

Unique Optical and Electronic Effects of RE-DOBDC MOFs with Acid Gas Adsorption

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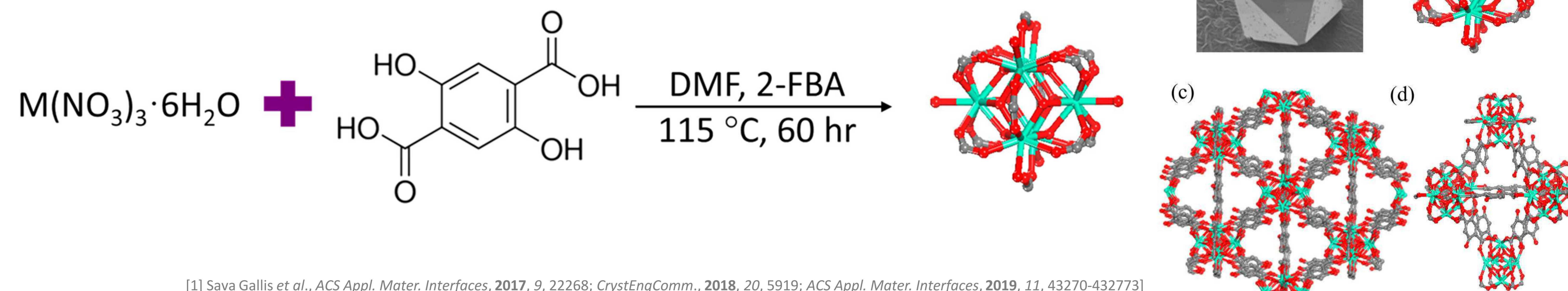
Introduction

AIM: Design and synthesize a RE-containing MOF for the selective adsorption of NO_x and SO_x from flue streams

UNCAGE-ME Center exemplar "Complex Mixtures":

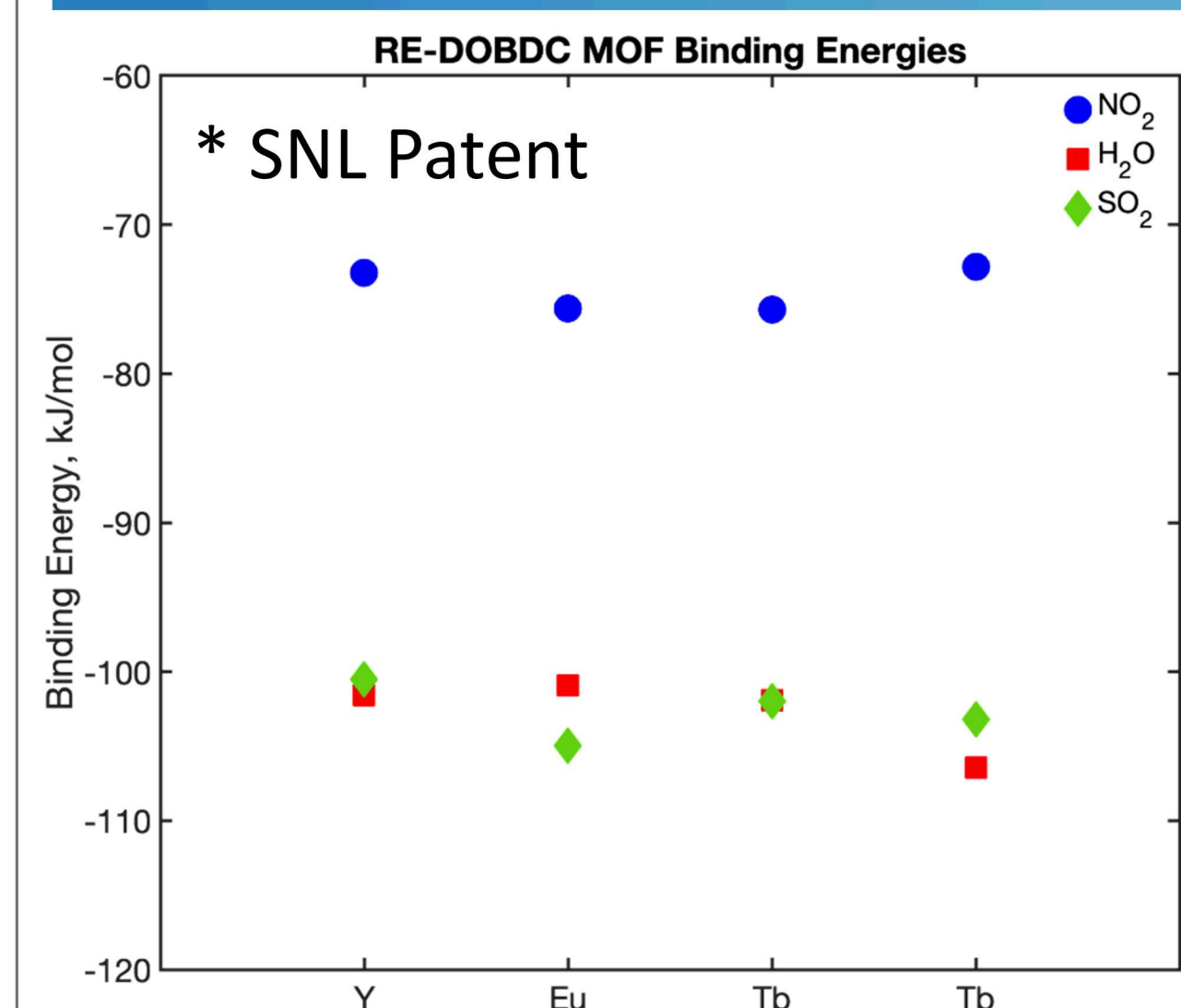
- Coal-fired power plant flue gas (13% CO₂, 6% H₂O, ~4% O₂, 50 ppm CO, 420 ppm NO_x, 420 ppm SO₂, 76% N₂)

- MOFs are inorganic-organic hybrid materials with extremely high surface area and the ability to form CUSs
- Rare earth (RE) elements selectively bind to acid gases (NO_x and SO_x)¹
- RE-DOBDC (RE – Eu, Y, Tb, Yb) MOFs selectively adsorb acid gases from a humid gas stream – In collab. with GATech
- Retention of structure by PXRD after exposure to acid gases. Investigations ongoing with neutron PDF – In collab. with ORNL
- RE-DOBDC shows a reduction in paramagnetism with adsorption of NO_x – In collab. with CINT
- Fluorescence of MOF dependent on the presence of free hydroxyls

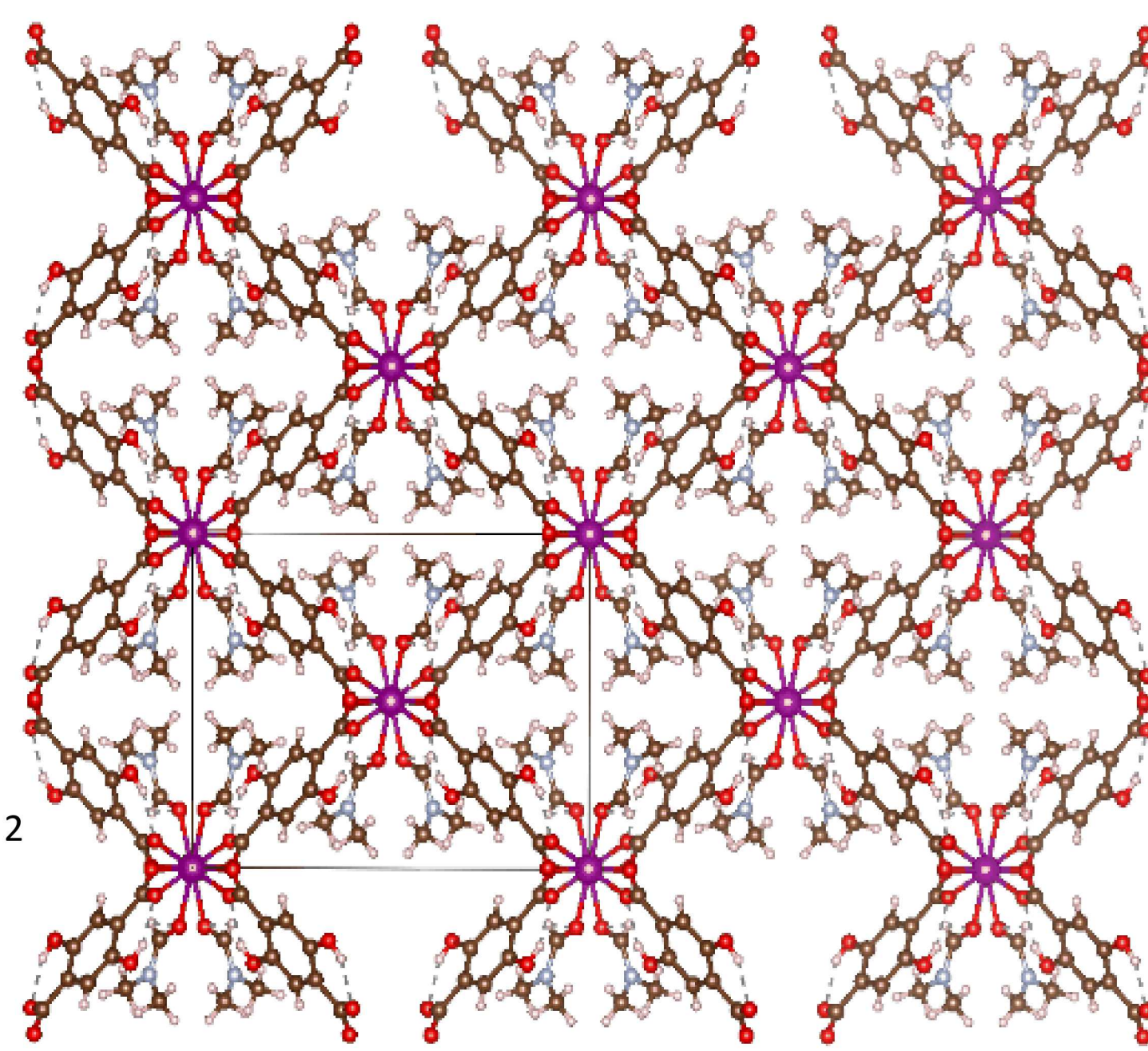


[1] Sava Gallis et al., ACS Appl. Mater. Interfaces, **2017**, 9, 22268; CrystEngComm., **2018**, 20, 5919; ACS Appl. Mater. Interfaces, **2019**, 11, 43270-432773]

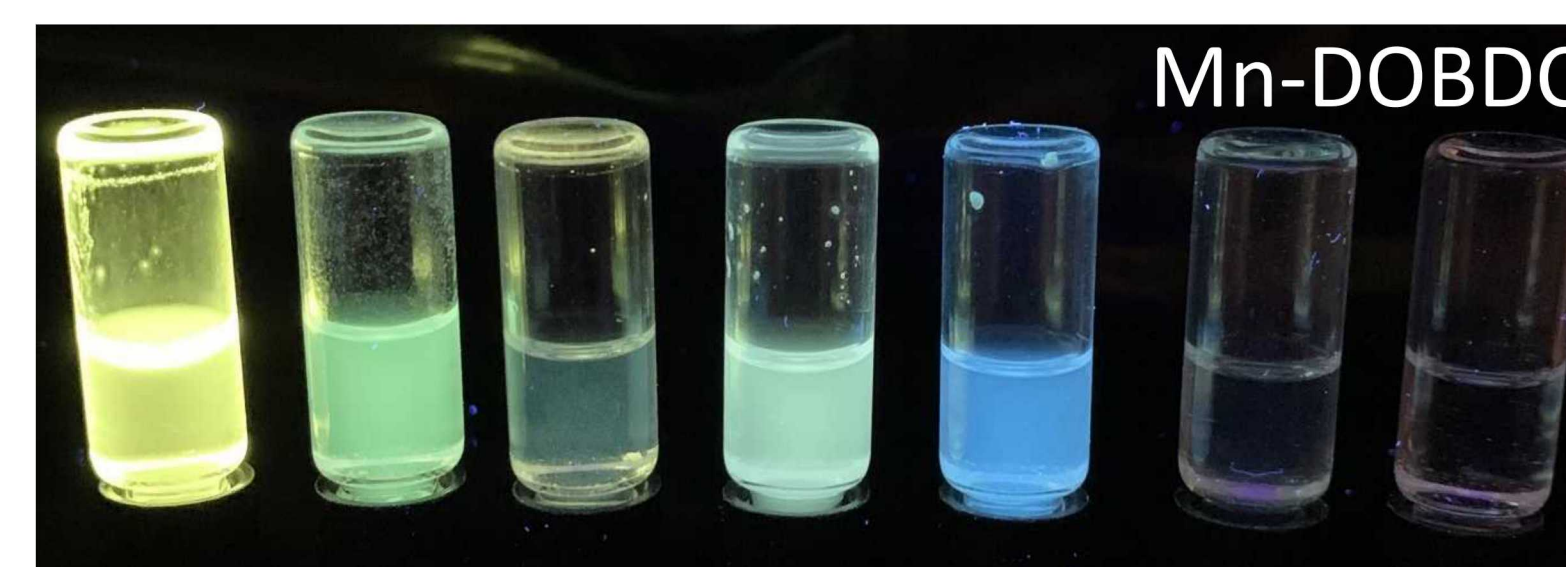
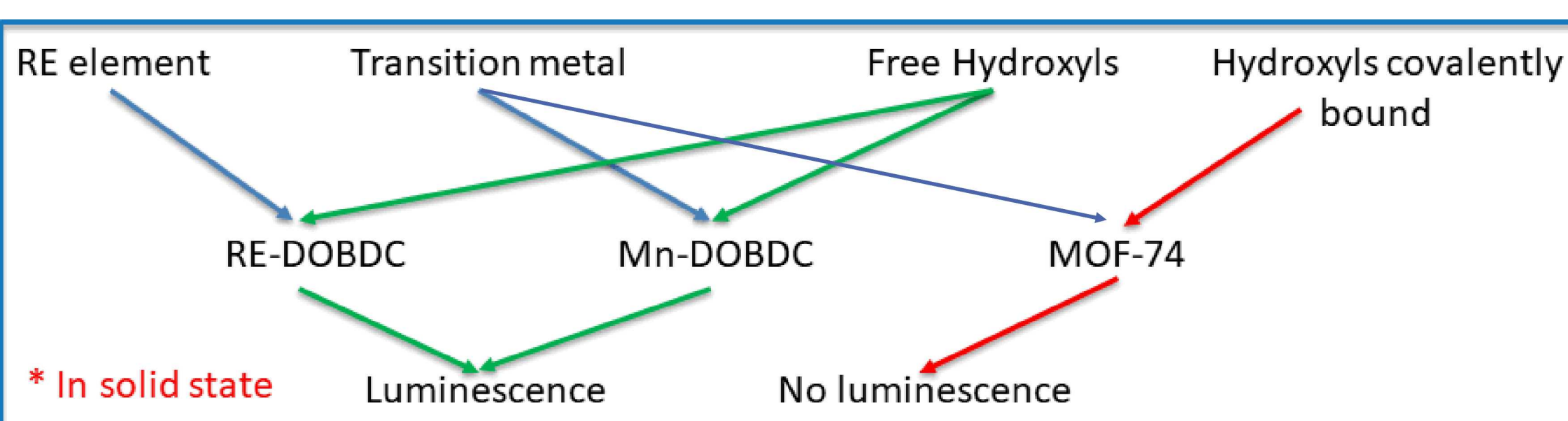
Photoluminescence



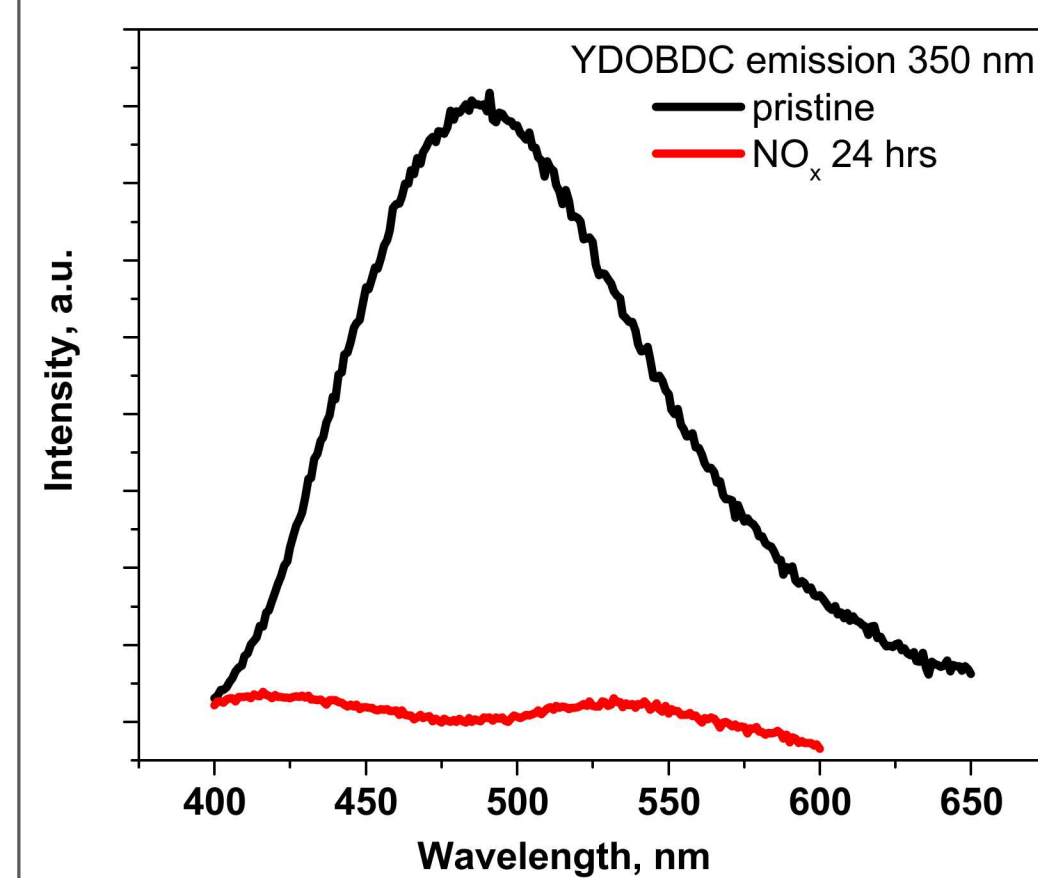
- Competitive binding to metal site between SO_x and H₂O
- NO_x: lower binding E than H₂O
- In humid NO_x, H₂O binds to the metal so NO_x only has the ability to bind to the linker
- Luminescence in MOF arises from the free linker hydroxyls
- Corroborated by comparing with MOF-74 and Mn-DOBDC²



Binding energies rationalized through DFT modelling
(*** See Jon Vogel's Poster ***)



[2] Henkelis et al., Tuned Luminescent Properties of DOBDC Containing MOFs: the Role of Free Hydroxyls, ACS Appl. Mater. & Interfaces, **2020**, Submitted

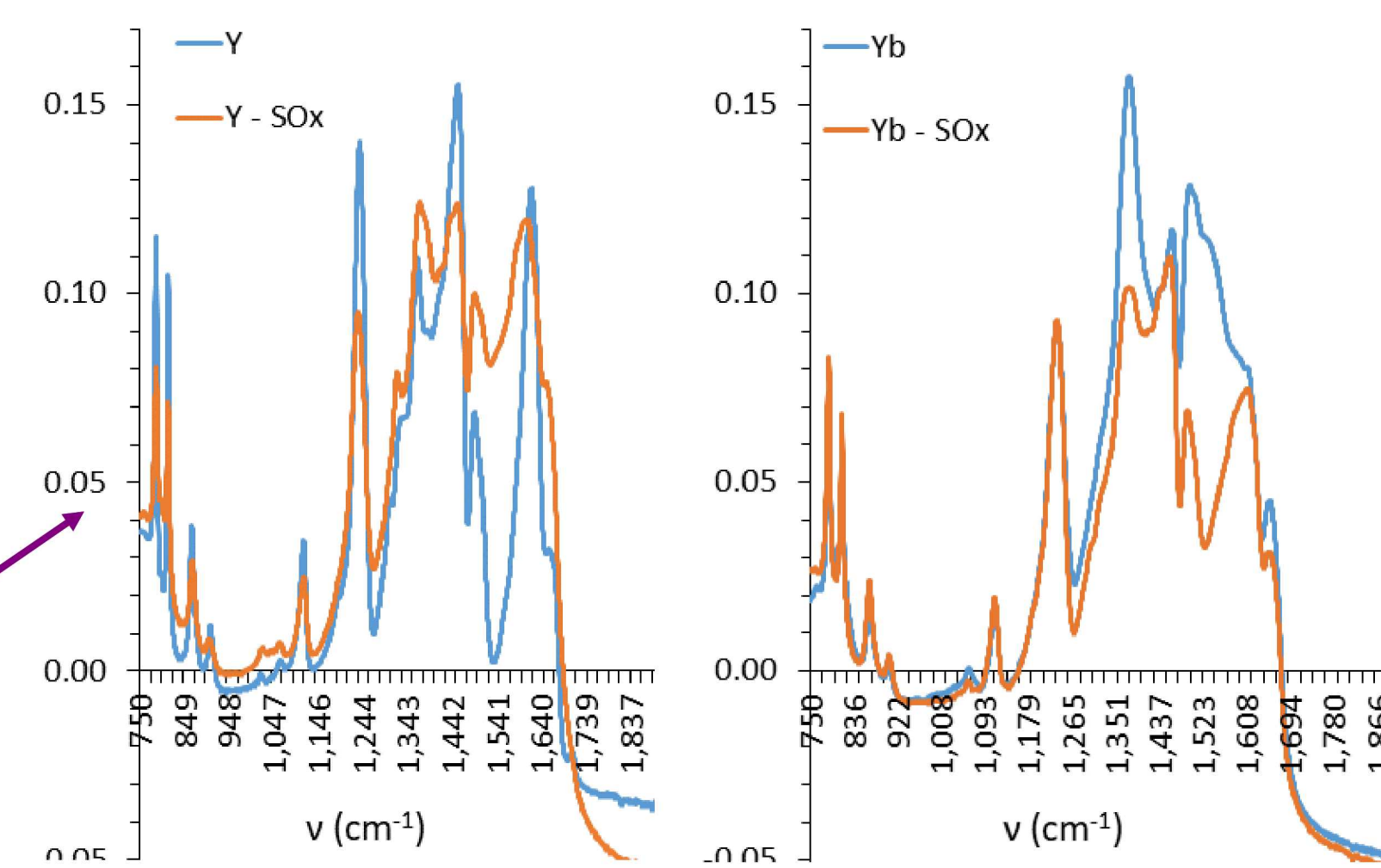


RE-DOBDC with NO_x:

- Binds to linker hydroxyls
- Quenches photoluminescence
- No change in PXRD

Y-DOBDC and Yb-DOBDC:

- Clear change in IR pre- and post-SO_x adsorption
- SO_x binding to metal center

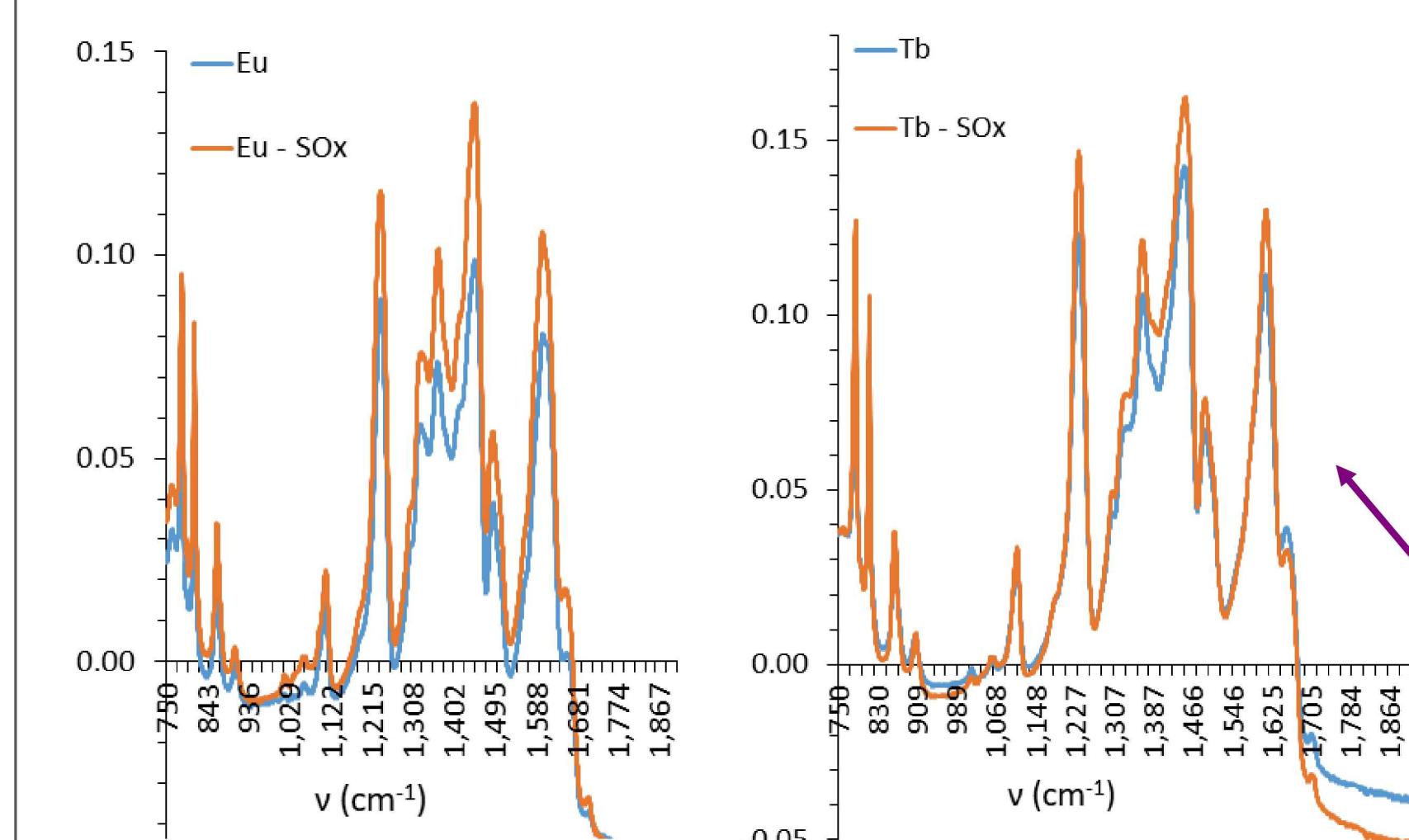


RE-DOBDC with SO_x:³

- Binds to the RE metal center
- No quenching in photoluminescence
- No change in PXRD
- Clear metal dependency: Eu and Tb vs Y and Yb

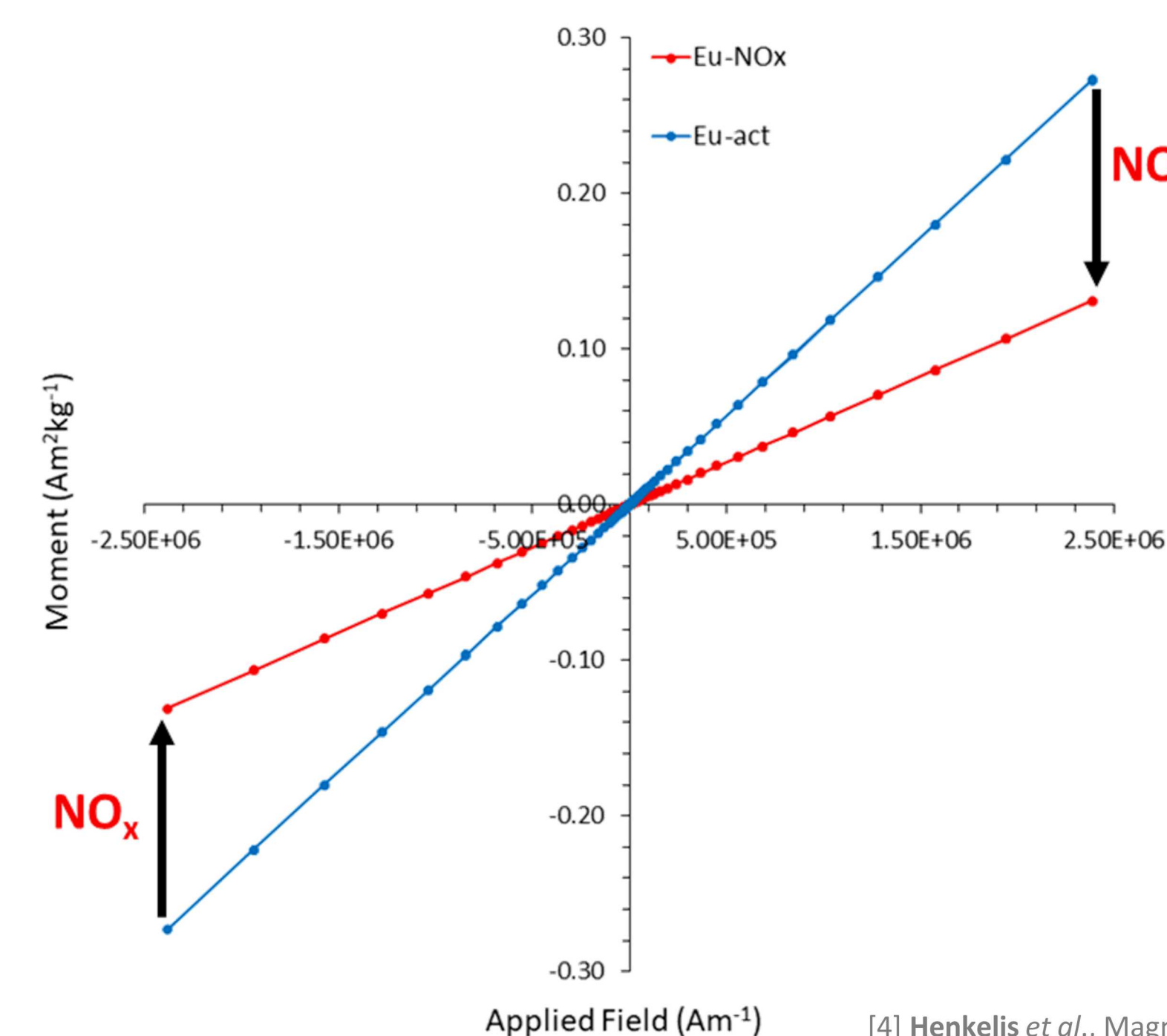
Eu-DOBDC and Tb-DOBDC:

- no change in IR
- SO_x physisorbed within pores



[3] Henkelis et al., Selective adsorption of SOx from humid gas streams by DOBDC-containing MOFs, **2020**, In Preparation;

Magnetism

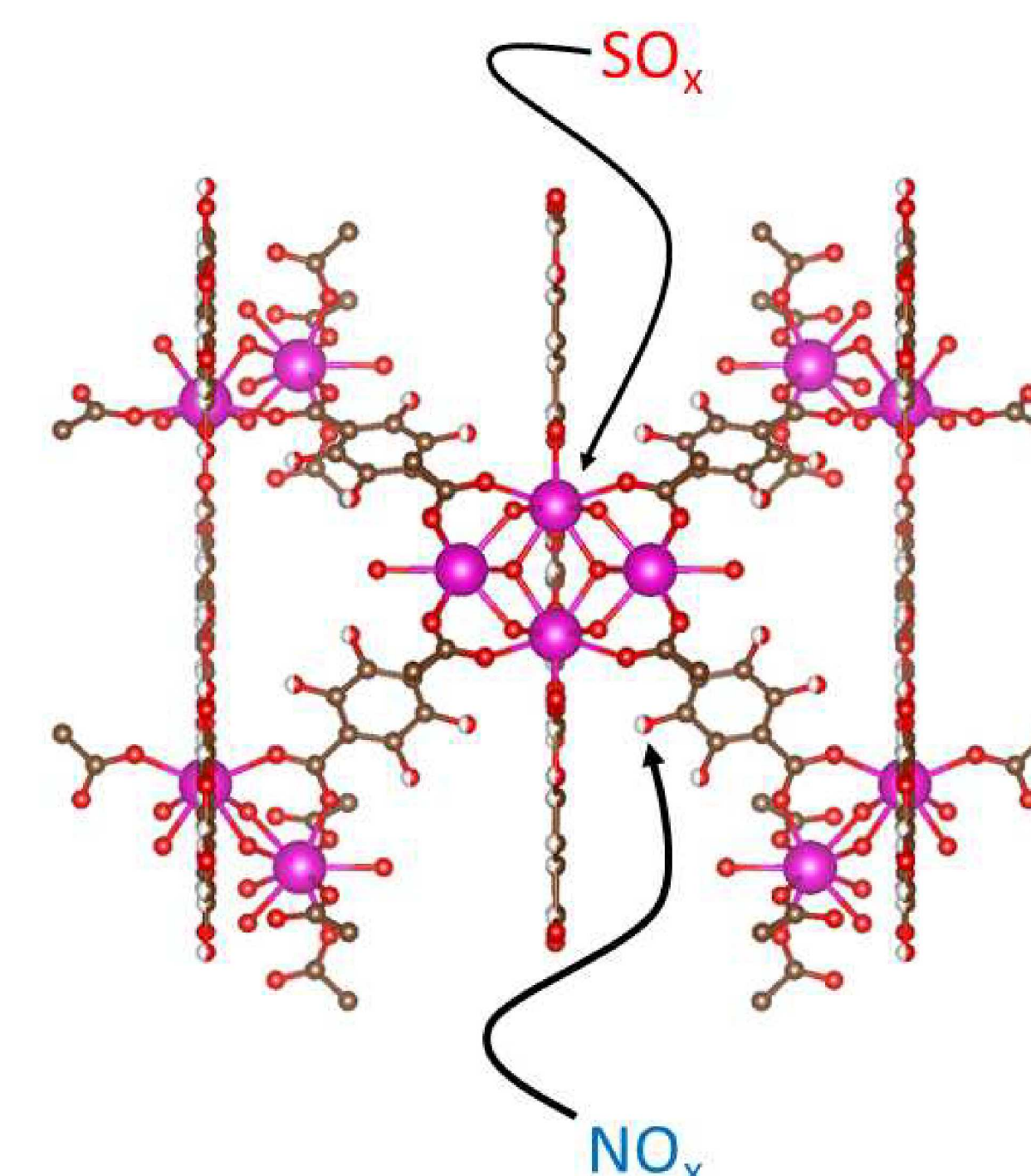


[4] Henkelis et al., Magnetic Tunability in RE-DOBDC MOFs via NOx Acid Gas Adsorption, ACS Appl. Mater. & Interfaces, **2020**, Submitted

- Magnetic moment of metal center is imparted to MOF framework
- NO_x is an electron withdrawing group
- Net result: reduction in magnetic susceptibility (χ) of overall MOF framework
- Eu, Tb, Yb-DOBDC – paramagnetic susceptibility
- Eu > Tb > Yb $\Delta\chi$ with addition of NO_x
- Charge transfer of 1 upe from: metal → ring → EW group ∴ reducing magnetism
- Y-DOBDC – diamagnetic; susceptibility ↓ with NO_x⁴

Conclusions

- RE-DOBDC MOFs are structurally stable to humid acid gas streams – studies into the structure/property relationships are on-going
- Free linker hydroxyls = fluorescent MOF
- Quenching of photoluminescence with adsorption of NO_x – ligand binding
- No quenching with adsorption of SO_x – metal binding
- NO_x reduces overall magnetism of RE-DOBDC
- Eu & Tb-DOBDC – use as a dual optical and magnetic sensor for selectively adsorbing NO_x from mixed acid gas streams
 - Quenching of photoluminescence and large $\Delta\chi$
- Y & Yb-DOBDC – use as an adsorbent for SO_x from NO_x/SO_x streams
 - S-species seen in IR
- On-going research – Investigation into competitive NO_x/SO_x binding in DOBDC containing MOFs (RE-DOBDC, MOF-74) for selective removal of acid gases from flue streams



Acknowledgements