

## Ongoing Research and Development: Cement Filler Testing and Analysis

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# Key Attributes for DPC Fillers

- Material Compatibility
- Ease of Injectability
- Moderator Displacement
- Minimal Intrinsic Neutron Moderation
- Minimal Gas Generation
- Long-Term Chemical Stability
- Radionuclide Sequestration



Phosphate-Based Cements



Low Melting Point Metals

# Phosphate Cements as DPC Fillers

## Features of Phosphate Cements:

- Intrinsic neutron moderation is limited.
- Addition of neutron poisons possible ( $B_4C$ , Gd-oxides)
- Slurry properties vary but can be modified.
- Strengths, porosities, etc. vary but can be modified.
- Near Neutral pH
- Very low solubilities (at near neutral pH)
- Self-Bonding
- Radionuclide Sequestration



# Phosphate Cements Under Evaluation

- Aluminum Oxide / Aluminum Phosphate ( $\text{Al}_2\text{O}_3$  /  $\text{AlPO}_4$ ) Cements (APCs)
- Wollastonite / Aluminum Phosphate ( $\text{CaSiO}_3$  /  $\text{AlPO}_4$ ) Cements (WAPCs)
- Calcium Phosphate ( $\text{Ca}_5(\text{PO}_4)_3(\text{OH})$ ) Cements (CPCs)
- Fly Ash / Aluminum Phosphate Cements
- Calcium Aluminum Phosphate Cements

# DPC Filler Target Property Requirements

- Slurry Properties
  - Low viscosity  $< 1,000$  cPi.
  - Non-Newtonian fluid flow (shear thinning) is likely beneficial.
  - Slow set or thermal set – slurry working time of 8 hours.
  - Aggregate grain size  $< 100$   $\mu\text{m}$ .
  - Minimal (or beneficial) reactions between slurry and DPC components.
  - Thermal stability over a wide range of temperatures  $50 - 250$   $^{\circ}\text{C}$ .



# DPC Filler Target Property Requirements

## ■ Cement Properties

- Set times of hours to days.
- Porosity well connected and  $\leq 35\%$  with permeability  $\approx 10$  mD
- Expansion Ratio (solid volume / fluid volume)  $\approx 1.0$
- Strength  $\geq 5$  MPa unconfined compressive strength (UCS).
- Near neutral pH.
- Chemically stable or alterable to stable phases over significant periods of time under repository conditions.

# Aluminum Phosphate Cements (APCs)



\*  $\text{Al}_2\text{O}_3$  to  $\text{H}_3\text{PO}_4$  ~5:1

- Simple Acid-Base Reaction with 2 Compounds (4 elements) in Water. (Wagh et al., 2003). Patented!
- Inexpensive Starting Materials ( $\text{Al}_2\text{O}_3$  and  $\text{H}_3\text{PO}_4$ ).
- Reactants form Smooth Pourable Slurries in Water that are Stable for Days at RT.
- Near Neutral pH Post Set.
- Thermal Set Cement, Reaction Initiates at ~130 °C at Room Pressure.
- Chemical water is generated as  $\text{AlPO}_4$  binder forms.



# Aluminum Phosphate Cements (APCs)

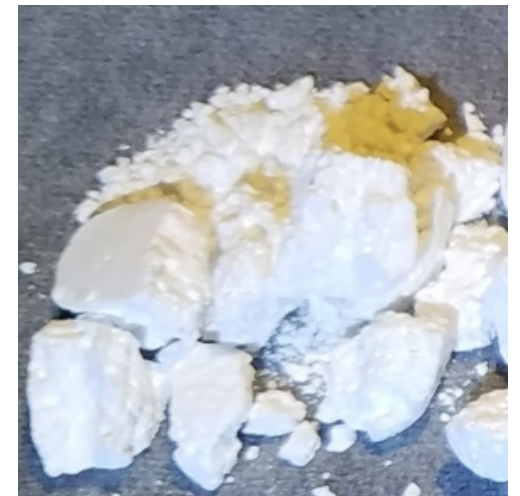


- Simple ambient pressure and hydrothermal experiments produce poor products.
  - Cements with significant macro-porosity observed in ambient pressure experiments at various temperatures and times.
  - Hydrothermal experiments in Parr Vessels produce poorly consolidated cements.

Early Attempts...



0.1 MPa Pressure 150 °C



~0.2 MPa Pressure 150 °C

# Modified APC Formulations

- The reaction  $\text{Al}_2\text{O}_3 + 2\text{H}_3\text{PO}_4 \rightarrow 2\text{AlPO}_4 + 3\text{H}_2\text{O}$  takes place at  $\geq 130^\circ\text{C}$ . Product water as steam causes large voids as APCs set at ambient pressure.
- Additional aluminum sources such as gibbsite ( $\text{Al}(\text{OH})_3$ ) and metakaolin reduce or eliminate expansion and large void formation during setting of the cement.
- Al sources react with phosphates at room temperature, causing APCs to begin setting below  $100^\circ\text{C}$ .
- $\text{NH}_4\text{H}_2\text{PO}_4$ ,  $\text{NaH}_5(\text{PO}_4)_2$ , and  $\text{NH}_4\text{H}_5(\text{PO}_4)_2$  were also tested as alternative phosphate sources.



Standard APC at Ambient Pressure



APC with metakaolin and  $\text{NaH}_5(\text{PO}_4)_2$  at Ambient Pressure

# APCs: Challenges and Opportunities

- Porosity tends to be high ( $> 40\%$ ) and permeability low (0.1-1 mD) in the APC variants.
  - Varying the  $\text{Al}_2\text{O}_3$  starting material (reactive vs calcined  $\text{Al}_2\text{O}_3$ ) and the use of  $\text{NH}_4\text{H}_2\text{PO}_4$  as a phosphate source ( $\text{pH} = 4$ ) shows promise for porosity/permeability and expansion control.
  - Aluminum modifiers, particularly metakaolin, also show promise for porosity/permeability control.
  - The addition of neutron poisons can mitigate porosity concerns but permeability must improve so that water can escape during cement setting. The addition of B-based materials and Gd-oxides tend to reduce strength, more research is necessary.

# APCs: Challenges and Opportunities

A large number of aluminum phosphate phases are observed as products ( $\alpha$ - $\text{AlPO}_4$ ,  $\beta$ - $\text{AlPO}_4$ ,  $\gamma$ - $\text{AlPO}_4$ ,  $\text{AlPO}_4 \cdot \text{H}_2\text{O}$ ,  $\text{AlPO}_4$  zeolite, amorphous aluminum phosphate phases)...

- $\text{AlPO}_4$  polymorphs and even the amorphous aluminum phosphate phases can be effective binders.

And...

- Preliminary geochemical modeling indicates  $\text{AlPO}_4$  polymorphs and  $\text{AlPO}_4 \cdot \text{H}_2\text{O}$  likely to alter to variscite  $\text{AlPO}_4 \cdot 2(\text{H}_2\text{O})$  while  $\text{Al}_2\text{O}_3$  will likely alter to boehmite ( $\text{AlOOH}$ ) with a significant increase in cement solid volume.

# Calcium Aluminate Phosphate Cements (CAPCs)

- Calcium aluminates are synthesized by heating CaO and  $\text{Al}_2\text{O}_3$  blends at high temperatures.
- Krotite  $\text{CaAl}_2\text{O}_4$  aka 'CA' reacts with phosphate to form a cement (developed and patented by Sugama, BNL).
- CA also reacts with water to form CAH bonded cements but Grossite ( $\text{CaAl}_4\text{O}_7$ ) aka 'CA2' is only weakly hydraulic.
- Grossite ( $\text{CaAl}_4\text{O}_7$ ) aka 'CA2' & Hibonite ( $\text{CaAl}_{12}\text{O}_{19}$ ) aka 'CA6' were synthesized and reacted with phosphoric acid.



Grossite  
Phosphate Cement

# Calcium Aluminate Phosphate Cements (CAPCs)

## Grossite ( $\text{CaAl}_4\text{O}_7$ ) reactions with $\text{H}_3\text{PO}_4$

- Binder phase is currently unknown and likely amorphous.
- Shrinkage can be minimal (ER=0.98).
- Recent experiments with grossite and alternative phosphate sources (sodium hexametaphosphate) are increasing working times to days.
- Starting pH's range between ~3-6.
- Binder phase is likely hydrous and an assessment of water and/or  $\text{H}_2$  gas generation upon irradiation will be initiated soon.



# Summary and Next Steps

- Currently APCs and CAPCs show the greatest promise for continued development.
- Continue process and formulation optimization of both cements.
- Future work includes:
  - Radiation stability and long-term solubility testing on optimized products.
  - Develop in-package chemistry models with fillers.
  - Small scale testing of fillers in DPC mock-ups.



# Selected References

- Achelhi, K., S. Masse, G. Laurent, A. Saoiabi, A. Laghzizil, & T. Coradin (2010). Role of carboxylate chelating agents on the chemical, structural and textural properties of hydroxyapatite. *Dalton Transactions*, 39(44), 10644-10651. doi:10.1039/C0DT00251H.
- Colorado, H.A., J. Pleitt, C. Hiel, J.M. Yang H.T. Hahn and C.H. Castro (2012) Wollastonite based-Chemically Bonded Phosphate Ceramics with lead oxide contents under gamma irradiation. *Journal of Nuclear Materials*, 425, p.197-204.
- Hardin, E.L., P.V. Brady and C. Bryan, “Joint Workplan on Filler Investigations for DPCs,” SFWD-SFWST-2018-000481 Rev. 0, SAND2017-13727 R, Sandia National Laboratories, Albuquerque, New Mexico, December 2017.
- Hardin, E.L. and P.V. Brady (2018). *Recommendations for Filler Material Composition and Delivery Method for Bench- Scale Testing*. SFWD-SFWST-2018-000490 Rev. 0. U.S. Department of Energy, Office of Spent Fuel and Waste Science and Technology. March, 2018.
- Krogstad, D.V., D. Wang & S. Lin-Gibson (2017). Polyaspartic Acid Concentration Controls the Rate of Calcium Phosphate Nanorod Formation in High Concentration Systems. *Biomacromolecules*, 18(10), 3106-3113. doi:10.1021/acs.biomac.7b00772
- SNL (Sandia National Laboratories) 2017. *Joint Workplan on Filler Investigations for DPCs*. SFWD-SFWST-2018-000481 Rev. 0. U.S. Department of Energy, Office of Spent Fuel and Waste Science and Technology. December, 2017.
- Wagh, A.S., S. Grover, and S.Y. Jeong, *Chemically Bonded Phosphate Ceramics: II Warm Temperature Process for Alumina Ceramics*. *Journal of the American. Ceramic. Society.*, 86, 1845-1849, 2003.
- Wagh, A.S. *Chemically Bonded Phosphate Ceramics: 21st Century Materials with Diverse Applications (1st Edition)*. Elsevier Science. 2004.

# Selected References

- Wagh, A.S. *Chemically Bonded Phosphate Ceramics: Twenty-First Century Materials with Diverse Applications (2nd edition)*. Elsevier. 2016.
- Zhang, J., W. Liu, V. Schnitzler, F. Tancret & J.-M. Bouler (2014). Calcium phosphate cements for bone substitution: Chemistry, handling and mechanical properties. *Acta Biomaterialia*, 10(3), 1035-1049. doi:<https://doi.org/10.1016/j.actbio.2013.11.001>

Questions?