

Atomic Scale Observations of Dislocation Core Structures in Bismuth Telluride

**D.L. Medlin¹, N. Yang¹, K. Erickson¹,
M. Siegal², G. Yelton², S. Limmer²**

¹Sandia National Laboratories, Livermore CA

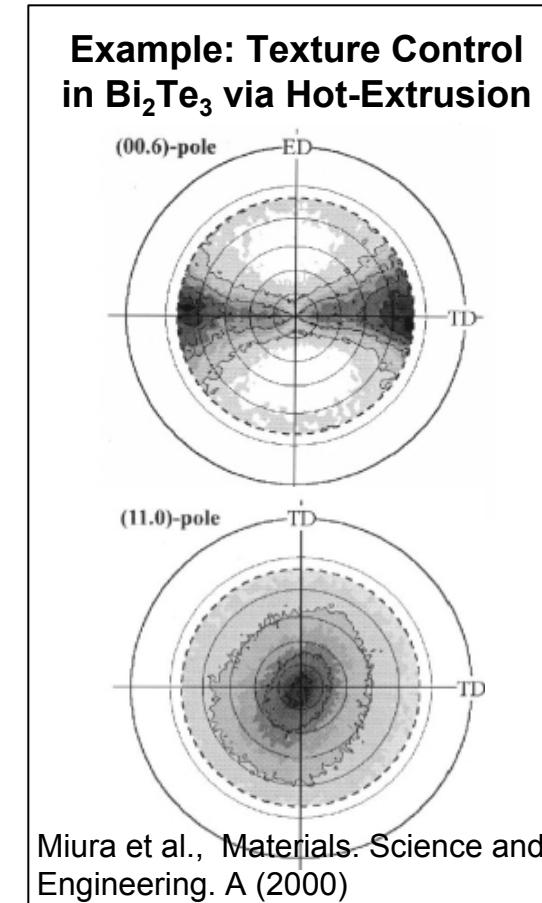
²Sandia National Laboratories, Albuquerque NM

dlmedli@sandia.gov

Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

Why care about dislocations in Bi_2Te_3 ?

- Impact on transport properties
- Processing:
 - Deformation processing is common:
 - hot pressing, extrusion
 - Thin film nanostructure growth, epitaxy,
 - coherency strains accommodated by interfacial defects.
 - non-stoichiometric defect cores.
- Elementary interest:
 - Weak van der Waals bonding and complex crystal structures yield interesting defect structures.



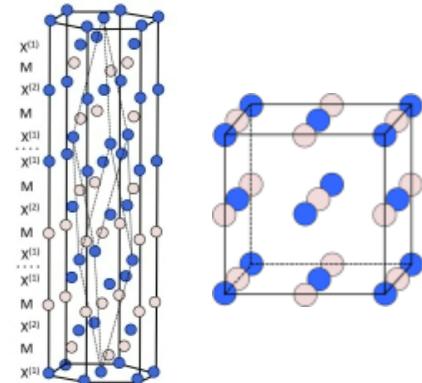
Focus for this presentation:

-Application of atomic resolution microscopy and dislocation theory to understand structure of dislocations in Bi_2Te_3

Outline

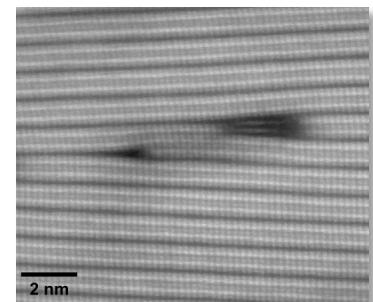
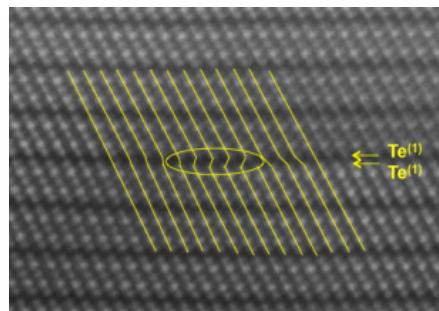
- **Crystal structure in tetradymite-type compounds**

- Layered structure gives flexibility in accommodating compositional variations
- Close structural relationship to rock-salt chalcogenides



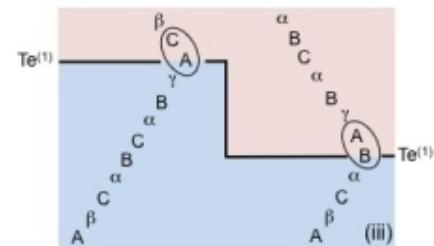
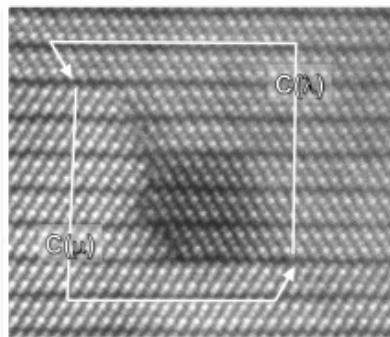
- **Dislocation Structure in Bi_2Te_3**

- Basal dislocations
- Non-basal dislocations
- Low angle grain boundaries

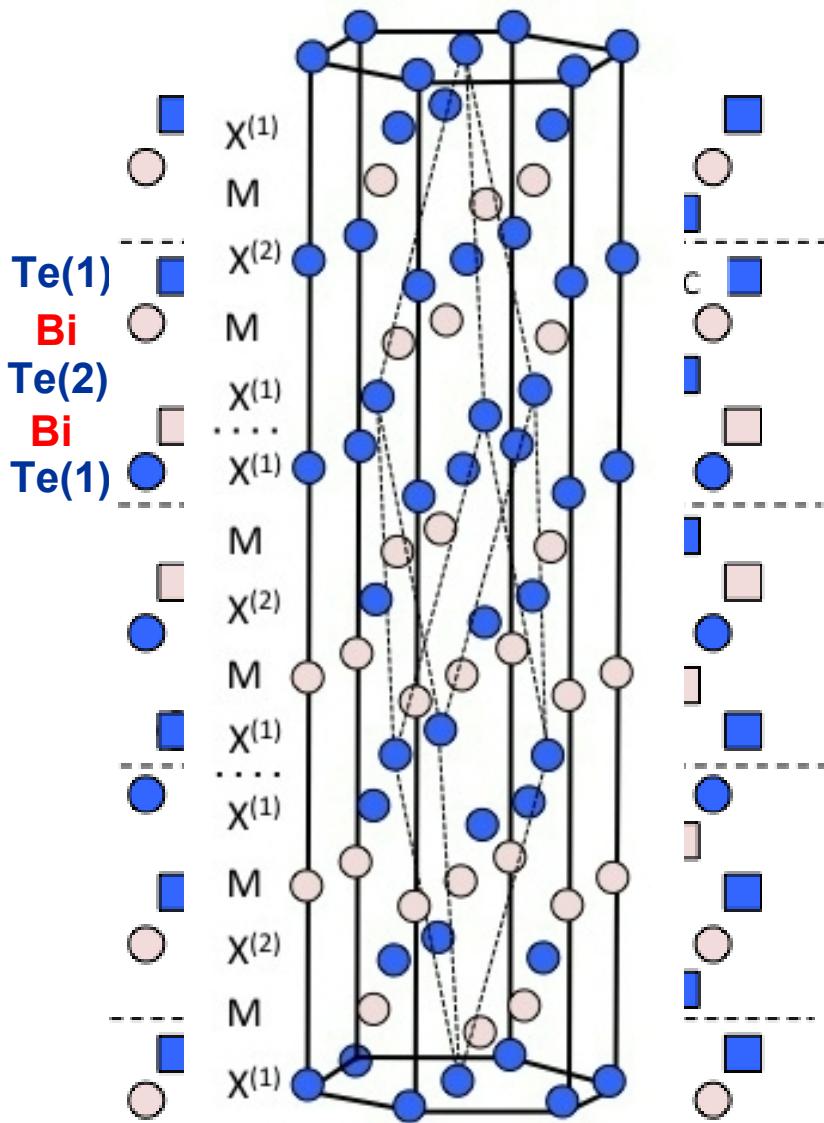


- **Interfacial Dislocations**

- High angle grain boundaries
- Example: step defect at twin boundary

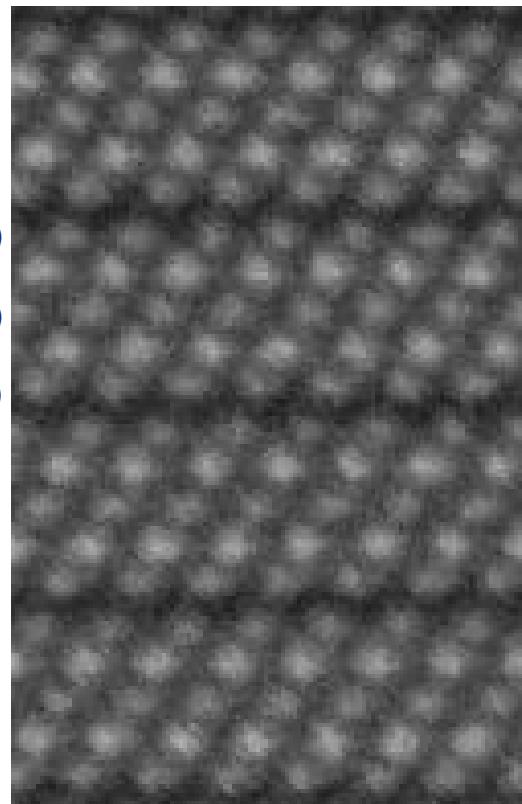


Bismuth Telluride (Bi_2Te_3): Crystal Structure



- Rhombohedral (R-3m) structure
- Based on tetradymite (Bi_2STe_2) prototype
- Three crystallographically distinct atomic sites
- $\text{Te}^{(1)}\text{-Te}^{(1)}$ layers: van der Waals bonding

HAADF-STEM



Bi: $Z=83$
Te: $Z=52$

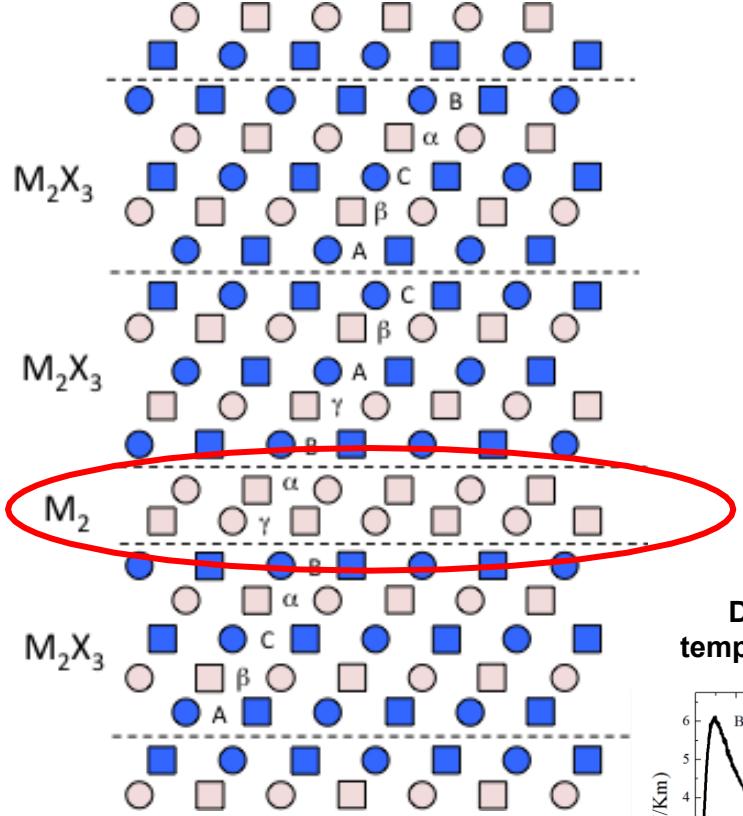
Atomic number difference enables Bi and Te to be distinguished in HAADF-STEM



Sandia
National
Laboratories

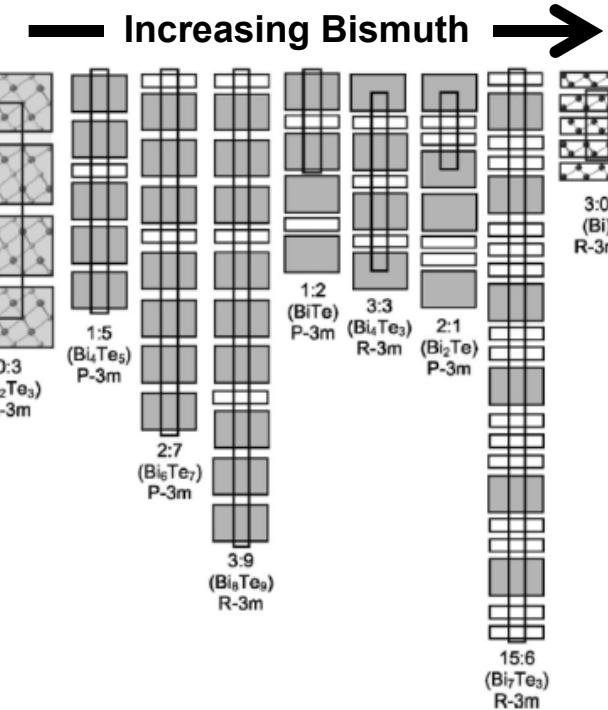
Layered structure allows flexibility in accommodating variations in composition

Insertion of metal bi-layers

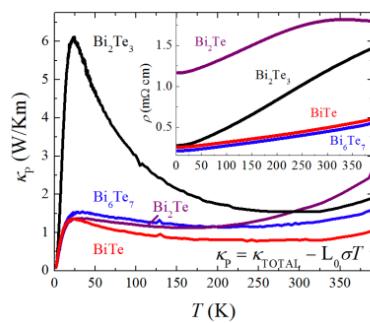


Example:
 $(Bi_2)_m(Bi_2Te_3)_n$
 homologous series

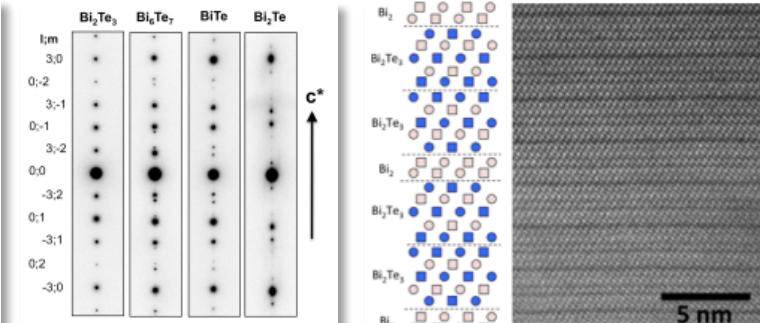
J.W.G. Bos, H.W.
 Zanderbergen,
 M.-H. Lee, N.P. Ong,
 R.J. Cava,
 Phys. Rev B 2007



Dramatic reduction in low-temperature thermal conductivity

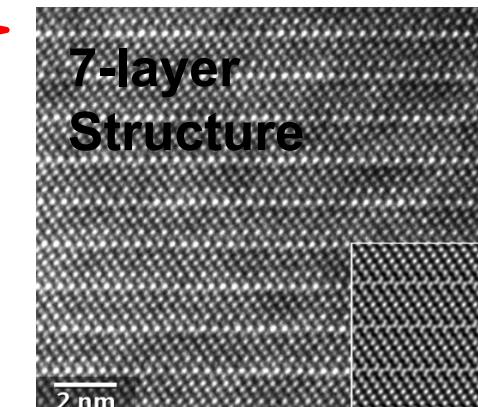
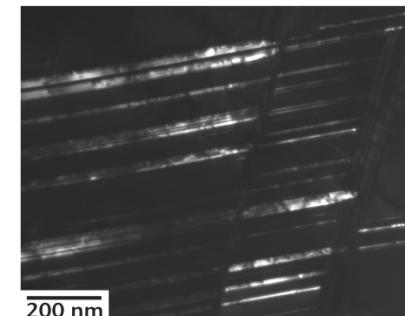
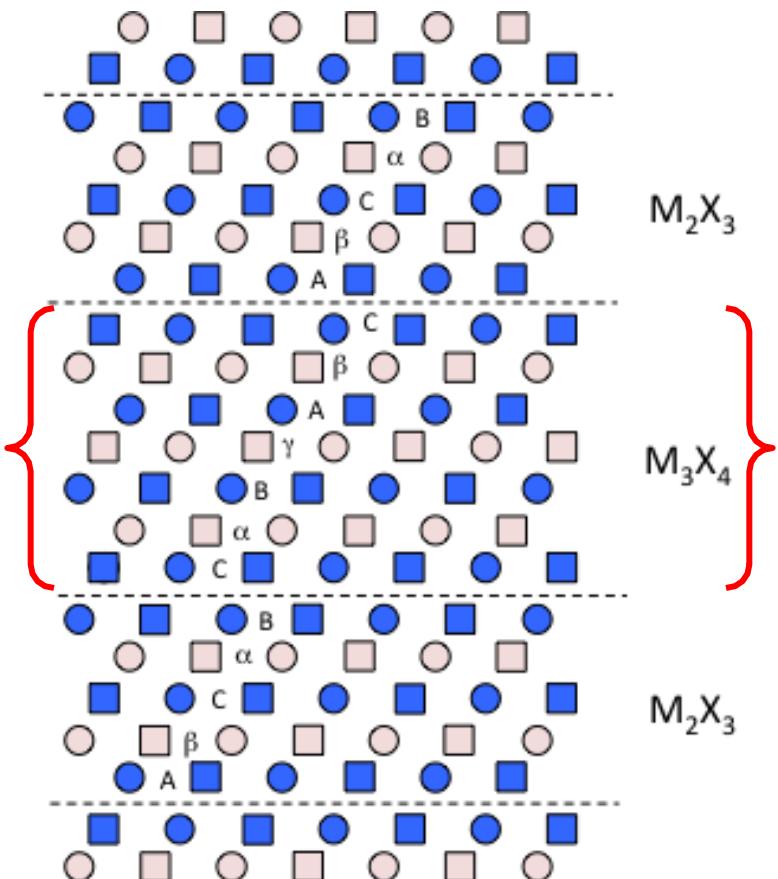


Electron diffraction and HRSTEM show ordering in $(Bi_2)_m(Bi_2Te_3)_n$ series

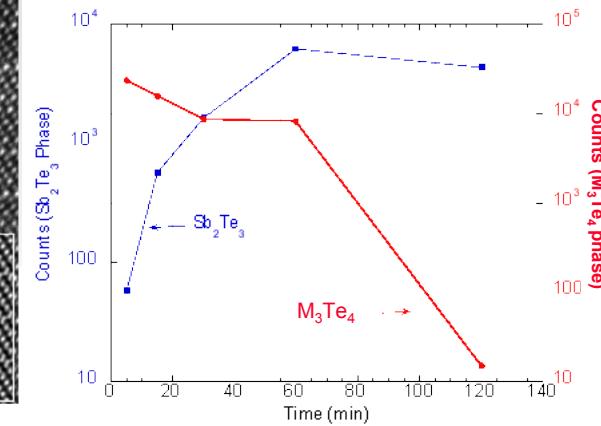
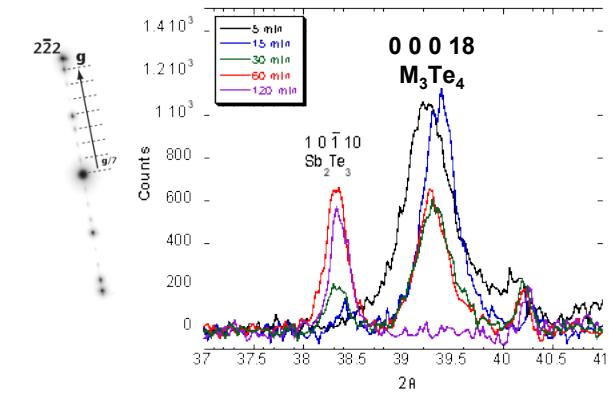


Layered structure allows flexibility in accommodating variations in composition

Metal rich, 7-Layer M_3X_4 fault



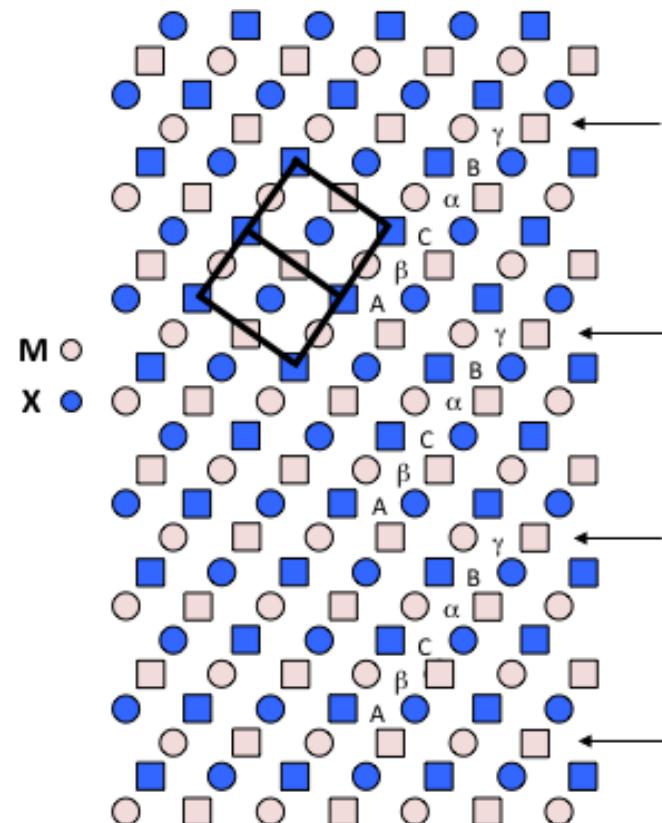
Example: $(Ag, Sb)_3Te_4$ transition phase during nucleation of Sb_2Te_3 precipitates in $AgSbTe_2$



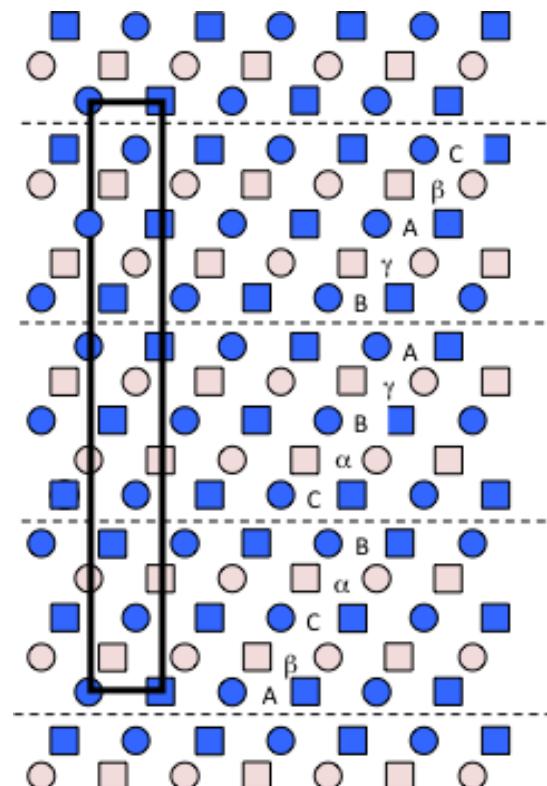
Sharma, Sugar, & Medlin, Journal of Applied Physics (2010).
Sugar and Medlin, Journal of Materials Science (2011)

Tetradymite and Rocksalt structures are closely related

Rocksalt (MX) Structure

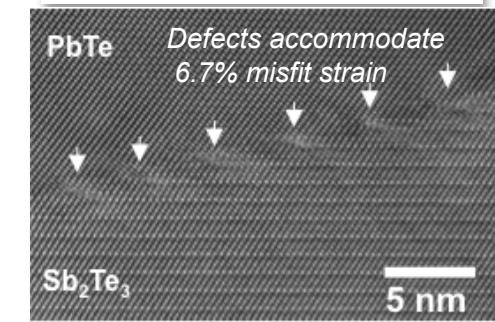
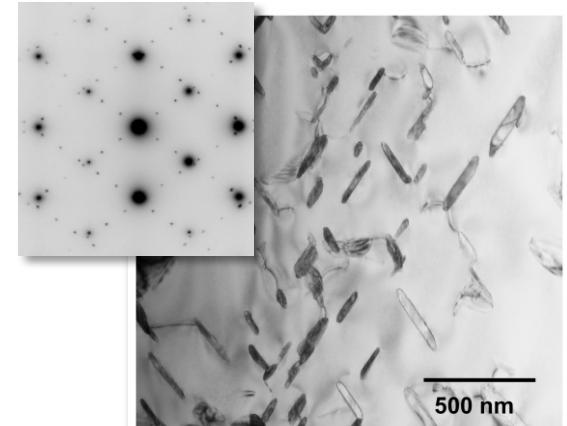


Tetradymite (M_2X_3) Structure



Example:

Crystallographically Aligned Sb_2Te_3 precipitates in PbTe

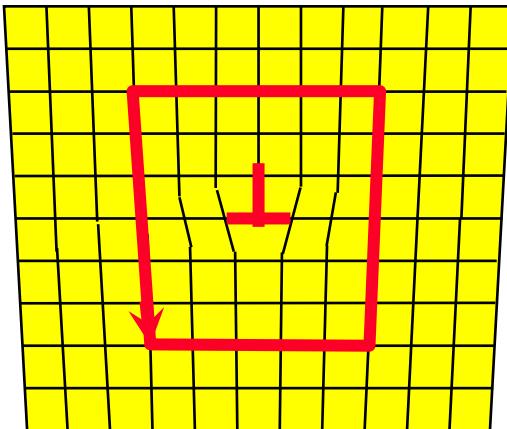


Heinz, Snyder, Ikeda, and Medlin, Acta Mat. 2011

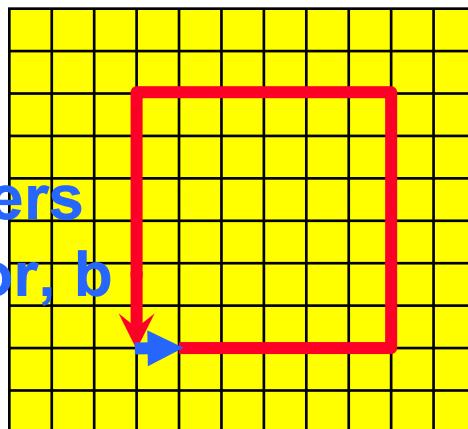
- Solid-state phase transformations in chalcogenides
- Structural interpretation of extended defects

How do these structural considerations impact Dislocations and Interfaces?

Edge Dislocation

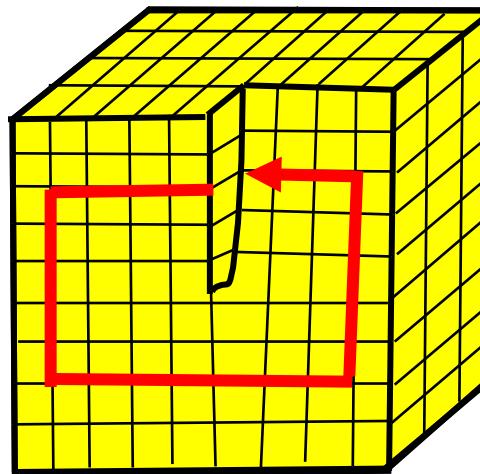


Reference Frame:
Perfect Crystal

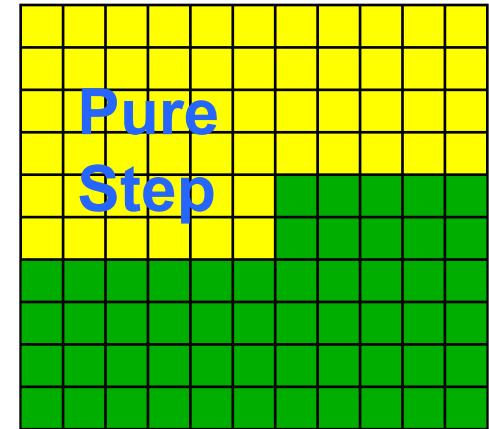


**Burgers
Vector, \mathbf{b}**

Screw Dislocation



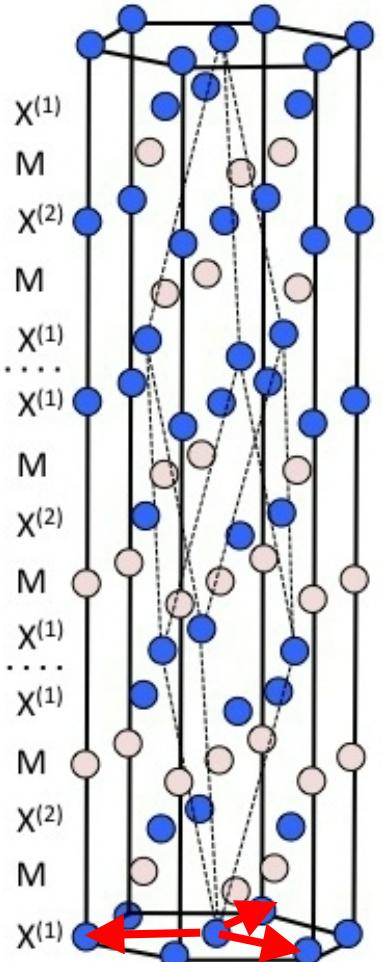
Interfacial Disconnection



Dislocations in Bismuth Telluride

Basal dislocations: (1/3) <2-1-10>

Burgers vectors lying in basal plane



Array of $1/3<2-1-1 0>$ Dislocations in Bi_2Te_3

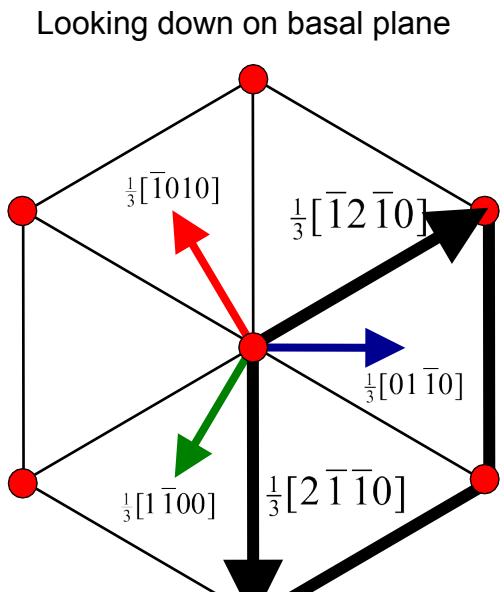
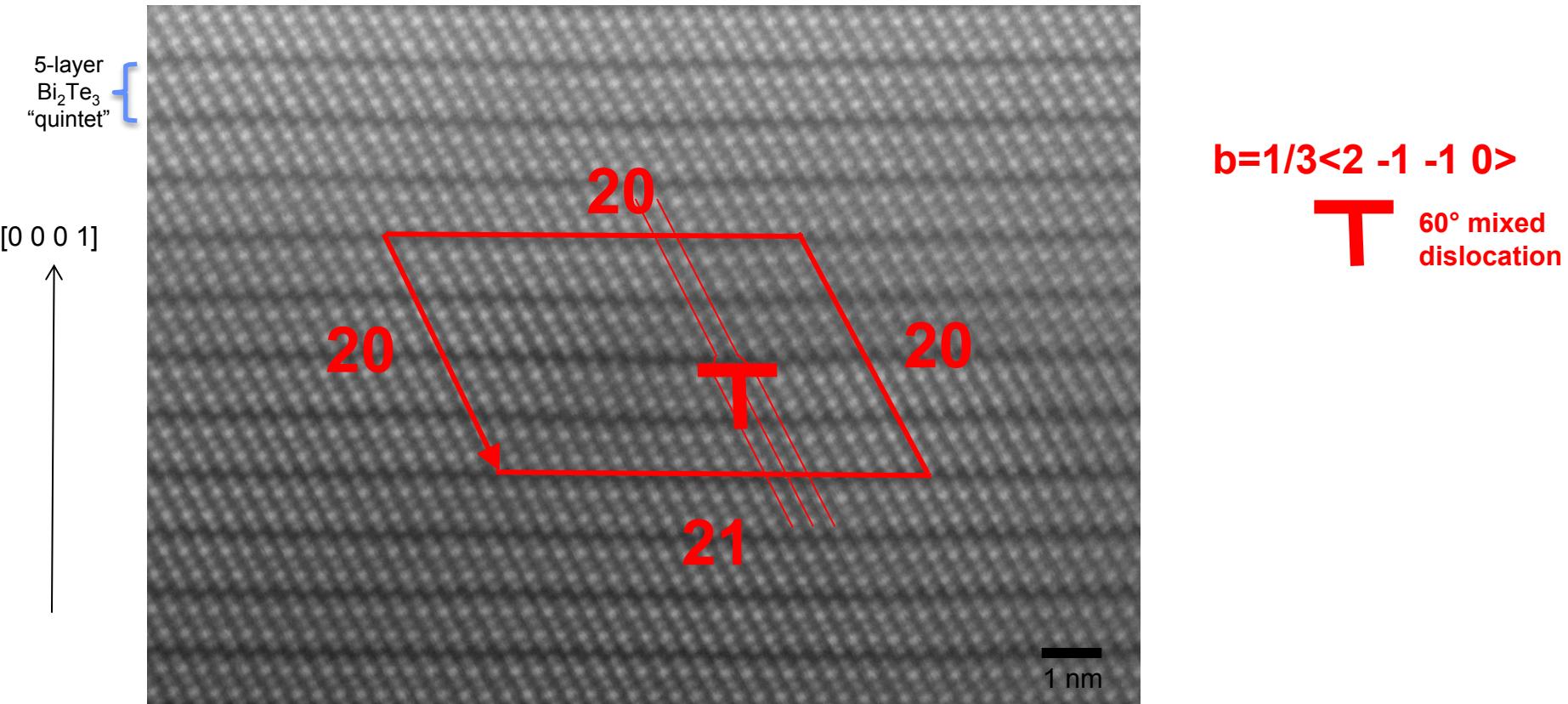


Fig. 1. Dislocation network in Bi_2Te_3 . Note that certain segments of dislocations and certain node-points have left the foil

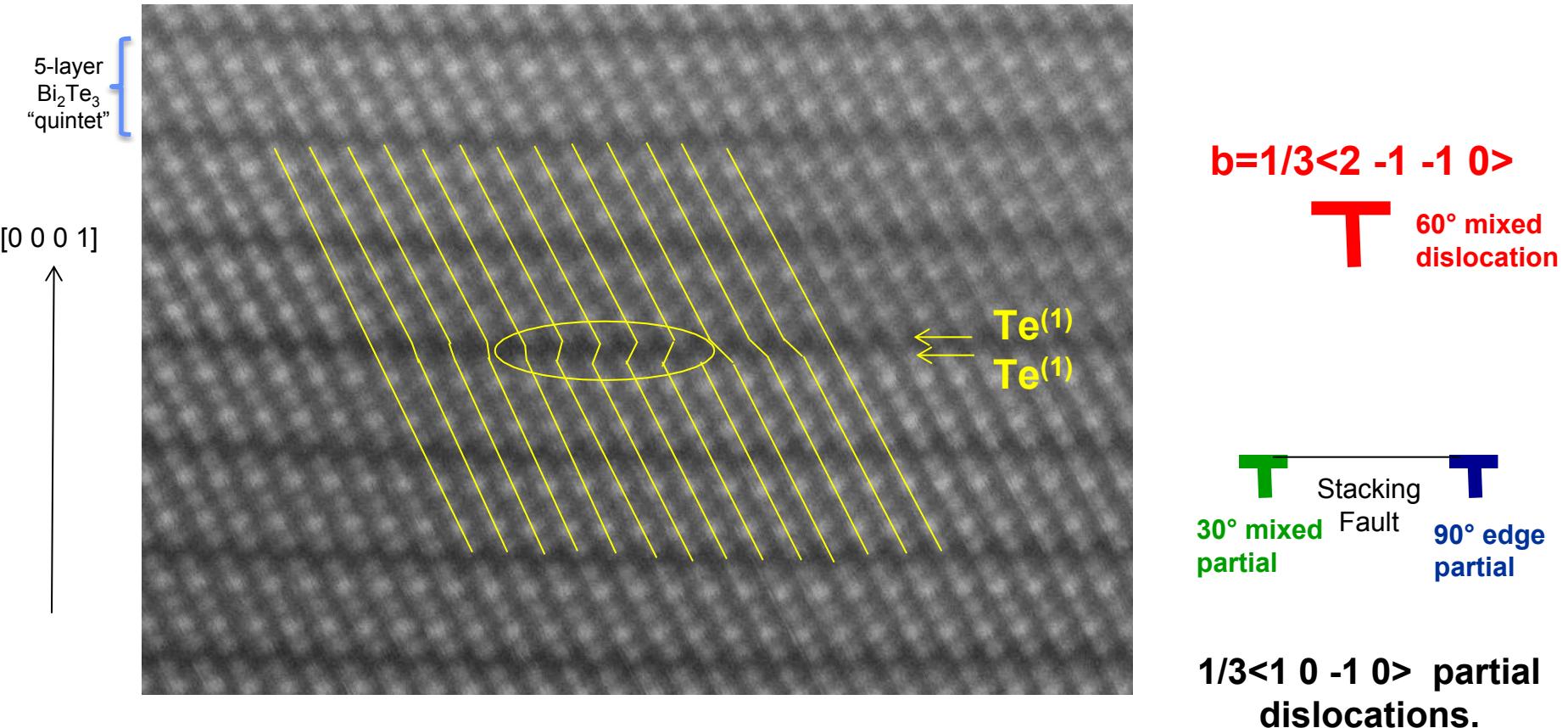
Dislocation Core structure: termination at Te⁽¹⁾-Te⁽¹⁾ layer, dissociation



HAADF-STEM
300 keV

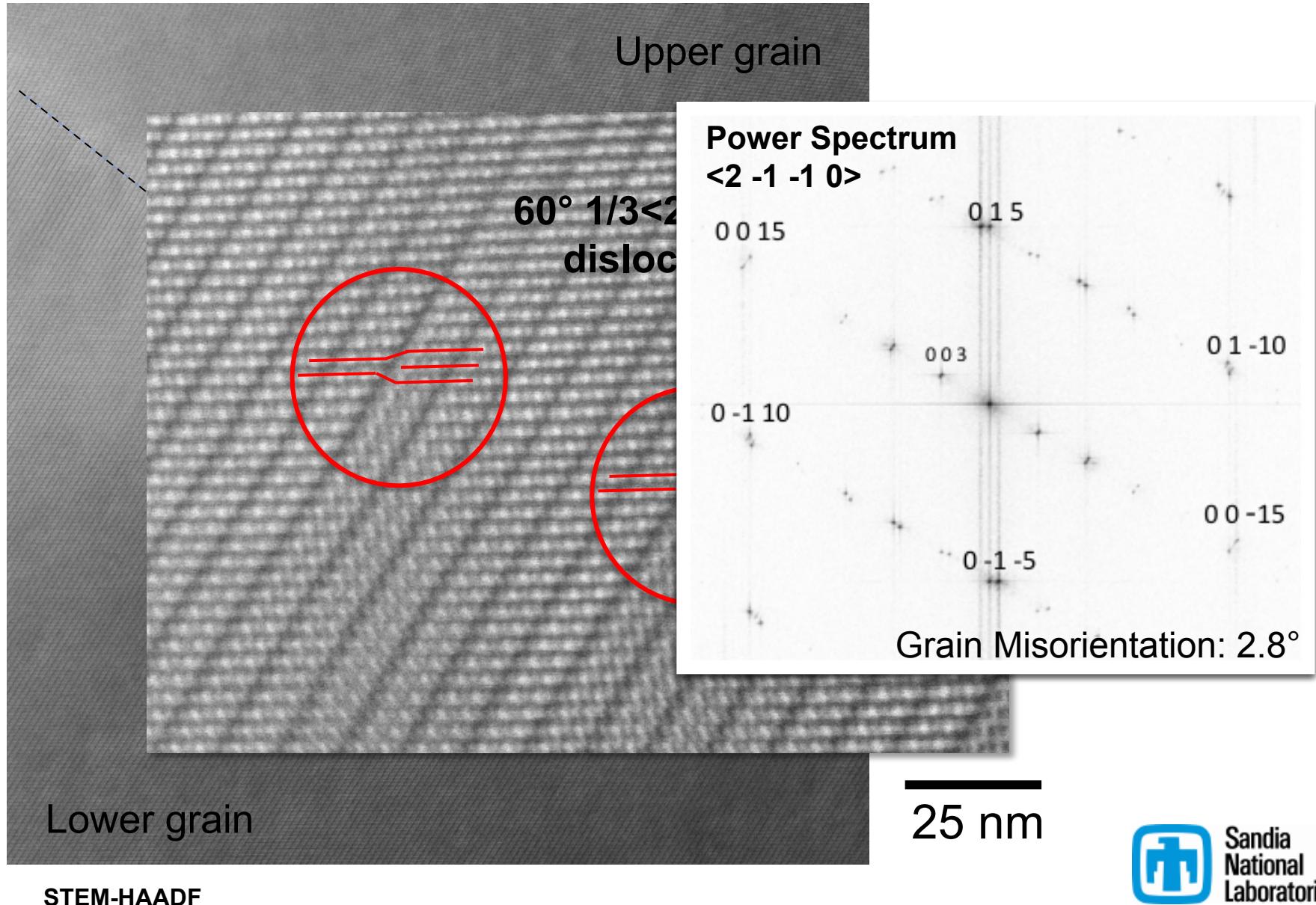
<2 -1 -1 0 > projection

Dislocation Core structure: termination at Te⁽¹⁾-Te⁽¹⁾ layer, dissociation

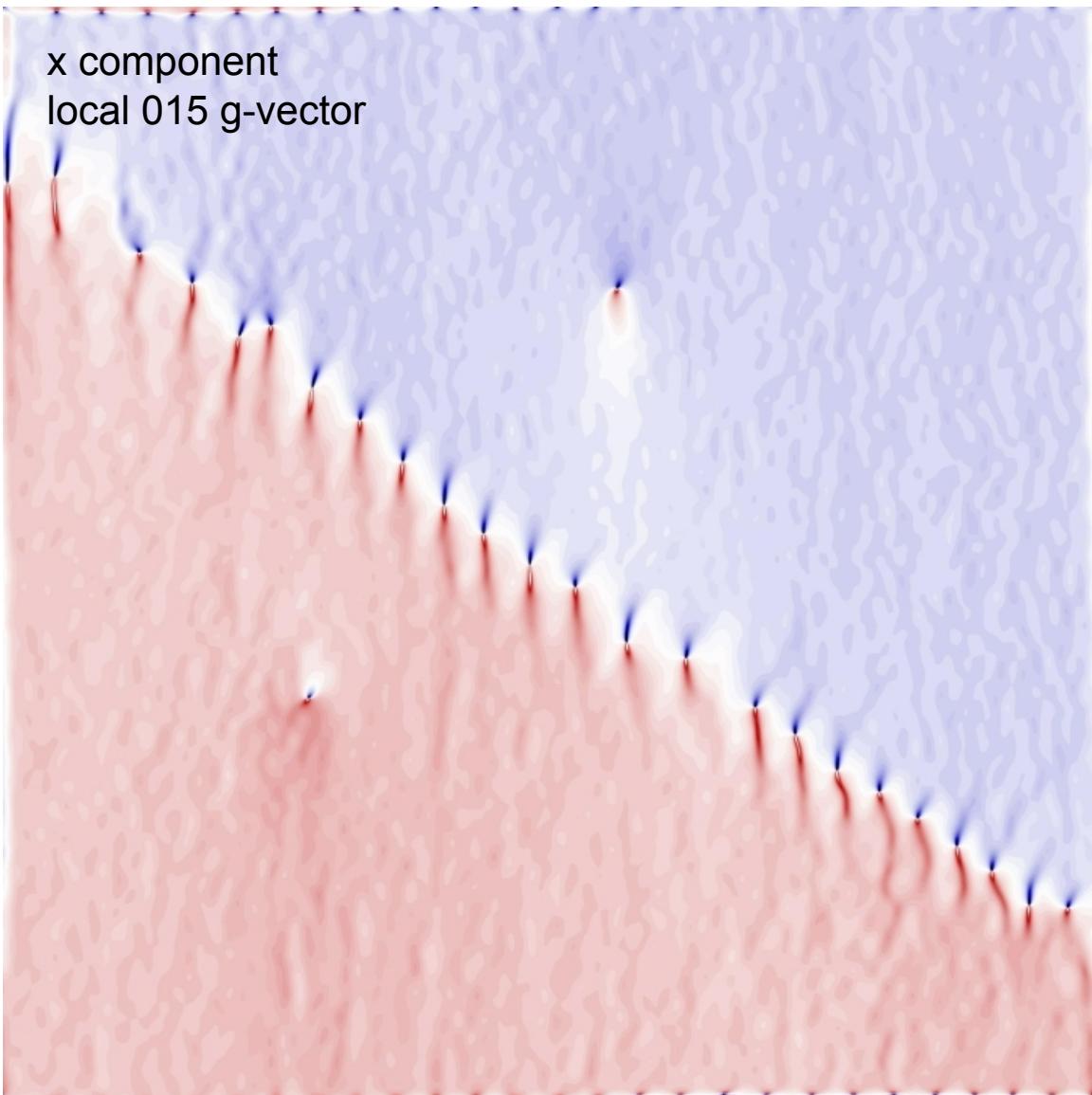


Weak, van der Waals bonding likely favors dissociated dislocation core at Te⁽¹⁾-Te⁽¹⁾ layer

A Low Angle Tilt Boundary



A Low Angle Tilt Boundary

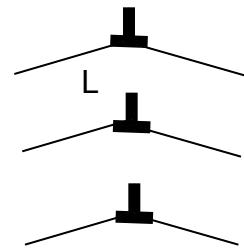


Calculate tilt rotation
from dislocation density

$$\theta = 2 \sin^{-1} \left(\frac{b_{edge}}{2L} \right)$$

= 2.7° (calculated)
(2.8° measured)

$$\begin{aligned}L_{avg} &= 8.0 \text{ nm} \\b_{edge} &= 0.379 \text{ nm} \\(60^\circ 1/3 <2-1-10> \\b &= 0.438 \text{ nm})\end{aligned}$$

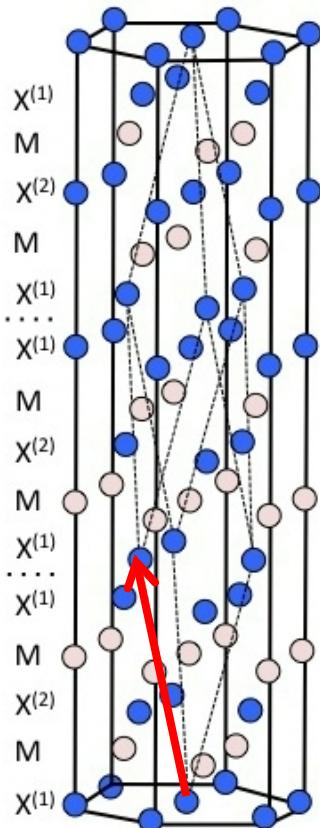


(for array of 60° dislocations,
to be pure tilt, screw components
must cancel)

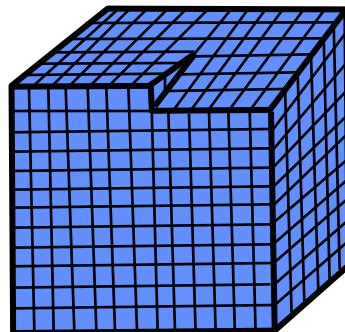
25 nm

Non-basal dislocations:

Screw dislocations important to crystal growth

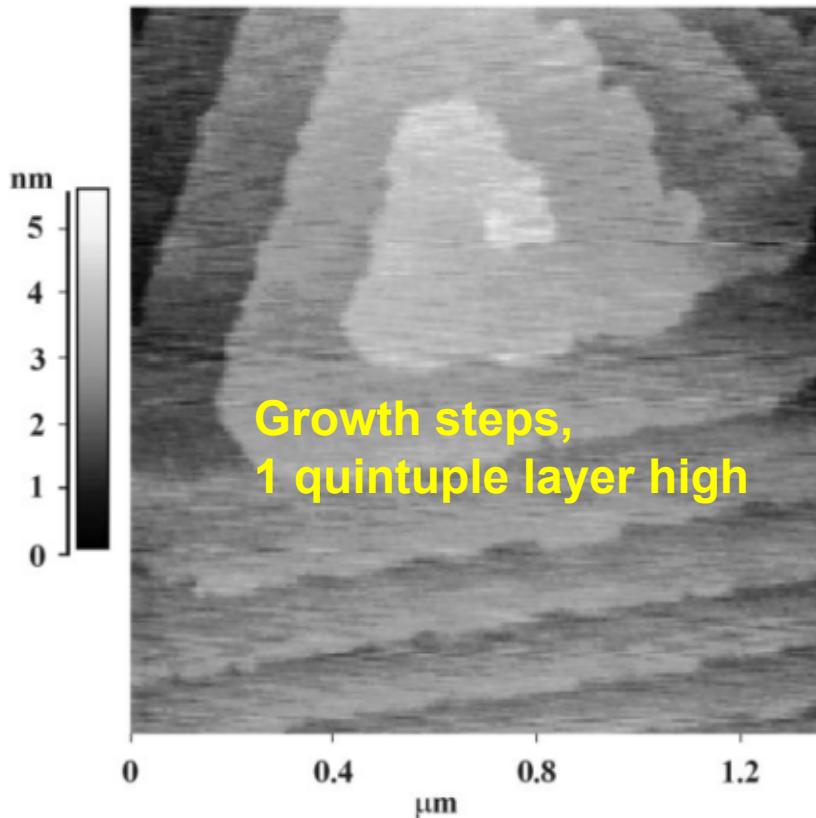


Screw Dislocation



$$\mathbf{b} = (1/3)[0-111]_H = [001]_R$$

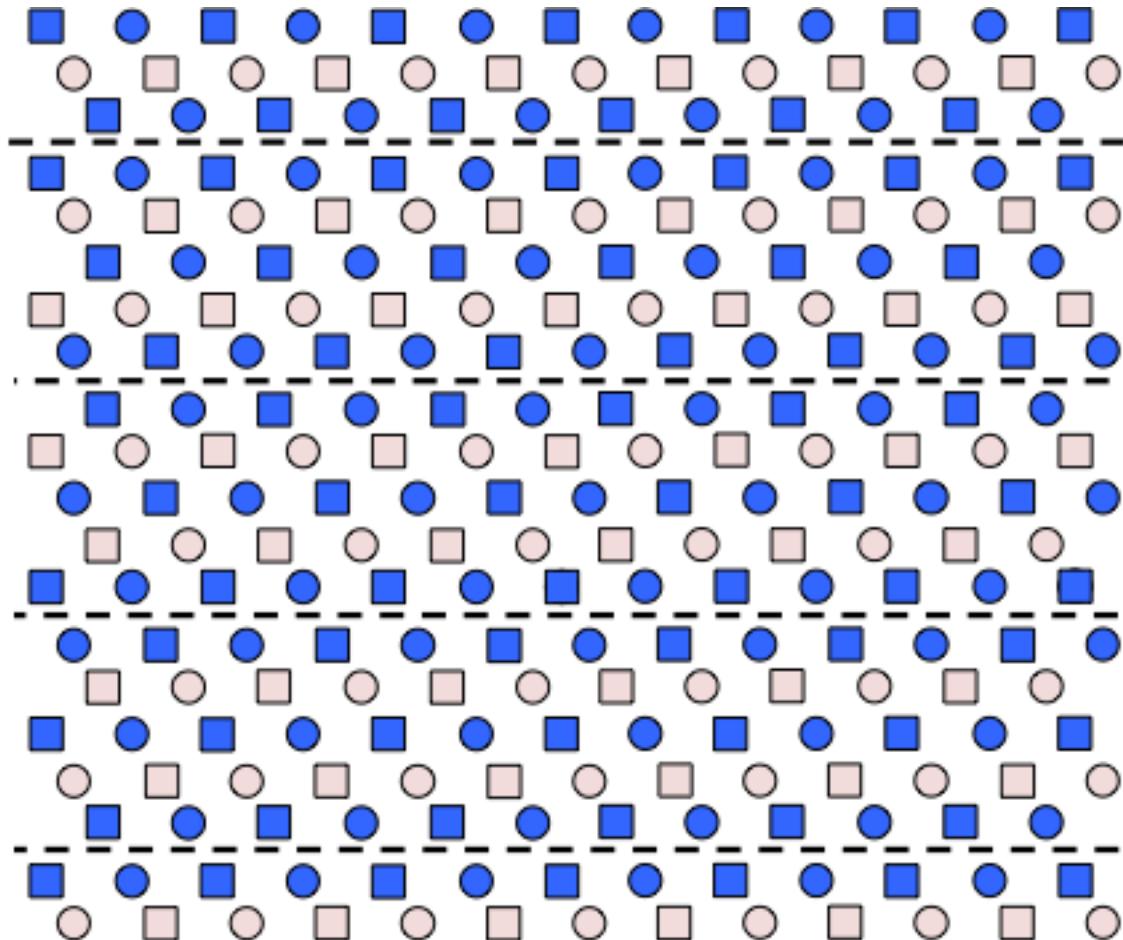
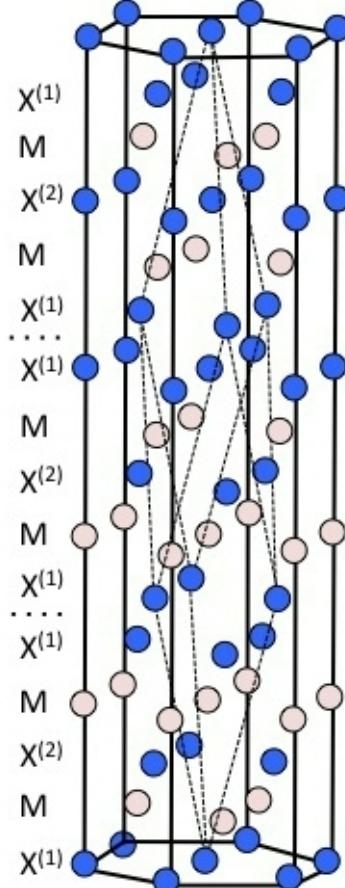
Example: spiral growth steps at screw dislocation in Bi_2Te_3 thin film



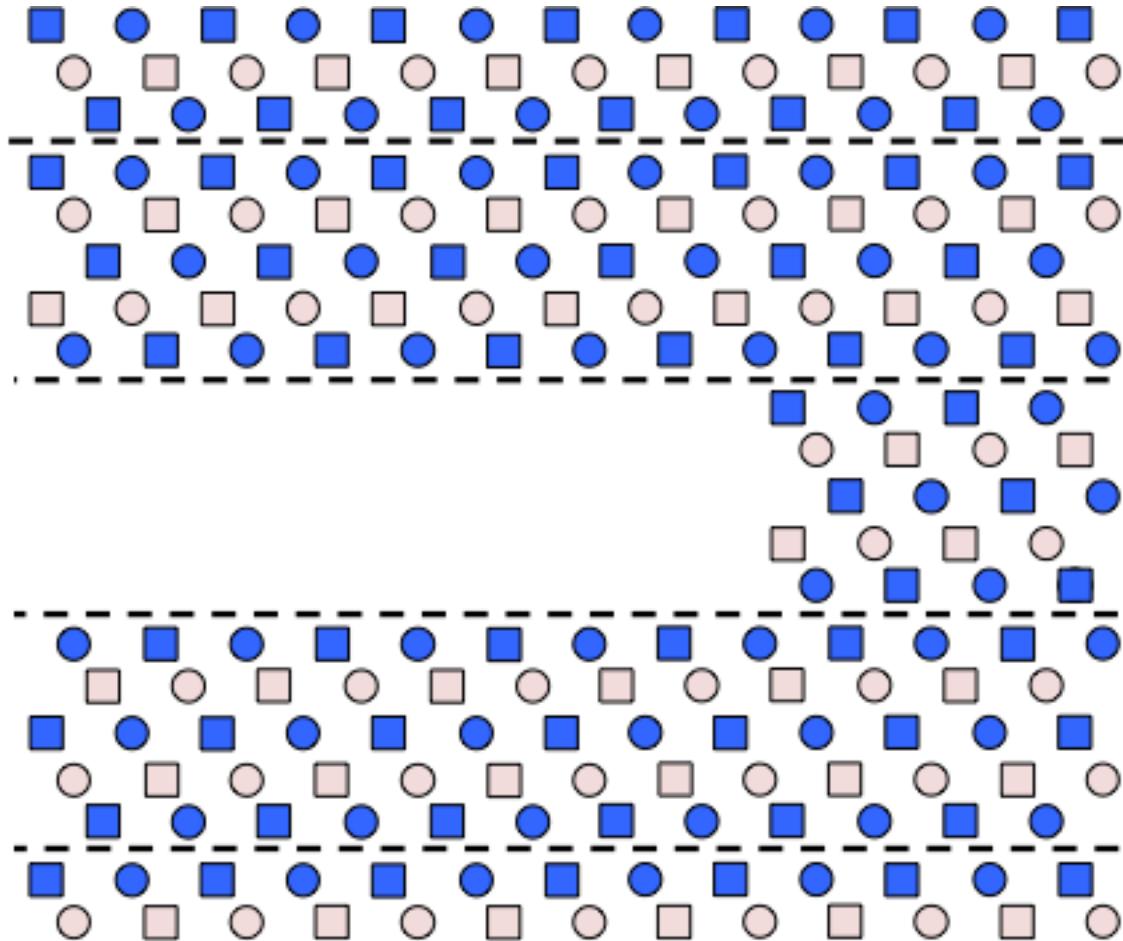
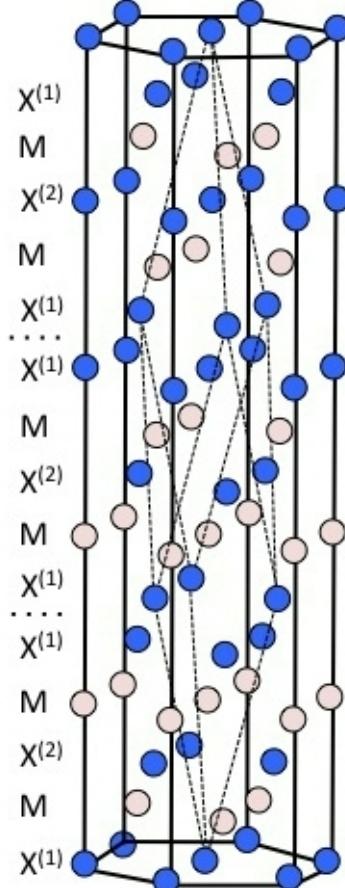
What about **edge** dislocations with non-basal Burgers vectors?

M. Ferhat, J.C. Tedenac, J. Nagao,
J. Crystal Growth (2000)

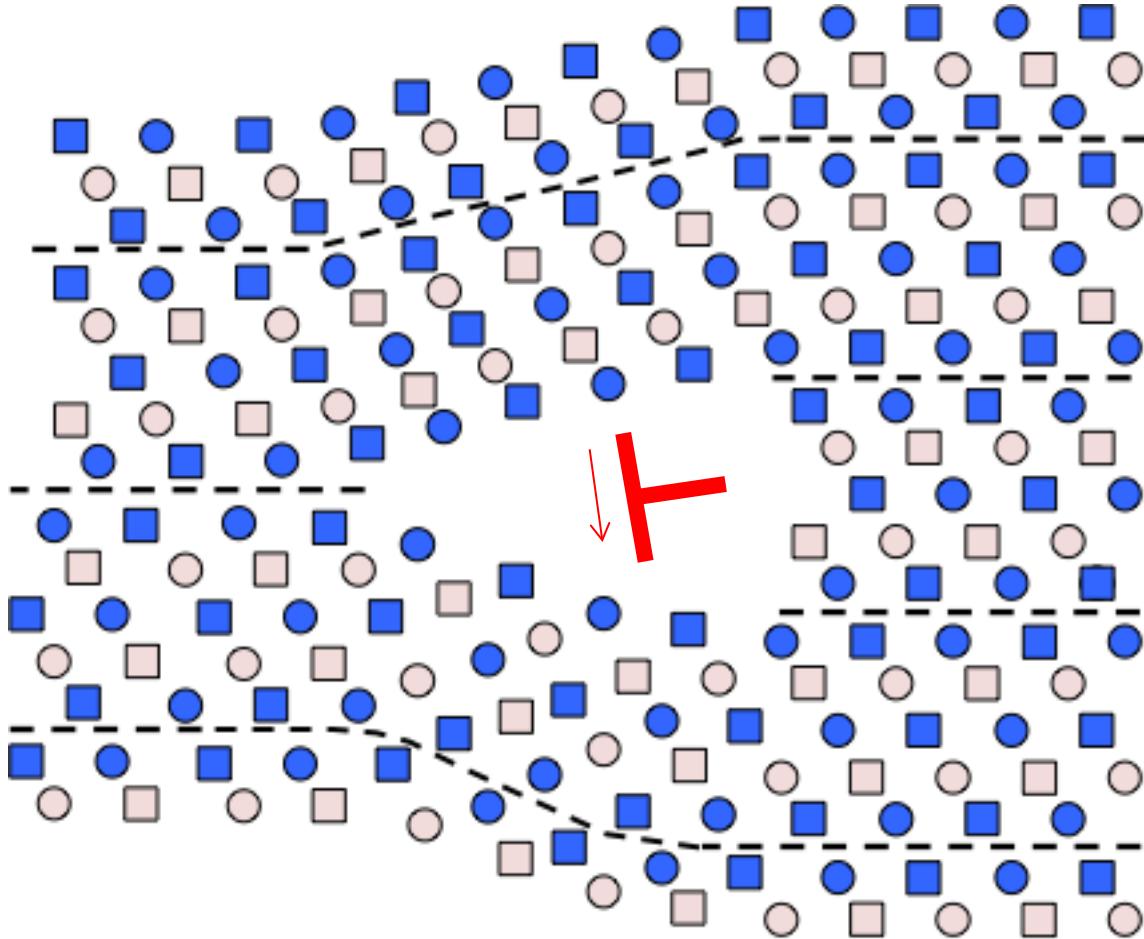
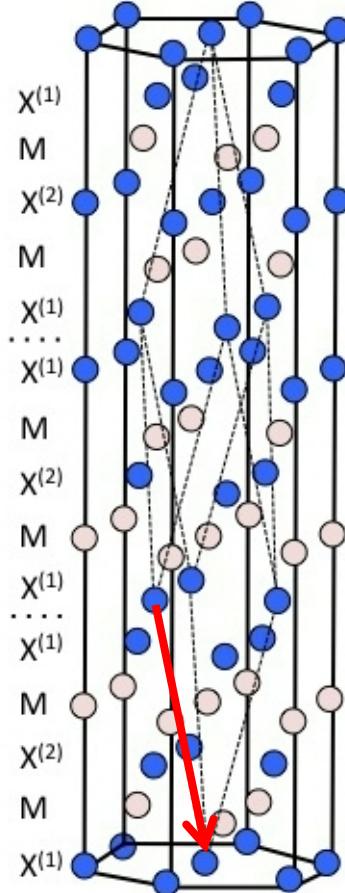
Non-basal edge dislocations: What happens if we pull out a quintuple unit?



Non-basal edge dislocations: What happens if we pull out a quintuple unit?

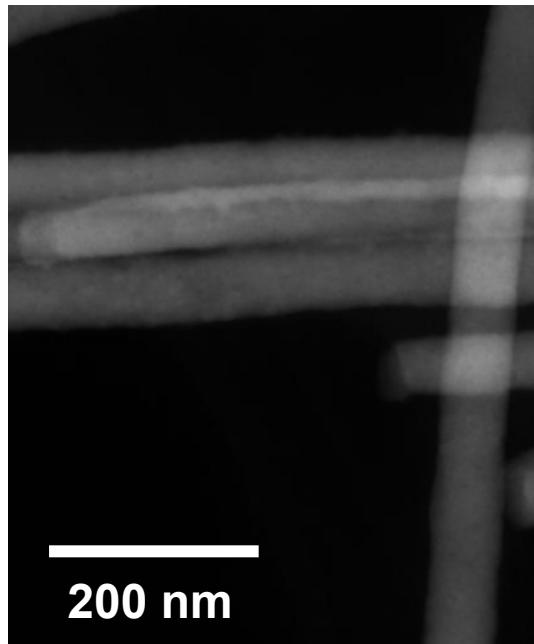


Non-basal edge dislocations: What happens if we pull out a quintuple unit?



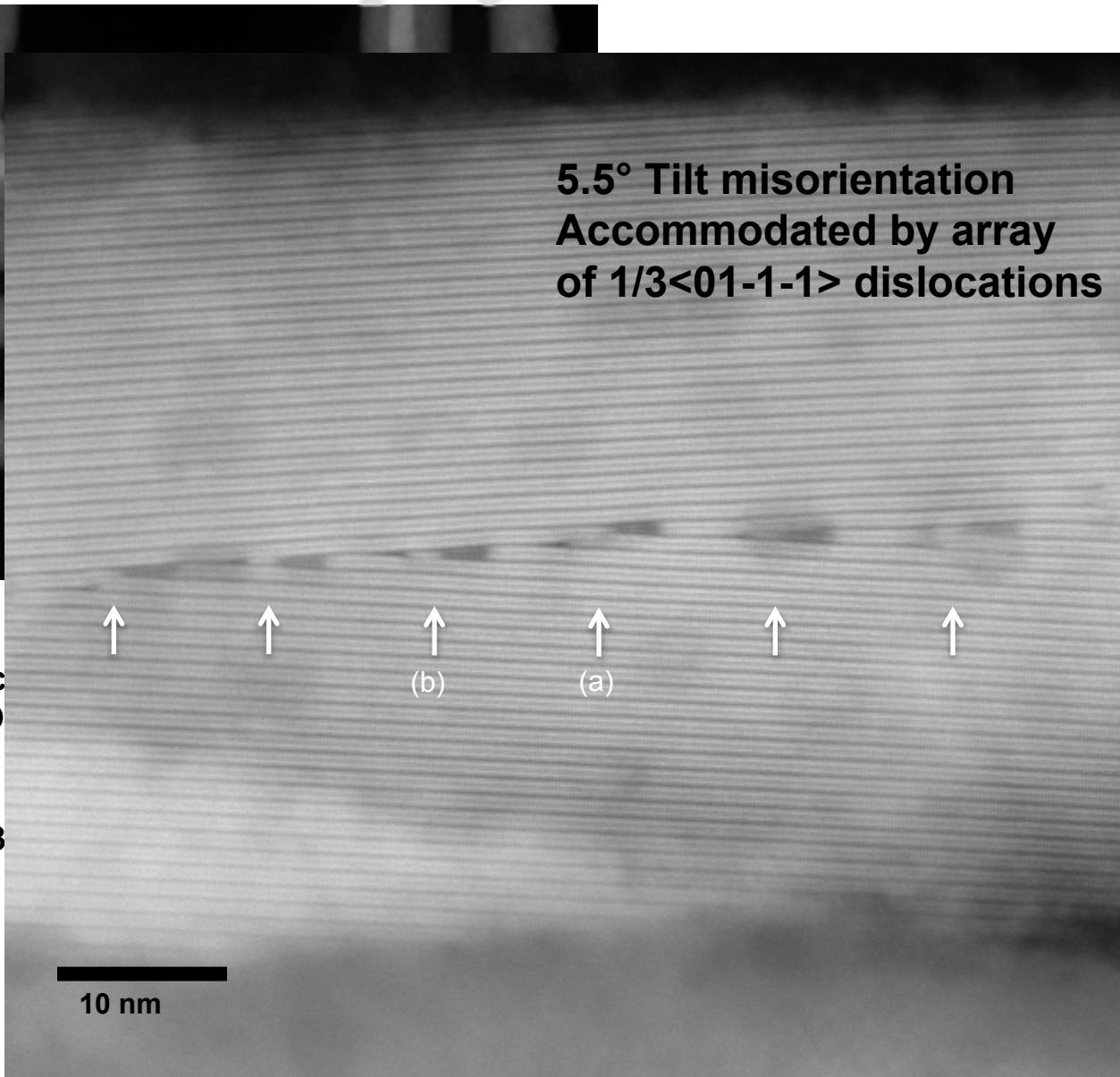
$$\mathbf{b} = (1/3)[01-1-1]_H = [00-1]_R$$

Dislocations in Bi_2Te_3 Nanowires

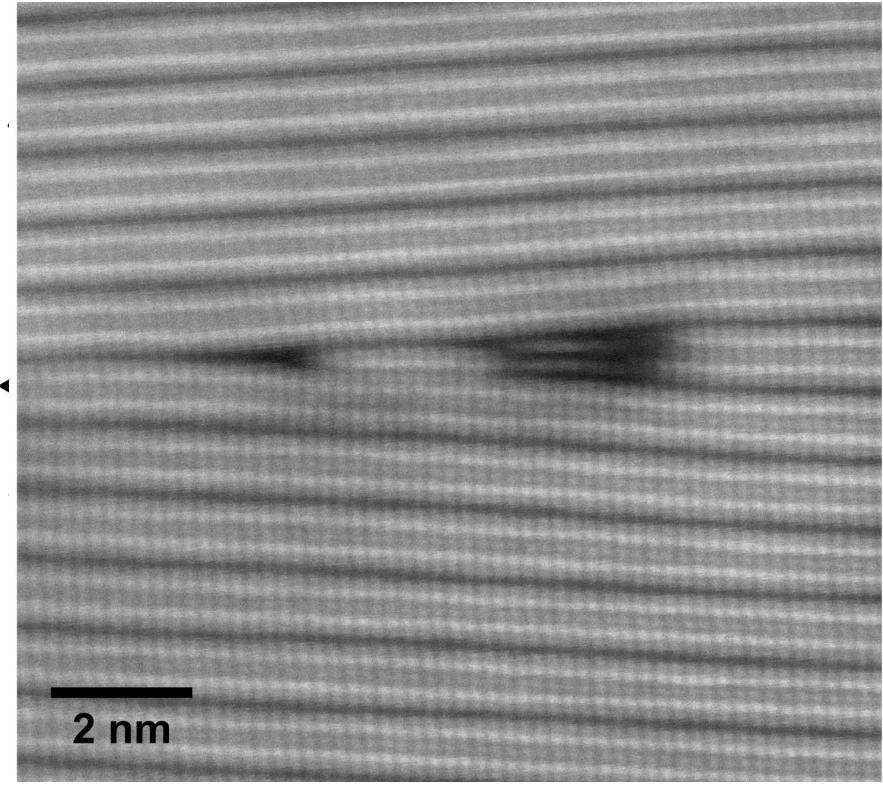
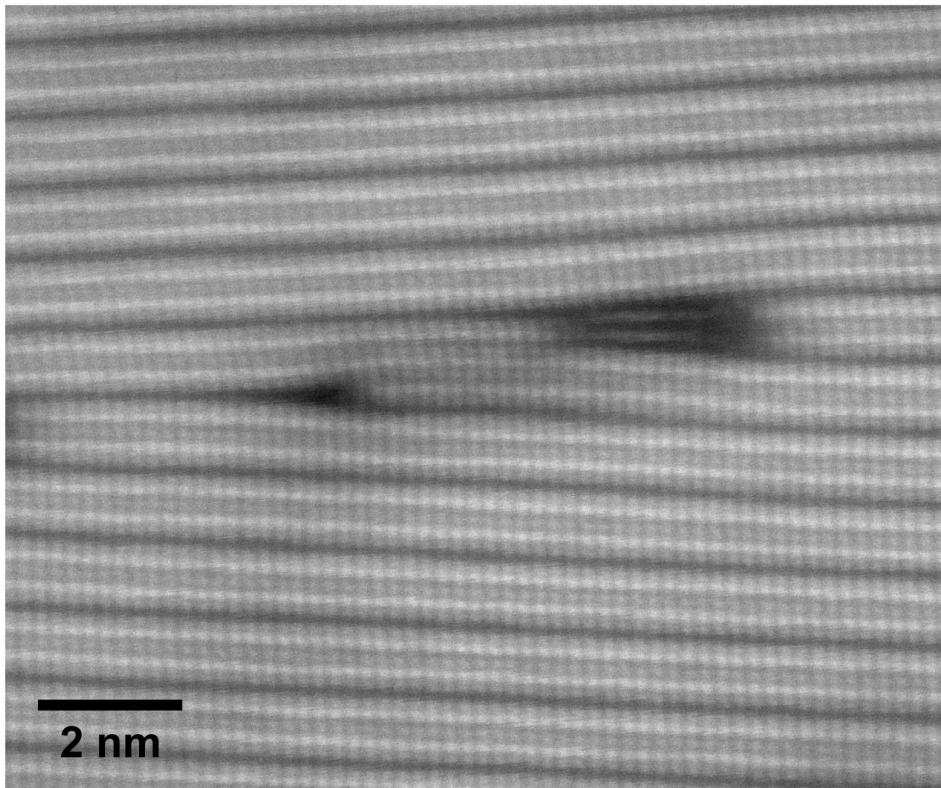


Wires formed by electrochemical deposition in nanoporous AAO templates.

Free standing wires annealed 3 minutes at 300°C in Ar-3%H₂.

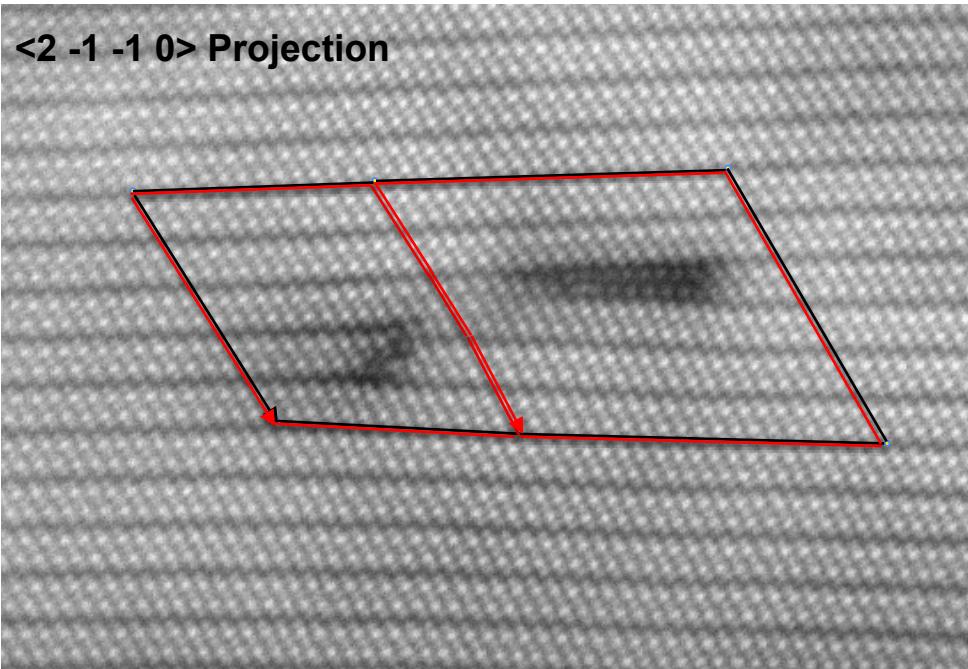


Dislocations have dissociated core: two configurations

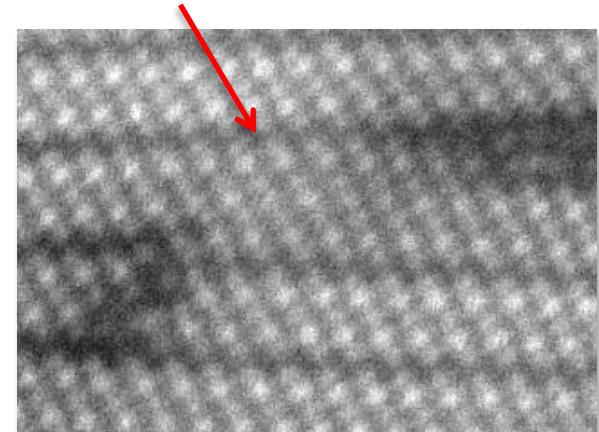


$<1\ 0\ -1\ 0>$ Projection

$(1/3)[0\ 1\ -1\ -1]$ Dislocation in Bi_2Te_3 :



Core structure:
 Bi_3Te_4 7-layer fault



$$\mathbf{b} = (1/3) [0\ 1\ -1\ -1] \rightarrow (1/15) [0\ 5\ -5\ -2] + (3/15) [0\ 0\ 0\ -1]$$

$$\mathbf{b}^2 > \mathbf{b}_1^2 + \mathbf{b}_2^2$$
$$109.8 \text{ \AA}^2 > 22.9 \text{ \AA}^2 + 37.2 \text{ \AA}^2 = 60.2 \text{ \AA}^2$$

Reduced strain energy with dissociation

7-Layer Bi_3Te_4 faults: Mechanism to accommodate Te loss during annealing

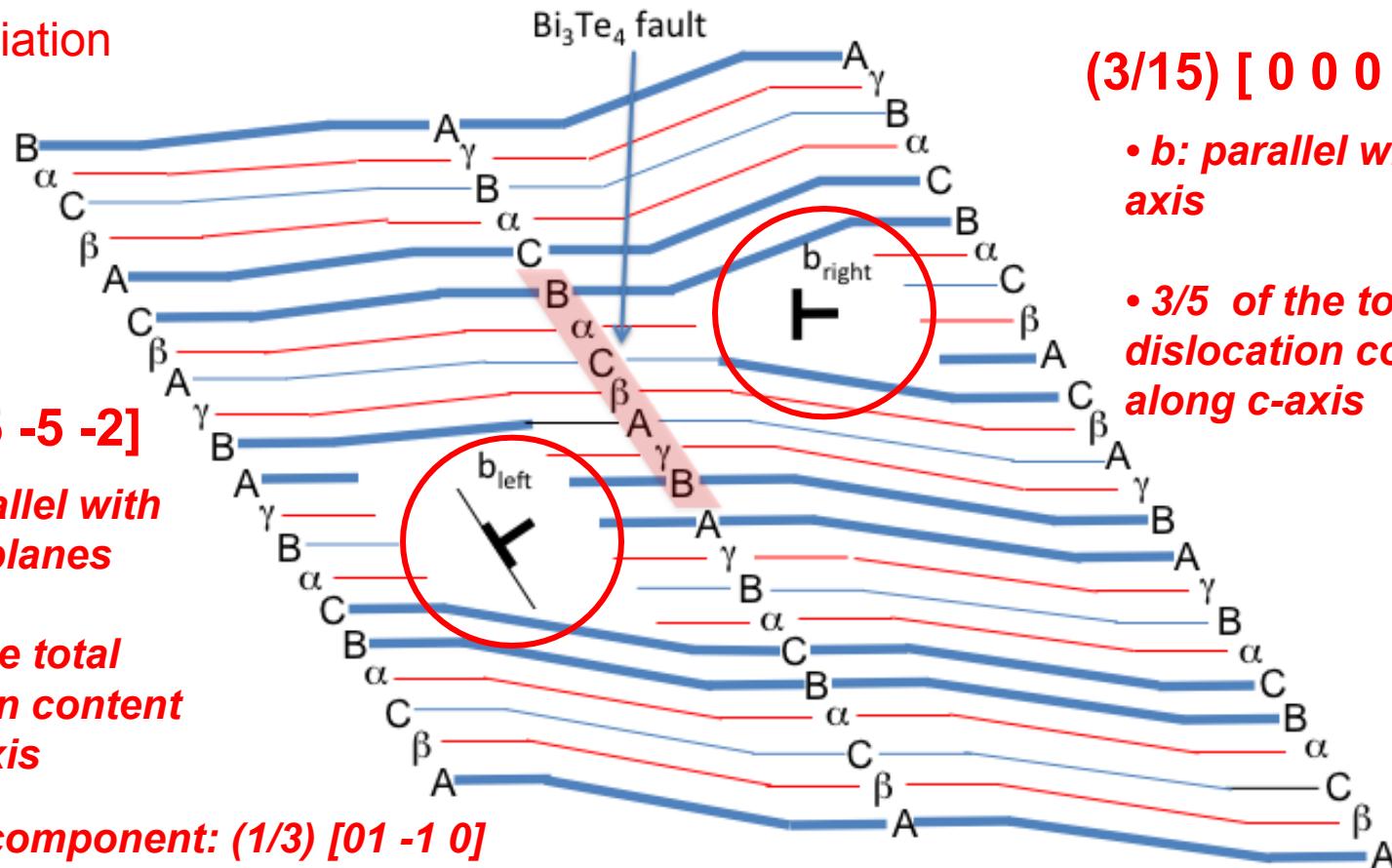
Climb dissociation

(1/15) [0 5 -5 -2]

- b is parallel with $(0\ 1\ -1\ 5)$ planes
- 2/5 of the total dislocation content along c -axis

Horizontal component: (1/3) [01 -1 0]

- analogous to Shockley partial.
- avoids fault in stacking resulting from additional 2 planes at Bi_3Te_4 fault



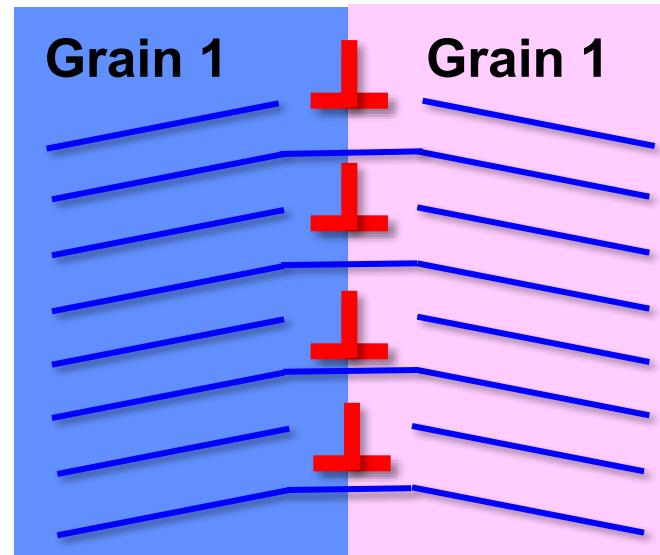
(3/15) [0 0 0 -1]

- b : parallel with c -axis
- 3/5 of the total dislocation content along c -axis

Dislocations at Grain Boundaries

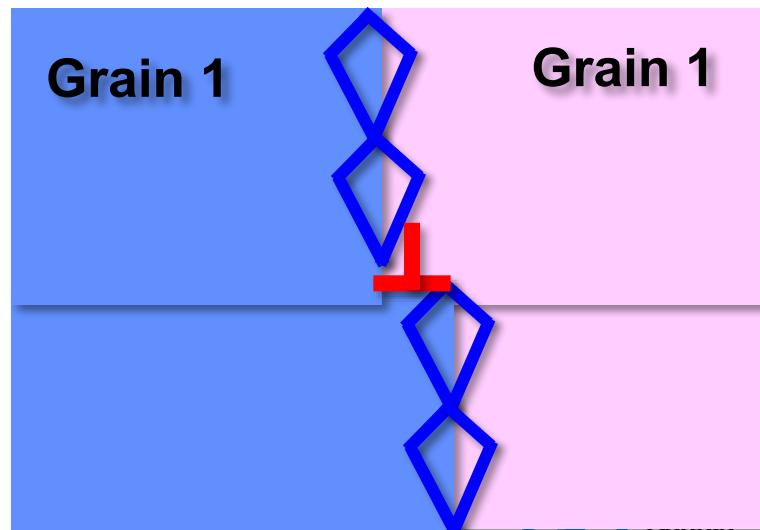
Low Angle Grain Boundaries

- Arrays of discrete lattice dislocations
- Reference state: single crystal



High Angle Grain Boundaries

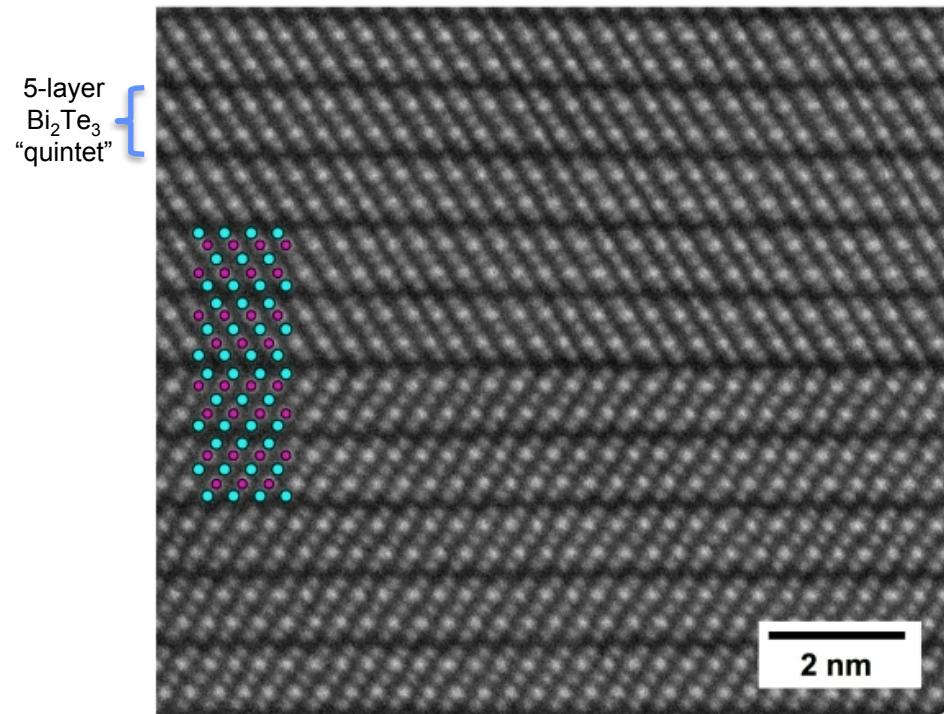
- Lattice dislocation cores too close to be meaningful.
- Reference state: singular, low-energy reference orientations.
- Burgers vector and step content are important.



Bi₂Te₃ Basal Twin: A singular, low-energy interface structure

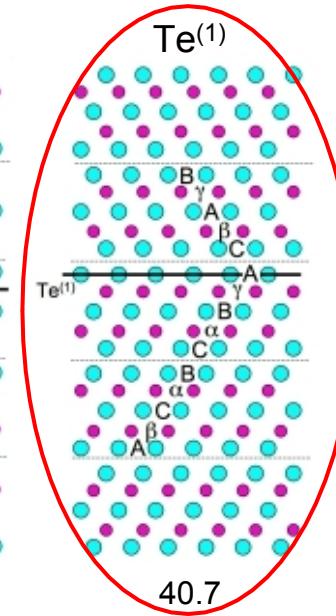
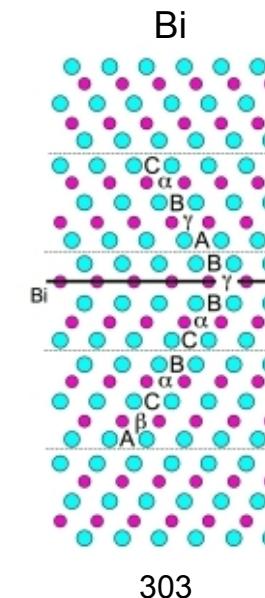
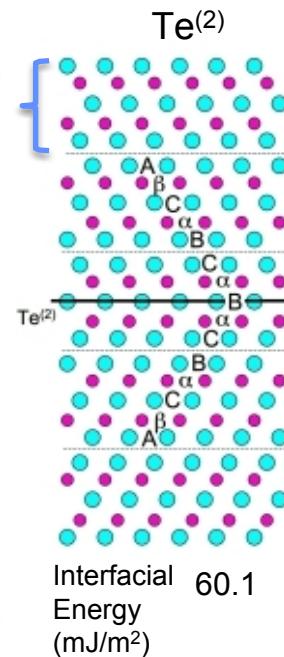
HAADF-STEM Imaging:

Twin Boundary Terminated at Te⁽¹⁾ layer



DFT Calculations:

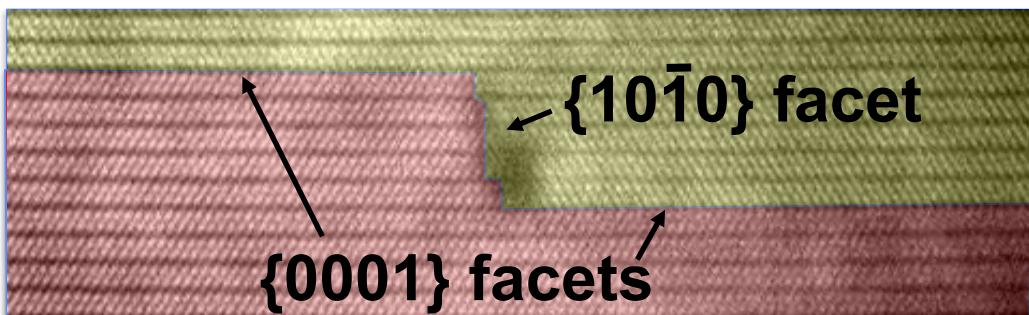
Three Possible Compositional Terminations



Energetic preference for termination at Te⁽¹⁾ sites

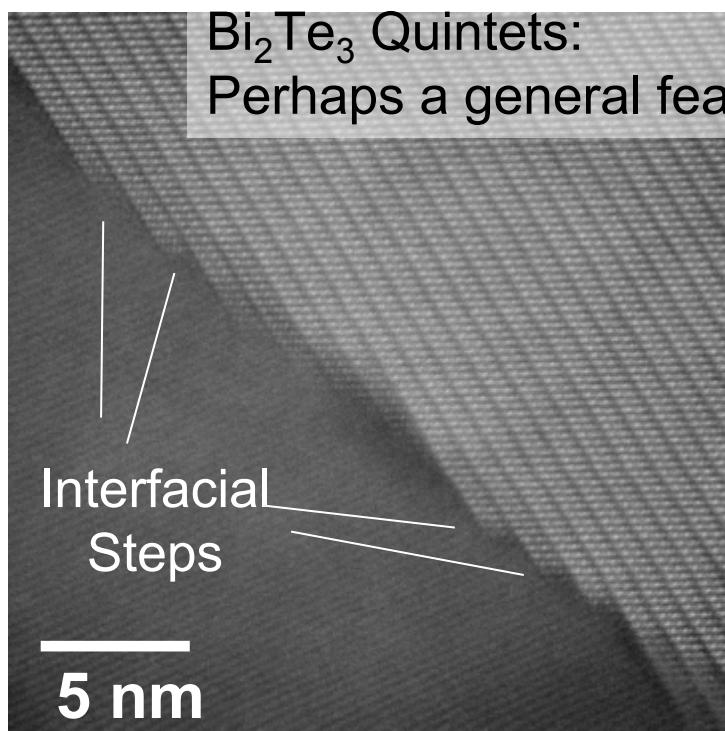
D.L. Medlin, Q.M. Ramasse, C. D. Spataru, N.C. Yang, J. Appl. Phys. (2010)

Steps in the Bi_2Te_3 Twin Boundary



Steps of integral 5-plane

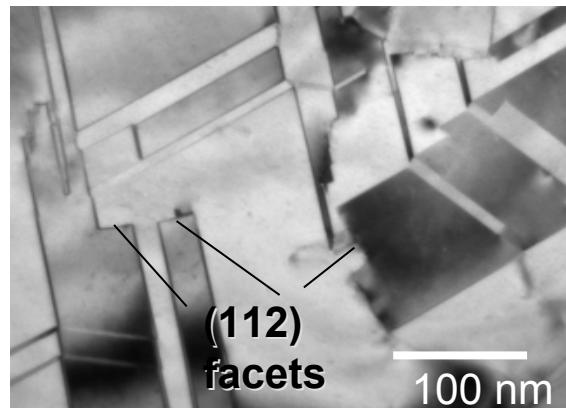
Bi_2Te_3 Quintets:
Perhaps a general feature?



NBT20/21mar11/18.15.52

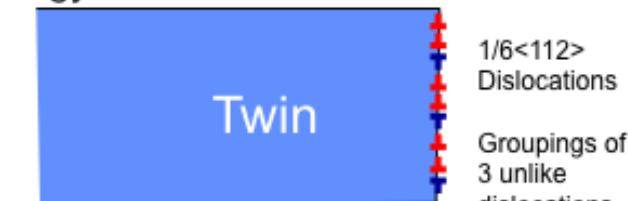
Morphology analogous to annealing and growth twins in FCC materials.

Example: Twins in Electrodeposited Ni

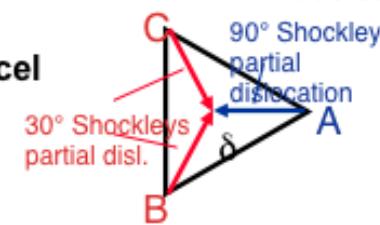


Lucadamo, Medlin, Talin, Yang, Kelly, Phil Mag 2005

Morphology Related to Dislocation Structure

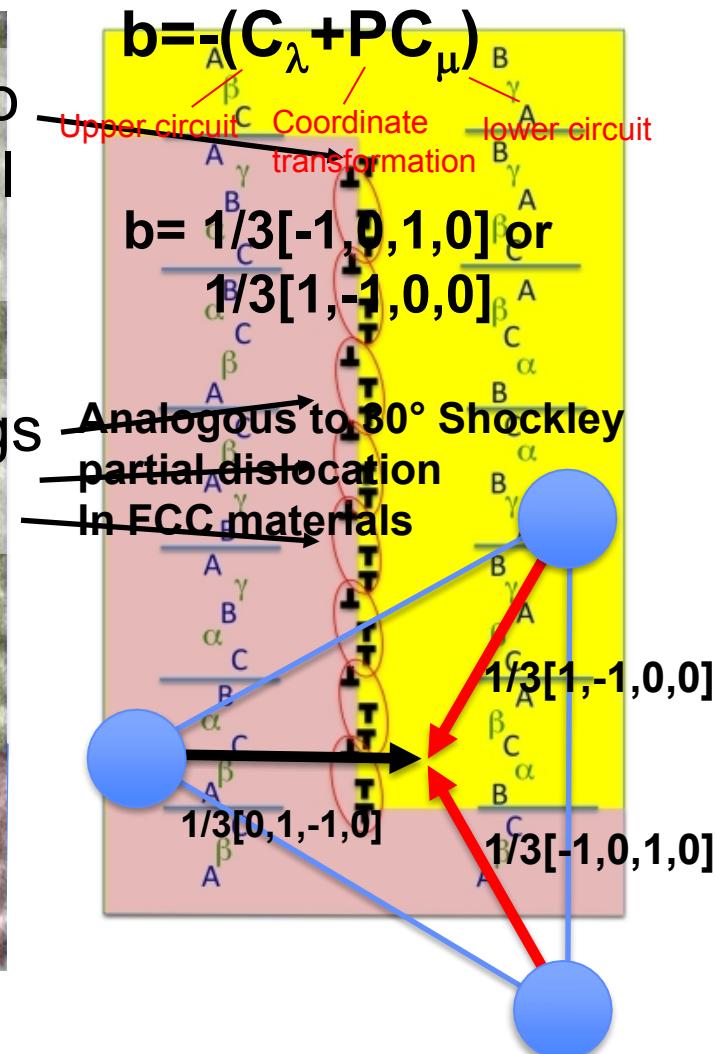
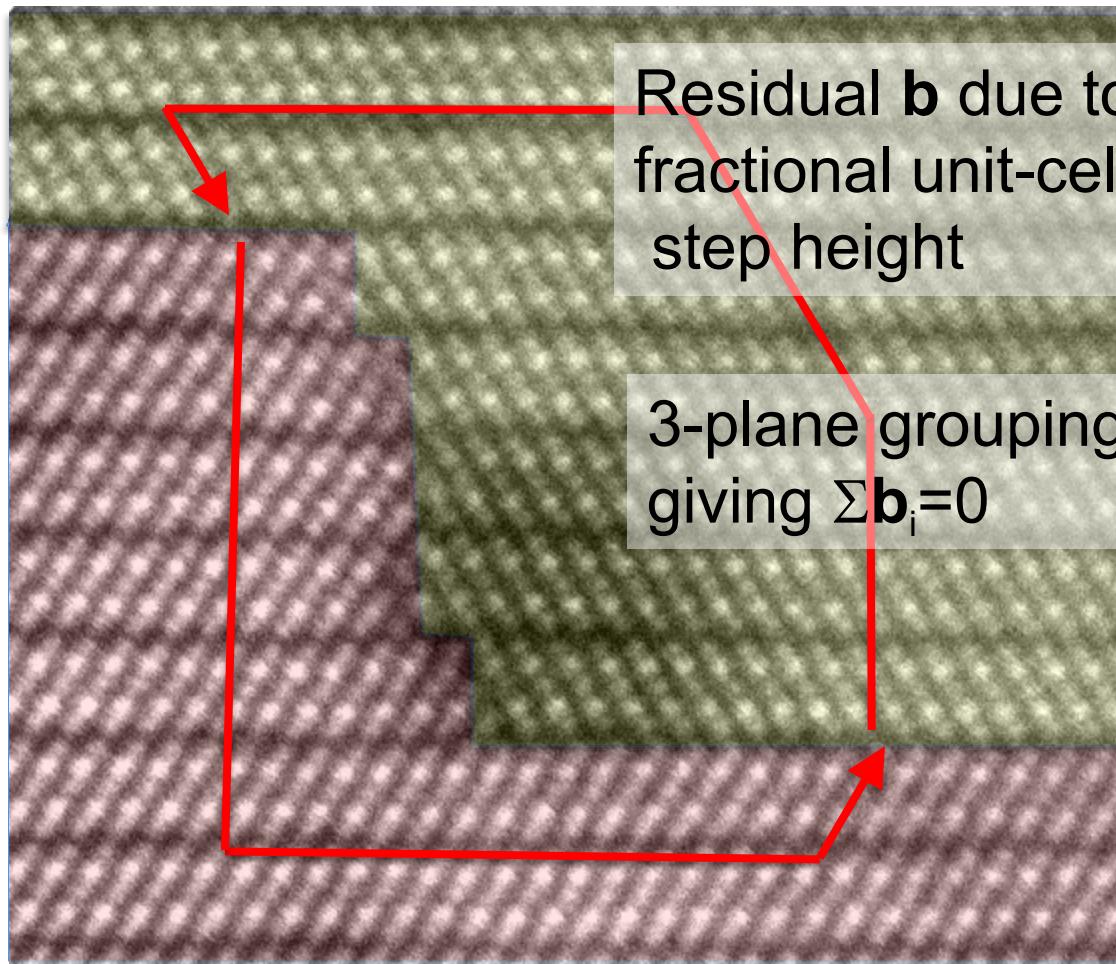


Burgers vectors cancel
eliminating long-range
strain.

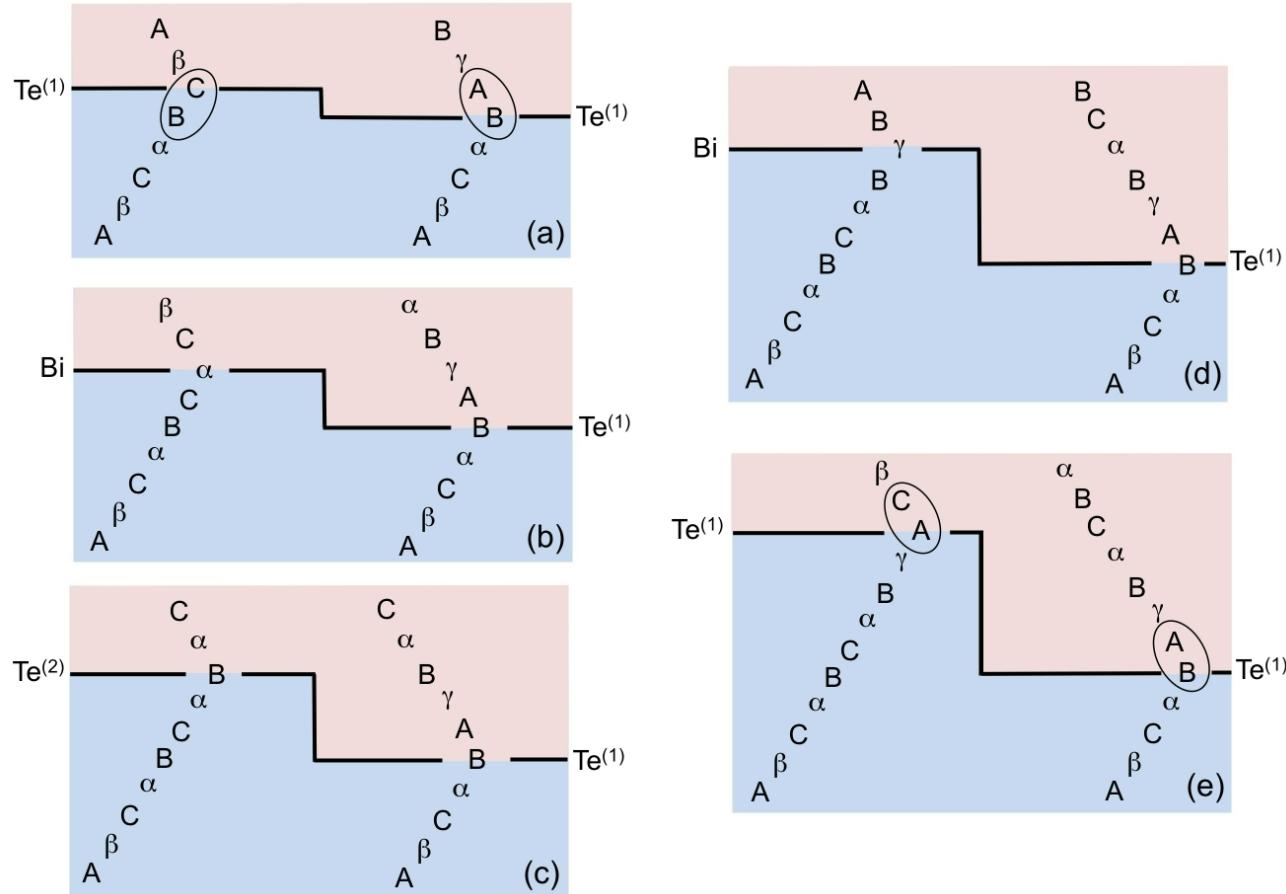


India
National
Laboratories

What is the dislocation content of the step?



Consideration of other possible (0001) twin defects



Consideration of other possible (0001) twin defects

Step Height, h	Terminating Planes on Either Side of Step	Equivalency of Adjacent Twin Terraces	Burgers Vector, b
1	$\text{Te}^{(1)}/\text{Te}^{(1)}$	Equivalent (inverted)	$\frac{1}{3}\langle 10\bar{1}0 \rangle$
	$\text{Bi}/\text{Te}^{(1)}$	Nonequivalent	
	$\text{Te}^{(2)}/\text{Bi}$	Nonequivalent	
	$\text{Te}^{(1)}/\text{Bi}$	Nonequivalent	
2	$\text{Te}^{(2)}/\text{Te}^{(1)}$	Nonequivalent	$\frac{1}{3}\langle 10\bar{1}0 \rangle$
	Bi/Bi	Equivalent (inverted)	
	$\text{Te}^{(1)}/\text{Te}^{(2)}$	Nonequivalent	
	Bi/Bi	Equivalent (inverted)	0
3	$\text{Bi}/\text{Te}^{(1)}$	Nonequivalent	
	$\text{Te}^{(1)}/\text{Bi}$	Nonequivalent	
	$\text{Bi}/\text{Te}^{(2)}$	Nonequivalent	
	$\text{Te}^{(1)}/\text{Te}^{(1)}$	Equivalent (inverted)	
4	Same	Equivalent	$\frac{1}{3}\langle 10\bar{1}0 \rangle$
10	Same	Equivalent	$\frac{1}{3}\langle 10\bar{1}0 \rangle$
15	Same	Equivalent	0

Low Energy: $\text{Te}^{(1)}/\text{Te}^{(1)}$ termination.

Intermediate energy:

Termination at either Te site: $\text{Te}^{(1)}$ and $\text{Te}^{(2)}$

Conclusions

Key structural aspects of Bi_2Te_3 and related tetradymite-type chalcogenides are manifested in the detailed structures of extended defects in these materials.

Weak, van der Waals bonding across double chalcogenide layers

Ability to accommodate non-stoichiometry through altering the layer stacking

Close inter-relationship between the rocksalt and tetradymite structural types.

Attention to the topological properties and detailed structure of extended defects in the chalcogenides is critical.

Understanding interfacial formation and stability and, ultimately, interfacial transport properties in thermoelectrics.

Set of elementary "building blocks" for a general picture of interfacial structure in chalcogenide thermoelectrics

Acknowledgements

Sandia/Livermore

N.Y.C. Yang, C.Spataru, K. Erickson, J.D. Sugar

Sandia/Albuquerque

M. Siegal, S. Limmer, W.G. Yelton

LBNL/NCEM

Q. Ramasse (present address, SuperSTEM Laboratory, Daresbury, UK)

California Institute of Technology

N.A. Heinz, T. Ikeda, G.J. Snyder

Special thanks to:

-LLNL: John Bradley, for use for LLNL's Titan 80/300 instrument

-LBNL: User program, National Center for Electron Microscopy

-UCD: Z. Zhang and E. Lavernia, for assistance with bulk Bi₂Te₃ processing

Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.