



## Solar Thermochemical Water Splitting (STCH): Tony McDaniel (lead)

**Participating Labs: SNL, NREL, LLNL, INL**

Project ID # P148D



# HydroGEN STCH Seedling Projects & Lab Collaboration

- Barriers
- Cost
  - Efficiency
  - Durability

## STCH Node Labs



Sandia  
National  
Laboratories



Lawrence Livermore  
National Laboratory



Support  
through:



Personnel  
Equipment  
Expertise  
Capability  
Materials  
Data

## Interactive STCH Projects



Arizona State  
University



University of Colorado  
Boulder



Massachusetts  
Institute of  
Technology



Washington  
University  
in St. Louis



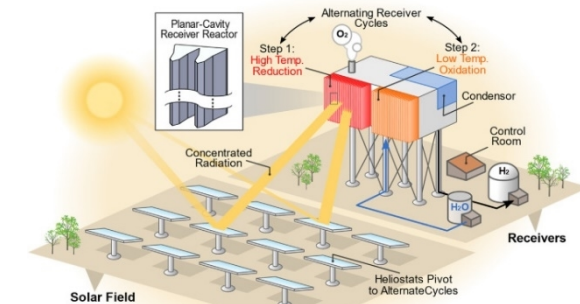
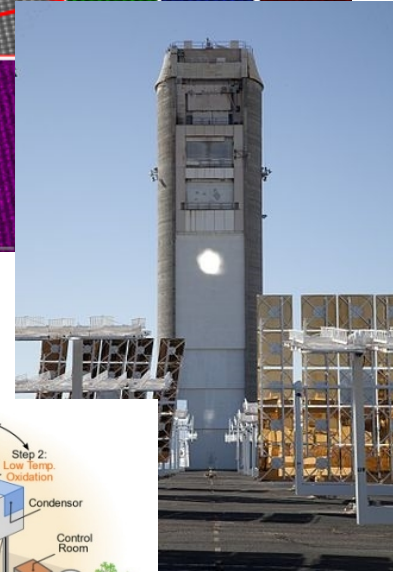
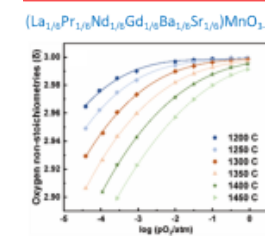
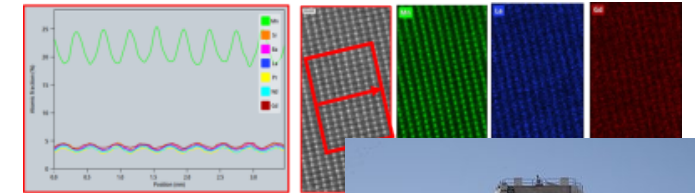
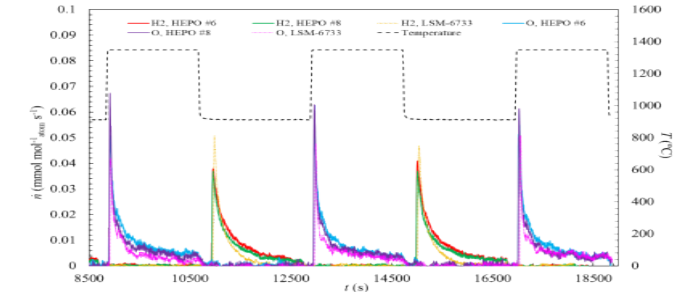


# HydroGEN STCH Seedling Projects with Lab Capability Support

## Technical Accomplishment Highlights

### 10 Lab capabilities support 5 new FOA-awarded projects:

- (P211, ASU, SNL, INL and NREL) **Design of Perovskite Materials for Solar Thermochemical Hydrogen Production:** Initial multiscale modeling of STCH redox reactors with comprehensive thermal-chemical models predicting component design and performance
- (P208, CU, NREL) **Non-Intermittent, Solar-Thermal Processing to Split Water Continuously via a Near-Isothermal, Pressure-Swing Redox Cycle:** Preliminary multiscale modeling evaluating STCH materials for commercial scale-up with TEA assessment
- (P210, CU, SNL and NREL) **Accelerated Discovery and Development of Perovskites for Solar Thermochemical Hydrogen Production:** Identified path forward to on-sun testing using prior reactor development.
- (P217, St. Gobain, SNL, LLNL, and NREL) **Scalable Solar Fuels Production in A Reactor Train System by Thermochemical Redox Cycling of Novel Nonstoichiometric Perovskites:** Computed energy barriers for water splitting process on STM and performed preliminary analysis of on-sun reactor testing
- (P212, WASHU, SNL and NREL) **Ca-Ce-Ti-Mn-O-Based Perovskites for Two-Step Solar Thermochemical Hydrogen Production Cycles:** Performed initial evaluation of balance of plant needs for on-sun testing.





Goals: Comprehensively validate known STCH material properties and demonstrate theory-guided design of materials approach that optimizes the capacity/yield tradeoff.

- Develop computational toolset to define and establish material performance targets.
- Rigorously assess selected material formulations.
- Develop a materials search strategy for optimizing the capacity/yield tradeoff using DFT + Machine Learning (ML).
- Discover new materials using the ML model and characterize by detailed calculations, synthesis, and experimental validation.

STCH  
R&D:Q4  
Annual  
Milestone

STCH Material Down-Select.  
Criteria: Use the technology assessment methodology derived from FY23 work to critically assess new ML-based solid solution STCH materials. The metric accounts for material-specific cycle dynamics and plant operational modality. (~10 new solid solution materials).

DFT = density functional theory  
 $T_{RED}$  = reduction temperature



# Exemplar Material Commercial Viability Study

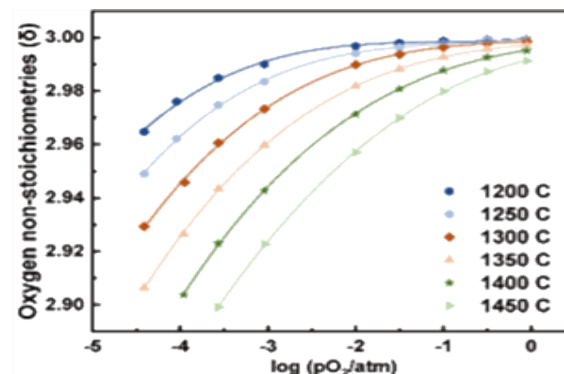
## STCH Lab R&D Accomplishments

- Exemplar materials, methods and metrics determined by community consensus (via Benchmarking).
- Software platform developed for “end to end” processing of experimental data.
  - Will be made available to public

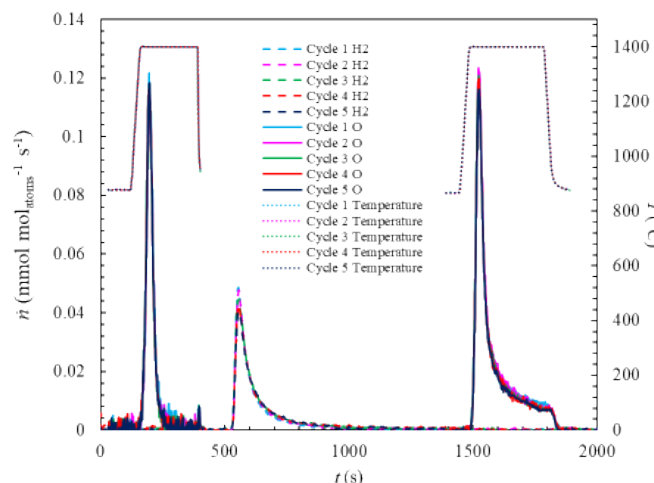
**Key accomplishment:** Measured and analyzed thermodynamics and H<sub>2</sub> production of five exemplar materials

**Next step:** Complete exemplar evaluation of cycle efficiency with estimator tool and publicize results

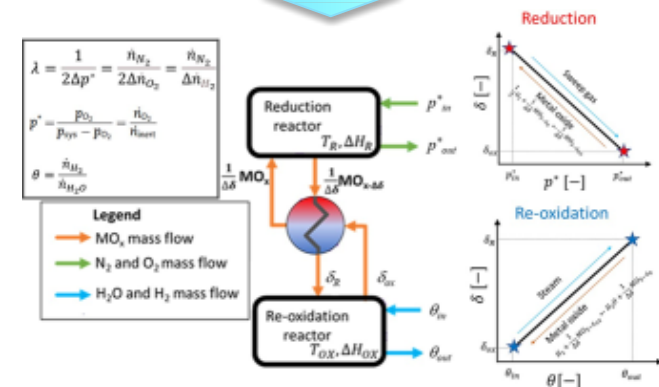
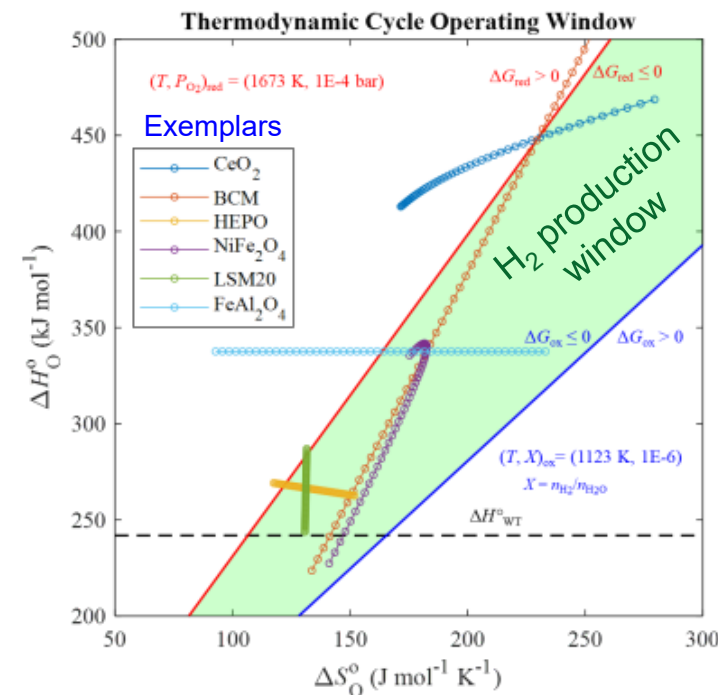
### Thermodynamic parameters (from thermo-gravimetric analysis)



### Hydrogen production and kinetic parameter (from flow reactor)



BCM: BaCe<sub>0.25</sub>Mn<sub>0.75</sub>O<sub>3</sub>  
HEPO: La<sub>1/6</sub>Pr<sub>1/6</sub>Nd<sub>1/6</sub>Gd<sub>1/6</sub>(Ba<sub>1/6</sub>Sr<sub>1/6</sub>)MnO<sub>3</sub>  
LSM20: La<sub>0.8</sub>Sr<sub>0.2</sub>MnO<sub>3</sub>



### Cycle efficiency estimation





# Exemplar Material Commercial Viability Study

## STCH Lab R&D Accomplishments

### Metrics

	Descriptor	Target Values
Cycle Efficiency (STH)	Solar-to-hydrogen conversion efficiency derived from detailed cycle analysis using a thermodynamic model based on specific plant operational assumptions	$\eta_{\text{STH}} > 26\%$
Material Efficiency	$\frac{\Delta G_{\text{WS}}^0}{\Delta H_0^0}$ is the maximum possible thermal efficiency of the two-step process. ( $\Delta G_{\text{WS}}^0$ evaluated at 25 °C)	$\eta_{\text{Max}} > 50\%$
Reduction Capacity	mmol O / mol atom in solid reduced @ neutral low condition	$\alpha_{\text{O}} > 5$
STCH Capacity (Maximum Yield)	mmol H <sub>2</sub> / mol atom in solid reduced @ neutral low condition, oxidized in pure H <sub>2</sub> O @ optimal T <sub>OX</sub> for material	$\alpha_{\text{H}_2, \text{Max}} > 5$
STCH Capacity (Low Yield)	mmol H <sub>2</sub> / mol atom in solid reduced @ neutral low condition, oxidized in steam-to-fuel ratio H <sub>2</sub> O/H <sub>2</sub> = 1000 @ optimal T <sub>OX</sub> for material	$\alpha_{\text{H}_2, \text{Low}} > 2.5$
STCH Capacity (Moderate Yield)	mmol H <sub>2</sub> / mol atom in solid reduced @ neutral low condition, oxidized in steam-to-fuel ratio H <sub>2</sub> O/H <sub>2</sub> = 100 @ optimal T <sub>OX</sub> for material	$\alpha_{\text{H}_2, \text{Mod}} > 1$
Kinetic Performance	Time to 90% of $\alpha_{\text{H}_2, \text{Max}}$ in pure H <sub>2</sub> O at optimal T <sub>OX</sub> for specific material in a dispersed powder configuration	$\tau > 0.20$

### Key accomplishment:

- **Evaluation framework created and metrics identified**
- **Weakness of exemplars in low steam/H<sub>2</sub> ratio → critical need for new materials**

Exemplars normalized to state-of-the-art CeO<sub>2</sub>

CeO<sub>2</sub> = 1

Cycle Eff.

Mat. Eff.

$\eta_{\text{Max}}$  (-)

Kinetics

$\tau$  (-)

Reduction

$\alpha_{\text{O}}$  (-)

Large reduction

High H<sub>2</sub> production in high steam/H<sub>2</sub>

H<sub>2</sub> Prod. (Max.)

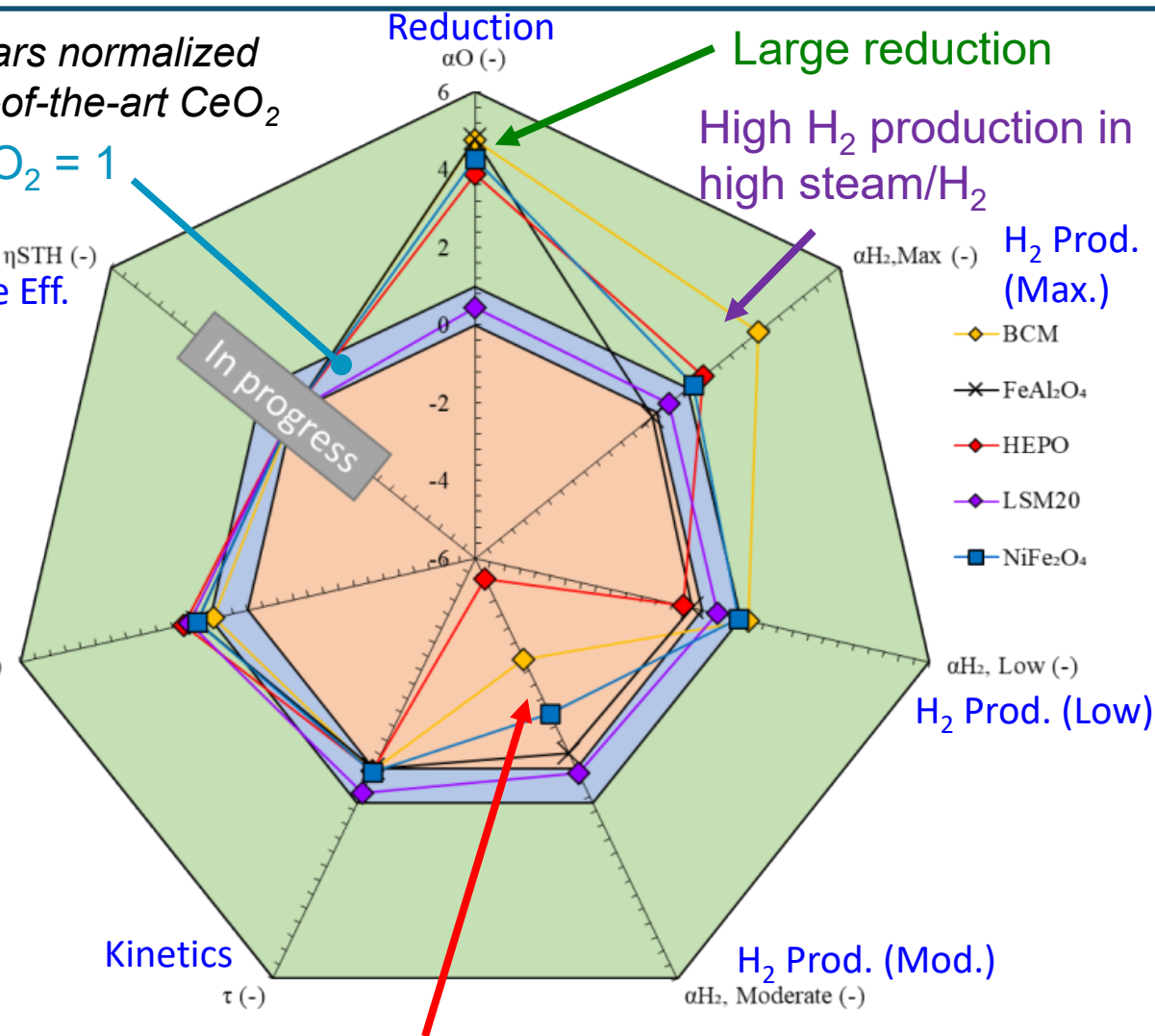
$\alpha_{\text{H}_2, \text{Max}}$  (-)

H<sub>2</sub> Prod. (Low)

$\alpha_{\text{H}_2, \text{Low}}$  (-)

H<sub>2</sub> Prod. (Mod.)

$\alpha_{\text{H}_2, \text{Moderate}}$  (-)



...but H<sub>2</sub> “consumption” in low steam/H<sub>2</sub> → only LSM20 competitive with CeO<sub>2</sub>

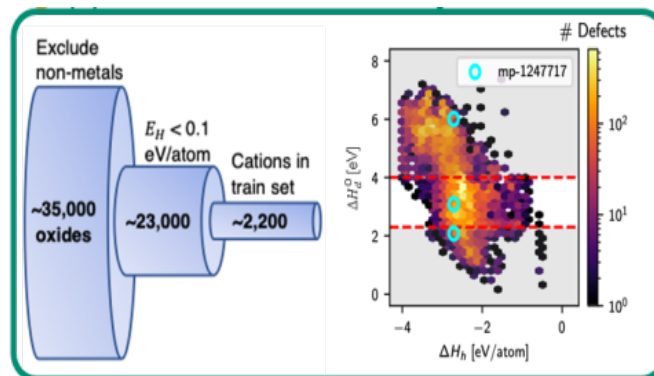


# High Throughput Screening of Materials Project: Version 2

## STCH Lab R&D Accomplishments

ML screens 10,000's of MP structures in minutes that would take 1,000's of DFT months

Metric	Requirement
Frac. of defects w/ $\Delta H_d^0 > 2.3$ eV	$x_{\min} = 1$
Frac. of defects w/ $\Delta H_d^0 \in [2.3, 4.0]$ eV	$x_{\text{rng}} > 0$
STCH operating range conditions ( $P_{O_2}$ )	$\Delta \mu'_{O_2}$
Compound stability range	$\Delta \mu_{O_2}^{\phi_H} < \{0, 0.1, \dots\}$
Stable in the target range	$\Delta \mu_{O_2}^{\phi_H} <^X \cap \Delta \mu'_{O_2}$

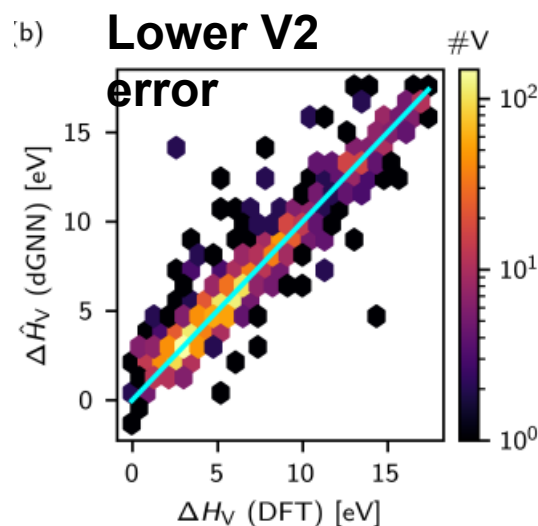


➤ **V2 training data**  
= V1 + V1  
validation + lit. +  
new cations

Data Source	#Hosts	#V <sub>O</sub>	#V <sub>M</sub>
V1	199	795	686
{Ga,Cr,Pr,Sn}-containing	23	75	57
(SrLa)(AlCoFeMn)O <sub>3</sub> alloys	12	12	0
SCM	2	86	0
BXM	4	18	9
Quat.+ Perovskites	4	8	13
ABO <sub>3</sub>	29	43	0
V2 Totals	273	1037	765

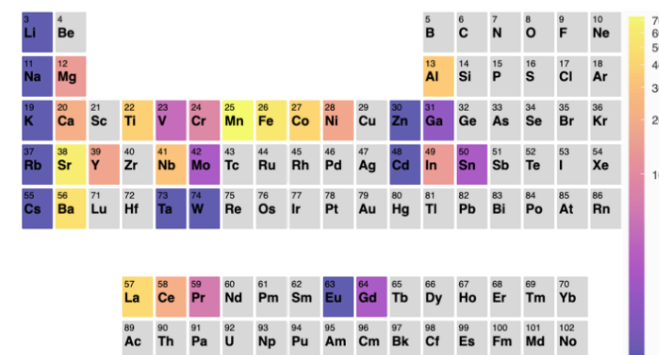
### Key accomplishments:

- Version 2 (V2): More compounds for training and more elements included
- Improved  $\Delta H_V$  MAE for unseen compounds from  $\sim 0.5$  eV to  $\sim 0.4$  eV.
- New approach  $\rightarrow$  target “unexpected” STCH compounds for experimental validation



➤ New cations in V2: V, Pr, Sn, Cr, Ga, Gd, Cs, Rb, Eu, Li, Na, K, Zn, Cd, Mo, W, and Ta

➤ Prevalence of training compounds containing a given cation

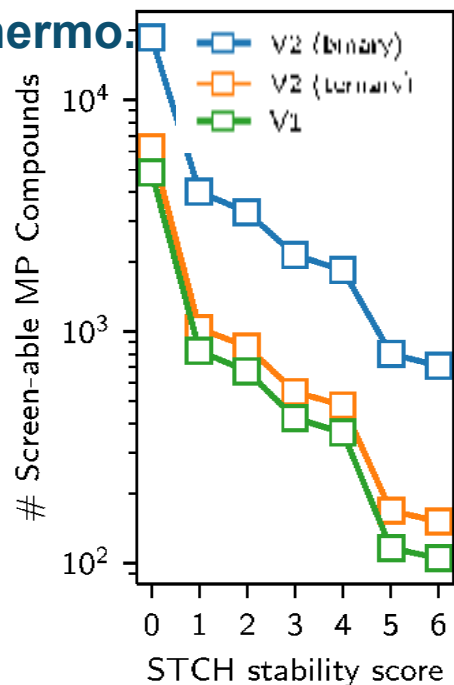




# High Throughput Screening of Materials Project: Version 2

## STCH Lab R&D Accomplishments

**V2: 10x increase in high stability oxides with desired thermo.**



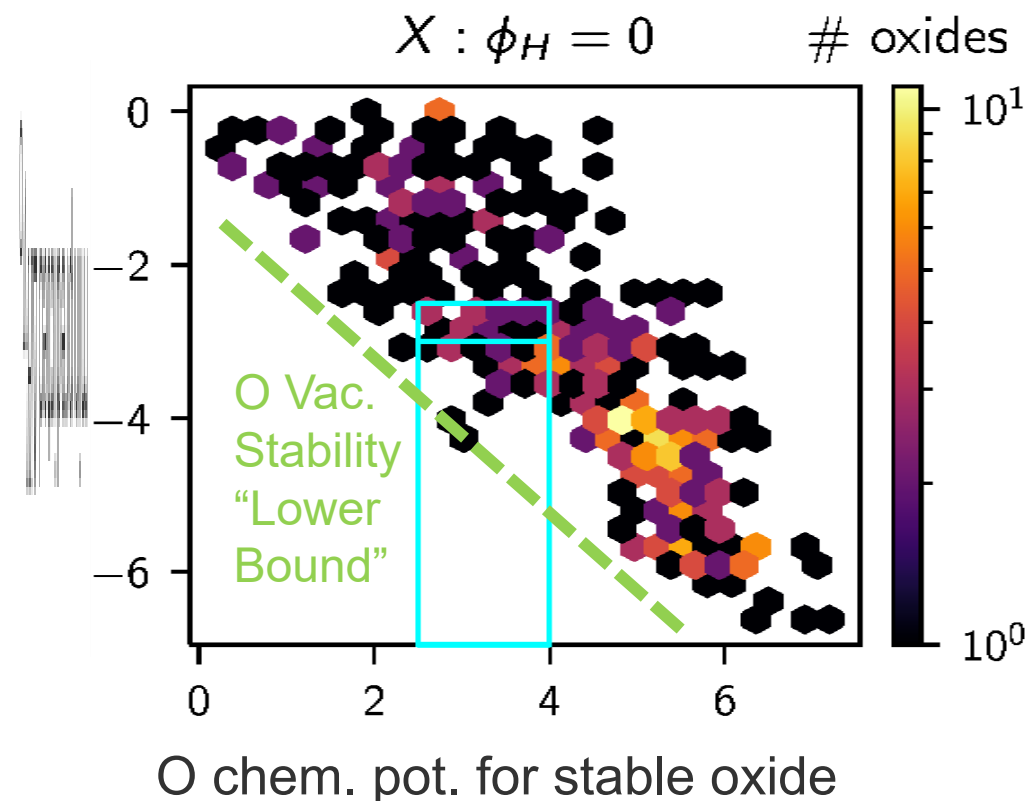
➤ **~30x reduction** in # of oxides with STCH  
Stability Score = 6 vs. 0

➤ BUT... V2 screening will  
**~10x increase** # of oxides from V1  
screening with STCH  
Stability Score = 6

**Computationally validate with high-throughput DFT**

➤ Between ~50-400 new oxides targeted for DFT calc. of  $\Delta H_{V_O}$

Identify candidates for experiment in  
STCH “Goldilocks zone”



**10x more predicted oxides → next step of model validation with computation and experiment**



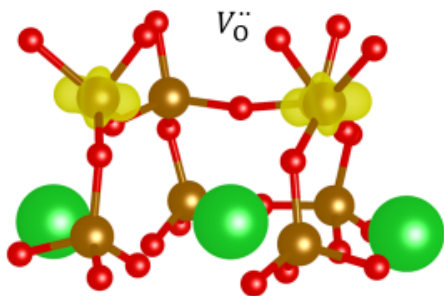


# High Throughput Screening of MP Identified New STCH Materials

## STCH Lab R&D Accomplishments

### Predicted V1 STCH Compounds → Water-splitters!

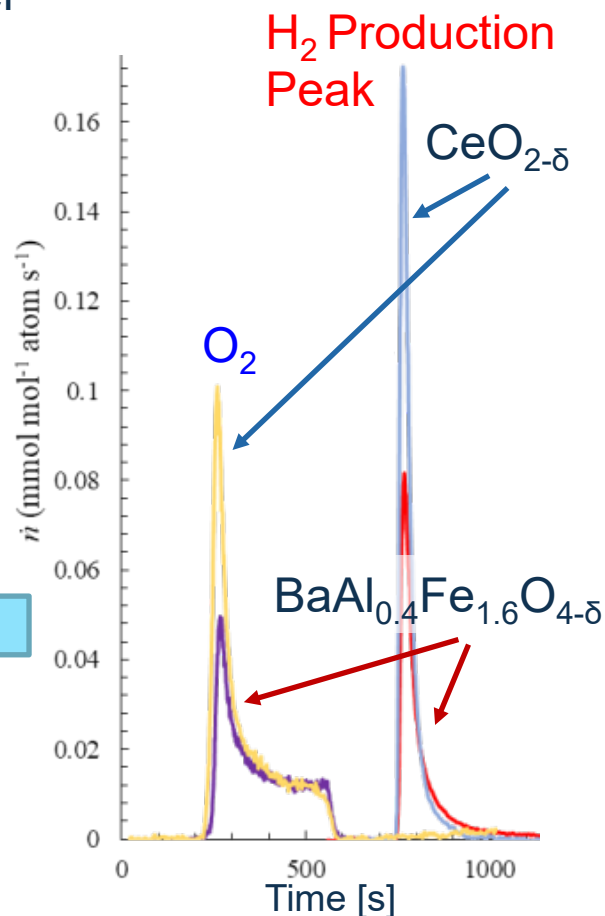
**BaFe<sub>2</sub>O<sub>4</sub>** – predicted water splitter  
(Al → increased *hi-T* stability)



DFT → oxygen vacancy preferred vs. cation defects

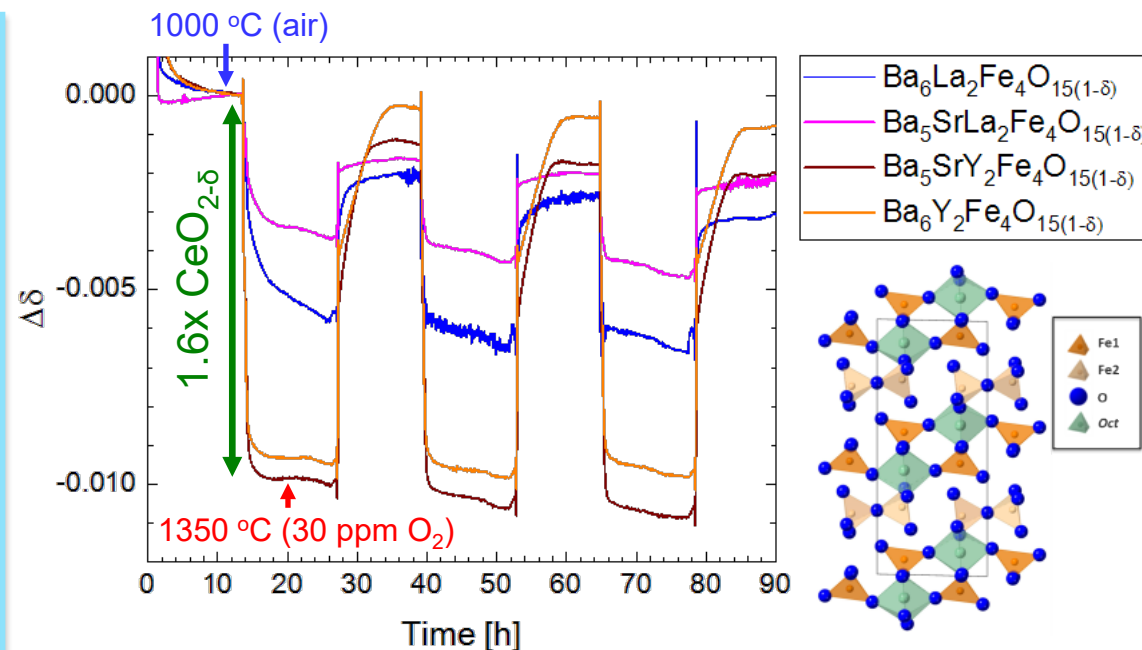


Fe-rich particles formed *in situ* TEM



*Identified fabrication and stability challenges in other predicted V1 compounds*

**(Ba,Sr)<sub>6</sub>Oct<sub>2</sub>Fe<sub>4</sub>O<sub>15</sub>** family →  $\Delta\delta > \text{CeO}_2$



STCH screening protocol<sup>1</sup> → Ba<sub>6</sub>Y<sub>2</sub>Fe<sub>4</sub>O<sub>15</sub> is best  
(flow reactor testing next)

<sup>1</sup>Sanders et al., *Front. Energy Res.* 10:856943 (2022)

**Milestone accomplishment:** >10 new compositions identified from V2 for STCH validation

- Selected by “ease” of fabrication, melting point,  $\Delta H_V$ , and stability
- Examining unexpected STCH compounds



# STCH Summary and Proposed Future Work

## Summary:

- Evaluated exemplar materials' potential to meet DOE STCH technology performance targets using a technology assessment methodology developed in this project. Exemplars have attractive  $H_2$  production in dilute  $H_2$ /steam, but in concentrated conditions, only one is competitive with state of the art ceria. → Need for new materials
- Successfully demonstrated a water splitting material predicted from theory-guided design of materials using a Machine Learning algorithm developed in this project.

## Proposed Future Work:

- STCH Lab R&D
  - Apply technology assessment methodology derived in this project to evaluate viability of the >10 predicted V2 STCH materials to meet DOE STCH technology performance targets.
  - Complete technology assessment framework by evaluating exemplars using cycle efficiency model and publicize
  - Continue validation and development of machine learning model for theory-guided design of materials
- Leverage HydroGEN nodes to enable successful completion of new seedling projects.

(future work subject to available funding)



# Content incorporated in other parts of the main presentation

---



# Responses to Reviewers

*Degradation mechanisms studies should be planned. Most durability data was short, which limits the significance of project progress.*

- Per the DOE requirement for seedlings, long-term (multi-week) testing of materials on-sun is planned for STCH.

*In the future, the consortium can provide a year-to-year progress (since 2016) that would provide an overall outlook of the accomplishments.*

- Metrics have now been established in STCH to which new materials will be benchmarked against



# Responses to Previous Year Reviewers' Comments

- It is hard to really get a feel for how HydroGEN, as a project, functions at the overall coordination level, as much of the discussion (with the exception of the cross-cutting work) was about the seedlings. It would be good to understand **how TEA is informing the choice of performance metrics so that there is a clearer focus on the critical ones**. The impression is that there is a certain inflexibility, or at least inertia, in the work program; so it would be interesting to better understand **how agility could be brought to the portfolio and whether there should be an element of tactical, as opposed to strategic, work**. The **PFAS bans** will need to be addressed, for instance, and the consortium could look for ways to support industry and/or use operational data to inform the work program.
  - All five STCH seedling projects have requested TEA support from the NREL node



# Acknowledgements



## SNL Team

**Anthony McDaniel, Lead  
Principal Investigators:**

Andrea Ambrosini	Cy Fujimoto
Kenneth Armijo	Pinwen Guan
Sean Bishop	Keith King
Arielle Clauser	Mark Rodriguez
Eric Coker	Josh Sugar
Bert Debusschere	Matthew Witman
Tyra Douglas	



*Sandia National Laboratories is a multimission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.*