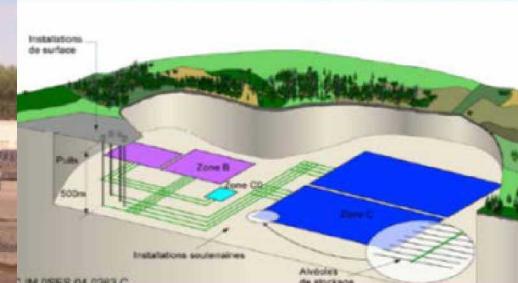




# A Stochastic Model of Degradation Behavior of Tristructural-isotropic Coated Particle Spent Fuels



PRESENTED BY

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- Introduction
  - U.S. Program Wastes
  - Tristructural Isotropic (TRISO) Particle Fuels
- Considerations for TRISO Spent Fuels
  - Features, Events, and Processes (FEP)
  - Coupled SiC Layer Corrosion and Radionuclide Diffusion  
(coauthor Fred Gelbard, SNL)
- Summary and Conclusions

# Spent Nuclear Fuel and High-Level Radioactive Waste Disposal: The Goal

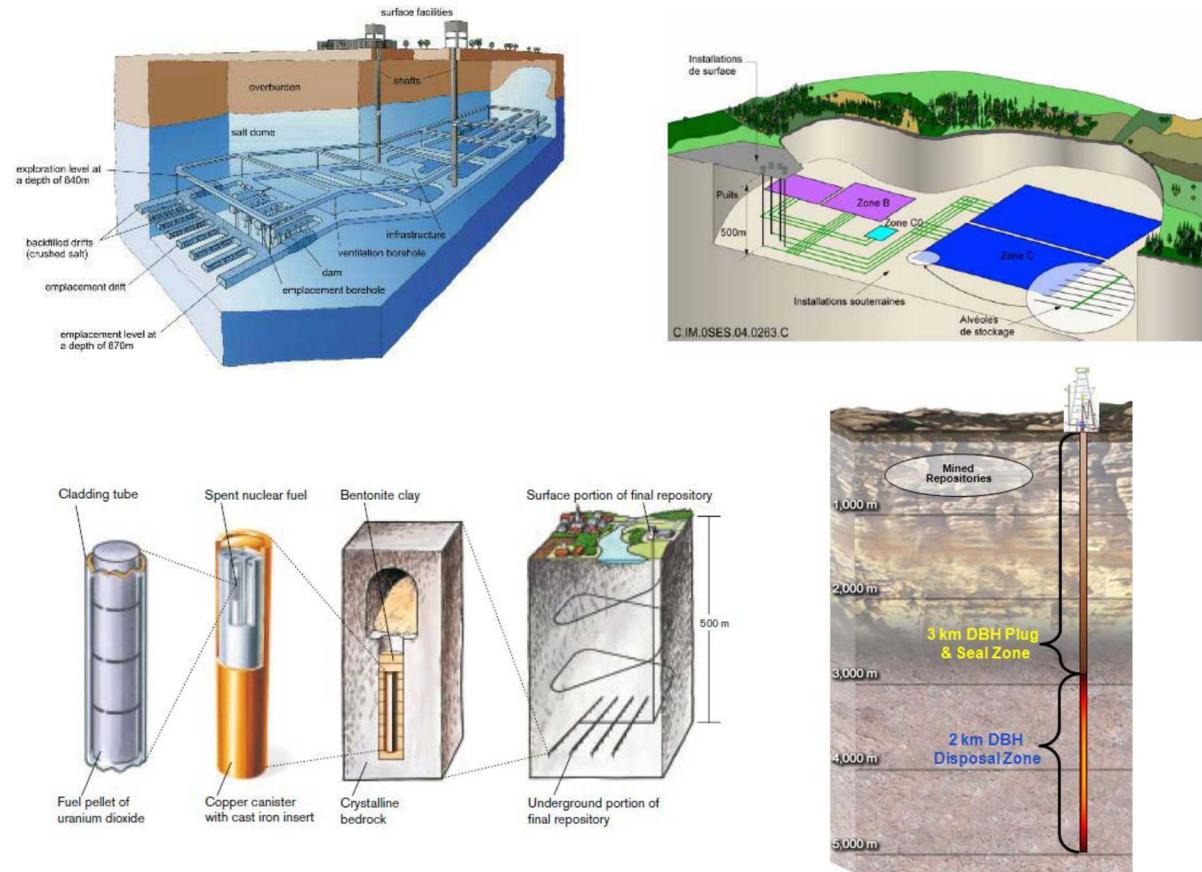


“There has been, for decades, a worldwide consensus in the nuclear technical community for disposal through geological isolation of high-level waste (HLW), including spent nuclear fuel (SNF).”

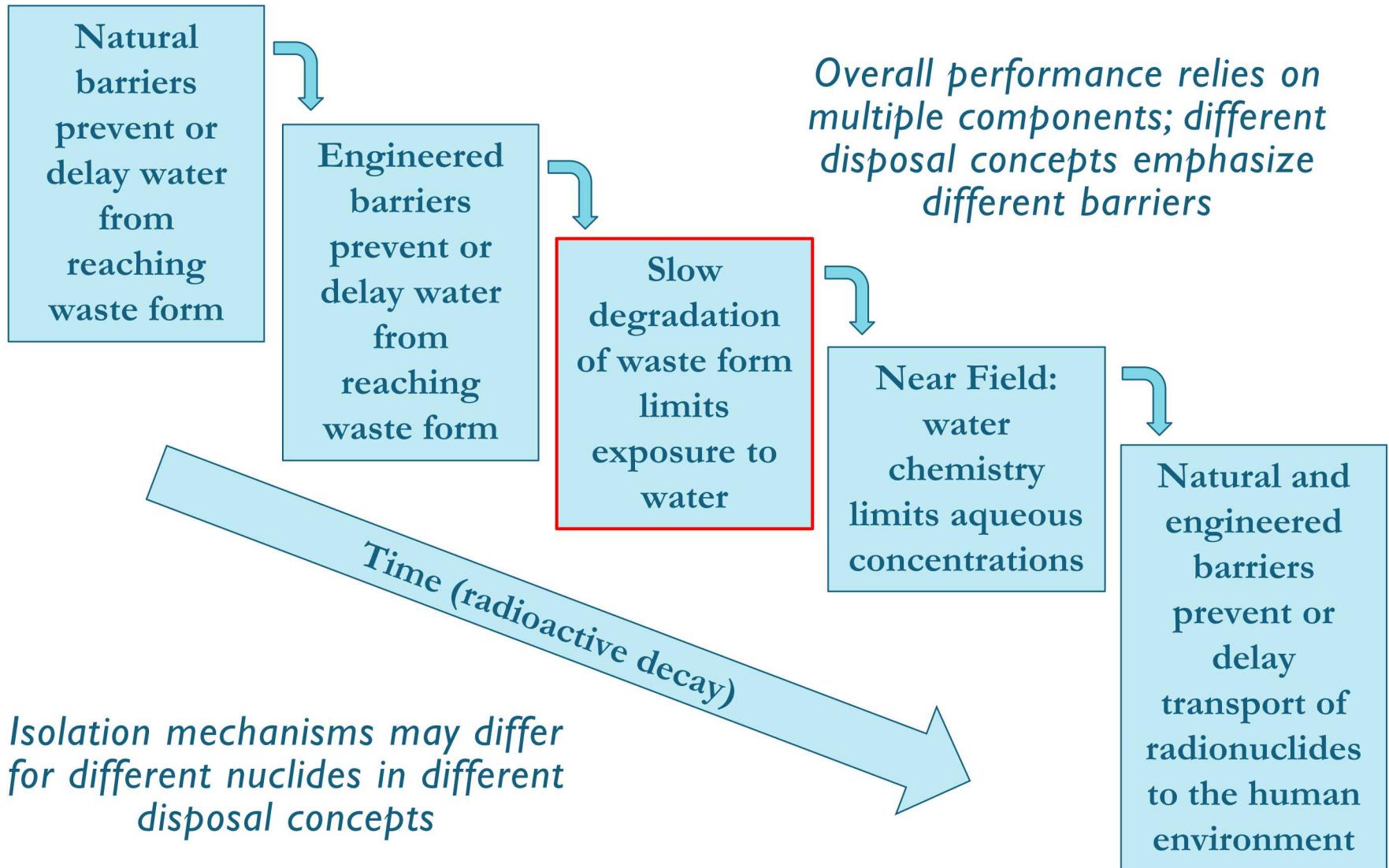
“Geological disposal remains the only long-term solution available.”

National Research Council, 2001

*Deep geologic disposal has been planned since the 1950s (SNL, 2014 provides recent analysis of disposal options)*



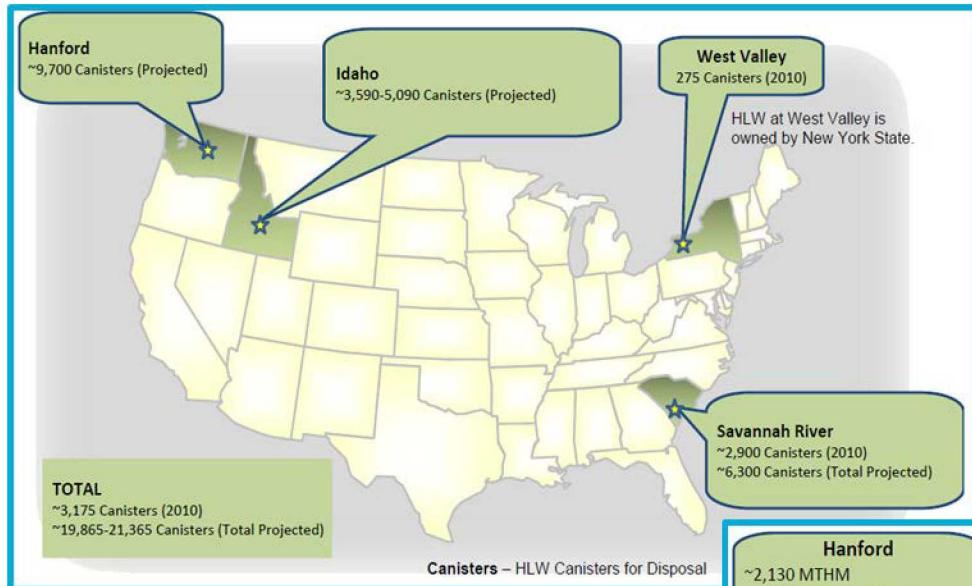
# How Repositories Work



# Geologic Disposal in the US: The Reality



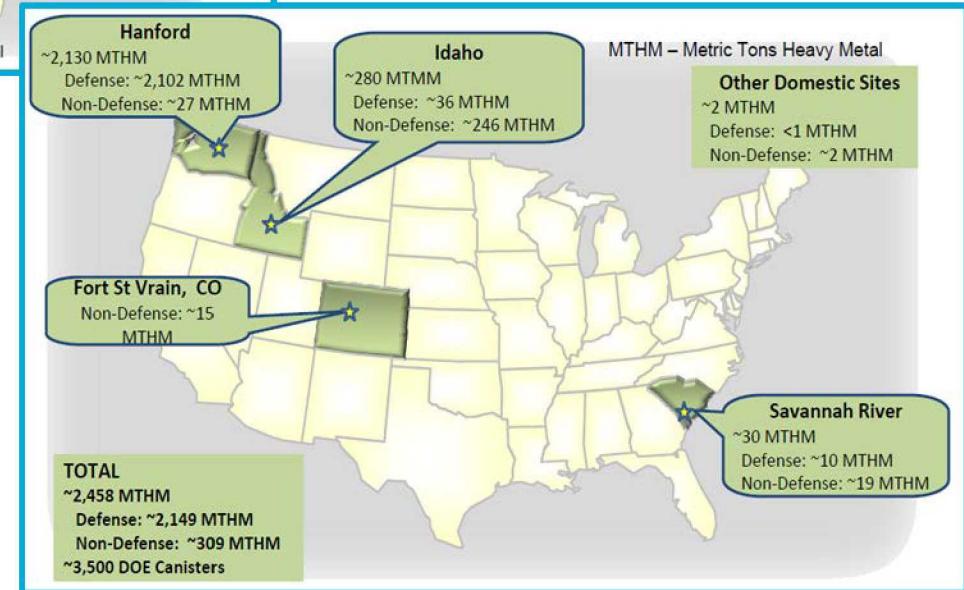
*DOE-managed SNF and HLW is in Temporary Storage at 5 Sites in 5 States*



**DOE-Managed HLW**

~20,000 total canisters (projected)

**DOE-Managed SNF**  
~2,458 Metric Tons  
FSVR TRISO (~23.5 MTHM)



Source: Marcinowski, F., "Overview of DOE's Spent Nuclear Fuel and High-Level Waste," presentation to the Blue Ribbon Commission on America's Nuclear Future, March, 25, 2010, Washington, DC.

# TRISO Particle Fuels Overview



Characteristics of TRISO fuel with a  $UO_x$  core of radius 250 - 300  $\mu\text{m}$

Layer	Nominal Thickness ( $\mu\text{m}$ )	Purpose/function
porous pyrolytic carbon buffer	60 - 95	- allows kernel to swell - stops recoiling fission products from reaching SiC layer - provides void volume for gases
inner dense pyrolytic carbon (IPyC)	30 - 40	- barrier to gaseous fission products - slows down metallic fission product transport
Silicon Carbide (SiC)	25 - 35	- main fission product barrier - structural support to contain gas pressure
Outer dense pyrolytic carbon (OPyC)	40 - 45	- protects SiC layer from chemical and mechanical damage - adds to support to contain gas pressure

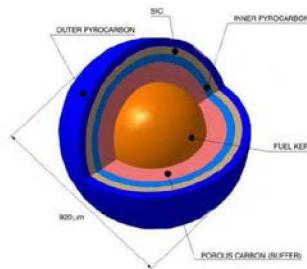
Sources: Minato et al., (1994); Moormann, et al., (2001); Nabielek et al., (2010); Fachinger (2006).

- Previous work by van den Akker and Ahn (2013) evaluated releases in repository setting
  - Relies mainly on graphite matrix chemical longevity (oxidation)
  - Fuel element graphite
  - Individual graphite compacts

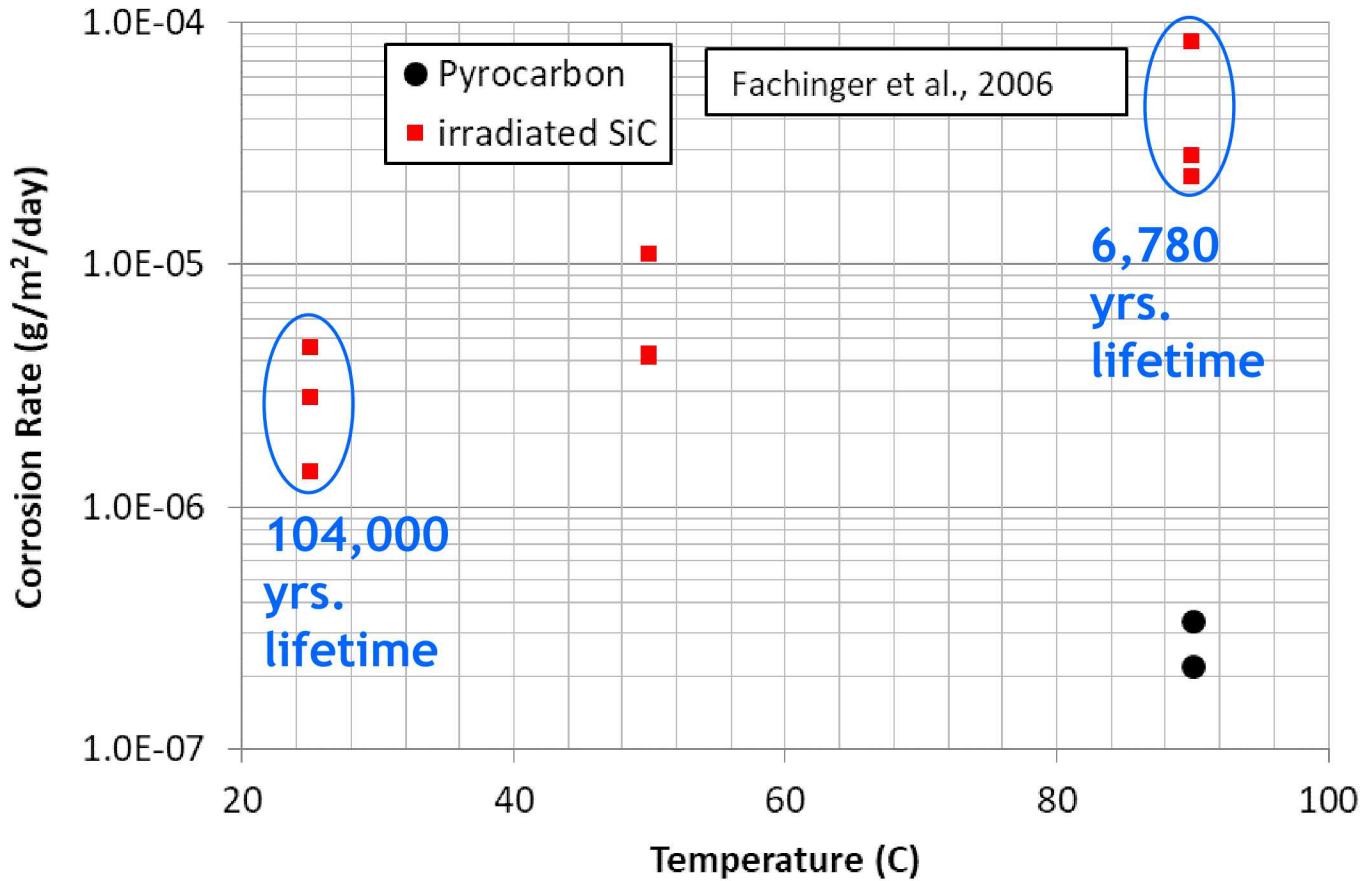
# 7 Degradation and Release Mechanisms for TRISO Particles



- Corrosion of SiC and PyC layers
  - Using Fachinger et al. (2006) data for various fluids and temperatures
- Models of van den Akker and Ahn (2013) to assess SiC layer rupture dependent on
  - Corrosion rates;
  - Possible protection via outer PyC
  - Helium internal pressure buildup
  - Statistical variability of SiC strength
- Additional FEP Considerations
  - Degradation mechanisms for graphite matrix (elements and compacts)
    - Seismic disruption: (compacts likely more durable than elements)
      - Porosity/permeability evolution over time – advective pathways
  - Diffusive release through graphite matrix
    - Compare to lifetimes  $10^6$  –  $10^8$  yrs
    - Condition of particles (e.g., location of radionuclides; Demkowicz et al., 2017)
  - Diffusion of radionuclides through particles (Gelbard, 2002)
    - Diffusivities not readily available for SiC
    - Sensitivity study of diffusion compared to SiC layer corrosion
    - Coupled diffusion and corrosion of SiC layer
    - Kinetic models used to estimate/assess magnitude of diffusivities



# SiC layer corrosion lifetimes for TRISO fuel



- SiC layer corrosion data for different brines indicate:
  - At  $90^\circ\text{C}$  the  $35\ \mu\text{m}$  layer would last  $\sim 7000$  years (average at constant T)
  - At  $25^\circ\text{C}$  the layer would last  $\sim 100,000$  years (average at constant T)
- Estimated layer lifetime will depend on thermal evolution
  - Uncertainties: corrosion rate, thermal history, hydrologic condition ( $\sim 10^4$  to  $10^5$  years)
  - Pyrolitic carbon layer protection may add to lifetime ( $\sim 10^6$  years; van den Akker and Ahn, 2013)

- **Repositories:**
  - $>90^\circ\text{C}$  for 100's to  $\sim 1500$  yrs
  - Lack of water in WP to corrode SiC
  - $<50^\circ\text{C}$   $\sim 10^3$  to  $10^4$  yrs

# Example radionuclides of interest for geological repositories



## Radionuclide Half-life (years) Decay Product

$^{129}\text{I}$   $1.57 \times 10^7$   $^{129}\text{Xe}$

$^{36}\text{Cl}$   $3.01 \times 10^5$   $^{36}\text{Ar}$

$^{226}\text{Ra}$   $1.60 \times 10^3$   $^{222}\text{Rn} \rightarrow ^{218}\text{Po} \rightarrow ^{214}\text{Pb} \rightarrow ^{214}\text{Bi} \rightarrow ^{214}\text{Po} \rightarrow ^{210}\text{Pb}$   
 $\rightarrow ^{210}\text{Bi} \rightarrow ^{210}\text{Po} \rightarrow ^{206}\text{Pb}$

$^{79}\text{Se}$   $2.95 \times 10^5$   $^{79}\text{Br}$

$^{99}\text{Tc}$   $2.13 \times 10^5$   $^{99}\text{Ru}$

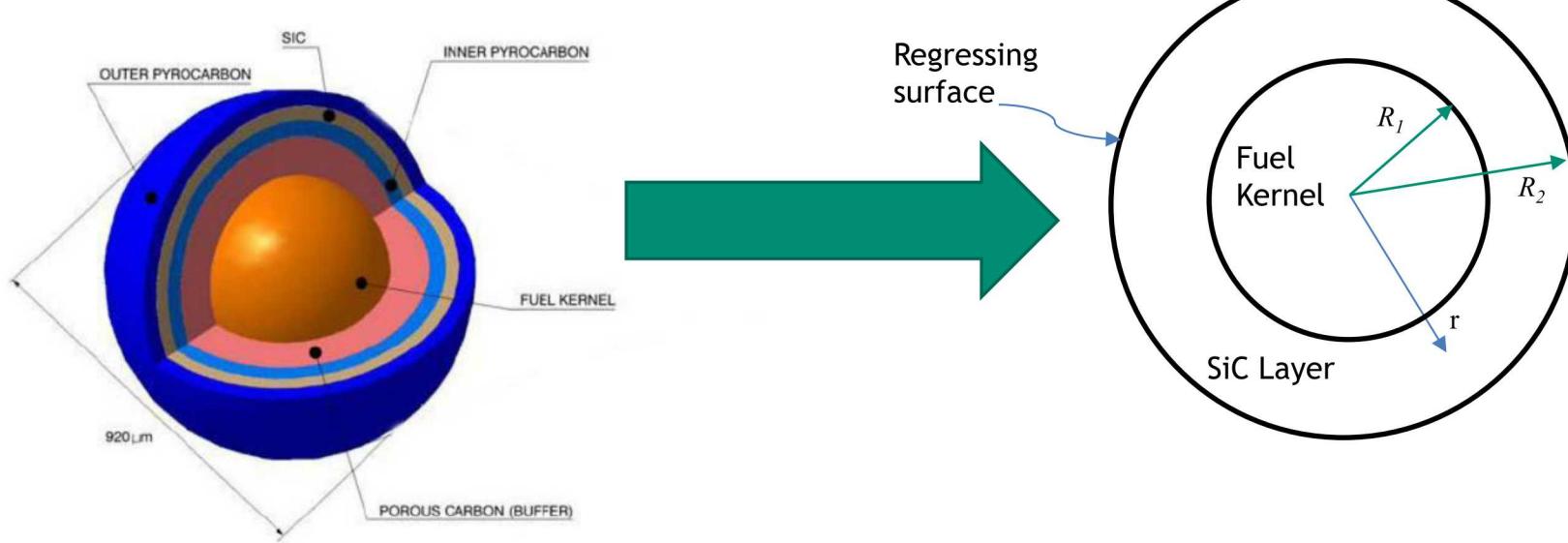
## Actinides

- Data for diffusivities through SiC
- Not readily available in general, and specifically at repository temperatures
- Conduct sensitivity analyses for coupled radionuclide diffusion and SiC layer corrosion to identify threshold diffusivities

# Simplifications for simultaneous SiC layer corrosion and radionuclide diffusion

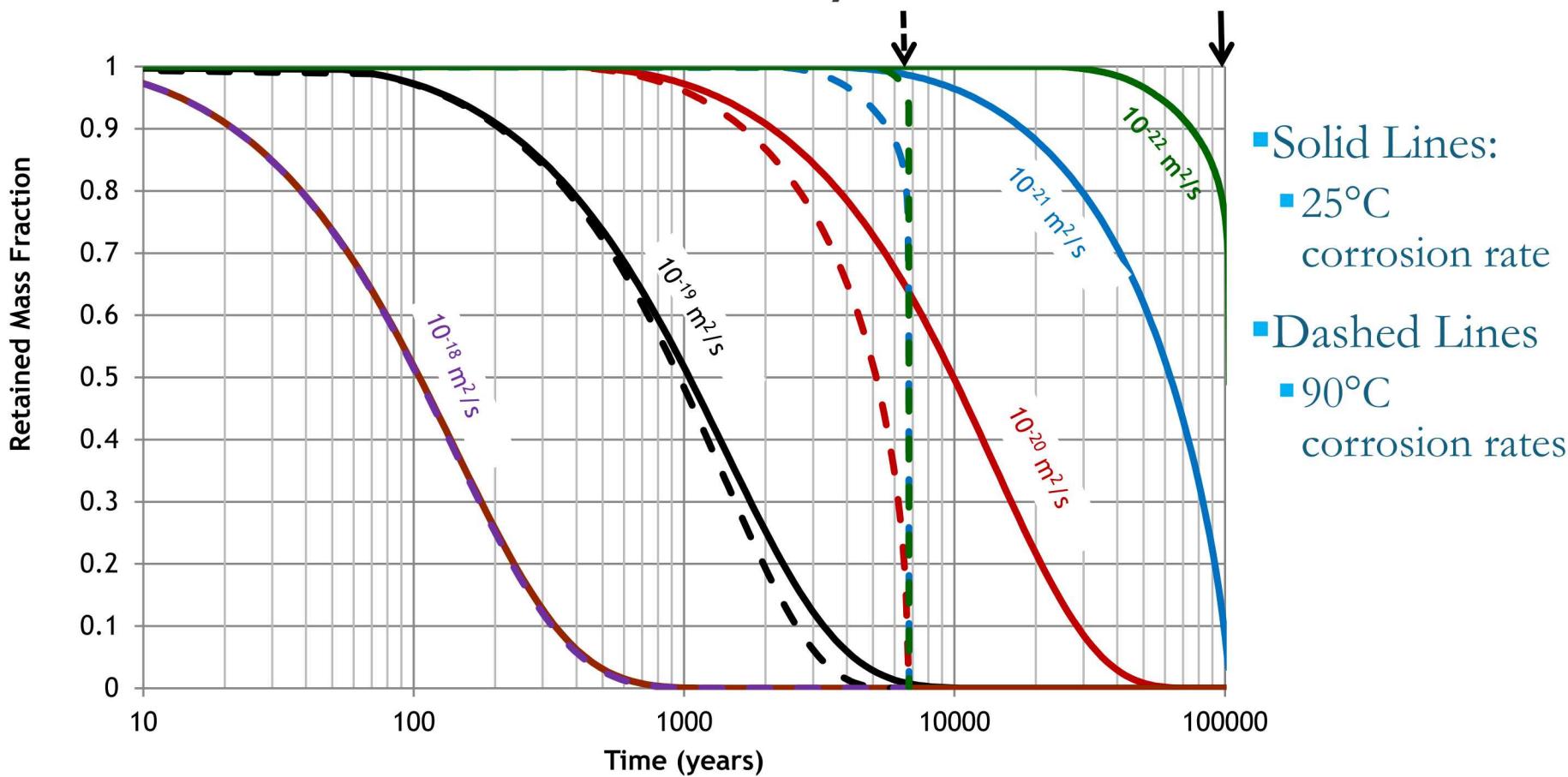


- Diffusivities of  $^{36}\text{Cl}$ ,  $^{90}\text{Sr}$ , and  $^{134}\text{Cs}$  through graphite in different brines range from  $1.2 \times 10^{-13} \text{ m}^2/\text{s}$  to  $6.3 \times 10^{-13} \text{ m}^2/\text{s}$  (Fachinger et al., 2006)
  - Likely faster than for denser SiC, so assume instant transport for carbon layers



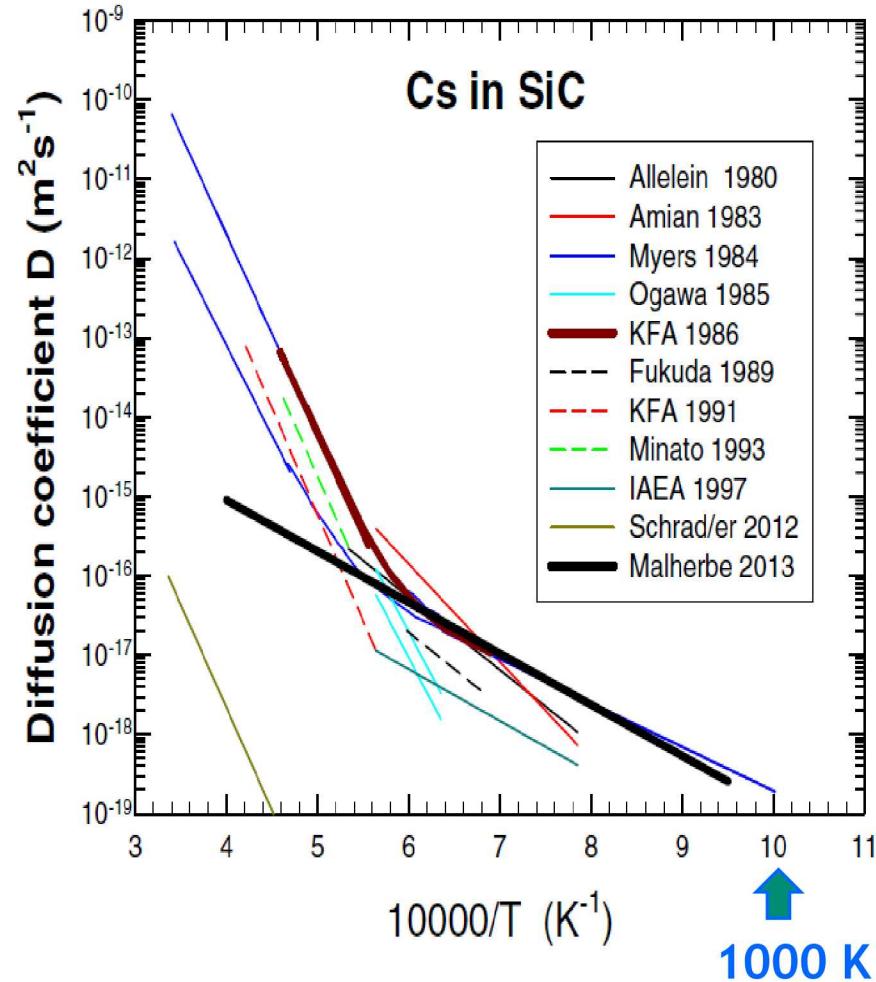
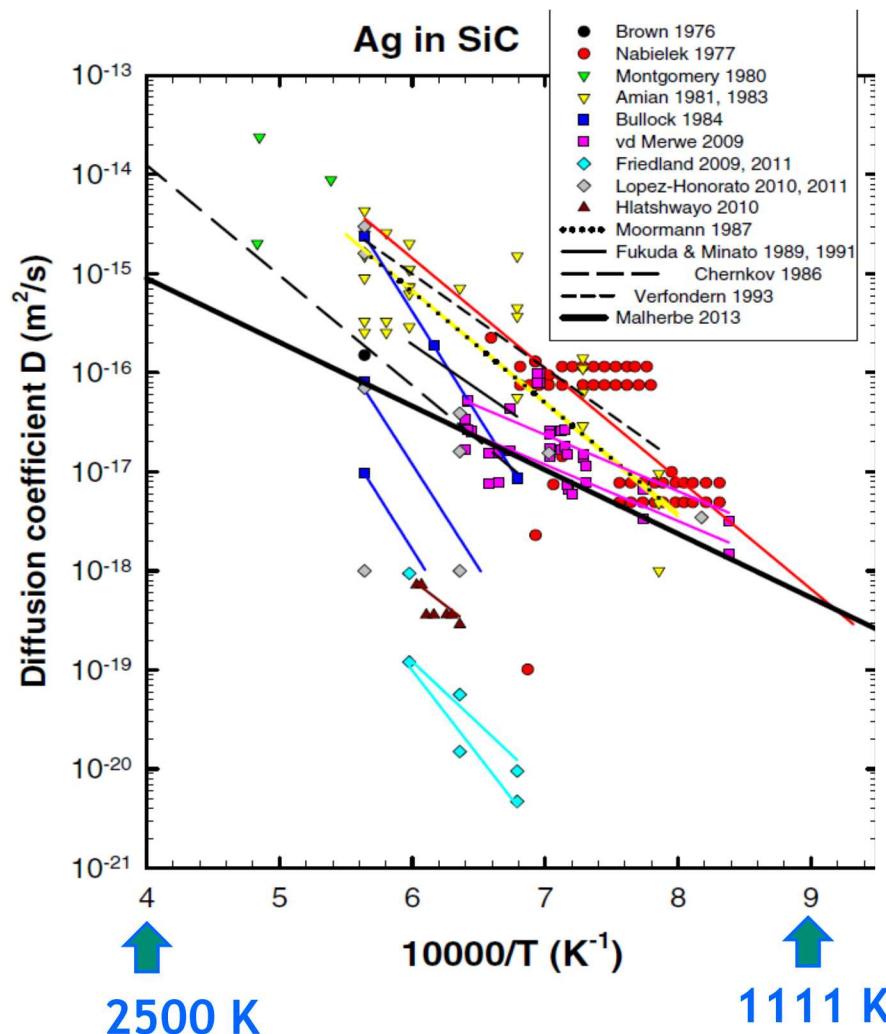
- Evaluate range of diffusivities through SiC, while layer corrodes
  - Compare releases for 25 °C and 90 °C corrosion rates.

# Simultaneous radionuclide diffusion through, and corrosion of, the SiC layer



- If diffusivity through SiC is  $10^{-18} \text{ m}^2/\text{s}$  => diffusive release dominates
- If diffusivity through SiC is  $10^{-21} \text{ m}^2/\text{s}$  => diffusive release contributes only for lower T's
  - This is  $10^{-8}$  times than that for the carbon/graphite materials ( $\sim 10^{-13} \text{ m}^2/\text{s}$ )

# Diffusivity data analyzed by Malherbe (2013) are at higher temperatures than disposal conditions



- Extrapolate to  $T$  where  $D = > 10^{-21} \text{ m}^2/\text{s}$
- Diffusivities for both Ag and Cs hit this at  $\sim 750 \text{ K}$  ( $\sim 480^\circ\text{C}$ )

# Summary and Conclusions



- SiC Layer Lifetimes
  - About 100,000 to 7,000 years from SiC average corrosion rates
    - Fluid composition and thermal history
    - If OPyC is protective, may be longer ( $\sim 10^6$  yr; van den Akker and Ahn, 2013)
- Simultaneous SiC Corrosion and Radionuclide Diffusion
  - Simplified moving boundary problem with diffusion
  - If diffusivity through SiC is  $10^{-21}$  m<sup>2</sup>/s
    - Diffusive release contributes to release only for lower temperatures
    - High T data for Ag and Cs suggest values would be well below this value
    - Diffusion in graphite/pyrolytic carbon may be much higher
- Next Steps
  - Develop model further, coupling stochastics
    - Added layers, including diffusion through compact graphite matrix
    - Evolving temperatures for repository
  - Assess major uncertainties
    - Diffusivities, especially in SiC
    - Mechanical behavior/evolution of graphite matrix (elements and compacts)

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