

Entrapment of VOCs in Defective UiO-66 MOF: Ab Initio Molecular Dynamics Simulations at 300K

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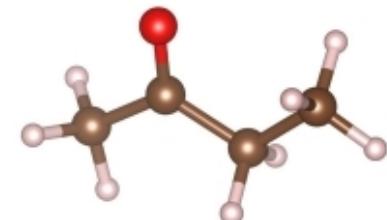
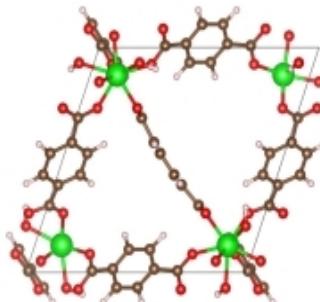


Fall 2024 LDRD Virtual Poster Session

Introduction and Motivation

Metal Organic Frameworks (MOF) have been shown to be effective for Volatile Organic Compound (VOCs) absorption. 2-Butanone is one such VOC that is typically found in lung cancer patients. If a MOF could be shown to interact favorably with this VOC, it could be used as an indicator.

UiO-66 is well known for its use in gas capture due to its structural and thermal stability. It is composed of $Zr_6O_4(\mu_3-OH)_4$ nodes and BDC linkers. Frequently, defects in UiO-66 manifest as missing BDC linkers.



This project asks if UiO-66 could be effective in capturing 2-butanone molecules and how we can understand the influences of defects, temperature, and moisture on this interaction.

Approach

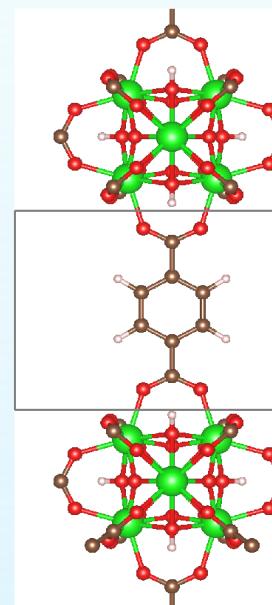
-Observation: UiO-66 has proven to be effective for certain VOC absorption

-Curiosity: Could UiO-66 be used to absorb 2-Butanone?

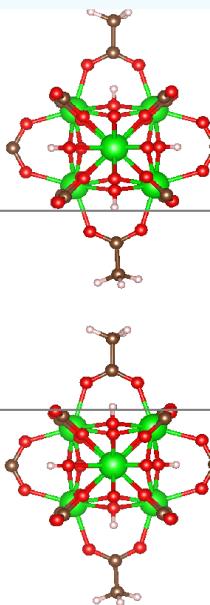
-Application: 2-Butanone is often found in lung cancer patients. If a MOF were to be found to interact favorably with the VOC, it could be used to determine if butanone is found in their breath.

Capping species at missing linker sites

w/ BDC linker



Acetate



Ab initio molecular dynamics (AIMD) was performed using the Vienna *ab initio* simulation package (VASP)

The time-evolution of interaction energy between 2-butanone and MOF was quantified at 300K using the expression:

$$\Delta E_{MOF+VOC+water}^{Interaction}(t) = \Delta E_{Defective-MOF+VOC}^t - \Delta E_{defective-MOF}^{Time Avg} - n \times \Delta E_{VOC}^{Time Avg},$$

Current Status

Research has been completed and a paper is currently under review with Langmuir

Four molecule loading with preexisting H₂O showed favorable energies.

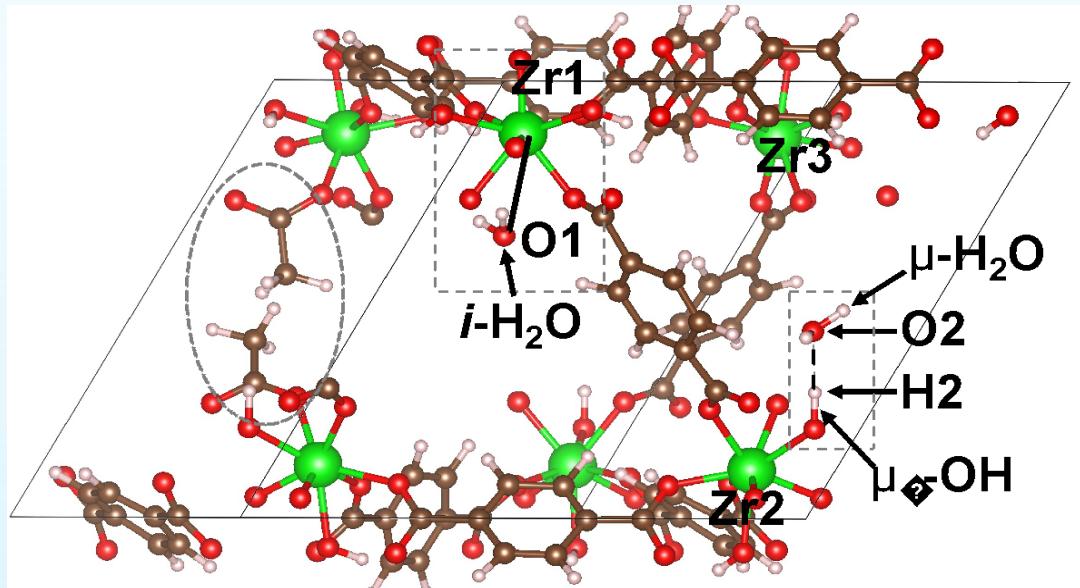
Bonds were formed from a hydrogen bond between 2-butanone and H₂O, and then to the Hydroxyl Zr Nodes

At higher loadings, 2 butanone has been observed vacating the octahedral cage.



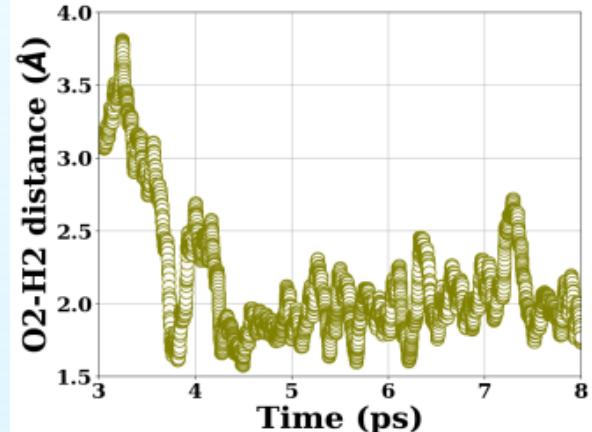
Results

AIMD snapshot of defective UiO-66 at 300K

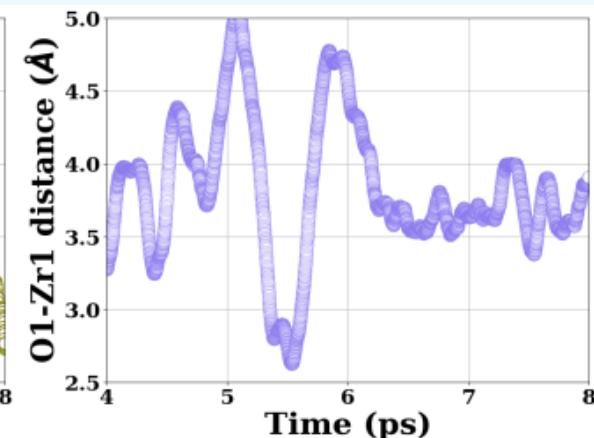


Energetically favorable structure at 300K: Missing BDC linker compensated by acetate groups and two neighboring water molecules (μ -H₂O & *i*-H₂O)

μ -H₂O (near μ_3 -OH)

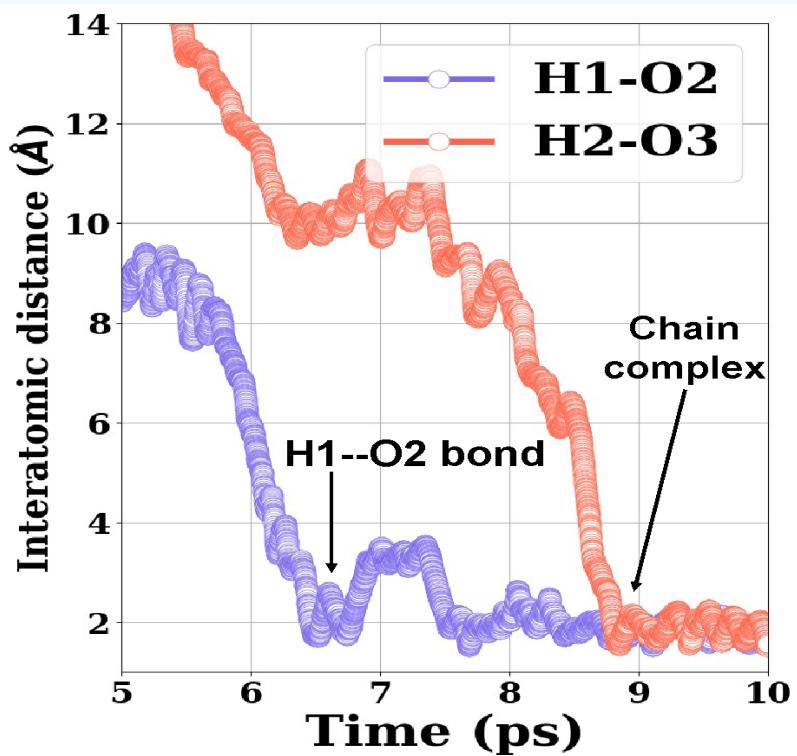
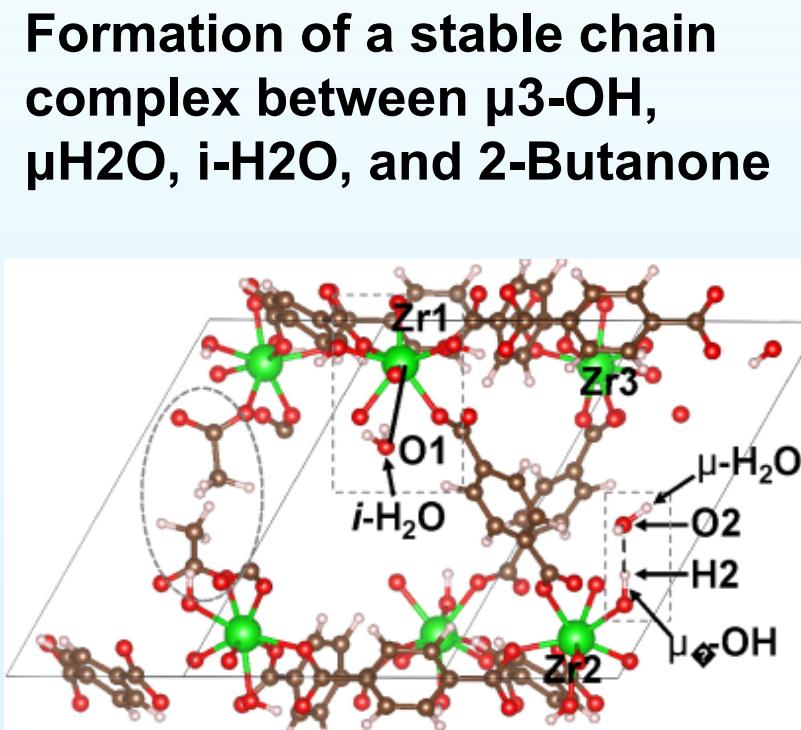
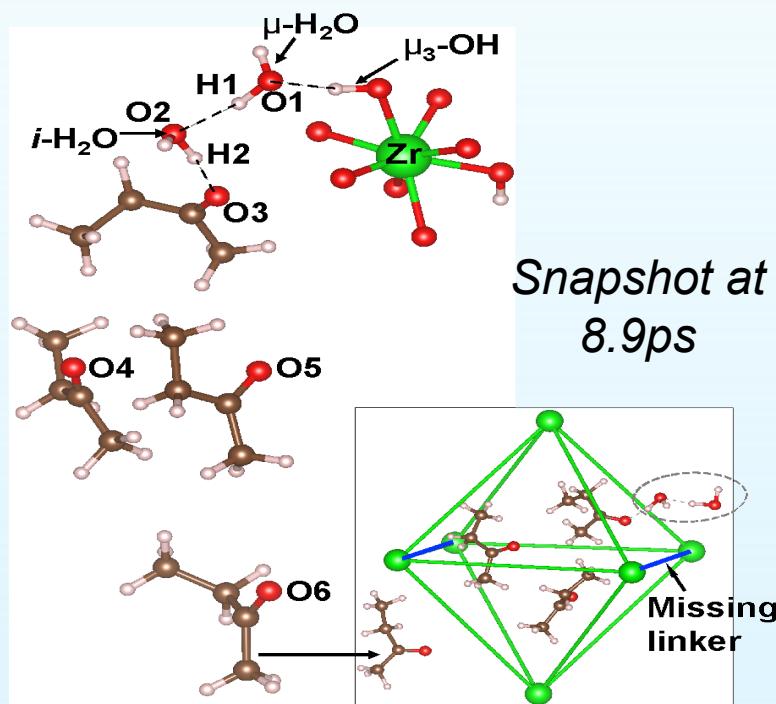


i-H₂O



Results

Trapping mechanism of 2-butanone VOC within defective UiO-66



Impact of Work

This project will provide insights to the usage of 2-butanone as a biomarker by encapsulating the compound within the voids of UiO-66

Challenges and Risks / Next Steps and Future Work

Next Steps:

Consider UiO-66 for absorption of CO₂