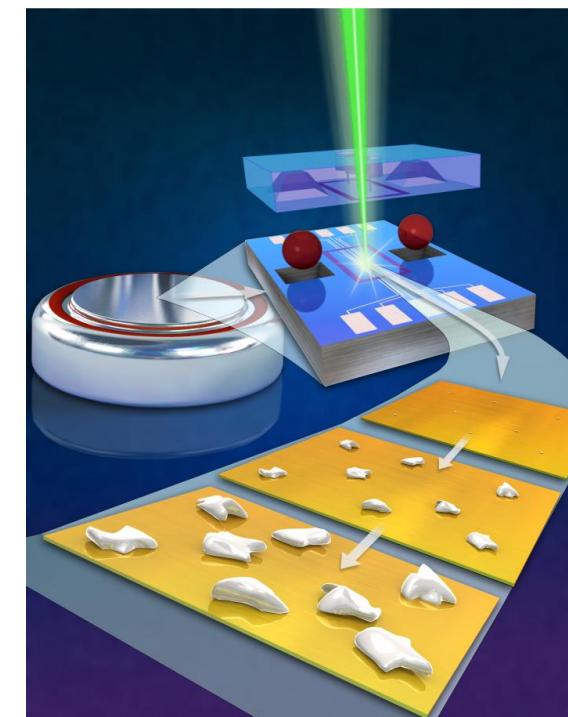
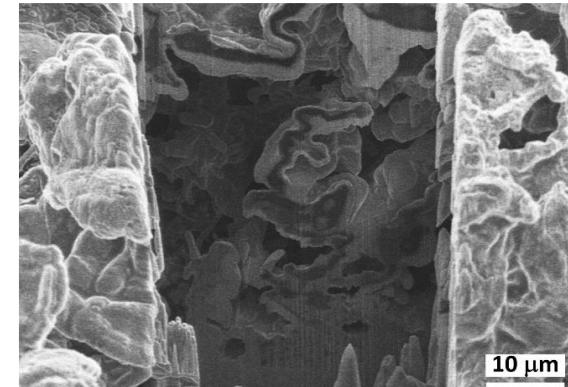


# Nanoscale Characterization of Electrochemical-Mechanical Mechanisms with Electron Microscopy



Katherine Jungjohann

National Renewable Energy Laboratory

Formerly at Sandia National Laboratories

This work was performed, in part, at the Center for Integrated Nanotechnologies, an Office of Science User Facility operated for the U.S. Department of Energy (DOE) Office of Science. Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. DOE's National Nuclear Security Administration under contract DE-NA-0003525. The views expressed in the article do not necessarily represent the views of the U.S. DOE or the United States Government.



# About the Presenter

- **Katherine Jungjohann, National Renewable Energy Laboratory**
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  - Microscopy Society of America, EMLG FIG Lead
- **Characterization of Solid-Liquid Interfaces in Energy Systems**
- **Co-authors:**
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  - **Katharine Harrison**, Sandia National Laboratories
  - **Laura Merrill**, Sandia National Laboratories
  - **Steven Randolph**, Oak Ridge National Laboratory
  - **Renae Gannon**, Thermo Fisher Scientific
  - **Zac Milne**, Sandia National Laboratories
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**ThermoFisher**  
SCIENTIFIC



**Sandia**  
National  
Laboratories



**LDRD**

Laboratory Directed  
Research and Development



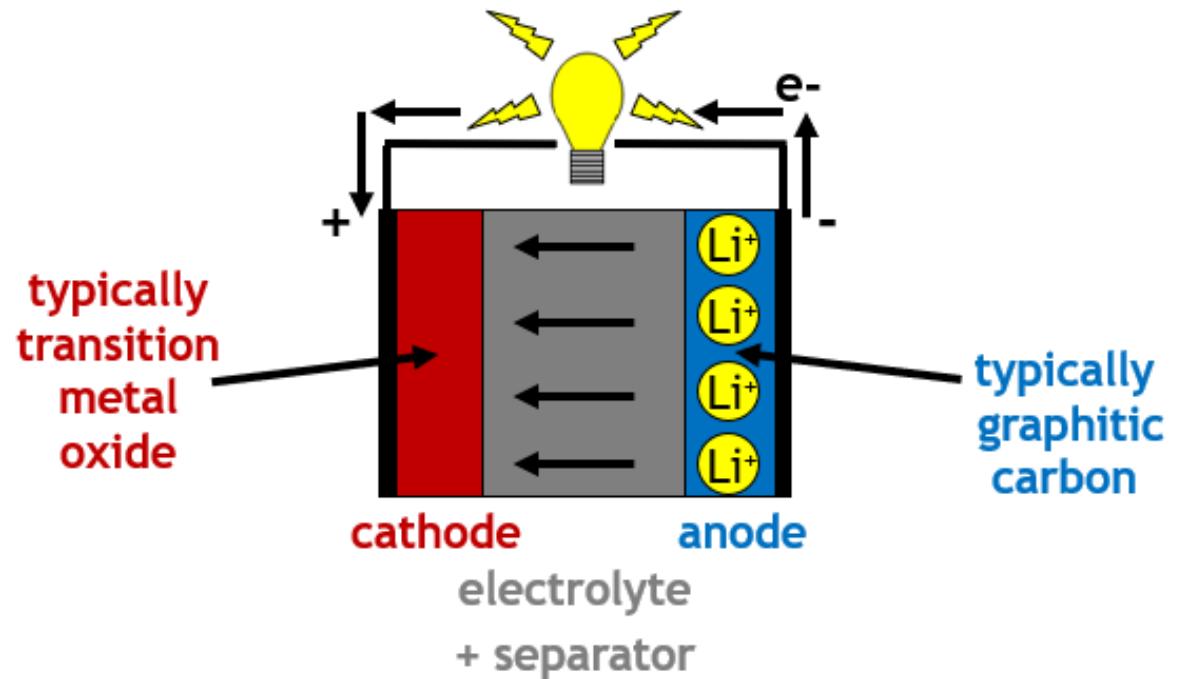
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Lithium metal anodes can theoretically increase battery capacity by

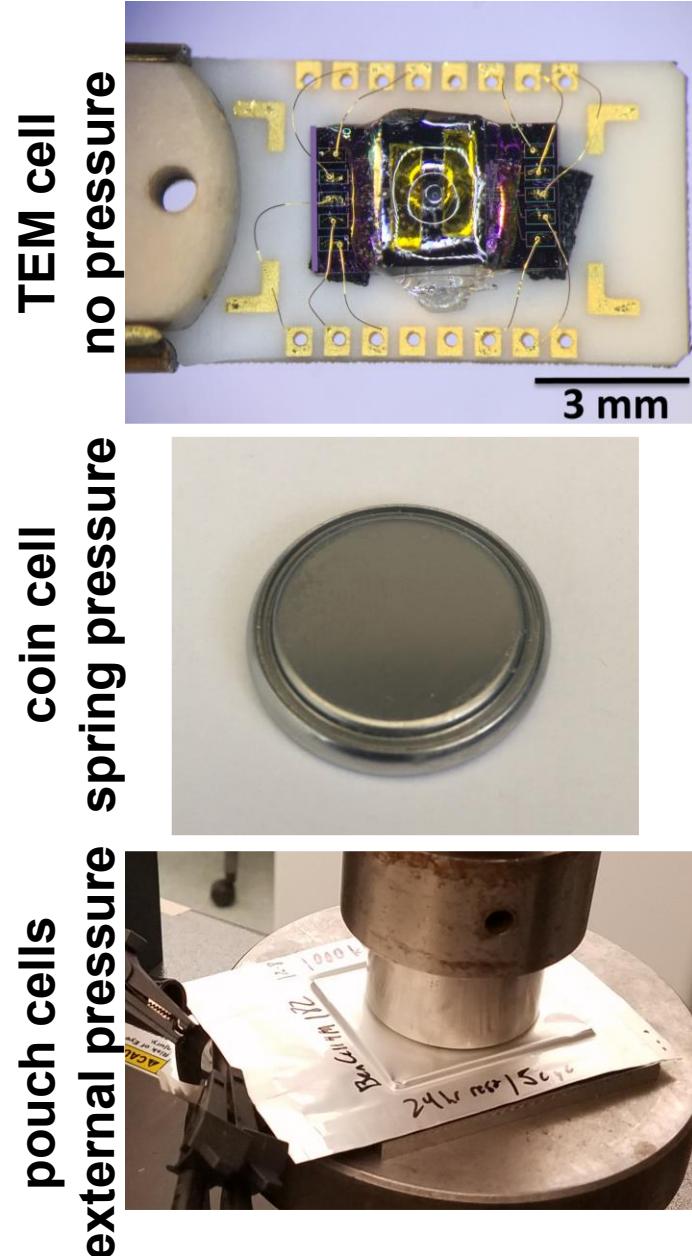
10 X

# Traditional Li-Ion Batteries

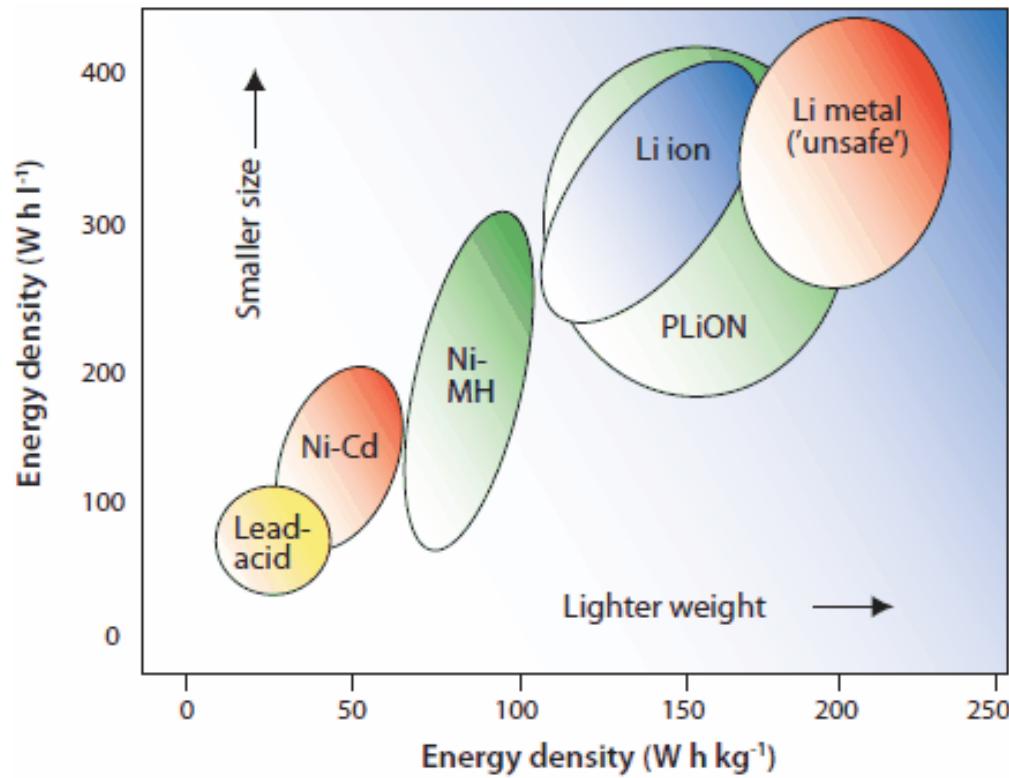


## Discharging Battery

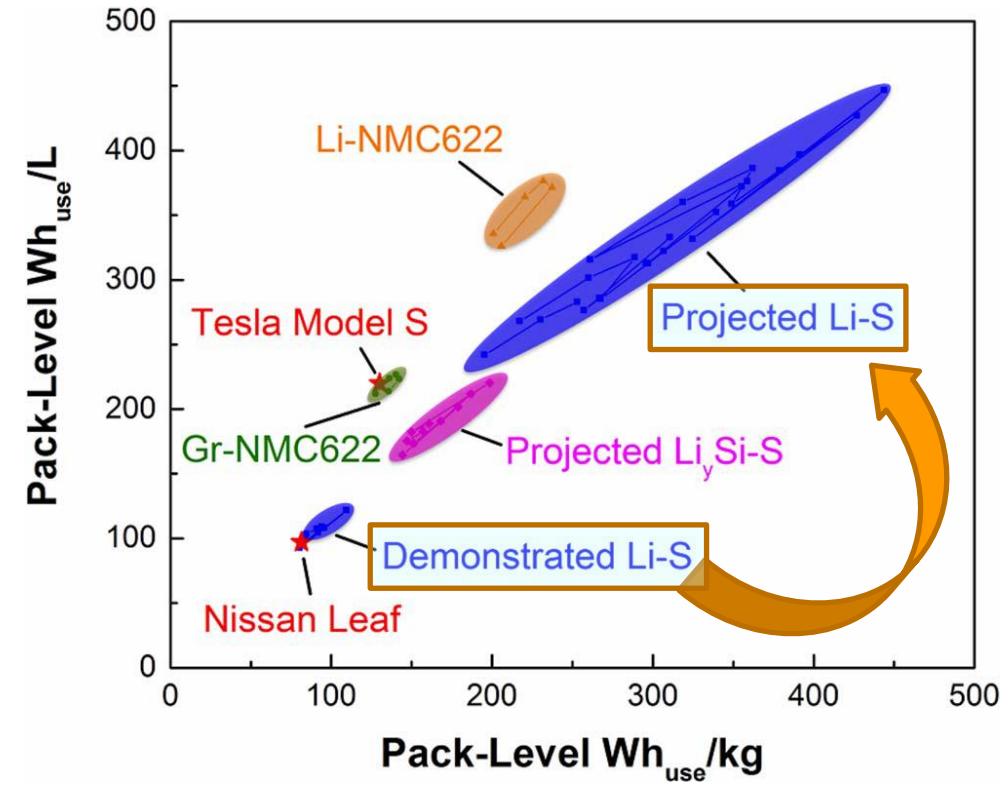
- $\text{Li}^+$  from the **anode** through the **electrolyte** and into the **cathode**
- $e^-$  move through the external circuit from the **anode** to the **cathode** (from - to + charge)



# How Can We Improve Rechargeable Batteries?



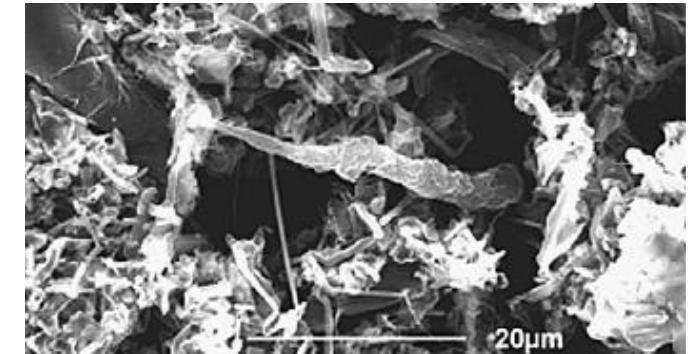
J.-M. Tarascon and M. Armand *Nature*  
DOI: 10.1038/35104644



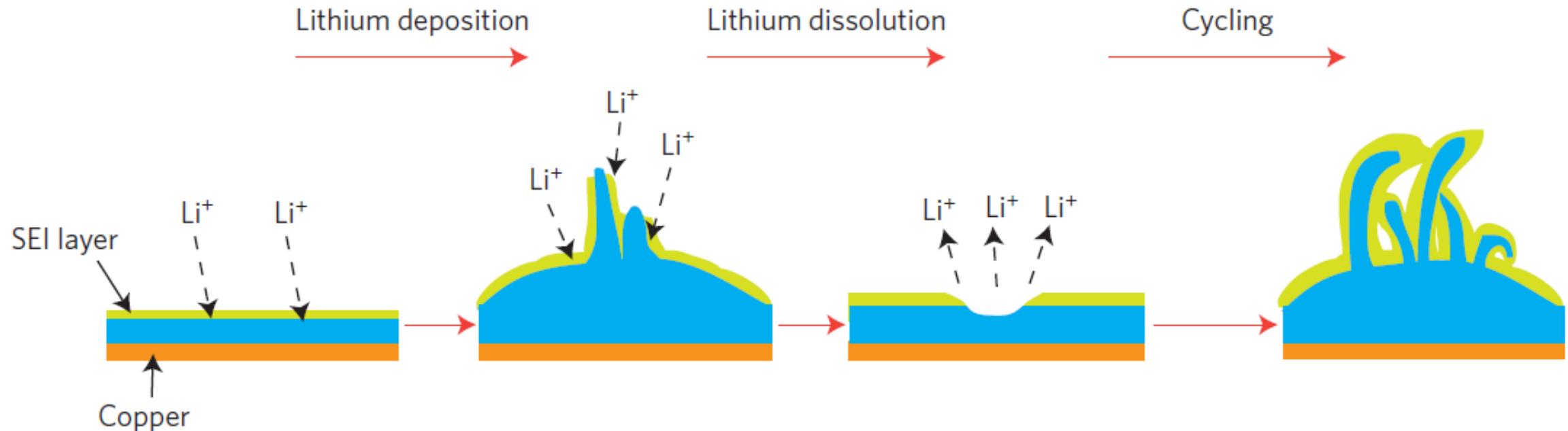
K.G. Gallagher et al. *J. Electrochem. Soc.*  
DOI: 10.1149/2.0611506jes

# Rechargeable Li Metal: Chemical & Mechanical

- Uncontrolled morphology → many problems
  - Short circuits = fire
  - Excessive solid electrolyte interphase (SEI) = low Coulombic efficiency, high impedance, and Li consumption
  - Li gets stranded and disconnected = “dead” Li

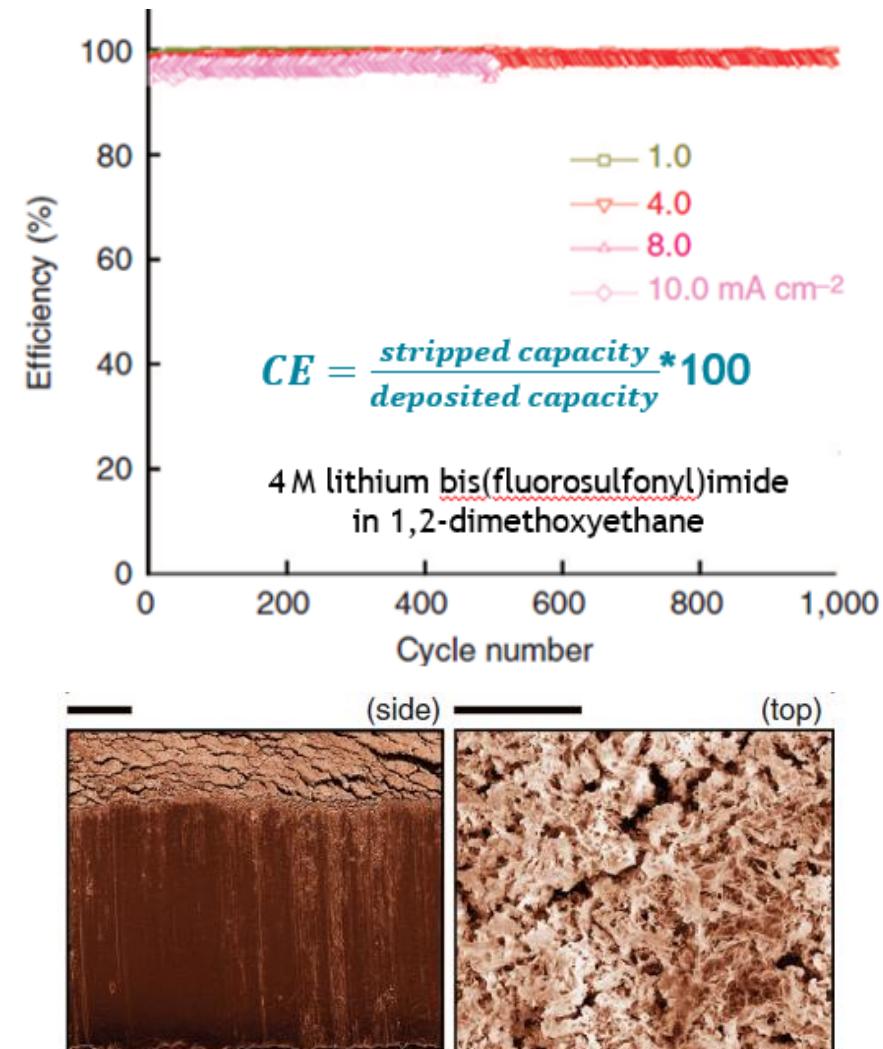
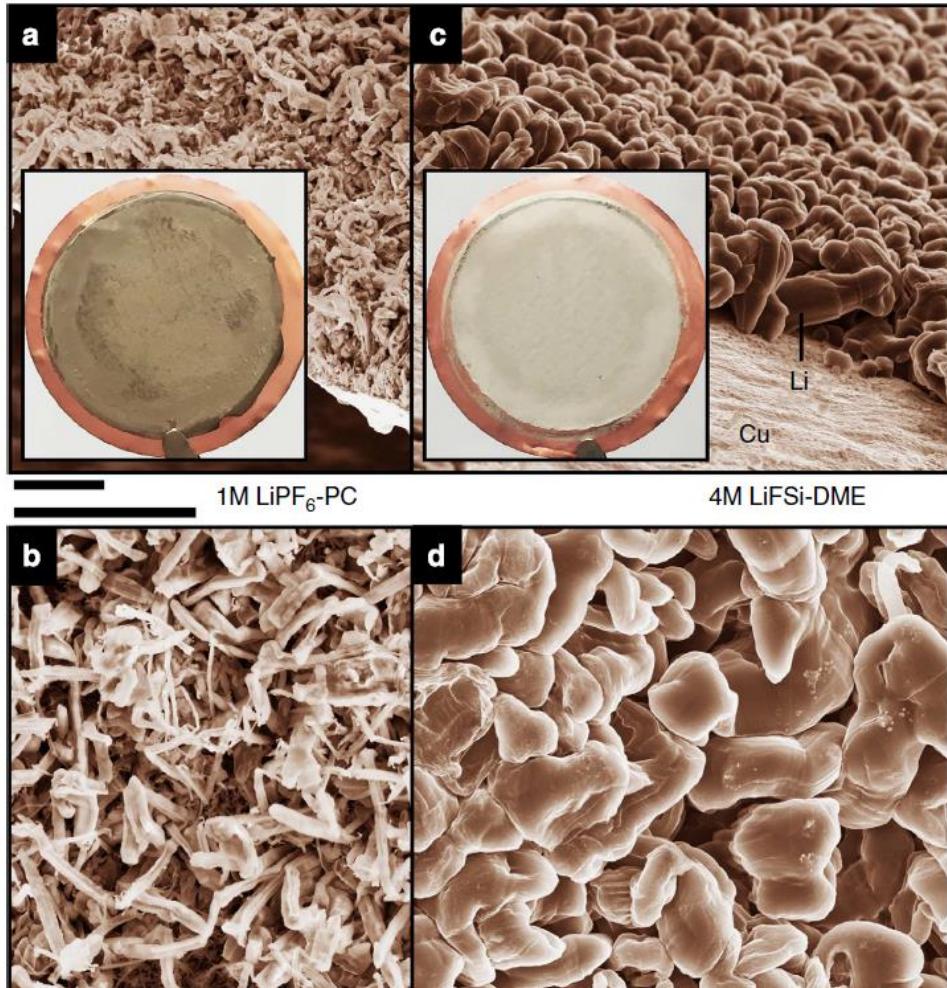


Gireaud et al., *Electrochim. Comm.* 8, 1639 (2006)



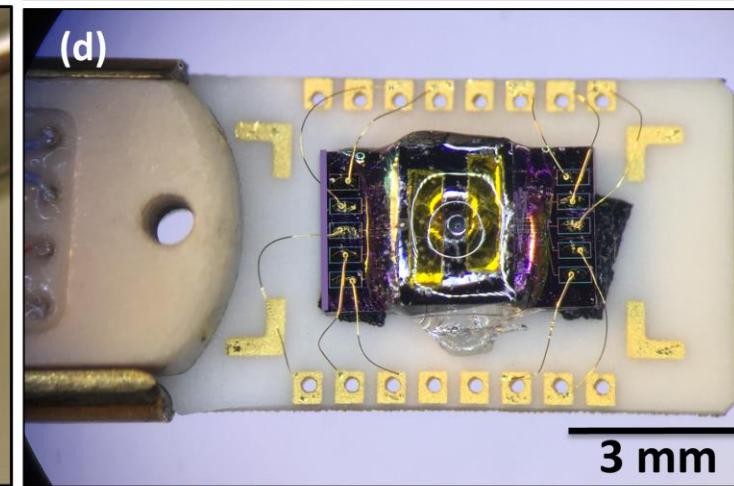
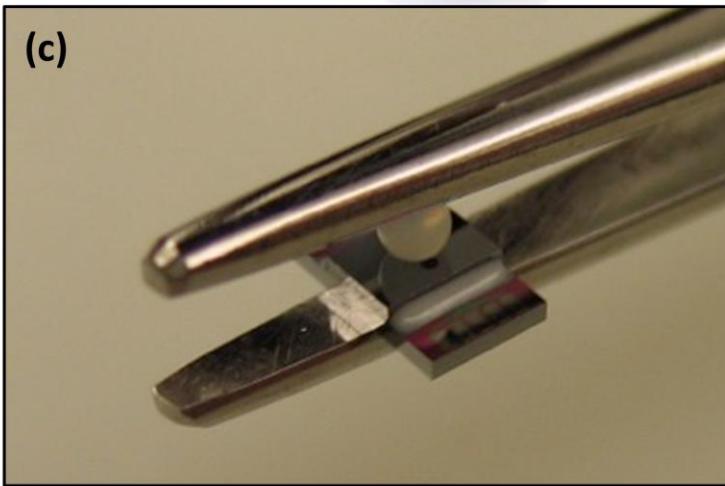
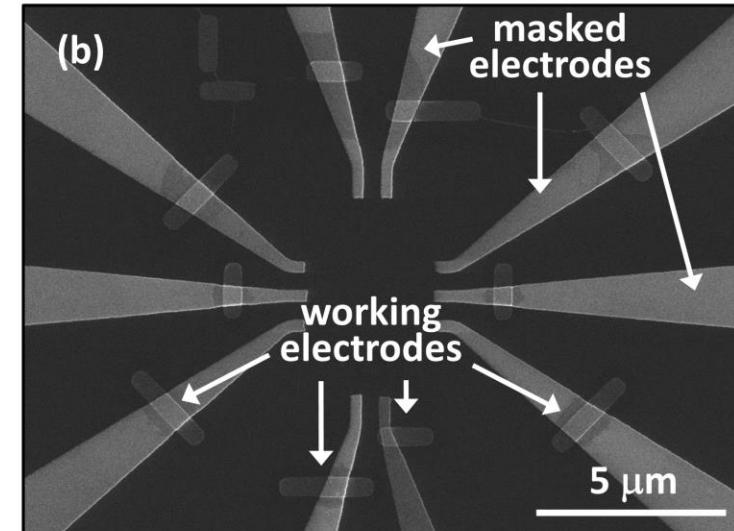
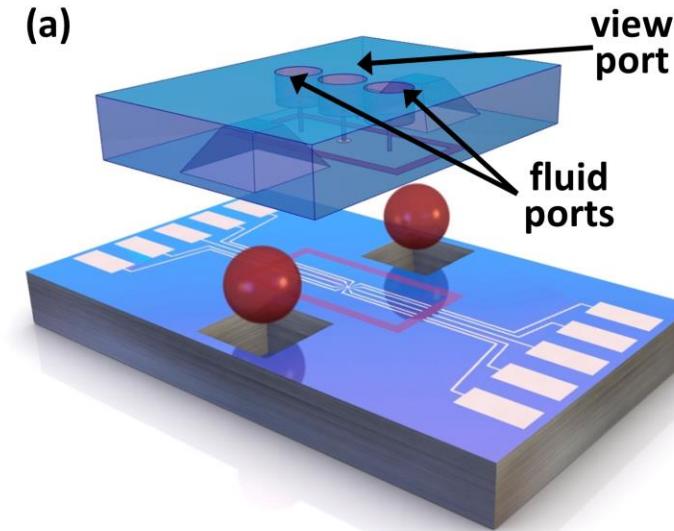
# Lithium Self-Discharge and Prevention

Qian et al. *Nature Communications* DOI: 10.1038/ncomms7362

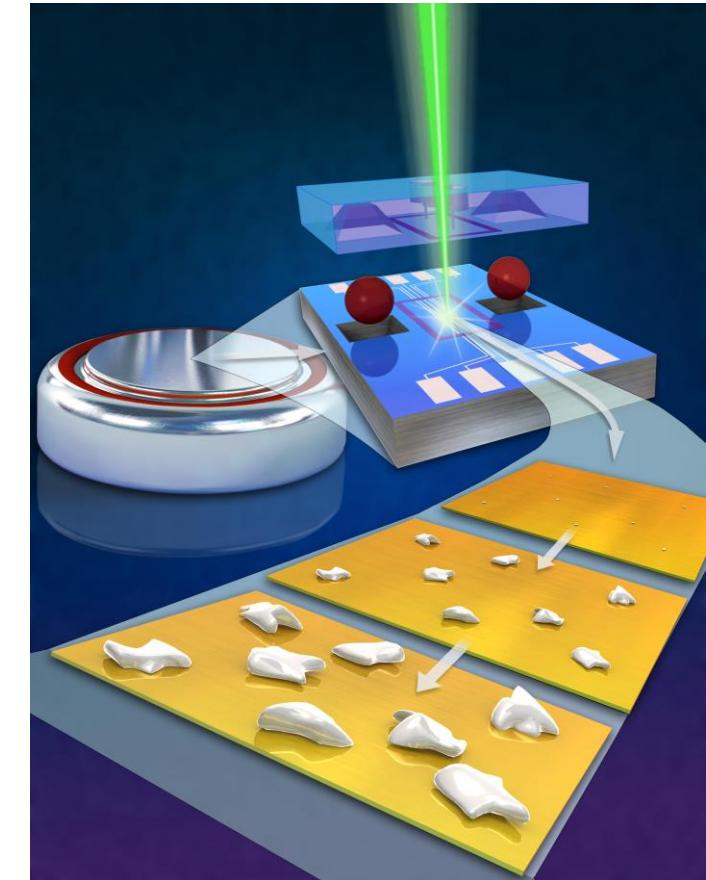


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# CINT's Electrochemical TEM Discovery Platform



\*planar electrodes on insulated chip surface with no separator

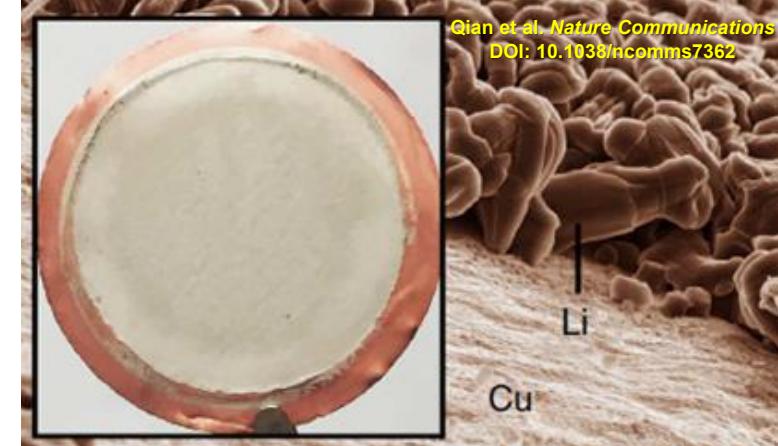
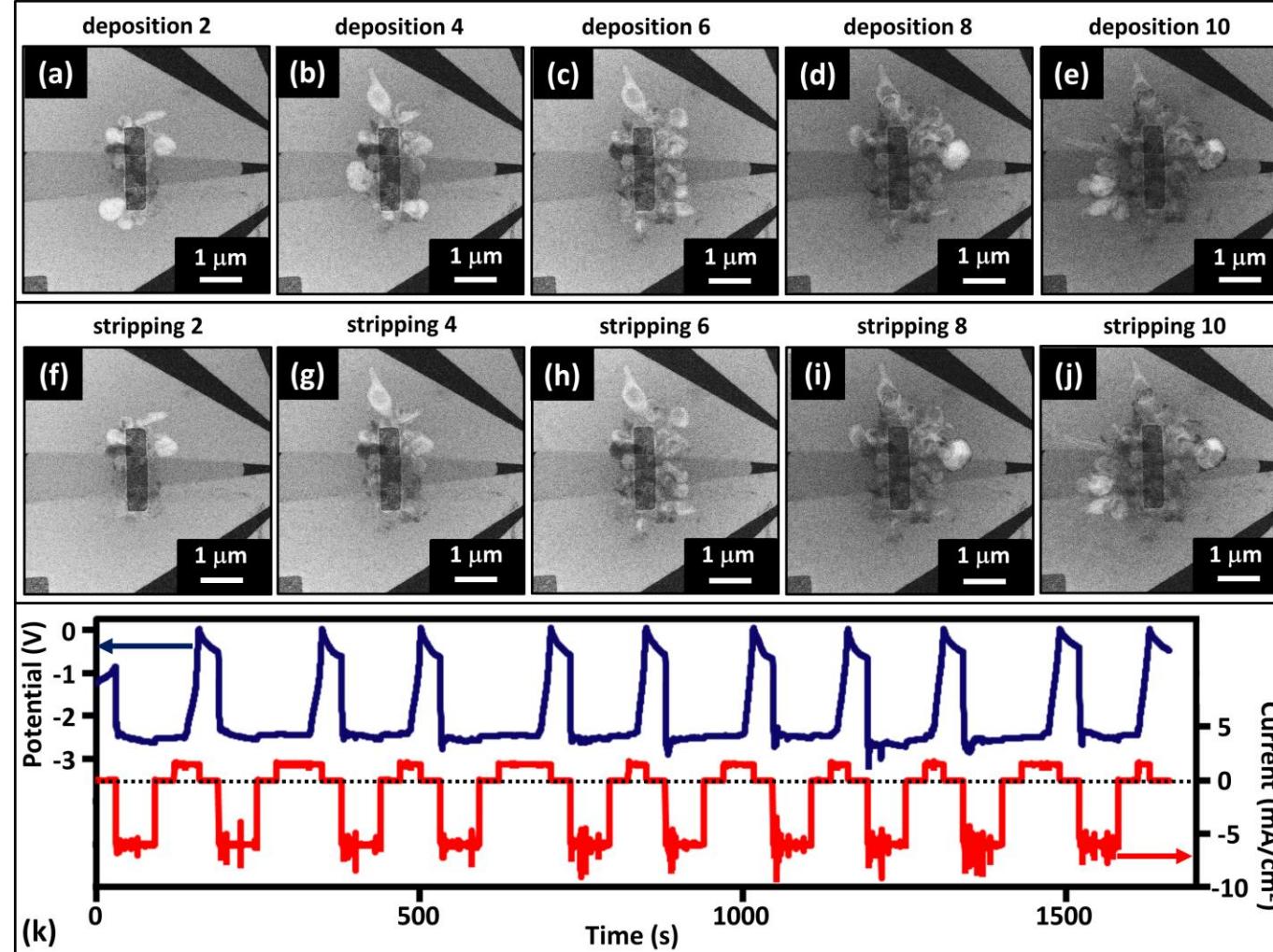


Harrison et al. *ACS Nano*  
10.1021/acsnano.7b05513



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# In-Situ Nanoscale Li Electrodeposition



- Morphology images after every deposition and stripping step (Li = white)
- CE VERY low ( $18\% \pm 9\%$ )
- Widely varied morphology

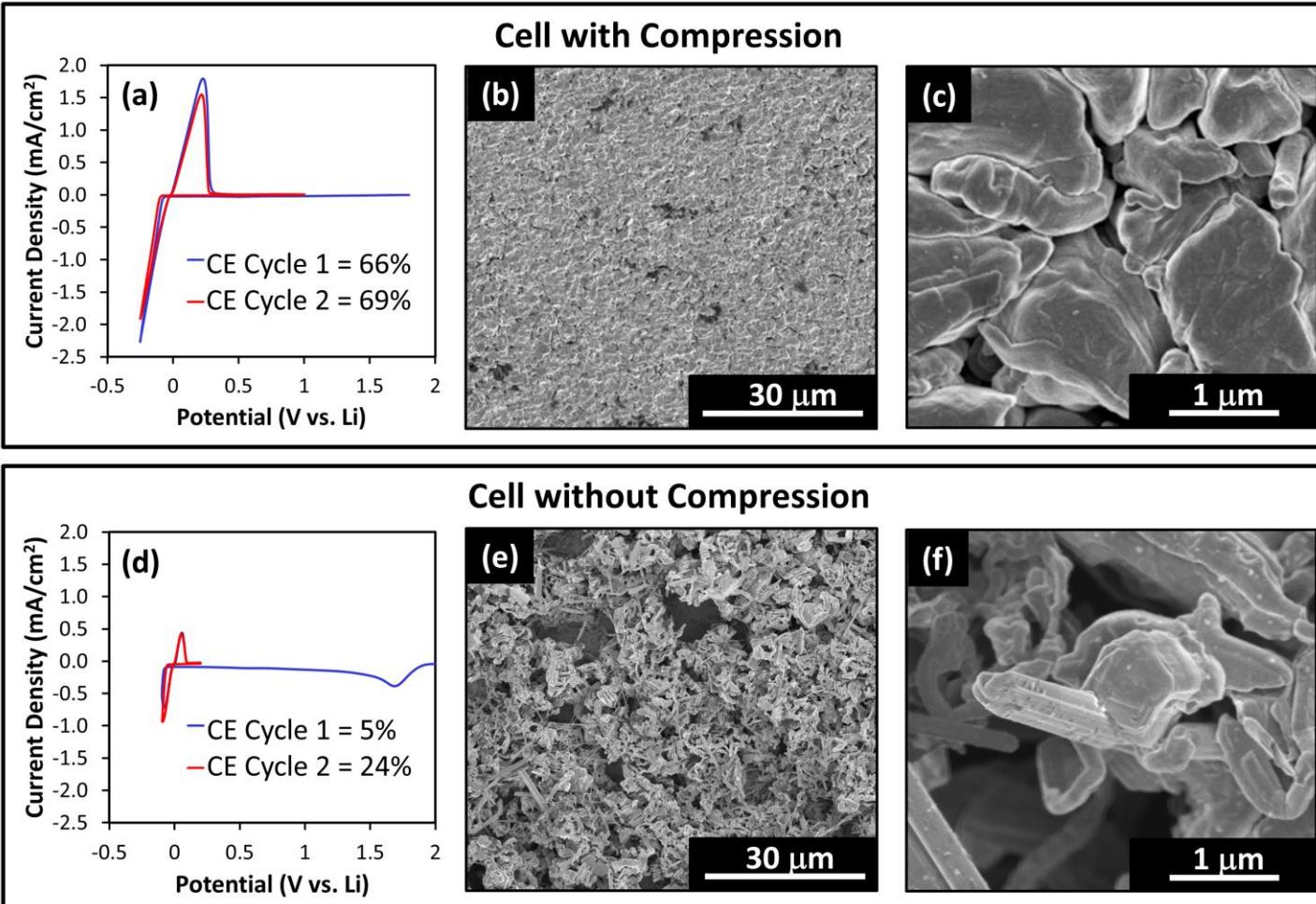
bright field imaging  
WE = 60 nm Ni/5 nm Ti  
deposit 6  $\text{mA}/\text{cm}^2$   
strip 1.5  $\text{mA}/\text{cm}^2$   
capacity = 0.1  $\text{mAh}/\text{cm}^2$

Harrison et al. *ACS Nano*  
DOI: 10.1021/acsnano.7b05513

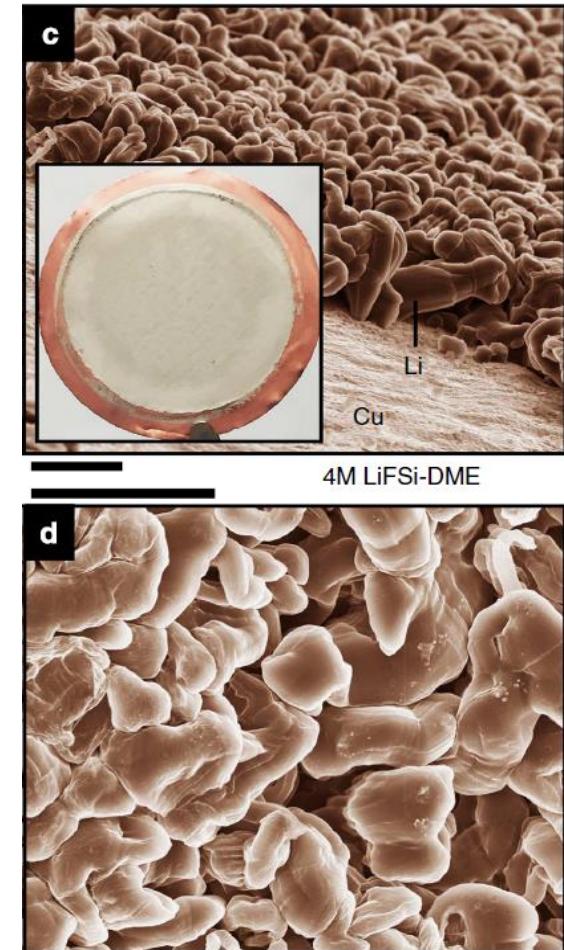


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# Applied Pressure Critical for High-Density Li

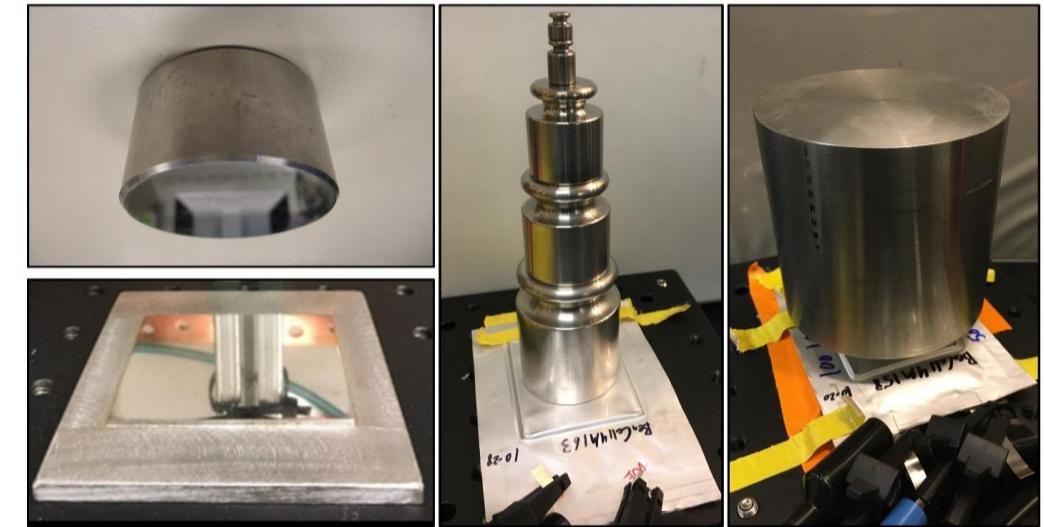
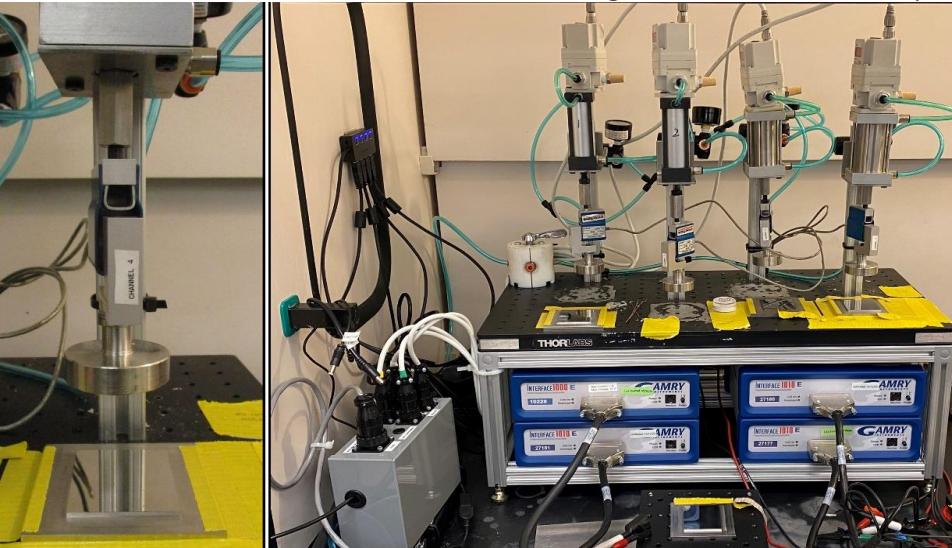
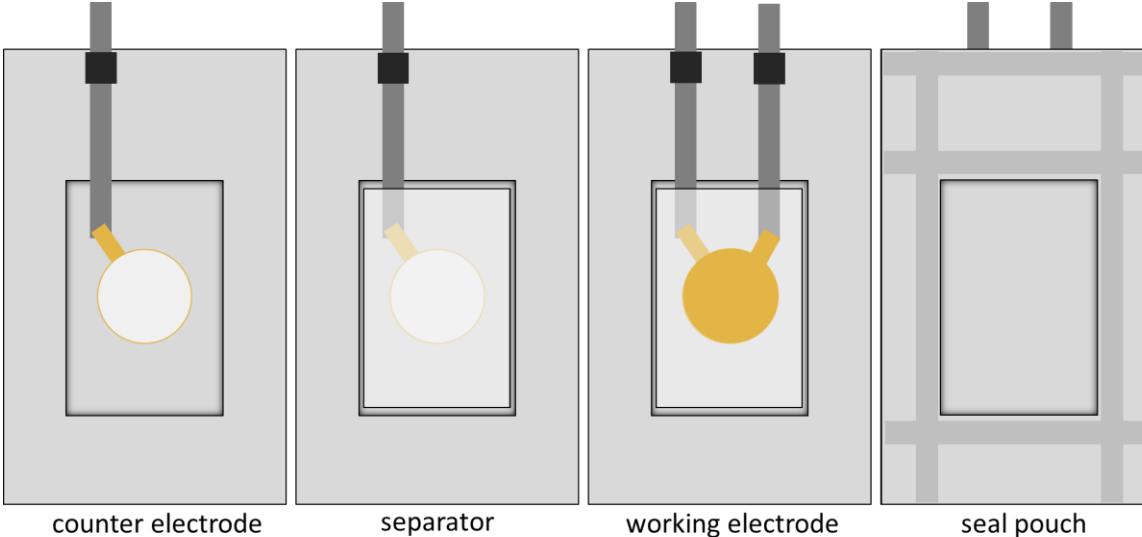


Qian et al. *Nature Communications*  
DOI: 10.1038/ncomms7362



Harrison et al. *ACS Nano*  
10.1021/acsnano.7b05513

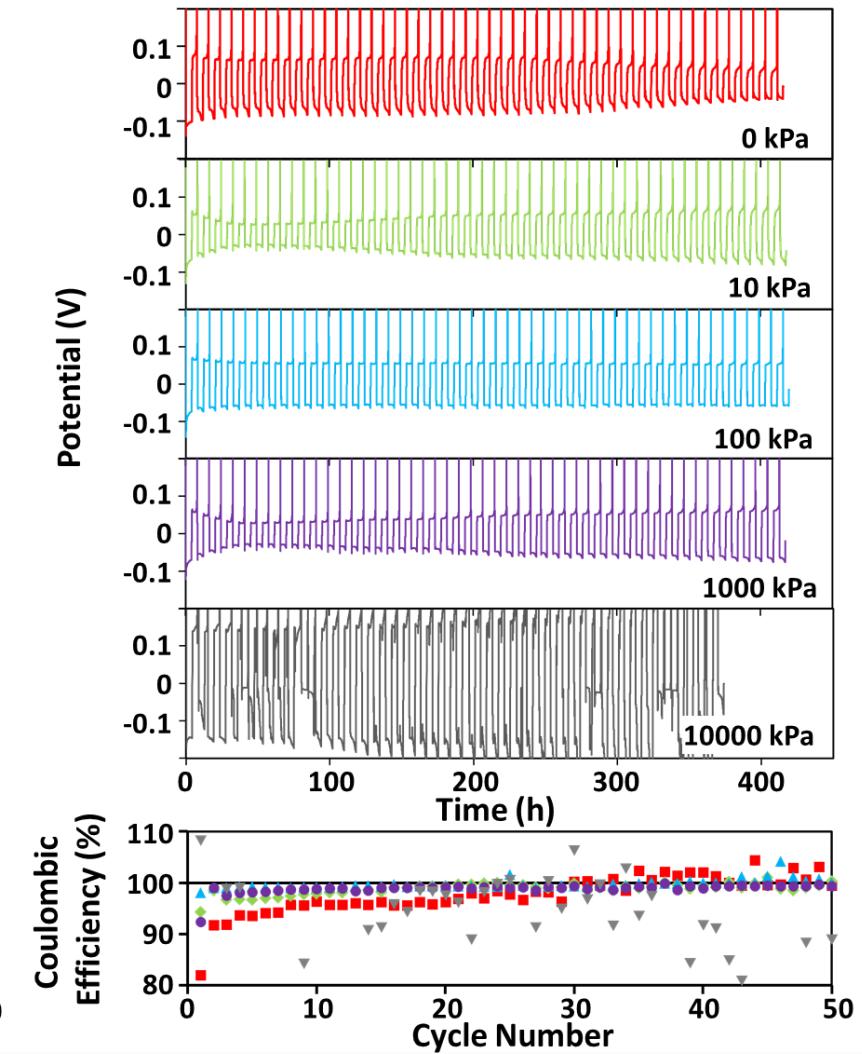
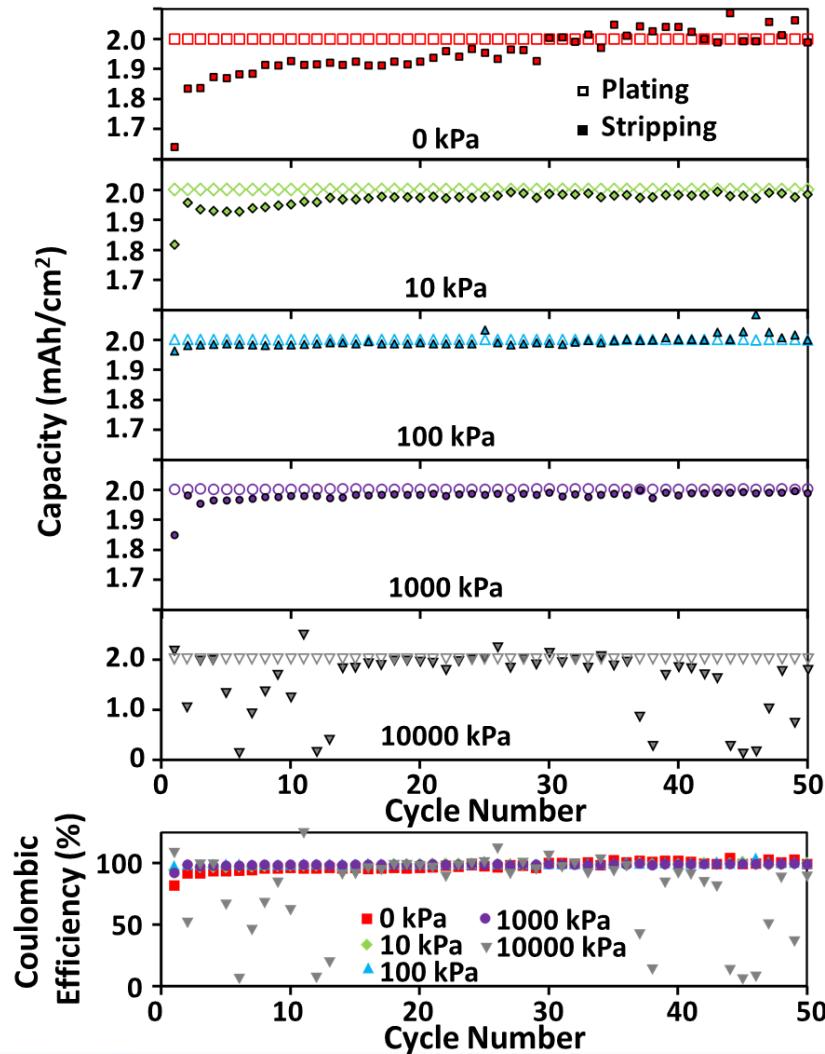
# Controlled Pressure on Electrodeposited Li



# Effects of Pressure on Li Metal: Low Current

Low Current:  $0.5 \text{ mA/cm}^2$

- Cycling stability generally increases with increasing pressure until 10,000 kPa
- 10,000 kPa is too high and causes increased overpotential and loss of cycling stability
- Transport might be limited locally at high pressure where pores can close
- CE generally improves with pressure but 100 and 1,000 kPa are similar

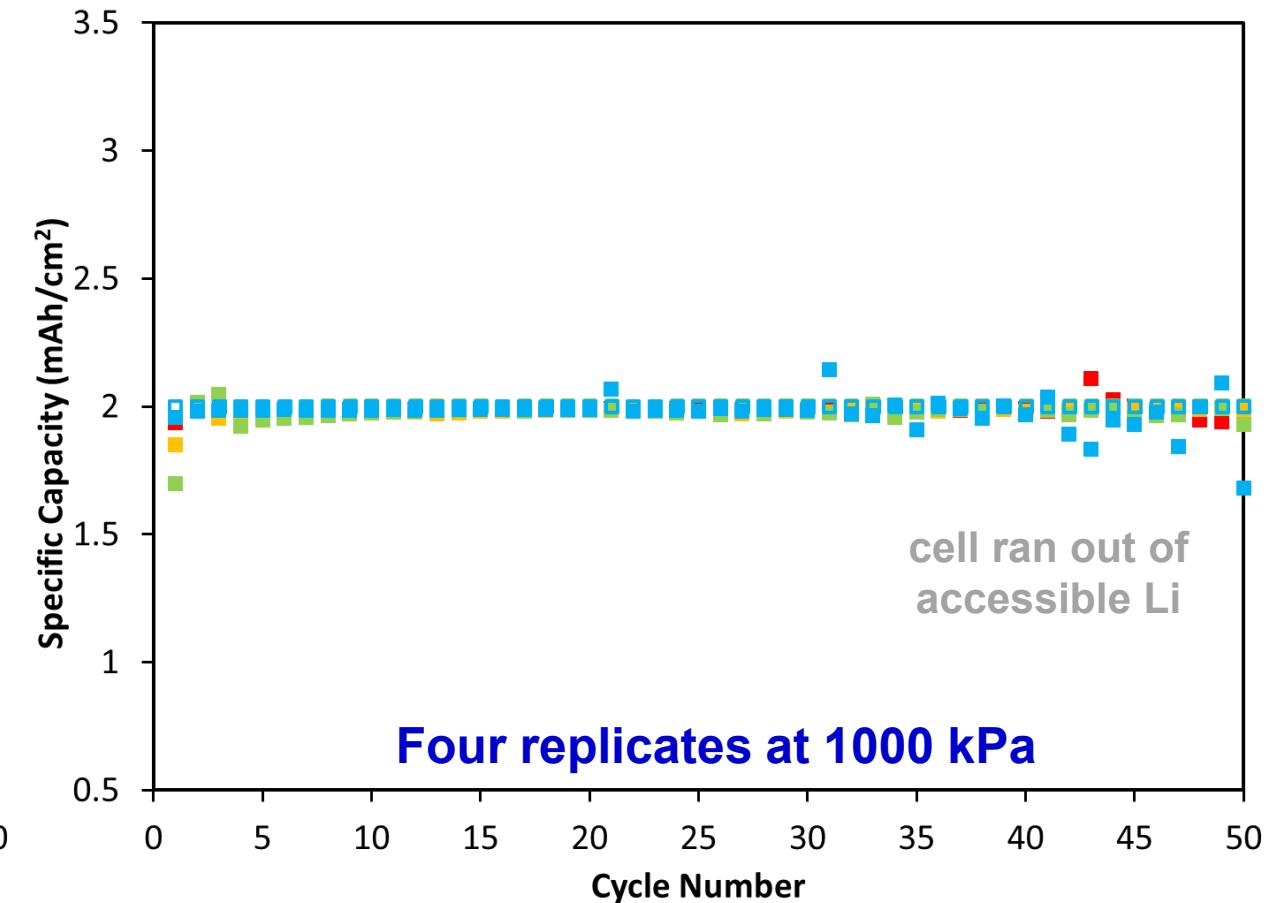
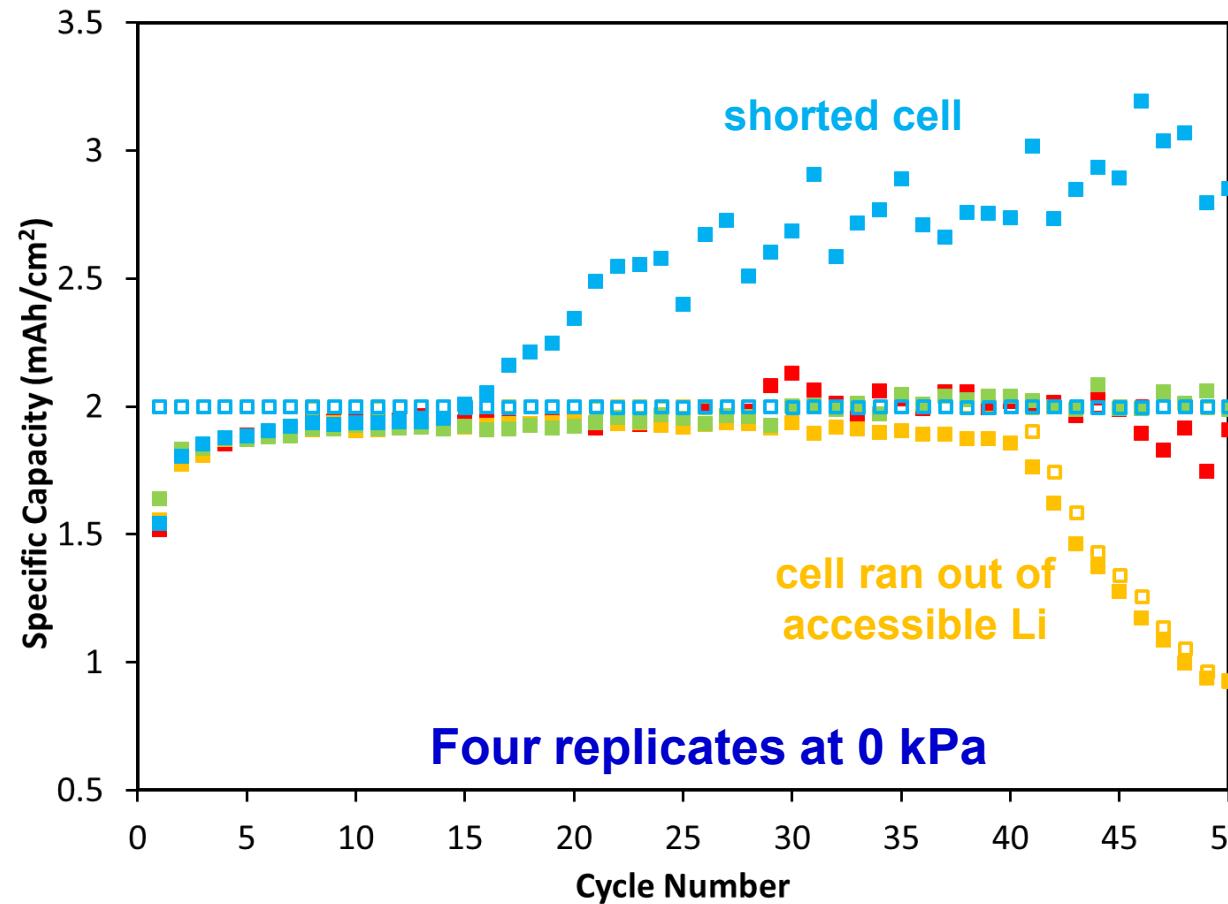


Harrison et al., ACS Appl. Mater. Interfaces 13, 31668 (2021).  
DOI: 10.1021/acsami.1c06488



# Effects of Pressure on Li Metal: Low Current

Low Current:  $0.5 \text{ mA/cm}^2$



Harrison et al., *ACS Appl. Mater. Interfaces* 13, 31668 (2021).  
DOI: 10.1021/acsami.1c06488

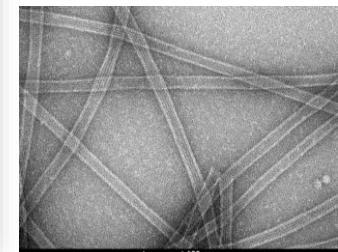
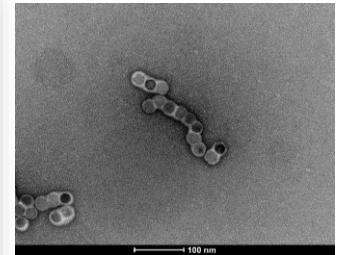


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# Characterization: Ex-Situ Cryogenic SEM

- Plunge-freeze or slowly freeze coin cell battery electrodes
- Inert transfer from glovebox into cryo SEM/FIB
- Cross-sectioning in cryo SEM/FIB to observe electrodeposited Li metal

Dr. John Watt  
watt@lanl.gov  
cint.lanl.gov or  
nsrcportal.sandia.gov



**Vitrobot**  
Captures native, solvated state

**Scios FIB/SEM**  
Analysis of surfaces & buried interfaces; 3D tomography

**Leica Cryo SEM Stage**  
Includes cryo-FIB milling, lift out, and transfer to the TEM

**Talos L120C CryoTEM**  
Dedicated low dose, low keV TEM for imaging of beam sensitive materials



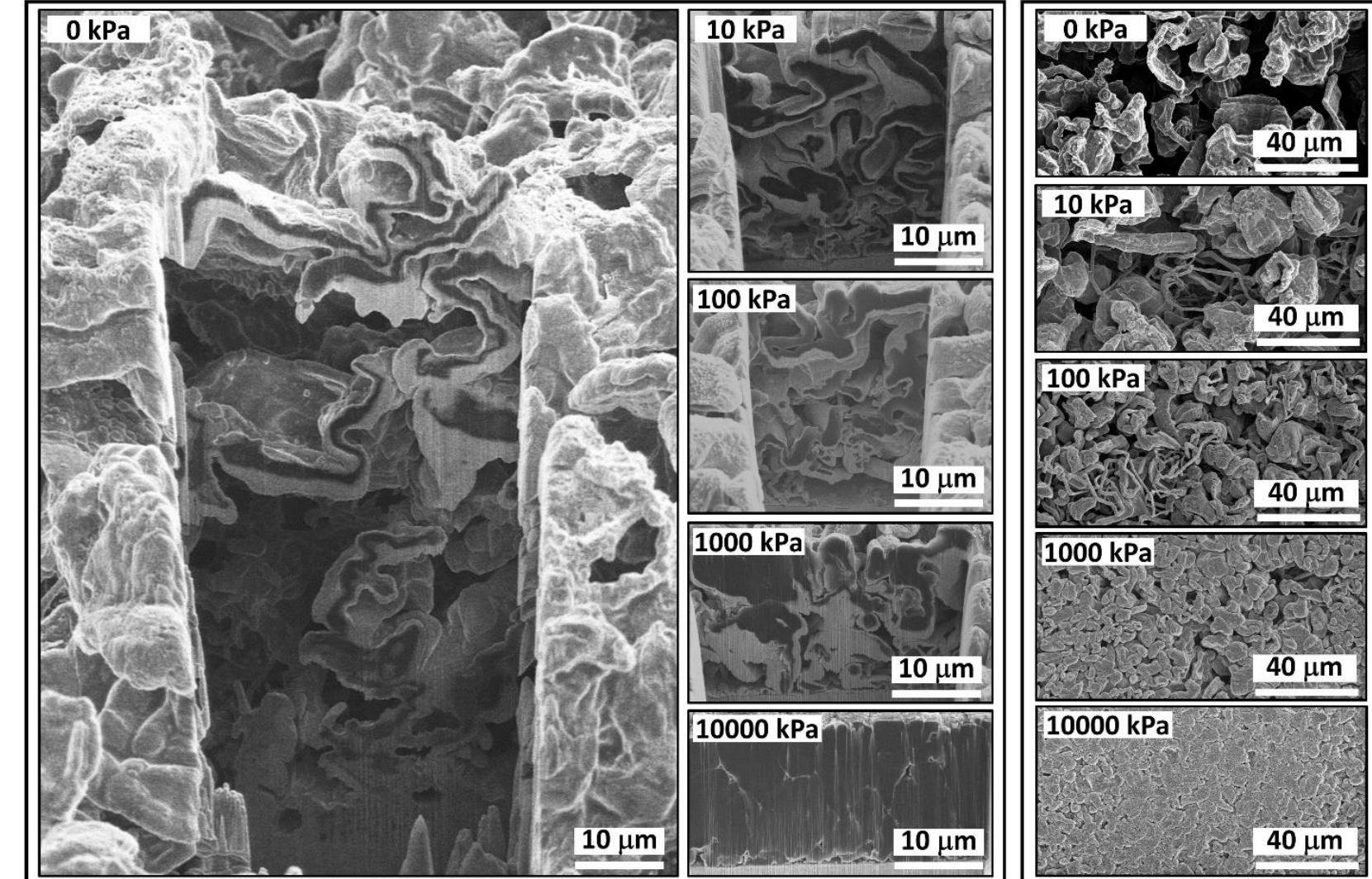
# Pressure at Low Current: 1<sup>st</sup> Li Deposition Step

Low Current: 0.5 mA/cm<sup>2</sup>

- Morphology improves drastically with pressure (even for 10000 kPa)

Pressure (kPa)	Average CE (%) First Cycle
0	82.3 ± 6.2
10	90.5 ± 4.1
100	97.5 ± 0.6
1000	93.6 ± 5.3
10000	106.2 ± 1.6

Pressure (kPa)	Thickness 51 <sup>st</sup> Plating (μm)
0	91
10	33
100	30
1000	22
10000	17

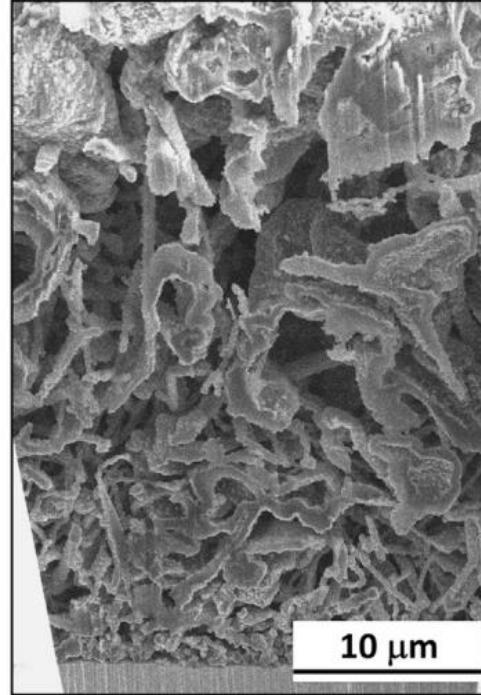


Harrison et al., *ACS Appl. Mater. Interfaces* 13, 31668 (2021).  
 DOI: 10.1021/acsami.1c06488

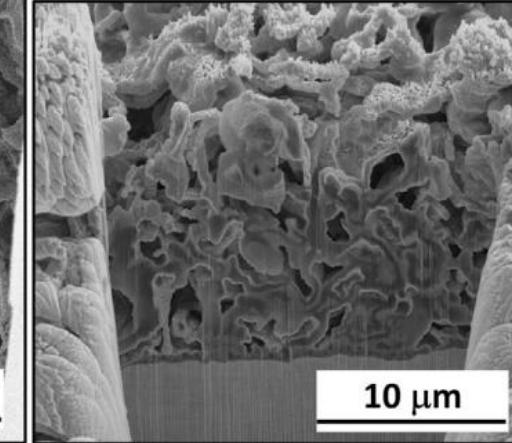


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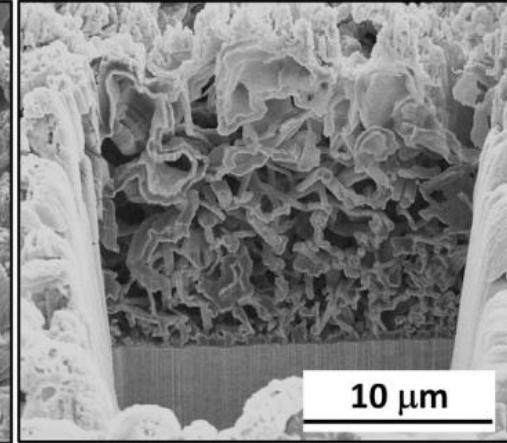
# Pressure at High Current: 1<sup>st</sup> Li Deposition Step



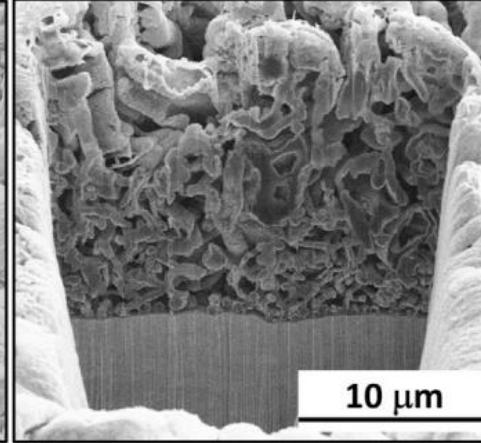
0 MPa



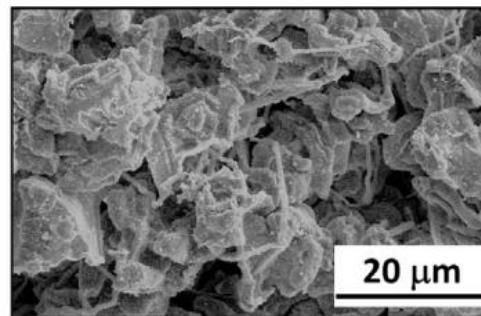
0.01 MPa



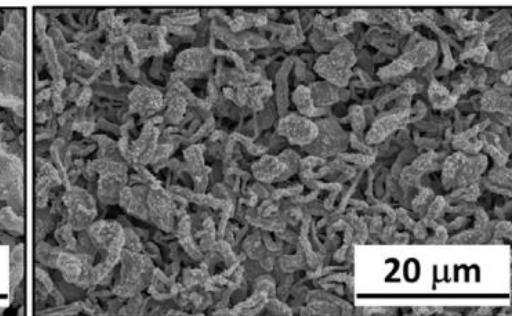
0.1 MPa



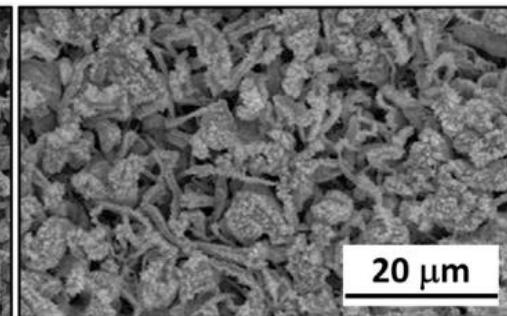
1 MPa



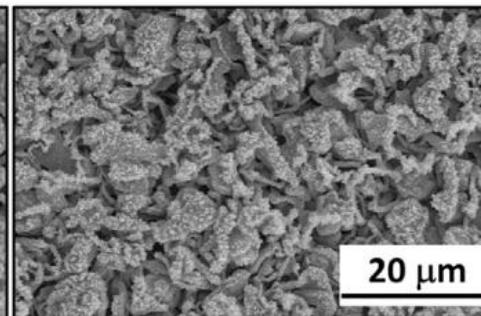
20 μm



20 μm



20 μm

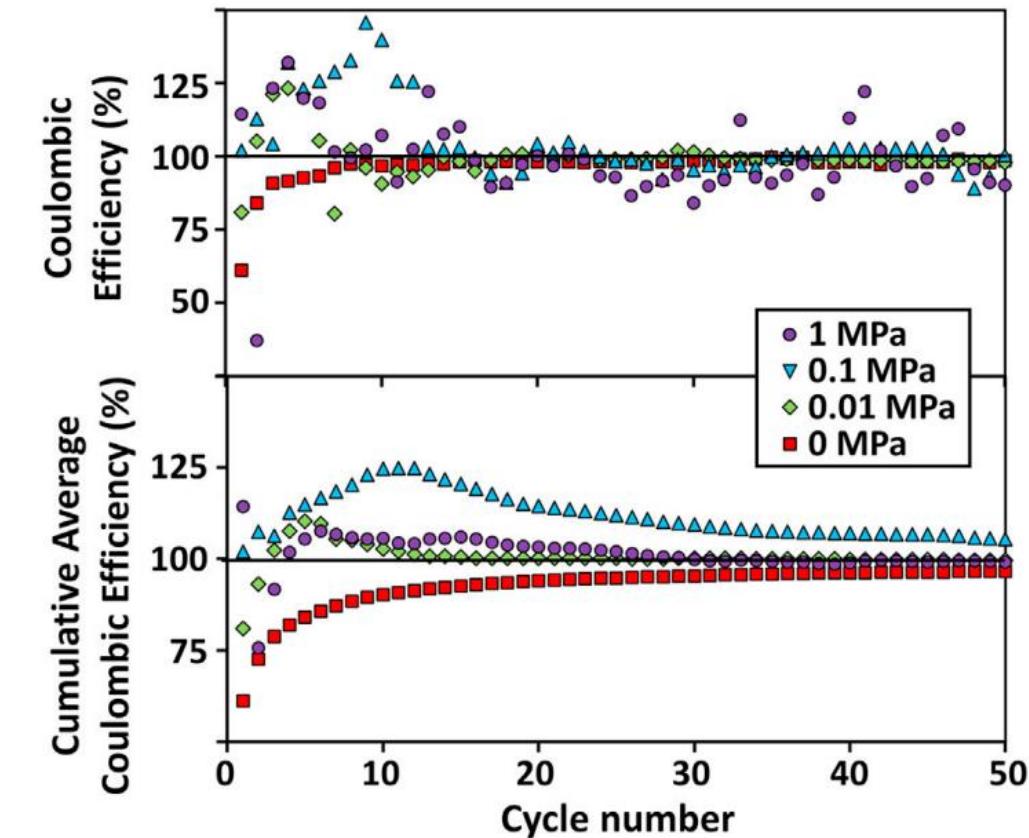
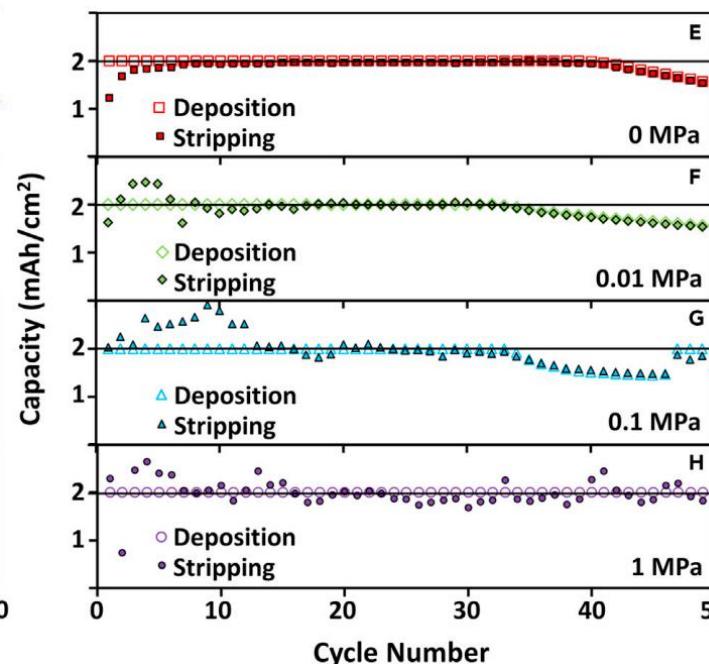
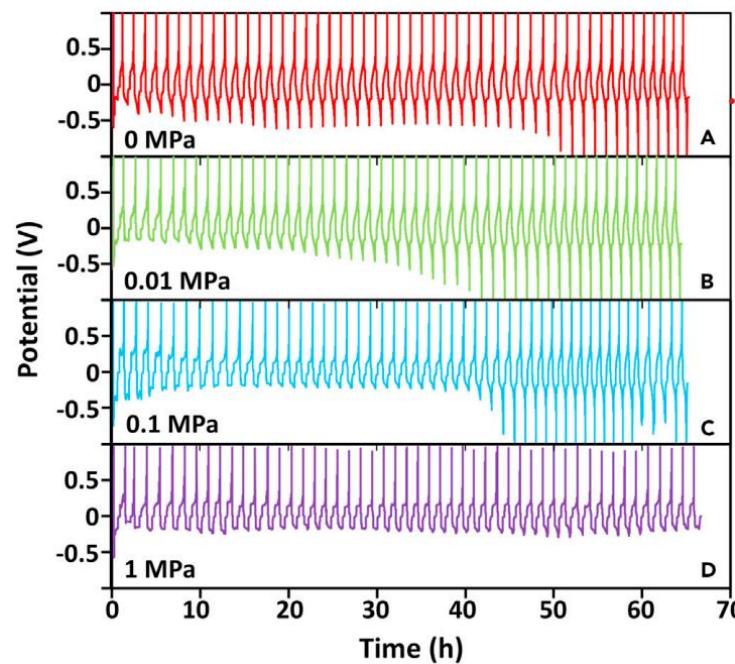


20 μm



# Effects of Pressure on Li Metal: High Current

High Current:  $4 \text{ mA/cm}^2$

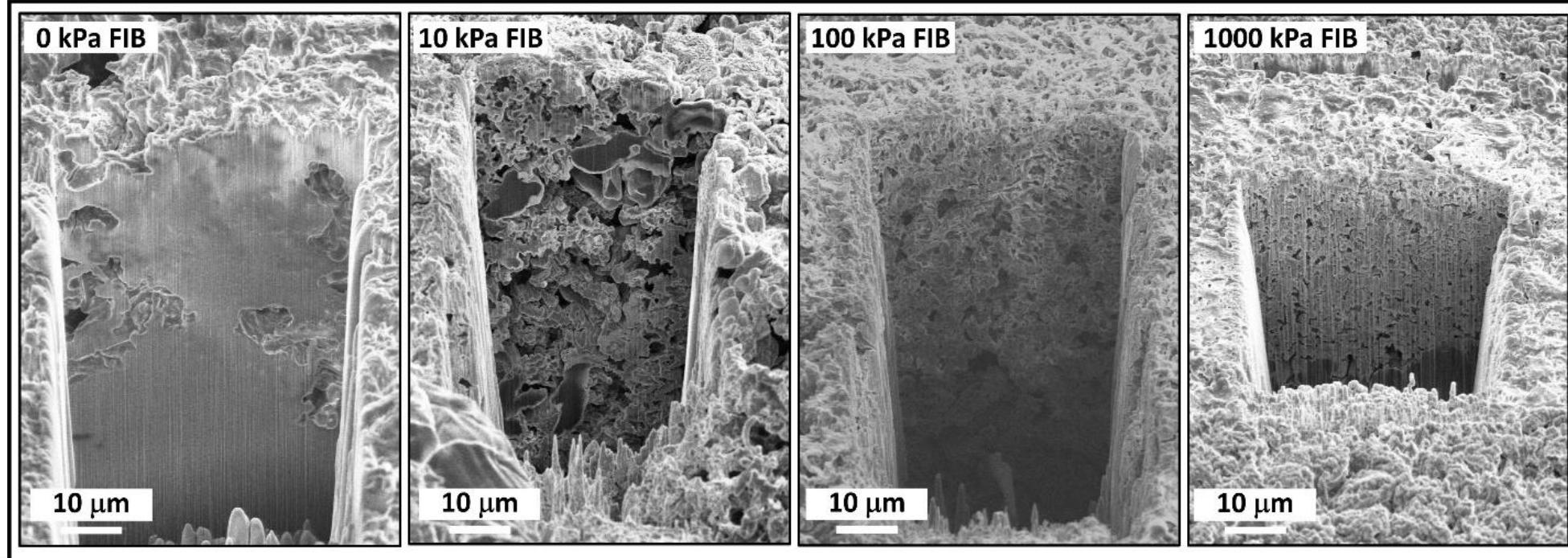
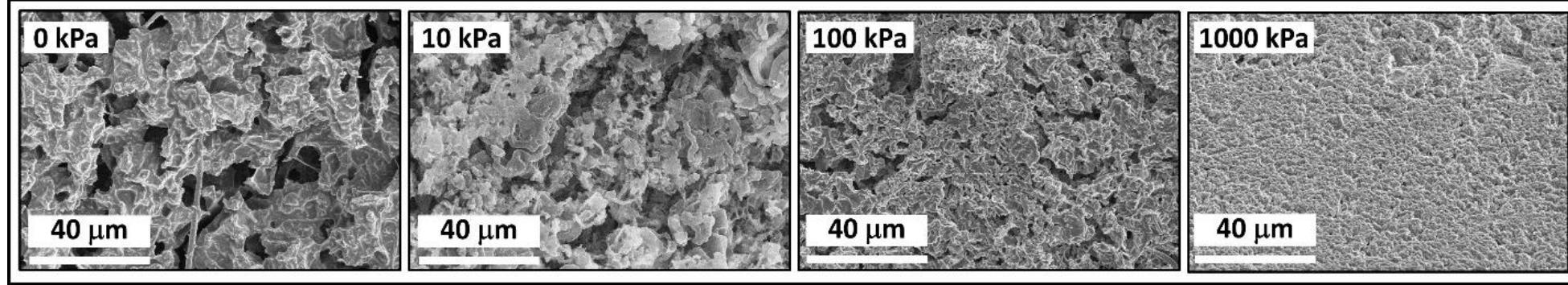


Harrison et al., *iScience* 24, 103394 (2021).  
DOI: [10.1016/j.isci.2021.103394](https://doi.org/10.1016/j.isci.2021.103394)

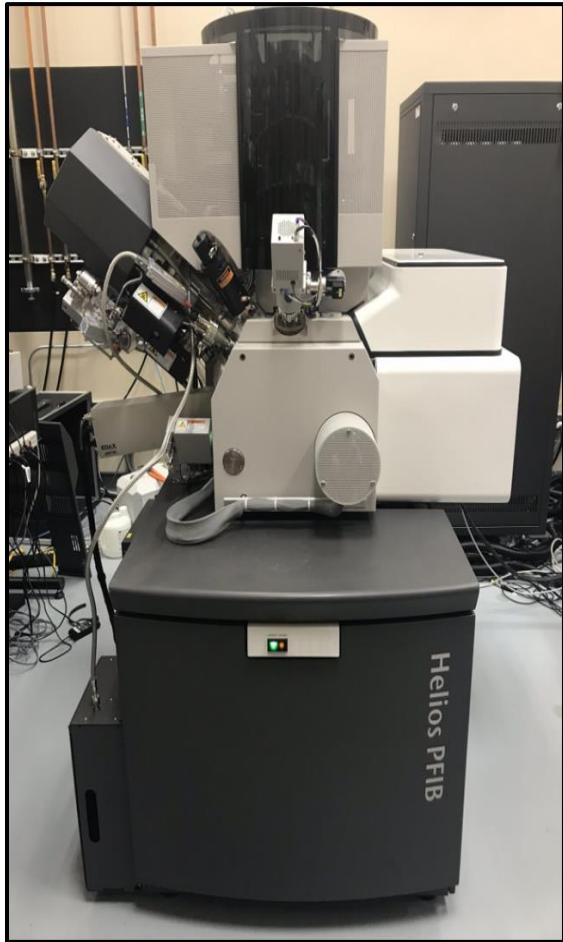


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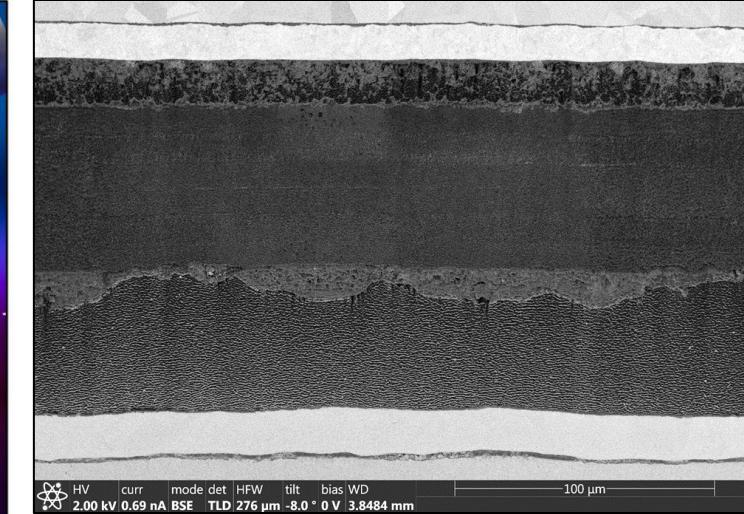
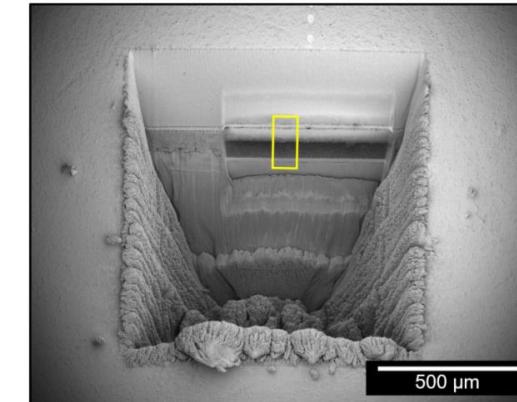
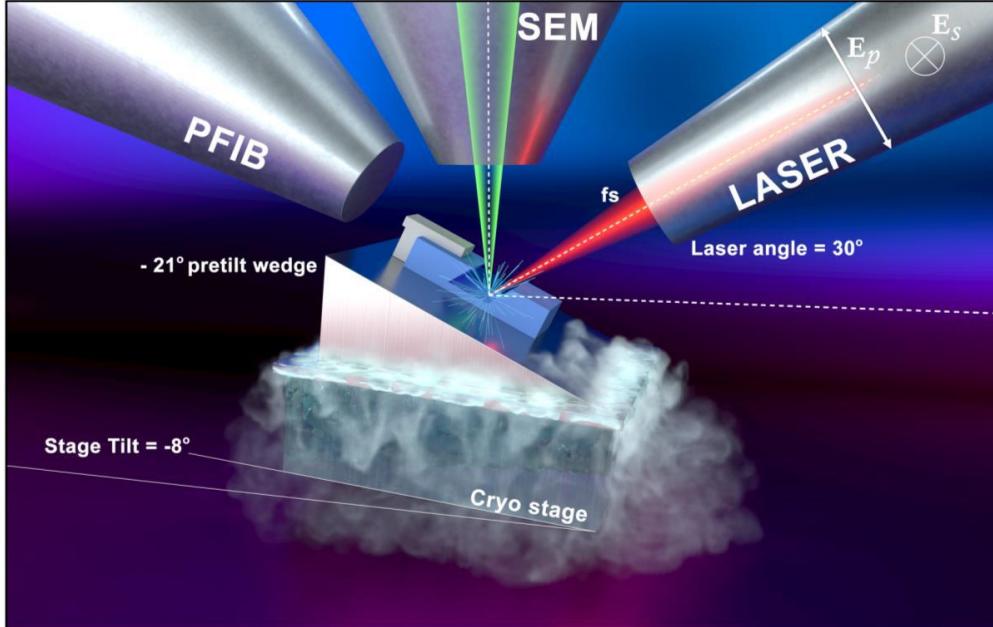
# Pressure at Low Current: 51<sup>st</sup> Li Deposition Step



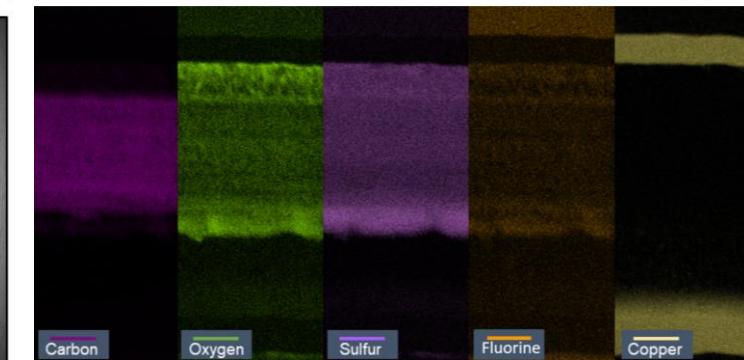
# Cross-sectioning without Battery Disassembly



**Helios Laser Plasma FIB**  
fs laser mills 15,000x faster than Ga-ion FIB



- Two Celgard 2325 Separators
- Li/SEI/electrolyte layers found



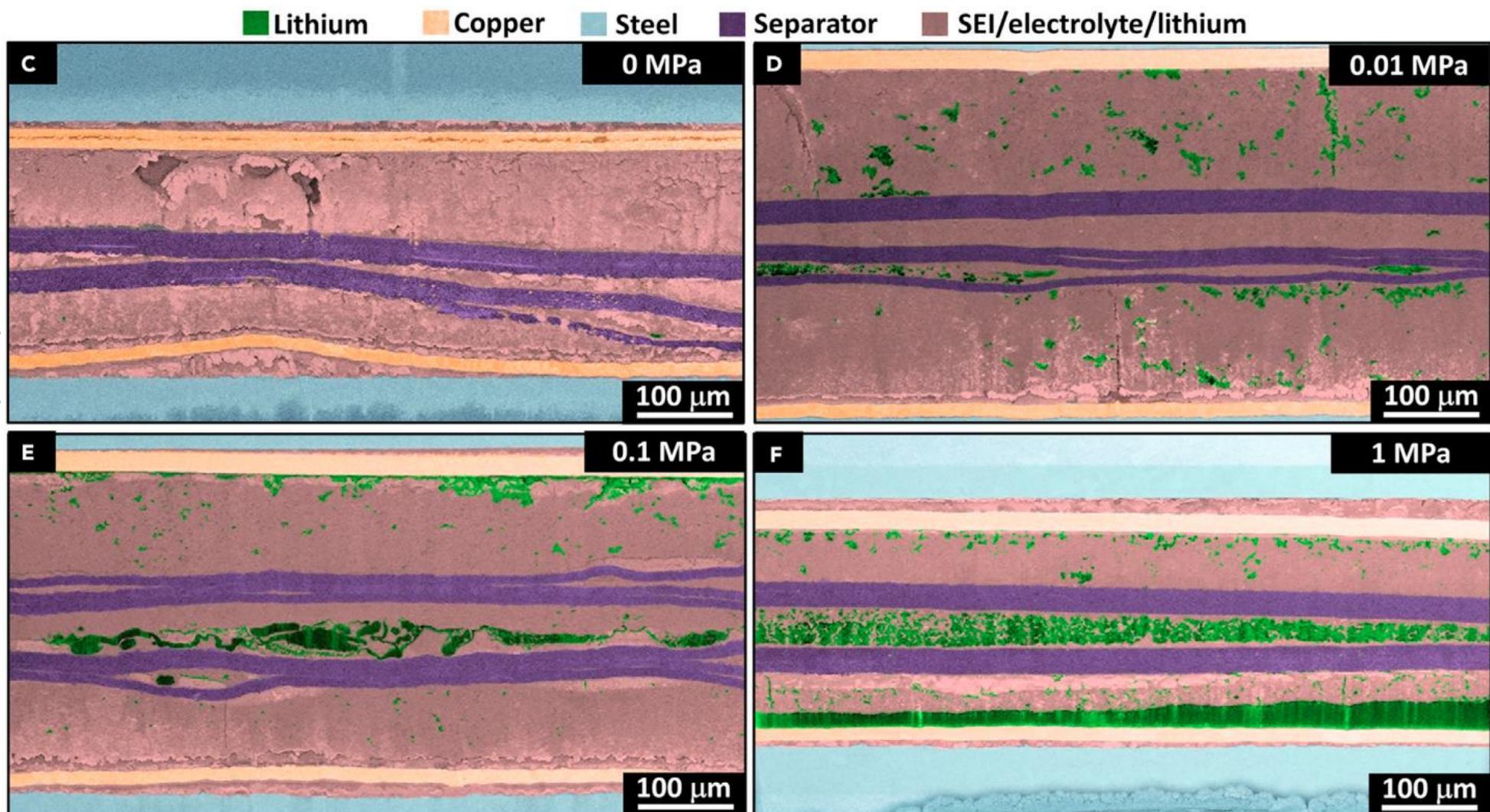
Jungjohann et al., *ACS Energy Lett.* 6, 2138 (2021).  
DOI: 10.1021/acsenergylett.1c00509



# Pressure at High Current: 51<sup>st</sup> Li Deposition Step

High Current: 4 mA/cm<sup>2</sup>

Pressure (MPa)	Thickness 51 <sup>st</sup> Plating (μm)
0	189
0.01	318
0.1	274
1.0	188



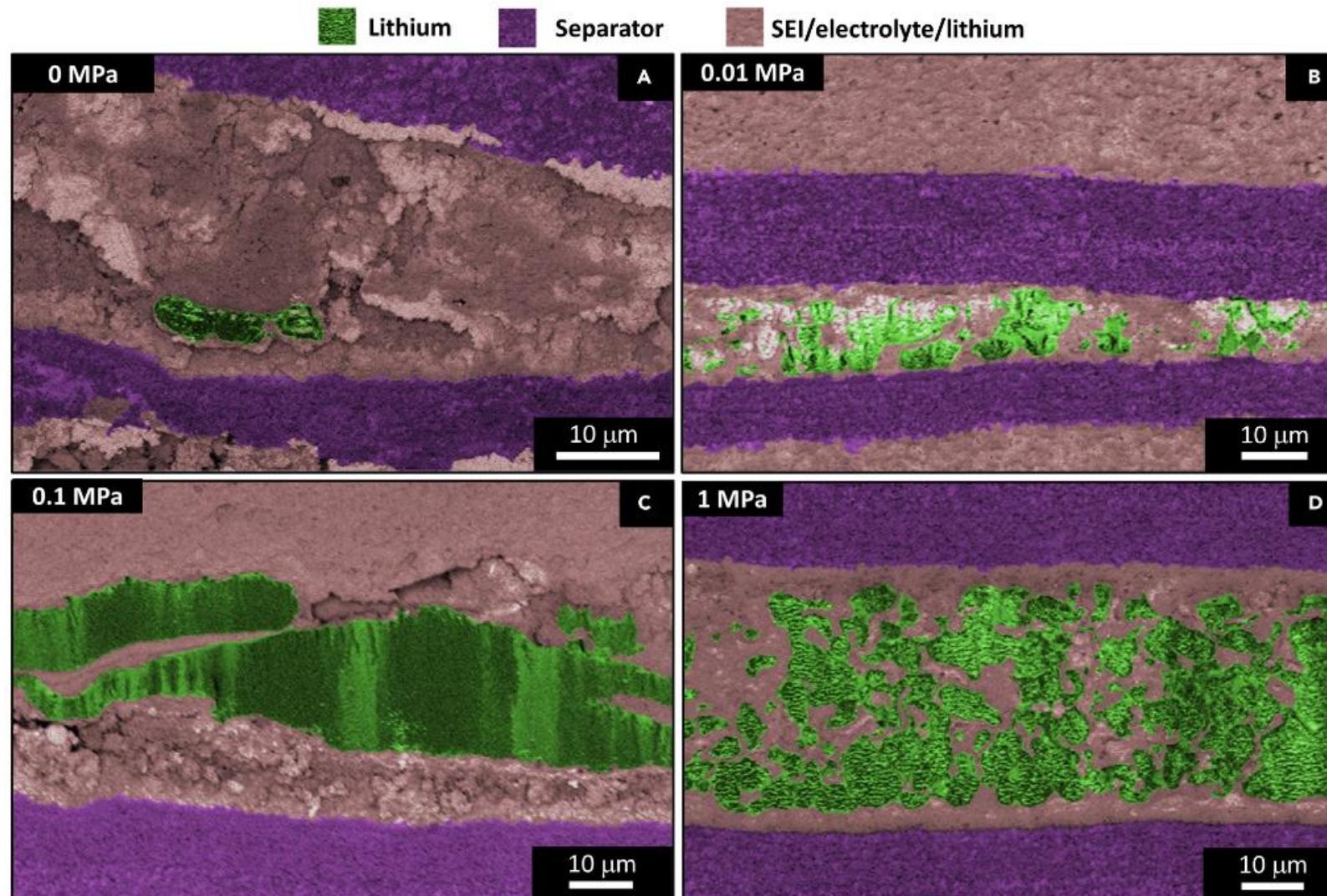
Harrison et al., *iScience* 24, 103394 (2021).  
DOI: 10.1016/j.isci.2021.103394



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# All Pressures: Li Deposition within Separator

High Current:  $4 \text{ mA/cm}^2$

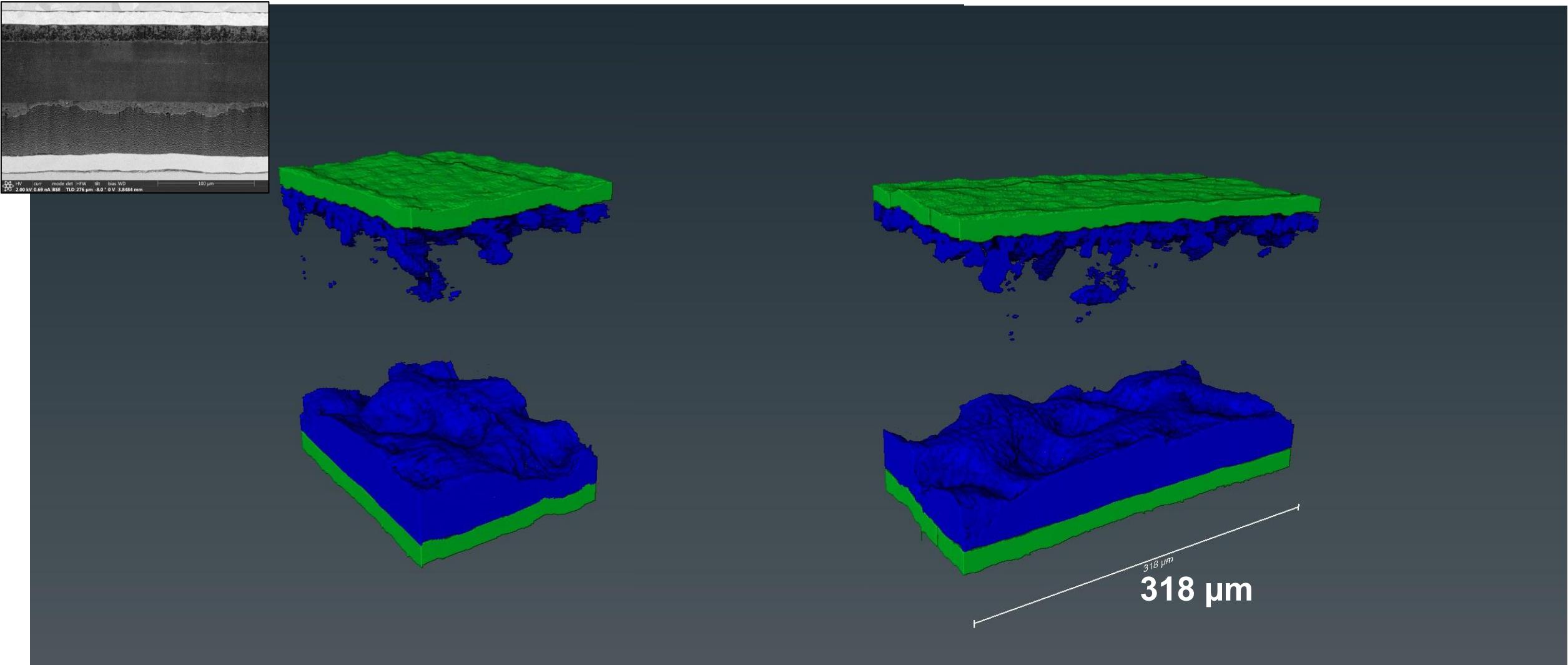


Harrison et al., *iScience* 24, 103394 (2021).  
DOI: [10.1016/j.isci.2021.103394](https://doi.org/10.1016/j.isci.2021.103394)



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# fs Laser Slice-N-View of Battery Stack



Dr. Khalid Hattar

khattar@sandia.gov



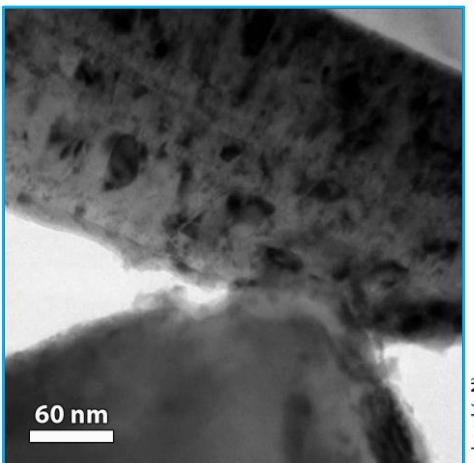
# PI-95 Picoindenter in TEM



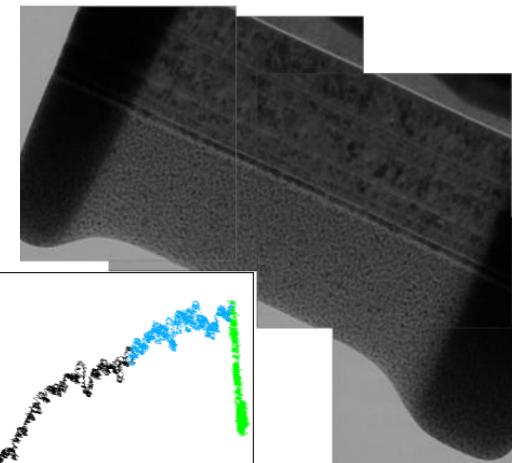
Notched Bar



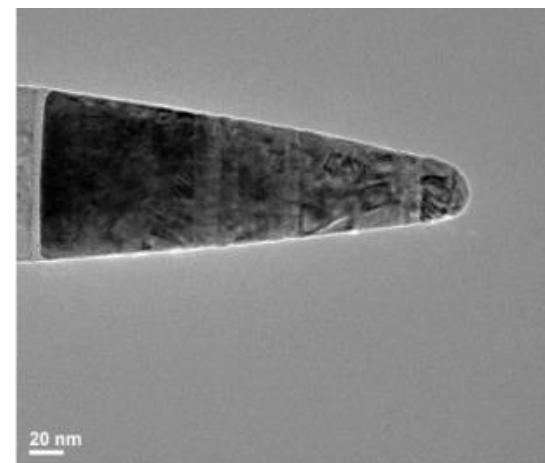
Nanoindentation



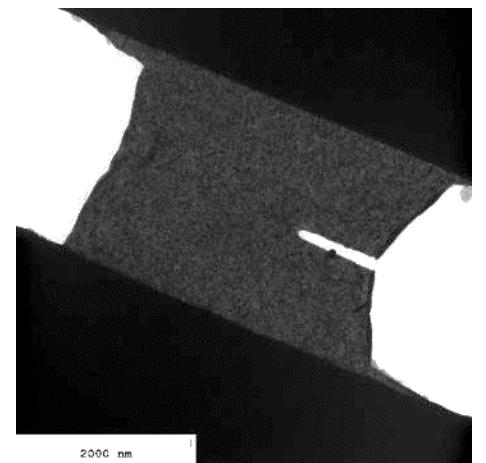
I Beams



Nanopillars



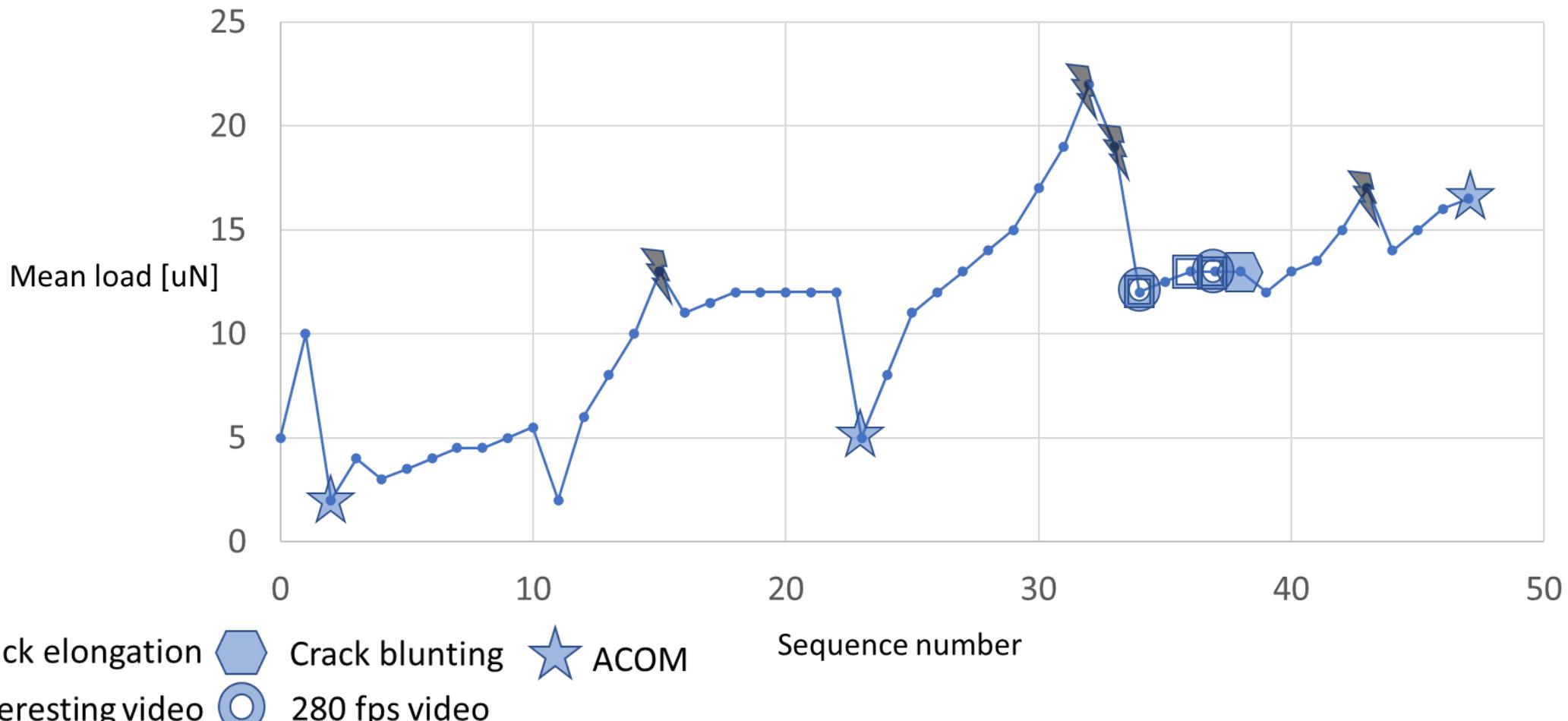
Notched Micro Tension Bars



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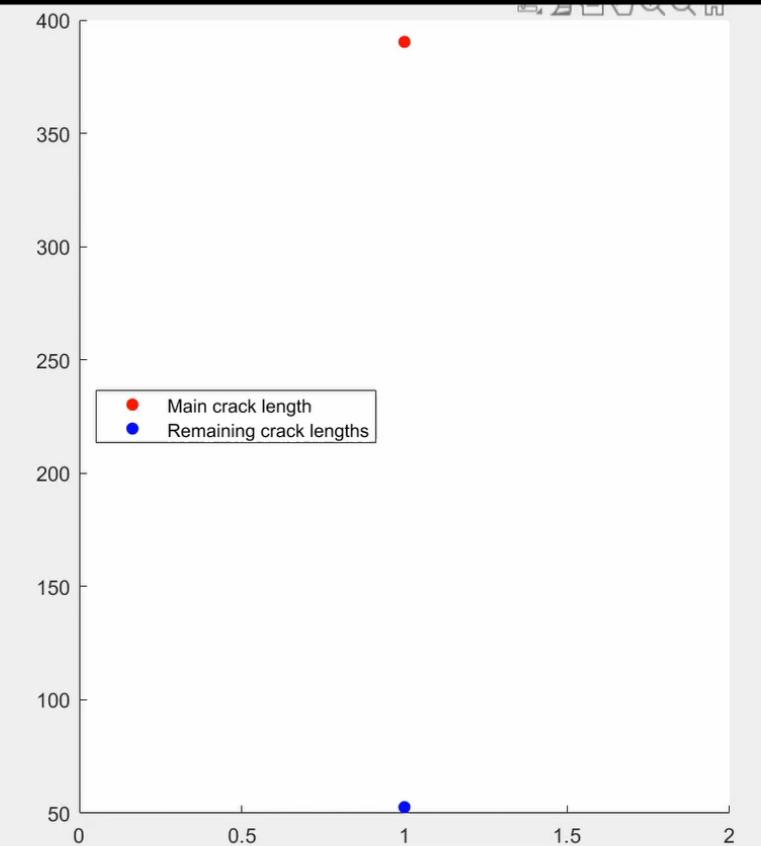
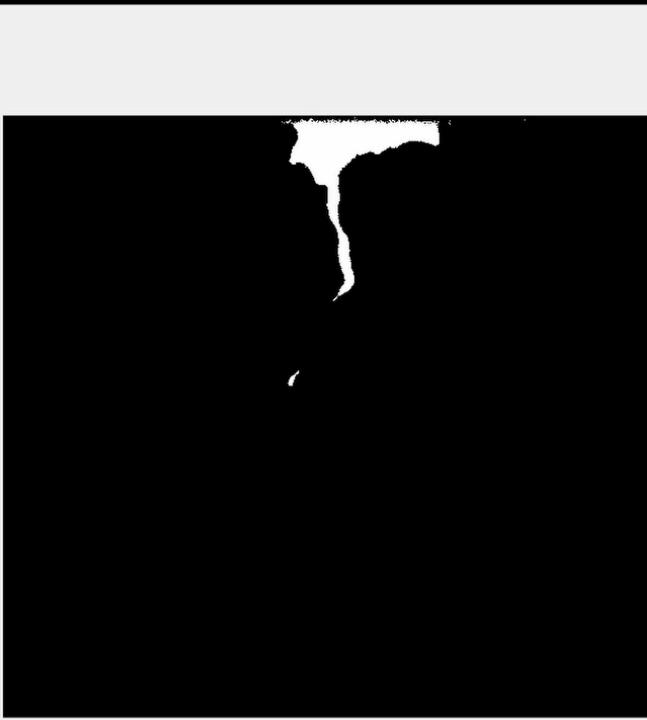
# A6-2 Experiment

Mean load vs. test number (R=0.4 for all tests)



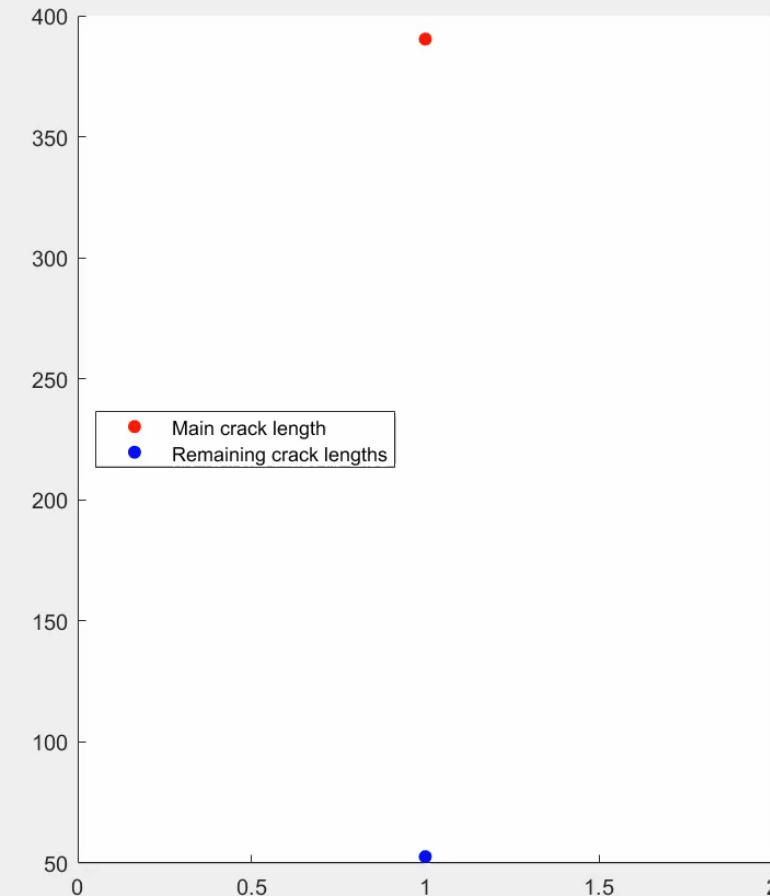
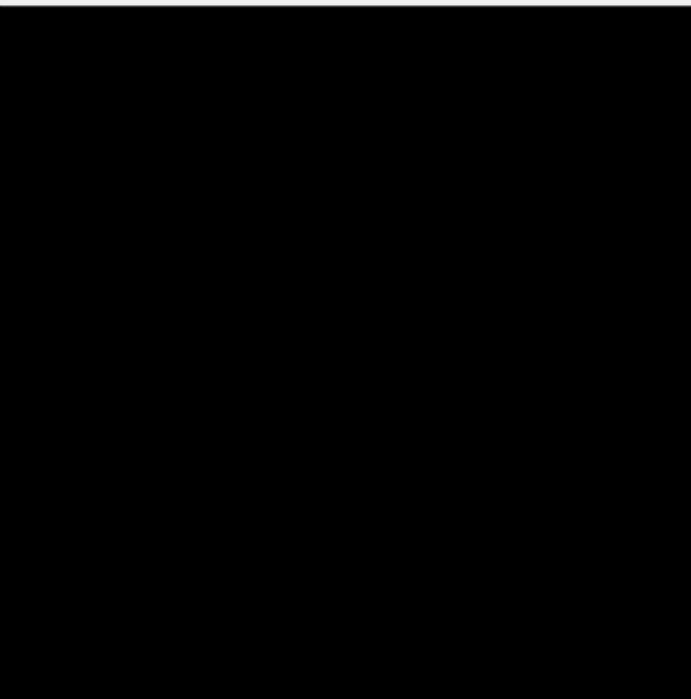
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# A6-2 Experiment



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# A6-2 Experiment



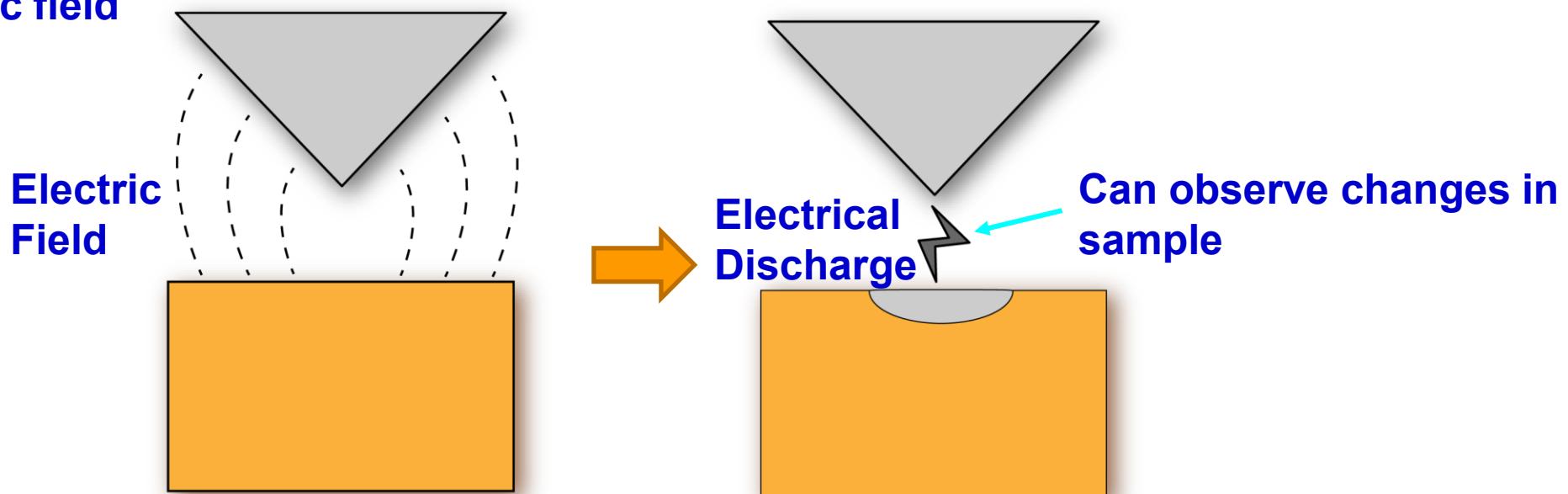
TMS 2022 150<sup>th</sup> ANNUAL MEETING & EXHIBITION  
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# Future: Electrical-Mechanical TEM

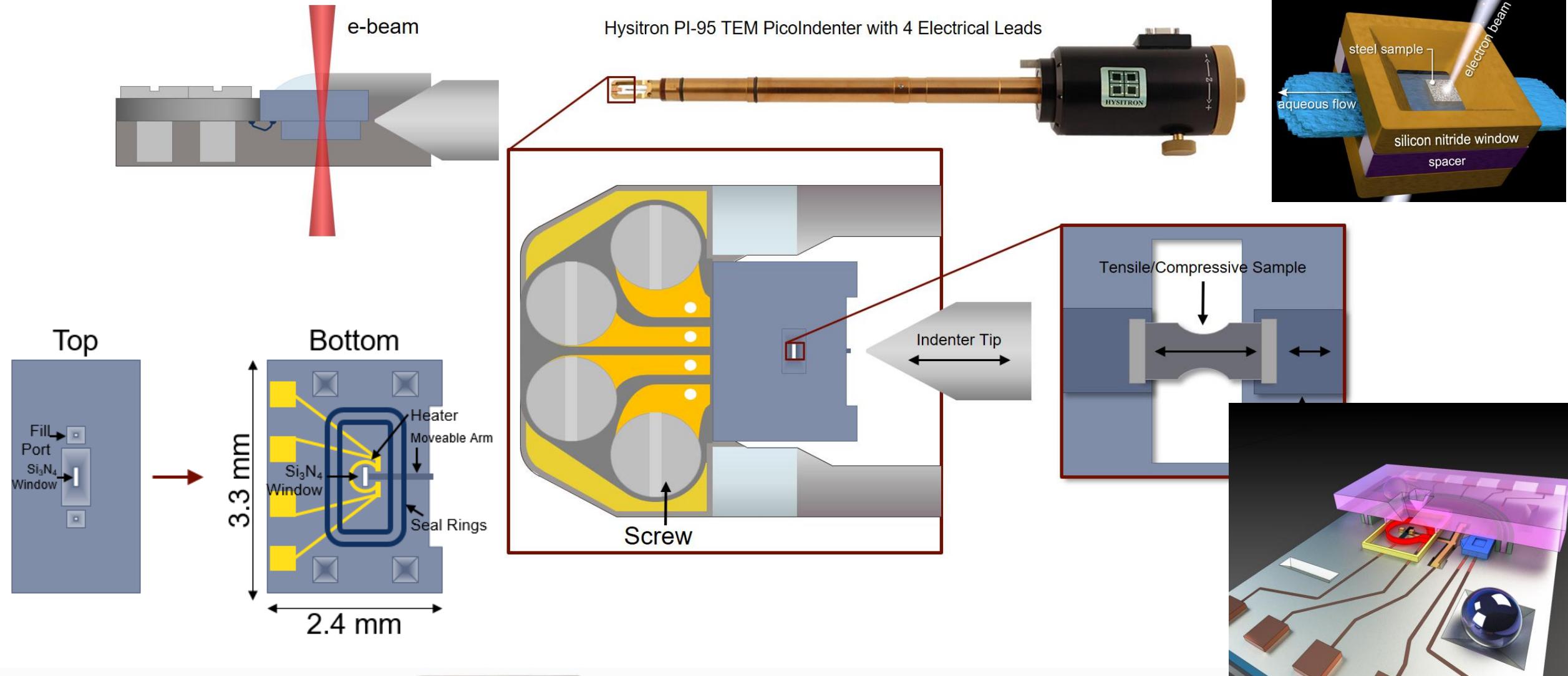
- Electromigration
- Electrical discharge
- Can combine with laser stimulus

Nanoindentation tip from PI-95  
that applies the electric field

Sample can be at  
high temperature



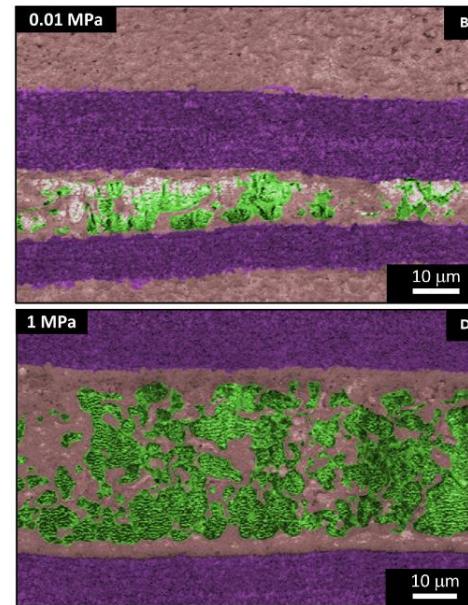
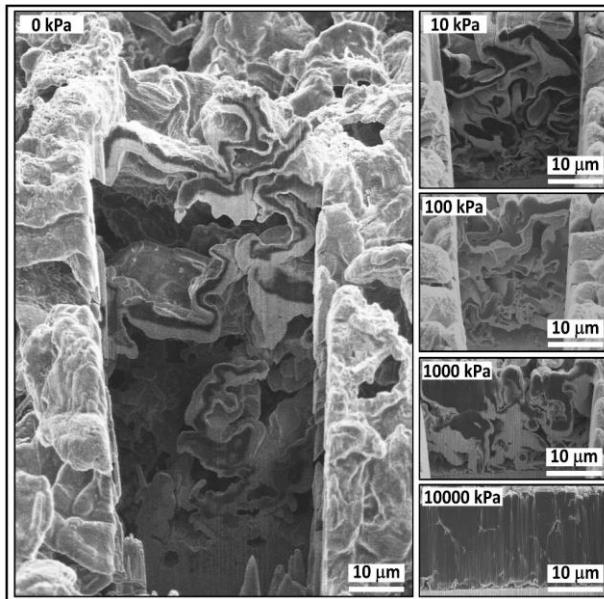
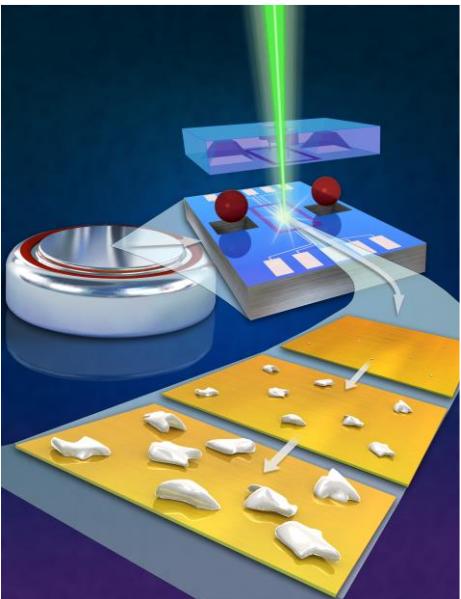
# Future: Electrochemical-Mechanical TEM



# Thank You! Questions?



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