

Theoretical analysis of QOOH combustion reaction pathways

Madison D. Fellows, Judit Zádor

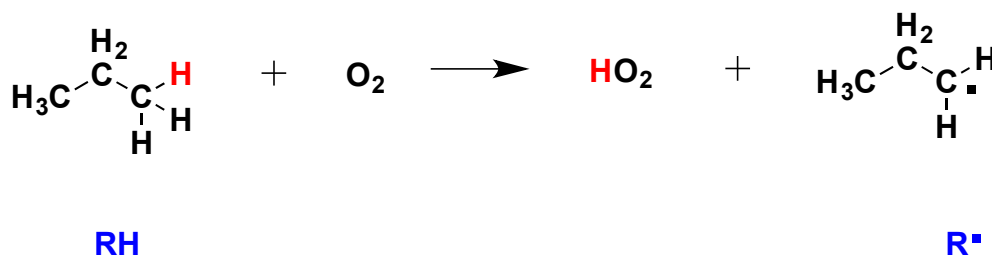
Abstract

QOOH radicals are key intermediates in the chain of reactions leading to the autoignition of hydrocarbons and oxygenated organic compounds. They are thought to undergo two main reactions: OH elimination to form a cyclic ether and HO₂ elimination to form an alkene. However, theoretical analysis of various substituted hydroperoxyalkyl radicals has found two new pathways: OH transfer and internal H abstraction assisted OH elimination. To determine the importance of these new pathways, their barrier heights for several substituted alkanes were calculated using various quantum chemical theories and compared to those of the well-known pathways. Several cases revealed possible competition with the well-known pathways. Rate coefficients were calculated for propyl systems but further studies will need to complete rate coefficients and branching fractions for all systems analyzed to understand these new reactions' role in autoignition.

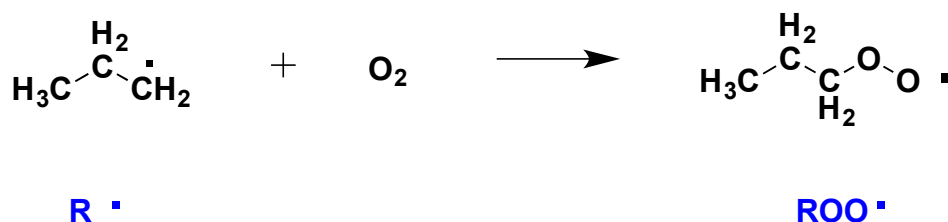
Introduction

The need to increase engine efficiencies and lower unwanted emissions has resulted in recent attention on low-temperature hydrocarbon oxidation.¹⁻⁷ Engines employing low-temperature hydrocarbon oxidation (500-1900 K) utilize lower fuel concentrations compared to gasoline and diesel engines, effectively reducing soot formation, while the lower temperatures eliminate nitrogen oxides that are produced above 2000 K in gasoline and diesel engines.⁸ Rather than spark ignition, these low-temperature hydrocarbon oxidation engines rely on autoignition prompted by compression.

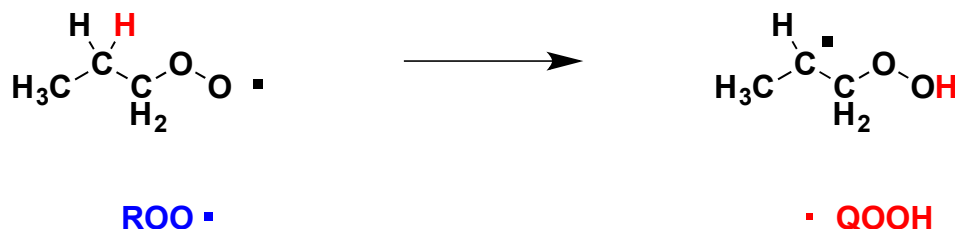
Autoignition chemistry is driven by chemical chain propagation. The initial reaction proceeds via formation of an alkyl radical (R) from an alkane fuel molecule (RH) and oxygen



From here, R reacts with oxygen to form an alkylperoxy radical (ROO)



While ROO can undergo direct elimination to form an alkene + HO₂, it can also isomerize to form QOOH, a hydroperoxy alkyl radical



QOOH is a very important species because it is very unstable, and therefore very reactive. For this reason, QOOH formation leads to chain propagation and chain branching, which are ultimately responsible for autoignition. However, these steps comprise less than half of what is accepted as the autoignition pathway (Figure 1). From here, these reactions get very complicated.

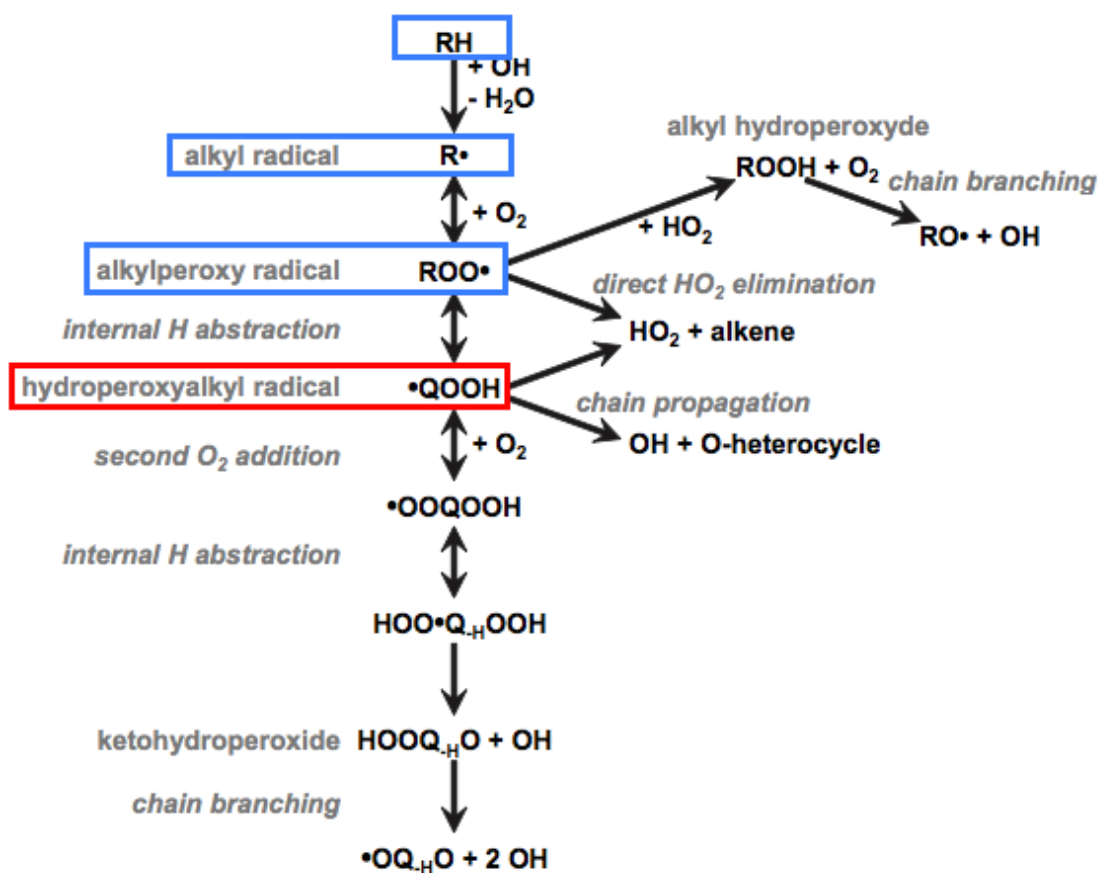


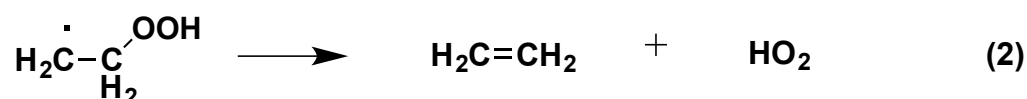
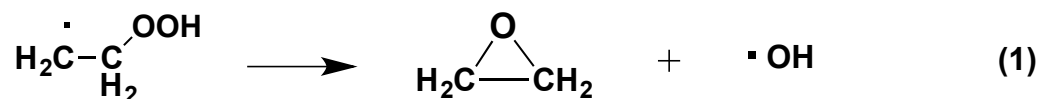
Figure 1. Schematic mechanism for low-temperature hydrocarbon oxidation and autoignition chemistry. Figure adopted from Zádor, Taatjes, and Fernandes 2011.⁷

In order to optimize new fuel or engine models, these reactions must be fully explored. These elementary reactions, however, are very hard to observe experimentally due to fast propagation—making it easy to miss important pathways or short-lived

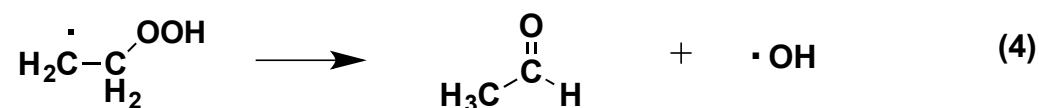
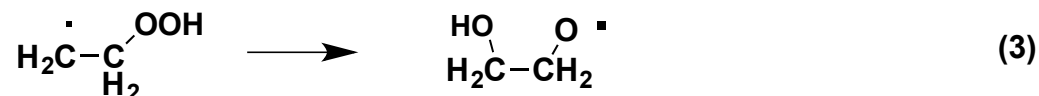
intermediates. In this sense, theoretical computations can help characterize these complicated pathways.

KinBot is a software that can predict gas-phase combustion reactions—allowing the modeling of several reaction pathways in much less time and with fewer holes that may be found experimentally or using less efficient computational methods.

As an overview, QOOH radicals are generally thought to undergo OH elimination to form a cyclic ether (1) or HO₂ elimination to form an alkene (2).



However, KinBot found two new reactions that have not yet been observed or investigated computationally—OH transfer (3) and internal H abstraction assisted OH elimination (4).



While DeSain et al.², Miller et al.⁶, and Goldsmith et al.³ report direct aldehyde + OH formation from RO₂, no studies have observed these products originating from QOOH as a reactant. Both the OH and H transfer found have the potential to contribute to autoignition. To understand whether these findings are specific to just certain QOOH radicals and how the energy of these new pathways compares to the already known ones, the barrier heights for several substituted alkanes were calculated using various quantum chemical theories.

Though barrier heights can suggest a lot about a given reaction, kinetics are required to understand the entire scope of a reaction. Both OH and H transfer reaction pathways introduced generate chain-propagating products that may speed up the autoignition process. To optimize fuel and engine models, the rate of these reactions must be known. As propane has received considerable attention as a possible prototype for low-temperature hydrocarbon oxidation,¹ rate coefficients were calculated for propyl systems undergoing OH and H transfer to better contribute to the prototype already in place.

Methods

To determine the barrier height for various substituted alkanes (Figure 2), saddle points were located using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), and CBS-CB3 quantum chemical theories using the Gaussian 09 program package. The saddle points were proven to connect the proposed reactants and products by following the pathways along the intrinsic reaction coordinates (IRC). Besides the above methods, highly accurate energies were calculated at the UCCSD(T)-F12/cc-pVnZ//M06-2X/6-311++G(d,p) level of theory for each species, where $n = T$ or Q .

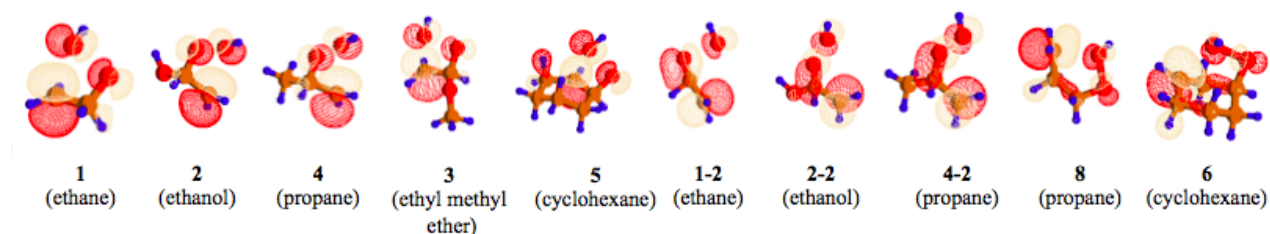


Figure 2. Geometries and highest occupied molecular orbitals (HOMOs) of tested transition states for various QOOH species. Species denoted with ‘-2’ undergo H-transfer (reaction 4), the others OH-transfer (reaction 3).

Species 4, 4-2, and 8 were investigated further, in addition to species 9 ($\text{CH}_3\text{CHCH}_2\text{OOH}$ undergoing OH transfer) and 9-22 ($\text{CH}_3\text{CHCH}_2\text{OOH}$ undergoing H transfer) to develop a greater understanding of the OH and H transfer reaction pathways for propyl systems. Lowest energy conformers for 4reactant/4-2reactant, 4-2product and 8reactant were adapted from Goldsmith et al.³ (QOOH-3, acetone + OH, and QOOH-1, respectively) and studied using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), CBS-CB3, and UCCSD(T)-F12/cc-pVQZ//M06-2X/6-311++G(d,p) levels of theory. The lowest energy conformer for 9reactant and 9-22reactant was found to be the 9-22reactant conformer generated in this work, as it was lower in energy than its respective QOOH-2 conformer from Goldsmith et al.³ The energy difference at the M06-2X level of theory was 0.486 kcal/mol. For transition states and remaining products not discussed in Goldsmith et al.³, 3^n possible conformers were generated for each species, where n is the number of dihedral angles, and optimized using M06-2X/6-311++G(d,p) level of theory. Lowest energy conformers were selected and reoptimized using the above methods. The potential energy surfaces for propyl systems undergoing OH and H transfer were similarly compared to those undergoing the well-known cyclic ether and alkene formation pathways.

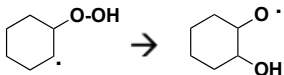
Hindered rotor energy scans were calculated for each propyl product using KinBot/Gaussian, whereas the rotors for propyl transition states were generated manually and run in Gaussian. These rotor energies, along with geometries and frequencies optimized at the M06-2X/6-311++G(d,p) level of theory for each transition state were incorporated into the propane master-equation provided by Michael P. Burke (Columbia University) to calculate temperature and pressure dependent rate coefficients. The master equation is equivalent to the one published by Burke et al.³ 2015. The master equation was run using the PAPER code of Argonne National Laboratory.

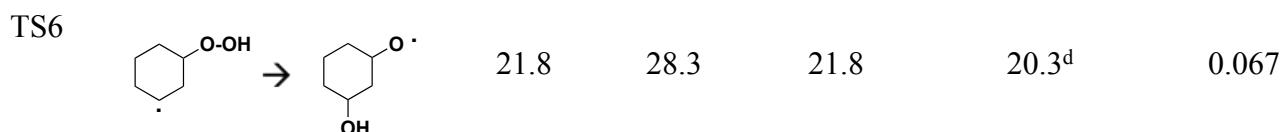
Results and Discussion

Energies for new OH and H transfer reaction pathways were calculated for ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems using B3LYP/6-311++G(d,p), M06-2X/6-

311++G(d,p), CBS-CB3, and UCCSD(T)-F12/cc-pVnZ//M06-2X/6-311++G(d,p) levels of theory ($n = D, T$ or Q). Energies for transition states and products are listed in Table 1 and 2, respectively.

Table 1. Transition state energies for ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems.

Species	Reaction	Relative energy ^a				T1 Diagnostic
		B3LYP ^b	M06-2X ^b	CBS-QB3	UCCSD(T)-F12	
TS1	CH ₂ CH ₂ OOH → CH ₂ (OH)CH ₂ O	24.6	37.8	30.7	30.1 ^c	0.068
TS1-2	CH ₂ CH ₂ OOH → CH ₃ CHO + OH	21.3	32.6	30.7	28.8 ^d	0.071
TS2	CH ₂ CH(OOH)OH → CH ₂ (OH)CH(O)OH	23.8	36.7	25.2	28.4 ^e	0.068
TS2-2	CH ₂ CH(OOH)OH → CH ₃ COOH + OH	17.9	30.0		26.0 ^d	0.066
TS3	CH ₂ CH(OOH)OCH ₃ → CH ₂ (OH)CH(O)OCH ₃	27.0	39.8	26.2	31.0 ^e	0.055
TS5		20.1	31.6	21.2	20.4 ^d	0.088



^aAll energies are relative to respective reactant in kcal/mol.

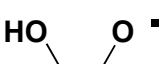
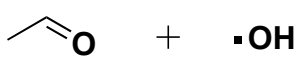
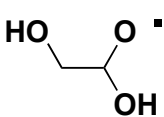
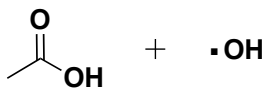
^bPresent work employing 6-311++G(d,p) basis set.

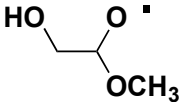
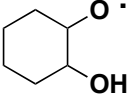
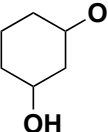
^cPresent work employing cc-pVQZ//M06-2X/6-311++G(d,p) basis set and geometry.

^dPresent work employing cc-pVDZ//M06-2X/6-311++G(d,p) basis set and geometry.

^ePresent work employing cc-pVTZ//M06-2X/6-311++G(d,p) basis set and geometry.

Table 2. Product energies for ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems.

Species	Structure	Relative Energy ^a	
		B3LYP ^b	M06-2X ^b
1product		-51.0	-52.1
1-2product		-50.8 ^c	-49.4
2product		-54.2	-54.3
2-2product		-70.1 ^c	-67.7

3product		-57.0	-56.0
5product		-49.7	-51.3
6product		-49.4	-57.6

^aAll energies are relative to respective reactant in kcal/mol.

^bPresent work employing 6-311++G(d,p) basis set.

^cProducts not calculated separately.

Overall, the transition state barrier for H transfer reaction pathways were lower in energy than those undergoing OH transfer for the above systems, while the latter reaction pathways were more exothermic. The new reaction pathways for ethyl and cyclohexyl systems were compared to the more common reaction pathways (Figure 3 and 4).

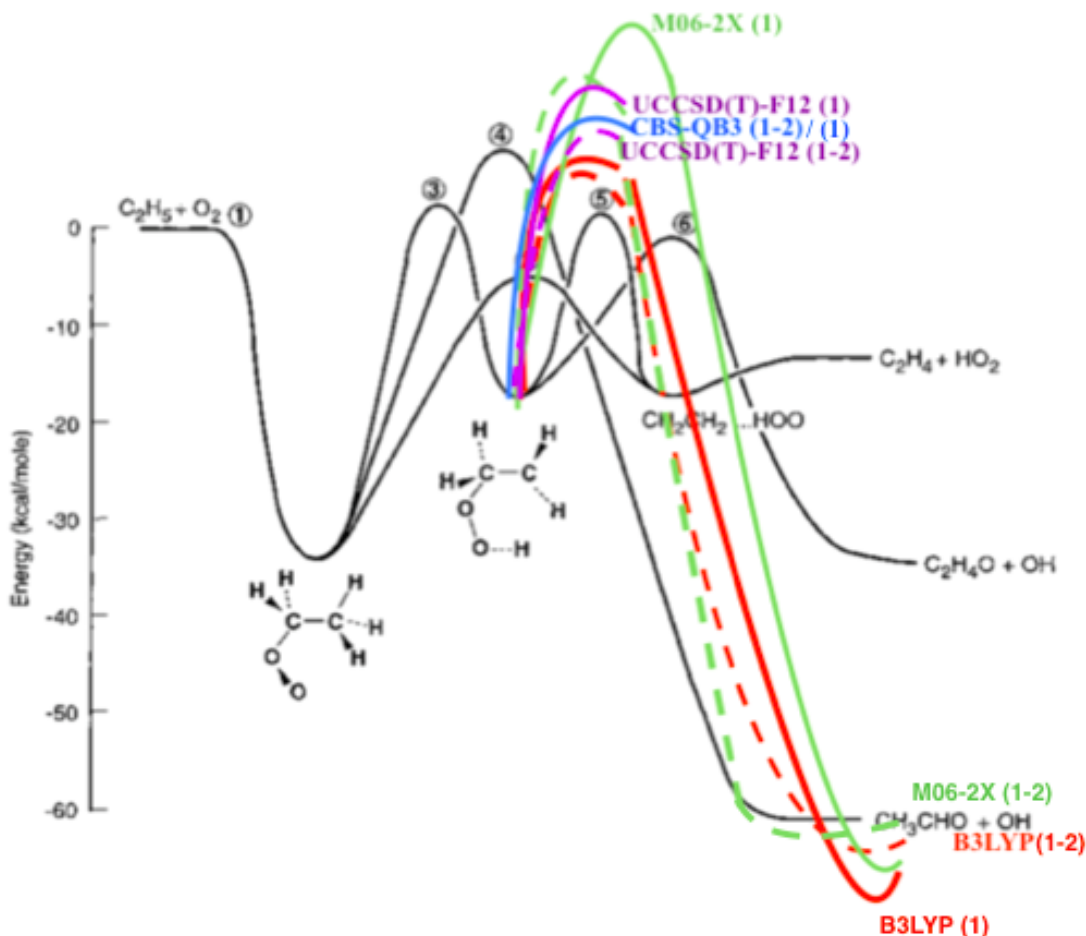


Figure 3. Schematic potential energy surfaces comparing calculated energy barriers of new OH transfer (1) and H transfer (1-2) pathways and those of common HO₂ and OH elimination pathways from CH₂CH₂OOH (1 and 1-2) using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), and UCCSD(T)-F12 levels of theory (barriers for 1 and 1-2 were calculated at cc-pVQZ and cc-pVDZ basis sets, respectively). The drawn barrier heights are given relative to the ethyl + O₂ entrance channel: TS1_{B3LYP} = 7.6, TS1_{M06-2X} = 20.8, TS1_{CBS-QB3} = 13.7, TS1_{UCCSD} = 13.0; TS1-2_{B3LYP} = 4.3, TS1-2_{M06-2X} = 15.6, TS1-2_{CBS-QB3} = 13.7, and TS1-2_{UCCSD} = 11.8. 1-product_{B3LYP} = -68.0, 1-product_{M06-2X} = -69.1; 1-2-product_{B3LYP} = -67.8, 1-2-product_{M06-2X} = -66.4. Base figure adapted from Miller, Klippenstein, and Robertson 2000.⁶

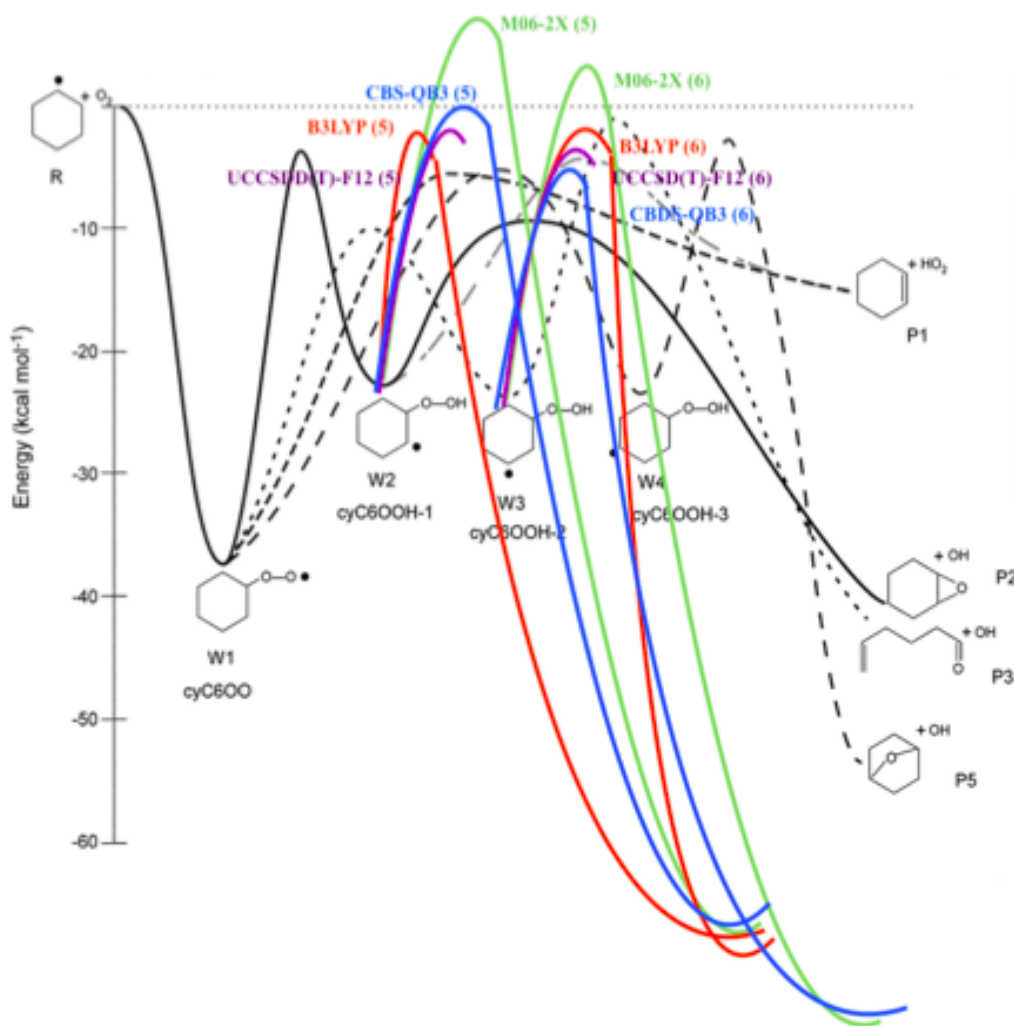


Figure 4. Schematic potential energy surfaces comparing calculated energy barriers of new OH transfer pathway and those of common HO₂ and OH elimination pathways from cyC₆OOH-1 (5) and cyC₆OOH-2 (6) using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), CBS-CB3, and UCCSD(T)-F12/cc-pVDZ//M06-2X/6-311++G(d,p) levels of theory. Drawn barriers represent E + ZPE values in kcal/mol. The barrier heights relative to the cyclohexyl + O₂ entrance channel: Ts5_{B3LYP} = -1.9, Ts5_{M06-2X} = 9.6, Ts5_{CBS-QB3} = -0.8, Ts5_{UCCSD} = -1.6; Ts6_{B3LYP} = -1.4, Ts6_{M06-2X} = 5.1, Ts6_{CBS-QB3} = -1.4, and Ts6_{UCCSD} = -2.9. 5product_{B3LYP} = -71.7, 5product_{M06-2X} = -73.3; 6product_{B3LYP} = -72.6, 6product_{M06-2X} = -80.8. Base figure adapted from Knepp et al.⁵

While the OH transfer and H transfer reaction pathways were much higher in energy compared to the well-known reaction pathways for the ethyl system (1 and 1-2) and TS5, there is a good comparison between those of TS6 and the well-known pathways. However, both OH transfers for TS5 and TS6 are extremely exothermic, suggesting the likelihood of chain propagation. The product of TS6 is likely to dissociate into P3 on the cyclohexyl PES (Figure 4), while we have not investigated the likely dissociation pathways for the product of TS5.

As a general observation, the OH transfer is competitive in γ -QOOH radicals likely because the OH transfer involves a 5-member ring TS, which has significantly less ring strain than the related 4-membered-ring cyclic ether formation pathway in this case.

All of the energies at the saddle points exhibit unusually large T1 diagnostic values (meaning high multi-reference character), which means that the barrier heights are more uncertain than usual.

Propyl system:

Energies for new OH and H transfer reaction pathways were calculated for propyl systems using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), CBS-CB3, and UCCSD(T)-F12/cc-pVT//M06-2X/6-311++G(d,p) levels of theory. Energies for transition states and products are listed in Table 3 and 4, respectively, relative to the lowest energy conformer.

Table 3. Transition state energies for propyl systems. Note that the H-transfer equivalent for pathway TS8 was not found.

Species	Reaction	Relative energy ^a				T1 diagnostic
		B3LYP ^b	M06-2X ^b	CBS-QB3	UCCSD(T)-F12	
TS4	$\text{CH}_3\text{CH}(\text{CH}_2)\text{OOH} \rightarrow \text{CH}_2(\text{OH})\text{CH}(\text{O})\text{CH}_3$	24.2	37.4	30.0	28.7 ^c	0.066
TS4-2	$\text{CH}_3\text{CH}(\text{CH}_2)\text{OOH} \rightarrow (\text{CH}_3)_2\text{CO} + \text{OH}$	20.4	31.9	29.6	27.7 ^d	0.065
TS8	$\text{CH}_2\text{CH}_2\text{CH}_2\text{OOH} \rightarrow \text{CH}_2(\text{OH})\text{CH}_2\text{CH}_2\text{O}$	20.3	30.6	23.7	23.8 ^c	0.069
TS9	$\text{CH}_3\text{CHCH}_2\text{OOH} \rightarrow \text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{O}$	19.5	32.4	22.3	22.9 ^c	0.094
TS9-22	$\text{CH}_3\text{CHCH}_2\text{OOH} \rightarrow \text{CH}_3\text{CH}_2\text{CHO} + \text{OH}$	15.7	27.7	26.5	23.7 ^c	0.072

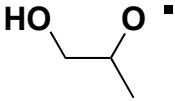
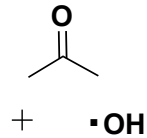
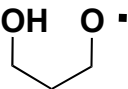
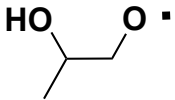
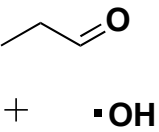
^aAll energies are relative to respective reactant in kcal/mol.

^bPresent work employing 6-311++G(d,p) basis set.

^cPresent work employing cc-pVTZ//M06-2X/6-311++G(d,p) basis set.

^dPresent work employing cc-pVDZ//M06-2X/6-311++G(d,p) basis set.

Table 4. Product energies of propyl systems.

Species	Structure	Relative energy ^a				T1 diagnostic
		B3LYP ^b	M06-2X ^b	CBS-QB3	UCCSD(T)-F12	
4product		-53.1	-53.9	-50.6	-51.2 ^d	0.083
4-2product		-50.4	-46.8	-46.0	-47.4 ^c	
8product		-52.3	-53.8	-50.3	-50.5 ^c	0.069
9product		-52.8	-55.1	-52.1	-52.7 ^c	0.094
9-22product		-42.2	-40.1	-39.7	-41.1 ^c	

^aAll energies are relative to respective reactant in kcal/mol.

^bPresent work employing 6-311++G(d,p) basis set.

^cPresent work employing cc-pVTZ//M06-2X/6-311++G(d,p) basis set and geometry.

^dPresent work employing cc-pVDZ//M06-2X/6-311++G(d,p) basis set and geometry.

^ePresent work employing cc-pVQZ//M06-2X/6-311++G(d,p) basis set and geometry.

Similar to the ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems, the OH transfer reaction pathway was very exothermic and higher in energy than the H transfer pathway. The new reactions of species 4, 4-2, 8, 9, and 9-22 were also compared to the more common pathways (Figure 5 and 6).

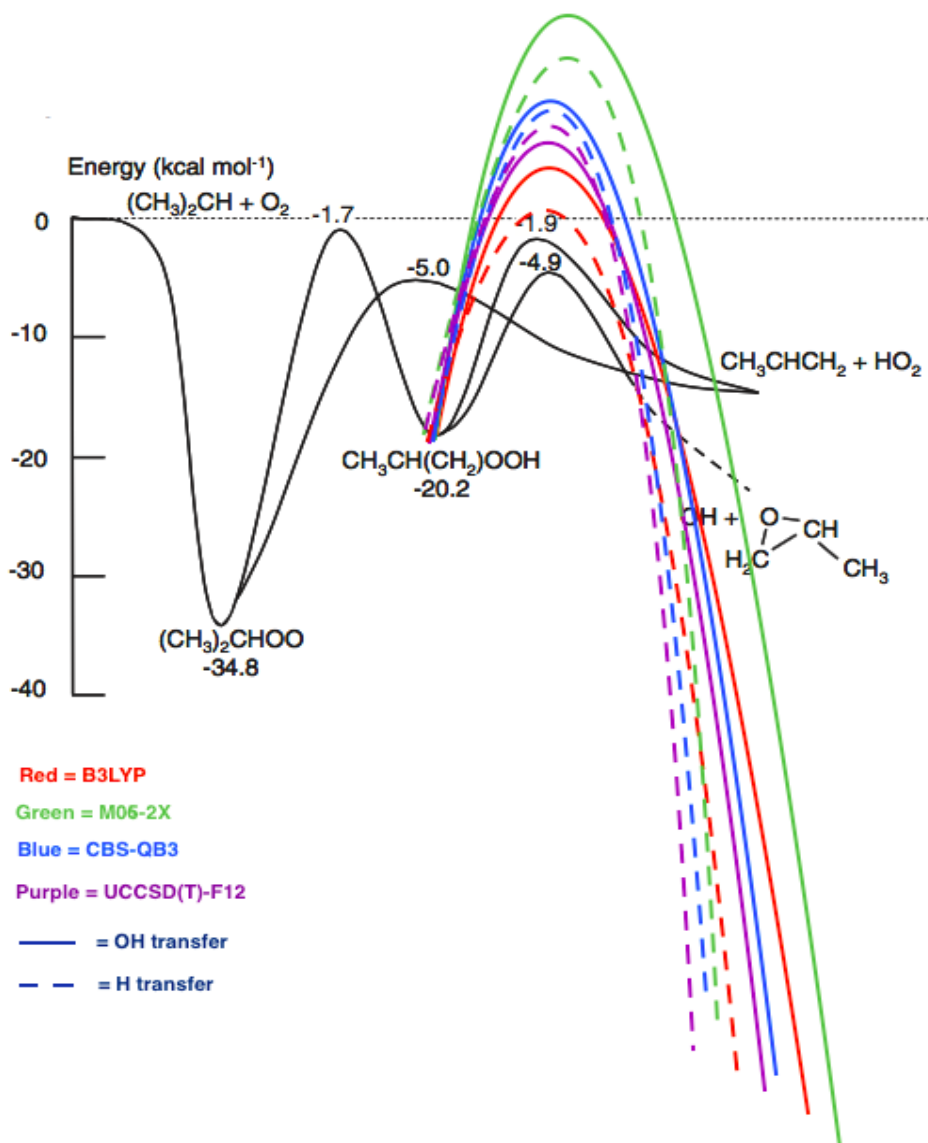


Figure 5. Schematic potential energy surfaces comparing calculated energy barriers of new OH transfer (4) and H transfer (4-2) pathways and those of common HO₂ and OH elimination pathways from CH₃CH₂(CH₂)OOH using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), CBS-CB3, and UCCSD(T)-F12/cc-pVDZ//M06-2X/6-31++G(d,p) levels of theory. Drawn barriers represent E + ZPE values in kcal/mol relative to the i-propyl + O₂ entrance channels; lowest energy conformers from Goldsmith et al.³ are used for reactants and relevant products. TS4_{B3LYP} = 4.0, TS4_{M06-2X} = 17.2, TS4_{CBS-QB3} = 9.8, TS4_{UCCSD(T)-F12} = 6.5; TS4-2_{B3LYP} = 0.2, TS4-2_{M06-2X} = 11.7, TS4-2_{CBS-QB3} = 9.4, TS4-2_{UCCSD(T)-F12} = 7.5. 4product_{B3LYP} = -73.3,

$4\text{product}_{\text{M06-2X}} = -74.1$, $4\text{product}_{\text{CBS-QB3}} = -70.8$, $4\text{product}_{\text{UCCSD(T)-F12}} = -71.4$; $4\text{-2product}_{\text{B3LYP}} = -70.6$, $4\text{-2product}_{\text{M06-2X}} = -67.0$, $4\text{-2product}_{\text{CBS-QB3}} = -66.2$, $4\text{-2product}_{\text{UCCSD(T)-F12}} = -67.9$. Base figure adapted from Huang et al.⁴

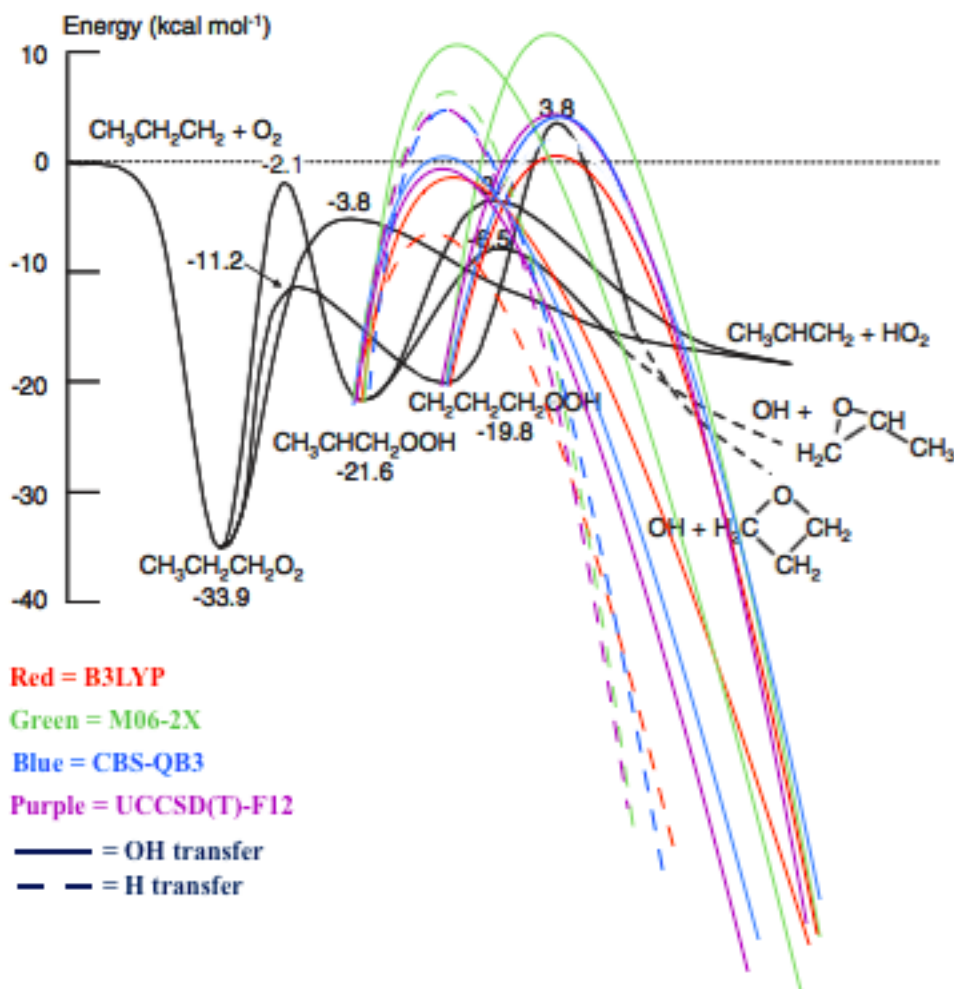


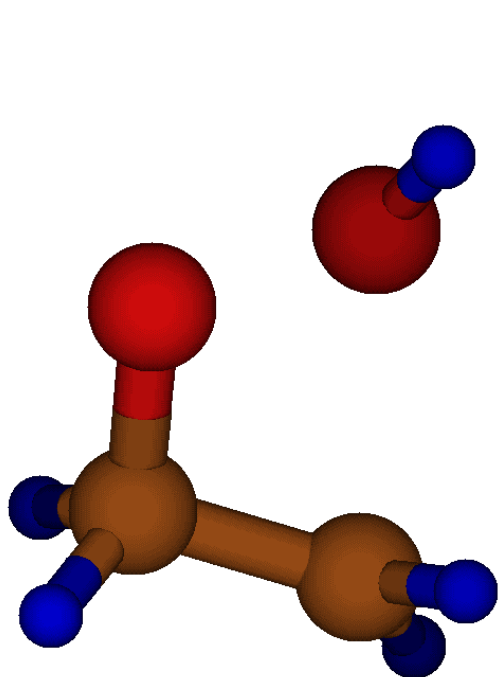
Figure 6. Schematic potential energy surfaces comparing calculated energy barriers of new OH transfer (8 and 9) and H transfer (9-22) pathways and those of common HO₂ and OH elimination pathways from CH₂CH₂CH₂OOH (8) and CH₃CHCH₂OOH (9 and 9-22) using B3LYP/6-311++G(d,p), M06-2X/6-311++G(d,p), CBS-CB3, and UCCSD(T)-F12/cc-pVDZ//M06-2X/6-31++G(d,p) levels of theory. Drawn barriers represent E + ZPE values in kcal/mol relative to the n-propyl + O₂ entrance channels; lowest energy conformers from Goldsmith et al.³ are used for reactants and relevant products. TS8_{B3LYP} = 0.5, TS8_{M06-2X} = 10.8, TS8_{CBS-QB3} = 3.9, TS8_{UCCSD(T)-F12} = 4.0; TS9_{B3LYP} = -1.8, TS9_{M06-2X} = 10.4, TS9_{CBS-QB3} = 0.5, TS9_{UCCSD(T)-F12} = -0.2; TS9-22_{B3LYP} = -5.6, TS9-22_{M06-2X} = 5.6, TS9-22_{CBS-QB3} = 4.8, TS9-22_{UCCSD(T)-F12} = 4.8. 8product_{B3LYP} = -72.1, 8product_{M06-2X} = -73.6, 8product_{CBS-QB3} = -70.1, 8product_{UCCSD(T)-F12} = -70.3; 9product_{B3LYP} = -74.1, 9product_{M06-2X} = -77.2, 9product_{CBS-QB3} = -73.9, 9product_{UCCSD(T)-F12} = -75.9; 9-22product_{B3LYP} = -63.6, 9-22product_{M06-2X} = -62.2, 9-22product_{CBS-QB3} = -66.0, 9-22product_{UCCSD(T)-F12} = -60.0. Base figure adapted from Huang et al.⁴

While the OH transfer and H transfer reaction pathway for the i-propyl system (4 and 4-2) and 9 and 9-22 were high in energy, the OH transfer for 8 revealed considerable comparison with the cyclic ether formation—while the cyclic ether formation undergoes a 4-membered ring transition state, the OH transfer undergoes a 5-membered ring transition state, which is much more favorable in energy. Though the H transfer was higher in energy than most OH transfer reactions, a H transfer transition structure could not be found for species 8. Several slight changes in transition state conformations for each species and reaction type often lead to very different reactions, highlighting the uniqueness of these two new reaction pathways.

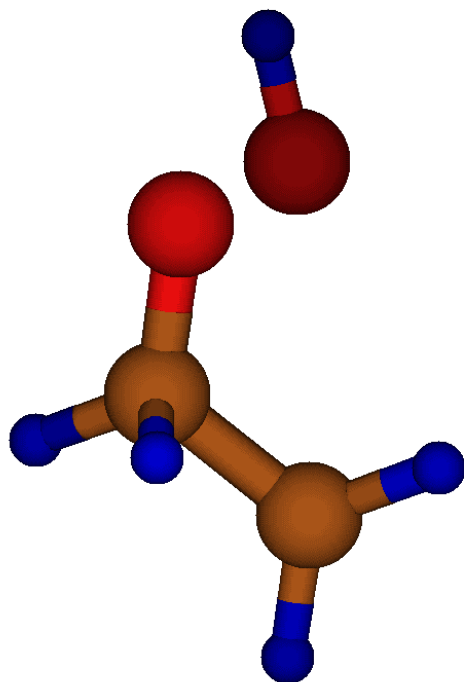
References

- (1) Burke, M.P.; Goldsmith, C.F.; Klippenstein, S.J.; Welz, O.; Huang, H.; Antonov, I.O.; Savee, J.D.; Osborn, D.L.; Zádor, J.; Taatjes, C.A.; Sheps, L. Multiscale informatics for low-temperature propane oxidation: Further complexities in studies of complex reactions. *J. Phys. Chem. A* **2015**, 119, 7095-7115.
- (2) DeSain, J. D.; Klippenstein, S.J.; Miller, J.A.; Taatjes, C.A.; Measurements, Theory, and Modeling of OH formation in Ethyl + O₂ and Propyl + O₂ Reactions. *J. Phys. Chem. A* **2003**, 107, 4415-4427.
- (3) Goldsmith, C.F.; Green, W.H.; Klippenstein, S.J. Role of O₂ + QOOH in low-temperature ignition of propane. 1. Temperature and pressure dependent rate coefficients. *J. Phys. Chem. A* **2012**, 116, 3325-3346.
- (4) Huang, H.; Merthe, D.J.; Zádor, J.; Jusinski, L.E.; Taatjes, C.A. New experiments and validated master-equation modeling for OH production in propyl + O₂ reactions. *Proc. Combust. Inst.* **2011**, 33, 293-299.
- (5) Knepp, A.M.; Meloni, G.; Jusinski, L.E.; Taatjes, C.A.; Cavallotti, C.; Klippenstein, S. J. Theory, measurements, and modeling of OH and HO₂ formation in the reaction of cyclohexyl radicals with O₂. *Phys. Chem. Chem. Phys.* **2007**, 9, 4315-4331.
- (6) Miller, J.A.; Klippenstein, S.J.; Roberston, S.H. A theoretical analysis of the reaction between ethyl and molecular oxygen. *Proc. Combust. Inst.* **2000**, 28, 1479-1486.
- (7) Zádor, J.; Fernandes, R.X.; Georgievskii, Y.; Meloni, G.; Taatjes, C.A.; Miller, J.A. The reaction of hydroxyethyl radical with O₂: A theoretical analysis and experimental study. *Proc. Combust. Inst.* **2009**, 32, 271-277.
- (8) Manley, D.K.; McIlroy, A.; Taatjes, C.A. Research needs for future internal combustion engines. *Physics Today, Scientific American* **2008**, Nov., 47-52.

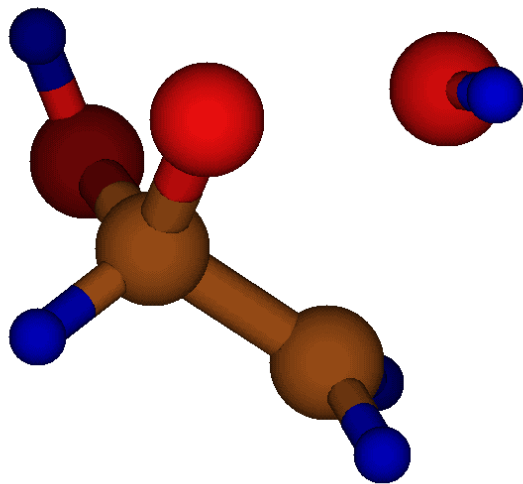
Appendix A: Images of M06-2x/6-311++G(d,p) optimized transition states of ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems.



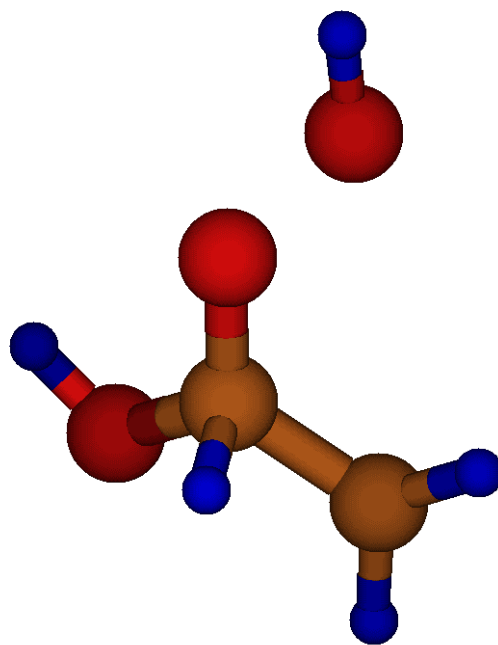
TS1



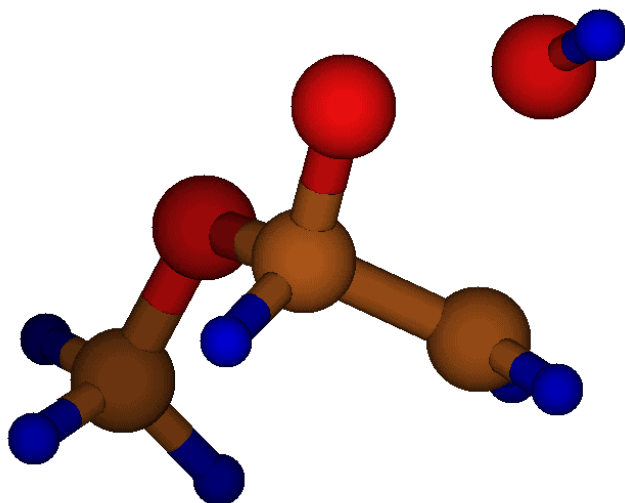
TS1-2



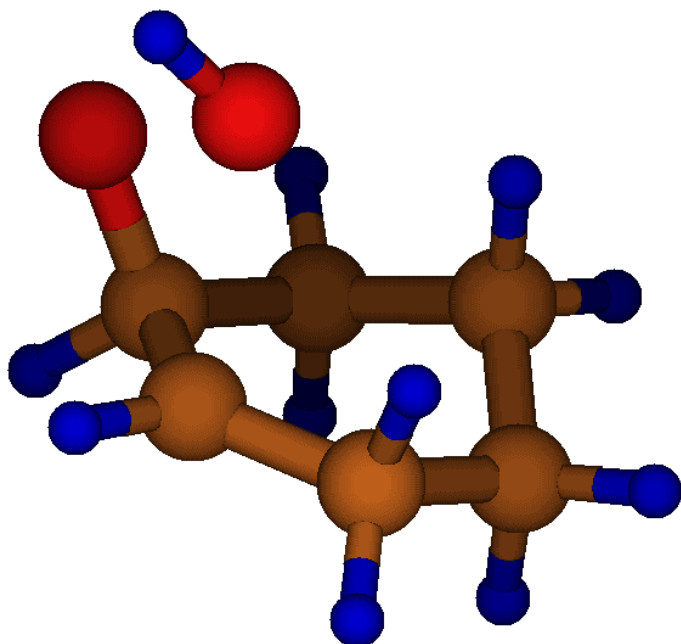
TS2



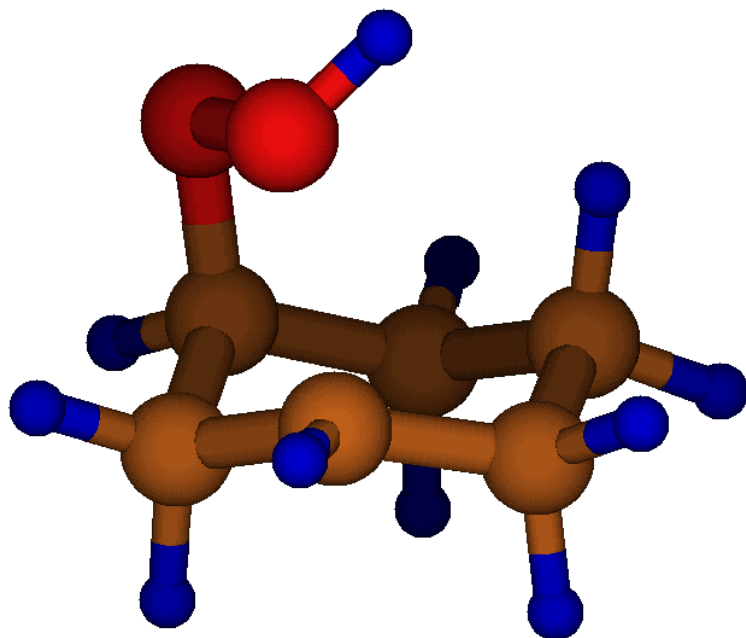
TS2



TS3

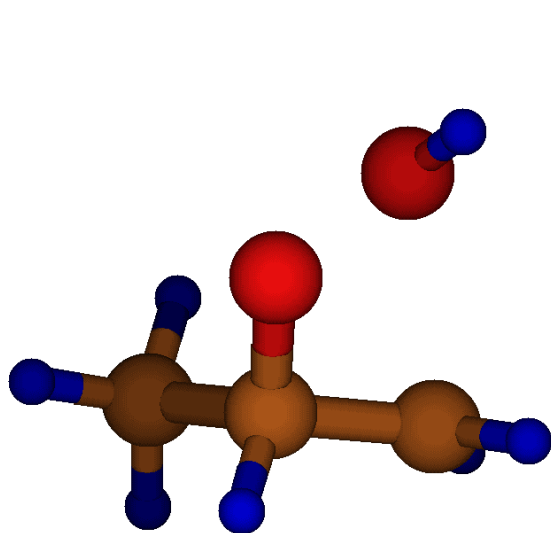


TS5

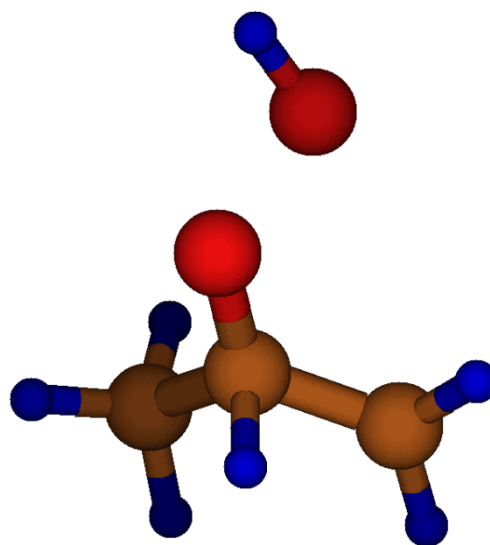


TS6

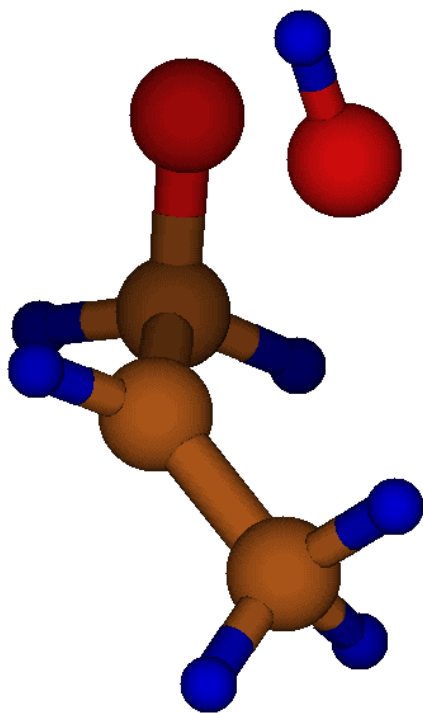
Appendix B: Images of M06-2x/6-311++G(d,p) optimized transition states of propyl systems.



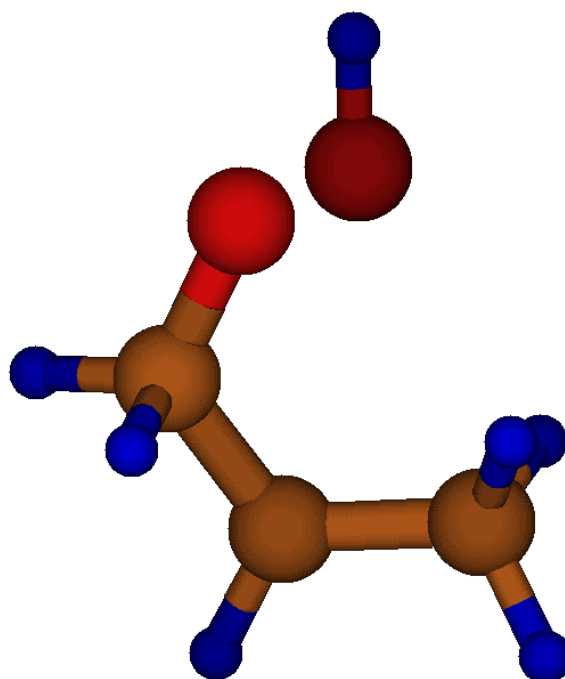
TS4



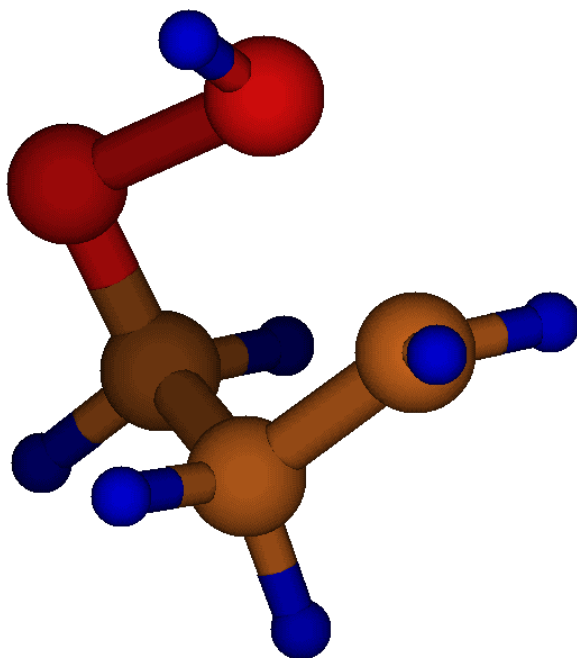
TS4-2



TS9



TS9-22



TS8

Appendix C: M06-2x/6-311++G(d,p) optimized geometries of ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems in Cartesian coordinates (Angstroms).

1reactant

C	-0.057625	0.180951	-0.128217
H	-0.185634	0.227242	0.959785
O	1.355906	0.041011	-0.262939
C	-0.804693	-0.957915	-0.721825
H	-0.364654	1.137158	-0.558409
H	-1.600003	-0.804143	-1.436466
H	-0.565277	-1.968256	-0.415418
O	1.667752	0.196643	-1.640961
H	1.816194	-0.715284	-1.921199

TS1

C	1.046292	-1.351071	-0.176859
C	-0.000973	-0.482963	0.382785
O	1.615829	0.901455	0.040841
O	1.743991	-0.516743	-1.042293
H	1.482135	1.476670	-0.722373
H	-0.711564	-0.035136	-0.299709
H	0.628393	-2.165858	-0.788292
H	1.684005	-1.777375	0.605132
H	-0.221740	-0.449998	1.441372

1product

C	0.034240	-0.003771	-0.010058
H	0.116041	-0.028324	1.089149
O	1.294686	0.128826	-0.528662
C	-0.723760	-1.220331	-0.503577
H	-0.511452	0.926961	-0.241337
H	-0.644061	-1.264201	-1.595425
O	-0.134416	-2.352376	0.105894
H	-1.782528	-1.130137	-0.228122
H	-0.413065	-3.146496	-0.353525

1-2reactant

H	1.245160	-0.716359	-0.708317
C	0.835123	-0.371067	0.245632
C	1.156848	1.055569	0.495537
O	-0.563657	-0.632939	0.288297
O	-1.151071	0.101362	-0.786116
H	-1.649727	-0.595578	-1.228453
H	1.213558	-1.021618	1.044481
H	1.834530	1.600877	-0.144298
H	0.678302	1.568174	1.318700

TS1-2

H	1.003490	-1.551255	0.344618
C	0.352693	-1.042559	1.066689
C	0.679308	0.343877	1.336316
O	-0.864015	-1.482622	1.224911
O	-1.762677	-0.503698	-0.064952
H	-2.505110	-1.121974	-0.049637
H	0.851730	-1.143101	2.117405
H	1.692093	0.694397	1.186210
H	-0.027541	0.941231	1.897792

1-2product

H	1.906549	-1.178359	0.383968
C	0.996071	-0.581079	0.189505
C	1.124549	0.897312	0.394776
O	-0.011680	-1.139926	-0.164092
O	-2.030377	0.818386	-0.500682
H	-1.609610	-0.064987	-0.488661
H	1.428646	1.088173	1.427675
H	1.922755	1.278674	-0.247934
H	0.190158	1.414810	0.178470

2reactant

C	-0.081813	0.111220	-0.102511
C	-0.133482	0.226228	1.372283
O	1.244842	0.051680	-0.518313
O	-0.786479	-1.084023	-0.427559
H	-0.611038	0.932028	-0.606034
O	-0.843766	-1.121527	-1.857829
H	-0.533198	-2.018782	-2.026911
H	1.251292	0.068271	-1.480178
H	-1.081879	0.383044	1.862892
H	0.762356	0.027684	1.941167

TS2

C	1.000284	-1.396125	-0.220933
C	0.006365	-0.499232	0.393733
O	1.682048	0.833507	0.045794
O	1.764587	-0.588292	-1.021338
H	1.543180	1.406992	-0.718448
H	-0.728829	-0.032010	-0.247034
H	0.529612	-2.130899	-0.900260

O	1.725389	-2.051943	0.785611
H	-0.147137	-0.512274	1.463997
H	2.629661	-2.135682	0.468264

2product

C	0.000219	-0.066415	-0.183757
H	0.201190	-0.014964	0.906448
O	1.183508	-0.365508	-0.891938
C	-0.950194	-1.276115	-0.410113
O	-0.575190	1.087393	-0.517061
O	-1.165367	-1.497004	-1.771256
H	-0.481234	-2.137757	0.077788
H	-1.909358	-1.061192	0.057515
H	-0.301289	-1.572304	-2.190802
H	1.600157	0.464973	-1.141146

2-2reactant

O	1.274239	-0.652640	-0.519327
C	0.501162	-0.157159	0.524350
C	0.700013	1.310178	0.603841
O	-0.869158	-0.532918	0.384819
O	-1.404106	0.085180	-0.779272
H	-1.975392	0.761805	-0.395320
H	0.733740	-0.650564	1.479257
H	1.163283	1.829989	-0.221013
H	0.329730	1.851931	1.462369
H	1.062745	-1.584096	-0.636215

TS2-2

O	1.022773	-1.491222	0.394236
C	0.180510	-0.878230	1.294139
C	0.511406	0.509892	1.553535
O	-1.030931	-1.330457	1.411756
O	-1.904770	-0.492658	0.063263
H	-2.746527	-0.903721	0.300695
H	0.564043	-0.989508	2.405954
H	1.530291	0.838719	1.406283
H	-0.184362	1.086054	2.145778
H	0.613475	-2.331907	0.163011

2-2product

O	1.745869	-0.934564	0.146206
C	0.584353	-0.270426	0.076765

C	0.701455	1.129462	0.598673
O	-0.415182	-0.784542	-0.360894
O	-2.505874	1.113911	-0.222457
H	-2.029080	0.277121	-0.397699
H	0.999320	1.094866	1.647632
H	1.482593	1.655628	0.048860
H	-0.248038	1.650106	0.496583
H	1.597252	-1.825172	-0.200668

3reactant

C	0.302419	0.199345	-0.382072
C	-0.370588	0.169306	0.942748
O	1.655406	0.494906	-0.148238
O	0.268693	-1.051568	-1.027676
H	-0.117319	0.966216	-1.048705
O	2.279603	0.686637	-1.410139
H	2.389803	-0.225169	-1.715566
C	-0.969273	-1.321509	-1.652090
H	-0.877302	-2.291876	-2.135735
H	-1.783739	-1.360856	-0.920218
H	-1.201890	-0.560890	-2.406597
H	-0.796661	1.069727	1.360290
H	-0.253047	-0.711417	1.559015

TS3

O	1.179292	-1.326720	-0.154280
H	0.350988	-0.834513	-0.216039
O	2.093440	0.183903	-0.239667
C	3.037607	-0.268943	-1.117656
H	3.376751	0.608284	-1.702006
C	2.309647	-1.253517	-1.952208
H	1.555168	-0.894813	-2.639384
O	4.127694	-0.836798	-0.441653
C	5.259755	-0.998879	-1.263182
H	6.050246	-1.413077	-0.640482
H	5.069095	-1.687686	-2.096412
H	5.592705	-0.036913	-1.673963
H	2.683856	-2.261902	-2.069096

3product

C	-0.082989	0.137308	-0.172126
H	-0.005167	0.033770	0.926142
O	1.043296	0.426128	-0.742291
O	-0.824927	-0.943344	-0.653010

C	-1.009344	1.463444	-0.335480
C	-0.462804	-1.386521	-1.949062
H	-1.191899	-2.145530	-2.223712
H	-0.493120	-0.565840	-2.673790
H	0.543456	-1.810566	-1.954416
H	-1.181342	1.563430	-1.411174
H	-1.937523	1.246239	0.188263
O	-0.404506	2.559558	0.244931
H	0.456084	2.685462	-0.169271

5reactant

C	1.038201	-1.346619	-0.282303
C	-0.876010	0.024589	0.750823
C	0.096712	1.153083	0.845275
C	1.431511	1.097770	0.178482
C	2.041817	-0.308479	0.218671
H	1.488536	-2.342595	-0.268126
H	0.765571	-1.124622	-1.316723
H	-1.362133	1.272368	-1.765296
H	-0.274299	2.101390	1.218729
H	1.320566	1.392010	-0.878073
H	2.108329	1.828399	0.630236
H	2.953966	-0.331956	-0.382385
H	2.327621	-0.547482	1.249611
H	-1.525442	0.015746	1.632764
O	-1.869501	0.262312	-0.271732
O	-1.212001	0.347973	-1.533988
C	-0.214886	-1.348031	0.597513
H	0.067054	-1.687644	1.600869
H	-0.965903	-2.045367	0.218972

TS5

C	1.029281	-1.408975	-0.229274
C	-0.894865	-0.078451	0.782570
C	-0.027415	1.099567	0.547766
C	1.398219	1.048636	0.171876
C	2.025073	-0.347154	0.238949
H	1.506259	-2.391702	-0.243045
H	0.694951	-1.182553	-1.245539
H	-1.637419	1.161157	-1.845535
H	-0.495064	2.066928	0.692939
H	1.383146	1.389373	-0.880992
H	1.981861	1.793529	0.722266
H	2.932390	-0.362003	-0.368204
H	2.326927	-0.567324	1.268902

H	-1.400825	0.034699	1.756108
O	-1.915915	0.069906	-0.162474
O	-0.866017	0.721321	-1.470533
C	-0.174316	-1.429397	0.713475
H	0.167399	-1.703950	1.718382
H	-0.912013	-2.171684	0.403303

5product

C	1.079973	-1.396739	-0.205939
C	-0.778799	0.016575	0.861563
C	-0.199223	1.077217	-0.081868
C	1.330792	1.100553	-0.002970
C	1.931663	-0.265792	0.368115
H	1.559449	-2.364523	-0.041863
H	0.979932	-1.253382	-1.283597
H	-1.542727	0.671792	-1.414819
H	-0.602573	2.054018	0.211252
H	1.688199	1.418307	-0.985319
H	1.647404	1.859341	0.716208
H	2.956916	-0.332112	-0.000771
H	1.985055	-0.373307	1.457333
H	-0.439928	0.197747	1.892457
O	-2.142418	-0.032486	0.798056
O	-0.585966	0.788875	-1.414747
C	-0.316707	-1.422670	0.436505
H	-0.335893	-2.044600	1.332498
H	-1.049206	-1.829750	-0.262028

6reactant

C	0.842345	-0.660718	-0.453821
C	-0.521339	-0.827277	-1.131978
C	-1.625422	0.108351	-0.623975
C	-1.143333	1.573564	-0.617672
C	0.142100	1.693846	0.129887
C	1.259790	0.817787	-0.327163
O	0.958713	-1.358544	0.782595
O	0.106175	-0.741687	1.749578
H	-2.505714	0.007720	-1.265561
H	-1.922544	-0.170316	0.388071
H	-0.400093	-1.503414	2.055298
H	2.130294	0.879968	0.327854
H	0.175622	2.156151	1.106383
H	-1.912683	2.224659	-0.199126
H	-0.991712	1.883595	-1.663265
H	1.596589	-1.193533	-1.042013

H	1.581714	1.132281	-1.331873
H	-0.366704	-0.629017	-2.199730
H	-0.825089	-1.874734	-1.048304

TS6

C	2.452924	-1.396908	0.547380
C	1.155298	-1.444913	-0.264846
C	0.010891	-0.659391	0.381235
C	0.433558	0.772418	0.748353
C	1.758925	0.819863	1.436992
C	2.887045	0.054598	0.797191
O	2.350533	-2.092384	1.782582
O	1.739451	-0.832193	2.702213
H	-0.843619	-0.623844	-0.298984
H	-0.327992	-1.183159	1.277976
H	0.921650	-1.197559	3.081421
H	3.769463	0.061451	1.438563
H	2.005506	1.687717	2.037812
H	-0.336191	1.265471	1.346162
H	0.520360	1.353998	-0.185024
H	3.238140	-1.930436	-0.003976
H	3.163068	0.520564	-0.160372
H	1.364856	-1.031839	-1.259228
H	0.864784	-2.489099	-0.403168

6product

C	0.882951	-0.868858	-0.710757
C	-0.427496	-0.785401	-1.506182
C	-1.546786	-0.149629	-0.682940
C	-1.143483	1.242606	-0.194916
C	0.153250	1.206839	0.617938
C	1.283926	0.544062	-0.169861
O	0.762739	-1.653535	0.404745
O	-0.030780	0.596996	1.882197
H	-2.459099	-0.088675	-1.280208
H	-1.778305	-0.792079	0.172162
H	-0.125805	-0.354757	1.756437
H	2.173044	0.452767	0.455720
H	0.458483	2.232347	0.844184
H	-1.928402	1.678899	0.426366
H	-0.998429	1.903488	-1.057965
H	1.712798	-1.222429	-1.339043
H	1.536268	1.143622	-1.049766
H	-0.235497	-0.192189	-2.407844
H	-0.704098	-1.790505	-1.833439

Appendix D: M06-2X/6-311++G(d,p) optimized geometries of propyl systems in Cartesian coordinates (Angstroms).

9reactant/9-22reactant

O	0.009985	-0.313206	-0.674872
H	-0.074107	0.467856	-0.115496
O	1.415911	-0.336488	-0.916703
C	1.908268	-1.574347	-0.385403
H	1.312859	-2.384519	-0.823760
H	1.778253	-1.579678	0.700555
C	3.334150	-1.675817	-0.784227
H	4.099167	-1.433092	-0.057884
C	3.698171	-1.824565	-2.219048
H	4.695954	-2.249666	-2.341013
H	2.977378	-2.460351	-2.740885
H	3.688507	-0.853773	-2.732051

TS9

H	1.055929	-1.398718	-0.155759
O	0.369301	-0.875736	0.273809
O	1.313101	0.617335	-0.064542
C	0.729292	1.326137	0.974294
H	1.413165	2.154706	1.217638
C	0.627074	0.368907	2.089495
H	1.556803	-0.088131	2.409487
H	-1.000408	-0.872073	2.349988
H	-0.467618	-0.129244	3.870277
C	-0.619007	0.057625	2.806257
H	-1.380815	0.825052	2.662330
H	-0.254331	1.736280	0.711170

9product

H	-0.147340	-0.874184	-1.475452
O	-0.761749	-0.913776	-0.734238
O	1.328180	0.885181	-1.084407
C	0.531082	0.963248	0.025264
H	1.058438	1.435264	0.866822
C	-0.101438	-0.381308	0.393585
H	0.704531	-1.057552	0.711146
H	-0.651092	0.179188	2.406839
H	-1.924722	0.429299	1.189688
C	-1.116191	-0.232640	1.508549
H	-1.545742	-1.204229	1.753055

H	-0.281603	1.655904	-0.271975
---	-----------	----------	-----------

TS9-22

O	0.441203	-1.041042	-1.583935
H	-0.181295	-0.307191	-1.506157
O	1.623451	-0.147899	-0.378662
C	2.258026	-1.205313	0.082474
H	1.771526	-1.770937	0.892846
H	3.221266	-0.808290	0.551170
C	3.008572	-2.044735	-0.836403
H	3.454196	-2.950151	-0.435789
C	3.294961	-1.588938	-2.203288
H	4.161612	-2.073904	-2.648093
H	2.397701	-1.780793	-2.810125
H	3.393275	-0.498454	-2.214824

9-22product1

O	-2.084821	-0.018006	-0.945809
C	-2.018836	1.040461	-0.382450
H	-1.152626	1.271772	0.270545
C	-3.056521	2.127083	-0.491630
H	-3.387514	2.356952	0.528376
H	-2.538780	3.030435	-0.836183
C	-4.221818	1.759472	-1.397776
H	-3.870681	1.543945	-2.407547
H	-4.726222	0.864546	-1.031616
H	-4.946113	2.573096	-1.446382

OH

O	-2.007932	-0.022230	-0.953107
H	-2.007092	0.830003	-0.485636

8reactant

O	0.028575	0.000122	-0.038476
H	-0.006319	0.054084	0.923756
O	1.437616	0.000717	-0.260034
C	1.723225	1.093306	-1.114091
H	2.815338	1.119362	-1.143605
H	1.358018	2.021386	-0.659389
C	1.145187	0.911146	-2.514159
H	1.547873	-0.006600	-2.949604
H	0.061402	0.761489	-2.409051
C	1.435189	2.086439	-3.379988

H	1.301071	3.090974	-2.998004
H	1.630641	1.970845	-4.436228

TS8

C	0.998336	-1.348445	-0.324618
C	-0.175611	-0.699323	0.411039
O	1.909077	0.652354	0.277422
O	1.745101	-0.373230	-1.019921
H	1.736297	1.493326	-0.174326
H	0.659687	-2.067158	-1.079510
H	1.639955	-1.877060	0.392236
H	-0.809086	-1.485989	0.843367
C	0.314567	0.231050	1.486992
H	-0.785719	-0.154102	-0.314130
H	-0.266795	1.112566	1.730885
H	0.919029	-0.178605	2.285898

8product

C	-0.016356	-1.331103	-0.496794
C	-0.997745	-0.596549	0.414391
O	-0.227490	1.542501	-0.234088
O	1.151046	-1.690673	0.118175
H	0.257535	2.329136	0.022357
H	0.297846	-0.649190	-1.310831
H	-0.469792	-2.199371	-0.997054
H	-1.244989	-1.233196	1.268064
C	-0.416258	0.714226	0.906712
H	-1.921751	-0.397432	-0.136214
H	-1.100112	1.187629	1.619479
H	0.538039	0.521588	1.409339

4reactant/4-22reactant

C	0.004028	-0.023998	-0.069557
H	0.214168	0.870991	0.504486
H	0.717230	-0.835456	-0.054749
C	-1.270260	-0.106671	-0.837162
H	-1.336675	-1.058302	-1.372209
C	-2.497955	0.100961	0.041079
H	-3.404268	0.074186	-0.565234
H	-2.548881	-0.684365	0.796347
H	-2.438205	1.067317	0.545759
O	-1.358038	0.945872	-1.809474
O	-0.368967	0.711194	-2.800358
H	0.334488	1.317487	-2.535550

TS4

O	1.202831	-1.314895	-0.108923
H	0.352994	-0.859710	-0.145024
O	2.047423	0.250047	-0.328251
C	3.021907	-0.282269	-1.172093
H	3.364967	0.564818	-1.790990
C	2.261494	-1.241339	-1.994520
H	1.440757	-0.869882	-2.593908
C	4.181972	-0.909901	-0.413053
H	2.624519	-2.243010	-2.189269
H	4.928031	-1.317647	-1.099540
H	4.657087	-0.160585	0.220129
H	3.800761	-1.710640	0.222919

4product

O	-1.159283	-1.148048	0.267028
H	-1.506951	-0.417125	0.789511
O	-0.084556	1.439043	0.327403
C	0.505167	0.468880	-0.447935
H	0.976038	0.920628	-1.333220
C	-0.526386	-0.588014	-0.857991
H	-1.256953	-0.127771	-1.534272
C	1.617249	-0.135567	0.437898
H	-0.031149	-1.401536	-1.392336
H	2.161675	-0.870583	-0.157091
H	2.304793	0.641382	0.768955
H	1.174022	-0.630868	1.300539

TS4-2

O	1.788253	-0.106745	-0.436950
C	2.936490	-0.441211	-0.971968
C	2.936085	-1.023590	-2.304472
C	4.071426	-0.793884	-0.031629
O	1.034254	-1.738469	-0.165266
H	3.871671	-1.295266	-2.775437
H	2.029723	-0.955084	-2.891653
H	3.189141	0.481097	-1.655912
H	4.164881	-0.029966	0.738999
H	3.807910	-1.740954	0.442573
H	5.021653	-0.907306	-0.555303
H	0.411004	-1.380907	0.480034

4-2product1

H	-1.713198	-1.129448	1.236811
H	-2.210525	0.574333	1.458159
C	-1.474395	-0.088119	1.007731
H	-1.501606	0.017767	-0.080227
H	0.873864	-1.653502	1.505820
C	-0.090647	0.258396	1.512594
C	1.031057	-0.659554	1.078071
H	1.030936	-0.772256	-0.008754
H	1.986671	-0.263372	1.414905
O	0.107050	1.212677	2.221518

Appendix E: M06-2x/6-311++G(d,p) frequencies of ethane, ethanol, ethyl methyl ether, and cyclohexane radical systems (cm⁻¹).

1reactant

Frequencies --	134.3884	180.8361	195.3431
Frequencies --	368.6969	460.8797	577.2108
Frequencies --	835.4136	923.5550	1036.5740
Frequencies --	1095.6598	1158.8520	1298.4736
Frequencies --	1383.4176	1399.2603	1458.8713
Frequencies --	1478.6609	3050.9221	3112.2493
Frequencies --	3175.5225	3289.7020	3837.6513

TS1

Frequencies --	-2692.5586	219.8235	380.4119
Frequencies --	401.8513	522.6151	567.8120
Frequencies --	800.1372	897.9513	967.4354
Frequencies --	1023.6982	1079.2689	1176.8579
Frequencies --	1189.8525	1348.9769	1465.8124
Frequencies --	1491.2997	3013.2133	3070.8073
Frequencies --	3172.2589	3290.2523	3875.2597

1product

Frequencies --	151.0160	229.1646	330.1338
Frequencies --	441.3968	602.3490	899.2497
Frequencies --	998.3926	1092.7105	1110.0191
Frequencies --	1165.3803	1273.2958	1291.2220
Frequencies --	1370.5950	1400.7403	1458.4927
Frequencies --	1511.3812	2975.6414	3007.4826
Frequencies --	3027.1359	3078.0110	3925.6365

1-2reactant

Frequencies --	76.0708	164.3557	174.9088
Frequencies --	366.8620	440.9606	555.1638
Frequencies --	843.5011	920.8022	1034.2229
Frequencies --	1108.8939	1149.7451	1292.9017
Frequencies --	1381.1032	1411.4625	1459.2177
Frequencies --	1476.4598	3037.0199	3090.7942

Frequencies	--	3185.4656	3300.1934	3859.1753
TS1-2				
Frequencies	--	-1429.9086	123.1217	169.2146
Frequencies	--	232.5445	279.2340	477.9464
Frequencies	--	559.1713	768.3719	853.0946
Frequencies	--	959.0765	969.2477	1157.6176
Frequencies	--	1242.3405	1319.7683	1411.7247
Frequencies	--	1492.6939	2512.2235	3051.8014
Frequencies	--	3175.8505	3302.9597	3862.2768
1-2product				
Frequencies	--	43.8601	79.1542	139.8010
Frequencies	--	193.1655	382.1823	434.8174
Frequencies	--	529.5616	780.0241	904.6960
Frequencies	--	1144.5814	1145.9263	1388.3519
Frequencies	--	1432.2515	1464.3556	1469.0866
Frequencies	--	1853.8687	2977.8433	3060.5426
Frequencies	--	3129.7968	3168.6101	3672.7566
2reactant				
Frequencies	--	134.5453	163.9720	205.2613
Frequencies	--	290.0180	381.4138	443.3031
Frequencies	--	475.6743	568.3200	615.7175
Frequencies	--	916.8501	970.2709	1028.3587
Frequencies	--	1082.8369	1183.4240	1273.3097
Frequencies	--	1358.7953	1413.6518	1421.7053
Frequencies	--	1494.9797	3027.9471	3199.4478
Frequencies	--	3323.1134	3864.1577	3889.0325
TS2				
Frequencies	--	-2739.4163	91.2877	171.5683
Frequencies	--	385.6128	407.7106	425.7622
Frequencies	--	514.5046	520.4302	686.8628
Frequencies	--	865.1141	922.5557	988.6767
Frequencies	--	1025.6875	1088.8946	1158.0275
Frequencies	--	1203.7679	1322.3115	1368.0104
Frequencies	--	1474.2943	2969.8397	3181.9752
Frequencies	--	3309.9954	3869.1027	3894.1564
2product				
Frequencies	--	131.9682	241.0974	260.9302
Frequencies	--	369.6104	440.2605	529.4257
Frequencies	--	652.3762	798.6393	878.2748
Frequencies	--	995.6712	1065.7187	1145.7089
Frequencies	--	1182.6856	1224.1214	1286.3943
Frequencies	--	1315.0659	1387.6192	1418.4825
Frequencies	--	1503.1239	2903.6336	3056.2500
Frequencies	--	3164.7452	3865.6778	3891.9488
2-2reactant				

Frequencies	--	147.4682	166.4723	184.0060
Frequencies	--	288.3444	339.8698	409.4306
Frequencies	--	456.5743	515.1630	749.6285
Frequencies	--	887.4336	929.2265	1033.9713
Frequencies	--	1081.8156	1189.1685	1276.4692
Frequencies	--	1375.1376	1402.6038	1414.0565
Frequencies	--	1478.4199	3027.5749	3191.5819
Frequencies	--	3314.4318	3843.6124	3884.9280

TS2-2

Frequencies	--	-1492.5425	85.0638	155.9554
Frequencies	--	181.7564	287.0215	355.2472
Frequencies	--	426.3809	499.8224	545.3628
Frequencies	--	710.4290	790.9762	917.6866
Frequencies	--	937.5134	1001.7026	1128.9908
Frequencies	--	1202.7931	1348.7117	1412.6504
Frequencies	--	1508.6656	2383.2276	3196.3594
Frequencies	--	3327.1368	3861.6947	3881.6409

2-2product

Frequencies	--	24.7980	53.9114	91.7252
Frequencies	--	182.7856	419.9848	428.6693
Frequencies	--	442.6472	550.8454	613.1925
Frequencies	--	674.7585	895.0223	1023.2164
Frequencies	--	1074.5118	1248.5329	1365.8475
Frequencies	--	1436.4406	1471.1347	1480.5459
Frequencies	--	1852.8107	3086.9966	3161.4090
Frequencies	--	3192.4338	3664.4589	3825.7430

3reactant

Frequencies	--	90.6404	154.7992	185.5531
Frequencies	--	204.2937	277.0237	304.4426
Frequencies	--	395.3079	415.0083	527.8332
Frequencies	--	587.2433	606.0466	872.4493
Frequencies	--	1003.3325	1084.8679	1112.4426
Frequencies	--	1143.7540	1174.3760	1189.1739
Frequencies	--	1246.4165	1361.0424	1397.6657
Frequencies	--	1426.6546	1458.2707	1489.3118
Frequencies	--	1502.1302	1518.2655	3024.3236
Frequencies	--	3030.6338	3086.4904	3168.8219
Frequencies	--	3184.6472	3304.6123	3807.3817

TS3

Frequencies	--	-3353.5410	106.8077	140.4692
Frequencies	--	203.1877	317.4333	379.0618
Frequencies	--	421.0181	443.4890	510.2739
Frequencies	--	541.2179	705.8511	849.1668
Frequencies	--	894.6562	1008.9616	1043.3849
Frequencies	--	1079.7254	1145.0910	1177.5561
Frequencies	--	1187.7295	1214.7647	1263.1317
Frequencies	--	1371.7984	1463.5068	1488.1043
Frequencies	--	1503.1902	1516.2168	2951.5386
Frequencies	--	3011.5112	3062.5748	3164.2113

Frequencies	--	3175.8131	3299.8442	3867.2911
3product				
Frequencies	--	84.9368	151.9196	193.8173
Frequencies	--	247.4069	325.5466	372.4008
Frequencies	--	416.7327	467.5985	548.3928
Frequencies	--	704.1092	907.5440	972.6793
Frequencies	--	1050.3257	1150.4763	1169.5226
Frequencies	--	1183.7598	1196.9351	1215.4331
Frequencies	--	1259.6603	1300.7406	1365.5317
Frequencies	--	1408.5971	1482.6300	1494.7488
Frequencies	--	1511.1985	1515.5882	2939.9416
Frequencies	--	3046.2748	3072.7490	3115.0399
Frequencies	--	3175.8101	3178.7477	3867.4445
5reactant				
Frequencies	--	93.9635	138.1567	180.8401
Frequencies	--	201.2285	262.1757	324.1889
Frequencies	--	389.9435	421.4351	458.1070
Frequencies	--	478.7687	612.4818	728.5223
Frequencies	--	765.1765	833.3155	864.8330
Frequencies	--	894.4354	916.6683	929.1824
Frequencies	--	1006.3496	1040.3655	1066.6022
Frequencies	--	1078.2597	1091.9434	1139.2490
Frequencies	--	1168.5446	1198.7817	1276.0736
Frequencies	--	1304.3925	1343.0388	1353.5302
Frequencies	--	1366.3956	1376.9542	1390.0535
Frequencies	--	1392.4594	1395.2545	1417.6189
Frequencies	--	1466.3295	1473.4333	1494.4942
Frequencies	--	1503.4491	2977.0278	3046.9875
Frequencies	--	3050.4299	3072.5666	3074.9039
Frequencies	--	3084.9030	3104.6156	3112.2284
Frequencies	--	3124.8900	3186.1325	3849.6133
TS5				
Frequencies	--	-1644.6112	98.1503	145.5022
Frequencies	--	198.3531	291.4494	301.1606
Frequencies	--	383.2755	406.1628	450.6552
Frequencies	--	474.7208	587.4131	678.4493
Frequencies	--	736.8184	778.3394	840.5335
Frequencies	--	865.5082	878.5253	924.9305
Frequencies	--	954.3240	989.7485	1050.2041
Frequencies	--	1058.8924	1066.3850	1107.6176
Frequencies	--	1121.5112	1153.8329	1210.0632
Frequencies	--	1223.9287	1274.2971	1302.6150
Frequencies	--	1339.7467	1374.4344	1381.8174
Frequencies	--	1386.5177	1389.2491	1416.3879
Frequencies	--	1439.7369	1478.2555	1495.8909
Frequencies	--	1504.6156	2947.6305	2994.7265
Frequencies	--	3050.1082	3052.8566	3073.1601
Frequencies	--	3075.4376	3113.0521	3119.5534
Frequencies	--	3125.1889	3209.3164	3887.5605

5product

Frequencies	--	120.9339	146.1401	251.4844
Frequencies	--	271.3855	348.1585	376.4851
Frequencies	--	411.0405	462.2299	486.6225
Frequencies	--	595.1277	681.6280	775.4923
Frequencies	--	828.3427	850.4012	878.9459
Frequencies	--	905.8709	952.5543	1020.3379
Frequencies	--	1026.2679	1050.7002	1112.2113
Frequencies	--	1116.9735	1152.7739	1181.9429
Frequencies	--	1211.6326	1248.1626	1263.6229
Frequencies	--	1293.6492	1301.5498	1331.4436
Frequencies	--	1335.5712	1356.8135	1375.2616
Frequencies	--	1387.2601	1394.2696	1436.9304
Frequencies	--	1479.2344	1485.8742	1501.2427
Frequencies	--	1510.6951	3006.8319	3037.4427
Frequencies	--	3052.0896	3078.8258	3080.1308
Frequencies	--	3094.3914	3111.5439	3123.8979
Frequencies	--	3128.3283	3146.2668	3856.2970

6reactant

Frequencies	--	100.5033	129.1444	164.7855
Frequencies	--	177.9654	256.7105	280.6461
Frequencies	--	385.9478	406.0604	430.1018
Frequencies	--	488.1940	538.5475	744.6091
Frequencies	--	777.2369	841.2743	867.2523
Frequencies	--	894.1891	896.9302	947.9071
Frequencies	--	1002.9967	1034.3154	1049.7511
Frequencies	--	1081.7597	1137.7266	1140.3631
Frequencies	--	1155.4241	1198.1639	1274.2638
Frequencies	--	1295.1199	1315.9565	1328.1543
Frequencies	--	1352.8307	1369.7462	1382.0211
Frequencies	--	1386.9576	1400.8457	1426.3252
Frequencies	--	1462.3374	1474.1892	1485.0407
Frequencies	--	1498.5305	2992.5884	2998.0784
Frequencies	--	3043.4011	3069.8533	3080.8957
Frequencies	--	3096.2419	3113.3489	3123.3991
Frequencies	--	3131.9312	3229.0706	3855.4204

TS6

Frequencies	--	-2111.1823	148.3433	235.1455
Frequencies	--	252.0870	304.6727	339.5359
Frequencies	--	382.0866	400.2804	426.7500
Frequencies	--	524.7395	551.6211	663.3613
Frequencies	--	734.4559	804.9066	855.2690
Frequencies	--	859.7320	883.7300	904.7584
Frequencies	--	975.3778	1004.7350	1032.5982
Frequencies	--	1041.9054	1084.9084	1105.9892
Frequencies	--	1109.0771	1139.3238	1197.6189
Frequencies	--	1240.9189	1277.2839	1326.2148
Frequencies	--	1339.2850	1347.2271	1360.7557
Frequencies	--	1383.0202	1390.0526	1403.6324
Frequencies	--	1461.7533	1468.5938	1482.7410
Frequencies	--	1504.5818	2973.3412	3017.1307
Frequencies	--	3037.3260	3050.4494	3075.9223
Frequencies	--	3103.0174	3107.2982	3117.5674

Frequencies	--	3126.5813	3202.5916	3767.4919
6product				
Frequencies	--	145.9965	193.9363	280.0503
Frequencies	--	323.6979	359.5668	394.5738
Frequencies	--	423.7222	433.2925	511.6757
Frequencies	--	577.7863	699.4126	770.3358
Frequencies	--	835.3385	853.9004	863.1033
Frequencies	--	886.4663	947.6827	976.2977
Frequencies	--	1013.8341	1036.9596	1083.3918
Frequencies	--	1090.3876	1106.0383	1146.8025
Frequencies	--	1187.6971	1241.2975	1269.3491
Frequencies	--	1277.6242	1301.6213	1327.1106
Frequencies	--	1348.5544	1363.9869	1378.7230
Frequencies	--	1394.0475	1412.4649	1437.0614
Frequencies	--	1465.6646	1477.1962	1492.3172
Frequencies	--	1503.0022	3007.4831	3047.9961
Frequencies	--	3054.5124	3070.9324	3073.6790
Frequencies	--	3084.9278	3110.5819	3114.4571
Frequencies	--	3118.7707	3139.9738	3831.0121

Appendix F: M06-2X/6-311++G(d,p) frequencies of propyl systems (cm⁻¹).

9reactant/9-22reactant				
Frequencies	--	54.9228	117.2363	158.7089
Frequencies	--	180.3643	244.5609	395.0853
Frequencies	--	418.1970	613.4150	897.3526
Frequencies	--	907.9319	955.9129	996.7870
Frequencies	--	1047.0856	1146.7354	1159.9698
Frequencies	--	1291.4273	1373.6078	1385.2289
Frequencies	--	1406.5548	1420.1829	1470.3341
Frequencies	--	1479.1574	1490.3505	3021.2838
Frequencies	--	3035.9773	3080.8231	3096.1285
Frequencies	--	3138.7340	3220.9646	3860.1502
TS9				
Frequencies	--	-1637.9893	129.3294	158.1825
Frequencies	--	204.8463	293.6728	391.4578
Frequencies	--	410.3579	552.1176	734.5471
Frequencies	--	834.5185	927.1390	950.9434
Frequencies	--	1017.7321	1078.6461	1161.4663
Frequencies	--	1165.5382	1211.4050	1311.5617
Frequencies	--	1371.2929	1424.9381	1461.3727
Frequencies	--	1477.2631	1499.2022	2978.8111
Frequencies	--	3004.8084	3051.9878	3102.1514
Frequencies	--	3154.6793	3211.5965	3887.4457
9product				
Frequencies	--	145.2608	224.9649	255.4696
Frequencies	--	361.3161	444.9897	471.0767
Frequencies	--	535.6387	807.8801	861.7841
Frequencies	--	949.7601	1053.4704	1083.3053
Frequencies	--	1099.7174	1178.9787	1206.1786

Frequencies	--	1290.3872	1301.0730	1365.7701
Frequencies	--	1389.4780	1402.3701	1447.0183
Frequencies	--	1488.4249	1505.6199	2931.8770
Frequencies	--	3016.5852	3031.8824	3067.6630
Frequencies	--	3142.9636	3157.7500	3860.4447

TS9-22

Frequencies	--	-905.8659	62.1018	151.5825
Frequencies	--	190.1748	195.4832	269.4643
Frequencies	--	306.3600	424.7664	655.3347
Frequencies	--	664.8892	796.4329	903.8932
Frequencies	--	942.9357	982.6491	1063.3684
Frequencies	--	1128.6269	1249.1710	1293.8776
Frequencies	--	1367.5059	1397.9832	1429.8628
Frequencies	--	1461.4397	1469.7346	2678.3516
Frequencies	--	3002.1692	3014.9732	3080.6137
Frequencies	--	3168.2981	3198.1221	3874.8332

9-22product1

Frequencies	--	142.4716	235.4660	264.6445
Frequencies	--	671.5778	676.8170	871.4291
Frequencies	--	905.2937	1008.1648	1120.0020
Frequencies	--	1152.7801	1282.0674	1372.0026
Frequencies	--	1413.0607	1427.9953	1457.4228
Frequencies	--	1498.6071	1504.7635	1866.2973
Frequencies	--	2944.5214	3049.3176	3078.4716
Frequencies	--	3080.3699	3155.6148	3157.1796

OH

Frequencies	--	3791.3262
-------------	----	-----------

8reactant

Frequencies	--	-17.0689	104.9061	131.0604
Frequencies	--	168.7153	280.3456	385.2386
Frequencies	--	463.8027	555.5004	755.5831
Frequencies	--	889.2136	943.8851	1038.7686
Frequencies	--	1093.4770	1107.8433	1155.4026
Frequencies	--	1248.7185	1317.7966	1329.3857
Frequencies	--	1396.9289	1425.5793	1467.2922
Frequencies	--	1479.6631	1487.4042	3017.8064
Frequencies	--	3049.2994	3101.9037	3125.6728
Frequencies	--	3171.6253	3279.7018	3857.8339

TS8

Frequencies	--	-2302.0403	177.6279	309.9302
Frequencies	--	376.8441	417.2915	456.7378
Frequencies	--	517.2297	549.9250	800.9400
Frequencies	--	881.0847	890.4918	929.0770
Frequencies	--	1018.3940	1046.6534	1084.8021
Frequencies	--	1173.1477	1220.1502	1275.2487
Frequencies	--	1341.9102	1389.6706	1460.1862

Frequencies	--	1476.1752	1493.8029	3024.6123
Frequencies	--	3031.2548	3080.1077	3103.4477
Frequencies	--	3162.1461	3268.5177	3817.3895

8product

Frequencies	--	90.8349	194.3094	261.6371
Frequencies	--	278.4277	423.2728	539.9612
Frequencies	--	679.2851	857.4001	902.9635
Frequencies	--	994.1940	1070.4007	1079.1517
Frequencies	--	1103.5387	1153.7946	1230.7941
Frequencies	--	1260.1809	1318.5766	1338.6287
Frequencies	--	1386.7025	1390.5177	1456.2137
Frequencies	--	1471.6596	1527.6024	2943.3456
Frequencies	--	3015.1778	3040.4375	3070.4506
Frequencies	--	3085.6586	3122.6229	3921.6317

4reactant/4-2reactant

Frequencies	--	96.0736	156.6839	224.6578
Frequencies	--	230.9351	291.9919	343.3192
Frequencies	--	452.2597	491.3033	569.5799
Frequencies	--	859.4024	925.1691	938.7899
Frequencies	--	1036.2950	1081.9259	1176.3800
Frequencies	--	1190.2371	1339.8936	1382.6862
Frequencies	--	1400.7675	1406.9540	1462.0035
Frequencies	--	1488.3815	1500.5468	3071.6710
Frequencies	--	3083.2047	3152.6339	3158.5068
Frequencies	--	3167.7952	3285.2069	3831.6297

TS4

Frequencies	--	-2757.1288	161.2912	230.6796
Frequencies	--	358.7476	368.7994	403.0927
Frequencies	--	457.4167	512.9285	620.1167
Frequencies	--	842.2562	874.6074	932.4731
Frequencies	--	954.2761	1013.6832	1063.8265
Frequencies	--	1105.9031	1193.8926	1241.4659
Frequencies	--	1347.0276	1408.7092	1470.2198
Frequencies	--	1493.0707	1502.1829	2989.8975
Frequencies	--	3067.9606	3145.7949	3161.6019
Frequencies	--	3165.3583	3284.6558	3878.9763

4product

Frequencies	--	136.4161	199.6087	252.9285
Frequencies	--	365.1578	433.5410	457.5183
Frequencies	--	523.7577	845.2331	892.9553
Frequencies	--	941.4572	1043.1778	1118.4303
Frequencies	--	1130.7295	1147.1070	1178.5803
Frequencies	--	1212.9916	1290.9612	1361.9938
Frequencies	--	1398.8079	1426.5496	1489.5094
Frequencies	--	1499.2799	1513.8741	2922.2915
Frequencies	--	3025.3125	3070.6209	3126.8259
Frequencies	--	3146.1515	3162.3586	3860.3596

TS4-2

Frequencies	--	-1537.7731	102.4331	171.0155
Frequencies	--	185.7567	209.9395	324.3020
Frequencies	--	394.8929	478.8801	486.9993
Frequencies	--	623.6646	767.0262	846.7825
Frequencies	--	903.7445	955.3184	1025.1933
Frequencies	--	1060.9384	1128.3521	1291.9503
Frequencies	--	1391.8699	1401.4809	1458.6431
Frequencies	--	1485.9943	1506.7098	2458.8835
Frequencies	--	3076.6420	3157.6481	3169.7553
Frequencies	--	3176.9498	3303.1034	3870.6446

4-2product1

Frequencies	--	45.6309	132.4199	382.0581
Frequencies	--	488.4321	541.5059	804.9754
Frequencies	--	882.0415	898.2109	1085.5339
Frequencies	--	1121.4370	1249.5419	1388.3001
Frequencies	--	1398.2947	1465.3664	1468.1872
Frequencies	--	1475.7869	1491.7736	1857.4396
Frequencies	--	3063.9396	3069.1408	3129.9880
Frequencies	--	3136.1421	3182.7538	3183.6816