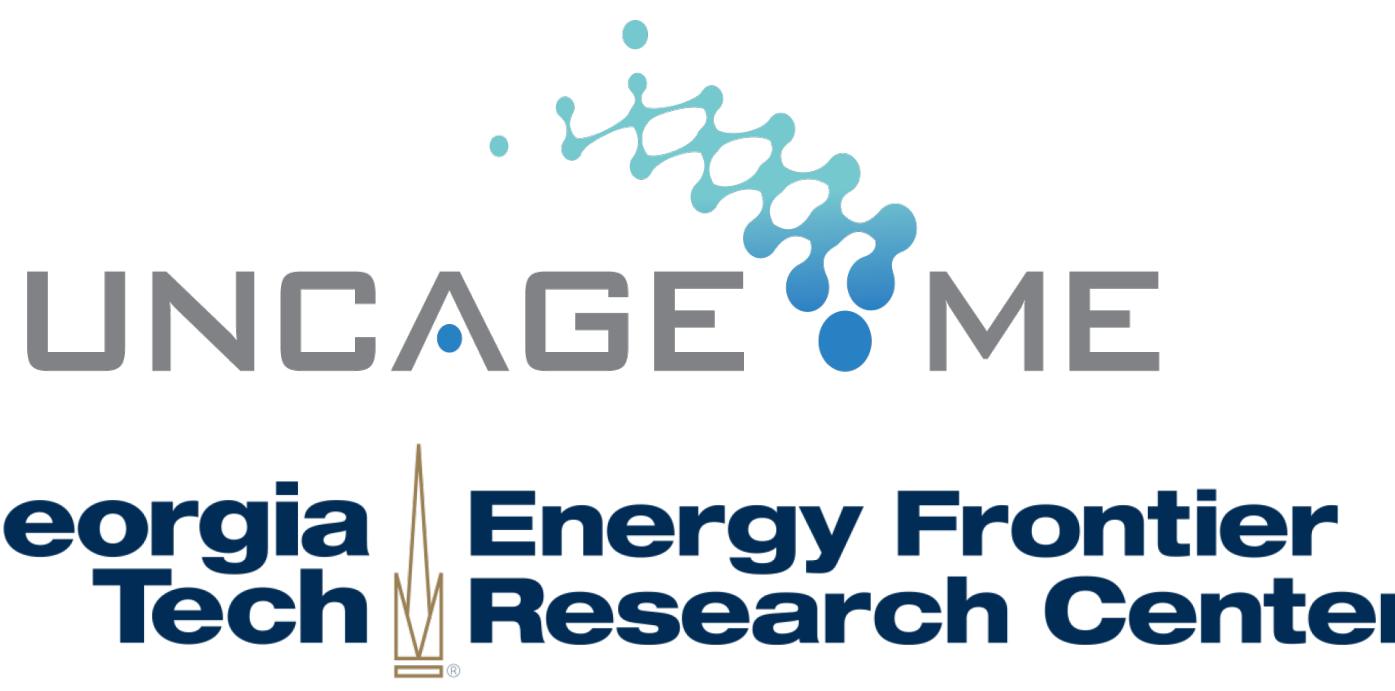


# Unique Magnetic Responses in RE-DOBDC MOFs with $\text{NO}_x$ Adsorption

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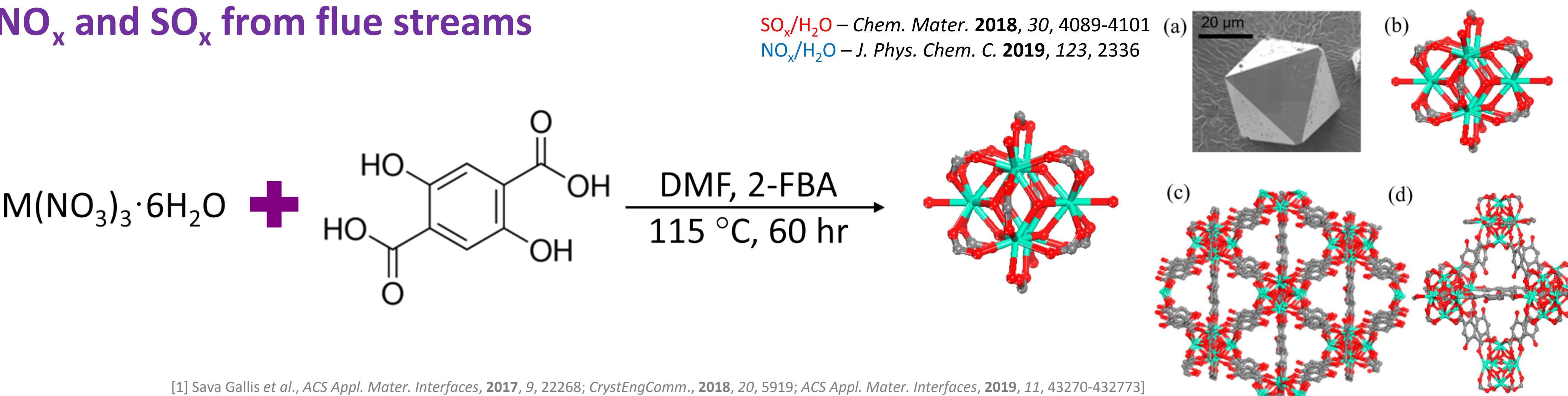
## Introduction

**AIM: Design and synthesize a RE-containing MOF for the selective adsorption of  $\text{NO}_x$  and  $\text{SO}_x$  from flue streams**

UNCAGE-ME Center exemplar "Complex Mixtures":

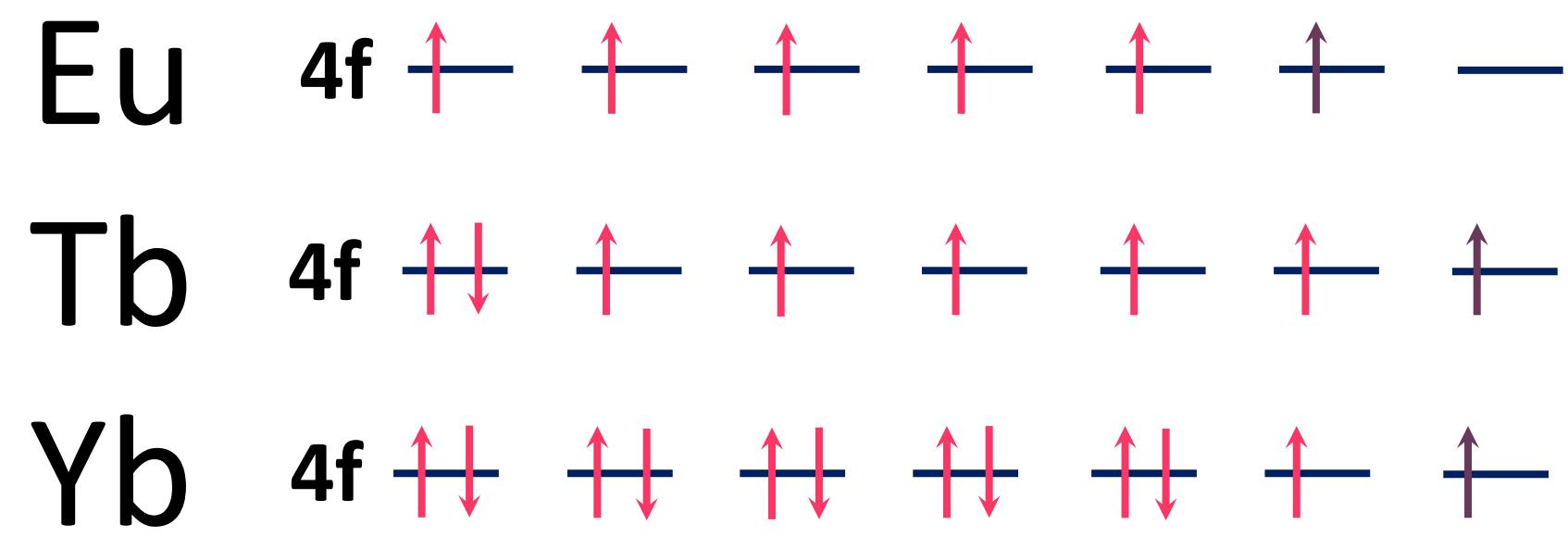
• Coal-fired power plant flue gas (13%  $\text{CO}_2$ , 6%  $\text{H}_2\text{O}$ , ~4%  $\text{O}_2$ , 50 ppm  $\text{CO}$ , 420 ppm  $\text{NO}_2$ , 420 ppm  $\text{SO}_2$ , 76%  $\text{N}_2$ )

- MOFs are inorganic-organic hybrid materials with extremely high surface area and the ability to form CUSs
- Rare earth (RE) elements selectively bind to acid gases ( $\text{NO}_x$  and  $\text{SO}_x$ )<sup>1</sup>
- Flue gases are produced from combustion processes and are a large environmental concern
- **Detection and capture of toxic  $\text{NO}_x$**  is important for emissions control of exhaust gases and human health
- RE-DOBDC shows a reduction in paramagnetism with adsorption of  $\text{NO}_x$  – In collab. with CINT
- RE-DOBDC show potential for use as a **magnetic sensor to detect  $\text{NO}_x$**

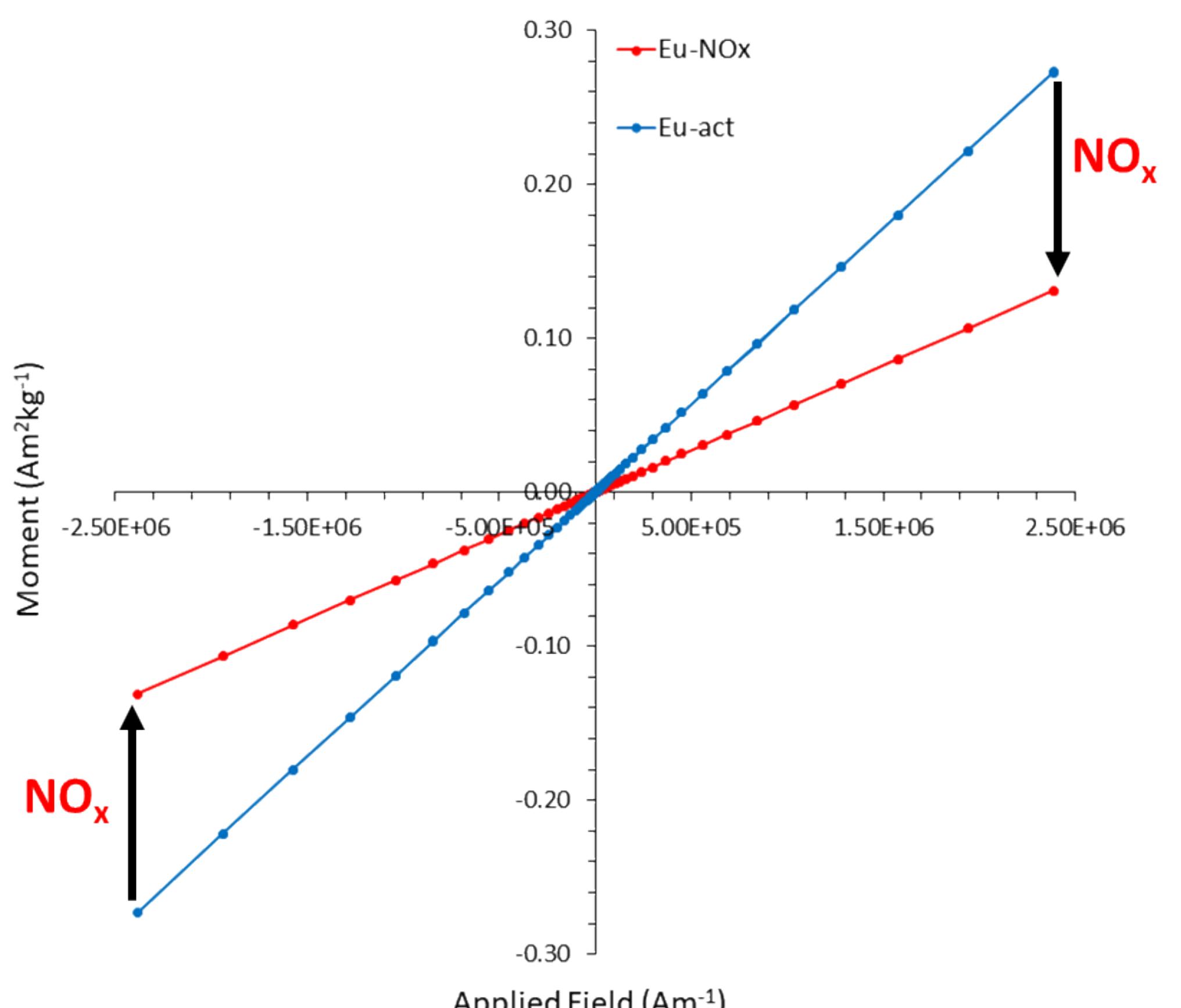


[1] Sava Gallis et al., ACS Appl. Mater. Interfaces, 2017, 9, 22268; CrystEngComm., 2018, 20, 5919; ACS Appl. Mater. Interfaces, 2019, 11, 43270-432773]

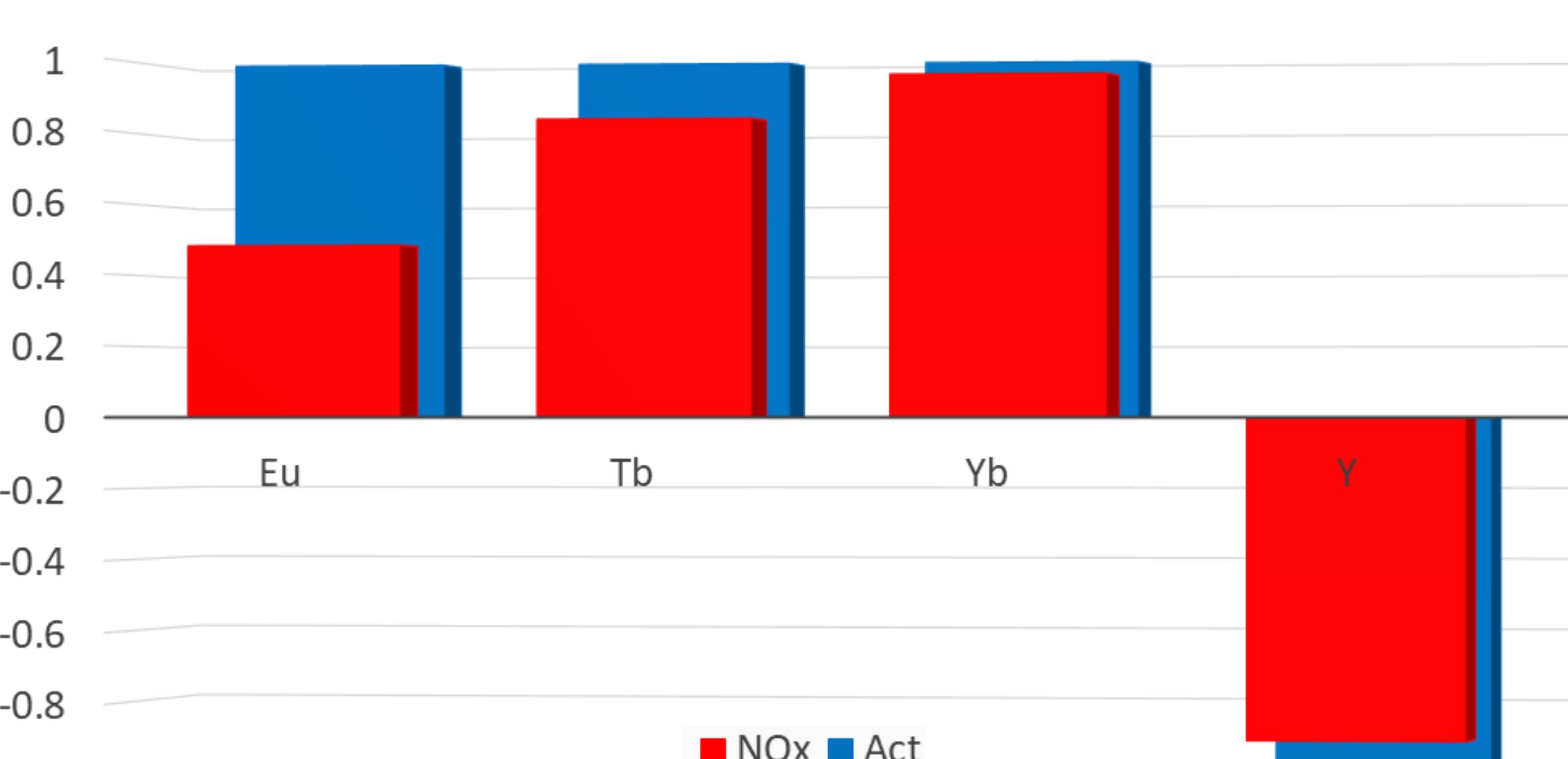
## The effect of 1 unpaired f electron



- RE-DOBDC MOFs were produced under solvothermal methods in DMF at 120 °C for 3 days
- Each MOF was activated (blue) and then subsequently loaded with humid  $\text{NO}_x$  (red)
- Each MOF was investigated upon using a VersaLab VSM as a function of metal center.

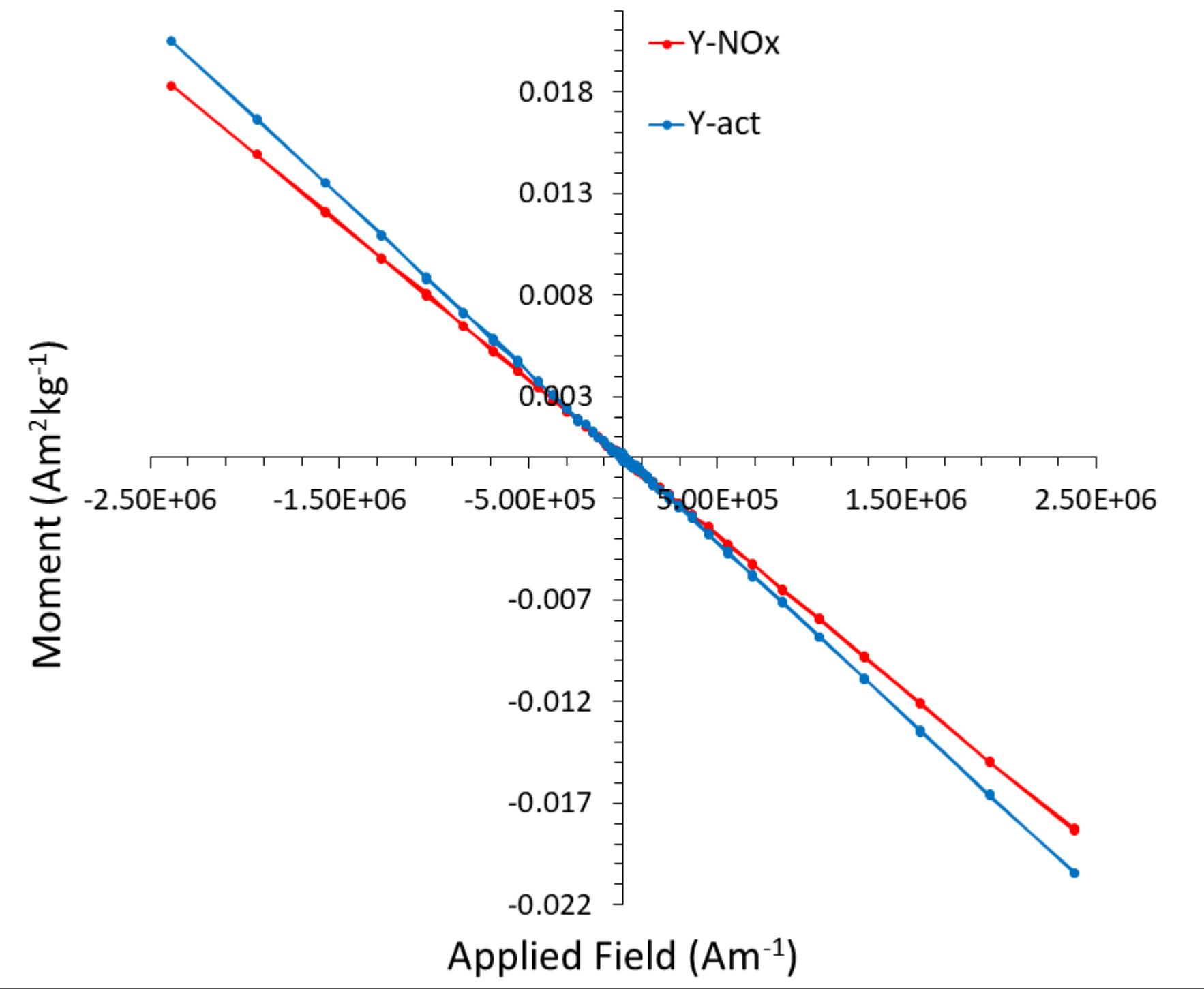


- Eu, Tb, Yb-DOBDC highlight a classic paramagnetic susceptibility with the susceptibility reduced upon addition of  $\text{NO}_x$
- $\text{Eu} > \text{Tb} > \text{Yb}$   $\Delta\chi$  with addition of  $\text{NO}_x$  (EW group)
- $\Delta\chi$  increases with increase in amount of unpaired electrons
- Charge transfer of 1 unpaired electron (purple) from: metal  $\rightarrow$  ring  $\rightarrow$  EW group  $\therefore$  reducing magnetism



- Comparison of the change in magnetic susceptibility for RE-DOBDC MOFs pre- and post- $\text{NO}_x$  exposure
- **Largest change = Eu-DOBDC**
- Positive values = paramagnetic
- Negative values = diamagnetic

## Paramagnetic vs. Diamagnetic



[4] Henkelis et al., Magnetic Tunability in DOBDC MOFs via Acid Gas Adsorption, ACS Appl. Mater. Interfaces, 2020, 12, 19504-19510

- Y contains no f electrons
- $\text{Y}^{3+}$  is paramagnetic
- Y donates its 1 up to form the MOF framework
- Y-DOBDC – diamagnetic; susceptibility  $\downarrow$  with  $\text{NO}_x$ <sup>4</sup>

## Conclusions

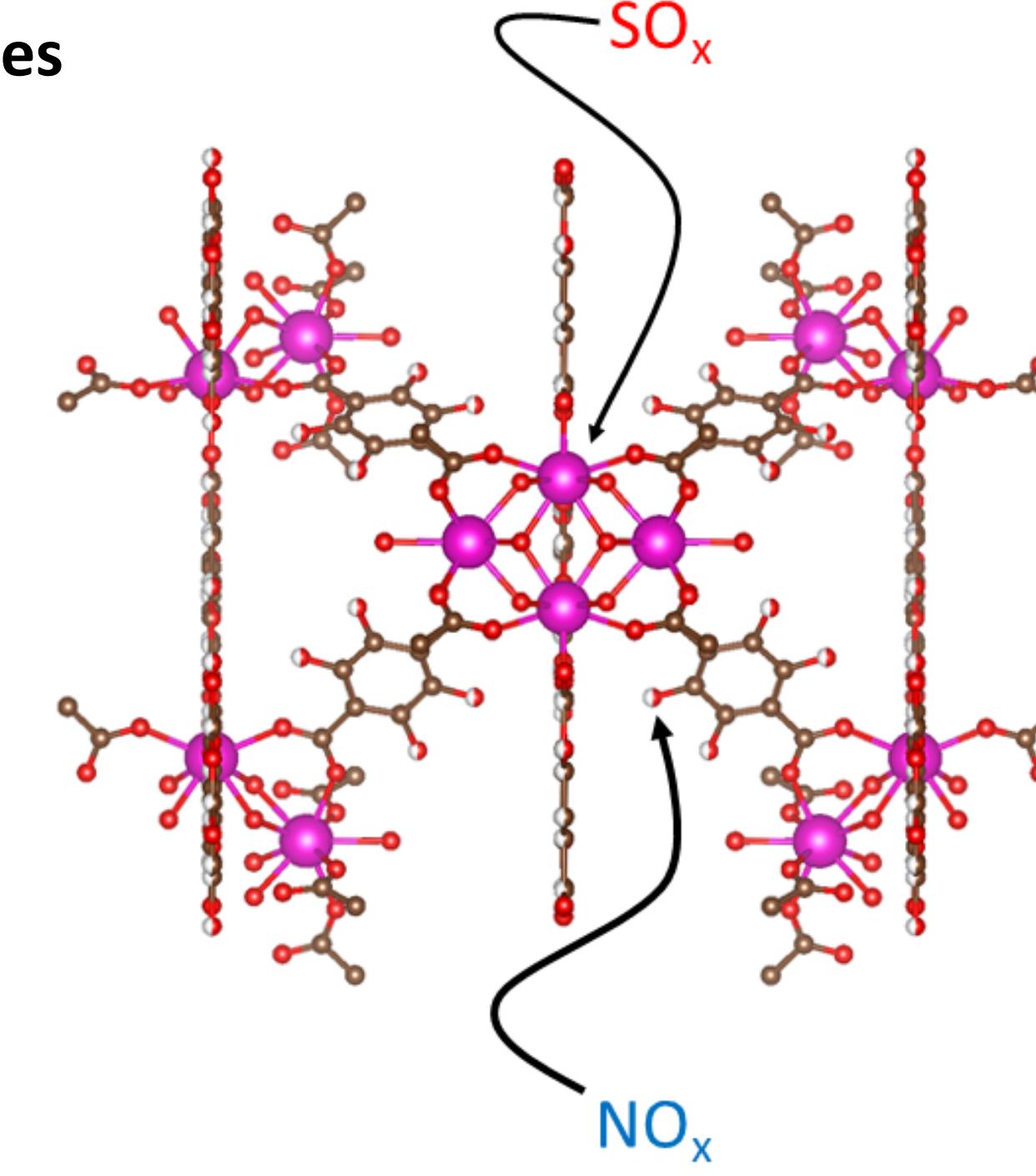
- RE-DOBDC MOFs are structurally stable to humid acid gas streams – studies into the structure/property relationships are on-going
- Potential ability for use as a magnetic sensor for  $\text{NO}_x$
- Complimentary to optical and electronic sensing methods for  $\text{NO}_x$ <sup>1</sup>
- $\text{NO}_x$  reduces overall magnetism of RE-DOBDC
  - Eu, Tb, Yb – paramagnetic susceptibility
  - $\text{Eu} > \text{Tb} > \text{Yb}$   $\Delta\chi$  with addition of  $\text{NO}_x$
  - Y – diamagnetic susceptibility



- Quenching of photoluminescence with adsorption of  $\text{NO}_x$  – ligand binding
- No quenching with adsorption of  $\text{SO}_x$  – metal binding

**On-going research – Investigation into competitive  $\text{H}_2\text{O}/\text{NO}_x/\text{SO}_x$  binding in DOBDC containing MOFs (RE-DOBDC, MOF-74) for selective removal of acid gases from flue streams**

[1] Sava Gallis et al., ACS Appl. Mater. Interfaces, 2019, 11, 43270-432773, Henkelis et al., ACS Appl. Mater. Interfaces, 2020, 12, 22845-22852, Small et al., US Patent Application, 2021



## Acknowledgements



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