

MASTER

UCRL-8289

4079

Decay Schemes

D. STROMINGER
J. M. HOLLANDER

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June 1958 · Berkeley, California

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Berkeley, California

Contract No. W-7405-eng-48

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DECAY SCHEMES

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 Radiation Laboratory
 University of California, Berkeley, California

The disintegration schemes presented here have been constructed from information contained in the Table of Isotopes, by Strominger, Hollander, and Seaborg, published as a supplement to the April, 1958, issue of the Reviews of Modern Physics.

They are arranged according to mass number so as to show more clearly than was possible in the Table of Isotopes the various isobaric and parent-daughter relationships encountered in alpha and beta decay. References to the original literature are not given here.

In these drawings, energies of excited states above the ground state are given in Mev to the right of the levels, and spin and parity assignments are indicated to the left. Directly measured spins are underlined, as 1/2+. If the spin is uniquely determined by other means, the quantum number is given without modification, as 3-. Probable spin values are indicated with parentheses, as (2+). This nomenclature does not apply to the parity assignments, which are heavily dependent upon theory.

The disintegration energy, or Q value, of a nuclear transformation is the mass difference (expressed in Mev) between the initial and final systems. For radioactive decay processes, Q equals the sum of the particle kinetic energy, nuclear recoil energy, and the energies of any gamma-transitions necessary to de-excite the final nucleus to its ground state. In β^+ decay the Q value is given as Q_{EC} , to avoid ambiguity associated with the addition of $2 m_0 c^2$ to the positron energy. Alpha and beta disintegration energies are given as Q_α and Q_β respectively.

The percentage figures given in the drawings total 100 percent for the total decay of each nuclide. In some cases the symbol † is used to designate the relative branchings of one mode of decay to different final states of the daughter nucleus.

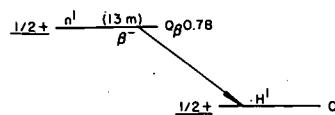
The half-lives of levels, when known, are given directly above the levels. A double arrow drawn from the ground state to a higher level shows that

the higher level has been Coulomb excited.

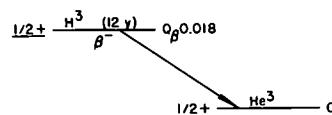
We are indebted to Virginia Shirley and Eileen Doyle for their assistance in the preparation of the manuscript.

This work was performed under the auspices of the United States Atomic Energy Commission.

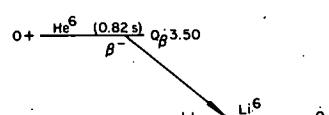
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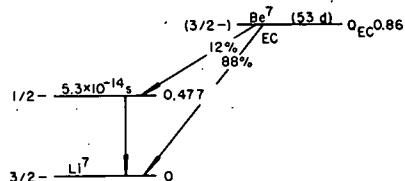
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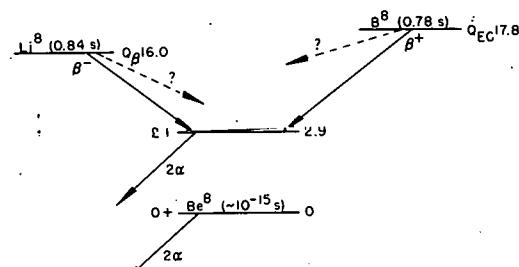
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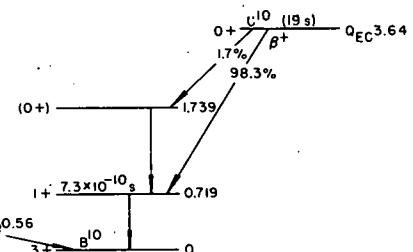
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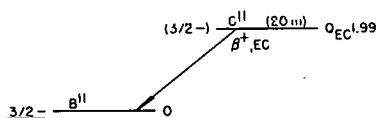
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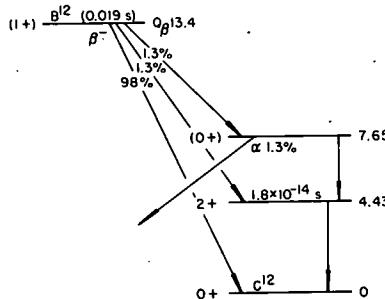
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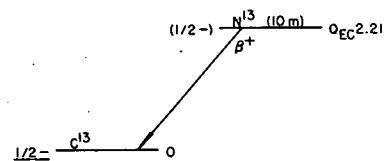
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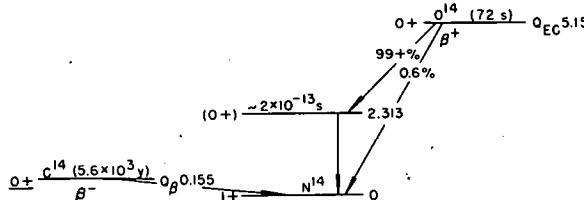
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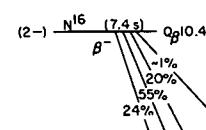
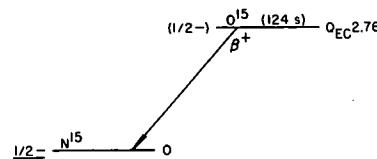
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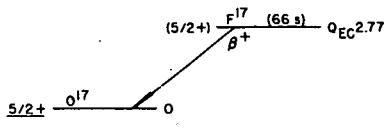
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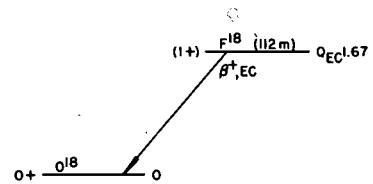
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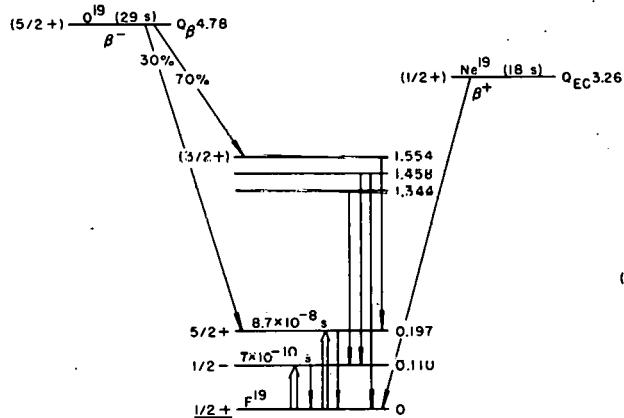
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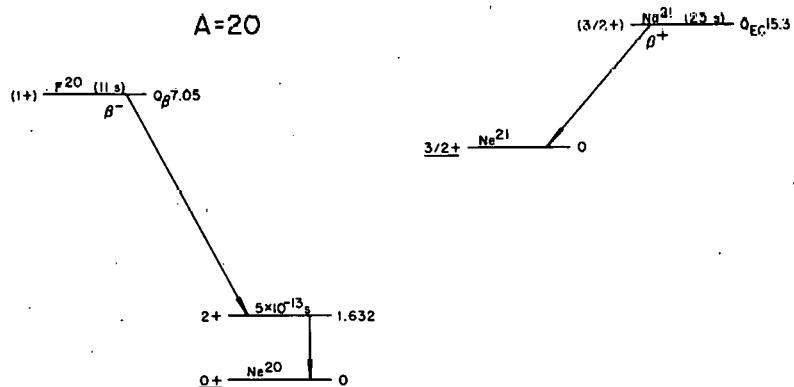
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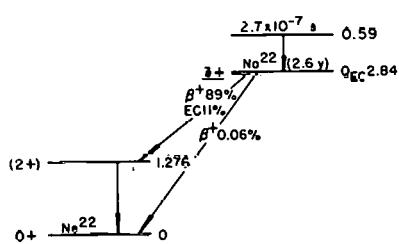
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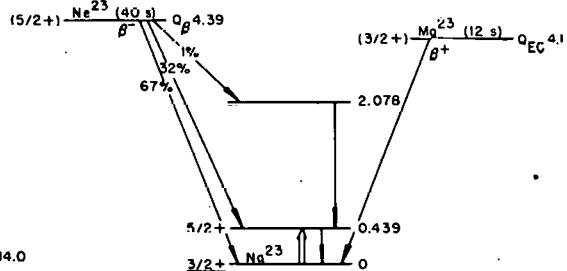
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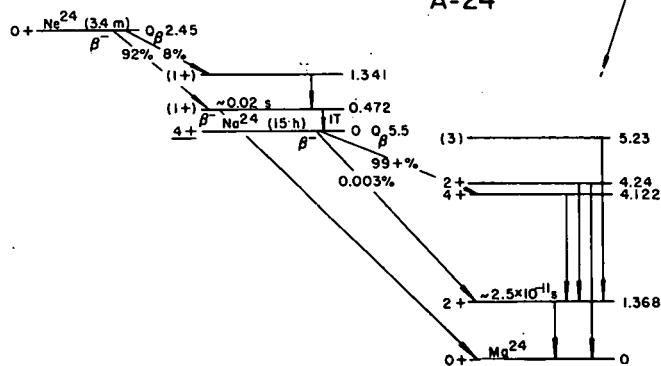
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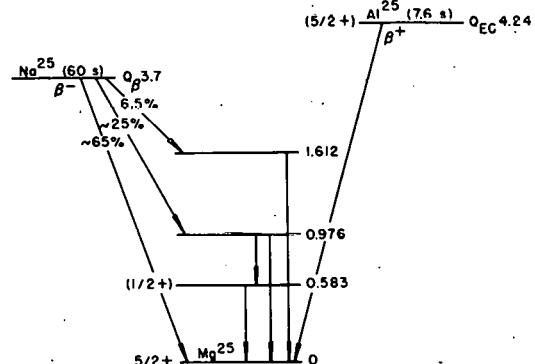
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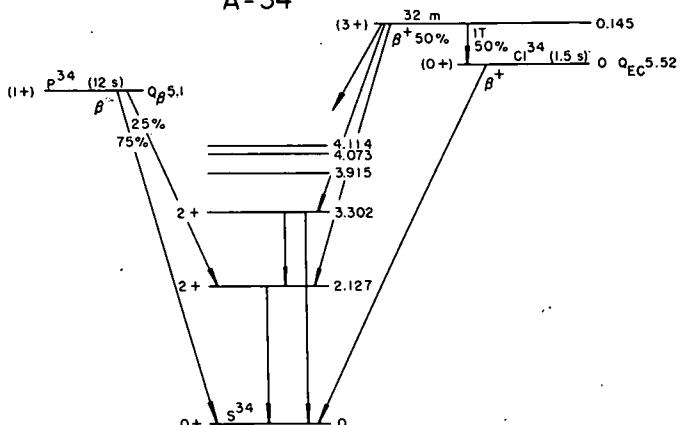
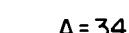
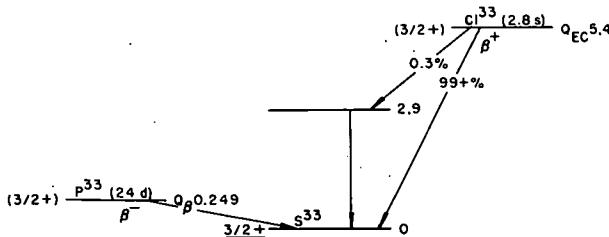
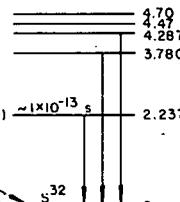
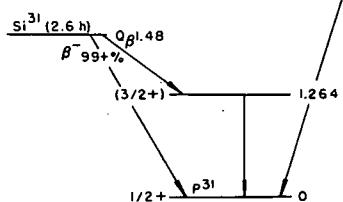
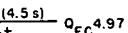
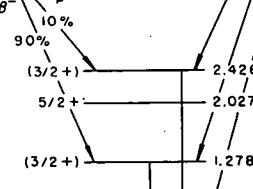
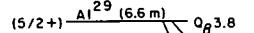
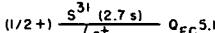
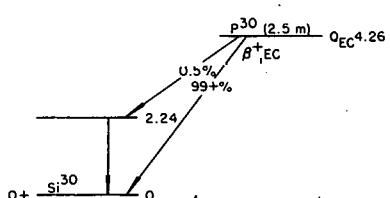
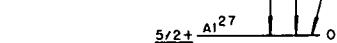
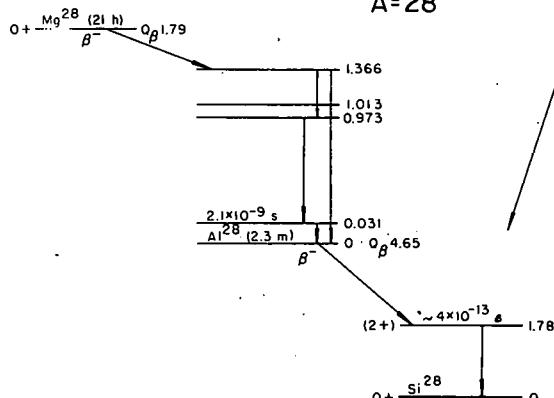
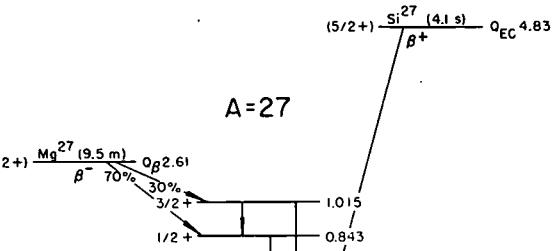
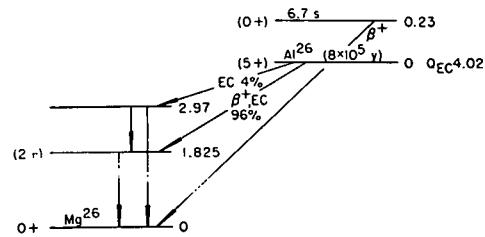


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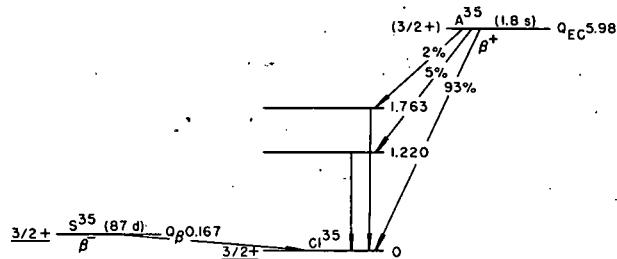


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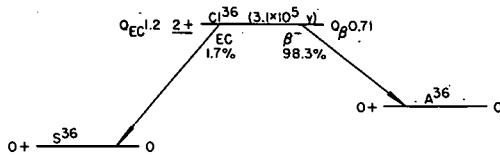




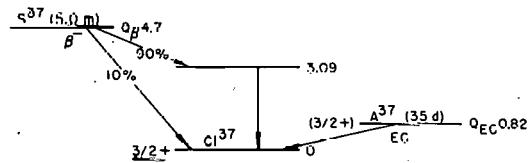
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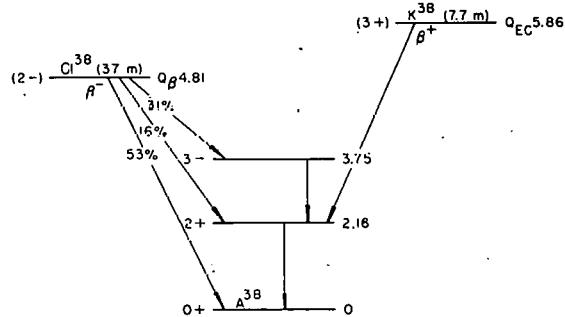
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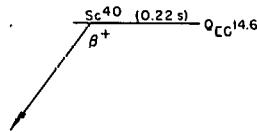
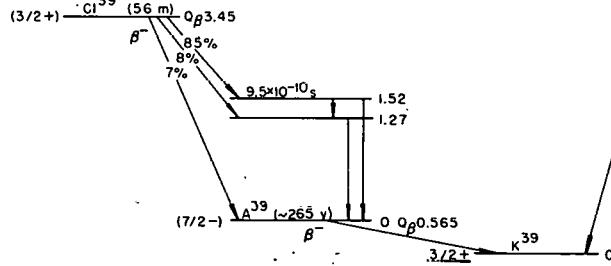
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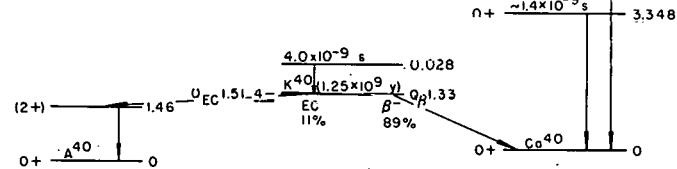
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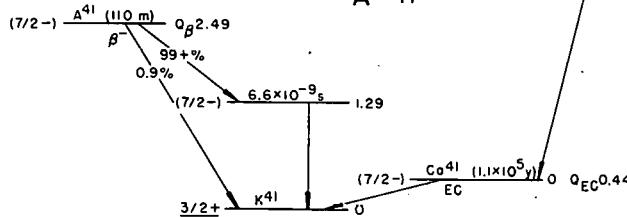
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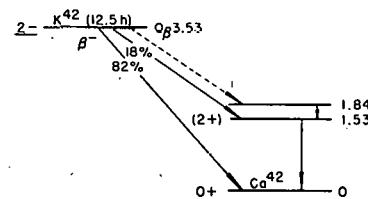
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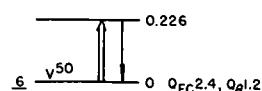
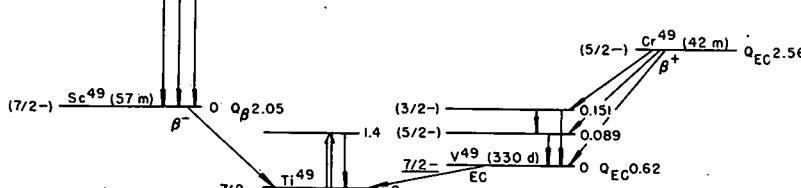
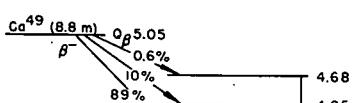
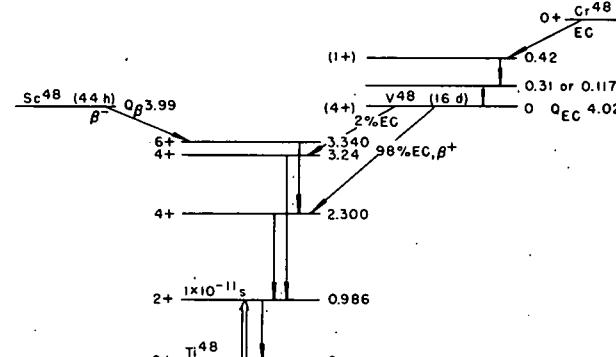
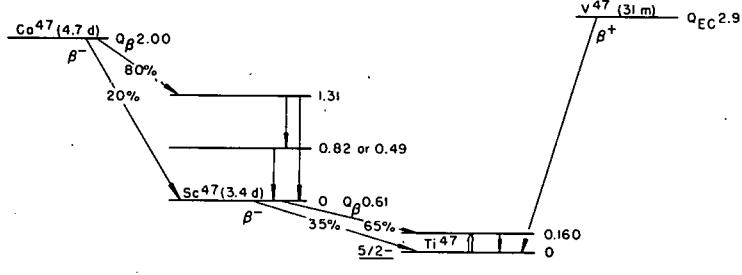
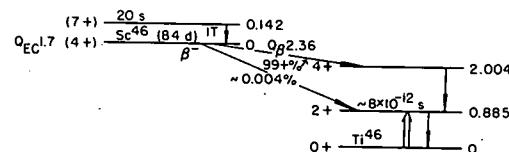
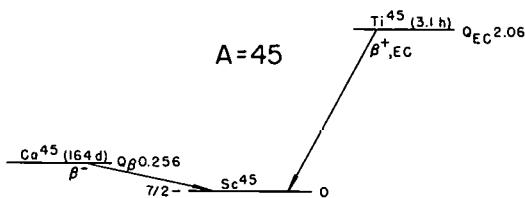
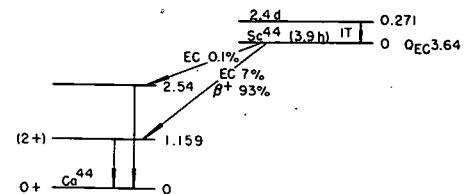
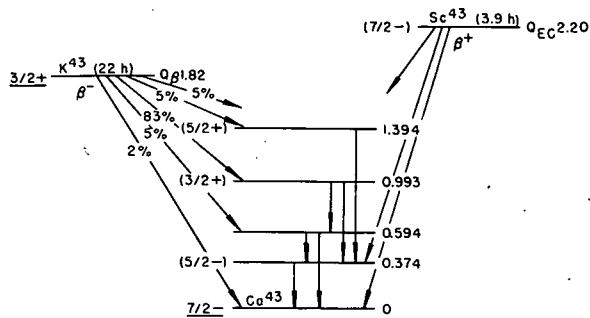


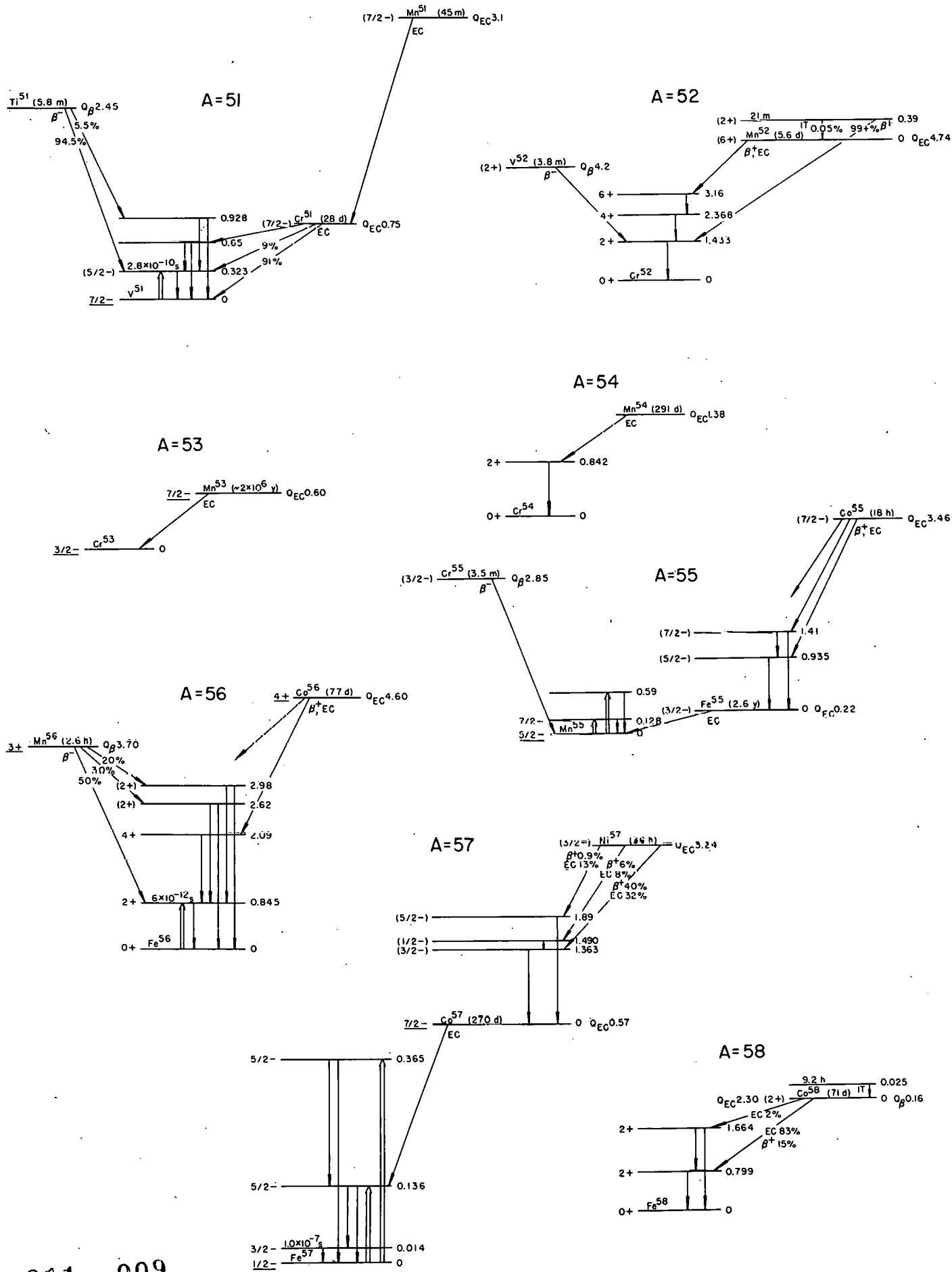
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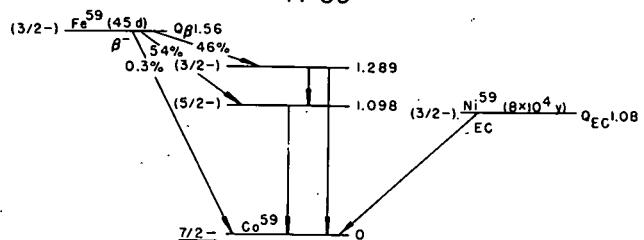
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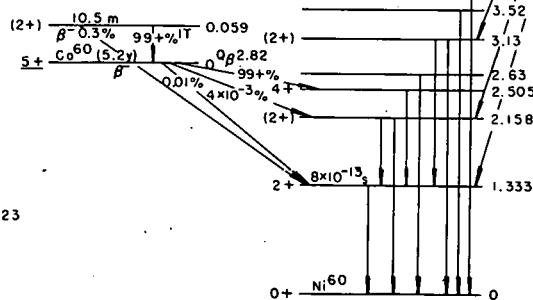




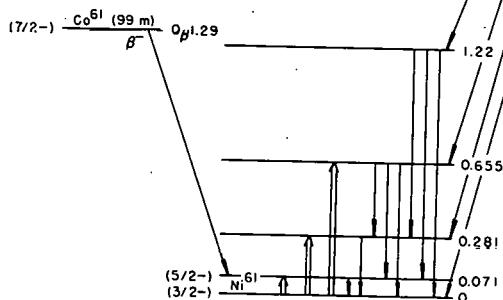
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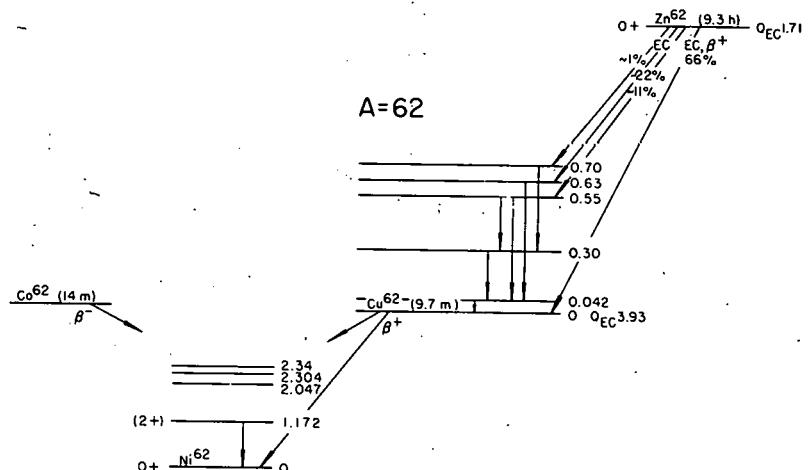
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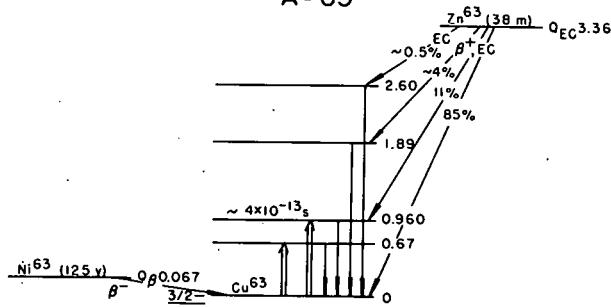
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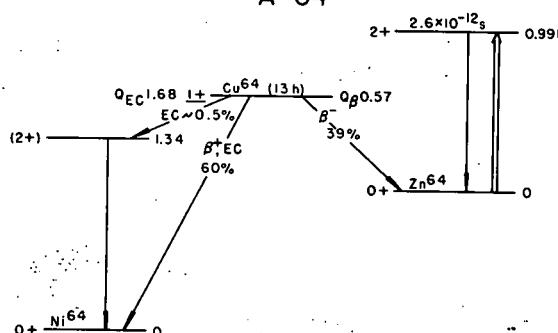
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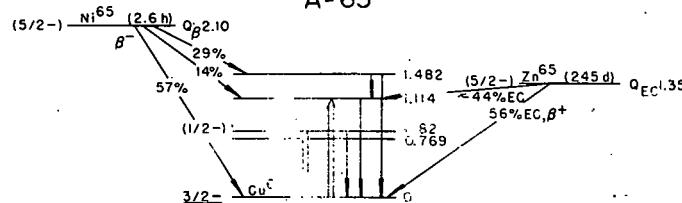
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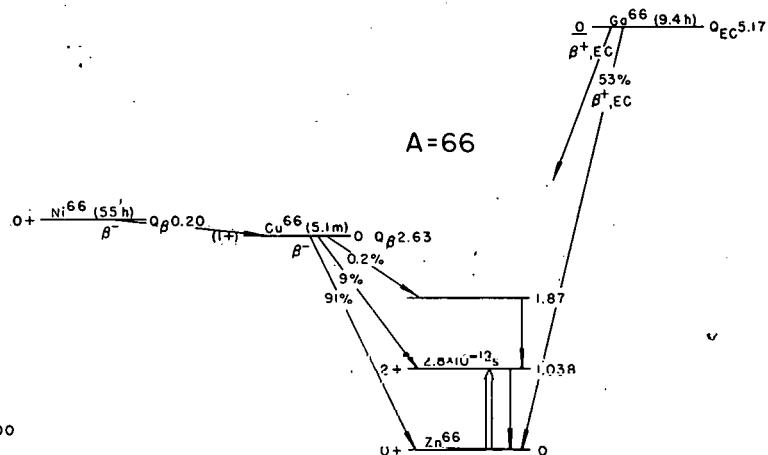
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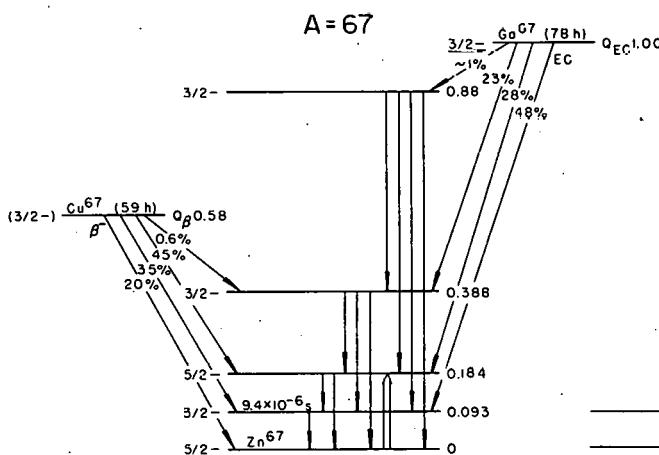
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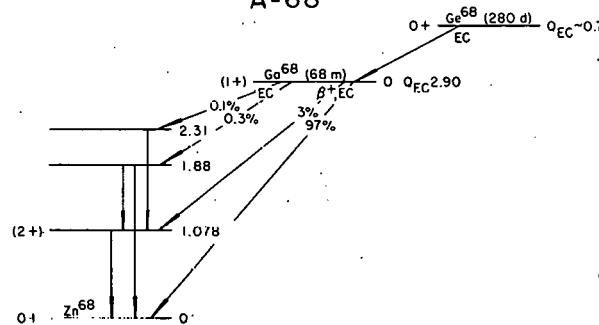
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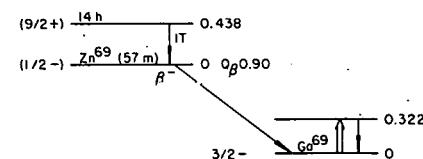
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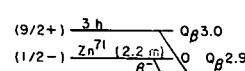
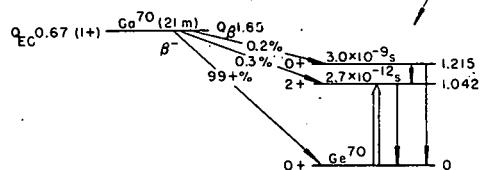
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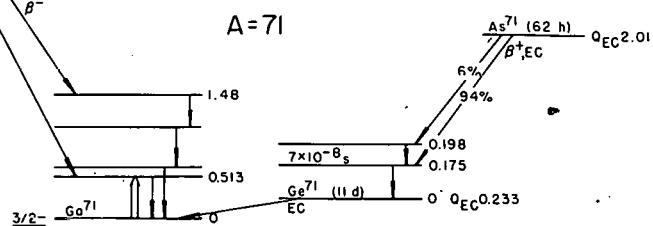
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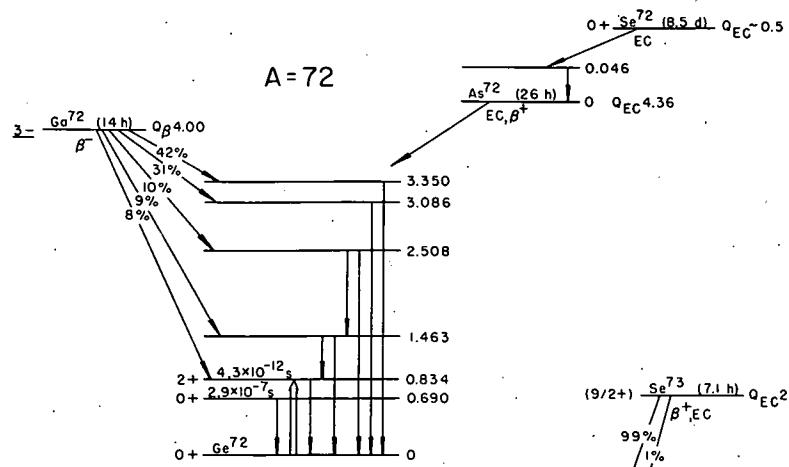


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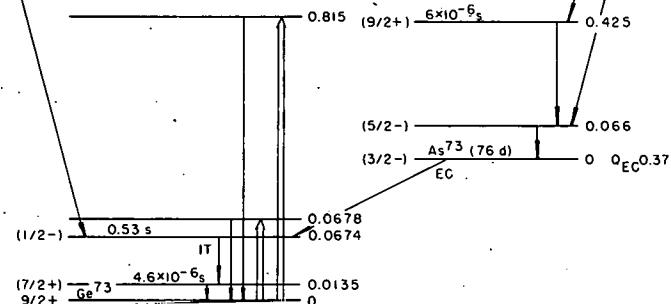
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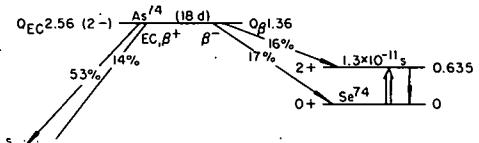


(3/2-) Ge-73 (5.0 h) $0\beta 1.5$

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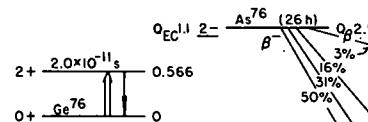
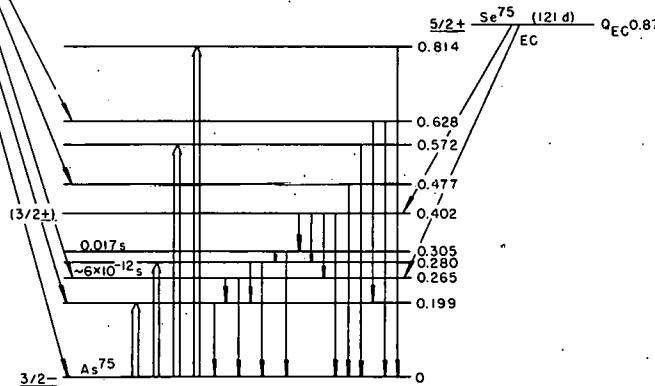


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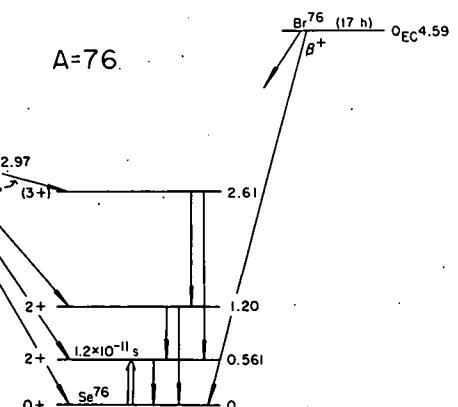


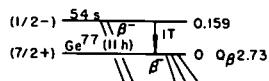
(7/2+) Ge-75 (82 m) $0\beta 0.14$

A = 75

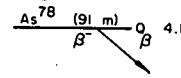
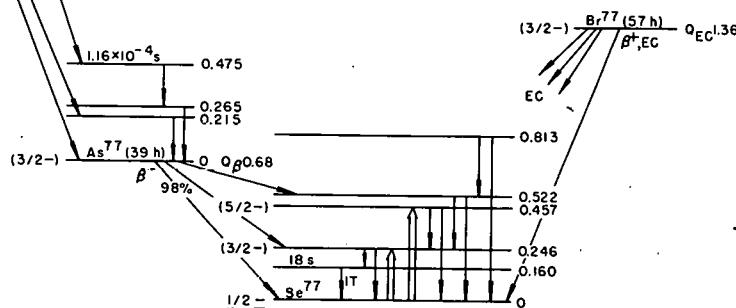


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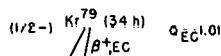
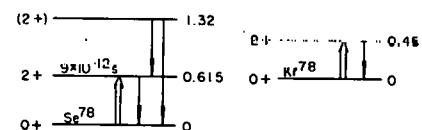




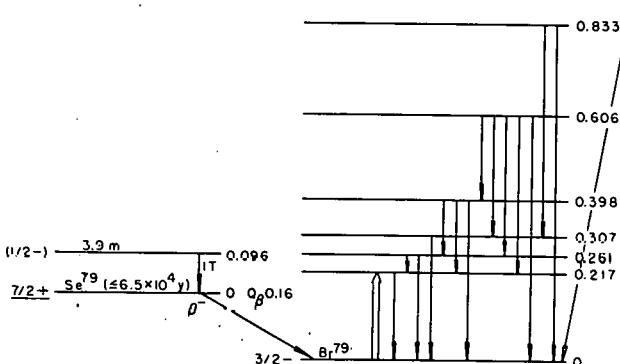
A = 77



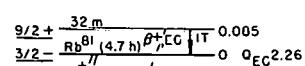
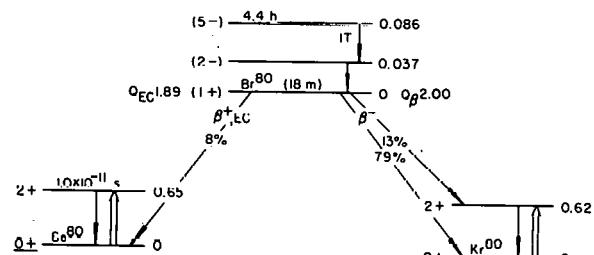
A = 78



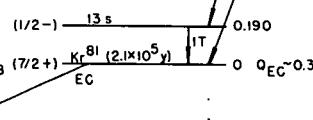
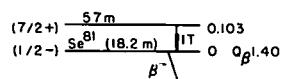
A = 79

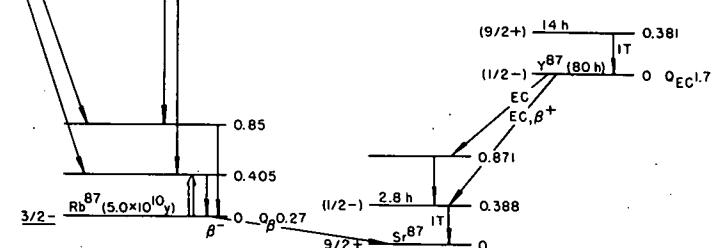
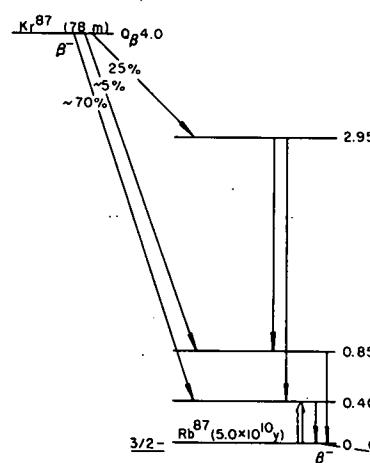
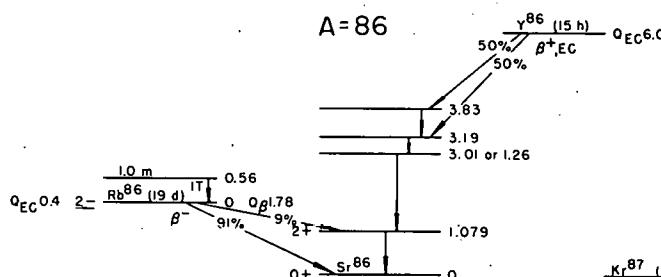
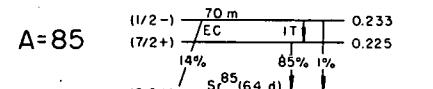
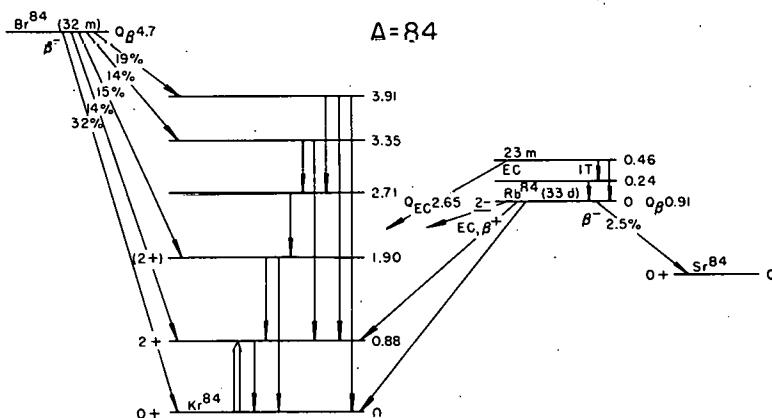
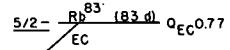
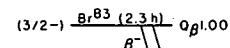
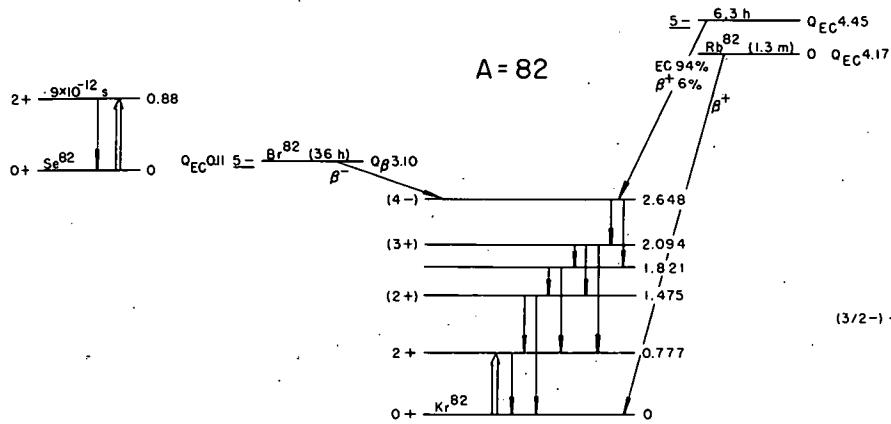


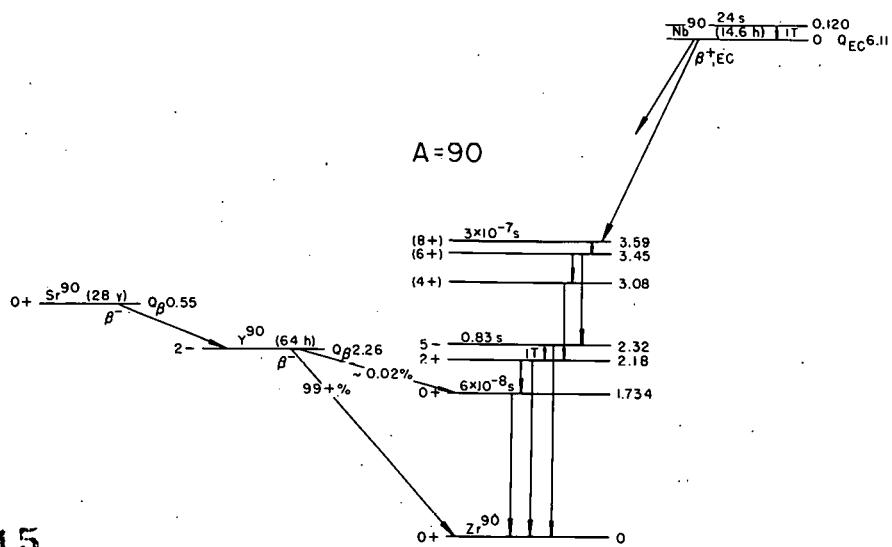
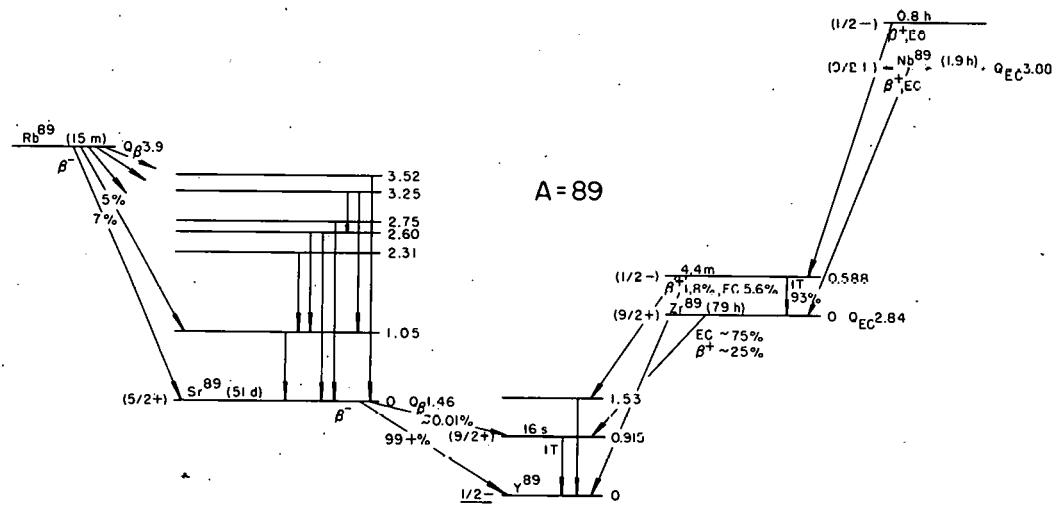
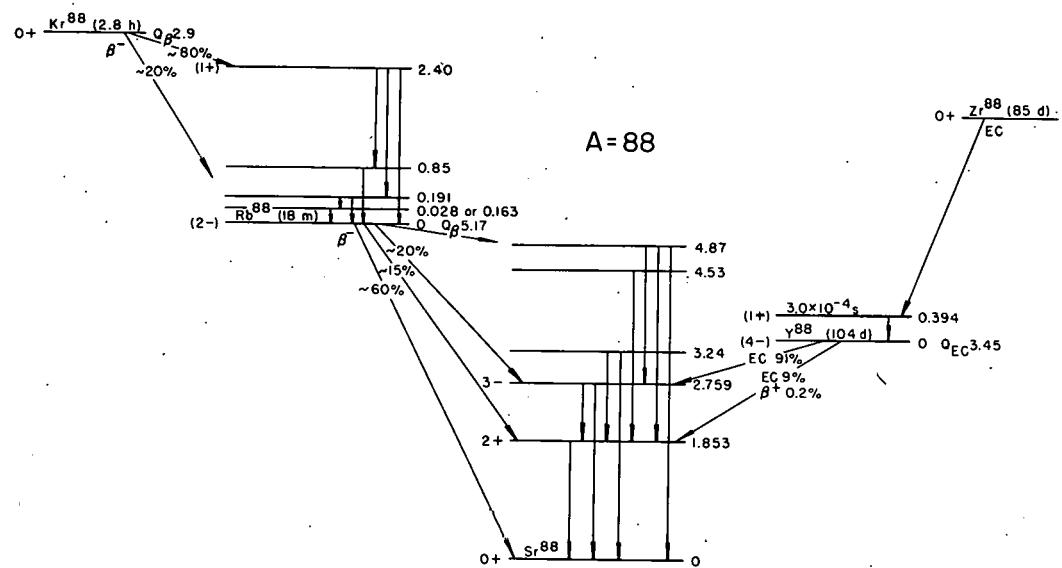
A = 80

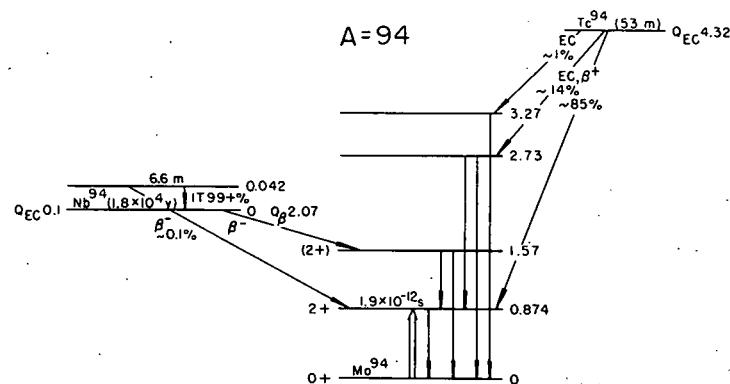
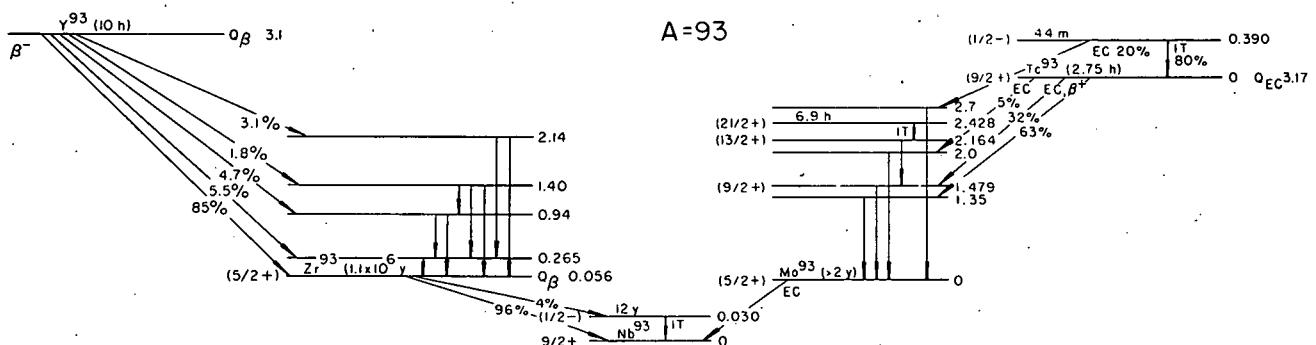
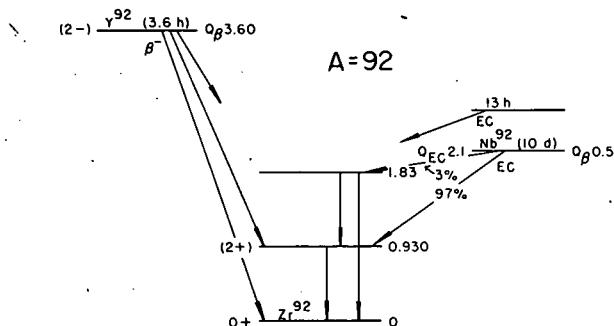
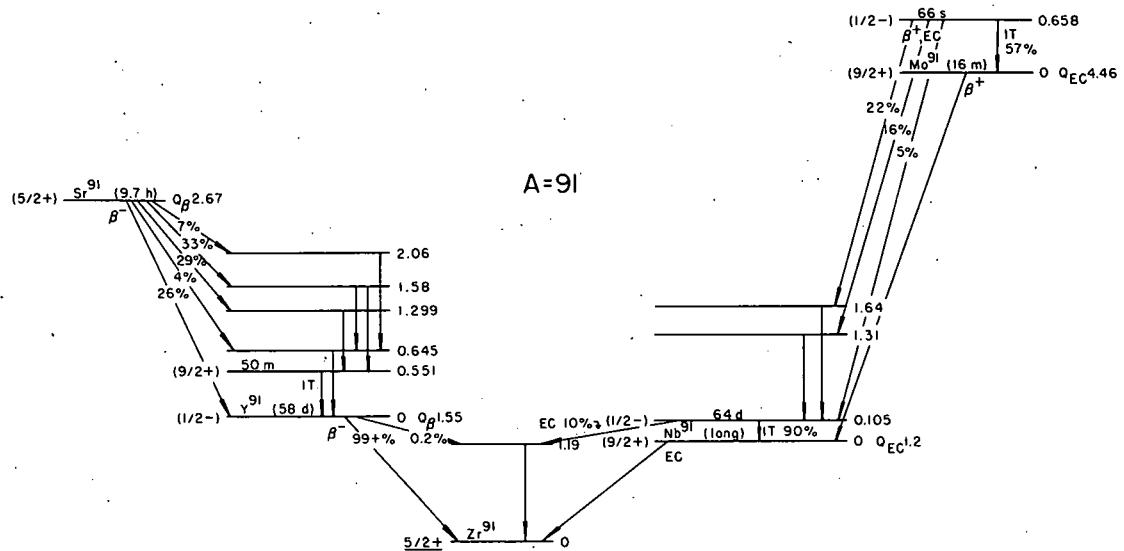


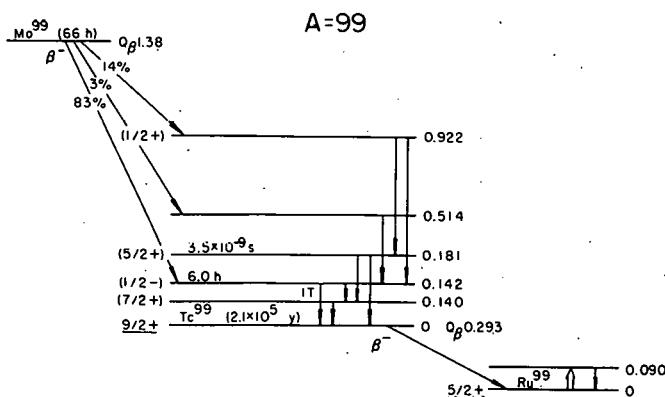
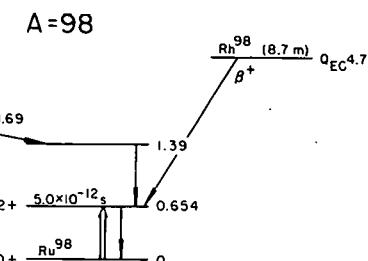
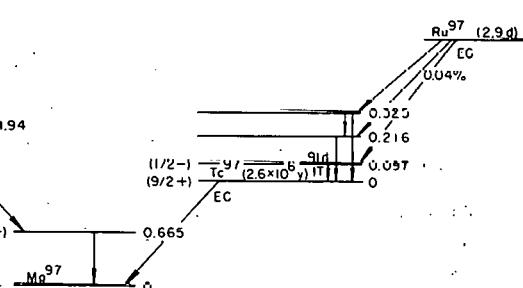
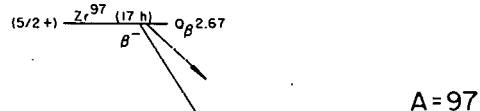
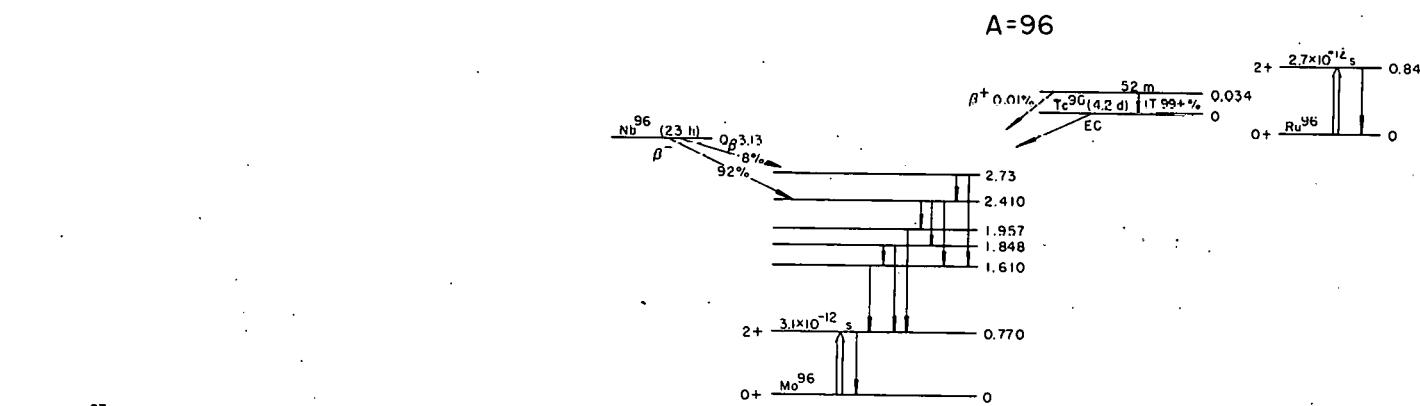
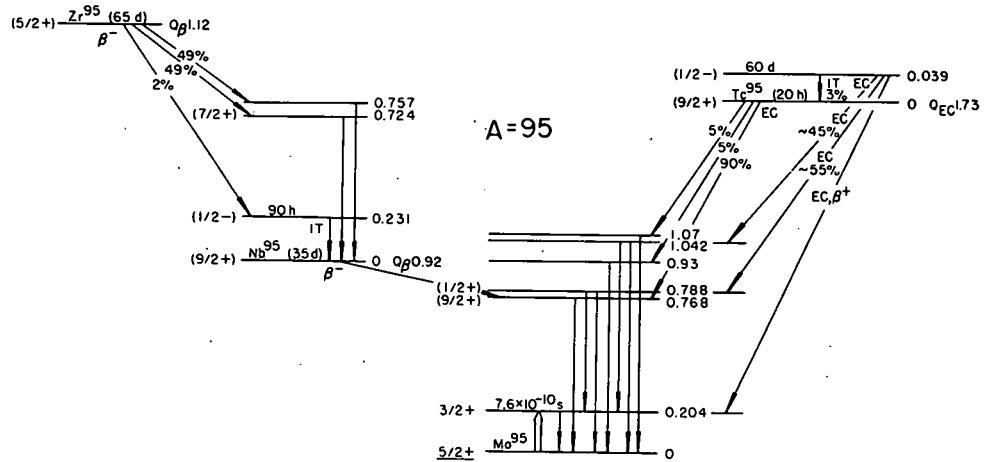
A = 81



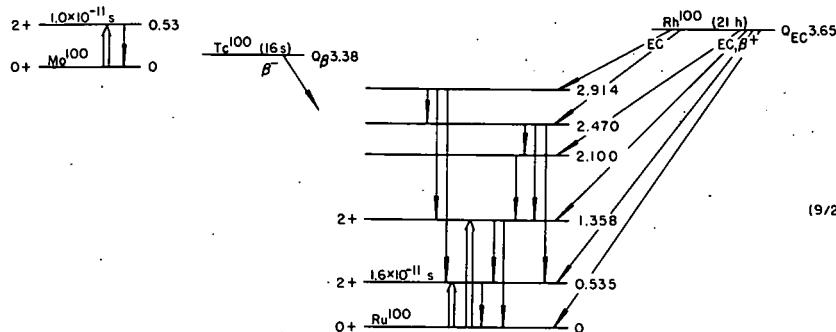




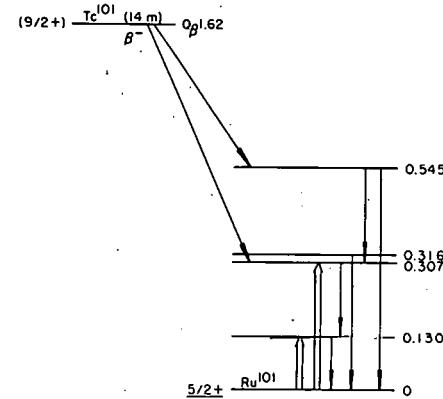




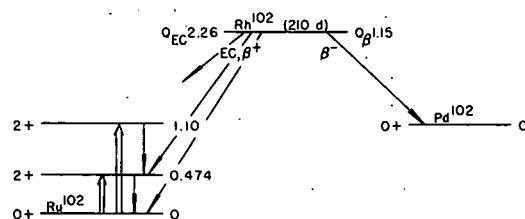
A=100



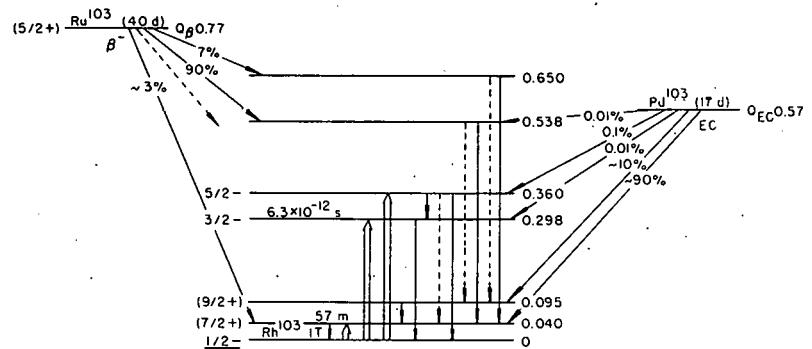
A=101



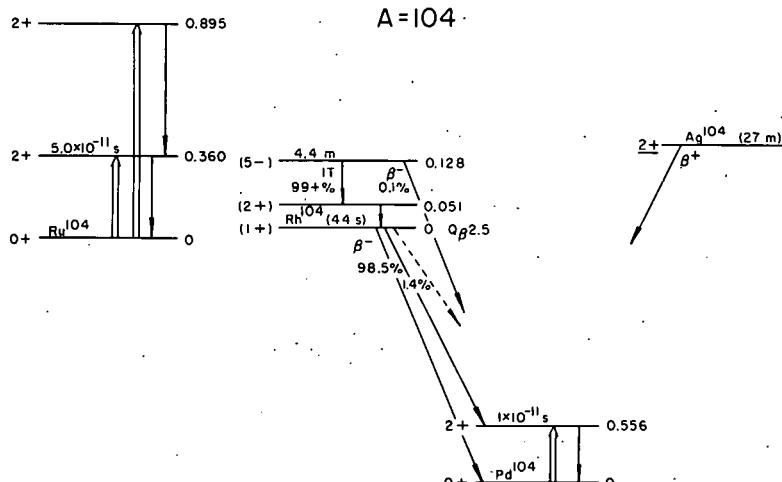
A=102



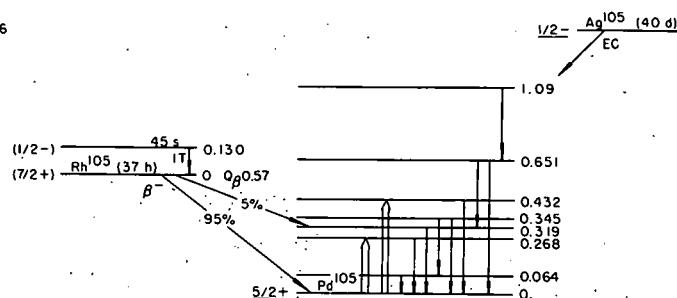
A=103

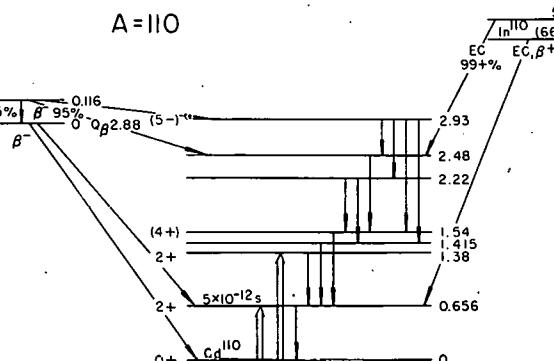
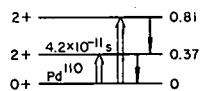
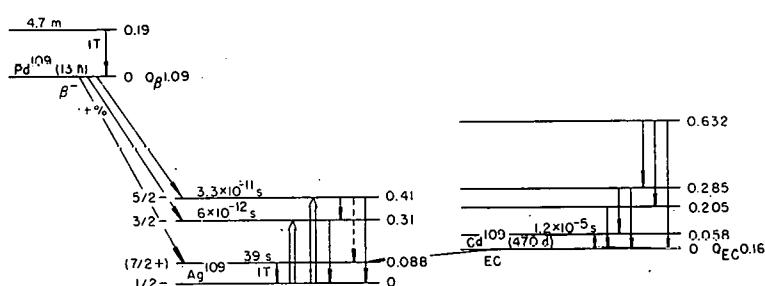
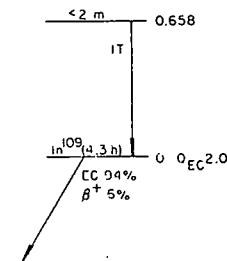
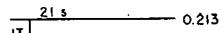
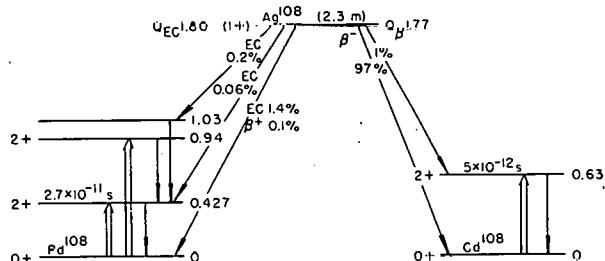
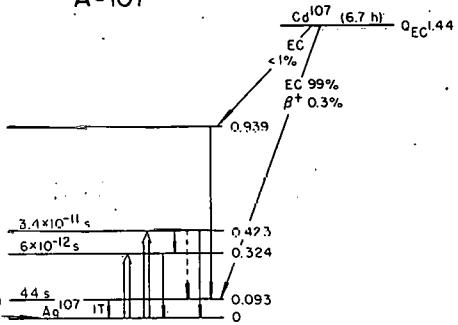
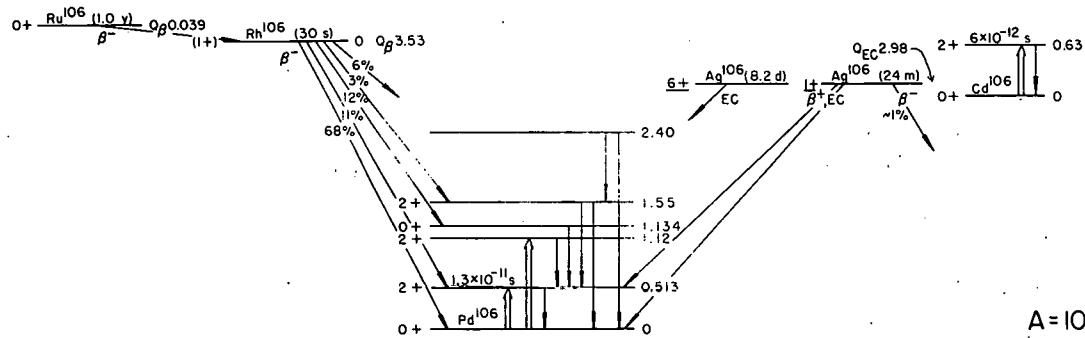


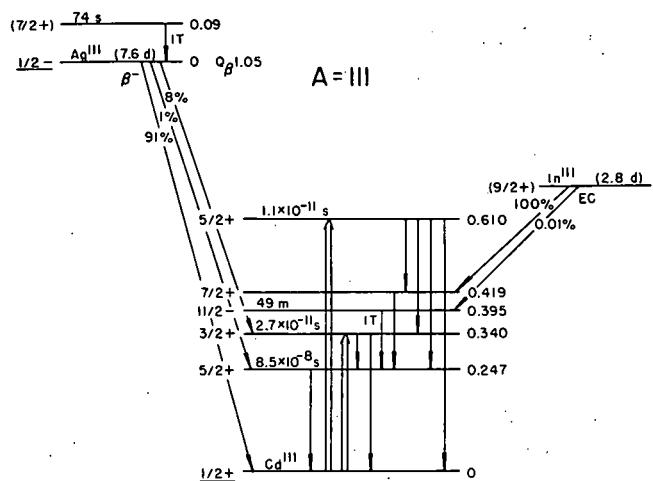
A=104



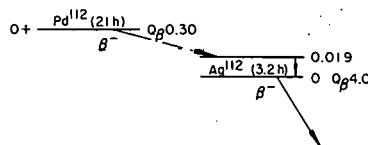
A=105



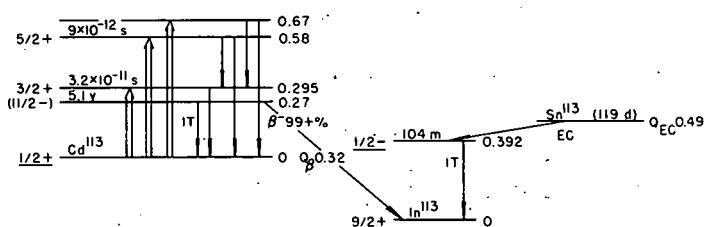




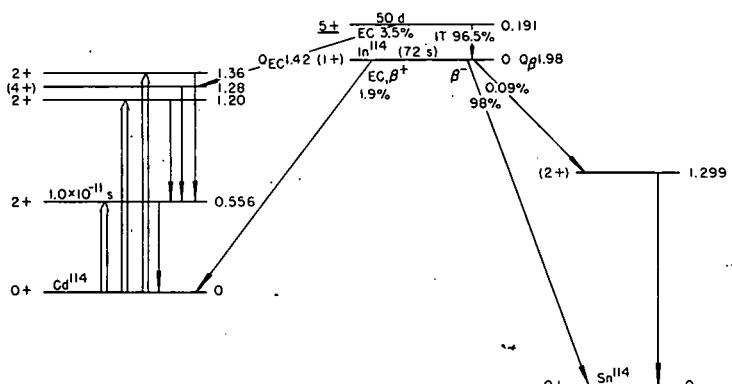
$$A = 112$$



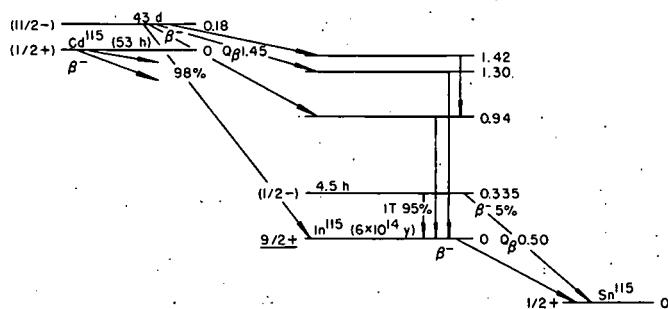
$$A = 113$$



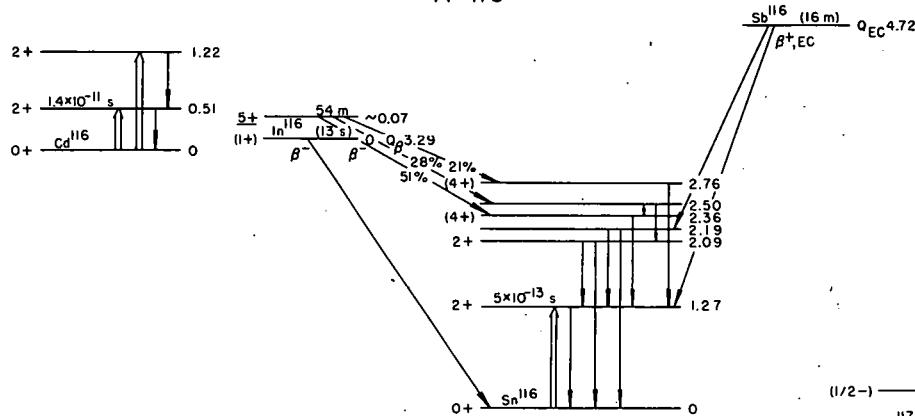
$$A = 114$$



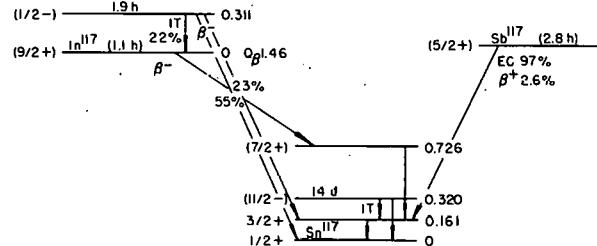
A=115



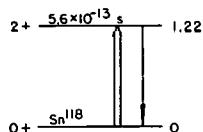
$A=116$



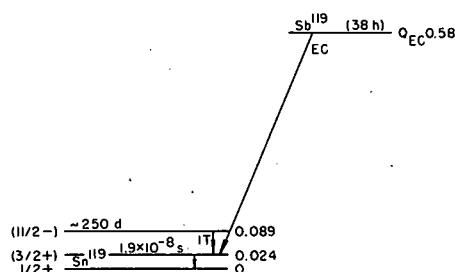
$A=117$



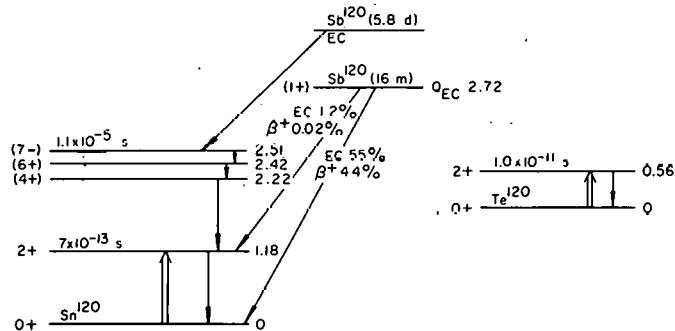
$A=118$



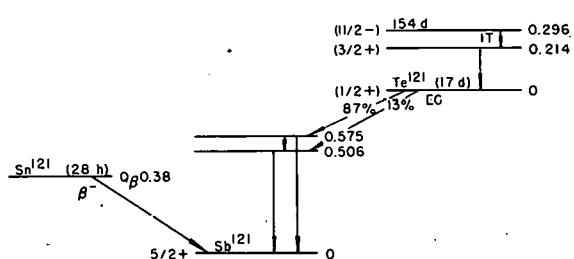
$A=119$



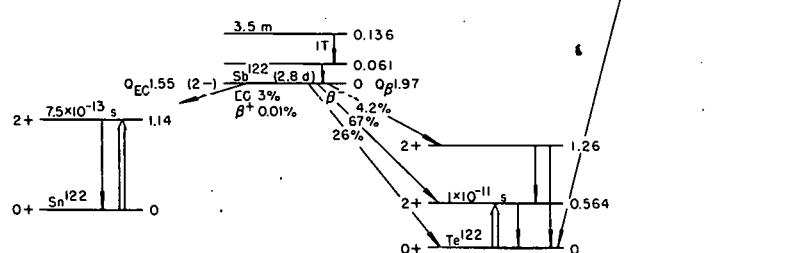
$A=120$



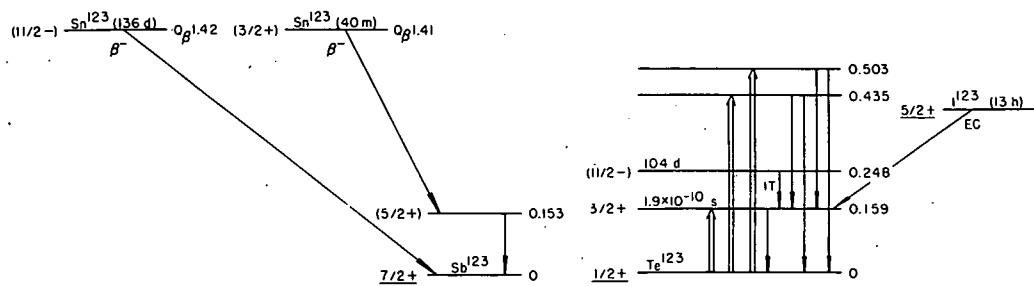
$A=121$



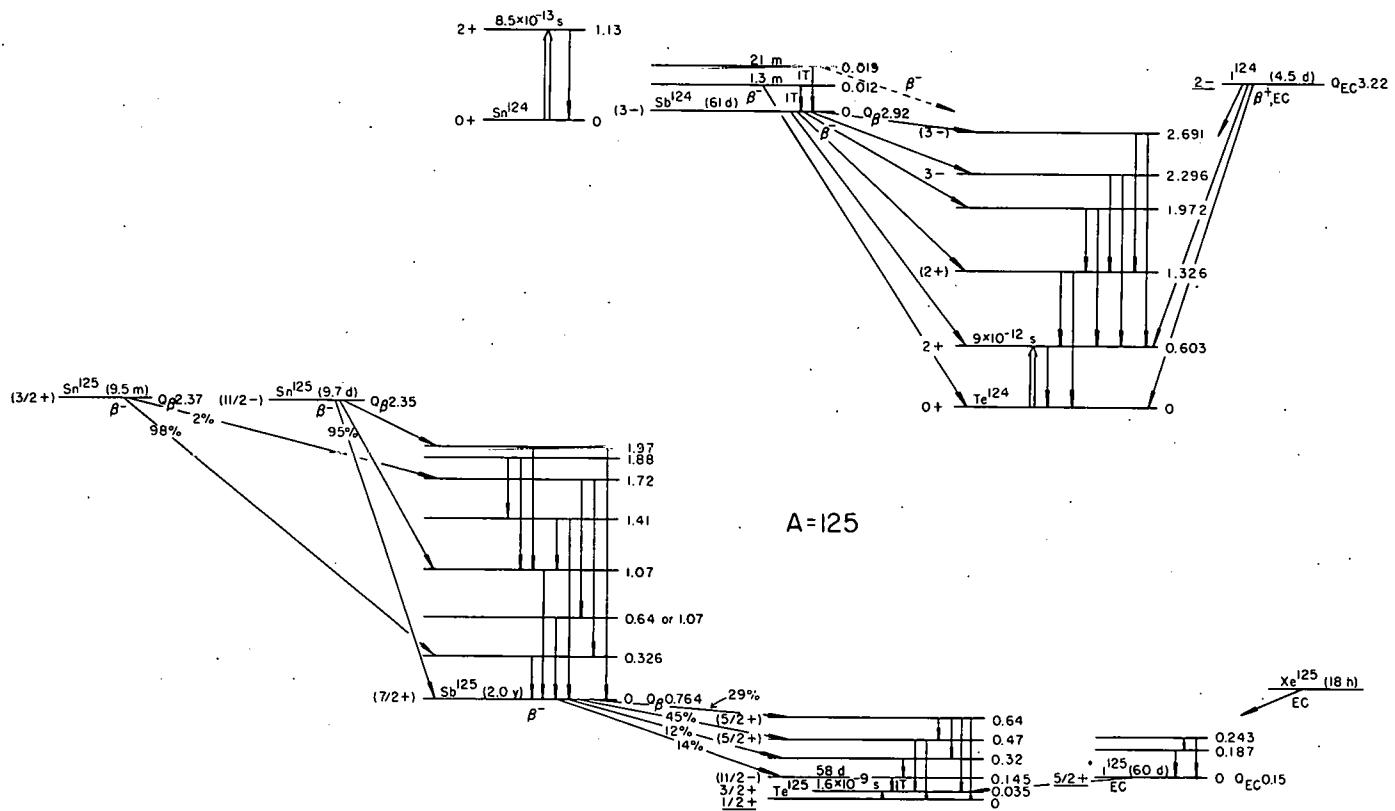
$A=122$



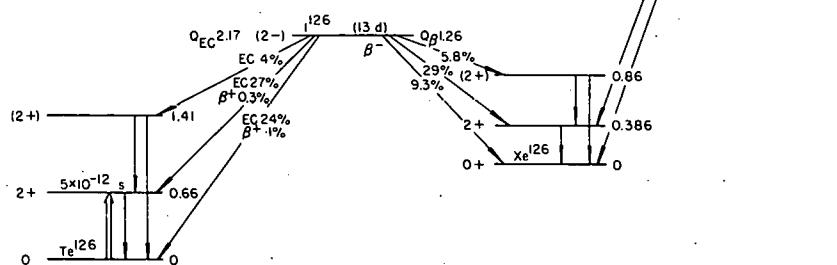
$A=123$

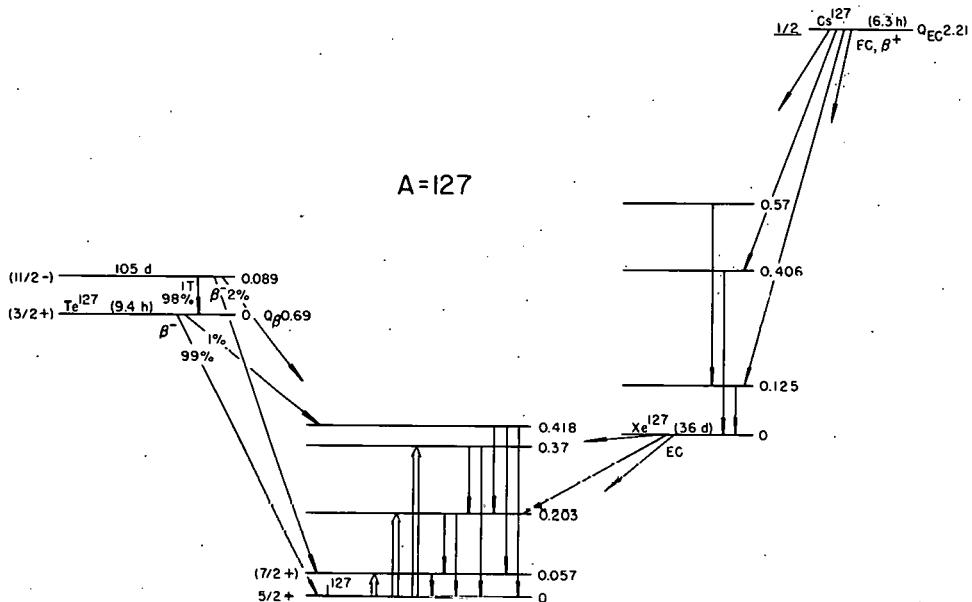


$A=124$

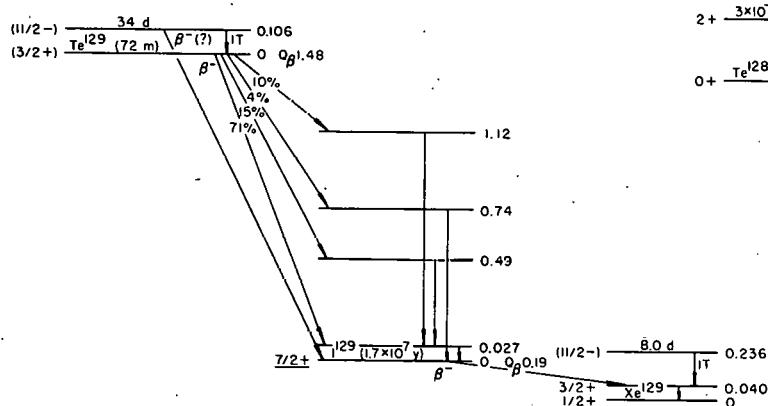


$A=126$

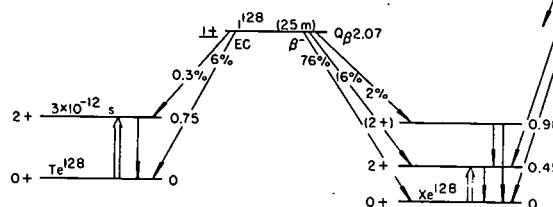




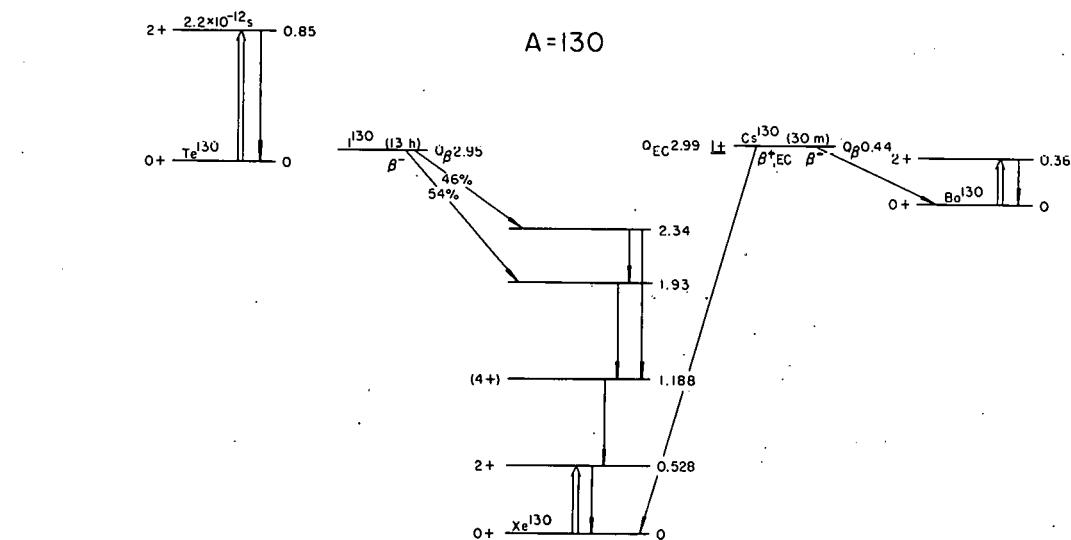
$$\Delta = 129$$

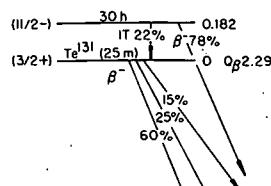


A=128

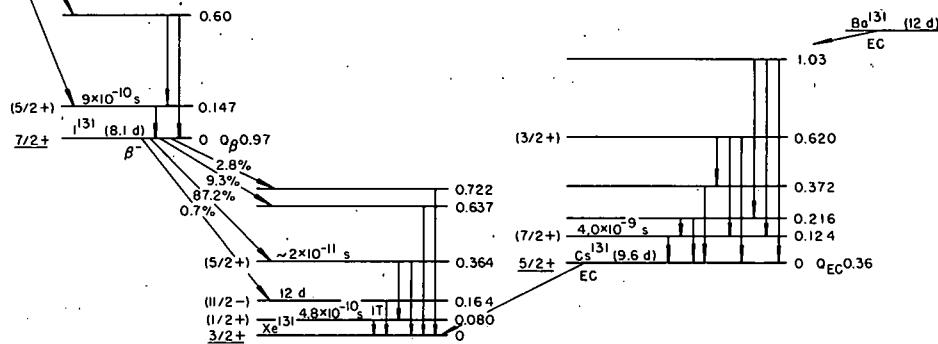


$$\Delta = 129$$

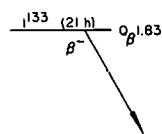




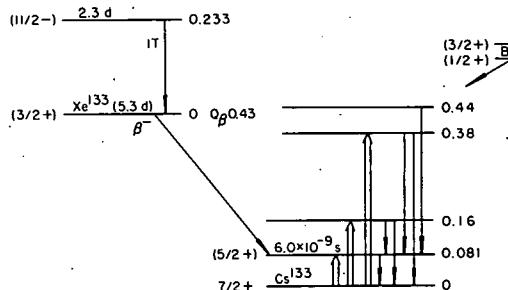
A=131



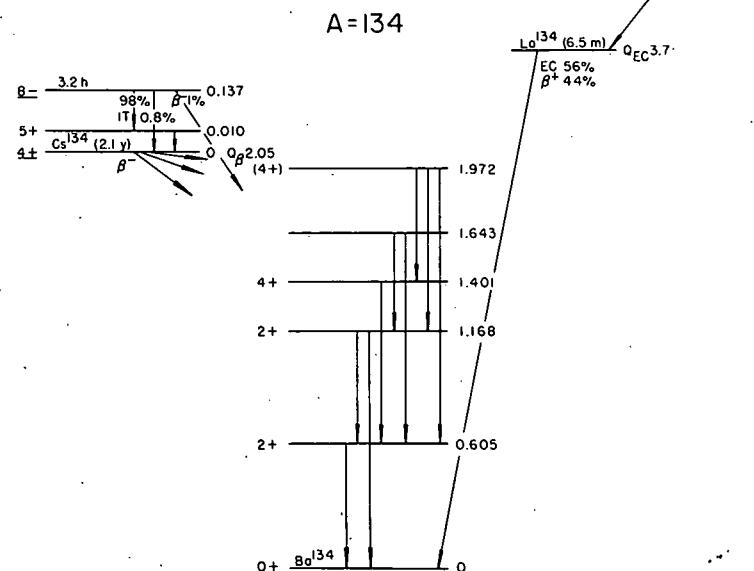
A=132

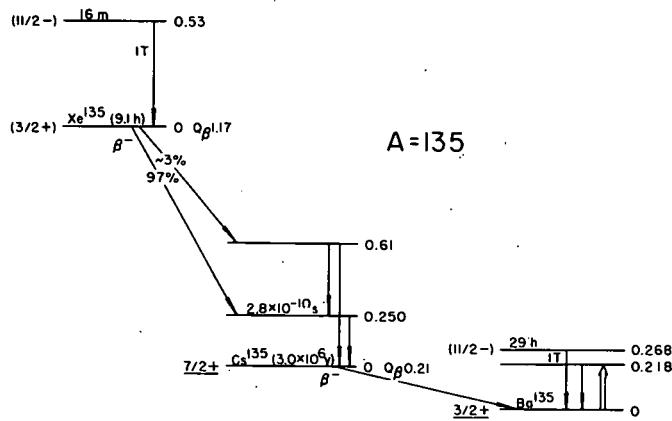


A=133

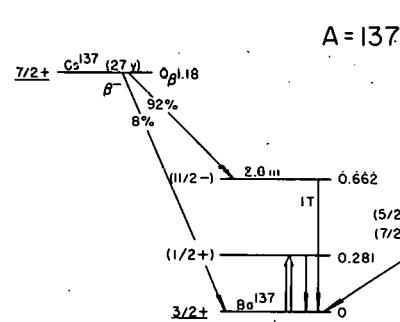
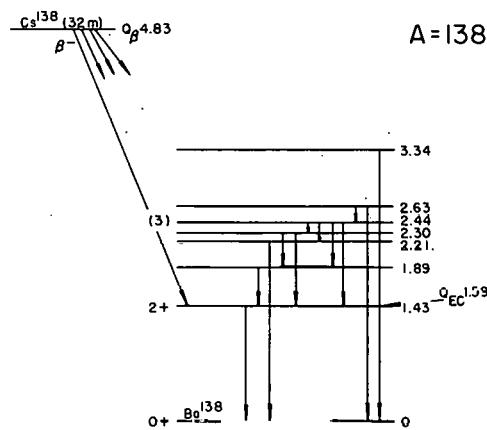
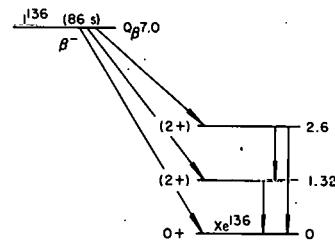


A=134

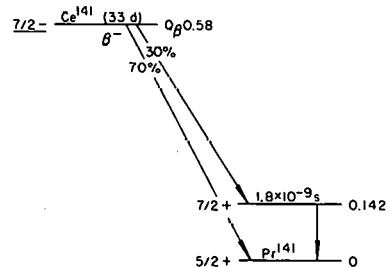




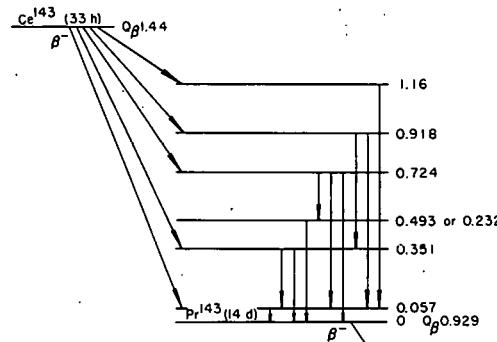
A = 136



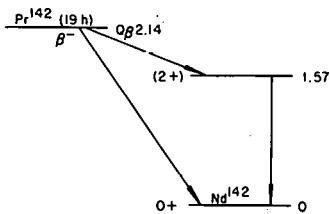
A = 141

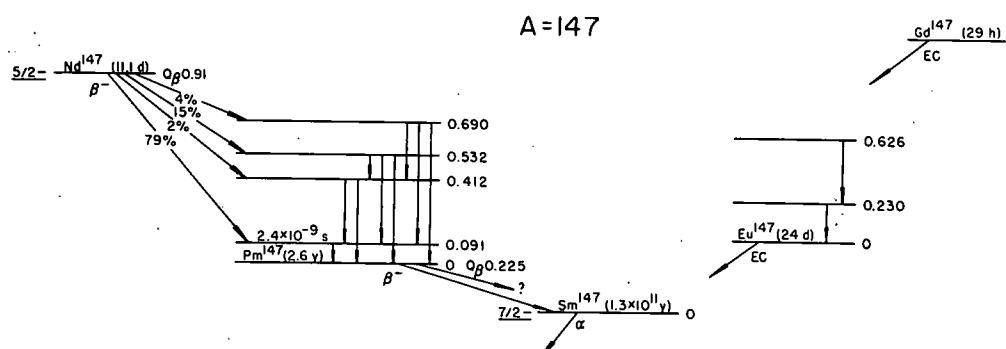
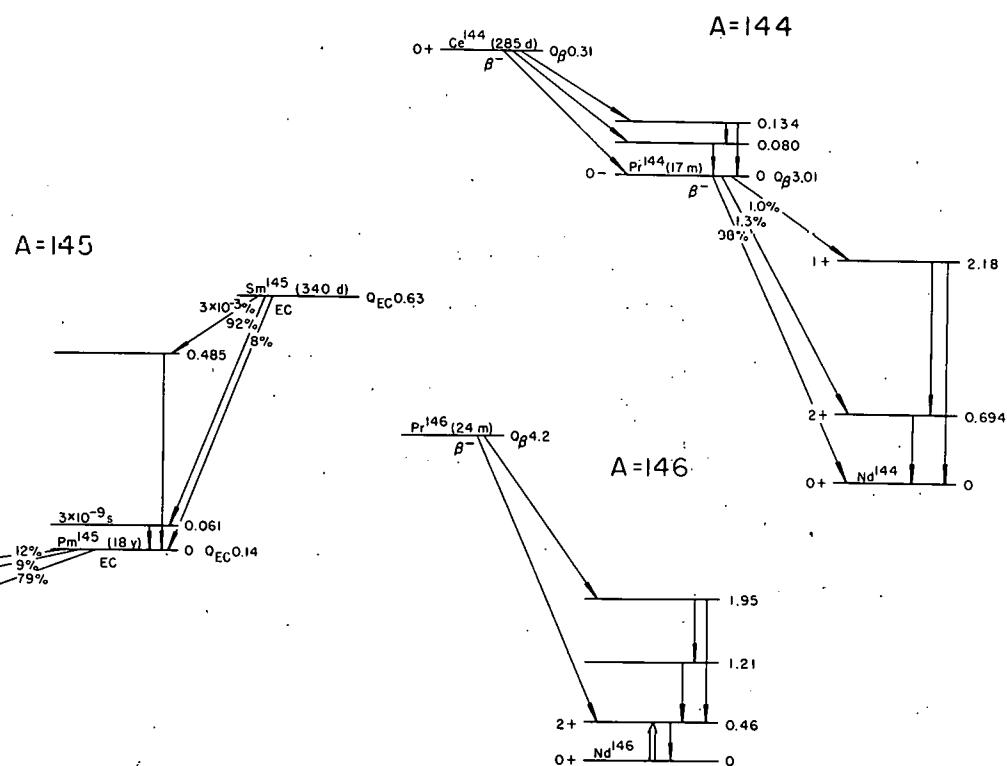
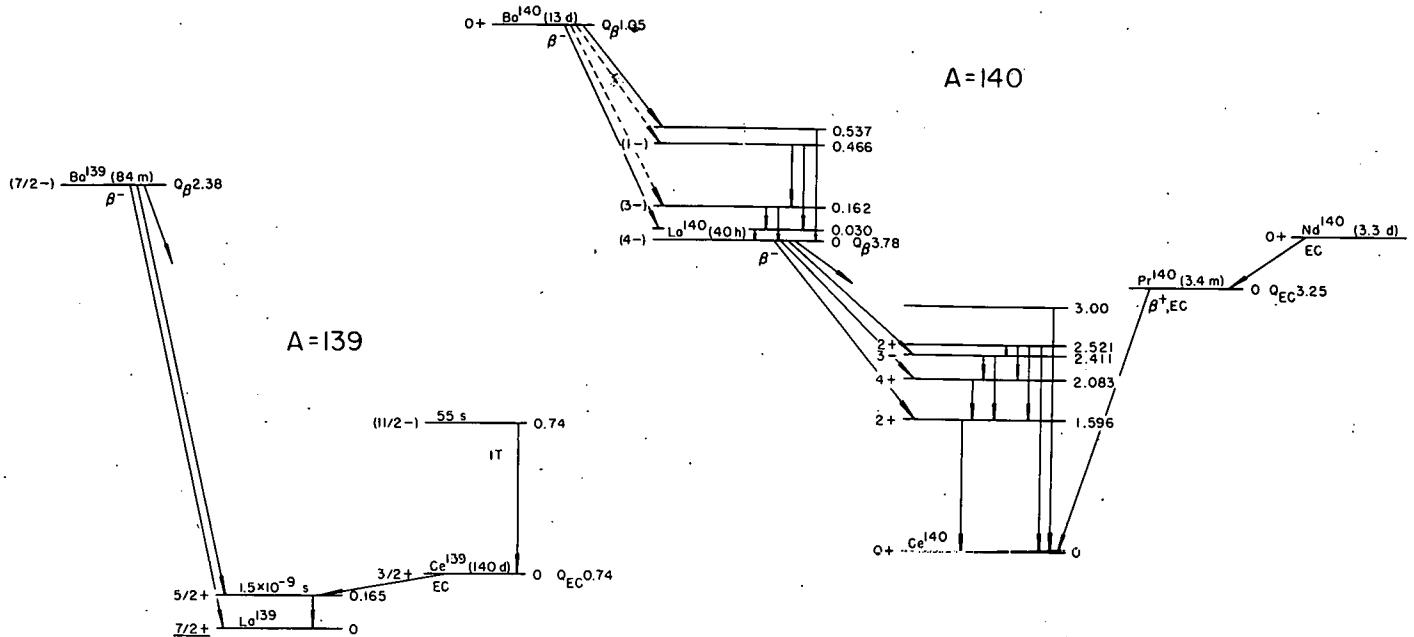


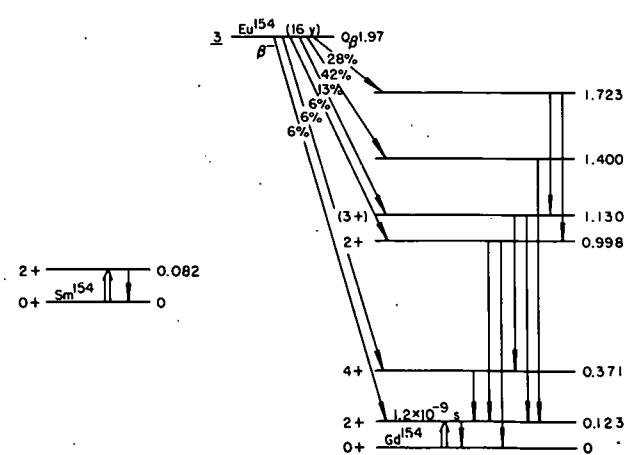
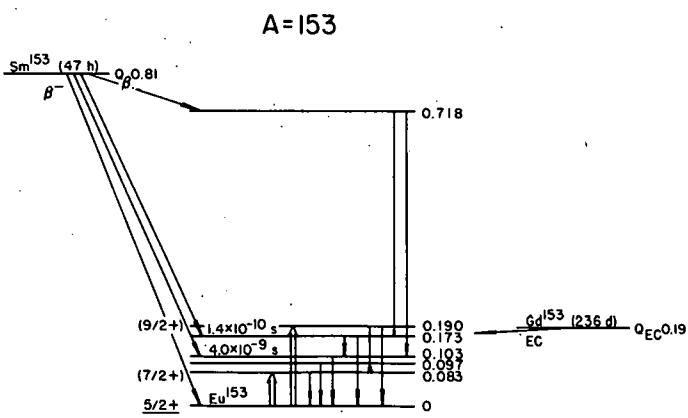
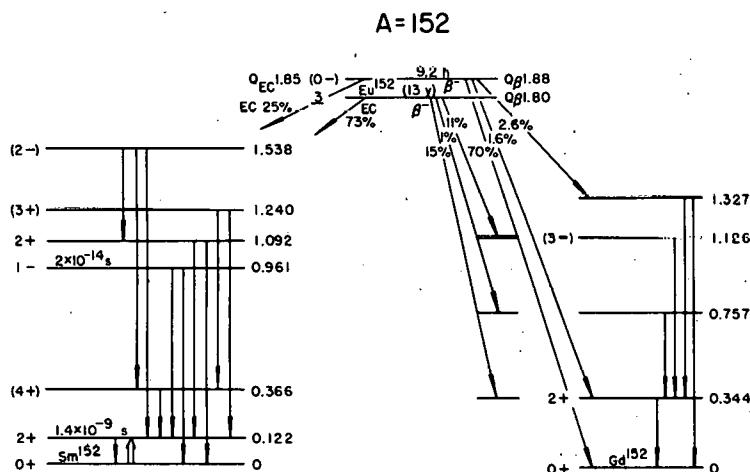
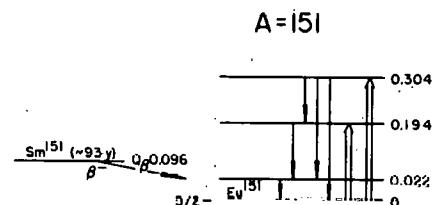
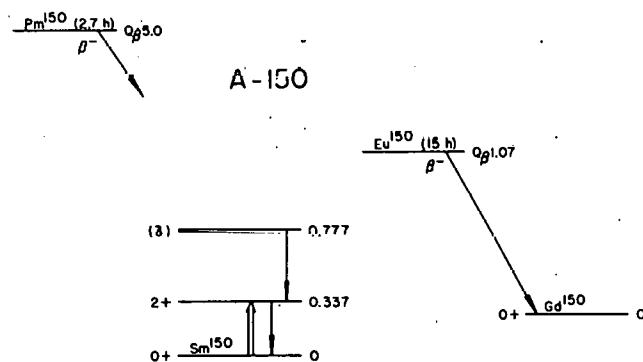
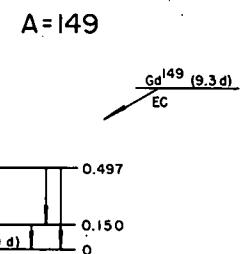
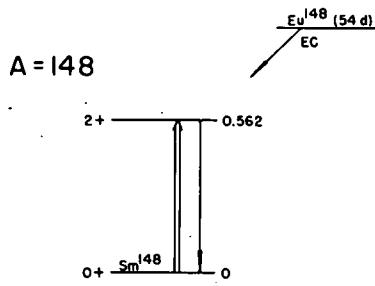
A = 143



A = 142



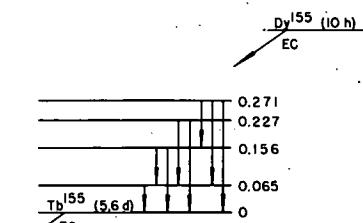
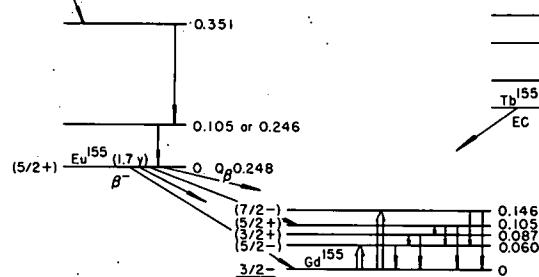




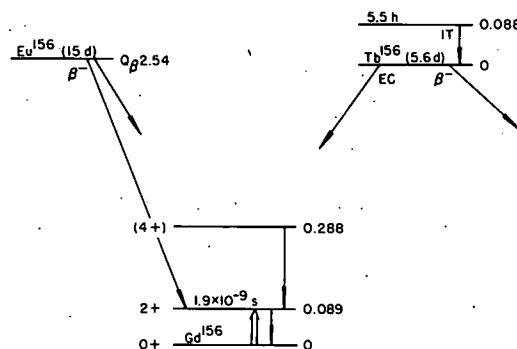
Sm^{155} (23 m) $0_{\beta}^{+} 2.2$

β^-

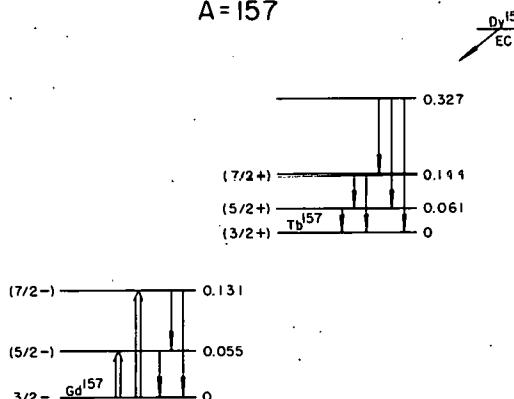
A=155



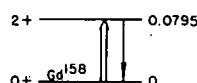
A=156



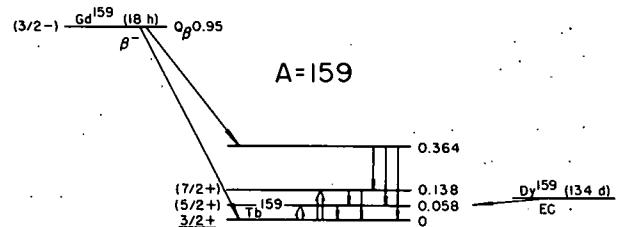
A=157



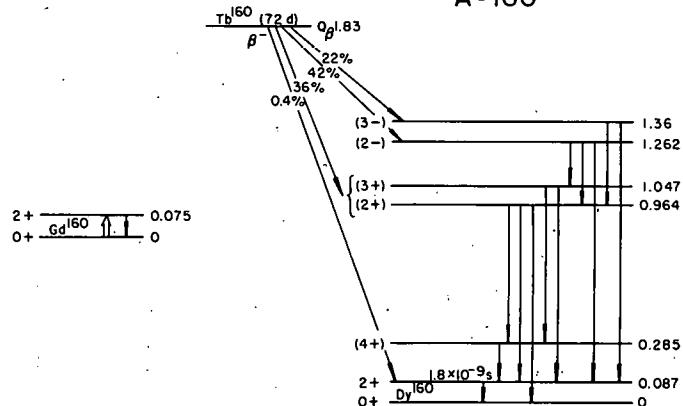
A=158

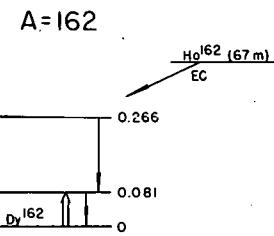
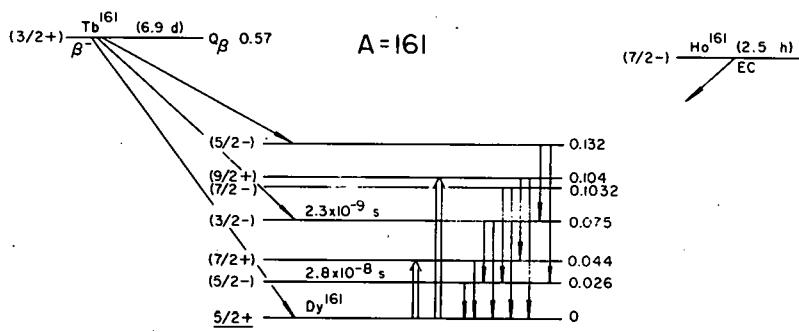


A=159

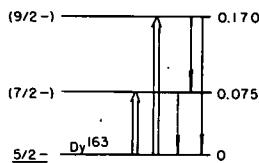


A=160

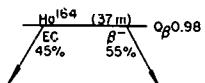




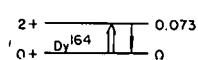
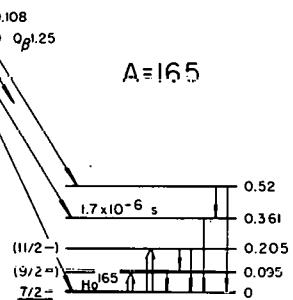
A = 163



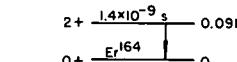
A = 164



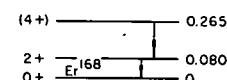
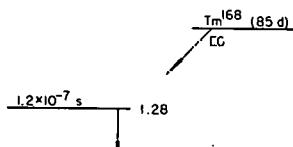
A = 165



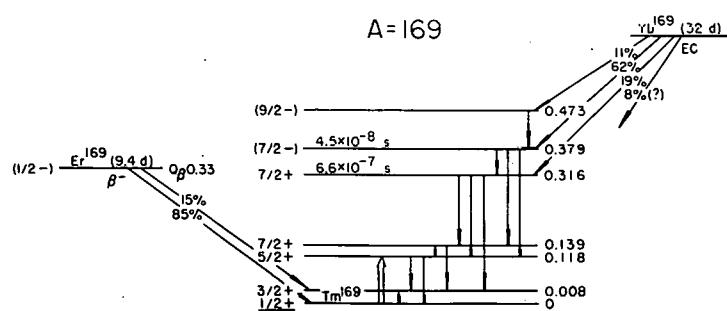
A = 166

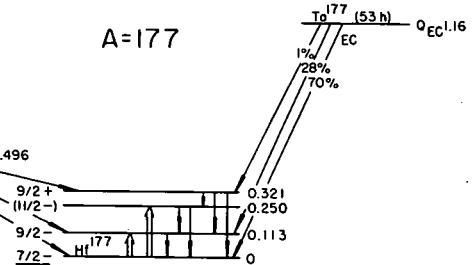
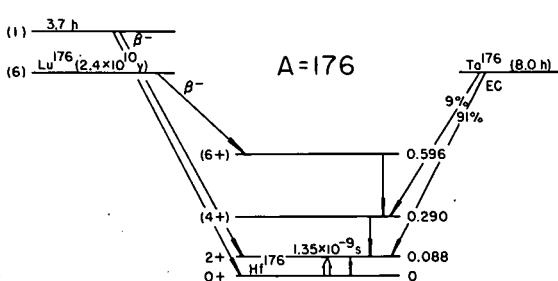
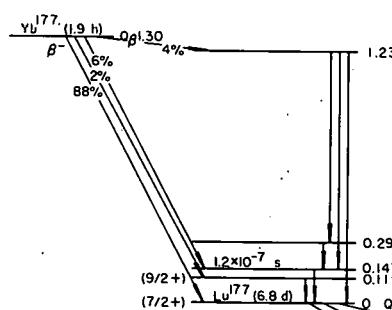
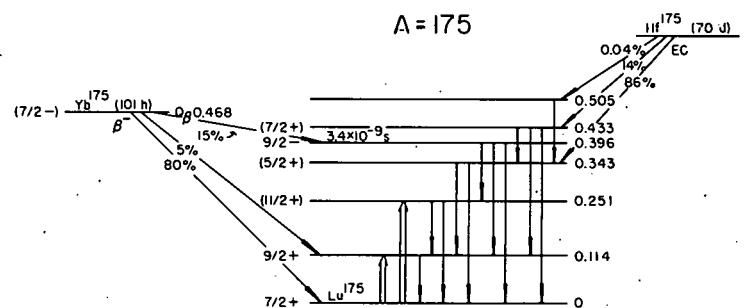
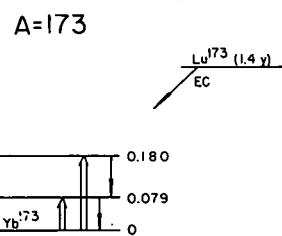
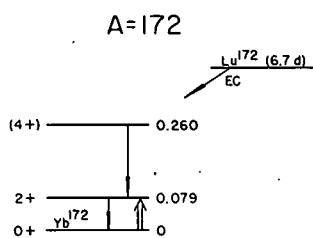
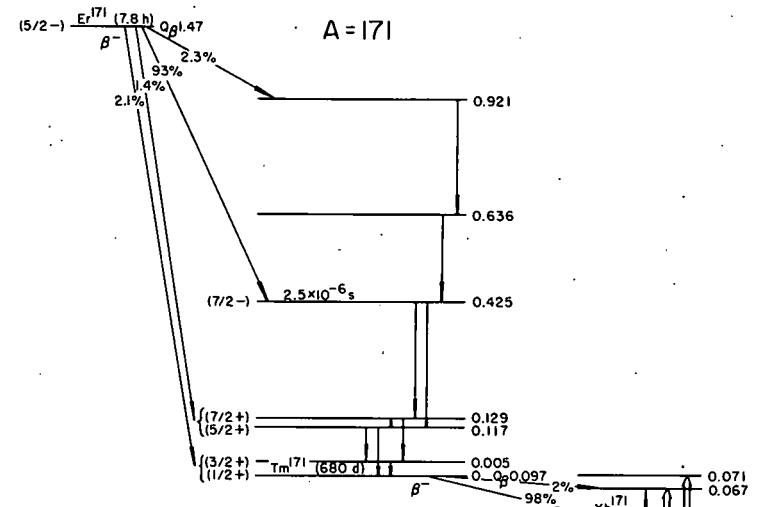
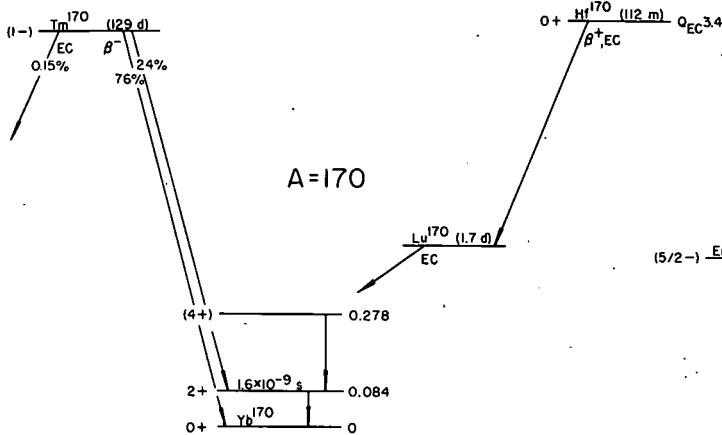


A = 168

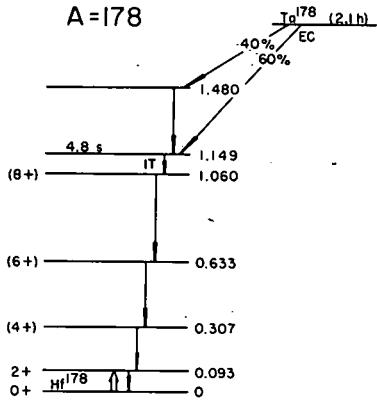


A = 169

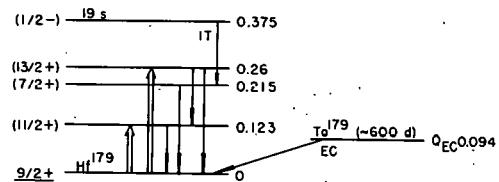




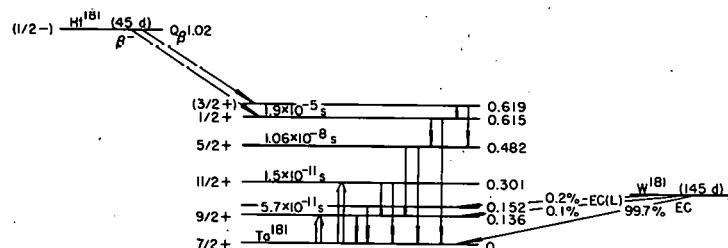
A=178



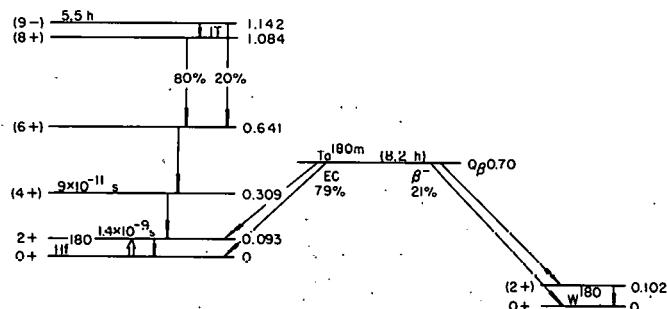
A=179



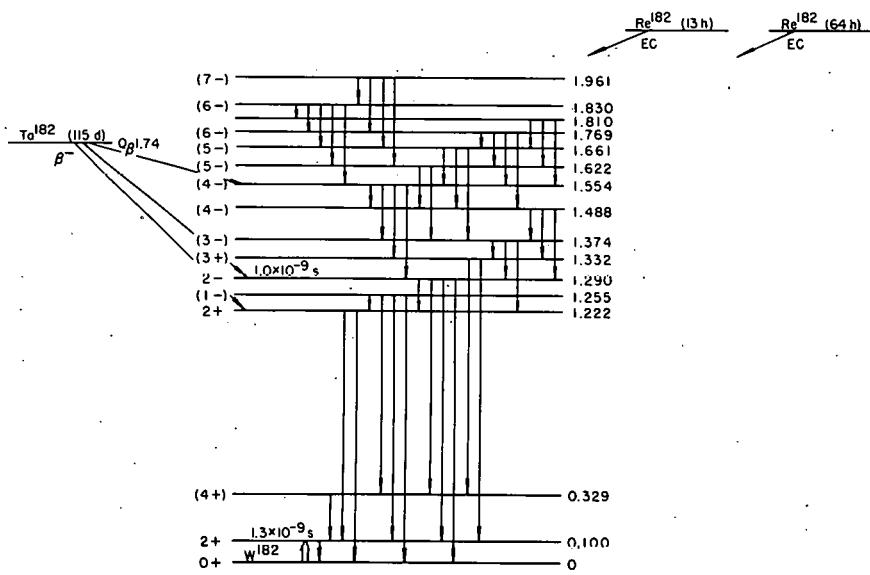
A=181

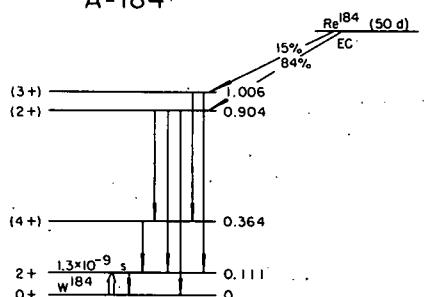
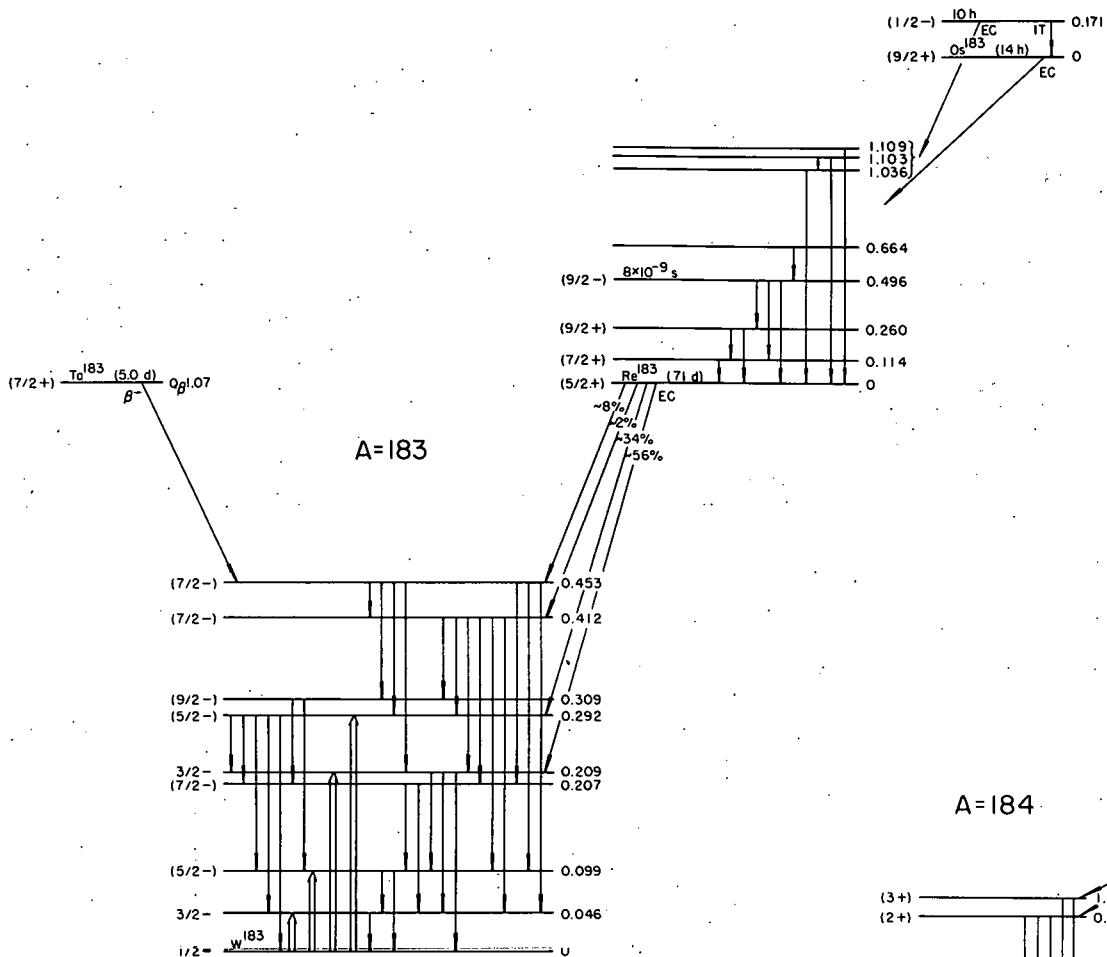


A=180

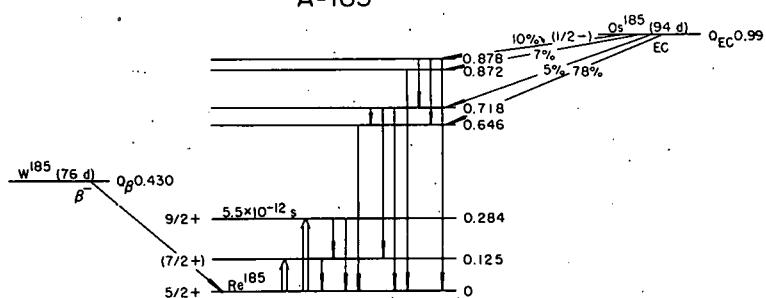


A=182

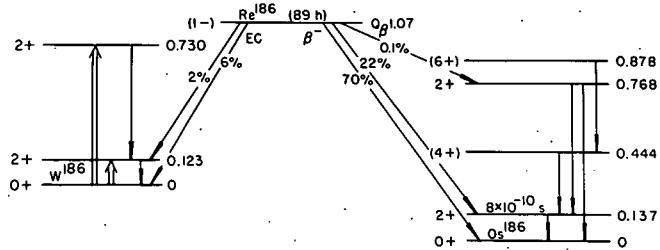




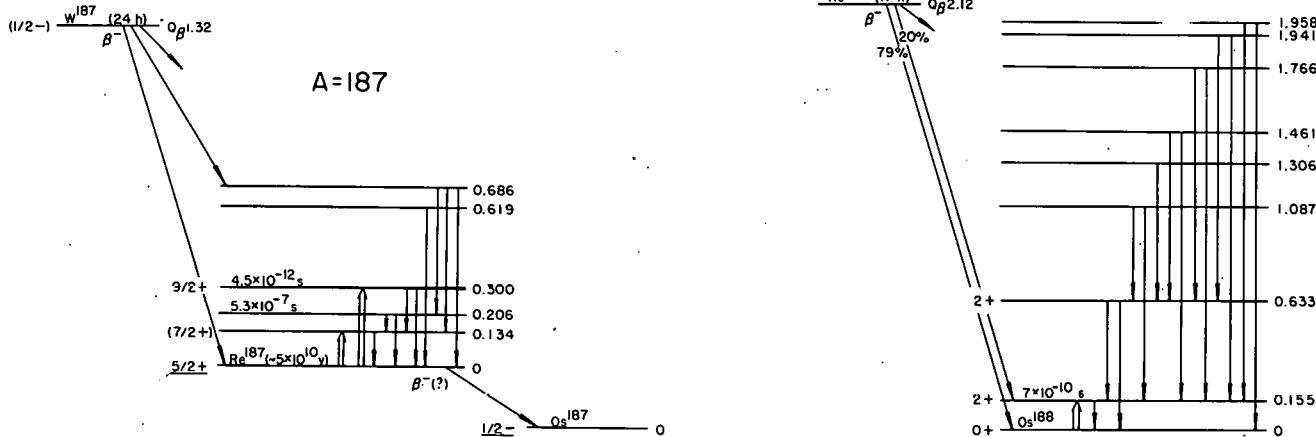
A=185



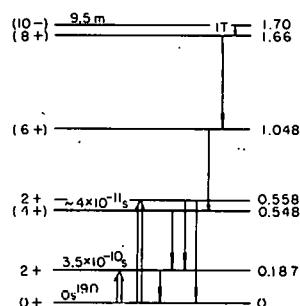
$$A = 186$$



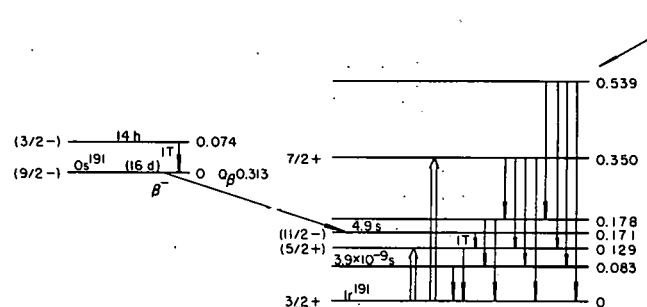
A=188



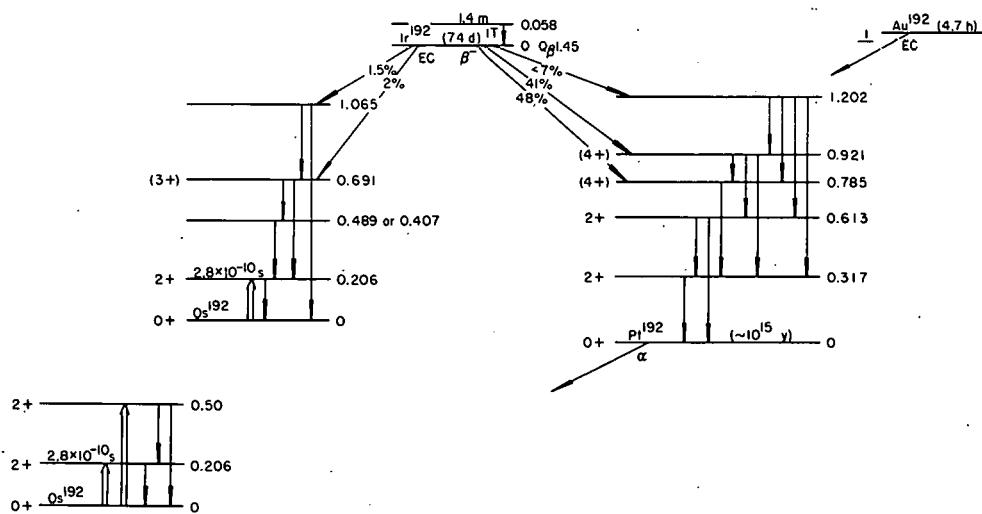
A=190

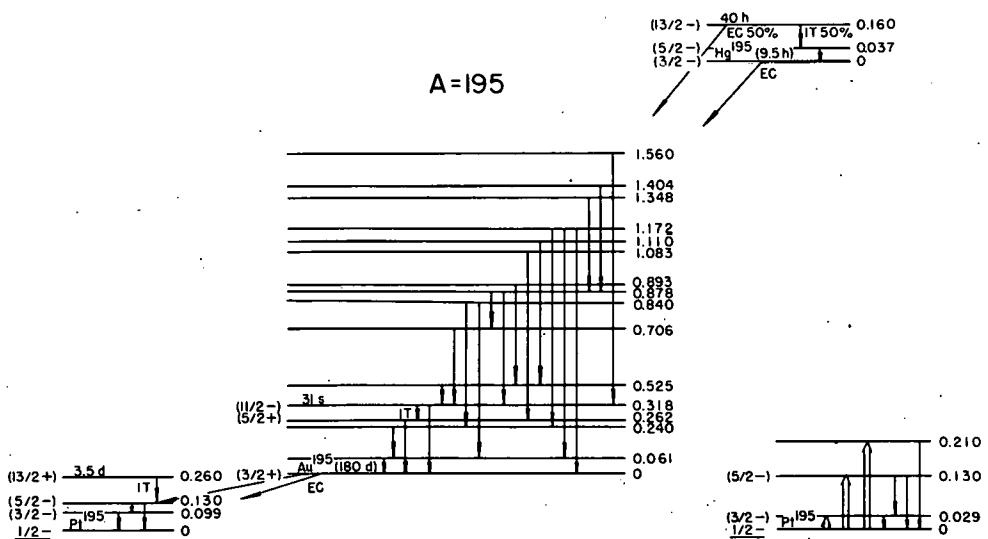
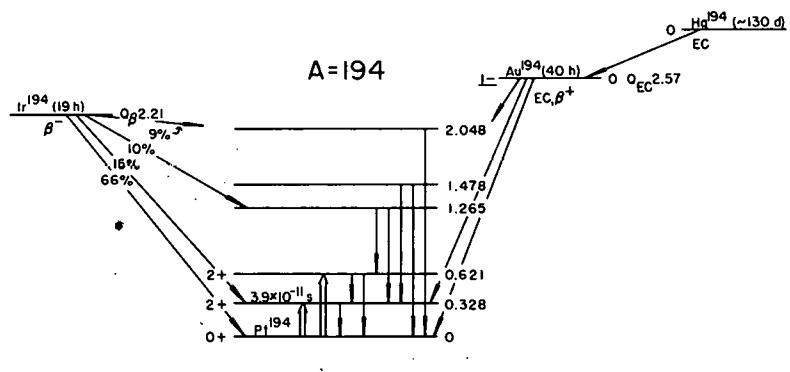
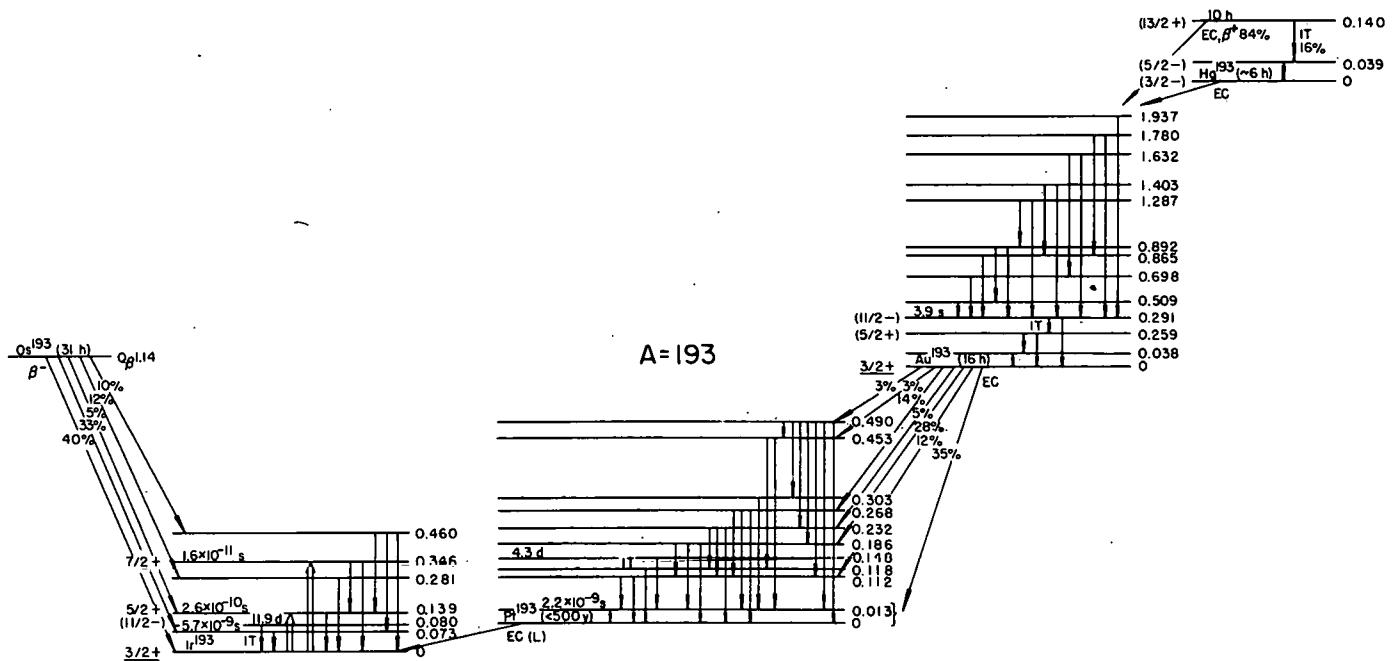


A=191

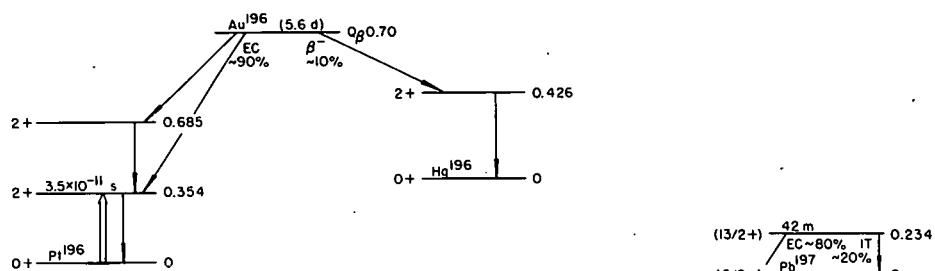


A=192

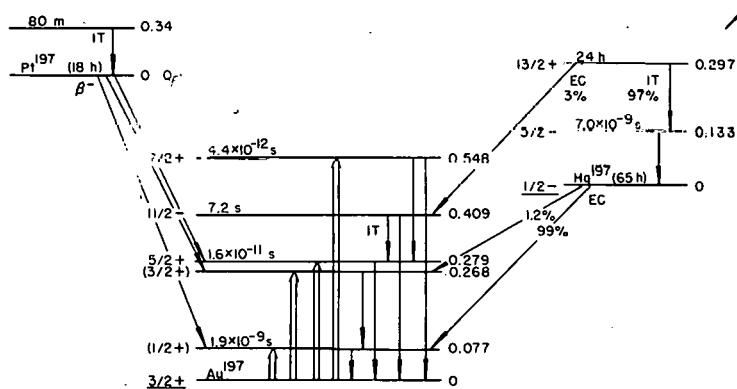




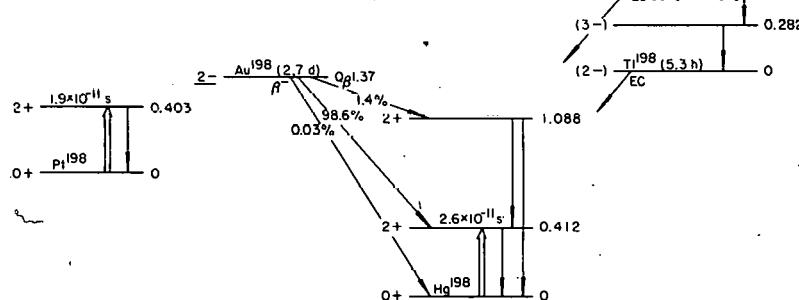
$$A = 196$$



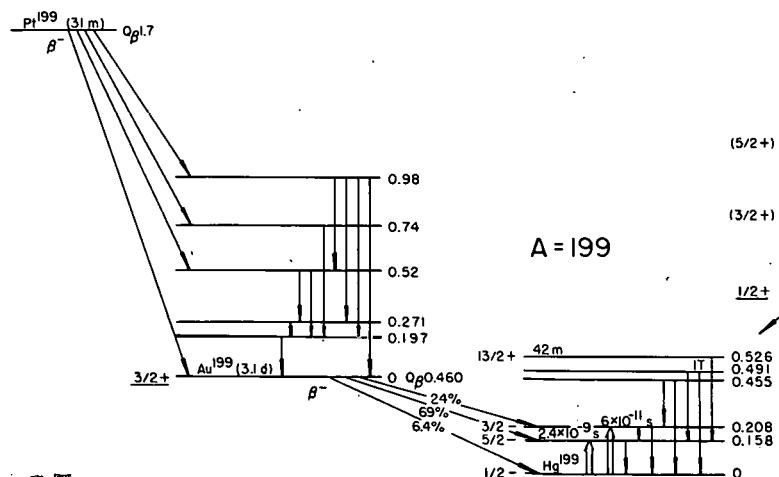
$$A=197$$

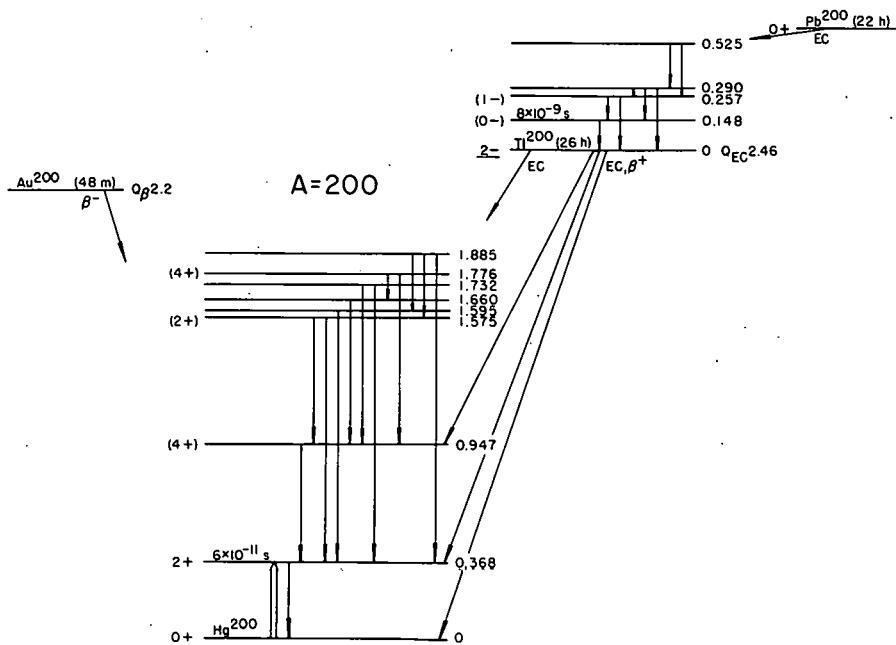


A = 198

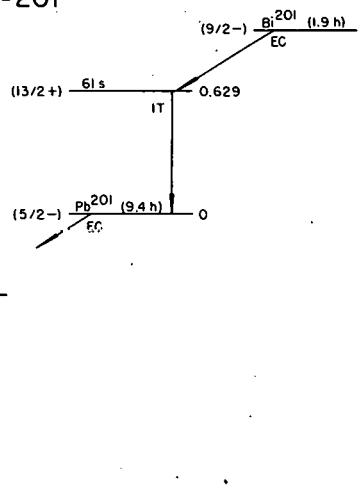


A = 199

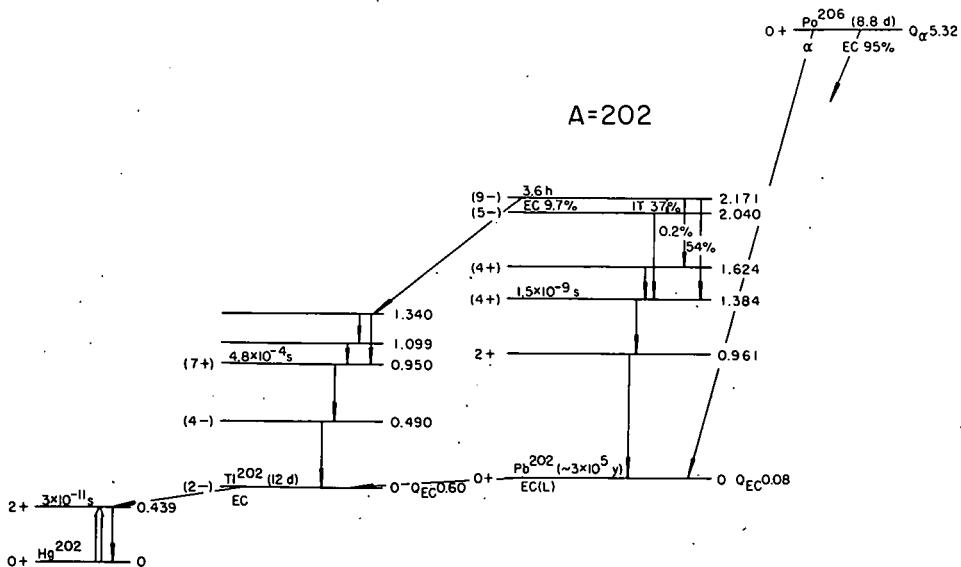




$A = 201$

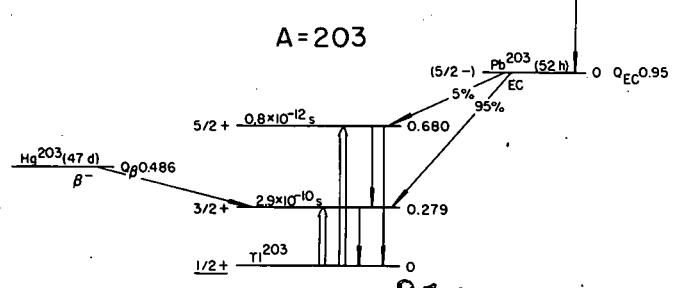


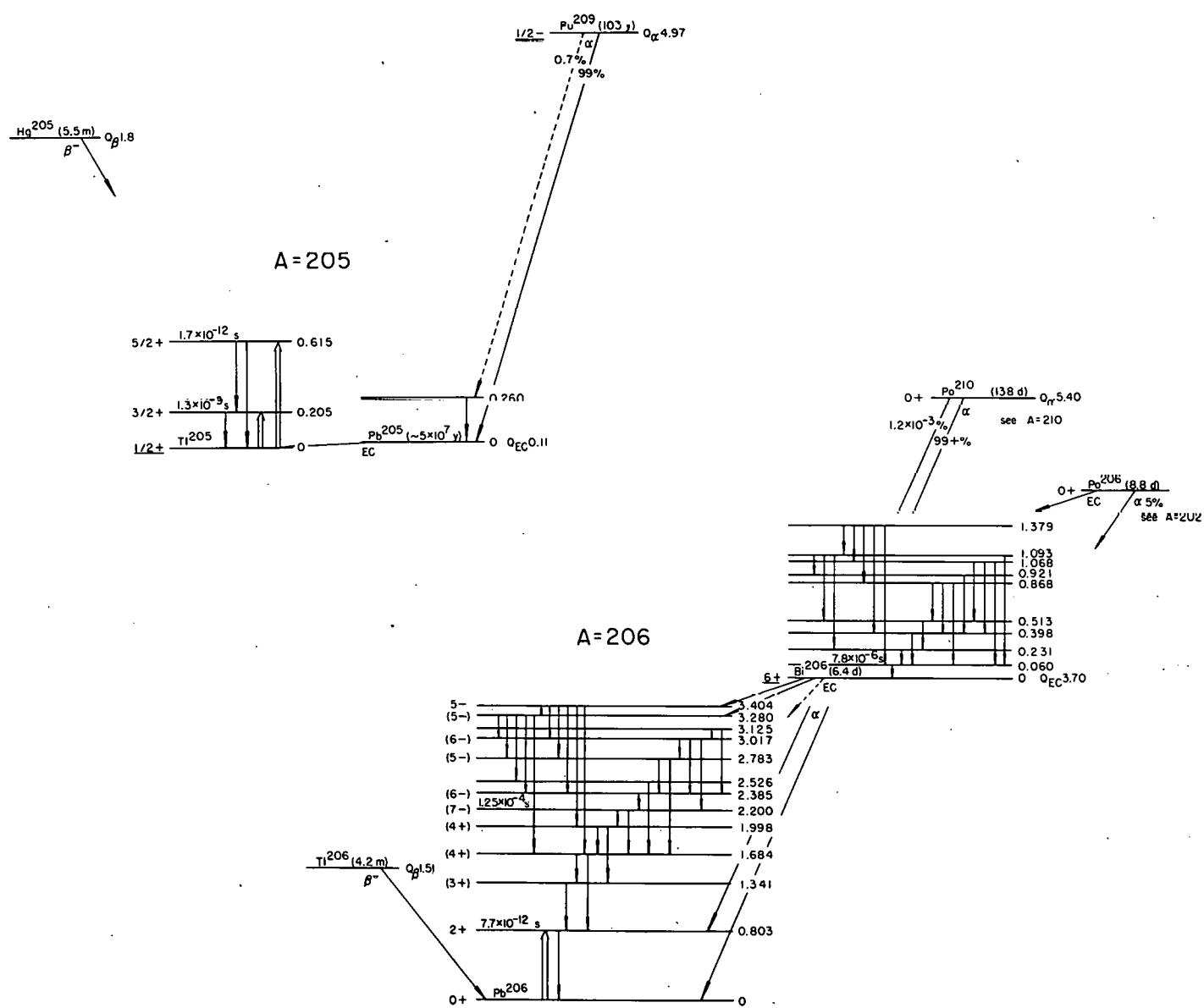
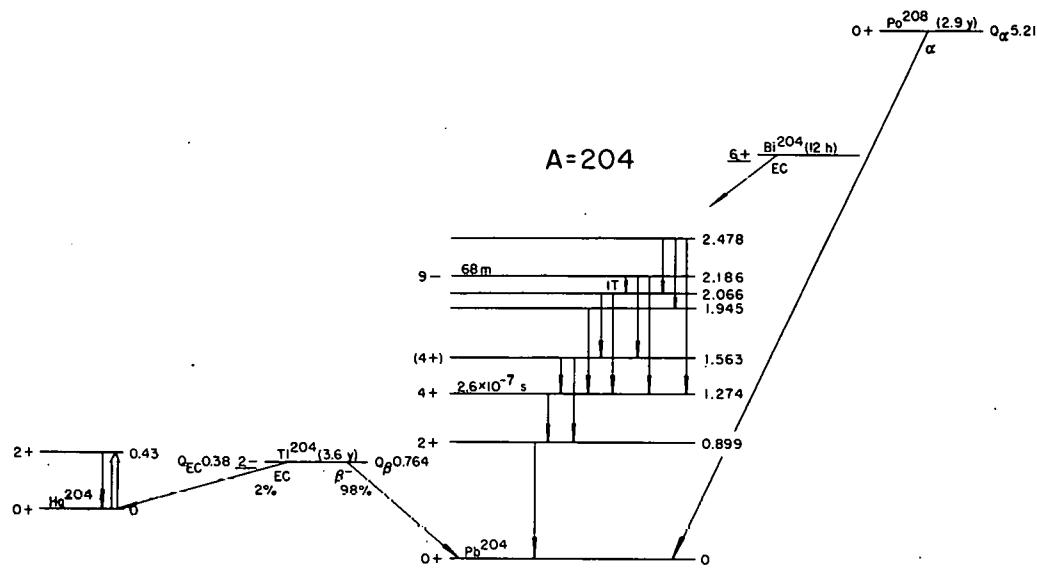
$A = 202$



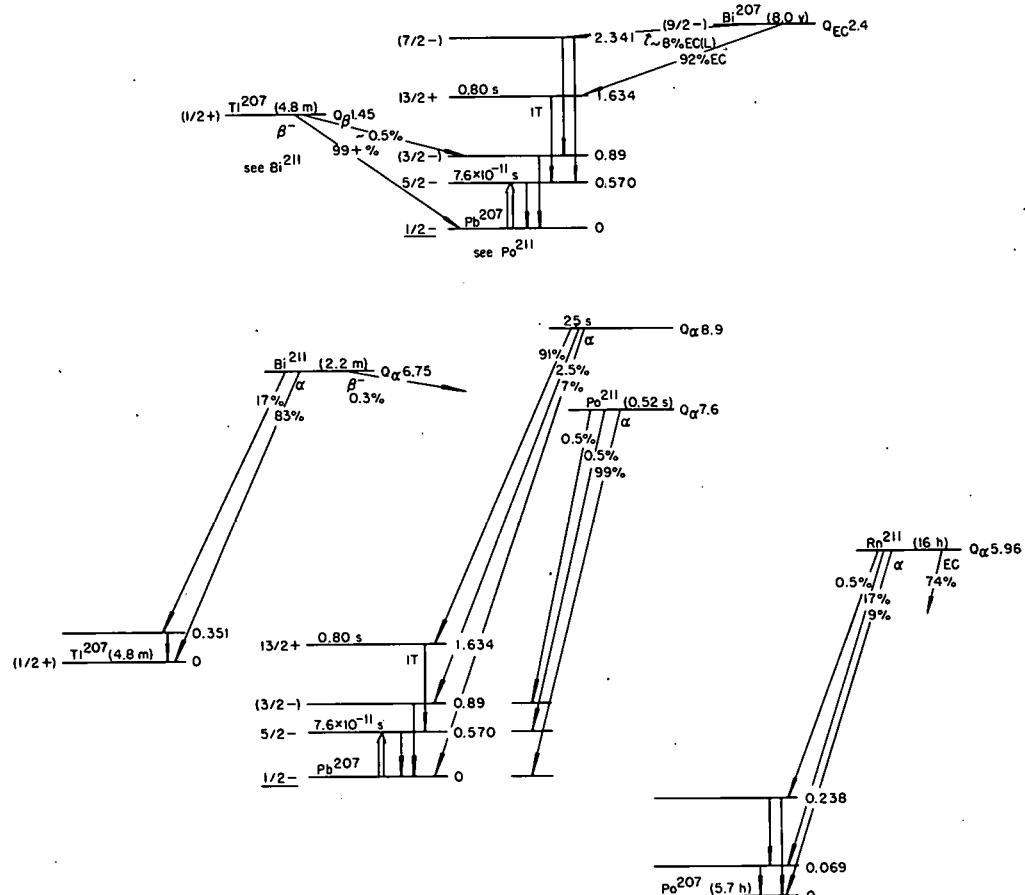
$(13/2^+) \xrightarrow{\beta^+} Pb^{202} (\sim 3 \times 10^5 \text{ y}) \xrightarrow{\beta^-} 0^+$

$A = 203$

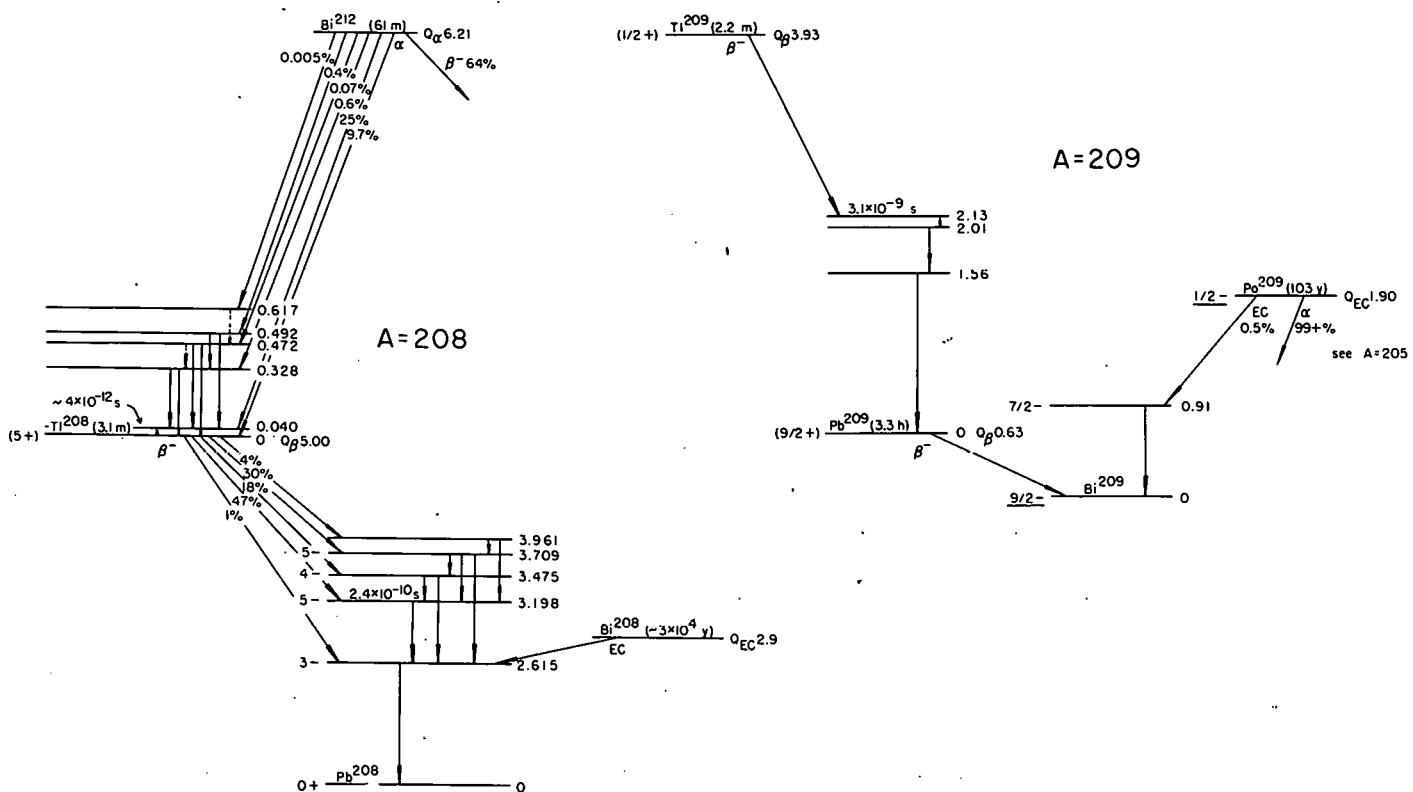




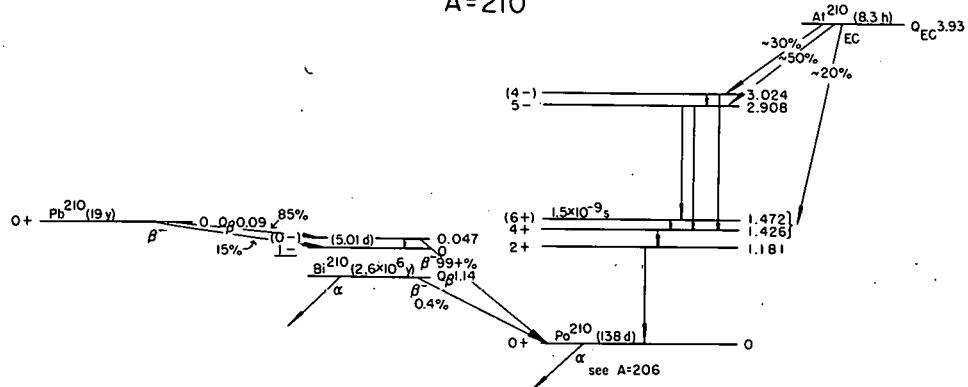
A=207



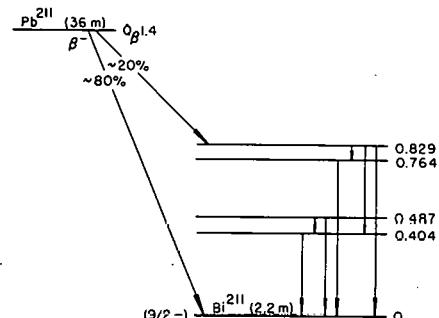
A=209



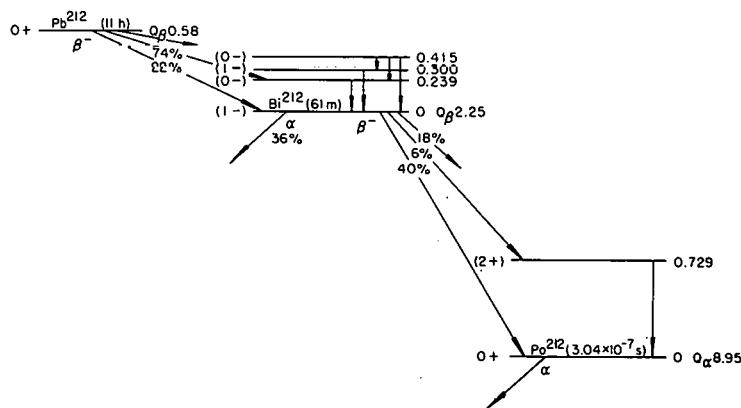
A=210



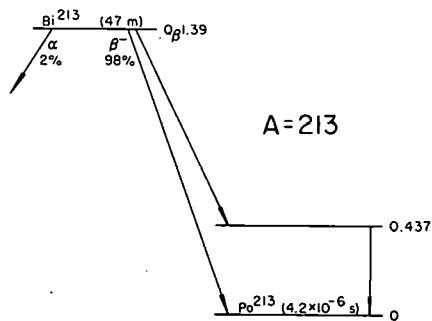
A=211



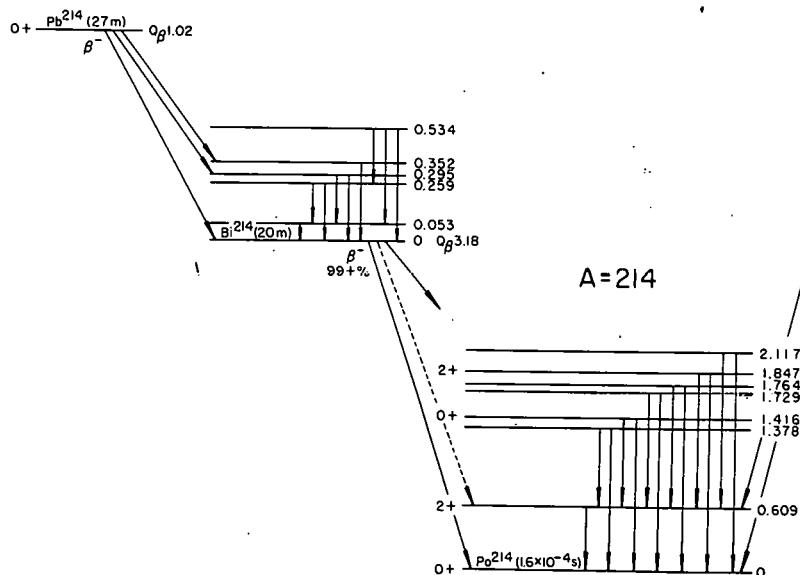
A=212



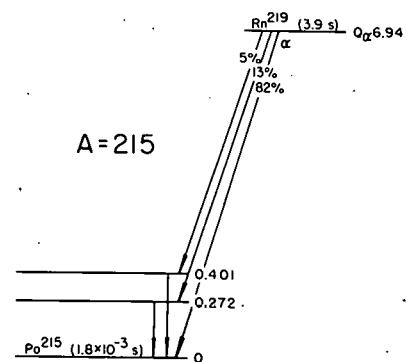
A=213

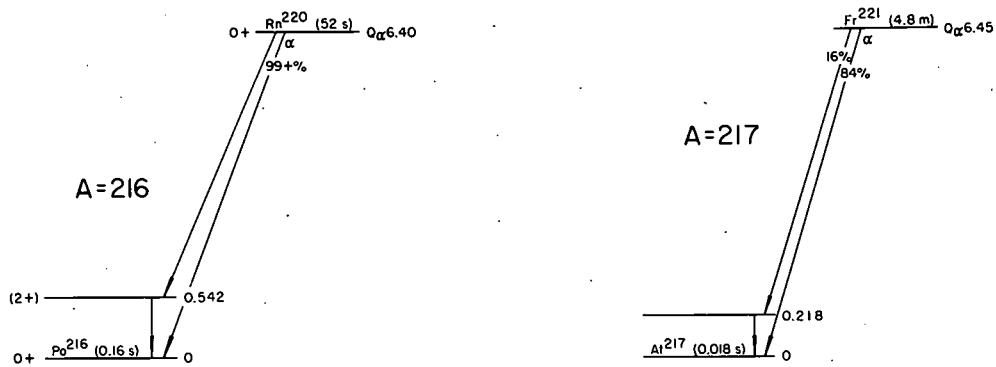


A=214

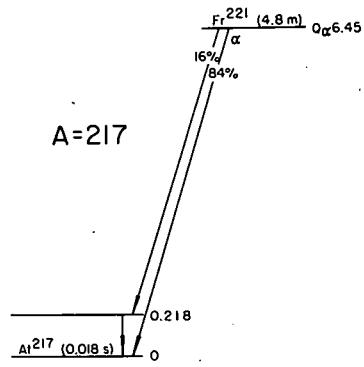


A=215

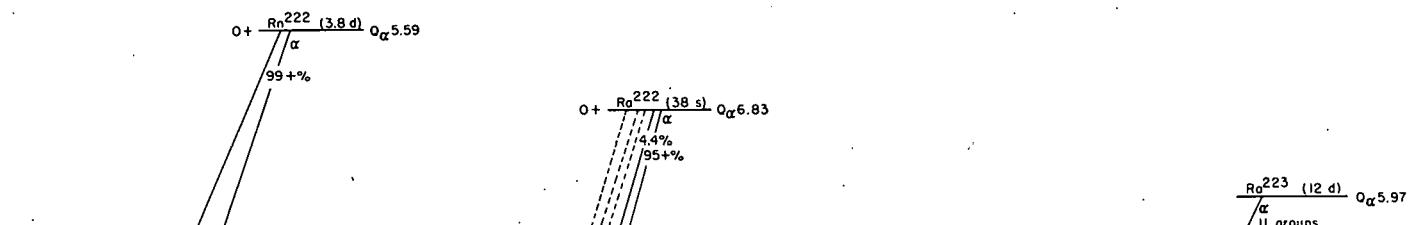




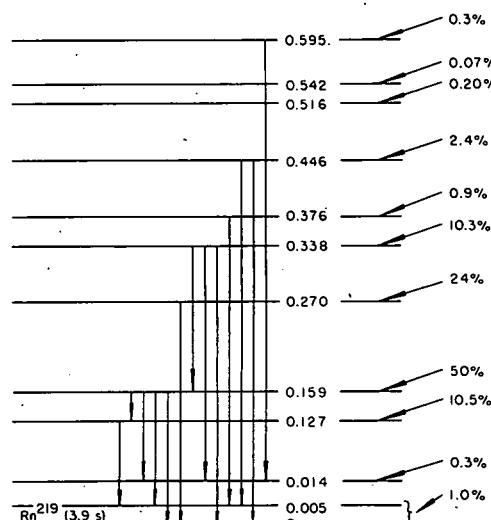
A = 217



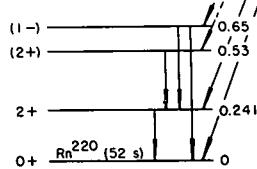
A = 218



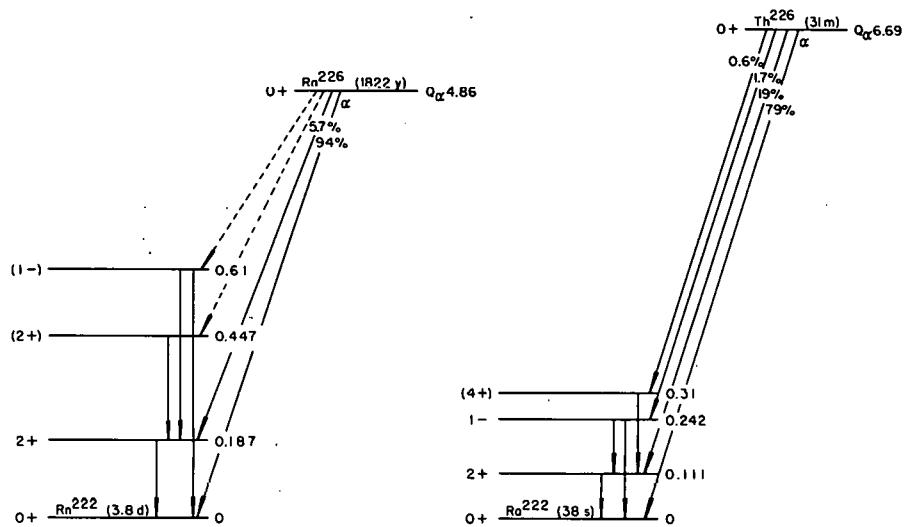
A = 219



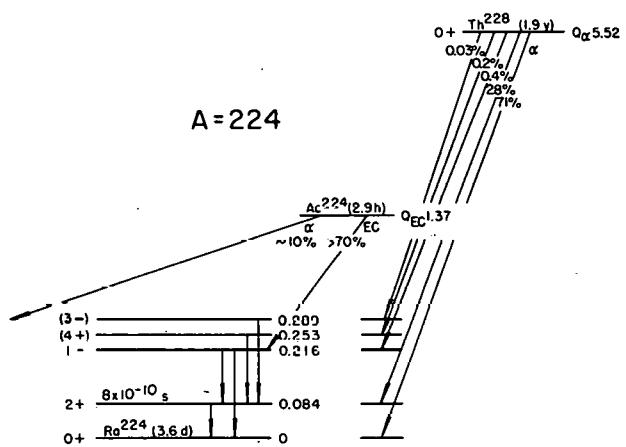
A = 220



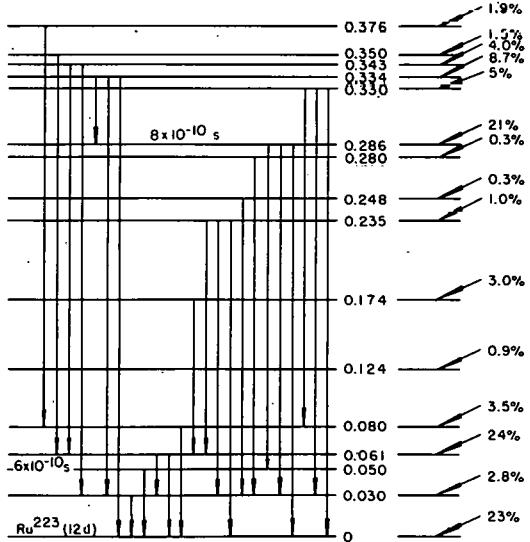
A=222



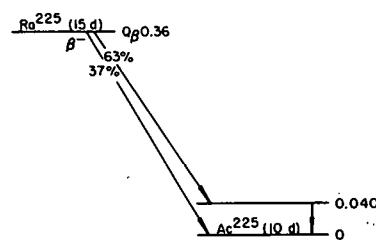
A=224



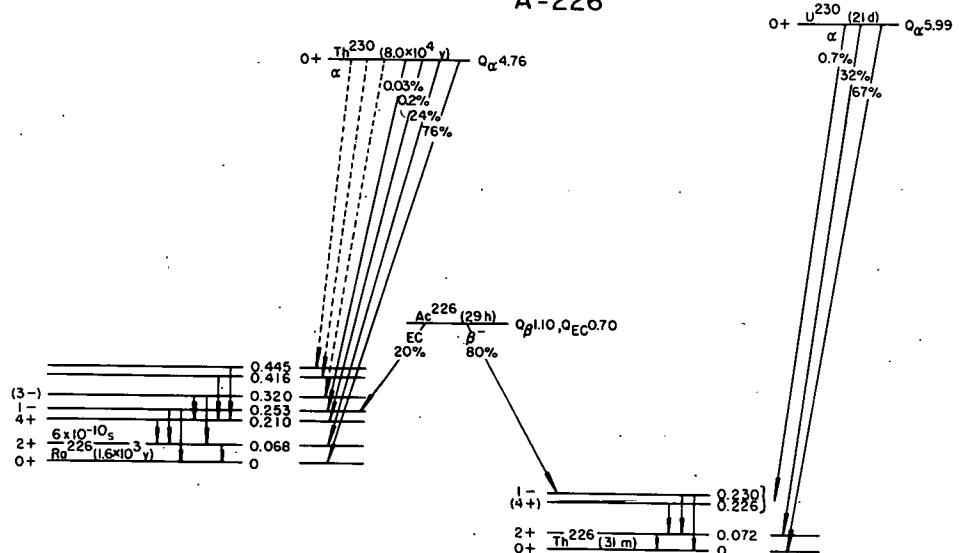
A=223



A=225

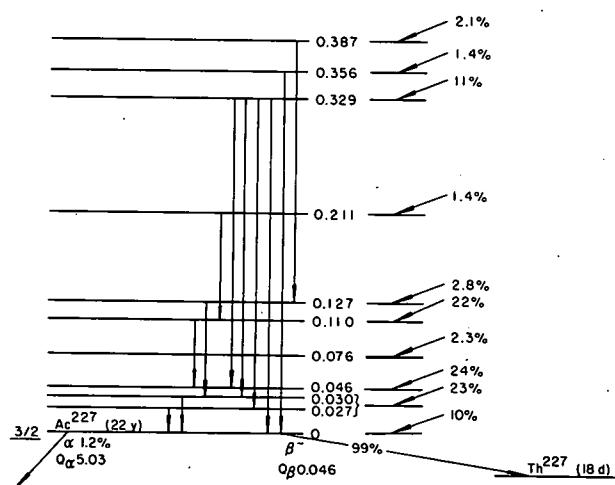


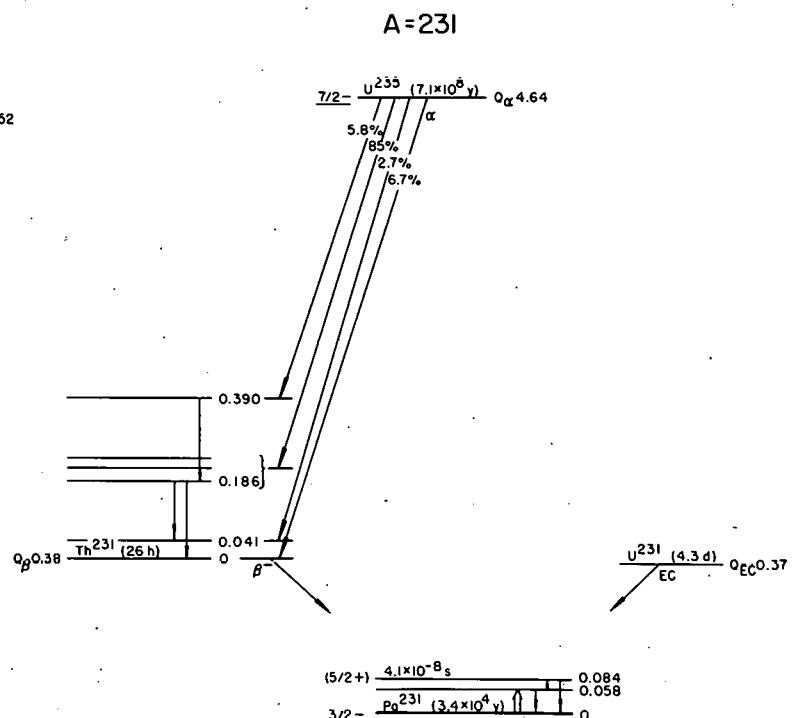
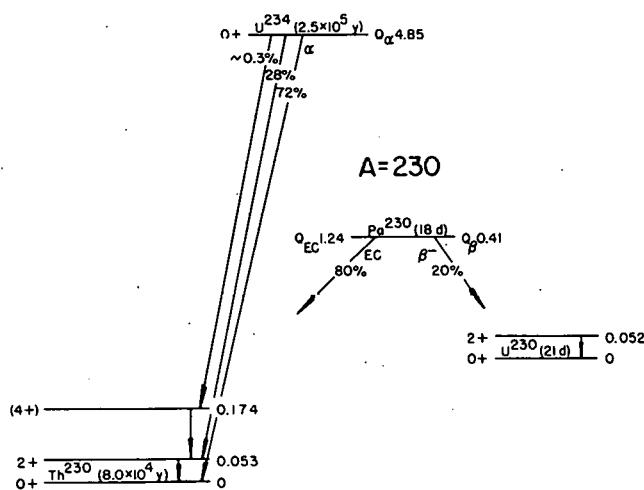
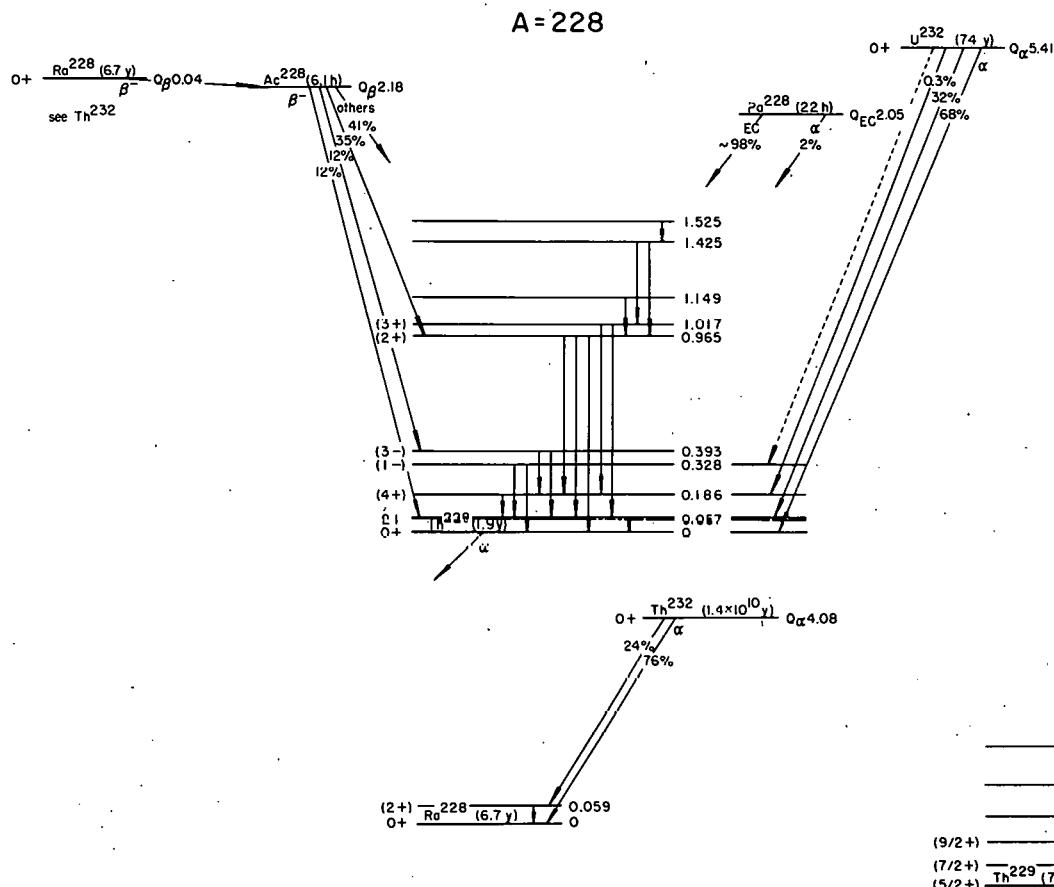
A=226



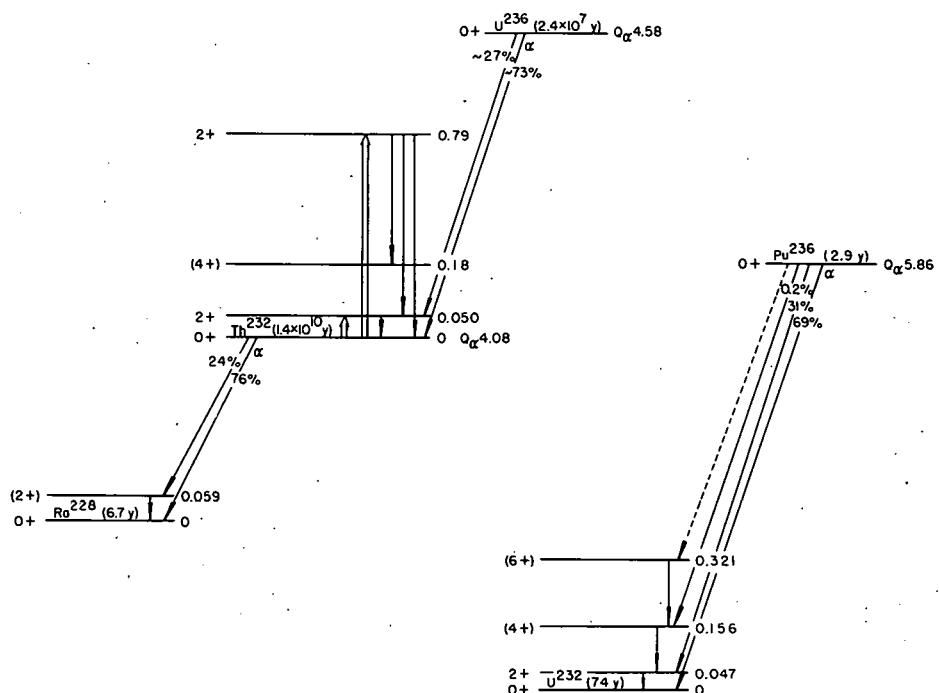
$3/2 - \text{Ra}^{231} (3.4 \times 10^4 \text{ y}) \quad 0\alpha 5.14$
10 groups

A=227

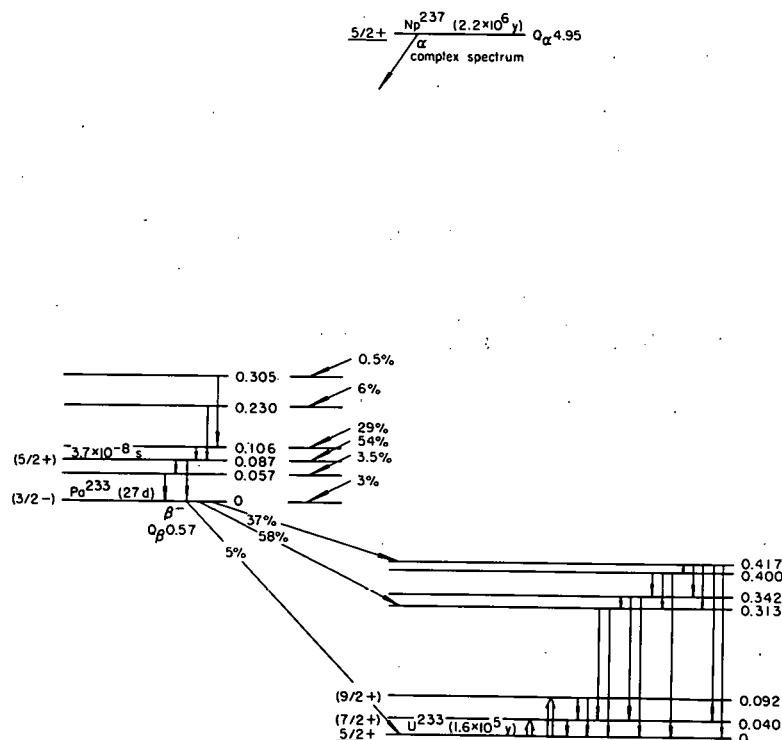




A=232



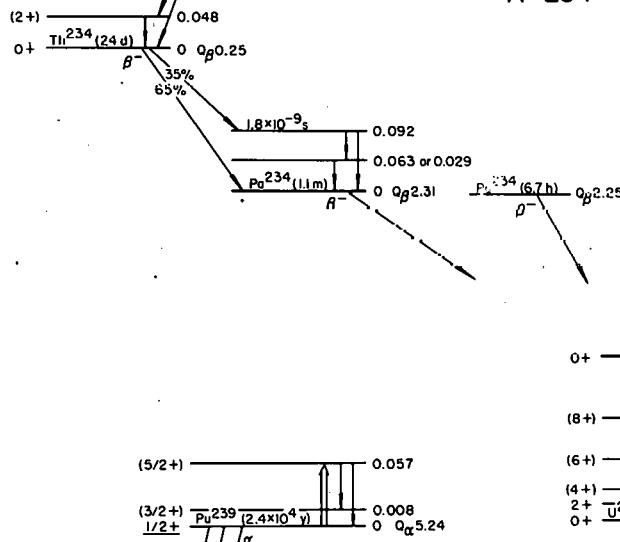
A=233



0+ $\xrightarrow{\text{U}^{238} (4.5 \times 10^9 \text{ y})}$ $\xrightarrow{\alpha}$ 0 α 4.27

23%
77%

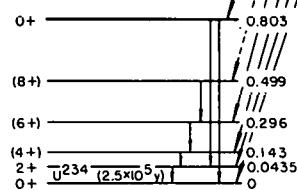
A = 234



0+ $\xrightarrow{\text{Pu}^{238} (86 \text{ y})}$ $\xrightarrow{\alpha}$ 0 α 5.59

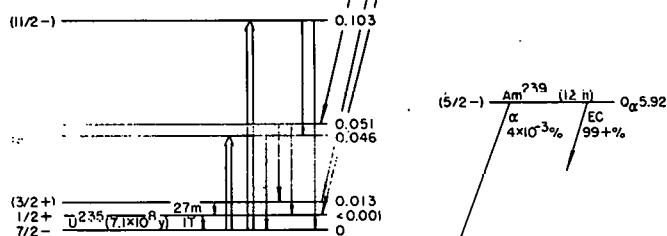
0.005%
0.1%
28%
72%

$\xrightarrow{\text{Pu}^{234} (4.4 \text{ d})}$ 0 EC 1.8



$\xrightarrow{\text{Pu}^{235} (24 \text{ m})}$ 0 β 1.4

A = 235



$\xrightarrow{\text{Np}^{235} (410 \text{ d})}$ 0 EC 0.21

see Am^{235}

$\xrightarrow{\text{Np}^{235} (410 \text{ d})}$ 0

0.048

0+ $\xrightarrow{\text{Pu}^{240} (6.6 \times 10^3 \text{ y})}$ $\xrightarrow{\alpha}$ 0 α 5.25

3 $\times 10^{-3} \text{ y}$

0.1%

24%

76%

A = 236

$\xrightarrow{\text{Np}^{236} (22 \text{ h})}$ 0 β 0.52, 0 EC 0.91

~9%
~34%
~45%

~12%

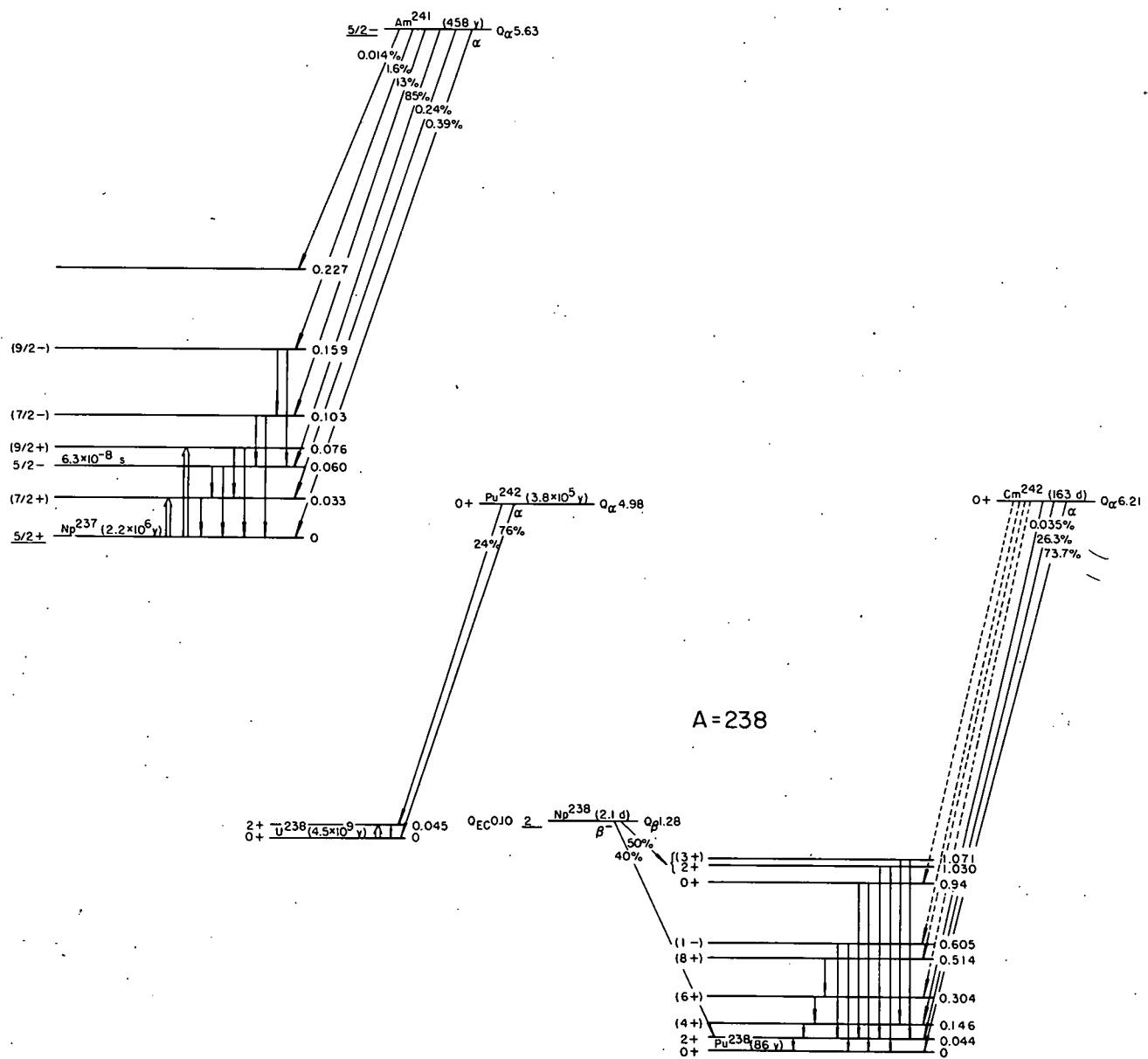
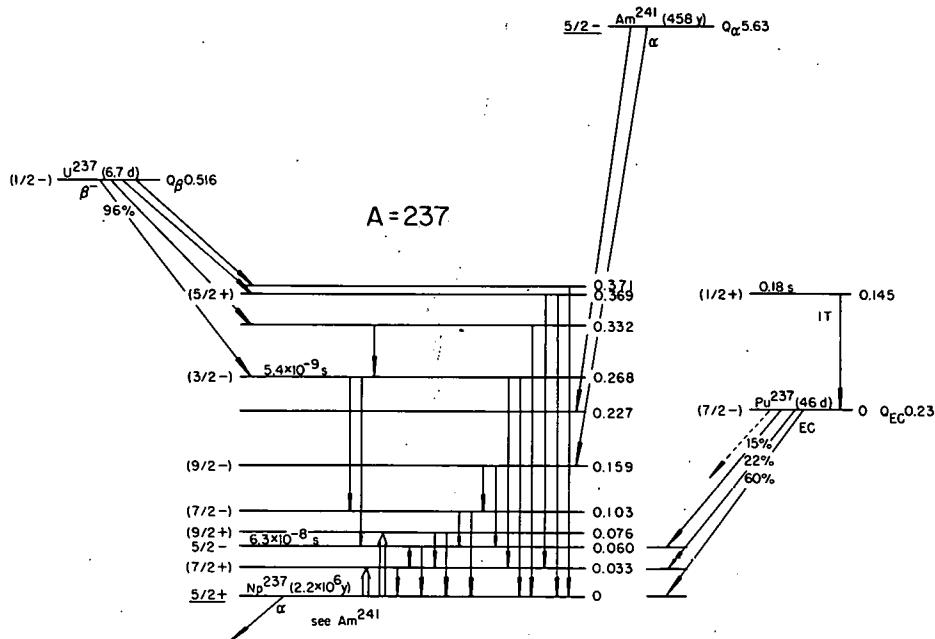
~45%

2+ $\xrightarrow{\text{Pu}^{236} (2.8 \text{ y})}$ 0

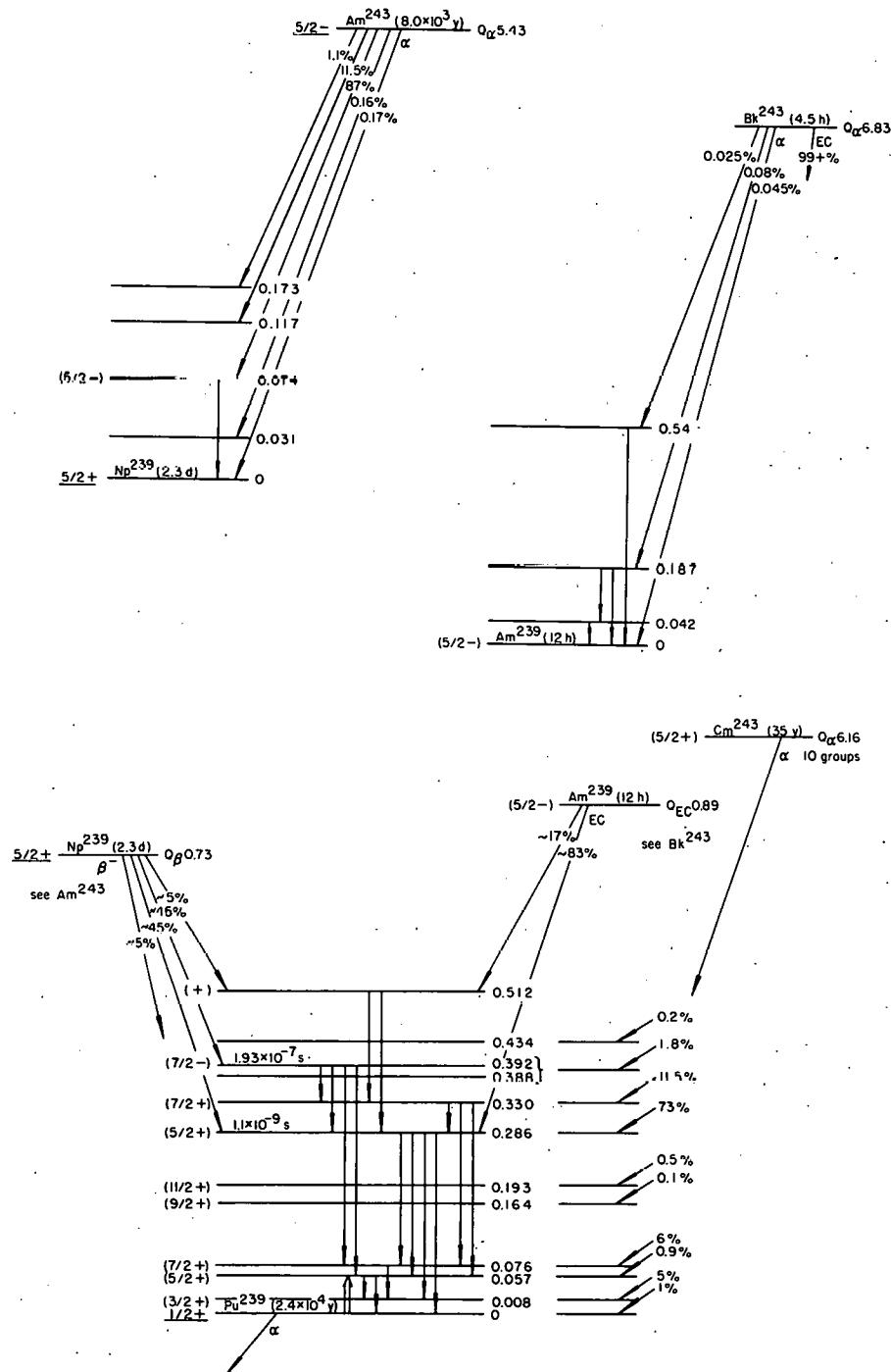
0.045

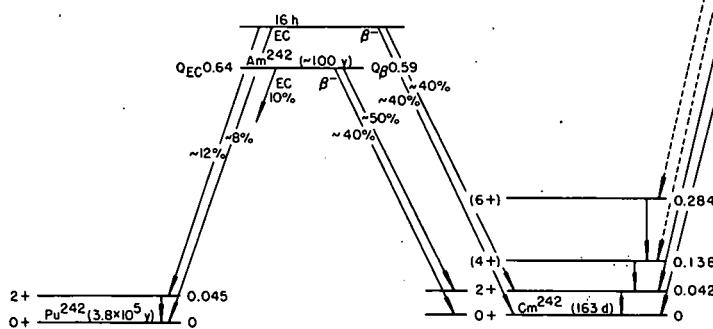
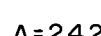
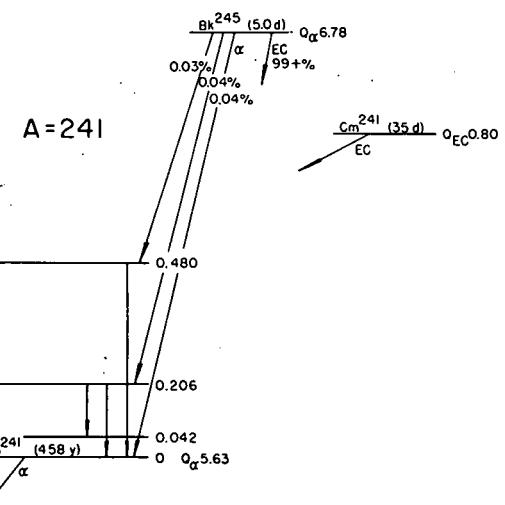
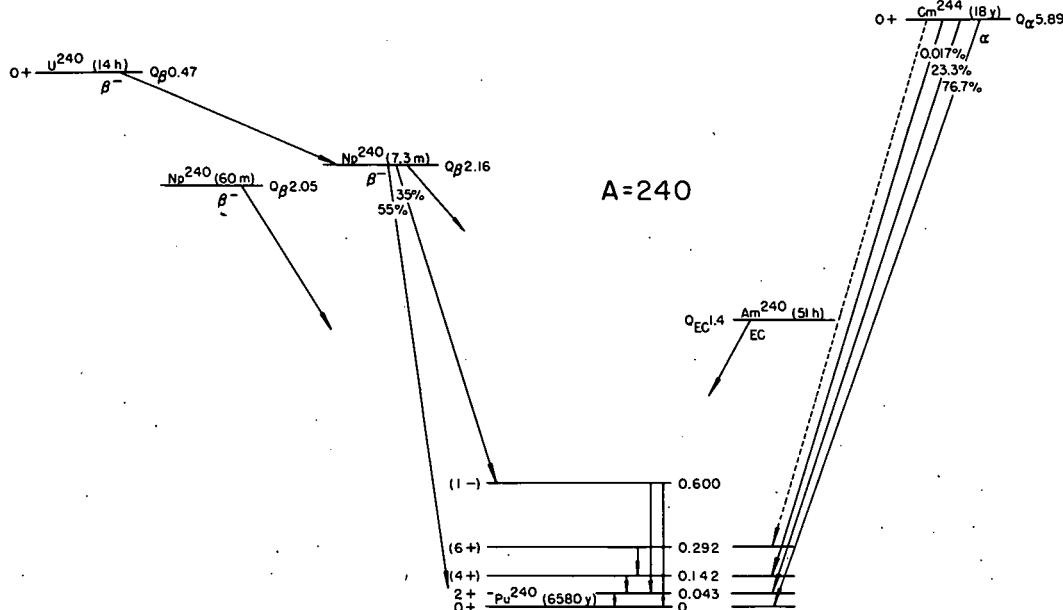
2+ $\xrightarrow{\text{U}^{236} (2.4 \times 10^7 \text{ y})}$ 0

0.045

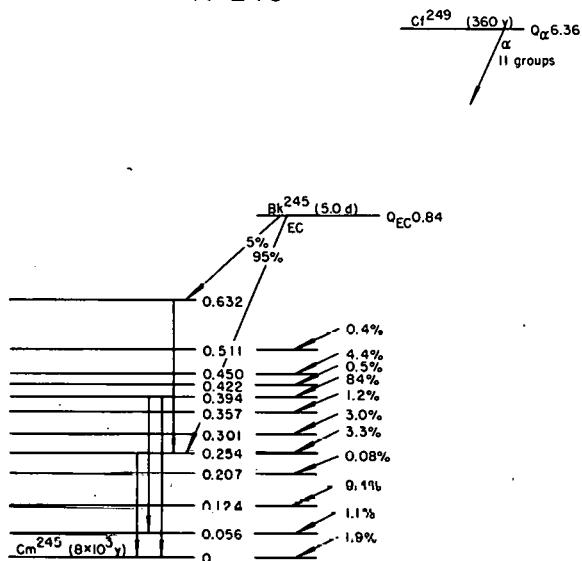


A = 239

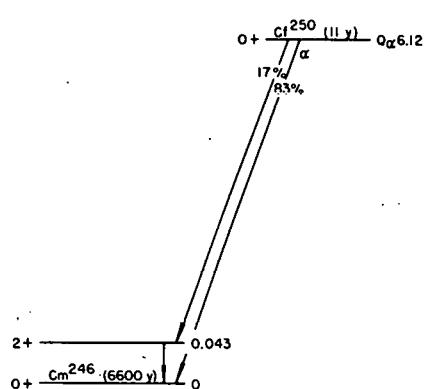




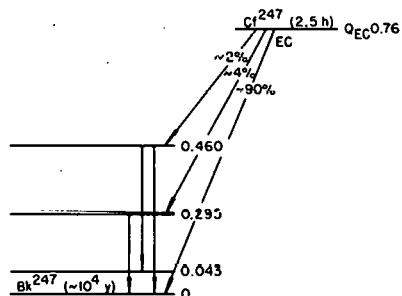
A=245



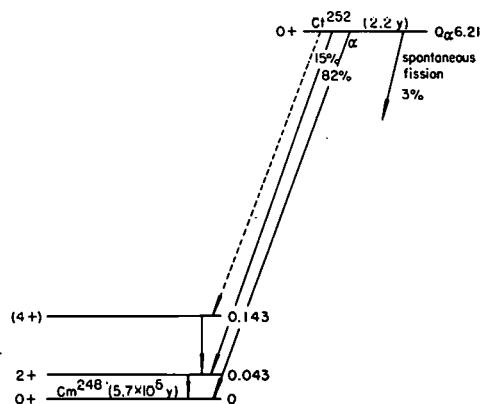
A=246



A=247



A=248



A=250

