

LightWorks®

# Techno-Economic Analysis of Solar-Thermal Ammonia Production

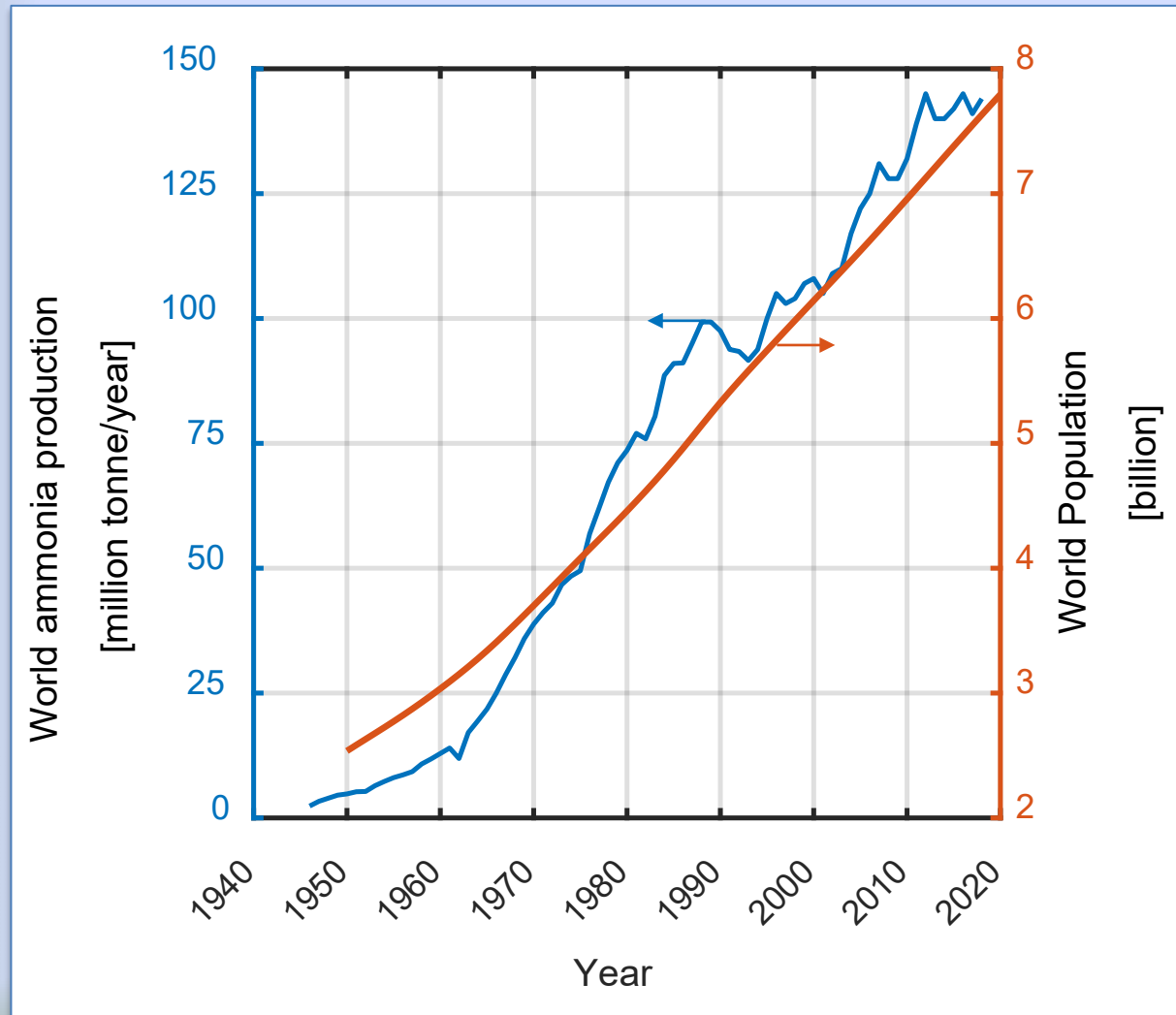
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# Ammonia Production



- Ammonia is the second most produced industrial chemical in the world
- About 80% of the nitrogen in our body comes from synthetic nitrogen fertilizer
- Responsible for over **1.4%** of global CO<sub>2</sub> emissions

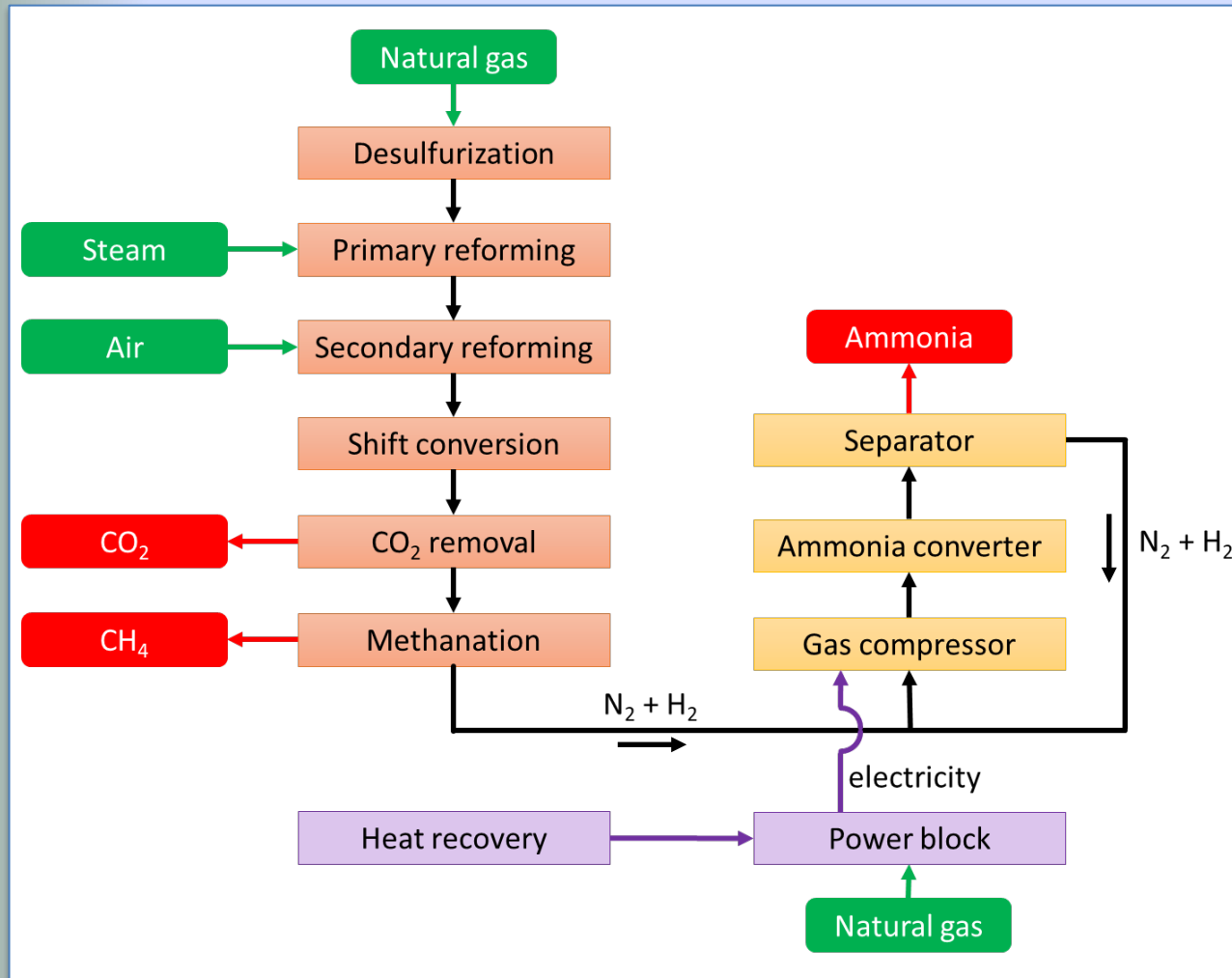
Data from:

[1] U.S. Geological Survey, National Minerals Information Center

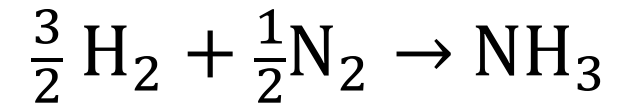
[2] United Nations, Department of Economic and Social Affairs



# The Haber-Bosch Process



Ammonia synthesis:



Exothermic reaction:  $\Delta H_{298\text{K}} = -45 \text{ kJ/mol}$

High pressure: 150-300 bar

Moderately elevated temperature: 350-500 °C

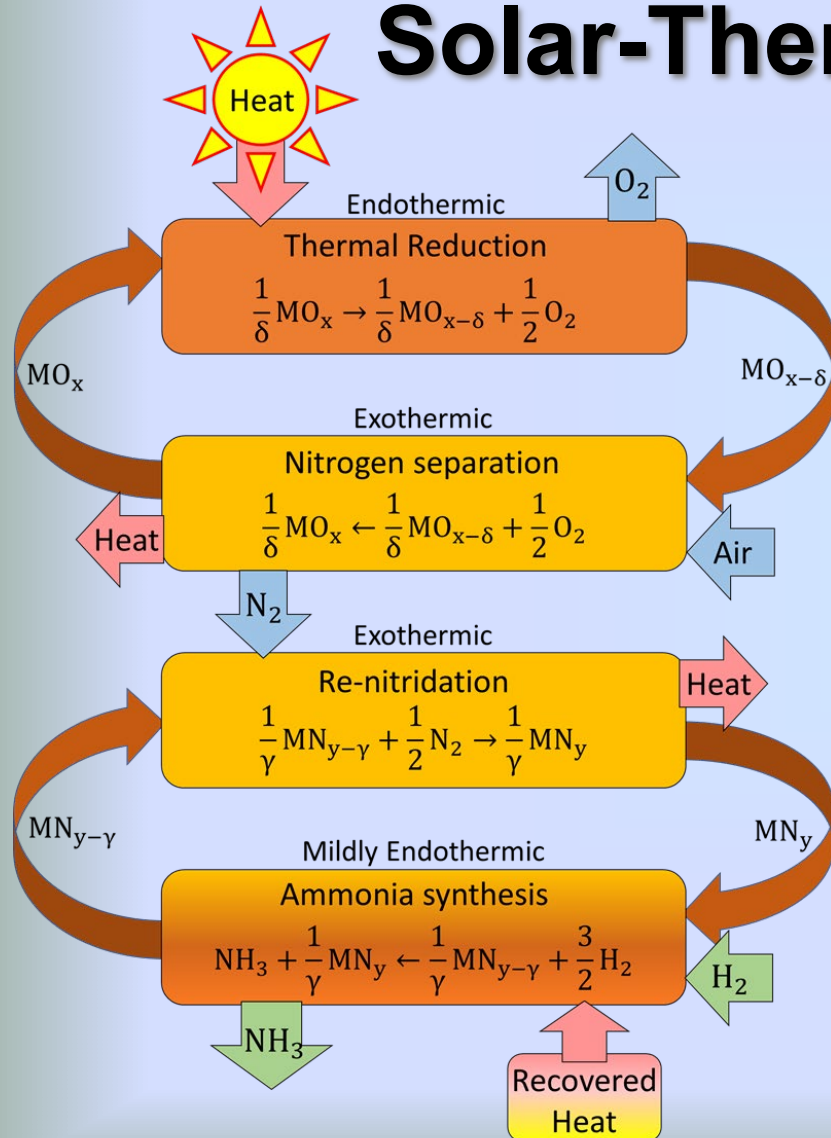
Low conversion: ~10-20%

- H<sub>2</sub> is produced via steam methane reforming
- N<sub>2</sub> is produced via methane combustion in air





# Solar-Thermal Ammonia Production

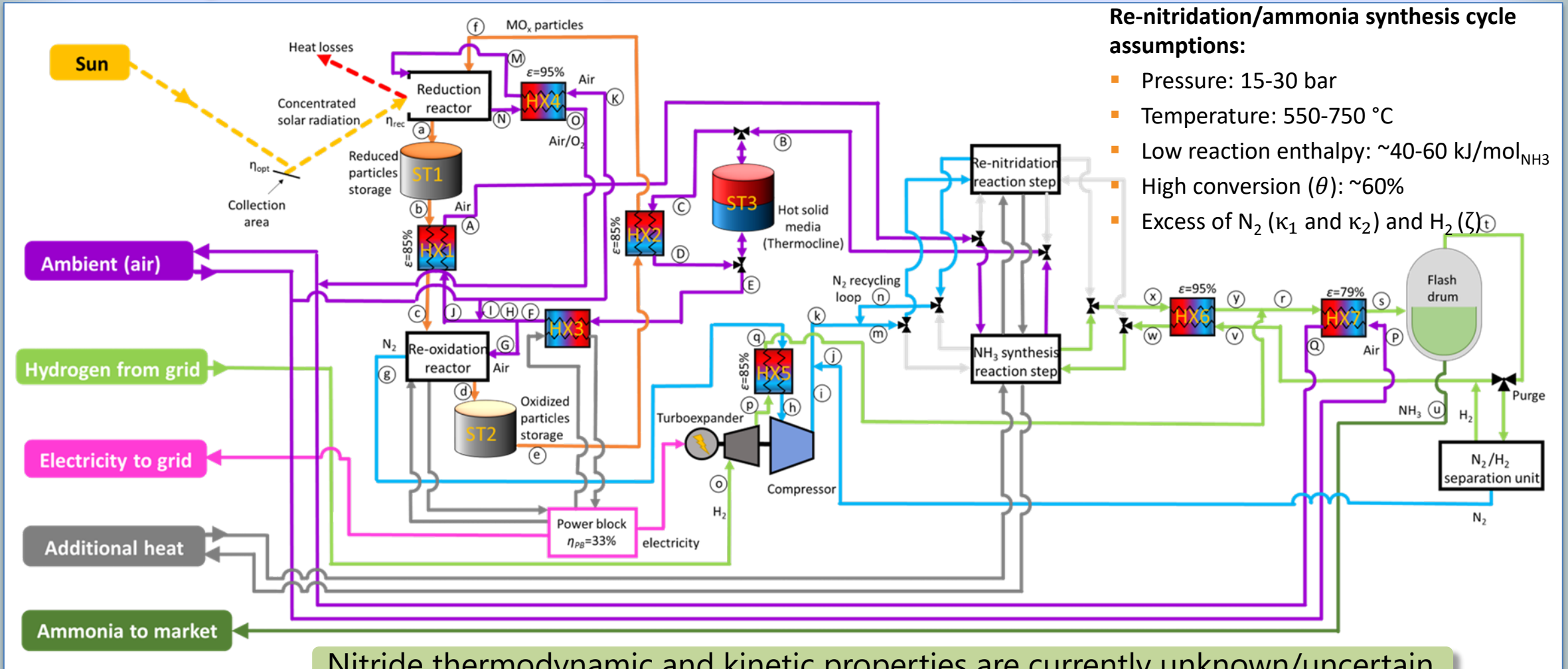


Goal: CO<sub>2</sub>-neutral ammonia production with concentrated solar technology

- **First stage:** nitrogen production from air using concentrated solar heat
- **Second stage:** ammonia production targeting pressures lower than HB by an order of magnitude

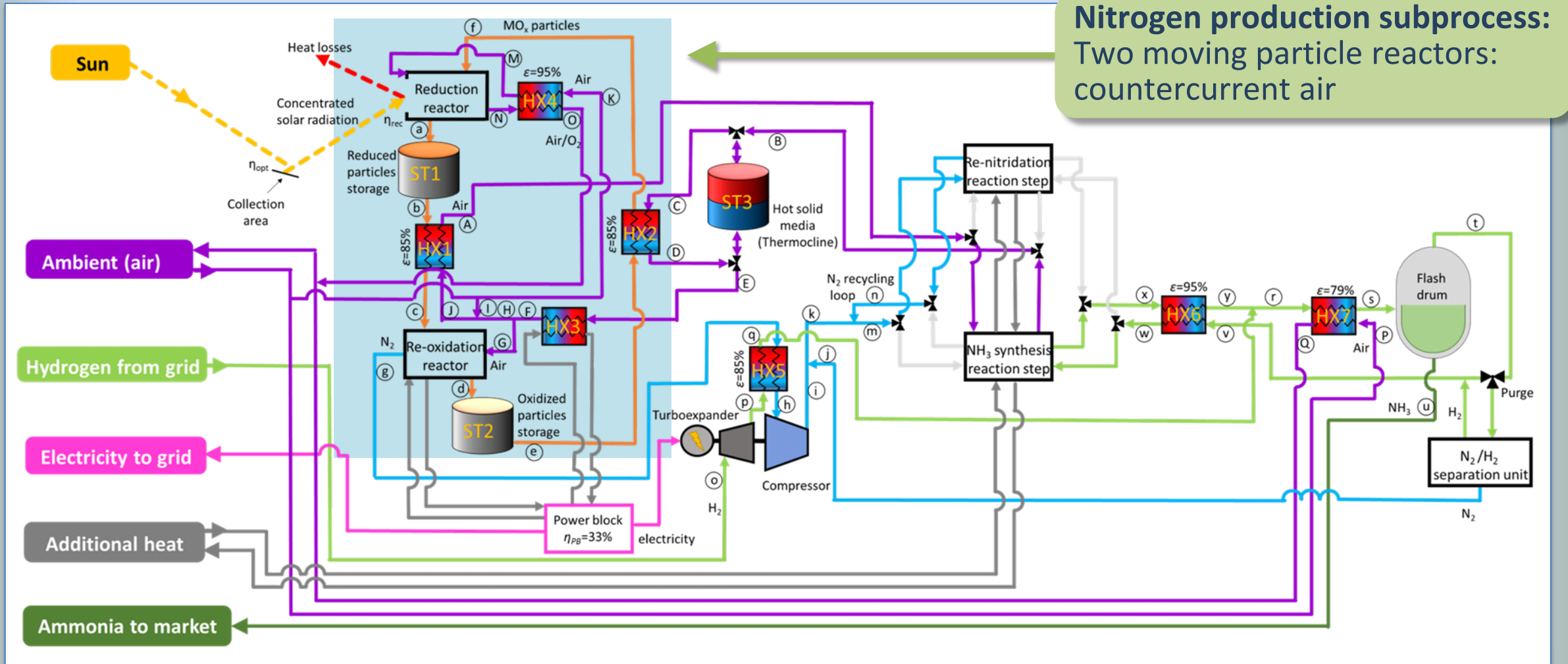


# System Description: Assumptions





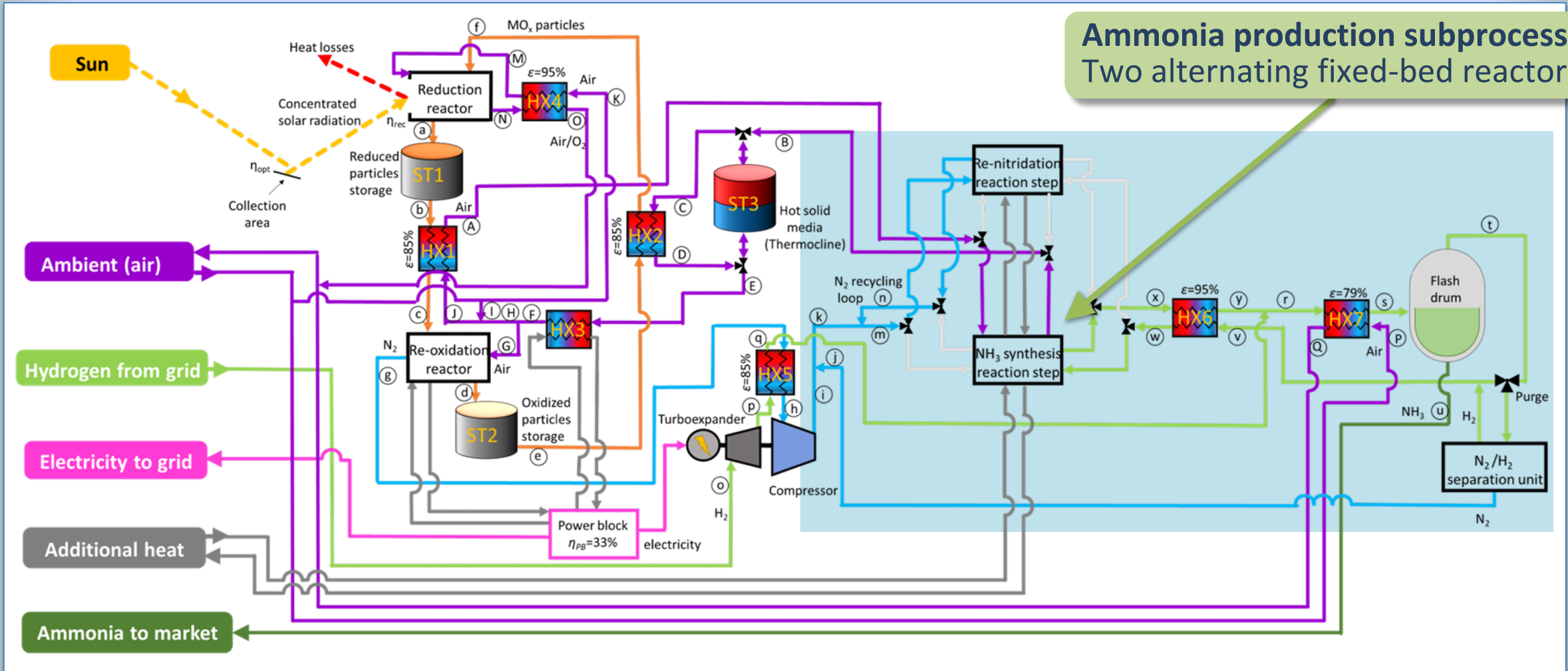
# System Description: Thermal Reduction and Air Separation





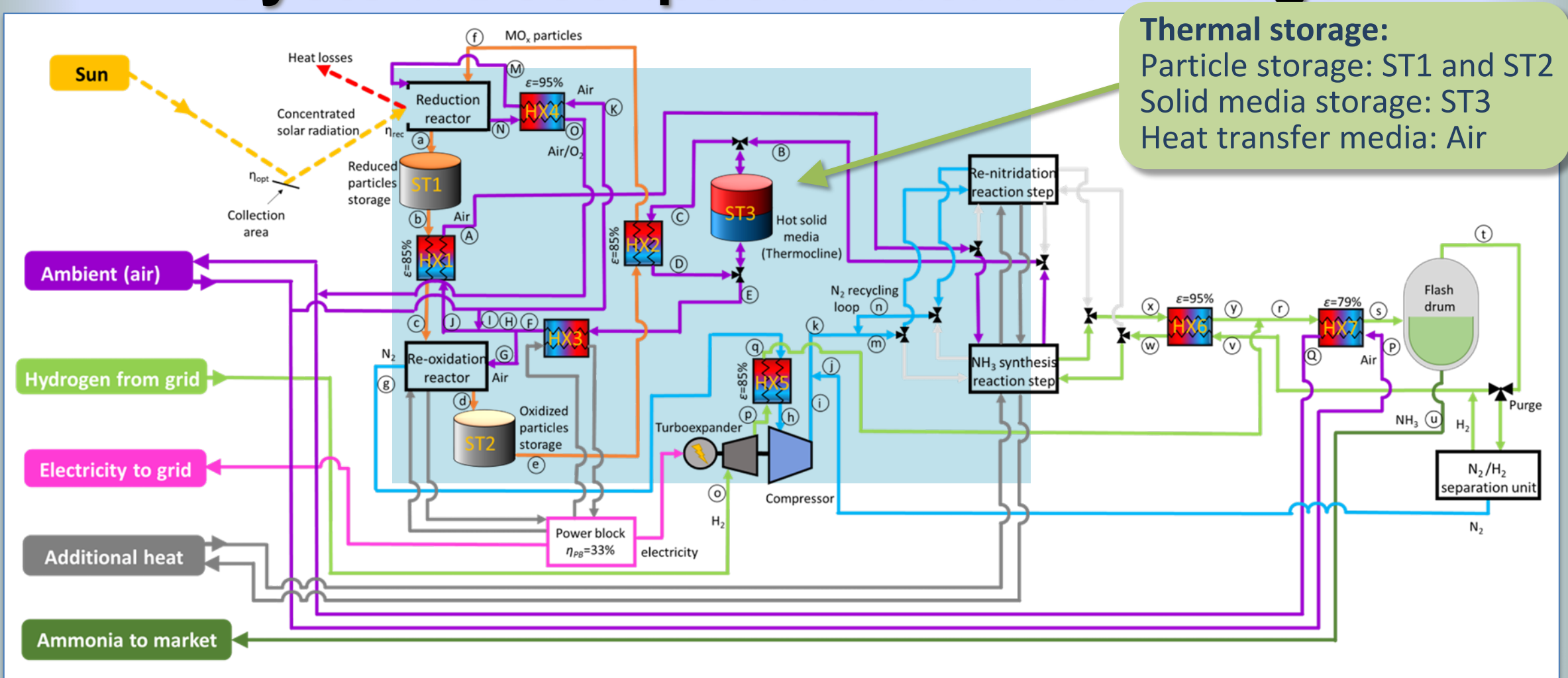


# System Description: Ammonia Synthesis and Re-nitridation





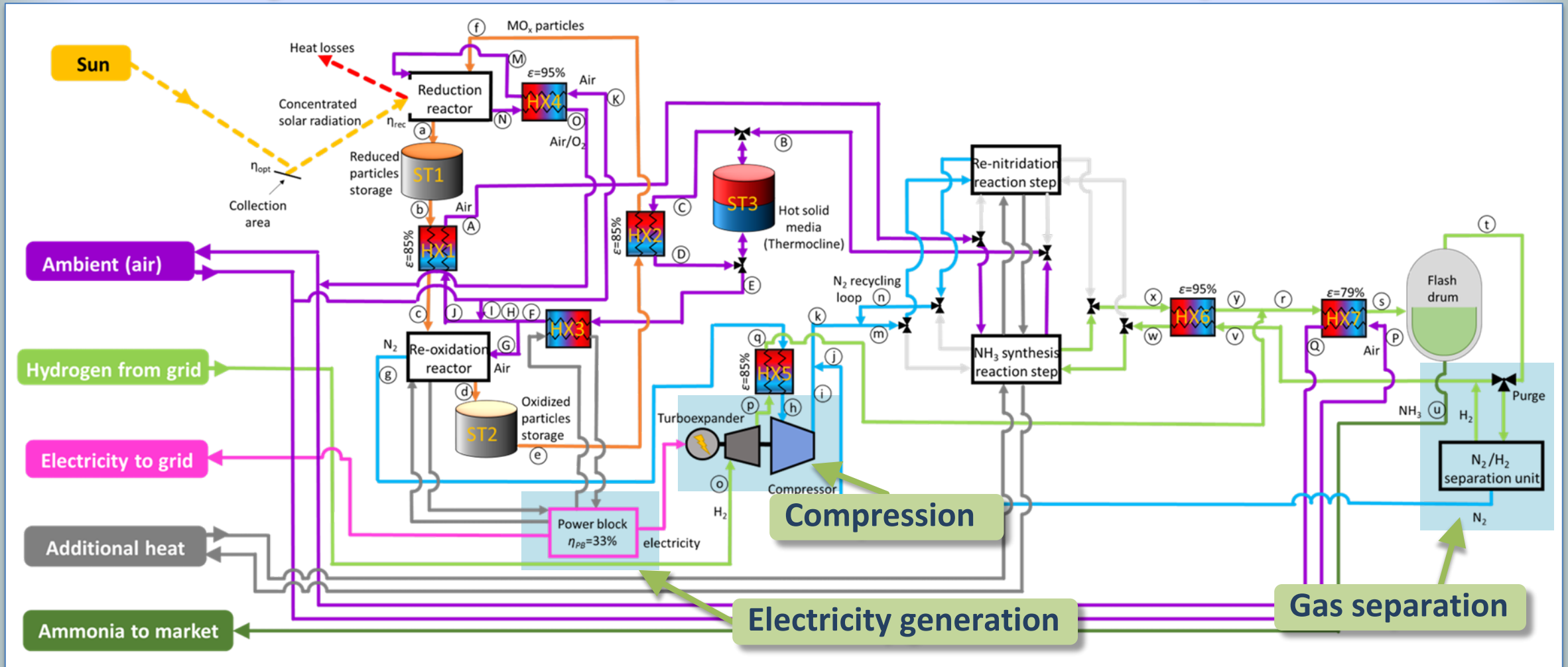
# System Description: Thermal storage







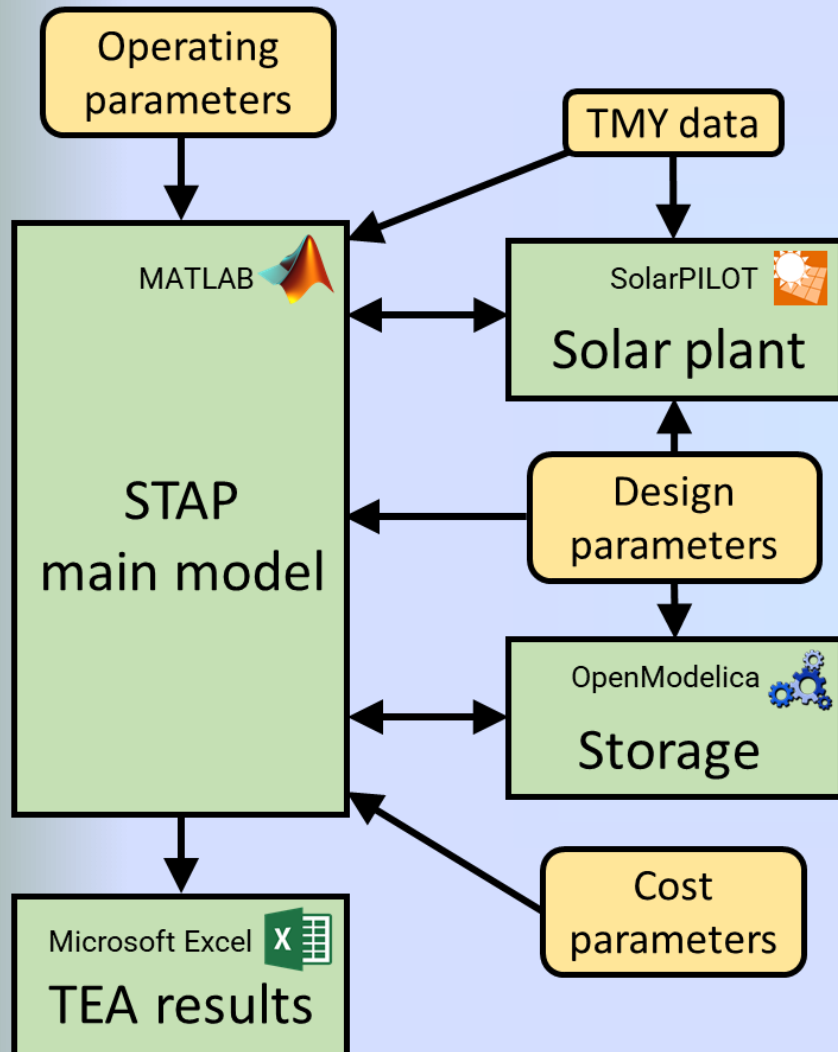
# System Description: Balance of plant





## Model design

Integration completed in a single Matlab script that communicates with other support software to perform the simulation.



- **SolarPILOT:** Design and simulation of solar field + receiver optical efficiency
- **Matlab:** System design, annual simulation, TEA, data processing and modeling integration
- **OpenModelica:** Fast annual simulation of the dynamic model (storage)
- **Microsoft Excel:** TEA result display



# Cost parameters

## Key parameters

- 30 MW<sub>th</sub> solar tower
- Three receivers
- Solar multiple 3
- 14 h storage
- Metal oxide: SrFeO<sub>3</sub>
- N<sub>2</sub> purity: 10 Pa, T<sub>OX</sub> = 500 °C
- Metal nitride: Co<sub>3</sub>Mo<sub>3</sub>N
- T<sub>AS</sub> = T<sub>RN</sub> = 700 °C
- NH<sub>3</sub> yield: 60%
- RN/AS cycle time: 120 min

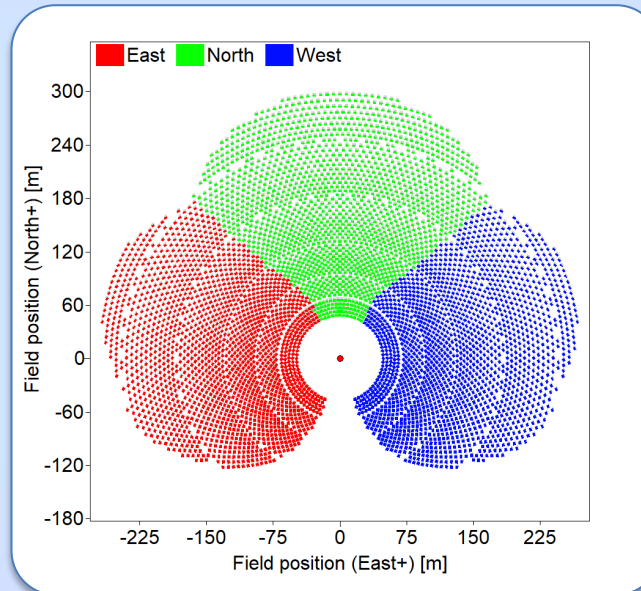


Table 1. Cost parameters

Parameter	Units	Value
Heliostat cost	\$/m <sup>2</sup>	75
Site preparation	\$/m <sup>2</sup>	10
Land cost	\$/m <sup>2</sup>	2.47
Power block	\$/kW <sub>e</sub>	1000
Additional heat	\$/MWh	15
Electricity	\$/MWh	50
MOx particle loss	%/y	10
EPC and owner cost	%	11
Contingency	%	7
O&M	%/y	2
Real discount rate	%	7
Lifetime	y	30





# Techno-economic analysis

Costs Calculation	Units	Value
Heliostat field	\$	3,975,900.50
Tower	\$	6,251,716.68
Receiver	\$	1,009,800.00
OX reactor	\$	336,600.00
Lift	\$	259,740.16
Storage tanks ST1 & ST2	\$	1,344,389.90
MO particles	\$	3,551,288.54
Storage tank ST3 and material	\$	834,750.51
Heat Exchangers	\$	1,550,681.99
Turboexpander	\$	283,220.00
Power Block	\$	2,758,295.81
Separation NH3	\$	107,307.00
Separation N2/H2	\$	-
AS & RN Reactors	\$	3,326,583.31
MN particles	\$	91,336,698.62
<b>Subtotal direct cost</b>	<b>\$</b>	<b>116,926,973.02</b>
Contingency	\$	8,184,888.11
<b>Total direct cost</b>	<b>\$</b>	<b>125,111,861.14</b>
Land cost	\$	1,007,571.02
EPC and owner cost	\$	13,762,304.72
<b>Total indirect cost</b>	<b>\$</b>	<b>14,769,875.74</b>
<b>Total CapEx</b>	<b>\$</b>	<b>139,881,736.88</b>
OpEx (fixed)	\$/y	2,797,634.74
Particle loss	\$/y	355,128.85
Additional heat	\$/y	-
OpEx (variable)	\$/y	355,128.85
<b>Total OpEx</b>	<b>\$/y</b>	<b>3,507,892.45</b>
<b>Total revenue</b>	<b>\$/y</b>	<b>474,058.12</b>
<b>LCOA w/o H2</b>	<b>\$/tonne</b>	<b>213.11</b>

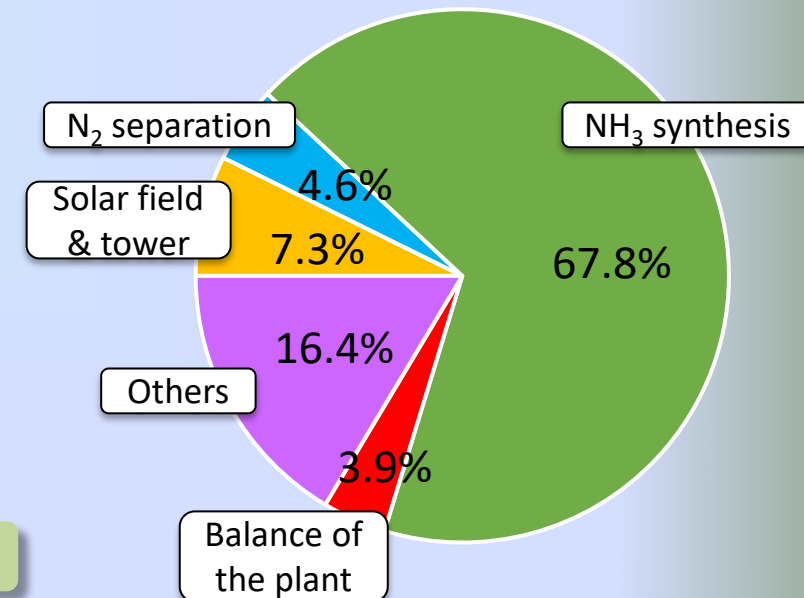
NH<sub>3</sub> yield and cycle time have a high impact on the total cost of the plant

The nitride cost is the most significant system expense, accounting for more than the 50% of the total CapEx

...but, it is also the most uncertain variable

LCOA without including the H<sub>2</sub>

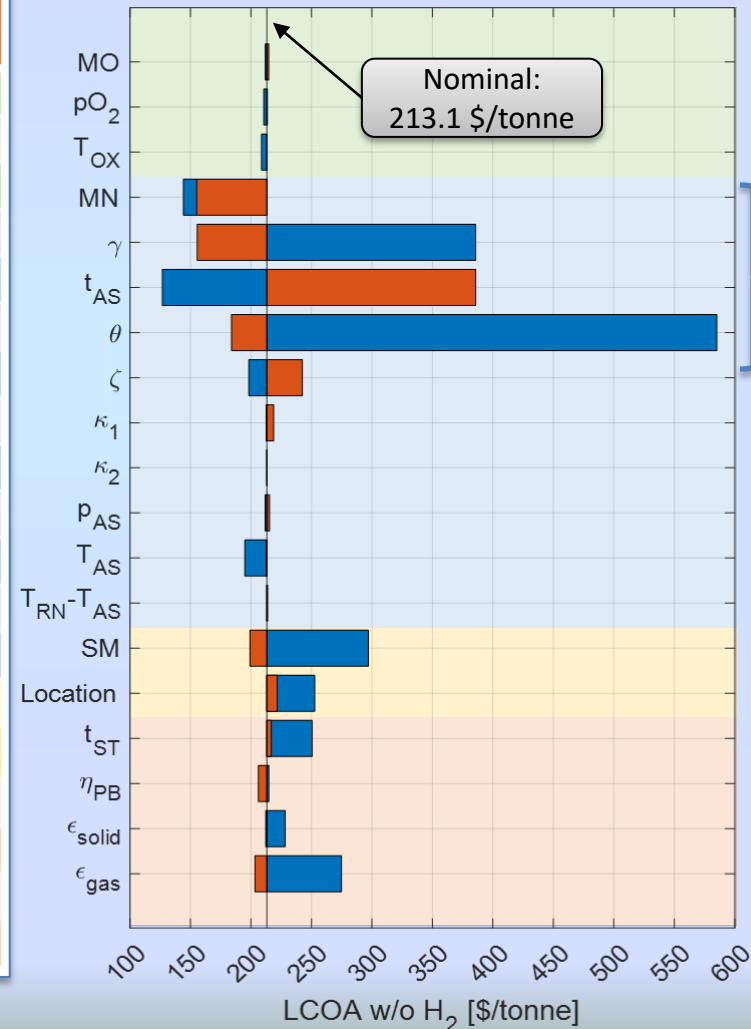
Capital expenses distribution





# Sensitivity analysis

	Variable	Nominal	Range (low)	Range (high)
<b>N<sub>2</sub> separation sub-system</b>	MO	SF	LSF15	BSF15
	pO <sub>2</sub> [Pa]	10	1	100
	T <sub>OX</sub> [°C]	500	450	600
<b>NH<sub>3</sub> synthesis sub-system</b>	MN	Co <sub>3</sub> Mo <sub>3</sub> N	Fe <sub>3</sub> Mo <sub>3</sub> N	Ni <sub>2</sub> Mo <sub>3</sub> N
	γ [-]	0.5	0.25	0.75
	t <sub>AS</sub> [h]	2	1	4
	θ [%]	60	20	70
	ζ [-]	10	0	50
	κ <sub>1</sub> [-]	1	0	20
	κ <sub>2</sub> [-]	5	0	20
	p <sub>AS</sub> [bar]	20	15	30
	T <sub>AS</sub> [°C]	700	550	750
	T <sub>RN</sub> - T <sub>AS</sub> [°C]	0	0	150
<b>Solar capture sub-system</b>	SM [-]	3	2	4
	Location	Daggett CA	Tonopah NV	Tucson AZ
<b>Auxiliary sub-system</b>	t <sub>ST</sub> [h]	14	8	24
	η <sub>PB</sub> [%]	33	25	65
	ε <sub>solid</sub> [%]	85	50	99
	ε <sub>gas</sub> [%]	95	50	99

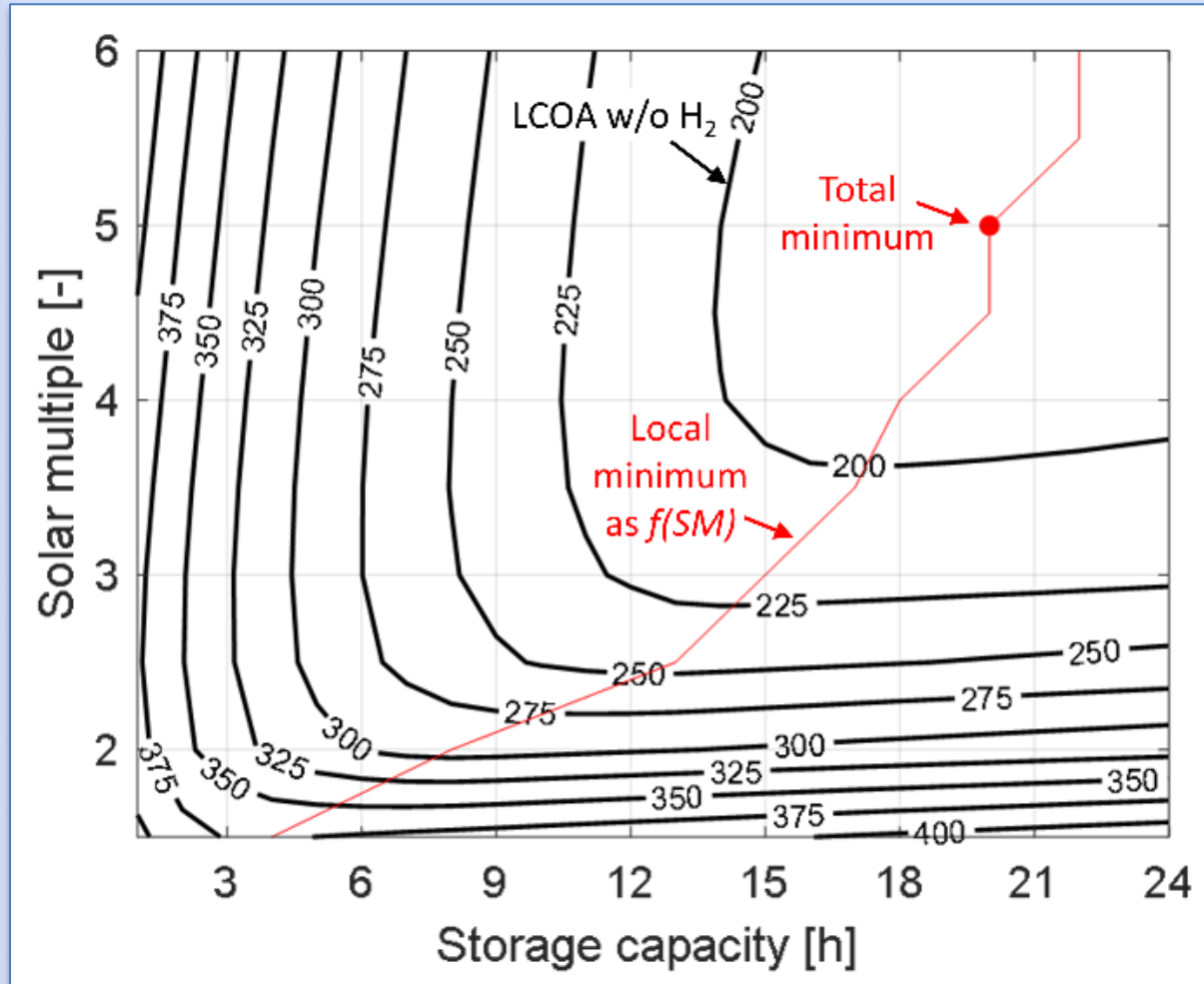


Quantity of metal nitride and its price dominate the variability in LCOA

Little room for improvement in other parts of the plant



# Solar multiple and storage capacity

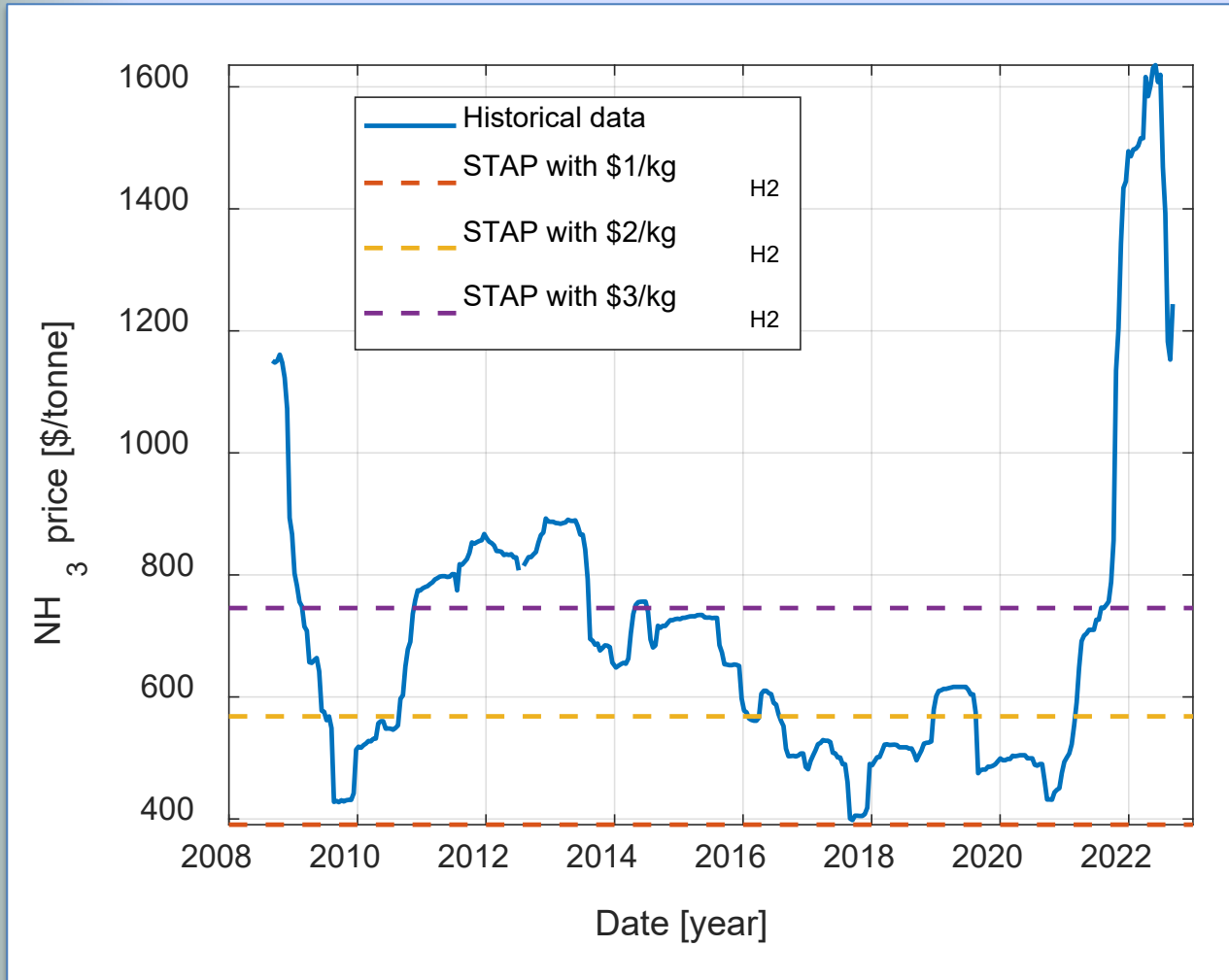


STAP plant benefits from a large solar multiple and large storage capacity





# Ammonia price context



Natural gas costs comprise ~ 70%–80% of HB production cost

Main drivers in the ammonia price in the US:

- record natural gas output
- a decline in domestic fertilizer production
- an increase in global demand for fertilizers
- an increase in the demand for fertilizers by

**STAP can offer price stability**

Data from:

[3] U.S. Department of Agriculture, Agricultural Marketing Service, Illinois Production Cost Report (Bi-weekly)



## Summary

- CO<sub>2</sub>-neutral ammonia production with concentrated solar technology is theoretically possible based on advanced solar thermochemical looping technology
- STAP offers price stability achieving a target price <250 \$/tonne NH<sub>3</sub> without including the H<sub>2</sub>
- The nitride cost is the most significant expense, accounting for more than the 50% of the total CapEx



# Thank you for your attention!!

We would like to acknowledge the team and institutions involved in this work



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TECHNOLOGIES OFFICE**  
U.S. Department Of Energy



**Sandia  
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If you like this work...

...there is more STAP in SolarPACES 2022:

- Wednesday 28 – 17:40: **Demonstration of a Solar Air Separation Process to Produce High-Purity N<sub>2</sub> via Ba<sub>0.15</sub>Sr<sub>0.85</sub>FeO<sub>3-δ</sub> Reduction/Oxidation Cycles.** Evan Bush, Matthew Kury, Kevin Albrecht, and Andrea Ambrosini
- Wednesday 28 – 17:00: **Solar-Thermal Ammonia Production: A Renewable, Carbon-Neutral Route to Ammonia via Concentrating Solar Thermochemistry.** Andrea Ambrosini, H. Evan Bush, Xiang Gao, Nhu "Ty" Nguyen, Alberto de la Calle, Ivan Ermanoski, Tyler Farr, Kevin Albrecht, Matthew W. Kury, Peter Loutzenhiser, and Ellen B. Stechel
- Friday 30 - 9:10: **Investigation of Co<sub>3</sub>Mo<sub>3</sub>N reduction/re-nitridation extents as a function of temperature and N<sub>2</sub> partial pressure for solar thermochemical NH<sub>3</sub> production.** Nhu Pales Nguyen, Shaspreet Kaur, H. Evan Bush, Xiang Gao, James E. Miller, Andrea Ambrosini, Peter G. Loutzenhiser
- Friday 30 - 9:50: **Solar Ammonia Production via Novel Two-step Thermochemical Looping of a Co<sub>3</sub>Mo<sub>3</sub>N/Co<sub>6</sub>Mo<sub>6</sub>N pair.** Xiang Gao, Ivan Ermanoski, Alberto de la Calle, Andrea Ambrosini, and Ellen B. Stechel