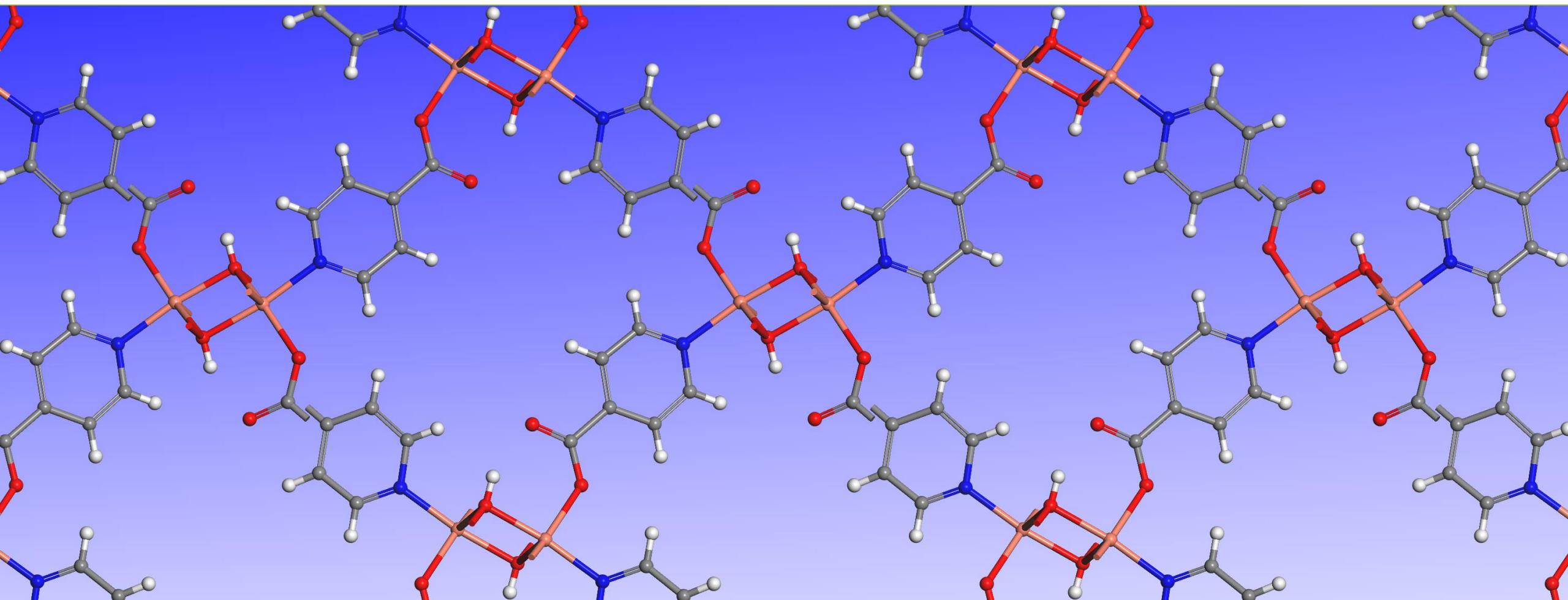


# Machine Learned Force Field Modeling of Metal Organic Frameworks for CO<sub>2</sub> Direct Air Capture

John Findley, Samir Budhathoki, Jan Steckel



ACS Meeting, August 2024



# Acknowledgements



## Co-authors:

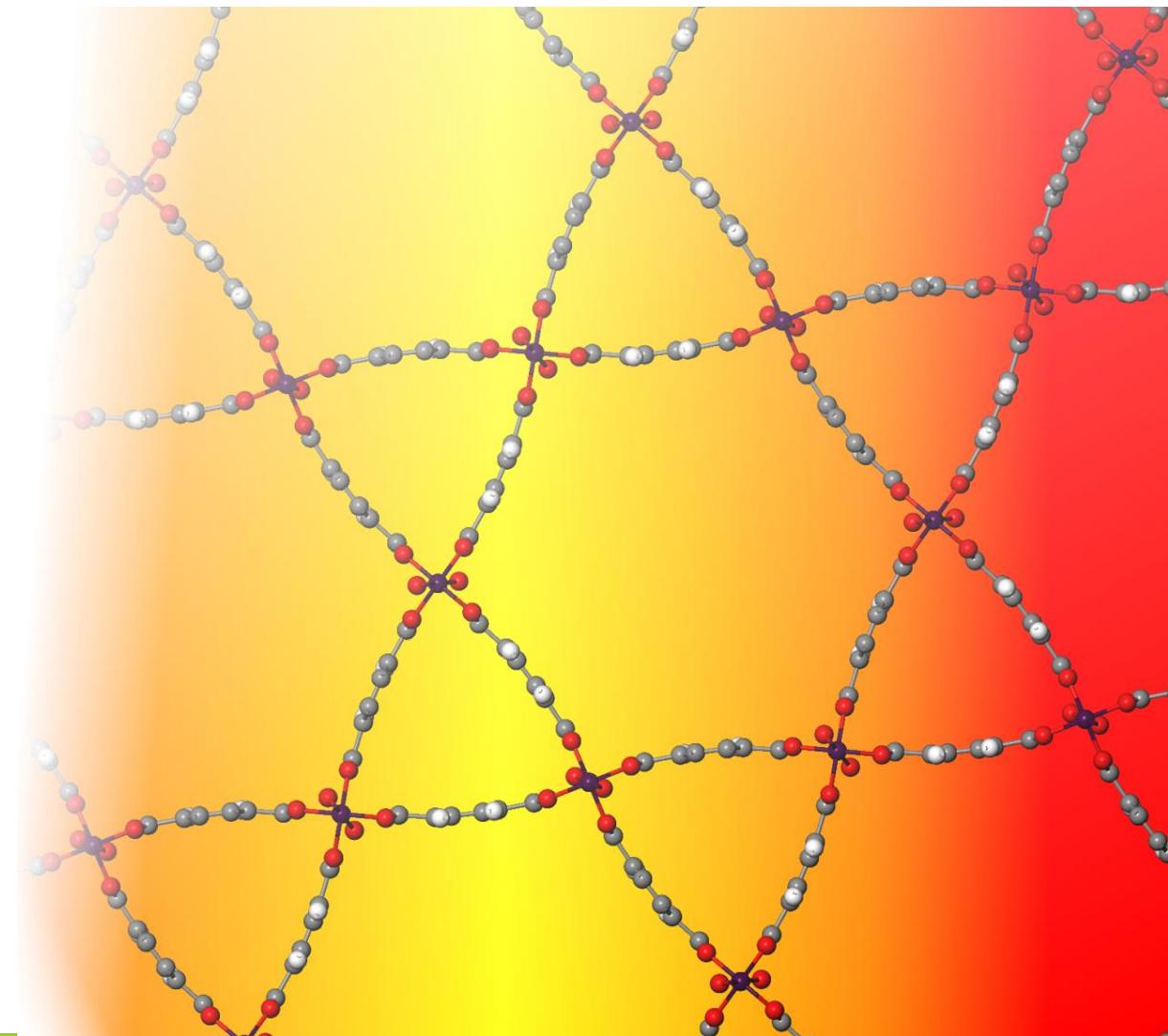
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- Clare Martin, Ugrad, currently at University of Michigan
- Laine Roper, Ugrad, currently at Harvard

## Collaborators:

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- Dan Sorescu - NETL



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**ENERGY**

# Outline and Disclaimer



Outline: Metal Organic Framework Modeling using three methods:

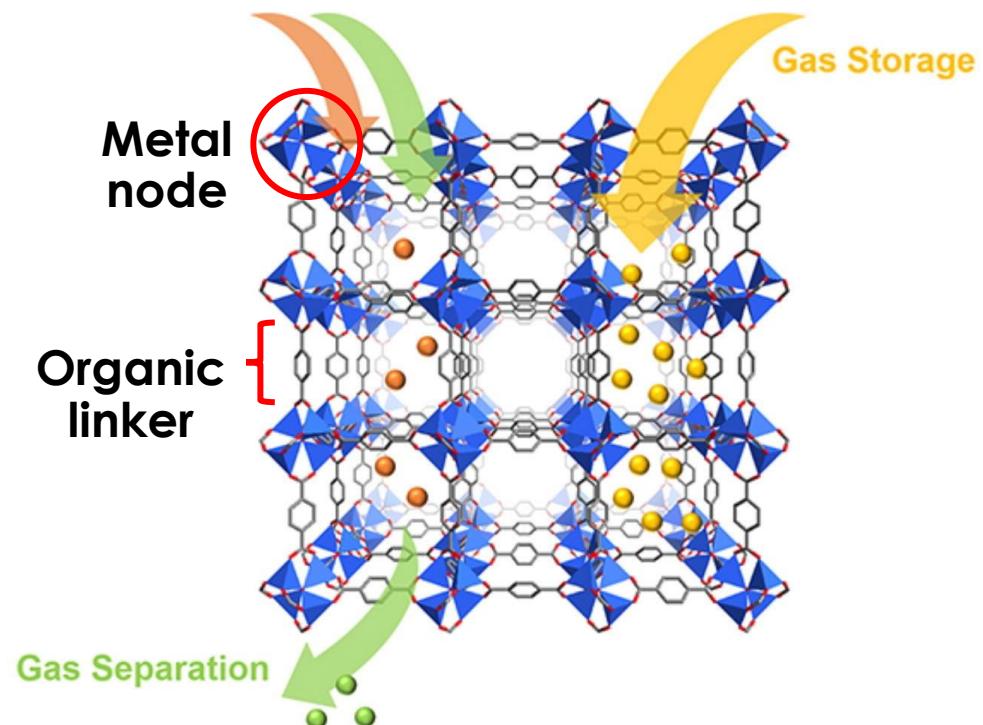
1. Classical (Rigid) Force Field: Universal Force Field (UFF)
2. Classical Flexible Force Field: QuickFF
3. Machine Learned Force Field: FitSNAP

## Disclaimer:

This project was funded by the United States Department of Energy, National Energy Technology Laboratory, in part, through a site support contract. Neither the United States Government nor any agency thereof, nor any of their employees, nor the support contractor, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.

# Metal Organic Frameworks (MOFs)

- Crystalline, porous materials
- Good for gas storage/separation applications
- Large, diverse class of molecules
  - >100,000 synthesized
  - >500,000 predicted



# Physisorbent MOFs for CO<sub>2</sub> Capture

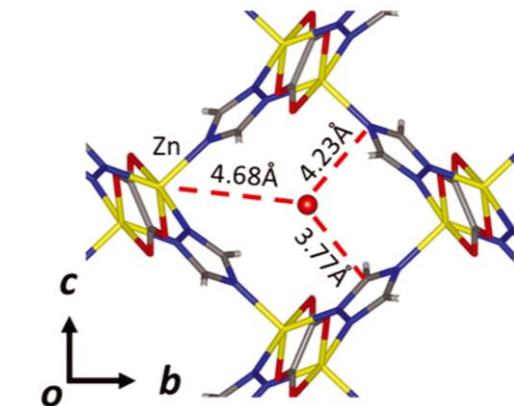
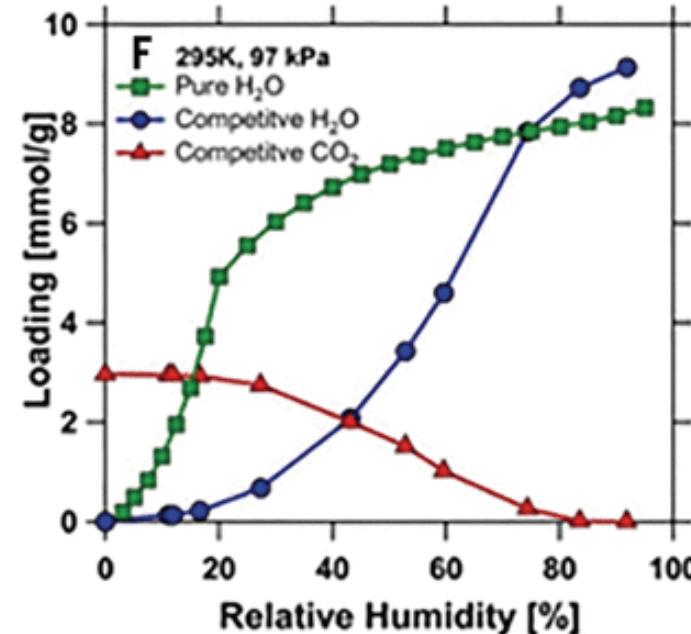


Characteristics of a good capture sorbent

- Strong CO<sub>2</sub> adsorption
- Regenerable
- Not hindered by humidity

Example: CALF-20

- Successfully used for point source
- Little effect of humidity up to 30% RH
- Strong **dispersion** interactions (CO<sub>2</sub> > H<sub>2</sub>O)

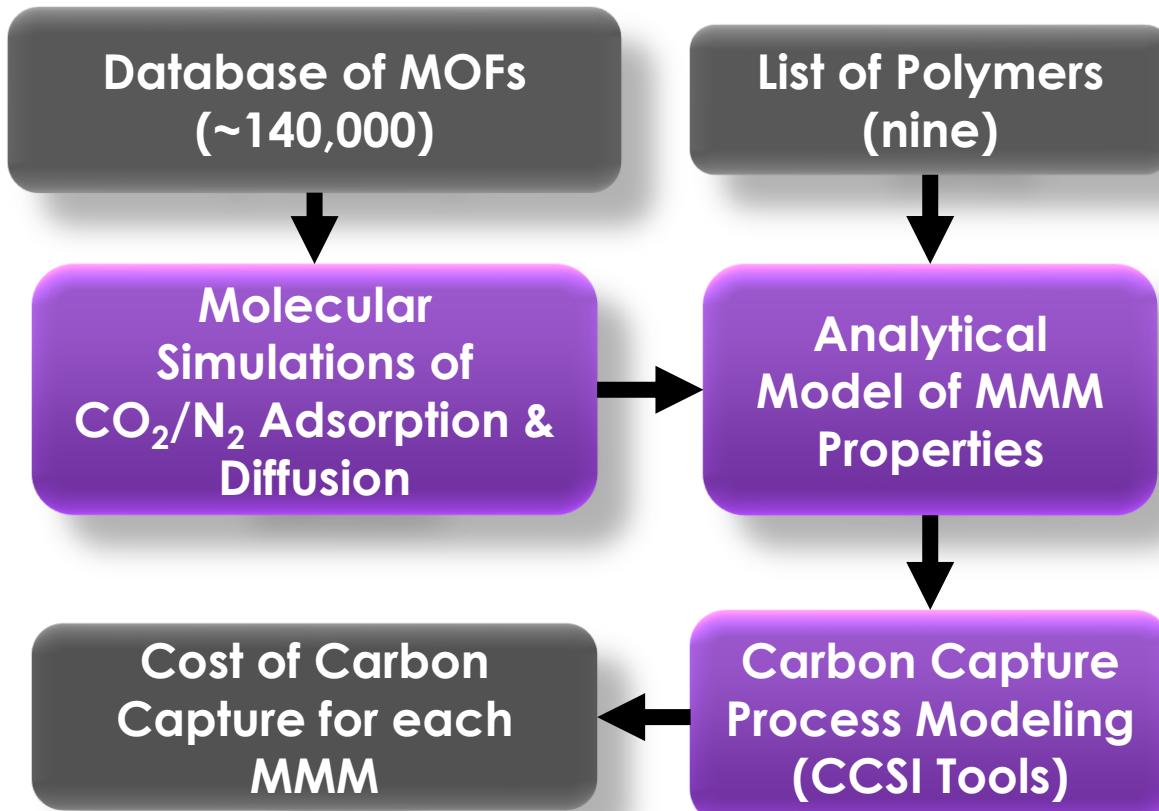


# Computational Screening for Mixed Matrix Membranes (MMMs)

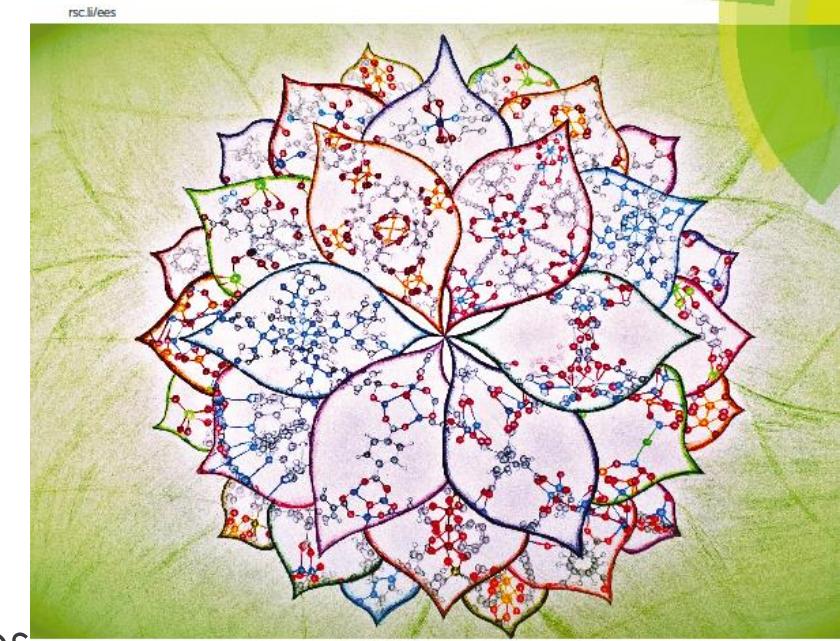
MMMs can be made from Combining MOFs and Polymer Materials



Volume 12 Number 4 April 2019 Pages 1111–1430



Energy & Environmental Science



Predicted Cost of Capture Process for >1 million membranes

Samir Budhathoki, Kayode Ajayi, Christopher E. Wilmer



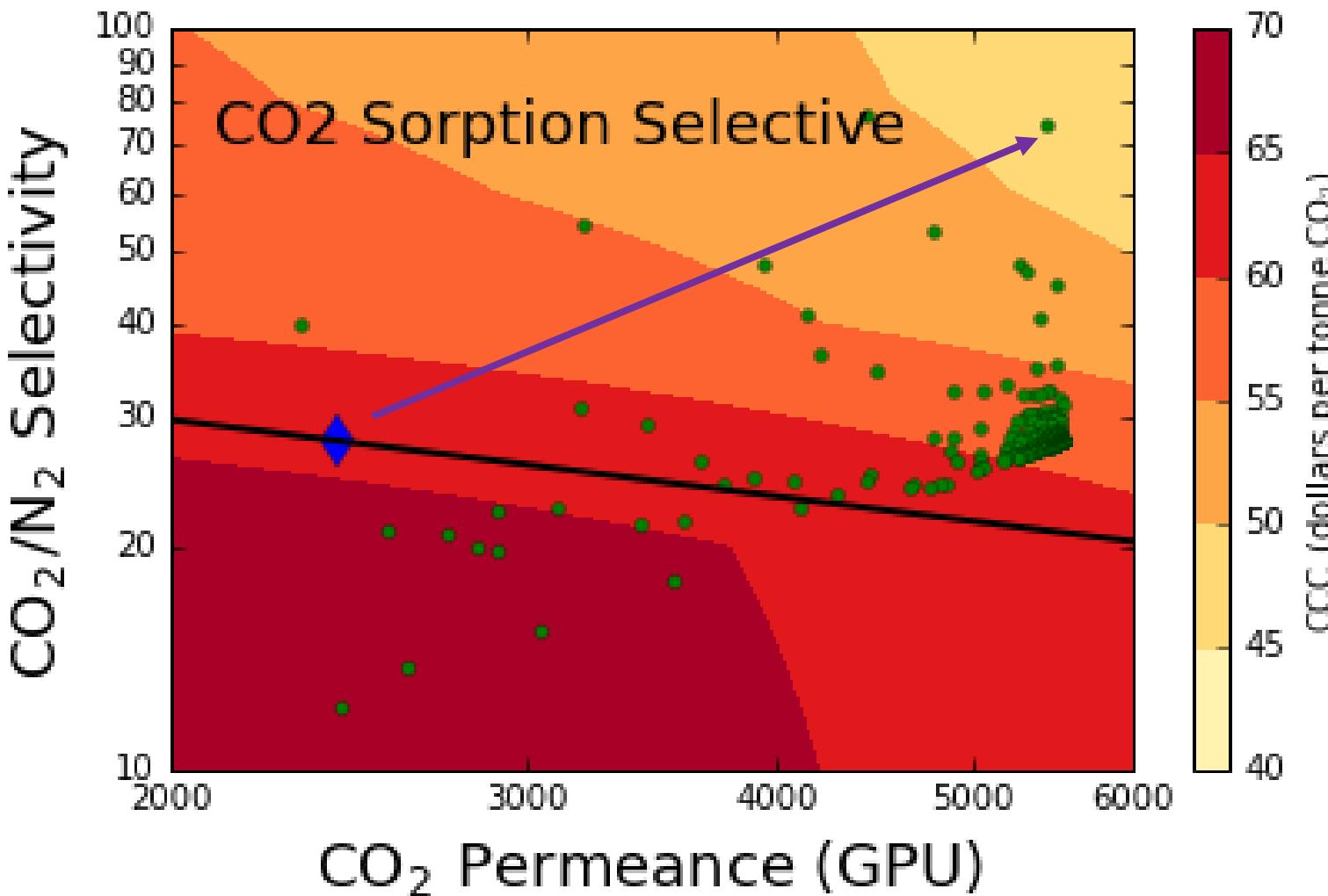
ROYAL SOCIETY  
OF CHEMISTRY

Celebrating  
IYPT 2019

PAPER

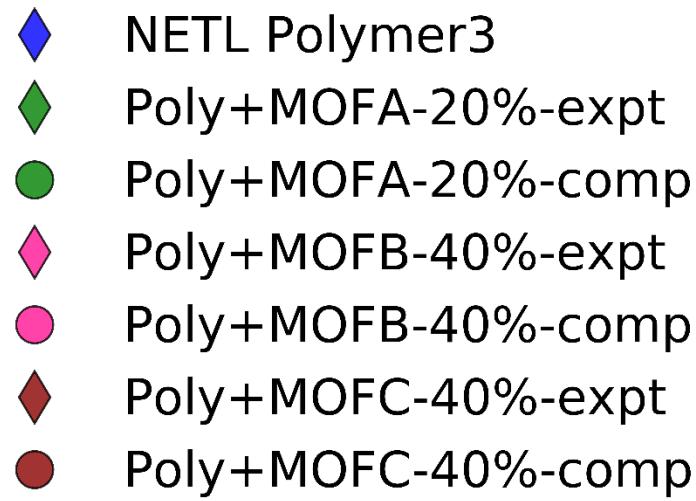
James A. Steckel, Christopher E. Wilmer et al.  
High-throughput computational prediction of the cost  
of carbon capture using mixed matrix membranes

# Predictions from High Throughput Computational Screening

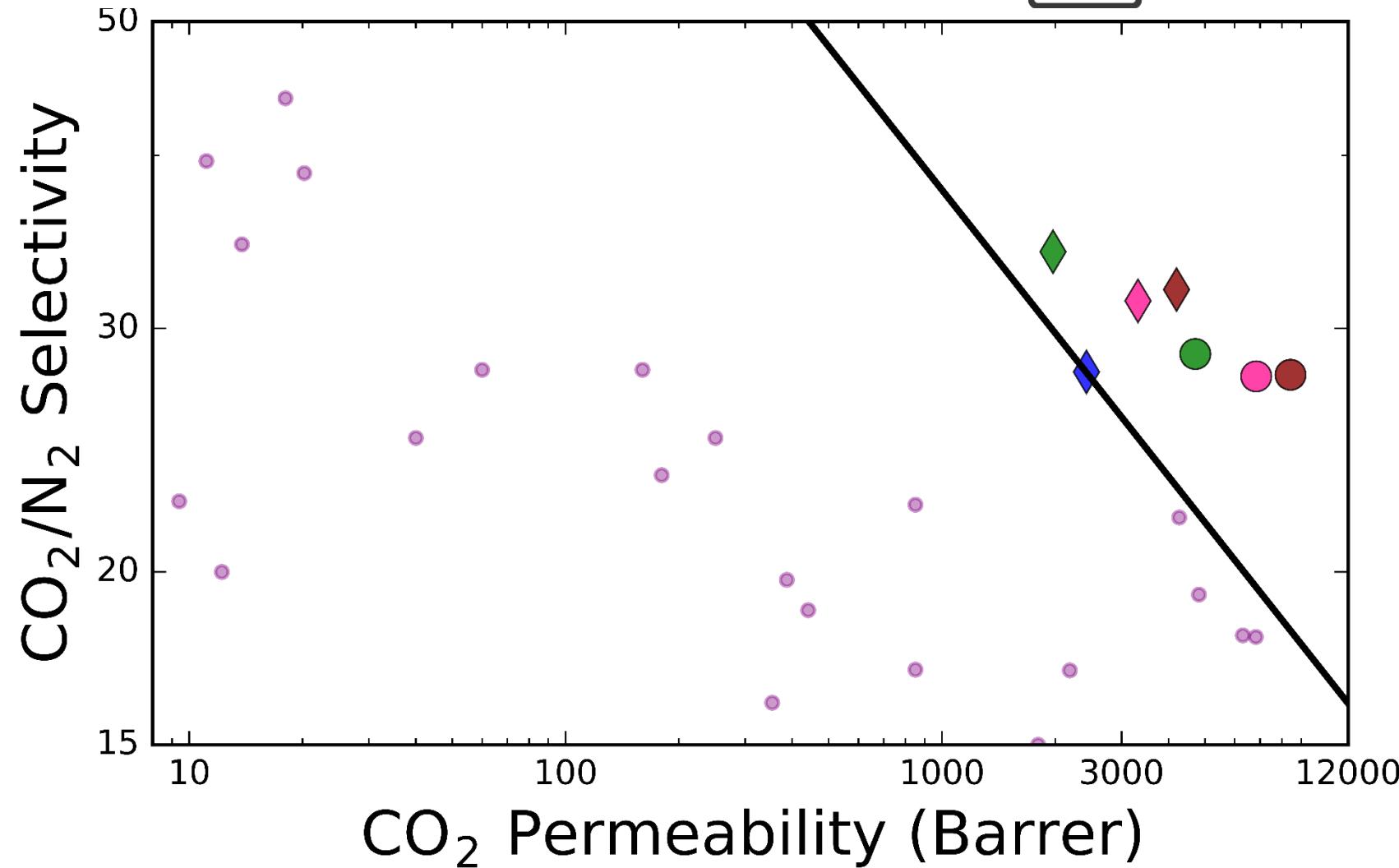


- MMMs using NETL Polymer 3 (blend polymer)
- Best MMM in this set:
  - Predicted CCC Reduction from \$62.9 to \$42.7 per tonne CO<sub>2</sub>
- Tool for selecting MOFs to pair with polymers

# Comparison with Experiment

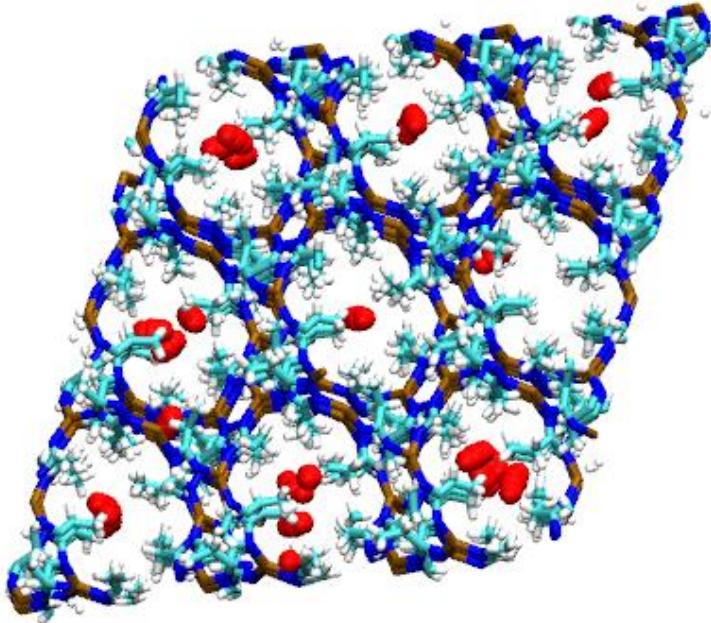


- Predictions have the correct trend and reasonable accuracy



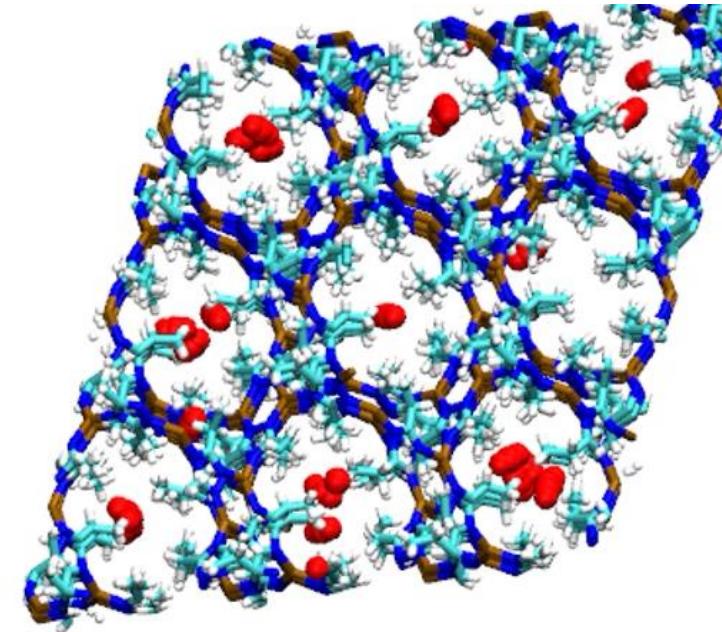
# MOF Computational Screening for CO<sub>2</sub> Capture

Past large scale screening studies have modeled MOFs atoms as rigid



## Rigid Force Field

- Electrostatic + dispersion
- Low computational cost
- Good for rigid MOFs but most MOFs are flexible
- Easy to obtain

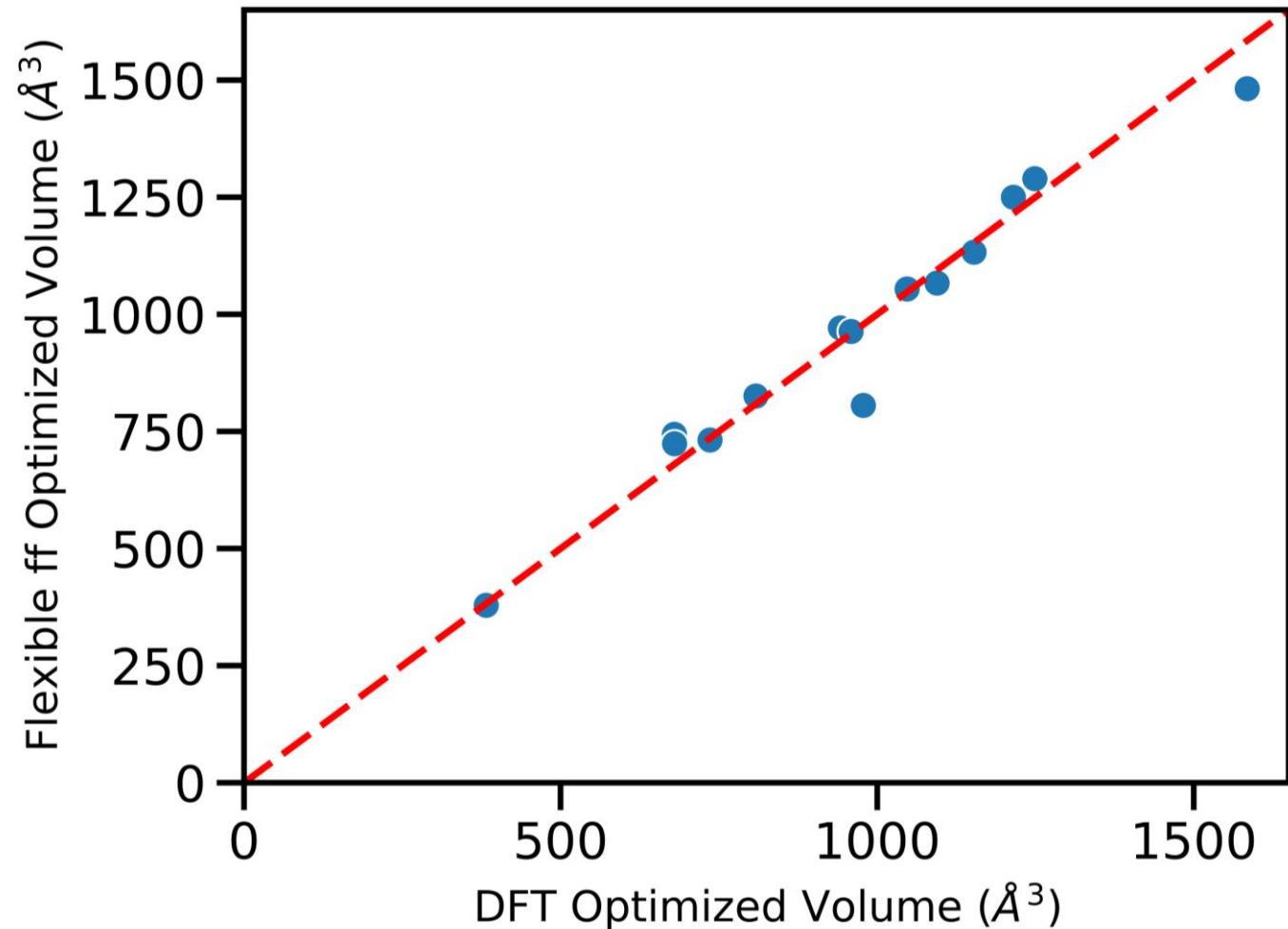
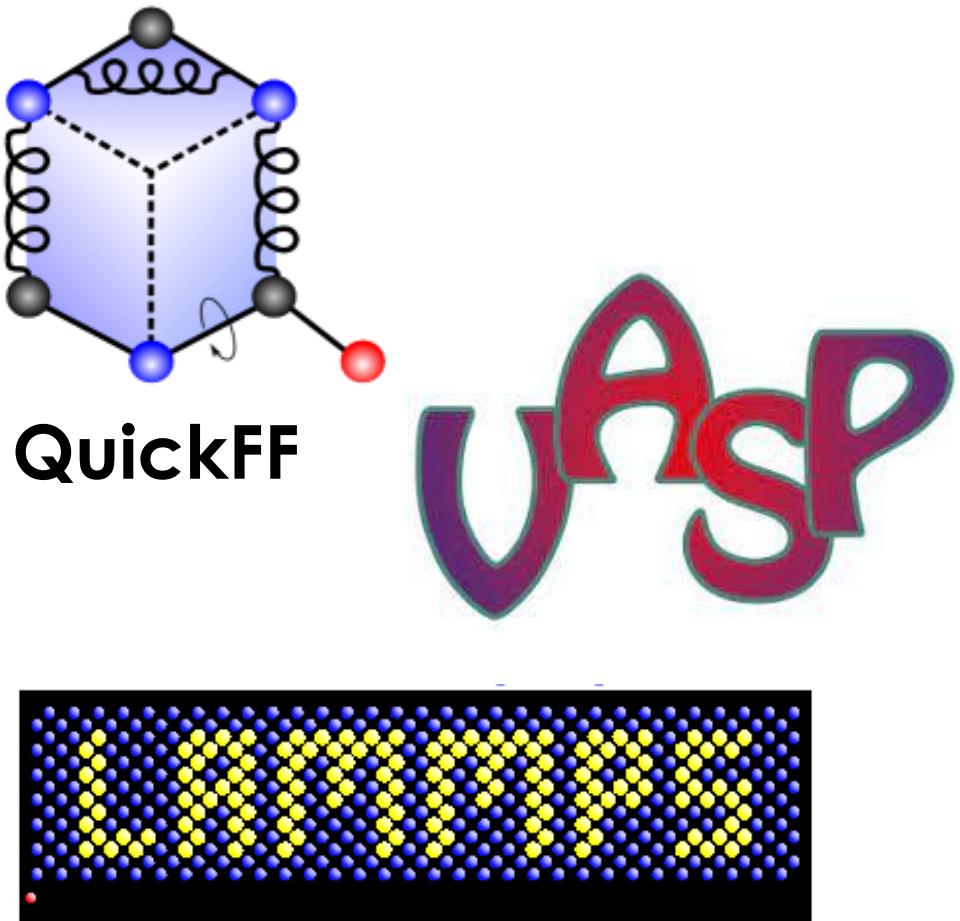


## Flexible Force Field

- Electrostatic + dispersion + bond + angle + torsion
- High computational cost
- Accounts for MOF flexibility
- Hard to obtain for a large set of materials

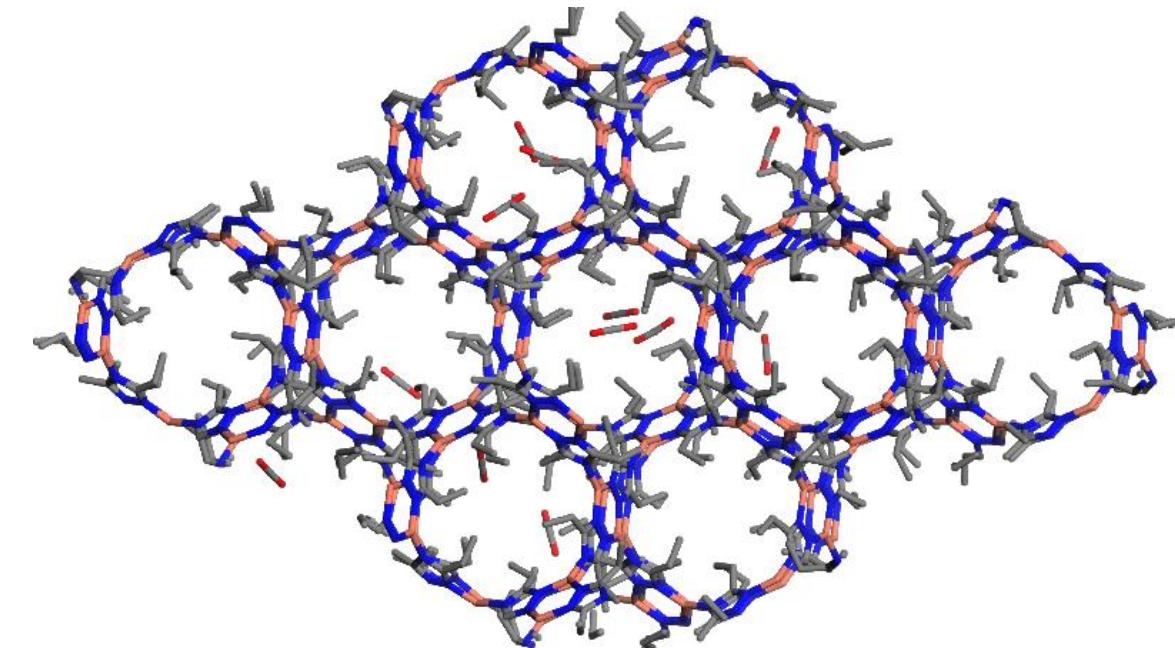
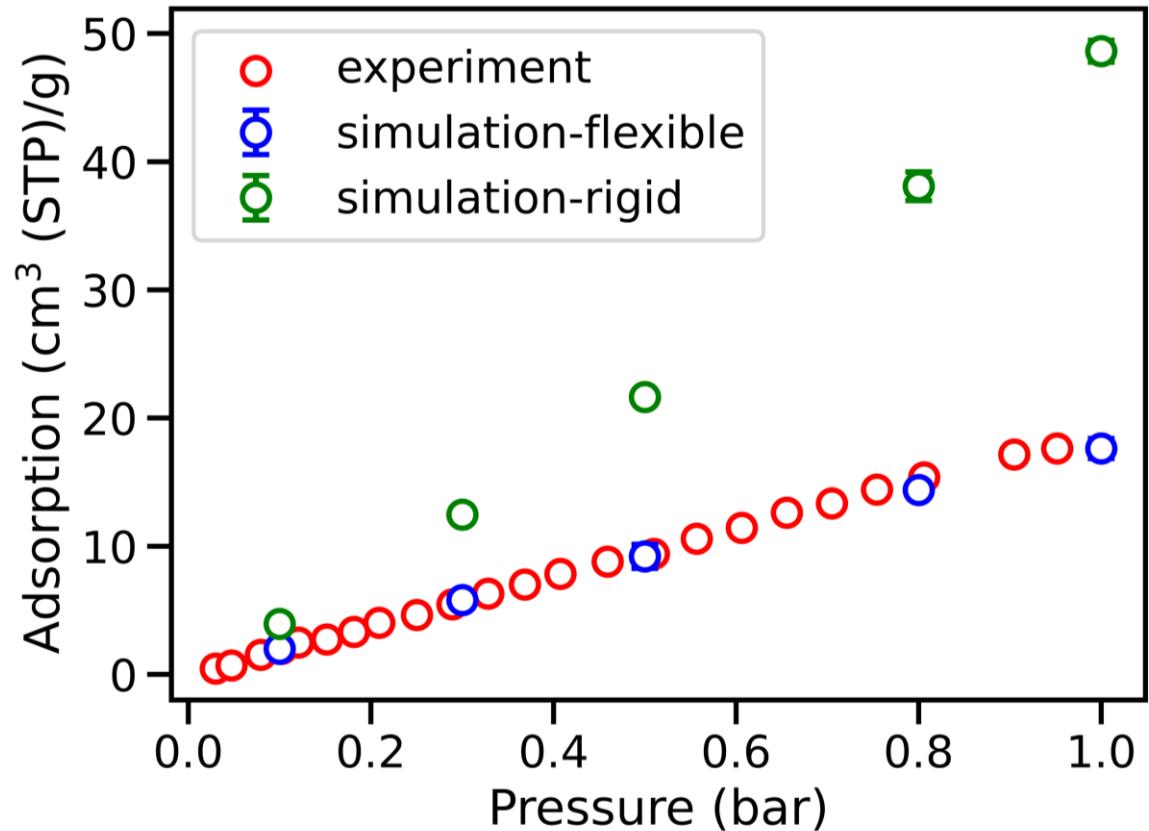
# Flexible Force Fields for MOFs

Quick FF used with DFT Vibrational Frequency Calculations to fit flexible FFs



# Calculated Isotherm: Flexible vs Rigid FF

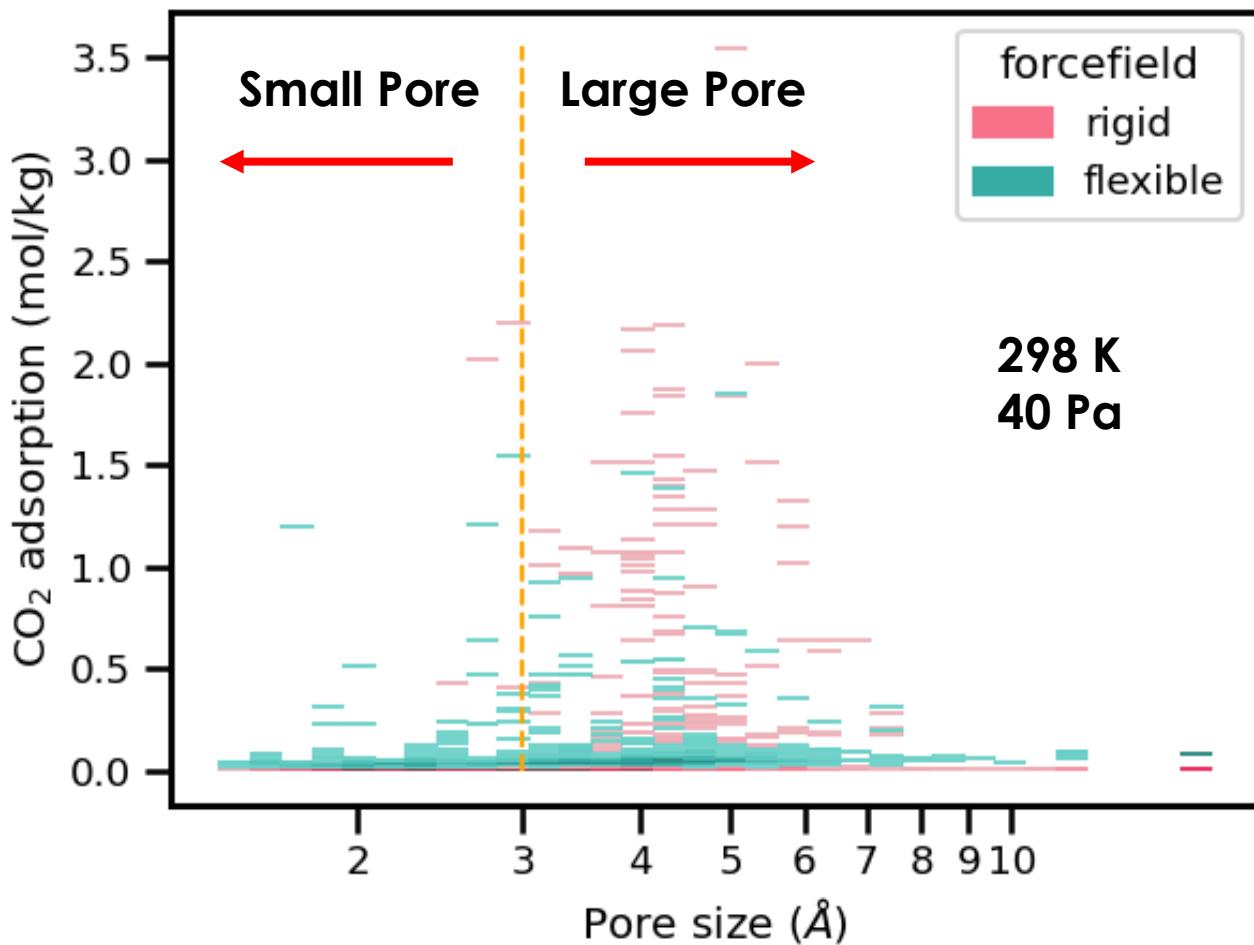
CO<sub>2</sub> adsorption in MAF-2 (BOGXIF) at 298 K



# CO<sub>2</sub> Adsorption at Direct Air Capture Conditions

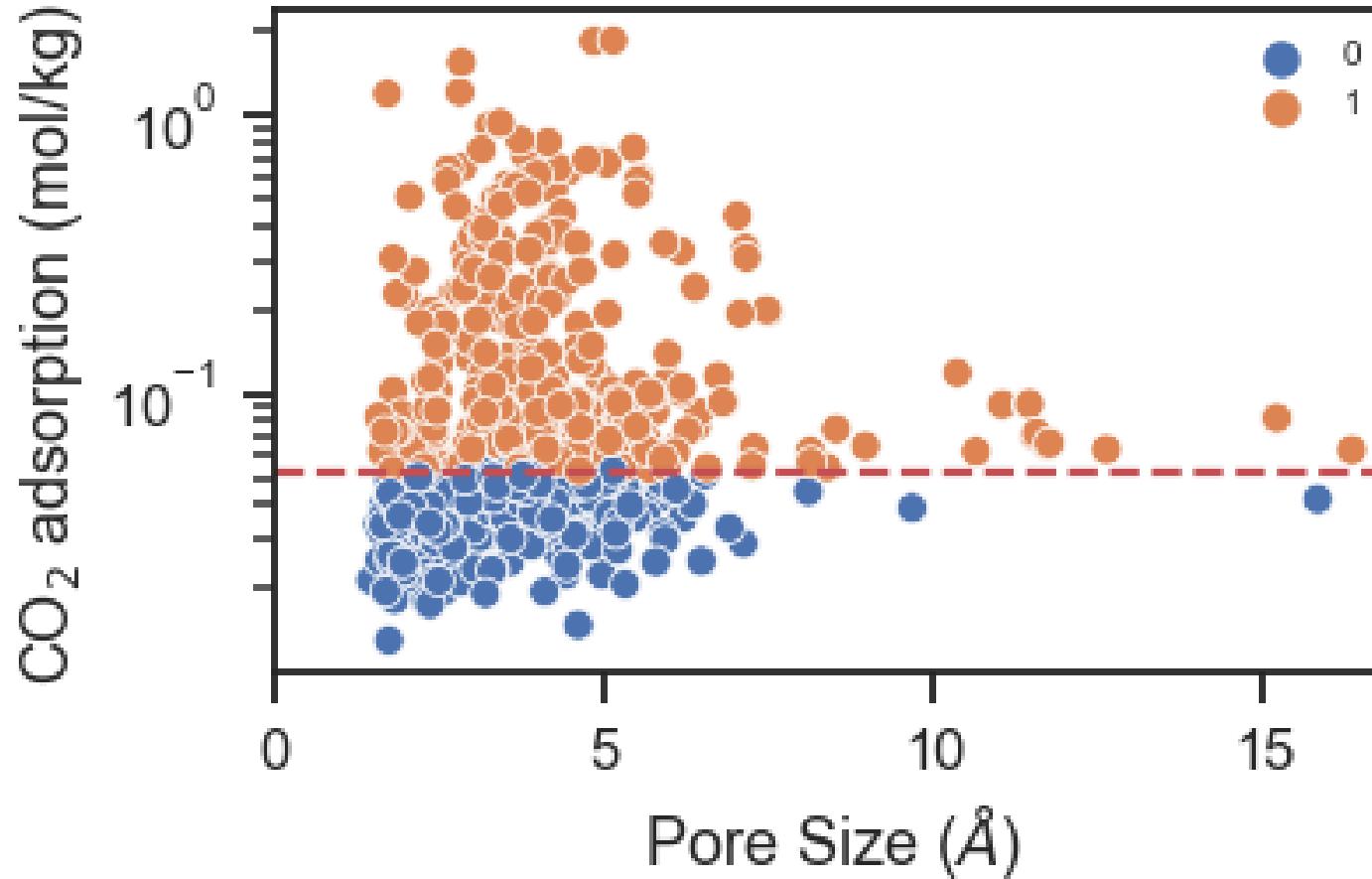


## Comparison Between Rigid and Flexible Force Field Results



- Flexible forcefields yield CO<sub>2</sub> adsorption in MOFs with small pore sizes
- Rigid forcefields yield no CO<sub>2</sub> adsorption for MOFs with pore size less than 3 Å in most of the cases
- For some MOFs with pore size greater than 3 Å, rigid forcefields overestimate the CO<sub>2</sub> adsorption

# Machine Learning model for MOF classification



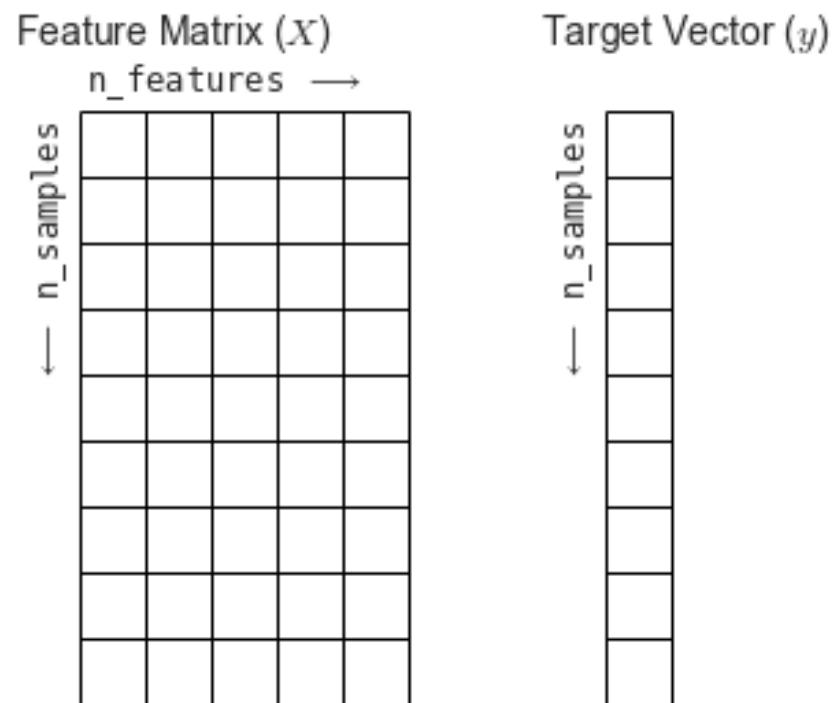
1 – CO<sub>2</sub> ads  $\geq 0.052$  mol/kg  
0 – CO<sub>2</sub> ads  $< 0.052$  mol/kg

# MOF Featurization

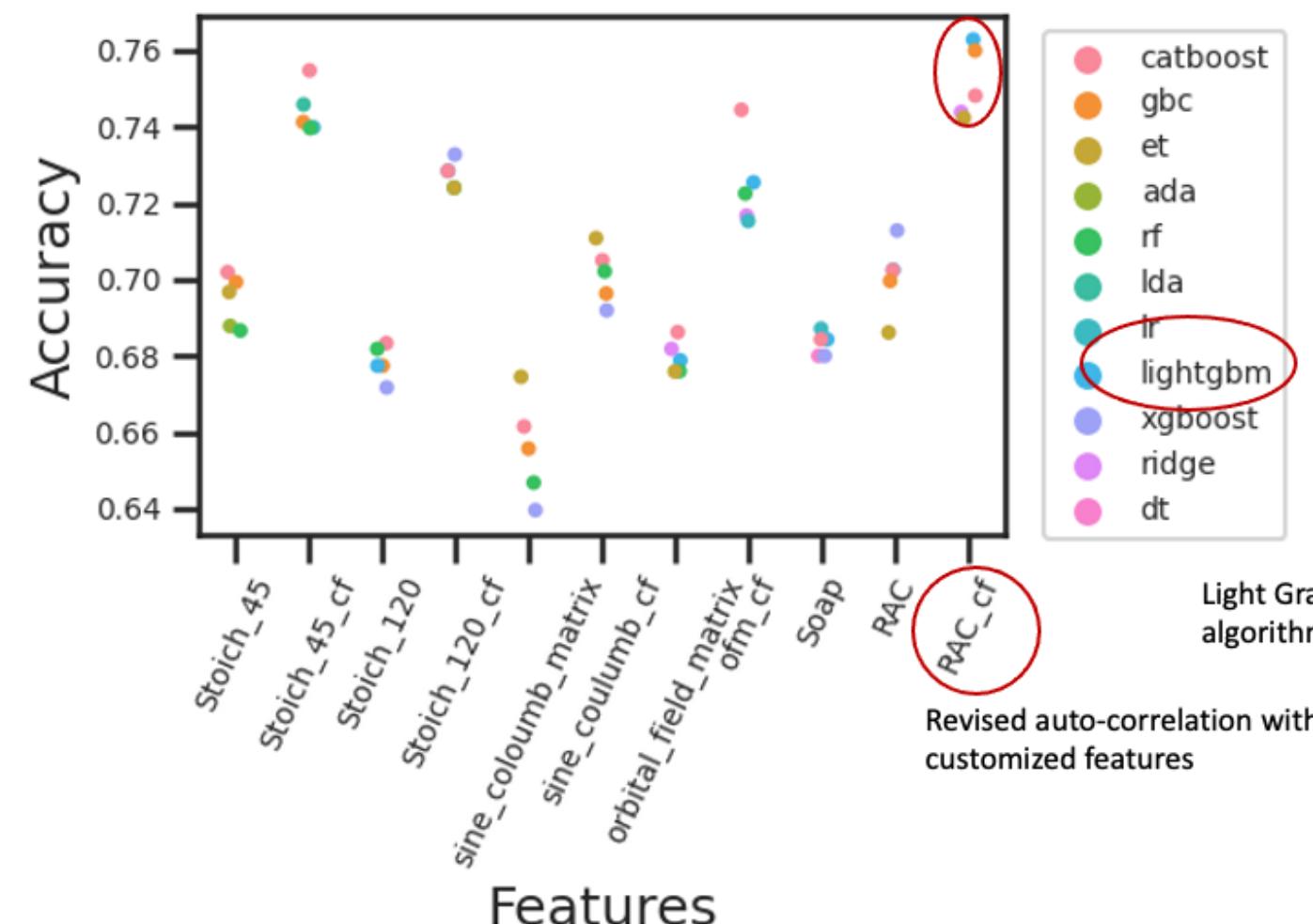


**Feature:** n-dimensional numerical vector that represents each MOF

1. **Stoichiometric-45**
  - 45 statistical attributes of elemental properties
2. **Stoichiometric-120**
  - 103 attributes describing elemental fractions
  - 7 statistical attributes of elemental properties
3. **Sine Coulomb Matrix**
  - pairwise electrostatic interactions between nuclei
4. **Orbital Field Matrix**
  - distribution of valence electrons
  - interaction of valence subshells between atoms
5. **Smooth Overlap of Atomic Positions (SOAP)**
  - similarity between a pair of local atomic environments
6. **Revised Autocorrelation (RAC) values + Custom features**
  - molecular revised autocorrelation (RAC) values, surface area, volume, density, pore-limiting diameter (PLD), charge difference, epsilon



# Results: ML Classifiers validation results

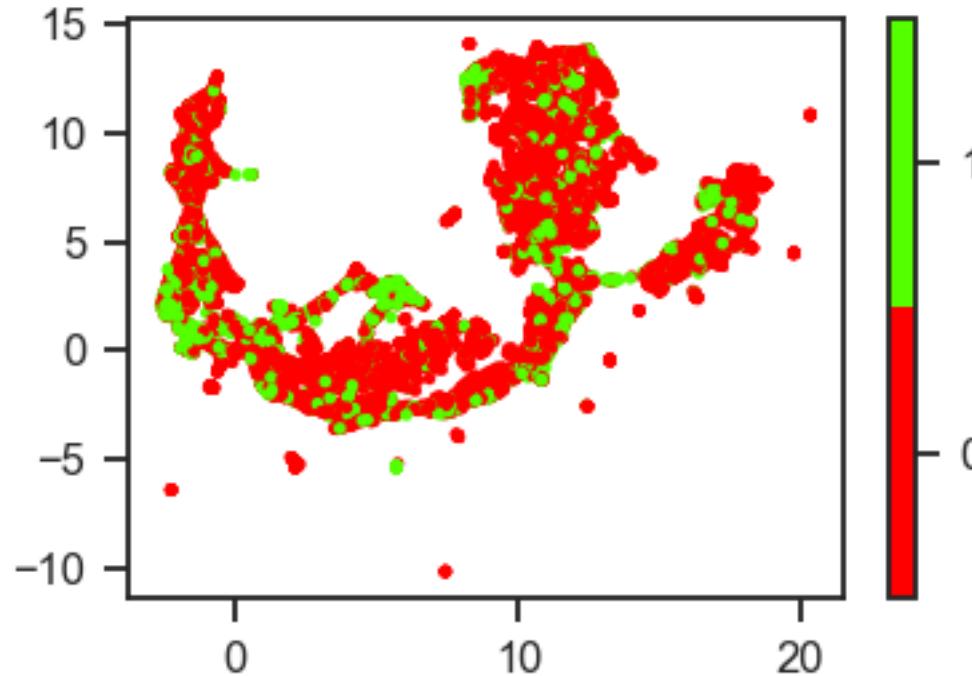


Train size : 80 %  
Test size: 20%  
Results are averaged from 10 cv folds

- 11 different classification ML models
- All the features were customized (denoted by '\_cf' in plot ) with geometric features such as surface area, pore size, electrostatics and dispersion term.
- Light Gradient Boosting algorithm performed the best

$$\text{Accuracy} = \frac{\text{True Positive} + \text{True Negative}}{\text{True Positive} + \text{False Positive} + \text{True Negative} + \text{False Negative}} = 76\%$$

# Results: Predictions on rest of QMOF Database



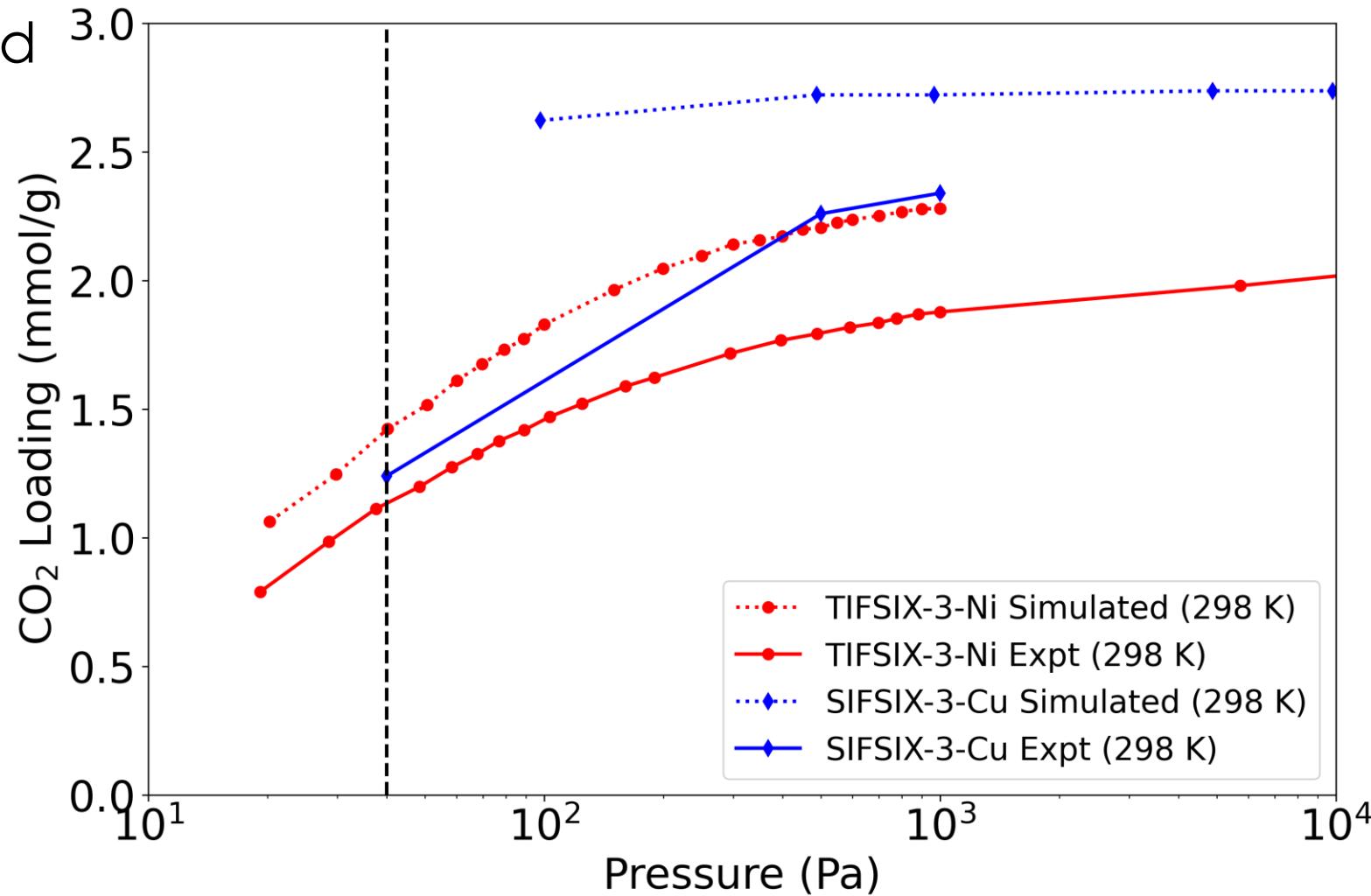
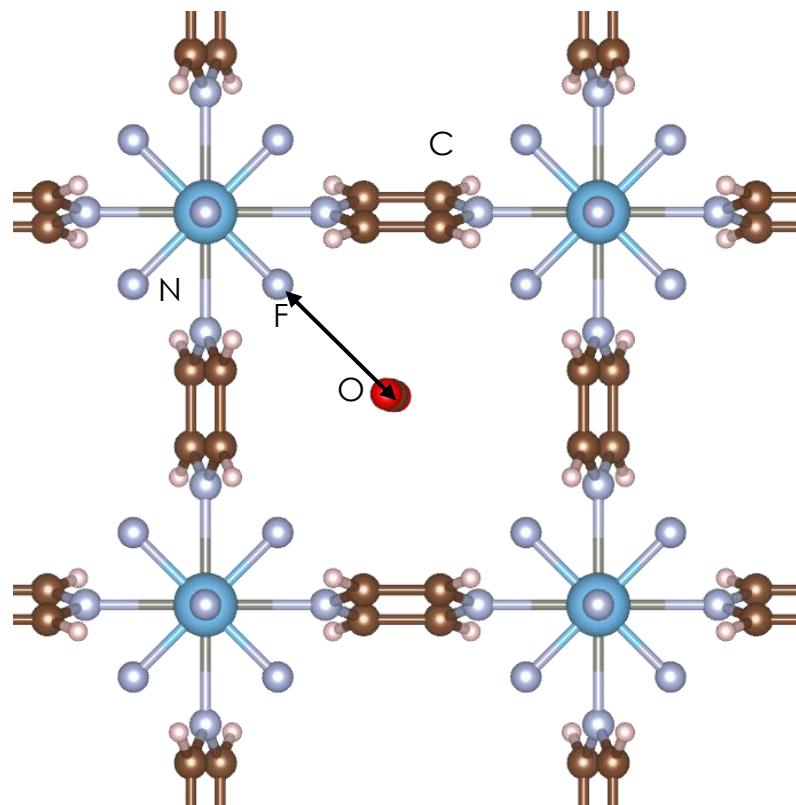
Total MOFs	High adsorbing MOFs (1) (CO <sub>2</sub> adsorption >= 0.05 mol/kg)	Low adsorbing MOFs (0) (CO <sub>2</sub> < 0.05 mol/kg)
10,645	2447 (~23 %)	8198 (~77%)

# Performance of Classical Force Fields: TIFSIX, SIFSIX



## Performance at Low Pressure

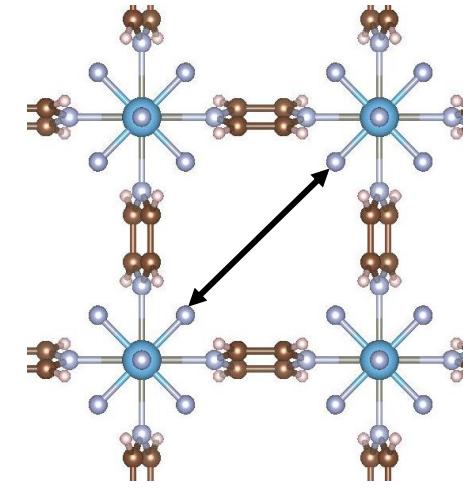
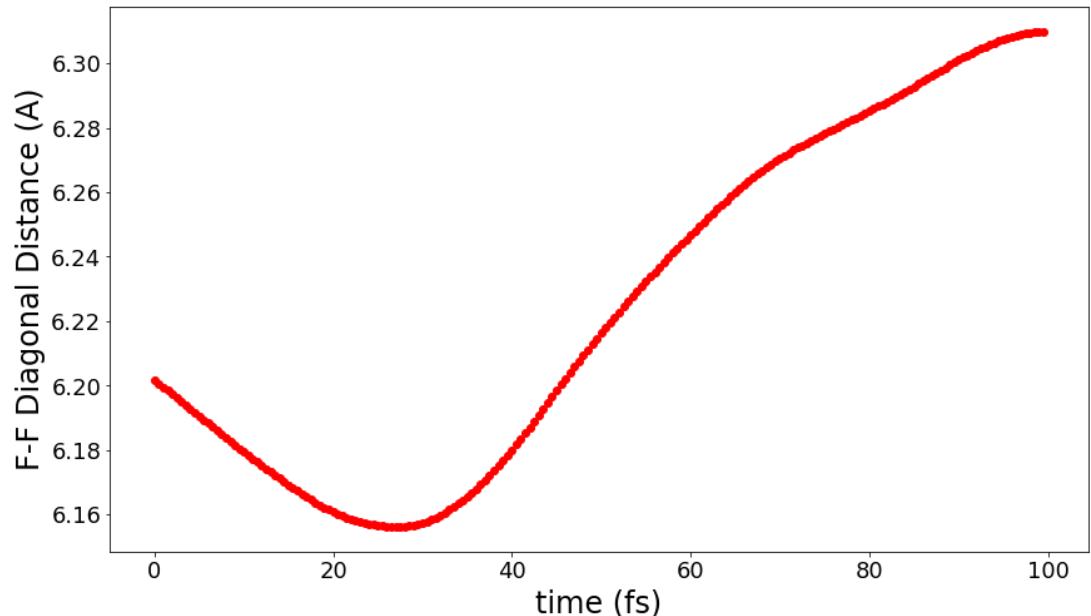
$\text{CO}_2$  adsorption overpredicted using rigid force fields



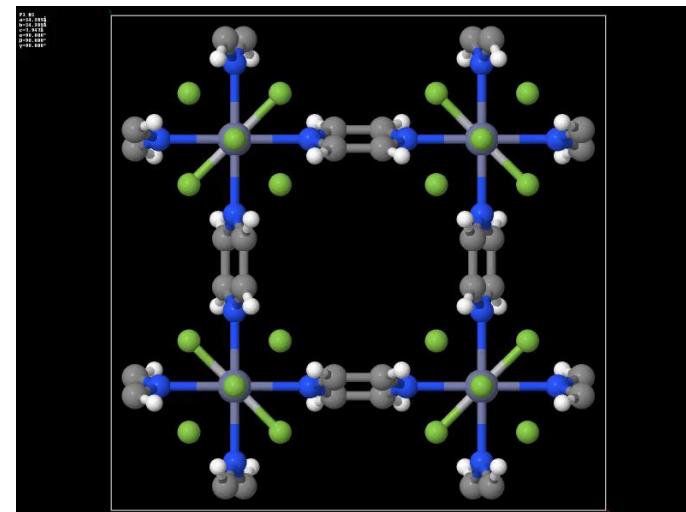
# Importance of Flexibility and Accuracy

Flexible  $\text{SiF}_6$  and  $\text{TiF}_6$  can Cause Changes in Window Size

- Poor performance of rigid force fields
  - Overprediction at low  $P_{\text{CO}_2}$
- DFT molecular dynamics:
  - Significant motion of fluorine atoms
  - Fluorine location affects adsorption strength



F-F diagonal

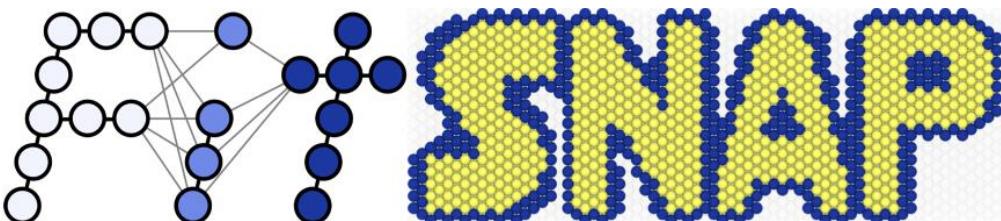


# MLFF Training: Spectral Neighbor Analysis Potential

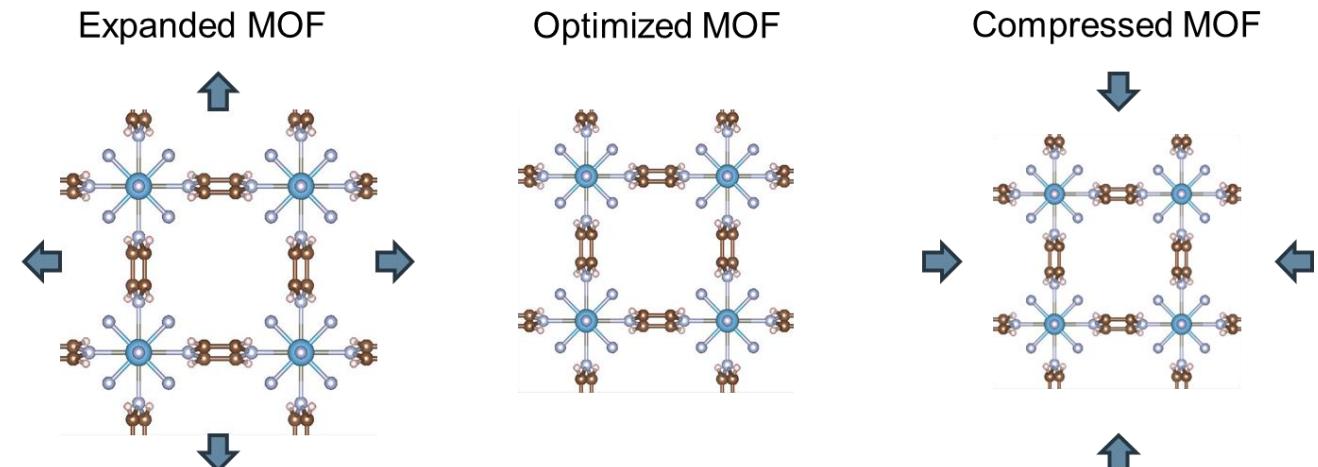


## MLFFs Trained on Accurate DFT Data

- **MLFF Method: SNAP**
  - regression model maps local atomic environments to atomic energies and forces
- **Training Configurations**
  - Favorable (low energy) and unfavorable (high energy)
  - MOF with and without  $\text{CO}_2$
- **Our Approach**
  - Hybrid SNAP/classical potential
  - MLFF handles short range
  - Classical handles long range



### Sample Volume Changes



### Sample $\text{CO}_2$ Adsorption

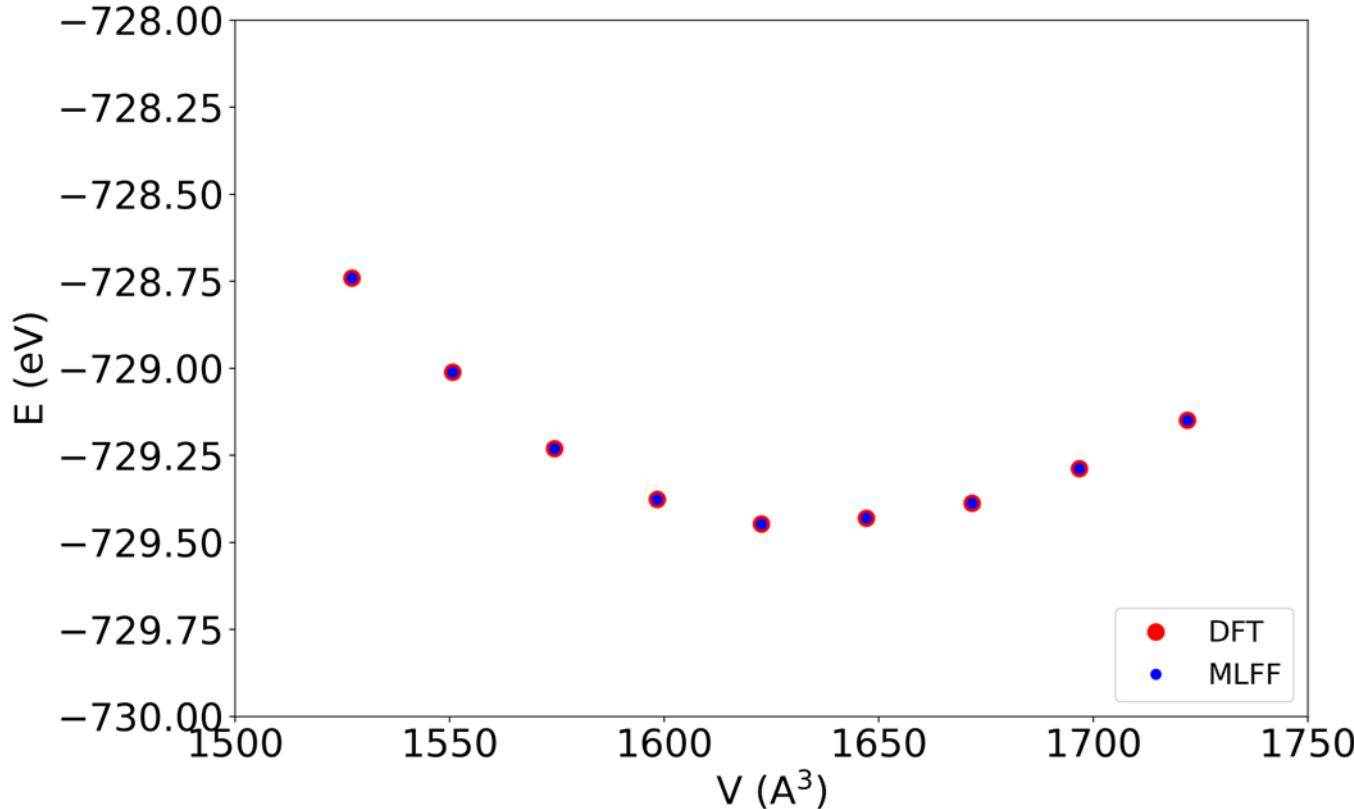


# Performance of MLFF Model (Empty)

## MLFF Describes MOF Structure and Flexibility (TIFSIX\_3\_Zn)

- Prediction of energy vs. volume curve is almost perfect (bottom)
  - Related to bulk modulus
- Compared energies/forces in QM-based dynamics calculations, performance is good
  - 7000 training configurations
  - 1800 testing configurations

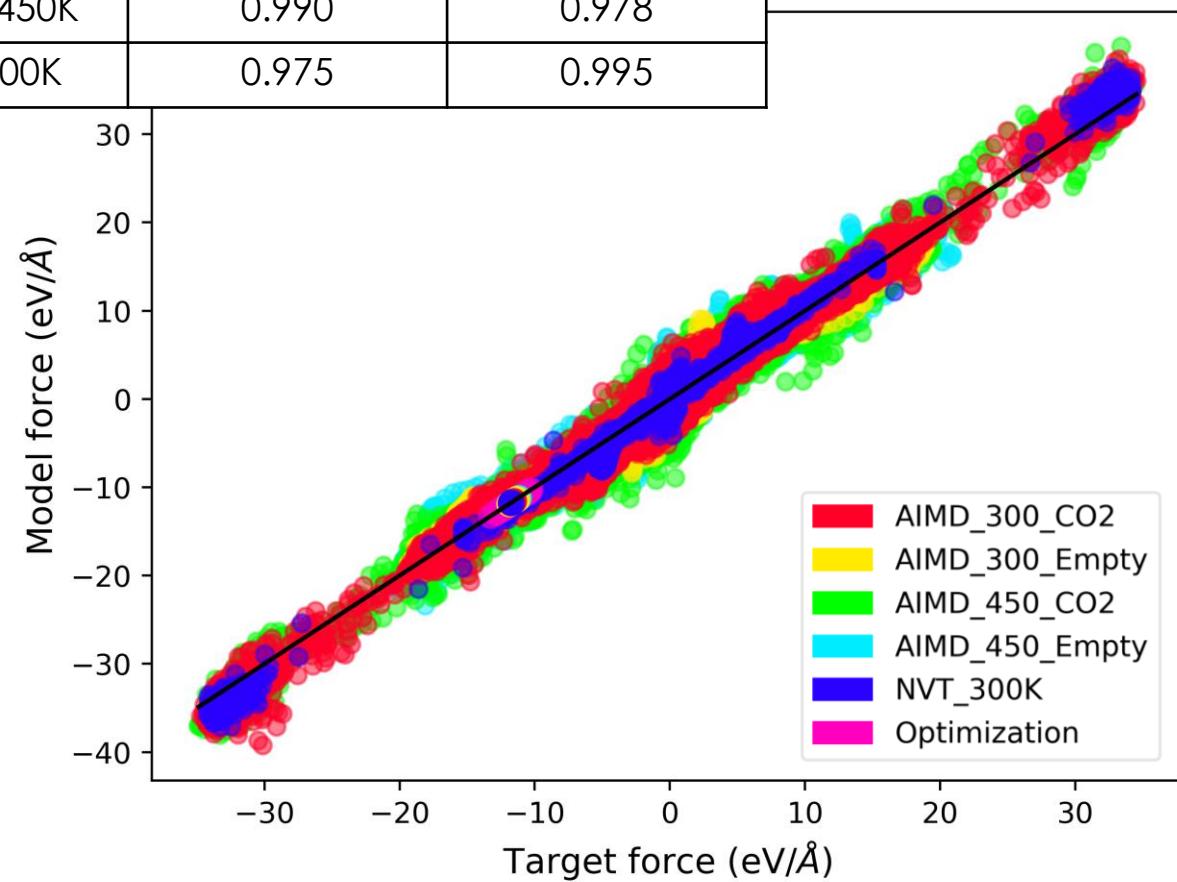
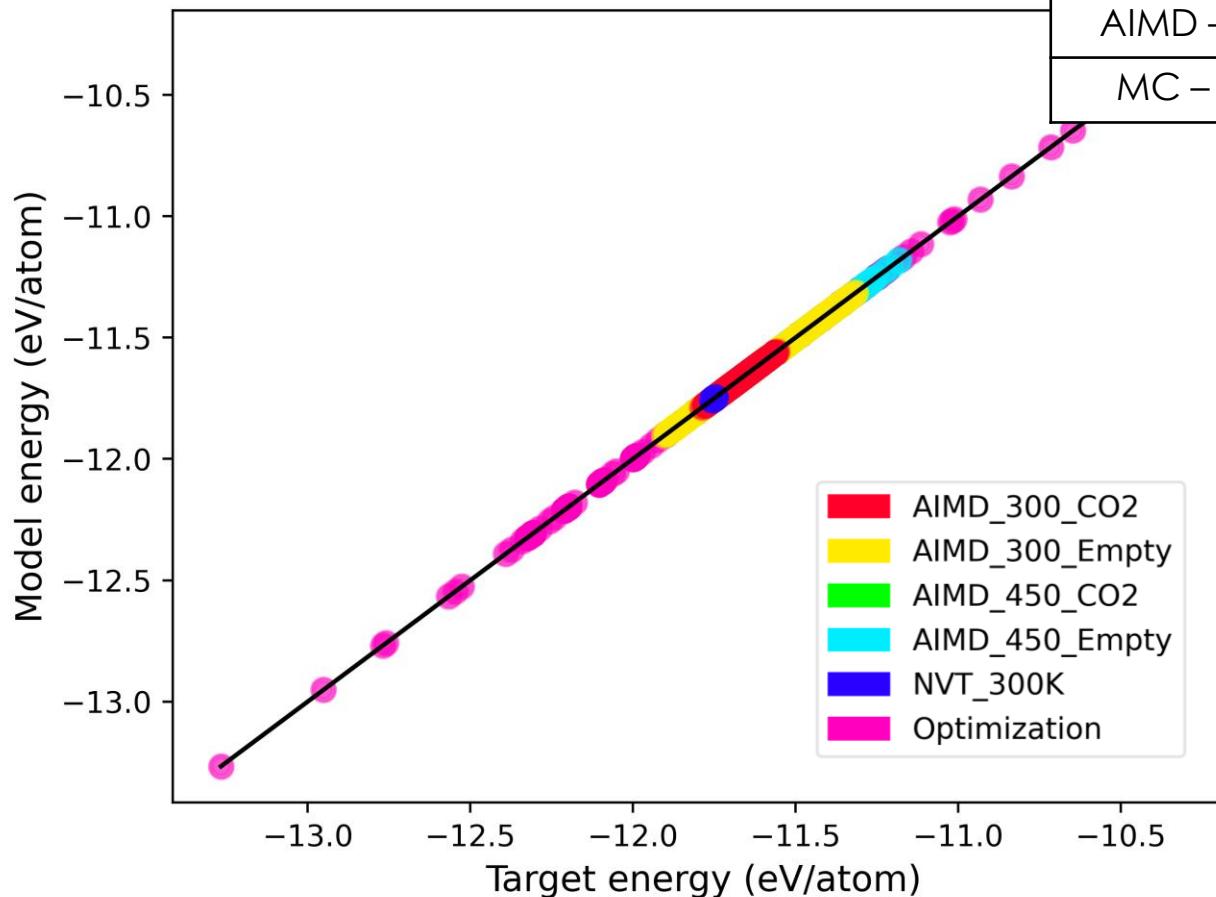
Data Empty MOF (Testing set)	$R^2$ Energies	$R^2$ Forces
Structure Optimization	0.998	0.990
AIMD – 300 K	0.991	0.981
AIMD – 450 K	0.990	0.975



# Performance of MLFF: MOF + CO<sub>2</sub>

## Parity Plots for Energies and Forces

Data MOF + CO <sub>2</sub> (Testing set)	R <sup>2</sup> Energies	R <sup>2</sup> Forces
AIMD - 300K	0.990	0.983
AIMD - 450K	0.990	0.978
MC - 300K	0.975	0.995



# Performance of MLFF Force Field: TIFSIX-3-Zn



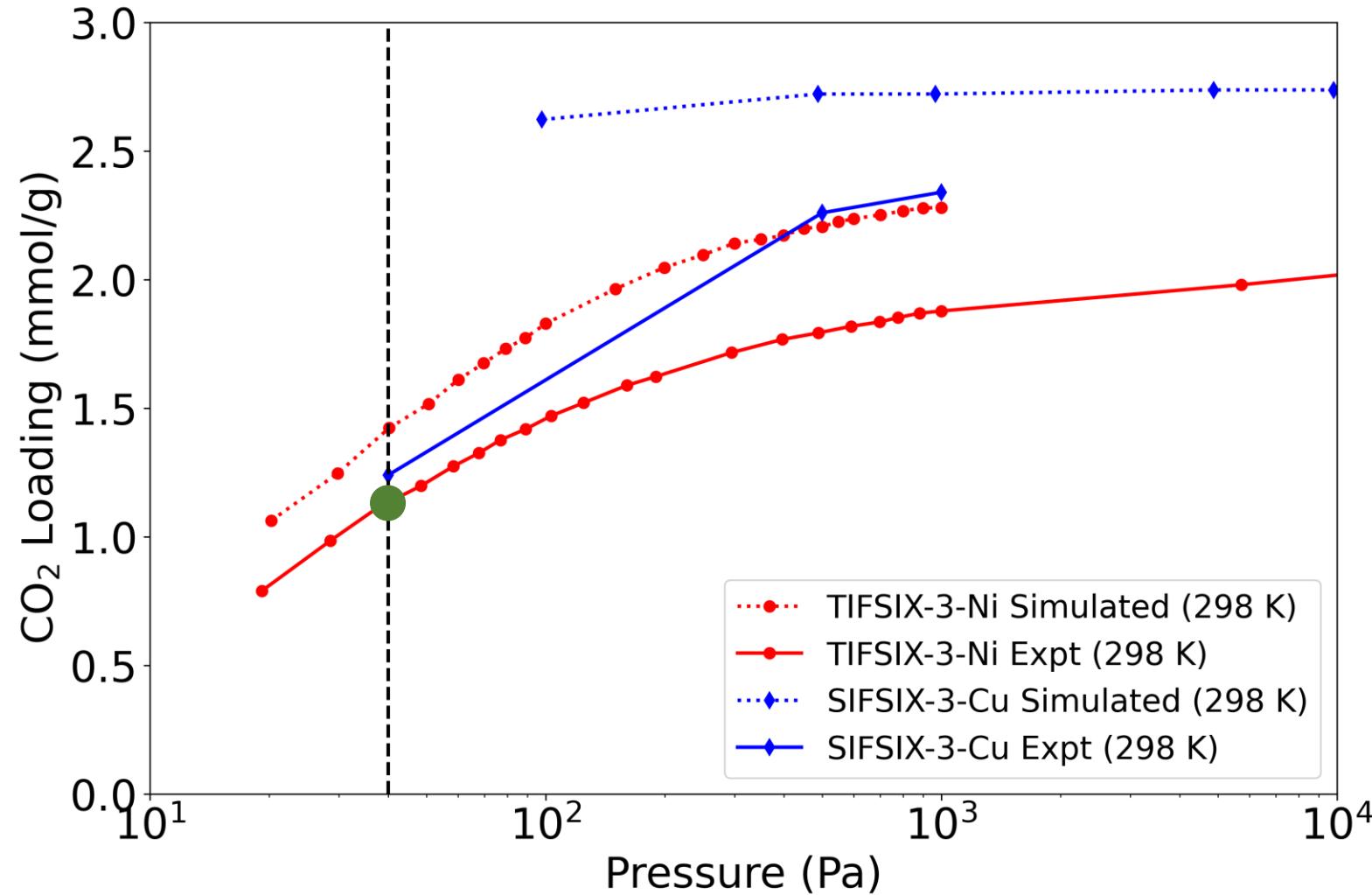
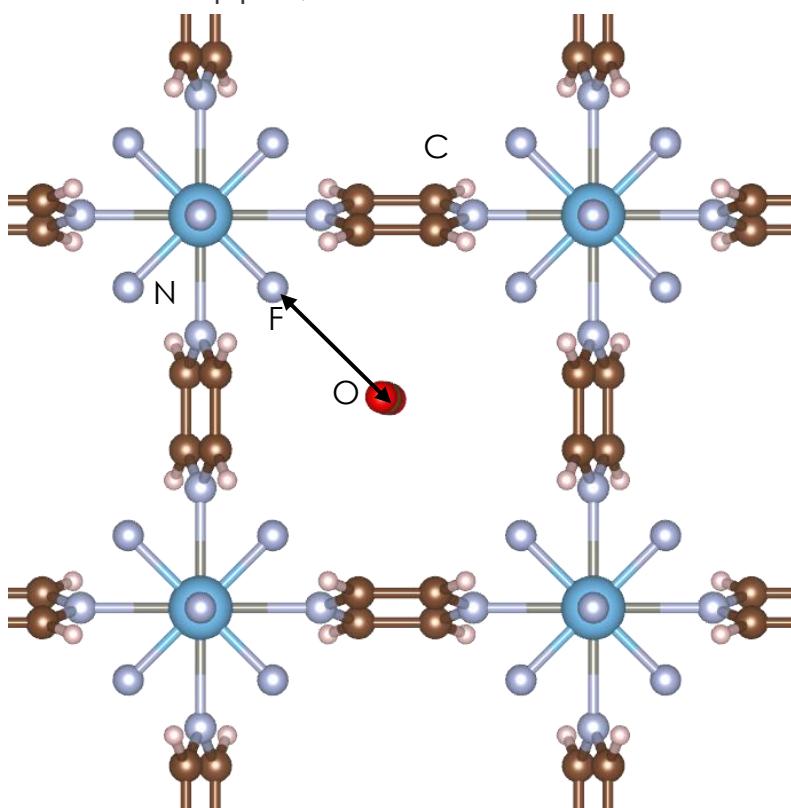
## Performance at Low Pressure

One point on isotherm so far

### MLFF Simulation:

TIFSIX-3-Zn adsorbs 1.08 mmol/g\*

- 400 ppm, 298 K



# Conclusions

- Inclusion of MOFs in polymer membranes can lower the cost of membrane-based CO<sub>2</sub> capture by up to 1/3
- Rigid FF can lead to large overprediction of CO<sub>2</sub> isotherm at low pressure
- MLFF flexible potential accurately describes energies and forces in a MOF loaded with CO<sub>2</sub>
- Training on the error of the classical force field improves the MLFF model fit
- TIFSIX-3-Zn  $\Delta E_{\text{ads, CO}_2}$ , MLFF = -52.9 kJ/mol, DFT -52.1 kJ/mol
- MLFF Prediction: TIFSIX-3-Zn adsorbs 1.08 mmol/g\* at 400 ppm, 298 K

