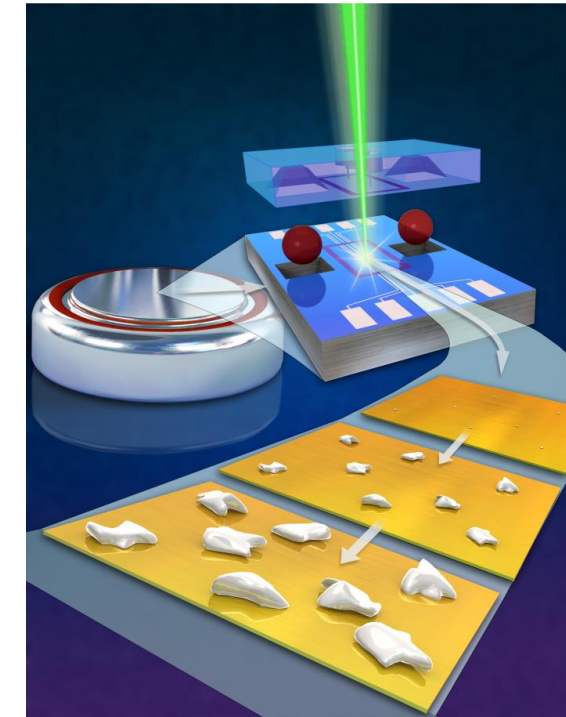
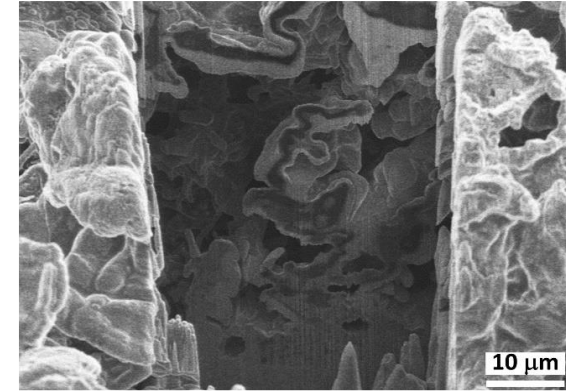


Nanoscale Characterization of Electrochemical- Mechanical Mechanisms with Electron Microscopy



Katherine Jungjohann
National Renewable Energy Laboratory
Formerly at Sandia National Laboratories

This work was performed, in part, at the Center for Integrated Nanotechnologies, an Office of Science User Facility operated for the U.S. Department of Energy (DOE) Office of Science. Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. DOE's National Nuclear Security Administration under contract DE-NA-0003525. The views expressed in the article do not necessarily represent the views of the U.S. DOE or the United States Government.



About the Presenter

- **Katherine Jungjohann**, National Renewable Energy Laboratory
 - Group Manager, Analytical Microscopy and Imaging Sciences
 - Microscopy Society of America, EMLG FIG Lead
- **Characterization of Solid-Liquid Interfaces in Energy Systems**
- **Co-authors:**
 - **Daniel Long**, Air Force Research Laboratory
 - **Katharine Harrison**, Sandia National Laboratories
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 - **Zac Milne**, Sandia National Laboratories
 - **Khalid Hattar**, Sandia National Laboratories
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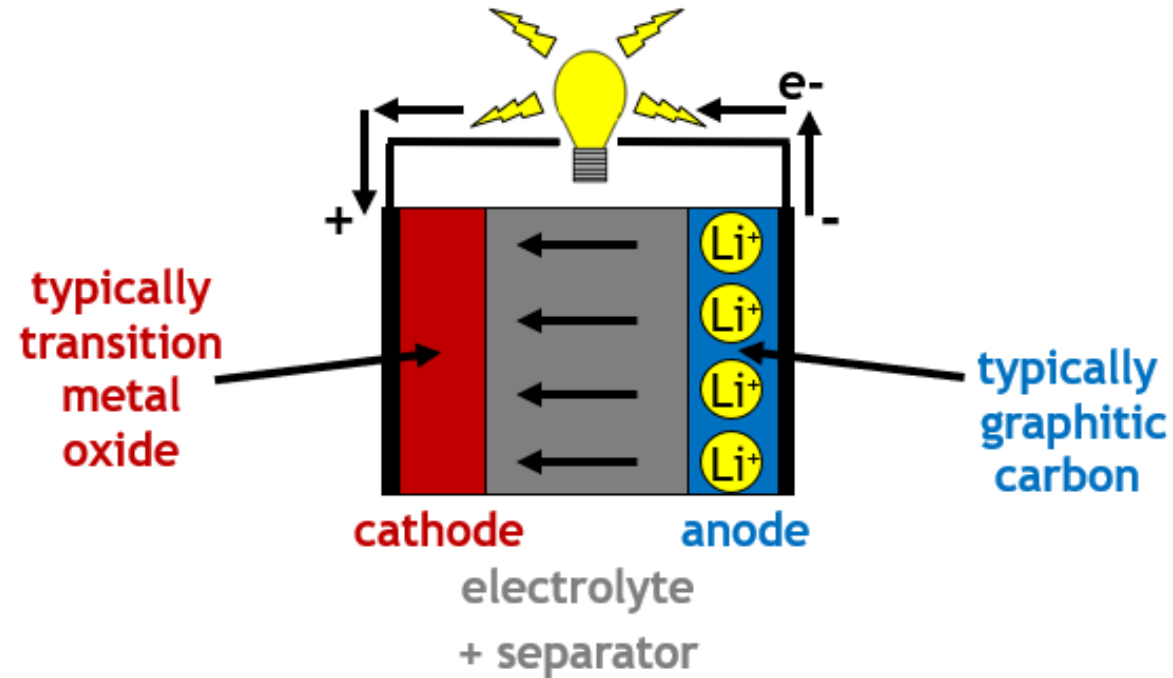


Lithium metal anodes can theoretically increase battery capacity by

10x



Traditional Li-Ion Batteries



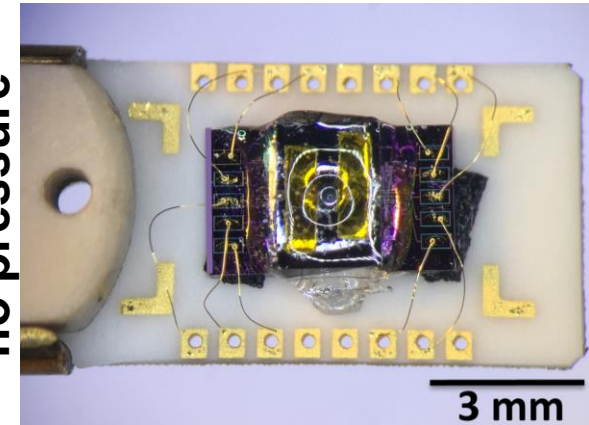
Discharging Battery

- Li^+ from the **anode** through the electrolyte and into the **cathode**
- e^- move through the external circuit from the **anode** to the **cathode** (from - to + charge)

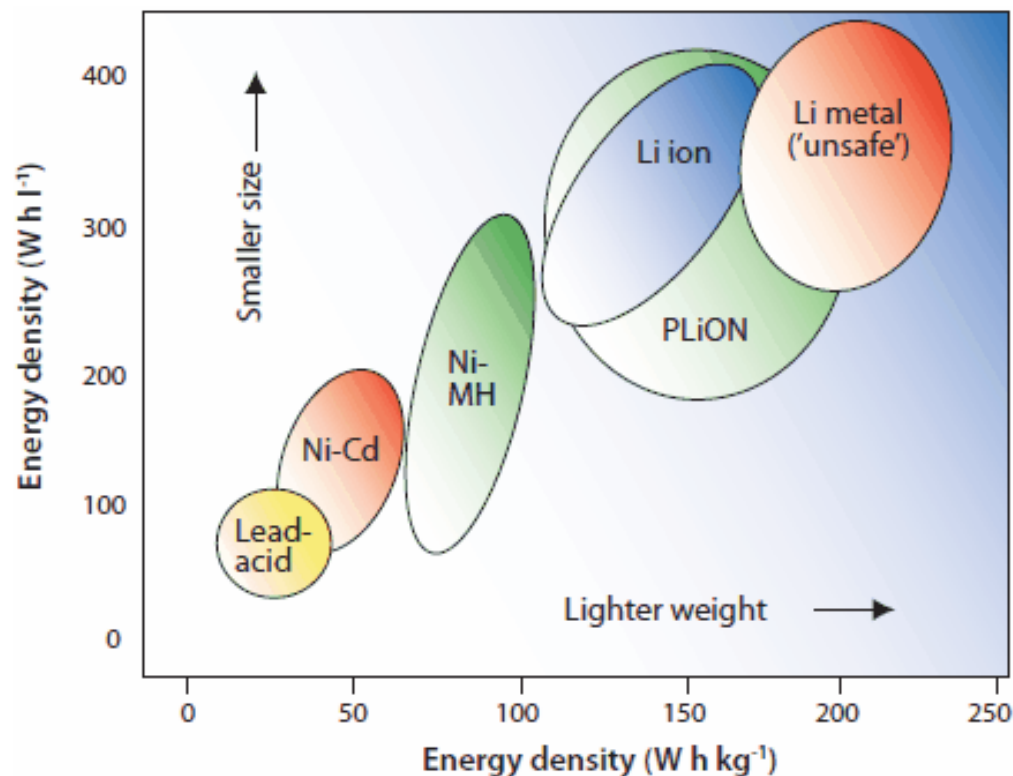
TEM cell
no pressure

coin cell
spring pressure

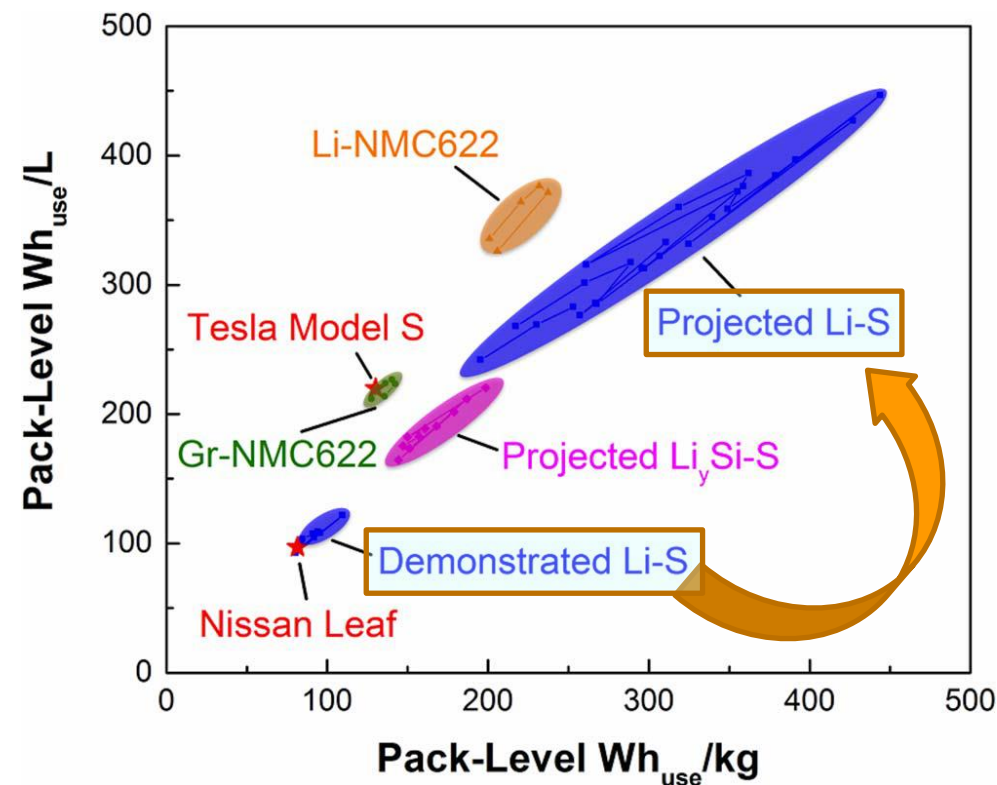
pouch cells
external pressure



How Can We Improve Rechargeable Batteries?



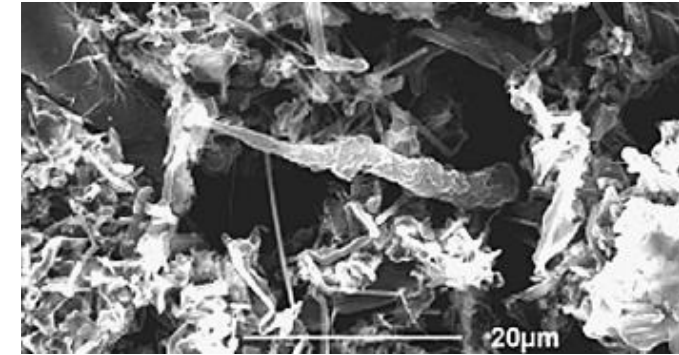
J.-M. Tarascon and M. Armand *Nature*
DOI: 10.1038/35104644



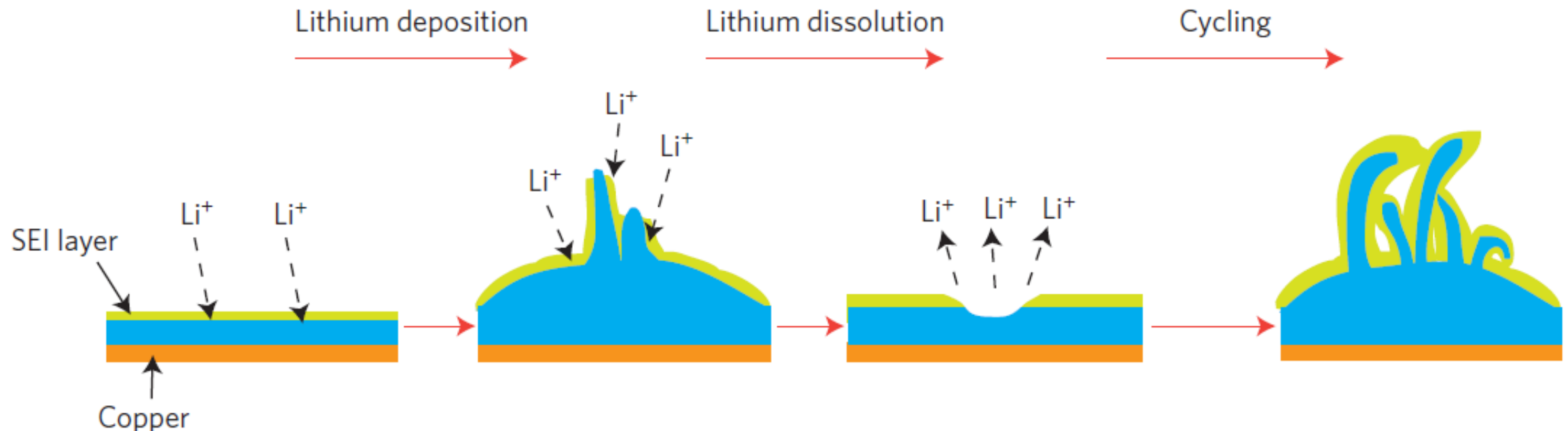
K.G. Gallagher et al. *J. Electrochem. Soc.*
DOI: 10.1149/2.0611506jes

Rechargeable Li Metal: Chemical & Mechanical

- Uncontrolled morphology → many problems
 - Short circuits = fire
 - Excessive solid electrolyte interphase (SEI) = low Coulombic efficiency, high impedance, and Li consumption
 - Li gets stranded and disconnected = “dead” Li

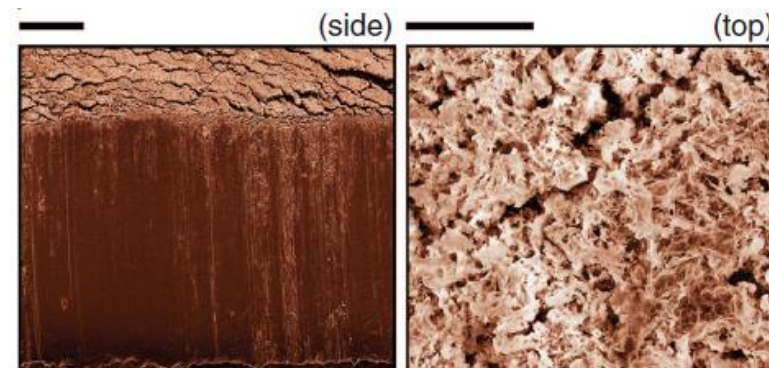
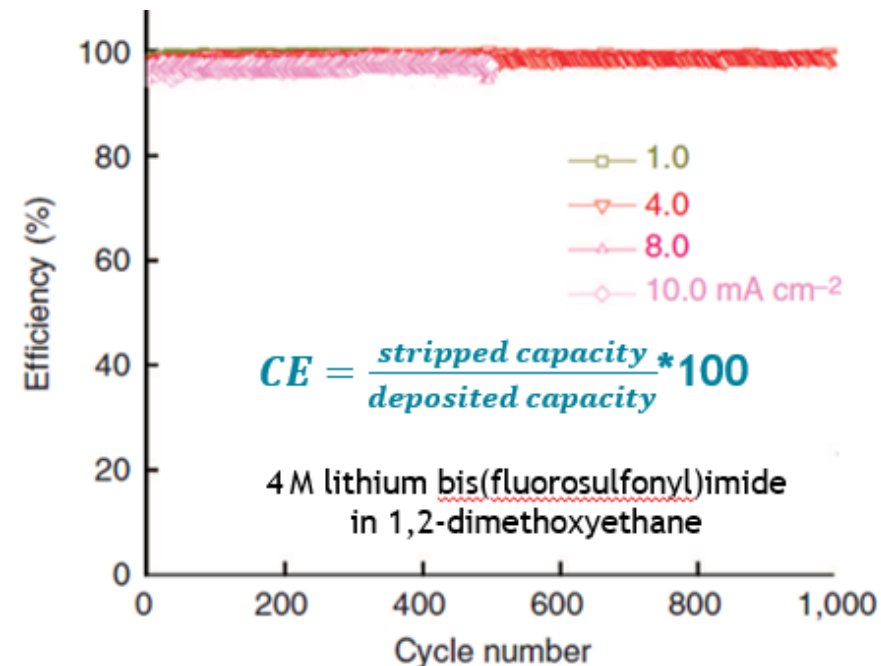
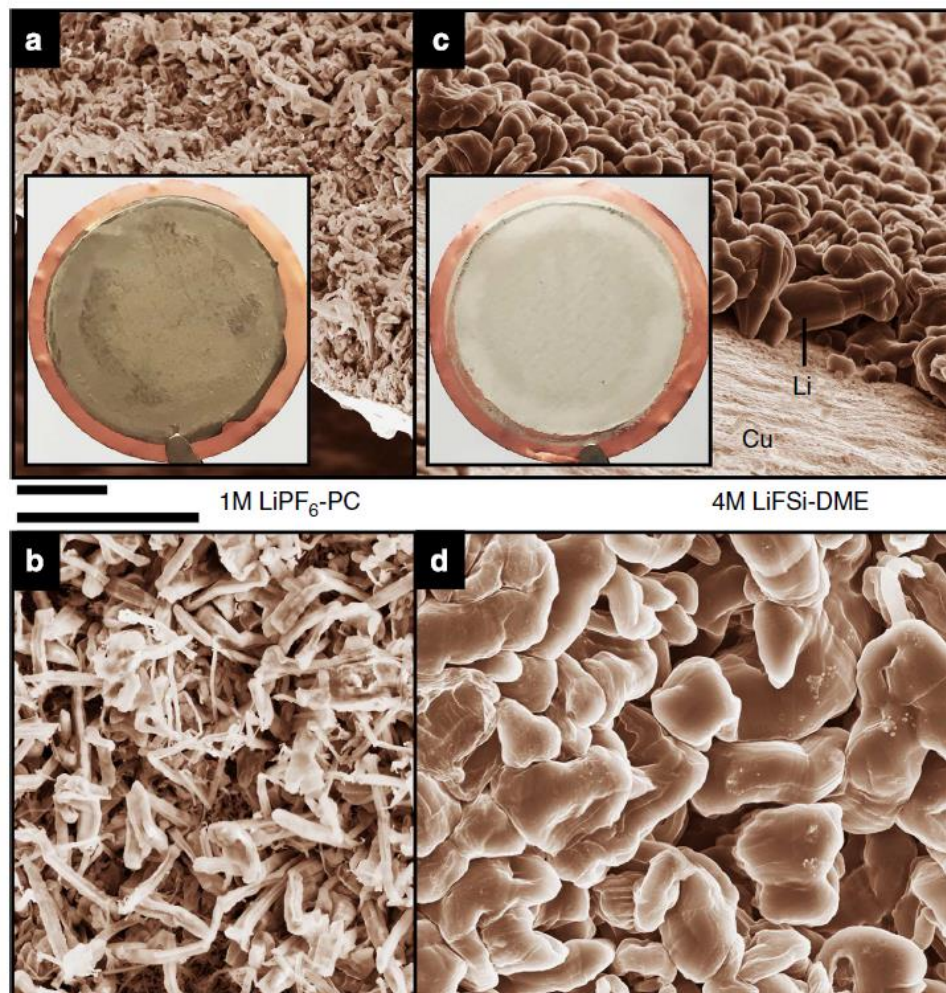


Gireaud et al., *Electrochem. Comm.* 8, 1639 (2006)



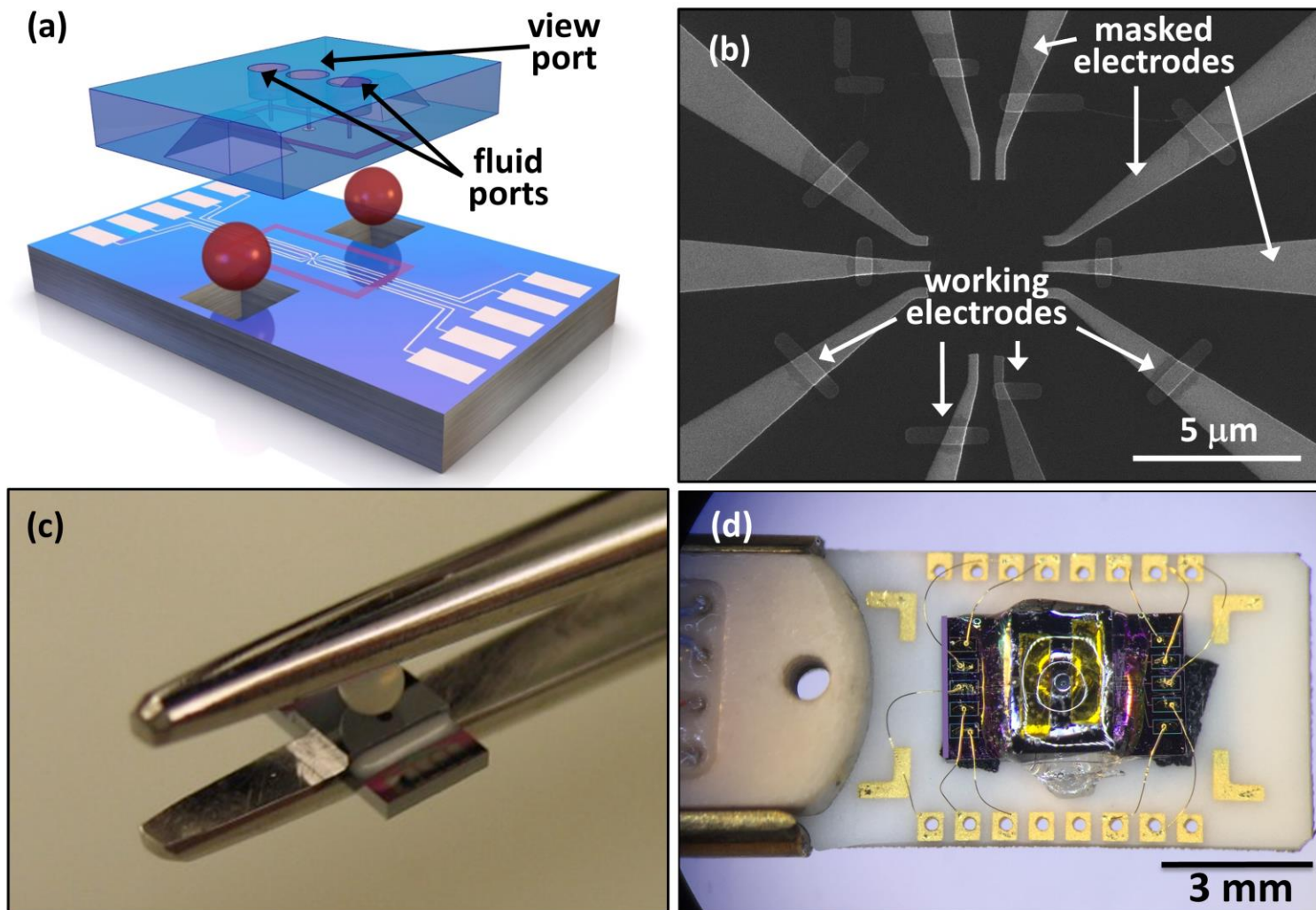
Lithium Self-Discharge and Prevention

Qian et al. *Nature Communications* DOI: 10.1038/ncomms7362

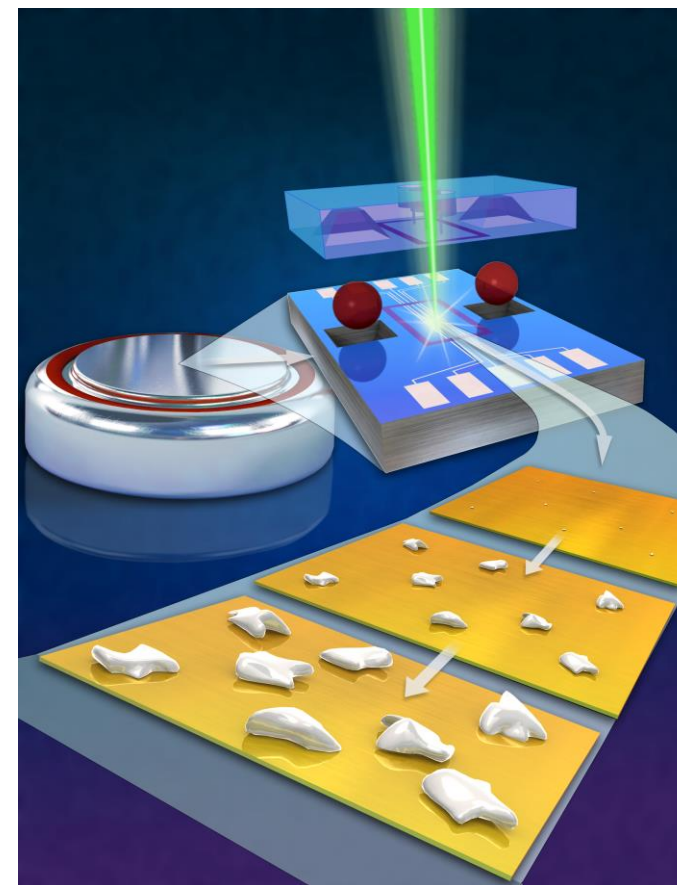




CINT's Electrochemical TEM Discovery Platform



***planar electrodes on insulated chip surface with no separator**

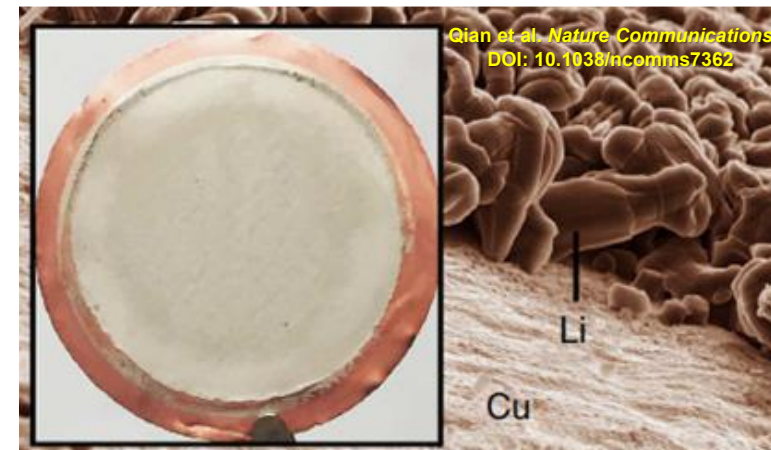
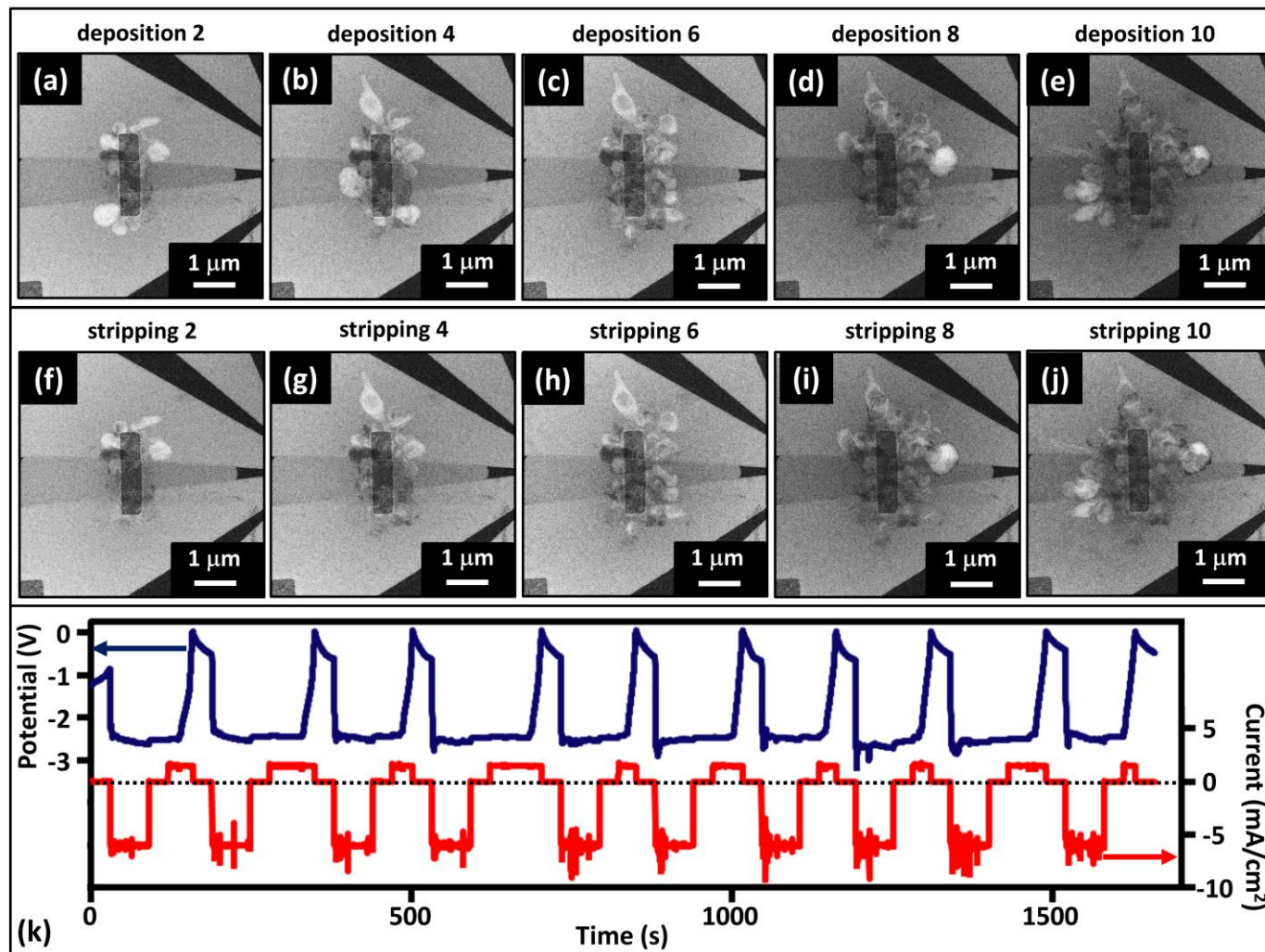


Harrison et al. *ACS Nano*
10.1021/acsnano.7b05513



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In-Situ Nanoscale Li Electrodeposition

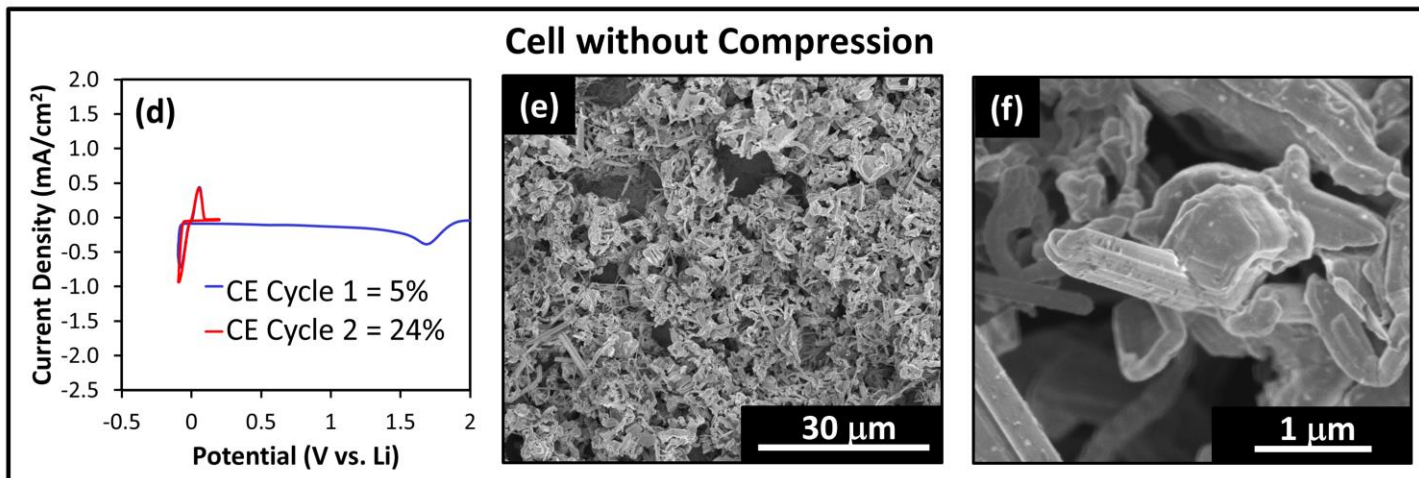
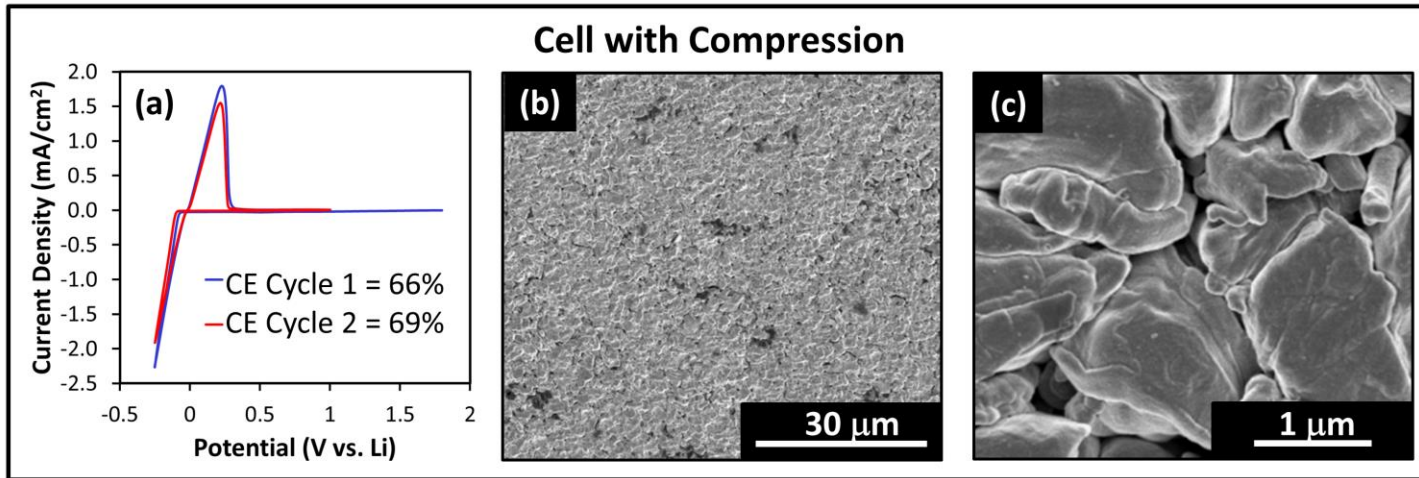


- Morphology images after every deposition and stripping step (Li = white)
- CE VERY low ($18\% \pm 9\%$)
- Widely varied morphology

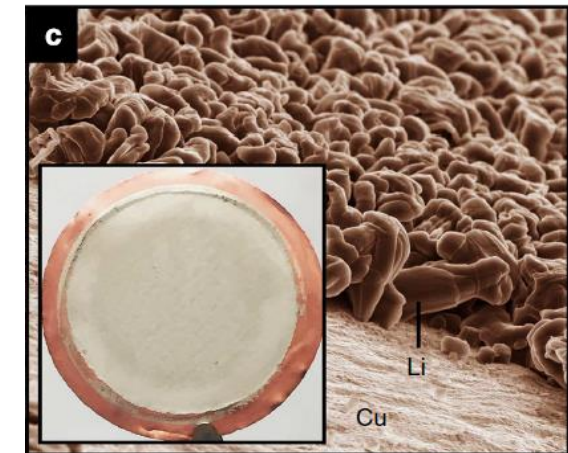
bright field imaging
 WE = 60 nm Ni/5 nm Ti
 deposit 6 mA/cm²
 strip 1.5 mA/cm²
 capacity = 0.1 mAh/cm²

Harrison et al. *ACS Nano*
 DOI: 10.1021/acsnano.7b05513

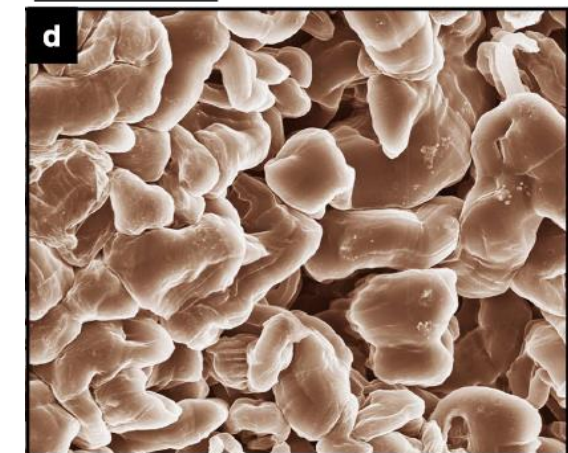
Applied Pressure Critical for High-Density Li



Qian et al. *Nature Communications*
DOI: 10.1038/ncomms7362



4M LiFSi-DME

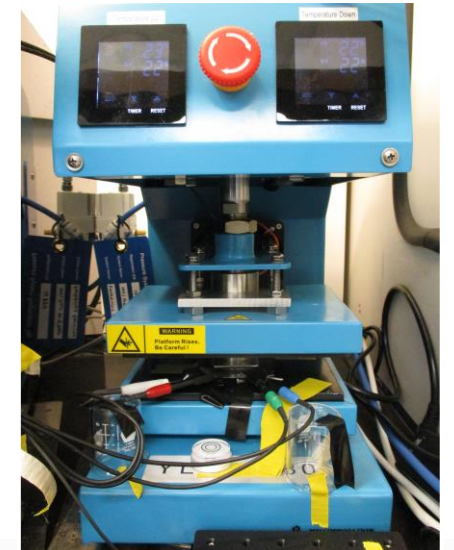
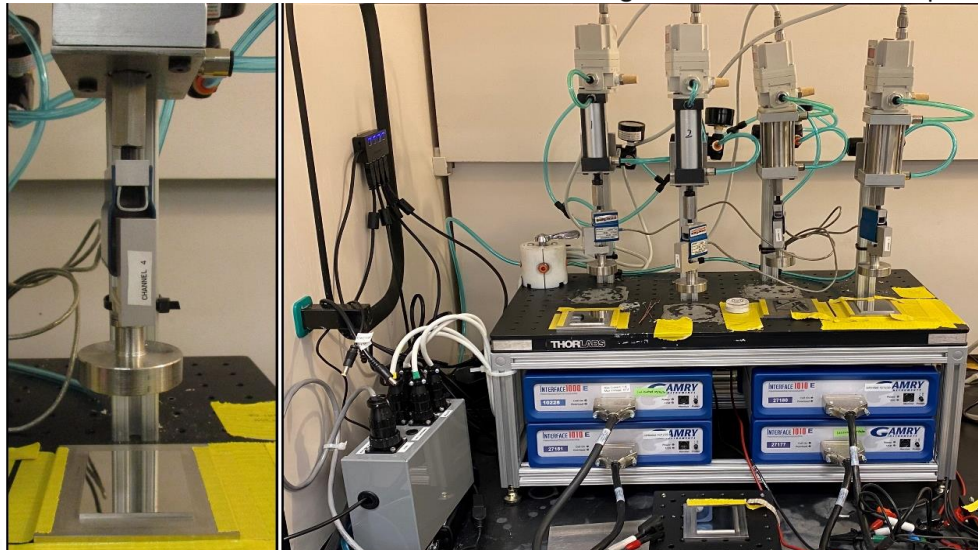
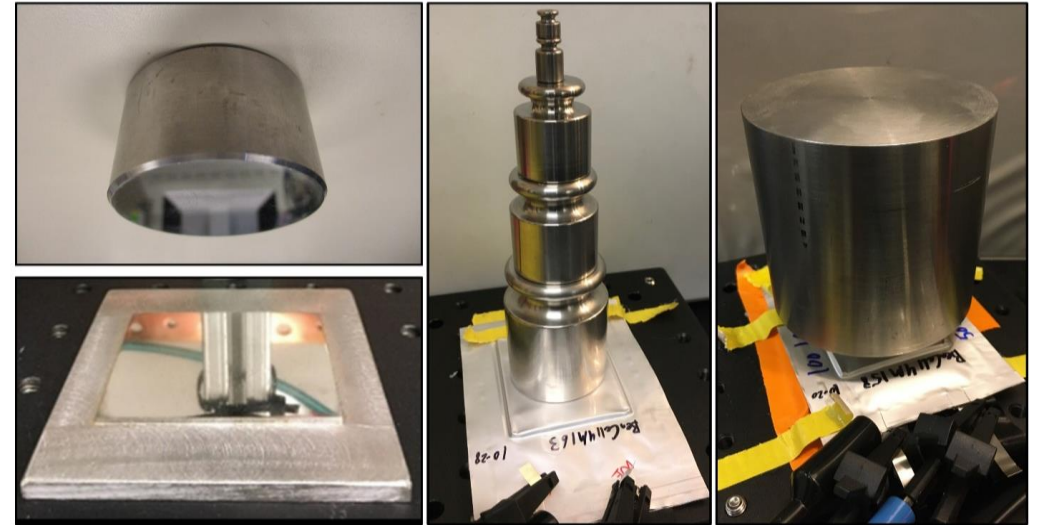
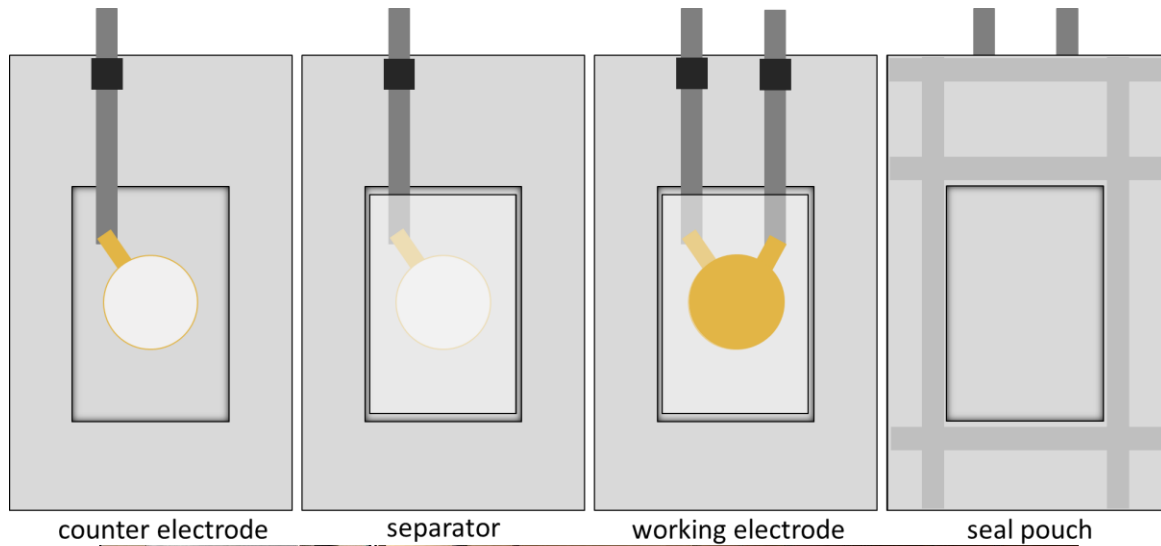


Harrison et al. *ACS Nano*
10.1021/acsnano.7b05513



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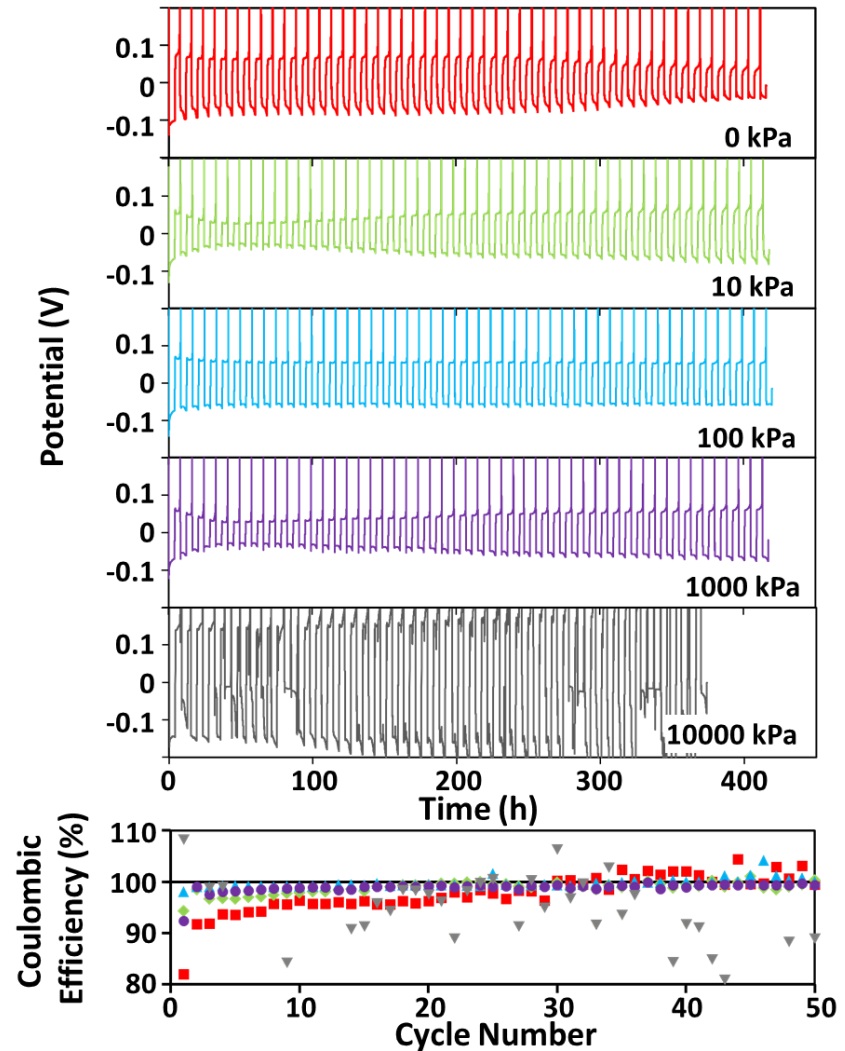
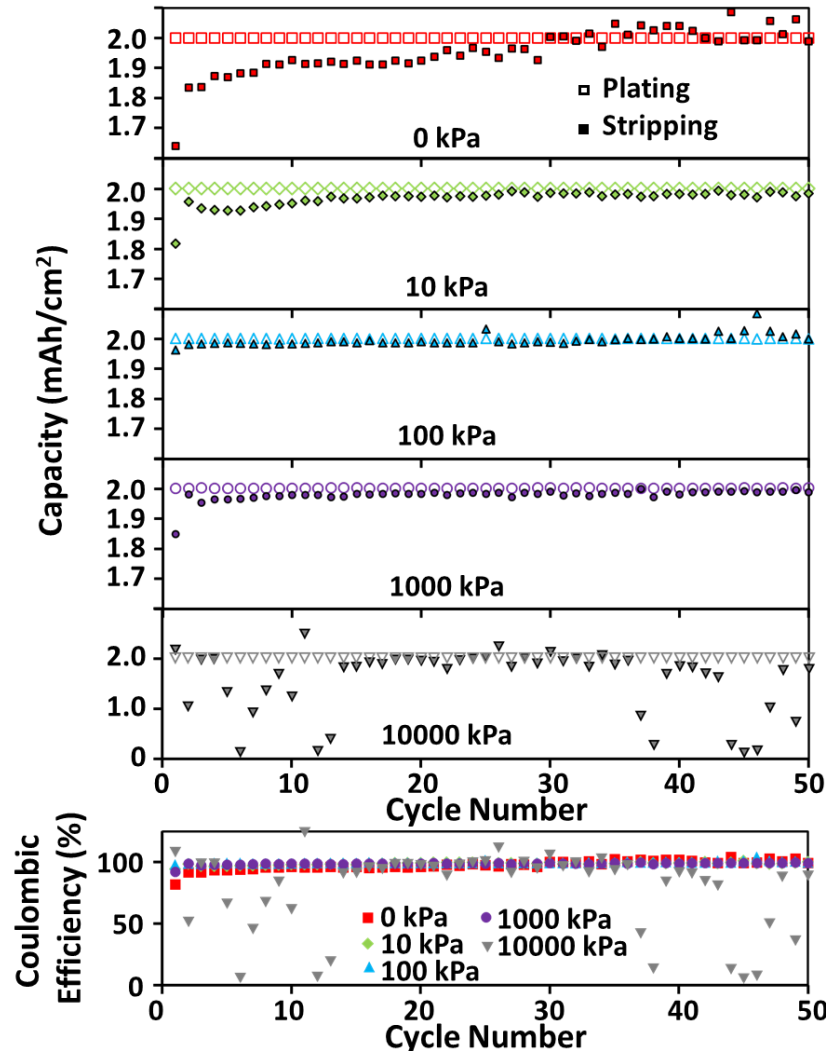
Controlled Pressure on Electrodeposited Li



Effects of Pressure on Li Metal: Low Current

Low Current: 0.5 mA/cm^2

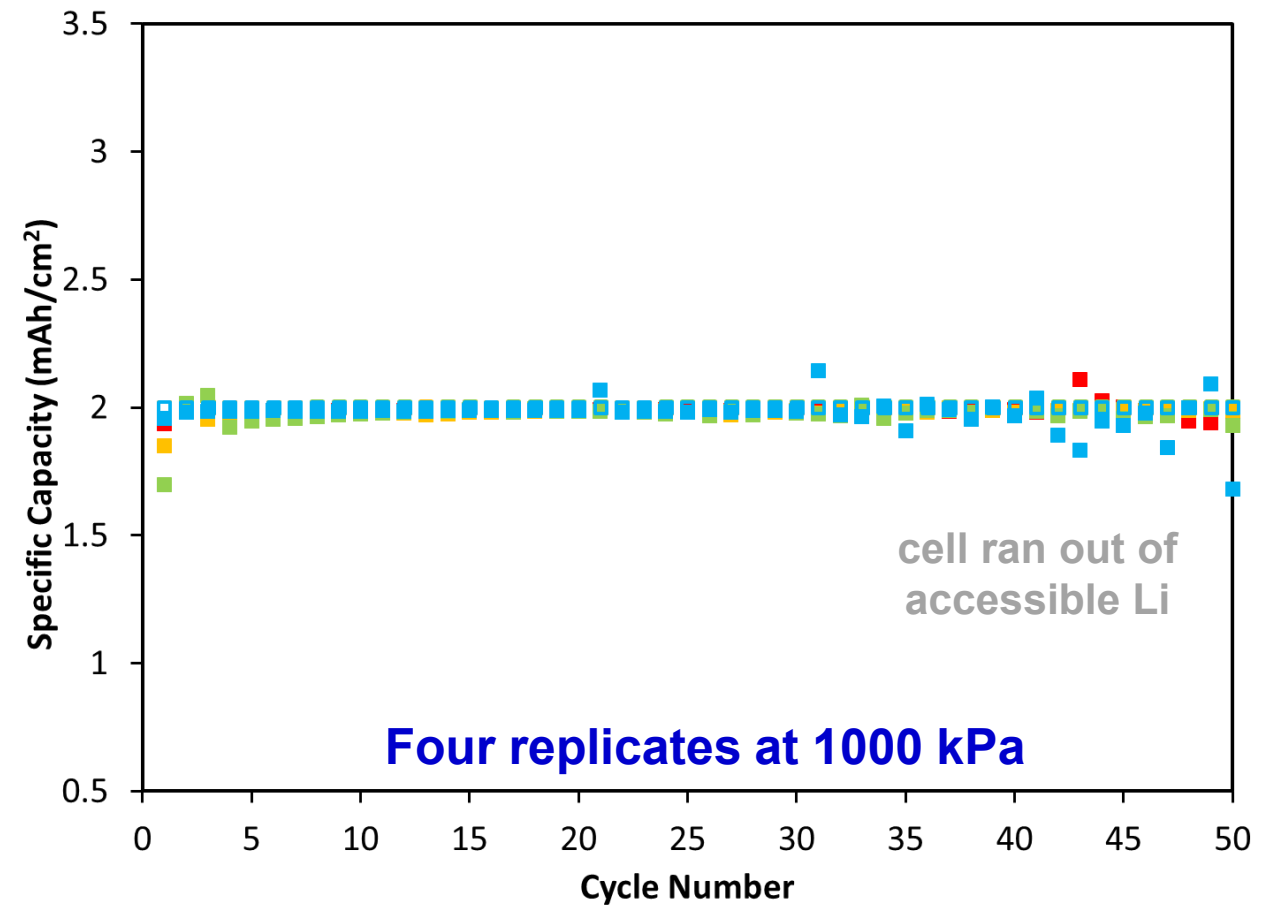
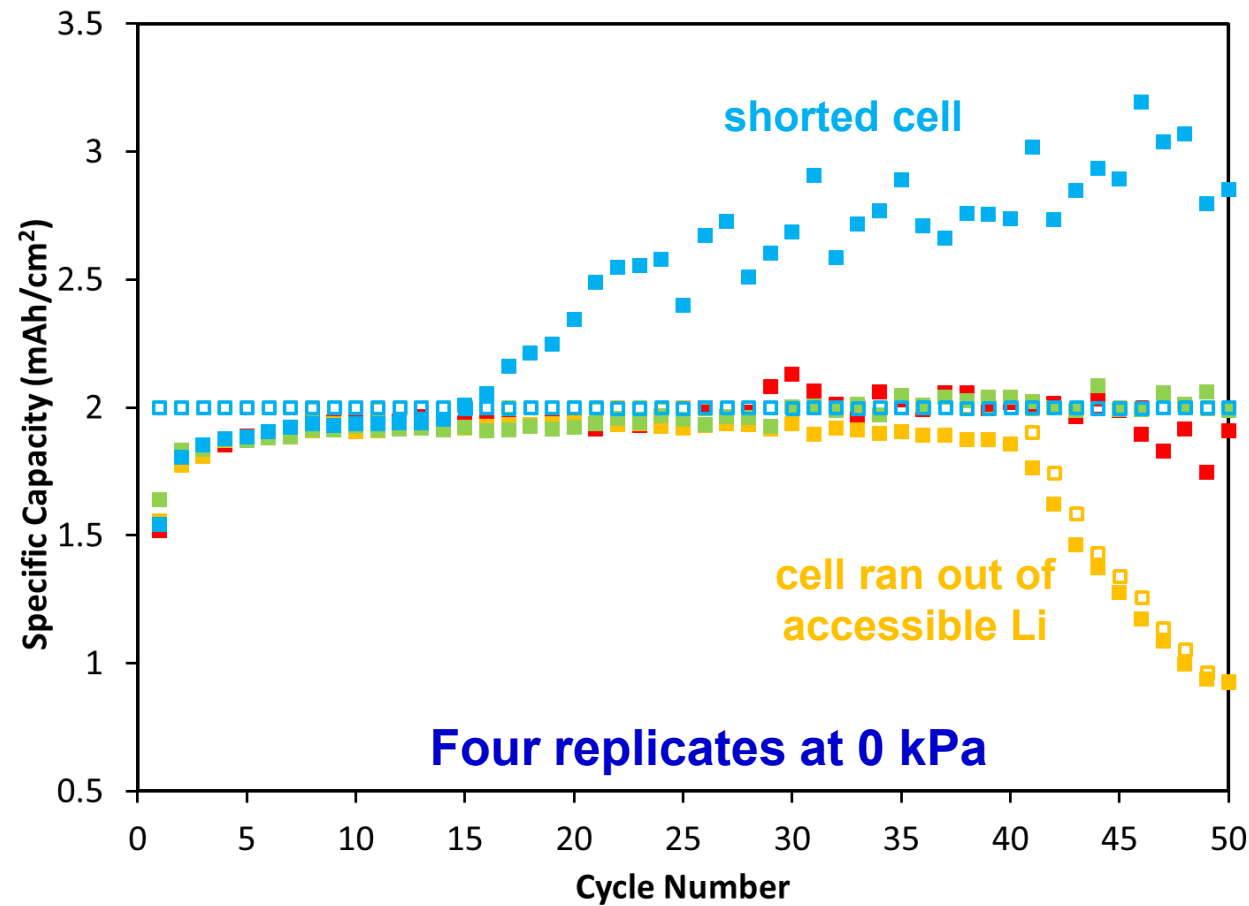
- Cycling stability generally increases with increasing pressure until 10,000 kPa
- 10,000 kPa is too high and causes increased overpotential and loss of cycling stability
- Transport might be limited locally at high pressure where pores can close
- CE generally improves with pressure but 100 and 1,000 kPa are similar



Harrison et al., *ACS Appl. Mater. Interfaces* 13, 31668 (2021).
DOI: 10.1021/acsami.1c06488

Effects of Pressure on Li Metal: Low Current

Low Current: 0.5 mA/cm^2





Characterization: Ex-Situ Cryogenic SEM

- Plunge-freeze or slowly freeze coin cell battery electrodes
- Inert transfer from glovebox into cryo SEM/FIB
- Cross-sectioning in cryo SEM/FIB to observe electrodeposited Li metal

Dr. John Watt

watt@lanl.gov

cint.lanl.gov or

nsrportal.sandia.gov



Vitrobot

Captures native, solvated state



Scios FIB/SEM

Analysis of surfaces & buried interfaces; 3D tomography



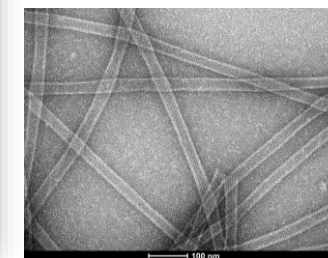
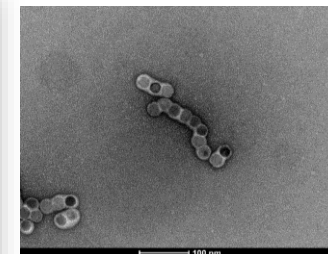
Leica Cryo SEM Stage

Includes cryo-FIB milling, lift out, and transfer to the TEM



Talos L120C CryoTEM

Dedicated low dose, low keV TEM for imaging of beam sensitive materials



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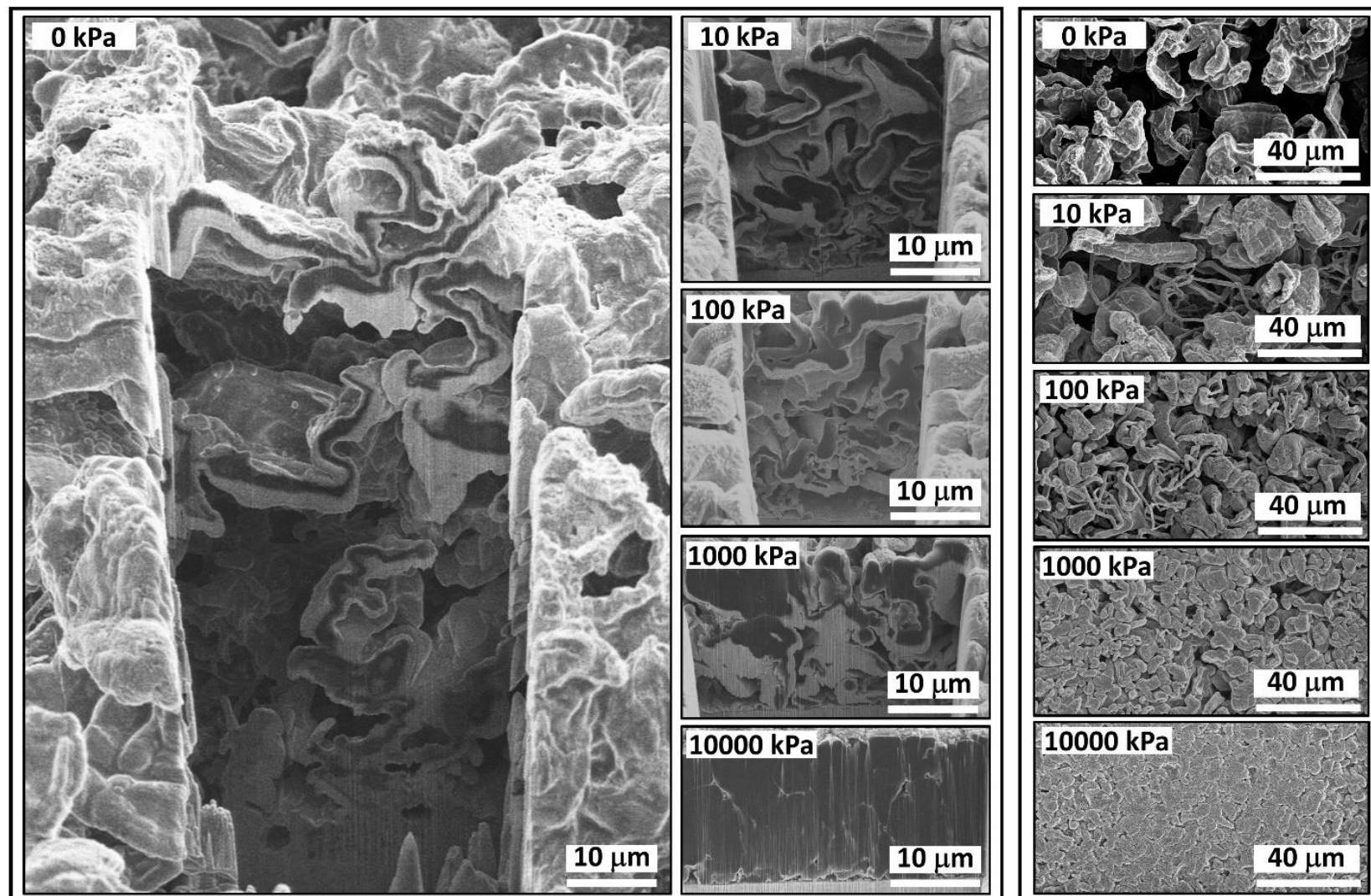
Pressure at Low Current: 1st Li Deposition Step

Low Current: 0.5 mA/cm²

- Morphology improves drastically with pressure (even for 10000 kPa)

Pressure (kPa)	Average CE (%) First Cycle
0	82.3 ± 6.2
10	90.5 ± 4.1
100	97.5 ± 0.6
1000	93.6 ± 5.3
10000	106.2 ± 1.6

Pressure (kPa)	Thickness 51 st Plating (μm)
0	91
10	33
100	30
1000	22
10000	17

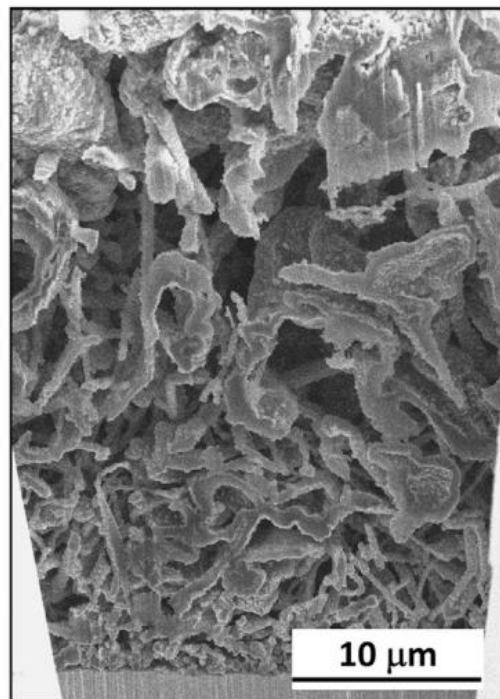


Harrison et al., *ACS Appl. Mater. Interfaces* 13, 31668 (2021).
DOI: 10.1021/acsami.1c06488

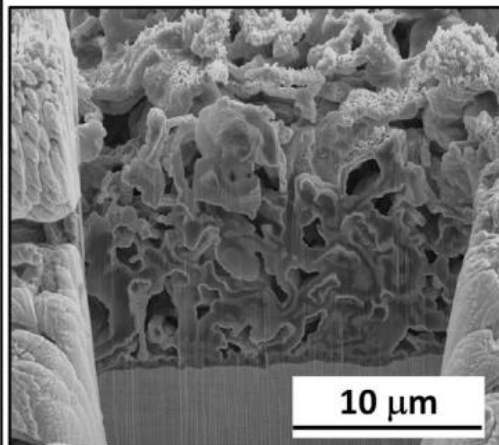
Pressure at High Current: 1st Li Deposition Step

- Li deposits are denser with increased pressure but slight difference
- Low current, no transport limitations, Li deposits at most favorable sites
- High current, transport severely limited, Li will deposit everywhere

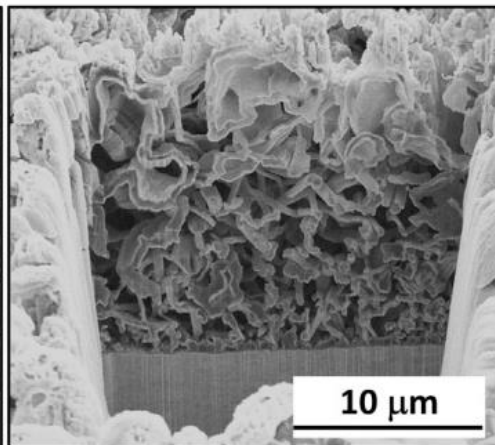
Harrison et al., *iScience* 24, 103394 (2021).
DOI: 10.1016/j.isci.2021.103394



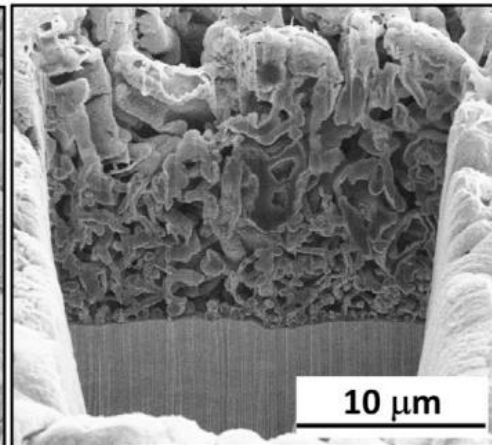
0 MPa



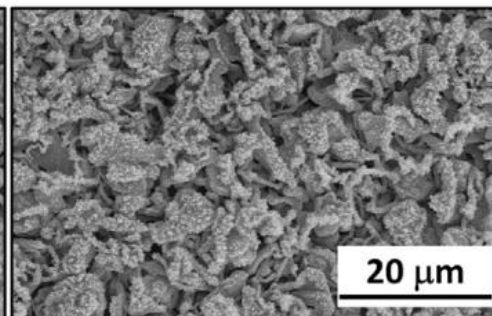
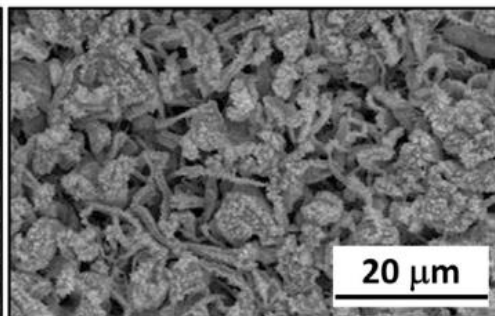
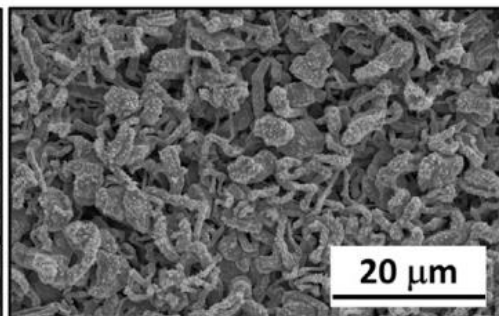
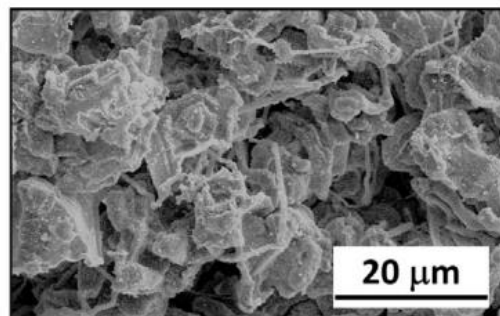
0.01 MPa



0.1 MPa

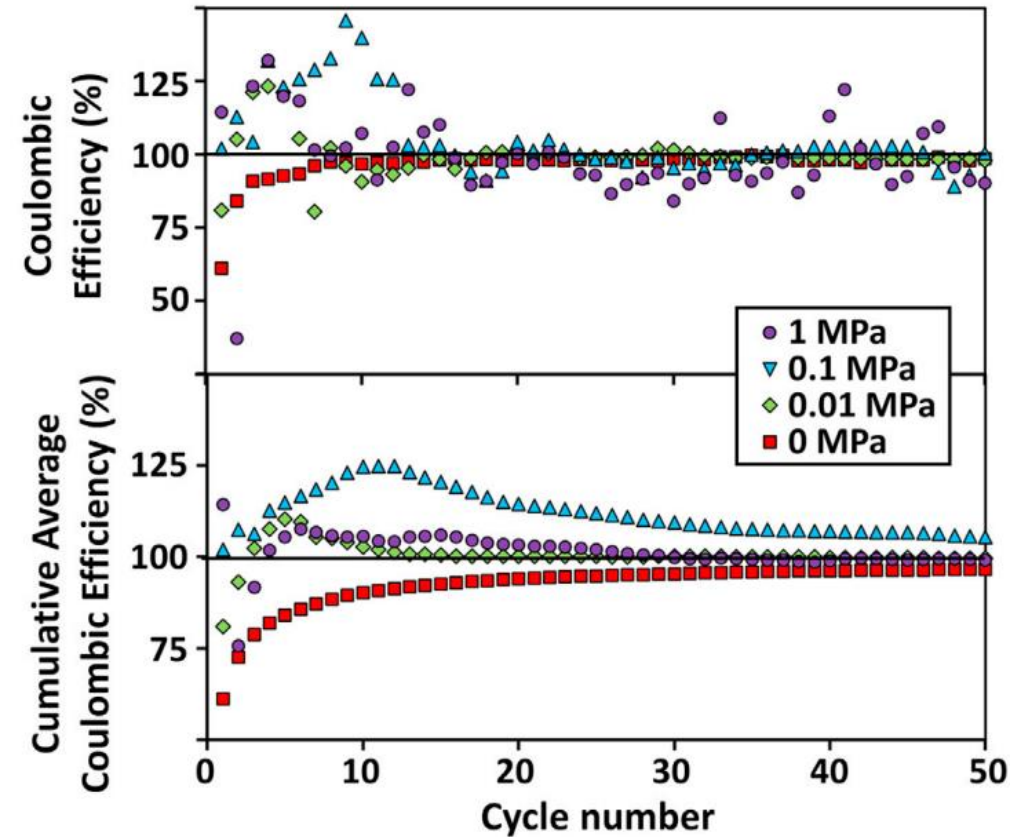
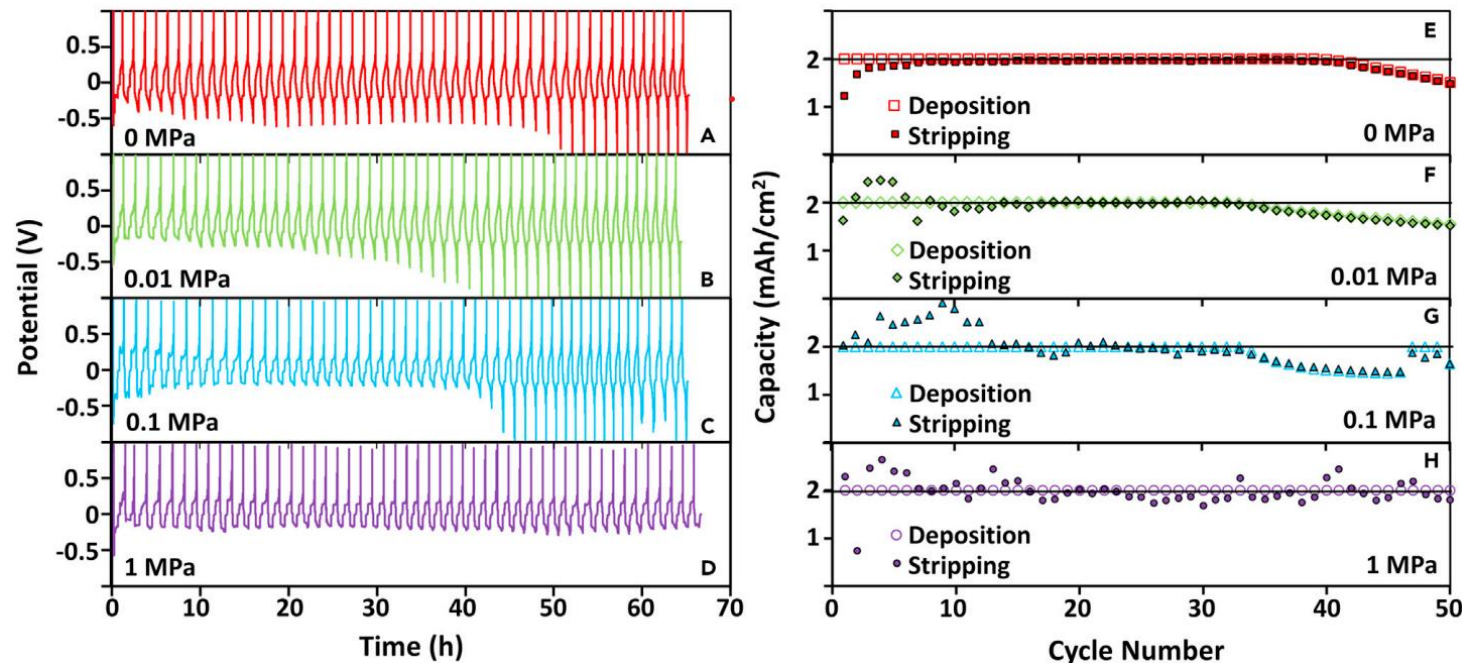


1 MPa



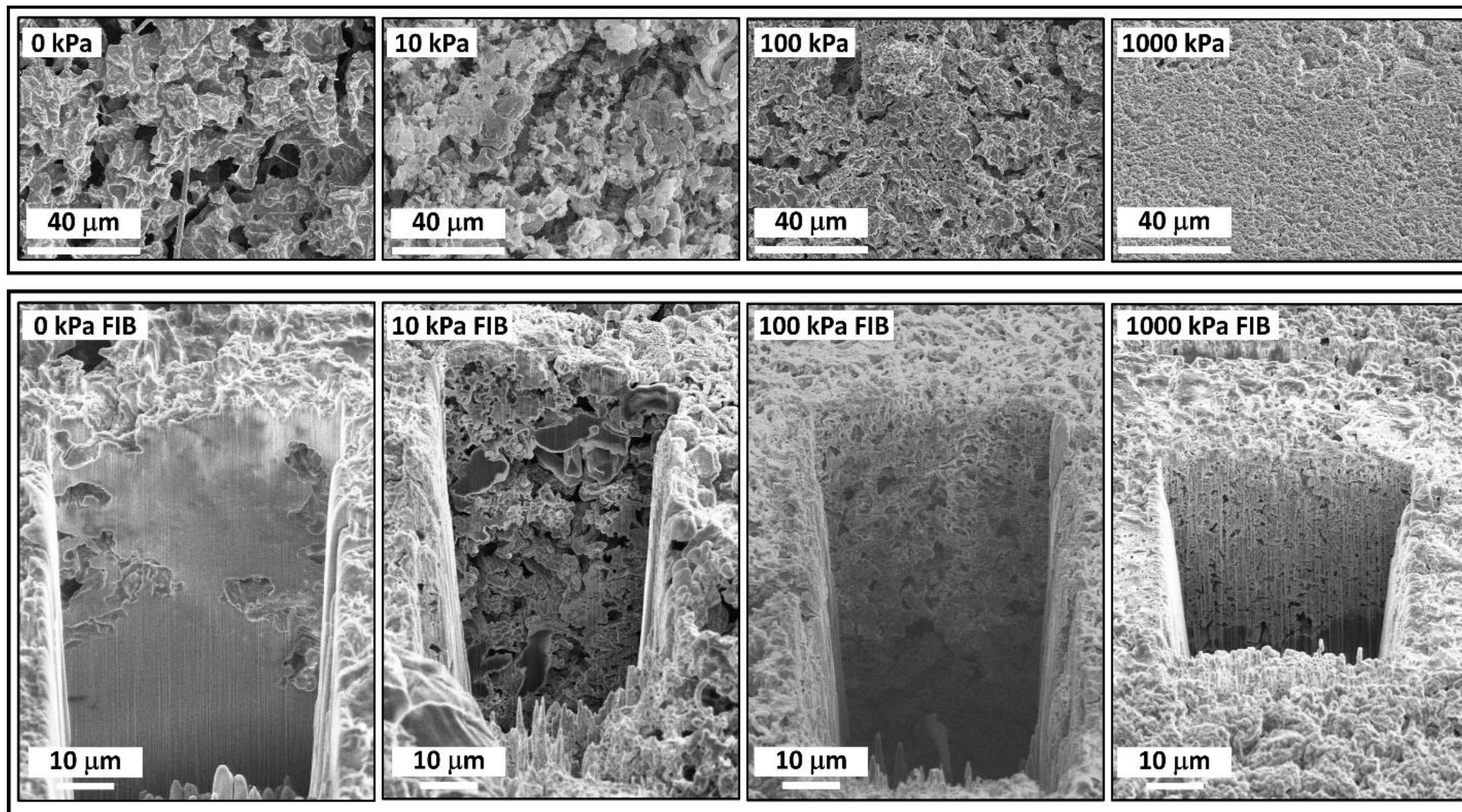
Effects of Pressure on Li Metal: High Current

High Current: 4 mA/cm²

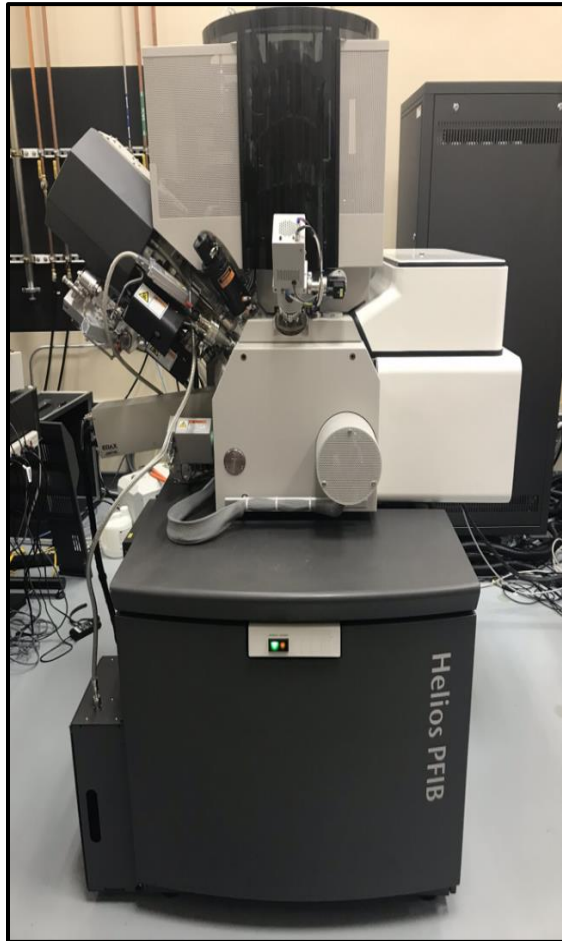


Harrison et al., *iScience* 24, 103394 (2021).
DOI: 10.1016/j.isci.2021.103394

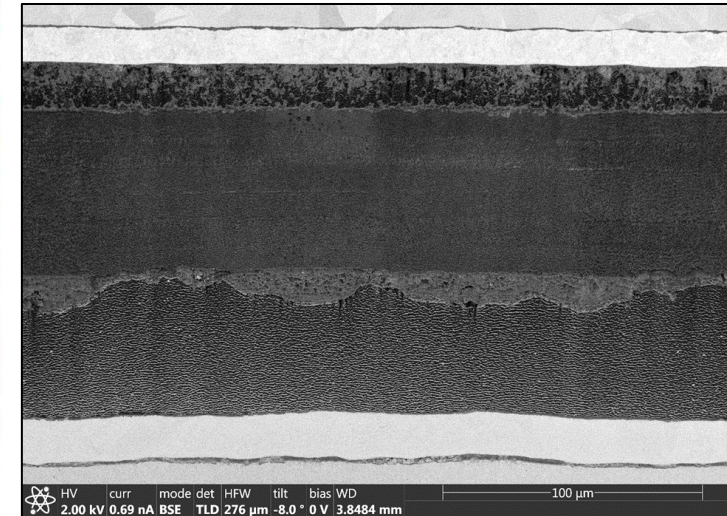
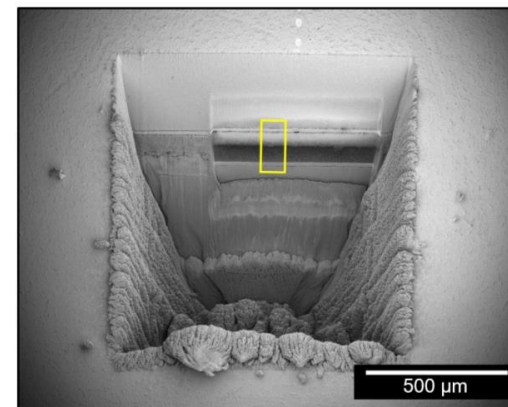
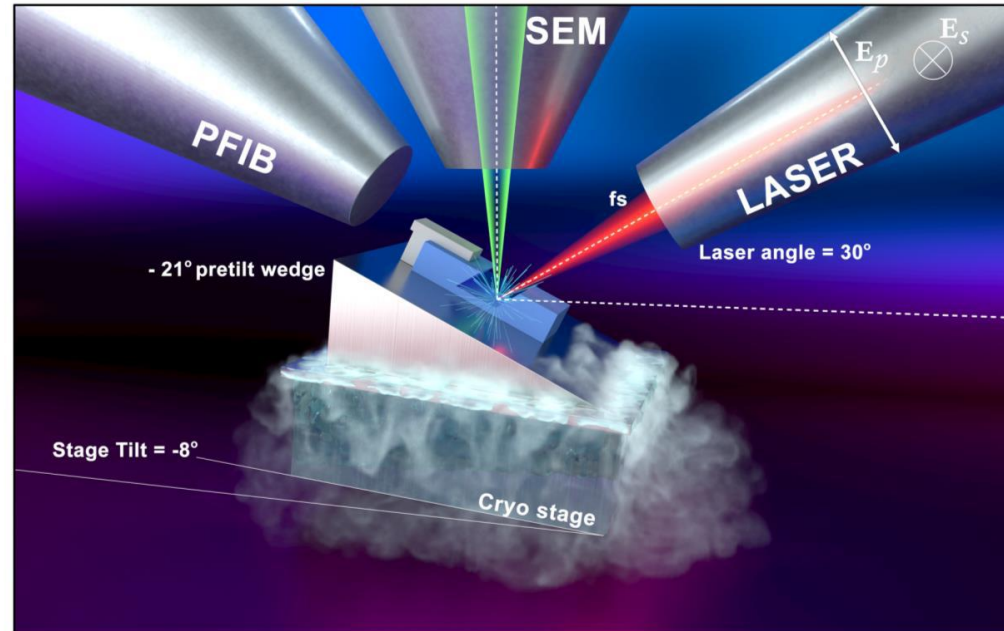
Pressure at Low Current: 51st Li Deposition Step



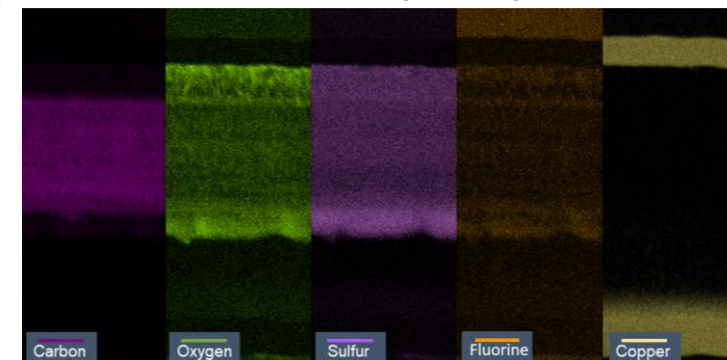
Cross-sectioning without Battery Disassembly



Helios Laser Plasma FIB
fs laser mills 15,000x faster than Ga-ion FIB



- Two Celgard 2325 Separators
- Li/SEI/electrolyte layers found



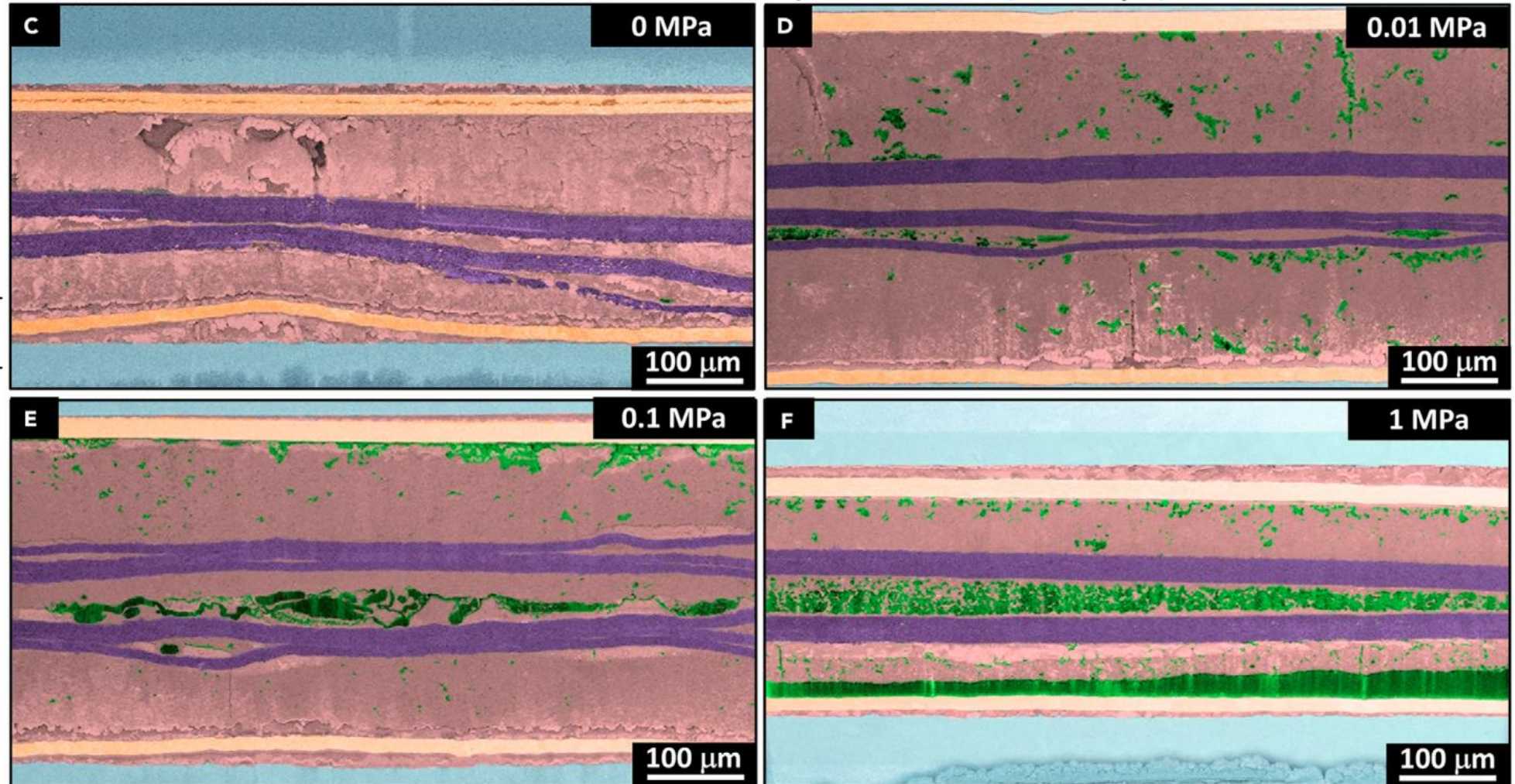
Jungjohann et al., *ACS Energy Lett.* 6, 2138 (2021).
DOI: 10.1021/acsenenergylett.1c00509



Pressure at High Current: 51st Li Deposition Step

High Current: 4 mA/cm²

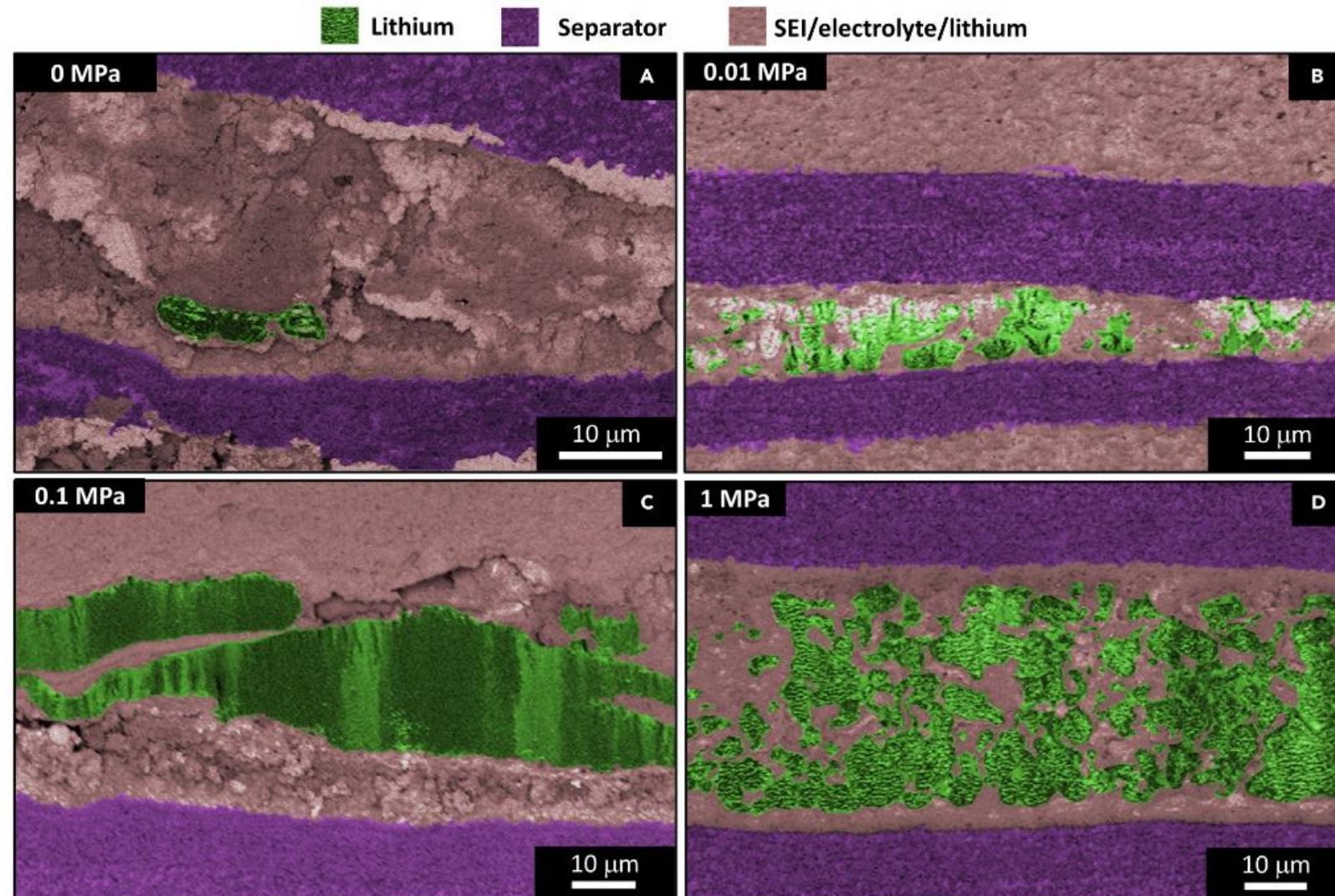
■ Lithium ■ Copper ■ Steel ■ Separator ■ SEI/electrolyte/lithium



Harrison et al., *iScience* 24, 103394 (2021).
DOI: 10.1016/j.isci.2021.103394

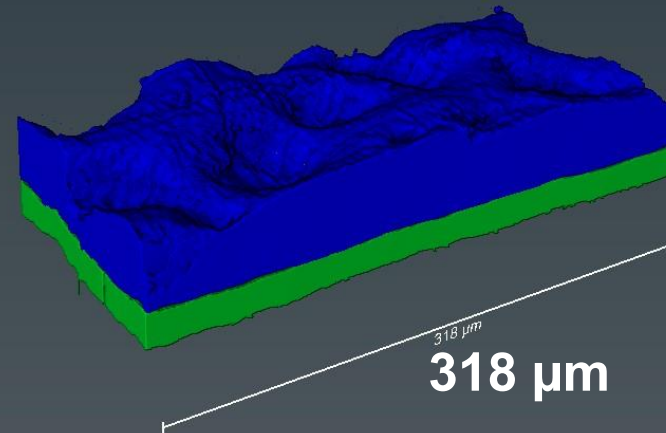
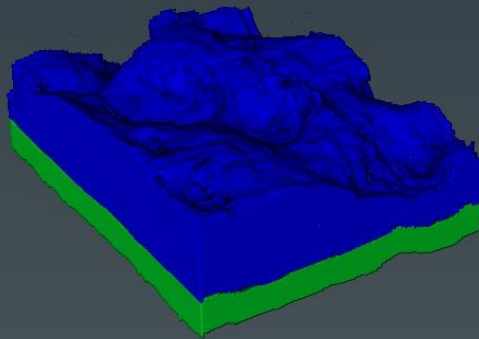
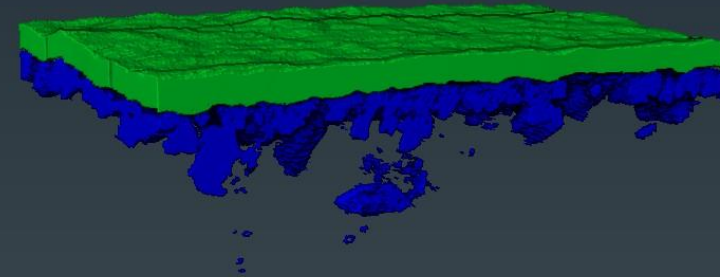
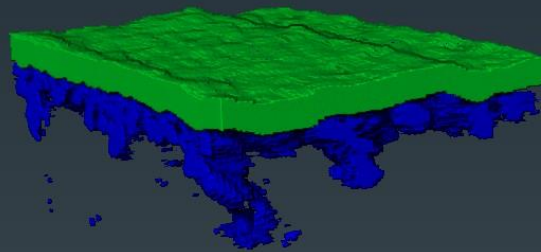
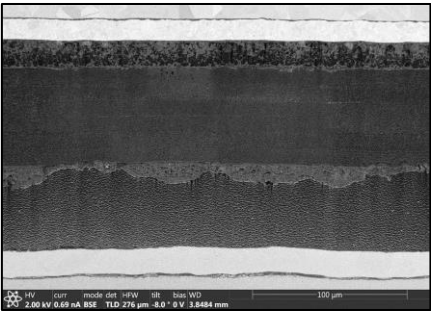
All Pressures: Li Deposition within Separator

High Current: 4 mA/cm^2



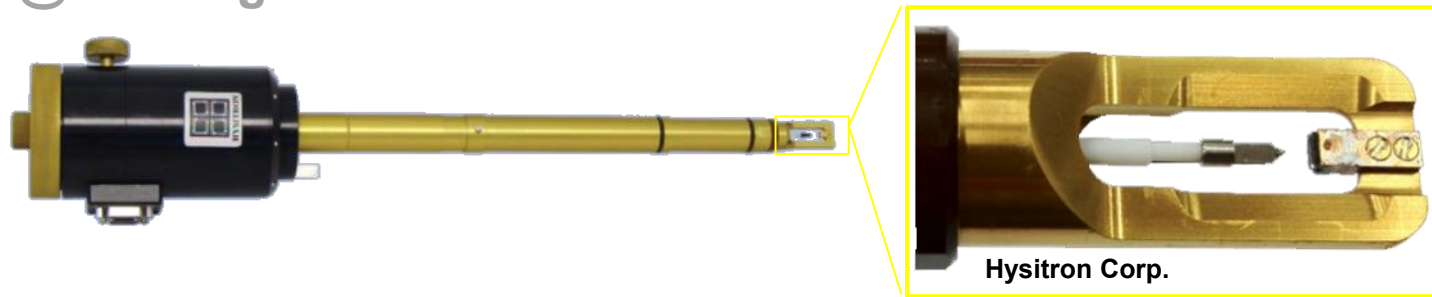
Harrison et al., *iScience* 24, 103394 (2021).
DOI: 10.1016/j.isci.2021.103394

fs Laser Slice-N-View of Battery Stack



Dr. Khalid Hattar
khattar@sandia.gov

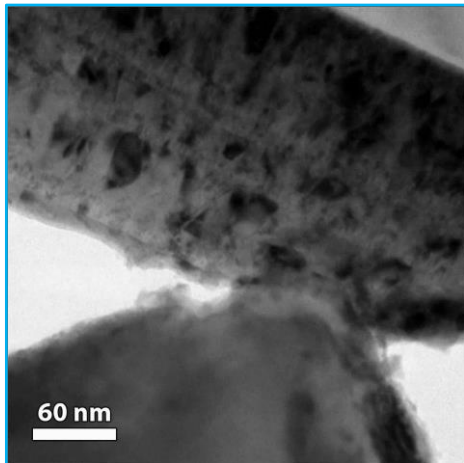
PI-95 Picoindenter in TEM



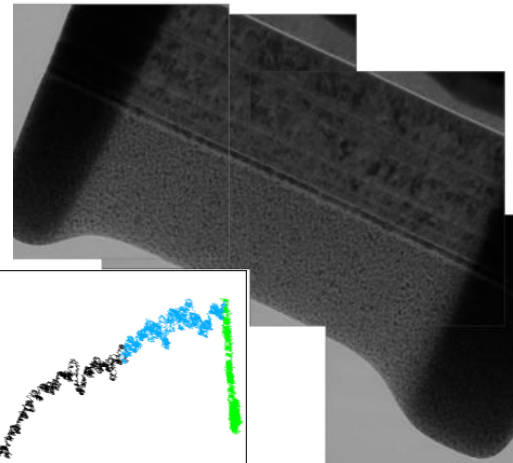
Notched Bar



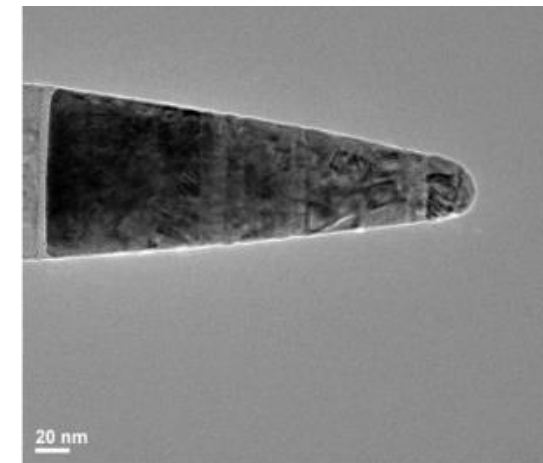
Nanoindentation



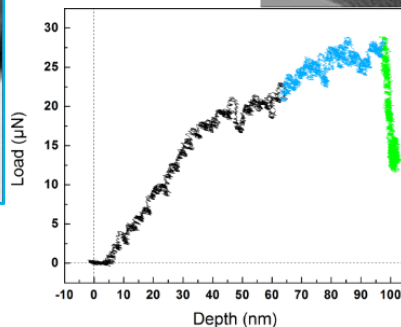
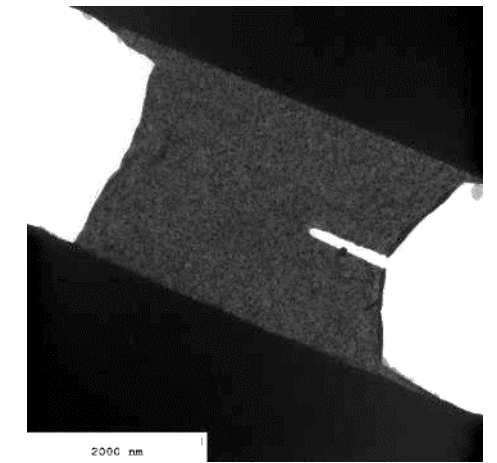
I Beams



Nanopillars

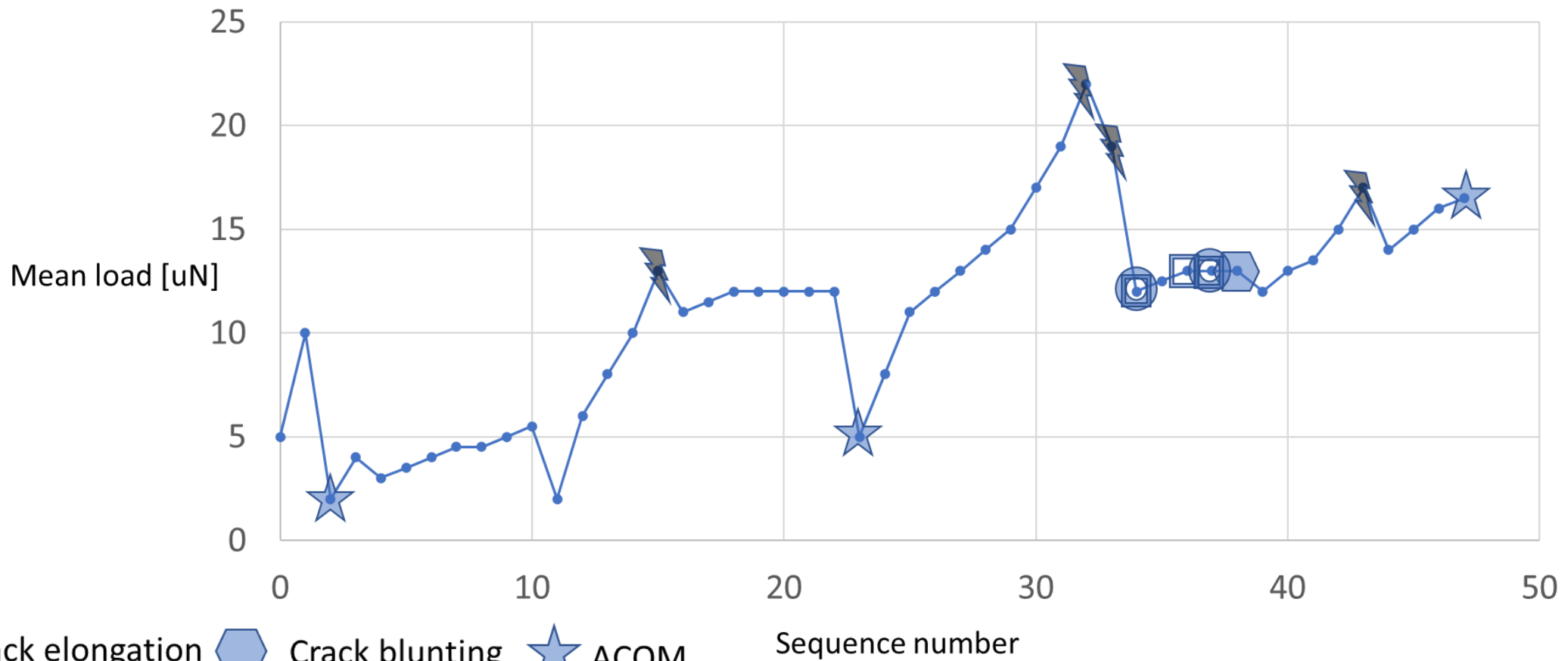


Notched Micro Tension Bars



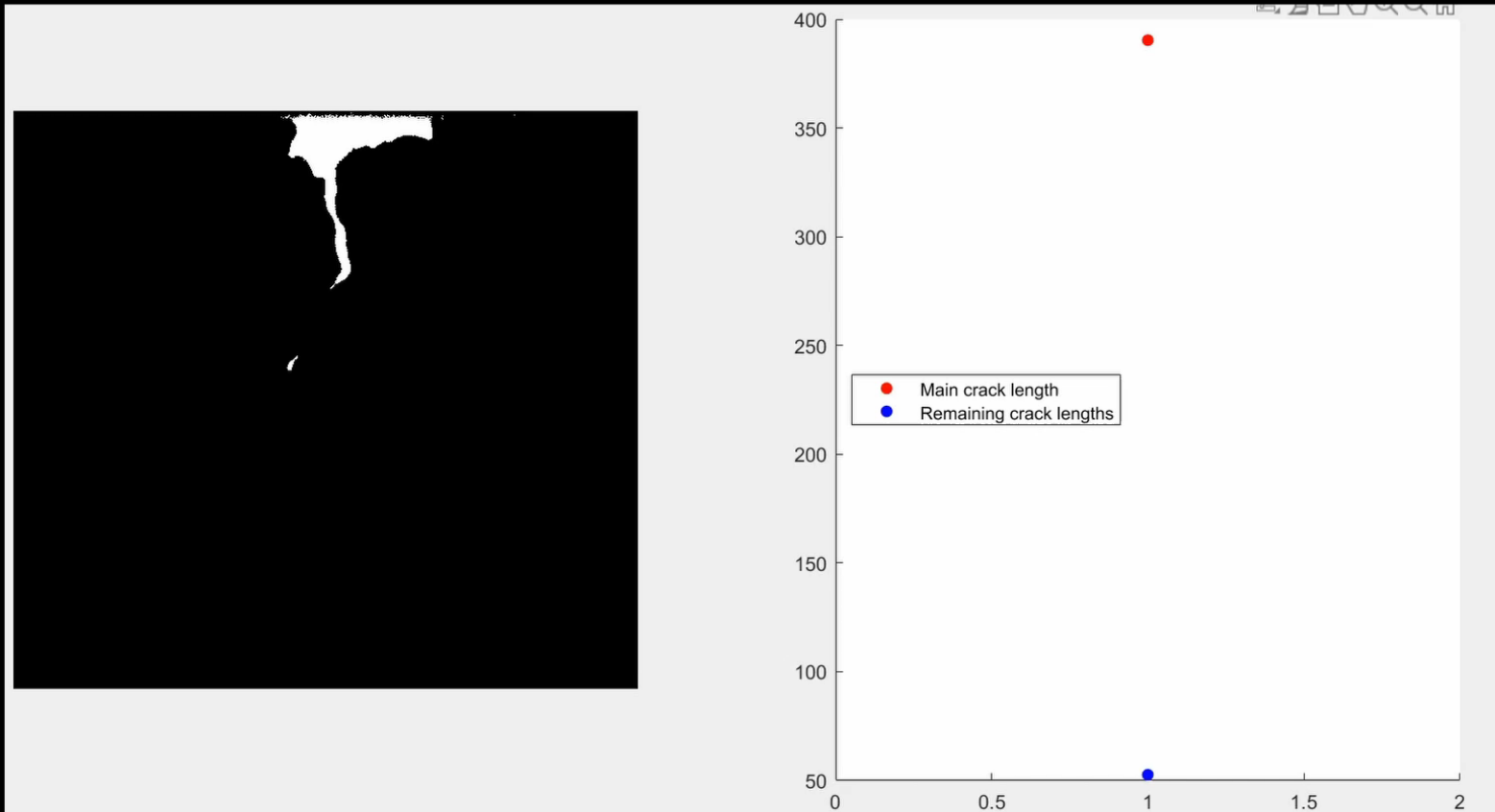
A6-2 Experiment

Mean load vs. test number (R=0.4 for all tests)

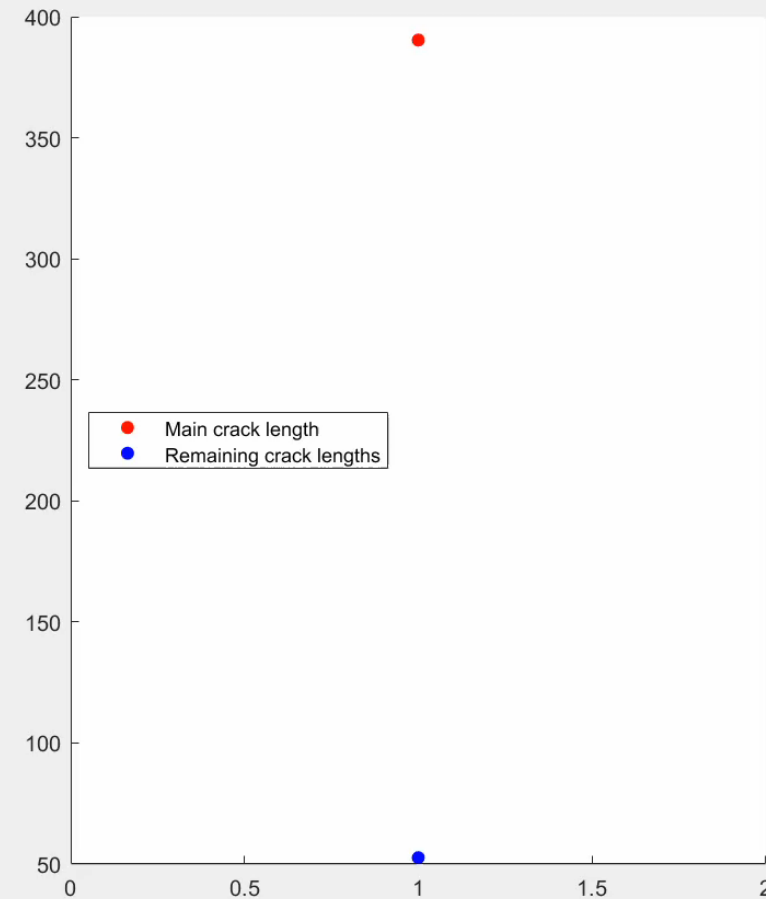
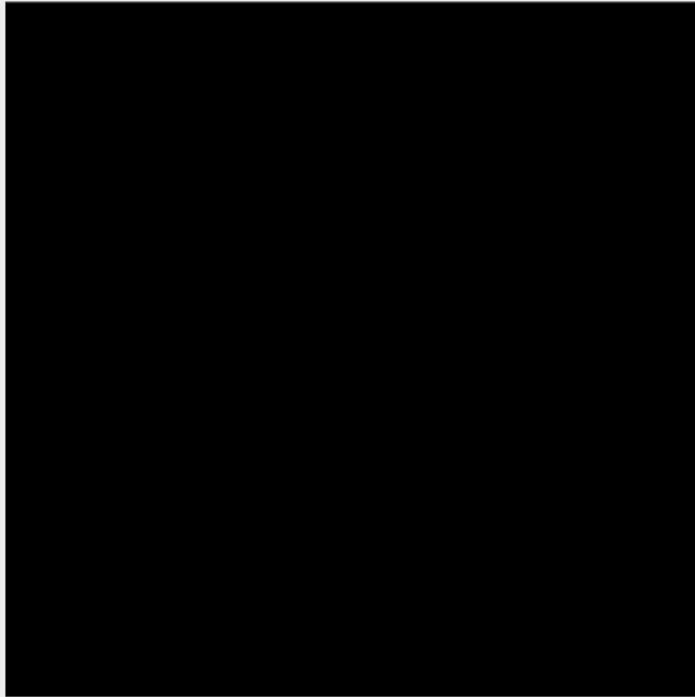


 Crack elongation
  Crack blunting
  ACOM
  Interesting video
  280 fps video

A6-2 Experiment



A6-2 Experiment

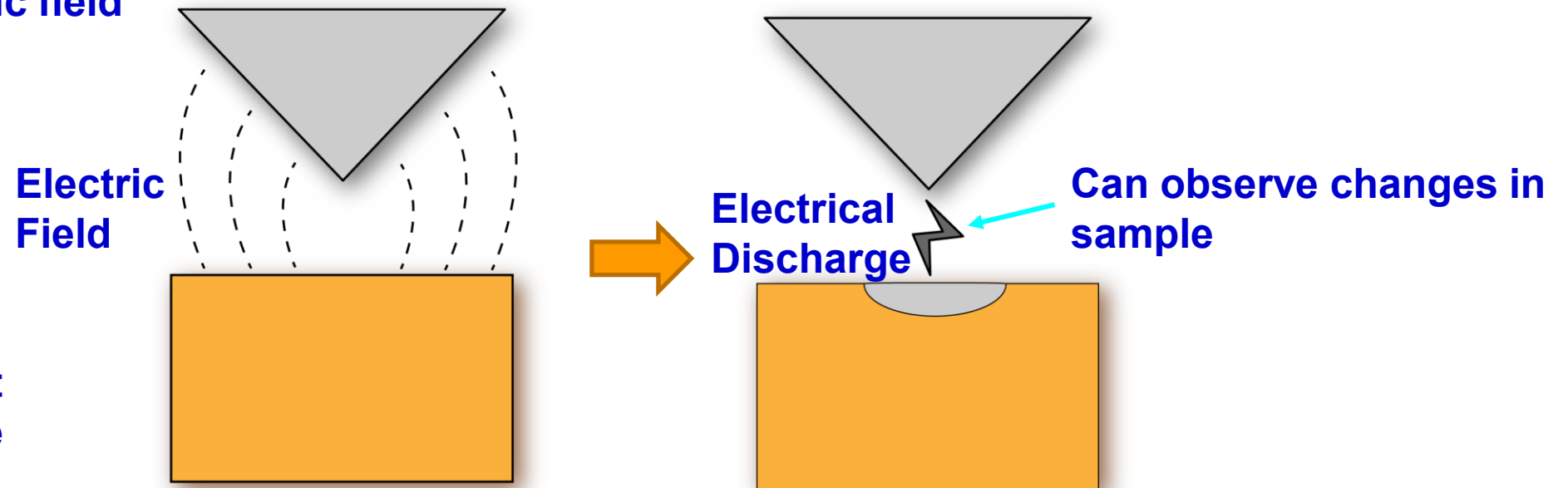


Future: Electrical-Mechanical TEM

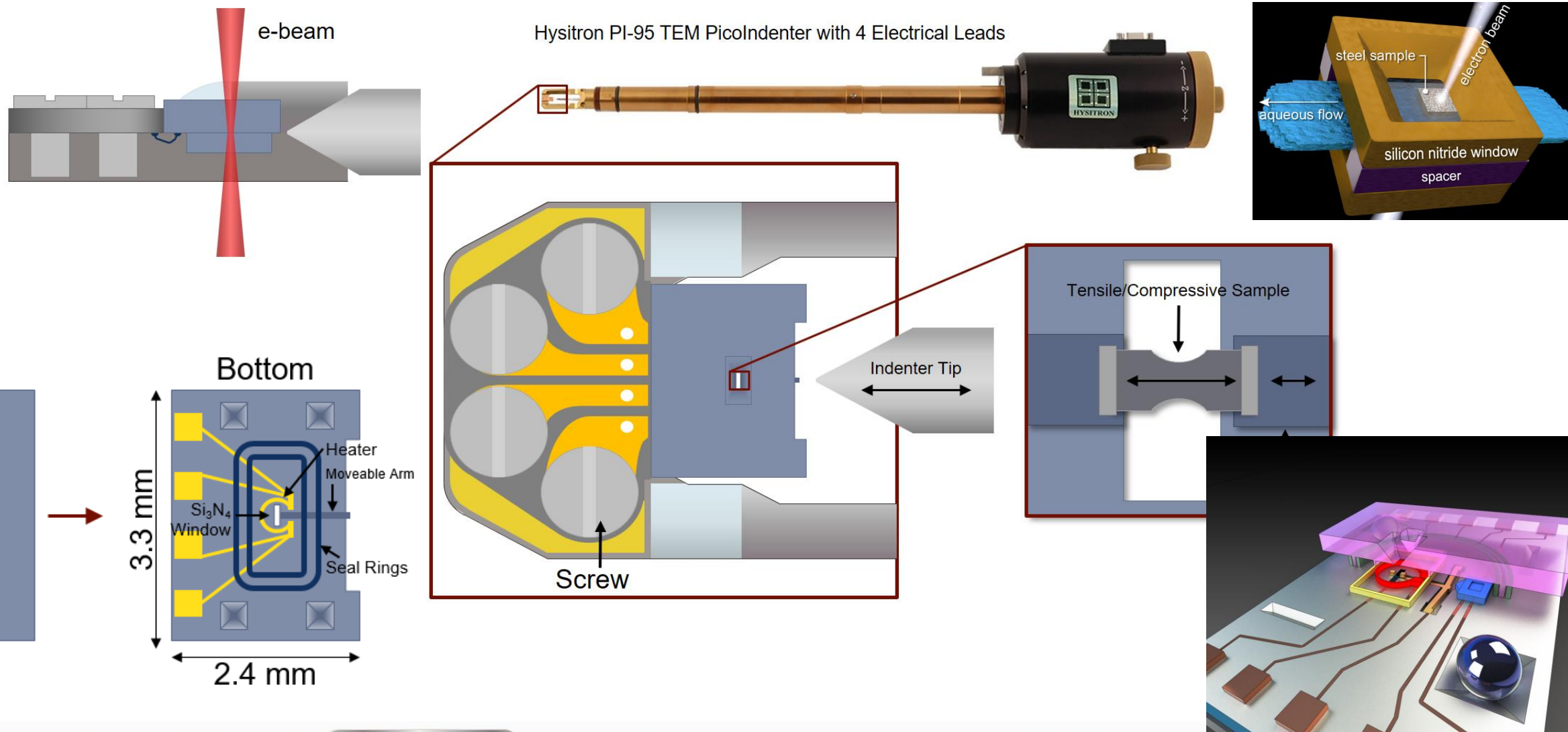
- Electromigration
- Electrical discharge
- Can combine with laser stimulus

Nanoindentation tip from PI-95
that applies the electric field

Sample can be at
high temperature



Future: Electrochemical-Mechanical TEM



Thank You! Questions?



This work was performed, in part, at the Center for Integrated Nanotechnologies, an Office of Science User Facility operated for the U.S. Department of Energy (DOE) Office of Science. Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. DOE's National Nuclear Security Administration under contract DE-NA-0003525. The views expressed in the article do not necessarily represent the views of the U.S. DOE or the United States Government.

