

# Pore-scale Microenvironments Control Anthropogenic Carbon Mineralization Outcomes in Basalt

Charles T. Depp<sup>a\*†</sup>, Quin R.S. Miller<sup>a</sup>, Jarrod V. Crum<sup>b</sup>, Jake A. Horner<sup>b</sup>, H. Todd Schaefer<sup>a\*</sup>

<sup>a</sup>Physical and Computational Sciences Directorate, Pacific Northwest National Laboratory, Richland, Washington, 99354, USA

<sup>b</sup>Energy and Environment Directorate, Pacific Northwest National Laboratory, Richland, Washington, 99354, USA

<sup>†</sup>Current affiliation is Department of Chemistry, Colorado School of Mines, Golden, Colorado, 80401, USA

**KEYWORDS:** *carbon sequestration, carbonate geochemistry, carbonate structure, supercritical CO<sub>2</sub>, basalt reservoir, basalt geochemistry, geologic sequestration, climate change mitigation*

**ABSTRACT:** Thin sections and hand samples from 50 sidewall cores from the Wallula Basalt Pilot Demonstration, a basaltic carbon sequestration demonstration, provided the opportunity for in-depth analysis carbon mineralization induced by the injection of supercritical CO<sub>2</sub>. In this study we used optical petrography and scanning electron microscopy to characterize the physical and chemical characteristics of the basalt components influenced by carbon mineralization reactions from all available hand samples and thin sections within the three CO<sub>2</sub> injection zones and caprock flow interiors. We found extensive carbonate mineralization, mostly in the form of nodules that were shown to be chemically zoned; Ca-dominant in the core regions and Ca-bearing Fe-dominant in the outer regions. Carbonate mineralization also took the form of fracture filling carbonate cement and acicular aragonite were also observed. Overall, we clarified structural and paragenetic relationships between newly formed minerals, identifying a new fibro-palagonite-like, poorly crystalline silicate phase that grew on the carbonate nodules and pore-lining zeolites. We observed Fe-dominant carbonate precipitates surrounding acicular aragonite and rhombohedral Ca-carbonate cores, whereas previous studies of these zoned nodules did not observe these structures. A comprehensive accounting of the carbon mineralization products is vital to understand and predict the behavior of supercritical CO<sub>2</sub> in the subsurface given both the diversity of the host rock between and within injection zones, especially considering that the morphology and chemistry of the diverse precipitates and be influenced by pore-scale microenvironments of the basalt.

## INTRODUCTION

The Wallula Basalt Pilot Demonstration was initiated in 2009 when an injection well was drilled into the Miocene Columbia River Basalt Group (CRBG) to a total depth of 1,253 m at a paper mill site near the town of Wallula, Washington, located in southeastern Washington State<sup>1-4</sup>. Within the ~828–887 m below the ground surface injection interval, were three brecciated, porous flowtop zones, that underwent extensive hydrologic characterization before CO<sub>2</sub> injection<sup>2</sup>. Zone 1 was the thickest and most porous zone and received an estimated 80% of injected CO<sub>2</sub>, while Zone 2, being the next porous zone received and estimated 20% of injected CO<sub>2</sub><sup>5</sup>. Zone 3 had less than 1% porosity and received a comparatively miniscule amount of CO<sub>2</sub><sup>5</sup>. Zones 1, 2, and 3 are at depths of approximately 828–846 m, 854–862 m, and 866–875 m, respectively<sup>6</sup>. All three zones are capped by impermeable Umtanum member flows<sup>4</sup>. Zones 1 and 2 belong to the Indian Ridge member while Zone 3 is at the top of the Ortley member<sup>6-8</sup>. The basalt sequestration pilot borehole received ~977 tons of supercritical CO<sub>2</sub> over a 3-week period starting in August of 2013<sup>6,9</sup>. After 24 months, the well was re-entered to

collect samples to be analyzed for post-injection characterization data. The samples included 50 sidewall rotary cores from the injection interval. Examination of these sidewall cores (SWC) showed amber colored, carbonate nodules that were identified as ankerite, Ca(Fe, Mn)(CO<sub>3</sub>)<sub>2</sub> and were shown to be chemically and isotopically distinct from the naturally occurring pre-injection calcite veins<sup>6</sup>. These ankerite nodules are the first and only reported occurrence of anthropogenic CO<sub>2</sub> mineralization in basalt pores. This kind of mineralization was observed during laboratory studies<sup>10</sup> with CRBG samples exposed to scCO<sub>2</sub>, and one of few CO<sub>2</sub> injection studies in the world<sup>6,11</sup>, which includes injection into Icelandic basalt for the CarbFix Project. Together, these endeavors are demonstrating permanent carbon storage technologies that leverages rapid reactivity in mafic formations.<sup>12-18</sup>

The objective of this study is to examine those sidewall cores collected post-CO<sub>2</sub> injection from the Wallula Basalt Pilot Demonstration and to identify the associated mineralogy. We also documented lithological features in hand samples and thin sections to better understand the paragenesis occurring post-CO<sub>2</sub> injection

73 and led to construction of a petrographic roadmap, that  
74 will help future researchers identify areas of interest and  
75 contextualize samples in a broader lithological context.

## 76 METHODS AND MATERIALS

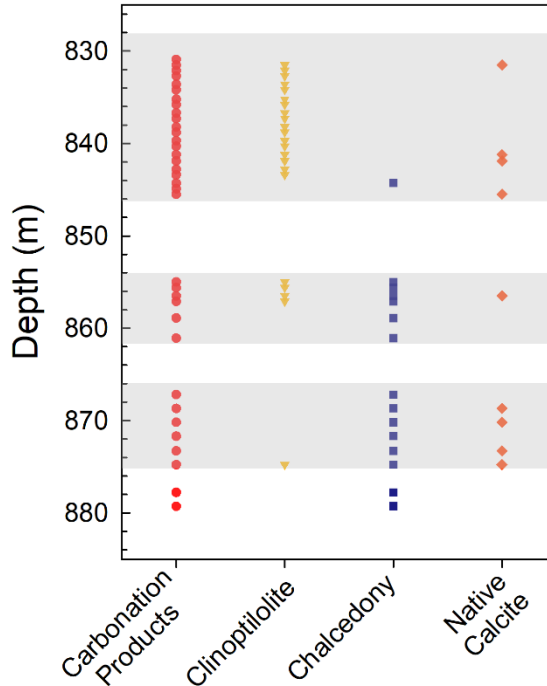
77 Twenty-four sub-samples were taken from the 50  
78 SWCs and sent to Intertek Westport Technology Center  
79 where thin sections were created. In total, 42 post-  
80 injection thin sections were created. To distinguish the  
81 carbonate composition, thin sections were stained with  
82 alizarin red S and potassium ferricyanide, causing the  
83 Ca-rich carbonate nodules to appear red and the Fe-rich  
84 to appear blue in thin section<sup>19,20</sup>. All depths expressed  
85 in this paper are below the ground surface.

86 For this present study, thin sections and hand  
87 samples were analyzed using a Leica DMLM polarizing  
88 microscope, a Clemex motorized stage with a JS-Tango  
89 joystick and Clemex Vision Lite and PE 8.1 image  
90 analysis software, and a Nikon Optiphot2-Pol with a  
91 0.5x C-mount adapter and a Jenoptik Gryphas AVIOR  
92 8MP USB3 digital microscope camera with Gryphax  
93 software. Hand samples also were examined using a  
94 Wild Heerbrugg optical microscope and a Cambridge  
95 Instruments StereoZoom7. Hand samples were  
96 analyzed uncoated at low vacuum (30 Pa) in the JEOL  
97 7001F SEM equipped with two opposing Bruker X-  
98 flash 6|60 EDS instruments. Samples were imaged  
99 using a backscattered electron detector. The EDS  
100 detectors were used to collect elemental spot analysis  
101 sample features and elemental maps were collected on  
102 the samples at an accelerating voltage of 15 kV and  
103 variable probe currents.

## 104 RESULTS

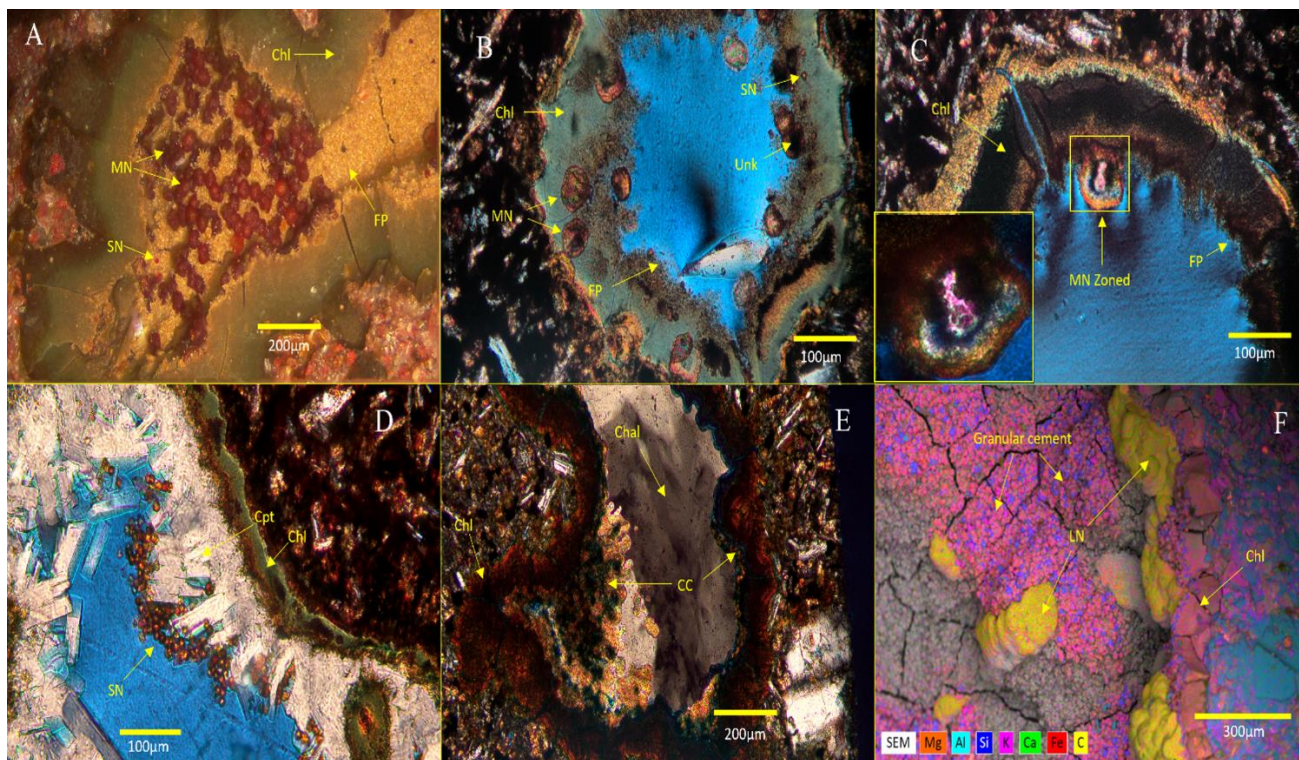
105 The results are presented sequentially starting with  
106 Zone 1. Each of these sections begins with the general  
107 characteristics of the host rock (i.e., porosity) and  
108 descriptions of thin section textures. This is followed by  
109 descriptions of notable minerals and cements, some of  
110 which serve as substrates for carbonate mineralization.  
111 Last, descriptions of the carbonate nodules themselves  
112 and how they fit into the broader context of the host rock  
113 are provided. This section ends with a discussion of  
114 important features regarding carbonation. A general  
115 overview of specific depths at which carbonation  
116 products were found, and the location of other important  
117 minerals commonly found in pores are represented in  
118 **Figure 1**.

119 **Zone 1.** (828–846 m) The shallowest and thickest  
120 injection zone (18 m thick) has the highest pre-injection  
121 porosity at 22%<sup>5</sup>. Visual inspection reveals abundant  
122 clinoptilolite and groundmass that varies in color from  
123 light gray at 828 m, down 3 m, then transitions red brown  
124 for the next 8 m, and light-dark gray in the last 7 m. Thin  
125 sections tend to have varying textures both within  
126 samples and among them. Many thin sections in Zone 1



**Figure 1.** Overview of key minerals, including carbonation products, clinoptilolite, chalcedony (including silica spherulites) and native (naturally occurring) calcite, identified in the recovered sidewall cores. The gray areas delineate the injection zones intervals.

127 contain multiple distinct textures, such as the sample  
128 from 837.3 m, seen in **Figure S1**. The multi-textured  
129 samples in Zone 1 typically range from hypocrySTALLINE  
130 regions, where the dominant phenocryst is plagioclase in  
131 a matrix dominated by glass, to cryptocrystalline where  
132 plagioclase is also dominant, but inequigranular, anhedral  
133 pyroxene is more common in this texture. The  
134 cryptocrystalline, intersertal groundmass is often an  
135 altered brown or red color, possibly due to devitrification  
136 or other alteration. These regions typically contain large  
137 plagioclase grains, small plagioclase grains, intersertal  
138 glass and small, relatively equigranular opaque minerals.  
139 The samples with the most heterogeneous textures are  
140 those from depths of 831.5, 832.7, 836.7, 837.3, 838.8,  
141 and 842.8 m. The rest are comparatively homogenous,  
142 and contain only one texture, which can generally be  
143 described as being comprised mostly of inequigranular  
144 plagioclase and pyroxenes, often contain opaque  
145 minerals and some glass. Plagioclase is the dominant  
146 phenocryst in almost all textures (most are subophitic or  
147 nesophitic) and typically is subhedral to anhedral,  
148 relatively equigranular in regions dominated by glass and  
149 inequigranular elsewhere. Some plagioclase grains are  
150 oikocrysts, containing Fe- and Ca-rich carbonate  
151 inclusions. Plagioclase-hosted carbonate inclusions are  
152 found in all three injection zones but are most common



**Figure 2.** **A:** (835.8 m). Optical image of small and medium red carbonate nodules and small orange carbonate nodules growing on fibro-palagonite (light green) in a chlorophaeite (dark green) coated pore. **B:** (835.2 m). Thin section PPL image of medium and small carbonate nodules growing in pores. **C:** (835.2 m). Thin section XPL image of a medium carbonate nodule exhibiting zoning. **D:** (832.7 m). Thin section PPL image of clusters of small carbonate nodules growing on clinoptilolite. **E:** (844.3 m). Thin section XPL image of Fe and Ca rich carbonate cement and chalcedony cement. **F:** (844.3 m). SEM image and EDS mapping of oblong carbonate nodules on granular cement. (SN = small carbonate nodule, MN = medium carbonate nodule, LN = large carbonate nodule, FP = fibro-palagonite, Unk = unknown, CC = carbonate cement, Chal = chalcedony, Chl = chlorophaeite, Cpt = clinoptilolite)

153 in Zone 3 (**Figure S2**). Within Zone 1, carbonate nodules  
154 were not observed above 831.5 m or below 845.5 m.

155 In Zone 1, the most common cements found growing  
156 from the edges of pores, and even entirely filling small  
157 pores, are green and green-brown chlorophaeite as seen  
158 in **Figure 2A**. On top of the chlorophaeite is a layer of  
159 fibro-palagonite (light green) and carbonate nodules (red  
160 and brown). In this paper, chlorophaeite refers generally  
161 to the chemically variable, clay mineral-like, green,  
162 yellow and brown, cement that is an alteration product  
163 that often coats pores and also is found in the groundmass  
164 of many samples and in all zones<sup>21</sup>. We use the term  
165 fibro-palagonite to specify a granular mat-like phase that  
166 is sometimes found growing on chlorophaeite<sup>22, 23</sup>. The  
167 chlorophaeite, which is rich in Mg, Fe, Si, and Al,  
168 according to EDS data (**Table S2-S4**), is present from a  
169 depth of 831.5 m down 10.7 m. The chlorophaeite in  
170 samples shallower than 835.2 m tends to be thinner.  
171 Between 835.2 m and 838.8 m, it is thicker and zoned  
172 with a thin outer brown layer followed by a thick inner  
173 green layer. In hand samples the chlorophaeite of  
174 different samples look nearly identical, although there are  
175 differences that are more easily distinguished in thin  
176 sections. **Figure 2B-E** demonstrate this variability.  
177 Cross polarized light (XPL) and plane polarized light

178 (PPL) images of **2B-E** all show the many colors and  
179 textures found in Zone 1, from the thick green cement of  
180 **2B**, thick zoned cement of **2C**, the thin green cement of  
181 **2D** and the brown cement of **2E**. The samples at depths  
182 839.7 m and 840.3 m have a thin cement that is yellow-  
183 brown in hand samples, but turquoise colored in thin  
184 sections in both XPL and PPL images. At depths from  
185 841.9 m through 843.4 m, there is a thin gray-green  
186 cement that, in thin sections, is like the cement observed  
187 in the thin section from 839.7 m. The hand sample from  
188 841.9 m also has a light green cement that fills small  
189 pores throughout the groundmass. The pores in the  
190 sample at depths of 844.3 m were coated with a thick  
191 brown layer of chlorophaeite and many of them are filled  
192 with chalcedony. In thin section, it is apparent that there  
193 is both an Fe-rich Ca-rich carbonate cement layer  
194 between the brown chlorophaeite and the chalcedony in  
195 many of the pores. The hand sample from 845.5 m was  
196 not very porous but did contain green chlorophaeite in a  
197 fracture, and some red-brown cement that potentially  
198 may be a carbonation product.

199 Clinoptilolite is ubiquitous throughout much of  
200 Zone 1 as one of the most common minerals growing in  
201 pores that is not a cement. In some samples, such as those  
202 from 835.8, 842.8 and 843.4 m, it dominates an estimated

203 50 and 75% of the pores. It is most abundant in samples  
204 at depths of 835.8, 839.7–840.3, and 842.8–843.4 m, and  
205 is at very low abundance at 841.2 m and practically  
206 nonexistent in Zone 1 beneath 843.4 m. It generally  
207 grows on chlorophaeite, and other pore coating cements  
208 as can be seen in **Figure 2D**, not directly on the  
209 groundmass. In pores where it is not present, there is  
210 often the yellow-green, fibro-palagonite mentioned  
211 earlier. This material is especially abundant from depths  
212 of 832.1 m down to 835.8 m, but it also is found  
213 throughout the next 3 m of depth beneath that. It does not  
214 appear to be more abundant than clinoptilolite in any  
215 sample in Zone 1. In some samples, such as the one from  
216 833.6 m, they occur together, but this is not common.  
217 Generally, the fibro-palagonite is observed growing on  
218 top of the clinoptilolite as shown in **Figure S3**. It is also  
219 worth noting that some fibro-palagonite appears to be  
220 growing on the carbonate nodules as well as can also be  
221 seen in **Figure S3-S4**, indicating the reprecipitation of  
222 this phase was induced by CO<sub>2</sub> injection.

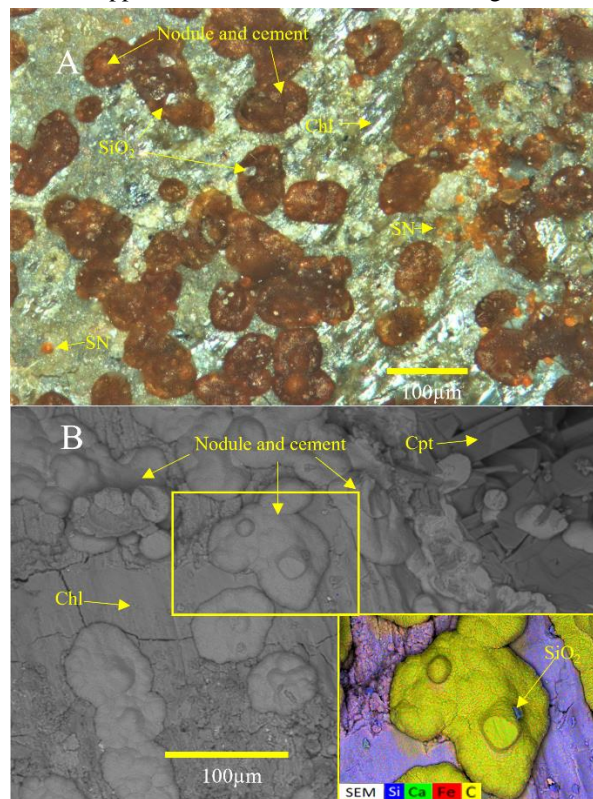
223 Chalcedony is present in Zone 1 in the sample from  
224 844.3 m, where it takes the form of pore filling  
225 chalcedony, and silica spherulites. In the thin section,  
226 shown in **Figure 2E**, the pores are coated with dark  
227 brown chlorophaeite, and most are filled with chalcedony  
228 and both Ca- and Fe-rich carbonate cement. In some  
229 pores, the carbonate cement and chalcedony appear  
230 intergrown, while in others the carbonate cement is  
231 between the chlorophaeite and chalcedony. In some  
232 clusters of carbonate nodules, there is also a granular  
233 form of SiO<sub>2</sub> per EDS that may be a coprecipitate (**Figure**  
234 **3B**).

235 The carbonate nodules in this zone are abundant yet  
236 small compared to other zones, yet still exhibit a wide  
237 range of sizes, colors, and habits. In general, they tend to  
238 range from small, orange, and vitreous, to medium, dark  
239 red, and dull, with many intermediates between these two  
240 endmembers (i.e., small, and red or medium and orange).  
241 The only hand sample with carbonate nodules that can be  
242 observed unaided was from 844.3 m. In Zone 1 the Ca-  
243 to Fe-rich zoning is not really visible in the small  
244 carbonate nodules, possibly due to their size (if it even  
245 occurs at all) but can be observed in some medium-sized  
246 nodules such as the one in **Figure 2C**. In the center of the  
247 nodule is the characteristic pink color shown by Ca-  
248 rich carbonate minerals, followed by blue colors as the Fe  
249 becomes the dominant cation. The carbonate nodules  
250 occur in three size groups: small, medium, and large. The  
251 diameters of the small nodules are <35 μm, the medium  
252 nodules are <200 μm and the large are ≥200 μm. The  
253 distinction between small and medium initially was made  
254 to distinguish extremely small nodules that often grew  
255 with the zeolite mineral clinoptilolite from larger nodules  
256 that tended to grow in pores without clinoptilolite.

257 Pores dominated by fibro-palagonite are often  
258 observed with many medium, red, opaque carbonate  
259 nodules on them. Other sizes and colors also grow on it

260 but are more difficult to see (**Figure S5**). **Figure 2A**  
261 shows a pore surface with medium-sized reddish  
262 carbonate nodules on top of the light green pore-lining  
263 fibro-palagonite. **Figure 2B** is a PPL image of medium  
264 and some small carbonate nodules growing on  
265 chlorophaeite, with what appears to be fibro-palagonite.  
266 The only carbonate nodules that commonly grow on  
267 clinoptilolite are small, orange, and vitreous (**Figure S6**)  
268 and are usually either dispersed throughout, or growing  
269 in botryoidal clusters as in **Figure 2D**. The average  
270 carbonate nodule size in this cluster is approximately 8.8  
271 μm, and most are Ca-rich (based on dye, not EDS). This  
272 can be observed at depths of 838.8 and 831.5 m but are  
273 found throughout the shallower half of this zone.

274 There were some unique morphologies observed in  
275 this zone. The large carbonate nodules from 844.3 m  
276 have a unique shape. They are represented in the yellow-  
277 orange color in the EDS chemical map in **Figure 2F**.  
278 They have an elongate shape, botryoidal texture, are  
279 composed of many carbonate nodules and are growing on  
280 a granular cement composed of silica spherulites and  
281 other silicate minerals that contain Mg, and Fe (**Table**  
282 **S5**). In some samples, such as shown in **Figure 3A**, the  
283 carbonate material looks like an intermediate between a  
284 nodule and cement. In **Figure 3A**, the circular looking  
285 material appears to be flattened or smeared together in

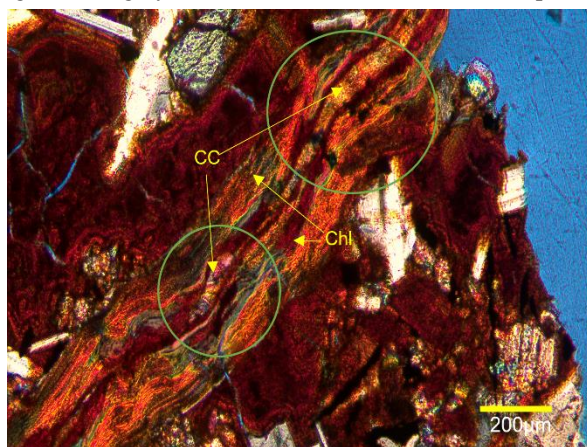


**Figure 3. A:** (838.2 m). Optical image of small and medium carbonate nodules exhibiting a morphology that is a mix of cement and nodules. **B:** (838.2 m). SEM image of cement/carbonate nodule morphology and an EDS chemical map. Abbreviations here are the same as in **Figure 2**.

286 some places, similar to a “pancake” siderite  
287 morphology.<sup>24, 25</sup> In some examples, carbonate nodules  
288 can be seen growing from the circular smear. These  
289 nodules tend to be red to orange in color, and  
290 measurements of 131 of these carbonate nodules/cements  
291 revealed a mean diameter of 60 μm, with a minimum and  
292 maximum of 25 and 111 μm respectively. **Figure 3B** is  
293 an SEM image and EDS map of the morphology shown  
294 in **Figure 3A**. The Ca- and Fe-rich carbonate material is  
295 represented by the yellow color, while the chlorophaeite  
296 is purple. This material is common from depths of 838.8  
297 m and shallower.

298 Calcite also was found in Zone 1, particularly in the  
299 thin section from 831.5 m and the hand sample from  
300 841.9 and 843.4 m. It is unclear whether this calcite was  
301 naturally occurring, however, the calcite from the hand  
302 sample from 841.9 m had some red carbonate cement on  
303 its edges, so it must have occurred before the carbonation  
304 products associated with the project (**Figure S7**).

305 **Zone 2.** (854–862 m) Zone 2 has a thickness of 8 m  
306 and a pre-injection porosity of 13%<sup>5</sup>. The samples in this  
307 zone are less obviously altered than the samples from  
308 Zone 1. Most samples from 856.5 m to the bottom of the  
309 zone are a light gray color, and the groundmass in the  
310 corresponding thin sections are very homogenous with  
311 little alteration compared to samples from 855.6 and 855  
312 m and those from Zone 1. Overall, the Zone 2 thin  
313 sections are relatively homogenous except for the  
314 samples from 855.6 and 855 m (**Figure S1**). The thin  
315 sections from 855 m have three different textures, all of  
316 which are hypocrystalline; however, two are composed  
317 mostly of glass and the other has some cryptocrystalline  
318 matrix. Of the two composed of glass, they can be  
319 distinguished by color, one has a red hue (that is visible  
320 on the thin section itself) and the other is black. The  
321 mineralogy of these two textures is nearly identical, with  
322 plagioclase being the main phenocryst, with some  
323 pyroxenes present. The third texture has only intersertal  
324 glass on a gray-brown matrix. The cements in the pores

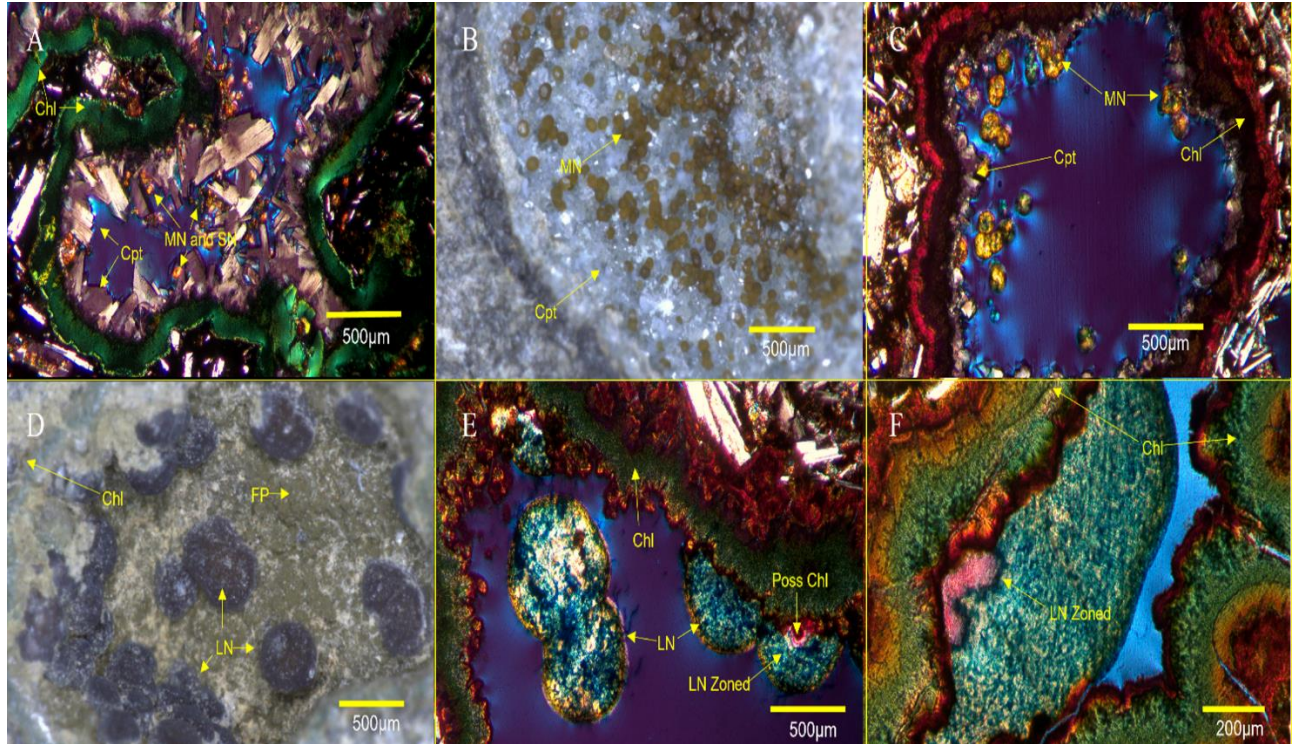


**Figure 4.** (858.9 m). Thin section PPL image of a fracture containing carbonate cements in the green circles. Abbreviations here are the same as in **Figure 2**.

325 also vary with different textures. At 855.6 m, there are  
326 two different matrices distinguished by either red or black  
327 colors. There are mostly small subhedral to anhedral  
328 plagioclase and pyroxene grains dispersed throughout the  
329 groundmass. The rest of the thin sections are very  
330 homogenous and almost identical to each other in terms  
331 of mineralogy. They consist of large and small  
332 plagioclase grains with interstitial pyroxene, and opaque  
333 minerals, also intersertal glass and brown chlorophaeite.  
334 No carbonate nodules were found below depths of 861.1  
335 m.

336 There are cements similar to those found in Zone 1,  
337 although there are also many instances of unique cement.  
338 At depths of 855 m and 855.6 m, there is a darker green,  
339 almost turquoise cement. In the corresponding thin  
340 section for the sample taken at 855 m, this cement is  
341 bright green and vibrant in XPL and PPL, although in the  
342 red, glassy region, the cement is duller and thicker.  
343 Overall, the hand sample from 855.6 m is not ideal for  
344 most notable features are on the side of the sample where  
345 the core was cut so the features are difficult to see. The  
346 corresponding thin section for this sample contains a  
347 unique material. The pore spaces are mostly filled with  
348 some material that is opaque in XPL and exhibits a range  
349 of different colors in PPL. The hand sample from 856.5  
350 m has a cement that is like the thick brown cement found  
351 at 844.3 m and is found growing around chalcedony  
352 likewise. The hand sample from 857.1 m has a cement  
353 that is turquoise but shows zoning. The outside is yellow,  
354 and in some cases, there is another yellow layer on the  
355 inner most part. For PPL in thin section, the yellow layer  
356 is very visible. In both XPL and PPL, the cement is a  
357 bright blue color that is unique to this sample. The  
358 sample from 858.9 m is not very porous compared to the  
359 other samples, however there are some fractures, most of  
360 which are filled with multiple materials that are easily  
361 distinguished in thin section. The outer-most part of the  
362 fracture is a layer of chalcedony, followed by some other  
363 cement that, in thin section, is bright green and orange in  
364 XPL and PPL. Finally in the center of the fracture (in  
365 some sections) is a carbonate cement that has regions of  
366 that are Fe-rich and some that are Ca-rich (**Figure 4**). In  
367 the thin section there are also regions where carbonate  
368 cement is found intergrown with chalcedony.

369 Clinoptilolite is abundant at 855 m and is similar to  
370 Zone 1 in this regard; however, the rest of Zone 2  
371 generally has less clinoptilolite than most Zone 1  
372 samples. From 855 m, the general trend is a decrease in  
373 clinoptilolite with depth to 858.9 m where none was  
374 observed in either hand sample or thin section. At 855.6  
375 m there is a small amount that can be seen in the hand  
376 sample, but none was observed in thin section, possibly  
377 due to the yellow cement dominating most pores. A large  
378 amount of the pore space in the 856.5 m hand sample is  
379 dominated by chalcedony; although, in thin section,  
380 clinoptilolite grains present in the same pores look like  
381 the chalcedony grew around them. The hand sample does



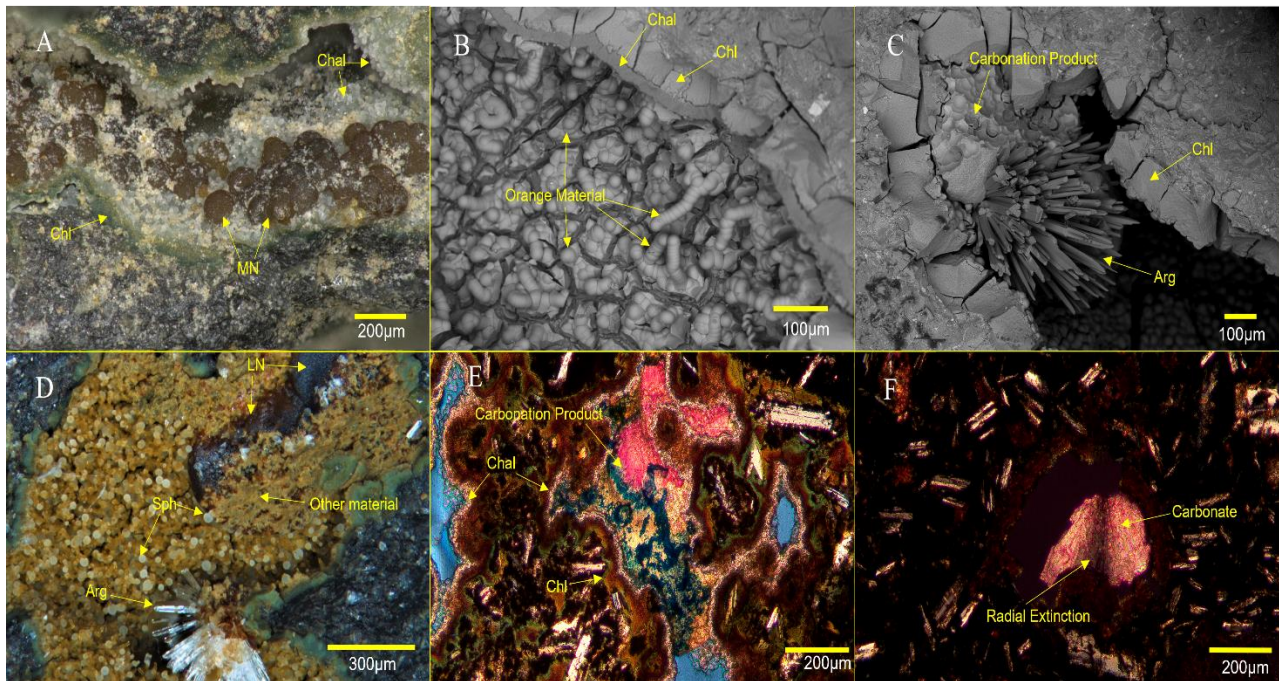
**Figure 5.** **A:** (855 m). Thin section XPL image of medium carbonate nodules growing in clinoptilolite on a unique vibrant green cement. **B:** (856.5 m). Optical image of medium carbonate nodules growing in a clinoptilolite dominated pore. **C:** (856.5 m). Thin section XPL image of medium carbonate nodules and clinoptilolite. **D:** (857.1 m). Optical image of large carbonate nodules growing in pores with fibro-palagonite. **E and F:** (857.1 m) Thin section XPL and PPL images (respectively) of large carbonate nodules growing in pores. Zoning is especially visible in F. Abbreviations here are the same as in **Figure 2**.

382 contain a few pores that are dominated by clinoptilolite,  
 383 but there are also a few large pores that contain only  
 384 fibro-palagonite. The hand sample from 855.6 m is the  
 385 first one that appears to contain more fibro-palagonite  
 386 than clinoptilolite.

387 The sample from 855 m possibly has the most  
 388 diverse size distribution of carbonate nodules of the entire  
 389 project area (**Figure S8**). There are small and medium  
 390 carbonate nodules, growing individually and in clusters,  
 391 usually in the same pores as clinoptilolite as can be seen  
 392 in **Figure 5A**. Some medium carbonate nodules grow  
 393 around clinoptilolite grains and some exhibit zoning.  
 394 Large carbonate nodules also are present, but typically  
 395 grow by themselves or with other large nodules in open  
 396 pores without clinoptilolite. Most of the carbonate  
 397 nodules observed in the hand sample are medium and  
 398 small and orange and brown in color. In the hand sample  
 399 from 855.6 m, there are what look like medium and small  
 400 carbonate nodules that range in color from orange to dark  
 401 red, but this is difficult to tell because of the issue with  
 402 the hand sample mentioned earlier. No carbonate  
 403 nodules were observed in the corresponding thin section.  
 404 Almost all carbonate nodules from 856.5 m are visible  
 405 unaided. The carbonate nodules that grow in the  
 406 clinoptilolite pores, as seen in **Figure 5B** and **C**, are  
 407 mostly medium-sized and grow in clusters and

408 individually, although compared to most of the carbonate  
 409 nodules observed in Zone 1 are still massive. The largest  
 410 carbonate nodules are found in a pore containing brown-  
 411 yellow fibro-palagonite. The carbonate nodules here are  
 412 brown to orange in color. The carbonate nodules growing  
 413 in the fibro-palagonite, which can be seen in **Figure 5D**  
 414 and **E** are darker than the ones growing in clinoptilolite,  
 415 which are more orange in color and appear slightly less  
 416 opaque. In thin section, it appears the carbonate nodules  
 417 are more likely to grow on chlorophaeite that is light  
 418 brown or green, compared to chalcedony and  
 419 clinoptilolite that grow mostly on brown chlorophaeite.  
 420 The sample from 857.1 m has carbonate nodules that are  
 421 comparable in size to those taken at 856.5 m. The large  
 422 carbonate nodules here grow almost exclusively on fibro-  
 423 palagonite. There also are some very small, orange  
 424 carbonate nodules in the fibro-palagonite; however, these  
 425 are difficult to see given the nature of the substrate and  
 426 the size of the nodules.

427 Chemical zoning in these larger carbonate nodules,  
 428 such as shown in **Figure 5F**, is much easier to observe  
 429 and is common in most of the large nodules. Medium  
 430 carbonate nodules in Zone 2 show chemical zoning more  
 431 commonly than in Zone 1, where a small number of  
 432 nodules exhibit chemical zoning, some are ambiguous,  
 433 and others do not appear to be chemically zoned at all.



**Figure 6.** **A:** (870.2 m). Optical image of medium carbonate nodules covered in drilling mud (tan) growing on botryoidal chalcidony cement. **B:** (873.3 m). SEM image of orange, vitreous cement growing in a pore. **C:** (873.3 m). SEM image of one instance of aragonite. **D:** (870.2 m). Optical image of semi-spherical carbonate nodules with silica spherulites and some tan material growing on the open face and another instance of aragonite. **E:** (870.2 m) Thin section PPL image of an oddly shaped carbonation product **F:** (873.3 m). Thin section XPL image Ca-rich carbonate material that is angular and has a radial extinction. (Sph = silica spherulite). Other abbreviations are the same as in **Figure 2**.

434 In the hand sample, there is a clinoptilolite-bearing pore  
 435 with small and medium carbonate nodules ranging in  
 436 color from orange to dark red, and many of the nodules  
 437 are covered and enveloped in a tan cement (**Figure S9**)  
 438 that has not been identified. The hand sample from 858.9  
 439 m is much less porous than shallower samples, and only  
 440 contained clusters of small, orange carbonate nodules that  
 441 appear to grow in the fractures, but most carbonate  
 442 material was the cement mentioned earlier. No carbonate  
 443 nodules were observed in this thin section, only cement.

444 **Zone 3.** (866–875 m) Zone 3 has a thickness of 9 m  
 445 and a pre-injection porosity of 1%<sup>5</sup>. This zone has a  
 446 unique variety of cements and accessory features not  
 447 found in any of the other zones. Overall, this zone  
 448 appears to have experienced more alteration than Zone 2.  
 449 This degree of alteration is apparent on the mostly red  
 450 hand sample from 867.2 m but is much more apparent in  
 451 many of the thin sections that appear mottled and have  
 452 matrices of varying colors. Plagioclase is once again the  
 453 dominant phenocryst. Like Zone 1, many of the samples  
 454 exhibit multiple textures, ranging from mostly glass to  
 455 mostly altered, cryptocrystalline groundmass, with  
 456 intersertal glass; however, overall, the texture mostly is  
 457 hypocrySTALLINE. Samples from shallower depths (867.2  
 458 and 870.2 m) contain the most glass and contain very  
 459 little pyroxene. Large grains of plagioclase also are rare.  
 460 Samples from 873.3 and 874.8 m are much more

461 crystalline. Those samples contain pyroxene and large  
 462 grains of plagioclase are much more common. Samples  
 463 from 887.8 m are almost entirely crystalline, with some  
 464 intersertal glass and opaque minerals present.

465 In the hand sample from 867.2 m, there is a light  
 466 green cement. Additionally, there is a gray-white  
 467 botryoidal chalcidony cement that is also present in  
 468 samples through 874.8 m. At a depth of 1.5 m lower,  
 469 there are three different cements: a dark green pore-  
 470 coating cement that is present through the sample at 871.7  
 471 m, the same gray-white botryoidal chalcidony mentioned  
 472 above, and the same light green cement (**Figure 6A**). In  
 473 thin section, the green cement is a bright lime green in  
 474 both XPL and PPL, with undulose extinction. At 873.3  
 475 m, most pore-lining cement is relatively thin, brown  
 476 chlorophaeite that is dark brown in XPL and light brown  
 477 in PPL, but there is also the spherulitic chalcidony  
 478 cement. There is an orange translucent, vitreous material  
 479 that grows on top of the chalcidony in some areas. SEM  
 480 revealed that this material is composed of a layer of small  
 481 carbonate nodules. They display a unique morphology  
 482 where they are growing on top of one another into the  
 483 pore space and have a worm-like texture to them as seen  
 484 in **Figure 6B** (chemical data shown in **Table S7**). There  
 485 is a similar looking material in the sample from 874.8 m  
 486 (**Figure S10**). At this same depth, there are thin brown,  
 487 and yellow cements, also clinoptilolite can be found

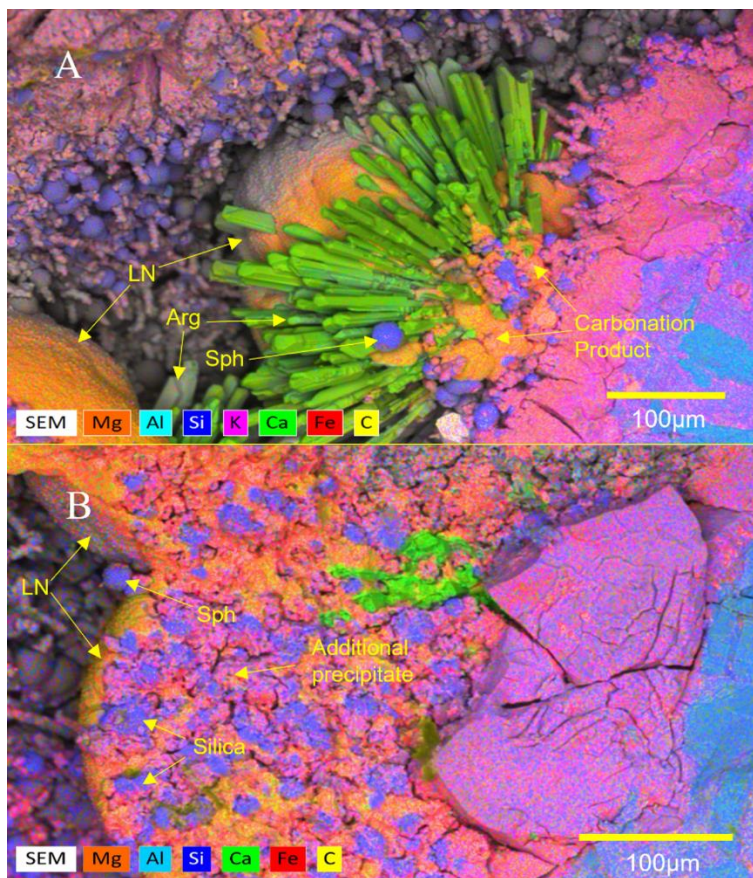
488 growing in pores and fractures. This is the  
489 only sample in Zone 3 in which clinoptilolite  
490 was found. Silica spherulites were observed  
491 growing among the clinoptilolite grains in  
492 thin section.

493 The area from 870.2 m to the bottom of  
494 the zone is one of the most chalcedony and  
495 silica rich regions in the entire target area and  
496 is distinct from most of the chalcedony and  
497 silica observed in other zones. Here, it is  
498 coating the insides of pores and is botryoidal  
499 and spherulitic chalcedony, in contrast to the  
500 silica observed, for example, in samples from  
501 around 856.5 m. In that region, chalcedony  
502 tends to fill the pores in such a way that the  
503 original orientation of the rock can be  
504 distinguished. In the thin section from 858.9  
505 m, there is an instance of silica spherulites  
506 and the hand sample for 855 m has botryoidal  
507 chalcedony cement, but it is significantly less  
508 abundant.

509 Four instances of aragonite also were  
510 found in hand samples in this zone (**Figure**  
511 **6C** and **D**): three from the sample from 870.2  
512 m and one from 873.3 m. In all four instances  
513 there are other carbonation products growing  
514 on and around the aragonite grains (**Table**  
515 **S6**). We believe that the aragonite is  
516 anthropogenic in origin, as the occurrence of  
517 aragonite in basaltic carbonation reactions  
518 has been demonstrated in laboratory  
519 experiments <sup>26</sup>.

520 In both the hand sample and thin  
521 section from 867.2 m, there are  
522 filamentations algal-bacterial features that  
523 are yellow-brown in the hand sample and  
524 opaque in the thin section. There also are relatively large  
525 pieces of semi-transparent naturally occurring calcite in  
526 samples from 868.7 m and 870.2 m. In the thin section,  
527 the calcite is pink, large, and tabular and fills fractures  
528 and vesicles. On some edges, there are Fe-rich carbonate  
529 nodules that have grown on the calcite. Almost all the  
530 calcite grains go extinct at the same angle, denoting that  
531 they have the same crystallographic orientation.

532 In terms of size, the carbonate nodules in Zone 3 are  
533 more like in Zone 2 than those in Zone 1. In the sample  
534 from 867.2 m, there are large, brown, spherical minerals  
535 found in the red groundmass of the sample that look like  
536 carbonate nodules, but EDS revealed them to be a  
537 ferromagnesian silicate instead. Large, dark red  
538 carbonate nodules were found growing around the  
539 algal/bacterial filaments as can be seen in (**Figure S11**).  
540 This also can be observed in the thin section. There also  
541 are smaller carbonate nodules here, found growing on the  
542 light green cement. The sample from 870.2 m has large,  
543 dark red carbonate nodules, many of which are half-



**Figure 7. A:** (870.2 m). SEM image and EDS chemical map of two instances of aragonite and carbonate nodules in a pore. Silica spherulites grow on and between aragonite crystals, possibly a product of the carbonation reaction. **B:** (870.2 m). SEM image and EDS map of a semispherical carbonate nodule with silica spherulites, Si-rich material and some other Fe-, Mg-, and Si-rich material growing on top of it. Abbreviations here are the same as in **Figure 2**.

544 formed semi-spheres, and some even have silica  
545 spherulites and another mineral growing on the flat,  
546 unformed half as can be seen in **Figure 6D** (additional  
547 photos in **Figure S12** and other chemical data in **Table**  
548 **S8**). At 873.3 m, there are a few large carbonate nodules  
549 in the hand sample, most of which have grown on the  
550 botryoidal chalcedony, but the bulk of the carbonate  
551 material is more easily examined in thin section. The  
552 carbonate nodules have a light orange-brown color, and  
553 almost appear to have an inner gray-white portion in the  
554 center, but that is difficult to say for sure.

555 Zone 3 contains some unique morphologies that are  
556 exclusive to this zone and best observed in the thin  
557 sections (**Figure S13**). Here, however, most of the pore  
558 space looks more like irregular, angular fractures as  
559 shown in **Figure 6E**. There are many instances of Ca-  
560 rich, carbonate material (no outer Fe-rich area) that fill  
561 entire small fractures. However, some of this material  
562 has a radial shape rather than the tabular shape observed  
563 in the native (naturally occurring) calcite but unlike the

564 typical rounded nodules observed elsewhere, it has  
565 angular edges (**Figure 6F**). When the stage is rotated,  
566 some of this material exhibits radial extinction, and in  
567 other instances, it appears to be composed of many  
568 crystals that all go extinct at different angles. Although  
569 the origin of these Ca-rich carbonates is unclear, previous  
570 laboratory based basalt carbonation experiments did  
571 produce calcites with a radiating texture<sup>26</sup>. There are also  
572 rounded, Fe-rich carbonate nodules in these samples, and  
573 in some instances these nodules are attached to the ends  
574 of, or intergrown with, the pink carbonate material (e.g.,  
575 the sample from 870.2 m).

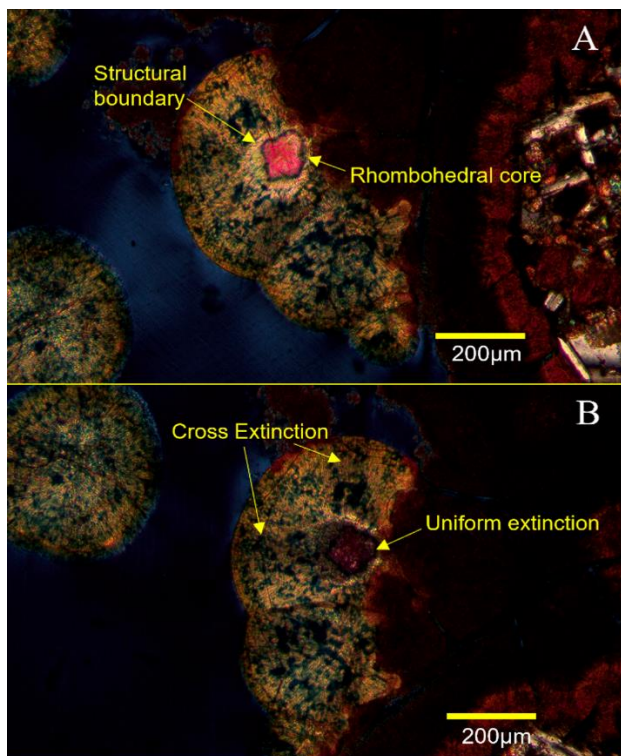
## 576 DISCUSSION

577 **Key Findings.** Each zone contained unique  
578 carbonation products (although there were similar  
579 products as well) that distinguish them from each other.  
580 One apparent difference between zones is the  
581 disappearance of botryoidal clusters of small carbonate  
582 nodules commonly observed in Zone 1 as depth  
583 increases. There are few of these clusters observed in  
584 Zones 2 and 3, although clusters of medium-sized  
585 carbonate nodules were observed in all zones. On the  
586 other hand, there are no large carbonate nodules observed  
587 in Zone 1 thin sections and only one hand sample in  
588 which they were observed (844.3 m). In Zone 1 only  
589 small carbonate nodules tend to grow in the same pores  
590 as clinoptilolite, while medium and small carbonate  
591 nodules in the same sample will grow on pores with fibro-  
592 palagonite. The circular smears of **Figure 3** along with  
593 some other products in **Figure S5**, only tend to grow in  
594 specific samples, on specific green chlorophaeite and not  
595 on fibro-palagonite or clinoptilolite. This green  
596 chlorophaeite cement found in much of the shallower half  
597 of this Zone 1 appears to be the best surface for growing  
598 a variety of different carbonate nodules and carbonate  
599 cement in Zone 1. The oblong carbonate nodules from  
600 844.3 m are another example. These carbonate nodules  
601 grow on a granular cement composed of silica spherulites  
602 and some other material (**Figure S5**) that is unique to that  
603 sample.

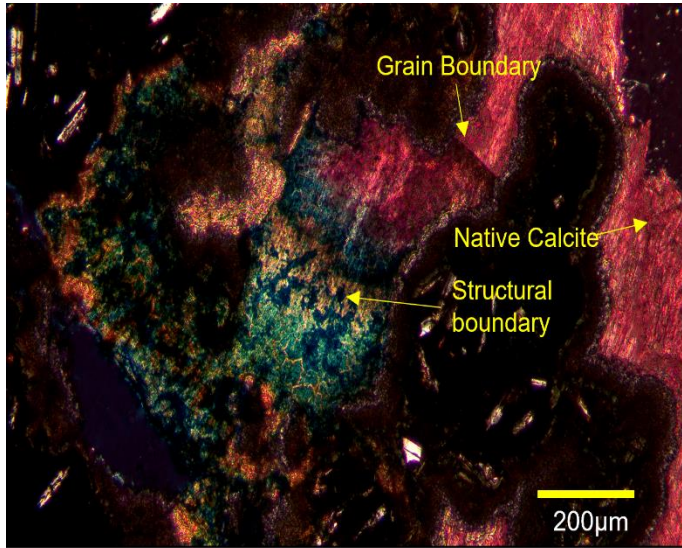
604 In order to understand the behavior of CO<sub>2</sub> in the  
605 subsurface it is important to identify any co-precipitates  
606 that formed during carbonation. In Zone 1 it is common  
607 to find small silica-rich materials growing on or near  
608 carbonate nodules, and can be seen in **Figure 3A** and **B**.  
609 Another instance of possible carbonate and silica co-  
610 precipitation in Zone 1 are the carbonate cements in the  
611 sample from 844.3 m, in which the carbonate cement  
612 appears closer to the edge of the pore, suggesting it may  
613 have occurred first, and in some cases, it even appears to  
614 be intergrown. In Zone 2, small and medium carbonate  
615 nodules appear to be covered or enveloped in a tan  
616 cement mentioned earlier. In Zone 3, silica spherulites  
617 and other Si-rich material is observed to be growing  
618 around the aragonite and carbonate nodules (**Figure 7A**).  
619 As mentioned in the **Zone 3** section, we believe the  
620 aragonite to have formed before the carbonate nodule,

621 and there are silica spherulites present on both. There are  
622 also large carbonate nodules that are half formed with  
623 silica spherulites and some other tan-colored material  
624 growing on its unformed face, shown in **Figure 7B**,  
625 providing evidence that the reaction is producing silica  
626 from reacting with the basalt mass.

627 In all three flow top targets there are carbonate  
628 nodules that exhibit chemical zoning. In thin section it  
629 appears that this is not a gradual change, and the Ca-rich  
630 centers can be seen with a distinct line of blue dye around  
631 its edges, beyond which no more pink is observed.  
632 There appears to be no blending of the two in most cases.  
633 There are some rare instances in Zones 2 and 3 where  
634 there does appear to be a purple color that could represent  
635 a transition area. It is not clear if zoning occurs at all in  
636 small carbonate nodules, but it is visible in many of the  
637 medium nodules although most visible in the large ones.  
638 The small, clustered carbonate nodules in Zone 1, appear  
639 to be overwhelmingly red in color (**Figure 2D**). In the  
640 EDS data seen in **Figure 3B** it appears there is an even  
641 distribution of Fe and Ca throughout, with no anomalous  
642 concentrations observed. The blue carbonate nodules  
643 may be zoned and, therefore just difficult to see. Some  
644 medium and, in some cases, large carbonate nodules also  
645 show only one color. Medium carbonate nodules can be



**Figure 8.** **A:** (856.5 m). Thin section XPL image of a large carbonate nodule positioned 45 degrees from extinction to show the rhombohedral shape of the core and the structural boundary. **B:** The same carbonate nodule, rotated to show extinction features (cross extinction in outer Fe-rich area, uniform extinction in the core).



**Figure 9.** (870.2 m). Thin section XPL image of a carbonate nodule growing on a native calcite grain.

646 either entirely blue or entirely red (**Figure 2B**), while  
 647 large ones have been observed to be only blue. For the  
 648 large and medium blue carbonate nodules, this could just  
 649 be the result of how the thin section was cut and the fact  
 650 that the dye is only surficial<sup>20</sup>.

651 The Ca-rich portion of chemically zoned carbonate  
 652 nodules exhibit a range of morphologies themselves.  
 653 Many of them, and a small part of the Fe rich region that  
 654 immediately surrounds it that makes up the core of the  
 655 carbonate nodule, often appear to be rhombohedral in  
 656 shape (**Figure 8A** and **Figure S16**), or an irregular shape.  
 657 The differences in shape may likely arise from how the  
 658 thin sections were cut, and it is possible that they are all  
 659 relatively rhombohedral. Other than rhombohedral cores,  
 660 aragonite also sometimes serves as the Ca-rich core  
 661 around which the Fe-dominated carbonate grows. This  
 662 can be observed in SEM in **Figure 6C** and **Figure 7A** and  
 663 in thin section in **Figure S15**. Aragonite cores appear far  
 664 less common than rhombohedral cores, with only one  
 665 instance being observed in thin section compared to  
 666 countless instances of rhombohedral cores. There is a  
 667 difference in optical extinction between the core and  
 668 outer portions that can be observed in chemically zoned  
 669 carbonate nodules. The core (Ca-rich and inner Fe-rich  
 670 area) exhibits uniform extinction, suggesting that it is one  
 671 crystal, or possibly multiple large crystals all oriented in  
 672 the same direction (**Figure 8B**). The outer Fe-rich parts,  
 673 on the other hand, typically exhibit undulose extinction,  
 674 although it can sometimes be difficult to see at all;  
 675 extinction crosses can also be exhibited (**Figure 8B**).  
 676 Overall, this suggests that the outer part of the carbonate  
 677 nodule is crystallographically distinct from the core. The  
 678 extinction crosses suggest they have a spherulitic texture,  
 679 which is composed of radially oriented crystallites<sup>27, 28</sup>,  
 680 suggestive of rapid growth at supersaturated conditions<sup>29</sup>.

1 30. This texture can be observed in thin section  
 2 (**Figure S17**) and SEM images (**Table S5** and **Table**  
 3 **S6**). **Figure 9** is an excellent example of the structural  
 4 and compositional diversity the carbonate nodules  
 5 exhibit. The larger Ca-rich region on the right is a  
 6 piece of native calcite. The grain boundary between  
 7 the carbonate nodule and the calcite is visible, and  
 8 when rotated, the Ca-rich part of the carbonate nodule,  
 9 and the inner Fe-rich part go extinct with the larger  
 10 calcite grain, possibly representing syntaxial growth  
 11 with that grain and a similar crystal lattice (unlike the  
 12 products growing on non-carbonate substrates), or  
 13 identical mineral phase, and the rest of the outer  
 14 carbonate nodule exhibits extinction crosses. This  
 15 may suggest that regardless of what the specific  
 16 mineral identity of the Ca-rich core is, the Fe-rich  
 17 carbonate crystallites still show a tendency to grow  
 18 from them.

695 **Comparisons with previous studies.** Each zone  
 700 was unique regarding its physical characteristics as  
 701 well as mineralogical makeup and degree of alteration.  
 702 Each zone also yielded unique carbonation products as  
 703 mentioned earlier. It seems likely that controls are  
 704 exerted by the host rock that influences the size, shape  
 705 and composition of the carbonation products. Studies of  
 706 carbonate morphologies and growth in soils have shown  
 707 a dependence on matrix composition and texture, and  
 708 particularly the amount of clay in the environment<sup>31</sup>. It  
 709 also has been shown that the presence of specific clays,  
 710 such as montmorillonite and other smectites, can affect  
 711 the nucleation, growth, and even the specific phase of  
 712 CaCO<sub>3</sub> that precipitates<sup>32 33 34</sup>. It is possible that similar  
 713 effects occur in basalts with different textures and  
 714 compositions, especially given the abundance of clay  
 715 mineral-like chlorophaeite from prior alteration which  
 716 varies, sometimes significantly, from sample to sample  
 717 (and also within the same samples) (**Figure S14**). It's  
 718 also been shown in lab experiments that cation  
 719 substitution varied between different basalts (i.e., Deccan  
 720 flood basalt and CRBG) and also that some samples had  
 721 distinct chemical boundaries, some had none at all and  
 722 others had gradual compositional changes<sup>35</sup>. The  
 723 porosity of the host rock is also highly variable between  
 724 zones, which could also influence nucleation and growth.  
 725 One important feature of a geologic carbon storage  
 726 project is injectivity, which is directly related to the  
 727 thickness and permeability of the host rock<sup>36</sup>. Zone 1,  
 728 being the thickest and most permeable zone, received  
 729 most of the injected CO<sub>2</sub> and as mentioned earlier, had a  
 730 tendency toward abundant, small and medium carbonate  
 731 nodules compared to the other zones. Some models of  
 732 nucleation and growth in a porous media show higher  
 733 initial concentrations can have lower inductions times<sup>37</sup>.  
 734 Other models have found higher reservoir conductivity  
 735 can significantly increase carbonate mineralization in  
 736 basalt<sup>38</sup>. It is possible due to the higher concentrations,  
 737 that there was a period of rapid nucleation, followed by

738 transport of CO<sub>2</sub> further from the inlet and hence no  
739 opportunity for larger nodules to grow.

740 Some experimental data shows and stoichiometric  
741 models predict that silica as well as aluminosilicates are  
742 produced in carbonation reactions with basalt<sup>15, 26, 39-42</sup>.  
743 Therefore, the co-precipitation of silica and other  
744 materials observed in several instances in all three zones  
745 suggests that the supercritical CO<sub>2</sub> did react with and  
746 dissolve the host rock, rather than just precipitating out of  
747 solution or from dissolving and reprecipitating  
748 preexisting carbonate minerals.

749 Carbonate minerals associated with carbonation of  
750 basalts that have distinct chemical boundaries are not  
751 unique and have been observed in multiple laboratory  
752 studies<sup>35, 42, 43</sup>. It is also important to note the non-zoned,  
753 Ca-rich carbonates of Zone 3. Although they are unique  
754 among the three zones, Ca-rich carbonates like calcite  
755 were found experimentally in multiple laboratory  
756 carbonation reactions and were also reported by  
757 Carbfix<sup>16, 40, 44</sup>. The unique feature of these carbonates  
758 seems to be the structural heterogeneities observed in  
759 large and medium carbonate nodules. These structural  
760 features haven't been reported in other experiments or  
761 from field data and appear to be unique to this project.  
762 Just as the changing composition could track changes in  
763 the pore fluid composition, the spherulitic texture also  
764 may suggest some sort of change in the environment as  
765 well<sup>45, 46</sup>.

#### 766 *Additional Considerations and Future Research.*

767 Additional research is also needed to understand the  
768 structural heterogeneities of many of the carbonate  
769 nodules. Although we know the chemical composition  
770 of the different regions, understanding the exact mineral  
771 identities of the crystallites that make up the outer Fe-rich  
772 region, may be important in understanding how  
773 conditions change as the reactions proceed.

774 It was noted earlier that algal/bacterial filaments  
775 were found in Zone 3. Although we collected no data on  
776 this subject, it would be useful to understand possible  
777 interplay between biological and inorganic components  
778 in the basalt. It is known that mafic reservoirs can host  
779 diverse microbial biospheres<sup>47</sup> and it has been shown  
780 experimentally that organic acids can affect the type,  
781 extent and rates of carbon mineralization outcomes  
782 through complexation<sup>48</sup>. Microbes can have complex  
783 interactions with CO<sub>2</sub> injections that may be  
784 advantageous or deleterious depending on the  
785 ecosystem<sup>49</sup>, but are important to understand in order to  
786 maximize mineralization efficiency and for potential  
787 selection of injection sites. Our group is actively  
788 investigating biomineralization in basaltic carbon storage  
789 settings.

## 790 CONCLUSION

791 Post-injection analysis of thin sections and hand  
792 samples from the Wallula Basalt Pilot Demonstration has  
793 revealed extensive carbonate mineralization throughout  
794 the target zones. Although carbonation products were  
795 identified in each zone, there were many differences  
796 between them regarding morphology, size, and  
797 abundance of products. It is quite possible that the  
798 chemical and physical variations of the zones are  
799 responsible for the different carbonation products.  
800 Structural and chemical heterogeneities of the carbonate  
801 nodules may also hold information on how the  
802 carbonation reaction effects the host rock and pore fluid  
803 and vice versa. The goal of this study was to examine all  
804 available samples, catalog all carbonation products in the  
805 context of the host rock, and increase our understanding  
806 of the fate of supercritical CO<sub>2</sub> in the subsurface. With  
807 increasing interest in using basalt for carbon  
808 sequestration<sup>50 51</sup>, it is important to collect as much data  
809 as possible from existing demonstrations. This study will  
810 aid researchers in parameterizing models for future and  
811 possibly commercial carbon sequestration projects.

## 812 ASSOCIATED CONTENT

### 813 Supporting Information

814 Depths at which samples were collected (Table S1);  
815 Additional XPL and PPL thin section photographs  
816 (Figures S1, S2, S4, S7, S8, S11, S13, S14, S15, S16,  
817 and S17); Additional hand sample photographs  
818 (Figures S3, S5, S6, S9, and S12); SEM images and  
819 EDS data (Tables S2, S3, S4, S5, S6, S7 and S8 and  
820 Figure S10)

### 821 Abbreviations

- 822 • Arg: Aragonite
- 823 • CC: Carbonate cement
- 824 • Chal: Chalcedony
- 825 • Chl: Chlorophaeite
- 826 • Cpt: Clinoptilolite
- 827 • CRBG: Columbia River Basalt Group
- 828 • FP: Fibro-palagonite
- 829 • LN: Large carbonate nodule
- 830 • MN: Medium carbonate nodule
- 831 • PPL: Plane-polarized light
- 832 • SN: Small carbonate nodule
- 833 • Sph: Silica spherulite
- 834 • Unk: Unknown
- 835 • XPL: Cross-Polarized Light

### 836 Supplemental Information

## 837 AUTHOR INFORMATION

### 838 Corresponding Authors

839 \*Corresponding Authors  
840 Charles T. Depp, [ctdepp@mines.edu](mailto:ctdepp@mines.edu)  
841 H. Todd Schaefer, [todd.schaef@pnnl.gov](mailto:todd.schaef@pnnl.gov)

### 842 Notes

843 The authors declare no competing financial interests.

#### 844 ORCID

845 C.T. Depp: 0000-0002-7763-4066  
846 Q.R.S. Miller: 0000-0003-3009-9702  
847 J. V. Crum: 0000-0001-9197-0878  
848 J.A. Horner: 0000-0003-4418-0675  
849 H.T. Schaefer: 0000-0002-4546-3979

#### 851 ACKNOWLEDGEMENTS

852 This work was performed at the Pacific Northwest  
853 National Laboratory (PNNL) and was supported by the  
854 U.S. Department of Energy's Carbon Storage Program.  
855 We acknowledge support from Darin Damiani (DOE  
856 HQ) and the Carbon Utilization and Storage Partnership  
857 (CUSP). CTD was partially supported by the U.S.  
858 Department of Energy, Office of Science, Office of  
859 Workforce Development for Teachers and Students  
860 (WDTS), under the Science Undergraduate Laboratory  
861 Internships Program (SULI) at PNNL. QRSM was  
862 partially supported by the U.S. Department of Energy  
863 (DOE), Office of Science, Office of Basic Energy  
864 Sciences (BES), Chemical Sciences, Geosciences, and  
865 Biosciences Division through its Geosciences program at  
866 the Pacific Northwest National Laboratory (PNNL). We  
867 would like to thank the two anonymous reviewers whose  
868 insightful comments helped improve our article. We  
869 appreciate their time and attention. We also thank Dr.  
870 Ross Cao (PNNL) for a helpful review of an earlier draft.

#### 871 REFERENCES

872 1. McGrail, B. P.; Spane, F. A.; Sullivan, E. C.; Bacon, D.  
873 H.; Hund, G. In *The Wallula Basalt Sequestration Pilot*  
874 *Project*, 10th International Conference on Greenhouse Gas  
875 Control Technologies, Amsterdam, Netherlands, Sep 19-  
876 23, 2010; Amsterdam, Netherlands, 2010; pp 5653-5660.  
877 2. McGrail, B. P.; Sullivan, C. E.; Spane, F. A.; Bacon, D.  
878 H.; Hund, G.; Thorne, P. D.; Thompson, C. J.; Reidel, S.  
879 P.; Colwell, F. S., Preliminary Hydrogeologic  
880 Characterization Results from the Wallula Basalt Pilot  
881 Study. In Pacific Northwest National Laboratory: Richland,  
882 WA PNWD-4129, 2009.  
883 3. McGrail, B. P.; Sullivan, C. E.; Spane, F. A.; Bacon, D.  
884 H.; Hund, G.; Thorne, P. D.; Thompson, C. J.; Reidel, S.  
885 P.; Colwell, F. S., Preliminary Hydrogeologic  
886 Characterization Results from the Wallula Basalt Pilot  
887 Study, PNWD-4129. In PNNL: Richland, Washington,  
888 2009.  
889 4. McGrail, B. P.; Spane, F. A.; Sullivan, E. C.; Bacon, D.  
890 H.; Hund, G., The Wallula Basalt Sequestration Pilot  
891 Project. *Energy Procedia* **2011**, *4*, 5653-5660.  
892 5. White, S. K.; Spane, F. A.; Schaefer, H. T.; Miller, Q. R.  
893 S.; White, M. D.; Horner, J. A.; McGrail, B. P.,  
894 Quantification of CO<sub>2</sub> Mineralization at the Wallula Basalt  
895 Pilot Project. *Environmental Science & Technology* **2020**,  
896 *54*, (22), 14609-14616.  
897 6. McGrail, B. P.; Schaefer, H. T.; Spane, F. A.; Cliff, J. B.;  
898 Qafoku, O.; Horner, J. A.; Thompson, C. J.; Owen, A. T.;

899 Sullivan, C. E., Field Validation of Supercritical CO<sub>2</sub>  
900 Reactivity with Basalts. *Environ. Sci. & Technol. Lett.*  
901 **2017**, *4*, (1), 6-10.  
902 7. Reidel, S. P.; Tolan, T. L., The Grande Ronde Basalt,  
903 Columbia River Basalt Group. *Geological Society of*  
904 *America Special Papers* **2013**, *497*, 117-153.  
905 8. Reidel, S. P.; Spane, F. A.; Johnson, V. G. *Natural gas*  
906 *storage in basalt aquifers of the Columbia Basin, Pacific*  
907 *Northwest USA: A guide to site characterization*; Pacific  
908 Northwest National Lab.(PNNL), Richland, WA (United  
909 States): 2002.  
910 9. McGrail, B. P.; Spane, F. A.; Amonette, J. E.;  
911 Thompson, C. R.; Brown, C. F., Injection and Monitoring  
912 at the Wallula Basalt Pilot Project. *Energy Procedia* **2014**,  
913 *63*, 2939-2948.  
914 10. McGrail, B. P.; Schaefer, H. T.; Ho, A. M.; Chien, Y. J.;  
915 Dooley, J. J.; Davidson, C. L., Potential for carbon dioxide  
916 sequestration in flood basalts. *Journal of Geophysical*  
917 *Research-Solid Earth* **2006**, *111*, (B12).  
918 11. Matter, J. M.; Stute, M.; Snaebjornsdottir, S. O.;  
919 Oelkers, E. H.; Gislason, S. R.; Aradottir, E. S.; Sigfusson,  
920 B.; Gunnarsson, I.; Sigurdardottir, H.; Gunnlaugsson, E.;  
921 Axelsson, G.; Alfredsson, H. A.; Wolff-Boenisch, D.;  
922 Mesfin, K.; Taya, D. F. D.; Hall, J.; Dideriksen, K.;  
923 Broecker, W. S., Rapid carbon mineralization for  
924 permanent disposal of anthropogenic carbon dioxide  
925 emissions. *Science* **2016**, *352*, (6291), 1312-1314.  
926 12. Qomi, M. J. A.; Miller, Q. R. S.; Zare, S.; Schaefer, H.  
927 T.; Kaszuba, J. P.; Rosso, K. M., Molecular-Scale  
928 Mechanisms of CO<sub>2</sub> Mineralization in Nanoscale  
929 Interfacial Water Films. *Nature Reviews Chemistry* **2022**, *6*,  
930 598-613.  
931 13. Snaebjornsdottir, S. Ó.; Sigfusson, B.; Marieni, C.;  
932 Goldberg, D.; Gislason, S. R.; Oelkers, E. H., Carbon  
933 dioxide storage through mineral carbonation. *Nature*  
934 *Reviews Earth & Environment* **2020**, *1*, (2), 90-102.  
935 14. Tutolo, B. M.; Awolayo, A.; Brown, C., Alkalinity  
936 Generation Constraints on Basalt Carbonation for Carbon  
937 Dioxide Removal at the Gigaton-per-Year Scale. *Environ.*  
938 *Sci. Technol.* **2021**, *55*, (17), 11906-11915.  
939 15. Xiong, W.; Wells, R. K.; Menefee, A. H.; Skemer, P.;  
940 Ellis, B. R.; Giammar, D. E., CO<sub>2</sub> mineral trapping in  
941 fractured basalt. *International Journal of Greenhouse Gas*  
942 *Control* **2017**, *66*, 204-217.  
943 16. Xiong, W.; Wells, R. K.; Horner, J. A.; Schaefer, H. T.;  
944 Skemer, P. A.; Giammar, D. E., CO<sub>2</sub> Mineral Sequestration  
945 in Naturally Porous Basalt. *Environ. Sci. & Technol. Lett.*  
946 **2018**, *5*, (3), 142-147.  
947 17. Wells, R. K.; Xiong, W.; Giammar, D.; Skemer, P.,  
948 Dissolution and surface roughening of Columbia River  
949 flood basalt at geologic carbon sequestration conditions.  
950 *Chem. Geol.* **2017**, *467*, 100-109.  
951 18. Menefee, A. H.; Giammar, D. E.; Ellis, B. R.,  
952 Permanent CO<sub>2</sub> Trapping through Localized and Chemical  
953 Gradient-Driven Basalt Carbonation. *Environ. Sci. Technol.*  
954 **2018**, *52*, (15), 8954-8964.  
955 19. Dickson, J. A. D., Carbonate identification and genesis  
956 as revealed by staining. *Journal of Sedimentary Research*  
957 **1966**, *36*, (2), 491-505.  
958 20. Dickson, J. A. D., A Modified Staining Technique for  
959 Carbonates in Thin Section. *Nature* **1965**, *205*, (4971), 587-  
960 587.

- 961 21. Peacock, M. A.; Fuller, R. E., Chlorophaeite,  
962 sideromelane and palagonite from the Columbia River  
963 Plateau. *American Mineralogist* **1928**, *13*, (7), 360-383.
- 964 22. Stroncik, N. A.; Schmincke, H.-U., Palagonite – a  
965 review. *International Journal of Earth Sciences* **2002**, *91*,  
966 (4), 680-697.
- 967 23. Stroncik, N.; Schmincke, H., Evolution of palagonite:  
968 Crystallization, chemical changes, and element budget.  
969 *Geochemistry Geophysics Geosystems* **2001**, *2*.
- 970 24. Antoshkina, A. I.; Ryabinkina, N. N.; Valyaeva, O. V.,  
971 Genesis of siderite nodules from the lower carboniferous  
972 terrigenous sequence in the Subpolar Urals. *Lithol. Mineral*  
973 *Resour.* **2017**, *52*, (2), 111-124.
- 974 25. Antoshkina, A. I.; Ryabinkina, N. N. In *Lower*  
975 *Carboniferous Siderites: A Product of Bottom Seeps and*  
976 *Bacterial Metanogenesis (Subpolar Urals)*, 2018, Springer:  
977 pp 139-142.
- 978 26. Schaefer, H. T.; McGrail, B. P.; Owen, A. T., Basalt  
979 Reactivity Variability with Reservoir Depth in Supercritical  
980 CO<sub>2</sub> and Aqueous Phases. *Energy Procedia* **2011**, *4*, 4977-  
981 4984.
- 982 27. Sun, C.-Y.; Gránásy, L.; Stiffler, C. A.; Zaquin, T.;  
983 Chopdekar, R. V.; Tamura, N.; Weaver, J. C.; Zhang, J. A.  
984 Y.; Goffredo, S.; Falini, G.; Marcus, M. A.; Pusztai, T.;  
985 Schoeppler, V.; Mass, T.; Gilbert, P. U. P. A., Crystal  
986 nucleation and growth of spherulites demonstrated by coral  
987 skeletons and phase-field simulations. *Acta Biomaterialia*  
988 **2021**, *120*, 277-292.
- 989 28. Beck, R.; Andreassen, J.-P., Spherulitic Growth of  
990 Calcium Carbonate. *Crystal Growth & Design* **2010**, *10*,  
991 (7), 2934-2947.
- 992 29. De Yoreo, J. J.; Gilbert, P. U. P. A.; Sommerdijk, N.  
993 A. J. M.; Penn, R. L.; Whitlam, S.; Joester, D.; Zhang, H.;  
994 Rimer, J. D.; Navrotsky, A.; Banfield, J. F.; Wallace, A. F.;  
995 Michel, F. M.; Meldrum, F. C.; Cölfen, H.; Dove, P. M.,  
996 Crystallization by particle attachment in synthetic,  
997 biogenic, and geologic environments. *Science* **2015**, *349*,  
998 (6247).
- 999 30. Spencer, E., On some occurrences of spherulitic  
1000 siderite and other carbonates in sediments. *Quarterly*  
1001 *Journal of the Geological Society* **1925**, *81*, (1-4), 667-705.
- 1002 31. Wieder, M.; Yaalon, D. H., Effect of matrix  
1003 composition on carbonate nodule crystallization. *Geoderma*  
1004 **1974**, *11*, (2), 95-121.
- 1005 32. Kralj, D.; Vdović, N., The influence of some naturally  
1006 occurring minerals on the precipitation of calcium  
1007 carbonate polymorphs. *Water Research* **2000**, *34*, (1), 179-  
1008 184.
- 1009 33. Fodor, M. A.; Ható, Z.; Kristóf, T.; Pósfai, M., The  
1010 role of clay surfaces in the heterogeneous nucleation of  
1011 calcite: Molecular dynamics simulations of cluster  
1012 formation and attachment. *Chemical Geology* **2020**, *538*,  
1013 119497.
- 1014 34. Molnár, Z.; Pekker, P.; Dódony, I.; Pósfai, M., Clay  
1015 minerals affect calcium (magnesium) carbonate  
1016 precipitation and aging. *Earth and Planetary Science*  
1017 *Letters* **2021**, *567*, 116971.
- 1018 35. Schaefer, H. T.; McGrail, B. P.; Owen, A. T., Carbonate  
1019 mineralization of volcanic province basalts. *International*  
1020 *Journal of Greenhouse Gas Control* **2010**, *4*, (2), 249-261.
- 1021 36. Raza, A.; Rezaee, R.; Gholami, R.; Bing, C. H.;  
1022 Nagarajan, R.; Hamid, M. A., A screening criterion for  
1023 selection of suitable CO<sub>2</sub> storage sites. *Journal of Natural*  
1024 *Gas Science and Engineering* **2016**, *28*, 317-327.
- 1025 37. Fazeli, H.; Masoudi, M.; Patel, R. A.; Aagaard, P.;  
1026 Hellevang, H., Pore-Scale Modeling of Nucleation and  
1027 Growth in Porous Media. *ACS Earth and Space Chemistry*  
1028 **2020**, *4*, (2), 249-260.
- 1029 38. Liu, D.; Agarwal, R.; Li, Y.; Yang, S., Reactive  
1030 transport modeling of mineral carbonation in unaltered and  
1031 altered basalts during CO<sub>2</sub> sequestration. *International*  
1032 *Journal of Greenhouse Gas Control* **2019**, *85*, 109-120.
- 1033 39. Olajire, A. A., A review of mineral carbonation  
1034 technology in sequestration of CO<sub>2</sub>. *Journal of Petroleum*  
1035 *Science and Engineering* **2013**, *109*, 364-392.
- 1036 40. Raza, A.; Glatz, G.; Gholami, R.; Mahmoud, M.;  
1037 Alafnan, S., Carbon mineralization and geological storage  
1038 of CO<sub>2</sub> in basalt: Mechanisms and technical challenges.  
1039 *Earth-Science Reviews* **2022**, *229*, 104036.
- 1040 41. Polites, E. G.; Schaefer, H. T.; Horner, J. A.; Owen, A.  
1041 T.; Holliman, J. E.; McGrail, B. P.; Miller, Q. R. S., Exotic  
1042 Carbonate Mineralization Recovered from a Deep Basalt  
1043 Carbon Storage Demonstration. *Environmental Science &*  
1044 *Technology* **2022**.
- 1045 42. Kanakiya, S.; Adam, L.; Esteban, L.; Rowe, M. C.;  
1046 Shane, P., Dissolution and secondary mineral precipitation  
1047 in basalts due to reactions with carbonic acid. *J. Geophys.*  
1048 *Res. Solid Earth* **2017**, *122*, (6), 4312-4327.
- 1049 43. Gysi, A. P.; Stefánsson, A., Mineralogical aspects of  
1050 CO<sub>2</sub> sequestration during hydrothermal basalt alteration —  
1051 An experimental study at 75 to 250°C and elevated pCO<sub>2</sub>.  
1052 *Chemical Geology* **2012**, *306-307*, 146-159.
- 1053 44. Clark, D. E.; Oelkers, E. H.; Gunnarsson, I.; Sigfússon,  
1054 B.; Snæbjörnsdóttir, S. Ó.; Aradóttir, E. S.; Gíslason, S. R.,  
1055 CarbFix2: CO<sub>2</sub> and H<sub>2</sub>S mineralization during 3.5 years of  
1056 continuous injection into basaltic rocks at more than  
1057 250 °C. *Geochimica et Cosmochimica Acta* **2020**, *279*, 45-  
1058 66.
- 1059 45. Shtukenberg, A. G.; Punin, Y. O.; Gunn, E.; Kahr, B.,  
1060 Spherulites. *Chemical Reviews* **2012**, *112*, (3), 1805-1838.
- 1061 46. Beck, R.; Andreassen, J.-P., The onset of spherulitic  
1062 growth in crystallization of calcium carbonate. *Journal of*  
1063 *Crystal Growth* **2010**, *312*, (15), 2226-2238.
- 1064 47. Lavalleur, H. J.; Colwell, F. S., Microbial  
1065 characterization of basalt formation waters targeted for  
1066 geological carbon sequestration. *FEMS microbiology*  
1067 *ecology* **2013**, *85*, (1), 62-73.
- 1068 48. Miller, Q. R.; Kaszuba, J. P.; Schaefer, H. T.; Bowden,  
1069 M. E.; McGrail, B. P., Impacts of organic ligands on  
1070 forsterite reactivity in supercritical CO<sub>2</sub> fluids.  
1071 *Environmental Science & Technology* **2015**, *49*, (7), 4724-  
1072 4734.
- 1073 49. Trias, R.; Ménez, B.; le Campion, P.; Zivanovic, Y.;  
1074 Lecourt, L.; Lecoeuvre, A.; Schmitt-Kopplin, P.; Uhl, J.;  
1075 Gíslason, S. R.; Alfreðsson, H. A.; Mesfin, K. G.;  
1076 Snæbjörnsdóttir, S. Ó.; Aradóttir, E. S.; Gunnarsson, I.;  
1077 Matter, J. M.; Stute, M.; Oelkers, E. H.; Gérard, E., High  
1078 reactivity of deep biota under anthropogenic CO<sub>2</sub> injection  
1079 into basalt. *Nature Communications* **2017**, *8*, (1), 1063.
- 1080 50. Arvanitis, A.; Koutsovitis, P.; Koukouzas, N.;  
1081 Tyrologou, P.; Karapanos, D.; Karkalis, C.; Pomonis, P.,  
1082 Potential Sites for Underground Energy and CO<sub>2</sub> Storage  
1083 in Greece: A Geological and Petrological Approach.  
1084 *Energies* **2020**, *13*, (11).

1085 51. Shrivastava, J.; Rani, N.; Pathak, V., Geochemical  
1086 modeling and experimental studies on mineral carbonation  
1087 of primary silicates for long-term immobilization of CO<sub>2</sub> in  
1088 basalt from the eastern Deccan volcanic province. *Jour.*  
1089 *Indian Geophys. Union, Sp* **2016**, *1*, 42-58.

1090