



LightWorks®

Techno-Economic Analysis of Solar-Thermal Ammonia Production

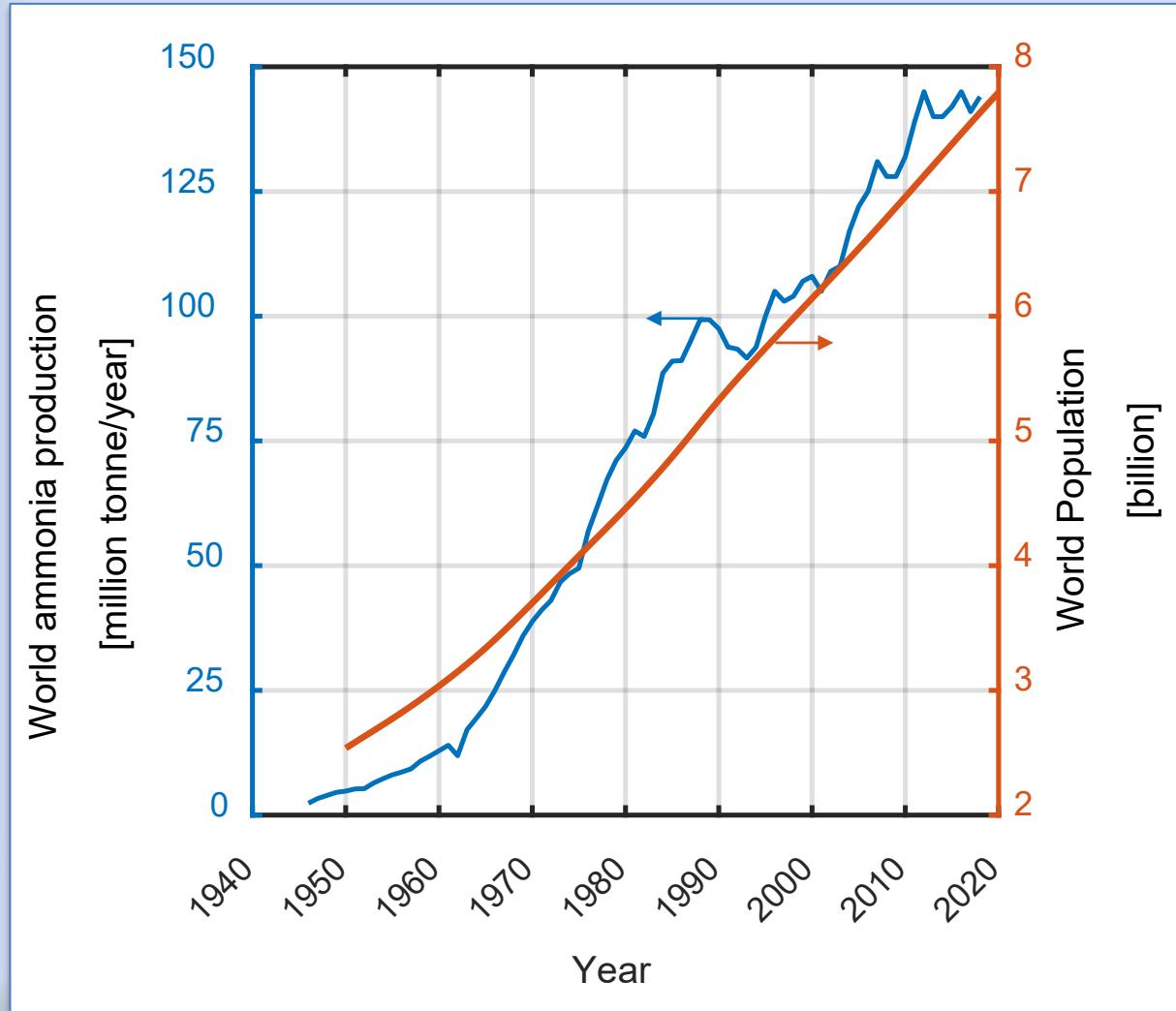
**Alberto de la Calle¹, H. Evan Bush², Ivan Ermanoski¹,
Andrea Ambrosini² and Ellen B. Stechel¹**

¹**ASU LightWorks®, Arizona State University, Tempe AZ, USA**

²**Sandia National Laboratories, Albuquerque NM, USA**



Ammonia Production



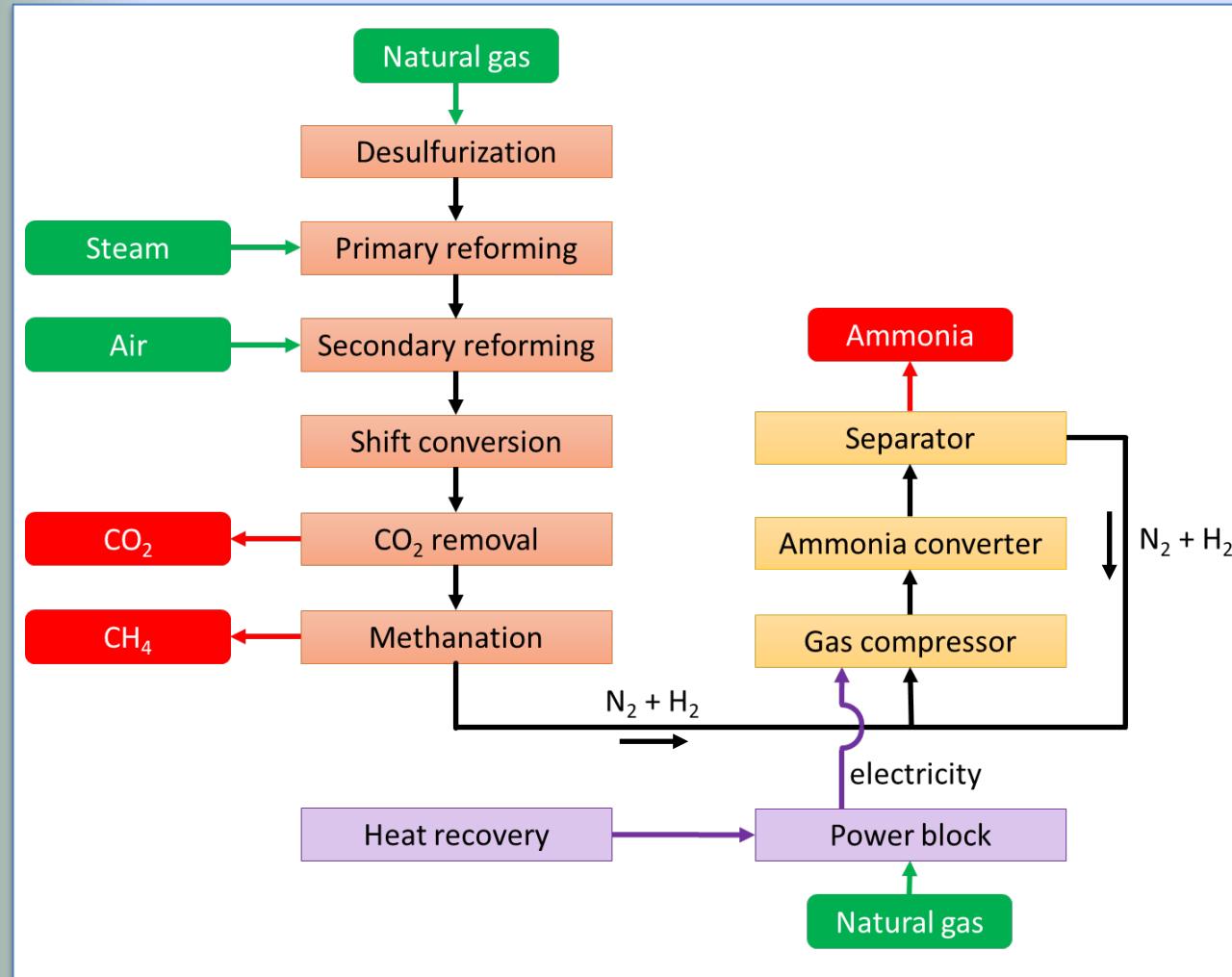
- Ammonia is the second most produced industrial chemical in the world
- About 80% of the nitrogen in our body comes from synthetic nitrogen fertilizer
- Responsible for over **1.4%** of global CO₂ emissions

Data from:

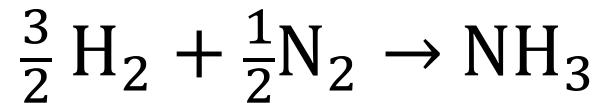
- [1] U.S. Geological Survey, National Minerals Information Center
- [2] United Nations, Department of Economic and Social Affairs



The Haber-Bosch Process



Ammonia synthesis:



Exothermic reaction: $\Delta H_{298\text{K}} = -45 \text{ kJ/mol}$

High pressure: 150-300 bar

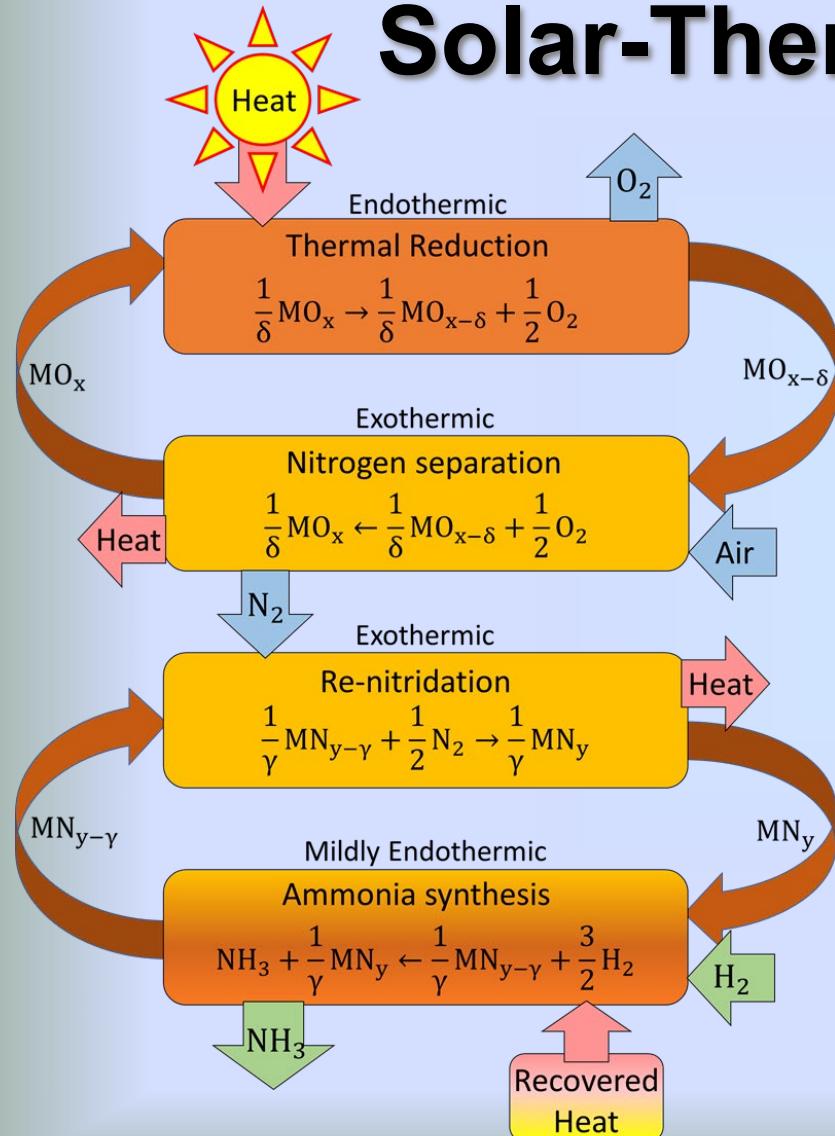
Moderately elevated temperature: 350-500 °C

Low conversion: ~10-20%

- H₂ is produced via steam methane reforming
- N₂ is produced via methane combustion in air



Solar-Thermal Ammonia Production

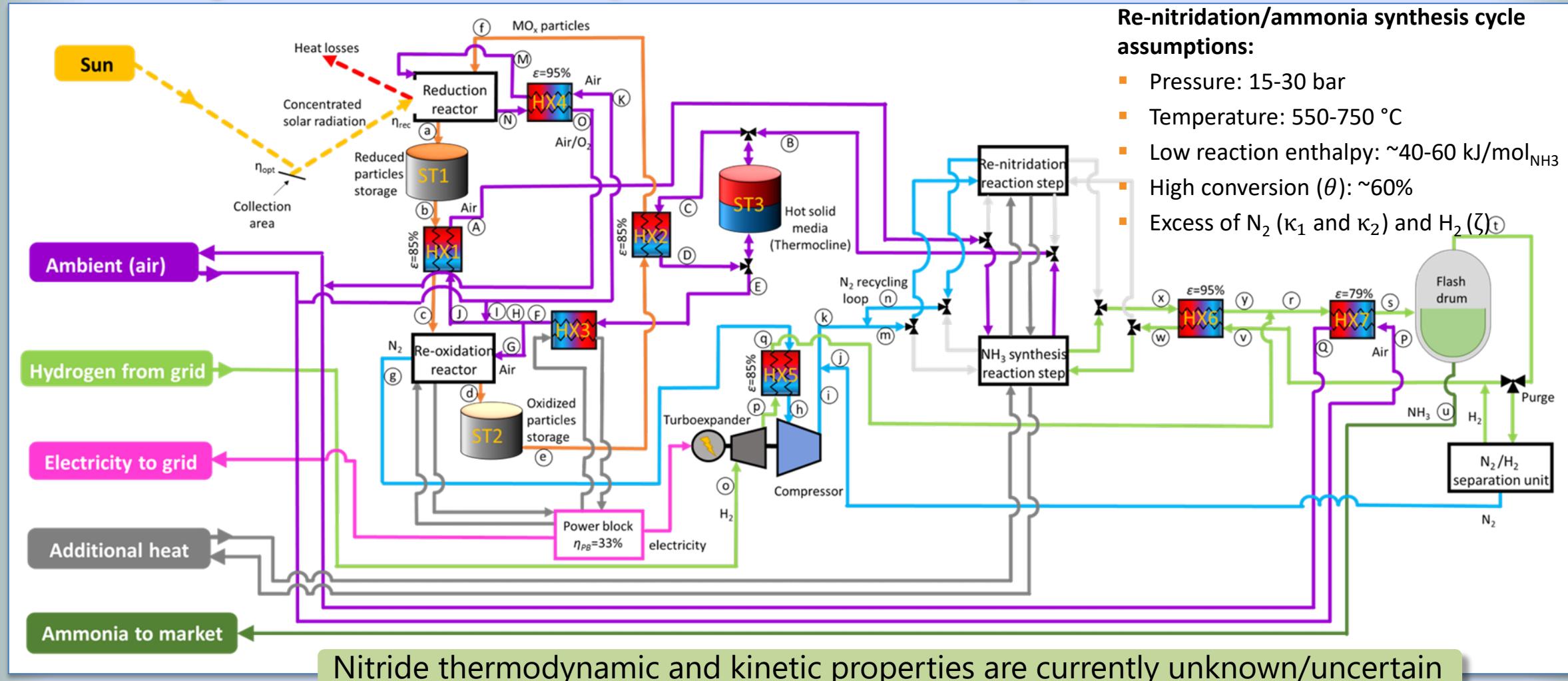


Goal: CO₂-neutral ammonia production with concentrated solar technology

- **First stage:** nitrogen production from air using concentrated solar heat
- **Second stage:** ammonia production targeting pressures lower than HB by an order of magnitude

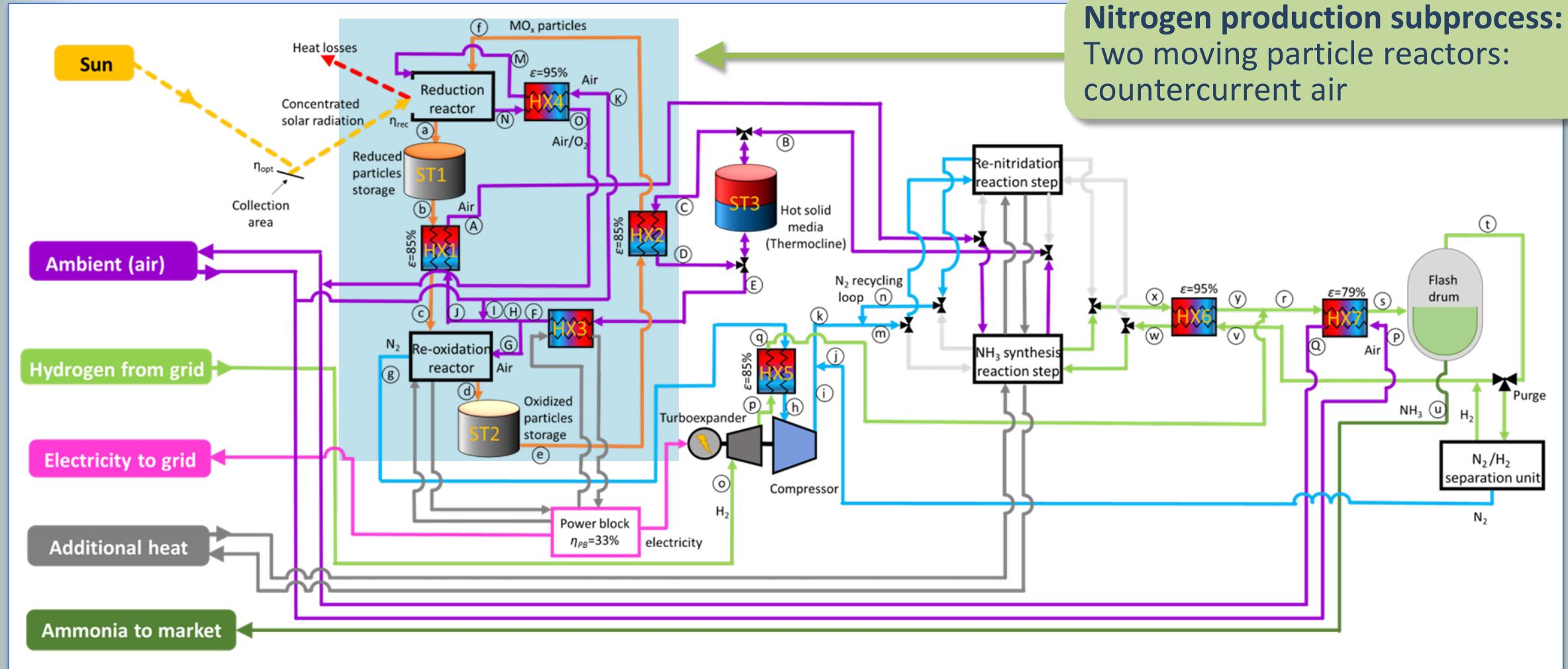


System Description: Assumptions



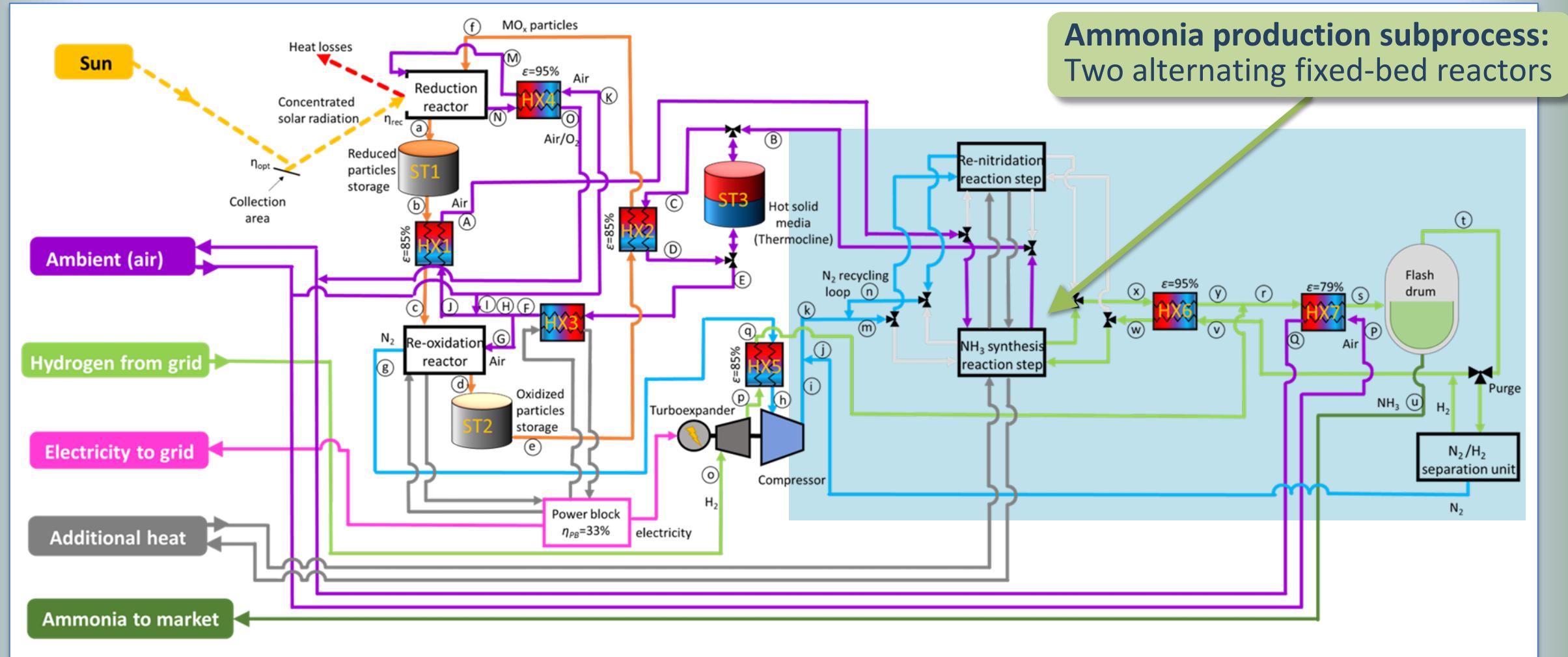


System Description: Thermal Reduction and Air Separation



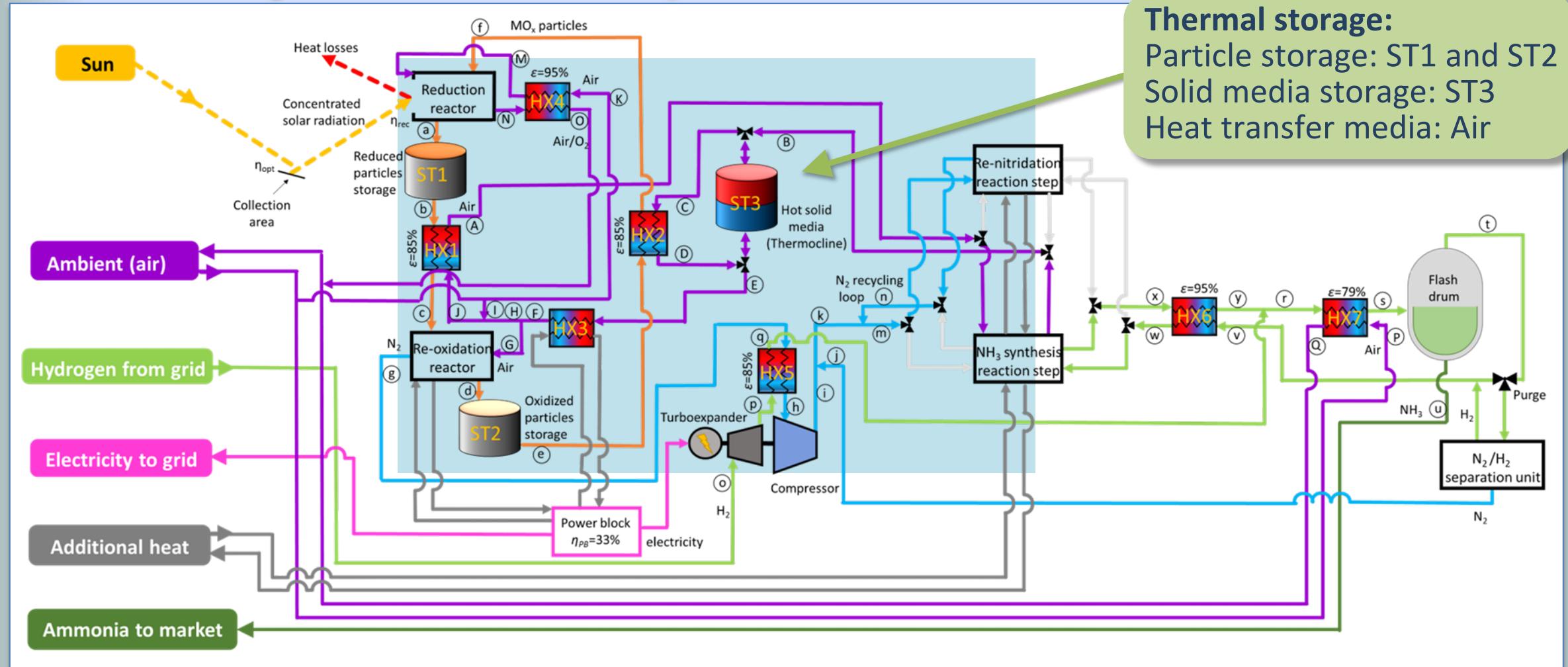


System Description: Ammonia Synthesis and Re-nitridation



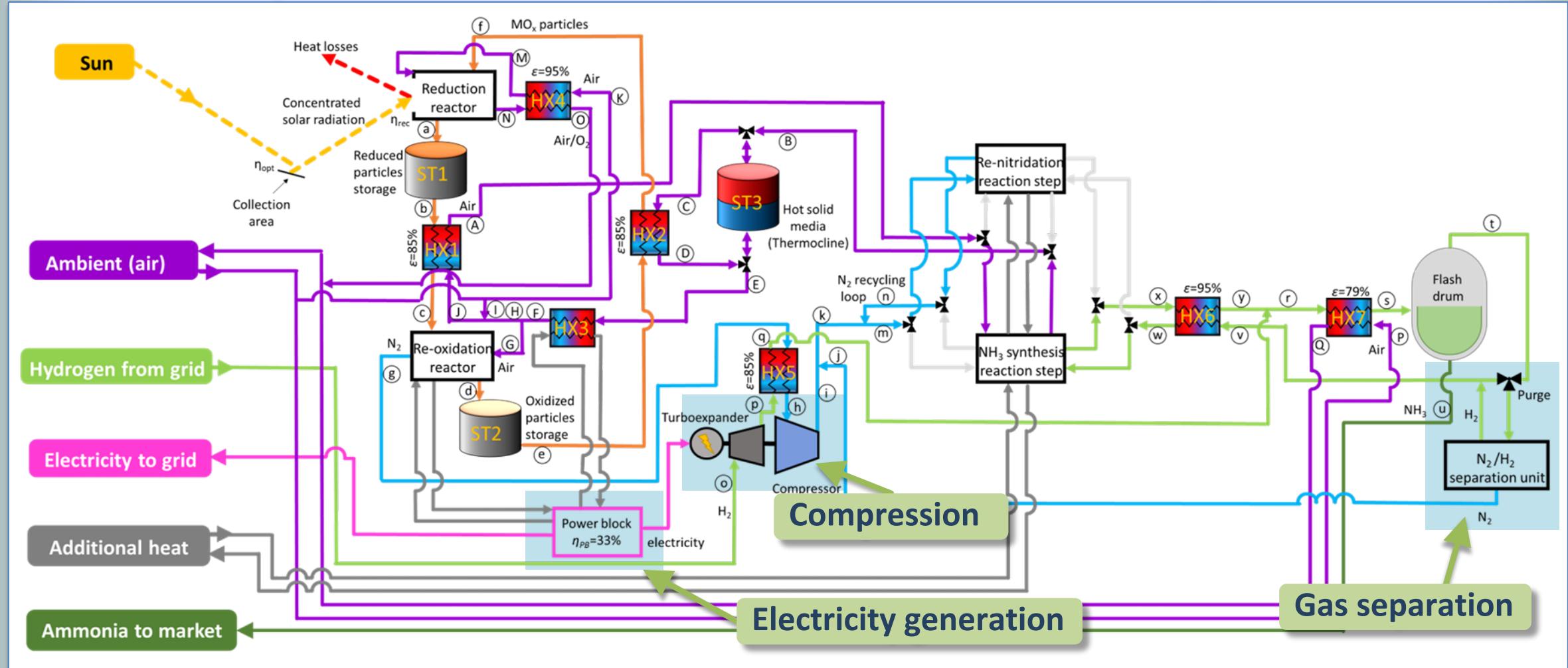


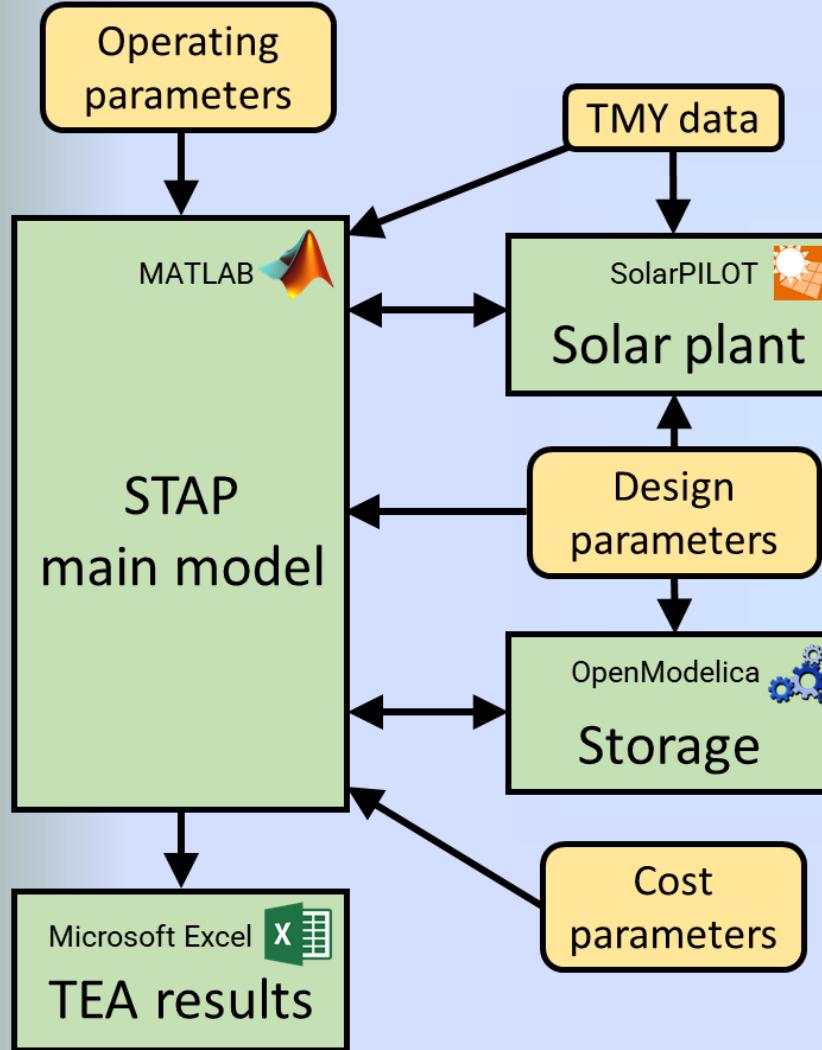
System Description: Thermal storage





System Description: Balance of plant





Model design

Integration completed in a single Matlab script that communicates with other support software to perform the simulation.

- **SolarPILOT:** Design and simulation of solar field + receiver optical efficiency
- **Matlab:** System design, annual simulation, TEA, data processing and modeling integration
- **OpenModelica:** Fast annual simulation of the dynamic model (storage)
- **Microsoft Excel:** TEA result display



Cost parameters

Key parameters

- 30 MW_{th} solar tower
- Three receivers
- Solar multiple 3
- 14 h storage
- Metal oxide: SrFeO₃
- N₂ purity: 10 Pa, T_{OX}= 500 °C
- Metal nitride: Co₃Mo₃N
- T_{AS}=T_{RN}=700 °C
- NH₃ yield: 60%
- RN/AS cycle time: 120 min

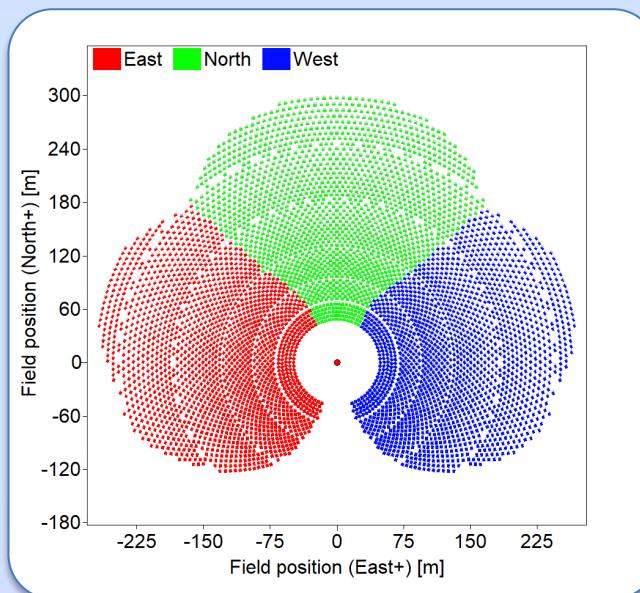


Table 1. Cost parameters

Parameter	Units	Value
Heliostat cost	\$/m ²	75
Site preparation	\$/m ²	10
Land cost	\$/m ²	2.47
Power block	\$/kW _e	1000
Additional heat	\$/MWh	15
Electricity	\$/MWh	50
MOx particle loss	%/y	10
EPC and owner cost	%	11
Contingency	%	7
O&M	%/y	2
Real discount rate	%	7
Lifetime	y	30



Techno-economic analysis

Costs Calculation	Units	Value
Heliosstat field	\$	3,975,900.50
Tower	\$	6,251,716.68
Receiver	\$	1,009,800.00
OX reactor	\$	336,600.00
Lift	\$	259,740.16
Storage tanks ST1 & ST2	\$	1,344,389.90
MO particles	\$	3,551,288.54
Storage tank ST3 and material	\$	834,750.51
Heat Exchangers	\$	1,550,681.99
Turboexpander	\$	283,220.00
Power Block	\$	2,758,295.81
Separation NH3	\$	107,307.00
Separation N2/H2	\$	-
AS & RN Reactors	\$	3,326,583.31
MN particles	\$	91,336,698.62
Subtotal direct cost	\$	116,926,973.02
Contingency	\$	8,184,888.11
Total direct cost	\$	125,111,861.14
Land cost	\$	1,007,571.02
EPC and owner cost	\$	13,762,304.72
Total indirect cost	\$	14,769,875.74
Total CapEx	\$	139,881,736.88
OpEx (fixed)	\$/y	2,797,634.74
Particle loss	\$/y	355,128.85
Additional heat	\$/y	-
OpEx (variable)	\$/y	355,128.85
Total OpEx	\$/y	3,507,892.45
Total revenue	\$/y	474,058.12
LCOA w/o H₂	\$/tonne	213.11

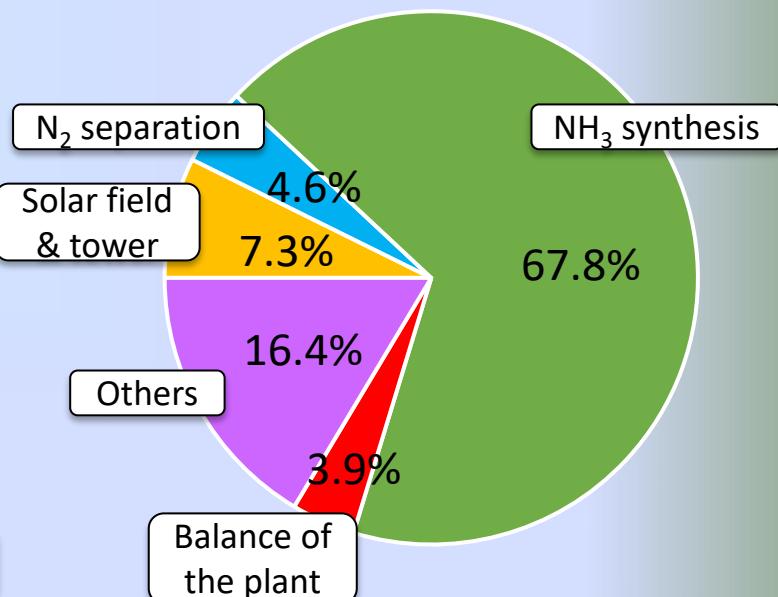
NH₃ yield and cycle time have a high impact on the total cost of the plant

The nitride cost is the most significant system expense, accounting for more than the 50% of the total CapEx

...but, it is also the most uncertain variable

LCOA without including the H₂

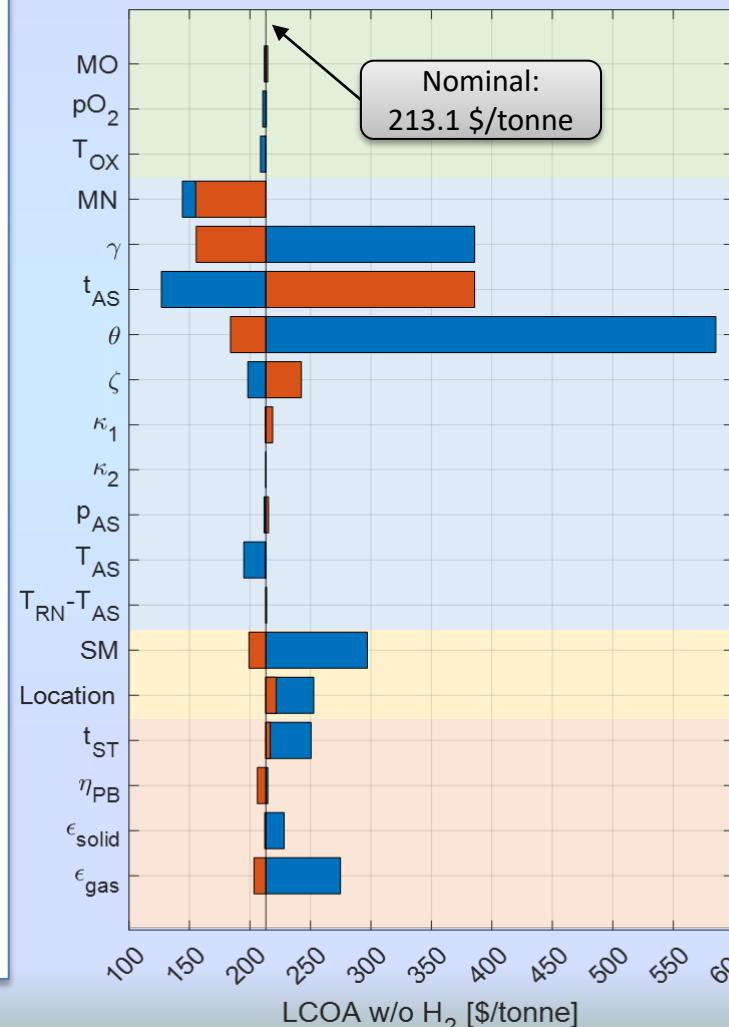
Capital expenses distribution





Sensitivity analysis

	Variable	Nominal	Range (low)	Range (high)
N ₂ separation sub-system	MO	SF	LSF15	BSF15
	pO ₂ [Pa]	10	1	100
	T _{ox} [°C]	500	450	600
NH ₃ synthesis sub-system	MN	Co ₃ Mo ₃ N	Fe ₃ Mo ₃ N	Ni ₂ Mo ₃ N
	γ [-]	0.5	0.25	0.75
	t _{AS} [h]	2	1	4
	θ [%]	60	20	70
	ζ [-]	10	0	50
	κ ₁ [-]	1	0	20
	κ ₂ [-]	5	0	20
	p _{AS} [bar]	20	15	30
	T _{AS} [°C]	700	550	750
	T _{RN} - T _{AS} [°C]	0	0	150
Solar capture sub-system	SM [-]	3	2	4
	Location	Daggett CA	Tonopah NV	Tucson AZ
Auxiliary sub-system	t _{ST} [h]	14	8	24
	η _{PB} [%]	33	25	65
	ε _{solid} [%]	85	50	99
	ε _{gas} [%]	95	50	99

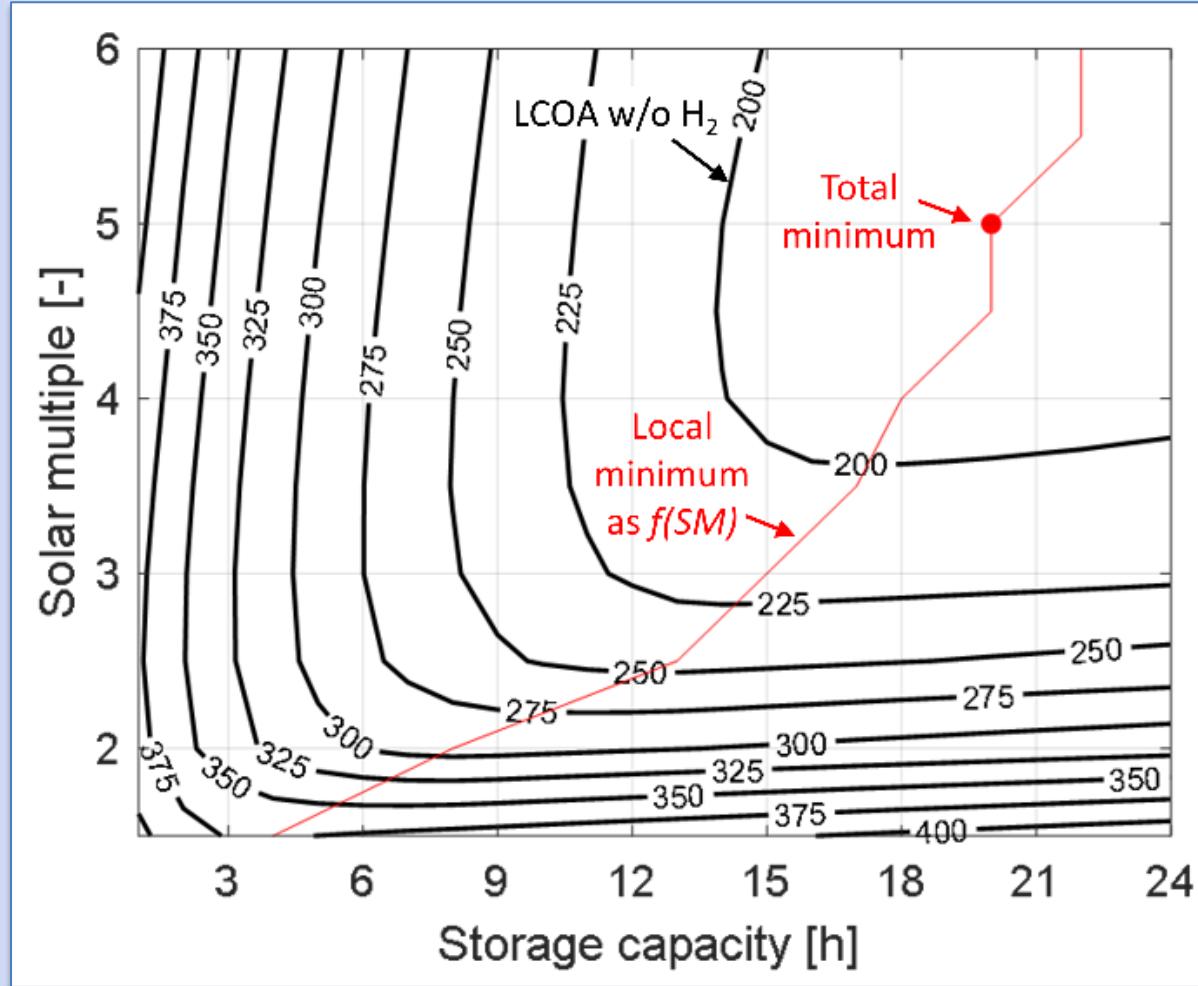


Quantity of metal nitride and its price dominate the variability in LCOA

Little room for improvement in other parts of the plant



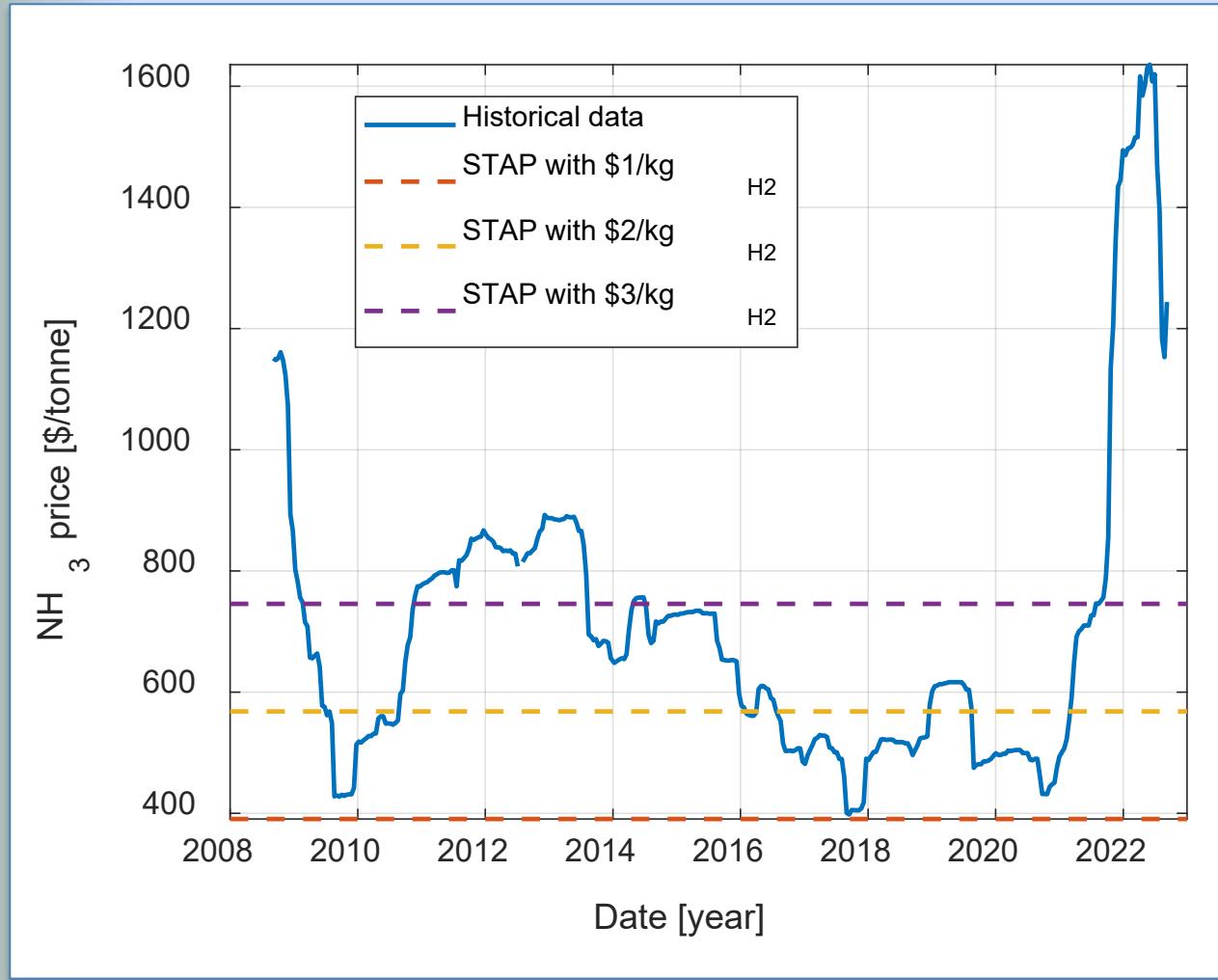
Solar multiple and storage capacity



STAP plant benefits from a large solar multiple and large storage capacity



Ammonia price context



Natural gas costs comprise ~ 70%–80% of HB production cost

Main drivers in the ammonia price in the US:

- record natural gas output
- a decline in domestic fertilizer production
- an increase in global demand for fertilizers
- an increase in the demand for fertilizers by

STAP can offer price stability

Data from:

[3] U.S. Department of Agriculture, Agricultural Marketing Service, Illinois Production Cost Report (Bi-weekly)



Summary

- CO₂-neutral ammonia production with concentrated solar technology is theoretically possible based on advanced solar thermochemical looping technology
- STAP offers price stability achieving a target price <250 \$/tonne NH₃ without including the H₂
- The nitride cost is the most significant expense, accounting for more than the 50% of the total CapEx



Thank you for your attention!!

We would like to acknowledge the team and institutions involved in this work



**SOLAR ENERGY
TECHNOLOGIES OFFICE**
U.S. Department Of Energy



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If you like this work...

...there is more STAP in SolarPACES 2022:

- Wednesday 28 – 17:40: Demonstration of a Solar Air Separation Process to Produce High-Purity N₂ via Ba_{0.15}Sr_{0.85}FeO_{3-δ} Reduction/Oxidation Cycles. Evan Bush, Matthew Kury, Kevin Albrecht, and Andrea Ambrosini
- Wednesday 28 – 17:00: Solar-Thermal Ammonia Production: A Renewable, Carbon-Neutral Route to Ammonia via Concentrating Solar Thermochemistry. Andrea Ambrosini, H. Evan Bush, Xiang Gao, Nhu "Ty" Nguyen, Alberto de la Calle, Ivan Ermanoski, Tyler Farr, Kevin Albrecht, Matthew W. Kury, Peter Loutzenhiser, and Ellen B. Stechel
- Friday 30 - 9:10: Investigation of Co₃Mo₃N reduction/re-nitridation extents as a function of temperature and N₂ partial pressure for solar thermochemical NH₃ production. Nhu Pailes Nguyen, Shaspreet Kaur, H. Evan Bush, Xiang Gao, James E. Miller, Andrea Ambrosini , Peter G. Loutzenhiser
- Friday 30 - 9:50: Solar Ammonia Production via Novel Two-step Thermochemical Looping of a Co₃Mo₃N/Co₆Mo₆N pair. Xiang Gao, Ivan Ermanoski, Alberto de la Calle, Andrea Ambrosini, and Ellen B. Stechel