



Solar Thermochemical Water Splitting (STCH): Tony McDaniel (lead)

Participating Labs: SNL, NREL, LLNL, INL

Project ID # P148D



HydroGEN STCH Seedling Projects & Lab Collaboration

Barriers

- Cost
- Efficiency
- Durability

STCH Node Labs



Sandia
National
Laboratories



Idaho National Laboratory



Lawrence Livermore
National Laboratory



NATIONAL RENEWABLE ENERGY LABORATORY

Support
through:



Personnel
Equipment
Expertise
Capability
Materials
Data

Interactive STCH Projects



Arizona State
University



University of Colorado
Boulder



Massachusetts
Institute of
Technology





HydroGEN STCH Seedling Projects with Lab Capability Support Technical Accomplishment Highlights

10 Lab capabilities support 5 new FOA-awarded projects:

- (P211, ASU, SNL, INL and NREL) **Design of Perovskite Materials for Solar**

Thermochemical Hydrogen Production: Initial multiscale modeling of STCH redox reactors with comprehensive thermal-chemical models predicting component design and performance

- (P208, CU, NREL) **Non-Intermittent, Solar-Thermal Processing to Split Water**

Continuously via a Near-Isothermal, Pressure-Swing Redox Cycle: Preliminary multiscale modeling evaluating STCH materials for commercial scale-up with TEA assessment

- (P210, CU, SNL and NREL) **Accelerated Discovery and Development of Perovskites for**

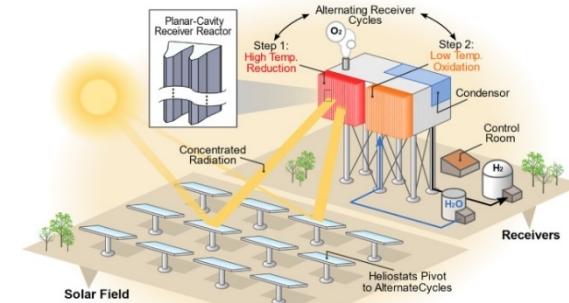
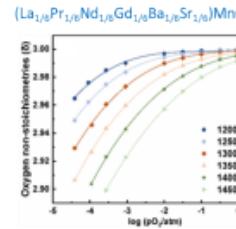
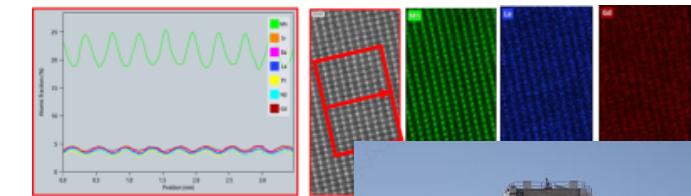
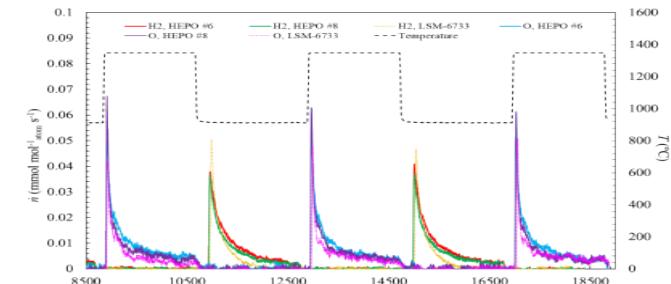
Solar Thermochemical Hydrogen Production: Identified path forward to on-sun testing using prior reactor development.

- (P217, St. Gobain, SNL, LLNL, and NREL) **Scalable Solar Fuels Production in A Reactor**

Train System by Thermochemical Redox Cycling of Novel Nonstoichiometric Perovskites: Computed energy barriers for water splitting process on STM and performed preliminary analysis of on-sun reactor testing

- (P212, WASHU, SNL and NREL) **Ca-Ce-Ti-Mn-O-Based Perovskites for Two-Step Solar**

Thermochemical Hydrogen Production Cycles: Performed initial evaluation of balance of plant needs for on-sun testing.





Goals: Comprehensively validate known STCH material properties and demonstrate theory-guided design of materials approach that optimizes the capacity/yield tradeoff.

- Develop computational toolset to define and establish material performance targets.
- Rigorously assess selected material formulations.
- Develop a materials search strategy for optimizing the capacity/yield tradeoff using DFT + Machine Learning (ML).
- Discover new materials using the ML model and characterize by detailed calculations, synthesis, and experimental validation.

STCH R&D:Q4 Annual Milestone	STCH Material Down-Select. Criteria: Use the technology assessment methodology derived from FY23 work to critically assess new ML-based solid solution STCH materials. The metric accounts for material-specific cycle dynamics and plant operational modality. (~10 new solid solution materials).
---------------------------------------	--

DFT = density functional theory
 T_{RED} = reduction temperature



Exemplar Material Commercial Viability Study

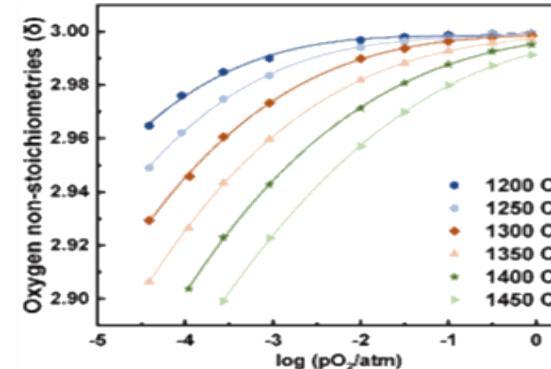
STCH Lab R&D Accomplishments

- Exemplar materials, methods and metrics determined by community consensus (via Benchmarking).
- Software platform developed for “end to end” processing of experimental data.
 - Will be made available to public

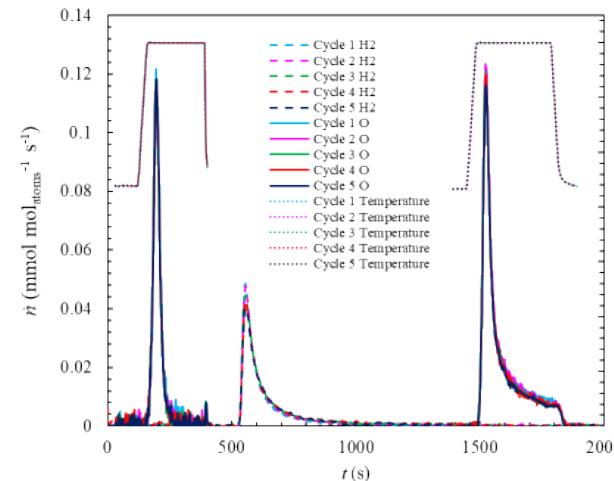
Key accomplishment: Measured and analyzed thermodynamics and H₂ production of five exemplar materials

Next step: Complete exemplar evaluation of cycle efficiency with estimator tool and publicize results

Thermodynamic parameters (from thermo-gravimetric analysis)

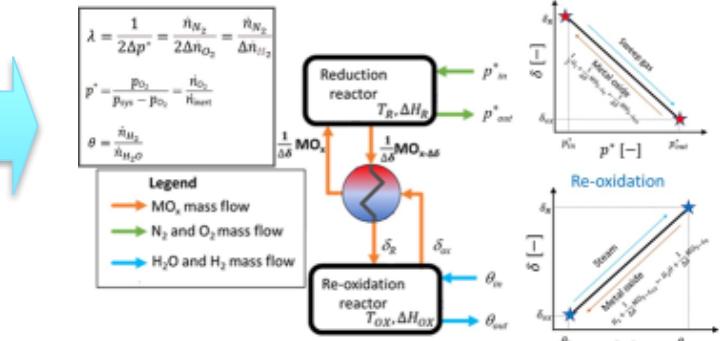
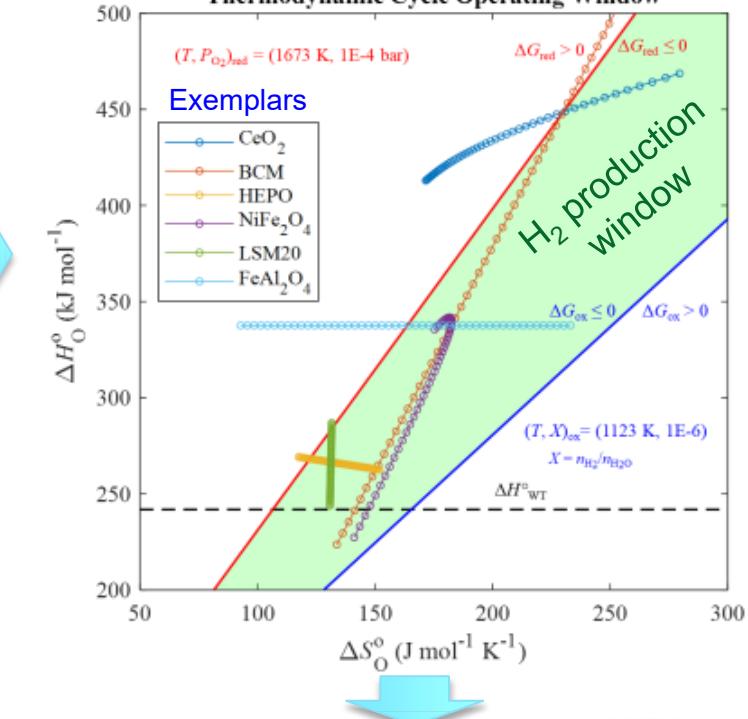


Hydrogen production and kinetic parameter (from flow reactor)



BCM: BaCe_{0.25}Mn_{0.75}O₃
 HEPO: La_{1/6}Pr_{1/6}Nd_{1/6}Gd_{1/6}Ba_{1/6}Sr_{1/6})MnO₃
 LSM20: La_{0.8}Sr_{0.2}MnO₃

Thermodynamic Cycle Operating Window



Cycle efficiency estimation



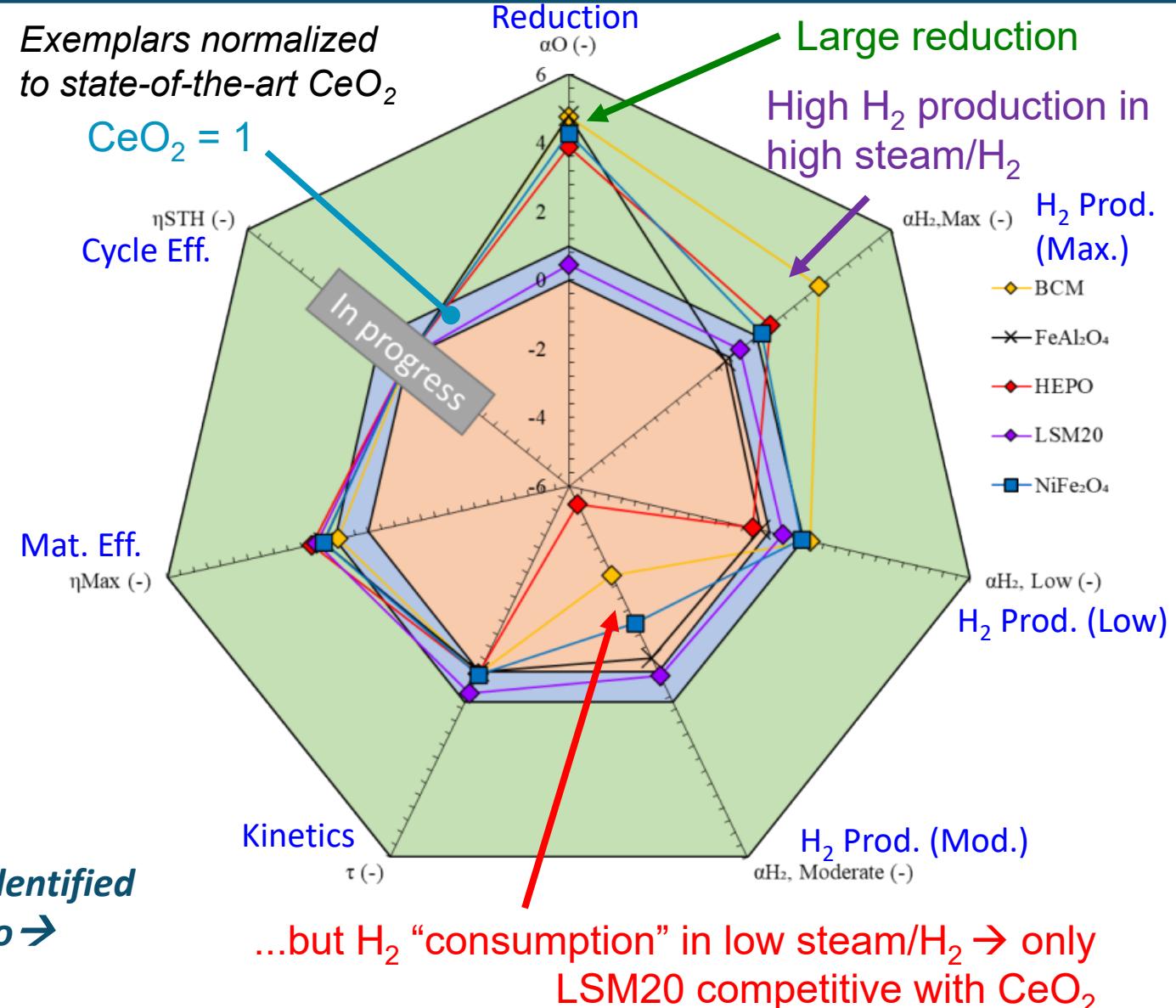
Exemplar Material Commercial Viability Study

STCH Lab R&D Accomplishments

Metrics	Descriptor	Target Values
Cycle Efficiency (STH)	Solar-to-hydrogen conversion efficiency derived from detailed cycle analysis using a thermodynamic model based on specific plant operational assumptions	$\eta_{STH} > 26\%$
Material Efficiency	$\frac{\Delta G_{WS}^o}{\Delta H_0^o}$ is the maximum possible thermal efficiency of the two-step process. (ΔG_{WS}^o evaluated at 25 °C)	$\eta_{Max} > 50\%$
Reduction Capacity	mmol O / mol atom in solid reduced @ neutral low condition	$\alpha_O > 5$
STCH Capacity (Maximum Yield)	mmol H ₂ / mol atom in solid reduced @ neutral low condition, oxidized in pure H ₂ O @ optimal T _{OX} for material	$\alpha_{H2,Max} > 5$
STCH Capacity (Low Yield)	mmol H ₂ / mol atom in solid reduced @ neutral low condition, oxidized in steam-to-fuel ratio H ₂ O/H ₂ = 1000 @ optimal T _{OX} for material	$\alpha_{H2,Low} > 2.5$
STCH Capacity (Moderate Yield)	mmol H ₂ / mol atom in solid reduced @ neutral low condition, oxidized in steam-to-fuel ratio H ₂ O/H ₂ = 100 @ optimal T _{OX} for material	$\alpha_{H2,Mod} > 1$
Kinetic Performance	Time to 90% of $\alpha_{H2,Max}$ in pure H ₂ O at optimal T _{OX} for specific material in a dispersed powder configuration	$\tau > 0.20$

Key accomplishment:

- Evaluation framework created and metrics identified**
- Weakness of exemplars in low steam/H₂ ratio → critical need for new materials**



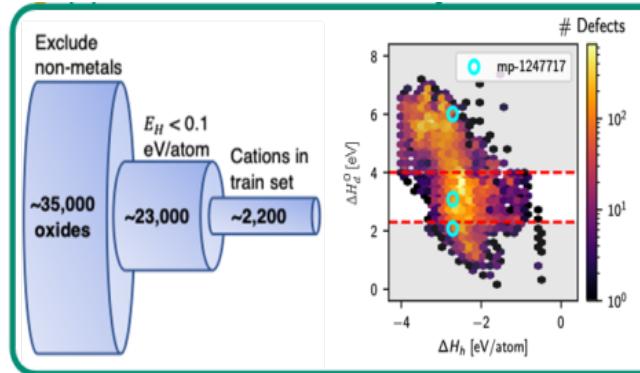


High Throughput Screening of Materials Project: Version 2

STCH Lab R&D Accomplishments

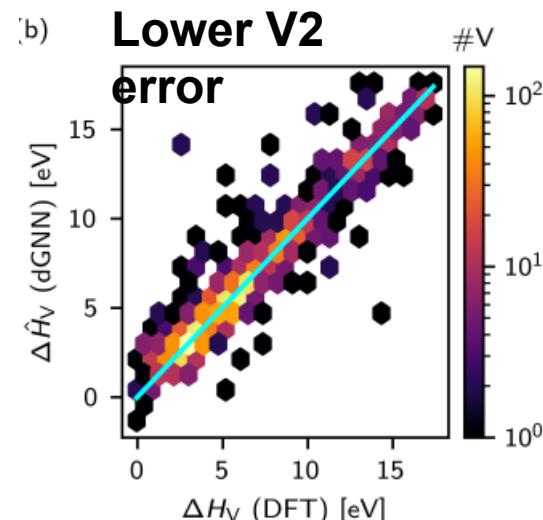
ML screens 10,000's of MP structures in minutes that would take 1,000's of DFT months

Metric	Requirement
Frac. of defects w/ $\Delta H_d^0 > 2.3$ eV	$x_{\min} = 1$
Frac. of defects w/ $\Delta H_d^0 \in [2.3, 4.0]$ eV	$x_{\text{rng}} > 0$
STCH operating range conditions (P_{O_2})	$\Delta \mu_{O_2}'$
Compound stability range	$\Delta \mu_{O_2}^{\phi_H < \{0, 0.1, \dots\}}$
Stable in the target range	$\Delta \mu_{O_2}^{\phi_H < X} \cap \Delta \mu_{O_2}'$



Key accomplishments:

- Version 2 (V2): More compounds for training and more elements included
- Improved ΔH_V MAE for unseen compounds from ~ 0.5 eV to ~ 0.4 eV.
- New approach \rightarrow target “unexpected” STCH compounds for experimental validation

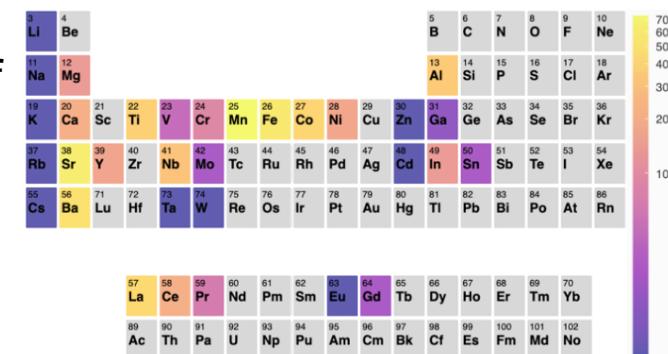


➤ **V2 training data**
= V1 + V1
validation + lit. +
new cations

Data Source	#Hosts	#V _O	#V _M
V1	199	795	686
{Ga,Cr,Pr,Sn}-containing (SrLa)(AlCoFeMn) ₃ alloys	23	75	57
SCM	12	12	0
BXM	2	86	0
Quat.+ Perovskites	4	18	9
ABO ₃	4	8	13
V2 Totals	29	43	0
	273	1037	765

➤ New cations in V2: V, Pr, Sn, Cr, Ga, Gd, Cs, Rb, Eu, Li, Na, K, Zn, Cd, Mo, W, and Ta

➤ Prevalence of training compounds containing a given cation



MAE = mean absolute error

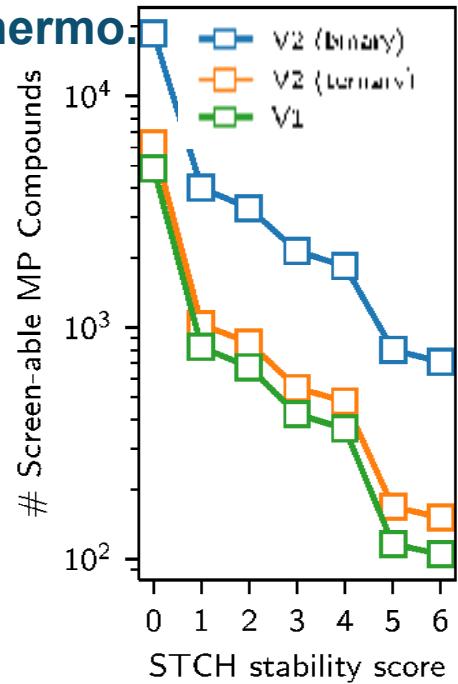
MP = Materials Project (<https://materialsproject.org/>)



High Throughput Screening of Materials Project: Version 2

STCH Lab R&D Accomplishments

V2: 10x increase in high stability oxides with desired thermo.



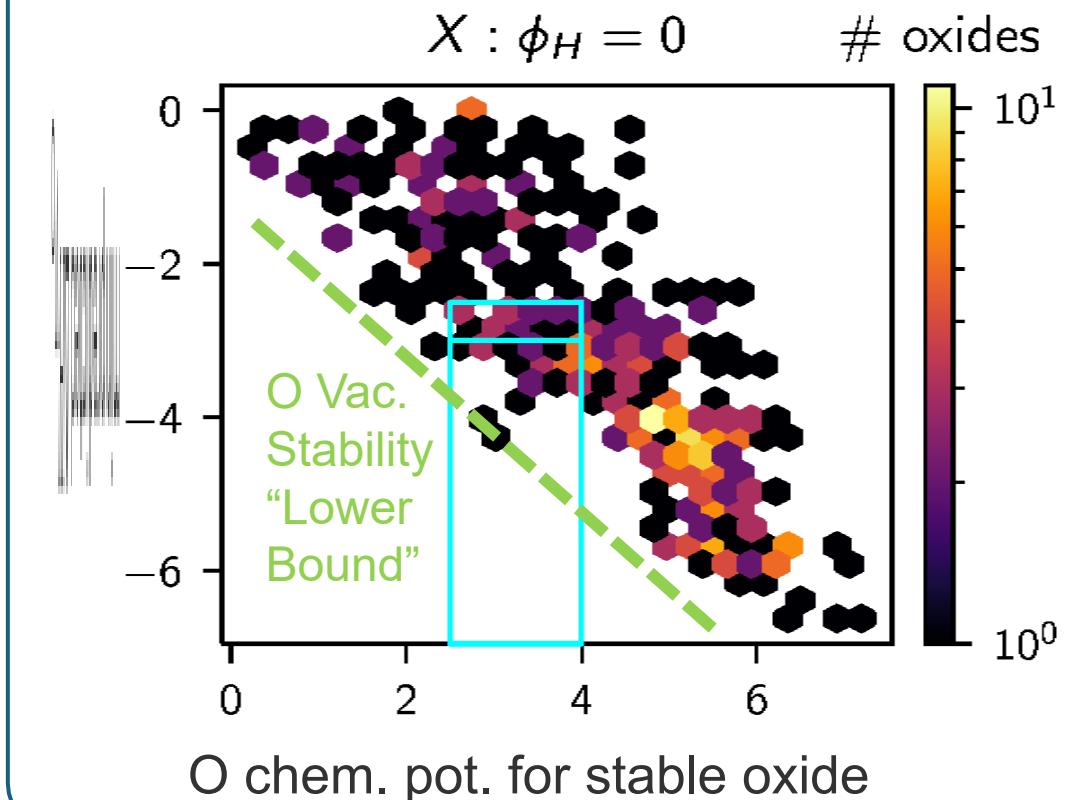
➤ ~30x reduction in # of oxides with STCH Stability Score = 6 vs. 0

➤ BUT... V2 screening will ~10x increase # of oxides from V1 screening with STCH Stability Score = 6

Computationally validate with high-throughput DFT

➤ Between ~50-400 new oxides targeted for DFT calc. of ΔH_{V_O}

Identify candidates for experiment in STCH “Goldilocks zone”



10x more predicted oxides → next step of model validation with computation and experiment

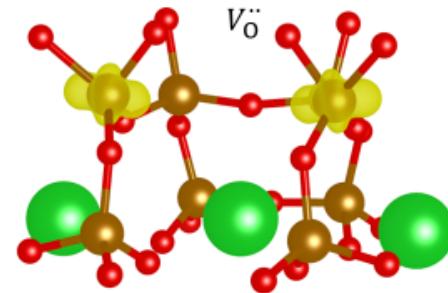


High Throughput Screening of MP Identified New STCH Materials

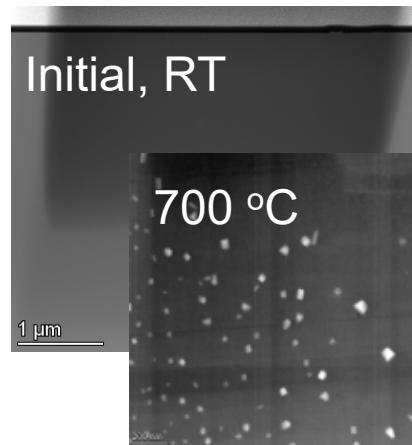
STCH Lab R&D Accomplishments

Predicted V1 STCH Compounds → Water-splitters!

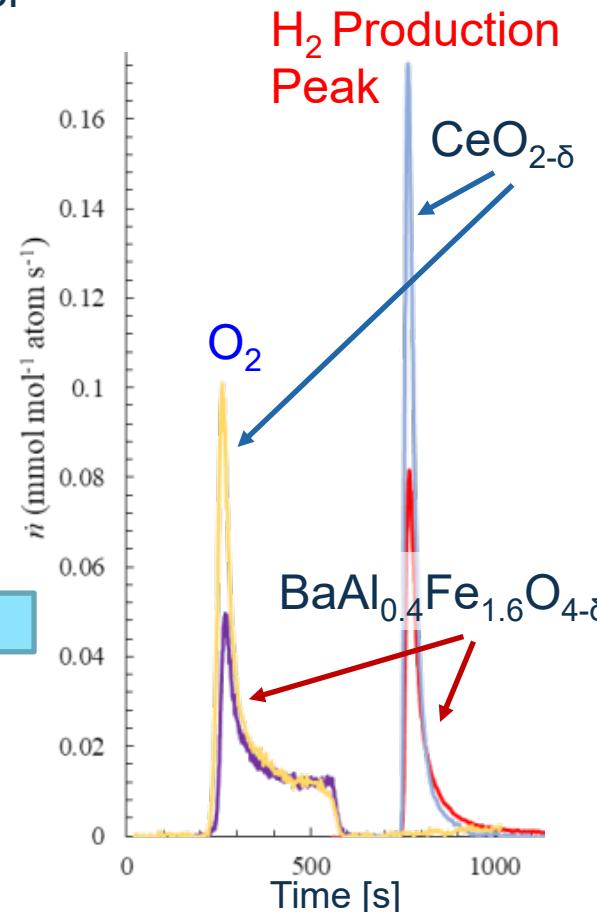
BaFe_2O_4 – predicted water splitter
(Al → increased hi-T stability)



DFT → oxygen vacancy preferred vs. cation defects

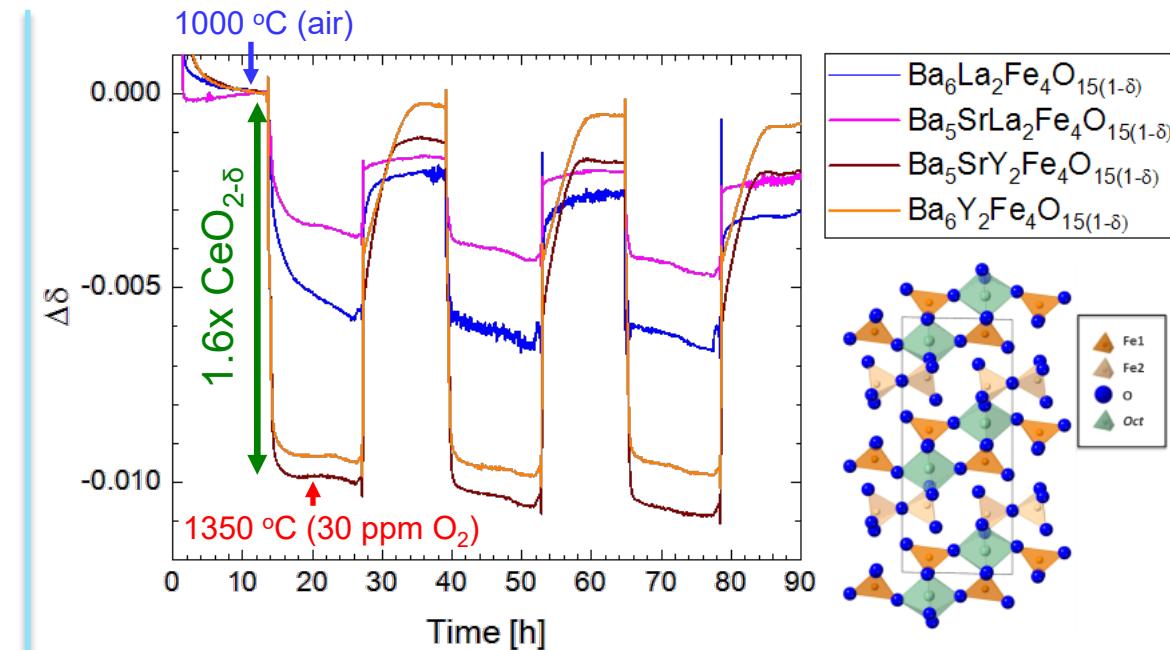


Fe-rich particles formed *in situ* TEM



Identified fabrication and stability challenges in other predicted V1 compounds

$(\text{Ba},\text{Sr})_6\text{Oct}_2\text{Fe}_4\text{O}_{15}$ family → $\Delta\delta > \text{CeO}_2$



STCH screening protocol¹ → $\text{Ba}_6\text{Y}_2\text{Fe}_4\text{O}_{15}$ is best (flow reactor testing next)

¹Sanders *et al.*, *Front. Energy Res.* 10:856943 (2022)

Milestone accomplishment: >10 new compositions identified from V2 for STCH validation

- Selected by “ease” of fabrication, melting point, ΔH_V , and stability
- Examining unexpected STCH compounds



STCH Summary and Proposed Future Work

Summary:

- Evaluated exemplar materials' potential to meet DOE STCH technology performance targets using a technology assessment methodology developed in this project. Exemplars have attractive H₂ production in dilute H₂/steam, but in concentrated conditions, only one is competitive with state of the art ceria. → Need for new materials
- Successfully demonstrated a water splitting material predicted from theory-guided design of materials using a Machine Learning algorithm developed in this project.

Proposed Future Work:

- STCH Lab R&D
 - Apply technology assessment methodology derived in this project to evaluate viability of the >10 predicted V2 STCH materials to meet DOE STCH technology performance targets.
 - Complete technology assessment framework by evaluating exemplars using cycle efficiency model and publicize
 - Continue validation and development of machine learning model for theory-guided design of materials
- Leverage HydroGEN nodes to enable successful completion of new seedling projects.

(future work subject to available funding)



Content incorporated in other parts of the main presentation



Responses to Reviewers

Degradation mechanisms studies should be planned. Most durability data was short, which limits the significance of project progress.

- Per the DOE requirement for seedlings, long-term (multi-week) testing of materials on-sun is planned for STCH.

In the future, the consortium can provide a year-to-year progress (since 2016) that would provide an overall outlook of the accomplishments.

- Metrics have now been established in STCH to which new materials will be benchmarked against



Responses to Previous Year Reviewers' Comments

- It is hard to really get a feel for how HydroGEN, as a project, functions at the overall coordination level, as much of the discussion (with the exception of the cross-cutting work) was about the seedlings. It would be good to understand **how TEA is informing the choice of performance metrics so that there is a clearer focus on the critical ones**. The impression is that there is a certain inflexibility, or at least inertia, in the work program; so it would be interesting to better understand **how agility could be brought to the portfolio and whether there should be an element of tactical, as opposed to strategic, work**. The **PFAS bans** will need to be addressed, for instance, and the consortium could look for ways to support industry and/or use operational data to inform the work program.
 - All five STCH seedling projects have requested TEA support from the NREL node

Acknowledgements



SNL Team

Anthony McDaniel, Lead Principal Investigators:

Andrea Ambrosini	Cy Fujimoto
Kenneth Armijo	Pinwen Guan
Sean Bishop	Keith King
Arielle Clauser	Mark Rodriguez
Eric Coker	Josh Sugar
Bert Debusschere	Matthew Witman
Tyra Douglas	



Sandia National Laboratories is a multimission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.