

1 **Ultrahigh Performance LC/FT-MS Non-Targeted Screening for Biomass Burning**
2 **Organic Aerosol with MZmine2 and MFAssignR**

3
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8 Key words: Fourier transform mass spectrometry, Non-targeted screening, Ion mass extraction,
9 Molecular formula assignment, Biomass burning organic aerosol

10 **Highlights:**

- 11 • A new non-targeted data processing strategy suitable for environmental complex
12 mixtures is presented.
- 13 • Results of the LC/FT-MS approach are consistent with direct infusion FT-MS results.
- 14 • Liquid smoke is a potential surrogate for biomass burning organic aerosol.

15 **Abstract**

16 In recent years, ultrahigh performance liquid chromatography Fourier transform mass
17 spectrometry (LC/FT-MS) based non-targeted screening (NTS) methods have become
18 increasingly popular for comprehensive analysis of complex organic mixtures. However, applying
19 these methods for environmental complex mixture analysis is challenging due to the extreme
20 complexity of natural samples and a lack of standard samples or surrogates for environmental
21 complex mixtures. Furthermore, limited molecular markers in the databases and insufficient data
22 processing software workflows make the application of these methods more challenging for
23 environmental complex mixtures. In this work, we implement a new NTS data processing
24 workflow to process data collected from ultrahigh performance liquid chromatography and Fourier
25 transform Orbitrap Elite Mass Spectrometry (LC/FT-MS) by combining MZmine2 and
26 MFAssignR, two opensource data processing tools and commercial Mesquite liquid smoke as a
27 surrogate for biomass burning organic aerosol. MZmine2.53 data extraction followed MFAssignR
28 molecular formula assignment offered noise free and highly accurate 1733 individual molecular
29 formulas presented in liquid smoke with 4906 molecular species, including isomers. The results
30 of this new approach were consistent with the results of direct infusion FT-MS analysis confirming
31 its reliability. Over 90% of the molecular formulas presented in mesquite liquid smoke were
32 matched with the molecular formulas of ambient biomass burning organic aerosol. This suggests

33 the potential use of commercial liquid smoke as a surrogate for biomass burning organic aerosol
34 research. The presented method significantly improves the identification of the molecular
35 composition of biomass burning organic aerosol by successfully addressing some of the limitations
36 related to the data analysis and giving a semi quantitative insight into the analysis.

37 **1. Introduction**

38 Natural and anthropogenic biomass combustion contributes a significant amount of organic aerosol
39 to the atmosphere. Model estimates have indicated that the annual emission of organic carbon from
40 biomass burning in 2008 ranged between 15.65 to 51.91 Tg (Pan et al., 2020). Smoke emissions
41 from fires usually contain simple organic acids, sugars, phenolic compounds, polycyclic aromatic
42 hydrocarbon (PAHs), and thousands of unknowns depending on the condition of the fire (Brege et
43 al., 2018; Lin et al., 2016; Mazzoleni et al., 2007). Transportation of these fire plumes through the
44 atmosphere can yield high levels of air pollution impacting human health as well as the Earth's
45 climate (Kenward et al., 2013; Shiraiwa et al., 2017; Zhang et al., 2020). Unraveling the molecular
46 composition of biomass burning organic aerosol is challenging due to its extreme complex nature
47 (Brege et al., 2021). Although several analytical methods for a variety of organic aerosol analyses
48 are available in the literature, none yet provide a comprehensive qualitative and quantitative
49 assessment of the molecular composition (Duarte et al., 2021). Most LC/MS or GC/MS based
50 quantitative studies in the literature use targeted approaches to analyze specific compounds
51 (Duarte et al., 2021), which limits the analysis to the selected analytical standards available. In
52 recent years, direct infusion Fourier transform mass spectrometry (DI-FT-MS), frequently with
53 electrospray ionization (ESI; Brege et al., 2018, 2021; Ijaz et al., 2022; Lin et al., 2016; Schum et
54 al., 2018), have been used technique for detailed analysis of organic aerosol. Such studies have
55 demonstrated extreme molecular complexity ranging from saturated to highly unsaturated products
56 of incomplete combustion with a wide variety of transformations from atmospheric oxidation
57 processes (Brege et al., 2021; Ijaz et al., 2022; Lin et al., 2016; Schum et al., 2018). Although very
58 powerful, direct infusion mass spectrometry cannot provide either structural or quantitative
59 information of the studied analytes. The analysis can be subject to matrix effects and ionization
60 suppression due to the competition for charge during the electrospray ionization (Chekmeneva et
61 al., 2017). Therefore, relative intensities of direct infusion mass spectrometry may not directly
62 proportional to the relative concentration of the analysts (Guillemant et al., 2021). The matrix

63 effects and ionization can be reduced using chromatographic separation to reduce the molecular
64 complexity of species and their concentrations arriving at the ionization source (Forcisi et al.,
65 2013). Due to the analogy of a molecular partitioning in the chromatographic column and within
66 the ESI droplet, Cech et al., 2001 observed the variation in the ionization response with the
67 chromatographic retention time. Such that, the fast eluent species can have low ESI response, and
68 later eluant species have a high possibility of having high ESI response (Cech et al., 2001). At the
69 same time, changing the mobile phase composition can lead to changes in the ionization response
70 (Ghosh and Jones, 2015; Liigand et al., 2014). Due to these observations, compounds with similar
71 ionization efficiencies are expected to elute with similar retention times. The reduced complexity
72 and the lower concentration of analytes together minimize the detection artifacts caused by
73 ionization competition (Forcisi et al., 2013; Kim et al., 2019; Lin et al., 2010). Therefore, the peak
74 area values obtained from LC/MS analysis are more reliable for quantitative or semi-quantitative
75 assessments than the relative ion abundance in mass spectra from direct infusion analysis (Forcisi
76 et al., 2013; Lin et al., 2010).

77 In Fourier Transform Mass Spectrometry (FT-MS), the accuracy of the detected masses is directly
78 proportional to a function of data acquisition time. On the other hand, in LC/FT-MS analysis,
79 longer data acquisition times can result in a lower number of detection peaks (Kim et al., 2019;
80 Michalski et al., 2012). Thus, balancing the two factors is crucial to obtain optimal results. In our
81 study, we used ultrahigh performance LC to achieve maximum peak capacity and fast separation,
82 which is compatible with the relatively fast data acquisition of ultrahigh resolution Fourier
83 Transform Orbitrap Elite Mass Spectrometry (Michalski et al., 2012). Similar analytical
84 approaches are widely used for non-targeted screening of metabolites and trace environmental
85 contaminants (Hollender et al., 2019; Michalski et al., 2012; Zhang et al., 2020).

86 LC/FT-MS and high-resolution time of flight (ToF) mass spectrometry non targeted
87 screening is a continuously developing technique in environmental complex mixture analysis
88 (Hollender et al., 2017; Minkus et al., 2022). There are numerous open source and commercial
89 software tools available for LC/MS post data processing in the literature. However, data post-
90 processing remains as the main challenge when applying NTS approach in environmental complex
91 mixture analysis (Hohrenk et al., 2020; Houriet et al., 2022; Minkus et al., 2022). Because LC/FT-
92 MS data acquisition generates large quantities of data points within a short time frame. Collected
93 data can be highly complexed due to the isotopes, adducts, and multicharged species detected and

94 sheer number of different molecules present in such samples (Hollender et al., 2017). The peak
95 detection and mass extraction are highly dependent on the data processing software program used.
96 For example, a study conducted by Hohrenk et al. observed a low level of coherence among the
97 extracted masses extracted from different data processing software and algorithms (Hohrenk et al.,
98 2020). Database based compound identification can lead to incorrect or uncertain molecular
99 identifications due to the lack of observation molecular markers of the databases (Hollender et al.,
100 2019). Further, signal threshold value also critical when considering the quality of the data,
101 detection of low abundance species and the computational time (Hollender et al., 2019, 2017;
102 Houriet et al., 2022). Therefore, data processing tools and parameters need to be carefully selected
103 and standardized depending on the study (Hollender et al., 2017; Houriet et al., 2022).

104 The goal of this work is to develop an LC/FT-MS method to identify biomass burning
105 organic aerosol components using commercial liquid smoke. Liquid smoke is a commercial food
106 flavoring substance, created by dissolving the volatile components collected from the smoke
107 produced during the slow pyrolysis of wood (Montazeri et al., 2013; Simon et al., 2005). This
108 process resembles how organic aerosols are generated through pyrolysis during plant burning
109 events (Sekimoto et al., 2018). However, the production of liquid smoke does not involve
110 atmospheric oxidation and multiphase processing that would typically occur in ambient conditions.
111 Laboratory oxidation and multiphase processing of liquid smoke to more comprehensively mimic
112 ambient biomass burning organic aerosol is the subject of a forthcoming paper (Divisekera et al,
113 in prep). Therefore, liquid smoke is a potential surrogate for biomass burning organic aerosol. In
114 this study the obstacles in the LC/FT-MS data processing are addressed by combining the existing
115 LC/MS data processing software MZmine2.53 (Pluskal et al., 2010) with MFAssignR (Schum et
116 al., 2020) software for molecular formula assignments of environmental complex mixtures.
117 MFAssignR was originally designed for the post-processing of direct infusion ultrahigh resolution
118 mass spectra and has been adapted to handle LC-MS data as well. It calculates molecular formula
119 using the expected chemical relationships between ions in a complex mixture, including CH₂, H₂O,
120 O, H₂, and CH₂O which provides robust identification of unambiguous molecular formulas as
121 described (Schum et al., 2020). In this approach, the exact masses of the detected ions and their
122 peak areas were extracted using MZmine2.53 and then MFAssignR was used for molecular
123 formula assignments, noise estimation, isotope filtration, and internal mass recalibration. The
124 resulting data set contains assigned and noise-free unambiguous molecular formula along with the

125 recalibrated exact masses, chromatographic retention times, and chromatographic peak areas
126 representing analyte abundance.

127 **2. Material and methods**

128 **2.1 LC/FT-MS analysis of liquid smoke**

129 Wright's commercial mesquite liquid smoke solution was filtered using 0.20 μm PTFE syringe
130 filter and was diluted by two times using LC/MS grade water. Chromatographic separation was
131 done using C18 column (Kinetex® 1.7 μm XB-C18 column) with an UltiMate 3000 UHPLC
132 System (Thermo Fisher Scientific) operated with Chromeleon™ chromatography data system
133 software. Compounds were separated using a solvent gradient at 35°C with the flowrate of
134 0.2 mL min⁻¹ starting with 5% ACN and 95% water (both with 0.1% formic acid) and increased
135 up to 70% ACN over 45 min with a linear gradient. Separated liquid smoke analytes were analyzed
136 with both positive and negative electrospray ionization (ESI) modes. We used 0.1% formic acid
137 to improve the chromatographic separation in both ionization modes. Hence, a decrease in the
138 sensitivity of acidic and neutral compounds detection in the -ESI analysis might happen.

139 The ESI settings include: temperature of 200 °C, sheath gas flow rate of 20, auxiliary gas flow rate
140 of 8, sweep gas flow rate of 1, spray voltage of 2.8 kV, capillary temperature of 300 °C and S-lens
141 RF level of 60% were used in the -ESI analysis. In +ESI mode analysis, the same initial ESI
142 settings were used except the sprayed voltage (3.8 kV) and S-lens RF level of 65%. Data
143 acquisition was done using Fourier Transform Orbitrap Elite Hybrid Ion trap-Orbitrap Mass
144 spectrometer (FT-OTE, Thermo Scientific™) at resolving power of 120,000 and 240,000 defined
145 at m/z 400 Da. The instrument was operated with XCalibur 3.0 software in the range of 100-600
146 Da. MS/MS data were acquired by 15,000 resolutions at m/z 400 with the data dependent selected
147 ion monitoring (SIM) for the 10 most abundant ions. Fragmentation was achieved with 25% and
148 35% of the normalized collision energy in the ion trap. Dynamic exclusion duration was disabled
149 to obtain the fragments of the isomeric species. Triplicated analysis of the liquid smoke sample
150 and the corresponding blank (LS/MS grade water) were performed using the discussed procedure.

151 **2.1.2 Data extraction using MZmine2.**

152 In the first step of ion extraction, Thermo raw data files were directly imported into the
153 MZmine2.53 software platform without any file conversion. Baseline correction was done to
154 remove the gradual shift of the baseline of the chromatogram. Then, the mass detection was carried

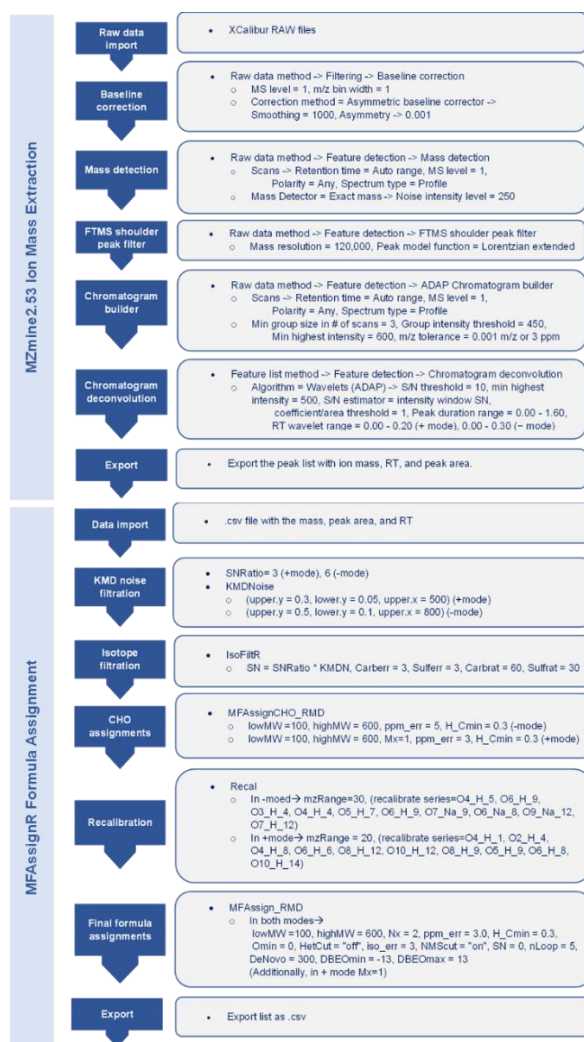
155 out by centroiding the profile masses using Exact mass detector algorithm (Pluskal et al., 2010)
156 keeping an optimum noise intensity level for mass detection. The noise intensity level was kept at
157 250 during the entire retention time range to avoid filtering out the analyte peaks with low
158 intensity. Shoulder peaks generated during the Fourier transformations were filtered out using the
159 Lorentzian extended algorithm. Subsequently, chromatograms were constructed using the ADAP
160 chromatogram builder (Myers et al., 2017) with a minimum group size of 3 scans, a group intensity
161 threshold of 450, a minimum highest intensity of 600, and a mass error tolerance of 0.001 Da.
162 Obtained chromatograms were then deconvoluted using the ADAP wavelet algorithm with a
163 minimum feature height of 500, signal to noise ratio of 10, coefficient/area threshold of 1, peak
164 duration of 0.00-1.6 minutes, and RT wavelet range of 0.2 in positive mode ion deconvolution
165 and 0.3 in the negative mode ion deconvolution step. Then the deconvoluted ion masses, retention
166 times and the peak area values were exported as a .csv file. The optimized data extraction steps
167 are shown in Figure 1.

168 We used a small minimum group number of 3 scans and a low noise level of 250 during
169 the ion mass extraction procedure. Keeping a low signal threshold is more helpful in detecting low
170 intense and isotopic peaks (Houriet et al., 2022). At the same time, it can contribute to a higher
171 chance of extracting false peaks. In this scenario, Kendrick mass defect (KMD) based noise
172 estimation function in MFAssignR package (Schum et al., 2020) is useful to filter out the
173 incorrectly extracted masses with smallest peak area values. Further, the isotope filtration function
174 successfully identified the isotopic masses extracted. These two functions in the MFAssignR
175 software package successfully adapted to processing LC/FT-MS data, yielding high accurate
176 molecular formula assignment. (Further discussed in the supplementary section S1 and S2)

177 **2.1.3 Molecular formula assignment using MFAssignR.**

178 Molecular formulas for the extracted masses were assigned separately for the replicate liquid
179 smoke and blank mass lists. The chromatographic peak area of each extracted mass was selected
180 to represent the analyte abundance as opposed to the relative abundance in the mass spectra.
181 Kendrick mass defect noise estimation and polyisotope filtration were done to avoid false
182 assignments. During the formula assignment in $^{12}\text{C}_{1-c}$, $^1\text{H}_{0-h}$, $^{16}\text{O}_{0-o}$, $^{14}\text{N}_{0-2}$ and in +ESI, Na^+ elements
183 were selected within the following ranges: double bond equivalence minus number of oxygen
184 atoms (DBE-O) (EqS4) between -13 to +13, $0.3 \leq \text{H/C} \leq 2.5$, $0.0 \leq \text{O/C} \leq 2$ de novo cut-off of 300

185 m/z, and mass error of ≤ 3 ppm using MFAssignR. the peak area of the Na⁺ adducted species and
 186 H⁺ adducted species with the same retention time and same molecular formulas were added
 187 together to yield the total peak area. Then all three replicate assignments were aligned according
 188 to the formula assignments and the retention time using a custom R script with a range of retention
 189 time ± 0.05 min. The molecular formulas in at least two out of the three data sets were retained.
 190 The same procedure was followed for the blank data sets. Subsequently, we subtracted the average
 191 peak area of the blank from that of the sample to eliminate any contaminants present. Then, all
 192 species with positive peak area values were selected for further analysis. Each peak area's average
 193 peak area and the corresponding relative standard deviation (RSD) were calculated.



194

Figure 1: MZmine2.53 and MFAssignR data processing workflow and parameters used.

195

196 **2.2 Direct infusion FT-MS and molecular formula assignment of liquid smoke**

197 In direct infusion FT-MS analysis, a more diluted sample is typically used to reduce matrix effects,
198 ion suppression, and interference, therefore the same mesquite liquid smoke sample diluted 10
199 times in 55% of ACN medium to best match the average chromatographic solvent matrix. Then
200 the obtained exact masses were assigned using MFAssignR software using the same formulas
201 assignment parameters discussed.

202 **3. Results and Discussion**

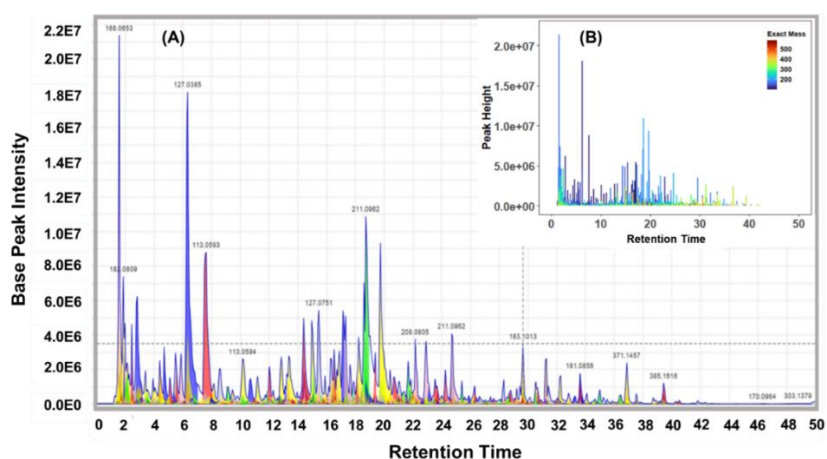
203 **3.1 Ion mass extraction and molecular formula assignment**

204 The C18 column and the linear gradient reasonably separated the liquid smoke complex
205 mixture. Approximately 8000 and 7000 ion masses were extracted in both ionization modes.
206 Figure 2 shows the extracted ion chromatogram generated by MZmine2 software (Figure 2A) and
207 the extracted data plotted by peak height vs retention time and colored by observed ion mass in a
208 reconstructed chromatogram for the +ESI analysis with a resolving power of 120,000 (Figure 2B).
209 The similarity in peak intensities and distributions supports the observation that our approach
210 accurately extracted the ions with their respective masses and retention times.

211 Accurate extraction of low intense noise peaks a challenge that we faced with a low noise
212 level during the ion mass extraction. However, KMD noise estimation method (section S1)
213 successfully removed the noise peak generated during the mass ion extraction (Figure S1)
214 especially in the –ESI ionization mode analysis. According to the KMD plot generated with the
215 +ESI mode analysis, very few noise peaks were extracted with the parameters used (Figure S2).
216 MFAssignR isotope filtration function (IsoFiltR) (section S2) identified isotopic ion masses
217 presented in the extracted mass list. For higher abundance species, ¹³C isotopic masses, which have
218 the same retention time as the monoisotopic masses, showed approximately equal isotopic
219 abundance (peak area) ratio with the theoretical isotopic abundance (peak area) ratio (Figure S3).
220 This behavior was more noticeable in the 120,000 resolving power than in the 240,000, consistent
221 with the expected fewer species in the higher resolution.

222 MZmine2 ion mass extraction followed by MFAssignR formula assignment yielded about
223 3000 and 1500 unambiguous molecular species in both +ESI and –ESI ionization modes at 120,000
224 and 240,000 resolving power settings. The method showed good reproducibility, such that, the
225 relative standard deviation of the peak area was below 20% for 90% of the aligned formulas that

226 were detected in at least two out of the tree replicate sets along with their retention times. Overall
227 results obtained at 120,000 and 240,000 resolving power were similar in each ionization mode
228 (Table S1). However, the higher number of total assignments obtained with 120,000 is likely due
229 to the faster detection times at the lower resolving power allowing the instrument to detect more
230 analyte signals (Michalski et al., 2012). The results obtained at 120,000 resolving power will be
231 discussed for the remainder of this article.



232
233 Figure 2: (A) Extracted ion chromatogram of liquid smoke sample and (B) retention time mass
234 spectra of extracted masses with the RT and peak height. Both are positive ion data collected with
235 a mass resolving power of 120,000.

236 It is known that MZMine2 identifies adduct ion species with two satisfactory conditions. (1) The
237 adduct ion needs to have the same retention time as the protonated species, and (2) the difference
238 between those two ions needs to be equal to the mass of the adduct (Pluskal et al., 2010). So, the
239 analytes exclusively presented as Na^+ adducts (without protonated species) cannot be identified.
240 Our method successfully solved the challenge of accurately identifying multiple adducted species
241 without H adducts and we observed 338 molecular species with 185 molecular formulas
242 exclusively presented as Na^+ adducted species in the +ESI analysis.

243 Most of the molecules detected in the liquid smoke samples were CHO species and
244 represented 69% and 99% of the total peak area of the identified species in +ESI and -ESI analysis,
245 respectively. Elemental ratios of the identified species follow the reversed phase chromatographic
246 retention order, as the species with relatively low O/C and H/C values have longer retention times,
247 whereas molecules with high O/C or higher H/C values elute early (Figure S5). A larger number
of nitrogen-containing formulas were detected in positive ionization mode. These molecules may
be valid CHNO/CHN species or ammonium adducts of the molecules in the liquid smoke sample.

248 So, the N-containing formulas were re-evaluated by removing NH_4^+ from the existing formula to
249 check for non-adducted valid assignments. This led to 95 out of 1771 species with a matching
250 protonated molecular formula at the same retention time suggesting that they were adducts while
251 majority of (1676) species were valid CHN/CHNO assignments.

252 **3.2 Comparison of LC/FT-MS with the direct infusion FT-MS analysis**

253 According to the positive mode analysis 1098 individual molecular formulas,
254 corresponding to 3209 molecular species, were successfully identified including structural
255 isomers. Of these, 970 molecular formulas were common among the molecular formulas obtained
256 from direct infusion mass spectrometric analysis of the same liquid smoke sample. This represents
257 99% of total peak area obtained from the molecular species identified by liquid chromatographic
258 ion mass extraction and 95% with respect to the total abundance of the direct infusion mass
259 spectrometric (DI-FT-MS) identified species (Figure 3A). These 970 common molecular formulas
260 correspond to 3043 total molecular species due to the numerous isomers present. This is an inherent
261 advantage of the LC/FT-MS because isomeric structures corresponding to only one molecular
262 formula can be separated. Most of the commonly identified species were CHO species
263 corresponding to about 70% of the total LC/FT-MS peak area and 86% of the direct infusion FT-
264 MS peak abundance. The LC/FT-MS method uniquely detected 128 molecular formulas with 166
265 molecular species, and they were responsible for 1% of the total peak area of the LC/FT-MS
266 identified species.

267 In negative mode ESI analysis, 1697 molecular species corresponding to 635 individual
268 molecular formulas were obtained. Of these 578 molecular formulas corresponding to 1619
269 molecular species are common with the $-$ ESI direct infusion analysis. This represents 98% of the
270 total LC/FT-MS peak area and 96% of the total direct infusion FT-MS peak abundance (Figure
271 3B). CHO species accounted for 97% of the LC/FT-MS peak area of the commonly identified
272 species. The summary of the molecular formulas detected in both methods is shown in Table 1.

273 The direct infusion analysis identified a significant number of unique molecular formulas,
274 relative to the liquid chromatography analysis. Specifically, 1119 and 1183, in $+ESI$ and $-ESI$
275 analysis, respectively (Figure 3). These direct infusion species were less intense than the majority
276 of the common species (Figure S4) likely due to lower concentrations or lower ionization
277 efficiencies. Many of the formulas uniquely identified species with direct infusion FT-MS methods

278 had relatively higher masses with relatively higher C and O content in comparison to the
 279 commonly detected species by both methods (Figure S4). While we tuned the Orbitrap to be as
 280 sensitive as possible to higher mass-to-charge values and liquid chromatography can improve
 281 ionization by minimizing the matrix effects, there is an increased sensitivity to lower mass-to-
 282 charge values as has been seen in (Hawkes et al., 2020). This observation in conjunction with
 283 potentially lower ionization efficiency, possible low species concentration, and the additional
 284 dilution encountered during LC/FT-MS analysis (due to 5 uL injection) likely explain why these
 285 species are not observed.

286 Although many of the compounds observed in direct infusion are not also observed in
 287 LC/FT-MS, those compounds unique to direct infusion make up only ~5% of the total abundance
 288 in direct infusion, highlighting that although we observe fewer compounds, we are still able to see
 289 the compounds that make up 95% of the total abundance in this sample and may be most important
 290 to the overall composition. The trade-off between more formulas and more specific information
 291 about the major species in a sample needs to be considered. Despite this, the LC/FT-MS analysis
 292 provided several additional pieces of information that were not obtained from direct infusion mass
 293 spectrometry, including: C18 column retention, isomeric complexity, the ability to automate
 294 fragmentation scans for structural analysis, and better semi-quantitative information, which allows
 295 us to assess the changes in compound concentration in a relative way (i.e., a compound has higher
 296 or lower concentration in one sample vs. another).

297 The proposed approach is not intended as a substitute for direct infusion mass
 298 spectrometry. Instead, this method enhances our understanding of the sample's composition with
 299 the relative quantities, isomers, and tentative structure. Therefore, by performing both methods,
 300 more comprehensive analysis of a complex sample can be achieved.

301 Table 1. Number of species detected by LC/FT-MS and direct infusion FT-MS methods.

	Number of Formulas							
	CHO		CHNO		CHN		CH	
	LC-MS	DI-MS	LC-MS	DI-MS	LC-MS	DI-MS	LC-MS	DI-MS
Positive	405	952	605	1329	83	116	5	50
	(1425)		(1495)		(276)		(13)	
Negative	579	1428	56	330	-	-	-	-
	(1619)		(78)					

302 The number of the species including isomers are in the parentheses.

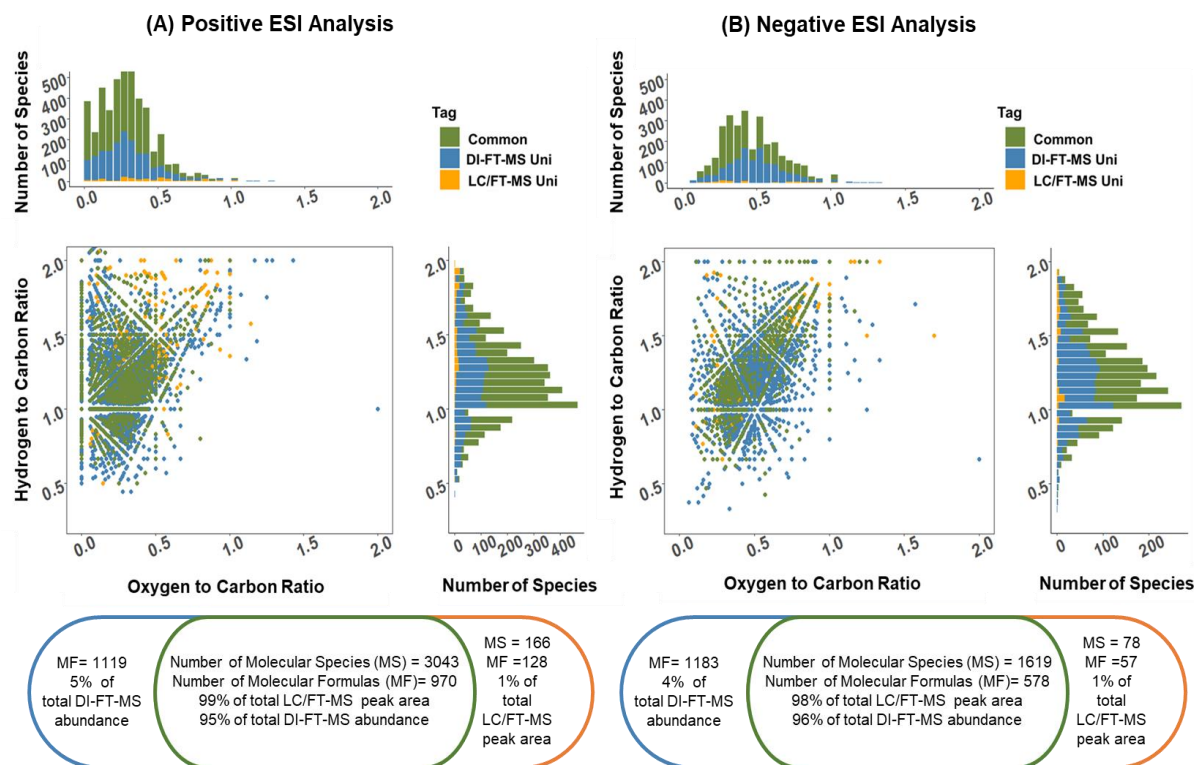


Figure 3: Comparison of the elemental ratios identified with LC/FT-MS analysis and direct infusion FT-MS analysis and the distribution of the identified species with O/C and H/C ratios, and comparison of the number of formulas, peak area, and peak abundances identified each method. (A) Positive ESI analysis and (B) Negative ESI analysis

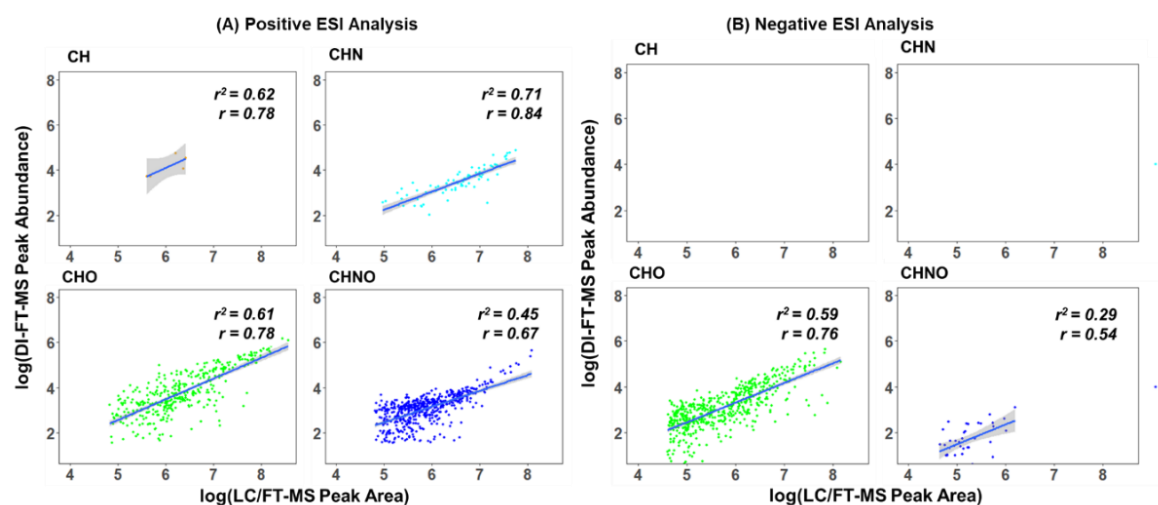
303

304 Overall, the results obtained from LC/FT-MS, and direct infusion FT-MS agreed with each
 305 other in both ionization methods. This further highlights the reliability of using the novel method
 306 in environmental organic mixture analysis.

307 Although the spectral abundances obtained from direct infusion ESI-FT-MS are not
 308 necessarily quantitative representation of the concentration of species, relative quantitative
 309 information such as normalized peak intensity abundances and weighted averages are widely used
 310 in the literature for sample trend comparison in environment complex mixture analysis (Brege et
 311 al., 2018; Hawkins et al., 2018; Wozniak et al., 2014; Xu et al., 2020)

312 In our study, we observed an interesting correlation between the chromatographic peak area
 313 (cumulative peak area of its all isomers) and the direct infusion peak abundance in commonly
 314 detected species in both methods (Figure 4). According to the Pearson correlation coefficient value
 315 ($r = \sqrt{r^2}$), $\log(\text{LC/FT-MS Peak Area})$ and $\log(\text{DI-FT-MS Peak Intensity Abundance})$ shown a
 316 positive linear relationship ($r > 0$). However, this is not sufficient to predict the concentration of the

317 species. Further method development with analytical standards is recommended and is beyond the
 318 scope of this study. Nonetheless, the correlation is valuable because it suggests that direct infusion
 319 peak intensities are not strongly impacted by matrix effects and thus may be related to the actual
 320 abundance of the analytes. This observation supports the concept of using peak intensity
 321 abundance for giving semi-quantitative insight of direct infusion peak intensities when comparing
 322 the environmental samples as has been done in several other studies in the literature (M. Brege et
 323 al., 2018; Hawkins et al., 2018; Wozniak et al., 2014; Xu et al., 2020).



324 Figure 4: Variation of the logarithmic value of direct infusion-FT-MS peak abundance with
 325 logarithmic value of LC/FT-MS peak area with their corresponding trendlines, r^2 values and,
 Pearson correlation coefficient r . (A) Positive ESI analysis and (B) Negative ESI analysis

326 3.3 Comparison of two ionization modes

327 Pairwise comparison of the molecules obtained
 328 from both ionization mode detected 344 molecular
 329 formulas in common. These common species are
 330 detected between H/C ratio of 0.75 -1.50 and O/C ratio
 331 of 0.0 - 0.5 (Figure 5) and they might correspond to the
 332 phenolic compounds obtained the decomposition of
 333 lignin which are expected to be present in smoke (Rivas-
 334 Ubach et al., 2018). In accordance with the
 335 literature, the trend of compounds uniquely
 336 identified using the -ESI analysis are the

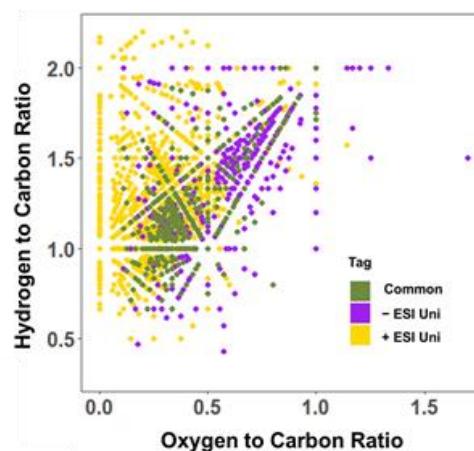


Figure 5: Elemental ratios of the molecular formulas presented in the liquid smoke sample detected by the new proposed LC/FT-MS analysis in both positive and negative ESI mode.

337 relatively highly oxygenated species with high O/C ratio values and most of the formulas unique
338 to the positive mode are consist with CHN and CHNO species present in the sample. Mesquite is
339 a leguminous tree and has the ability of fixing nitrogen in the environment (Lyons and Rector,
340 2015) That may be the reason for having noticeable amount of CHN and CHNO molecules (~20%
341 of the total peak area abundance) in the liquid smoke sample in the +ESI mode analysis.
342 Additionally, these N species include 8 out of the 18 pyrazine-based chromophores present in
343 biomass burning organic aerosol reported in the literature (Hawkins et al., 2018). These molecules
344 can be produced with the Maillard type reaction of amino acids and the reducing sugars during the
345 pyrolysis of mesquite plants. Among the 1389 identified molecular formulas from both +ESI and
346 –ESI mode analysis 95 molecular formulas with 155 possible structures are matched with Phenol
347 Explorer database (Rothwell et al., 2013). Of these, 21 structures belonged to the class of
348 flavonoids, 25 belonged to lignans. 54, 50, and 5 are classified as other phenols, phenolic acids,
349 and stilbenes, respectively.

350 **3.4 Detection of isomeric species**

351 Environmental complex mixtures can contain several isomeric species with the same
352 molecular formula. LC/MS analysis is valuable in this scenario to separate the isomeric species to
353 be independently identified. We observed 689 (in +ESI mode) and 341 (–ESI mode) molecular
354 formulas present in at least two different retention times, supporting the existence of the isomers
355 in the liquid smoke sample (Figure S6). MS/MS fragmentation patterns allow for the tentative
356 structural identification of these isomeric molecules. For example, $C_6H_6O_3$ appeared in three
357 different retention times in the +ESI chromatogram (Figure 6). Three of these peaks elute very
358 close to each other with close retention time of 4.4 min, 5.5 min and 6.3 min. Hence, they are
359 likely the isomeric species of $C_6H_6O_3$. The corresponding MS/MS pattern of these peaks showed
360 similarities to the characteristic's fragmentations of benzenetriol reported in the literature (Fahmey
361 et al., 2001) Consequently, the suspected molecules are likely phloroglucinol, hydroxyquinol, and
362 pyrogallol. Predicted octanol to water partition coefficients (logP) from Molinspiration interactive
363 logP calculator, (Molinspiration.com, n.d.) log P values are 0.43, 0.49 and 0.73 respectively. When
364 considering chromatographic affinity of the molecules in reversed phase chromatography, those
365 peaks could be identified as phloroglucinol (at 4.4 min), hydroxyquinol (at 5,5 min), and pyrogallol
366 (at 6.3 min) as shown in Figure 6. Fragmentation of trihydroxybenzene molecules result significant
367 peaks at 81 ($MH^+ - H_2O - CO$), 98 ($MH^+ - CHO$) with the 25% percent normalized collision energy.

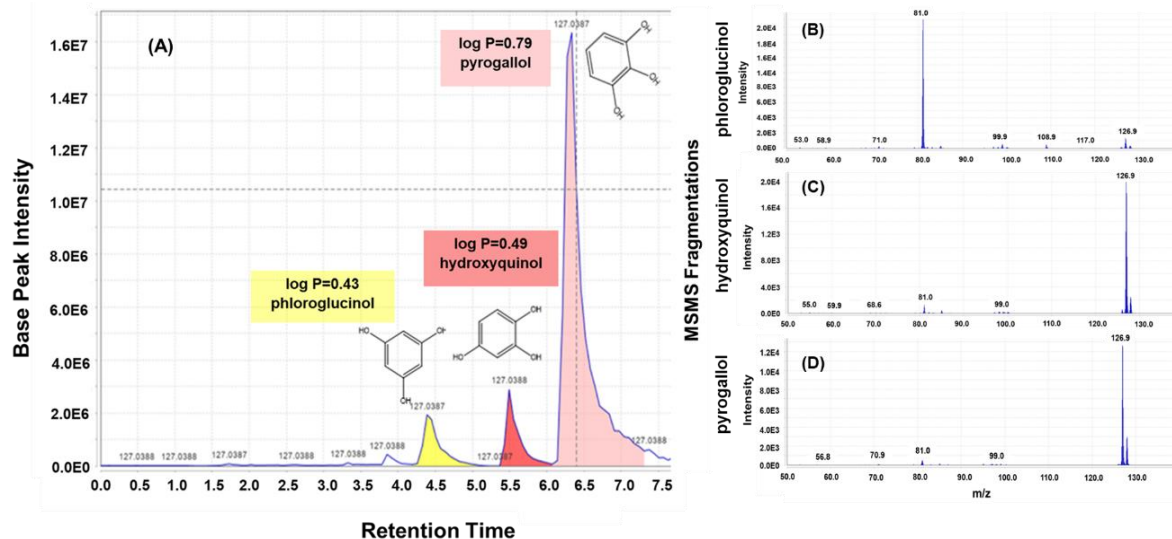


Figure 6: (A) Extracted ion chromatogram of $C_6H_6O_3$ species and MS/MS fragmented spectra of the three isomers with highest intense peaks, (B) phloroglucinol (m/z 81.0, 126.9, 99.9, 108.9), (C) phloroglucinol (m/z 126.9, 81.0, 99.0, 68.6), (D) pyrogallol (m/z 126.9, 81.0, 70.9)

368 Unlike the other two isomeric species of trihydroxybenzene, the significant fragmentation of
 369 pyrogallol (RT at 6.3 min) was not observed with the 25% normalized collision energy CID
 370 fragmentation. This could be due to the proximity of the H atom to OH group present in the
 371 benzene ring of the molecule. The similar way we could use our method for tentative structural
 372 isomeric species identification in biomass burning organic aerosol. This approach allows for the
 373 tentative identification of isomeric species in a complex mixture and is significant because there
 374 is only very limited database information. Reference standards are presented for detailed
 375 identifications. However, none of the tentative structures were confirmed.

376 3.5 Liquid smoke as a surrogate for biomass burning organic aerosol.

377 All detected molecular formulas in liquid smoke continuously spanning from C_3 to C_{32}
 378 with a DBE (Eq S3) range of 1-15 were observed (Figure S8) and the species' retention time
 379 increases with the C number and the DBE value (Figure S8B). Most identified molecular species
 380 are defined as olefins based on the modified aromaticity index AI_{mod} (Eq S5) (Figure S7, Figure
 381 S8C). Aromatic species were the next most common class detected related to +ESI ionization
 382 (Figure S7C, Figure S8). While aliphatic species were the second most common class observed in
 383 the -ESI analysis. This may be due to the ionization preference for relatively oxygenated
 384 functional groups (Figure S7).

385 Many molecular formulas identified in our study matched the molecular formulas of known
386 biomass burning organic aerosol chemical fingerprints found in the literature, including
387 levoglucosan, methoxy phenols, and organic acids (Bertrand et al., 2018; Brege et al., 2018;
388 Mazzoleni et al., 2007; Qi et al., 2020). Molecular formulas presented in the liquid smoke sample
389 detected by LC/FT-MS study were compared with an ambient wildfire influenced organic aerosol
390 sample, namely BBO6 (Brege et al., 2021). According to the pairwise comparison with the ambient
391 biomass burning organic aerosol sample, and liquid smoke, 97% of the total peak area (with 1036
392 out of 1098 total individual formulas) and 78% of the total peak area (with 510 out of 612 total
393 individual formula) matched with each other in +ESI and -ESI analysis respectively (Figure S9).
394 The carbon, DBE continuum, and degree of unsaturation of liquid smoke were analogous to those
395 trends of the ambient biomass burning aerosol in the literature (Brege et al., 2021). Though, unlike
396 the ambient sample the maximum frequency cluster of species were observed with relatively low
397 C number and DBE values ($C_8 - C_{12}$ and DBE 5-6). The C number above 32 or DBE above 15
398 formulas were not detected in the liquid smoke. This may be due to the removal of water insoluble
399 aromatics or condensed aromatics during liquid smoke production (Montazeri et al., 2013). The
400 species common between liquid smoke and the ambient biomass burning aerosol sample
401 correspond to only 38% and 48% of the ambient biomass burning organic aerosol peak abundances
402 in the positive mode and negative mode respectively (Figure S9). Still, ambient biomass burning
403 organic aerosol sample was more complex than liquid smoke. This could be due to the different
404 levels of oxidation, transformations, and the addition of other pollutants occurring in the natural
405 environment. Sample extracting solvent, and direct infusion analysis of the ambient biomass
406 burning organic aerosol sample also contribute to the relatively low percentage match of the
407 common species with respect to it.

408 In the research literature, most of the chemical reactions and transformations of organic
409 aerosol are carried out using single compounds or simple mixtures of species related to biomass
410 burning organic aerosol, such as methoxy phenols, vanillin, and levoglucosan-like compounds. (Li
411 et al., 2014; Rodriguez et al., 2022; Zhao et al., 2014). Even though these molecules are ideal for
412 targeted studies, the complex environmental behavior cannot be observed. Currently, no standard
413 reference material is available for the research and development of methods focused on organic
414 aerosol. According to our analysis, over 90% of the molecular formulas detected in the LC/FT-
415 MS analysis of liquid smoke were also detected in the ambient sample. Furthermore, the

416 characteristics and trends observed in the commercial liquid smoke sample show good agreement
417 with the biomass burning organic aerosol. As a result, commercial liquid smoke may be a valuable
418 substitute for organic aerosol research and method development. Liquid smoke is inexpensive and
419 easily obtainable in large quantities, making it an excellent candidate for further studies, including
420 inter-laboratory comparisons and characterization. These efforts may eventually lead to its
421 development as a standard reference material for biomass-burning organic aerosols.

422 **4. Conclusion**

423 LC/FT-MS nontargeted screening is a powerful semi-quantitative technique for
424 environmental complex mixture analysis. Adapting existing techniques, matching molecular
425 markers in databases, and finding representative samples for method development are some key
426 challenges. To overcome these problems, we present a novel approach to separate and identify the
427 compounds in biomass related organic aerosol using the surrogate mesquite liquid smoke. LC/FT-
428 MS data were processed, and molecular formulas were assigned using the combination of
429 MZmine2.53 and MFAssignR software tools.

430 Using the new approach, more than 1700 compounds in mesquite liquid smoke have been
431 identified with their isomers. Formula obtained from this approach showed a high consistency with
432 direct infusion LC/FT-MS results, confirming the accuracy of it. A linear correlation between the
433 logarithmic values of the chromatographic mass spectrometric peak area and direct infusion mass
434 spectrometric peak abundance was observed in the study favoring the use of relative quantitative
435 insight from direct infusion mass spectrometric peak intensity abundance. Over 90% of the
436 molecular formulas identified in mesquite liquid smoke matched with the ambient biomass burning
437 organic aerosol sample. So, commercial liquid smoke can be used as a suitable surrogate for
438 organic aerosol research and method developments. Despite the complexity of biomass burning
439 organic aerosol found in the literature, our study foresees the further complexity of organic aerosol
440 with their isomers. Thus, the method significantly improves the molecular identification of
441 biomass combustion aerosol components and obtains some semi quantitative insight. Further,
442 MS/MS fragmentation information can be used for tentative structural identification of the existing
443 compounds, but further studies need to be done for the structural confirmation.

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