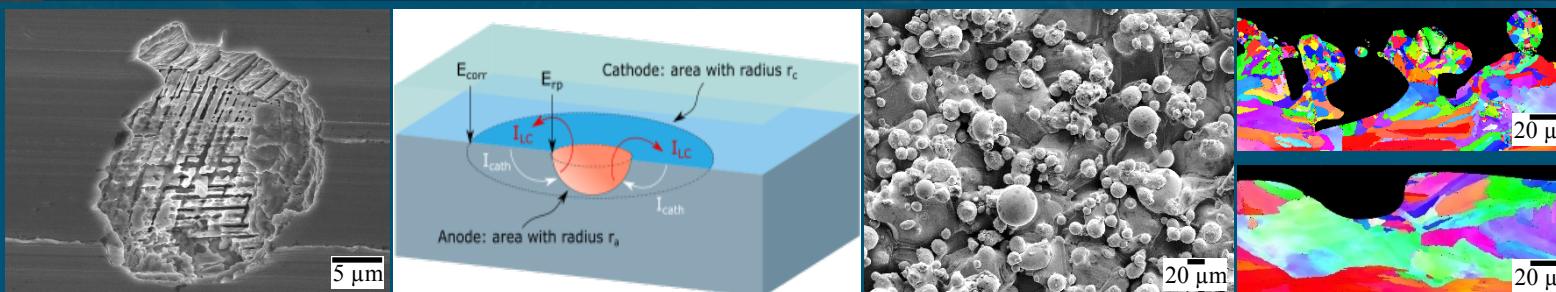
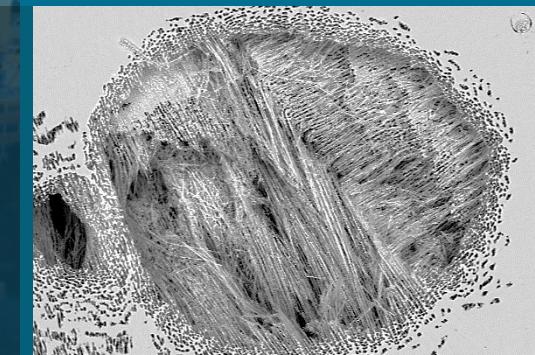


# Atmospheric Corrosion Of Laser Powder Bed Fusion 316L Stainless Steel



Sandia  
National  
Laboratories



AMPP 2022 Annual  
Conference + Expo  
March 6<sup>th</sup> to 10<sup>th</sup>, 2022 -  
San Antonio, TX



PRESENTED BY

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**UVA:** Rebecca S. Marshall, Danyil Kovalov, Ryan Katona, Ji Ma, Robert Kelly

**LANL:** Courtney Clark, Tim Gorey, Jamie Stull, Colt Montgomery, Daniel Hooks

**KCNSC:** Lucas Rice, Jonathan Dwyer, Andy Deal, Zach Rueger, Francisco Garcia-Moreno

**SRNL:** Patrick Kuzbary, Savidra Lucatero, Prabhu Ganesan, Hector Colon-Mercado, Paul Korinko

**LLNL:** Margaret Wu, Seongkoo Cho, Monika M. Biener, Y. Morris Wang, Justin Jones, and S. Roger Qiu, ...

***Thanks to the symposium organizers for this opportunity.***

Publications on topic:

E.J. Schindelholz, et al., Corrosion of Additively Manufactured Stainless Steels—Process, Structure, Performance: A Review, *Corrosion*, 77 (2021).

M.A. Melia, et al., Marine Atmospheric Corrosion of Additively Manufactured Stainless Steels, *Corrosion*, (2021).

M.A. Melia, et al., How build angle and post-processing impact roughness and corrosion of additively manufactured 316L stainless steel, *npj Materials Degradation*, 4 (2020).

M.A. Melia, et al., Corrosion properties of 304L stainless steel made by directed energy deposition additive manufacturing, *Corros. Sci.*, (2019).

R.F. Schaller, et al., The Role of Microstructure and Surface Finish on the Corrosion of Selective Laser Melted 304L, *J. Electrochem. Soc.*, 165 (2018).

R.F. Schaller, et al., Corrosion Properties of Powder Bed Fusion Additively Manufactured 17-4 PH Stainless Steel, *CORROSION*, 73 (2017).

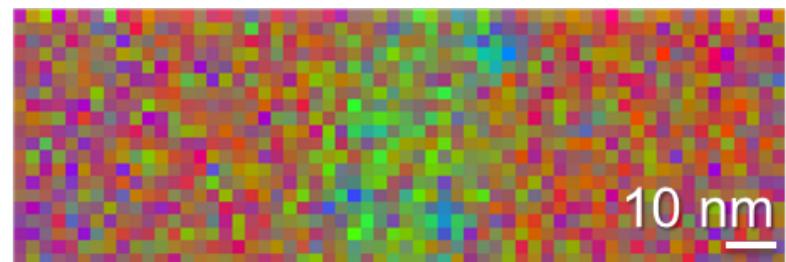
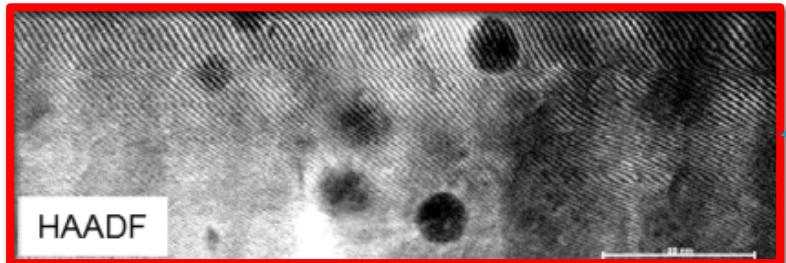


# Background - Powder metal additively manufactured (AM) microstructures

## Chemical heterogeneities

### Cell walls decorated with oxides

Matrix      Cell Boundary      Matrix

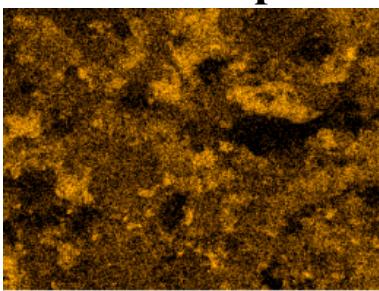


- Increased Fe (+ 9%)
- Increased Si
- Increased Cr (+ 4%)

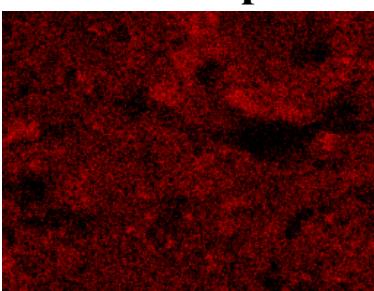
Adapted from Schaller, 2018

### Recast layer from a part removed with wire EDM

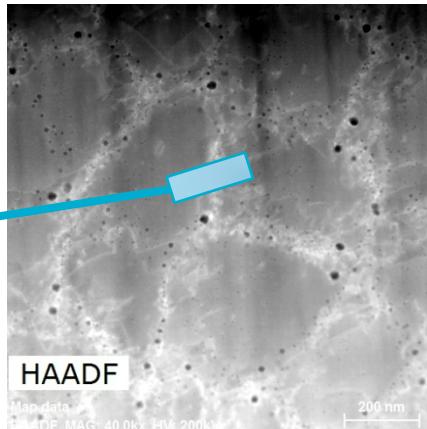
#### Cu map



#### Zn map

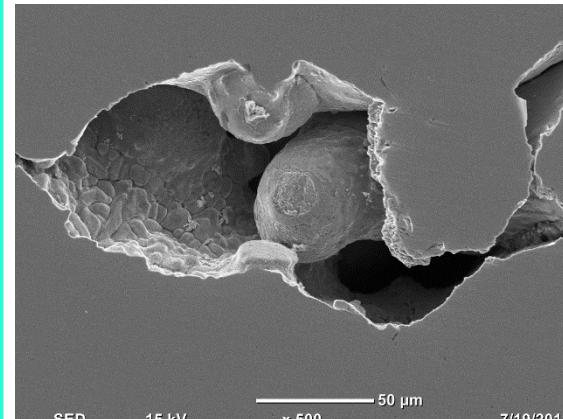


Cu and Zn penetrate 5 to 10 μm deep.

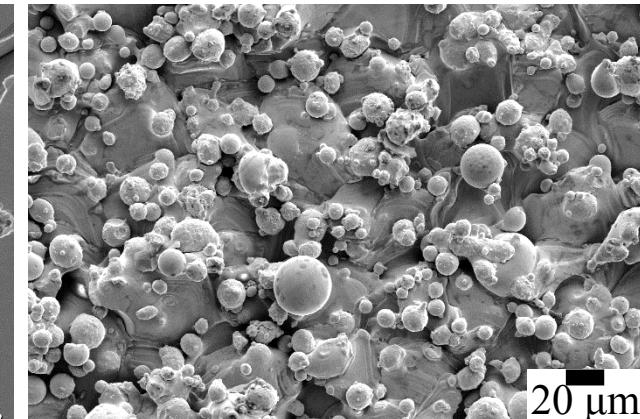


## Processing defects

### Porosity



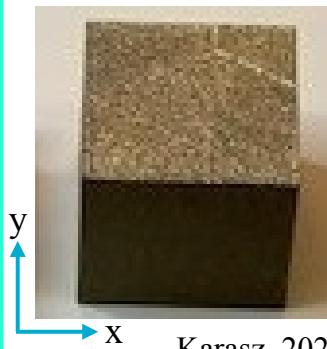
Melia, 2019



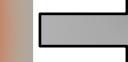
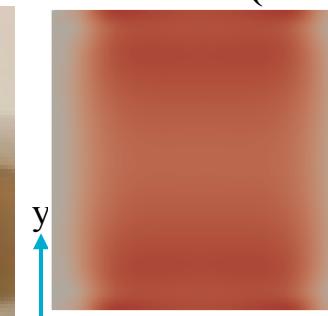
Melia, 2020

## Residual stress of 316L cubes

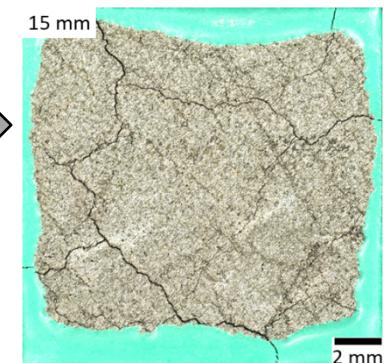
### Stress XX (MPa)



Karasz, 2021



200  
100  
0  
-100  
-200



(UUR)

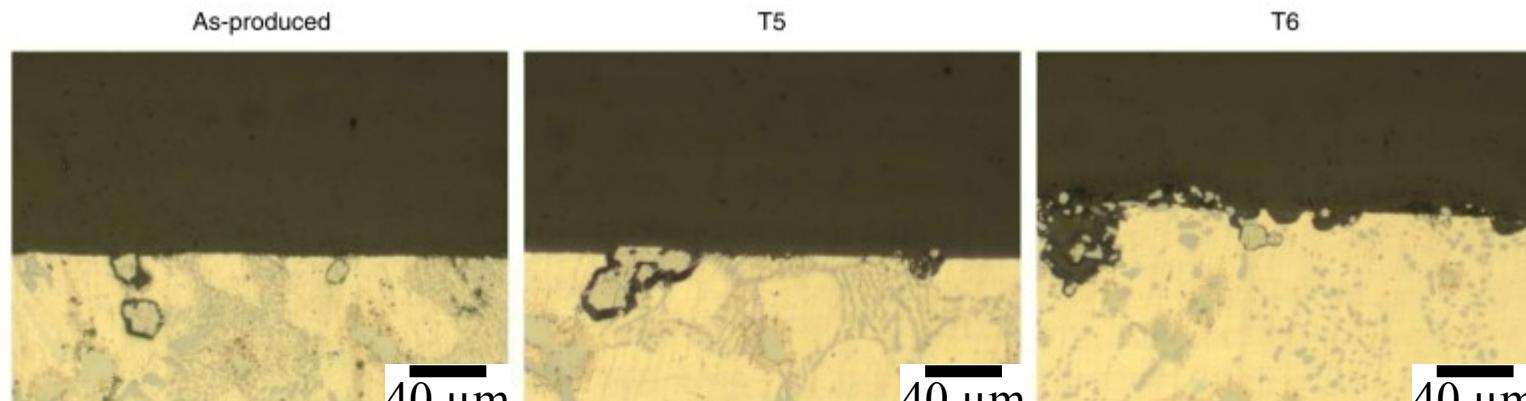
Leads to susceptibility to stress corrosion cracking

# Background – AM Al-10Si-Mg exposed to ASTM G85-A2

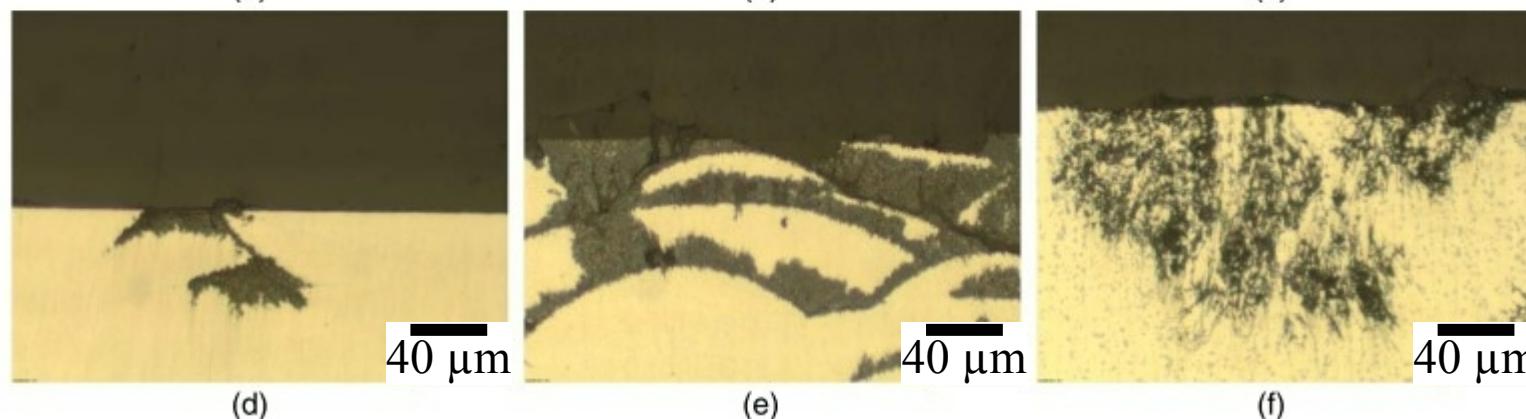


Cross-section images after 31 days of ASTM G85-A2 exposures

Cast Al-10Si-Mg

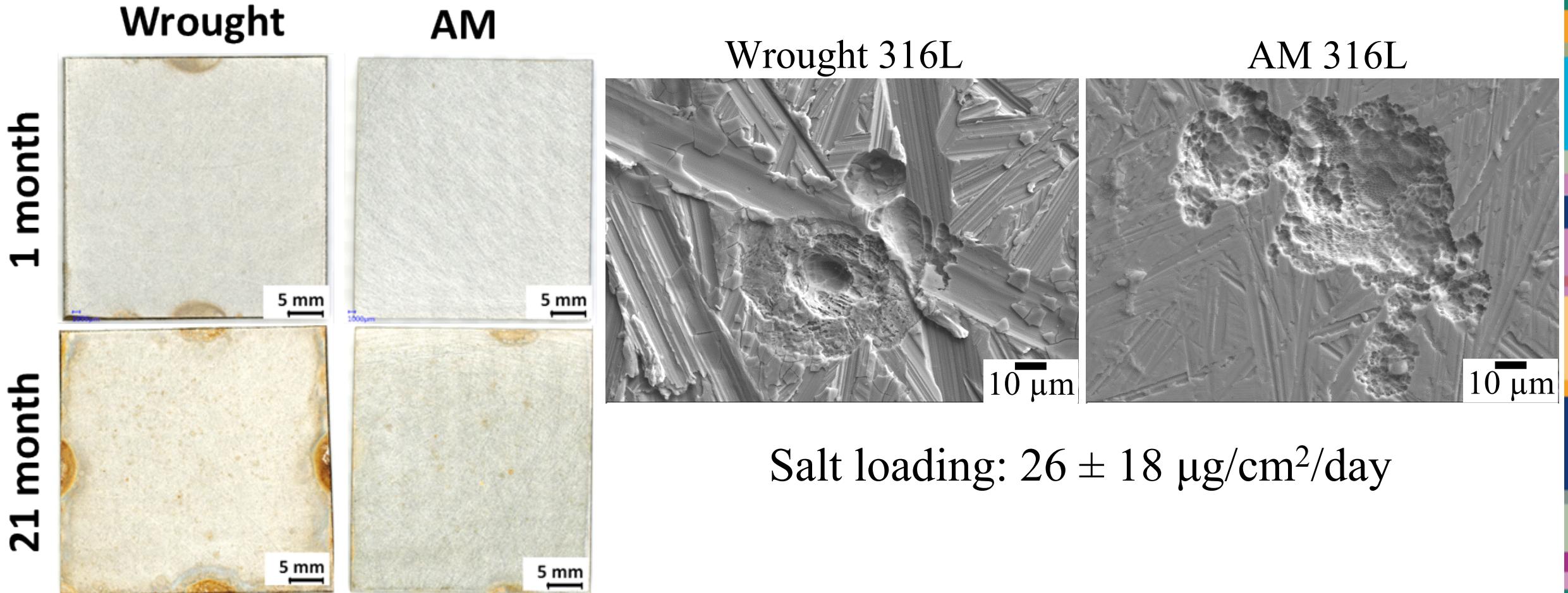


AM Al-10Si-Mg



- Corrosion damage morphology was dictated by AM microstructure. They hypothesize that melt pool boundaries act as barriers to corrosion propagation.
- Maximum corrosion depth was greater for AM compared to cast.

# Background – Coastal exposures of AM 304L/316L



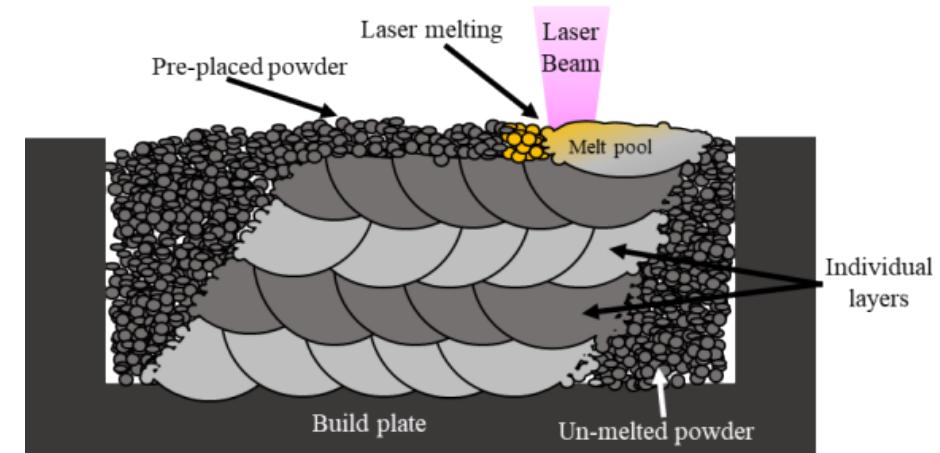
- The pit morphology for AM 316L showed significant preferential corrosion attack of the cellular microstructure, the cell boundaries were not as vulnerable.
- Pit depth of AM samples were smaller than wrought samples (often larger pit density).

# What will we go over in this talk?



- Constant humidity and contamination exposures of **ground surfaces** for wrought and powder bed fusion (PBF) 316L.
  - Pit morphologies.
  - Pit statistic analysis.
- Explore corrosion behavior of **as-printed** 316L parts when exposed to the accelerated corrosion test, ASTM G85-A2.

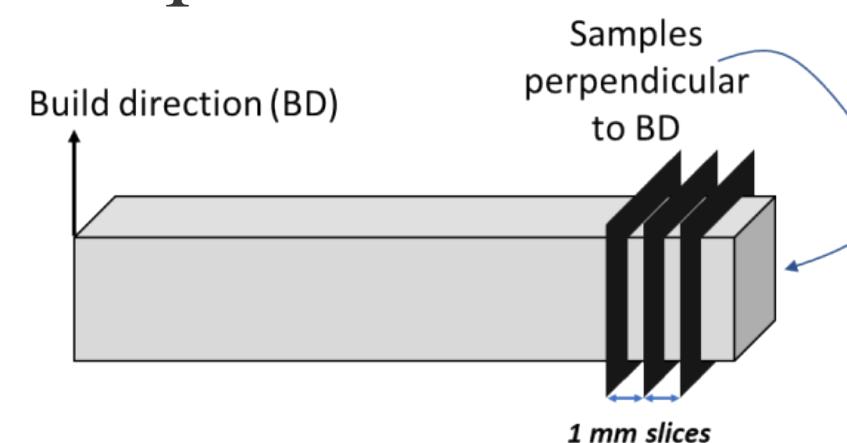
Samples were prepared using 316L powder with a powder bed fusion (PBF) technique.



# Constant humidity/contamination exposures



Alloy type	Rel. Hum. (%)		Salt loading ( $\mu\text{g}/\text{cm}^2$ )		Salt		Build	
	76	40	20	300	NaCl	ASW*	Parallel	Perpendicular
304AM	x		x		x			x
	x			x	x			x
	x			x		x		x
	x		x			x		x
		x		x		x		x
	x	x				x		x
	x		x			x	x	
316AM	x		x		x			x
	x			x	x			x
	x			x		x		x
	x		x			x		x
		x		x		x		x
	x	x				x		x
	x		x			x		x
304W	x		x		x			
	x			x	x			
	x			x		x		
	x		x			x		
		x		x		x		
	x	x				x		
316W	x		x		x			
	x			x	x			
	x			x		x		
	x		x			x		
		x		x		x		
	x	x				x		



ASW composition	
Composition	Concentration (g/L)
NaCl	24.53
MgCl <sub>2</sub> .6H <sub>2</sub> O	5.20
Na <sub>2</sub> SO <sub>4</sub>	4.09
CaCl <sub>2</sub>	1.16
KCl	0.695
NaHCO <sub>3</sub>	0.201
KBr	0.101
H <sub>3</sub> BO <sub>3</sub>	0.027
SrCl <sub>2</sub> .6H <sub>2</sub> O	0.025
NaF	0.003

\*Artificial Sea Water (ASW)

\*\*All samples polished to 600 grit prior to salt deposition.

\*\*\*Held at 35°C for all atmospheric exposures.

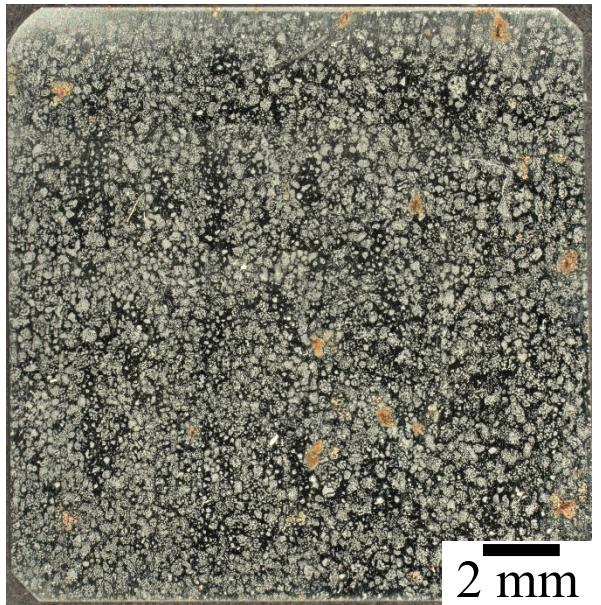
We will mostly show results from 316L material at 40% RH and 300  $\mu\text{g}/\text{cm}^2$  ASW. (UUR)

# Optical images of 316L specimen with $300 \mu\text{g}/\text{cm}^2$ ASW at 40% RH

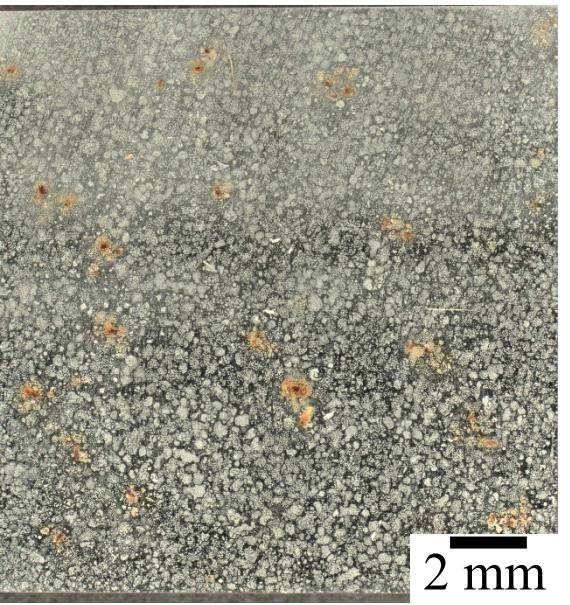
8



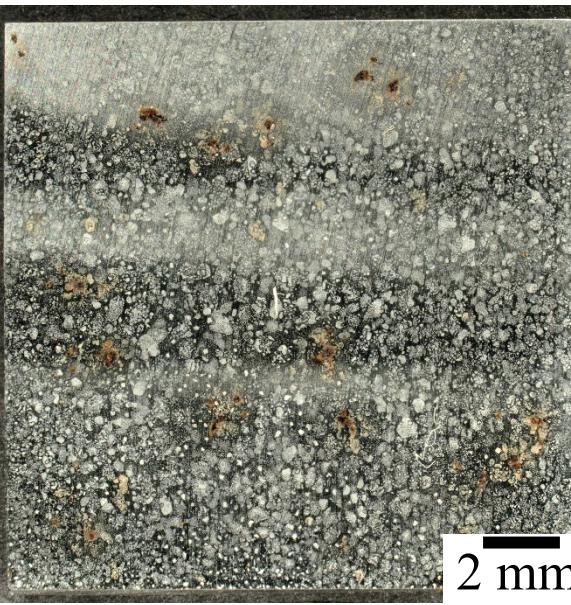
1 week



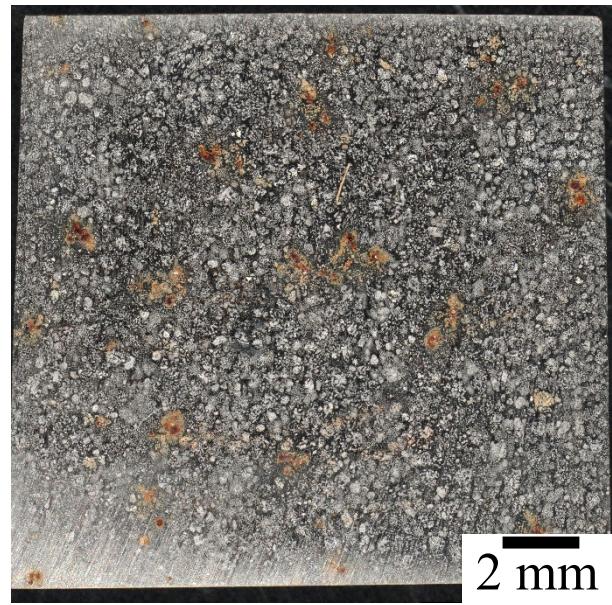
1 month



6 month



12 month



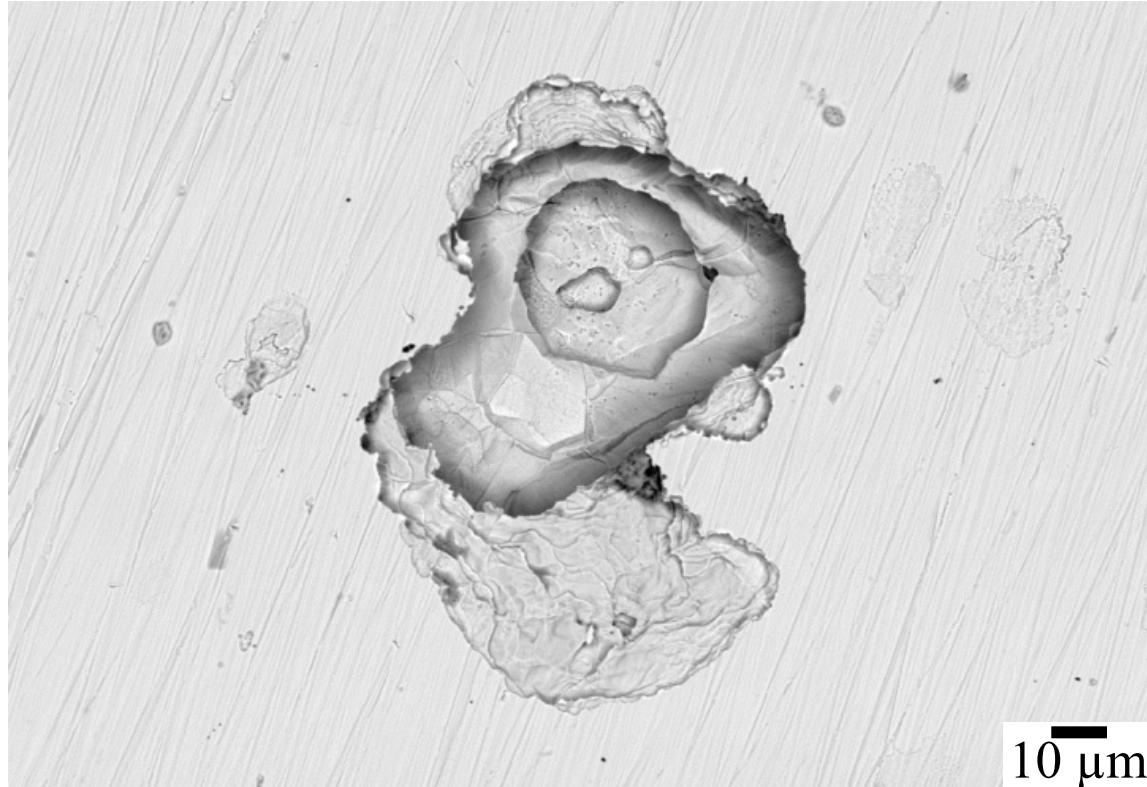
Wrought

AM

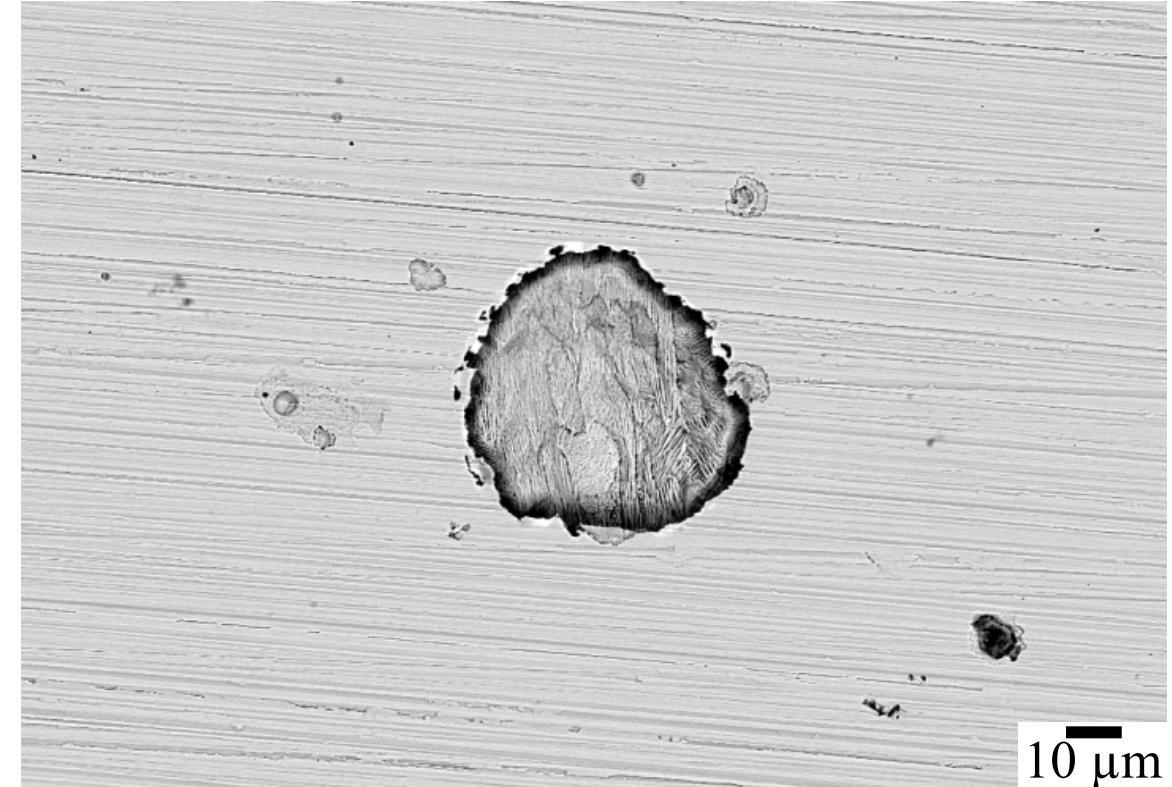
2 mm



**Wrought**



**AM**

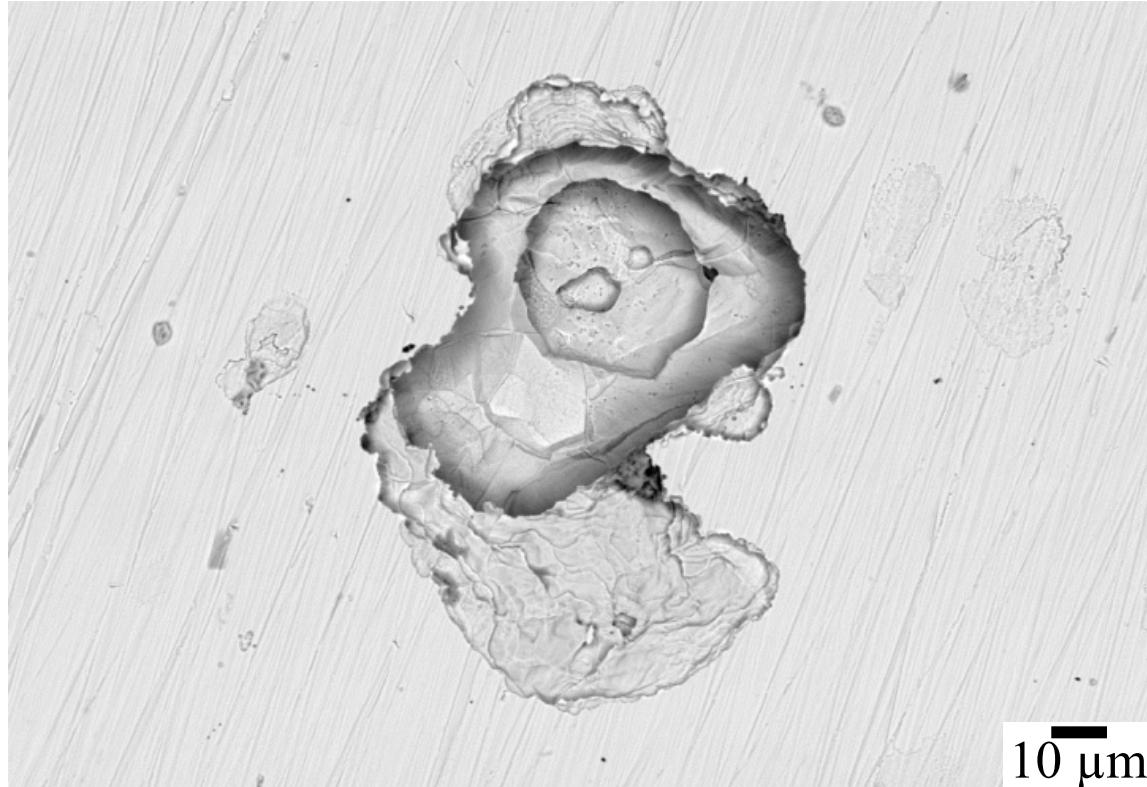


Cleaned surfaces with nitric acid solution.

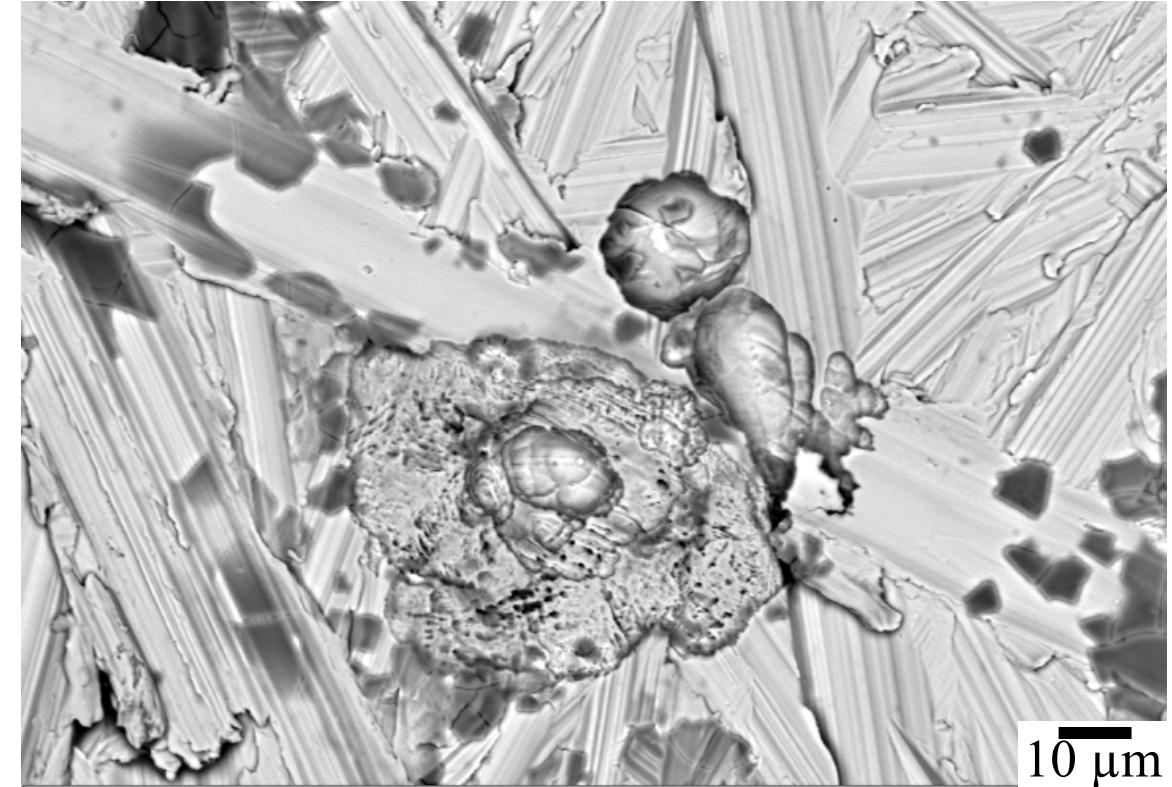
# SEM images of Wrought 316L specimen - Comparison



**40% RH - 300  $\mu\text{g}/\text{cm}^2$  of ASW**



**From 21 month exposure off coast of Florida**

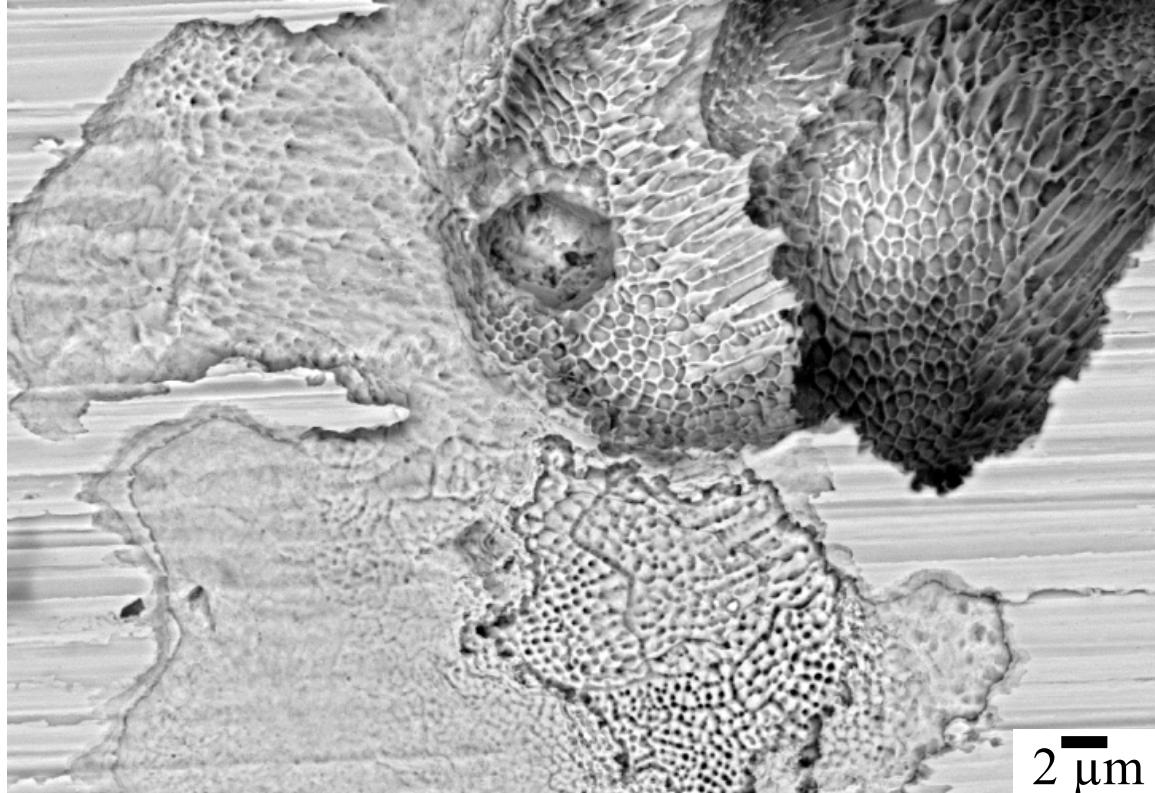


*These samples were prepared to 120 grit ground finish.*

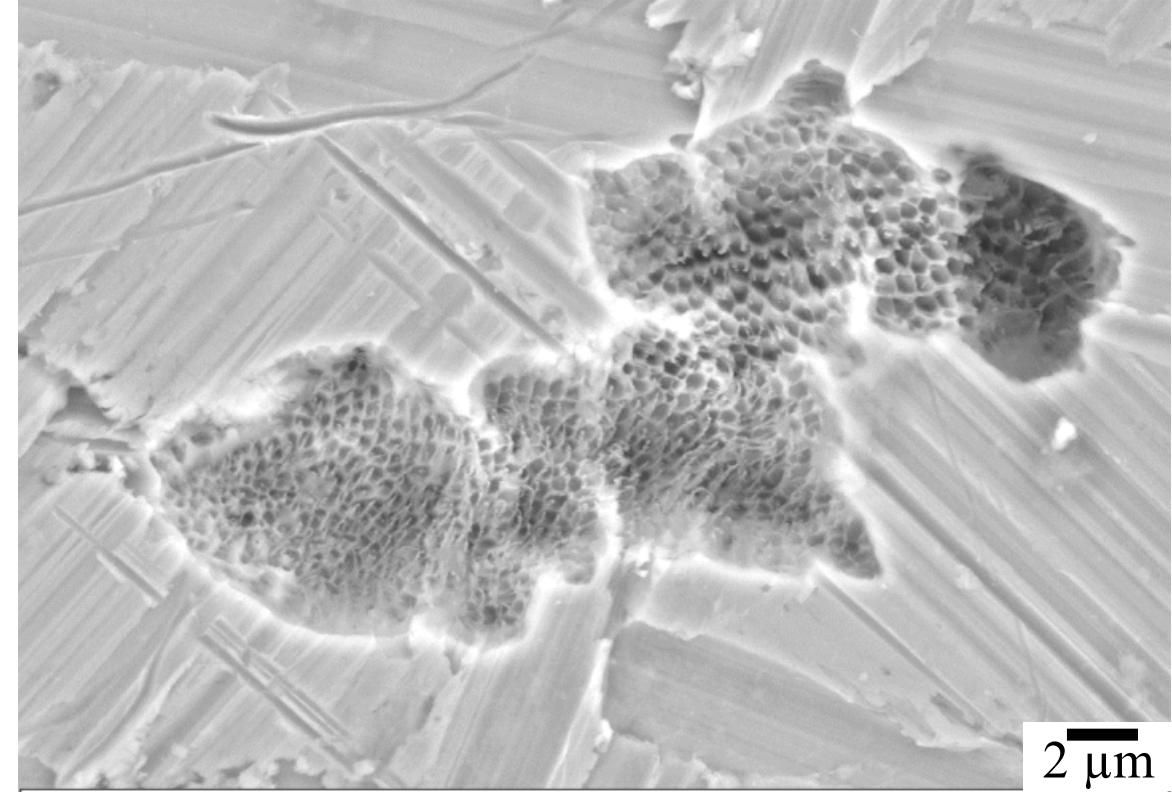
# SEM images of AM 316L specimen - Comparison



**40% RH - 300  $\mu\text{g}/\text{cm}^2$  of ASW**



**From 21 month exposure off coast of Florida**

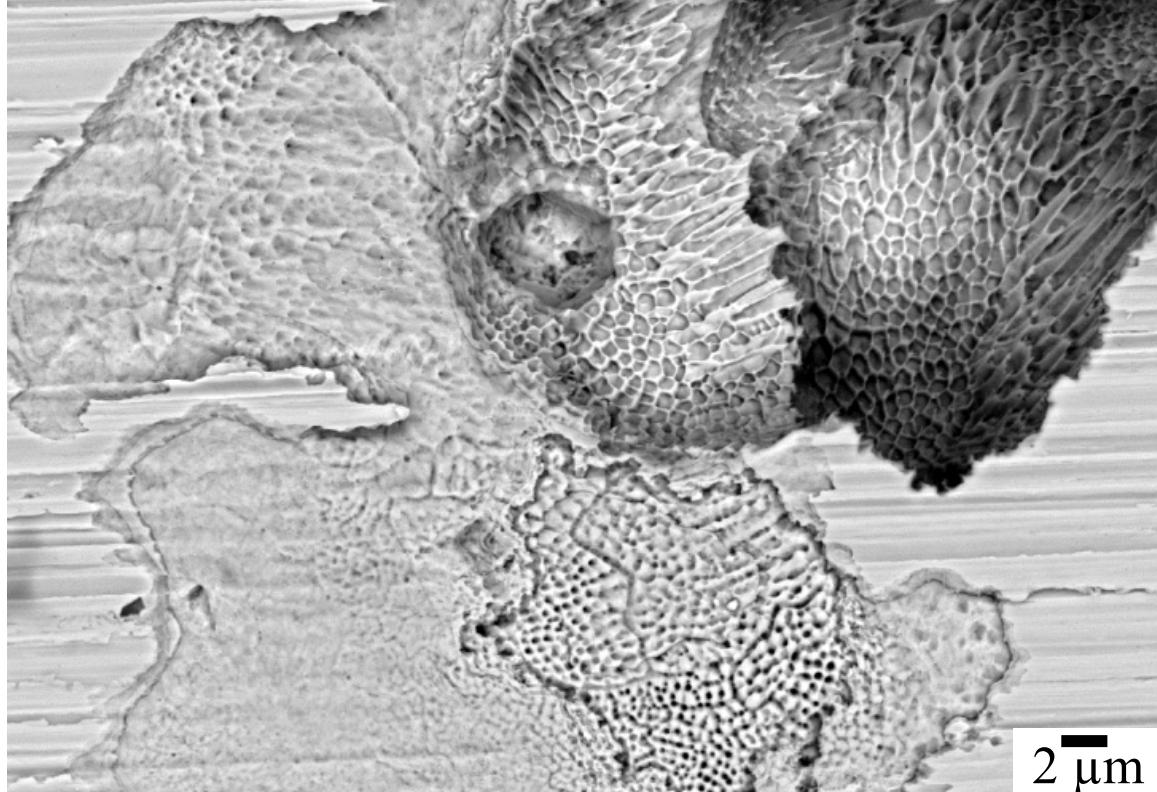


*These samples were prepared to 120 grit ground finish.*

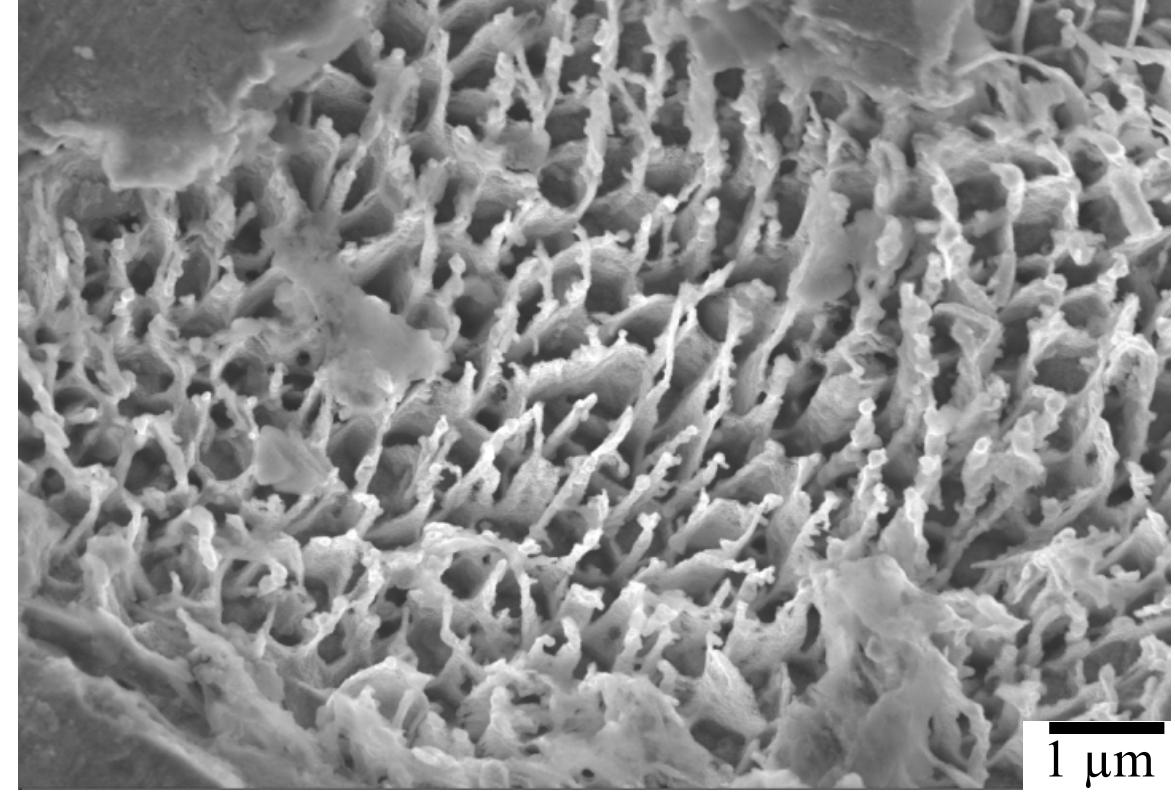
## SEM images of AM 316L specimen - Comparison



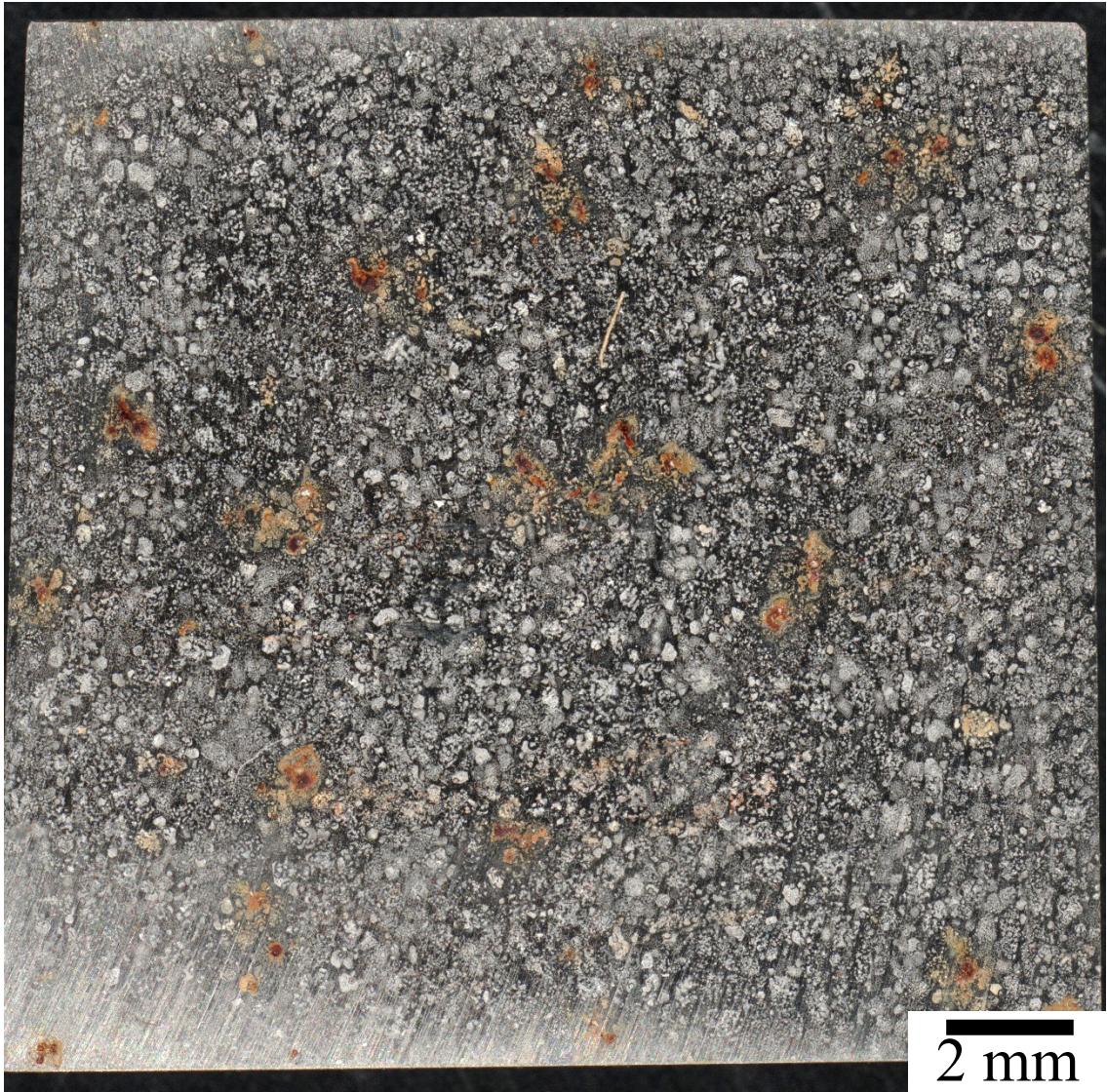
**40% RH - 300  $\mu\text{g}/\text{cm}^2$  of ASW**



**From 21 month exposure off coast of Florida**



# Pit Analysis Process

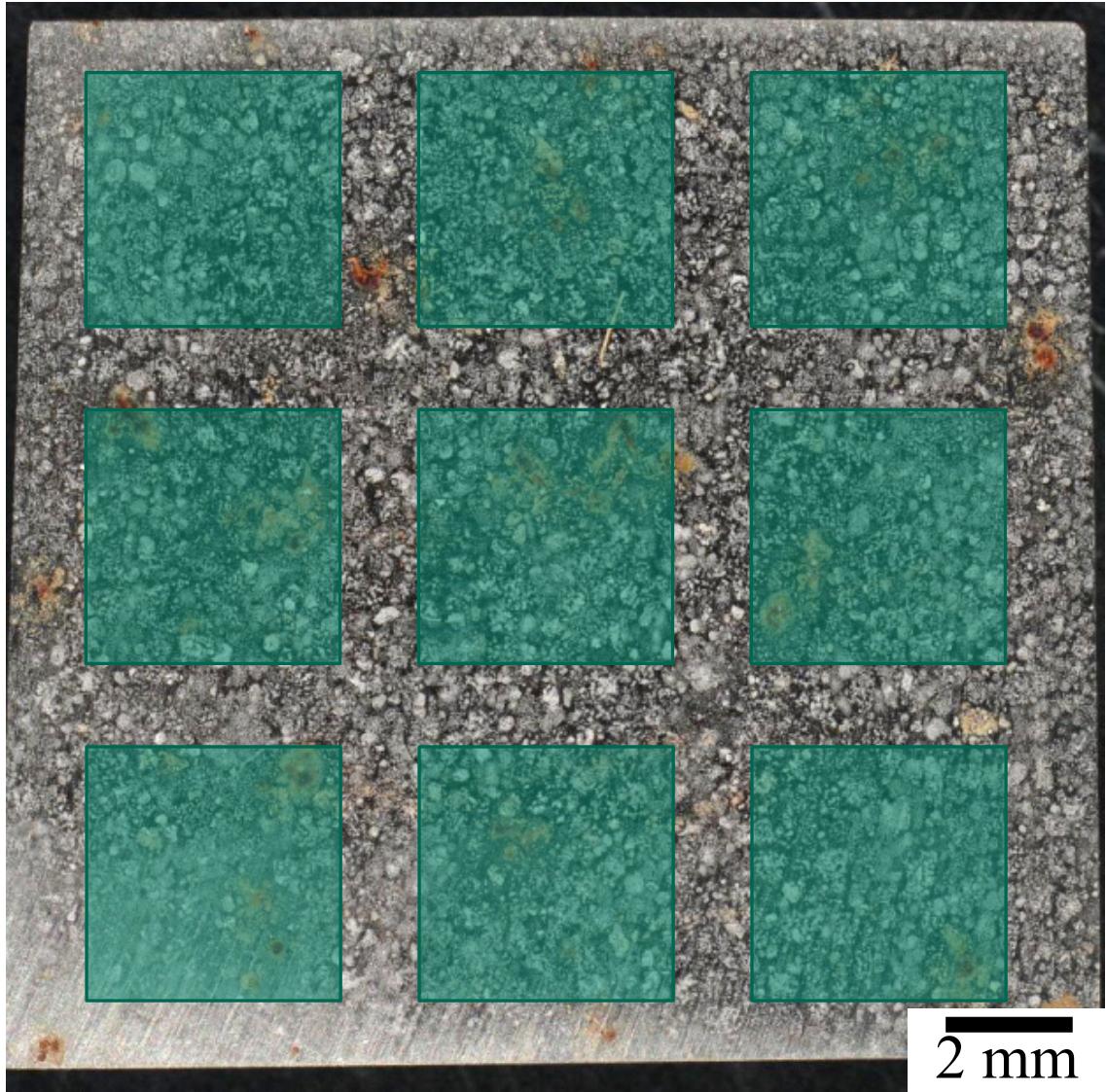


(UUR)

White light interferometry  
(Nexview™)



# Pit Analysis Process

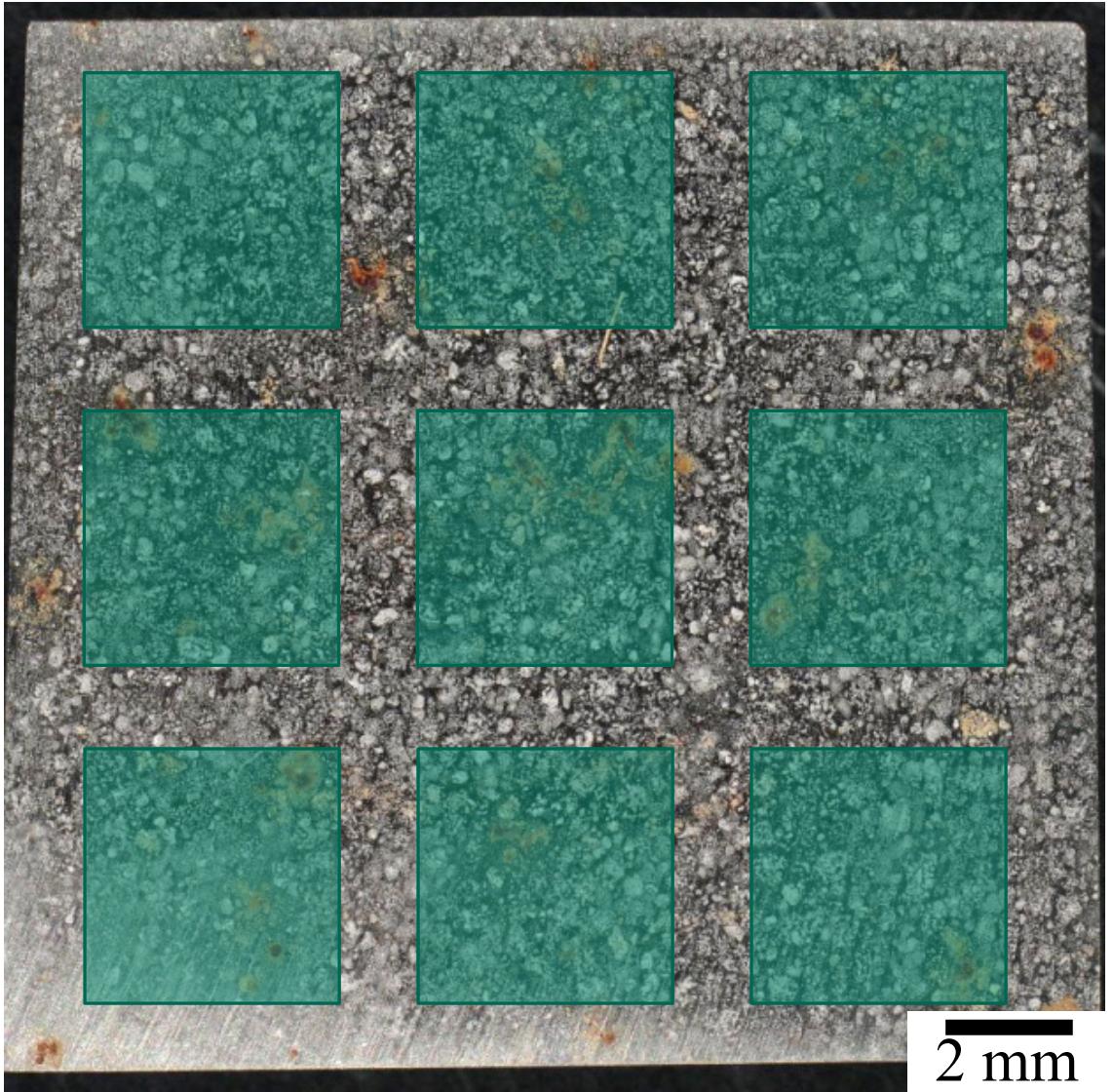


(UUR)

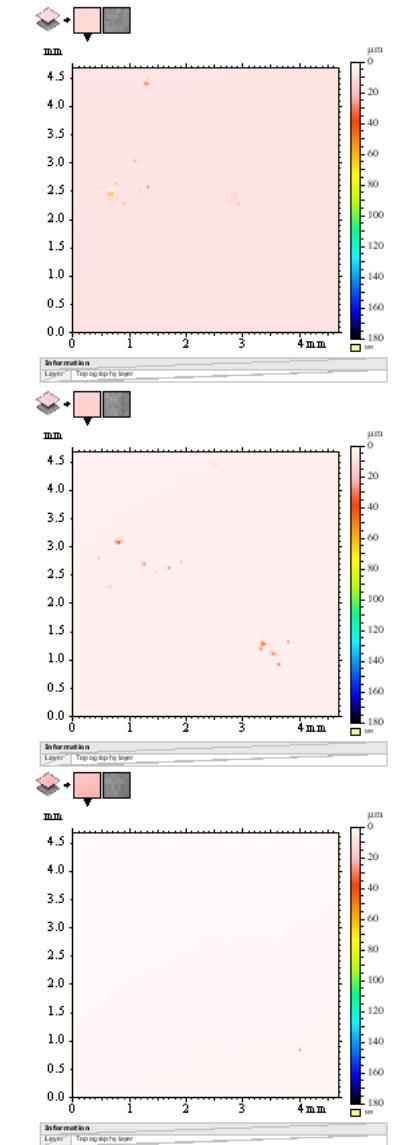
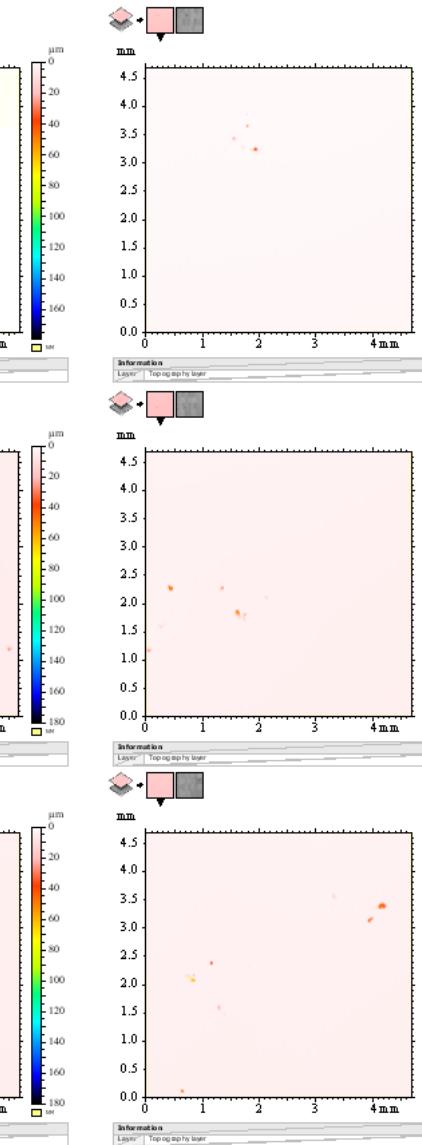
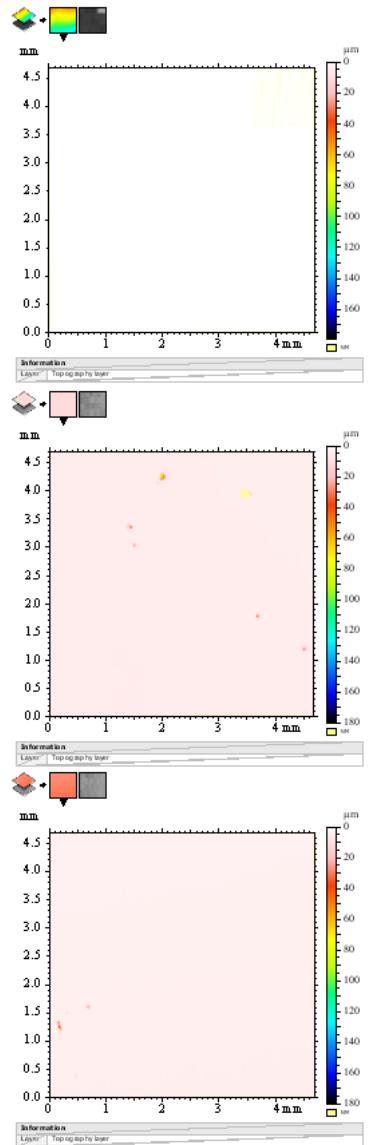
**White light interferometry  
(Nexview™)**



# Pit Analysis Process



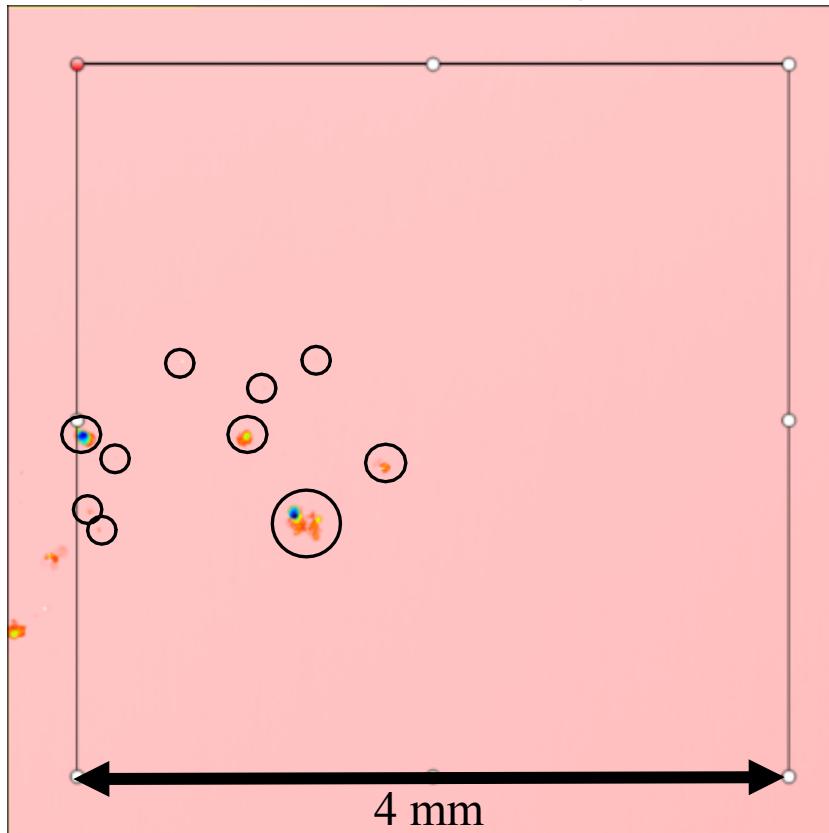
(UUR)



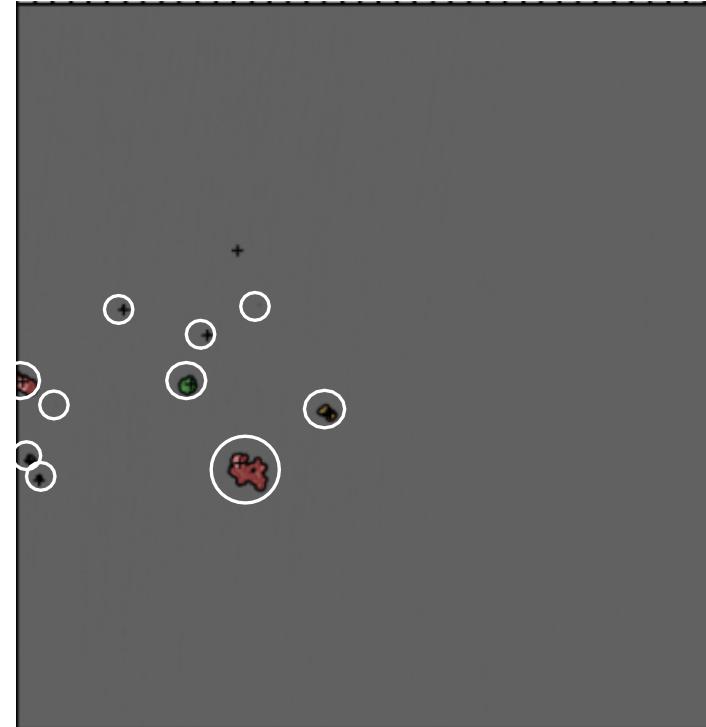
## Raw profilometry data

# Pit Verification Process

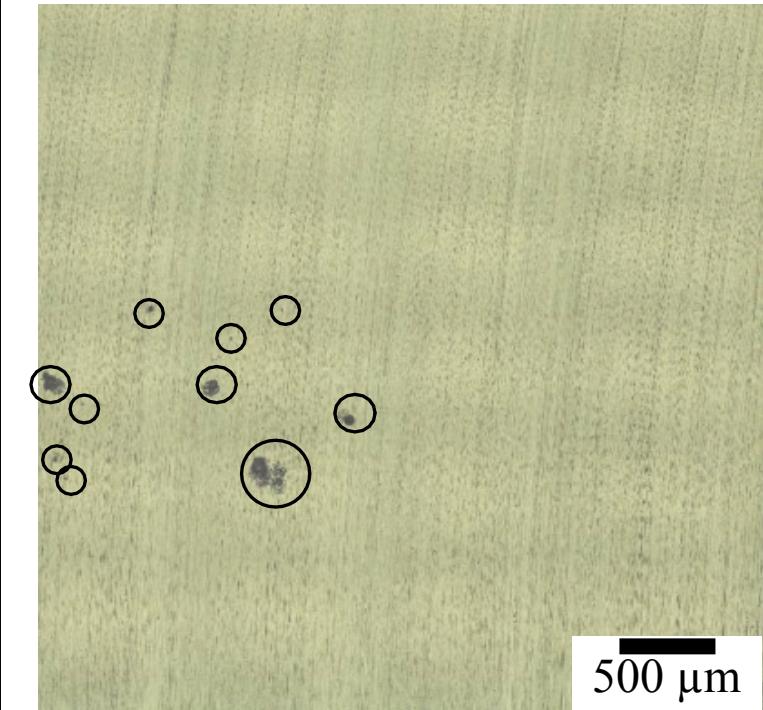
## Raw profilometry data



Processed particle data



Optical image

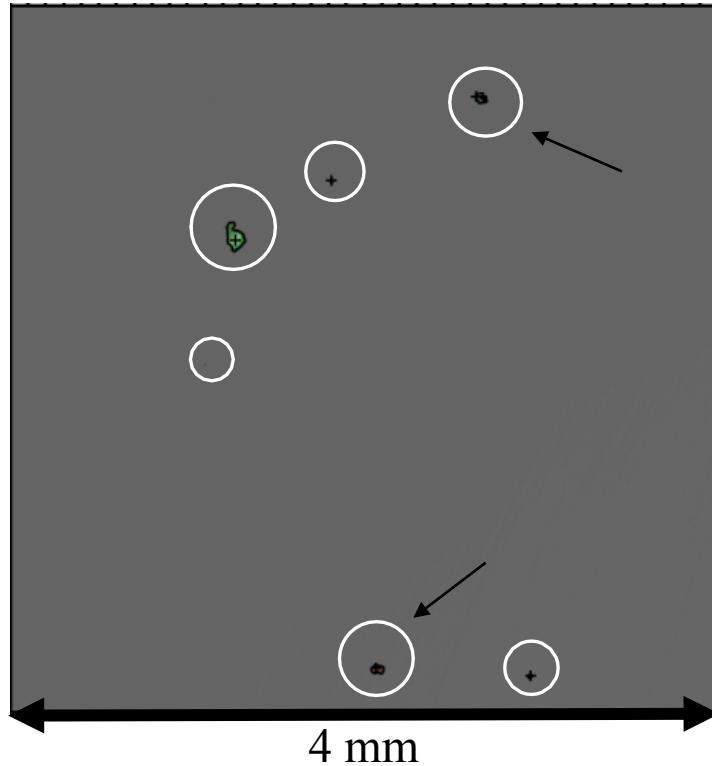


- White light interferometry measurements of cleaned surfaces to detect pits were processed to smooth surfaces.
- Mean plane threshold was used to eliminate scratch marks: **0.5 μm below surface**.
- Filtering criteria: **Maximum Height (Depth) > 0.2 μm** and **Volume/Area ratio > 0.2 μm**.
- Each circle contains one pit. *There were a total of 10 pits through the filtering process.*
- Through visual inspection, the particles are verified to be pits.

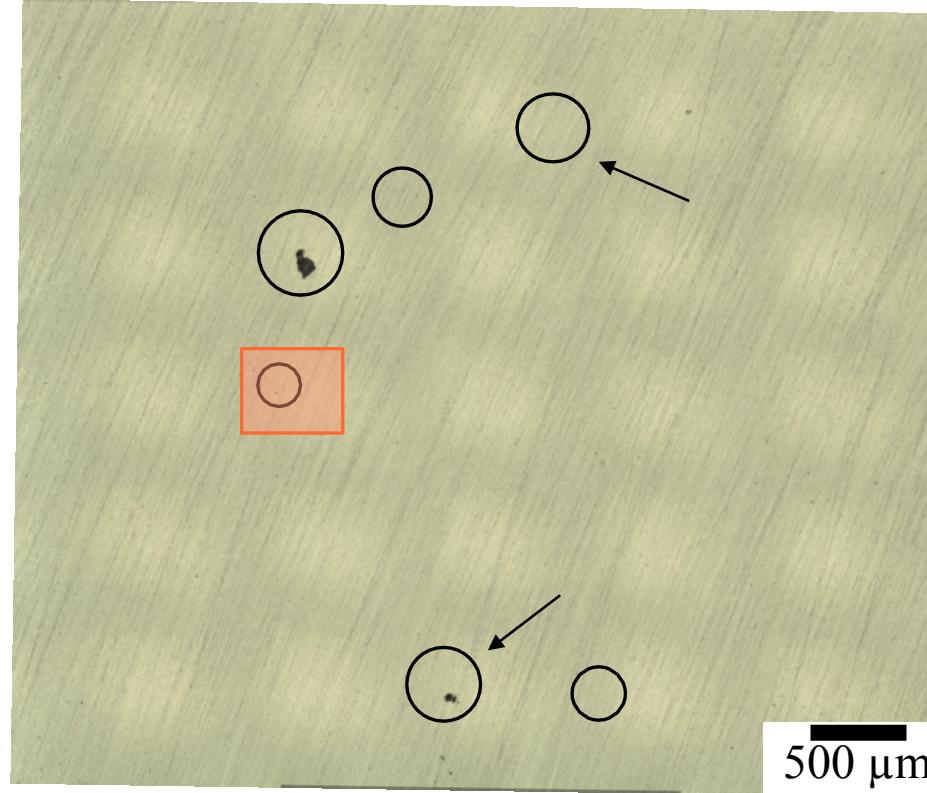
# Pit Verification Process – what are we missing?



Processed particle data



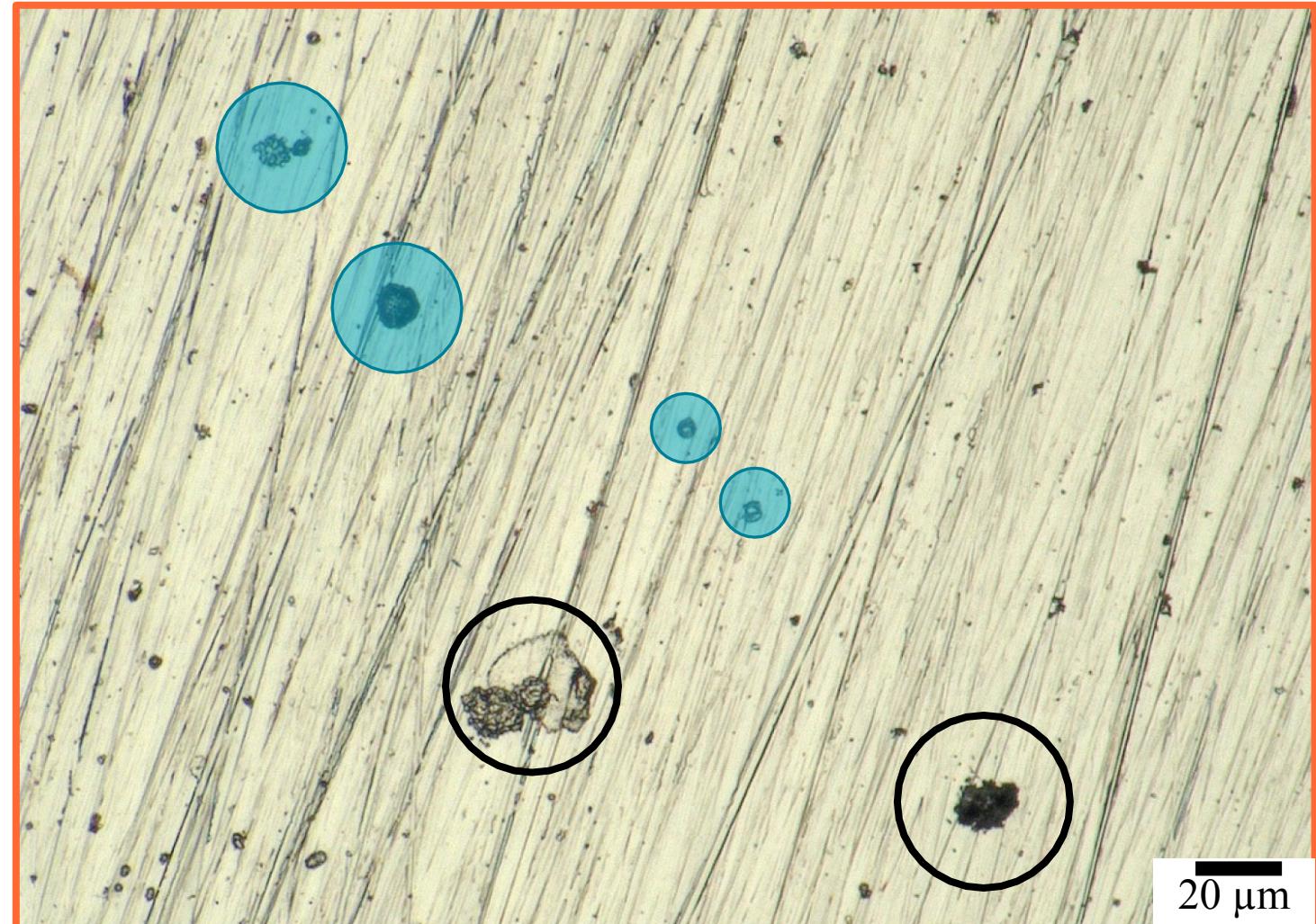
Optical image



# Pit Verification Process – what are we missing?

## Optical image

- Pits not picked up from interferometry analysis because they are **too small**.
- Specific to AM – lack of fusion **porosity** can cause outlier pit information.
- **Contamination** of the surface can lead to unexpected pits – make sure surfaces are clean before measurements!



Another filter based on Height/Area ratio was used to eliminate contamination/pore outliers.

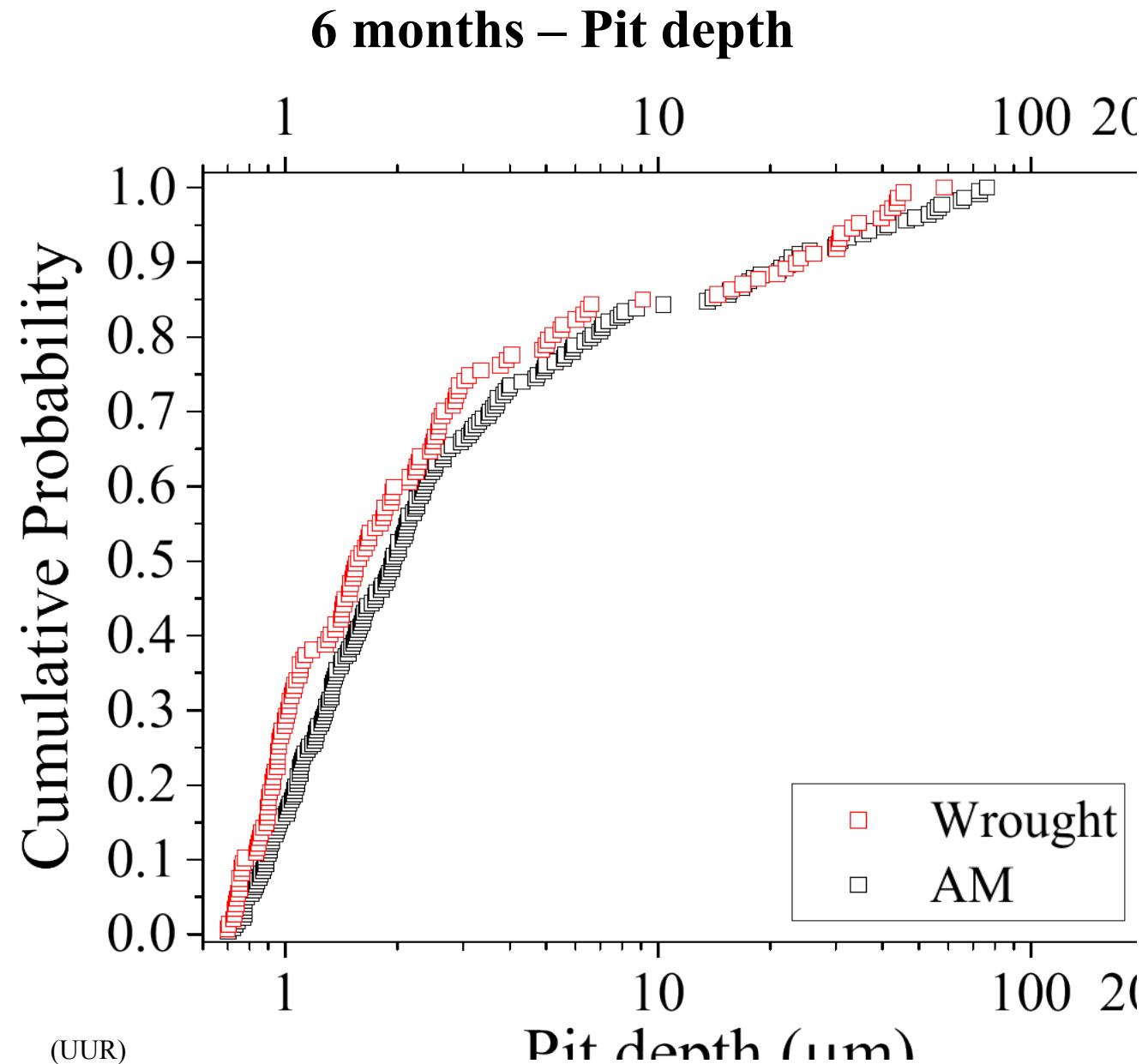
**Wrought**

	Average pit depth ( $\mu\text{m}$ )	Measured deepest pit ( $\mu\text{m}$ )
1 week	2.9	35.1
1 month	7.0	41.8
6 months	6.3	58.5
12 months	11.7	118.1

**AM**

	Average pit depth ( $\mu\text{m}$ )	Measured deepest pit ( $\mu\text{m}$ )
1 week	4.0	32.6
1 month	10.1	210.4 (73.3)
6 months	7.6	76.2
12 months	9.4	80.2

Likely porosity



## Wrought

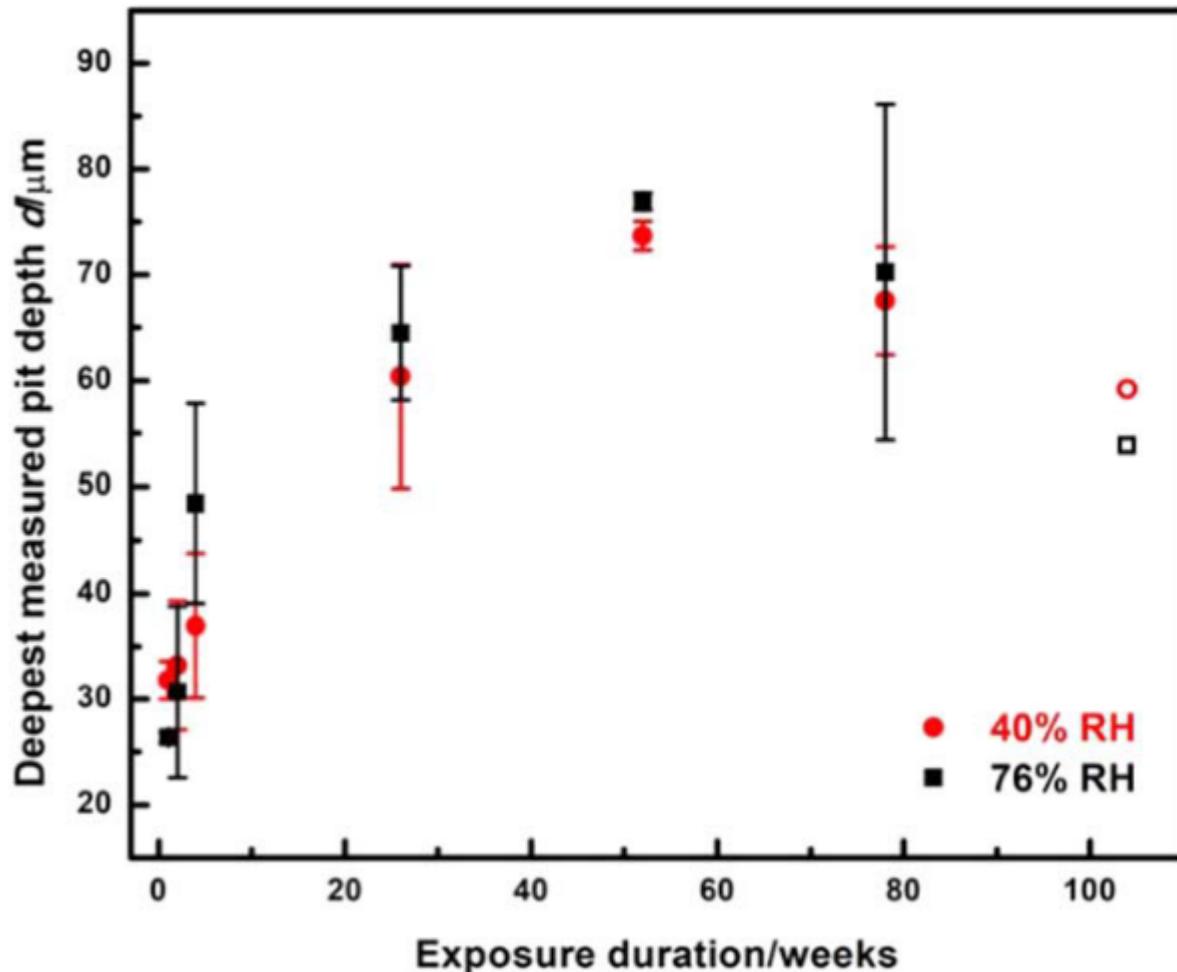
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## AM

	Average pit depth ( $\mu\text{m}$ )	Measured deepest pit ( $\mu\text{m}$ )
1 week	4.0	32.6
1 month	10.1	210.4 (73.3)
6 months	7.6	76.2
12 months	9.4	80.2

Likely porosity

## 10 deepest pits/area from literature – 304L



Source: Srinivasan (2021)

# Pit statistics of 316L specimen - 300 $\mu\text{g}/\text{cm}^2$ ASW at 40% RH



## Wrought

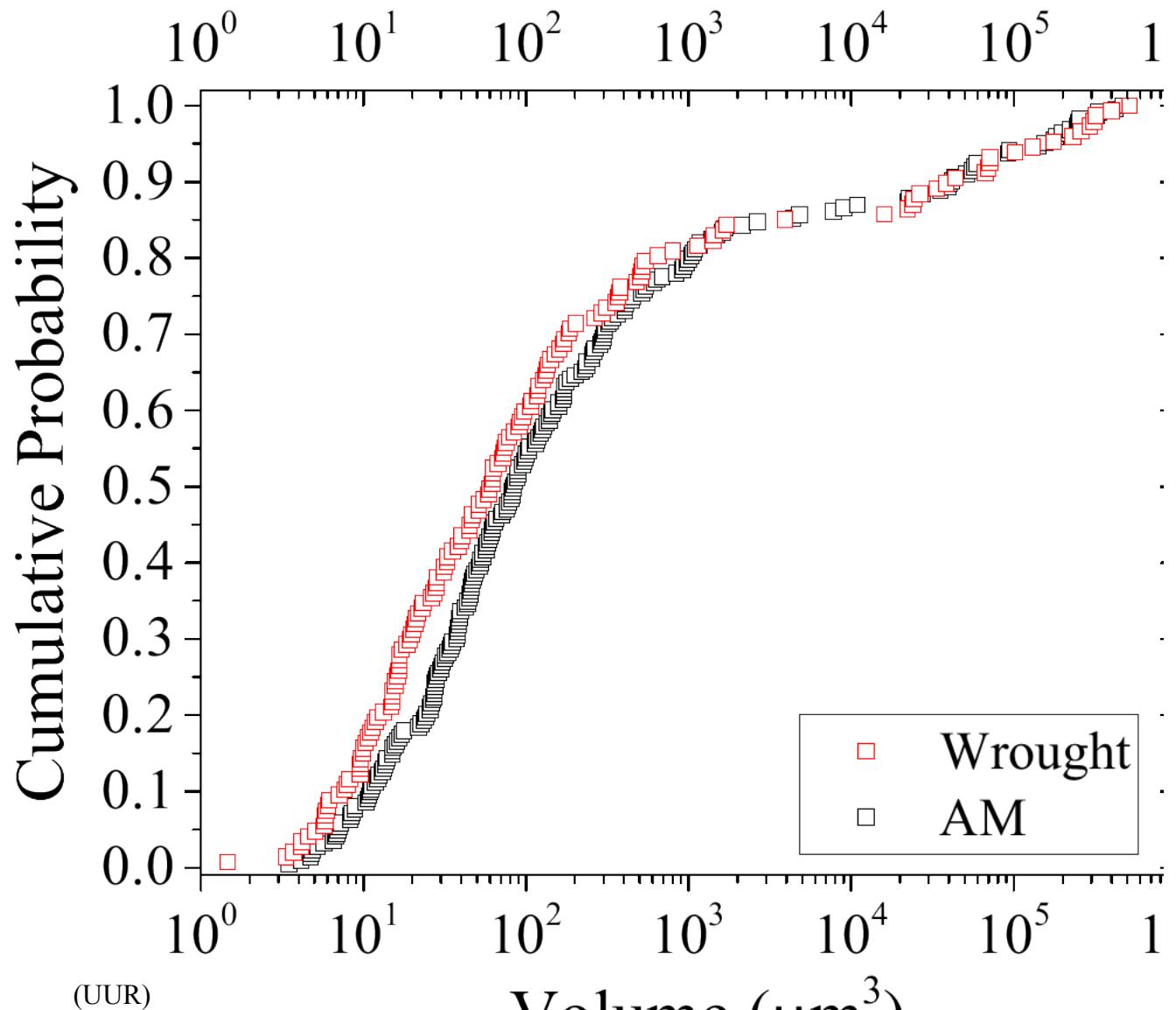
	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$1.15 \times 10^5$	50
1 month	$9.28 \times 10^5$	65
6 months	$2.34 \times 10^6$	105
12 months	$2.75 \times 10^6$	96

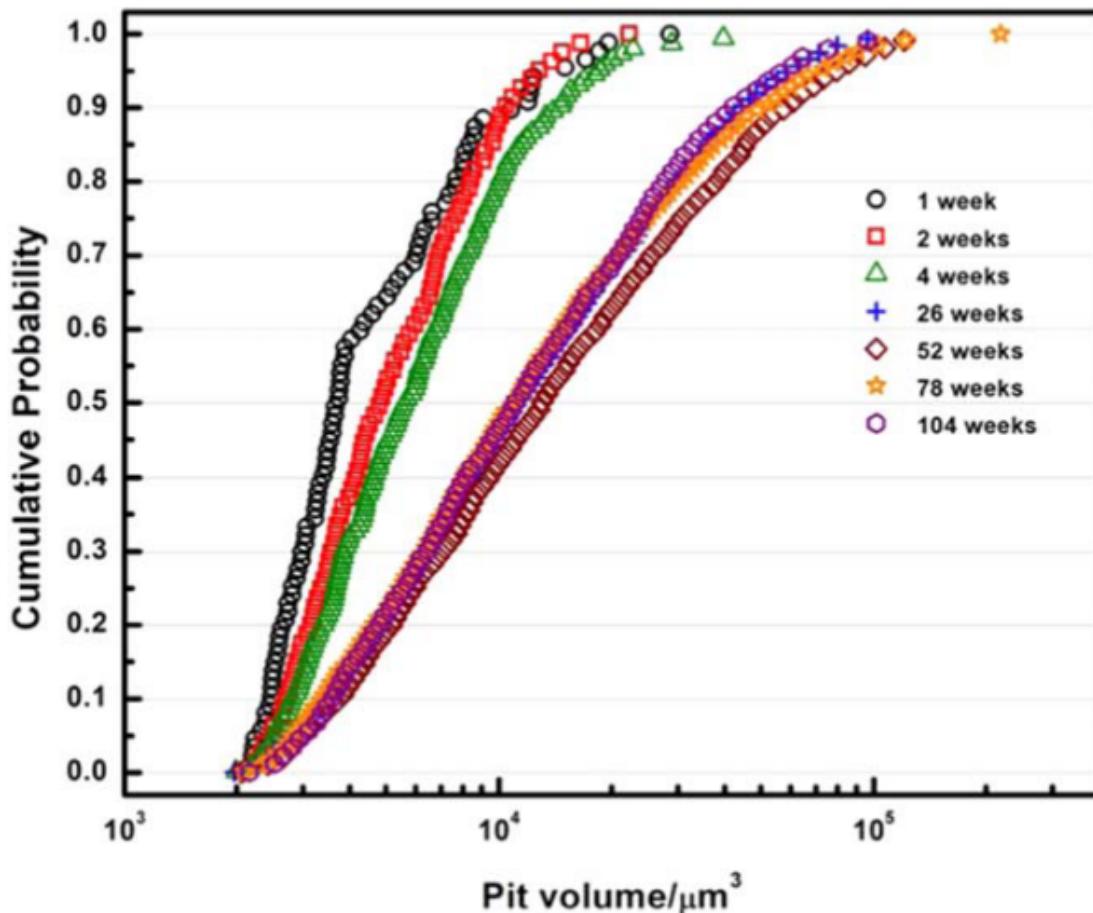
## AM

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$3.66 \times 10^5$	89
1 month	$9.35 \times 10^5$	88
6 months	$2.97 \times 10^6$	155
12 months	$1.74 \times 10^6$	53

Total area analyzed: 1.3 to 1.44  $\text{cm}^2$

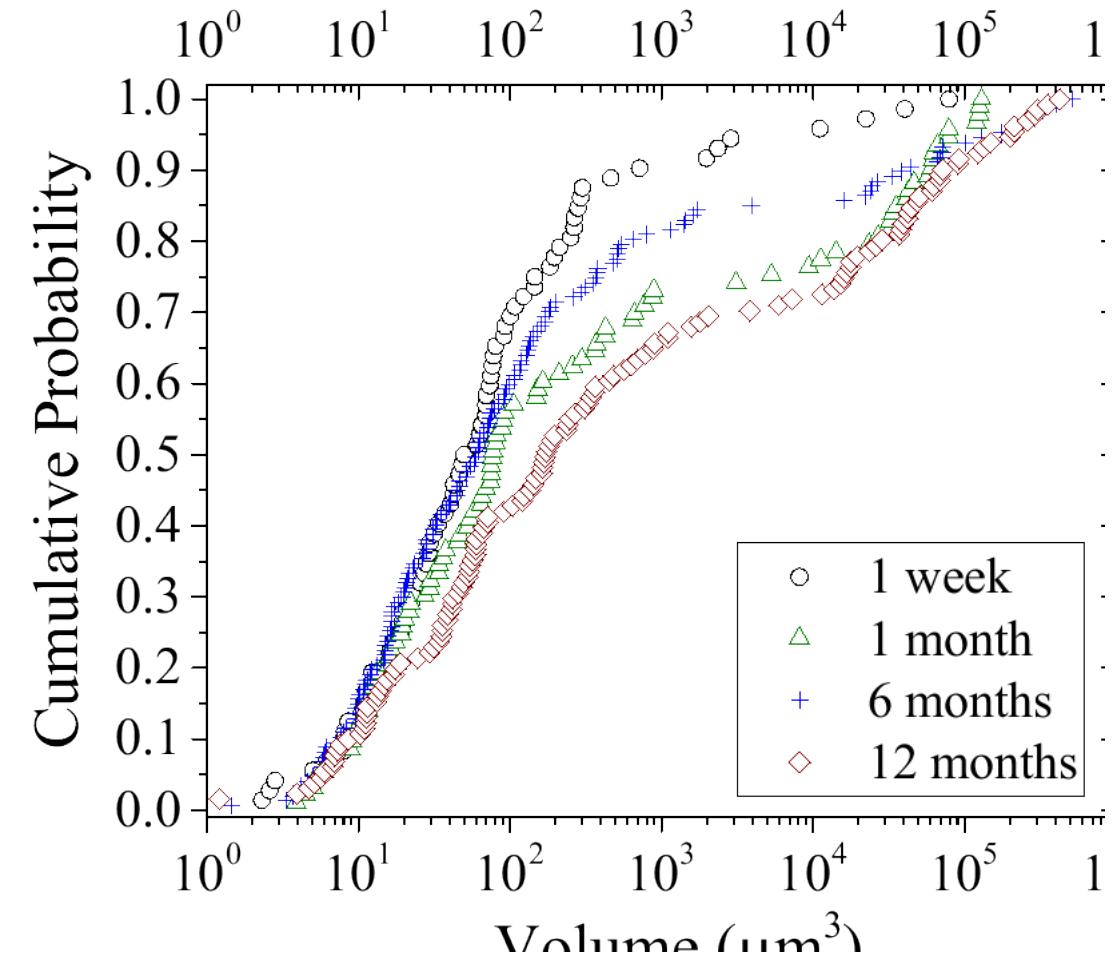
## 6 months – Pit volume

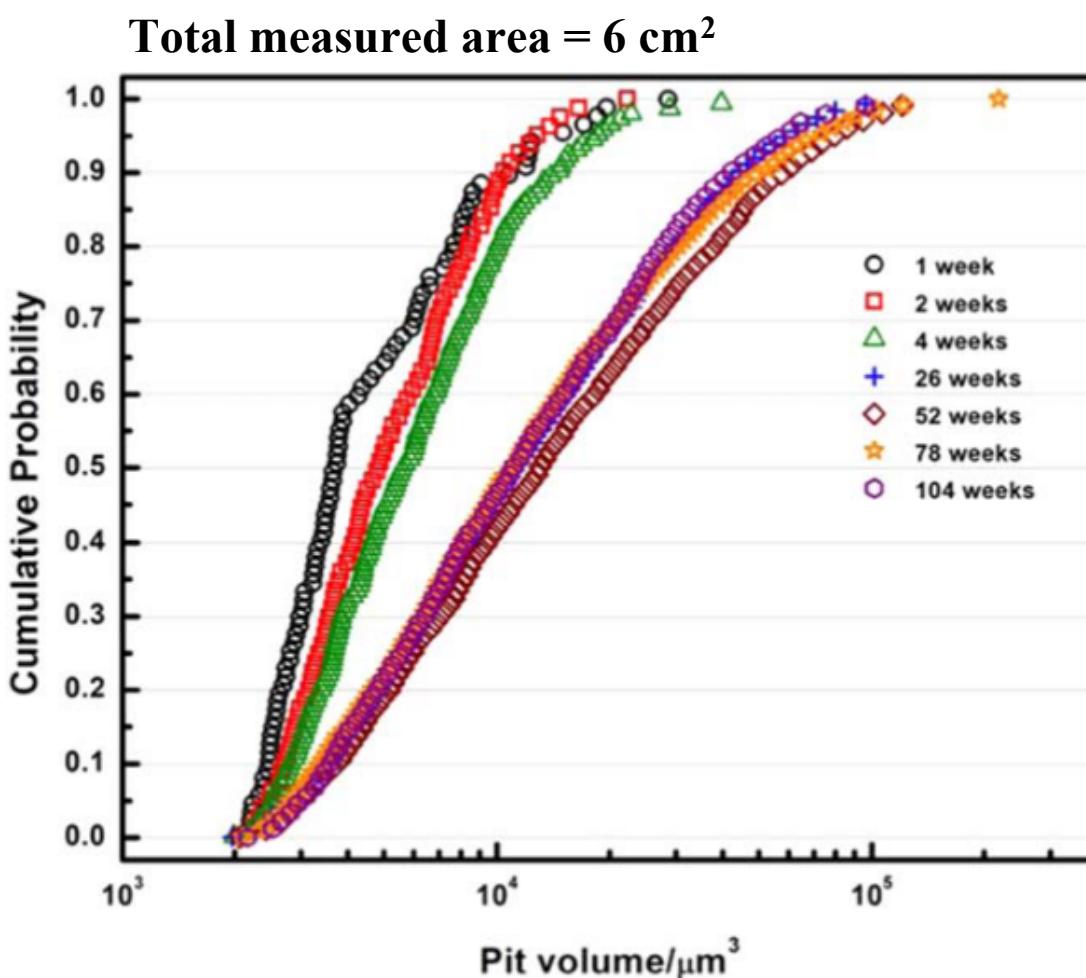




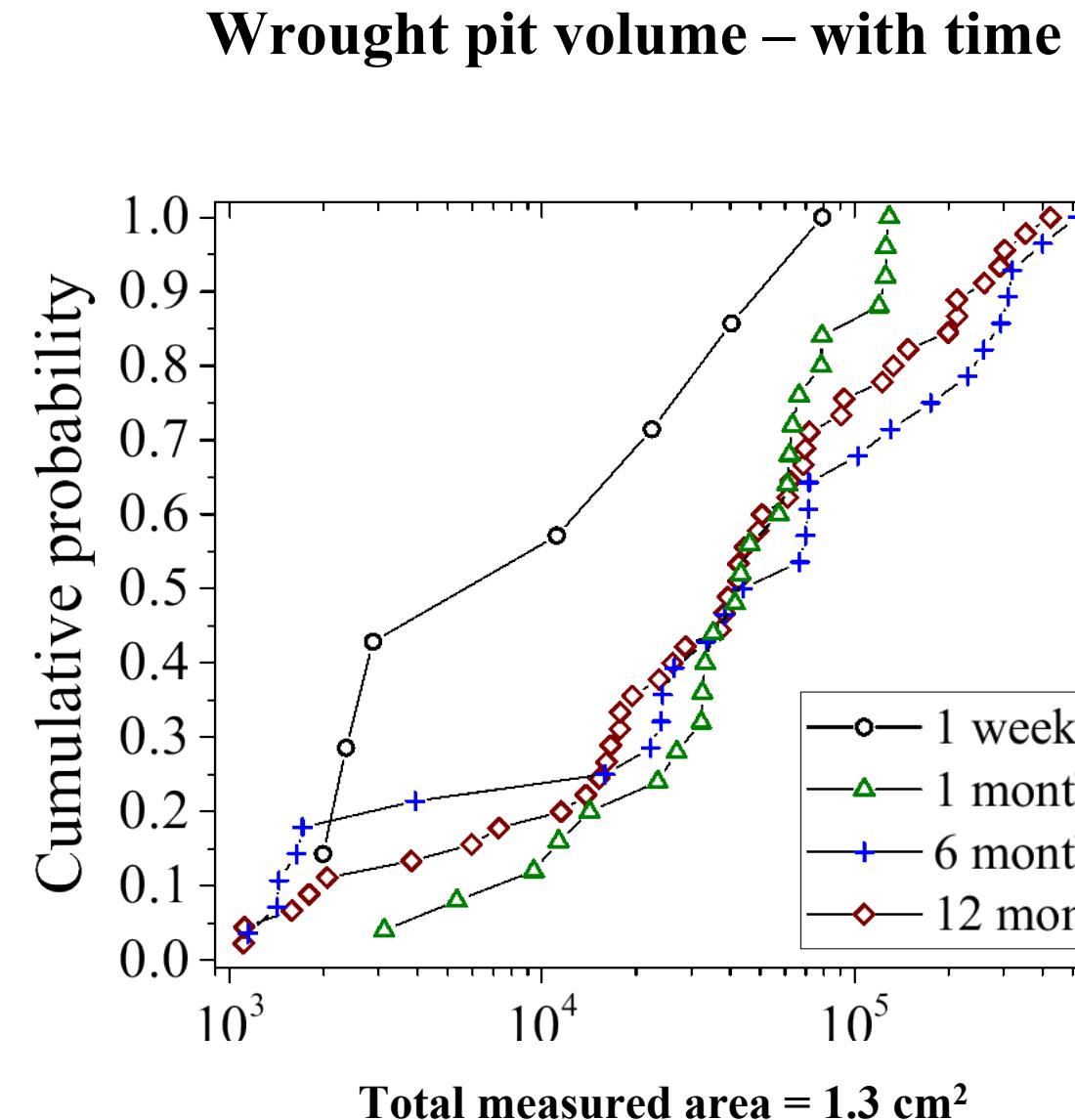
J. Srinivasan, et al., Long-Term Effects of Humidity on Stainless Steel Pitting in Sea Salt Exposures, *J. Electrochem. Soc.*, 168 (2021).

## Wrought pit volume – with time



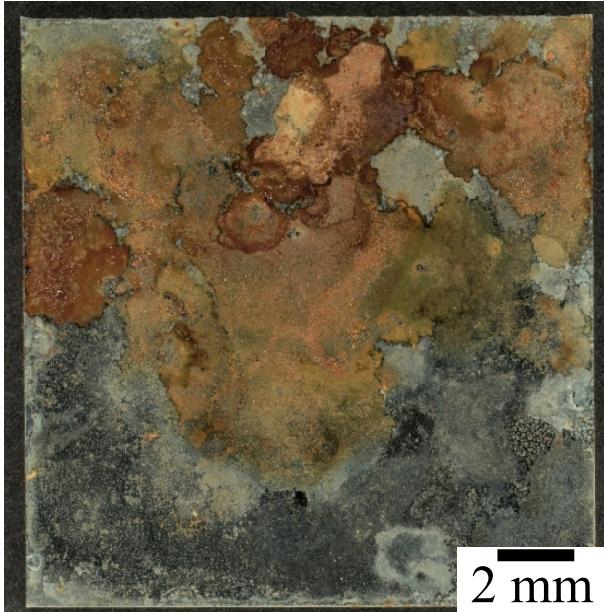


J. Srinivasan, et al., Long-Term Effects of Humidity on Stainless Steel Pitting in Sea Salt Exposures, *J. Electrochem. Soc.*, 168 (2021).

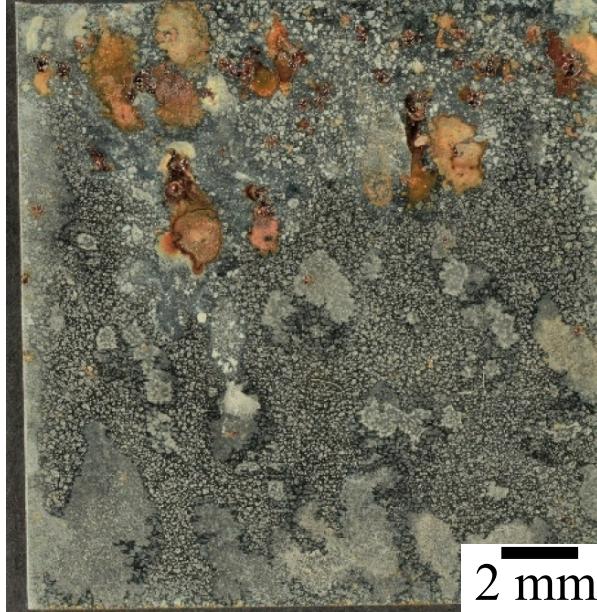


Optical images of AM 304L specimen with  $300 \mu\text{g}/\text{cm}^2$  ASW at 76% RH

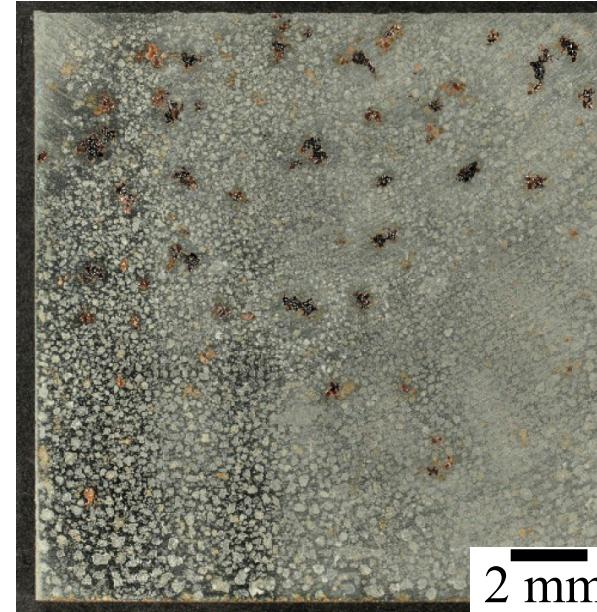
1 week



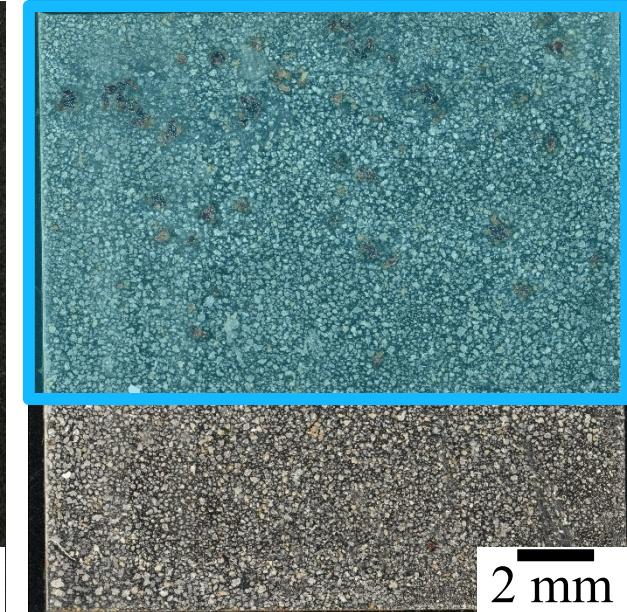
1 month



6 month



12 month

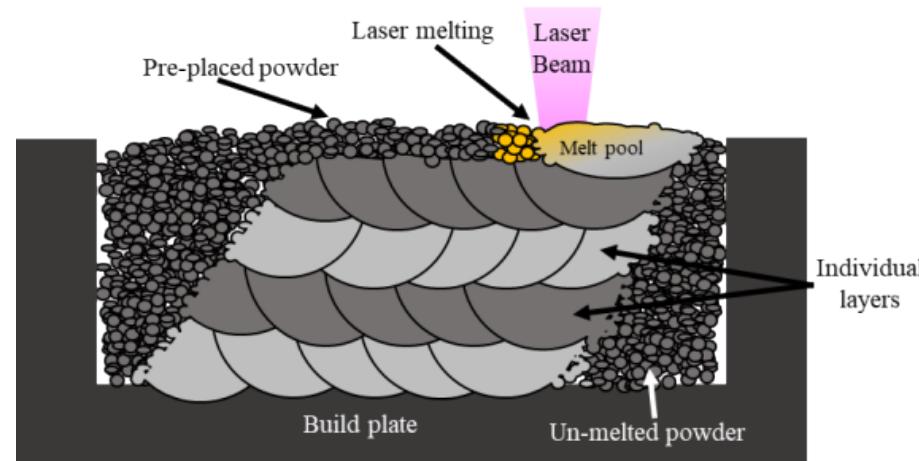


# Constant humidity/contamination experiments

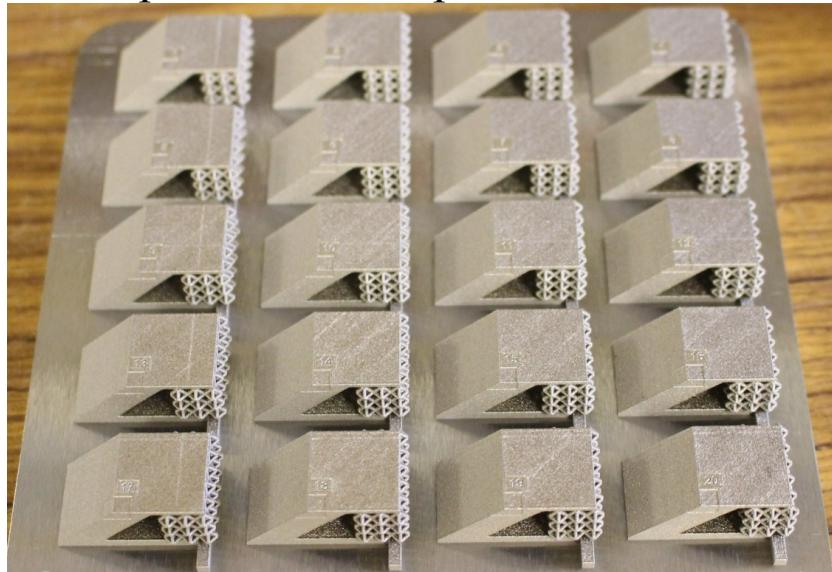


- Pit morphologies were similar to coastal exposures, besides the generally deeper pits for the constant humidity exposures.
- Pit density seems to be higher for AM samples, working on three potential hypotheses:
  - More initiation because of more preferential sites (oxides, cell boundaries, melt pool boundaries, etc.).
  - More initiation because of more efficient cathode (heterogeneous microstructure effects).
  - Or, more pits die (stop corroding) faster because of the heterogeneous microstructure (Cr/Mo enriched cell boundaries block propagation).
- Max pit depth was greater for AM at early times, but switched to Wrought after 12 month exposures.
- Initiation of local corrosion is most likely to occur in lack of fusion pores for samples at elevated humidity ( $\geq 76\%$ ). Most obvious for AM 304L samples.

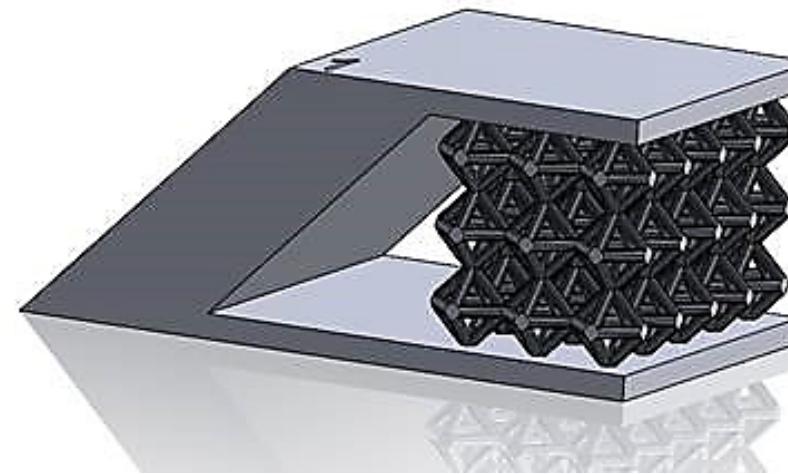
# Powder bed fusion 316L samples



Build plate of 20 samples

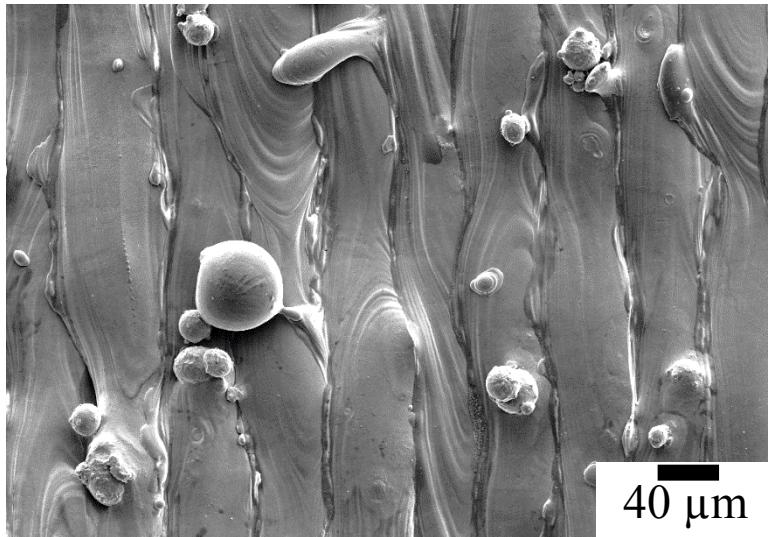


Samples were prepared using 316L powder with a powder bed fusion (PBF) technique.



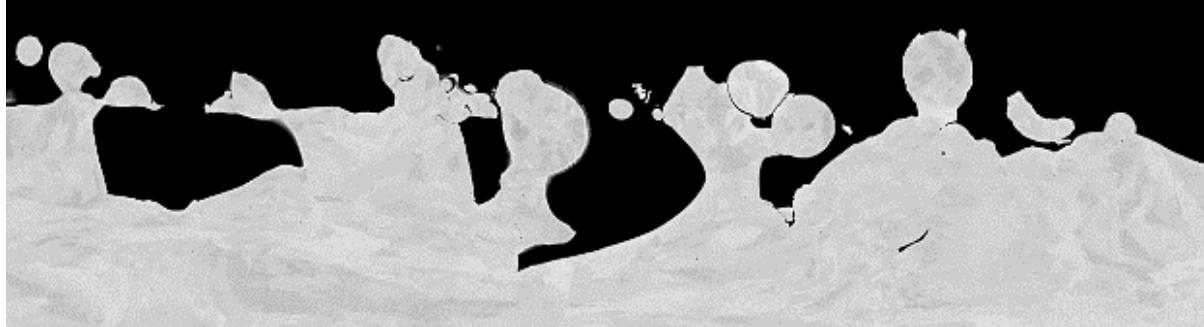
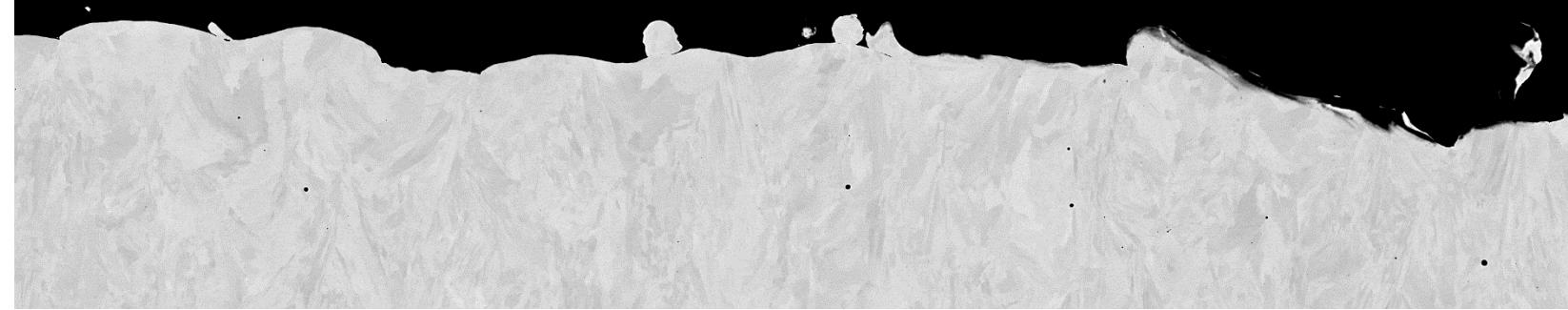
- Samples were printed on 5 different PBF machines, all with the machines optimized parameters for part density.
- SEM images of powder.
- Composition determined by ICP-MS and LECO.
- Powder analysis with laser diffraction.

# What do as-printed surfaces look like?

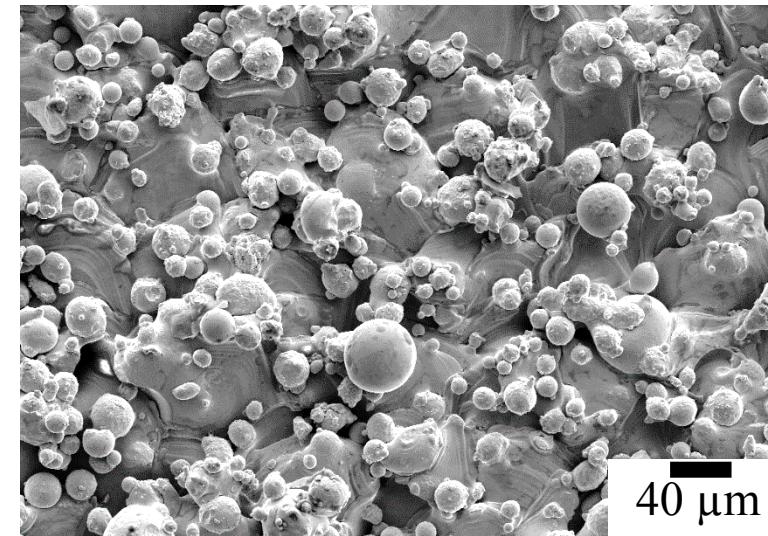


40  $\mu\text{m}$

Top surface

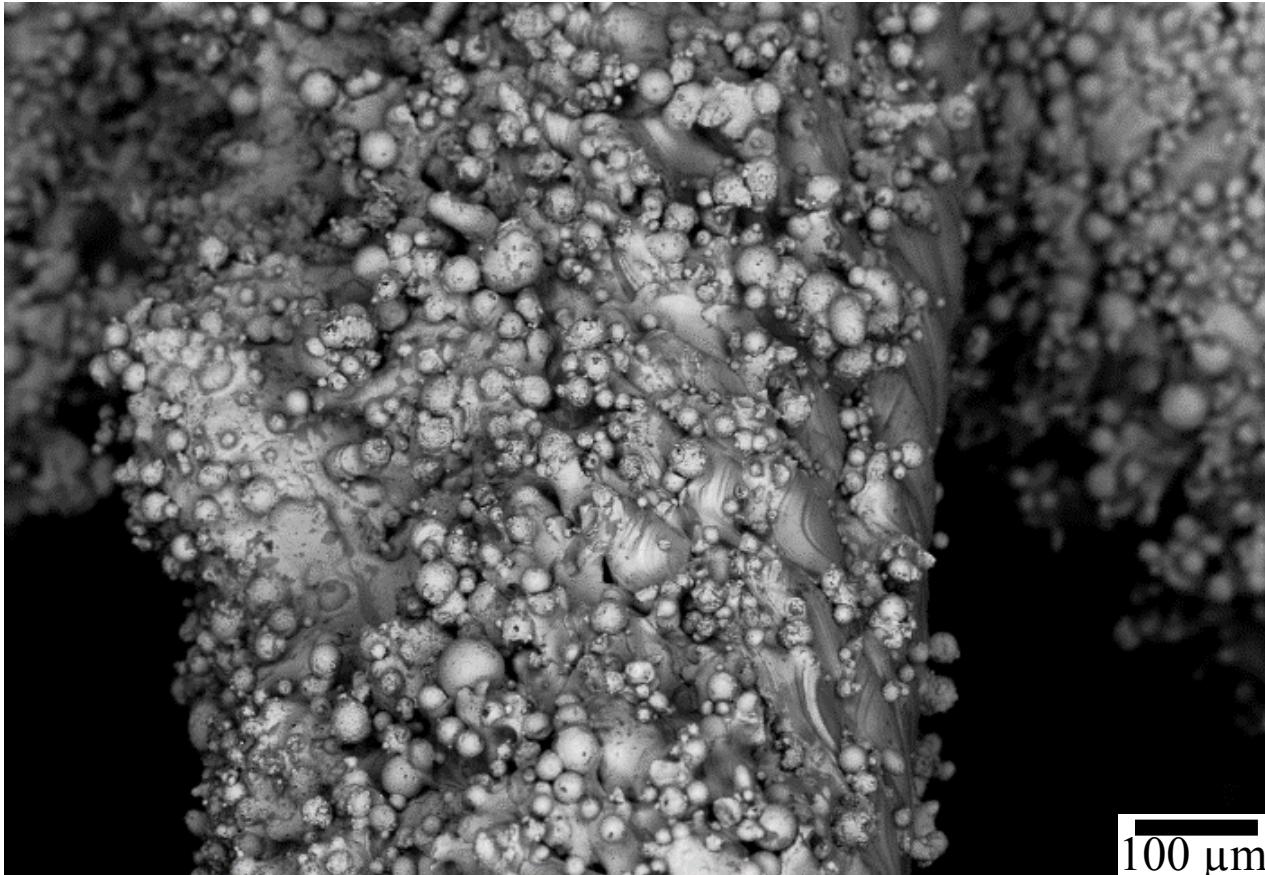


Side surface

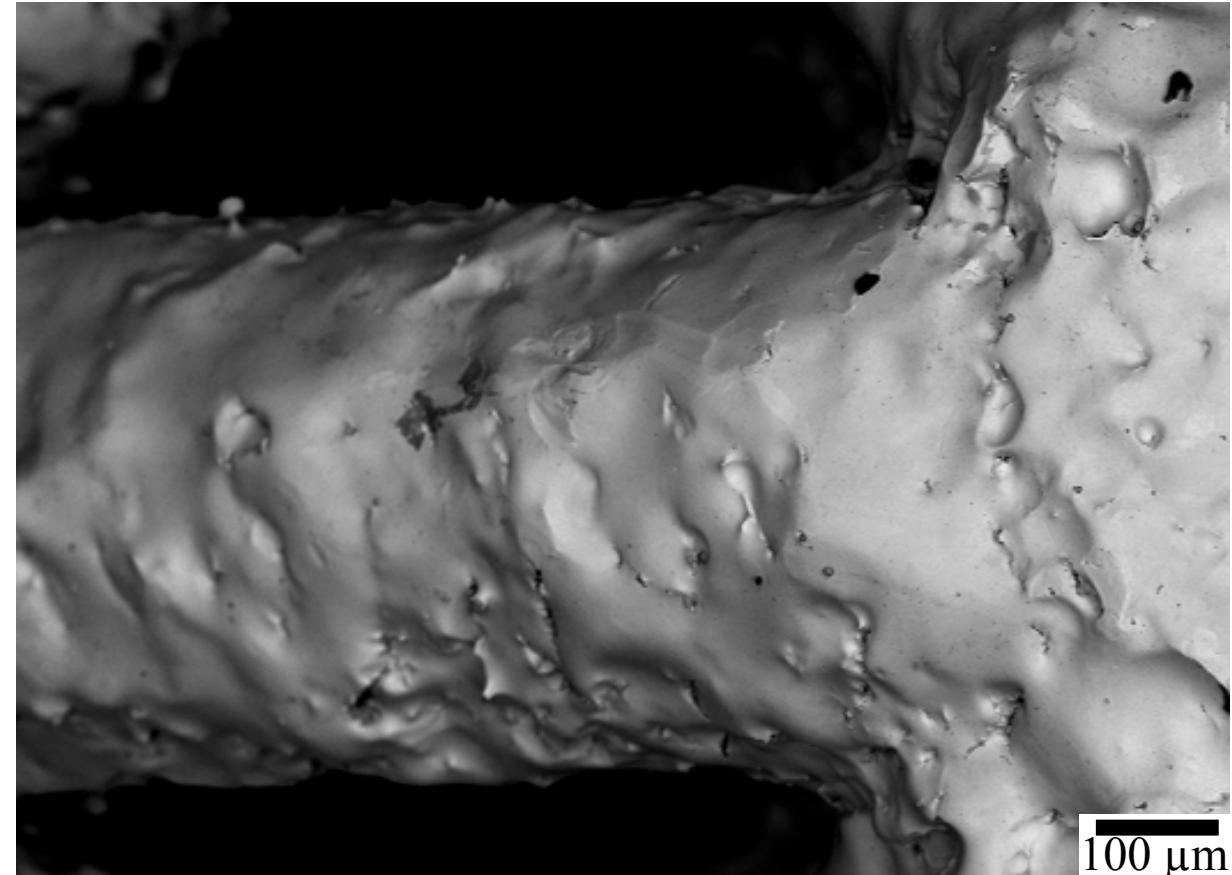


40  $\mu\text{m}$

# As-printed lattices?



As-printed



DC Electropolished

# ASTM G85-A2: Sample ID



1<sup>st</sup> set – ProX200 As-printed

2<sup>nd</sup> set – ProX200

3<sup>rd</sup> set – EOS M290 As-printed

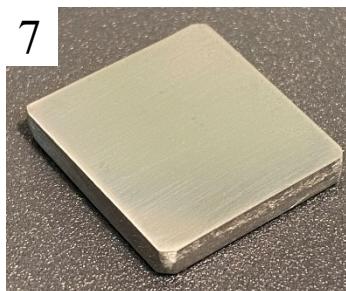
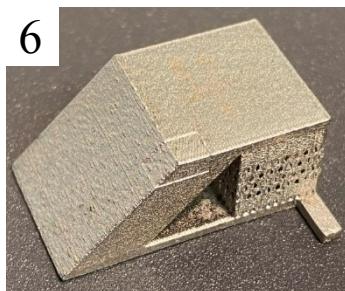
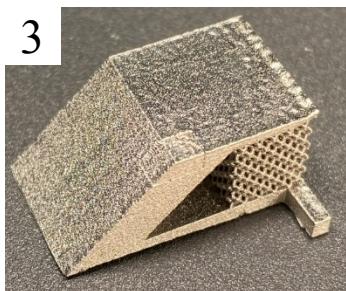
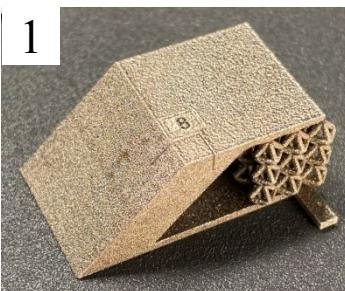
4<sup>th</sup> set – Renishaw AM250 As-printed

5<sup>th</sup> set – SLM280 As-printed

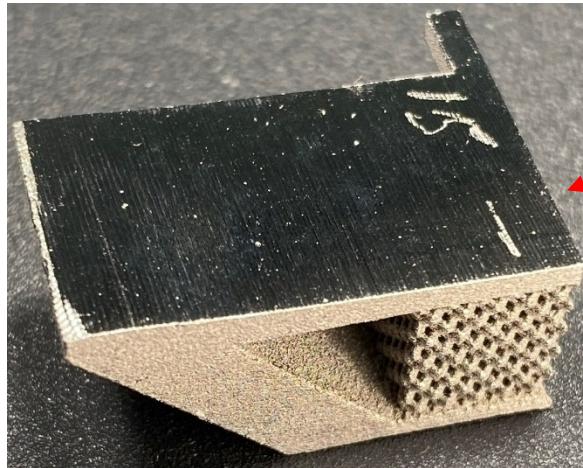
6<sup>th</sup> set – Concept Laser M2 As-printed

7<sup>th</sup> set – Wrought 316L

- *Wire EDM cut surface*
- *Electropolished*
- *Bandsaw cut surface*
- *Ground with 600 grit sand paper*
- *Wire EDM cut surface*
- *Bandsaw cut surface*
- *Ground with 600 grit sand paper*



# ASTM G85-A2: Sample ID

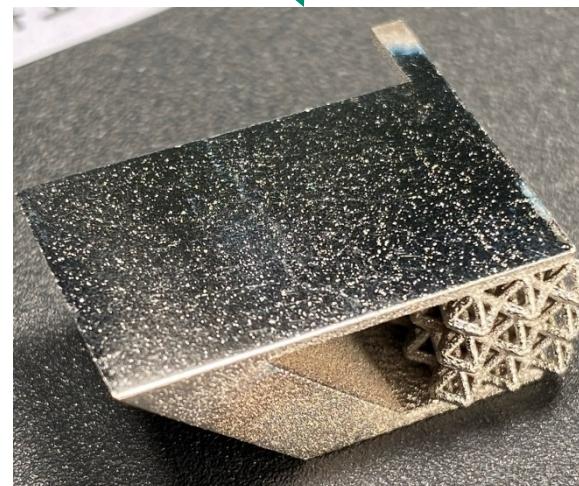
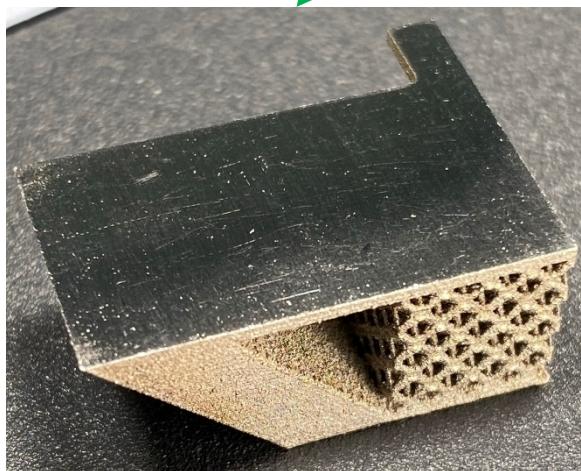


As-cut Bandsaw

As-cut wire EDM

Electropolishing

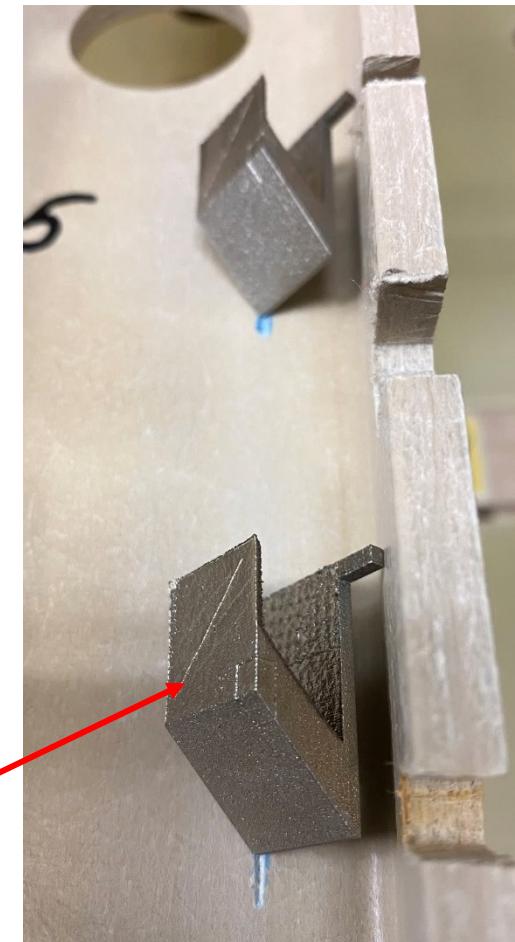
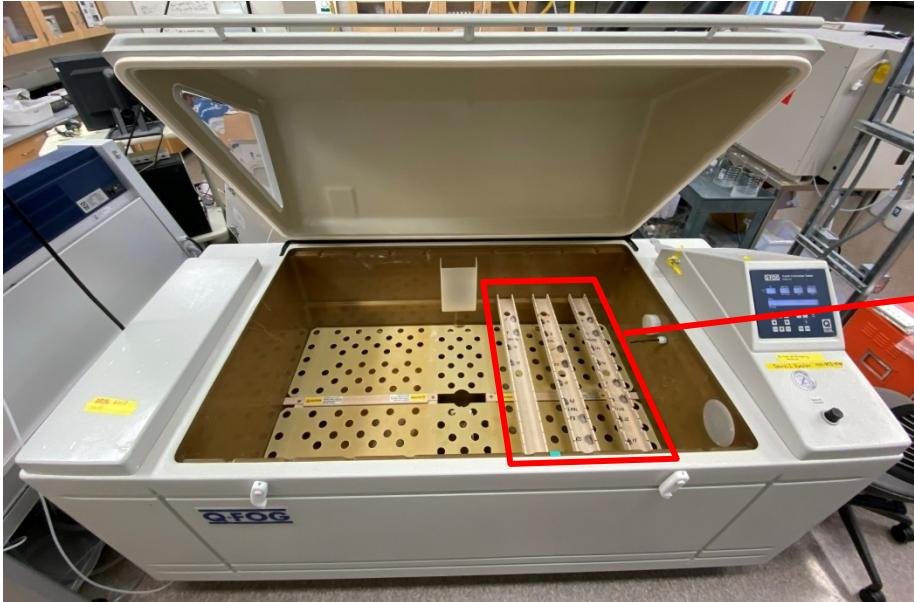
Planar grinding



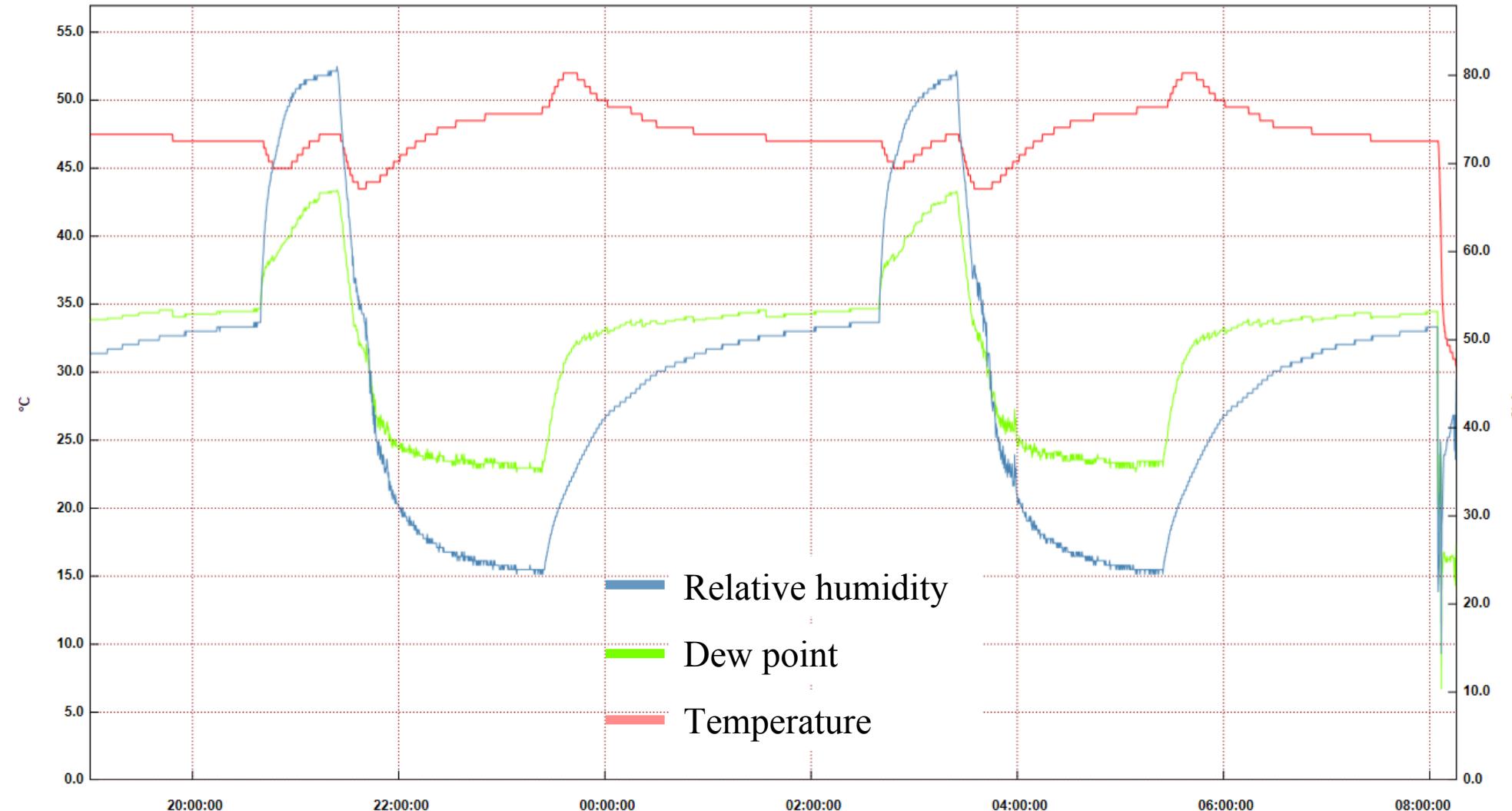
# ASTM G85-A2: Setup



- **ASTM for sample testing: G85 A2 with wet bottom**
  - Equipment: Q-fog cyclic corrosion tester.
  - Solution: 0.9M NaCl acidified to pH 3 with acetic acid.
  - Test temperature: 49°C.
  - Duration: 2000 hr.
  - Sample holder: PEEK rack with milled paths to place AM 316 samples.



## ASTM G85-A2: RH and Temp.



# Samples after 2000 hours – Top down



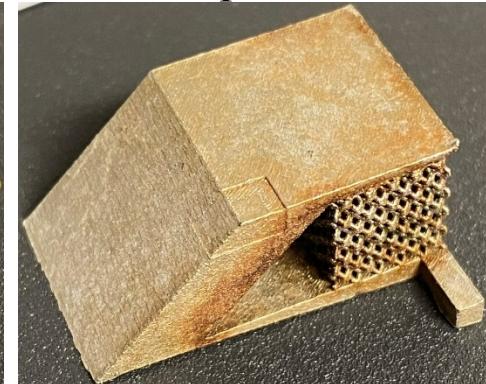
ProX 200  
As-printed



ProX 200  
Electropolish



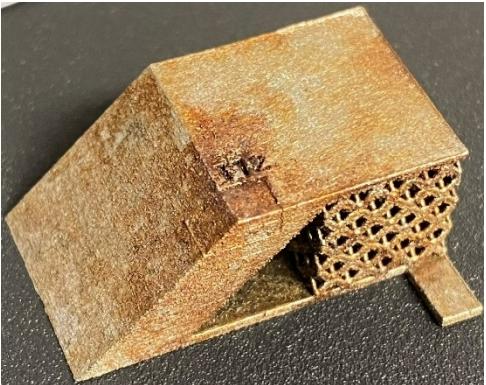
EOS M290  
As-printed



Wrought surface  
was polished



Renishaw  
As-printed



Concept M2  
As-printed



SLM®280  
As-printed



5 mm

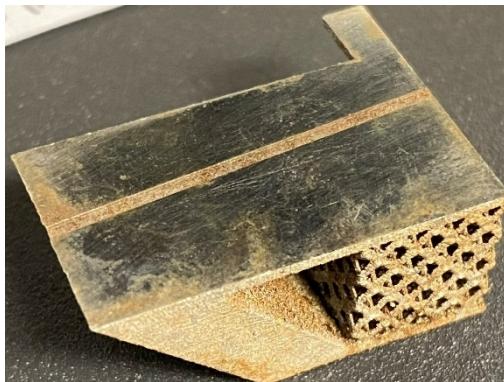
# Samples after 2000 hours – Cut surface



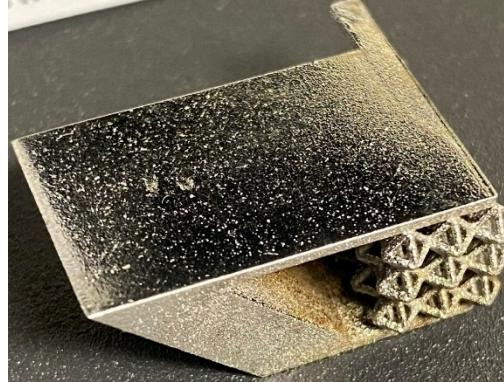
ProX 200  
As-cut wire EDM



Renishaw  
Ground to 600 grit



ProX 200  
Electropolish



Concept M2  
As-cut bandsaw



EOS M290  
As-cut bandsaw



SLM®280  
As-cut wire EDM

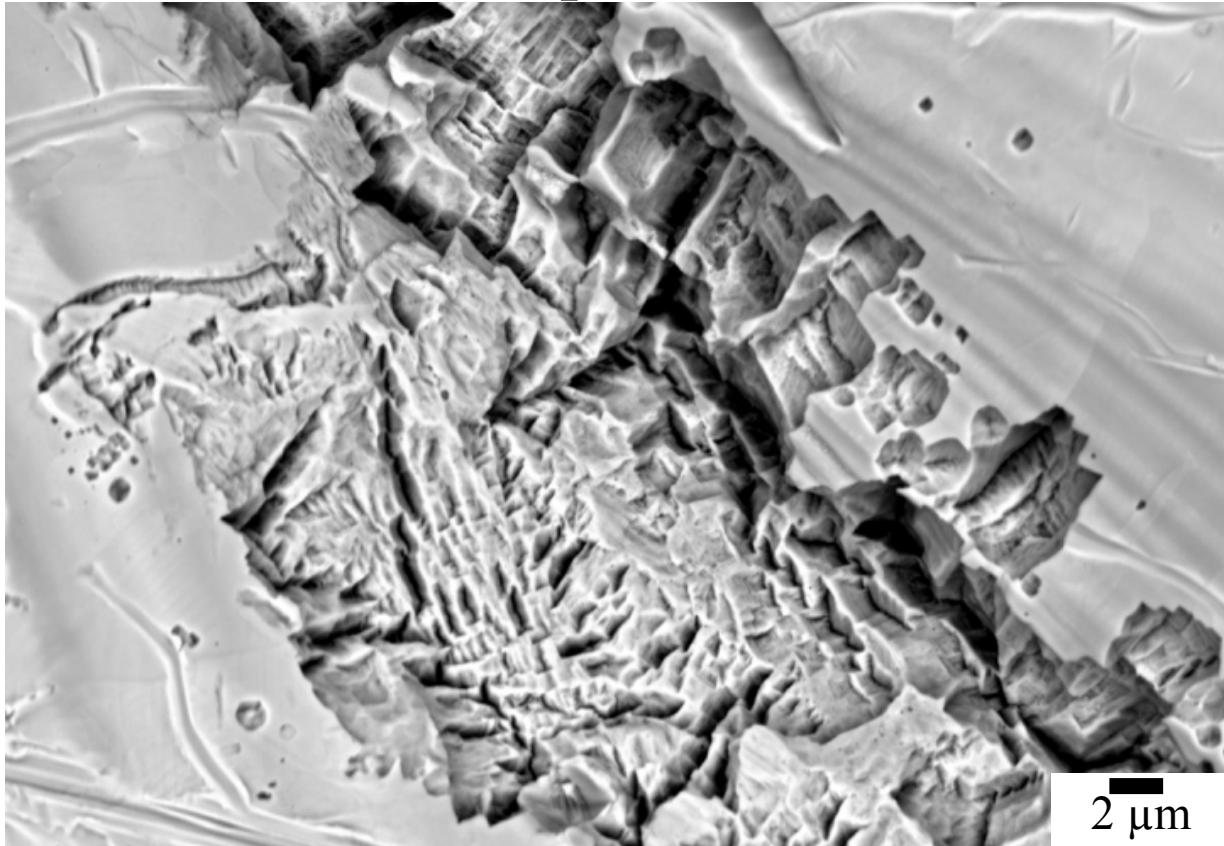


Wrought  
Ground to 600 grit

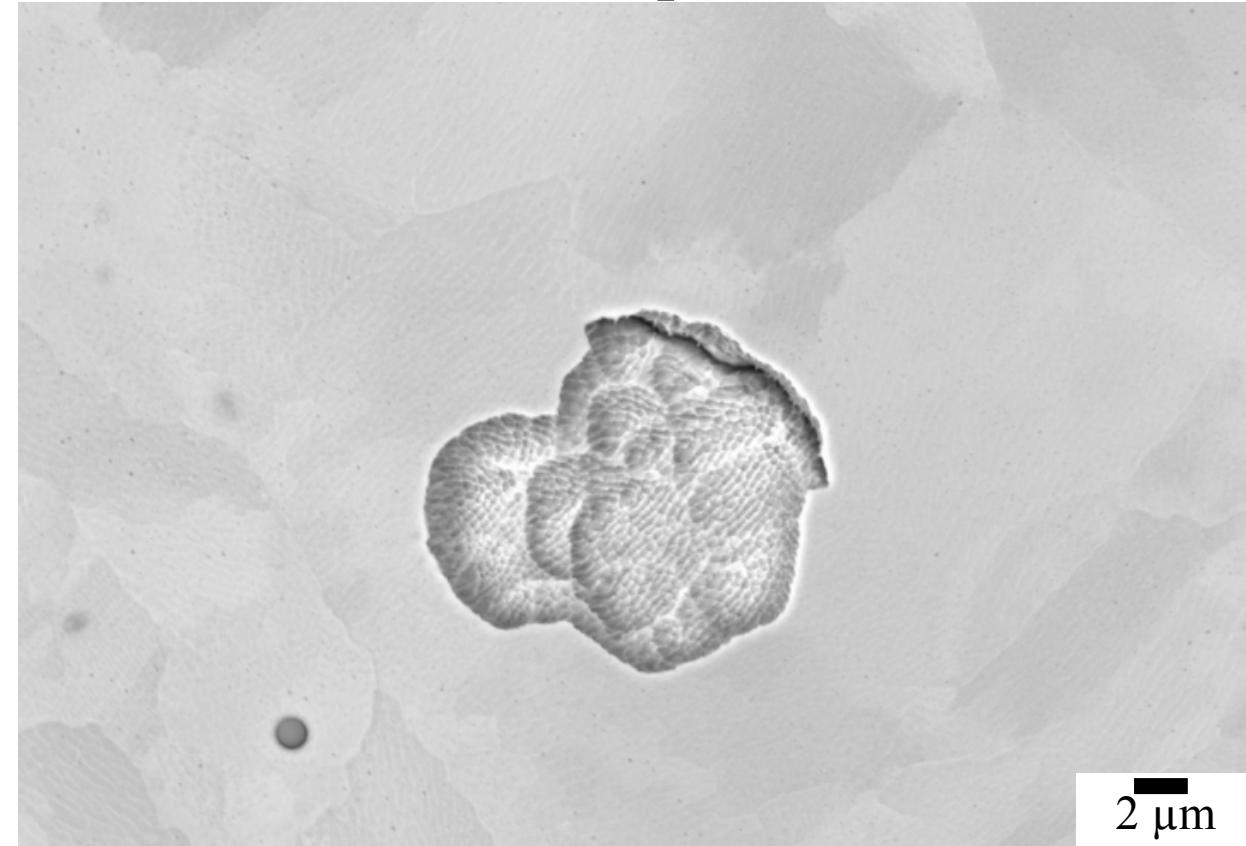


# Pit morphology – Top surface

ProX 200  
As-printed



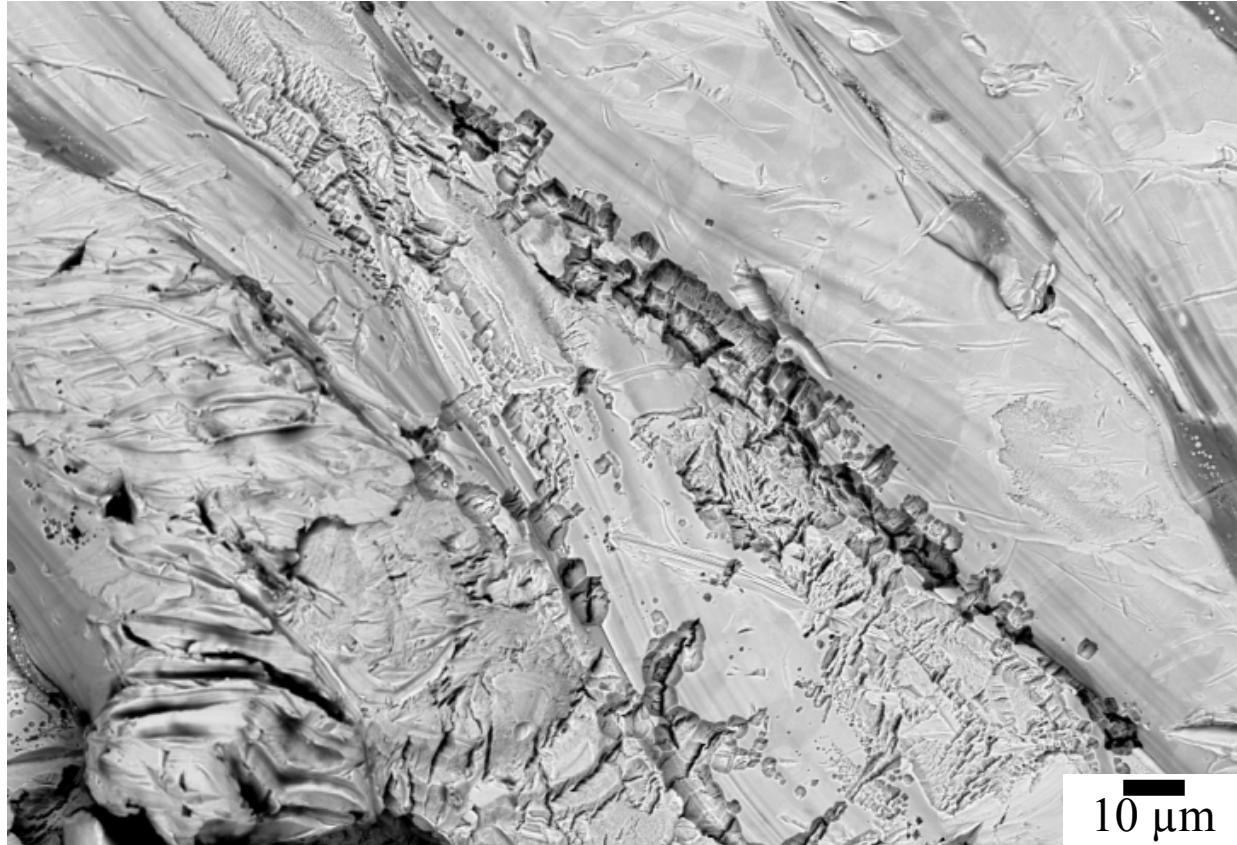
ProX 200  
Electropolish



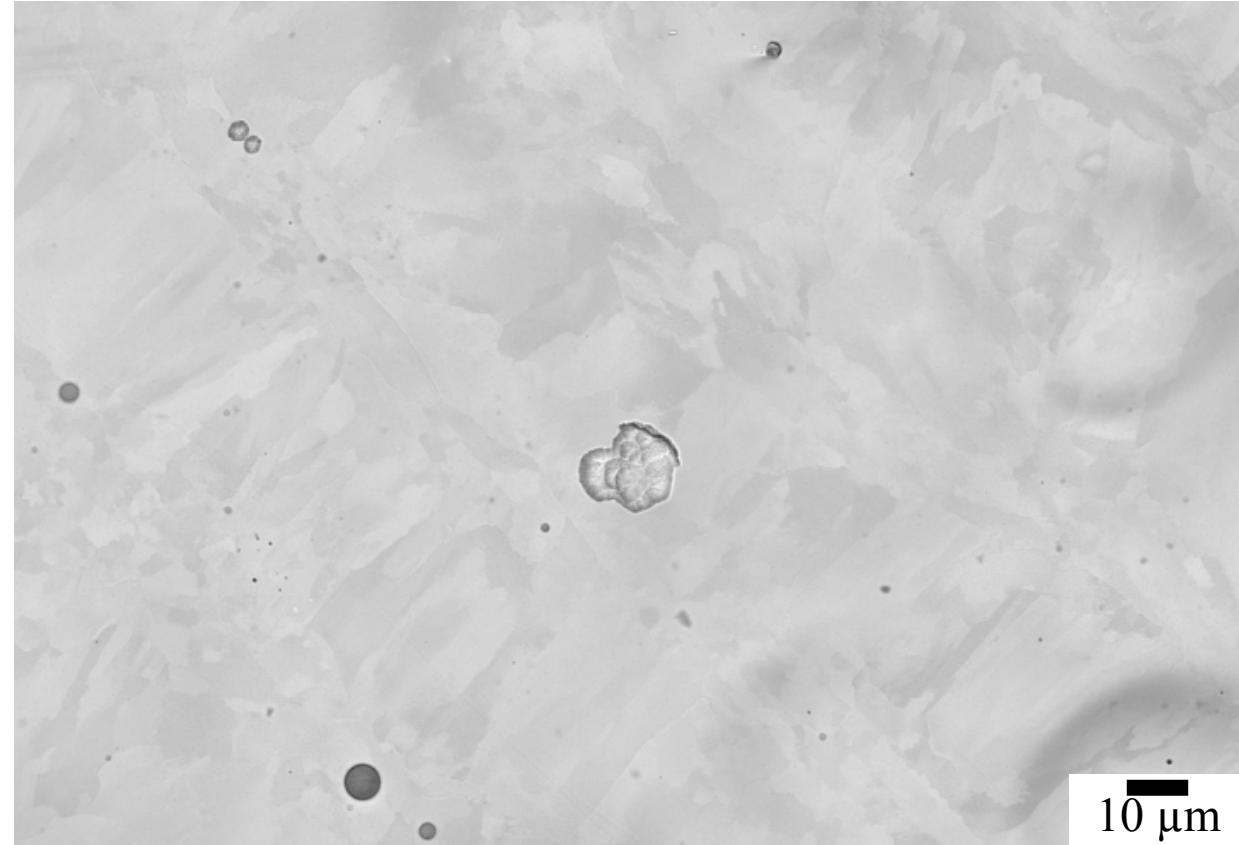
# Pit morphology – Top surface



ProX 200  
As-printed

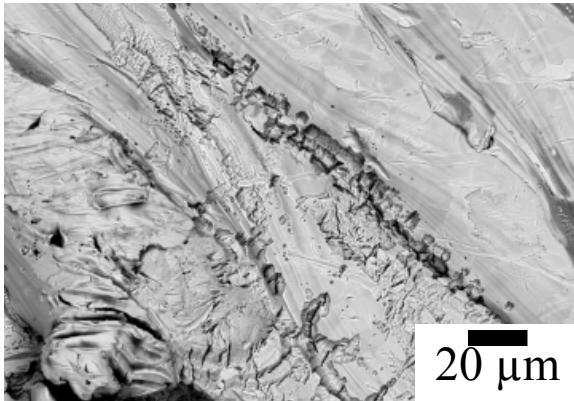


ProX 200  
Electropolish

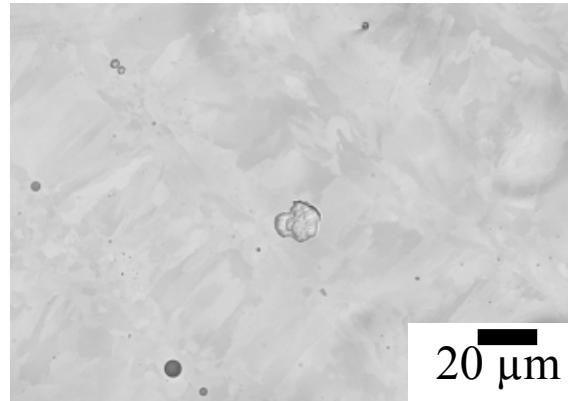


# Pit morphology – Top surface

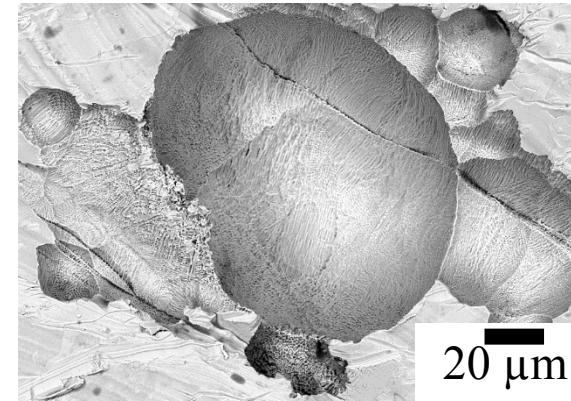
ProX 200  
As-printed



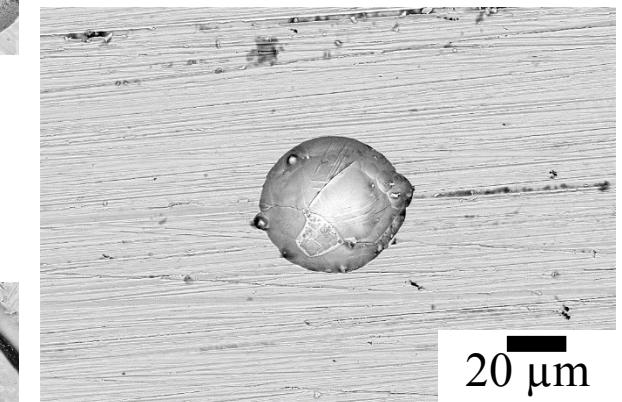
ProX 200  
Electropolish



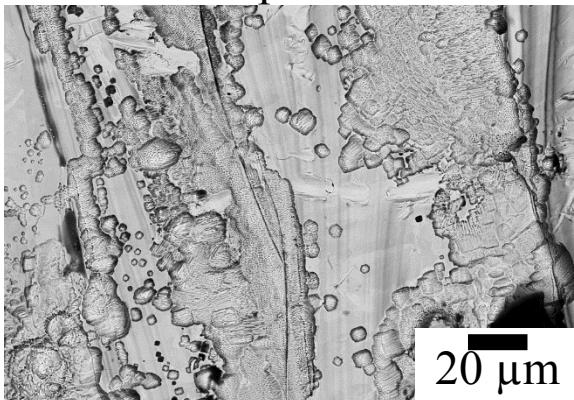
EOS M290  
As-printed



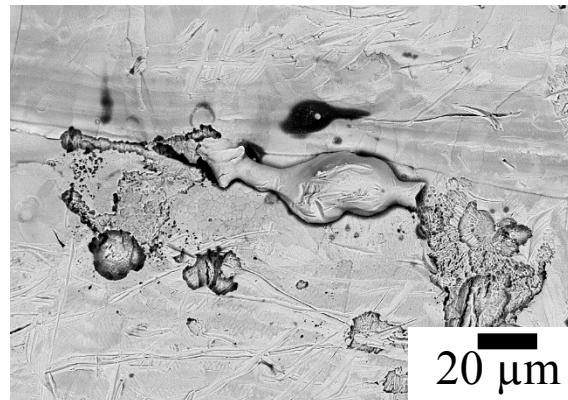
Wrought surface  
was polished



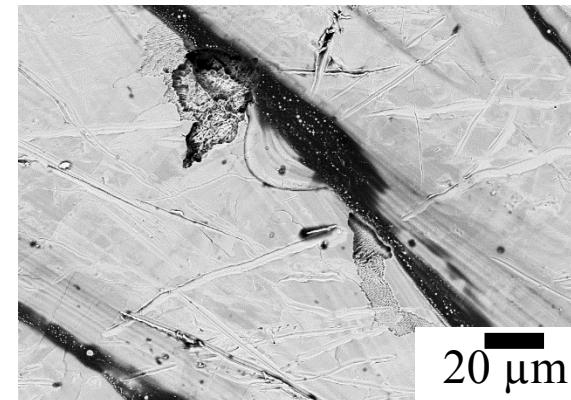
Renishaw  
As-printed



Concept M2  
As-printed



SLM®280  
As-printed



# ASTM G85-A2 conclusions (so far)



- Wire EDM as-cut surface had the most aggressive corrosion based on optical analysis.
- Electropolished surfaces were least susceptible to corrosion.
- As-printed samples had different quantities of corrosion based on optical, but similar morphology (many pits appeared to initiate from melt pool boundaries).

*Current results correlate well with electrochemical experiments (breakdown potential, etc.).*

Thank you for your attention!



Questions...?

Please contact with questions/comments:

Michael Melia

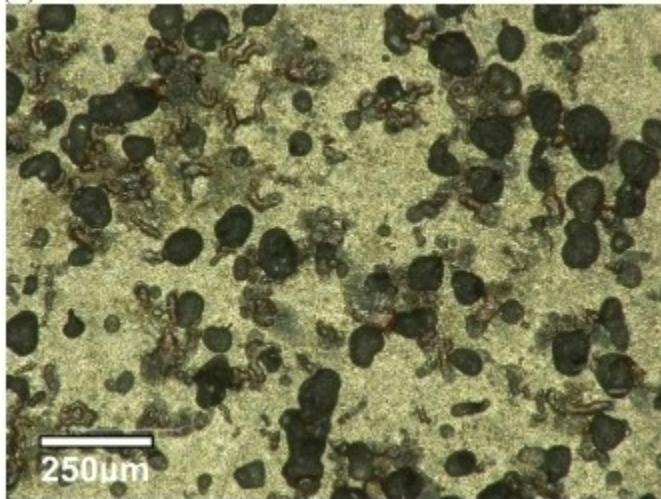
[mamelia@sandia.gov](mailto:mamelia@sandia.gov)

# Background – AM 4340 exposure to ASTM-B117

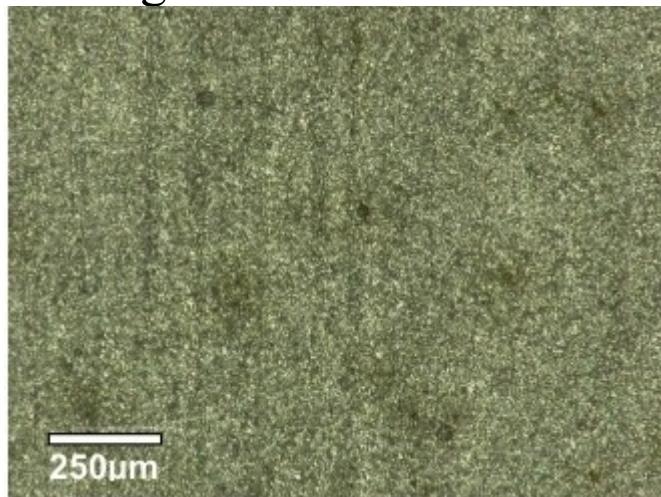


## Images after electrochemical tests

AM4340

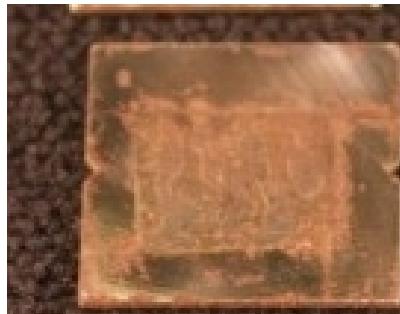


Wrought 4340



## Images after 48 hrs of ASTM B117 exposures

AM4340



Wrought 4340

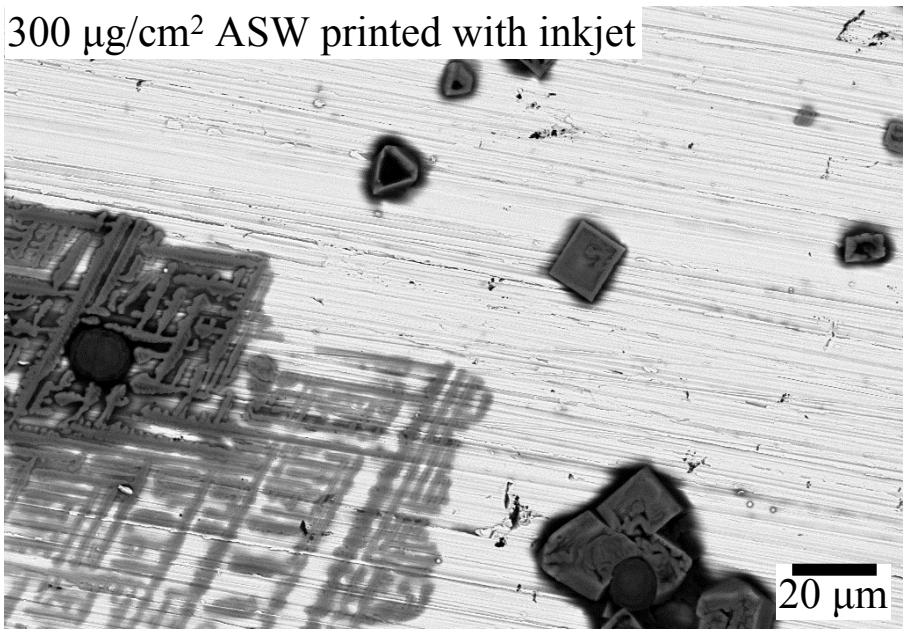


- No noticeable difference in electrochemical response, but surface corrosion morphology was dramatically different.
- Initial ASTM B117 results suggest minimal difference (only analysis was with optical imaging).

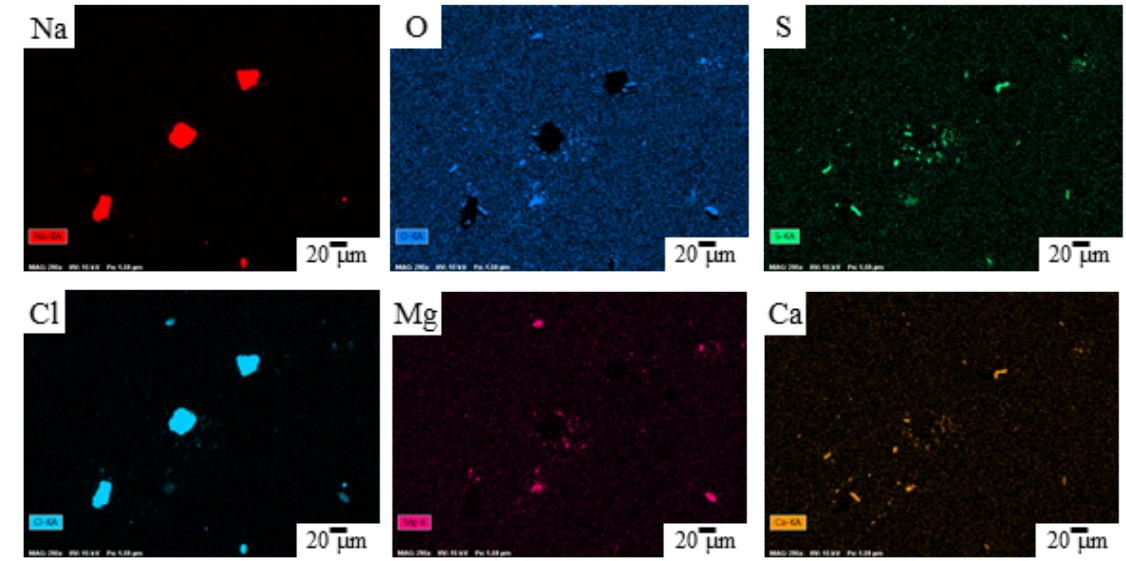
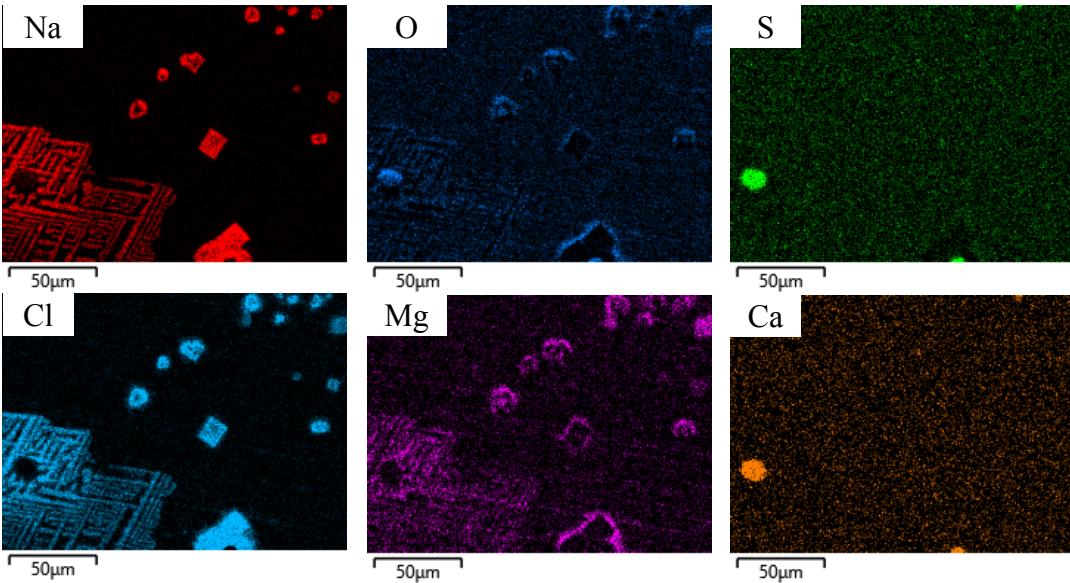
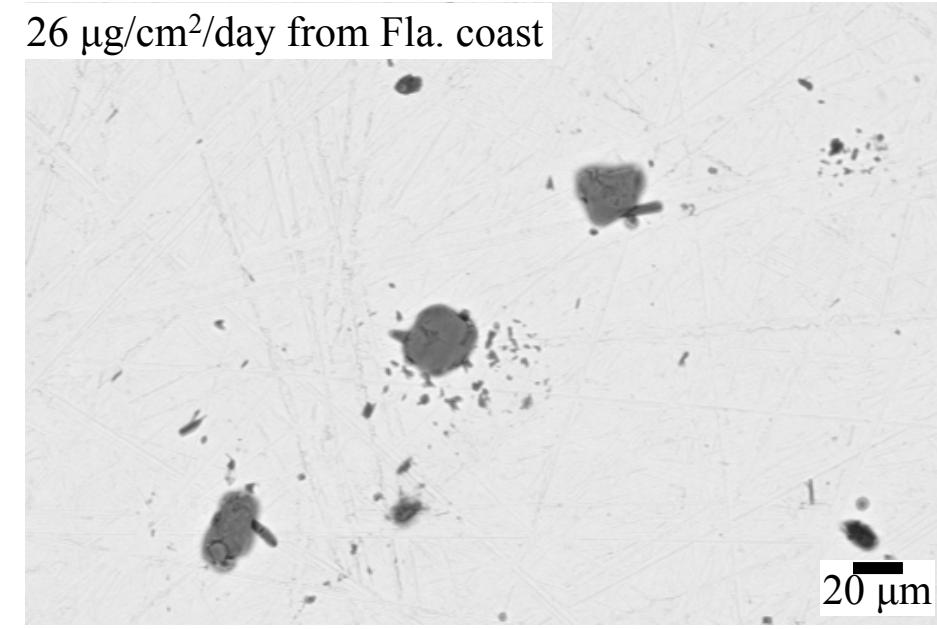
Source: Schmidt, MS&T  
Proceedings, 2015.

# Salt deposited on samples

300  $\mu\text{g}/\text{cm}^2$  ASW printed with inkjet

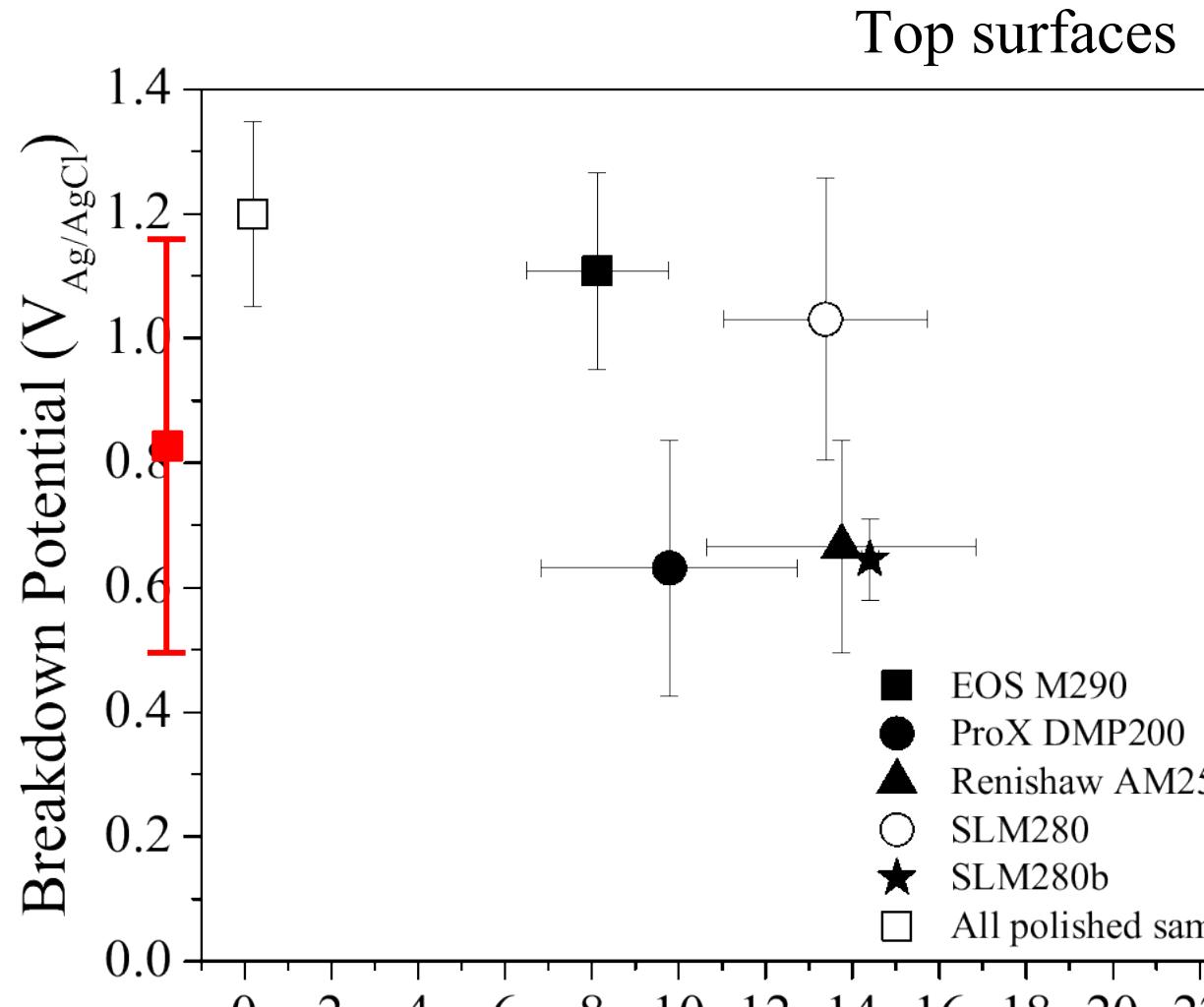


26  $\mu\text{g}/\text{cm}^2/\text{day}$  from Fla. coast



(UUR)

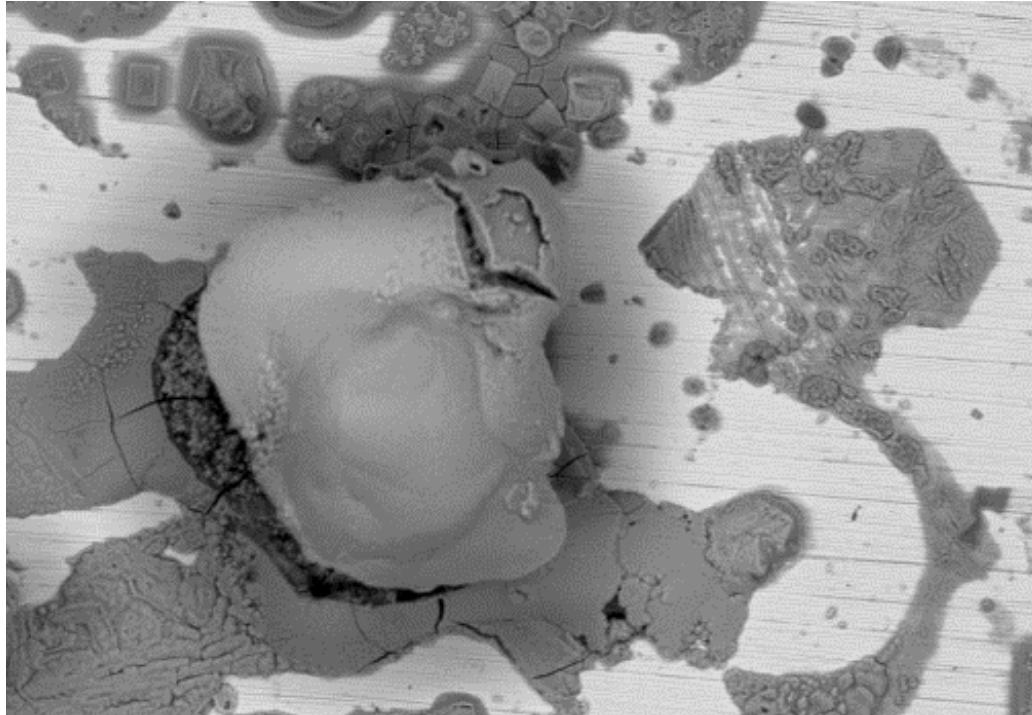
# Summary of data from all sites



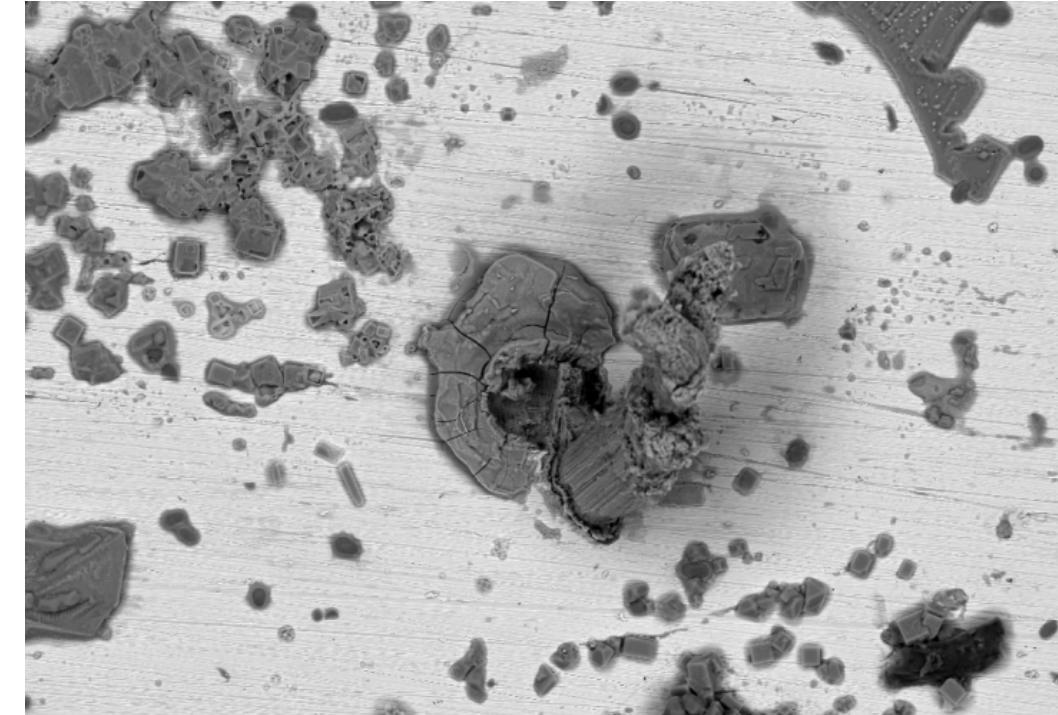
Courtesy of Courtney Clark (LANL)

SEM images of 316L specimen with 300  $\mu\text{g}/\text{cm}^2$  ASW at 40% RH

Wrought



AM

20  $\mu\text{m}$ EHT = 15.00 kV WD = 8.8 mm Signal A = BSD Width = 400.0  $\mu\text{m}$

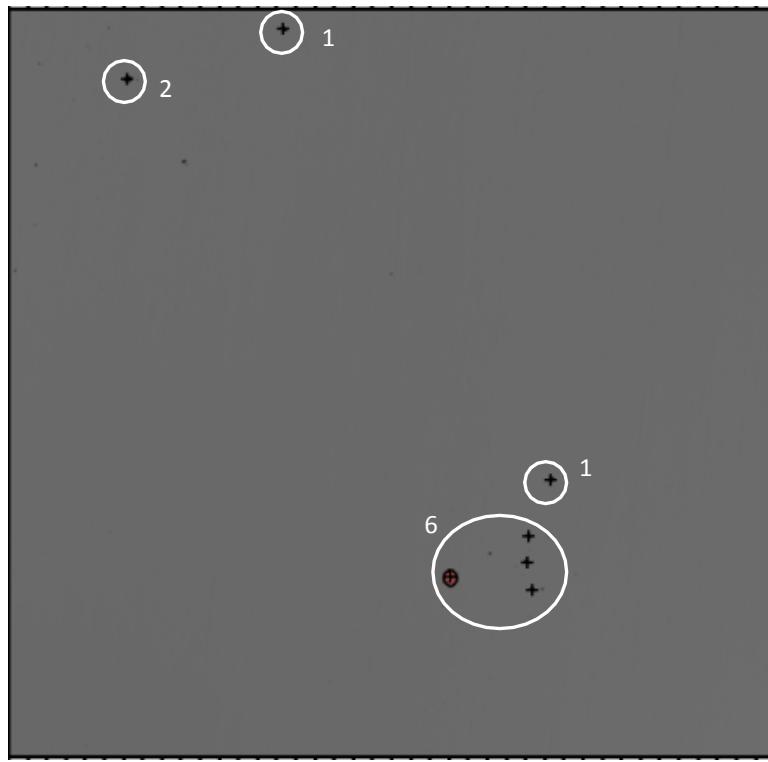
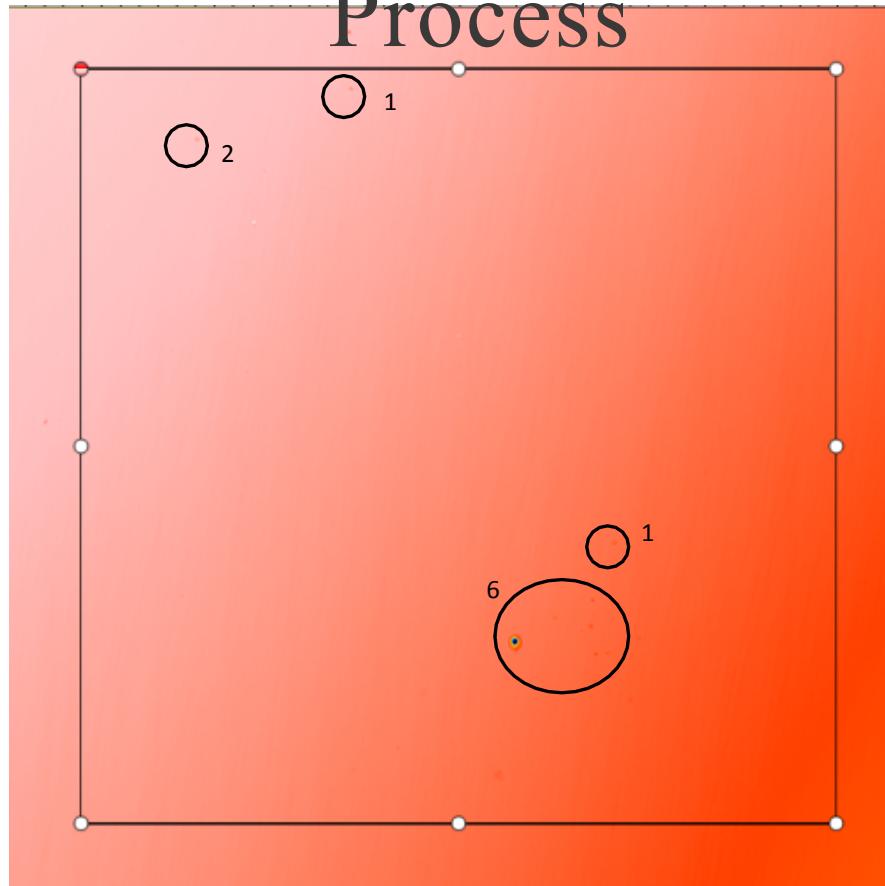
# Pit Verification Process

Raw Data

Processed Data

316L\_W\_35C\_40RH\_300ug  
ASW\_Perp\_1wk-Sample51

Mid Mid



The process for pit verification begins by filtering data taken from a Particle Analysis in MountainsMap. The filtering criteria depends on the Maximum Height (Depth) of a pit exceeding  $0.2 \mu\text{m}$  and the Volume/Area ratio exceeding  $0.2 \mu\text{m}$ .

Once the data is filtered, the remaining particles are verified by comparing the raw data to the processed data. Through visual inspection, the particles are verified to be pits.

The particles that were considered pits through the filtering process are outlined in the above images. The numbers next to the circles indicate the amount of pits located within the circle. For example, the largest circle contains 6 pits. This results in a total of 10 pits.

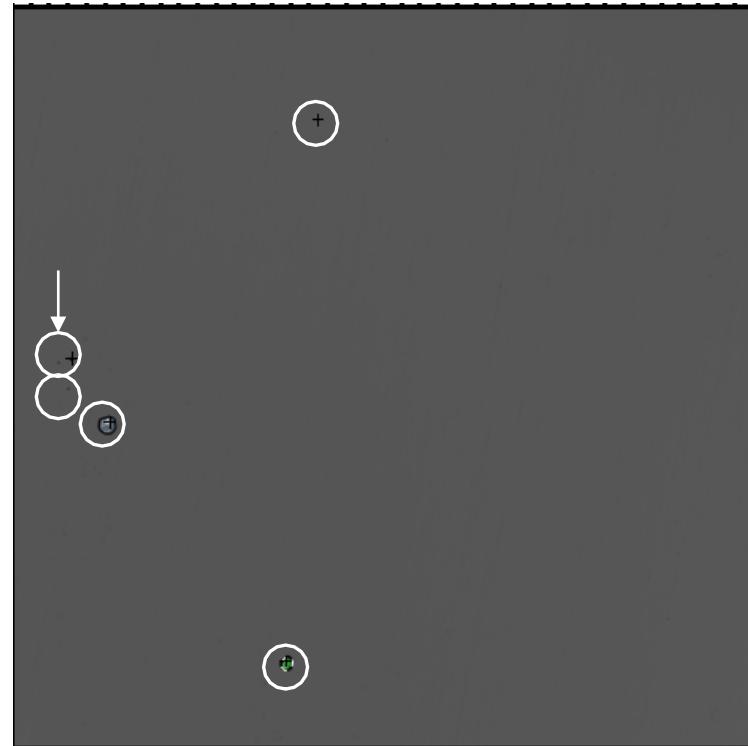
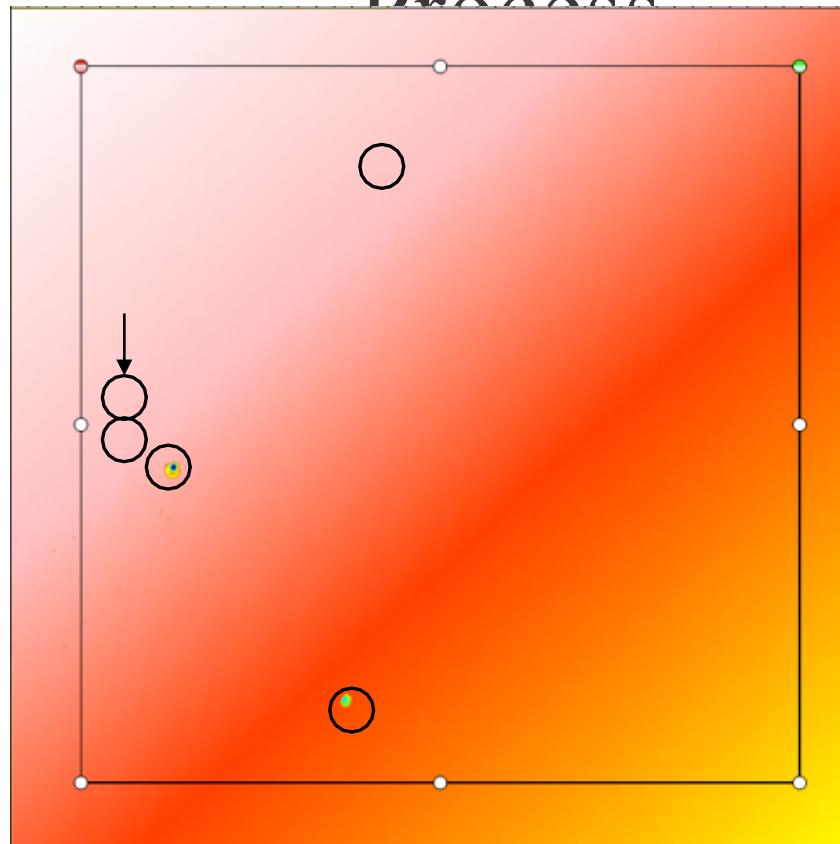
# Pit Verification

Raw Data

Processed Data

316L\_AM\_35C\_40RH\_300  
ASW\_Perp\_1wk-Sample116

Mid Mid



The process for pit verification begins by filtering data taken from a Particle Analysis in MountainsMap. The filtering criteria depends on the Maximum Height (Depth) of a pit exceeding  $0.2 \mu\text{m}$  and the Volume/Area ratio exceeding  $0.2 \mu\text{m}$ .

Once the data is filtered, the remaining particles are verified by comparing the raw data to the processed data. Through visual inspection, the particles are verified to be pits.

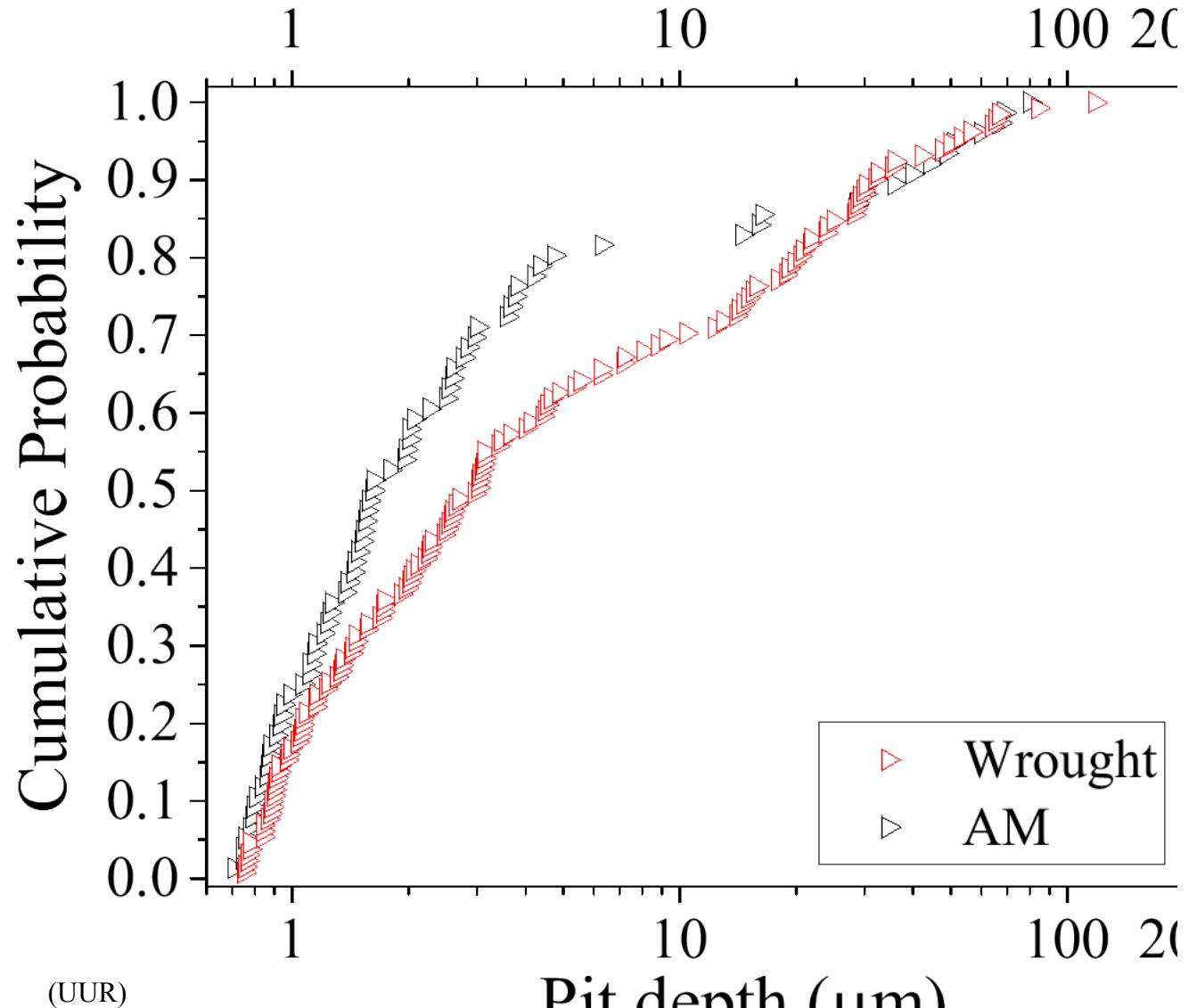
The particles that were considered pits through the filtering process are outlined in the above images. The circle with the arrow pointing to it contains two pits, all others contain 1 pit. This created a total of 6 pits.

**Wrought**

	Average pit depth ( $\mu\text{m}$ )	Max pit depth ( $\mu\text{m}$ )
1 week	2.9	35.1
1 month	7.0	41.8
6 months	6.3	58.5
12 months	11.7	118.1

**AM**

	Average pit depth ( $\mu\text{m}$ )	Max pit depth ( $\mu\text{m}$ )
1 week	4.0	32.6
1 month	10.1	210.4 (73.3)
6 months	7.6	76.2
12 months	9.4	80.2

**12 months – Pit depth**

# Pit statistics of 316L specimen - 300 $\mu\text{g}/\text{cm}^2$ ASW at 40% RH



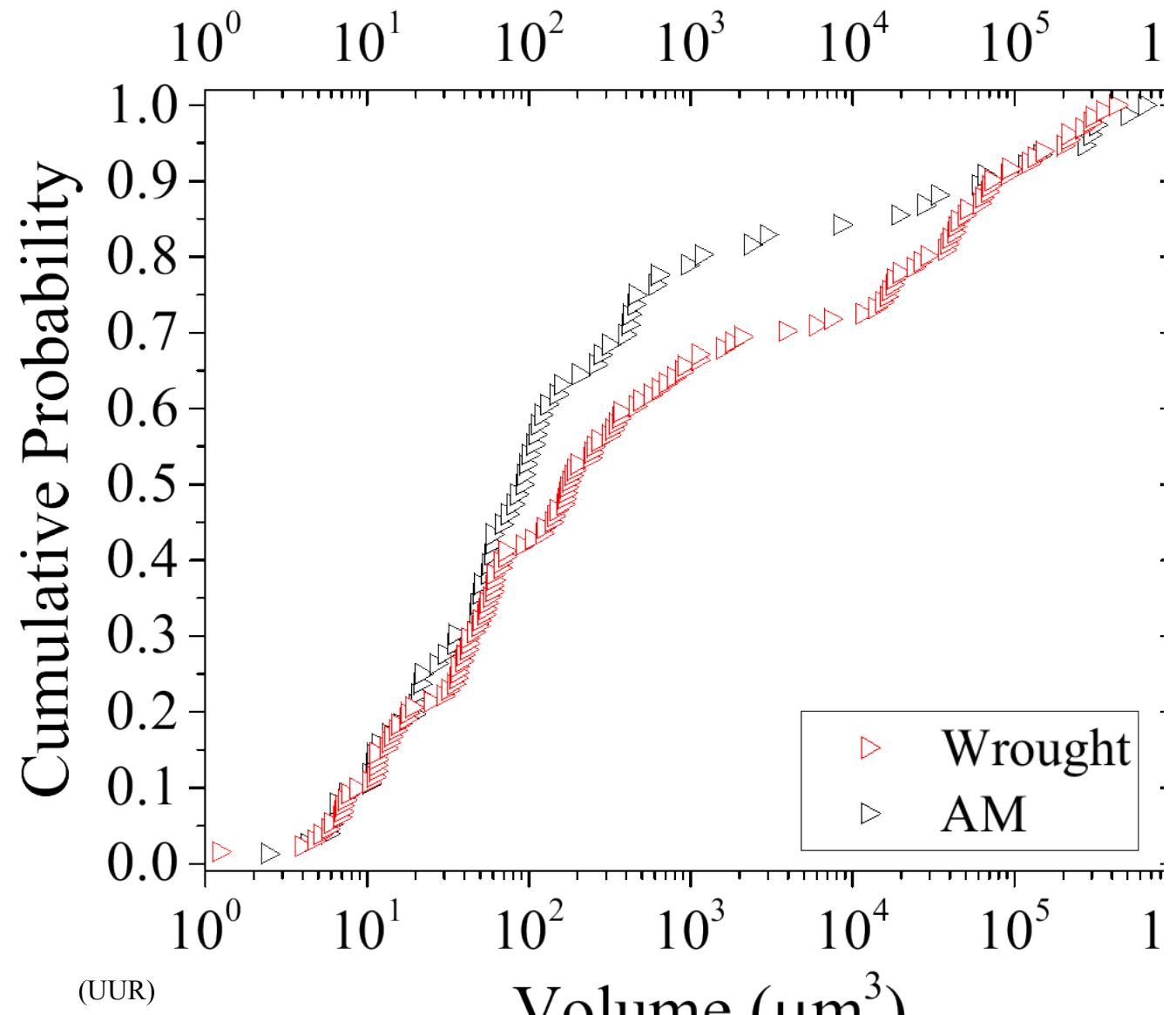
## Wrought

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$1.15 \times 10^5$	50
1 month	$9.28 \times 10^5$	65
6 months	$2.34 \times 10^6$	105
12 months	$2.75 \times 10^6$	96

## AM

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$3.66 \times 10^5$	89
1 month	$9.35 \times 10^5$	88
6 months	$2.97 \times 10^6$	155
12 months	$1.74 \times 10^6$	53

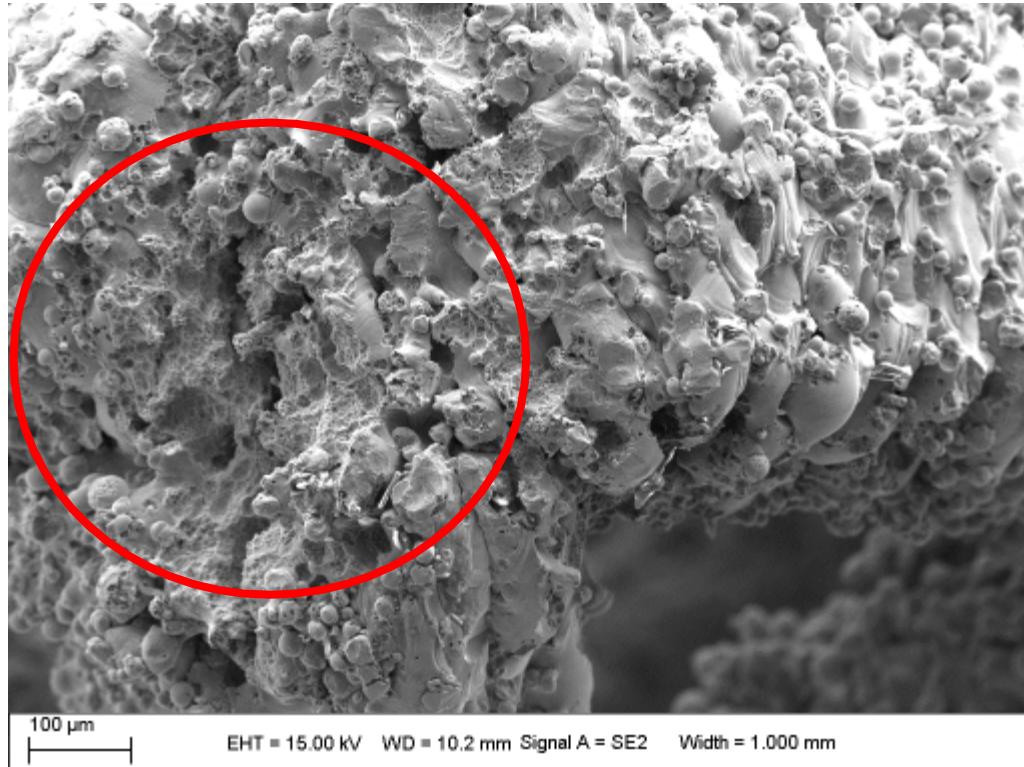
## 12 months – Pit volume



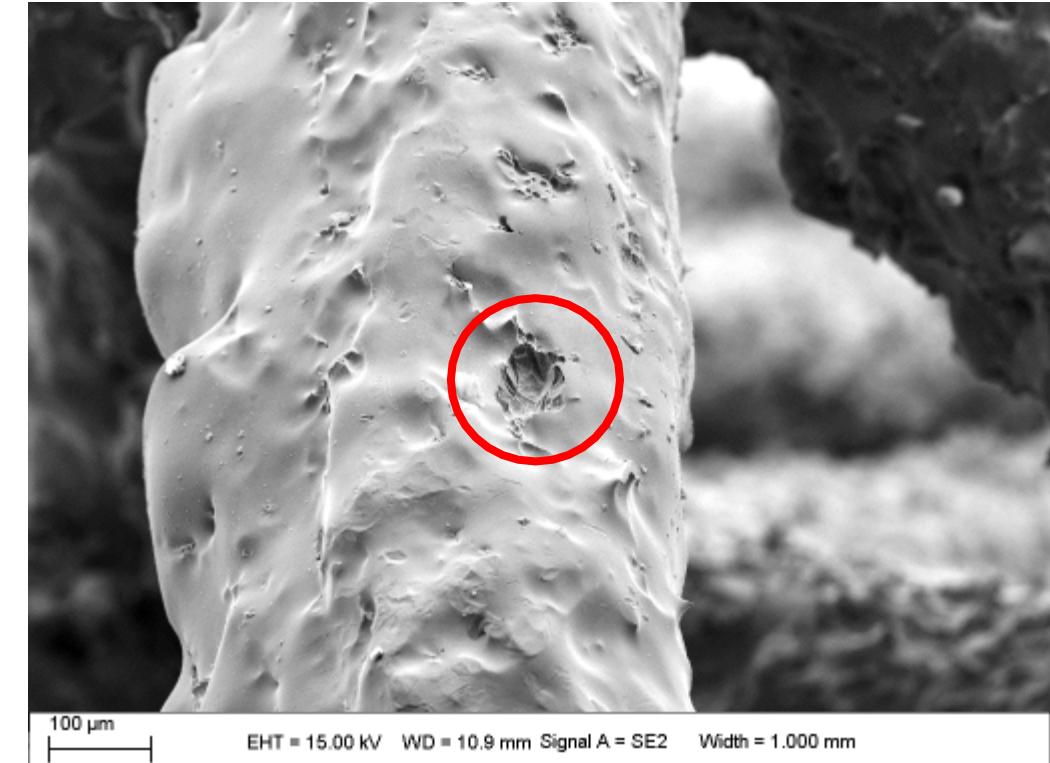
# Pit morphology - Lattices



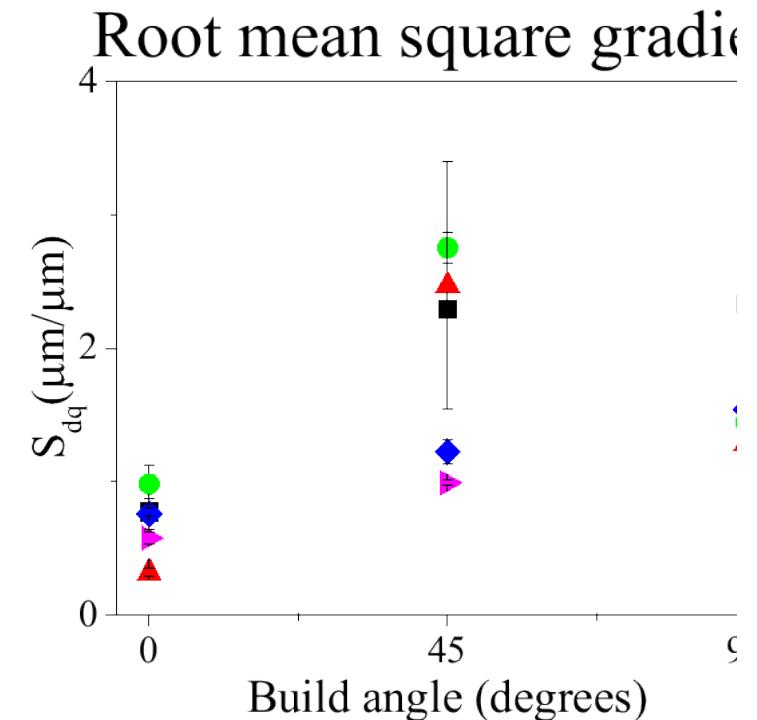
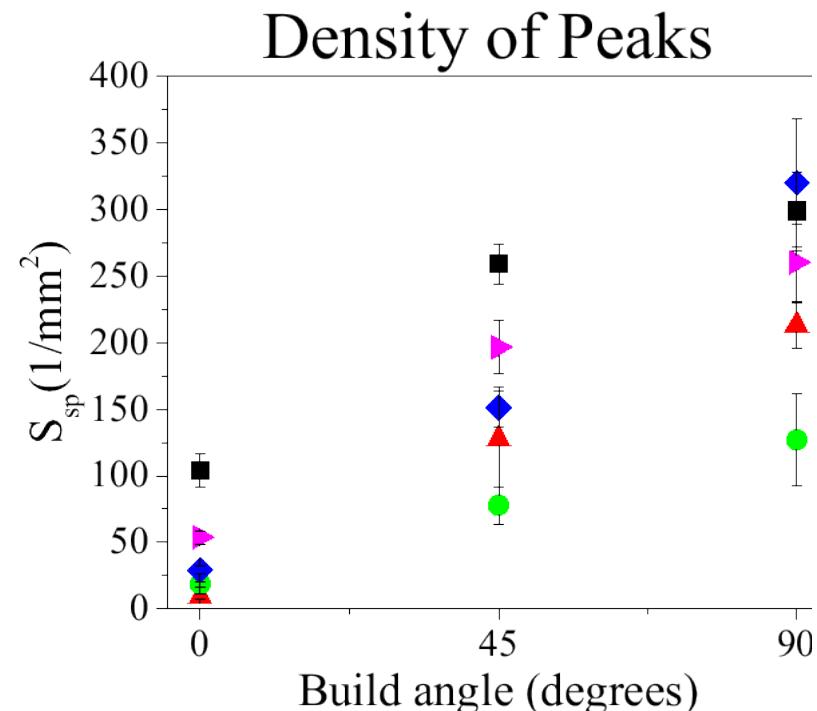
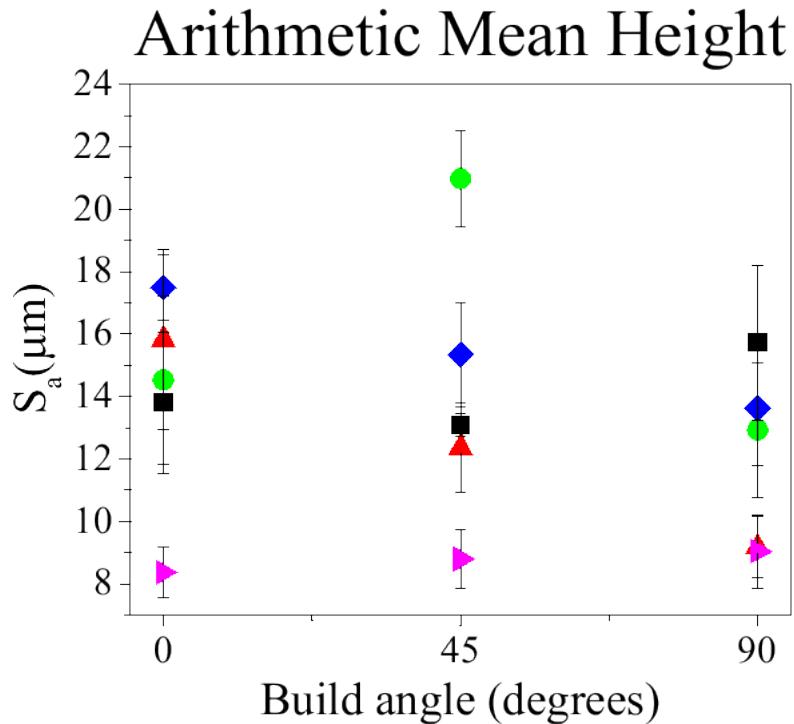
ProX 200  
As-printed



ProX 200  
Electropolish



# <sup>49</sup> Zygo measurements from all material

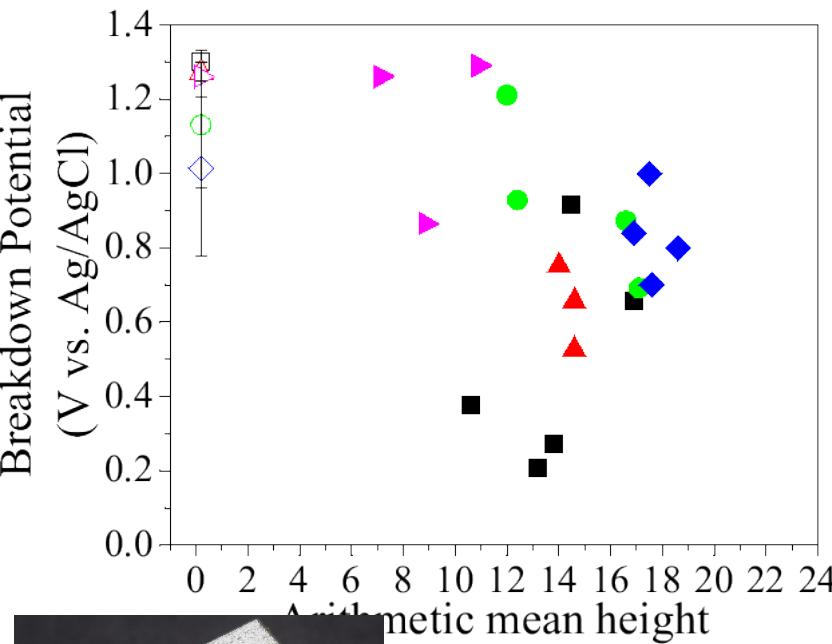


- ▲ SLM 280b
- SLM 280
- ▶ EOS M290
- ProX DMP200
- ◆ Renishaw AM250

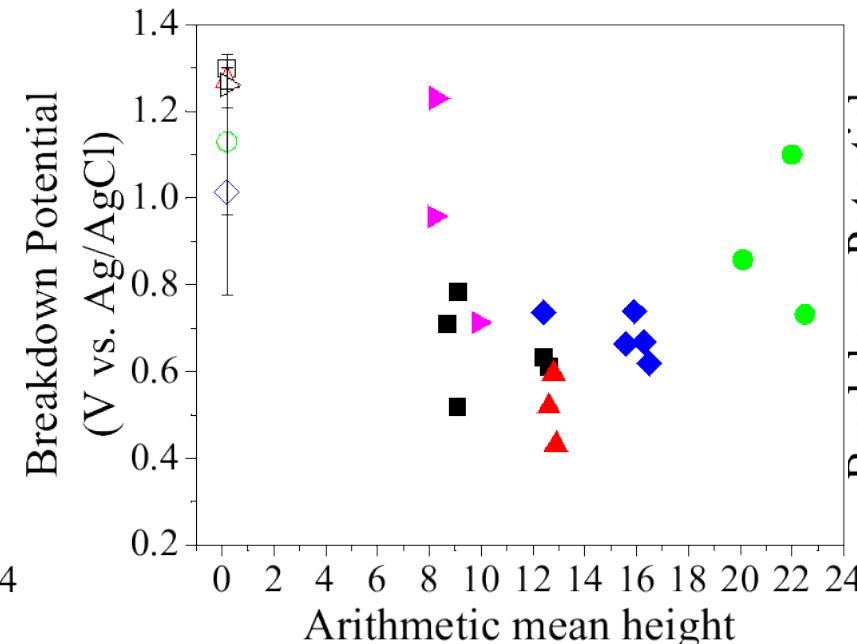
- Without any filtering (but some data processing),  $S_{pd}$  shows a linear trend with respect to build angle for all printed machines.
- The  $S_a$  values are generally all dissimilar when comparing machine to machine (influence from waviness).

# Summary of Breakdown potential ( $E_B$ )

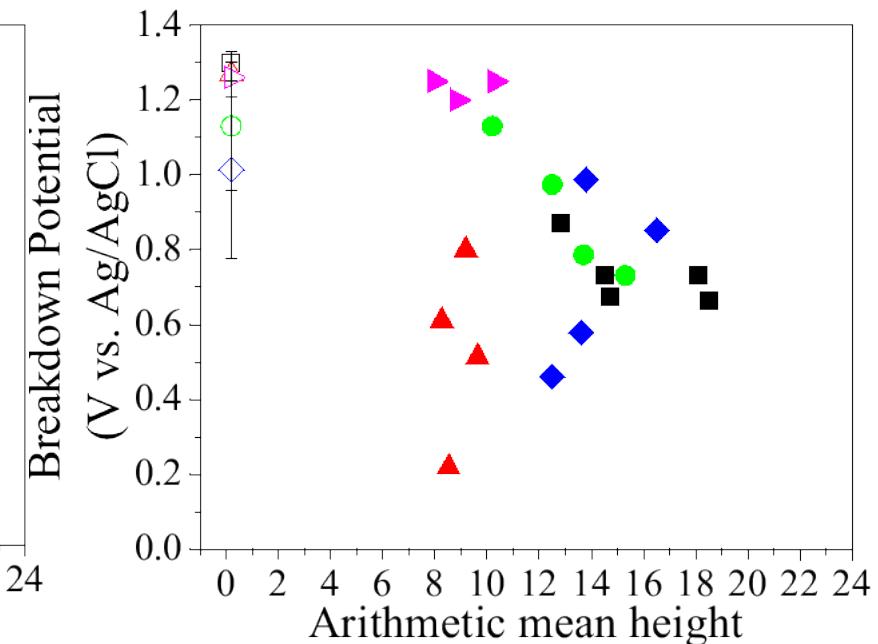
Top Surfaces



Upskin Surfaces



Side Surfaces



- ProX DMP200  $S_a$  ( $\mu\text{m}$ )
- ▲ SLM280b
- SLM280
- ◆ Renishaw AM250
- ▶ EOS M290

# Pit statistics of 316L specimen - 300 $\mu\text{g}/\text{cm}^2$ ASW at 40% RH



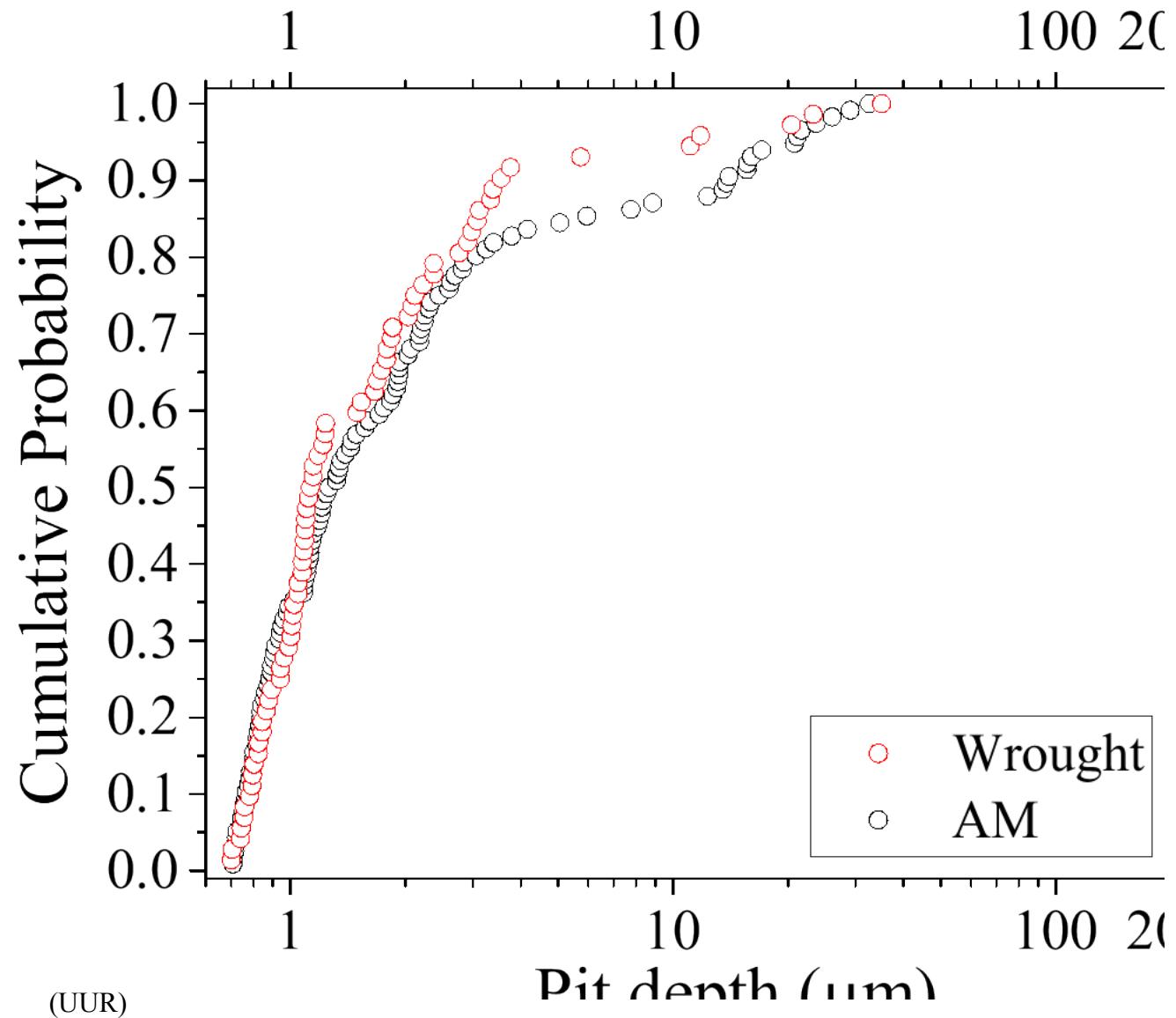
## Wrought

	Average pit depth ( $\mu\text{m}$ )	Max pit depth ( $\mu\text{m}$ )
1 week	2.9	35.1
1 month	7.0	41.8
6 months	6.3	58.5
12 months	11.7	118.1

## AM

	Average pit depth ( $\mu\text{m}$ )	Max pit depth ( $\mu\text{m}$ )
1 week	4.0	32.6
1 month	10.1	210.4 (73.3)
6 months	7.6	76.2
12 months	9.4	80.2

## 1 week – Pit depth

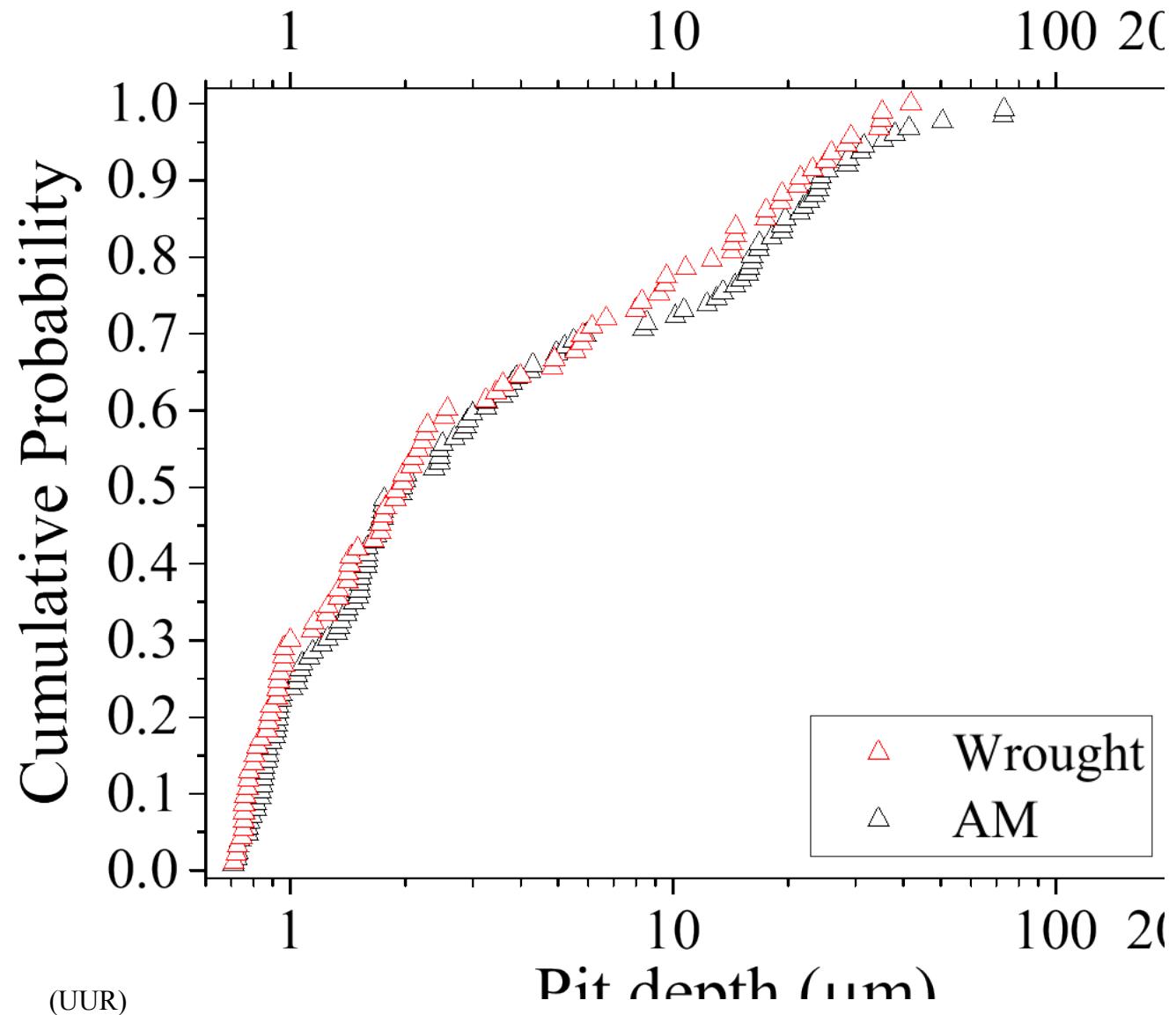


**Wrought**

	Average pit depth ( $\mu\text{m}$ )	Max pit depth ( $\mu\text{m}$ )
1 week	2.9	35.1
1 month	7.0	41.8
6 months	6.3	58.5
12 months	11.7	118.1

**AM**

	Average pit depth ( $\mu\text{m}$ )	Max pit depth ( $\mu\text{m}$ )
1 week	4.0	32.6
1 month	10.1	210.4 (73.3)
6 months	7.6	76.2
12 months	9.4	80.2

**1 month – Pit depth**

# Pit statistics of 316L specimen - 300 $\mu\text{g}/\text{cm}^2$ ASW at 40% RH



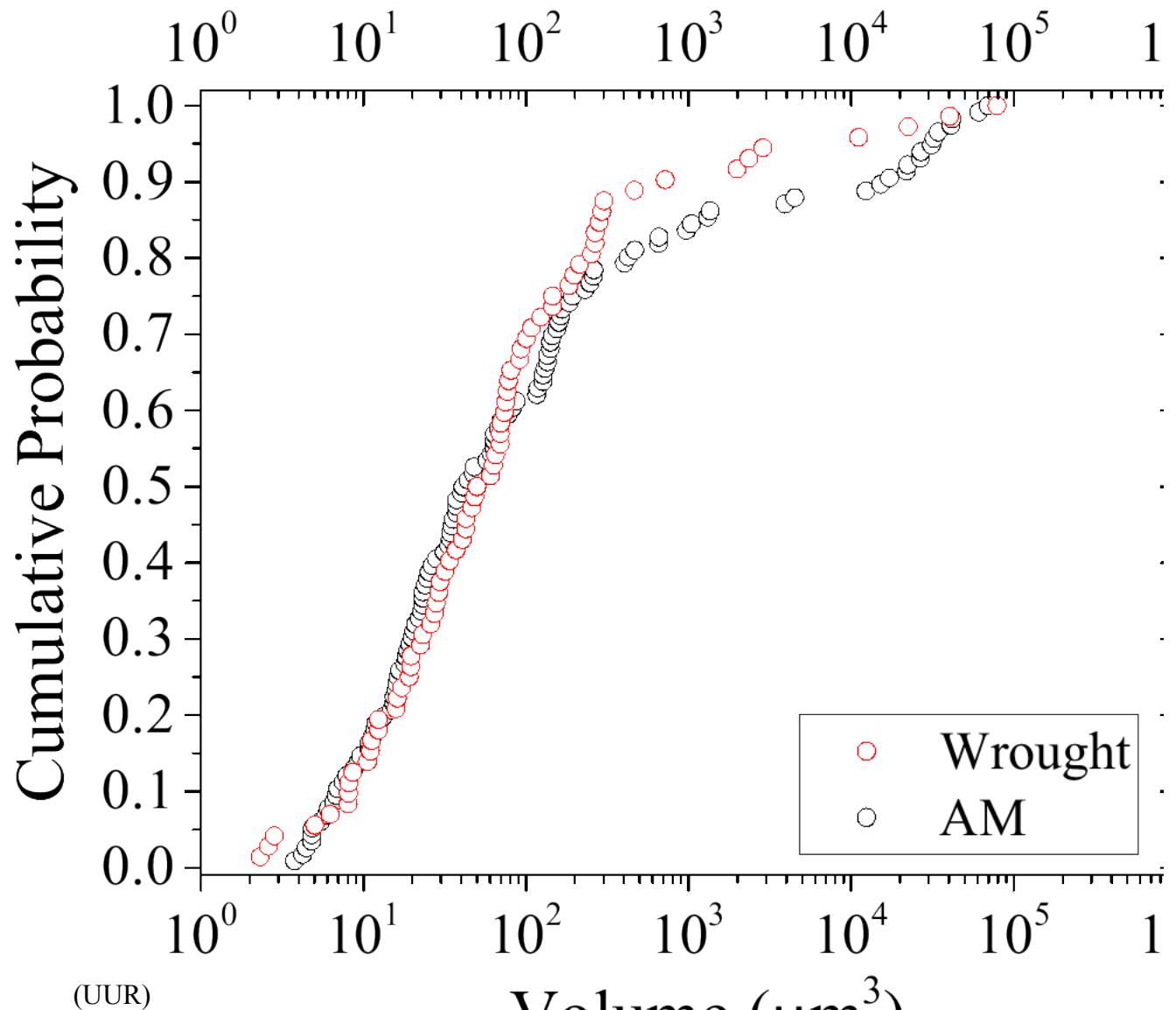
## Wrought

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$1.15 \times 10^5$	50
1 month	$9.28 \times 10^5$	65
6 months	$2.34 \times 10^6$	105
12 months	$2.75 \times 10^6$	96

## AM

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$3.66 \times 10^5$	89
1 month	$9.35 \times 10^5$	88
6 months	$2.97 \times 10^6$	155
12 months	$1.74 \times 10^6$	53

## 1 week – Pit volume

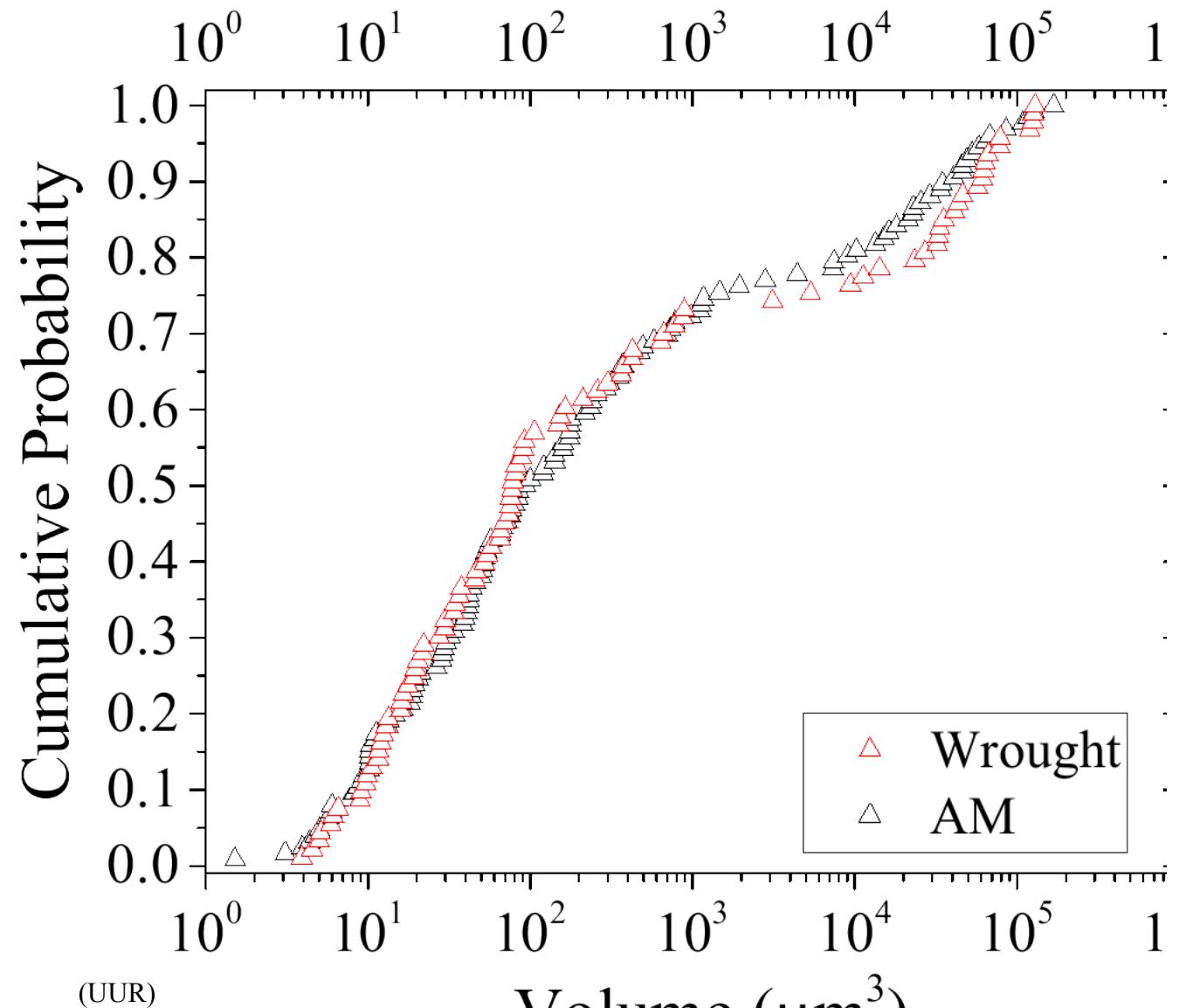


**Wrought**

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$1.15 \times 10^5$	50
1 month	$9.28 \times 10^5$	65
6 months	$2.34 \times 10^6$	105
12 months	$2.75 \times 10^6$	96

**AM**

	Volume per area ( $\mu\text{m}^3/\text{cm}^2$ )	# of Pits/ $\text{cm}^2$
1 week	$3.66 \times 10^5$	89
1 month	$9.35 \times 10^5$	88
6 months	$2.97 \times 10^6$	155
12 months	$1.74 \times 10^6$	53

**1 month – Pit volume**

Pit statistics of 316L specimen - 300  $\mu\text{g}/\text{cm}^2$  ASW at 40% RH