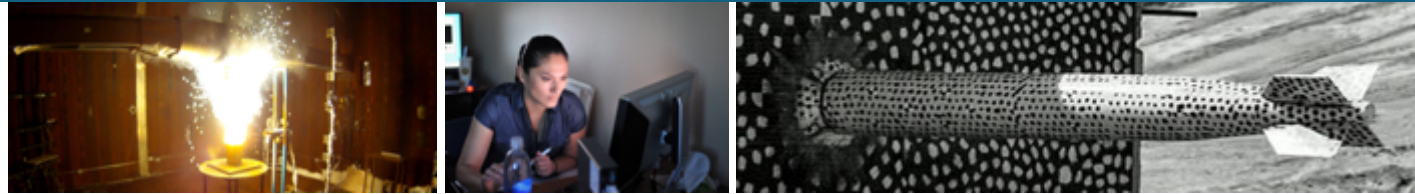




Atomistic Modeling of Beryllium and Helium Implantation in Tungsten Using Machine Learned Interatomic Potentials



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2021 APS-DPP

November 8, 2021

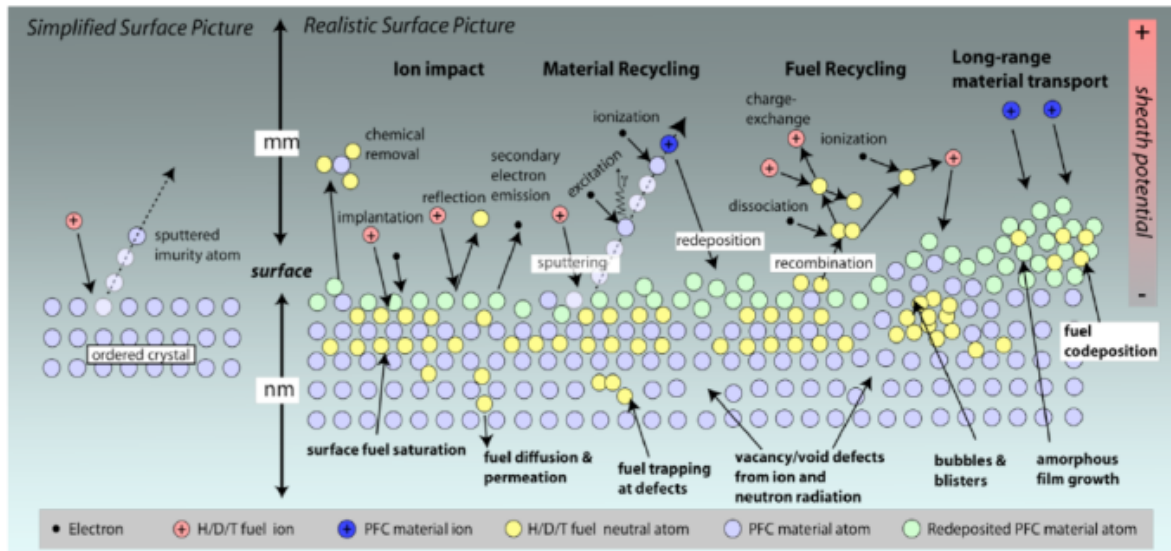
Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.



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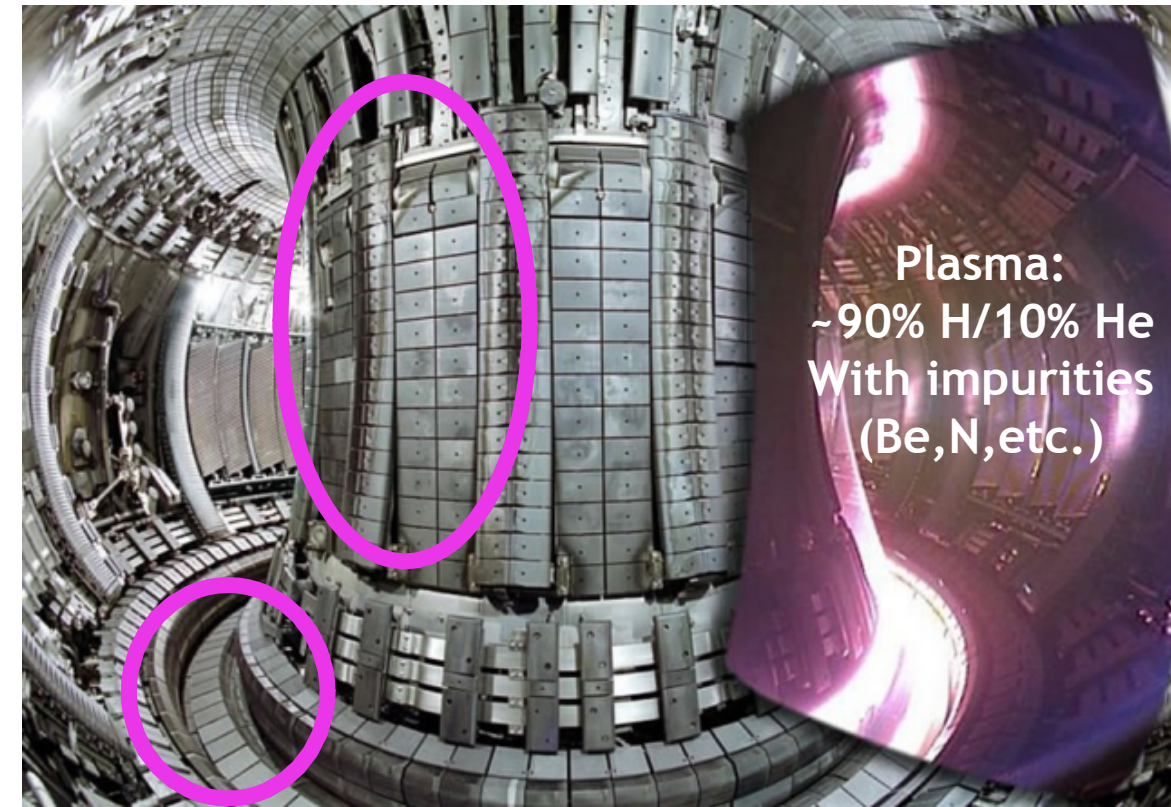
2 Materials for Fusion Energy

- Difficult to develop materials to handle extreme conditions within tokamak
- Large heat loads of 10-20 MW/m³
- High particles fluxes of $\sim 10^{24} \text{ m}^{-2}\text{s}^{-1}$ of mixed ion species (H/He/Be/N etc.)



Wirth, et al. MRS Bulletin 36 (2011) 216-222

Beryllium First Wall



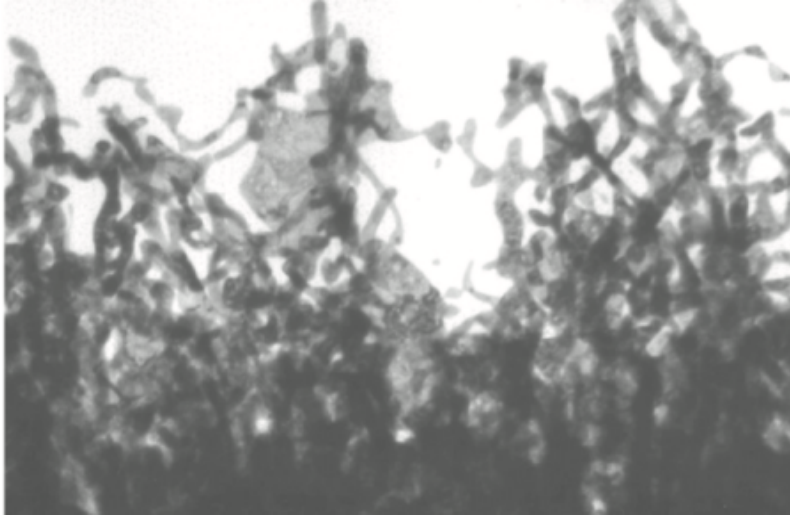
Tungsten Divertor

iter.org

- Many complex processes that occur at the plasma/material interface that can lead to material degradation

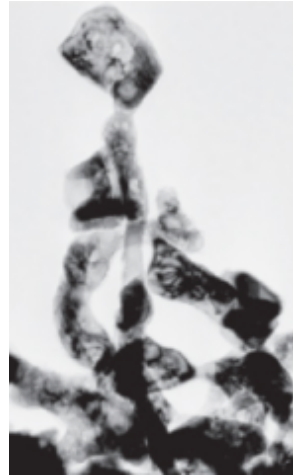
Plasma Material Interactions in Tungsten

Helium Fuzz Growth



Kajita, et al. J. Nucl. Mater. 418, (2011) 152-158

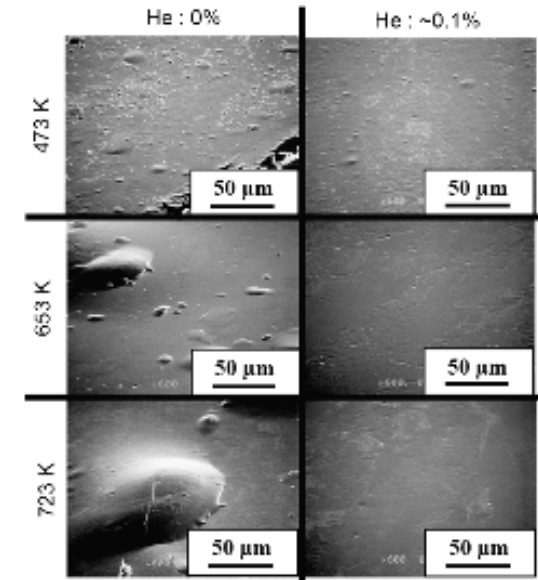
Material Degredation



Kajita, et al. Nucl. Fus. 471, 886-890 (2007)

Tritium Retention

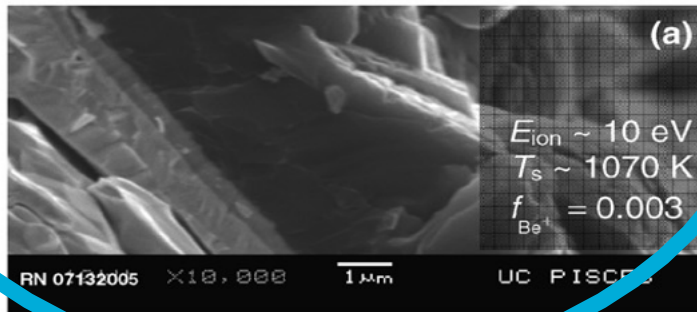
Effect of He on H Blistering



Ueda, et. al. J. Nucl. Mater. 386-388 (2009) 725-728

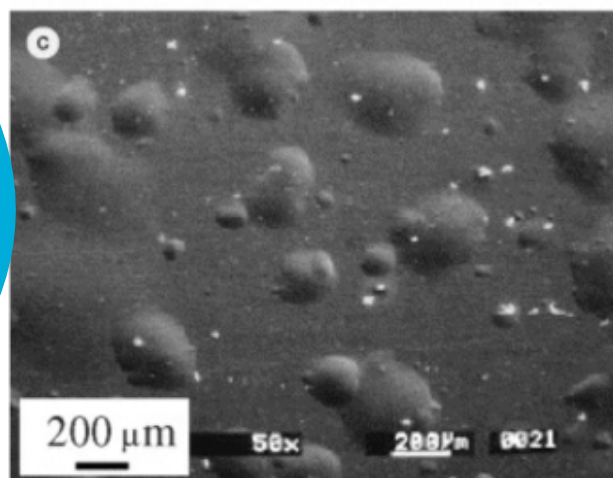
W-Be Intermetallics

Be₁₂W Be deposits (surface)



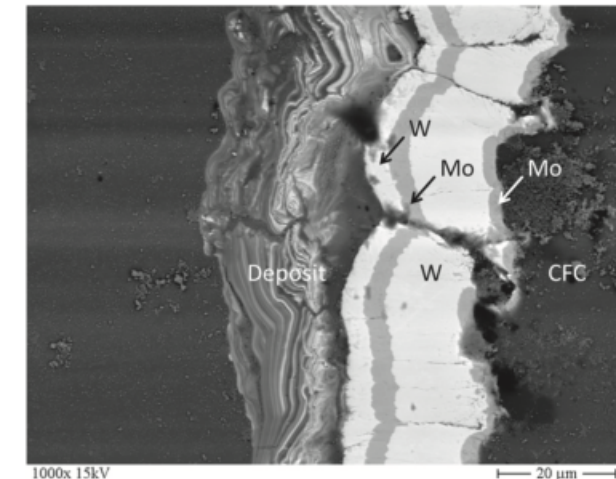
Baldwin, et. al. J. Nucl. Mater. 362-365 (2007) 1179-1183

Hydrogen Blisters



Ye, et al. J. Nucl. Mater. 313-316, 72-76 (2003)

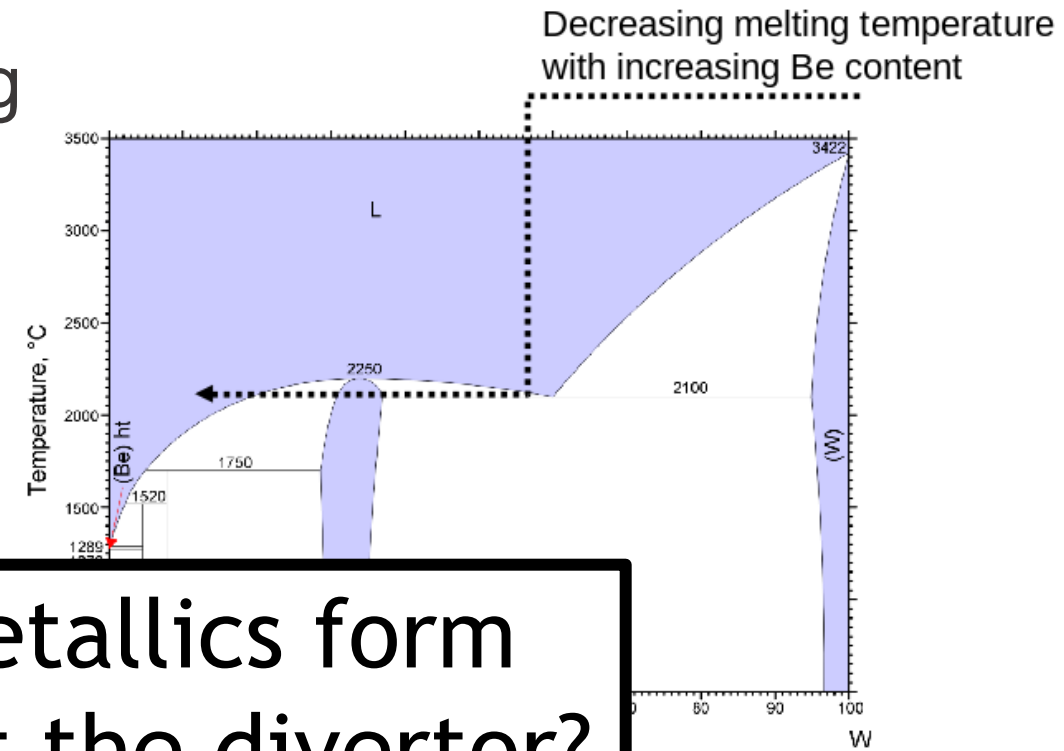
Co-Deposit Layer at Divertor Surface



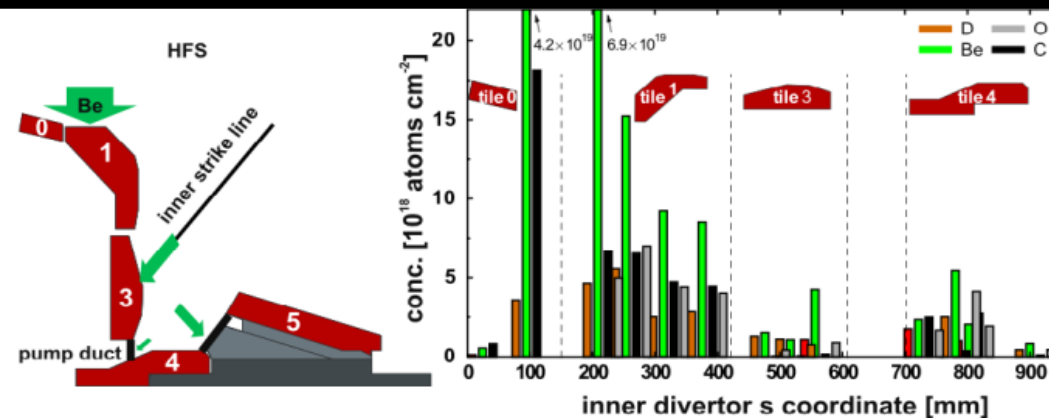
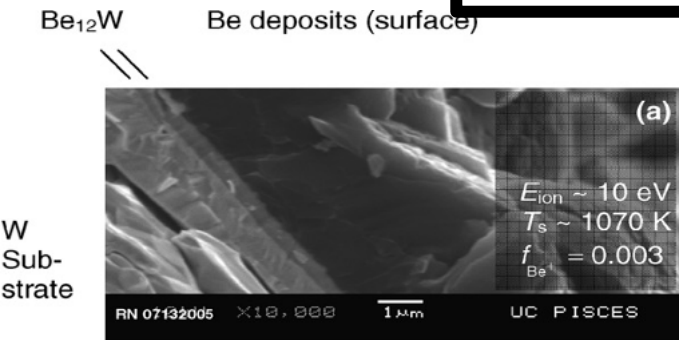
M Mayer et al 2016 Phys. Scr. 2016 014051

4 Beryllium Effect on Tungsten Melting

- W-Be intermetallics observed in linear plasma experiments
- Tokamak experiments indicated beryllium deposition on the divertor
- Phase diagram indicates decreasing melting temperature with increasing Be content

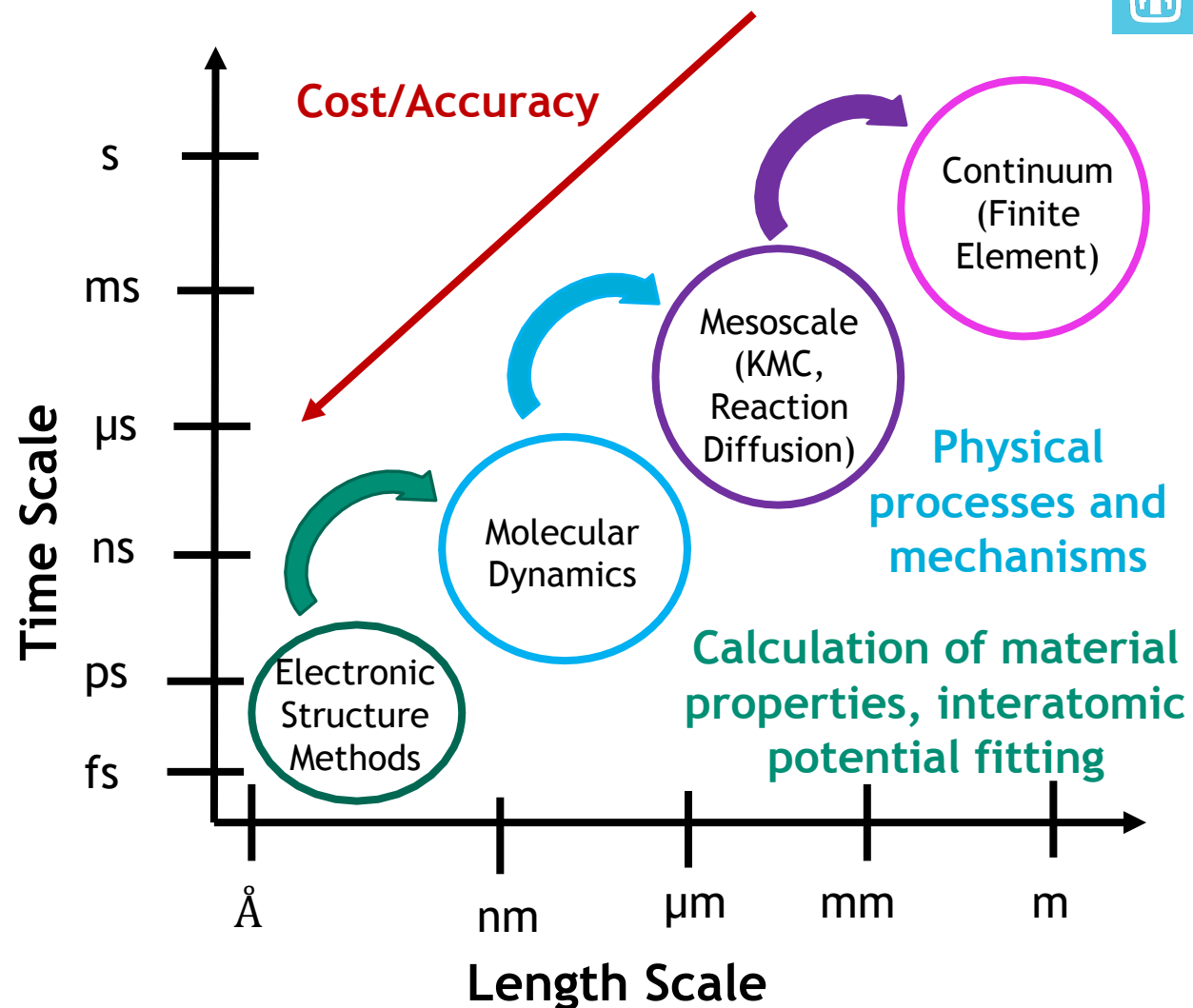
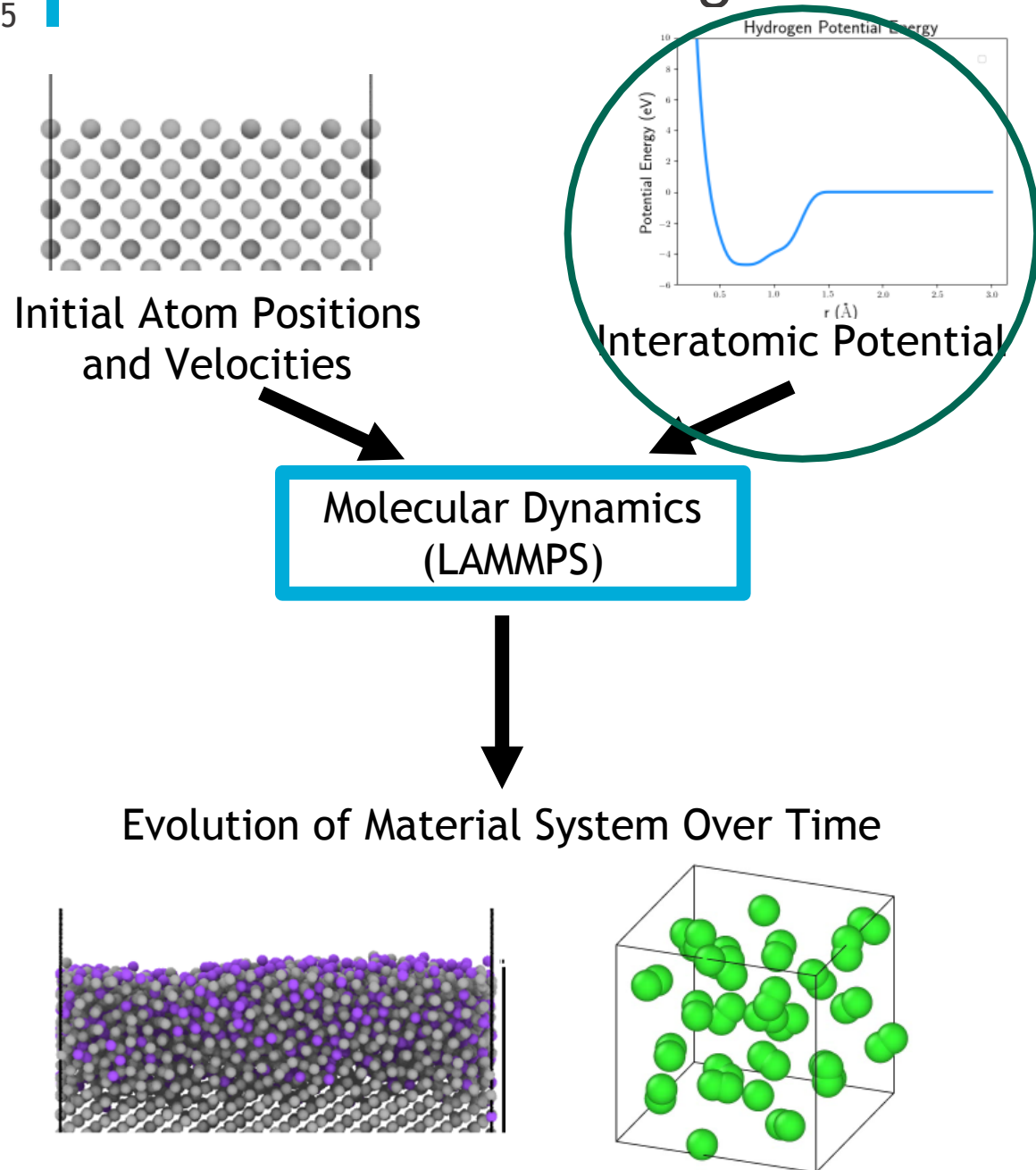


How do W-Be Intermetallics form and how do they affect the divertor?



	MD Melt T. (C)	Expt. Melt T (C)
W	3130	3422
Be	1630	1289
WBe ₂	1830	2250

Multiscale Modeling of Materials



Each simulation technique can provide information to the next scale up

Training Data

- Generated using quantum methods
- Can include:
 - Energies
 - Forces
 - Stresses
- Variety of atomic configurations
 - Bulk structures, liquids, surfaces, defects, etc.

Descriptor

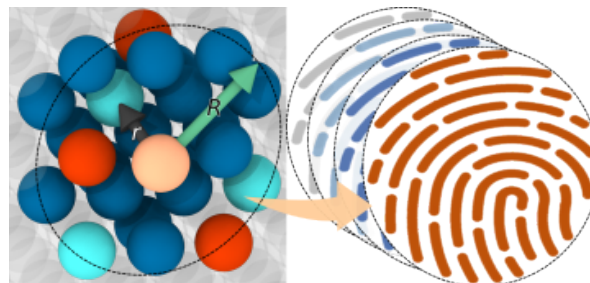
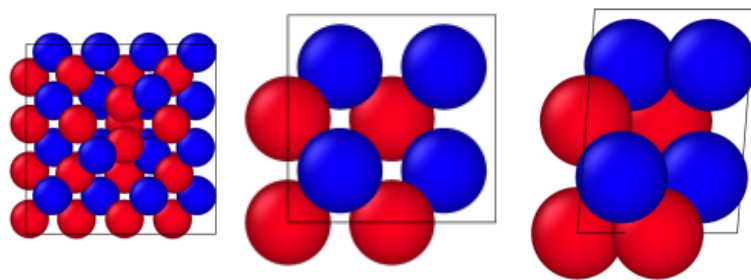
- Describes the local atomic environment
- Requirements
 - Rotation/Translation/. Permutation invariant
 - Equivariant forces
 - Smooth differentiable
 - Extensible
- Some Examples
 - Bispectrum, SOAP, ACE, Moment Tensors, etc.

Regression Method

- Linear regression
- Kernel ridge regression
- Gaussian process
- Non-linear optimization
- Neural Networks

SNAP

- Energies, forces, and stresses from DFT
- Bispectrum component descriptors
- Linear regression



SNAP Definition and Work Flow

Model Form

No
Image

$$E_{SNAP}^i = \beta_0 + \sum_{k=1}^K \beta_k (B_k^i - B_{k0}^i)$$

Regression Method

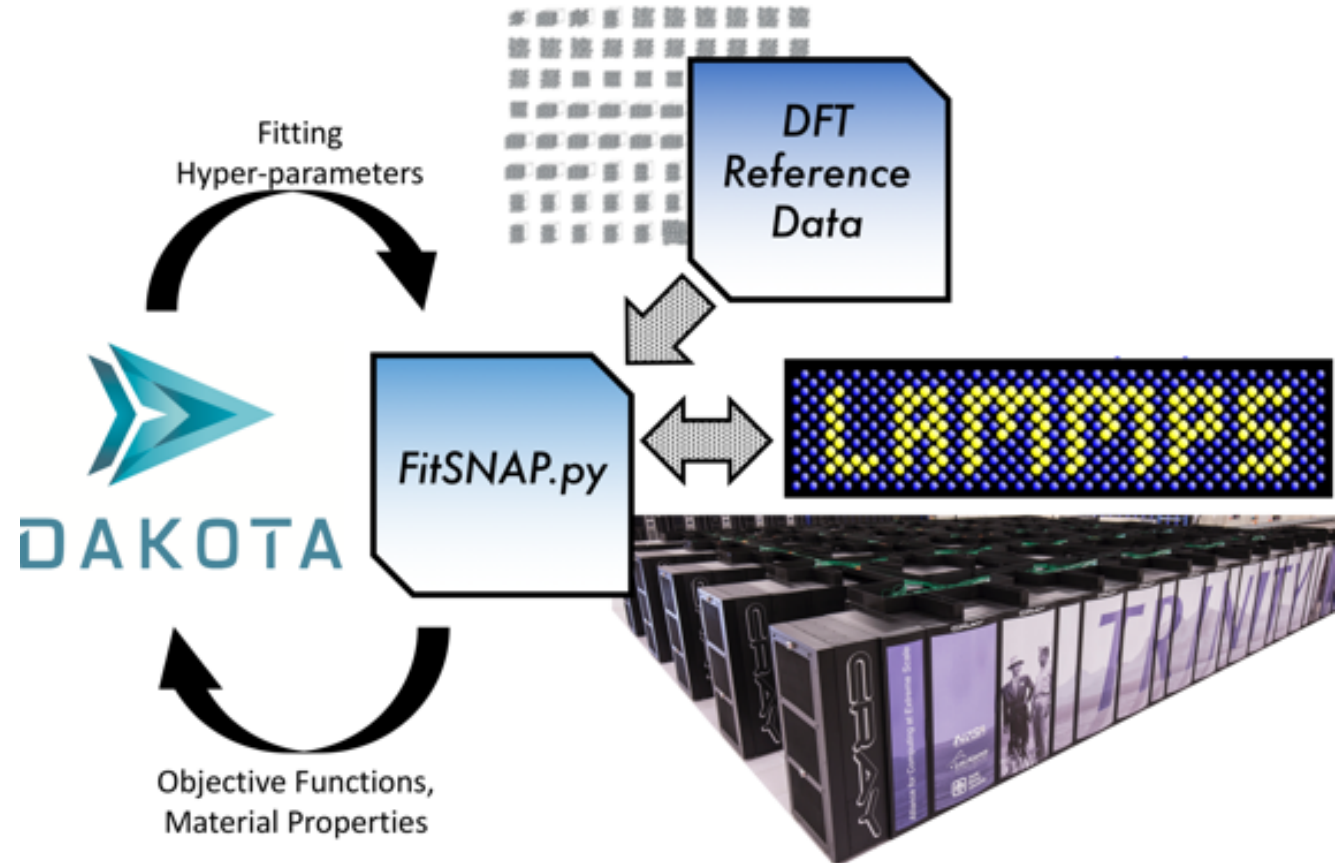
- β vector fully describes a SNAP potential
- Decouples MD speed from training set size

$$\min(\|\mathbf{w} \cdot D\boldsymbol{\beta} - T\|^2 - \gamma_n \|\boldsymbol{\beta}\|^n)$$

Weights

Set of Descriptors

DFT Training



Code available: <https://github.com/FitSNAP/FitSNAP>

Tungsten-Beryllium SNAP Fitting

W-Be Intermetallic Formation Energies (eV)



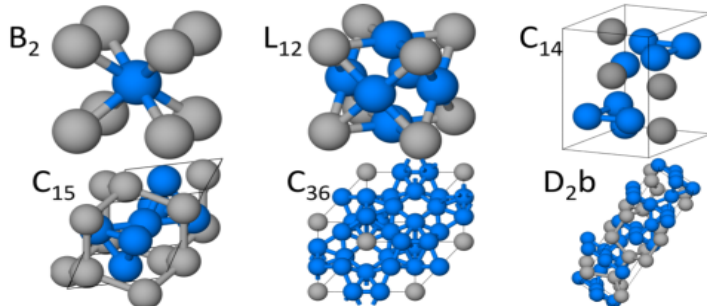
- Initially fit SNAP potential for pure elements
- Making a multi-element SNAP potential does sacrifice some accuracy from either pure component fit.
- Training set includes W-Be intermetallic structures

Description	N_E	N_F	σ_E	σ_F
-------------	-------	-------	------------	------------

W-Be:

Elastic Deform [†]	3946	68040	$3 \cdot 10^5$	$2 \cdot 10^3$
Equation of State [†]	1113	39627	$2 \cdot 10^5$	$4 \cdot 10^4$
DFT-MD [†]	3360	497124	$7 \cdot 10^4$	$6 \cdot 10^2$
Surface Adhesion	381	112527	$2 \cdot 10^4$	$9 \cdot 10^4$

[†] Multiple crystal phases included in this group:



Phase	Composition	DFT ¹	SNAP ¹	BOP ²
B ₂	WBe	0.67	0.30	-2.20
C ₁₄	WBe ₂	-0.87	-1.27	-4.20
C ₁₅	WBe ₂	-0.92	-1.15	-4.19
C ₁₆	WBe ₂	-0.90	-1.22	-4.20
L ₁₂	WBe ₃	-0.51	-0.15	-4.58
D ₂ B	WBe ₁₂	-0.96	-0.34	-6.69

Be Defect Formation Energies in W (eV)

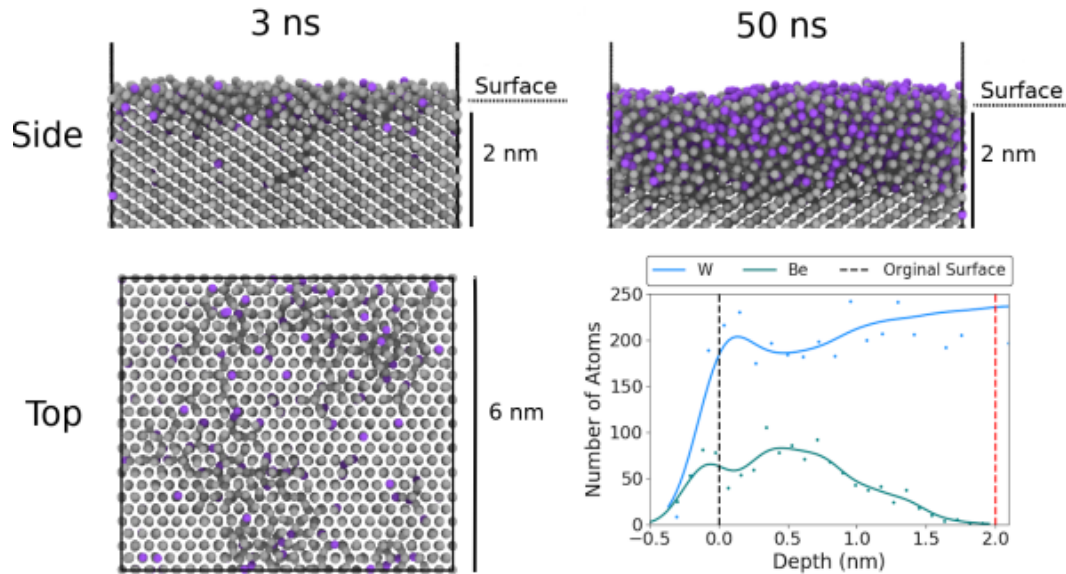
	DFT ¹	SNAP ¹	BOP ²
[111] Dumbbell	4.30	3.66	0.67
Substitution	3.11	3.29	-2.00
Surface Hollow Site	-1.05	-1.39	-3.52
Tetrahedral	4.13	4.20	-0.28
[110] Dumbbell	4.86	4.29	-0.03
Octahedral	3.0	5.11	0.34
Surface Bridge Site	1.01	0.44	-1.30

[1] M. A. Wood, M.A. Cusentino, B.D. Wirth and A.P. Thompson, Phys. Rev. B 99, 184305

[2] C. Björkas et al 2010 J. Phys.: Condens. Matter 22 352206

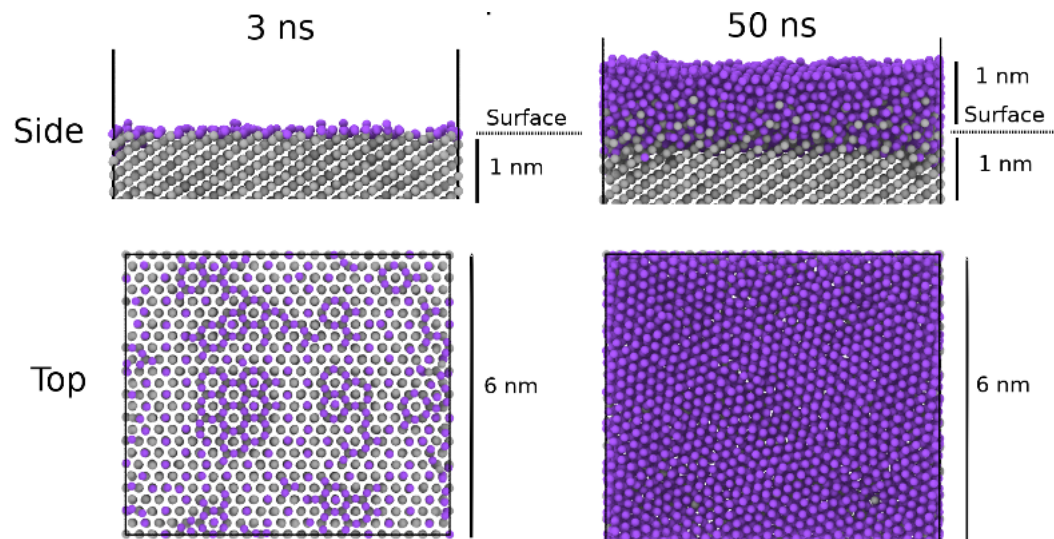
Beryllium Deposition Results in Near Surface Mixed Layer

75 eV Implantation

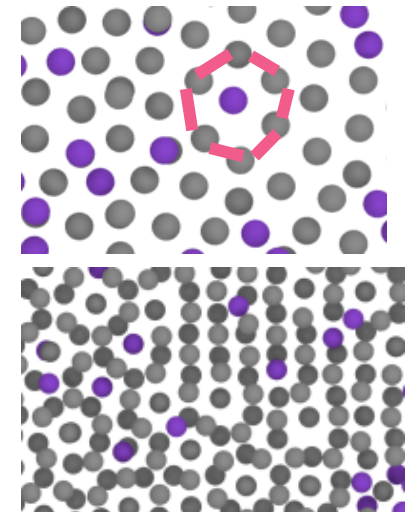
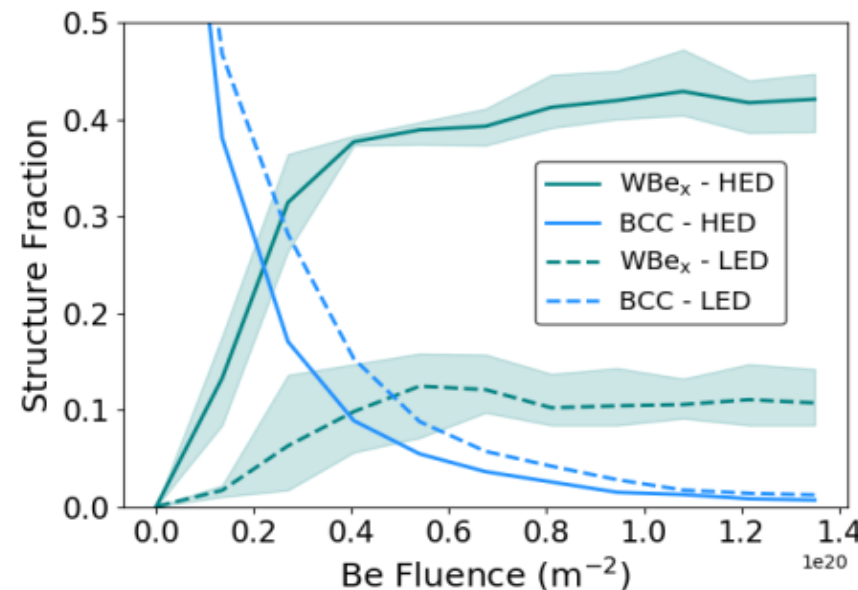


- High energy (75 eV) and low energy (0 eV) beryllium deposition on tungsten surfaces
- Formation of disordered mixed materials layer in first 2 nm of surface
- Some intermetallic growth observed within mixed materials layer
- However, mixed materials layer appears to be kinetically trapped at MD time scales

Athermal Deposition



Intermetallic Growth



Fluence: $1.4 \times 10^{20} \text{ m}^{-2}$

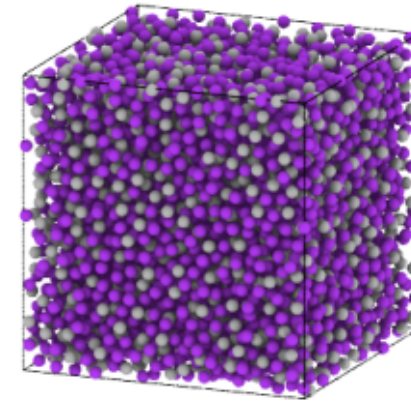
How Does Be Diffuse Within the Mixed Materials Layer?

Purple: Be
Gray: W

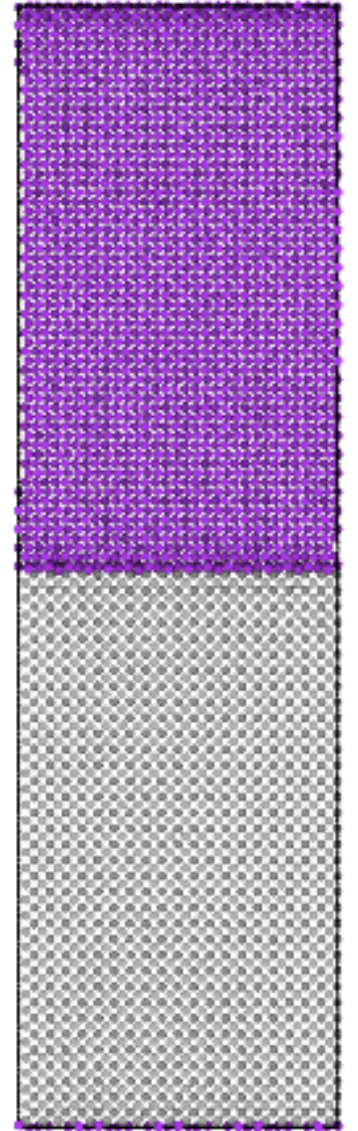


- How quickly does beryllium diffuse within the MML and how does this impact intermetallic formation over longer periods of time?
- Diffusion was found to be limited in the pre-mixed case and higher in the interface
- However, intermetallic growth saturates and further growth is likely beyond MD timescales

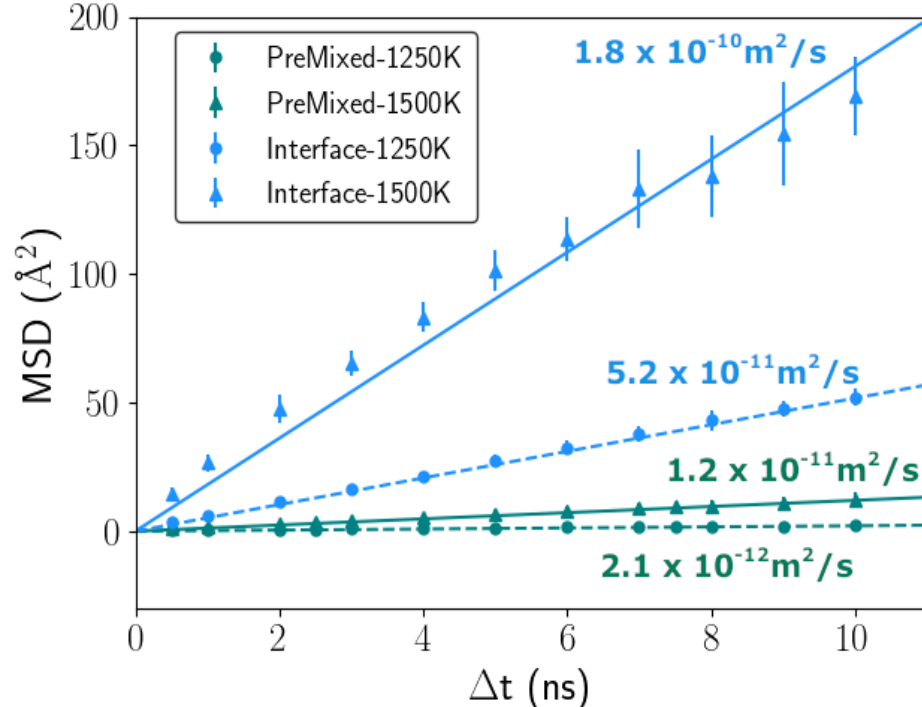
Pre-Mixed W-Be



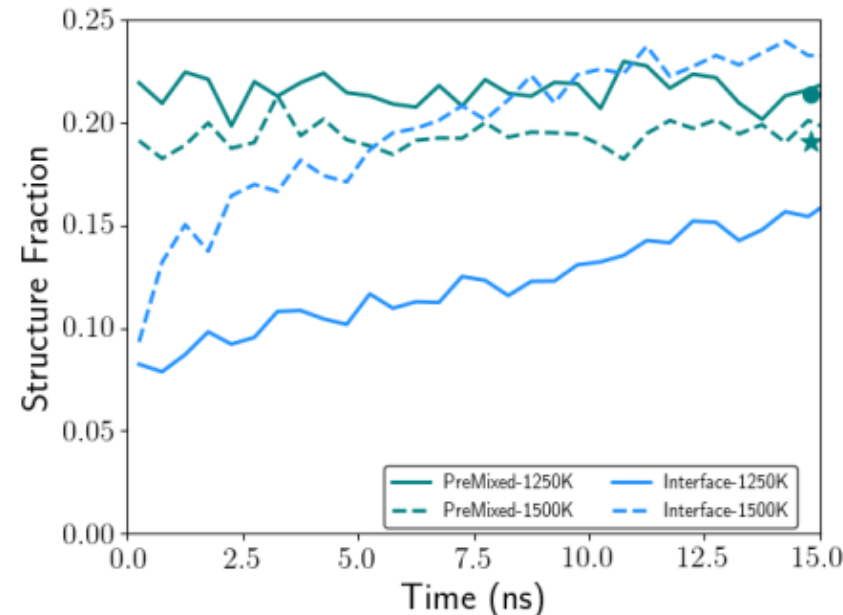
W-Be Interface



Be MSD vs. Time

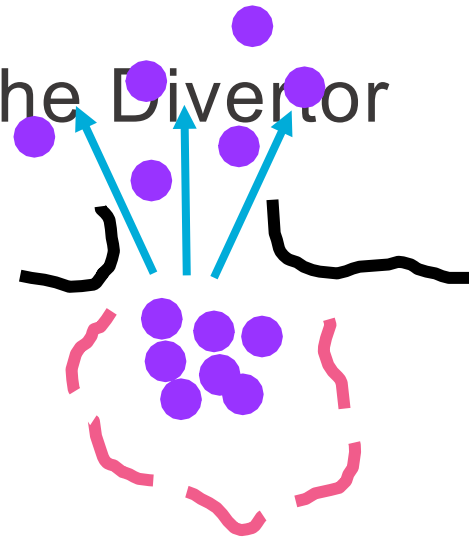
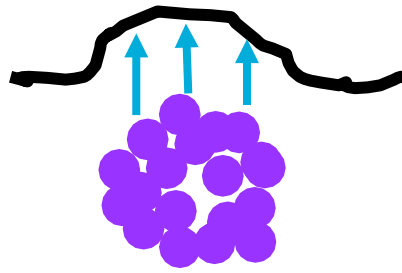
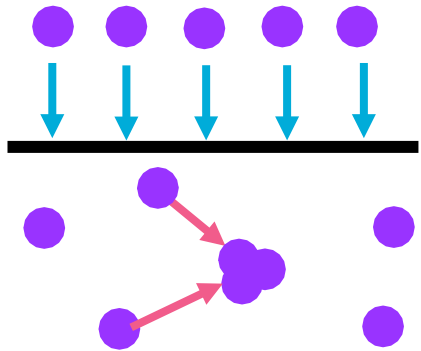


Intermetallic Growth



11

Mixed Material Effects in the Divertor

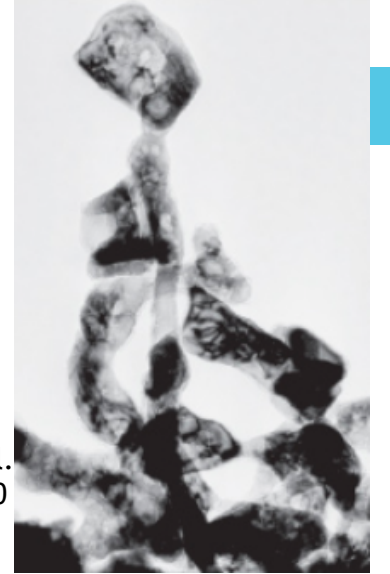


Purple: He Atom

?



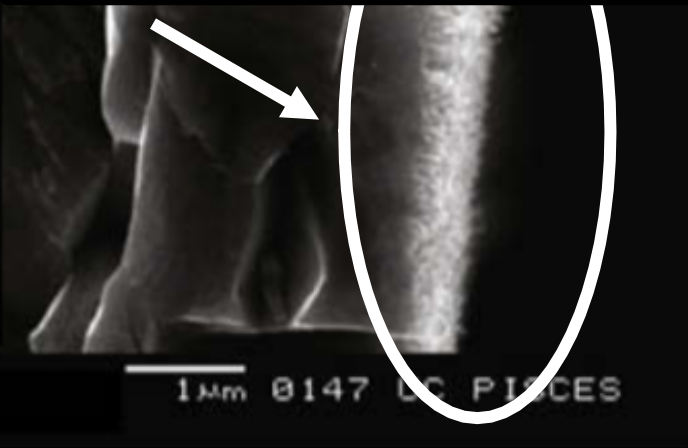
Kajita, et al. Nucl.
Fus. 471, 886-890
(2007)



D₂ - 0.1 He Plasma

W bulk

$E_{\text{ion}} \sim 60 \text{ eV}$
 $T_s \sim 1150 \text{ K}$
 $t = 4200 \text{ s}$
D₂-0.1He



RN 02022007

1mm Ø147 UC PISCES

How does beryllium disrupt the helium fuzz process?

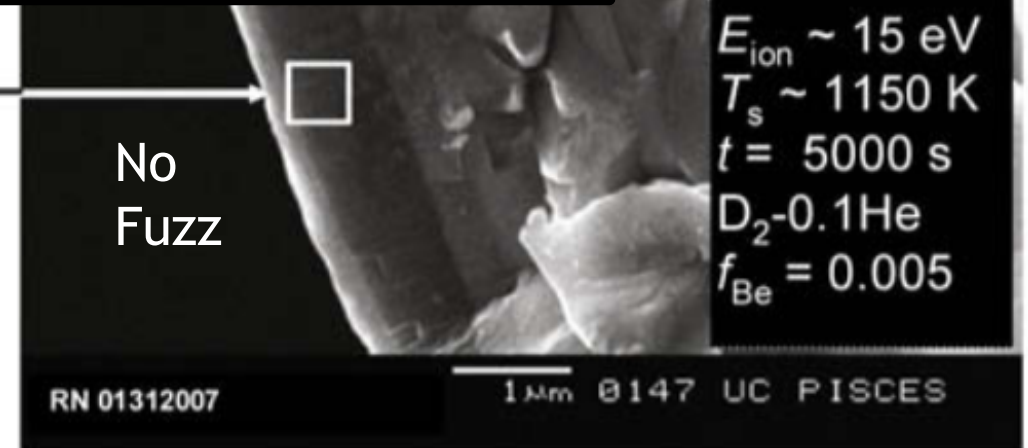
D₂ - 0.1 He with Be Plasma

W bulk

Be₉₃W₇

No
Fuzz

$E_{\text{ion}} \sim 15 \text{ eV}$
 $T_s \sim 1150 \text{ K}$
 $t = 5000 \text{ s}$
D₂-0.1He
 $f_{\text{Be}} = 0.005$



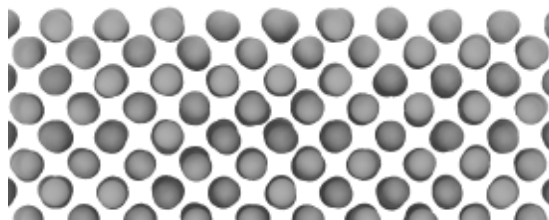
RN 01312007

1mm Ø147 UC PISCES

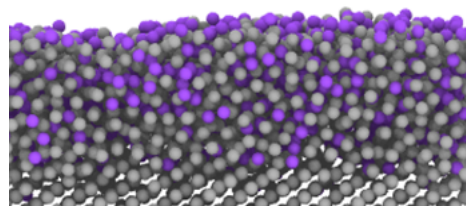


Increasing Time

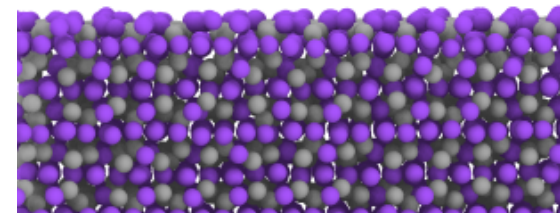
Crystalline W



Amorphous W-Be



Blue: He
Purple: Be
Gray: W

WBe₂ C14 Structure

12 nm

6 nm

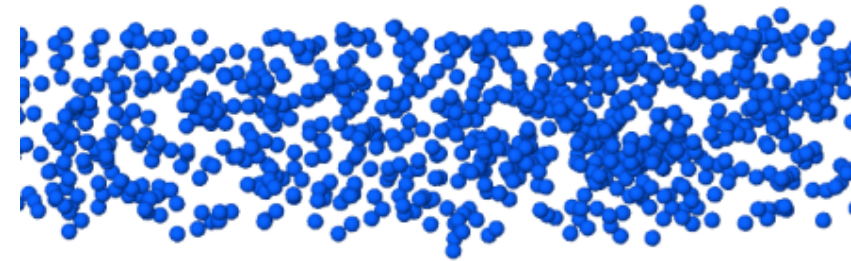
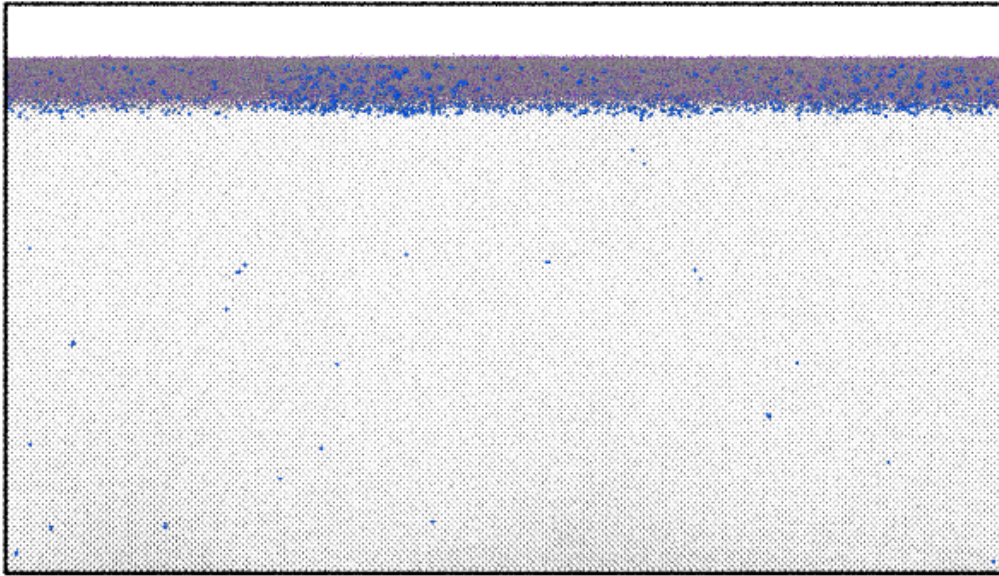
Tungsten:
Larger He clusters
distributed
throughout
simulation cell

Laves/Deposited
Layer:
Smaller He
clusters mostly
located near the
surface

Large Scale Be/He Implantation Simulations in W



He Implantation in Be Pre-Implanted W Surface

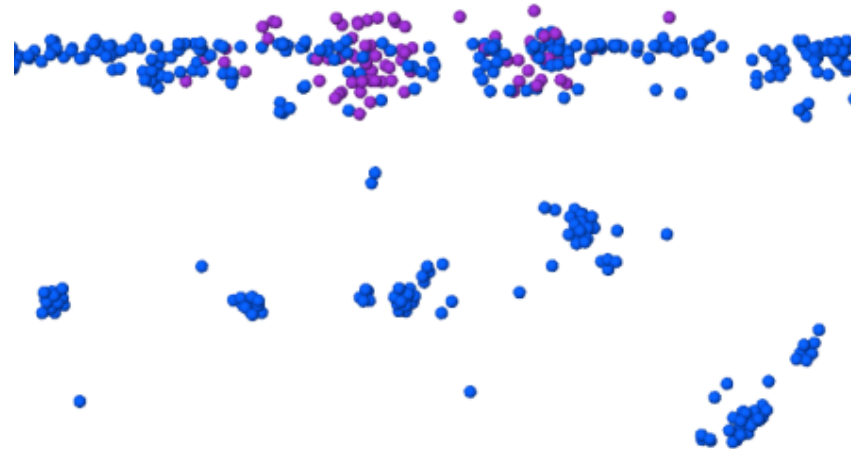
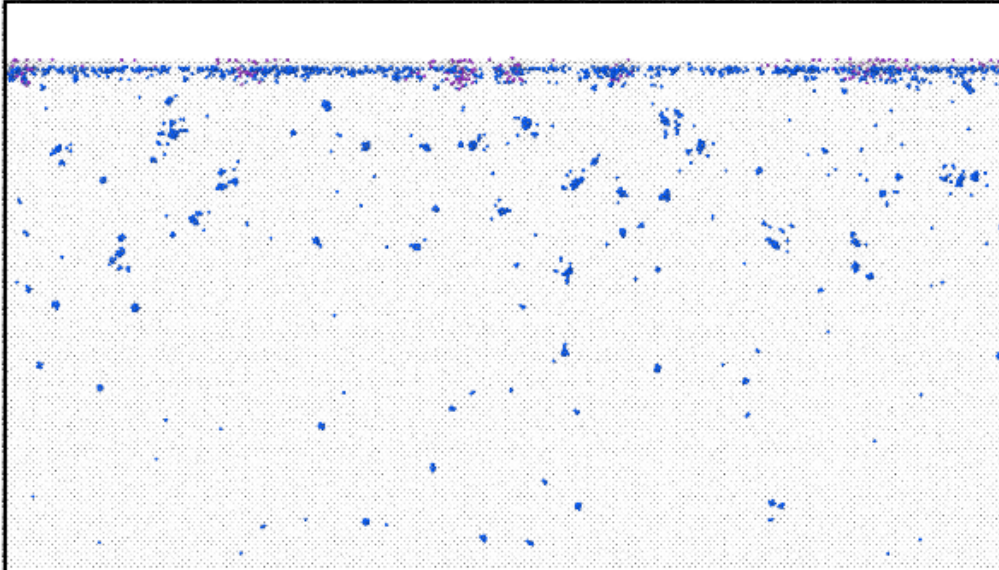


Blue: He
Purple: Be
Gray: W

Time: 90 ns
Fluence:
 $3.56 \times 10^{18} \text{ m}^{-2}$

- Helium resides in W-Be layer
- No clustering of He atoms

90% He/10% Be Implantation onto Clean W Surface

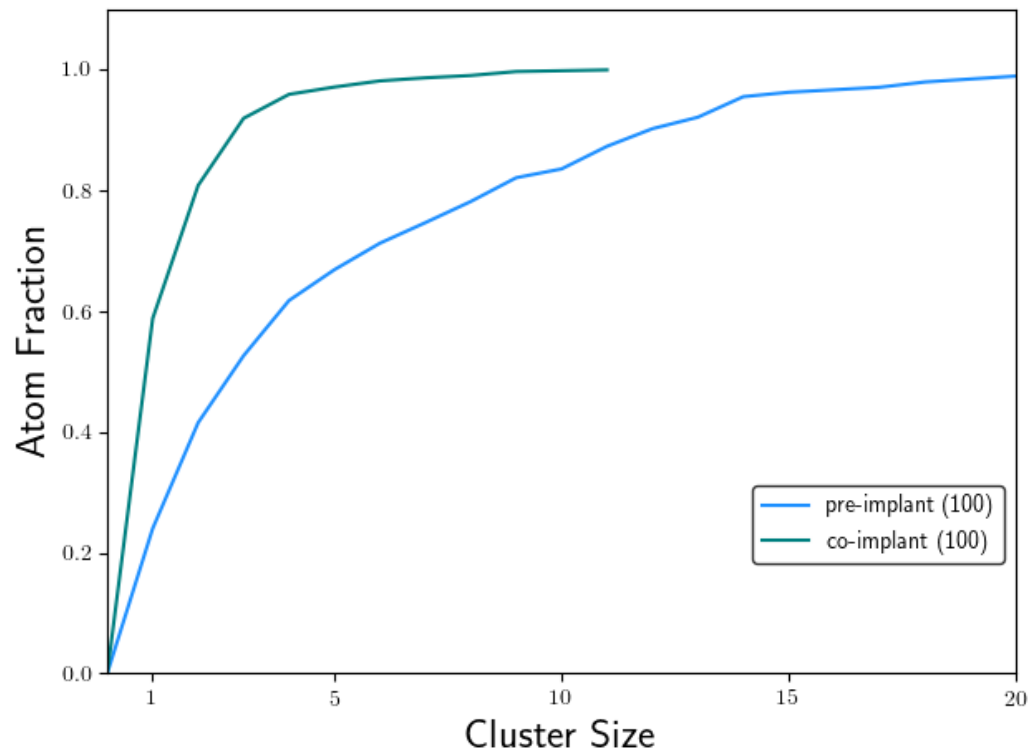


- Helium distributed throughout material
- He bubbles forming
- Be has created some local disorder

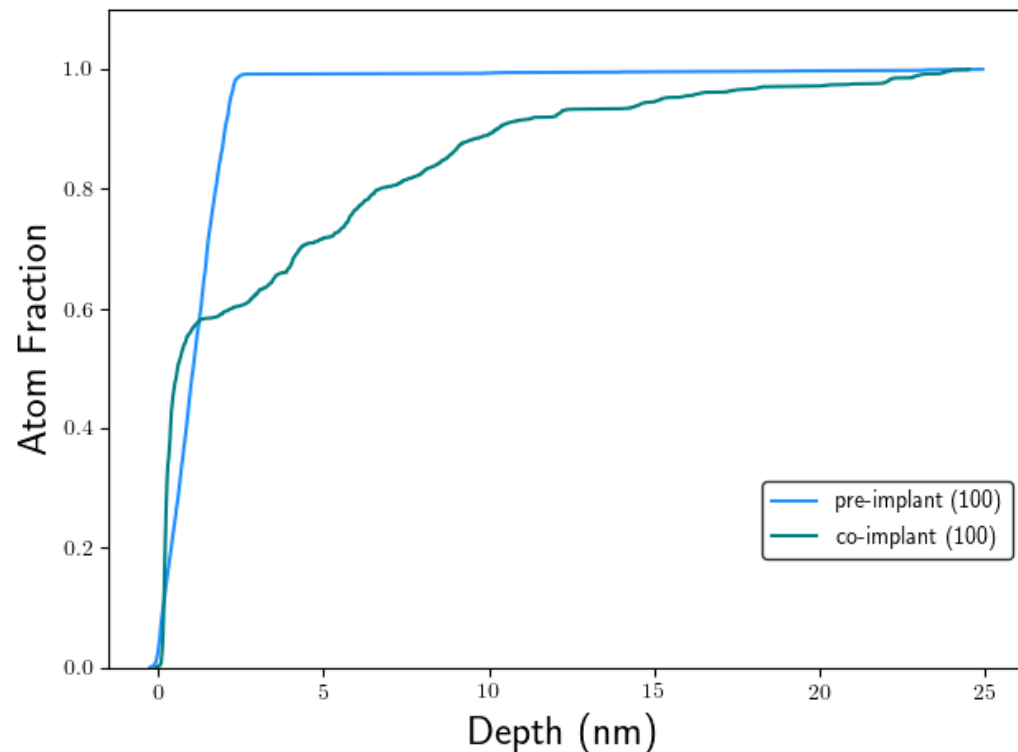
W-Be Mixed Materials Layer Alter He Bubble Nucleation



He Cluster Distribution



He Depth Profile



	Pre-(100)	Co-(100)
% Retention	46.4	23.1

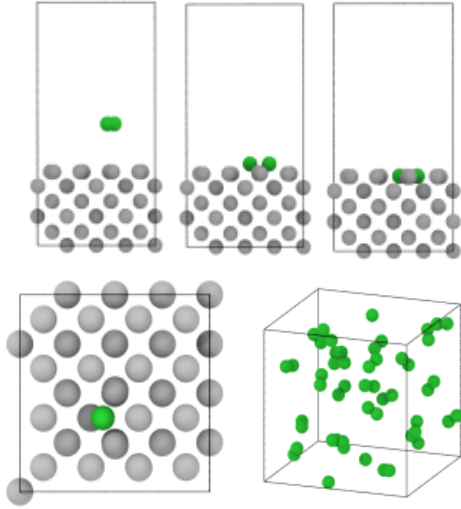
m ² /s	W	WBe ₂
Diffusion Coefficient	1.2 x 10 ⁻⁸	1.8 x 10 ⁻¹¹

- W-Be mixed materials layer results in:
 - Smaller helium clusters
 - Helium remaining near the surface
 - Higher helium retention
 - Slower helium diffusion
- This will results in changes to helium bubble nucleation and will likely affect fuzz growth

Next Steps: Incorporating Other Plasma Species in SNAP



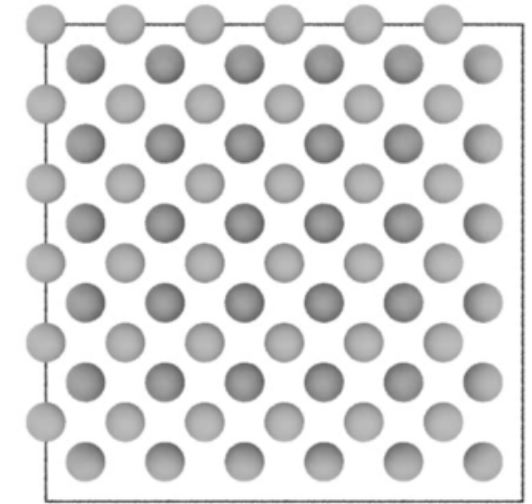
Hydrogen



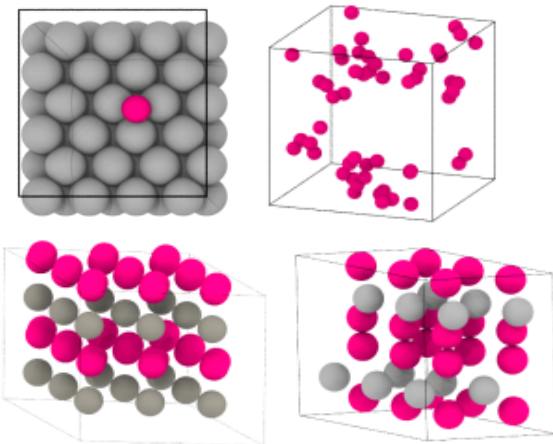
H Defect Formation Energies

H Defect _x	DFT	SNAP
E_f^{Tet} (eV)	0.88	0.74
E_f^{Oct} (eV)	1.26	1.06
E_f^{Sub} (eV)	4.08	4.05
E_f^{H2} (eV)	-4.75	-4.77

H Diffusion in Bulk W



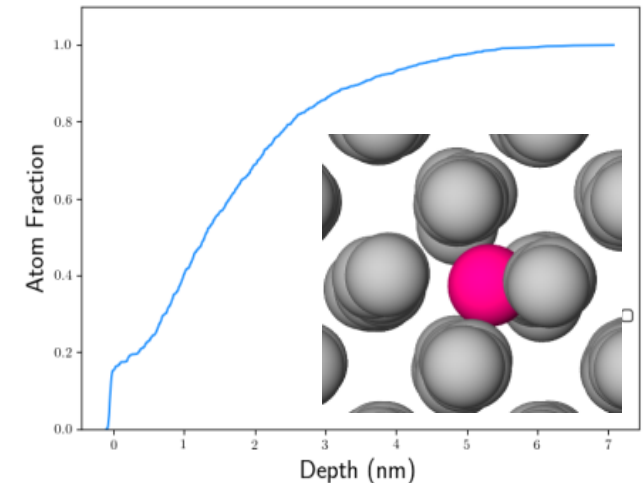
Nitrogen



W_xN_y Formation Energies

	DFT (eV)	SNAP (eV)
WN_2 - P62mmc	-1.82	-2.17
WN_2 - P6m2	-0.91	-1.91
WN - NiAs	-0.84	-1.36
WN - WC	-0.23	-0.77
W_2N	-0.03	0.7

75 eV N Depth Profile



Summary

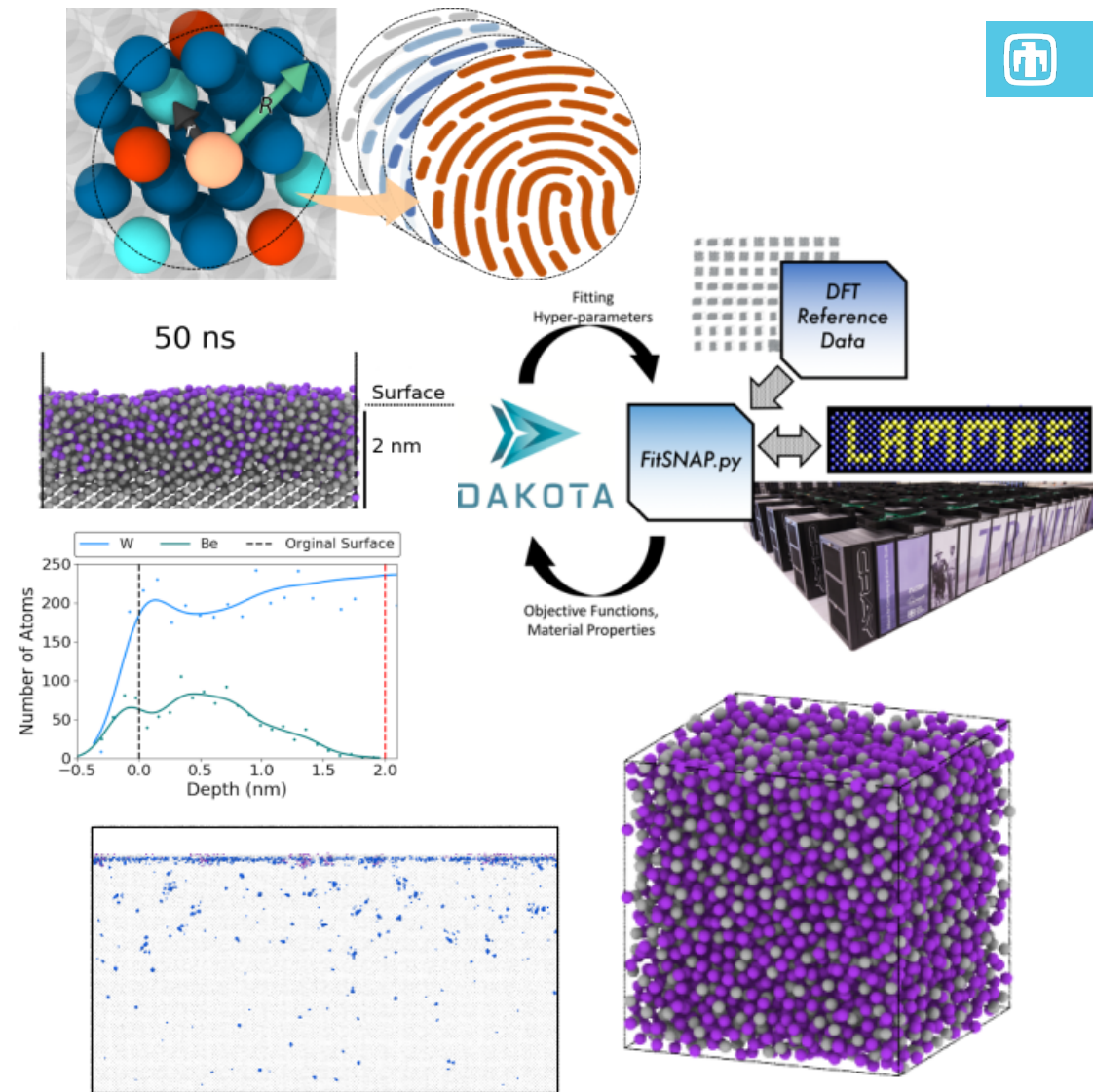
- Understanding material degradation in PFM is critical for designing viable fusion reactors
- Atomistic modeling plays a key role in understanding relevant physical mechanisms for material degradation at the divertor
- A W-Be SNAP potential has been developed and used to study Be implantation in W and extended to simulation He implantation W-Be materials
- Beryllium will quickly form a mixed material layer in the near surface region which has some intermetallic structure
- The W-Be mixed layer alters helium diffusion, cluster formation, and depth profile
- Future work entails the development of a W-Be-H-He-N potential for simulation of interactions

Contact:

mcusent@sandia.gov



Office of
Science





MD Approximations Change Over Time



Twobody (B.C.)

Lennard-Jones, Hard
Sphere, Coulomb,

Rounded

Manybody (1980s)

Stillinger-Weber,
Tersoff, Embedded

Atom Method
GPU Timings

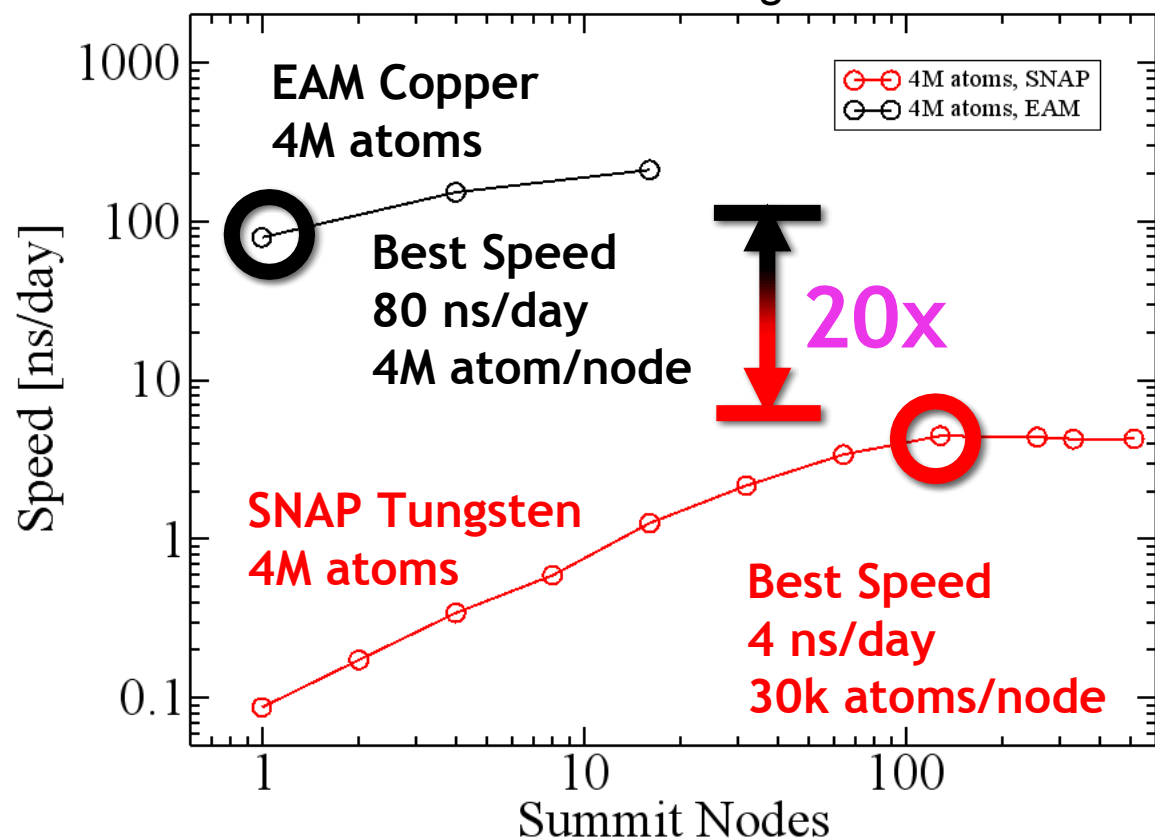
Advanced (90s- 2000s)

REBO, BOP, COMB,

ReaxFF

Big Data / Deep / Machine Learning (2010s)

GAP, SNAP, NN,...



Gayatri, Moore *et al.* (2020) <https://arxiv.org/abs/2011.12875>

Resources are limited, which is your best choice?

