

# In-situ TEM laser heating for manipulation of cooling rates and observation of precipitate dissolution kinetics



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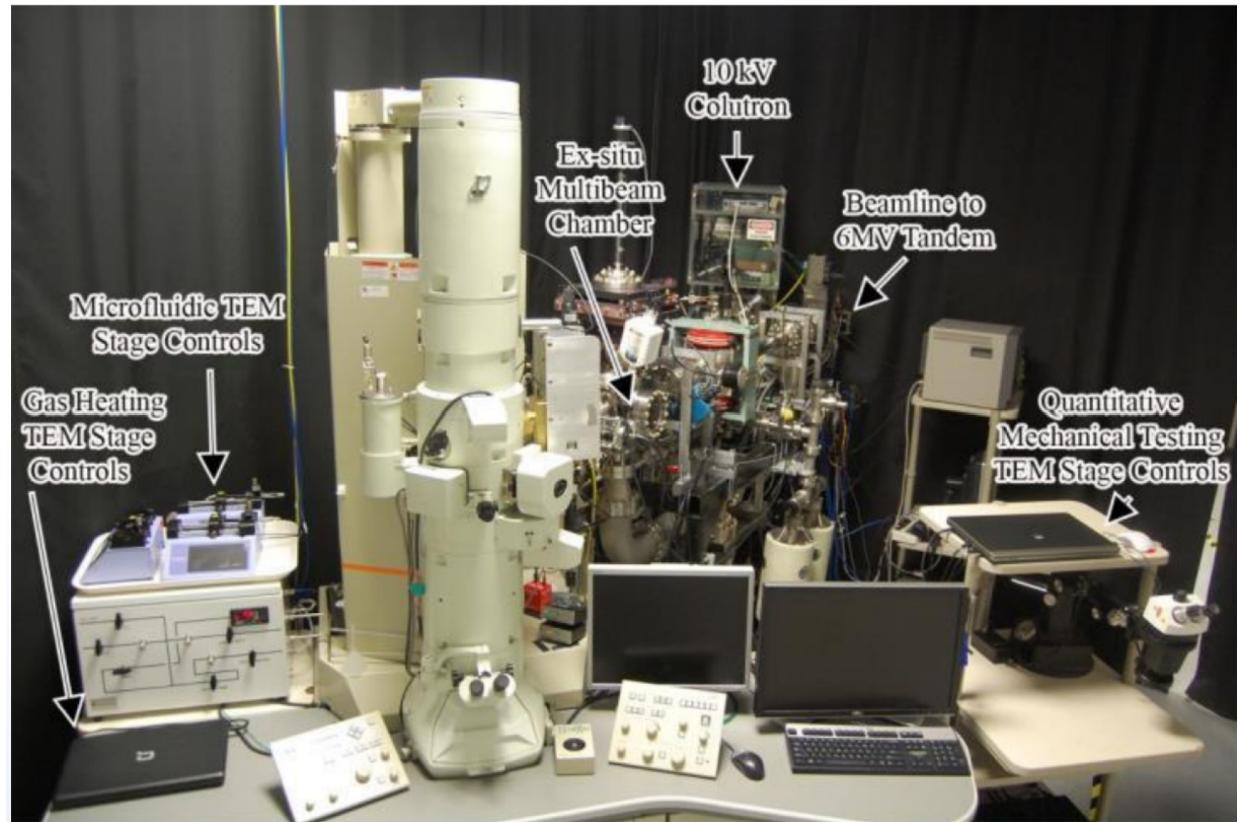
Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.



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# In-situ TEM: Capabilities

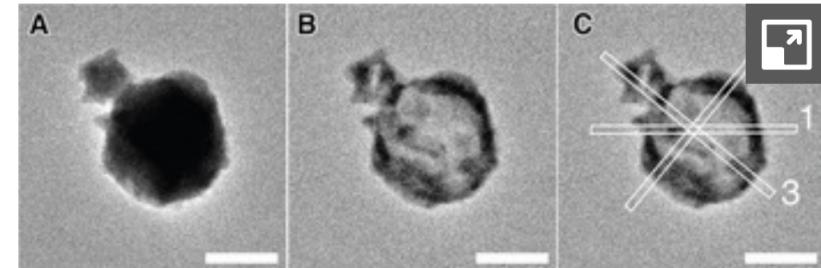
- The In-situ Ion Irradiation TEM (I3TEM) at the Ion Beam Laboratory at Sandia National Labs
- Variety of holders
  - △ Thermal: heating, cryo
  - △ Mechanical: straining, fatigue
  - △ Electro/chemical: liquid/gas cell with biasing
- Recent WAVIKS installation
- IR laser 1064nm wavelength, 20W



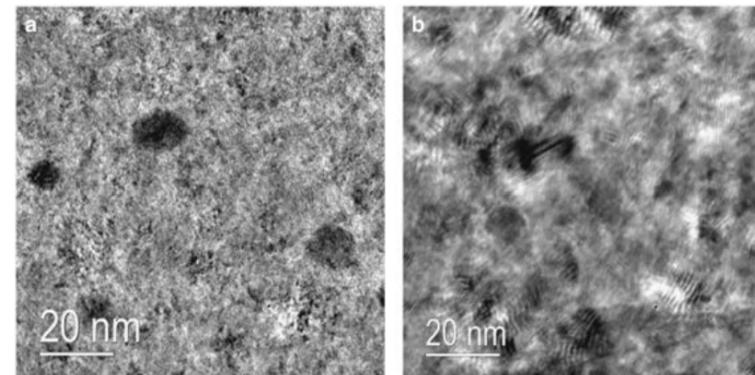
# Current heating work with lasers

- Observe thermally activated microstructural processes
- Crystallization of amorphous materials
- Synthesis of nanowires
- Nanoparticle synthesis through laser ablation
- Hollow zinc oxide nanoparticle generation using infrared heating

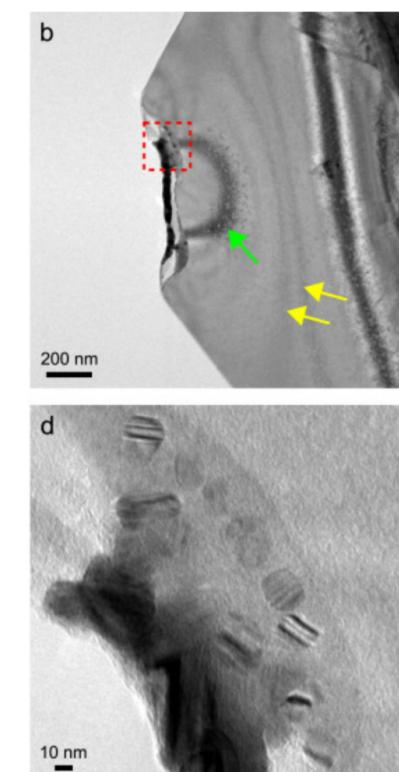
Taheri, Mitra L., et al. "Laser-based *in situ* techniques: Novel methods for generating extreme conditions in TEM samples." *Microscopy research and technique* 72.3 (2009): 122-130.



Mehraeen, Shareghe, et al. *Microscopy and Microanalysis* 19.2 (2013)

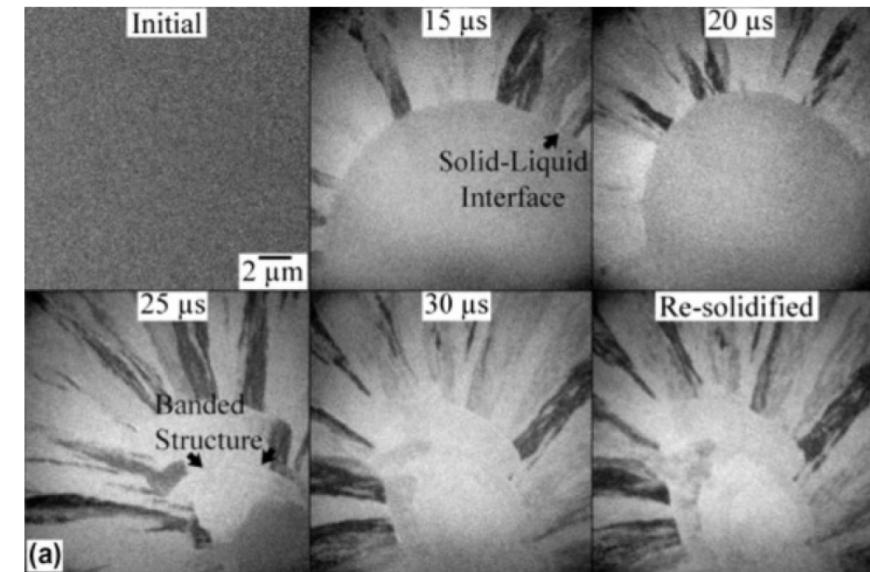


Wu, Yueying, et al. *Microscopy and Microanalysis* 24.6 (2018): 647-656.

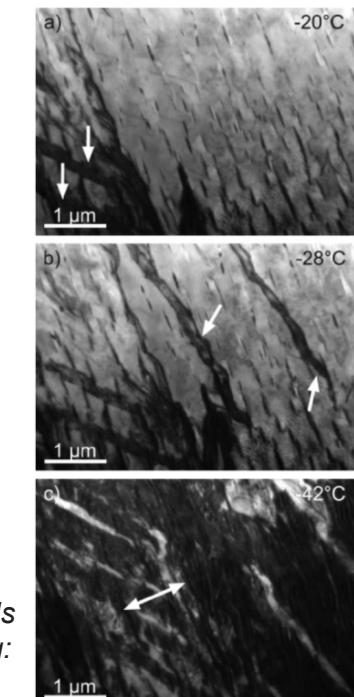


# Observations of cooling

- Rapid solidification after laser processing
- Martensitic transformations
  - Shape memory alloys
- Phase transformation at various cooling rates



McKeown, Joseph T., et al. *Acta Materialia* 65 (2014): 56-68.



**In-situ TEM lasers can be used to observe microstructural changes in real time during ultrafast cooling**

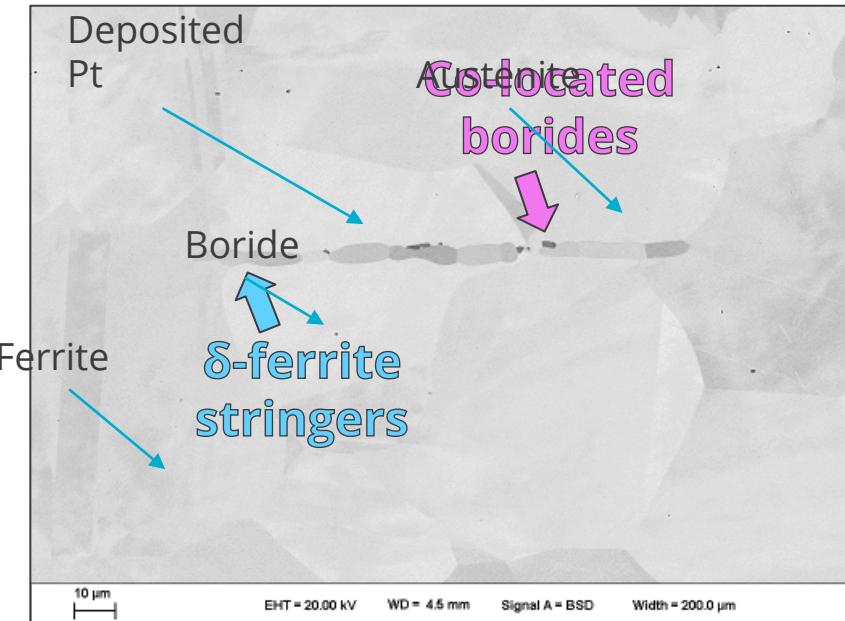
Dlouhy, Antonin, Jafar Khalil-Allafi, and Gunther Eggeler. *Philosophical Magazine* 83.3 (2003): 339-363.

Kröger, A., et al. *Materials Science and Engineering: A* 481 (2008): 452-456.

# Methods and Materials

- FIB lift-out of 304L VAR stainless steel
- Known to contain boride (CR2B) precipitates at Austenite/Ferrite phase boundaries
- After heat treatment, GB borides cause liquation cracking in welded SS
- Increased laser power incrementally to minimize thermal drift and allow equilibration
- Waited for features indicating melting of Pt
- Peak laser power differed between holder temperatures

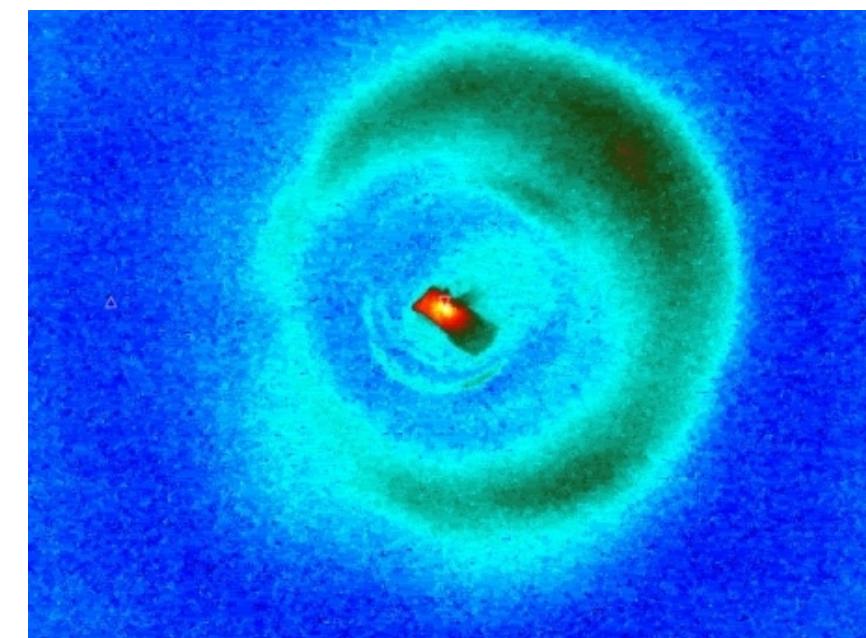
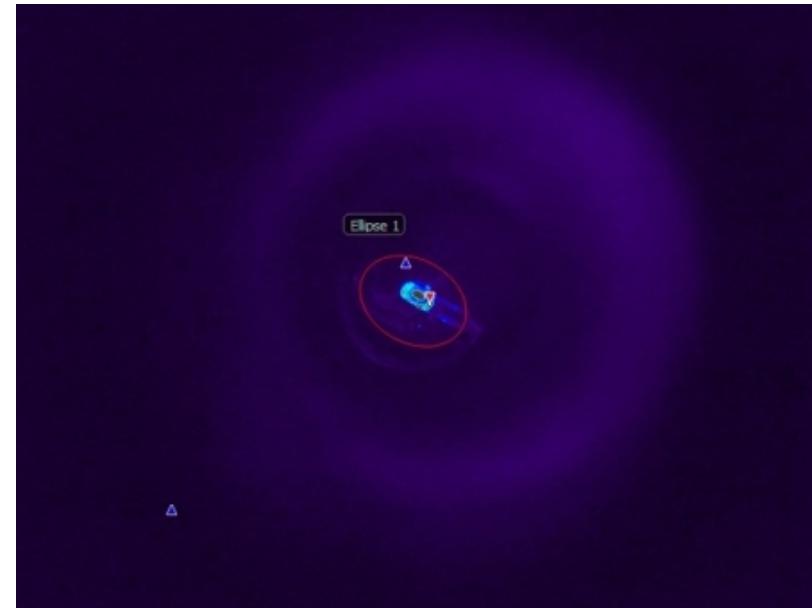
Combining in-situ laser heating with heating and cooling holders results in varying cooling rates



Experiment Temperature	Holder Used	Peak Laser Power (W)	Laser Power Increment (W)
Room	Double Tilt	4	1
Cryogenic	Cryo DT	10	1

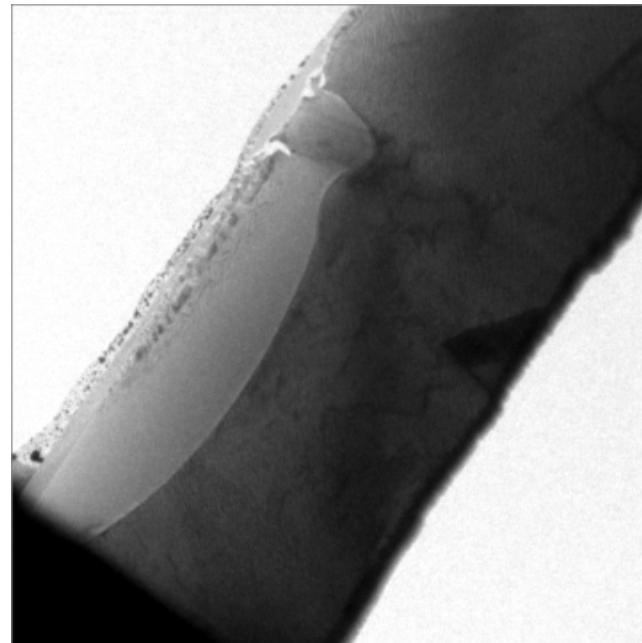
# Temperature Estimation

- FLIR Camera
  - Not high enough resolution to distinguish sample from laser beam
  - Calibration required for each phase due to different emissivity and absorptivity
- Diffraction pattern
  - Measurement of d-spacings at higher laser power
  - MATLAB code determines lattice strain through identification of DP spot centroids
    - Manually choose DP spots for each image
  - Strain and temperature results vary widely due to manual spot selection



# Change during laser exposure

- Further/faster change with increase from 4.8-5.4W
  - Ferrite degrades more than boride
- Overall change from pre- to post-exposure
  - Some recovery of Austenite phase visible
  - Degradation of ferrite and, to a lesser extent, boride
  - Most degradation of Pt deposition
    - may be helpful to remove Pt layer prior to laser heating to make results more clear



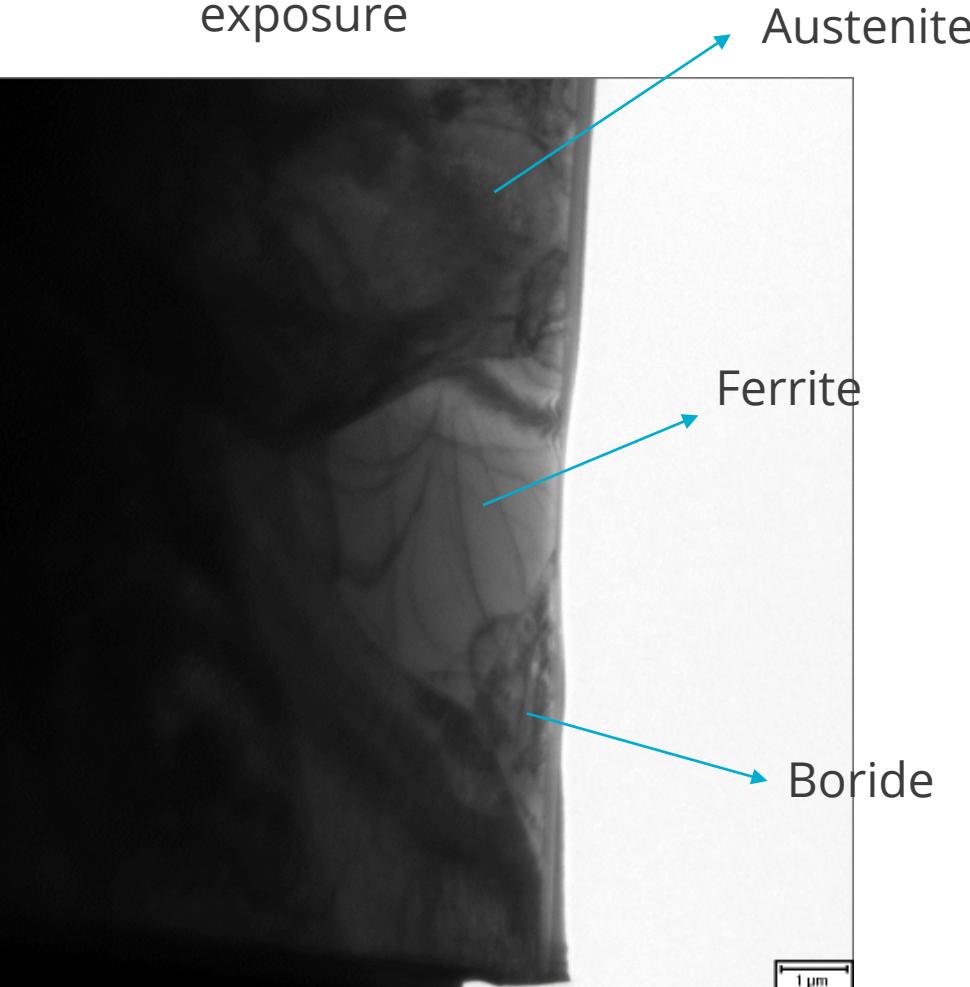
10 fps

Pre-laser heating at  
[112] zone axis

Post-laser heating (up to  
5.4W laser power for 10  
minutes) at [112] zone axis

# Room temperature

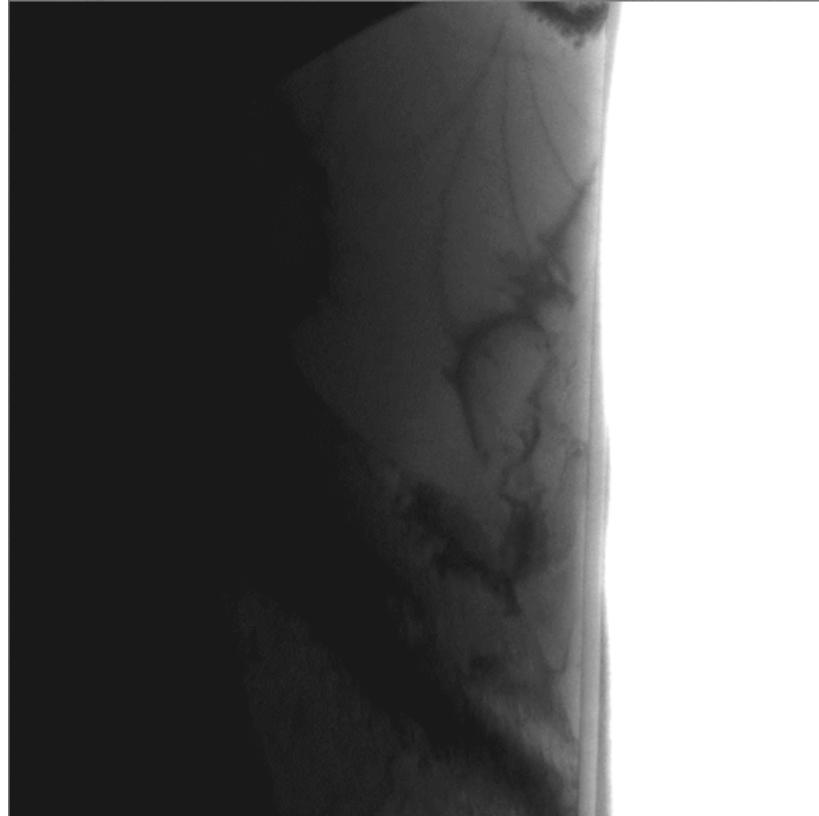
Sample at RT before laser exposure



At 0W

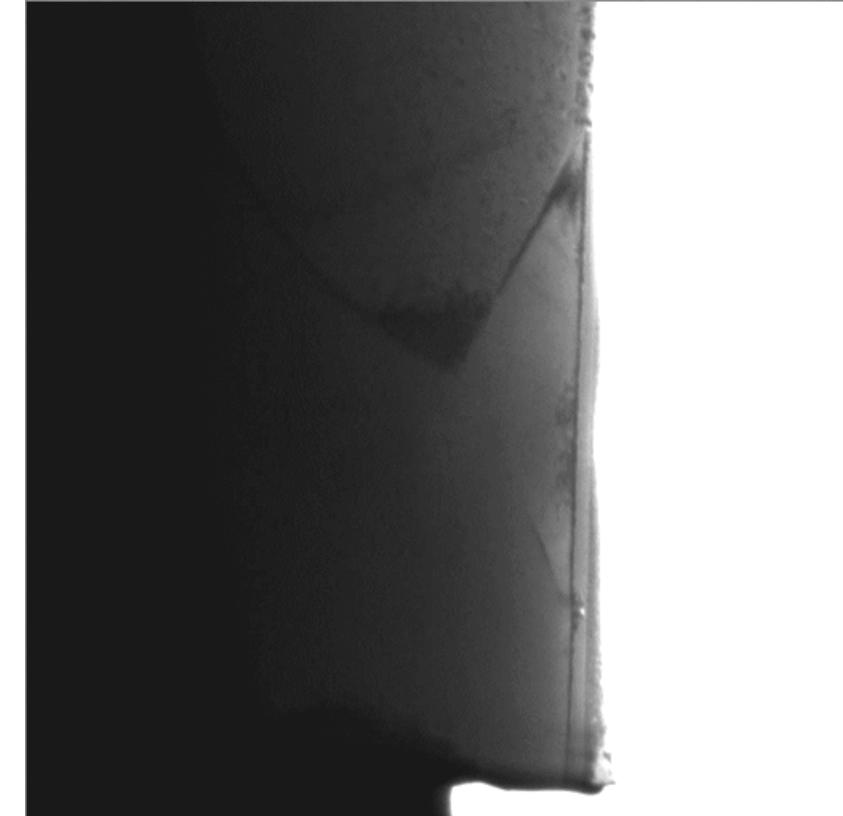
# Cooling to Room Temperature

20 fps



$3W \rightarrow 4W \rightarrow 0W$

5 fps



$4W \rightarrow 0W$

# Cryogenically cooled



Sample at -170°C before  
laser exposure

At 9W

Austenite

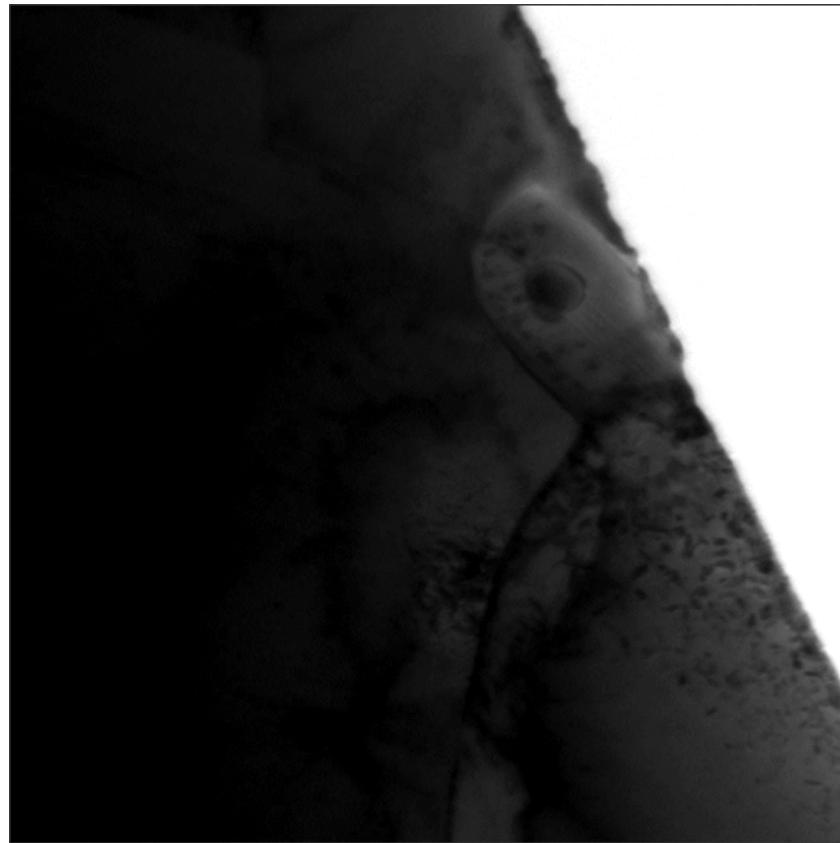
At 0W

Boride

Ferrite

# Cooling to Cryogenic Temperature (-170C)

28 fps



5 fps



9W → 10W → 0W

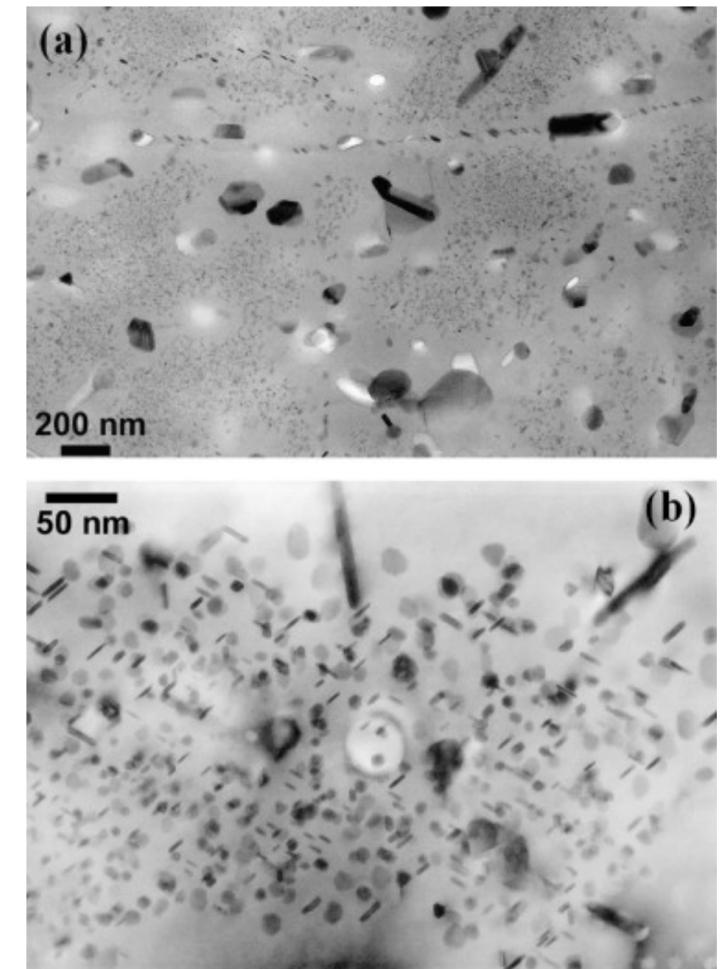
10W → 0W

# Microstructural Effects

Cooled to RT after 10W laser power

Cooled to -170C after 10W laser power

**Precipitates only form in ferrite; size seems to increase with increased cooling rate**



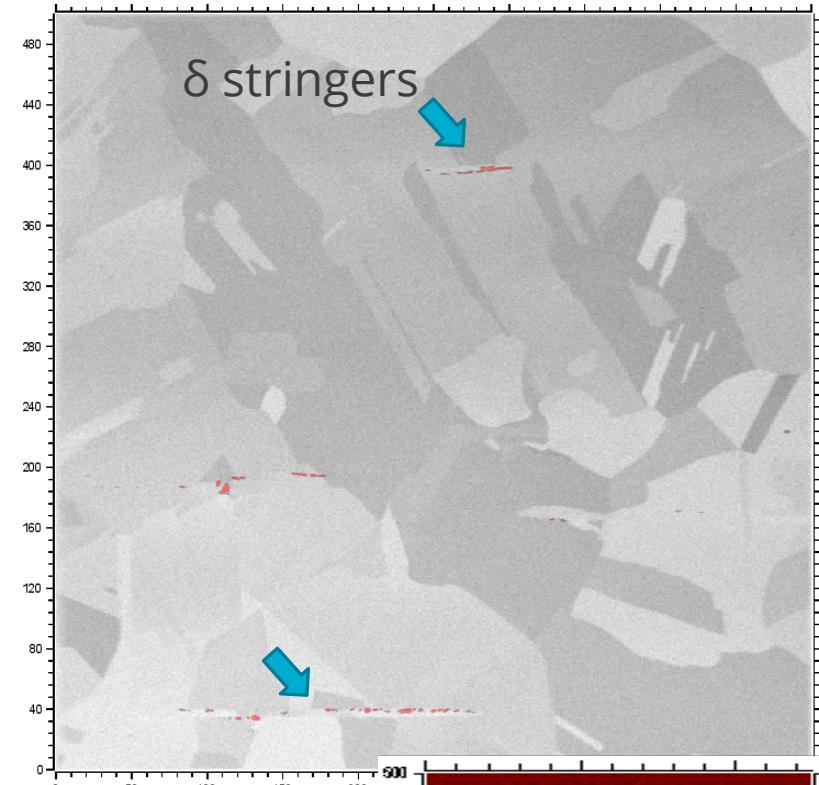
Al-Zn-Mg alloy quenched at 10C/min

Deschamps, A., et al. *Materials Science and Engineering: A* 501.1-2 (2009): 133-139.

# Boride dissolution occurs between 1000°C and 1100°C

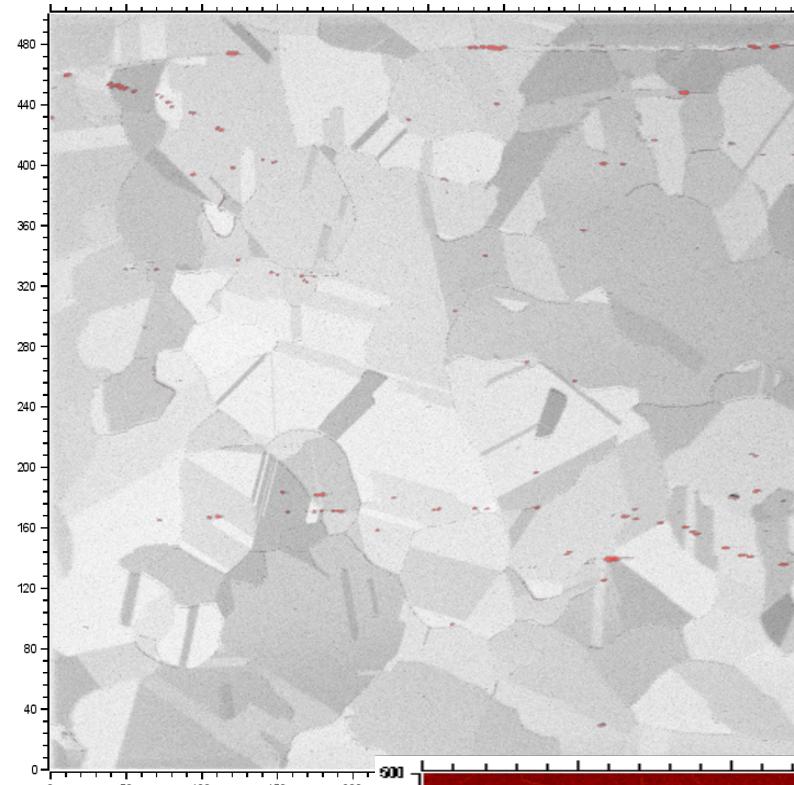


As-Received



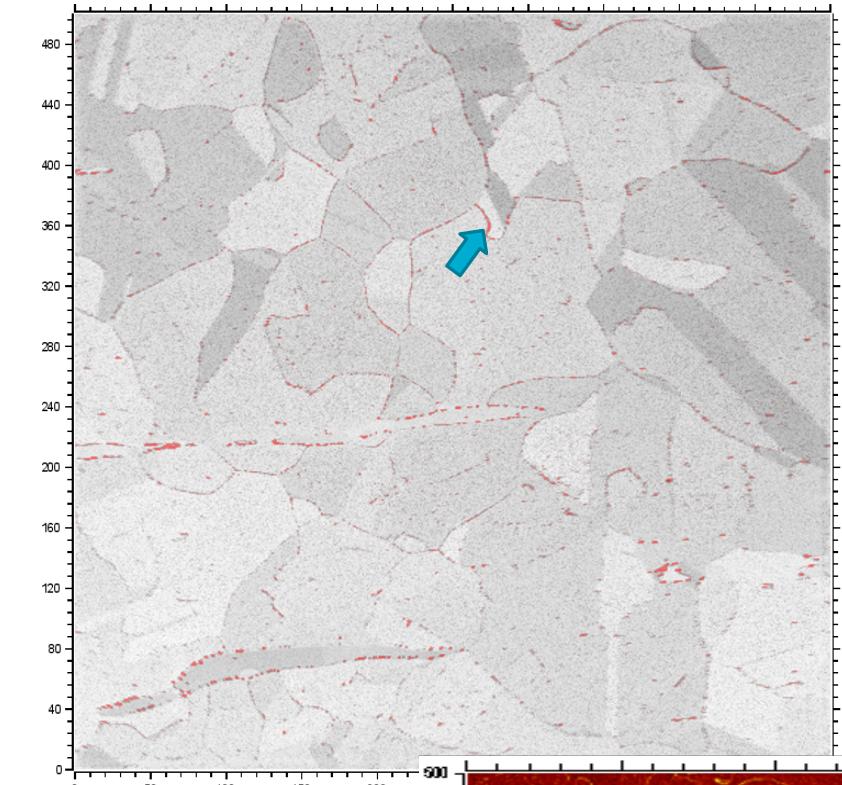
$\delta$  stringers

1000°C 32 min



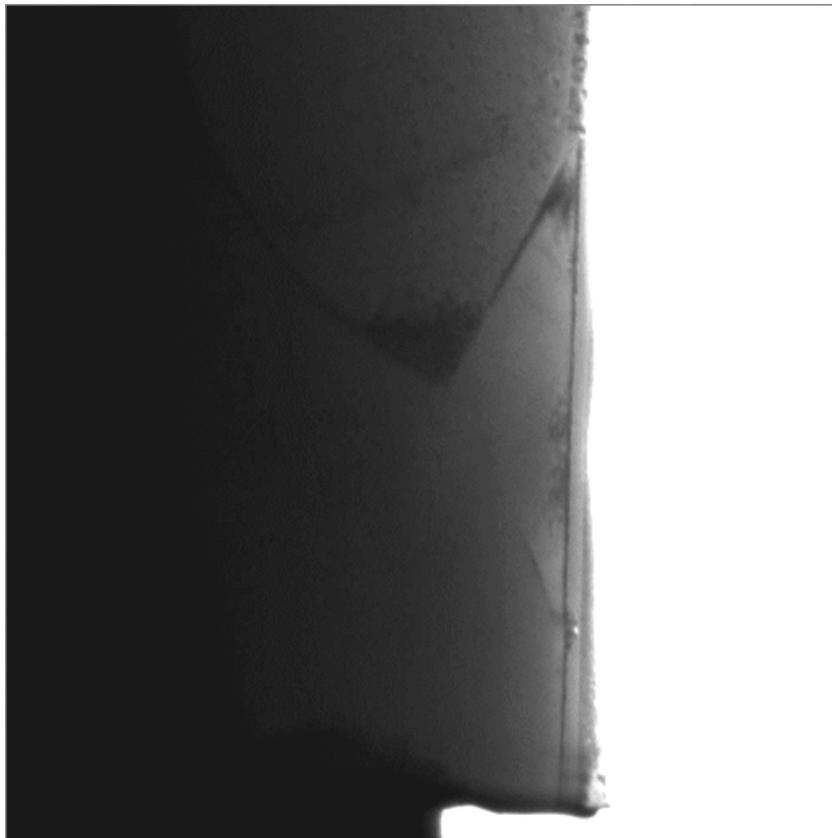
Little (if any) observable changes between as-received and 1000°C 32 min

1100°C 1 min



Some boron migration to  $\gamma$  grain boundaries; some remains on  $\delta$

# Approximation of Cooling Rate Observed

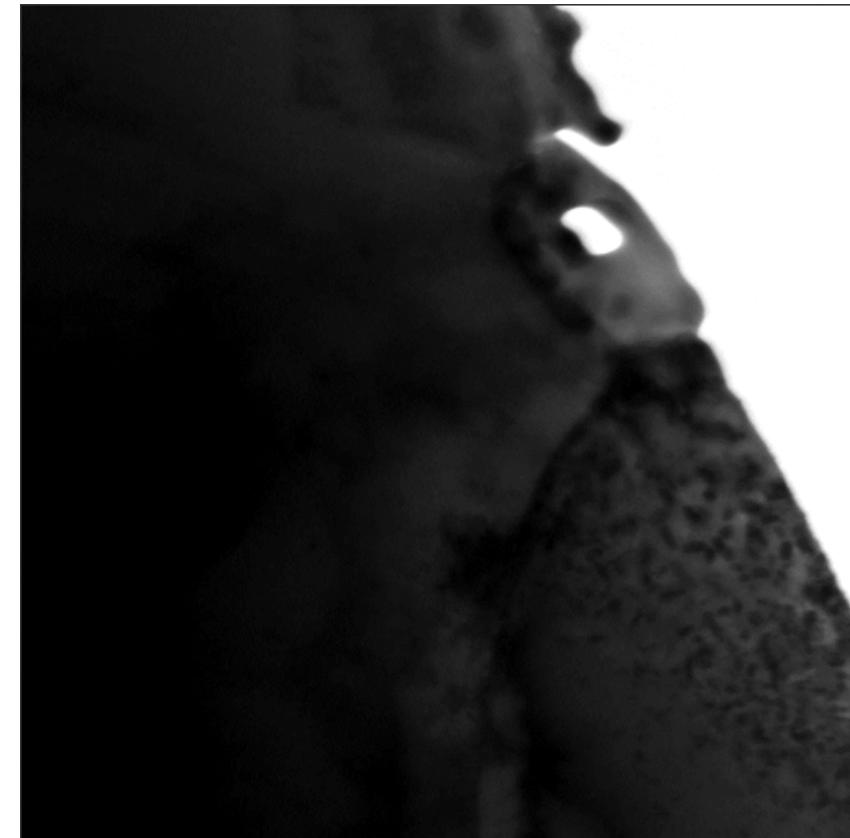


4W → 0W

Using RT holder, cooling  
lasted between 3-5 seconds

← ~366°C/s  
(457K/s)

Using cryo holder, cooling  
lasted no longer than .6  
seconds →  
~1833°C/s  
(2288K/s)

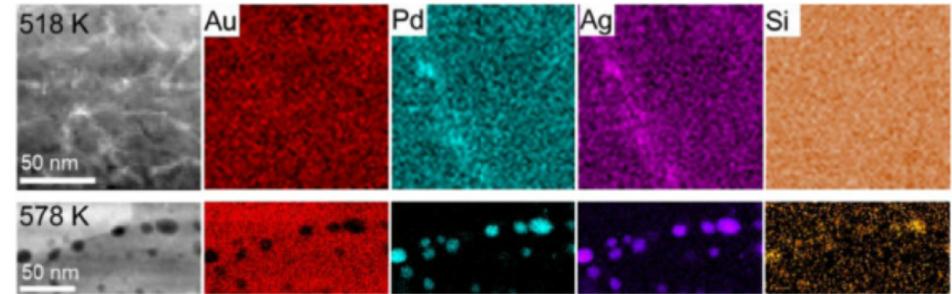


10W → 0W

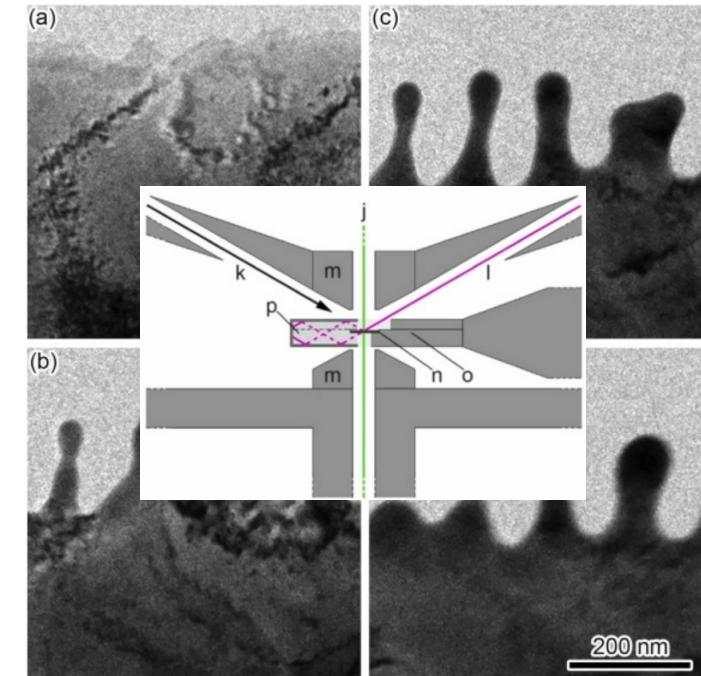
Decreasing temperature of holder  
is an easy way to increase cooling  
rate of sample

# Future cooling work with lasers

- Temperature measurement
- Precipitate identification in ferrite
- Application to laser fabrication techniques
  - Discrepancy between thermal conductivities (e.g. vacuum vs. metal powder)
- Crystallization of metallic glasses
  - Requires cooling rates  $\sim$ 1000s of K/s
- Development of heat resistant alloys
  - Protection of TEM components with specialized holders



Ivanov, Yu, et al. *Acta Materialia* 196 (2020) 52-60



Uemura, Naoki, et al. *Micron* 157 (2022): 103244.

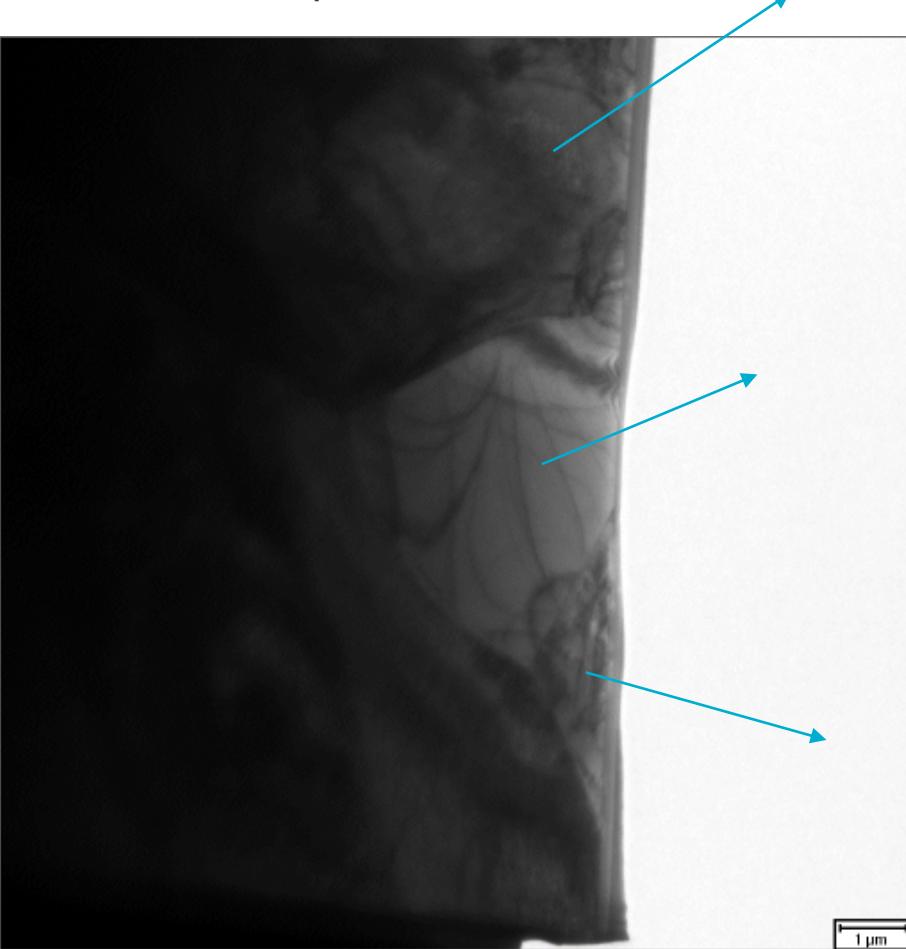


# Extra Slides

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# Room temperature

Sample at RT before laser exposure



Austenite

Ferrite

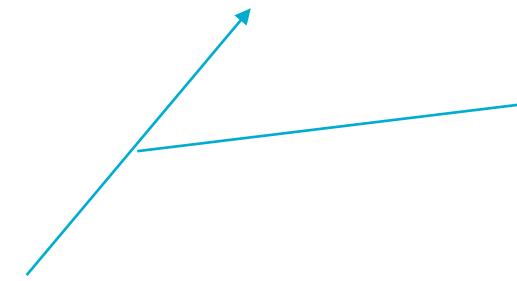
Boride

At 0W

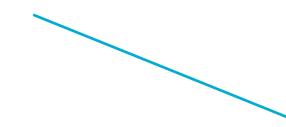
# Cryogenically cooled



At 9W



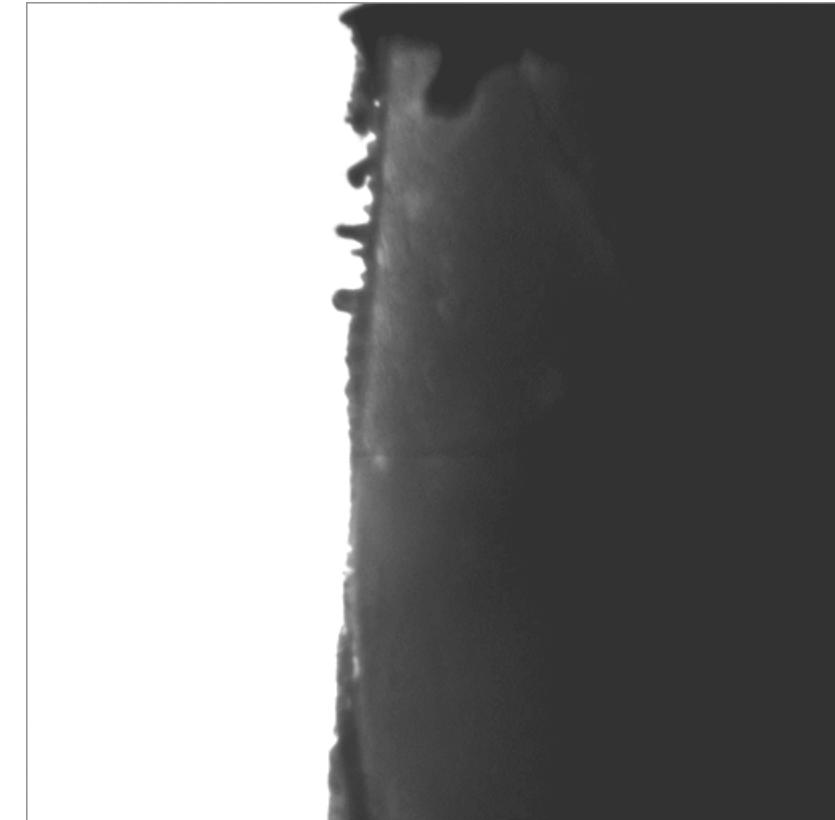
At 0W



$[\bar{1}22]$   
Austenite

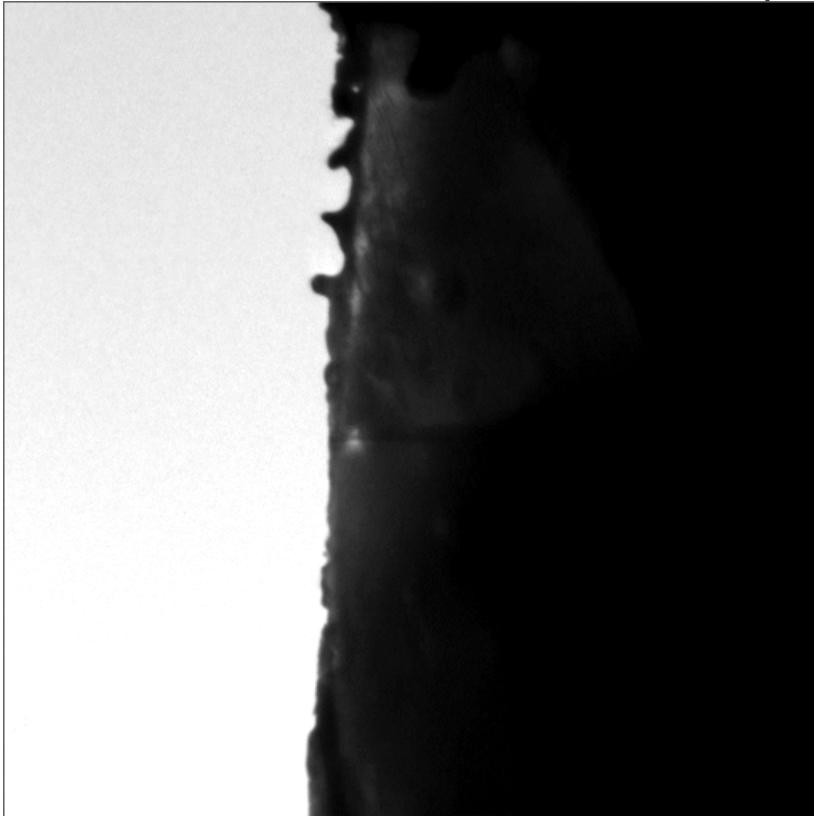
$[001]$   
Ferrite

# Heating Holder Cooling Experiment



# Heating Holder Cooling Experiment

30 fps



10 fps

