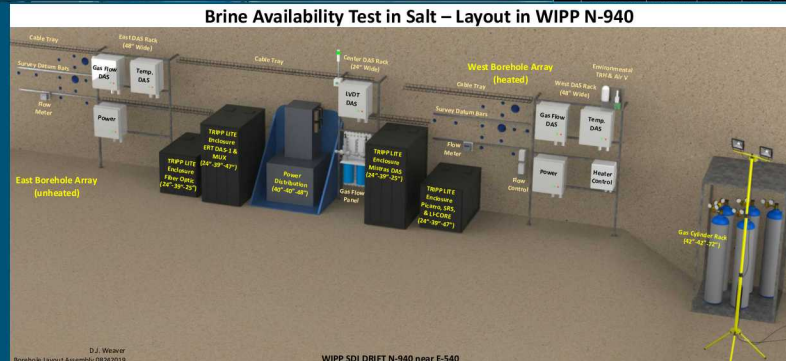
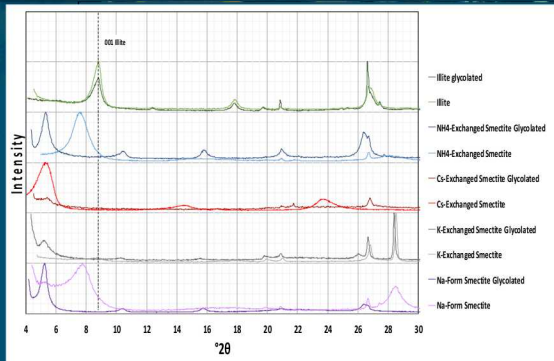
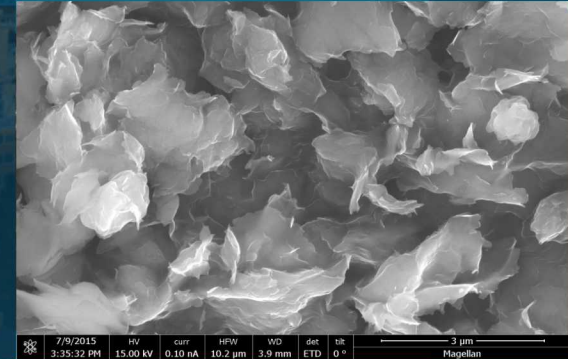


SAND2020-2238C

Investigation and Characterization of Geotechnical Barrier Materials at Elevated Temperatures



Melissa Mills

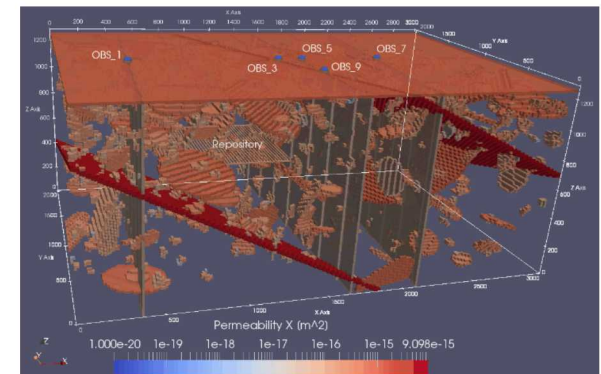
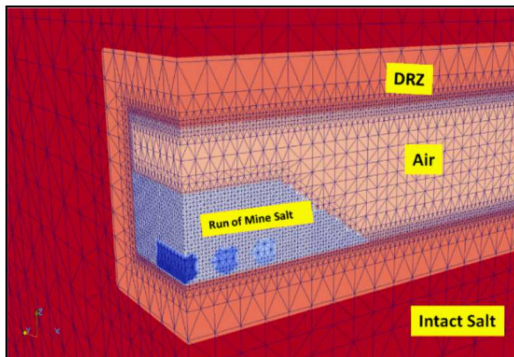
Sandia National Laboratories
Albuquerque, NM, USA

Friday, February 28, 2020

Hannover, Germany



- Department of Energy Office of Nuclear Energy focused on research and development activities related to storage, transportation, and disposal of spent nuclear fuel and high-level nuclear waste.
- Within disposal campaign, various host rock media are being investigated to advance the understanding and performance of long-term isolation



Research presented here focuses on characterizing buffer materials proposed for geotechnical barriers with respect to heat-generating nuclear waste

Motivation

- Ongoing uncertainty in long-term performance of geologic repositories for nuclear waste disposal revolves around construction and temporal evolution of geotechnical barriers
 - In salt, barriers will have permeability and porosity approaching those of the surrounding host rock
 - In crystalline, barriers are critical for isolating emplaced waste given the properties of the surrounding host rock
 - Few large-scale, long-term in situ barrier experiments involve heat, especially including permeability and chemistry data
- Striving to add confidence in future nuclear waste repository design and performance with respect to heat generating waste

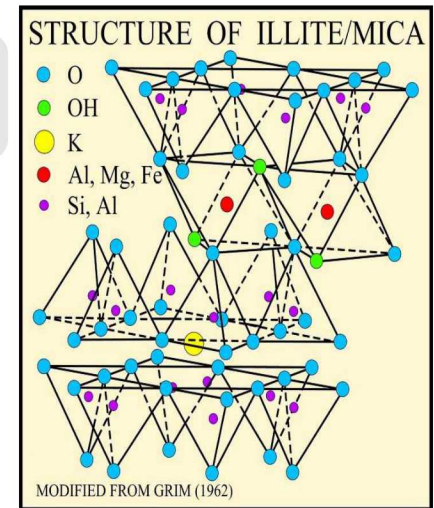
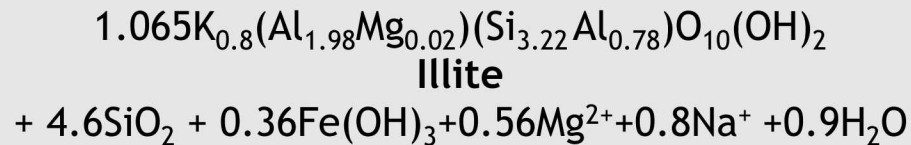
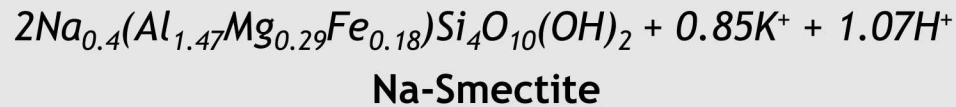
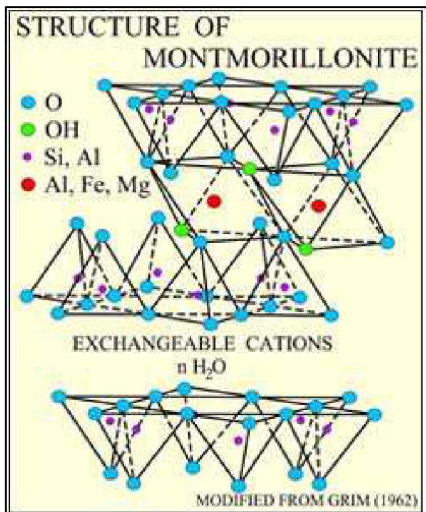


Clay Barriers: Illitization Studies



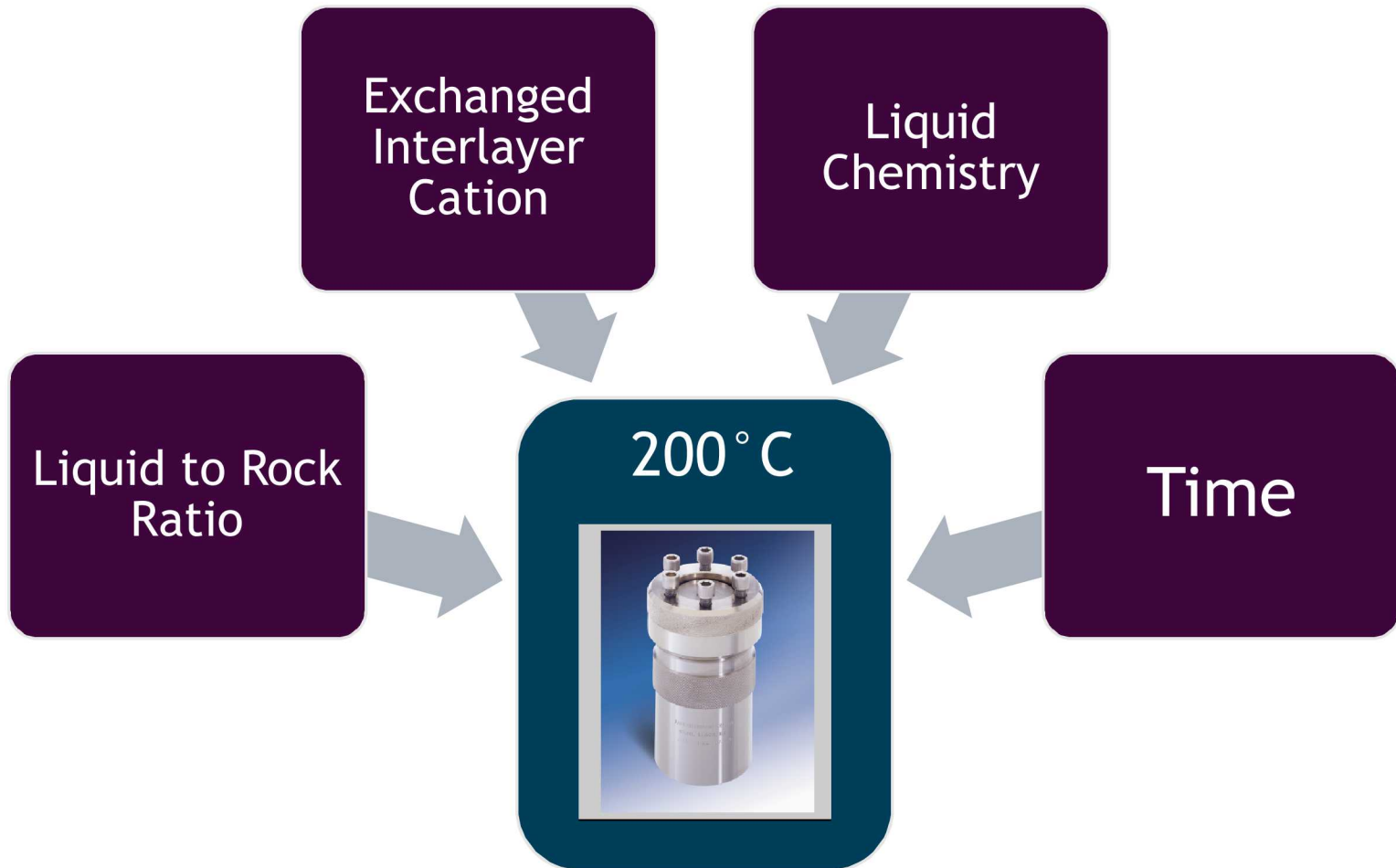
5 Background: Illitization

- To predict barrier functions over time within repository relevant temperatures, important to understand thermal alteration effects on smectite (montmorillonite), a main constituent of bentonite barriers
 - Illitization is a type of thermal alteration which weakens barrier functions



- Path to illitization is complex
 - Dependent on Si, K, interlayer cations, temperature, solution chemistry, and time
- Important to safety case in crystalline repository since buffer break down is the main source of any release

Experimental Approach for Illitization



Volume limit for vessel: 150 mL

7 **Current Sample Matrix**

Temperature	200° C			
Time	1 Week	2 Week	2 Month	
Ratio of Liquid to Rock	100	500	1000	
Exchanged Cation	Sodium	Potassium	Cesium	Ammonium
Liquid Chemistry	1M KCl	1M Parent Cation	DI Water	



= 108 different samples

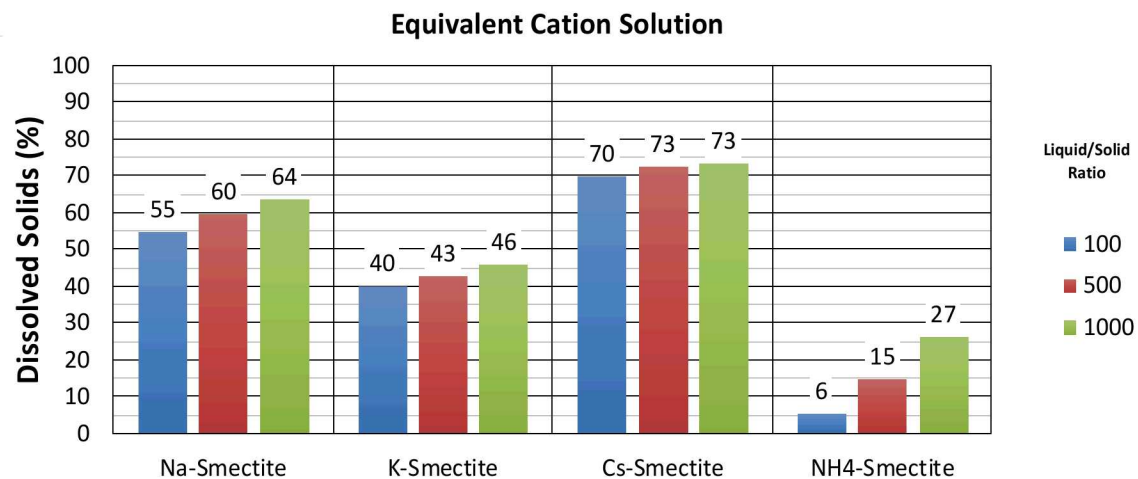
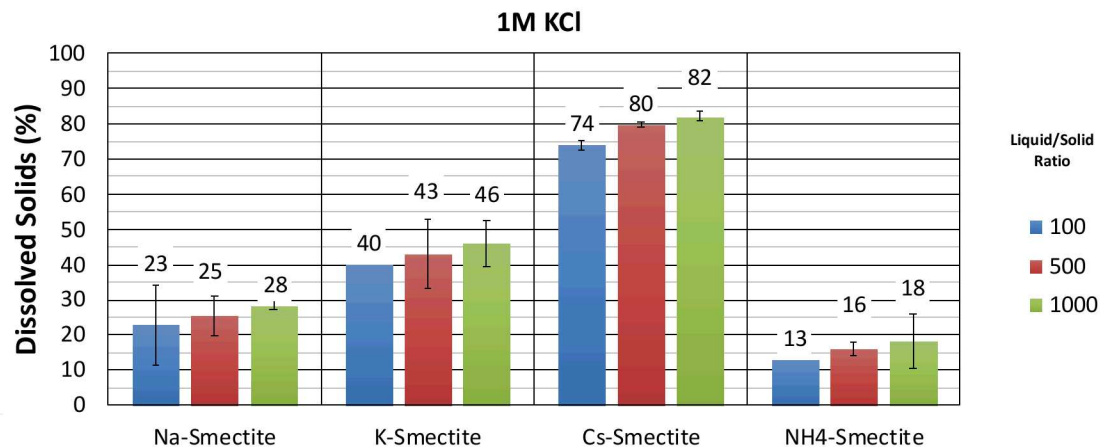
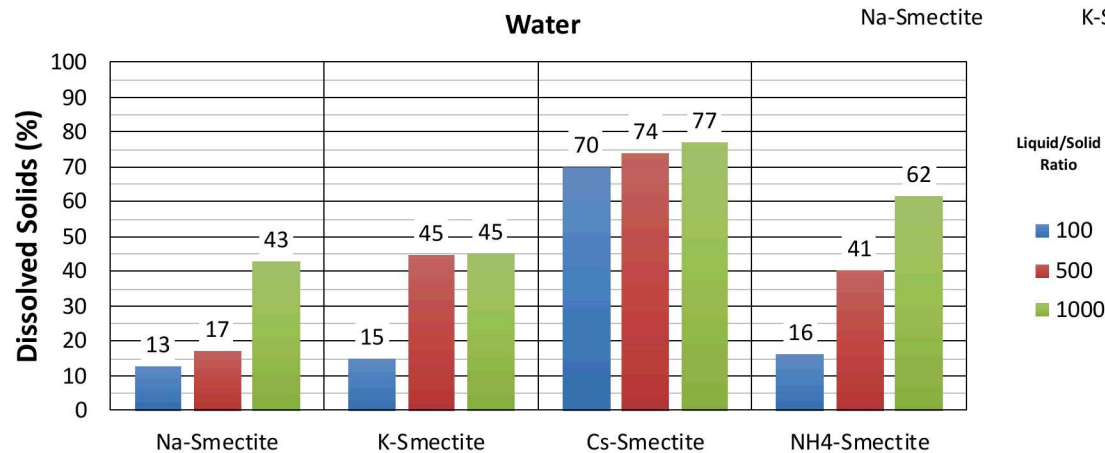
Experimental Techniques

- XRD for basal spacing and lattice structure
- Cation exchange capacity (CEC)
- Recorded dissolved solid data and pH of reacted solutions
- Surface area by BET nitrogen adsorption
- Chemical analysis of solutions after reaction by IC, ICP-MS, ICP-OES to determine dissolved elements
- Particle size by Malvern Zetasizer
- XRF for composition
- SEM-EDS for morphology and compositional mapping

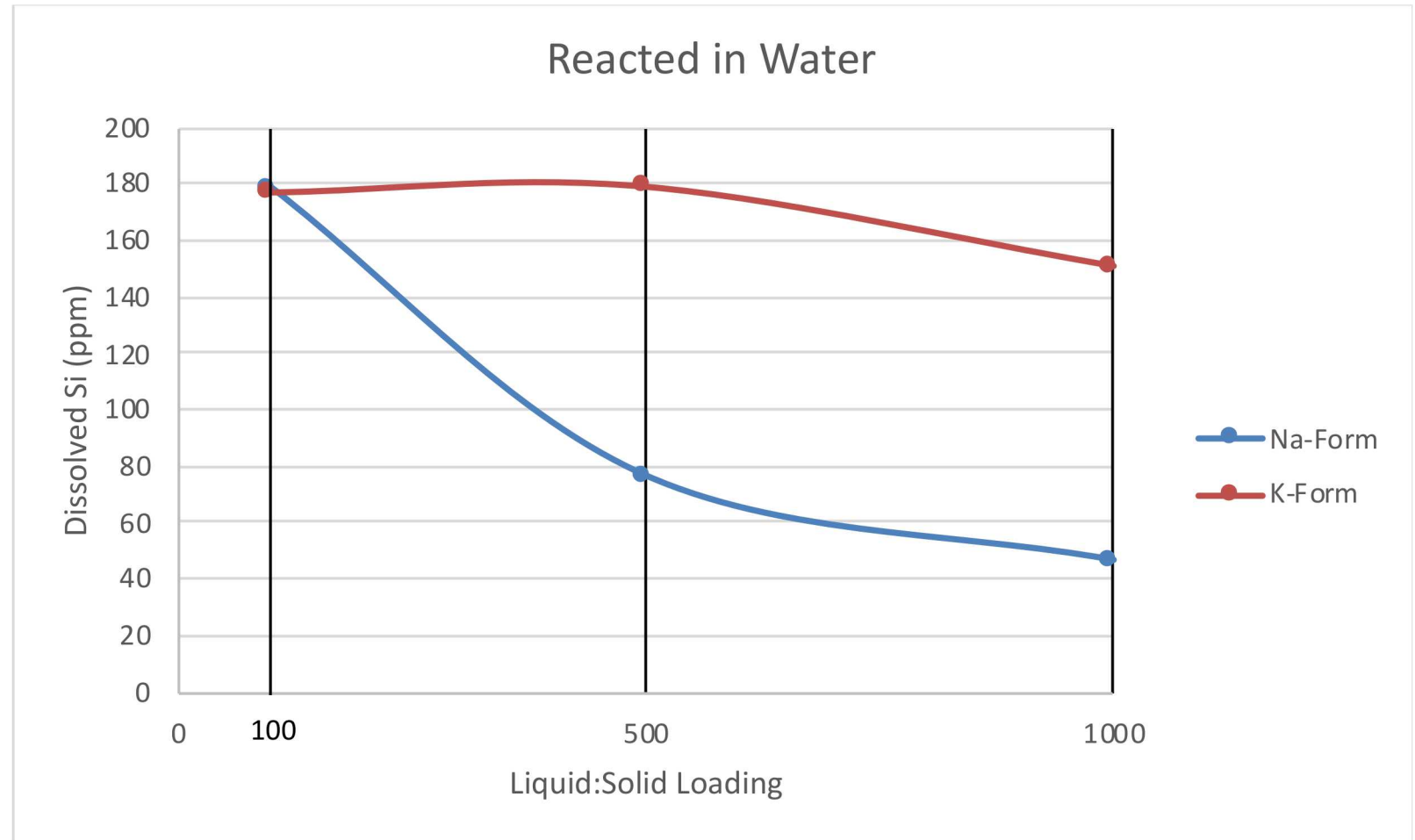
Observations

- Variable dissolution amounts
 - Solid in \neq solid out
 - Dependent on liquid:rock loading and cation
- Basal spacing changes
 - XRD peaks shift
- Surface area and particle size increase in certain cases
- No significant change in pH
- Variability among concentration of Silica in solution

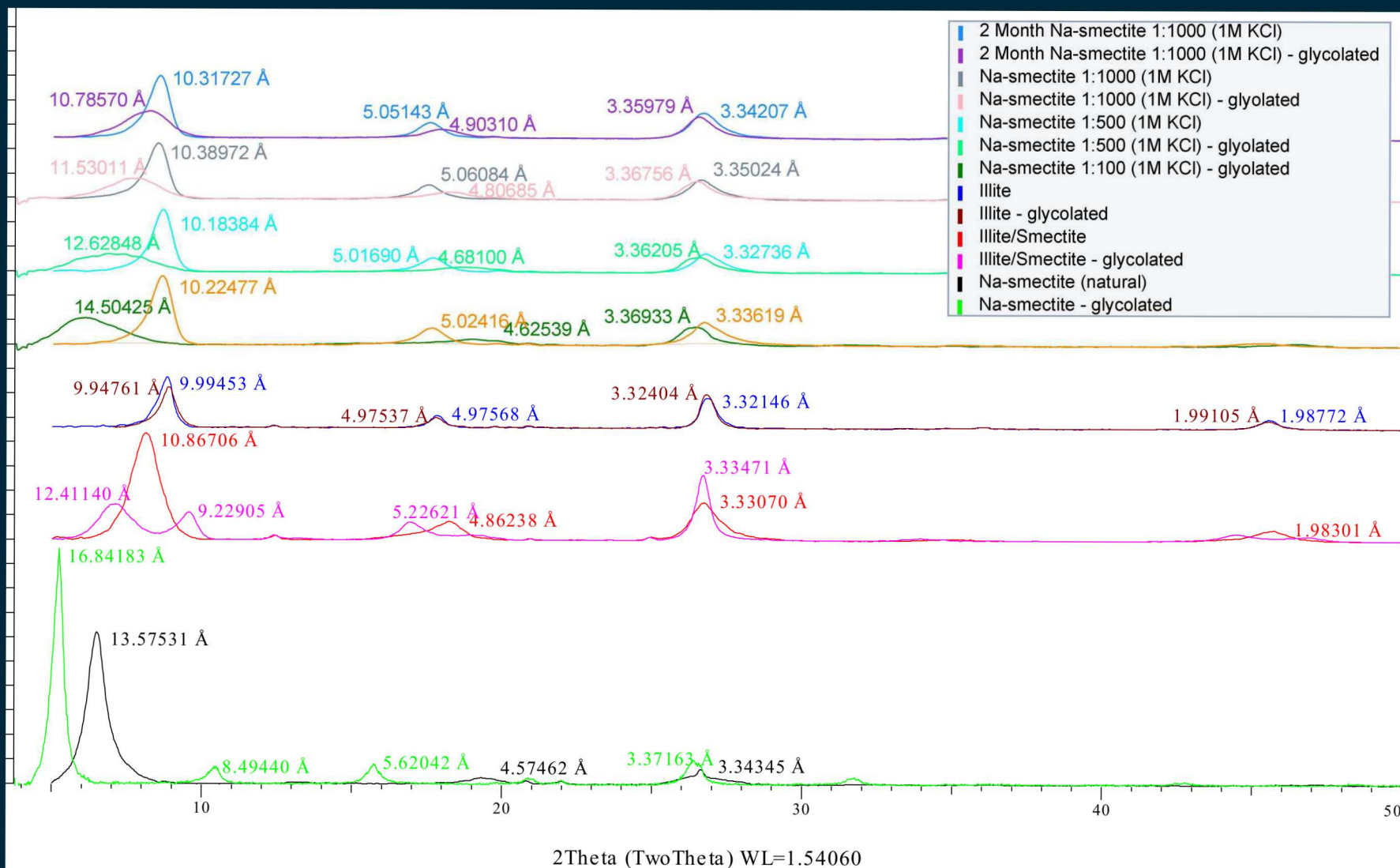
Results: Dissolution



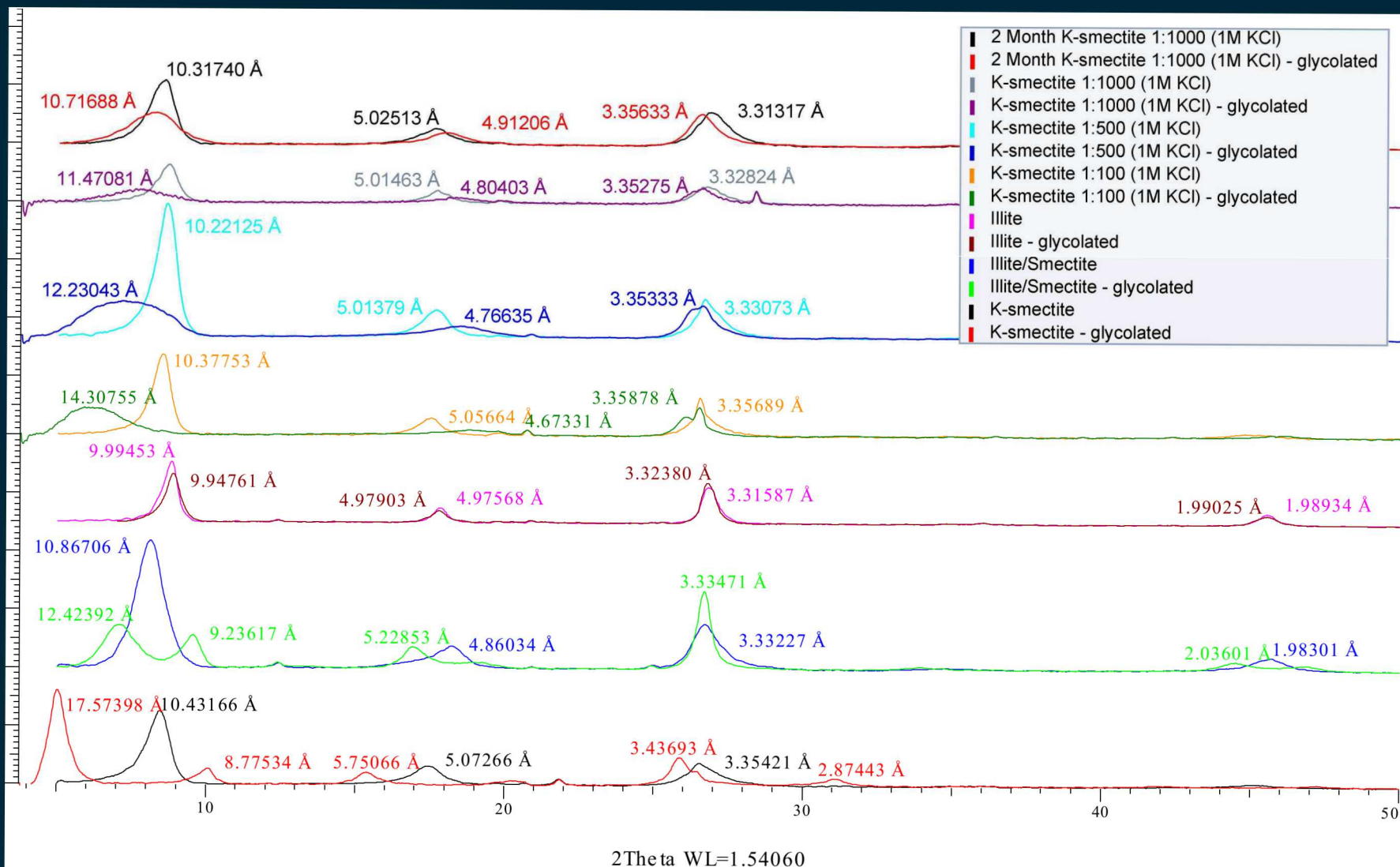
Dissolved Silica



Results: XRD- Sodium Form Smectite in 1M KCl

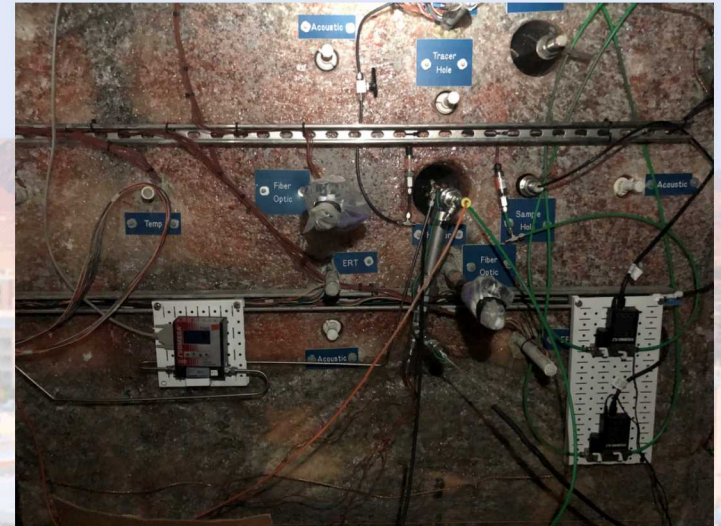


Results: XRD- Potassium Form Smectite in 1M KCl

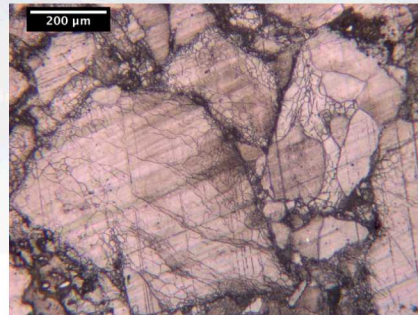


Remaining Research

- Explore different and longer time scales
- Continue analysis of solution chemistry
 - Effect of changes in liquid chemistry
- Continue analysis of composition (XRF)
- Changes in morphology by SEM
- Investigate behavior of iron in reaction
- Addition of mica to possibly escalate conversion
- Addition of quartz to inhibit conversion



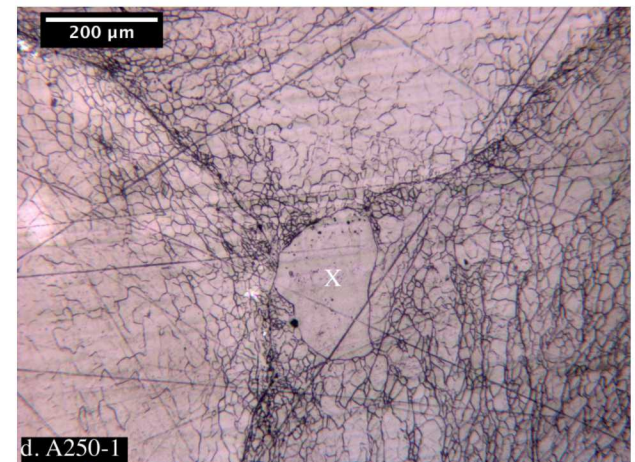
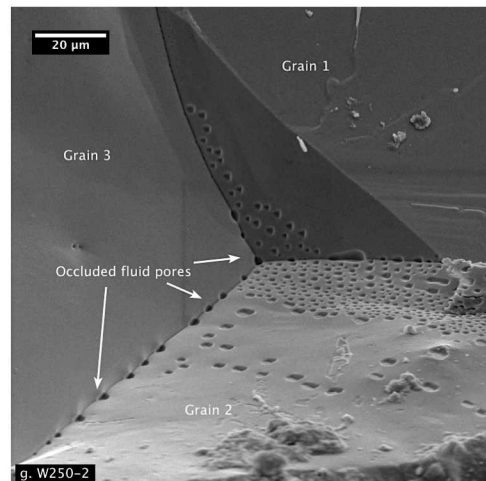
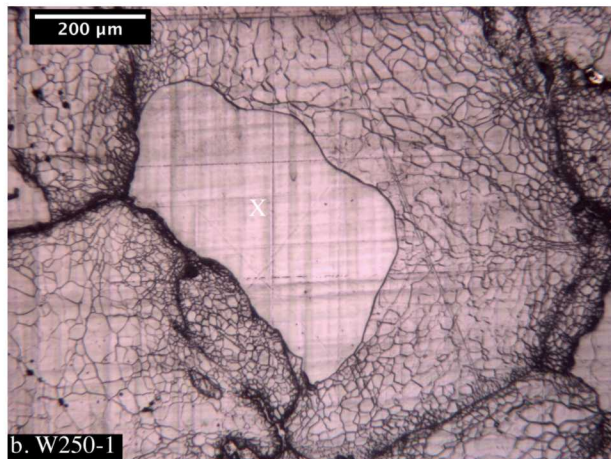
DOE-NE Salt Seal System Research



Barrier Systems in a Salt Repository

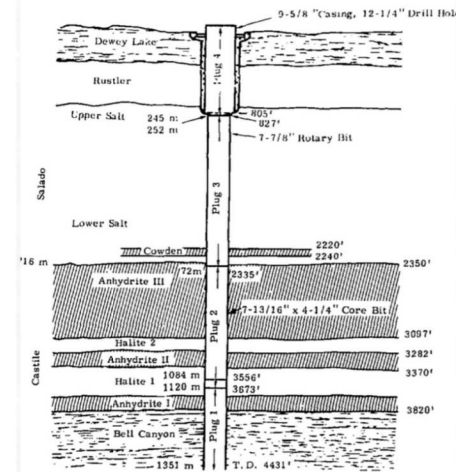
- Many programs on barriers systems in salt formations for repository applications
- Main types of barriers consist of either granular salt backfill or cementitious materials
 - Many different compositions, techniques of emplacement, and response of material depending on the formation (bedded, domal, or pillow)
- Focus will be on research conducted at elevated temperatures in relation to the Waste Isolation Pilot Plant (WIPP) in Carlsbad, NM.

- Stormont et al. 2017: *Improving the Understanding of the Coupled Thermal-Mechanical-Hydrologic Behavior of Consolidating Granular Salt*
 - Granular domal and bedded salt consolidation at temperatures up to 250 °C, pressures up to 40 MPa, and addition of moisture
 - Porosity, permeability, and thermal conductivity was measured
 - Microscopic investigations were conducted to document the deformation mechanisms associated with reconsolidation
 - Results show samples exposed to higher temperatures of 250°C experience recrystallization regardless of moisture conditions



Previous Cement Seal Experiments related to WIPP

- 1977- ERDA No. 10 Test (near WIPP)
- 1979- Bell Canyon Test (near WIPP)



Gulick (1979)

Figure 1. ERDA No. 10 Drill Hole

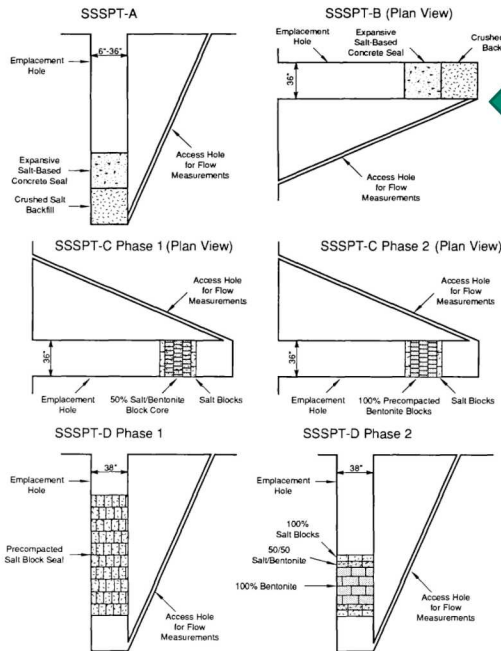


Figure 2. Generalized Test Configurations.

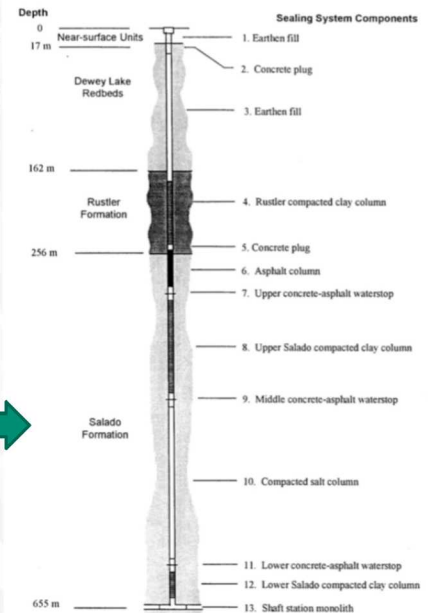
Finley et al. (1991)

- 1980's- Small-Scale Seal Performance Tests (at WIPP)

- Used very specific formulation of "Expansive Salt Concrete"
- Key ingredients are unavailable and potentially difficult to reproduce

- 1990's- Salado Mass Concrete

- Design for closure

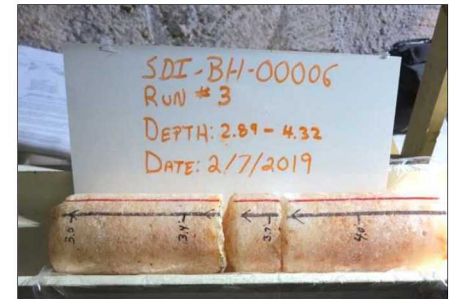
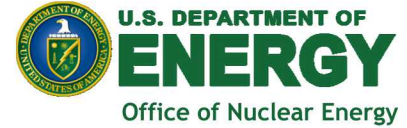


Hansen and Knowles (1999)

None focused on effects of elevated temperature exposure

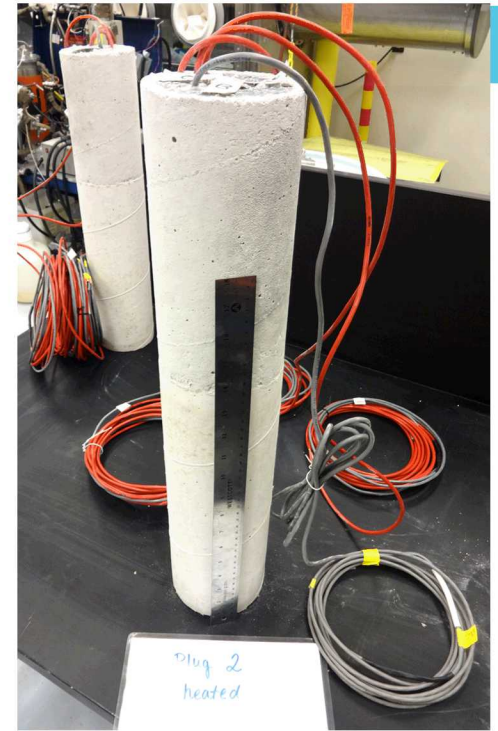
Brine Availability Test in Salt (BATS)

- Heated field test in ~6-m horizontal boreholes at WIPP
 - Began heating: January 2020
 - Parallel heated/unheated tests
 - Temperatures up to 140°C
- Only active repository research underground test in the US
- Collaboration of Sandia, Los Alamos, and Lawrence Berkeley National Labs
- Includes cement seal components

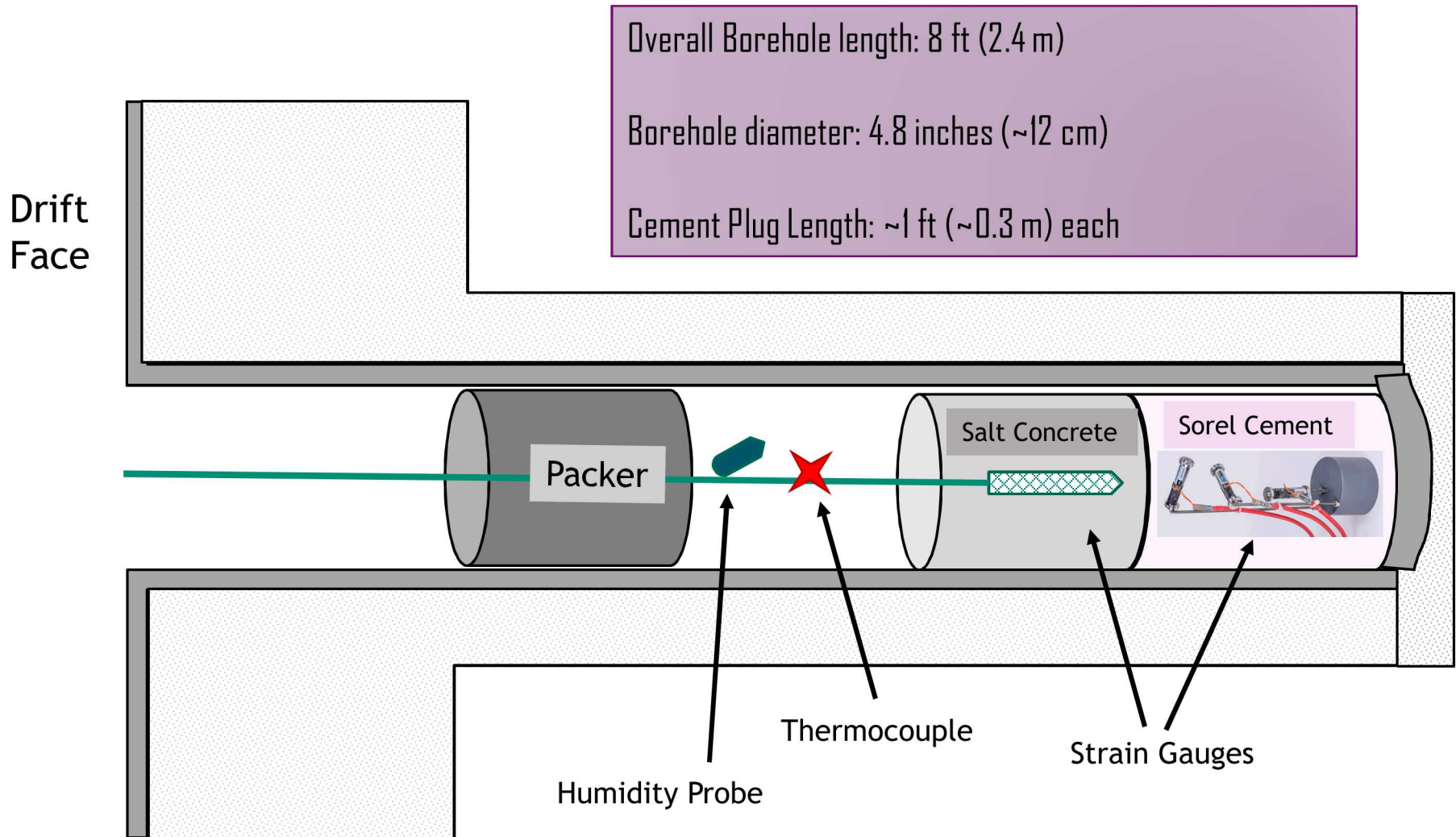


BATS: Cement Seal Borehole

- Two types: modified salt concrete and Sorel cement
 - Align with German programs
 - Complementary with laboratory tests
- Embedded strain gauges during fabrication
- Installed in borehole 1- 2 ft (0.3-0.6 m) from heated borehole
- Strain, temperature, moisture conditions, and brine composition monitored throughout the test
- Will overcore post-test to investigate salt/cement interface
- Understand the behavior of cement/salt/brine system



Side Profile View of Cement Seal Borehole



BATS: Recipes of Emplaced Plugs

Sorel Cement D4 (5-1-8 phase)

- Proportions similar to Popp et al. 2018
 - Density = 2240 kg/m³
- Desired fast setting
- Used MgO currently emplaced with waste at WIPP
 - Crushed and sieved <75 μ m
 - reactivity = 272 s
- Aggregate: Run of mine WIPP salt <4 mm grain size

Sorel D4	Composition (mass- (mass-%)
5 M MgCl ₂	18.3
< 75 μ m MgO	18.3
Salt Aggregate	63.4

Salt Concrete

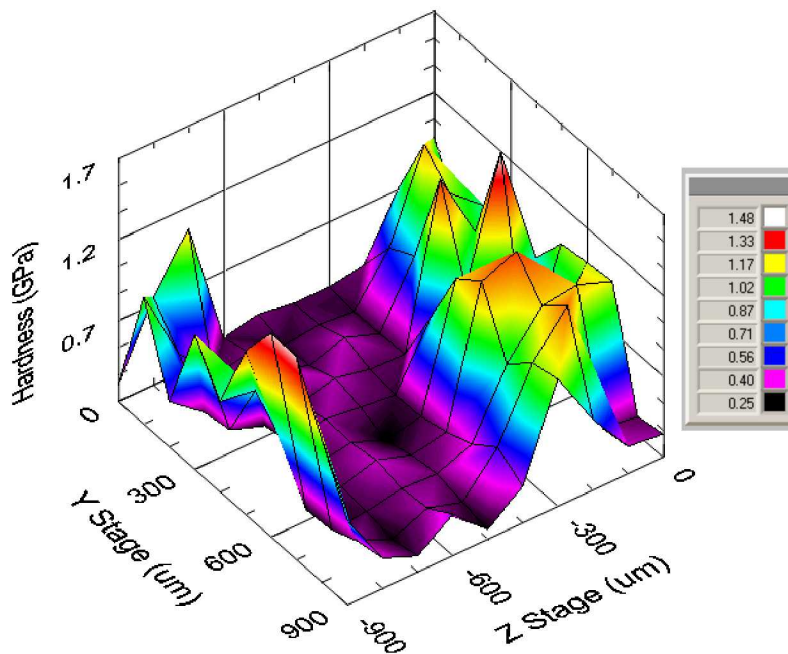
- Followed recipe from Muller-Hoppe et al. 2010 (LAVA2)
- Ground Blast Furnace Slag
- Aggregate: Run of mine WIPP salt <4 mm grain size
 - Impurities affected mixture
- Required to be mixed in glove box

Salt Concrete	Composition (mass-%)
Saturated NaCl	14.7
Blast Furnace Slag	28.4
Salt Aggregate	56.9

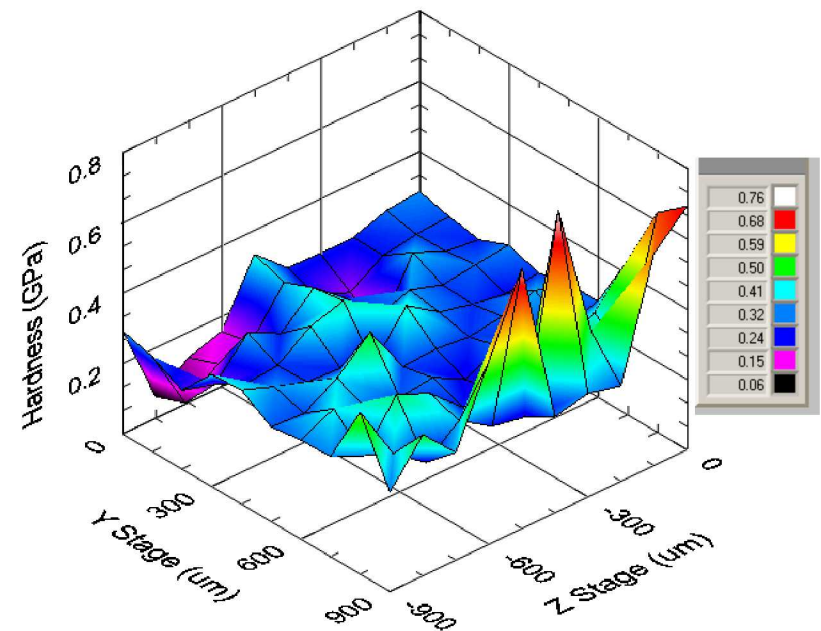
BATS: Characterization of Fabricated Plugs

- Currently analyzing mechanical and chemical properties of sub-samples from installed plugs
- Nano-indentation (250 mN constant load, 10x10 grid, 100 μm spacing)

Sorel Cement

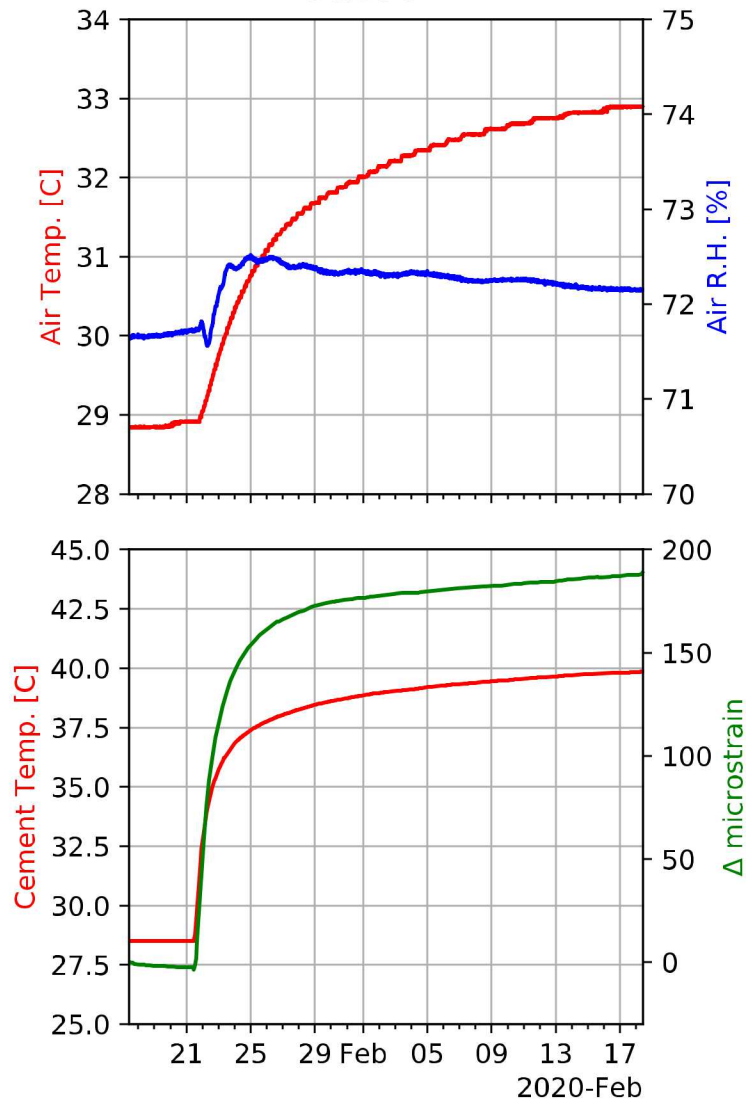


Salt Concrete

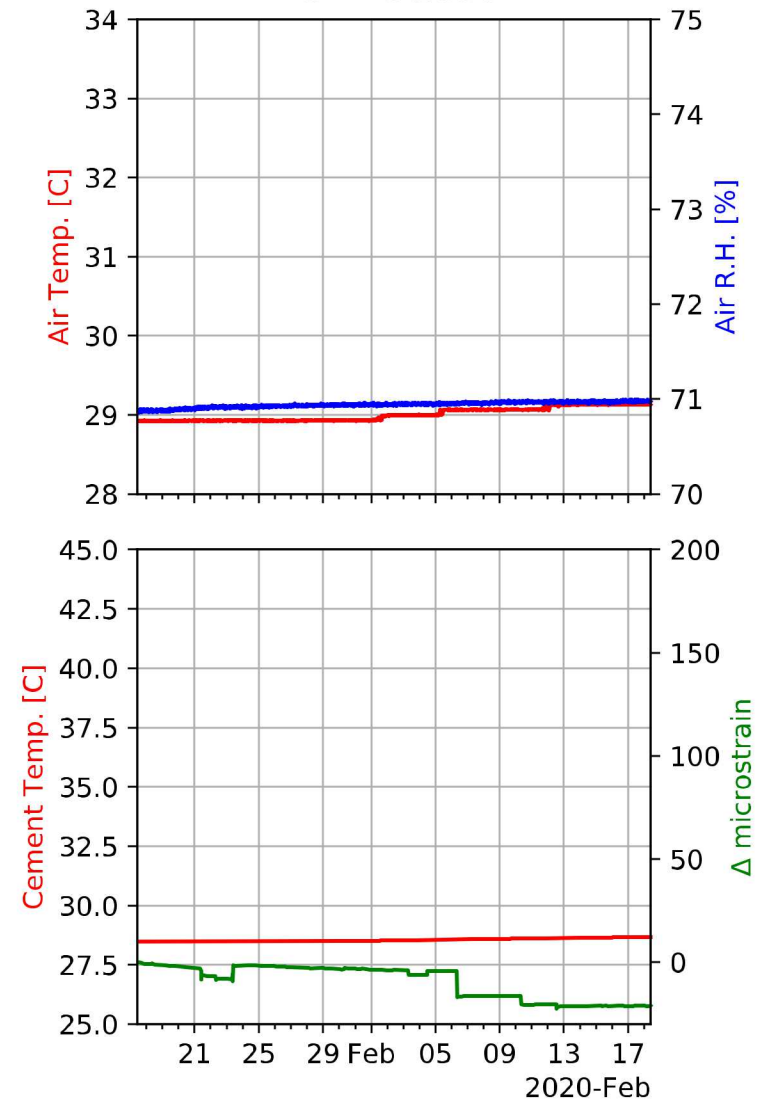


Preliminary In-Situ Data from Seal Borehole

Heated



Unheated



Future Goals and Endeavors of Cement Seals

- Further investigation into other seal material recipes
- Lab-scale tests
 - Installing into hollowed bedded salt core (similar to DOPAS)
 - Effects of heat
- Analysis of over-cored plug
- Additional plugs deployed in other boreholes at WIPP
 - *In situ* permeability measurements desired

Questions?