentiometer on the current source panel. Feedback to the microcomputer is provided by input into the A/D converter, of cell potential, zinc-reference potential, iron-reference potential, and current, as potentials developed across a precision current shunts and amplified by an instrumentation amplifier as the circuitry in Appendix A details.

The control program, written in "Tiny Basic" with slower portions written in assembly language consists of over 424 lines of Tiny Basic code (Appendix B) and 484 lines of assembly code (Appendix C) and utilizes virtually all of the available 17 Kbytes of RAM while the Tiny Basic interpreter resides in 3 Kbytes of E-PROM. The program is capable of controlling any number of batteries simultaneously but is limited to six with the present input/output devices and RAM size. The microcomputer controls the charge/discharge cycles of the rechargeable cells based upon the operator-determined cell parameters of charge time, overcharge potential, usable energy potential, zinc depletion potential versus a reference electrode. The current is integrated in each computer cycle and the amp-hours are continually updated so that at the end of the charge/discharge cycle the voltaic, coulombic, and energy efficiency is computed. These efficiencies and the cell parameters, voltage and current, at several points in the cycle are then data logged to a teletype and onto a digital tape

cassette recorder for later data manipulation and graphic representation using an IBM 360 Tymshare computer. The logic flow diagram for the control program is detailed in Figure 3.3.3. A more concise description of the microcomputer and its operations is given in Reference 56.

3.4 ELECTROLYTE PREPARATION

The redox electrolyte was prepared using reagent grade sodium hydroxide obtained from J. T. Baker, sodium ferrocyanide decahydrate (99%) from Research Inorganic Chemical Co., and deionized water.

The zincate electrolyte was prepared using reagent grade sodium hydroxide and zinc oxide obtained from J. T. Baker, and deionized water.

Reagents were introduced into the electrolyte reservoir and thoroughly mixed by pumping through the electrolyte loop to obtain saturation and thermal equilibrium.

3.5 EXPERIMENTAL METHODOLOGY

The electrolytes are circulated for 24 hours to achieve equilibrium prior to cell assembly. The cell is assembled in the following steps:

1. Install o-ring on electrical stud of zinc substrate.
2. Insert zinc substrate into position in acrylic half-cell.
3. Install o-ring, acrylic seal, and nut on electric stud then seal tight.
4. Repeat above procedures with the sintered nickel electrode modified by using several coats of Trucor chemical milling mask to seal the edges to prevent liquid leakage.
5. Cut and install a wet separator of proper dimensions onto o-ring of zinc half-cell.

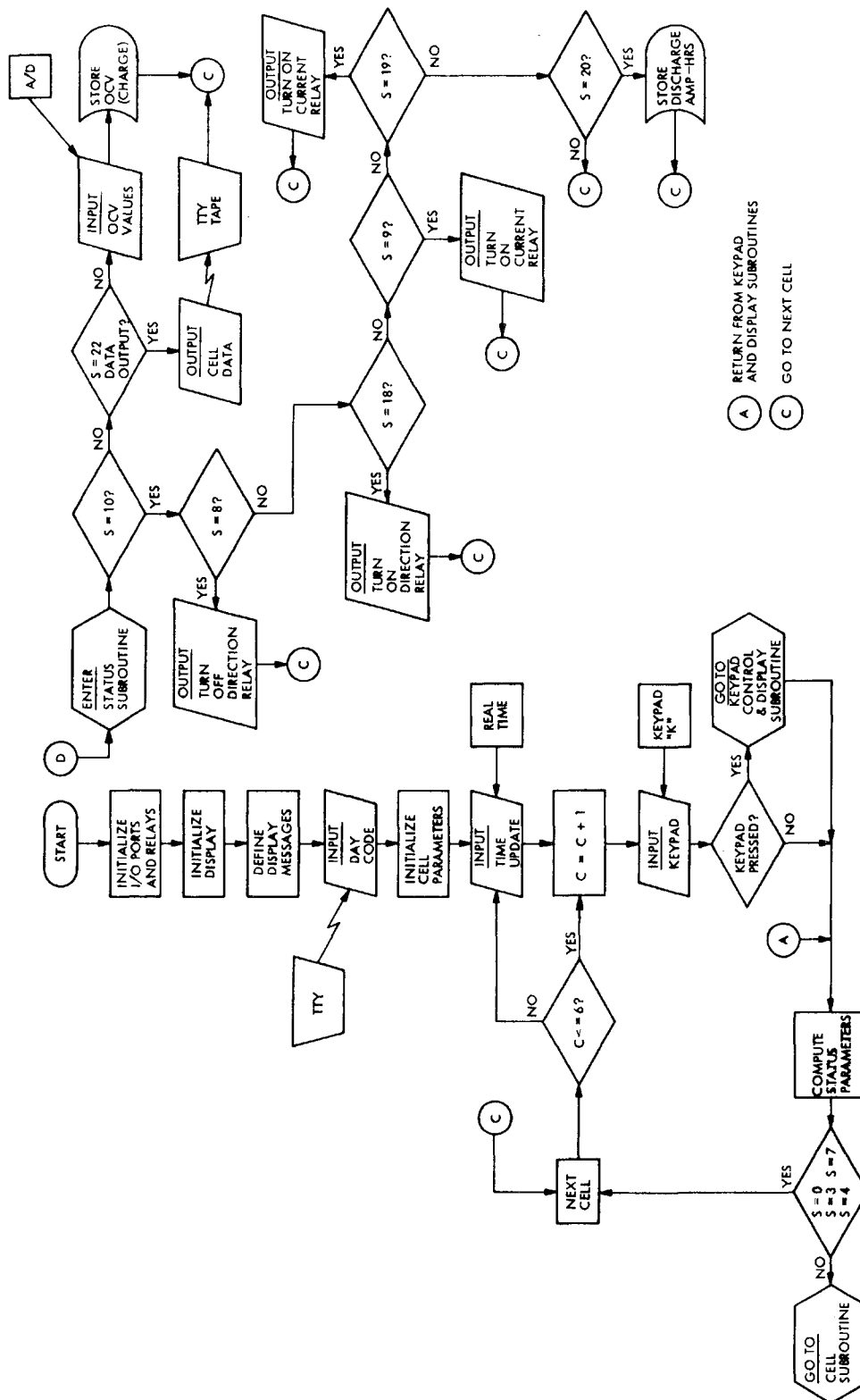


Figure 3.3.3 Microcomputer control program
Logic flow diagram (Start and Status Subroutine)

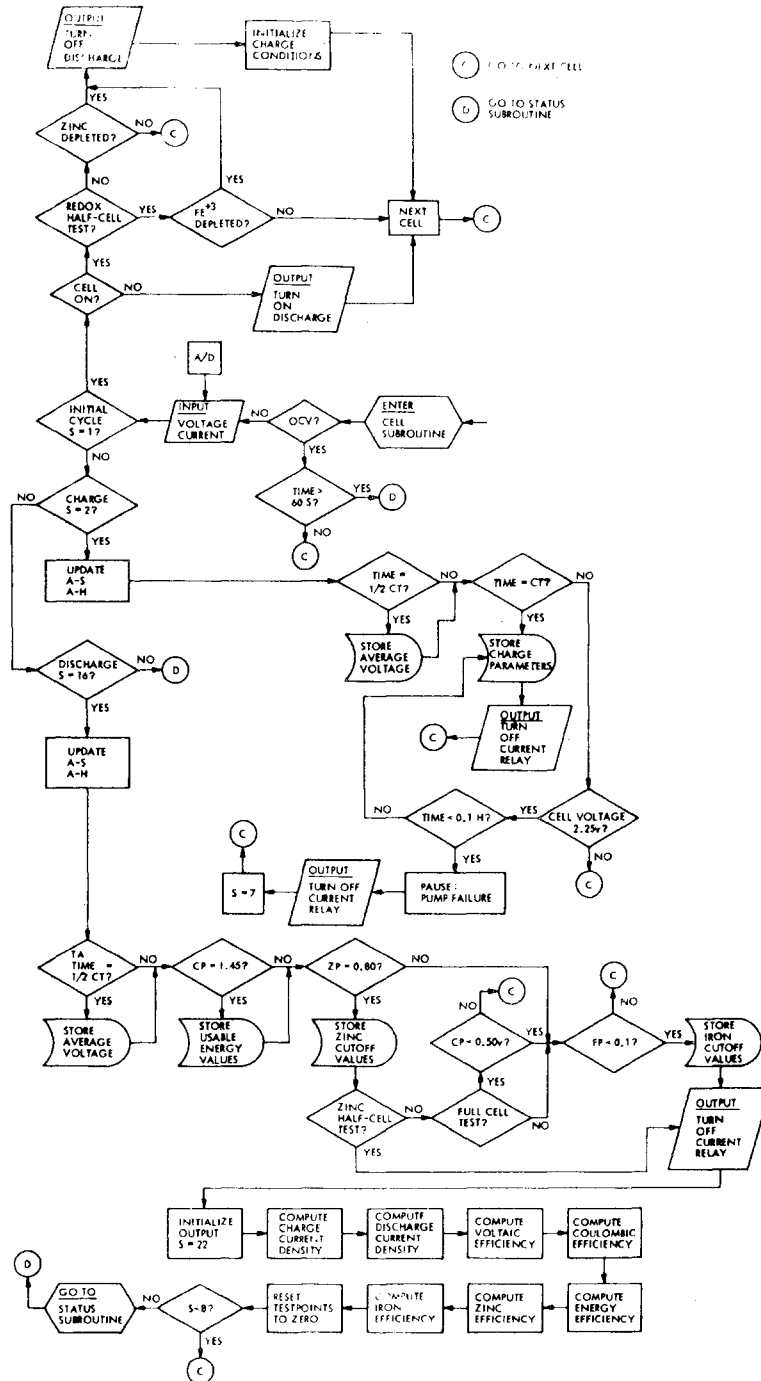


Figure 3.3.3 (continued) (Cell Control Subroutine)

6. Complete assembly with 14 bolts and nuts for total closure.
7. Install thermojacket water tube between two halves.
8. Install Hg/HgO reference electrode.
9. Place on test stand and complete electrolyte linkages and electrical hook-ups, and thermojacketing linkages.

A close-up of a cell assembled and in position for testing is shown in Figure 3.5.1. Separate wires are connected to each electrode to carry the current and to monitor voltage.

Unless otherwise noted, all testing was done at 40°C with equal charge/discharge periods of 2.0 hours; an overcharge potential of 2.25 volts for full cell tests and 2.50 volts for half-cell tests; an arbitrary usable energy potential of 1.45 V; a zinc depletion cutoff potential vs Hg/HgO of 0.80 V; and an iron depletion cutoff potential vs Hg/HgO of 0.10 V. Redox electrolyte flow rate was nominally 8 cm/sec while the zincate electrolyte was nominally 10 cm/sec, decreasing during charge due to decreased cell volume.

3.6 ANALYTICAL PROCEDURES

Chemical analyses of electrolytes were made using the following procedures:

Hydroxide - A standard acid-base titration with 1 N HCl, obtained from Ricca Chemical Co. and having NBS traceable standardization, to a phenolphthalein endpoint

Ferrocyanide - A standard ceric sulfate titration of an acidified (4M H_2SO_4) aliquot with 0.1 N ceric sulfate, obtained from Fischer Scientific and having NBS traceable standardization, to a ferroin endpoint

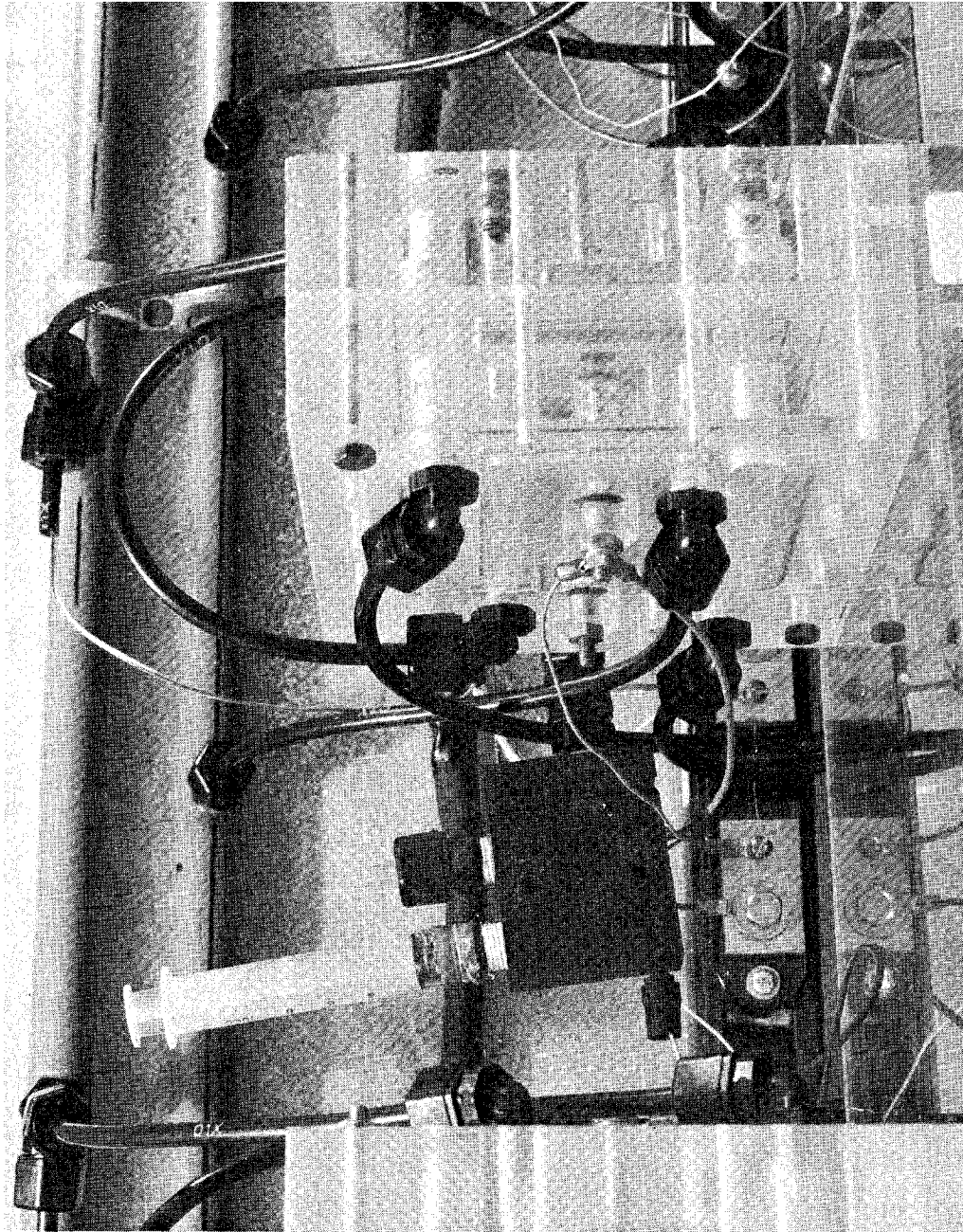


Figure 3.5.1 Close-up of 60 cm² cell undergoing cyclic testing

Ferricyanide - A standard indirect iodine-starch titration of the acidified (4M H_2SO_4) aliquot with 0.1 N sodium thiosulfate, obtained from Ricca Chemical Co. and having NBS traceable standardization, to a starch endpoint.

Zincate - Atomic absorption analysis of the acidified supernatant at 3075.9 Å using the method of standard additions on a Varian model 1200 Atomic Absorption Spectrophotometer.

Aquopentacyano iron (II or III)

Concentrations of aquopentacyano iron(II) and(III) were determined by an adaptation of a published method involving spectrophotometric determination of the violet complex with nitrosobenzene $[\text{Fe}(\text{CN})_5\text{C}_6\text{H}_5\text{NO}]^{3-}$ (ref 35). Because Fe (III) oxidizes nitrosobenzene, and prevents the violet complex from forming, samples must be treated with a noncoordinating reducing agent before adding the nitrosobenzene reagent. For the alkaline samples usually studied, 37% formalin solution was a convenient and effective reducing agent which did not interfere with the color reaction. After reduction of a sample was judged complete by the disappearance of any yellow color, the pH of the solution was adjusted to 9.9 in a carbonate buffer, and treated with at least a two-fold excess of nitrosobenzene as a 0.4% ethanol solution. After at least one hour, the sample was diluted and its absorbance at 5300 Å was measured on a Cary model 14 UV-visible spectrophotometer. Absorbance was related to $[\text{Fe}(\text{CN})_5\text{C}_6\text{H}_5\text{NO}]^{3-}$ concentration through a standard curve. Standards were prepared from $\text{Na}_2\text{NH}_4\text{Fe}(\text{CN})_5\text{NH}_3$ followed by confirmation of Fe concentration by atomic absorption analysis.

Infrared solution spectra were recorded in a CaF_2 cell using a Nicolet 7000 FTIR system. The spectrum of a NaOH blank was subtracted from that of the samples dissolved in NaOH solution before the spectrum was plotted.

3.7 BASELINE CALCULATIONS AND DATA PROCESSING

The calculations performed by the microcomputer are the voltaic, coulombic, and energy efficiencies of the full cell, and coulombic efficiency of each half-cell. The voltaic efficiency is calculated by dividing the cell potential at half discharge by the cell potential at half charge. The coulombic efficiencies are computed by dividing the amp-hours discharge by the amp-hours charge. The amp-hours discharge is picked up at three points during the discharge cycle; the usable energy potential of 1.45 volts; the zinc depletion potential of 0.80 volts vs Hg/HgO; and the iron depletion potential of 0.10 volts vs Hg/HgO. The usable energy efficiency for the cell is calculated by multiplying the voltaic efficiency by the usable energy coulombic efficiency. Another calculation done by the microcomputer is the average current density on charge and discharge, which is computed by dividing the accumulated amp-hours by the total hours and 60 cm² to report a current density in units of mA/cm². It should be noted that the voltaic and energy efficiencies obtained from half-cells are not meaningful in an absolute sense, since the voltage difference measured is from the electrode to a reference electrode. In a relative sense, they are valuable for following single-electrode performance with cycle life. The product of the two half-cell voltaic, coulombic, or energy efficiencies, will, of course, give the true corresponding single cell efficiency. At the end of each discharge cycle, cell data is outputted to a Texas Instruments model 735 silent terminal and a Tymshare model 600 digital data cassette recorder. Typical cell data output by the microcomputer is shown in figure 3.7.1 with the following identification.

a	b	c	d	e	f	g	h	i	j	k	l	
79/253	12:41:55	6	606	2.0	37.0	36.7	91.5	96.8	88.3	97.7	97.7	
1.43	0.47	1.89	1.82	1.31	0.43	1.13	0.33	0.78	0.23	0.78	0.23	1.70
m	n	o	p	q	r	s	t	u	v	w	x	y

Figure 3.7.1 Typical Cell Data Output by Microcomputer

First line

- | | |
|---------------------------|------------------------------|
| a. Year/Day | g. Current density discharge |
| b. Time | h. Voltaic efficiency |
| c. Cell number | i. Coulombic efficiency |
| d. Number of cycles | j. Usable energy efficiency |
| e. Charge time (hours) | k. Zinc coulombic efficiency |
| f. Current density charge | l. Iron coulombic efficiency |

Second Line

- m. Zinc potential vs Hg/HgO at 1/2 charged
- n. Iron potential vs Hg/HgO at 1/2 charged
- o. Cell potential at end of charge
- p. Open circuit potential charged
- q. Zinc potential vs Hg/HgO at 1/2 discharged
- r. Iron " at 1/2 discharged
- s. Zinc " at usable energy point of discharge
- t. Iron " at usable energy point of discharge
- u. Zinc " at Zinc depletion point of discharge
- v. Iron " at Zinc depletion point of discharge
- w. Zinc " at Iron depletion point of discharge
- x. Iron " at Iron depletion point of discharge
- y. Open circuit potential discharged

Due to the 8-bit resolution of the A/D converter the accuracy of all potentials is ± 20 mV while the accuracy of the efficiency calculations is $\pm 10\%$.

Every four days, or when necessary, the data stored on the digital cassette is transferred to a Tymshare IBM 360 computer. At the end of each month the data is sorted into individual cell data using the program detailed in Appendix D. After merging the month's data for a particular cell with data from previous months of testing, the cell data report program, detailed in Appendix E, is used to itemize and average the accumulated cell data and then generate a report form as shown in Appendix F for the cycle life test in progress. The program also calculates statistical averages and denotes when a particular cycle energy efficiency exceeds 1σ , 2σ , or 3σ 's. Cycles exceeding 3σ are not included in the calculations and are indicative of thermal control, microcomputer control, reference electrode plugging, or pump malfunction problems.

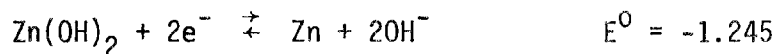
The report program creates a reduced data file which a plotting program, Appendix G, uses to compute a 5, 7, or 11-point smoothing of the data and creates a

plot data file. The plot data file consists of x and y values for cycle number and efficiency, respectively, which are then used to create a final graph on a Zeta model 500 digital plotter using the plotting routines shown in Appendix H. The program prints a point for each individual cycle and then draws a line which represents 5- or 7-point smoothing, less than 100 cycles, or 11-point smoothing, for more than 100 cycles of data.

Section 4
RESULTS AND DISCUSSION

4.1 ZINC SUBSTRATE EVALUATION

Using those factors affecting zinc-zincate electrode substrate selection discussed in Section 2.3.1 as guidelines, the following substrates were evaluated for polarization characteristics by cyclic voltammetry, with platinum included as a control: Pt, Cu, Ag, Pb, Sn, Cd, and four types of anode-grade graphite. Typical cyclic voltammograms at 10 mV/s are shown in figures 4.1.1 - 4.1.7 and are indicative of the reaction:



Other effects are indicated by several of the voltammograms such as the formation of a copper-zinc alloy on the copper substrate, Figure 4.1.1, and a porous silver coating on the silver substrate prepared by electrodeposition of silver onto iron with an intermediate copper strike. Evidence for copper-zinc alloy appears as a shoulder at -1.2V in figure 4.1.2. Platinum, figure 4.1.3, is known to exhibit a low hydrogen overvoltage which the cyclic voltammogram confirms. Onset of hydrogen evolution occurs at -.95 V vs SHE in alkaline zincate electrolyte which compares with the standard potential of -0.828 V in alkaline electrolyte. The graphites all present very poor performances except for the Stackpole 6077 graphite, figure 4.1.7, commonly used in the chlor-alkali industry. The other graphites evaluated were all from Stackpole Carbon Co. and consisted of 6015 and 6008 graphites and baked LXB carbon. The remaining cyclic voltammograms were similar and indicative of good zinc cyclic conditions. Figure 4.1.8 details the polarization characteristics of the candidate substrates which shows decreasing overvoltage on charge for the sequence lead, tin, cadmium, and 6077 graphite, while on discharge the order is slightly different; namely, lead, cadmium, tin, and then 6077 graphite.

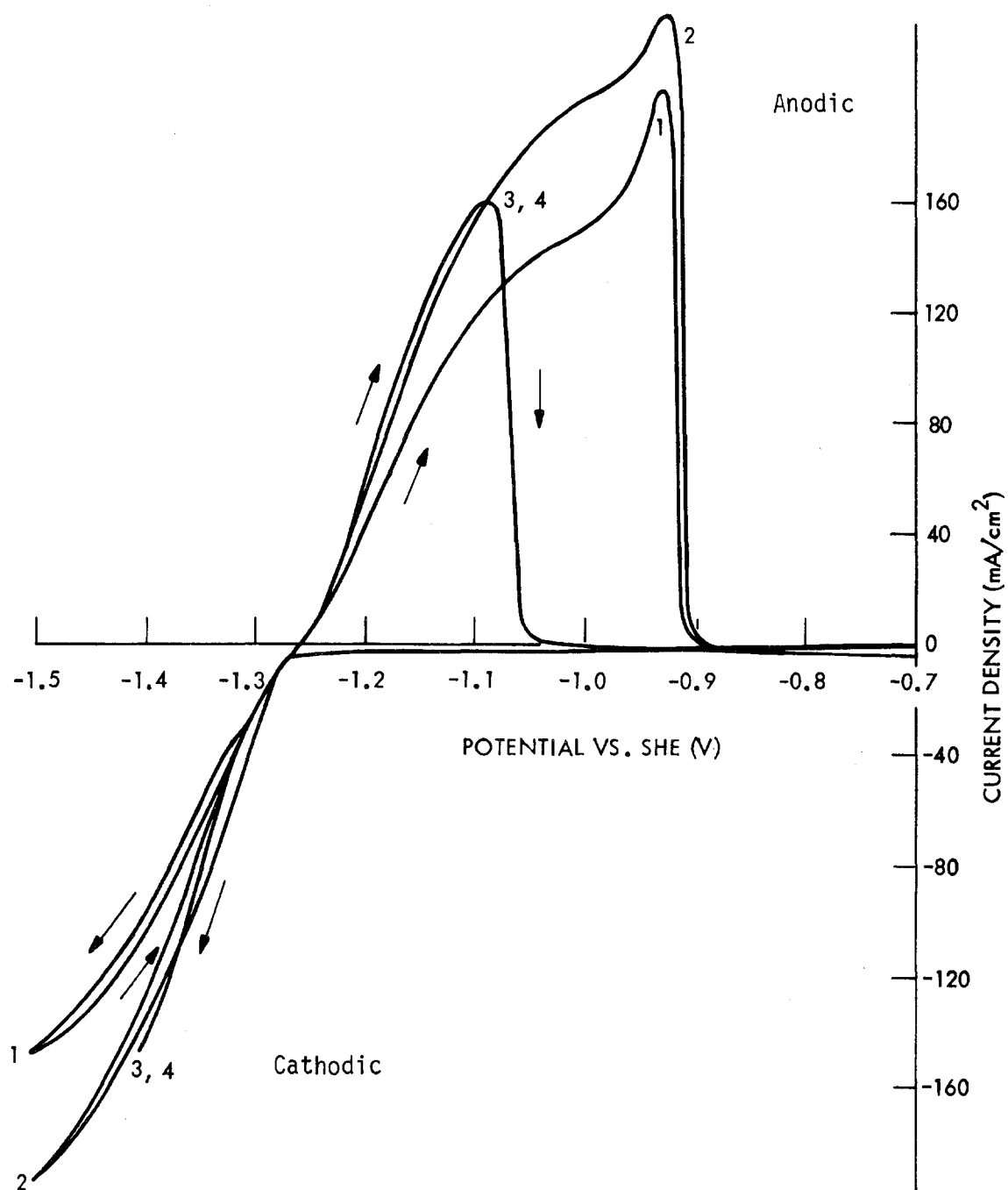


Figure 4.1.1 Zinc cyclic Voltammetry on Copper

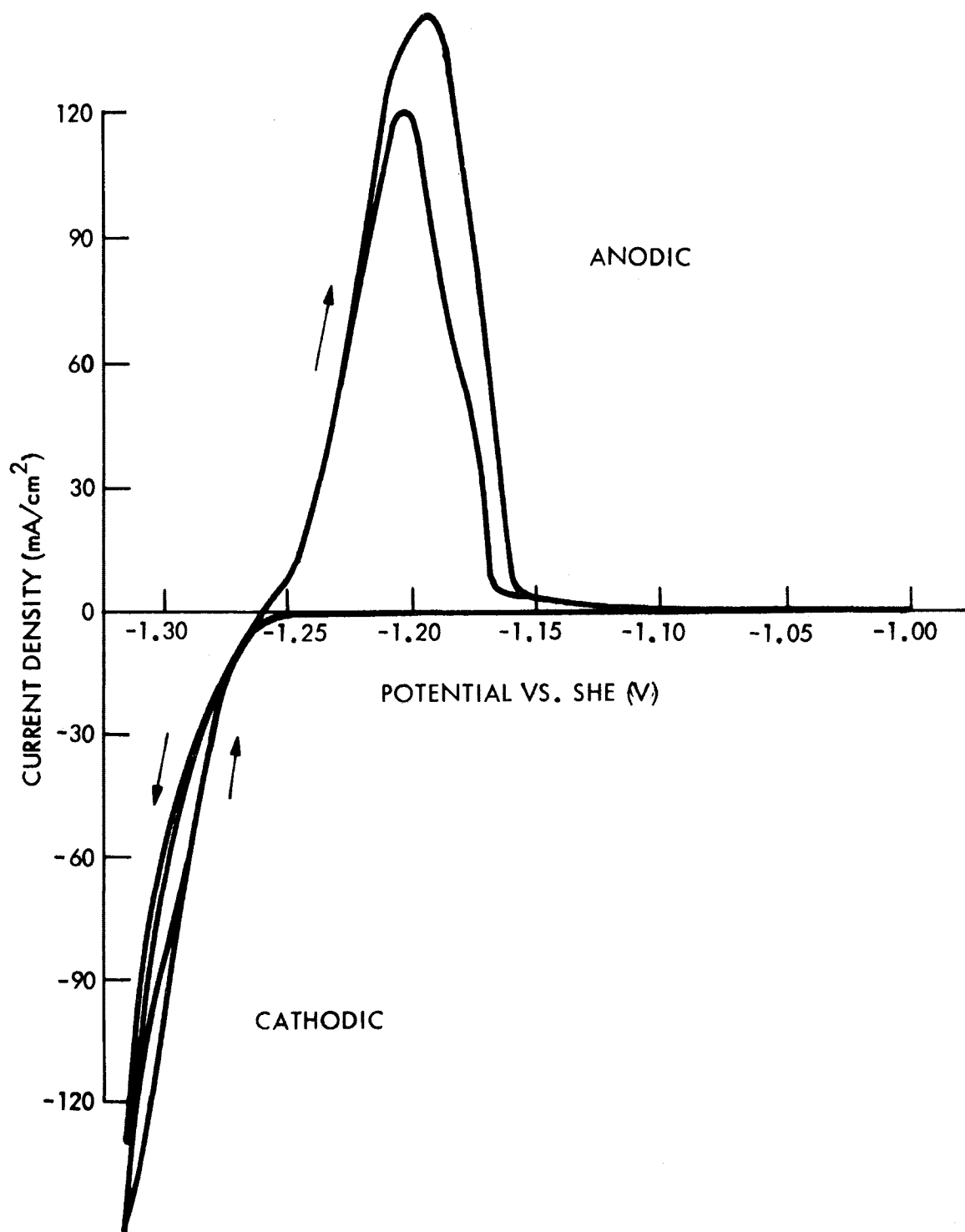


Figure 4.1.2 Zinc Cyclic Voltammetry on Electrodeposited Silver(Ag/Cu/Fe)

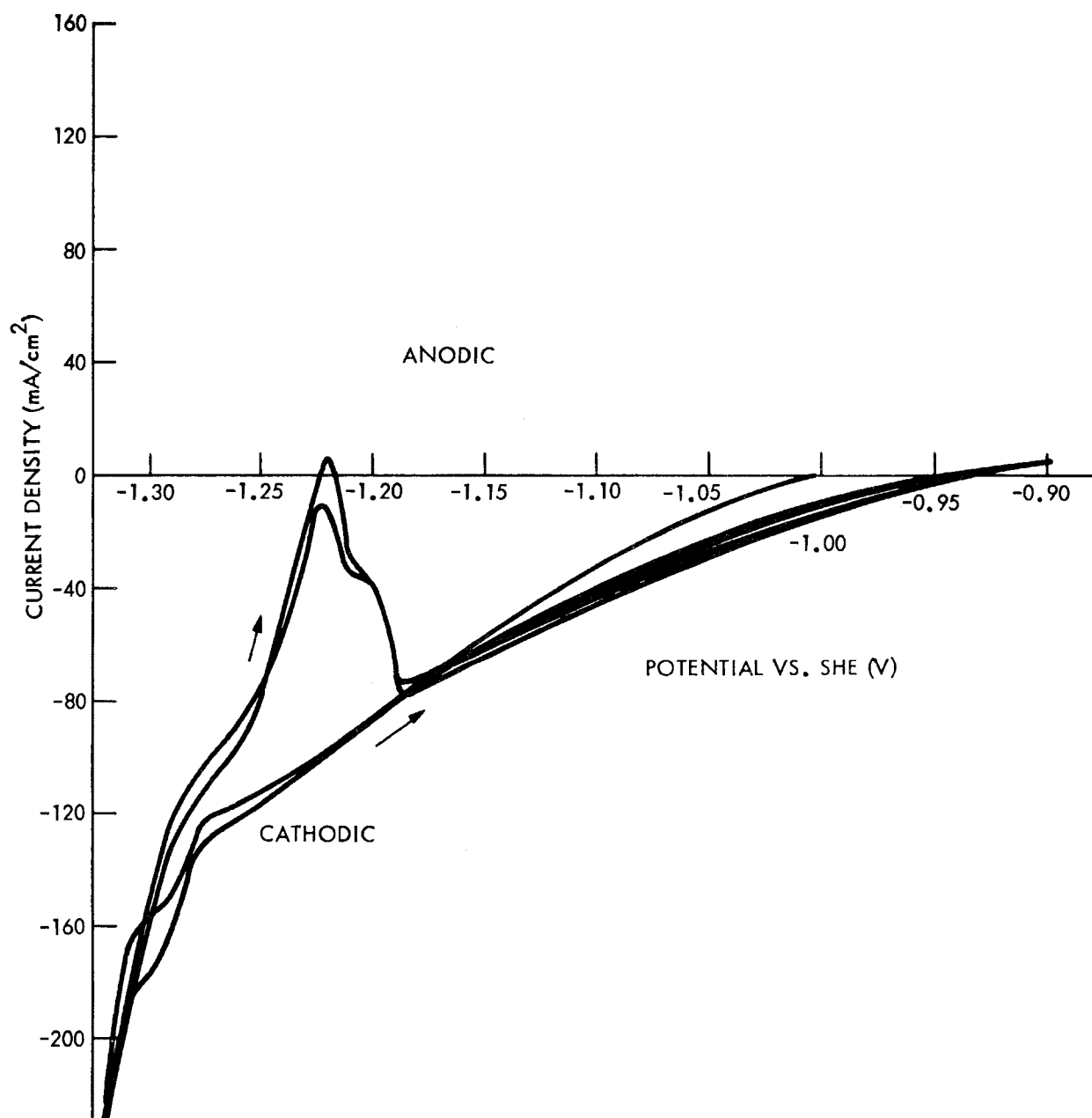


Figure 4.1.3 Zinc Cyclic Voltammetry on Platinum

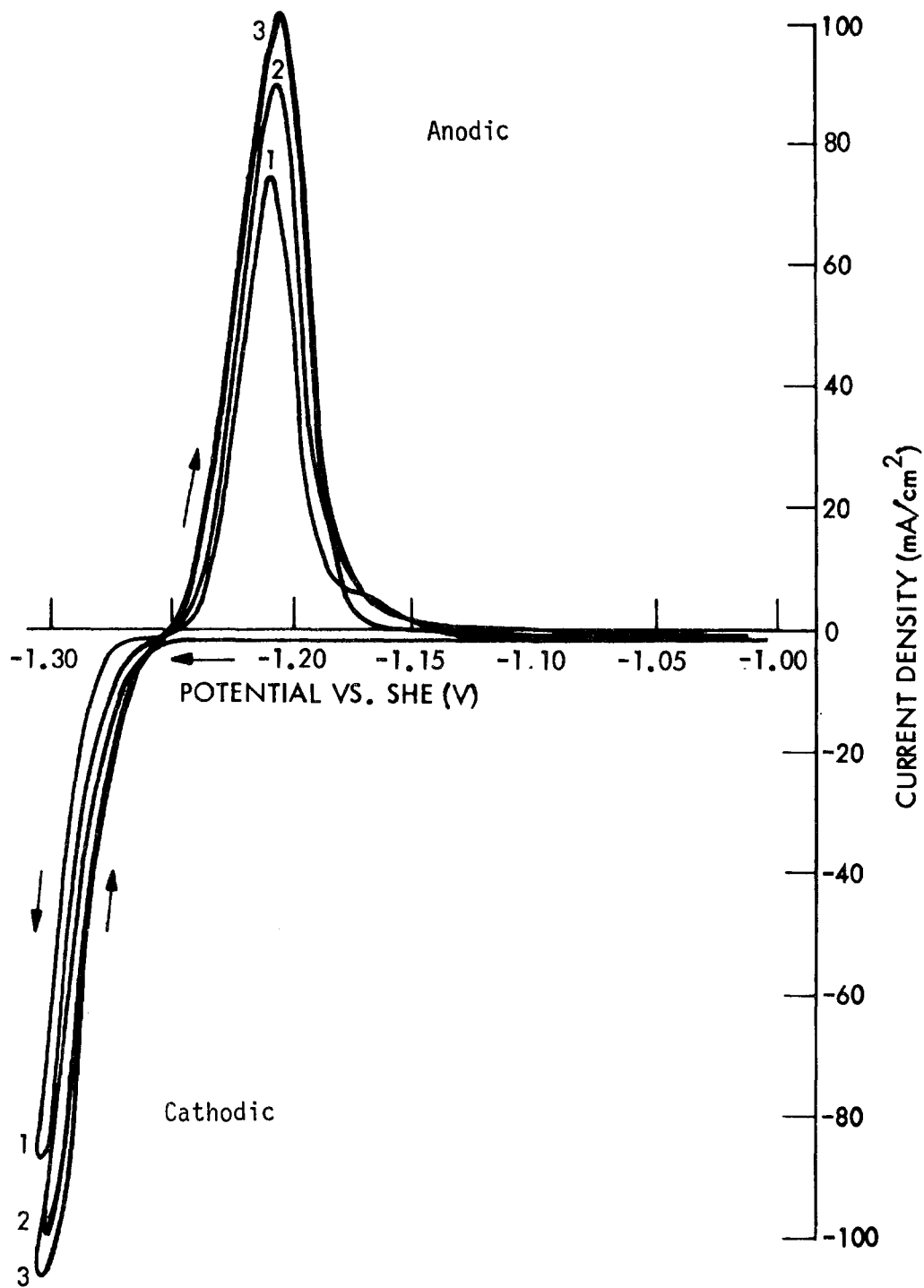


Figure 4.1.4 Zinc Cyclic Voltammetry on Lead

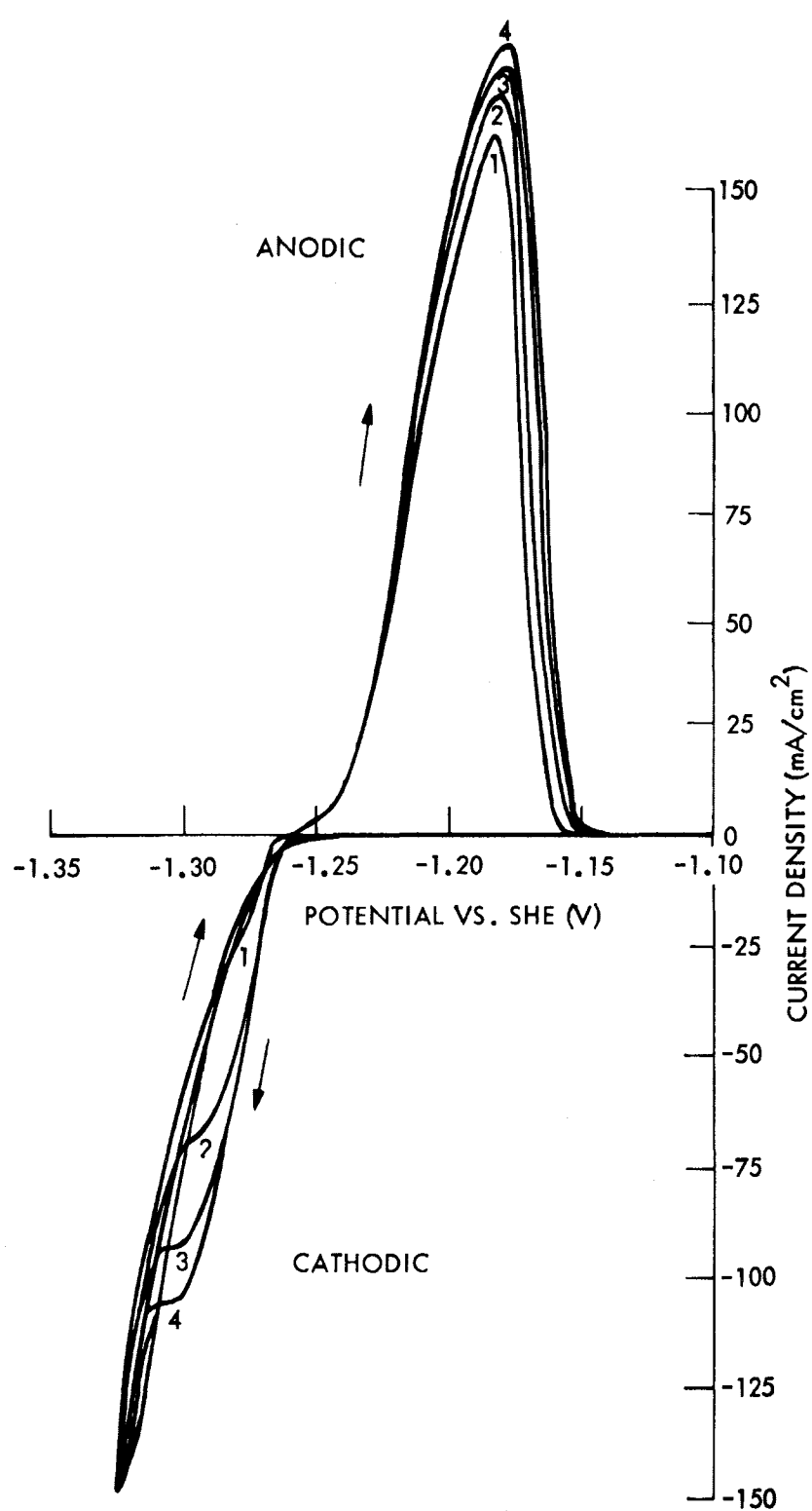


Figure 4.1.5 Zinc Cyclic Voltammetry on Tin

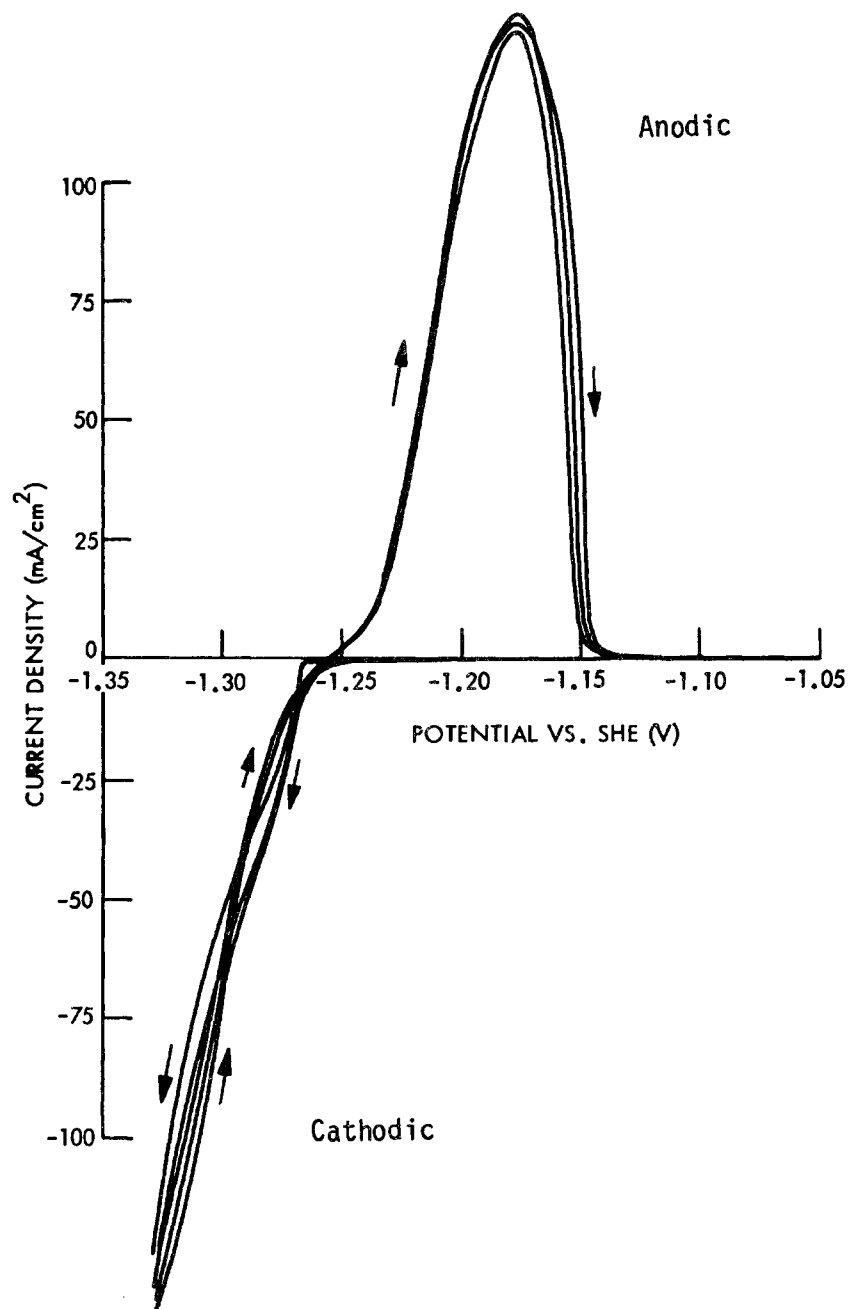


Figure 4.1.6 Zinc Cyclic Voltammetry on Cadmium

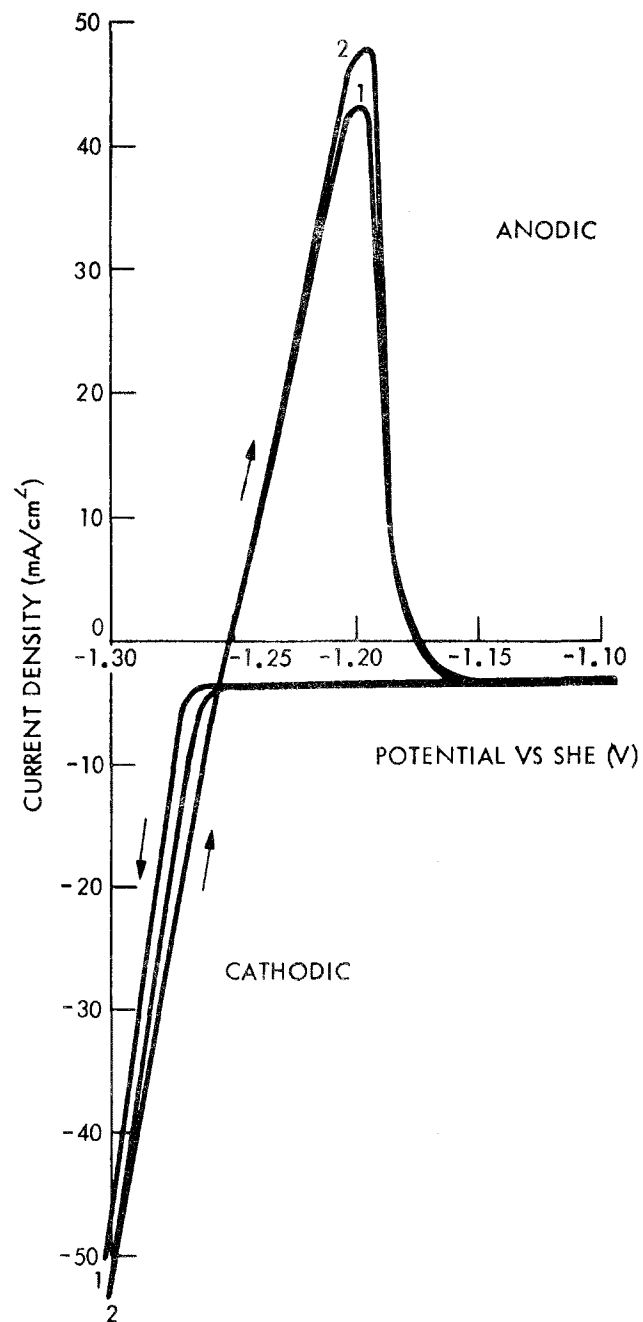


Figure 4.1.7 Zinc Cyclic Voltammetry on Stackpole 6077 Graphite

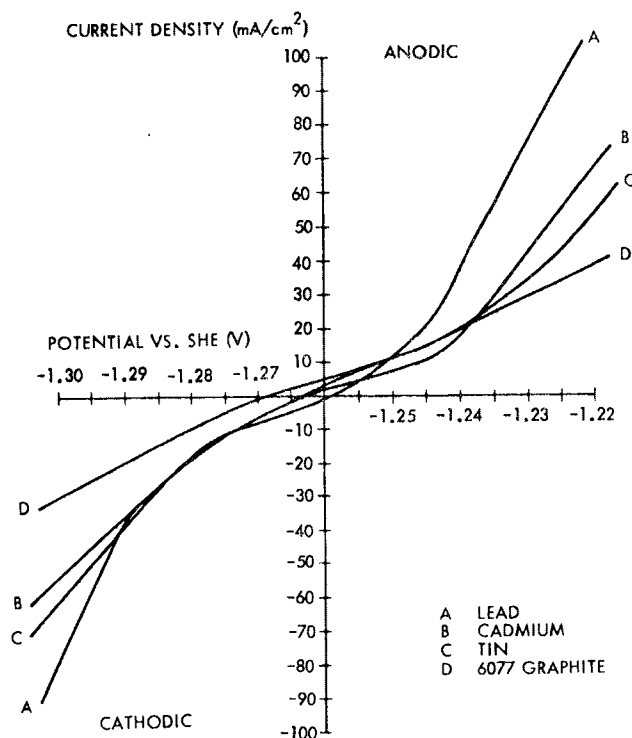


Figure 4.1.8 Zinc Polarization on Selected Substrates

In an actual battery, the selected substrate material would be electrodeposited upon an inexpensive backing material such as iron. Candidate substrates prepared by electrodeposition via various intermediate strikes were evaluated by cyclic voltammetry and the results are shown in figures 4.1.9 - 4.1.13. Except for tin electrodeposited over copper and nickel strikes, which displays a porous deposit as evidenced by the appearance of a copper-zinc alloy peak at -1.17 V vs SHE , all candidate substrates display encouraging polarization characteristics as shown in figure 4.1.14.

Comparing the current densities with those obtained for the pure metal evaluation, the roughness factor for the electrodeposited substrates is ca. 3. Cadmium plated via two different procedures displays similar overpotentials, whereas lead electrodeposited with a copper strike exhibits a slightly higher over-voltage on charge.

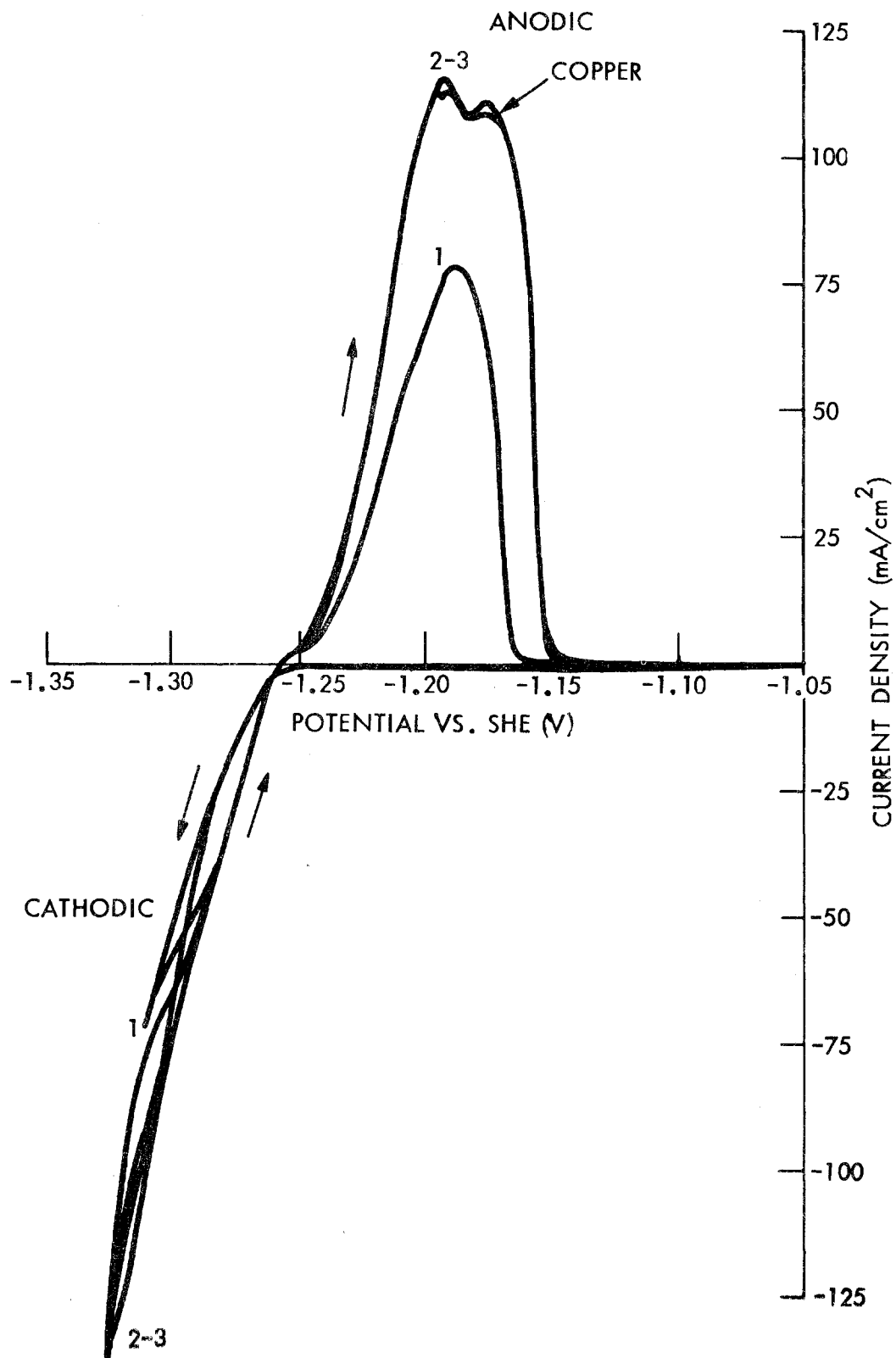


Figure 4.1.9 Zinc Cyclic Voltammogram on Electrodeposited Tin Sn/Cu/Fe

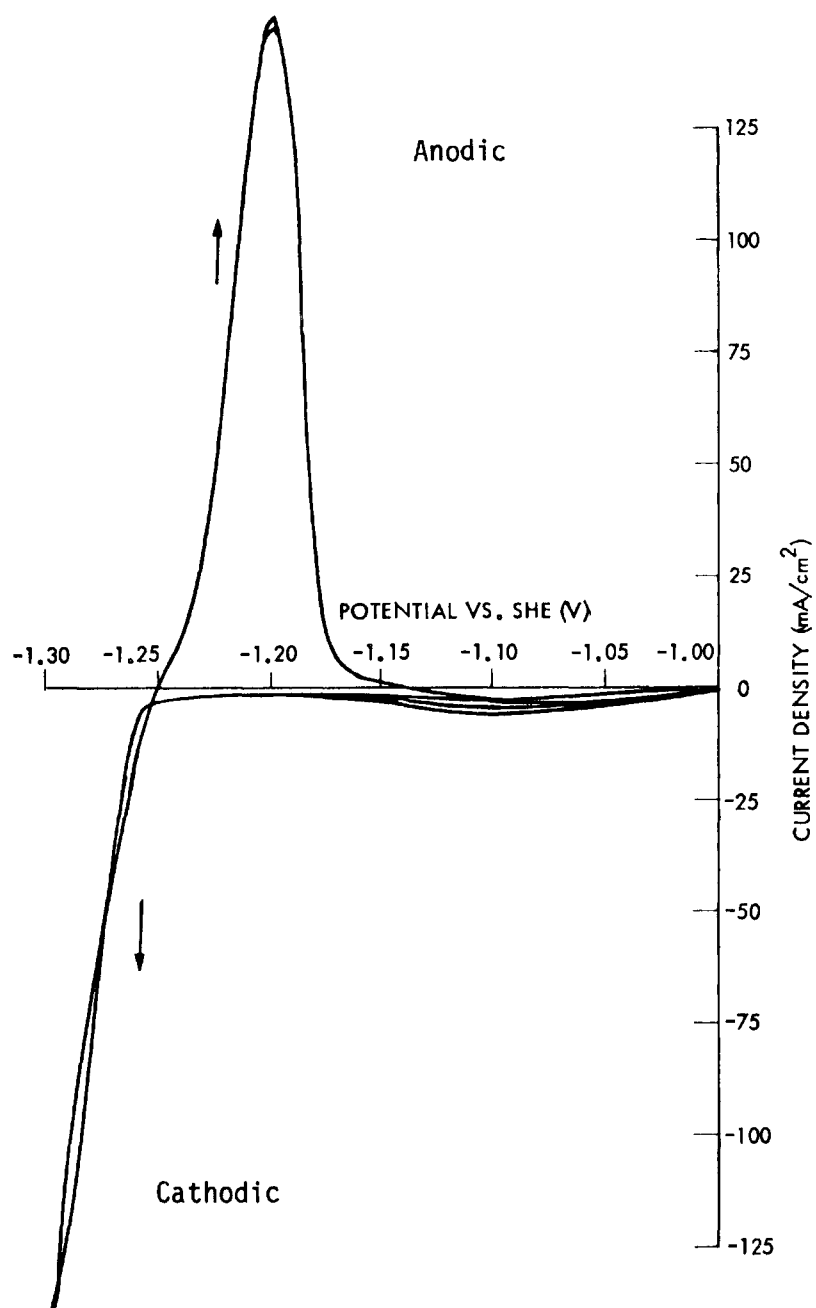


Figure 4.1.10 Zinc Cyclic Voltammogram on Electrodeposited Cadmium (Cd/Cu/Fe)

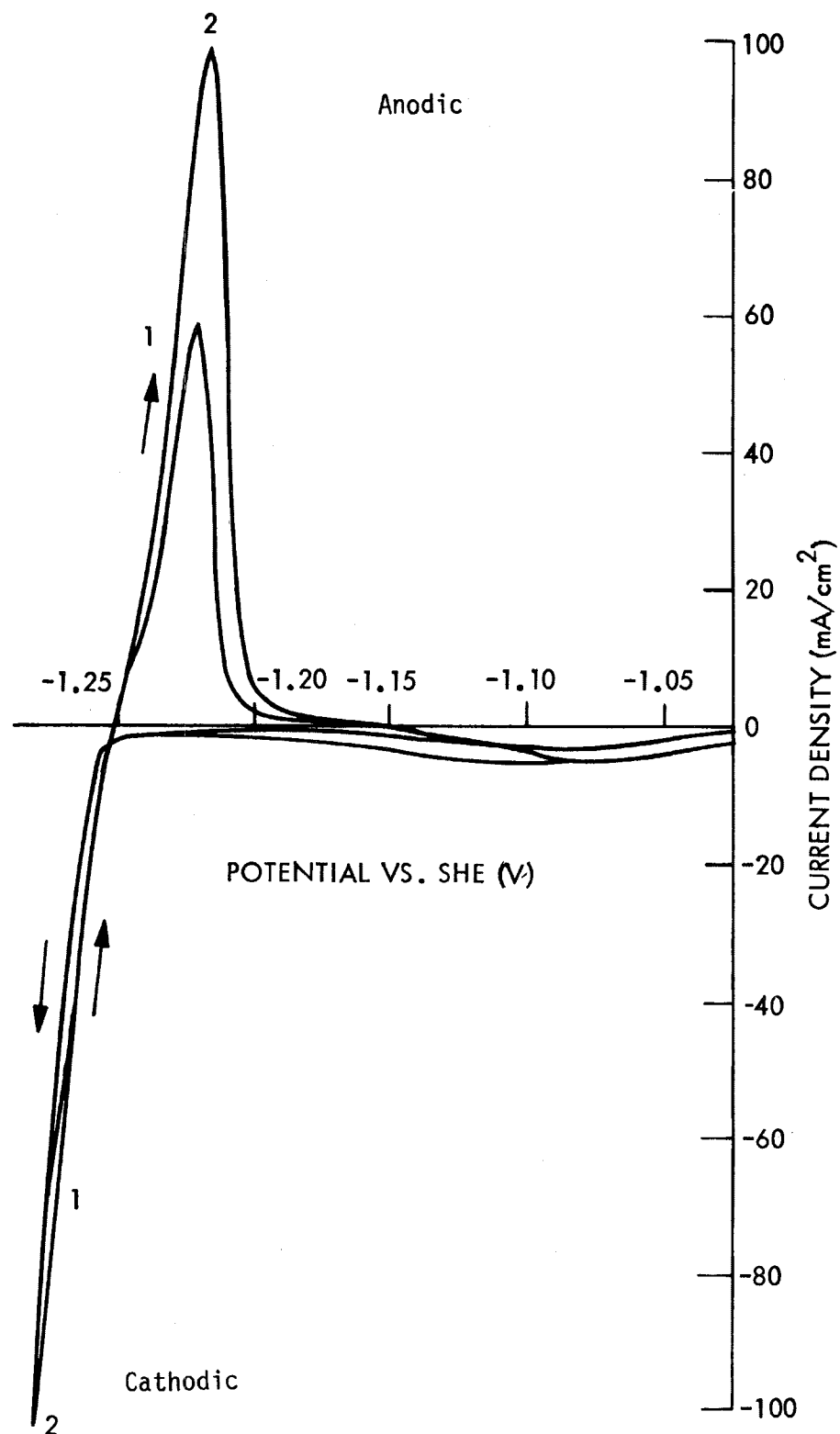


Figure 4.1.11 Zinc Cyclic Voltammogram on Electrodeposited Cadmium (Cd/Pb/Ni/Fe)

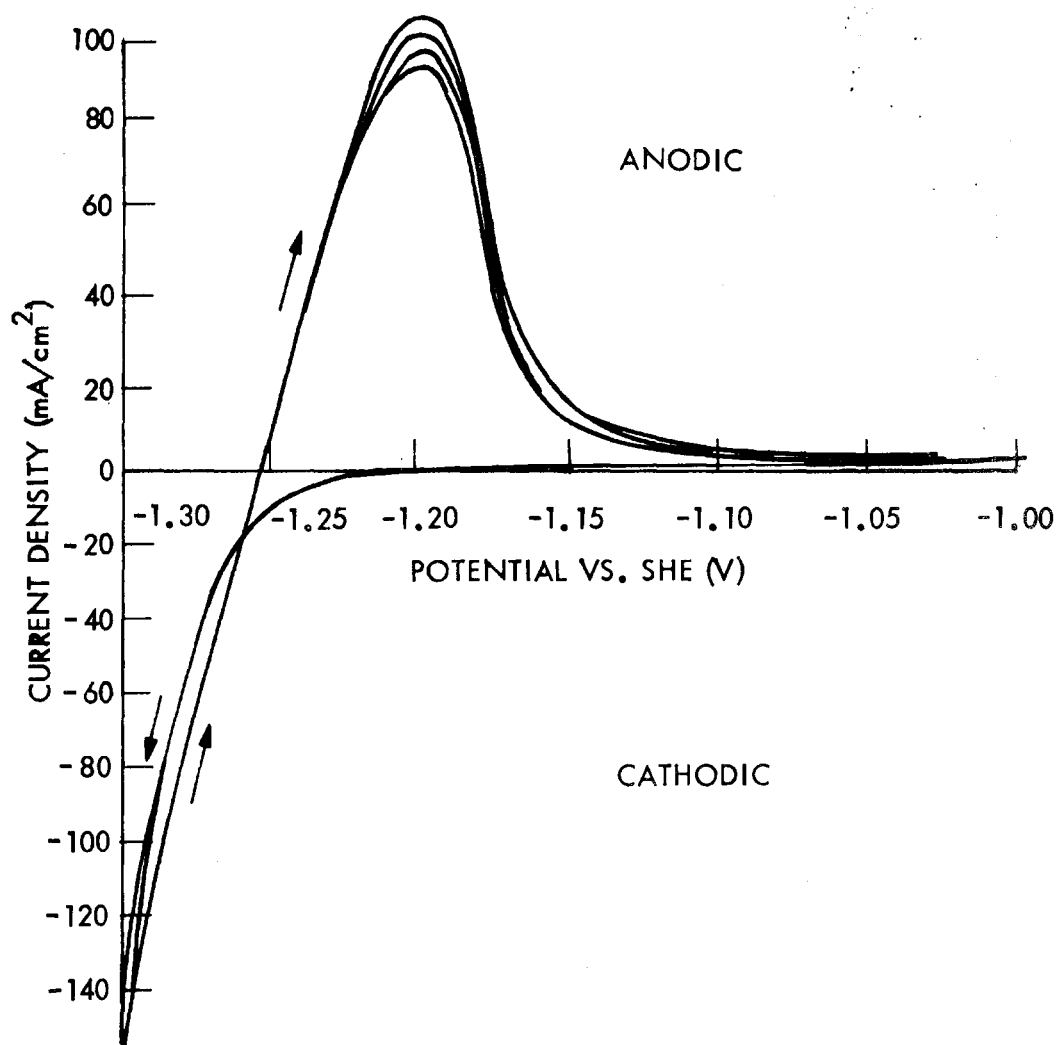


Figure 4.1.12 Zinc Cyclic Voltammogram on Electrodeposited Lead (Pb/Cu/Fe)

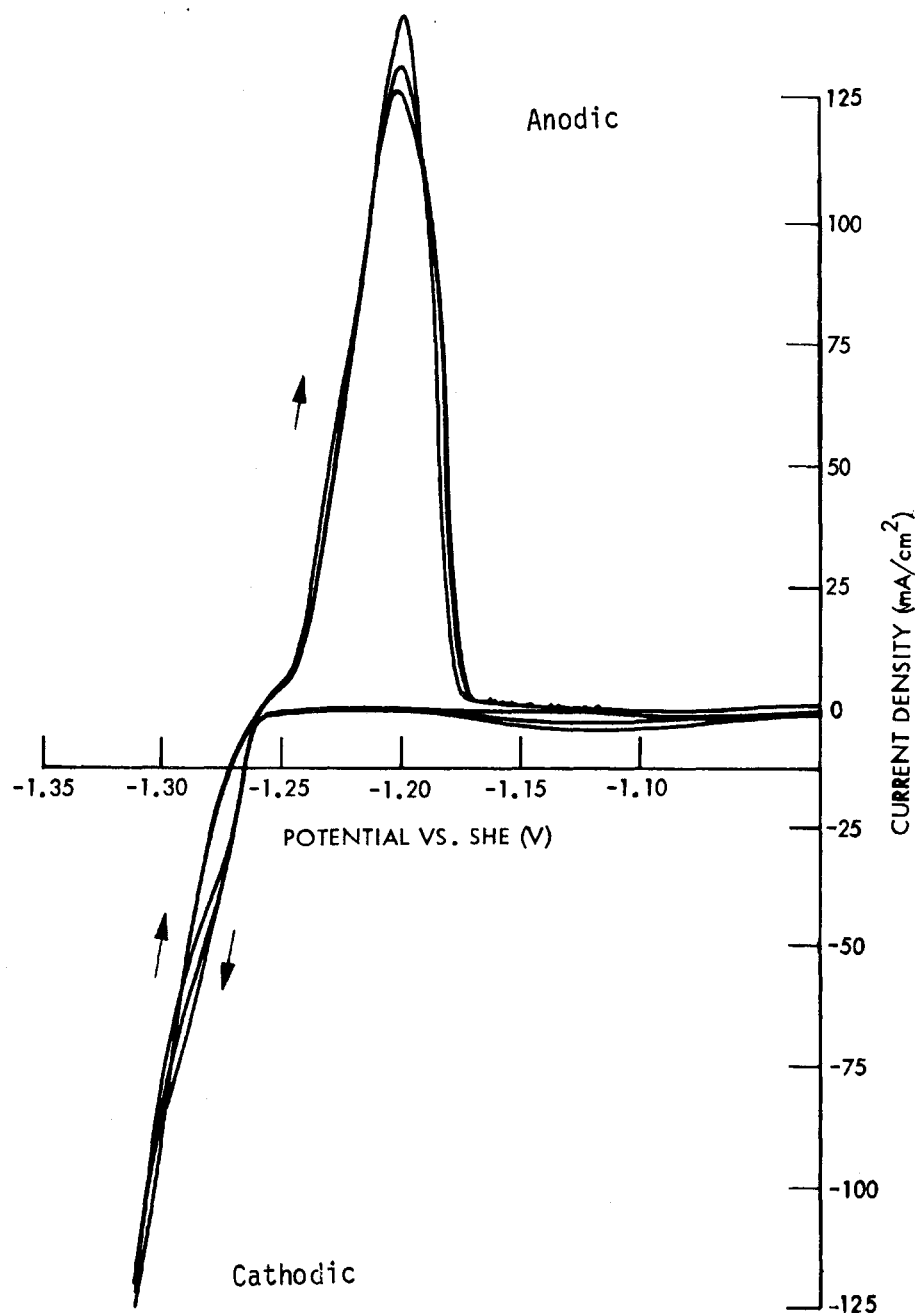


Figure 4.1.13 Zinc Cyclic Voltammogram on Electrodeposited Tin (Sn/Ni/Fe)

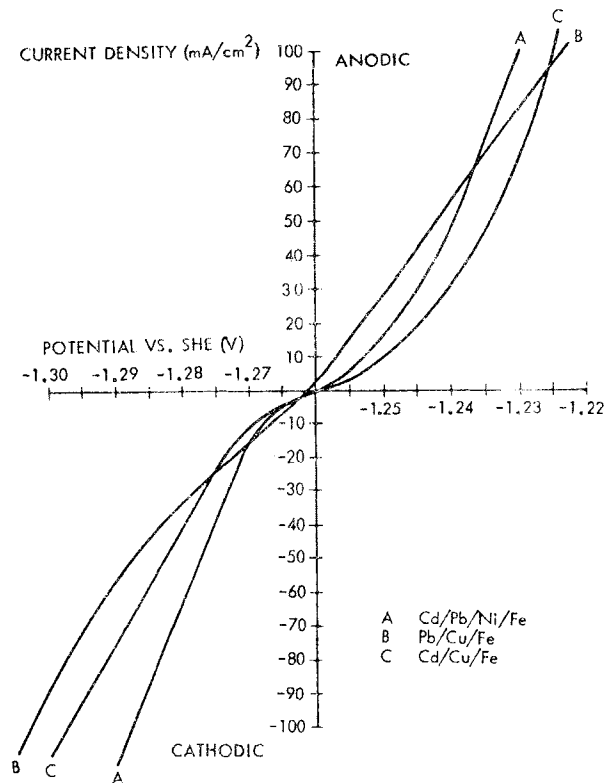


Figure 4.1.14 Zinc Electrode Polarization on Electrodeposited Metallic Substrates

Substrates prepared with cadmium plated iron (Cd/Cu/Fe), lead plated iron (Pb/Cu/Fe), and 6077 graphite were evaluated for adherency of zinc deposit using cyclic efficiency measurements which are discussed in section 4.5.1.

4.2 FERRO-FERRICYANIDE REDOX ELECTRODE INVESTIGATION

4.2.1 Substrate Evaluation

The prime requirements for this redox substrate are: inertness to anodic oxidation in sodium hydroxide electrolyte; a high oxygen overvoltage; reversibility with the ferro-ferricyanide redox couple; and chemical inertness to ferricyanide. Nine candidate substrates were evaluated by cyclic voltammetry using the procedure described in section 3.1. Substrates examined were nickel sheet, sintered nickel, copper, lead dioxide, Stackpole 6077, 6015, and 6008 graphites, Stackpole baked LXB carbon, and platinum as a control. In each case the sub-

strate exhibited low limiting current densities and low oxygen overvoltages as shown representatively in figures 4.2.1 and 4.2.2. Notice that for sintered nickel oxygen evolution occurs at potentials very close to ferrocyanide oxidation while the 6077 graphite displays a well defined oxidation plateau but the hysteresis indicates an oxygen reduction taking place on the surface of the graphite which would be counterproductive. The results of the subscale cyclic voltammetry studies were contrary to results from precontractual investigations in which a porous nickel flow-through electrode was used with good redox performance.

Transferring the redox substrate evaluations to the 60 cm^2 half-cell configuration described in section 3.2, and using laminar flow parallel to the substrate, similar results to those with sheet nickel were obtained using a sintered nickel substrate, as shown in figure 4.2.3. However, when the flow was directed through the porous nickel much better results were obtained, as shown in figure 4.2.4.

The redox substrate investigation is summarized in figure 4.2.5. It is evident that by operating the porous electrode in the flow-through mode, concentration polarization is reduced dramatically. Nickel is the only porous substrate evaluated to date. Tests with a 100 pore per inch Reticulated Vitreous Carbon obtained from FluoroCarbon Corp. were unsuccessful because of the inability to obtain a low resistance bond for attachment of conductive tabs with this fragile material, either with epoxy or contact bonding. Porous carbon or graphite is preferred to nickel on a cost basis.

4.2.2 Chemical Stability

During the precontractual work on the hybrid redox battery, slow decomposition of the alkaline ferricyanide electrolyte was observed. This was characterized by a gradual darkening of the solution, accumulation of ferric species which were not electrochemically reducible, eventual precipitation of ferric hydroxide, and loss of cell capacity. While slow, such decomposition could be significant

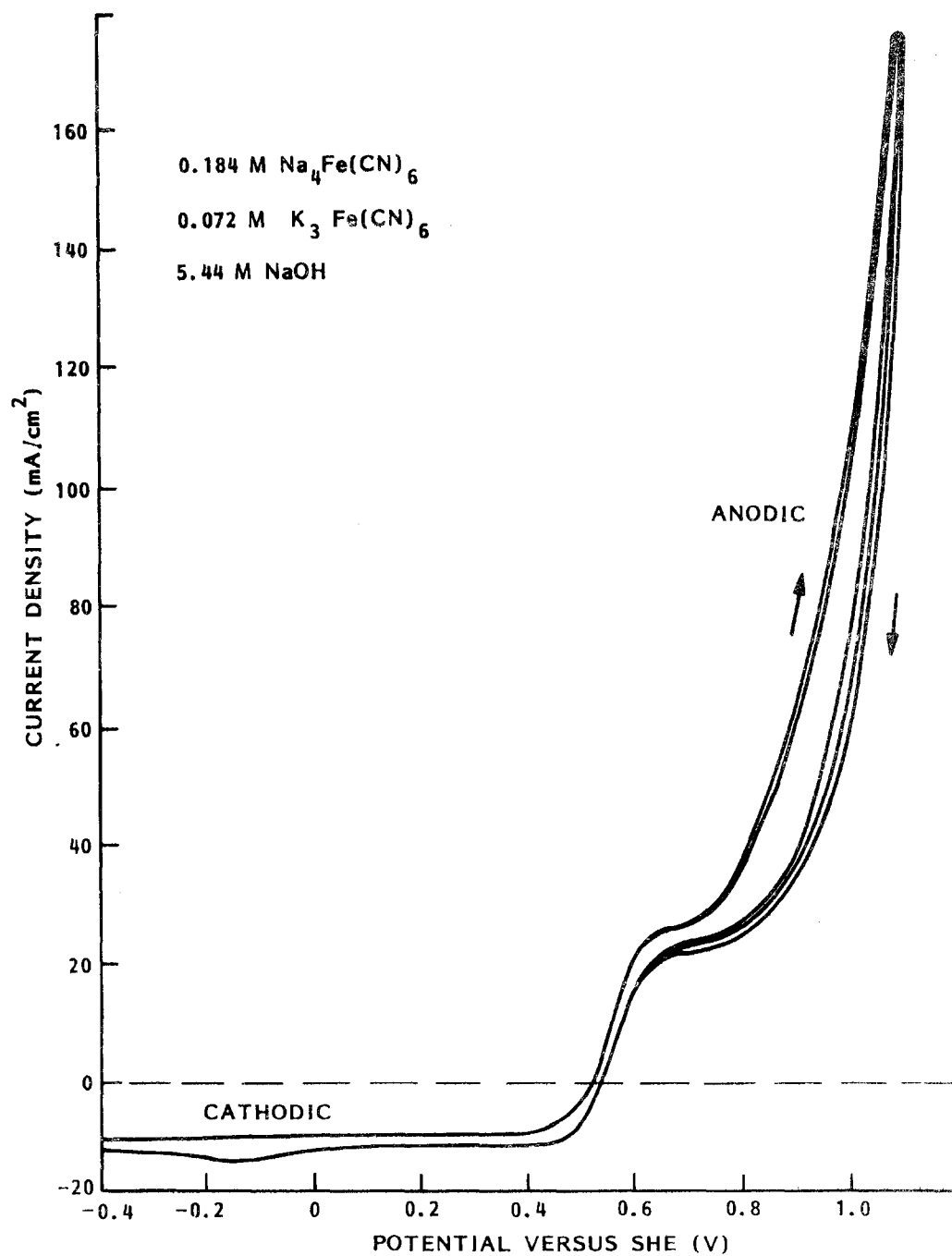


Figure 4.2.1 Ferro- Ferricyanide Cyclic Voltammetry on Sintered Nickel (1.0 cm^2)

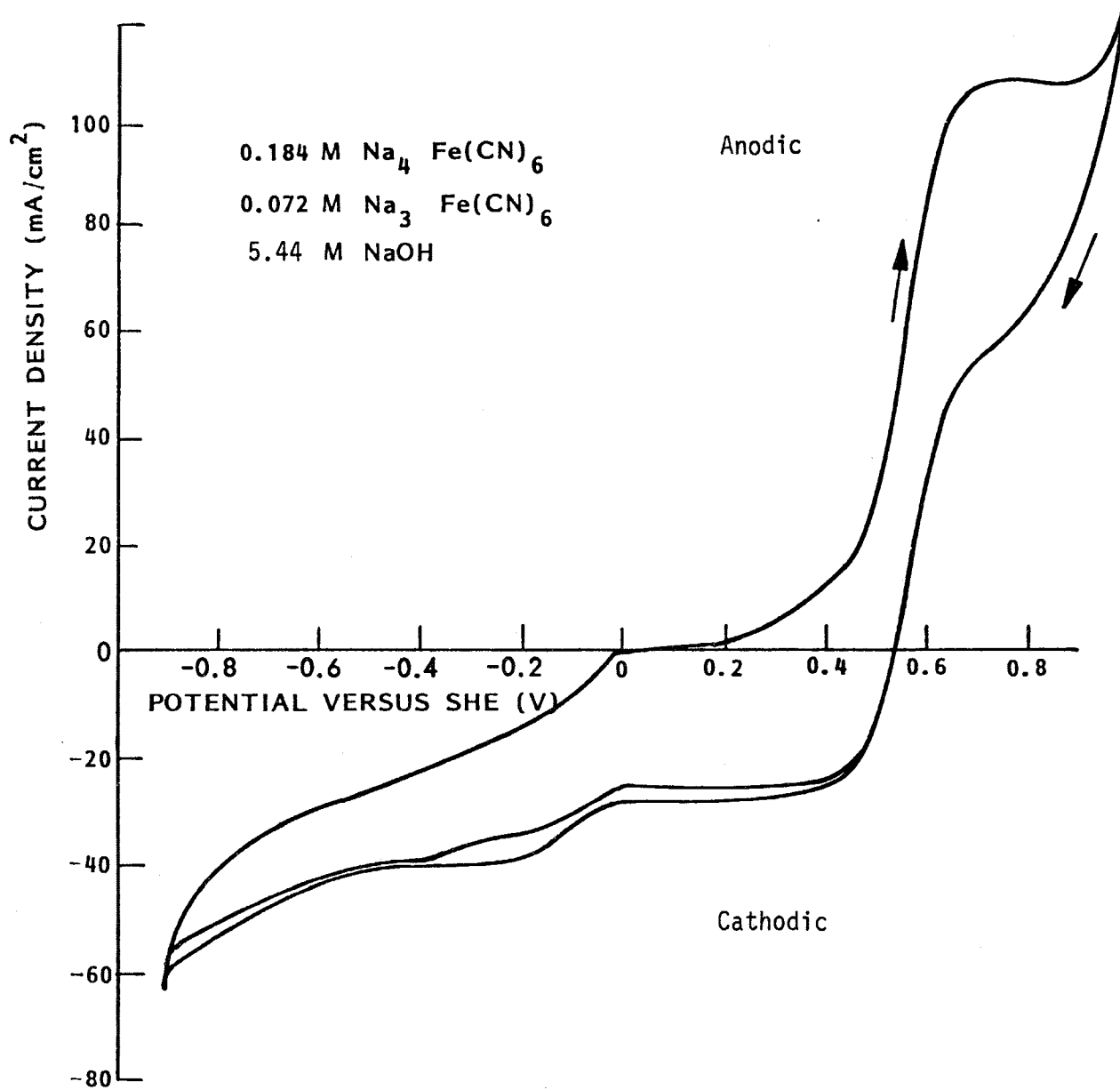


Figure 4.2.2 Ferro-Ferricyanide Cyclic Voltammetry on Stackpole 6077 Graphite (1.0 cm^2)

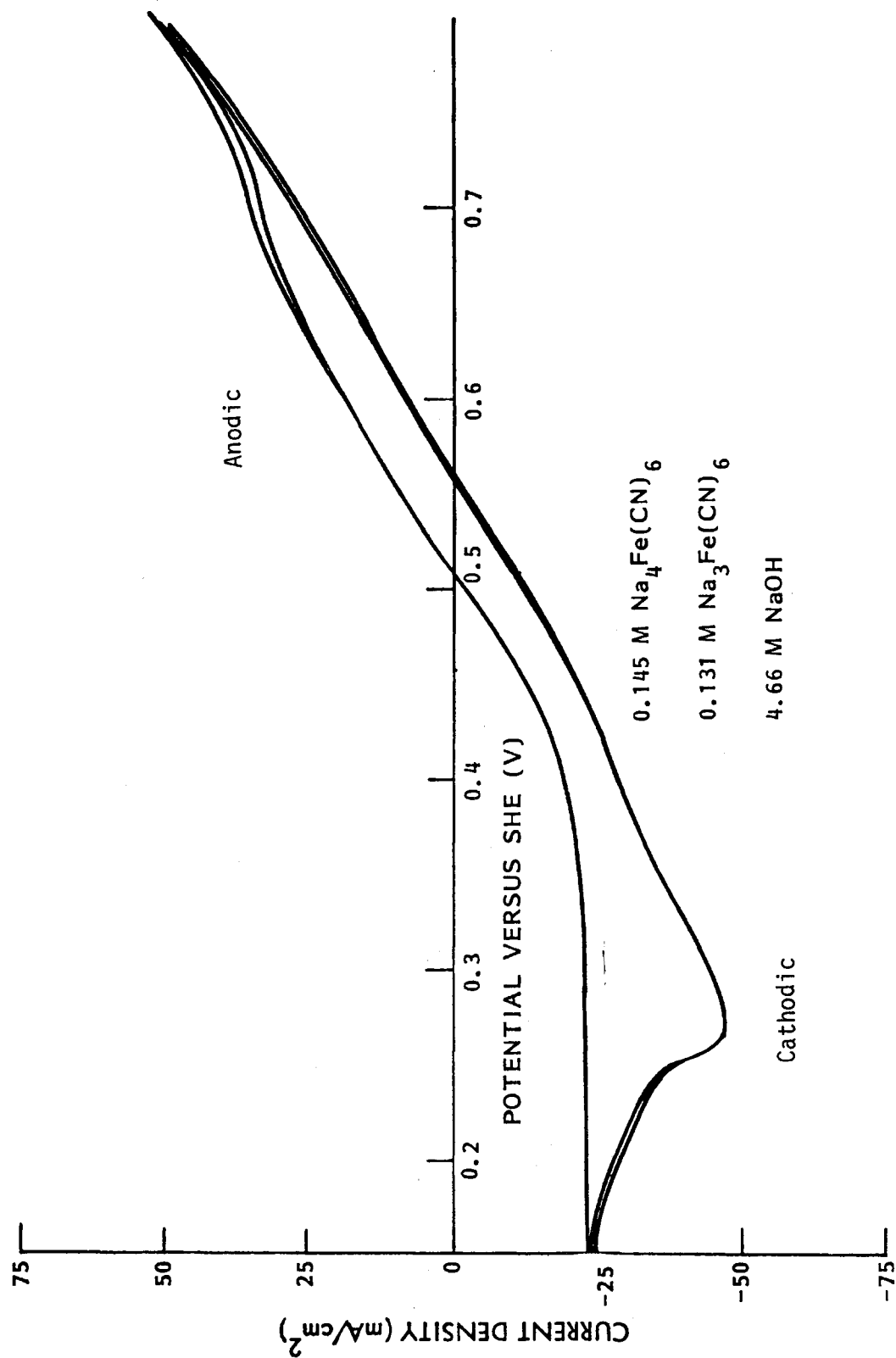


Figure 4.2.3 Ferro-Ferricyanide Cyclic Voltammetry on Sintered Nickel
(60 cm^2 Flow-by configuration)

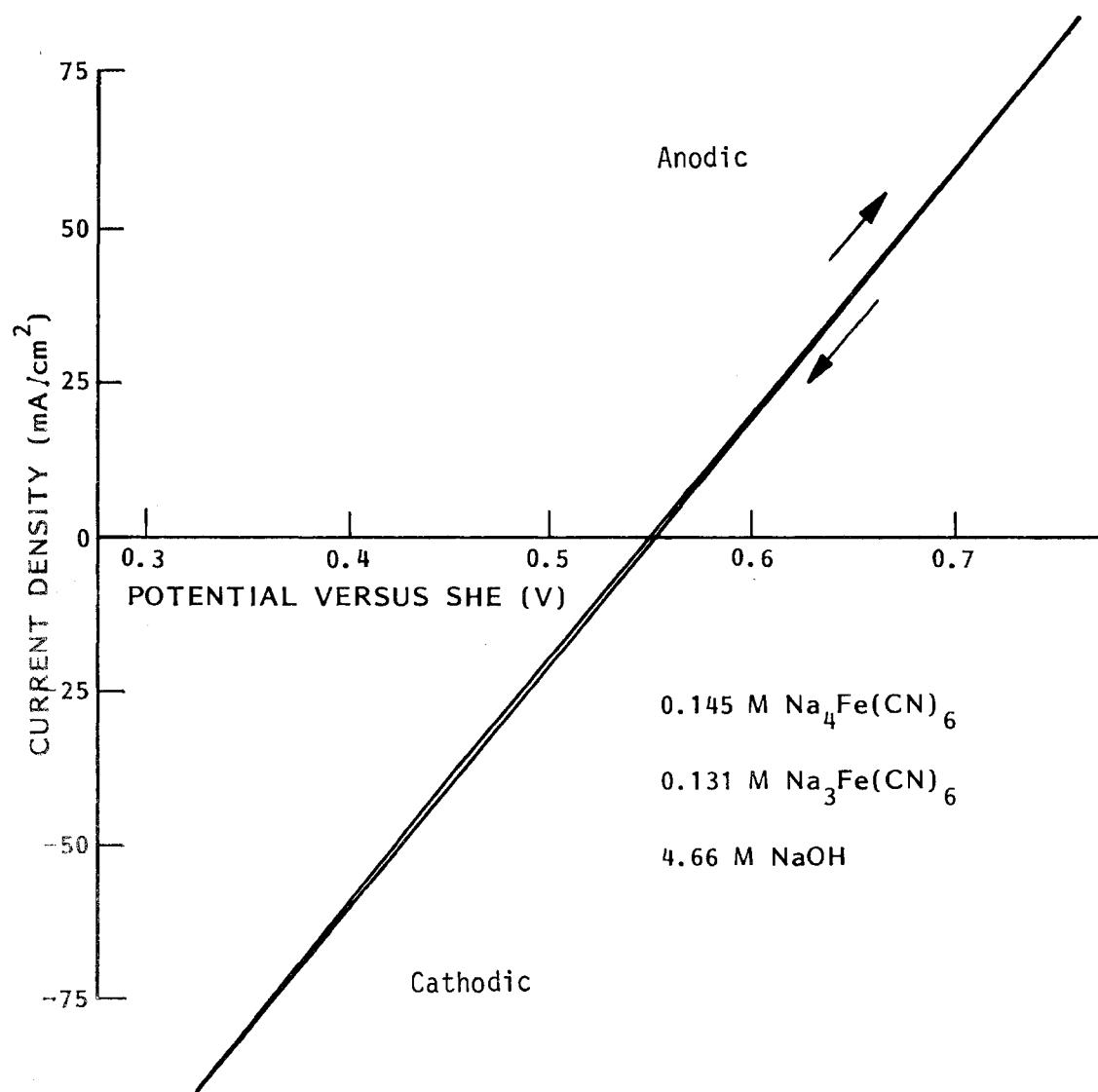


Figure 4.2.4 Ferro-Ferricyanide Cyclic Voltammetry on Sintered Nickel
(60 cm² Flow-through Configuration)

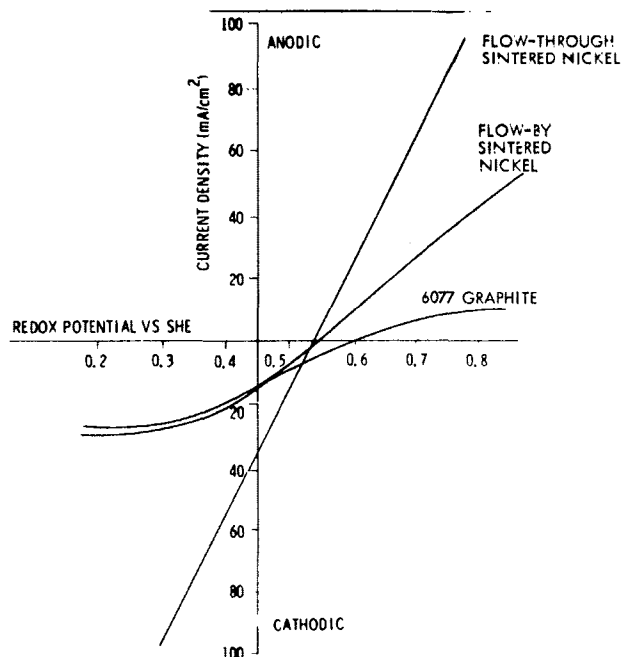


Figure 4.2.5 Ferro-Ferricyanide Electrode Polarization on Nickel and Graphite Substrates

over the lifetime projected for the hybrid redox battery, and so an investigation of the stability of alkaline ferro-ferricyanide solutions was conducted.

Since aquopentacyano iron(II) and (III) have been identified as the most stable intermediates in the decomposition of ferro- and ferricyanide by light³⁶, acids^{8, 33, 34}, and metal ions³⁷, it was reasonable to propose their involvement in the decomposition of the alkaline electrolyte. Using the nitrosobenzene colorimetric method of pentacyano iron determination⁸, studies of the unpromoted decomposition of ferrocyanide and of ferricyanide in various concentrations of NaOH at 20⁰ and 50⁰C were made. These results are presented in figures 4.2.6 and 4.2.7.

The commercial $\text{Na}_4\text{Fe}(\text{CN})_6 \cdot 10 \text{H}_2\text{O}$ and $\text{Na}_3\text{Fe}(\text{CN})_6 \cdot \text{H}_2\text{O}$ used in these experiments contain pentacyano iron as an impurity, so that some aquopentacyano iron is present in solution at the start of the experiments.

In static tests using polyethylene containers, the level of pentacyano iron in saturated solutions of ferrocyanide remained nearly constant under all conditions examined, indicating that ferrocyanide is stable up to 7N NaOH at temperatures as high as 50°C. Solutions saturated with $\text{Na}_3\text{Fe}(\text{CN})_6 \cdot \text{H}_2\text{O}$ at 20°C displayed a gradual decrease in the amount of pentacyano iron III initially present, indicating that at this temperature, its decomposition is more rapid than its formation. However, at 50°C, aquopentacyano iron production outpaces its decomposition and there is a gradual increase in its concentration. Ferricyanide is, therefore, the species through which the electrolyte decomposes, initially producing $\text{Fe}(\text{CN})_5\text{OH}^{3-}$ and, eventually, $\text{Fe}(\text{O})\text{OH}$, which may be recovered from the filter of an operating cell after several hundred cycles.

The change in $\text{Fe}(\text{CN})_5\text{OH}^{3,4-}$ concentration in an operating cell was also studied over the course of a 4-hour charge/discharge cycle of an iron redox half-cell containing a sintered nickel electrode. The data, detailed in figure 4.2.8, shows a rise in pentacyano iron concentration during charge, and a decrease during discharge. Subsequent static experiments showed that the nickel electrode material did not in itself cause loss of cyanide ion from electrolyte but did appear to promote the decomposition of aquopentacyano iron. It seems likely that the rise in pentacyano iron during charge is due mostly to an increased concentration of ferricyanide in solution and that the decrease during discharge is due at least in part to nickel-promoted decomposition, although other processes, such as electrode-assisted cyanide ion loss cannot be ruled out at this time.

Figures 4.2.9 and 4.2.10 show the appearance of a new peak in the infrared spectrum of the electrolyte after many cycles. Combined with other studies, this provides evidence for the formation of an electrochemically inactive Fe(III)binuclear complex from aquopentacyano iron(II) in NaOH solution which contains a bridging cyanide group. Further experimentation is planned to elucidate the relationship between this complex and the electrochemically nonreducible Fe(III) observed in old electrolyte.

In our system, the major pathway for electrolyte decomposition seems to be slow deligation of cyanide from ferricyanide by hydroxide. Photochemical reactions

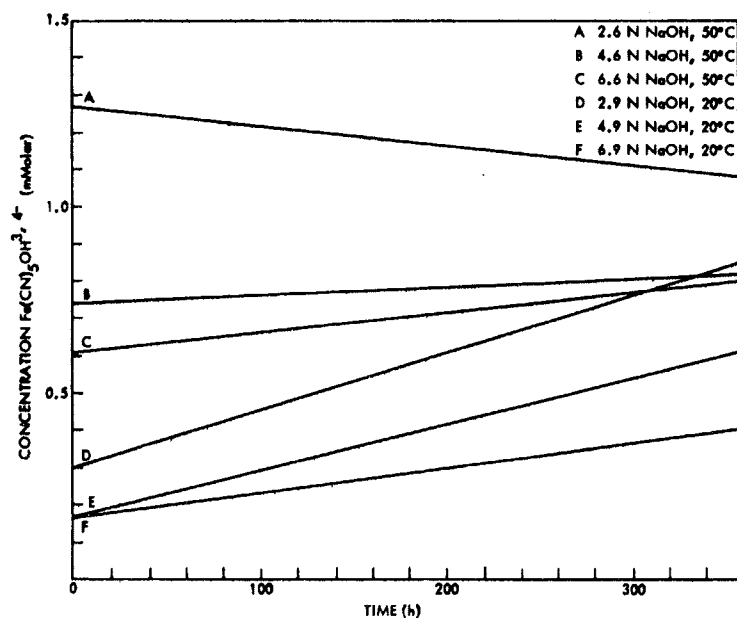


Figure 4.2.6 Variation of $\text{Fe}(\text{CN})_5\text{OH}^{3-}_4$ Concentration with Time in $\text{Na}_4\text{Fe}(\text{CN})_6$ -saturated NaOH solutions at 20° and 50°C

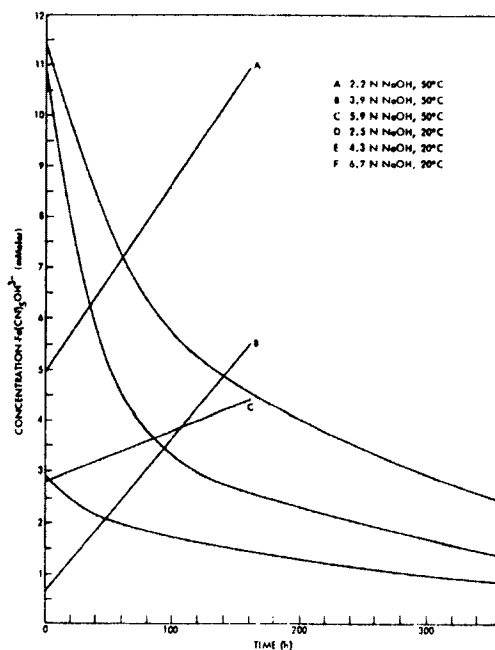


Figure 4.2.7 Variation of $\text{Fe}(\text{CN})_5\text{OH}^{3-}_3$ Concentration with Time in $\text{Na}_3\text{Fe}(\text{CN})_6$ -Saturated NaOH Solutions at 20° and 50°C

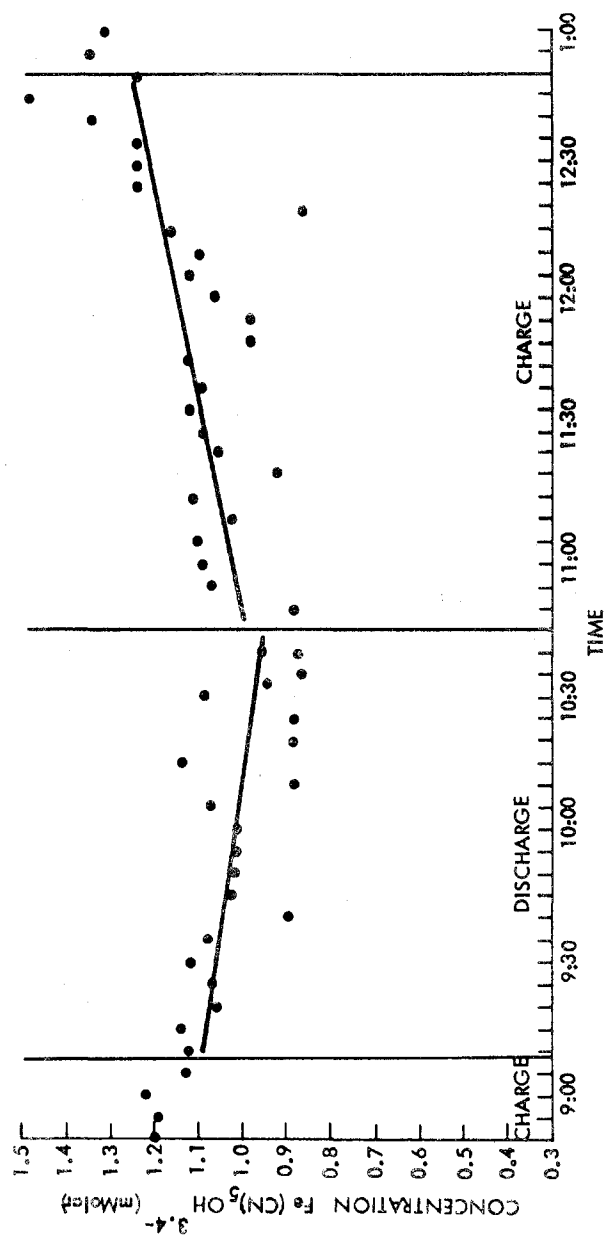


Figure 4.2.8 Variation of $\text{Fe(CN)}_5\text{OH}^{3,4-}$ Concentration with Time During Cell Cycling

Figures 4.2.9 and 4.2.10:

Infrared spectra in CN stretching region of Ferro-ferricyanide electrolyte taken from operating cells after 5 and 852 4-hour charge/discharge cycles

2039.9 cm^{-1} $\text{Fe}(\text{CN})_6^{4-}$ terminal CN stretch

2117.5 cm^{-1} $\text{Fe}(\text{CN})_6^{3-}$ terminal CN stretch

2173.9 cm^{-1} possible stretch of CN bridging two Fe(III) centers.

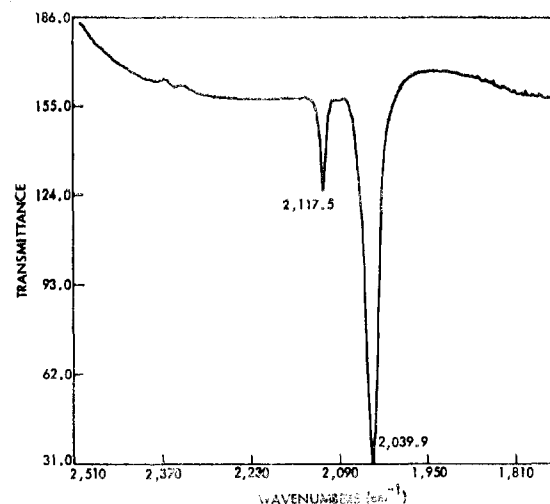


Figure 4.2.9 Electrolyte after 5 Cycles

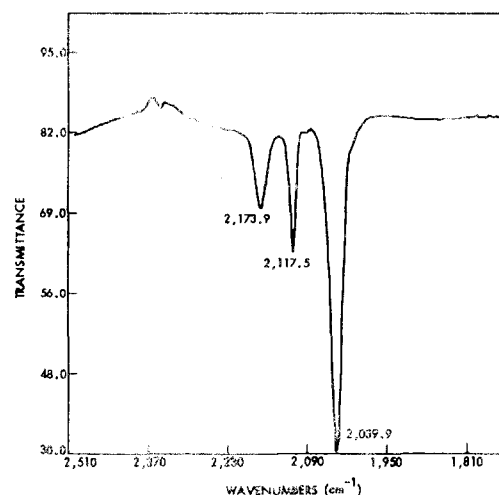


Figure 4.2.10 Electrolyte after 852 Cycles

are prevented by the use of opaque materials of construction. Acid and metal ion-catalyzed decompositions, which proceed through tight ion pairs with a cyanide ligand promoting its loss from iron as HCN or MCN^{37} are precluded by the high hydroxide concentration. A possible source of contamination, HHgO_2^- estimated to be $6 \times 10^{-5} \text{ M}$ in the Hg/HgO reference electrode compartment, is minimized by the capillary connection between the reference electrode and the cell compartment.

4.2.3 Religation of Pentacyano Iron

If hydroxide can displace cyanide from an iron complex, then microscopic reversibility requires that a pathway exists for cyanide replacement under similar conditions. Benchtop experiments showed that 7 - 10 equivalents of NaCN would rapidly convert $\text{Fe(CN)}_5\text{OH}^{4-}$ to ferrocyanide in NaOH solution at 50°C . Encouraged by this result, a procedure was developed for fully reducing the electrochemically inert Fe(III) to Fe(II) and then religating the pentacyano iron species, using NaCN.

In an early trial, arbitrarily conducted on a cell electrolyte after 74 cycles for concept evaluation, NaCN was added in small portions over 2 days until free cyanide ion persisted in solution; in this experiment, cyanide served as both reducing agent and religating agent. After oxidizing excess cyanide with $\text{K}_3\text{Fe(CN)}_6$ additions, the cell was placed back in operation. This procedure decreased the $\text{Fe(CN)}_5\text{OH}^{3,4-}$ level from 3×10^{-3} to $8 \times 10^{-4} \text{ M}$; coulombic efficiency of the cell increased from 85% to 95%. In a more recent trial, cell electrolyte which had operated for 460 cycles, was treated with formalin to reduce all Fe(III), and then only one equivalent of NaCN, calculated from the measured concentration of $\text{Fe(CN)}_5\text{OH}^{4-}$, was added. After waiting two hours for completion, ferricyanide was added to destroy any unreacted free cyanide, and cell cycling was reinitiated. This method, which requires considerably less NaCN than the first approach, reduced the $\text{Fe(CN)}_5\text{OH}^{4-}$ concentration from $1 \times 10^{-3} \text{ M}$ to less than $2 \times 10^{-6} \text{ M}$, and immediately increased the redox half-cell coulombic efficiency from 71% to 88%.

This approach to electrolyte rejuvenation appears promising, both technically and economically. Further work is planned to define optimal religation conditions.

4.3 ELECTROLYTE INVESTIGATION

4.3.1 Zincate Electrolyte

The solubilities of zincate in sodium hydroxide are shown in figure 4.3.1 and were determined by two solubilization procedures: (1) chemical dissolution of ZnO powder introduced into the electrolyte; (2) anodic dissolution of 99.99% pure zinc plate. In each case the particulates were allowed to settle for two weeks at a given temperature (28, 35, and 50°C) at which time a 5 ml aliquot of the clear supernatant was removed, acidified, and the zinc content determined by atomic absorption analysis.

Resistances of the above zincate saturated solutions (figures 4.3.2) were measured using an Industrial Instruments conductivity dip cell model CEL-BB1, having a cell constant of 1.07, in conjunction with a General Radio Co. type 1650-A Impedance Bridge.

From this study the optimal NaOH electrolyte concentration was taken as 5 N.

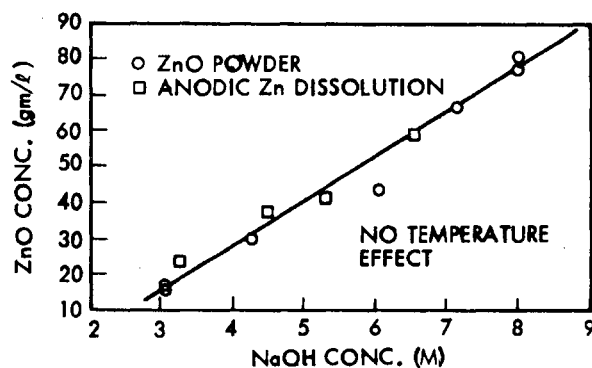


Fig. 4.3.1 Solubilities of ZnO in Sodium Hydroxide Electrolytes

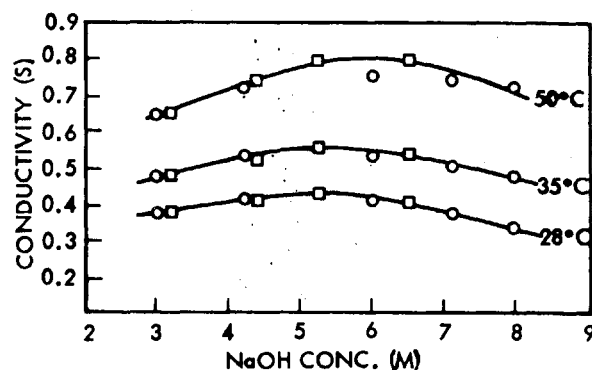


Fig. 4.3.2 Conductivities of ZnO-saturated Sodium Hydroxide Electrolytes

Since zinc ferrocyanide is insoluble in neutral solutions and could therefore be a potential passivator for a zinc anode, its solubility in sodium hydroxide electrolyte was estimated from available equilibria data. It was found that at pH 10.8 or greater, all zinc is present as the zincate ion and zinc ferrocyanide is completely soluble under these conditions. Passivation of zinc would not therefore be expected under the moderate current densities used in this cell.

4.3.2 Ferro-Ferricyanide Redox Electrolyte

Using the techniques described above, the solubilities of sodium ferrocyanide decahydrate and sodium ferricyanide monohydrate in NaOH and conductivities of these saturated solutions were determined; they are shown in figures 4.3.3 through 4.3.6.

Solubility of the ferrocyanide salt is relatively insensitive to sodium hydroxide concentration at 22°C, but at 50°C solubility increases rapidly with decreasing sodium hydroxide concentration. With sodium ferricyanide solubility rises very rapidly with decreasing sodium hydroxide concentration at both 20°C and 50°C. Interestingly, the conductivity of a sodium ferrocyanide-saturated sodium hydroxide is higher than that of the equivalent ferricyanide-saturated electrolyte, even though the solubility of the ferricyanide salt is much higher. This is no doubt due to the overriding conductivity of the 5N NaOH. The above data for 5N NaOH reduced to a 22°C temperature basis are as follows:

Table 4.3.1
Conductivities of 5N Sodium Hydroxide/Reactant Salt
Saturated Electrolytes at 22°C

Reactant Salt	Concentration (M)	Conductivity ($\Omega^{-1}\text{cm}^{-1}$)(S)
ZnO	0.48	0.31
$\text{Na}_4\text{Fe}(\text{CN})_6 \cdot 10\text{H}_2\text{O}$	0.05	0.35
$\text{Na}_3\text{Fe}(\text{CN})_6 \cdot \text{H}_2\text{O}$	0.40	0.27
None	n.a	0.38 ⁽⁵⁴⁾

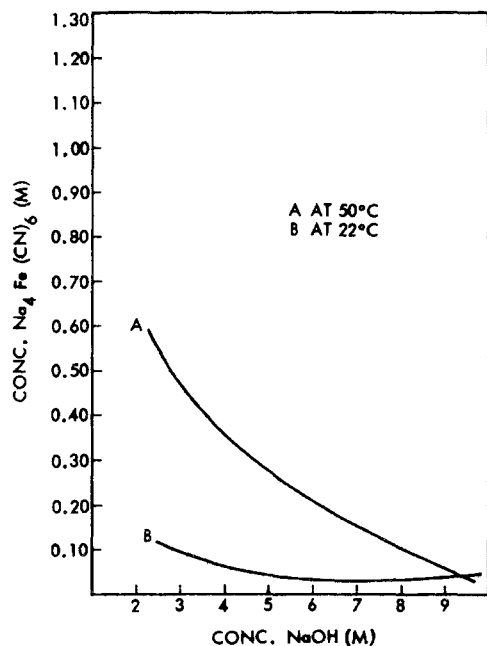


Fig. 4.3.3 Solubilities of Sodium Ferrocyanide in Sodium Hydroxide Electrolytes

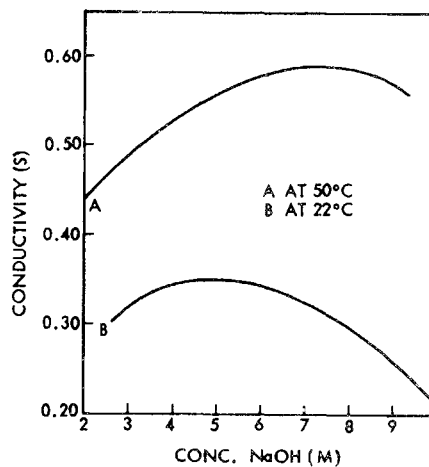


Fig. 4.3.4 Conductivities of Sodium Ferrocyanide-Saturated Sodium Hydroxide Electrolytes

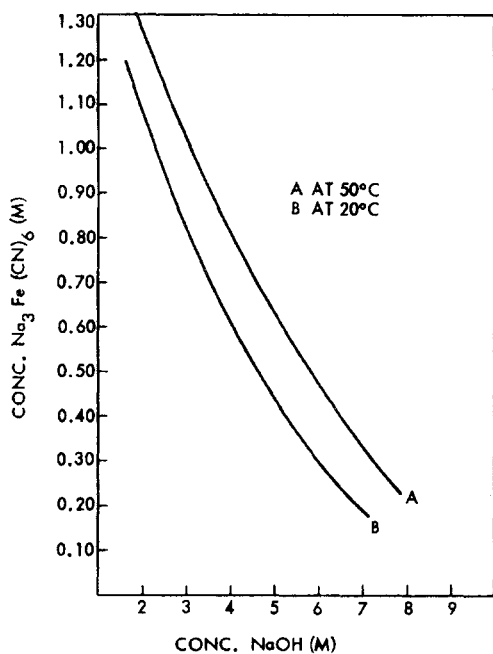


Fig. 4.3.5 Solubilities of Sodium Ferricyanide in Sodium Hydroxide Electrolytes

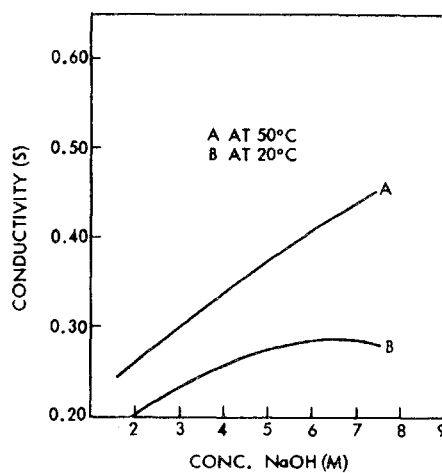


Fig. 4.3.6 Conductivities of Sodium Ferricyanide-Saturated Sodium Hydroxide Electrolytes

The conductivities of the reactant-saturated 5N NaOH electrolytes are all lower than that for 5 N NaOH. Sodium ferricyanide lowers the conductivity considerably more than zincate for equivalent concentrations. This is probably due to cationic pairing with the ferricyanide ion. A 5N NaOH electrolyte concentration has been tentatively selected as optimal, taking into consideration conductivity, electrode concentration polarization, electrolyte stability, and solubility of the ferrocyanide salt.

The results of an examination of the mixed solubilities at 40°C, of ferro- and ferricyanide as the sodium or potassium salt in sodium hydroxide or potassium hydroxide are shown in Table 4.3.2.

Table 4.3.2
Solubilities of Sodium and Potassium Ferro- and
Ferricyanide in Sodium and Potassium Hydroxide
Electrolytes at 40°C

Hydroxide conc. salt		Ferrocyanide conc. salt		Ferricyanide conc. salt	
4.38 M	NaOH	0.106 M	Na	0.358 M	Na
4.40	NaOH	0.341	K	0.299	K
4.70	KOH	0.045	Na	0.36	Na
4.95	KOH	0.029	K	0.13	K

Note that the potassium salts are more soluble in NaOH electrolyte than in a KOH electrolyte. The sodium salt solubilities appear to be largely unaffected by the supporting electrolyte. Hence sodium hydroxide is preferred over potassium hydroxide on a solubility basis and the sodium ferrocyanide salt is preferred to the potassium salt because of lower unit cost.

4.4 SEPARATOR SELECTION

The criteria for a suitable separator are low resistivity, high chemical stability, mechanical strength, and moderate cost. To date, six separators have been

evaluated for resistivity in 5N NaOH at 25⁰ and 50⁰C. The five low resistivity separators were used in the half-cell cyclic testing phase to determine mechanical integrity and chemical stability. Table 4.4.1 lists the separators evaluated, their specific resistances, costs in small quantities, and observations regarding performance.

Table 4.4.1

EVALUATION OF SEPARATOR MATERIALS

Manufacturer	Material	Area Resistivity (mΩ/cm ²)	Cost (\$/m ²)	Comments
RAI Industries	P-1010	10	113	Low tear strength
RAI Industries	P-1040	2	57	Low tear strength
RAI Industries	X1125-34-4	4		Experimental separator. High electrolyte intermixing rate.
DuPont	Nafion - 002	108	323	High resistance
DuPont	Nafion - 115	18	323	Effective
DuPont	Nafion - 114	14	323	In use

One of the problems not addressed as yet is the tearing of the RAI membranes caused by separator flutter in the flowing electrolyte streams. It is believed that lower cost, thinner materials will be viable if support techniques are employed to eliminate flutter. These techniques will be investigated in a Phase II Study.

4.5 HALF-CELL TESTING

The purpose of the half-cell testing was to gain insight into the performance of individual half-cells with regard to substrates, electrolyte flow rate and concentration, current density, and charge capacity. Optimum configurations were then used in full-cell cyclic testing. A nickel counterelectrode in 5N NaOH electrolyte is used in the opposing half-cell.

4.5.1 Zinc Electrode Half-Cell Testing

4.5.1.1 Substrate Evaluation. Three substrates, selected from preliminary cyclic voltammetric scanning were installed in the 60 cm² half-cell configuration with nickel-plated counterelectrodes and a separator in place. These were tested for cyclic energy efficiency at similar flow rates and current density (35 mA/cm²) conditions. The results are shown in Table 4.5.1. The lead substrate test was terminated after 14 cycles due to extreme variations in coulombic efficiency.

4.5.1.2 Effects of Current Density and Sodium Hydroxide Concentration. Using a cadmium substrate, the cyclic energy efficiency was then determined as a function of current density (Figure 4.5.1) and NaOH concentration (Figure 4.5.2). At the lower NaOH concentrations, the cell performance becomes very erratic and large quantities of gas are observed to evolve during charge periods. An electrolyte concentration 5N in NaOH appears optimal for the concentration range investigated.

4.5.1.3 Adherency. Adherency and the nature of the zinc electrodeposit were found qualitatively to be functions of face velocity in that at a face velocity of ca. 5 cm/sec the deposit was nodular (figure 4.5.3); whereas above 12 cm/sec it was primarily crystalline in appearance (figure 4.5.4) for similar charge capacities of 4.4 A-Hr. In these photographs, direction of electrolyte flow is upward. Quantitative data was not obtained due to the face velocity decrease and increase as the zinc was charged and discharged.

4.5.1.4 Ferrocyanide Effect. Ferrocyanide ion concentration may build up gradually in the zinc electrolyte by diffusion through microporous membranes used as separators. The ferrocyanide ion was found to cause initial erratic fluctuations in the coulombic efficiency of the zinc electrode (Figure 4.5.5) but these anomalies gradually disappeared with time on cycling. The mean coulombic efficiency remained constant at a high value.

Table 4.5.1

CYCLIC ZINC SUBSTRATE HALF-CELL RESULTS
(35 mA/cm²)

Substrate	Cycles (4 Hr)	Voltaic Efficiency (%)	Coulombic Efficiency (%)	Energy Efficiency (%)
Cadmium	85	94.3	94.9	89.4
Graphite (6077)	84	93.6	85.8	80.4
Lead	14	89.3	88.0	79.0

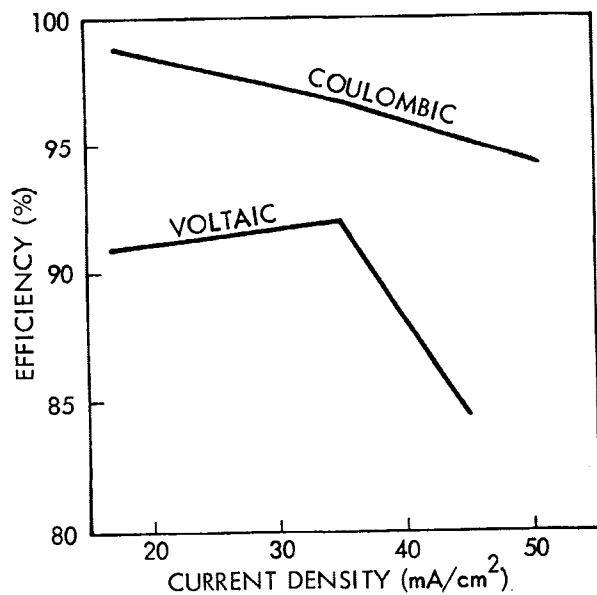


Fig. 4.5.1 Effect of Current Density on Zinc Half-cell Cyclic Efficiency

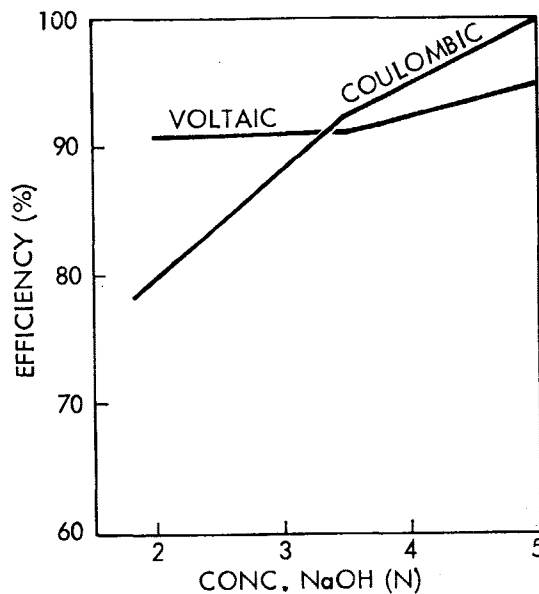


Fig. 4.5.2 Effect of NaOH Concentration on Zinc Half-cell Cyclic Efficiency

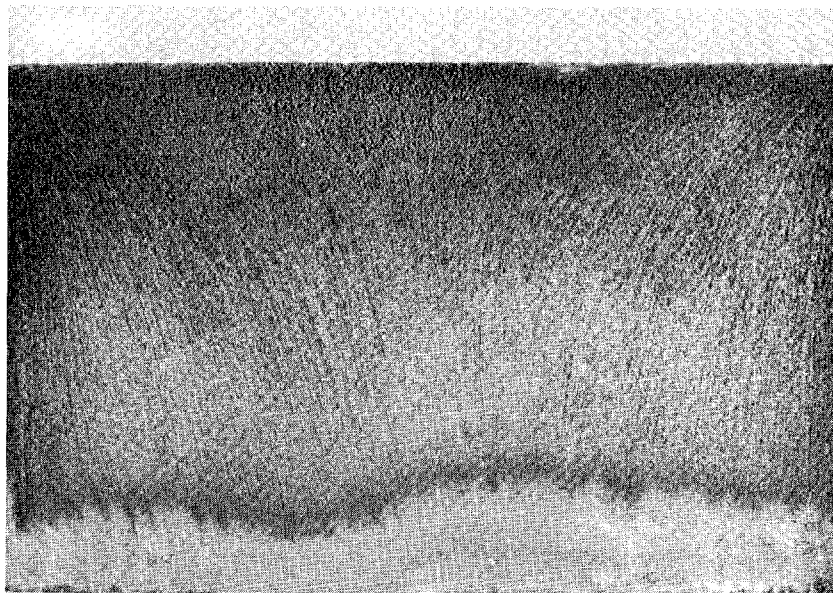


Figure 4.5.3 Electrodeposited Zinc Morphology at Low Face Velocity (5 cm/sec)

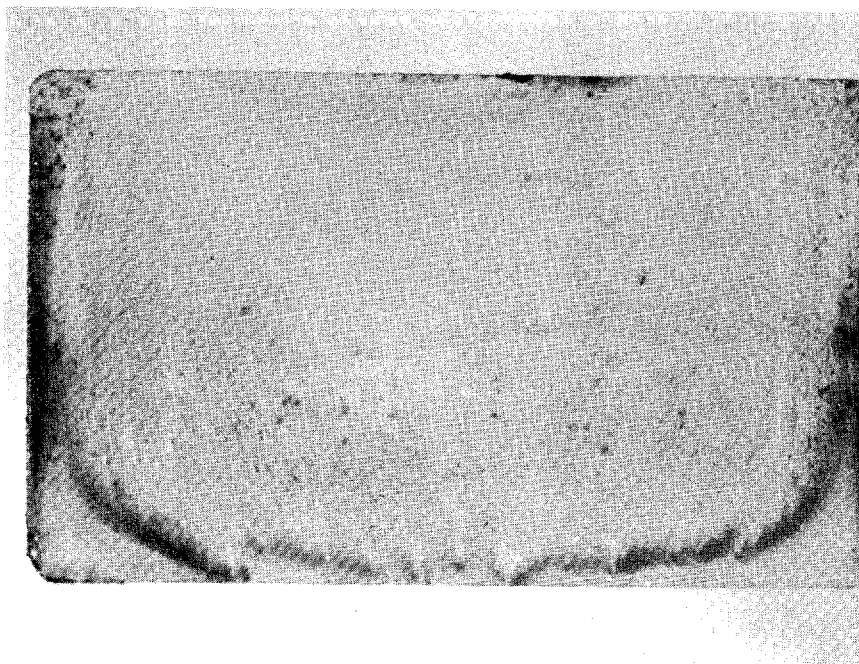


Figure 4.5.4 Electrodeposited Zinc Morphology at High Face Velocity (12.5 cm/sec)

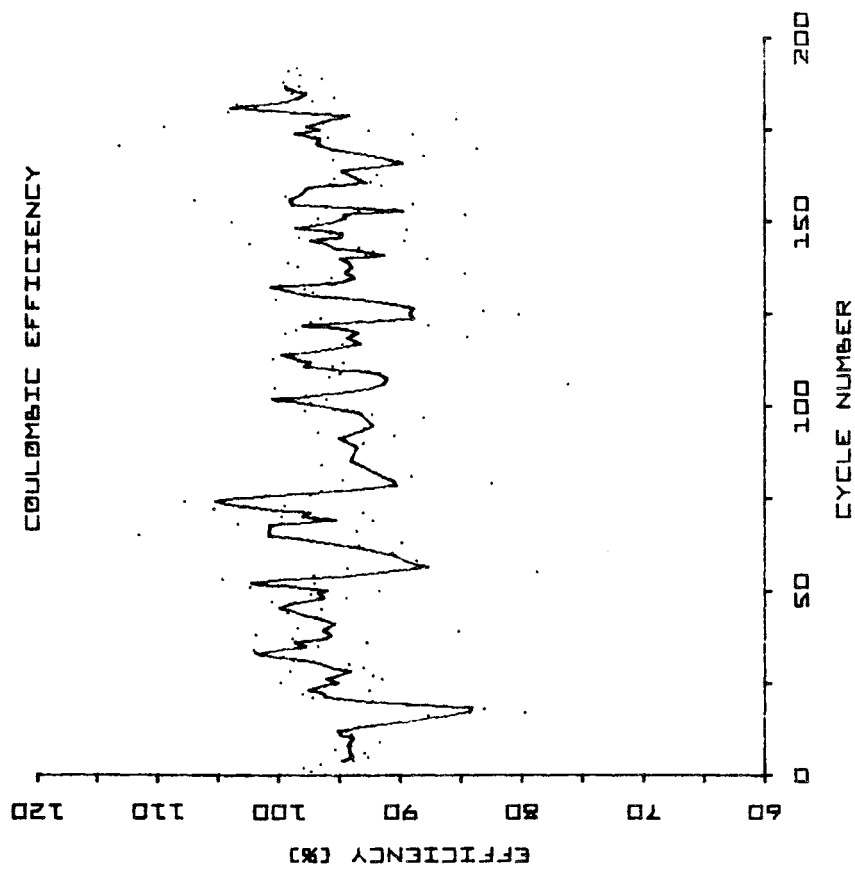


Figure 4.5.5 Cyclic Coulombic Efficiency of Zinc Half-cell with Addition of $\text{Na}_4\text{Fe}(\text{CN})_6$

4.5.1.5 Temperature Study. Evaluation of a zinc-zincate half-cell at 30⁰, 40⁰, and 50⁰C showed that cyclic coulombic efficiencies were identical at each temperature

4.5.1.6 Charge Capacity. The present battery design uses a zinc substrate-to-separator distance of 1.9 mm, for a fully discharged substrate. Assuming a crystalline deposit having a porosity as great as that typical of statically electrodeposited alkaline zinc, and occupying 50% of the internal half-cell volume, this design should have a minimum zinc capacity of 167 mA-hr/cm². Experimentally the time of charge was slowly increased until blockage of the electrolyte flow channels occurred, as shown in figure 4.5.6. This resulted in a charge capacity for the present design of 92 mA-hr/cm². As the figure shows, the electrode area of this fully discharged substrate is encouragingly clear of zinc deposit; however, both inlet and outlet flow channels are encrusted with zinc with the heaviest concentration at the top exit channel. It is felt that the low capacity limit is due to a combination of poor cell geometry, which induces the deposition of dendritic and/or mossy zinc instead of dense crystalline deposits, the adhesion of particulates (zinc and zinc oxide) to the walls of the acrylic flow channels leading to low flow rates, and flow blockage due to mossy zinc deposition. A microscopic study of this phenomenon is proposed in the Phase II investigations to elucidate the flow and geometry factors contributing to flow blockage. It is believed that once blockage is eliminated capacities of 160 to 300 mA-hr/cm² can be obtained with the present electrode clearances.

4.5.1.7 Electrolyte Additives Study. Studies to evaluate the effects of various electrolyte additives upon zinc half-cell cyclic efficiency have been completed using sodium stannate, lead acetate, sodium silicate, and polyethylene glycol of mean molecular weight 600.

Sodium stannate was added to the zincate electrolyte at a concentration of 10⁻⁴, 10⁻³, and 10⁻² Sn⁺⁴M. Cyclic coulombic efficiency increased from 86% to 99%, as the tin concentration was increased. However, the zinc half-cell charge potential increased from a normal 1.25 V vs SHE to 1.9 V vs SHE during the initial

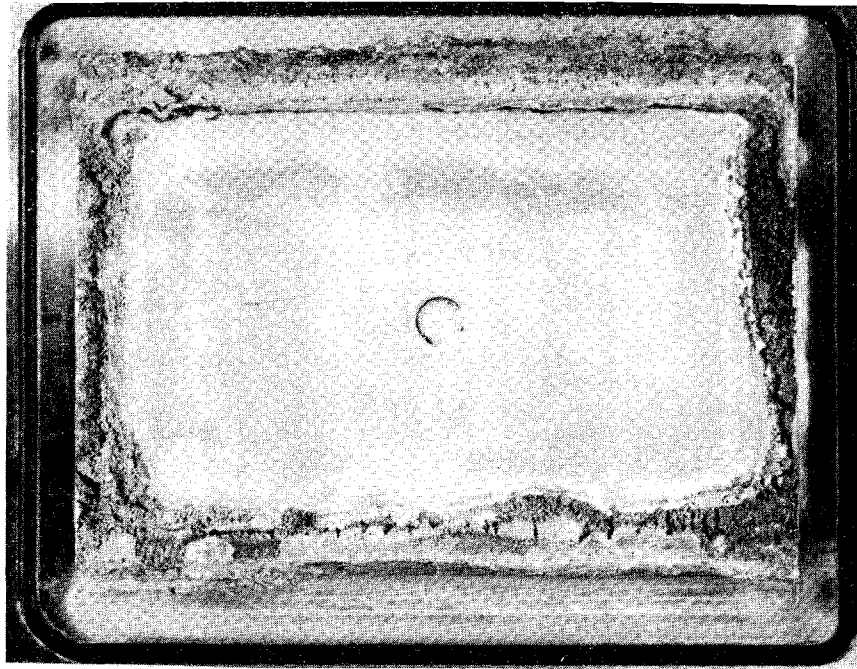


Figure 4.5.6 Cell Channel Blockage Leading to Reduced Charge Capacity.

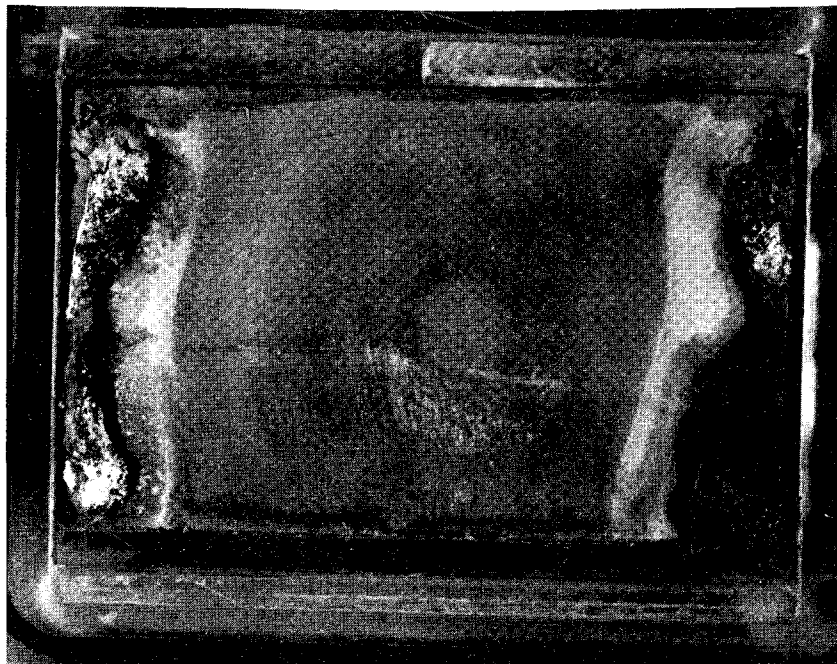


Figure 4.5.7 Zinc Ridges Caused by Lead Additives

thirty minutes of charge and then decreased to the normal level. High resistance film formation during discharge might be implied; however, cyclic voltammetry of the tin substrates, Section 4.1, indicates the absence of such a film. Another possible explanation is that a power source anomaly occurred. This particular test was ultimately terminated due to a microcomputer malfunction.

Lead acetate was studied at the 10^{-4} M Pb level. Prior to introduction, the half-cell coulombic efficiency was constant at $90 \pm 5\%$ over 50 cycles; however, for the first cycle after Pb addition, the efficiency was 137%, followed by 319% and 336% for the second and third cycles. Post mortem examination revealed two ridges of zinc, figure 4.5.7, which accounted for the abnormally high coulombic efficiency, because the reference electrode continued to sense the presence of zinc and maintained a constant current discharge.

Sodium silicate was added to a zincate solution at the 10^{-2} M level with a resultant increase in cyclic coulombic efficiency to 99.9%. After 50 cycles, however, the flow channels became increasingly encrusted with zinc, and flow blockage resulted. Post mortem examination revealed zinc deposited in the nonsubstrate flow area blocking the flow, with greatest blockage at the upper electrolyte exit channel. A second test of silicate in zincate solution gave similar results.

Polyethylene glycol of mean molecular weight 600 was tested at the 0.05wt% level in zincate solution. Immediately after addition the coulombic efficiency decreased to 75%, the half-cell potential increased, and the power supply was unable to maintain a constant 35 mA/cm^2 . This would indicate a greatly increased cell resistance, either at the separator due to pore blockage or at the substrate surface, due to deposition of a poorly conducting organic film. Post mortem examination revealed a very dense, crystalline zinc deposit, but the separator showed signs (discoloration, abnormal wrinkling) of attack by the polyethylene glycol additive.

Of the additives tested, only sodium silicate gave results worthy of further investigation. More work with this additive is planned in a Phase II study.

4.5.2 Ferro- Ferricyanide Redox Half-Cell Testing

4.5.2.1 Substrate Study. Data comparison of cyclic efficiencies of a sintered porous nickel substrate in the flow-by configuration and the flow-through configuration are given in Table 4.5.2 and illustrate concentration polarization effects which necessitate the use of porous electrodes for dilute redox electrolytes.

4.5.2.2 Flow Rate and Current Density Studies. Using the flow-through configuration in half-cell tests, the effects of flow rate and current density upon the cyclic efficiencies have been determined and are shown in Figures 4.5.8 and 4.5.9 respectively. Energy efficiency is not seriously affected by flow rate or current density at face velocities above 1.5 cm/sec and usable current densities up to 50 mA/cm².

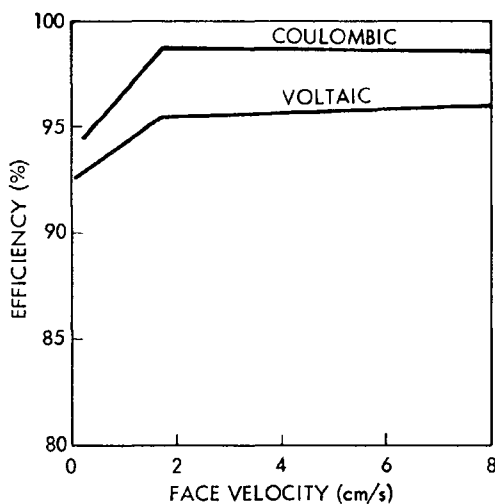


Fig. 4.5.8 Effect of Face Velocity on Redox Half-cell Cyclic Efficiency

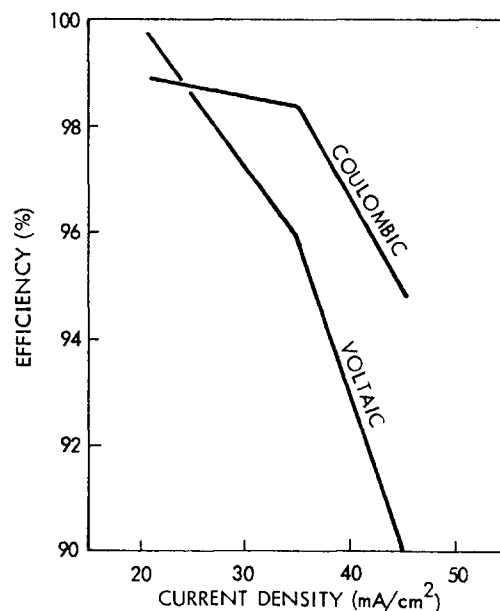


Fig. 4.5.9 Effect of Current Density on Redox Half-cell Cyclic Efficiency

Table 4.5.2
EFFECT OF FLOW GEOMETRY ON REDOX ELECTRODE EFFICIENCIES
(35 mA/cm²)

Geometry	Limiting Current Density (mA/cm ²)	Voltage Efficiency (%)	Coulombic Efficiency (%)	Energy Efficiency (%)
Flow-By	20	91.5	78.8	72.1
Flow-Through	100	95.9	98.9	94.8

4.5.2.3 Charge Capacity Variation with Temperature and NaOH Concentration.

The variables of temperature and NaOH concentration are important to the overall capacity of the redox half-cell, since both control reactant concentration which, in turn, affects the half-cell potential, moving it either toward or away from the oxygen evolution potential. Figure 4.5.10 details the effect of NaOH concentration and temperature upon the redox half-cell potential as a function of state of charge. As noted in figures 4.3.3 and 4.3.4 the solubilities of sodium ferro- and ferricyanide are directly related to temperature and inversely related to NaOH concentration, so that at low temperatures and high NaOH concentrations the evolution of oxygen becomes a serious competing reaction, degrading the coulombic efficiency. Thus the cyclic coulombic efficiency is only 85% for a 5N NaOH electrolyte over a 12-hour cycle at 12.1 A-hr and 73% state of charge, whereas, with 2.2 N NaOH, the coulombic efficiency is 99% during a 16-hour cycle at 15.6 A-hr and 70.3% state of charge.

Testing at 30°C resulted in 65% cyclic coulombic efficiencies due to low solubility of sodium ferrocyanide whereas, at 40 and 50°C, there was sufficient redox reactant concentration due to increased solubilities to prevent concentration polarization-induced oxygen evolution, and good cyclic efficiencies (90-99%) were realized.

4.5.2.4 Redox Half-cell Cyclic Life Study. At the present time, cyclic life testing of a ferro- ferricyanide redox half-cell, operating at a charge/discharge current density of 35 mA/cm^2 , stands at 1040 4-hour cycles (4160 hours) with a statistical mean energy efficiency of $86 \pm 13\%$. Figure 4.5.11 shows the cycle-by-cycle energy efficiency with 11-point data smoothing applied. The data scatter is due to several contributors listed in order of importance: (1) Gassing or crystallization of ferrocyanide leading to blockage of the reference electrode causing erroneous discharge cutoffs; (2) occasional thermo-control malfunctions with resultant flow blockage due to crystallization of ferrocyanide at lower ambient temperatures; (3) variations in electrolyte flow rate in the counterelectrode resulting in erroneous half-cell potentials due to gas blockage of current paths; and (4) poor resolution of the 8-bit A/D converter which can only resolve the 400 mV controlling redox half-cell potential to $\pm 20 \text{ mV}$. The downward trend in energy efficiency occurring after 600 4-hour cycles is tentatively attributed to a decreased volumetric flow rate caused by partial plugging of the sintered nickel electrode with consequent increased concentration polarization and decreased voltaic efficiency (figure 4.5.12). Coulombic efficiency remains unaffected (figure 4.5.13).

4.6 SINGLE CELL CYCLIC TESTING

4.6.1 Current Density

Full cell cyclic testing to evaluate the influence of current density reveals an optimum energy efficiency of 88% at a current density of 25 mA/cm^2 as shown in figure 4.6.1. Extrapolation of the energy efficiency curve down to 70% implies a maximum current density of 55 mA/cm^2 for both the charge and discharge cycles. However, higher discharge rates are obtainable if an unbalanced charge/discharge power profile is used; i.e., 35 mA/cm^2 charge and 100 mA/cm^2 discharge. Thus, from Figures 4.6.3 and 4.6.2 the voltaic efficiency for this charge/discharge profile is 83%. Figure 4.6.2 shows the variation of cell polarization with current density for charge and discharge modes.

The difference in the open circuit potential (OCV) of a charged and a discharged cell is due to the redox electrode. This half-cell OCV is a function of the state of charge.

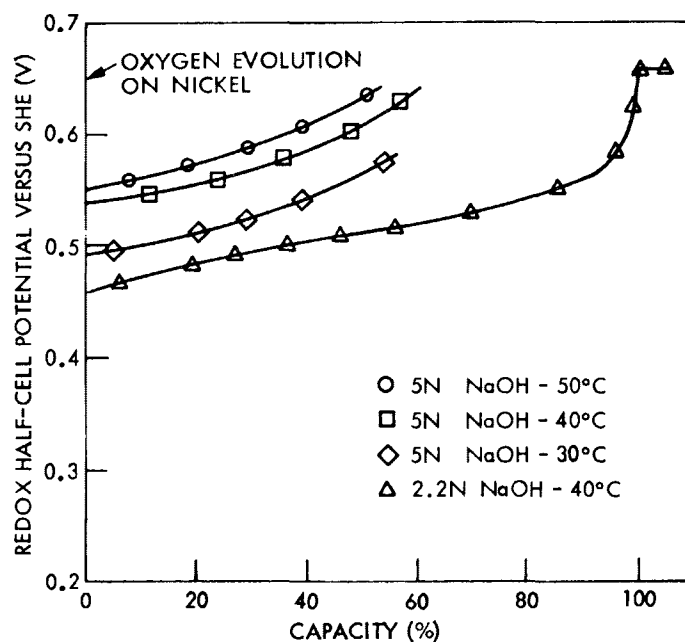


Figure 4.5.10 Effect of NaOH Concentration and Temperature upon Redox Half-cell Potential as a Function of State of Charge

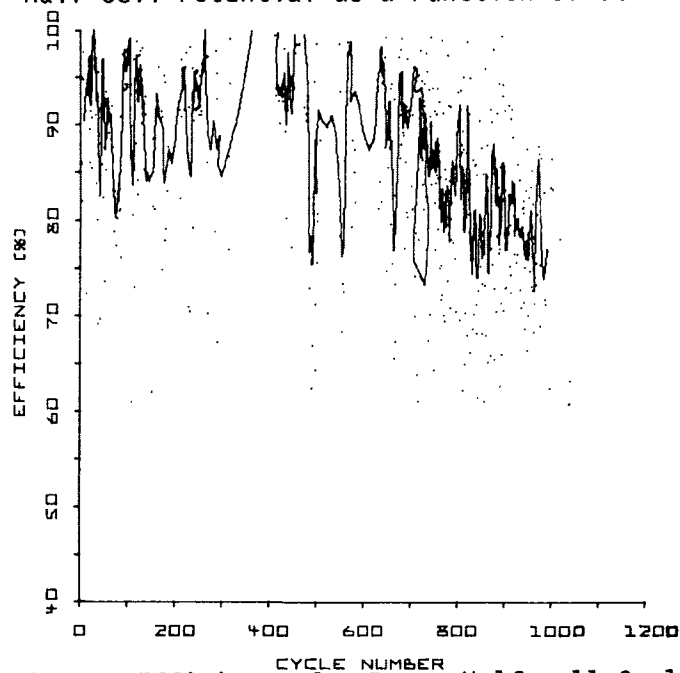


Figure 4.5.11 Energy Efficiency for Redox Half-cell Cyclic Life Test (continuing)

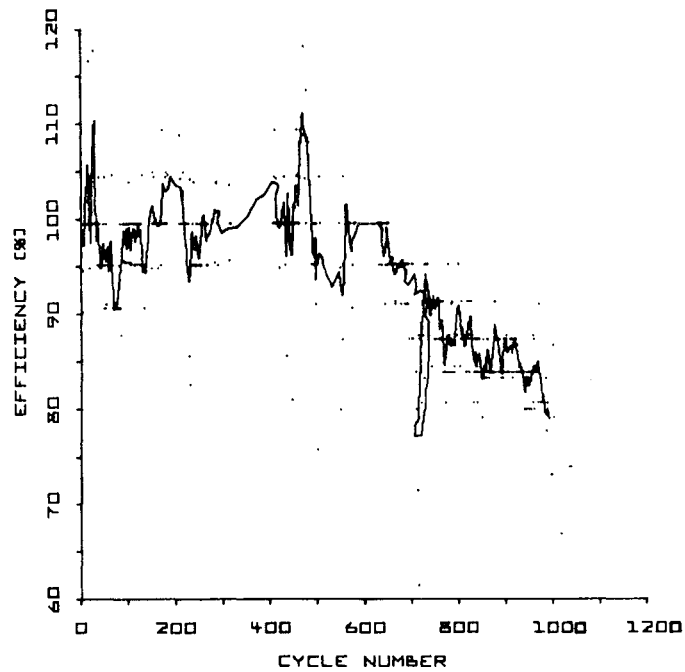


Figure 4.5.12 Voltaic Efficiency for Redox Half-cell Cyclic Life Test (continuing)

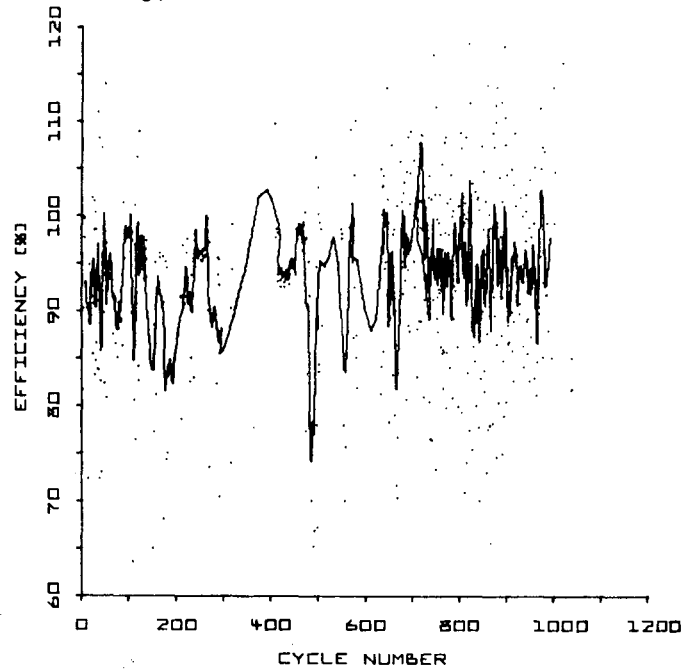


Figure 4.5.13 Coulombic Efficiency for Redox Half-cell Cyclic Life Test (continuing)

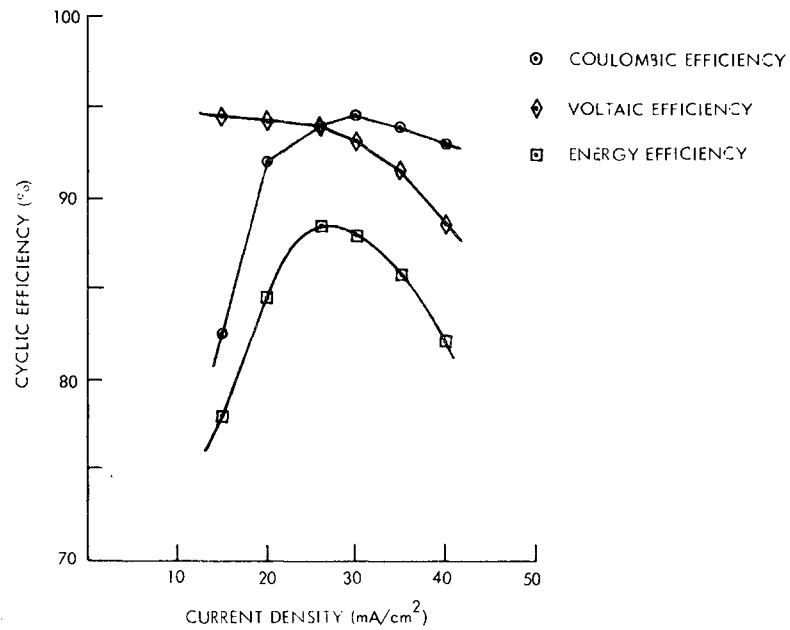


Fig. 4.6.1 Variation of Voltaic and Coulombic Efficiencies with Current Density for a Zinc-ferricyanide Hybrid Redox Battery

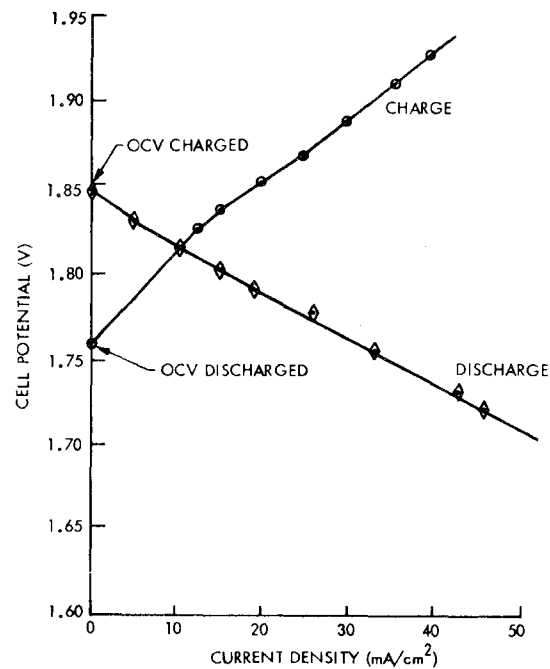


Fig. 4.6.2 Variation of Cell Polarization with Current Density

Figure 4.6.3 shows the power output curve for the zinc-ferricyanide hybrid redox battery. As can be seen, this battery system is capable of high energy discharge rates. From this curve, assuming an energy efficiency cut-off at 70% and a coulombic efficiency of 85%, approximately 217 watts/ft² is obtainable. The present design of the electronic control hardware limits the discharge rates to 56 watts/ft²; thus, battery performance should be investigated at high discharge rates and, since zinc passivation may occur at high current densities, it will be necessary to determine the effect of electrolyte flow rate upon the efficiency of discharge.

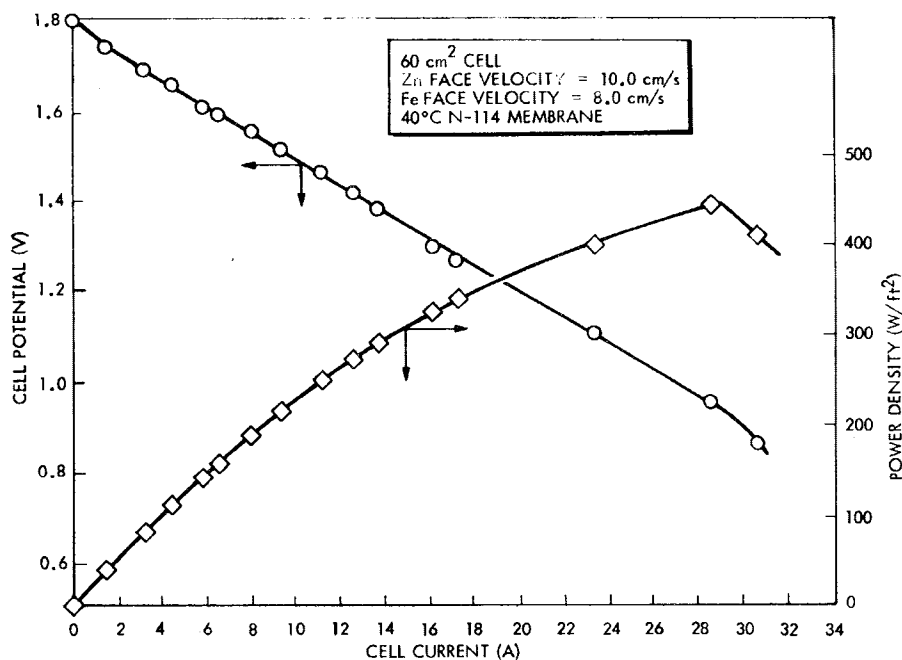


Fig. 4.6.3 Power Output of Zinc-Ferricyanide Hybrid Redox Battery

4.6.2 Charge Capacity

Due to a zinc capacity limit of 92 mA-hr/cm² for the present cell design, section 4.5.1.6, tests were not conducted upon the full cell configuration beyond this capacity limit; however, up to this limit, the cell operates with uniform cyclic efficiency.

4.6.3 Sodium Hydroxide Concentration

Similarly, due to low zinc half-cell coulombic efficiency at lower sodium hydroxide concentrations, full cell tests were not conducted at NaOH electrolyte concentrations below 5N.

4.6.4 Temperature

Half-cell cyclic efficiency evaluations as a function of temperature showed that zinc is not affected by temperature whereas the redox electrolyte solubility varies with temperature and hence the cyclic efficiency is affected. Full cell cyclic testing to study the influence of temperature upon cyclic efficiencies again supports these observations as shown in figure 4.6.4 where good performance is obtained at or above 40°C. Redox half-cell potentials indicate low solubility of sodium ferrocyanide is the prime cause for low efficiency at 30°C.

4.6.5 Full Cell Cyclic Life Testing

Cyclic life testing of two full cells has been conducted. The first test ran for 267 cycles at 4.2 - 4.6 A-Hr capacity, an average current density of 35 mA/cm² and had the following full-cell efficiencies.

Voltaic	87±6%
Coulombic	90±8%
Energy	79±7%

The test was terminated due to an electrolyte leakage problem. Upon disassembly, the electrode surfaces were seen to be in very good condition with no signs of corrosion or nonuniform zinc deposition.

Figure 4.6.5 displays the cycle-by-cycle energy efficiencies with 11-point data smoothing applied. Data scatter has been discussed in section 4.5.2. Figures 4.6.6 and 4.6.7 display the voltaic and coulombic efficiencies respectively.

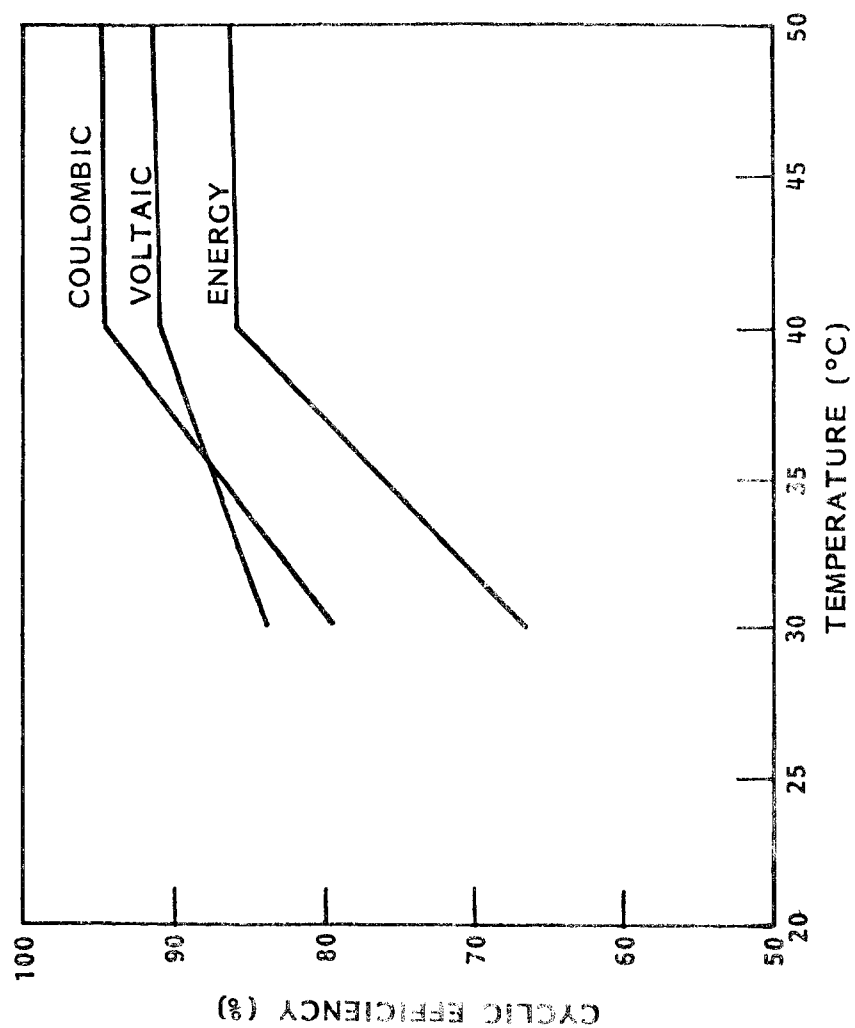


Figure 4.6.4 Variation of Full-Cell Cyclic Efficiency with Temperature

A second full cell test currently has accumulated 698 cycles with an appreciable upward trend in coulombic efficiency as shown in Figure 4.5.8. Figures 4.5.9 and 4.5.10 depict the voltaic and energy efficiencies for this battery. Test cycling is continuing with the current statistical mean efficiencies of

Voltaic 91 ± 2 %
Coulombic 92 ± 9 %
Energy 84 ± 9 %

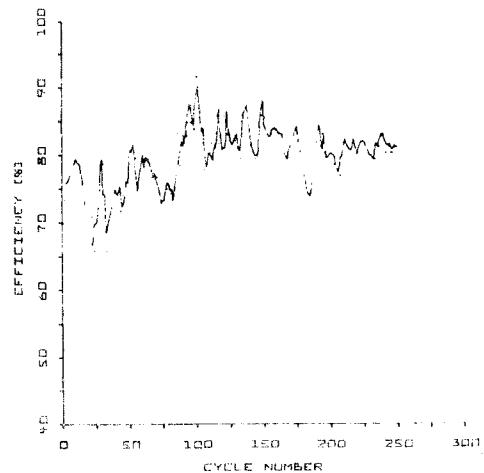


Figure 4.6.5 Energy Efficiency for Full Cell Cyclic Life Test (terminated)

4.6.6 Cell Operation Without Separator

A test was conducted to evaluate cell performance without a separator and having a common electrolyte reservoir. The cell was constructed having the highly permeable RAI experimental X1125-34-4 separator to maintain flow patterns, but to allow easy intermixing of electrolyte. The cell performed normally during a 2-hour charge having a cell potential of 1.9 volts; however, upon discharge the cell went into voltage reversal after one minute with the half-cell potentials indicating normal zinc operation but no ferricyanide reduction occurring. Apparently, ferricyanide is reduced about as rapidly as it is produced; hence it appears that a separator will be required for the battery.

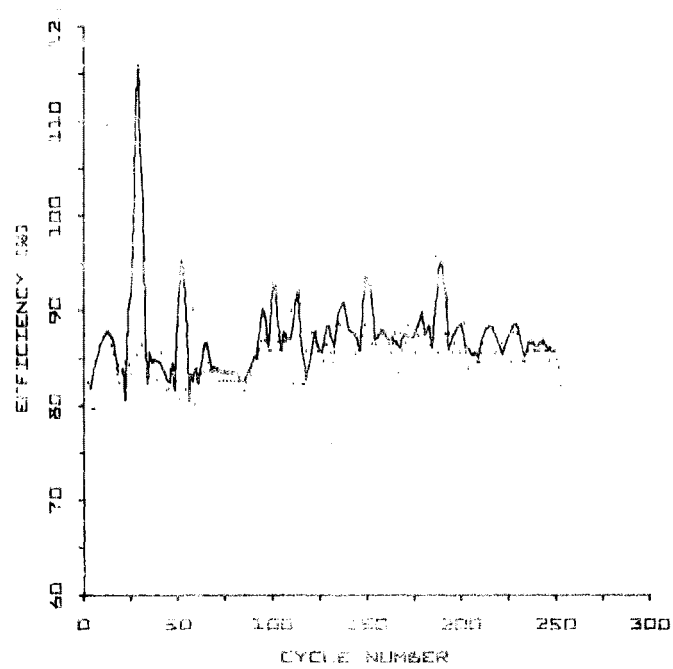


Figure 4.6.6 Voltaic Efficiency for Full Cell Cyclic Life Test (terminated)

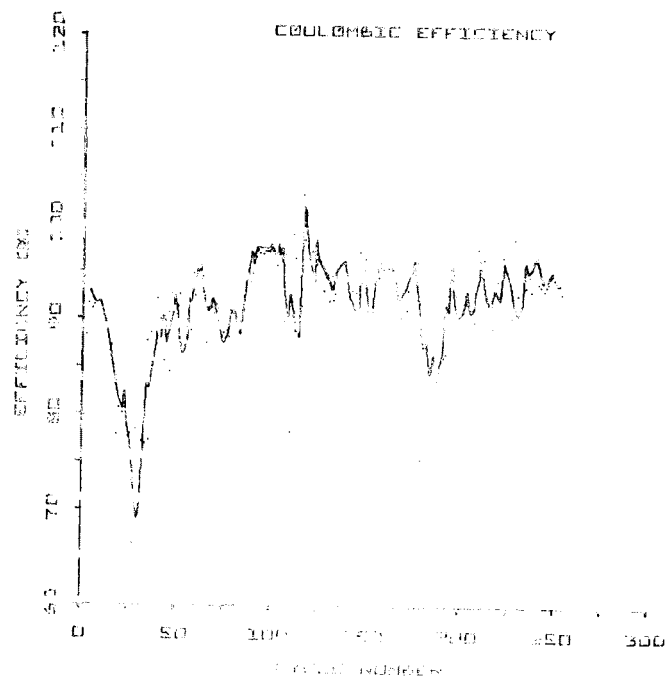


Figure 4.6.7 Coulombic Efficiency for Full Cell Cyclic Life Test (terminated)

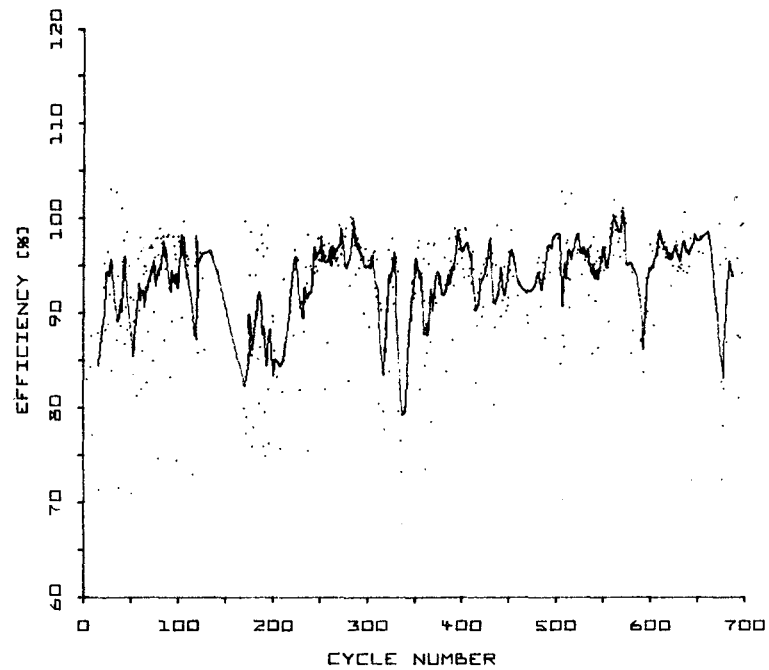


Figure 4.6.8 Coulombic Efficiency for Full Cell Cyclic Life Test (continuing)

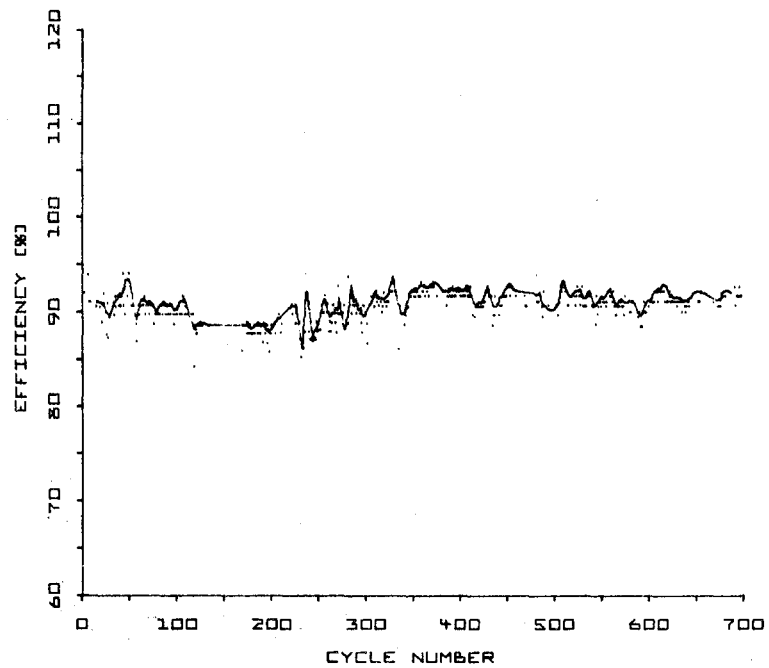


Figure 4.6.9 Voltaic Efficiency for Full Cell Cyclic Life Test (continuing)

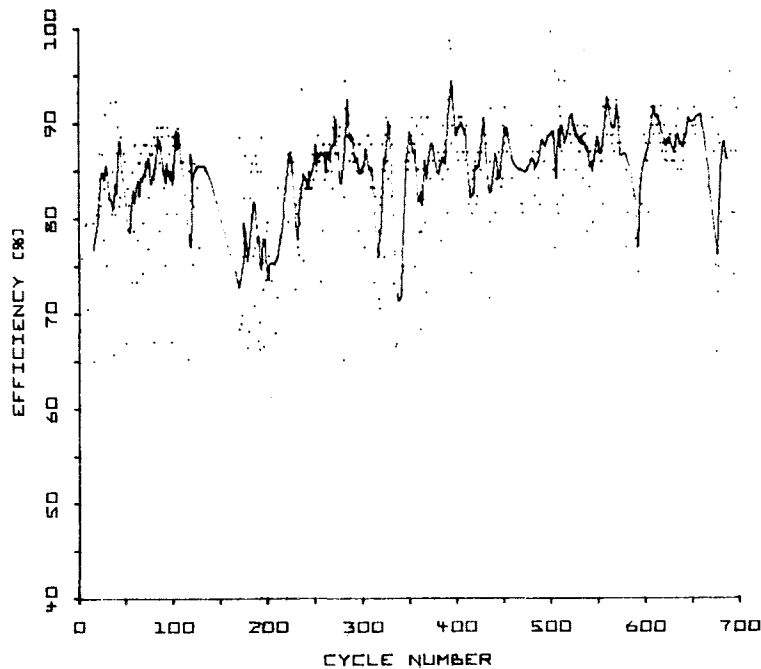


Figure 4.6.10 Energy Efficiency for Full Cell Cyclic Life Test (continuing)

4.7 COST ESTIMATE FOR LMSC MOD 00 ZINC/FERRICYANIDE 20 MW, 100 MWh BATTERY

Prior to the award of the present contract, a preliminary cost estimate for a 10MW, 85 MWh bulk electrical energy storage system was made in a preproposal study, based on the zinc/ferrocyanide electrochemical system. This estimate was modeled after that made previously by Warshaw and Wright (Ref. 57) for the NASA-LeRC flow redox battery, in which external centralized storage of electrolyte was a key feature. In this rough initial capital cost estimate, total system cost and size including a \$40/kW allowance for power and conditioning, but not including labor, ROI, taxes, overhead, and without a detailed battery design, was \$32.9/kWh. Based on the improved voltaic efficiency achieved with the system since that first estimate, this figure was reduced to \$30.0/kWh.

In the present cost estimate these figures are updated and upgraded using the Gould Mod 1 battery (Ref. 58) for a model to incorporate the zinc/ferrocyanide system, since operating voltage and current density are very similar, and a

separator is required. Certain changes in the overall Mod 1 battery design are then necessary. These are as follows:

- Centralized storage of electrolytes external to the battery modules is required for both the zincate and redox electrolytes. For zincate, centralized storage allows recovery, if required, of the sodium ferrocyanide (via procedure outlined in Section 2.2) redox reactant, which may gradually diffuse through the separator with extended cycling. For periodic religation of the redox electrolyte to maintain chemical stability, centralized redox storage is also required.
- Centralized storage is environmentally feasible with this system, due to low toxicities of electrolyte components (compared, for example, to chlorine and bromine).
- Costs for reactants and electrolyte are based on the zinc/ferricyanide system.
- Costs for electrodes and separators are based on electrode areas derived from the Mod 1 battery with both sides of each electrode used. Porous carbon electrodes of the type used in the EDA Mod 4 zinc-chloride battery (Ref. 59) are assumed.
- Costs for tankage, pumps, heat exchangers, and filters are estimated from the NASA-LeRC cost project (Ref. 57) increased by the MWH ratio 100/85. Costs projected for pumps and filters are based on an electrolyte volume equal to the volumes of reactants used, computed as solids. That is, half of the volume of each storage tank is assumed to be electrolyte.*
- Internal plumbing costs are assumed to be 50% of those used in the Gould Mod 1 estimate, due to centralized electrolyte storage.

*This electrolyte volume may be reduced considerably; the minimal volume would be the internal battery volume plus piping volume plus, say, 10% of the tankage volume. The actual minimal value will depend upon dissolution rates of the stored solids relative to conversion rate in the battery.

- Cell diodes for protecting the negative electrode from irreversible modification effects on the event of cell reversal have been eliminated, since graphite electrodes are used.
- Module length has been reduced by 1/3 to 11 feet, since individual electrolyte and bromide storage tanks are not used in the LMSC Mod 00 design.
- Costs for module racks are those for the lower cost design used in the EDA Mod 4 zinc/chlorine system (Ref. 59), adjusted upward by the factor 1.25 to account for the power density difference between the Gould and EDA systems.

These racks would be sectioned into 11-foot truckable lengths (5 feet wide by 6 feet high) to accept one 400-kWh LMSC Mod 00 module per section, which would be factory installed. These sections would be stacked three high on site, and would be bolted together. Two adjacent racks of these sections would constitute a string, and would have electrolyte piping and bus bars centrally located along stacks, running the length of each string. Strings would be 5 modules long.

The overall modular make-up of the LMSC Mod 00 battery would be similar to the Gould Mod 1 design. Stack size would be closely comparable. Electrode area is 0.0929 m^2 (or 1 ft^2) on a side. The Mod 1 20 MW, 100 MWh, battery is structured as indicated in Table 4.7.1.

Thus, the battery consists of 240 421-kWh modules arranged in 16 strings of 15 modules each, or 8 "doubled" strings to accommodate the rack design discussed above. With three tiers per string, each double string would be 55 feet long by 10-1/2 feet wide by 18 feet high. Estimated footprint for the on-site battery layout shown in Fig. 4.7.1 is 8.7 kWh/ft.^2 *

*This figure is increased to 9 kWh/ft.^2 if rectangular-shaped electrolyte storage tanks are used.

Table 4.7.1

LMSC MOD 00 20MW, 100-MWh UTILITY BATTERY COMPONENT RATINGS

Component	Discharge Voltage	Capacity* (kAh)	Stored Energy (kWh)	Number of Units
Electrode Pair	1.74	0.372	0.65	14 negatives } in parallel/ 13 positives } cell
Cell	1.74	4.84	8.42	5 cells in series/submodule
Submodule	8.70	4.84	42.1	10 submodules in series/module
Module	87.0	4.84	421	15 modules in series/string
String	1305	4.84	6315	16 strings in parallel/battery
Battery	1305	77.4	101,000	

*Negative electrode/loading is assumed to be 2000 Ah/m².

The estimated purchased materials and components of the LMSC Mod 00 modular battery in 1978 dollars per kilowatt-hour are summarized in Table 4.7.2. Estimated capital costs of a manufacturing plant capable of producing 25 100-MWh utility energy storage facilities per year are included (Ref. 58, 59). Calculation of these costs follows EPRI guidelines for standardized costing (Ref. 61). Cost computations for individual components are detailed in Appendix I.

To the summation of the individual component costs is then added costs for labor plus overhead, profit and tax, to develop figures for production factory cost and battery selling price. These items are detailed in Table 4.7.3. It is immediately apparent from an inspection of this table that Purchased Materials Costs constitute the major contribution to battery price.

From Table 4.7.2 it is seen that major contributions to Purchased Materials and Components costs derive from:

	(2000 Ah/m ²)	(3000 Ah/m ²)
Redox reagent	21.2%	26.4%
Electrodes	13.0	10.8
Separators	11.9	9.9
Busbars	12.5	10.4
Other cell-related	<u>14.2</u>	<u>11.8</u>
	72.8%	69.3%

(Above figures are computed as percentages of total
Purchased Materials and Components Cost.)

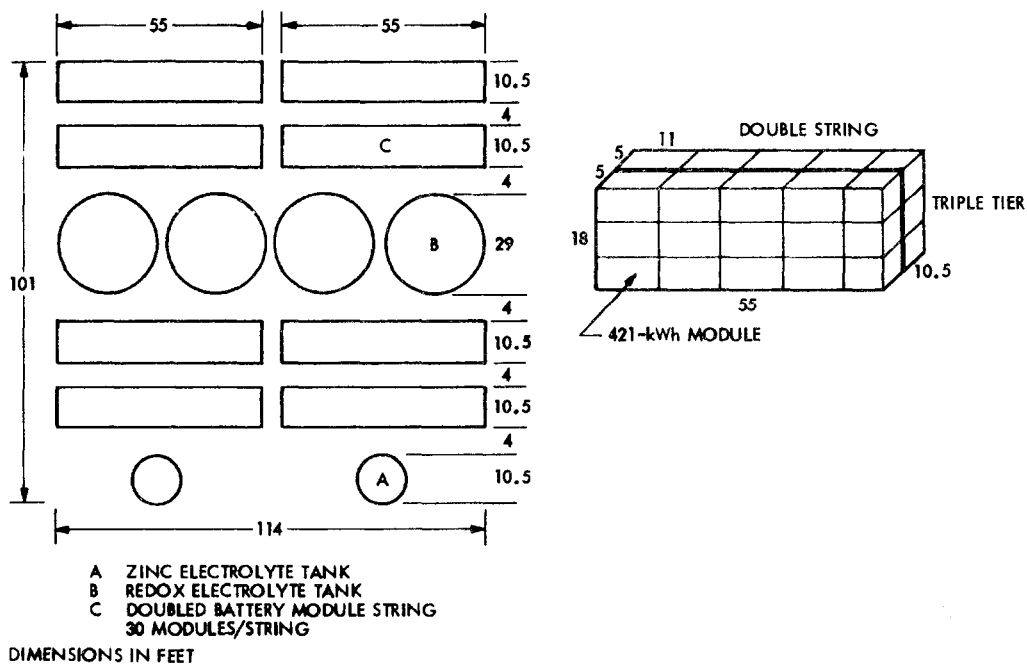


Figure 4.7.1 On-Site Layout of 20 MW, 100-MWh Bulk Electrical Energy Storage Facility

Table 4.7.2
PURCHASED MATERIALS AND COMPONENTS
COSTS FOR LHSC MOD 00 BATTERY
(\$/kWh)

Discharge Capacity (Ah/m ²)	<u>2000</u>	<u>3000</u>
<u>Power Related</u>		
Electrodes		
positive	2.22	1.48
negative	1.33	0.39
Separators	3.24	2.16
Other Cell Components	1.87	1.25
Internal plumbing	2.00	1.33
Busbars	3.41	2.27
Racks and Panels	<u>2.15</u>	<u>1.43</u>
	16.22	10.31
<u>Energy Related</u>		
Reagents		
Na ₄ Fe(CN) ₆ · 10H ₂ O	5.78	5.78
ZnO	0.83	0.83
NaOH	0.43	0.43
Tanks	1.16	1.16
Exterior Plumbing	0.32	0.32
Filter	0.82	0.82
Pumps	1.11	1.11
Heat Exchanger	<u>0.58</u>	<u>0.58</u>
	11.08	11.08
Total Cost	27.30	21.39

Since the cost of the redox reagent, sodium ferrocyanide decahydrate, is 21-26% of the total purchased materials cost, it should be manufactured rather than purchased, unless price discounts substantially larger than 15% can be obtained for 50 kilo ton lots. The salt contains 37% water of crystallization and the chemical components, iron, carbon, and nitrogen are abundant; hence, manufacturing cost should be nominal. Plastic lined, reinforced concrete tanks should be less costly than steel and may be adequate.

It should be more economical to use one large rectangular tank for the redox reactant and two smaller rectangular tanks for the zincate electrolyte. Footprint for this geometry is estimated at 9 kWh/ft².

The selection of electrode material has not yet been finalized. An electroformed matte material developed on government contract at LMSC previously for fuel cell research applications should be relatively inexpensive and easier to incorporate into the present design than porous carbon or graphite. Analysis of final costs given in Table 4.7.3 shows that major contributions to capital cost arose from electrode costs in terms of labor and processing equipment costs. With this substitute electrode material, a considerable savings in labor and processing equipment costs should be realized.

The battery selling costs compare well with equivalent costs for the Gould Mod 1 battery, which are \$64/kWh and \$47/kWh for 2000 Ah/m² and 3000 Ah/m² capacities, respectively.

At the 3000 Ah/m² capacity level, the LMSC Mod 00 Zinc/ferrocyanide battery cost at \$31.44 compares favorably with four other utility battery candidates (Ref. 60), Sodium/Sulfur (G.E.) \$35.98; Sodium/Antimony Trichloride (E.S.B.) \$39.60; Lithium/Iron Sulfide (A.T.) \$34.27* and Zinc/Chlorine (EDA) \$34.87.

The present cost estimate lacks the firm basis of a battery module designed specifically for the zinc/ferrocyanide system; and therefore can only be considered a first approximation to costs which will be derived from a finalized

*Cell-related costs only

design. It is realized that a bipolar rather than a monopolar battery design would be more cost-effective for the system proposed. These costs, however, are more realistic than those projected in a 1977 ROM preproposal estimate, and serve to identify major cost components. Such an analysis can be used as a guide in the final selection of module and battery design as scale-up proceeds in the development of this electrochemical system.

Table 4.7.3
ESTIMATED MANUFACTURING COST FOR LMSC MOD 00
20 MW, 100 MWh, ZINC/FERRICYANIDE BATTERY
(costs are \$/kWh unless noted otherwise)

Discharge Capacity (Ah/m ²)*	<u>2000</u>	<u>3000</u>
Direct Labor	1.86	1.36
Labor Overhead @ 150%	<u>2.79</u>	<u>2.04</u>
Direct Labor + Overhead	4.65	3.40
Purchased Materials & Components	27.30	21.89
Materials Overhead @ 10%	<u>2.73</u>	<u>2.19</u>
Materials, Components + Overhead	30.03	24.08
Floor Space Rental (Ref. 3)	0.19	0.13
Equipment Depreciation (Ref 3)	<u>0.58</u>	<u>0.39</u>
Factory Cost	35.45	28.00
Initial Equipment Costs (\$)	11.37 M	7.58 M
Working Capital (\$) @ 30% of annual production factory cost	<u>26.59 M</u>	<u>21.00 M</u>
Initial Invested Capital (\$)	37.96 M	28.58 M
After-Tax Return on Investment @ 15%	5.69 M	4.29 M
Factory Cost	35.45	28.00
After-Tax Return on Investment	2.28	1.72
Taxes	<u>2.28</u>	<u>1.72</u>
FOB BATTERY SELLING PRICE	40.01	31.44

*Based on negative electrode area

Section 5

CONCLUSIONS AND RECOMMENDATIONS

This 12-month applied research and development program has been directed primarily to the technical advancement of the rechargeable alkaline zinc/ferricyanide hybrid redox battery. The key technical aspects of this system have been investigated and overall system feasibility has been demonstrated.

This electrochemical system is well suited to utility load leveling and solar voltaic/wind applications, with advantages of high cell voltage, near ambient temperature operation, alkaline electrolyte, low toxicity, low-cost reactant storage, potentially long cycle life and low projected capital costs.

A preliminary economic assessment, updated to include improved performance demonstrated in the current applied research program, suggests that an initial capital cost of \$160/kW or \$32/kWh are reasonable current figures for a 20-MW (100 MWh) electrically rechargeable bulk energy storage plant, based on this electrochemical system.

The system in its present early state of development demonstrates excellent electrochemical performance. Cell voltages are 1.94 V on charge and 1.78 V on discharge at 35 mA/cm². In full-cell cyclic testing at 40°C, mean energy efficiency stands at 84% after 760 four-hour cycles with energy efficiency currently at 87% , as cycling continues. Mean coulombic and voltaic half-cell efficiencies are 93% coulombic and 91% voltaic for the iron redox electrode and 99% coulombic and 90% voltaic for the zinc electrode. A ferro-ferricyanide redox half cell cyclic test now stands at 1110 four-hour cycles with mean energy efficiency at 86% ; with cyclic testing continuing.

Based upon the results of the above study certain key technical areas have been identified as requiring additional effort. These areas, which are the basis of a follow-on proposal to DOE for a Phase II Study (Ref. 62) are summarized here:

- With the redox electrode, further work is required in two areas. There is a need for a lower cost electrode substrate than the porous nickel plaque now used. Activated porous carbon and carbon felts, as well as nickel-plated porous carbon and a special low-cost nickel electroformed mesh developed previously on government contract in this laboratory, will be evaluated. Additional information is required on the long-term stability of alkaline sodium ferricyanide electrolyte. This will be obtained under appropriate long-term cyclic half-cell testing, using extended charge-discharge cycles. A quantitative technique has now been perfected for measuring the degradation rate of alkaline ferricyanide solutions. Using this technique, concepts for maintaining the long-term capacity of the redox electrode will be evaluated.

- In order to extend long-term cycle life and to meet the zinc electrode loading capacity goal of 2000 Ah/m^2 at 90% energy efficiency, further work is necessary. This work will encompass microscopic studies on the quality of zinc electrodeposition obtained using electrolyte additives and flow rate/flow distribution modification in half-cell cycling. In addition, alternative substrate materials will be evaluated. These include carbon felt, porous carbons and graphites, and an alloy of copper with cadmium, tin, or zinc electrodeposited onto the above or an iron substrate.

- The evaluation of alternative low-cost separator materials, such as uniformly cross-linked and radiation grafted polyethylene and polypropylene will include studies to determine specific resistance, iron and zinc permeation rates, as well as mechanical and chemical stabilities. Attractive separators will be evaluated further under cyclic conditions to determine voltaic losses and mechanical strength. Emphasis will be upon the selection of a low-cost, economical separator having the desired characteristics.

- A mathematical model of the zinc-ferricyanide battery will be developed with the emphasis upon optimizing the system design.

Section 6
REFERENCES

1. P. A. Nelson and N. P. Tao, "Batteries for Utility Load Leveling," Preprint, 38th American Power Conference, Chicago (Apr 1976)
2. S. Arouete, K. F. Blurton, and H. G. Oswin, J. Electrochem. Soc., 116, 166 (1969)
3. M. D. Zhouldey & V. V. Stender, Zhur. Priklad. Khim., 31, 711 (1958)
4. S. Arouete & K. F. Blurton, Final Report, N67-39054 (Oct. 31, 1966)
5. J. E. Oxley & C. W. Fleischmann "Improvement of Zinc Electrodes for Electrochemical Cells", Third Quarterly Report, 30 pp, N66-26870 (March 1966)
6. A. I. Samokhotskii & A. P. Byudov, Vestnik Metalloprom, 15 (12), 122-124 (1935) CA 31:2155
7. J. Goodkin & F. Solomon, "A Zinc-Silver Oxide Cell for Extreme Temperature Application", in Batteries 2, Research and Development in Non-Mechanical Electrical Power Sources, Proc. 4th Internat'l Symposium, D. H. Collins, editor, Pergamon Press, N.Y., pp 475-487 (1964)
8. J. Legros, J. Chim Phys., 61, 909 (1964)
9. R. D. Naybour, J. Electrochem Soc., 116, 520 (1969)
10. E. D. Farmer & A. H. Webb, J. Applied Electrochem., 2, 123 (1972)
11. F. Mansfield, S. Gilman, J. Electrochem. Soc., 117, 1154 (1970)
12. J. W. Diggle & A. Damjanovic, J. Electrochem. Soc., 117, 65, (1970)
13. R. V. Bobker, "Zinc in Alkali Batteries", The Society for Electrochemistry, p. 13, University of Southampton, Great Britain (1973)
14. T. F. Sharpe, J. Electrochem. Soc., 122, 845 (1975)
15. J. Bressan, P. Gaultochet, & R. Wiert, Extended Abstracts, 27th ISE Meeting, Zurich, Switzerland, Abstract No. 78 (Sept. 6 - 11, 1976)
16. O. Wagner & A. Hiny, Proceedings 27th Power Sources Symposium, p. 135 (1976)
17. P. P. Gregory, P. C. Jones, D. P. Redfearn, J. Electrochem. Soc., 119, 1288 (1972)
18. F. Mansfield & S. Gilman, J. Electrochem. Soc., 117, 1328 (1970)

19. A. L. Hampartzumian & R. V. Moshkev, Power Sources 3, 495 (1970)
20. R. W. Lewis & J. Turner, J. Applied Electrochem., 5, 343 (1975)
21. R. W. Powers, et al., "Plating from Alkaline Solutions," International Lead Zinc Research Organization Research Reports, ILZRO Project No. ZE-120 (1969)
22. J. W. Diggle and A. Damjanovic, J. Electrochemical Soc., 119, 1649 (1972)
23. Great Britain Patent #1364874, Union Carbide (8-29-74)
24. H. Creutz, Ger. Offen. 2,517,781 (17 Jan 74)
25. H. G. Oswin and K. F. Blurton, in Zinc-Silver Oxide Batteries, A. Fleischer and J. J. Lander, eds., New York, John Wiley & Sons, Inc., p. 184 (1971)
26. G. Dalin, in Zinc-Silver Oxide Batteries, A. Fleischer and J. J. Lander, eds., New York, John Wiley & Sons, Inc., p.94 (1971)
27. W. M. Latimer, Oxidation Potentials, 2nd ed., Prentice-Hall, Inc., (1956)
28. L. M. Peter, et al., J. Electroanal. Chem., 71, 31-50 (1976)
29. D. M. Tench and E. Yeager, J. Electrochem. Soc., 121, 318-372 (1974)
30. Yu. S. Gurinov, et al., Zh. Fiz. Khim., 37 (5), 1141-1143 (1963)
31. R. Sohn, et al., Electroanal. Chem. and Interfacial Electrochem., 50, 55-63 (1974)
32. L. R. Sharma and J. Dutt, Indian J. Chem., 8, 170-173 (1970)
33. A. G. MacDiarmid & N. F. Hall, J. American Chem. Soc., 76, 4222 (1954)
34. G. Emschwiller, J. Legros, Compt. Rend., 261, 1535 (1965)
35. K. A. Hoffman, Ann., 312, 1 (1900)
36. A. G. Sharpe, "The Chemistry of Cyano Complexes of the Transition Metals" Academic Press, N.Y. p. 107 (1976)
37. S. Asperger, I. Murati, D. Pavlovic, JCS, (A), 2044 (1969)
38. A. G. Sharpe, loc. cit., p. 126
39. F. Holze, W. Stockman, Monatash, 56 253 (1930)
40. G. Emschwiller, Compt. Rend., Ser. C, 274, 1500 (1972)
41. G. Emschwiller, Rev. Roum. Chim., 17, 131 (1971)
42. K. D. Schleinitz, R. Landsberg, G. V. Louis, J. Electroanal. Chem., 28, 287 (1970)
43. G. Emschwiller, Compt. Rend., 238, 341 (1954)
44. G. Emschwiller, C. K. Jorgensen, Chem. Phys. Letters, 5, 561 (1970)
45. R. Gale, A. J. McCaffery, JCS Dalton, 1344 (1973)
46. G. Emschwiller, Compt. Rend., 260, 4333 (1965)

47. J. H. Espenson, S. G. Wolenuk, Inorg. Chem., 11, 2034 (1972)
48. G. Emschwiller, Compt. Rend., 259, 4281 (1964)
49. J. Dupessis-Legros, G. Emschwiller, Compt. Rend., 273C, 452 (1971)
50. R. Glauser et al., JACS, 95, 8457 (1973)
51. D. A. Dows, A. Haim, W. K. Wilmarth, J. Inorg. Nuc. Chem., 21, 33 (1961)
52. D. Hall et al., JCS, (A) 800 (1971)
53. A. G. Sharpe, loc. cit., pp 108, 115
54. H. E. Toma & J. M. Malin, Inorganic Chem., 12, 2080 (1973)
55. International Critical Tables V6, p 254
56. R. P. Hollandsworth, G. B. Adams, Role of Microcomputers in Secondary Battery Development and Testing (DOE Report to be published 12/79)
57. M. Warshay and L. W. Wright, Cost and Size Estimate for an Electrochemical Bulk Storage Concept, NTIS N 75-32593 (NASA-TM-X-71805) (Oct. 1975)
58. Assessment of Technical and Economic Feasibility of Zinc/Bromine Batteries for Utility Load Levelling, Electric Power Research Institute, EM-1059 (May 1979)
59. Development of Zinc-Chlorine Battery for Utility Applications, Mar 1978 - Electric Power Research Institute, EM-1051, Parts 1-3 (Apr 1979)
60. Interim Cost Estimates for Advanced Battery Systems, Dec. 1973, Electric Power Research Institute, EM-742 (July 1978)
61. J. H. B. George, Proposed Criteria for Estimating the Capital Costs of Advanced Battery Systems for Utility Energy Storage, EPRI Project 787-1, Arthur D. Little, Inc., Cambridge, Mass. (Nov 1976)
62. Proposal for Research and Development of a Rechargeable Alkaline Zinc/Ferro-Ferricyanide Hybrid Redox Battery, LMSC-DU85667 (Submitted DOE Aug. 1979)
63. A. Savitzky and M. J. E. Golay, Anal. Chem., 36 (8), 1627 (1964)

Appendix A

ELECTRONIC INSTRUMENTATION

The instrumentation constructed on this program was based on a National Semiconductor Single Board Computer System model RMC 80/11 with appropriate Input/Output interfacing. The input interfacing consisted of a Burr-Brown model MP8632 8-bit, 32 differential channel analog to digital converter board, a current-to-voltage amplifier section, a hexadecimal (16 key) keypad for operator control and monitoring of cells and cell parameters, and a real-time clock for continuous update of time. The output interfacing consisted of teletype and digital tape cassette storage of reduced cell data transmitted via a RS-232 serial data interface, a 12-digit alphanumeric display enabling operator monitoring of cell operation, setpoints, and cell parameters, and a constant current source to charge and discharge the individual cells.

Current Instrumentation Amplifier

The current-to-voltage amplifier section, circuitry shown in figure A.1, was necessary in order to amplify the mV signal present at the current shunt to a signal level having sufficient resolution for A to D conversion. The current shunt having 150 mV output while passing 3 Amps was amplified 33.3X to provide the A/D with a signal of 4.995 volts. The resolution of the 8-bit converter is ± 20 mV, thus inaccuracy is 0.4% instead of 13.3% for the sensed mV potential. The input of the mV signal to the Burr-Brown 3629BP instrumentation amplifier is controlled by the relay logic controlling the constant current source relays because the A/D converter will only recognize a positive input. The OR signal from the E8 port of the microcomputer controls whether or not current is being passed through the current shunt, while the DR signal from the E9 port determines whether the cell is charging or discharging. These signals, via the TTL logic shown, control a relay to invert the shunt input signal upon discharge providing a positive output to the A/D.

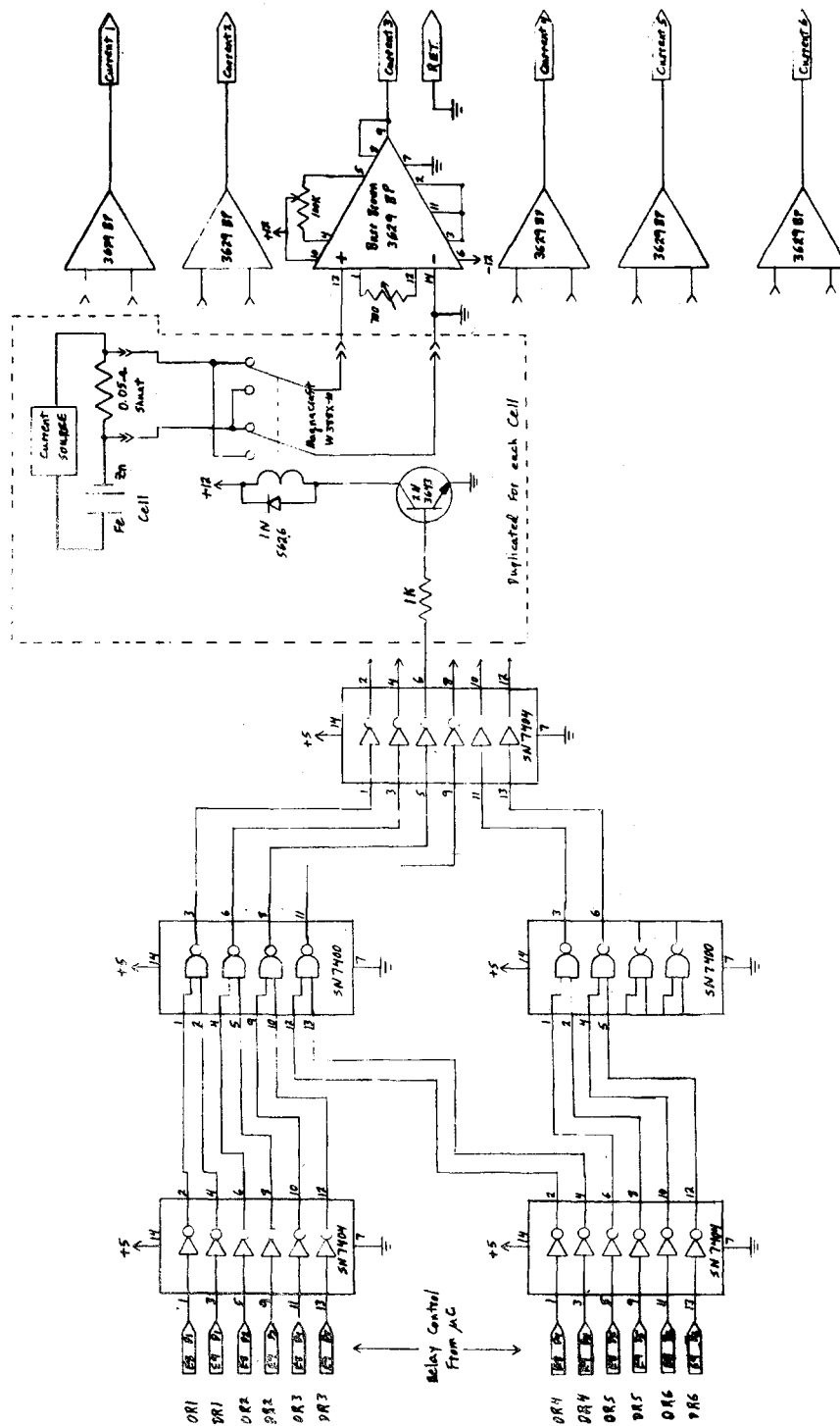


Figure A.1 Current-to-Voltage Instrumentation Amplifier Circuitry

Hexadecimal Keypad

The hexadecimal keypad circuitry shown in figure A.2 is based upon a 16-key, Centralab model M-1650-1 keypad. The SN 74147's and SN 7400 provide BCD coding of the keypad. The output from pin 8 of the SN 7400 signals that a key has been pressed and the SN 7405 and sonalert provide audible reinforcement to the operator. The switch at the bottom provides two choices for the operator. In the normal mode the display is used as a parameter monitor, whereas in the program mode, cell control parameters can be changed, and cell operation can be controlled by the operator. Input to the microcomputer is via the E6 port.

Real-Time Clock Interface

The real-time clock is required for continuous update of elapsed time while the computer performs various operations. The clock, shown in figure A.3, is essentially a frequency divider dividing the 12 VAC 60 Hz signal from the transformer of the μ C 12 VDC power supply down to one second periods and inputting to port EA an 8-bit digital code representing one to 255 seconds. As no microcomputer operation takes more than 60 seconds there is sufficient overlap for real-time updating of the microcomputer.

Alphanumeric Display

The monitoring of cell information and control parameters is made possible by a 12-digit alphanumeric display. Based upon three Litronix DL-1416 modules having 4 digits each, the display circuitry, shown in figure A.4, is relatively easy to interface with the microcomputer. The displays consist of seven data lines (D_0 - D_6), a chip enable ($\overline{CE}$), a cursor enable ($\overline{CU}$), a write line ($\overline{W}$), and two address lines (A_0, A_1). The data lines, address lines, and write lines are common for each module while each digit is controlled by the cursor and chip enable lines. Overall control of the display is via the Tiny Basic control program (Appendix B) outputting to ports E4 and E5. The SN 7407 hex inverter buffers are required due to the noisy industrial environment.



LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

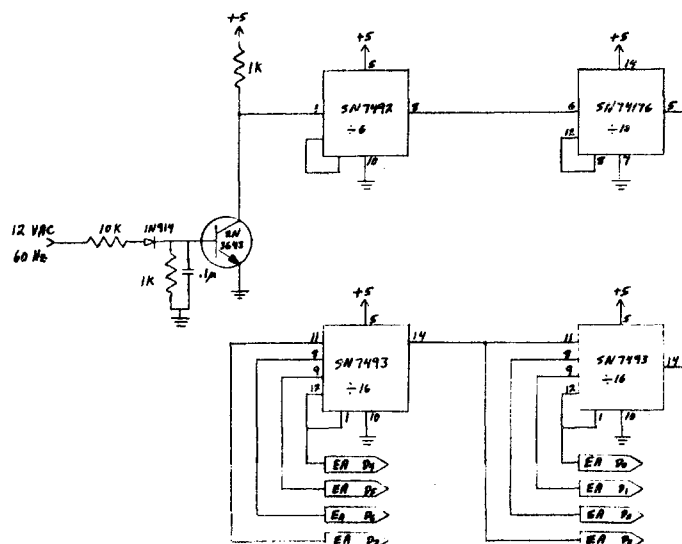


Figure A.3 Real-time clock interface

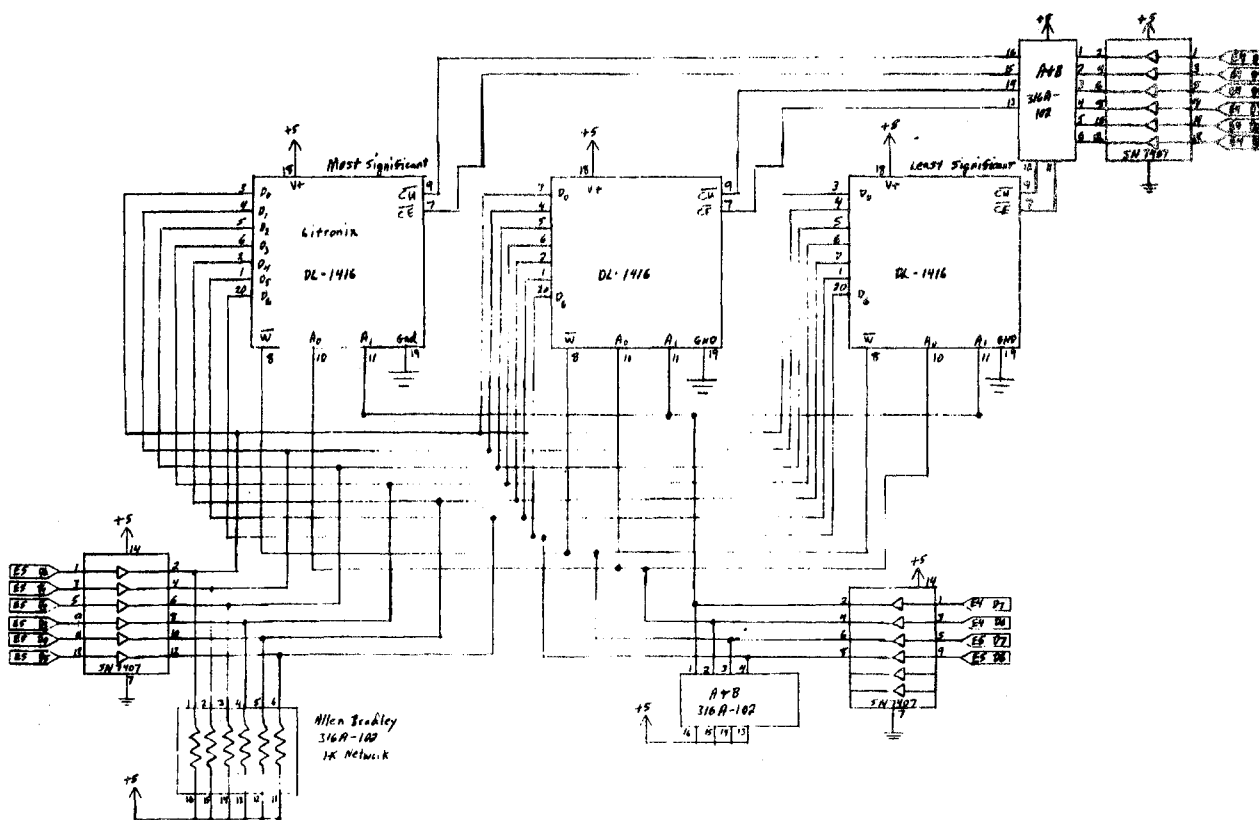


Figure A.4 12-digit Alphanumeric Display circuitry

If they are not installed as close to the displays as physically possible, electrical noise can destroy the CMOS display. During the course of this instrumentation development, six displays costing \$30 each were destroyed before remedial action could be devised and implemented.

Constant Current Sources

The constant current sources are the heart of the battery test stand. They are required to charge and discharge the cells at constant current under command of the microcomputer. The microcomputer will command the charge sequence by outputting on port E8 a digital word which turns on the "OR" (open circuit) relay via an optical isolator. The relay allows current to flow from the LM 350 voltage regulator into the cell. The current is maintained constant by a constant voltage drop across the 6Ω variable resistor with current limiting provided by the 0.5Ω resistor. Similarly the discharge sequence is started by having "OR" on and turning "DR" (Direction relay) on via a digital command from the microcomputer on port E9. This action turns on the right-hand discharge circuit. Open circuit periods are controlled by turning "OR" off. Figure A.5 details the circuitry for one of the six current source circuits having common $\pm 12V$ power supplies.

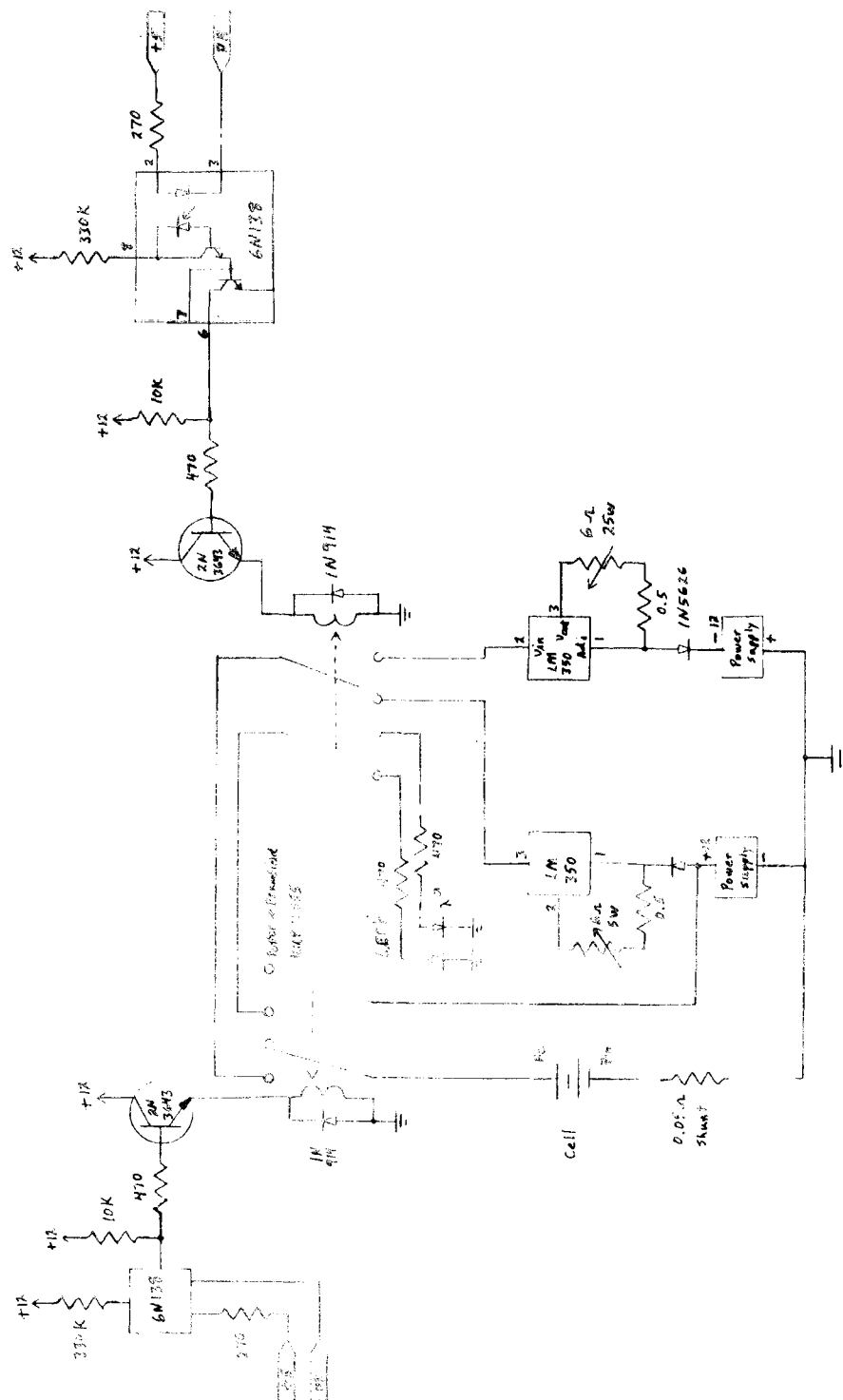


Figure A.5 Constant Current Sources for one cell

Appendix B

"TINY BASIC" CELL CONTROL PROGRAM

The Palo Alto Tiny Basic language is an interpretive language in which each line of code is independently translated to machine language and executed in turn. The assembly language code for this translation is given in Appendix C.

The language is a simplified set of the high order Basic language written and developed by Dr. Li-Chen Wang of LMSC. It was first described in the May 1976 issue of Dr. Dobb's Journal of Computer Calisthenics and Orthodontia. The present program was written for LMSC by Lincoln Semiconductor of Sunnyvale, CA whom we thank for their efforts.

In order to understand the control program it is first necessary to define the limits of the Palo Alto Tiny Basic language.

1. Numbers

All numbers are integers and must be less than 32767. Floating point math is not allowed. Numbers from arithmetic operations in excess of 32767 will result in stoppage of program execution via an error message.

2. Functions

There are three functions in the published language with six additional functions added by Lincoln Semiconductor.

ABS(X)	gives the absolute value of X
RND(X)	gives a random number between 1 and X (inclusive)
SIZE	gives the number of bytes of RAM unused by the program (includes variables of matrixes)
PEEK(X)	gives the value stored at memory location X
POKE(X,Y)	puts the value of Y into memory location X

PUT(X,Y)	Output to port X the value of Y
GET (X)	Input from port X
VOLT (X)	The voltage from A/D converter channel X
AMP(X)	The current from A/D converter channel X (This signal is corrected for shunt and instrumentation amplifier gains.)

3. Arithmetic and Compare Operators

<u>Operation</u>	<u>Function</u>
/	divide
*	multiply
-	subtract
+	add
>	greater than (compare)
<	less than (compare)
=	equal to (compare)
#	not equal to (compare)
>=	greater than or equal to (compare)
<=	less than or equal to (compare)

The +, -, *, and / operations result in a value between -32767 and 32767.
All compare operators result in a 1 if true and a 0 if not true.

4. Expressions

The common expressions found in the high order basics are found here,
namely:

LET, FOR, NEXT, IF, STOP, RUN, GOTO, GOSUB, LIST, PRINT,
INPUT, LIST, REMARK, NEW, RETURN, STEP.

The majority of the above can be truncated such as P. for Print, GOS. for
GOSUB, and R. for RETURN. The expression NEW will delete all prior statements.

5. Error Messages

The present Tiny Basic Code has been modified so that upon an error Port E8 is turned off which turns off all cell operations. If this was not done the cells would continue charging or discharging with no control.

The three error conditions in Tiny Basic are What, How, and Sorry.

<u>What?</u>	means it does not understand you and is usually a coding error
<u>How?</u>	means it understands you but does not know how to do it; usually due to arithmetic operation exceeding 32767
<u>Sorry?</u>	means it understands you and knows how to do it but there is not enough memory to do it.

The following listing details the Tiny Basic Control program developed to control the charge/discharge of 6 cells. Table B.1 details the identity of the @(J+X) matrix variable names, Table B.2 details the identity of operator variables, and Table B.3 gives the listing of the control program.

Table B.1 Identity of @ (J+X) Matrix Variables

X	Identification	X	Identification
1	S Status	45	TDU Time at T6
2	CN Cycle Number	46	UCU Ah at T6
3	CT Charge Time (h)	47	CPU Cell Pot. at T6
4	OC Overcharge Pot.	48	ZPU Zinc Pot. at T6
5	UC Usable Energy Pot.	49	FPU Iron Pot. at T6
6	ZC Zinc Depletion Pot.	50	CCT A-s Charge
7	FC Iron Depletion Pot.	51	DCT A-s Discharge
8	TAH Accumulated Time (h)	52	TAS Accumulated Time (s)
9	CP Cell Potential	53	TCT Temporary Time
10	ZP Zinc Potential	54	CD Charge Current Density
11	FP Iron Potential	55	DD Discharge Current Density
12	CV Average Charge Pot.	56	VE Voltaic Energy Efficiency
13	DV Average Discharge Pot.	57	UE Usable Coulombic Efficiency
14	+I Charge Current	58	OE Energy Conversion Efficiency
15	CC Charge (Ah)	59	FE Iron Coulombic Efficiency
16	UC Discharge (Ah)	60	ZE Zinc Coulombic Efficiency
17	-I Discharge Current	61	Not Used
18	SSF Secondary Status Flag	62	OCF Open Circuit Pot. Charged
19	V Wait Time OCV	63	OZP OCV Zinc Charged
20	TCC Time for T1	64	OFD OCV Iron Charged
21	CCC Ah at T1	65	O Zero Output
22	CPC Cell Pot. at T1	66	O Zero Output
23	ZPC Zinc Pot. at T1	67	DCP OCV Discharged
24	FPC Iron Pot. at T1	68	DZP OCV-Zinc Discharged
25	TCP Time at T2	69	DFP OCV-Iron Discharged
26	CCP Ah at T2	70	O Zero Output
27	CPP Cell Pot. at T2	71	CCC Ah at T1
28	ZPP Zinc Pot. at T2	72	CCP Ah at T2
29	FPP Iron Pot. at T2	73	UCD Ah at T5
30	TDZ Time for T3	74	UCZ Ah at T3
31	UCZ Ah at T3	75	UCF Ah at T4
32	CPZ Cell Pot. at T3	76	UCU Ah at T6
33	ZPZ Zinc Pot. at T3	77	Current at T1
34	FPZ Iron Pot. at T3	78	Current at T2
35	TDZ Time at T4	79	Current at OCV
36	UCF Ah at T4	80	Current at T3
37	CPF Cell Pot. at T4	81	Current at T4
38	ZPF Zinc Pot. at T4	82	Current at T5
39	FPF Iron Pot. at T4	83	Current at T6
40	TDD Time at T5	84	Ah DCP
41	UCD Ah at T5	85	Ah Discharge
42	CPD Cell Pot. at T5	86	Ah Charge
43	ZPD Zinc Pot. at T5	87	DD Discharge Current Density
44	FPD Iron Pot. at T5	88	TD Time at End Discharge

- T1 Test for 30 min for average charge voltage
 T2 Test for end of charge (either time or potential controlled)
 T3 Test for zinc depletion on discharge
 T4 Test for iron depletion on discharge
 T5 Test for 30 min for average discharge voltage
 T6 Test for usable energy potential

Table B.2
Table of Operator Variables

A	=	Status function
B	=	Base of 1st Battery Array
C	=	Cell number
D	=	Secondary Quotient
E	=	Enter function
F	=	Display function
G	=	Temporary
H	=	N/A
I	=	Temporary Index
J	=	Index to Array
K	=	Key pressed
L	=	Length of array for each cell
M	=	N/A
N	=	N/A
O	=	N/A
P	=	Direction relay bit pattern
Q	=	Temporary
R	=	Open circuit relay bit pattern
S	=	Status of cell
T	=	Timer
V	=	OCV Time
W	=	Temporary
X	=	Temporary
Y	=	N/A
Z	=	Base for A/D

Table B.3
TINY BASIC CONTROL PROGRAM

```

6 PUT235,137;PUT231,137;PUT232,0;REM INITIALIZE HARTS, OPEN CURRENT
9 LET B=305,C=1,F=1,L=90,Z=-9*256,V=60
10 FOR I=1TOB+6*L;LET @ (1)=0;NEXT I
11 @ (1)=239;@ (2)=238;@ (3)=237;@ (4)=236;@ (5)=247;@ (6)=246
12 @ (7)=245;@ (8)=244;@ (9)=251;@ (10)=250;@ (11)=249;@ (12)=248
13 GOSUB 1180
15 FOR I=13TO197;LET @ (1)=32;NEXT I
16 LET @ (16)=83;REM F=1
17 LET @ (27)=67,@ (28)=78;REM F=2
18 LET @ (39)=67,@ (40)=84,@ (47)=72,@ (48)=82,@ (44)=46;REM F=3
19 LET @ (51)=79,@ (52)=67,@ (55)=46,@ (59)=86;REM F=4
20 LET @ (63)=85,@ (64)=69,@ (67)=46,@ (71)=86;REM F=5
21 LET @ (75)=90,@ (76)=67,@ (79)=46,@ (83)=86;REM F=6
22 LET @ (87)=70,@ (88)=67,@ (91)=46,@ (95)=86;REM F=7
23 LET @ (99)=84,@ (100)=65,@ (104)=46,@ (107)=72,@ (108)=82;REM F=8
24 LET @ (111)=67,@ (112)=80,@ (115)=46,@ (119)=86;REM F=9
25 LET @ (123)=90,@ (124)=80,@ (127)=46,@ (131)=86;REM F=9
26 LET @ (135)=70,@ (136)=80,@ (139)=46,@ (143)=86;REM F=11
27 LET @ (147)=67,@ (148)=86,@ (151)=46,@ (155)=86;REM F=12
28 LET @ (159)=68,@ (160)=86,@ (163)=46,@ (167)=86;REM F=13
29 LET @ (172)=73,@ (175)=46,@ (179)=65;REM F=14
30 LET @ (183)=67,@ (184)=67,@ (188)=46,@ (191)=65,@ (192)=72;REM F=14
31 LET @ (195)=68,@ (196)=67,@ (200)=46,@ (203)=65,@ (204)=72;REM F=15
32 INPUT"YEAR"@ (230),"DAY"@ (229),"HR"@ (228),"MIN"@ (227);@ (226)=0
33 LET @ (232)=1
40 U=GET(234)
50 FOR J=B TO B+5*LSTEPL;LET @ (J+1)=0,@ (J+2)=0,@ (J+3)=20,@ (J+4)=225
52 LET @ (J+5)=145,@ (J+6)=80,@ (J+7)=30;NEXT J
60 LET P=126,R=0;PUT233,P
100 T=GET(234);IFT=U GOTO 100
101 IFT<0LET T=T+256
102 LET T=T-U,U=U+T;IFU>=256LET U=U-256
110 LET @ (226)=@ (226)+T;IF@ (226)<60 GOTO 200
120 LET @ (226)=@ (226)-60,@ (227)=@ (227)+1;IF @ (226)>=60 GOTO 120
125 IF @ (227)<60 GOTO 200
130 LET @ (227)=@ (227)-60,@ (228)=@ (228)+1;IF@ (228)<24 GOTO 200
140 LET @ (228)=@ (228)-24,@ (229)=@ (229)+1;IF@ (229)<366 GOTO 200
150 LET @ (229)=1,@ (230)=@ (230)+1
200 FOR @ (225)=1TO6
202 PUT228,255-@ (2);PUT229,170;PUT229,42;PUT229,170
210 LET Q=0,K=GET(230)-48;IFK>=128LET Q=128
212 PUT229,160;PUT229,32;PUT229,160
220 LET K=K-Q;IFK>=64 GOTO 500

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230 IFF<8 GOTO 240
232 LET @ (231)=0; GOTO 500
240 IFF#0 GOTO 250
242 IF (@ (231)=0)*(F#1) GOTO 3000
244 IFF#1 GOTO 2900
246 GOTO 1000
250 IFK=14 GOTO 400
260 IFK<10 GOTO 270
262 LET @ (231)=0; GOTO 500
270 IFF=1 GOTO 330
280 IF @ (231)#0 GOTO 290
282 FOR I=1 TO 12; LET @ (204+I)=@ (12*F+I); NEXT I; LET @ (231)=1
290 LET @ (210)=@ (211); IF @ (211)=46 LET @ (210)=@ (212)
300 IF @ (211)=46 GOTO 310
302 LET @ (211)=@ (212); IF @ (212)=46 LET @ (211)=@ (213)
310 IF @ (212)=46 GOTO 320
312 LET @ (212)=@ (213)
320 LET @ (213)=K; GOTO 2900
330 LET X=R+L*(C-1); IFK#0 GOTO 340
332 LET @ (X+1)=0, @ (X+18)=0, @ (X+19)=0; GOSUB 354; LET R=R-Q; PUT 232, R
334 LET R=64, W=0; FOR I=1 TO 18; IF I=C GOTO 337
336 IFF<0 GOTO 338
338 LET P=P-Q, W=W+Q; GOTO 338
337 IF P>=Q LET P=P-Q
339 LET Q=Q/2; NEXT I; LET P=W; PUT 233, P; GOTO 1000
340 IF (K#1)*(K#2) GOTO 360
342 IF @ (X+1)#0 GOTO 350
344 LET @ (X+1)=1, @ (X+19)=0; GOSUB 354; LET R=R+Q; PUT 232, R; GOTO 1000
350 IF (@ (X+1)=4)+(@ (X+1)=7)=0 GOTO 3130
352 IF @ (X+1)=4 GOTO 356
354 LET @ (X+1)=@ (X+18), @ (X+18)=0; GOSUB 354; LET R=R+Q; PUT 232, R; GOTO 1
356
358 LET G=1; FOR I=1 TO C; LET Q=Q+Q; NEXT I; RETURN
359 LET @ (X+1)=@ (X+18), @ (X+18)=0; GOTO 3130
360 IFK#4 GOTO 364
362 LET @ (X+18)=4; GOTO 1000
364 IFF#7 GOTO 370
366 IF (@ (X+1)*2)+(@ (X+1)*2)+(@ (X+1)*16) GOTO 1000
368 LET @ (X+18)=@ (X+1), @ (X+1)=7; GOSUB 354; LET R=R-Q; PUT 232, R; GOTO 10
370
372 IFF#4 GOTO 380
374 INPUT "NEW WAIT TIME" V; LET V=60*V; GOTO 3130
380 IFF#0 GOTO 382
382 LET Q=230; GOTO 3002
390 IFF#9 GOTO 3130

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395 LET Q=292; GOTO 3002
400 LET @ (231)=0
410 IFF#2 GOTO 420
412 LET @ (B+L*(C-1)+F)=1000*@ (210)+100*@ (211)+10*@ (212)+@ (213)
414 GOTO 1000
420 LET @ (B+L*(C-1)+F)=100*@ (210)+@ (213)
430 IF (F#3)*(F#8)#0 GOTO 440
432 LET @ (B+L*(C-1)+F)=@ (B+L*(C-1)+F)+10*@ (211); GOTO 1000
440 LET @ (B+L*(C-1)+F)=@ (B+L*(C-1)+F)+10*@ (212); GOTO 1000
500 IF Q=0 GOTO 1000
510 IF K>=64 LET K=K-64
520 IF K<10 GOTO 3130
530 GOTO 530+10*(K-9)
540 K=GET (230)-48; IF K<192 GOTO 548
542 PUT 228,255-@ (2); PUT 229,193; PUT 229,65; PUT 229,193
544 K=GET (230)-48; IF K#64 GOTO 544
545 K=GET (230)-112; IF K=0 GOTO 545
546 K=K-128; LET F=K+1; GOTO 1000
548 LET F=1; GOTO 1000
550 IF @ (232)#0 GOTO 552
551 @ (232)=1; PRINT "TAPE READY"; GOSUB 8500; GOTO 3130
552 @ (232)=0; PRINT "OK TO REMOVE TAPE"; GOTO 3130
560 LET C=C+1; IF C>6 LET C=1
562 GOTO 1000
570 IF @ (232)#0 GOSUB 7000
572 GOTO 3130
580 GOTO 3130
590 LET F=F+1; IFF>16 LET F=1
1000 LET W=12*F, Q=B+L*(C-1), X=@ (Q+F), @ (W+1)=C; IF F=1 GOTO 1100
1010 IFF=2 GOTO 1200
1020 IF F=15 GOTO 1220
1022 IF F#16 GOTO 1030
1024 LET X=@ (Q+85); GOTO 1220
1030 IF (F=3)+(F=8)>0 GOTO 1300
1032 IFF#14 GOTO 1040
1034 LET @ (W+5)=32; IF @ (Q+1)=2 GOTO 1040
1036 LET @ (W+5)=45, X=@ (Q+17)
1040 LET @ (W+6)=X/100, X=X-100*@ (W+6), @ (W+8)=X/10, @ (W+9)=X-10*@ (W+8)
1050 GOTO 3000
1100 IF @ (Q+18)=4 GOTO 1170
1102 IF (X=2)+(X=8)+(X=9)+(X=10)+(X=20) GOTO 1112
1104 IF (X=16)+(X=17)+(X=18)+(X=19)+(X=22) GOTO 1120
1105 IF X=3 GOTO 1160

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1106 IFX=7 GOTO 1162
1107 IFX=8 GOTO 1164
1108 IFX=1 GOTO 1130
1109 IFX=4 GOTO 1140
1110 IFX=0 GOTO 1150
1111 GOTO 3130
1112 LET @ (18)=67,@ (19)=72,@ (20)=65,@ (21)=82,@ (22)=71,@ (23)=69; GOTO 30
00
1118 LET @ (18)=67,@ (19)=72,@ (20)=65,@ (21)=82,@ (22)=71,@ (23)=69; GOTO 30
00
1120 LET @ (18)=68,@ (19)=73,@ (20)=83,@ (21)=67,@ (22)=72,@ (23)=71; GOTO 30
00
1130 LET @ (18)=73,@ (19)=78,@ (20)=73,@ (21)=84,@ (22)=32,@ (23)=32; GOTO 30
00
1140 LET @ (18)=73,@ (19)=78,@ (20)=84,@ (21)=82,@ (22)=69,@ (23)=68; GOTO 30
00
1150 LET @ (18)=83,@ (19)=84,@ (20)=79,@ (21)=80,@ (22)=32,@ (23)=32; GOTO 30
00
1160 LET @ (18)=80,@ (19)=82,@ (20)=73,@ (21)=78,@ (22)=84,@ (23)=32; GOTO 30
00
1162 LET @ (18)=80,@ (19)=65,@ (20)=85,@ (21)=83,@ (22)=69,@ (23)=32; GOTO 30
00
1164 LET @ (18)=80,@ (19)=82,@ (20)=73,@ (21)=78,@ (22)=84,@ (23)=32; GOTO 30
00
1170 LET @ (18)=73,@ (19)=78,@ (20)=84,@ (21)=82,@ (22)=32,@ (23)=32; GOTO 30
00
1180 LET @ (280)=32,@ (281)=32,@ (282)=32,@ (283)=78,@ (284)=65,@ (285)=78
1182 @ (286)=79,@ (287)=32,@ (288)=78,@ (289)=65,@ (290)=78,@ (291)=79
1184 LET @ (292)=32,@ (293)=32,@ (294)=32,@ (295)=83,@ (296)=72,@ (297)=65
1186 @ (298)=90,@ (299)=66,@ (300)=79,@ (301)=84,@ (302)=32,@ (303)=63
1187 @ (304)=32
1188 RETURN
1200 LET @ (30)=X/1000,X=X-1000*@ (30),@ (31)=X/100,X=X-100*@ (31)
1210 LET @ (32)=X/10,@ (33)=X-10*@ (32); GOTO 3000
1220 LET @ (W+6)=X/1000,X=X-1000*@ (W+6),@ (W+7)=X/100,X=X-100*@ (W+7)
1230 LET @ (W+9)=X/10,@ (W+10)=X-10*@ (W+9)
1240 GOTO 3000
1300 LET @ (W+6)=X/100,X=X-100*@ (W+6),@ (W+7)=X/10
1310 LET @ (W+9)=X-10*@ (W+7); GOTO 3000
2900 LET Q=204; GOTO 3002
3000 LET Q=12*F
3002 FOR I=12 TO 15.-1
3010 PUT228,255-@ (I)
3012 LET W=0; IF @ (Q+1)<32 LET W=48
3020 PUT229,@ (Q+1)+W+128
3030 PUT229,@ (Q+1)+W
3040 PUT229,@ (Q+1)+W+128
3060 NEXT I
3130 LET J=B+L*(@ (225)-1),X=Z+5*(@ (225)-1)
3132 LET S=@ (J+1); IF (S=0)+(S=3)+(S=4)+(S=7) GOTO 4900
3140 IF @ (J+19)>0 @ (J+19)=@ (J+19)-T
3150 IF @ (J+19)>0 GOTO 4900
3160 IFS=17 GOTO 3930

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3162 IF(S#1)*(S#2)*(S#16) GOTO 4000
3170 LET @ (J+9)=(V.(X+1)+50)/100,@ (J+10)=(V.(X+2)+50)/100
3180 LET @ (J+11)=(V.(X+3)+50)/100,@ (J+17)=(AMP(X+4)+50)/100
3190 LET @ (J+14)=(AMP(X+5)+50)/100
3192 GOTO 3230
3195 IF @ (225)<=3 GOTO 3200
3197 IF @ (J+8)#@ (J+3) GOTO 3230
3198 GOTO 3210
3200 IF @ (J+10)#0 GOTO 3230
3210 LET @ (J+18)=@ (J+1),@ (J+1)=7;GOSUB 4510;LET R=R-Q;PUT232,R; GOTO 49
00
3230 IFS#1GOTO3400
3232 IF @ (225)>=5 GOTO 3240
3234 IF @ (225)=2 GOTO 3240
3236 IF @ (225)=3 GOTO 3240
3238 IF @ (J+11)>@ (J+7) GOTO 4900
3239 GOTO 3242
3240 IF@ (J+6)<@ (J+10)GOTO4900
3242 GOSUB 4510;LET R=R-Q;PUT232,R
3250 PRINT "CELL ",#1,@ (225)," READY @ ",@ (230)," / ",@ (229)," ",
3252 PRINT #1,@ (228)," : ",@ (227)," : ",@ (226)
3260 LET @ (J+8)=0,@ (J+1)=20,@ (J+19)=60
3262 NEXT @ (225); GOTO 100
3400 IFS#16GOTO3500
3410 @ (J+51)=@ (J+51)+T*@ (J+17)
3420 IF @ (J+51)<0GOTO3540
3430 LET Q=@ (J+51)/3600,@ (J+85)=@ (J+85)+Q,@ (J+51)=@ (J+51)-3600*Q
3500 IFS=2LET @ (J+50)=@ (J+50)+T*@ (J+14)
3510 IF@ (J+50)<=0GOTO3540
3520 LET Q=@ (J+50)/3600,@ (J+15)=@ (J+15)+Q,@ (J+50)=@ (J+50)-3600*Q
3540 @ (J+52)=@ (J+52)+T
3550 IF@ (J+52)<=0GOTO3600
3552 GOTO 3560
3553 REM LINE 3555 SPEEDS TA CLOCK UP 10X
3554 REM LINE 3552 BYPASSES 10X CLOCK
3555 @ (J+52)=@ (J+52)-36;@ (J+8)=@ (J+8)+1;GOTO 3550
3560 LET D=@ (J+52)/360,@ (J+8)=@ (J+8)+D,@ (J+52)=@ (J+52)-360*D
3600 IFS#2GOTO3700
3610 IF @ (J+8)>@ (J+3)/2 GOTO 3612
3611 IF @ (J+9)<@ (J+4) GOTO 4900
3612 IF@ (J+20)#0GOTO3640
3620 @ (J+20)=@ (J+8);@ (J+21)=@ (J+15);@ (J+22)=@ (J+9);@ (J+12)=@ (J+9)
3630 LET @ (J+23)=@ (J+10),@ (J+24)=@ (J+11),@ (J+77)=@ (J+14),@ (J+71)=@ (J+15)
3640 IF@ (J+9)>@ (J+4) GOTO 3650

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3642 IF @ (J+8) < @ (J+3) GOTO 4900
3650 @ (J+25) = @ (J+8); @ (J+26) = @ (J+15); @ (J+27) = @ (J+9)
3660 LET @ (J+28) = @ (J+10), @ (J+29) = @ (J+11), @ (J+78) = @ (J+14), @ (J+86) = @ (J+15)
3665 IF @ (J+25) < 1 GOTO 3210
3670 LET @ (J+1) = 10, @ (J+8) = 0, @ (J+52) = 0, @ (J+19) = V; GOSUB 4510; LET R = R - Q; PUT
232, R
3672 GOTO 4900
3700 IFS#16GOTO4000
3702 IF @ (J+8) < 5 GOTO 3708
3703 IF @ (J+40) # 0 GOTO 3708
3704 LET @ (J+40) = @ (J+8), @ (J+41) = @ (J+85), @ (J+42) = @ (J+9), @ (J+73) = @ (J+85)
3706 LET @ (J+43) = @ (J+10), @ (J+44) = @ (J+11), @ (J+82) = @ (J+17)
3707 LET @ (J+13) = @ (J+9)
3708 IF @ (J+9) > @ (J+5) GOTO 3714
3709 IF @ (J+45) # 0 GOTO 3714
3710 LET @ (J+45) = @ (J+8), @ (J+46) = @ (J+85), @ (J+47) = @ (J+9)
3712 LET @ (J+48) = @ (J+10), @ (J+49) = @ (J+11), @ (J+76) = @ (J+85), @ (J+16) = @ (J+85)
3714 IF @ (J+30) # 0 GOTO 3738
3716 IF @ (J+10) > @ (J+6) GOTO 3739
3720 LET @ (J+30) = @ (J+8), @ (J+31) = @ (J+85), @ (J+32) = @ (J+9), @ (J+74) = @ (J+85)
3730 LET @ (J+33) = @ (J+10), @ (J+34) = @ (J+11), @ (J+80) = @ (J+17)
3732 IF ((@ (225) = 5) + (@ (225) = 6)) = 0 GOTO 3738
3733 IF @ (J+9) > @ (J+5) GOTO 3739
3734 IF @ (J+35) # 0 GOTO 3830
3736 GOTO 3750
3738 IF (@ (J+30) # 0) * (@ (J+35) # 0) # 0 GOTO 3830
3739 IF @ (J+35) # 0 GOTO 4900
3740 IF @ (J+11) > @ (J+7) GOTO 4900
3750 LET @ (J+35) = @ (J+8), @ (J+36) = @ (J+85), @ (J+37) = @ (J+9), @ (J+75) = @ (J+85)
3760 LET @ (J+38) = @ (J+10), @ (J+39) = @ (J+11), @ (J+81) = @ (J+17)
3765 IF @ (J+9) > @ (J+5) GOTO 4900
3770 IF @ (225) = 4 GOTO 3830
3772 IF @ (225) = 1 GOTO 3830
3780 IF @ (J+30) = 0 GOTO 4900
3830 LET S = 17, @ (J+1) = 17, @ (J+19) = V
3832 GOSUB 4510; LET R = R - Q; PUT 232, R
3840 LET @ (J+8) = 0, @ (J+51) = 0; GOTO 4900
3930 LET @ (J+67) = (V * (X+1) + 50) / 100, @ (J+68) = (V * (X+2) + 50) / 100
3932 LET @ (J+69) = (V * (X+3) + 50) / 100
3934 LET X = @ (J+41), Q = @ (J+40); GOSUB 5000; LET X = G, Q = 60; GOSUB 5000
3936 LET @ (J+87) = G / 10, G = G - 10 * @ (J+87); IF G >= 5 LET @ (J+87) = @ (J+87) + 1
3938 LET X = @ (J+26), Q = @ (J+25); GOSUB 5000; LET X = G, Q = 600; GOSUB 5000
3940 LET @ (J+54) = G; IF D / 100 > 5 LET @ (J+54) = @ (J+54) + 1
3944 LET X = @ (J+42), Q = @ (J+22); GOSUB 5000

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3946 LET @ (J+56)=10*G+(D+50)/100
3950 LET X=@ (J+76),Q=@ (J+26);GOSUB 5000
3952 LET @ (J+57)=10*G+(D+50)/100
3954 LET X=@ (J+56)/10;IF@ (J+56)-10*X<=5 GOTO 3958
3956 LET X=X+1
3958 LET Q=@ (J+57)/10;IF@ (J+57)-10*Q<=5 GOTO 3962
3960 LET Q=Q+1
3962 LET X=X*Q,Q=100;GOSUB 5000
3972 LET @ (J+58)=G/10,G=G-10*@ (J+58);IFG>5LET @ (J+58)=@ (J+58)+1
3974 LET X=@ (J+75),Q=@ (J+26);GOSUB 5000
3976 LET @ (J+59)=10*G+(D+50)/100
3980 LET X=@ (J+74),Q=@ (J+26);GOSUB 5000
3982 LET @ (J+60)=10*G+(D+50)/100
3994 LET @ (J+30)=0,@ (J+35)=0,@ (J+40)=0,@ (J+45)=0,@ (J+50)=0,@ (J+51)=0
3995 LET @ (J+2)=@ (J+2)+1
3996 GOTO 4050
4000 IFS<8GOTO4900
4020 IFS#10 GOTO 4100
4030 IFS=22 GOTO 4050
4034 @ (J+62)=(V.(X+1)+50)/100;@ (J+63)=(V.(X+2)+50)/100
4036 LET @ (J+64)=(V.(X+3)+50)/100
4040 LET @ (J+72)=@ (J+15),@ (J+1)=8,@ (J+19)=30
4042 GOTO 4900
4050 LET @ (J+84)=@ (J+16),@ (J+1)=18,@ (J+19)=30
4056 GOSUB 8700;IF@ (J+1)=3 GOTO 4900
4058 IF@ (J+18)#4 GOTO 4900
4060 @ (J+18)=@ (J+1),@ (J+1)=4; GOTO 4900
4100 IFS#8GOTO4200
4110 GOSUB4510
4120 LET P=P+Q;PUT233,P
4130 LET @ (J+1)=9,@ (J+19)=30
4132 GOTO 4900
4200 IFS#18GOTO4300
4210 GOSUB 4510;LET P=P-Q;PUT233,P
4220 LET @ (J+1)=19,@ (J+19)=30
4222 GOTO 4900
4300 IFS#9GOTO4400
4310 GOSUB 4510;LET R=R+Q;PUT232,R;REM CLOSE CURRENT RELAY
4320 LET @ (J+1)=16,@ (J+52)=0
4322 GOTO 4900
4400 IFS#19 GOTO 4430
4410 GOSUB 4510;LET R=R+Q;PUT232,R;REM CLOSE CURRENT RELAY
4420 LET @ (J+1)=2;FOR I=8TOL;LET @ (J+1)=0;NEXT I

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4422 GOTO 4900
4430 IFS#20 GOTO 4900
4440 LET @ (J+84)=@ (J+16), @ (J+1)=18, @ (J+19)=30; GOTO 4900
4500 GOTO 4900
4510 LET Q=1; FOR I=1 TO @ (225); LET Q=Q+Q; NEXT I; RETURN
4900 NEXT @ (225)
4910 GOTO 100
5000 IF Q#0 GOTO 5010
5002 LET G=99, D=999; RETURN
5010 LET W=X/Q, X=10*(X-W*Q), G=W, W=X/Q, X=10*(X-W*Q), G=10*G+W, W=X/Q
5020 LET X=10*(X-W*Q), G=10*G+W
5030 LET W=X/Q, X=10*(X-W*Q), D=W, W=X/Q, X=10*(X-W*Q), D=10*D+W, W=X/Q
5040 LET X=10*(X-W*Q), D=10*D+W
5060 RETURN
7000 G=B+L*(C-1)
7002 PRINT
7004 PRINT ; PRINT #1, @ (230), "/", @ (229), " ", @ (228), ":", @ (227), ":", @ (226)
, " ",
7006 PRINT #2, "CELL #", ((G-B)/L)+1, " CYCLE #", @ (G+2)-1
7011 PRINT " HRS AMP/HRS CELL ZINC IRON I"
7012 PRINT "CHARGE"; IF @ (G+1)#2 GOTO 7074
7014 PRINT "CURRENT ",
7020 LET Q=@ (G+8); GOSUB 8600
7030 LET Q=@ (G+15); GOSUB 8610
7040 LET Q=@ (G+9); GOSUB 8610
7050 LET Q=@ (G+10); GOSUB 8610
7060 LET Q=@ (G+11); GOSUB 8610
7070 LET Q=@ (G+14); GOSUB 8610
7074 PRINT
7080 PRINT "30 MIN ", ; LET Q=@ (G+20); GOSUB 8600
7090 LET Q=@ (G+71); GOSUB 8610
7100 LET Q=@ (G+22); GOSUB 8610
7110 LET Q=@ (G+23); GOSUB 8610
7120 LET Q=@ (G+24); GOSUB 8610
7130 LET Q=@ (G+77); GOSUB 8610
7132 PRINT ; PRINT "END ", ; LET Q=@ (G+25); GOSUB 8600; LET Q=@ (G+26); GOS
UB 8610
7134 LET Q=@ (G+27); GOSUB 8610; LET Q=@ (G+28); GOSUB 8610
7136 LET Q=@ (G+29); GOSUB 8610
7138 LET Q=@ (G+78); GOSUB 8610
7140 PRINT ; PRINT "OCV ", ; LET Q=0; GOSUB 8600
7150 LET Q=@ (G+72); GOSUB 8610
7160 LET Q=@ (G+62); GOSUB 8610
7170 LET Q=@ (G+63); GOSUB 8610
7180 LET Q=@ (G+64); GOSUB 8610

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7190 LET Q=@(G+79);GOSUB 8610;PRINT
7200 PRINT "DISCHARGE";IF@(G+1)#16 GOTO 7290
7210 PRINT "CURRENT ",
7230 LET Q=@(G+8);GOSUB 8600
7240 LET Q=@(G+85);GOSUB 8610
7250 LET Q=@(G+9);GOSUB 8610
7260 LET Q=@(G+10);GOSUB 8610
7270 LET Q=@(G+11);GOSUB 8610
7280 LET Q=@(G+17);GOSUB 8610
7290 PRINT
7300 PRINT "30 MIN ";LET Q=@(G+40);GOSUB 8600
7310 LET Q=@(G+41);GOSUB 8610
7320 LET Q=@(G+42);GOSUB 8610
7330 LET Q=@(G+43);GOSUB 8610
7340 LET Q=@(G+44);GOSUB 8610
7350 LET Q=@(G+82);GOSUB 8610;PRINT
7360 PRINT "ZINC LIM";LET Q=@(G+30);GOSUB 8600
7370 LET Q=@(G+74);GOSUB 8610
7380 LET Q=@(G+32);GOSUB 8610
7390 LET Q=@(G+33);GOSUB 8610
7400 LET Q=@(G+34);GOSUB 8610
7410 LET Q=@(G+80);GOSUB 8610;PRINT
7420 PRINT "IRON LIM";LET Q=@(G+35);GOSUB 8600
7430 LET Q=@(G+75);GOSUB 8610
7440 LET Q=@(G+37);GOSUB 8610
7450 LET Q=@(G+38);GOSUB 8610
7460 LET Q=@(G+39);GOSUB 8610
7470 LET Q=@(G+81);GOSUB 8610;PRINT
7480 PRINT "CELL LIM";LET Q=@(G+45);GOSUB 8600
7490 LET Q=@(G+76);GOSUB 8610
7500 LET Q=@(G+47);GOSUB 8610
7510 LET Q=@(G+48);GOSUB 8610
7520 LET Q=@(G+49);GOSUB 8610
7530 LET Q=@(G+83);GOSUB 8610;PRINT
7540 PRINT "DCV ";LET Q=@(G+6);GOSUB 8600
7550 LET Q=0;GOSUB 8610
7560 LET Q=@(G+67);GOSUB 8610
7570 LET Q=@(G+68);GOSUB 8610
7580 LET Q=@(G+69);GOSUB 8610
7590 LET Q=0;GOSUB 8610;PRINT ;RETURN
8500 IF@(233)=0RETURN
8510 LET @(234)=J;FOR J=225 TO 230;LET @(J+10)=@(J);NEXT J
8520 FOR @(217)=1 TO @(233);LET Q=241+6*(@(217)-1),J=B+L*(@(Q)-1),@(225)=Q
(Q)

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8530 LET Q(226)=Q(Q+1),Q(227)=Q(Q+2),Q(228)=Q(Q+3),Q(229)=Q(Q+4)
8540 LET Q(23)=Q(Q+5),Q(J+1)=Q(J+18),Q(J+18)=0;GOSUB 8700;NEXT Q(17)
8550 FOR J=225 TO 230;LET Q(J)=Q(J+10);NEXT J
8560 LET J=Q(234),Q(233)=0;RETURN
8600 LET W=Q/10,Q=Q-10*W;PRINT #3,W,".",#1,Q;RETURN
8610 W=Q/100;Q=Q-100*W;PRINT #5,W,".",
8612 IFQ=0 GOTO 8618
8614 IFQ>=10 GOTO 8616
8616 PRINT "0",#1,Q;RETURN
8618 PRINT #2,Q;RETURN
8619 PRINT "00",;RETURN
8700 IF Q(232)#0 GOTO 8720
8702 IF Q(233)>=6RETURN
8704 LET Q(233)=Q(233)+1,Q=241+6*(Q(233)-1)
8706 LET Q(Q)=Q(225),Q(Q+1)=Q(226),Q(Q+2)=Q(227),Q(Q+3)=Q(228)
8708 LET Q(Q+4)=Q(229),Q(Q+5)=Q(230)
8710 LET Q(J+18)=Q(J+1),Q(J+1)=3;RETURN
8720 PRINT ;PRINT #1,Q(230),"/",Q(229),",",Q(228),",",Q(227),",",Q(226)
8721 PRINT #1," ",Q(225),",",Q(224),
8722 LET Q=Q(J+25);GOSUB 8950
8723 LET Q=Q(J+54);GOSUB 8950
8724 LET Q=Q(J+87);GOSUB 8950
8730 LET Q=Q(J+56);GOSUB 8950
8740 LET Q=Q(J+57);GOSUB 8950
8750 LET Q=Q(J+58);GOSUB 8950
8760 LET Q=Q(J+60);GOSUB 8950
8770 LET Q=Q(J+59);GOSUB 8950;PRINT
8775 LET Q=Q(J+23);GOSUB 8900;LET Q=Q(J+24);GOSUB 8900
8780 LET Q=Q(J+27);GOSUB 8900;LET Q=Q(J+62);GOSUB 8900
8790 LET Q=Q(J+43);GOSUB 8900;LET Q=Q(J+44);GOSUB 8900
8800 LET Q=Q(J+48);GOSUB 8900;LET Q=Q(J+49);GOSUB 8900
8810 LET Q=Q(J+33);GOSUB 8900;LET Q=Q(J+34);GOSUB 8900
8820 LET Q=Q(J+38);GOSUB 8900;LET Q=Q(J+39);GOSUB 8900
8830 LET Q=Q(J+67);GOSUB 8900
8880 PRINT ;RETURN
8885 W=Q/100;Q=Q-100*W
8890 PRINT #0,W,GOTO 8920
8900 W=Q/100;Q=Q-100*W
8905 IF Q>=0 GOTO 8910
8907 LET Q=Q,W=0
8910 PRINT #3,W,".",
8920 IFQ=0 GOTO 8940
8930 IFQ>=10 GOTO 8938
8932 PRINT "0",#1,Q;RETURN
8934 PRINT #2,Q;RETURN
8940 PRINT "00",;RETURN
8950 LET V=Q/10,Q=Q-10*W
8960 PRINT #0,W,".",#1,Q;RETURN

```


Appendix C

ASSEMBLY LANGUAGE INTERPRETER FOR TINY BASIC

The following listing for an 8080 microcomputer is in assembly language compiled by Lincoln Semiconductor, and is included for report completeness. For an explanation of this listing, the reader is encouraged to consult a microcomputer expert. Key additions to Dr. Li-Chen Wang's Palo Alto Tiny Basic Interpreter with their address are: Open Current Relays (00C0); PEEK (01ED); POKE (01A8); PUT (01B6); GET (01F4); VOLT (01CE); and AMP (01D4).

0000 0000 F3 0001 3ECF 0003 D3ED 0005 C3BA00 0008 E3 0009 EF 000A BE 000B C36800 000E 3E0D 0010 F5 0011 3A003C 0014 B7 0015 C36807 0018 CD9F04 001B E5 001C C3B04 001F 57 0020 7C 0021 BA 0022 C0 0023 7D 0024 BB 0025 C9 0026 41 0027 4E 0028 1A 0029 FE20 002B C0 002C 13 002D C32800	0001 PON 0005 DI 0010 MVI A,BAUD+CHRLN+STOPS 0015 OUT COMMND 0020 JMP STO 0025 TSTC XTHL 0030 RST 5 0035 CMP M 0036 JMP TC1 0040 CRLF MVI A,ODH 0045 OUTC PUSH AF 0050 LDA CCSW 0055 ORA A 0060 JMP OC2 0065 EXPR CALL EXPR2 0070 PUSH H 0075 JMP EXPR1 0080 DB 'W' 0085 COMP MOV A,H 0090 CMP D 0095 RNZ 0100 MOV A,L 0105 CMP E 0110 RET 0115 DB 'A' 0116 DB 'N' 0120 SS1 LDAX D IGNBLK 0125 CPI ' ' 0130 RNZ 0135 INX D 0140 JMP SS1	0030 F1 0006 0031 CDD805 0034 C3EE05 0037 47 0038 EF 0039 D640 003B D8 003C C25800 003F 13 0040 CD4505 0043 29 0044 DA9F00 0047 D5 0048 EB 0049 CD8705 004C E7 004D DA1E06 0050 21007F 0053 CDAA05 0056 D1 0057 C9 0058 FE1B 005A 3F 005B D8 005C 13 005D 21007F 0060 07 0061 85 0062 6F 0063 3E00	0145 FINISH POP AF 0146 AF EQU P 0150 CALL FIN 0155 JMP QWHAT 0160 DB 'G' 0165 TSTV RST 5 0170 SUI 'e' 0175 RC 0180 JNZ TV1 0185 INX D 0190 CALL PARN 0195 DAD H 0200 JC QHOW 0205 PUSH D 0210 XCHG 0215 CALL SIZE 0220 RST 4 0225 JC ASCRRY 0230 LXI H,VARBGN 0235 CALL SUBDE 0240 POP D 0245 RET 0250 TV1 CPI 1BH 025A CMC 0260 RC 0265 INX D 0270 LXI H,VARBGN 0275 RLC 0280 ADD L 0285 MOV L,A 0290 MVI A.0
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0065 8C	0295 ADC H	00AA 0D	0669 DB 0DH
0066 67	0300 MOV H,A	00AB 4F	0670 OK DB 'O'
0067 C9	0305 RET	00AC 4B	0671 DB 'K'
0068 23	0310 TC1 INX H	00AD 0D	0672 DB 0DH
0069 CA7300	0315 JZ TC2	00AE 57	0673 WHAT DB 'W'
006C C5	0320 PUSH B	00AF 48	0674 DB 'H'
006D 4E	0330 MOV C,M	00B0 41	0676 DB 'A'
006E 0600	0335 MVI B,0	00B1 5A	0678 DB 'T'
0070 09	0340 DAD B	00B2 3F	0680 DB '?'
0071 C1	0345 POP B	00B3 0D	0682 DB 0DH
0072 1B	0350 DCX D	00B4 53	0684 SORRY DB 'S'
0073 13	0355 TC2 INX D	00B5 4F	0686 DB 'O'
0074 23	0360 INX H	00B6 52	0688 DB 'R'
0075 E3	0365 XTHL	00B7 52	0690 DB 'R'
0076 C9	0370 RET	00B8 59	0692 DB 'Y'
0077 210000	0380 TSTNUM LXI H,0	00B9 0D	0694 DB 0DH
007A 44	0390 MOV B,H	00BA 3E25	0696 STO MVI A,RIS+RXE+TXEN
007B EF	0400 RST 5	00BC D3ED	0698 OUT COMMND
007C FE30	0410 TNL CPI 'O'	00BE 3E89	0700 MVI A,137
007E D8	0420 RC	00C0 D3EB	0702 OUT 235 OPEN CURRENT RELAYS
007F FE3A	0430 CPI 3AH	00C2 21153C	0703 RSTRT LXI H,XTBGN
0081 D0	0440 RNC	00C5 22133C	0704 SHLD TXTUNF
0082 3EF0	0450 MVI A,0F0H	00C8 3EFF	0705 MVI A,0FFH
0084 A4	0460 ANA H	00CA 32003C	0706 STA OCSW
0085 C29F00	0470 JNZ QHOW	00CD 310080	0707 START LXI S,STACK
0088 04	0480 INR B	00D0 210000	0708 LXI H,0
0089 C5	0490 PUSH B	00D3 22113C	0709 SHLD RANPNT
008A 44	0500 MOV B,H	00D6 11AB00	0710 LXI D,OK
008B 4D	0505 MOV C,L	00D9 97	0715 SUB A
008C 29	0510 DAD H	00DA CD8A06	0720 CALL PRSTG
008D 29	0520 DAD H	00DD 21E400	0725 LXI H,ST2+1
008E 09	0530 DAD B	00E0 22013C	0730 SHLD CURRNT
008F 29	0540 DAD H	00E3 210000	0735 ST2 LXI H,0
0090 1A	0550 LDAX D	00E6 22073C	0740 SHLD LOPVAR
0091 13	0560 INX D	00E9 22033C	0745 SHLD STKGOS
0092 E60F	0570 ANI 0FH	00EC 3E3E	0750 ST3 MVI A,'>'
0094 85	0580 ADD L	00EE CD2406	0755 CALL GETLN
0095 6F	0590 MOV L,A	00F1 D5	0760 PUSH D
0096 3E00	0600 MVI A,0	00F2 11377F	0765 LXI D,BUFFER
0098 8C	0610 ADC H	00F5 CD7700	0770 CALL TSTNUM
0099 67	0620 MOV H,A	00F8 EF	0780 RST 5
009A C1	0630 POP B	00F9 7C	0790 MOV A,H
009B 1A	0635 LDAX D	00FA B5	0795 ORA L
009C F27C00	0640 JP TNL	00FB C1	0800 POP B
009F D5	0645 QHOW PUSH D	00FC CA3B02	0805 JZ DIRECT
00A0 11A600	0650 AHOW LXI D,HOW	00FF 1B	0810 DCX D
00A3 C3F205	0660 JMP ERROR	0100 7C	0815 MOV A,H
00A6 4B	0662 HOW DB 'H'	0101 12	0820 STAX D
00A7 4F	0664 DB 'O'	0102 1B	0825 DCX D
00A8 57	0666 DB 'W'	0103 7D	0830 MOV A,L
00A9 3F	0668 DB '?'	0104 12	0835 STAX D

0105 C5	0840	PUSH B	0156 45	1056	DB 'E'
0106 D5	0845	PUSH D	0157 57	1058	DB 'W'
0107 79	0850	MOV A,C	0158 80	1059	DB 80H
0108 93	0855	SUB E	0159 6C02	1060	DW NEW
0109 F5	0860	PUSH AF	015B	1070	TAB2 DS 0 DIRECT/STATEMENT
010A CD6206	0870	CALL FNDLN	015B 4E	1072	DB 'N'
010D D5	0875	PUSH D	015C 45	1074	DB 'E'
010E C22101	0880	JNZ ST4	015D 58	1076	DB 'X'
0111 D5	0885	PUSH D	015E 54	1078	DB 'T'
0112 CD7E06	0890	CALL FNDNXT	015F 80	1079	DB 80H
0115 C1	0895	POP B	0160 9103	1080	DW NEXT
0116 2A133C	0900	LHLD TXTUNF	0162 4C	1082	DB 'L'
0119 CD0B07	0905	CALL MVUP	0163 45	1084	DB 'E'
011C 60	0910	MOV H,B	0164 54	1086	DB 'T'
011D 69	0916	MOV L,C	0165 80	1087	DB 80H
011E 22133C	0920	SHLD TXTUNF	0166 5104	1088	DW LET
0121 C1	0925	ST4 POP B	0168 49	1090	DB 'I'
0122 2A133C	0930	LHLD TXTUNF	0169 46	1092	DB 'F'
0125 F1	0935	POP AF	016A 80	1093	DB 80H
0126 E5	0940	PUSH H	016B E203	1094	DW IF
0127 FE03	0945	CPI 3	016D 47	1096	DB 'G'
0129 CACD00	0950	JZ START	016E 4F	1098	DB 'O'
012C 85	0955	ADD L	016F 54	1100	DB 'T'
012D 6F	0960	MOV L,A	0170 4F	1102	DB 'O'
012E 3E00	0965	MVI A,0	0171 80	1103	DB 80H
0130 8C	0970	ADC H	0172 9A02	1104	DW COTO
0131 67	0975	MOV H,A	0174 47	1106	DB 'G'
0132 11007F	0980	LXI D,TXTEND	0175 4F	1108	DB 'O'
0135 E7	0985	RST 4	0176 53	1110	DB 'S'
0136 D21D06	0990	JNC QSCRRY	0177 55	1112	DB 'U'
0139 22133C	0995	SHLD TXTUNF	0178 42	1114	DB 'B'
013C D1	1000	POP D	0179 80	1115	DB 80H
013D CD1407	1005	CALL MVDOWN	017A F902	1116	DW CORUS
0140 D1	1010	POP D	017C 52	1118	DB 'R'
0141 E1	1015	POP H	017D 45	1120	DB 'E'
0142 CD0B07	1020	CALL MVUP	017E 54	1122	DB 'T'
0145 C3EC00	1025	JMP ST3	017F 55	1124	DB 'U'
0148	1030	*TABLES * DIRECT * & EXEC	0180 52	1126	DB 'R'
0148	1035	TAB1 DS 0 DIRECT COMMANDS	0181 4E	1128	DB 'N'
0148 4C	1036	DB 'L'	0182 80	1129	DB 80H
0149 49	1038	DB 'I'	0183 1903	1130	DW RETURN
014A 53	1040	DB 'S'	0185 52	1132	DB 'R'
014B 54	1042	DB 'T'	0186 45	1134	DB 'E'
014C 80	1043	DB 80H	0187 4D	1136	DB 'M'
014D A902	1044	DW LIST	0188 80	1137	DB 80H
014F 52	1046	DB 'R'	0189 DE03	1138	DW REM
0150 55	1048	DB 'U'	018B 46	1140	DB 'F'
0151 4E	1050	DB 'N'	018C 4F	1142	DB 'O'
0152 80	1051	DB 80H	018D 52	1144	DB 'R'
0153 7B02	1052	DW RUN	018E 80	1145	DB 80H
0155 4E	1054	DB 'N'	018F 3203	1146	DW FOR

0191 49	1148 DB 'I'	01DA 52	1221 DB 'R'
0192 4E	1150 DB 'N'	01DB 4E	1222 DB 'N'
0193 50	1152 DB 'P'	01DC 44	1223 DB 'D'
0194 55	1154 DB 'U'	01DD 80	1224 DB 80H
0195 54	1156 DB 'T'	01DE 5005	1225 DW RND
0196 80	1157 DB 80H	01E0 41	1226 DB 'A'
0197 FB03	1158 DW INPUT	01E1 42	1227 DB 'B'
0199 50	1160 DB 'P'	01E2 53	1228 DB 'S'
019A 52	1162 DB 'R'	01E3 80	1229 DB 80H
019B 49	1164 DB 'I'	01E4 7B05	1230 DW ABS
019C 4E	1166 DB 'N'	01E6 53	1231 DB 'S'
019D 54	1168 DB 'T'	01E7 49	1232 DB 'I'
019E 80	1169 DB 80H	01E8 5A	1233 DB 'Z'
019F C102	1170 DW PRINT	01E9 45	1234 DB 'E'
01A1 43	1171 DB 'C'	01EA 80	1235 DB 80H
01A2 41	1172 DB 'A'	01EB 8705	1236 DW SIZE
01A3 4C	1173 DB 'L'	01ED 50	1238 DB 'P'
01A4 4C	1174 DB 'L'	01EE 45	1239 DB 'E'
01A5 80	1175 DB 80H	01EF 45	1240 DB 'E'
01A6 A807	1176 DW CALL	01F0 4B	1241 DB 'K'
01A8 50	1177 DB 'P'	01F1 80	1242 DB 80H
01A9 4F	1178 DB 'O'	01F2 BE07	1243 DW PEEK
01AA 4B	1179 DB 'K'	01F4 47	1244 DB 'G'
01AB 45	1180 DB 'E'	01F5 45	1245 DB 'E'
01AC 80	1181 DB 80H	01F6 54	1246 DB 'T'
01AD C507	1182 DW POKE	01F7 80	1247 DB 80H
01AF 53	1183 DB 'S'	01F8 D407	1248 DW GET
01B0 54	1184 DB 'T'	01FA 80	1254 DB 80H
01B1 4F	1185 DB 'O'	01FB 3605	1255 DW XP40
01B2 50	1186 DB 'P'	01FD 54	1256 TAB5 DB 'T'
01B3 80	1187 DB 80H	01FE 4F	1257 DB 'O'
01B4 7502	1188 DW STOP	01FF 80	1258 DB 80H
01B6 50	1189 DB 'P'	0200 4203	1259 DW FR1
01B7 55	1190 DB 'U'	0202 80	1260 DB 80H
01B8 54	1192 DB 'T'	0203 EE05	1261 DW QWHAT
01B9 80	1194 DB 80H	0205 53	1262 TAB6 DB 'S'
01BA EB07	1196 DW PUT	0206 54	1264 DB 'T'
01BC 80	1203 DB 80H	0207 45	1266 DB 'E'
01BD 4B04	1204 DW DEF LT	0208 50	1268 DB 'P'
01BF	1206 DS 14	0209 80	1269 DB 80H
01CD 56	1210 TAB4 DB 'V'	020A 4C03	1270 DW FR2
01CE 4F	1211 DB 'O'	020C 80	1271 DB 80H
01CF 4C	1212 DB 'L'	020D 5003	1272 DW FR3
01D0 54	1213 DB 'T'	020F	1274 TAB8 DS 0
01D1 80	1214 DB 80H	020F 3E	1276 DB '>'
01D2 1408	1215 DW VOLT	0210 3D	1278 DB '='
01D4 41	1216 DB 'A'	0211 80	1279 DB 80H
01D5 4D	1217 DB 'M'	0212 6104	1280 DW XP11
01D6 50	1218 DB 'P'	0214 3D	1282 DB '='
01D7 80	1219 DB 80H	0215 3E	1284 DB '>'
01D8 0A08	1220 DW AMP	0216 80	1285 DB 80H

0217 6104	1286	DW XP11
0219 23	1288	DB '4'
021A 80	1289	DB 80H
021B 6704	1290	DW XP12
021D 3E	1292	DB '>'
021E 80	1293	DB 80H
021F 6D04	1294	DW XP13
0221 80	1295	DB '4'
0222 80	1297	DB 80H
0223 7C04	1298	DW XP15
0225 3C	1300	DB '<'
0226 3D	1302	DB '='
0227 80	1303	DB 80H
0228 7404	1304	DW XP14
022A 3D	1306	DB '='
022B 3C	1308	DB '<'
022C 80	1309	DB 80H
022D 7404	1310	DW XP14
022F 3C	1312	DB '<'
0230 80	1313	DB 80H
0231 8204	1314	DW XP16
0233 3C	1316	DB '<'
0234 3E	1318	DB '>'
0235 80	1319	DB 80H
0236 6701	1320	DW XP12
0238 80	1321	DB 80H
0239 8804	1322	DW XP17
023B 214701	1330	DIRECT EX1 H,TAB1-1 *** DIRECT ***
023E	1335	EXEC DS 0 *** EXEC ***
023E EF	1340	EX0 RST 5
023F DS	1345	PUSH D
0240 1A	1350	EX1 LDAX D
0241 13	1355	INX D
0242 FE2E	1360	CPI '.'
0244 CA5E02	1365	JZ EX3
0247 23	1370	INX H
0248 BE	1375	CMP M
0249 CA4002	1380	JZ EX1
024C 3E7F	1385	MVI A,7FH
024E 1B	1390	DCX D
024F BE	1395	CMP M
0250 DA6502	1400	JC EX5
0253 23	1405	EX2 INX H
0254 BE	1410	CMP M
0255 D25302	1415	JNC EX2
0258 23	1420	INX H
0259 23	1422	INX H FOR EXTRA BYTE
025A D1	1425	POP D
025B C33E02	1430	JMP EX0
025E 3E7F	1435	EX3 MVI A,7FH
0260 23	1440	EX4 INX H

0251 B7	1445 CMP H
0262 D26002	1450 JNC EX4
0265 23	1455 EX5 INX H
0266 7E	1460 MOV A,M
0267 23	1465 INX H
0268 66	1470 MOV H,M
0269 6F	1475 MOV L,A
026A F1	1480 POP AF
026B E9	1485 PCHL
026C	1490 * NEW * STOP * RUN (& FRIENDS) * GOTO
026C CDEA05	1495 NEW CALL ENDCHK
026F 21153C	1500 LXI H,TXTBGN
0272 22133C	1505 SHLD TXTUNF
0275 CDEA05	1510 STOP CALL ENDCHK
0278 C3CD00	1515 JMP START
027B CDEA05	1520 RUN CALL ENDCHK
027E 11153C	1525 LXI D,TXTBGN
0281 210000	1530 RUNNXL LXI H,0
0284 CD6A06	1535 CALL FNDLNP
0287 DACD00	1540 JC START
028A EB	1545 RUNTSL XCHG
028B 22013C	1550 SHLD CURRNT
028E EB	1555 XCHG
028F 13	1560 INX D
0290 13	1565 INX D
0291 CD8A07	1570 RUNSML CALL CHKIO
0294 215A01	1575 LXI H,TAB2-1
0297 C33E02	1580 JMP EXEC
029A DF	1585 GOTO RST 3
029B D5	1590 PUSH D
029C CDEA05	1595 CALL ENDCHK
029F CD6206	1600 CALL FNDLN
02A2 C2A000	1605 JNZ AHOW
02A5 F1	1610 POP AF
02A6 C38A02	1615 JMP RUNTSL
02A9 CD7700	1620 LIST CALL TSTNUM
02AC CDEA05	1625 CALL ENDCHK
02AF CD6206	1630 CALL FNDLN
02B2 DACD00	1635 LS1 JC START
02B5 CDF806	1640 CALL PRTLN
02B8 CD8A07	1645 CALL CHKIO
02BB CD6A06	1650 CALL FNDLNP
02BE C3B202	1655 JMP LS1
02C1 0E06	1660 PRINT MVI C,6
02C3 CF	1665 RST 1
02C4 3B	1670 DB ';
02C5 06	1672 DB PR2-1-\$
02C6 CD0E00	1675 CALL CRLF
02C9 C39102	1680 JMP RUNSML
02CC CF	1685 PR2 RST 1
02CD 0D	1690 DB 0DH

C-6

0200 06	1695	DB PR0-1-0
0207 CD0E00	1700	CALL CRLF
02D2 C38102	1705	JMP RUNNO.
02D5 CF	1710	PR0 RST 1
02D6 23	1715	DB '1'
02D7 05	1720	DB PR1-1-0
02D8 DF	1725	RST 3
02D9 4D	1730	MOV C,L
02DA C3E302	1735	JMP PR3
02DD CD9606	1740	PR1 CALL C*STG
02E0 C3F002	1745	JMP PR8
02E3 CF	1750	PR3 RST 1
02E4 2C	1755	DB '1'
02E5 06	1760	DB PR6-1-0
02E6 CDD805	1765	CALL FIN
02E9 C3D502	1770	JMP PR0
02EC CD0E00	1775	PR6 CALL CRLF
02EF F7	1780	RST 6
02F0 DF	1785	PR8 RST 2
02F1 C5	1790	PUSH B
02F2 CDB005	1795	CALL PR0HUM
02F5 C1	1800	POP R
02F6 C3E302	1805	JMP PR3
02F9 CD3F07	1810	GOSUB CALL PUSHA
02FC DF	1815	RST 3
02FD D5	1820	PUSH D
02FE CD6206	1825	CALL FNDLN
0301 C2A000	1830	JNZ AHOW
0304 2A013C	1835	LHLD CURRNT
0307 E5	1840	PUSH H
0308 2A033C	1845	LHLD STKGOS
030B E5	1850	PUSH H
030C 210000	1855	LXI H,0
030F 22073C	1860	SHLD LOPVAR
0312 39	1865	DAD S
0313 22033C	1870	SHLD STKGOS
0316 C38A02	1875	JMP RUNESL
0319 CDEA05	1880	RETURN CALL ENDCHK
031C 2A033C	1885	LHLD STKGOS
031F 7C	1890	MOV A,M
0320 85	1895	ORA L
0321 CAEE05	1900	JZ QMIAT
0324 F9	1905	SPHL
0325 E1	1910	POP H
0326 22033C	1915	SHLD STKGOS
0329 E1	1920	POP H
032A 22013C	1925	SHLD CURRNT
032D D1	1930	POP D
032F CD2307	1935	CALL POPA
0331 F7	1940	RST 6
0332 CD3F07	1945	FOR CALL PUSHA

0335 CDC805	1950	CALL SEIVAL
0338 2B	1955	DCX H
0339 22073C	1960	SHLD LOPVAR
033C 21FC01	1965	LXI H,TAB5-1
033F C33F02	1970	JMP EXEC
0342 DF	1975	PR1 RST 3
0343 22083C	1980	SHLD LOELMT
0346 210402	1985	LXI H,TAB6-1
0349 C33E02	1990	JMP EXEC
034C DF	1995	PR2 RST 3
034D C35303	2000	JMP FR4
0350 210100	2005	PR3 LXI H,1
0353 22093C	2010	PR4 SHLD LOELMT
0356 2A013C	2015	PR5 LHLD CURRNT
0359 220D3C	2020	SHLD LOELMT
035C EB	2025	XCHG
035D 220F3C	2030	SHLD LOELMT
0360 010A00	2035	LXI B,10
0363 2A073C	2040	LHLD LOPVAR
0366 3B	2045	XCHG
0367 60	2050	MOV H,B
0368 68	2055	MOV L,B
0369 39	2060	DAD S
036A 3E	2065	DZ 3EH
036B 09	2070	PR7 DAD S
036C 7E	2075	MOV A,M
036D 23	2080	INX H
036E B6	2085	ORA M
036F CA8C03	2090	JZ PR8
0372 7E	2095	MOV A,M
0373 2B	2100	DCX H
0374 BA	2105	CMP D
0375 C2GB03	2110	JNZ PR7
0378 7E	2115	MOV A,M
0379 BB	2120	CMP B
037A C2GB03	2125	JNZ PR7
037D EB	2130	XCHG
037E 210000	2135	LXI H,0
0381 39	2140	DAD S
0382 44	2145	MOV B,M
0383 40	2150	MOV C,L
0384 210A00	2155	LXI B,10
0387 19	2160	DAD D
0388 CD1407	2165	CALL MDOOWN
038B F9	2170	SPHL
038C 2A0F3C	2175	PR8 LHLD LOELMT
038F EB	2180	XCHG
0390 F7	2185	RST 6
0391 FF	2190	NEXT RST 7
0392 DAEE05	2195	JC QMIAT
0395 220D3C	2200	SHLD LOPVAR

0398 D5	2205 NX0 PUSH D	03F9 D1	2465 POP D
0399 EB	2210 XCHG	03FA D1	2470 POP D
039A 2A073C	2215 LHLD LOPVAR	03FB	2475 INPUT DS 0
039D 7C	2220 MOV A,H	03FB D5	2480 IP1 PUSH D
039E B5	2225 ORA L	03FC CD9606	2485 CALL QTSTG
039F 0A0F05	2230 JZ AWHAT	03FF C30904	2490 JMP IP2
03A2 E7	2240 RST 4	0402 FF	2495 RST 7
03A3 CAB003	2245 JZ NX3	0403 DA4304	2500 JC IP4
03A6 D1	2250 POP D	0406 C31904	2505 JMP IP3
03A7 CD2307	2255 CALL POPA	0409 D5	2510 IP2 PUSH D
03AA 2A053C	2260 LHLD VARNXT	040A FF	2515 RST 7
03AD C39803	2265 JMP NX0	040B DA0005	2520 JC QWHAT
03B0 5E	2270 NX3 MOV E,M	040E 1A	2525 LDAX D
03B1 23	2275 INX H	040F 4F	2530 MOV C,A
03B2 56	2280 MOV D,M	0410 97	2535 SUB A
03B3 2A093C	2285 LHLD LOPINC	0411 12	2540 STAX D
03B6 E5	2290 PUSH H	0412 D1	2545 POP D
03B7 19	2295 DAD D	0413 CD8A06	2550 CALL PRSTG
03B8 EB	2300 XCHG	0416 79	2555 MOV A,C
03B9 2A073C	2305 LHLD LOPVAR	0417 18	2560 DCX D
03BC 73	2310 MOV M,E	0418 12	2565 STAX D
03BD 23	2315 INX H	0419 D5	2570 IP3 PUSH D
03BE 72	2320 MOV M,D	041A EB	2575 XCHG
03BF 2A0B3C	2325 LHLD LOHMT	041B 2A013C	2580 LHLD CURRNT
03C2 F1	2330 POP AF	041E E5	2585 PUSH H
03C3 B7	2335 ORA A	041F 21F803	2590 LXI H,IP1
03C4 F2C803	2340 JP NX1	0422 22013C	2595 SHLD CURRNT
03C7 EB	2345 XCHG	0425 210000	2600 LXI H,0
03C8 CDC005	2350 NX1 CALL CKHLDE	0428 39	2605 DAD S
03CB D1	2355 POP D	0429 22053C	2610 SHLD STKINP
03CC DADA03	2360 JC NX2	042C D5	2615 PUSH D
03CF 2A0D3C	2365 LHLD LOHLN	042D 3E3A	2620 MVI A,':'
03D2 22013C	2370 SHLD CURRNT	042F CD2406	2625 CALL GETIN
03D5 2A0F3C	2375 LHLD LOPPT	0432 11377F	2630 LXI D,BUFFER
03D8 EB	2380 XCHG	0435 DF	2635 RST 3
03D9 F7	2385 RST 6	0436 00	2640 NOP
03DA CD2307	2390 NX2 CALL POPA	0437 00	2645 NOP
03DD F7	2395 RST 6	0438 00	2650 NOP
03DE 210000	2400 REM LXI H,0	0439 D1	2655 POP D
03E1 3E	2405 DB 3EH SKIP NXT INST	043A EB	2660 XCHG
03E2 DF	2410 IF RST 3	043B 73	2675 MOV M,E
03E3 7C	2415 MOV A,H	043C 23	2680 INX H
03E4 B5	2420 ORA L	043D 72	2685 MOV M,D
03E5 C29102	2425 JNZ RUNSML	043E E1	2690 POP H
03E8 CD8006	2430 CALL FND SKP	043F 22013C	2695 SHLD CURRNT
03EB D28A02	2435 JNC RUNTSL	0442 D1	2700 POP D
03EE C3CD00	2440 JMP START	0443 F1	2705 IP4 POP AF
03F1 2A053C	2445 INPERR LHLD STKINP	0444 CF	2710 RST 1
03F4 F9	2450 SPHL	0445 2C	2715 DB ',,'
03F5 E1	2455 POP H	0446 03	2720 DB IP5-1-\$
03F6 22013C	2460 SHLD CURRNT	0447 C3FB03	2725 JMP IP1

044A F7	2730 IP5 RST 6	0498 D1	3210 POP D
044B 1A	2735 DEFLT LDAX D	0499 210000	3220 LXI H,0
044C FE0D	2740 CPI ODH	049C 3E01	3230 MVI A,1
044E CA5A04	2745 JZ LT1	049E C9	3240 RET
0451 CDC805	2750 LET CALL SETVAL	049F CF	3250 EXPR2 RST 1
0454 CF	2755 RST 1	04A0 2D	3260 DB '-'
0455 2C	2760 DB ',,'	04A1 06	3270 DB XP21-1-1-\$
0456 03	2765 DB LT1-1-1-\$	04A2 210000	3280 LXI H,0
0457 C35104	2770 JMP LET	04A5 C3C904	3290 JMP XP26
045A F7	2775 LT1 RST 6	04A8 CF	3300 XP21 RST 1
045B 210E02	2800 EXPR1 LXI H,TAB8-1	04A9 2B	3305 DB '+'
045E C33E02	2810 JMP EXEC	04AA 00	3310 DB XP22-1-1-\$
0461 CD8A04	2820 XP11 CALL XP18	04AB CDD304	3320 XP22 CALL EXPR3
0464 D8	2830 RC	04AE CF	3330 XP23 RST 1
0465 6F	2840 MOV L,A	04AF 2B	3335 DB '+'
0466 C9	2850 RET	04B0 15	3340 DB XP25-1-1-\$
0467 CD8A04	2860 XP12 CALL XP18	04B1 E5	3350 PUSH H
046A C8	2870 RZ	04B2 CDD304	3360 CALL EXPR3
046B 6F	2880 MOV L,A	04B5 EB	3370 XP24 XCHG
046C C9	2890 RET	04B6 E3	3380 XTHL
046D CD8A04	2900 XP13 CALL XP18	04B7 7C	3390 MOV A,H
0470 C8	2910 RZ	04B8 AA	3400 XRA D
0471 D8	2920 RC	04B9 7A	3405 MOV A,D
0472 6F	2930 MOV L,A	04BA 19	3410 DAD D
0473 C9	2940 RET	04BB D1	3415 POP D
0474 CD8A04	2950 XP14 CALL XP18	04BC FAAE04	3420 JM XP23
0477 6F	2960 MOV L,A	04BF AC	3425 XRA H
0478 C8	2970 RZ	04C0 F2AE04	3430 JP XP23
0479 D8	2980 RC	04C3 C39F00	3435 JMP QHOW
047A 6C	2990 MOV L,H	04C6 CF	3440 XP25 RST 1
047B C9	3000 RET	04C7 2D	3445 DB '-'
047C CD8A04	3010 XP15 CALL XP18	04C8 83	3450 DB XP42-1-1-\$
047F C0	3020 RNZ	04C9 E5	3455 XP26 PUSH H
0480 6F	3030 MOV L,A	04CA CDD304	3460 CALL EXPR3
0481 C9	3040 RET	04CD CDB405	3465 CALL CHKSGN
0482 CD8A04	3050 XP16 CALL XP18	04D0 C3B504	3470 JMP XP24
0485 D0	3060 RNC	04D3 CD3005	3475 EXPR3 CALL EXPR4
0486 6F	3070 MOV L,A	04D6 CF	3480 XP31 RST 1
0487 C9	3080 RET	04D7 2A	3485 DB '*'
0488 E1	3090 XP17 POP H	04D8 2C	3490 DB XP34-1-1-\$
0489 C9	3100 RET	04D9 E5	3495 PUSH H
048A 79	3110 XP18 MOV A,C	04DA CD3005	3500 CALL EXPR4
048B E1	3120 POP H	04DD 0600	3505 MVI B,0
048C C1	3130 POP B	04DF CDB105	3510 CALL CHKSGN
048D E5	3140 PUSH H	04E2 EB	3515 XCHG
048E C5	3150 PUSH B	04E3 E3	3520 XTHL
048F 4F	3160 MOV C,A	04E4 CDB105	3525 CALL CHKSGN
0490 CD9F04	3170 CALL EXPR2	04E7 7C	3530 MOV A,H
0493 EB	3180 XCHG	04E8 B7	3535 ORA A
0494 E3	3190 XTHL	04E9 CAF204	3540 JZ XP32
0495 CDC005	3200 CALL CHKHLDE	04EC 7A	3545 MOV A,D

04ED B2	3550 ORA D	0545 CF	3790 PARN RST 1
04EE EB	3555 XCHG	0546 28	3795 DB '('
04EF C2A000	3560 JNZ AHOW	0547 05	3800 DB XP43-1-\$
04F2 7D	3565 XP32 MOV A,L	0548 DF	3805 RST 3
04F3 210000	3570 LXI H,0	0549 CF	3810 RST 1
04F6 B7	3575 ORA A	054A 29	3815 DB ')
04F7 CA2205	3580 JZ XP35	054B 01	3820 DB XP43-1-\$
04FA 19	3585 XP33 DAD D	054C C9	3825 XP42 RET
04FB DAA000	3590 JC AHOW	054D C3EE05	3830 XP43 JMP QWHAT
04FE 3D	3595 DCR A	0550 CD4505	3835 RND CALL PARN *** RND(EXPR) ***
04FF C2FA04	3600 JNZ XP33	0553 7C	3840 MOV A,H
0502 C32205	3605 JMP XP35	0554 B7	3845 ORA A
0505 CF	3610 XP34 RST 1	0555 FA9F00	3850 JM QHOW
0506 2F	3615 DB '/'	0558 B5	3855 ORA L
0507 44	3620 DB XP42-1-\$	0559 CA9F00	3860 JZ QHOW
0508 B5	3625 PUSH H	055C D5	3865 PUSH D
0509 CD3005	3630 CALL EXPR4	055D E5	3870 PUSH H
050C 0600	3635 MVI B,0	055E 2A113C	3875 LHLD RANPNT
050E CDB105	3640 CALL CHKSGN	0561 116408	3880 LXI D,LSTROM
0511 EB	3645 XCHG	0564 E7	3885 RST 4
0512 E3	3650 XTHL	0565 DA6B05	3890 JC RAL
0513 CDB105	3655 CALL CHKSGN	0568 210000	3895 LXI H,0
0516 7E	3660 MOV A,D	056B 5E	3900 RAL MOV E,M
0517 B3	3665 ORA E	056C 23	3905 INX H
0518 CA0000	3670 JZ AHOW	056D 56	3910 MOV D,M
051D C5	3675 PUSH B	056E 22113C	3915 SHLD RANPNT
051E CD3005	3680 CALL DIVIDE	0571 E1	3920 POP H
051F 6C	3685 MOV H,B	0572 EB	3925 XCHG
0520 6A	3690 MOV L,C	0573 C5	3930 PUSH B
0521 C3	3695 POP B	0574 CD9405	3935 CALL DIVIDE
0527 00	3700 XP35 POP D	0577 C1	3940 POP B
0528 7C	3705 MOV A,H	0578 D1	3945 POP D
052A C3	3710 ORA A	0579 23	3950 INX H
052B FA9F00	3711 JM QHOW	057A C9	3955 RET
052E 7E	3712 MOV A,B	057B CD4505	3960 ABS CALL PARN
0529 B7	3713 ORA A	057E CDB105	3965 CALL CHKSGN
052A FCB405	3715 CM CHGSGN	0581 7C	3970 MOV A,H
052D C3D804	3720 JMP XP31	0582 B4	3975 ORA H
0530 21CC01	3725 EXPR4 LXI H,TAB4-1	0583 FA9F00	3980 JM QHOW
0533 C33E02	3730 JMP EXEC	0586 C9	3985 RET
0536 FF	3735 XP40 RST 7	0587 2A133C	3990 SIZE LHLD TXTUNF
0537 DB3F05	3740 JC XP41	058A D5	3995 PUSH D
053A 7E	3745 MOV A,M	058B EB	4000 XCHG
053B 23	3750 INX H	058C 21007F	4005 LXI H,VARBGN
053C 66	3755 MOV H,M	058F CDAA05	4010 CALL SUBDE
053D 6F	3760 MOV L,A	0592 D1	4015 POP D
053E C9	3765 RET	0593 C9	4020 RET
053F C37700	3770 XP41 CALL TSTNUM	0594 E5	4025 DIVIDE PUSH H
0542 7E	3775 MOV A,B	0595 6C	4030 MOV L,H
0543 B7	3780 ORA A	0596 2600	4035 MVI H,0
0544 C0	3785 RNZ	0598 CD9F05	4040 CALL DVI

059B 41	4045 MOV B,C	05D8 C3EE05	4300 SV1 JMP QWHAT
059C 7D	4050 MOV A,L	05DB CF	4305 FIN RST 1
059D E1	4055 POP H	05DC 3B	4310 DB ','
059E 67	4060 MOV H,A	05DD 04	4315 DB FI1-1-\$
059F 0EFF	4065 DV1 MVI C,-1	05DE F1	4320 POP AF
05A1 0C	4070 DV2 INR C	05DF C39102	4325 JMP RUNSNL
05A2 CDAA05	4075 CALL SUBDE	05E2 CF	4330 FI1 RST 1
05A5 D2A105	4080 JNC DV2	05E3 0D	4335 DB 0DH
05A8 19	4085 CAD D	05E4 04	4336 DB FI2-1-\$
05A9 C9	4090 RET	05E5 F1	4337 POP AF
05AA 7D	4095 SUBDE MOV A,L	05E6 C38102	4340 JMP RUNNXL
05AB 93	4100 SUB E	05E9 C9	4345 FI2 RET
05AC 6F	4105 MOV L,A	05EA EF	4350 ENDCHK RST 5
05AD 7C	4110 MOV A,H	05EB FE0D	4355 CPI 0DH
05AE 9A	4115 SBB D	05ED C8	4360 RZ
05AF 67	4120 MOV H,A	05EE D5	4365 QWHAT PUSH D
05B0 C9	4125 RET	05EF 11AE00	4370 AWHAT LXI D,WHAT
05B1 7C	4130 CHKSGN MOV A,H	05F2 97	4375 ERROR SUB A
05B2 B7	4135 ORA A	05F3 D3E8	4376 OUT 232
05B3 F0	4140 RP	05F5 CD8A06	4380 CALL PRISTG
05B4 7C	4145 CHGSGN MOV A,H	05F8 D1	4385 POP D
05B5 2F	4150 CMA	05F9 1A	4390 LDAX D
05B6 67	4155 MOV H,A	05FA F5	4395 PUSH AF
05B7 7D	4160 MOV A,L	05FB 97	4400 SUB A
05B8 2F	4165 CMA	05FC 12	4405 STAX D
05B9 6F	4170 MOV L,A	05FD 2A013C	4410 LHLD CURRNT
05BA 23	4175 INX H	0600 E5	4415 PUSH H
05BB 78	4180 MOV A,B	0601 7E	4420 MOV A,M
05BC EE80	4185 XRI 80H	0602 23	4425 INX H
05BE 47	4190 MOV B,A	0603 B6	4430 ORA M
05BF C9	4195 RET	0604 D1	4435 POP D
05C0 7C	4200 CKHLDE MOV A,H	0605 CACD00	4440 JZ START
05C1 AA	4205 XRA D	0608 7E	4445 MOV A,M
05C2 F2C605	4210 JP CK1	0609 B7	4450 ORA A
05C5 EB	4215 XCHG	060A FAF103	4455 JM INPERR
05C6 E7	4220 CK1 RST 4	060D CDF806	4460 CALL PRITIN
05C7 C9	4225 RET	0610 1B	4465 DCX D
05C8 FF	4230 SETVAL RST 7	0611 F1	4470 POP AF
05C9 DAEE05	4235 JC QWHAT	0612 12	4475 STAX D
05CC E5	4240 PUSH H	0613 3E3F	4480 MVI A,'?'
05CD CF	4245 RST 1	0615 D7	4485 RST 2
05CE 3D	4250 DB '='	0616 97	4490 SUB A
05CF 08	4255 DB SV1-1-\$	0617 CD8A06	4495 CALL PRISTG
05D0 DF	4260 RST 3	061A C3CD00	4500 JMP START
05D1 44	4265 MOV B,H	061D D5	4505 QSORRY PUSH D
05D2 4D	4270 MOV C,L	061E 11B400	4510 ASORRY LXI D,SORRY
05D3 E1	4275 POP H	0621 C3F205	4515 JMP ERROR
05D4 71	4280 MOV M,C	0624 D7	4520 GETIN RST 2
05D5 23	4285 INX H	0625 11377F	4525 LXI D,BUFFER
05D6 70	4290 MOV M,B	0628 CD8A07	4530 GL1 CALL CHKIO
05D7 C9	4295 RET	062B CA2806	4535 JZ GL1
		062E D7	4540 RST 2

062F FE0A	4545	CPI OAH	0683 C27F06	4800	JNZ FL2
0631 CA2806	4550	JZ GL1	0686 13	4805	INX D
0634 B7	4555	ORA A	0687 C36A06	4810	JMP FL1
0635 CA2806	4560	JZ GL1	068A 47	4815	PR1STG MOV B,A
0638 FE7F	4565	CPI 7FH	068B 1A	4820	PS1 LDAX D
063A CA4D06	4570	JZ GL3	068C 13	4825	INX D
063D FE7D	4575	CPI 7DH	068D B8	4830	CMP B
063F CA5A06	4580	JZ GL4	068E C8	4835	RZ
0642 12	4585	STAX D	068F D7	4840	RST 2
0643 13	4590	INX D	0690 FE0D	4845	CPI 0DH
0644 FE0D	4595	CPI 0DH	0692 C28B06	4850	JNZ PS1
0646 C8	4600	RZ	0695 C9	4855	RET
0647 7B	4605	MOV A,E	0696 CF	4860	QTSTG RST 1
0648 FE7F	4610	CPI 7FH	0697 22	4865	DB '''
064A C22B06	4615	JNZ GL1	0698 0F	4870	DB QT3-1-\$
064D 7B	4620	GL3 MOV A,E	0699 3E22	4875	MVI A,'''
064E FE37	4625	CPI 37H	069B CD8A06	4880	QT1 CALL PR1STG
0650 CA5A06	4630	JZ GL4	069E FE0D	4885	CPI 0DH
0653 1B	4635	DCX D	06A0 E1	4890	POP H
0654 3E5C	4640	MVI A,5CH	06A1 CA8102	4895	JZ RUNNXL
0656 D7	4645	RST 2	06A4 23	4900	QT2 INX H
0657 C32806	4650	JMP GL1	06A5 23	4905	INX H
065A CD0E00	4655	GL4 CALL CRLF	06A6 23	4910	INX H
065D 3E5E	4660	MVI A,5EH	06A7 E9	4915	PCHL
065F C32406	4665	JMP GETLN	06A8 CF	4920	QT3 RST 1
0662 7C	4670	FNDLN MOV A,H	06A9 27	4925	DB 27H
0663 B7	4675	ORA A	06AA 05	4930	DB QT4-1-\$
0664 FA9F00	4680	JM QHOW	06AB 3E27	4935	MVI A,27H
0667 11153C	4685	LXI D,TXTBGN	06AD C39B06	4940	JMP QT1
066A	4690	FNDLNP DS 0	06B0 CF	4945	QT4 RST 1
066A E5	4695	FL1 PUSH H	06B1 5F	4950	DB 5FH
066B 2A133C	4700	LHLD TXTUNF	06B2 08	4955	DB QT5-1-\$
066E 2B	4705	DCX H	06B3 3E8D	4960	MVI A,8DH
066F E7	4710	RST 4	06B5 D7	4965	RST 2
0670 E1	4715	POP H	06B6 D7	4970	RST 2
0671 D8	4720	RC	06B7 E1	4975	POP H
0672 1A	4725	LDAX D	06B8 C3A406	4980	JMP QT2
0673 95	4730	SUB L	06BB C9	4985	QT5 RET
0674 47	4735	MOV B,A	06BC D5	4990	PR1NUM PUSH D
0675 13	4740	INX D	06BD 110A00	4995	LXI D,10
0676 1A	4745	LDAX D	06C0 D5	5000	PUSH D
0677 9C	4750	SBB H	06C1 42	5005	MOV B,D
0678 DA7F06	4755	JC FL2	06C2 0D	5010	DCR C
067B 1B	4760	DCX D	06C3 CDB105	5015	CALL CHKSGN
067C B0	4765	ORA B	06C6 F2CC06	5020	JP FN1
067D C9	4770	RET	06C9 062D	5025	MVI B,'-'
067E	4775	FNDNXT DS 0	06CB 0D	5030	DCR C
067E 13	4780	INX D	06CC C5	5035	FN1 PUSH B
067F 13	4785	FL2 INX D	06CD CD9405	5040	FN2 CALL DIVIDE
0680 1A	4790	FND5KP LDAX D	06D0 78	5045	MOV A,B
0681 FE0D	4795	CPI 0DH	06D1 B1	5050	ORA C

06D2 CAD006	5055 JZ PN3	071C 1B	5310 POP DCX D
06D5 E3	5060 XTHL	071D 2B	5315 DCX H
06D6 2D	5065 DCR L	071E 1A	5320 LDAX D
06D7 E5	5070 PUSH H	071F 77	5325 MOV M,A
06D8 60	5075 MOV H,B	0720 C31407	5330 JMP MVDOWN
06D9 69	5080 MOV L,C	0723 C1	5335 POPA POP B
06DA C3CD06	5085 JMP PN2	0724 E1	5340 POP H
06DD C1	5090 PN3 POP B	0725 22073C	5345 SHLD LOPVAR
06DE 0D	5095 PN4 DCR C	0728 7C	5350 MOV A,H
06DF 79	5100 MOV A,C	0729 B5	5355 ORA L
06E0 B7	5105 ORA A	072A CA3D07	5360 JZ PPI
06E1 FAEA06	5110 JM PN5	072D E1	5365 POP H
06E4 3E20	5115 MVI A,' '	072E 22093C	5370 SHLD LOPINC
06E6 D7	5120 RST 2	0731 E1	5375 POP H
06E7 C3DE06	5125 JMP PN4	0732 220B3C	5380 SHLD LOFLMT
06EA 78	5130 PN5 MOV A,B	0735 E1	5385 POP H
06EB D7	5135 RST 2	0736 220D3C	5390 SHLD LOFLN
06EC 5D	5140 MOV E,L	0739 E1	5395 POP H
06ED 7B	5145 PN6 MOV A,E	073A 220F3C	5400 SHLD LOPPT
06EE FE0A	5150 CPI OAH	073D C5	5405 PPI PUSH B
06F0 D1	5155 POP D	073E C9	5410 RET
06F1 C8	5160 RZ	073F 21A77F	5415 PUSHA LXI H,STKLMT
06F2 C630	5165 ADI '0'	0742 CDB405	5420 CALL CHGSGN
06F4 D7	5170 RST 2	0745 C1	5425 POP B
06F5 C3ED06	5175 JMP PN6	0746 39	5430 DAD S
06F8 1A	5180 PRTIN LDAX D	0747 D21D06	5435 JNC QSCRRY
06F9 6F	5185 MOV L,A	074A 2A073C	5440 LHLD LOPVAR
06FA 13	5190 INX D	074D 7C	5445 MOV A,H
06FB 1A	5195 LDAX D	074E B5	5450 ORA L
06FC 67	5200 MOV H,A	074F CA6507	5455 JZ PUI
06FD 13	5205 INX D	0752 2A0F3C	5460 LHLD LOPPT
06FE 0E04	5210 MVI C,4	0755 E5	5465 PUSH H
0700 CDBC06	5215 CALL PRTNUM	0756 2A0D3C	5470 LHLD LOFLN
0703 3E20	5220 MVI A,' '	0759 E5	5475 PUSH H
0705 D7	5225 RST 2	075A 2A0B3C	5480 LHLD LOFLMT
0706 97	5230 SUB A	075D E5	5485 PUSH H
0707 CD8A06	5235 CALL PRISTG	075E 2A093C	5490 LHLD LOPINC
070A C9	5240 RET	0761 E5	5495 PUSH H
070B E7	5245 MVUP RST 4	0762 2A073C	5500 LHLD LOPVAR
070C C8	5250 RZ	0765 E5	5505 PUI PUSH H
070D 1A	5255 LDAX D	0766 C5	5510 PUSH B
070E 02	5260 STAX B	0767 C9	5515 RET
070F 13	5265 INX D	0768 C26D07	5520 OC2 JNZ OC3
0710 03	5270 INX B	076B F1	5525 POP AF
0711 C30B07	5275 JMP MVUP	076C C9	5530 RET
0714 78	5280 MVDOWN MOV A,B	076D DBED	5535 OC3 IN STATUS
0715 92	5285 SUB D	076F E601	5540 ANI TXRDY
0716 C21C07	5290 JNZ MD1	0771 CA6D07	5545 JZ CC3
0719 79	5295 MOV A,C	0774 F1	5550 POP AF
071A 93	5300 SUB E	0775 D3EC	5555 OUT DATA
071B C8	5305 RZ	0777 FE0D	5560 CPI ODH

0779 C0	5565 RAZ	07CE C00B05	5910 CALL FIN
077A 3E0A	5570 MVI A,0AH	07D1 C38E05	5920 JMP QWHAT
077C D7	5575 RST 2	07D4 CD4505	7000 GET CALL PARAM
077D 3E7F	5576 MVI A,7FH	07D7 65	7010 MOV H,L
077E D7	5577 RST 2	07D8 E5	7040 PUSH H
0780 D7	5578 RST 2	07D9 21377F	7110 LXI H,BUFFER
0781 D7	5579 RST 2	07DC 36DB	7120 MVI M,ODSH
0782 D7	5580 RST 2	07DE 23	7130 INX H
0783 D7	5581 RST 2	07DF F1	7140 POP AF
0784 D7	5582 RST 2	07E0 77	7150 MOV M,A PORT ID
0785 D7	5583 RST 2	07E1 23	7160 INX H
0786 D7	5584 RST 2	07E2 36C9	7190 MVI M,OC9H RET
0787 3E0D	5585 MVI A,0DH	07E4 CD377F	7220 CALL BUFFER
0789 C9	5586 RET	07E7 6F	7230 MOV L,A
078A DBED	5590 CHKIO IN STATUS	07E8 2600	7240 MVI H,0
078C E602	5595 ANI RXRDY	07EA C9	7250 RET
078E C8	5600 RZ	07EB DF	7300 PUT RST 3 EXPR
078F DBEC	5605 IN DATA	07EC 65	7310 MOV H,L
0791 E67F	5610 ANI 7FH	07ED E5	7312 PUSH H
0793 FE0F	5615 CPI OFH	07EE CF	7320 RST 1
0795 C2A207	5620 JNZ CIL	07EF 2C	7330 LB ',,'
0798 3A003C	5625 LDA OCSW	07F0 15	7340 DB BQWHAT-1->\$
079B 2F	5630 CMA	07F1 DF	7350 RST 3 EZPR
079C 32003C	5635 STA OCSW	07F2 65	7352 MOV H,L
079E C38A07	5640 JMP CHKIO	07F3 E3	7356 XTML
07A2 FE03	5650 CIL CPI 3	07F4 E5	7370 PUSH H
07A4 C0	5655 RAZ	07F5 21377F	7410 LXI H,BUFFER
07A5 C3CD00	5660 JMP START	07F8 36D3	7440 MVI M,ODSH
07A8 DF	5670 CALL RST 3 EXPR	07FA 23	7450 INX H
07A9 22377F	5680 SHLD BUFFER WHERE	07FB F1	7460 POP AF
07AC 3EC3	5690 MVI A,OC3H	07FC 77	7470 MOV M,A
07AE 32377F	5700 STA BUFFER	07FD 23	7480 INX H
07B1 CF	5710 RST 1	07FE 36C9	7490 MVI M,OC9H RET
07B2 2F	5720 DB ',,'	0800 F1	7500 POP AF
07B3 07	5730 DB CQWHAT-1->\$	0801 CD377F	7520 CALL BUFFER
07B4 DF	5740 RST 3 EXPR	0804 CD0B05	7522 CALL FIN
07B5 CD377F	5750 CALL BUFFER	0807 C38E05	7530 BQWHAT JMP QWHAT
07B8 C00B05	5760 CALL FIN	080A CD4505	7540 AMP CALL PARAM
07BB C38E05	5770 CQWHAT JMP QWHAT	080D D5	7552 PUSH D
07BE CD4505	5780 PEEK CALL PARAM	080E 114508	7554 LXI D,TABLE3
07C1 6F	5790 MOV L,M	0811 C31808	7556 JMP VOLTA
07C2 2800	5800 MVI H,0	0814 CD4505	7570 VOLT CALL PARAM
07C4 C9	5810 RET	0817 D5	7572 PUSH D
07C5 DF	5820 POKE RST 3 EXPR	081D 114508	7574 LXI D,TABLE3
07C6 E5	5830 PUSH H WHERE	081E C9	7576 VOLTA PUSH H
07C7 CF	5840 RST 1	081F 02	7578 MOV C,M
07C9 2C	5850 LB ',,'	081D 7D	7580 MOV A,C
07CB 3D	5860 DB BQWHAT-1->\$	081E 210000	7582 LXI H,0
07CA E5	5870 RST 3 EXPR	0821 FE00	7584 CPI 0
07CB 2D	5880 MOV A,1 WHAT	0823 CA4306	7586 JZ PSPORB
07CC E1	5890 POP H WHERE	0826 E5	7588 PUSH H
07CD 77	5900 MOV M,A POKE IT		

0827 B8	7592 XCHT	3C11 0000	9055 RASBNC DW 0
0828 0680	7592 MVL B,80H	3C13 153C	9060 TXTLNE DW 7800H
082A 78	7594 MOV A,B	3C15	9065 TXTBGN DS 0
082B A1	7596 P5ANAC ANA C	7F00	9075 ORG 7F00H
082C CA3908	7598 JZ P5INXH	7F00	9080 VAMBGN DS 54
082F 5E	7600 MOV E,M	7F00	9082 TXTEND EQU VAMBGN
0830 23	7602 INX H	7F36	9085 DS 1
0831 55	7604 MOV D,M	7F37	9090 BUFFER DS 72
0832 23	7606 INX H	7F7F	9095 BUFEND DS 0
0833 E3	7608 XTHL	7F7F	9100 DS 40
0834 19	7610 LAD D	7FA7	9105 STRLMT DS 0
0835 E3	7612 XTHL	8000	9110 ORG 8000H
0836 C33B08	7614 JMP P5MVAB	8000	9115 STACK DS 0
0839 23	7616 P5INXH INX H	8000	9120 *PORT DEFINITIONS
083A 23	7618 INX H	00ED	9122 STATUS EQU 0EDH
083B 78	7620 P5MVAB MOV A,B	00ED	9124 COMMND EQU 0EDH
083C 0F	7622 RRC	00EC	9126 DATA EQU 0ECH
083D 47	7624 MOV B,A	8000	9128 *MODE DEFINITIONS
083E D22B08	7626 JNC P5ANAC	0003	9130 BAUD EQU 3
0841 E1	7628 POP H	000C	9132 CHARLEN EQU 0CH
0842 C1	7630 P5POPB POP B	00C0	9134 STOPS EQU 0C0H
0843 D1	7632 POP D	8000	9136 *COMMAND DEFINITIONS
0844 C9	7634 RET	0020	9138 RTS EQU 20H
0845 A861	7650 TABL5 DW 25000	0004	9140 RXE EQU 4
0847 D430	7652 DW 12500	0001	9142 TXEN EQU 1
0849 6A18	7654 DW 3250	0002	9144 DTR EQU 2
084B 350C	7656 DW 3125	8000	9146 *STATUS DEFINITIONS
084D 1A06	7658 DW 1562	0001	9148 TXRDY EQU 1
084F 0D03	7660 DW 781	0002	9150 RXRDY EQU 2
0851 8601	7662 DW 390	0004	9152 TXE EQU 4
0853 C300	7664 DW 195		
0855 983A	7666 TABL3 DW 15000		
0857 4C1D	7668 DW 7500		
0859 A60E	7670 DW 3750		
085B 5507	7672 DW 1875		
085D AA03	7674 DW 750		
085F B701	7676 DW 450		
0861 EA00	7678 DW 234		
0863 7500	7680 DW 117		
0864	8999 LSTRM EQU \$-1		
3C00	9000 ORG 3C00H		
3C00 FF	9005 OCMW DB 0FFH		
3C01 0000	9010 CURRNT DW 0		
3C03 0000	9015 STKCRB DW 0		
3C05	9020 VAPINT DS 0		
3C05 0000	9025 STKINP DW 0		
3C07 0000	9030 LOPVAL DW 0		
3C09 0000	9035 LOPINC DW 0		
3C0B 0000	9040 LOPLMT DW 0		
3C0D 0000	9045 LOPLEN DW 0		
3C0F 0000	9050 LOPPT DW 0		

Appendix D
DATA SORTING PROGRAM

The following program is written in TYMBASIC for use on a TYMSHARE system IBM 360 computer. The program will sort the preliminary data stored on the tape cassette unit from random cell records into specific cell record files.

```
10 PRINT DATE
20 OPEN "R1",OUTPUT,1
30 OPEN "R2",OUTPUT,2
40 OPEN "R3",OUTPUT,3
50 OPEN "R4",OUTPUT,4
60 OPEN "R5",OUTPUT,5
70 OPEN "R6",OUTPUT,6
80 STRING A
90 DIM A2(1000),A3(1000),A4(1000),A5(1000),A6(1000),A7(1000)
100 DIM A8(1000),A9(1000),B1(1000),B2(1000),B3(1000),B4(1000),B5(1000)
110 DIM B6(1000),B7(1000),B8(1000),B9(1000),C1(1000),C2(1000),C3(1000)
120 DIM C4(1000),C5(1000),C6(1000)
130 TEXT A1(1000)
140 A="16% 23( # )/"
150 OPEN "D248",INPUT,7
160 FOR I=1 TO 1000
170 ON ENDFILE(7) GO TO 220
180 ON ERROR GO TO 290
190 INPUT FROM 7 IN FORM A:A1(I),A2(I),A3(I),A4(I),A5(I),A6(I),A7(I),A8(
I),A9(I),B1(I),B2(I),B3(I),B4(I),B5(
**I),B6(I),B7(I),B8(I),B9(I),C1(I),C2(I),C3(I),C4(I),C5(I),C6(I)
200 J=I
210 NEXT I
220 PRINT DATE
230 FOR I=1 TO J
240 WRITE ON A2(I) IN FORM A:A1(I),A2(I),A3(I),A4(I),A5(I),A6(I),A7(I),A
8(I),A9(I),B1(I),B2(I),B3(I),B4(I),B
**5(I),B6(I),B7(I),B8(I),B9(I),C1(I),C2(I),C3(I),C4(I),C5(I),C6(I)
250 NEXT I
260 PRINT DATE
270 CLOSE 1,2,3,4,5,6,7
280 GO TO 300
290 PRINT I,A1(I)
300 PRINT
```


Appendix E

DATA REPORT PROGRAM

The following program, written in TYMBASIC, is for generation of reports on the performance of individual cells. The program first computes a statistical mean for the cyclic energy efficiency. Any cycle having an energy efficiency that is more than one standard deviation from the mean is denoted and ignored for a second calculation of statistical means. This allows for a rejection of cyclic data in which reference electrode, thermal, or electronic malfunctions occur. This data is marked with a * on the report output. The second calculation of statistical means and standard deviations is outputted at the end of the report and individual cycle data is marked if it is one standard deviation (S) or two standard deviations (D) from the mean energy efficiency.

The program generates two reports, the first is a cycle-by-cycle efficiency report while the second gives a cycle-by-cycle report of voltages (cell, iron, and zinc) at several points in the charge/discharge cycle.

The last function of this program is to generate a data file from which data can be plotted.


```

430 IF L#2 THEN GO TO 520
440 IF F(L)#1 THEN GO TO 2670
450 IF FA#0 THEN GO TO 870
460 OPEN "REFFE2",OUTPUT,2
470 OPEN "RV2",OUTPUT,3
480 FA=1
490 GOSUB 2700
500 L=2
510 GO TO 870
520 IF L#3 THEN GO TO 610
530 IF F(L)#1 THEN GO TO 2670
540 IF FA#0 THEN GO TO 870
550 OPEN "REFFE3",OUTPUT,2
560 OPEN "RV3",OUTPUT,3
570 FA=1
580 GOSUB 2700
590 L=3
600 GO TO 870
610 IF L#4 THEN GO TO 700
620 IF F(L)#1 THEN GO TO 2670
630 IF FA#0 THEN GO TO 870
640 OPEN "REFFE4",OUTPUT,2
650 OPEN "RV4",OUTPUT,3
660 FA=1
670 GOSUB 2700
680 L=4
690 GO TO 870
700 IF L#5 THEN GO TO 790
710 IF F(L)#1 THEN GO TO 2670
720 IF FA#0 THEN GO TO 870
730 OPEN "REFFE5",OUTPUT,2
740 OPEN "RV5",OUTPUT,3
750 FA=1
760 GOSUB 2700
770 L=5
780 GO TO 870
790 IF L#6 THEN GO TO 2680
800 IF F(L)#1 THEN GO TO 2670
810 IF FA#0 THEN GO TO 870
820 OPEN "REFFE6",OUTPUT,2
830 OPEN "RV6",OUTPUT,3
840 FA=1
850 GOSUB 2700

```

14

```

1290 IF ZPT6(K)<FPT6(K) THEN GO TO 1310
1300 ZP=ZPT6(K)
1310 FPT6(K)=ZPT6(K)+FPT6(K)
1320 ZPT6(K)=ZP
1330 GO TO 1360
1340 ZPT6(K)=FPT6(K)
1350 FPT6(K)=0
1360 A=A+1
1370 CPT1(K)=ZPT1(K)+FPT1(K)
1380 CPT3(K)=ZPT3(K)+FPT3(K)
1390 CPT4(K)=ZPT4(K)+FPT4(K)
1400 CPT5(K)=ZPT5(K)+FPT5(K)
1410 CPT6(K)=ZPT6(K)+FPT6(K)
1420 IF CL="R" THEN GO TO 1450
1430 IF CL="F" THEN GO TO 1540
1440 GO TO 1500
1450 VE(K)=(FPT5(K)/FPT1(K))*100
1460 UE(K)=FE(K)
1470 CPT6(K)=FPT3(K)
1480 GE(K)=(VE(K)*UE(K))/100
1490 GO TO 1540
1500 VE(K)=(ZPT5(K)/ZPT1(K))*100
1510 UE(K)=FE(K)
1520 CPT6(K)=ZPT3(K)
1530 GE(K)=(VE(K)*UE(K))/100
1540 S02=S02+GE(K)
1550 D02=D02+0.01(K)**2
1560 NEXT K
1570 H02=S02/A
1580 S02=S02+((D02-(C02**2)/A))/A**3
1590 H=0
1600 S7=0
1610 S1=0
1620 T0=0
1630 S16=0
1640 S17=0
1650 C1=0
1660 D0=0
1670 C0=0
1680 D1=0
1690 D4=0
1700 A=0
1710 DO K=1 TO J

```

17

```

1720 C=CELL(K)
1730 IF C#L THEN GO TO 1880
1740 IF ABS(OE(K)-MOE)<1+SDOE THEN GO TO 1770
1750 PD(K)="#"
1760 GO TO 1880
1770 SV=SV+VE(K)
1780 SU=SU+UE(K)
1790 XO=XO+OE(K)
1800 SIC=SIC+CD(K)
1810 SID=SID+DD(K)
1820 DV=DV+VE(K)**2
1830 DU=DU+UE(K)**2
1840 XD=XD+OE(K)**2
1850 DIC=DIC+CD(K)**2
1860 DID=DID+DD(K)**2
1870 A=A+1
1880 NEXT K
1890 MV=SV/A
1900 MU=SU/A
1910 MO=XO/A
1920 MC=SIC/A
1930 MD=SID/A
1940 TDVE=SQRT(ABS((DV-((SV**2)/A))/(A-1)))
1950 TDUE=SQRT(ABS((DU-((SU**2)/A))/(A-1)))
1960 TDOE=SQRT(ABS((XD-((XO**2)/A))/(A-1)))
1970 TDC=SQRT(ABS((DIC-((SIC**2)/A))/(A-1)))
1980 TDD=SQRT(ABS((DID-((SID**2)/A))/(A-1)))
1990 FOR K=1 TO J
2000 C=CELL(K)
2010 IF C#L THEN GO TO 2350
2020 M=M+1
2030 IF M<35 THEN GO TO 2110
2040 PRINT IN FORM "3/"
2050 PRINT "...."
2060 PRINT IN FORM "3/"
2070 GOSUB 3050
2080 N=12
2090 M=-16
2100 FA=0
2110 IF CL="R" THEN GO TO 2140
2120 IF CL="F" THEN GO TO 2230
2130 GO TO 2190
2140 VE(K)=(FPTS(K)/FPTI(K))*100

```

```

2150 UE(K)=FE(K)
2160 CPT6(K)=FPT3(K)
2170 OE(K)=(VE(K)*UE(K))/100
2180 GO TO 2230
2190 VE(K)=(ZPT5(K)/ZPT1(K))*100
2200 UE(K)=ZE(K)
2210 CPT6(K)=ZPT3(K)
2220 OE(K)=(VE(K)*UE(K))/100
2230 IF PD(K)="*" THEN GO TO 2310
2240 IF ABS(OE(K)-MO)<TD0E THEN GO TO 2300
2250 IF ABS(OE(K)-MO)<2*TD0E THEN GO TO 2280
2260 PD(K)="D"
2270 GO TO 2310
2280 PD(K)="S"
2290 GO TO 2310
2300 PD(K)=" "
2310 PRINT IN IMAGE V:DCODE(K),CT(K),CPT6(K),CYCLE(K),CD(K),DD(K),VE(K),
UE(K),OE(K),ZE(K),FE(K),PD(K)
2320 IF PD(K)="*" THEN GO TO 2350
2330 IF PD(K)="D" THEN GO TO 2350
2340 WRITE ON 2 IN FORM T:DCODE(K),TCODE(K),CYCLE(K),CD(K),DD(K),VE(K)
,UE(K),OE(K),ZE(K),FE(K)
2350 NEXT K
2360 CLOSE 2,3
2370 PRINT IN FORM "3/"
2380 PRINT IN IMAGE "STATISTICAL MEAN          %%.%   %%.%   %%.%   %%.%
%%.% ":MC,MD,MV,MU,MO
2390 PRINT
2400 PRINT IN IMAGE "STANDARD DEVIATION        %%.%   %%.%   %%.%   %%.%
%%.% ":TDC,TDD,TDVE,TDUE,TD0E
2410 FOR X=1 TO (46-N-M)
2420 PRINT
2430 NEXT X
2440 PRINT"....."
2450 PRINT IN FORM "3/"
2460 PRINT"....."
2470 GOSUB 3150
2480 K=0
2490 FOR K=1 TO J
2500 C=CELL(K)
2510 IF C#L THEN GO TO 2610
2520 M=M+1
2530 IF M<50 THEN GO TO 2600
2540 PRINT IN FORM "3/"
2550 PRINT "....."
2560 PRINT IN FORM "3/"
2570 GOSUB 3150

```

E-7

```

2580 N=4
2590 M=1
2600 PRINT IN IMAGE U:CYCLE(K),CPT1(K),ZPT1(K),FPT1(K),CPT2(K),OCV(K),CP
T5(K),ZPT5(K),FPT5(K),CPT6(K),ZPT6(K
**),FPT6(K),CPT3(K),DCV(K)
2610 NEXT K
2620 FOR X=1 TO (61-M-N)
2630 PRINT
2640 NEXT X
2650 PRINT "....."
2660 PRINT IN FORM "3/"
2670 NEXT L
2680 CLOSE 1
2690 GO TO 3110
2700 PRINT IN IMAGE"CELL = #":L
2710 STRING DA,INI,TEMP,FLOW,FEANL,FEOH,ZNANL,ZNOH,MEMB,IR,ZN
2720 INPUT IN IMAGE"INITIALS : #":INI
2730 INPUT IN IMAGE"DATE      : #":DA
2740 INPUT IN IMAGE"TEMPERATURE : #":TEMP
2750 INPUT IN IMAGE "FACE VELOCITY : #":FLOW
2760 INPUT IN IMAGE "MEMBRANE : #":MEMB
2770 INPUT IN IMAGE "FE ELECTRODE : #":IR
2780 INPUT IN IMAGE "  ANALYSIS : #":FEANL
2790 INPUT IN IMAGE "  OH ANALYSIS : #":FEOH
2800 INPUT IN IMAGE "ZN ELECTRODE : #":ZN
2810 INPUT IN IMAGE "  ANALYSIS : #":ZNANL
2820 INPUT IN IMAGE "  OH ANALYSIS : #":ZNOH
2830 PRINT "....."
2840 PRINT
2850 PRINT
2860 PRINT
2870 PRINT
2880 PRINT"                                **HYBRID REDOX BATTERY**"
2890 PRINT
2900 PRINT
2910 PRINT
2920 PRINT
2930 PRINT IN IMAGE "BATTERY CONSTRUCTION ( # )":L
2940 PRINT "*****"
*****"
2950 PRINT
2960 PRINT IN IMAGE"# * [# FE(2+):# NaOH] * # * [# ZNO:# NaOH] * #":IR,F
EANL,FEOH,MEMB,ZNANL,ZNOH,ZN
2970 PRINT
2980 PRINT "*****"
*****"
2990 PRINT

```

```

3000 PRINT IN IMAGE"INITIALS  ???      DATE  ???????  TEMPERATURE  %
%.% C":INI,DA,TEMP
3010 PRINT
3020 PRINT IN IMAGE"FLOW RATE # CM/SEC":FLOW
3030 PRINT
3040 PRINT
3050 PRINT " DATE  C.T. C.P. CYCLE  +      -      VOLT  COUL  ENERGY
ZN      FE"
3060 PRINT "      HRS  E  NO.  C.D.  C.D.  EFF  EFF  EFF
EFF  EFF"
3070 PRINT "      MA/CM2      %      %      %
%      %"
3080 PRINT "-----"
---
3090 PRINT
3100 RETURN
3110 PRINT
3120 PRINT
3130 PRINT
3140 GO TO 3230
3150 PRINT IN FORM "3/"
3160 PRINT IN IMAGE "CELL UNDER TEST (# )":L
3170 PRINT
3180 PRINT" CN  CP  ZP  FP  CP  CP  CP  ZP  FP  CP  ZP  FP
CP  CP"
3190 PRINT" #      30 MIN      END  OCV      30 MIN      USABLE
ZN  OCV"
3200 PRINT"-----"
---
3210 PRINT
3220 RETURN
3230 PRINT

```


Appendix F
REDUCED DATA REPORT

The following listing details the typical output from the Data Report Program (Appendix E). The data presented is for the full cell cycle life test which is still in progress. The output for the first section is self-explaining except for C.P. E which is the usable energy cell potential, nominally set by the operator at 1.45 volts. Deviations from this value are indicative of various malfunctions. The second set of data gives the cycle number, CN, the voltages at half-way into charge, the cell potential at the end of charge, the charged OCV, the average discharge voltages, the voltages when the cell potential was 1.45 (usable energy point), the cell potential when the zinc potential was 0.80 vs Hg/HgO (zinc depletion), and the open circuit potential (OCV) of the discharged cell. Both OCV potentials are recorded after one minute at OCV to allow for attainment of equilibrium.

LOCKHEED P-7800X BATTERY

EXPERIMENTAL CONSTRUCTION (G)

PL - SAT'D FeCl₂ + 1.5 NaOH * W-114 * SAT'D ZnO:5 NaOH * Cd/Fe

INITIALS IPH DATE 9/26/77 TEMPERATURE 40 C

FLOW RATE 10 CM/SEC

TIME	C.F.	C.P.	CYCLE	+	-	VOLT	COUL	ENERGY	AM	FE
HR:MIN	SEC	SEC	NO.	C.D.	C.D.	EFF	EFF	EFF	EFF	EFF
-----	-----	-----	-----	-----	-----	-----	-----	-----	-----	-----
79/142	2.0	1.41	1	36.1	35.7	92.4	70.4	64.4	95.6	57.5 D
79/142	2.0	1.46	2	36.3	35.7	90.4	67.7	61.2	94.7	55.5 D
79/142	2.0	1.54	3	37.3	35.7	92.5	83.5	76.4	97.1	6.3
79/143	2.0	1.33	4	31.2	37.7	89.9	35.4	31.5	71.1	5.5 *
79/143	.5	1.54	5	29.7	26.7	0.0	123.6	0.0	279.8	24.7 *
79/143	2.0	1.25	6	36.4	35.0	3.6	24.7	1.0	85.6	4.5 *
79/143	2.0	1.47	7	23.5	35.3	94.4	84.8	79.9	91.8	52.1
79/143	2.0	1.31	8	39.3	36.0	91.5	88.1	80.1	91.7	16.9
79/143	2.0	1.29	9	36.4	28.7	91.5	77.6	71.0	96.3	6.6 S
79/144	2.0	1.46	10	37.7	35.0	91.5	58.3	52.8	93.2	5.7 D
79/144	1.8	1.50	11	43.2	36.0	86.3	40.3	34.4	87.2	25.9 *
79/144	2.0	1.45	12	37.8	35.3	90.1	67.6	61.2	84.6	7.9 D
79/144	2.0	1.45	12	36.2	35.3	88.2	61.4	53.7	96.6	53.6 D
79/144	2.0	1.44	15	35.2	35.7	91.0	88.4	80.1	94.8	64.5
79/145	2.0	1.56	16	37.2	40.7	91.5	71.8	65.5	106.8	11.8 S
79/145	2.0	1.46	17	35.1	35.3	90.1	55.6	50.4	97.4	36.8 D
79/145	2.0	1.51	18	41.9	36.7	83.9	55.1	46.2	85.4	14.4 S
79/145	2.0	1.46	19	39.7	35.9	91.0	88.6	81.0	96.7	83.6
79/146	2.0	2.48	20	47.0	27.3	91.5	119.6	109.8	129.1	56.2 D
79/146	2.0	1.41	21	29.9	37.0	89.4	93.2	82.8	94.2	5.3
79/146	2.0	1.42	22	25.7	30.7	91.4	95.8	87.4	97.1	6.8
79/146	2.0	1.44	23	25.0	36.3	92.4	88.7	81.9	92.7	7.7
79/147	2.0	1.47	24	25.2	36.7	89.9	69.2	62.1	90.1	8.2 D
79/147	2.0	1.43	25	24.7	37.3	90.4	97.0	87.3	99.7	46.3
79/147	2.0	1.44	26	25.7	37.3	88.1	89.3	78.3	92.9	36.8
79/147	2.0	1.46	27	24.5	36.7	91.4	92.5	83.7	95.1	28.6
79/147	.6	1.40	28	38.3	39.0	87.3	103.6	91.5	108.0	17.4
79/147	1.4	1.35	29	37.4	37.3	89.9	84.4	75.6	86.2	2.5
79/147	2.0	1.49	30	37.8	35.3	91.0	69.7	63.7	98.4	18.8 D
79/148	2.0	1.43	31	41.2	38.7	87.2	47.5	40.9	92.3	1.3 *
79/148	2.0	1.36	32	37.3	37.3	88.2	58.9	51.9	100.2	47.1 D
79/148	2.0	1.35	33	37.3	37.7	90.1	69.2	62.1	98.7	31.7 D
79/148	2.0	1.43	34	37.6	37.3	89.1	103.3	92.7	106.7	25.5
79/148	2.0	1.46	35	41.7	38.6	92.1	27.4	81.0	89.2	62.7

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DATE	TIME	C.P. %	CYCLE NO.	+ C.D. MAZGMR	- C.D.	VOLT EFF %	COB EFF %	WINDUP EFF %	SE EFF %	RE EFF %
797143	2.0	1.44	36	38.7	37.0	91.0	93.4	85.5	95.3	91.7
797143	2.0	1.43	37	37.5	37.0	92.5	92.0	86.2	93.3	90.5
797143	2.0	1.27	38	42.1	40.7	91.1	83.4	57.3	91.5	83.4
797143	2.0	1.45	39	40.2	45.7	91.1	69.2	51.0	90.4	82.7
797143	2.0	1.43	40	36.8	35.7	91.1	101.5	92.2	100.4	110.3
797144	2.0	1.41	41	42.7	40.7	91.1	95.7	87.6	93.9	90.1
797143	2.0	1.47	42	41.8	39.7	92.1	93.2	90.2	93.4	91.7
797150	2.0	1.35	43	32.8	22.3	90.5	87.2	81.8	83.4	83.2
797150	2.0	1.44	44	35.7	34.7	91.1	91.1	85.0	91.0	91.1
797150	2.0	1.44	45	38.2	36.7	91.2	99.1	82.1	105.5	90.7
797150	2.0	1.47	46	35.2	39.7	90.6	81.1	53.8	91.9	90.1
797150	2.0	1.33	47	22.1	21.7	95.5	108.4	102.6	109.7	107.4
797150	2.0	1.46	48	39.0	40.0	90.1	95.5	87.6	90.8	93.4
797151	2.0	1.39	49	35.1	35.7	90.1	92.1	86.5	91.4	90.1
797151	2.0	1.37	50	32.5	17.7	95.3	71.4	67.4	91.1	100.1
797151	2.0	1.43	51	17.0	26.0	90.5	109.0	105.6	110.1	112.1
797151	2.0	1.41	52	40.0	36.7	91.1	91.9	84.1	91.7	92.7
797151	2.0	1.44	53	37.8	37.3	91.1	59.6	51.9	90.5	87.4
797151	1.2	1.44	54	43.0	39.3	85.9	53.2	45.6	65.5	66.4
797152	2.0	1.43	55	36.8	35.7	90.0	35.7	50.4	102.8	65.8
797152	2.0	1.27	56	34.2	36.3	90.2	51.1	73.8	121.7	91.7
797152	2.0	1.38	57	33.7	36.3	89.9	26.3	65.4	97.0	5.0
797152	2.0	1.52	58	37.7	36.3	87.4	90.7	79.2	92.7	92.7
797152	2.0	1.47	59	34.4	36.7	90.2	90.5	85.2	99.5	94.5
797152	2.0	1.46	60	35.7	36.0	91.2	97.0	88.3	98.6	96.6
797153	2.0	1.23	61	36.6	36.3	90.2	82.4	73.8	95.5	92.4
797153	2.0	1.33	62	35.7	36.3	91.2	86.5	78.3	98.4	96.5
797153	2.0	1.47	63	36.8	36.3	91.2	94.6	86.4	96.4	96.1
797153	2.0	1.47	64	35.5	37.7	91.2	95.5	86.4	96.7	96.1
797153	2.0	1.47	65	36.0	36.0	91.0	95.0	86.4	97.2	97.1
797153	2.0	1.47	66	35.8	35.7	92.1	94.9	87.4	96.7	96.7
797154	2.0	1.27	67	36.0	36.0	90.0	80.1	74.7	97.2	98.7
797154	2.0	1.27	68	35.5	36.7	91.0	80.5	80.1	97.7	98.1
797154	2.0	1.47	69	35.7	36.7	91.2	97.4	85.3	99.6	99.0
797154	2.0	1.47	70	35.0	36.7	91.2	97.4	85.3	98.6	98.0
797154	2.0	1.45	71	35.7	35.3	90.2	95.7	85.0	96.0	98.1
797154	2.0	1.47	72	35.7	36.3	91.2	97.0	85.3	96.1	97.1
797155	2.0	1.38	73	39.3	36.2	91.2	80.4	71.0	99.1	95.7
797155	2.0	1.48	74	35.4	36.3	91.2	90.6	82.8	98.8	98.9
797155	2.0	1.45	75	35.2	36.3	91.2	96.4	87.4	91.4	90.7
797155	2.0	1.41	76	31.0	36.0	90.0	98.2	87.2	100.4	100.0
797155	2.0	1.47	77	36.7	36.3	90.1	97.4	88.2	100.0	100.0
797155	2.0	1.44	78	36.7	36.3	90.2	96.0	87.3	99.0	96.6
797156	2.0	1.45	79	35.5	36.0	90.2	95.1	87.5	97.5	95.1
797156	2.0	1.44	80	26.3	36.0	90.3	99.5	89.1	100.6	99.2
797156	2.0	1.47	81	36.3	35.7	90.3	98.0	85.0	107.5	100.0
797156	2.0	1.45	82	35.5	35.7	90.0	97.9	88.2	99.0	99.5
797156	2.0	1.47	83	35.4	35.7	91.2	98.6	90.1	100.0	100.0
797157	2.0	1.47	84	36.4	35.7	91.2	96.0	87.7	100.7	100.1
797157	2.0	1.31	85	26.0	35.7	91.2	97.5	87.3	98.0	85.0
797157	1.0	1.35	86	36.1	35.7	91.1	96.7	87.4	97.4	96.1

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C-T. HRS	C.P. E	CYCLE NO.	C.D. MA/CM2	C.D. MA/CM2	VOLT EFF %	COIL EFF %	ENERGY EFF %	2N EFF %	FE EFF %
79/157	2.0	1.33	87	36.0	35.7	91.2	98.6	90.1	100.2	99.6
79/157	2.0	1.48	88	36.0	35.3	90.2	98.6	89.1	100.0	99.8
79/157	2.0	1.45	89	35.7	35.7	91.2	98.8	90.1	100.2	100.2
79/158	2.0	1.25	90	36.0	35.7	90.2	93.1	83.7	98.6	93.1
79/158	2.0	1.27	91	36.2	36.7	90.2	83.7	75.6	101.5	83.0
79/158	2.0	1.42	92	36.0	35.3	91.2	84.4	85.5	98.4	94.4
79/158	2.0	1.31	93	36.0	35.7	90.2	89.6	81.0	98.6	89.6
79/158	2.0	1.47	94	35.7	35.3	91.2	98.6	90.1	100.9	100.9
79/158	2.0	1.45	95	35.4	35.3	91.2	97.6	89.2	99.1	99.1
79/159	2.0	1.47	96	35.6	35.3	90.2	96.7	87.3	98.1	98.1
79/159	2.0	1.44	97	35.3	35.3	89.2	96.2	85.4	97.9	85.6
79/159	2.0	1.46	98	35.2	35.3	90.2	74.9	67.5	96.2	74.9 S
79/159	2.0	1.49	99	35.0	35.7	90.2	98.6	89.1	100.2	100.2
79/159	2.0	1.45	100	34.9	35.2	90.2	98.1	88.2	99.3	99.3
79/159	2.0	1.47	101	35.1	35.3	91.1	97.6	89.2	99.0	99.0
79/160	2.0	1.45	102	35.1	35.3	91.1	97.1	88.3	98.8	95.8
79/160	2.0	1.41	103	35.0	35.3	90.2	90.9	81.0	98.1	89.8
79/160	2.0	1.48	104	34.8	35.3	90.2	70.3	63.0	97.1	70.1 D
79/160	2.0	1.45	105	34.8	35.3	90.2	100.2	90.0	102.2	102.2
79/160	2.0	1.43	106	35.5	35.3	92.1	96.9	89.2	98.1	98.1
79/160	2.0	1.43	107	35.8	35.7	92.1	96.5	88.2	97.7	97.7
79/161	2.0	1.47	108	35.5	35.7	93.2	96.5	86.4	95.1	98.1
79/161	2.0	1.44	109	35.3	35.7	91.1	95.5	86.4	97.0	95.5
79/161	2.0	1.46	110	35.2	35.7	90.2	95.2	85.2	97.0	95.7
79/161	2.0	0.90	111	35.0	35.7	90.2	0.0	0.0	0.0	0.0 S
79/161	2.0	0.80	112	35.1	35.7	0.0	0.0	0.0	0.0	0.0 S
79/161	2.0	1.68	113	35.1	35.7	88.4	228.7	201.5	240.7	3.0 S
79/162	2.0	1.68	114	35.2	35.3	90.1	55.1	49.5	97.4	5.0 D
79/162	2.0	1.66	115	35.1	35.3	90.2	73.4	65.7	97.9	5.0 S
79/162	2.0	1.41	117	35.2	35.3	90.2	95.5	85.5	98.1	5.0
79/162	2.0	1.76	118	34.9	35.7	84.7	96.5	82.4	168.5	6.4
79/162	2.0	1.83	119	35.2	35.3	89.1	87.7	78.3	93.1	87.7
79/163	2.0	1.68	120	35.2	36.0	88.2	80.3	75.7	96.4	86.3
79/163	2.0	1.64	121	35.0	35.0	86.2	90.0	81.2	98.3	90.4
79/163	2.0	1.41	122	35.0	36.0	89.1	96.9	86.3	98.6	9.0
79/163	1.5	1.50	123	36.1	35.3	88.1	60.7	56.2	71.7	6.5 D
79/163	2.0	1.57	124	35.1	35.3	89.1	114.5	101.5	116.8	5.0 S
79/163	2.0	1.56	125	35.2	35.0	89.1	94.1	83.7	96.7	5.0
79/164	2.0	1.62	126	35.0	36.0	89.1	88.5	80.5	97.6	8.0 D
79/164	2.0	1.62	127	35.5	36.0	89.1	86.0	79.4	96.9	90.2
79/171	2.0	1.41	169	35.3	35.7	86.4	80.4	68.8	101.7	80.4 S
79/171	2.0	1.46	170	35.2	35.7	89.1	100.2	89.0	104.5	104.5
79/171	2.0	1.46	171	35.3	35.7	89.1	73.8	69.4	95.1	113.0 S
79/172	2.0	1.38	172	35.4	35.7	89.1	73.5	68.3	98.8	19.7
79/172	2.0	1.36	173	35.2	35.7	89.1	86.1	76.5	98.7	17.2
79/172	2.0	1.42	174	35.3	36.0	88.2	89.9	81.2	99.1	90.1
79/172	2.0	1.44	175	35.1	36.3	88.2	94.5	88.7	100.2	95.8
79/172	2.0	1.46	176	35.2	36.0	88.2	99.1	87.1	100.9	100.9
79/172	2.0	1.46	177	35.2	36.3	89.1	86.0	76.5	99.5	90.8
79/173	2.0	1.36	178	35.3	36.3	88.2	70.4	66.9	97.0	70.7 S
79/173	2.0	1.38	179	35.2	36.0	88.2	73.4	68.6	98.6	73.7 S

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C.T. HRS	C.P. E	CYCLE NO.	+	-	VOLT	COUL	ENERGY	ZN	FE
				C.D. MA/CM2	C.D.	EFF %	EFF %	EFF %	EFF %	EFF %
79/173	2.0	1.46	180	35.1	36.0	88.2	88.1	77.4	99.5	88.4
79/173	2.0	1.46	181	35.2	36.0	88.2	94.5	82.7	99.3	94.8
79/173	2.0	1.46	182	35.2	35.7	88.2	97.2	85.4	99.1	99.1
79/173	2.0	1.46	183	35.2	35.7	89.1	97.2	86.3	98.8	98.3
79/174	2.0	1.38	184	35.2	36.0	89.1	85.6	76.5	99.3	85.6
79/174	2.0	1.44	185	35.3	35.7	88.2	80.9	71.3	99.5	81.8 S
79/174	2.0	1.46	186	35.1	36.0	89.1	86.9	77.4	98.8	87.2
79/174	2.0	1.46	187	34.9	36.7	88.2	98.8	87.1	100.5	99.5
79/174	2.0	1.44	188	35.1	36.3	88.2	97.9	86.2	99.5	99.3
79/174	2.0	1.42	189	35.2	36.0	89.1	96.5	85.4	98.3	97.2
79/175	2.0	1.44	190	35.3	36.3	89.1	76.4	67.6	99.8	76.9 S
79/175	2.0	1.44	191	35.4	36.3	87.3	80.0	69.6	99.3	80.5 S
79/175	2.0	1.46	192	35.5	36.3	89.1	75.4	66.7	96.5	76.1 S
79/175	2.0	1.42	193	35.2	36.3	88.2	97.4	85.4	99.1	98.1
79/175	2.0	1.40	194	35.2	36.7	89.1	99.8	89.0	101.7	101.2
79/175	2.0	1.48	195	35.4	36.0	89.1	82.4	73.0	98.4	82.8 S
79/176	2.0	1.42	196	35.4	36.0	87.2	76.7	67.0	98.4	76.9 S
79/176	2.0	1.42	197	35.2	36.0	87.2	72.8	63.5	98.6	73.5 D
79/176	2.0	1.43	198	35.2	36.3	86.3	85.6	74.0	99.3	85.6 S
79/176	2.0	1.48	199	35.6	35.7	88.2	90.2	79.2	92.5	92.0
79/176	2.0	1.46	200	35.4	35.7	89.1	89.6	80.1	92.5	91.8
79/176	2.0	1.46	201	35.2	36.3	89.1	86.5	76.5	90.1	87.9
79/177	2.0	1.48	202	35.3	36.3	88.2	68.2	59.8	97.4	69.3 D
79/177	2.0	1.47	203	35.3	36.0	86.3	63.0	54.2	102.1	63.0 D
79/177	2.0	1.44	204	35.2	36.0	88.2	83.7	73.9	91.5	85.1 S
79/177	1.3	1.23	205	36.0	1.7	0.0	7.3	0.0	31.0	7.8 *
79/178	2.0	1.40	207	34.7	36.7	89.2	118.2	105.0	124.0	118.5 D
79/178	2.0	1.31	208	35.6	36.3	90.2	76.1	68.4	101.2	75.9 S
79/179	2.0	1.29	209	35.7	36.0	89.2	61.0	54.3	97.7	61.0 D
79/179	2.0	1.33	210	35.7	36.0	88.3	82.1	72.2	92.5	82.1 S
79/179	2.0	1.43	211	35.2	36.0	89.2	94.1	83.7	100.0	94.1
79/179	0.0	0.00	217	1.6	1.7	0.0	100.0	0.0	100.0	00.0 *
79/179	2.0	1.51	222	36.4	36.0	91.2	88.1	80.1	89.2	89.2
79/179	2.0	1.47	223	36.0	36.0	90.2	97.2	87.3	98.4	98.4
79/180	2.0	1.53	224	35.9	35.7	91.1	95.1	86.4	97.0	97.0
79/180	2.0	1.49	225	36.0	35.7	91.2	96.3	87.4	97.7	97.7
79/180	2.0	1.51	226	36.0	36.3	89.2	96.1	85.4	97.7	97.7
79/180	2.0	1.51	227	35.9	35.7	90.3	96.1	86.4	97.9	91.5
79/180	.1	1.49	228	238.3	46.0	89.1	141.3	125.5	144.1	144.1 D
79/181	2.0	1.52	229	35.7	35.7	89.2	82.3	73.0	84.6	84.6 S
79/181	2.0	1.50	230	36.1	36.3	89.3	91.7	81.9	96.3	96.5
79/182	2.0	1.49	231	35.9	36.0	87.6	91.6	81.0	97.0	97.0
79/182	2.0	1.49	232	36.1	36.0	85.7	90.5	77.4	94.0	94.0
79/182	2.0	1.45	233	35.5	33.0	86.7	88.5	76.6	92.7	92.7
79/182	2.0	1.47	234	36.0	19.0	86.7	91.7	80.0	94.2	94.2
79/182	2.0	1.49	235	35.9	34.7	88.4	95.4	83.6	97.4	97.4
79/182	2.0	1.47	236	35.1	35.7	88.4	93.6	82.7	96.0	96.0
79/183	2.0	1.47	237	27.1	25.3	94.6	98.8	94.0	102.8	102.8 S
79/183	2.0	1.31	238	2.4	3.7	100.0	75.9	76.0	96.6	96.6
79/183	2.0	1.31	239	5.7	1.7	0.0	87.0	0.0	88.4	88.4 *
79/183	2.0	1.47	240	35.5	35.3	88.4	95.5	83.6	97.4	97.4

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C.T. HRS	C.P. E	CYCLE NO.	+	-	VOLT	COUL	ENERGY	ZN	FE
				C.D.	C.D.	EFF	EFF	EFF	EFF	EFF
				MA/CM2		%	%	%	%	%
-----	---	---	----	----	----	----	----	----	----	-----
79/183	2.0	1.49	241	35.4	35.0	87.6	95.1	83.6	97.6	97.6
79/183	2.0	1.47	242	36.1	35.0	89.4	94.7	84.5	96.8	96.8
79/184	2.0	1.48	243	35.8	35.3	87.6	96.3	84.5	97.9	97.9
79/184	2.0	1.47	244	36.0	35.3	87.7	94.7	83.6	96.8	96.8
79/184	2.0	1.47	245	35.7	36.0	87.7	96.5	84.5	98.1	98.1
79/184	2.0	1.49	246	36.0	35.3	87.7	95.4	83.6	97.2	97.2
79/184	2.0	1.48	247	35.9	35.7	87.6	94.9	83.6	97.0	97.0
79/184	2.0	1.41	248	36.0	35.7	89.4	98.4	87.2	100.2	100.2
79/185	2.0	1.47	249	36.0	35.3	88.4	96.1	84.5	97.7	97.7
79/185	2.0	1.47	250	35.9	35.7	88.6	97.0	86.3	98.6	98.6
79/185	2.0	1.43	251	35.7	35.3	88.6	97.2	86.3	98.4	98.4
79/185	2.0	1.47	252	35.7	35.3	88.6	96.0	85.4	97.7	97.7
79/185	2.0	1.48	253	35.8	35.3	90.4	97.0	87.3	98.6	98.6
79/185	2.0	1.45	254	35.7	35.7	91.3	97.7	89.2	98.8	98.8
79/186	2.0	1.39	255	35.4	35.7	90.4	96.5	86.4	97.6	97.6
79/186	2.0	1.33	256	35.5	35.3	90.4	82.4	73.8	84.0	84.0 S
79/186	2.0	1.47	257	35.7	35.3	92.2	111.4	102.1	115.2	115.2 S
79/186	2.0	1.47	258	35.3	35.7	91.2	87.3	79.2	93.9	93.9
79/186	2.0	1.47	259	35.8	35.3	90.4	97.9	88.2	99.8	99.8
79/186	2.0	1.43	260	36.0	35.3	91.3	96.8	88.3	98.6	98.6
79/187	2.0	1.47	261	35.7	35.3	88.4	95.8	84.5	98.1	98.1
79/187	2.0	1.46	262	35.7	35.3	89.4	95.8	85.4	98.1	98.1
79/187	2.0	1.46	263	35.5	35.7	88.4	96.7	85.4	98.8	98.8
79/187	2.0	1.47	264	35.8	35.7	89.3	96.5	85.4	98.1	98.1
79/187	2.0	1.47	265	35.8	36.0	91.2	96.3	87.4	98.1	98.1
79/187	2.0	1.47	266	35.8	36.0	91.2	97.2	88.3	98.8	98.8
79/188	2.0	1.48	267	35.7	36.0	90.3	96.0	86.4	98.4	98.4
79/188	2.0	1.46	268	35.6	35.7	89.3	96.3	85.4	98.6	98.6
79/188	2.0	1.46	269	35.5	35.7	88.3	95.5	83.6	97.2	97.2
79/188	2.0	1.43	270	35.6	35.7	90.3	99.8	90.0	101.2	101.2
79/188	2.0	1.45	271	35.8	35.3	93.2	98.1	91.1	99.8	99.8
79/188	2.0	1.47	272	35.8	35.3	90.2	96.7	87.3	97.9	97.9
79/189	2.0	1.47	273	35.8	35.3	90.2	97.4	87.3	99.1	99.1
79/189	2.0	1.45	274	35.7	35.3	90.3	97.4	87.3	98.6	98.6
79/189	2.0	1.45	275	36.8	38.0	106.2	98.2	103.9	99.3	99.3 D
79/189	2.0	1.47	276	35.8	36.0	90.3	98.6	89.1	99.8	99.8
79/189	2.0	1.45	277	35.7	35.7	91.2	99.5	90.1	100.5	100.5
79/189	2.0	1.47	278	34.7	36.0	87.0	45.3	39.1	57.6	57.6 *
79/190	2.0	1.48	279	35.0	35.3	77.1	82.4	63.1	92.1	92.1 D
79/190	2.0	1.48	280	35.0	35.3	79.4	83.3	65.6	94.3	94.3 S
79/190	2.0	1.46	281	35.3	35.3	89.3	93.4	82.8	101.2	101.2
79/190	2.0	1.46	282	36.3	15.7	94.2	100.7	94.9	105.7	105.7 S
79/190	2.0	1.44	283	36.4	36.0	92.2	100.0	92.0	102.5	102.5
79/191	2.0	1.45	284	36.4	36.0	92.1	100.5	92.0	102.1	102.1
79/191	2.0	1.46	285	36.3	36.0	90.3	98.6	89.1	100.7	100.7
79/191	2.0	1.44	286	36.5	35.7	91.2	97.7	89.2	99.1	99.1
79/191	2.0	1.46	287	36.5	36.0	92.1	94.5	86.5	98.6	94.5
79/192	2.0	1.45	288	36.2	36.0	91.1	98.2	89.2	99.5	99.5
79/192	2.0	1.45	289	36.2	35.3	91.2	98.4	89.2	99.5	99.5
79/192	2.0	1.46	290	36.1	35.7	90.3	97.5	87.3	98.8	98.8
79/192	2.0	1.46	291	36.1	35.7	91.2	95.2	86.4	98.6	95.8

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C.T. HRS	C.P. E	CYCLE NO.	+	-	VOLT	COUL	ENERGY	ZN	FE
-----	---	---	----	C.D.	C.D.	EFF	EFF	EFF	EFF	EFF
				MA/CM2		%	%	%	%	%
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79/192	2.0	1.48	292	36.6	36.0	91.2	97.5	88.3	98.9	98.9
79/193	2.0	1.48	294	36.8	35.7	90.3	96.4	86.4	98.4	98.4
79/193	2.0	1.46	295	36.7	36.0	89.3	96.4	85.4	98.6	88.4
79/193	2.0	1.46	296	36.5	36.0	88.3	90.4	79.2	93.4	91.3
79/193	2.0	1.46	297	36.6	35.7	91.2	97.3	88.3	99.8	98.9
79/193	2.0	1.44	298	36.7	36.0	91.1	97.3	88.3	99.1	99.1
79/194	2.0	1.48	301	36.5	36.0	89.2	96.1	85.4	99.1	95.7
79/194	2.0	1.44	302	36.3	36.0	87.2	89.0	77.4	92.7	87.4
79/195	2.0	1.48	305	34.8	36.7	90.4	201.7	181.8	206.5	206.2 *
79/195	2.0	1.46	306	37.2	36.0	92.1	96.9	89.2	98.9	97.3
79/195	2.0	1.46	307	37.1	36.3	92.1	97.1	89.2	98.7	98.7
79/195	2.0	1.48	308	37.2	36.7	92.1	93.1	85.6	96.6	79.9
79/195	2.0	1.48	309	37.2	36.7	91.0	92.2	83.7	94.6	92.4
79/195	2.0	1.48	310	37.3	36.3	92.1	96.2	88.3	97.5	97.5
79/197	2.0	1.47	312	37.0	35.7	91.5	80.9	73.7	86.9	79.5 S
79/197	2.0	1.45	313	37.1	36.3	92.1	91.5	83.7	96.9	89.0
79/197	2.0	1.47	314	37.2	36.7	91.1	90.8	82.8	95.5	89.0
79/201	2.0	1.48	316	36.3	35.7	91.5	84.2	76.4	101.6	4.8
79/201	2.0	1.47	317	35.7	36.0	90.4	80.1	72.0	99.5	6.1 S
79/201	2.0	1.47	318	36.0	35.7	91.0	78.2	71.0	98.3	76.4 S
79/201	2.0	1.47	319	36.0	35.3	92.5	88.9	81.9	99.1	86.6
79/201	2.0	1.47	320	36.2	35.3	91.5	90.8	82.8	97.2	91.5
79/202	2.0	1.46	321	36.4	35.3	91.5	92.0	83.7	94.5	94.5
79/202	2.0	1.46	322	36.3	35.7	91.0	97.2	88.3	101.4	101.4
79/202	2.0	1.49	323	36.1	36.0	91.5	84.3	76.4	96.8	85.5
79/202	2.0	1.45	324	36.1	35.3	91.5	93.8	85.5	99.1	94.0
79/202	2.0	1.48	325	36.4	32.7	92.6	96.3	89.3	99.1	98.4
79/202	2.0	1.40	326	36.7	33.0	92.6	98.4	91.1	99.8	99.8
79/203	2.0	1.31	327	36.7	33.0	93.5	93.2	86.5	98.0	93.2
79/203	2.0	1.43	328	36.8	32.0	92.6	91.4	84.6	95.9	91.6
79/203	2.0	1.46	329	36.7	32.0	93.6	96.4	90.2	99.1	98.2
79/203	2.0	1.42	330	36.3	32.0	93.6	95.9	90.2	97.2	97.2
79/203	2.0	1.45	331	36.2	32.0	92.6	89.2	82.8	95.9	89.6
79/203	2.0	1.47	332	36.1	32.3	92.6	94.9	88.3	98.4	96.1
79/204	2.0	1.48	333	36.1	32.3	92.1	80.1	73.6	84.5	82.2 S
79/204	2.0	1.46	334	35.6	32.3	82.2	31.1	25.4	109.4	34.9 *
79/204	2.0	1.33	335	36.2	32.7	86.6	77.0	67.0	100.2	77.0 S
79/204	2.0	1.47	336	37.1	32.0	91.1	73.7	67.3	95.5	75.3 S
79/204	2.0	1.41	337	36.9	32.0	89.1	72.2	64.1	97.5	72.2 D
79/205	2.0	1.47	338	36.7	32.3	89.2	66.2	58.7	97.5	66.9 D
79/285	2.0	1.47	340	36.5	32.7	87.3	62.1	53.9	99.3	62.6 D
79/205	2.0	1.46	341	36.7	33.0	89.1	82.1	73.0	93.2	87.8 S
79/205	2.0	1.50	342	36.8	32.7	90.1	83.9	75.6	92.3	90.7
79/205	2.0	1.46	343	37.2	32.3	91.0	83.0	75.5	92.6	86.4
79/206	2.0	1.46	344	37.2	33.0	91.0	87.2	79.2	94.2	91.3
79/206	2.0	1.47	345	37.0	33.0	91.0	89.9	81.9	97.1	91.2
79/206	2.0	1.48	346	37.1	33.0	92.1	91.7	84.6	95.3	95.3
79/206	2.0	1.46	347	37.2	33.3	92.1	93.7	86.5	97.1	97.1
79/206	2.0	1.48	348	37.2	33.0	92.1	93.9	86.5	96.9	96.9
79/207	2.0	1.46	349	37.0	33.0	92.6	92.1	85.6	96.6	93.5
79/207	2.0	1.48	350	37.1	33.0	92.6	92.6	86.5	95.5	93.5

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C.T. HRS	C.P. E	CYCLE NO.	+	-	VOLT	COUL	ENERGY	ZN	FE
				C.D. MA/CM2	C.D.	EFF %	EFF %	EFF %	EFF %	EFF %
79/207	2.0	1.42	351	37.0	32.7	92.1	94.8	87.4	96.8	96.8
79/207	2.0	1.44	352	37.2	33.0	92.6	95.1	88.3	96.9	96.9
79/207	2.0	1.46	353	36.8	33.3	92.1	97.5	89.2	99.5	99.3
79/207	2.0	1.46	354	36.7	33.0	92.6	98.2	91.1	99.8	99.8
79/208	2.0	1.46	355	36.8	33.3	92.6	94.8	88.3	97.3	97.1
79/208	2.0	1.47	356	36.8	33.7	92.1	82.4	75.4	95.7	82.6
79/208	2.0	1.42	357	36.5	33.7	93.1	94.7	88.3	98.2	97.5
79/208	2.0	1.46	358	35.7	33.3	92.1	92.8	85.6	97.2	93.9
79/208	2.0	1.39	359	35.9	33.3	93.6	96.3	90.2	98.4	96.3
79/208	2.0	1.45	360	35.7	33.0	92.6	91.8	85.6	95.8	92.1
79/209	2.0	1.49	361	35.9	33.3	92.6	84.9	79.0	93.3	85.2
79/209	2.0	1.43	362	35.6	32.7	93.1	74.0	68.8	92.0	74.5 S
79/209	2.0	1.47	363	35.9	33.0	92.1	86.3	79.1	93.5	87.5
79/209	2.0	1.44	364	35.0	33.0	93.1	97.9	91.1	101.0	101.0
79/209	2.0	1.46	365	37.2	33.0	92.6	95.7	89.3	98.0	97.5
79/210	2.0	1.45	366	36.9	33.3	92.6	91.6	85.6	96.8	92.1
79/210	2.0	1.47	367	36.9	32.3	92.1	87.4	80.0	94.1	89.2
79/210	2.0	1.47	368	36.9	33.0	92.1	79.0	72.7	94.4	79.9 S
79/210	2.0	1.47	369	36.4	32.7	92.1	70.9	65.3	95.4	71.2 D
79/210	2.0	1.46	370	36.5	32.7	93.1	97.7	91.1	103.7	21.0
79/210	2.0	1.46	371	35.4	33.0	93.1	97.9	91.1	99.8	37.2
79/211	2.0	1.60	372	35.9	32.7	93.6	92.1	86.5	95.8	4.4
79/211	2.0	1.46	373	35.5	33.0	93.1	91.5	84.6	94.8	12.2
79/211	2.0	0.00	374	35.4	1.7	0.0	0.0	0.0	21.6	4.5 *
79/211	.6	1.48	375	35.5	33.0	92.1	345.3	317.4	356.3	356.3 *
79/211	2.0	1.44	376	36.6	33.3	92.1	93.6	86.5	97.5	94.8
79/211	2.0	1.31	377	36.0	34.7	93.1	28.5	26.0	33.8	28.5 *
79/292	2.0	1.52	378	36.4	35.7	93.2	155.6	145.1	159.5	139.8 *
79/212	1.6	.92	379	39.0	1.7	0.0	5.6	0.0	5.9	5.6 *
79/213	.9	1.87	380	37.2	1.7	0.0	10.4	8.0	23.9	10.4 *
79/213	2.0	1.46	381	35.7	36.3	91.3	188.1	971.1	198.1	189.5 *
79/213	2.8	1.46	382	35.5	37.0	92.6	96.4	87.4	98.8	98.8
79/214	2.0	1.46	383	35.5	37.8	92.6	93.4	86.5	97.7	97.7
79/214	2.0	1.47	384	35.5	36.7	92.6	84.3	78.1	114.1	85.0
79/214	2.0	1.48	386	35.7	37.0	91.1	93.5	84.6	98.4	98.4
79/214	2.0	1.46	387	35.7	37.0	92.1	98.4	90.2	102.1	102.1
79/214	2.0	1.44	388	34.9	37.0	92.1	95.0	87.4	98.1	98.1
79/215	2.0	1.46	389	35.6	37.7	92.9	96.0	88.3	98.8	98.4
79/293	2.0	1.45	390	35.6	37.8	93.1	94.6	87.4	100.0	96.0
79/215	2.0	1.47	391	35.7	37.7	92.1	87.6	81.0	97.4	90.0
79/215	2.8	1.45	392	35.4	37.7	92.1	93.4	85.6	99.8	94.6
79/215	2.0	1.44	393	36.1	38.0	93.1	99.5	92.9	102.5	102.3
79/215	2.0	1.44	394	35.6	36.7	92.1	98.6	99.1	180.7	100.5 S
79/216	2.0	1.48	395	35.7	37.3	92.1	98.4	98.2	100.5	100.2 S
79/216	2.0	1.44	396	35.2	37.0	92.1	95.7	88.3	98.8	98.8
79/216	2.0	1.31	397	34.4	37.3	92.6	97.3	90.2	100.5	97.3
79/216	2.0	1.42	398	34.7	37.0	92.6	99.3	92.1	100.7	100.7
79/216	2.0	1.39	399	36.0	36.7	92.6	98.1	91.1	100.2	99.3
79/216	2.0	1.44	400	36.2	37.0	92.6	96.3	89.3	98.8	98.8
79/217	2.0	1.43	401	35.7	37.0	92.1	97.0	89.2	99.3	97.4
79/217	2.0	1.45	402	35.7	36.7	92.1	95.1	87.4	99.5	97.0

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C.T. HRS	C.P. E	CYCLE NO.	+	-	VOLT	CORR.	ENERGY	EN	ED
				C.D. MA/CM2	C.D.	EFF %	EFF %	EFF %	MPF %	EFF %
79/229	2.0	2.42	471	37.1	35.3	91.3	89.6	81.3	0.0	6.7
79/231	2.0	1.44	481	16.2	27.0	94.4	95.2	90.2	104.1	94.3
79/231	2.0	1.46	482	36.7	34.0	91.5	96.4	87.4	100.7	87.8
79/231	2.0	1.45	483	37.1	36.3	91.5	94.8	86.4	99.8	93.9
79/231	2.0	1.47	484	36.7	36.0	91.5	91.1	82.8	0.0	55.2
79/232	2.0	1.46	487	36.9	36.0	91.0	95.9	87.4	104.5	94.6
79/232	2.0	1.48	488	36.5	35.0	89.9	94.7	84.6	98.4	94.3
79/232	2.0	1.45	489	36.9	35.3	92.6	87.8	81.8	96.6	86.5
79/233	2.0	1.43	494	36.6	37.0	91.1	94.8	86.4	97.7	87.7
79/234	2.0	1.46	501	36.2	34.3	86.7	119.7	100.0	0.0	117.7 S
79/234	2.0	1.37	502	36.3	34.0	91.1	93.6	89.2	0.0	46.3
79/235	2.0	1.49	504	36.2	41.0	90.1	95.2	85.3	100.7	43.6
79/235	2.0	1.43	505	36.4	41.7	91.5	95.1	77.2	93.4	84.3
79/235	2.0	1.45	506	36.3	31.7	92.6	103.4	95.8	104.4	104.4 S
79/236	2.0	1.46	507	36.4	31.3	92.6	91.3	84.6	97.3	92.2
79/236	2.0	1.44	508	36.3	31.3	92.6	88.1	81.8	91.5	91.5
79/236	2.0	1.47	509	36.3	30.7	93.1	86.9	80.9	0.0	85.1
79/236	2.0	1.46	510	36.4	30.0	93.1	102.1	94.9	104.1	104.1 S
79/236	2.0	1.46	511	36.4	29.0	93.6	97.9	92.1	98.9	98.9
79/236	2.0	1.44	512	36.5	29.3	92.8	98.2	91.1	99.3	97.9
79/237	2.0	1.47	513	36.4	27.7	92.6	94.1	87.4	99.3	93.1
79/237	2.0	1.48	514	36.2	28.3	91.1	91.0	82.8	93.5	92.9
79/237	2.0	1.46	515	36.2	29.3	91.1	94.0	85.5	98.4	94.9
79/237	2.0	1.46	516	36.2	29.7	92.1	103.2	94.8	106.5	106.5 S
79/238	2.0	1.35	522	36.0	28.0	92.1	96.8	89.2	0.0	96.5
79/238	2.0	1.46	523	36.3	34.0	91.1	97.2	88.3	101.4	96.8
79/238	2.0	1.46	524	36.5	35.7	92.1	98.9	91.1	100.7	99.1
79/238	2.0	1.30	525	36.5	34.3	93.1	97.0	90.2	97.9	97.0
79/239	2.0	1.42	526	36.5	39.0	92.6	97.0	90.2	99.1	96.6
79/239	2.0	1.48	527	36.5	40.0	92.1	96.8	89.2	97.9	96.8
79/239	2.0	1.48	528	36.3	38.3	91.1	94.5	85.5	97.2	93.1
79/239	2.0	1.47	529	36.4	34.3	92.1	97.5	89.2	100.5	96.6
79/239	2.0	1.44	530	36.6	32.3	93.1	97.3	90.2	99.1	95.7
79/239	2.0	1.46	531	36.7	39.3	91.0	96.1	87.4	98.2	96.1
79/240	2.0	1.48	532	36.7	39.0	91.0	96.4	87.4	97.7	96.6
79/240	2.0	1.46	533	36.6	39.0	91.0	96.6	88.3	97.7	96.8
79/240	2.0	1.46	534	36.6	38.7	91.0	97.0	88.3	97.7	97.7
79/240	2.0	1.47	535	36.7	37.0	92.1	92.3	84.6	0.0	91.3
79/240	2.0	1.42	536	36.8	32.7	92.1	97.5	89.2	98.9	98.0
79/240	2.0	1.46	537	36.8	29.0	92.1	97.7	90.2	98.6	98.6
79/241	2.0	1.46	538	36.7	33.3	92.6	95.7	89.3	0.0	93.7
79/241	2.0	1.45	539	36.7	33.7	92.1	96.2	79.1	0.0	89.1
79/241	2.0	1.46	540	36.7	33.7	91.1	98.6	90.1	100.0	98.0
79/241	2.0	1.41	541	36.8	37.7	91.0	98.7	87.4	97.7	95.2
79/241	2.0	1.46	542	37.1	38.0	91.0	91.7	83.7	97.3	91.3
79/241	2.0	1.46	543	37.1	39.0	91.0	96.2	87.4	97.3	97.3
79/242	2.0	1.46	544	36.8	38.7	89.1	97.1	86.3	99.8	98.0
79/242	2.0	1.46	545	36.7	34.3	92.1	86.8	80.0	0.0	83.7
79/242	2.0	1.44	546	36.9	34.7	90.2	96.4	86.4	0.0	91.2
79/242	2.0	1.47	547	36.9	35.7	91.0	93.0	86.4	98.9	94.1
79/242	2.0	1.42	548	37.4	39.7	91.5	96.4	87.2	97.3	97.2

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATA	C.P. HRS	C.P. E	COIL NO.	C.D. C.D.	C.D. C.D.	VOLT EFF	COIL EFF	ENRGEY EFF	EN EFF	CE EFF
-----	---	-----	-----	-----	-----	-----	-----	-----	-----	-----
79/243	2.0	1.44	549	37.3	41.3	91.5	96.0	87.4	96.9	96.9
79/243	2.0	1.44	550	36.7	40.3	91.0	97.3	88.3	98.4	98.4
79/243	2.0	1.46	551	36.7	37.3	92.1	95.0	87.4	97.7	96.8
79/243	2.0	1.42	552	36.5	38.7	91.5	97.5	88.3	98.6	98.6
79/243	2.0	1.44	553	37.2	40.3	91.5	95.7	87.4	98.6	98.6
79/243	2.0	1.40	554	37.4	38.7	91.5	95.5	86.0	98.0	96.0
79/244	2.0	1.48	555	37.3	42.0	91.0	95.3	86.0	98.7	98.2
79/244	2.0	1.31	556	36.8	40.7	91.5	98.0	81.7	98.4	98.9
79/244	2.0	1.46	557	36.6	34.7	92.6	98.7	96.1	100.7	99.5
79/244	2.0	1.46	558	36.8	35.3	92.6	99.5	92.1	101.1	101.1
79/245	2.0	1.42	559	36.4	36.0	92.1	100.0	98.0	100.7	100.7
79/245	2.0	1.38	560	36.4	36.0	92.1	99.1	91.1	100.2	100.2
79/245	2.0	1.46	561	36.7	36.0	92.1	96.6	89.2	98.2	98.0
79/245	2.0	1.46	562	36.4	38.0	91.5	102.5	92.3	104.5	104.0
79/245	2.0	1.46	563	36.5	37.3	92.6	99.3	92.1	102.9	101.1
79/245	2.0	1.45	564	37.2	37.7	91.0	100.7	91.0	102.5	103.0
79/246	2.0	1.49	565	36.5	37.3	90.1	99.3	89.1	103.0	100.0
79/246	2.0	1.49	566	36.9	37.7	90.1	98.0	88.2	100.7	98.4
79/246	2.0	1.44	567	37.2	37.7	91.0	98.4	89.2	103.0	99.8
79/246	2.0	1.45	568	37.3	37.0	91.0	96.4	87.4	99.1	96.7
79/246	2.0	1.39	569	37.2	38.7	91.0	99.1	90.1	100.7	99.6
79/246	2.0	1.44	570	37.2	38.7	91.5	101.3	91.9	102.7	102.5
79/247	2.0	1.43	571	37.0	38.7	91.5	101.6	92.5	102.9	103.0
79/247	2.0	1.45	572	36.8	38.7	91.0	97.1	88.3	100.0	97.5
79/247	2.0	1.47	573	36.8	38.7	91.0	100.5	91.0	103.3	98.4
79/247	2.0	1.64	574	37.4	38.7	90.1	100.0	90.0	101.6	98.1
79/248	2.0	1.82	575	36.7	37.0	92.1	95.7	88.3	106.1	93.0
79/249	2.0	1.46	580	36.8	36.7	90.4	86.2	77.4	93.0	96.0
79/249	2.0	1.48	581	36.7	36.7	90.4	94.8	85.5	98.0	96.8
79/249	2.0	1.46	582	36.7	37.0	91.5	88.6	81.9	98.1	95.5
79/249	2.0	1.46	583	34.8	36.7	91.5	111.0	101.0	112.0	110.0 S
79/250	2.0	1.44	589	35.9	36.7	89.9	92.3	82.8	103.0	94.0
79/250	2.0	1.47	590	35.9	36.0	90.4	87.5	78.3	98.4	90.0
79/251	2.0	1.46	591	35.6	36.7	88.9	87.7	78.3	98.5	90.0
79/251	2.0	1.44	592	36.2	39.0	88.9	87.3	77.4	98.5	96.0
79/251	2.0	1.47	593	36.9	38.3	88.9	84.2	74.8	98.3	94.0
79/251	2.0	1.44	594	36.7	38.3	90.4	90.4	81.0	99.0	90.0
79/251	2.0	1.45	595	36.9	38.3	91.5	92.1	80.7	100.7	98.3
79/251	2.0	1.31	596	37.3	38.3	90.4	94.2	84.6	98.4	94.2
79/251	2.0	1.44	597	37.2	38.3	90.4	90.6	81.9	99.3	91.8
79/252	2.0	1.46	600	37.1	37.0	91.0	97.3	89.2	99.3	99.0
79/252	2.0	1.48	601	37.3	37.0	92.6	114.1	106.0	117.0	117.0 D
79/252	2.0	1.43	602	37.2	36.3	91.5	97.1	85.3	92.5	97.3
79/253	2.0	1.43	603	37.5	36.7	91.5	95.3	86.4	98.7	95.0
79/253	2.0	1.41	604	37.1	36.7	91.5	89.2	81.0	94.7	90.2
79/253	2.0	1.46	606	37.3	36.7	91.5	96.8	88.3	97.1	97.7
79/253	2.0	1.46	607	36.7	36.3	92.6	98.0	91.1	98.6	98.5
79/253	2.0	1.46	608	36.7	36.7	92.6	98.9	92.1	99.5	99.5
79/254	2.0	1.47	609	37.2	36.7	92.6	97.5	90.2	98.2	99.0
79/254	2.0	1.45	610	37.3	36.7	91.5	96.4	89.2	98.7	98.7
79/254	2.0	1.47	611	37.2	36.7	92.6	98.6	92.1	99.1	99.3

DATE	C.T. HRS	C.P. Σ	CYCLE NO.	+	-	VOLT EFF	COOL EFF	ENERGY EFF	ZN EFF	FE EFF
-----	----	-----	-----	C.D. MA/CM2	C.D.	%	%	%	%	%
79/254	2.0	1.46	612	37.0	37.0	90.5	90.9	91.1	98.6	98.6
79/254	2.0	1.48	613	37.1	37.0	90.6	97.3	90.8	97.8	97.8
79/254	2.0	1.44	614	37.2	36.7	92.1	97.2	99.2	97.8	97.8
79/255	2.0	1.46	615	36.7	36.7	92.6	99.0	91.1	99.1	99.1
79/255	2.0	1.46	616	36.7	37.0	92.5	98.9	99.3	98.6	97.3
79/255	2.0	1.43	617	36.9	37.3	91.5	95.1	99.5	98.6	97.3
79/255	2.0	1.47	618	27.7	0.0	92.7	97.6	92.2	100.0	97.5
79/255	2.0	1.46	619	36.5	37.0	91.5	94.0	87.4	97.5	97.3
79/256	2.0	1.40	620	36.3	37.7	91.5	98.1	88.3	97.5	97.5
79/256	2.0	1.48	621	36.2	36.0	91.6	96.7	90.2	98.4	96.8
79/256	2.0	1.42	622	36.2	35.7	98.6	91.1	86.5	98.4	93.8
79/256	2.0	1.46	623	36.0	38.2	91.5	96.1	87.4	98.1	98.1
79/256	2.0	1.44	624	35.4	37.8	91.5	95.5	86.4	96.7	96.7
79/256	2.0	1.40	625	35.6	36.9	91.5	97.1	88.3	97.9	97.9
79/257	2.0	1.34	626	35.6	38.3	91.5	97.2	88.3	97.7	97.7
79/257	2.0	1.30	627	35.7	38.7	91.5	99.1	99.9	98.6	98.8
79/257	2.0	1.44	628	36.4	38.7	91.5	97.1	89.2	98.4	98.4
79/257	2.0	1.44	629	35.9	38.7	91.5	94.9	88.4	98.1	98.1
79/257	2.0	1.46	630	36.0	38.7	91.0	95.4	86.4	97.7	97.7
79/257	2.0	1.48	631	36.2	38.0	91.5	95.1	87.4	97.7	97.7
79/257	2.0	1.46	632	36.2	40.0	91.0	97.0	86.1	97.7	97.7
79/258	2.0	1.44	633	36.2	39.3	91.0	98.1	87.4	96.8	96.8
79/258	2.0	1.46	634	36.2	39.0	91.5	99.1	99.1	100.9	100.9
79/258	2.0	1.48	635	36.1	39.0	90.4	94.9	85.5	98.2	98.2
79/258	2.0	1.48	636	36.2	39.0	91.5	100.0	91.9	102.5	102.5
79/258	2.0	1.46	637	36.2	39.6	91.5	97.8	88.3	98.5	98.6
79/258	2.0	1.48	638	36.2	39.4	91.0	99.0	88.5	98.2	98.2
79/259	2.0	1.48	639	36.0	38.7	90.4	96.7	85.5	96.8	96.8
79/259	2.0	1.46	640	36.0	39.3	91.0	97.0	88.3	98.1	98.1
79/259	2.0	1.44	641	35.9	40.8	91.5	95.8	86.4	96.5	96.5
79/259	2.0	1.46	642	36.0	40.1	91.5	98.8	89.2	100.2	100.2
79/259	2.0	1.46	643	36.3	40.0	91.0	97.1	88.3	99.1	99.1
79/260	2.0	1.48	651	36.3	40.4	92.6	97.1	90.2	100.0	100.0
79/260	2.0	1.44	652	36.3	40.4	91.5	98.8	90.1	100.5	100.5
79/261	2.0	1.44	653	36.2	40.2	91.5	98.7	92.1	99.8	99.8
79/261	2.0	1.42	654	36.3	40.1	91.5	98.7	89.2	99.3	99.3
79/261	2.0	1.46	655	36.3	39.4	94.6	96.8	90.2	97.7	97.7
79/263	2.0	1.44	668	36.4	40.2	91.9	98.0	88.7	93.8	93.8
79/264	2.0	1.46	673	36.2	39.8	91.0	100.0	91.0	109.9	181.4
79/264	2.0	1.48	674	36.2	39.3	91.0	95.1	87.4	97.0	97.5
79/264	2.0	1.45	675	36.2	39.1	91.0	83.8	75.5	102.5	83.4
79/265	2.0	1.33	676	36.2	40.1	91.0	78.9	66.4	99.5	72.9 S
79/265	2.0	1.41	677	36.2	40.0	92.1	78.6	72.7	99.3	78.6 S
79/265	2.0	1.46	678	36.2	39.9	92.1	82.3	75.4	99.3	82.7
79/265	2.0	1.45	679	36.2	39.6	92.1	97.5	89.2	98.6	98.6
79/265	2.0	1.46	680	36.2	39.6	92.1	96.3	88.3	98.6	98.6
79/265	2.0	1.43	681	36.3	39.7	92.1	87.4	80.0	98.4	87.6
79/267	2.0	1.48	691	36.3	37.8	93.1	102.8	95.8	109.2	103.7 S
79/267	2.0	1.82	692	36.3	35.9	91.1	96.3	87.4	89.9	96.3
79/267	2.0	1.46	693	36.4	37.8	92.1	88.1	81.0	101.7	88.6
79/268	2.0	1.46	694	36.4	39.8	92.1	81.0	74.5	97.9	81.7 S

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

DATE	C.T. HRS	C.P. E	CYCLE NO.	+	-	VOLT	COUL	ENERGY	ZN	FE
				C.D.	C.D.	EFF	EFF	EFF	EFF	EFF
				MA/CM2		%	%	%	%	%
-----	---	----	----	-----	-----	-----	-----	-----	-----	-----
79/268	2.0	1.42	695	36.4	38.1	92.1	87.9	81.0	94.1	88.3
79/268	2.0	1.44	696	36.3	37.4	93.1	99.8	93.0	103.0	100.2
79/268	2.0	1.46	697	36.5	36.6	92.1	94.7	87.4	180.2	95.2
79/268	2.0	1.46	698	36.4	35.6	92.1	100.0	92.0	182.7	102.7
STATISTICAL MEAN				36.3	36.0	91.0	92.3	83.9		
STANDARD DEVIATION				9.3	3.5	2.0	9.5	9.2		

CELL UNDER TEST (6)

CH #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP OCV	CP 30 MIN	ZP 30 MIN	FP	CP 30 MIN	ZP USABLE	FP	CP ZN	CP OCV
1	1.88	1.45	.43	1.88	1.82	1.74	1.31	.43	1.41	1.27	.14	.80	1.62
2	1.89	1.46	.43	1.91	1.80	1.72	1.29	.43	1.46	1.23	.23	.96	1.74
3	1.88	1.45	.43	1.91	1.82	1.76	1.33	.43	1.54	1.25	.29	.80	1.74
4	1.89	1.46	.43	1.91	1.82	1.78	1.33	.45	1.33	1.33	0.00	.76	.37
5	2.33	1.88	.45	2.32	1.84	1.54	1.54	0.00	1.54	1.54	0.00	0.00	.63
6	2.24	1.48	.76	2.19	1.91	1.25	1.25	0.00	1.25	1.25	0.00	.74	.18
7	1.80	1.41	.39	1.89	1.80	1.74	1.31	.43	1.47	1.43	.04	.80	1.56
8	1.89	1.46	.43	1.89	1.80	1.76	1.31	.45	1.31	1.21	.10	.80	1.74
9	1.90	1.45	.45	1.89	1.82	1.74	1.31	.43	1.29	1.29	0.00	.74	1.64
10	1.90	1.45	.45	1.89	1.80	1.74	1.31	.43	1.46	1.25	.21	.78	1.70
11	1.92	1.43	.55	2.91	1.80	1.70	1.25	.45	1.50	1.27	.23	.80	1.76
12	1.92	1.46	.47	1.91	1.82	1.74	1.29	.45	1.45	1.27	.18	.80	1.72
13	1.97	1.46	.51	1.89	1.80	1.72	1.31	.41	1.45	1.25	.20	.76	1.74
14	1.92	1.46	.47	1.89	1.82	1.74	1.31	.43	1.44	1.23	.21	.80	1.72
15	1.90	1.45	.45	1.89	1.82	1.76	1.35	.41	1.56	1.29	.27	.80	1.70
16	1.93	1.46	.47	1.91	1.82	1.76	1.33	.43	1.46	1.25	.21	.80	1.70
17	2.11	1.48	.63	1.91	1.82	1.72	1.31	.41	1.51	1.33	.18	.76	1.76
18	1.94	1.45	.49	1.89	1.82	1.74	1.31	.43	1.46	1.21	.25	.80	1.72
19	1.91	1.48	.43	1.91	1.82	1.78	1.35	.43	2.42	2.42	0.00	0.00	1.11
20	1.90	1.86	.04	1.88	1.80	1.70	1.37	.33	1.41	1.02	.39	1.17	1.35
21	1.84	1.45	.43	1.88	1.80	1.68	1.33	.35	1.42	1.05	.37	1.17	1.35
22	1.35	1.43	.43	1.86	1.80	1.72	1.33	.39	1.44	1.19	.25	.76	1.70
23	1.98	1.45	.47	1.88	1.80	1.70	1.31	.39	1.47	1.31	.16	.80	1.74
24	1.88	1.45	.43	1.89	1.82	1.72	1.31	.41	1.43	1.23	.20	.76	1.72
25	1.95	1.45	.51	1.88	1.80	1.74	1.31	.43	1.44	1.11	.33	.88	1.74
26	1.88	1.43	.45	1.91	1.82	1.72	1.31	.41	1.46	1.19	.27	.78	1.74
27	1.92	1.45	.47	2.66	1.72	1.66	1.27	.39	1.40	1.05	.35	1.05	1.37
28	1.93	1.46	.47	2.79	1.80	1.70	1.31	.39	1.35	1.21	.14	.76	1.72
29	1.90	1.45	.45	1.91	1.82	1.74	1.33	.41	1.29	1.29	0.00	.72	1.74
30	1.99	1.46	.53	1.93	1.82	1.68	1.29	.39	1.43	1.27	.16	.80	.68
31	1.97	1.46	.51	1.93	1.82	1.76	1.31	.45	1.46	1.25	.21	.80	.55
32	1.94	1.45	.49	1.97	1.84	1.74	1.31	.43	1.35	1.25	.10	.80	.76
33	1.94	1.45	.49	1.99	1.86	1.74	1.31	.43	1.43	1.00	.43	1.21	1.64
34	1.92	1.43	.49	1.97	1.84	1.76	1.33	.43	1.46	1.05	.41	1.19	1.37
35	1.92	1.45	.47	1.97	1.84	1.74	1.31	.43	1.44	1.05	.39	1.17	1.37
36	1.94	1.45	.49	1.97	1.84	1.76	1.33	.43	1.43	1.25	.18	.78	.64
37	1.92	1.41	.51	1.97	1.86	1.76	1.31	.45	1.27	1.27	0.00	.80	.68
38	1.94	1.43	.51	1.97	1.86	1.74	1.29	.45	1.48	1.23	.25	.80	1.35
39	1.94	1.45	.49	1.97	1.86	1.76	1.31	.45	1.43	1.02	.41	1.15	1.56
40	1.98	1.45	.47	1.97	1.86	1.78	1.33	.45	1.41	1.00	.41	1.19	1.46
41	1.96	1.43	.53	1.97	1.86	1.78	1.33	.45	1.47	1.04	.43	1.19	1.39
42	1.92	1.35	.43	1.93	1.82	1.72	1.29	.43	1.35	.92	.43	1.19	1.39
43	1.94	1.45	.49	1.97	1.86	1.74	1.31	.43	1.44	1.21	.23	.74	.82
44	1.94	1.45	.49	1.97	1.86	1.76	1.31	.45	1.44	1.21	.23	.80	1.76
45	1.86	1.39	.47	1.91	1.84	1.76	1.31	.45	1.47	1.02	.45	1.13	1.41
46	1.94	1.45	.49	1.82	1.82	1.82	1.37	.45	1.33	.90	.43	1.23	1.37
47	1.92	1.41	.49	1.91	1.84	1.76	1.35	.41	1.48	1.05	.43	1.11	1.41
48	1.85	1.39	.47	1.93	1.82	1.72	1.31	.41	1.39	.94	.45	1.05	1.37
49	1.92	1.43	.49	1.93	1.84	1.86	1.39	.47	1.37	1.27	.10	.78	.61

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

CELL UNDER TEST (6)

CN #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP OCV	CP 30 MIN	ZP 30 MIN	FP	CP 30 MIN	ZP 30 MIN	FP	CP 30 MIN	CP OCV
51	1.84	1.39	.45	1.91	1.82	1.70	1.27	.43	1.43	1.04	.39	1.11	1.78
52	1.94	1.45	.49	1.97	1.86	1.76	1.33	.43	1.41	1.00	.41	1.11	1.41
53	1.92	1.45	.47	1.95	1.84	1.74	1.33	.41	1.44	1.21	.23	.76	.42
54	1.94	1.45	.49	2.52	1.84	1.70	1.31	.39	1.44	1.17	.27	.80	1.78
55	1.94	1.45	.49	1.99	1.86	1.74	1.31	.43	1.43	1.23	.20	.78	.41
56	1.96	1.45	.51	1.97	1.88	1.78	1.33	.45	1.27	1.27	0.00	0.00	0.00
57	1.97	1.89	.08	1.97	1.86	1.76	1.56	.20	1.38	1.05	.33	1.03	1.78
58	1.97	1.54	.43	1.97	1.86	1.76	1.33	.43	1.52	1.05	.47	1.17	1.76
59	1.95	1.46	.49	1.95	1.86	1.76	1.33	.43	1.47	1.04	.43	1.15	1.41
60	1.95	1.46	.49	1.95	1.86	1.78	1.33	.45	1.46	1.05	.41	1.17	1.43
61	1.97	1.46	.51	1.97	1.88	1.76	1.33	.43	1.23	1.23	0.00	.80	.43
62	1.95	1.46	.49	1.97	1.88	1.78	1.33	.45	1.33	1.15	.18	.80	.41
63	1.94	1.45	.49	1.97	1.88	1.78	1.33	.45	1.47	1.04	.43	1.13	1.43
64	1.95	1.46	.49	1.97	1.88	1.78	1.33	.45	1.47	1.04	.43	1.19	1.72
65	1.94	1.45	.49	1.95	1.86	1.78	1.33	.45	1.47	1.04	.43	1.23	1.66
66	1.94	1.45	.49	1.95	1.86	1.78	1.33	.45	1.47	1.04	.43	1.17	1.43
67	1.94	1.45	.49	1.97	1.88	1.76	1.31	.45	1.27	1.27	0.00	.80	.38
68	1.96	1.45	.51	1.97	1.88	1.78	1.33	.45	1.27	1.25	.02	.60	.41
69	1.96	1.45	.51	1.97	1.88	1.78	1.33	.45	1.47	1.04	.43	1.17	1.66
70	1.94	1.45	.49	1.99	1.88	1.80	1.33	.47	1.49	1.04	.45	1.11	1.81
71	1.96	1.45	.51	1.95	1.86	1.78	1.33	.45	1.45	1.02	.43	1.23	1.41
72	1.94	1.45	.49	1.95	1.86	1.78	1.33	.45	1.47	1.04	.43	1.17	1.43
73	1.95	1.46	.49	1.97	1.88	1.76	1.33	.43	1.38	1.17	.21	.78	.43
74	1.98	1.45	.53	1.97	1.86	1.76	1.33	.43	1.48	1.23	.25	.73	.41
75	1.95	1.46	.49	1.97	1.88	1.74	1.33	.41	1.46	1.15	.31	.78	1.66
76	1.97	1.46	.51	1.97	1.88	1.76	1.31	.45	1.41	.98	.43	1.11	1.39
77	1.97	1.46	.51	1.97	1.88	1.76	1.31	.45	1.47	1.04	.43	1.13	1.73
78	1.97	1.46	.51	1.99	1.88	1.76	1.31	.45	1.44	1.19	.25	.80	.76
79	1.96	1.45	.51	1.99	1.88	1.76	1.31	.45	1.45	1.25	.20	.80	.41
80	1.97	1.46	.51	2.01	1.88	1.76	1.31	.45	1.44	1.11	.33	.80	1.11
81	1.97	1.46	.51	1.99	1.88	1.78	1.33	.45	1.47	1.04	.43	1.19	1.73
82	1.96	1.45	.51	1.99	1.88	1.76	1.31	.45	1.45	1.02	.43	1.21	1.66
83	1.96	1.45	.51	1.97	1.88	1.76	1.31	.45	1.47	1.04	.43	1.19	1.66
84	1.96	1.45	.51	1.97	1.88	1.78	1.33	.45	1.47	1.04	.43	1.11	1.41
85	1.97	1.46	.51	1.99	1.88	1.78	1.31	.47	1.31	1.31	0.00	.78	.31
86	1.96	1.45	.51	1.99	1.88	1.76	1.31	.45	1.33	1.27	.06	.80	.39
87	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.33	1.15	.18	.80	1.25
88	1.97	1.46	.51	1.97	1.86	1.76	1.31	.45	1.48	1.07	.41	.83	1.68
89	1.96	1.45	.51	1.95	1.86	1.76	1.31	.45	1.45	1.02	.43	1.23	1.75
90	1.96	1.45	.51	1.97	1.86	1.76	1.31	.45	1.25	1.23	.02	.80	.47
91	1.96	1.45	.51	2.13	2.03	1.78	1.33	.45	1.27	1.27	0.00	.80	.67
92	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.42	1.17	.25	.80	.31
93	1.97	1.46	.51	1.97	1.86	1.78	1.33	.45	1.31	1.31	0.00	.80	.35
94	1.94	1.43	.51	1.97	1.86	1.78	1.33	.45	1.47	1.04	.43	1.21	1.60
95	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.45	1.04	.41	1.23	1.48
96	1.96	1.45	.51	1.95	1.86	1.78	1.33	.45	1.47	1.02	.45	1.23	1.43
97	1.97	1.46	.51	1.95	1.86	1.78	1.33	.45	1.44	1.07	.37	.76	1.68
98	1.96	1.45	.51	1.95	1.86	1.76	1.31	.45	1.46	1.25	.21	.80	.43
99	1.96	1.45	.51	1.97	1.86	1.80	1.33	.47	1.49	1.04	.45	1.23	1.78

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CELL UNDER TEST (6)

CN #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP OCV	CP 30 MIN	ZP 30 MIN	FP	CP USABLE	ZP USABLE	FP	CP ZN	CP OCV
100	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.45	1.00	.45	1.23	1.54
101	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.47	1.04	.43	1.19	1.39
102	1.94	1.43	.51	1.95	1.86	1.78	1.33	.45	1.45	1.02	.43	1.17	1.41
103	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.41	1.25	.16	.80	.39
104	1.96	1.45	.51	1.97	1.86	1.78	1.33	.45	1.48	1.27	.21	.80	.45
105	1.96	1.45	.51	1.97	1.88	1.78	1.33	.45	1.45	1.02	.43	1.13	1.41
106	1.96	1.45	.51	1.95	1.88	1.78	1.33	.45	1.43	1.00	.43	1.19	1.39
107	1.94	1.45	.49	1.95	1.86	1.78	1.33	.45	1.43	1.00	.43	1.23	1.41
108	1.94	1.43	.51	1.95	1.86	1.76	1.31	.45	1.47	1.04	.43	1.19	1.41
109	1.94	1.43	.51	1.97	1.88	1.76	1.31	.45	1.44	1.07	.37	.78	1.45
110	1.96	1.45	.51	1.95	1.86	1.76	1.31	.45	1.46	1.09	.37	.76	1.70
111	1.96	1.45	.51	1.95	1.86	2.56	2.56	0.00	0.00	0.00	0.00	1.78	1.84
112	2.44	2.44	0.00	1.99	1.88	0.00	0.00	0.00	0.00	0.00	0.00	1.78	1.88
113	2.01	1.89	.12	2.01	1.89	1.78	1.74	.04	1.68	1.68	0.00	.80	.41
114	1.96	1.88	.08	1.93	1.84	1.74	1.72	.02	1.68	1.68	0.00	.80	.33
115	1.96	1.88	.08	1.95	1.84	1.76	1.72	.04	1.66	1.66	0.00	.72	.41
117	1.95	1.70	.25	1.95	1.86	1.76	1.74	.02	1.41	1.37	.04	.74	1.60
118	2.07	1.95	.12	1.97	1.84	2.21	1.70	.51	1.76	1.43	.33	.80	0.00
119	1.96	1.86	.10	1.95	1.84	2.13	1.68	.45	1.83	1.50	.33	.68	.37
120	2.45	1.88	.57	1.95	1.84	2.17	1.70	.47	1.68	1.56	.12	.72	.33
121	2.35	1.88	.47	1.95	1.84	2.09	1.72	.37	1.64	1.58	.06	.80	.21
122	1.98	1.88	.10	1.95	1.84	1.76	1.72	.04	1.41	1.39	.02	0.00	1.39
123	1.96	1.86	.10	2.29	1.88	1.72	1.72	0.00	1.50	1.50	0.00	.80	1.76
124	1.98	1.88	.10	1.97	1.86	1.76	1.72	.04	1.57	1.41	.16	0.00	1.70
125	2.15	1.86	.29	1.95	1.84	1.76	1.72	.04	1.56	1.46	.10	.78	1.72
126	1.98	1.84	.14	1.95	1.84	1.76	1.72	.04	1.62	1.58	.04	.76	.37
127	2.00	1.88	.12	1.97	1.86	1.91	1.66	.25	1.62	1.48	.14	.80	.37
169	2.01	1.46	.55	1.97	1.86	1.74	1.31	.43	1.41	1.23	.18	.80	.45
170	2.04	1.45	.59	1.97	1.86	1.74	1.31	.43	1.46	1.05	.41	9.11	1.66
171	1.96	1.45	.51	1.95	1.84	1.74	1.31	.43	1.46	1.13	.33	.80	.43
172	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.38	1.15	.23	.78	.45
173	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.38	1.15	.23	.80	.43
174	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.42	1.11	.31	.80	.39
175	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.44	1.15	.29	.80	.45
176	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.46	1.05	.41	1.05	1.68
177	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.46	1.13	.33	.78	.41
178	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.36	1.13	.23	.80	.47
179	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.38	1.17	.21	.80	.47
180	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.46	1.17	.29	.78	.41
181	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.46	1.19	.27	.78	.43
182	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.46	1.07	.39	1.05	1.70
183	1.96	1.45	.51	1.97	1.84	1.74	1.31	.43	1.46	1.09	.37	.64	1.68
184	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.38	1.15	.23	.80	.41
185	1.96	1.43	.53	1.97	1.84	1.74	1.31	.43	1.44	1.13	.31	.74	.39
186	1.96	1.45	.51	1.97	1.86	1.74	1.31	.43	1.46	1.25	.21	.80	.41
187	1.98	1.45	.53	1.97	1.86	1.74	1.31	.43	1.46	1.13	.33	.78	1.72
188	1.98	1.45	.53	1.97	1.84	1.74	1.31	.43	1.44	1.11	.33	.70	1.68
189	1.96	1.45	.51	1.97	1.84	1.74	1.31	.43	1.42	1.13	.29	.72	1.72
190	1.96	1.45	.51	1.97	1.84	1.74	1.31	.43	1.44	1.17	.27	.80	.47

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

CELL UNDER TEST (3)

SN #	CP	ZP	FP	CP	CP	CP	ZP	FP	CP	ZP	FP	CP	CP
	30 MIN			END	OCV		20 MIN			USABLE		20	OCV
250	2.03	1.54	.49	1.93	1.86	1.80	1.31	.49	1.47	1.04	.43	1.21	1.43
251	2.01	1.52	.49	1.95	1.86	1.78	1.31	.47	1.43	1.00	.43	1.15	1.41
252	2.05	1.54	.51	1.93	1.86	1.80	1.31	.45	1.47	1.04	.43	1.17	1.41
253	1.99	1.50	.49	1.93	1.86	1.80	1.31	.49	1.48	1.05	.43	1.19	1.43
254	1.97	1.48	.49	1.93	1.86	2.15	1.58	.57	1.45	1.02	.43	1.15	1.41
255	1.99	1.50	.49	1.93	1.86	1.80	1.31	.49	1.39	.96	.43	1.21	1.78
256	1.88	1.39	.49	1.95	1.86	1.80	1.35	.45	1.33	.98	.35	1.11	1.78
257	1.94	1.45	.49	1.95	1.88	1.68	1.21	.47	1.47	1.04	.43	1.19	1.45
258	1.95	1.46	.49	1.97	1.88	1.78	1.31	.47	1.47	1.04	.43	1.21	1.50
259	1.97	1.50	.47	1.93	1.86	1.78	1.31	.47	1.47	1.04	.43	1.21	1.43
260	1.97	1.48	.49	1.93	1.86	1.78	1.31	.47	1.43	1.00	.43	1.17	1.43
261	1.99	1.50	.49	1.93	1.86	1.78	1.31	.47	1.47	1.04	.43	1.15	1.43
262	1.99	1.50	.49	1.95	1.86	1.78	1.31	.47	1.46	1.05	.41	1.11	1.43
263	2.03	1.52	.51	1.95	1.88	1.80	1.37	.43	1.46	1.09	.37	1.09	1.43
264	1.99	1.52	.47	1.95	1.88	1.78	1.31	.47	1.47	1.04	.43	1.11	1.43
265	1.95	1.46	.49	1.93	1.86	1.78	1.31	.47	1.47	1.04	.43	1.15	1.43
266	1.95	1.46	.49	1.55	1.86	1.78	1.31	.47	1.47	1.04	.43	1.15	1.43
267	1.97	1.52	.45	1.95	1.86	1.78	1.33	.45	1.48	1.07	.41	1.11	1.80
268	1.99	1.52	.47	1.95	1.86	1.76	1.37	.39	1.46	1.09	.37	1.07	1.43
269	1.99	1.52	.47	1.99	1.86	1.74	1.29	.45	1.46	1.05	.41	1.15	1.64
270	1.97	1.48	.49	1.93	1.84	1.78	1.31	.47	1.43	1.02	.41	1.19	1.38
271	1.87	1.46	.41	1.91	1.84	1.80	1.33	.47	1.45	1.02	.43	1.13	1.39
272	1.95	1.48	.47	1.93	1.84	1.78	1.31	.47	1.47	1.02	.45	1.23	1.39
273	1.95	1.48	.47	1.93	1.84	1.78	1.31	.47	1.47	1.04	.43	1.21	1.39
274	1.97	1.48	.49	1.91	1.84	1.78	1.35	.43	1.45	1.02	.43	1.19	1.35
275	1.97	1.52	.45	2.23	2.15	1.78	1.33	.45	1.45	1.02	.43	1.21	1.39
276	1.95	1.48	.47	1.89	1.84	1.80	1.33	.47	1.47	1.04	.43	1.23	1.39
277	1.93	1.46	.47	1.91	1.84	1.80	1.33	.47	1.45	1.00	.45	1.15	1.37
278	1.90	1.48	.47	2.13	1.86	1.68	1.23	.45	1.47	1.04	.43	1.21	1.38
279	2.23	1.76	.49	2.33	1.86	1.74	1.27	.47	1.48	1.05	.43	1.21	1.56
280	2.11	1.62	.49	2.01	1.88	1.70	1.31	.39	1.49	1.07	.41	1.19	1.56
281	1.99	1.52	.47	1.93	1.86	1.78	1.33	.45	1.46	1.07	.39	1.17	1.70
282	1.96	1.46	.47	1.93	1.86	1.82	1.35	.47	1.46	1.08	.41	1.19	1.70
283	1.98	1.48	.47	1.93	1.84	1.80	1.31	.47	1.46	1.08	.39	1.19	1.64
284	1.95	1.46	.47	1.93	1.84	1.80	1.31	.47	1.45	1.04	.41	1.17	1.58
285	1.99	1.52	.51	1.91	1.84	1.80	1.30	.47	1.46	1.05	.41	1.09	1.51
286	1.99	1.48	.49	1.92	1.84	1.78	1.33	.45	1.44	1.05	.39	1.11	1.56
287	1.97	1.46	.45	1.93	1.84	1.76	1.33	.45	1.46	1.19	.27	.76	1.79
288	1.94	1.46	.49	1.95	1.84	1.76	1.31	.45	1.45	1.04	.41	1.19	1.79
289	1.95	1.46	.49	1.91	1.84	1.76	1.31	.45	1.45	1.04	.41	1.05	1.71
290	1.97	1.47	.49	1.91	1.84	1.76	1.31	.45	1.45	1.07	.39	1.15	1.71
291	1.98	1.48	.49	1.91	1.84	1.78	1.31	.45	1.45	1.13	.35	.77	1.70
292	1.94	1.46	.49	1.91	1.84	1.76	1.33	.45	1.45	1.07	.41	1.17	1.71
293	1.97	1.48	.51	1.89	1.84	1.70	1.33	.45	1.46	1.13	.35	1.17	1.37
294	1.95	1.46	.51	1.91	1.84	1.78	1.33	.45	1.46	1.05	.21	.22	1.61
295	1.99	1.50	.47	1.91	1.84	1.76	1.37	.45	1.46	1.11	.35	1.20	1.58
296	1.95	1.46	.49	1.91	1.84	1.78	1.33	.45	1.46	1.11	.35	1.03	1.54
297	1.99	1.48	.49	1.93	1.86	1.76	1.33	.49	1.46	1.11	.33	.67	1.66
298	1.97	1.48	.49	1.91	1.84	1.76	1.31	.45	1.46	1.21	.27	.72	1.52

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

SN #	CP	CP 30 MIN	FP	CP END	CP OCV	CP	CP 30 MIN	FP	CP	CP USABLE	FP	CP ZN	CP OCV
302	1.97	1.46	.51	1.93	1.84	1.72	1.31	.41	1.44	1.23	.21	.78	1.64
305	1.97	1.46	.53	2.01	1.89	1.78	1.31	.47	1.48	1.13	.35	1.57	1.70
306	1.91	1.46	.45	1.89	1.82	1.75	1.33	.43	1.46	1.13	.33	.99	1.60
307	1.91	1.43	.43	1.89	1.82	1.76	1.35	.43	1.44	1.13	.33	.90	1.50
308	1.91	1.46	.45	1.89	1.82	1.76	1.33	.43	1.46	1.17	.31	.94	1.59
309	1.89	1.46	.43	1.91	1.82	1.74	1.31	.43	1.48	1.17	.31	1.07	1.56
310	1.91	1.46	.45	1.89	1.82	1.76	1.33	.43	1.48	1.13	.35	1.07	1.52
312	1.90	1.45	.45	1.91	1.82	1.74	1.33	.41	1.47	1.27	.20	.74	1.69
313	1.92	1.45	.47	1.95	1.84	1.74	1.31	.43	1.45	1.29	.16	.76	1.68
314	1.92	1.45	.47	1.93	1.84	1.76	1.33	.43	1.47	1.27	.20	.80	1.66
316	1.89	1.58	.31	1.91	1.82	1.71	1.46	.25	1.48	1.46	.02	.74	.31
317	1.89	1.62	.27	1.91	1.82	1.71	1.50	.21	1.47	1.43	.04	.80	1.02
318	1.93	1.48	.45	1.93	1.82	1.74	1.31	.43	1.47	1.27	.20	.78	1.31
319	1.90	1.45	.45	1.93	1.84	1.78	1.33	.45	1.47	1.29	.18	.78	1.52
320	1.90	1.45	.45	1.91	1.84	1.76	1.33	.43	1.47	1.27	.20	.80	1.48
321	1.90	1.45	.45	1.93	1.84	1.74	1.31	.43	1.46	1.15	.31	1.01	1.68
322	1.91	1.46	.45	1.95	1.84	1.74	1.31	.43	1.46	1.21	.25	.86	1.68
323	1.90	1.45	.45	1.93	1.84	1.76	1.33	.43	1.49	1.31	.18	.78	1.62
324	1.90	1.43	.47	1.95	1.84	1.76	1.33	.43	1.45	1.29	.16	.78	1.66
325	1.90	1.43	.47	1.93	1.84	1.78	1.33	.45	1.48	1.19	.29	.74	1.66
326	1.88	1.46	.45	1.91	1.84	1.75	1.33	.43	1.40	1.15	.25	.78	1.64
327	1.88	1.46	.45	1.91	1.82	1.78	1.33	.45	1.31	1.31	0.00	.72	1.50
328	1.90	1.43	.47	1.93	1.84	1.76	1.33	.43	1.43	1.31	.12	.68	1.50
329	1.88	1.43	.45	1.93	1.84	1.78	1.33	.45	1.46	1.19	.27	.68	1.58
330	1.88	1.43	.45	1.93	1.84	1.78	1.33	.45	1.42	1.11	.31	1.09	1.46
331	1.90	1.45	.45	1.95	1.88	1.75	1.33	.43	1.45	1.29	.16	.74	1.58
332	1.88	1.43	.45	1.95	1.88	1.78	1.33	.45	1.47	1.27	.20	.68	1.60
333	1.92	1.45	.41	1.97	1.86	1.76	1.31	.45	1.48	1.13	.35	.76	1.74
334	1.97	1.50	.47	2.01	1.83	1.68	1.23	.39	1.46	1.23	.23	.70	1.02
335	1.93	1.48	.45	1.95	1.85	1.76	1.33	.43	1.33	1.31	.02	.86	.27
336	1.91	1.46	.45	2.03	1.89	1.76	1.31	.45	1.47	1.27	.20	.80	.21
337	1.96	1.49	.51	1.97	1.84	1.74	1.31	.43	1.44	1.33	.08	.80	.20
338	1.97	1.48	.49	2.05	1.89</								

CELL UNDER TEST (6)

ON #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP GCV	CP 30 MIN	FP	CP 30 MIN	CP GCV	CP 30 MIN	CP GCV	CP 30 MIN	CP GCV
356	1.88	1.43	.45	1.89	1.82	1.76	1.33	.43	1.47	1.35	.12	.80	1.64
357	1.92	1.45	.47	1.89	1.82	1.80	1.35	.43	1.42	1.37	.25	.76	1.71
358	1.92	1.45	.47	1.89	1.82	1.78	1.33	.45	1.46	1.25	.21	.60	1.74
359	1.90	1.43	.47	1.89	1.82	1.76	1.33	.43	1.39	1.21	.18	.74	1.70
360	1.90	1.43	.47	1.89	1.82	1.76	1.33	.43	1.45	1.31	.14	.78	1.62
361	1.92	1.45	.47	1.89	1.82	1.76	1.33	.43	1.49	1.33	.16	.80	1.64
362	1.92	1.45	.47	1.89	1.82	1.78	1.33	.45	1.43	1.29	.14	.80	1.60
363	1.93	1.46	.47	1.91	1.84	1.80	1.35	.45	1.47	1.27	.20	.80	1.72
364	1.92	1.45	.47	1.89	1.84	1.78	1.33	.45	1.44	1.15	.29	.90	1.74
365	1.92	1.45	.47	1.89	1.82	1.76	1.33	.45	1.46	1.23	.23	.72	1.72
366	1.90	1.43	.47	1.89	1.82	1.76	1.33	.43	1.45	1.29	.16	.80	1.70
367	1.90	1.45	.45	1.89	1.82	1.76	1.33	.43	1.47	1.27	.20	.78	1.76
368	1.92	1.45	.47	1.89	1.82	1.78	1.33	.45	1.47	1.33	.14	.80	1.64
369	1.92	1.45	.47	1.91	1.84	1.76	1.33	.43	1.47	1.35	.12	.80	1.54
370	1.92	1.45	.47	1.89	1.84	2.99	2.99	0.00	1.46	1.13	.33	.96	1.74
371	1.91	1.46	.45	1.89	1.82	1.78	1.37	.41	1.46	1.17	.29	.68	1.66
372	1.92	1.45	.47	1.89	1.82	2.75	2.75	0.00	1.60	1.60	0.00	.80	1.58
373	1.91	1.46	.45	1.89	1.84	1.76	1.33	.43	1.46	1.21	.25	.76	1.74
374	1.92	1.84	.08	1.91	1.84	0.00	0.00	0.00	0.00	0.00	0.00	1.74	1.82
375	1.94	1.43	.51	1.89	1.84	1.80	1.33	.47	1.48	1.15	.32	.52	1.74
376	1.94	1.45	.45	1.91	1.84	1.76	1.33	.45	1.44	1.23	.21	.74	1.70
377	1.94	1.43	.63	1.91	1.84	1.76	1.33	.47	1.51	1.31	.33	.74	1.55
378	1.95	1.46	.49	1.91	1.84	1.76	1.33	.47	1.52	1.33	0.00	1.58	1.62
379	1.92	1.43	.49	2.25	1.80	0.00	0.00	0.00	1.92	.52	0.00	1.56	.45
379	2.77	1.66	.54	2.34	1.76	0.00	0.00	0.00	1.87	1.37	.80	.76	.31
381	1.98	1.45	.51	1.97	1.89	1.82	1.33	.41	1.43	1.22	.23	.80	1.74
382	1.90	1.45	.45	1.88	1.82	1.76	1.33	.43	1.45	1.11	.33	1.01	1.74
383	1.91	1.46	.45	1.89	1.82	1.76	1.33	.43	1.46	1.19	.29	.62	1.78
384	1.90	1.45	.45	1.89	1.82	1.75	1.33	.43	1.47	1.31	.16	.35	1.68
385	1.91	1.46	.45	1.93	1.84	1.76	1.31	.41	1.48	1.13	.35	1.07	1.76
387	1.92	1.45	.49	1.93	1.84	1.76	1.31	.45	1.46	1.17	.29	.90	1.74
388	1.90	1.45	.45	1.91	1.82	1.76	1.31	.49	1.46	1.15	.29	.86	1.76
389	1.97	1.46	.45	1.89	1.84	1.76	1.31	.41	1.45	1.15	.31	1.54	1.76
390	1.92	1.45	.47	1.89	1.84	1.76	1.31	.45	1.45	1.25	.20	.80	1.74
391	1.92	1.45	.47	1.99	1.84	1.76	1.33	.45	1.47	1.29	.18	.78	1.74
392	1.91	1.46	.45	1.91	1.84	1.78	1.33	.45	1.43	1.27	.18	.80	1.74
393	1.90	1.45	.45	1.91	1.84	1.76	1.31	.45	1.43	1.15	.29	.80	1.74
394	1.90	1.45	.45	2.59	1.82	1.76	1.31	.41	1.44	1.15	.24	.68	1.72
395	1.90	1.45	.45	1.89	1.82	1.76	1.31	.41	1.48	1.19	.29	.80	1.74
396	1.90	1.45	.45	1.89	1.82	1.76	1.31	.45	1.44	1.23	.21	.68	1.74
397	1.88	1.45	.43	1.89	1.82	1.76	1.31	.43	1.31	1.25	.06	.84	1.74
398	1.90	1.45	.45	1.89	1.84	1.76	1.31	.47	1.42	1.07	.35	.97	1.72
399	1.90	1.45	.45	1.89	1.82	1.76	1.31	.49	1.39	1.21	.16	.49	1.65
400	1.90	1.45	.45	1.91	1.84	1.76	1.31	.41	1.44	1.19	.25	.75	1.78
401	1.90	1.45	.45	1.91	1.82	1.76	1.31	.43	1.43	1.23	.20	.72	1.74
402	1.90	1.45	.47	1.91	1.82	1.76	1.31	.41	1.45	1.27	.16	.70	1.76
403	1.92	1.45	.47	1.91	1.84	1.76	1.31	.41	1.46	1.23	.30	.60	1.74
404	1.92	1.45	.47	1.91	1.84	1.76	1.31	.41	1.46	1.17	.30	1.00	1.74
405	1.92	1.45	.47	1.91	1.84	1.76	1.31	.45	1.46	1.07	.32	1.03	1.72

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CELL UNDER TEST (6)

CN #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP OCV	CP 30 MIN	ZP 30 MIN	FP	CP USABLE	ZP USABLE	FP	CP ZN	CP OCV
406	1.92	1.45	.47	1.89	1.84	1.78	1.33	.45	1.38	1.11	.27	.84	1.74
407	1.90	1.43	.47	1.91	1.84	1.76	1.31	.45	1.46	1.21	.25	.43	1.70
408	1.90	1.43	.47	1.89	1.84	1.76	1.31	.45	1.46	1.25	.21	.80	1.76
409	1.92	1.45	.47	1.91	1.84	1.76	1.31	.45	1.45	1.27	.18	.70	1.72
410	1.92	1.45	.47	1.91	1.84	1.78	1.33	.45	1.46	1.09	.37	1.03	1.72
411	1.88	1.43	.45	1.91	1.84	2.71	2.71	0.00	1.45	1.31	.14	.72	1.70
412	1.89	.70	1.19	1.91	1.84	1.78	1.33	.45	1.44	1.11	.33	.79	1.72
413	1.92	1.43	.49	1.91	1.84	1.76	1.33	.43	2.71	2.71	0.00	.14	1.54
414	1.91	1.46	.45	1.95	1.84	1.74	1.29	.45	2.79	2.79	0.00	0.00	0.00
416	1.93	1.46	.47	1.93	1.82	1.72	1.33	.39	1.43	1.35	.08	.78	1.62
417	1.90	1.43	.47	1.91	1.82	1.72	1.31	.41	1.46	1.23	.23	.70	1.72
418	1.90	1.45	.45	1.91	1.84	1.74	1.31	.43	1.47	1.27	.20	.78	1.72
419	1.92	1.45	.47	1.91	1.84	1.74	1.31	.43	1.44	1.19	.25	.76	1.74
420	1.92	1.45	.47	1.93	1.84	1.74	1.31	.43	1.47	1.29	.18	.70	1.72
421	1.92	1.45	.47	1.91	1.84	1.74	1.33	.41	2.83	2.83	0.00	0.00	.80
422	1.92	1.45	.47	1.93	1.82	1.74	1.31	.43	1.47	1.35	.12	.80	1.66
423	1.92	1.45	.47	1.91	1.82	1.74	1.31	.43	1.47	1.35	.12	.80	1.66
424	1.92	1.43	.49	1.93	1.84	1.72	1.31	.41	1.47	1.27	.20	.80	1.66
425	1.90	1.43	.47	1.93	1.84	1.74	1.31	.43	1.45	1.31	.14	.78	1.62
426	1.90	1.43	.47	1.93	1.84	2.70	2.70	0.00	1.45	1.33	.12	.76	1.60
427	1.90	1.43	.47	1.93	1.84	1.74	1.31	.43	1.41	1.27	.14	.78	1.64
428	1.90	1.43	.47	1.93	1.84	1.74	1.33	.41	1.45	1.33	.12	.80	1.62
429	1.92	1.43	.49	1.93	1.84	1.74	1.33	.41	1.47	1.33	.14	.80	1.64
430	1.90	1.43	.47	1.93	1.84	1.76	1.31	.45	1.42	1.13	.29	.78	1.62
431	1.92	1.43	.49	1.93	1.84	1.74	1.31	.43	1.39	1.25	.14	.80	1.60
432	1.90	1.43	.47	1.93	1.84	1.76	1.33	.43	1.45	1.29	.16	.78	1.56
433	1.90	1.43	.47	1.93	1.84	1.74	1.31	.43	1.41	1.31	.10	.74	1.56
434	1.92	1.43	.49	1.93	1.84	1.76	1.33	.43	2.66	2.66	0.00	0.00	0.00
435	1.91	1.50	.41	1.93	1.82	1.72	1.31	.41	1.45	1.27	.18	.78	1.54
436	1.93	1.46	.47	1.93	1.84	1.72	1.31	.41	1.45	1.29	.16	.74	1.70
437	1.93	1.46	.47	1.93	1.84	1.86	1.86	0.00	1.44	1.15	.29	.80	1.72
438	1.91	1.46	.45	1.93	1.84	1.74	1.39	.35	1.86	1.86	0.00	.78	0.00
442	1.93	1.50	.43	1.95	1.84	1.74	1.35	.39	1.47	1.29	.18	0.00	1.70
443	1.93	1.58	.35	1.95	1.86	1.76	1.35	.41	1.45	1.27	.18	.72	1.68
444	1.92	1.45	.47	1.95	1.86	1.74	1.31	.43	1.47	1.29	.18	.80	1.70
445	1.92	1.45	.47	1.95	1.86	1.74	1.31	.43	1.45	1.29	.16	0.00	1.72
446	1.92	1.45	.47	1.95	1.86	1.74	1.31	.43	1.48	1.27	.21	.80	1.70
448	1.90	1.43	.47	1.95	1.86	1.74	1.64	.10	1.45	1.27	.18	.76	1.68
449	1.92	1.43	.49	1.95	1.86	1.76	1.31	.45	1.46	1.25	.21	.78	1.68
450	1.92	1.43	.49	1.95	1.86	1.76	1.74	.02	1.43	1.41	.02	.76	1.76
451	1.91	1.89	.02	1.95	1.86	1.78	1.76	.02	1.47	1.45	.02	.56	1.74
452	1.91	1.89	.02	1.95	1.86	1.78	1.76	.02	1.43	1.41	.02	.72	1.72
453	1.91	1.89	.02	1.95	1.88	1.78	1.76	.02	1.47	1.45	.02	.78	1.72
454	1.93	1.91	.02	1.93	1.88	1.78	1.76	.02	1.46	1.25	.21	.80	1.70
455	1.92	1.45	.47	1.93	1.86	1.78	1.33	.45	1.44	1.23	.21	.80	1.70
460	1.92	1.31	.61	1.91	1.84	2.01	2.01	0.00	1.45	1.25	.20	0.00	1.76
469	1.92	1.45	.47	1.91	1.82	1.76	1.33	.43	1.47	1.29	.18	0.00	1.72
470	1.92	1.45	.47	1.91	1.82	1.76	1.33	.43	1.37	1.33	.04	0.00	1.56
471	1.91	1.27	.64	1.89	1.82	1.74	1.70	.04	2.42	2.42	0.00	0.00	1.56

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CELL UNDER TEST (6)

ON #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP OCV	CP 30 MIN	ZP 30 MIN	FP	CP USABLE	ZP USABLE	FP	CP ZN	CP OCV
481	1.80	1.37	.43	1.88	1.78	1.70	1.31	.39	1.44	1.19	.25	.80	1.72
482	1.88	1.43	.45	1.89	1.82	1.76	1.33	.43	1.46	1.25	.21	.80	1.70
483	1.92	1.45	.47	1.91	1.82	1.80	1.31	.49	1.45	1.27	.18	.76	1.70
484	1.92	1.45	.47	1.91	1.82	1.74	1.31	.43	1.47	1.27	.20	0.00	1.70
487	1.92	1.45	.47	1.91	1.82	1.74	1.31	.43	1.46	1.21	.25	.76	1.70
488	1.92	1.45	.47	1.93	1.82	1.74	1.31	.43	1.48	1.17	.31	.80	1.72
489	1.90	1.43	.47	1.91	1.82	1.74	1.31	.43	1.46	1.19	.27	.78	1.72
494	1.93	1.46	.47	1.95	1.84	1.76	1.31	.45	1.48	1.13	.35	1.07	1.66
501	2.03	1.54	.49	1.93	1.84	1.78	1.33	.45	1.46	1.09	.37	0.00	1.76
502	1.94	1.45	.49	1.89	1.82	1.78	1.33	.45	1.37	1.27	.10	0.00	1.76
504	1.92	1.43	.49	1.91	1.82	1.76	1.31	.45	1.49	1.29	.20	.80	1.43
505	1.90	1.43	.47	1.91	1.82	1.72	1.29	.43	1.43	1.31	.12	1.11	1.74
506	1.90	1.43	.47	1.89	1.82	1.76	1.31	.45	1.45	1.04	.41	1.21	1.54
507	1.90	1.43	.47	1.89	1.82	1.76	1.31	.45	1.46	1.15	.31	1.09	1.72
508	1.90	1.43	.47	1.91	1.82	1.74	1.31	.43	1.44	1.05	.39	1.19	1.74
509	1.90	1.43	.47	1.91	1.82	1.76	1.31	.45	1.47	1.27	.20	0.00	1.74
510	1.92	1.43	.49	1.89	1.82	1.78	1.33	.45	1.46	1.05	.41	1.17	1.74
511	1.90	1.43	.47	1.89	1.82	1.78	1.33	.45	1.46	1.05	.41	1.21	1.70
512	1.88	1.43	.45	1.89	1.82	1.76	1.31	.45	1.44	1.23	.21	.74	1.64
513	1.90	1.43	.47	1.89	1.82	1.76	1.31	.45	1.47	1.27	.20	.78	1.62
514	1.93	1.46	.47	1.91	1.82	1.76	1.31	.45	1.48	1.13	.35	1.07	1.76
515	1.92	1.45	.47	1.89	1.82	1.76	1.31	.45	1.46	1.13	.33	1.01	1.70
516	1.92	1.45	.47	1.91	1.82	1.76	1.31	.45	1.46	1.07	.39	1.19	1.68
522	1.94	1.45	.49	1.91	1.84	1.78	1.31	.47	1.35	1.29	.06	0.00	1.66
523	1.94	1.45	.49	1.89	1.84	1.74	1.31	.43	1.46	1.21	.25	1.09	1.72
524	1.92	1.43	.49	1.89	1.82	1.76	1.31	.45	1.46	1.15	.31	1.09	1.68
525	1.90	1.43	.47	1.89	1.82	1.78	1.33	.45	1.30	1.05	.25	.96	1.70
526	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.42	1.21	.21	.72	1.58
527	1.90	1.41	.49	1.89	1.82	1.74	1.31	.43	1.48	1.19	.29	.84	1.70
528	1.94	1.45	.49	1.91	1.82	1.76	1.31	.45	1.48	1.19	.29	.76	1.76
529	1.92	1.43	.49	1.91	1.82	1.74	1.31	.43	1.47	1.29	.18	.78	1.64
530	1.94	1.45	.49	1.91	1.82	1.78	1.33	.45	1.44	1.23	.21	1.09	1.66
531	1.92	1.43	.49	1.89	1.82	1.74	1.31	.43	1.46	1.19	.27	.84	1.56
532	1.92	1.43	.49	1.89	1.82	1.74	1.31	.43	1.48	1.15	.33	.92	1.60
533	1.92	1.43	.49	1.89	1.82	1.72	1.29	.43	1.46	1.13	.33	1.05	1.62
534	1.92	1.43	.49	1.89	1.82	1.74	1.31	.43	1.46	1.07	.39	1.15	1.68
535	1.92	1.43	.49	1.89	1.82	1.74	1.31	.43	1.47	1.31	.16	0.00	1.58
536	1.90	1.43	.47	1.91	1.82	1.74	1.31	.43	1.42	1.07	.35	1.03	1.52
537	1.92	1.43	.49	1.91	1.82	1.76	1.33	.43	1.46	1.05	.41	1.09	1.56
538	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.21	.25	0.00	1.70
539	1.92	1.45	.47	1.91	1.82	1.76	1.31	.45	1.45	1.29	.16	0.00	1.70
540	1.92	1.45	.47	1.91	1.82	1.76	1.31	.45	1.46	1.13	.33	.80	1.64
541	1.92	1.45	.47	1.89	1.82	1.74	1.31	.43	1.41	1.23	.18	.72	1.52
542	1.90	1.43	.47	1.91	1.82	1.74	1.31	.43	1.46	1.25	.21	.80	1.56
543	1.92	1.43	.49	1.93	1.84	1.72	1.29	.43	1.46	1.07	.39	1.17	1.74
544	1.93	1.46	.47	1.93	1.84	1.72	1.29	.43	1.46	1.09	.37	.76	1.45
545	1.92	1.45	.47	1.93	1.84	1.76	1.31	.45	1.46	1.25	.21	0.00	1.74
546	1.93	1.46	.47	1.91	1.82	1.76	1.31	.45	1.44	1.21	.23	0.00	1.70
547	1.90	1.43	.47	1.91	1.82	1.74	1.31	.43	1.47	1.27	.20	.78	1.60

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

CELL UNDER TEST (6)

CN #	CP 30 MIN	ZP	FP	CP END	CP OCV	CP 30 MIN	ZP	FP	CP	ZP USABLE	FP	CP ZN	CP OCV
548	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.42	1.05	.37	1.13	1.72
549	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.44	1.07	.37	1.01	1.72
550	1.92	1.45	.47	1.89	1.82	1.74	1.31	.43	1.44	1.07	.37	.99	1.74
551	1.92	1.45	.47	1.89	1.82	1.74	1.31	.43	1.46	1.19	.27	.76	1.72
552	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.42	1.11	.31	.47	1.64
553	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.44	1.13	.31	.87	1.64
554	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.40	1.05	.35	.77	1.72
555	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.48	1.23	.25	.02	1.60
556	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.31	1.31	0.00	.53	1.48
557	1.90	1.43	.47	1.89	1.82	1.78	1.33	.45	1.46	1.25	.21	.53	1.58
558	1.92	1.43	.49	1.89	1.82	1.74	1.31	.43	1.46	1.17	.29	.96	1.68
559	1.92	1.45	.47	1.89	1.82	1.74	1.31	.43	1.42	1.05	.37	1.09	1.72
560	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.38	1.15	.23	.97	1.68
561	1.92	1.45	.47	1.89	1.82	1.76	1.33	.43	1.46	1.15	.31	.96	1.74
562	1.92	1.43	.49	1.89	1.82	1.74	1.33	.41	1.46	1.11	.35	1.05	1.74
563	1.90	1.43	.47	1.89	1.82	1.76	1.33	.43	1.46	1.21	.25	.76	1.62
564	1.92	1.43	.49	1.91	1.82	1.76	1.60	.16	1.45	1.27	.18	.78	1.56
565	1.94	1.39	.55	1.91	1.82	1.76	1.39	.37	1.49	1.31	.18	.78	1.48
566	1.92	1.39	.53	1.95	1.84	1.72	1.37	.35	1.49	1.31	.18	.80	1.60
567	1.92	1.41	.51	1.91	1.82	1.74	1.37	.37	1.44	1.23	.21	.78	1.72
568	1.90	1.39	.51	1.89	1.82	1.74	1.37	.37	1.45	1.29	.16	.76	1.72
569	1.92	1.41	.51	1.91	1.82	1.74	1.37	.37	1.39	1.21	.18	.76	1.72
570	1.90	1.39	.51	1.89	1.82	1.74	1.35	.39	1.44	1.23	.21	.51	1.58
571	1.92	1.41	.51	1.89	1.82	1.74	1.37	.37	1.43	1.27	.16	.80	1.62
572	1.90	1.39	.51	1.89	1.82	1.74	1.37	.37	1.45	1.29	.16	.80	1.74
573	1.92	1.39	.53	1.89	1.82	1.76	1.39	.37	1.47	1.33	.14	.80	1.58
574	1.93	1.25	.68	1.91	1.82	1.84	1.84	0.00	1.64	1.64	0.00	.70	1.64
575	1.92	1.45	.47	1.89	1.82	2.52	2.52	0.00	1.82	1.82	0.00	.76	0.00
580	1.88	1.43	.45	1.89	1.80	1.72	1.31	.41	1.46	1.19	.27	.80	1.72
581	1.88	1.43	.45	1.89	1.80	1.72	1.31	.41	1.48	1.19	.29	.80	1.74
582	1.88	1.43	.45	1.89	1.80	1.72	1.31	.41	1.46	1.21	.25	.80	1.72
583	1.90	1.43	.47	1.89	1.82	1.72	1.31	.41	1.46	1.17	.29	.76	1.70
589	1.92	1.45	.47	1.89	1.80	1.70	1.31	.39	1.44	1.21	.23	.80	1.52
590	1.90	1.43	.47	1.89	1.80	1.70	1.31	.39	1.47	1.31	.16	.78	1.50
591	1.90	1.43	.47	1.91	1.80	1.70	1.31	.39	1.46	1.19	.27	.80	1.66
592	1.90	1.43	.47	1.91	1.80	1.70	1.31	.39	1.44	1.17	.27	.80	1.72
593	1.90	1.43	.47	1.91	1.80	1.70	1.31	.39	1.47	1.27	.20	.80	1.70
594	1.88	1.43	.45	1.89	1.80	1.72	1.31	.41	1.44	1.17	.27	.78	1.72
595	1.88	1.43	.45	1.89	1.80	1.72	1.31	.41	1.45	1.31	.14	.80	1.58
596	1.88	1.43	.45	1.89	1.82	1.72	1.31	.41	1.31	1.31	0.00	.78	1.62
597	1.88	1.43	.45	1.89	1.80	1.72	1.31	.41	1.44	1.19	.25	.80	1.54
600	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.11	.35	1.01	1.76
601	1.90	1.43	.47	1.89	1.82	1.75	1.52	.23	1.48	1.19	.29	.70	1.54
602	1.87	1.23	.64	1.89	1.82	1.71	1.50	.21	1.43	1.23	.20	.76	1.50
603	1.87	1.23	.64	1.89	1.80	1.71	1.50	.21	1.43	1.31	.12	.80	1.43
604	1.87	1.23	.64	1.89	1.80	1.72	1.52	.20	1.41	1.37	.04	.80	1.62
606	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.13	.33	1.01	1.70
607	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.46	1.07	.39	1.13	1.66
608	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.09	.37	1.15	1.68

CELL UNDER TEST (6)

CP	CP	CP	CP	CP	CP	CP	CP	CP	CP	CP	CP	CP	CP
#	30 MIN	30 MIN	END	OCV	30 MIN	30 MIN	USABLE	ZN	OCV				
609	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.41	1.04	.37	1.03	1.60
610	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.35	.98	.37	1.13	1.70
611	1.90	1.43	.47	1.89	1.82	1.76	1.31	.45	1.41	1.04	.37	1.15	1.70
612	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.09	.37	.94	1.60
613	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.48	1.13	.35	1.07	1.68
614	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.44	1.09	.35	1.13	1.70
615	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.46	1.17	.29	1.09	1.70
616	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.46	1.23	.23	.61	1.56
617	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.43	1.23	.20	.72	1.39
618	1.90	1.43	.47	1.80	1.89	1.82	1.37	.45	1.47	1.27	.20	.76	1.54
619	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.46	1.17	.29	.61	1.56
620	1.88	1.43	.45	1.88	1.82	1.74	1.31	.43	1.40	1.11	.29	.68	1.70
621	1.88	1.43	.45	1.88	1.82	1.76	1.31	.45	1.48	1.25	.23	.74	1.64
622	1.88	1.43	.45	1.88	1.82	1.72	1.29	.43	1.42	1.21	.21	.74	1.54
623	1.90	1.43	.47	1.88	1.82	1.74	1.31	.43	1.46	1.15	.31	1.01	1.60
624	1.88	1.43	.45	1.88	1.82	1.74	1.31	.43	1.44	1.11	.33	.78	1.72
625	1.88	1.43	.45	1.88	1.82	1.74	1.31	.43	1.40	1.05	.35	.99	1.72
626	1.88	1.43	.45	1.88	1.82	1.74	1.31	.43	1.34	1.07	.27	1.01	1.72
627	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.30	1.05	.25	.90	1.72
628	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.44	1.07	.37	1.09	1.72
629	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.44	1.07	.37	1.13	1.72
630	1.92	1.45	.47	1.91	1.82	1.74	1.31	.43	1.46	1.13	.33	1.11	1.72
631	1.90	1.45	.45	1.89	1.82	1.74	1.31	.43	1.48	1.11	.37	1.13	1.74
632	1.88	1.43	.45	1.89	1.82	1.72	1.31	.41	1.46	1.11	.35	1.11	1.74
633	1.88	1.43	.45	1.88	1.82	1.74	1.31	.43	1.44	1.09	.35	1.13	1.74
634	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.46	1.11	.35	1.09	1.74
635	1.90	1.45	.45	1.89	1.82	1.72	1.29	.43	1.48	1.11	.37	1.13	1.74
636	1.90	1.43	.47	1.88	1.82	1.74	1.31	.43	1.48	1.11	.37	1.11	1.72
637	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.09	.37	1.13	1.72
638	1.93	1.46	.47	1.89	1.82	1.72	1.29	.43	1.48	1.11	.37	1.13	1.72
639	1.92	1.45	.47	1.97	1.84	1.72	1.29	.43	1.48	1.11	.37	1.13	1.74
640	1.92	1.45	.47	1.89	1.82	1.74	1.31	.43	1.46	1.09	.37	1.13	1.72
641	1.92	1.45	.47	1.91	1.82	1.74	1.31	.43	1.44	1.09	.35	.72	1.74
642	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.09	.37	1.11	1.72
643	1.90	1.43	.47	1.89	1.82	1.74	1.31	.43	1.46	1.09	.37	1.07	1.72
651	1.90	1.43	.47	1.89	1.82	1.74	1.33	.41	1.48	1.09	.39	1.15	1.74
652	1.92	1.45	.47	1.89	1.82	1.74	1.33	.41	1.44	1.07	.37	1.15	1.74
653	1.88	1.43	.45	1.89	1.82	1.76	1.33	.43	1.44	1.07	.37	1.13	1.72
654	1.90	1.43	.47	1.89	1.82	1.74	1.33	.41	1.42	1.05	.37	1.11	1.74
655	1.92	1.45	.47	1.89	1.82	1.76	1.31	.45	1.46	1.09	.37	1.13	1.74
668	1.88	1.43	.45	1.89	1.82	1.74	1.31	.43	1.44	1.11	.33	.96	1.74
673	1.92	1.45	.47	1.91	1.84	1.74	1.29	.45	1.46	1.19	.27	.80	1.60
674	1.92	1.45	.47	1.91	1.82	1.74	1.29	.45	1.48	1.09	.39	.88	1.76
675	1.92	1.45	.47	1.89	1.82	1.74	1.29	.45	1.45	1.27	.18	.80	1.23
676	1.92	1.45	.47	1.89	1.82	1.74	1.29	.45	1.33	1.25	.08	.78	.80
677	1.90	1.43	.47	1.89	1.82	1.74	1.29	.45	1.41	1.25	.16	.80	.64
678	1.94	1.45	.49	1.91	1.84	1.76	1.29	.47	1.46	1.25	.21	.80	.74
679	1.96	1.45	.51	1.91	1.84	1.74	1.29	.45	1.45	1.02	.43	1.11	1.76
680	1.92	1.45	.47	1.89	1.82	1.76	1.31	.45	1.46	1.05	.41	1.23	1.76

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LOCKHEED PALO ALTO RESEARCH LABORATORY
LOCKHEED MISSILES & SPACE COMPANY, INC.

CELL UNDER TEST (6)

CN #	CP 30 MIN	ZP 30 MIN	FP	CP END	CP OCV	CP 30 MIN	ZP 30 MIN	FP	CP USABLE	ZP USABLE	FP	CP ZN	CP OCV
681	1.90	1.43	.47	1.89	1.82	1.76	1.31	.45	1.43	1.27	.16	.88	1.48
691	1.94	1.43	.51	1.91	1.84	1.80	1.31	.49	1.48	1.23	.25	.74	1.64
692	1.94	1.43	.51	1.91	1.84	1.84	1.33	.51	1.82	.80	1.02	1.68	1.78
693	1.94	1.45	.49	1.89	1.84	1.78	1.31	.47	1.46	1.25	.21	.76	.78
694	1.92	1.43	.49	1.89	1.84	1.78	1.31	.47	1.46	1.25	.21	.78	.82
695	1.92	1.43	.49	1.91	1.84	1.76	1.29	.47	1.42	1.17	.25	.80	1.68
696	1.94	1.43	.51	1.91	1.84	1.78	1.31	.47	1.44	1.15	.29	.80	1.70
697	1.90	1.43	.47	1.89	1.84	1.76	1.31	.45	1.46	1.19	.27	.80	1.64
698	1.92	1.45	.47	1.89	1.84	1.76	1.31	.45	1.46	1.05	.41	1.17	9.74

Appendix G
PROGRAM FOR DATA SMOOTHING AND GRAPHICS

The following program, written in TYMBASIC, applies a data smoothing routine (5, 7, or 11 point - Ref.63) which eliminates data noise caused by reference electrode, thermal, or electronic malfunctioning. The program outputs a data file which is comprised of a cycle number and an efficiency in the following data block order.

- Voltaic efficiency
- Voltaic data smoothing
- Coulombic efficiency
- Coulombic data smoothing
- Energy efficiency
- Energy data smoothing

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10 STRING PLOT
20 DIM NP(2000),NDATA(2000),MDATA(2000),NC(2000),CDATA(2000)
30 DIM CYCLE(2000),CD(2000),DD(2000),VE(2000)
40 DIM UE(2000),OE(2000),ZE(2000),FE(2000)
50 TEXT DCODE(2000),TCODE(2000)
60 INPUT IN IMAGE "CELL NUMBER ?  #":L
70 IF L#1 THEN GO TO 130
80 OPEN "REFF1",INPUT,4
90 OPEN "PLOT1",OUTPUT,5
100 GOSUB 430
110 L=1
120 GO TO 420
130 IF L#2 THEN GO TO 190
140 OPEN "REFF2",INPUT,4
150 OPEN "PLOT2",OUTPUT,5
160 GOSUB 430
170 L=2
180 GO TO 420
190 IF L#3 THEN GO TO 250
200 OPEN "REFF3",INPUT,4
210 OPEN "PLOT3",OUTPUT,5
220 GOSUB 430
230 L=3
240 GO TO 420
250 IF L#4 THEN GO TO 310
260 OPEN "REFF4",INPUT,4
270 OPEN "PLOT4",OUTPUT,5
280 GOSUB 430
290 L=4
300 GO TO 420
310 IF L#5 THEN GO TO 370
320 OPEN "REFF5",INPUT,4
330 OPEN "PLOT5",OUTPUT,5
340 GOSUB 430
350 L=5
360 GO TO 420
370 IF L#6 THEN GO TO 420
380 OPEN "REFF6",INPUT,4
390 OPEN "PLOT6",OUTPUT,5
400 GOSUB 430
410 L=6
420 GO TO 1040
430 STRING W,Y

```

```

440 DIM CP(2000)
450 W="6%B 8%B # # # # # # # #/"
460 Y="2(12#)/"
470 FOR I=1 TO 1000
480 ON ENDFILE (4) GO TO 520
490 INPUT FROM 4 IN FORM W:DCODE(I),TCODE(I),CYCLE(I),CD(I),DD(I),VE(I),
UE(I),OE(I),ZE(I),FE(I) .
500 G=I
510 NEXT I
520 PRINT IN IMAGE "NO. DATA POINTS : #":G
530 INPUT IN IMAGE "5 OR 7 POINT SMOOTHING ? #":P
540 FOR J=1 TO G
550 WRITE ON 5 IN FORM Y:CYCLE(J),VE(J)
560 NDATA(J)=VE(J)
570 NEXT J
580 WRITE ON 5:"9999.,0."
590 GOSUB 740
600 FOR J=1 TO G
610 WRITE ON 5 IN FORM Y:CYCLE(J),UE(J)
620 NDATA(J)=UE(J)
630 NEXT J
640 WRITE ON 5:"9999.,0."
650 GOSUB 740
660 FOR J=1 TO G
670 WRITE ON 5 IN FORM Y:CYCLE(J),OE(J)
680 NDATA(J)=OE(J)
690 NEXT J
700 WRITE ON 5:"9999.,0."
710 GOSUB 740
720 CLOSE 4,5
730 RETURN
740 B=G-P-1
750 FOR AK=2 TO P
760 JK=AK-1
770 NP(AK)=NDATA(JK)
780 NC(AK)=CYCLE(JK)
790 NEXT AK
800 FOR K=1 TO B
810 JK=K+P-1
820 FOR KK=1 TO P-1
830 KA=KK+1
840 NP(KK)=NP(KA)
850 NC(KK)=NC(KA)
860 NEXT KK

```

```

870 NP(P)=NDATA(JK)
880 NC(P)=CYCLE(JK)
890 IF P=7 THEN GO TO 950
900 NSUM=17*NP(3)+12*(NP(2)+NP(4))-3*(NP(1)+NP(5))
910 NCSUM=17*NC(3)+12*(NC(2)+NC(4))-3*(NC(1)+NC(5))
920 MDATA(K)=NSUM/35
930 CDATA(K)=NCSUM/35
940 GO TO 990
950 NSUM=7*NP(4)+6*(NP(3)+NP(5))+3*(NP(2)+NP(6))-2*(NP(1)+NP(7))
960 NCSUM=7*NC(4)+6*(NC(3)+NC(5))+3*(NC(2)+NC(6))-2*(NC(1)+NC(7))
970 MDATA(K)=NSUM/21
980 CDATA(K)=NCSUM/21
990 WRITE ON 5 IN FORM Y:CDATA(K),MDATA(K)
1000 NEXT K
1010 WRITE ON 5:"9999.,0."
1020 PRINT
1030 RETURN
1040 PRINT

```

Appendix H

EZPLOT PROGRAM FOR GRAPHICS GENERATION

The following program, written in THMSHARE Editor, is an EZPLOT program for generating a plot of cyclic data by the Tymshare computer. The program takes the data file generated by the Data Smoothing program, Appendix G, and plots the data. The particular listing generates a plot for voltaic, coulombic, and energy efficiencies for a cell having less than 600 cycles. Simple modification to the first INCREMENT of each plot allows for scaling of plot to the number of cycles.

```

ROTATE TO 270
MOVE TO 3,1
TYPE "OPEN DATA FILE"
PAUSE
X SCALE
    START=0
    INCREMENT =100
Y SCALE
    START=60
    INCREMENT=10
X AXIS
    LENGTH=6
    SPACING=.5,1
    FORMAT="%%%"
    LABEL="CYCLE NUMBER"
    LSIZE=.1
Y AXIS
    LENGTH=6
    SPACING=.5,1
    FORMAT="%%%"
    LABEL="EFFICIENCY (%)"
    LSIZE=.1
DATA
PLOT
    SYMBOL=25
    SIZE=.1
EXECUTE
SUPPRESS X AXIS
SUPPRESS Y AXIS
DELETE PLOT
DATA
PLOT
    LINES
    SIZE=.1
TITLE 1
    LOC=2,6
    TEXT="VOLTAIC EFFICIENCY"
    SIZE=.1
EXECUTE
DELETE PLOT
MOVE TO 0,8.5
SUPPRESS TITLE 1
X SCALE

```

```

        START=0
        INCREMENT =100
Y SCALE
        START=60
        INCREMENT=10
X AXIS
        LENGTH=6
        SPACING=.5,1
        FORMAT="%%%"
        LABEL="CYCLE NUMBER"
        LSIZE=.1
Y AXIS
        LENGTH=6.5,1
        SPACING=.5,1
        FORMAT="%%%"
        LABEL="EFFICIENCY (%)"
        LSIZE=.1
DATA
PLOT
        SYMBOL=25
        SIZE=.1
EXECUTE
SUPPRESS X AXIS
SUPPRESS Y AXIS
DELETE PLOT
DATA
PLOT
        LINES
        SIZE=.1
TITLE 2
        LOC=2,6
        TEXT="COULOMBIC EFFICIENCY"
        SIZE=.1
EXECUTE
DELETE PLOT
MOVE TO 0,8.5
SUPPRESS TITLE 2
X SCALE
        START=0
        INCREMENT =100
Y SCALE
        START=40
        INCREMENT=10

```

```

X AXIS
  LENGTH=6
  SPACING=.5,1
  FORMAT="%%%"
  LABEL="CYCLE NUMBER"
  LSIZE=.1
Y AXIS
  LENGTH=6
  SPACING=.5,1
  FORMAT="%%%"
  LABEL="EFFICIENCY (%)"
  LSIZE=.1
DATA
PLOT
  SYMBOL=25
  SIZE=.1
EXECUTE
SUPPRESS X AXIS
SUPPRESS Y AXIS
DELETE PLOT
DATA
PLOT
  LINES
  SIZE=.1
TITLE 3
  LOC=2,6
  TEXT="ENERGY EFFICIENCY"
  SIZE=.1
EXECUTE
MOVE TO -3,7.5

```

Appendix I

COST ESTIMATES FOR COMPONENTS OF THE LMSC MOD 00 ZINC/FERROCYANIDE 20 MW, 100 MWh BATTERY

1. COST BASES

The following cost bases were used in these computations:

Reagent Costs

- The unit price for sodium ferrocyanide decahydrate was taken as \$0.69/kg based on (1) a price quotation* of \$0.37/lb in 24,000 lb lots and (2) that for 50,000 ton lots it was conservatively assumed that it could be obtained at a 15% discount from this quotation. Using EPRI cost estimating guidelines (ref 61), this cost was further reduced by 24.8%, which is the present percentage value of the salvaged reagent using a 15% per annum discount rate on the 100% value over 10 years. The 100% figure has not been adjusted for inflation. It is assumed that any slow decomposition of the ferrocyanide salt has been reversed periodically by religation over the 10-year period.
- Zinc was priced as zinc oxide at \$1.23/kg**. The salvage value is also taken as 24.8%, as described above.
- Sodium hydroxide was priced at \$0.24/kg (100% basis) as the 73% solution.**

Electrode Costs

- The negative electrode substrate is selected as Union Carbide CS graphite, at a cost of \$1.98/kg and a cutting efficiency of 0.514 and a thickness of 0.050 inch (ref 61). Density is 1.76 g/cc.

*American Cyanamid Co., Sept. 1979

**Chemical Marketing Reporter, May 28, 1979

- For positive electrode substrate, Airco Speer grade 37G porous carbon, at \$5.08/kg was selected. Following the electrode design used in the EDA zinc/chlorine battery, a cutting efficiency of 0.563 and electrode thickness of 0.050 inch were assumed (ref. 59). Density is 1.35 g/cm^3 .

Separator Costs

- Separator material was assumed to be an RAI Research Corp. grafted, cross-linked polyethylene or polypropylene material having essentially the same physical properties and pore size distribution as their teflon-based material, which we found to be satisfactory, except for cost and tear strength. For sales volume at the million square feet per year level, RAI Research Corp. projects a price of $\$1.00/\text{ft}^2$ for this material.*

Bus Bar Costs

The Gould figure of \$3.80/kWh is used for fabricated copper bus bars (ref. 58); after reduction by the ratio of the number of modules in the LMSC Mod 00 battery to that in the Mod 1 Gould battery (240/250). This gives a unit price of \$3.65/kWh. This price was reduced for the salvage value of the copper involved, $300 \times \frac{240}{250} = 288 \text{ kg/module}$. At \$1.43/kg, salvage value is $288 \times \$1.43 \times .248 = \$102/\text{module}$ or \$0.24/kWh. This gives a reduced cost of \$3.41/kWh.

2. COMPONENT COSTS

Electrodes

Referring to Table 4.7.1, negative electrode area is: $14 \text{ (plates/cell)} \times .0929 \text{ (m}^2\text{/plate)} \times 5 \text{ (cells/submodule)} \times 10 \text{ (submodule/module)} \times 15 \text{ (modules/string)} \times 16 \text{ (strings/battery)} = 1.56 \times 10^4 \text{ m}^2$.

*Private communication. R. Facciola, RAI Research Corp., May 1979

Total Cost:

$$15,600 \text{ (m}^2\text{/battery)} \times 0.05 \text{ (inch thickness)} \times 2.54 \text{ (cm/in.)} \\ \times \frac{10,000 \text{ (cm}^2\text{/m}^2\text{)}}{1000 \text{ (g/kg)}} \times \frac{1.76 \text{ (g/cc)}}{0.514 \text{ (cutting efficiency)}} = 67,840 \text{ kg graphite required}$$

$$67,840 \times \$1.98 = \underline{\$134,300} \text{ (negative electrode substrate)}$$

Positive electrode area is:

$$13 \text{ (plates/cell)} \times 0.0929 \text{ (m}^2\text{/plate)} \times 12,000 \text{ (cells/battery)} = 1.45 \times 10^4 \text{ m}^2$$

Weight of porous carbon required:

$$1.45 \times 10^8 \text{ (cm}^2\text{)} \times \frac{1.35 \text{ (g/cc)} \times 0.05 \text{ (inch)} \times 2.54 \text{ (cm/in)}}{1000 \text{ (g/kg)} \times 0.563 \text{ cutting efficiency}} \\ = 44,157 \text{ kg porous carbon required}$$

$$44,160 \times \$5.08 = \$224,300 \text{ (positive electrode substrate)}$$

Total cost for all electrode material is: \$358,600
or: \$3.55/kWh

Separator

Separator material cost is \$1.00/ft² or \$10.76/m². Assume area of a separator is 5% larger than that of an electrode. Total area is: 0.0975 m²

$$26 \text{ (separators/cell)} \times 0.0975 \text{ m}^2\text{/separator} \times 12,000 \text{ (cells/battery)} = 30,420 \text{ m}^2$$

$$\text{Cost is: } 30,420 \times \$10.76 = \underline{\$327,300} \text{ or } \$3.24/\text{kWh}$$

Reagents

Cell voltage is 1.74 V, and turnaround coulombic efficiencies are 96% and 95%, for redox and zinc electrodes, respectively. Assume 3% excess capacity for each electrode.

- Sodium ferrocyanide decahydrate

$$1.03 \text{ (excess capacity)} \times \frac{1.01 \times 10^8 \text{ (watthrs/battery)} \times 484.0 \text{ (g/equiv)} \times 0.69 \text{ (\$/kg)}}{1.74 \text{ (V/cell)} \times 0.96 \text{ (eff.)} \times 26.8 \text{ (Ah/equiv)} \times 1000 \text{ (g/kg)}}$$

$$= \$776,100$$

This cost is reduced by 24.8% for salvage value.

$$\$776,100 \times 0.752 = 583,600 \text{ redox reactant cost or } \$5.78/\text{kWh}$$

- Zinc Oxide

$$1.03 \text{ (excess capacity)} \times \frac{1.01 \times 10^8 \text{ (watts/battery)} \times 40.7 \text{ (g/equiv)} \times 1.23 \text{ (\$/kg)}}{1.74 \text{ (V/cell)} \times 0.95 \text{ (eff.)} \times 26.8 \text{ (Ah/equiv.)} \times 1000 \text{ (g/kg)}}$$

$$= \$117,600$$

This cost is reduced by 24.8% for salvage value.

$$\$117,600 \times 0.752 = \$88,400 \text{ or } 0.88 \text{ kWh}$$

- Sodium Hydroxide

Assume electrolyte volume is battery cell volume plus piping volume plus 50% of electrolyte storage tank volume. Assume a 5N NaOH electrolyte (See below storage tanks for volume estimation).

Cell electrolyte volume is: $0.35 \text{ (cm/interelectrode gap)}^{(\text{Ref. 58})} \times 0.0929 \text{ (m}^2/\text{electrode)}$
 $\times 26 \text{ (interelectrode gap/cell)} \times 10^4 \text{ (cm}^2/\text{m}^2)$
 $= 8454 \text{ cm}^3/\text{cell}$

$$\frac{8454 \text{ (cm}^3) \times 12,000 \text{ cells/battery}}{10^6 \text{ (cm}^3/\text{m}^3)} = 101.4 \text{ m}^3/\text{battery}$$

Piping contains 12.5 m^3

Half of Tankage Volume is 797.0 m^3

Cost is: $\frac{910.9 \text{ (m}^3) \times 10^3 \text{ (1/m}^3) \times 5 \text{ (equiv/l)} \times 40 \text{ g/equiv} \times 0.24 \text{ (\$/kg)}}{1000 \text{ (g/kg)}}$

$= \$43,720 \text{ or } \$0.43/\text{kWh}$

Reactant Tanks

● Catholyte Tank

Compute volume of solid $\text{Na}_4\text{Fe(CN)}_6 \cdot 10 \text{ H}_2\text{O}$ used:

$$\frac{776,100 \text{ (\$/batt)} \times 454 \text{ (g/lb)}}{0.31 \text{ (\$/lb)} \times 10^6 \text{ (cm}^3/\text{m}^3) \times 1.458 \text{ (g/cm}^3)} = 779.6 \text{ m}^3$$

In crystal form, assume 50% porosity. Then need tank volume of $2 \times 780 = 1560 \text{ m}^3$.
 Half of this volume would be electrolyte. Assume four cylindrical tanks, 20 feet tall. Diameter is:

$$((1560) (20 \times \pi \times 0.30455)^{-1})^{1/2}$$

$= 9.03 \text{ m or } 29.7 \text{ ft.}$

$$\text{Area of tank} = \frac{2\pi d^2}{4} + \pi dh$$

$$= \frac{2\pi(9.03)^2}{4} + \pi(9.03)(20)(0.30455) = 300.9 \text{ m}^2$$

Four tanks have total area: 1204 m^2

Anolyte Tank

$$\begin{aligned}\text{Volume is } & \frac{117,600 (\$/\text{batt}) \times 454 (\text{g/lb})}{0.56 (\$/\text{lb}) \times 1000 (\text{cm}^3/\text{l}) \times 5.47 (\text{g/cm}^3)} \\ & = 17,430 \text{ l or } \underline{17.4 \text{ m}^3}\end{aligned}$$

Double this for 50% solids porosity and double again for two tanks (one empty)

$$17.4 \times 4 = 69.6 \text{ m}^3 \text{ total volume}$$

Diameter of each tank is taken as 10 ft; with a corresponding height of 14.9 ft.

$$\begin{aligned}\text{Tankage area is: } & 2\left(\frac{2\pi d^2}{4} + \pi dh\right) \\ & = 2\left(\frac{2\pi(3.05)^2}{4} + (3.05\pi)(4.57)\right) = 116.8 \text{ m}^2\end{aligned}$$

$$\text{Total tankage area: } 1204 + 117 = 1321 \text{ m}^2$$

Using structural carbon steel at \$0.47/kg with a 3/8 inch wall thickness, cost is:

$$\begin{aligned}& \frac{2.54 (\text{cm/in}) \times 0.47 (\$/\text{kg}) \times 1321 (\text{m}^2) \times 0.375 (\text{in}) \times 10^4 (\text{cm}^2/\text{m}^2) \times 7.87 (\text{g/cm}^3)}{1000 (\text{g/kg})} \\ & = \$46,540\end{aligned}$$

Tanks will be lined with sulfonated polyethylene lining at \$53.80/m² (ref. 57).
Hence, add: $53.80 \times 1321 = \$71,070$ for a total cost of \$117,600 or \$1.16/kWh.

Filters, Heat Exchangers, and Pumps

These costs are taken as those given in the NASA-LeRC Cost Estimate (ref. 57) increased by the factor 100/85 to account for increased energy stored in a 100 MWh plant.

Filters (0.48 m ³ /sec)	\$82,500	or	\$0.82/kWh
---------------------------------------	----------	----	------------

Heat Exchangers (stainless steel tubes at \$43/m ²)	\$58,800	or	0.58/kWh
---	----------	----	----------

Pumps @ \$48,000/96.9 kW
in stainless steel

4 @ 50 kW and 2 @ 5kW	\$112,000	or	\$1.11 kWh
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Racks

EDA (ref. 59) Costs (adjusted upward by the power density ratio for the two systems, 1.25) were used for the racks upon which the battery modules are mounted, each string being three tiers high by five modules long. Two such strings are placed adjacent to each other for a double string, with electrolytes piped to each doubled string. Strings are 55 feet long by five feet wide by 18 feet in height.

These costs are:

	<u>Materials</u>	<u>Labor</u>
Rack structure	\$148,500	
Rack Assembly		\$54,000
Weatherproof panels (@ \$2.00/ft ² est.)	68,700	
Panel Assembly (est.)		4,000
	<u>\$217,200</u>	<u>\$58,000</u>
	<u>\$2.15 kWh</u>	<u>\$0.57/kWh</u>

Miscellaneous Costs2000 Ah/m² basis

	<u>Materials</u>	<u>Labor</u>
		(\$/kWh)
Internal plumbing (ref. 58)	2.00	-
Cell Components (other than electrodes and separators (ref. 58)	1.87	-
Cell Assembly (ref. 58)	-	0.36
Cell Stack Assembly (ref. 58)	-	0.27
Module Assembly (ref. 58)	-	0.30
External Plumbing (racks) Pipe and valves (est.)	0.32	0.10
Tanks (fabrication and erection) (est.)		0.25
Rack and Panel Assembly (ref. 59)	-	0.58
Process Equipment (for Electrode fab) (ref. 59)	4.55	
Physical Plant (ref. 59)	1.25	
Floor Space Rental (ref. 59)	0.19	

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