



# Copper Electrodeposition in Mesoscale Through-Silicon-Vias

## Unique 3D Integration Applications for SOI Substrates



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# Presentation Outline

## Unique Application

- Ion Trap Basics
- Wire Bond Issues
- TSVs in Ion Traps
- Cu TSV ECD Electrical Connection & Isolation

## Conformal Copper TSV Filling

- Wetting Issues
- ECD Basics
- Fill Profile Requirements
- Chemical Analysis
- CEAC Mechanism Scalability

## Bottom-Up Filling Technique

- Fundamentals
- Parameter Modification Optimization
- Ion Trap TSV Application
- Plating Approach
- Transition to MSA



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# RF (or Paul) Ion Trap Basics

- A potential well is formed via an oscillating RF electromagnetic field to control ion(s)
- The traps discussed here are operated under cryogenic conditions (3-4 K and ultra-high-vacuum)
- Optical access is required, for cooling lasers to drop the ions into their respective ground states

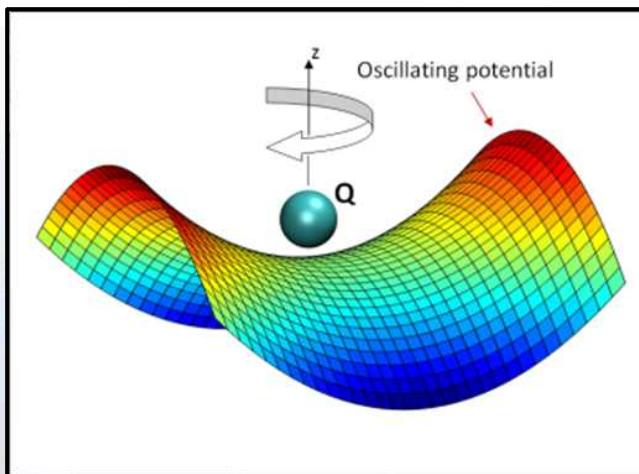


Image Retrieved From: <http://www.kurzweilai.net/nanopores-may-lead-to-fast-low-cost-dna-sequencing-device>

## Doppler Effect

Blue Shift

Red Shift

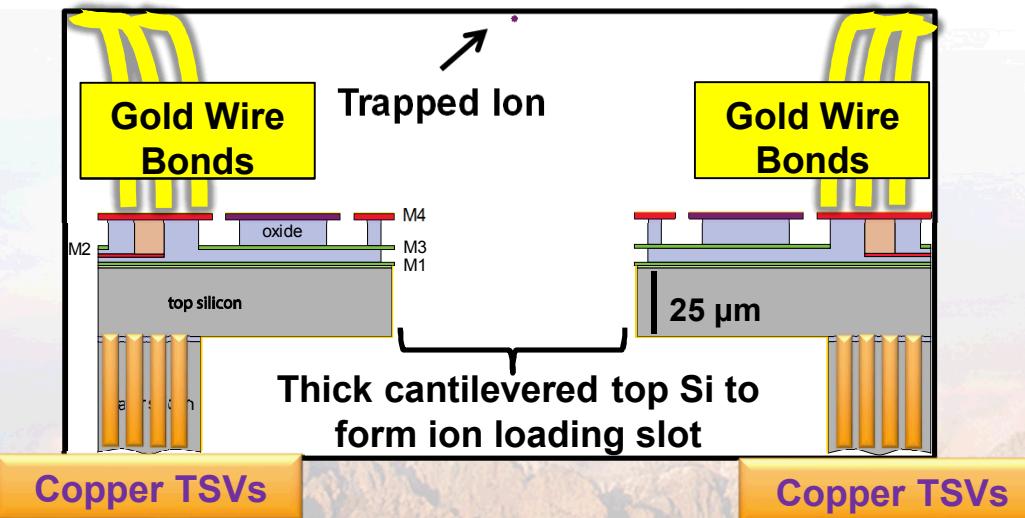


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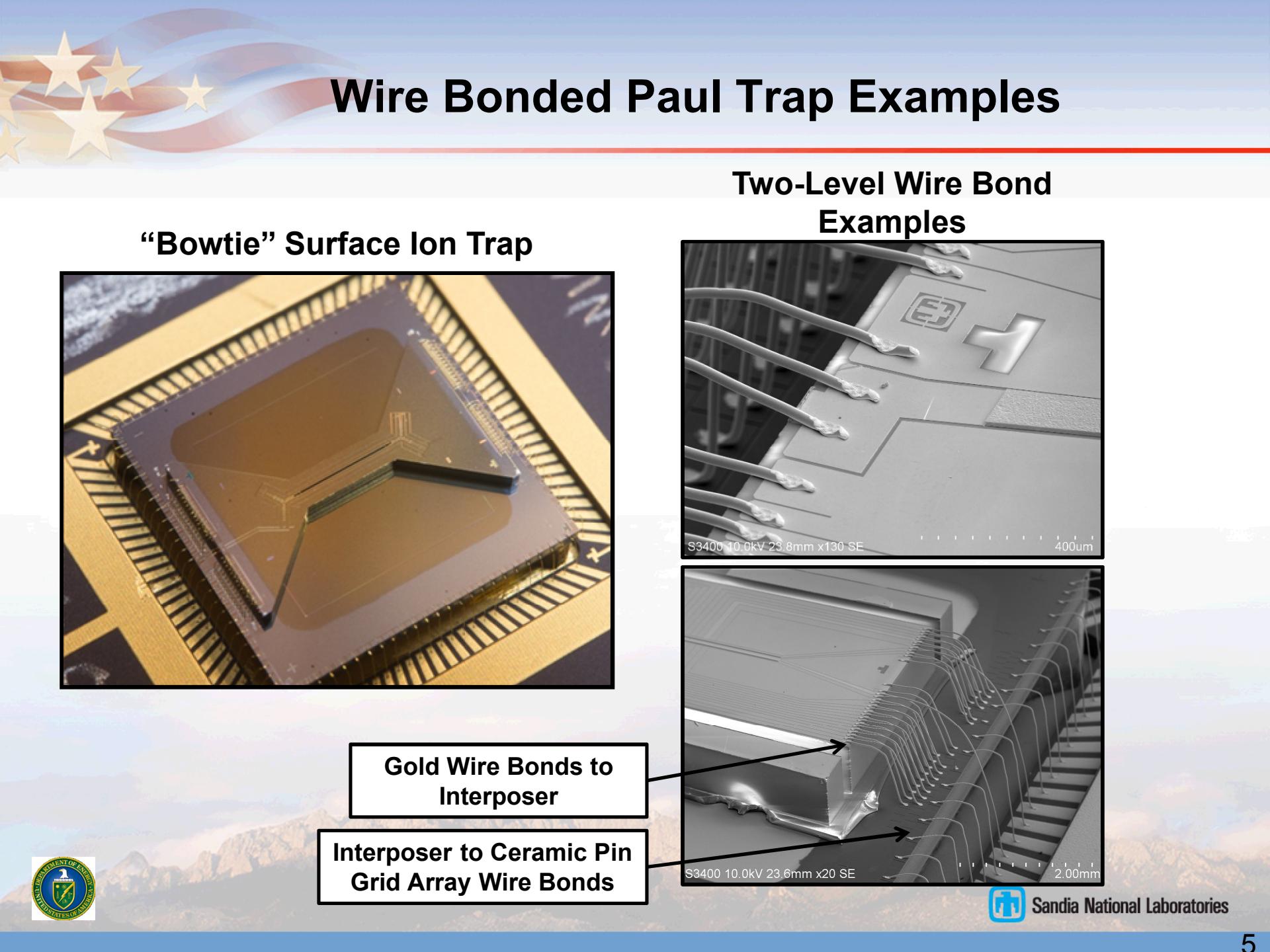


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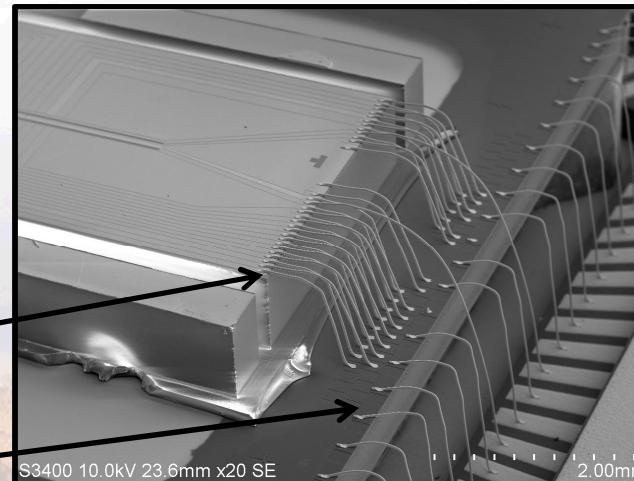
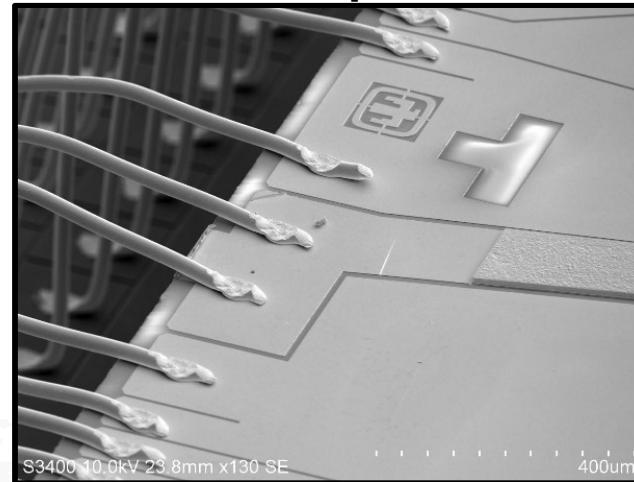
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Gold Wire Bonds to  
Interposer

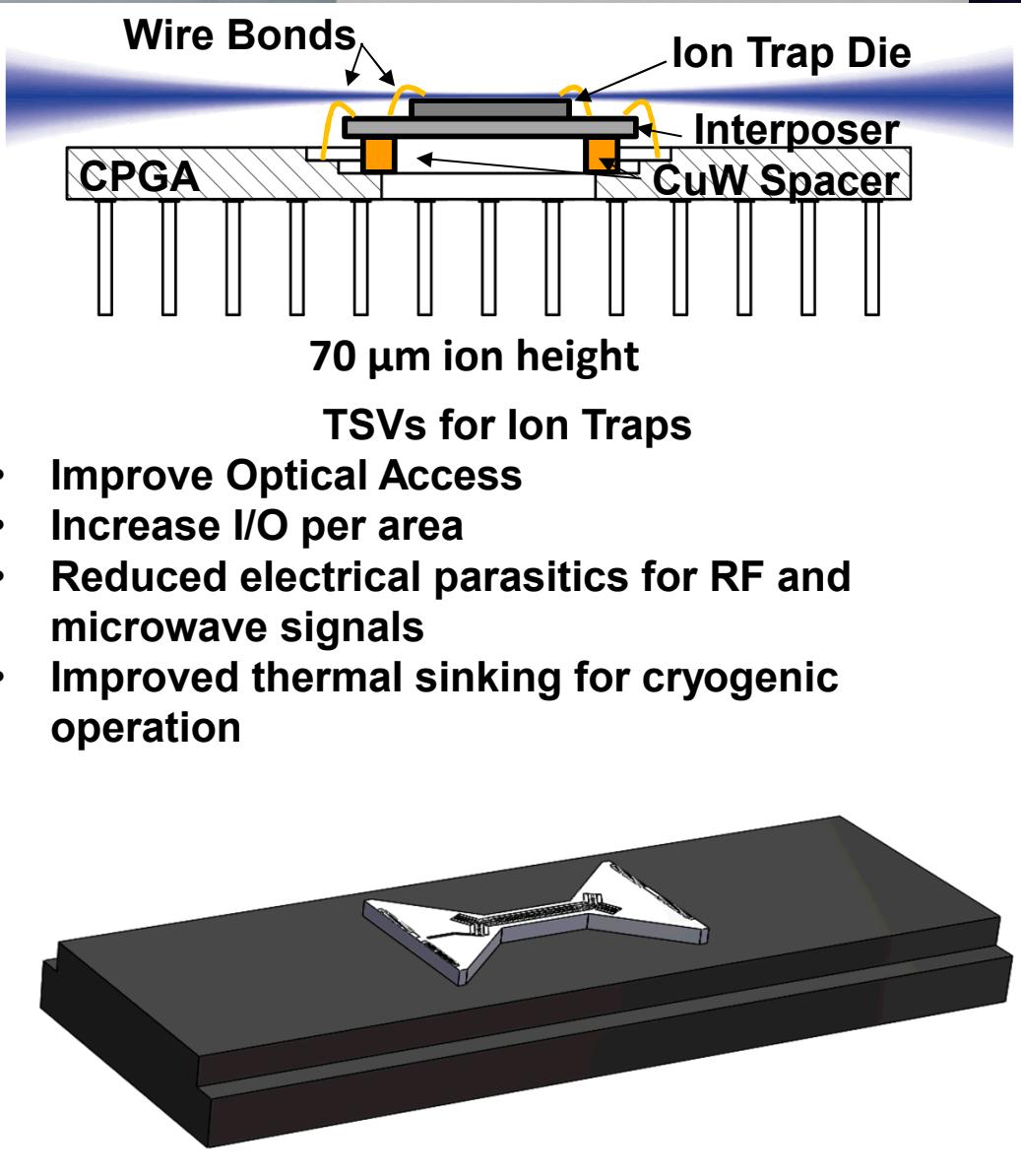
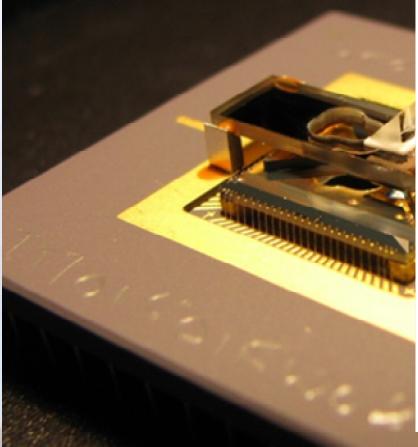
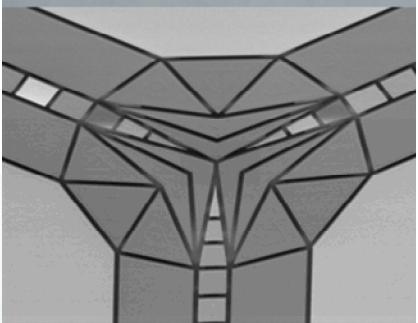
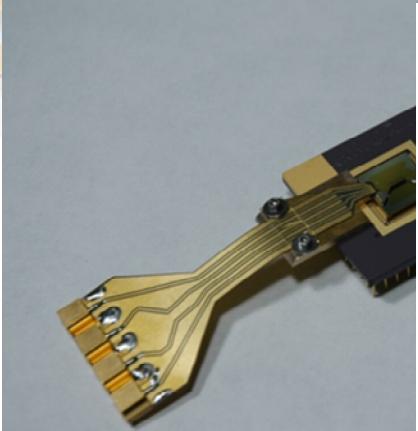
Interposer to Ceramic Pin  
Grid Array Wire Bonds

Two-Level Wire Bond  
Examples



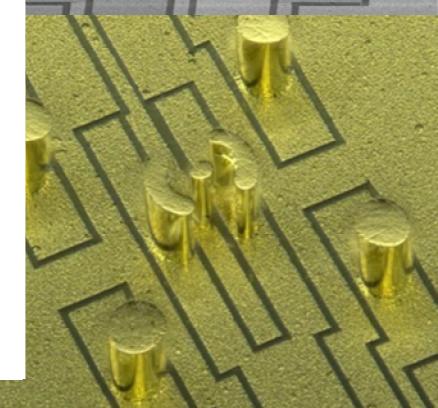
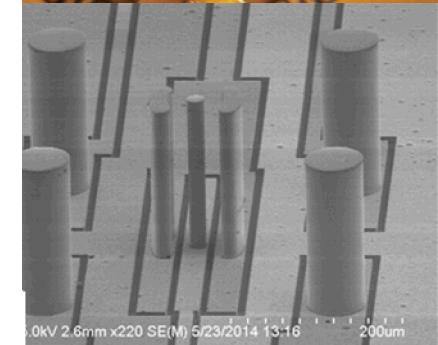
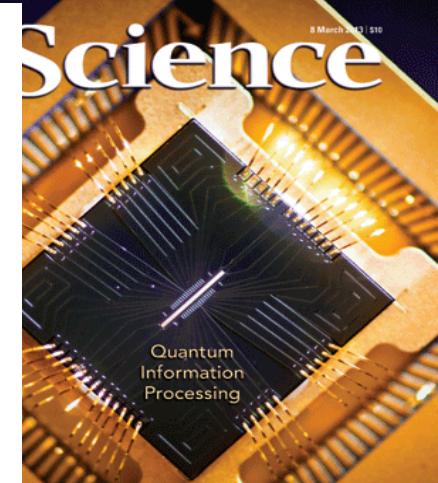
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# Ion Traps as a Quantum Information Platform

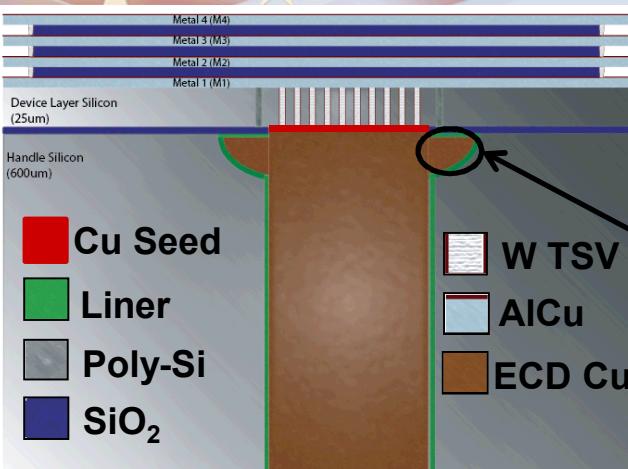


## TSVs for Ion Traps

- Improve Optical Access
- Increase I/O per area
- Reduced electrical parasitics for RF and microwave signals
- Improved thermal sinking for cryogenic operation



# Hybrid W Via First, Cu Via Last Approach for SOI Substrates



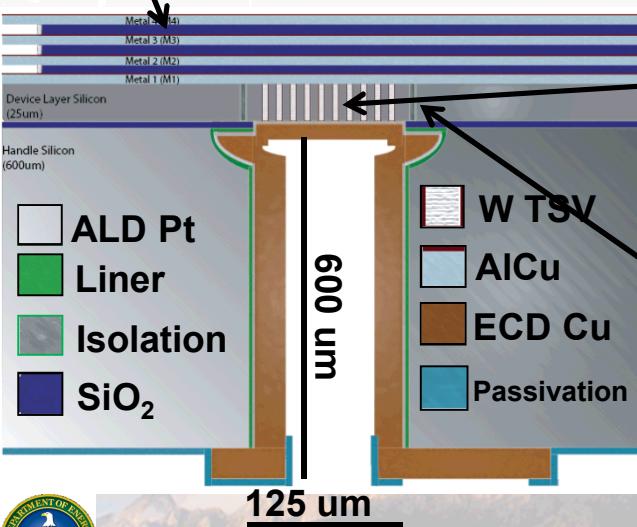
## 1st Approach: Through mask via fill: plating from evaporated seed metal only at the bottom of the via

- Cu seed metal directional evaporation to contact W implanted vias
- Plating is initiated from Cu seed

**A void free fill of the notch will be difficult to achieve**

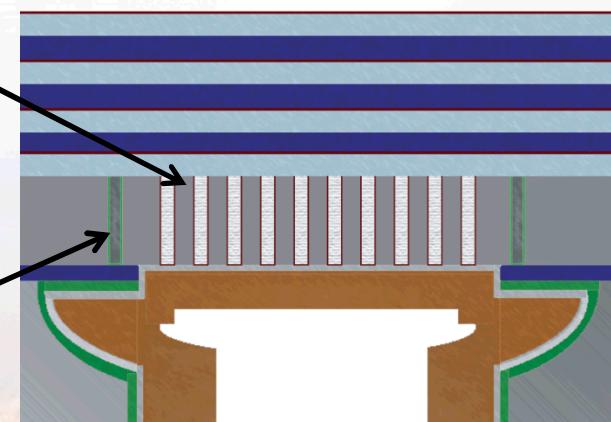
## 2nd Approach: Conformal via fill: plating from a ALD Pt film, filling the via from the outside in.

### Multi-layer routing



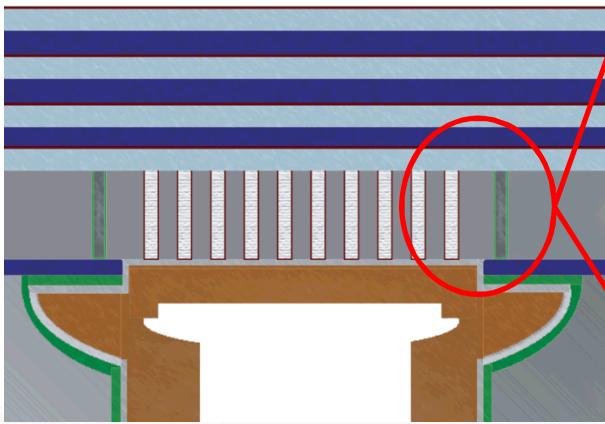
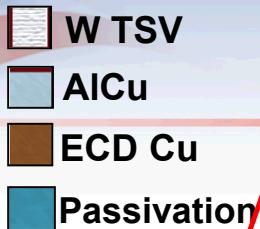
**W vias to connect Cu TSV through BOX and top Si layer**

**Isolation trench to electrically isolate the vias from the device Si layer**

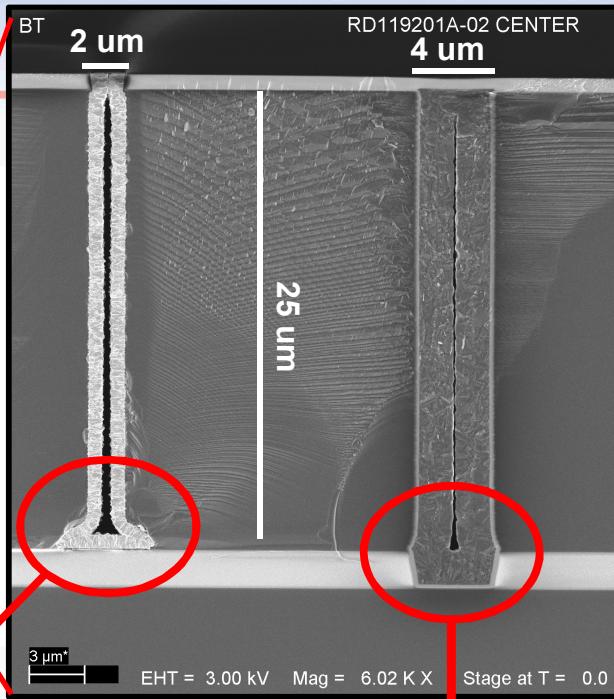


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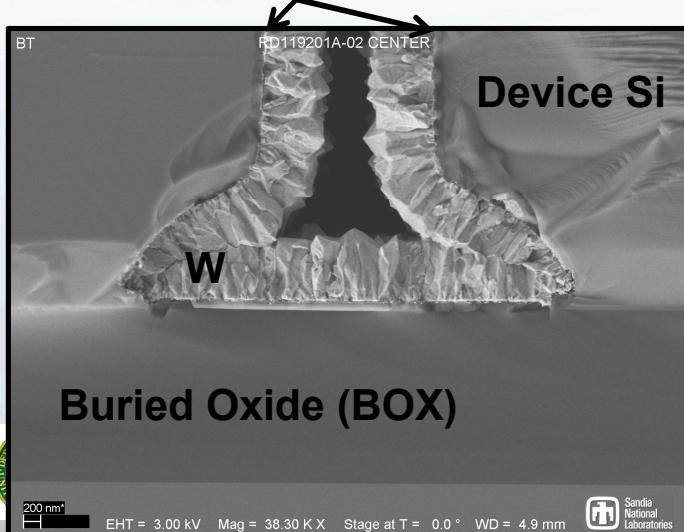
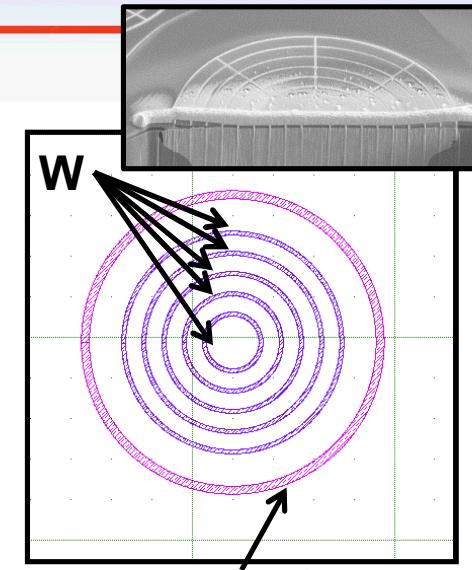
# Tungsten Vias and Electrical Isolation



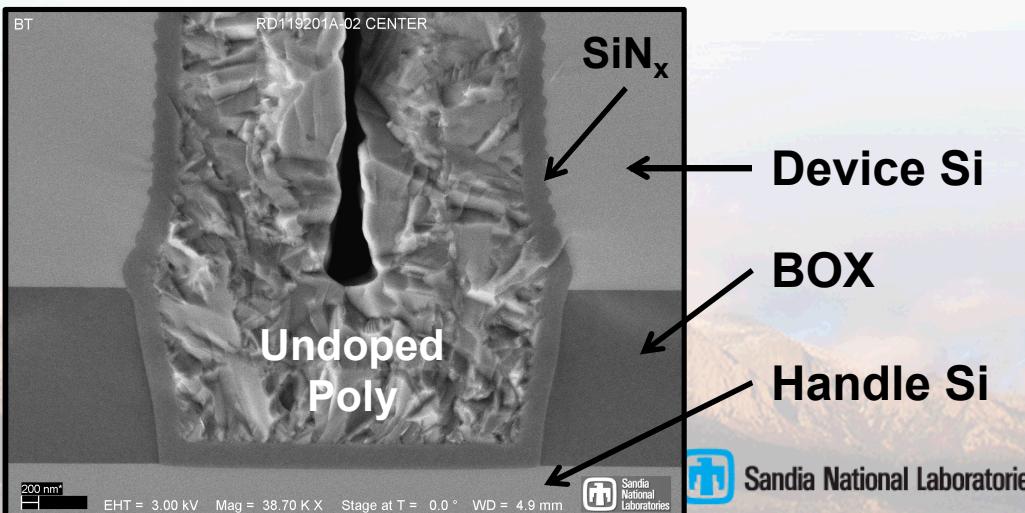
Degenerately doped Poly Si



Top Si Layout



Buried Oxide (BOX)



EHT = 3.00 kV Mag = 38.30 K X Stage at T = 0.0 ° WD = 4.9 mm



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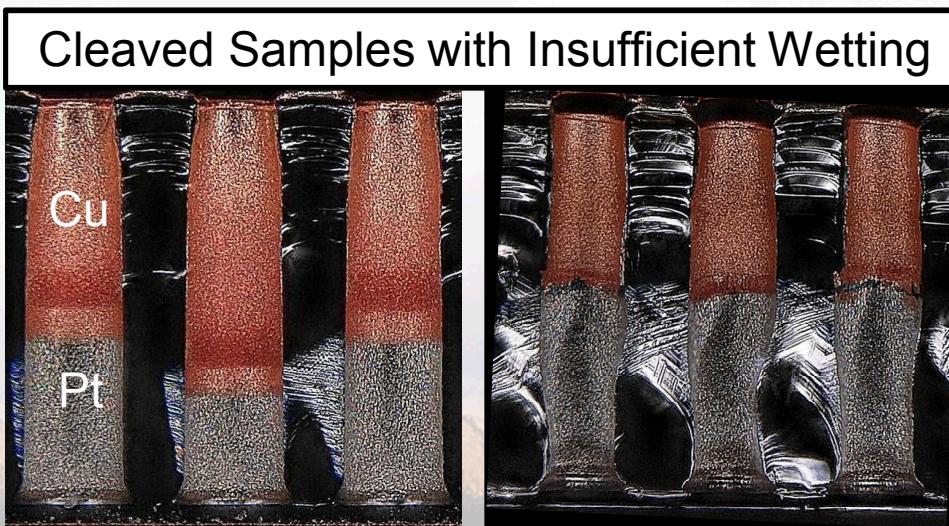


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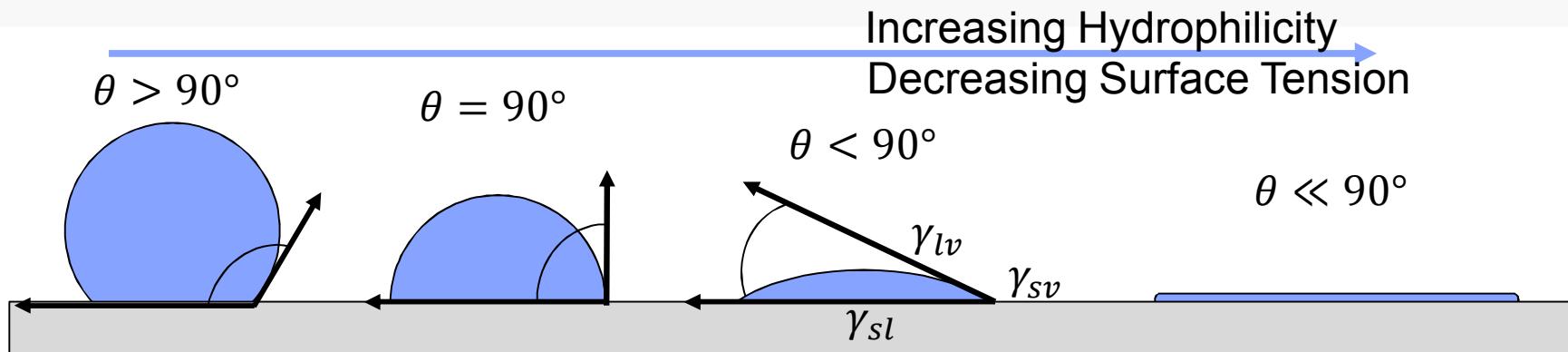
# TSV Wetting Issue

- ALD Pt should also be completely hydrophilic, but carbonaceous contamination can yield a hydrophobic surface<sup>2,3</sup>
- Potential solutions for achieving full via wetting:
  - a) Improve substrate hydrophilicity through surface preparation (or cleaning) techniques
  - b) Pre-wet the vias with some fluid under vacuum or and/or positive pressure and submerge in chemistry



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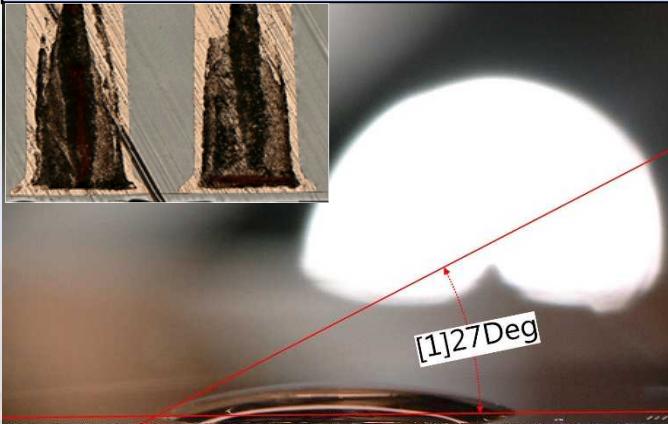
# Platinum Substrate Wetting<sup>2,3</sup>



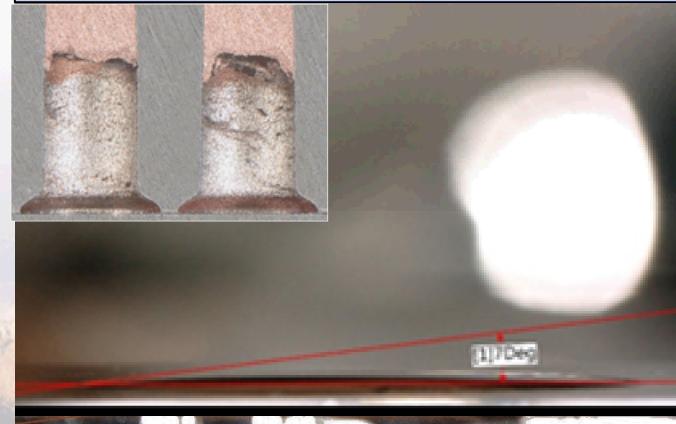
$$\gamma_{lv} \cos \theta = \gamma_{sv} - \gamma_{sl}$$

$\gamma_{lv}$ ,  $\gamma_{sv}$ ,  $\gamma_{sl}$  = liquid-vapor, solid-vapor, solid-liquid interfacial tensions

Room Temp NH<sub>4</sub>OH Clean



O<sub>2</sub> Plasma Clean



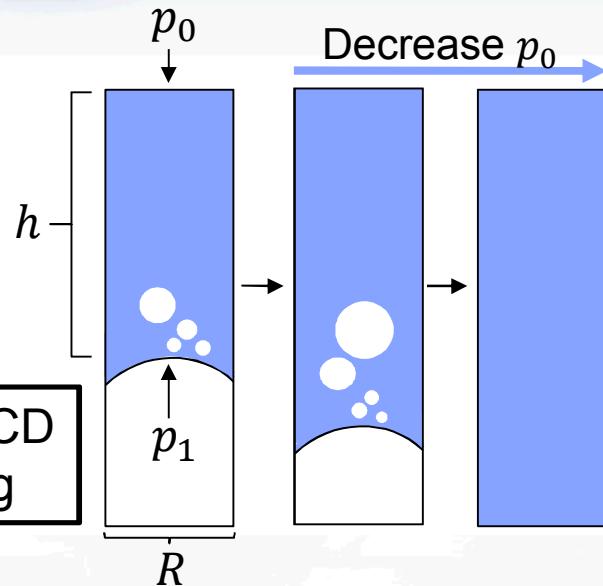
<sup>2</sup>Zhang, Junhong, et al. "Wetting process of copper filling in through silicon vias." *Applied Surface Science* 359 (2015): 736-741.

<sup>3</sup>Yuan, Yuehua, and T. Randall Lee. "Contact angle and wetting properties." *Surface science techniques*. Springer Berlin Heidelberg, 2013. 3-34.



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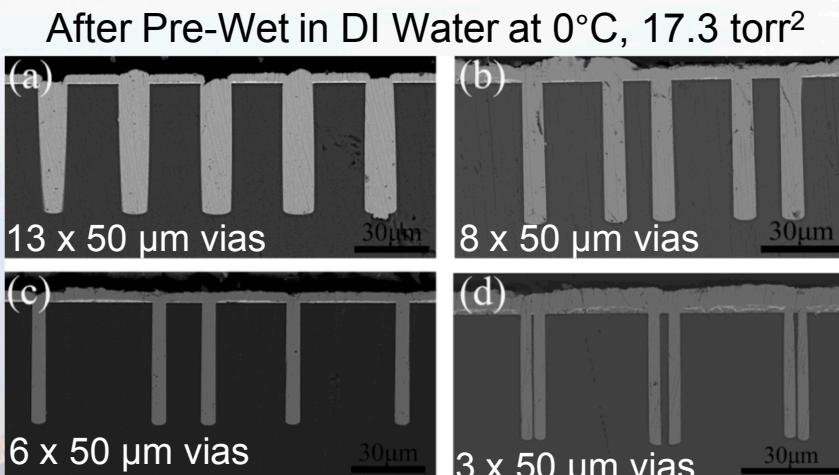
# Further Research into TSV Wetting<sup>2</sup>



$$P_1 = P_0 + \rho gh + \sigma \left[ \frac{1}{R} + \frac{1}{R} \right]$$

$\sigma$  = Surface Tension (Determined using contact angle measurement)

$So, P_1 > P_0 + \sigma \left[ \frac{1}{R} + \frac{1}{R} \right]$  for full wetting to occur  
(this can be accomplished by decreasing external pressure or fluid temperature)<sup>4</sup>



After Pre-Wet in EtOH  
at 22°C, 10<sup>-3</sup> torr



<sup>2</sup>Zhang, Junhong, et al. "Wetting process of copper filling in through silicon vias." *Applied Surface Science* 359 (2015): 736-741.

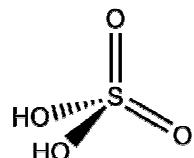


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# Copper Electrodeposition Basics

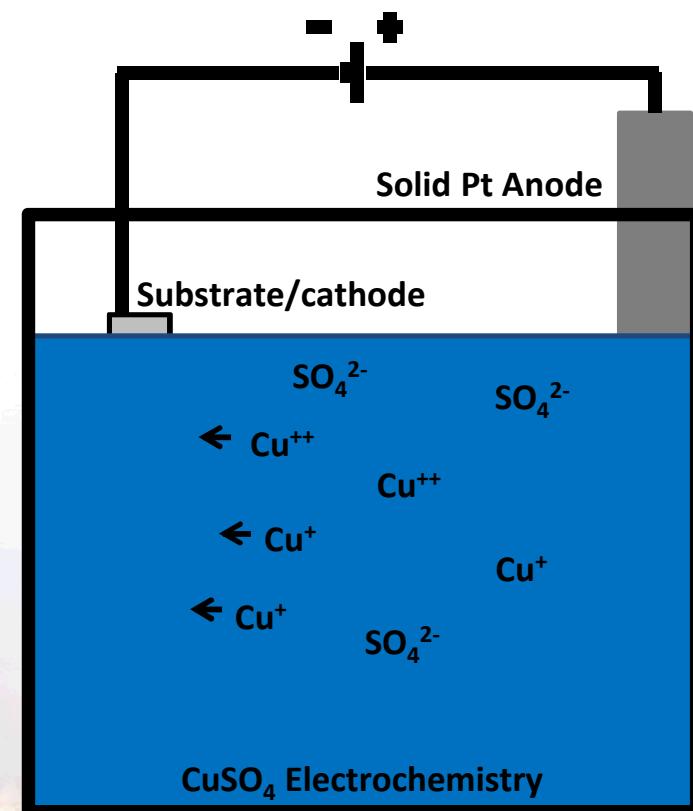
Additive-free Acid Sulfate Copper Electrolyte			
Constituent	Form	Concentration	Purpose
Copper Sulfate Pentahydrate	Solid (salt)	1-1.3 mol/L	Metal ion source
Sulfuric Acid Or Methane Sulfonic Acid (MSA)	Liquid	0.25-1 mol/L	Cu reduction & bath stability aid
Sodium Chloride Or Hydrochloric Acid	Liquid	50-100 mg/L	Complexing agent for additives

## Copper Sulfate Pentahydrate



## $\text{H}_2\text{SO}_4$ or MSA

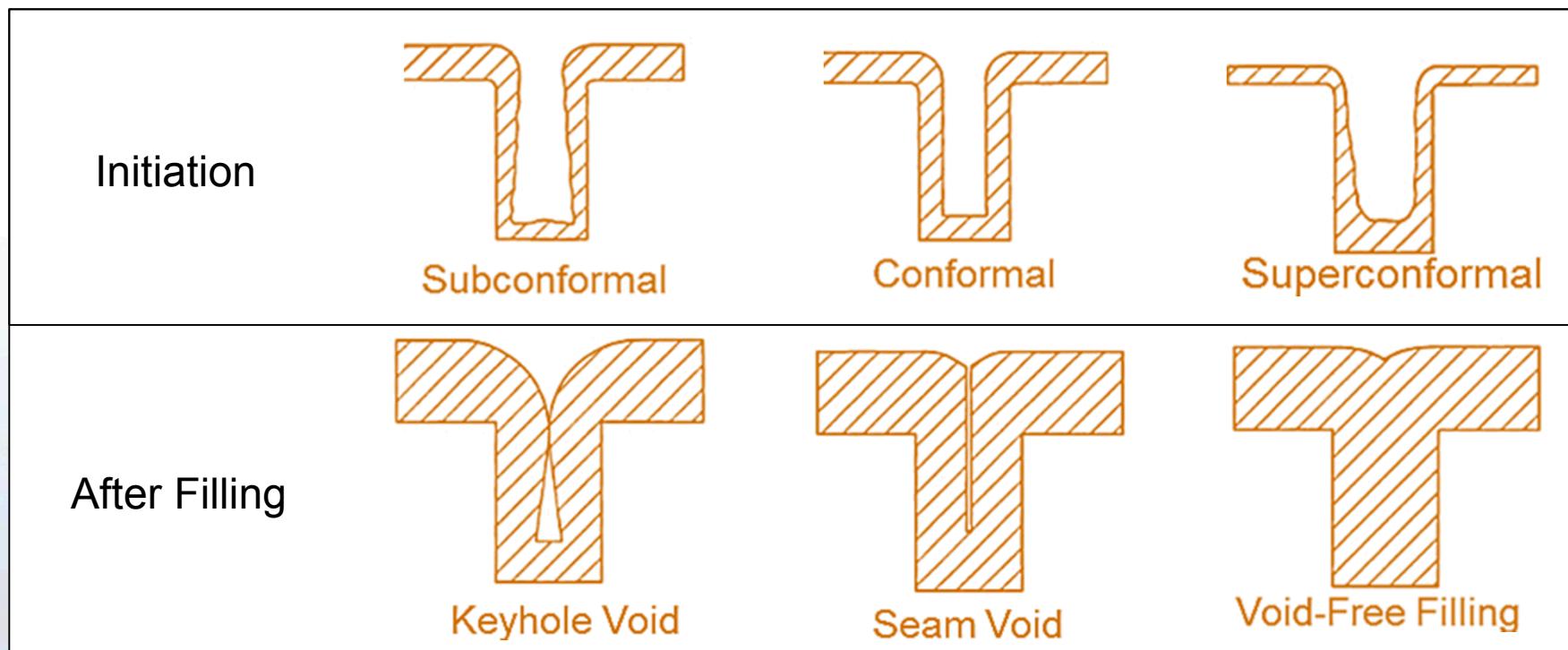
- Prevents Cu from reacting with oxygen
- Allows for an energetically favorable Cu reduction
- Improves film quality
- Improves bath stability





# Growth Front Importance in TSVs

- Any void formation is problematic for these features due to the environment they operate at (ultra high vacuum, low temperature)
- Voids can also trap wet chemistry, leading to device degradation over time

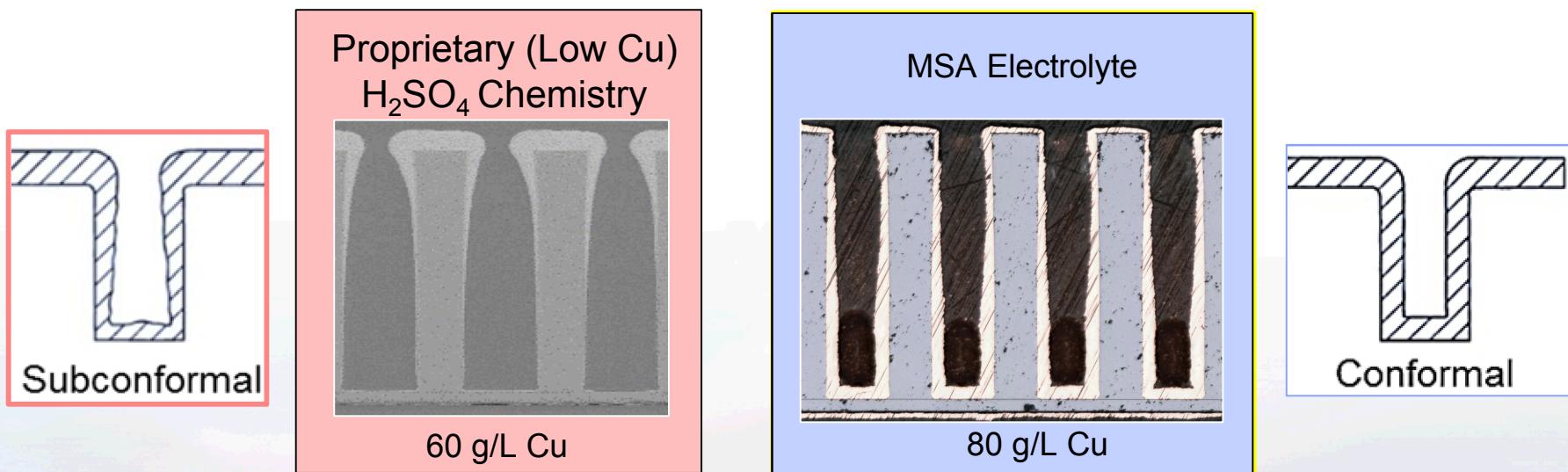


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# Copper Concentration Analysis

- Methanesulfonic acid (MSA) and sulfuric acid ( $H_2SO_4$ ) electrolytes have been widely researched for TSV filling<sup>4,5</sup>.
- Higher solubility of Cu in MSA as compared to  $H_2SO_4$  suggests that MSA is the optimal electrolyte for the purpose of filling these TSVs.



- More Cu ions available in each via decreases mass transport limited depletion of Cu, leading to increased plating conformality.

<sup>4</sup>Cho, Sung Ki, Myung Jun Kim, and Jae Jeong Kim. "MSA as a supporting electrolyte in copper electroplating for filling of damascene trenches and through silicon vias." *Electrochemical and Solid-State Letters* 14.5 (2011): D52-D56.

<sup>5</sup>Moffat, T. P., and D. Josell. "Extreme bottom-up superfilling of through-silicon-vias by damascene processing: suppressor disruption, positive feedback and turing patterns." *Journal of The Electrochemical Society* 159.4 (2012): D208-D216.

<sup>6</sup>Sun, Jian-Jun, et al. "High-aspect-ratio copper via filling used for three-dimensional chip stacking." *Journal of The Electrochemical Society* 150.6 (2003): G355-G358.

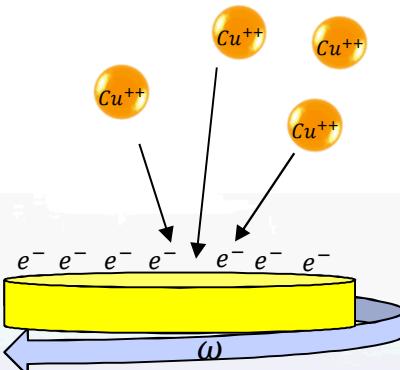


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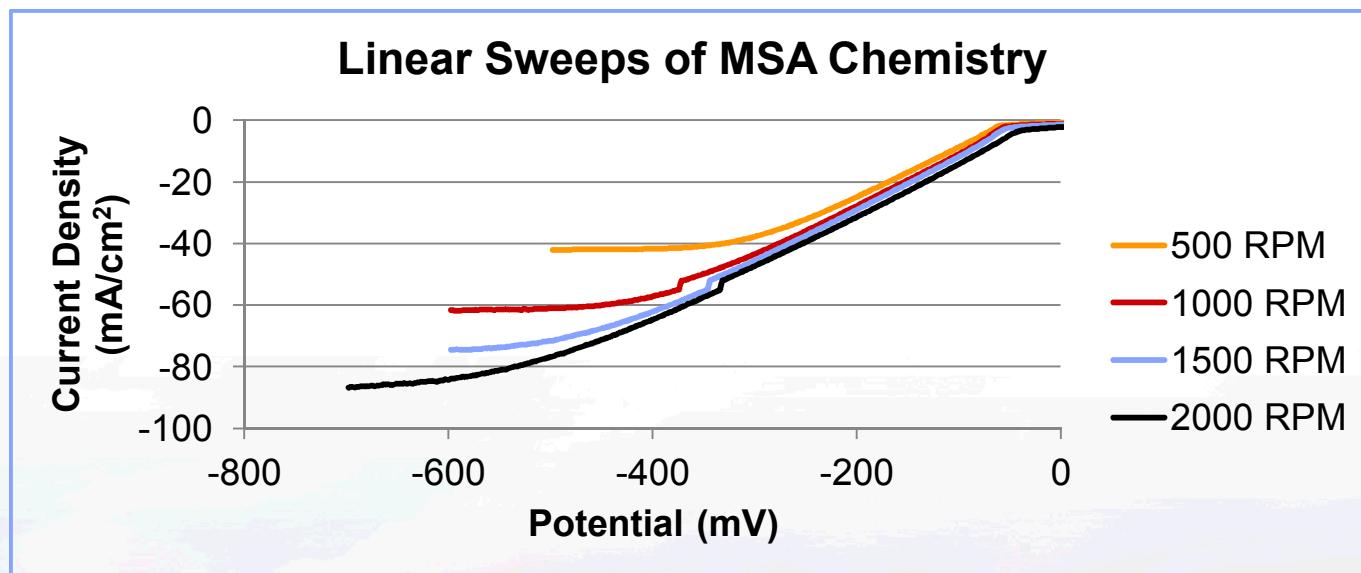
# Plating Parameter/Diffusion Coefficient Study

- A linear voltammetric study was performed using a gold (Au) rotating disk electrode (RDE) to determine the diffusion coefficient of Cu ions in the MSA electrolyte.

Rotating Disk Electrode (RDE)



Diffusion Layer  
Thickness ( $\delta_f$ )  $\propto \omega^{-1/2}$



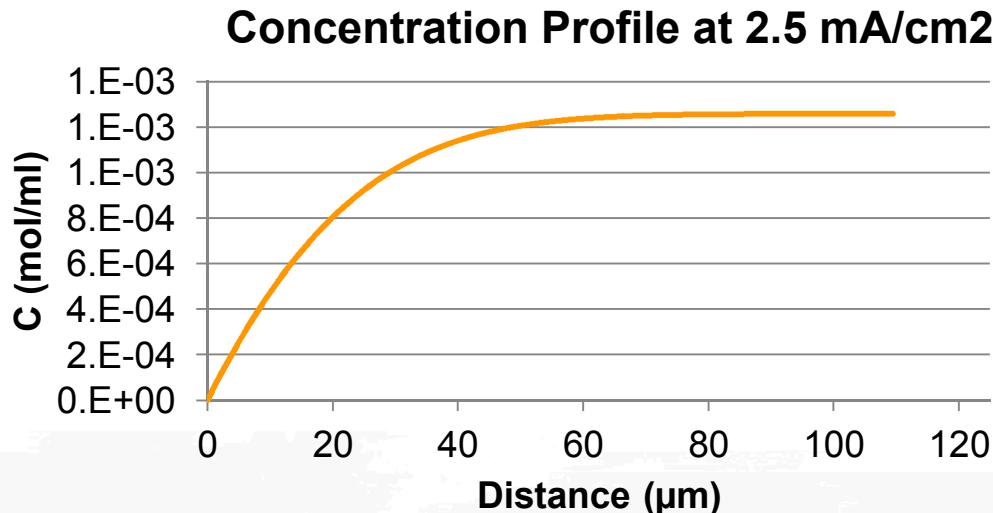
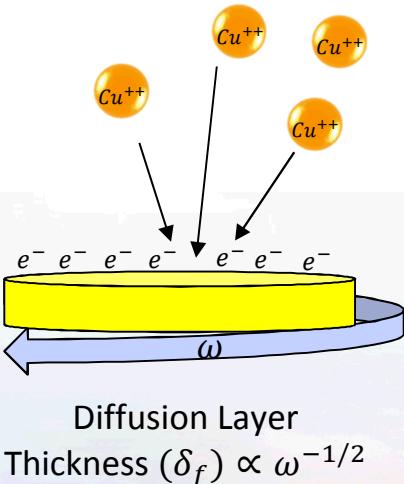
$$I_L = 0.62nFAD^{\frac{2}{3}}\omega^{\frac{1}{2}}v^{\frac{-1}{6}}C$$

Experimentally Determined	Roha, D. Et. Al. Provided	Chuan Seng Tan, K.-N. C. Provided
$D_{Cu}$	$D_{Cu}^{13}$	$D_{Cu}^{14}$
<b>985 <math>\mu\text{m}^2/\text{sec}</math></b>	<b>800 <math>\mu\text{m}^2/\text{sec}</math></b>	<b>1000 <math>\mu\text{m}^2/\text{sec}</math></b>



# Plating Parameter/Diffusion Coefficient Study

Rotating Disk Electrode (RDE)



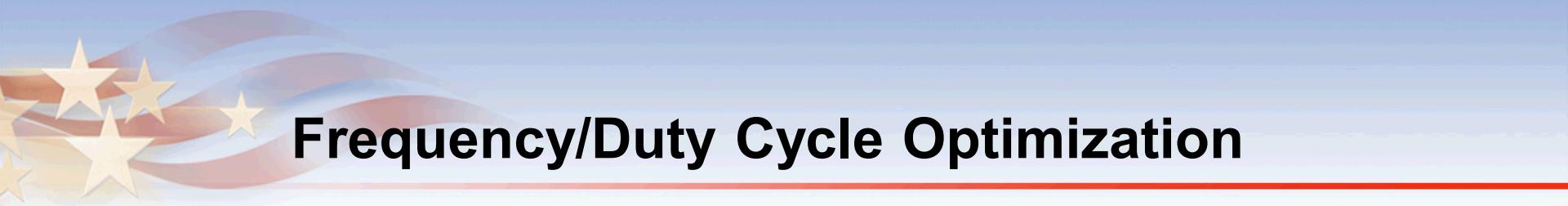
$$C(x, t) = C^* - \frac{i}{nFAD} \left\{ 2 \left( \frac{Dt}{\pi} \right)^{\frac{1}{2}} \exp \left( -\frac{x^2}{4Dt} \right) - x \operatorname{erfc} \left[ \frac{x}{2(Dt)^{\frac{1}{2}}} \right] \right\}$$

**Cu Diffusion Time for Replenishment in TSV Bottom = 1.6 sec**  
**Applied Potential on Time for Cu Depletion = 0.105 sec**

$$\frac{\text{On Time}}{\text{Off Time}} = \frac{105}{1600} = 6.6\% \text{ Duty Cycle}$$

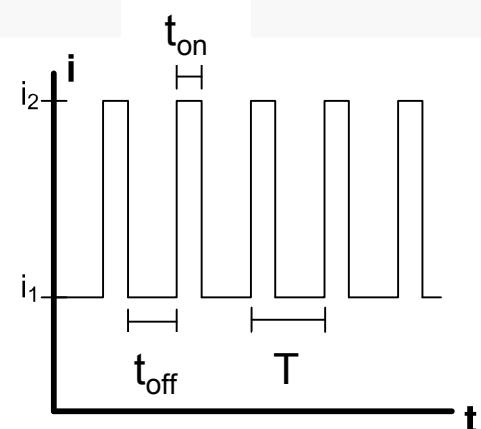


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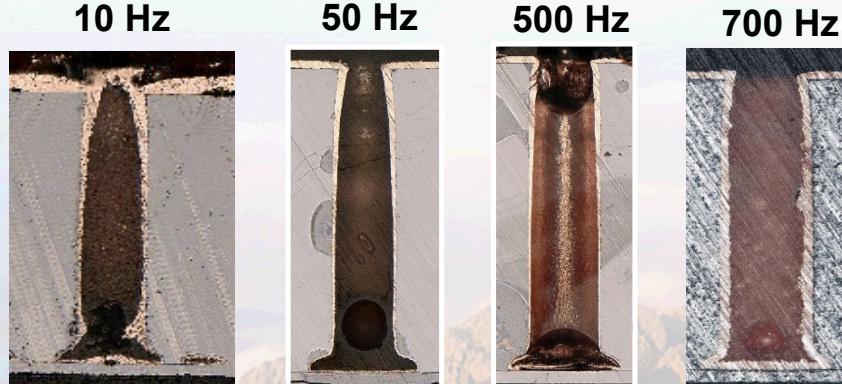


# Frequency/Duty Cycle Optimization

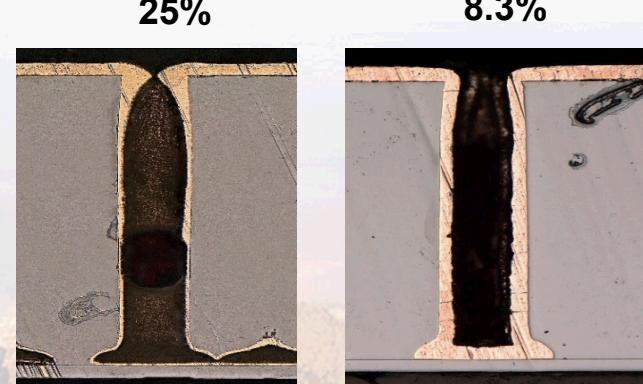
- Pulse plating can mitigate copper depletion issues down the features as “off” time allows for Cu replenishment
- Using pulse plating will require a longer fill time



Frequency Comparison at 50% Duty Cycle



Duty Cycle Comparison at 700 Hz

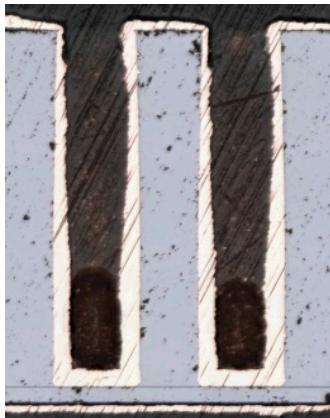


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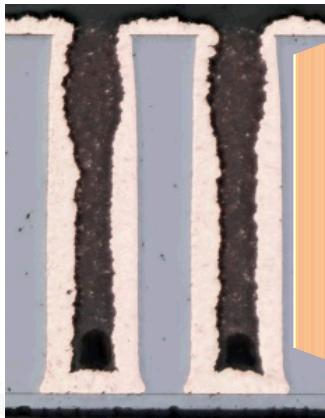
# Further Process Optimization

## Current Density Comparison

5 mA/cm<sup>2</sup>



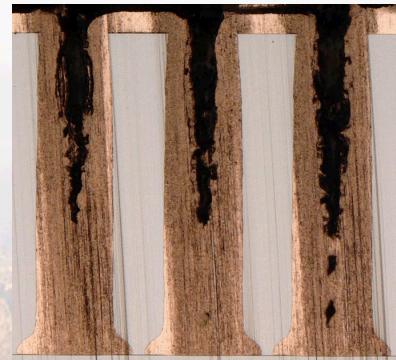
2.5 mA/cm<sup>2</sup>



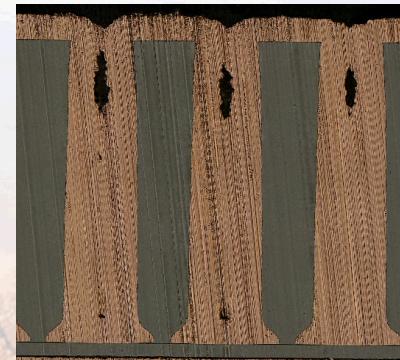
Large grains lead to void formation

Void formation ultimately occurs even after optimizing conditions:

**High Cu concentration, 2.5 mA/cm<sup>2</sup>, 700 Hz, 8.3% Duty Cycle**



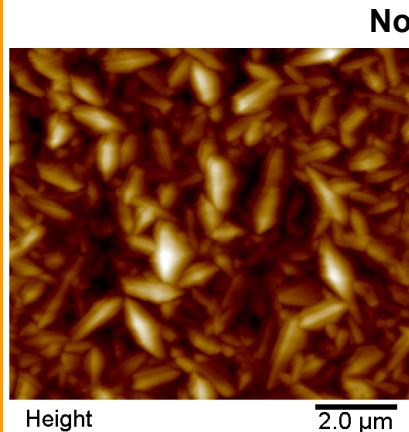
Seam/Keyhole voids



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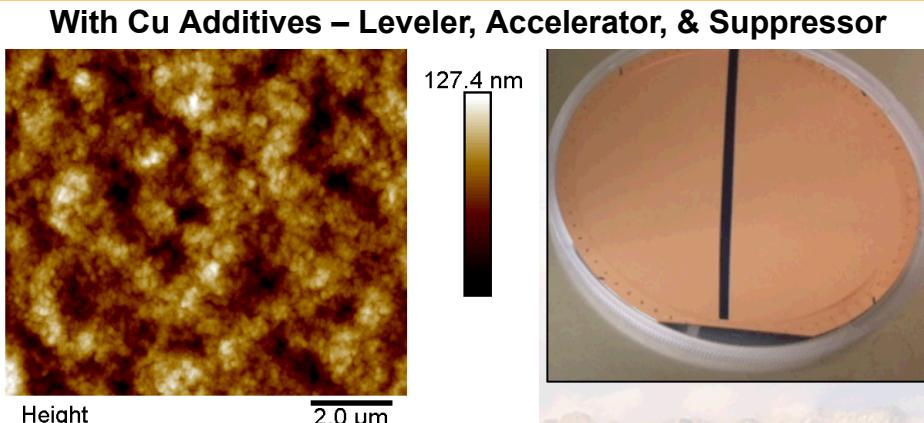
# Effect of Additives on Planar Substrates

## Optical and Atomic Force Microscopy



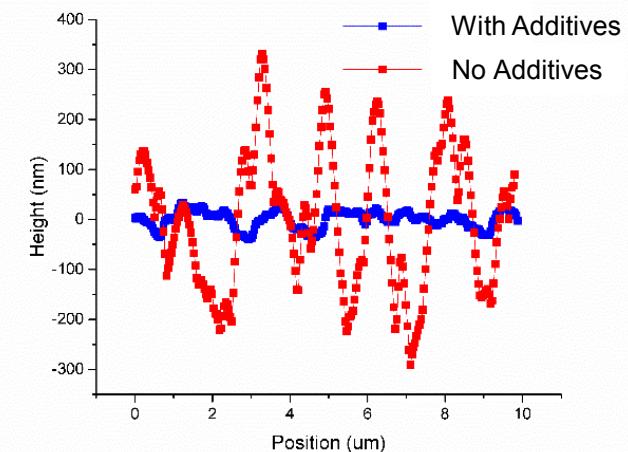
**RMS Roughness**  
134 nm

**Uniformity (Center to Edge)**  
8% - 100%



**RMS Roughness**  
18 nm

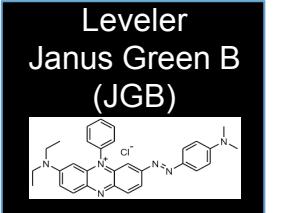
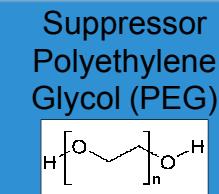
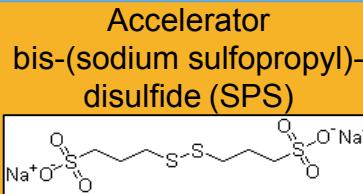
**Uniformity (Center to Edge)**  
< 5%



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# Feature Filling Profiles and Additives

## Common Additives to Achieve Void-Free Fill Profile



HCl  
Or  
NaCl  


- Accelerator – Surfactant molecule that adsorbs on the surface and, by coverage increase with area loss, preferentially increases plating rate in concave regions
- Suppressor - Large chain polymer (1k-20k mW) whose gradient of concentration yields an associated gradient in deposition rate (slower higher in via)
- Leveler - Disables accelerator to reduce overburden thickness; grain refiner
- Chloride – Competitively complexes with suppressor and accelerator species at electrode surface; required for suppressor function



<sup>7</sup> Moffat, Thomas P., et al. "Superconformal film growth: Mechanism and quantification." *IBM Journal of Research and Development* 49.1 (2005): 19-36.



# Curvature Enhanced Adsorbate Coverage<sup>7,8,9</sup>

- More strongly adsorbed molecules (accelerator) agglomerate on concave surfaces during Cu growth by<sup>7,8</sup>:

$$\frac{d\theta}{dt} = k_{ads}(1 - \theta_{SPS})C_{SPS} - k_{inc}\theta^q + v\kappa\theta_{SPS}$$

Where,  $k_{ads} \equiv SPS$  Adsorption Rate Const.

$\theta_{SPS} \equiv SPS$  Surface Coverage

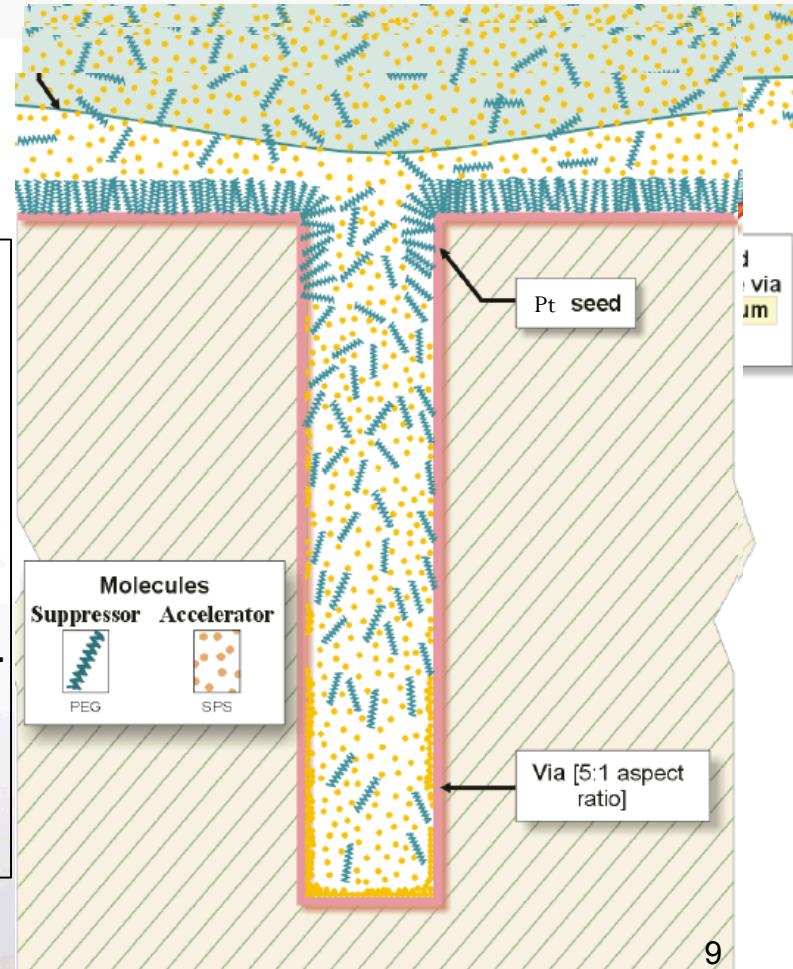
$C_{SPS} \equiv SPS$  Concentration

$k_{inc} \equiv SPS$  Incorporation Rate Const.

$q \equiv$  Reaction Order

$v \equiv$  Interface Velocity

$\kappa \equiv$  Local Curvature



<sup>7</sup> Moffat, Thomas P., et al. "Superconformal film growth: Mechanism and quantification." *IBM Journal of Research and Development* 49.1 (2005): 19-36.

<sup>8</sup> Moffat, T. P., et al. "Curvature enhanced adsorbate coverage model for electrodeposition." *Journal of The Electrochemical Society* 153.2 (2006): C127-C132.

<sup>9</sup> Keigler, A., et al. "Sematech 3D Equipment Challenges: 300mm Copper Plating." *NEXX Systems* (2008).



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# CEAC Mechanism Scaling

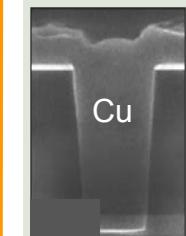
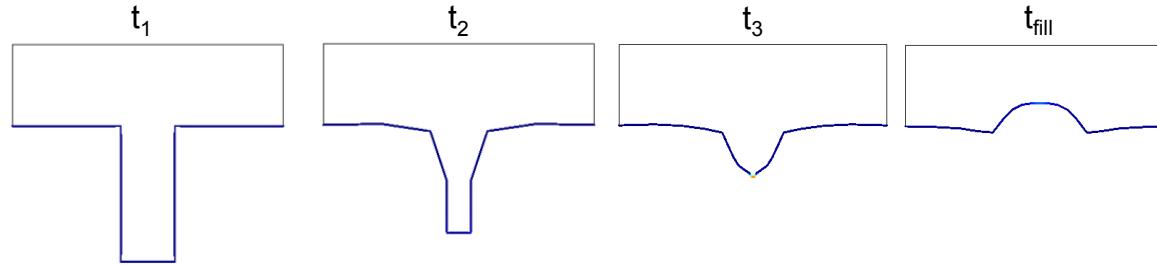
## COMSOL Modeling

Scale

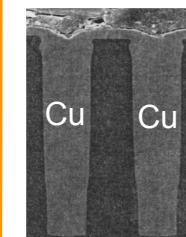
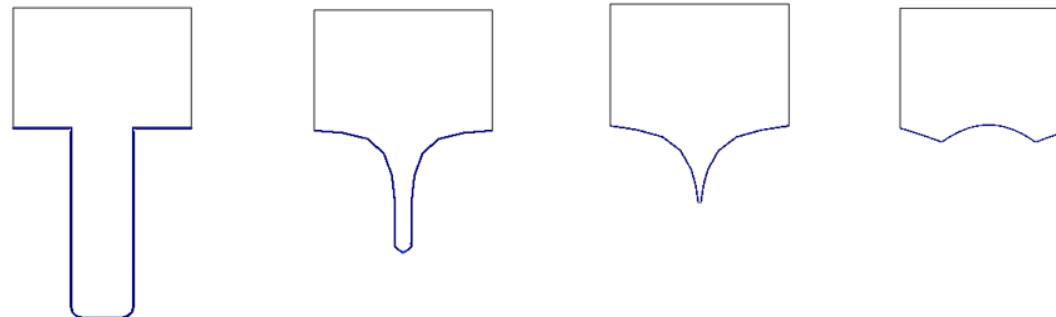
Modeled Growth Front Tracking

Experimental Results<sup>6,7</sup>

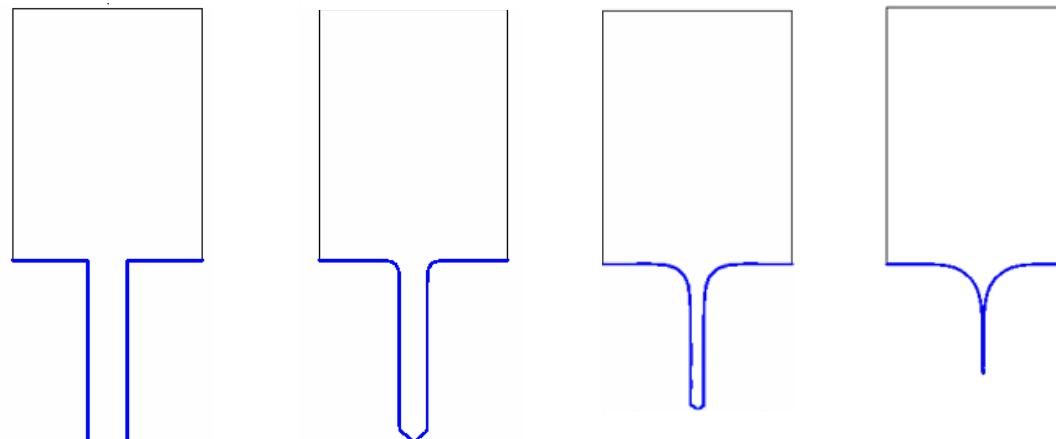
0.5 x 3.25  $\mu\text{m}$   
TSV



10 x 70  $\mu\text{m}$   
TSV



Ion Trap 125 x 600  $\mu\text{m}$   
TSV





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- Wire Bond Issues
- TSVs in Ion Traps
- Cu TSV ECD Electrical Connection & Isolation

## Conformal Copper TSV Filling

- ECD Basics
- Wetting Issues
- Fill Profile Requirements
- Chemical Analysis
- CEAC Mechanism Scalability

## Bottom-Up Filling Technique

- Fundamentals
- Parameter Modification Optimization
- Ion Trap TSV Application
- Plating Approach
- Transition to MSA



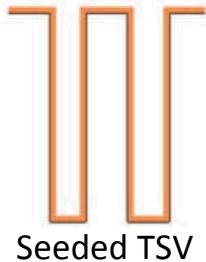
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# Alternative Approach to TSV Filling

## S-Shaped Negative Differential Resistance (S-NDR) Approach<sup>10</sup>

- Single suppressor additive system
- Bottom-up growth mode even with conformal seed metal

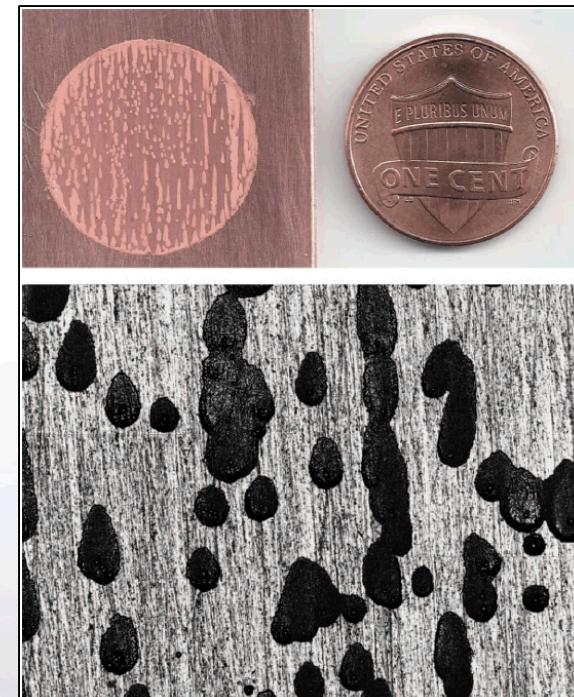
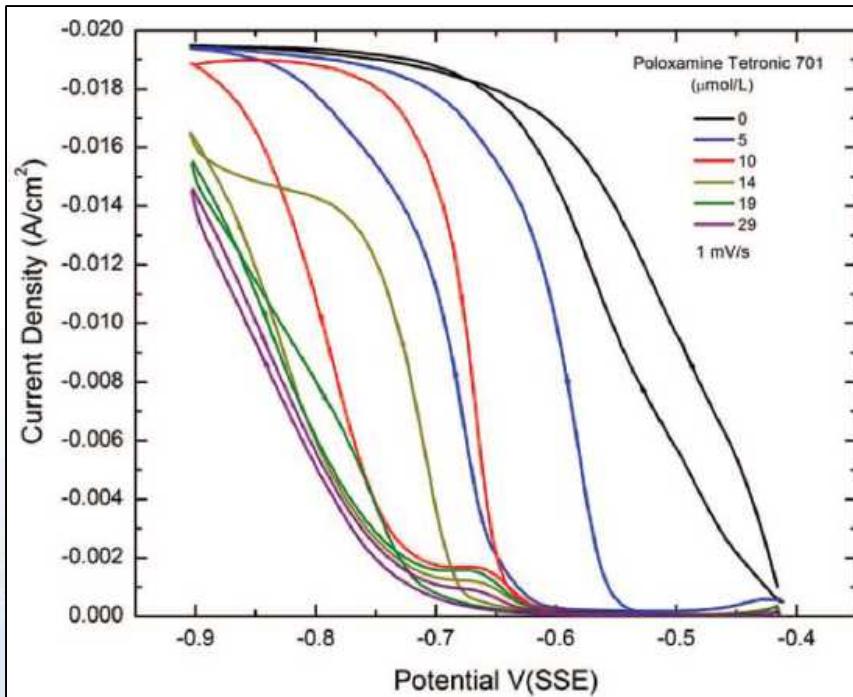


<sup>10</sup> Moffat, T. P., and D. Josell. "Extreme bottom-up superfilling of through-silicon-vias by damascene processing: suppressor disruption, positive feedback and turing patterns." *Journal of The Electrochemical Society* 159.4 (2012): D208-D216.



# S-NDR Research<sup>10</sup>

- Hysteresis represents breakdown of suppressor blocking layer
- This hysteresis causes 'Turing Patterns' on planar electrodes

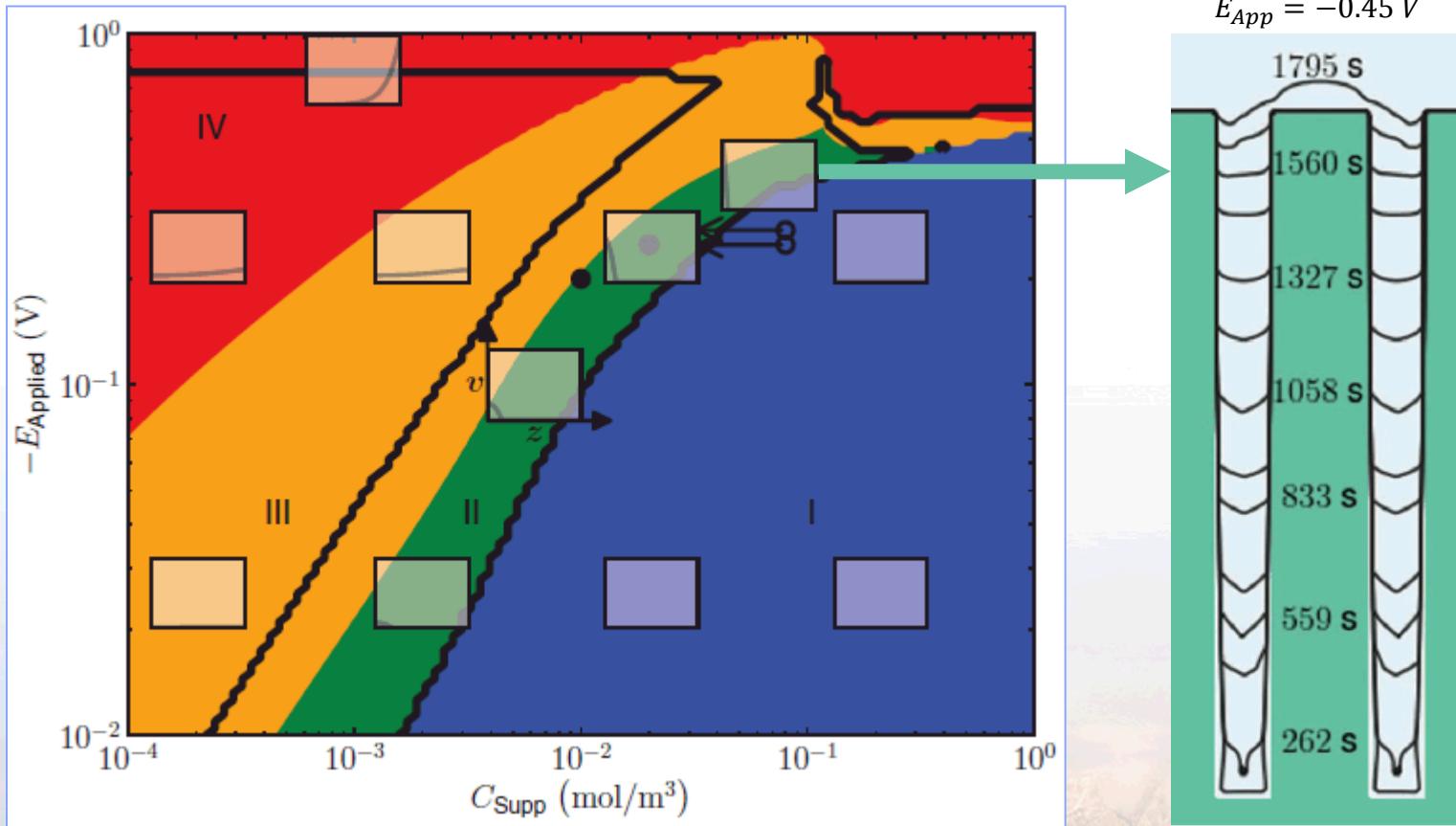


<sup>10</sup> Moffat, T. P., and D. Josell. "Extreme bottom-up superfilling of through-silicon-vias by damascene processing: suppressor disruption, positive feedback and turing patterns." *Journal of The Electrochemical Society* 159.4 (2012): D208-D216.



# S-NDR Technique Fundamentals<sup>11,12</sup>

- Suppressor concentration and applied potential are two key parameters



$$C_{\text{Supp}} = 0.06 \frac{\text{mol}}{\text{m}^3}, \quad E_{\text{App}} = -0.45 \text{ V}$$

<sup>11</sup> Josell, D., D. Wheeler, and T. P. Moffat. "Modeling extreme bottom-up filling of through silicon vias." *Journal of The Electrochemical Society* 159.10 (2012): D570-D576.

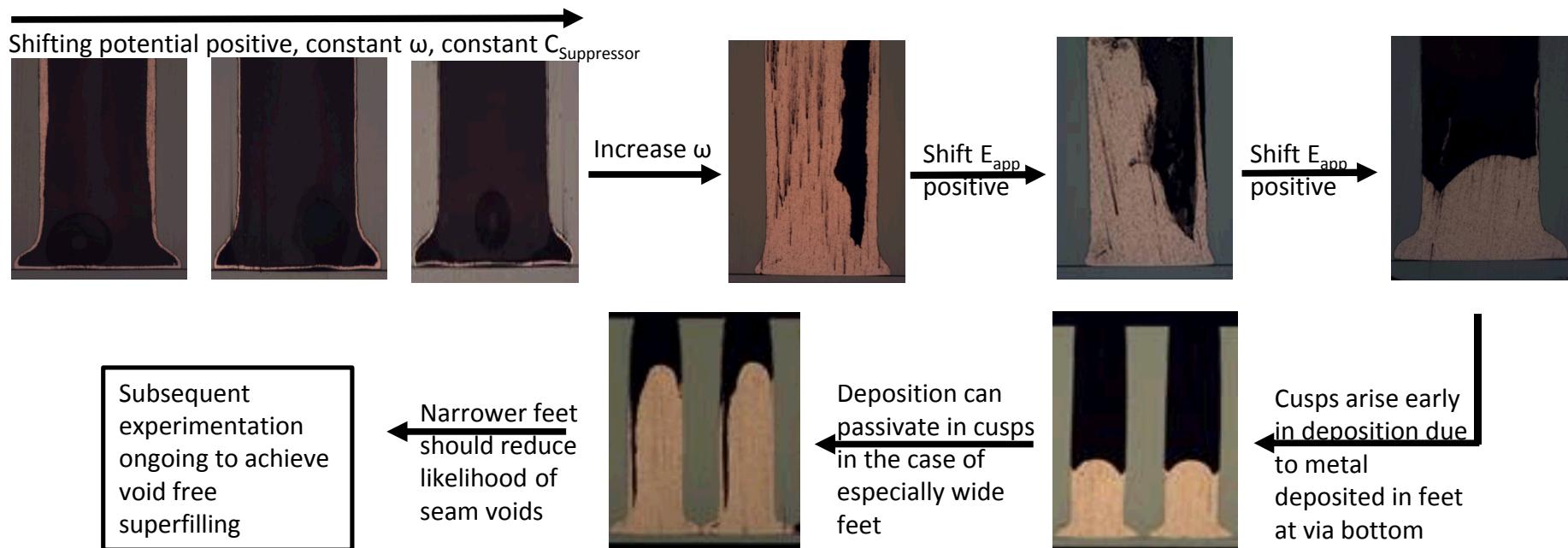
<sup>12</sup> Wheeler, D., T. P. Moffat, and D. Josell. "Spatial-temporal modeling of extreme bottom-up filling of through-silicon-vias." *Journal of The Electrochemical Society* 160.12 (2013): D3260-D3265.



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# S-NDR Approach on Ion Trap TSVs

Three key variables manipulated: Applied potential, electrolyte fluid flow, and suppressor concentration

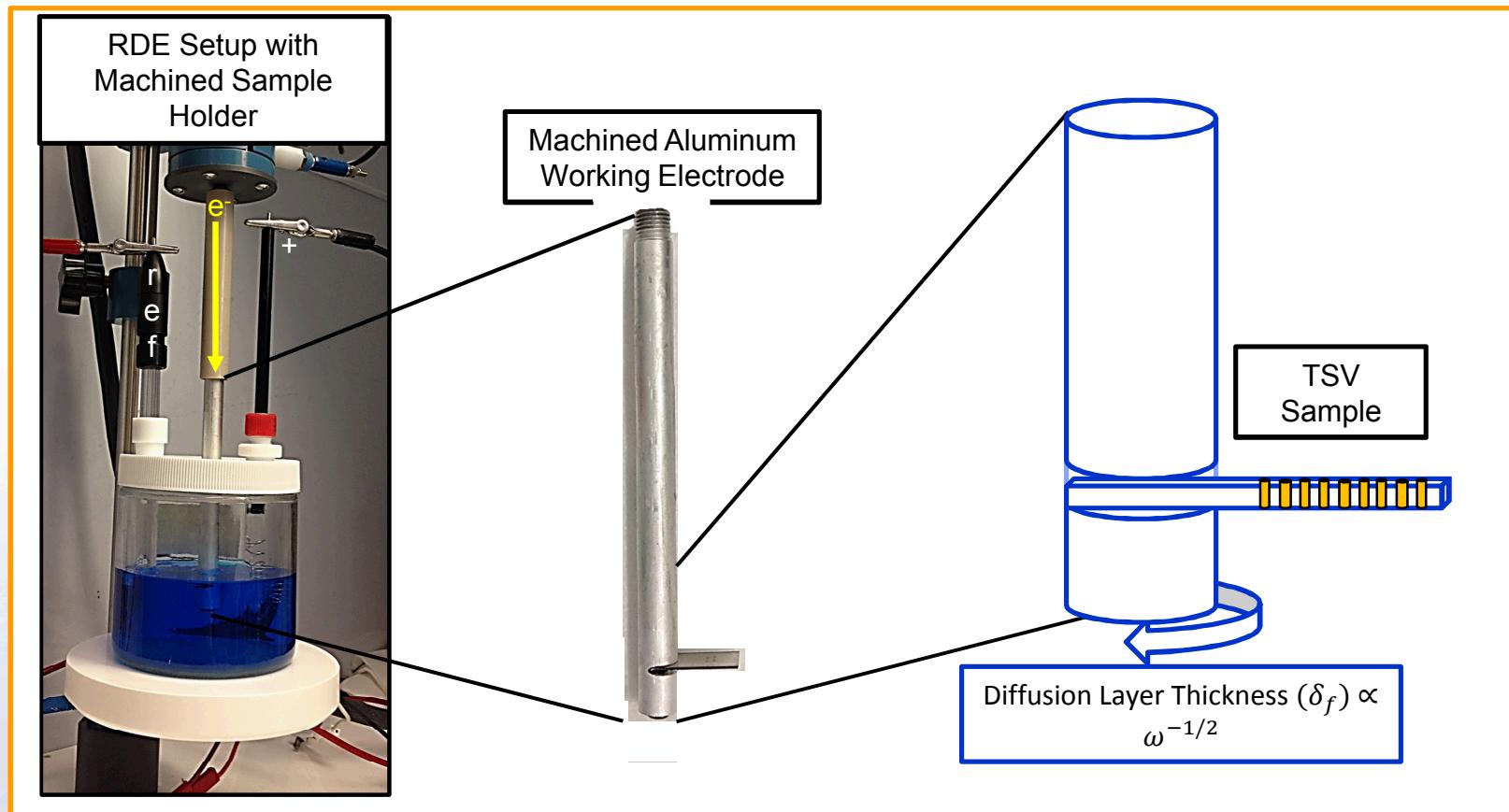


$$\text{Diffusion Layer Thickness} \propto \omega^{-1/2} = \delta_f = 1.61 (D_f)^{1/3} v^{1/6} \omega^{-1/2}$$



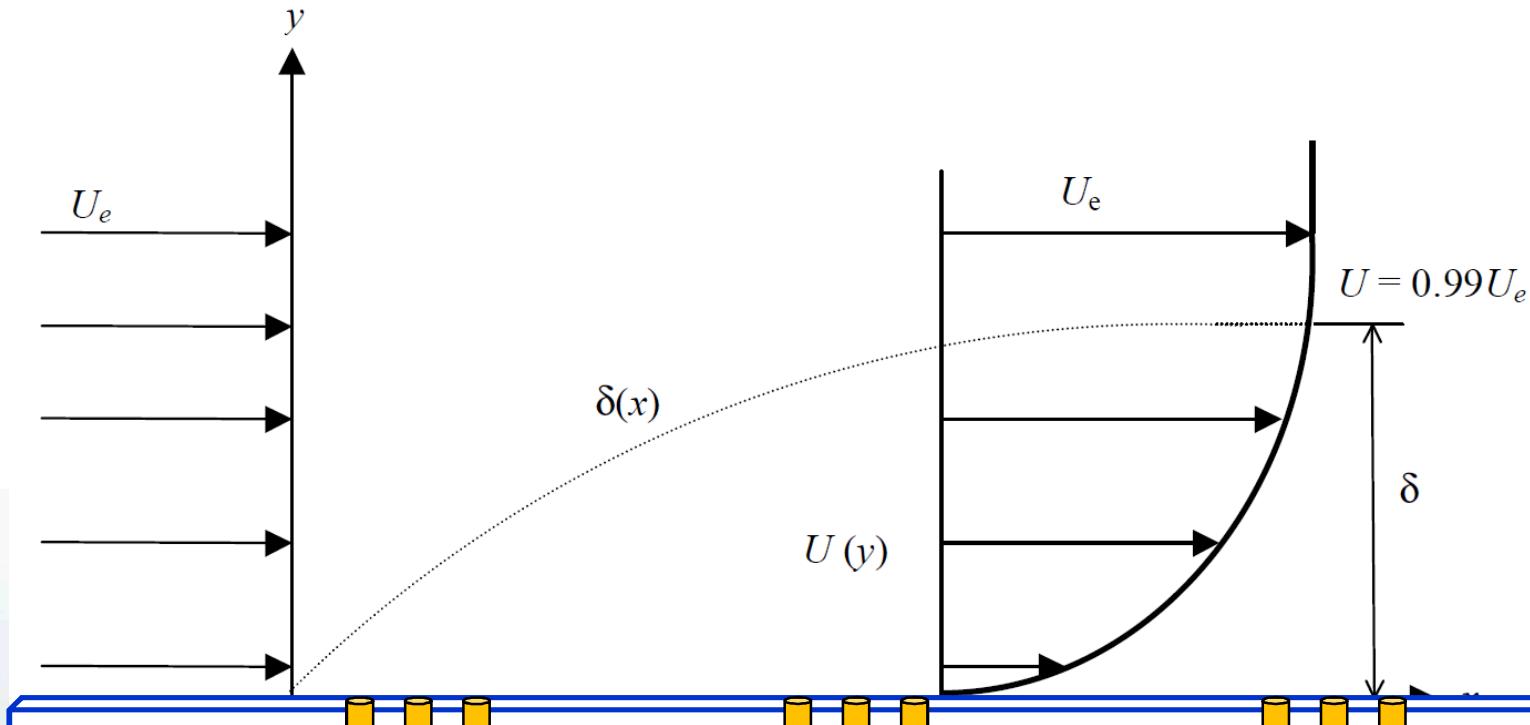
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# S-NDR Plating Setup



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# Boundary Layer Thickness Above Rotating Planar Substrate



**Fluid Velocity ( $u$ )**

$$u = 2\pi r \omega_{cyc}$$

**Boundary Layer Thickness ( $\delta$ )**

$$\delta = \sqrt{\frac{vx}{u}}$$

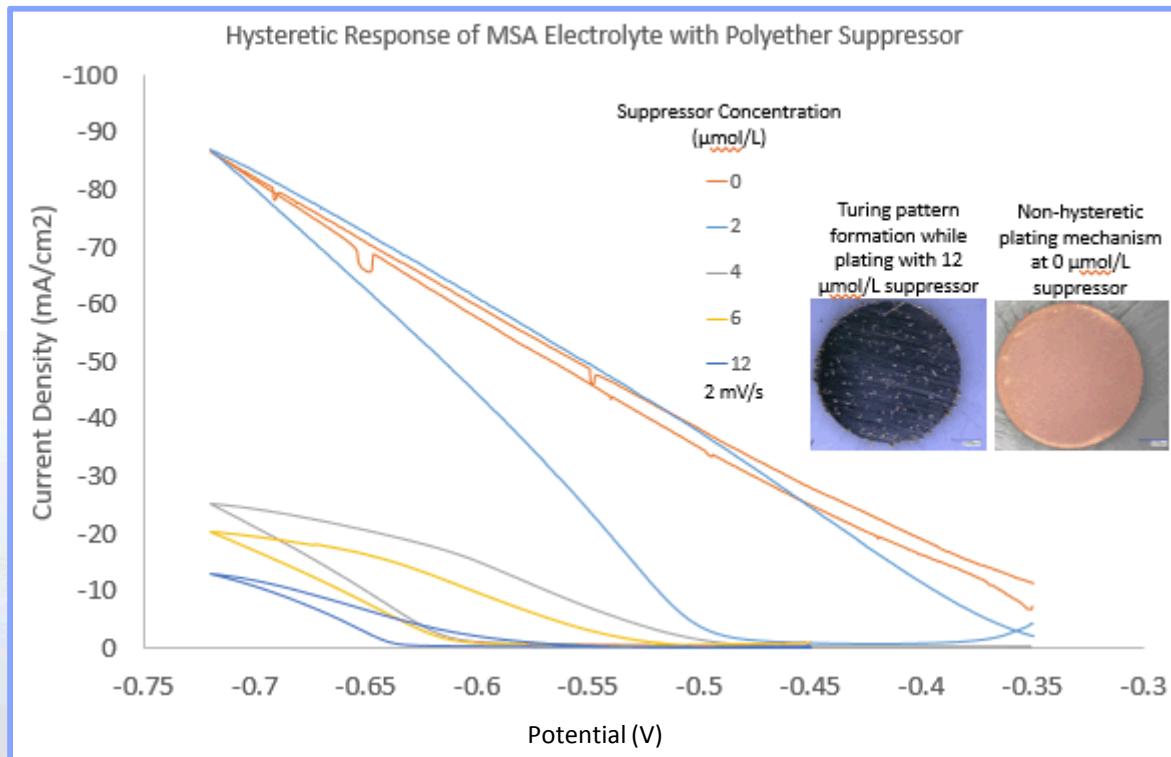
$$\delta_{avg} = \frac{1}{x} \sqrt{\frac{v}{u}} \int_0^x \sqrt{x} dx$$



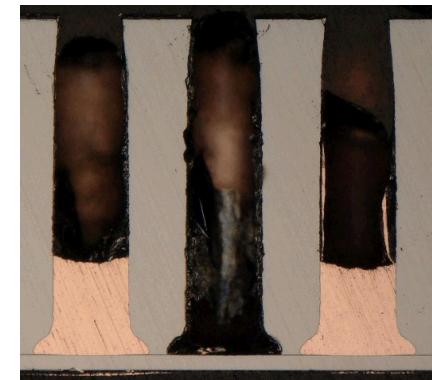
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# S-NDR Results with MSA Electrolyte

- S-NDR mechanism can likely be applied to high Cu, MSA electrolyte to increase plating rate



R8, -0.6 V (SSE),  
2 hr, 400 rpm



R11, -0.6 V  
(SSE), 4.5 hr,  
400 rpm



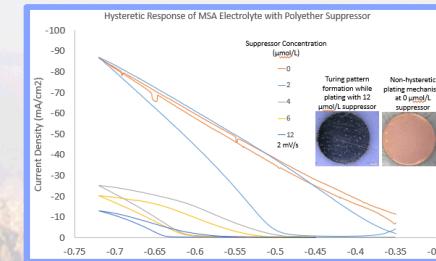
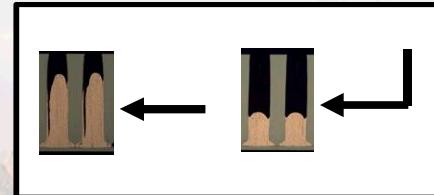
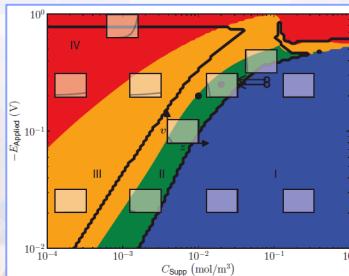
# Summary

## Established:

- TSV wetting procedure
- MSA as electrolyte for sufficient conformal jacketing
- Traditional additive system deficiencies

## Future Work:

- Examine alternative TSV wetting procedures
- Continue replicating S-NDR approach with  $\text{H}_2\text{SO}_4$  electrolyte
- Develop S-NDR process with MSA electrolyte



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