

CONF-750913--15

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FOR PRESENTATION AT

4th International Transplutonium Element Symposium

Baden-Baden, Germany

September 10-13, 1975

MASTER

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ARGONNE NATIONAL LABORATORY, ARGONNE, ILLINOIS

operated under contract W-31-109-Eng-38 for the
U. S. ENERGY RESEARCH AND DEVELOPMENT ADMINISTRATION

CRYSTAL STRUCTURE AND MAGNETIC PROPERTIES OF AMERICIUM LAVES PHASES*

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ABSTRACT

We have investigated a series of binary americium alloys with AB₂ stoichiometry (B = Al, Co, Fe, Rh and Ru) prepared from the ²⁴³Am isotope. The C-15 (MgCu₂-type) structure appears in AmAl₂, AmCo₂, AmFe₂ and AmRh₂, whereas AmRu₂ is isostructural with AmOs₂ which has the C-14 (MgZn₂-type) structure.

The magnetic properties of the cubic americium Laves phases have been studied by means of magnetization and nuclear gamma-ray resonance (Mössbauer effect) measurements between 2.5 and 300 K. All the cubic Laves phases studied, except AmFe₂, exhibit almost temperature independent paramagnetism. The AmFe₂ sample is ferromagnetic with an estimated Curie temperature of ~400 K. The hyperfine field at the Am site in all the compounds is small. The bulk magnetic moment of AmFe₂ (~3 μ_B /F.U.) is thus associated only with the Fe atoms. The magnetic properties of the compounds are consistent with the assumption that the Am ion is in the +3 state(5f⁶).

I. INTRODUCTION

The importance of Laves phases to the alloy chemistry of intermetallic compounds arises from their widespread occurrence (they form the largest known family of intermetallic compounds). Lam and Mitchell [1] showed that the Laves phases of the actinides Th, U, Np, and Pu obey the same general rules which govern the systematics of other Laves phases. In the present paper we report the existence of some new americium Laves phases and show that their alloy chemistry is consistent with the earlier results. In addition, we have studied the magnetic properties of some of these compounds by means of magnetization and NGR (nuclear gamma-ray resonance) experiments. Our interest in the magnetic properties arises from similar studies of the corresponding U, Np, and Pu Laves phases [2-5] that show a variety of magnetic behavior.

II. EXPERIMENTAL

The compounds were prepared by arc melting the requisite amounts of the constituent metals in an argon-helium atmosphere; the resultant buttons weighed ~0.2g. Chemical analysis of the ²⁴³Am metal used in this investigation (obtained from Hollifield National Laboratory) yielded the following impurities (in ppm by weight). Al 1000; B 0.5; Ca 500; Cr 30; Cu 40; Fe 300; K <0.2; Mg <1; Mn 10; Mo <0.3; Na <0.1; Ni 2; P <0.1; Pb 30; Si 50; Sr 4; Ti <1; V <0.1; Zn <0.2; Zr <0.2; Am²⁴¹ 100; Pu²⁴⁰ 1. The remaining materials used for sample preparation were zone-refined Al (99.999% pure) and high-purity cobalt (99.99%), iron (99.99%), rhodium (99.96%) and ruthenium (99.99%). Most of the samples were arc melted twice, homogenized at 1000°C for several hours and furnace cooled. X-ray powder diffraction patterns (Straumanis-type Debye-Scherrer camera, 114.6 mm diameter) indicated that the samples were single phase, except for the AmCo₂ and particularly the AmFe₂ samples which were found to have small amounts of a second phase. The AmFe₂ sample was re-annealed at 1000°C for one month to insure homogenization.

*Work supported by the U. S. Energy Research and Development Administration.

Approximately 100 mg of each compound was sealed in an aluminum capsule for magnetization measurements and a further 30 mg was sealed in a separate aluminum capsule for the NGR experiments. Magnetizations were measured between 2.5 and 300 K in fields up to 14 kOe by a force method; the experimental technique has been described in detail [2]. NGR measurements were made at 4.2 K by means of the 83.9 keV resonance in ^{243}Am .

III. RESULTS AND DISCUSSION

The crystal structures and lattice parameters of americium Laves phases are given in Table 1. The stability of Laves phases is usually discussed in terms of a hard-sphere packing model [1]. The ideal ratio of the radii of the constituent atoms R_A/R_B (A would correspond to americium) is 1.225. The fact that phases exist with ratios between 1.05 and 1.68 suggests that substantial changes occur in the size of the component atoms when the structure forms. The radius ratios of americium Laves phases (Table 1) are all greater than the ideal value. The parameters $(D_A - d_A)$ and $(D_B - d_B)$, where D_A and D_B are the CN12 atomic diameters and d_A and d_B are the atomic diameters of the A and B atoms in the Laves phase, are usually used to indicate the degree of atomic size adjustment [1]. The values of these parameters, also given in Table 1, are consistent with the earlier results for actinide Laves phases (see Fig. 1 of ref. [1]) and indicate that when the radius ratio is greater than the ideal value (1.225) the actinide atoms contract [$(D_A - d_A) > 0$], whereas the B atoms remain unchanged in size [$(D_B - d_B) \approx 0$].

TABLE 1
Crystal structures and parameters of Americium Laves phases

Compound	Crystal Structure	Lattice Parameters		R_A/R_B	$D_A - d_A$ (Å)	$D_B - d_B$ (Å)
		a (Å)	c (Å)			
AmAl_2	MgCu_2 -type	7.861(1) ^a		1.26	0.20	0.08
AmCo_2	"	-	b			
AmFe_2	"	7.30(1)		1.42	0.44	-0.04
AmIr_2^c	"	7.550(3)		1.33	0.33	0.03
AmNi_2^d	"	6.99(1)		1.45	0.57	0.01
AmPt_2^d	"	7.66(1)		1.30	0.28	0.05
AmRh_2	"	7.51(1)		1.34	0.35	0.02
AmOs_2^d	MgZn_2 -type	5.320	8.849			
AmRu_2	"	5.26(1)	8.73(1)			

- Number in parenthesis represents estimated error in last significant figure.
- The high-angle lines were too diffuse for a lattice parameter determination.
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The compounds AmAl_2 , AmRh_2 and AmRu_2 are all paramagnetic over the entire temperature range studied. Magnetic susceptibilities χ (emu/g) were calculated by fitting the magnetization σ (emu/g) versus field H (Oe) data at different temperatures to an equation of the form

$$\sigma = \sigma_0 + \chi H. \quad (1)$$

All the samples had significant positive values of σ_0 that are presumably associated with either ferromagnetic impurities in the ^{243}Am starting material or the ^{239}Np daughter product of the alpha decay of ^{243}Am . The molar susceptibilities of the three samples are shown as a function of temperature in Fig. 1. The rather large experimental error is associated with the presence of the σ_0 term together with the small intrinsic values of the susceptibility. The compounds show a relatively temperature-independent susceptibility and no evidence of magnetic order.

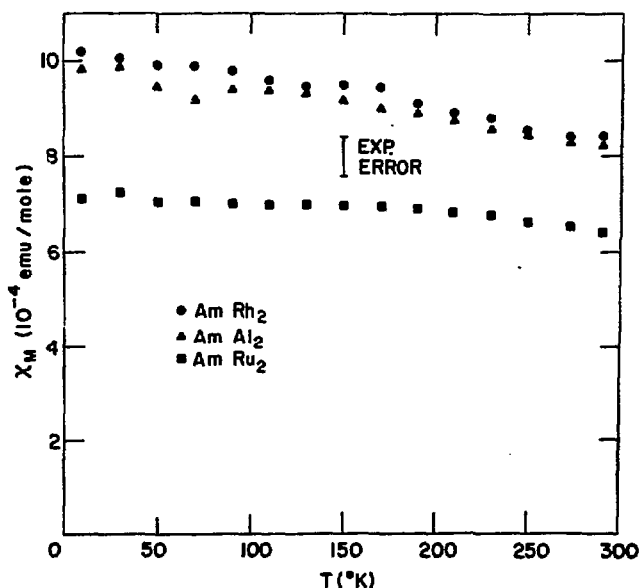


Fig. 1. Temperature dependence of the molar susceptibility of AmAl_2 , AmRh_2 and AmRu_2 .

On the other hand, AmFe_2 is ferromagnetic over the entire temperature range and some magnetic isotherms are given in Fig. 2. On the basis of the temperature dependence of the magnetization at constant field, we would estimate that the Curie temperature T_C is in the range 350–400 K. This temperature is substantially lower than those of NpFe_2 (~500K) [3] and PuFe_2 (~600K) [5]. However, the maximum magnetization value (Fig. 2) is ~50% greater than for NpFe_2 and PuFe_2 and corresponds to a moment of $2.9\mu_B/\text{mole}$. The large, slowly-saturating field dependence is typical of actinide ferromagnets [2,3,5] and presumably reflects the presence of a large magnetocrystalline anisotropy. In the absence of single crystals, it will be necessary to use very high fields to obtain reliable bulk moments for these compounds. The decrease in the relative field dependence of the magnetization as the temperature approaches T_C (Fig. 2) again suggests the influence of anisotropy.

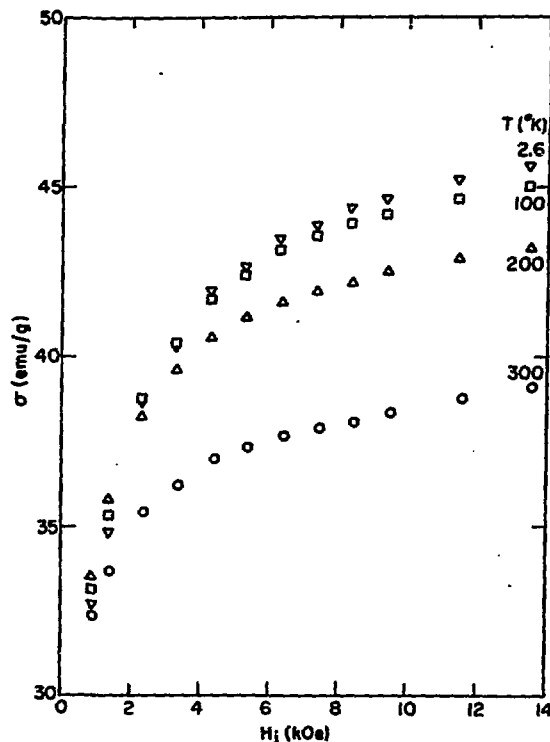


Fig. 2. Magnetization versus internal field isotherms for AmFe_2 .

The NGR experiments showed a single resonance line for each of the samples and the results are summarized in Table 2. There is some line broadening in AmFe_2 , which is probably due to a transferred hyperfine field, but there is no evidence of a magnetic moment at the americium site - indicating a non-magnetic ($5f^6$) ground state of the Am^{3+} ion. The moment per iron atom thus deduced for AmFe_2 from the magnetization results ($\sim 1.5\mu_B$) is higher than in either UFe_2 ($\sim 0.5\mu_B$) [5], NpFe_2 ($\sim 1\mu_B$) [3], or PuFe_2 ($\sim 1\mu_B$) [6]. The variation in magnetic properties among the actinide-iron Laves phases is not understood at this time.

TABLE 2

NCR results for americium Laves phases; source ^{243}Am in PuAl_2

Compound	Isomer Shift ^a (mm/s)	Linewidth (mm/s)
AmAl_2	+1.6	9.1
AmFe_2	-6.0	31.5
AmRh_2	-4.8	7.7
AmRu_2	-10.9	10.0
	(<u>+0.5</u>)	(<u>+0.5</u>)

^aRelative to PuAl_2

The authors would like to acknowledge the capable experimental work of A. W. Mitchell and S. D. Smith.

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