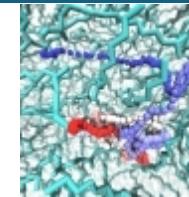
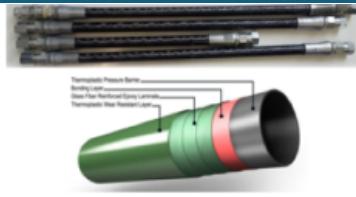




Sandia
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H-Mat: Task P2 (SNL) Computational Polymer modeling Accomplishments



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¹Computational Materials and Data Science (org 1864)

²Nanostructure Physics (org 1881)

1

FY21 Accomplishments



Two classes of hydrogen gas are observed in pressurized EPDM

The H_2 gas with slower diffusional dynamics localizes away from crosslinks

Increasing crosslinks can:

- Slow the structural rearrangement of EPDM after decompression
- Increases gas mobility in the EPDM

Publications / presentations:

- *High-Pressure Hydrogen Decompression in Sulfur Crosslinked Elastomers*; Wilson M., Frischknecht, A. (In technical review)
- *Sub-diffusive High-Pressure Hydrogen Gas Dynamics in Elastomers*; Brownell M., Frischknecht A., Wilson M. (in technical review)
- *MRS Fall Meeting 2021: Atomistic modeling of elastomer performance during exposure to high-pressure hydrogen*, Brownell M., Frischknecht A., Wilson M.

Atomistic modeling of EPDM and H₂



SUCCESSES:

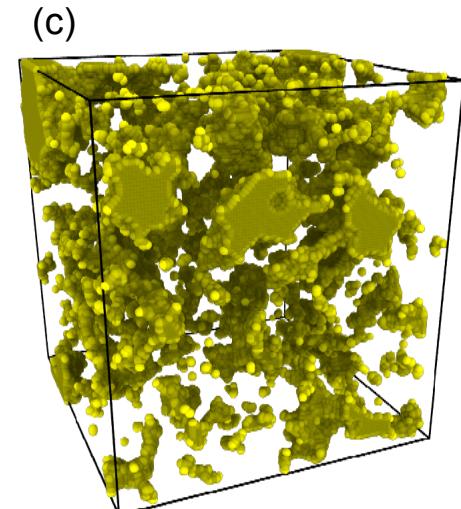
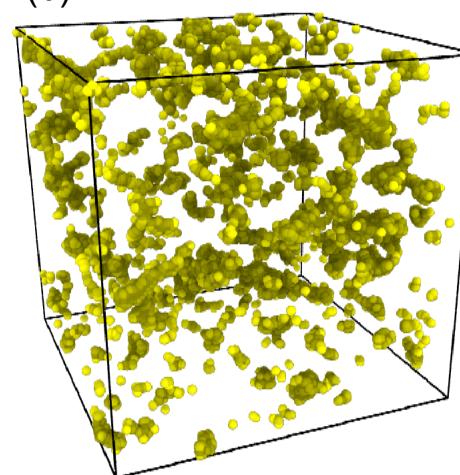
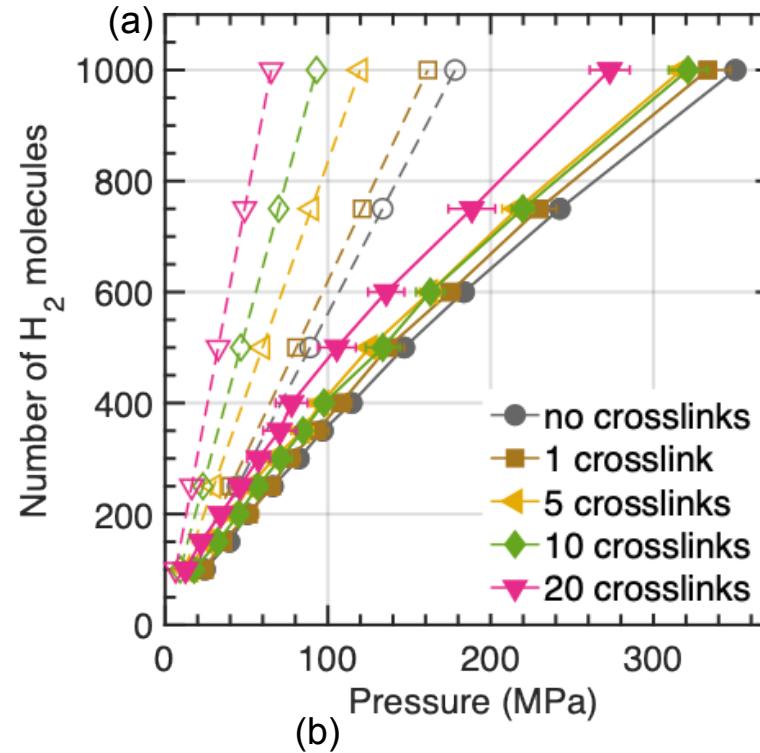
- Calculated gas solubility-pressure relationships in dilute limit.
- Free volume is calculated before and after decompression and found to correlate to crosslink density.

METHOD:

- Molecular dynamics simulations are performed for all-atom representations of EPDM
- Various degrees of crosslinking are imposed in initial configurations
- H₂ gas is inserted into EPDM configurations at various pressures
- Simulations model rapid decompression to 1 atm

SIGNIFICANCE:

- Structural rearrangement of polymer chains is slowed with increased content of crosslinks
- Suggests a composition with increased percentage of crosslinks could reduce H₂ susceptibility to bubble formation



(a) Plot of the pressure of each model vs the number of H₂ molecules. Solid lines show our results while dashed lines show Henry's law. (b) Hydrogen accessible free volume of the 5 crosslinks EPDM model with 1000 H₂ molecules at pressure and (c) after decompression.

Sub-diffusive high pressure H₂ gas dynamics



SUCCESSES:

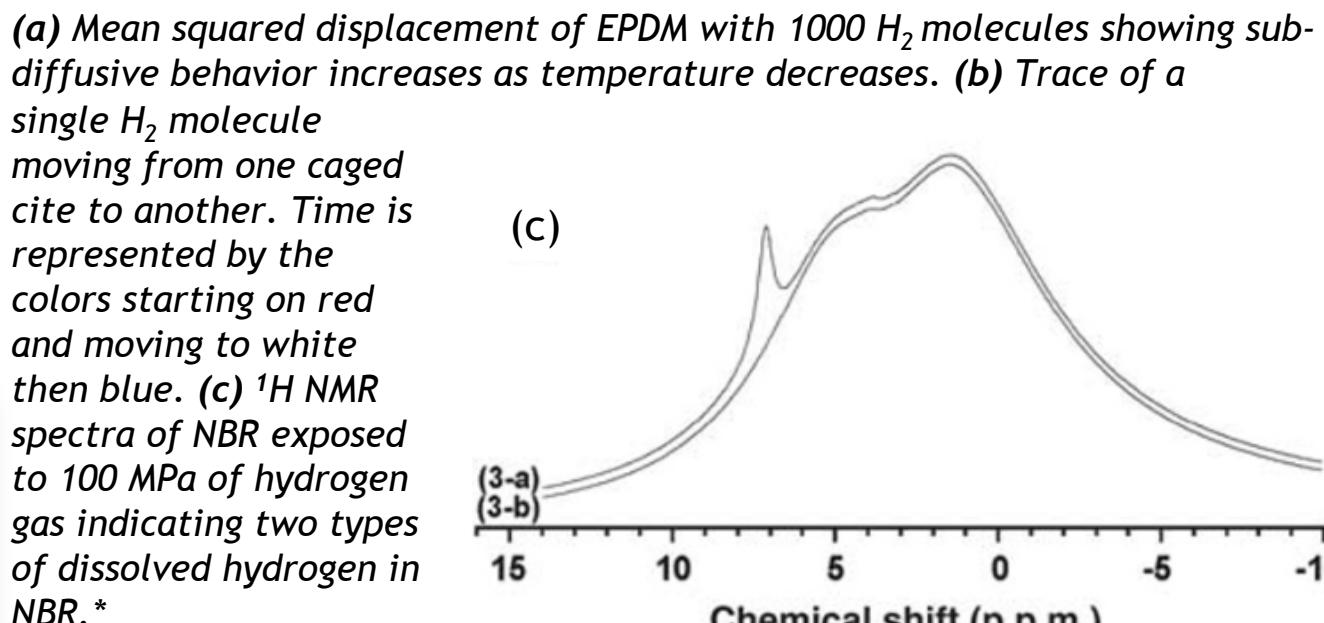
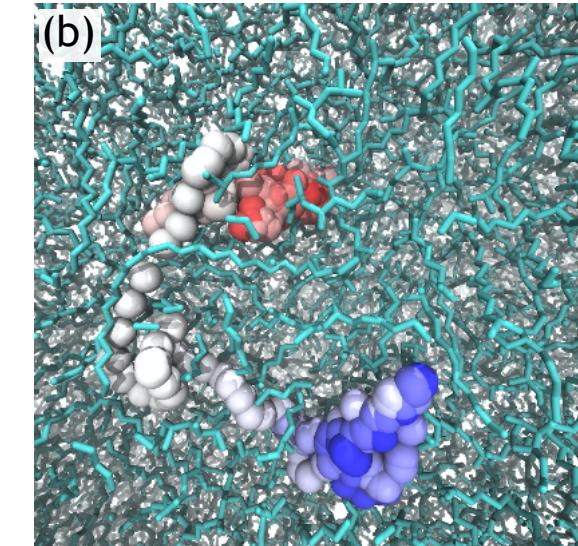
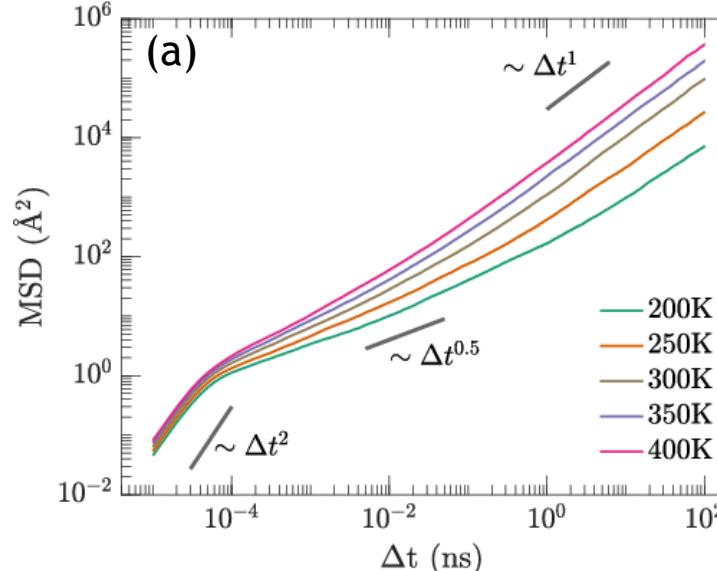
- Identified sub-diffusive behavior of H₂ which was larger at temperatures below glass transition temperature.
- Showed that the slower moving H₂ atoms were localizing away from sulfur on the crosslinks.

METHOD:

- Movement of H₂ gas is measured and analyzed over 2 ns of simulations time.
- Displacement of H₂ gas is measured to classify caged/mobile hydrogen.
- RDF is used to measure distribution of hydrogen throughout EPDM.

SIGNIFICANCE:

- Greater gas mobility occurs with greater crosslinks, higher temperatures, and lower gas concentrations.
- Caged hydrogen exhibits slow dynamics and localization, possible precursors to cavitation.



Analysis of dynamic free volume



SUCCESSES:

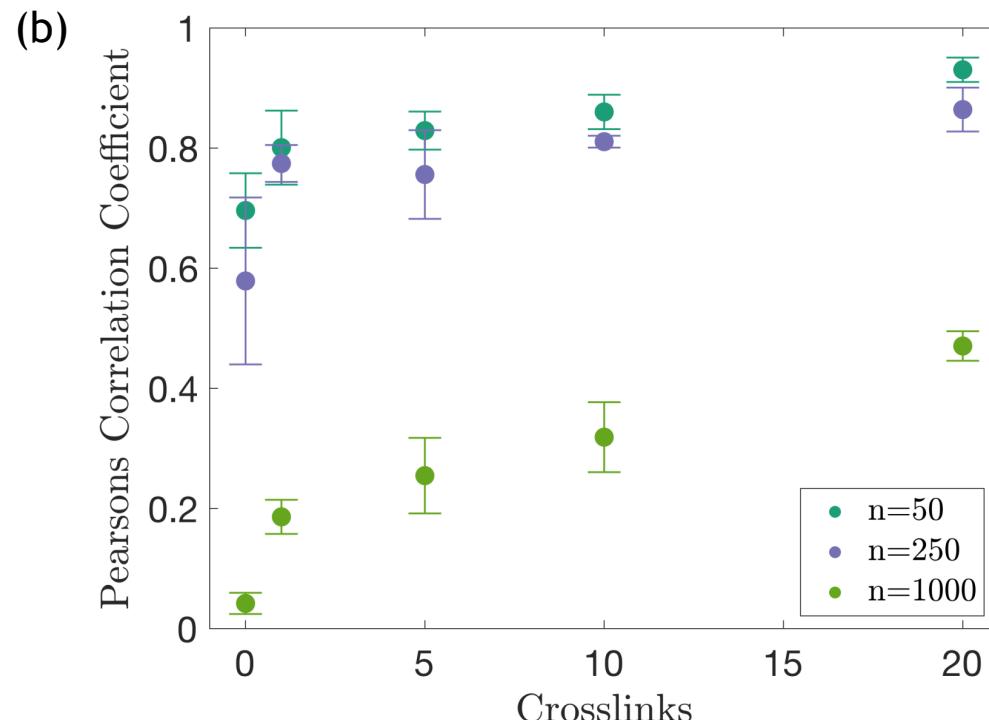
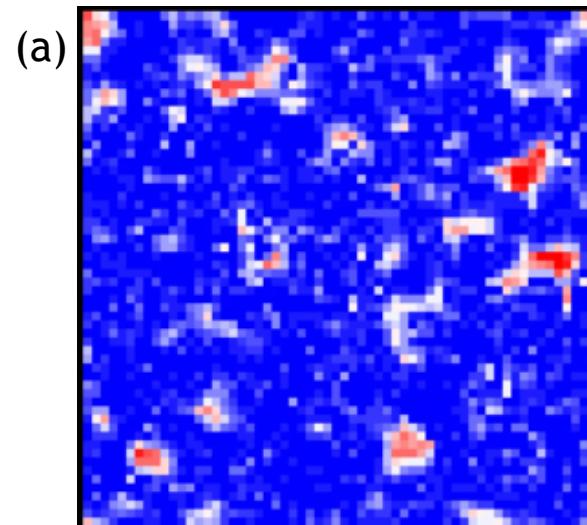
- Developed postprocessing method to analyze free volume much faster than previously
- Observed dynamic behavior of free volume and identified persistent areas of free volume
- Identified a correlation between free volume before and after decompression which increases with greater crosslinking

METHOD:

- A new tool was created to analyze the free volume of a system and to compare the free volume with atom displacements and locations.

SIGNIFICANCE:

- Increasing crosslinks increases the correlation of free volume before and after decompression.
- Now able to know how free volume is created during decompression



(a) Video of a 2D slice of free volume movement with time. The color map indicates the average time a pixel is free volume. (b) Pearson's correlation coefficient of the free volume before and after decompression, showing increased correlation with increased crosslinks.