

## EFFECTS OF GAS RELATIVE PERMEABILITY HYSTERESIS AND SOLUBILITY ON ASSOCIATED CO<sub>2</sub> STORAGE PERFORMANCE

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24 **ABSTRACT**

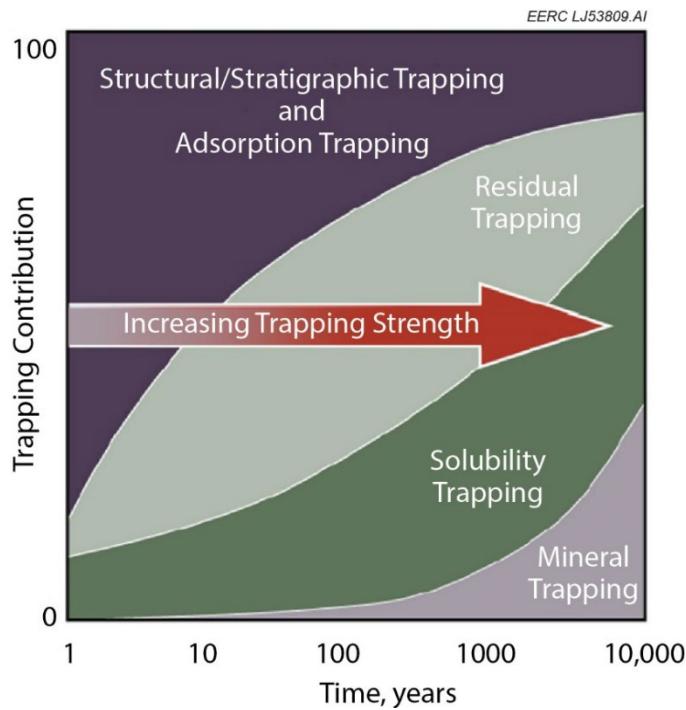
25 CO<sub>2</sub> enhanced oil recovery (EOR) has been carried out in the Bell Creek oil field since 2013.  
26 Together with the encouraging oil production results, a considerable quantity of CO<sub>2</sub> has also been  
27 trapped in the reservoir as a normal part of the EOR process, also referred to as associated storage.  
28 Because of the complex geologic conditions in the field, a series of experimental and modeling  
29 work have been conducted to better understand the CO<sub>2</sub> EOR and associated storage performance  
30 in the reservoir. Effects of gas relative permeability hysteresis and solubility on associated CO<sub>2</sub>  
31 storage performance are thoroughly investigated in this study.

32 A proportion of injected CO<sub>2</sub> remains behind through residual and solubility trapping  
33 mechanisms when CO<sub>2</sub> flows through a reservoir during a CO<sub>2</sub> EOR process. Over 50 core plugs  
34 were collected from the reservoir to characterize the rock properties. Mineralogical analysis and  
35 capillary pressure measurements showed that the mineral composition and pore-size distribution  
36 in the reservoir are favorable for residual trapping of CO<sub>2</sub>. The hysteresis of gas relative  
37 permeability was measured to assess the effect of residual trapping on associated CO<sub>2</sub> storage  
38 using steady-state relative permeability tests and reservoir simulation. The reservoir oil was  
39 characterized based on pressure–volume–temperature experiments and Peng–Robinson equation  
40 of state modeling, which showed that CO<sub>2</sub> solubility in oil is much greater ( $\geq 5$  times) than in  
41 water. Results indicated that depleted oil reservoirs have great potential to store a huge quantity of  
42 CO<sub>2</sub> associated with EOR operations, as residual oil saturation is 0.3 or greater in most  
43 conventional oil reservoirs after water flooding.

44 **KEYWORDS:** CO<sub>2</sub> enhanced oil recovery, associated CO<sub>2</sub> storage, relative permeability  
45 hysteresis, residual trapping, solubility trapping

46 **1. INTRODUCTION**

47 CO<sub>2</sub> trapping and associated storage processes are important to enhanced oil recovery (EOR)  
48 performance, as they can affect the oil recovery and CO<sub>2</sub> utilization factor (Belhaj et al., 2013;  
49 Gozalpour et al., 2005; Kovscek, 2002; Malik and Islam, 2000; Verma, 2015; Soltanian et al.  
50 2017). For instance, more trapped CO<sub>2</sub> may lead to a higher CO<sub>2</sub> utilization factor since less CO<sub>2</sub>  
51 is available to contact oil and sweep it from the reservoir; therefore, more CO<sub>2</sub> needs to be  
52 purchased and injected for an equivalent oil recovery (Gao et al., 2013; Jin et al., 2017b; Li et al.,  
53 2006; Zhang et al., 2010). The trapping mechanisms also determine the state of the associated CO<sub>2</sub>  
54 storage during and after CO<sub>2</sub> flooding. Fig. 1 is a generalized illustration which shows how the  
55 contribution from different trapping mechanisms change over time leading to an increase in CO<sub>2</sub>  
56 trapping strength (or security) (Metz et al., 2005). Effective trapping mechanisms ensure injected  
57 CO<sub>2</sub> will remain, in permanence, within the area of review (limited lateral migration) and contained  
58 within the zone of interest (limited vertical migration). The four primary CO<sub>2</sub>-trapping  
59 mechanisms include structural/stratigraphic, residual, solubility, and mineral trapping in most  
60 conventional petroleum reservoirs (Jia et al., 2016). Adsorption trapping is an important  
61 mechanism in unconventional reservoirs, such as shale oil, shale gas, and coalbed methane (CBM)  
62 reservoirs, as these reservoirs have higher percentages (>5%) of organic content and a large  
63 number of nanometer-size pores (Wong et al., 2007; Busch et al., 2008; Gale and Freund, 2001;  
64 Jia et al., 2017; Jin et al., 2016, 2017a; Khosrokhavar et al., 2014; Ross and Bustin, 2009). Mineral  
65 trapping, with the exception of a small number of documented instances of CO<sub>2</sub> storage in basalt  
66 formations (McGrail et al., 2016, 2006), is thought to occur over an extended time frame (hundreds  
67 to thousands of years).



68

69 **Fig. 1.** Increase of CO<sub>2</sub> trapping strength with time (modified from Intergovernmental Panel on  
70 Climate Change, 2005).

71

72 Studies have shown that residual CO<sub>2</sub> saturation may be on the order of 5%–30%, varying  
73 with reservoir conditions (Ennis-King and Paterson, 2001; Juanes et al., 2006; Niu et al., 2015;  
74 Zuo and Benson, 2014). Therefore, understanding residual trapping in a reservoir can provide a  
75 conservative estimate of CO<sub>2</sub> storage potential for sequestration projects (Burnside and Naylor,  
76 2014; Krevor et al., 2015; Al-Menhali and Krevor, 2016). The residual trapping of CO<sub>2</sub> in a small  
77 pore space has been visualized and analyzed accurately at core scale (Iglauer et al., 2011; Ruprecht  
78 et al., 2014). Using core-flooding with x-ray computed tomography, the results indicated that the  
79 hysteretic nonwetting phase behavior (i.e., relative permeability hysteresis of CO<sub>2</sub>) would be a  
80 significant factor in determining long-term immobilization of injected CO<sub>2</sub> in the reservoir.  
81 Accurate determination of relative permeability hysteresis is also important for CO<sub>2</sub>-based EOR

82 projects since many of them use water alternating gas (WAG) operations, where CO<sub>2</sub> hysteresis  
83 directly relates to the displacing efficiency (Fatemi et al., 2012).

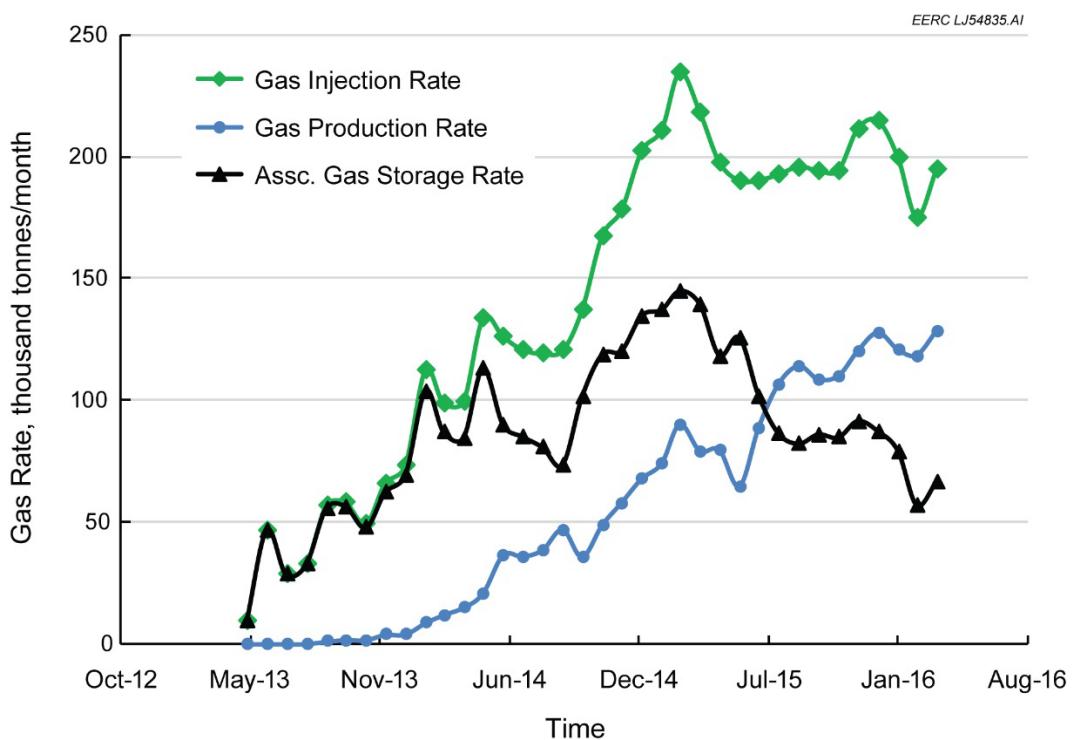
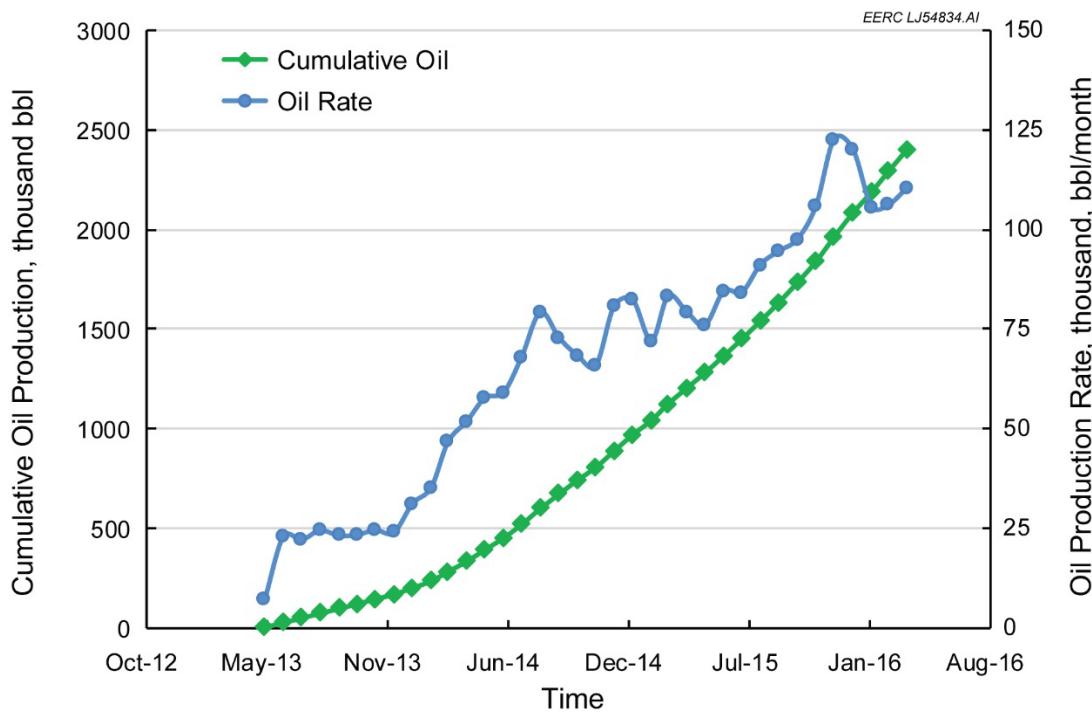
84 CO<sub>2</sub> dissolves in other formation fluids when injected into a reservoir, a process termed  
85 solubility trapping. The density of oil increases when CO<sub>2</sub> is dissolved in the oil (Holm and  
86 Josendal, 1974), which may create gravitational instability in the reservoir, leading to convective  
87 mixing of fluids. The mixing of fluids with differing dissolved CO<sub>2</sub> content will further enhance  
88 the dissolution process in the long run (Li and Jiang, 2014; Shelton et al., 2016; Szulczewski et al.,  
89 2013). Therefore, CO<sub>2</sub> dissolution is considered a significant trapping mechanism in deep geologic  
90 formations, with potential to permanently store large amounts of CO<sub>2</sub> (Ampomah et al., 2016;  
91 Bachu and Adams, 2003; Bachu and Bennion, 2007; Metz et al., 2005; [Holubnyak et al., 2018](#)).

92 A literature review showed that most of the studies on CO<sub>2</sub>-trapping mechanisms are based  
93 on dedicated CO<sub>2</sub> storage in deep saline formations (Al-Khdheeawi et al., 2018; Bachu and Adams,  
94 2003; Bachu and Bennion, 2007; Burnside and Naylor, 2014; Iglauer, 2011; Juanes et al., 2006;  
95 Krevor et al., 2015; Szulczewski et al., 2013). The effects of these mechanisms on associated CO<sub>2</sub>  
96 storage during EOR operations have not been discussed thoroughly. There is still a lack of rock  
97 and fluid characterization data from actual oil fields to clearly demonstrate the correlations  
98 between these trapping mechanisms and associated CO<sub>2</sub> storage performance. In this study, a series  
99 of experimental and simulation work has been conducted to investigate the effects of residual and  
100 solubility trapping mechanisms on associated CO<sub>2</sub> storage performance in the Bell Creek oil field,  
101 where CO<sub>2</sub>-based EOR operations are in progress. This is part of a larger study on CO<sub>2</sub> associated  
102 storage being conducted by the Plains CO<sub>2</sub> Reduction (PCOR) Partnership at the Bell Creek oil  
103 field in southeastern Montana (Braunberger et al., 2014; Gorecki et al., 2013; Hamling et al., 2013;  
104 2016).

105 **2. CO<sub>2</sub> EOR AND ASSOCIATED STORAGE IN THE BELL CREEK OIL FIELD**

106 Since discovery in 1967, the Bell Creek oil field has undergone primary production (solution  
107 gas drive), waterflooding, and two micellar–polymer pilot tests and CO<sub>2</sub>-based EOR since 2013.  
108 Over 40 years of waterflooding in the field has resulted in a reservoir saturated with water and  
109 with oil at residual saturation levels ~30%–45%. Fig. 2 shows the encouraging oil production  
110 performance of the field from the beginning of CO<sub>2</sub> injection. The oil production rate has increased  
111 from 7 thousand barrels per month (bpm) to 110 thousand bpm, which has yielded over 2.4 MMbbl  
112 of oil in the first 3 years of EOR operations. CO<sub>2</sub> injection increased from 9.5 to 234.5 thousand  
113 tonnes per month (tpm) in the first 2 years of injection and then fluctuated around 195 thousand  
114 tpm after June 2015, as shown in Fig. 3 (Montana Board of Oil and Gas Conservation, 2017). CO<sub>2</sub>  
115 production lagged behind the injection for about 9 months, indicating that CO<sub>2</sub> can effectively  
116 displace oil in the pore space and remain in place during flooding operations. About 5 million  
117 tonnes of CO<sub>2</sub> has been injected into the reservoir, and over 3 million tonnes has been stored there  
118 since the beginning of CO<sub>2</sub> flooding (as of March 2016). From Fig. 3, it is also clear that the gas  
119 storage rate related closely to the injection rate, decreasing rapidly when the injection became  
120 stable, after June 2015, while the production rate continued increasing. This observation indicates  
121 that CO<sub>2</sub> dominated the flow networks between injection and production wells and likely means  
122 gas production will continue to increase as the flooding goes on.

123



129 Of the primary CO<sub>2</sub>-trapping mechanisms shown in Fig. 1, the mineral trapping is of  
130 decreased importance when immediate containment/conformity of injected CO<sub>2</sub> is considered, and  
131 the effect of adsorption trapping is minimal since there is a lack of organic content in this  
132 conventional reservoir. Therefore, these two mechanisms will likely have no impacts on  
133 operational activities in the Bell Creek oil field. As such, CO<sub>2</sub> mineralization and adsorption have  
134 not been a focus and will not be discussed further. Since the reservoir is strongly heterogeneous  
135 and the CO<sub>2</sub> floods operations are conducted using WAG, the residual and solubility trapping  
136 mechanisms are important for the CO<sub>2</sub> flow behavior in the Bell Creek oil field and, therefore,  
137 were investigated and are discussed in the following sections.

### 138 **3. RESIDUAL TRAPPING OF CO<sub>2</sub>**

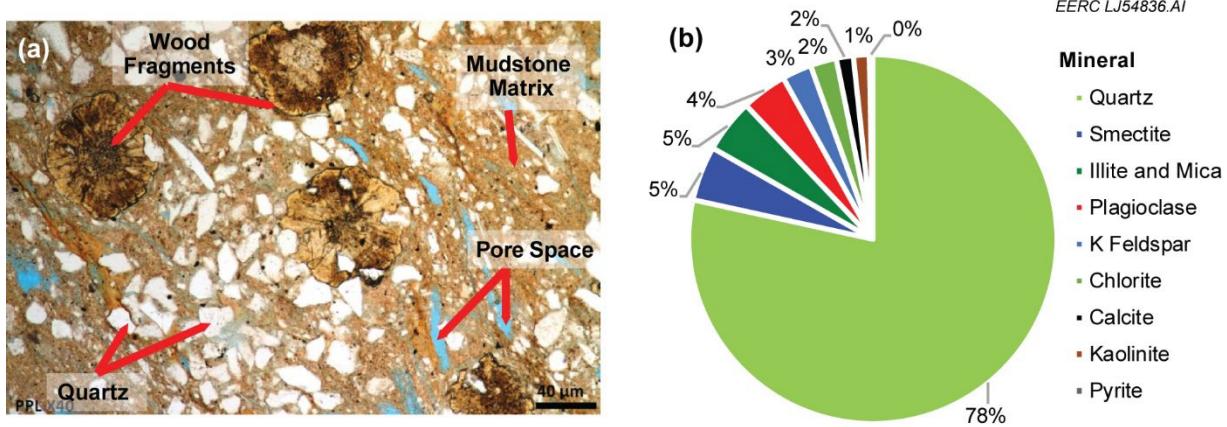
139 Residual trapping occurs rapidly after CO<sub>2</sub> is injected into the formation under the effects of  
140 wettability and capillary pressure, resulting in immobilization of CO<sub>2</sub> in the pore space (Afonja et  
141 al., 2012; Al-Khdheeawi et al., 2018; Krevor et al., 2012, 2015; Raza et al., 2015, 2016). Relative  
142 permeability is a concept used to describe individual fluid-phase mobility when multiple fluid  
143 phases are present while accounting for wettability and capillary pressure phenomena. Injection of  
144 CO<sub>2</sub> results in increasing near-wellbore CO<sub>2</sub> saturation (which continues to increase away from  
145 the injection point as injection progresses) accompanied by a decrease in brine/oil saturation, in  
146 which case relative permeability of CO<sub>2</sub> increases. After injection ends as CO<sub>2</sub> migrates updip, or  
147 if water is injected after CO<sub>2</sub>, the near wellbore CO<sub>2</sub> saturation will decrease away from the  
148 injection point with (accompanied by an increase in brine/oil saturation), so the relative  
149 permeability of CO<sub>2</sub> decreases. As CO<sub>2</sub> saturation decreases, a “residual” saturation will eventually  
150 be reached at which CO<sub>2</sub> is effectively immobilized and, therefore, considered stabilized under the  
151 effects of residual CO<sub>2</sub> trapping (Spiteri et al., 2008). Thus predicting the extent of CO<sub>2</sub> migration

152 within the reservoir under the effects of residual trapping requires an estimate of residual CO<sub>2</sub>  
153 saturation.

154 An additional complexity is that the shape of relative permeability curves may be different  
155 depending on the directionality of changing fluid saturations (imbibition versus drainage), termed  
156 relative permeability hysteresis. The replacement of in situ liquid by injected CO<sub>2</sub> is termed  
157 drainage (nonwetting gas phase replaces the wetting liquid phase). In the WAG injection process,  
158 the gas and liquid phases alternately displace each other, meaning the drainage and imbibition  
159 processes occur in cycles. Hysteresis occurs under the effects of wettability and capillary pressure  
160 when CO<sub>2</sub> is present. This is important to understand in investigations of CO<sub>2</sub> storage, as the effect  
161 is usually pronounced when multiple fluids occupy the same system and may have direct  
162 implications to CO<sub>2</sub> migration and the trapping of CO<sub>2</sub> in the pore space (Burnside and Naylor,  
163 2014). Rock properties, such as mineral composition and pore-size distribution (PSD), play  
164 fundamental roles in understanding the capillary effects and residual trapping in the reservoir.

165 Over 50 core plugs were collected from different wells which penetrate through the main  
166 sandstone of the reservoir. A detailed evaluation of rock properties was conducted using  
167 photomicrography and x-ray diffraction (XRD) mineralogical analysis to visualize the rock  
168 framework and determine the mineral composition of the rock. The Bruker D8 x-ray diffractometer  
169 was used to make the XRD measurements. The model can examine samples in situ using  
170 noncorrosive gas environments from vacuum to 147 psi and up to 900°C in temperature. Ten rock  
171 samples (including seven sandstone cores from the pay zone and three shale cores from the cap  
172 rock) were characterized using XRD to ensure that the results are representative of the reservoir.  
173 Generally, the results of sandstone cores are very similar. Figs. 4a and 4b show the typical  
174 photomicrograph and mineral composition of a rock sample from the main sandstone in the

175 reservoir, respectively. The figure shows that quartz is the main mineral component (78 wt%) in  
 176 the oil-bearing sand. The quartz grains are poorly to moderately sorted, and most of them are  
 177 angular to subangular, with relatively sharp edges. The framework elements also include pebble-  
 178 size petrified wood fragments, and the matrix is mainly mudstone which constitutes about 13 wt%  
 179 of the rock. The clay mineralogy mainly includes smectite, illite, and mica, which makes the rock  
 180 more favorable for residual trapping of CO<sub>2</sub> because of the high capillary pressure caused by the  
 181 small pore throat sizes between the clay particles.



182

183 **Fig. 4.** Photomicrograph (a) and mineral composition (b) of a rock sample from the main sandstone  
 184 in the reservoir.

185

186 Eq. 1 shows the relation between capillary pressure and pore throat size (Ahmed, 2006):

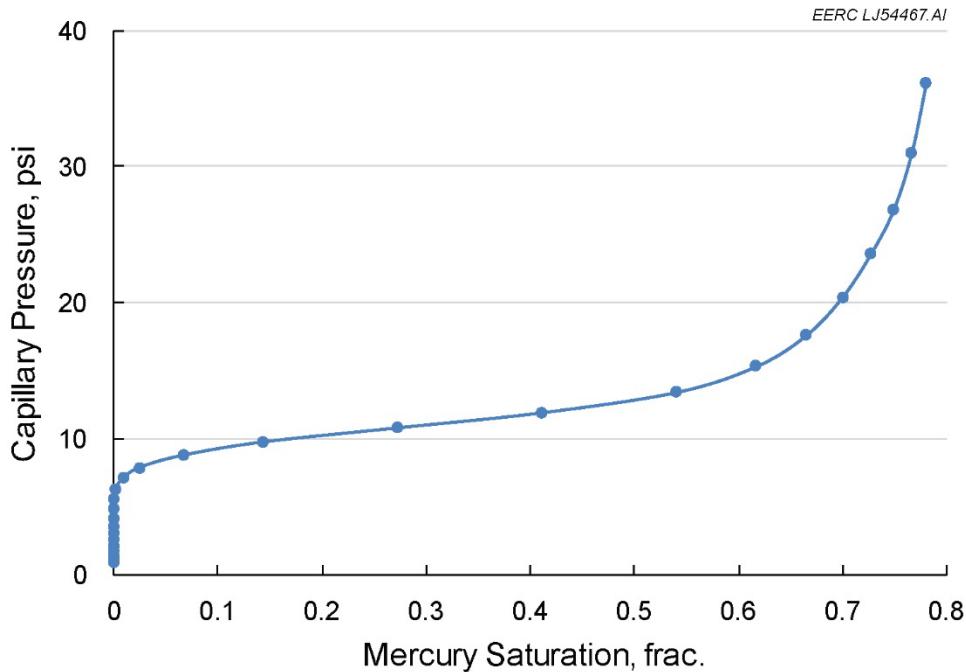
$$187 \quad P_c = \frac{2\sigma\cos\theta}{r} \quad [\text{Eq. 1}]$$

188 Where  $P_c$  is the capillary pressure, kPa;  $\sigma$  is the interfacial tension (IFT), dyne/cm;  $\theta$  is the contact  
 189 angle between two phases, degree; and  $r$  is the pore throat radius,  $\mu\text{m}$ .

190 The equation shows that tiny pore throats can generate a considerable capillary pressure  
 191 between phases when two or more fluids, i.e., CO<sub>2</sub>, oil, and/or water, coexist in the rock. Capillary  
 192 pressure curves were measured using a high-pressure mercury injection (HPMI) method for the

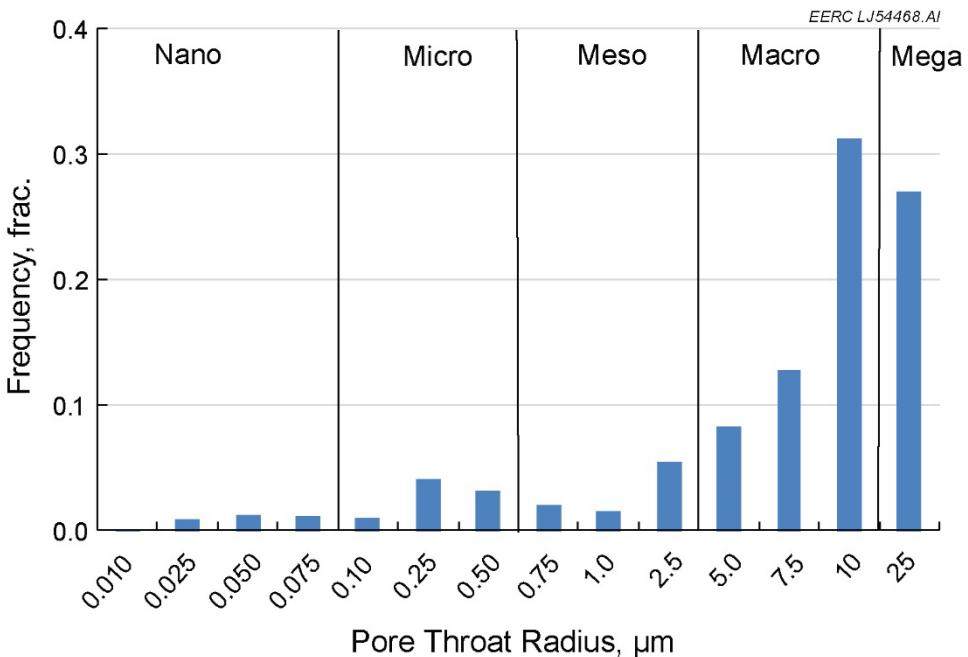
193 selected samples. A typical curve is shown in Fig. 5, which indicates that the capillary effect could  
194 be quite strong when the nonwetting phase enters the small pores.

195 PSD can be determined based on the rock–fluid properties and capillary pressure curve, as  
196 shown in Fig. 6. The figure indicates that most of the pores in the rock belong to macro- and  
197 megapores, which have a throat radius greater than 5  $\mu\text{m}$ . However, 20% of the pores have a throat  
198 radii of less than 2.5  $\mu\text{m}$ . These small pores may have effects on residual trapping of  $\text{CO}_2$   
199 associated with the flooding process. Relative permeability hysteresis curves provide a convenient  
200 way to evaluate these effects. The curves can be measured directly from experiments with  
201 sufficient data points or generated from empirical correlations by fitting them to limited data  
202 (Juanes et al., 2006; Land, 1968; Larsen and Skauge, 1998).



203  
204 **Fig. 5.** Capillary pressure curve of a rock sample from the main sandstone in the reservoir.  
205

206



207

208 **Fig. 6.** PSD of a rock sample from the main sandstone in the reservoir.  
209

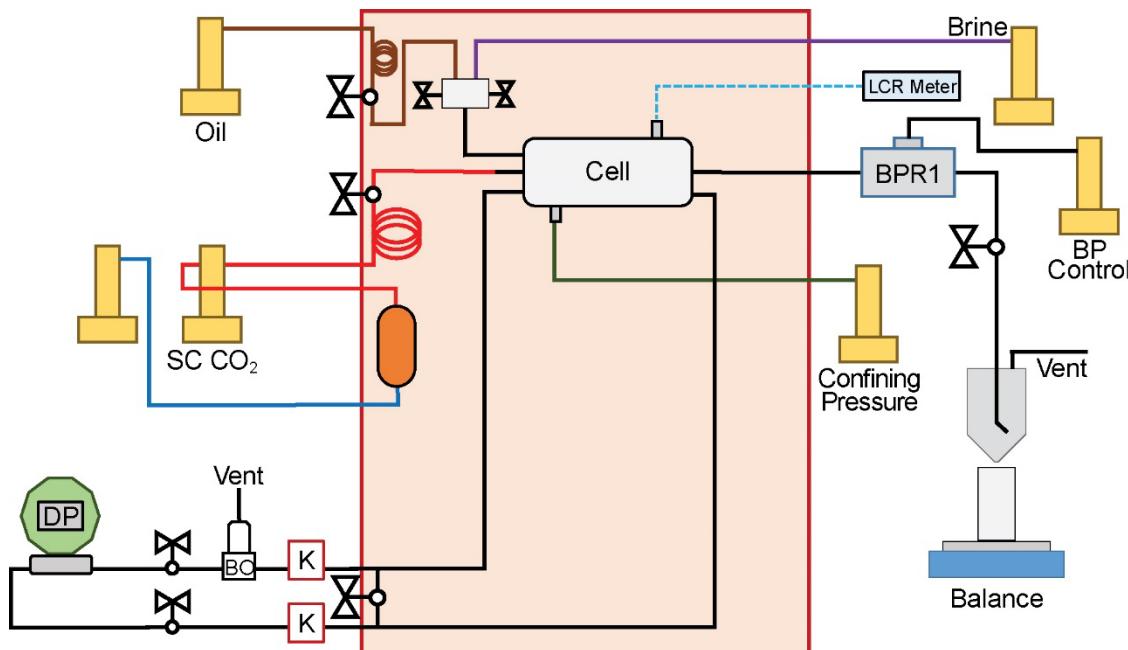
210 In this study, relative permeability hysteresis was measured using a clean sandstone core  
211 sample, collected from a monitoring well at a depth of 4533 ft. Table 1 contains the measured  
212 physical properties of the core sample and the oil used in the procedure. Based upon the reservoir  
213 and fluid properties, multicontact miscible flooding occurs in the Bell Creek oil field, therefore,  
214 the IFT between CO<sub>2</sub> and oil is minimized during the CO<sub>2</sub> EOR operations. However, the CO<sub>2</sub>-oil  
215 IFT may decrease gradually in the reservoir since CO<sub>2</sub> injection is conducted progressively across  
216 the field. The decreasing IFT is of great interest for miscible/near-miscible CO<sub>2</sub> EOR processes  
217 including WAG injection scenarios (Fatemi et al., 2012). Because gas-liquid relative  
218 permeabilities change with IFT, especially when IFT becomes low, it is necessary to allow enough  
219 contact between CO<sub>2</sub> and oil in the gas-liquid relative permeability hysteresis measurement  
220 process. Following the experimental procedure outlined by Fatemi et al. (2012) for low CO<sub>2</sub>-oil  
221 IFT conditions, CO<sub>2</sub> injection and oil injection were selected for the drainage and imbibition  
222 cycles, respectively. Steady-state relative permeability tests were performed using the

223 experimental setup shown in Fig. 7 to derive the relative permeability curves of the gas phase. The  
 224 experiments were conducted under reservoir conditions (2350 psi and 108°F for confining pressure  
 225 and temperature, respectively). The pressure profiles for the drainage (CO<sub>2</sub> injection) and  
 226 imbibition (oil injection) processes are shown in Fig. 8 and Fig. 9, respectively. The plots  
 227 demonstrate that steady-state displacement was in both drainage and imbibition processes after  
 228 1000 seconds.

229 **Table 1**  
 230 Physical Properties of the Core Used in Relative Permeability Hysteresis Measurements.

Parameter	Value	Unit
Diameter	0.97	in.
Length	1.91	in.
Weight	44.37	g
Grain Density	2.65	g/cm <sup>3</sup>
Porosity	0.26	fraction
Permeability	1052	md
Residual Oil Saturation	0.31	fraction

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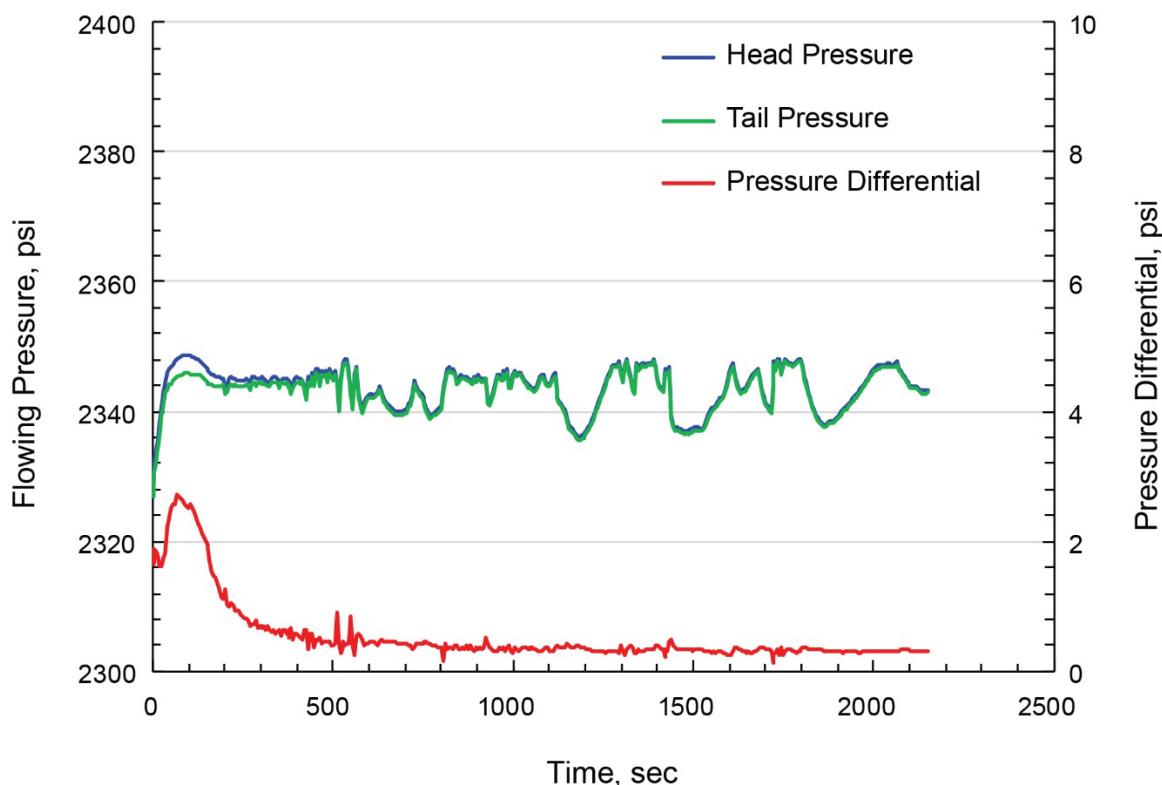


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 233 **Fig. 7.** Experimental setup for relative permeability hysteresis measurement.  
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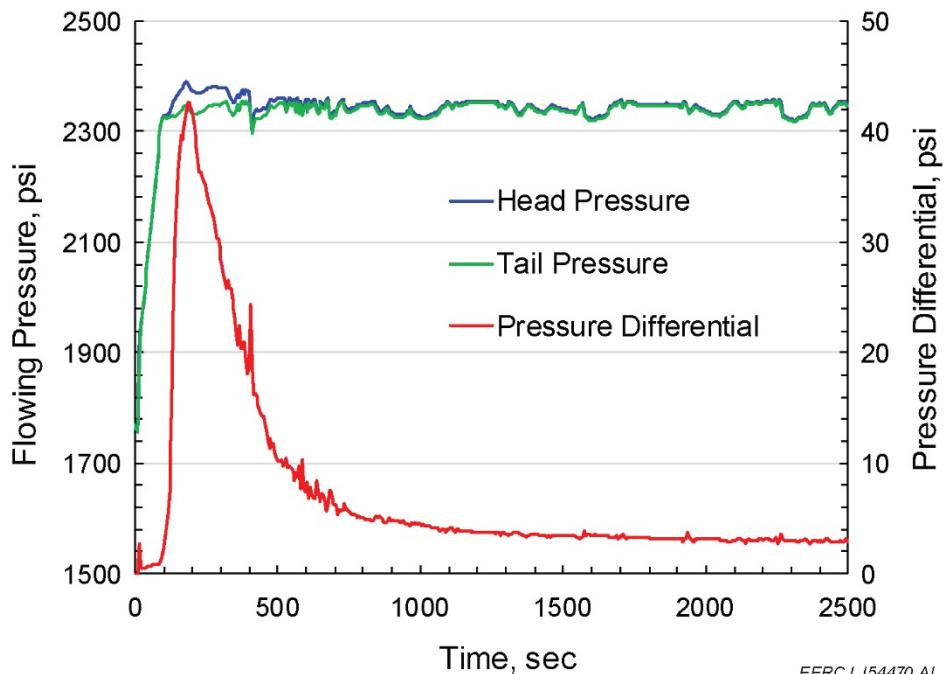
235 The measured relative permeability curves for the gas branch (Fig. 10) clearly show a  
236 hysteretic effect between the CO<sub>2</sub> relative permeability curves during drainage and imbibition  
237 processes. The irreducible (or trapped) gas saturation increases from 0.07 in the drainage process  
238 to 0.19 in the imbibition process, which means a considerable amount of CO<sub>2</sub> was trapped in the  
239 core sample during the cycle.

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241 **Fig. 8.** Pressure profile in the drainage process (CO<sub>2</sub> injection).  
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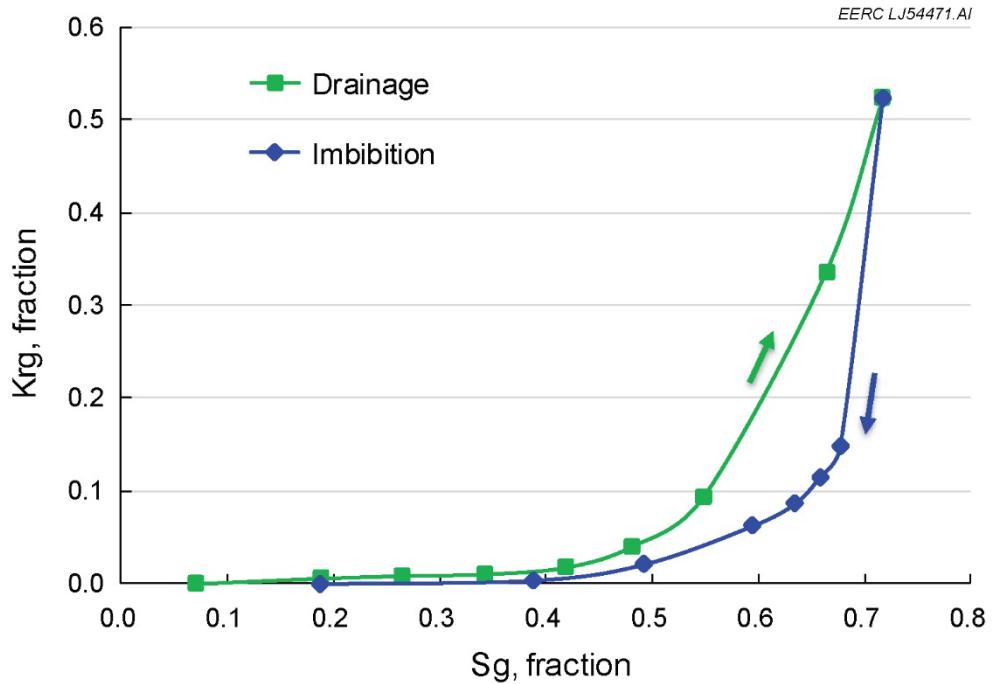


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244 **Fig. 9.** Pressure profile in the imbibition process (oil injection).

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246

247 **Fig. 10.** Relative permeability curves for  $\text{CO}_2$  in the drainage and imbibition processes showing a  
248 clear hysteretic effect.

249

250 **4. SOLUBILITY TRAPPING OF CO<sub>2</sub>**

251        Aside from residual trapping, solubility trapping of CO<sub>2</sub> is also critical for CO<sub>2</sub> EOR and  
252    associated storage. CO<sub>2</sub> dissolution in oil is one of the primary mechanisms for CO<sub>2</sub> EOR, in which  
253    dissolved CO<sub>2</sub> changes the oil's physical properties, yielding important benefits to recovery.  
254    Through this process, oil volume swells and viscosity reduces, which effectively increase the oil  
255    mobility and, thus, oil recovery (Emera and Sarma, 2007). However, the results of this process  
256    differ with changing pressure, oil composition, and impurities in the CO<sub>2</sub> stream (Srivastava et al.,  
257    1999). Another complication is posed by changing fluid saturations within the reservoir  
258    (decreasing oil saturation relative to water saturation). Within the oil phase specifically, the CO<sub>2</sub>  
259    EOR process preferentially mobilizes "lighter" hydrocarbon species (short-chain hydrocarbons) in  
260    comparison to "heavier" hydrocarbon species (long-chain hydrocarbons) (Hawthorne et al., 2014),  
261    resulting in changing oil composition over time. Therefore, fluid characterization and CO<sub>2</sub>  
262    solubility need to be studied carefully for a reasonable prediction of associated CO<sub>2</sub> storage which  
263    occurs during CO<sub>2</sub>-based EOR.

264        Detailed fluid characterization work has been conducted for the PCOR Partnership study oil  
265    using various PVT (pressure–volume–temperature) experiments at reservoir temperature  
266    (108°F), including saturation pressure, separator, constant composition expansion, differential  
267    liberation, and swelling tests. These tests accurately measured the oil/gas composition, saturation  
268    pressure, fluid density, viscosity, formation volume factor, and oil swelling with CO<sub>2</sub>, etc. Based  
269    on the experimental results, the physical properties of the reservoir fluids can be precisely  
270    characterized (Hawthorne et al., 2016). The original and residual oil compositions are shown in  
271    Table 2, where considerable medium hydrocarbons are left in the residual oil after pressure

272 depletion. The interactions between CO<sub>2</sub> and these hydrocarbons are important for CO<sub>2</sub> EOR and  
273 associated storage performance.

274 **Table 2**  
275 Composition of the Crude Oil in the Bell Creek Oil Field.

Oil Composition	Mole Fraction	
Component	Original	Residual
CO <sub>2</sub>	0.0042	0
N <sub>2</sub>	0.0019	0
CH <sub>4</sub>	0.1909	0
C <sub>2</sub> H <sub>6</sub>	0.0033	0.0009
C <sub>3</sub> H to NC <sub>4</sub>	0.0428	0.0370
IC <sub>5</sub> to C07	0.1526	0.1881
C08 to C13	0.2860	0.3606
C14 to C24	0.1997	0.2523
C25 to C36+	0.1184	0.1612

276

277 A series of swelling tests were performed by Core Laboratories Inc. to determine the  
278 interactions between CO<sub>2</sub> and oil, especially for CO<sub>2</sub> solubility and the oil-swelling factor, which  
279 is defined as the volume of fluid at current saturation pressure divided by the volume of reservoir  
280 oil at initial saturation pressure. CO<sub>2</sub> solubility in oil is affected by reservoir temperature, oil  
281 saturation pressure, and density, etc. The solubility generally decreases with temperature but  
282 increases with oil saturation pressure and density (Perera et al., 2016). Table 3 clearly illustrates  
283 the relationship between CO<sub>2</sub> solubility and key parameters of oil under reservoir conditions. The  
284 reservoir oil has a strong ability to dissolve CO<sub>2</sub>: 0.48 mole fraction of CO<sub>2</sub> can be dissolved in  
285 the oil when the oil saturation pressure increases to 1505 psi. Meanwhile, oil viscosity decreases  
286 from 2.22 to 0.86 and oil volume swells to 23%, respectively.

287 Both analytical and numerical correlations have been developed to predict the interactions  
288 between oil and CO<sub>2</sub> (Emera and Sarma, 2007; Mulliken and Sandler, 1980). Analytical  
289 correlations can be used to calculate the parameters quickly when the system is simple, while

290 numerical correlations are usually used with simulation models to compute the thermodynamic  
291 properties of fluids in complex systems.

292 **Table 3**  
293 Interactions Between CO<sub>2</sub> and Reservoir Oil at 108°F.

CO <sub>2</sub>	Oil			
Solubility, mol. frac.	Saturation Pressure, psi	Density, lb/ft <sup>3</sup>	Viscosity, cP	Swelling Factor, vol/vol
0.00	925	49.23	2.221	1.0000
0.18	1038	48.64	1.766	1.0668
0.33	1231	49.19	1.316	1.1377
0.48	1505	50.20	0.859	1.2301

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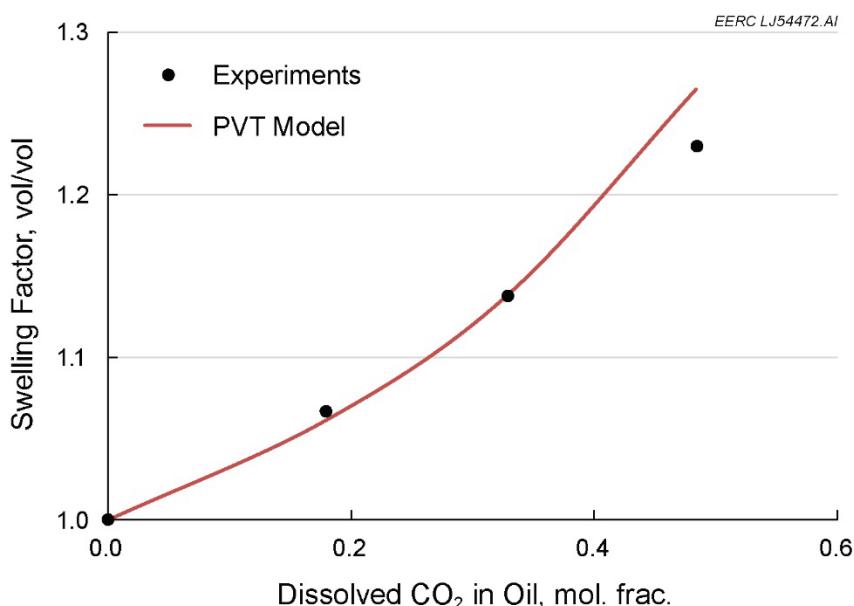
295 To couple with the complicated geologic conditions and strong heterogeneity in the  
296 reservoir, a numerical correlation (PVT model with cubic equation of state [EOS]) was developed  
297 to investigate the CO<sub>2</sub>–oil–water interactions in this study. The PVT model has seven components:  
298 CO<sub>2</sub> as a single component and other six components lumped together (N<sub>2</sub>–C2, C3–C4, C5–C7,  
299 C8–C13, C14–C24, and C25–C36). The Peng–Robinson (PR) EOS was applied to fine tune the  
300 model using Computer Modelling Group's (CMG's) WINPROP<sup>®</sup> module. Experimental data from  
301 saturation pressure, separator, constant composition expansion, differential liberation, and  
302 swelling tests were matched at reservoir temperature (108°F) to make sure the model can  
303 accurately predict the phase behavior of the reservoir fluids. Fig. 11 shows that the PVT model  
304 can capture the CO<sub>2</sub> solubility and oil-swelling behavior satisfactorily.

305 **5. CASE STUDY**

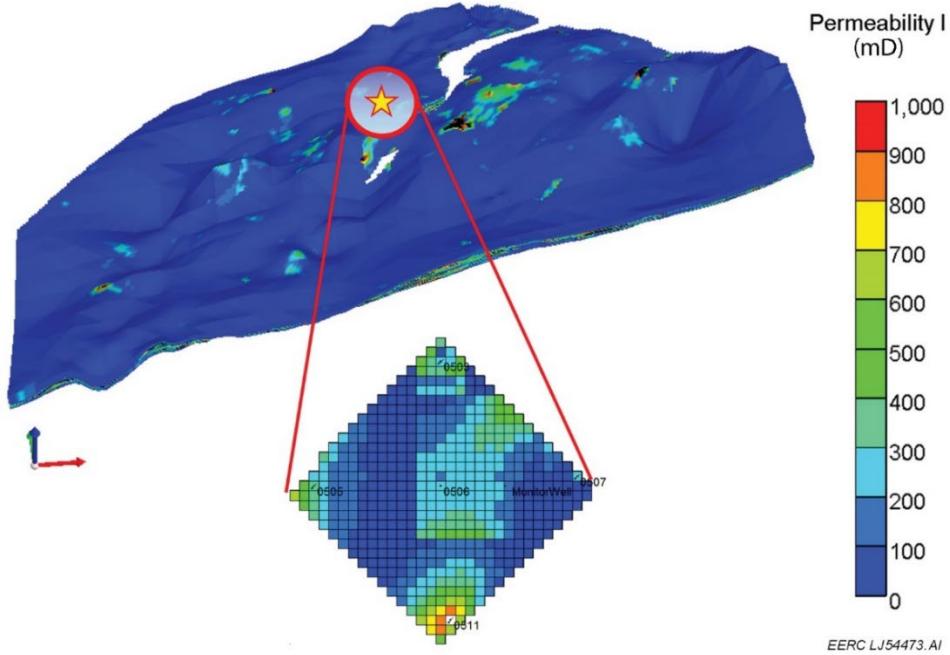
306 Reservoir simulation provides a useful means to predict fluid flow behavior in reservoirs  
307 with strong heterogeneity and complicated phase behavior. A large-scale simulation model with a  
308 total of 859,362 cells (259 × 158 × 21) and 102 wells was constructed to simulate the reservoir  
309 performance with CO<sub>2</sub> EOR operations. Satisfactory history-matching results through the primary  
310 production, waterflooding, and CO<sub>2</sub>-flooding stages showed that the model is able to capture the

311 flow dynamics in the reservoir. Details of the modeling and simulation work have been reported  
 312 by Jin et al. (2017b). Since relative permeability hysteresis requires greater simulation time and  
 313 sensitivity analysis of hysteretic effect requires a model with a fast running speed, a smaller five-  
 314 spot simulation model, as shown in Fig. 12, was clipped from the comprehensive reservoir-scale  
 315 model to investigate the effects of relative permeability hysteresis and solubility on associated CO<sub>2</sub>  
 316 storage performance. The five-spot model has five wells for fluid injection/production and also  
 317 keeps the original reservoir heterogeneity in the model. Its fast running speed makes it possible to  
 318 conduct sensitivity analysis on residual and solubility trapping effects efficiently.

319 Several relative permeability hysteresis models, including Land's trapping model, Carlson's  
 320 hysteresis model, and Killough's hysteresis model, are available to predict the effects of hysteresis  
 321 on oil recovery and associated storage (Fatemi et al., 2012; Land, 1968; Larsen and Skauge, 1998).  
 322 In this study, Land's model was used in simulation cases to evaluate the effect of hysteresis on  
 323 CO<sub>2</sub> flood performance. Based on the measured CO<sub>2</sub> relative permeability hysteresis curves shown  
 324 in Fig. 10, three different residual CO<sub>2</sub> saturations (0.1, 0.2, and 0.3) were considered in the study



325  
 326  
 327 **Fig. 11.** Correlation between CO<sub>2</sub> solubility and oil-swelling factor.

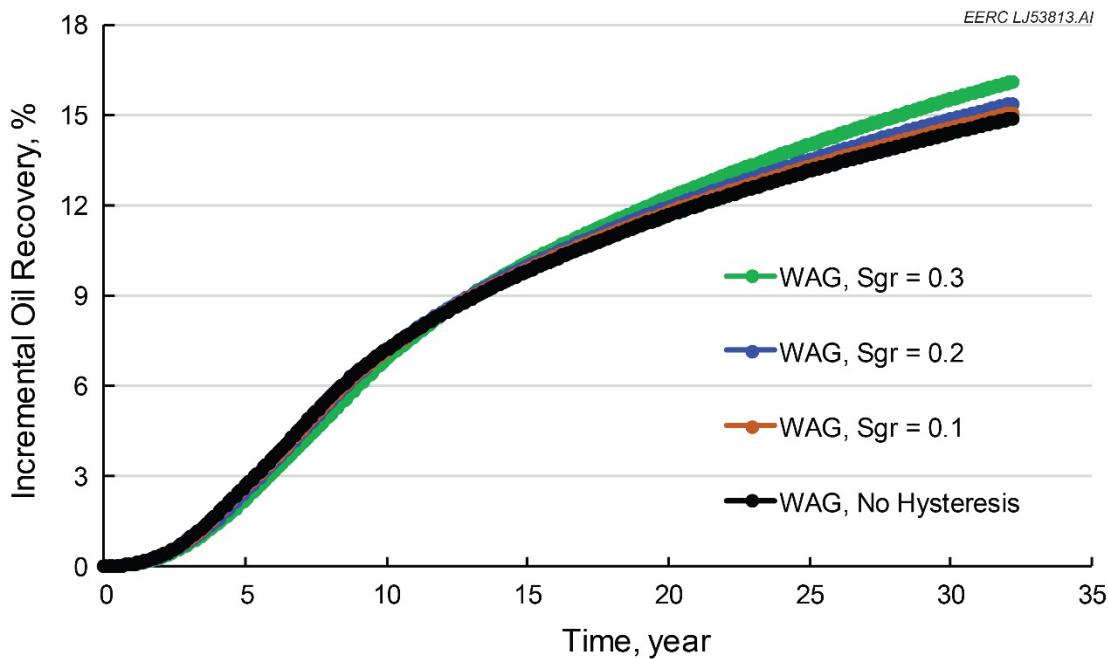


328

329 **Fig. 12.** Simulation model with different scales for the Bell Creek oil field.

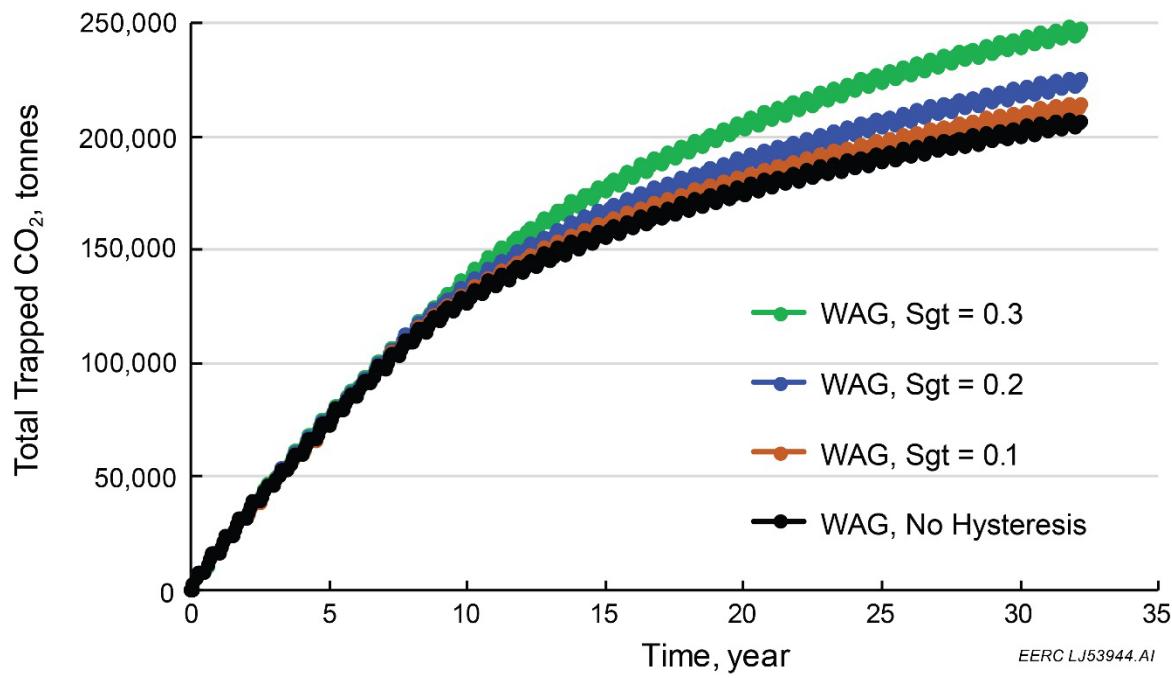
330

331 to span a range of possible CO<sub>2</sub>-trapping scenarios. Fig. 13 shows the comparison of incremental  
 332 oil recovery for the five-spot model CO<sub>2</sub> EOR simulation cases with and without relative  
 333 permeability hysteresis. The results indicate relative permeability hysteresis does not have a  
 334 significant impact on oil recovery in this model. Oil recovery factor is slightly higher when the  
 335 residual CO<sub>2</sub> saturation is 0.3, but the difference is negligible between other cases. However, the  
 336 effect of relative permeability hysteresis on total associated CO<sub>2</sub> storage is obvious, as shown in  
 337 Fig. 14. More CO<sub>2</sub> is stored in the reservoir when residual CO<sub>2</sub> saturation is high, as is expected.  
 338 Quantitatively, a difference of approximately 20% of total trapped CO<sub>2</sub> was noted between a case  
 339 without hysteresis (referred as base case hereinafter) applied and a case with hysteresis applied  
 340 and an assumed residual CO<sub>2</sub> saturation of 0.3 as indicated in Fig. 15. Fig. 16 clearly shows that  
 341 the residually trapped CO<sub>2</sub> has increased over 220% for the case of 0.3 residual gas saturation  
 342 compared to the base case.



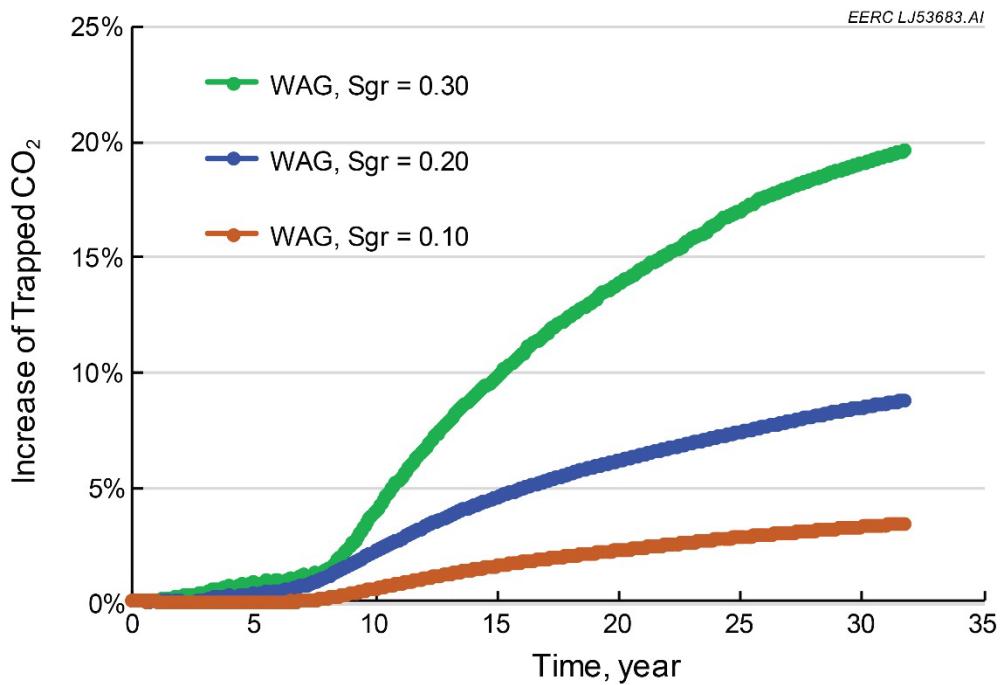
343

344 **Fig. 13.** Comparison of incremental oil recovery for cases with different residual CO<sub>2</sub> saturations.  
 345



346

347 **Fig. 14.** Comparison of total CO<sub>2</sub> trapped for cases with different residual CO<sub>2</sub> saturations.  
 348

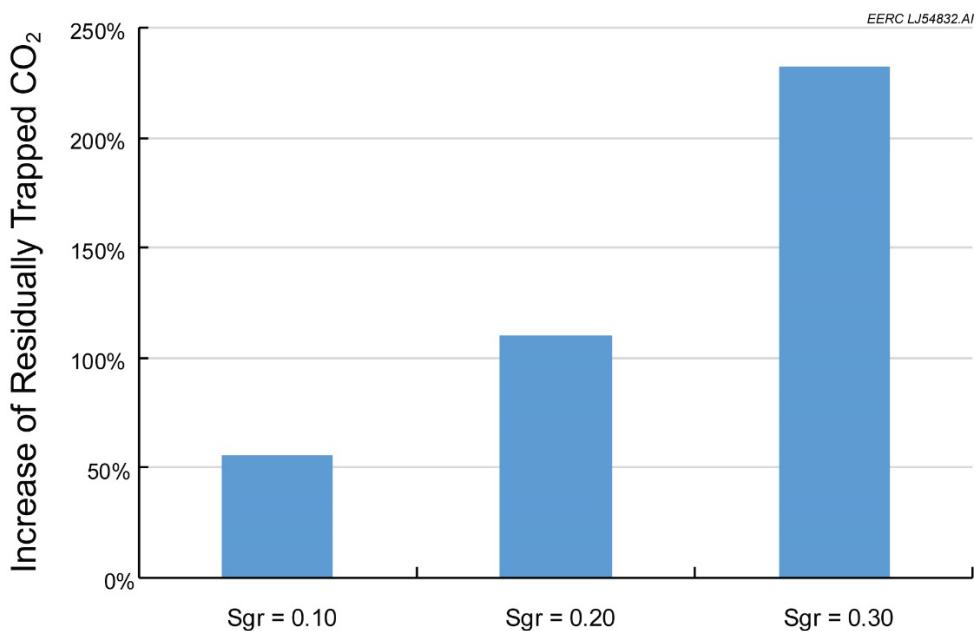


352

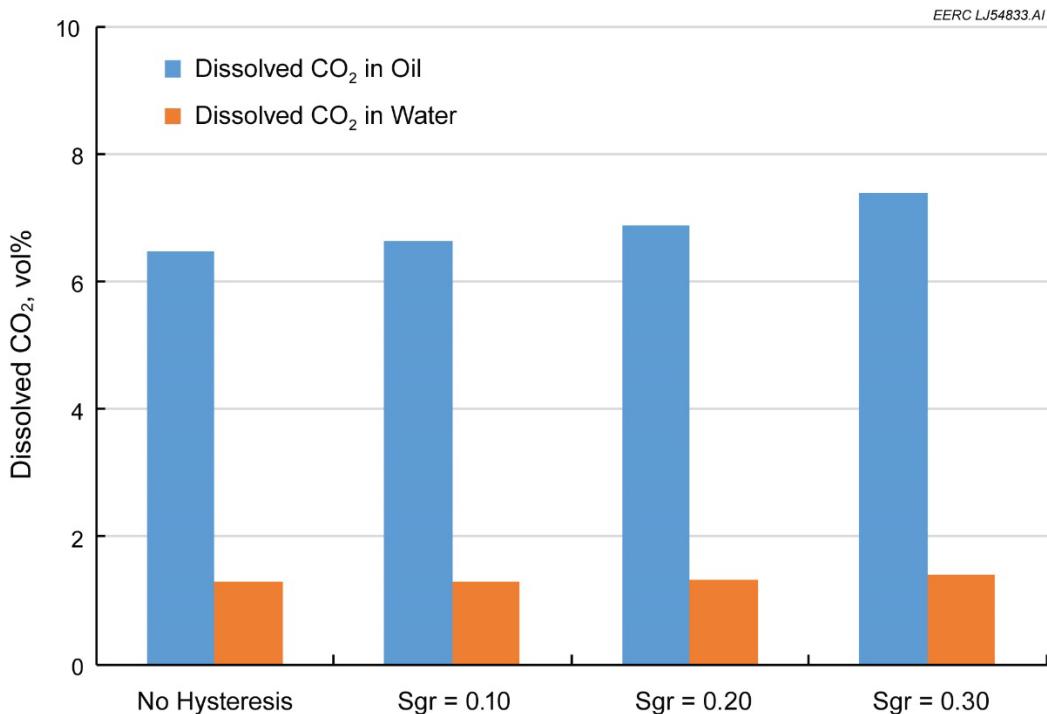
353 **Fig. 16.** Increase of residually trapped  $\text{CO}_2$  compared to the case without hysteresis effect.

354

355



356       Simulations accounting for CO<sub>2</sub> dissolution in water and oil were conducted together with  
357   the hysteresis cases since the solubility correlation has been included in the simulation model. The  
358   results of cases assuming different residual CO<sub>2</sub> saturations are shown in Fig. 17. The results  
359   indicate CO<sub>2</sub> solubility in oil is much greater ( $\geq 5$  times) than that of water in the pore volume. A  
360   part of residual oil after waterflooding (S<sub>orw</sub>) is moved by CO<sub>2</sub> and becomes movable oil (S<sub>om</sub> =  
361   S<sub>orw</sub> – S<sub>orm</sub>) in the reservoir. This movable oil is then produced to the surface via oil producers,  
362   and the CO<sub>2</sub> in the produced oil is separated and continually recycled, i.e., reinjected into the  
363   reservoir. However, not all of the movable oil is produced to the surface because of the limitation  
364   of producing time (32 years in this study). CO<sub>2</sub> continues to interact with the remaining residual  
365   oil (S<sub>orm</sub>), dissolving into it and being trapped there after EOR operations cease. As such, the  
366   higher the S<sub>orm</sub>, the greater the CO<sub>2</sub> trapping potential in residual oil. Residual oil saturation after  
367   waterflooding is usually 0.3 or greater in most conventional oil reservoirs, and a considerable  
368   quantity of residual oil still remains in the pore space after CO<sub>2</sub> EOR operations. Thus these oil  
369   reservoirs could be great candidates for CO<sub>2</sub> storage. The simulation results also show that more  
370   CO<sub>2</sub> is dissolved when the trapped CO<sub>2</sub> saturation is higher, as more CO<sub>2</sub> is available to interact  
371   with oil and water in the pore space. [The dissolved CO<sub>2</sub> in water under different residual gas](#)  
372   saturations is also different, but the difference is too small to distinguish in Fig. 17 as the quantities  
373   of dissolved CO<sub>2</sub> are 20045, 20278, 20829, and 21871 tonnes for the cases displayed from left to  
374   right in the figure, respectively.



375

376 **Fig. 17.** Comparison of simulated dissolved CO<sub>2</sub> for cases with different residual CO<sub>2</sub> saturations.  
377

378 **6. CONCLUSION**

379 Large-scale CO<sub>2</sub>-flooding operations are in progress in the Bell Creek oil field. Encouraging  
380 oil production performance shows the success of the EOR project. Simultaneously, a considerable  
381 quantity of associated CO<sub>2</sub> storage has occurred in the reservoir. In this study, CO<sub>2</sub>-trapping  
382 mechanisms in the reservoir associated with EOR operations were analyzed. Two of the primary  
383 CO<sub>2</sub>-trapping mechanisms responsible for associated CO<sub>2</sub> storage in the Bell Creek oil field,  
384 residual trapping and solubility trapping, were discussed in detail. The main findings include the  
385 following:

386 1. Production and injection data were analyzed in the CO<sub>2</sub>-flooding stage. Despite the  
387 continued improvement of oil production during flooding, the rate of associated CO<sub>2</sub>  
388 storage decreased after 2 years. The results indicate the flow network/channels for CO<sub>2</sub>  
389 has been well established between injectors and producers in the reservoir.

- 390 2. Over 50 core plugs were collected from the reservoir to characterize rock properties.
- 391 Mineralogical analysis and capillary pressure measurements showed that the mineral
- 392 composition and PSD in the reservoir are favorable for both CO<sub>2</sub> EOR and associated
- 393 storage.
- 394 3. The reservoir oil was characterized based on PVT experiments and PR EOS modeling.
- 395 Results showed that the reservoir oil has a strong ability to dissolve CO<sub>2</sub>, which not only
- 396 improves the mobility of residual oil in the reservoir, but also traps a considerable
- 397 quantity of CO<sub>2</sub> in the reservoir – [over 100 thousand tonnes in a five-spot pattern reservoir](#)
- 398 [section under study.](#)
- 399 4. Steady-state relative permeability tests were performed to derive gas-phase relative
- 400 permeability curves using a clean sandstone core sample collected from a monitoring well
- 401 in the reservoir. [The irreducible \(or trapped\) gas saturation increases from 0.07 in the](#)
- 402 [drainage process to 0.19 in the imbibition process due to the relative permeability](#)
- 403 [hysteresis effects.](#)
- 404 5. The relative permeability hysteresis curves were integrated within a five-spot simulation
- 405 model to investigate the effect of residual trapping on CO<sub>2</sub> EOR and storage performance.
- 406 Results showed that oil recovery factor and associated CO<sub>2</sub> storage could increase 1.21%
- 407 and 20%, respectively, considering relative permeability hysteresis with a residual CO<sub>2</sub>
- 408 saturation of 0.3.
- 409 6. The five-spot simulation model was also used to investigate solubility trapping of CO<sub>2</sub> in
- 410 the reservoir. Based on the fluid properties and reservoir conditions in the Bell Creek oil
- 411 field, CO<sub>2</sub> solubility in oil is much greater ( $\geq 5$  times) than that in water.

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419 possible.

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428 by the United States Government or any agency thereof. The views and opinions of authors  
429 expressed herein do not necessarily state or reflect those of the United States Government or any  
430 agency thereof.

431 **CONVERSIONS**

$C$ :	compressibility, $psi^{-1}$	$\times 0.145 \text{ kPa}^{-1}$
$d$ :	diameter, $in$	$\times 0.0254 \text{ m}$
$k$ :	permeability, $mD$	$\times 10^{-12} \text{ m}^2$
$l$ :	length, $in$	$\times 0.0254 \text{ m}$
$m$ :	weight, $g$	$\times 10^{-3} \text{ kg}$
$p_b$ :	saturation pressure, $psi$	$\times 6.895 \text{ kPa}$

$p_c$ :	capillary pressure, <i>psi</i>	× 6.895 <i>kPa</i>
$p_e$ :	reservoir pressure, <i>psi</i>	× 6.895 <i>kPa</i>
$q$ :	liquid production rate, <i>bpd</i>	× $6.625 \times 10^{-3} \text{ m}^3/\text{hr}$
$Q$ :	cumulative liquid production, <i>bbl</i>	× 0.159 $\text{m}^3$
$r$ :	pore throat radius, $\mu\text{m}$	× $10^{-6} \text{ m}$
$S_{om}$	movable oil saturation	× 1 <i>fraction</i>
$S_{orm}$	residual oil saturation after CO <sub>2</sub> flooding	× 1 <i>fraction</i>
$S_{orw}$	residual oil saturation after water flooding	× 1 <i>fraction</i>
$T$ :	temperature, °F	= $([\text{°F}] + 459.67) \times \frac{5}{9} \text{ K}$
$FVF$ :	formation volume factor, <i>rb/stb</i>	× 1 $\text{m}^3/\text{sm}^3$
$GOR$ :	gas-oil ratio, <i>scf/stb</i>	× 0.178 $\text{m}^3/\text{m}^3$
$\rho$ :	density, <i>lb/ft</i> <sup>3</sup>	× 16.02 $\text{kg/m}^3$
$\sigma$ :	interfacial tension, <i>dyne/cm</i>	× 1 <i>dyne/cm</i>
$\theta$ :	contact angle between two phases, <i>degree</i>	× 1 <i>degree</i>
$\phi$ :	porosity, <i>fraction</i>	× 1 <i>fraction</i>
$\mu$ :	viscosity, <i>cP</i>	× $10^{-3} \text{ Pa}\cdot\text{s}$

432

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