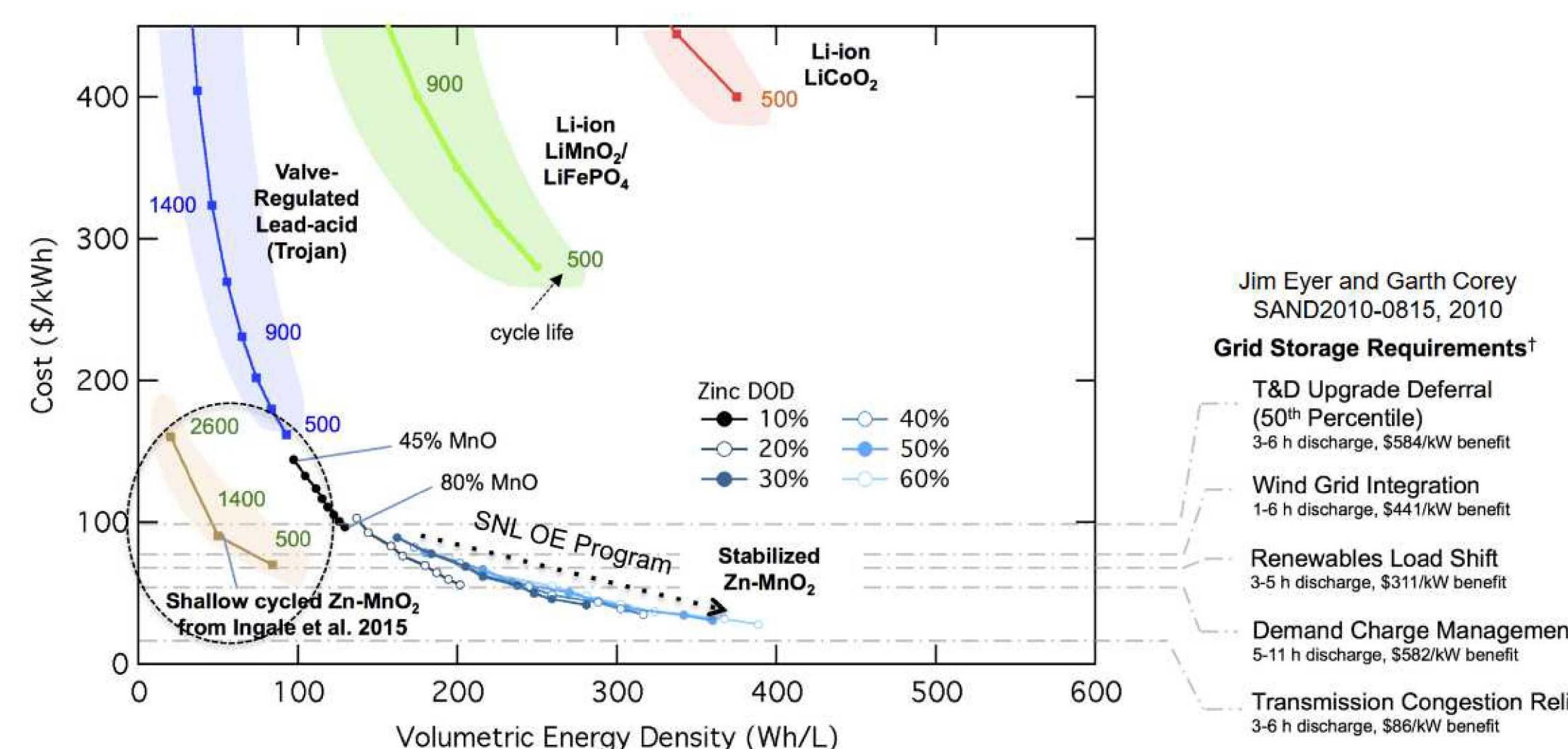


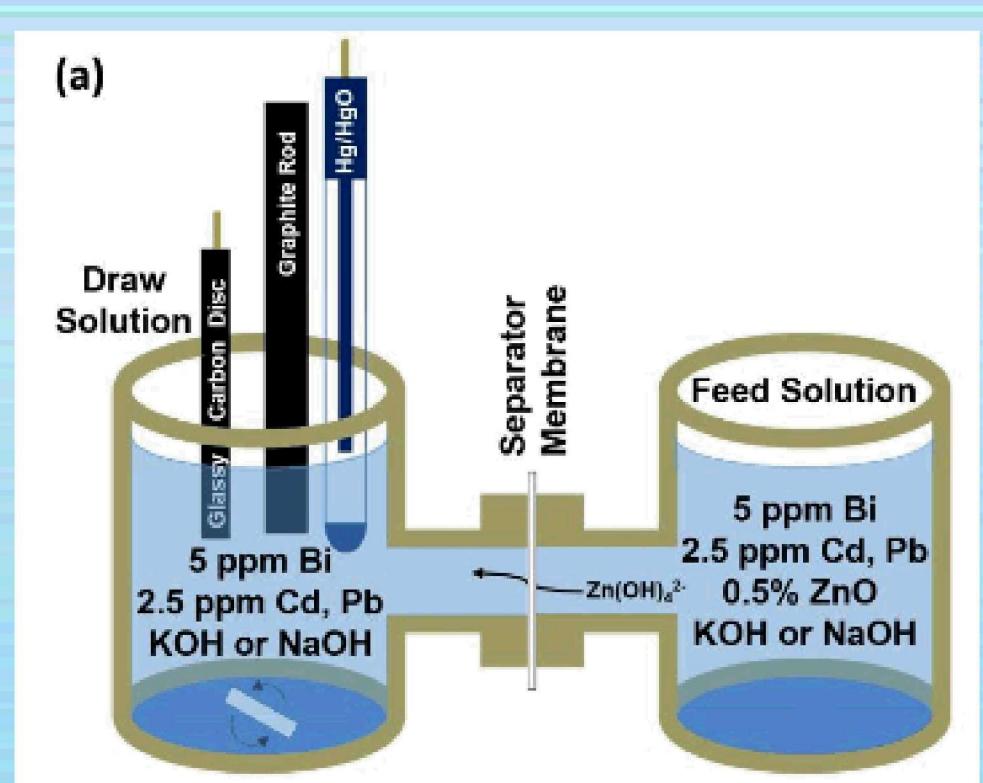
Ion Selective Separators for Rechargeable Alkaline Zn/MnO<sub>2</sub> BatteriesDavid Arnot<sup>1,2</sup>, Igor Kolesnichenko<sup>2</sup>, Jonathon Duay<sup>2</sup>, Timothy N. Lambert<sup>2\*</sup><sup>1</sup>Department of Chemical and Biological Engineering, University of New Mexico, Albuquerque, New Mexico 87131, USA<sup>2</sup>Department of Photovoltaics & Materials Technologies, Sandia National Laboratories, Albuquerque, New Mexico 87185, USA\*Email: [tnlambe@sandia.gov](mailto:tnlambe@sandia.gov)

## Abstract

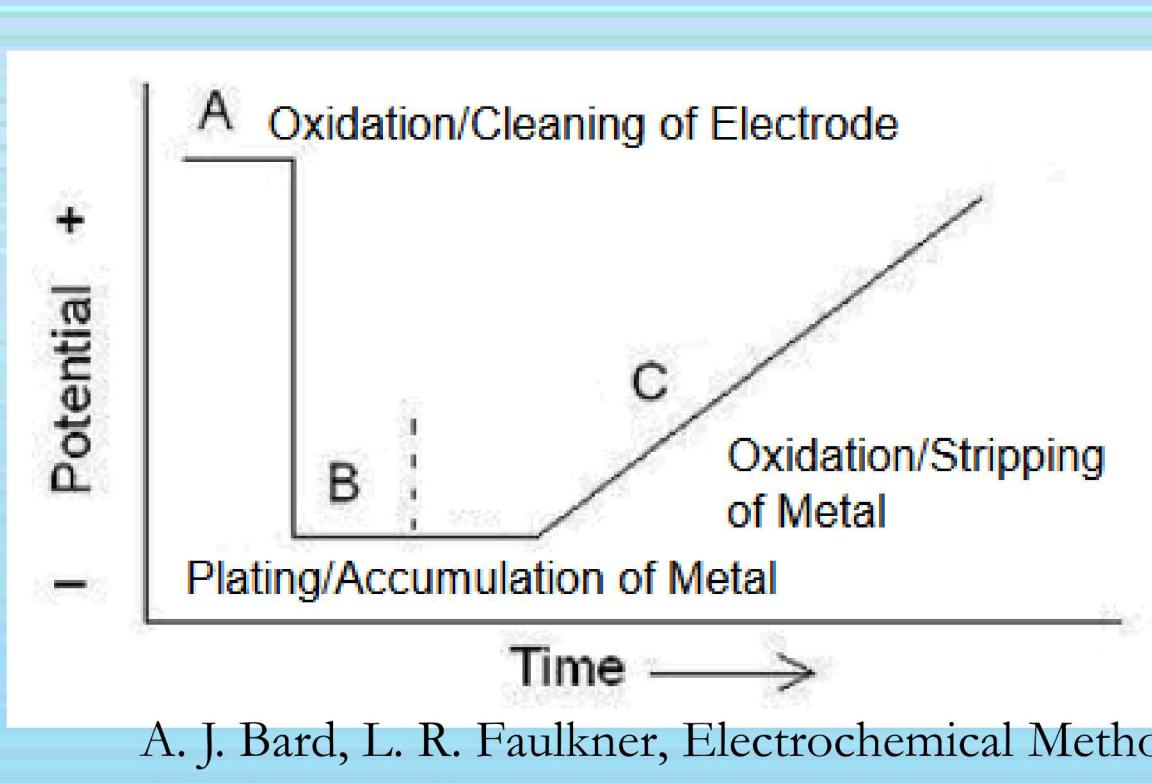
Rechargeable alkaline Zn/MnO<sub>2</sub> batteries are a good candidate for grid-level energy storage due to their low cost, theoretically high energy density, and environmental compatibility. However, performance has been hindered by the formation of electrochemically inactive ZnMn<sub>2</sub>O<sub>4</sub> phases at the cathode. Previous experimentation from our group showed that a commercial ceramic sodium ion conductor (NaSICON) separator which is impervious to zincate, increased battery cycle life by over 22%. Here, an ion selective polymeric separator which prevents the transport of zincate [Zn(OH)<sub>4</sub><sup>2-</sup>] from anode to cathode is presented.

Potential for Zn/MnO<sub>2</sub> Batteries

## Zinc Diffusion Analysis

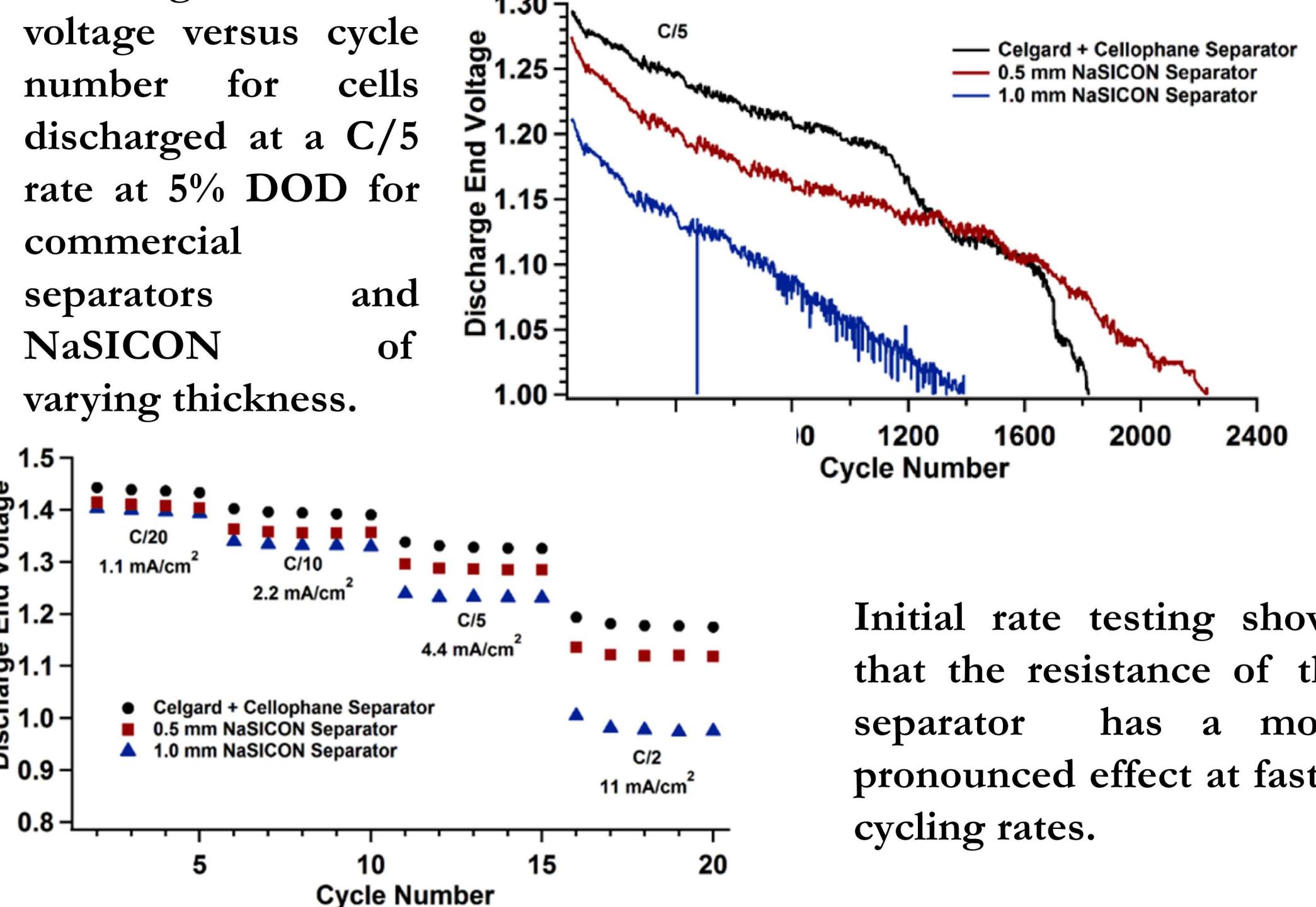


Anodic stripping voltammetry (ASV) allows for much faster screening of separators compared to ICP-MS, with similar limits of detection.

J. Duay, T.N. Lambert, R. Aidun, *Electroanalysis* 29 (2017) 1-8.A. J. Bard, L. R. Faulkner, *Electrochemical Methods*, (2001) 458-466.

## NaSICON Separator

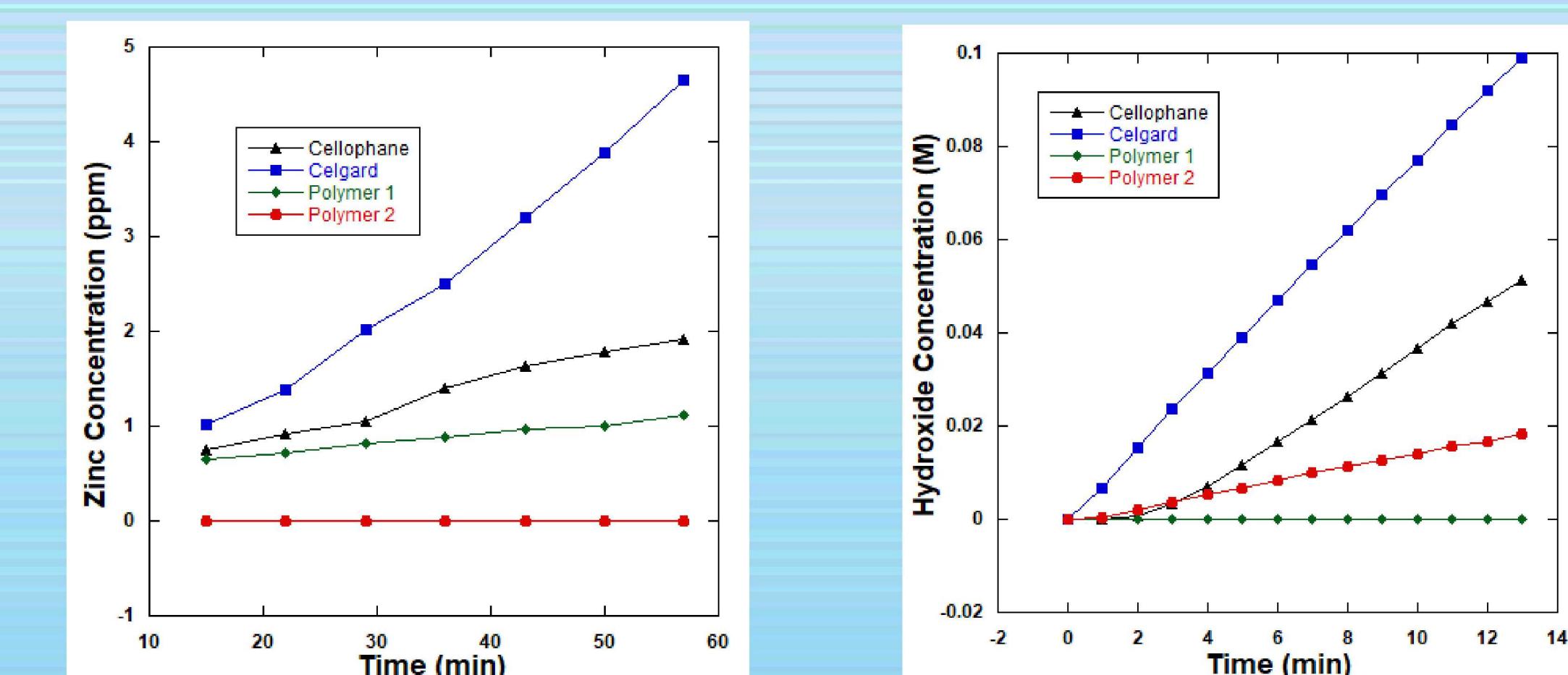
Discharge end voltage versus cycle number for cells discharged at a C/5 rate at 5% DOD for commercial and NaSICON separators of varying thickness.

J. Duay, M. Kelly, T.N. Lambert, *J. Power Sources* 395 (2018) 430-438.

Initial rate testing shows that the resistance of the separator has a more pronounced effect at faster cycling rates.

## Polymeric Separators

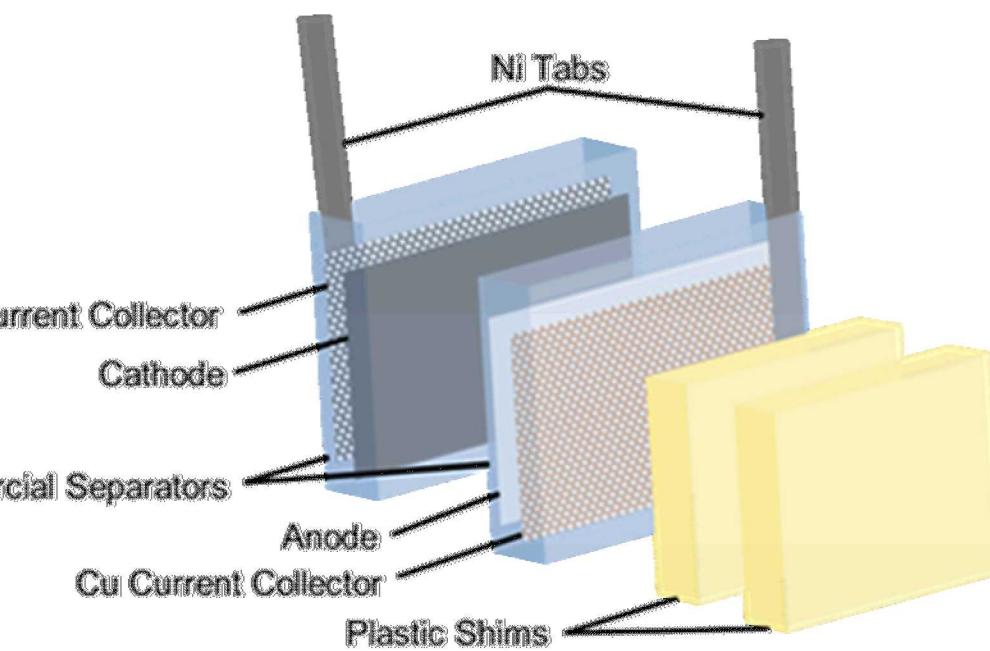
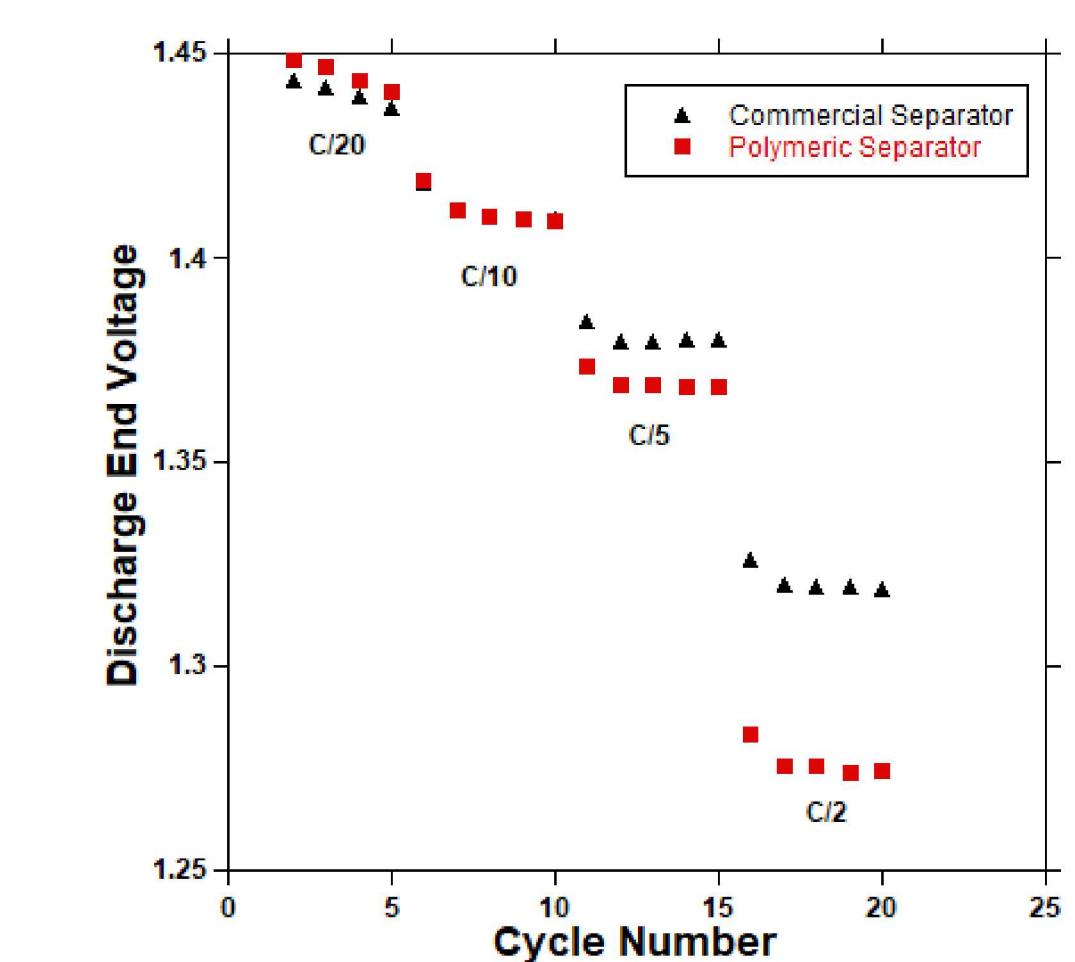
	Thickness (mm)	Resistance (Ω)	
Celgard 3501	0.025	0.1	
Cellophane 350POO	0.025	0.2	
NaSICON	0.5	9.8	Polymer separators offer lower thickness and less resistance compared to ceramic NaSICON separators.
Polymer 2	0.025	2.7	



Polymer 2 was chosen to be tested in batteries based on nearly complete zincate blocking after a period of three weeks with only minor effects on hydroxide diffusion rates.

## Battery Assembly and Testing

Batteries are assembled as shown and placed inside a sealed prismatic polysulfone case.

*Journal of Power Sources* 395 (2018) 430-438

Cells cycled at 5% DOD relative to 1e- MnO<sub>2</sub> discharge and varying rates with commercial and polymeric separators show smaller differences in discharge end voltage than those with ceramic NaSICON separators.

## Conclusion

Accelerating the integration of intermittent renewable power sources is an important part of reducing our dependence on carbon-burning sources of energy. Rechargeable Zn/MnO<sub>2</sub> batteries represent an exciting grid-level energy storage solution. In this work, flexible polymeric separators were evaluated and displayed favorable diffusion properties. Zincate diffusion across the separator was slowed down to an immeasurable rate with more moderate effects on conductivity and hydroxide transport. In the future, these polymeric separators will be improved further through new fabrication techniques and decreases in thickness. Lowering the resistance of the separators will allow for higher rates and depths of discharge to be obtained, increasing power output and volumetric energy density.

## Acknowledgements

This work was supported by Sandia National Laboratories; Sandia National Laboratories is a multi-mission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.

Dr. Imre Gyuk, Energy Storage Program Manager, Office of Electricity Delivery and Energy Reliability and the SNL-LDRD program are thanked for their financial support of this project. The authors also thank Dr. Erik Spoerke for the procurement of the NaSICON discs, Dr. Eric Allcorn for help with the design and 3D-Printing of the H-cell, and Alice Kilgore for mechanically thinning the NaSICON discs used here.