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A System-Level Approach to Li Metal/Iron Fluoride Batteries Enabled by Optimal Electrolyte Choice

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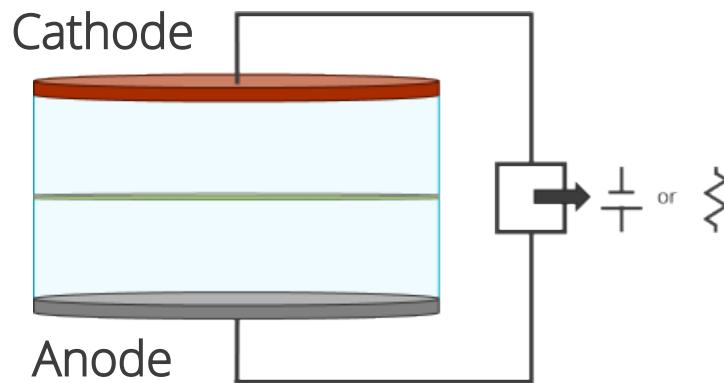
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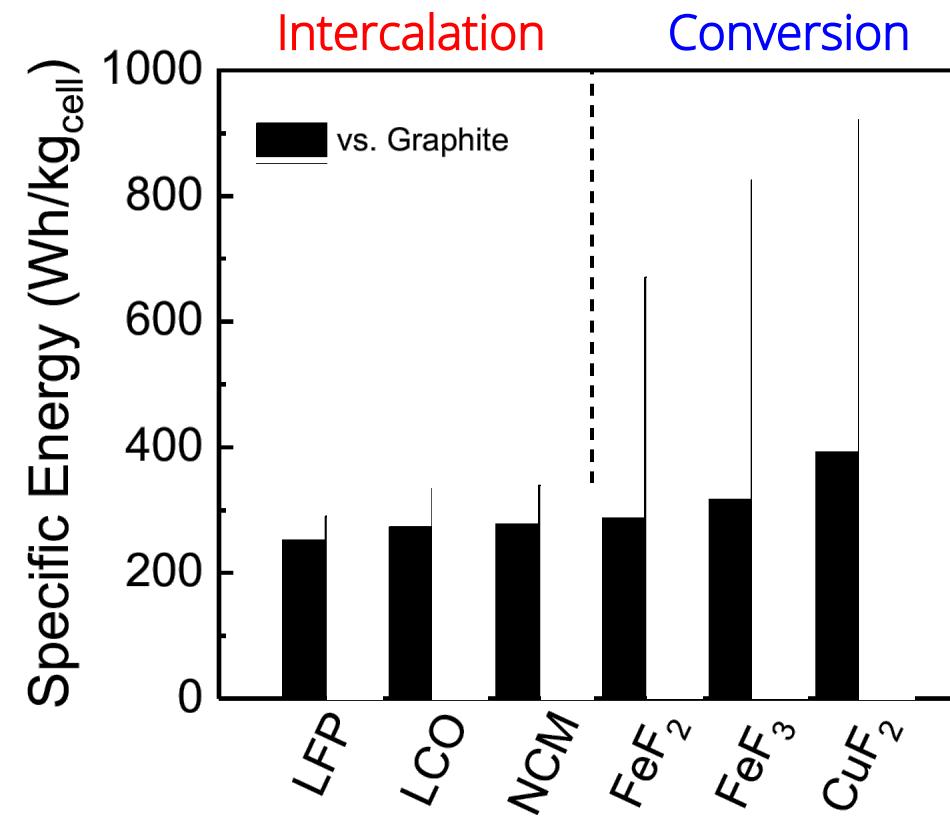


Pairing of high-capacity conversion cathodes with common anodes

- Choices of cathode **and** anode matter when building/testing a full cell battery
 - Conversion cathodes have higher capacities than intercalation cathodes
 - Need to balance energy density and system compatibility

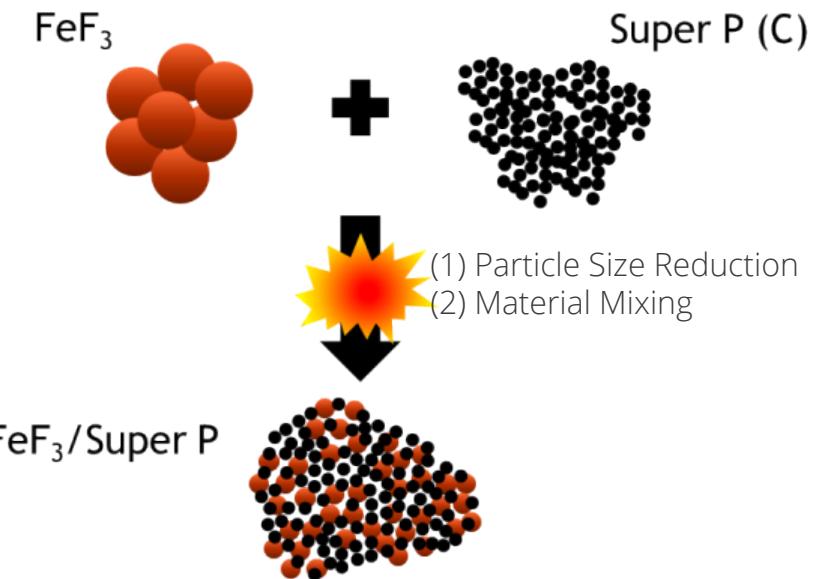


- Cathodes and Anodes are typically treated and optimized separately in studies
 - Is what's good for one electrode good for the other?



Compatibility of FeF_3 /Li full cell batteries with electrolyte systems

- What is the most compatible electrolyte for a FeF_3 /Li full cell?
 - Using Li metal anodes and ball-milled FeF_3 /C cathodes
- Testing 4 electrolytes with known half-cell compatibilities
 - $\text{Pyr}_{13}\text{FSI}$ compatible with FeF_2
 - TTE/DME and Bisalt compatible with Li
- Using electrochemical, chemical, and physical characterization to study each of the electrodes separately and together



$\text{Pyr}_{13}\text{FSI}$ ¹ : 1M LiFSI in *N*-propyl-*N*-methylpyrrolidinium FSI ionic liquid

TTE/DME ² : 1 mol eq. LiFSI in mixed perfluorinated/unfluorinated ethers (3:1.2 mol eq.)

Bisalt ³ : 2M LiFSI + 1M LiTFSI in 1:1(vol) DOL/DME

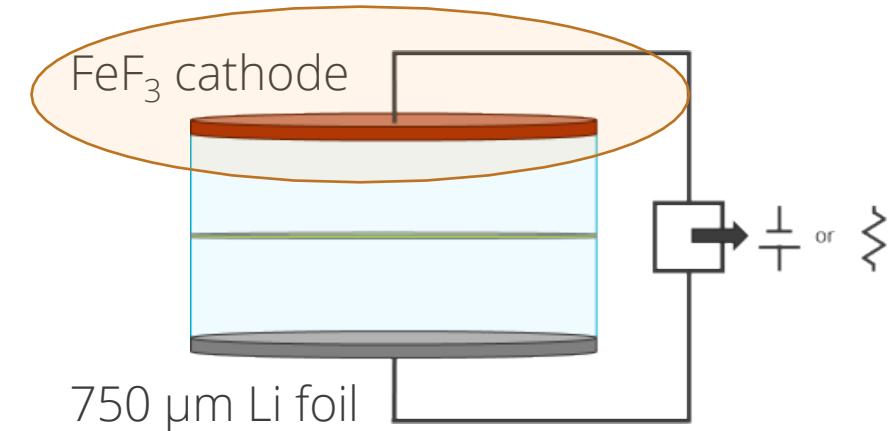
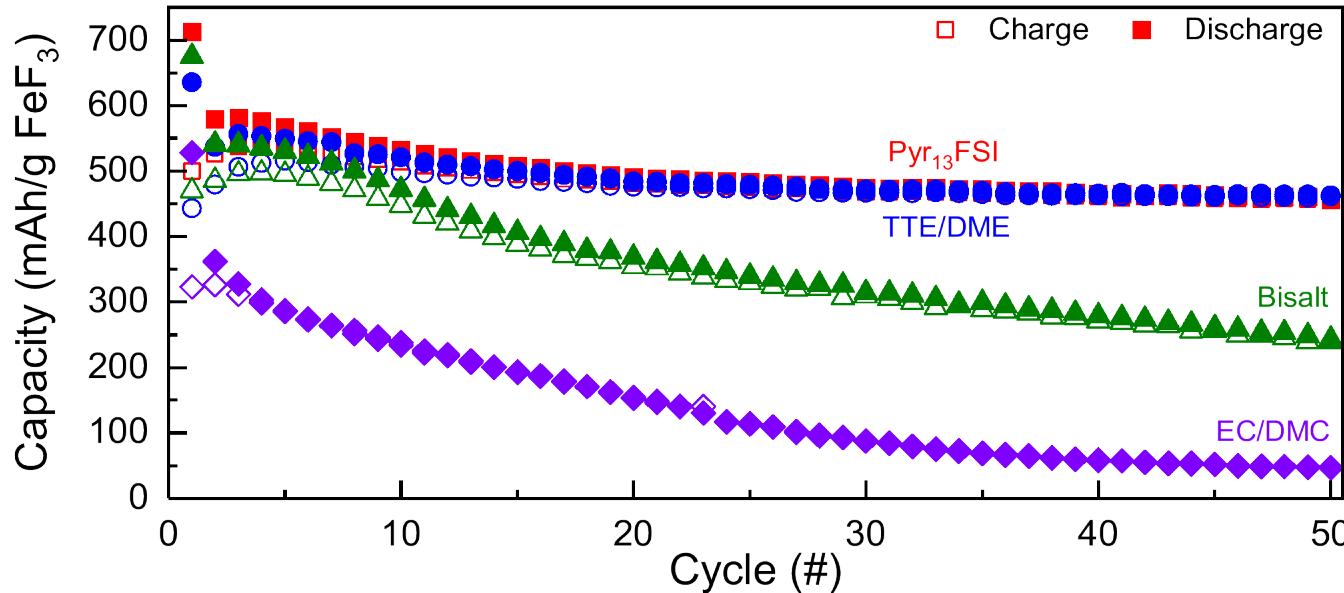
EC/DMC : commercial 1M LiPF₆/carbonate electrolyte

¹Xiao *et al.*, *Nature Mater.*, 2020,

²Ren *et al.*, *Joule*, 2019, 19, 644-654

³Merrill *et al.*, *ACS Appl. Energy Mater.*, 2021, 4, 7589-7598

Compatibility of electrolytes with FeF_3

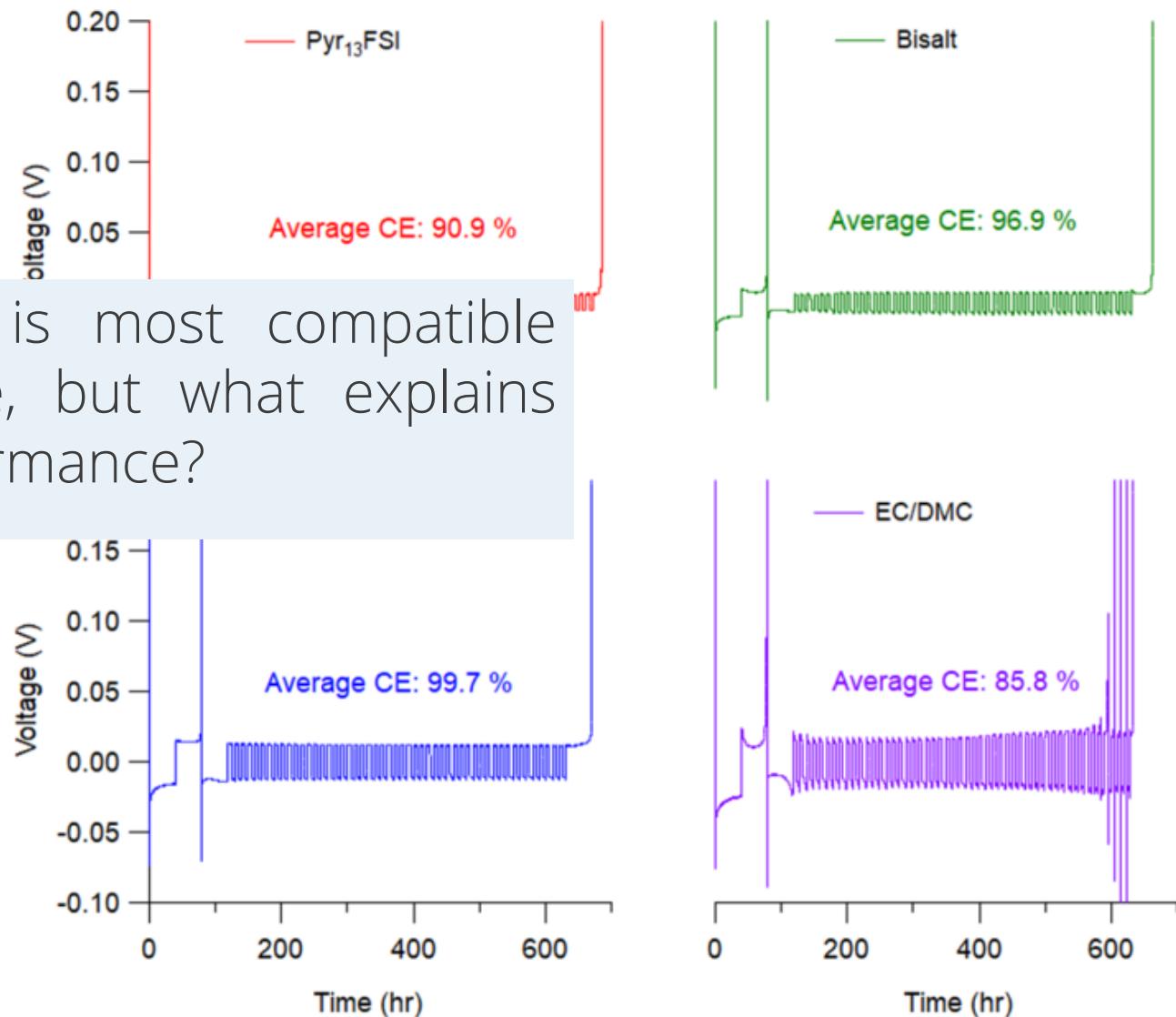
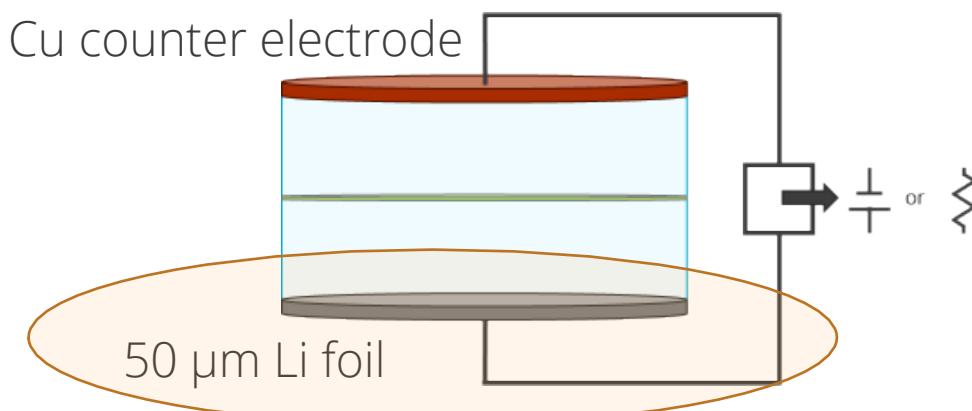


- FeF_3 -limited cells made with excess Li, cycled at relatively slow rate (C/20)
- $\text{Pyr}_{13}\text{FSI}$ and TTE/DME show highest capacity retention
- Bisalt and EC/DMC show poor FeF_3 compatibility

Compatibility of electrolytes with Li metal

- Evaluated Li-compatibility of each electrolyte by measuring coulombic efficiency (CE) in a Cu/Li cell
 - $CE = (\#e^- \text{ out}) / (\#e^- \text{ in})$
- TTE/DME and Bisalt show high CE and good behavior
- Pyr₁₃FSI and EC/DMC are incompatible with Li metal

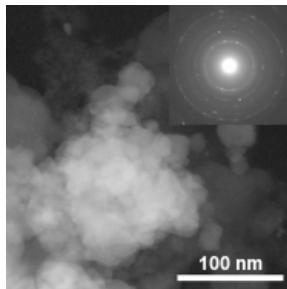
TTE/DME is most compatible electrolyte, but what explains this performance?



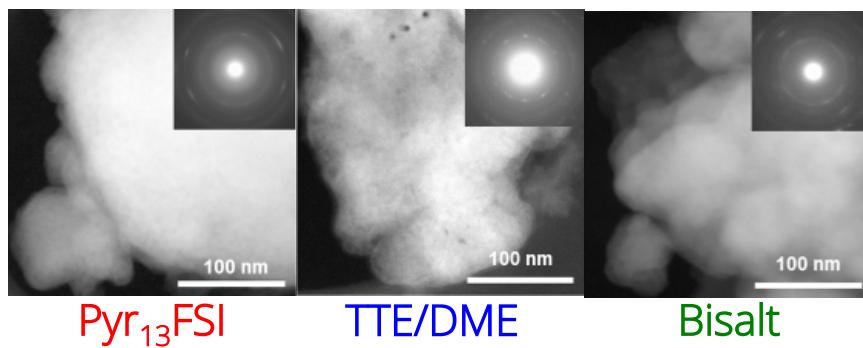
Micro/Nanoscopic changes to the electrodes

- Initial physical characterization by SEM and STEM showed important differences (and a lack of differences)

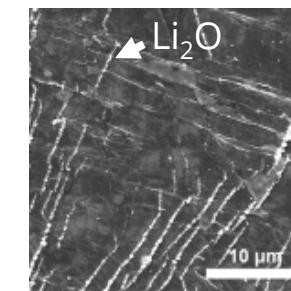
Minimal differences in final structure of FeF_3 cathodes



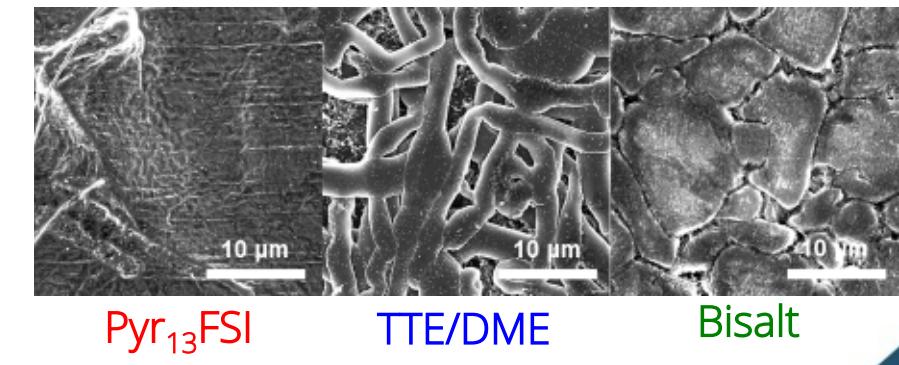
2 cycles



Notable differences in Li deposition morphology

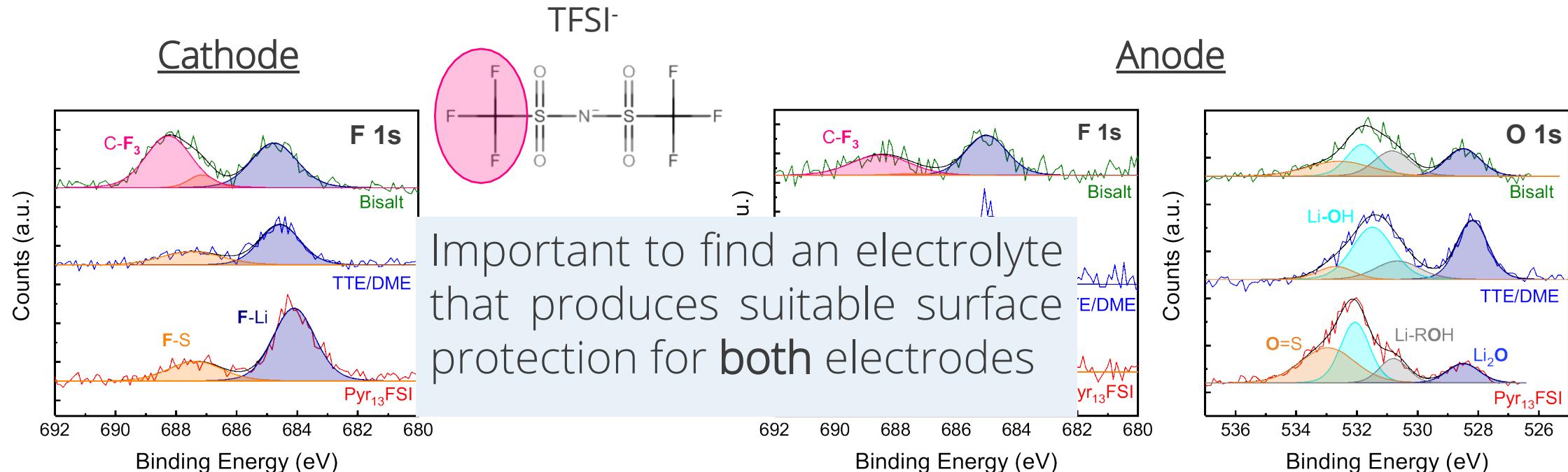


2 cycles



XPS shows important chemical differences in the CEI/SEI

- X-ray photoelectron spectroscopy (XPS) to probe differences in the chemistry of the electrode surfaces



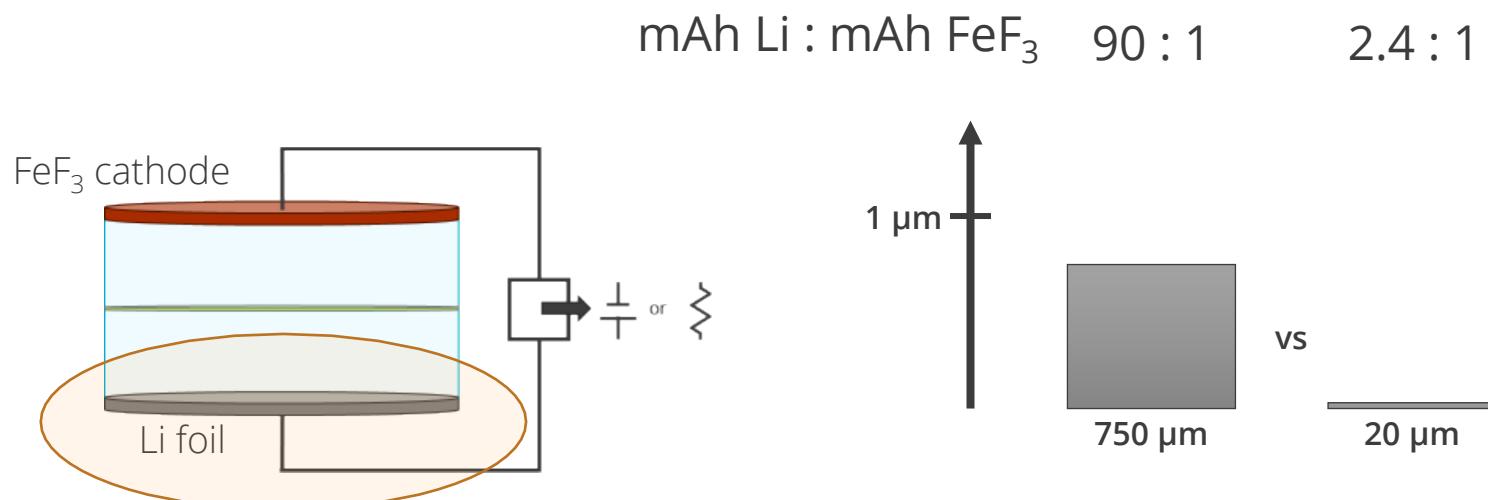
- Quantity of organic fluorine (C-F₃) in the cathode electrolyte interphase (CEI) appears correlated to worse cathode performance

- Most fluorine in the solid electrolyte interphase (SEI) is from Li-F for all electrolytes
- Most significant difference is the relative Li₂O richness of the Li-optimized electrolytes relative to the Pyr₁₃FSI

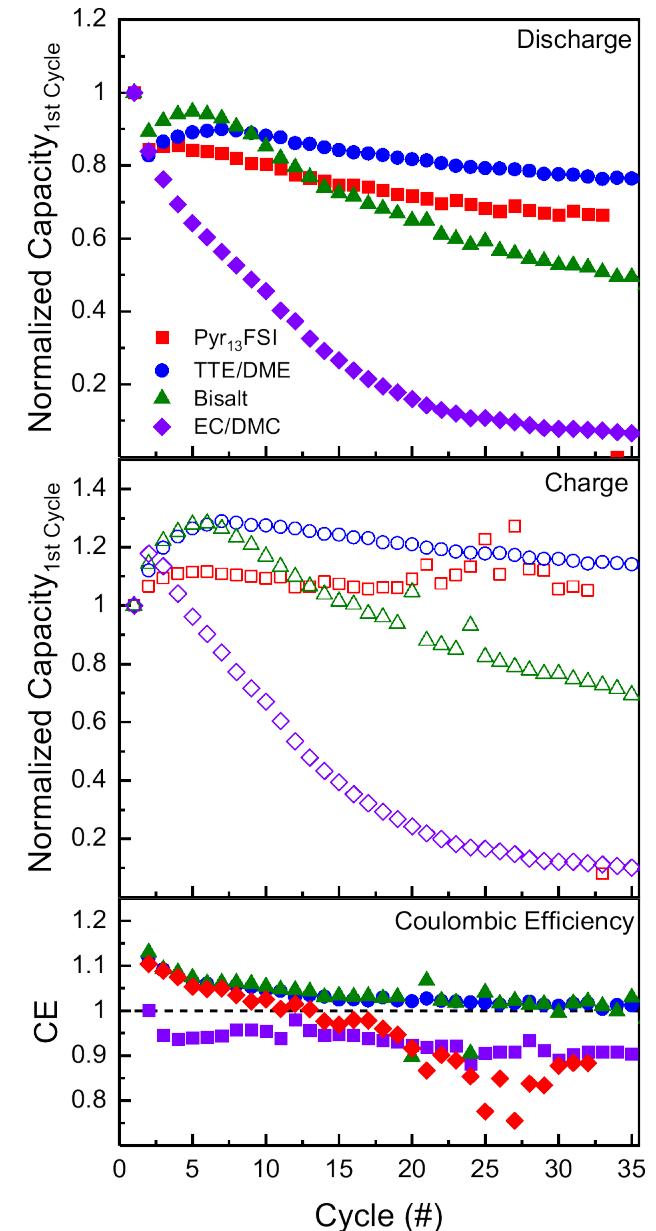
	Pyr ₁₃ FSI	TTE/DME
Li ₂ O:LiOH	- 1.1:1	and 2.8:1
Li ₂ O:LiF	- 0.63:1	and 1.3:1

Application to limited-Li full cells

- Tested each electrolyte in a full FeF_3/Li cell with limited Li supply

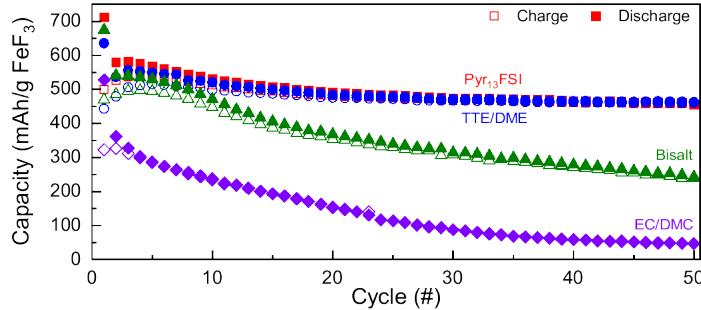


- Results agree with predictions
 - $\text{Pyr}_{13}\text{FSI}$ discharge appears stable, but begin to observe issues on charge and low coulombic efficiency due to Li

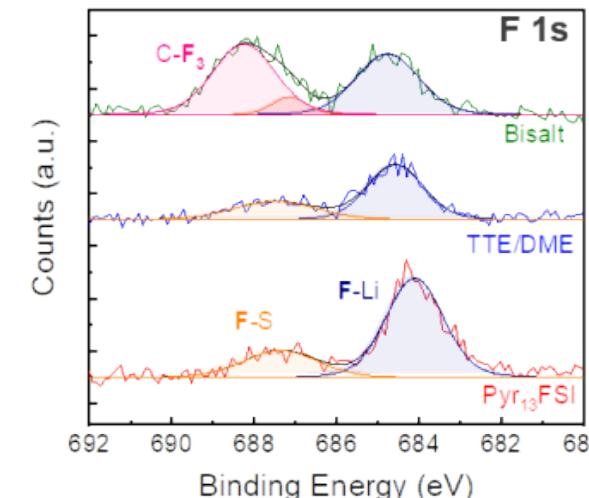
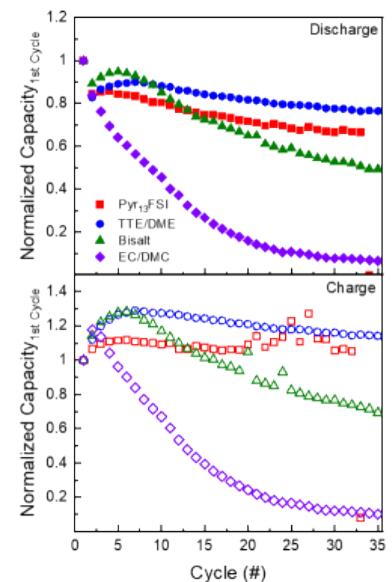


Summary

- Full cell compatibility should be taken into account when testing conversion cathode half cells
- Of the electrolytes tested, TTE/DME appears to be the best choice for a full cell FeF_3/Li secondary battery
- Chemical composition of CEI and SEI are corelated to electrode performance



Vs.





Acknowledgements

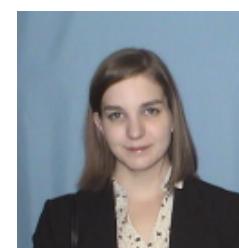
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LDRD

Laboratory Directed Research and Development

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Thank you for
your attention!