

# MR13A-3183: Microbial and Geochemical Characterization of Groundwater: Implications for Underground Hydrogen Storage Leakage



Allison Clark<sup>1</sup>, Arkajyoti Pathak<sup>1</sup>, Kara Tinker<sup>2</sup>, Djuna Gulliver<sup>2</sup>, & Shikha Sharma<sup>1</sup>

<sup>1</sup> West Virginia University Department of Geology and Geography, Morgantown WV, 26506

<sup>2</sup> National Energy Technology Laboratory, U.S. Department of Energy, Pittsburgh PA, 15236



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## ABSTRACT

Underground hydrogen storage (UHS) in geological formations is a key element of the clean energy transition as it enables the decarbonization of the transportation and industrial sectors by decoupling hydrogen production and storage. UHS has many benefits, including low cost, much wider availability, large storage capacity, well-established infrastructure, and increased safety because of geological sealing capabilities. However, the impact of hydrogen (H<sub>2</sub>) biogeochemical interactions in the presence of subsurface microorganisms is largely neglected from UHS perspectives. These interactions might affect the effectiveness of storage and can even cause H<sub>2</sub> to leak into the shallow aquifers through storage reservoirs. Leakage of H<sub>2</sub> into groundwater can change the geochemistry and induce several microbial-driven processes. Microorganisms, such as sulfate-reducers, are naturally abundant in groundwater and consume H<sub>2</sub> to produce hydrogen sulfide (H<sub>2</sub>S), which can contaminate the freshwater drinking groundwater and cause damage to infrastructure. Hydrogen leakage can also trigger microbial reactions responsible for metal mobility, which can impact the water quality. However, the kinetics of these reactions and the temporal impact of hydrogen leakage in groundwater are still unknown. Therefore, a time series hydrogen-groundwater interaction experiment was conducted, and the changes in fluid chemistry and headspace gas composition will be analyzed along with DNA sequencing results to understand the extent and kinetics of biogeochemical reactions that occur if hydrogen leaks into groundwater. In the experiments, Ultra High Purity (UHP) hydrogen gas will be injected into glass vials with groundwater samples, for a designated time period. For each glass-sealed vial, 16S rRNA gene sequencing, IC, ICP-MS, and GC-TCD will be performed. The experiments provide insights into plausible impacts of hydrogen leakage into shallow drinking water aquifers.

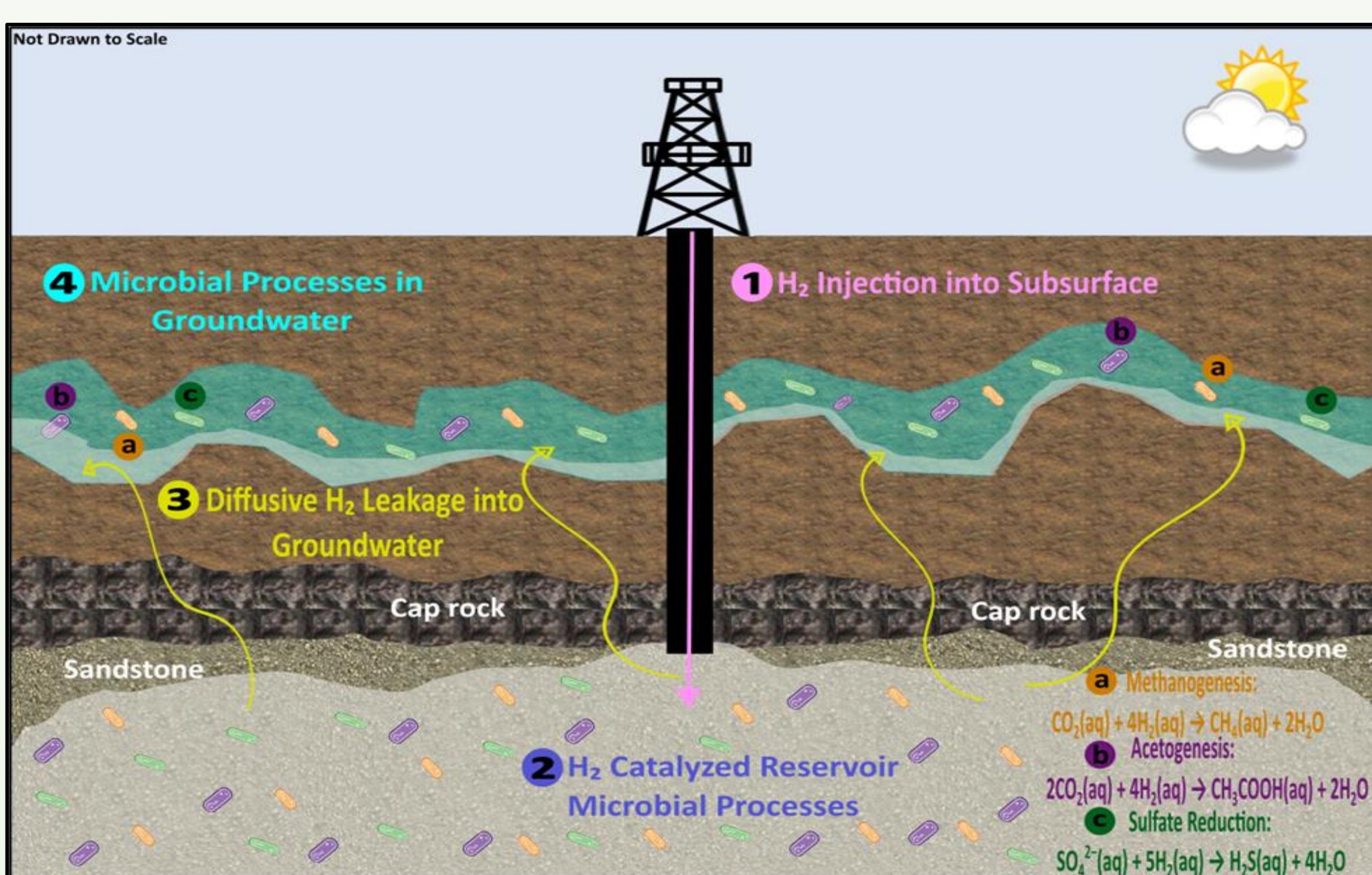


Fig 1: Overview of interactions between microorganisms and stored H<sub>2</sub> in the subsurface. Microbial processes in reservoirs consume H<sub>2</sub>. Diffusion can lead to increased concentration of H<sub>2</sub> in groundwater. Microbial processes in groundwater consume H<sub>2</sub> leading to possible contamination of groundwater.

## Key Microbial Processes

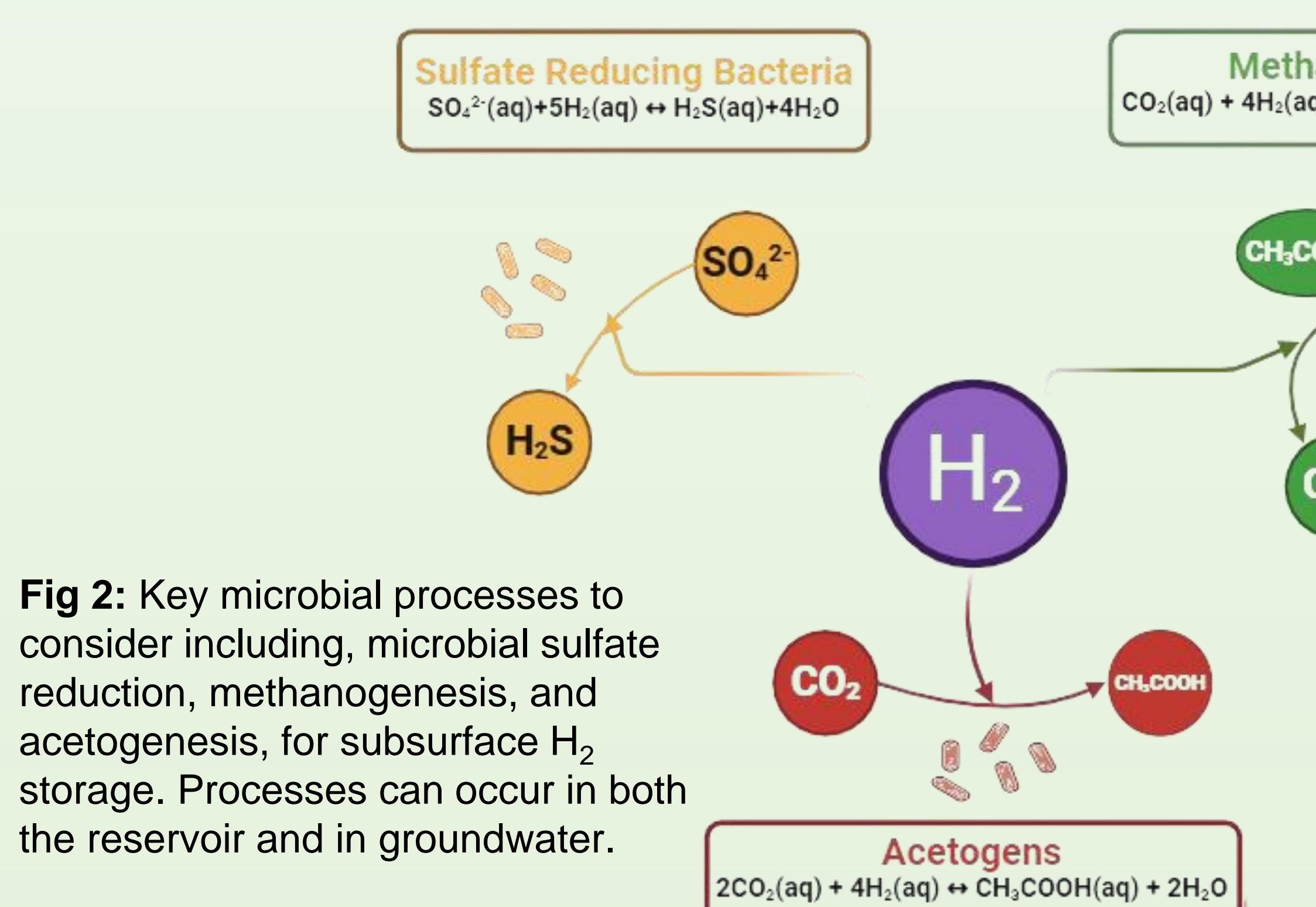


Fig 2: Key microbial processes to consider including, microbial sulfate reduction, methanogenesis, and acetogenesis, for subsurface H<sub>2</sub> storage. Processes can occur in both the reservoir and in groundwater.

## EXPERIMENTAL APPROACH

### Experimental Setup

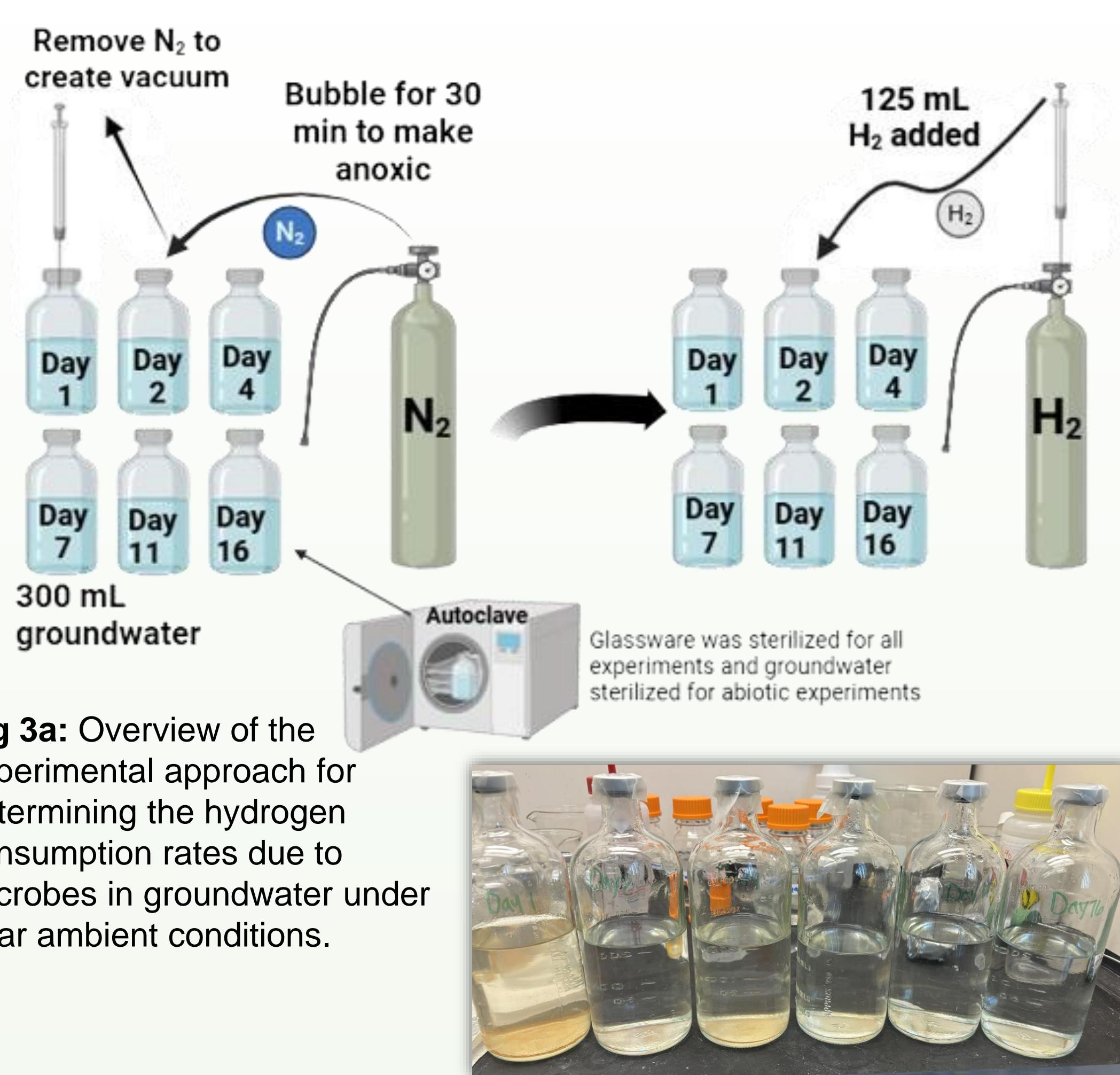


Fig 3a: Overview of the experimental approach for determining the hydrogen consumption rates due to microbes in groundwater under near ambient conditions.

### Gas and Fluid Sampling

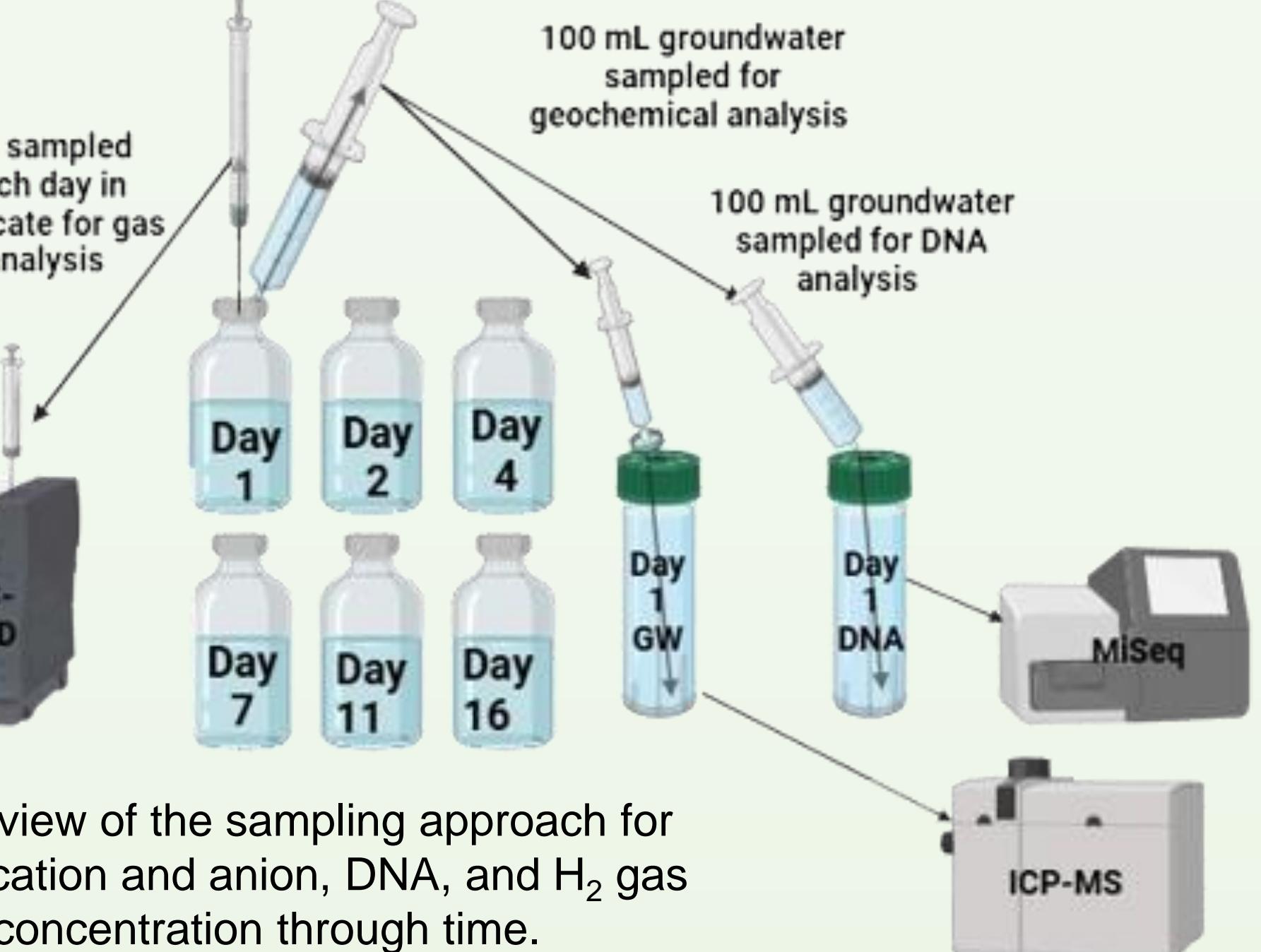


Fig 3b: Overview of the sampling approach for determining cation and anion, DNA, and H<sub>2</sub> gas composition/concentration through time.

## GAS RESULTS

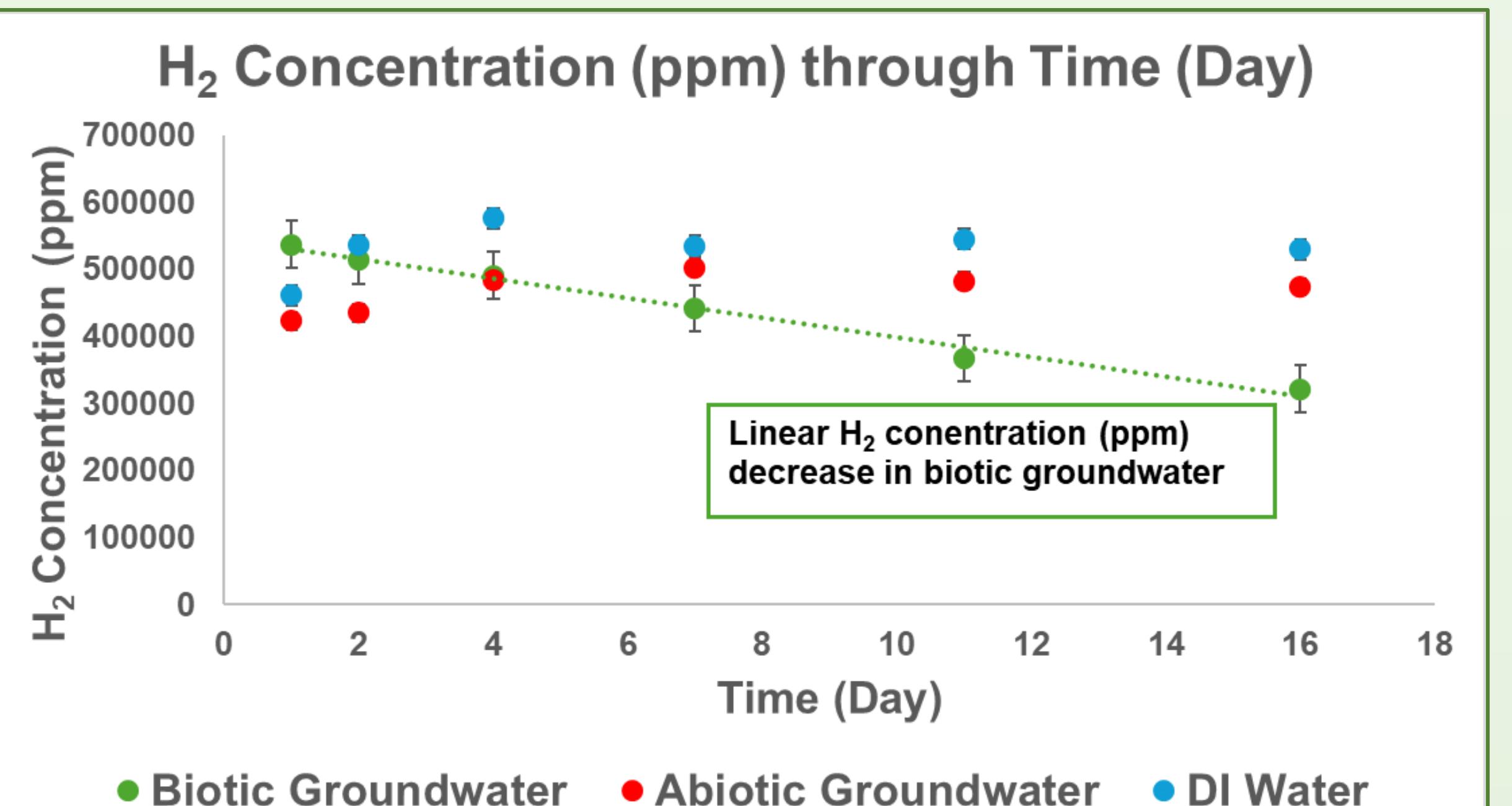


Fig 4: H<sub>2</sub> gas concentration (ppm) through time (day) for biotic and abiotic batch reactions containing groundwater and UHP H<sub>2</sub> gas headspace. DI water batch reaction conducted as blank. H<sub>2</sub> concentration adjusted to 4 mL of sample injected into GC-TCD.

## PRELIMINARY GEOCHEMICAL RESULTS

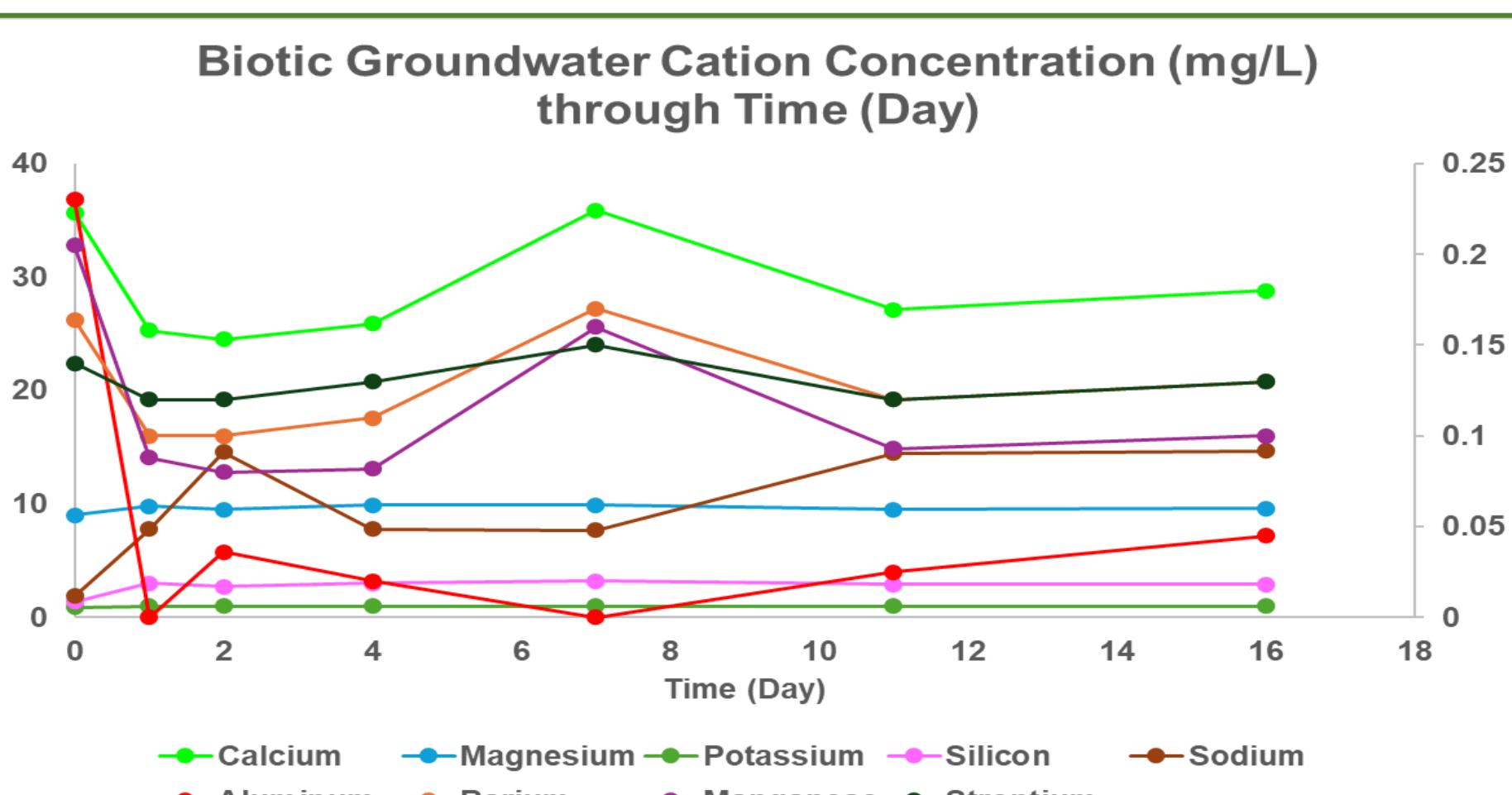


Fig 5a: Comparison of cation concentration (mg/L) of biotic groundwater subjected to UHP grade H<sub>2</sub> (ppm) through time (Day). Day 0 groundwater had no exposure to H<sub>2</sub>.

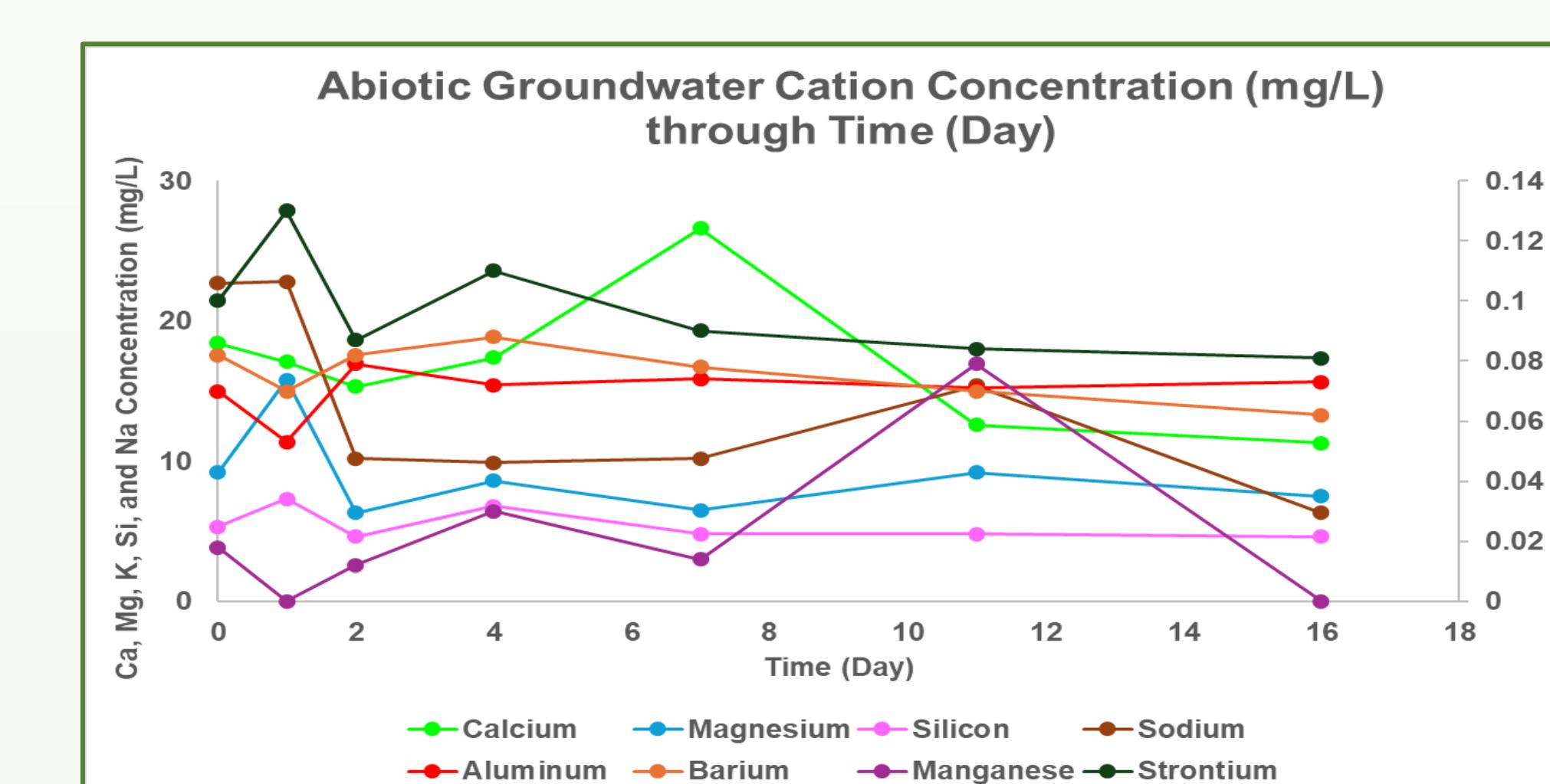


Fig 5b: Comparison of cation concentration (mg/L) of abiotic groundwater subjected to UHP grade H<sub>2</sub> (ppm) through time (Day). Day 0 groundwater had no exposure to H<sub>2</sub>.

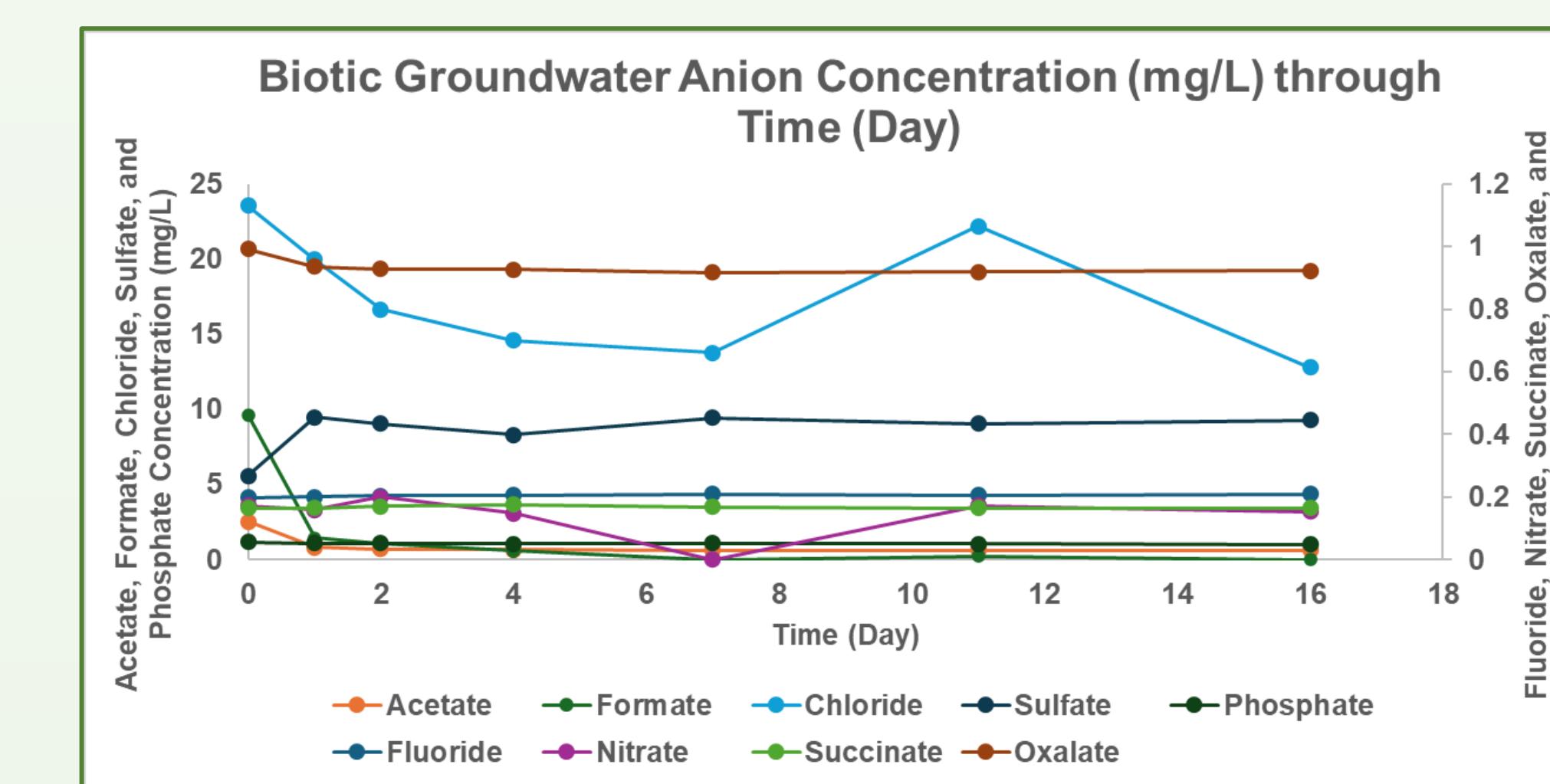


Fig 5c: Comparison of anion concentration (mg/L) of biotic groundwater subjected to UHP grade H<sub>2</sub> (ppm) through time (Day). Day 0 groundwater had no exposure to H<sub>2</sub>.

Aluminum has significant decrease in concentration from Day 0 to Day 1. Magnesium, silicon, and potassium remain constant

Barium, calcium, manganese, strontium follow a similar trend with a decrease from Day 0 to Day 1 and a peak at Day 7

Calcium, magnesium, barium, and silicon show similar trend to biotic groundwater cation concentrations

Compared to biotic aluminum does not decrease from Day 0 to Day 1. Unlike biotic sodium decreases from Day 1 to Day 2 in abiotic

Oxalate, phosphate, sulfate, succinate, fluoride, nitrate, and acetate remain constant. Chloride decreases through time and peak at Day 11

Need to complete abiotic anion analysis to make comparisons. Geochem analysis for DI water also needs to be completed

## FUTURE WORK

Collect the rest of the geochemical data for the biotic, abiotic, and DI water blank experiments

Complete DNA analysis for biotic and DI water samples

Repeat groundwater experiments with added clay minerals to explore clay mineral-H<sub>2</sub>microbe interactions



## Acknowledgements

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