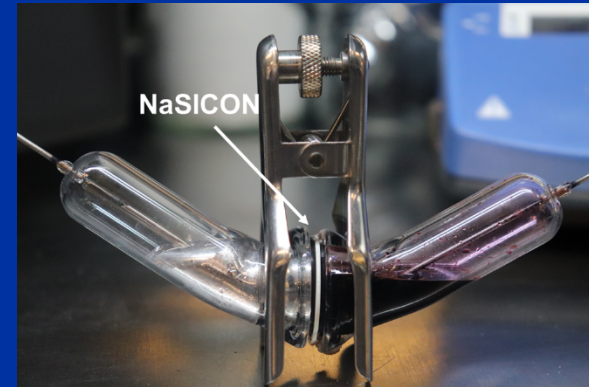
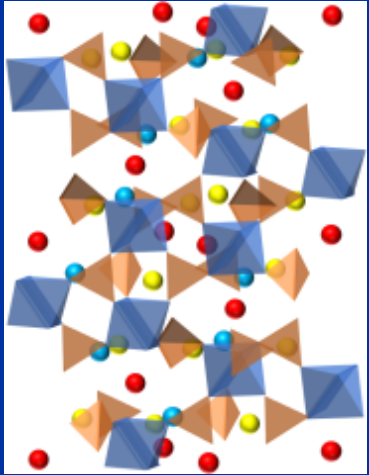


Electrolytes Exposed to Thermal and Electrochemical Cycling in Molten Sodium Environment

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*University of Kentucky and **Sandia National Laboratories



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Part of SNL's Sodium Battery Program (PI: Leo Small)

May 8th-13th, 2021



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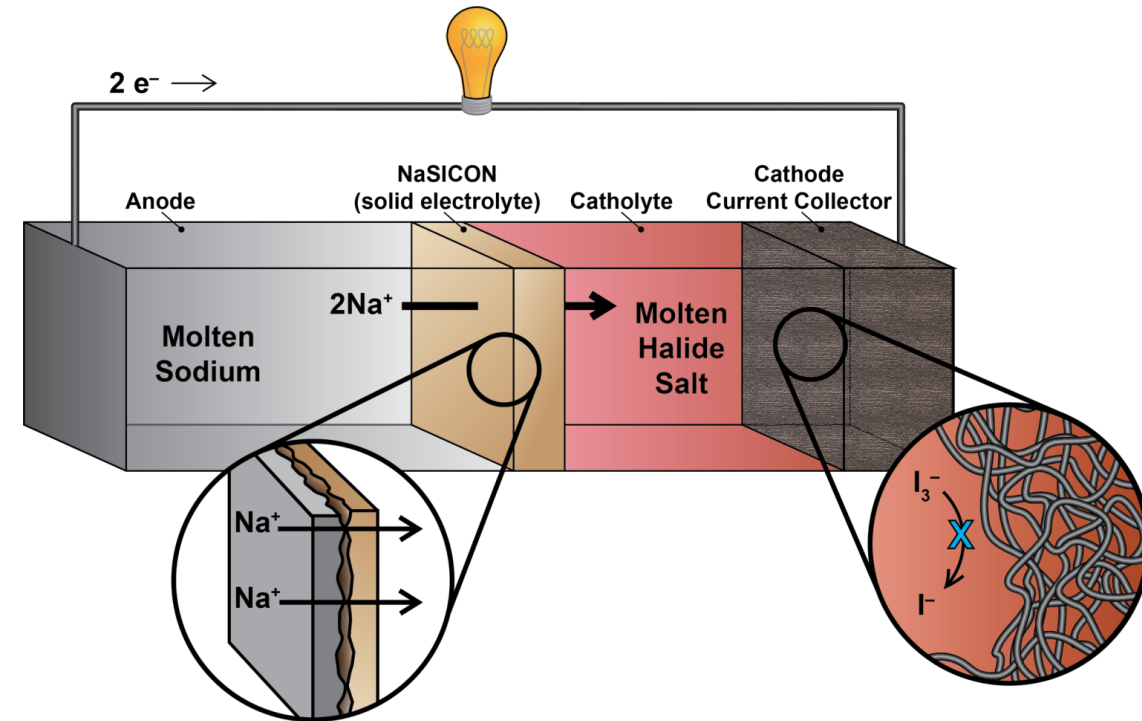
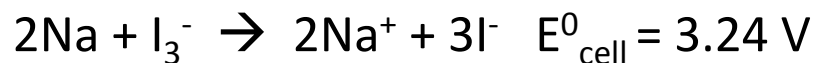
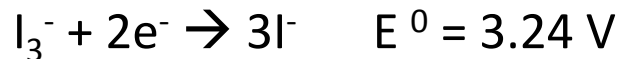
Motivation: Low Temperature Molten Sodium (Na-NaI) Batteries

*Realizing a new, low temperature molten Na battery requires new battery materials and chemistries – particularly in **sodium ion conductors***

Important electro-chemo-mechanical properties

- Highly Na⁺-conductive
 - Zero-crossover
 - (Electro)chemical compatibility with Na and halide salts
 - Mechanical integrity and “dendrite” suppression
- ✓ Important for large-scale, long-duration, long-life applications

Na-NaI battery:

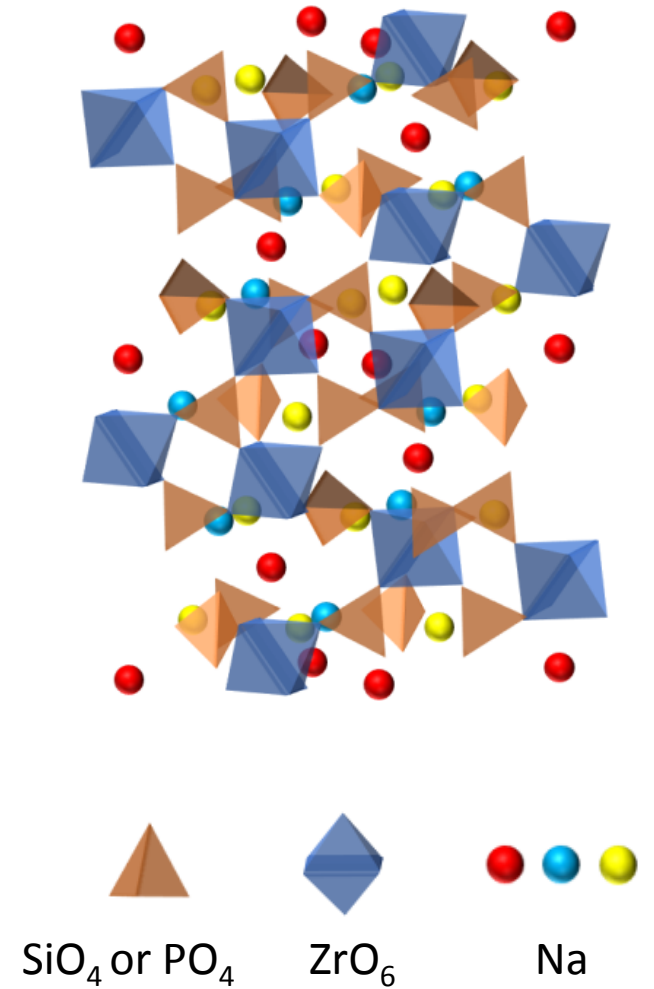
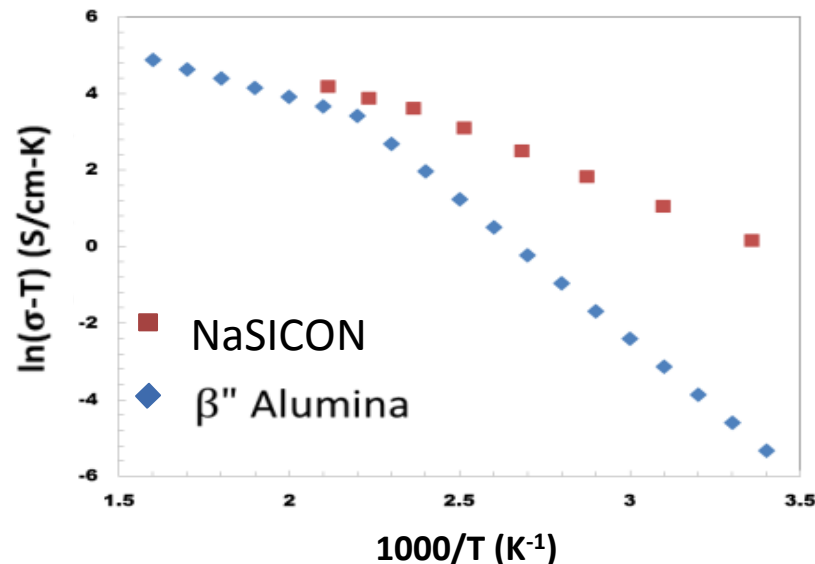


Martha Gross

NaSICON Solid Electrolytes

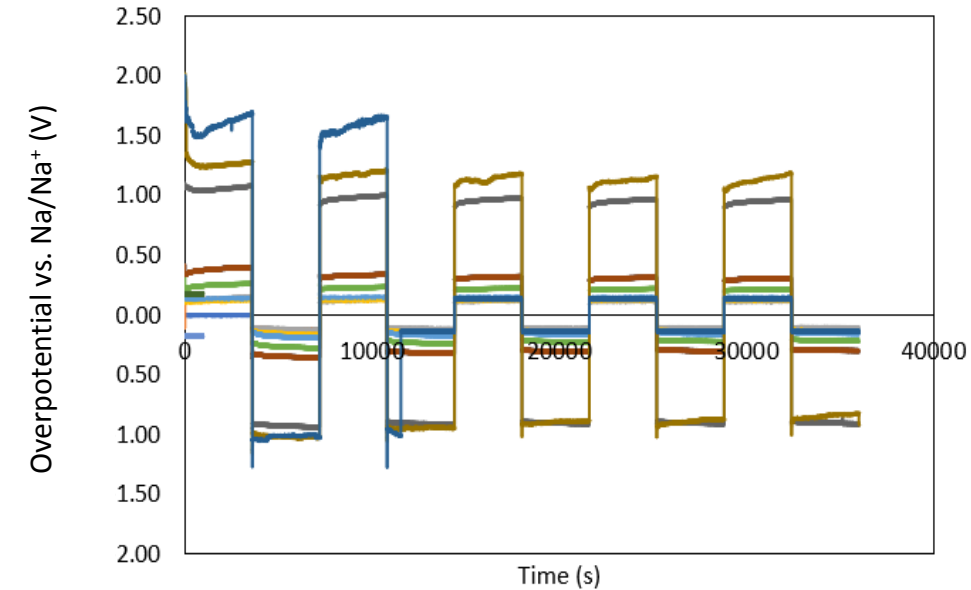
Key Qualities of NaSICON Ceramic Ion Conductors

- $\text{Na}_{3.4}\text{Zr}_2\text{Si}_{2.4}\text{P}_{0.6}\text{O}_{12}$
- High Na-ion conductivity ($>10^{-3}$ S/cm at 25°C)
- Zero-crossover (high-density after sintering)
- Chemically compatible with molten Na and halide salts (at low T!)
- Mechanical integrity? After exposure to battery materials?



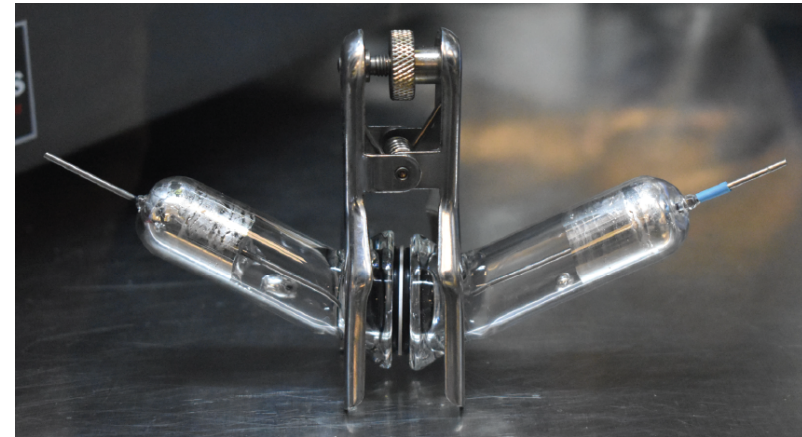
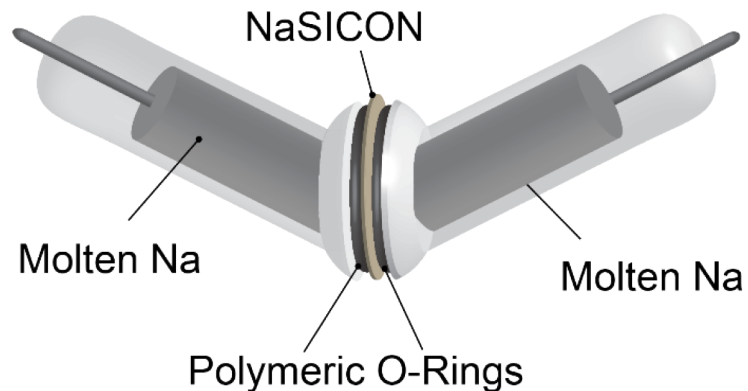
Understanding NaSICON Behavior in Na Batteries

- Electro-chemo-mechanical phenomenon
 - Must understand all three aspects
- UK-SNL team has the capability to explore these!
 - Cycling molten sodium batteries (electrochemical)
 - Compositional analysis (chemical)
 - Indentation (mechanical)
 - Microstructure analysis



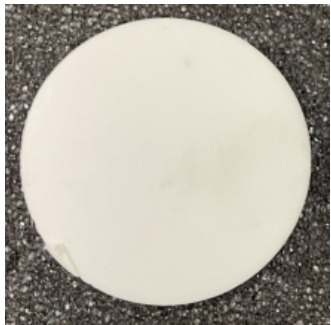
Mechanical Behavior of NaSICON

- Exploring mechanical properties after:
 - NaSICON heated to 110 °C
 - NaSICON exposed to molten Na at 110 °C
 - NaSICON (x2) cycled up to 50 mA/cm² in Na|NaSICON|Na cell at 110 °C

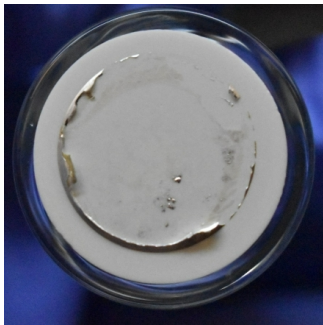


Mechanical Behavior of NaSICON

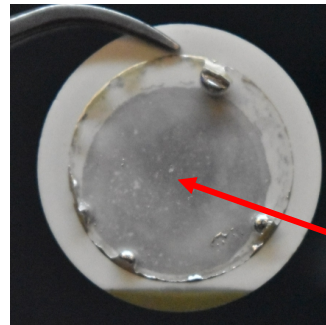
- Saw changes in NaSICON's mechanical response
 - Modulus increased after exposure to molten Na metal
 - Modulus and hardness decreased after cycling in Na|NaSICON|Na



Pristine

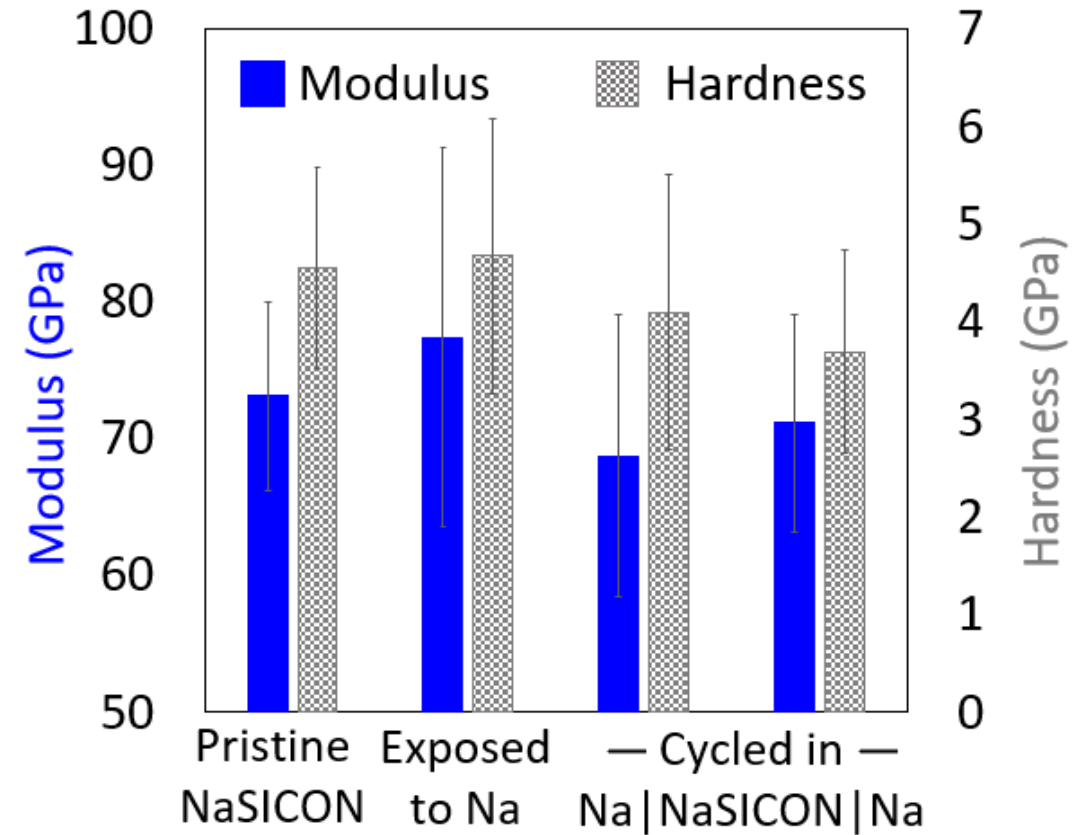


Exposed to Na



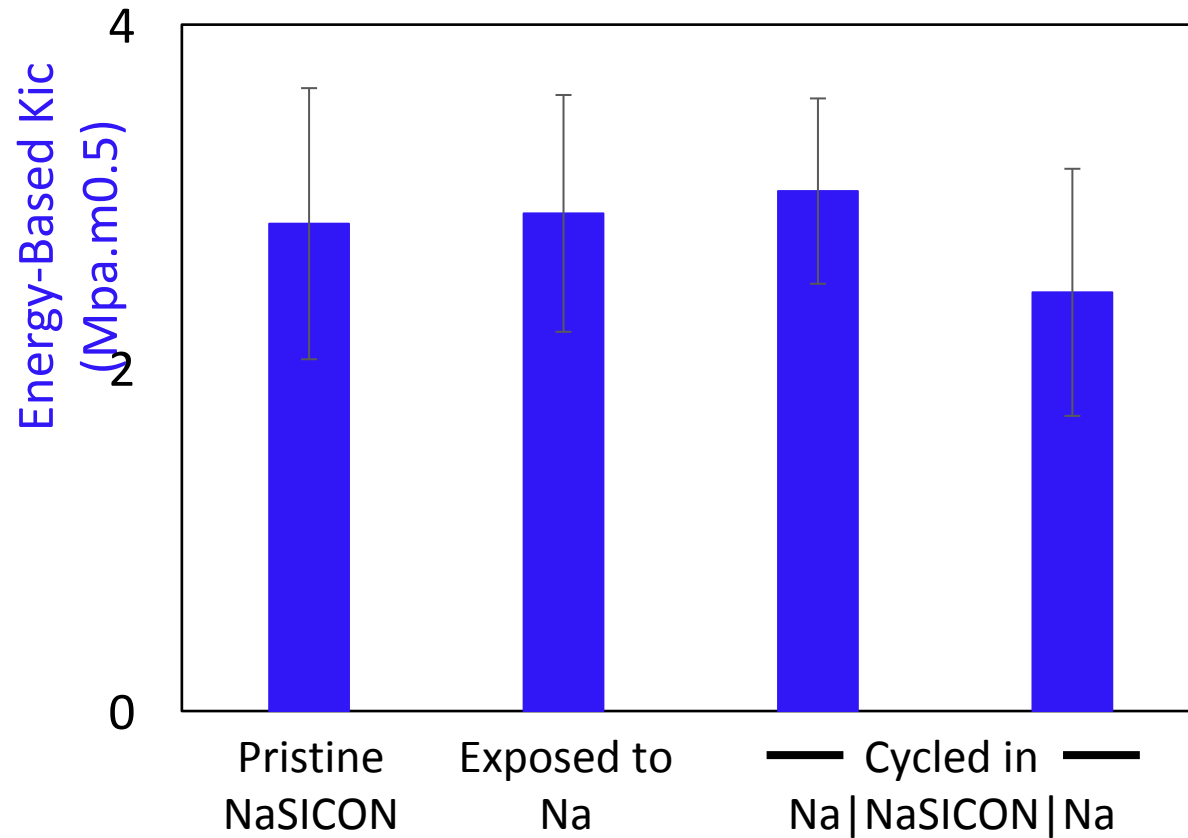
Cycled

Color disappeared upon air exposure



Fracture Toughness of NaSICON

Fracture toughness is intimately related to critical current density (CCD)



$$W_{crack} = W_{irrev} - W_{pp}$$

$$K_{ic} = \sqrt{\frac{E_r W_{crack}}{A_m}}$$

K_{ic}: Fracture Toughness

W_{irrev}: Total irreversible indentation work

W_{pp}: Purely plastic indentation work

W_{crack}: Work done to create cracks

E_r: Material reduced modulus

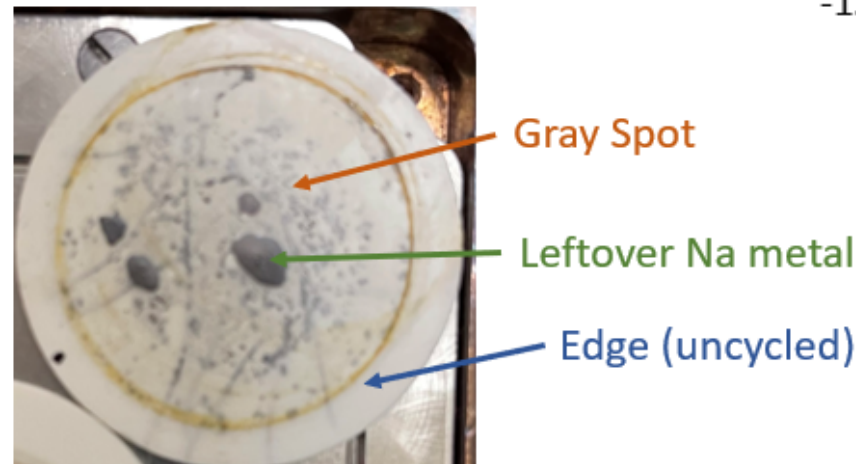
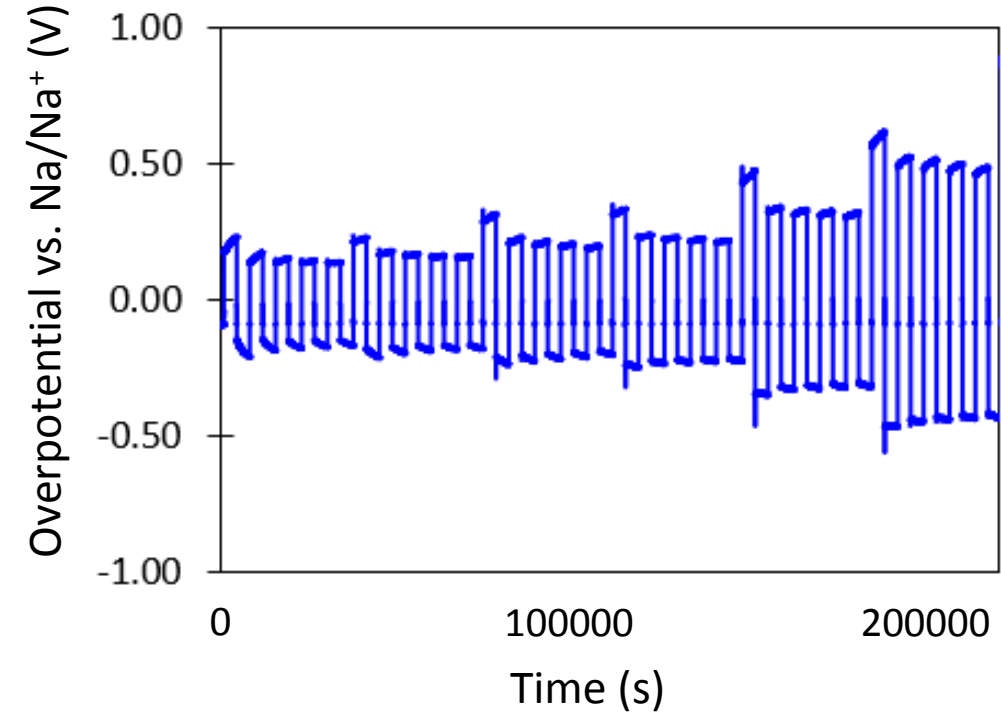
A_m: Indent contact area

Fracture toughness not significantly impacted by Na⁺ conduction

Cycling NaSICON should not affect ability to handle higher current densities

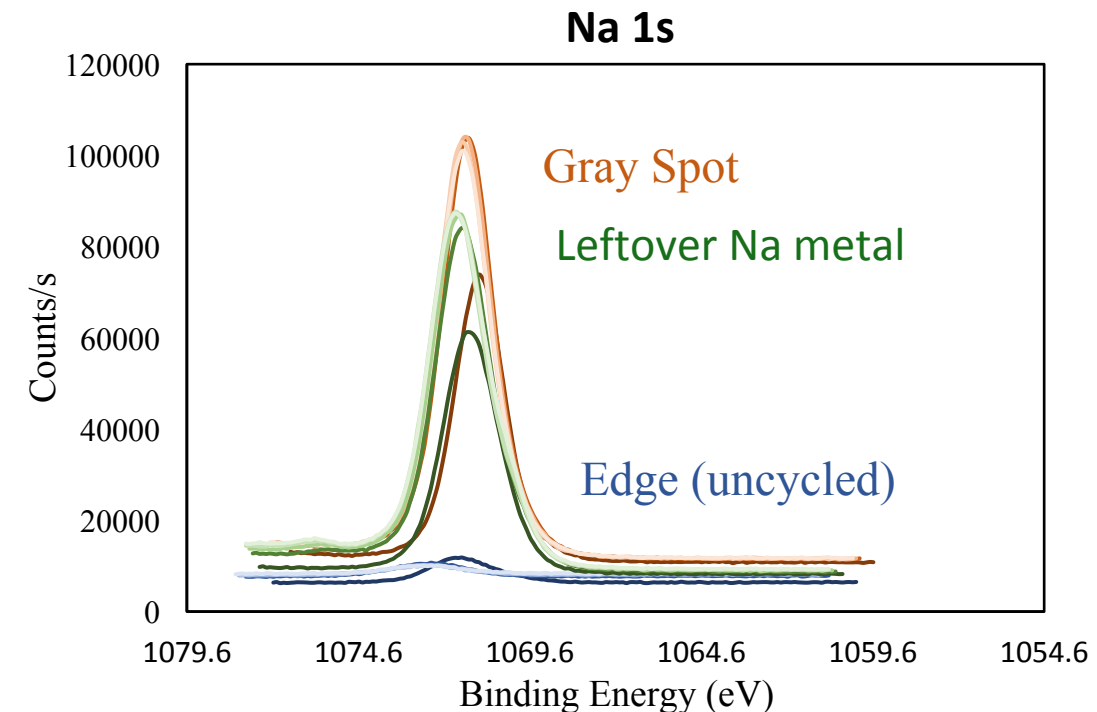
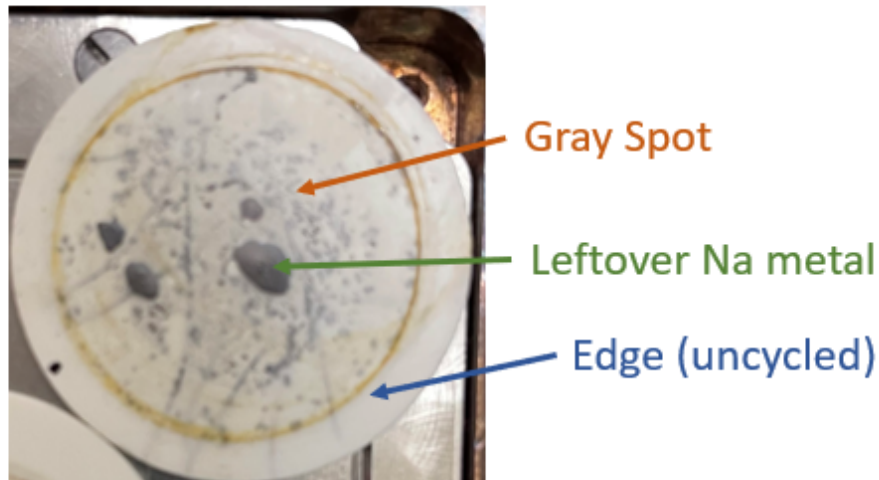
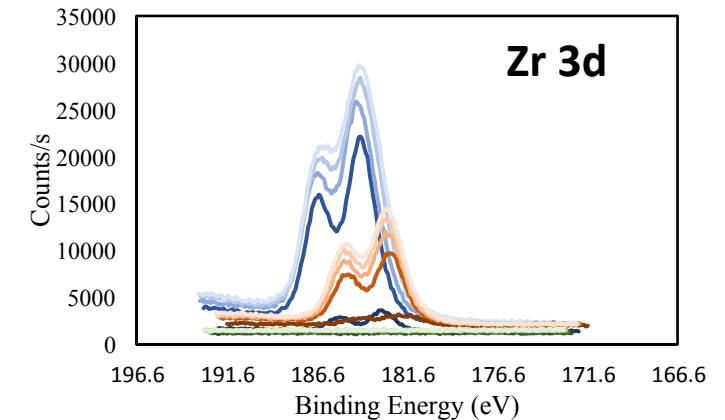
Chemical Changes in NaSICON

- NaSICON cycled up to 225 mA/cm² in Na | NaSICON | Na symmetric cell at 110 °C
 - High current density to induce chemical changes
 - Typical CCD in SEs @ RT is ~0.1-1 mA/cm²
- Permanent gray spots (and lines) appeared across NaSICON surface



Chemical Changes in NaSICON

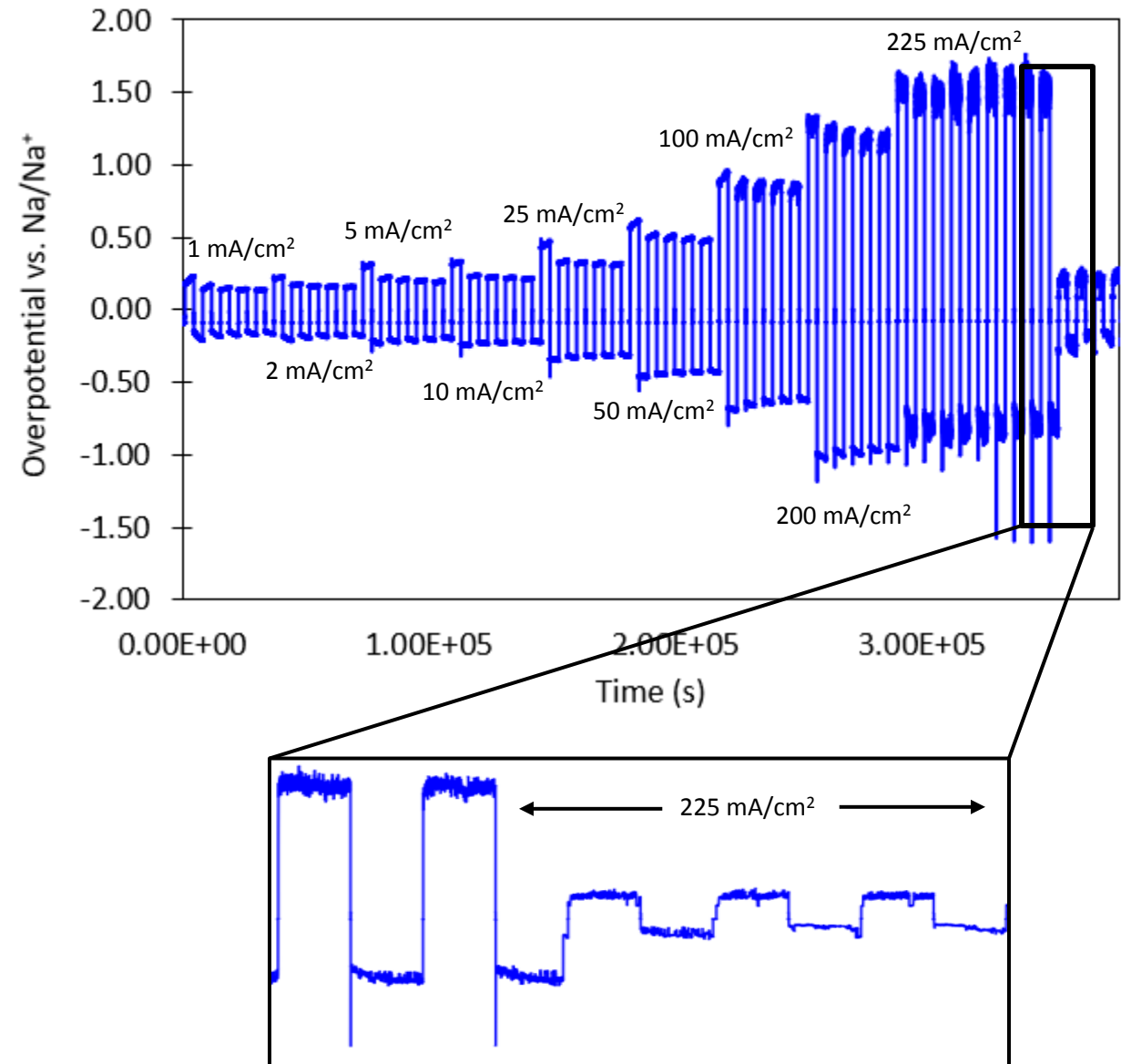
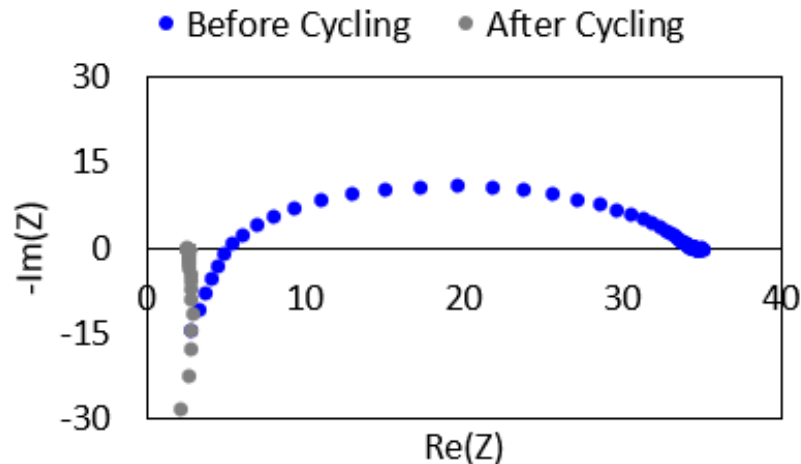
- XPS to determine composition in various areas
 - Gray spots have much higher Na1s signal than edge (uncycled) area - more similar to leftover “pure” Na metal
 - Gray spots still show expected Zr and Si signals



“Dendrites”?

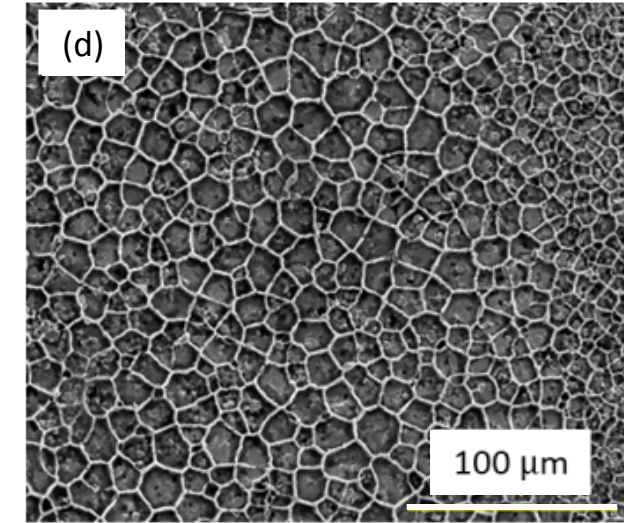
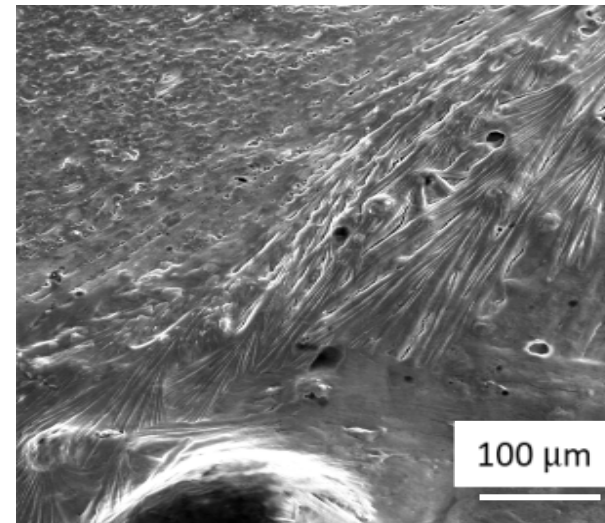
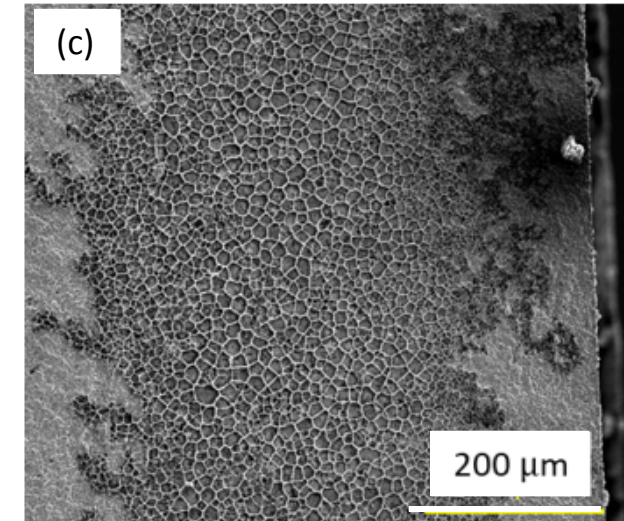
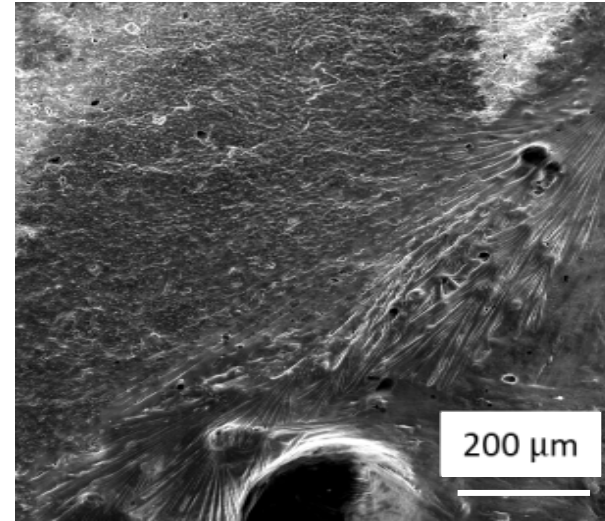
Electrochemical Behavior of NaSICON

- Cycling above 100 mA/cm^2 led to unstable voltage plateaus
- Continued cycling at 225 mA/cm^2 led to eventual voltage drop
 - Very small impedance measured by impedance spectroscopy
 - Consistent with electronic short



Electrochemical Behavior of NaSICON

- Na can propagate through NaSICON causing an electronic short
- Morphology of Na within NaSICON is vastly different than Li within LLZO
 - Needle-like vs. grain boundary accumulation



Na propagation in NaSICON

Li propagation in LLZO

In Short



- NaSICON solid electrolytes will play an integral role in next-generation energy storage technology
- NaSICON SEs exhibit changes in their mechanical, chemical, and electrochemical behavior during cycling
 - These all contribute to NaSICON failure in Na batteries
- Failure of NaSICON at high-temperature is significantly different than in RT batteries
 - A fundamental understanding of this failure will guide improved electrolyte fabrication

Acknowledgments

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