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# Imaging of Hydrogen Peroxide and Methyl in Nanosecond Pulsed Plasmas by Photofragmentation Laser-Induced Fluorescence

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# Need a technique for spatially and temporally resolved imaging measurements of hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) and methyl ( $\text{CH}_3$ )



Key reactive species in surface decontamination, biomedical treatments, combustion, catalysis

- Single-point and line-of-sight measurements
  - Molecular beam mass spectrometry (MBMS)
  - Absorption
  - Degenerate four-wave mixing (DFWM)
  - Resonance-enhanced multi-photon ionization (REMPI)

How to image hydrogen peroxide and methyl?

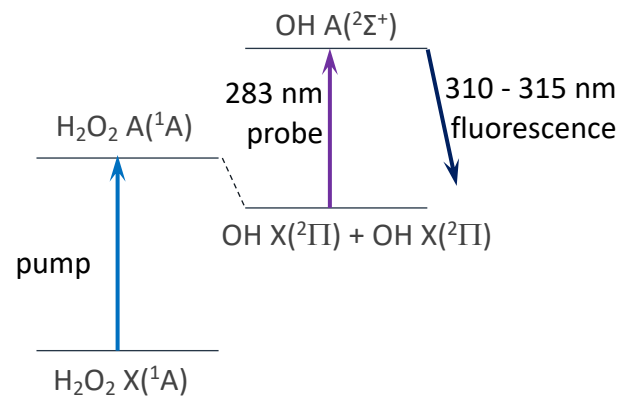
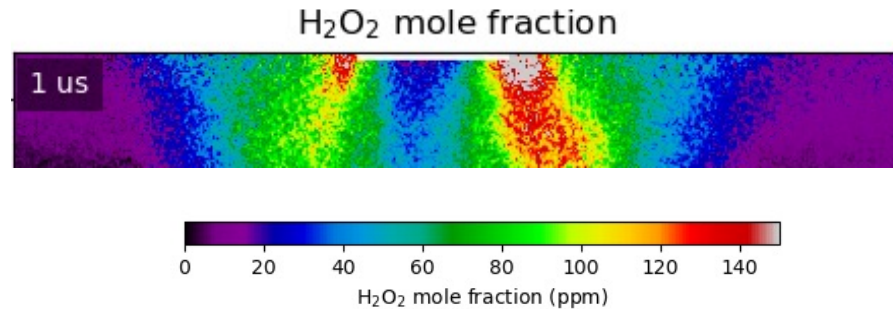
- $\text{H}_2\text{O}_2$  and  $\text{CH}_3$  not directly detectable with laser-induced fluorescence (LIF) due to strongly predissociative excited states
- Detect OH product of photodissociation with OH-LIF:  $\text{H}_2\text{O}_2 + h\nu \rightarrow 2\text{OH}$
- Detect CH product of photodissociation with CH-LIF:  $\text{CH}_3 + h\nu \rightarrow \text{CH} + \text{H}_2$
- Previously demonstrated in combustion [Li et al. Proc. Combust. Inst. 34, 3573 (2013); Li et al. Proc. Combust. Inst. 37, 4487 (2017)]

# Photofragmentation Laser-Induced Fluorescence Enables Detection of Key Species that Don't Fluoresce



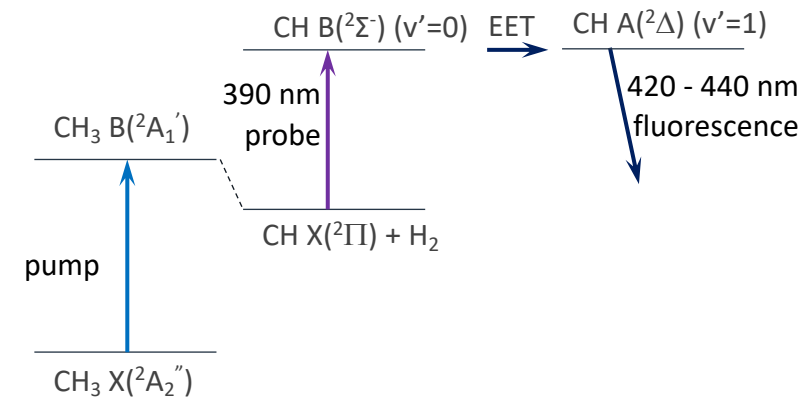
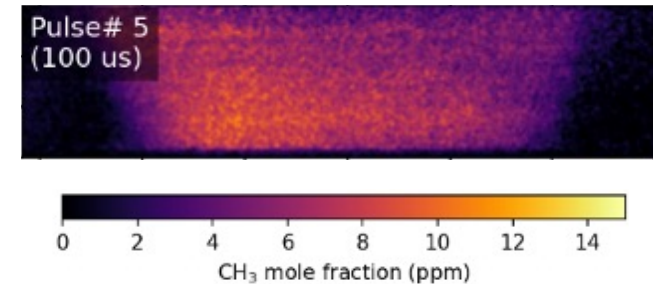
## Hydrogen peroxide ( $\text{H}_2\text{O}_2$ )

Key species in interactions of plasma with water and tissue: decontamination/sterilization, plasma medicine



## Methyl ( $\text{CH}_3$ )

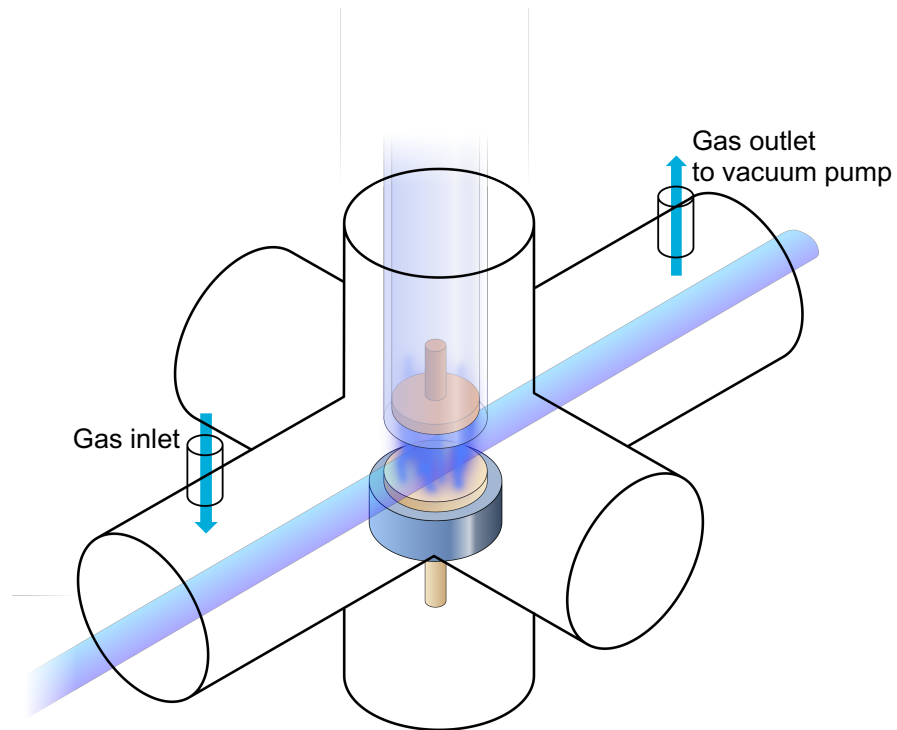
Key intermediate for plasma-assisted catalysis, reforming, combustion



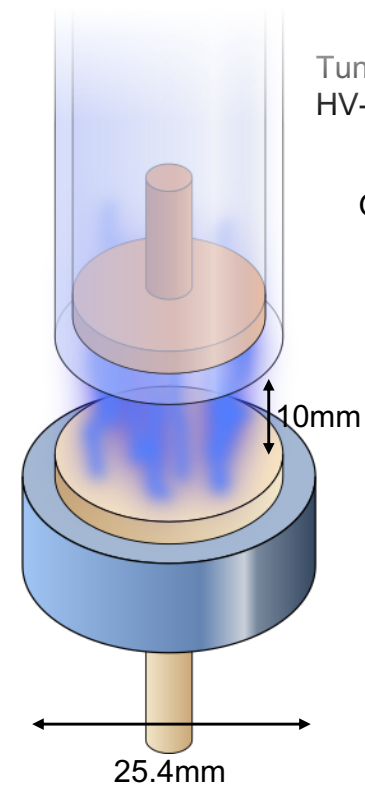
# Configuration for 2D in-situ imaging of plasmas and plasma-surface interactions



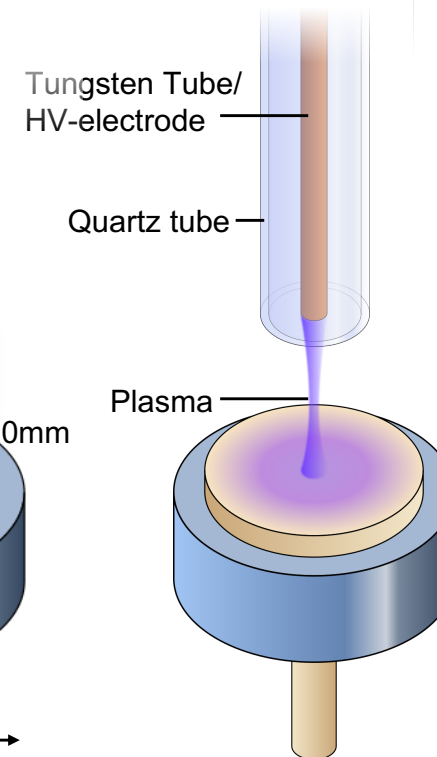
*Our design:* Direct plasma-catalyst interaction with optical access for in situ laser diagnostics



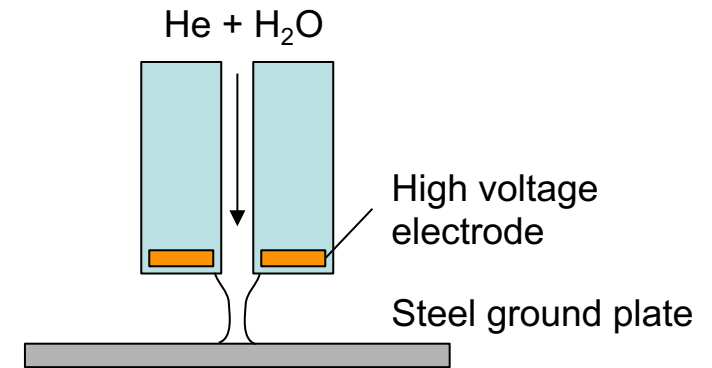
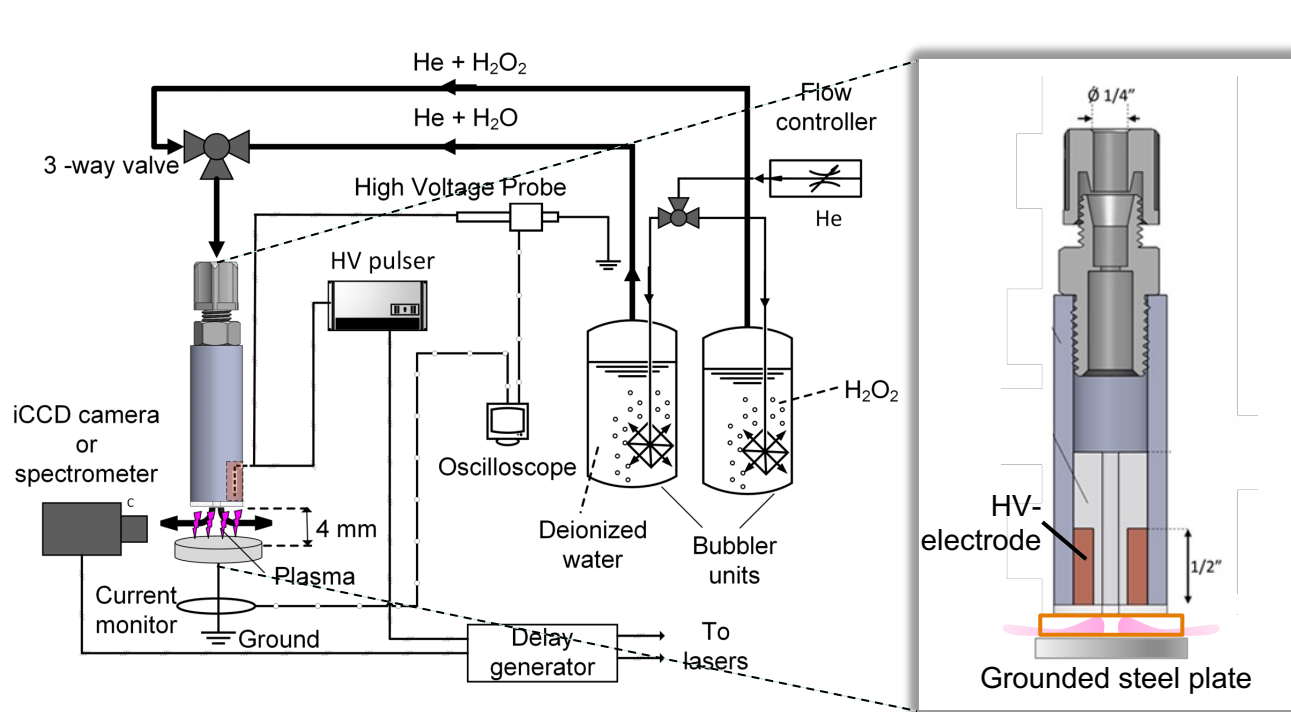
Plane-to-plane DBD Plasma



Plasma Jet



# Quantification of Reactive Species and Characterization of High-Water Content in Dielectric Barrier Discharge using Photofragmentation and Laser-Induced Fluorescence



- Water saturated helium jet
- H<sub>2</sub>O<sub>2</sub> PF-LIF ( $\text{H}_2\text{O}_2 + h\nu_1 \rightarrow 2\text{OH}$ )
- OH-LIF for measuring photolytically generated and plasma-generated OH

Measurements of key reactive species (OH, H<sub>2</sub>O<sub>2</sub>) needed for insights into properties of water-laden plasmas in decontamination, biomedical treatments

# Experimental Configuration for H<sub>2</sub>O<sub>2</sub> PF-LIF Imaging



## PF-LIF lasers:

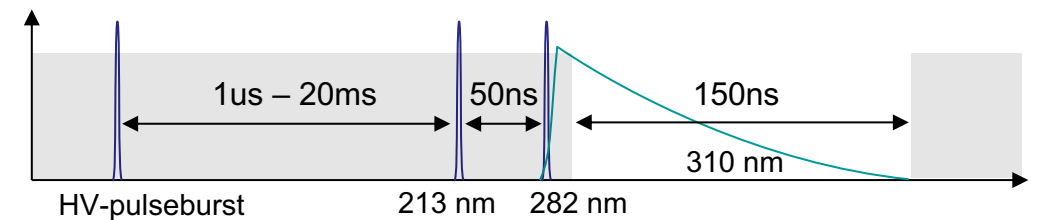
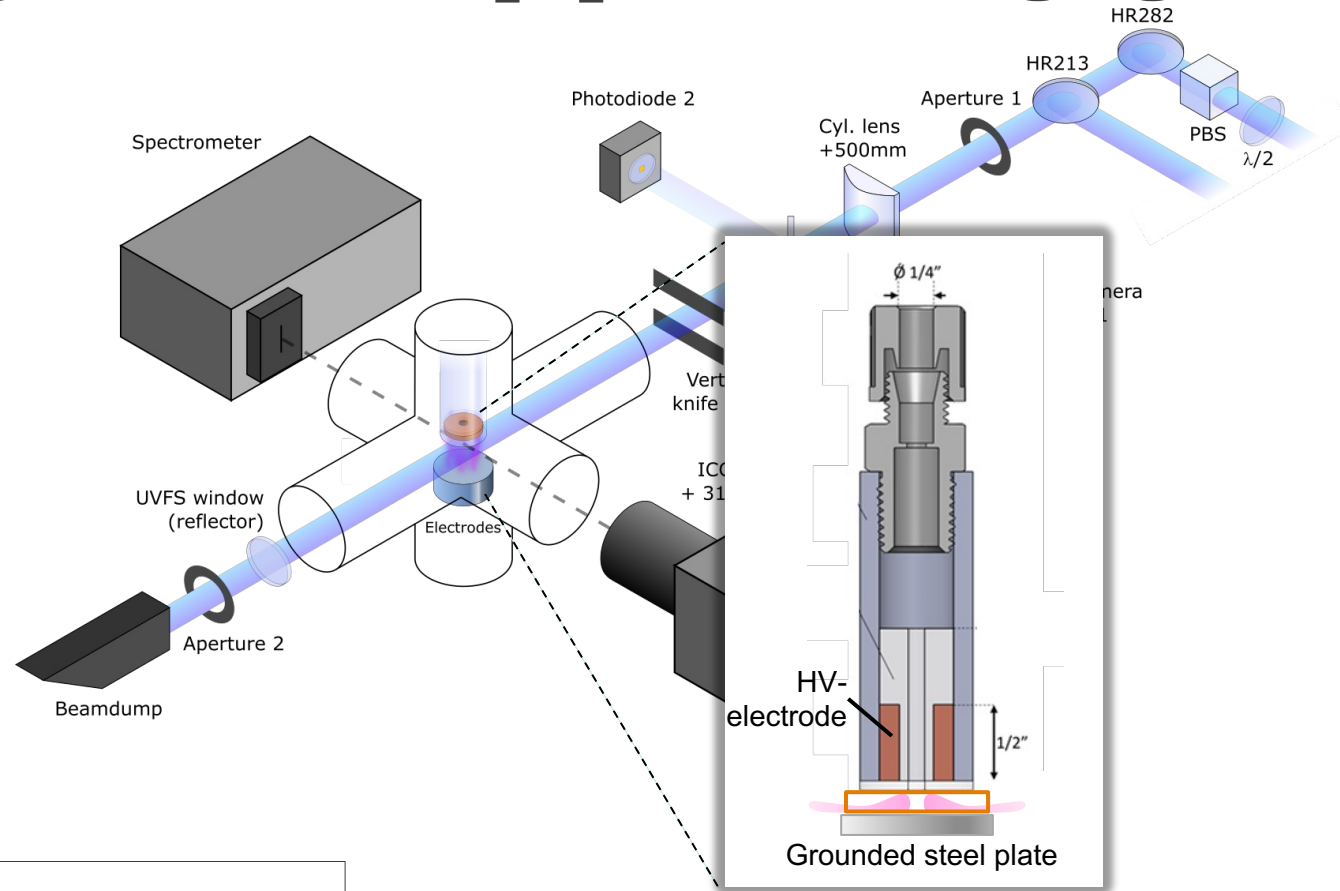
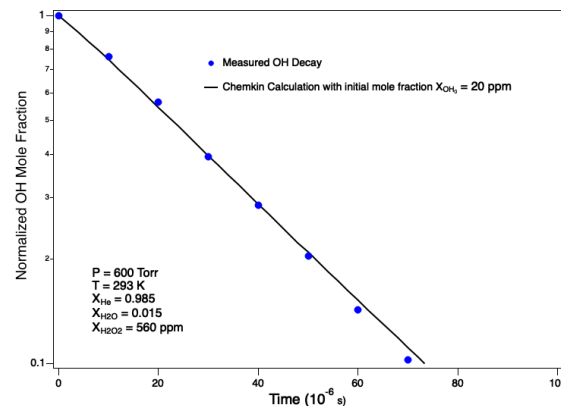
- PF: 213 nm (5<sup>th</sup> HG YAG)
- LIF: 282 nm (2<sup>nd</sup> HG 564nm dye)

## Conditions:

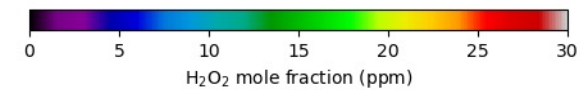
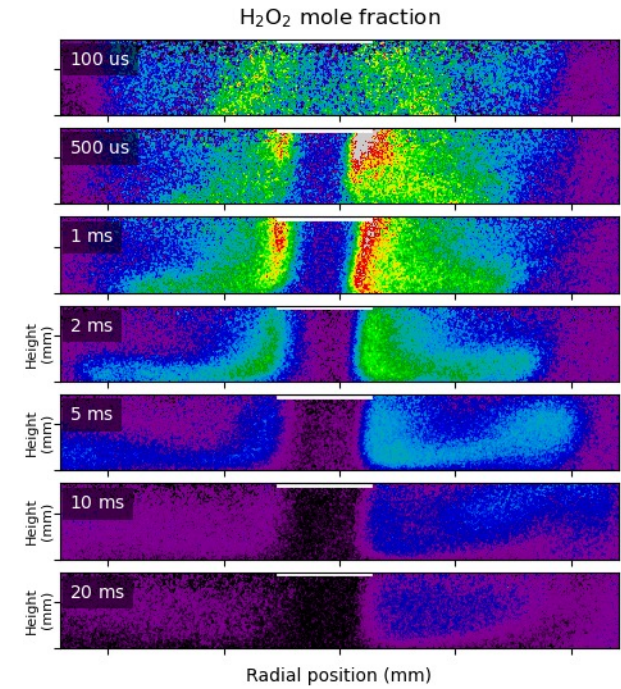
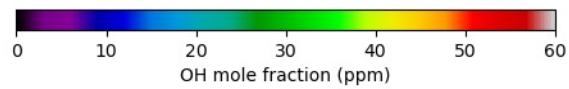
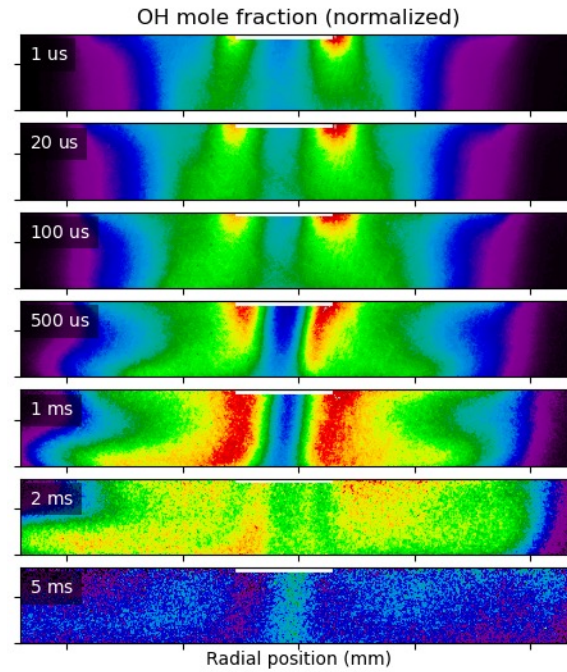
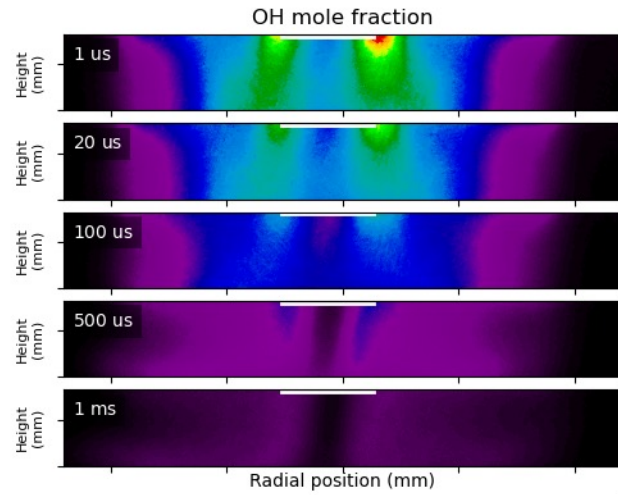
- P = 600 torr
- 4 slm He through H<sub>2</sub>O bubbler (10°C)
- 10kV, 5 pulse burst 50kHz/50Hz

## Calibration:

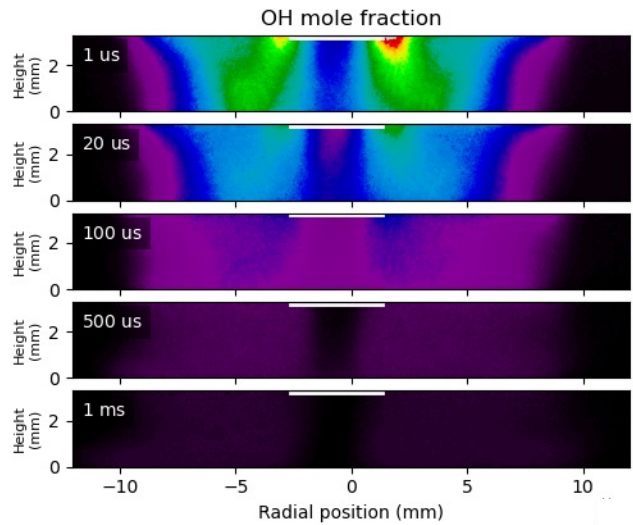
- Reference mixture H<sub>2</sub>O<sub>2</sub>/H<sub>2</sub>O/He
- Reactive decay rate of OH compared to Chemkin simulations



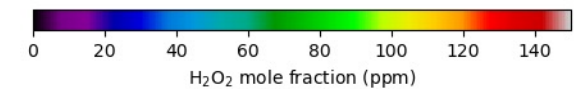
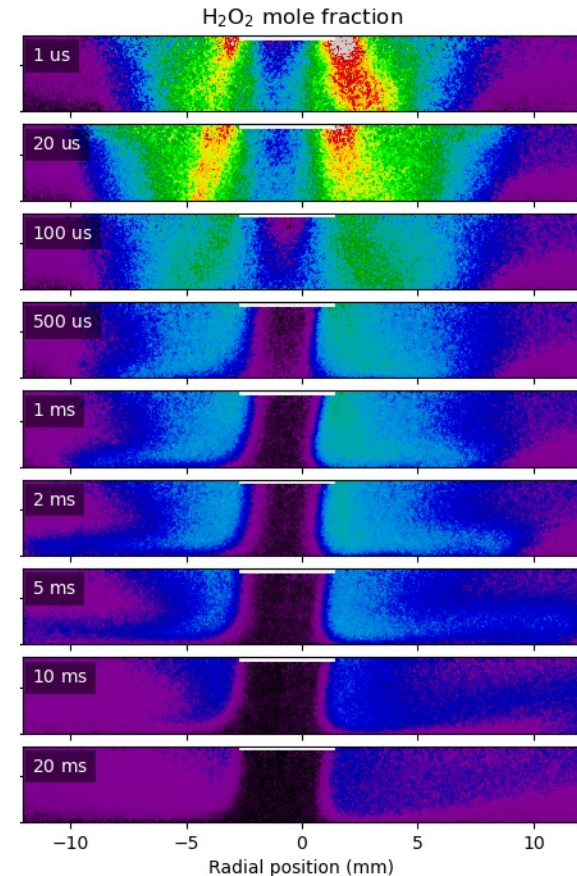
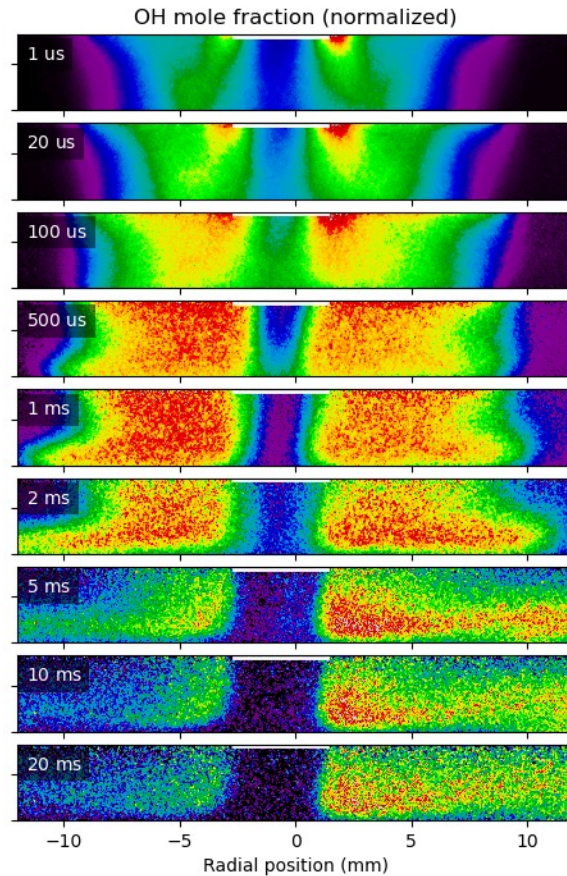
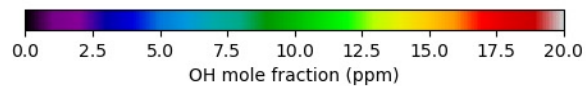
# Imaging of $\text{H}_2\text{O}_2$ and OH in He- $\text{H}_2\text{O}$ Jet



# Imaging of H<sub>2</sub>O<sub>2</sub> and OH in He-H<sub>2</sub>O Jet with 5% O<sub>2</sub> Added

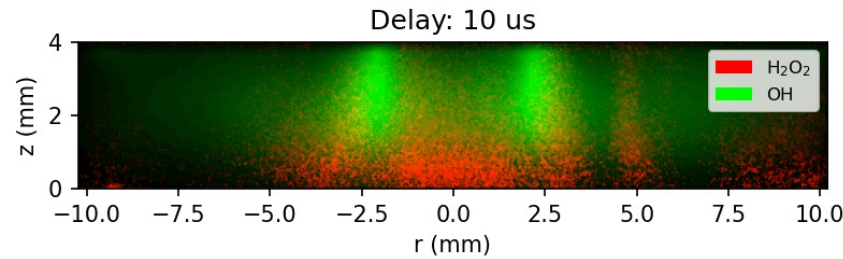


Peak OH decreases by ~ 3x



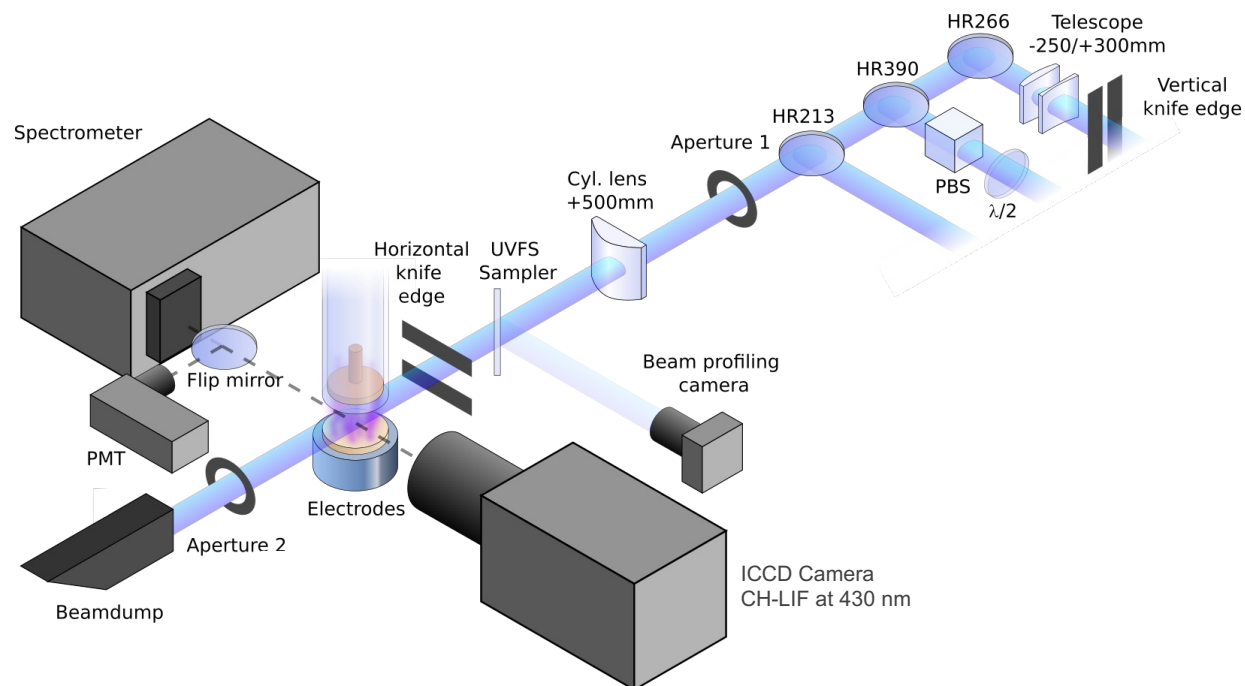
- Peak H<sub>2</sub>O<sub>2</sub> increases by ~ 5x
- H<sub>2</sub>O<sub>2</sub> production starts at earlier time

# Summary H<sub>2</sub>O<sub>2</sub>/OH Imaging in Humid Plasma Jet

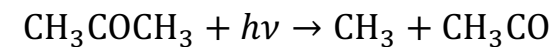


- Demonstrated quantitative 2D (PF)-LIF for imaging of OH and H<sub>2</sub>O<sub>2</sub> distributions in a ns-plasma jet close to a surface
- Spatial and temporal evolution of H<sub>2</sub>O<sub>2</sub> and OH show production, consumption, and transport of key reactive species near surface
- Effects of O<sub>2</sub> addition, flow rate, and pulse burst duration explored.
- Provides target case for modeling by Tanvir Farouk's group
- Enables further investigations of plasma-liquid/solid interactions

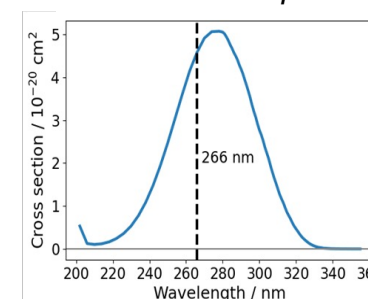
# Configuration for Methyl Photofragmentation LIF



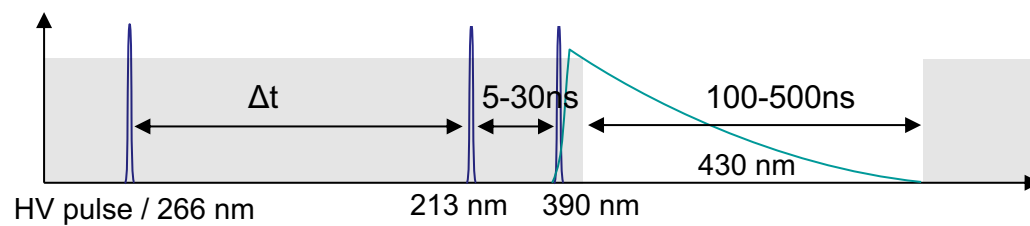
Methyl source for calibration:  
Photodissociation of Acetone



Acetone absorption



Acetone LIF used to measure  
temperature distribution and account  
for T dependence of absorption.

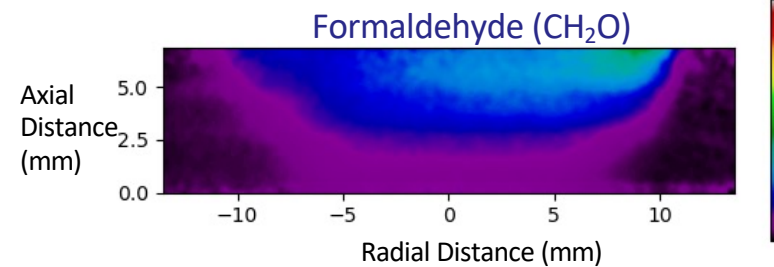
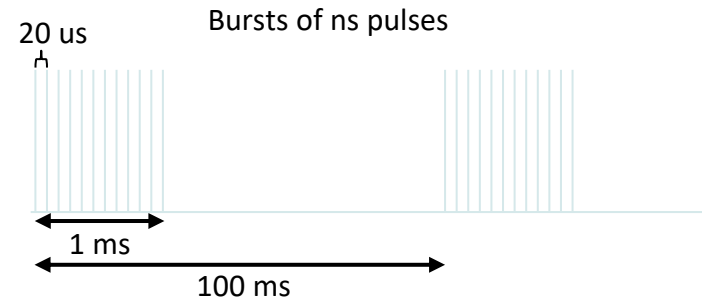
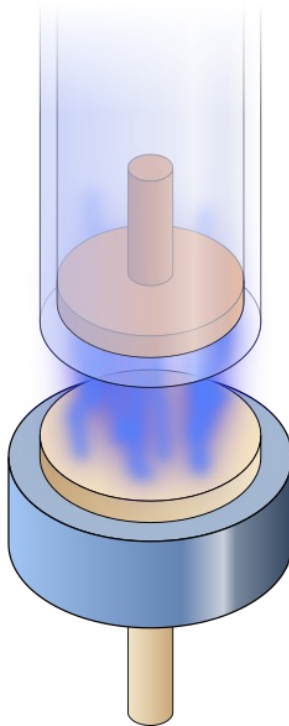


# Temporal Evolution of Methyl and Formaldehyde in Plane-to-Plane Pulsed Dielectric Barrier Discharge Plasma



Plasma conditions:

- 1 ms burst of 16kV ns pulses @ 50 kHz
- 200 Torr
- 40% CH<sub>4</sub>, 40% CO<sub>2</sub>, 20% He
- Lower surface: V<sub>2</sub>O<sub>5</sub> Catalyst @ 200 C



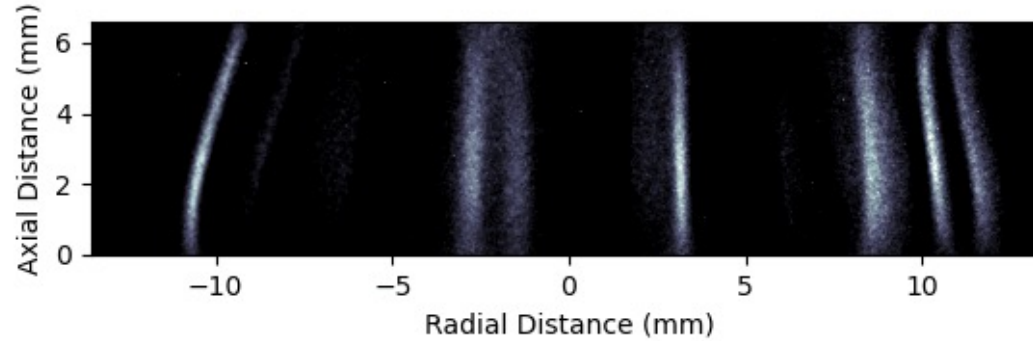
With Subtraction of Residual Signals from Previous Burst

# Methyl and Formaldehyde Imaging in a Filamentary Discharge

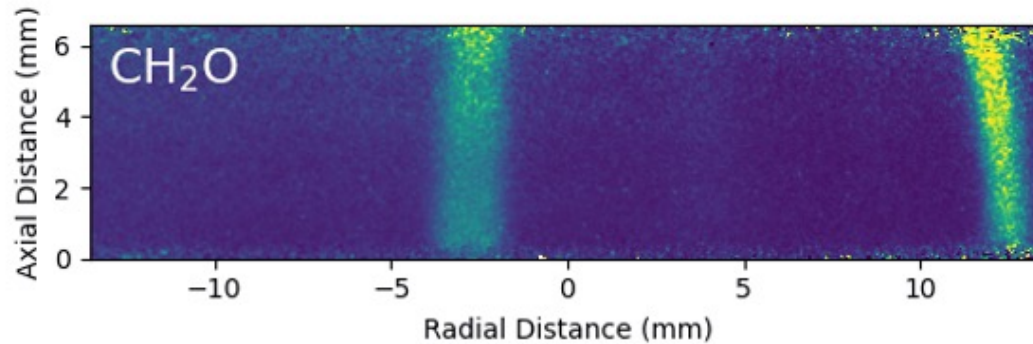
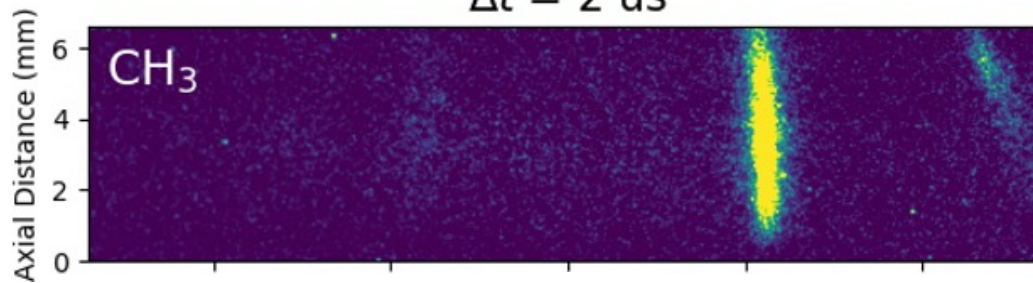


## Plasma conditions:

- 1 ms burst of 16kV ns pulses @ 50 kHz
- 100 Torr
- 83% CH<sub>4</sub>, 17% O<sub>2</sub>
- Lower surface: V<sub>2</sub>O<sub>5</sub>



$\Delta t = 2 \text{ us}$



# Summary CH<sub>3</sub> Imaging in Plasmas



- Demonstrated spatially and temporally resolved 2D imaging of methyl profiles in a plasma
- Near-surface detection using non-resonant LIF scheme
- Methyl PF-LIF calibration by photo-dissociation of acetone
- Quenching dominated by methane doesn't vary between calibration and experiment
- Acetone LIF used for temperature corrections
- Absolutely calibrated time resolved 2D-methyl profiles were measured in a ns-plasma, obtaining
- Current collaboration with U Mich.: Comparisons of measurements and modelling of methyl production in nanosecond pulsed plasma
- Sets stage for studying plasma-assisted catalysis and pyrolysis

# Acknowledgements



This research used resources of the Low Temperature Plasma Research Facility at Sandia National Laboratories, which is a collaborative research facility supported by the U.S. Department of Energy, Office of Science, Office of Fusion Energy Sciences.