

H05-2025: Nanoporous-Based Sensors for Electrical Detection of Gaseous Pollutants



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Introduction

- The ability to sense and identify **individual gaseous pollutants** from the complexity of the environment requires highly selective materials
 - Avoidance of interference from real-world air components
- Current conductivity-based devices generally fall into two categories:
 - Solid state – (oxide based) require higher temperatures ($>200^{\circ}\text{C}$) for interaction of the gas with the surface oxides; heating devices are needed
 - Fuel cell – room temperature liquid electrolyte, easily fouled, short lifetime
- Electrical **metal organic framework (MOF) based sensors** have previously been used for direct electrical sensing of gases; however, none for **NO₂** have been reported in open literature
- By tuning the composition of MOFs, selective chemical adsorption and/or catalysis can be achieved
- Typical sensors for this application are hard-wired or require frequent battery replacement– nanoporous MOFs allows for “near-zero” long lived sensing in a wider range of environments

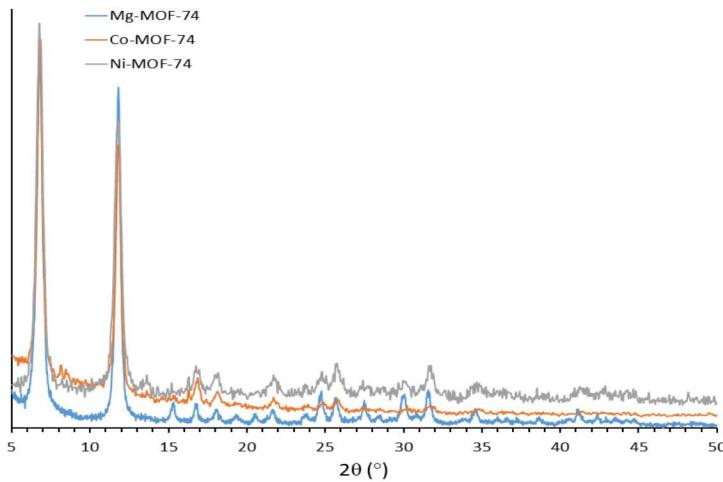
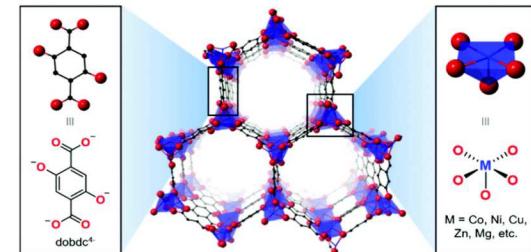
Direct Electrical Readout Sensors Combined with Nanoporous Adsorption Materials

Reasoning for this construction:

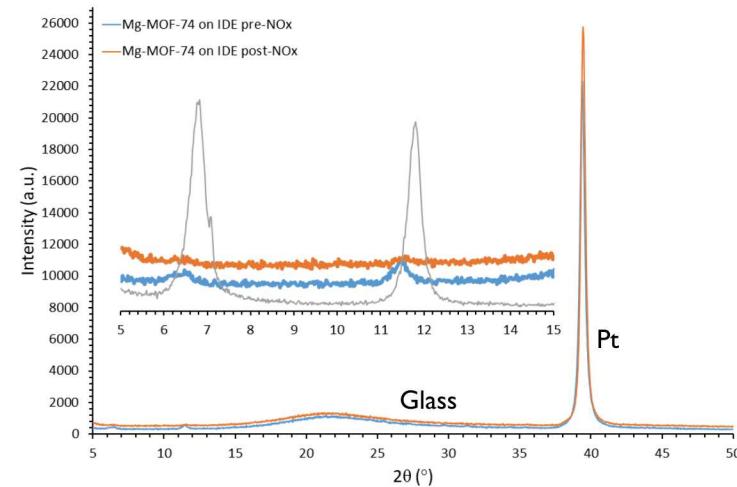
- Increase the chemical sensitivity of the overall sensor through the incorporation to its platform of nanoporous materials
- Nanoporous adsorption materials chosen for ability to selectively adsorb target gas molecules
- Electrical readout sensor of this design:
 - Decreased power consumption
 - Ability to interrogate for specified gases selectively in real-time or as an integrating sensor for delayed/later testing
- Design of an integrated sensor:
 - Record whether any degradation product was ever present during the sensor's lifetime
- Integrated sensor is useful in cases where degradation products may:
 - Subsequently react with other components,
 - Gradually leak out of the system,

Nanoporous Materials Targeted for the Selective Adsorption of NO_x

- M-MOF-74 (M= Co, Mg, Ni) was targeted for its selectivity to NO_2
- MOF-74 materials were synthesized and investigated as bulk materials and dropcast onto an interdigitated electrode (IDE)
- Each powder pattern highlighted two primary diffraction peaks corresponding to the MOF pore (intensities reduced for dropcast samples, with the large peak corresponding to the platinum IDE)

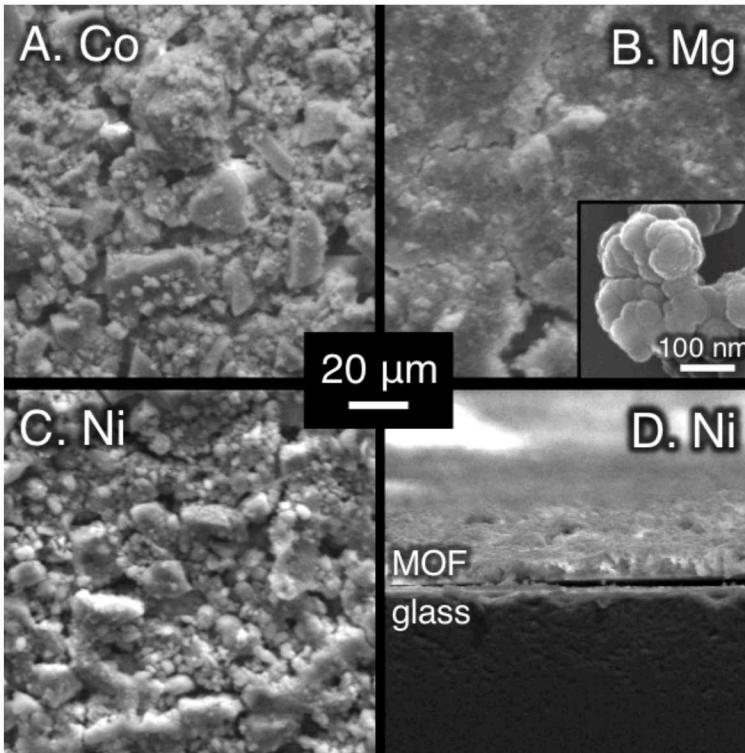


Powder XRD patterns for as-synthesized MOF-74 in the bulk phase.



Powder XRD patterns for Mg-MOF-74 dropcast onto IDE pre- NO_2 (blue) and post- NO_2 (orange). Inset: zoomed in region compared to bulk powder Mg-MOF-74.

SEM Characterization of Dropcast M-MOF-74 Thin Films

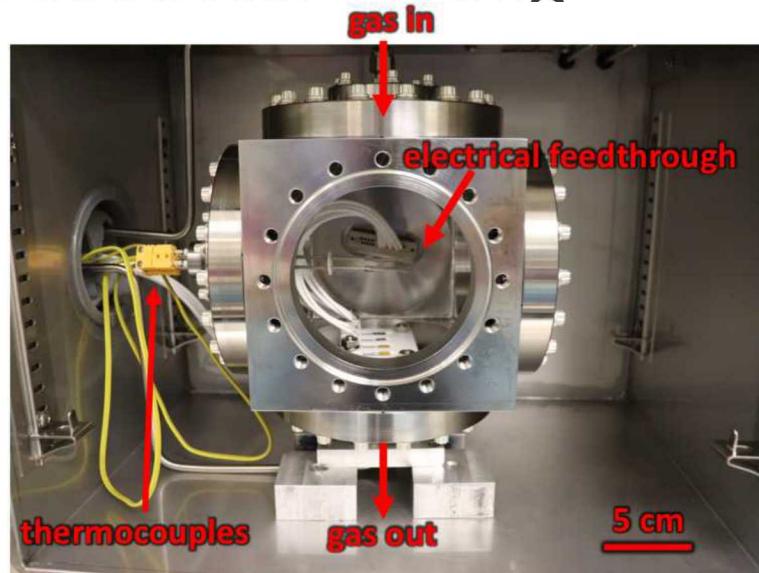


Plan-view SEM micrographs of (A) Co-MOF-74, (B) Mg-MOF-74, (C) Ni-MOF-74 powders dropcast onto IDEs. (D) Cross-sectional micrograph of Ni-MOF-74 film from (C).

- Co- and Ni-MOF-74 contained a wide range of crystallite sizes, from 100's of μm to 100 nm
- Mg-MOF-74 crystallites were on the order of 100 nm
- Film thickness was $\sim 10 \mu\text{m}$

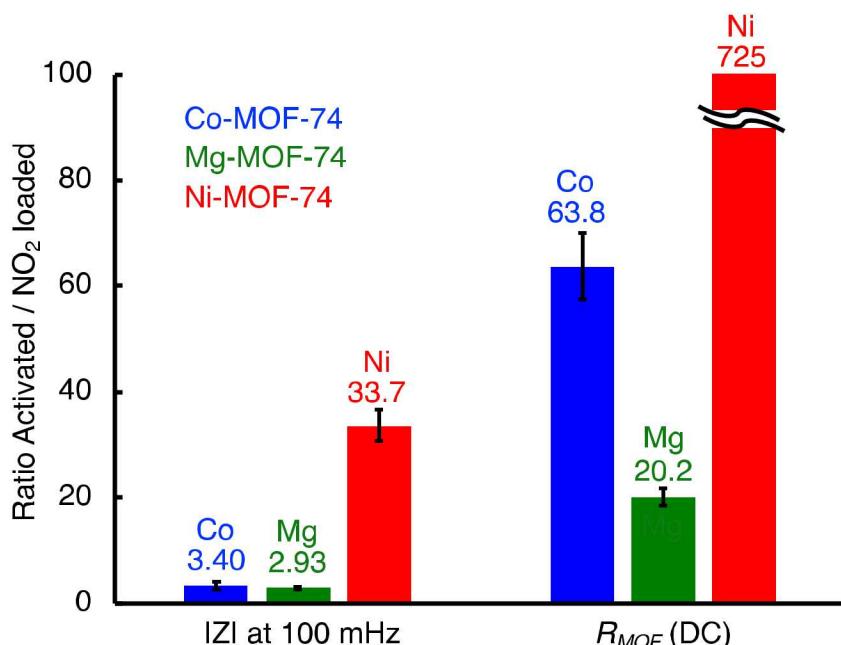
NO_x Exposure and In Situ Electrical Testing

- Custom-built NO_x exposure chamber enabled MOF activation and subsequent in situ electrical testing under varying NO₂ concentrations without exposure to lab atmospheres
- Variable NO₂ concentrations (0.5-5 ppm) were achieved by diluting 5 ppm NO₂ gas stream with pure UHP N₂ at 500 sccm total gas flow
- Impedance spectra recorded at 0 V DC and 100 mV (RMS) AC over 1 MHz - 10 mHz
- All electrical measurements and NO₂ exposures occurred at 50°C

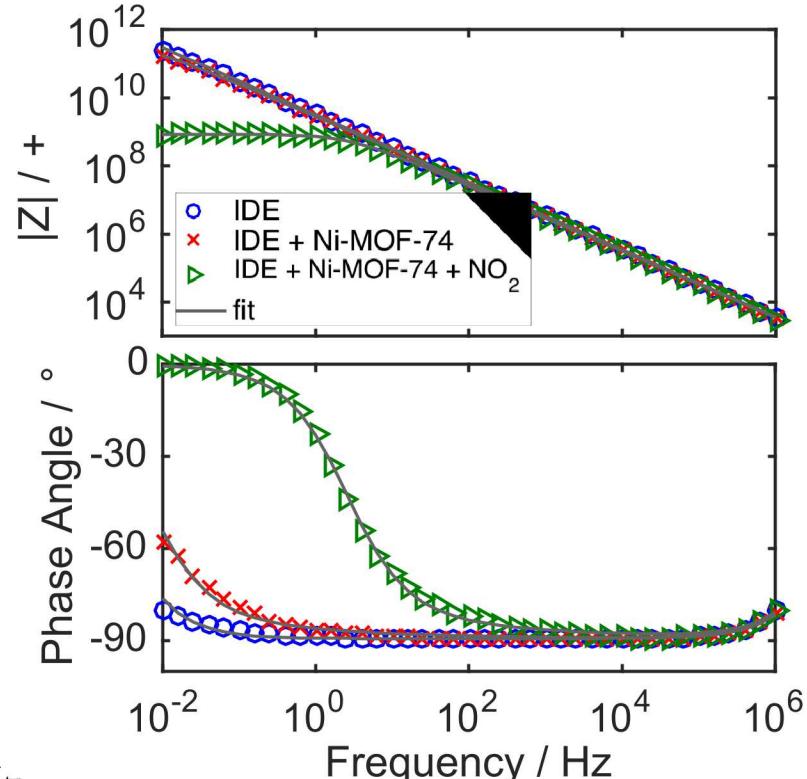


Typical Impedance Responses of M-MOF-74-Based Sensors

- Exposed M-MOF-74-based sensors to 5 ppm NO_2 for 8 h at 50° C.



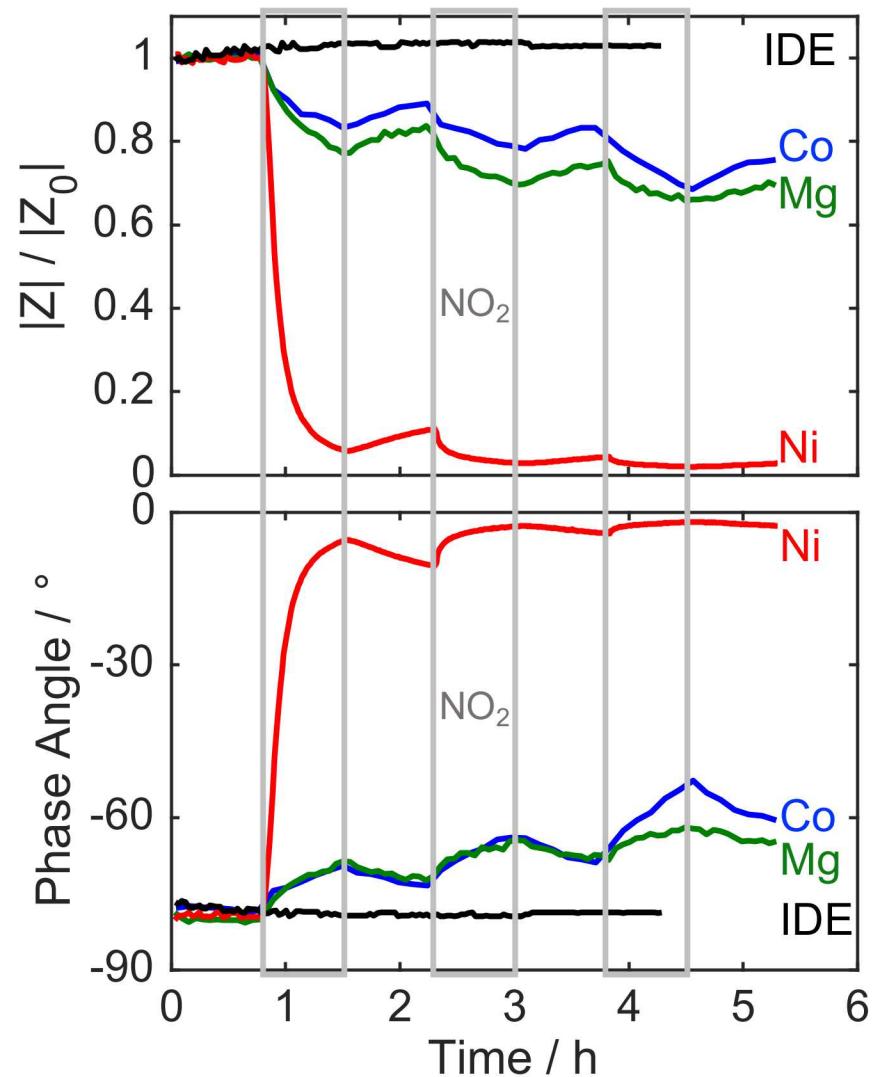
Ratio of response as-activated to NO_2 -exposed for (1) impedance magnitude ($|Z_{\text{activated}}| / |Z_{\text{NO}_2}|$) at 100 mHz and (2) MOF DC film resistance ($R_{\text{activated}} / R_{\text{NO}_2}$) for IDEs coated with M-MOF-74 (M= Co, Mg, Ni).



Example impedance spectra for Ni-MOF-74-based sensor

Impedance Responses as a Function of NO_2 Concentration

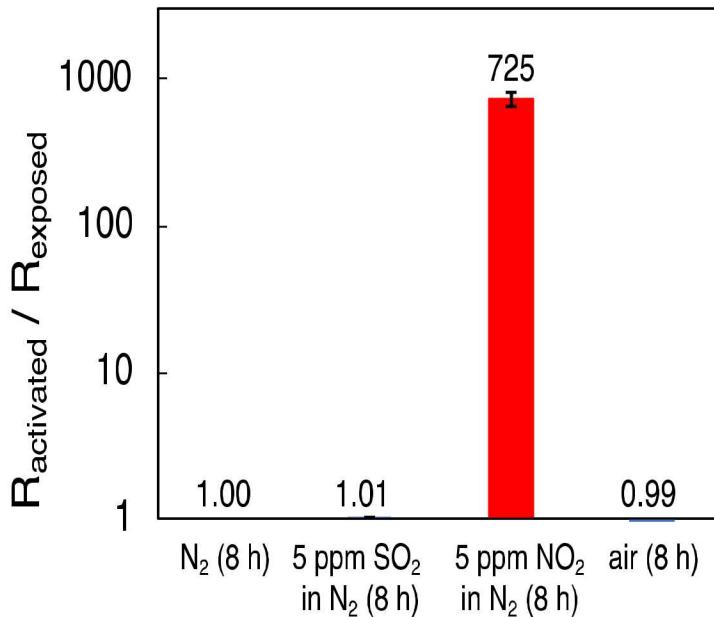
- Blank IDEs and IDEs coated in M-MOF-74 (M= Co, Mg, Ni) were activated and exposed to alternating 0.75 h flows of pure N_2 or N_2 containing trace NO_2 , while impedance was constantly measured at 100 mHz
- The magnitude of the impedance change observed is ordered Ni > Co > Mg
 - Explained by each variant's NO_2 adsorption capacity and specific chemical interaction
- Use of Ni-MOF-74 provided the highest sensitivity to NO_2 , with a 725× decrease in resistance at 5 ppm NO_2 and a NO_2 detection limit <0.5 ppm



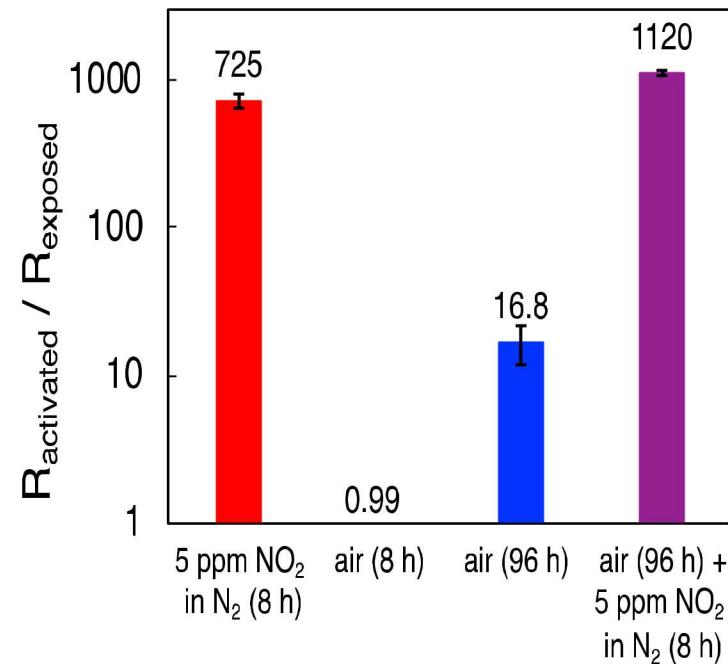
NO₂ Selectivity

- A Ni-MOF-74-based sensor was activated and exposed to 5 ppm SO₂ in N₂, and ambient air (25 °C, 50% RH, 400 ppm CO₂) heated to 50 °C, and its response compared to previous exposures to 5 ppm NO₂ in N₂
- An extended air exposure (96 hours) followed by subsequent NO₂ exposure was also performed
- The Ni-MOF-74-based sensor demonstrated selectivity to NO₂ versus N₂, SO₂, and air.

A. Ni-MOF-74



B. Ni-MOF-74



Conclusions

- M-MOF-74 (M = Co, Mg, Ni)-based sensors for selective detection of trace (0.5–5 ppm) NO_2 were successfully demonstrated
- Ni-MOF-74-based sensor exhibited a superior electrical response in its selectivity to NO_2 over interfering gases such as N_2 , SO_2 , and ambient air
- Differences in electrical response to NO_2 between the M-MOF-74 analogues were attributed to both the adsorption capacity and chemical interactions between the NO_2 and MOF
- The magnitude of the electrical response observed is ordered $\text{Ni} > \text{Co} > \text{Mg}$, with Ni-MOF-74 providing $725\times$ decrease in resistance at 5 ppm NO_2 and a NO_2 detection limit <0.5 ppm
- The high impedance of these materials enables applications requiring an ultralow power sensor or dosimeter, with the active material dissipating <15 pW, despite being a macroscale device

Future Work

- Investigations are underway for the following:
 - To increase mechanical durability of these sensors via stronger adhesion of nanoporous materials to the IDE
 - To incorporate new phases onto the sensors for selective detection of relevant competing gases
 - To evaluate sensor response over a range of environmental conditions (temp., humidity) as a function of time