

NBS CIRCULAR *562*

SUPPLEMENT 1

**Bibliography of Research
on Deuterium and Tritium Compounds
1953 and 1954**

UNITED STATES DEPARTMENT OF COMMERCE

NATIONAL BUREAU OF STANDARDS

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Information on the Bureau's publications can be found in NBS Circular 460, Publications of the National Bureau of Standards (\$1.25) and its Supplement (\$0.75), available from the Superintendent of Documents, Government Printing Office, Washington 25, D. C.

NBS-C-562 (Supp. 1)

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1953 and 1954**

Virginia R. Johnson, Lawrence M. Brown
and Abraham S. Friedman



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II

Preface

Late in 1951, the Thermodynamics Section of the National Bureau of Standards became actively engaged in an Isotope Exchange Data Program, sponsored by the Atomic Energy Commission Division of Research, which involved, in part, the preparation of bibliographies of research on the hydrogen isotopes for the years subsequent to 1945. This bibliography supplements Circular 562, which covered the years 1945 to 1952. The importance of the hydrogen isotopes as research tools and the utility of these bibliographies as an aid to scientific research relating to them has prompted the compilation of this Circular.

A. V. ASTIN, *Director.*



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NBS-C-562 (SUPPL.)

Bibliography of Research on Deuterium and Tritium Compounds

Virginia R. Johnson, Lawrence M. Brown, and Abraham S. Friedman

A bibliography of 720 references to published research on the properties of deuterium and tritium compounds is given. The subject matter of each entry in the bibliography is indicated by letter codes related to a list of broad subject headings as shown in the Introduction. An index of deuterium and tritium compounds and a subject index are included.

1. Introduction

This Circular is a bibliography of research on deuterium and tritium compounds for the years 1953 and 1954. It is divided into three sections: the first contains a bibliography and author index, and the second and third contain the subject and compound indexes, respectively.

The 720 references were obtained from Chemical Abstracts for the years 1953 to 1954, inclusive.

The references in the bibliography are arranged alphabetically according to the last name of the leading author. The names of the coauthors are also included and are cross-referenced.

Each entry in the bibliography is numbered by a letter-number symbol that corresponds to the first letter of the leading author's last name and the numerical listing of the reference under that letter. This numbering system is used to identify the entries in the compound and subject indexes.

The subject matter of the references is indicated at the end of each entry by means of one or more of the letter codes related to the subject categories shown in the Principal Topics and Subject Code. In the letter codes the main headings of the subject categories are designated by two letters of the heading title, and the subheadings are represented by these two letters and one or two additional, appropriate letters. For example, the code symbol for Chemical Kinetics is Ki and that for the kinetics of photochemical reactions is KiP. The subject content was determined generally from the abstracts of the references; the original articles were consulted when the abstracts were not sufficiently informative. Review articles and abstracts are indicated as such. Research articles related to nuclear properties have been excluded, except where other pertinent information on deuterium or tritium was reported. Also the topic of Isotope Effects has been restricted pri-

marily to the isotopes of hydrogen, whereas the topic of Isotope Separation still contains references to works on the separation of isotopes of various elements.

In preparing the compound index, the nomenclature of Lange's Handbook of Chemistry¹ has been followed. However, common synonyms of many of the compound names used therein have been entered and cross-indexed. Compounds containing tritium have been named in accordance with the modified Boughton system,² in which the name of the hydrogen compound is followed by "-t". No subscript has been appended to the "-t" to indicate the extent of tritium substitution. The chemical formula for each compound containing deuterium or tritium is given.

No listing has been made in the compound index under Water or Water-t for the use of water as a solvent. A lower case "s" has been used to indicate other substances used as solvents (see, for example, under Ethyl alcohol). In addition, no listing has been made in the compound index for Hydrogen, Tritium, Water, or Water-t under the subjects of Abundance, Analytical Methods, General and Review, Nuclear Properties, Separation and Atomic Spectra. References to such research may be found under these headings in the subject index.

The arrangement and the methods of coding and indexing used here and in Circular 562 are consistent with those in the Bibliography of Research on Heavy Hydrogen Compounds, by Kimball, Urey, and Kirshenbaum,³ to which this work and Circular 562 are sequels.

¹ Handbook Publishers, Inc., Sandusky, Ohio (1949).

² E. J. Crane, Ind. Eng. Chem. News, Ed. 13, 200-01 (1935).

³ McGraw-Hill Book Company, Inc., New York, N. Y., Ed. 1 (1949).

2. Principal Topics and Subject Code

Ab	Abundance. AbG Geological. AbO Organic.	Me	Mechanical Properties. MeAc Acoustic properties. MeD Density and molar volume. MeDf Diffusion. MeSt Surface tension. MeV Viscosity.
Ad	Adsorption and Sorption. AdC Chromatography. AdG Gases on solids. AdL Liquids on solids.	No	Nomenclature.
An	Analytical Methods. AnC Counters, cloud chambers, electrometers, ionization chambers, and photographic emulsions. AnCl Colorimetric methods. AnDn Density methods. AnEl Optical rotation. AnMg Magnetic and magneto-optic methods. AnMs Mass spectrograph and mass spectrometer. AnRf Refractive index. AnSp Absorption spectra. AnTh Thermal conduction.	Nu	Nuclear Properties. NuB Beta ray spectra. NuH Hyperfine structure. NuIn Interactions (absorption of radiation, ranges, and scattering). NuM Masses and binding energies. NuMg Magnetic moments. NuQ Quadrupole moments. NuR Reactions. NuRe Magnetic resonances. NuS Spins, states, and wave functions. NuSt Statistics.
Bi	Biological Effects of Deuterium and Tritium Compounds and of Deuterons and Tritons. BiB Botanical. BiC Biochemical. BiZ Zoological.	Sd	Solid State. SdCr Crystal structure (including electron, neutron, and X-ray diffraction). SdEc Elastic constants. SdEl Electromagnetic properties. SdNu Nuclear properties. SdSp Spectra. SdTr Transitions (including phase transitions).
Ec	Electrochemical Properties. EcC Conductivities and mobilities. EcO Overvoltage. EcP Electrode potentials.	Se	Isotope Separation. SeAc Acoustics. SeAd Adsorption (including chromatography and ion exchange). SeCf Centrifuging. SeCh Chemical reaction. SeCr Crystallization. SeDf Diffusion (including thermal diffusion). SeDs Distillation. SeEl Electrolysis. SeEm Electromagnetic methods. SeMs Mass spectrometer and mass spectrograph. SeSo Solubility.
El	Electromagnetic and Optical Properties (Except Spectra). ElC Color Effects. ElD Dielectric constants and dipole moments. ElGd Gas discharges. ElMg Magnetic susceptibilities and Curie constants. ElMm Magnetic moments. ElMr Magnetic rotatory power. ElP Polarization. ElRe Paramagnetic resonances. ElRf Refractive index and molar refraction. ElRo Optical rotatory power. ElSc Light scattering. ElT Relaxation times.	So	Solubility. SoG Gases in solids. SoH In H ₂ O, HDO, and D ₂ O. Sol In inorganic solvents. SoO In organic solvents.
Eq	Chemical Equilibria. EqG Gaseous. EqH Heterogeneous. EqI Ionic. EqL Liquid and solution.	Sp	Spectra and Spectroscopic Constants. SpA Atomic (line). SpEl Molecular electronic. SpFl Fluorescence. SpM Microwave. SpVi Vibrational (including Raman). SpX X-ray.
Ge	General and Review.	Sr	Mass Spectroscopy and Mass Spectrography.
In	Indicator and Tracer Techniques. InA Age determinations. InBi Biological. InKi Reaction kinetics. InSo Solubility determinations. InSp Spectra. InSt Structure determinations.	St	Molecular Structure. StA Molecular association. StD Molecular constants (interatomic distances, bond angles, moments of inertia, and force constants). StDi Electron, neutron, and X-ray diffraction.
Is	Isotope Effects. IsCr Crystal structure. IsEl Electromagnetic properties. IsEq Chemical equilibria. IsKi Reaction kinetics. IsMs Mass spectra. IsSp Spectra. IsTh Thermodynamic properties.	Sy	Synthesis and Preparation of Compounds.
Ki	Chemical Kinetics. KiB Biochemical. KiG Gaseous. KiH Heterogeneous. KiI Ionic. KiL Liquid and solution. KiP Photochemical. KiR Radiochemical. KiS Solid state.	Th	Thermodynamic and Related Properties. ThD Diffusion and heat conduction. ThF Thermodynamic functions for pure substances and reactions between them (E , H , S , C_p , C_p , F , K , ΔH , ΔS , ΔE , ΔC_p , ΔF , data of state, and thermal expansion). ThP Phase equilibria (melting points, triple points, boiling points, heats of transition, critical constants, and vapor pressures). ThS Statistical mechanics and statistical thermodynamics. ThSo Properties of solutions (activities, fugacities, pH , vapor pressures, heats of solution and dilution, and colligative properties).

3. Bibliography and Author Index

A

- Aamodt, L. C. See Miller, S. L.
 Ahlberg, J. E. Stabilization of reduced metal catalysts.
A1 U. S. Patent 2,677,668, May 4, 1954. C. A. 48:9586 h. **KiH**.
 Albenesius, E. L. See Roe, A.
 Alder, B. J. See Harris, F. E.
 Alford, W. P. See Hamilton, D. R.
 Allen, H. C., Jr. Pure quadrupole spectra of molecular
A2 crystals. *J. Phys. Chem.* **57**, 501-04 (1953); C.A. 47:7888 c. **SpM**.
 Allen, I. See Roberts, J. D.
 Allen, K. W., and Almquist, E. Recovery and purification of tritium and helium³ used in ion accelerators. *Rev. Sci. Instr.* **24**, 70-71 (1953); C.A. 47:7908 g. **SeAd**.
 Almin, K. E. The reaction between cellulose and heavy
A4 water. *Svensk Papperstidn.* **55**, 767-70 (1952); C.A. 48:11777 h. **EqL, KiL, SpVi**.
 Almquist, E. See Allen, K. W.
 Alten, L. van. Infrared intensities and bond moments in
A5 C₂H₂, C₂D₂, and C₂HD. Univ. Microfilms (Ann Arbor, Mich.), Pub. No. 8107, 102 pp. Dissertation *Abstr.* **14**, 770-71 (1954); C.A. 48:9811 g. **SpVi**.
 Andersen, F. A., and Bak, B. Infrared absorption spectra
A6 of SiD₃Cl, SiH₃F, and SiD₃F. *Acta. Chem. Scand.* **8**, 738-43 (1954); C.A. 48:13427 f. **SpVi, StD**.
 Anderson, J. R., and Kemball, C. Catalysis on evaporated metal films. III. The efficiency of different metals for the reaction between ethane and deuterium. *Proc. Roy. Soc. (London)* **A223**, 361-77 (1954); C.A. 48:7965 g. **EqH, KiH**.
 Anderson, R. C. See Cerwonka, E.
 Anderson, R. C., Delabarre, Y., and Bothner-By, A. A.
A8 Chemical analysis and isotopic assay of organic compounds. *Anal. Chem.* **24**, 1298-1303 (1952); C.A. 47:1543 b. **An**.
 Andrew, E. R., and Eades, R. G. Separation of the intramolecular and intermolecular contributions to the second moment of the nuclear magnetic resonance spectrum. *Proc. Phys. Soc. (London)* **66A**, 415-17 (1953); C.A. 47:9143 e. **NuRe, Sd**.
A10 A nuclear magnetic resonance investigation of three solid benzenes. *Proc. Roy. Soc. (London)* **218A**, 537-52 (1953); C.A. 47:9768 g. **NuRe, Sd**.
 Andrussov, L. Thermal conductivity, viscosity, and
A11 diffusion in the gas phase. VII. Hydrogen, deuterium, helium, neon, and their mixtures. *J. chim. phys.* **49**, 599-604 (1952); C.A. 47:9699 h. **Ge, ThD, MeV, MeDf**.
A12 Thermal conductivity, viscosity, and diffusion in the gas phase. VIII. Calculation of temperature coefficients. *Z. Elektrochem.* **57**, 124-30 (1953); C.A. 47:9699 i. **Ge, ThD, MeV, MeDf**.
 Anger, H. O. See Rosenthal, D. J.
 Angler, H. O. See Tobias, C. A.
 Antipina, S. A. See Maslov, P. G.
 Arison, B. See Trenner, N. R.
 Armstrong, G. T., Brickwedde, F. G., and Scott, R. R.
A13 The vapor pressures of the deuteriomethanes. *J. Chem. Phys.* **21**, 1297-98 (1953); C.A. 47:10932 h. **ThP**.
 Arnheim, J. See Eidinoff, M. L.
 Arnold, J. R. See Johnston, W. H.
 Aschner, T. C. See Doering, W. v. E.
 Auty, R. P., and Cole, R. H. Dielectric properties of ice
A14 and solid D₂O. *J. Chem. Phys.* **20**, 1309-14 (1952); C. A. 47:9694 g. **EID, SdEl**.
 Avery, E. C. See Smaller, B.
 Azcona, J. M. L. de See López de Azcona, J. M.
 Azevedo, M. D. de See Deodata de Azevedo, M.

B

- Baden, H. See Edelman, I. S.
 Bader, R. F. W. See Sheppard, W. A.
 Badger, R. M. See Moore, G. E.
 Bak, B. See Anderson, F. A.
 Bak, B., Bruhn, J., and Rastrup-Andersen, J. Microwave spectrum and structure of SiD₃F. *J. Chem. Phys.* **21**, 752-53 (1953); C.A. 47:6770 f. **SpM, StD**.
B1 Microwave spectrum and structure of SiD₃Cl. *J. Chem. Phys.* **21**, 753-54 (1953); C.A. 47:7324 a. **SpM, StD**.
B2 Stereochemistry of silicon. *Acta Chem. Scand.* **8**, 367-73 (1954); C.A. 48:10437 a. **SpM, StD, Sy**.
 Ball, C. D. See Dewey, L. J.
 Ballou, J. E. See Thompson, R. C.
 Balmain, J. H., Folley, S. J., and Glascock, R. F. Stimulation by insulin of in vitro fat synthesis by mammary tissue studied with carbon¹⁴ and tritium. *Nature*, **168**, 1083-84 (1951); C.A. 47:4462 e. **BiC, InBi**.
B3 Relative utilization of glucose and acetate carbon for lipogenesis by mammary gland slices, studied with tritium, C¹³ and C¹⁴. *Biochem. J.* **56**, 234-39 (1954); C. A. 48:5240 c. **InBi**.
 Bannard, R. A. B., Morse, A. T., and Leitch, L. C.
B4 Synthesis of organic deuterium compounds. VI. Formaldehyde-d₂ and formaldehyde-d. *Can. J. Chem.* **31**, 351-60 (1953); C.A. 48:2560 b. **Sy**.
 Bardwell, J., and Herzberg, G. Laboratory experiments
B7 on the detectability of silane (SiH₄) and methyl deuteride (CH₃D) in the atmospheres of the outer planets. *Astrophys. J.* **117**, 462-65 (1953); C.A. 47:7314 d. **SpVi**.
 Barnes, F. W., Jr. See Hilton, M. A.
 Barnes, R. G. See Bray, P. J.; Harrick, N. J.
 Barnes, R. G., and Bray, P. J. Radiofrequency spectroscopy of deuterium. *Phys. Rev.* **87**, 229 (1952); C. A. 48:8025 b. **NuRe, NuMg**.
B8 Variations of hydrogen rotational magnetic moments with rotational quantum number and with isotopic mass. *Phys. Rev.* **94**, 893-902 (1954); C.A. 48:9190 a. **NuRe, NuMg, NuQ, ElMg**.
 Barnett, M. K. See Heiks, J. R.
 Barr, F. T. Catalytic apparatus for isotope exchange.
B10 U. S. Patent 2,676,875, Apr. 27, 1954. C.A. 48:P9123 a. **EqH, SeCh**.
 Barrow, G. M. Heat capacity, gas imperfection, infrared spectra, and internal rotation barriers of ethyl alcohol. *J. Chem. Phys.* **20**, 1739-44 (1952); C.A. 47:4722 a. **SpVi**.
 See Searles, S.
 Barsh, M. K. See Wilmarth, W. K.
 Bartlett, P. D., and Tate, F. A. The polymerization of allyl compounds. VI. The polymerization of allyl-1,1-d₂ acetate and the mechanism of its chain termination. *J. Am. Chem. Soc.* **75**, 91-95 (1953); C.A. 48:580 b. **KiL**.
 Barton, D. H. R., and Mayo, P. De. Triterpenoids.
B13 XIII. Phyllanthol, the first hexacarboxylic triterpenoid. *J. Chem. Soc.* 2178-81 (1953); C.A. 48:12712 e. **InSt**.
 Bassham, J. A. See Calvin, M.
 Bauer, E. Effects of the dielectric relaxation during
B14 phase changes in polar materials. *Compt. rend. reunion ann. avec comm. thermodynam.*, Union intern. phys. (Paris) 1952; C.A. 47:7273 h. **EID, SdCr, SdTr**.
 Becker, E. W., Misenta, R., and Stehl, O. Difference in
B15 viscosity of ortho- and para-deuterium at low temperatures. *Phys. Rev.* **91**, 414 (1953); C.A. 47:10930 f. **MeV**.

- B16** Viscosity difference of ortho and para deuterium at low temperatures. *Z. Physik* **136**, 457-62 (1953); C.A. **48**:4910 g. **MeV**.
- Beckerle, J. C. Exchange between rotational and translational energy in gas molecules. *J. Chem. Phys.* **21**, 2034-41 (1953); C.A. **48**:1751 g. **ElT**.
- Beckey, H. D., and Groth, W. E. A new method of isotope enrichment in the transition region between Knudsen and Poiseuille flow. *Z. Naturforsch.* **7a**, 474-80 (1952); C.A. **48**:50 i. **SeDf**.
- Beckey, H. D., Groth, W. E., and Welge, K. H. Enrichment of isotopic molecules in the direct-current glow discharge. II. The multiplication process. Enrichment of heavy hydrogen. *Z. Naturforsch.* **8a**, 556-62 (1953); C.A. **48**:4305 g. **SeEm**.
- Beeck, O. See Stevenson, D. P.
- Beers, Y., and Weisbaum, S. An ultra-high-frequency rotation line of HDO. *Phys. Rev.* **91**, 1014 (1953); C.A. **47**:11991 b. **SpM**.
- Beischer, D. E. Absolute β -measurement of tritium monolayers. *Nucleonics* **11**, No. 12, 24-27 (1953); C.A. **48**:3802 g. **AnC**.
- Belford, R. L. See Yankwich, P. E.
- Bell, E. E. See Stroup, R. E.
- Bell, F. K. See Carr, C. J.
- Benedict, W. S., Gailar, N., and Plyler, E. K. The vibration-rotation spectrum of D₂O. *J. Chem. Phys.* **21**, 1301-02 (1953); C.A. **47**:11002 c. **SpVi**.
- B23** The vibration-rotation spectrum of HD. *J. Chem. Phys.* **21**, 1302-03 (1953); C.A. **47**:11002 d. **SpVi**.
- Bennett, J. M. See Wiggins, T. A.
- Bennett, W. See Bernstein, I. A.
- Benoist, S. Attempt to identify Schuler's unidentified system of the OH molecule. *Compt. rend.* **238**, 883-85 (1954); C.A. **48**:6249 c. **SpEl**.
- Berglund-Larsson, U., and Melander, L. Isotope effect of hydrogen and mechanism of aromatic sulfonation. *Arkiv. Kemi* **6**, 219-29 (1953); C.A. **48**:11377 i. **IsKi, KiL**.
- Bernal, M. J. M. Analytical wave functions for methane and the ammonium ion. *Proc. Phys. Soc. (London)* **66A**, 514-18 (1953); C.A. **47**:9687 f. **StD**.
- Bernas, A. See Fiquet, F.
- Bernhard, K. Absorption of aliphatic hydrocarbons, carotene, and vitamin A by the rat. *Fette u. Seifen* **55**, 160-66 (1953); C.A. **47**:8861 e. **InBi**.
- Bernhard, K., Gloor, U., and Scheitlin, E. Metabolism in the animal body of aliphatic hydrocarbons with 8-18 carbon atoms. *Helv. Chim. Acta* **35**, 1908-13 (1952); C.A. **47**:5499 b. **BiC, InBi, Sy**.
- Bernstein, H. J. See Davidson, D. W.
- Bernstein, H. J., and Pullin, A. D. E. Vibration spectra of cis- and trans-dichloroethylene-d₁. *Can. J. Chem.* **30**, 963-72 (1952); C.A. **47**:3116 g. **SpVi**.
- Bernstein, H. J., Pullin, A. D. E., Rabinovitch, B. S., and Larson, N. R. Rotational isomerism and the vibrational spectrum of sym-dideuteriodibromethane. *J. Chem. Phys.* **20**, 1227-31 (1952); C.A. **47**:9775 h. **SpVi, Sy**.
- Bernstein, I. A., Bennett, W., and Fields, M. Triation of multiple bonds—synthesis of tritiated styrene. *J. Am. Chem. Soc.* **74**, 5763-64 (1952); C.A. **48**:4944 a. **Sy**.
- Bernstein, R. B. Enrichment of isotopes by difference in rates for irreversible isotopic reactions. *J. Phys. Chem.* **56**, 893-96 (1952); C.A. **47**:2053 e. **IsKi, SeCh**.
See Gordus, A. A.; Pontarelli, D. A.; Semelek, G. P.
- Bernstein, R. B., Gordus, A. A., and Cleveland, F. F. **B33** Substituted methanes. XII. Infrared spectral data for deuteriotrichloromethane. *J. Chem. Phys.* **20**, 1979 (1952); C.A. **47**:2600 a. **SpVi**.
- Berry, C. E. See Washburn, H. W.
- Berstein, I. A., Bennett, W., Fields, M., and Farmer, E. C. **B34** Preparation of a tritiated self-luminescent material. *Nucleonics* **11**, No. 2, 64-66 (1953); C.A. **47**:5257 a. **Sy, SpFl**.
- Berstein, I. A. See Farmer, E. C.
- Bevege, E. E., Renaud, R., and Leitch, L. C. Synthesis of **B35** organic deuterium compounds. IX. Bistriderteromethyl mercury. *Can. J. Chem.* **31**, 1259-60 (1953); C.A. **48**:3248 b. **Sy**.
- Bévuillard, P. Applications of isotopes to the study of **B36** structures and mechanisms of reactions. A review. *Bull. soc. chim. France*, D40-8, D55-79 (1954); C.A. **48**:11886 c. **Ge**.
- Bianco, D. R. See Jen, C. K.; Massey, J. T.
- Biddulph, O. Translocation of radioactive nutrients in **B37** plants. U. S. Atomic Energy Comm. TID-5098 (Apr., 1953); C.A. **48**:4637 i. **BiB**.
- Bier, A. See Ketelaar, J. A. A.
- Bieri, R. See Mattauich, J.
- Bigeleisen, J. Disproportionation in deuteriated ammonia. **B38** *J. Chem. Phys.* **20**, 1495-96 (1952); C.A. **47**:2587 d. **EqG**.
- B39** The effect of isotopic substitution on the entropy, enthalpy, and heat capacity of ideal gases. I. Systems in thermal equilibrium. II. Chemically reacting systems. *J. Chem. Phys.* **21**, 1333-39 (1953); C.A. **47**:11855 a. **IsEq, IsTh, IsKi, ThF, ThS**.
- B40** The effects of isotopic substitution on the rates of chemical reactions. *J. Phys. Chem.* **56**, 823-28 (1952); C.A. **47**:2021 i. **Ge, IsKi**.
- B41** Isotopes. A review. *Ann. Rev. Nuclear Sci. (Annual Revs., Inc., Stanford, Calif.)* **2**, 1-40 (1953); C.A. **47**:2045 d. **Ge**.
- B42** Theoretical evaluation of the N-isotope effect in the thermal deamination of phthalamide. *Can. J. Chem.* **30**, 443-47 (1952); C.A. **47**:3821 e. **IsKi**.
See Perlman, M. L.; Weston, R. E., Jr.
- Bigeleisen, J., Perlman, M. L., and Prosser, H. C. **B43** Conversion of hydrogenic materials to hydrogen for isotopic analysis. *Anal. Chem.* **24**, 1356-57 (1952); C. A. **47**:1531 e. **AnMs**.
- Bigeleisen, J., and Wolfsberg, M. Temperature-independent factor in the relative rates of isotopic three-center reactions. *J. Chem. Phys.* **21**, 1972-74 (1953); C.A. **48**:1848 i. **IsKi**.
- B45** Fractionation of the carbon isotopes in decarboxylation reactions. VI. Comparison of the intermolecular isotope effects of a pair of isotopic isomers. *J. Chem. Phys.* **21**, 2120-21 (1953); C.A. **48**:4944 f. **IsKi**.
- B46** Errata: Temperature-independent factor in the relative rates of isotopic three-center reactions. *J. Chem. Phys.* **22**, 1264 (1954); C.A. **48**:12570 c. **IsKi**.
- Biggs, M. W., and Kritchevsky, D. The suitability of **B47** tritium-labeled cholesterol for the study of cholesterol metabolism. *Arch. Biochem. Biophys.* **36**, 430-33 (1952); C.A. **47**:736 g. **InBi**.
- Biggs, M. W., Lemmon, R. M., and Pierce, F. T., Jr. **B48** Δ^1 -Cholesterol metabolism in the rabbit. *Arch. Biochem. Biophys.* **51**, 155-60 (1954); C.A. **48**:12998 b. **BiC**.
- Billings, B. H., Hitchcock, W. J., and Zelikoff, M. The photochemical separation of isotopes. *J. Chem. Phys.* **21**, 1762-66 (1953); C.A. **48**:454 d. **SeCh**.
- Bird, G. R., and Townes, C. H. Sulfur bonds and the quadrupole moments of oxygen, sulfur, and selenium isotopes. *Phys. Rev.* **94**, 1203-08 (1954); C.A. **48**:9126 a. **SpM**.
- Birnbaum, G., and Maryott, A. A. Change in the inversion spectrum of ND₃ from resonant to non-resonant absorption. *Phys. Rev.* **92**, 270-73 (1953); C.A. **48**:1813 i. **SpM**.
- Blatt, J. M. The β -decay of the triton. *Phys. Rev.* **89**, 86-92 (1953); C.A. **47**:4209 b. **NuB**.
- Bleaney, B., Bowers, K. D., and Trenam, R. S. The nuclear electric quadrupole moments of copper ^{63,65}. *Proc. Phys. Soc. (London)* **66A**, 410-12 (1953); C.A. **47**:9144 e. **NuQ**.
- Block, K. See Langdon, R. G.
- Blout, E. R. See Lenormant, H.
- Blue, R. W. The liberational heat capacity of ice and of

- B54** heavy ice. *J. Chem. Phys.* **22**, 280-83 (1954); C.A. **48**:6227 f. **ThF, IsTh.**
- Boato, G. The primeval cosmic abundance of deuterium. *Phys. Rev.* **93**, 640-41 (1954); C.A. **48**:6243 d. **AbG.**
- Boato, G., Careri, G., Cimino, A., Molinari, E., and Volpi, G. G. A perturbing factor on the kinetics of the homogeneous hydrogen-deuterium exchange reaction. *Nuovo cimento* **10**, 993-94 (1953); C.A. **47**:10973 b. **EqG, KiG.**
- Bockris, J. O'M. Parameters of electrode kinetics. *Natl. Bur. Standards Circ. No. 524*, 243-62 (1953); C.A. **48**:4338 b. **EcP, KiH.**
- Boer, D. H. W., and Borg, W. A. J. Mass spectrometric determination of deuterium in organic compounds. *Rec. trav. chim.* **71**, 120-24 (1952); C.A. **47**:7374 a. **AnMs.**
- Boer, D. H. W. den See Kögl, F.
- Boer, J. de Theories of the liquid state. *Proc. Roy. Soc. (London)* **215A**, 4-28 (1952); C.A. **47**:4156 d. **Th, ThD, ThF, ThP, ThS, SdTr.**
- Bok, L. D. C., and Cohen, M. D. The hydrogen exchange in aqueous solutions of sodium formate. *J. S. African Chem. Inst. [N. S.]* **4**, 37-50 (1951); C.A. **47**:6744 g. **EqL, KiL.**
- Bok, L. D. C., and Mitchell, L. G. The rate of hydrogen exchange in aqueous solutions of aminoacetic acid. *J. S. African Chem. Inst. [N. S.]* **4**, 51-55 (1951); C.A. **47**:6744 h. **EqL, KiL.**
- Bok, L. D. C., and Petters, L. B. Hydrogen-deuterium exchange in aqueous solutions of glycolic acid and of sodium formate. *J. Chem. Soc.* 1524-25 (1952); C.A. **46**:9389 e. **EqL, KiL.**
- Bond, G. C., Sheridan, J., and Whiffen, D. H. Heterogeneous catalysis. V. Exchange reactions involving dideuterioacetylene. *Trans. Faraday Soc.* **48**, 715-23 (1952); C.A. **47**:2583 i. **EqH, KiH.**
- Bond, G. C., and Turkevich, J. Reaction of propylene with deuterium over a platinum catalyst. *Trans-Faraday Soc.* **49**, 281-91 (1953); C.A. **47**:9741 c. **KiH, EqH.**
- Bond, V. P., Swift, M. N., Taketa, S. T., Welch, G. P., and Tobias, C. A. Indirect effects of localized deuterium irradiation of the rat. *Am. J. Physiol.* **174**, 259-63 (1953); C.A. **47**:11291 i. **BiZ.**
- Bonner, T. G., Bowyer, F., and Williams, G. Nitration in sulfuric acid. IX. The rates of nitration of nitrobenzene and pentadeuterionitrobenzene. *J. Chem. Soc.* 2650-52 (1953); C.A. **48**:1783 a. **KiL.**
- Boozer, C. E. See Lewis, E. S.
- Borg, W. A. J. See Boer, D. H. W.
- Bothner-By, A. A. See Anderson, R. C.
- Boudart, M. See Mikovsky, R. J.
- Bourns, A. N. See Sheppard, W. A.
- Bovey, L. F. H. Rotation-vibration spectra of diatomic and simple polyatomic molecules with long absorbing paths. X. The spectrum of trideuteriomethane in the photographic infrared. *J. Chem. Phys.* **21**, 830-36 (1953); C.A. **47**:7892 c. **SpVi, StD.**
- Bowers, K. D. See Bleaney, B.
- Bowyer, F. See Bonner, T. G.
- Boyd, D. R. J., and Thompson, H. W. A vibration-rotation band of monodeuteriomethane. *Proc. Roy. Soc. (London)* **216A**, 143-52 (1953); C.A. **47**:5797 c. **SpVi.**
- Boyd, D. R. J., and Thompson, H. W. Rotational constants of monodeuteriomethane. *Trans. Faraday Soc.* **49**, 1281-84 (1953); C.A. **48**:8048 d. **SpVi.**
- Bradlow, H. L. See Fukushima, D. K.
- Bradshaw, W. See Silverman, L.
- Brandes, G. See Goubeau, J.
- Braune, H., and Zehle, F. Self-diffusion of HCl and DCl. *Z. physik. Chem.* **49B**, 247-56 (1941); C.A. **36**:6388. **MeDf.**
- Bray, P. J. See Barnes, R. G.; Harrick, N. J.
- Bray, P. J., Barnes, R. G., and Harrick, N. J. Magnetic moment ratios in hydrogen-like molecules. *Phys. Rev.* **87**, 229 (1952); C.A. **48**:8040 f. **NuRe, NuMg.**
- Breger, I. A. See Burton, V. L.
- Breit, G., and Thaler, R. M. Relativistic corrections to the magnetic moments of nuclear particles. *Phys. Rev.* **89**, 1177-86 (1953); C.A. **47**:8525 f. **NuMg.**
- Brickman, W. J., Dunford, H. B., Tory, E. M., Morrison, J. L., and Brown, R. K. The reactivity of cellulose. II. Water sorption, heats of wetting, and the reactions with thallose ethylate in ether, nitration mixtures, and heavy water of cotton linters alternately wetted with water and dried. *Can. J. Chem.* **31**, 550-63 (1953); C. A. **47**:12800 h. **Ad, In.**
- Brickwedde, F. G. See Armstrong, G. T.
- Brin, G. P. See Krasnovskii, A. A.
- Brobeck, J. R. See Mankin, H.
- Brodskii, A. I. Mechanism of hydrogen exchange and some protolytic reactions in solutions. *Zhur. Obshchei Khim.* **24**, 413-22 (1954); C.A. **48**:8621 hg. **Ge, Eq, Ki.**
- See Gragerov, I. P.; Kukhtenko, I. I.
- Brodskii, A. I., and Kukhtenko, I. I. Kinetics and mechanism of hydrogen migration in phenols. *Zhur. Fiz. Khim.* **25**, 920-26 (1951); C.A. **47**:3814 f. **EqG, KiG.**
- Brodskii, A. I., and Sulima, L. V. Tautomerism of hypophosphorous and phosphorous acids. *Doklady Akad. Nauk. S.S.S.R.* **85**, 1277-80 (1952); C.A. **47**:930 e. **EqL, KiL, InSt.**
- Broeker, W. S. See Grosse, A. V.
- Broida, H. P. See Morowitz, H. J.
- Broida, H. P., and Gaydon, A. G. The mechanism of formation of OH, CH, and HCO in flame spectra with deuterium as tracer. *Proc. Roy. Soc. (London)* **218A**, 60-69 (1953); C.A. **47**:9780 g. **InKi, KiG, SpEl.**
- Broida, H. P., and Kane, W. R. Rotational intensity distributions of OH and OD in an electrodeless discharge through water vapor. *Phys. Rev.* **89**, 1053-59 (1953); C.A. **47**:8511 d. **ElGd, SpVi, SpEl.**
- Broida, H. P., and Morgan, G. H. Optical spectrophotometric analysis of hydrogen-deuterium mixtures in the presence of air. *Anal. Chem.* **24**, 799-804 (1952); C.A. **47**:6307 e. **AnSp, SpA.**
- Broida, H. P., Morowitz, H. J., and Selgin, M. Optical spectroscopic determination of hydrogen isotopes in aqueous mixtures. *J. Research Natl. Bur. Standards* **52**, 293-301 (1954); C.A. **48**:11972 e. **AnSp.**
- Brokaw, R. W., and Pease, R. N. Effect of water on the burning velocities of cyanogen-oxygen-argon mixtures. *J. Am. Chem. Soc.* **75**, 1454-57 (1953); C.A. **47**:10972 e. **InKi.**
- Brouckere, L. de See Prigogine, I.
- Brown, E. V. See Cerwonka, E.
- Brown, F., and Holland, D. A. Isotope effects: reaction at the carbonyl group. *Can. J. Chem.* **30**, 438-42 (1952); C.A. **47**:2149 h. **IsKi.**
- Brown, H. C., and Russell, G. A. The photochlorination of 2-methylpropane-2-d and α -d₂-toluene; the question of free radical rearrangement or exchange in substitution reactions. *Am. Chem. Soc.* **74**, 3995-98 (1952); C.A. **47**:4852 i. **KiP, Sy.**
- Brown, R. K. See Brickman, W. J.
- Brown, W. G. See Wilzbach, K. E.
- Brown, W. G., Kaplan, L., and Wilzbach, K. E. The exchange of hydrogen gas with lithium and sodium borohydrides. *J. Am. Chem. Soc.* **74**, 1343-44 (1952); C.A. **47**:4686 b. **EqH, SeCh.**
- Brun, J. Dissolving velocity of metals in deuterio-electrolytes. I. *Kgl. Norske Videnskab. Selskab. Forh.* **24**, 114-17 (1951); C.A. **47**:9112 h. **Ge, KiH, Sol.**
- B86** Dissolving velocity of metals in deuterioelectrolytes. II. *Kgl. Norske Videnskab. Selskab. Forh.* **24**, 118-21 (1951); C.A. **47**:9112 i. **Sol, KiH, Eco.**

- B87** Heavy water. A review. *Tidsskr. Kjemi, Bergvesen Met.* **13**, 156-61 (1953); *C.A.* **48**: 8608 c. **Ge.**
- Buess, R. See Prigogine, I.
- Bugnard, L., Chevallier, F., and Coursaget, J.
- B88** The preparation, extraction, and determination of cholesterol labeled with deuterium or carbon¹⁴. *J. physiol.* **45**, 463-69 (1953); *C.A.* **48**:1457 f. **Sy.**
- Burg, A. B. The preparation and characterization of some deuterio-boron compounds. *J. Am. Chem. Soc.* **74**, 1340-41 (1952); *C.A.* **48**:1247 h. **Sy.**
- B89** Burgerjon, J. J. New deuterium supply for cyclotron ion sources. *Rev. Sci. Inst.* **25**, 522-23 (1954); *C.A.* **48**:1344 a. **SeEl**
- Burgison, R. M. See Carr, C. J.
- Burke, B. F., and Strandberg, M. W. P. Zeeman effect in rotational spectra of asymmetric-rotor molecules. *Phys. Rev.* **90**, 303-08 (1953); *C.A.* **47**:9146 f. **SpM.**
- Burrus, C. A., Jr. See Gordy, W.
- Burrus, C. A., Jr., and Gordy, W. One-to-two-millimeter wave spectroscopy. III. NO and DI. *Phys. Rev.* **92**, 1437-39 (1953); *C.A.* **48**:4978 i. **SpM.**
- B92** Burrus, C. A., Jr., Jache, A., and Gordy, W. One-to-two millimeter wave spectroscopy. V. PH₃ and PD₃. *Phys. Rev.* **95**, 706-08 (1954); *C.A.* **48**:11929 d. **SpM.**
- Burrus, R. H. See Grisolia, S.
- Burton, M. See Gordon, S.; Patrick, W. N.; Wiener, H.
- Burton, M., and Patrick, W. N. Radiation chemistry of mixtures: cyclohexane and benzene-d₆. *J. Phys. Chem.* **58**, 421-23 (1954); *C.A.* **48**:11935 h. **KiR.**
- B94** Burton, V. L., and Breger, I. A. Radiochemical preparation of 8-heptadecene. *Science* **116**, 477-79 (1953); *C.A.* **48**:7531 d. **KiR, Sy.**
- B95** Burwell, R. L., Jr., and Briggs, W. S. Reaction of alkanes with hydrogen and deuterium. Racemization and exchange. *J. Am. Chem. Soc.* **74**, 5096-5102 (1952); *C.A.* **48**:6371 i. **EqH, KiH.**
- Buu-Hoi, Ng. ph. See Lacassagne, A.
- Byerrum, R. U. See Dewey, L. J.
- C**
- Cagle, F. W., Jr. See Eyring, H.
- Calvin, M. See Fry, A.
- Calvin, M., and Bassham, J. A. Studies in photosynthesis with isotopes. U. S. Atomic Energy Comm. TID-5098 (April 1953); *C.A.* **48**:4638 b. **Ge, InKi.**
- C1** Cardwell, H. M. E., Dunitz, J. D., and Orgel, L. E. The structure of the hydrogen maleate anion: a symmetric hydrogen bond? *J. Chem. Soc.*, 3740-42 (1953); *C.A.* **48**:4975 h. **SpVi, InSt.**
- Careri, G. See Boato, G.
- C2** Carr, C. J., Burgison, R. M., Bell, F. K., and Krantz, J. C., Jr. Anesthesia. XXXVI. Anesthesia with deuteriocarbons and cardiac automaticity. *Anesthesiology* **12**, 230-33 (1951); *C.A.* **47**:3976 e. **BiC.**
- C3** Carr, H. Y., and Purcell, E. M. Interaction between nuclear spins in HD gas. *Phys. Rev.* **88**, 415-16 (1952); *C.A.* **47**:1486 h. **NuRe, NuS, SpA.**
- Cartwright, W. F. See Oxley, C. L.
- C4** Caskey, F. E., and Driekamer, H. G. Thermal diffusion in isotopic mixtures in the critical region. *J. Chem. Phys.* **21**, 153-56 (1953); *C.A.* **47**:3640 h. **ThD.**
- C5** Cavert, H. M. Metabolism of carbon¹³-labeled substrates and kinetics of deuterium oxide exchange in the isolated heart. Univ. Microfilms (Ann Arbor, Mich.), Pub. No. 4924, 54 pp. Dissertation Abstr. **13**, 248-49 (1953); *C.A.* **47**:7074 a. **KiB.**
- See Johnson, J. A.
- Ceccaldi, M. See Lecompte, J.
- Cerral, E., Marchetti, C., Renzoui, R., Roseo, L., Silvestri, M., and Villani, S. A thermal method for concentrating heavy water. *Chem. Eng. Progr. Symposium Ser.* **50**, No. 11, 271-80 (1954); *C.A.* **48**:9757 f. **SeCh, SeDf.**
- Cerwonka, E., Anderson, R. C., and Brown, E. V. Mechanism of the Willgerodt reaction. III. Nature of the labile intermediate. *J. Am. Chem. Soc.* **75**, 30-33 (1953); *C.A.* **48**:585 h. **InKi.**
- C8** Chang, T. L. The density of heavy water. *Chinese Sci.* **2**, 185-90 (1951); *C.A.* **48**:5584 h. **MeD.**
- Chapiro, A. See Prevot-Bernas, A.
- Chapman-Andresen, C. Autoradiographs of algae and C10 ciliates exposed to tritiated water. *Exptl. Cell Research A*, **239-42** (1953); *C.A.* **47**:10038 e. **BiB.**
- Charette, J. See Hemptinne, M. de
- Charette, J., Courtoy, C., and Hemptinne, M. de. Infrared spectra of cis- and trans-C₂H₂D₂ at high dispersion. *Ann. soc. sci. Bruxelles Ser. I*, **67**, 122-30 (1953); *C.A.* **48**:1808 i. **SpVi.**
- C11** Charette, J., and Hemptinne, M. de. Infrared spectra of deuterium-substituted vinyl bromide. *Acad. roy. Belg., Bull. classe sci.* **38**, 934-56 (1952); *C.A.* **47**:4198 f. **SpVi.**
- C12** Charette, J., and Manneback, C. Potential function of ethylene. *Ann. soc. sci. Bruxelles, Ser. I*, **66**, 178-93 (1952); *C.A.* **47**:9687 d. **StD, SpVi.**
- Chester, P. F., and Dugdale, J. S. Melting curves of deuterium and hydrogen. *Phys. Rev.* **95**, 278-79 (1954); *C.A.* **48**:9776 d. **ThP.**
- Chevallier, F. See Bugnard, L.
- Chinard, F. P., and Enns, T. Preparation of water samples for deuterium analysis in the mass spectrometer. *Anal. Chem.* **25**, 1413-14 (1953); *C.A.* **48**:1078 f. **AnMs.**
- C15** C16 Relative rates of passage of deuterium and tritium oxides across capillary walls in the dog. *Am. J. Physiol.* **178**, 203-05 (1954); *C.A.* **48**:13900 d. **BiZ, InBi.**
- Cimino, A. See Boato, G.
- Clark, A. Oxides of the transition metals as catalysts. *Ind. Eng. Chem.* **45**, 1476-80 (1953); *C.A.* **47**:10151 h. **EqH.**
- C17** Cleek, G. W. See Hubbard, D.
- Cleveland, F. F. See Bernstein, R. B.; Pontarelli, D. A.
- Clewett, G. H. Chemical separation of stable isotopes. *Ann. Rev. of Nuclear Sci. (Annual Revs., Inc., Stanford, Calif.)* **1**, 293-300 (1952); *C.A.* **47**:409 h. **SeCh.**
- C18** Cline, J. F. Absorption and metabolism of tritium oxide and tritium gas by bean plants. *Plant Physiol.* **28**, 717-23 (1953); *C.A.* **48**:2190 gh. **BiB, BiC.**
- C19** Cohen, M., and Westheimer, F. H. The chronic acid oxidation of isopropyl alcohol in 86.5% acetic acid solution. The chemistry of the chlorochromate ion. *J. Am. Chem. Soc.* **74**, 4387-91 (1952); *C.A.* **47**:8633 i. **KiL.**
- C20** Cohen, M. D. See Bok, L. D. C.
- Cohen, V. W. Spectroscopy of radioactive molecules. A review. *Ann. N. Y. Acad. Sci.* **55**, 904-14 (1952); *C.A.* **47**:2605 g. **Ge, NuMg, NuQ, NuS, NuRe, Sp.**
- C21** Cohn, H., Ingold, C. K., and Poole, H. G. Infrared spectrum of nitric and deuterionitric acid vapor. Completion of identification of the fundamental frequencies. Entropy of nitric acid. Barrier resisting rotation of the hydroxyl group. *J. Chem. Soc.*, 4272-82 (1952); *C.A.* **47**:4740 d. **SpVi.**
- C22** Coillet, D. W., and Harris, G. M. Exchange reactions of methane and monodeuteriomethane with atomic deuterium. *J. Am. Chem. Soc.* **75**, 1486-87 (1953); *C.A.* **47**:12083 c. **EqG, KiG.**
- Cole, A. R. H. See Jones, R. N.
- Cole, R. H. See Auty, R. P.
- Cole, R. W. Exchange magnetic moments of nuclei with almost closed shells of nucleons. *Phys. Rev.* **89**, 883-84 (1953); *C.A.* **47**:8524 e. **NuMg.**
- Collin, J. See D'Or, L.
- Collinson, E., and Dainton, F. S. γ -ray and x-ray induced polymerization of aqueous solutions and acrylonitrile. *Discussions Faraday Soc.* [12] 212-26 (1952); *C.A.* **47**:9812 i. **InKi.**
- C25** Colowick, S. P. See Rafter, G. W.
- Conn, E. E. See Fisher, H. F.

- Cooke-Yarborough, E. H., and Russell, M. C. B. **A**
C26 simple mass spectrometer for analysis of stable
tracer elements. *J. Sci. Instr.* **30**, 474-80 (1953);
C.A. **48**:7937 h. **AnMs**.
- Cooper, C. D., and Sponer, H. The absorption system of
C27 light and heavy naphthalene vapor at 2900-2500
A. *Phys. Rev.* **87**, 213, 169 (1952); C.A. **48**:8024
i. **SpEl**.
- Cooper, H. G. See Marx, J. W.
- Cooper, H. G., Koehler, J. S., and Marx, J. W. Re-
C28 sistivity changes in copper, silver, and gold
produced by deuteron irradiation near 10° K.
Phys. Rev. **94**, 496 (1954); C.A. **48**:7457 b.
NuIn.
- Corbella, J. M. See Stephenson, C. C.
- Corrsin, L., Fax, B. J., and Lord, R. C. The vibrational
C29 spectra of pyridine and pyridine-d₅. *J. Chem.*
Phys. **21**, 1170-76 (1953); C.A. **47**:9776 g. **SpVi**.
- Corval, M. A method for chemical detection of γ -rays in
C30 the presence of slow neutrons. *Compt. rend.* **235**,
799-801 (1952); C.A. **47**:2054 e. **KiR**.
See Viallard, R.
- Corval, M., and Piolet, C. Preparation and character-
C31 istics of methyl acetate-d₃. *Bull. soc. chim.*
France, 234 (1954); C.A. **48**:8727 f. **Sy**, **ThP**,
EIRf.
- Corval, M., Viallard, R., and Piolet, C. Determination
C32 of deuterium in organic compounds. *Mikrochim.*
Acta, 231-39 (1954); C.A. **48**:6317 c. **AnDn**.
- Coulon, A. See Sprinson, D. B.
- Coursaget, J. See Bugnard, L.
- Courtroy, C. Infrared spectrum under high dispersion of
C33 CD₂Br (2900 to 3400 cm⁻¹). *Ann. soc. sci.*
Bruxelles, Ser. I, **66**, 173-77 (1952); C.A. **47**:9149
e. **SpVi**.
See Charette, J.; Riet, R. van
- Courtroy, C., and Hemptinne, M. de. High-dispersion
C34 infrared spectra of CH₂=CHD. II. *Ann. soc. sci.*
Bruxelles, Ser. I, **67**, 285-95 (1953); C.A.
48:6255 h. **SpVi**.
- Cousin, C. See Prévot-Bernas, A.
- Cox, J. D. Phase relations in the pyridine series. II. The
C35 miscibility of some pyridine homologs with deu-
terium oxide. *J. Chem. Soc.*, 4606-08 (1952);
C.A. **47**:5236 h. **SoH**, **ThP**.
- Cox, J. T. See Williams, Q.
- Craig, D., and Shipman, J. J. Maleic anhydried adducts of
C36 sorbic acid and methyl sorbate. *J. Am. Chem.*
Soc. **74**, 2905-07 (1952); C.A. **48**: 8173 i. **InSt**,
SpVi, **SpEl**, **Sy**.
- Crawford, B. L., Jr., Lancaster, J. E., and Inskip, R. G.
C37 The potential function of ethylene. *J. Chem.*
Phys. **21**, 678-86 (1953); C.A. **47**:6253 i. **SpVi**,
StD.
- Crawford, H. D. Microwave spectroscopy in the X band
C38 region—the X band microwave spectrum of heavy
water. *Univ. Microfilms*, Ann Arbor, Mich., Pub.
No. 6377, 58 pp.; *Dissertation Abstracts* **13**, 1288
(1953); C.A. **48**:4307 i. **SpM**.
- C39** Two new lines in the microwave spectrum of
heavy water. *J. Chem. Phys.* **21**, 2099 (1953);
C.A. **48**:2477 i. **SpM**.
- Crawford, M. F. See Stansbury, E. J.
- Crespin, D. The abundance of deuterium in the solar
C40 atmosphere. *Acad. roy. Belg. classe sci.* **38**, 838-50
(1952); C.A. **47**:5792 g. **AbG**, **SpA**.
- Crist, R. H. See Taylor, A. H.
- Cristol, S. J., and Fix, D. D. Mechanism of elimination
C41 reactions. X. Deuterium exchange in base-promo-
ted dehydrochlorination of β -benzene hexa-
chloride. *J. Am. Chem. Soc.* **75**, 2647-48 (1953);
C.A. **48**:7586 d. **EqL**.
- Cuddeback, R. B., Koeller, R. C., and Drickamer, H. G.
C42 The effect of pressure on diffusion in water and in
sulfate solutions. *J. Chem. Phys.* **21**, 589-97
(1953); C.A. **47**:6223 h. **MeDf**.
- Curran, S. C. Measurement of β -particles of low energy
C43 and allied phenomena. *Physica*, **18**, 1161-70
(1952); C.A. **47**:9164 g. **Ge**, **NuB**.
- Curtis, W., and Ordal, E. J. Hydrogenase and hydrogen
C44 metabolism in micrococcus aerogenes. *J. Bacteriol.*
68, 351-61 (1954); C.A. **48**:13812 f. **EqL**, **KiB**.

D

- Dabbs, J. W. T., and Roberts, L. D. Polarization of
D1 manganese⁵⁵ nuclei: cryogenic aspects. *Phys. Rev.*
95, 970-74 (1954); C.A. **48**:11953 f. **EIP**.
- Dailey, B. P. Hindered rotation and microwave spectro-
D2 scopy. *Ann. N. Y. Acad. Sci.* **55**, 915-27 (1952);
C.A. **47**:2563 g. **SpM**, **St**.
- Dainton, F. S. See Collinson, E.
- Dainton, F. S., Tomlinson, R. H., and Batke, T. L.
D3 Polymerization of styrene and α -methylstyrene
catalyzed by SnCl₄. *Cationic Polymerisation and*
Related Complexes (Proc. Conf. Univ. Coll. North
Staffordshire, Engl.), 80-83 (1952), discussion 84
(Pub. 1953); C.A. 48:5547 e. KiL.
- Daniels, F. See Schmitt, J. A.
- Dardel, G. F. v. Interaction of neutrons with moderating
D4 materials. *Phys. Rev.* **94**, 1272-83 (1954); C.A.
48:9215 c. **NuIn**.
- Darwent, B. deB., and Roberts, R. The reactions of
D5 hydrogen atoms with hydrocarbons. *Discussions*
Faraday Soc. [14], 55-63 (1953); C.A. **48**:9895 d.
KiP.
- Dauben, W. G., and Eastham, J. F. The mechanism of
D6 the reduction of steroidal enol acetates by lithium
aluminum hydride. *J. Am. Chem. Soc.* **75**, 1718-21
(1953); C.A. **48**:4574 h. **InKi**.
- David, E. Separation of isotopes with continuously
D7 working separation tubes. *Z. Physik.*, **134**, 377-96
(1953); C.A. **47**:9805 g. **Se**.
- David, H. G., and Hamann, S. D. Quantum behavior of
D8 compressed gases. *Trans. Faraday Soc.* **49**, 711-16
(1953); C.A. **48**:2468 g. **MeD**, **ThF**.
- Davidson, D. W., Stoicheff, B. P., and Bernstein, H. J.
D9 The infrared and Raman spectra of formaldehyde-
d₁ vapor. *J. Chem. Phys.* **22**, 289-94 (1954);
C.A. **48**:6255 d. **SpVi**.
- Davies, M., and Evans, J. C. The infrared absorptions of
D10 asparagine and glutamine. *J. Chem. Soc.* 480-82
(1953); C.A. **47**:5797 f. **SpVi**.
- Davies, T. H. See Wexler, S.
- Davis, M., and Pollard, E. C. The inactivation of dry
D11 penicillin by deuteron bombardment. *Arch.*
Biochem. Biophys. **37**, 112-16 (1952); C.A. **47**:
779 h. **BiC**.
- Davis, R. T., Jr., and Schiessler, R. W. Optical dispersion
D12 of perdeuteriobenzene and perdeuteriocyclo-
hexane. *J. Am. Chem. Soc.* **75**, 2763-64 (1953);
C.A. **48**:416 a. **EIRf**.
- D13** Vapor pressures of perdeuteriobenzene and of
perdeuteriocyclohexane. *J. Phys. Chem.* **57**,
966-68 (1953); C.A. **48**:3089 a. **ThP**.
- Davis, T. W. See McNesby, J. R.
- Davydova, S. Y. See Konikova, A. S.
- Day, A. R. See Williams, E. D.
- Dayhoff, E. S. See Triebwasser, S.
- Dayhoff, E. S., Triebwasser, S., and Lamb, W. E., Jr.
D14 Fine structure of the hydrogen atom. VI. *Phys.*
Rev. **89**, 106-15 (1953); C.A. **47**:4193 f. **SpA**.
- Dayton, J. C. See Wilmarth, W. K.
- Dean, H. See Mohler, F. L.
- De Azcona, J. M. L. See López de Azcona, J. M.
- De Azevedo, M. D. See Deodata de Azevedo, M.
- De Boer, J. See Boer, J. de
- De Brouckere, L. See Brouckere, L. de
- Decuis, J. C., and Pearson, D. P. The infrared absorption
D15 of crystalline and liquid hydrazine monochloride
and monobromide. *J. Am. Chem. Soc.* **75**, 2436-
39 (1953); C.A. **47**:9776 a. **SpVi**, **SdSp**.
- De Flines, J. See Flines, J. de
- De Hemptinne, M. See Hemptinne, M. de
- De Jager, C. See Jager, C. de
- Delabarre, Y. See Anderson, R. C.
- De Launay, J. See Launay, J. de
- Del Marmol, P. See Marmol, P. del

- De Long, C. W. See Long, C. W. de
 De Mayo, P. See Mayo, P. de
 Den Boer, D. H. W. See Boer, D. H. W. den
 Deodata de Azevedo, M. See Jacobaohn, K. P.
 Deser, S. Electromagnetic effects in two-nucleon systems.
D16 Phys. Rev. **92**, 1542-47 (1953); C.A. **48**:4987 b. **NuQ**.
 Dewey, L. J., Byerrum, R. U., and Ball, C. D. The origin of the methyl group of nicotine through transmethylation. J. Am. Chem. Soc. **76**, 3997-99 (1954); C.A. **48**:12931 h. **InBi**.
 Diamond, A. E. See Pollard, E. C.
 Dibeler, V. H. See Mohler, F. L.
 Dieke, G. H. New results for the spectra of molecular
D18 hydrogen and its isotopes. J. phys. radium **15**, 393-97 (1954); C.A. **48**:11187 f. **SpVi**.
 Dimond, A. E. See Pollard, E.
 Dixon, J. A., and Schiessler, R. W. Viscosities of benzene-
D19 d_6 and cyclohexane- d_{12} . J. Phys. Chem. **58**, 430-32 (1954); C.A. **48**:12511 b. **MeV, SpVi**.
 Djurle, E. See Ingelstam, E.
 Dobriner, K. See Fukushima, D. K.
 Doering, W. v. E., and Aschner, T. C. Mechanism of the
D20 alkoxide-catalyzed carbinol-carbonyl equilibrium. J. Am. Chem. Soc. **75**, 393-97 (1953); C.A. **48**:1950 g. **InKi**.
 Dogramadzi, N. Mass spectrometric determination of
D21 deuterium. Rec. trav. inst. recherches Structure Matiere (Belgrade) **2**, 17-20 (1953); C.A. **47**:12119 c. **AnMs**.
 Dole, M., and Keeling, C. D. Long-range migration of
D22 chemical activity in the solid state. J. Am. Chem. Soc. **75**, 6082-83 (1953); C.A. **48**:3161 e. **KiH**.
 Done, J. See Payne, P. R.
 D'Or, L., and Collin, J. Behavior of aliphatic nitrites
D23 under the action of electronic bombardment. Bull. soc. roy. sci. Liège **22**, 285-304 (1953); C.A. **48**:3796 h. **KiR, Sr**.
 D'Or, L., and Tarte, P. Spectroscopic investigation on nitrous acid, the ultraviolet spectrum and its interpretation. Bull. soc. roy. sci. Liège, **20**, 685-92 (1951); C.A. **47**:401 b. **SpEl, SpVi**.
 Dorfman, L. M. Absorption of tritium β -particles in
D25 hydrogen and other gases. Phys. Rev. **95**, 393-96 (1954); C.A. **48**:11206 h. **NuIn**.
 Dorfman, L. M., and Mattraw, H. C. Exchange reaction
D26 of hydrogen and tritium. J. Phys. Chem. **57**, 723-25 (1953); C.A. **48**:465 b. **EqG, KiG**.
 Dostrovsky, I., Gillis, J., Llewellyn, D. E., and Vromen,
D27 B. H. Separation of isotopes by fractional distillation. II. Determination of parameters from production data. Value of the unit process separation factor for the $H_2^{16}O-H_2^{18}O$ system. J. Chem. Soc. 3517-24 (1952); C.A. **47**:1434 h. **SeDs**.
 Dostrovsky, I., Gillis, J., and Vromen, B. H. Optimal
D28 flow in fractionating columns for isotope separation. Bull. Research Council Israel **2**, No. 1, 68-69 (1952); C.A. **48**:1733 f. **Se**.
 Douglas, A. E., and Sharma, D. Rotation-vibration
D29 spectra of diatomic and simple polyatomic molecules with long absorbing paths. IX. The spectra of the HCN and DCN molecules from 2.5 to 0.5 μ . J. Chem. Phys. **21**, 448-58 (1953); C.A. **47**:5797 i. **SpVi, StD**.
 Douglas, J. E. See Rabinovitch, B. S.
 Douglas, J. E., and Rabinovitch, B. S. Some catalytic
D30 exchange and hydrogenation reactions of acetylene and ethylene. J. Am. Chem. Soc. **74**, 2486-89 (1952); C.A. **47**:6336 a. **EqH, KiH, AnMs, AnSp**.
 Dousmanis, G. C. See Miller, S. L.
 Dousmanis, G. C., Sanders, T. M., Jr., Townes, C. H., and
D31 Zeiger, H. J. Structure of HNCs from microwave spectra. J. Chem. Phys. **21**, 1416-17 (1953); C.A. **47**:10919d. **StD, SpM**.
 Dowling, J. M., and Meister, A. G. Substituted methanes.
D32 XX. Potential constants and calculated thermo-
 dynamic properties for some dibromomethanes. J. Chem. Phys. **22**, 1042-44 (1954); C.A. **48**:12491 f. **ThF, SpVi**.
 Downes, A. M. Isotope effect in the reaction of C^{14} -
D33 formaldehyde with dimedon. Australian J. Sci. Research **5A**, 521-29 (1952); C.A. **47**:1475 d. **Ge, IsKi**.
 Doyle, B. See Setlow, R. B.
 Dreyfus-Alain, B. See Viallard, R.
 Drickamer, H. G. See Caskey, F. E.; Cuddeback, R. B.; Jeffries, Q. R.
 Dubbs, C. A. Determination of deuterium. Anal. Chem. **D34** **25**, 828-29 (1953); C.A. **47**:9062 g. **AnMs**.
 Duchesne, J. Nuclear quadrupole coupling constants and
D35 molecular vibrations. J. Chem. Phys. **20**, 1804-05 (1952); C.A. **47**:9144 d. **NuQ**.
 Dugdale, J. S. See Chester, P. F.
 Duncan, A. B. F. See Nostrand, E. D.
 Dunford, H. B. See Brickman, W. J.
 Dumitz, J. D. See Cardwell, H. M. E.
 Duval, C., and Lecomte, J. Infrared spectrum of H_2O in
D36 solids. II. Experimental determinations. J. chim. phys. **50**, C64-71 (1953); C.A. **48**:4977 i. **SpVi**.
 Du Vigneaud, V. See Vigneaud, V. du
 Dyke, D. C. van See Tobias, C. A.
 Dyken, A. R. van See Wilzbach, K. E.
 Dykhno, N. M., and Shatenshtein, A. I. Mobility of
D37 hydrogen in toluene. Zhur. Fiz. Khim. **28**, 11-13 (1954); C.A. **48**:10413 e. **EqL, KiL**.
D38 Mobility of hydrogen in Tetralin. Zhur. Fiz. Khim. **28**, 14-18 (1954); C.A. **48**:11139 e. **EqL, KiL**.
 Dzhelepov, B. S., and Zyryanova, L. N. Masses of light
D39 nuclei. Uspekhi Fiz. Nauk **48**, 465-530 (1952); C.A. **48**:3159 e. **NuM**.

E

- Eades, R. G. See Andrew, E. R.
 Eastham, J. F. See Dauben, W. G.
 Eaton, R. S. See Giguere, P. A.
 Eckert, B. See Lacassagne, A.
 Edelman, I. S. See Wang, J. H.
 Edelman, I. S., James, A. H., Baden, H., and Moore, F. D.
E1 Electrolyte composition of bone and the penetration of radiosodium and deuterium oxide into dog and human bone. J. Clin. Invest. **33**, 122-31 (1954); C.A. **48**:5986 e. **BiC**.
 Ehrenberg, W., and Jaffke, H. Concentration of heavy
E2 water by means of fractional distillation. Z. angew. Phys. **5**, 375-76 (1953); C.A. **48**:6747 h. **SeDs**.
 Ehrlich, G., and Sutherland, G. B. B. M. Contribution
E3 of side chains to the infrared spectra of proteins: the 6.5μ band. Nature **172**, 671-72 (1953); C.A. **48**:4017 e. **SpVi**.
 Eichhorn, G. Magnetic microwave spectroscopy.
E4 Physik. Bl. **5**, 320-25 (1949); C.A. **47**:6764 e. **Ge, SpM, SpA, NuH**.
 Eidinoff, M. L. Some fractionation effects involving the
E5 isotopes of hydrogen. Trans. N. Y. Acad. Sci. **16**, 76-82 (1953); C.A. **48**:7953 b. **Ge**.
 See Fukushima, D. K.; Perri, G. C.; Rachele, J. R.; Verly, W. G.
 Eidinoff, M. L., Knoll, J. E., Fukushima, D. K., and
E6 Gallagher, T. F. Equilibrium between protium, deuterium, and tritium in the system hydrogen-acetic acid; isotopic fractionation factors in a catalytic hydrogenation. J. Am. Chem. Soc. **74**, 5280-84 (1952); C.A. **47**:1477 f. **EqH, EqG, KiH, IsKi**.
 Eidinoff, M. L., Perri, G. C., Knoll, J. E., Marano, B. J.,
E7 and Arnheim, J. The fractionation of hydrogen isotopes in biological systems. J. Am. Chem. Soc. **75**, 248-49 (1953); C.A. **47**:4939 e. **BiC**.

- Eidinoff, M. L., Reilly, H. C., Knoll, J. E., and Marrian, E. H. Hydrolysis products of nuclei acids labeled with tritium; preparation by biosynthesis. *J. Biol. Chem.* **199**, 511-16 (1952); C.A. **47**:3371 b. **EqL, Sy.**
- Eigen, M. See Meyer, F.
- Elliot, N. See Perlman, M. L.
- Elwyn, D., and Sprinson, D. B. Certain interrelationships in the metabolism of glycine and serine. *J. Biol. Chem.* **207**, 459-65 (1954); C.A. **48**:6540 c. **BiC, InBi.**
- E10 The synthesis of thymine and purines from serine and glycine in the rat. *J. Biol. Chem.* **207**, 467-76 (1954); C.A. **48**:6540 i. **BiC, KiB.**
- Emerson, R. J. See Smith, G. N.
- Enns, T. See Chinard, F. P.; Hilton, M. A.
- Enomoto, S. Kinetic investigations on the ammonia-synthetic iron catalyst. I. Temperature dependence of the rate of exchange reaction between deuterioammonia and hydrogen. *Shokubai* **8**, 47-49 (1952); C.A. **47**:11920 d. **EqH, KiH.**
- Erivanskaya, L. A. See Kagan, M. Y.
- Ertaud, A., and Zaleski, P. Distribution of the thermal neutron density inside a cell of natural uranium and heavy water. *J. phys. radium* **15**, 75A-77A (1954); C.A. **48**:11211 c. **NuIn.**
- Essen, L. The refractive indexes of water vapor, air, oxygen, nitrogen, hydrogen, deuterium, and helium. *Proc. Phys. Soc. (London)* **66B**, 189-93 (1953); C.A. **47**:9083 f. **EIRf.**
- Evans, J. C. See Davies, M.
- Ewald, H. Accuracy of mass-spectrographic isotope-mass measurements. *Natl. Bur. Standards (U. S.) Circ.* **522**, 3746 (1953); C.A. **47**:10367 a. **Sr, NuM.**
- Eyring, H. See Wheeler, A.
- Eyring, H., and Cagle, F. W. Jr. The significance of isotope reactions in rate theory. *J. Phys. Chem.* **56**, 889-92 (1952); C.A. **47**:2021 g. **IsKi, SeCh, Ki.**

F

- Faltings, V. Tritium counter with hydrogen filling and toluene addition. *Naturwissenschaften* **40**, 409 (1953); C.A. **48**:6851 d. **AnC.**
- Farmer, E. C. See Berstein, I. A.
- Farmer, E. C., and Berstein, I. A. Determination of specific activities of tritium-labeled compounds with liquid scintillators. *Science* **117**, 279-80 (1953); C.A. **47**:5805 i. **AnC.**
- Favarger, P., and Gerlach, J. Variations in the manner of absorption of different fats and fat acids in the rat. *Helv. Physiol. et Pharmacol. Acta* **11**, 239-50 (1953); C.A. **48**:1507 b. **InBi.**
- Fax, B. J. See Corrsin, L.
- Feldmann, K. See Goubeau, J.
- Feshbach, H. See Pease, R. L.
- Feyten, J. See Mund, W.
- Fickett, W. Zero-point vibrational contributions to the optical rotatory power of isotopically dissymmetric molecules. *J. Am. Chem. Soc.* **74**, 4204-05 (1952); C.A. **47**:9688 i. **ElRo.**
- Fields, M. See Bernstein, I. A.
- Finke, J. See Selak, P. J.
- Fiquet, F., and Bernas, A. Primary products of the radiolysis of water. *J. chim. phys.* **51**, 47-51 (1954); C.A. **48**:10440 c. **KiR.**
- Fireman, E. L. Measurement of the (n, H^3) cross section in nitrogen and its relationship to tritium production in the atmosphere. *Phys. Rev.* **91**, 922-6 (1953); C.A. **47**:12017 i. **Ab.**
- Fireman, E. L., and Schwarzer, D. Measurement of the tritium concentration in natural waters by a diffusion cloud chamber. *Phys. Rev.* **94**, 385-88 (1954); C.A. **48**:7443 g. **AnC, AbG.**
- Fisher, H. F. See Loewus, F. A.

- Fisher, H. F., Conn, E. E., Vennesland, B., and Westheimer, F. H. The enzymic transfer of hydrogen. I. The reaction catalyzed by alcohol dehydrogenase. *J. Biol. Chem.* **202**, 687-97 (1953); C.A. **47**:8798 d. **InBi, EqL, KiB.**
- Fitzgibbon, G. C. See Zeltmann, A. H.
- Fix, D. D. See Cristol, S. J.
- Fleischmann, R. See Hengevoss, J.
- Flines, J. de. See Kögl, F.
- Florent, R., and Leach, S. The emission spectra of ammonia and heavy ammonia: the transition $^{11}I \rightarrow ^1\Delta$ of the radicals NH and ND. *J. phys. radium* **13**, 377-85 (1952); C.A. **48**:1146 f. **SpEl.**
- Flournoy, J. M. See Wilmarth, W. K.
- Foldy, L. L. A new rigorous lower bound on the range of the triplet neutron-proton interaction. *Phys. Rev.* **89**, 1251-54 (1953); C.A. **47**:8523 d. **NuS, NuQ.**
- Folley, S. J. See Balmain, J. H.
- Fraenkel, G. See Yankwich, P. E.
- Freis, E. D., Higgins, T. F., and Morowitz, H. J. Transcapillary exchange rates of deuterium oxide and thiocyanate in the forearm of man. *J. Appl. Physiol.* **5**, 527-32 (1953); C.A. **47**:7064 g. **BiC.**
- Friedman, A. S., Trzeciak, M., and Johnston, H. L.
- F12 Pressure-volume-temperature relationships of liquid normal deuterium. *J. Am. Chem. Soc.* **76**, 1552-53 (1954); C.A. **48**:6762 i. **ThP, ThF.**
- Friedman, I. Deuterium content of natural waters and other substances. *Geochim. et Cosmochim. Acta* **4**, 89-103 (1953); C.A. **48**:4152 a. **AbG, AnMs.**
- Friis-Hansen, B. Measurement of deuterium oxide at low concentrations by the gradient-tube method. *Scand. J. Clin. and Lab. Invest.* **6**, 65-69 (1954); C.A. **48**:9268 i. **An.**
- Fry, A., and Calvin, M. Isotope effect in the decomposition of oxalic acid. *J. Phys. Chem.* **56**, 897-901 (1952); C.A. **47**:2055 e. **IsKi.**
- F16 The carbon-14 isotope effect in the decarboxylation of 1-naphthyl- and phenylmalonic acids. *J. Phys. Chem.* **56**, 901-05 (1952); C.A. **47**:8653 f. **IsKi.**
- Fukushima, D. K. See Eidinoff, M. L.
- Fukushima, D. K., Bradlow, H. L., Dobriner, K., and Gallagher, T. F. The fate of testosterone infused intravenously in man. *J. Biol. Chem.* **206**, 863-74 (1954); C.A. **48**:5993 i. **InBi.**
- Fukushima, D. K., Dobriner, K., and Gallagher, T. F.
- F18 Studies with testosterone-d in normal men. *J. Biol. Chem.* **206**, 845-61 (1954); C.A. **48**:5993 g. **InBi.**
- Fukushima, D. K., and Gallagher, T. F. The platinum-catalyzed exchange of hydrogen isotopes with steroids. *J. Biol. Chem.* **198**, 871-84 (1952); C.A. **47**:6966 d. **EqH, Sy.**
- F20 Isotopic distribution in cholesterol after platinum-catalyzed hydrogen-deuterium exchange. *J. Biol. Chem.* **198**, 861-69 (1952); C.A. **47**:6965 e. **EqH, Sy.**
- Fukushima, D. K., Kritchevsky, T. H., Eidinoff, M. L., and Gallagher, T. F. Synthesis of radioactive cortisone. *J. Am. Chem. Soc.* **74**, 487-90 (1952); C.A. **47**:9340 b. **Sy.**
- Fukushima, S. See Otozai, K.
- Fuson, N., and Josien, M. L. Structure of the associated OH valence vibration band in light and heavy acetic, trichloroacetic, and trifluoroacetic acids. *J. Opt. Soc. Amer.* **43**, 1102-05 (1953); C.A. **48**:1154 e. **SpVi.**
- Fuson, N., Josien, M. L., Jones, E. A., and Lawson, J. R.
- F23 Infrared and Raman spectroscopy studies of light and heavy trifluoroacetic acids. *J. Chem. Phys.* **20**, 1627-34 (1952); C.A. **47**:7321 f. **SpVi.**

G

- Gailar, N. See Benedict, W. S.
- Galbraith, T. W. See Smith, G. N.

- Gallagher, K. J., Ubbelohde, A. R., and Woodward, I.
G1 The hydrogen bond in crystals. IX. The isotope effect in acetylenedicarboxylic acid dihydrate. *Proc. Roy. Soc. (London)* **A222**, 195-206 (1954); C.A. **48**:6192 i. **IsCr, SdCr**.
- Gallagher, T. F. See Eidinoff, M. L.; Fukushima, D. K.
G2 The fundamental vibrations of ammonia and the ammonium ion. *Compt. rend.* **236**, 911-12 (1953); C.A. **47**:6199 e. **SpVi, StD**.
- G3** The fundamental vibrations in stibine. *Compt. rend.* **238**, 2305-07 (1954); C.A. **48**:11191 f. **SpVi, StD**.
- Garby, L., and Linderholm, H. Permeability of frog skin to heavy water and to ions, with special reference to the effect of some diuretics. *Acta Physiol. Scand.* **28**, 336-46 (1953); C.A. **47**:12442 d. **BiZ**.
- Garland, C. W. See Pimentel, G. C.
G5 Some effects of heavy-particle irradiation of aqueous acetic acid. *Radiation Research* **1**, 97-108 (1954); C.A. **48**:5670 a. **KiR**.
- Gaydon, A. G. See Broidon, H. P.
 Gerlach, J. See Favarger, P.
 Gierer, A., and Wirtz, K. Proton and deuteron transfer over hydrogen bridges as cause of conductivity effects. *J. Phys. Chem.* **56**, 914-17 (1952); C.A. **47**:949 i. **EcC**.
- Giguère, P. A. Properties and structure of hydrogen peroxide and of deuterium peroxide. A review. *Bull. soc. chim. France*, 720-23 (1954); C.A. **48**:11856 a. **Ge, St**.
- Giguère, P. A., and Secco, E. A. Hydrogen peroxide and its analogs. V. Phase equilibria in the system $D_2O-D_2O_2$. *Can. J. Chem.* **32**, 550-56 (1954); C.A. **48**:10416 e. **ThP**.
- Giguère, P. A., Secco, E. A., and Eaton, R. S. Production of hydrogen and deuterium peroxides in the electrodeless discharge. *Discussions Faraday Soc. No.* **14**, 104-11 (1953); C.A. **48**:469 h. **ElGd, Sy**.
- Gilley, L. W. See Lagemann, R. T.
 Gillis, J. See Dostrovsky, I.
 Glascock, R. F. Combustion technique for the assay of **G10** tritium, carbon¹³, and carbon¹⁴ in a single 10-mg. sample of biological material. *Biochem. J.* **52**, 699-704 (1952); C.A. **47**:2243 d. **AnC**.
 See Balmain, J. H.
 Glemser, O. See Hartert, E.
 Glemser, O., and Hartert, E. "Nod" vibrations of the **G11** OH group in the lattice of hydroxides. *Naturwissenschaften* **40**, 552-53 (1953); C.A. **48**:8654 e. **SpVi, SdSp**.
- Gloor, U. See Bernhard, K.
 Gold, V., and Long, F. A. Hydrogen-isotope exchange **G12** between anthracene and sulfuric acid. *J. Am. Chem. Soc.* **75**, 4543-47 (1953); C.A. **48**:3116 h. **EqL, KiL**.
- Goldstein, J. S. See Salpeter, E. E.
 Goodall, C. A., and Alkire, G. J. Mass-spectrometric **G13** determination of deuterium in water. U. S. Atomic Energy Comm., Tech. Inform. Service, **HW-25258**, 3-8 (1952); C.A. **48**:464 a. **AnMs**.
- Gordon, A. S. See McNesby, J. R.
 Goulon, S., and Burton, M. Radiation chemistry of pure **G14** organic compounds: benzene and benzene-d₆. *Discussions Faraday Soc. No.* **12**, 88-89 (1952); C.A. **48**:8079 e. **KiR, Sr**.
- Gordus, A. A. See Bernstein, R. B.
 Gordus, A. A., and Bernstein, R. B. Isotope effect in **G15** continuous ultraviolet absorption spectra: methyl bromide-d₃ and chloroform-d. *J. Chem. Phys.* **22**, 790-95 (1954); C.A. **48**:11923 d. **IsSp, SpEl**.
- Gordy, W. Microwave spectroscopy above 60,000 Me. **G16** *Ann. N. Y. Acad. Sci.* **55**, 774-88 (1952); C.A. **47**:3115 a. **SpM**.
 See Burrus, C. A., Jr.; Williams, Q.
 Gordy, W., and Burrus, C. A. Spectrum of DBr in the one-**G17** millimeter wave region. *Phys. Rev.* **93**, 419-20 (1954); C.A. **48**:6834 f. **SpM**.
- Goubeau, J., Luther, H., Feldmann, K., and Brandes, G.
G18 Molecular vibration spectra of naphthalene and derivatives. I. Production of deuterionaphthalenes; their Raman spectra. *Chem. Ber.* **86**, 214-26 (1953); C.A. **48**:13424 i. **SpVi, Sy**.
- Gould, R. G. See Hayes, F. N.
 Gragerov, I. P., and Brodskii, A. I. Experimental deter-
G19 mination of the height of potential barrier in hydrogen bonds of benzoquinhydrone. *Zhur. Obshchei Khim.* **23**, 1193-99 (1953); C.A. **47**:11826 d. **InKi**.
- Grenon, M. A study of the functioning of counters with
G20 an external cathode. The influence of heat treatment and the role of water bound to the glass. *Compt. rend.* **236**, 1772-73 (1953); C.A. **47**:10991 g. **In**.
 See Viillard, R.
 Grenon, M., and Viillard, R. Determination of tritium.
G21 A review. *J. chim. phys.* **19**, 623-28 (1952); C.A. **47**:3754 e. **Ge, AnC**.
- Grew, K. E., Johnson, F. A., and Neal, W. E. J. The
G22 thermal diffusion factor and temperature. *Proc. Roy. Soc. (London)* **A224**, 513-26 (1954); C.A. **48**:13349 b. **ThD**.
- Grisenthwaite, R. J., and Thompson, H. W. Infrared
G23 spectrum of deuteriomethylacetylene $CH_3C:CD$. *Trans. Faraday Soc.* **50**, 212-17 (1954); C.A. **48**:13431 a. **SpVi**.
- Grisolia, S., and Burris, R. H. Preparation of glutamate
G24 and carbamyl glutamate selectively labeled with deuterium. *J. Biol. Chem.* **210**, 109-17 (1954); C.A. **48**:12844 hi. **Sy, EqL**.
- Gross, E. F., and Val'kov, V. I. Vibrations of the hydro-
G25 gen bond and its raman spectrum. *Doklady Akad. Nauk S. S. S. R.* **67**, 619-22 (1949); C.A. **48**:6829 h. **SpVi**.
- Gross, L. See Hamilton, R. R.
 Grosse, A. V., Kirshenbaum, A. D., Kulp, J. L., and
G26 Broecker, W. S. The natural tritium content of atmospheric hydrogen. *Phys. Rev.* **93**, 250-51 (1954); C.A. **48**:6174 g. **AbG**.
- Groth, W., and Hardeck, P. Separation of isotopes by
G27 gaseous diffusion. *Z. physik. Chem.* **199**, 114-24 (1952); C.A. **47**:3108 a. **ThD, SeDf, AnMs**.
- Groth, W. E. See Beckey, H. D.
 Gurnee, E. F. See Magee, J. L.
 Gustin, S. T. See Marx, W.
 Gutmann, J. R. The exchange reaction between deuterium
G28 and ammonia on the surface of metal powders. *J. Phys. Chem.* **57**, 309-12 (1953); C.A. **47**:6233 h. **EqH, KiH**.
- Gwinn, W. D. See Myers, R. J.

H

- Hadži, D. Electronic influences on the quasi-character-
H1 istic infrared absorption of the carbonyl groups in the quinone series. *Archiv Kem.* **25**, 33-38 (1953); C.A. **48**:10436 b. **SpVi**.
- Hadži, D., and Sheppard, N. The infrared absorption
H2 bands associated with the COOH and COOD groups in dimeric carboxylic acids. I. The region from 1500 to 500 cm^{-1} . *Proc. Roy. Soc. (London)* **A216**, 247-66 (1953); C.A. **47**:5797 i. **SpVi, SdSp**.
- Haigh, C. P. Analysis for deuterium based on the photo-
H3 neutron effect. *Nature* **172**, 359 (1953); C.A. **48**:2491 e. **AnC**.
- Haissinsky, M., and Puchault, J. Radiochemical reac-
H4 tions in nuclear reactors. II. The influence of electrolytes on the formation and decomposition of hydrogen peroxide. *J. chim. phys.* **49**, 294-301 (1952); C.A. **47**:976 g. **KiR**.
- Haldeman, R. G. Determination of heavy water with
H5 the hot-wire gage. *Anal. Chem.* **25**, 787-90 (1953); C.A. **37**:8419 e. **AnTh**.
- Hall, L. G. See Washburn, H. W.
 Hamann, S. D. See David, H. G.
 Hamill, W. H. See Schwarz, H. A.

- Hamilton, D. R., Alford, W. P., and Gross, L. Limit on **H6** neutrino mass from tritium β -spectrum. *Phys. Rev.* **83**, 215 (1951); C.A. 47:6773 i. **NuB**.
- H7** Upper limits on the neutrino mass from the tritium β -spectrum. *Phys. Rev.* **92**, 1521-25 (1953); C.A. 48:4986 b. **NuB**.
- Hanke, M. E. See Mandeles, S.
- Hansler, R. L., and Oetjen, R. A. The infrared spectra of **H8** HCl, DCl, HBr, and NH₃ in the region from 40 to 140 microns. *J. Chem. Phys.* **21**, 1340-43 (1953); C.A. 47:11983 c. **SpVi**.
- Harrick, N. J. See Bray, P. J.
- Harrick, N. J., Barnes, R. G., Bray, P. J., and Ramsey, **H9** N. F. Nuclear radiofrequency spectra of D₂ and H₂ in intermediate and strong magnetic fields. *Phys. Rev.* **90**, 260-66 (1953); C.A. 47:9143 g. **NuRe**.
- Harris, F. E., and Alder, B. J. Dielectric polarization in **H10** polar substances. *J. Chem. Phys.* **21**, 1031-38 (1953); C.A. 47:9081 h. **EID**.
- Harris, G. M. See Collett, D. W.; Stranks, D. R.
- Harris, P. M. See Stambaugh, C. K.
- Harteck, P. See Groth, W.
- Hartert, E. See Glemser, O.
- Hartert, E., and Glemser, O. Position of hydrogen in **H11** lattice of crystalline hydroxides. *Naturwissenschaften* **40**, 199-200 (1953); C.A. 47:11971 a-**SpVi**, **SdSp**.
- Hauser, C. R. See Hill, D. G.
- Hayes, F. N., and Gould, R. G. Liquid scintillation counting of tritium-labeled water and organic compounds. *Science* **117**, 480-82 (1953); C.A. 47:11014 d. **AnC**.
- Hayes, F. N., Hiebert, R. D., and Schuch, R. L. Low-energy counting with a new liquid scintillation solute. *Science* **116**, 140 (1952); C.A. 47:3132 g. **AnC**, **NuB**.
- Haymond, H. R. See Garrison, W. M.
- Haynie, W. H., and Nielsen, H. H. The fundamental absorption bands in the infrared spectrum of stibine and deuteriated stibine. *J. Chem. Phys.* **21**, 1839-42 (1953); C.A. 48:1151 c. **SpVi**, **StD**.
- Heiks, J. R., Barnett, M. K., Jones, L. V., and Orban, E. **H15** The density, surface tension, and viscosity of deuterium oxide at elevated temperatures. *J. Phys. Chem.* **58**, 488-91 (1954); C.A. 48:11862 e. **MeD**, **MeSt**, **MeV**.
- Helvey, T. C. The natural concentration of deuterium in **H16** honey. *Science* **117**, 276-77 (1953); C.A. 47:6065 f. **AbO**, **AnMs**.
- Hemptinne, M. de. See Charette, J.; Courtoy, C.; Riet, R. van.
- Hemptinne, M. de, and Charette, J. Raman spectrum of **H17** gaseous ethylene. *Bull. classe sci., Acad. roy. Belg.* **39**, 622-28 (1953); C.A. 48:2477 f. **SpVi**.
- Henderson, I. H. S. See Lossing, F. P.
- Henderson, J. W. See Marx, J. W.
- Henderson, L. M. See Schayer, R. W.
- Henry, S. S. See Hilton, M. A.
- Hengevoss, J., and Fleischmann, R. Pilot-experiment of **H18** a continuous isotope separation. *Z. Physik* **137**, 265-73 (1954); C.A. 48:8670 ab. **Se**.
- Herber, R. H. Isotopic exchange. *J. Chem. Educ.* **31**, **H19** 359-61 (1954); C.A. 48:11886 b. **Ge**, **Eq**.
- Herrmann, J. See Viillard, R.
- Herr, W. Activation analysis. *Angew. Chem.* **64**, 679-85 **H20** (1952); C.A. 47:2624 g. **Ge**, **AnC**.
- Herzberg, G. See Bardwell, J.
- Herzberg, G., and Ramsay, D. A. The absorption spectrum of free NH₂ radicals. *Discussions Faraday Soc.* No. 14, 11-16 (1953); C.A. 48:444 e. **EIGd**, **SpEl**.
- Hevesy, G. Use of radioactive indicators. A lecture. **H22** *Chimia (Switz.)* **6**, 201-11 (1952); C.A. 47:2045 i. **Ge**, **In**.
- Hicks, H. G. See Stevenson, P. C.
- Hiebert, R. D. See Hayes, F. N.
- Hiebert, R. D., and Watts, R. J. Fast-coincidence circuit **H23** for hydrogen³ and carbon¹⁴ measurements. *Nucleonics* **11**, No. 12, 38-41 (1953); C.A. 48:3800 g. **AnC**.
- Higatsberger, M. J. See Wagner, G.
- Higgins, T. F. See Freis, E. D.
- Higuchi, I., and Kawana, Y. Calcium-hydrogen system. **H24** III. A theoretical consideration of the mechanism for the growth of a nucleus consisting of calcium hydride or deuteride. *Science Repts. Tohoku Univ. Ser. I*, **37**, 179-84 (1953); C.A. 48:5619 g. **KiS**.
- Hill, D. G., Judge, W. A., Skell, P. S., Kantor, S. W., and **H25** Hauser, C. R. Evidence for α - and β -elimination from deuterioalkyl halides with potassium amide. *J. Am. Chem. Soc.* **74**, 5599-602 (1952); C.A. 48:1940 g. **KiL**.
- Hilton, M. A., Barnes, F. W., Jr., Henry, S. S., and Enns, **H26** T. Mechanisms in enzymic transamination. Rate of exchange of the hydrogen of aspartate. *J. Biol. Chem.* **209**, 743-54 (1954); C.A. 48:12204 i. **EqL**, **KiB**.
- Hine, J., Peck, R. C., Jr., and Oakes, B. D. The kinetics of the base-catalyzed deuterium exchange of chloroform in aqueous solution. *J. Am. Chem. Soc.* **76**, 827-29 (1954); C.A. 48:11893 i. **EqL**, **KiL**.
- Hine, J., and Thomas, C. H. The rate of deuterium exchange between ethanol and water. A reinvestigation. *J. Am. Chem. Soc.* **75**, 739-40 (1953); C.A. 47:11912 e. **EqL**, **KiL**.
- H29** Rate of deuterium exchange of certain amines and alcohols. *J. Am. Chem. Soc.* **76**, 612 (1954); C.A. 48:4943 i. **EqL**, **KiL**.
- Hiskey, C. F. Recovery of heavy water from solutions with neutron reactors. U. S. Patent 2,681,883, June 22, 1954; C.A. 48:9834 g. **Se**.
- Hitchcock, W. J. See Billings, B. H.
- Hojman, J. M. The effect of nitrate ion and ammonia on **H31** the electrolytic separation of hydrogen-deuterium isotopes. *Bull. Inst. Nuclear Sci. "Boris Kidrich"* **3**, No. 46, 121-26 (1953); C.A. 48:1852 a. **SeEl**.
- Holland, D. A. See Brown, F.
- Hollander, J. M., Perlman, I., and Seaborg, G. T. Table **H32** of isotopes. *Rev. Mod. Phys.* **25**, 469-651 (1953); C.A. 47:8429 b. **Ge**.
- Hoomissen, J. E. van See Noyes, J. C.
- Horwitz, L. Some effects of deuterium oxide on the **H33** quinone Hill reaction of *Chlorella pyrenoidosa*. *Plant Physiol.* **29**, 215-19 (1954); C.A. 48:11576 bc. **BiB**, **BiC**.
- Hrostowski, H. J., and Myers, R. J. The microwave. **H34** spectra, structure, and dipole moment of stable pentaborane. *J. Chem. Phys.* **22**, 262-65 (1954); C.A. 48:5653 g. **SpM**, **EID**, **StD**.
- Hrostowski, H. J., and Pimentel, G. C. The infrared **H35** spectra of stable pentaborane and deuteriated pentaborane. *J. Am. Chem. Soc.* **76**, 998-1003 (1954); C.A. 48:7434 i. **SpVi**, **Sy**, **ThF**.
- Hubbard, D., and Cleek, G. W. Deuterium- and hydrogen-electrode characteristics of lithium-silica **H36** glasses. *J. Research Natl. Bur. Standards* **49**, 267-71 (1952) RP2363; C.A. 47:5222 g. **EcP**, **ThSo**.
- Hubbard, J. C. See Sette, D.
- Hutchinson, D. L. See Plentl, A. A.
- Hutchinson, D. L., Plentl, A. A., and Taylor, H. C., Jr. **H37** The total body water and the water turnover in pregnancy studied with deuterium oxide as isotopic tracer. *J. Clin. Invest.* **33**, 235-41 (1954); C.A. 48:5987d. **BiZ**.
- Hutchinson, F. Inactivation of unimolecular layers of **H38** bovine serum albumin with deuterons. *Arch. Biochem. Biophys.* **41**, 317-24 (1952); C.A. 47:2869 a. **InBi**, **BiC**. See McNulty, W. P., Jr.
- Hutchinson, F., and Mosbury, E. R., Jr. Deuteron **H39** inactivation of adsorbed unilayers of bovine serum albumin. *Arch. Biochem. and Biophys.* **51**, 436-43 (1954); C.A. 48:12840 c. **BiC**.

Hutchison, D. A. See Smaller, B.
Hvidt, A., and Linderström-Lang, K. Exchange of
H40 hydrogen atoms in insulin with deuterium atoms
in aqueous solutions. *Biochim. et Biophys. Acta*
14, 574-75 (1954); C.A. **48**:12828 g. **EqL**.

I

Ingelstam, E., Djurle, E., and Johansson, L. Precision
I1 concentration analysis of D₂O/H₂O by means of
phase-contrast refractometry. *J. Opt. Soc. Amer.*
44, 472-77 (1954); C.A. **48**:11236d. **AnRf**.
Ingersoll, L. R., and Liebenberg, D. H. The Faraday
I2 effect in gases and vapors. I. *J. Opt. Soc. Amer.*
44, 566-71 (1954); C.A. **48**:10423f. **ElMr**.
Ingerson, E. Nonradiogenic isotopes in geology: a review.
I3 *Bull. Geol. Soc. Am.* **64**, 301-74 (1953); C.A.
47:4810 a. **Ge, Se**.
Ingold, C. K. See Cohn, H.
Ingold, C. K., and King, G. W. Excited states of C₂H₂.
I4 II. Experimental methods of recording the near-
ultraviolet absorption spectra of acetylenes. *J.*
Chem. Soc., 2704-07 (1953); C.A. **48**:1806 h.
SpEl.
I5 Excited states of C₂H₂. IV. Description and analy-
sis of the near-ultraviolet absorption spectra of
C₂H₂ and C₂D₂: nature of the excited state. *J.*
Chem. Soc., 2725-44 (1953); C.A. **48**:1806 i.
SpEl, StD.
I6 Excited states of C₂H₂. V. Measured and calculat-
ed frequencies in the near-ultraviolet absorp-
tion spectra of C₂H₂ and C₂D₂: vibrational and
rotational energies of the trans bent excited state.
J. Chem. Soc., 2745-55 (1953); C.A. **48**:1806 i.
SpEl, SpVi.
Ingold, K. U. See Lossing, F. P.
Inskeep, R. G. See Crawford, B. L., Jr.
Isberg, P., and Lundberg, L. Density of heavy water and
I7 electrolytic enrichment of the oxygen isotopes. *Z.*
Naturforsch. **9a**, 472-73 (1954); C.A. **48**:12575 i.
MeD.
Ivanenko, D. D., and Kolesnikov, N. Theory of isotopic
I8 displacement. *Doklady Akad. Nauk S. S. S. R.* **89**,
253-56 (1953); C.A. **48**:5639f. **IsSp**.
Izrailevich, E. A. See Shatenshtein, A. I.

J

Jache, A. See Burrus, C. A., Jr.
Jacobsohn, B. A. β -Decay of para- and ortho-tritium.
J1 An abstract. *Phys. Rev.* **83**, 885-86 (1951); C.A.
47:6775 g. **NuB**.
Jacobsohn, K. P., and Azevedo, M. D. de. Action of
J2 deuterium oxide on urease. *Rev. fac. ciênc., Univ.*
Lisboa 2a Sér., B, **2**, 65-71 (1952-53); C.A. **48**:
9418 d. **KiB**.
Jaffke, H. See Ehrenberg, W.
Jager, C. de. A new determination of the abundance of
J3 deuterium in the solar atmosphere. *Processus*
nucléaires dans astres. 5^e Colloque. intern. astro-
phys., Liège, Sept. 1953 in Mém. soc. roy. sci.
Liège **14**, 460-67 (1954); C.A. **48**:11917 f. **AbG**.
James, A. H. See Edelman, I. S.
James, H. M. See Kreiger, T. J.
Jean, M. Method of second quantization and configura-
J4 tion space in relativistic theory of a system of par-
ticles. Application to the derivation of relativistic
equations for the deuteron. *Ann. phys.* **8**, 338-91
(1953); C.A. **47**:11943f. **NuS**.
Jeffries, T. O., and Owen, M. E. A tritium monitor. *J.*
J5 *Sci. Instr.* **30**, 387-88 (1953); C.A. **48**:9216 b. **AnC**.
Jeffries, Q. R., and Drickamer, H. G. Diffusion in the
J6 system CH₄-TCH₃ to 300 atmospheres pressure.
J. Chem. Phys. **21**, 1358 (1953); C.A. **47**:11854 h.
MeDf, ThD.
Jen, C. K. Molecular and nuclear magnetic moments.
J7 *Ann. N. Y. Acad. Sci.* **55**, 822-30 (1952); C.A.
47:3056 g. **Ge, ElMm, NuMg, SpM**.

Jen, C. K., Bianco, D. R., and Massey, J. T. Some heavy-
J8 water rotation absorption lines. *J. Chem. Phys.*
21, 520-25 (1953); C.A. **47**:5793 e. **SpM**.
Jenkins, W. A. Estimating the tritium content of tritiated
J9 water. *Anal. Chem.* **25**, 1477-80 (1953); C.A.
48:914 d. **AbG, AnC**.
Jensen, M. L. Geologic importance of variations in stable
J10 isotopic abundances. A review. *Econ. Geol.* **48**,
161-76 (1953); C.A. **48**:11997 g. **AbG**.
Jernakoff, G. See Morgan, W. A.
Johansson, L. See Ingelstam, E.
Johnson, F. A. See Grew, K. E.
Johnson, J. A., Cavert, H. M., and Lifson, N. Kinetics
J11 concerned with distribution of isotopic water in
isolated perfused dog heart and skeletal muscle.
Am. J. Physiol. **171**, 687-93 (1952); C.A. **47**:4462 e.
KiB.
Johnson, J. R., and Shiner, V. J., Jr. The structure of
J12 ketene dimer. *J. Am. Chem. Soc.* **75**, 1350-55
(1953); C.A. **48**:3253 e. **EqL, InSt**.
Johnston, H. See Friedman, A. S.
Johnston, W. H., and Arnold, J. R. The existence of
J13 multiply charged molecular ions of hydrobromic
and hydrochloric acids and deuterium chloride.
J. Chem. Phys. **21**, 1499-1502 (1953); C.A.
47:11961 h. **Sr**.
Joiner, E. E. See Prentice, T. C.
Jones, E. A. See Fuson, N.
Jones, L. H. The infrared spectra and structure of LiOH,
J14 LiOH.H₂O, and the deuterium species. Funda-
mental frequency of OH⁻. *J. Chem. Phys.* **22**,
217-19 (1954); C.A. **48**:4975 g. **SpVi**.
Jones, L. H., and Penneman, R. A. Infrared absorption
J15 spectra of aqueous HF₂⁻, DF₂⁻, and HF. *J.*
Chem. Phys. **22**, 781-82 (1954); C.A. **48**:11925 i.
SpVi.
Jones, L. V. See Heiks, J. R.
Jones, R. N. See Nolin, B.
Jones, R. N., Cole, A. R. H., and Nolin, B. Characteriza-
J16 tion of methyl and methylene groups in steroids by
infrared spectrometry. II. Methyl and methylene
bending frequencies in steroids labeled with
deuterium. *J. Am. Chem. Soc.* **74**, 5662-68
(1952); C.A. **47**:5800 d. **IsSp, SpVi, St**.
Jones, R. N., and Lauzon, R. The selection of solvents for
J17 infrared spectrometry. *Natl. Research Council,*
Can. N. R. C. Bull. No. 3, 3-9 (1953); C.A.
47:9154 i. **SpVi**.
Jones, W. M. The triple-point temperature of tritium
J18 oxide. *J. Am. Chem. Soc.* **74**, 6065-66 (1952);
C.A. **47**:3675 e. **ThP**.
J19 Luminescence behavior in tritium oxide. *J. Chem.*
Phys. **20**, 1974 (1952); C.A. **47**:4747 f. **SdSp,**
SpFl, NuB.
Josien, M. L. See Fuson, N.
Judge, W. A. See Hill, D. G.
Jura, G. See Pimentel, G. C.

K

Kagan, M. Y., Erivanskaya, L. A., and Trofimova, I. V.
K1 Mechanism of the catalytic dehydrocyclization of
paraffin hydrocarbons. *Doklady Akad. Nauk*
S.S.S.R. **82**, 913-16 (1952); C.A. **47**:6333 h. **Eq**.
Kahn, B. See Lacy, W. J.
Kakiuchi, Y. Normal vibrations of benzene. *J. Chem.*
K2 *Soc. Japan, Pure Chem. Sect.* **75**, 143-46 (1954);
C.A. **48**:4976 h. **SpVi**.
Kanda, E. See Sugawara, T.
Kanda, T. See Sugawara, T.
Kane, W. R. See Broida, H. P.
Kaneko, Y. The platinum catalyst. III. The mechanism
K3 of the exchange reaction of hydrogen between
heavy ammonia and hydrogen in the presence of
platinum. *Shokubai, No. 7*, 98-105 (1951);
C.A. **47**:11924 i. **EqH, KiH**.
Kantor, S. W. See Hill, D. G.

- Kaplan, L. See Brown, W. G.; Wilzbach, K. E.
- Kaplan, L., and Wilzbach, K. E. Hydrogen-isotope effect
K4 in the hydrolysis of triphenylsilane. *J. Am. Chem. Soc.* **74**, 6152-53 (1952); C.A. **47**:11917 f. **IsKi, KiL**.
- Kast, W. The particle properties of the crystalline regions
K5 of the cellulose fiber. *Z. Elektrochem.* **57**, 525-30 (1953); C.A. **48**:4217 h. **Eq**.
- Kauder, L. Mass spectrometric examination of the catalytic isomerization of n-propylchloride and the catalytic exchange of acetone and propane with deuterium. Univ. Microfilms (Ann Arbor, Mich.), Pub. No. 4262, 61pp. Dissertation Abstracts, **12**, 651-52 (1952); C.A. **47**:1592 h. **EqH, KiH**.
- Kaufman, S., and Libby, W. F. Natural distribution of
K7 tritium. *Phys. Rev.* **93**, 1337-44 (1954); C.A. **48**:7443 d. **AbG**.
- Kaveeschwar, S. N. See Ramaiah, N. A.
- Kawana, Y. Calcium-hydrogen system. II. The rate of
K8 absorption of heavy hydrogen by metallic calcium. Science Repts. Tohoku Univ. Ser. I, **37**, 175-78 (1953); C.A. **48**:5619 gh. **KiS**.
 See Higuchi, I.
- Keeling, C. D. See Dole, M.
- Keii, T. Rates of formation of different deuterioethylenes and deuterioethanes during contact deuteration of light ethylene in the presence of nickel catalyst. *J. Research Inst. Catalysis, Hokkaido Univ.*, **3**, 36-44 (1953); C.A. **48**:8006 ed. **KiH**.
- K10** Rates of formation of different deuterioethylenes and deuterioethanes in the reaction of ethylene and deuterium. *J. Chem. Phys.* **22**, 144-45 (1954); C.A. **48**:4847 hg. **KiH**.
- Keim, C. P. Electromagnetic separation of stable isotopes. *Ann. Rev. of Nuclear Sci.* **1**, 263-92 (1952); C.A. **47**:409 g. **Se**.
- K12** Enriching stable isotopes electromagnetically. A review. *J. Appl. Phys.* **24**, 1255-61 (1953); C.A. **48**:464 c. **SeMs**.
- Keller, F. L., and Nielsen, A. H. The infrared spectrum and molecular constants of DBr. *J. Chem. Phys.* **22**, 294-99 (1954); C.A. **48**:6248 i. **SpVi, StD**.
- Kelly, P., Tigg, T., and Weiss, J. Chemical primary processes in the action of ionizing radiations on water. Evidence from experiments with heavy water. *Nature*, **173**, 1130-32 (1954); C.A. **48**:11214 hg. **KiR**.
- Kemball, C. Catalysis on evaporated metal films. II. The efficiency of different metals for the reaction between methane and deuterium. *Proc. Roy. Soc. (London)* **A217**, 376-89 (1953); C.A. **47**:8490 f. **EqH, KiH**.
- K16** Catalysis on evaporated metal films. IV. Exchange between propane and deuterium and isobutane and deuterium. *Proc. Roy. Soc. (London)* **A233**, 377-92 (1954); C.A. **48**:7965 i. **EqH, KiH**.
 See Anderson, J. R.
- Kessler, H. K., and Sutherland, G. B. B. M. Out-of-plane deformation frequency of the NH group in the peptide link. *J. Chem. Phys.* **21**, 570-71 (1953); C.A. **47**:7893 b. **IsSp**.
- Ketelaar, J. A. A., Bier, A., and Vlaar, H. T. Behavior of
K18 1,3,5-trinitrobenzene in alkaline solution. *Rec. trav. chim.* **73**, 37-38 (1954); C.A. **48**:6793 d. **EqL**.
- Khan, N. A. Selective reduction of the unsaturated compounds: deuteration of the Raney nickel catalysts. *Science* **117**, 130 (1953); C.A. **47**:12208 f. **KiH**.
- K20** Preparation of deuterized Raney Nickel and selective deuteration of the triple bond. *J. Am. Chem. Soc.* **74**, 3018-22 (1952); C.A. **48**:9896 g. **Sy, KiH**.
- K21** Application of tracer techniques to studies on autoxidation reactions. *J. Am. Oil Chemists' Soc.* **30**, 273-78 (1953); C.A. **48**:4437 a. **InKi**.
- Khaskin, I. G. Mobility of hydrogen in some organosilicon compds. *J. Gen. Chem. U. S. S. R.* **23**, 29-32 (1953); C.A. **48**:12671 f. **KiH**.
- K23** Mobility of hydrogen in some organosilicon compounds. *Zhur. Obshchei Khim.* **23**, 32-37 (1953); C.A. **48**:563 h. **EqL**.
- Kierstead, H. A. See McDonnell, W. R.
- Kigoshi, K. The separation of isotopes by a counter-current gaseous exchange column. I. Transport equation and over-all separation. *Bull. Chem. Soc. Japan* **26**, 288-93 (1953); C.A. **48**:4943d f. **SeCh**.
- King, G. W. See Ingold, C. K.
- Kirshenbaum, A. D. See Grosse, A. V.
- Kistiakowsky, G. B., and Roberts, E. K. Rate of association of methyl radicals. *J. Chem. Phys.* **21**, 1637-43 (1953); C.A. **48**:1123 f. **KiP**.
- Klein, J. A., and Nethercot, A. H., Jr. Microwave spectrum of DI at 1.5 mm. wave length. *Phys. Rev.* **91**, 1018 (1953); C.A. **47**:11990 i. **SpM**.
- Kleman, B. The 2700-A system in the band spectra of
K27 AlH and AlD. *Arkiv Fysik* **6**, 407-14 (1953); C.A. **47**:8512 b. **SpEl**.
- Klemm, A. Separation of isotopes by electrolytic migration of ions. *Colloques intern. centr. natl. recherche sci.* **39**, Electrolyse C18-C24 (1952); C.A. **48**:13464 a. **SeEl**.
- Klemperer, W., and Pimentel, G. C. Hydrogen bonding in sodium trifluoroacetate-trifluoroacetic acid compounds. *J. Chem. Phys.* **22**, 1399-1402 (1954); C.A. **48**:13362 c. **SpVi, ThF**.
- Kluyver, J. C., and Milatz, J. M. W. An infrared isotope analyzer. *Physica* **19**, 401-11 (1953); C.A. **48**:450 i. **AnSp**.
- Knauss, H. J. See Porter, J. W.
- Knioll, J. See Perri, G. C.
- Knoll, J. E. See Eidinoff, M. L.; Rachele, J. R.; Verly, W. G.
- Koefoed-Johnsen, V., and Ussing, H. H. Contributions of diffusion and flow to the passage of D₂O through living membranes. Effect of neurohypophysial hormone on isolated anuran skin. *Acta Physiol. Scand.* **28**, 60-76 (1953); C.A. **47**:5972 h. **BiZ**.
- Kögl, F., Flines, J. de, and Boer, D. H. W. den. Protein problems. I. Synthesis of some optically active α , α' -imino acids. *Rec. trav. chim.* **72**, 1009-19 (1953); C.A. **48**:5096 h. **Sy**.
- Kögl, F., Flines, J. de, Boer, D. H. W. den, and Salemink, C. A. Protein problems. II. The behavior of some α , α' -imino acids as substrates in biological systems. *Rec. trav. chim.* **72**, 1020-26 (1953); C.A. **48**:5233 h. **BiC**.
- Kochler, J. S. See Cooper, H. G.; Marx, J. W.
- Kochler, J. S., and Seitz, F. Radiation dislocations of
K34 crystals. A review. *Z. Physik* **138**, 238-45 (1954); C.A. **48**:13410 e. **NuIn, Sd**.
- Kocler, R. C. See Cuddeback, R. B.
- Kolesnikov, N. See Ivanenko, D. D.
- Konikova, A. S., and Davydova, S. Y. Investigation of protein metabolism in normal and in tumorous rats by means of deuterium. *Ukrain. Biokhim. Zhur.* **22**, 420-24 (1950); C.A. **48**:3494 c. **InBi**.
- Koppelman, R. See Mandeles, S.
- Kornberg, H. A. See Long, C. W. de
- Korshak, V. V. See Kursanov, D. N.
- Koski, W. S. See Maybury, P. C.
- Kovner, M. A. Vibrational spectra of benzene and
K36 deuterobenzenes: computation and interpretation. *Doklady Akad. Nauk S. S. R.* **91**, 499-502 (1953); C.A. **48**:6833 c. **SpVi, StD**.
- Kraitchman, J. See Miller, S. L.
- Krantz, J. C., Jr. See Carr, C. J.
- Krasna, A. I., and Rittenberg, D. The mechanism of the enzyme hydrogenase. *J. Am. Chem. Soc.* **76**, 3015-20 (1954); C.A. **48**:9430 h. **EqH, KiB**.
- K38** Inhibition of hydrogenase by nitric oxide. *Proc. Natl. Acad. Sci. U. S.* **40**, 225-27 (1954); C.A. **48**:9430 i. **EqH, KiB**.

Krasnovskii, A. A., and Brin, G. P. Action of heavy water on the reaction of photoreduction of chlorophyll and the photochemical activity of the substance of green leaves. Doklady Akad. Nauk S. S. R. **96**, 1025-28 (1954); C.A. **48**:13831 c. **BiB**, **BiC**, **KiB**.

Krieger, K. A. See Williams, E. D.

Krieger, T. J., and James, H. M. Successive orientational transitions in crystals. J. Chem. Phys. **22**, 796-814 (1954); C.A. **48**:11867 h. **SdTr**.

Kritchevsky, D. See Biggs, M. W., Fukushima, D. K.

Kruse, U. E. See Ramsey, N. F.

Kuchinskas, E. J. See Rachele, J. R.

Kukhtenko, I. I. Mechanism of the reaction of bromination of toluene in the methyl group. Doklady Akad. Nauk S. S. R. **92**, 77-78 (1953); C.A. **48**:10628 d. **InKi**, **KiP**. See Brodskii, A. I.

Kukhtenko, I. I., and Brodskii, A. I. Mobility of hydrogen atoms in the toluene nucleus. Doklady Akad. Nauk S. S. R. **86**, 733-35 (1952); C.A. **47**:4852 e. **EqG**, **EqL**, **Sy**.

Kulp, J. L. See Grosse, A. V.

Kume, S. See Ootzai, K.

Kuratani, K. Infrared spectra of deuterium compounds. I. Trichloroacetic acid. J. Chem. Soc. Japan. Pure Chem. Sect. **73**, 758-61 (1952); C.A. **47**:3696 h. **SpVi**.

K44 Infrared spectra of deuterium compounds. II. Pentachlorophenol. J. Chem. Soc. Japan. Pure Chem. Sect. **73**, 928-31 (1952); C.A. **47**:3696 i. **SpVi**.

K45 The infrared dichroism of amides. Bull. Chem. Soc. Japan. **25**, 258-61 (1952); C.A. **47**:8515 i. **SpVi**, **St**.

Kursanov, D. N. See Nazarov, I. N.; Setkina, V. N.

Kursanov, D. N., Korshak, V. V., and Vinogradova, S. V. High-molecular weight compounds. II. Exchange reaction in polyesters studied by means of the heavy isotope of hydrogen. Bull. Acad. Sci. U. S. S. R., Div. Chem. Sci., 125-28 (1953); C.A. **48**:12674 f. **EqH**, **InKi**.

K47 High-molecular compounds. II. Exchange reaction in polyesters studied by means of the heavy isotope of hydrogen. Invest. Akad. Nauk S. S. R., Otdel. Khim. Nauk, 140-44 (1953); C.A. **48**:3912 g. **EqH**, **InKi**.

Kursanov, D. N., and Parnes, Z. N. Reactions of hydrogen exchange of α , β -unsaturated ketones. Doklady Akad. Nauk S. S. R. **91**, 1125-28 (1953); C.A. **48**:10549 d. **EqL**.

Kursanov, D. N., Parnes, Z. N., Zaretskaya, I. I., and Nazarov, I. N. Reaction mechanism of the cyclization by means of deuterium. I. Cyclization of isopropenyl allyl ketone. Bull. Acad. Sci. U. S. S. R., Div. Chem. Sci. 103-07 (1953); C.A. **48**:12786 i. **InKi**.

K50 Reaction mechanism of the cyclization by means of deuterium. I. Cyclization of isopropenyl allyl ketone. Izvest. Akad. Nauk S. S. R., Otdel. Khim. Nauk, 114-20 (1953); C.A. **48**:3271 b. **InKi**.

Kusch, P. See Prodell, A. G.

L

Lacassagne, A., Buu-Hoi, Ng. Ph., Xuong, Ng. D., Zajdela, F., and Eckert, B. Synthesis of hexestrol labeled with two atoms of tritium. Compt. rend. **235**, 589-90 (1952); C.A. **47**:8050 f. **Sy**.

Lacy, W. J., and Kahn, B. Survey meters and electroscopes for monitoring radio-activity in water. J. Am. Water Works Assoc. **46**, 55-65 (1954); C.A. **48**:4734 c. **AnC**.

Lagemann, R. T., Gilley, L. W., and McLeroy, E. G. L3 The ultrasonic velocity, density, and compressibility of supercooled water and D₂O. J. Chem. Phys. **21**, 819-21 (1953); C.A. **47**:7849 h. **MeAc**, **MeD**, **ThF**, **SdTr**.

Laidler, K. J. See Markham, M. C.

Laidler, K. J., Wall, M. C., and Markham, M. C. Molecular kinetics of hydrogenation and exchange reactions of ethylene at surfaces. J. Chem. Phys. **21**, 949-50 (1953); C.A. **47**:11912h. **EqH**, **KiH**.

Lamb, W. E., Jr. Data on level shifts (in hydrogen). L5 deuterium, singly ionized helium, and positronium. Physica **19**, 832 (1953); C.A. **48**:3082 g. **SpEl**

See Dayhoff, E. S.; Triebwasser, S.

Lancaster, J. E. See Crawford, B. L., Jr.

Landler, Y. See Prévot-Bernas, A.

Langdon, R. G., and Bloch, K. The utilization of squalene L6 in the biosynthesis of cholesterol. J. Biol. Chem. **200**, 135-44 (1953); C.A. **47**:3953 a. **Sy**.

Langer, L. M., and Moffat, R. J. D. The β -spectrum of L7 tritium and the mass of the neutrino. Phys. Rev. **88**, 689-94 (1952); C.A. **47**:3136 f. **NuB**.

Lanius, K. The properties of the new Afga K2 plates for L8 nuclear tracks. Z. wiss. Phot. **48**, 243-62 (1953); C.A. **48**:8096 h. **AnC**.

Lanseau, K. P. See Morgan, W. A.

Larnaudie, M. Isomerism of monosubstituted derivatives of cyclohexane. Compt. rend. **235**, 154-56 (1952); C.A. **47**:366 e. **SpVi**, **St**.

Larson, N. R. See Bernstein, H. J.; Schubert, W. M.

Lauer, W. M., and Noland, W. E. The nitration of L10 monodeuterobenzene. J. Am. Chem. Soc. **75**, 3689-92 (1953); C.A. **48**:10628 f. **IsKi**, **KiL**.

Launay, J. de. The isotope effect in superconductivity. L11 Phys. Rev. **93**, 661-65 (1954); C.A. **48**:7358 b. **IsEl**.

Lauzon, R. See Jones, R. N.

Lawson, J. R. See Fuson, N.

Leach, S. An experimental value of the force constant of the SH radical. J. Chem. Phys. **22**, 1261 (1954); C.A. **48**:12551 d. **StD**, **SpEl**.

Leach, S., and Pannetier, G. The origin of the α -band of ammonia. J. phys. radium **15**, 413-16 (1954); C.A. **48**:11188 a. **SpEl**, **ElGd**.

Lecomte, J. The infrared spectrum of water in solids. L14 I. General introduction and formulation of the question. A review. J. chim. phys. **50**, C53-64 (1953); C.A. **48**:4977 h. **Ge**, **SpVi**.

See Duval, C.; Manneback, C.

Lecomte, J., Ceccaldi, M., and Roth, E. The isotopic determination of water by measuring the absorption in the infrared. J. chim. phys. **50**, 166-69 (1953); C.A. **47**:9149 a. **AnSp**.

Lee, G. H. Mass spectrometer for separation of isotopes. L16 U. S. Patent 2,659,822, Nov. 17, 1953. C.A. **48**:1172 h. **SeMs**.

Leitch, L. C. Synthesis of organic deuterium compounds. L17 V. The chlorination of propyne and propyne-d₄. Can. J. Chem. **31**, 385-86 (1953); C.A. **48**:2559 i. **Sy**.

See Bannard, R. A. B.; Bevege, E. E.; Nolin, B.

Leitch, L. C., and Morse, A. T. Synthesis of organic deuterium compounds. III. 1,2-Dibromoethane-d₄ and its derivatives. Can. J. Chem. **30**, 924-32 (1952); C.A. **47**:10455 d. **Sy**, **ThP**, **IsTh**.

L18 Synthesis of organic deuterium compounds. VII. Deuteriated 3-pentanones. Can. J. Chem. **31**, 785-92 (1953); C.A. **48**:10547 e. **Sy**.

Lemmon, R. M. See Biggs, M. W.

Lenormant, H., and Blout, E. R. Origin of the absorption band at 1550 cm⁻¹ in proteins. Nature **172**, 770-71 (1953); C.A. **48**:4017 e. **InSt**, **SpVi**.

L21 The infrared spectra of proteins in solution. Effect of pH and temperature increase. Bull. soc. chim. France, 859-62 (1954); C.A. **48**:12557 f. **InSt**, **SpVi**, **EqL**.

Leo, A., and Westheimer, F. H. The chemistry of diisopropyl chromate. J. Am. Chem. Soc. **74**, 4383-86 (1952); C.A. **47**:9253 d. **KiL**, **Sy**.

- Levi, C. See Marx, W.
- Levy, H. A., and Peterson, S. W. Neutron-diffraction determination of the crystal structure of ammonium bromide in four phases. *J. Am. Chem. Soc.* **75**, 1536-42 (1953); C.A. 47:7854a. **SdCr.**
- L23**
- L24** Neutron-diffraction study of sodium chloride-type modification of ammonium-d₄ bromide and iodide. *J. Chem. Phys.* **21**, 366 (1953); C.A. 47:9097 e. **SdCr.**
- Lewis, E. S., and Boozer, C. E. Isotope effects in the ionization of alkylchlorosulfites. *J. Am. Chem. Soc.* **74**, 6306-07 (1952); C.A. 47:11917 a. **KiL, IsKi, Sy.**
- L25**
- L26** Lewis, F. A., Roberts, G. E., and Ubbelohde, A. R. Electrolytic cold-working and internal friction in palladium-hydrogen alloys. *Proc. Roy. Soc. (London)* **A220**, 279-89 (1953); C.A. 48:5059 h. **SoG.**
- Libby, W. F. Potential usefulness of natural tritium. *Proc. Natl. Acad. Sci. U. S.* **39**, 245-47 (1953); C.A. 47:7906 e. **Ge.**
- L27** See Kaufman, S.
- Lieberman, A. L. See Setkina, V. N.
- Liebenberg, D. H. See Ingersoll, I. L. R.
- Lifson, N. See Johnson, J. A.
- Linderholm, H. See Garby, L.
- Linderström-Lang, K. See Hvidt, A.
- Lipscomb, W. N. See Nordman, C. E.
- Livingston, R. See Weigl, J. W.
- Llewellyn, D. R. See Dostrovsky, I.
- Loewus, F. A., Ofner, P., Fisher, H. F., Westheimer, F. H., and Vennesland, B. The enzymic transfer of hydrogen. II. The reaction catalyzed by lactic dehydrogenase. *J. Biol. Chem.* **202**, 699-704 (1953); C.A. 47:8798 f. **InBi, KiB.**
- L28**
- Long, C. N. H. See Mankin, H.
- Long, C. W. de, Thompson, R. C., and Kornberg, H. A. Percutaneous absorption of tritium oxide. *Am. J. Roentgenol. Radium Therapy, Nuclear Med.* **71**, 1038-45 (1954); C.A. 48:1168 f. **BiZ.**
- L29**
- Long, F. A. See Gold, V.
- Looney, F. S., Jr. The homogeneous catalysis of the thermal trans-cis isomerization of dideuterioethylene. *Univ. Microfilms (Ann Arbor, Mich.)*, Publ. No. 8354, 113 pp. Dissertation Abstr. **14**, 1153 (1954); C.A. 48:13374 b. **KiG.**
- See Rabinovitch, B. S.
- López de Azcona, J. M. Geophysical interpretation of natural isotopic variations of the chemical elements. *Rev. cienc. apl. (Madrid)* **7**, 193-202 (1953); C.A. 48:4321 d. **AbG.**
- L31**
- Lord, R. C. Synthesis of cyclobutene-d₈ and cyclobutane-d₈. *J. Chem. Phys.* **21**, 378 (1953); C.A. 48:139 e. **Sy.**
- L32** See Corrsin, L.
- Lord, R. C., and Merrifield, R. E. Strong hydrogen bonds in crystals. *J. Chem. Phys.* **21**, 166-67 (1953); C.A. 47:9777 i. **SpVi, SdSp, IsSp, StA.**
- L33**
- Lord, R. C., and Phillips, W. D. Exchange reactions of γ -pyrone and synthesis of deuterated pyrones. *J. Am. Chem. Soc.* **74**, 2429-30 (1952); C.A. 48:7605 e. **Sy, EqL.**
- L34**
- Lord, R. C., and Steese, C. M. The infrared spectra of monochlorogermane and monochlorogermane-d₃. *J. Chem. Phys.* **22**, 542-46 (1954); C.A. 48:7435 a. **SpVi, ThF.**
- L35**
- Lord, R. C., and Venkateswarlu, P. The rotation-vibration spectra of allene and allene-d₄. *J. Chem. Phys.* **20**, 1237-47 (1952); C.A. 47:9775 f. **SpVi, StD.**
- L36**
- L37** The infrared spectra of propylene and propylene-d₆. *J. Opt. Soc. Amer.* **43**, 1079-85 (1953); C.A. 48:1152 i. **SpVi.**
- L38** Lossing, F. P., Ingold, K. U., and Henderson, I. H. S. Free radicals by mass spectrometry. V. The ionization potentials of methyl, allyl, and benzyl radicals. *J. Chem. Phys.* **22**, 621-25 (1954); C.A. 48:7995 e. **Sr.**
- Lundberg, L. See Isberg, P.
- Luther, H. See Goubeau, J.
- Lyons, H. See Nuckolls, R. G.

M

- Ma, S. T. Interpretation of the virtual level of the deuteron. *Revs. Mod. Phys.* **25**, 853-60 (1953); C.A. 48:4321 e. **NuS.**
- M1**
- McCoubrey, J. C., Parke, J. B., and Ubbelohde, A. R. The transfer of vibrational energy in hydrocarbons. *Proc. Roy. Soc. (London)*, **A223**, 155-66 (1954); C.A. 48:7946 d. **MeAc.**
- M2**
- McDonald, G. Calculated approximate values of the free-energy function for the OD molecule. *J. Am. Chem. Soc.* **74**, 5539-40 (1952); C.A. 47:10919 b. **ThF.**
- M3**
- McDonell, W. R., and Kierstead, H. A. Expansion of copper bombarded by 21-m. e. v. deuterons. *Phys. Rev.* **93**, 247 (1954); C.A. 48:6289 a. **NuIn.**
- M4**
- Mackenzie, H. A. E., and Malherbe, P. N. Isotopes. **M5** II. Isotopic equilibrium during the fractional distillation with appreciable hold-up in the stillpot. *J. S. African Chem. Inst. [N. S.]* **4**, 63-68 (1951); C.A. 47:5742 e. **SeDs.**
- M6** Isotopes. III. Performance of stainless steel gauze ring packing in the isotopic fractionation of water. *J. S. African Chem. Inst. [N. S.]* **4**, 69-78 (1951); C.A. 47:5742 e. **SeDs.**
- Mackenzie, H. A. E., and Milner, A. M. Isotopes. I. **M7** Isotopic composition of water which has been enriched in heavy isotopes by fractionation. *J. S. African Chem. Inst. [N. S.]* **4**, 57-62 (1951); C.A. 47:5742 d. **SeDs.**
- McLeroy, E. G. See Lagemann, R. T.
- McNally, J. R., Jr. Atomic spectroscopy and separated isotopes. *Am. J. Phys.* **20**, 152-60 (1952); C.A. 47:38 g. **IsSp.**
- M8**
- McNesby, J. R., Davis, T. W., and Gordon, A. S. The mechanism of acetone pyrolysis. *J. Chem. Phys.* **21**, 956 (1953); C.A. 47:8483 b. **KiG.**
- M9**
- M10** Pyrolysis of mixtures of acetone and acetone-d₆. *J. Am. Chem. Soc.* **76**, 823-27 (1954); C.A. 48:6213 d. **KiG.**
- McNesby, J. R., and Gordon, A. S. Photolysis of acetone. **M11** *J. Am. Chem. Soc.* **76**, 1416-18 (1954); C.A. 48:6213 e. **KiP.**
- McNulty, W. P., Jr., and Hutchinson, F. Deuteron inactivation of dry bovine serum albumin. *Arch. Biochem. Biophys.* **50**, 92-103 (1954); C.A. 48:8841 d. **BiC.**
- M12**
- Magat, M. See Prévot-Bernas, A.
- Magee, J. L. Theory of the chemical reaction rate constant. *Proc. Natl. Acad. Sci. U. S.* **38**, 764-70 (1952); C.A. 47:2579 e. **EqG, KiG.**
- M13**
- Magee, J. L., and Gurnee, E. F. Mechanism of bond rupture in HBr⁸⁰ following isomeric transition. *J. Chem. Phys.* **20**, 894-98 (1952); C.A. 47:2055 b. **KiG.**
- M14**
- Majury, T. G., and Steacie, E. W. R. The reactions of methyl and methyl-d₃ radicals with hydrogen and deuterium. *Can. J. Chem.* **30**, 800-14 (1952); C.A. 47:2582 i. **KiG, KiP, IsKi.**
- M15**
- M16** The reactions of methyl radicals with the hydrogen isotopes. *Discussions Faraday Soc.*, No. 14, 45-49 (1953); C.A. 48:431 b. **KiP.**
- Malenka, B. J. See Ramsey, N. F.
- Malherbe, P. N. See Mackenzie, H. A. E.
- Mandeles, S., Koppelman, R., and Hanke, M. E. Deuterium studies on the mechanism of enzymic amino acid decarboxylation. *J. Biol. Chem.* **209**, 327-36 (1954); C.A. 48:10811 d. **InBi, KiB.**
- M17**
- Mankin, H., Stevenson, J. A. F., Brobeck, J. R., Long, C. N. H., and Stetten, D., Jr. Turnover of body fat in obesity resulting from hypothalamic injury studied with the aid of deuterium. *Endocrinology*, **47**, 443-47 (1950); C.A. 47:1248 i. **InBi.**
- M18**

- Mann, D. E. The vibrational spectra of N-dimethyl-**M19** aminodiborane. *J. Chem. Phys.* **22**, 70-78 (1954); C.A. **48**:4976c. **SpVi, St.**
- Mann, J. See Marrinan, H. J.
- Manneback, C., and Lecomte, J. Totally symmetric modes of vibration of ethylene (substituted or unsubstituted), principally of molecules of the type $R_2C:CR_2$. *J. phys. radium* **13**, 428 (1952); C.A. **48**:1152 c. **SpVi.** See Charette, J.
- Manneback, C., and Rahman, A. Potential function for the out-of-the-plane vibrations of vinyl bromide and the seven deuteriovinyl bromides [including stereoisomers], and the effect of anharmonicity on the calculation. *Ann. soc. sci. Bruxelles, Ser. I*, **67**, 28-67 (1953); C.A. **47**:6767 b. **SpVi, StD.**
- M22** Potential function for the vibrations of $H_2C=CBR_2$, $HDC=CBR_2$, $D_2C=CBR_2$. *Ann. soc. sci. Bruxelles Ser. I*, **67**, 140-76 (1953); C.A. **48**:1809 a. **SpVi.**
- Marano, B. See Perri, G. C.
- Marano, B. J. See Eidinoff, M. L.
- Marchetti, C. See Cerral, E.
- Markham, M. C. See Laidler, K. J.
- Markham, M. C., Wall, M. C., and Laidler, K. J. Molecular kinetics and mechanism of methane-deuterium exchange reactions on nickel. *J. Phys. Chem.* **57**, 321-23 (1953); C.A. **47**:6234 a. **EqH, KiH.**
- M24** Mechanisms of exchange and hydrogenation reactions of olefins. *J. Chem. Phys.* **20**, 1331-32 (1952); C.A. **48**:1938 i. **EqH, KiH.**
- Marmol, P. del See Mund, W.
- Marrian, D. H. See Eidinoff, M. L.
- Marrinan, H. J., and Mann, J. A study by infrared spectroscopy of hydrogen bonding in cellulose. *J. Appl. Chem. (London)* **4**, 204-11 (1954); C.A. **48**:12401 e. **SpVi, St.**
- Marshall, R. O. See Smith, G. N.
- Martin, R. E. Hydrogen chloride—the exchange reaction between tritium hydride and hydrogen chloride—red hydrogen chloride, and impurity color center. *Univ. Microfilms (Ann Arbor, Mich.) Publ. No. 8052*, 172 pp. *Dissertation Abstr.* **14**, 936 (1954); C.A. **48**:9793 g. **EqG, KiG.**
- Marx, J. W. See Cooper, H. G.
- Marx, J. W., Cooper, H. G., and Henderson, J. W. Radiation damage and recovery in copper, silver, gold, nickel, and tantalum. *Phys. Rev.* **88**, 106-12 (1952); C.A. **47**:410 h. **Ge.**
- Marx, J. W., Koehler, J., and Wert, C. Effect of deuteron bombardment on the electrical resistivity of copper, silver, and gold. An abstract. *Phys. Rev.* **86**, 643 (1952); C.A. **47**:6780 c. **Ge.**
- M28**
- Marx, W., Gustin, S. T., and Levi, C. Effects of thyroxine, thyroidectomy, and lowered environmental temperature on incorporation of deuterium into cholesterol. *Proc. Soc. Exptl. Biol. Med.* **83**, 143-46 (1953); C.A. **47**:9458 c. **BiC.**
- M29**
- Maryott, A. A. See Birnbaum, G.
- Maslov, P. G., and Antipina, S. A. Application of weighted coefficients in the determination of the vibrational spectra of polyatomic atoms. I. Formulas for finding the distribution of forces for molecular bonds and the interacting frequencies of vibrating molecules. II. Weighted coefficients, form of the oscillations, distribution of forces, and the analysis of interacting frequencies for acetylene and deuteriated acetylenes. *Zhur. Exptl. i Teoret. Fiz.* **22**, 172-83 (1952); C.A. **48**:3142 g. **SpVi.**
- Mason, W. P. Properties of a tetragonal antiferroelectric crystal. *Phys. Rev.* **88**, 480-84 (1952); C.A. **47**:2559 f. **EID, SdCr, SdTr.**
- M31**
- Mason, W. P., and Matthias, B. T. The piezoelectric, dielectric, and elastic properties of $(ND_4)_2D_2PO_4$ (deuteriated ADP). *Phys. Rev.* **88**, 477-79 (1952); C.A. **47**:2559 d. **EID, SdCr, SdTr.**
- M32**
- M33** Piezoelectric-transducer material. U. S. Patent **2,669,666**, Feb. 16, 1954; C.A. **48**:8684 h. **EID, SdTr, Sy.**
- Massey, J. T. See Jen, C. K.
- Massey, J. T., and Bianco, D. R. The microwave spectrum of hydrogen peroxide. *J. Chem. Phys.* **22**, 442-48 (1954); C.A. **48**:7437 c. **SpM, EID, St.**
- M34**
- Masuda, Y. See Sugawara, T.
- Matheson, M. S. See Smaller, B.
- Matsubara, T., and Yoshimori, A. Energy of the lowest state of molecular crystals. *Busseiron Kenkyū No. 59*, 80-87 (1953); C.A. **47**:5244 g. **SdCr.**
- M35**
- Matsuda, H. See Ogata, K.
- Matsuura, T. Lithium aluminum hydride. A review. *Kagaku no Ryōiki*, **7**, 733-49, 782 (1953); C.A. **48**:13507 a. **Ge.**
- M36**
- Mattauch, J., and Bieri, R. New doublet measurements. *Natl. Bur. Standards (U. S.) Circ.* **522**, 71-74 (1953); C.A. **47**:10341 b. **Sr.**
- M37**
- M38** A new mass-spectrographic determination of the masses of H^1 , D^2 , He^4 , C^{12} , and N^{14} . *Z. Naturforsch.* **9a**, 303-23 (1954); C.A. **48**:9805 h. **Sr, NuM.**
- Matthias, B. T. Isotope effect in RbD_2PO_4 . An abstract. *Phys. Rev.* **85**, 723-24 (1952); C.A. **47**:6716 i. **IsCr.**
- M39** See Mason, W. P.
- Matraw, H. C. See Dorfman, L. M.
- Maxwell, E. The isotope effect in superconductivity. *Physics Today*, No. 12, 5, 14-18 (1952); C.A. **47**:10922 g. **IsEL.**
- M40**
- Maybury, P. C., and Koski, W. S. Kinetics of two exchange reactions involving diborane. *J. Chem. Phys.* **21**, 742-47 (1953); C.A. **47**:6745 h. **EqG, EqH, KiG, KiH.**
- M41**
- Mayo, P. de See Barton, D. H. R.
- Mazia, D. See Prescott, D. M.
- Meal, H. C. Correlation of chlorine³⁵ nuclear quadrupole coupling frequencies with Hammett's sigma. *J. Am. Chem. Soc.* **74**, 6121-22 (1952); C.A. **47**:11973 i. **NuRe.**
- M42**
- Megaw, H. D. Ferroelectricity and crystal structure. II. *Acta Cryst.* **7**, 187-89 (1954); C.A. **48**:7362 g. **SdCr.**
- M43**
- Meister, A. G. See Dowling, J. M.; Pontarelli, D. A.
- Meister, A. G., and Voelz, F. L. Substituted methanes. XI. Potential constants for chlorotribromomethane. *J. Chem. Phys.* **21**, 158 (1953); C.A. **47**:3633 i. **SpVi.**
- M44**
- Melander, L. See Berglund-Larsson, U.; Olsson, S.
- Merrifield, R. E. See Lord, R. C.
- Meyer, F., and Eigen, M. Thermal conductivity of heavy water. *Z. Naturforsch.* **8a**, 500-01 (1953); C.A. **48**:5633 e. **ThD.**
- M45**
- Michel, A. See Schuler, H.
- Michels, A., Schipper, A. C. J., and Rintoul, W. H. The viscosity of hydrogen and deuterium at pressures up to 2000 atmospheres. *Physica* **19**, 1011-28 (1953); C.A. **48**:7365 i. **MeV.**
- M46**
- Miklukhin, G. P., and Rekasheva, A. F. Mechanism of the reduction of diazonium salts. Reduction with hypophosphorus acid. *Doklady Akad. Nauk. S.S.S.R.* **85**, 827-30 (1952); C.A. **47**:4855 f. **EqL, KiL.**
- M47**
- Mikovsky, R. J., Boudart, M., and Taylor, H. S. Hydrogen-deuterium exchange on copper, silver, gold, and alloy surfaces. *J. Am. Chem. Soc.* **76**, 3814-19 (1954); C.A. **48**:13387 bc. **EqH, KiH.**
- M48**
- Milatz, J. M. W. See Kluyver, J. C.
- Miller, S. L., Aamodt, J. C., Dousmanis, G. C., Townes, C. H., and Kraitchman, J. Structure of the methyl halides. *J. Chem. Phys.* **20**, 1112-16 (1952); C.A. **47**:4745 g. **SpM, StD, Is.**
- M49**
- Miller, W. C. See Noyes, J. C.
- Milner, A. M. See Mackenzie, H. A. E.
- Misenta, R. See Becker, E. W.
- Mitchell, L. G. See Bok, J. D. C.
- Miyashiro, S. See Uemura, Y.

- Moffat, R. J. D. β -Spectrum of tritium and the mass of the neutrino. Univ. Microfilms (Ann Arbor, Mich.), Publ. No. 5872, 59 pp. Dissertation Abstr. **13**, 848-49 (1953); C.A. **48**:1819 h. **NuB**. See Langer, L. M.
- Mohler, F. L., Dibeler, V. H., Williamson, L., and Dean, H. Mass spectra of deuterioacetylenes, mono-deuteriobenzene, and deuterionaphthalenes. J. Research Natl. Bur. Standards **48**, 188-92 (1952); RP 2304; C.A. **47**:2034 e. **Sr**, **IsMs**.
- Molinari, E. See Boato, G.
- Molinari, E., and Parravano, G. The hydrogen-deuterium exchange reaction on zinc oxide catalysts. J. Am. Chem. Soc. **75**, 5233-37 (1953); C.A. **48**:2457 gh. **EqH**, **KiH**.
- Momigny, J. The mass spectra of monosubstituted benzene derivatives. Phenol, monodeuteriophenol, thiophenol, and aniline. Bull. soc. Roy. sci. Liège **22**, 541-60 (1953); C.A. **48**:9806 b. **Sr**.
- Moore, G. E., and Badger, R. M. The infrared spectra and structure of the chloramines and nitrogen trichloride. J. Am. Chem. Soc. **74**, 6076-80 (1952); C.A. **47**:3697 b. **SpVi**, **StD**.
- Morgan, G. H. See Broida, H. P.
- Morgan, W. A., and Jernakoff, G. Ion resonance mass spectrometer. Ind. Eng. Chem. **46**, 1404-49 (1954); C.A. **48**:11850 g. **AnMs**, **Sr**.
- Morita, H. See Price, C. C.
- Morowitz, H. J. See Broida, H. P.; Freis, E. D.
- Morowitz, H. J., and Broida, H. P. Rapid spectroscopic determination of total water content. Anal. Chem. **24**, 1657-58 (1952); C.A. **47**:1006 a. **AnSp**.
- Morozov, V. P. Rule of order for nonlinear XY_2 molecules. Zhur. Fiz. Khim. **27**, 233-36 (1953); C.A. **48**:2477 b. **IsSp**, **StD**.
- Morrison, J. L. See Brickman, W. J.
- Morse, A. T. See Bannard, R. A. B.; Leitch, L. C.
- Mosburg, E. R., Jr. See Hutchinson, F.
- Moszkowski, S. A., and Townes, C. H. The tendency for positive nuclear quadrupole moments. Phys. Rev. **93**, 306-12 (1954); C.A. **48**:6270 i. **NuQ**, **NuS**.
- Mund, W., Feyten, J., and Marmol, P. del. Determination of vapor pressures as criterion for the purity of organic compounds. Bull. classe. sci., Acad. roy. Belg. **39**, 666-75 (1953); C.A. **48**:2431 h. **ThP**.
- Myers, R. J. See Hrostowski, H. J.
- Myers, R. J., and Gwinn, W. D. The microwave spectra, structure, dipole moment, and chlorine nuclear quadrupole coupling constants of methylene chloride. J. Chem. Phys. **20**, 1420-27 (1952); C.A. **47**:9697 e. **SpM**, **StD**, **EID**.
- Myerson, A. A. See Schmitt, J. A.
- ## N
- Nazarov, I. N. See Kursanov, D. N.
- Nazarov, I. N., Zaretskaya, I. I., Parnes, Z. N., and Kursanov, D. N. The mechanism of the cyclization reaction by means of deuterium. II. Izvest. Akad. Nauk S. S. S. R. Otdel. Khim. Nauk, 519-23 (1953); C.A. **48**:9930 b. **InKi**.
- Neal, W. E. J. See Grew, K. E.
- Nelson, L. C., and Obert, E. F. Generalized PVT properties of gases. Trans. Am. Soc. Mech. Engrs. **76**, 1057-66 (1954); C.A. **48**:13307 c. **Ge**, **ThF**.
- Nethercot, A. H., Jr. See Klein, J. A.
- Nevin, T. E., and Stephens, D. V. Band spectra of MnH and MnD. Proc. Roy. Irish Acad. **55A**, 109-16 (1953); C.A. **47**:7889 g. **SpEl**, **SpVi**.
- Nicholson, A. J. C. The photolysis of the aliphatic aldehydes and ketones. A review. Revs. Pure and Appl. Chem. (Australia) **2**, 174-84 (1952); C.A. **47**:1481 h. **Ge**, **KiP**.
- Nieklas, J. P. Isotope shift in neutral carbon. Phys. Rev. **95**, 1469-71 (1954); C.A. **48**:13415 i. **IsSp**.
- Nielsen, A. H. See Keller, F. L.
- Nielsen, H. H. Molecular structure of arsine. J. Chem. Phys. **20**, 1955-56 (1952); C.A. **47**:9774 c. **StD**. See Haynie, W. H.
- Nightingale, R. E., and Wagner, E. L. The vibrational spectra and structure of solid hydroxylamine and deuteriohydroxylamine. J. Chem. Phys. **22**, 203-08 (1954); C.A. **48**:4975 d. **SpVi**, **St**.
- Niini, R. A thermal diffusion apparatus with moving walls. Suomen Kemistilehti **26B**, 42-45 (1953); C.A. **48**:407 h. **SeDf**.
- Noetzelin, J. Measurement of isotopic ratios and its possible applications to volcanology. Bull. volcanol. **12**, 115-25 (1952); C.A. **47**:5322 d. **AbG**.
- Noland, W. E. See Lauer, W. M.
- Nolin, B. Synthesis of organic deuterium compounds. VIII. Deuterium substitution products of ethyl acetate. Can. J. Chem. **31**, 1257-59 (1953); C.A. **48**:3247 h. **Sy**. See Jones, R. N.
- Nolin, B., and Jones, R. N. The infrared absorption spectra of diethyl ketone and its deuterium substitution products. J. Am. Chem. Soc. **75**, 5626-28 (1953); C.A. **48**:1812 i. **SpVi**.
- Nolin, B. Preparation of some deuterated steroids. Can. J. Chem. **30**, 727-33 (1952); C.A. **48**:3379 i. **Sy**.
- Nolin, B., and Leitch, L. C. Synthesis of organic deuterium compounds. IV. Methyl- d_3 bromide. Can. J. Chem. **31**, 153-57 (1953); C.A. **48**:2559 h. **Sy**.
- Nordman, C. E., and Lipscomb, W. N. Isotope effect and thermal expansion coefficients of crystals. J. Chem. Phys. **21**, 2077 (1953); C.A. **48**:1791 e. **IsCr**, **StD**, **ThF**.
- Norton, F. J. Helium diffusion through glass. J. Am. Ceram. Soc. **36**, 90-96 (1953); C.A. **47**:4568 h. **MeDf**.
- Norton, F. J. Mass-spectrometer effects at higher pressures. Natl. Bur. Standards (U. S.) Circ. **522**, 201-04 (1953); C.A. **47**:10342 d. **Sr**.
- Norton, F. J. Diffusion of deuterium from deuterium oxide through steel. J. Appl. Phys. **24**, 499 (1953); C.A. **47**:10436 e. **In**, **MeDf**.
- Nostrand, E. D., and Duncan, A. B. F. Effect of pressure on intensity of some electronic transitions in SF_6 , C_2H_2 , and C_2D_2 vapors in the vacuum ultraviolet region. J. Am. Chem. Soc. **76**, 3377-79 (1954); C.A. **48**:13420 i. **SpEl**.
- Nottorf, R. W. Some problems in chemistry of uranium and thorium of interest to the development of atomic power. Iowa State Coll. J. Sci. **26**, 255-57 (1952); C.A. **47**:2052 g. **MeD**, **SeCh**, **ThP**, **IsTh**, **Sd**.
- Noyes, J. C., Hoomissen, J. E. van, Miller, W. C., and Waldman, B. Photodisintegration thresholds of deuterium and beryllium. Phys. Rev. **95**, 396-403 (1954); C.A. **48**:11212 h. **NuR**.
- Nuckolls, R. G. See Rueger, L. J.
- Nuckolls, R. G., Rueger, L. J., and Lyons, H. Measurement of the microwave inversion spectrum of ND_3 . An abstract. Phys. Rev. **83**, 880-81 (1951); C.A. **47**:6755 a. **SpM**.
- Nuckolls, R. G. Microwave absorption spectrum of ND_3 . Phys. Rev. **89**, 1101 (1953); C.A. **47**:8517 b. **SpM**.
- ## O
- Oakes, B. D. See Hine, J.
- Oakwood, T. S. See Shull, E. R.
- Oana, S. Distribution of heavy water in natural waters. J. Earth Sci. Nagoya Univ. **1**, 42-61 (1953); C.A. **47**:11607 g. **AbG**.
- Obert, E. F. See Nelson, L. C.
- Odeblad, E., and Tobias, C. A. Autoradiography after activation of histological sections with deuterons. Arch. Biochem. Biophys. **49**, 452-53 (1954); C.A. **48**:8288 i. **BiC**, **BiZ**.
- Oetjen, R. A. See Hansler, R. L.; Stroup, R. E.
- Ofer, P. See Loewus, F. A.

- Ogata, K., and Matsuda, H. Masses of light atoms. **O3** *Phys. Rev.* **89**, 27-32 (1953); *C.A.* **47**:4190 i. **NuM, Sr.**
- O4** Standards of atomic mass. *Natl. Bur. Standards (U. S.) Circ.* **522**, 59-66 (1953); *C.A.* **47**:10340 i. **Sr, NuM.**
- Olsson, S., and Melander, L. Acid-induced hydrogen exchange in benzene derivatives as electrophilic aromatic substitution. *Acta. Chem. Scand.* **8**, 523-25 (1954); *C.A.* **48**:11163 f. **EqL, KiL.**
- Ono, Y. See Uemura, Y.
- Orban, E. See Heiks, J. R.
- Ordal, E. J. See Curtis, W.
- O'Reilly, E. J., Jr. The infrared and Raman vibrational assignment of naphthalene and naphthalene-*d*₈. *Univ. Microfilms (Ann Arbor, Mich.), Pub. No. 6335*, 57 pp. *Dissertation Abstracts* **13**, 1005 (1953); *C.A.* **48**:4317 i. **SpVi.**
- Orgel, L. E. See Cardwell, H. M. E.
- Orsoni, B. Preconcentration of heavy water in electrolysis plants. *Proc. XIth Intern. Congr. Pure Appl. Chem. (London)* **5**, 1057-63 (1947); (*Pub.* 1953); *C.A.* **47**:7918 e. **SeEl.**
- Otozai, K., Kume, S., and Fukushima, S. Entropies of closure, resonance, bond, and activation. *Catalyst (Japan)* No. 9, 53-68 (1953); *C.A.* **48**:1742 g. **ThF, ThS.**
- Otvos, J. W. See Stevenson, D. P.; Wagner, C. D.; Wilson, J. N.
- Overend, J., and Thompson, H. W. Vibration-rotation bands and molecular structure of allene. *J. Opt. Soc. Amer.* **43**, 1065-70 (1953); *C.A.* **48**:1153 b. **SpVi, StD.**
- Overhauser, A. W. Stored-energy measurements in irradiated copper. *Phys. Rev.* **94**, 1551-57 (1954); *C.A.* **48**:9234 e. **NuIn.**
- Owen, M. E. See Jefferies, T. O.
- Oxley, C. L., Cartwright, W. F., and Rouvina, J. Double scattering of high-energy protons. *Phys. Rev.* **93**, 806-16 (1954); *C.A.* **48**:6854 g. **NuIn.**
- P**
- Pakhomova, N. L. See Sverdlov, L. M.
- Panchenko, V. G. Isotopes. *Priroda* [8], **39**, 16-26 (1950); *Chem. Zentr. I*, 1264 (1951); *C.A.* **47**:10353 d. **Ge.**
- Pancholy, M. Temperature variation of velocity and absorption coefficient of ultrasonic waves in heavy water. *J. Acoust. Soc. Amer.* **25**, 1003-06 (1953); *C.A.* **48**:2432 e. **MeAc.**
- Pannetier, G. See Leach, S.
- Parke, J. B. See McCoubrey, J. C.
- Parnes, Z. N. See Kursanov, D. N.; Nazarov, I. N.
- Parravano, G. See Molinari, E.
- Patrick, W. N. See Burton, M.
- Patrick, W. N., and Burton, M. Radiation chemistry of mixtures: propionaldehyde and benzene-*d*₆. *J. Phys. Chem.* **58**, 424-30 (1954); *C.A.* **48**:11936 a. **KiR.**
- Payne, P. R., and Done, J. Assay of tritium-labeled substances—combustion bomb method of preparation of gas for counting. *Nature (London)* **173**, 27-28 (1954); *C.A.* **48**:11524 b. **AnC.**
- Pearlman, M. R. J. See Pearlman, W. H.
- Pearlman, W. H., Pearlman, M. R. J., and Rakoff, A. E. Estrogen metabolism in human pregnancy; a study with the aid of deuterium. *J. Biol. Chem.* **209**, 803-12 (1954); *C.A.* **48**:13003 fg. **InBi.**
- Pearson, D. P. See Decius, J. C.
- Pease, R. L., and Feshbach, H. The theory of hydrogen³. *Phys. Rev.* **88**, 945-50 (1952); *C.A.* **47**:3127 f. **NuM, NuS.**
- Pease, R. N. See Brokaw, R. S.
- Peek, R. C., Jr. See Hine, J.
- Pelz, A. See Wagner, G.
- Penneman, R. A. See Jones, L. H.
- Perlman, I. See Hollander, J. M.
- Perlman, M. L. See Bigeleisen, J.
- Perlman, M. L., Bigeleisen, J., and Elliot, N. Equilibrium in the exchange of deuterium between ammonia and hydrogen. *J. Chem. Phys.* **21**, 70-72 (1953); *C.A.* **47**:3667 a. **EqG.**
- Perperot, H. Preliminary study of possible processes to prepare heavy water. *Mem. poudres* **32**, 429-36 (1950); *C.A.* **47**:8544 c. **SeCh.**
- Perri, G. C. See Eidinoff, M. L.
- Perri, G. C., Eidinoff, M. L., Knoll, J., and Marano, B. Preferential utilization of isotopes of hydrogen in biological systems. *Boll. soc. ital. biol. sper.* **30**, 134-35 (1954); *C.A.* **48**:10074 a. **BiC.**
- Person, W. B., and Pimentel, G. C. Infrared studies of mixed crystals: naphthalene-*d*₈ in naphthalene. *J. Chem. Phys.* **20**, 1913-16 (1952); *C.A.* **47**:10348 g. **SpVi, SdSp.**
- Peterson, S. W. See Levy, H. A.
- Petters, L. B. See Bok, L. C. D.
- Phillips, C. S. E. Dielectric properties of a sample of deuterium chloride containing an impurity. *Nature* **168**, 518-19 (1951); *C.A.* **47**:14 d. **EiD, SdEl, SdTr, Sy.**
- Phillips, W. D. See Lord, R. C.
- Pickworth, J., and Thompson, H. W. The fundamental vibration-rotation band of deuterium chloride. *Proc. Roy. Soc. (London)*, **A218**, 37-43 (1953); *C.A.* **47**:9769 g. **SpVi, IsSp, StD.**
- Pierce, F. T., Jr. See Biggs, M. W.
- Pimentel, G. C. See Hrostowski, H. J.; Klemperer, W.; Person, W. B.
- Pimentel, G. C., Garland, C. W., and Jura, G. Infrared spectra of heavy water adsorbed on silica gel. *J. Am. Chem. Soc.* **75**, 803-05 (1953); *C.A.* **47**:5253 a. **In, SpVi, EqH.**
- Piolet, C. See Corval, M.
- Plentl, A. A. See Hutchinson, D. L.
- Plentl, A. A., and Hutchinson, D. L. Determination of deuterium exchange rates between human maternal circulation and amniotic fluid. *Proc. Soc. Exptl. Biol. Med.* **82**, 681-84 (1953); *C.A.* **47**:8204 g. **KiB.**
- Plyler, E. K. See Benedict, W. S.
- Pollard, E. C. See Davis, M.; Preiss, J. W.; Woese, C.
- Pollard, E. C., and Dimond, A. E. Effect of deuterium bombardment on the serology of tobacco mosaic virus. An abstract. *Phys. Rev.* **86**, 612 (1952); *C.A.* **47**:6993 g. **BiB.**
- P16** Inactivation of southern bean mosaic virus by deuterons. *Phys. Rev.* **87**, 186 (1952); *C.A.* **48**:8332 a. **BiB.**
- Pollard, E. C., and Setlow, J. Effect of ionizing radiation on the serological affinity of T-1 bacteriophage. *Arch. Biochem. Biophys.* **50**, 376-82 (1954); *C.A.* **48**:10085 h. **BiC.**
- Polo, S. R., and Wilson, M. K. The infrared spectrum of CF₃D. *J. Chem. Phys.* **21**, 1129-31 (1953); *C.A.* **47**:9774 d. **SpVi, ThF, Sy.**
- Polyakov, Y. A. The possibility of toxic activity of heavy water on agricultural crops. *Pochvovedenie*, No. 7, 25-32 (1953); *C.A.* **48**:6061 d. **BiB.**
- Pontarelli, D. A., Meister, A. G., Cleveland, F. F., Voelz, F. L., Bernstein, R. B., and Sherman, R. H. Substituted methanes. IX. Raman and infrared spectra, assignments, potential constants, and calculated thermo-dynamic properties of CHClBr₂ and CCl₂Br₂. *J. Chem. Phys.* **20**, 1949-54 (1952); *C.A.* **47**:4740 i. **Sp.Vi, StD, ThF.**
- Poole, H. G. See Cohn, H.
- Pople, J. A. The viscosity of isotopic substances. *Physica* **19**, 668 (1953); *C.A.* **48**:3740 e. **MeV, Is.**
- Porter, J. W. See Weinberger, D.
- Porter, J. W., and Knauss, H. J. Inhibition of growth of *Chlorella pyrenoidosa* by β -emitting radioisotopes. *Plant. Physiol.* **29**, 60-63 (1954); *C.A.* **48**:6518 d. **BiZ.**
- Porter, J. W., and Watson, M. S. Gross effects of growth-inhibiting levels of tritium oxide on *Chlorella pyrenoidosa*. *Am. J. Botany* **41**, 550-55 (1954); *C.A.* **48**:12937 c. **BiB.**

- Posener, D. W., and Strandberg, M. W. P. Microwave spectrum of HDO. *J. Chem. Phys.* **21**, 1401-02 (1953); C.A. 47:11991 a. **SpM**.
- P25** Centrifugal distortion in asymmetric top molecules. III H_2O , D_2O , and HDO. *Phys. Rev.* **95**, 374-84 (1954); C.A. 48:11191 hi. **SpM**, **SpVi**.
- Powles, J. G. Dielectric relaxation and the internal field. *J. Chem. Phys.* **21**, 633-37 (1953); C.A. 47:6201 d. **EID**, **EIT**.
- Preiss, J. W., and Pollard, E. C. Energy loss of deuterons in hydrocarbons. An abstract. *Phys. Rev.* **86**, 612 (1952); C.A. 47:6757 e. **InSt**, **KiR**.
- Prentice, T. C., Siri, W., and Joiner, E. E. Quantitative studies of ascitic fluid circulation with tritium-labeled water. *Am. J. Med.* **13**, 668-73 (1952); C.A. 47:5519 g. **InBi**.
- Prescott, D. M., and Mazia, D. The permeability of nucleated and enucleated fragments of Amoeba proteus to D_2O . *Exptl. Cell Research* **6**, 117-26 (1954); C.A. 48:9564 h. **BiZ**.
- Prévot-Bernas, A., Chapiro, A., Cousin, C., Landler, Y., and Magat, M. Radiolysis of some organic liquids. *Discussions Faraday Soc.* No. 12, 98-132 (1952); C.A. 48:8079 e. **KiR**.
- Price, C. C., and Morita, H. The reaction of methyl radicals with isobutyryl and α -deuterioisobutyryl chlorides. *J. Am. Chem. Soc.* **75**, 3686-88 (1953); C.A. 48:10560 i. **KiL**.
- Prigogine, I., Brouckere, L. de, and Buess, R. Thermodiffusion in the liquid phase. V. Thermodiffusion of heavy water. *Physica* **18**, 915-20 (1952); C.A. 47:6723 i. **ThD**, **SeDf**.
- Prodell, A. G., and Kusch, P. The hyperfine structure of hydrogen and deuterium. *Phys. Rev.* **88**, 184-90 (1952); C.A. 47:965 b. **SpA**, **NuH**, **NuRe**, **NuMg**.
- Prosser, H. C. See Bigeleisen, J.
- Pucheault, J. See Haissinsky, M.
- Pullin, A. D. E. See Bernstein, H. J.
- Purcell, E. M. See Carr, H. Y.; Reif, F.
- Q**
- Quinan, J. R., and Wiberley, S. E. Assignment of the O-H deformation frequency. *J. Chem. Phys.* **21**, 1896-97 (1953); C.A. 48:1154 i. **SpVi**.
- R**
- Raaen, V. F. See Ropp, G. A.
- Rabinovitch, B. S. See Bernstein, H. J.; Douglas, J. E.; Schubert, W. M.
- Rabinovitch, B. S., Douglas, J. E., and Looney, F. S. Thermal trans-cis isomerization of dideuterioethylene. *J. Chem. Phys.* **20**, 1807-08 (1952); C.A. 48:30 e. **KiG**.
- Rabinovitch, B. S., and Looney, F. S. The stereochemistry of reduction of the acetylenic bond. *J. Am. Chem. Soc.* **75**, 2652-54 (1953); C.A. 48:7531 f. **KiH**, **Sy**.
- Rachele, J. R. See Verly, W. G.
- Rachele, J. R., Kuchinskas, E. J., Knoll, J. E., and Eidinoff, M. L. Isotopic selection in the neo-genesis of labile methyl groups from monodeuterio-, monotritio-, C^{14} -labeled methanol. *J. Am. Chem. Soc.* **76**, 4342-45 (1954); C.A. 48:12841 i. **BiC**.
- Rafter, G. W., and Colowick, S. P. The structure of reduced N^1 -methylnicotinamide. *J. Biol. Chem.* **209**, 773-77 (1954); C.A. 48:12205 c. **InKi**.
- Rahman, A. See Manneback, C.
- Rakoff, A. E. See Pearlman, W. H.
- Ramaiah, N. A., and Kaveeshwar, S. N. Joshi effect in H_2O and D_2O vapor. *J. Sci. Research Banaras Hindu Univ.* **2**, 1-42 (1951-52); C.A. 47:5245 g. **EIGd**.
- Ramsay, D. A. Absorption spectra of SH and SD produced by flash photolysis of H_2S and D_2S . *J. Chem. Phys.* **20**, 1920-27 (1952); C.A. 47:4734 a. **SpEl**. See Herzberg, G.
- Ramsey, N. F. Spin interactions of accelerated nuclei in molecules. *Phys. Rev.* **90**, 232-33 (1953); C.A. 47:9136 b. **NuMg**.
- R8** Electron coupled interactions between nuclear spins in molecules. *Phys. Rev.* **91**, 303-07 (1953); C.A. 47:10998 f. **NuS**. See Barnes, R. G.; Harrick, N. J.
- Ramsey, N. F., Malenka, B. J., and Kruse, U. E. Polarizability of the deuteron. *Phys. Rev.* **91**, 1162-64 (1953); C.A. 47:12013 g. **NuS**, **EIP**.
- Randolph, B. Temperature dependence of the resistivity of deuteron-irradiated and annealed molybdenum. An abstract. *Phys. Rev.* **85**, 710 (1952); C.A. 47:6776 c. **Ge**.
- Rank, D. H. See Shull, E. R.; Wiggins, T. A.
- Rastrup-Anderson, J. See Bak, B.
- Rebbert, R. E., and Steacie, E. W. R. Photolysis of mercury dimethyl with deuterium. *Can. J. Chem.* **32**, 113-16 (1954); C.A. 48:5656 h. **KiP**.
- Regan, C. M. See Roberts, J. D.
- Reif, F., and Purcell, E. M. Nuclear magnetic resonance in solid hydrogen. *Phys. Rev.* **91**, 631-47 (1953); C.A. 47:10998 a. **NuRe**, **SdEl**.
- Reilly, H. C. See Eidinoff, M. L.
- Reinebeck, L. An example of determination by approximation of the emission of a new spectrum produced by organic molecules. *J. phys. radium* **15**, 398-400 (1954); C.A. 48:12552 g. **SpVi**. See Schuler, H.
- Rekasheva, A. F. See Miklukhin, G. P.
- Renaud, R. See Bevege, E. E.
- Renzoni, R. See Cerral, E.
- Rice, F. O., and Varnerin, R. E. The mechanism of the thermal decomposition of ethane- d_6 . *J. Am. Chem. Soc.* **76**, 324-27 (1954); C.A. 48:6215 d. **KiG**.
- R15** The thermal decomposition of acetaldehyde. *J. Am. Chem. Soc.* **76**, 2629-33 (1954); C.A. 48:10412 i. **KiG**.
- Riet, R. van, Courtoy, C., and Hemptinne, M. de. Infrared spectra of CH_2D-CH_2D at high dispersion. *Ann. soc. sci. Bruxelles, Sér. I*, **67**, 105-21 (1953); C.A. 48:1808 h. **SpVi**.
- R17** High-dispersion infrared spectra of the molecule CD_3H-CD_2H . *Ann. soc. sci. Bruxelles, Sér. I*, **67**, 296-308 (1953); C.A. 48:6255 i. **SpVi**.
- R18** Experimental contribution to the study of the molecule CD_3CD_2H . *Ann. soc. sci. Bruxelles, Sér. I*, **68**, 108-18 (1954); C.A. 48:13428 i. **SpVi**.
- Riet, R. van, and Hemptinne, M. de. Infrared and Raman spectra of deuterioethanes. *Bull. classe sci., Acad. roy. Belg.* **39**, 273-84 (1953); C.A. 47:8515 g. **SpVi**.
- R20** The C—H, C—D, and C—C vibrations of the deuterio-substituted ethane molecule CH_3-DC_2D . *Ann. soc. sci. Bruxelles, Sér. I*, **68**, 92-107 (1954); C.A. 48:13428 g. **SpVi**.
- Rigg, T. See Kelly, P.
- Rintoul, W. H. See Michels, A.
- Rittenberg, D. See Krasna, A. I.
- Roberts, E. K. See Kistiakowsky, G. B.
- Roberts, G. E. See Lewis, F. A.
- Roberts, J. D., and Regan, C. M. Kinetics and some hydrogen-isotope effects of the reaction of diphenyldiazomethane with acetic acid in ethanol. *J. Am. Chem. Soc.* **74**, 3695-96 (1952); C.A. 47:9735 c. **IsKi**, **KiL**.
- Roberts, J. D., Regan, C. M., and Allen, I. Mechanism of acid catalysis. The kinetics and mechanisms of the acid-catalyzed reactions of ethyl diazoacetate with ethanol and acetic acid. *J. Am. Chem. Soc.* **74**, 3679-83 (1952); C.A. 47:392 c. **EqL**, **KiL**, **SpVi**.
- Roberts, L. D. See Dabbs, J. W. T.

- Roberts, R. See Darwent, B. deB.
 Robinson, C. V. See Wang, J. H.
 Roe, A., and Albenesius, E. L. Isotope effect. III. Reaction of acetone-1-C¹⁴ with alkaline hypiodite. *J. Am. Chem. Soc.* **74**, 2402-03 (1952); C.A. **48**:7533 i. **IsKi**.
- Rogers, T. F. Absolute intensity of water-vapor absorption at microwave frequencies. *Phys. Rev.* **93**, 248-49 (1954); C.A. **48**:6258 e. **SpM**.
- Rollefson, G. K. (Editor). *Ann. Rev. Phys. Chem.* **3**, **R25** 384 pp. (1952); C.A. **47**:11 i. **Ge**.
- Romaniw, M. Concentration of heavy water. *Belg.* **R26** 485, 904, Mar. 16, 1949; C.A. **48**:P8080 d. **SeCr**.
- Ronzio, A. R. Synthesis of deuteriopolythene. U. S. **R27** Atomic Energy Comm., Tech. Inform. Service, **LA-1478**, 3-36 (1952); C.A. **48**:11290 d. **Sy**.
- Ropp, G. A. Effect of isotope substitution on organic reaction rates. *Nucleonics* **10**, No. 10, 22-27 (1952); C.A. **48**:2559 b. **Ge, IsKi**.
- Ropp, G. A., and Raaen, V. F. A comparison of the magnitudes of the isotope intermolecular effects in the decarboxylations of malonic-1-C¹⁴ and malonic-2-C¹⁴ acids at 154°. *J. Am. Chem. Soc.* **74**, 4992-94 (1952); C.A. **47**:1476 d. **IsKi**.
- Ropp, G. A., Raaen, V. F., and Weinberger, A. J. Some carbon-14 isotope effects in organic chemistry. *J. Am. Chem. Soc.* **75**, 3694-97 (1953); C.A. **48**:13611 h. **IsKi**.
- Rosenthal, D. J., and Anger, H. O. Liquid scintillation counting of tritium- and carbon¹⁴-labeled compounds. U. S. Atomic Energy Comm., **UCRL-2320**, 17 pp (1953); C.A. **48**:5663 e. **AnC**.
- R32** Liquid scintillation counting of tritium- and carbon¹⁴-labeled compounds. *Rev. Sci. Instr.* **25**, 670-74 (1954); C.A. **48**:13451 i. **AnC**.
- Roseo, L. See Cerral, E.
 Ross, M. Evidence for nonadditivity of nucleon moments in heavy nuclei. *Phys. Rev.* **88**, 935-42 (1952); C.A. **47**:3112 c. **NuMg**.
- Roth, E. See Lecomte, J.
 Rouvina, J. See Oxley, C. L.
 Rowlinson, J. S. Viscosity of heavy-water vapor. **R34** *Physica* **19**, 303-04 (1953); C.A. **47**:10934 g. **MeV, Is**.
- Rueger, L. J. See Nuckolls, R. G.
 Rueger, L. J., and Nuckolls, R. G. Broad-band coaxial Stark cell for microwave spectroscopy. *Rev. Sci. Instr.* **23**, 635 (1952); C.A. **47**:3700 c. **SpM**.
- Rundle, R. E. The deuterium effect on hydrogen-bond distances in crystals. *J. Chem. Phys.* **21**, 937-38 (1953); C.A. **47**:8430 d. **IsCr**.
- Russell, G. A. See Brown, H. C.
 Russell, L. A. See Stephenson, C. C.
 Russell, M. C. B. See Cooke-Yarborough, E. H.
- S**
- Sakisaka, M. See Uemura, Y.
 Salemink, C. A. See Kögl, F.
 Salpeter, E. E. The Lamb shift for hydrogen and deuterium. *Phys. Rev.* **89**, 92-97 (1953); C.A. **47**:4193 b. **SpA**.
- S2** Salpeter, E. E., and Goldstein, J. S. Momentum space wave functions. II. The deuteron ground state. *Phys. Rev.* **90**, 983-86 (1953); C.A. **47**:9749 d. **NuS**.
- Sanders, T. M., Jr. See Dousmanis, G. C.
 Sandler, Y. L. The adsorption and the magnetic ortho-para conversion of hydrogen on diamagnetic solids. I. Some experiments in surface paramagnetism. *J. Phys. Chem.* **58**, 54-57 (1954); C.A. **48**:3751 f. **AdG**.
- S4** The ortho-para conversion of hydrogen and deuterium on inhomogeneous paramagnetic surfaces. *Can. J. Chem.* **32**, 249-60 (1954); C.A. **48**:11858 bc. **AdG, KiH**.
- Schaeffer, O. Mass-spectral patterns of isotopic molecules. *Natl. Bur. Standards (U. S.) Circ.* **522**, 249-53 (1953); C.A. **47**:10343 i. **Ge, Sr**.
- Schayer, R. W., Henderson, L. M., and Smiley, R. L. **S6** Conversion of deuterio-N¹⁵-tryptophan to quinolinic acid. *J. Biol. Chem.* **195**, 657-61 (1952); C.A. **47**:10036 e. **InBi**.
- Scheitlin, E. See Bernhard, K.
 Schemin, D. See Tatum, E. L.
 Schiessler, R. W. See Davis, R. T., Jr.; Dixon, J. A.
 Schipper, A. C. J. See Michels, A.
 Schmitt, J. A., and Daniels, F. The carbon-isotope effect in the acid hydrolysis of urea. *J. Am. Chem. Soc.* **75**, 3564-66 (1953); C.A. **48**:1125 d. **IsKi**.
- S8** Schmitt, J. A., Myerson, A. A., and Daniels, F. Relative rates of hydrolysis of urea containing carbon¹⁴, carbon¹³, and carbon¹². *J. Phys. Chem.* **56**, 917-20 (1952); C.A. **47**:2024 g. **IsKi**.
- Schneider, S. Tritium tracer experiment for investigating gas cleanup in hydrogen thyratrons. An abstract. *Phys. Rev.* **84**, 1073 (1951); C.A. **47**:6755 f. **In**.
- S10** Schubert, W. M., Rabinovitch, B. S., Larson, N. R., and Sims, V. A. The stereochemistry of the debromination of vicinal dibromides by metals. *J. Am. Chem. Soc.* **74**, 4590-92 (1952); C.A. **47**:10455 a. **InKi**.
- Schuch, R. L. See Hayes, F. N.
 Schüller, H., Reinebeck, L., and Michel, A. Is there a spectrum of the OH⁻ ion? *Z. Naturforsch.* **9a**, 279-85 (1954); C.A. **48**:9192 e. **SpEl**.
- Schuit, G. C. A. The adsorption of hydrogen and deuterium on nickel-silica catalysts. *Proc. Intern. Symposium Reactivity of Solids, Gothenburg, 1952*, 571-81 (Pub. 1954); C.A. **48**:11869 h. **AdG**.
- Schuler, H. A new spectrum attributed to the C₂H_x molecule (x=1, 2, or 3). *Mem. soc. roy. sci. Liege* **13**, 231-37 (1953); C.A. **47**:8511 i. **EIGd, SpVi**.
- Schull, E. R. See Wiggins, T. A.
 Schwarz, H. A., Williams, R. R., Jr., and Hamill, W. H. **S14** The kinetics of hot hydrogen atoms in the photolysis of the hydrogen halides. *J. Am. Chem. Soc.* **74**, 6007-12 (1952); C.A. **47**:3704 d. **KiP, KiG**.
- Schwarzer, D. See Fireman, E. L.
 Scott, R. B. See Armstrong, G. T.
 Seaborg, G. T. See Hollander, J. M.
 Searles, S., Tamres, M., and Barrow, G. M. Hydrogen-bonding of esters and lactones. Site of bonding and effect of ring size. *J. Am. Chem. Soc.* **75**, 71-73 (1953); C.A. **47**:12243 f. **InSt, SpVi**.
- Secco, E. A. See Giguère, P. A.
 Seitz, F. See Koehler, J. S.
 Selak, P. J., and Finke, J. Heavy water. A review of processes and plants for large-scale production. *Chem. Eng. Progr.* **50**, 221-29 (1954); C.A. **48**:7444 d. **Ge, Se**.
- Selgin, M. See Broida, H. P.
 Seligman, H. Production of radioactive isotopes in nuclear piles. *Angew. Chem.* **66**, 95-99 (1954); C.A. **48**:4985 i. **Se**.
- Semelek, G. P., and Bernstein, R. B. The thermal decomposition of chloroform. I. Products. *J. Am. Chem. Soc.* **76**, 3793-96 (1954); C.A. **48**:13375 a. **EqG, KiG**.
- Setkina, V. N., Kursanov, D. N., Sterlingov, O. D., and Liberman, A. L. Hydrogen exchange of saturated hydrocarbons under influence of sulfuric acid. *Doklady Akad. Nauk S. S. S. R.* **85**, 1045-48 (1952); C.A. **47**:851 f. **EqL, KiI**.
- Setlow, J. See Pollard, E. C.
 Setlow, R. B. The inactivation of urease by deuterons and heat. *Arch. Biochem. Biophys.* **36**, 328-35 (1952); C.A. **47**:644 e. **BiC**.
- S21** Setlow, R. B., and Doyle, B. The molecular weight of insulin and the inactivation of insulin by fast charged particles. *Arch. Biochem. Biophys.* **42**, 83-90 (1953); C.A. **47**:9381 a. **BiC**.
- Sette, D., and Hubbard, J. C. Thermal relaxation of carbon dioxide in presence of water and D₂O molecules. *J. Acoust. Soc. Amer.* **25**, 994-97 (1953); C.A. **48**:2431 e. **MeAc**.

- Sharma, D. See Douglas, A. E.
- Shatenshtein, A. I. Studies of the reactivity of hydrocarbons in solution by the method of isotopic exchange of hydrogen. A review. *Uspekhi Khim.* **21**, 914-39 (1952); C.A. 47:3055b. **Ge, Eq.** See Dykhno, N. M.
- S24 Shatenshtein, A. I., and Izrailevich, E. A. Rate of isotopic exchange reactions of hydrogen in benzene and in naphthalene upon catalysis by a solution of potassium amide in liquid deuteriated ammonia. *Zhur. Fiz. Khim.* **28**, 3-10 (1954); C.A. 48:10413 e. **EqL, KiL.**
- Sheppard, N. See Hadzi, D.
- S25 Sheppard, W. A., Bader, R. F. W., and Bourns, A. N. Sulfur isotope effects in the bisulfite addition reaction of aldehydes and ketones. II. Bond-formation effect. *Can. J. Chem.* **32**, 345-50 (1954); C.A. 48:7542 e. **IsKi.**
- Sheridan, J. See Bond, G. C.
- Sherman, R. H. See Pontarelli, D. A.
- Shiner, V. J. Jr. Substitution and elimination rate studies on some deuterioisopropyl bromides. *J. Am. Chem. Soc.* **74**, 5285-88 (1952); C.A. 47:8007 d. **KiL, Sy.**
- S27 Solvolysis rates of some deuterated tertiary amyl chlorides. *J. Am. Chem. Soc.* **75**, 2925-29 (1953); C.A. 48:7532 fhg. **KiL.** See Johnson, J. R.
- Shipman, J. J. See Craig, D.
- Shull, E. R. See Wiggins, J. A.
- S28 Shull, E. R., Oakwood, T. S., and Rank, D. H. Infrared and Raman spectra of tetramethylmethane- d_{12} . *J. Chem. Phys.* **21**, 2024-29 (1953); C.A. 48:1810 d. **SpVi.**
- Sidhu, S. S. Deuterium effect on hydrogen bond distances in hafnium dihydride. *J. Chem. Phys.* **22**, 1062-63 (1954); C.A. 48:12502 e. **IsCr, StD.**
- Siebert, H. The force constants for the methyl compounds of oxygen, sulfur, and selenium. *Z. anorg. u. allgem. Chem.* **271**, 65-75 (1952); C.A. 47:4668 f. **SpVi, StD.**
- Silverman, L., and Bradshaw, W. Precision determination of deuterium in aqueous solutions by a pycnometer method. *Anal. Chim. Acta* **10**, 68-77 (1954); C.A. 48:5728 e. **AnDn.**
- Silvestri, M. See Cerral, E.
- Simpson, M. E. See Tobias, C. A.
- Sims, V. A. See Schubert, W. M.
- Singer, S. F. Meteorites as cosmic-ray meters. *Phys. Rev.* **90**, 168 (1953); C.A. 47:9176 f. **Nu.**
- Siri, W. See Prentice, T. C.
- Sirvetz, M. H. See Weston, R. E., Jr.
- Sirvetz, M. H., and Weston, R. E., Jr. The structure of phosphine. *J. Chem. Phys.* **21**, 898-902 (1953); C.A. 47:7898 g. **SpM, StD, EID.**
- Skell, P. S. See Hill, D. G.
- Slater, N. B. Gaseous unimolecular reactions: theory of the effects of pressure and of vibrational degeneracy. *Trans. Roy. Soc. (London)* **A246**, 57-80 (1953); C.A. 47:9730 h. **IsKi.**
- Sliwa, A. See Trzebiatowski, W.
- Smaller, B., Matheson, M. S., and Yasaitis, E. L. Paramagnetic resonance in irradiated ice. *Phys. Rev.* **94**, 202 (1954); C.A. 48:7457 a. **ElRe.**
- S28 Smaller, B., Yasaitis, E. L., Avery, E. C., and Hutchison, D. A. Spin-spin interaction in HD nuclear resonance. *Phys. Rev.* **88**, 414-15 (1952); C.A. 47:1486 e. **SpA, NuRe, NuS.**
- Smiley, R. L. See Schayer, R. W.
- S27 Smith, C. L. The inactivation of deoxyribonuclease by electron bombardment, deuteron bombardment, and heat. *Arch. Biochem. Biophys.* **45**, 83-90 (1953); C.A. 47:10584 g. **BIC, KiB, KiR.**
- S38 The breakdown of deoxyribonucleic acid under deuteron and electron bombardment. *Arch. Biochem. Biophys.* **46**, 12-17 (1953); C.A. 48:203 f. **BiC.**
- Smith, G. N., Emerson, R. J., Temple, L. A., and Galbraith, T. W. The oxidation of molecular tritium in mammals. *Arch. Biochem. Biophys.* **46**, 22-31 (1953); C.A. 48:254 d. **BiC.**
- S40 Smith, G. N., and Marshall, R. O. Fixation of molecular tritium by bacterial suspensions. *Arch. Biochem. Biophys.* **39**, 395-405 (1952); C.A. 47:634 e. **BiC.**
- Smith, R. K. See Turkevich, J.
- S41 Smith, W. V. Pressure broadening. A review. *Ann. N. Y. Acad. Sci.* **55**, 891-903 (1952); C.A. 47:2603 e. **SpM.**
- S42 Solbrig, A. W., Jr. Effects of certain three-body nuclear interactions in hydrogen³ and helium³. *Univ. Microfilms (Ann Arbor, Mich.)*, Pub. No. 5517, 67 pp. *Dissertation Abstracts* **13**, 841-42 (1953); C.A. 48:1819 b. **Nu.**
- S43 Effects of certain three-body nuclear interactions in hydrogen³ and helium³. *Phys. Rev.* **95**, 831-36 (1954); C.A. 48:11940 i. **Nu.**
- Sponer, H. See Cooper, C. D.
- Springson, D. B. See Elwyn, D.
- S44 Sprinson, D. B., and Coulon, A. The precursors of sphingosine in brain tissue. *J. Biol. Chem.* **207**, 585-92 (1954); C.A. 48:7734 d. **InBi.**
- Sprinson, D. B. See Weissbach, A.
- Stalinski, B. See Trzebiatowski, W.
- S45 Stambaugh, C. K., and Harris, P. M. Charge distribution and vibration amplitudes in lithium hydride and lithium deuteride. An abstract. *Phys. Rev.* **86**, 651 (1952); C.A. 47:6717 i. **SdCr.**
- S46 Stansbury, E. J., Crawford, M. F., and Welsh, H. L. Determination of rates of change of polarizability from Raman and Rayleigh intensities. *Can. J. Phys.* **31**, 954-61 (1953); C.A. 48:4318 e. **SpVi, EIP.**
- Steacie, E. W. R. See Majury, T. G.; Rebbert, R. E.; Whittle, E.; Wijnen, M. H. J.
- Steese, C. M. See Lord, R. C.
- Stehl, O. See Becker, E. W.
- S47 Stein, R. S., and Sutherland, G. B. B. M. Interaction of methylene deformation frequencies in paraffin crystals. *J. Chem. Phys.* **21**, 370-71 (1953); C.A. 47:9775 e. **SpVi.**
- Stephens, D. V. See Nevin, T. E.
- S48 Stephenson, C. C., Corbella, J. M., and Russell, L. A. Transition temperatures in some dihydrogen and deuterio phosphates and arsenates and their solid solutions. *J. Chem. Phys.* **21**, 1110 (1953); C.A. 47:9092 d. **SdTr.**
- Sterlingov, O. D. See Setkina, V. N.
- Stetten, D., Jr. See Mankin, H.
- Stevenson, D. P. See Wagner, C. D.; Wilson, J. N.
- S49 Stevenson, D. P., Wagner, C. D., Beeck, O., and Otvos, J. W. Hydrogen exchange and isomerization of saturated hydrocarbons with sulfuric acid. *J. Am. Chem. Soc.* **74**, 3269-82 (1952); C.A. 47:6333 i. **EqL, KiL, InKi.**
- Stevenson, J. A. F. See Mankin, H.
- S50 Stevenson, P. C., and Hicks, H. G. Separation techniques used in radiochemistry. *Ann. Rev. Nuclear Sci.* **3**, 221-34 (1953); C.A. 48:1164 h. **Se.**
- S51 Stewart, E. S., and Stewart, J. L. Rotation dispersion in the velocity, attenuation, and reflection of ultrasonic waves in hydrogen and deuterium. *J. Acoust. Soc. Amer.* **24**, 194-98 (1952); C.A. 48:9132 d. **MeAc.**
- S52 Stewart, J. L. See Stewart, E. S.
- Stewart, J. W., and Swenson, C. A. Compression to 10,000 atmospheres of solid hydrogen and deuterium at 4.2° K. *Phys. Rev.* **94**, 1069-70 (1954); C.A. 48:9137 gh. **ThF, MeD.**
- Stivers, E. C. See Yankwich, P. E.
- S53 Stoicheff, B. P. High-resolution Raman spectroscopy of gases. II. Rotational spectra of C₆H₆ and C₆D₆ and the internuclear distances in the benzene molecule. *Can. J. Phys.* **32**, 380-88 (1954); C.A. 48:9195 a. **SpVi, StD.** See Davidson, D. W.

- Strandberg, M. W. P. Centrifugal distortion. *Ann. N. Y. Acad. Sci.* **55**, 808 (1952); C.A. **47**:2555 f. **StD, SpM**.
See Burke, B. F.; Posener, D. W.
- Stranks, D. R., and Harris, G. M. Isotope effect in some reactions of carbon¹⁴-labeled carbonatotetramine cobaltic complex ion. *J. Phys. Chem.* **56**, 906-10 (1952); C.A. **47**:2079 a. **IsKi**.
- S56** Predicted isotopic enrichment effects in some isotopic exchange equilibria involving carbon¹⁴. *J. Am. Chem. Soc.* **75**, 2015-16 (1953); C.A. **47**:12015 b. **Ge, SeCh, IsEq, ThS**.
- Stroup, R. E., and Oetjen, R. A. Ph₂D and PHD₂ rotational lines in the region between 50 and 100 K. *J. Chem. Phys.* **21**, 2092 (1953); C.A. **48**:2477 d. **SpM**.
- Stroup, R. E., Oetjen, R. A., and Bell, E. E. The rotational spectrum of ND₃ between 60 and 200° K. *J. Chem. Phys.* **21**, 2072-73 (1953); C.A. **48**:1808 g. **SpM**.
- S59** The rotational spectra of regular and deuteriated phosphine and arsine in the region between 50 and 200 kc. *J. Opt. Soc. Amer.* **43**, 1096-99 (1953); C.A. **48**:1151 a. **SpM**.
- Stuart, A. V., and Sutherland, G. B. B. M. Effect of hydrogen bonding on the hindered rotation of the hydroxyl group in alcohols. *J. Chem. Phys.* **20**, 1977 (1952); C.A. **47**:4741 f. **IsSp, SpVi**.
- S61** Effects of hydrogen bonding on the deformation vibrations of the hydroxyl group in alcohols. *J. phys. radium* **15**, 321-23 (1954); C.A. **48**:9195 f. **SpVi**.
- Sugawara, T., Masuda, Y., Kanda, T., and Kanda, E. **S62** Nuclear magnetic resonance in solid hydrogen with various ortho-concentrations. *Phys. Rev.* **95**, 1355-56 (1954); C.A. **48**:12550 e. **NuRe, SdNu**.
- Sulima, L. V. See Brodskii, A. I.
- Suma, K. See Yamada, N.
- Sutherland, G. B. B. M. See Ehrlich, G.; Kessler, H. K.; Stein, R. S.; Stuart, A. V.
- Sverdlov, L. M. Relations between moments of inertia and rotation frequencies of isotopic molecules. *Doklady Akad. Nauk S. S. S. R.* **88**, 249-52 (1953); C.A. **48**:1741 a. **IsSp**.
- S64** Vibration spectra of formic acid (monomer) and its deuterium-substituted forms—computation and interpretation. *Doklady Akad. Nauk S. S. S. R.* **91**, 503-05 (1953); C.A. **48**:6832 h. **SpVi**.
- S65** Calculations of vibration spectra of molecules containing a carbonyl group and their comparative characteristic. *Izvest. Akad. Nauk S. S. S. R., Ser. Fiz.* **17**, 567-73 (1953); C.A. **48**:5649 d. **SpVi**.
- Sverdlov, L. M., and Pakhomova, N. L. **S66** Vibrational spectra and computation of the potential energy parameters of deuterioethylenes. *Doklady Akad. Nauk S. S. S. R.* **91**, 51-54 (1953); C.A. **48**:6831 f. **SpVi**.
- Swan, P. Existence of a bound state of hydrogen. *Proc. Phys. Soc. (London)* **66A**, 1066-67 (1953); C.A. **48**:6266 a. **NuS**.
- Swenson, C. A. See Stewart, J. W.
- Swift, M. N. See Bond, V. P.
- T**
- Taketa, S. T. See Bond, V. P.
- Takeuchi, T. See Yamada, N.
- Tamres, M. Aromatic compounds as donor molecules in hydrogen bonding. *J. Am. Chem. Soc.* **74**, 3375-78 (1952); C.A. **47**:8026 i. **InSt, StA, SpVi**.
See Searles, S.
- Tarte, P. Spectroscopic investigation of the esters of nitric acid—infra-red spectrum of methyl nitrite-d. *Bull. soc. chim. Belges.* **62**, 401-10 (1953); C.A. **48**:4317 a. **SpVi, St**.
- Tate, F. A. See Bartlett, P. D.
- Tatum, E. L., and Shemin, D. Mechanism of tryptophan **T3** synthesis in *Neurospora*. *J. Biol. Chem.* **209**, 671-75 (1954); C.A. **48**:12236 a. **BiC**.
- Taylor, A. H., and Crist, R. H. Rate and equilibrium **T4** studies on the thermal reaction of hydrogen and iodine. *J. Am. Chem. Soc.* **63**, 1377-85 (1941); C.A. **35**:4268. **EqG, KiG**.
- Taylor, H. C., Jr. See Hutchinson, D. L.
- Taylor, H. S. See Mikovsky, R. J.
- Taylor, W. J. Formulation of isotopic splitting in terms **T5** of the reciprocal kinetic energy matrix. *J. Chem. Phys.* **21**, 763-64 (1953); C.A. **47**:7309 d. **IsSp**.
- Temple, L. A. See Smith, G. N.
- Thaler, R. M. See Breit, G.
- Thode, H. G. Isotopes. A review. *Ann. Rev. Phys.* **T6** *Chem.* **4**, 95-118 (1953); C.A. **47**:10984 h. **Ge**.
- Thomas, C. H. See Hine, J.
- Thompson, H. W. See Boyd, D. R. J.; Grisenthwaith, R. J.; Overend, J.; Pickworth, J.
- Thompson, R. C. Metabolic turnover with tritium as a **T7** tracer. I. Gross studies on the mouse. *J. Biol. Chem.* **197**, 81-87 (1952); C.A. **47**:189 e. **InBi**.
- T8** Metabolic turnover with tritium as a tracer. II. Gross studies on the rat. *J. Biol. Chem.* **200**, 731-43 (1953); C.A. **47**:8214 a. **InBi**.
- T9** Biological applications of tritium. A review. *Nucleonics* **12**, No. 9, 31-35 (1954); C.A. **48**:12827 e. **Ge, InBi**.
See Long, C. W. de
- Thompson, R. C., and Ballou, J. E. The metabolic **T10** equivalence of deuterium and tritium in animal experimentation. *Arch. Biochem. Biophys.* **42**, 219-20 (1953); C.A. **47**:8877 f. **BiZ**.
- T11** Metabolic turnover with tritium as tracer. III. Comparative studies with tritium and deuterium. *J. Biol. Chem.* **206**, 101-07 (1954); C.A. **48**:3520 e. **BiZ**.
- T12** Studies of metabolic turnover with tritium as a tracer. IV. Metabolically inert lipide and protein fractions from the rat. *J. Biol. Chem.* **208**, 883-88 (1954); C.A. **48**:10187 i. **BiZ**.
- Titani, T. Heavy hydrogen and oxygen. *Ann. Rept. Sci. Works Fac. Sci. Osaka Univ.* **1**, 39-51 (1952); C.A. **48**:4275 f. **AbG, AnDn, EqH**.
- Tobias, C. A. See Bond, V. P.; Odeblad, E.
- Tobias, C. A., Dyke, D. C. van, Simpson, M. E., Angler, **T14** H. O., Huff, R. L., and Koneff, A. A. Irradiation of the pituitary of the rat with high-energy deuterons. *Am. J. Roentgenol., Radium Therapy, Nuclear Med.* **72**, 1-21 (1954); C.A. **48**:11506 d. **BiZ**.
- Tomlinson, R. H. See Dainton, F. S.
- Topley, B. See Wheeler, A.
- Torkington, P. The nonplanar vibrations of some **T15** substituted ethylenes and related compounds. *Proc. Roy. Soc. (London)* **A206**, 17-39 (1951); C.A. **48**:1152 d. **StD, SpVi**.
- Tory, E. M. See Brickman, W. J.
- Townes, C. H. See Bird, G. R.; Dousmanis, G. C.; Miller, S. L.; Moszkowski, S. A.; White, R. L.
- Treanor, C. E. Isotope shift in neutral oxygen. *Phys. Rev.* **95**, 1472-73 (1954); C.A. **48**:13416 b. **IsSp**.
- Trenam, R. S. Nuclear magnetic moment of iron⁵⁷. **T17** *Proc. Phys. Soc. (London)* **A66**, 414-15 (1953); C.A. **47**:8542 g. **SoH**.
See Bleaney, B.
- Trenner, N. R., Arison, B., and Walker, R. W. A general **T18** infrared spectrophotometric technique for the determination of deuterium in organic compounds. *Appl. Spectroscopy* **7**, 166-71 (1953); C.A. **48**:2516 f. **AnSp**.
- Triebwasser, S. See Dayhoff, E. S.
- Triebwasser, S., Dayhoff, E. S., and Lamb, W. E., Jr. **T19** Fine structure of the hydrogen atom. *V. Phys. Rev.* **89**, 98-106 (1953); C.A. **47**:4193 d. **SpA**.
- Trofimova, I. V. See Kagan, M. Y.
- Trotman-Dickenson, A. F., and Steacie, E. W. R. The **T20** reactions of methyl radicals. A review. *J. Phys. and Colloid Chem.* **55**, 908-924 (1951); C.A. **46**:3488 b. **KiG, KiP**.

W

- Trzebiatowski, W., Śliwa, A., and Stalinski, B. Magnetic properties of uranium hydride and deuteride. *Roczniki Chem.* **28**, 12-20 (1954); C.A. **48**:11131 a. **ElMg, ElMm.**
- Trzeciak, M. See Friedman, A. S.
- Tuomikoski, P. The infrared spectrum of pyrrole in solution. *J. phys. radium* **15**, 318-20 (1954); C.A. **48**:9196 h. **SpVi.**
- Turkevich, J. See Bond, G. C.
- Turkevich, J., and Smith, R. K. Catalytic isomerization of 1-butene to 2-butene. *J. Chem. Phys.* **16**, 466-80 (1948); C.A. **42**:5409 h. **EqH, KiH.**

U

- Ubbelohde, A. R. See Gallagher, K. J.; Lewis, F. A.; McCoubrey, J. C.
- Ubisch, H. von. Age determination by means of tritium. *Fysik. Verden, Fra* **15**, 174-77 (1953); C.A. **48**:9209 e. **InA.**
- Uemura, Y., Sakisaka, M., Ono, Y., and Miyashiro, S. Evacuation of the large pumping system. *Bull. Inst. Chem. Research, Kyoto Univ.* **25**, 55-56 (1951); C.A. **47**:4144 c. **MeDf.**
- U3 The pumping speed ration of hydrogen to deuterium in an oil-diffusion pump. *Bull. Inst. Chem. Research, Kyoto Univ.* **27**, 54-55 (1951); C.A. **47**:5182 b. **MeDf.**
- Ussing, H. H. See Koefoed-Johnsen, V.

V

- Val'kov, V. I. See Gross, E. F.
- Van Alten, L. See Alten, L. van.
- Van Dyke, D. C. See Dyke, D. C. van.
- Van Dyken, A. R. See Dyken, A. R. van.
- Van Hoomissen, J. E. See Hoomissen, J. E. van.
- Van Riet, R. See Riet, R. van.
- Varnier, R. E. See Rice, F. O.
- Venkateswarlu, P. See Lord, R. C.
- Vennesland, B. See Fisher, H. F.; Loewus, F. A.
- Vennesland, B., and Westheimer, F. H. Hydrogen transport and steric specificity in reactions catalyzed by pyridine nucleotide dehydrogenases. Symposium mechanism enzyme action, McCollum-Pratt Inst., Johns Hopkins Univ., Contrib. No. 70, 3 (1954) 357-88; C.A. **48**:7071 h. **KiB.**
- Verly, W. G., Rachele, J. R., Vigneaud, V. du, Eidinoff, M. L., and Knoll, J. E. A test of tritium as a labeling device in a biological study. *J. Am. Chem. Soc.* **74**, 5941-43 (1952); C.A. **47**:11283 i. **InBi, BiC.**
- Viallard, R., Corval, M., Dreyfus-Alain, B., Grenon, M., and Hermann, J. Determination of tritium in triated organic compounds. *Chim. anal.* **36**, 102-04 (1954); C.A. **48**:6920 a. **AnC.**
- Viallard, R. See Corval, M.; Grenon, M.
- Vigneaud, V. du. See Verly, W. G.
- Villani, S. See Cerral, E.
- Vinogradov, A. P. Geochemistry of isotopes. *Vestnik Akad. Nauk. S. S. R.* **24**, No. 5, 26-43 (1954); C.A. **48**:13571 c. **AbG.**
- Vinogradova, S. V. See Kursanov, D. N.
- Vlaar, H. T. See Ketelaar, J. A. A.
- Voelz, F. L. See Meister, A. G.; Pontarelli, D. A.
- Volpi, G. G. See Boato, G.
- Voltz, S. E., and Weller, S. Catalytic activity and electrical conductivity of chromic oxide catalysts. *J. Am. Chem. Soc.* **75**, 5227-31 (1953); C.A. **48**:2458 c. **EqH, AdG.**
- V6 Effect of water on the catalytic activity of chromic oxide. *J. Am. Chem. Soc.* **75**, 5231-32 (1953); C.A. **48**:2458 d. **EqH, AdG.**
- Von Dardel, G. F. See Dardel, G. F. v.
- Von Ubisch, H. See Ubisch, H. von.
- Vromen, B. H. See Dostrovsky, I.

- Wagner, C. D. See Stevenson, D. P.; Wilson, J. N.
- Wagner, C. D., Wilson, J. N., Otvos, J. W., and Stevenson, D. P. Hydrogen redistribution during olefin hydrogenation. *J. Chem. Phys.* **20**, 338-89, 1331 (1952); C.A. **47**:2680 g. **EqH, KiH, AnMs.**
- Wagner, E. L. See Nightingale, R. E.
- Wagner, G., Pelz, A., and Higatsberger, M. J. Experimental contribution to the problem of isotopic separation by chemical means. *Monatsh.* **85**, 464-66 (1954); C.A. **48**:12576 b. **SeCh.**
- Wahl, W. A. Separating components of a gas or liquid mixture and enriching and separating isotopes. *Brit. Patent* 698, 941, Oct. 28, 1953; C.A. **48**:3732e. **Se.**
- Waldman, B. See Noyes, J. C.
- Walen, R. The continuous isotopic enrichment by cascades. *Rec. trav. inst., recherches structure matiere (Belgrade)* **1**, 35-67 (1952); C.A. **47**:5824 g. **SeEl.**
- Walker, R. W. See Trenner, N. R.
- Wall, M. C. See Laidler, K. J.; Markham, M. C.
- Wang, J. H., Robinson, C. V., and Edelman, I. S. Self diffusion and structure of liquid water. III. Measurement of the self diffusion of liquid water with hydrogen², hydrogen³, and oxygen¹⁸ as tracers. *J. Am. Chem. Soc.* **75**, 466-70 (1952); C.A. **47**:5198 i. **MeDf, In.**
- Washburn, H. W., Berry, C. E., and Hall, L. G. Measurement of deuterium oxide concentration in water samples by the mass spectrometer. *Anal. Chem.* **25**, 130-34 (1953); C.A. **47**:3755 i. **AnMs.**
- W7 Instrumental problems encountered in mass-spectrometer isotope analysis of water samples. *Natl. Bur. Standards (U. S.) Circ. No.* **522**, 141-49 (1953); C.A. **47**:10342 b. **AnMs.**
- Watson, M. S. See Porter, J. W.
- Watson, W. W. Methods of increasing the efficiency of thermal-diffusion columns. *Z. Physik* **133**, 272-75 (1952); C.A. **47**:4756 h. **SeDf.**
- Watts, R. J. See Hiebert, R. D.
- Weeks, B. M. See Garrison, W. M.
- Weigl, J. W. Infrared spectrum of deuteriated ascorbic acid—note on labile hydrogens. *Anal. Chem.* **24**, 1483-86 (1952); C.A. **47**:44 c. **SpVi.**
- Weigl, J. W., and Livingston, R. An attempt to detect hydrogen-deuterium exchange between chlorophyll and water. *J. Am. Chem. Soc.* **74**, 4160-62 (1952); C.A. **47**:3410 h. **EqL, InKi.**
- W11 Hydrogen transfer in a reaction sensitized by chlorophyll. *J. Am. Chem. Soc.* **74**, 4211-12 (1952); C.A. **47**:9738 d. **InKi.**
- Weinberger, A. J. See Ropp, G. A.
- Weinberger, D., and Porter, J. W. Incorporation of tritium oxide into growing *Chlorella pyrenoidosa* cells. *Science* **117**, 636-38 (1953); C.A. **47**:8191 c. **BiB, Is.**
- W13 Metabolism of hydrogen isotopes by rapidly growing *Chlorella pyrenoidosa* cells. *Arch. Biochem. Biophys.* **50**, 160-68 (1954); C.A. **48**:8870 d. **BiC.**
- Weisbaum, S. See Beers, Y.
- Weiss, J. See Kelly, P.
- Weissbach, A., and Sprinson, D. B. The metabolism of 2-carbon compounds related to glycine. II. Ethanolamine. *J. Biol. Chem.* **203**, 1031-37 (1953); C.A. **47**:11437 a. **InBi, Sy.**
- Welch, G. P. See Bond, V. P.
- Welge, K. H. See Beckey, H. D.
- Weller, S. See Voltz, S. E.
- Welsh, H. L. See Stansbury, E. J.
- Welt, I. D., and Wilhelmi, A. E. The effect of adrenalectomy and of the adrenocorticotropic (ACTH) and growth hormones on the synthesis of fatty acids. *Yale J. Biol. Med.* **23**, 99-111 (1950); C.A. **47**:12585 e. **InBi.**
- Wert, C. See Marx, J. W.
- Westheimer, F. H. See Cohen, M.; Fisher, H. F.; Leo, A.; Loewus, F. A.; Vennesland, B.

- Weston, R. E., Jr. Vibrational energy level splitting and optical isomerism in pyramidal molecules of the type XY_3 . *J. Am. Chem. Soc.* **76**, 2645-48 (1954); C.A. **48**:11191 d. **SpVi, EI**Ro.
See Sirvetz, M. H.
- Weston, R. E., Jr., and Bigeleisen, J. Equilibrium in the exchange of hydrogen between phosphine and water. *J. Chem. Phys.* **20**, 1400-02 (1952); C.A. **47**:9728 e. **EqH, EqG**.
- W18 Kinetics of the exchange of hydrogen between phosphine and water: a kinetic estimate of the acid and base strengths of phosphine. *J. Am. Chem. Soc.* **76**, 3074-78 (1954); C.A. **48**:10411 ba. **EqH, KiH**.
- Weston, R. E., Jr., and Sirvetz, M. H. Vibrational frequencies of phosphine-d and phosphine-d₂. *J. Chem. Phys.* **20**, 1820-21 (1952); C.A. **47**:2599 g. **SpVi, EqH**.
- Wexler, S., and Davies, T. H. Average electric charge of daughter atoms from β -decay and isomeric transition. *Phys. Rev.* **88**, 1203-04 (1952); C.A. **47**:3149 i. **NuB**.
- Wheeler, A., Topley, B., and Eyring, H. The absolute rates of reaction of hydrogen with the halogens. *J. Chem. Phys.* **4**, 178-87 (1936); C.A. **30**:2830. **IsKi, KiG**.
- Whiffen, D. H. See Bond, G. C.
- White, R. L. Quadrupole coupling of the deuteron in DC₂CCl and DCN. *Phys. Rev.* **91**, 1014 (1953); C.A. **47**:11954 e. **SpM**.
- White, R. L., and Townes, C. H. Spin of silicon²⁹ and the mass ratios of the stable silicon isotopes. *Phys. Rev.* **92**, 1256-57 (1953); C.A. **48**:2492 b. **SpM**.
- Whittle, E., and Steacie, E. W. R. The reactions of methyl radicals with the hydrogen isotopes. *J. Chem. Phys.* **21**, 993-99 (1953); C.A. **47**:9122 a. **KiP, KiG, IsKi**.
- Wiberg, K. B. The thermal decomposition of deuteriated barium butyrate. *J. Am. Chem. Soc.* **74**, 4381-82 (1952); C.A. **47**:2024 e. **EqL, KiL, Sy**.
- Wiberley, S. E. See Quinan, J. R.
- Wibenga, E. H. Crystal structure of cyanuric acid. *J. Am. Chem. Soc.* **74**, 6156-57 (1952); C.A. **47**:11964 d. **SdCr, SpVi**.
- Wiener, H., and Burton, M. Decomposition of methane in an electrical discharge. *J. Am. Chem. Soc.* **75**, 5815-23 (1953); C.A. **48**:3148 a. **KiR**.
- Wiggins, T. A., Shull, E. R., Bennett, J. M., and Rank, D. H. K fine structure in the 2ν band of CHD₃. *J. Chem. Phys.* **21**, 1940-42 (1953); C.A. **48**:1809 g. **SpVi**.
- Wiggins, T. A., Shull, E. R., and Rank, D. H. Rotation vibration bands of some symmetric-top molecules under high resolution. *J. Chem. Phys.* **21**, 1368-73 (1953); C.A. **47**:11969 i. **SpVi**.
- Wijnen, M. H. J., and Steacie, E. W. R. Photolysis of 2,2',4,4'-tetra deuteriodiethyl ketone. *Can. J. Chem.* **29**, 1092-1103 (1951); C.A. **47**:5257 e. **KiP, KiG**.
- Wijnen, M. H. J. The reaction of methyl radicals with aceto- and propionitrile. *J. Chem. Phys.* **22**, 1074-76 (1954); C.A. **48**:12560 g. **KiG**.
- Wilhelmi, A. E. See Welt, I. D.
- Williams, E. D., Krieger, K. A., and Day, A. R. The mechanism of the Meerwein-Ponndorf-Verley reaction. A deuterium tracer study. *J. Am. Chem. Soc.* **75**, 2404-07 (1953); C.A. **48**:6382 b. **InKi**.
- Williams, G. See Bonner, T. G.
- Williams, Q., Cox, J. T., and Gordy, W. Molecular structure of bromoform. *J. Chem. Phys.* **20**, 1524-25 (1952); C.A. **47**:4745 f. **StD, SpM**.
- Williams, R. R., Jr. See Schwarz, H. A.
- Williamson, C. W. Diffusion-distillation process for the separation of isotopes. U. S. At. Energy Comm. Tech. Inform. Service, Oak Ridge, Tenn. **SO-3252**, 5-33 (1953); C.A. **48**:2491 f. **SeDf, SeDs**.
- Williamson, L. See Mohler, F. L.
- Wilmarth, W. K., and Barsh, M. K. Conversion of W35 ortho-para hydrogen by cuprous acetate in quinoline. *J. Am. Chem. Soc.* **75**, 2237-42 (1953); C.A. **47**:9124 i. **EqH**.
- Wilmarth, W. K., and Dayton, J. C. Mechanism of the exchange of hydrogen gas with solutions of potassium amide in liquid ammonia. *J. Am. Chem. Soc.* **75**, 4553-56 (1953); C.A. **48**:3119 e. **EqH, KiH**.
- Wilmarth, W. K., Dayton, J. C., and Flournoy, J. M. Mechanism of exchange of hydrogen gas and aqueous alkali. *J. Am. Chem. Soc.* **75**, 4549-53 (1953); C.A. **48**:1780 i. **EqH, KiH**.
- Wilson, J. N. See Wagner, C. D.
- Wilson, J. N., Otvos, J. W., Stevenson, D. P., and Wagner, C. D. Hydrogenation of olefins over metals. *Ind. Eng. Chem.* **45**, 1480-87 (1953); C.A. **48**:6951 c. **KiH**.
- Wilson, M. K. See Polo, S. R.
- Wilzbach, K. E. See Brown, W. G.; Kaplan, L.
- Wilzbach, K. E., Dyken, A. R. van, and Kaplan, L. Determination of tritium by ion current measurement. *Anal. Chem.* **26**, 880-83 (1954); C.A. **48**:11205 i. **AnC**.
- Wilzbach, K. E., Kaplan, L., and Brown, W. G. Preparation of gas for the assay of tritium in organic compounds. *Science* **118**, 522-23 (1953); C.A. **48**:3156 h. **AnC**.
- Wirtz, K. See Gierer, A.
- Witzig, W. F. Creep of copper under deuteron bombardment. *J. Appl. Phys.* **23**, 1263-66 (1952); C.A. **47**:1504 c. **Ge**.
- Woese, C., and Pollard, E. C. The effect of ionizing radiation on various properties of Newcastle disease virus. *Arch. Biochem. Biophys.* **50**, 354-67 (1954); C.A. **48**:10085 f. **BiC**.
- Wolfsberg, M. See Bigeleisen, J.
- Woodward, I. See Gallagher, K. J.
- X**
- Xuong, Ng D. See Lacassagne, A.
- Y**
- Yamada, N., Suma, K., and Takeuchi, T. Exchange reaction of acetate groups between acetic acid and zinc acetate. *J. Chem. Soc. Japan, Pure Chem. Sect.* **74**, 1018-20 (1953); C.A. **48**:9794 d. **EqL, KiL**.
- Yamamuro, N. Isotope separation by radiofrequency electric field. *J. Phys. Soc. Japan* **6**, 411-13 (1951); C.A. **47**:5188 g. **Se**.
- Yankwich, P. E. Isotope effects in chemical reactions. *Rev. Nuclear Sci.* **3**, 235-48 (1953); C.A. **48**:1164 h. **IsKi**.
- Yankwich, P. E., and Belford, R. L. Intermolecular carbon isotope effect in the decarboxylation of normal malonic acid in quinoline solution. *J. Am. Chem. Soc.* **75**, 4178-82 (1953); C.A. **47**:11923 f. **IsKi**.
- Yankwich, P. E., Belford, R. L., and Fraenkel, G. Temperature coefficient of the intermolecular carbon isotope effect in the decarboxylation of normal malonic acid. *J. Am. Chem. Soc.* **75**, 832-35 (1953); C.A. **47**:5816 d. **IsKi**.
- Yankwich, P. E., and Stivers, E. C. Intramolecular carbon isotope effects in the decarboxylation of malonic acid and bromomalonic acid. *J. Chem. Phys.* **21**, 61-66 (1953); C.A. **47**:10975 d. **IsKi**.
- Yasaitis, E. L. See Smaller, B.
- Yoshimori, A. See Matsubara, T.
- Z**
- Zajdela, F. See Lacassagne, A.
- Zaleski, P. See Ertaud, A.
- Zaretskaya, I. I. See Kursanov, D. N.; Nazarov, I. N.
- Zehle, F. See Braune, H.
- Zeiger, H. J. See Dousmanis, G. C.
- Zelikoff, M. See Billings, B. H.
- Zeltmann, A. H., and Fitzgibbon, G. C. Physical properties of the deuteriogermanes. *J. Am. Chem. Soc.* **76**, 2021-22 (1954); C.A. **48**:8144. **ThI ThII Sy**.
- Zyryanova, L. N. See Dzhelepor, B. S.

4. Subject Index

This index is divided into a number of principal and subsidiary subject headings to which are assigned appropriate letter-code symbols. The letter-number symbols following each heading refer to references in the bibliography.

- Ab** **Abundance:** F6.
AbG (geological): B55, C40, F7, F13, G26, J3, J9, J10, K7, L31, N9, O1, S3, S12, T13, V4, V5, V6.
AbO (organic): H16.
- Ad** **Adsorption and sorption:** B73.
AdG (gases on solids): S4.
- An** **Analytical methods:** A8, F14.
AnC (counters, cloud chambers, electrometers, ionization chambers, and photographic emulsions): B21, F1, F2, F7, G10, G21, H3, H12, H13, H20, H23, J5, J9, L2, L8, P4, R31, R32, V3, W39, W40.
AnDn (density methods): C32, S31, T13.
AnMs (mass spectrograph and mass spectrometer): B43, B58, C15, C26, D21, D30, D34, F13, G13, G27, H16, M55, W1, W6, W7.
AnRf (refractive index): I1.
AnSp (absorption spectra): B79, B80, D30, K30, L15, M56, T18.
AnTh (thermal conduction): H5.
- Bi** **Biological effects of deuterium and tritium compounds and of deuterons and tritons.**
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BiC (biochemical): B4, B28, B48, C3, C19, D11, E1, E7, E9, E10, F11, H33, H38, H39, K33, K39, M12, M29, O2, P9, P17, R3, S20, S21, S37, S38, S39, S40, T3, V2, W13, W42.
BiZ (zoological): B65, C16, G4, H37, K31, L29, O2, P22, P29, T10, T11, T12, T14.
- Ec** **Electrochemical properties.**
EcC (conductivities and mobilities): G6.
EcO (overtoltage): B86.
EcP (electrode potentials): B57, H36.
- El** **Electromagnetic and optical properties (except spectra):**
EID (dielectric constants and dipole moments): A14, B14, H10, H34, M31, M32, M33, M34, M60, P11, P26, S33.
ElGd (gas discharges): B78, G9, H21, L13, R5, S13.
ElMg (magnetic susceptibilities and Curie constants): B9, T21.
ElMm (magnetic moments): J7, T21.
ElMr (magnetic rotatory power): I2.
EIP (polarization): D1, R9, S46.
ElRe (paramagnetic resonances): S35.
ElRf (refractive index and molar refraction): C31, D12, E13.
ElRo (optical rotatory power): F4, W16.
ElT (relaxation times): B17, P26.
- Eq** **Chemical equilibria:** B74, H19, K1, K5, S23.
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EqL (liquid and solution): A4, B60, B61, B62, B76, C41, C44, D37, D38, E8, F8, G12, G24, H26, H27, H28, H29, H40, J12, K18, K23, K42, K48, L21, L34, M47, O5, R22, S19, S24, S49, W10, W25, Y1.
- Ge** **General and review:** A11, A12, B36, B40, B41, B74, B85, B87, C1, C21, C43, D33, E4, E5, G7, H19, H20, H22, H32, I3, L14, L27, M27, M28, M36, N2, N4, P1, R10, R25, R28, S5, S16, S23, S56, T6, T9, W41.
- In** **Indicator and tracer techniques:** B73, G20, H22, P13, S9, W5.
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InBi (biological): B4, B5, B27, B28, B47, C16, D17, E9, F3, F8, F17, F18, H38, K35, L28, M17, M18, P5, P28, S6, S44, T7, T8, T9, V2, W14, W15.
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InSt (structure determinations): B13, B76, C2, C36, J12, L20, L21, P27, S15, T1.
- Is** **Isotope effects:** M49, P21, R34, W12.
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IsSp (spectra): G15, I8, J16, K17, L33, M8, M57, N5, P12, S60, S63, T5, T16.
IsTh (thermodynamic properties): B39, B54, L18, N19.
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KiL (liquid and solution): A4, B12, B25, B60, B61, B62, B66, B76, C20, D3, D37, D38, G12, H25, H27, H28, H29, K4, L10, L22, L25, M47, O5, P31, R21, R22, S19, S24, S26, S27, S49, W25, Y1.
KiP (photochemical): B83, D5, K25, K41, M11, M15, M16, N4, R11, S14, T20, W24, W30.
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	MeDf (diffusion): A11, A12, B70, C42, J6, N15, N17, U2, U3, W5.		SpA (atomic-line): B79, C4, C40, D14, E4, P33, S1, S36, T19.
	MeSt (surface tension): H15.		SpEl (molecular electronic): B24, B77, B78, C27, C36, D24, F9, G15, H21, I4, I5, I6, K27, L5, L12, L13, N3, N18, R6, S11.
	MeV (viscosity): A11, A12, B15, B16, D19, H15, M46, P21, R34.		SpFl (fluorescence): B34, J19.
Nu	Nuclear properties: S32, S42, S43.		SpM (microwave): A2, B1, B2, B3, B20, B50, B51, B91, B92, B93, C38, C39, D2, D31, E4, G16, G17, H34, J7, J8, K26, M34, M49, M60, N21, N22, P24, P25, R24, R35, S33, S41, S54, S57, S58, S59, W22, W23, W33.
	NuB (beta ray spectra): B52, C43, H6, H7, H13, J1, J19, L7, M50, W20.		SpVi (vibrational—including Raman): A4, A5, A6, B7, B11, B22, B23, B29, B30, B33, B67, B68, B69, B78, C2, C11, C12, C13, C22, C29, C33, C34, C36, C37, D9, D10, D15, D18, D19, D24, D29, D32, D36, E3, F22, F23, G2, G3, G11, G18, G23, G25, H1, H2, H8, H11, H14, H17, H35, I6, J14, J15, J16, J17, K2, K13, K29, K36, K43, K44, K45, L9, L14, L20, L21, L33, L35, L36, L37, M19, M20, M21, M22, M25, M30, M44, M54, N3, N7, N11, O6, O9, P10, P12, P13, P18, P20, P25, Q1, R13, R16, R17, R18, R19, R20, R22, S13, S15, S28, S30, S46, S47, S53, S60, S61, S64, S65, S66, T1, T2, T15, T22, W9, W16, W19, W26, W28, W29.
	NuH (hyperfine structure): E4, P33.		
	NuIn (interactions—absorption of radiation, ranges, and scattering): C28, D4, D25, E12, K34, M4, O10, O11.		Sr
	NuM (masses and binding energy): D39, E14, M38, O3, O4, P5.		Mass spectroscopy and mass spectrography: D23, E14, G14, J13, L38, M37, M38, M51, M53, M55, N16, O3, O4, S5.
	NuMg (magnetic moments): B8, B9, B71, B72, C21, C24, J7, P33, R7, R33.		St
	NuQ (quadrupole moments): B9, B53, C21, D16, D35, F10, M58.		Molecular structure: D2, G7, J16, K45, L9, M19, M25, M34, N7, T2.
	NuR (reactions): N20.		StA (molecular association): L33, T1.
	NuRe (magnetic resonances): A9, A10, B8, B9, B71, C4, C21, H9, M42, P33, R12, S36, S62.		StD (molecular constants—interatomic distances, bond angles, moments of inertia, force constants, and potential functions): A6, B1, B2, B3, B26, B67, C13, C37, D29, D31, G2, G3, H14, H34, I5, K13, K36, L12, L36, M21, M49, M54, M57, M60, N6, N14, O9, P12, P20, S29, S30, S33, S53, S54, T15, W33.
	NuS (spins, states, and wave functions): C4, C21, F10, J4, M1, M58, P6, R8, R9, S2, S36, S67.		Sy
Sd	Solid state: A9, A10, K34, N19.		Synthesis and preparation of compounds: B3, B6, B28, B30, B31, B34, B35, B83, B88, B89, B95, C31, C36, E8, F19, F20, F21, G9, G18, G24, H35, K20, K32, K42, L1, L6, L17, L18, L19, L22, L25, L32, L34, M33, N10, N12, N13, P11, P18, R27, S26, W14, W25, Z1.
	SdCr (crystal structure—including electron, neutron, and x-ray diffraction): B14, G1, L23, L24, M31, M32, M35, M43, S45, W26.		Th
	SdEl (electromagnetic properties): A14, P11, R12.		Thermodynamic and related properties: B59.
	SdNu (nuclear properties): S62.		ThD (diffusion and heat conduction): A11, A12, B59, C5, G22, G27, J6, M45, P32.
	SdSp (spectra): D15, G11, H2, H11, J19, L33, P10.		ThF (thermodynamic functions for pure substances and reactions between them—E, H, S, C_v, C_p, F, K, ΔH, ΔS, ΔE, ΔC_p, ΔF, data of state, and thermal expansion): B39, B54, B59, D8, D32, F12, H35, K29, L3, L35, M3, N2, N14, O8, P18, P20, S52, Z1.
	SdTr (transitions—including phase transitions): B14, B59, K40, L3, M31, M32, M33, P11, S48.		ThP (phase equilibria—melting points, triple points, boiling points, heats of transition, critical constants, and vapor pressures): A13, B59, C14, C31, C35, D13, F12, G8, J18, L18, M59, N19, Z1.
Se	Isotope separation: D7, D28, H18, H30, I3, K11, S16, S17, S50, W3, Y2.		ThS (statistical mechanics and statistical thermodynamics): B39, B59, O8, S56.
	SeAd (adsorption—including chromatography and ion exchange): A3.		ThSo (properties of solutions—activities, fugacities, pH, vapor pressures, heats of solution and dilution, and colligative properties): H36.
	SeCh (chemical reaction): B10, B32, B49, B84, C7, C18, E15, K24, N19, P8, S56, W2.		
	SeCr (crystallization): R26.		
	SeDf (diffusion—including thermal diffusion): B18, C7, G27, N8, P32, W8, W34.		
	SeDs (distillation): D27, E2, M5, M6, M7, W34.		
	SeEl (electrolysis): B90, H31, K28, O7, W4.		
	SeEm (electromagnetic methods): B19.		
	SeMs (mass spectrometer and mass spectrograph): K12, L16.		
So	Solubility.		
	SoG (gases in solids): L26.		
	SoH (in H₂O, HDO, and D₂O): C35, T17.		

5. Compound Index

This index will be useful in locating references dealing with a particular deuterium or tritium compound. Tritium compounds are designated by a "t" following the compound name. Each entry in the index contains one or more letter codes and a series of letter-number symbols indicating, respectively, the subject content and the location of the references in the bibliography. A description of the coding system may be found in the Introduction.

Acetaldehyde (C₂H₄O) **KiB**:V1; **KiP**:N4.

Acetic acid (C₂H₄O₂) **EqG**:E6, F19; **EqH**:E6, F19, F20; **EqL**:R22, Y1; **IsKi**:E6, R21; **KiH**:E6; **KiL**:R21, R22, Y1; **SdSp**:H2; **SpVi**:F22, H2, S65.

Acetic acid, cuprous salt of (C₂H₃O₂Cu) **EqH**:W35.

Acetic acid, methyl ester of (C₃H₆O₂) **EIRf**:C31; **KiR**:C30; **ThP**:C31; **Sy**:C31.

Acetic acid, zinc salt of (C₄H₆O₄Zn) **EqL**:Y1; **KiL**:Y1.

Acetic acid-t (C₂H₄O₂-t) **EqG**:E6; **EqH**:E6; **IsKi**:E6; **KiH**:E6.

Acetoacetic acid, methyl ester of (C₅H₈O₃) **EqL**:J12; **InSt**:J12.

Acetone (C₃H₆O) **EqH**:K6; **IsKi**:M15, W:24; **KiG**:M7, M10, M15, T20, W24; **KiH**:K6; **KiP**:M11, M15, M16, N4, T20, W24.

Acetone dicarboxylic acid (C₅H₆O₅) **EqL**:G24; **Sy**:G24.

Acetylene (C₂H₂) **AnMs**:D30; **AnSp**:D30; **BiC**:C3; **ElGd**:S13; **EqH**:B63, D30; **IsMs**:M51; **KiH**:B63, D30, R2; **KiR**:G14; **SpEl**:I4, I5, I6, N18; **SpVi**:A5, I6, M30, R13, T15; **Sr**:G14, M51; **StD**:I5, T15; **Sy**:L18.

Acetylene dibromide (C₂H₂Br₂) **SpVi**:M22.

Acetylenedicarboxylic acid (C₄H₂O₄) **SdSp**:H2; **SpVi**:H2.

Acetylene dicarboxylic acid dihydrate (C₄H₂O₄·2H₂O) **IsCr**:G1; **SdCr**:G1.

Acetylene dichloride (C₂H₂Cl₂) **SpVi**:B29.

Acetylglycine (C₄H₇NO₃) **IsSp**:L33; **SdSp**:L33; **SpVi**:L33; **StA**:L33.

Acids **KiH**:B85s; **SoI**:B85s.

Adenine-t (C₅H₅N₅-t) **EqL**:E8; **Sy**:E8.

Adipic acid (C₈H₁₀O₄) **SdSp**:H2; **SpVi**:H2.

Alanine (C₃H₇NO₂) **AnMs**:B58.

Allene (C₃H₄) **SpVi**:L32, O9; **StD**:L36, O9.

Allyl acetate (C₅H₈O₂) **KiL**:B12.

Allylene (C₃H₄) **KiH**:R2; **SpVi**:G23; **Sy**:L17.

Allyl radical (CH₂CHCH₂) **Sr**:L38.

Aluminum hydride (AlH) **SpEl**:K27.

Aluminum isopropionate ((C₃H₇O)₃Al) **InKi**:W32.

Amidogen (NH₂) **ElGd**:L13; **SpEl**:L13.

Aminoacetic acid. See Glycine.

Amino ethanol (C₂H₇NO) **InBi**:W14; **Sy**:W14.

Ammonia (NH₃) **ElGd**:H21, L13; **EIRo**:W16; **EqG**:B38, P7, **EqH**:E11, G28, K3, W36; **EqL**:D37, D38, S24; **KiH**:E11, G28, K3, W36; **KiL**:D37, D38, S24; **SpEl**:F9, H21, L13; **SpVi**:W16; **SpM**:B51, N21, N22, R35, S41, S54, S58; **StD**:S54.

Ammonium arsenate, dihydrogen ((NH₄)₂AsO₄) **SdTr**:S48.

Ammonium bromide (NH₄Br) **SdCr**:L23, L24.

Ammonium chloride (NH₄Cl) **StD**:B26.

Ammonium iodide (NH₄I) **SdCr**:L24.

Ammonium ion (NH₄⁺) **SpVi**:G2; **StD**:G2.

Ammonium manganous sulfate ((NH₄)₂Mn(SO₄)₂) **EIP**:D1.

Ammonium phosphate, dihydrogen ((NH₄)₂H₂PO₄) **EID**:M31, M32, M33; **SdCr**:M31, M32, M33, M43; **SdTr**:M31, M32, M33, S48; **Sy**:M33.

Amyl alcohol (C₅H₁₁OH) **EqL**:H29; **KiL**:H29; **SpVi**:Q1.

Amyl benzene (C₁₁H₁₆) **InKi**:C8.

Amyl chloride (C₅H₁₁Cl) **KiL**:S27.

Anethole hydrochloride-t (C₁₀H₂₀O·HCl-t) **Sy**:L1.

Anthracene (C₁₂H₁₀) **EqL**:G12; **KiL**:G12.

Arsine (AsH₃) **SpM**:S59; **StD**:N6.

Ascorbic acid (C₆H₈O₆) **InKi**:W11; **SpVi**:W9.

Asparagine (C₄H₈N₂O₃) **SpVi**:D10.

Aspartic acid (C₄H₇NO₄) **AnMs**:B58.

Benzene (C₆H₆) **EIRf**:D12; **IsKi**:L10; **IsMs**:M51; **KiL**:L10; **KiR**:B94, G14, P3; **MeV**:D19; **NuRe**:A9, A10; **Sd**:A9; **SpVi**:D19, G18, K2, K36, S53, T15; **Sr**:G14, M51; **StD**:K36, S53, T15; **Sy**:G18; **ThP**:D13.

Benzene hexachloride (C₆H₆Cl₆) **EqL**:C41.

Benzoic acid (C₇H₆O₂) **SdSp**:H2; **SpVi**:H2.

Benzoquinhydrone (C₁₂H₁₀O₄) **InKi**:G19.

Benzyl radical (C₆H₅CH₂) **Sr**:L38.

Bistrideuteromethyl mercury (CH₃)₂Hg **Sy**:B35.

Borine carbonyl (BH₃CO) **Sy**:B89.

Borine dimethylamine (C₂H₈NB) **Sy**:B89.

Borine methyl ether (C₂H₆OBH₃) **Sy**:B89.

Boron hydride (B₂H₆) **EqG**:M41; **EqH**:M41; **KiG**:M41; **KiH**:M41; **Sy**:B89.

Bromobenzene-t (C₆H₅Br-t) **IsKi**:B25; **KiL**:B25.

Bromoethylene (C₂H₃Br) **SpVi**:T15; **StD**:T15.

Bromoform (CHBr₃) **SpM**:W33; **StD**:W33.

Bromotoluene (C₇H₇Br) **InKi**:K41; **KiP**:K41.

Butane (C₄H₁₀) **KiL**:B83; **KiP**:D5; **Sy**:B83.

Butene. See Butylene.

Butyl alcohol (C₄H₉OH) **EqL**:H29; **KiL**:H29; **SpVi**:Q1.

Butylene (C₄H₈) **AnMs**:W1; **EqH**:W1; **KiH**:W1, W38; **KiP**:D5.

Butyric acid (C₄H₈O₂) **EqL**:W25.

Butyric acid, barium salt of (C₈H₁₄O₄Ba) **EqL**:W25; **KiL**:W25; **Sy**:W25.

Butyric acid, methyl ester of (C₅H₁₀O₂) **EqL**:W25.

Calcium hydride (CaH₂) **KiS**:H24, K8.

Calcium hydroxide (Ca(OH)₂) **EcP**:H36s; **ThP**:H36.

Calcium sulfate (CaSO₄· $\frac{1}{2}$ H₂O) (CaSO₄·2H₂O) **SpVi**:D36.

Cellulose ((C₆H₁₀O₅)_x) **EqL**:A4; **KiL**:A4; **SpVi**:A4, M25; **St**:M25.

Cesium arsenate, dihydrogen (CsH_2AsO_4) **SdTr**:S48.
 Chloramine (NH_2Cl) **SpVi**:M54; **StD**:M54.
 Chloroacetamide ($\text{CH}_2\text{ClCONH}_2$) **SpVi**:K45; **St**:K45.
 Chloroacetic acid (CH_2ClCOOH) **SpM**:A2.
 Chloroacetylene (C_2HCl) **SpM**:W22.
 Chlorobenzene ($\text{C}_6\text{H}_5\text{Cl}$) **NuRe**:M42.
 Chlorobutane ($\text{C}_4\text{H}_7\text{ClO}$) **KiL**:P31.
 Chlorodibromo methane (CHClBr_2) **SpVi**:M44, P20; **StD**:P20; **ThF**:P20.
 Chloroform (CHCl_3) **EqG**:S18; **EqL**:H27; **IsKi**:E15; **IsSp**:G15; **Ki**:E15; **KiL**:H27; **SpEl**:G15; **SpVi**:B33; **KiG**:S18.
 Chlorosilane (SiH_3Cl) **SpM**:B2, B3; **SpVi**:A6; **StD**:A6, B2, B3; **Sy**:B3.
 Cholesterol ($\text{C}_{27}\text{H}_{48}\text{O}$) **InKi**:D6.
 Δ^7 -Cholestenol-t **BiC**:B48.
 Cholesterol ($\text{C}_{27}\text{H}_{46}\text{O}$) **EqH**:F20; **Sy**:B88, F20, L6.
 Cholesterol-t ($\text{C}_{27}\text{H}_{46}\text{O-t}$) **BiC**:B48; **InBi**:B47.
 CH radical **KiG**:B77; **SpEl**:B77.
 C_2H radical **ElGd**:S13; **SpVi**:S13.
 C_2H_3 radical **ElGd**:S13; **SpVi**:S13.
 Chromic oxide **AdG**:V5, V6; **EqH**:V5, V6.
 Cortisone-t **Sy**:F21.
 Cyanuric acid ($(\text{HCNO})_3$) **SdCr**:W26; **SpVi**:W26.
 Cyclobutane (C_4H_8) **Sy**:L32.
 Cyclobutene (C_4H_6) **Sy**:L32.
 Cyclohexane (C_6H_{12}) **ElRf**:D12; **KiR**:B94; **MeV**:D19; **SpVi**:D19, L9; **St**:L9; **ThP**:D13.
 Cyclohexane carboxylic acid. See Hexahydrobenzoic acid.
 Cyclohexanone ($\text{C}_6\text{H}_{10}\text{O}$) **InKi**:W32.
 Cycloöctatetraene (C_8H_8) **Sy**:L32.
 Cycloöctatriene (C_8H_{10}) **Sy**:L32.
 Cytosine-t ($\text{C}_4\text{H}_5\text{N}_3\text{O-t}$) **EqL**:E8; **Sy**:E8.
 Decane ($\text{C}_{10}\text{H}_{22}$) **BiC**:B28; **InBi**:B27, B28; **Sy**:B28.
 Decyl alcohol ($\text{C}_{10}\text{H}_{21}\text{OH}$) **SpVi**:Q1.
 Diborane. See Boron hydride.
 Diborane dimethylamine ($\text{C}_2\text{H}_6\text{NB}_2\text{H}_5$) **Sy**:B89.
 Dibromo chloro methane. See Chloro dibromo methane.
 Dibromoethane ($\text{C}_2\text{H}_4\text{Br}_2$) **InKi**:S10; **SpVi**:B30; **Sy**:B30, L18.
 Dibromoethylene. See Acetylene dibromide.
 Dibromomethane. See Methylene bromide.
 Dibutyl isopropyl carbinol ($\text{C}_{12}\text{H}_{26}\text{O}$) **EqL**:H29; **KiL**:H29.
 Dichloroethylene. See Acetylene dichloride.
 Dichloropropene. See Dichloropropylene.
 Dichloropropylene ($\text{C}_3\text{Cl}_2\text{H}_1$) **Sy**:L17.
 Diethyl amine ($(\text{C}_2\text{H}_5)_2\text{NH}$) **EqL**:K23.
 Diethyl ketone ($(\text{C}_2\text{H}_5)_2\text{CO}$) **KiG**:W30; **KiP**:W30; **SpVi**:N11; **Sy**:L19.
 Digermane (Ge_2H_6) **ThF**:Z1; **ThP**:Z1; **Sy**:Z1.
 Dihydroxyanthraquinone ($\text{C}_{14}\text{H}_8\text{O}_4$) **SpVi**:H1.
 Diisopropyl chromate ($\text{C}_6\text{H}_{14}\text{CrO}_4$) **KiL**:L22; **Sy**:L22.
 Diphenyloxazole-t ($\text{C}_{15}\text{H}_{11}\text{NO-t}$) **AnC**:H13.
 Diphosphopyridine nucleotide **InBi**:F8, L28; **KiB**:F8, L28.
 Dodecane ($\text{C}_{12}\text{H}_{26}$) **BiC**:B28; **InBi**:B27, B28; **Sy**:B28.
 Estrone acetate **InBi**:P5.
 Ethane (C_2H_6) **EqH**:A7; **KiG**:R14, R15; **KiH**:A7, K9; **KiP**:D5; **SpVi**:R16, R17, R18, R19, R20.
 Ethanolamine. See Amino ethanol.
 Ethyl acetate ($\text{C}_4\text{H}_8\text{O}_2$) **Sy**:N10.
 Ethylacetoacetate ($\text{C}_8\text{H}_{10}\text{O}_3$) **SpVi**:R22.
 Ethyl alcohol ($\text{C}_2\text{H}_5\text{OH}$) **EqL**:C41, F8, H28, K23, R22; **InBi**:F8; **IsKi**:R21s; **KiB**:F8, V1; **KiL**:R21s, R22; **SpVi**:Q1.
 Ethyl amine ($\text{C}_2\text{H}_5\text{N}$) **EqL**:H29; **KiL**:H29.
 Ethyl benzene (C_8H_{10}) **ElRo**:F4.
 Ethyl diazoacetate ($\text{C}_4\text{H}_6\text{N}_2\text{O}_2$) **EqL**:R22.
 Ethylene (C_2H_4) **AnMs**:D30; **AnSp**:D30; **BiC**:C3; **EqH**:D30, L4, M24; **KiG**:L30, R1; **KiH**:D30, L4, M24, W38; **KiP**:D5; **SpVi**:C11, C13, C34, C37, H17, M20, S66, T15; **StD**:C13, C37, T15; **Sy**:B30, L18, R2.
 Ethyl naphthalene ($\text{C}_{12}\text{H}_{12}$) **SpVi**:G18; **Sy**:G18.
 Ethylene oxide ($\text{C}_2\text{H}_4\text{O}$) **IsTh**:L18; **Sy**:L18; **ThP**:L18.
 Ethyl ethoxyacetate ($\text{C}_8\text{H}_{12}\text{O}_3$) **SpVi**:R22.
 Fluoroform (CHF_3) **SpVi**:P18; **Sy**:P18; **ThF**:P18.
 Fluorosilane (SiH_3F) **SpM**:B1, B3, W23; **SpVi**:A6; **StD**:A6, B1, B3; **Sy**:B3.
 Formaldehyde (CH_2O) **SpVi**:D9, T15; **StD**:T15; **Sy**:B6.
 Formic acid (HCOOH) **SpVi**:S64.
 Formic acid dimer ($(\text{HCOOH})_2$) **SpVi**:S65.
 Formic acid, sodium salt of (CHO_2Na) **EqL**:B60; **KiL**:B60.
 Formyl radical (CHO) **KiG**:B77; **SpEl**:B77.
 Fumaric acid ($\text{C}_4\text{H}_4\text{O}_4$) **AnMs**:B58.
 Germane. See Germanium hydride.
 Germanium hydride (GeH_4) **ThF**:Z1; **ThP**:Z1; **Sy**:Z1.
 Germanium monochlorohydride (GeH_3Cl) **SpVi**:L35; **ThF**:L35.
 Glutamic acid ($\text{C}_5\text{H}_9\text{NO}_4$) **AnMs**:B58; **BiC**:K33; **EqL**:G24; **Sy**:G24, K32.
 Glutamic acid, N carbamyl ($\text{C}_6\text{H}_{10}\text{N}_2\text{O}_5$) **EqL**:G24; **Sy**:G24.
 Glutamic acid, N carboxyethyl ($\text{C}_8\text{H}_{13}\text{NO}_6$) **BiC**:K33; **Sy**:K32.
 Glutamic amide ($\text{C}_5\text{H}_{10}\text{N}_2\text{O}_3$) **SpVi**:D10.
 Glutamine. See Glutamic amide.
 Glutaric acid ($\text{C}_5\text{H}_8\text{O}_4$) **EqL**:G24; **SdSp**:H2; **SpVi**:H2; **Sy**:G24.

Glycine ($C_2H_5NO_2$) **EqL**:B61; **KiL**:B61.

Glycollic acid ($HOCH_2COOH$) **EqL**:B62; **KiL**:B62.

Guanine-t ($C_5H_5N_5O-t$) **EqL**:E8; **Sy**:E8.

Hafnium hydride (HfH_2) **IsCr**:S29; **StD**:S29.

Helium hydride (HeH) **NuM**:M38; **Sr**:M37, M38.

Heptadecane ($C_{17}H_{36}$) **InBi**:B27.

Heptane (C_7H_{16}) **EqH**:B96; **KiH**:B96.

Heptylamine ($C_7H_{17}N$) **EqL**:H29; **KiL**:H29.

Hexadecane ($C_{16}H_{34}$) **BiC**:B28; **InBi**:B28; **Sy**:B28.

Hexahydrobenzoic acid ($C_7H_{12}O_2$) **SdSp**:H2; **SpVi**:H2.

Hexane (C_6H_{14}) **KiL**:S49.

Hexestrol-t ($C_{18}H_{22}O_2-t$) **Sy**:L1.

Hexyl alcohol ($C_6H_{13}OH$) **SpVi**:Q1.

Hexyl bromide ($C_6H_{13}Br$) **KiL**:H25.

Hydrazine hydrobromide ($N_2H_4 \cdot HBr$) **SdSp**:D15; **SpVi**:D15.

Hydrazine hydrochloride ($N_2H_4 \cdot HCl$) **SdSp**:D15; **SpVi**:D15.

Hydriodic acid (HI) **EID**:B14; **EqG**:T4; **IsEq**:B39; **IsKi**:B39; **KiG**:S14, T4; **KiP**:S14; **SdCr**:B14; **SdTr**:B14; **SpM**:B392, K26; **ThF**:B39; **ThS**:B39.

Hydrobromic acid (HBr) **EID**:P26; **EIT**:P26; **EqG**:M13; **KiG**:M13, M14; **MeDf**:B70; **SpM**:G17; **SpVi**:K13; **Sy**:L18; **StD**:K13.

Hydrochloric acid (HCl) **EqG**:S18; **EcO**:B86s; **EID**:P11; **IsSp**:P12; **KiG**:S18; **KiH**:B86s; **MeDf**:B70; **SdEl**:P11; **SdTr**:P11; **SoI**:B86s; **SpM**:A2; **SpVi**:G23, H8, P12; **Sr**:J13; **StD**:P12; **Sy**:P11.

Hydrocyanic acid (HCN) **SpM**:G16, W22; **SpVi**:D29; **StD**:D29.

Hydrogen (H_2) **AdG**:S4, S12, V5, V6; **BIB**:P16; **BiC**:E7, H39, M12, P17, S38, W13, W42; **EcP**:B57; **ElMg**:B9; **ElMr**:I2; **EIP**:R9, S46; **ElRf**:E13; **EIT**:B17; **Eq**:K5; **EqG**:B56, C23, E6, M41, P7, T4; **EqH**:A7, B64, B84, B96, D30, E6, E11, F19, F20, G28, K3, K6, K15, K16, K46, K47, L4, M23, M24, M41, M48, M52, T13, V5, V6, W1, W35, W36, W37; **EqL**:C44, D37, D38, L21; **In**:H22; **InBi**:H38, M18; **InKi**:B77, K21, K46, K47, K49, K50, S49; **InSt**:L21, P27; **Is**:P21; **IsEq**:B39; **IsKi**:B39, M15, W24; **KiB**:C44; **KiG**:B56, C23, M15, M41, S18, T4, W21, W24; **KiH**:A1, A7, B57, B64, B96, D30, E6, G28, K6, K10, K15, K16, K20, K22, L4, M23, M24, M41, M48, M52, S4, W1, W36, W37, W38; **KiL**:D37, D38; **KiP**:M15, M16, R11, W24; **KiR**:G5, P27, S37, W27; **KiS**:K8; **MeAc**:M2, S51; **MeD**:D8, S52; **MeDf**:A11, A12, N15, U2, U3; **MeV**:A11, A12, B15, B16, M46, P21; **NuM**:M38; **Sd**:K34; **SdCr**:M35; **SdEl**:R12; **SdNu**:S62; **SdTr**:B59, K40; **SeDf**:B18; **SeEl**:B90; **SeEm**:B19; **SoG**:L26; **SpEl**:L5; **SpVi**:D18, S46; **Sr**:E14, M37, M38, M55, N16, O3, O4, S5; **Th**:B59; **ThD**:A11, A12, B59, G22, G27; **ThF**:B39, B59, D8, F12, N2, O8, S52; **ThP**:B59, C14, F12; **ThS**:B39, B59, O8.

Hydrogen peroxide (H_2O_2) **EID**:M34; **ElGd**:G9; **ElRe**:S35; **Ge**:G7; **SpM**:M34; **St**:G7, M34; **Sy**:G9; **ThP**:G8.

Hydrogen selenide (H_2Se) **IsSp**:M57; **StD**:M57.

Hydrogen sulfide (H_2S) **IsSp**:M57; **KiP**:D5; **SpEl**:R6; **SpM**:B50; **StD**:M57.

Hydronium ion (H_3O^+) **EcC**:G6.

Hydroxylamine (H_3NO) **SpVi**:N7; **St**:N7.

Hydroxyl ion (OH^-) **EcC**:G6; **EqH**:K38; **KiB**:K38; **SpEl**:S11.

Hydroxyl radical (OH) **ElGd**:B78; **KiG**:B77; **SpEl**:B77, B78; **SpVi**:B78; **ThF**:M3.

Hypophosphorous acid (H_3PO_2) **EqL**:M27; **KiL**:M27.

Imidogen (NH) **ElGd**:L13; **SpEl**:F9, L13.

Indium hydroxide (InO_3H_3) **SdSp**:G11; **SpVi**:G11.

Isobutane (C_4H_{10}) **EqH**:K16; **KiH**:K16.

Isothiocyanic acid (HNCS) **SpM**:D31; **StD**:D31.

Ketoglutaric acid. See Acetone dicarboxylic acid.

Lauric acid ($C_{12}H_{24}O_2$) **SdSp**:H2; **SpVi**:H2.

Lithium aluminum hydride ($LiAlH_4$) **Ge**:M36.

Lithium aluminum hydride-t ($LiAlH_4-t$) **Ge**:M36; **IsKi**:K4; **KiL**:K4.

Lithium borohydride ($LiBH_4$) **EqH**:B84.

Lithium borohydride-t ($LiBH_4-t$) **IsKi**:K4; **KiL**:K4.

Lithium hydride (LiH) **IsCr**:N14; **SdCr**:S45; **StD**:N14; **ThF**:N14.

Lithium hydride-t ($LiH-t$) **IsCr**:N14; **StD**:N14; **ThF**:N14.

Lithium hydroxide monohydrate ($LiOH \cdot H_2O$) **SpVi**:J14.

Magnesium hydroxide ($Mg(OH)_2$) **SdSp**:H11; **SpVi**:H11.

Maleic acid ($C_4H_4O_4$) **InSt**:C2; **SpVi**:C2.

Maleic acid, postassium salt of ($C_4H_3O_4K$) **InSt**:C2; **SpVi**:C2.

Malonic acid ($C_3H_4O_2$) **SdSp**:H2; **SpVi**:H2.

Manganese hydride (MnH) **SpEl**:N3; **SpVi**:N3.

Mercapto radical (HS) **SpEl**:L12, R6; **StD**:L12.

Methane (CH_4) **AnMs**:D21; **EqG**:C23; **EqH**:K15, M23; **Is**:P21; **KiG**:C23, W31; **KiH**:K15, M23; **KiP**:R11; **KiR**:W27; **MeV**:P21; **SpVi**:B7, B67, B68, B69, W28; **Sr**:L38; **StD**:B67; **ThP**:A13.

Methane-t (CH_4-t) **MeDf**:J6; **ThD**:C5, J6.

Methionine ($C_5H_{11}NO_2S$) **InBi**:D17.

Methyl acetate. See Acetic acid, methyl ester of.

Methyl acetylene. See Allylene.

Methyl alcohol (CH_3OH) **BiC**:R3, V2; **EqL**:J12; **InBi**:V2; **InSt**:J12, S15, T1; **IsSp**:S60; **KiR**:C30, P30; **SpVi**:B11, Q1, S15s, S30, S60, S61, T1; **StA**:T1; **StD**:S30.

Methyl alcohol-t (CH_3OH-t) **BiC**:R3, V2; **InBi**:V2.

Methyl aldehyde (CH_3CHO) **KiG**:R15.

Methyl bromide (CH_3Br) **Is**:M49; **IsSp**:G15; **SpEl**:G15; **SpM**:M49; **SpVi**:C33, W29; **StD**:M49; **Sy**:N13.

2-Methyl butyl alcohol ($C_5H_{12}O$) **InKi**:D20.

2-Methyl butyric acid ($C_5H_{10}O$) **InKi**:D20.

2-Methyl butyric acid, methyl ester of ($C_6H_{12}O_2$) **InKi**:D20.

Methyl chloride (CH_3Cl) **Is**:M49; **NuQ**:D35; **SpM**:M49; **StD**:M49.

Methylene chloride (CH_2Cl_2) **EID**:M60; **SpM**:M60; **StD**:M60.

Methyl fluoride (CH_3F) **SpM**:G16.

Methyl formamide (C_2H_5NO) **IsSp**:K17.

- Methyl iodide (CH_3I) **Is**:M49; **SpM**:M49; **StD**:M49.
- Methyl naphthalene ($\text{C}_{11}\text{H}_{10}$) **SpVi**:G18; **Sy**:G18.
- Methyl nitrite (CH_3NO_2) **KiR**:D23; **Sr**:D23.
- Methyl radical (CH_3) **IsKi**:M15, W24; **KiG**:M10, M15, T20, W24, W31; **KiP**:K25, M11, M15, M16, T20, W24; **Sr**:L38.
- Methyl silane (CH_3SiH_3) **SpM**:D2; **St**:D2.
- Methylene bromide (CH_2Br_2) **SpVi**:D32; **ThF**:D32.
- Naphthalene (C_{10}H_8) **IsMs**:M51; **SdSp**:P10; **SpEl**:C27; **SpVi**:G18, O6, P10; **Sr**:M51; **Sy**:G18.
- NH radical. See Imidogen.
- Nicotinamide, dihydromethyl ($\text{C}_7\text{H}_{10}\text{N}_2\text{O}$) **InKi**:R4.
- Nitric acid (HNO_3) **SpVi**:C22.
- Nitric acid, methyl ester of (CH_3NO_2) **SpVi**:T2; **St**:T2.
- Nitrobenzene ($\text{C}_6\text{H}_5\text{NO}_2$) **KiL**:B66.
- Nitrous acid (HNO_2) **SpEl**:D24; **SpVi**:D24.
- Octadecane ($\text{C}_{18}\text{H}_{38}$) **BiC**:B28; **InBi**:B27, B28; **Sy**:B28.
- Octadecene. See Octadecylene.
- Octadecylene ($\text{C}_{18}\text{H}_{36}$) **KiH**:K20; **Sy**:K20.
- Octane (C_8H_{18}) **BiC**:B28; **EqH**:B96; **InBi**:B27, B28; **KiH**:B96; **Sy**:B28.
- Octyl alcohol ($\text{C}_8\text{H}_{17}\text{OH}$) **SpVi**:Q1.
- Octyl bromide ($\text{C}_8\text{H}_{17}\text{Br}$) **KiL**:H25.
- Octyl chloride ($\text{C}_8\text{H}_{17}\text{Cl}$) **KiL**:H25.
- Oleic acid ($\text{C}_{18}\text{H}_{34}\text{O}_2$) **InBi**:F3; **KiH**:K20; **Sy**:K20.
- Oleic acid, methyl ester of ($\text{C}_{19}\text{H}_{36}\text{O}_2$) **InKi**:K21; **KiH**:K20; **Sy**:K20.
- Organic compounds-t **AnC**:V3, W39, W40; **AnDn**:C32.
- Oxalic acid ($\text{H}_2\text{C}_2\text{O}_4$) **SdSp**:H2; **SpVi**:H2.
- Pentaborane (B_5H_{10}) **EiD**:H34; **EqG**:M41; **EqH**:M41; **KiG**:M41; **KiH**:M41; **SpM**:H34; **SpVi**:H35; **StD**:H34; **Sy**:H35; **ThF**:H35.
- Pentachlorophenol ($\text{C}_6\text{Cl}_5\text{OH}$) **SpVi**:K44.
- Pentane (C_5H_{12}) **KiL**:S49; **SpVi**:S28.
- 2-Pentylchlorosulfite ($\text{C}_5\text{H}_{11}\text{ClSO}_2$) **IsKi**:L25; **KiL**:L25; **Sy**:L25.
- Phenol ($\text{C}_6\text{H}_5\text{OH}$) **EqG**:B75; **KiG**:B75; **Sr**:M53.
- Phosphine (PH_3) **EiD**:S33; **EqG**:W17; **EqH**:W17, W18, W19; **KiH**:W18; **SpM**:B93, S33, S57, S59; **SpVi**:W17; **StD**:S33.
- Phosphoric acid (H_3PO_4) **EcP**:H36s; **InKi**:K50, N1; **ThSo**:H36.
- Phosphoric acid-t ($\text{H}_3\text{PO}_4\text{-t}$) **EqH**:T23; **KiH**:T23.
- Phyllanthol **InSt**:B13.
- Picolene ($\text{C}_6\text{H}_7\text{N}$) **SoH**:C35; **ThP**:C35.
- Polythene ($(\text{C}_2\text{H}_2)_n$) **Sy**:R27.
- Potassium amide (KNH_2) **EqL**:D37, D38, S24; **KiL**:D37, D38, S24.
- Potassium arsenate, dihydrogen (KH_2AsO_4) **SdTr**:S48.
- Potassium fluoride (HF_2K) **SpVi**:J15.
- Potassium hydroxide (KOH) **EcO**:B86s; **KiH**:B86s; **SoI**:B86s.
- Potassium phosphate, dihydrogen (KH_2PO_4) **IsSp**:L33; **SdSp**:L33; **SdTr**:S48; **SpVi**:L33; **StA**:L33.
- Propane (C_3H_8) **EqH**:A7, K6, K16; **KiH**:A7, K6, K16; **KiP**:D5.
- Propene. See Propylene.
- Propionic acid ($\text{C}_3\text{H}_6\text{O}_2$) **SdSp**:H2; **SpVi**:H2.
- Propyl alcohol ($\text{C}_3\text{H}_7\text{OH}$) **KiL**:C20, L22s; **SpVi**:Q1.
- Propyl bromide ($\text{C}_3\text{H}_7\text{Br}$) **KiL**:S26; **Sy**:S26.
- Propylene (C_3H_6) **EqH**:B64; **KiH**:B64; **KiP**:D5; **SpVi**:L37; **Sy**:R2.
- Propyne. See Allylene.
- Pyridine ($\text{C}_5\text{H}_5\text{N}$) **SpVi**:C29.
- Pyridine hydrochloride ($\text{C}_5\text{H}_5\text{N}\cdot\text{HCl}$) **IsSp**:L33; **SdSp**:L33; **SpVi**:L33; **StA**:L33.
- Pyrone ($\text{C}_6\text{H}_4\text{O}_2$) **EqL**:L34; **Sy**:L34.
- Pyrrole ($\text{C}_4\text{H}_5\text{N}$) **SpVi**:T22.
- Rubidium arsenate, dihydrogen (RbH_2AsO_4) **SdTr**:S48.
- Rubidium phosphate, dihydrogen (RbH_2PO_4) **IsCr**:M39; **SdTr**:S48.
- Serine ($\text{C}_3\text{H}_7\text{O}_3\text{N}$) **BiC**:E9, E10, T3; **InBi**:E9, S44; **KiB**:E10.
- Sodium borohydride (NaBH_4) **EqH**:B84.
- Sodium formate (HCOONa) **EqL**:B62; **KiL**:B62.
- Sodium hydroxide (NaOH) **EqL**:H27; **KiL**:H27.
- Sorbic acid ($\text{CH}_3(\text{CHCH})_2\text{CO}_2\text{H}$) **InSt**:C36; **SpEl**:C36; **SpVi**:C36; **Sy**:C36.
- Stearic acid ($\text{C}_{18}\text{H}_{36}\text{O}_2$) **AnC**:B21, R31; **SdSp**:H2; **SpVi**:H2; **InBi**:F3.
- Steroids **EqH**:F19; **IsSp**:J16; **SpVi**:J16; **St**:J16; **Sy**:F19, N12.
- Stibine (SbH_3) **SpVi**:G3, H14; **StD**:G3, H14.
- Stilbene-t ($\text{C}_{14}\text{H}_{12}\text{-t}$) **SpFl**:B34; **Sy**:B34.
- Styrene-t (C_8H_8) **Sy**:B31.
- Succinic acid ($\text{C}_4\text{H}_6\text{O}_4$) **SdSp**:H2; **SpVi**:H2.
- Sulfuric acid (H_2SO_4) **EcO**:B86s; **EqL**:S19, S49; **KiH**:B86s; **KiL**:S19; **SoI**:B86s.
- Testosterone ($\text{C}_{19}\text{H}_{28}\text{O}_2$) **InBi**:F17, F18.
- Tetrachloropropane ($\text{C}_3\text{Cl}_4\text{H}_4$) **Sy**:L17.
- Tetradecane ($\text{C}_{14}\text{H}_{30}$) **BiC**:B28; **InBi**:B27, B28; **Sy**:B28.
- Toluene (C_7H_8) **EqG**:K42; **EqL**:D37, K42; **InKi**:K41; **KiL**:B83, D37; **KiP**:K41; **Sy**:B83, K42.
- Toluene-t ($\text{C}_7\text{H}_8\text{-t}$) **EqL**:O5; **KiL**:O5.
- p-Tolyldiazonium chloride ($\text{C}_7\text{H}_7\text{N}_2\text{Cl}$) **EqL**:M47; **KiL**:M47.

Trichloroacetic acid ($C_2HCl_3O_2$) **SdSp**:H2; **SpVi**:F22, H2, K43.

Triethoxy silane ($(C_2H_5O)_3SiH$) **EqL**:K23.

Triethylamine hydrochloride ($C_6H_{15}N \cdot HCl$) **IsSp**:L33; **SdSp**:L33; **SpVi**:L33; **StA**:L33.

Triethyl silane ($(C_2H_5)_3SiH$) **EqL**:K23.

Trifluoroacetic acid (CF_3COOH) **SpVi**:F22, F23.

Trifluoroacetic acid, trimer ($(CF_3COOH)_3$) **SpVi**:K29; **ThF**:K29.

Trigermane (Ge_3H_8) **ThF**:Z1; **ThP**:Z1; **Sy**:Z1.

Trimethyl acetic acid ($C_5H_{10}O_2$) **SdSp**:H2; **SpVi**:H2.

Trinitrobenzene ($C_6H_3N_3O_6$) **EqL**:K18.

Triphenylsilane ($(C_6H_5)_3SiH$) **EqL**:K23.

Triphenylsilane-t ($C_{18}H_{16}Si-t$) **IsKi**:K4; **KiL**:K4.

Tritium **BiB**:C19, W12; **BiC**:C19, E7, S39, S40, W13; **BiZ**:T11; **EqG**:D26, E6; **EqH**:E6, T13; **EqL**:O5; **Ge**:T9; **In**:S9; **InA**:U1; **InBi**:B5, T9; **Is**:W12; **KiG**:D26, M26; **KiH**:E6; **KiL**:O5; **KiR**:H4; **Sp**:C22; **SpVi**:D18.

Tryptophane ($C_{11}H_{12}N_2O_2$) **InBi**:S6.

Uracil-t ($C_4H_4N_2O_2-t$) **EqL**:E8; **Sy**:E8.

Uranium hydride (UH_3) **ElMg**:T21; **ElMm**:T21; **IsTh**:N19; **MeD**:N19; **Sd**:N19; **ThP**:N19.

Valeric acid ($C_5H_{10}O_2$) **SdSp**:H2; **SpVi**:H2.

Vinyl bromide (C_2H_3Br) **SpVi**:C12, M21; **StD**:M21.

Water (H_2O) **Ad**:B73; **AdG**:S3; **BiB**:H33, K39, P19; **BiC**:F11, H33, K39, M29, P9, W13; **BiZ**:C16, G4, H37, K31, L29, P29, T10, T11; **EcC**:G6; **EiD**:A14, H10; **ElGd**:B78, R5; **ElMm**:J7; **ElP**:D1; **ElRe**:S35; **Eq**:B74; **EqG**:W17; **EqH**:B10, K37, K38, P13, T13, W17, W18, W19; **EqL**:A4, B60, B61, B62, B76, F8, H27, H28, K18, K23, K42, K48, R22, W10; **In**:B73, N17, P13, W5; **InBi**:C16, F8, K35, M17, W15; **InKi**:B81, C1, C25, W10; **InSt**:B76; **Is**:P21, R34; **IsCr**:R36; **IsKi**:E15; **IsSp**:M57; **IsTh**:B54; **Ki**:B74, E15; **KiB**:C6, F8, J11, K37, K38, K39, M17, P14, S37; **KiH**:D22, W18; **KiL**:A4, B60, B61, B62, H27, H28; **KiR**:F5, K14; **MeAc**:L3, P2, S22; **MeD**:C9, H15, I7, L3; **MeDf**:N17, W5; **MeSt**:H15; **MeV**:H15, P21, R34; **SdEl**:A14; **SdTr**:L3; **SeCr**:R26; **SeCh**:B10, C7, E15; **SeDf**:C7; **SpEl**:B24, B78; **SpM**:B20, B91, C38, C39, J7, J8, P24, P25, R24; **SpVi**:B22, B23, G25, J17, L14, P13, P25; **StD**:M57; **ThD**:M45, P32; **ThF**:B54, L3; **ThP**:C35, G8, J18, M59.

Water-t (H_2O-t) **BiB**:B37, C10, C19, P23; **BiC**:B4, C19, P9, W13; **BiZ**:C16, L29, P22, T10, T11, T12; **In**:G20, W5, **InBi**:B4, C16, P28, T7, T8; **InSp**:M57; **MeDf**:C42, W5; **SdSp**:J19; **SpFl**:J19; **StD**:M57; **ThP**:J18.

WASHINGTON, January 4, 1957.

The National Bureau of Standards

The scope of activities of the National Bureau of Standards at its headquarters in Washington, D. C., and its major field laboratories in Boulder, Colorado, is suggested in the following listing of the divisions and sections engaged in technical work. In general, each section carries out specialized research, development, and engineering in the field indicated by its title. A brief description of the activities, and of the resultant publications, appears on the inside of the front cover.

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Electricity and Electronics. Resistance and Reactance. Electron Devices. Electrical Instruments. Magnetic Measurements. Dielectrics. Engineering Electronics. Electronic Instrumentation. Electrochemistry.

Optics and Metrology. Photometry and Colorimetry. Optical Instruments. Photographic Technology. Length. Engineering Metrology.

Heat and Power. Temperature Physics. Thermodynamics. Cryogenic Physics. Rheology. Engine Fuels. Free Radicals Research.

Atomic and Radiation Physics. Spectroscopy. Radiometry. Mass Spectrometry. Solid State Physics. Electron Physics. Atomic Physics. Neutron Physics. Nuclear Physics. Radioactivity. X-rays. Betatron. Nucleonic Instrumentation. Radiological Equipment. AEC Radiation Instruments.

Chemistry. Organic Coatings. Surface Chemistry. Organic Chemistry. Analytical Chemistry. Inorganic Chemistry. Electrodeposition. Gas Chemistry. Physical Chemistry. Thermochemistry. Spectrochemistry. Pure Substances.

Mechanics. Sound. Mechanical Instruments. Fluid Mechanics. Engineering Mechanics. Mass and Scale. Capacity, Density, and Fluid Meters. Combustion Controls.

Organic and Fibrous Materials. Rubber. Textiles. Paper. Leather. Testing and Specifications. Polymer Structure. Plastics. Dental Research.

Metallurgy. Thermal Metallurgy. Chemical Metallurgy. Mechanical Metallurgy. Corrosion. Metal Physics.

Mineral Products. Engineering Ceramics. Glass. Refractories. Enameled Metals. Concreting Materials. Constitution and Microstructure.

Building Technology. Structural Engineering. Fire Protection. Air Conditioning, Heating, and Refrigeration. Floor, Roof, and Wall Coverings. Codes and Specifications. Heat Transfer.

Applied Mathematics. Numerical Analysis. Computation. Statistical Engineering. Mathematical Physics.

Data Processing Systems. SEAC Engineering Group. Components and Techniques. Digital Circuitry. Digital Systems. Analog Systems. Application Engineering.

● Office of Basic Instrumentation

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Radio Propagation Physics. Upper Atmosphere Research. Ionospheric Research. Regular Propagation Services. Sun-Earth Relationships.

Radio Propagation Engineering. Data Reduction Instrumentation. Modulation Systems. Navigation Systems. Radio Noise. Tropospheric Measurements. Tropospheric Analysis. Radio Systems Application Engineering.

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