



September 15, 2020 - Carbon as Pollutant and Critical Feedstock

Microalgae Biomass Production for Utilization of CO₂ and Mitigation of Greenhouse Gas Emissions

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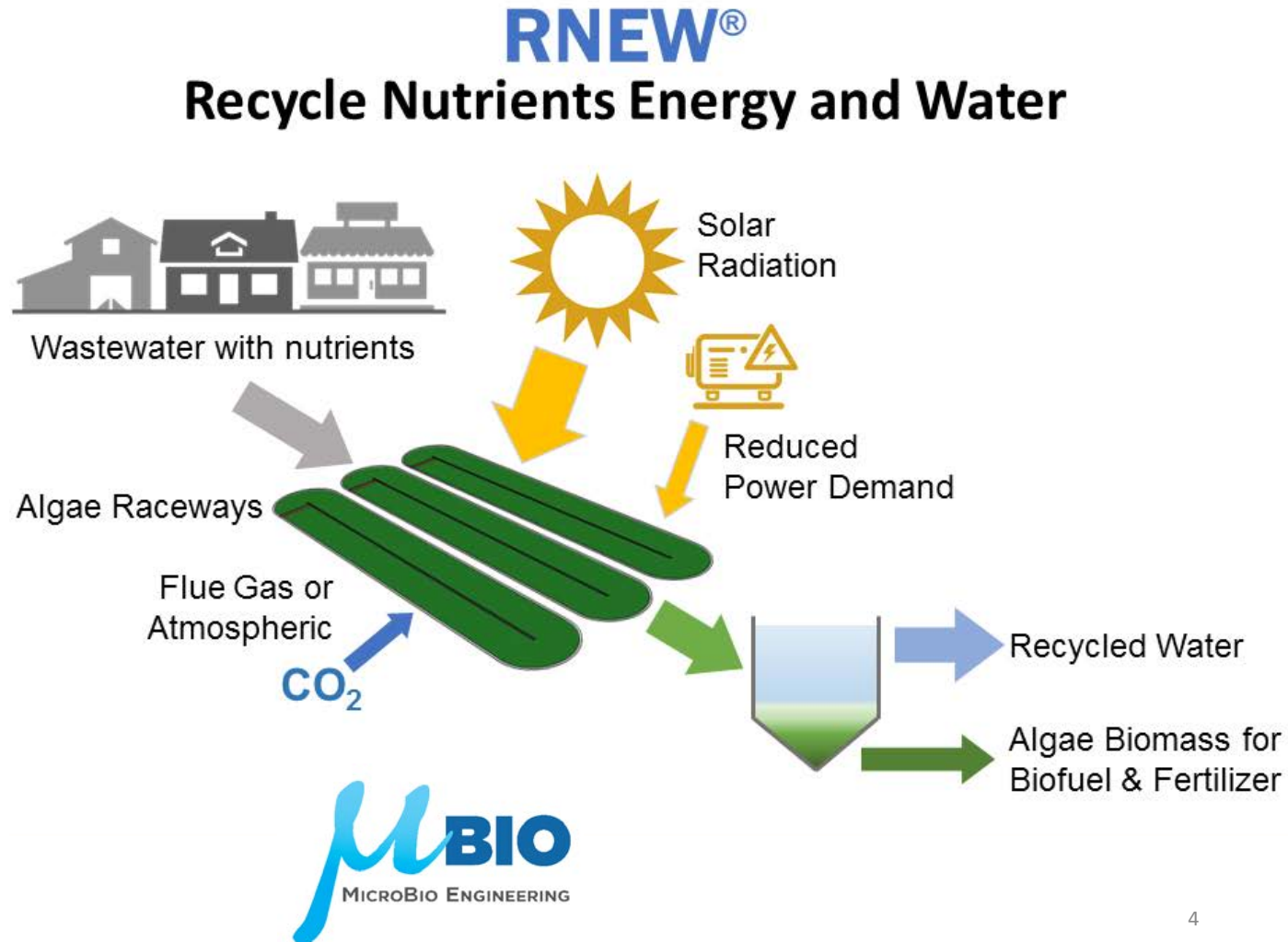
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Bioenergy Technologies Office (DOE BETO). Agreement DE-EE0008519**





- Research and Development
- R&D, Engineering, Consulting
- Facility Designs
- Algae Equipment Sales
- Techno-Economic Analyses / Life Cycle Assessments
- Wastewater treatment/ nutrients recovery
- Biofuels production and CO₂ utilization

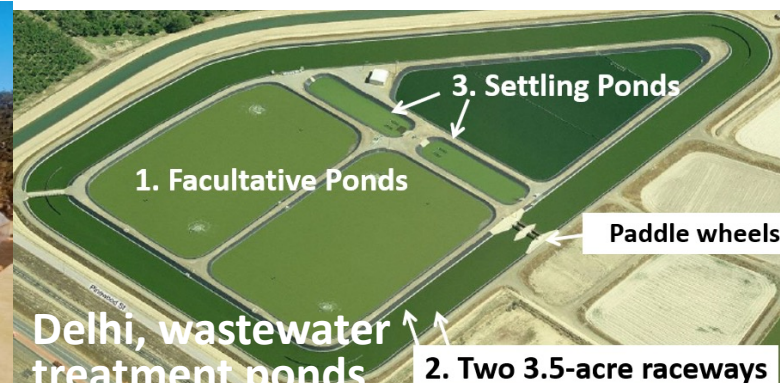
- RNEW®
Wastewater
treatment
technology for:
- ▶ Unrestricted
water reuse
 - ▶ Nutrients, N
and P, recovery
 - ▶ Co-production
of biofuels,
biofertilizers,
bioproducts



Recent and Current U.S. DOE Supported R&D Projects

- 2013 - 2016 US DOE Water & Nutrient Recycling (BETO - ASAP)*
- 2016 - 2017 Culture of Filamentous Algae on Wastewater (BETO, SBIR)*
- 2015 - 2021 CO₂ Use at Coal-Fired Power Plant (NETL (R. Spierling presentation))
- 2015 - 2018 Algae Harvesting by Bioflocculation (BETO, SBIR/STTR 1, 2)*
- 2014 - 2021 Algae Biomass Yield (BETO, ABY1,2) (See A. Davis Presentation, 9/22)*
- 2020 - 2021 Biorecovery of Nutrients from Municipal Wastewaters (SBIR)*
- **2015 - 2022 Algae Culture on air-level CO₂ (BETO 'AirFix')****

*in collaborations with Cal Poly, others ** Collaboration with PNNL, Qualitas Health, Global Thermostat



INTRODUCTION

Microalgal cultivation for production of fuels, feeds, bioproducts generally and wastewater treatment could help reduce CO₂ and other greenhouse gas emissions

CO₂ reductions would be relative to current biofuels, feed, or wastewater treatment.

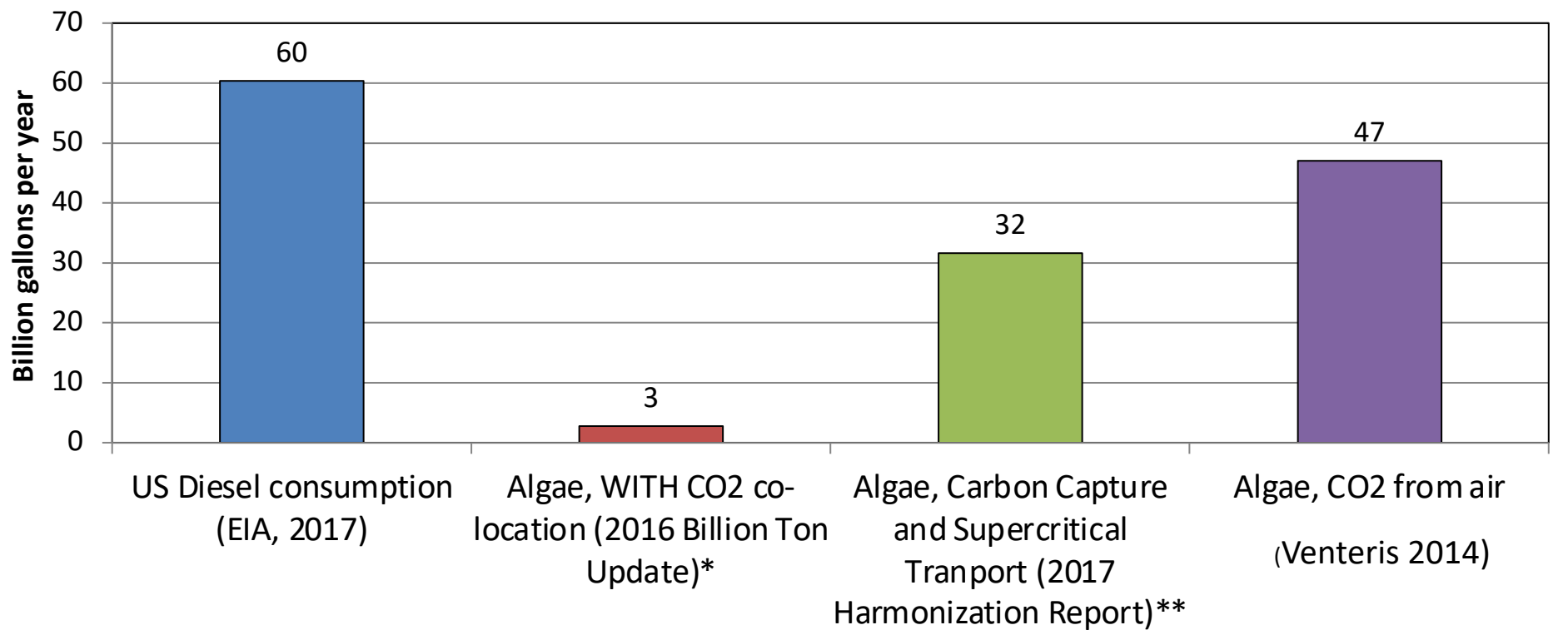
Unlike higher plants, microalgae currently require concentrated sources of CO₂.

Flue gas (~3 - 12 % CO₂) transport greatly reduces the microalgae biomass potential.

For maximum potential direct capture and utilization of CO₂ from air will be required.

This might be accomplished by the algal cultivation process itself, using large open-raceway ponds, through chemical and/or biological enhancements of CO₂ transfer.

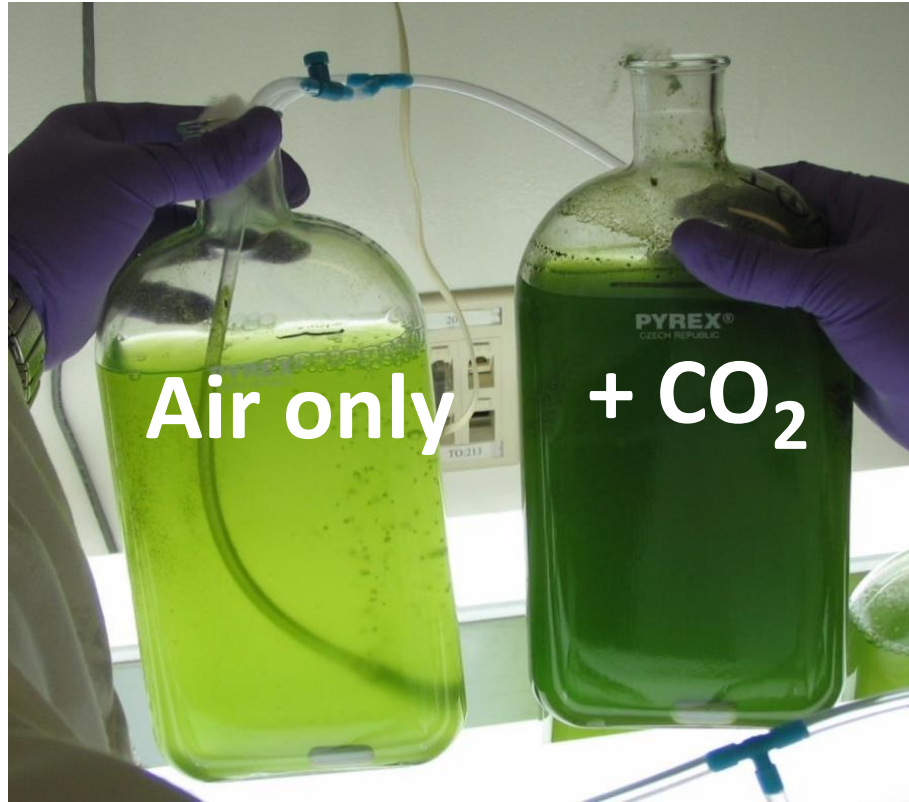
WHY PRODUCE ALGAE BIOMASS ON CO₂ FROM AIR? By decoupling algae cultivation from CO₂ point-sources, resource potential increases > 10-fold. (Alternative is flue gas carbon capture & long-distance supercritical transport)



*freshwater and saltwater scenario, present productivity. Assume 0.35 g diesel/gas per g of biomass.

**Assume 0.35 g diesel/gas per g of biomass (for comparison to the 2016 BTU case)

Growing algae on air vs. enriched CO₂ sources



Growing algae on higher CO₂ (>10x) than air levels maximizes growth rates and productivity, while allowing for complete nutrients (N, P) assimilation in wastewater treatment.

Algae can be cultivated on air-CO₂ but that reduces biomass productivity by ~25% and also requires energy for sparging with large volumes of air.

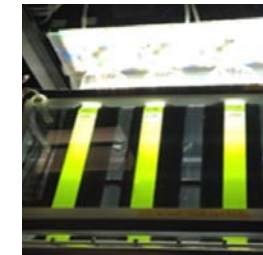
Q?: can we grow algae on air-CO₂ without sparging with air?

Increasing CO₂ Mass Transfer Rates from Air into Ponds

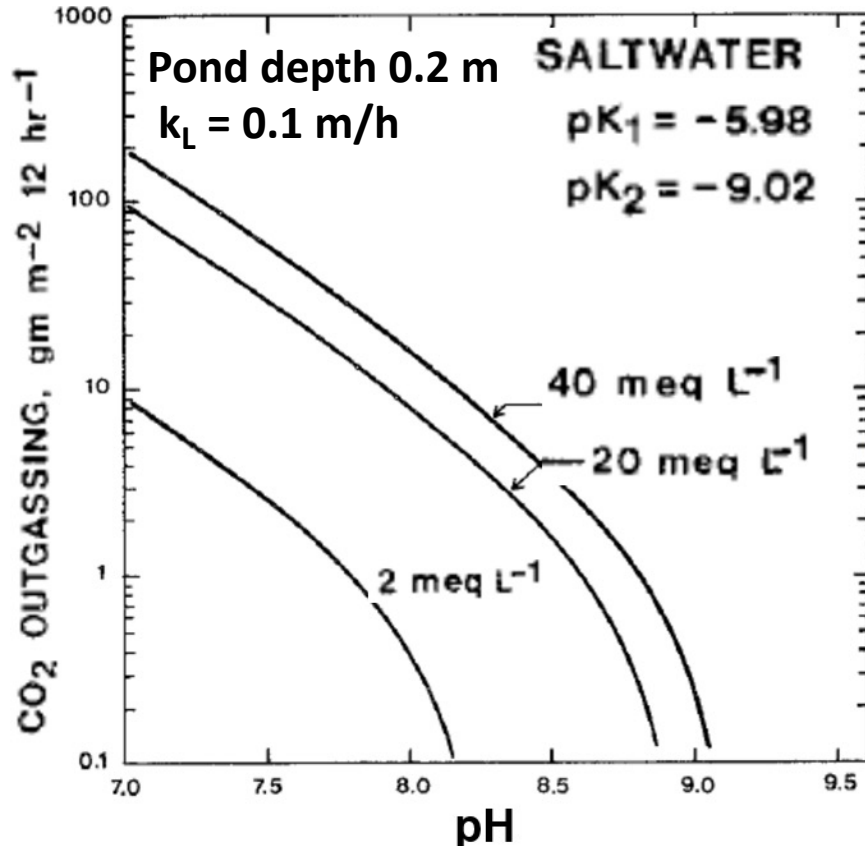
- ▶ Physical: Increased mixing (turbulence) results in a smaller boundary layer. Challenge: Energy intensive to transfer air-CO₂ at sufficient rate into large algal ponds.
- ▶ Biological: Addition of carbonic anhydrase could sufficiently increase CO₂ mass transfer. Challenge: Must be present and active in the boundary layer at the air-water interface.
- ▶ Chemical: High pH increases 'chemical-enhanced' CO₂ mass-transfer. Challenge: Requires strains with high growth rate-productivity at high pH (>11). **THIS PRESENTATION**

Objectives: Increase air-CO₂ mass transfer rates by chemical enhancement (high pH) with high biomass productivity with only air levels of CO₂

- Identify the extent to which chemical enhancement (high pH, alkalinity) would be able to support algae biomass production in small and large ponds.
- Select microalgae able to grow with air-CO₂ at the high pH required to achieve sufficiently high transfer rates
- Estimate air CO₂ mass transfer rates achievable in large production ponds (~10-acres), and their effects on production costs and the environment (TEA/LCA study)



CO₂ Flux to Atmosphere out of Algal Ponds

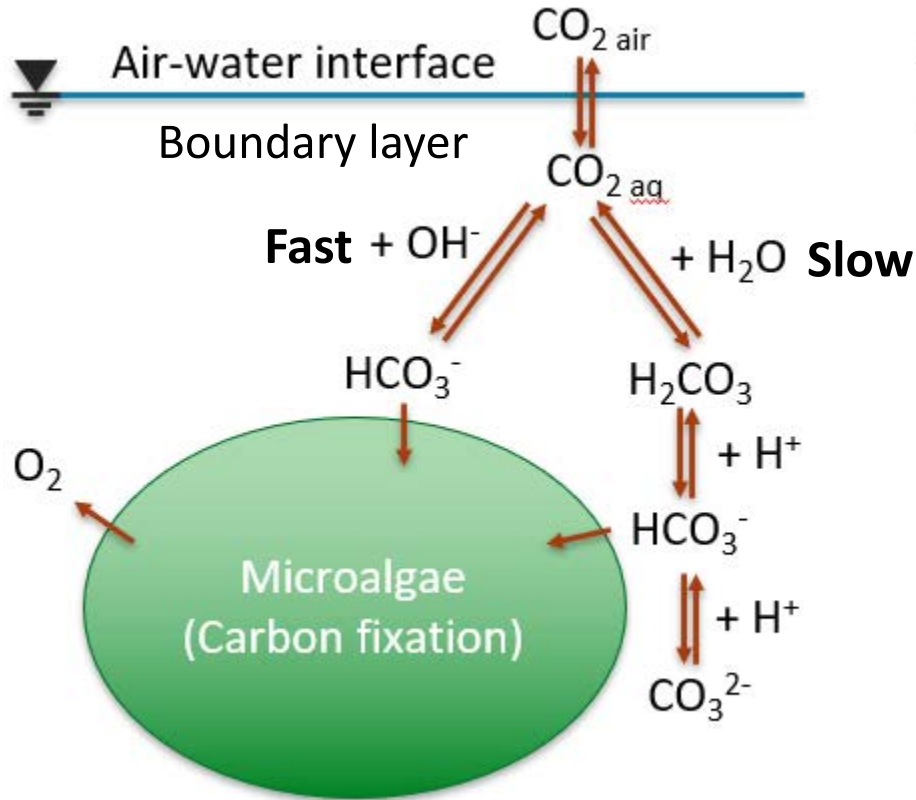


Weissman, Goebel, and Benemann, 1987.

CO₂ outgassing wastes CO₂. It is a function of pH and alkalinity and can be minimized, allowing for overall near 90% CO₂ utilization efficiency.

When pH is high enough, CO₂ will cease outgassing and start ingassing from air into the algal ponds. How high a pH? What alkalinity? What transfer rates? What is the effect of mixing?

Air- CO₂ mass transfer into ponds can be accelerated by high hydroxide ion concentrations (high pH)



Air-CO₂ flux (J_{CO_2}) can be described by Fick's law:

$$J_{\text{CO}_2} = k_L^0 * \underbrace{(P_{\text{CO}_2}^{\text{air}} * H - C_{\text{CO}_2}^*(x = \delta))}_{\text{"Driving force"}}$$

↓
"Piston velocity"
↓
"Back pressure"

Where:

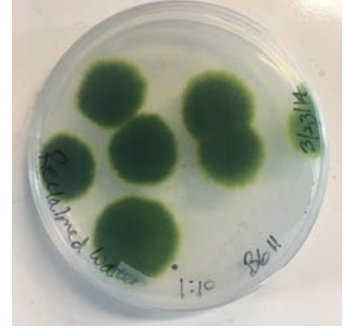
k_L^0 = mass transfer coefficient, m/hr, a function of pond hydraulics

$P_{\text{CO}_2}^{\text{air}} * H$ = Dissolved CO₂ concentration in equilibrium with air

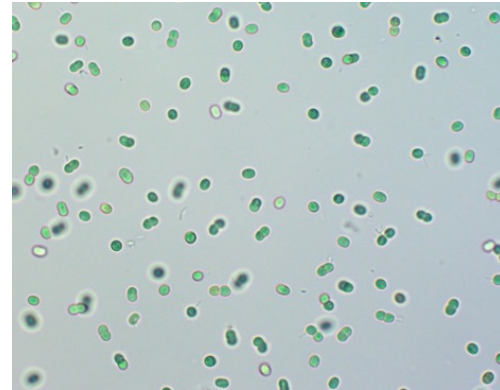
$C_{\text{CO}_2}^*(x = \delta)$ = Dissolved CO₂ concentration some distance from the air-water interface

Bioprospecting for alkaliphilic strains

- Obtain inoculum from high pH environments (possible locations: Soap Lake, Soda Lake, Mono Lake, Searles Lake, Walker Lake, Owen's Lake)



***Cyanobacterium* sp.
SSL1 (isolated by PNNL)**

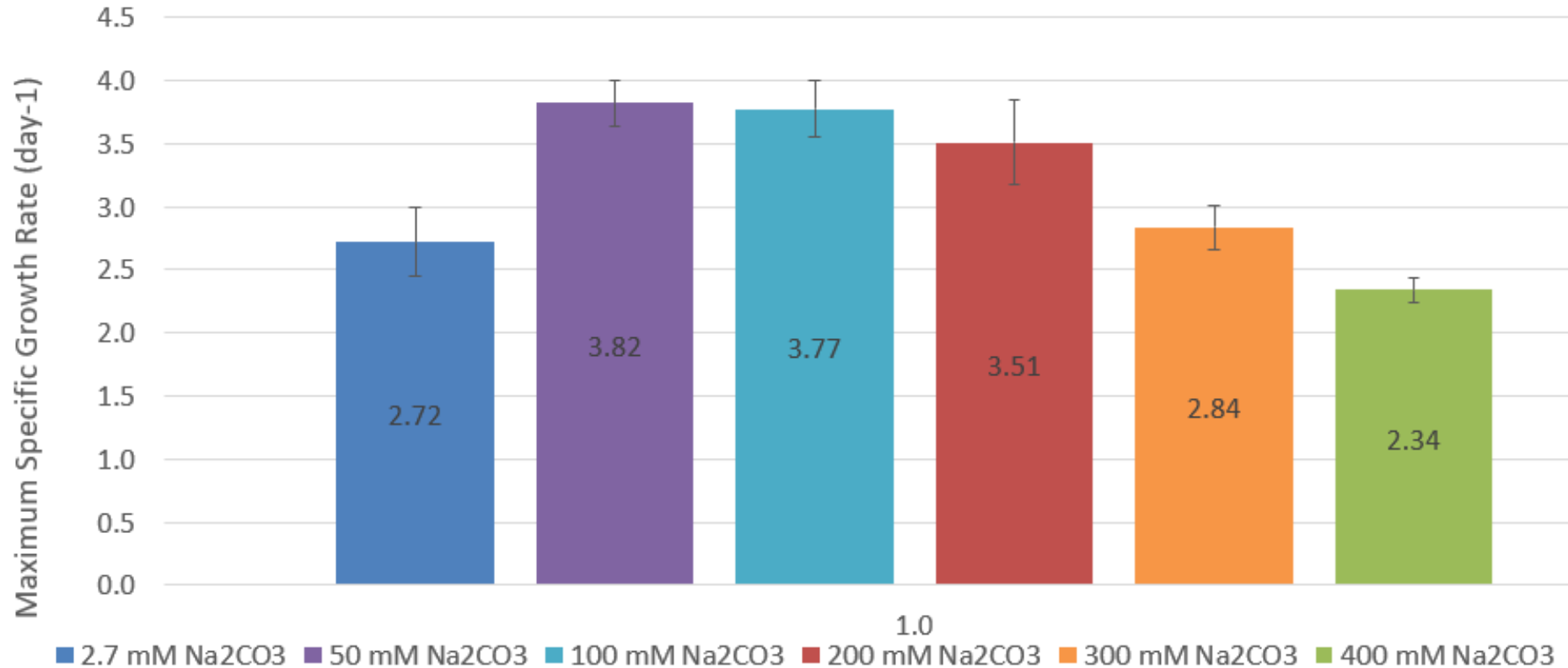


Left: Soap Lake, WA. Right: Searles Lake, CA

- Obtain isolates using plating
- or FACS sorting



Objective: select alkaliphilic strains with high productivity at high pH:
Maximum specific growth rate as function of alkalinity, pH 11



CO₂ ingassing from air: being studied at large ponds at Qualitas Health, Imperial, TX.



Carbon balances both calculated from known pH and alkalinity, and directly measured as DIC

By Calculation (known pH, alkalinity):

$$A_c = A_T - ([B(OH)_4^-] + [OH^-] + [HPO_4^{2-}] + 2[PO_4^{3-}] \\ + [SiO(OH)_3^-] + [NH_3] + [HS^-] + ... \\ - [H^+]_F - [HSO_4^-] - [HF] - [H_3PO_4] - ...)$$

$$[CO_3^{2-}] = \frac{A_c K_2}{[H^+] + 2K_2} \quad [HCO_3^-] = \frac{A_c [H^+]}{[H^+] + 2K_2} \quad [CO_2^*] = \frac{A_c [H^+]^2}{K_1 ([H^+] + 2K_2)}$$

DIC = Carbonate + Bicarbonate + Carbon dioxide

DIC is a function of K1, K2, and KW.

Water-source and temperature specific equilibrium coefficients must be used

Direct measurement of DIC:



MBE, Qualitas Health, and PNNL all use in-house TOC, DIC analyzers.

Analytical round robin completed.



**Abiotic ingassing-outgassing experiments
now underway at the Imperial, Texas,
facility of Qualitas Health.**



Bicarbonate Evaporation Pond Near Mexico City - Site of first Spirulina production facility (1974 -1995): CO₂ was provided by a combination of air and bicarbonate. NOTE: Chinese Spirulina producers still use this process. But productivity is low and bicarbonate addition expensive.



Conclusions

- Point-source CO₂ co-location limits resource potential
- 2nd generation carbon capture technologies combined with compressed CO₂ transport are expensive.
- CO₂ transfer from air requires very high pH (>11) for sufficient C transfer, which reduces productivity.
- Strains are required able to grow well at high pH.
- Abiotic & biotic studies required to establish feasibility.
- Feasibility requires low energy and chemical inputs.
- Small scale pond results must be verified, and at scale.

Acknowledgement:

Support by DOE-BETO,
Agreement DE-EE0008519



NEXT PROJECT: Microalgae Commodities Production with a Direct Air Capture Process

**Topic Area 3: Algae Bioproducts and CO₂ Direct-Air-Capture Efficiency (ABCDE)
DE-FOA-0002203 Control Number: 2203-1728. Award Number: DE-EE0009276**

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Charley O' Kelley, Gerry Cysewski – Cyanotech Inc.

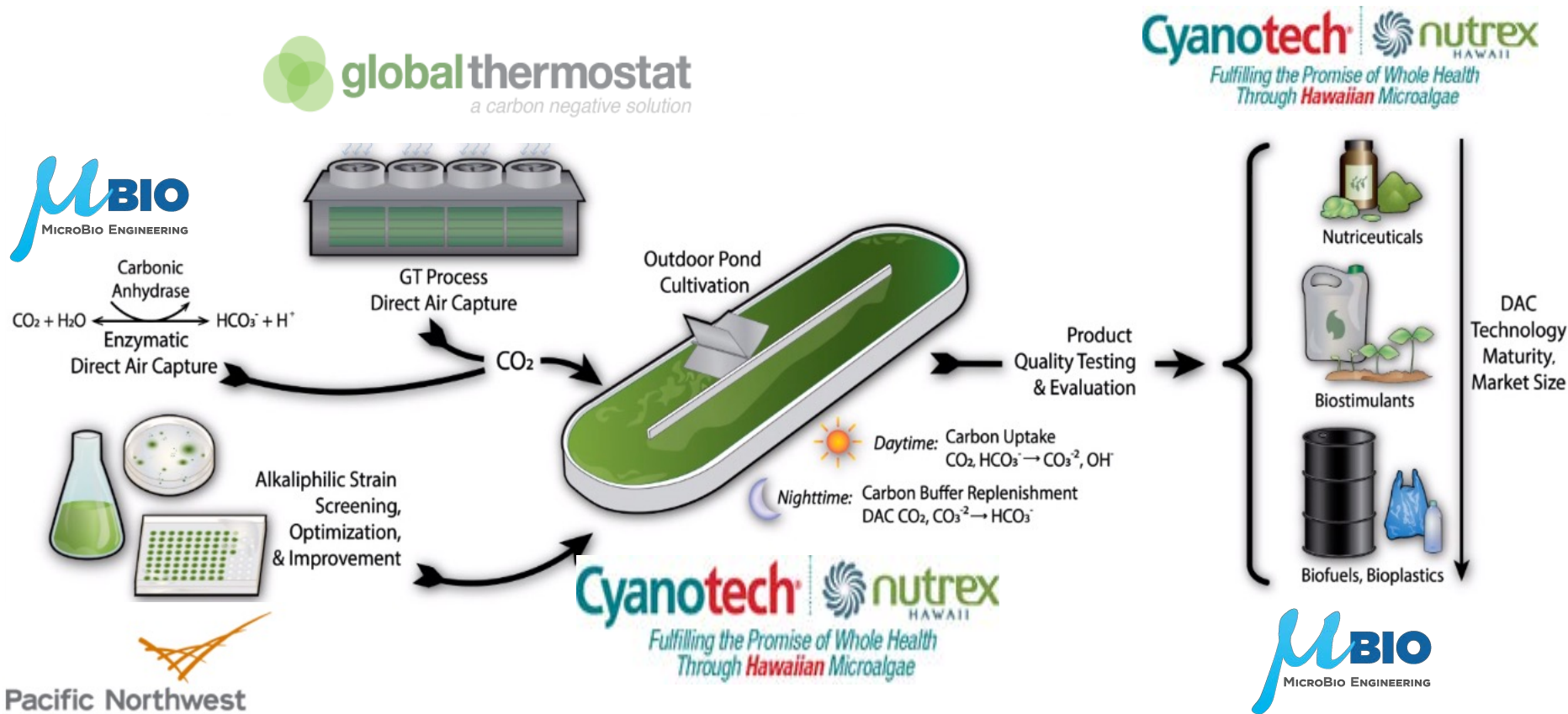
Ron Chance, Eric Ping, Hans Kistenmacher – Global Thermostat Inc.

Michael Huesemann, Scott Edmundson, Song Gao – PNNL

Period of Performance: October 2020 to December 2022.



Project Overview



Paddle wheels →



Cyanotech®


THANK YOU!





Scale-Up
Design
TEA-LCA

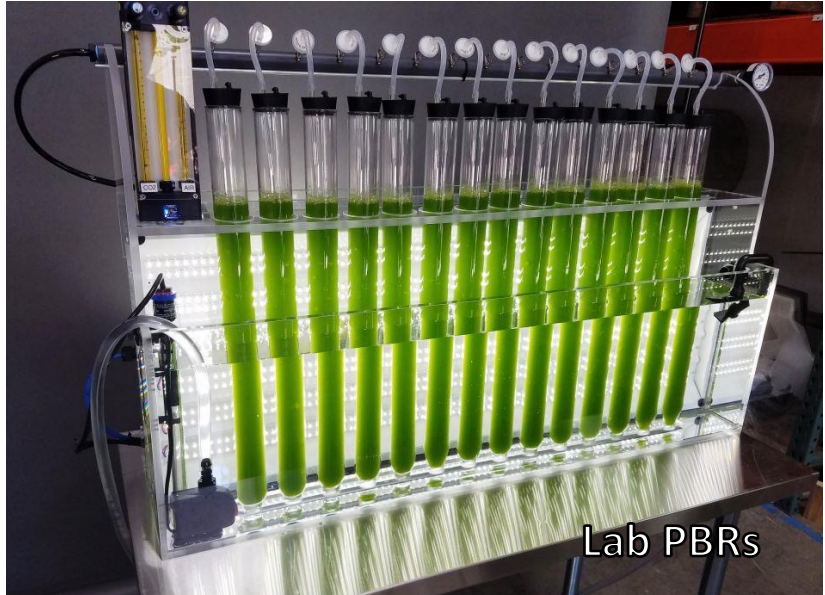
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Full-Scale
Equipment



Pilot
Facilities



Lab PBRs

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