



Italian Section of the Combustion Institute

# Understanding the interplay between pilot fuel mixing and auto-ignition chemistry in hydrogen-enriched environment

Taesong Lee, Rajavasanth Rajagegar, Ales Srna  
Sandia National Laboratories

# Understanding the interplay between pilot fuel mixing and auto-ignition chemistry in hydrogen-enriched environment

Taesong Lee<sup>1</sup>, Rajavasanth Rajasegar<sup>1,2</sup>, and Aleš Srna<sup>1</sup>

<sup>1</sup>Combustion Research Facility, Sandia National Laboratories, CA

<sup>2</sup>Department of Mechanical Engineering, Colorado School of Mines, CO

Sponsors: Vehicle Technologies Office (VTO),  
Department of energy (DOE)

2024 40<sup>th</sup> International Symposium on Combustion  
Allianz MiCo Congress Center, Milan, Italy

# Diesel-Piloted Dual Fuel (DPDF) Combustion Strategy

## Diesel-Piloted Dual-Fuel (DPDF) System

Combustion initiated by short pilot injection of highly reactive liquid fuel (diesel) into lean premixed primary fuel-air mixture charge.

## Inhibitive effect of H<sub>2</sub> on n-heptane auto-ignition chemistry

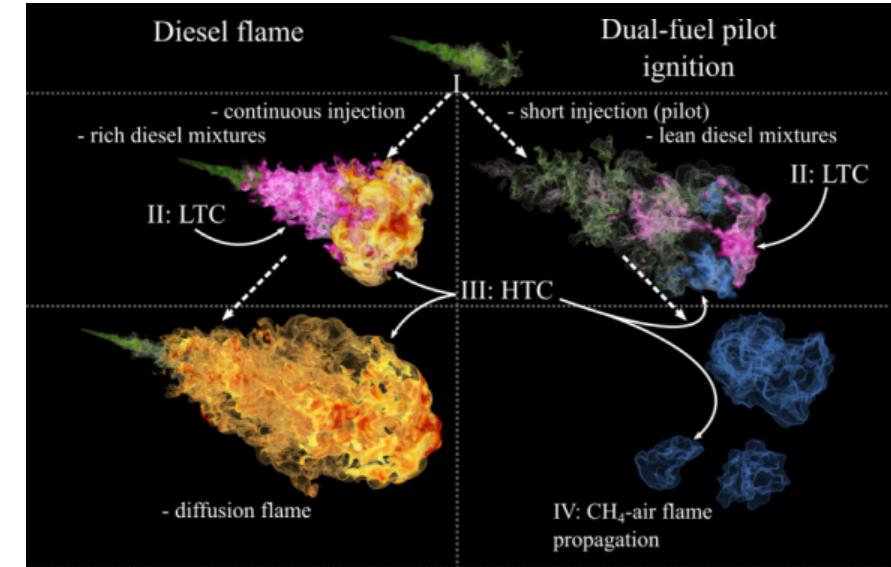
### Combustion in a highly stratified environment

- Temperature: pilot-fuel vaporization cooling
- Equivalence ratio: partial mixing of pilot-fuel

→ Complex physical and chemical effects govern a spatio-temporal evolution of dual fuel auto-ignition process.

However, it is challenging to anticipate the dual-fuel combustion characteristics of H<sub>2</sub> as a primary fuel

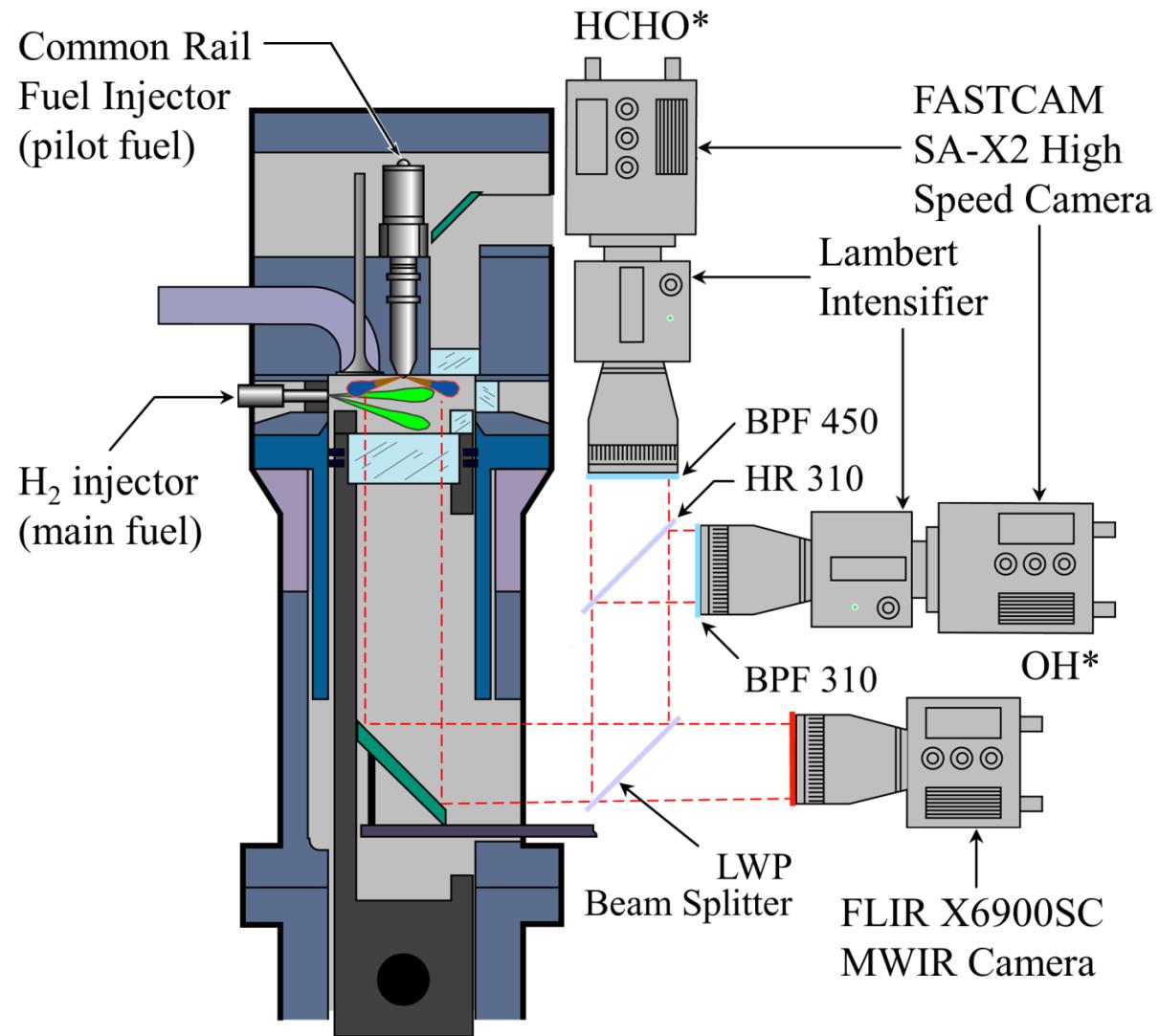
: Low minimum ignition energy (**Pre-ignition**) vs Strong inhibition effect (**Longer ignition delay**)



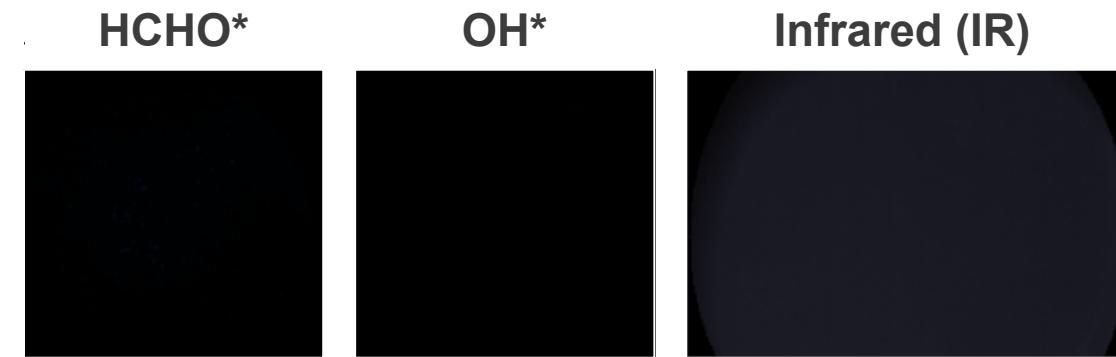
## Research Objective

“Understanding the interplay between the physical and chemical processes that govern ignition of pilot fuel jet in the presence of lean-premixed H<sub>2</sub>/air mixture, complemented with zero-dimensional chemical kinetics and one-dimensional spray dynamics simulation.”

# Experimental Setup



Schematic of heavy-duty optical engine with high-speed imaging setup



## Operating Conditions

Intake conditions  $100\text{ }^\circ\text{C}, 100\text{ kPa}, 21\% \text{ O}_2$

Engine speed  $1200\text{ RPM}$  ( $1\text{ }^\circ\text{CA} \approx 138\text{ }\mu\text{s}$ )

Fuel type  
Pilot fuel: **n-heptane**  
Main fuel: **Hydrogen**

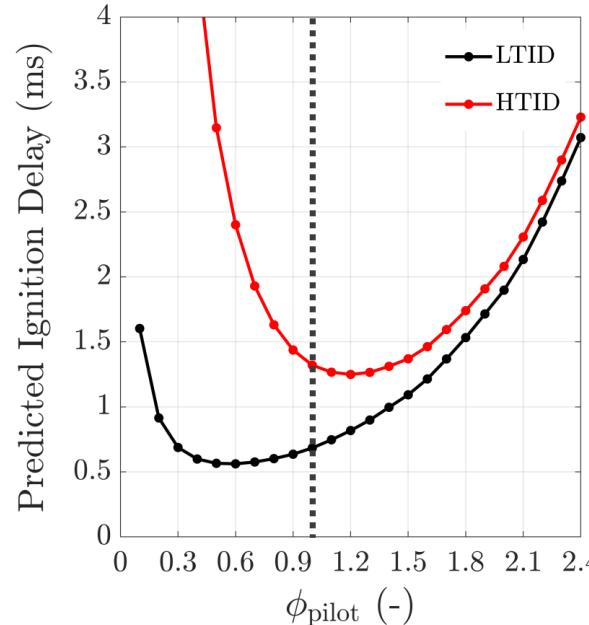
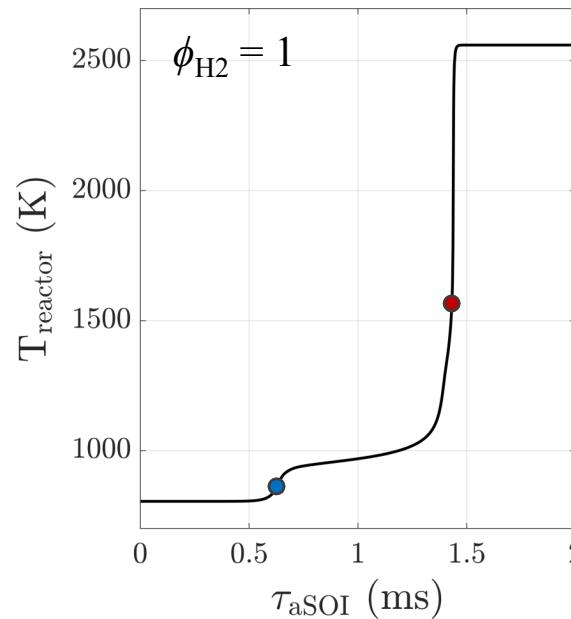
Injection timing  
Pilot: 347 CAD (\*TDC: 360 CAD)  
Main ( $\text{H}_2$ ): 60 CAD (\*homogeneous mixing)

Pilot fuel, Inj. parameters  
**Case 1:**  $P_{\text{inj}}: 800\text{ bar}, t_{\text{inj}}: 760\text{ }\mu\text{s}$  (21 mg)  
**Case 2:**  $P_{\text{inj}}: 800\text{ bar}, t_{\text{inj}}: 500\text{ }\mu\text{s}$  (7.9 mg)  
**Case 3:**  $P_{\text{inj}}: 400\text{ bar}, t_{\text{inj}}: 760\text{ }\mu\text{s}$  (8.8 mg)

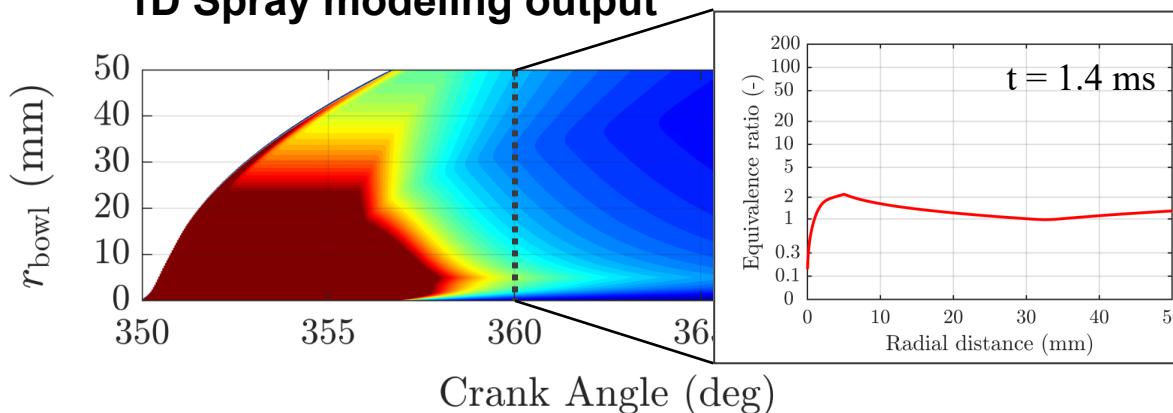
Main fuel,  $\text{H}_2$  Inj. parameters  
 $\phi_{\text{H}_2}: 0$  to  $0.4$  ( $\Delta\phi_{\text{H}_2} = 0.1$ )  
 $P_{\text{inj}}: 40\text{ bar}, t_{\text{inj}}: 711$  to  $2844\text{ }\mu\text{s}$

# Experimental Setup

## 0D CHEMKIN output



## 1D Spray modeling output



## 0D/1D Simulations

### Chemical Kinetics Simulation

ANSYS CHEMKIN-pro software using 0D Closed Homogeneous Reactor (CHR) model, coupled with *LLNL detailed n-heptane mechanism* ver. 3.1 (\*accounted for the pilot fuel vaporization cooling effect)

### Output (0D)

- Temperature
- Volumetric heat release rate
- Chemical species
- Reaction rates

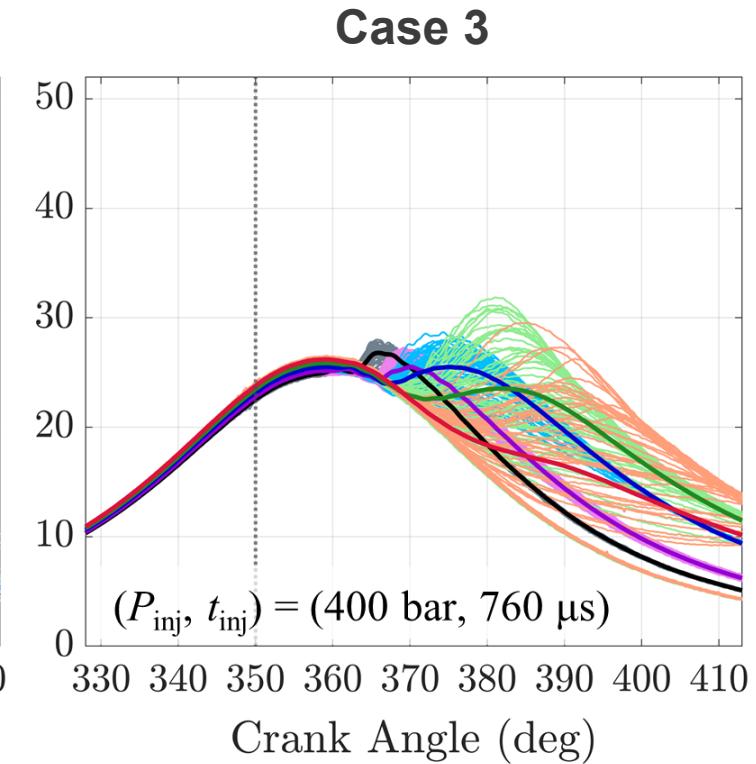
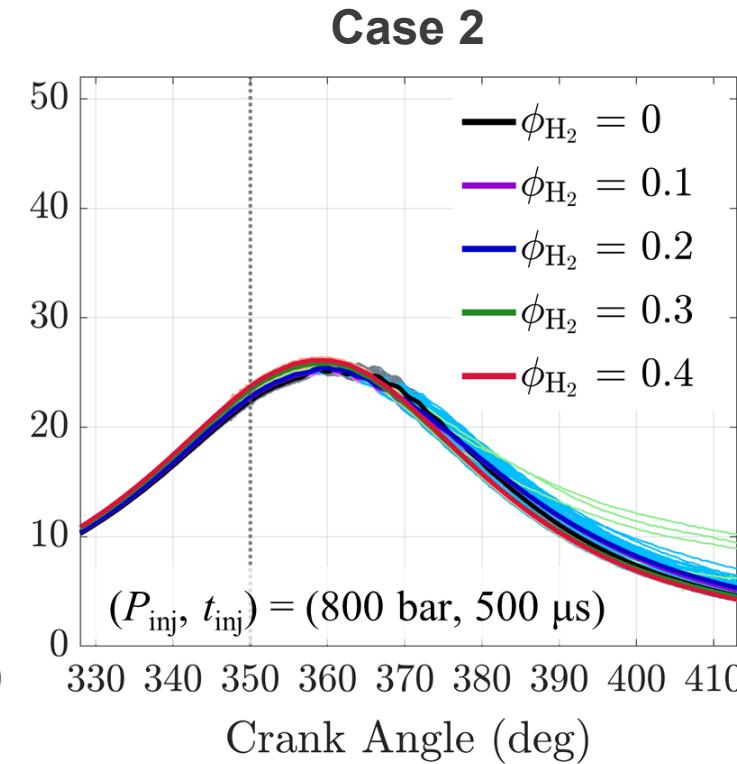
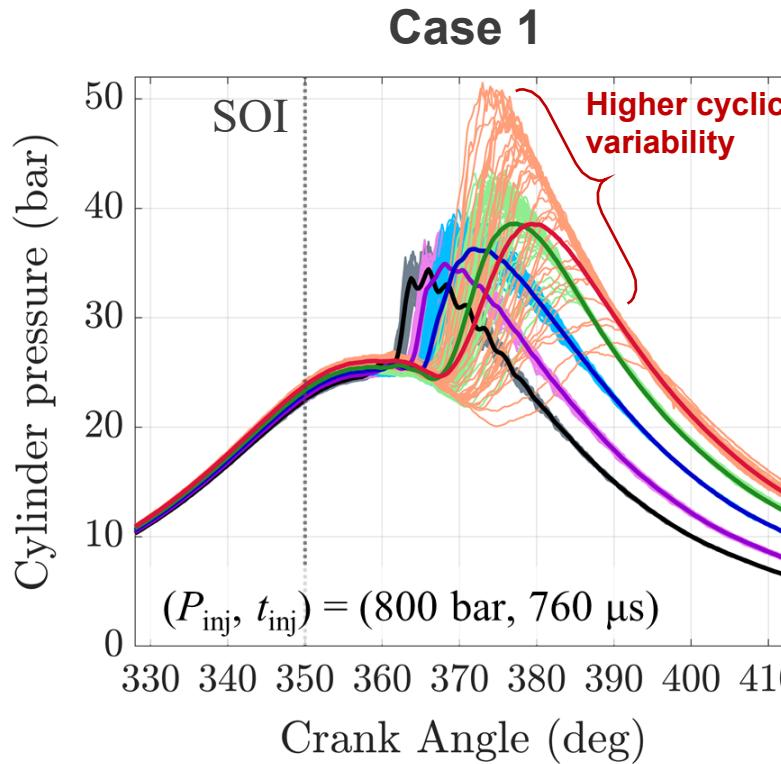
### Computational Spray Modeling

1D jet model with a variable profile approximation for radial mixing and velocity distribution (also known as the *Musculus and Kattke model*)

### Output (1D)

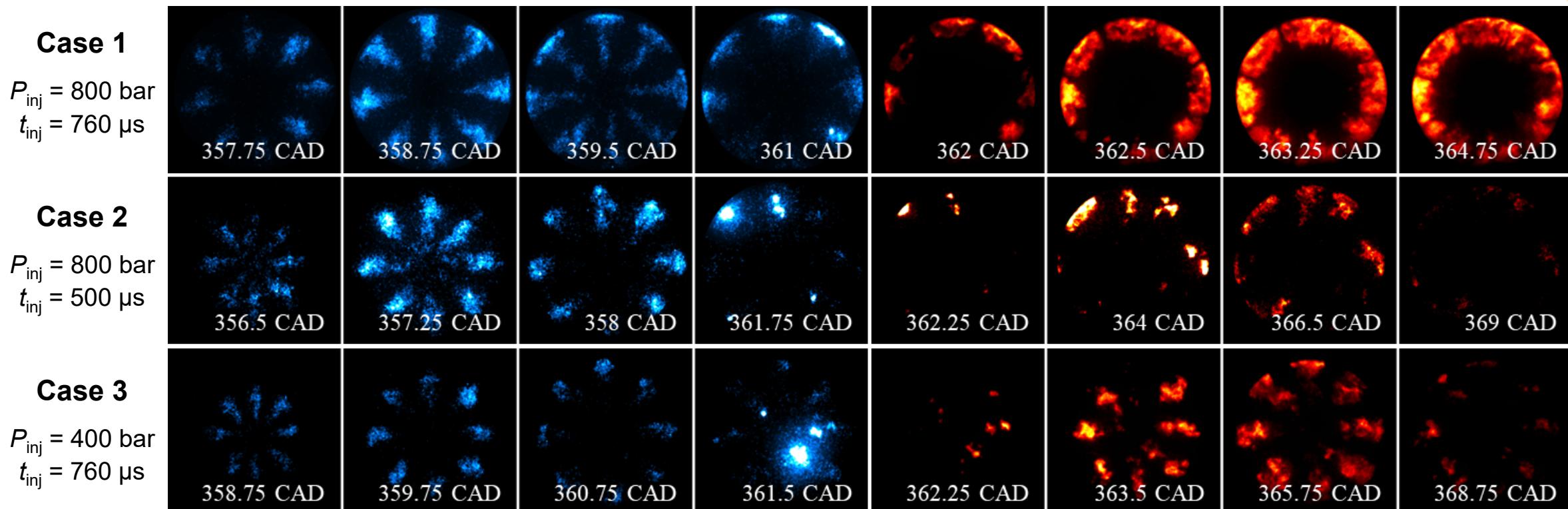
- Equivalence ratio distribution
- Jet penetration length
- Mean velocity
- Air entrainment rate

# Pressure traces with respect to the injection parameters



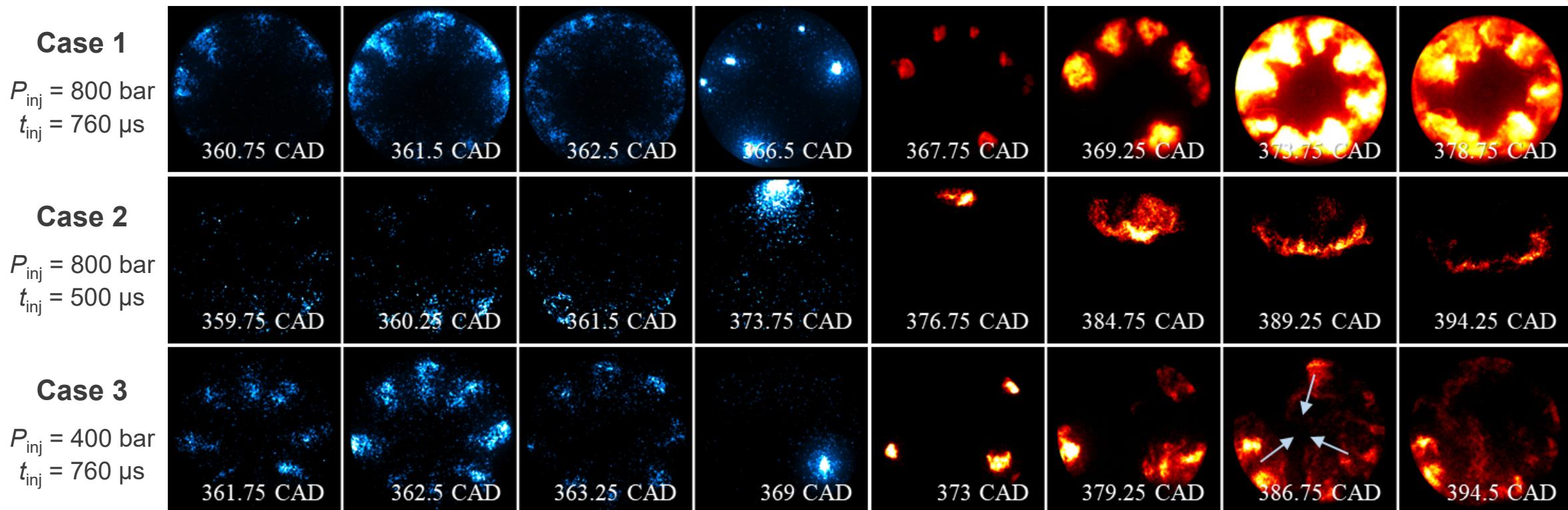
- Increasing the  $H_2$  concentration results in two noticeable effects: **1) prolonged ignition delay** (inhibitive effect on pilot fuel ignition) and **2) higher cyclic variability** (low minimum ignition energy & ultrafast flame speed)
- The impact of reduced fuel mass (~60% relative to Case 1) in Case 2 and 3 manifests as a decrease in the maximum in-cylinder pressure and pressure rise rates.
- Contrary to Case 2, the longer injection can initiate successful ignition in most cycles except for few misfires at  $\phi_{H_2} = 0.4$ , while the influence of  $H_2$  on pilot ignition delays is more pronounced in Case 3.

# Two-stage auto-ignition process of pure pilot injection case



- For the single pilot fuel injection, first-stage ignition starts downstream of the eight fuel jets and propagates towards upstream, followed by near-wall, second-stage combustion.
- Reduced injection duration in Case 2 results in enhanced fuel-air mixing, leading to an earlier onset of LTHR near nozzle (~356 CAD) and a limited flame kernel development (fuel leaning out).
- Lower injection pressure and associated slower mixing contribute to: 1) the delayed onset of LTC (~359 CAD) and 2) second-stage ignition predominantly occurring upstream of the spray tip.

# Impact of hydrogen on two-stage auto-ignition chemistry



- In general, due to the strong inhibitive effect of  $\text{H}_2$ , an increased mixing time induces the delayed onset of low temperature combustion in the vicinity of the wall without recession, and not all pilot fuel jets proceed to the second-stage auto-ignition.
- With reduced amount of pilot fuel injection (Case 2 & 3), the formation of flame kernel is significantly delayed and successfully initiated by few pockets with sufficient pilot fuel concentration.

→ “Importance of stochastic rich fuel pockets”

# Evolution of auto-ignition process: only pilot injection



Case 1

347.0 CAD

$P_{\text{inj}} = 800 \text{ bar}$ ,  $t_{\text{inj}} = 760 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

“LTC regime”

347.0 CAD

$P_{\text{inj}} = 800 \text{ bar}$ ,  $t_{\text{inj}} = 500 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

“HTC regime”

Case 2

347.0 CAD

$P_{\text{inj}} = 400 \text{ bar}$ ,  $t_{\text{inj}} = 760 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

# Evolution of auto-ignition process: H<sub>2</sub> dual-fuel operation



Case 1

347.0 CAD

$P_{\text{inj}} = 800 \text{ bar}$ ,  $t_{\text{inj}} = 760 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

“LTC regime”

Case 2

347.0 CAD

$P_{\text{inj}} = 800 \text{ bar}$ ,  $t_{\text{inj}} = 500 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

“HTC regime”

Case 3

347.0 CAD

$P_{\text{inj}} = 400 \text{ bar}$ ,  $t_{\text{inj}} = 760 \mu\text{s}$



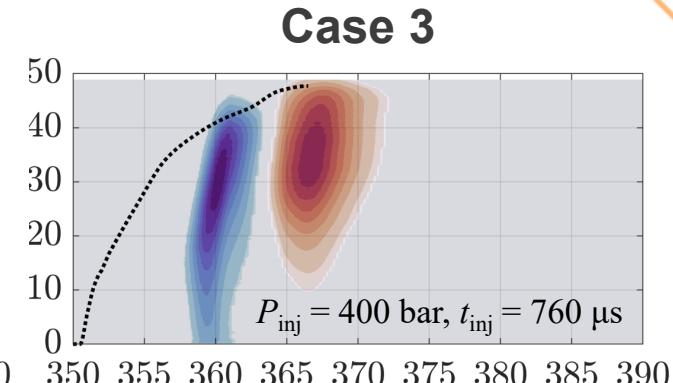
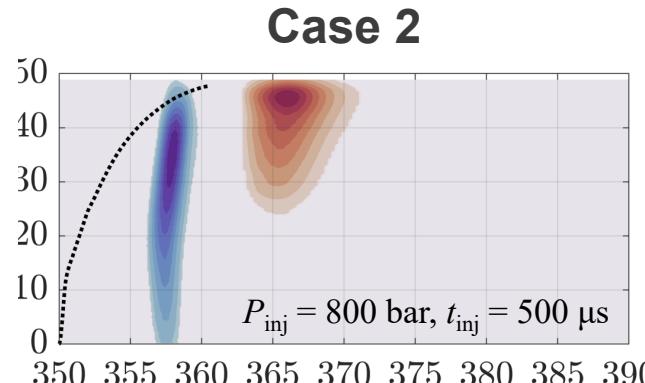
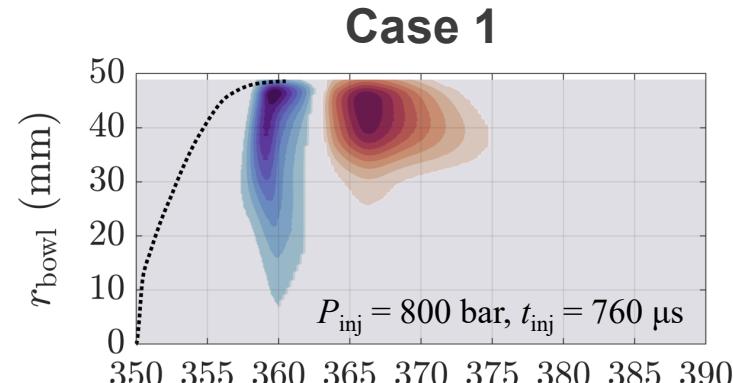
$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

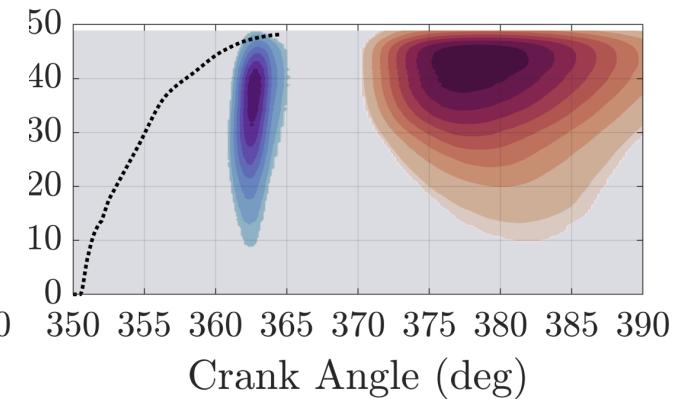
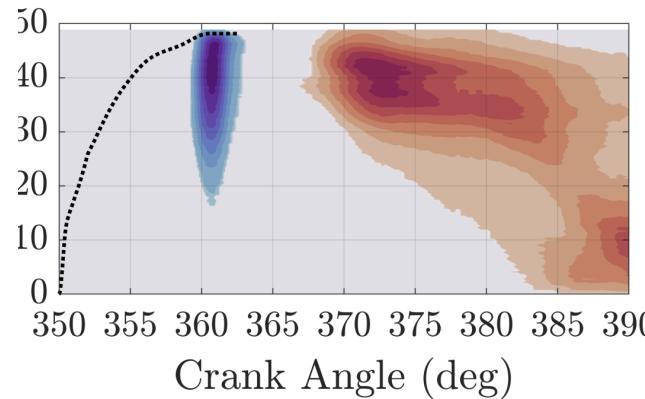
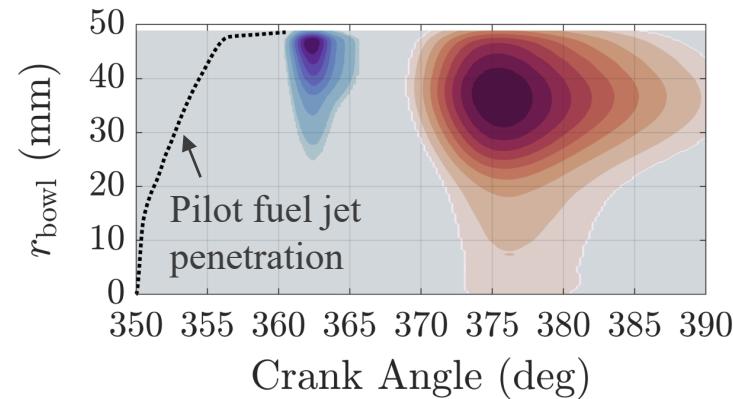
# Spatio-temporal flame evolution with pilot jet penetration



## Single pilot injection

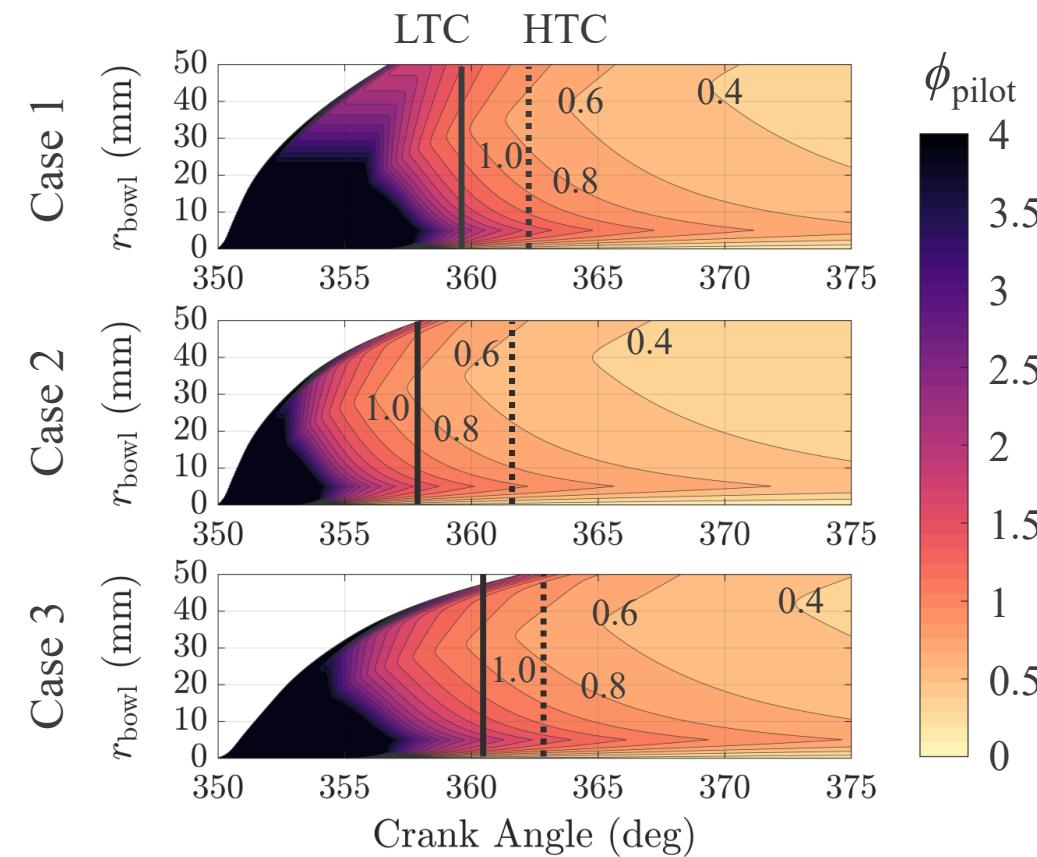
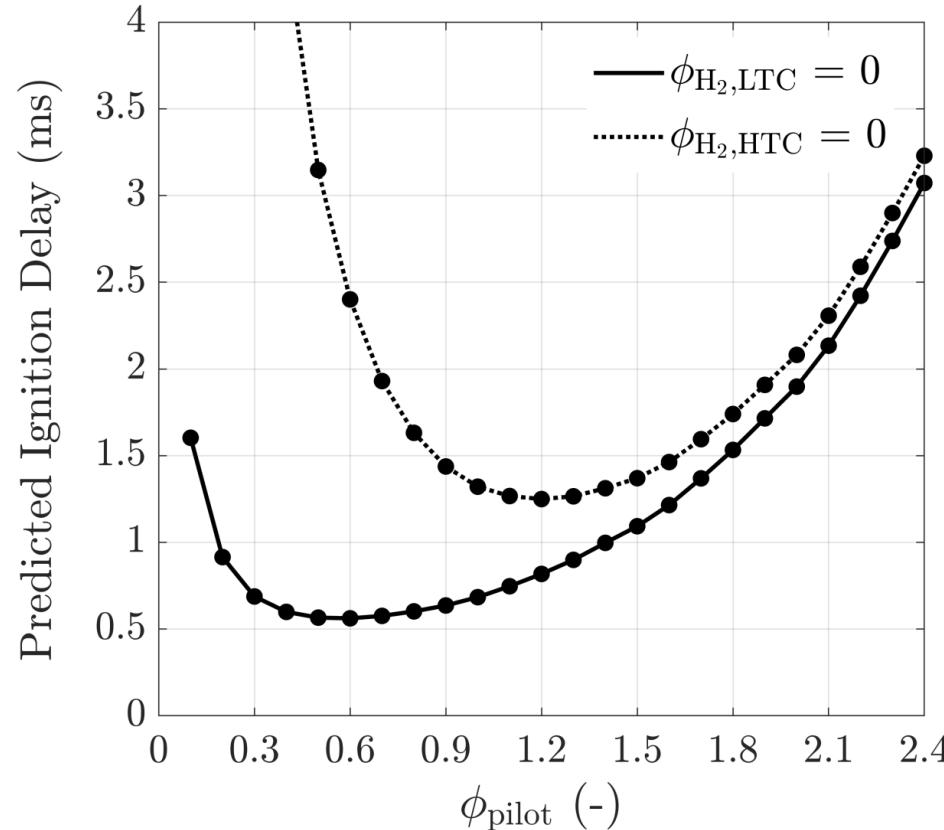


## H<sub>2</sub>-pilot dual fuel



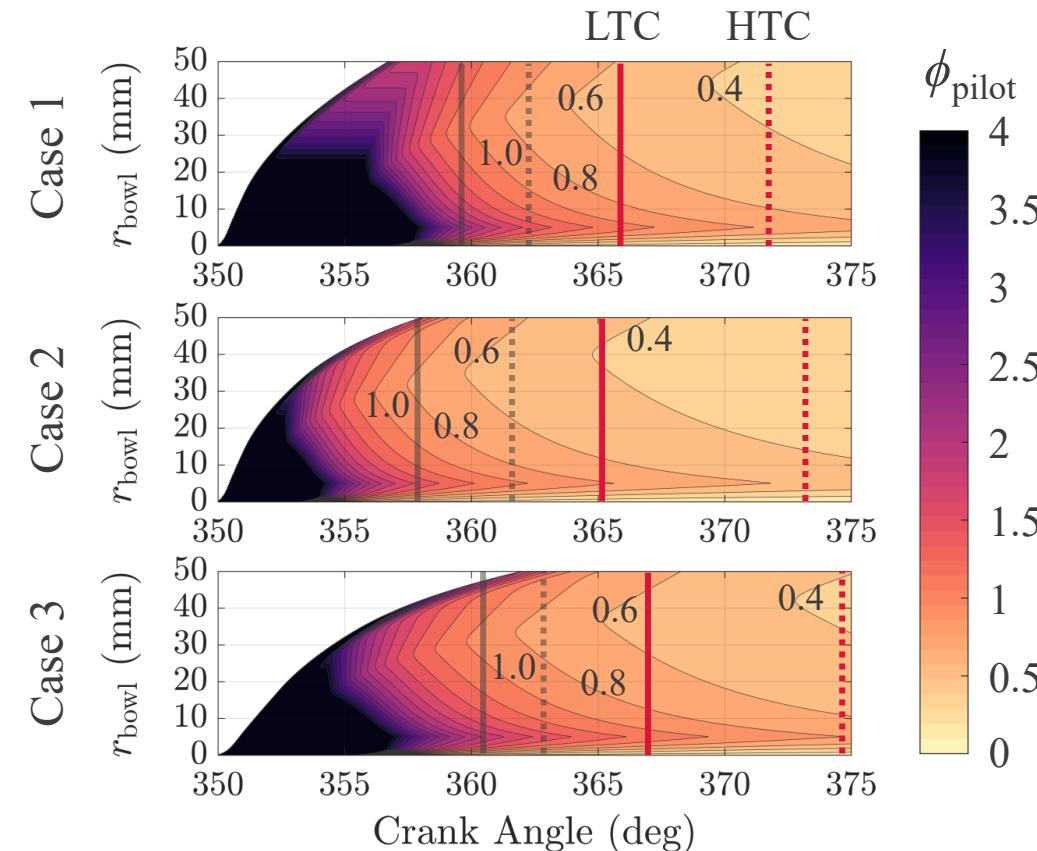
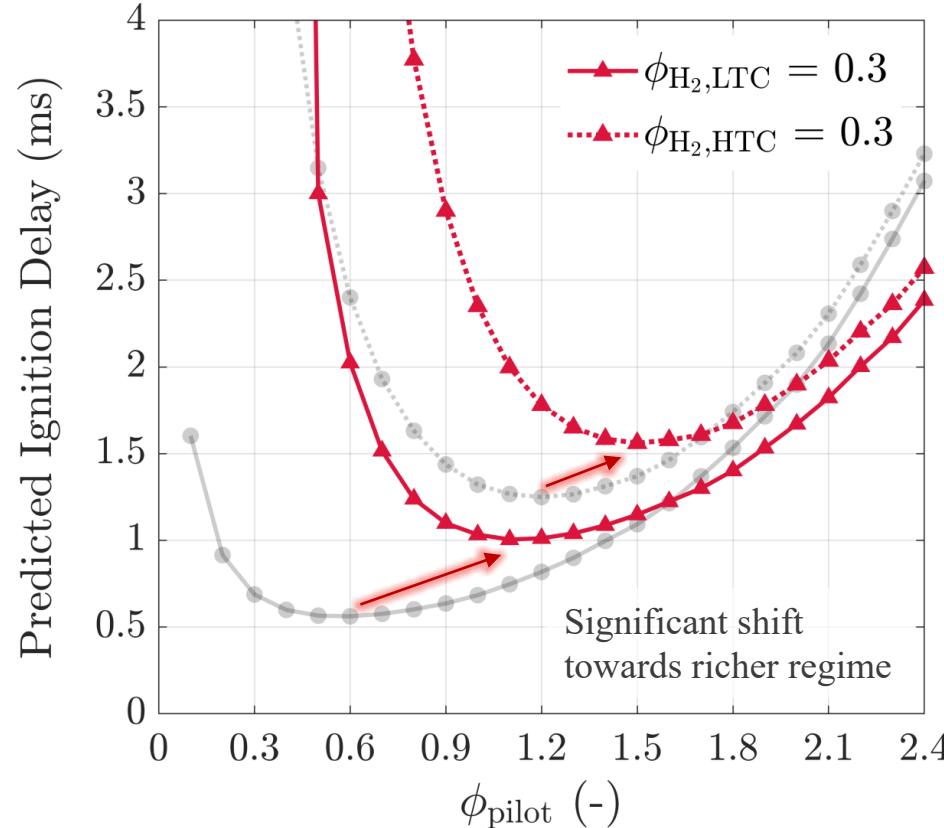
- The changes in jet penetration with respect to pilot injection parameter result in variations in mixing time and pilot fuel concentration, followed by spatial changes in LTHR (HCHO\*) and HTHR (OH\*).
- In H<sub>2</sub> environment with reduced ambient density, faster jet penetration is noticeable but the inhibitive effect of H<sub>2</sub> delays the overall reaction, increases dwell time, and confines the reaction to near-wall regions.
- The injection rate profile of Case 3 is more conducive to maintaining rich fuel pockets, highlighting the significance of fuel jet evolution on the ignition process in hydrogen dual-fuel combustion.

# Physico-chemical interplay between pilot injection and chemical kinetics



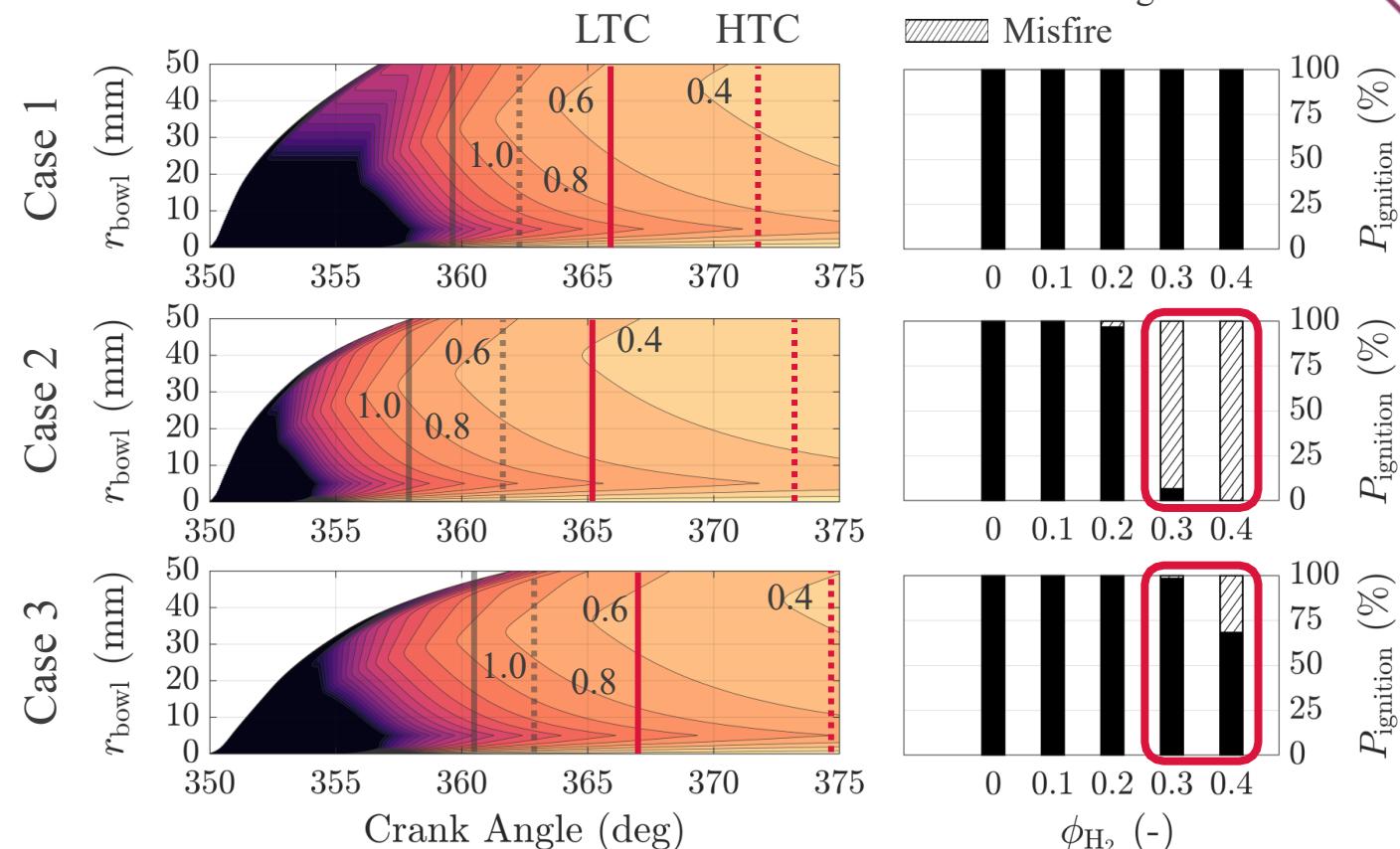
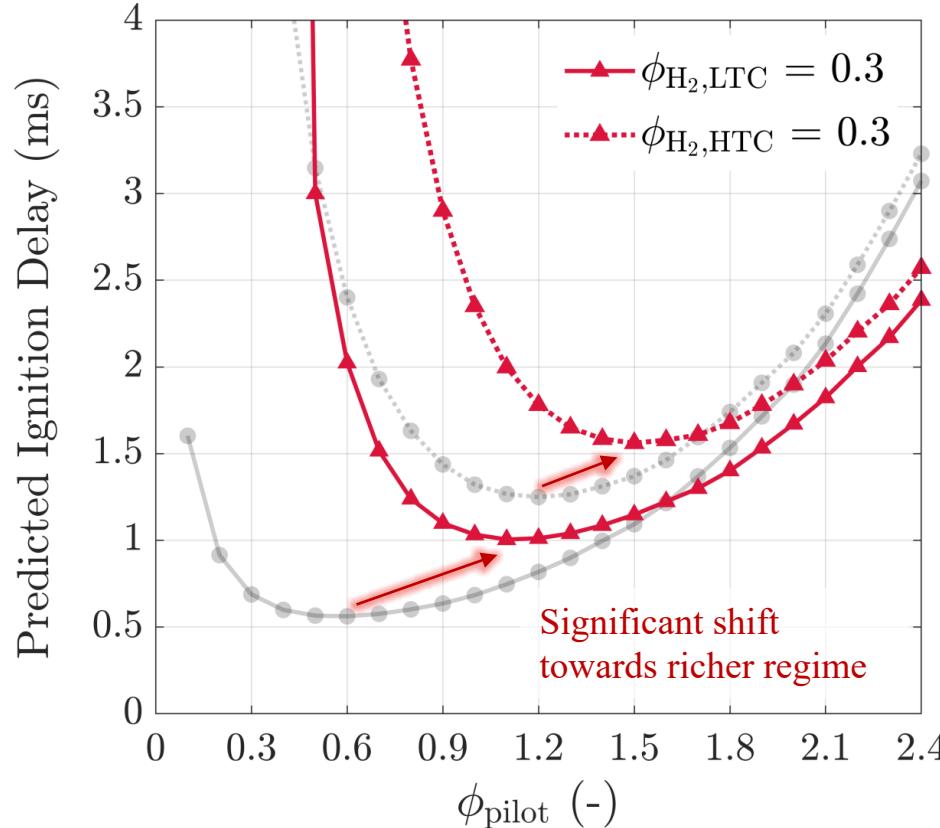
- The most-reactive mixture fraction (indicating shortest ignition delays) or highest pilot fuel concentration in the spray can possibly initiate the two-stage auto-ignition.
- With the pure pilot injection, a wide range of  $\phi_{\text{pilot}}$  is available in the spray domain, which is not expected to limit the ignition process; except for the increased dependence on stochastic rich fuel pockets in Case 2.

# Physico-chemical interplay between pilot injection and chemical kinetics



- The H<sub>2</sub> addition results in a significant shift of the most-reactive mixture fraction towards *richer regime* because of the low O<sub>2</sub> concentration ([O<sub>2</sub>]) and radical scavenging by the primary fuel (H<sub>2</sub>).
- Comparing all three cases, slower pilot fuel mixing induced by the lower injection pressure appears to enable more effective combustion with lower pilot fuel consumption, thus achieving a higher H<sub>2</sub> substitution rate.

# Physico-chemical interplay between pilot injection and chemical kinetics



- Frequent misfires under rich hydrogen environment (Case 2) can be explained by the considerably low pilot fuel concentration across the entire jet area at the onset of low- and high-temperature ignition.
- Comparing all three cases, slower pilot fuel mixing induced by the lower injection pressure appears to enable more effective combustion with lower pilot fuel consumption, thus achieving a higher H<sub>2</sub> substitution rate.

# Summary and Future Work

## Summary

- Increasing the hydrogen content substantially delays the first- and second-stage ignition, followed by many misfires especially when the pilot injection parameter is inappropriate to create rich fuel pockets.
- Dual-fuel chemical kinetics simulations reveal that to overcome the strong inhibitive effect of  $H_2$  under lean conditions, a pilot injection strategy should encourage significant local fuel concentration for reliable ignition.
- Supported by the 1D spray modeling and chemical kinetics simulation, higher  $H_2$  substitution rates can possibly be realized by using larger injector orifices with lower injection pressures, while maintaining engine performance.

## Future Work

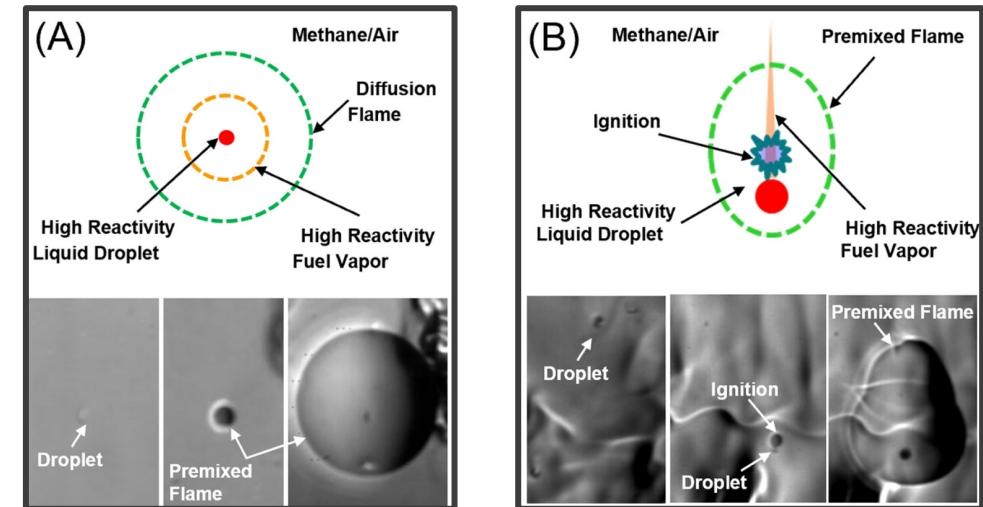
### ***Homogeneous hydrogen mixture formation***

- Measure the hydrogen mixing field with temperature field to quantify the tracer PLIF dataset (anisole or *p*-DFB).
- Combine  $H_2$  mixing field diagnostics with PIV measurement for the flow-field characterization.

### ***Oil-induced hydrogen pre-ignition***

- Oil droplet injector to explore the impact of  $H_2$  and the role of size and temperature on oil-induced pre-ignition.

### **Effect of diesel droplet size on flame evolution**



# Acknowledgements

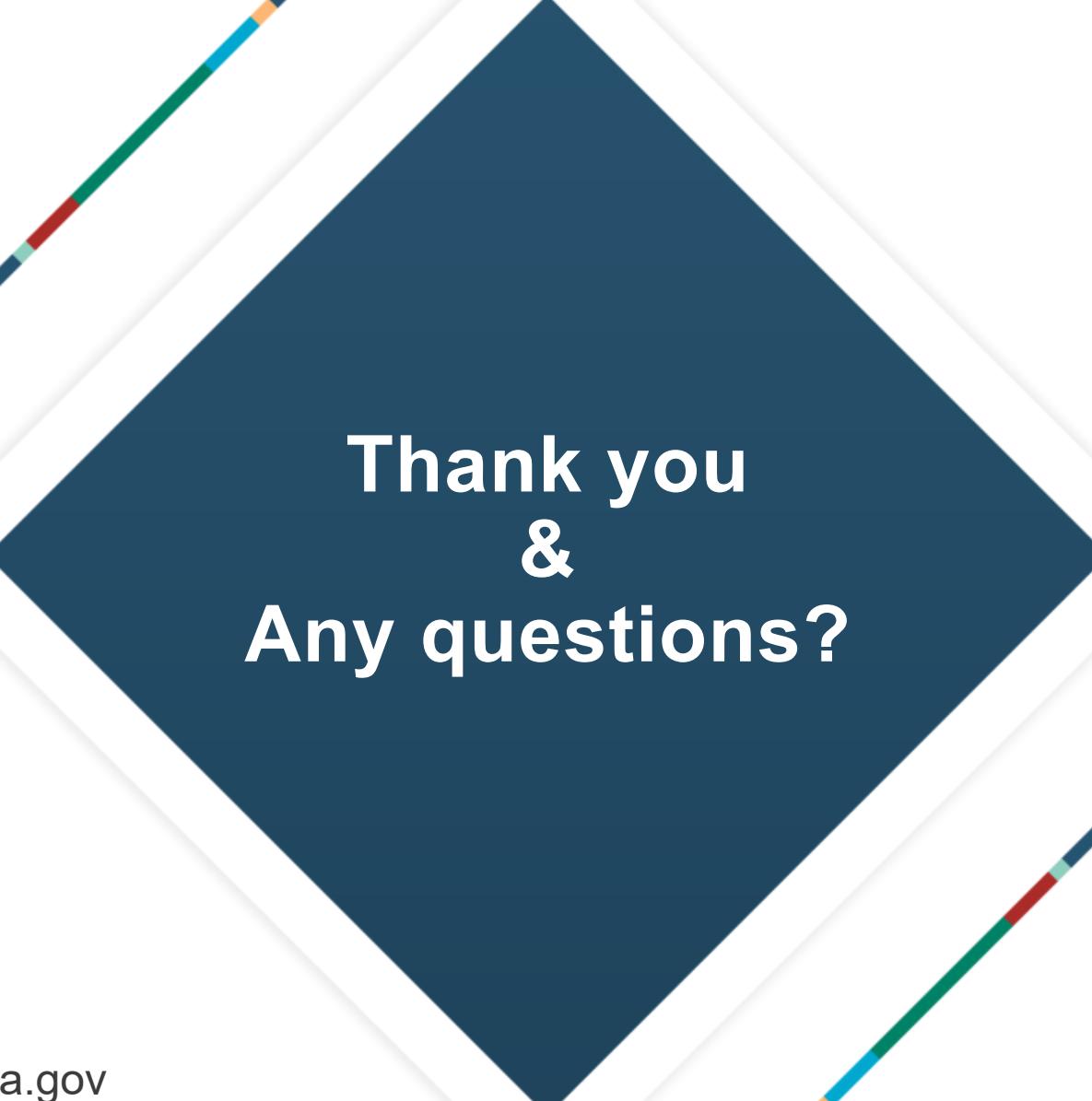
## *Combustion Research Facility (CRF), Sandia National Laboratories*

- Department Manager: Paul Miles
- Technologist: Kyra Schmidt

## *Department of Energy (DOE), Office of Vehicle Technologies (VTO)*

- Gurpreet Singh, DOE EERE – Funding support
- Kevin Stork, DOE EERE – Funding support

This research was sponsored by the U.S. Department of Energy (DOE) Office of Energy Efficiency and Renewable Energy (EERE). Optical engine experiments were conducted at the Combustion Research Facility (CRF), Sandia National Laboratories (SNL), Livermore, CA. Sandia National Laboratories is a multi-mission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC (NTESS), a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration (DOE/NNSA) under contract DE-NA0003525. This written work is authorized by an employee of NTESS. The employee, not NTESS, owns the right, title and interest in and to the written work and is responsible for its contents. Any subjective views or opinions that might be expressed in the written work do not necessarily represent the views of the U.S. Government. The publisher acknowledges that the U.S. Government retains a non-exclusive, paid-up, irrevocable, world-wide license to publish or reproduce the published form of this written work or allow others to do so, for U.S. Government purposes. The DOE will provide public access to results of federally sponsored research in accordance with the DOE Public Access Plan



Thank you  
&  
Any questions?

## Contact

Aleš Srna (P.I.): [asrna@sandia.gov](mailto:asrna@sandia.gov)

Taesong Lee: [taelee@sandia.gov](mailto:taelee@sandia.gov)

# Hydrogen Pre-ignition Events

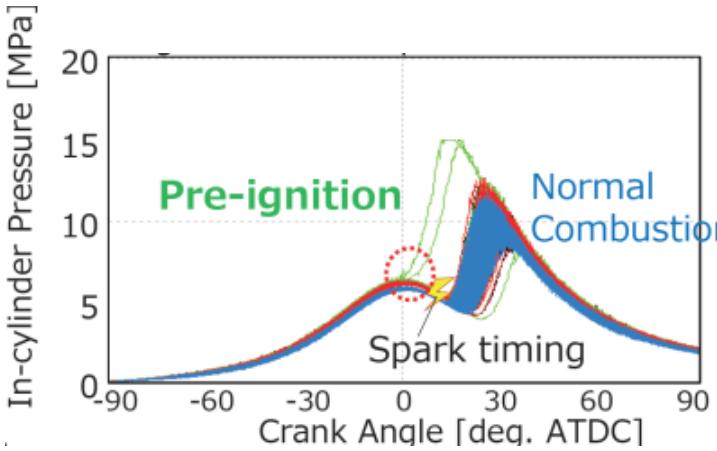


## Sporadic Pre-ignition

### Sources:

- Oil droplets
- Solid hot particles
- Carbonaceous deposits

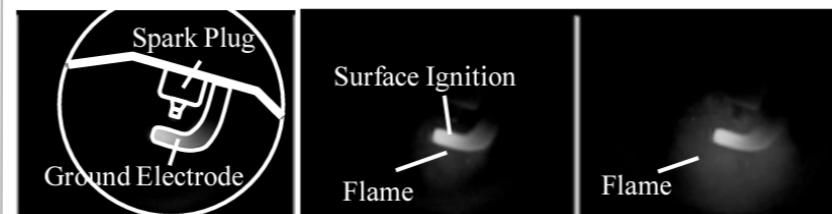
Little is known about the potential sources and related phenomenology; it seems to be caused by temperature increase during compression stroke.



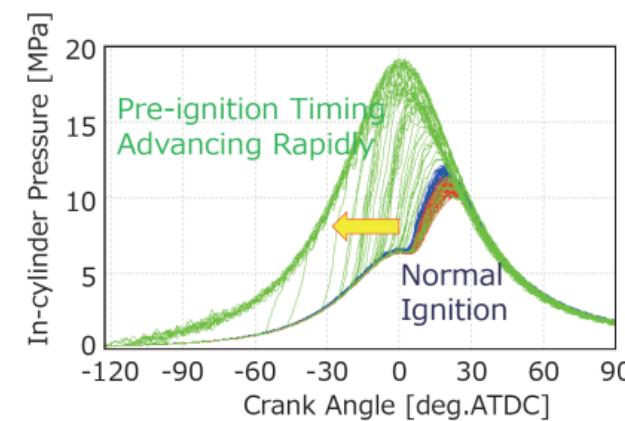
• Matsubara et. al, JSAE 20224660

## Runaway Pre-ignition

- Occurs particularly at high load
- Spark-plug electrode or exhaust valves are the most common source
- May appear like a “thermal runaway”, often requires fuel cut-off to stop.



• Matsubara et. al, JSAE 20234016

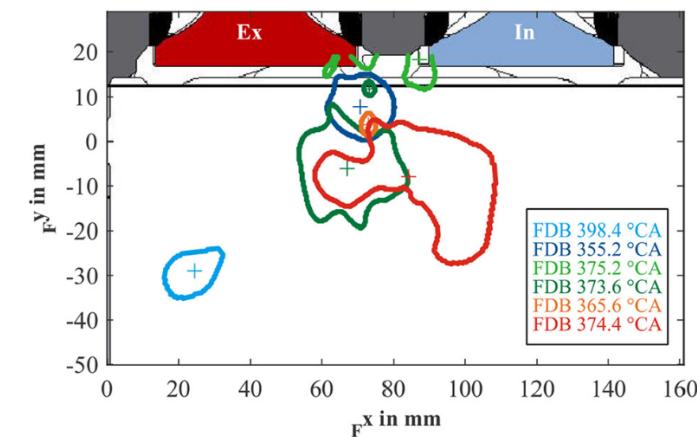


• Matsubara et. al, JSAE 20224660

## Back-fire

- Potential source of back-fire
- fresh mixture gets in contact with hot exhaust gas from previous cycle early during the intake stroke.

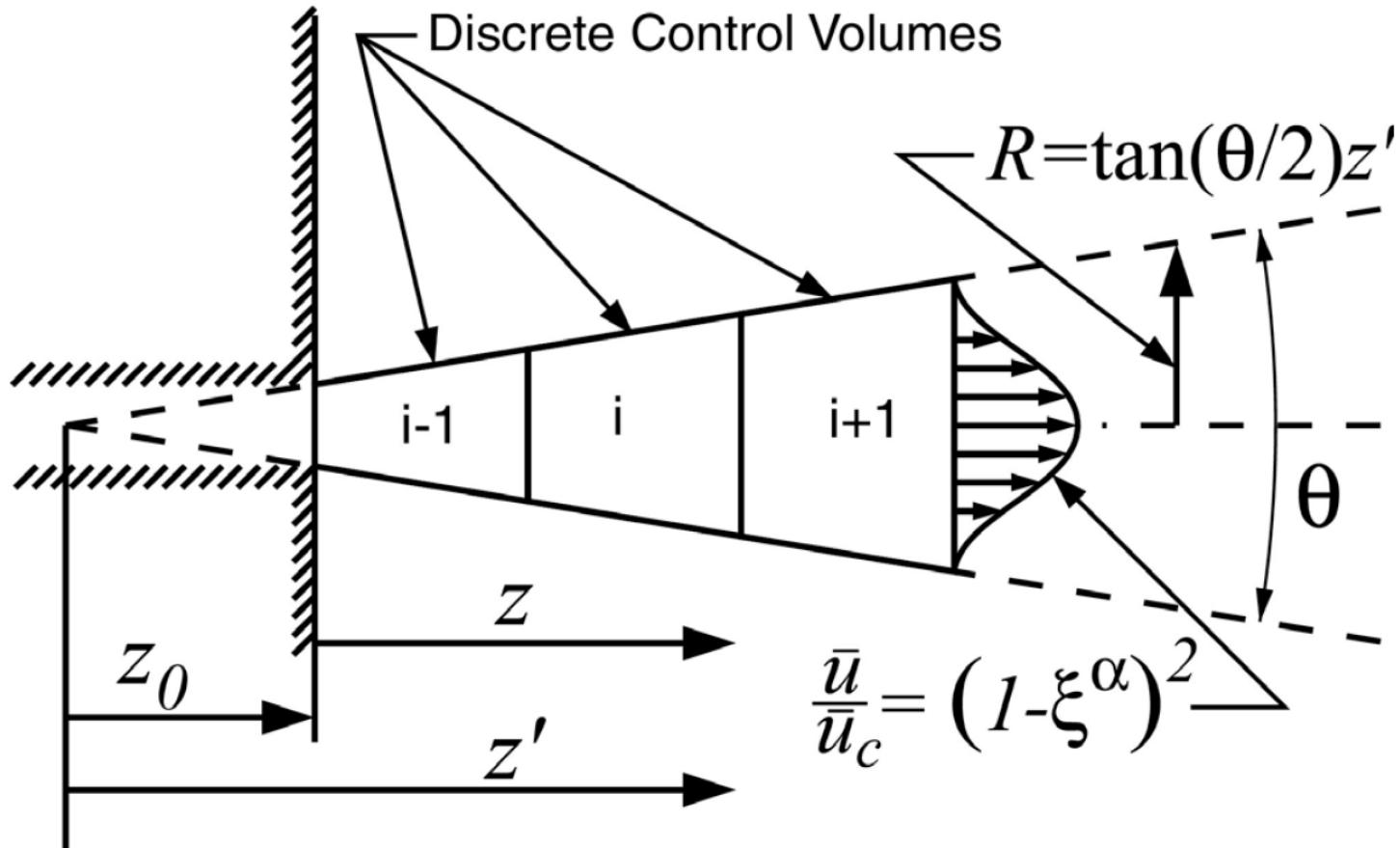
- A series of back-fire events
- back-fire heats up intake port mixture, subsequent cycles ignite almost immediately upon entering the cylinder.



• Eicheldinger et. al, IJER Vol 23, Issue 5

# Musculus-Kattke 1D Jet Model

1D simplified jet model based on the control volume analysis encompassing mixing and transient jet development



## Assumptions

- 1) The jet is not vaporizing.  
(No evaporative cooling effect)
- 2) Incompressible flow
- 3) Turbulent viscous forces are neglected.
- 4) Axial mixing of momentum due to molecular and turbulent diffusion is neglected.
- 5) The net force due to any axial pressure gradient is negligible.
- 6) The jet spreading angle is constant.
- 7) The radial profile of mean axial velocity remains unchanged during the EOI transient.

# Evolution of auto-ignition process: H<sub>2</sub> dual-fuel operation



Case 1

347.0 CAD

$P_{\text{inj}} = 800 \text{ bar}$ ,  $t_{\text{inj}} = 760 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

“LTC regime”

Case 2

347.0 CAD

$P_{\text{inj}} = 800 \text{ bar}$ ,  $t_{\text{inj}} = 500 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$

“HTC regime”

Case 3

347.0 CAD

$P_{\text{inj}} = 400 \text{ bar}$ ,  $t_{\text{inj}} = 760 \mu\text{s}$



$I_{\text{HCHO}^*}$

$I_{\text{OH}^*}$