



2020 DOE Hydrogen and Fuel Cells Program Review presentation

Dispenser Reliability: Materials R&D

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May 23, 2020

Project ID # in013

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T I M E L I N E

- **Start date: 10/1/2016**
- **End date: 09/30/2020***

* Project continuation and direction determined annually by DOE

B A R R I E R S

Multiyear RD&D Barriers ***Technology Validation Barriers***

- **D. Lack of Hydrogen Refueling Infrastructure Performance and Availability Data**
- **E. Codes and Standards** - Validation projects will be closely coordinated with Safety, Codes and Standards

B U D G E T

Project funding:

- **NREL**
 - FY19: \$266k (carryover)
 - Project Total: \$1,740k
- **SNL**
 - FY19: \$590k
 - Project Total: \$677k

P A R T N E R S

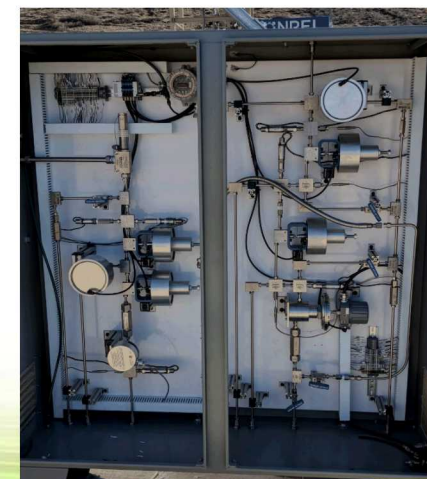
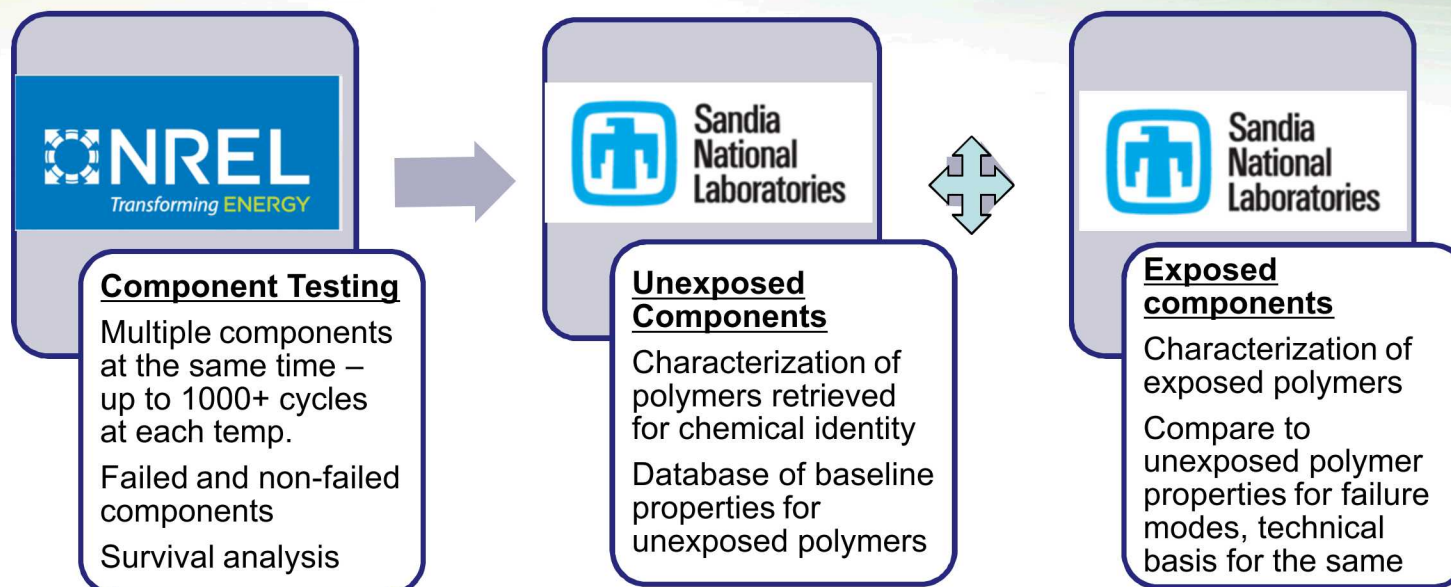
Funded

NREL: Hardware testing and lifetime analysis
SNL: Material Characterization

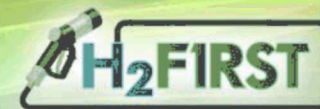
Close Collaboration

Walther-Präzision GmbH & Weh GmbH:
Material consulting and lifetime monitoring

Materials compatibility testing steps

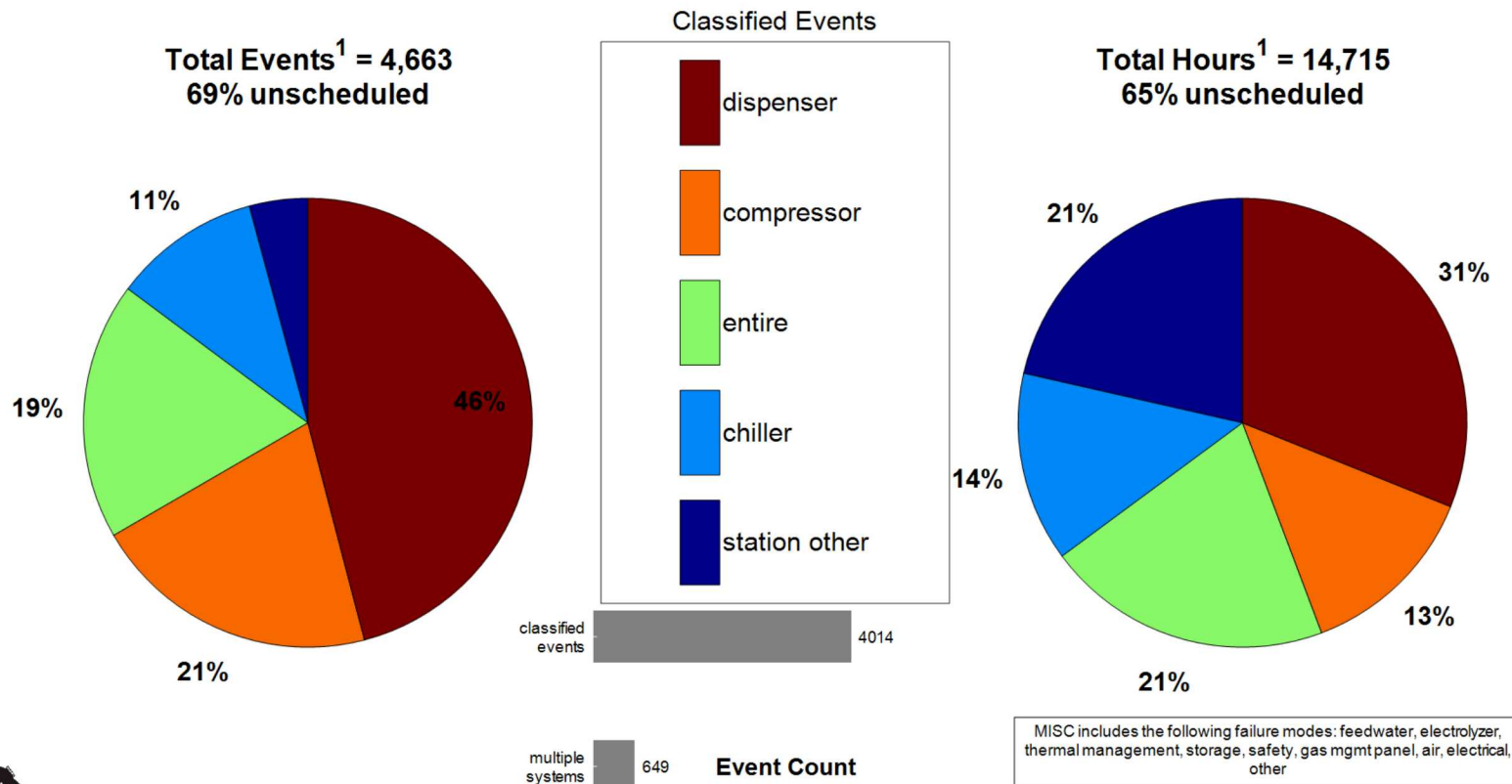


Relevance: Dispensers are #1 in Downtime



Dispensers are the top cause of maintenance events and downtime at retail hydrogen stations

Maintenance by Equipment Type - Retail Stations



NREL cdpRETAIL_infr_21

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Hydrogen Fueling Infrastructure Research Station Technology

Objective: Assess reliability and prediction of lifetimes of fueling and dispensing components exposed to pre-cooled hydrogen at high pressures based on component testing and material analyses

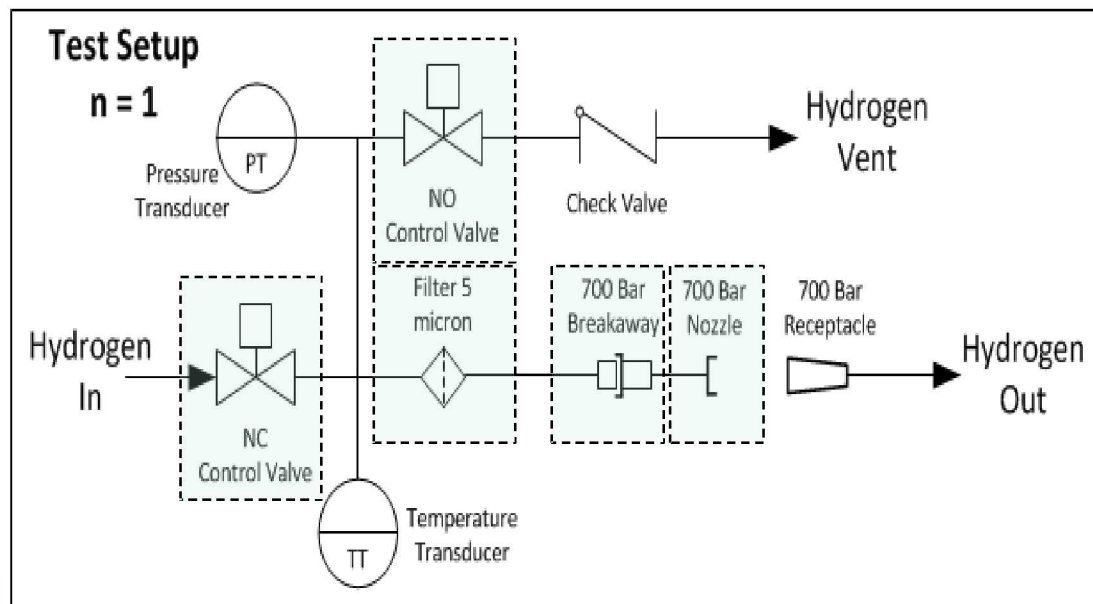
Multi-year Barriers RD and D	SNL Impact
<p>D. Lack of Hydrogen Refueling Infrastructure Performance and Availability Data</p> <ul style="list-style-type: none">• Only qualitative information available from earlier test campaigns on performance of piping components at fueling dispenser conditions• Results of specific qualification tests do not assess effects other than pass/fail	<ul style="list-style-type: none">• Material compatibility analyses pre-and post-exposure to hydrogen under dispenser operating conditions• Chemical and physical characterization of polymeric O-rings from failed and non-failed components to explore hydrogen effects• Dispenser reliability and lifetime prediction based on failure modes and degradation analyses

Approach: Accelerated Reliability Testing at NREL

Measure the mean fills between failures (MFBF) and mean kilograms between failures (MKBF) of hydrogen components subjected to pressures, ramp rates, and flow rates similar to light duty fuel cell electric vehicle fueling at -40°C, -20°C, and 0°C*

Devices Under Test (DUTs):

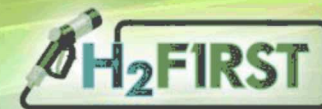
Nozzles, breakaways, normally closed valve, normally open valve, filter



Material analysis of polymeric O-rings from actual tested components to establish causes of failure modes and predict component lifetimes and dispenser reliability

* = Due to fueling practicality, budget, and schedule the team limited the testing to two temperature levels: -40°C and -20°C

Approach: Test factors and response variables for NREL testing



Fixed Factors

Controlled

H₂ pressure ramp rate
(> 17.6 MPa/min)

H₂ flow rate (0.8 kg/min)

H₂ pressure range
(14.7-77.9 MPa)

Variable Factors

Controlled

H₂ temperature -40°C, -20°C

Component types
Nozzles
Breakaways
NO valves
NC valves
Filters

Uncontrolled

Ambient temperature 10°C - 40°C

Ambient humidity 0 - 100%

Response Variables

H₂ leak (qualitative) Yes or No

Fills before failure (quantitative) Number

Amount of H₂ through component before failure (quantitative) Kilograms

Some unknowns

- Proprietary nature of polymeric materials per component type, different for different components – proper chemical identification of polymer needed
- For a given component design, identification of polymers' locations exposed to hydrogen – schematic of the component proprietary

Approach: Polymer characterization on NREL tested components at SNL



- SNL's primary role –Support of NREL's ALTA testing of components for materials level testing of unexposed and failed components for failure mode and degradation analyses
- Total of 69 tested components received: 11 unexposed for baseline , 32 (-40°C) and 26 (-20°C)
- Temperatures and number of cycles of test: -40°C, -20°C; 1000 + plus cycles for each
- Goal: To identify polymer chemistries vulnerable to cold H₂ cycling, failure modes for polymers common across all component types, manufacturers – degradation analyses
- Motivation: Elastomers and thermoplastics used as O-rings in H₂ service components – leaks possible due to:
 - Compression set at low temperatures leading to leaks as the temperature increases
 - Extreme temperature conditions - cold, brittle catastrophic failure
 - Combined influence of temperature, pressure, cycling times



Approach: Disassembly of components at SNL



Same sequence of steps used to process both exposed and unexposed components

1. Pictures taken of whole components received from NREL with NREL designation clearly depicted
2. Component disassembled carefully with special tools so as to not alter polymer physical form
3. Polymer O-rings retrieved bagged individually and assigned special combination of letter and number to indicate component source, entered into database
4. Polymer pictures taken and stored along with whole component pictures
5. Specimens subjected to non-destructive testing first, followed by destructive testing

Approach: Post-H₂ characterization methods for polymer degradation assessment



- **Microscopy (Non-destructive)**
 - Optical (Keyence) – blisters, external cracks, surface roughness/texturing, damage in the form of bubbles and/or tears or shredding
 - Assessment for permanent damage
- **Hardness (Non-destructive)**
 - Nano indentation for surface hardness and modulus changes due to H₂ exposure
 - Hardness changes are permanent; causes are plasticization or stress hardening of the matrix
- **Chemical characterization**
 - Fourier-transform infrared (FTIR) spectroscopy – polymer microstructure changes through functionalities identification (**Non-destructive**)
 - Dynamic mechanical and thermal analysis (DMTA) – T_g (glass transition temperature) changes and modulus changes (**Non-destructive**)

Multi-technique characterization is necessary for identification

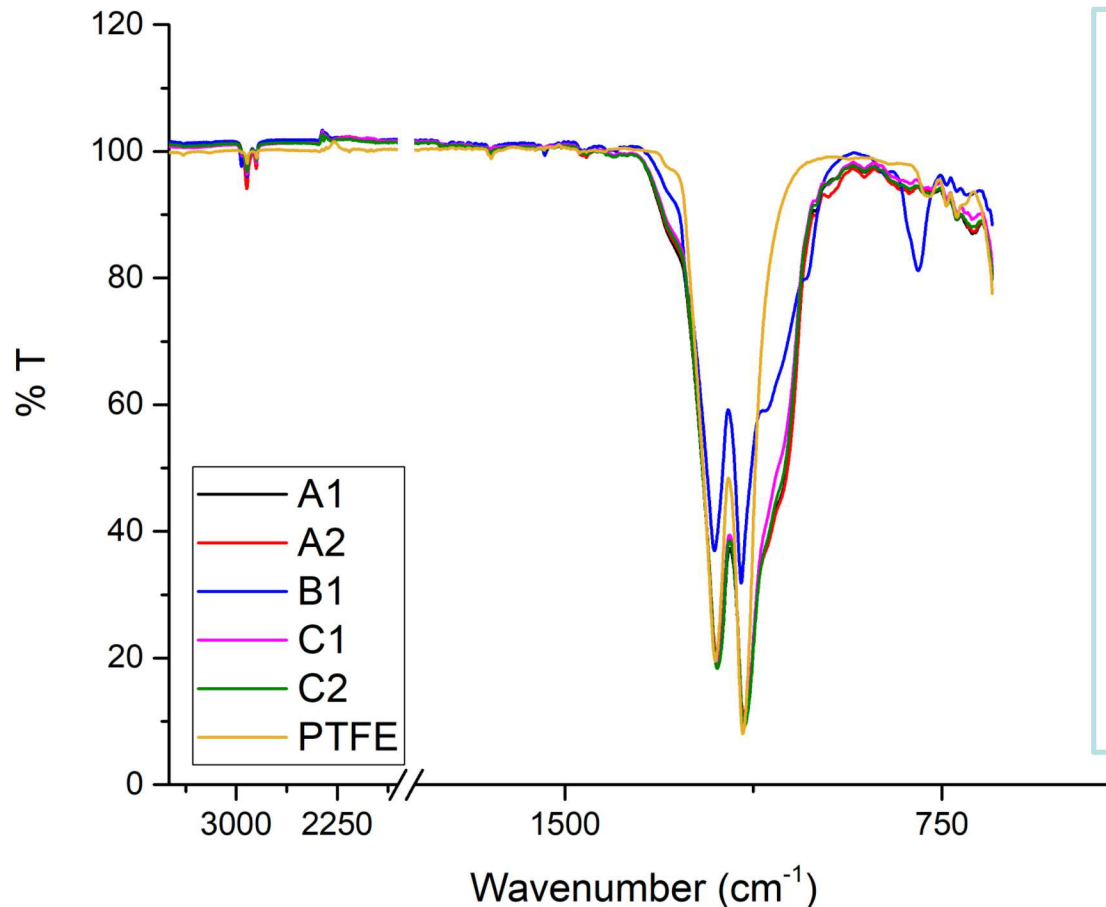
For e.g. Collection of FTIR-ATR spectra for elastomers is challenging but combined with T_g/modulus data from DMTA, confirmation of chemical changes possible

Keyence images of typical damage seen in polymer O rings after H₂ gas exposure



Approach:

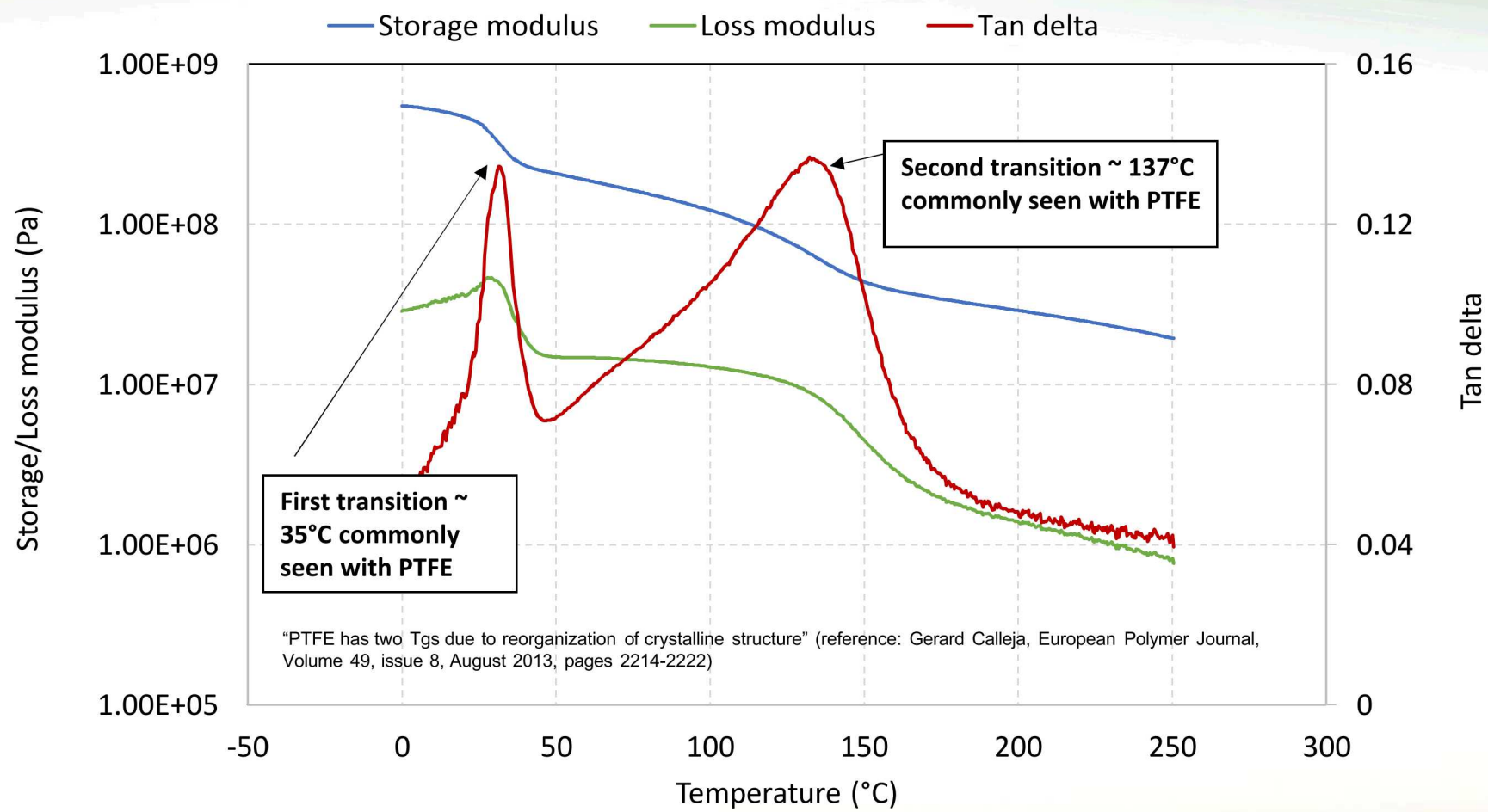
IR spectroscopy to detect O-ring chemistry changes



- Material characterization with Infra-red spectroscopy ATR (attenuated total reflectance)
- Non-destructive, surface-focused technique
- Precise identification possible in unique fingerprint region (1500 to 500 cm⁻¹)
- Assess complex bending, rotational and vibrational modes of molecular motion in response to IR, unique to each material type
- Elastomeric materials give weak signals, still discernable

For e.g.: in the figure shown, O-ring C2 (green) compares well to a standard spectrum of PTFE (orange)

H2 FIRST, PTFE thin gasket DMTA, 0.5% strain, 1 Hz, 5°C/min heating

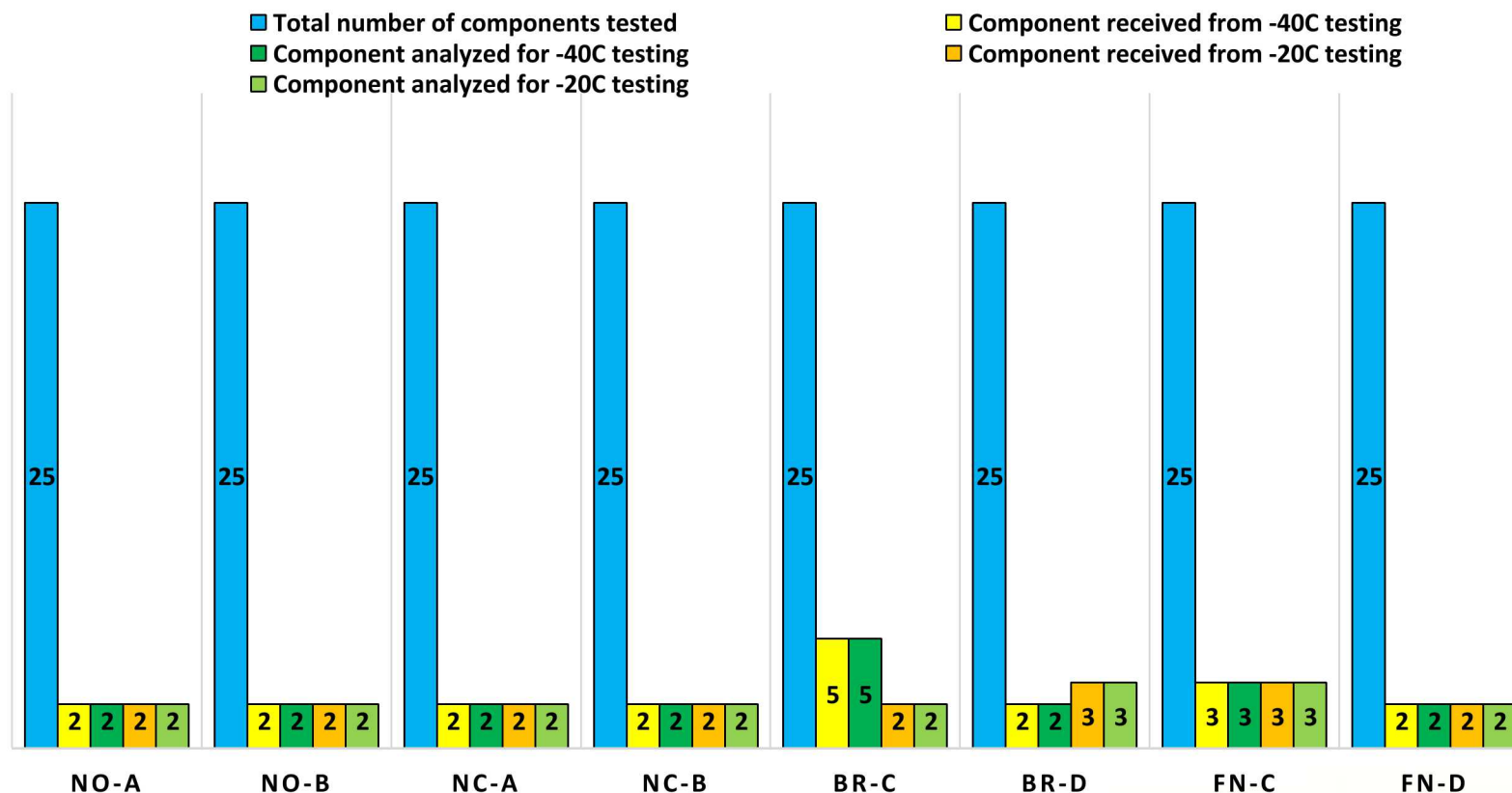


Glass transition temperature and modulus shifts reflect polymer microstructural changes in response to exposure to H₂

Accomplishment: Analyses of tested components



NREL-TESTED VS. SNL-ANALYZED COMPONENTS FOR EACH TEST TEMPERATURE



NO = normally open, NC = normally closed, BR = breakaways FN = fueling nozzles
A, B, C and D stand for protected manufacturer IDs

Accomplishment:

Polymers identified in tested components

Component*	NO Valves		NC Valves		Breakaways		Fueling Nozzles	
Polymer (numbered series)	NO-A (1-25)	NO-B (25-50)	NC-A (1-25)	NC-B (25-50)	BR-C (1-25)	BR-D (25-50)	FN-C (1-25)	FN-D (25-50)
Components analyzed/tested	8/25	8/25	8/25	8/25	14/25	10/25	12/25	8/25
PTFE								
NBR								
PEEK								
FKM								
Polyurethane PURs								
Butyl Rubber								
Neoprene								
POM								
HNBR								

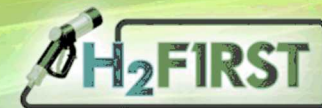
*= manufacturer
ID protected

Polymer susceptibility to damage in testing depends on

- Chemistry of polymer selections and changes in microstructure due to H₂ exposure
- Exposure and cycling at low temperatures (-40°C vs -20°C) and related thermal shock
- Number of cycles (1000 vs 500) and corresponding mechanical physical stress
- Design related factors - location within component dictates
 - Whether polymer exposed to H₂ when in service
 - How metal fixturing makes seal with polymer, moves relative to polymer, etc.

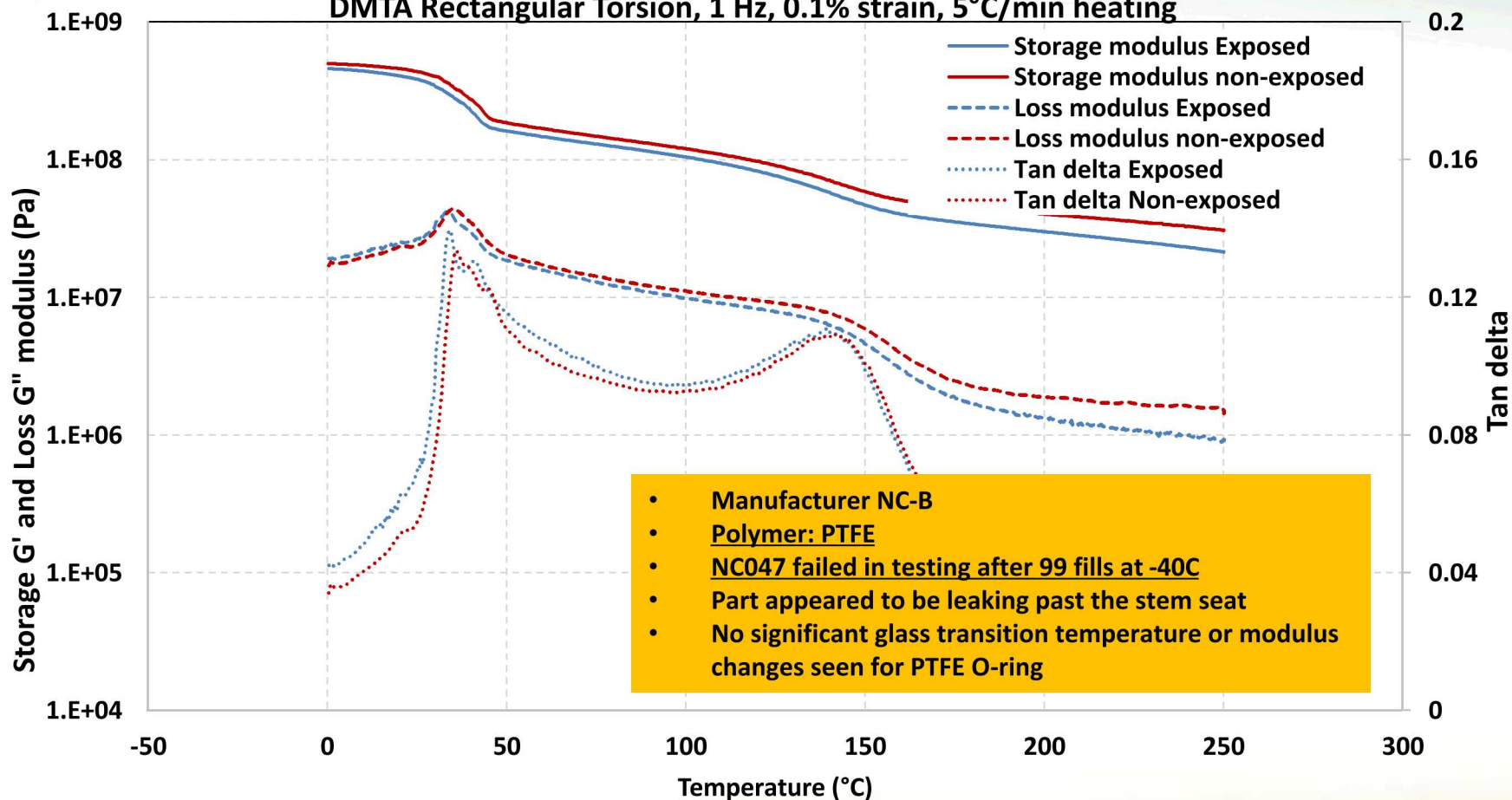
Accomplishment:

DMTA demonstrates that polymer chemical properties changes are not responsible for early failure of PTFE in normally closed valves NC-B valves



H2FIRST NC47-B PTFE gasket, compared to non-exposed NC26-B PTFE gasket

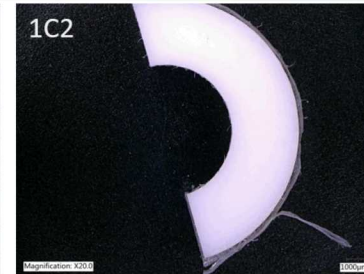
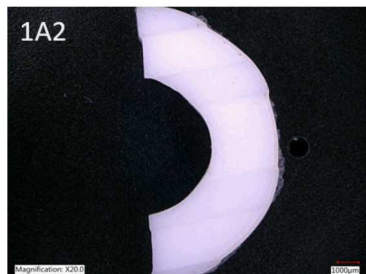
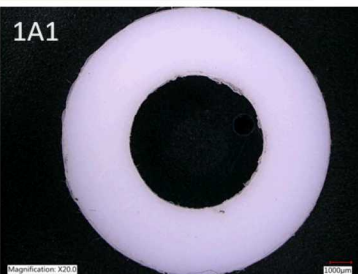
DMTA Rectangular Torsion, 1 Hz, 0.1% strain, 5°C/min heating



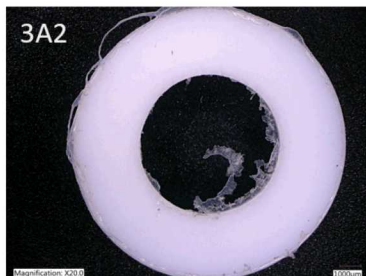
- Insignificant change in PTFE glass transition temperature and modulus, therefore not directly responsible for NC valve failure

Accomplishment:

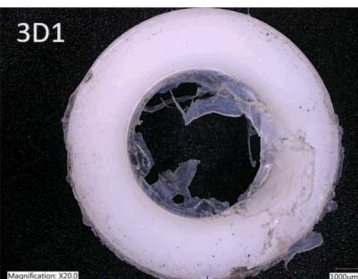
Optical microscopy demonstrates change (degradation) in physical properties of failed normally closed valve PTFE O-rings



NC026-B:
non H₂ exposed



NC049-B:
H₂ exposed,
failed at 99
cycles, -40°C



NC047-B:
H₂ exposed,
failed at 99
cycles, -40°C



Metallic particles and
other debris embedded
in failed PTFE O-rings

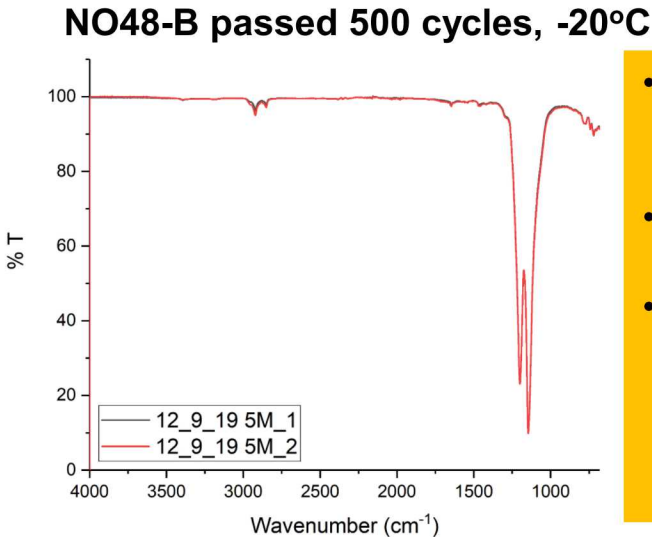
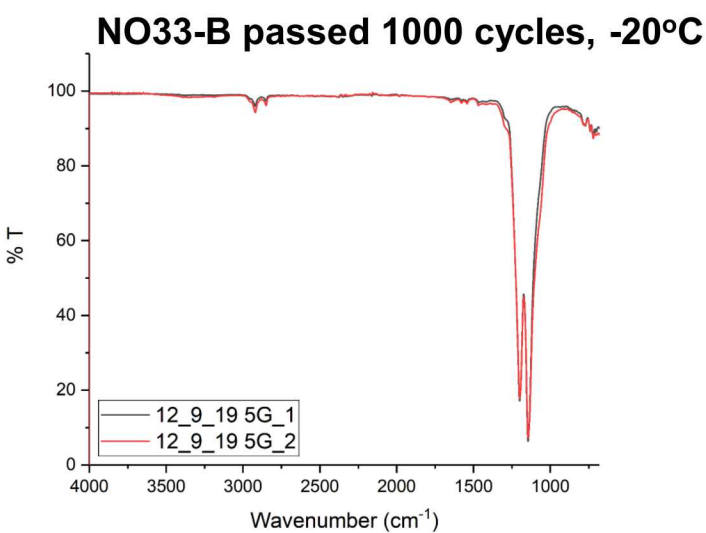
Significant increase in fraying and deposit of metallic and non-metallic debris with H₂ exposure

Accomplishment:

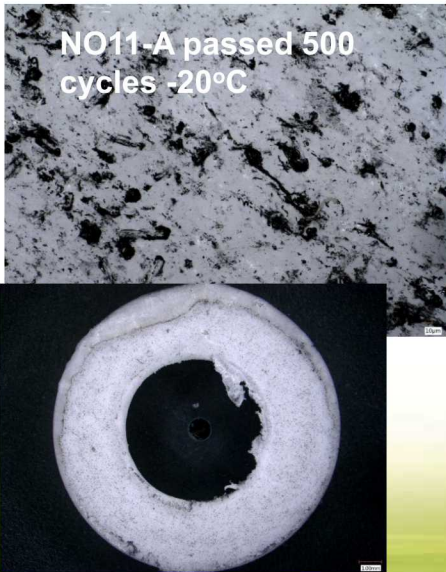
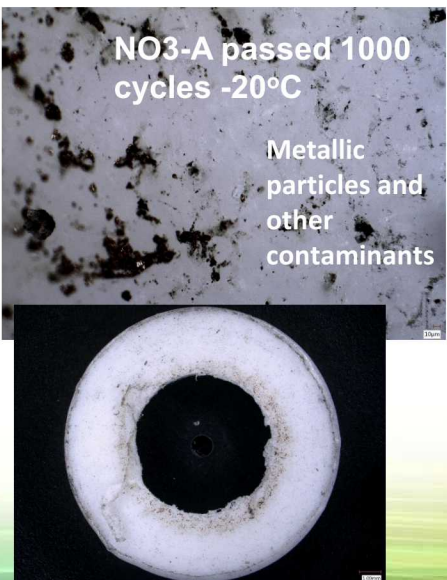
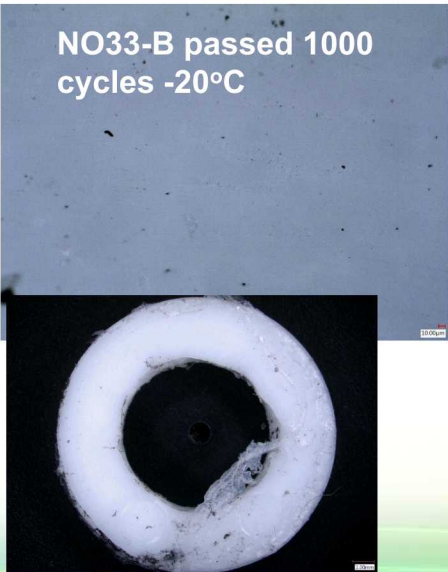
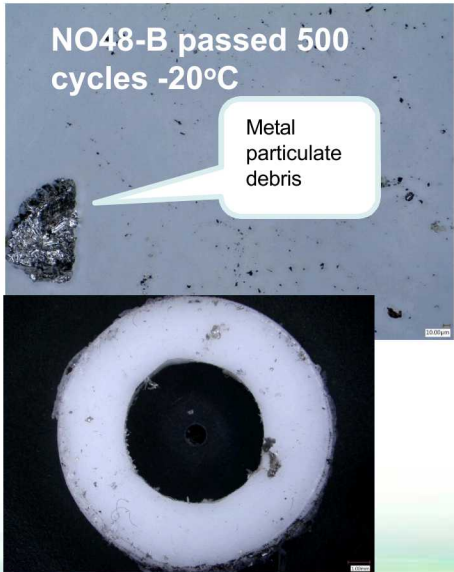
ATR-FTIR demonstrates no polymer chemical structural changes for PTFE O-rings in normally open valves NO-A after exposure to H₂



Optical microscopy evidence of damage



- PTFE O-rings in NO-A and NO-B show different types of damage
- The NO-B rings are cleaner but show greater fraying
- The NO-A PTFE show fraying and embedded metal particles and discolorations for the same number of cycles

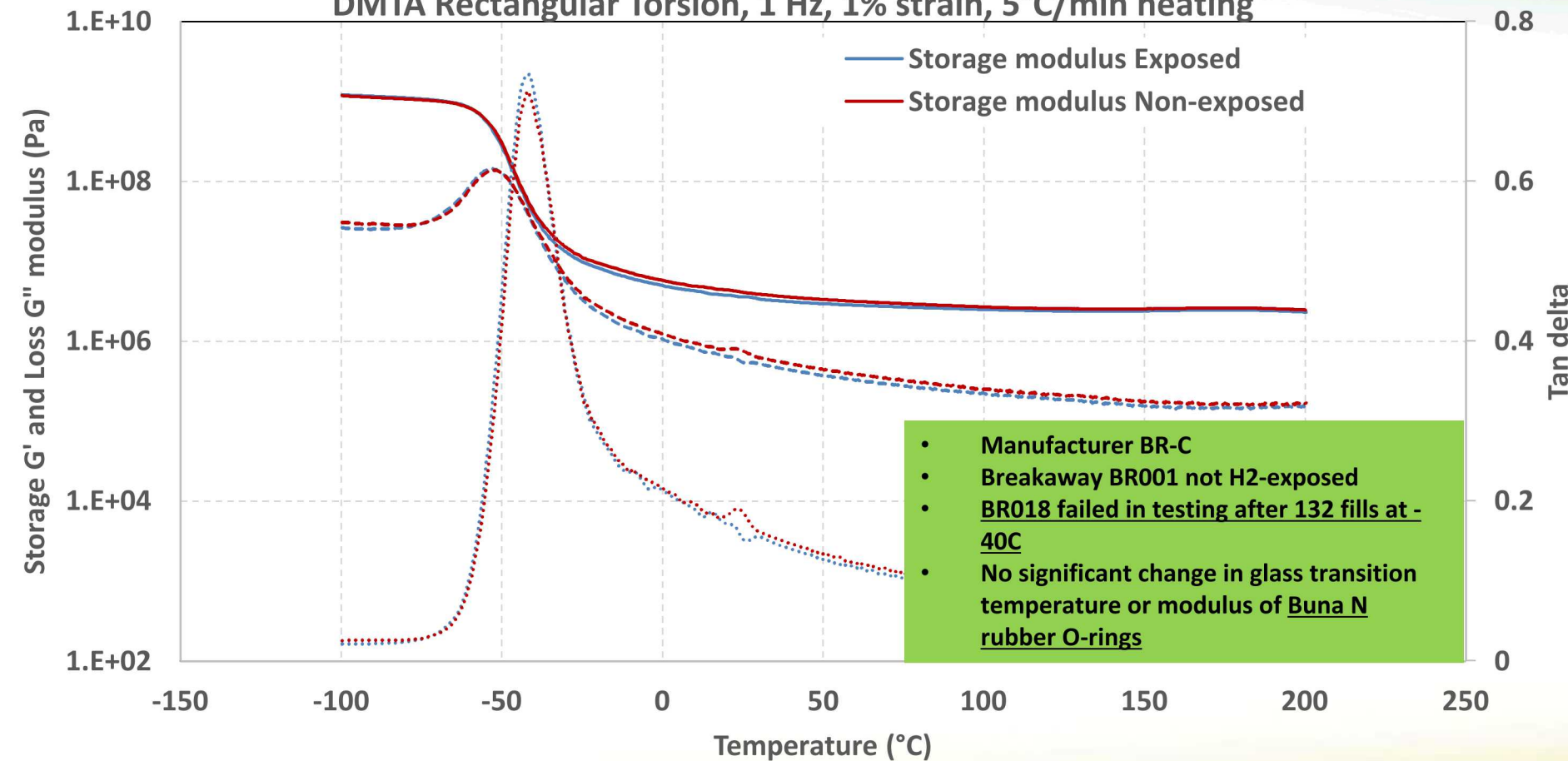


Accomplishment:

DMTA demonstrates that polymer chemical properties changes are not responsible for early failure of buna-n found in breakaway BR-C:



H2FIRST BR18-C Buna N O-ring, compared to similar Buna N O-ring BR01-C
DMTA Rectangular Torsion, 1 Hz, 1% strain, 5°C/min heating



• Change in Buna N thermomechanical properties not reason for breakaway failure

Accomplishment:

ATR-FTIR demonstrates no polymer chemical structural changes for Buna-N O-rings in breakaway BR-C after exposure to H₂



**Unexposed
BR001-C**



**BR018-C failed at
132 cycles, -40°C**



**BR023-C failed at
356 cycles, -40°C**



- ATR-FTIR spectra for Buna O rings from these failed breakaways do not show significant chemical differences
- No physical damage seen with any of the Buna N O-rings in the optical images

Accomplishment:

Nano indentation shows PTFE physical property (hardness, modulus) changes
Optical microscopy shows degradation of multiple polymers in breakaways



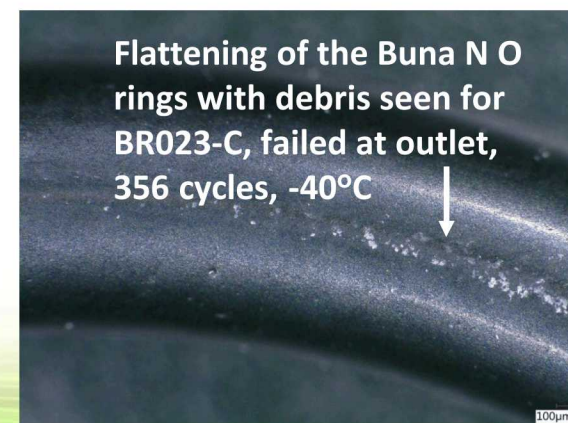
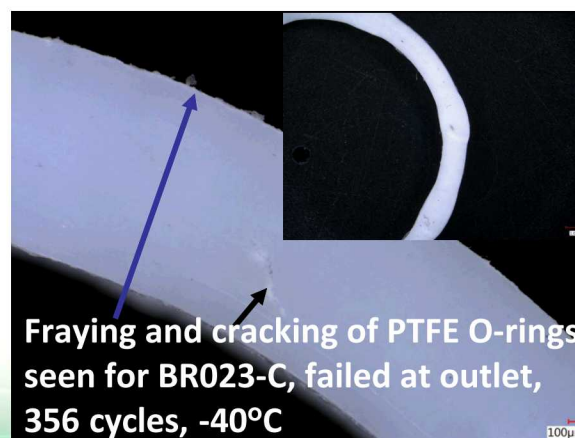
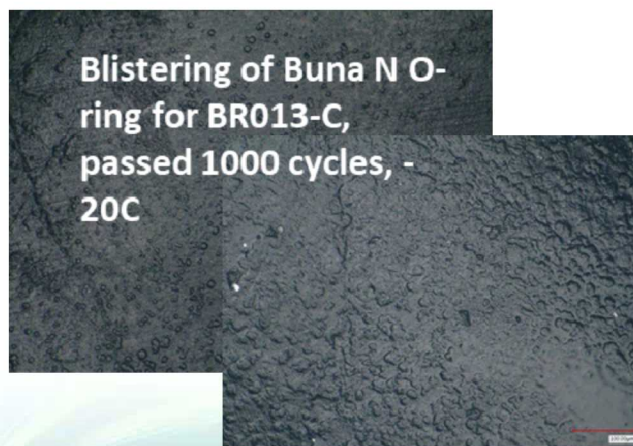
Nanoindentation of exposed and control PTFE O-rings from multiple components

Polymer physical property changes indicate either

- plasticization (lowering of hardness modulus)
- stress hardening (increase in hardness) by H₂

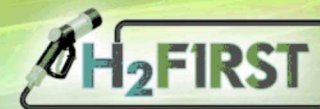
Typical degradation effects seen for O-rings in this study

Breakaway BR-C	Hardness (GPa)	Modulus (GPa)	Hydrogen exposed?	Number of cycles @ failure, test temperature	Failure modes
BR001-C PTFE	0.11 ± 0.02	1.68 ± 0.24	No	NA	NA
BR018-C PTFE	0.07 ± 0.02	1.26 ± 0.18	Yes	132 cycles, -40°C	Leak at outlet of breakaway coupling
Normally closed valve NC-B	Hardness (GPa)	Modulus (GPa)	Hydrogen exposed?	Number of cycles @ failure, test temperature	Failure modes
NC026-B	0.064 ± 0.02	1.58 ± 0.35	No	NA	NA
NC049	0.068 ± 0.01	1.53 ± 0.19	Yes	99 cycles, -40°C	Leak at packing gland and stem interface
NC047	0.065 ± 0.02	1.06 ± 0.26	Yes	99 cycles, -40°C	Leaking past stem seat



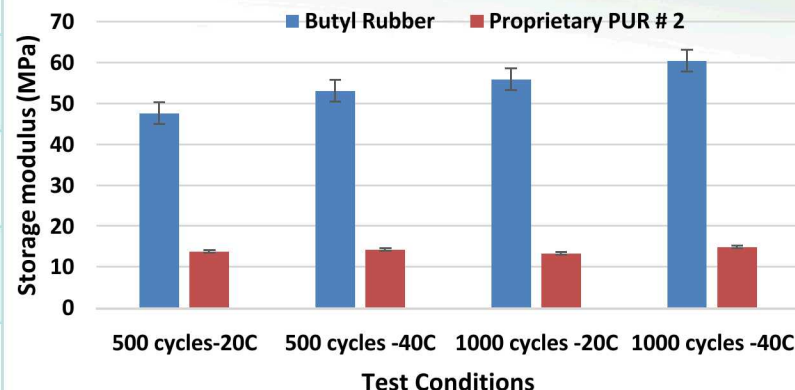
Accomplishment:

DMTA demonstrates chemical changes to butyl rubber in breakaway with trends based on number of cycles and exposure temperature

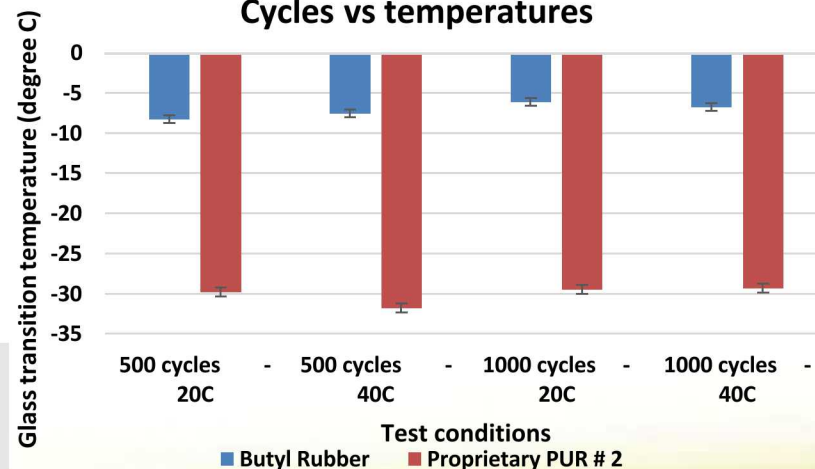


Breakaways	Polymer	Storage modulus (MPa)	Tan Delta Peak 1 (degree C)
BR031-D (1000 cycles, -40°C)	Butyl rubber	60.47 ± 13.34	-6.77 ± 0.44
	Proprietary PUR #2	14.87 ± 0.49	-29.31 ± 0.65
BR048-D (500 cycles, -40°C)	Butyl rubber	53.08 ± 1.58	-7.53 ± 0.07
	Proprietary PUR #2	14.30 ± 0.66	-31.75 ± 0.10
BR034-D (1000 cycles, -20°C)	Butyl rubber	55.9 ± 8.4	-6.1 ± 0.50
	Proprietary PUR #2	13.4 ± 0.5	-29.5 ± 0.40
BR056-D (500 cycles, -20°C)	Butyl Rubber	47.6 ± 7.5	-8.3 ± 0.4
	Proprietary PUR #2	13.80 ± 0.90	-29.80 ± 1.9

Storage modulus BR-D breakaways:
Cycles vs temperature

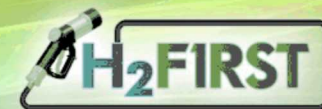


Glass transition temperature (degree C):
Cycles vs temperatures



- All the BR-D breakaways passed under test conditions shown
- Storage modulus of butyl rubber O-rings increases with more cycles and lower exposure temperatures
- Glass transition temperatures are steady under test conditions

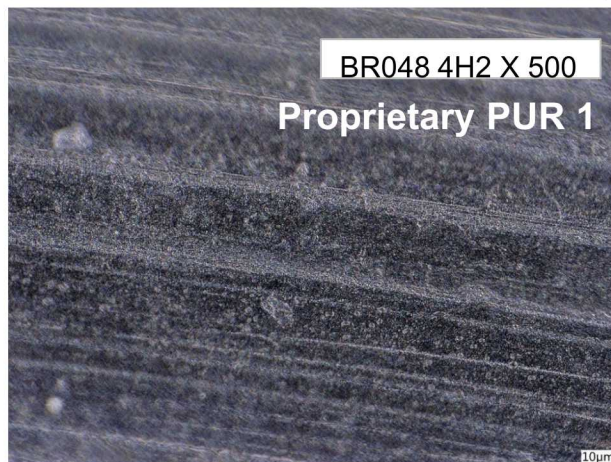
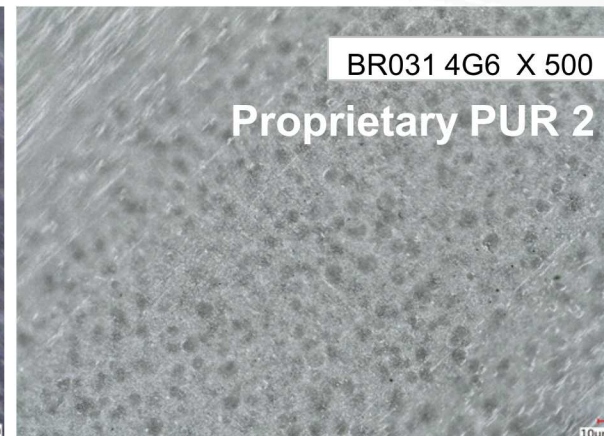
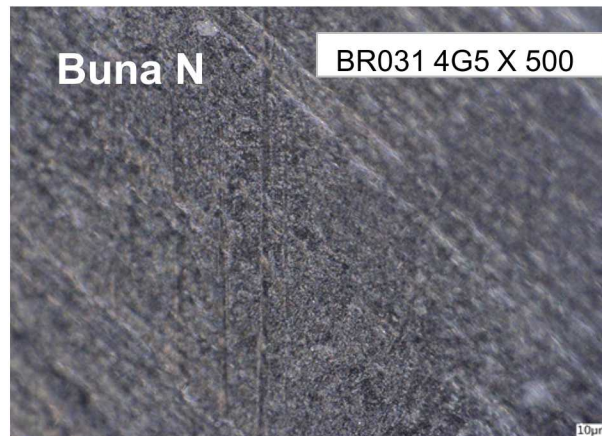
Accomplishment:



Optical microscopy shows increased blistering of Buna-N and increased wear of PUR with additional cycles

BR031-D (top row) passed 1000 cycles, -40C

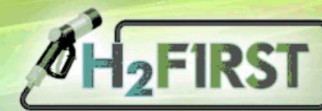
BR048-D (bottom row) passed 500 cycles, -40C



**At -40C test temperature, blisters in Buna N, PUR# 2 and wear in PUR# 1 is greater at 1000 cycles vs 500 cycles
This indicates an effect of number of cycles: greater the number of cycles, greater the damage**

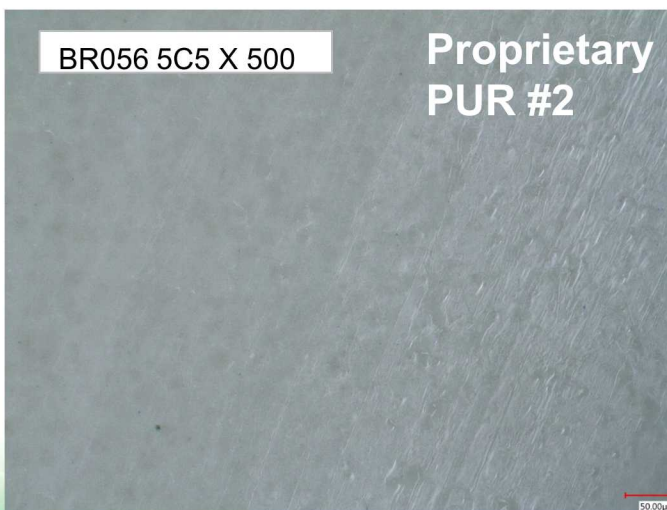
Accomplishment:

Optical microscopy shows less damage to polymers cycled at -20°C , independent of number of cycles



BR034-D (top row) passed 1000 cycles, -20°C

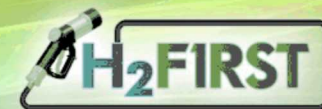
BR056-D (bottom row) passed 500 cycles, -20°C



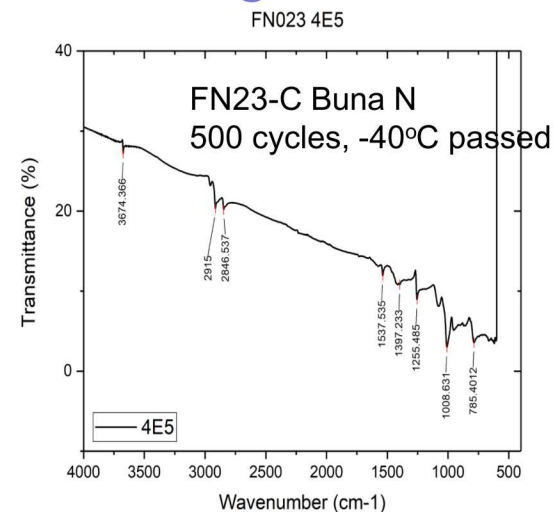
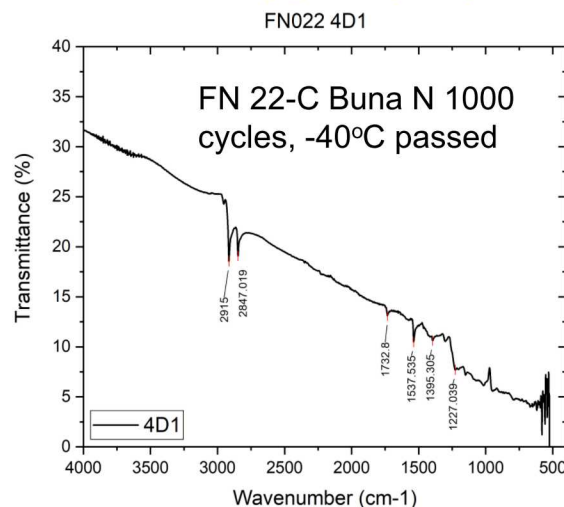
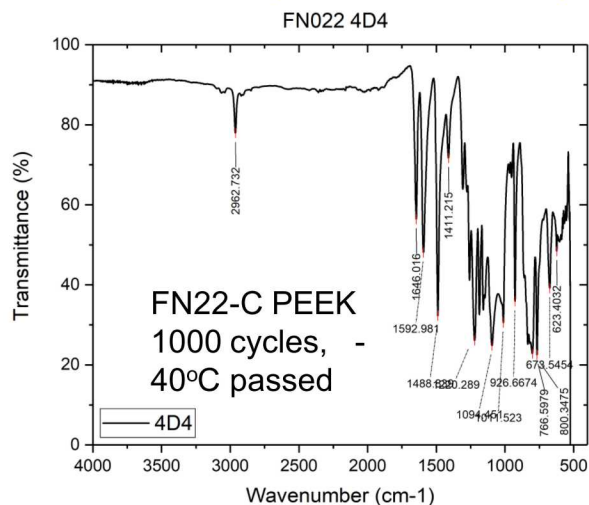
- No significant increase in damage with increased number of cycles at -20°C .
- Damage increased with cycling at -40°C
- Effect of test temperature and the number of cycles is evident for these breakaways and the polymers within

Accomplishment:

Polymer choice *critical* for performance in H₂



FTIR spectra and optical microscopy of polymers from fueling nozzle



FN22-C PEEK 1000
cycles, -40°C passed

PEEK does not show any
wear vs Buna N

100µm

FN22-C Buna N
1000 cycles, -40°C

Wear in the middle of the
Buna N O-ring

100µm

FN23-C Buna N
500 cycles -40°C

1000µm

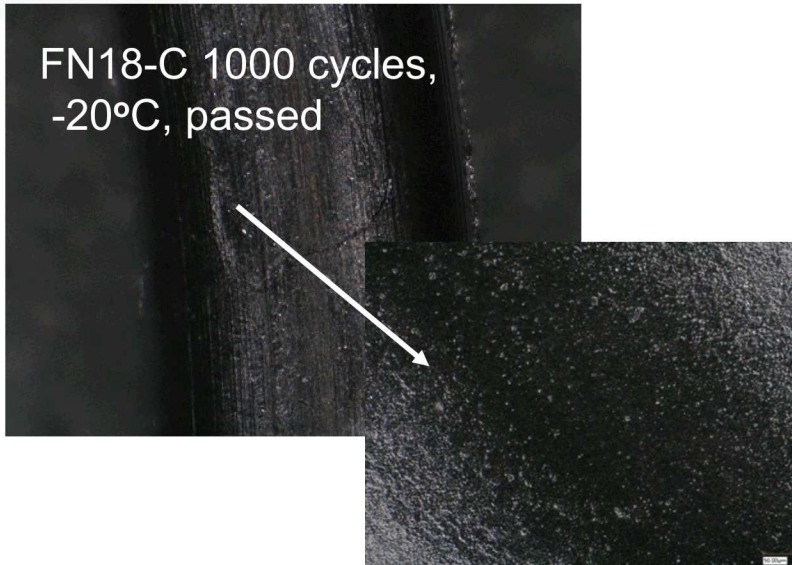
Increasing number of cycles at -40C shows increased wear in Buna N

Accomplishment:

Extent of blistering and wear, leading to failure, can vary for the same component exposed to the same conditions



FN18-C 1000 cycles,
-20°C, passed



FN19-C 5K 1000 cycles,
-20°C, failed



Leak from the nozzle
stopped the testing and
no fills were done after
this failure at 1,063
cycles

FN04-C 500 cycles, -20°C



Nature of damage here is
similar to the FN18-C

- Two fueling nozzles tested at -20°C from FN-C were compared for degradation modes (top row)
- Extent and nature of wear is different
- For FN19-C, extreme pitting and wear is seen with the Buna N rubber whereas with FN18-C had blisters
- FN19-C failed at 1000 cycles

Response to last year's Reviewer's comments



- This part of the project was not independently reviewed last year. See project in008 for reviewer responses.



SNL and NREL are using expertise at both labs with close coordination to execute this project

Component Manufacturers

- Team maintained an open communication with the breakaway/nozzle manufacturers as results became available
- NREL and SNL secured NDAs with both manufacturers and allowed them to review failure results + material analysis



Remaining Challenges and Future Work:



Publish findings and NREL findings (see in008) in a comprehensive final report

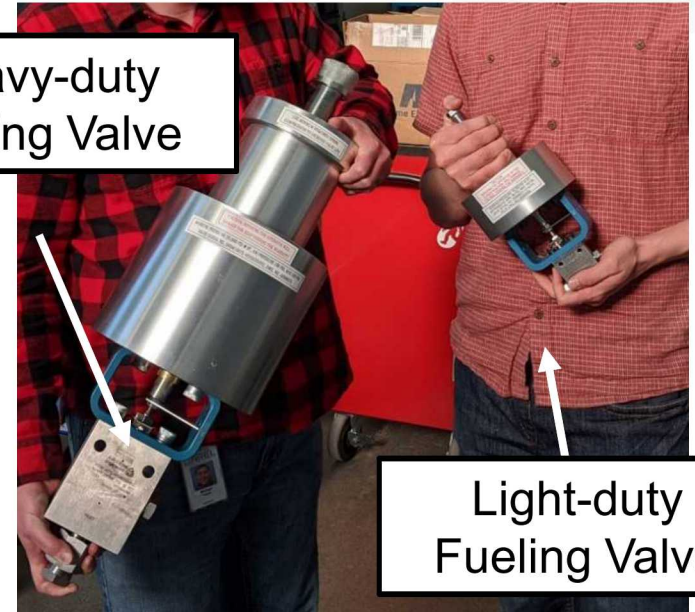
Move Towards the Future

- Develop novel polymeric materials or seal designs
- Develop additional work scope including coordination with H-MAT consortium

Coordinate additional testing at NREL and analysis at SNL with:

- Heavy-duty components (higher flows)
- Additional components and/or new component designs
- Higher cycle counts
- Increased thermal shock

Heavy-duty
Fueling Valve



Light-duty
Fueling Valve

Any proposed future work is subject to change based on funding levels.

- ❖ Right choice of polymers make all the difference in mitigating failure
 - ❖ Chemistry of polymers play a role in resistance to hydrogen cycling
 - ❖ e.g. Buna N and PTFE showed more damage compared to PUR and PEEK
- ❖ Degradation modes examined in components indicate mostly physical damage to the polymeric O-rings; no evidence of chemical changes for test conditions
 - ❖ Tested PTFE showed physical fraying and cracks whereas in Buna N, wear, blisters and pitting was observed
- ❖ Damage in fueling nozzles and breakaways increased from 500 to 1000 cycles at -40C for Buna N rubber
- ❖ Degradation or the extent of damage did not increase with cycling for -20C testing
- ❖ The nature and the extent of the degradation was much less at -20C as compared to -40 components
- ❖ More component testing with >>1000 cycles at -40C can help inform both degradation and statistical analyses in the future