

First-principles DFT study of the electronic and optical properties of lanthanum and magnesium-doped strontium titanates for advanced gas sensing applications

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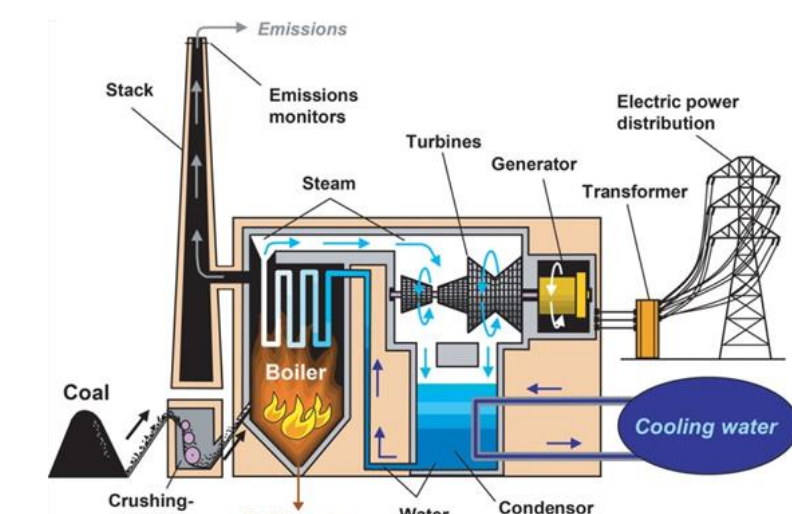
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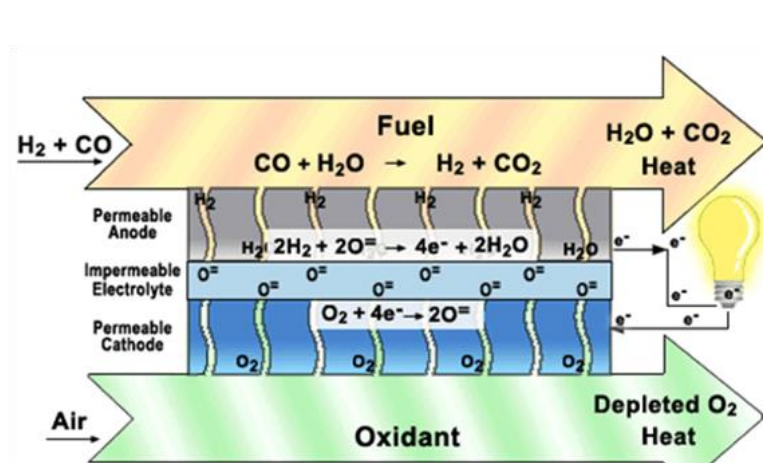
Introduction

- For advanced real-time monitoring and control of gas species in combustion environments, development of efficient sensing platforms and new sensor materials able to work under harsh environments are required
- Semiconducting optical-based sensor platform based on strontium titanate (SrTiO₃, STO) shows promise; cubic ABO₃ structure has tunable electronic, optical properties dependent on A-site, B-site doping, oxygen vacancies

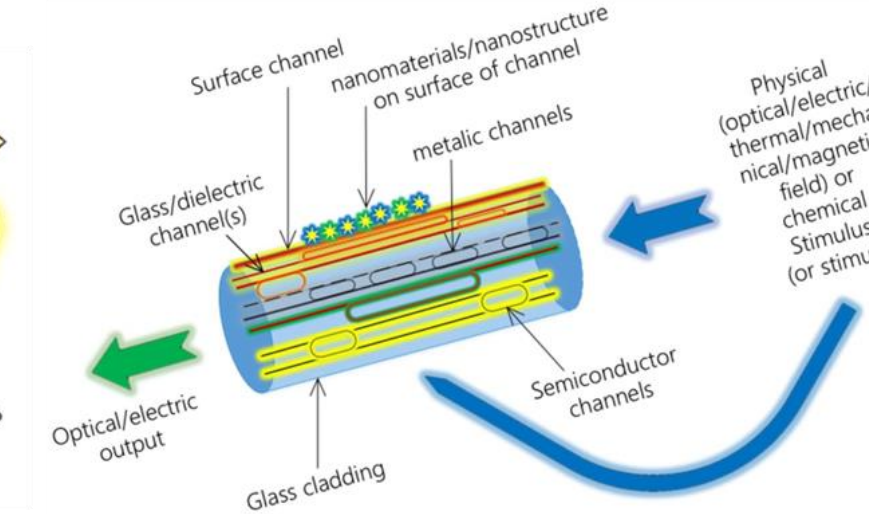
Fossil-/Nuclear-based power plants



Solid oxide fuel cells



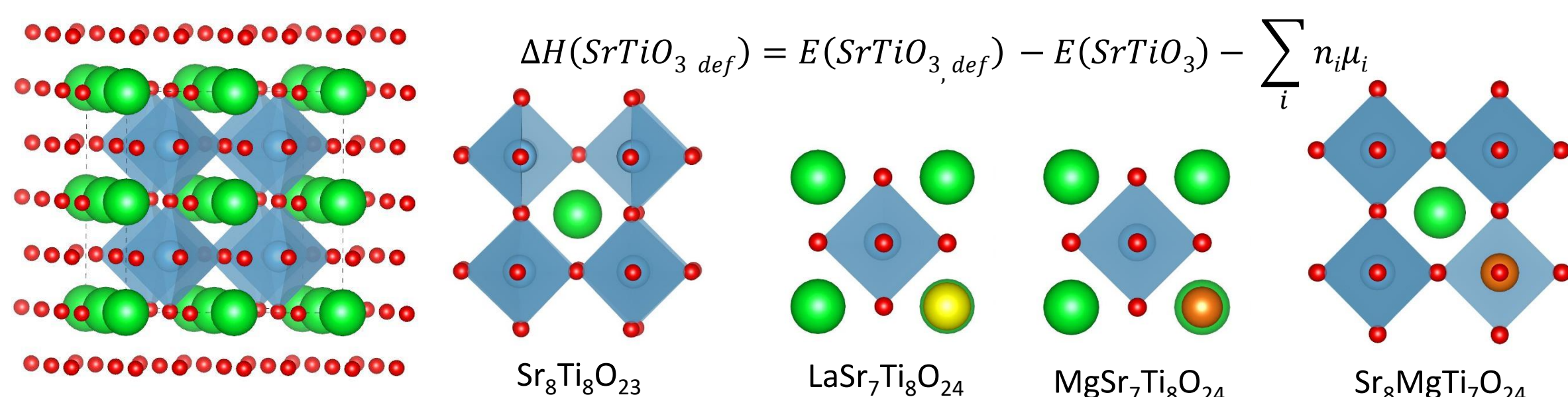
Optical fiber sensor



- Hydrogen (H) and oxygen (O) impurities are ubiquitous to metal oxide perovskites; both H and O are thought to contribute to room temperature, long lasting photoconductivity in STO, indicating the potential use of STO in H, O gas sensing applications
- Understanding how lanthanum (La) and magnesium (Mg) doping of STO and oxygen vacancy defects affect electronic and optical properties of SrTiO₃ is required to tailor STO-based materials for development of sensitive, selective gas sensors
- First-principles modeling assesses the tunability of STO material properties via incorporation of La, Mg impurities and emergence of oxygen vacancies

Methods

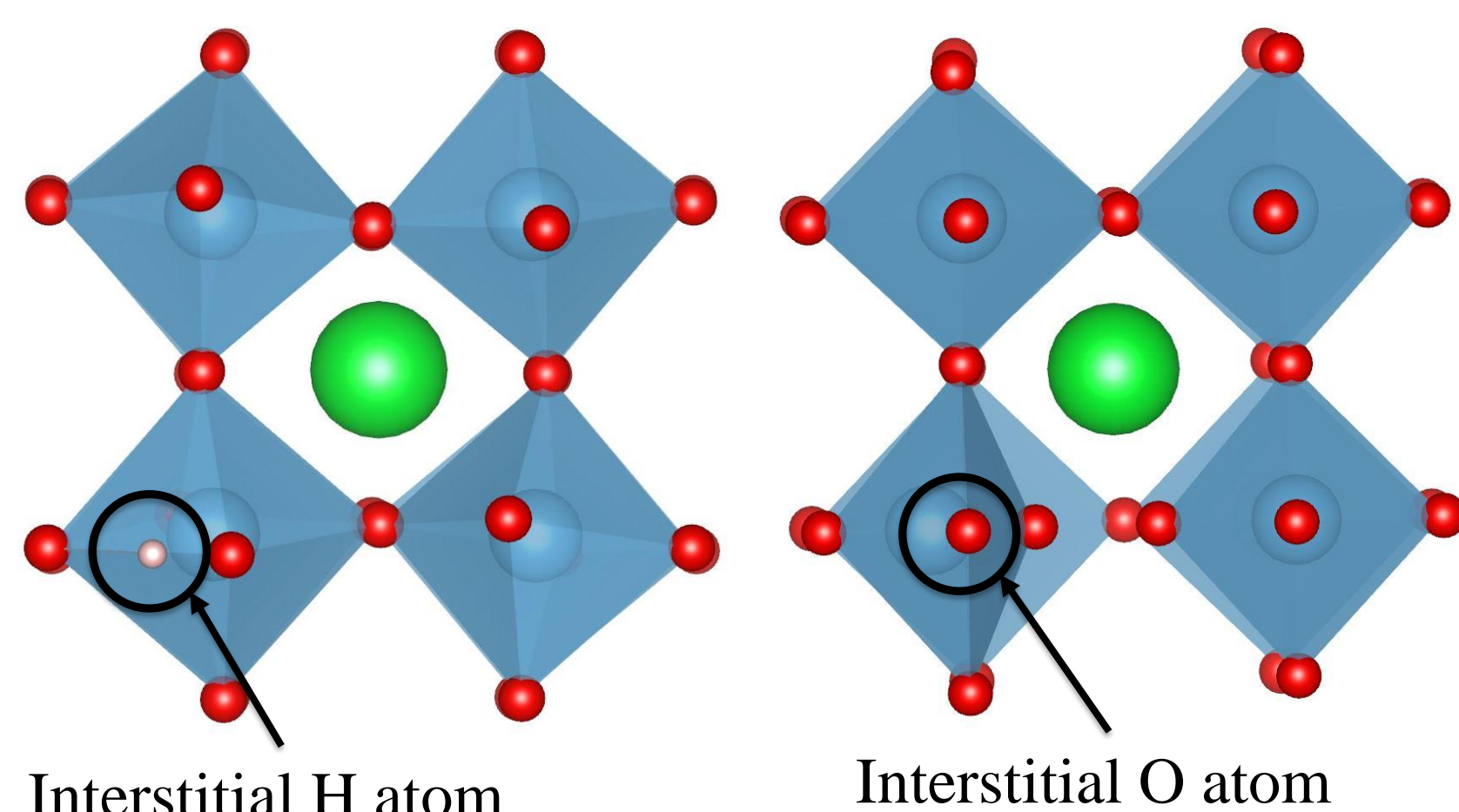
- Density functional theory (DFT): PAW-PBE(+U) XC in GGA
 - Energies of formation of point defects (La and Mg dopants, O vacancies)



- Optical properties calculated from frequency-dependent dielectric function

$$\epsilon(\omega) = \epsilon_1(\omega) + i\epsilon_2(\omega) = (n + ik)^2 = \frac{4\pi i}{\omega} \sigma(\omega)$$

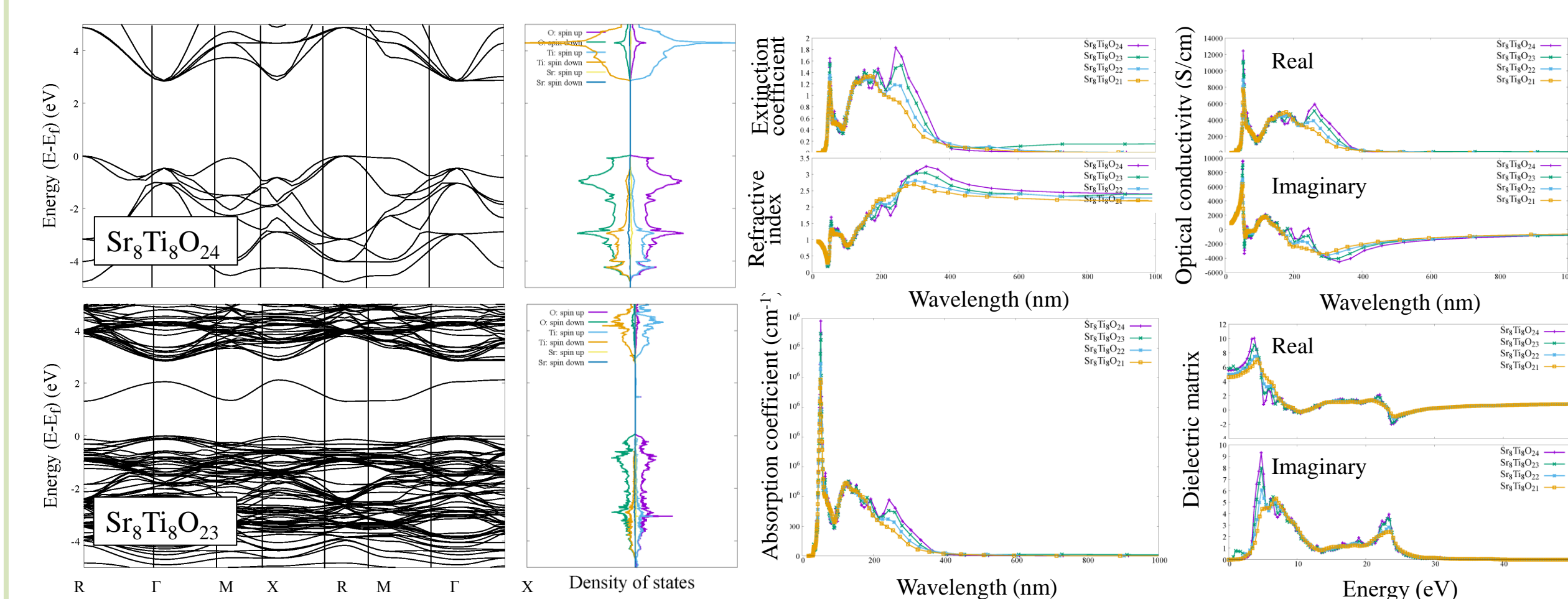
- Incorporation of H and O interstitial atoms



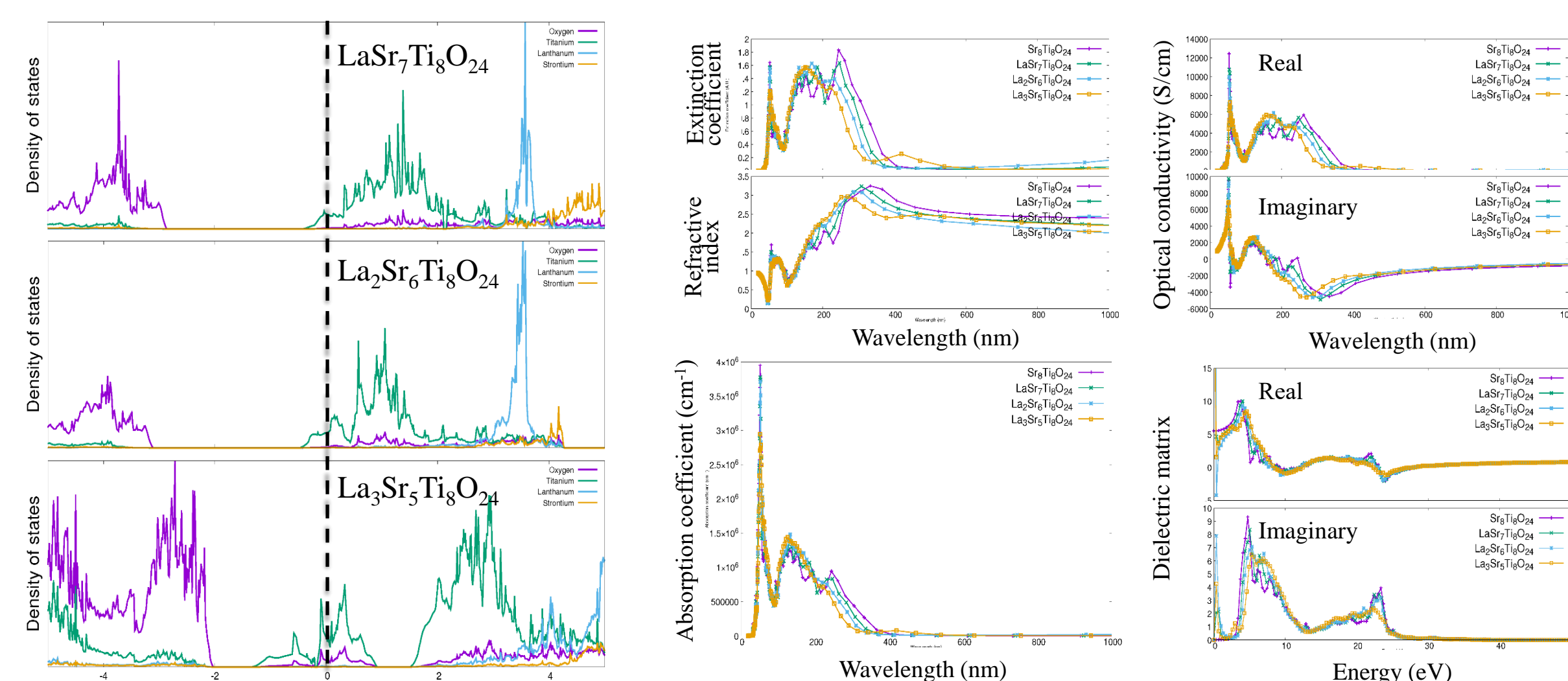
- Determine local energy minima of adsorbate H, O atom binding sites
- Elucidate electronic, optical properties of STO with interstitial H, O atoms
- Nudged elastic band calculations to determine relevant diffusion barriers

Perovskite oxide: La- and Mg-doped SrTiO₃

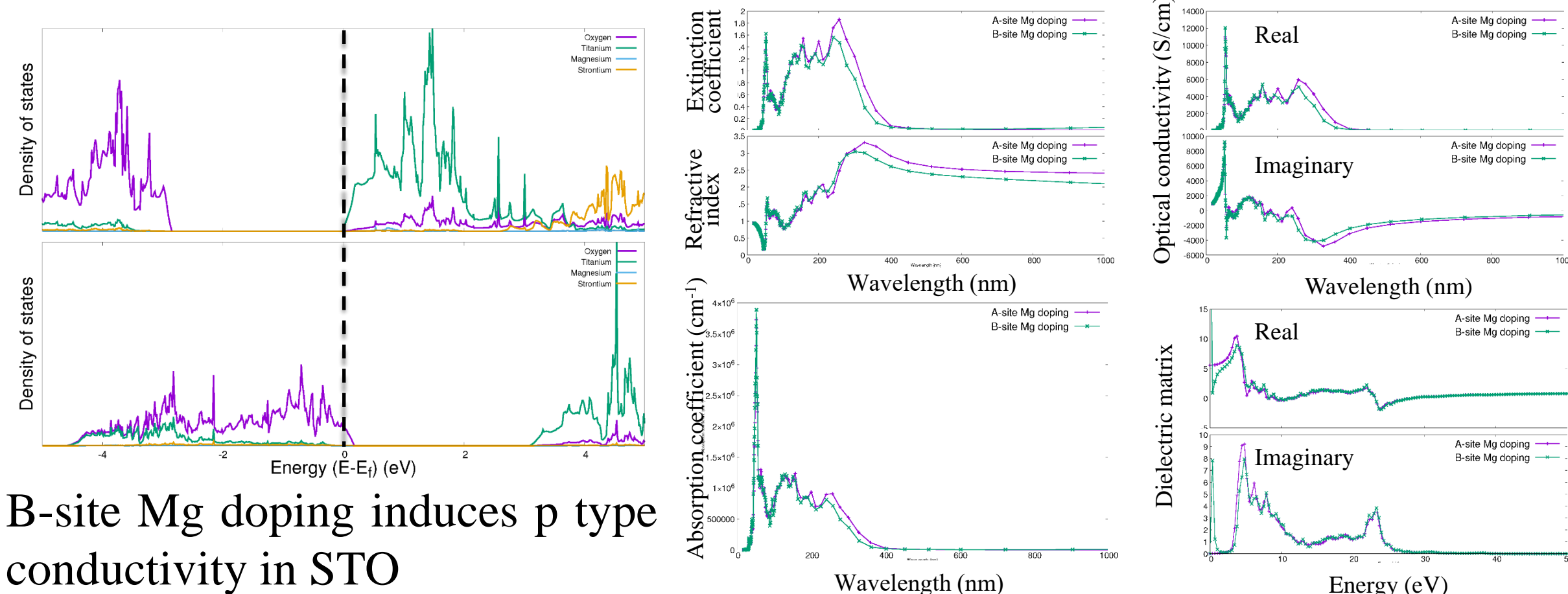
- Electronic structure of SrTiO₃ shows a bandgap of 2.98 eV; optical calculations indicate SrTiO₃ has absorption peak in the IR spectrum at 55 nm
- Incorporation of O vacancy creates an occupied defect state within the bandgap. Its location is sensitive to Hubbard parameter U; contributions to defect state are shown to be due to O 2p, Ti 3d hybridization
- Incorporation of O vacancies to SrTiO₃ due not cause peaks in the absorption coefficient within the range of visible light; increase in O vacancies causes lower-wavelength peak in optical conductivity below 200 nm



- La-doped STO shows n type behavior for all doping levels of La studied ranging from 12.5 to 37.5 at. %; La doping impacts free carrier concentration
- La doping causes shift in dielectric matrix to higher photon energy above 4 eV
- La doping causes loss in optical transparency at lower range of visible light



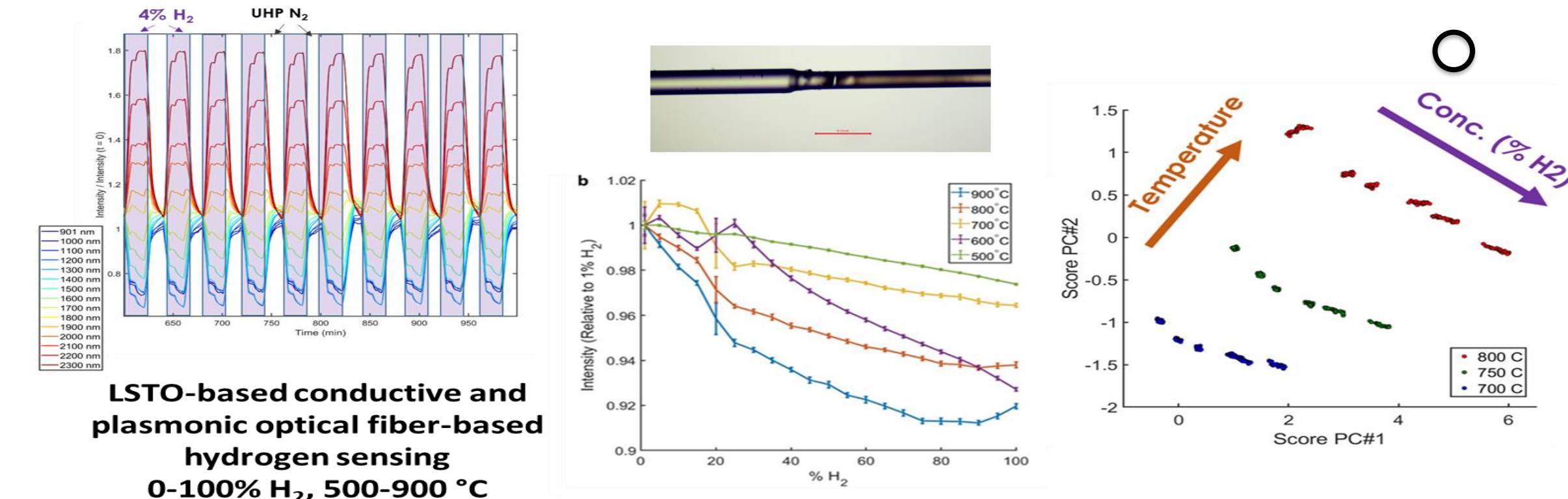
- Mg-doped STO exhibits varied electronic, optical properties dependent on site of Mg substitution; neither show absorption in visible light range
- B-site doped STO has occurrence of Drude peak in imaginary component of dielectric matrix, confirming the presence of free carriers



B-site Mg doping induces p type conductivity in STO

Doped Perovskite Sensing Layers on Optical Fiber

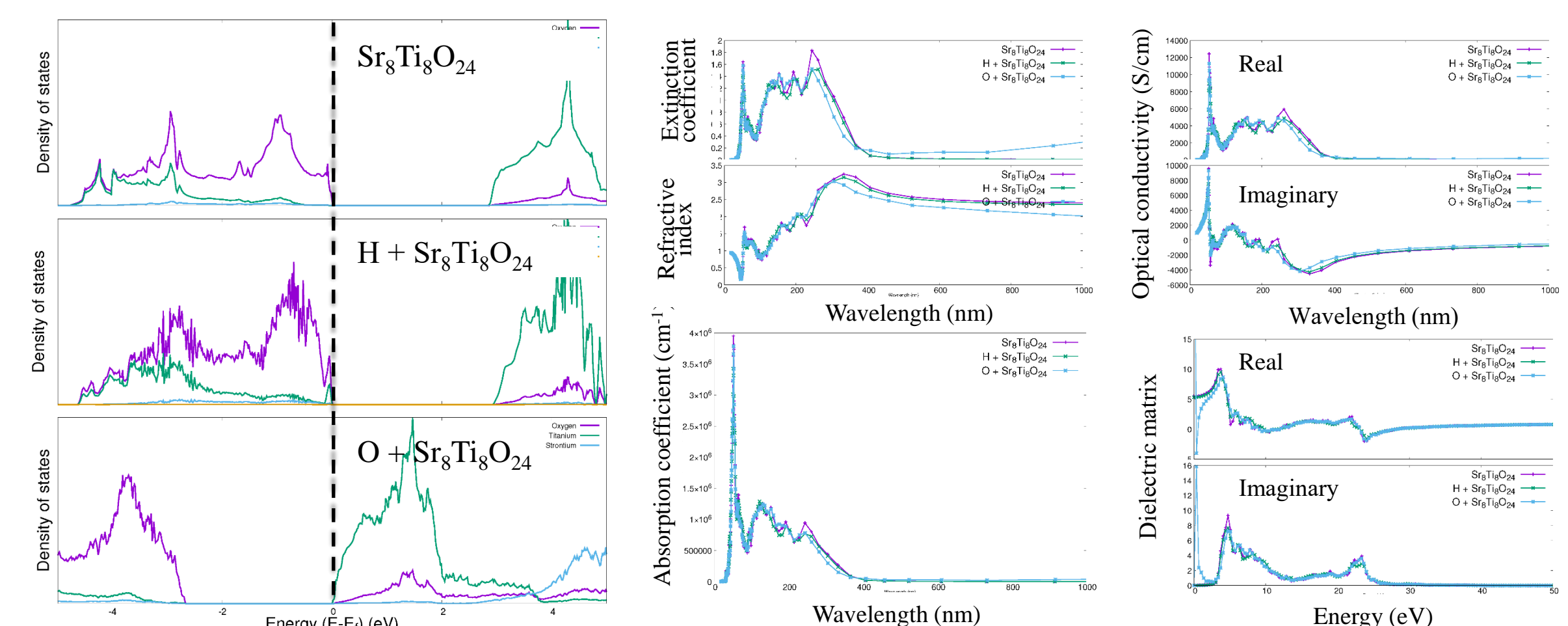
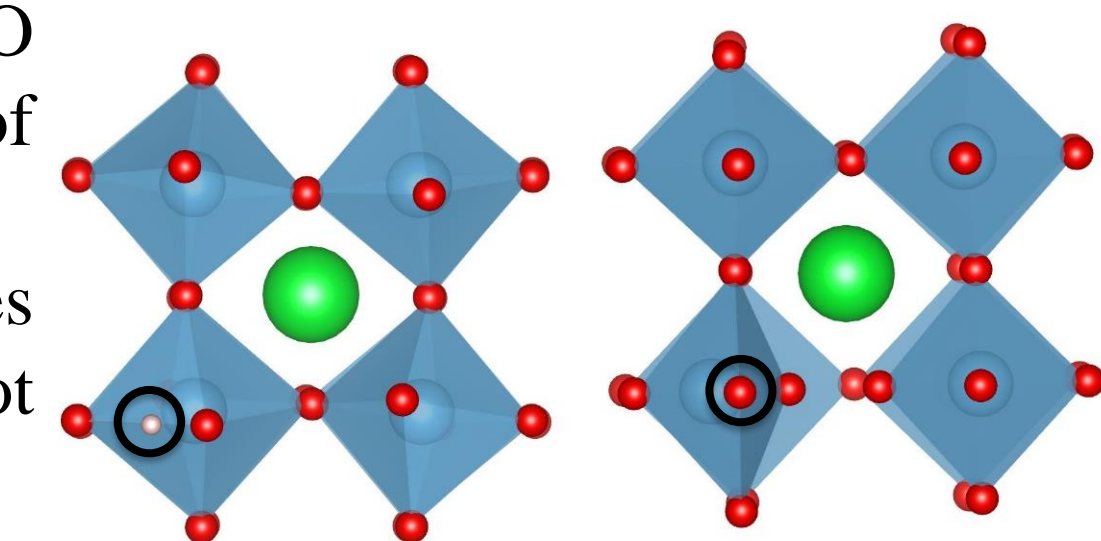
- As a functional sensing layer on evanescent-field based optical fiber sensors, A- or B-site doped can operate as a versatile, high-T sensor for reducing or oxidizing gas streams.
- La-doped SrTiO₃ acts like an n-type doped semiconductor under reducing conditions – demonstrating an effective high-T sensing material for H₂.



- Other SrTiO₃-based systems such as SrFe_xTi_{1-x}O₃ (SFTO) and Mg-doped SrTiO₃ can act as p-type doped semiconductors under oxidizing conditions and show promise for high-T stable oxygen sensing.

Interstitial hydrogen and oxygen impurities

- Interstitial H atom preferably binds to O atom in STO leading to breaking of octahedral symmetry
- Interstitial O atom bonds to apical O causes distortion of Ti octahedral but does not break symmetry of crystal
- Incorporation of H, O interstitials alters SrTiO₃ electronic, optical properties as both can act as electron donors to system
- H introduces defect state at VBM below Fermi level without significant change to bandgap, absorption, or dielectric matrix
- O induces n-type conductivity, evident in DOS and Drude peak in imaginary component of dielectric matrix; causes peak shift in optical conductivity to lower wavelength (higher photon energy)



Publications

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