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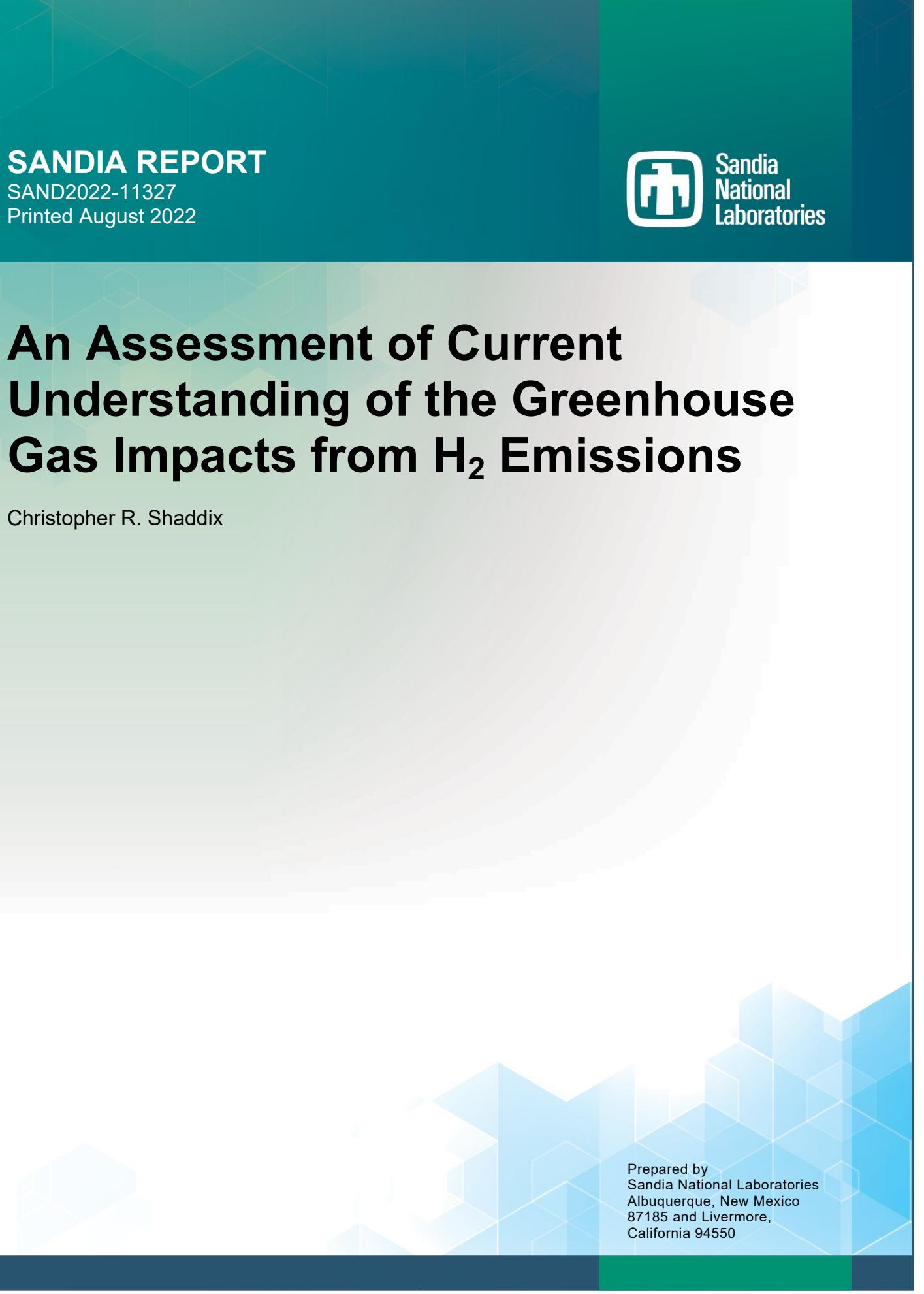
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An Assessment of Current Understanding of the Greenhouse Gas Impacts from H₂ Emissions

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A large, abstract graphic in the background consists of numerous light blue and white 3D hexagonal prisms of varying sizes, arranged in a way that suggests depth and perspective, resembling a city skyline or a complex data visualization.

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ABSTRACT

With the anticipated growth in hydrogen generation and use as part of a broad shift in energy use away from fossil fuels, concerns have been raised regarding the impact of increased H₂ emissions on global warming. Atmospheric scientists have long recognized that H₂ emissions into the atmosphere do have an indirect impact on global warming, largely because a portion of emitted H₂ is consumed by the hydroxyl radical (OH), which is the primary reactant that removes the potent greenhouse gas methane from the atmosphere. Therefore, increases in H₂ emissions will result in decreases in the average OH concentrations in the atmosphere and an increase in the atmospheric lifetime of methane. Various assessments of the impact of H₂ emissions on global warming have been performed over the past couple of decades. These assessments have yielded significant variability and recognized uncertainty in the magnitude of the warming effect of a given quantity of emitted H₂, and an even greater uncertainty in the magnitude of H₂ leakage and releases that can be anticipated with an expanded H₂ infrastructure. Consequently, definitive estimates of the magnitude of the warming effect of additional emitted H₂ are lacking. However, given the current understanding of the warming potential of emitted H₂ and given reasonable expectations of the emission rate of an expanded H₂ infrastructure, it is anticipated that warming effects from emitted H₂ will offset no more than 5% of the reduction in warming associated with avoided CO₂ emissions from using clean H₂. Further, it is highly unlikely that the warming effects from emitted H₂ will offset more than 10% of the benefit from avoided CO₂ emissions, at least as considered over a typical 100-year accounting period. Because of the short atmospheric lifetimes of H₂ and methane, however, the warming effect of emitted H₂ is enhanced over the first few years following increases in H₂ emission.

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EXECUTIVE SUMMARY

Over the past two decades there have been several assessments of the impact of H₂ emissions on atmospheric chemistry and overall global warming. There is consensus that H₂ emissions do have an overall warming effect, primarily because they enhance the lifetime of methane in the atmosphere through reactive competition for the hydroxyl radical. The magnitude of the warming effect is directly dependent on the quantity of H₂ emissions and on estimates of the global warming potential of H₂, which vary by about a factor of three. When comparing the warming effect from H₂ emissions to the warming avoided by using clean H₂ instead of fossil fuels, both straightforward computations and complex system analysis arrive at the conclusion that the warming effect from H₂ emissions is unlikely to be more than 5% of the warming effect if the equivalent energy use were provided by fossil fuels. However, in the near-term, the warming effect of H₂ emissions will be somewhat larger because of the short atmospheric lifetimes of H₂ and methane.

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1. BACKGROUND

Several recent analyses have suggested that global H₂ demand could increase by about an order of magnitude and exceed 500 Mt by 2050, to help meet net-zero CO₂ emission goals in energy sectors that are difficult to electrify.¹⁻³ Coincident with these predictions of expanded H₂ use, concerns have been raised regarding the impacts of hydrogen released from leaks or releases on global warming and whether such effects could offset the positive impacts of CO₂ emission reduction from switching from use of fossil fuel energy sources to clean (i.e. green or blue) H₂ sources. In fact, assessments of the climate consequences of H₂ emissions have been undertaken with various levels of scientific fidelity for many years, beginning in the early 2000's,⁴⁻⁸ when serious consideration was first given to the use of H₂ energy systems as an alternative to fossil fuels, particular for transportation applications. In contrast to direct greenhouse gases that participate in radiative exchange in the earth's atmosphere, hydrogen is considered an 'indirect' greenhouse gas, because while it does not directly participate in radiative exchange, its presence in the atmosphere is known to influence the concentrations of direct greenhouse gases through photochemical reactions, such that there is a net warming effect.

Current assessments of the inventory of H₂ in the earth's atmosphere place its concentration at approximately 530 ppb, on average, with a nominally balanced flux into and out of the atmosphere of 80 Tg/yr (i.e. 80 billion kg per year).^{6,7,9} The largest contributing source of H₂ in the atmosphere is through the atmospheric oxidation of methane and other volatile organic compounds (VOCs). Biomass combustion (e.g. wildfires) and fossil fuel combustion systems (primarily spark ignition engines during cold-start or without a catalytic converter) are assessed to be the next largest sources of atmospheric H₂. Bacterial nitrogen-fixing reactions also contribute some H₂ to the atmosphere. H₂ removal from the atmosphere is primarily accomplished through two pathways. The largest sink of H₂ is by bacterial respiration (i.e. biological oxidation) in the soil. The second largest sink is through oxidation of H₂ by reaction with the hydroxyl radical (OH) in the atmosphere. This latter route of H₂ consumption results in global warming impacts, primarily through three different effects, as suggested by the chemical mechanism in Fig. 1.

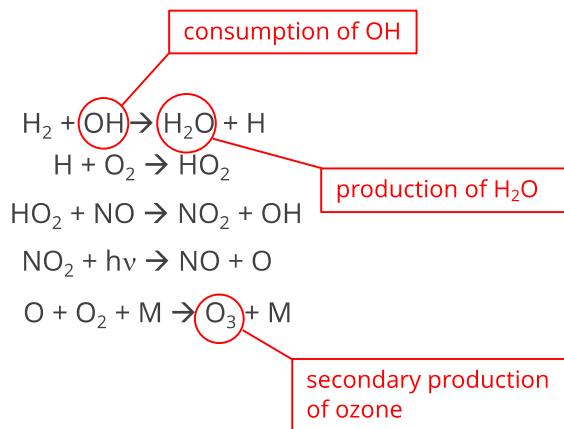


Figure 1. Primary consumption chemistry of H₂ in the earth's atmosphere, highlighting the most important reactants and products that influence the earth's energy balance.

The largest impact of H₂ oxidation on global warming is through its consumption of OH, which is the primary oxidant of methane in the atmosphere. As methane is a potent greenhouse gas (in total, methane currently contributes approximately 2/3rds as much as CO₂ to earth's warming),¹⁰ any

factor that tends to reduce OH concentrations in the atmosphere, leading to a longer methane lifetime, thereby contributes to global warming. The other two warming effects of H₂ consumption that have been identified are smaller: additional production of tropospheric ozone, which is itself a greenhouse gas, and additional generation and transport of water vapor (generated from H₂ oxidation) into the stratosphere. Water vapor in the stratosphere tends to warm the earth due to radiative trapping (i.e. absorption and re-emission) of long wavelength (IR) radiation emitted from the earth's surface. Currently, H₂ in the atmosphere has a lifetime of approximately 1.5-3 years.^{5,9,11-}
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2. ASSESSMENTS OF H₂ EMISSION IMPACTS

There have been several studies that have attempted to quantify the expected impact of hydrogen infrastructure leakage and releases on global warming.^{4-9,13-17} Of course, such impacts need to be weighed against any reduction in global warming that will result from reduced CO₂ emissions associated with use of a hydrogen energy source that avoids the use of fossil fuels. There are substantial uncertainties in these estimates of H₂ emission effects on global warming associated with (a) the magnitude of H₂ leakage and releases that can be expected from H₂ infrastructure, (b) the extent to which the current rate of bacterial respiration of H₂ will scale with a higher concentration of H₂ in the atmosphere due to the additional emission of H₂, (c) the effect of increased H₂ use on emissions of CH₄, VOCs, and NO_x into the atmosphere, all of which influence the concentration of the potent greenhouse gases CH₄ and O₃, and (d) the extent of increased water vapor transport into the stratosphere due to H₂ oxidation in the upper troposphere (which is difficult to predict). There are also significant variations in the predictions of how H₂ oxidation in the atmosphere affects OH concentrations and ozone production in the troposphere, due to differences in the transport-chemistry models that are employed, as well as differences in the various assumed tropospheric source and sink fluxes. As a result, there is substantial variation in the magnitude of warming effects predicted to result from H₂ leakage and releases.

A common metric used to measure the warming impact of gases introduced into the atmosphere is the Global Warming Potential (GWP), which is calculated as the change in the earth's temperature upon introducing a small pulse of the gas species in question, relative to the warming effect of an equal mass of CO₂ added to the atmosphere. By definition, CO₂ has a GWP of precisely 1.0, whereas the GWP of methane is approximately 30, underscoring its strong absorptivity of infrared radiation emitted by the surface of the earth. The GWP is typically quantified over the course of a 100-year timeframe, because the lifetime of CO₂ in the atmosphere is on the order of 100 years. Derwent et al.¹⁻² first calculated a GWP for H₂ as 6, and subsequently calculated values of 3.3 +/- 1.4¹⁶ and 5 +/- 1.¹⁵ However, these calculations were all based on a tropospheric 'box' model that ignored stratospheric impacts of increased H₂ in the atmosphere and also made rough extrapolations from shorter time period calculations to arrive at GWP₁₀₀ values. Recently, Ocko and Hamburg¹⁸ asserted that the results from the detailed full atmosphere study of Paulot et al.⁹ implied a H₂ GWP of 10, and the full atmosphere study of Warwick et al.¹³ reported a value of 11 +/- 5. While these larger GWP values reported for H₂ seem ominous, recall that GWP is defined on a *mass* basis. Consequently, a H₂ GWP of 11 implies that a H₂ molecule introduced into the atmosphere will have exactly one-half times the warming impact of a CO₂ molecule introduced into the atmosphere: $11 * (\text{MW}_{\text{H}_2}/\text{MW}_{\text{CO}_2}) = 0.5$. Noting that H₂ PEM fuel cells are approximately 60% energy efficient and heavy-duty diesel engines are approximately 40% energy efficient, it can be shown that for every 100 molecules of clean H₂ that are used in a fuel cell engine that replaces a heavy-duty diesel engine, there will be 60 fewer CO₂ molecules released into the atmosphere. (This simple analysis yields a result that is actually very similar to the detailed analysis for system-wide avoided CO₂ emission that was reported in the Hydrogen Council 2017 study,¹ wherein the avoided CO₂ emission/H₂ was estimated to be 0.5, on a molar basis.) Consequently, for each leak or release of 1% of the H₂ intended to be used in a fuel cell engine, the warming effect from that H₂ emission will offset 0.5/60 = 0.8% of the benefit of using H₂ to reduce global warming from avoided CO₂ emissions.

There is substantial uncertainty regarding the extent of H₂ leakage and releases that can be anticipated in an expanded H₂ infrastructure, which naturally influences the magnitude of the expected warming effect from such emissions. Cooper et al.¹⁷ have estimated that overall H₂ emissions are unlikely to be more than 4% of the H₂ that is ultimately used. Air Liquide has analyzed H₂ losses and have determined that the largest losses are associated with purging during compressed H₂ vehicle refueling and with boil-off and related losses when liquefying and transferring liquefied H₂.¹⁹ Total system losses for compressed H₂ used in vehicles is estimated to be around 4%, while pipeline H₂ use is estimated to have about a 1% loss rate, and liquefied H₂ used in vehicles currently is estimated to have a loss rate between 10-20%. This analysis suggests that from a global warming perspective, losses associated with liquefied H₂ need to be significantly reduced. Petitpas²⁰ has shown that with implementation of existing boil-off recovery strategies the loss associated with larger LH₂ dispensing stations could be reduced below 1%.

Based on the existing studies, an assumed 5% emission rate of H₂ in an expanded H₂ economy would appear to be a reasonable upper limit. Using the result of the analysis described previously for the CO₂ emission offset of using H₂ in a fuel cell engine, a 5% emission rate of H₂ corresponds to a 4% warming offset relative to avoided CO₂ emissions. This result is very similar to those that have been reported in the existing literature on the warming effects from H₂ emissions associated with a large global H₂ energy economy.^{5-7,9,13-15} All of these studies have found that the expected warming effect is less than 5% of the warming that is avoided by converting a fossil fuel energy system to a clean H₂ energy system. Given the uncertainties in the true GWP of hydrogen as well as uncertainties in the expected emission rate of H₂, a very conservative upper bound of the warming effect from emitted H₂ would be at the level of 10% of the warming that is avoided by converting a fossil fuel energy system to a clean H₂ energy system.

Ocko and Hamburg,¹⁸ and to a lesser extent Warwick et al.,¹³ have emphasized that the warming effect from H₂ emissions occurs much quicker than that from emitted CO₂, on account of the shorter lifetimes of H₂ and CH₄ in the atmosphere. Essentially one pays for the heating effect from leaked H₂ right away (over the course of several years), while you need to wait many years to fully experience the benefit of reduced CO₂ emissions associated with using H₂. In fact, as shown by Ocko and Hamburg,¹⁸ for very high H₂ leak rates (order of 10%) one can experience net warming over the first few years of switching from a fossil fuel system to a H₂ system. Over time, however, the benefits outweigh the short-term penalty and the avoided CO₂ emissions dominate the overall warming response.

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