

CENTER FOR ELECTROCHEMICAL SCIENCE AND ENGINEERING

Department of Materials Science and Engineering



Sensor Design, Construction and Validation for in-situ Water Layer Thickness Determination during Accelerated Corrosion Testing

R. M. Katona,^{1,2} S. Tokuda,³ and **R. G. Kelly**¹

¹Materials Science and Engineering, University of Virginia, Charlottesville, VA 22904 USA

²Sandia National Laboratories, Albuquerque, NM 87123 USA

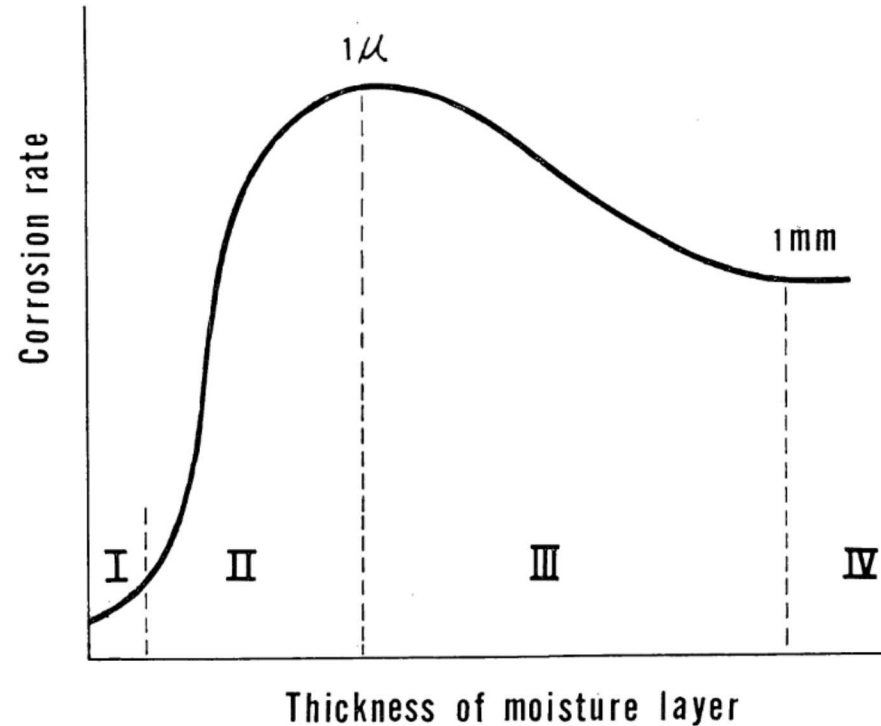
³Department of Materials Science, Tohoku University, Aoba-ku, Sendai 980-8579, Japan

Overview

- Motivation and Background
- Objective
- Modeling approach and Results
- Results and Discussion
- Implications and future work

Motivation

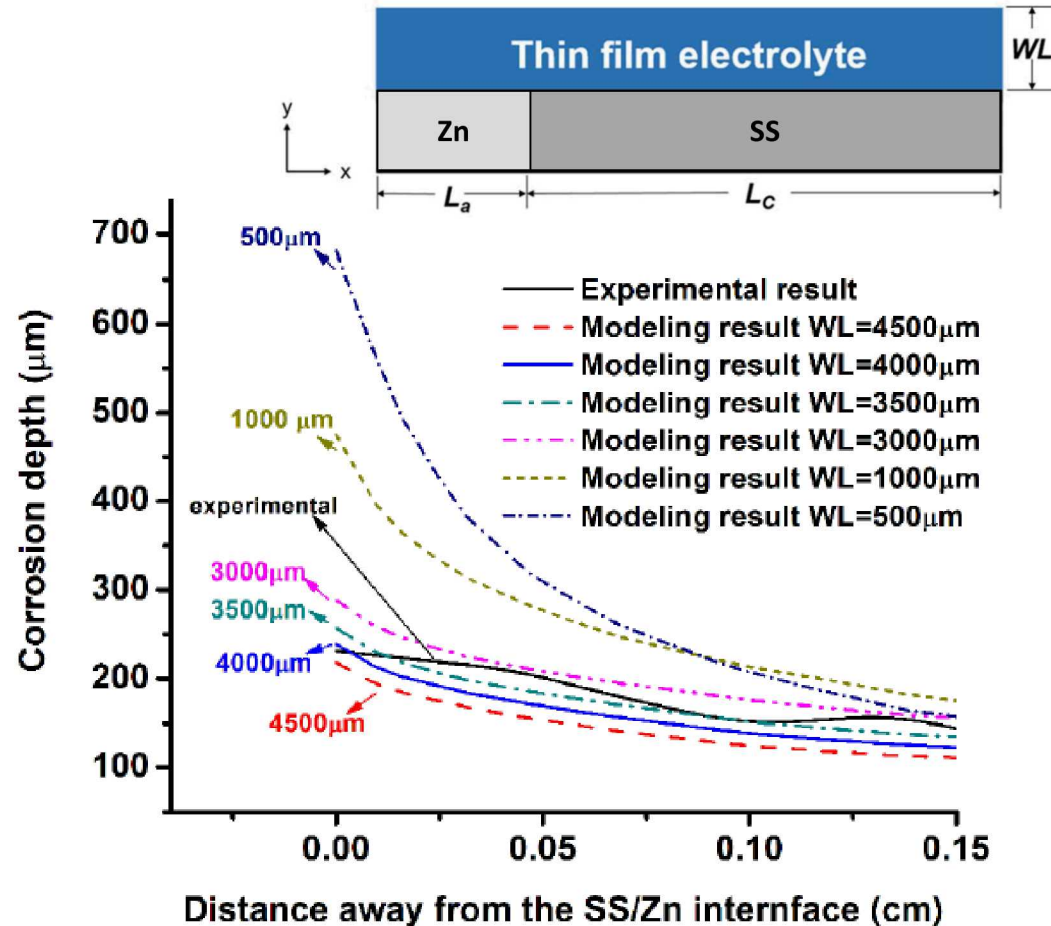
- The importance of the WL in corrosion testing has been long recognized



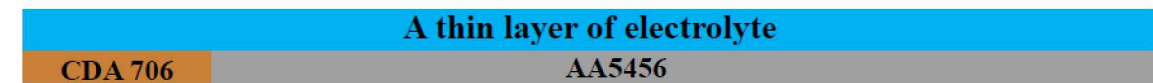
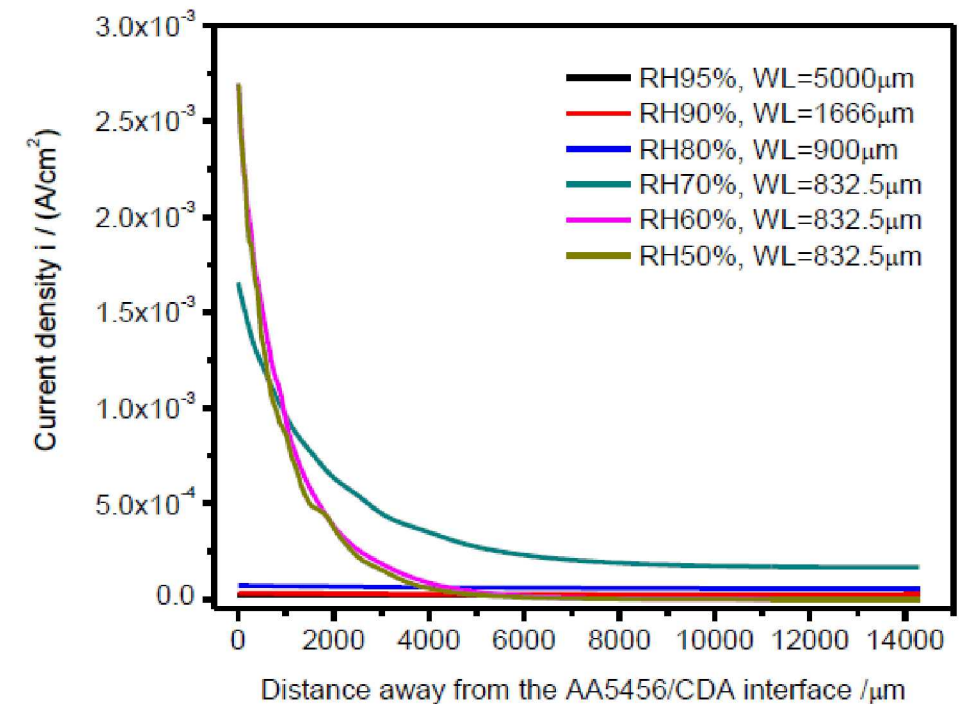
N.D. Tomashov,, CORROSION. 20 (1964) 7t-14t.

Motivation

- In galvanic couples, WL defines current distribution



Liu, C., PhD Thesis, UVA



Blohm, L. Masters Thesis, UVA

Motivation

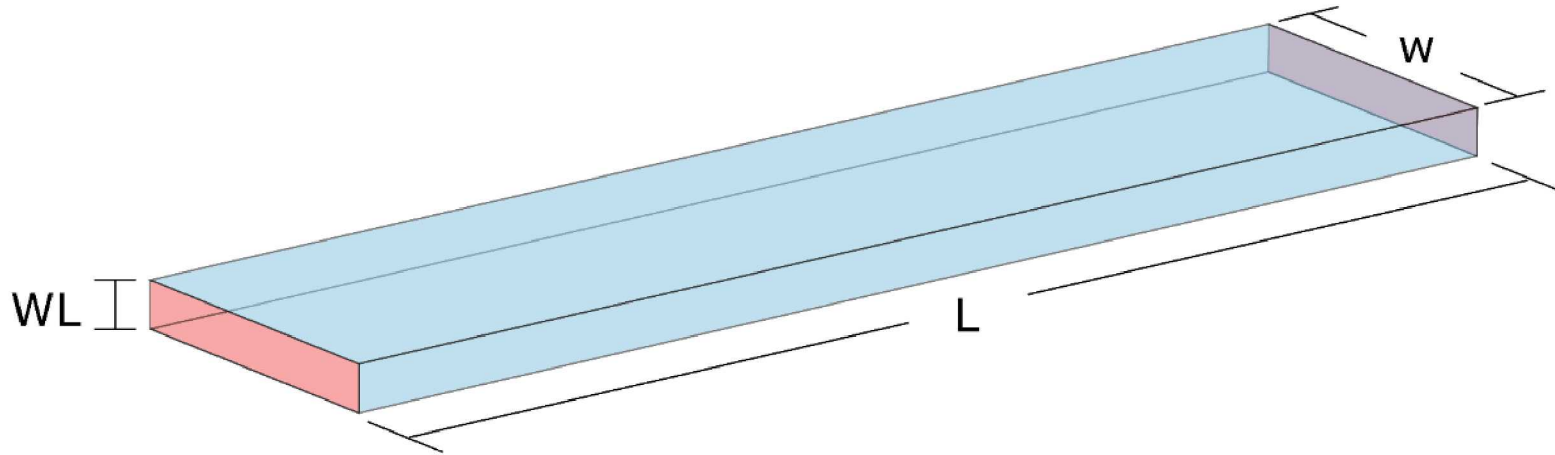
- Ability to measure WL in accelerated testing needed to understand corrosion damage distribution and improve test-to-test and chamber-to-chamber variability
- Thus a resistance based sensor will be created in which WL can be measured in a salt spray environment

The presented work is based on recently published paper:

- Katona, R.M.; Tokuda, S.; Perry, J.; Kelly, R.G., Design, “Construction, and Validation for in-situ Water Layer Thickness Determination during Accelerated Corrosion Testing,” *Corros. Sci.* 175 (2020) 108849.
<https://doi.org/10.1016/j.corsci.2020.108849>.

Governing Aspects of the Sensor

- For a parallelepiped WL, we know that WL is related to the solution resistance (R_s), length (L), width (W), and conductivity (κ) of the solution

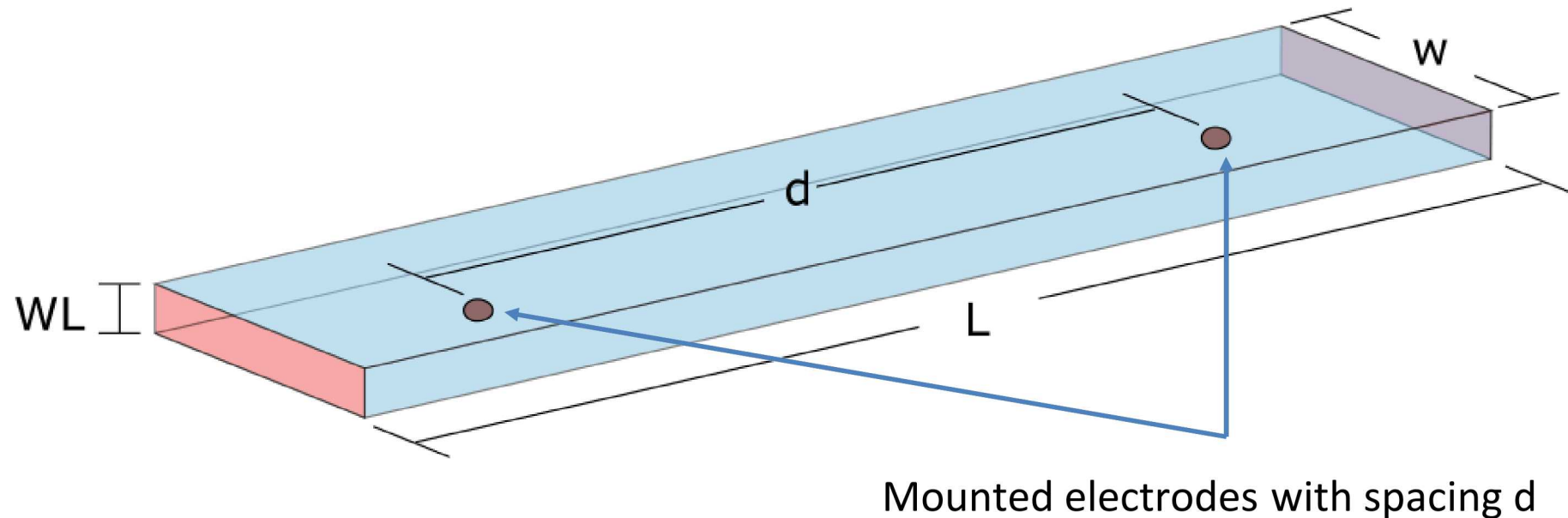


$$WL = \frac{L}{R_s * W} * \frac{1}{\kappa}$$

- Applies to measurement from end-to-end
- WL formation would be influenced by the walled sides

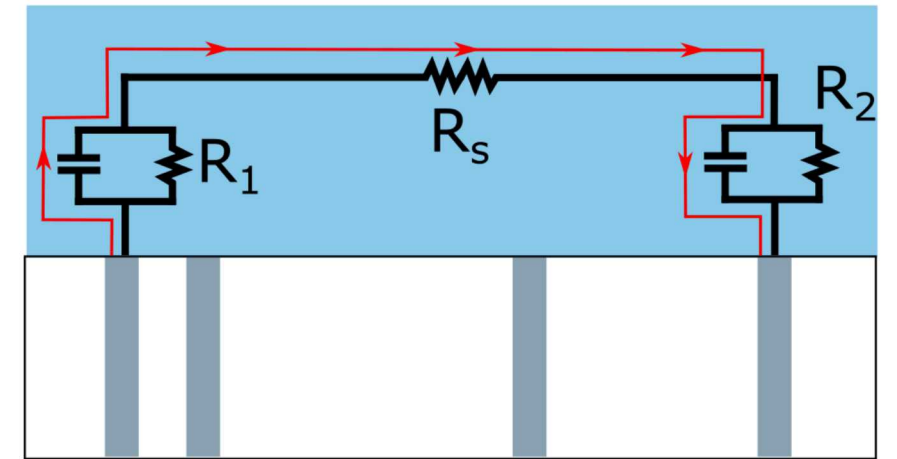
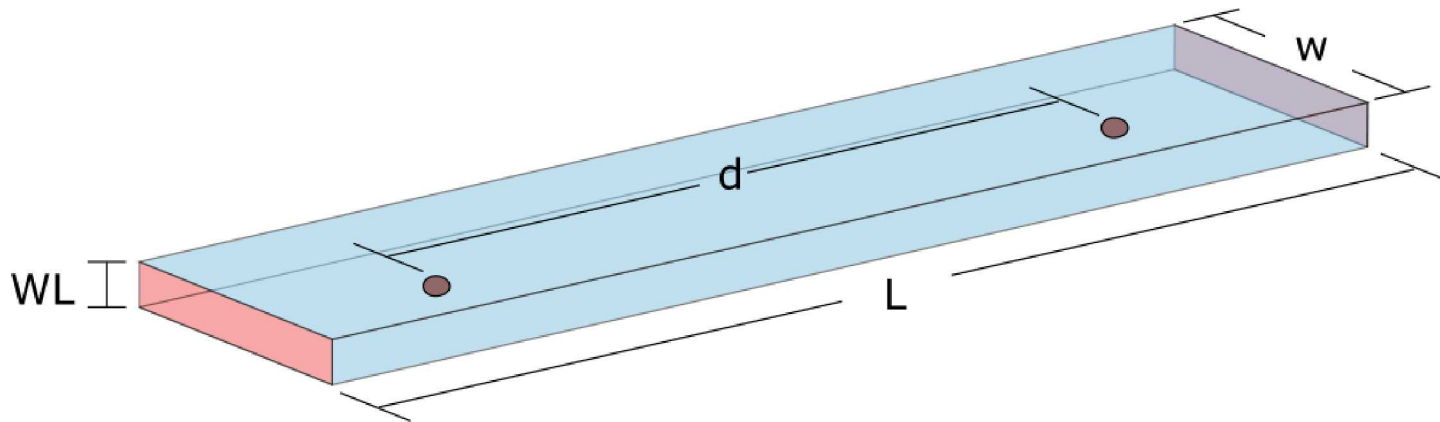
Governing Aspects of the Sensor

- A geometry that would avoid influencing the WL formation would involve mounting electrodes flush with the surface upon which the WL was being formed



Governing Aspects of the Sensor

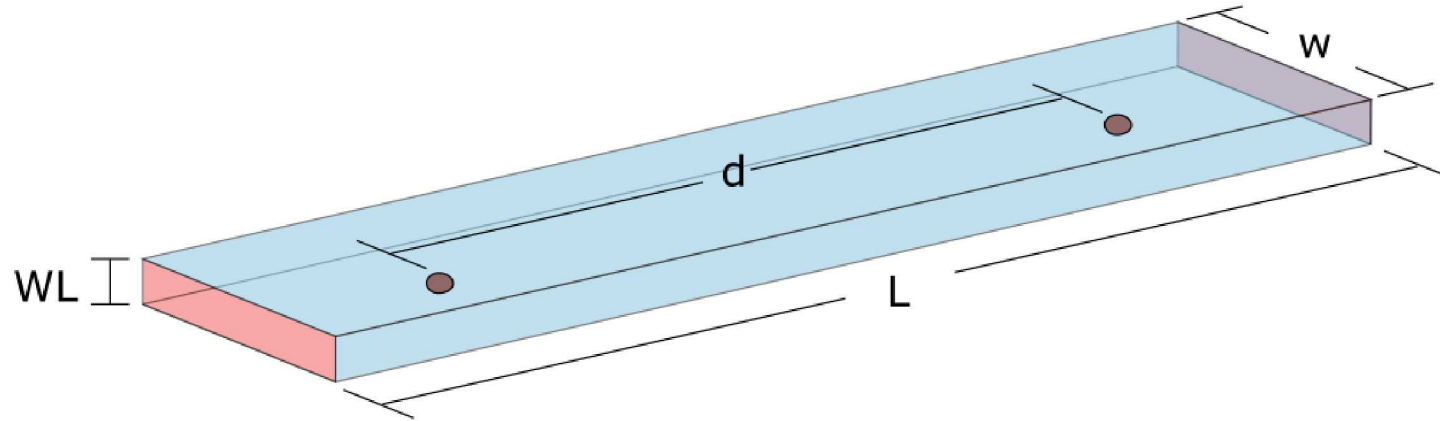
- EIS at high frequencies would measure the solution resistance



- No direct solution for the resistance as a function of WL which is a function of voltage (V) and current (I)
- FEM can be used to evaluate the expression

$$R_s = \frac{\Delta V}{\int I}$$

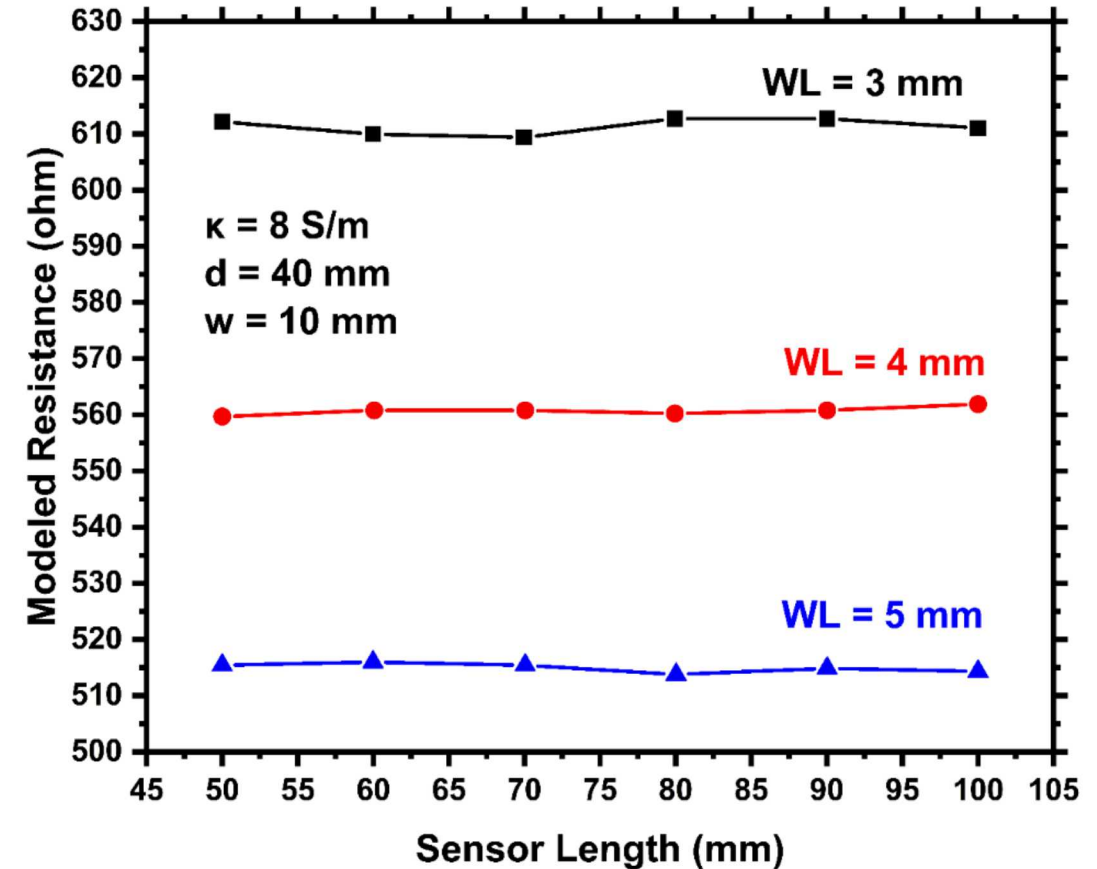
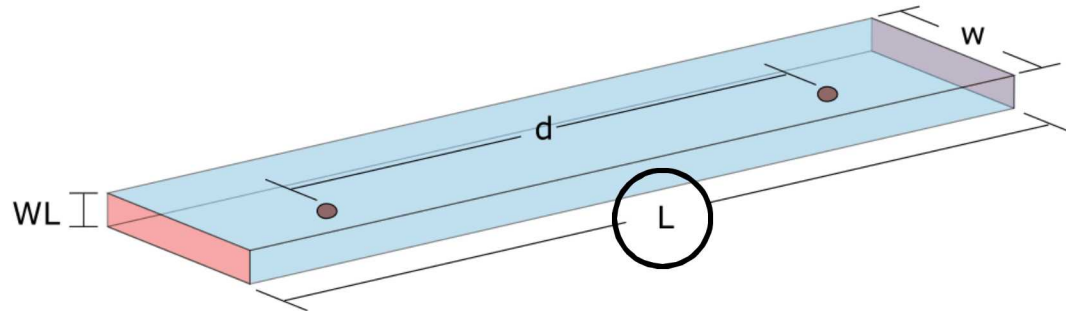
Goal



- Create a resistance-based sensor to measure WL thickness in continuous salt spray environments that is sensitive in the range of 0-5 mm
- Approach
 - Utilize FEM to create the sensor and explore various sensitivities (d , w , L , κ)
 - Construct sensor and utilize in accelerating testing scenarios

Sensor length does not influence calculated resistances

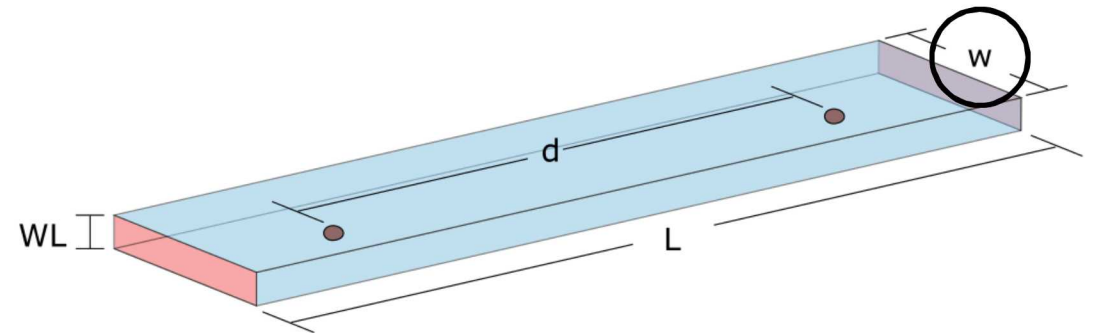
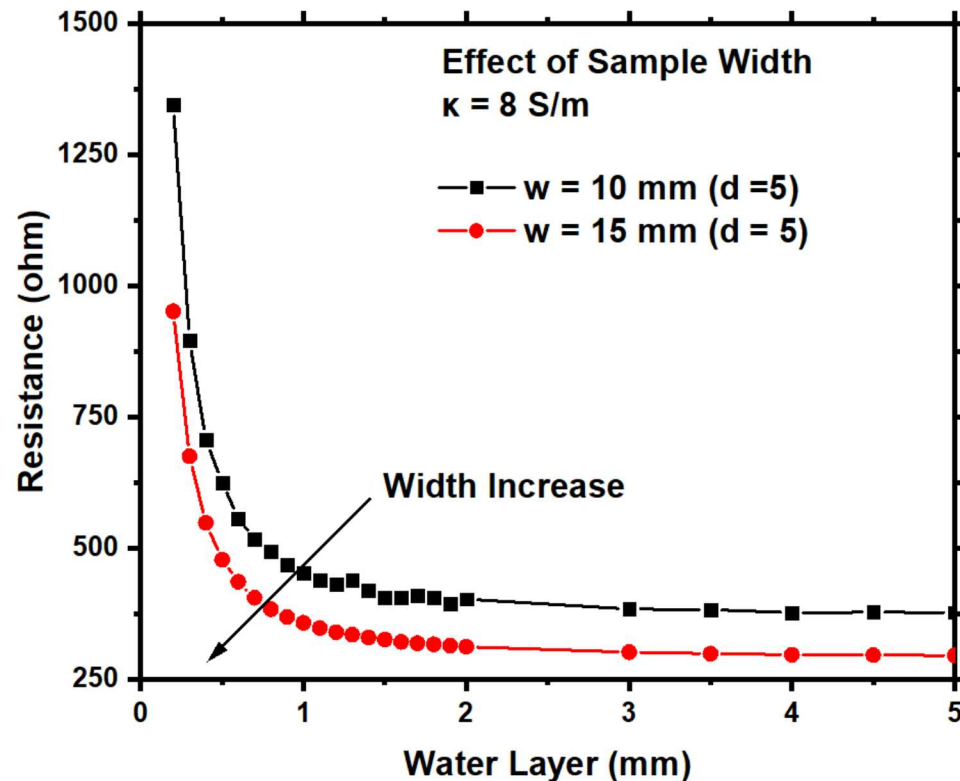
- Sensor length does not have an influence on the modeled resistance
- **However, continuous WL needed**



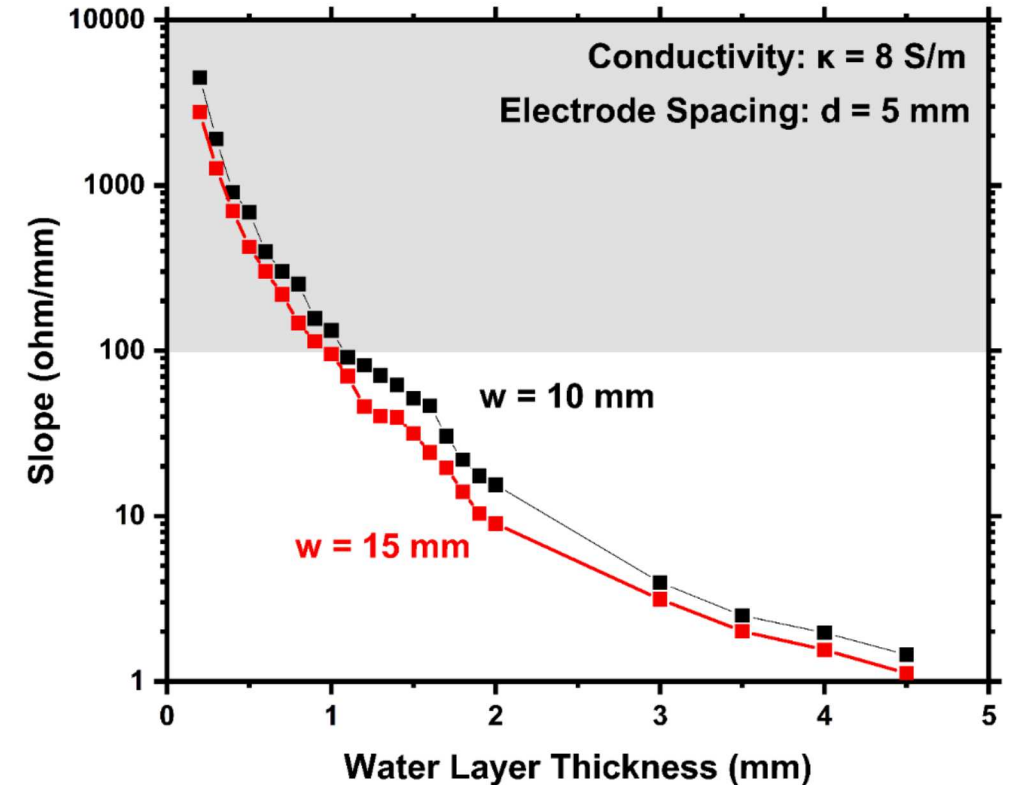
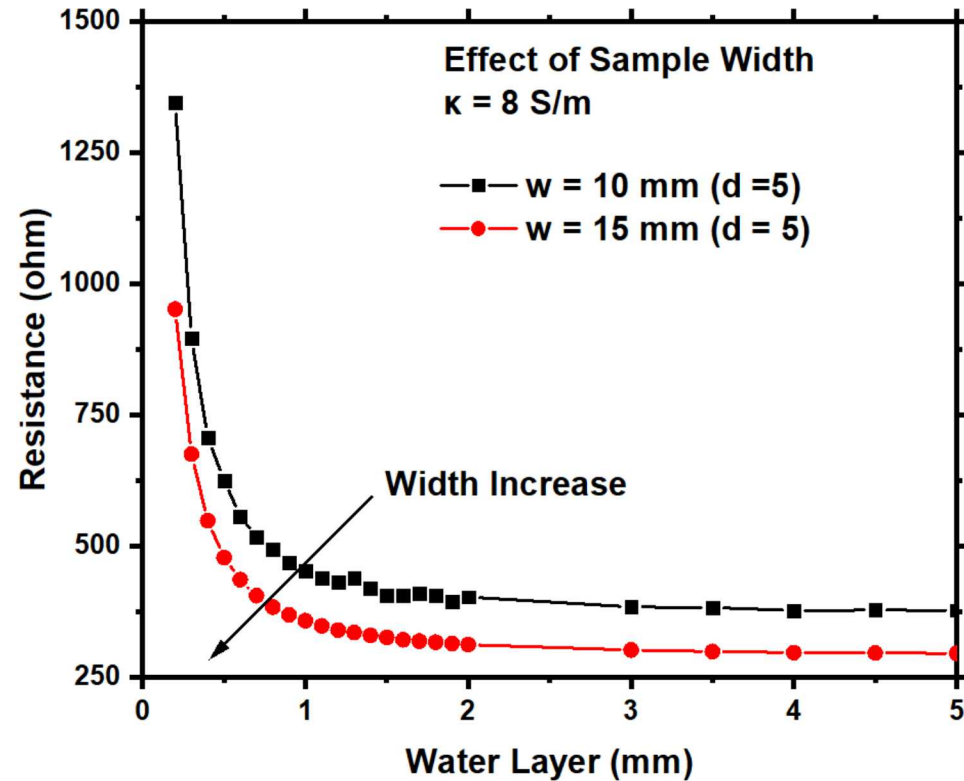
Model results

Smaller sensor widths increase sensitivity

- Increase in width decreases the calculated resistances
- Can determine sensitivity by taking slope of resistance vs. WL (ohm/mm)

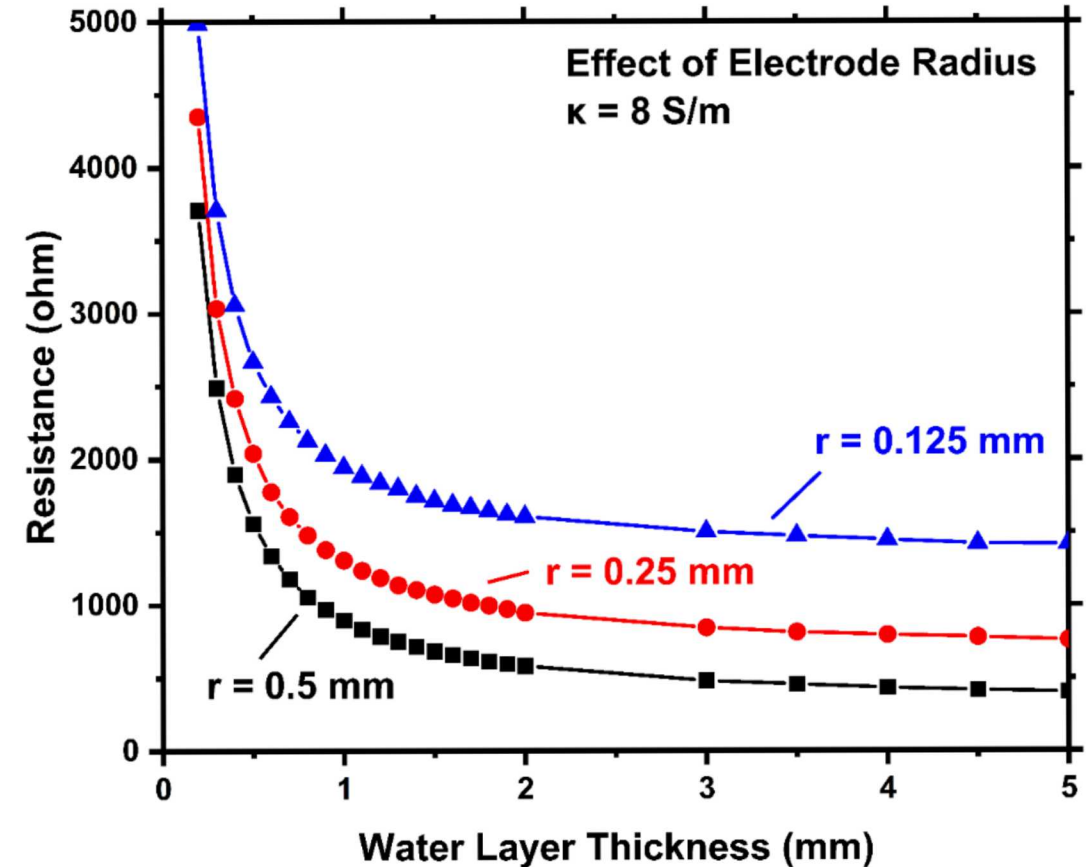
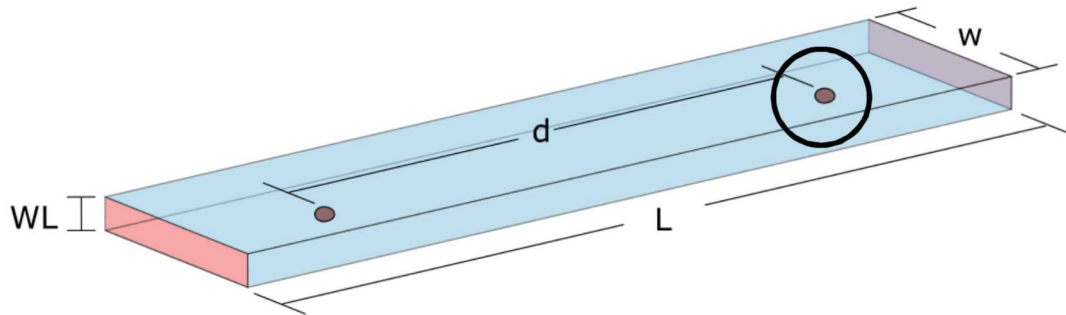


Smaller sensor widths increase sensitivity



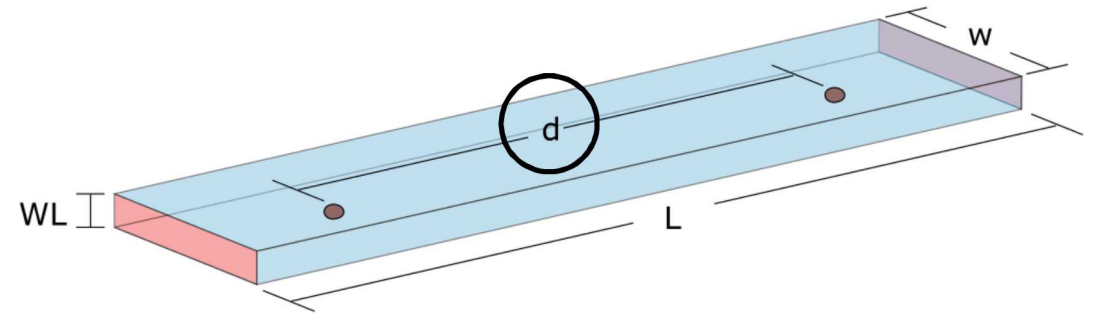
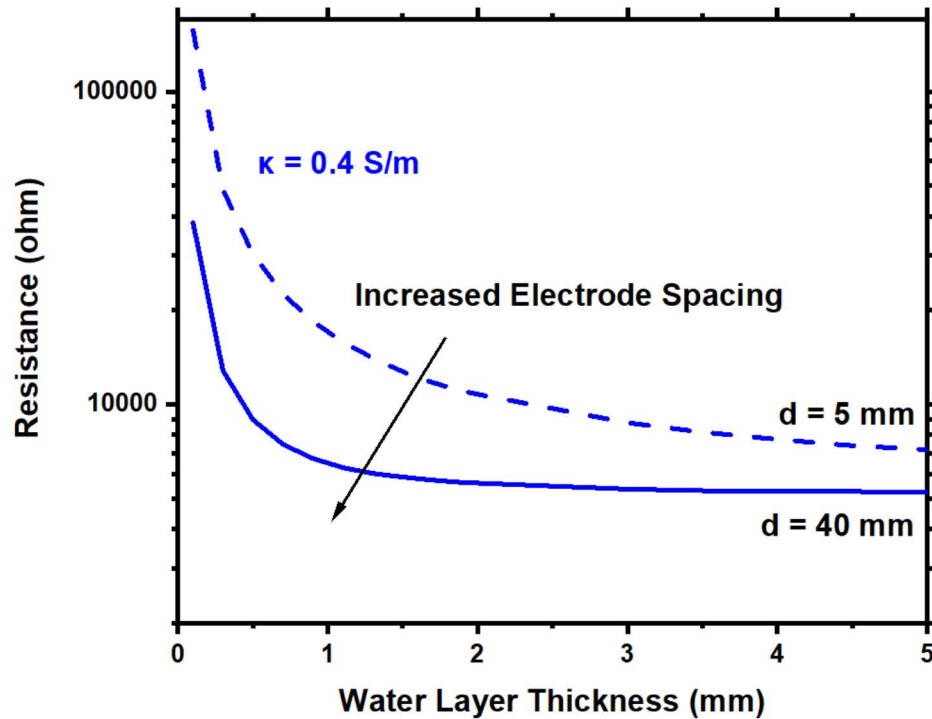
Resistances not sensitive to wire radius

- Increasing embedded wire radius, decreases modeled resistances
- But there is no change in the sensitivity (ohm/mm) due to the radius of the wire

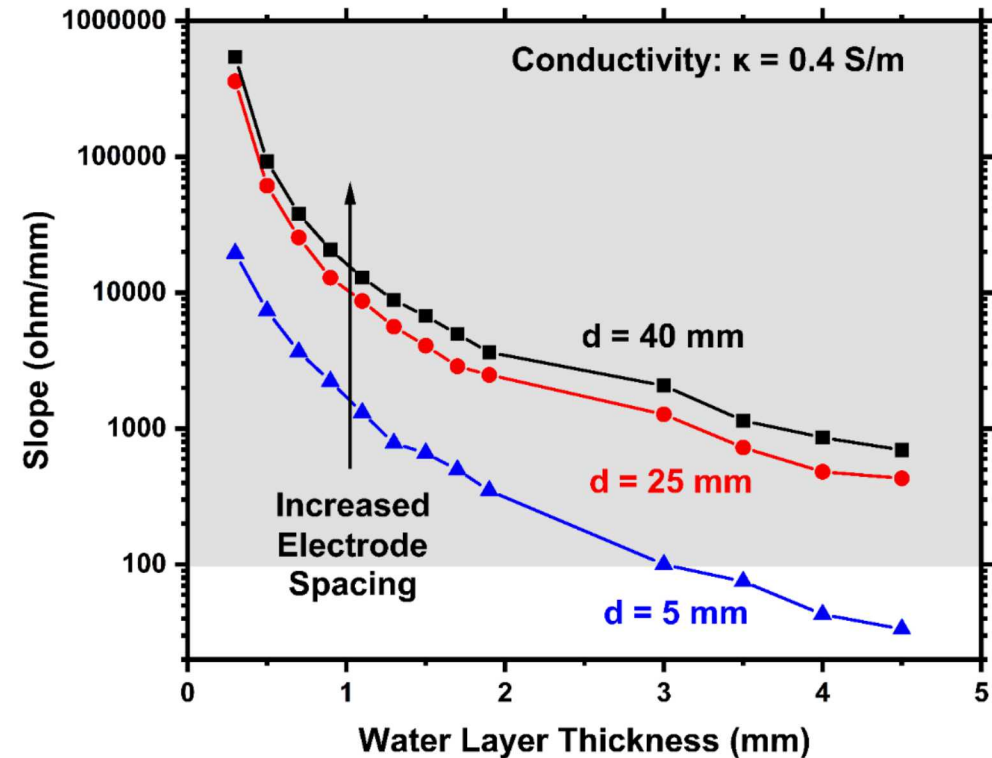
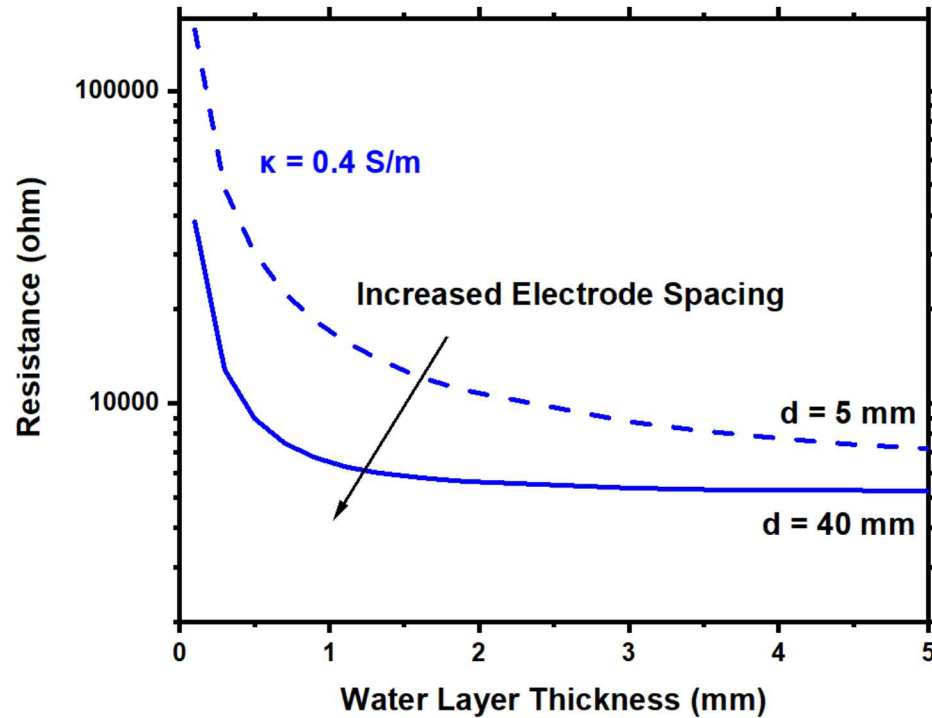


Wide electrode spacing provides highest sensitivity

- Increasing the distance between electrode decreases the measured resistances

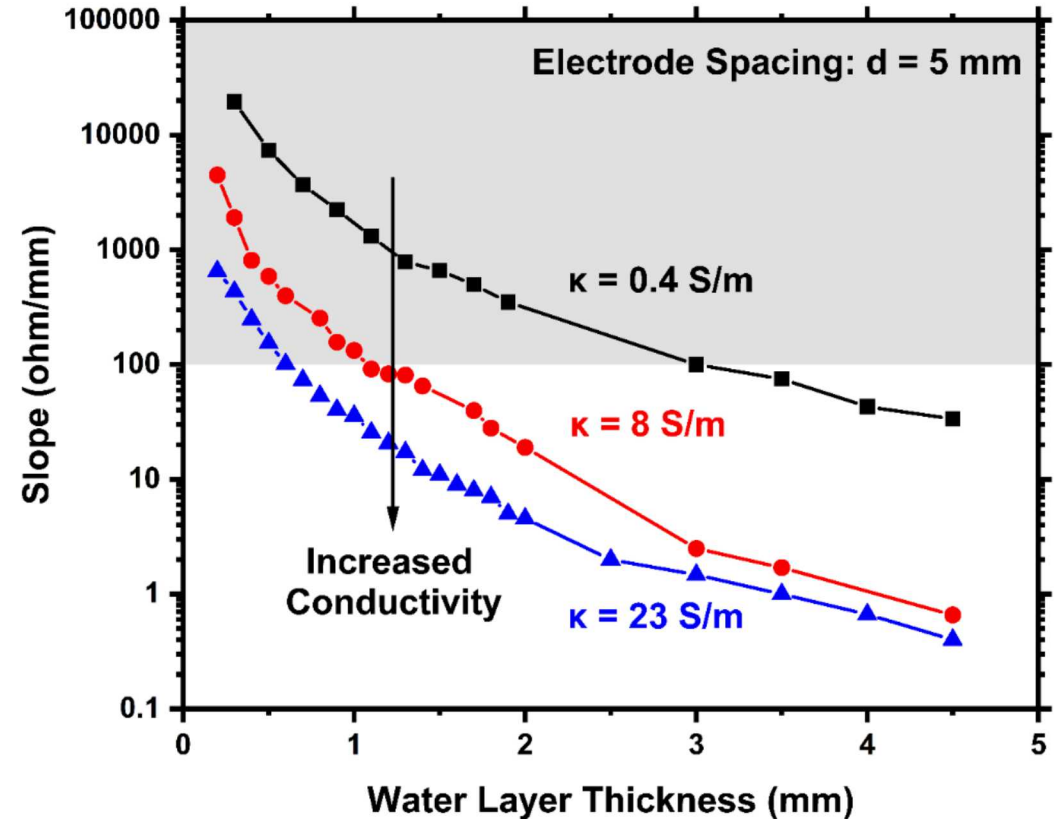
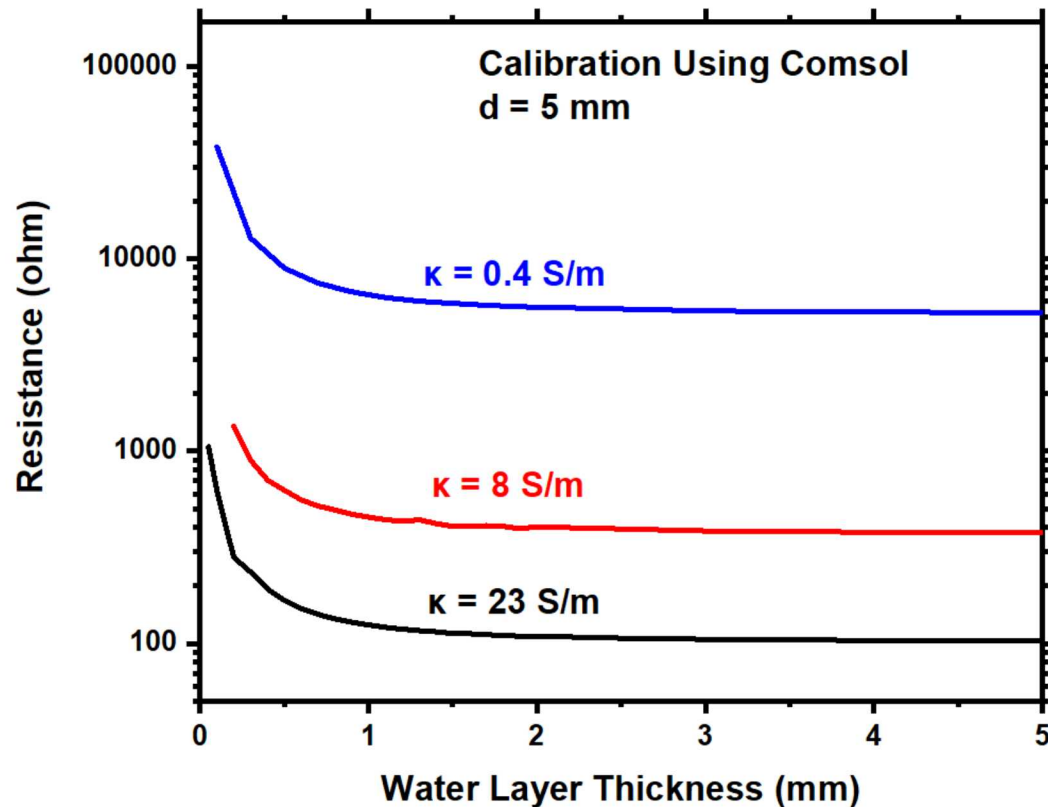


Wide electrode spacing provides highest sensitivity



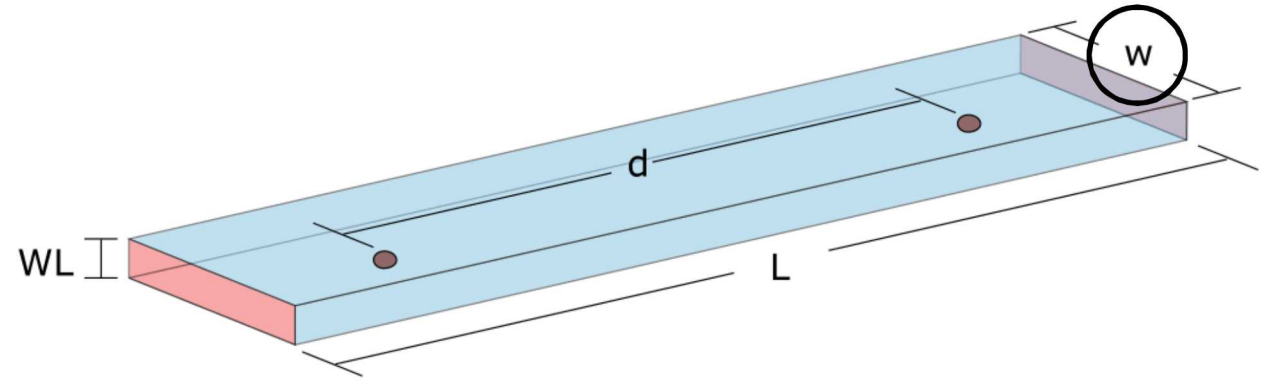
Dilute solutions provide highest sensitivity

- Decreasing solution conductivity increases modeled resistances
- **Sensitivity is increased for dilute solutions**



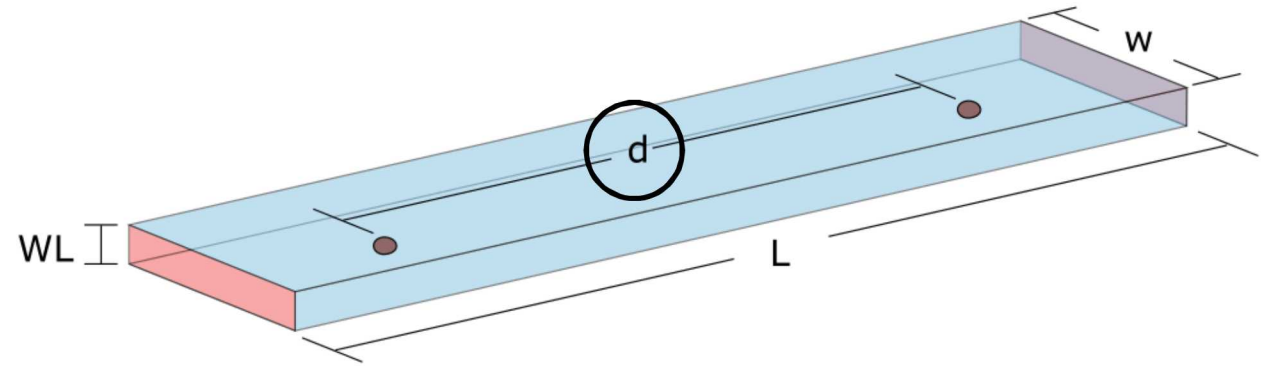
Best sensor qualities

- Small sensor width



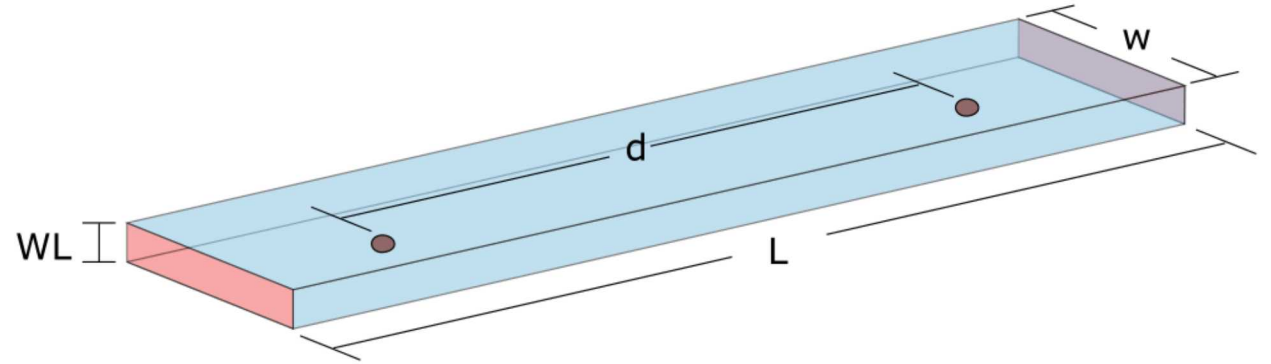
Best sensor qualities

- Small sensor width
- Large electrode spacing



Best sensor qualities

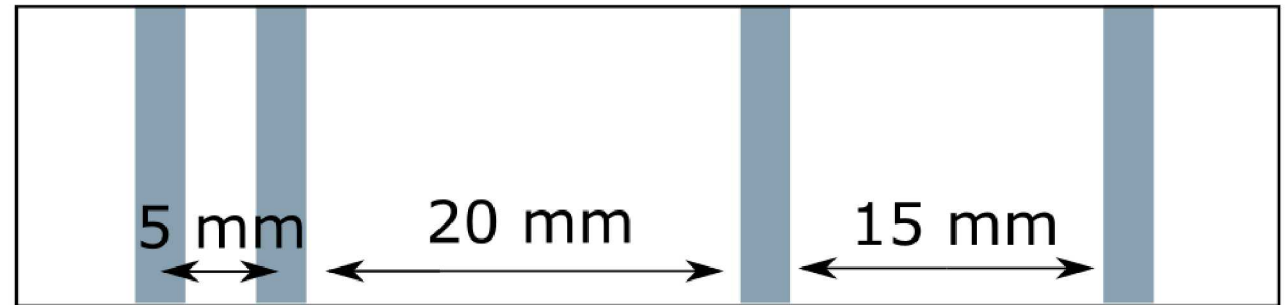
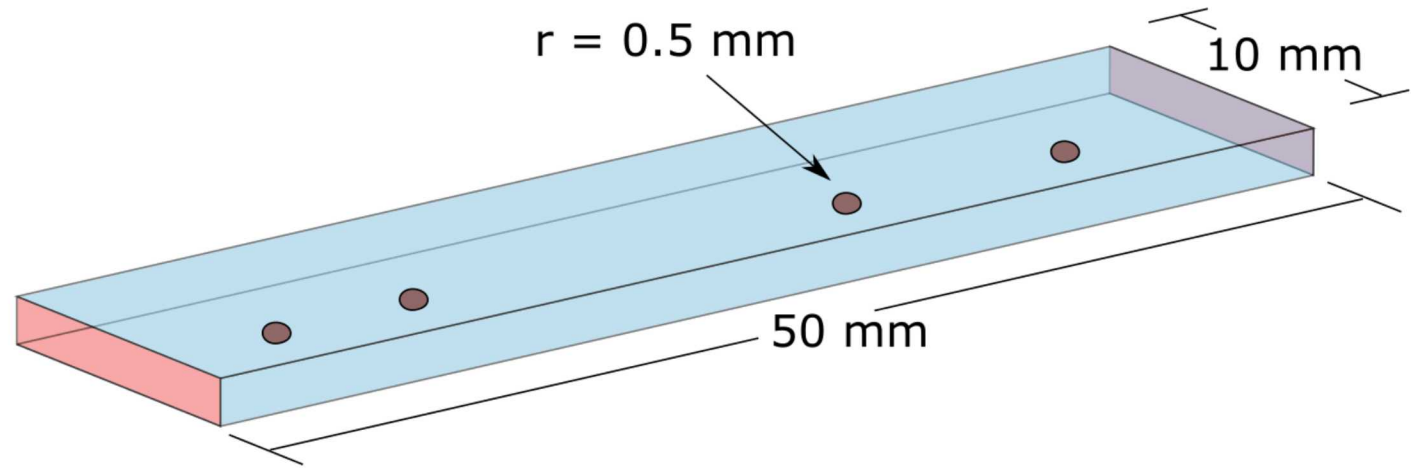
- Small sensor width
- Large electrode spacing
- Low conductivity solutions



- Several concerns related to the creation of a sensor:
 - Continuous WL formation
 - Influence of edge effects on WL formation (high angled meniscus)

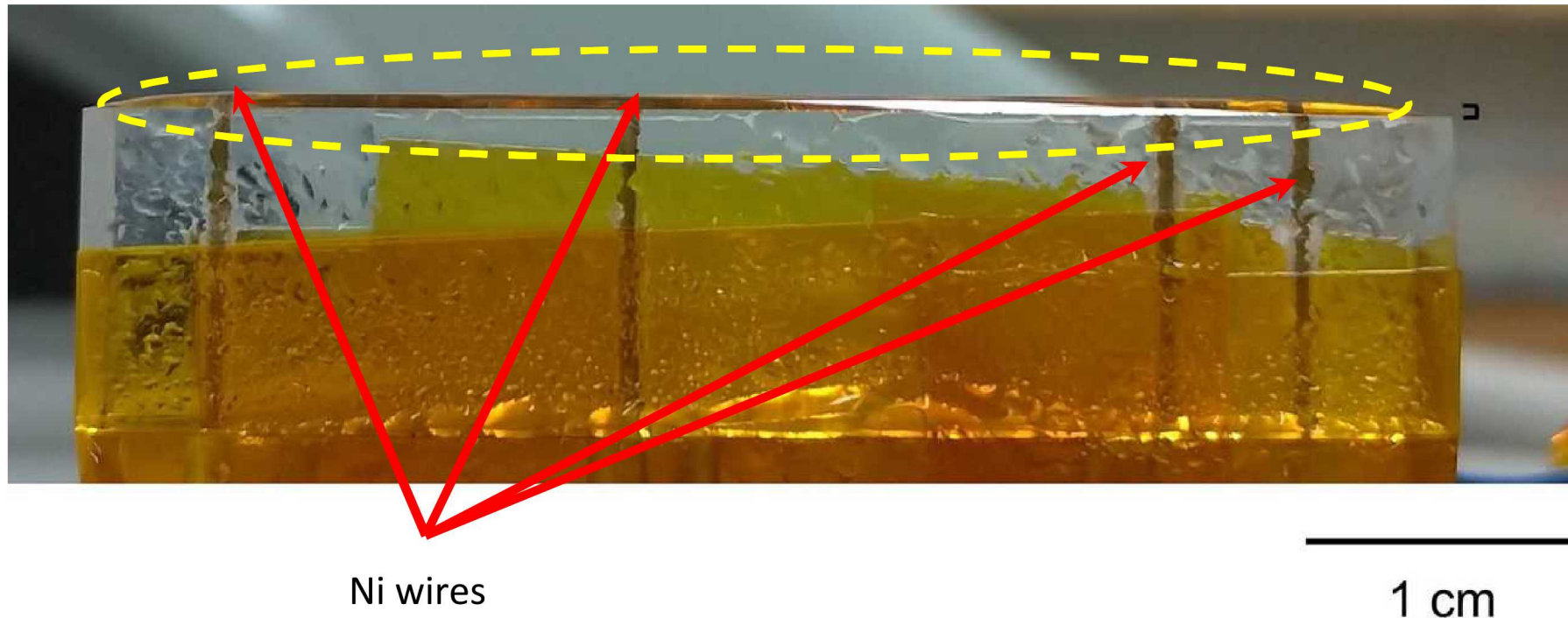
Final sensor dimensions

- Based on the FEM modeling, the sensor was created with the shown dimensions
- A variety of electrode spacings was created



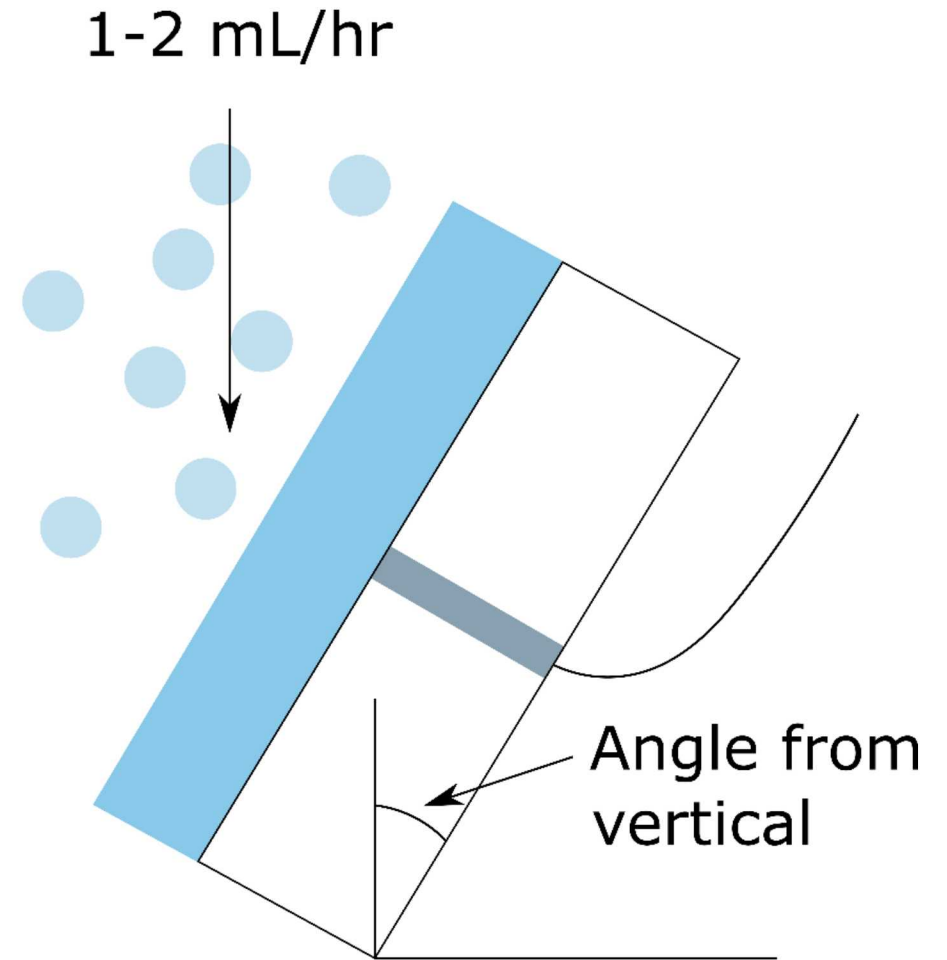
Created sensor

- To create the sensor, nickel wires were mounted in plexiglass
 - High conductivity nickel wires decreases influences to resistance measurements
- Wires electrically insulated to connect to external potentiostat



Experimental Approach

- Calibrations with known volume of liquid
 - Walled structure
 - Thin AAO membrane
- Continuous salt spray environment utilized
 - Various solution conductivities
- Extent of accelerated corrosion standards were explored
 - Deposition rate (ranging from 1-2 mL/hr)
 - Test interruptions
 - Angle of exposure



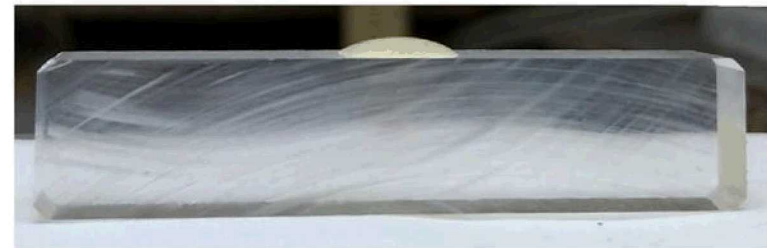
Experimental Approach

- In all experiments, the sensor was cleaned in a plasma cleaner for 1 hour, under 100 Watts in an oxygen-only environment

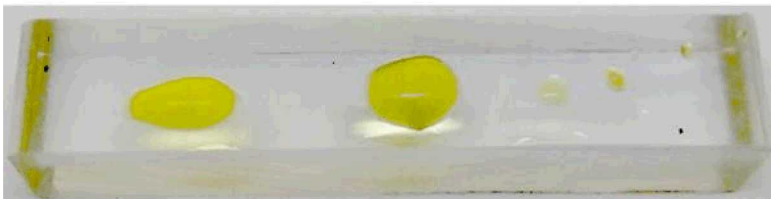
Before



After



After spreading out

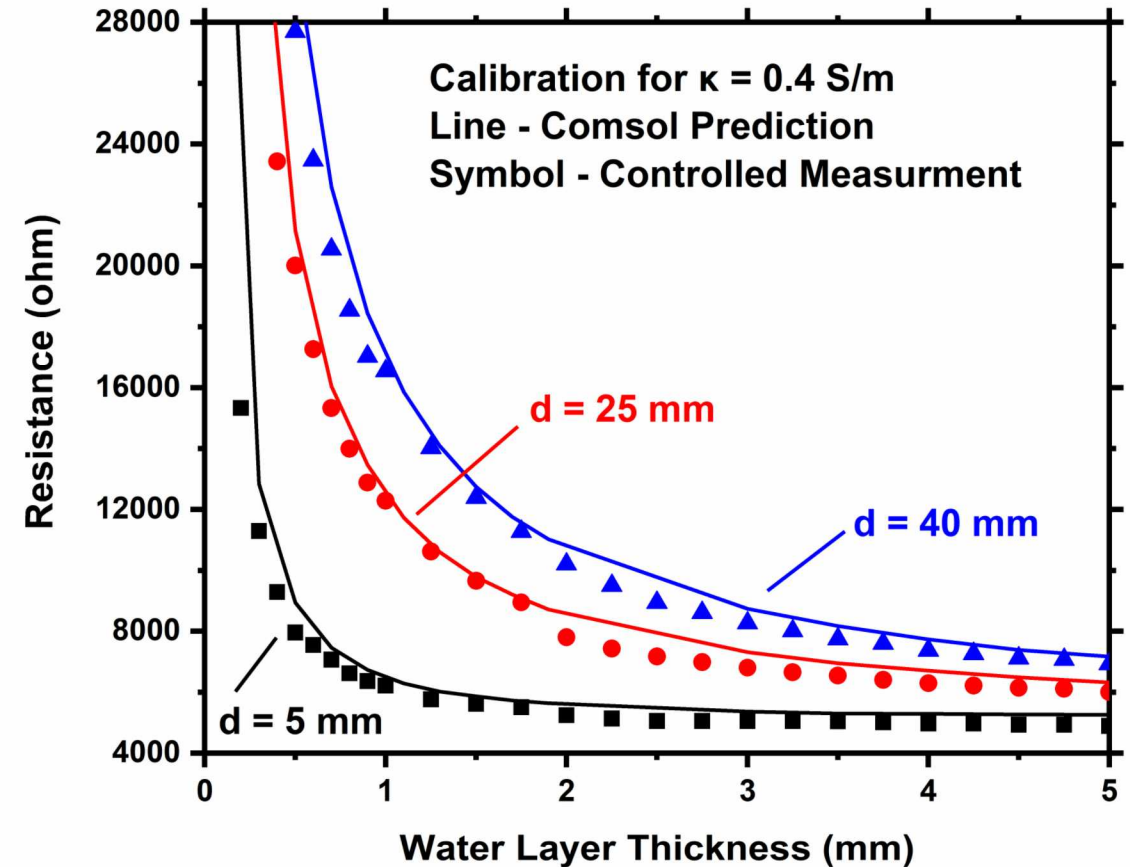
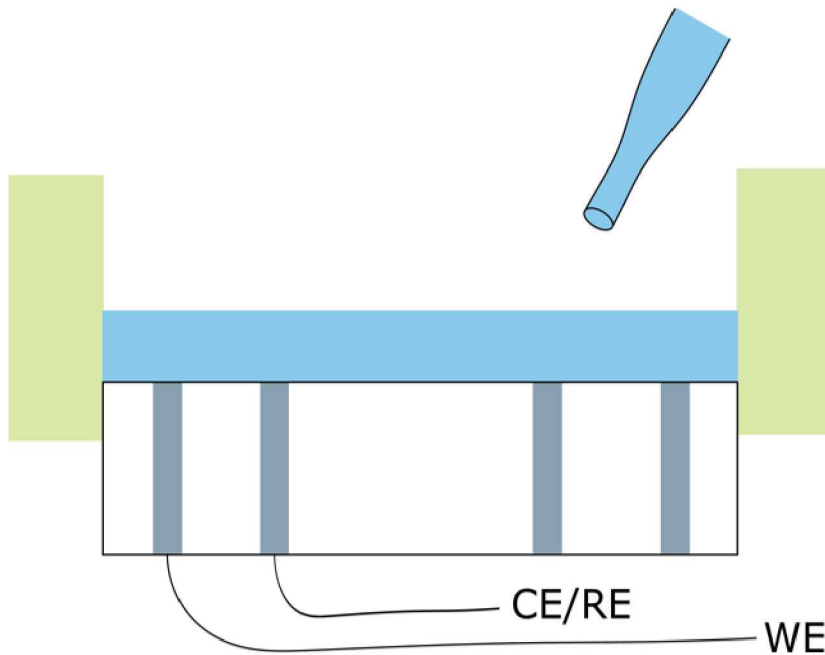


Experimental Approach

- High frequency EIS was utilized to determine WL thickness
- Frequency was scanned from 7MHz to 1kHz with six points per decade
- Sine waves of amplitude of 10 mV were applied about open circuit
- Each point was an average of 10 measurements

Calibration of Sensor

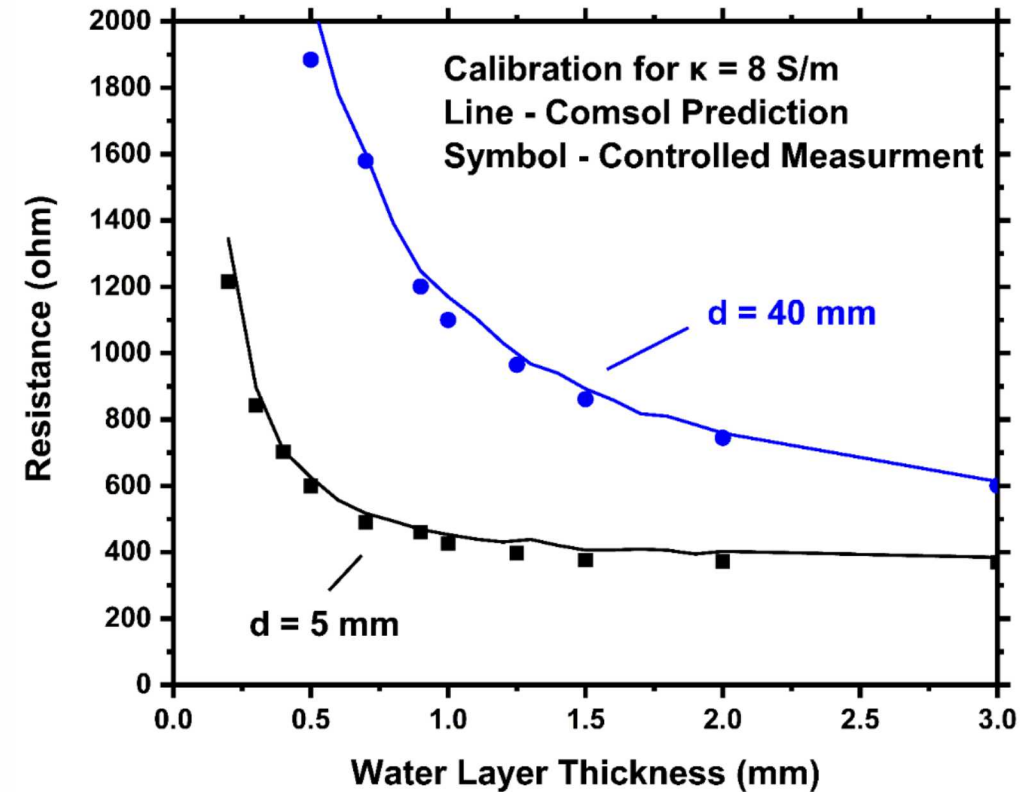
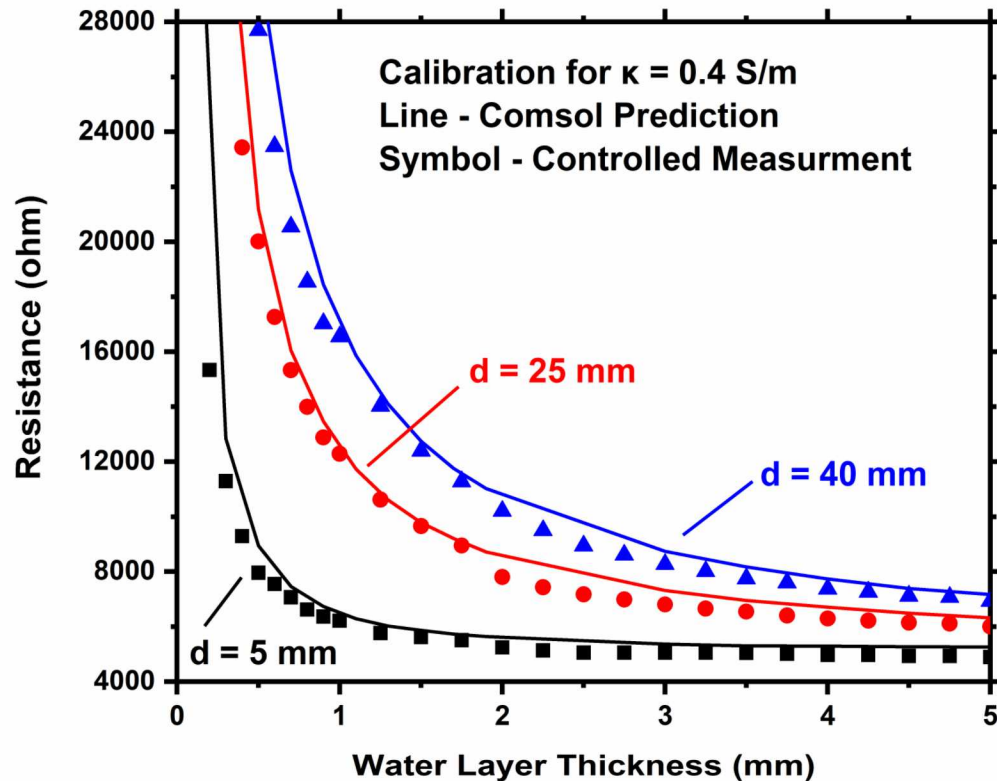
- Known volumes of solution were placed on top of the sensor
 - WL thickness calculated through known densities and exposure area



- Measured resistances match well with modeled resistances

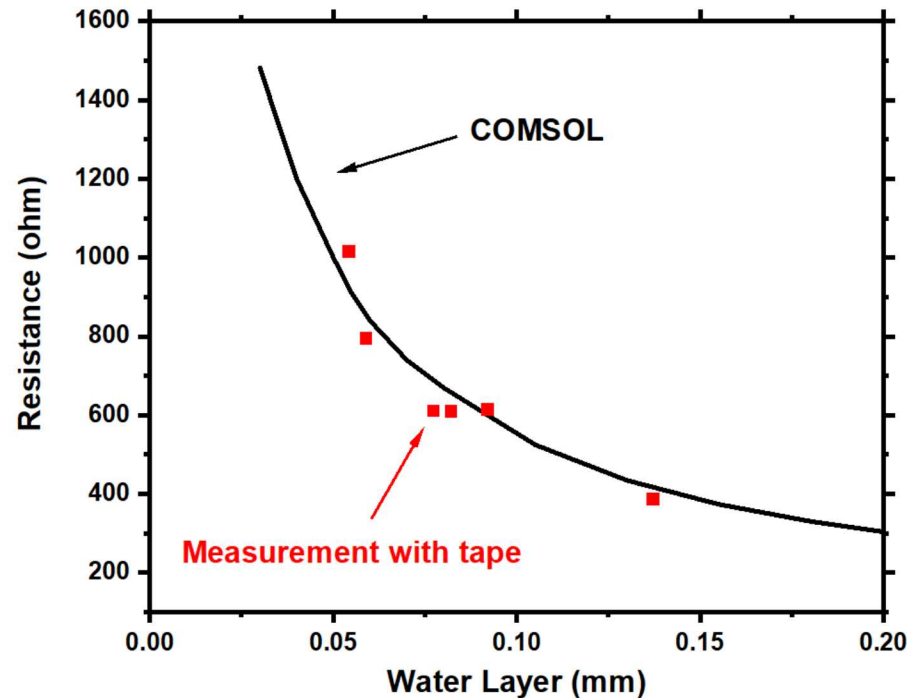
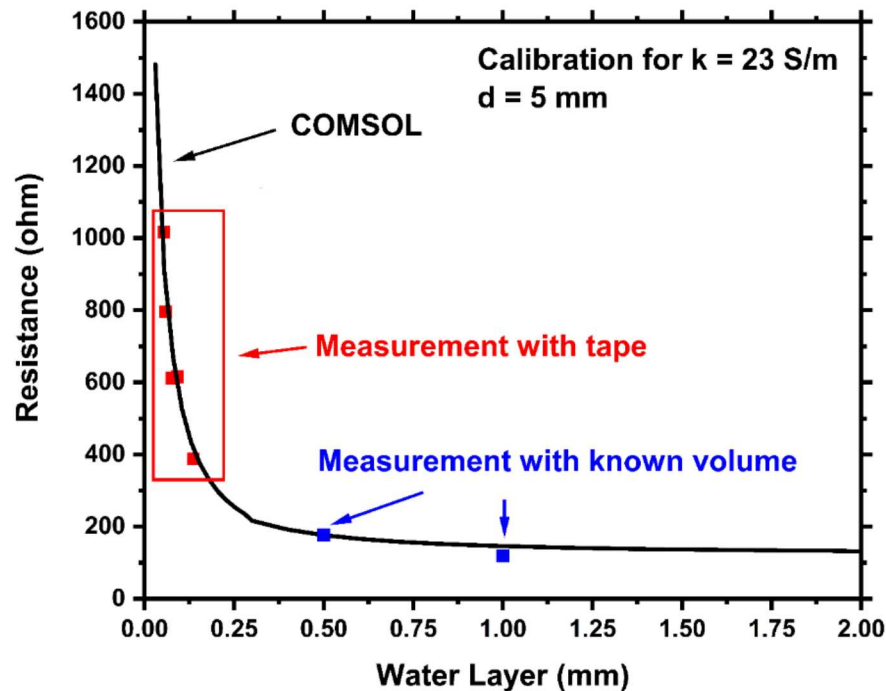
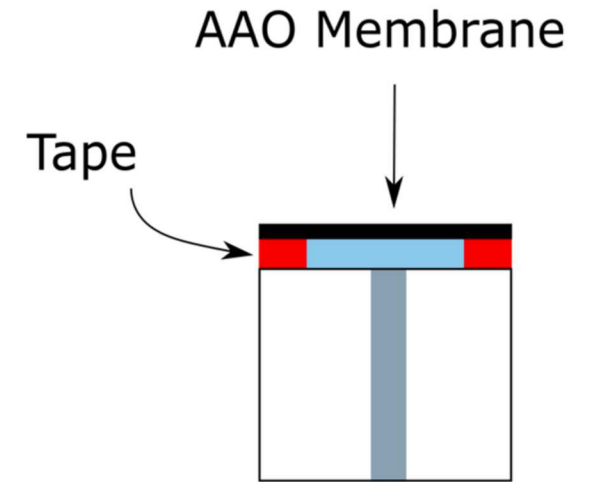
Calibration of Sensor

- Sensor efficacy shown across multiple solution conductivities and electrode spacings



Calibration of Sensor

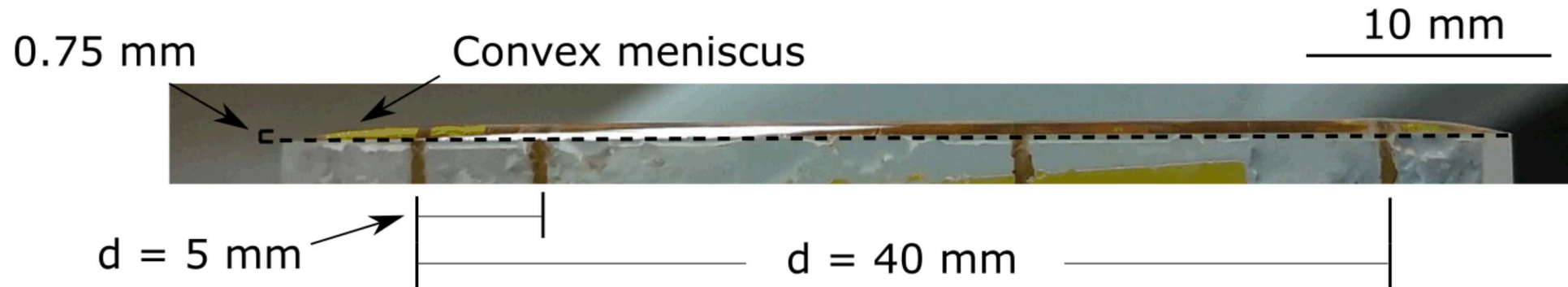
- Known tape thicknesses placed around edge of the sensor and an anodic aluminum oxide (AAO) placed on top
 - AAO was treated with trimethoxy(octyl)silane to increase the hydrophobic behavior



- Small water layer thicknesses also match well with modeled results
- Smaller error in resistances

Increased accuracy with membrane due to no meniscus

- Increased accuracy shown when using AAO membrane in comparison to known volume of liquid on surface
- Small convex meniscus seen when water layer present on the top of plexiglass

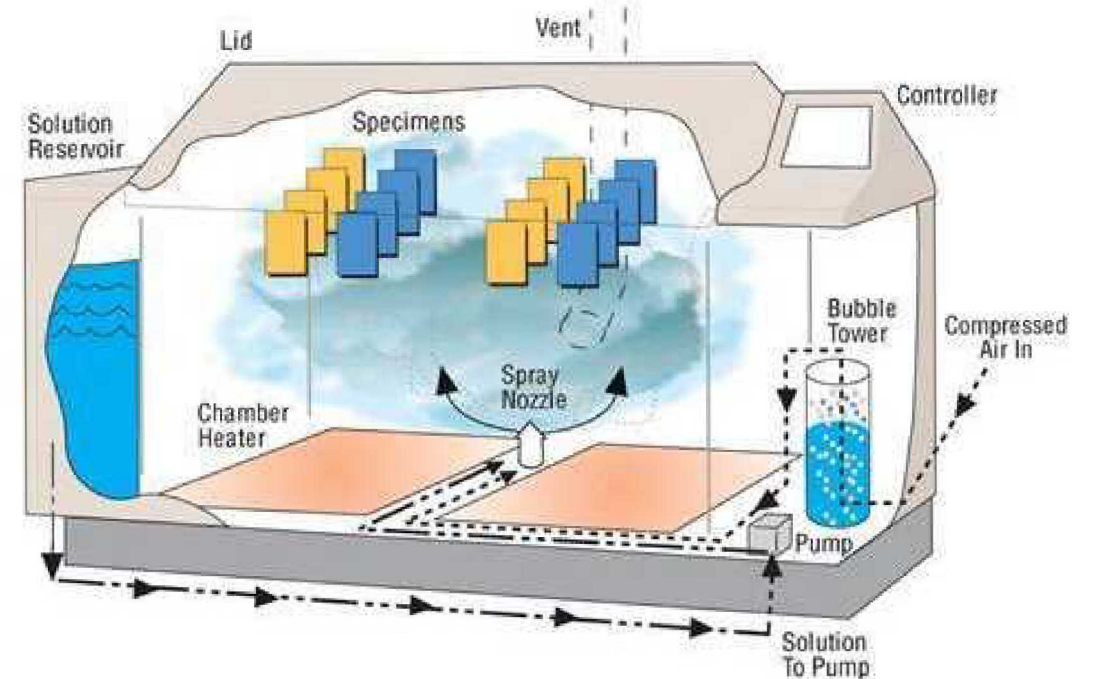
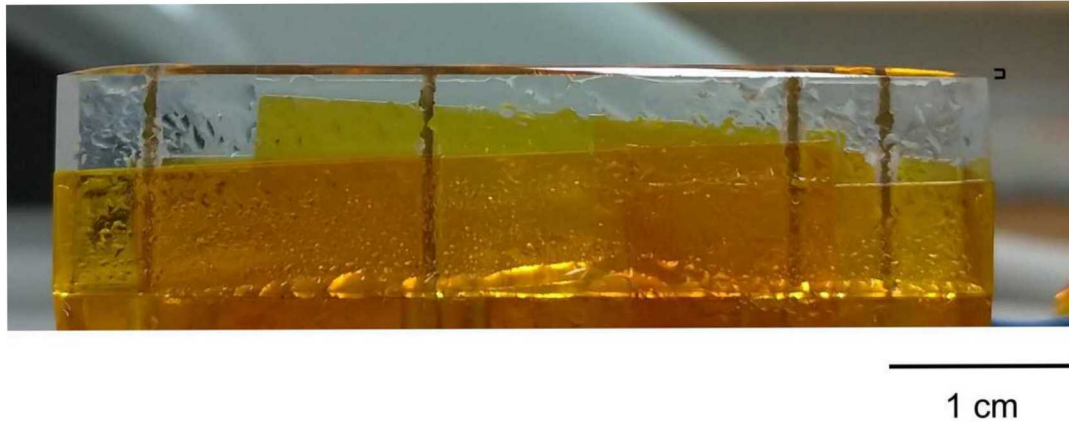


- Considering a simple parallelepiped, increased WL decreases R_s

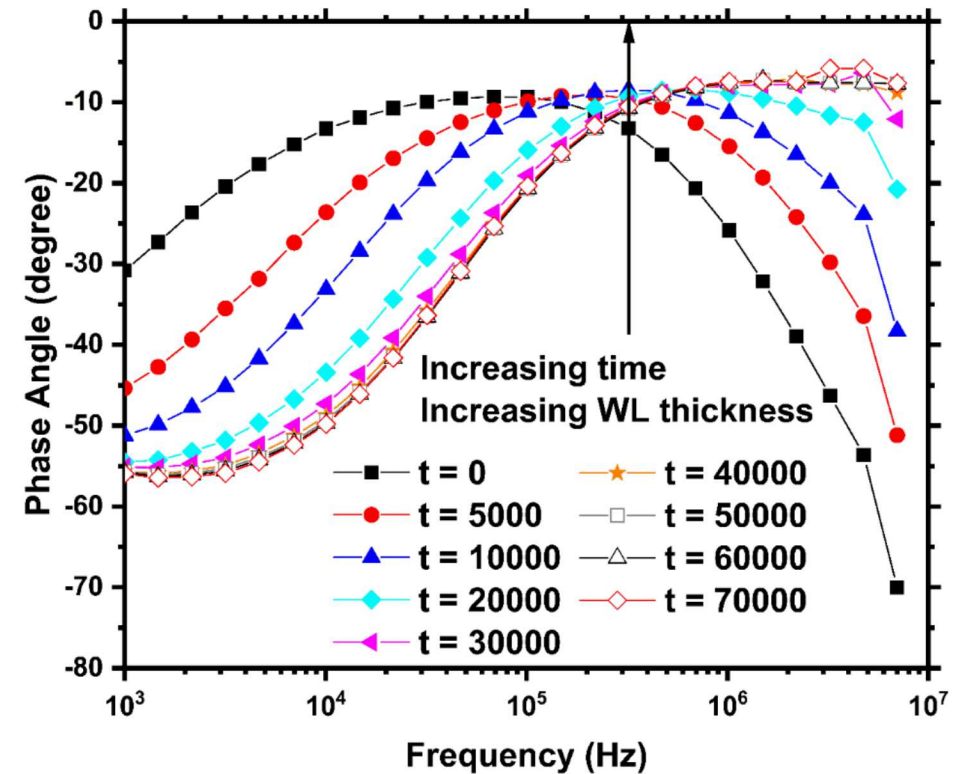
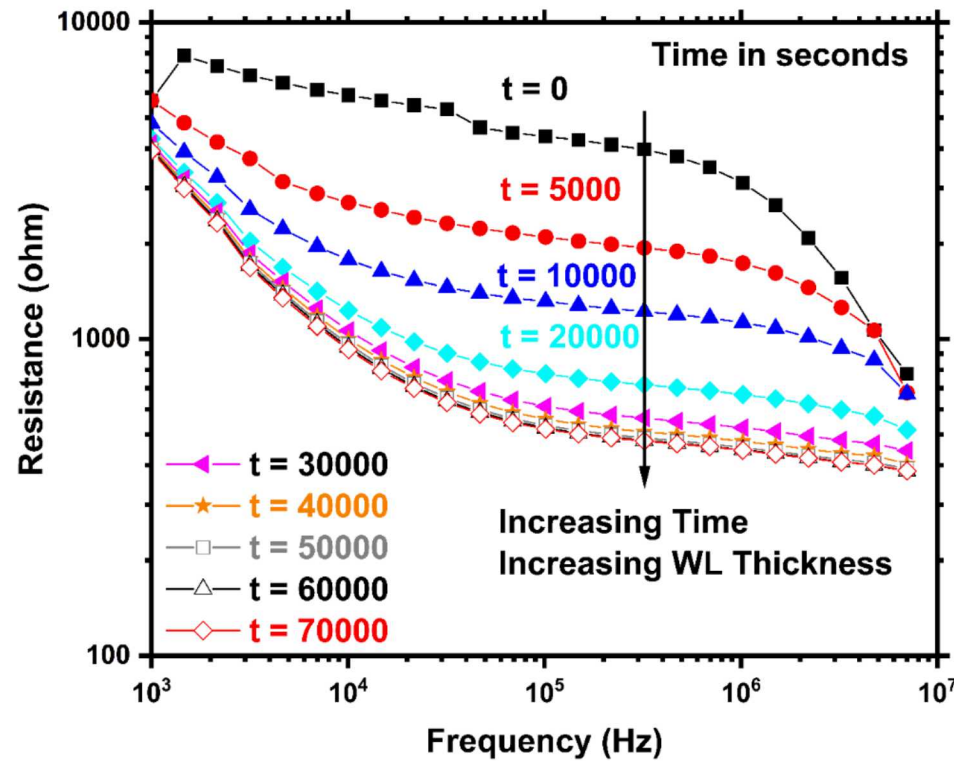
$$R_s = \frac{L}{WL * w} * \frac{1}{\kappa}$$

Application of sensor to salt spray test

- Sensor placed in constant salt spray chamber to measure WL as a function of test parameters



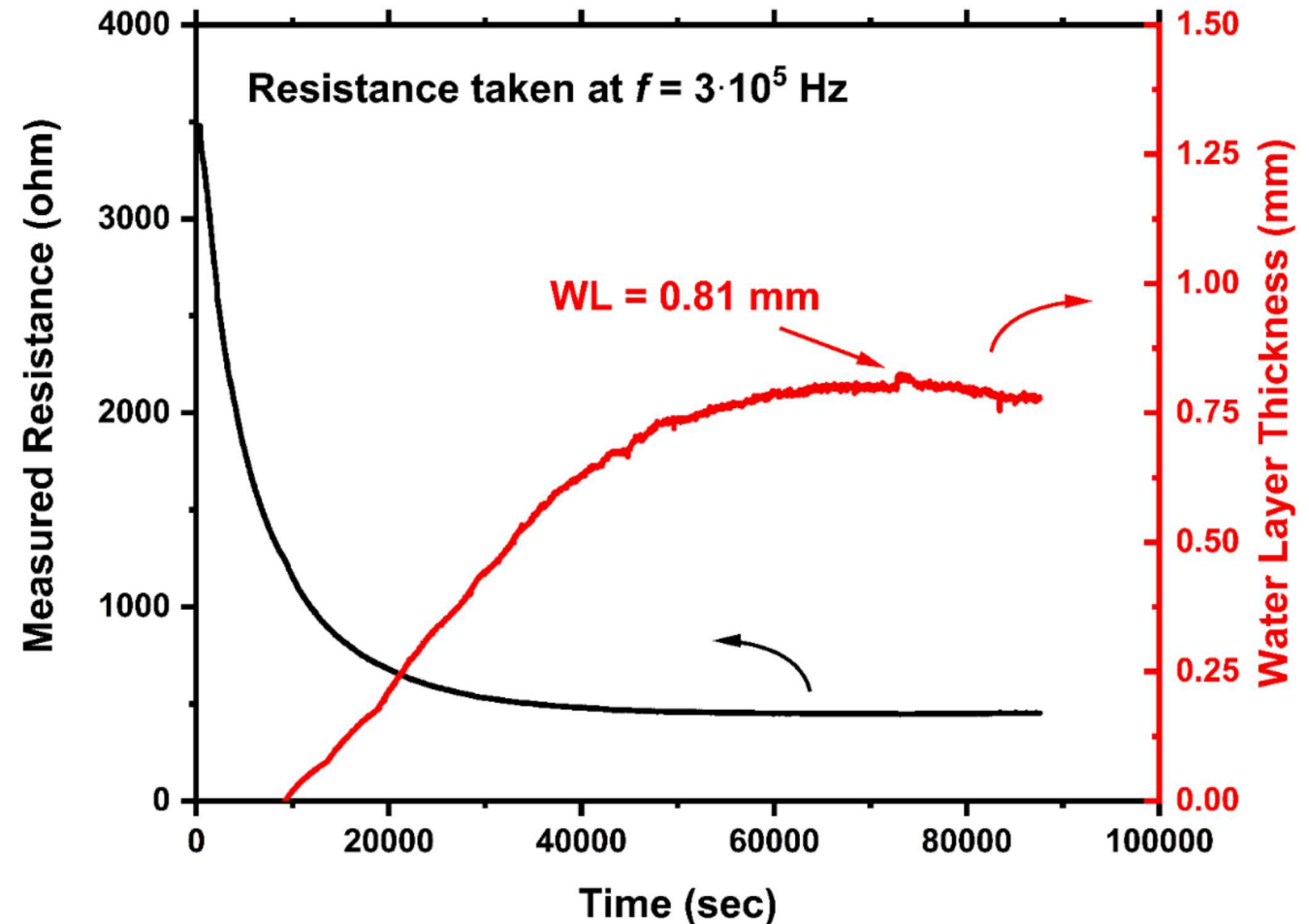
Decreased in resistance with increased exposure time



- Resistances were compiled at a relatively constant phase angle with time
- Initial rewetting of sensor took ~ 17 hr

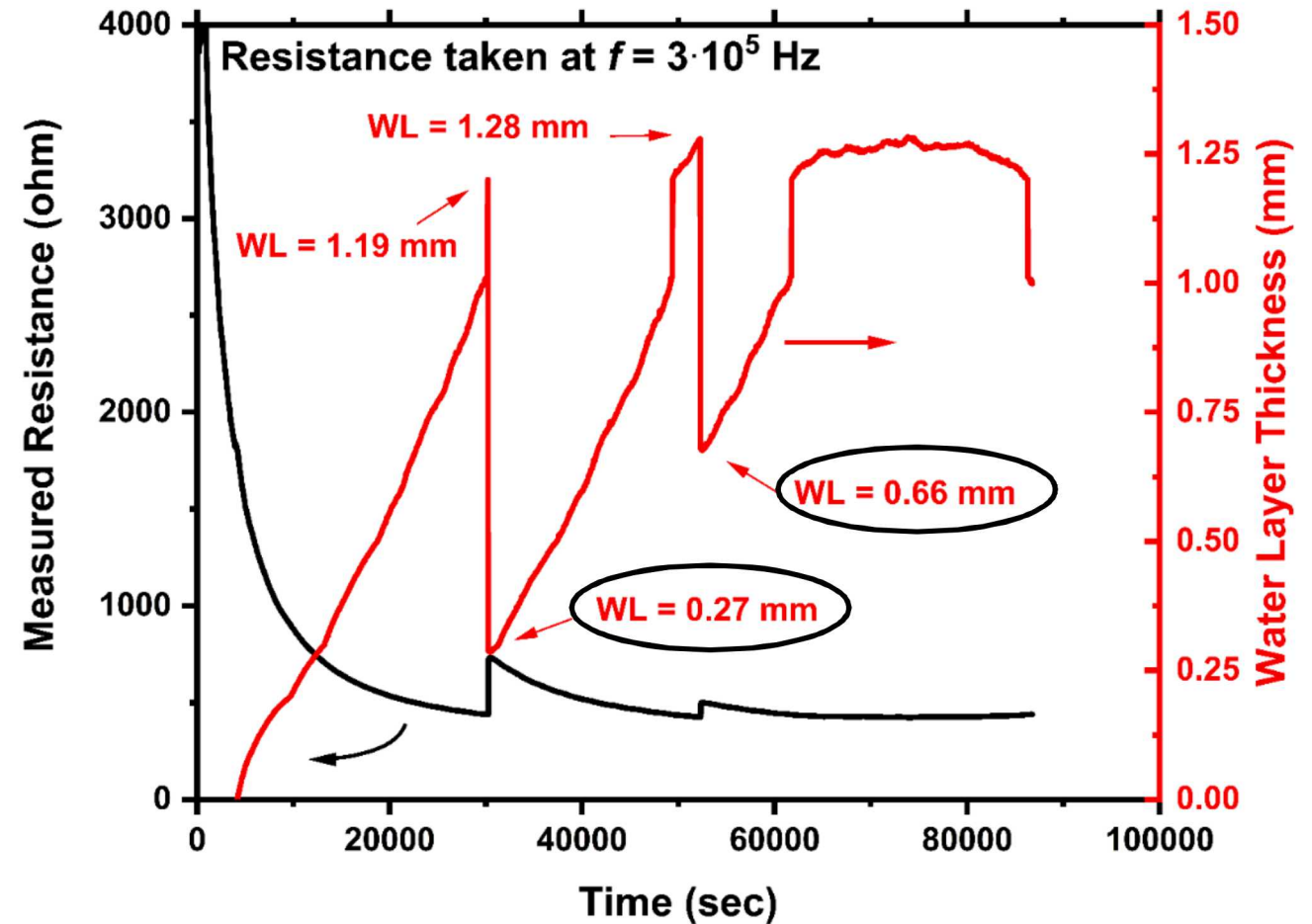
Plateau in WL thickness at long exposure time

- EIS measurements taken at 0.6 M NaCl ($\kappa = 8 \text{ S/m}$), 25 °C, and angle of 20°
- Resistances evaluated as a function of time at a constant frequency showing a plateau
- WL thicknesses calculated based on calibration curve
- Roughly 17 hours needed to achieve steady state
- All measurements taken for at least 24 hours in chamber



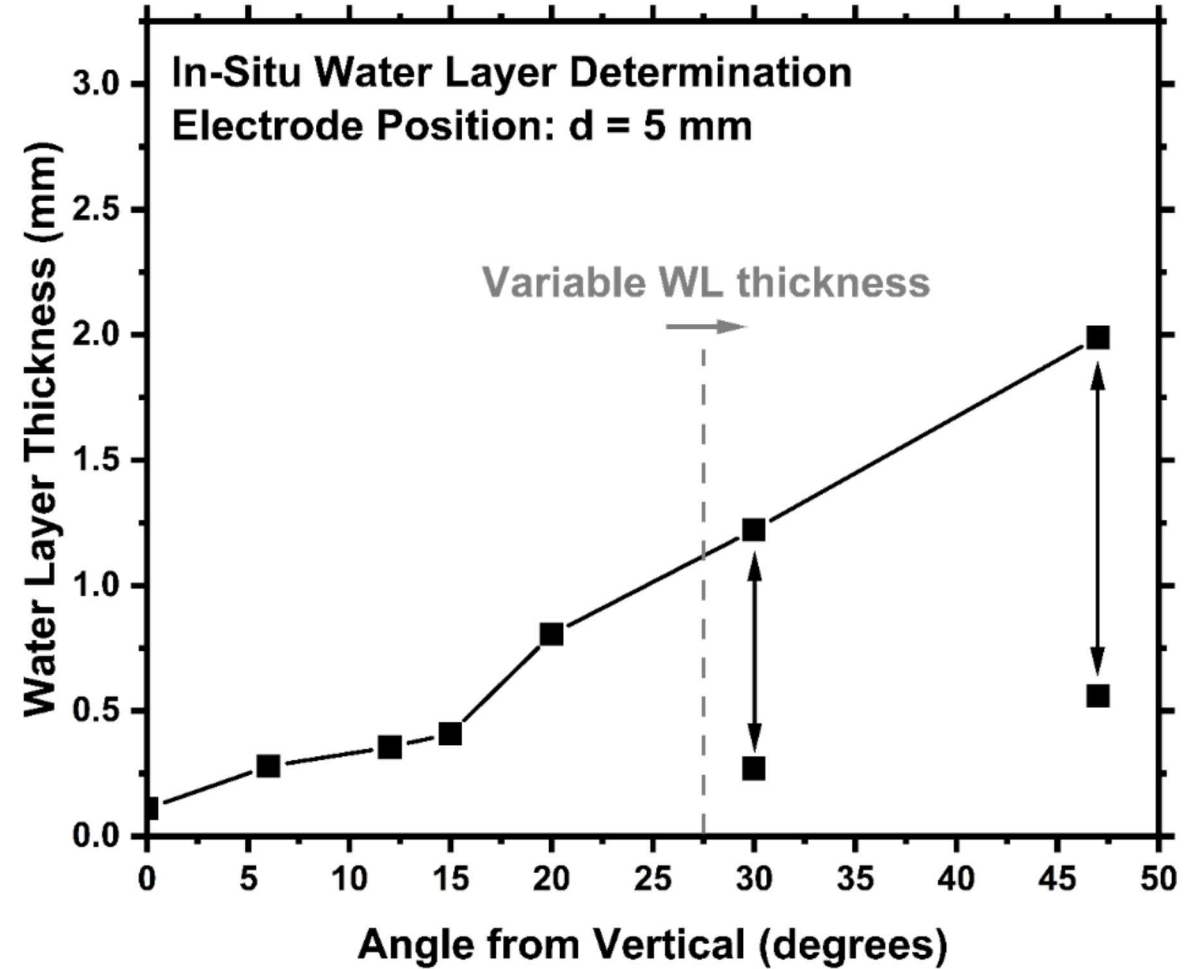
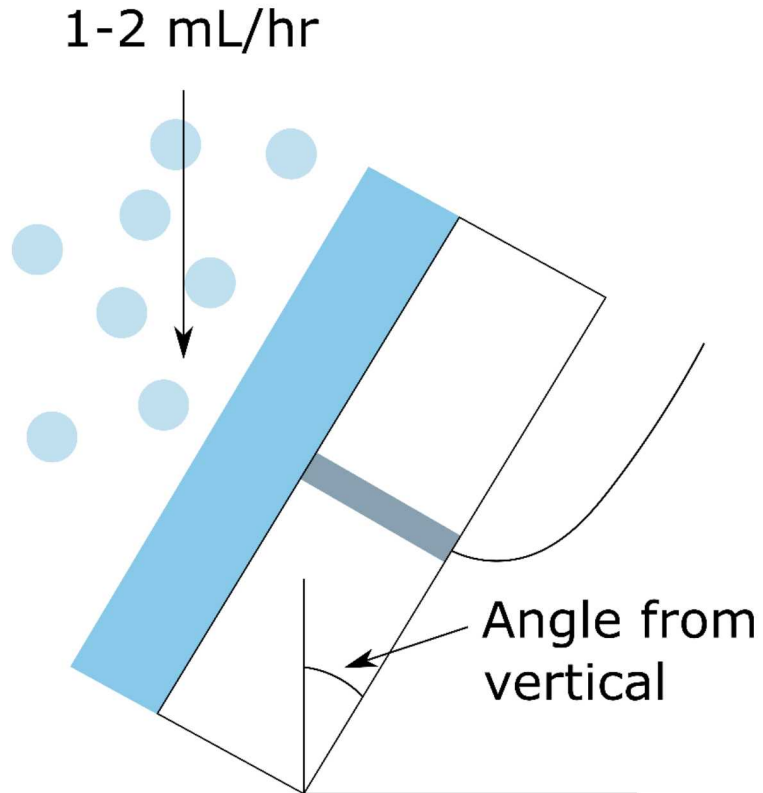
Increased exposure angles cause semi-periodic run-off

- EIS measurements taken at 0.6 M NaCl ($\kappa = 8$ S/m), 25 °C, and angle of 30°
- Variable WL at increased angles of exposure
- Semi-periodic in nature
- Seen at angles larger than 20°
 - Independent of κ or deposition rate
- Variable final WL thickness



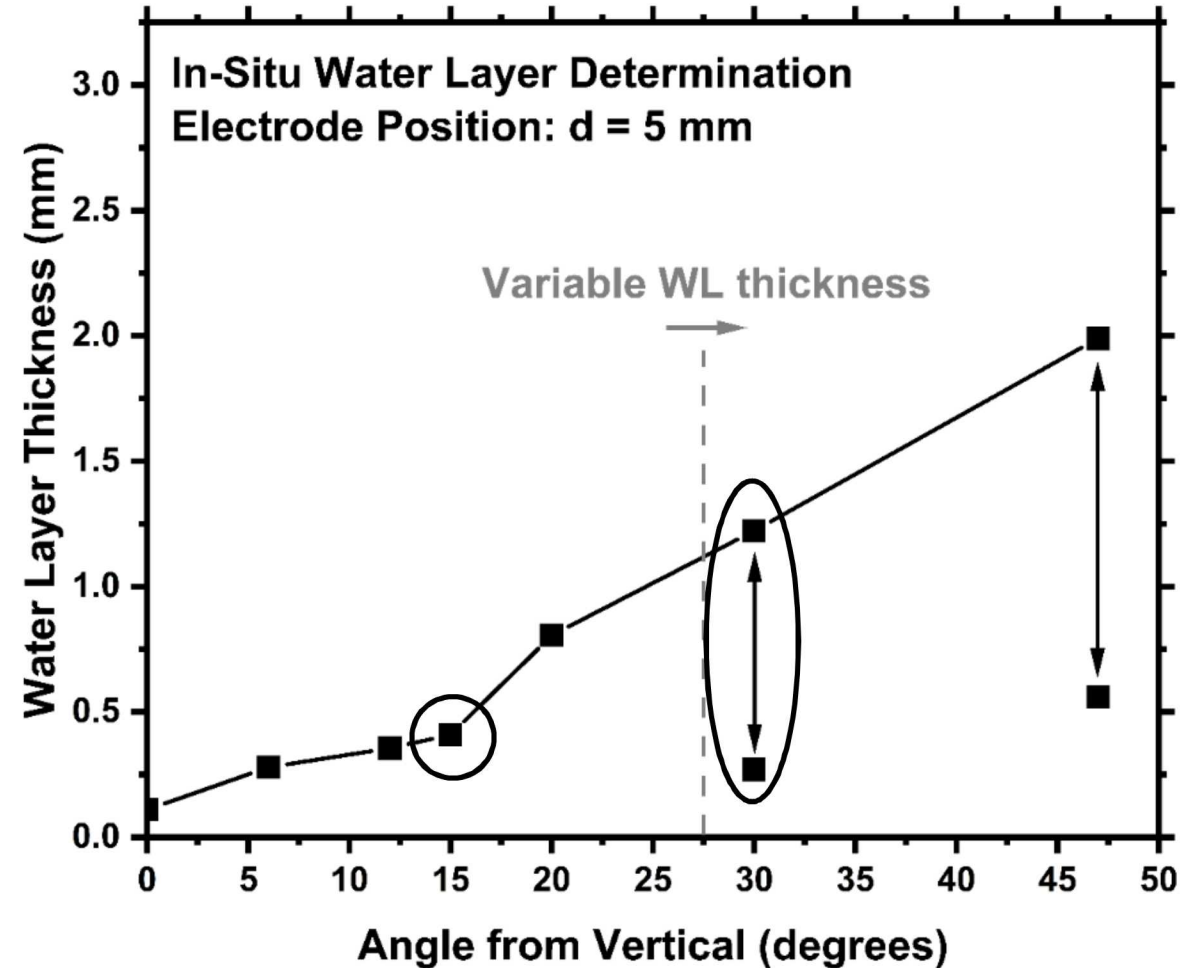
Angle of exposure varies WL thickness significantly

- Increasing angle of exposure increases WL thickness



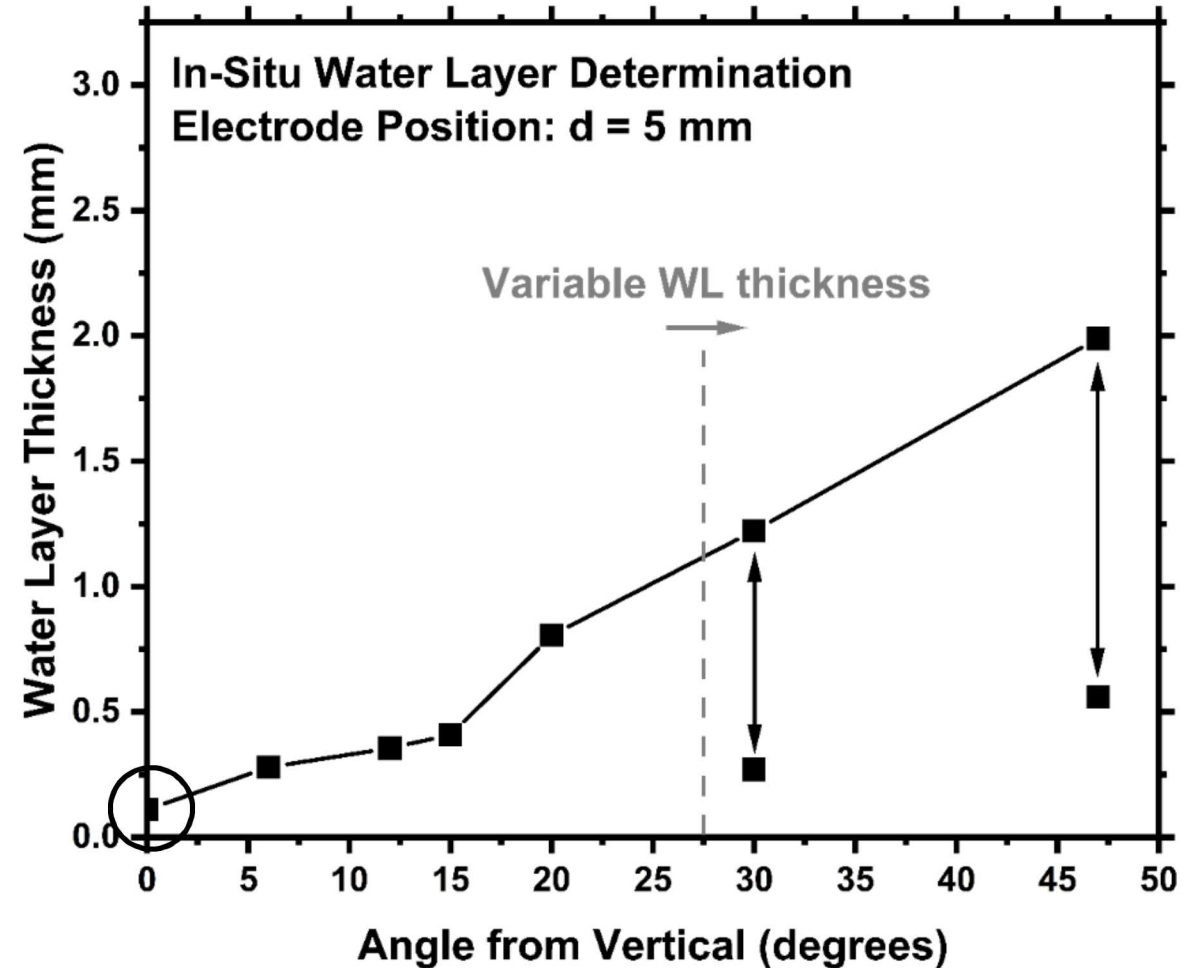
Angle of exposure varies WL thickness significantly

- Increasing angle of exposure increases WL thickness
- Semi-periodic run-off above an angle of 20°
- Within the realms of ASTM B117:
 - $15^\circ = 0.4 \text{ mm}$
 - $30^\circ = 1.25 \text{ mm}$ (but variable)

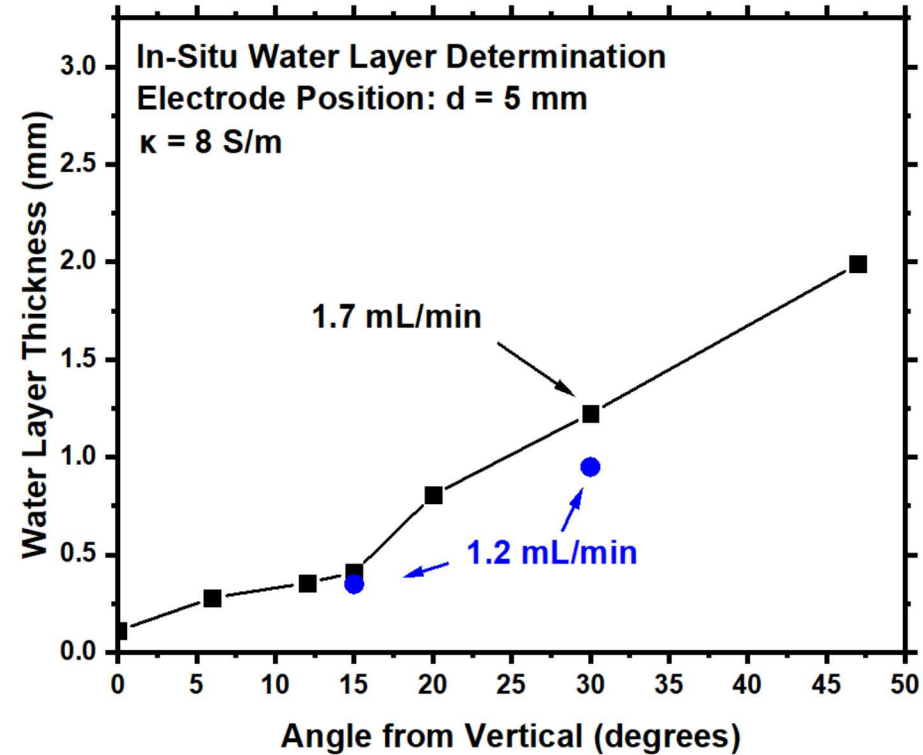


Angle of exposure varies WL thickness significantly

- Increasing angle of exposure increases WL thickness
- Semi-periodic run-off about an angle of 20°
- Within the realms of ASTM B117:
 - $15^\circ = 0.4 \text{ mm}$
 - $30^\circ = 1.25 \text{ mm}$ (but variable)
- When completely vertical, a WL of 0.11 mm was detected



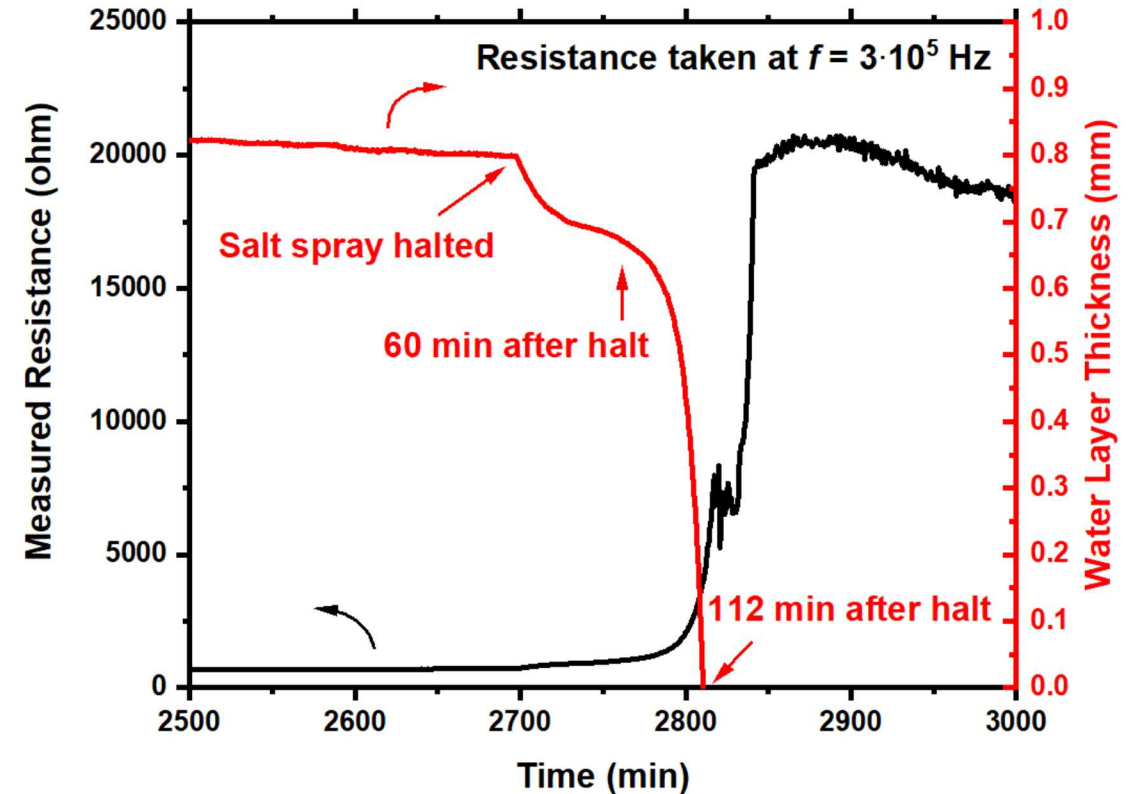
Deposition rate does not play a significant role in WL thickness



- Deposition rate does not appear to influence the WL significantly within ASTM standards

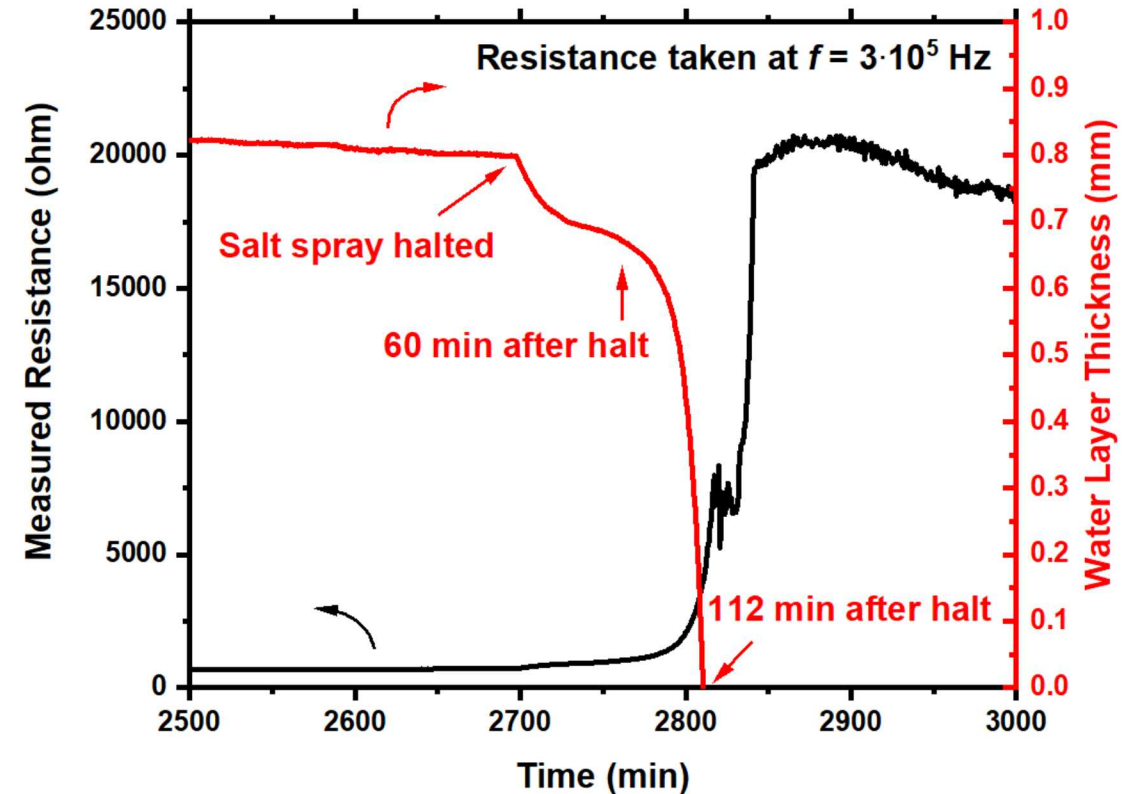
Evaporation occurs during shut-off periods

- Salt spray was stopped after roughly 45 hours of exposure and measurements were continuously taken
- Lid of chamber was left shut but fan started
- Within 1 hour of the shut-off, the WL thickness decreased 0.15 mm (18.5%)
- No WL after 112 min



Evaporation occurs during shut-off periods

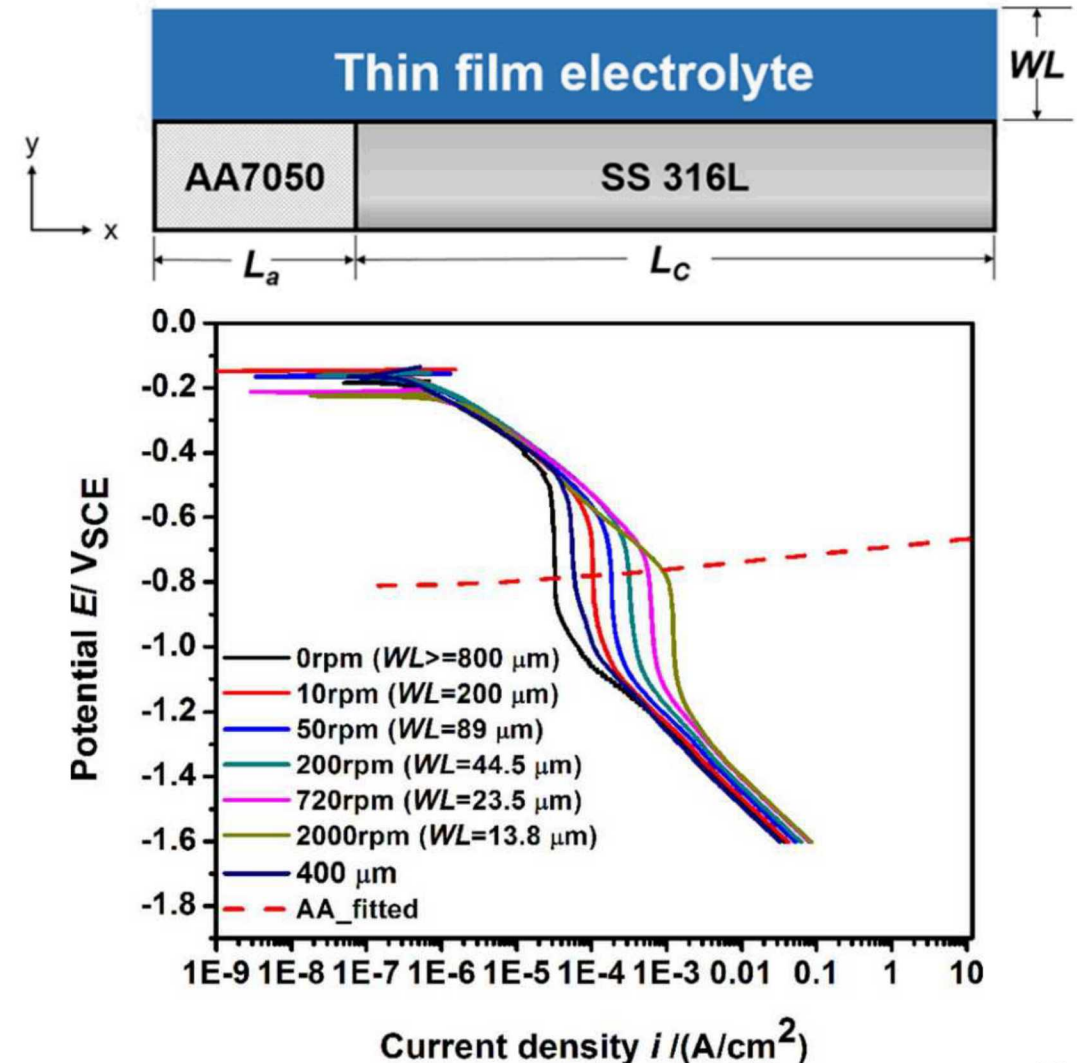
- Salt spray was stopped after roughly 45 hours of exposure and measurements were continuously taken
- Lid of chamber was left shut but fan started
- Within 1 hour of the shut-off, the WL thickness decreased 0.15 mm (18.5%)
- No WL after 112 min



Angle of sample and test interruptions are allowable within ASTM standards can drastically influence WL thickness and corrosion rate

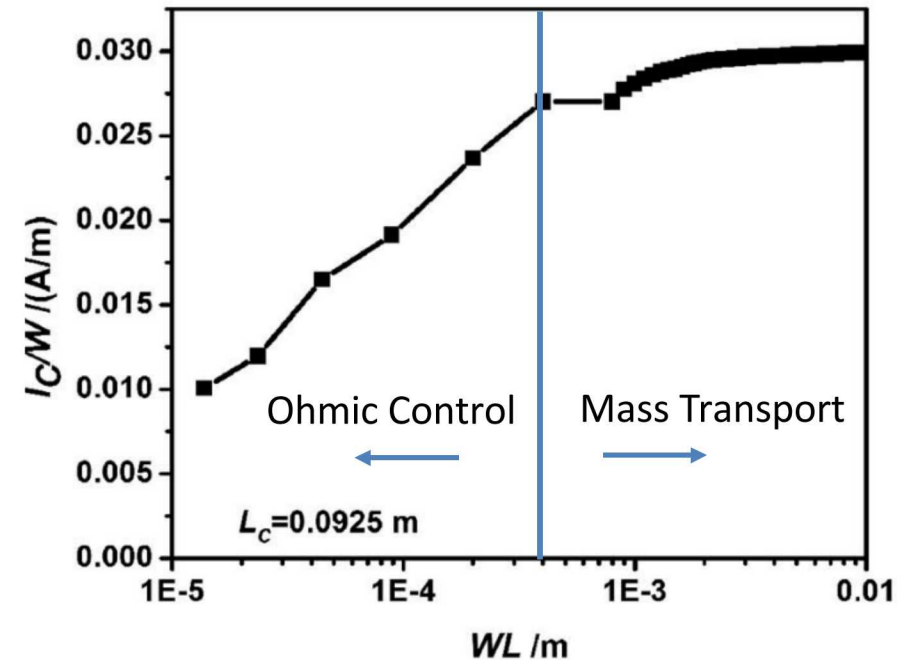
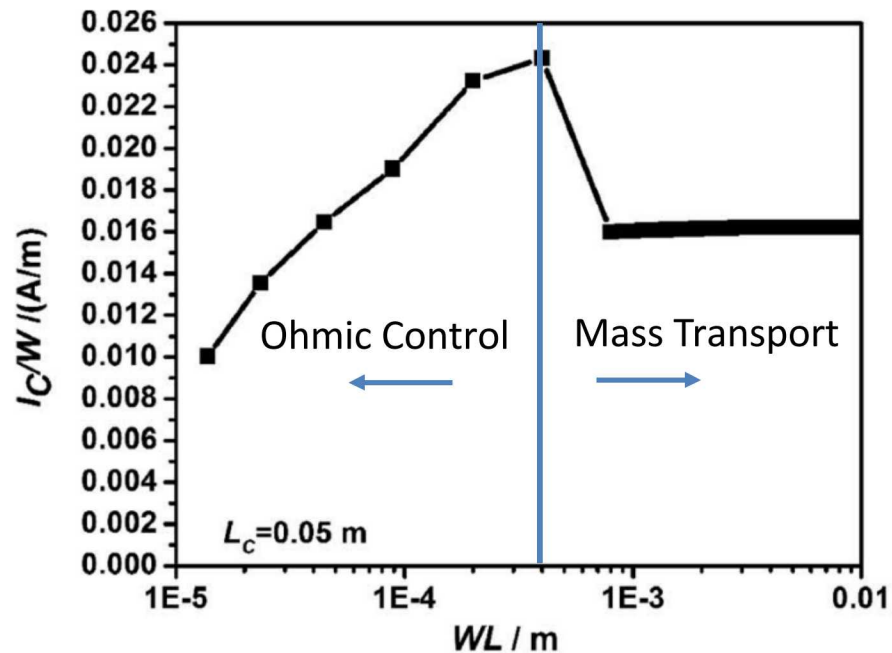
Water layer and cathode length impact cathodic current in a galvanic couple

- Liu et al. determined current in a galvanic couple as a function of cathode length and WL thickness
- Determined cathodic kinetics using rotating disk electrode
- Determined anodic kinetics using CPP
- Used FEM to model currents
 - 0.6 M NaCl
 - 25 °C



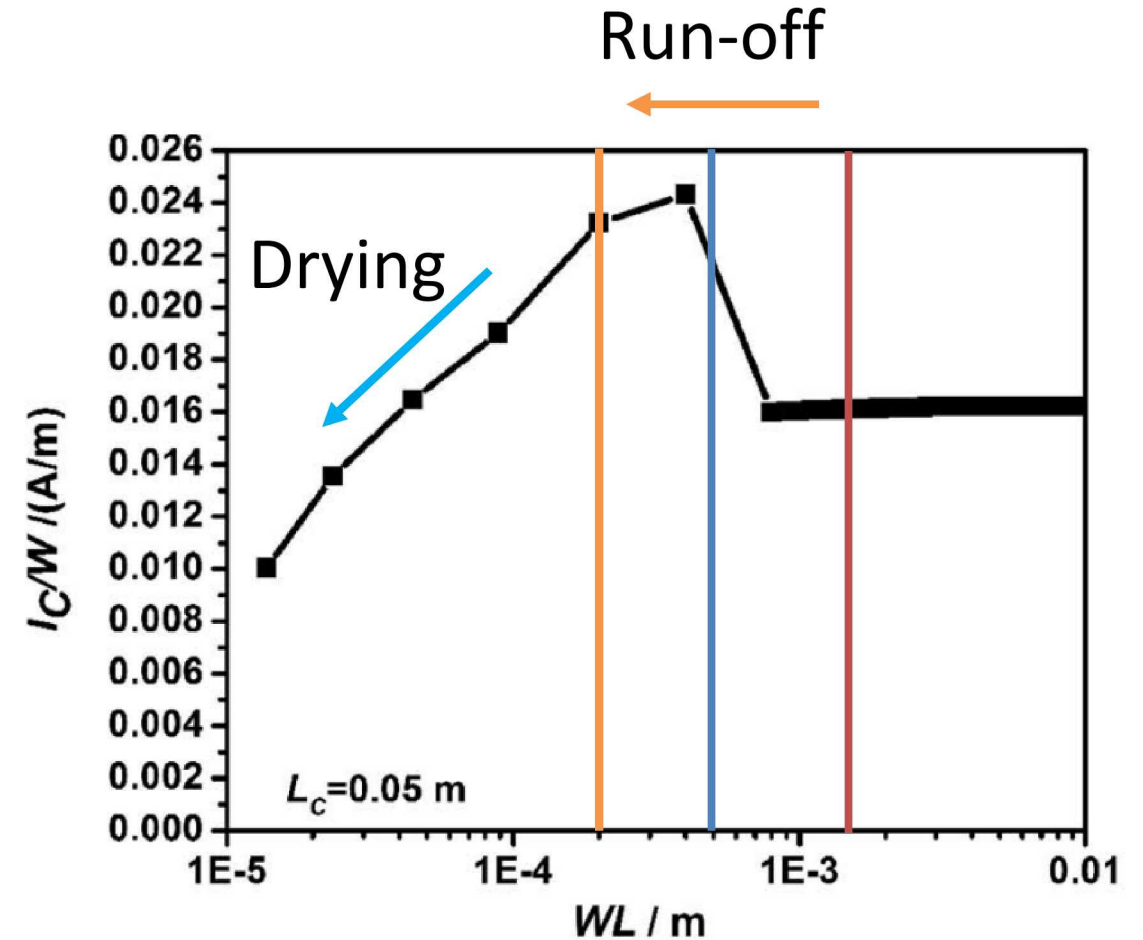
Water layer and cathode length impact cathodic current in a galvanic couple

- Both cathode size and WL influenced current available for dissolution
- Broadly, two regimes were found: (i) ohmic controlled region and (ii) mass transport regime



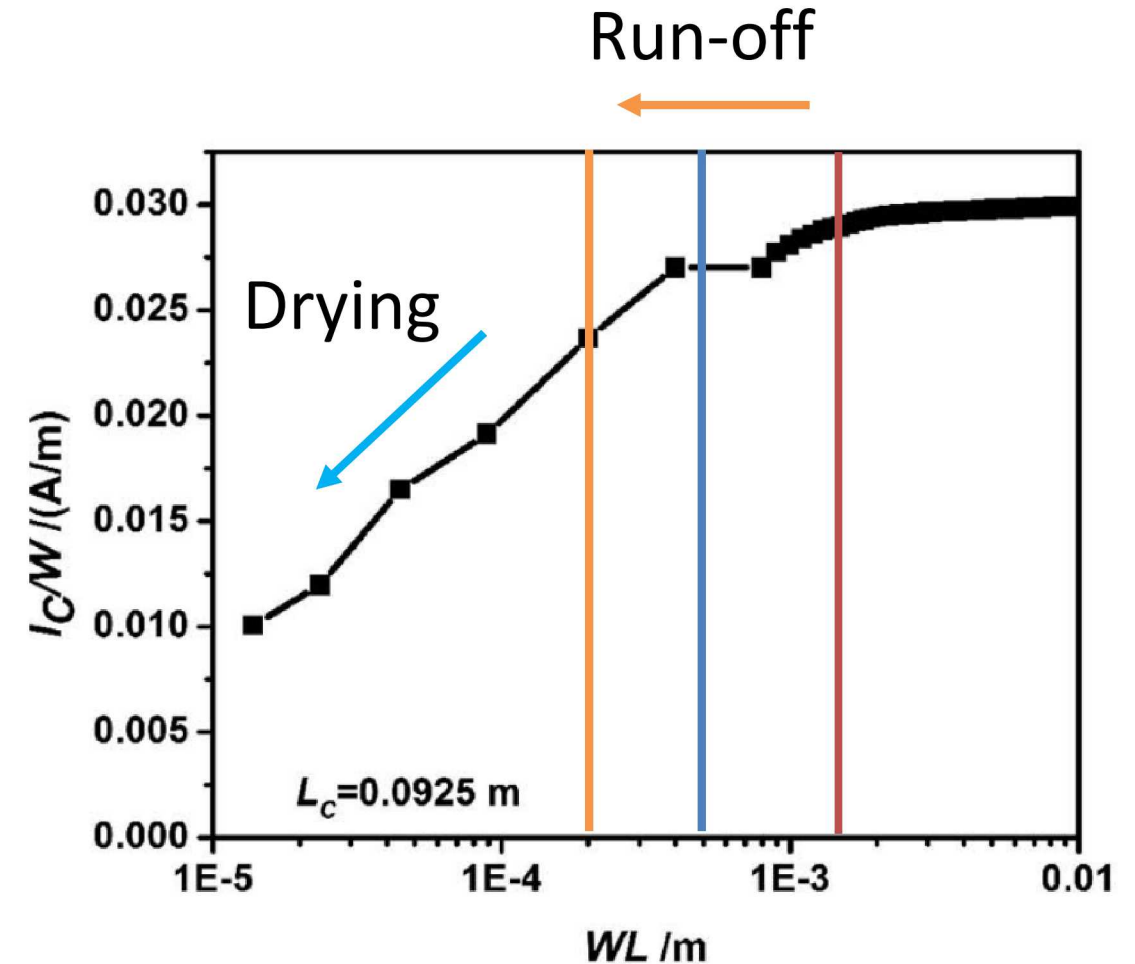
Water layer and cathode length impact cathodic current in a galvanic couple

- Increased current seen at 15°
- Mass transport limitations on current calculated at a WL corresponding to an angle of exposure of 30°
- But remember, there is run-off at high angles of exposure
- Drying will further decrease WL and current



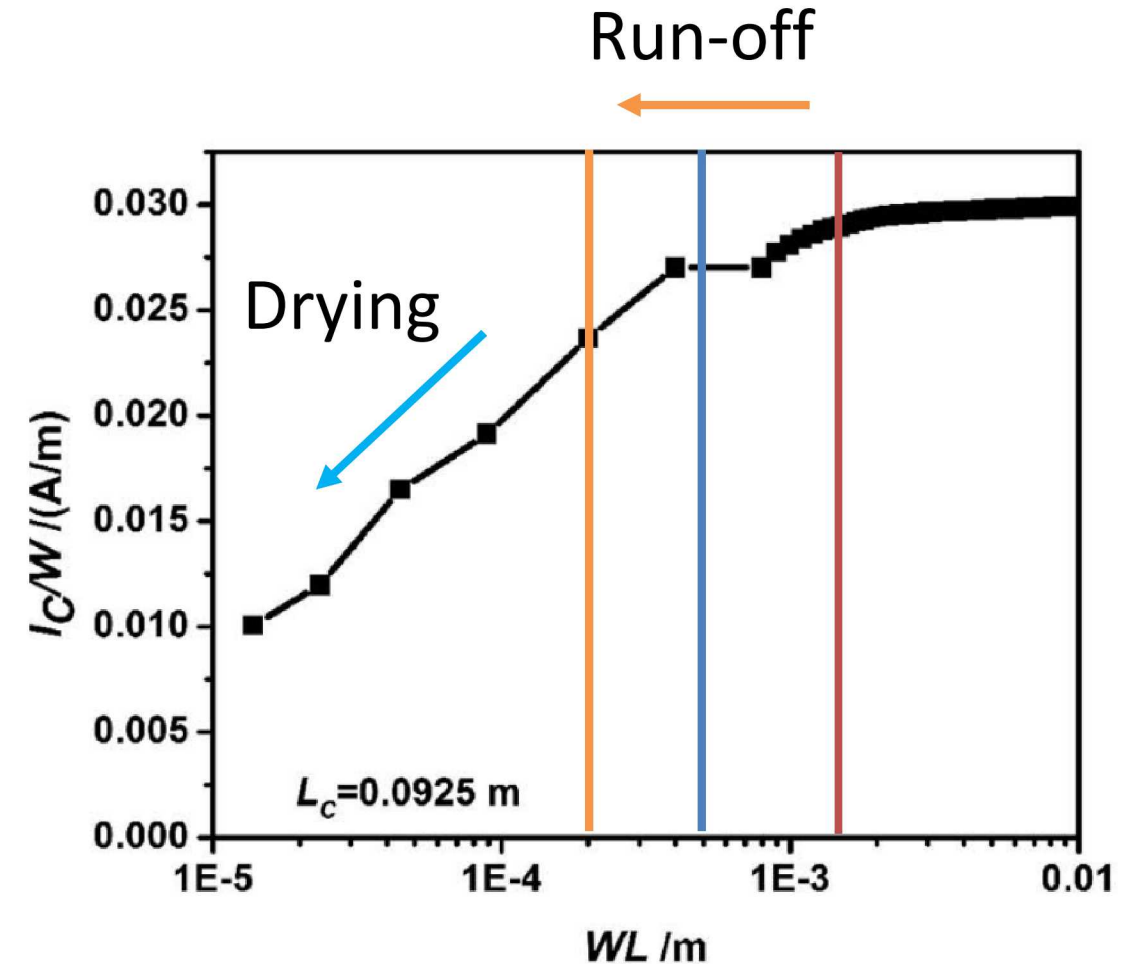
Water layer and cathode length impact cathodic current in a galvanic couple

- Highest current seen at an angle of exposure of 30°
- Decreasing angle to 15° decreases current
- Run-off and drying further decrease current



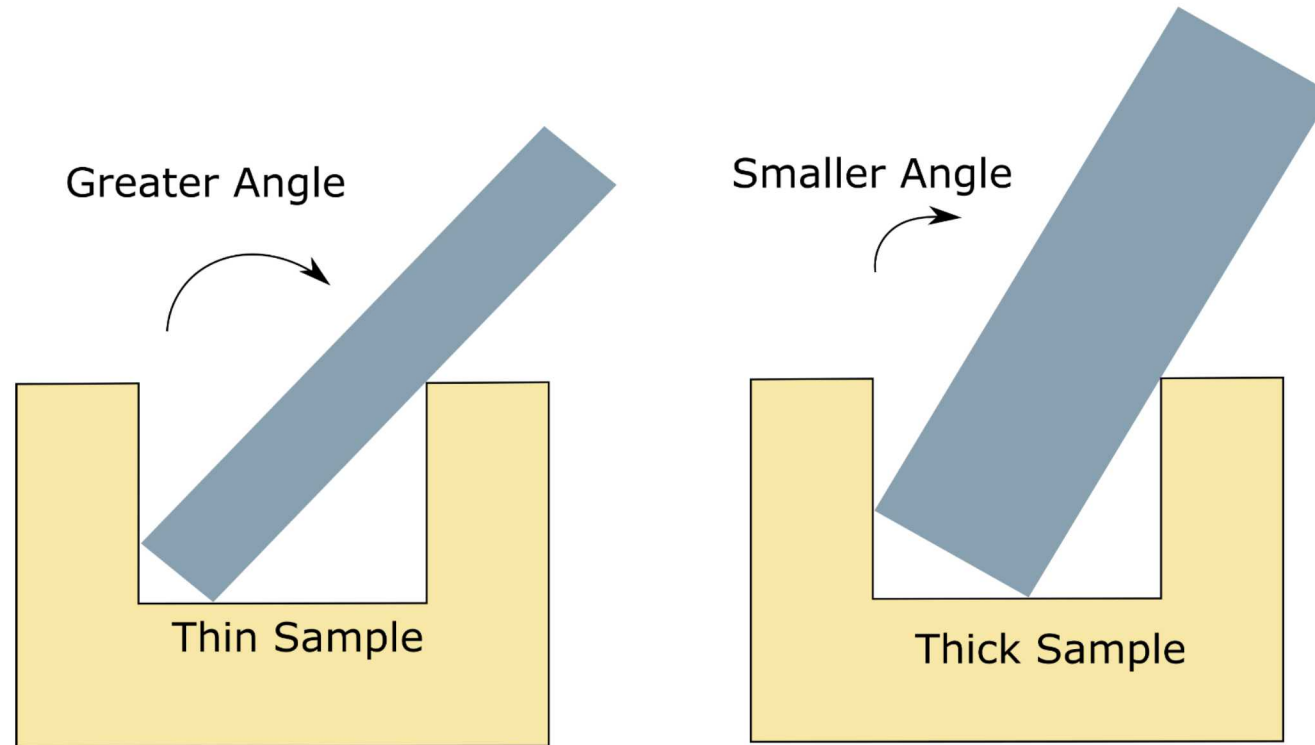
Water layer and cathode length impact cathodic current in a galvanic couple

- Highest current seen at an angle of exposure of 30°
 - Decreasing angle to 15° decreases current
 - Run-off and drying further decrease current
- Variations within ASTM accelerated standards can provide for varying currents, influencing extent of corrosion



Influence on accelerated testing

- Within standards no guidelines or restriction on the slot width for the samples which will have an impact on the sample angle



- Need for consistency in standards to avoid test-to-test and chamber-to-chamber variability

Conclusions

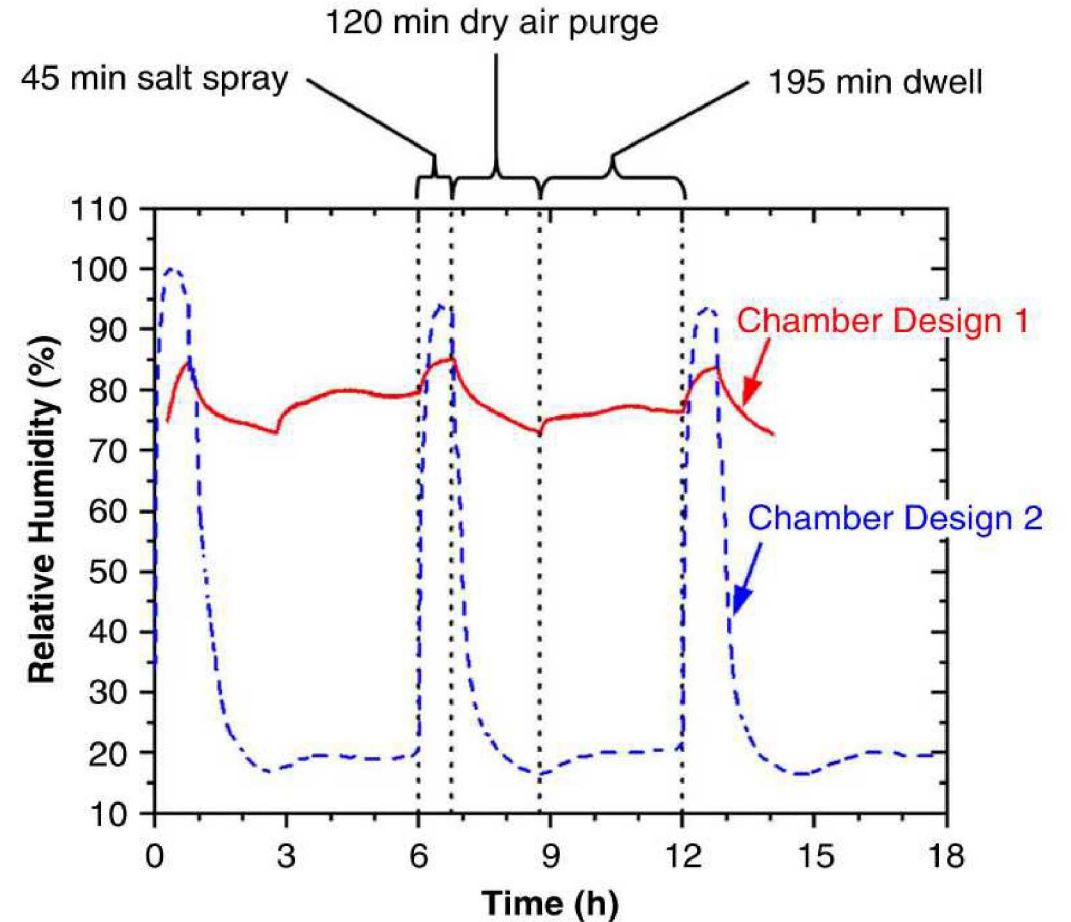
- Design, construction, validation, and application of a sensor to determine WL thickness in the range of 0 to 5 mm in salt spray testing was presented
- Ideal dimensions for sensor were established with FEM
- Utility of the sensor was shown by measuring WL thickness in a continuous salt spray test similar to ASTM B117
 - Angle of exposure plays the largest role in WL thickness
 - Semi-periodic solution run-off experienced at angles $> 20^\circ$
- Angle of sample exposure in salt spray environments determines if thin film conditions are achieved
- Need for tighter standards in accelerated testing

Acknowledgements

- P. Steiner, J. T. Burns (UVA)
- Sandia National Laboratories is a multi-mission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC., a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.
- Financial assistance from the U.S. Department of Energy's Nuclear Energy University Program under contract DE-NE0008901
- S. Tokuda was supported by JSPS KAKENHI Grant Numbers JP19J11684 and Interdepartmental Doctoral Degree Program for Multi- dimensional Materials Science Leaders in Tohoku University.

Future Work

- Creation of sensor to account for variable conductivity



M.E. Parker, R.G. Kelly, CORROSION. 76 (2020) 39–50.