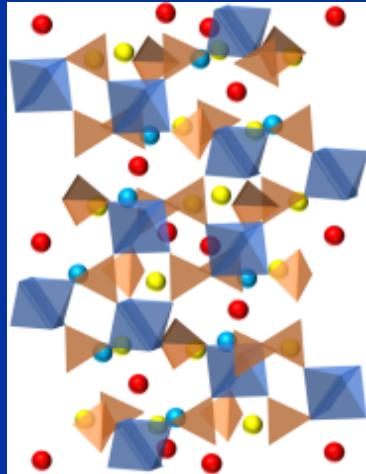
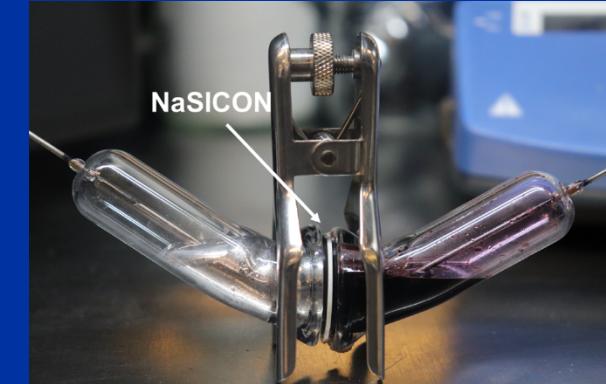


# Electrolytes Exposed to Thermal and Electrochemical Cycling in Molten Sodium Environment



Ryan Hill – University of Kentucky

Ryan.C.Hill@uky.edu



Additional Authors: Yang-Tse Cheng\*, Jacob Hempel\*, Erik Spoerke\*\*, Leo Small\*\*, Martha Gross\*\*, Amanda Peretti\*\*

\*University of Kentucky and \*\*Sandia National Laboratories



Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.

Part of SNL's Sodium Battery Program (PI: Leo Small)

May 8<sup>th</sup>-13<sup>th</sup>, 2021



Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.

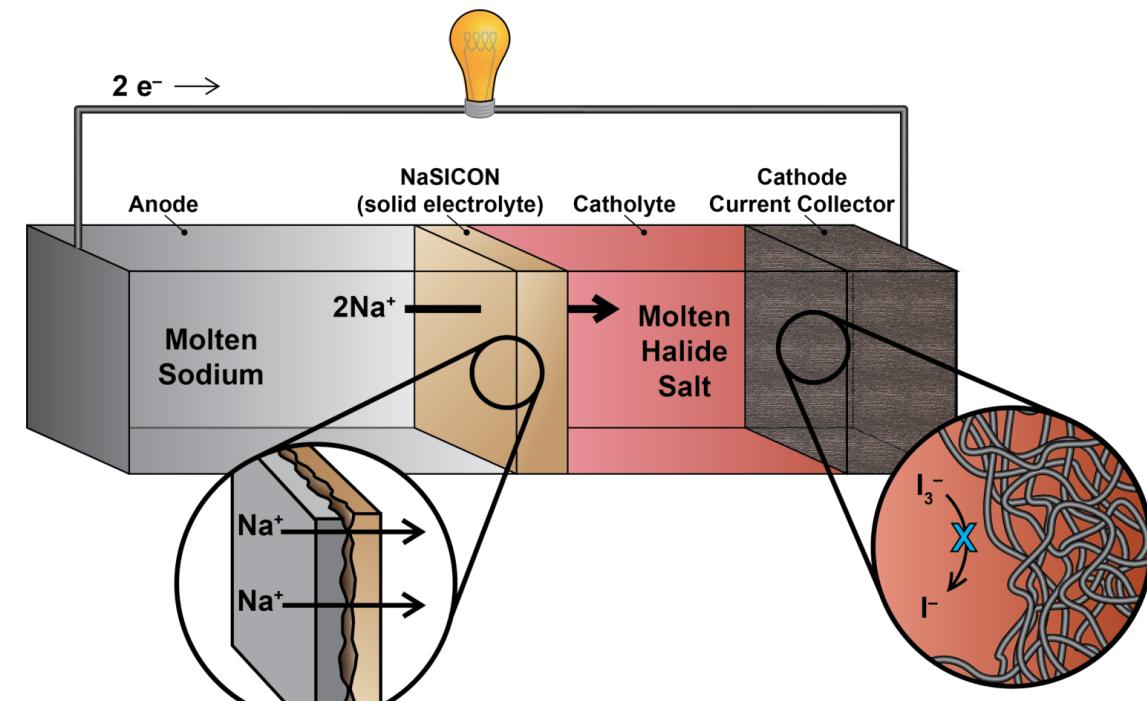
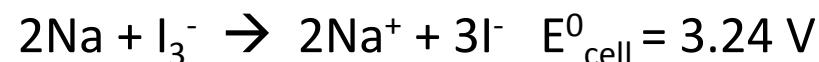
# Motivation: Low Temperature Molten Sodium (Na-NaI) Batteries

*Realizing a new, low temperature molten Na battery requires new battery materials and chemistries – particularly in **sodium ion conductors***

## Important electro-chemo-mechanical properties

- Highly  $\text{Na}^+$ -conductive
- Zero-crossover
- (Electro)chemical compatibility with Na and halide salts
- Mechanical integrity and “dendrite” suppression
  - ✓ Important for large-scale, long-duration, long-life applications

### Na-NaI battery:

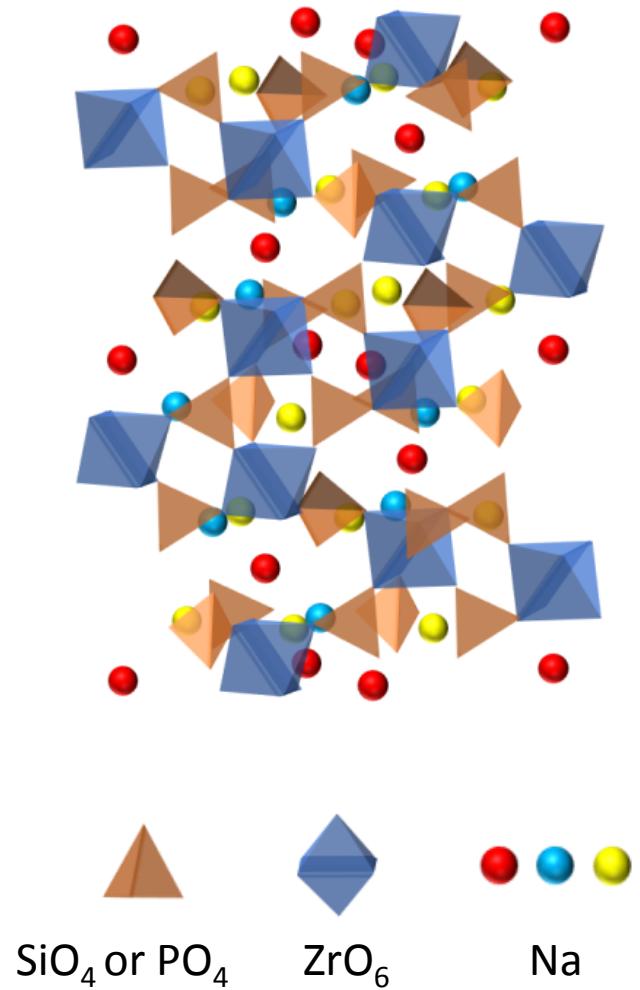
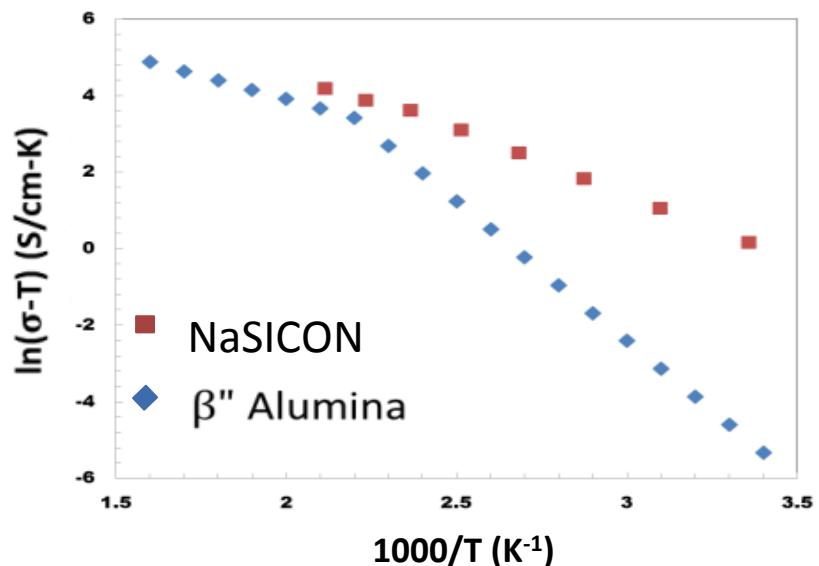


Martha Gross

# NaSICON Solid Electrolytes

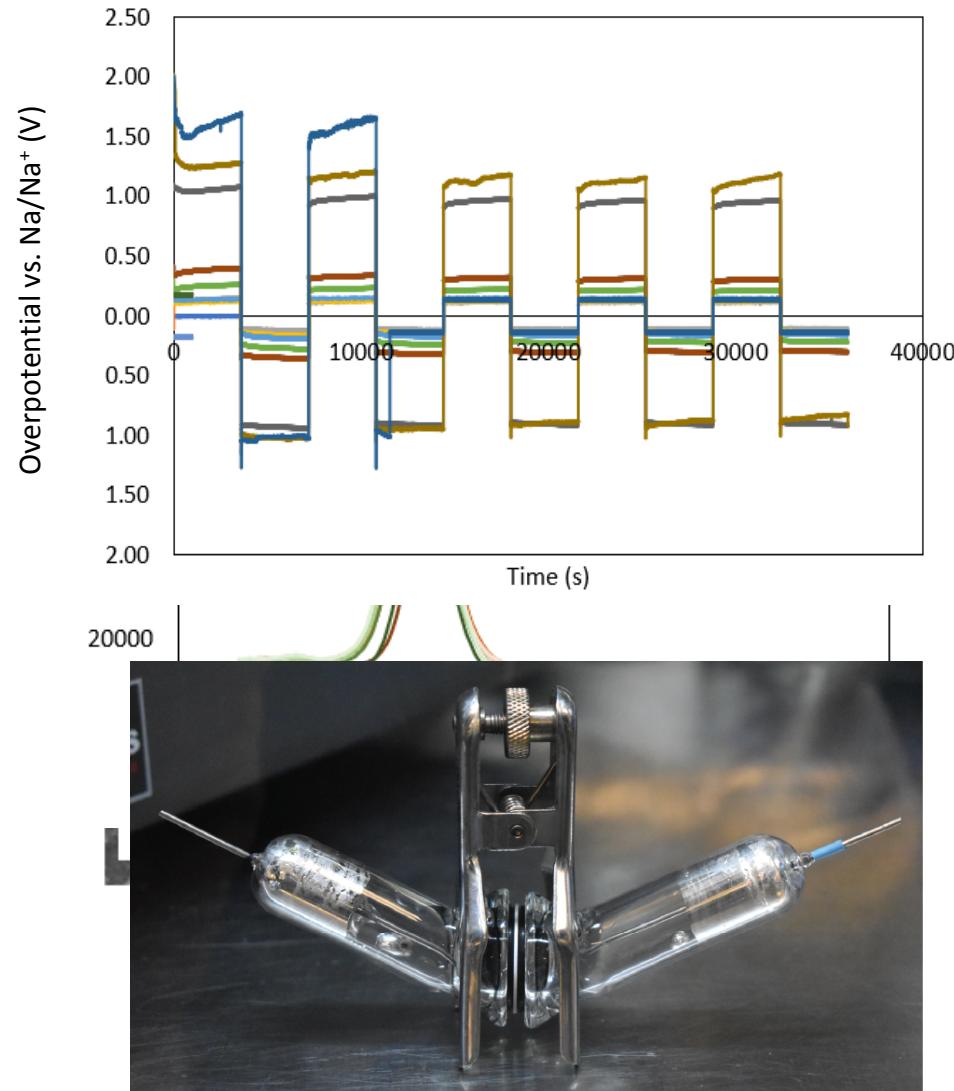
## Key Qualities of NaSICON Ceramic Ion Conductors

- $\text{Na}_{3.4}\text{Zr}_2\text{Si}_{2.4}\text{P}_{0.6}\text{O}_{12}$
- High Na-ion conductivity ( $>10^{-3}$  S/cm at 25°C)
- Zero-crossover (high-density after sintering)
- Chemically compatible with molten Na and halide salts (at low T!)
- Mechanical integrity? After exposure to battery materials?



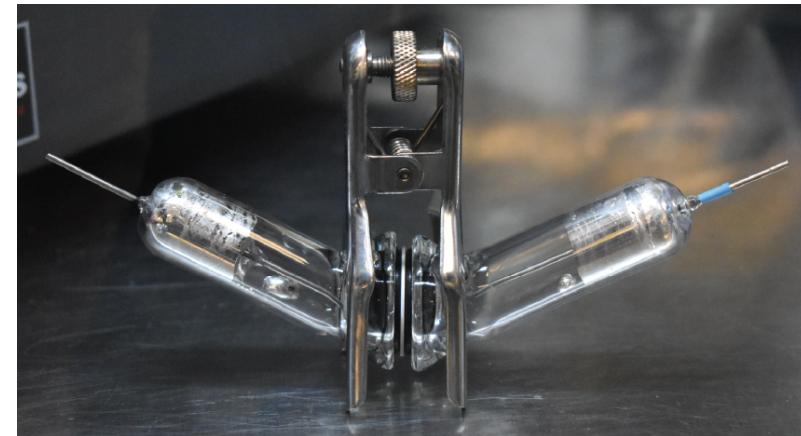
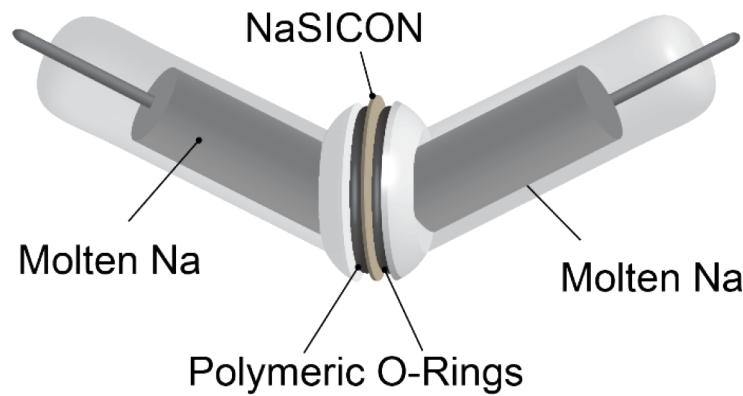
# Understanding NaSICON Behavior in Na Batteries

- Electro-chemo-mechanical phenomenon
  - Must understand all three aspects
- UK-SNL team has the capability to explore these!
  - Cycling molten sodium batteries (electrochemical)
  - Compositional analysis (chemical)
  - Indentation (mechanical)
  - Microstructure analysis



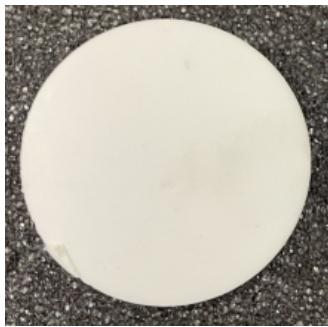
# Mechanical Behavior of NaSICON

- Exploring mechanical properties after:
  - NaSICON heated to 110 °C
  - NaSICON exposed to molten Na at 110 °C
  - NaSICON (x2) cycled up to 50 mA/cm<sup>2</sup> in Na | NaSICON | Na cell at 110 °C

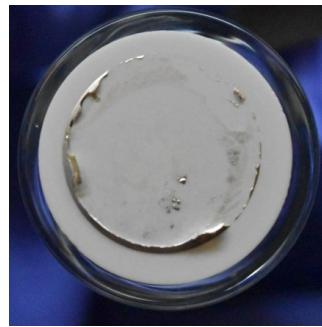


# Mechanical Behavior of NaSICON

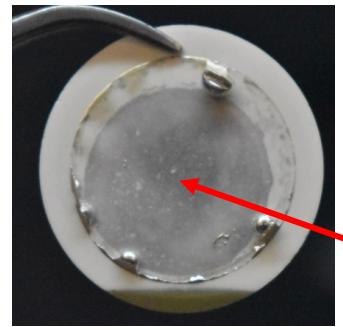
- Saw changes in NaSICON's mechanical response
  - Modulus increased after exposure to molten Na metal
  - Modulus and hardness decreased after cycling in Na|NaSICON|Na



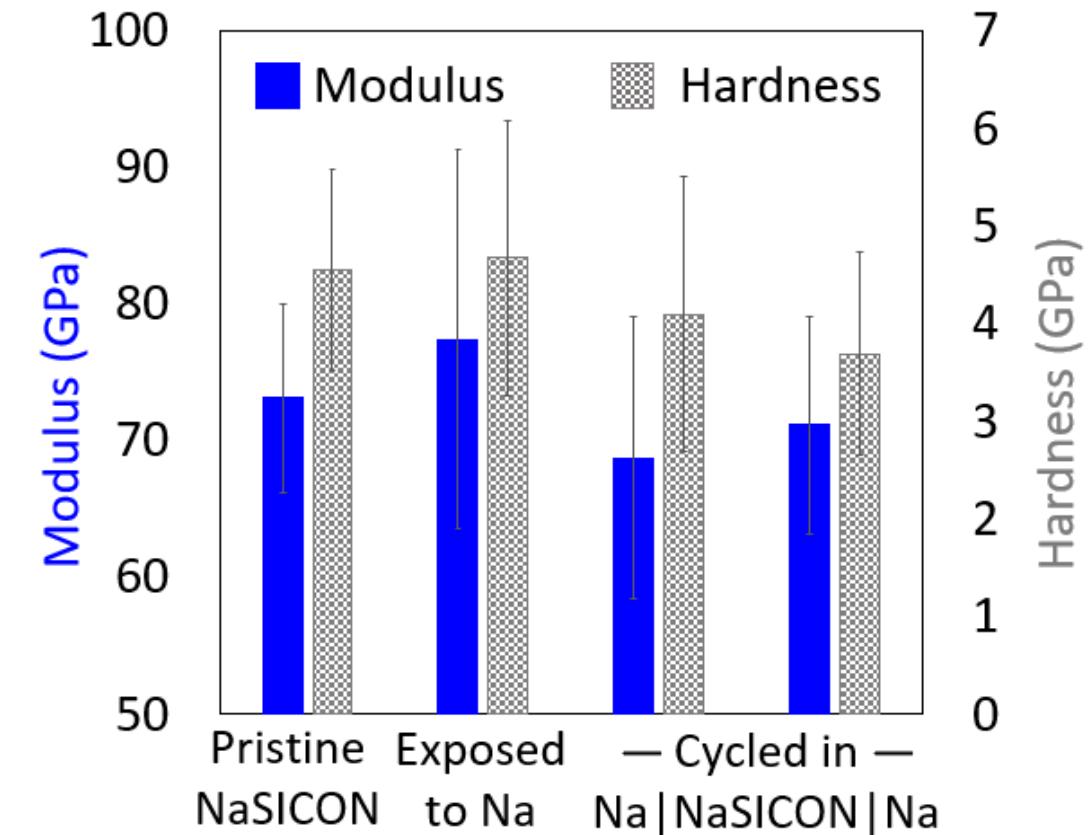
Pristine



Exposed to Na



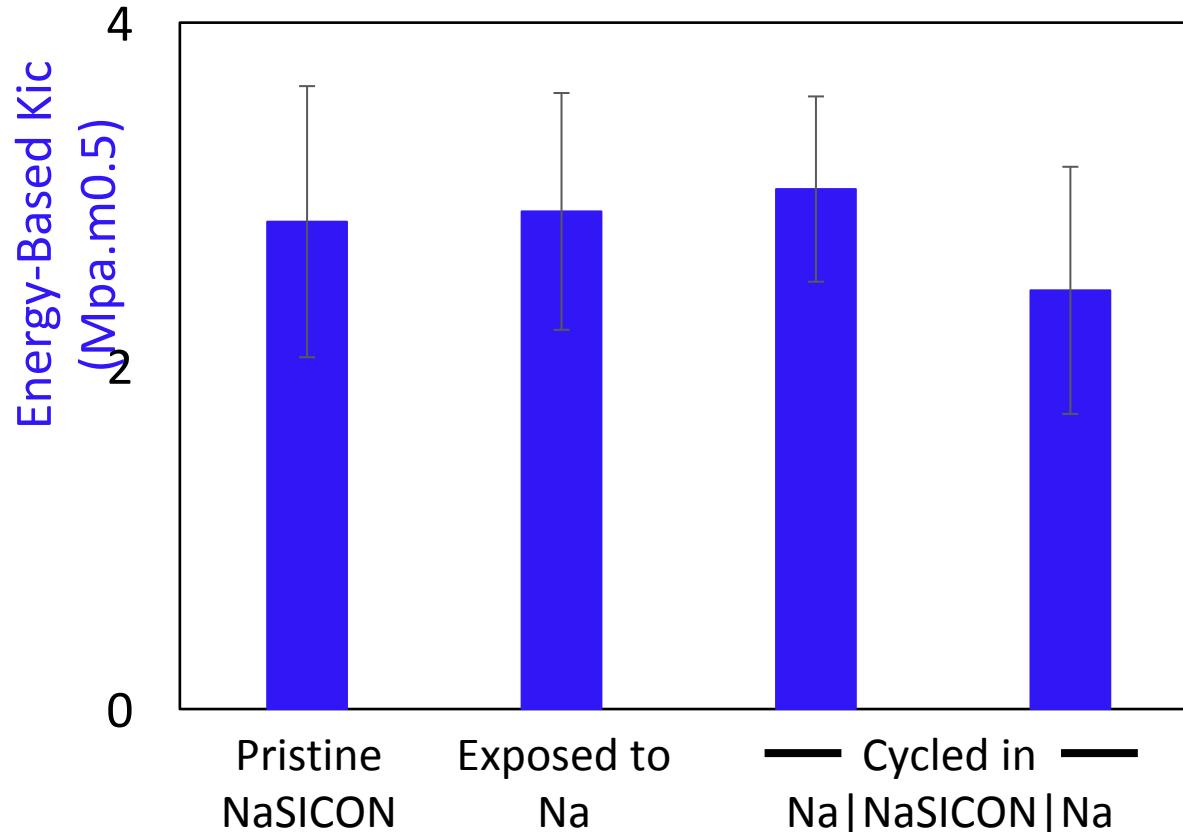
Cycled



Color disappeared  
upon air exposure

# Fracture Toughness of NaSICON

Fracture toughness is intimately related to critical current density (CCD)



$$W_{\text{irr}} = W_{\text{irrev}} - W_{\text{pp}}$$

$$K_{ic} = \sqrt{\frac{E_r W_{\text{crack}}}{A_m}}$$

$K_{ic}$ : Fracture Toughness

$W_{\text{irrev}}$ : Total irreversible indentation work

$W_{\text{pp}}$ : Purely plastic indentation work

$W_{\text{crack}}$ : Work done to create cracks

$E_r$ : Material reduced modulus

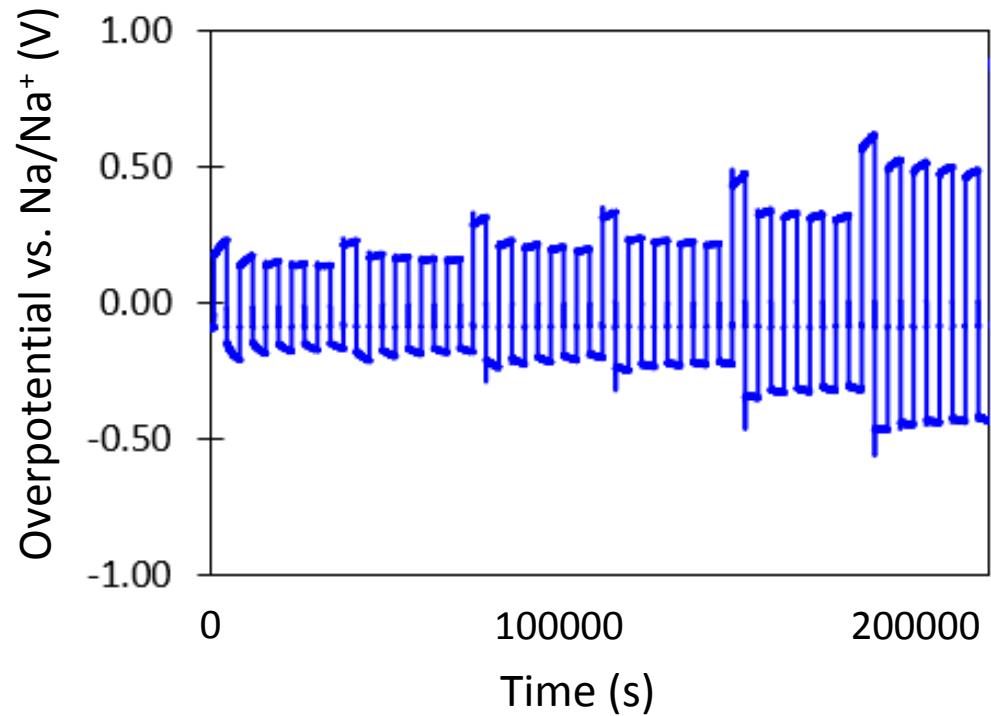
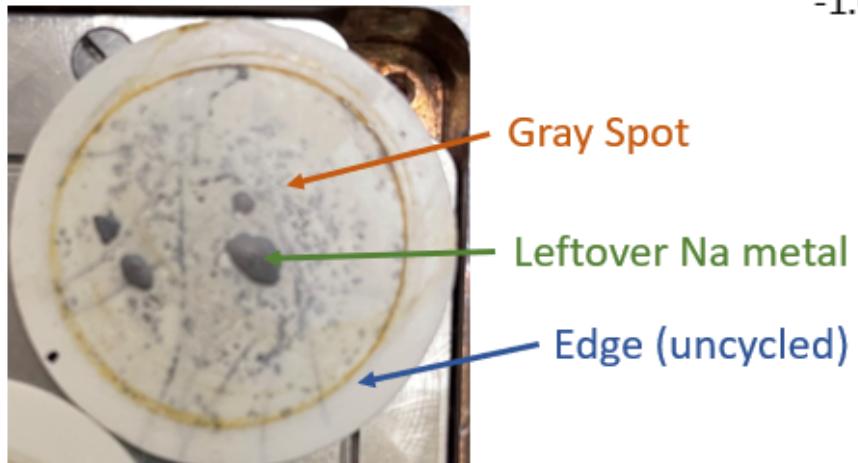
$A_m$ : Indent contact area

***Fracture toughness not significantly impacted by Na<sup>+</sup> conduction***

***Cycling NaSICON should not affect ability to handle higher current densities***

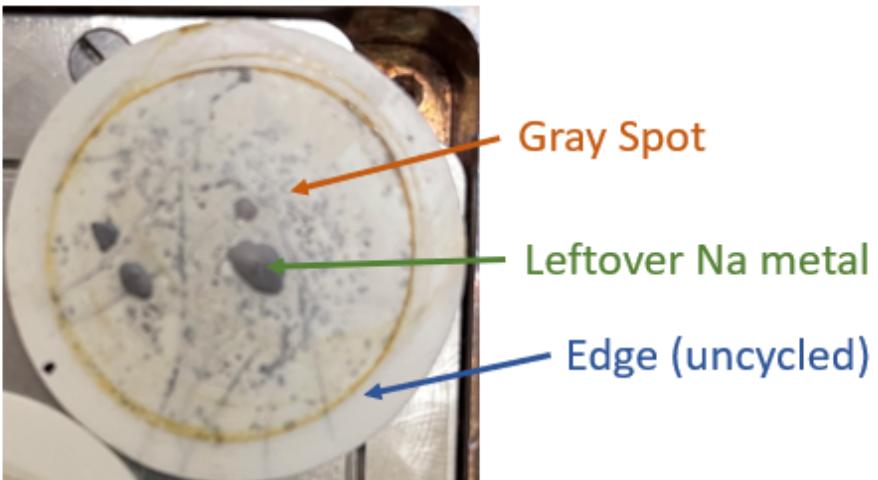
# Chemical Changes in NaSICON

- NaSICON cycled up to 225 mA/cm<sup>2</sup> in Na|NaSICON|Na symmetric cell at 110 °C
  - High current density to induce chemical changes
  - Typical CCD in SEs @ RT is ~0.1-1 mA/cm<sup>2</sup>
- Permanent gray spots (and lines) appeared across NaSICON surface

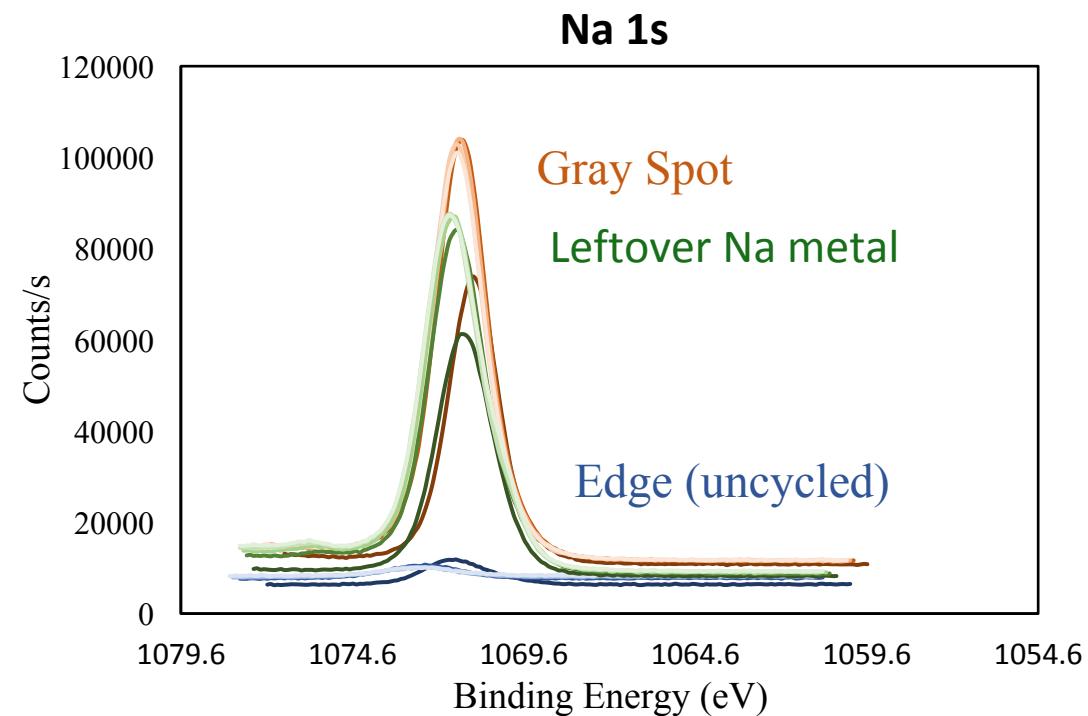
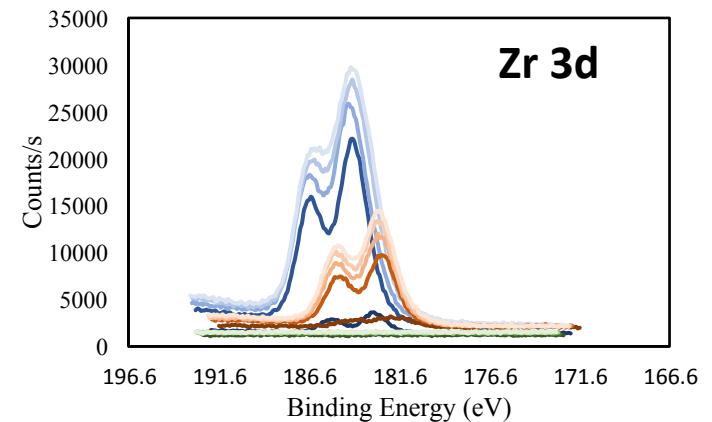


# Chemical Changes in NaSICON

- XPS to determine composition in various areas
  - Gray spots have much higher Na1s signal than edge (uncycled) area - more similar to leftover “pure” Na metal
  - Gray spots still show expected Zr and Si signals

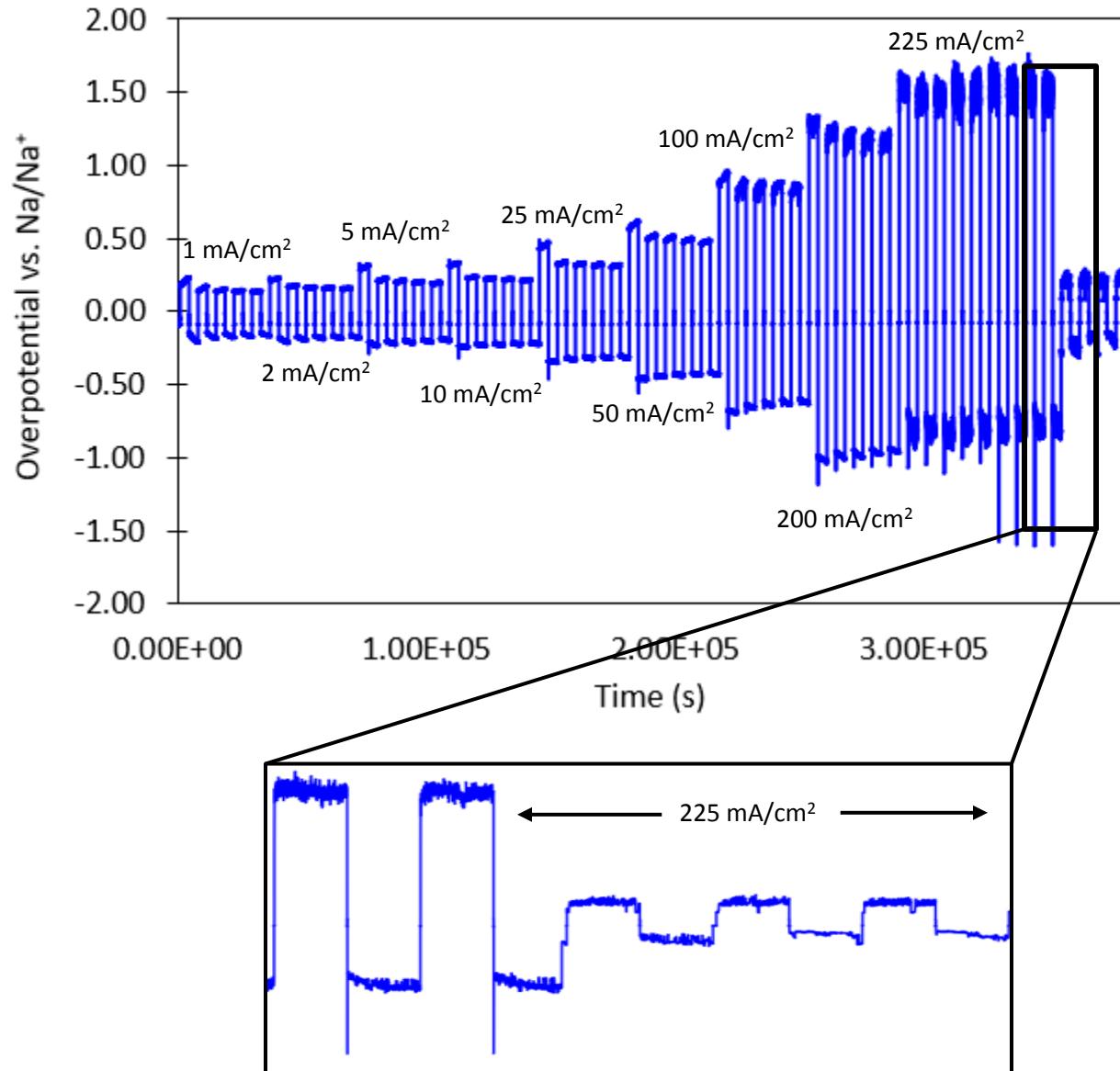
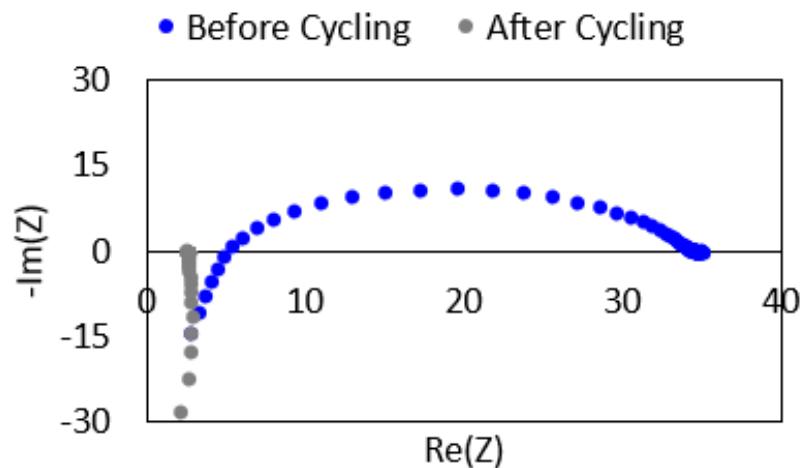


***“Dendrites”?***



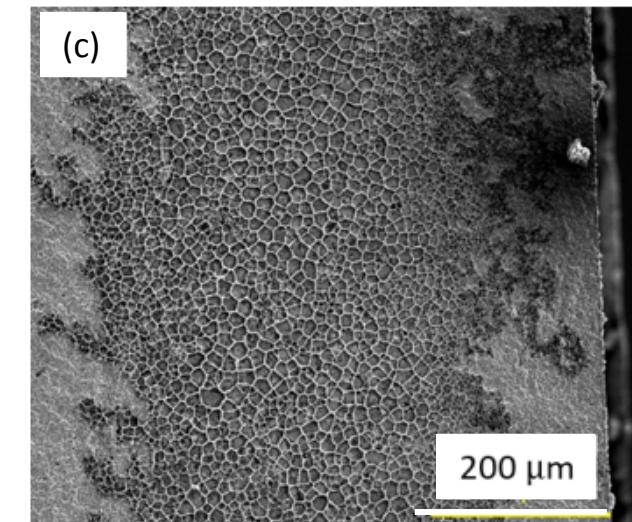
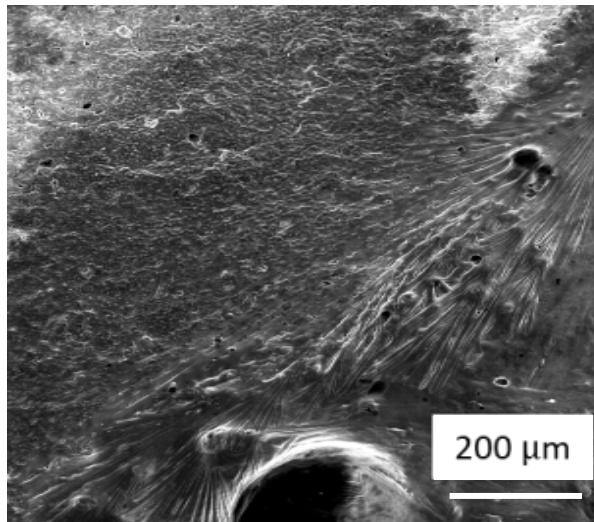
# Electrochemical Behavior of NaSiCON

- Cycling above  $100 \text{ mA/cm}^2$  led to unstable voltage plateaus
- Continued cycling at  $225 \text{ mA/cm}^2$  led to eventual voltage drop
  - Very small impedance measured by impedance spectroscopy
  - Consistent with electronic short

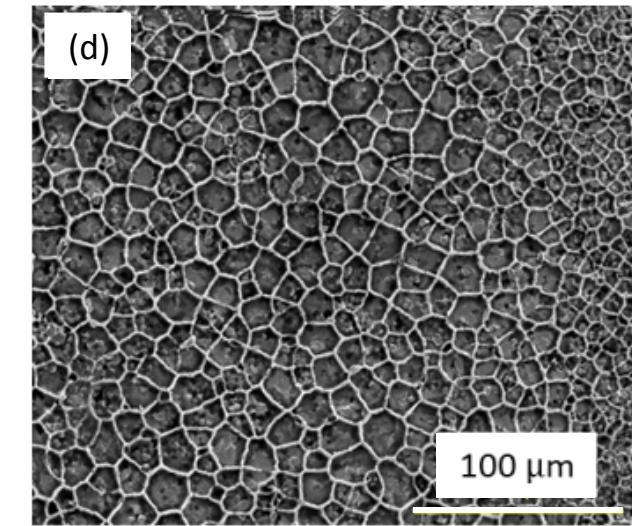
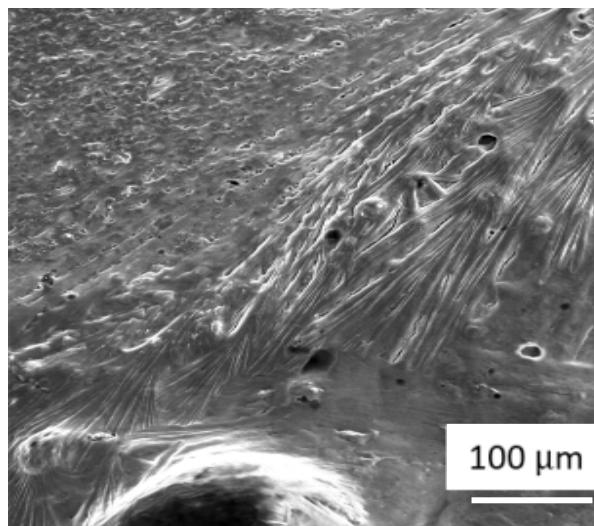


# Electrochemical Behavior of NaSICON

- Na can propagate through NaSICON causing an electronic short



- Morphology of Na within NaSICON is vastly different than Li within LLZO
  - Needle-like vs. grain boundary accumulation



Na propagation in NaSICON

Li propagation in LLZO



- NaSICON solid electrolytes will play an integral role in next -generation energy storage technology
- NaSICON SEs exhibit changes in their mechanical, chemical, and electrochemical behavior during cycling
  - These all contribute to NaSICON failure in Na batteries
- Failure of NaSICON at high-temperature is significantly different than in RT batteries
  - A fundamental understanding of this failure will guide improved electrolyte fabrication

# Acknowledgments

This work was in collaboration with Sandia National Labs and was supported by the U.S. Department of Energy Office of Electricity's Energy Storage Program, managed by Dr. Imre Gyuk.

## University of Kentucky

Bhamiti Sharma  
Jacob Hempel  
Jacob Bonta  
Kübra Uzun  
Bayode Dada  
Haidar Alolaywi  
Yang-Tse Cheng

## Sandia National Labs

Babu Chalamala  
Erik Spoerke  
Leo Small  
Martha Gross  
Amanda Peretti

Ryan Hill

[Ryan.C.Hill@uky.edu](mailto:Ryan.C.Hill@uky.edu)

