



# OPEN Superionic-like diffusion in yttrium dihydride

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For the next-generation high temperature microreactors, yttrium dihydride (YH<sub>2</sub>) is an attractive solid state neutron moderator. Despite a number of recent investigations, the mechanism of hydrogen transport remains poorly understood. Experimental evaluations of diffusivity are inconclusive with large variations in diffusivities and activation energies. In this work, we perform *ab initio* molecular dynamics (AIMD) simulations on YH<sub>2</sub> for temperatures spanning 300 K to 1200 K. Our main finding is that YH<sub>2</sub> shows a superionic-like behavior with hydrogen atoms hopping from one native site to another above a characteristic temperature of 800 K. This correlated motion results in quasi-one-dimensional string-like displacements that enable the hydrogen atoms to diffuse rapidly. We confirm that the octahedral sites are mostly unoccupied, although channeling through them is the most favored pathway between lattice hops above 800 K. At the highest temperature of 1200 K, the string relaxation time is merely of the order of a few picoseconds, which indicates a liquid-like diffusive behavior. Based on the formation of spontaneous thermal vacancies, an order-disorder crossover temperature  $T_{\alpha} \sim 800$  K is established for YH<sub>2</sub> with an activation energy of 0.83 eV for hydrogen diffusion in the superionic-like state.

The persistent reliance on fossil fuels has resulted in arduous environmental challenges, notably air pollution and climate change driven by carbon emissions. Nuclear power continues to hold the pole position for generating carbon-free electricity all year round with the highest capacity factor among all the clean energy sources. In the United States, nuclear energy contributes to roughly one fifth of the total electric power generated with an estimated 24 Gt of avoided CO<sub>2</sub> emissions over the past fifty years<sup>1</sup>. Most of the nuclear power in the United States comes from large light water reactors with a power rating of 1000 MW<sub>e</sub> or more built more than fifty years ago; construction of new plants has come to a standstill due to large upfront investments and protracted construction schedules. In recent years, there has been a shift to designing and building small modular reactors (SMRs) that are scalable, affordable and thermodynamically more efficient. A smaller subset of SMRs labeled as microreactors with a power rating of 20MW<sub>e</sub><sup>2</sup> are also being considered for deployment in areas that do not have access to the electric grid. These truck-transportable reactors require a compact reactor design that can be achieved by solid-state neutron moderators such as metal hydrides<sup>3</sup>. Yttrium dihydride (YH<sub>2</sub>) is a particularly suitable candidate due to its ability to retain hydrogen at temperatures exceeding 1000 °C and its remarkable thermal stability at elevated temperatures<sup>4,5</sup>. However, the mobility of hydrogen in YH<sub>2</sub> introduces special challenges such as redistribution or loss at high temperatures that can adversely impact the neutronic behavior and potentially reduce the operational lifespan of the microreactors. It's critical, therefore, to establish the hydrogen migration and retention mechanisms in YH<sub>2</sub> to ensure safe reactor operational limits under normal and off-normal conditions<sup>6</sup>.

Hydrogen diffusivity in metal hydrides has been investigated using nuclear magnetic resonance (NMR) spectroscopy<sup>7</sup> and neutron spectroscopy (NS)<sup>8,9</sup> through inelastic neutron scattering; other techniques such as mechanical spectroscopy<sup>10</sup> have also been reported. Both NMR and NS techniques are well-suited for probing hydrogen motion due to the gyromagnetic ratio of the proton for the former that is the largest among all known stable nuclei<sup>11</sup>, and the large incoherent neutron scattering cross section for the latter (~80 b), which is over ten times more than other common nuclei<sup>12</sup>. All the spectroscopic methods extract relaxation rates or times that are associated with hydrogen diffusion under various approximations. Most NMR methods rely on evaluating the spin-lattice ( $T_1$ ) and spin-spin ( $T_2$ ) relaxation rates, which are strongly sensitive to atomic motion such as translation and rotation; a third relaxation rate ( $T_{1\rho}$ ), which is associated with the relaxation of transverse component of magnetization, is also used to assess the atomic jump frequency<sup>11</sup>. The spin echo NMR technique, often employed with a pulsed field gradient (PFG-NMR)<sup>13</sup>, can estimate diffusivity directly without any reference to lattice paths or correlations while spin-lattice relaxation measurements need additional information usually supplied by jump diffusivity models<sup>11</sup>. In inelastic neutron scattering (INS) or quasi-elastic neutron scattering

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(QENS), which is a limiting case of INS, the broadening of the quasi-elastic peak provides the jump distances and diffusivities of mobile species in translational diffusion<sup>9,14</sup> typically using Gaussian diffusion models.

While hydrogen is known to have high mobilities in hydrides, the mechanism of transport is not well-elucidated. In recent years, group II metal hydrides such as  $\text{BaH}_2$  have been shown to be superionic conductors resulting from a (Ni,In) hexagonal structure at high temperatures. The current evidence for superionic-like transport in group III or IV dihydrides is suggestive but not conclusive<sup>15–18</sup>. An anomalous increase in the nuclear spin relaxation time at high temperatures has been interpreted as a transition to a superionic-like state with correlated motion among hydrogen atoms<sup>17</sup> although objections have been raised for using only the spin-lattice ( $\Gamma_1$ ) relaxation<sup>8,19</sup>. Alternate explanations involving the presence of hydrogen molecules or atom pairing at high temperatures<sup>20</sup>, deemed unlikely by Cotts<sup>21</sup>, or the presence of unidentified impurities have also been offered<sup>22</sup>.

Diffusivity is strongly sensitive to the stoichiometry; as a general trend, the diffusivity increases with increasing hydrogen concentration in  $\text{YH}_x$ , which is opposite to that observed in  $\text{ZrH}_x$  as shown in several investigations<sup>8,13</sup>. Using neutron spectroscopy, Stühr and coworkers<sup>8</sup> evaluated a diffusivity of  $\sim 10^{-9}$  m<sup>2</sup>/s for hydrogen atoms at a temperature  $\sim 1250$  K with  $\text{YH}_{1.97}$ . This work is notable because of liquid-like diffusivity, albeit at high temperatures, obtained with a stoichiometry that is nearly two. More interesting is a crossover behavior  $\sim 870$  K where the diffusivity starts increasing at a faster rate with a small but conspicuous change in activation energies. Curiously, this transition which went undetected previously, appears to conform with a peak in the specific heat at a slightly lower temperature of  $\sim 800$  K as well as a change in behavior for thermal expansivity observed in more recent measurements with  $\text{YH}_{1.92}$ <sup>23</sup>. A peak in the specific heat, or other thermodynamic response functions such as compressibility and thermal expansivity, is a signature of a second-order phase transition that is associated with Type II superionic conductors<sup>24</sup>. Commonly known as the  $\lambda$  transition, a sharp rise and drop in the specific heat is observed at a characteristic temperature ( $T_\lambda$ ) and is usually accompanied by a change in the ionic conductivity<sup>25,26</sup>. Stoichiometric-dependent peaks in the specific heat reported by Trofimov et al.<sup>27</sup> appear to suggest that this transition temperature decreases with increasing hydrogen content in  $\text{YH}_x$ . These new observations give a certain plausibility to the original speculation by Barnes and coworkers<sup>16,17</sup> that the long correlation time for proton dipolar interactions in NMR measurements may signify an onset of strongly correlated motion among the hydrogen atoms and the diffusive motion may be analogous to that in Type II fast ion conductors.

Verifying the existence of superionic-like diffusion in  $\text{YH}_x$  is challenging due to the inherent limitations in interpreting data and various experimental uncertainties. Electronic structure simulation methods such as those based on density functional theory (DFT)<sup>28</sup> along with energy landscape sampling methods<sup>29,30</sup> are useful for identifying potential pathways for diffusion. However, these methods primarily provide kinetic rates based on the potential energy barriers between presumed lattice/defect sites and do not typically incorporate thermal fluctuations that play a key role in correlated diffusive transport. In this work, we use ab initio molecular dynamics (AIMD) simulations based on DFT to probe the complex diffusive dynamics of hydrogen atoms in  $\text{YH}_2$  at finite temperatures.

Building upon previous molecular dynamics (MD) analysis on superionic fluorites<sup>25,26,31</sup>, we show that the hydrogen atoms start diffusing in a string-like manner beyond a characteristic temperature ( $T_a$ ) of  $\sim 800$  K. Correlated motion of hydrogen atoms across the tetrahedral (*tet*) sites by thermal activation is shown as the primary mechanism of diffusion at high temperatures. Unlike speculated in the early investigations<sup>8,13,16</sup>, the occupancy of octahedral (*oct*) sites is minimal, which is consistent with the assessment made in a more recent experimental-computational study<sup>28</sup>. The *oct* sites, however, are dynamically favored as plausible transition state points connecting two neighboring *tet* sites at high temperatures. Based on the formation of spontaneous thermal vacancies and string relaxation times, an order-disorder transition temperature  $T_a \sim 800$  K, similar to that in fluorite superionic conductors<sup>25,26,31</sup>, is established for  $\text{YH}_2$  with a possible superionic-like transition occurring at  $T_\lambda \sim 1000$  K.

## Results and analysis

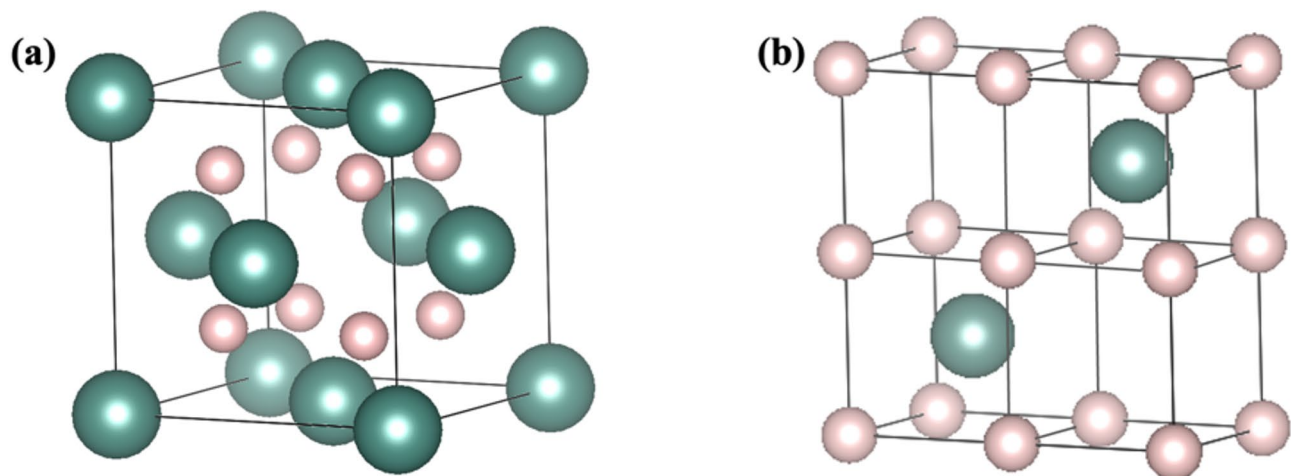
$\text{YH}_2$  has a fluorite structure and belongs to the cubic  $\text{Fm}\bar{3}\text{m}$  space group. The Y atoms form a *fcc* lattice with the H atoms occupying the *tet* sites as shown in Fig. 1(a). In an alternative representation, the H atoms are positioned in a simple cubic lattice with the Y atoms occupying alternate cube centers that are also the *oct* locations of the *fcc* lattice – see Fig. 1(b); this leaves two empty *oct* sites that can potentially accommodate the H atoms.

In this work, we use AIMD simulations as implemented in the Vienna Ab initio Simulation Package (VASP)<sup>32,33</sup> of  $\text{YH}_2$  for temperatures between 300 and 1200 K focusing on the mobility of hydrogen atoms. The details are given in the Methods section.

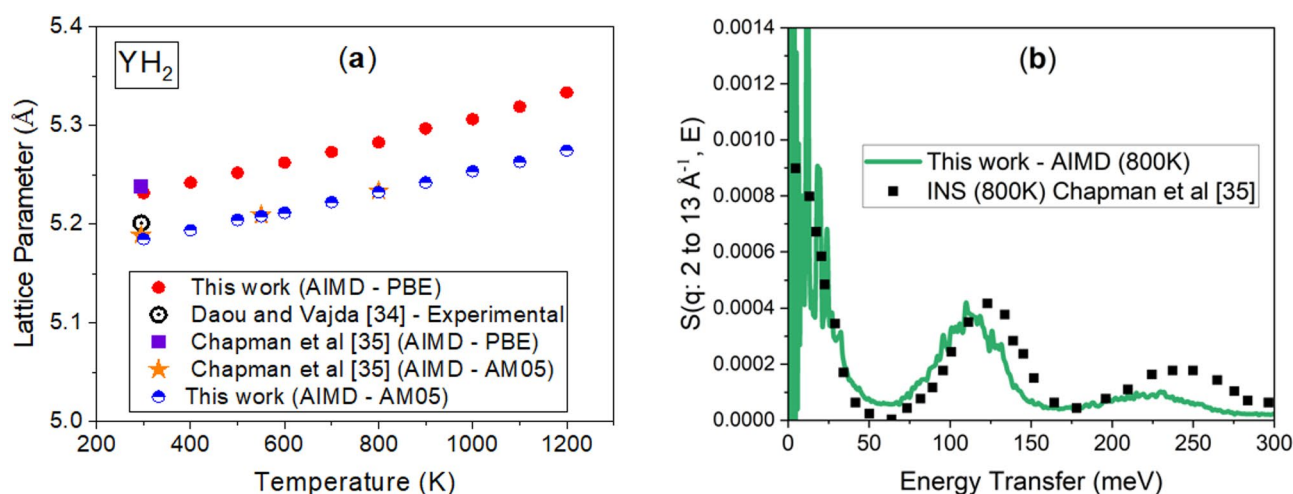
## Validation/verification of the AIMD simulations

To validate our simulations, the structural and dynamical features are compared against available experimental data; verification is provided by comparison with past simulation data. Fig. 2(a) delineates the variation of lattice parameter with temperature. At 300 K, a lattice parameter of 5.231 Å with the PBE functional is only 0.6% higher than the experimental value of 5.201 Å<sup>34</sup> (at 295 K); it also matches well with data from Chapman et al.<sup>35</sup> with the same functional. With the AM05 functional, our AIMD simulations underpredicts the lattice parameter at 295 K slightly, but are nearly identical to the simulation data from Chapman et al.<sup>35</sup> using the same functional. Given the greater computational expense with the AM05 functional, and the relatively good prediction with the PBE functional for the structural properties (see supplementary information), the rest of the results presented here are with the PBE functional.

Next, we compare the dynamic structure factor integrated over wavevectors ( $q$ ) ranging from 2 to 13 Å<sup>−1</sup> from neutron scattering experiments<sup>35</sup> against simulation data in Fig. 2(b). The AIMD data, although noisy



**Fig. 1.** Crystal structure of  $\text{YH}_2$ . (a) The *fcc* lattice formed by Y atoms and (b) the simple cubic lattice formed by the H atoms with alternating cube centers occupied by Y atoms.



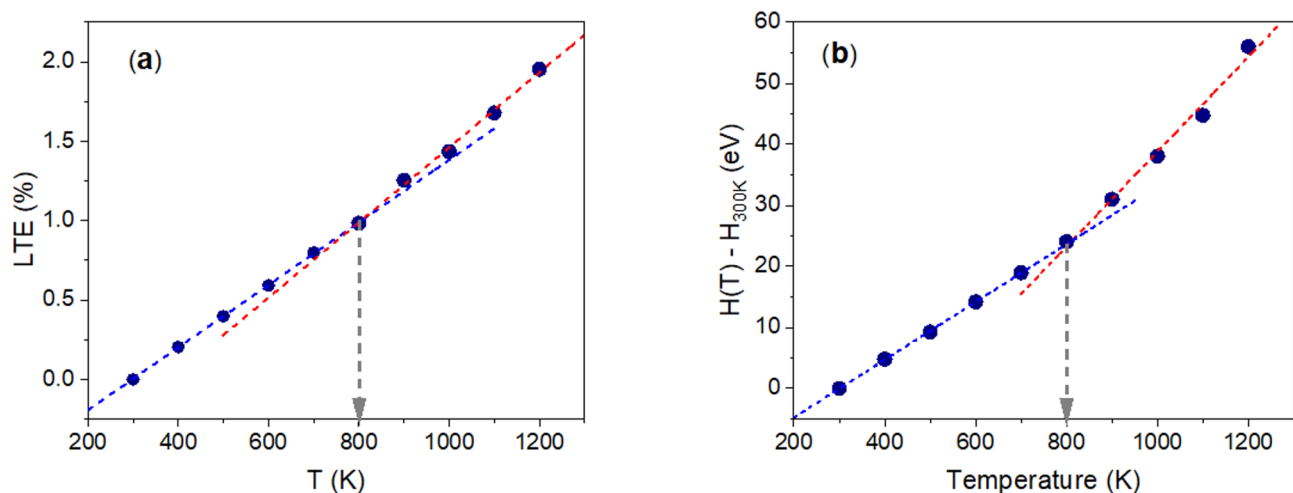
**Fig. 2.** Validation/verification of the AIMD simulations, (a) lattice parameter from the current simulations for different temperatures compared to experimental data<sup>34</sup> (295 K) and previous AIMD simulations<sup>35</sup>, (b) comparison of dynamic structure factor for  $\text{YH}_2$  integrated over wavevectors ( $q$ ) ranging from 2 to 13  $\text{\AA}^{-1}$  from inelastic neutron scattering (INS) experiments<sup>35</sup> at 800 K against current AIMD prediction.

at shorter wavevectors due to limited number of atoms in the simulation supercell (324) and lack of statistical averaging, is able to predict the first peak location and relative amplitude with good fidelity. While the second peak position is well-captured, there is a greater decay in the amplitude suggesting more thermal relaxation at larger frequencies with the AIMD simulations.

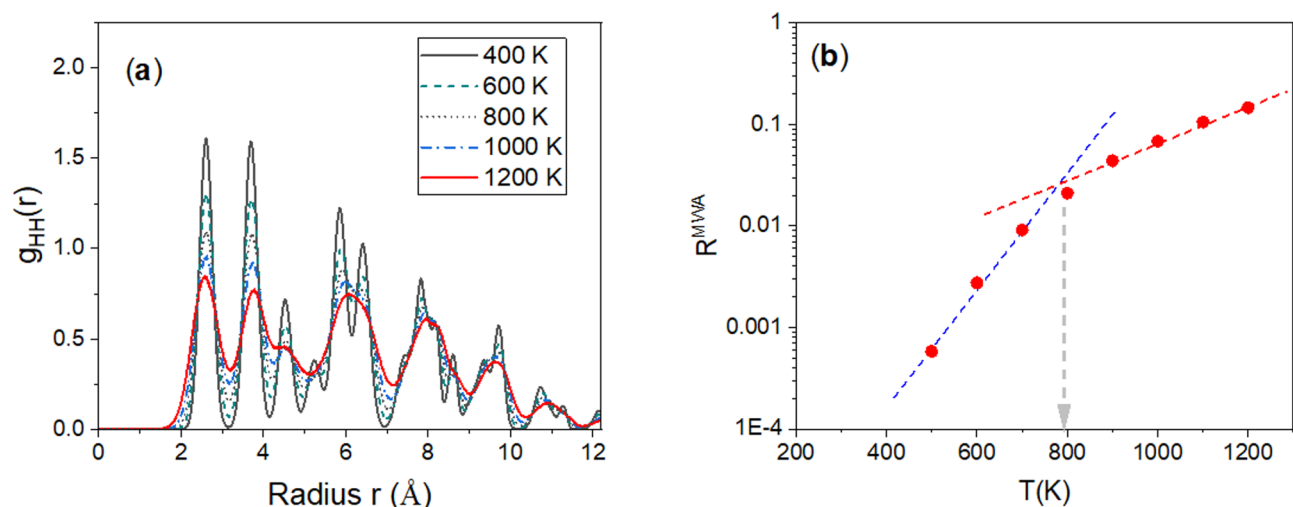
### Linear thermal expansion and enthalpy

The temperature dependence of linear thermal expansion (LTE) and enthalpy difference (relative to 300 K) are shown in Fig. 3. The LTE is computed based on the change in lattice parameter normalized to that at 300 K. The current simulations are performed at 0 pressure; so, the enthalpy is identically equal to the internal energy. Interestingly, a change in slope at 800 K can be seen in both LTE and the enthalpy variation, which is indicative of two different responses – one at low temperatures ( $\leq 800$  K) and another at higher temperatures ( $> 800$  K). To explore this transition further, the radial distribution function  $g_{HH}(r)$  which depicts the local arrangement of H atoms at various temperatures is delineated in Fig. 4(a). At 400 K,  $g_{HH}(r)$  exhibits well-defined peaks indicating an ordered lattice structure. As the temperature increases, the magnitude of the peaks diminishes significantly and the sharper peaks beyond the second neighbor merge to form fewer and more diffusive peaks reflecting a transition towards a more disordered state.

To identify a more specific temperature for a plausible transition, we compute the Wendt-Abraham (WA) parameter, which is defined as the ratio of the magnitude of the first minimum to the maximum of the first peak in  $g_{HH}(r)$ :  $R^{WA} = g_{min}/g_{max}$ <sup>36</sup>. Previously, this parameter has been used as a criterion for



**Fig. 3.** (a) Linear thermal expansion (LTE) as a function of temperature, and (b) calculated enthalpy difference for different temperatures. Both LTE and enthalpy exhibit a change of slope at 800 K demarcating a low temperature behavior from a high temperature one.

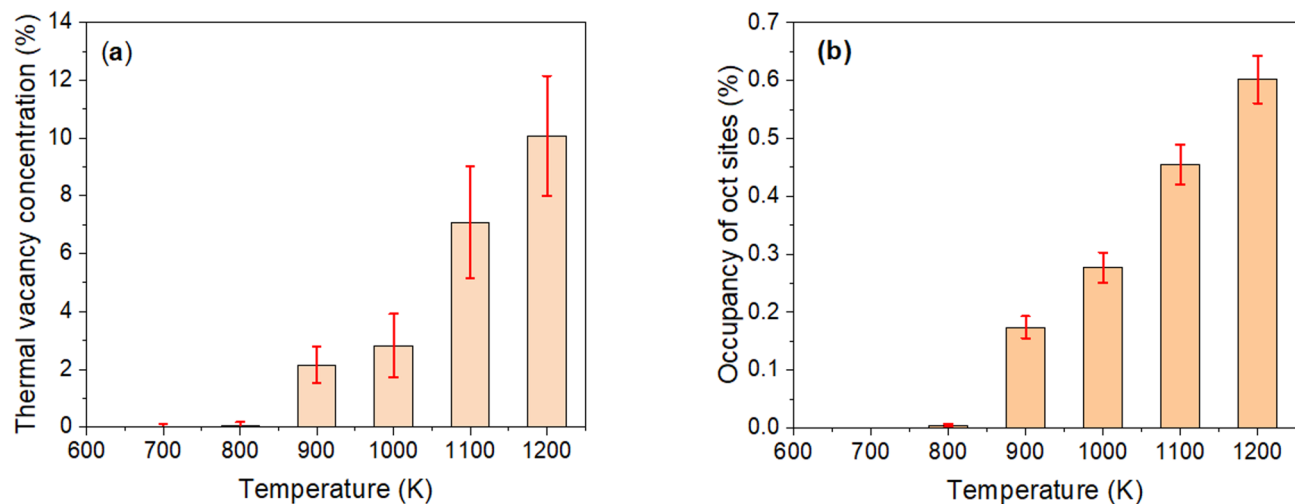


**Fig. 4.** (a) H-H radial distribution function at different temperatures, (b) modified Wendt-Abraham (MWA) parameter for the hydrogen atoms indicating a crossover from low to high temperature behavior at ~800 K.

identifying a transition to an amorphous phase<sup>36</sup>. More recently, Celtek et al.<sup>37</sup> proposed that the squared WA parameter,  $R^{MWA} = (g_{min}/g_{max})^2$ , is a more accurate predictor of the transition temperatures<sup>37</sup>. From Fig. 4(b), it can be observed that  $R^{MWA}$  has a smooth variation across the temperatures. However, a low temperature and a high temperature regime is clearly discernible with a crossover at ~800 K. Thus, the structural (LTE, radial distribution function) and thermodynamic (enthalpy) metrics strongly suggest an order-disorder transitioning in  $YH_2$  at ~800 K. In the next sections, we will investigate the dynamical changes across this temperature.

### Analyzing the motion of H atoms

There are several space-time correlation functions that can be used to inspect the dynamical behavior of the hydrogen atoms with AIMD. First, we use a simpler method to map their motion. Making an analogy with the fluorite superionic conductors, the crossover temperature  $T_a$  marks an order-disorder transition where the mobile atoms leave their native lattice sites<sup>26,31,38</sup>. Thus, vacancies (and interstitials) are spontaneously created; these are identified as dynamic Frenkel pairs in fluorites<sup>24</sup>. To identify such vacancies, we utilize the Wigner-Seitz cell method<sup>39</sup> to count the number of dynamic vacancies forming at each *tet* site, which is then averaged over the simulation time. Fig. 5(a) depicts the average vacancy concentration at different temperatures. Below 800 K, and during the simulation time of 30 ps, no significant number of vacancies are observed at the *tet* sites. With the temperature increasing above 800 K, a notable increase in the number of vacancies is observed, which indicates that H atoms are being thermally displaced from their original positions. Although there is a significant



**Fig. 5.** (a) The average vacancy concentration at different temperatures, (b) occupancy of H atoms at the *oct* sites.

number of vacancies (as high as ~10%) at 1200 K, the *oct* sites, as shown in Fig. 5(b), are meagerly occupied. The displaced H atoms thus are spatially distributed across the domain.

To qualitatively assess the path of the displaced H atoms, we examine the spatial distribution of the H atoms. The  $\text{YH}_2$  supercell is first discretized into a grid of  $30^3$  (27,000) equi-sized cells with each H atom assigned to one of these cells based on its spatial coordinates. Fig. 6(a) – (d) represents contour plots that depict the probability of locating H atoms on the (110) plane for temperatures ranging from 800 K to 1200 K. Expectedly, the highest probability of H atom occupancy, regardless of the temperature, is for the *tet* sites represented by the dark green regions. At 800 K, in accordance with the average concentration in Fig. 5(a), most of the H atoms are vibrating about their native *tet* sites with some evidence of displacement towards the *oct* sites. At 900 K, the probability contours clearly delineate the tendency of the displaced H atoms to move into the *oct* sites and out; this channeling is strengthened with increasing temperature. While not completely demonstrated, a discernible pathway through the *oct* sites appears to connect the neighboring *tet* sites, possibly facilitating a rapid diffusion of H atoms within the  $\text{YH}_2$  lattice.

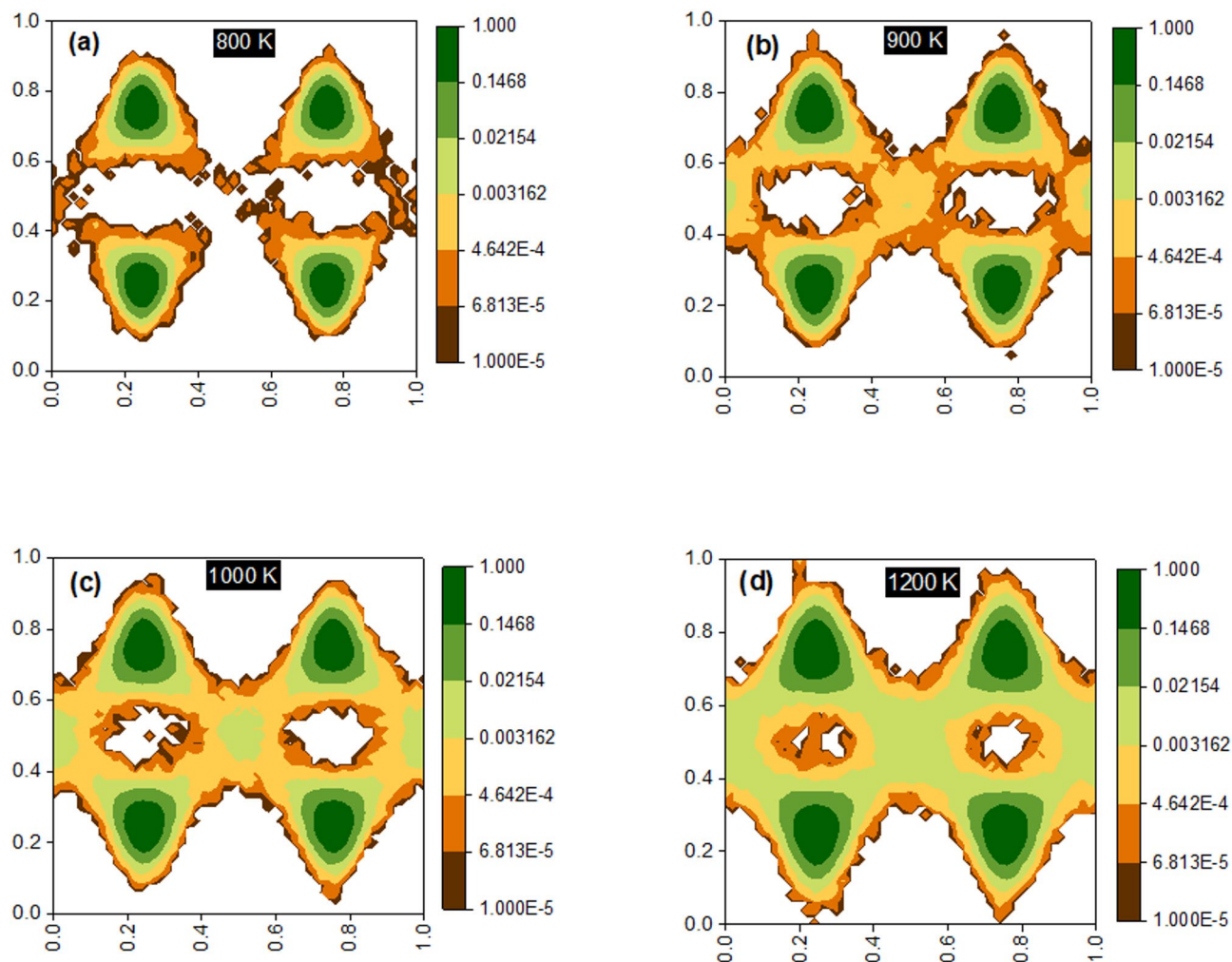
At the highest temperature of 1200 K, there is evidence of atoms making a tortuous path through and along the edges of the *oct* sites, which is strikingly similar to the convoluted migration path of anions coupled with large amplitude vibrations in fluorites<sup>40</sup>. The *oct* sites in  $\text{YH}_2$ , unlike in fluorites superionic conductors, tend to accommodate a larger number of mobile atoms with increasing temperature as shown in Fig. 5(b). The probability map of anions in fluorites thus is more homogeneous<sup>25</sup> compared to  $\text{YH}_2$ , which shows a small but significant propensity to escort the displaced H atoms through the *oct* sites. The rather small probability strongly suggests that the *oct* sites are merely transition states and not true local minima in the potential energy landscape – a conclusion that is reached in a recent study using DFT simulations albeit at 0 K<sup>29</sup>.

### Van-Hove self-correlation

The pathways of H atoms are better illustrated through the Van Hove self-correlation function  $G_s(\mathbf{r}, t)$ <sup>25,41</sup>, which gives the probability of finding a H atom in the neighborhood of a given displacement and time given that it was located at the origin initially. For an isotropic simple liquid, the spatial Fourier transform of  $G_s$ , which is the density correlator, at long (and short) times can be shown to be proportional to an exponentially decaying function of the mean square displacement  $\langle \mathbf{r}^2(t) \rangle$ , which in turn is proportional to the self-diffusivity of the species<sup>42,43</sup>. Within this approximation,  $G_s$  approaches a Gaussian function at long times. Neutron scattering evaluates self-diffusivity by measuring the incoherent scattering function, which is then fitted to a Lorentzian that corresponds to the Gaussian approximation for  $G_s$ <sup>44</sup>.

Fig. 7 depicts the Van Hove self-correlation function at 20 ps for temperatures ranging from 800 K to 1200 K (left axis). At 800 K, as shown in panel (a),  $G_s$  is mostly bounded indicating that H atoms are mostly vibrating at the *tet* sites. A short emergent tail suggests that a few atoms are moving to the nearby locations, which is consistent with the insignificant number of thermal vacancies generated at this temperature shown in Fig. 5(a). Along with the  $G_s$ , the radial distribution function  $g_{HH}(r)$  is also shown in Fig. 7 (right axis). At 800 K, the H atoms are mostly placed at the geometric nearest neighbor (*tet*) sites  $(a/2)[1, \sqrt{2}, \sqrt{3} \dots]$ , where  $a$  is the lattice parameter (~5.2 Å). As temperature increases, the minor shoulder at  $a$  disappears with the outer peaks merging together at ~6 Å. Quite remarkably, multiple peaks also appear for  $G_s$  as shown in panels (b) to (d) at higher temperatures and the  $G_s$  peaks align more or less precisely with the peaks observed in  $g_{HH}(r)$ . Although  $G_s$  is evaluated at 20 ps, the behavior can be observed at longer times; in the next section, we show that the relaxation time for correlated motion is of the order of few ps at 1200 K. The close correspondence between the peaks in  $G_s$  and  $g_{HH}(r)$  demonstrates that the H atoms jump from one lattice (*tet*) site to another simultaneously. This cooperative motion, which becomes more pronounced with increasing temperature, is the basis of diffusive





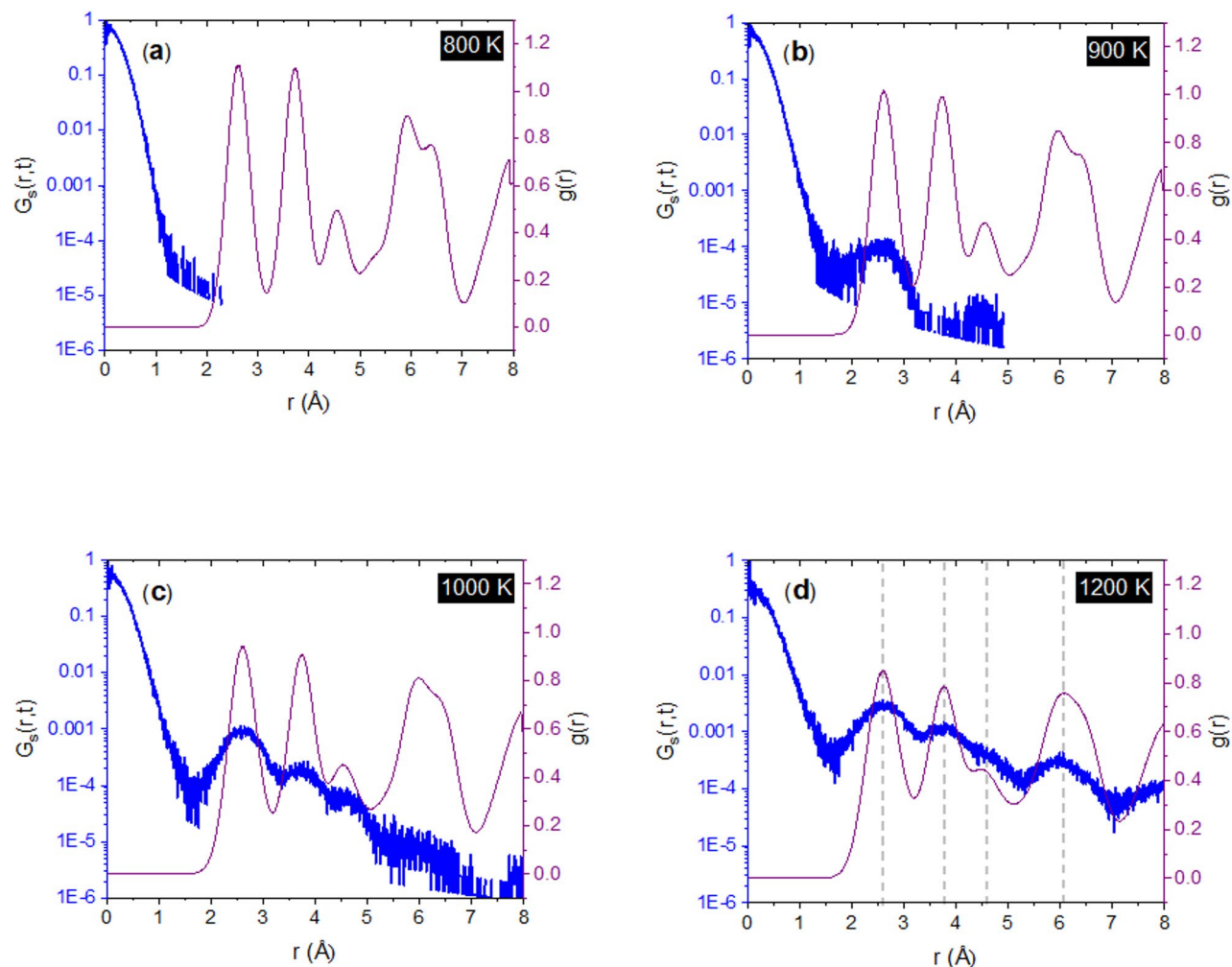
**Fig. 6.** Probability contours of finding H atoms on the (110) plane at different temperatures, (a) 800 K, (b) 900 K, (c) 1000 K and (d) 1200 K. The white spaces correspond to probabilities lower than  $10^{-5}$ .

motion in  $\text{YH}_2$  above a crossover temperature of 800 K. Although speculated before<sup>16,17</sup>, the multiple peaks in  $G_s$  draw out the collective motion among the H atoms unambiguously.

It is apparent that  $G_s$  does not have a Gaussian profile as commonly assumed in diffusion modeling or for interpreting experimental data. Instead, it decays with distance in a somewhat exponential manner that suggests the presence of dynamic heterogeneity (DH) or spatially-separated clustering of fast and slow atoms<sup>45</sup>. DH is commonly observed in supercooled liquids, colloids, and granular materials<sup>46</sup> and also in superionic conductors based on our previous work<sup>26,31</sup>. A common feature that is associated with DH is the emergence of low-dimensional or quasi-one-dimensional pathways for the diffusing species. In the next section, we will provide evidence for cooperative string-like motion of the mobile H atoms at high temperatures.

### String-like pathways for H atoms

We follow the procedure of an earlier work by Annamareddy and Eapen<sup>26</sup> and define strings as a set of ions constituted by pairs of mobile atoms where one atom gets replaced another over a certain time interval<sup>47,48</sup>. To test the presence of string-like motion, the most mobile H atoms are analyzed (20%); our main result is shown in Fig. 8(a). At 800 K or lower, the string pathways are not well-formed, which is consistent with the incipient generation of thermal vacancies. At 900 K, string formation is conspicuous and with increasing temperature, the number of participating atoms in the strings increases reaching a maximum at 1000 K (nearly 80%) followed by decreasing participation. Notably, the relaxation time monotonically decreases with increasing temperature reaching a few picoseconds at the highest temperature, which is suggestive of fast liquid-like diffusive behavior. Remarkably, the cooperative motion among the H atoms is completely analogous to that in fluorite superionic conductors<sup>26,31</sup> and our current results provide unequivocal support for a superionic-like diffusive process in  $\text{YH}_2$  above a crossover temperature of 800 K. The hydrogen atoms typically show large amplitude oscillations before making a diffusive hop and there are several ways to visualize this motion<sup>49,50</sup>. In Fig. 8(b) we have depicted a snapshot of simultaneous hops of hydrogen atoms at 1000 K that appear as string-like displacements;



**Fig. 7.** Van Hove self-correlation function  $G_s(\mathbf{r}, t)$  of the H atoms for different temperatures – (a) 800 K, (b) 900 K, (c) 1000 K and (d) 1200 K at 20 ps (left axis, blue), and the radial distribution function  $g_{HH}(r)$  (right axis, purple) for corresponding temperatures.

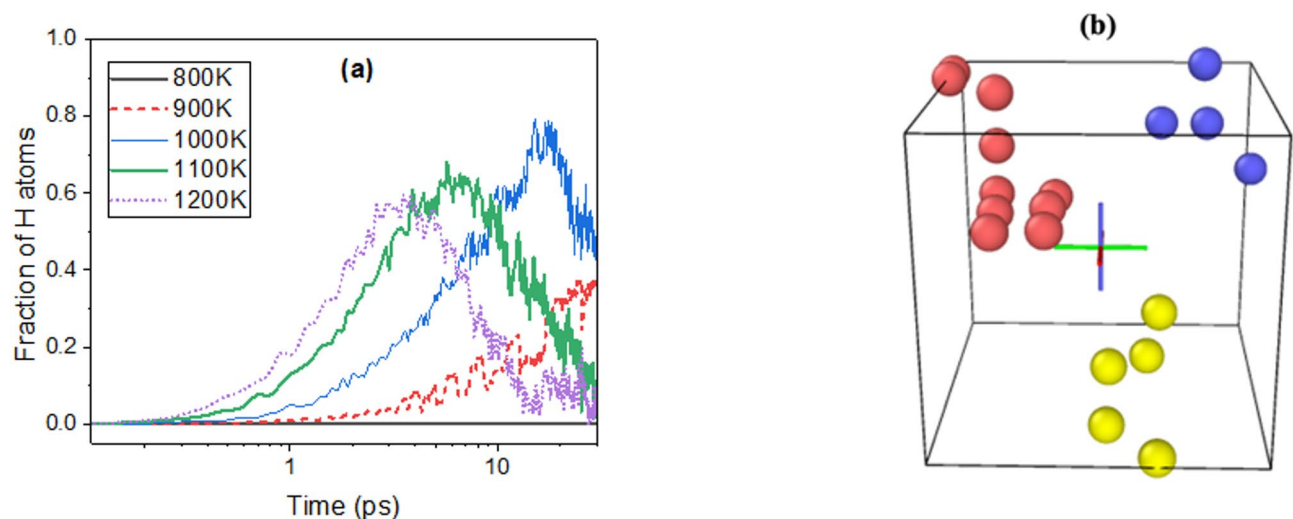
detailed analysis of the string metrics such as the distribution of string lengths and lifetimes will be reported in a future publication.

From the results thus far, there is considerable evidence for an order-disorder transition at  $T_a \sim 800$  K, which is similar to that observed for fluorite superionic conductors<sup>25,26,31</sup>. As mentioned previously, a second order phase transition marked by a theoretical discontinuity in the thermodynamic response functions (such as specific heat) is also operational in fluorites; more demanding simulations, however, at several more temperatures are needed for such evaluations. Nevertheless, there is one key evidence that foretells a superionic-like transition temperature ( $T_\lambda$ ). In earlier studies on fluorites,  $T_\lambda$  is observed to coincide approximately with the peak participation in the string formation<sup>26</sup>. Based on the variation shown in Fig. 8(a), it appears that  $T_\lambda$  is likely to be in the vicinity of 1000 K. We will leave it to a future study with larger simulation systems and enhanced statistical sampling for probing this second order thermodynamic transition and evaluating other string metrics such as the mean string length, probability distributions and dynamic heterogeneity.

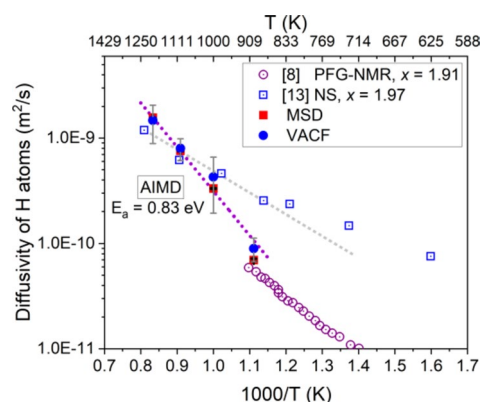
### Hydrogen diffusivity

The self-diffusivity of the hydrogen atoms is evaluated by two equivalent methods, one based on the mean square displacement (MSD) of the diffusing atoms and the other based on the velocity autocorrelation (VAC) function. The former falls into the Einstein formalism while the latter come under the Green-Kubo formalism<sup>43</sup>. As delineated in Fig. 9, both methods yield similar results with an average activation energy of 0.83 eV. We also note that the electronic density of states does not change significantly with temperature, and electronic temperature has a minimal effect on hydrogen diffusion.

Two sets of experimental data are also shown – data on  $\text{YH}_{1.97}$  using neutron spectroscopy<sup>8</sup> and data on  $\text{YH}_{1.91}$  from pulsed field gradient nuclear magnetic resonance<sup>13</sup> measurements. Diffusivity is strongly dependent on stoichiometry and as previously elucidated, the diffusivity typically increases with more hydrogen content<sup>8,13</sup>.



**Fig. 8.** (a) Fraction of H atoms participating in strings among the most mobile (20%) H atoms for temperatures ranging from 800 K to 1200 K in steps of 100 K, (b) typical quasi-one-dimensional string-like displacements of hydrogen atoms at 1000 K.



**Fig. 9.** Self-diffusivity of hydrogen in  $\text{YH}_2$  using mean square displacements (MSD) and velocity autocorrelation (VAC) from AIMD simulations (solid symbols). Error bars for MSD and VACF are based on the standard error of the mean across 300 and 2,400 time origins, respectively. The error bars for diffusivity from MSD are less than the symbol size. The open square symbols are experimental data on  $\text{YH}_{1.97}$  using neutron spectroscopy (NS)<sup>8</sup> while the open circles represent data from pulsed field gradient nuclear magnetic resonance spectroscopy (PFG-NMR)<sup>13</sup> with  $\text{YH}_{1.91}$ . A crossover can be noted in the NS data at  $\sim 870$  K.

with the exception of one recent experimental study<sup>51</sup>. The closest stoichiometric match is with  $\text{YH}_{1.97}$  used by Stühr and coworkers<sup>8</sup> in their neutron spectroscopic (NS) investigation. Interestingly, the NS data can be demarcated into high temperature and low temperature regimes with a crossover at  $\sim 870$  K, which is somewhat close to the estimate from the current AIMD simulations. More strikingly, the diffusivity data from NS experiments is of the same order of magnitude as that from AIMD simulations albeit with a larger activation energy. Below 800 K, the AIMD simulations do not generate reliable mean square displacements and thus the  $T_a$  crossover is not established from the diffusivity data alone. It is likely that below the crossover temperature, a different activated mechanism may be more dominant as shown in  $\text{TiH}_x$ <sup>15</sup>. The PFG-NMR measurements on  $\text{YH}_{1.91}$  reveal a considerably lower diffusivity that may be attributed to a lower stoichiometry. The sparsity of AIMD diffusivity data does not allow for an unambiguous identification of possible superionic transition temperatures. Future work with larger systems and more enhanced phase space sampling will draw out the transition temperatures more clearly.

## Discussion

The observed activation energy of 0.83 eV includes energy for generating the thermal vacancies and for diffusion across the *tet* sites. A previous DFT calculation at 0 K predicts a diffusion barrier of 0.64 eV for a *tet-oct* pathway<sup>29</sup>, which is smaller than the current value; however, when the formation energy of the vacancies is included, the total activation energy becomes more than 2 eV. Transition state calculations based on a static energy



landscape thus overestimate the average energy/migration barrier. At high temperatures, the effective barrier will include both enthalpic and entropic effects. As observed in the AIMD simulations, H atoms take a plethora of trajectories; with increasing temperatures, there are large amplitude vibrations followed by displacements that do not necessarily be described by a single reaction coordinate. The hopping itself in many instances is a complex collective process that involves the participation of several diffusing atoms simultaneously. The effective diffusion barrier thus needs to be assessed in a free energy landscape perspective that involves both vibrational and configurational entropy, the latter, which is non-trivial to estimate. When thermal energy itself is sufficient to generate a large number of displaced atoms, a static rigid potential energy landscape loses its traditional meaning; instead, a free energy landscape may be more appropriate as a dynamic or undulating construct that accommodates several competing and statistically different pathways.

Thermal effects may be incorporated in an ad hoc manner in the traditional landscape sampling methods by incorporating random displacements as done by Novoselov and Yanilkin<sup>15</sup> in their investigation of hydrogen diffusion in titanium dihydrides that showed distinct low and high temperature behaviors. Accelerator molecular dynamics (AMD) methods<sup>52</sup> are also better suited for analyzing activated diffusion mechanisms when the final state points are not known a priori. Nevertheless, methods such as AMD still rely on transition state theory that presume separate vibrational and diffusive timescales<sup>53</sup>. The dynamical trajectories from AIMD simulations reveal a somewhat tightly coupled vibrational and diffusive motion that may render the landscape methods less applicable above the order-disorder transition temperature. A fundamental partitioning of entropy into separate diffusive and vibrational contributions may also lead to erroneous results. The methods we have used here (MSD and VAC), on the other hand, do not presume any specific pathway or depend on the transition state theory, and thus can be considered as more reliable predictors of complex atomic diffusion processes within the simulation timescale. Given the close association between atomic vibrations and diffusion, it would be interesting to assess the role of specific phonons branches<sup>54–56</sup> in the transport processes, particularly above  $T_a$ .

The diffusion process identified in our study involves cooperative motion of many atoms, which is nearly identical to that in fluorite superionic conductors. In an earlier work, Annamareddy and Eapen<sup>26</sup> introduced the concept of *thermal jamming* to explain the dynamic origin of superionicity in *fcc* materials, which have the highest atomic packing density. Jamming can occur when accessible sites are energetically unfavorable such as the *oct* sites in fluorites or when mobile ions exceed the number of favorable sites. Jamming to a more perfect *fcc* structure through cation disorder is also seen to be responsible for increased oxygen diffusion in  $\text{Gd}_2\text{Zr}_2\text{O}_7$ <sup>57</sup>. At low levels of cation disorder, the oxygen atoms assume a defected fluorite lattice position with seven atoms sharing the eight available locations. With increasing cation disorder, the oxygen atoms tend to assume a nearly perfect cubic lattice structure with random occupancy, which enables a rapid increase in the diffusivity through concerted jumps across the lattice sites as in perfect fluorites with minimal transit through the *oct* sites<sup>57</sup>. In  $\text{YH}_2$  too, the *oct* sites are revealed to act as transitional states and not as true minima. Given the similarities, the thermal jamming perspective can be applied to  $\text{YH}_2$  to explain the large increase in H diffusivity with temperature. Based on the thermal jamming concept, we expect the collective string-like hopping process to diminish for sub stoichiometric compositions ( $x < 2$ ) and vice versa for super stoichiometric compositions ( $x > 2$ ). Indeed, most experimental data on  $\text{YH}_x$  portray this general trend and the rapid rise with increasing  $x$  can be attributed to the system being driven to the jammed state ( $x \geq 2$ ) without the need to presume an increased occupancy at the *oct* sites as done in the past.

## Conclusion

Ab initio molecular dynamics simulations are conducted to explore the diffusional properties of hydrogen in  $\text{YH}_2$  at high temperatures. An order – disorder transition takes place at  $T_a \sim 800$  K above which hydrogen atoms diffuse through a collective string-like mechanism. With the temperature increasing above 800 K, spontaneous thermally-generated vacancies are observed, which indicates that H atoms are being thermally displaced from their original positions. Rapid increase in the diffusivity is noted above 800 K through cooperative hops across the native *tet* sites with an activation energy of 0.83 eV; at the highest temperature of 1200 K, the diffusivity approaches that of a liquid state. Our study reveals several close similarities between the dynamics of the hydrogen atoms in yttrium metal hydride and anions in fluorite superionic conductors. We further confirm that the *oct* sites are mostly unoccupied even though channeling through them is the most favored pathway between lattice hops suggesting that *oct* sites are merely transitional states and not true energy minimas. Our work lays the groundwork for future investigations of the complex hydrogen dynamics in metal hydrides at high temperatures.

## Methods

### AIMD simulations

All the calculations were performed using the AIMD method implemented in the Vienna Ab initio Simulation Package (VASP); specifically, the core-valence electron interactions were treated with the projected augmented wave method (PAW)<sup>33,58,59</sup>, with PAW potentials sourced from the VASP PAW PBE 54 library. As in a previous DFT investigation on  $\text{YH}_2$ , the generalized gradient approximation (GGA) in the formulation of Perdew-Burke-Ernzerh (PBE) was employed for the exchange-energy correlation<sup>60,61</sup>; several test runs with AM05 functional<sup>62</sup> yielded very similar structural properties (see supplementary information). For yttrium, valence electron configuration  $4s^2 4p^6 4d^2 5s^1$  includes semicore effects. For hydrogen, the valence electron configuration is  $1s^1$ . Computations were performed on a  $3 \times 3 \times 3$   $\text{YH}_2$  supercell having 324 atoms. Partial electronic occupancies were handled using the first-order Methfessel–Paxton smearing method<sup>63</sup> with a smearing width of 0.2 eV, which helps improve convergence by smoothing the electronic density of states near the Fermi level. A plane wave basis cutoff of 500 eV and a Monkhorst-pack grid of  $1 \times 1 \times 1$  centered at  $\Gamma$  point within the Brillouin zone were

sufficient to have an error of 2 meV/atom or less between configurations. The convergence for electronic loop was set at  $10^{-6}$  eV. The initial atomic positions, based on the fluorite lattice structure, were relaxed using static DFT calculation to minimize forces below  $10^{-2}$  eV/Å. For the AIMD simulations, an NPT ensemble was used with the Langevin thermostat and Parrinello-Rahman barostat<sup>64,65</sup>. The AIMD simulations were conducted with a time step of 1 fs at zero pressure for temperatures ranging from 300 to 1200 K. After obtaining the relaxed lattice parameters, a canonical NVT ensemble was used for 30 ps with a Nose-Hoover thermostat with other parameters remaining unchanged.

### String analysis

We define strings as a set of ions constituted by pairs of mobile atoms where one atom gets replaced by another over a certain time interval<sup>47,48</sup>. Two atoms,  $i$  and  $j$ , are considered to form part of a string if their relative displacement satisfies one of the following conditions  $|\mathbf{r}_i(t) - \mathbf{r}_j(0)| < \delta$  or  $|\mathbf{r}_j(t) - \mathbf{r}_i(0)| < \delta$ ; in this analysis, we set the displacement threshold  $\delta$  as 1 Å. The string analysis is performed for the most mobile atoms in the system (top 20%) that are the most likely to exhibit cooperative, string-like motion.

### Data availability

Data generated from the current study are available from the corresponding author upon reasonable request.

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### References

- CO<sub>2</sub> emissions. Avoided by nuclear by country or region, 1971–2022. *Int. Energy Agency* (2023).
- Black, G., Shropshire, D., Araújo, K. & van Heek, A. Prospects for nuclear microreactors: A review of the technology, economics, and regulatory considerations. *Nucl. Technol.* 1–20 (2022).
- Vetrano, J. B. Hydrides as neutron moderator and reflector materials. *Nucl. Eng. Des.* **14**, 390 (1971).
- Shivprasad, A. P. et al. Advanced Moderator Material Handbook (FY23 Version). (Los Alamos National Laboratory (LANL), Los Alamos, NM (United States), United States (2023)).
- Hu, X. & Terrani, K. A. Thermomechanical properties and microstructures of yttrium hydride. *J. Alloys Compd.* **867**, 158992 (2021).
- Cinbiz, M. N., Taylor, C. N., Luther, E., Trellue, H. & Jackson, J. Considerations for hydride moderator readiness in microreactors. *Nucl. Technol.* **209**, S136–S145 (2023).
- Barnes, R. G. In *Hydrogen in Metals III: Properties and Applications* (ed Helmut Wipf). (Springer, 1997).
- Stuhr, U., Steinbinder, D., Wipf, H. & Frick, B. Hydrogen diffusion in f.c.c. TiH<sub>x</sub> and YH<sub>x</sub>: two distinct examples for diffusion in a concentrated lattice gas. *EPL* **20**, 117–123 (1992).
- Sköld, K. in *Hydrogen in Metals I: Basic Properties* (eds Georg Alefeld & Johann Völkl). Springer Berlin Heidelberg, 267 (1978).
- Wipf, H., Kappesser, B. & Werner, R. Hydrogen diffusion in titanium and zirconium hydrides. *J. Alloys Compd.* **310**, 190 (2000).
- Cotts, R. M. Hydrogen diffusion studies using nuclear magnetic resonance. *Ber. Bunsenges. Phys. Chem.* **76**, 760 (1972).
- Ramirez-Cuesta, A. J., Jones, M. O. & David, W. I. F. Neutron scattering and hydrogen storage. *Mater. Today*. **12**, 54 (2009).
- Majer, G., Gottwald, J., Peterson, D. T. & Barnes, R. G. Model-independent measurements of hydrogen diffusivity in the yttrium dihydrides. *J. Alloys Compd.* **330–332**, 438–442 (2002).
- Gissler, W. & Rother, H. Theory of the quasielastic neutron scattering by hydrogen in bcc metals applying a random flight method. *Physica* **50**, 380 (1970).
- Novoselov, I. I. & Yanilkin, A. V. Hydrogen diffusion in titanium dihydrides from first principles. *Acta Mater.* **153**, 250 (2018).
- Barnes, R. G. et al. Dynamical evidence of hydrogen sublattice melting in metal-hydrogen systems. *J. Less-Common Met.* **129**, 279–285 (1987).
- Barnes, R. G. et al. Dynamical evidence for a change in hydrogen-diffusion behavior in transition-metal hydrides at high temperatures. *Phys. Rev. B* **35**, 890 (1987).
- Barnes, R. G. et al. Normal and anomalous nuclear spin-lattice relaxation at high temperatures in Sc-H(D), Y-H, and Lu-H solid solutions. *Phys. Rev. B* **51**, 3503 (1995).
- Barnfather, K. J. et al. A study of high temperature hydrogen diffusion in YH<sub>1.98</sub> and V<sub>0.25</sub>Nb<sub>0.75</sub>H<sub>0.20</sub>. *Z. Phys. Chem.* **164**, 935–940 (1989).
- Baker, D. B., Conradi, M. S., Norberg, R. E., Barnes, R. G. & Torgeson, D. R. Explanation of the high-temperature relaxation anomaly in a metal-hydrogen system. *Phys. Rev. B* **49**, 11773 (1994).
- Cotts, R. M. Would hydrogen pairing at high temperatures account for nuclear magnetic resonance relaxation anomalies in metal hydrides? *J. Less-Common Met.* **172–174**, 467 (1991).
- Phua, T. T. et al. Paramagnetic impurity effects in NMR determinations of hydrogen diffusion and electronic structure in metal hydrides. Gd<sup>3+</sup> in YH<sub>2</sub> and LaH<sub>2.25</sub>. *Phys. Rev. B* **28**, 6227–6250 (1983).
- Shivprasad, A. P. et al. Thermophysical properties of high-density, sintered monoliths of yttrium dihydride in the range 373–773 K. *J. Alloys Compd.* **850**, 156303 (2021).
- Hull, S. Superionics: crystal structures and conduction processes. *Rep. Prog. Phys.* **67**, 1233 (2004).
- Annamareddy, A. & Eapen, J. Disorder and dynamic self-organization in stoichiometric UO<sub>2</sub> at high temperatures. *J. Nucl. Mater.* **483**, 132–141 (2017).
- Annamareddy, A. & Eapen, J. Low dimensional string-like relaxation underpins superionic conduction in fluorites and related structures. *Sci. Rep.* **7**, 44149 (2017).
- Trofimov, A. A., Hu, X., Wang, H., Yang, Y. & Terrani, K. A. Thermophysical properties and reversible phase transitions in yttrium hydride. *J. Nucl. Mater.* **542**, 152569 (2020).
- Mehta, V. K. et al. A density functional theory and neutron diffraction study of the ambient condition properties of sub-stoichiometric yttrium hydride. *J. Nucl. Mater.* **547**, 152837 (2021).
- Tunes, M. A. et al. Challenges in developing materials for microreactors: A case-study of yttrium dihydride in extreme conditions. *Acta Mater.* **280**, 120333 (2024).
- Sundar, A., Huang, Y., Yu, J. & Cinbiz, M. N. The impacts of charge transfer, localization, and metallicity on hydrogen retention and transport capacity. *Int. J. Hydrog. Energy*. **47**, 20194–20204 (2022).
- Annamareddy, V. A., Nandi, P. K., Mei, X. & Eapen, J. Waxing and waning of dynamical heterogeneity in the superionic state. *Phys. Rev. E* **89**, 010301 (2014).
- Kresse, G. & Furthmüller, J. Efficient iterative schemes for Ab initio total-energy calculations using a plane-wave basis set. *Phys. Rev. B* **54**, 11169 (1996).

33. Kresse, G. & Joubert, D. From ultrasoft pseudopotentials to the projector augmented-wave method. *Phys. Rev. B*. **59**, 1758 (1999).
34. Daou, J. N. & Vajda, P. Hydrogen ordering and metal-semiconductor transitions in the system  $\text{YH}_{2+x}$ . *Phys. Rev. B*. **45**, 10907 (1992).
35. Chapman, C. W. et al. Thermal neutron scattering measurements and modeling of yttrium-hydrides for high temperature moderator applications. *Ann. Nucl. Energy*. **157**, 108224 (2021).
36. Wendt, H. & Abraham, F. F. Empirical criterion for the glass transition region based on Monte Carlo simulations. *Phys. Rev. Lett.* **41**, 1244 (1978).
37. Celtek, M., Sengul, S., Domekeli, U. & Guder, V. Dynamical and structural properties of metallic liquid and glass  $\text{Zr}_{48}\text{Cu}_{36}\text{Ag}_8\text{Al}_8$  alloy studied by molecular dynamics simulation. *J. Non-Cryst. Solids*. **566**, 120890 (2021).
38. Hutchings, M. T. et al. Investigation of thermally induced anion disorder in fluorites using neutron scattering techniques. *J. Phys. C Solid State Phys.* **17**, 3903 (1984).
39. Stukowski, A. Visualization and analysis of atomistic simulation data with OVITO—the open visualization tool. *Model. Simul. Mater. Sci. Eng.* **18**, 015012 (2010).
40. Mohn, C. E., Stølen, S., Norberg, S. T. & Hull, S. Oxide-ion disorder within the high temperature  $\delta$  phase of  $\text{Bi}_2\text{O}_3$ . *Phys. Rev. Lett.* **102**, 155502 (2009).
41. Van Hove, L. Correlations in space and time and born approximation scattering in systems of interacting particles. *Phys. Rev.* **95**, 249 (1954).
42. McQuarrie, D. A. *Statistical Mechanics* (University Science Books, 2000).
43. Boon, J. P. & Yip, S. *Molecular Hydrodynamics* (Dover, 1991).
44. Boothroyd, A. T. *Principles of Neutron Scattering from Condensed Matter* (Oxford University Press, 2020).
45. Ediger, M. D. Spatially heterogeneous dynamics in supercooled liquids. *Ann. Rev. Phys. Chem.* **51**, 99 (2000).
46. Chaudhuri, P., Berthier, L. & Kob, W. Universal nature of particle displacements close to glass and jamming transitions. *Phys. Rev. Lett.* **99**, 060604 (2007).
47. Gebremichael, Y., Vogel, M. & Glotzer, S. C. Particle dynamics and the development of string-like motion in a simulated monoatomic supercooled liquid. *J. Chem. Phys.* **120**, 4415–4427 (2004).
48. Donati, C. et al. Stringlike cooperative motion in a supercooled liquid. *Phys. Rev. Lett.* **80**, 2338 (1998).
49. Sau, K., Ikeshoji, T., Kim, S., Takagi, S. & Orimo, S. I. Comparative molecular dynamics study of the roles of anion–cation and cation–cation correlation in cation diffusion in  $\text{Li}_2\text{B}_{12}\text{H}_{12}$  and  $\text{LiCB}_{11}\text{H}_{12}$ . *Chem. Mater.* **33**, 2357–2369 (2021).
50. Xu, M., Ding, J. & Ma, E. One-dimensional stringlike cooperative migration of lithium ions in an ultrafast ionic conductor. *Appl. Phys. Lett.* **101** (2012).
51. Cakmak, E. et al. Hydrogen motion in near stoichiometric yttrium dihydride at elevated temperatures. *J. Nucl. Mater.* **593**, 154972 (2024).
52. Voter, A. F., Montalenti, F. & Germann, T. C. Extending the time scale in atomistic simulation of materials. *Ann. Rev. Mater. Res.* **32**, 321–346 (2002).
53. Kushima, A., Eapen, J., Li, J., Yip, S. & Zhu, T. Time scale bridging in atomistic simulation of slow dynamics: viscous relaxation and defect activation. *Eur. Phys. J. B*. **82**, 271 (2011).
54. Raj, A. & Eapen, J. Phonon dispersion using the ratio of zero-time correlations among conjugate variables: computing full phonon dispersion surface of graphene. *Comput. Phys. Commun.* **238**, 124 (2019).
55. Raj, A. & Eapen, J. Deducing phonon scattering from normal mode excitations. *Sci. Rep.* **9**, 7982 (2019).
56. Raj, A. & Eapen, J. Exact diagonal representation of normal mode energy, occupation number, and heat current for phonon-dominated thermal transport. *J. Phys. Chem.* **151**, 104110 (2019).
57. Annamareddy, A. & Eapen, J. Decoding ionic conductivity and reordering in cation-disordered pyrochlores. *Philos. Trans. R. Soc. A*. **379**, 20190452 (2021).
58. Kresse, G. & Hafner, J. Ab initio molecular dynamics for open-shell transition metals. *Phys. Rev. B*. **48**, 13115 (1993).
59. Blöchl, P. E. Projector augmented-wave method. *Phys. Rev. B*. **50**, 17953 (1994).
60. Perdew, J. P., Burke, K. & Ernzerhof, M. Generalized gradient approximation made simple. *Phys. Rev. Lett.* **77**, 3865–3868 (1997).
61. Ernzerhof, M. & Scuseria, G. E. Assessment of the Perdew–Burke–Ernzerhof exchange–correlation functional. *J. Chem. Phys.* **110**, 5029 (1999).
62. Mattsson, A. E. et al. The AM05 density functional applied to solids. *J. Chem. Phys.* **128** (2008).
63. Methfessel, M. & Paxton, A. T. High-precision sampling for Brillouin-zone integration in metals. *Phys. Rev. B*. **40**, 3616–3621 (1989).
64. Parrinello, M. & Rahman, A. Crystal structure and pair potentials: A molecular-dynamics study. *Phys. Rev. Lett.* **45**, 1196–1199 (1980).
65. Parrinello, M. & Rahman, A. Polymorphic transitions in single crystals: A new molecular dynamics method. *J. Appl. Phys.* **52**, 7182–7190 (1981).

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## Author contributions

YH performed the simulations, analyses and prepared the first draft of the manuscript; all the authors contributed to writing the final manuscript. The project was conceived by JE, MNC and JY. Since the manuscript is co-authored by UT-Battelle, LLC, and partially supported by the US DOE, the US government retains and the publisher, by accepting the article for publication, acknowledges that the US government retains a nonexclusive, paid-up, irrevocable, worldwide license to publish or reproduce the published form of this manuscript, or allow others to do so, for US government purposes. DOE will provide public access to these results of federally sponsored research in accordance with the DOE Public Access Plan (<http://energy.gov/downloads/doe-public-access-plan>).

## Declarations

## Competing interests

The authors declare no competing interests.

### Additional information

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