



Dimethyl methylphosphonate Detection Using Zirconium Metal-Organic Framework Functionalized Quartz Crystal Microbalance

Anthony J. Chavez¹, Christian A. Pattyn¹, Erika C. Vreeland¹, Leah N. Appelhans^{1*}, Jason P. Sammon¹, Matthew W. Moorman¹, James Griego¹, Karl Westlake¹, Jayson L. Briscoe¹, Dale L. Huber², Jeremy B. Wright^{1*}

¹Sandia National Laboratories, Albuquerque, NM 87123; ²Center for Integrated Nanotechnologies, Sandia National Laboratories, Albuquerque, NM 87185

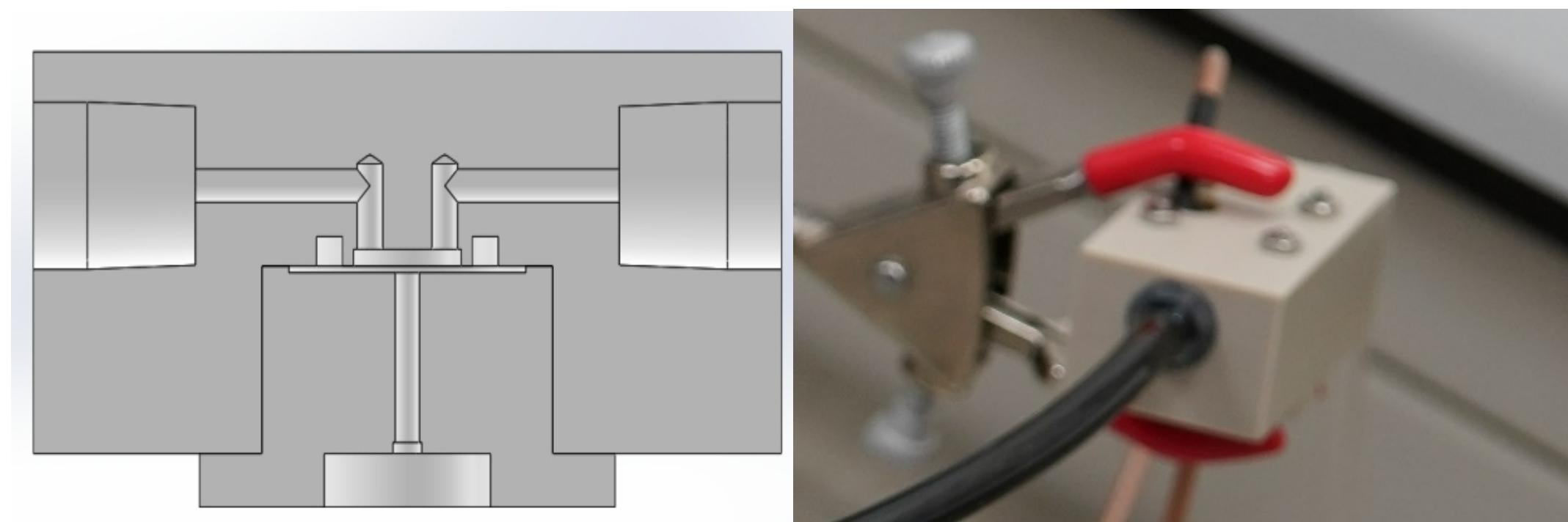


Introduction

- We face numerous challenges in detecting chemical threats both for civilians and our warfighters.
- A simple and rapid microwave synthesis method was developed to grow a thin film of metal organic frameworks (MOFs).
- UiO-66 or UiO-66-NH₂ were grown on unfunctionalized gold and silicon substrates.
- We have expanded this to the functionalization of quartz crystal microbalances (QCMs).
- We exposed our functionalized MOF-QCM to humid N₂ and dimethyl methylphosphonate (DMMP).
- DMMP is both a simulant and a precursor in CWA production and is used in other chemical processes.

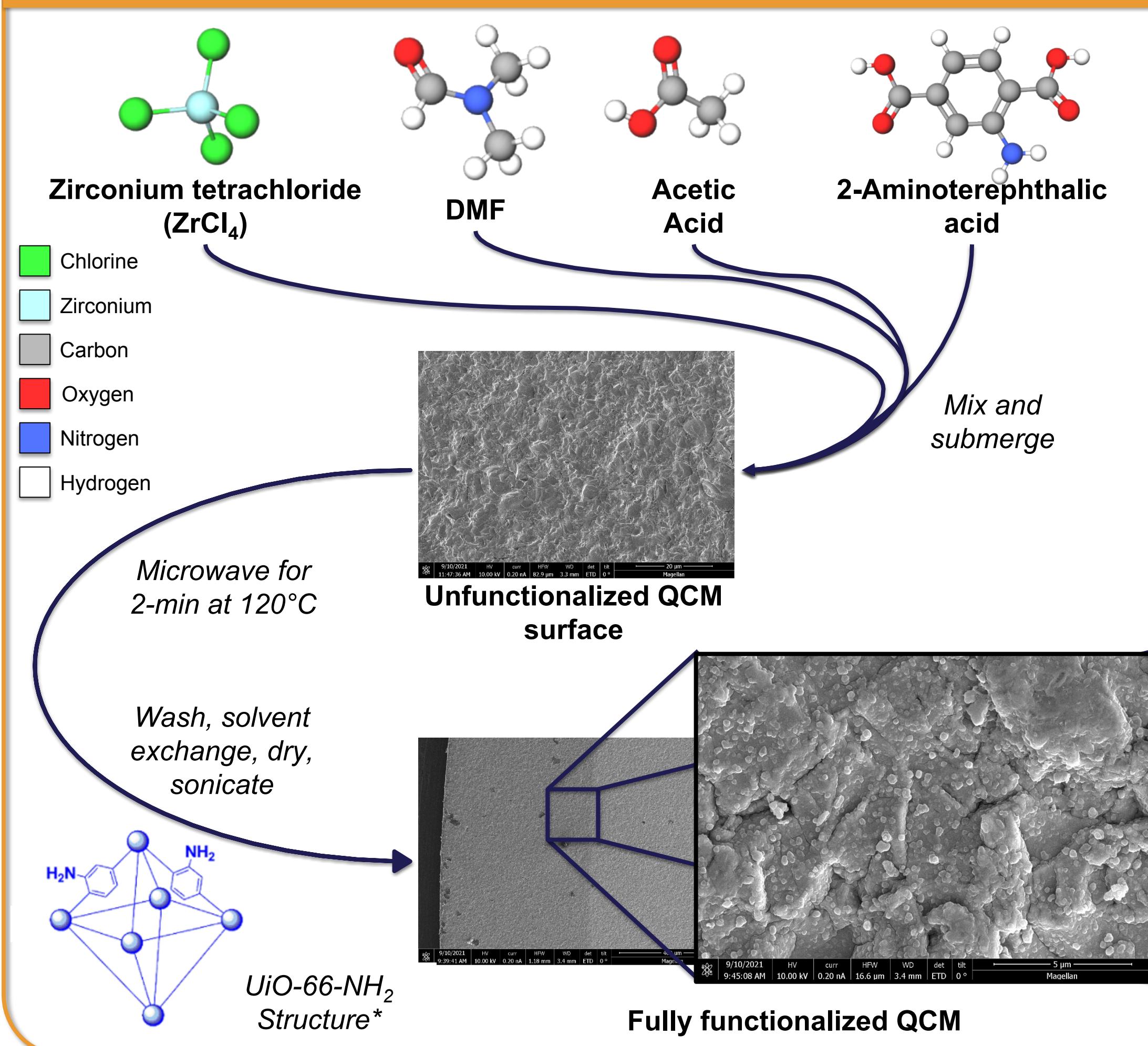


Quartz Crystal Microbalance



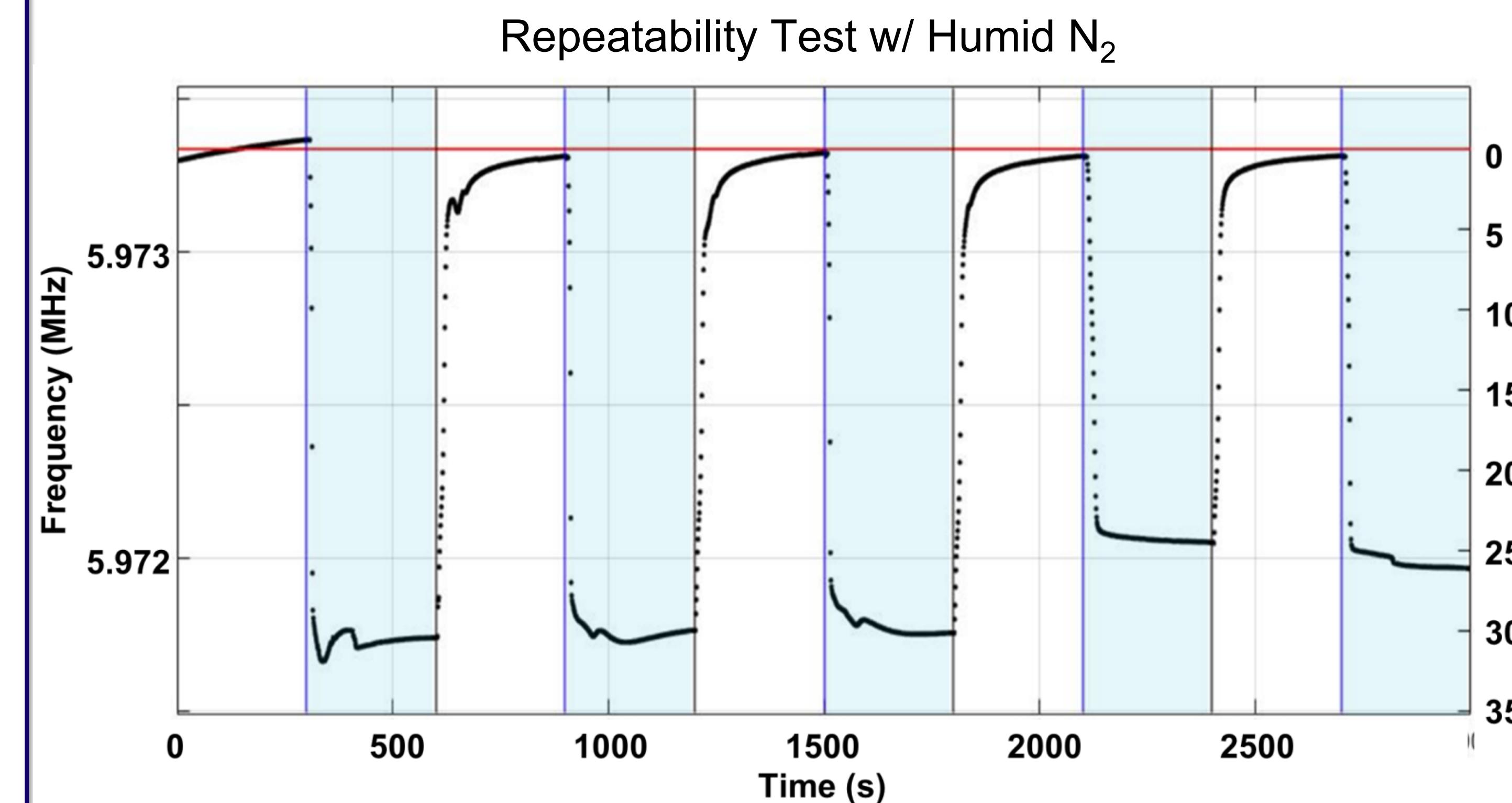
In house designed and manufactured flow cell with detachable QCM holder.

Synthesis



Zhu, J., et al. (2019). "Polyethyleneimine-Modified Uio-66-NH₂(Zr) Metal-Organic Frameworks: Preparation and Enhanced CO₂ Selective Adsorption." *ACS Omega* 4(2): 3188-3197.

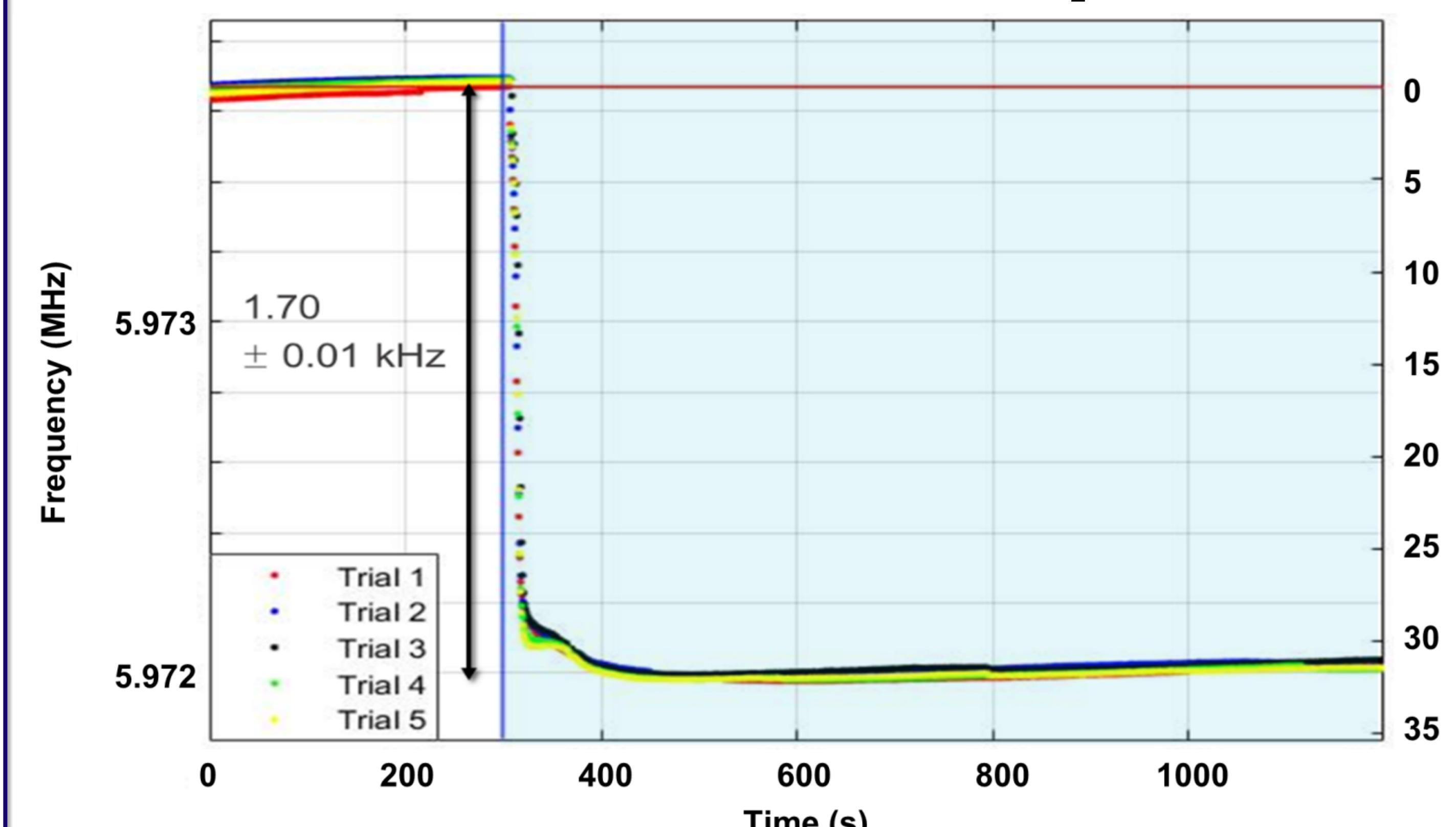
Results: Initial Humidity Testing



QCM output data following repeatability testing with water exposure.

- Bubbler used to add humidity to N₂.
- MOF-QCM used in all trials.
- Baseline found by running pure N₂ for 5 minutes.
- Red line indicates baseline frequency.**
- Humid N₂ for 5 minutes, pure N₂ for 5 minutes, 5 cycles.
- Blue highlights indicated humid air passing through the flow cell.**
- Nonhighlighted areas indicate pure N₂ passing through the flow cell.
- In the repeatability test we see that prolonged exposure could be causing degradation which is likely the reason we have a lower adsorption on the 4th and 5th cycle.

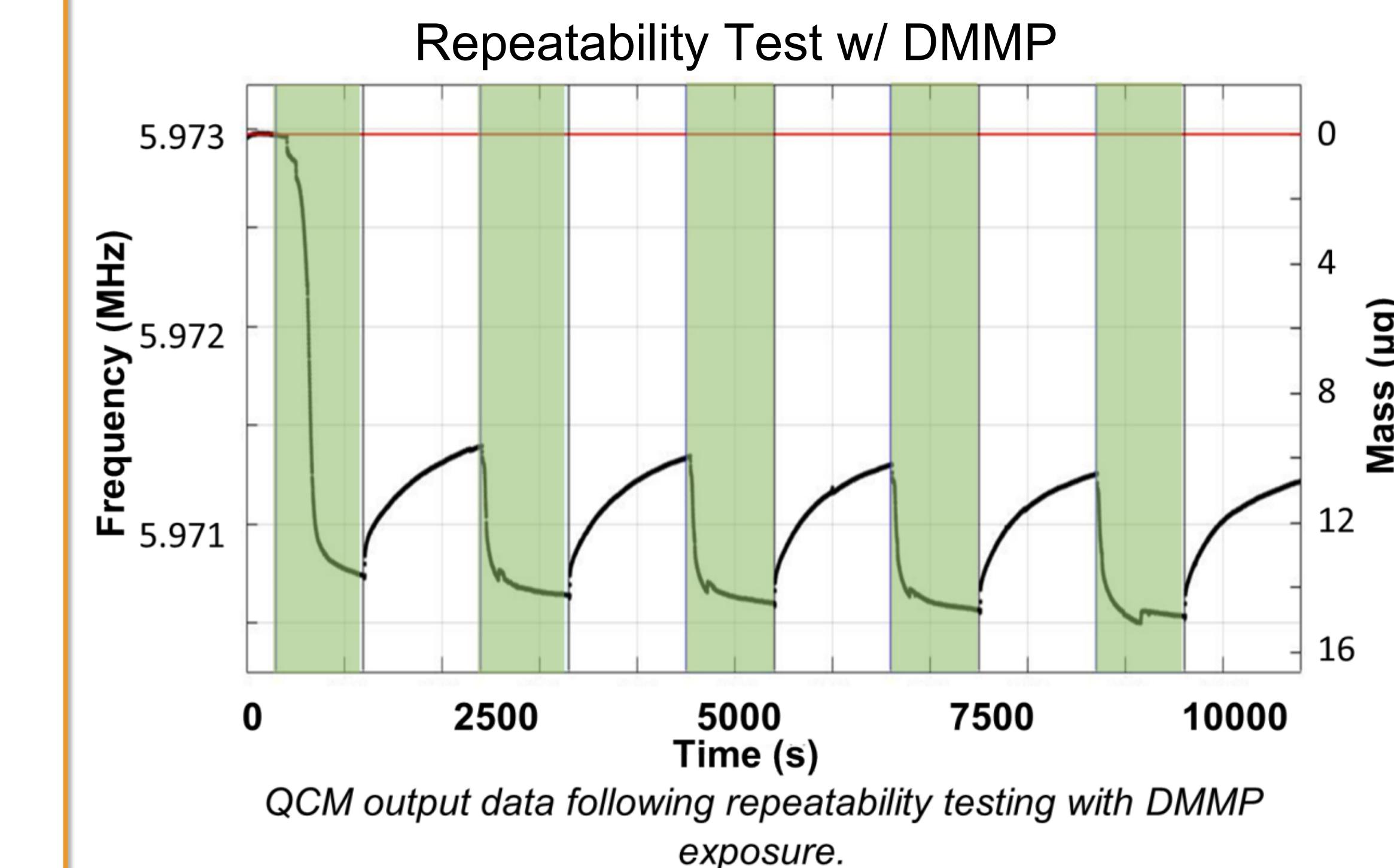
Reproducibility Test w/ Humid N₂



QCM output data following reproducibility testing with water exposure.

- Baseline with pure N₂ for 5 minutes.
- Humid N₂ for 15 minutes.
- With sufficient time (over 30 minutes of drying) between trials we return to our initial baseline.
- We were able to get similar adsorption levels when starting at the initial baseline.

Results: DMMP Testing



QCM output data following repeatability testing with DMMP exposure.

- DMMP trial run using a Kin-Tek.
- Baseline found by running pure N₂ for 5 minutes.
- DMMP for 15 minutes, pure N₂ for 20 minutes, 5 cycles.
- Green highlights indicated DMMP passing through the flow cell.**
- Remarkably, the DMMP exposure shows a difference in desorption rate from the humid N₂ trials.
- This indicates the selectivity our MOF-QCM has to DMMP.
- If we are not able to find a simple and fast method of removing the DMMP we will likely have a precursor to design for a dosimeter instead of a sensor.

Conclusions & Future Work

We have shown successful detection of DMMP using a Zr-MOF functionalized QCM. Further development will investigate functionalizing a more robust sensor using nanohole arrays (NHA). By functionalizing an NHA with the Zr-MOF we hope to detect and quantify a shift in the extraordinary optical transmission (EOT) spectrum when in the presence of DMMP.

