

# CINT

The Center for Integrated Nanotechnologies

*Nanomaterials*

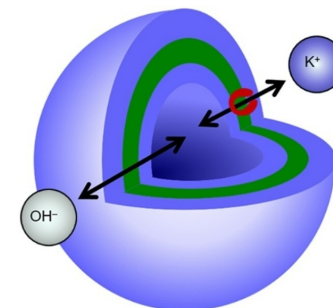
*Integration*

A U.S. DOE Nanoscale Science Research Center

## Monitoring and Modulating Ion Traffic in Hybrid Lipid/Polymer Vesicles

**Walter F. Paxton, Patrick T. McAninch, Komandoor E. Achyuthan, Sun Hae Ra Shin, Haley L. Monteith**  
(CINT / SNL)

255<sup>th</sup> ACS National Meeting  
New Orleans, LA / March 19, 2018

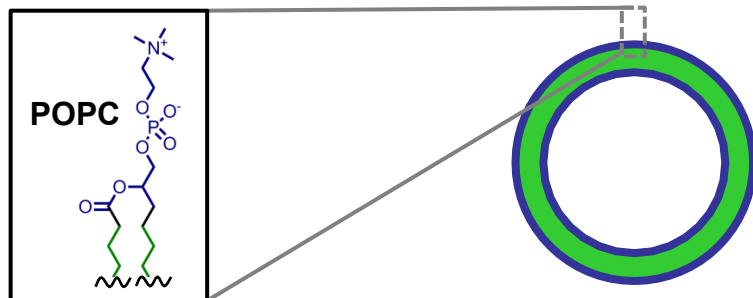


This work was performed in part, at the Center for Integrated Nanotechnologies, an Office of Science User Facility operated for the U.S. Department of Energy (DOE) Office of Science (project number 2016AU0018). Sandia National Laboratories is a multimission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC., a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA-0003525.



# Mimicking Biological Membranes

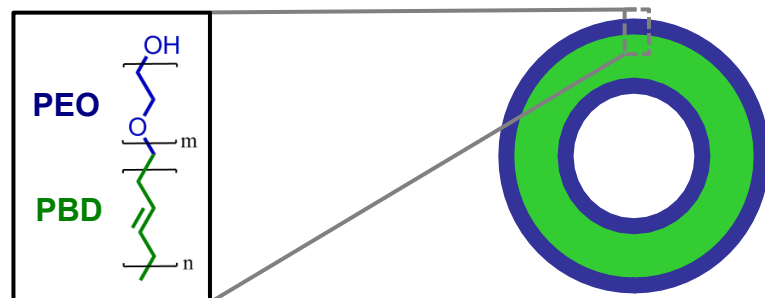
## Liposomes



### 2 Major Challenges:

Limited Chemical and Mechanical Stability  
Limited Modification Chemistries

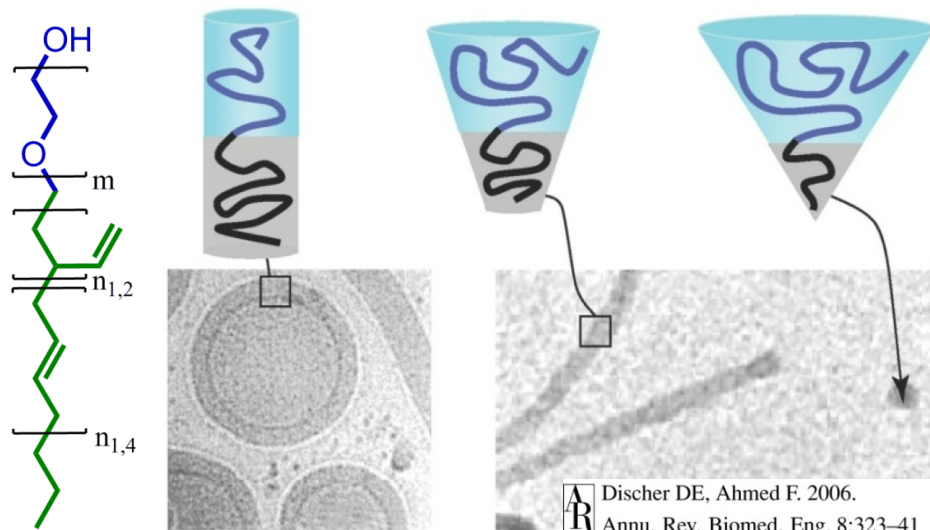
## Polymersomes



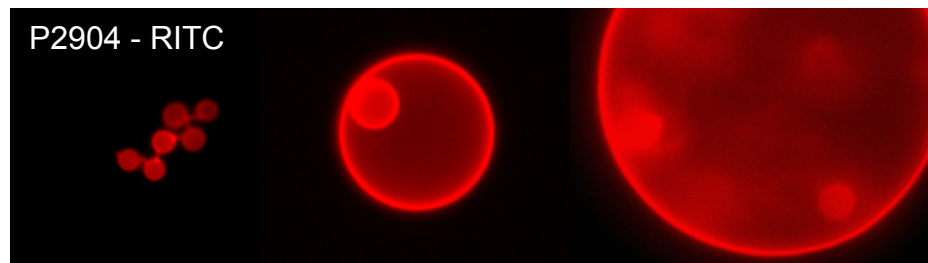
### Polymersomes Can Help

Enhanced Chemical and Mechanical Stability  
Unlimited Modification Chemistries

Can we incorporate or mimic **properties** and **functions** of biological cells to create robust **advanced materials**?



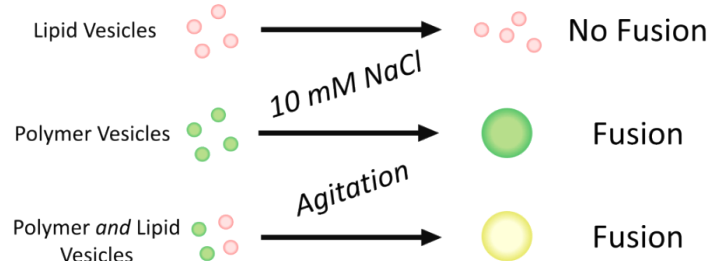
P2904 - RITC





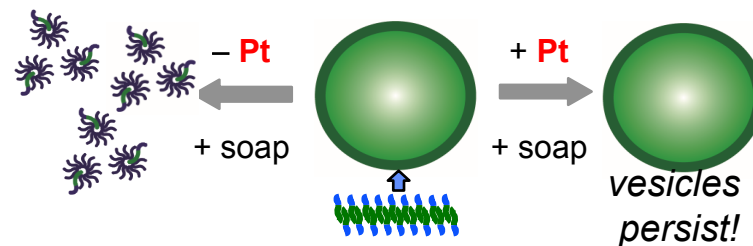
# Dynamic Polymer Vesicle Membranes

## Mechanically-Activated *Fusion*



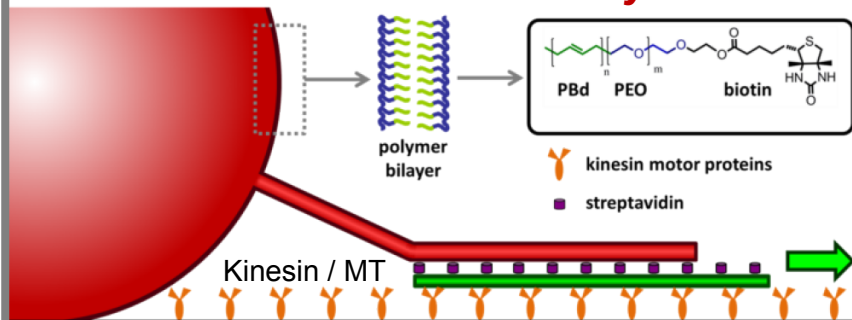
Angew. Chem. – Int. Ed. **2014**, 53, 3372–3376; J. Poly. Sci. B, **2014**, 53, 297–303

## Catalytically-Active Cross-Links (*Reactivity and Stability*)



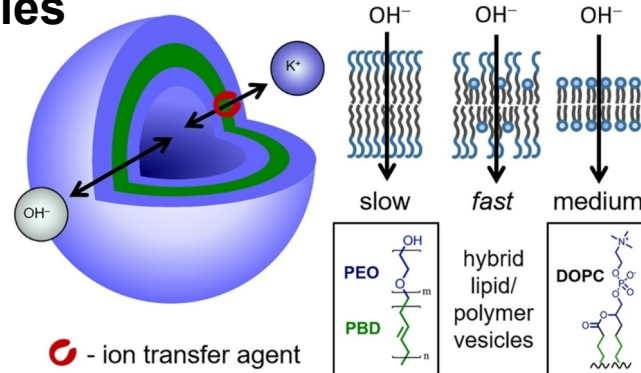
Chem. Mater, **2015**, 27, 4808–4813

## Dynamic Assembly of Polymer Nanotubes – *Fluidity*

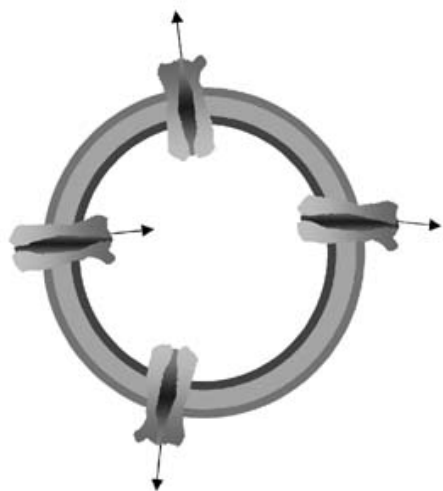


Nanoscale, **2015**, 7, 10998–11004

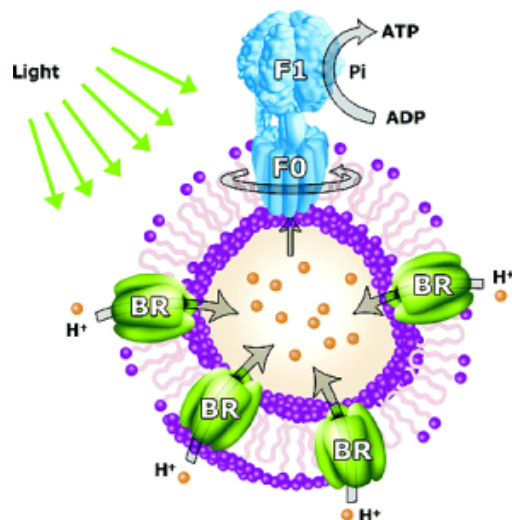
## Modulating *Permeability* in Hybrid Vesicles



Colloids and Surfaces B, **2017**, 159, 268–276

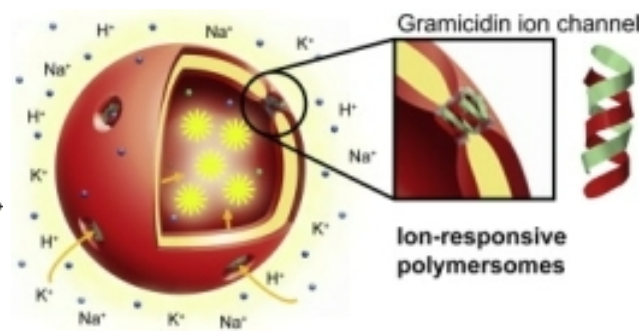


Aquaporin (Meier et al, 2004)



Bacteriorhodopsin

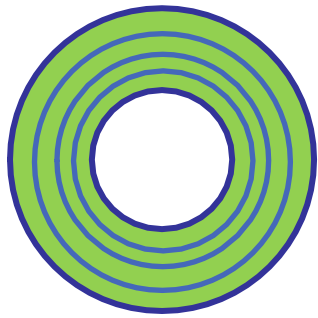
F1-ATP-ase (Montemagno et al, 2005)



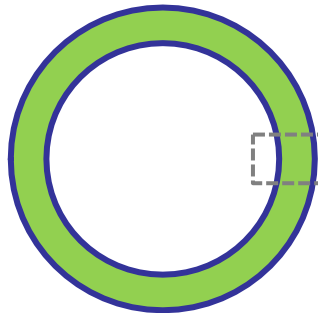
Gramicidin A (Palivan et al, 2015)

# Vesicle Permeability Model

Multilamellar Vesicles

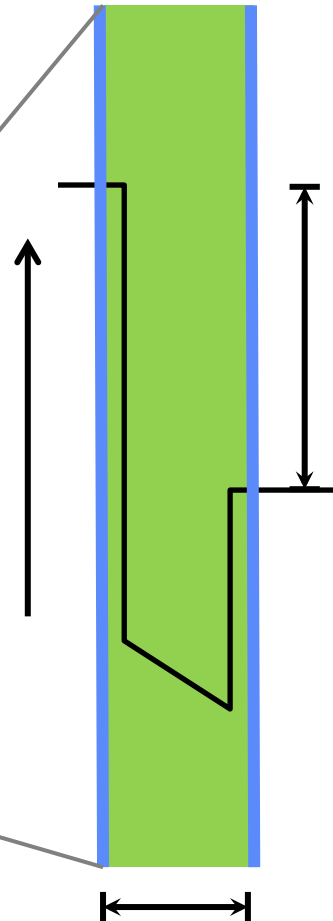


extrusion



Unilamellar Vesicles

relative solute conc.



Solubility-Diffusion Mechanism

$$P = \frac{DK}{d} = \frac{J}{\Delta c_w}$$

**$P$**  = permeability coeff.

**$D$**  = diffusion coeff.

**$K$**  = partition coeff.

**$d$**  = bilayer thickness

**$J$**  = solute flux

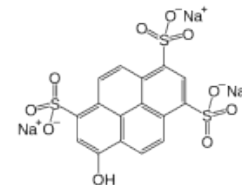
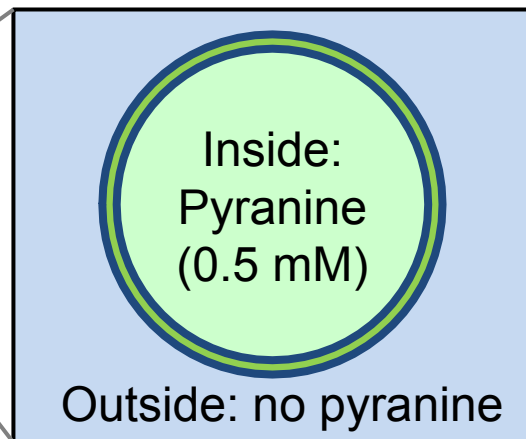
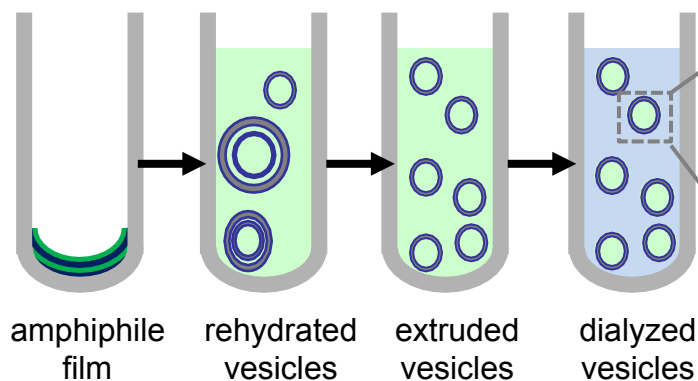
**$\Delta c_w$**  = solute gradient





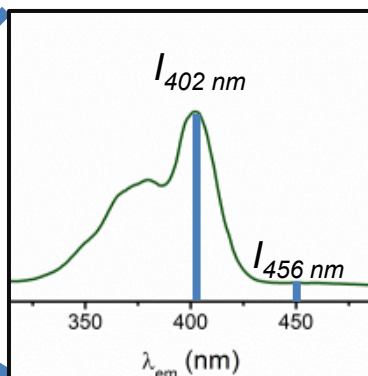
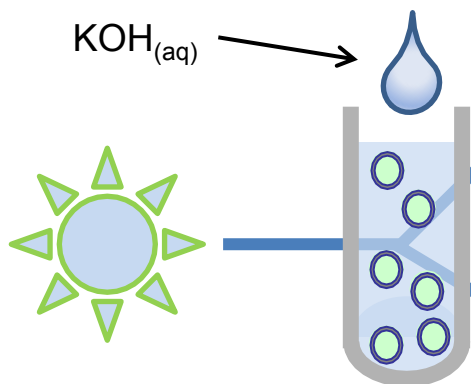
# Ion Flux Monitored by Fluorescence

## Vesicle Preparation



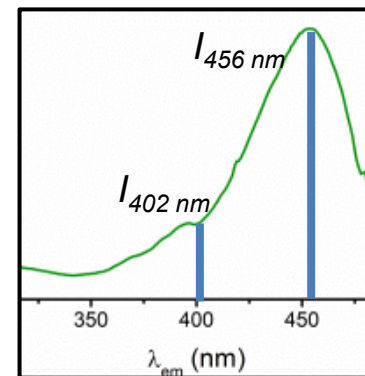
~200 nm vesicles  
("unilamellar")

pH = 7  
(PIPES/Sucrose)

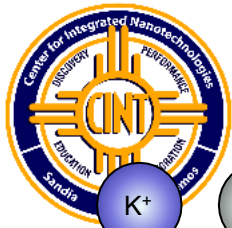


$$\frac{I_{456 \text{ nm}}}{I_{402 \text{ nm}}} \rightarrow \text{pH}_{t_1}$$

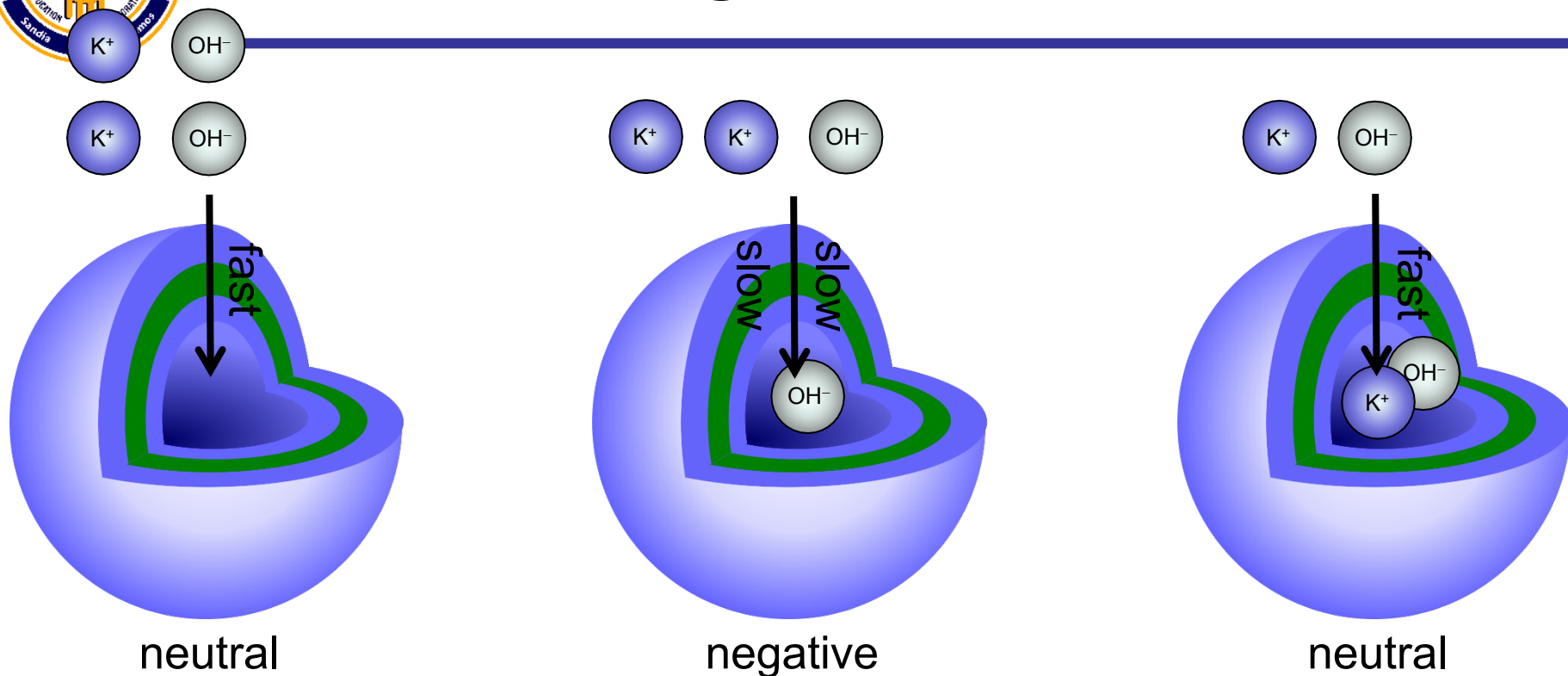
time  
→



$$\frac{I_{456 \text{ nm}}}{I_{402 \text{ nm}}} \rightarrow \text{pH}_{t_2}$$



# Measuring Ion Flux in Vesicles

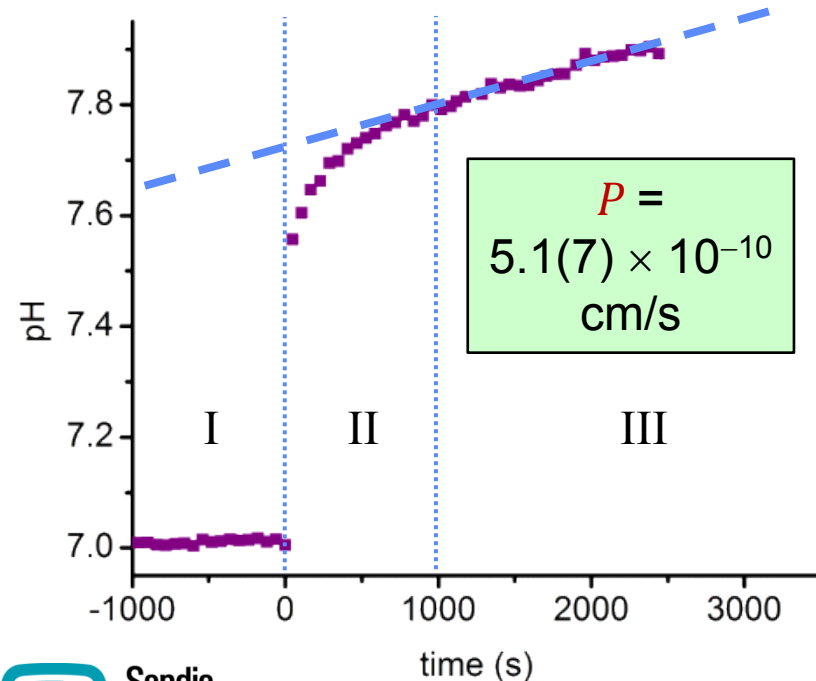
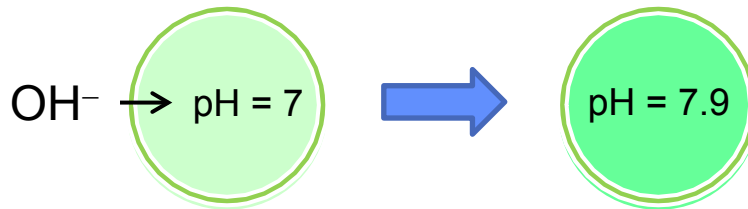


1. Permeability of  $\text{OH}^-$  is FAST  $\rightarrow$  buildup of negative charge
2. Negative charge compensated by flux of the  $\text{K}^+$  counterion...
3. ...but permeability of  $\text{K}^+$  is SLOW.
4.  $\text{K}^+$  flux is the rate limiting step for net flux of  $\text{OH}^-$

***pH can be used to determine flux of  $\text{K}^+$***

# Calculating Lipid Permeability

DOPC



$$\frac{\Delta n_{OH^-}}{\Delta t} = f([OH^-])$$

via  
Henderson-  
Hasselbalch

$$J_{OH^-} = \frac{\Delta n_{OH^-}}{\Delta t} \times \left( \frac{1}{S_{ave}} \right)$$

OH<sup>-</sup> flux  
across a  
membrane

$$P = \frac{J}{\Delta c_w}$$

Effective permeability  
coeff.

$$D^* = DK = \frac{J_{OH^-}}{\Delta c_w}$$

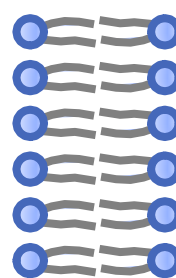
Effective diffusion coeff.



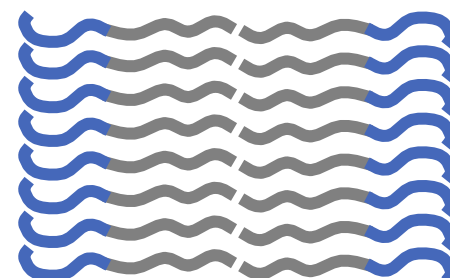
# Vesicle Permeability Summary

| Sample                             | Vesicle diameter (nm) <sup>a</sup> | Membrane Thickness (nm) | $P_{OH^-}$ ( $\times 10^{10}$ cm/s) | $D^*$ ( $\times 10^{16}$ cm <sup>2</sup> /s) |
|------------------------------------|------------------------------------|-------------------------|-------------------------------------|--|
| DOPS                               | 170(30)                            | 2.6 <sup>b</sup>        | 5.4(8)                              | 1.4(2)                                       |
| DOPC                               | 190(30)                            | 2.7 <sup>c</sup>        | 5.1(7)                              | 1.4(2)                                       |
| EO <sub>20</sub> BD <sub>33</sub>  | 200(40)                            | 6.8 <sup>d</sup>        | 1.5(3)                              | 1.0(2)                                       |
| EO <sub>89</sub> BD <sub>120</sub> | 260(70)                            | 22 <sup>d</sup>         | 0.4(1)                              | 0.9(2)                                       |

Permeability  $\leftrightarrow$  Thickness

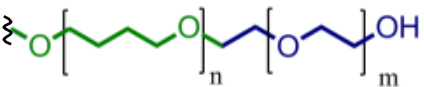
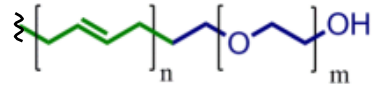
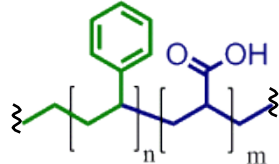


vs.



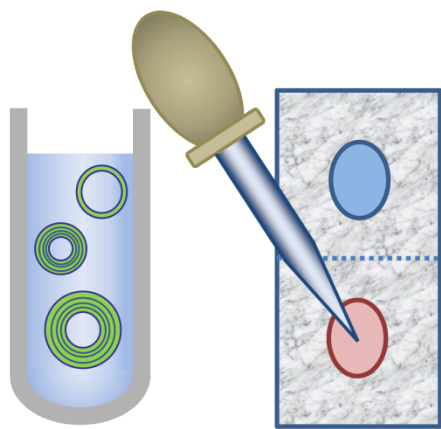
<sup>a</sup> DLS. <sup>b,c</sup> Literature values. <sup>d</sup> Estimated from published models.

Permeability  
 $\updownarrow$   
Chemical  
Composition

| Polymer  | Thickness (nm) | $P_{A^-}$ (cm/s) | $D^*$ (cm <sup>2</sup> /s) |
|--|----------------|------------------|----------------------------|
| PEO-PBO <sup>a</sup>  | 2.4            | $10^{-7}$        | $10^{-14}$                 |
| PEO-PBD              | 6.8            | $10^{-9}$        | $10^{-16}$                 |
| PS-PAA <sup>b</sup>  | 33             | $10^{-13}$       | $10^{-18}$                 |

<sup>a</sup> Battaglia et al. **2006**; <sup>b</sup> Eisenberg et al. **2006**

# Large Persistent $\Delta\text{pH}$ in Polymersomes

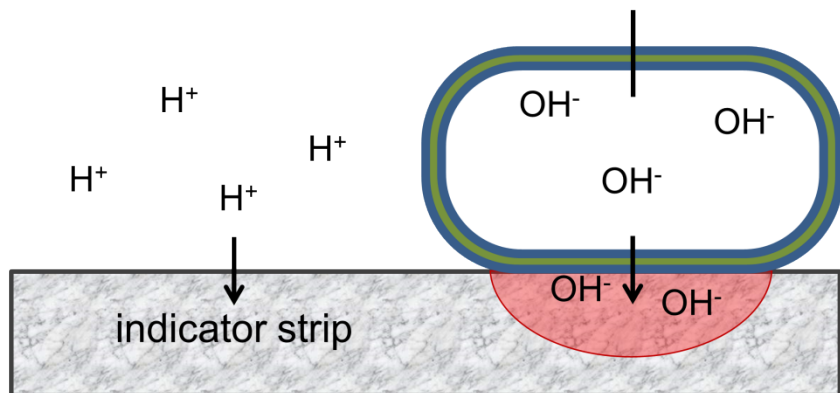


thymolphthalein  
( $\text{pK}_a \sim 10$ )

phenolphthalein  
( $\text{pK}_a \sim 9$ )

$\text{pH}_{\text{external}} = \text{low}$

$\text{pH}_{\text{internal}} = \text{high}$



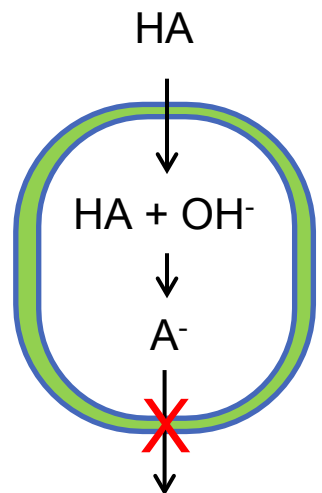
| EO <sub>20</sub> BD <sub>33</sub><br>1 M KOH | +3 eq H <sup>+</sup><br>H <sub>3</sub> PO <sub>4</sub> | After<br>3 weeks | +0.1%<br>TX100 |
|--|--|------------------|----------------|
|  |  |                  |                |
|  |  |                  |                |
| $\text{pH}_{\text{internal}} 14$             | $>10$  | $>10$            | 4.2            |
| $\text{pH}_{\text{external}} 14$             | 3.8  | 3.8              | 4.2            |

- OH<sup>-</sup>/H<sup>+</sup> gradients up to 6 orders...
- ...that Persist for several weeks

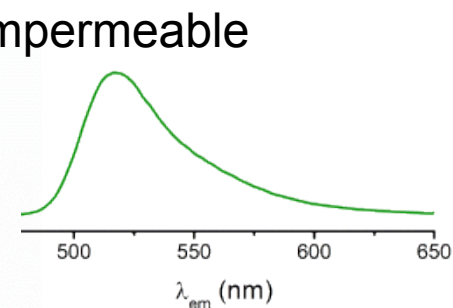
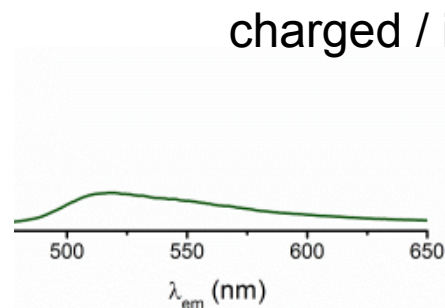
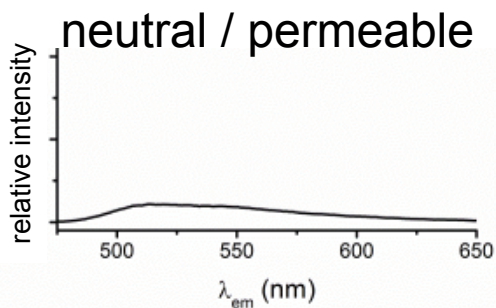
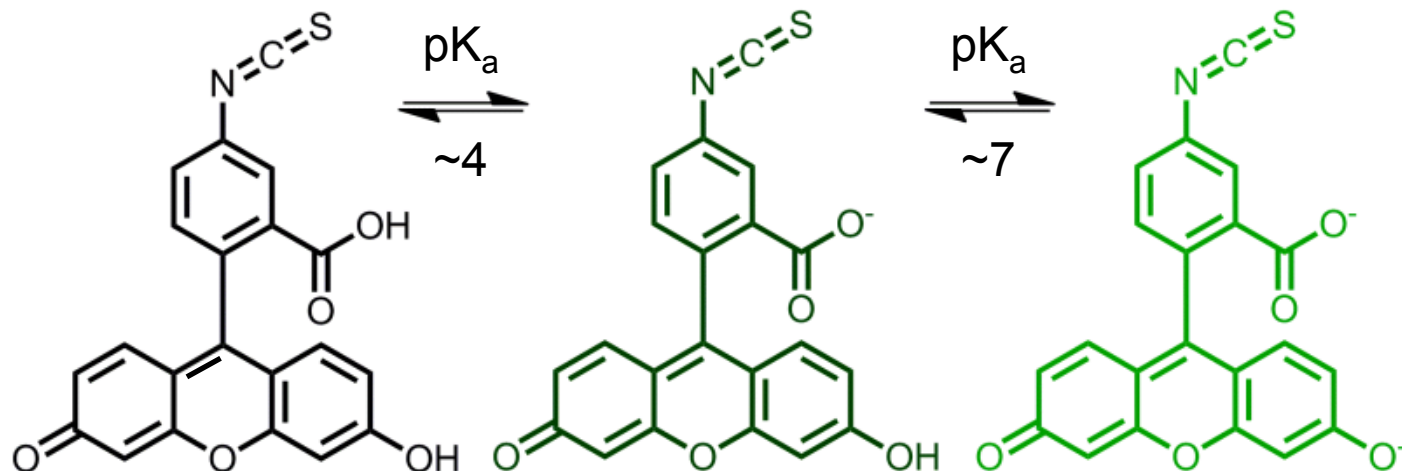
*How can we exploit this effect?*

# Uptake of a Fluorescent Reporter

Neutral species readily penetrates membrane

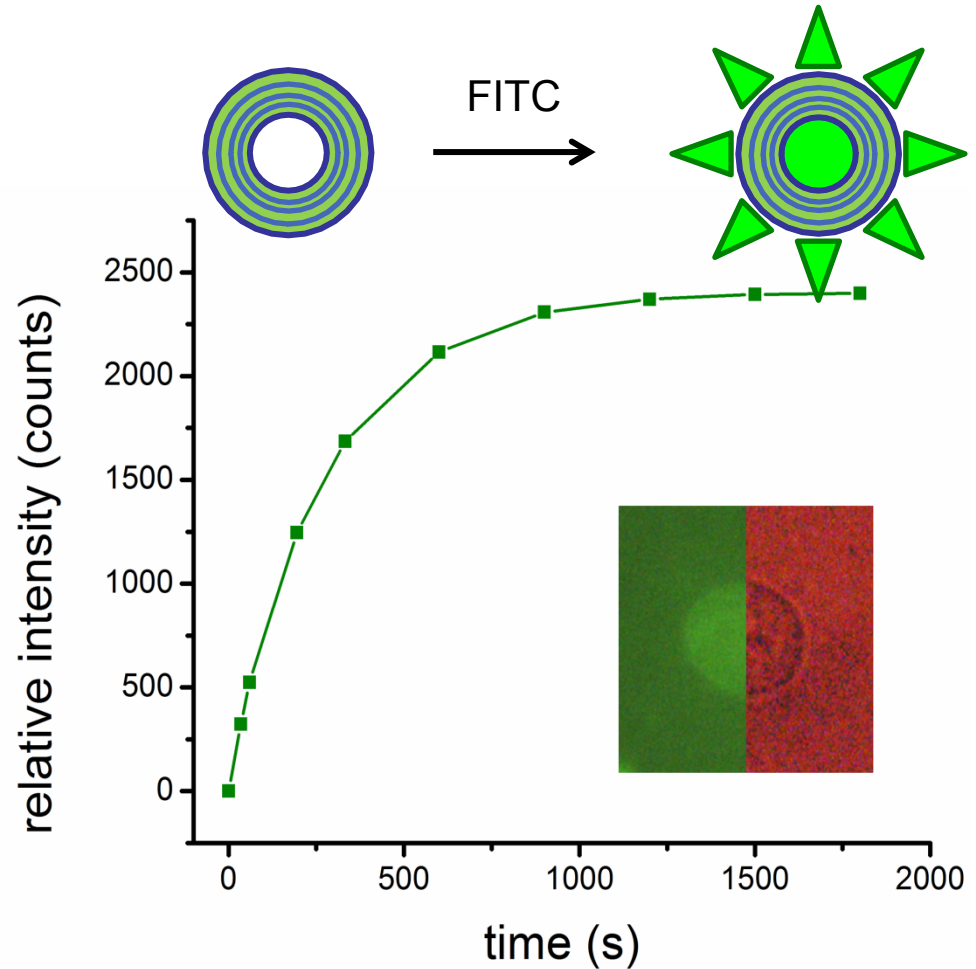
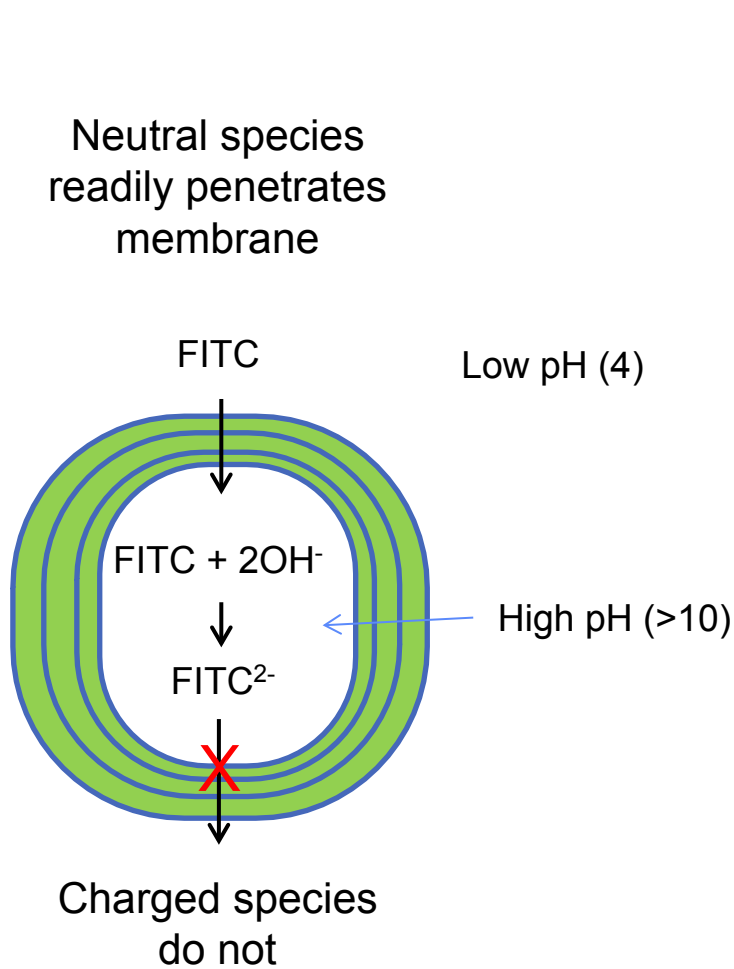


Charged species do not



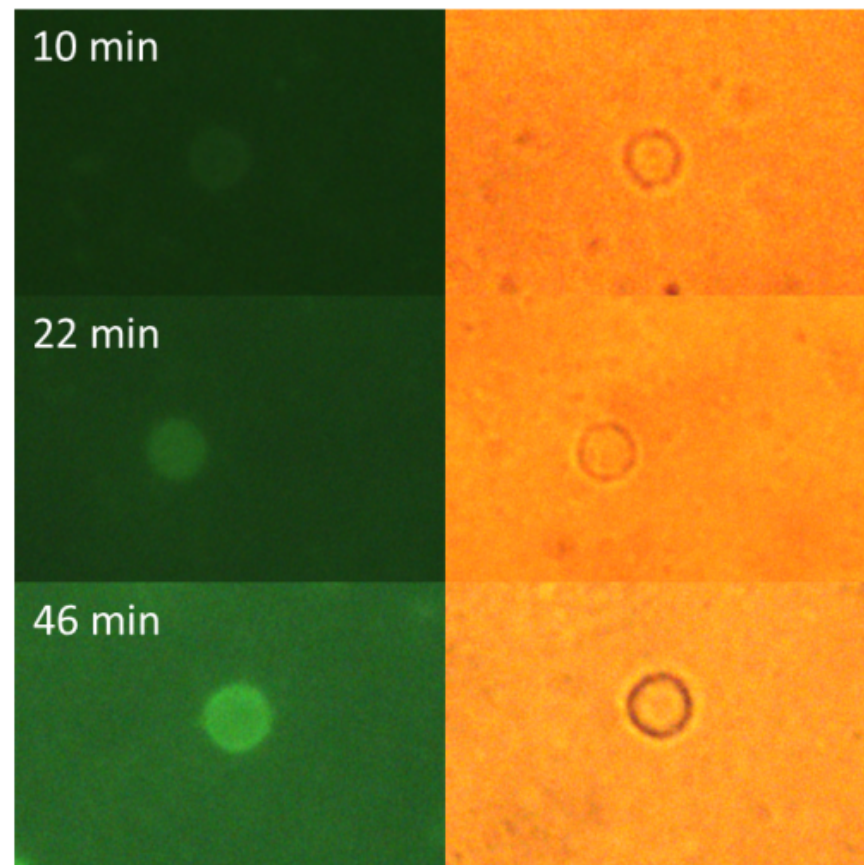
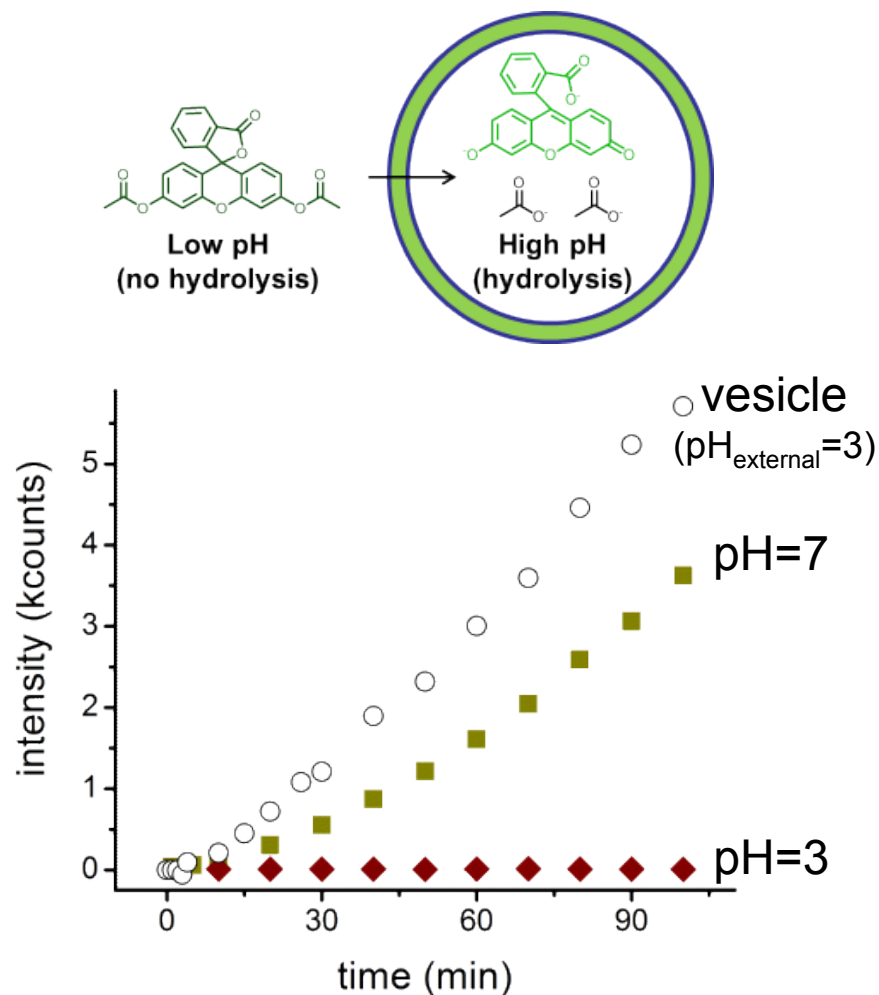
*Provides basis for enrichment of a fluorescent reporter.*

# FITC Uptake Results



*$\Delta$ pH in polymersomes to sequester acidic compounds*

# A Nanoreactor for Ester Hydrolysis



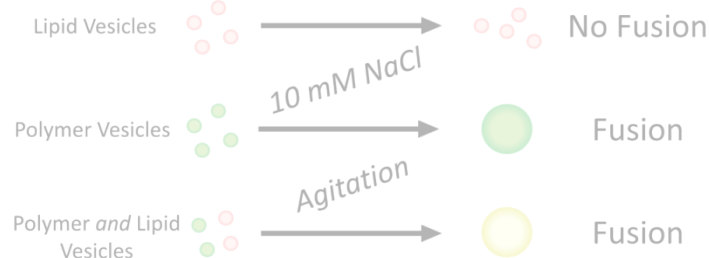
*Polymersomes act as an **artificial lysosome**, collecting and digesting hydrolysable materials*





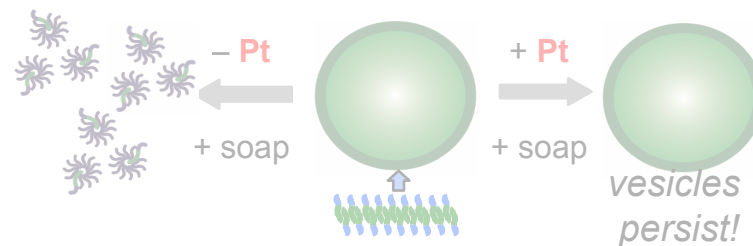
# Dynamic Polymer Vesicle Membranes

## Mechanically-Activated *Fusion*



Angew. Chem. – Int. Ed. **2014**, 53, 3372–3376; J. Poly. Sci. B, **2014**, 53, 297–303

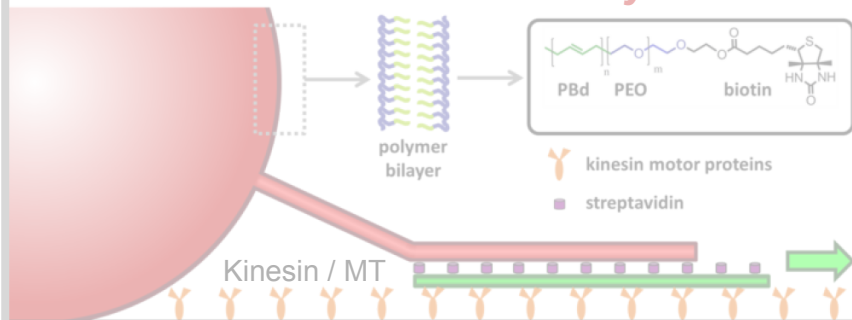
## Catalytically-Active Cross-Links (*Reactivity and Stability*)



Chem. Mater., **2015**, 27, 4808–4813

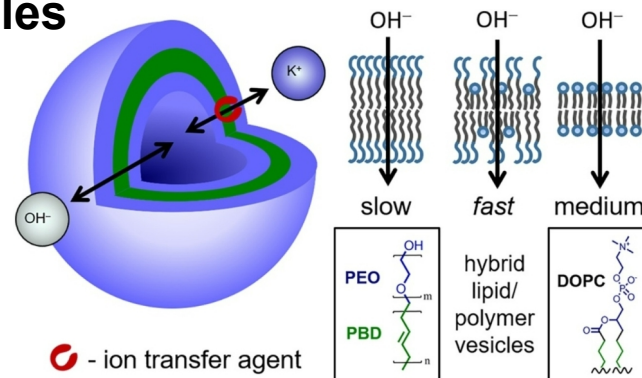
w/Hae Ra Shin & Patrick McAninch

## Dynamic Assembly of Polymer Nanotubes – *Fluidity*



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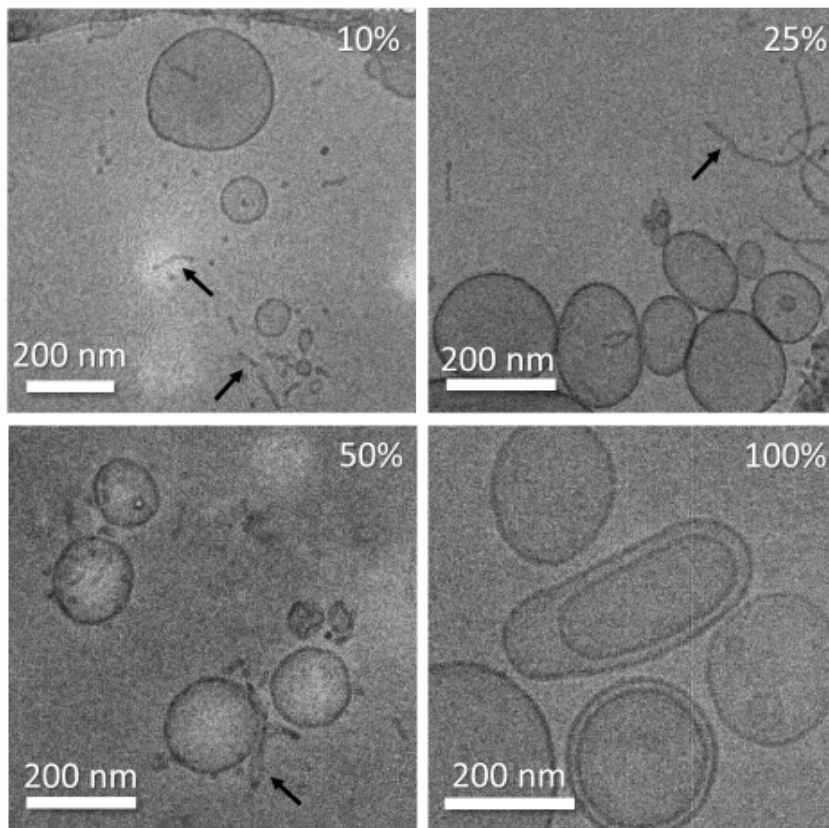
## Modulating *Permeability* in Hybrid Vesicles



Colloids and Surfaces B, **2017**, 159, 268–276



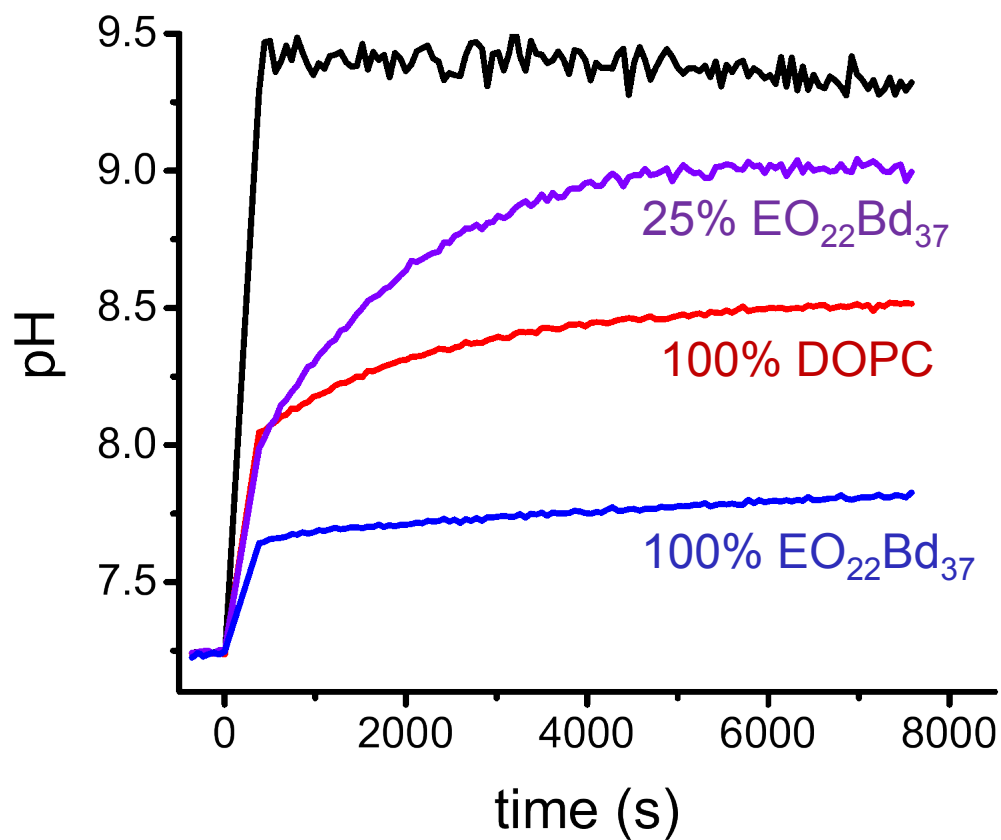
# Preparing and Characterizing \*Hybrid\* Vesicles



Good vesicles w/worm-like micelles



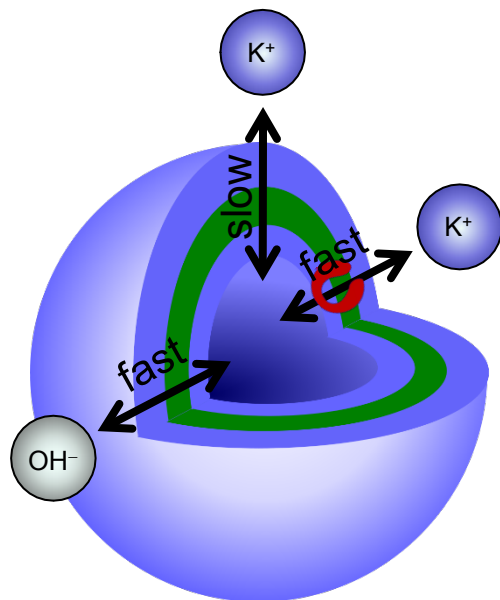
## Hybrid Vesicle Permeability



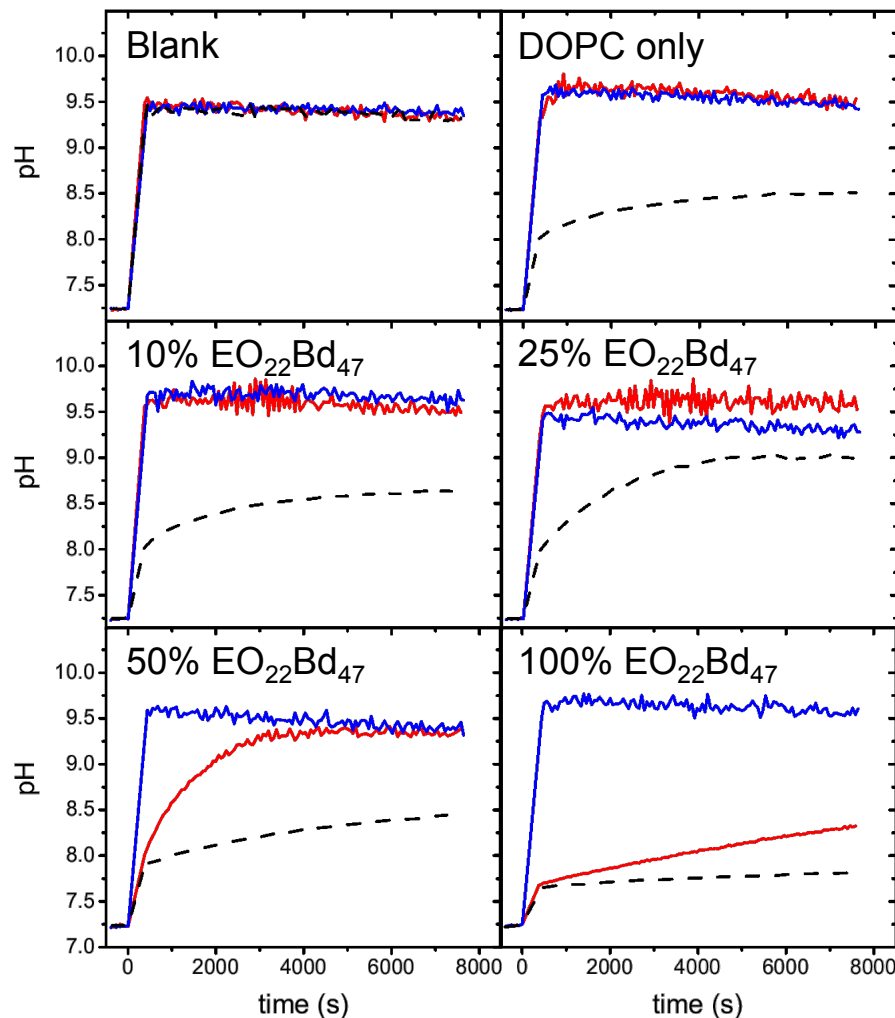
$$J = P\Delta C \rightarrow P \sim 10^{-9} \text{ cm/s}$$



# Modulating Permeability with Ionophores

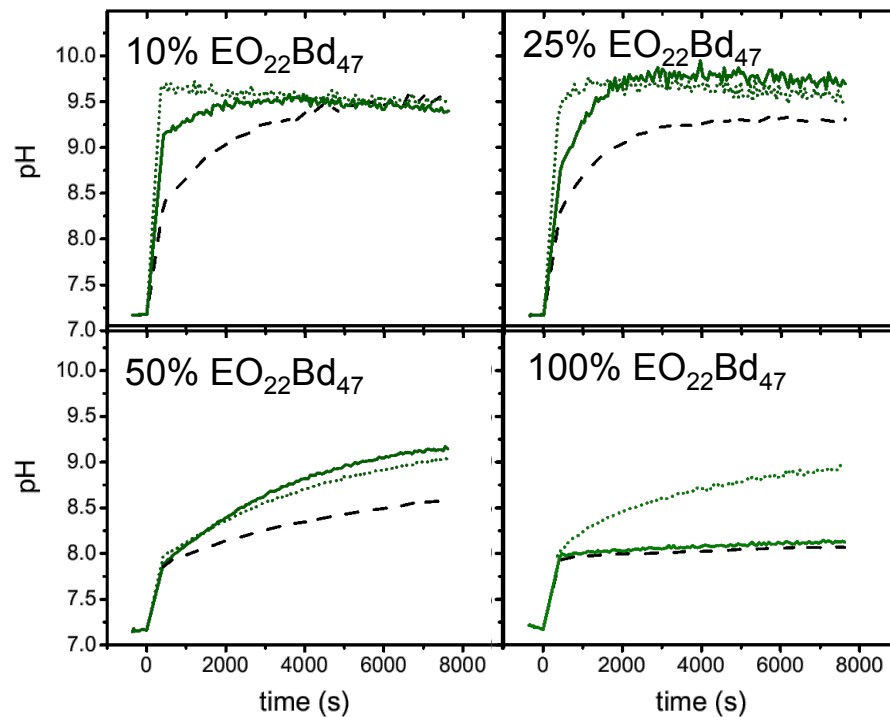
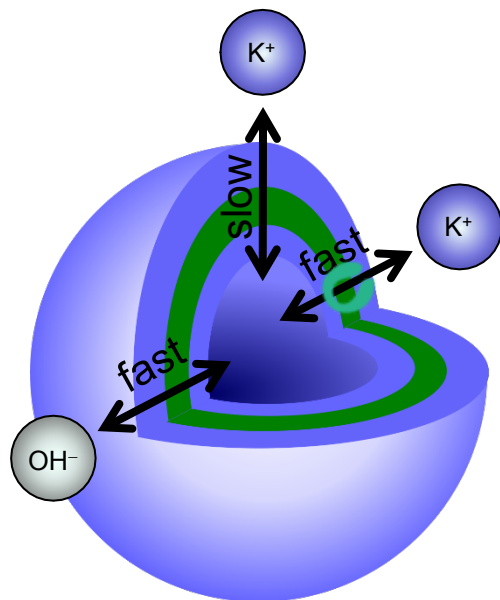


- Control (+KOH)
-  Nigericin
-  Valinomycin





# Modulating Permeability with Ion Channels



--- Control (+KOH)

— Gramicidin A (incubated <5 min)

... Gramicidin A (rehydrated >5 days)





# Summary and Conclusions

## Catalytically Active Cross-Links

Used organometallic interactions to modulate the properties of “normal” polymersomes:

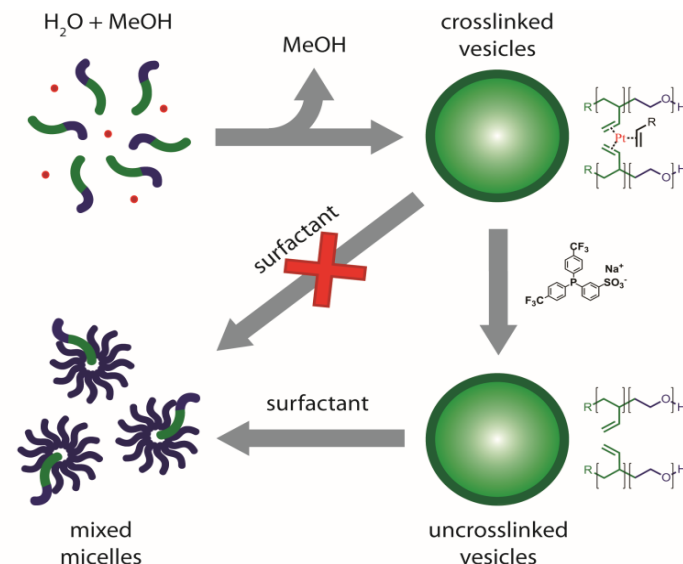
### 1. Enhanced Stability:

- Pt--II → Organometallic cross-links
- Resistant to destabilization
- Crosslinking can be selectively reversed w/ phosphine ligands

### 2. Catalytic Activity:

- Pt centers still active
- Enable hydrosilylation reactions

To produce soft self-assembling material that is both **robust** and **dynamic**.

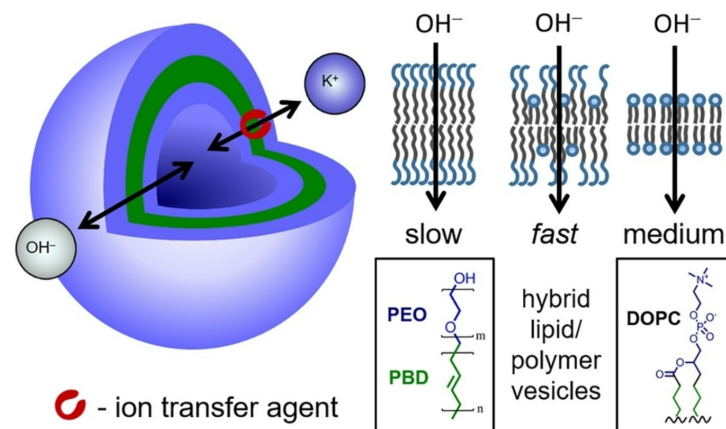


## Monitoring and Modulating Ion Flux in Hybrid Bilayers

Hybrids *MORE* permeable than 100% lipid or 100% polymer vesicles

Modulated ion flux in hybrid vesicles via reconstituted membrane proteins (Nigericin / Valinomycin / Gramicidin)

The permeability of hybrid bilayers is critical property for drug delivery, nanoreactor, and sensing applications.





# Team and Acknowledgments



**Dr. Walter Paxton (CINT Staff)**

Dr. Ian Henderson (Omphalos)

Hope Quintana (NMSU/LANL)

Dr. Julio Martinez (NMSU)

**Patrick McAninch (CINT intern)**

**Dr. Hae Ra Shin (CINT postdoc)**

**Dr. Komandoor Achyuthan (Sandia)**

Dr. George Bachand (CINT Staff)

Dr. Adrienne Greene (Sandia)

Dr. Nathan Bouxsein (Perspectives)

Dr. Sergei Ivanov (CINT Staff)

Dr. Gabriel Montano (NAU)

## CINT / DOE-BES / SNL-LDRD

# Thank You!



