

Entrapment of VOCs in Defective UiO-66 MOF: Ab Initio Molecular Dynamics Simulations at 300K

Authors: Brianne Boyd, Scott Bobbitt, Dr. Deep Choudhuri



NM Partnerships

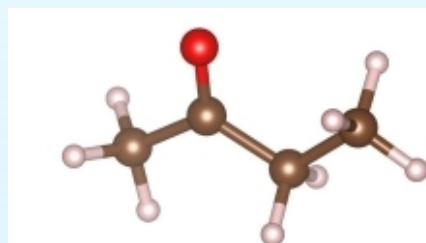
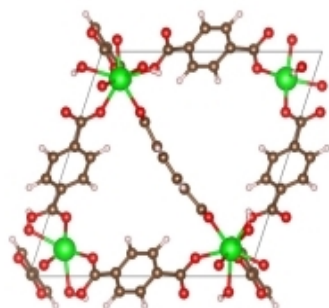


Fall 2024 LDRD Virtual Poster Session

Introduction and Motivation

Metal Organic Frameworks (MOF) have been shown to be effective for Volatile Organic Compound (VOCs) absorption. 2-Butanone is one such VOC that is typically found in lung cancer patients. If a MOF could be shown to interact favorably with this VOC, it could be used as an indicator.

UiO-66 is well known for its use in gas capture due to its structural and thermal stability. It is composed of $\text{Zr}_6\text{O}_4(\mu_3\text{-OH})_4$ nodes and BDC linkers. Frequently, defects in UiO-66 manifest as missing BDC linkers.



This project asks if UiO-66 could be effective in capturing 2-butanone molecules and how we can understand the influences of defects, temperature, and moisture on this interaction.

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Approach

-Observation: UiO-66 has proven to be effective for certain VOC absorption

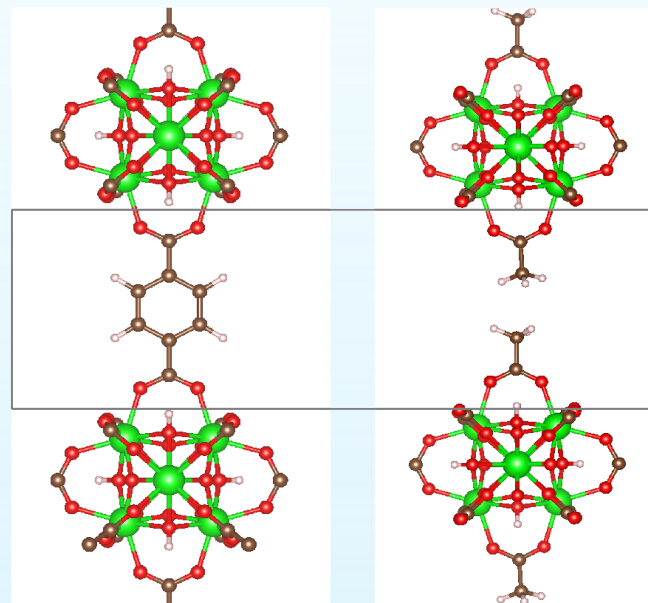
-Curiosity: Could UiO-66 be used to absorb 2-Butanone?

-Application: 2-Butanone is often found in lung cancer patients. If a MOF were to be found to interact favorably with the VOC, it could be used to determine if butanone is found in their breath.

Capping species at missing linker sites

w/ BDC linker

Acetate



Ab initio molecular dynamics (AIMD) was performed using the Vienna ab initio simulation package (VASP)

The time-evolution of interaction energy between 2-butanone and MOF was quantified at 300K using the expression:

$$\Delta E_{MOF+VOC+water}^{Interaction}(t) = \Delta E_{Defective-MOF+VOC}^t -$$

$$\Delta E_{defective-MOF}^{Time Avg} - n \times \Delta E_{VOC}^{Time Avg},$$

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Current Status

Research has been completed and a paper is currently under review with Langmuir

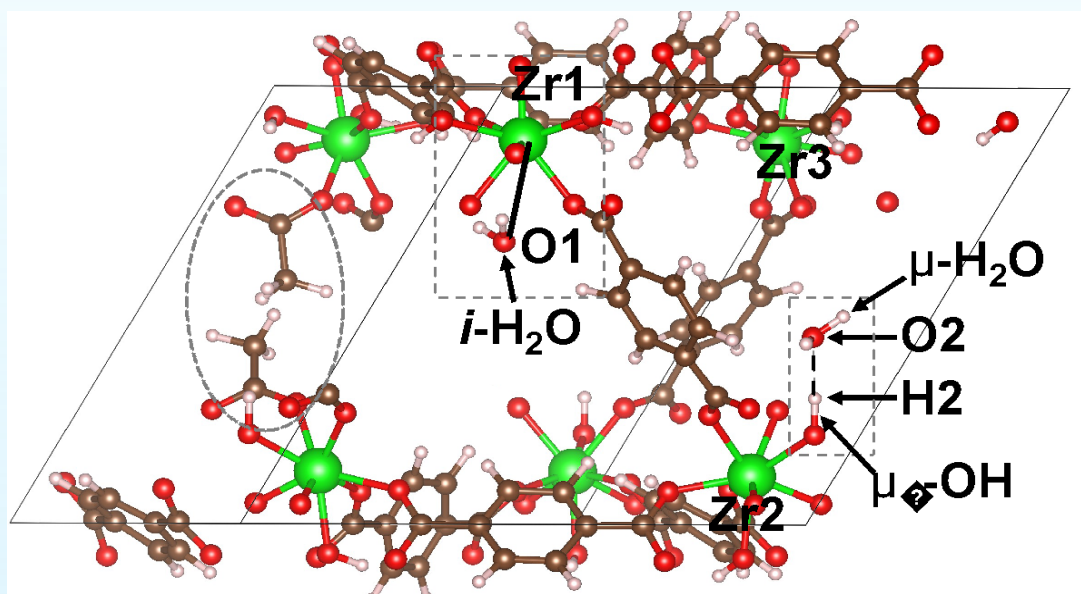
Four molecule loading with preexisting H₂O showed favorable energies.

Bonds were formed from a hydrogen bond tween 2-butanone and H₂O, and then to the Hydroxyl Zr Nodes

At higher loadings, 2 butanone has been observed vacating the octahedral cage.

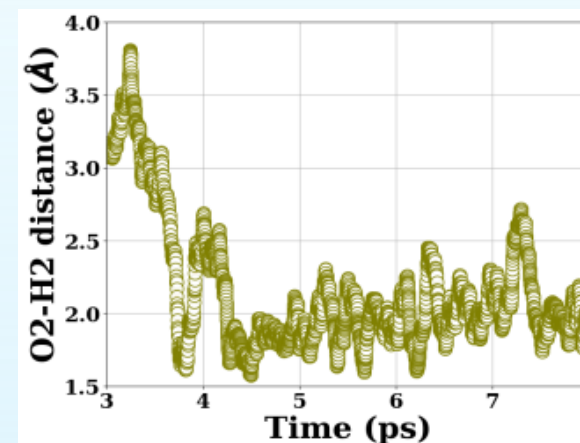
Results

AIMD snapshot of defective UiO-66 at 300K

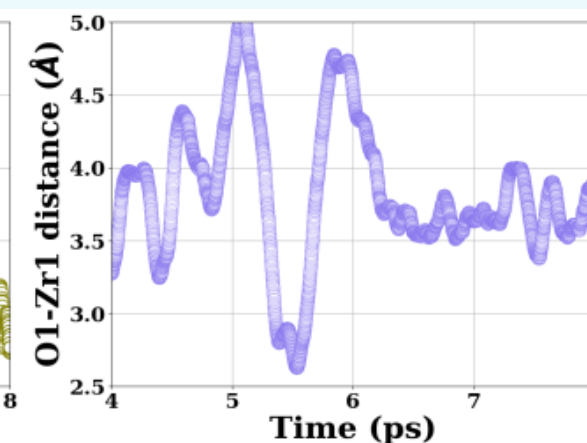


Energetically favorable structure at 300K: Missing BDC linker compensated by acetate groups and two neighboring water molecules (μ -H₂O & i -H₂O)

μ -H₂O (near μ_3 -OH)

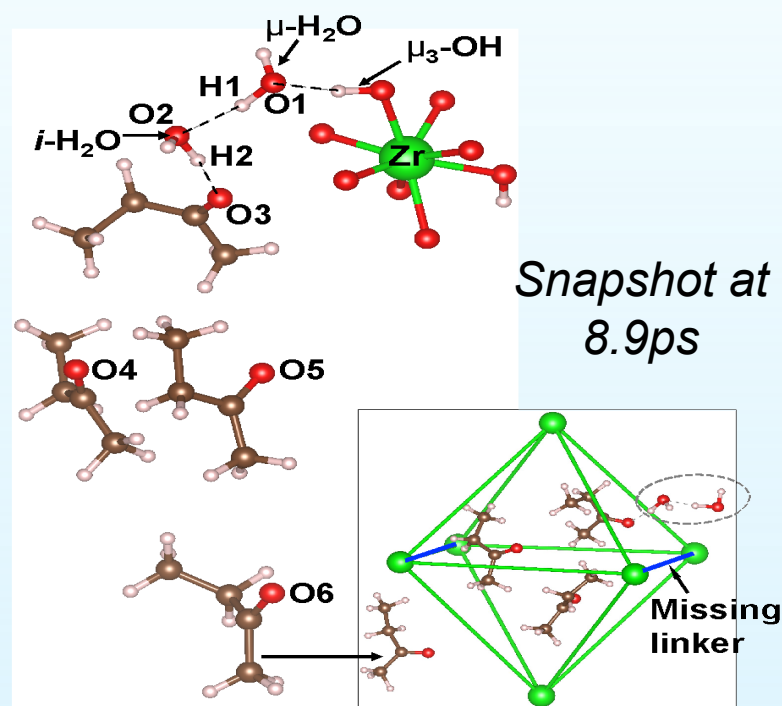


i -H₂O

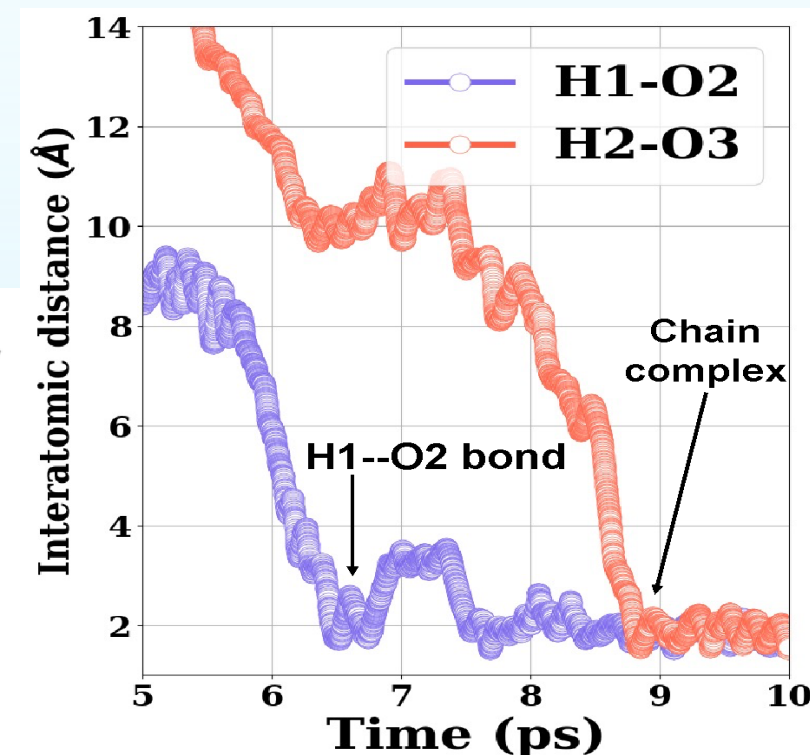
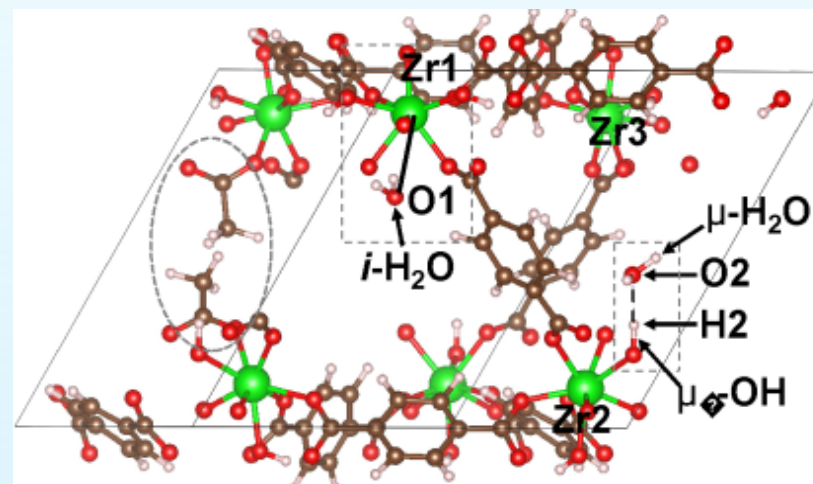


Results

Trapping mechanism of 2-butanone VOC within defective UiO-66



Formation of a stable chain complex between μ_3 -OH, μ H₂O, i-H₂O, and 2-Butanone



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Impact of Work

This project will provide insights to the usage of 2-butanone as a biomarker by encapsulating the compound within the voids of UiO-66

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Challenges and Risks / Next Steps and Future Work

Next Steps:

Consider UiO-66 for absorption of CO₂