



In-situ TEM laser heating for manipulation of cooling rates and observation of precipitate dissolution kinetics



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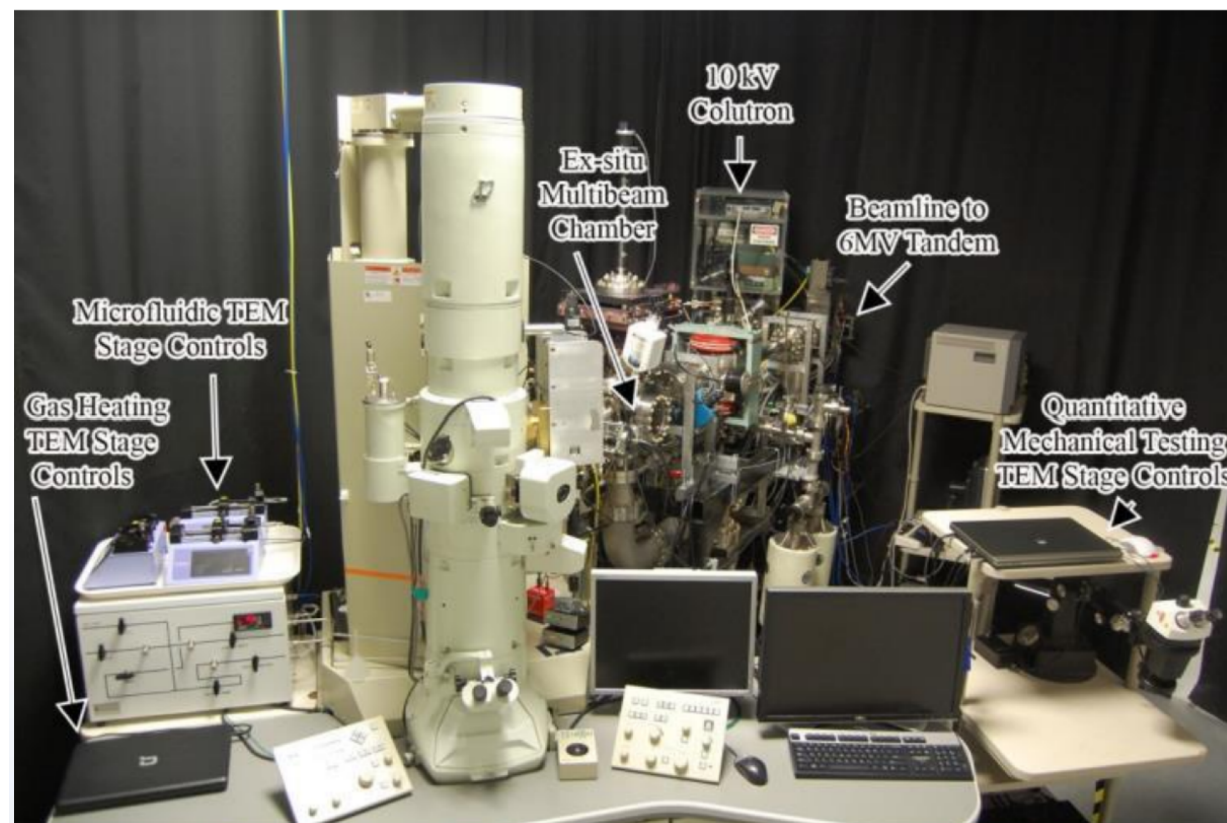
Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.



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In-situ TEM: Capabilities

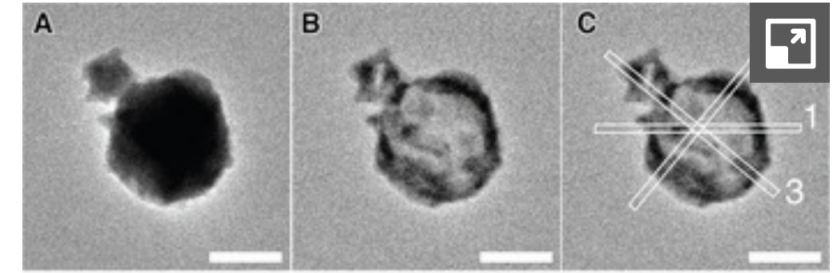
- The In-situ Ion Irradiation TEM (I³TEM) at the Ion Beam Laboratory at Sandia National Labs
- Variety of holders
 - △ Thermal: heating, cryo
 - △ Mechanical: straining, fatigue
 - △ Electro/chemical: liquid/gas cell with biasing
- Recent WAVIKS installation
- IR laser 1064nm wavelength, 20W



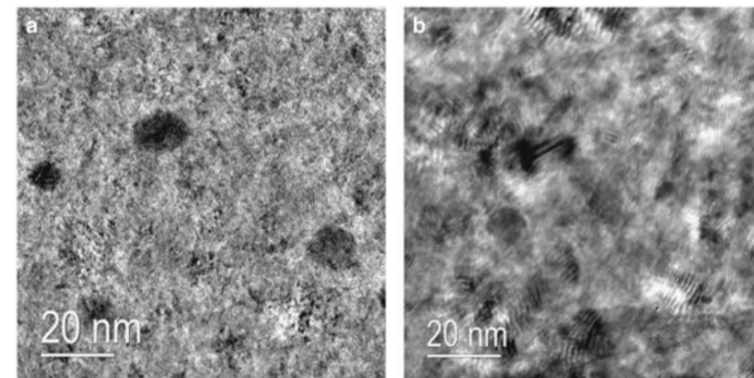
Current heating work with lasers

- Observe thermally activated microstructural processes
- Crystallization of amorphous materials
- Synthesis of nanowires
- Nanoparticle synthesis through laser ablation
- Hollow zinc oxide nanoparticle generation using infrared heating

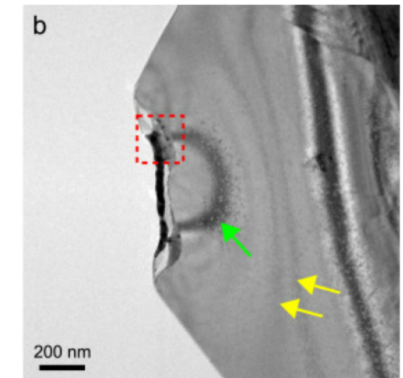
Taheri, Mitra L., et al. "Laser-based in situ techniques: Novel methods for generating extreme conditions in TEM samples." *Microscopy research and technique* 72.3 (2009): 122-130.



Mehraeen, Shareghe, et al. *Microscopy and Microanalysis* 19.2 (2013)



Wu, Yueying, et al. *Microscopy and Microanalysis* 24.6 (2018): 647-656.



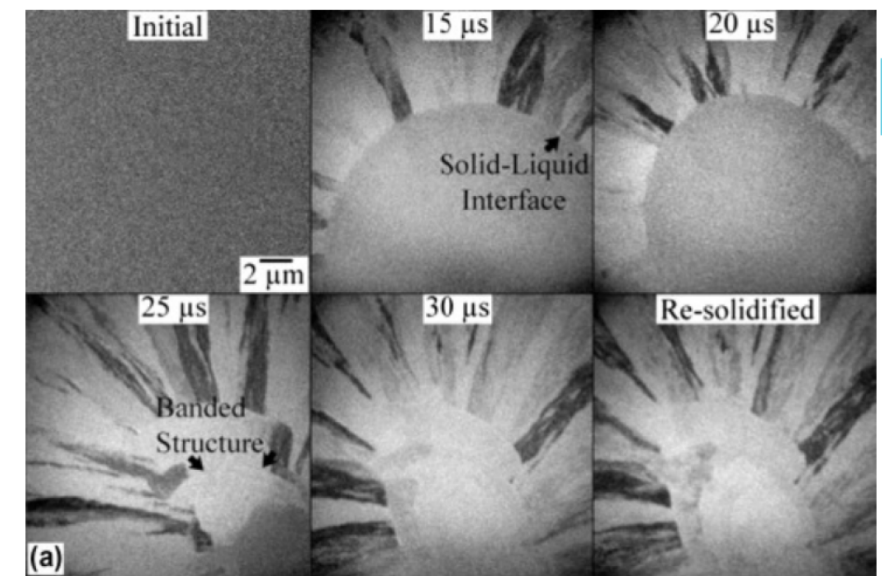
Allen, F. I., et al. *Ultramicroscopy* 178 (2017)

Observations of cooling

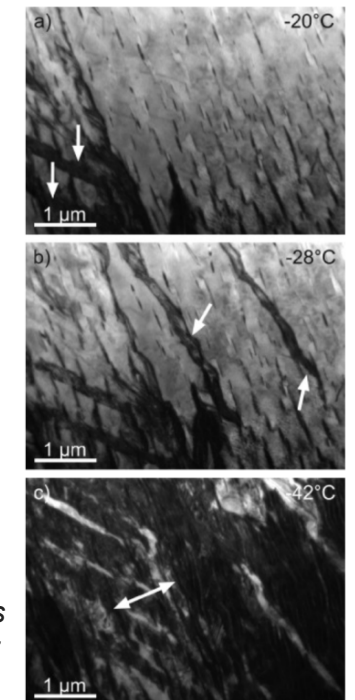
- Rapid solidification after laser processing
- Martensitic transformations
 - Shape memory alloys
- Phase transformation at various cooling rates

In-situ TEM lasers can be used to observe microstructural changes in real time during ultrafast cooling

Dlouhy, Antonin, Jafar Khalil-Allafi, and Gunther Eggeler. *Philosophical Magazine* 83.3 (2003): 339-363.



McKeown, Joseph T., et al. *Acta Materialia* 65 (2014): 56-68.

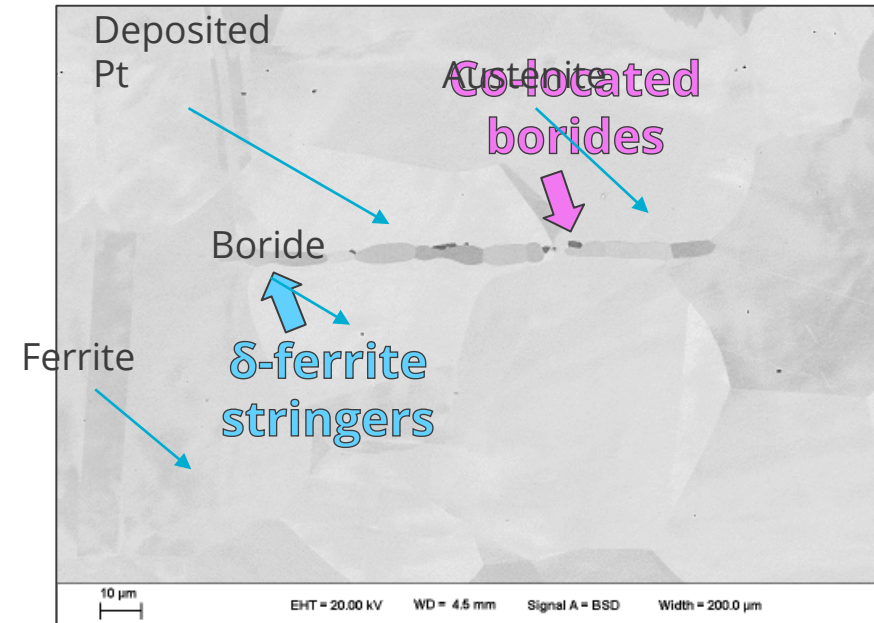


Kröger, A., et al. *Materials Science and Engineering: A* 481 (2008): 452-456.

Methods and Materials

- FIB lift-out of 304L VAR stainless steel
- Known to contain boride (CR2B) precipitates at Austenite/Ferrite phase boundaries
- After heat treatment, GB borides cause liquation cracking in welded SS
- Increased laser power incrementally to minimize thermal drift and allow equilibration
- Waited for features indicating melting of Pt
- Peak laser power differed between holder temperatures

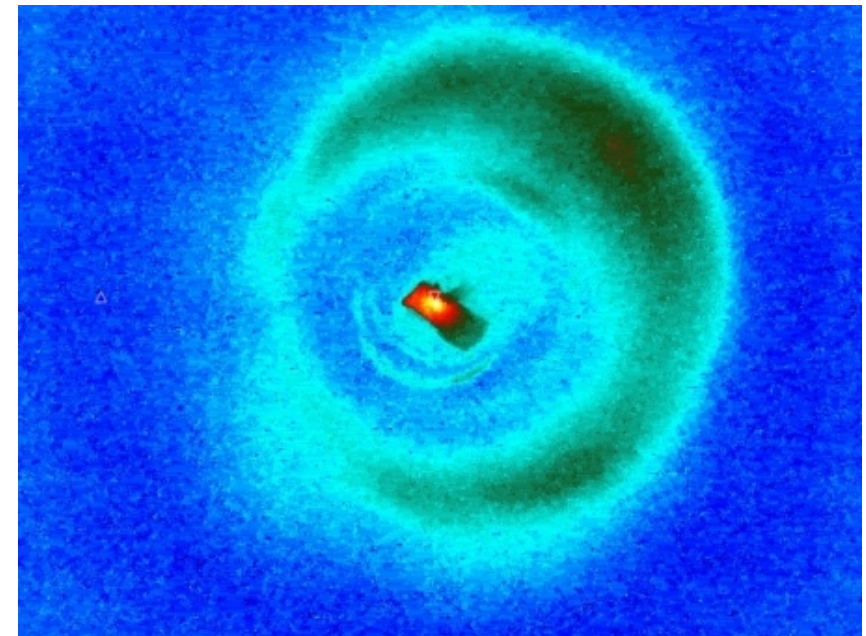
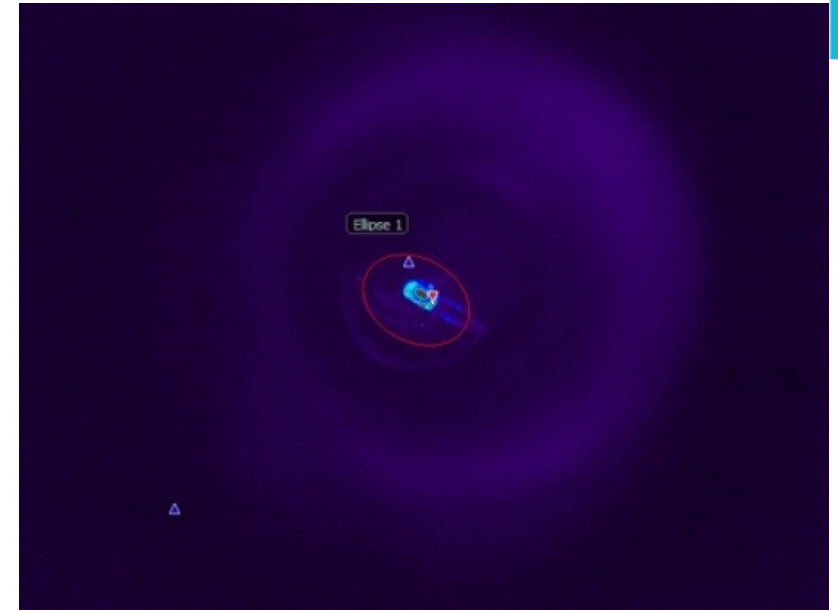
Combining in-situ laser heating with heating and cooling holders results in varying cooling rates



Experiment Temperature	Holder Used	Peak Laser Power (W)	Laser Power Increment (W)
Room	Double Tilt	4	1
Cryogenic	Cryo DT	10	1

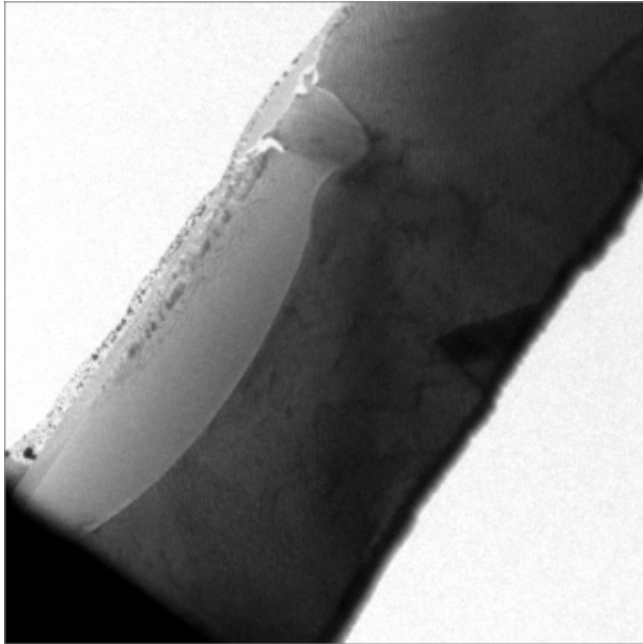
Temperature Estimation

- FLIR Camera
 - Not high enough resolution to distinguish sample from laser beam
 - Calibration required for each phase due to different emissivity and absorptivity
- Diffraction pattern
 - Measurement of d-spacings at higher laser power
 - MATLAB code determines lattice strain through identification of DP spot centroids
 - Manually choose DP spots for each image
 - Strain and temperature results vary widely due to manual spot selection



Change during laser exposure

- Further/faster change with increase from 4.8-5.4W
 - Ferrite degrades more than boride



10 fps

- Overall change from pre- to post-exposure
 - Some recovery of Austenite phase visible
 - Degradation of ferrite and, to a lesser extent, boride
 - Most degradation of Pt deposition
 - may be helpful to remove Pt layer prior to laser heating to make results more clear

Post-laser heating (up to 5.4W laser power for 10 minutes) at $[\bar{1}12]$ zone axis

Pre-laser heating at $[\bar{1}12]$ zone axis

Room temperature

At 3W



Sample at RT before laser exposure

Austenite

Ferrite

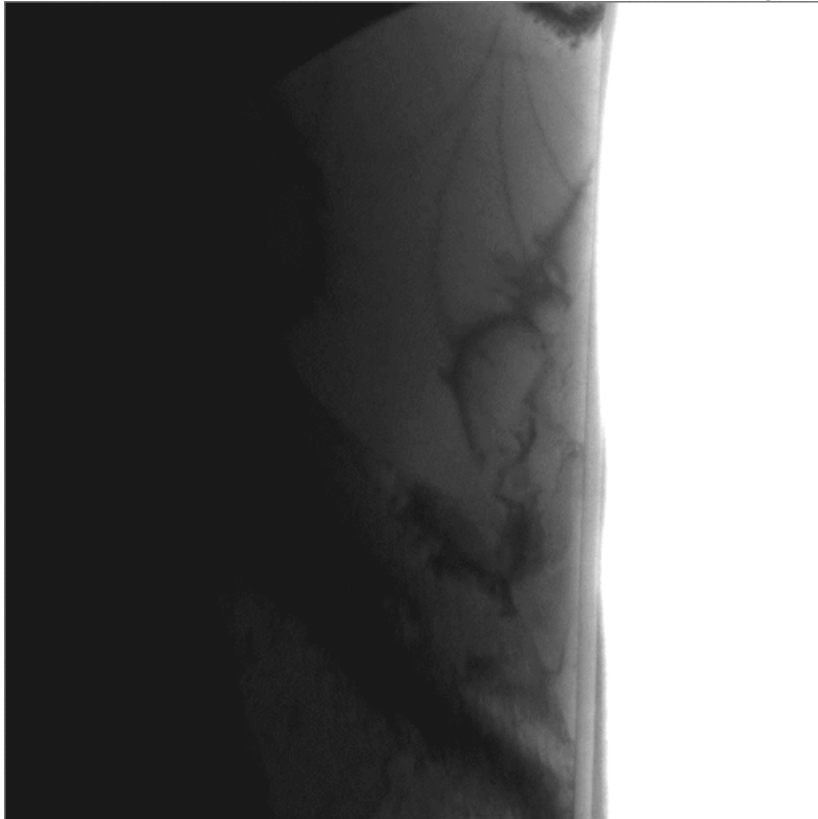
Boride

1 μm

At 0W

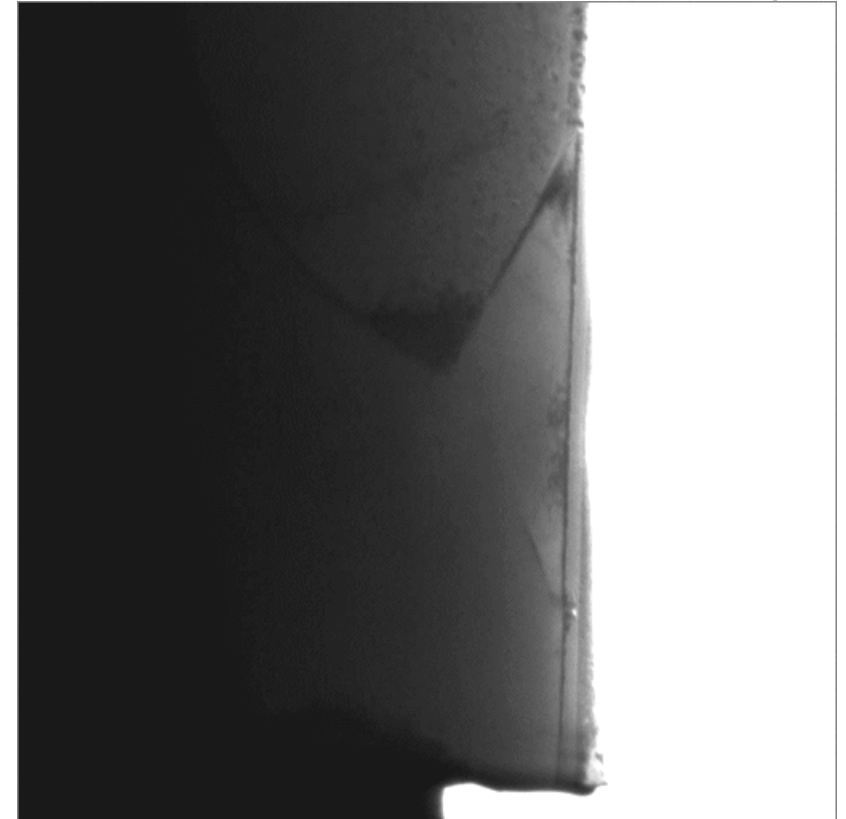
Cooling to Room Temperature

20 fps



3W → 4W → 0W

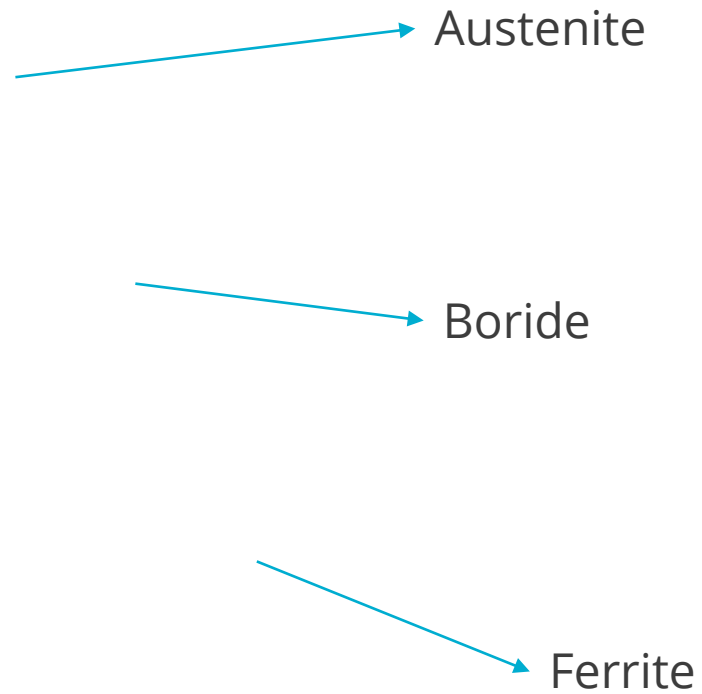
5 fps



4W → 0W

Cryogenically cooled

Sample at -170°C before
laser exposure



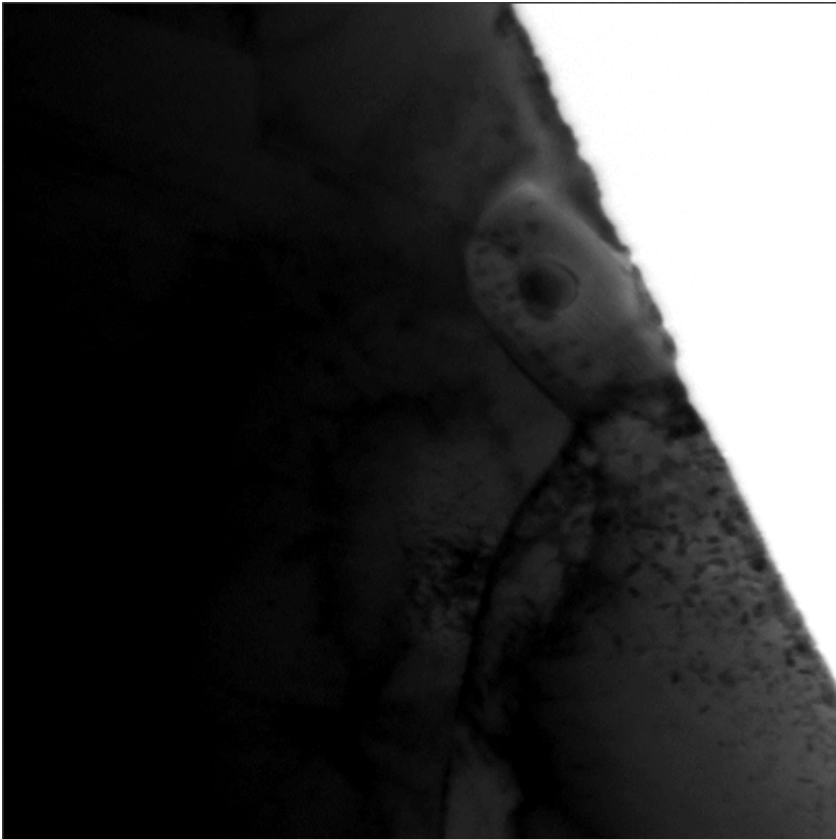
At 9W

At 0W



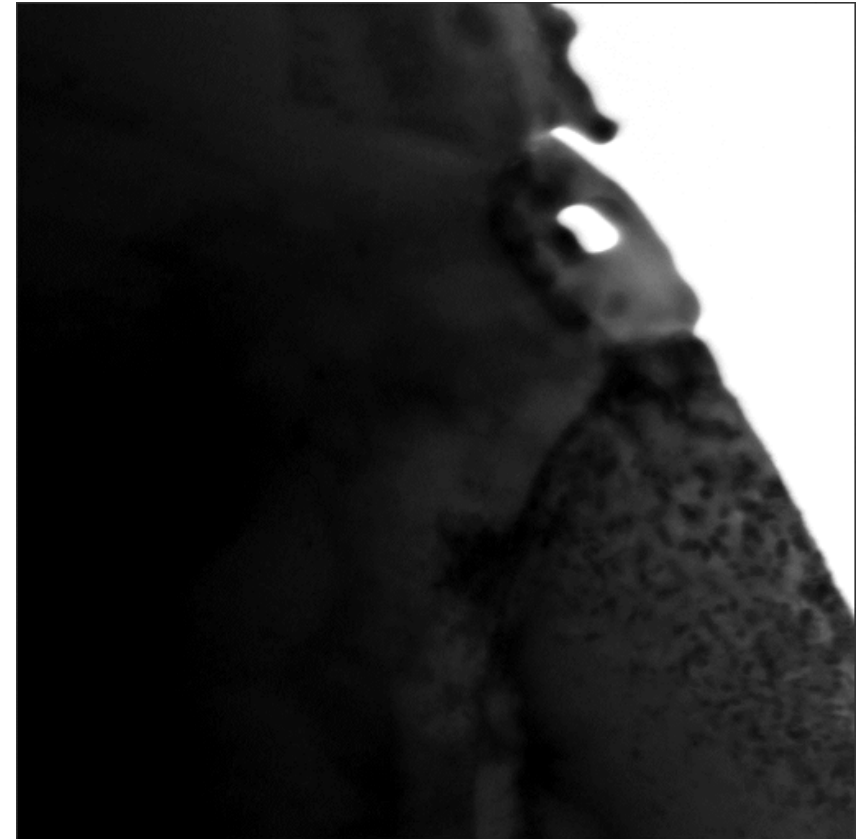
Cooling to Cryogenic Temperature (-170C)

28 fps



9W → 10W → 0W

5 fps



10W → 0W

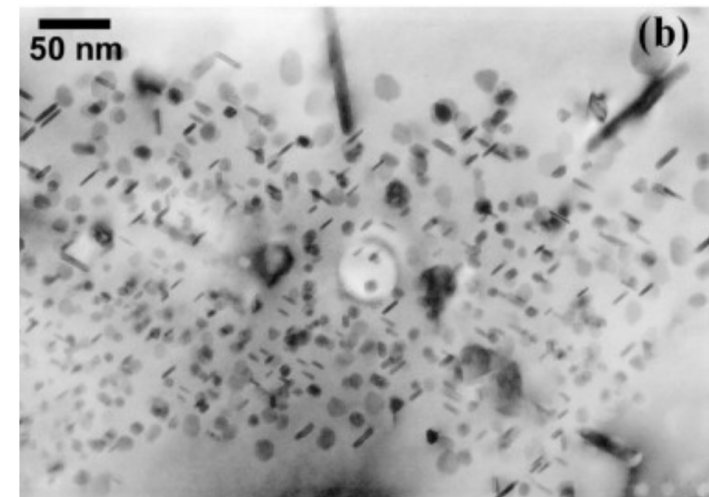
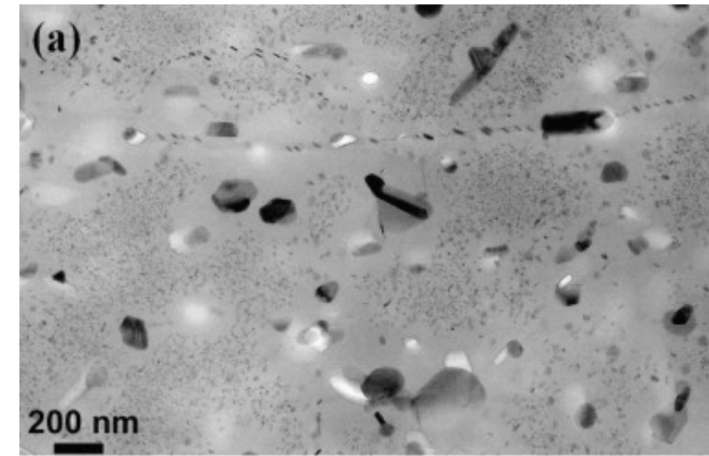
Microstructural Effects



Cooled to RT after 10W laser power

Cooled to -170C after 10W laser power

Precipitates only form in ferrite; size seems to increase with increased cooling rate



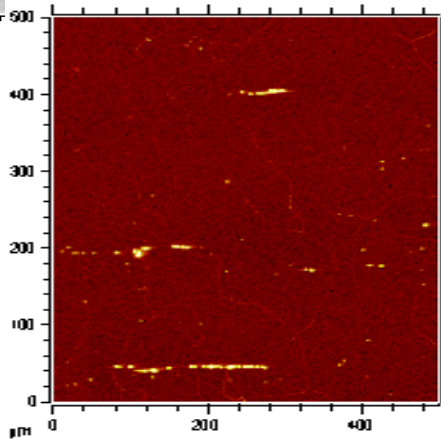
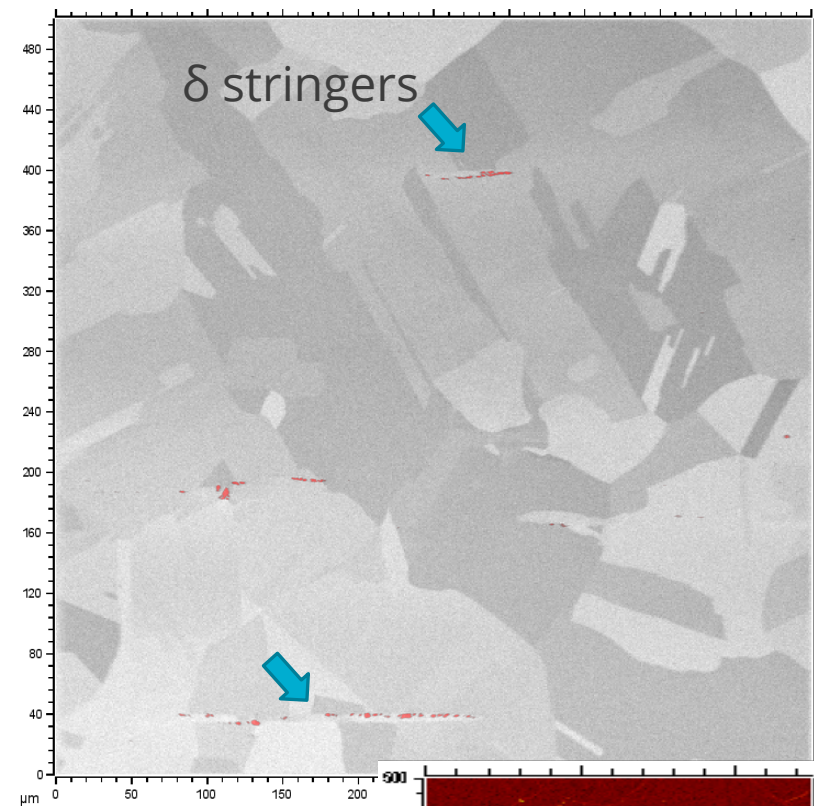
Al-Zn-Mg alloy quenched at 10C/min

Deschamps, A., et al. *Materials Science and Engineering: A* 501.1-2 (2009): 133-139.

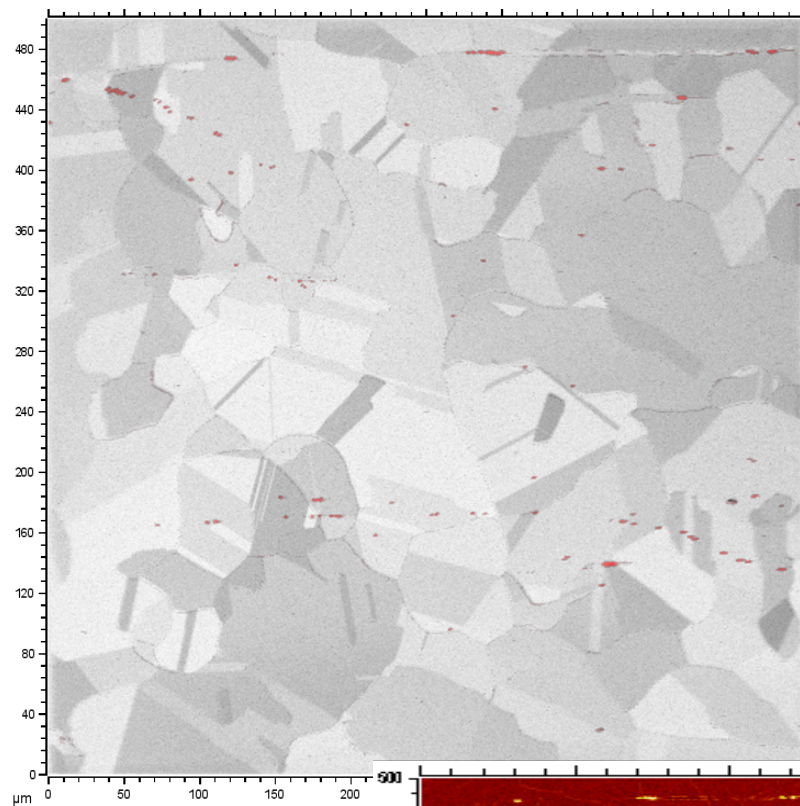
Boride dissolution occurs between 1000°C and 1100°C



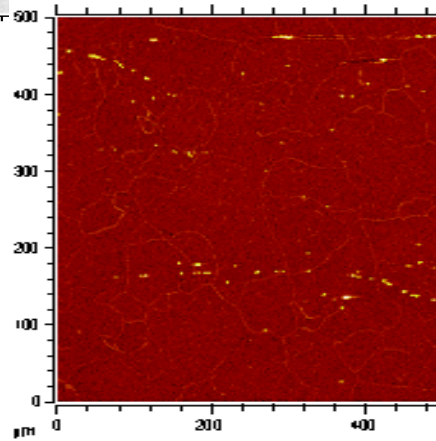
As-Received



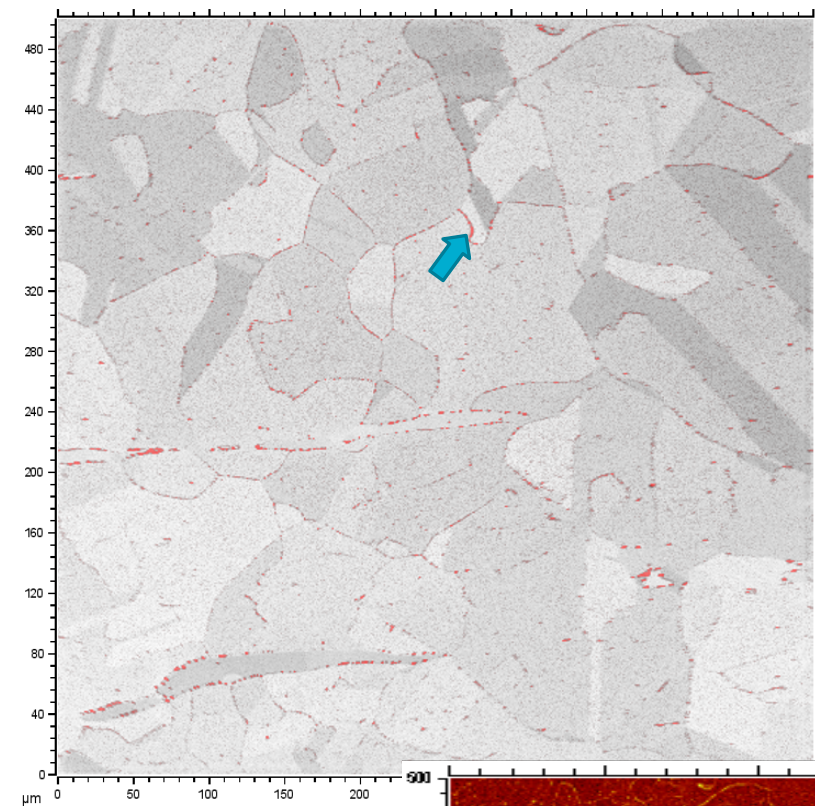
1000°C 32 min



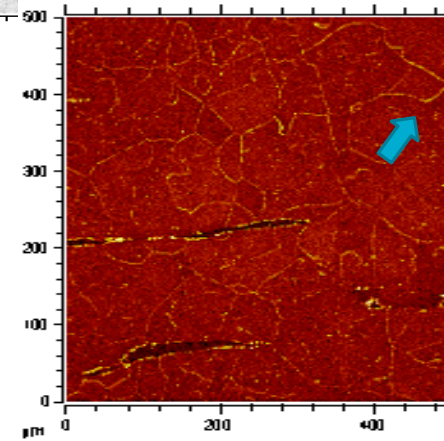
Little (if any) observable changes between as-received and 1000°C 32 min



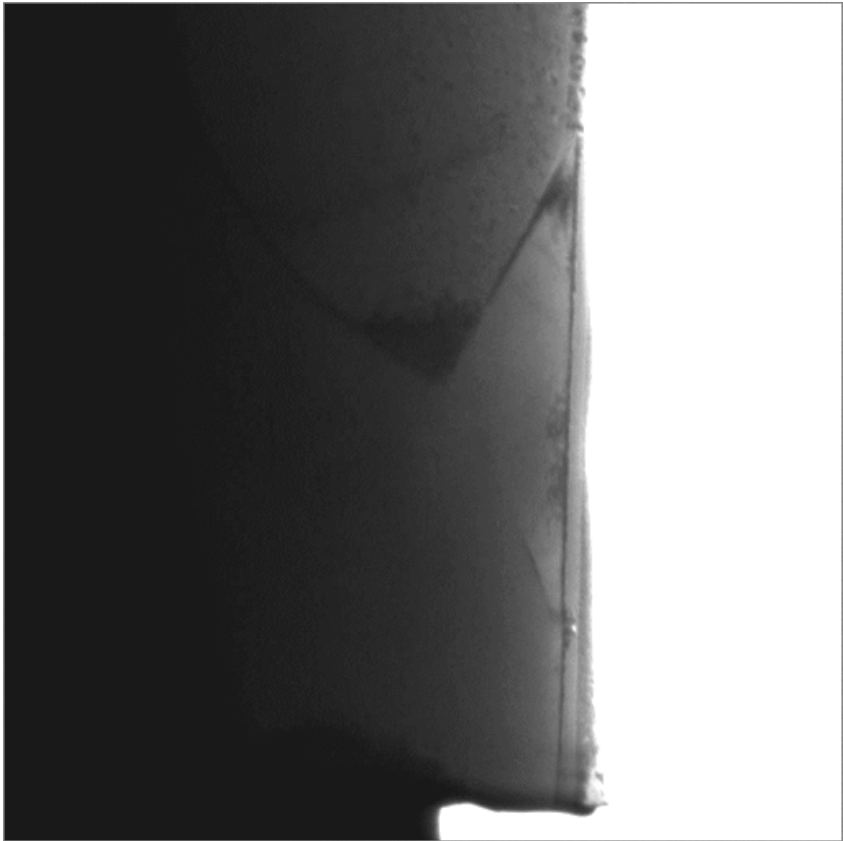
1100°C 1 min



Some boron migration to γ grain boundaries; some remains on δ



Approximation of Cooling Rate Observed



Using RT holder, cooling lasted between 3-5 seconds

← $\sim 366^{\circ}\text{C/s}$
(457K/s)

Using cryo holder, cooling lasted no longer than .6 seconds →
 $\sim 1833^{\circ}\text{C/s}$
(2288K/s)



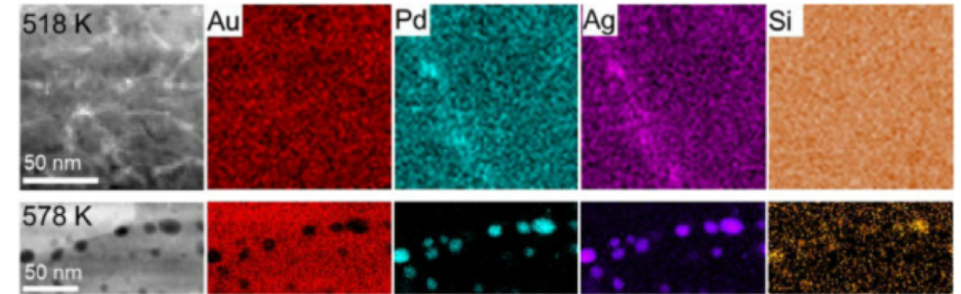
4W → 0W

Decreasing temperature of holder is an easy way to increase cooling rate of sample

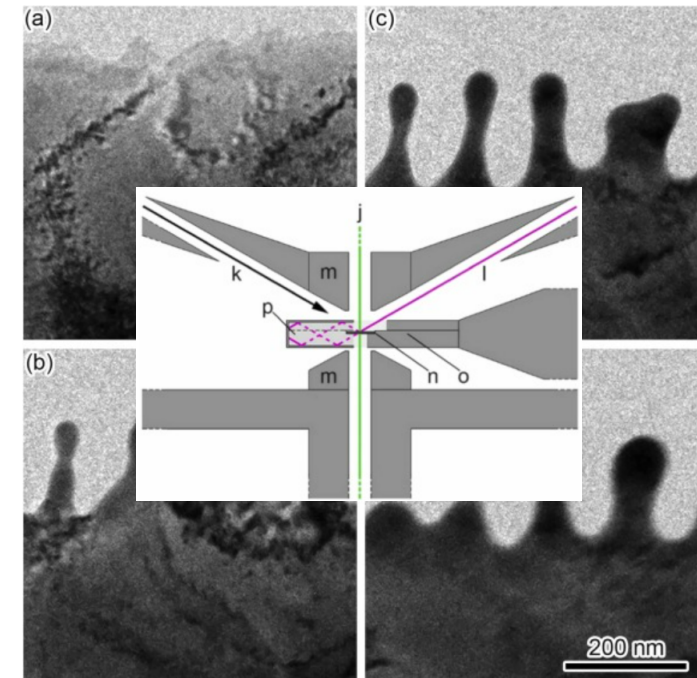
10W → 0W

Future cooling work with lasers

- Temperature measurement
- Precipitate identification in ferrite
- Application to laser fabrication techniques
 - Discrepancy between thermal conductivities (e.g. vacuum vs. metal powder)
- Crystallization of metallic glasses
 - Requires cooling rates ~ 1000 s of K/s
- Development of heat resistant alloys
 - Protection of TEM components with specialized holders



Ivanov, Yu, et al. *Acta Materialia* 196 (2020) 52-60



Uemura, Naoki, et al. *Micron* 157 (2022): 103244.



Extra Slides



Room temperature

At 3W



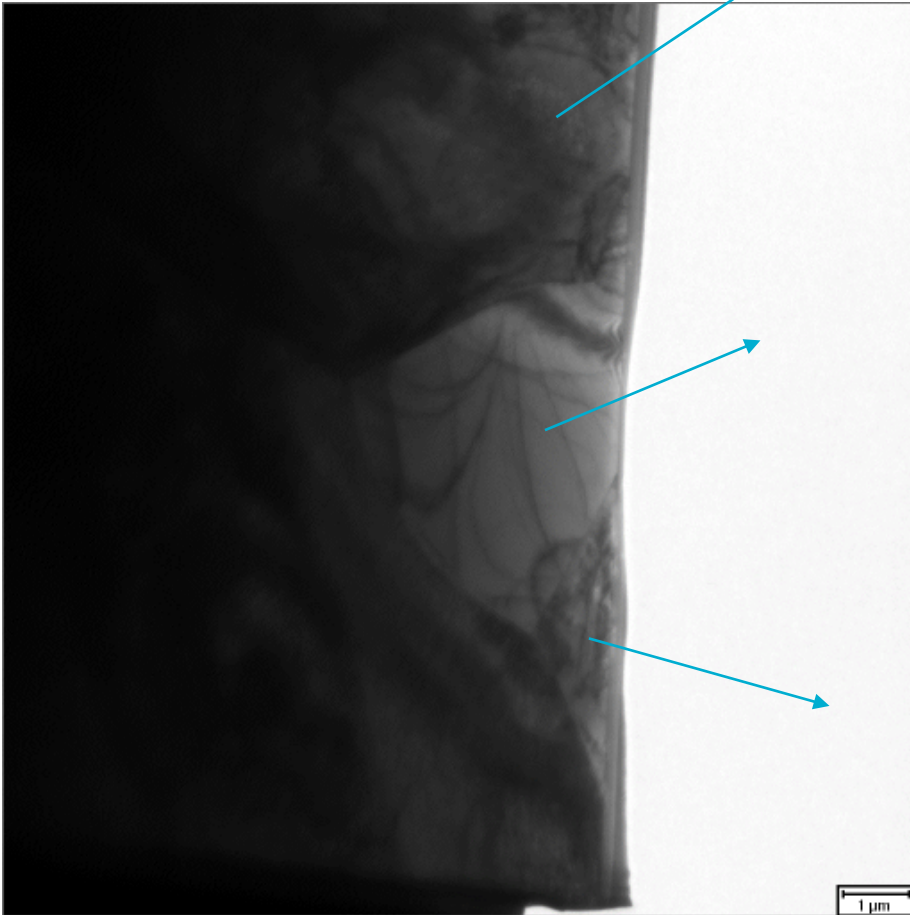
Sample at RT before laser exposure

Austenite

Ferrite

Boride

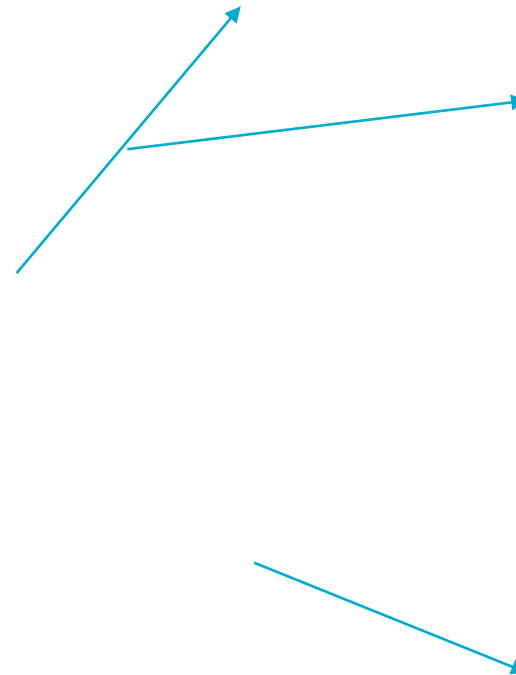
At 0W



Cryogenically cooled

At 9W

At 0W

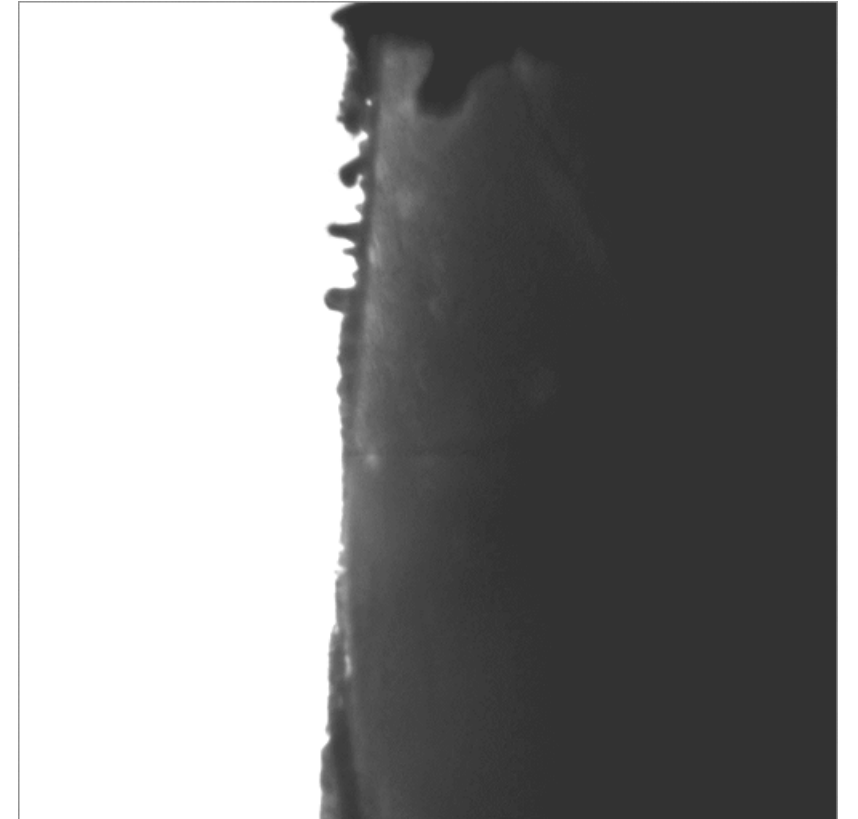


$[\bar{1}22]$
Austenite

$[001]$
Ferrite

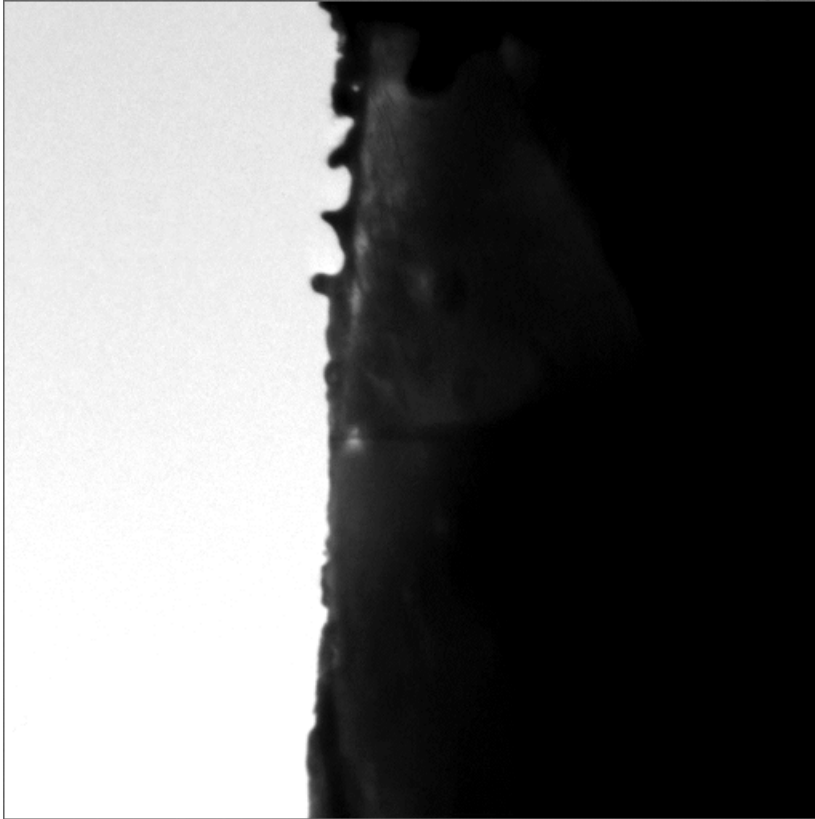


Heating Holder Cooling Experiment



Heating Holder Cooling Experiment

30 fps



10 fps

