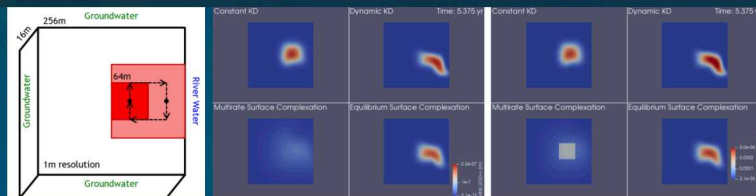
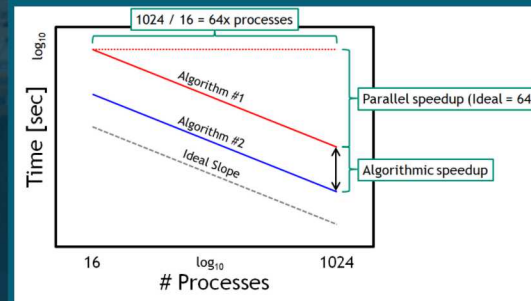




# Bounding Computation and Complexity for Reactive Transport on Supercomputers: A Perspective from Nuclear Waste Repository Performance Assessment



Sorption Model	Prescribed Uniform Flow		Heterogeneous Flow Field	
Constant $K_0$	Operator Split	Global Implicit	Operator Split	Global Implicit
Dynamic $K_0$	Operator Split	Global Implicit	Operator Split	Global Implicit
Equilibrium Surface Complexation	Operator Split	Global Implicit	Operator Split	Global Implicit
Kinetic Multirate Surface Complexation	Operator Split	Global Implicit	Operator Split	Global Implicit

PRESENTED BY

Glenn Hammond and Peter Lichtner\*

International Workshop on: How to integrate geochemistry at affordable costs into reactive transport for large-scale systems

February 5-7, 2020

\*OFM Research

Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.

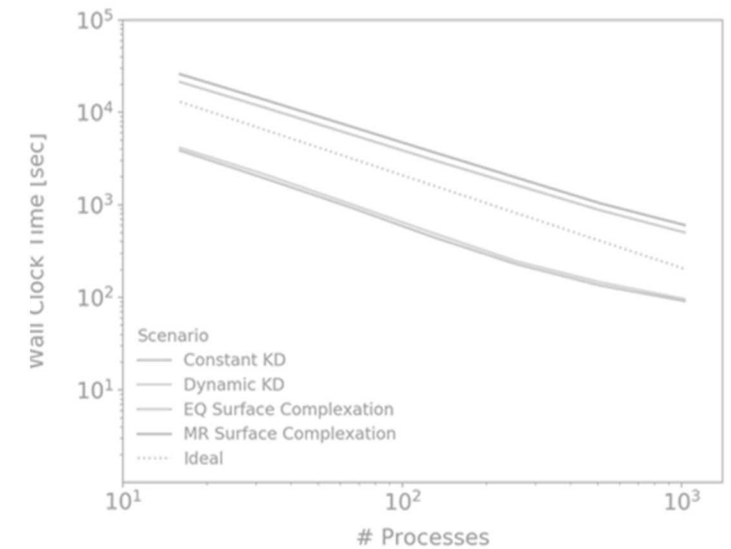
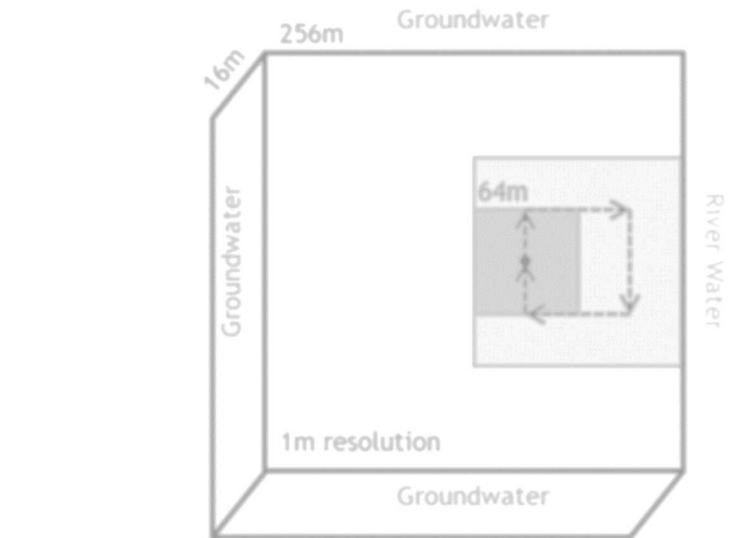
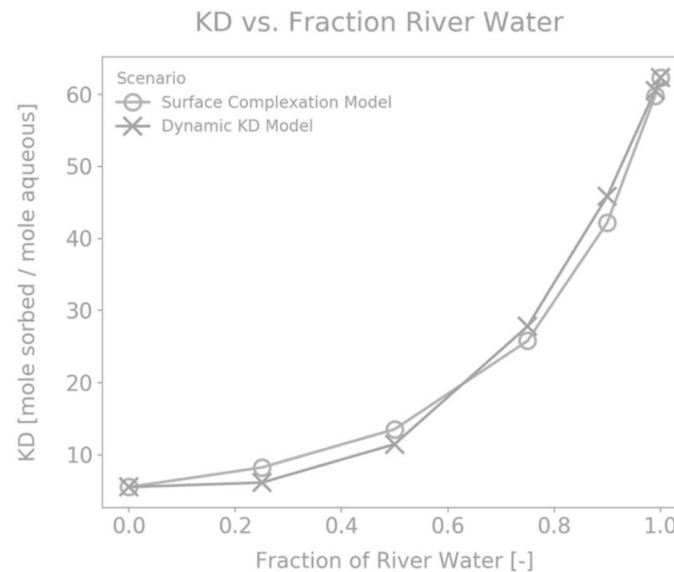
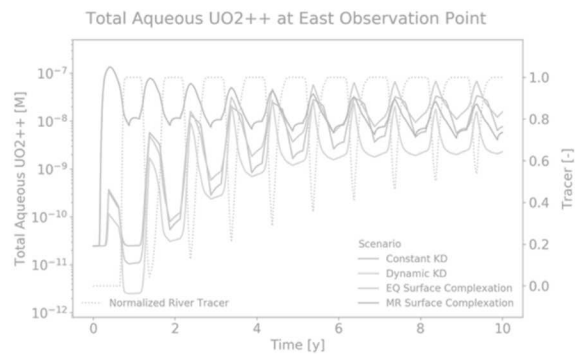
## Motivation

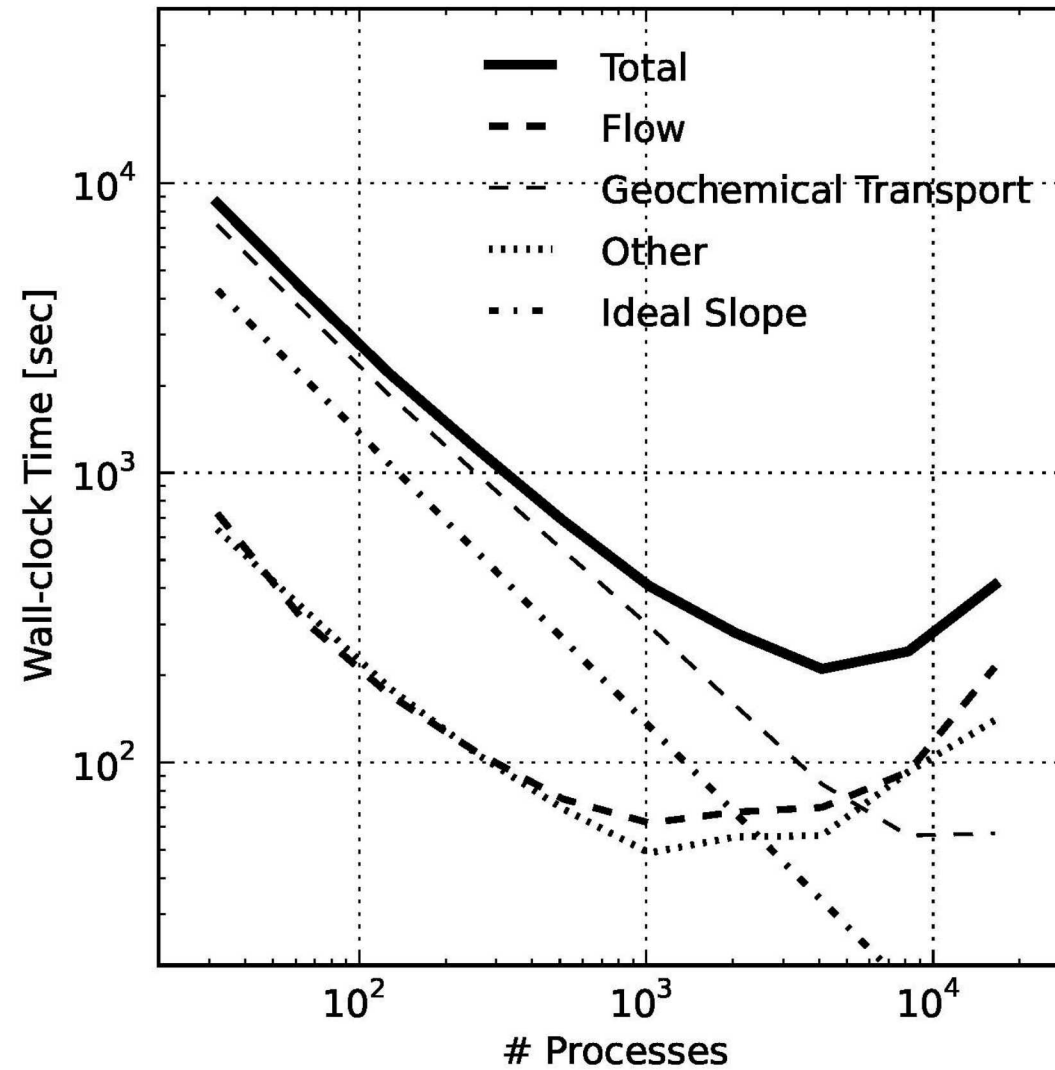
Hanford 300 Area test problem

Prototype dynamic  $K_D$  model

Parallel and algorithmic speedup results

## Conclusion





Hammond, G.E., P.C. Lichtner and R.T. Mills (2014) Evaluating the Performance of Parallel Subsurface Simulators: An Illustrative Example with PFLOTRAN, Water Resources Research, 50, doi:10.1002/2012WR013483.

For a representative reactive transport (surface complexation) problem:

- What is the maximum speedup obtainable through parallel processing?
- What is the maximum speedup obtainable through simplification of chemistry?

# PFLOTRAN

Petascale reactive multiphase flow and transport code

Open source license (GNU LGPL)

Object-oriented Fortran 2003/2008

Founded upon well-known ([supported](#)) open source libraries

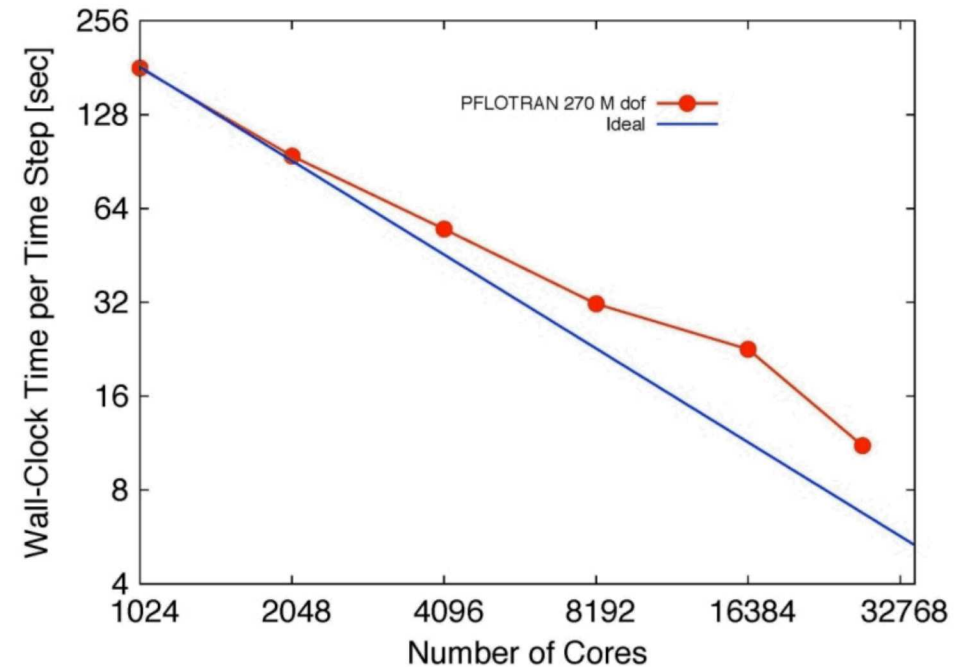
- MPI, PETSc, HDF5, METIS/ParMETIS/CMake

Demonstrated performance

- Maximum # processes: 262,144 (Jaguar supercomputer)
- Maximum problem size: 3.34 billion degrees of freedom
- [Scales well to over 10K cores](#)

More info.

- [documentation.pflotran.org](http://documentation.pflotran.org)

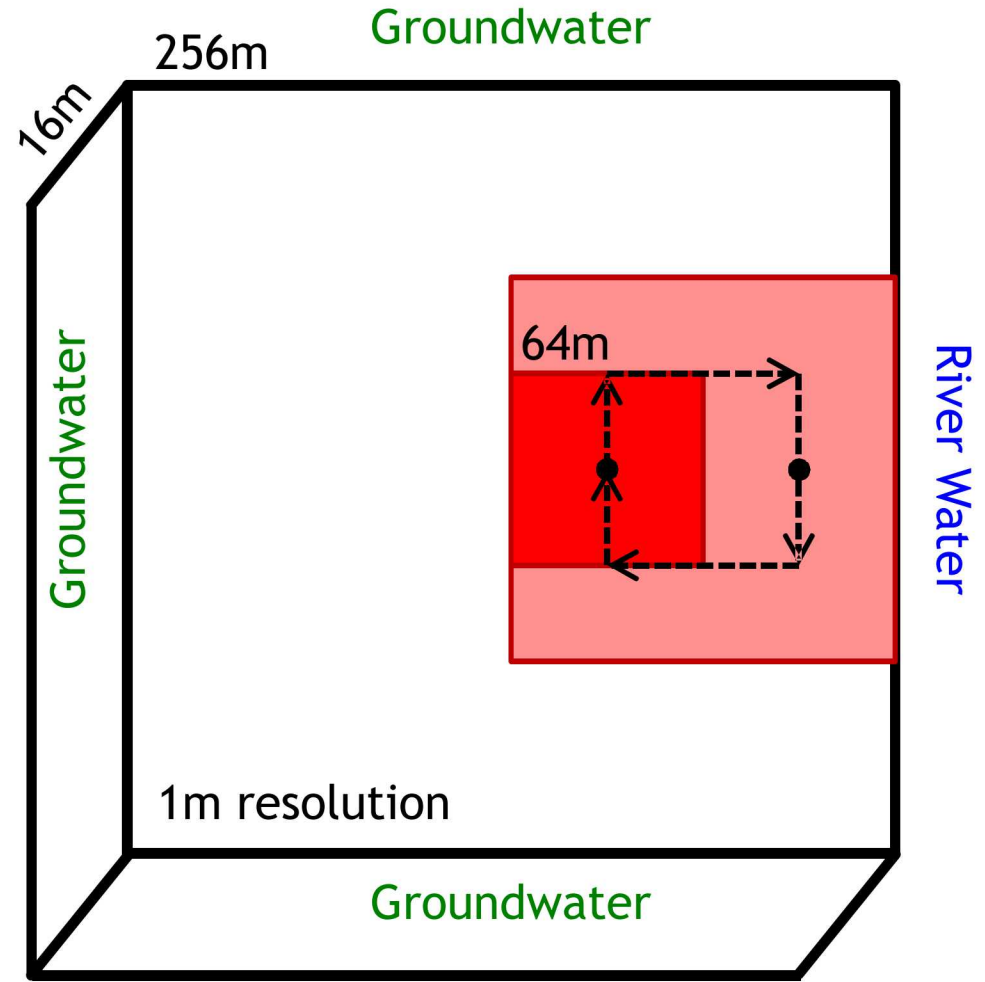
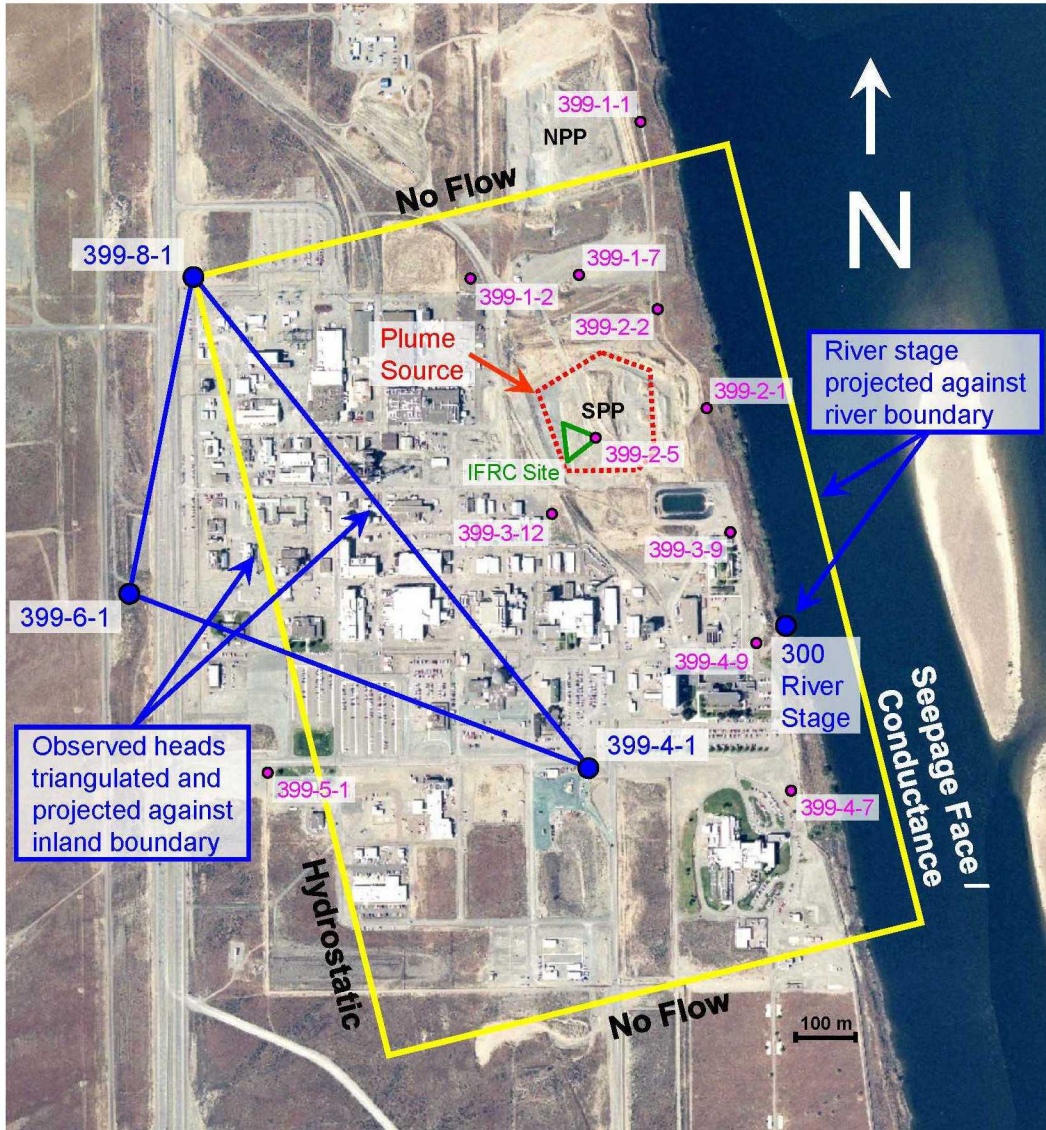




# Hanford 300 Area Test Problem



# 7 Idealized Conceptual Model



256 x 256 x 16 grid cells x 15 components = ~15.7M dof total

10 year simulation time

Run in parallel on 16 - 1024 processes (1M - 15K dof/process)

Hammond, G.E. and P.C. Lichtner (2010) Field-Scale Modeling for the Natural Attenuation of Uranium at the Hanford 300 Area using High Performance Computing, Water Resources Research, 46, W09527, doi:10.1029/2009WR008819.

### 15 primary aqueous species

- $\text{H}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{UO}_2^{2+}$ ,  $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{HCO}_3^-$ ,  $\text{Cl}^-$ ,  $\text{F}^-$ ,  $\text{HPO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ , and 2 tracers

### 88 secondary aqueous species

- Dominant U(VI) species:  $\text{CaUO}_2(\text{CO}_3)_3^{2-}$ ,  $\text{Ca}_2\text{UO}_2(\text{CO}_3)_3(\text{aq})$

### 1 mineral

- Calcite

### 2 surface complexes and 1 surface site

- $>\text{SOUO}_2\text{OH}$ ,  $>\text{SOHOU}_2\text{CO}_3$ ,  $>\text{SOH}$

# Groundwater and River Water Composition at Hanford 300 Area

	Groundwater†	River Water†
pH	7.6	7.8
Ionic Strength	6.88e-3	4.39e-4
Ca <sup>2+</sup>	1.21e-3	4.50e-4
Mg <sup>2+</sup>	5.10e-4	1.82e-4
HCO <sub>3</sub> <sup>-</sup>	2.57e-3	1.13e-3
K <sup>+</sup>	1.55e-4	1.90e-5
Na <sup>+</sup>	1.34e-3	5.12e-5
K <sub>D</sub> *	5.56	62.37
Equilibrating Mineral	Calcite	N/A

† = equilibrated with sediment

K<sub>D</sub>\* = ratio of sorbed to aqueous mass [dimensionless K<sub>D</sub>]

## Flow

$$\frac{\partial}{\partial t}(\phi s \rho) + \nabla \cdot \rho q = 0$$

$$q = -\frac{kk_r}{\mu} \nabla(p - \rho g z)$$

## Reactive Transport

$$\frac{\partial}{\partial t}(\phi s \Psi_j + \Psi_j^S) + \nabla \cdot \Omega_j = -\sum_m v_{jm}^{\min} I_m$$

$$\Omega_j = (q - \phi s \tau D \nabla) \Psi_j$$

## Total Aqueous Component Concentration

$$\Psi_j = C_j + \sum_i v_{ji}^{\text{aq}} C_i$$

$$C_i = \frac{K_i}{\gamma_i} \prod_j (\gamma_j C_j)^{v_{ji}^{\text{aq}}}$$

## Total Sorbed Component Concentration

$$\Psi_j^S = \sum_{k\sigma} v_{jk}^{\text{srf}} S_{k\sigma}$$

$$S_{k\sigma} = \frac{\omega_\sigma K_k Q_k}{1 + \sum_{k'} K_{k'} Q_{k'}}$$

$$Q_k = \prod_j (\gamma_j C_j)^{v_{jk}^{\text{srf}}}$$

## Mineral Precipitation-Dissolution

$$I_m = -k_m a_m \mathcal{P}_m (1 - K_m Q_m) \zeta_m$$

$$Q_m = \prod_j (\gamma_j C_j)^{v_{jm}^{\min}}$$



# Prototype Dynamic $K_D$ Model

---

## Hanford 300 Area Dynamic $K_D$ Model

$$\Psi_S = \phi s_l K_D \Psi$$

$$K_D = K_{D_{low}} + \left( K_{D_{high}} - K_{D_{low}} \right) C_{river\ tracer}^{K_{D_{power}}}$$

$\Psi$  = aqueous concentration

$\Psi_S$  = sorbed concentration

$C_{river\ tracer}$  = normalized tracer concentration (measure of river water fraction)

$\phi$  = porosity

$s_l$  = liquid saturation

$K_D$  = calculated ratio of sorbed to aqueous mass

$K_{D_{low}}, K_{D_{high}}, K_{D_{power}}$  = parameters read from input deck

# Hanford 300 Area Dynamic $K_D$ Model

$$\Psi_S = \phi s_l K_D \Psi$$

$$K_D = K_{D_{low}} + (K_{D_{high}} - K_{D_{low}}) C_{river\ tracer}^{K_{D_{power}}}$$

$$K_D = 5.56 + (62.37 - 5.56) C_{river\ tracer}^{3.26}$$

$\Psi$  = aqueous concentration

$\Psi_S$  = sorbed concentration

$C_{river\ tracer}$  = normalized tracer concentration

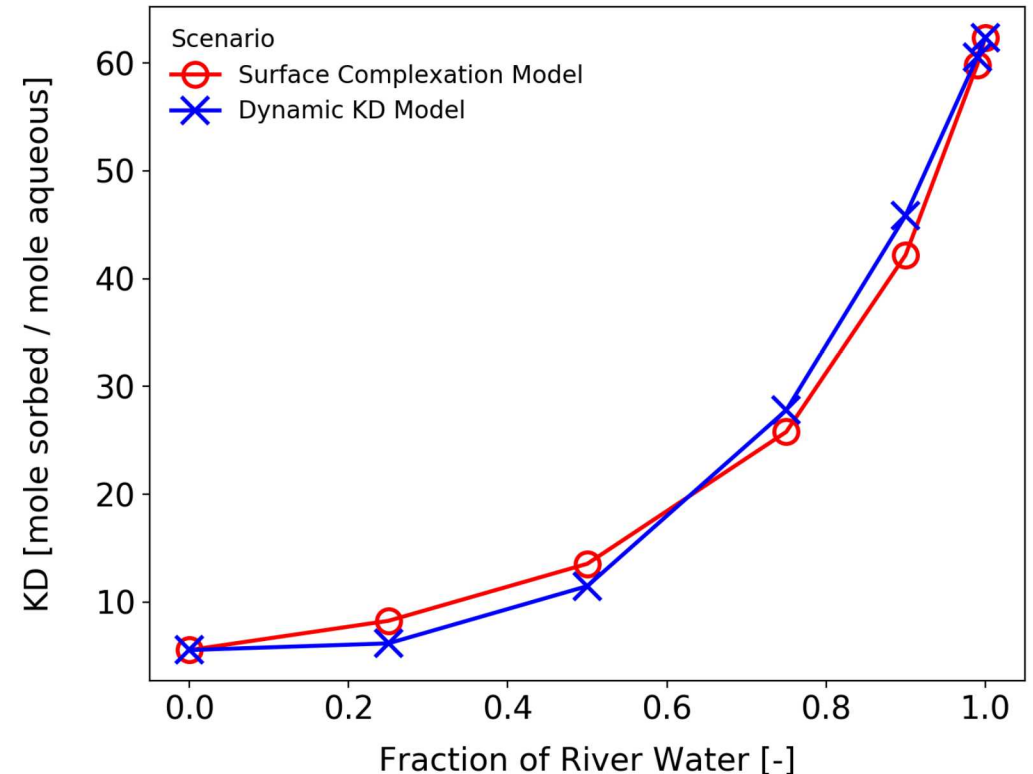
$\phi$  = porosity

$s_l$  = liquid saturation

$K_D$  = calculated ratio of sorbed to aqueous mass

$K_{D_{low}}, K_{D_{high}}, K_{D_{power}}$  = parameters read from input deck

### KD vs. Fraction River Water



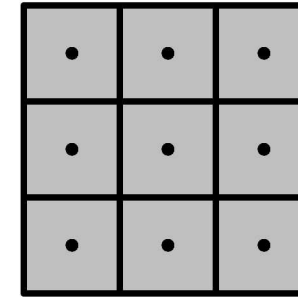
Sorption Model	Prescribed Uniform Flow		Heterogeneous Flow Field	
Constant $K_D$	Operator Split	Global Implicit	Operator Split	Global Implicit
Dynamic $K_D$	Operator Split	Global Implicit	Operator Split	Global Implicit
Equilibrium Surface Complexation	Operator Split	Global Implicit	Operator Split	Global Implicit
Kinetic Multirate Surface Complexation	Operator Split	Global Implicit	Operator Split	Global Implicit

Prescribed Uniform Flow = **Reactive transport only** in a uniform velocity field that changes over time (i.e. no flow calculation).

Heterogeneous Flow Field = **Flow and reactive transport** in a heterogeneous permeability field with dynamic piezometric head gradients that change over time.

# Global Implicit vs. Operator Split Coupling

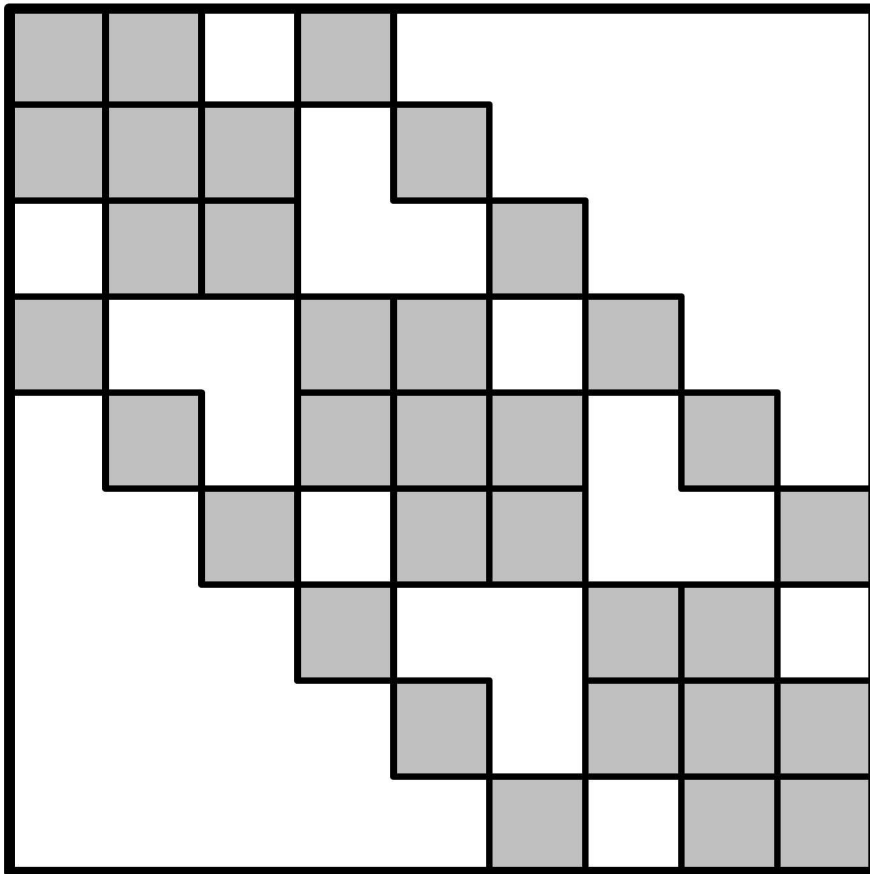
For each timestep:



9 cells  
3 chemical components

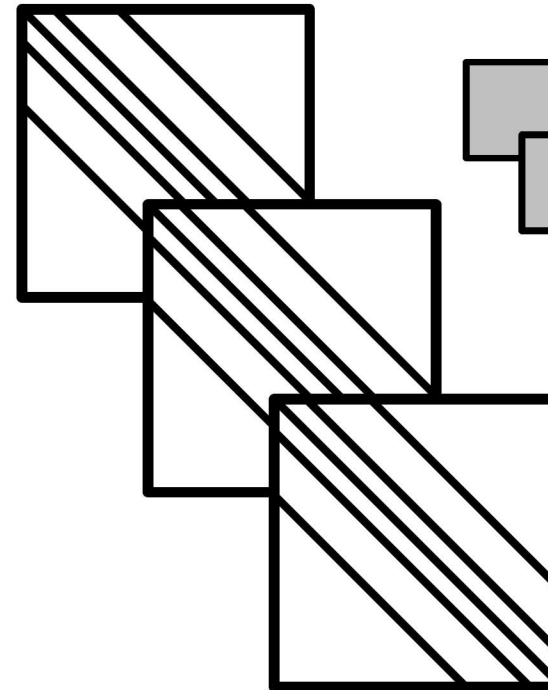
## Global Implicit Coupling

$N \geq 1$  27x27 coupled nonlinear solutions

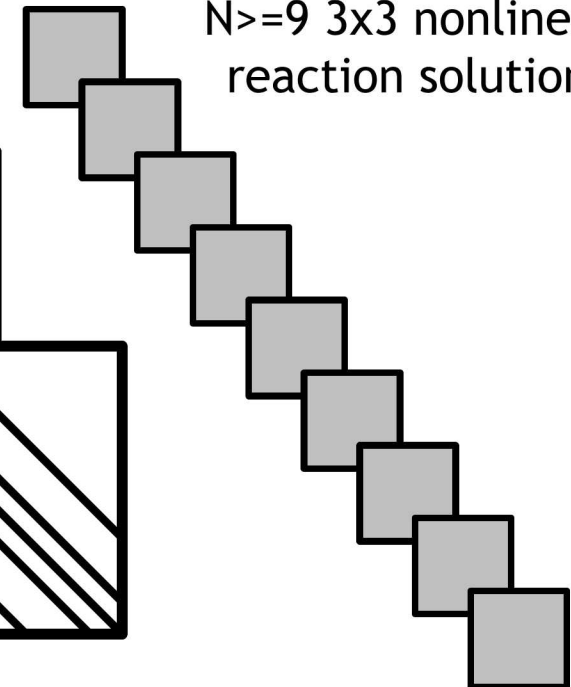


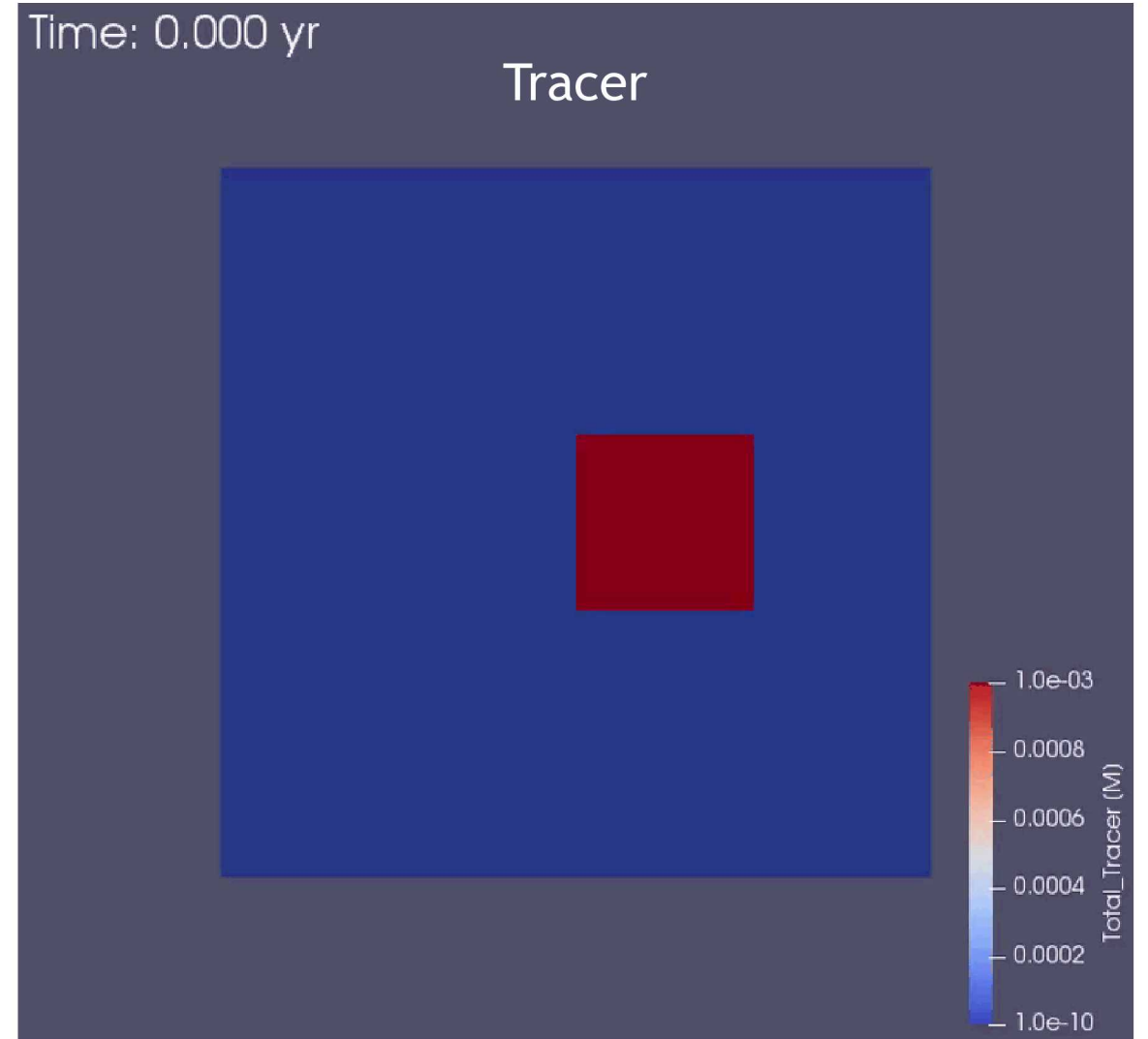
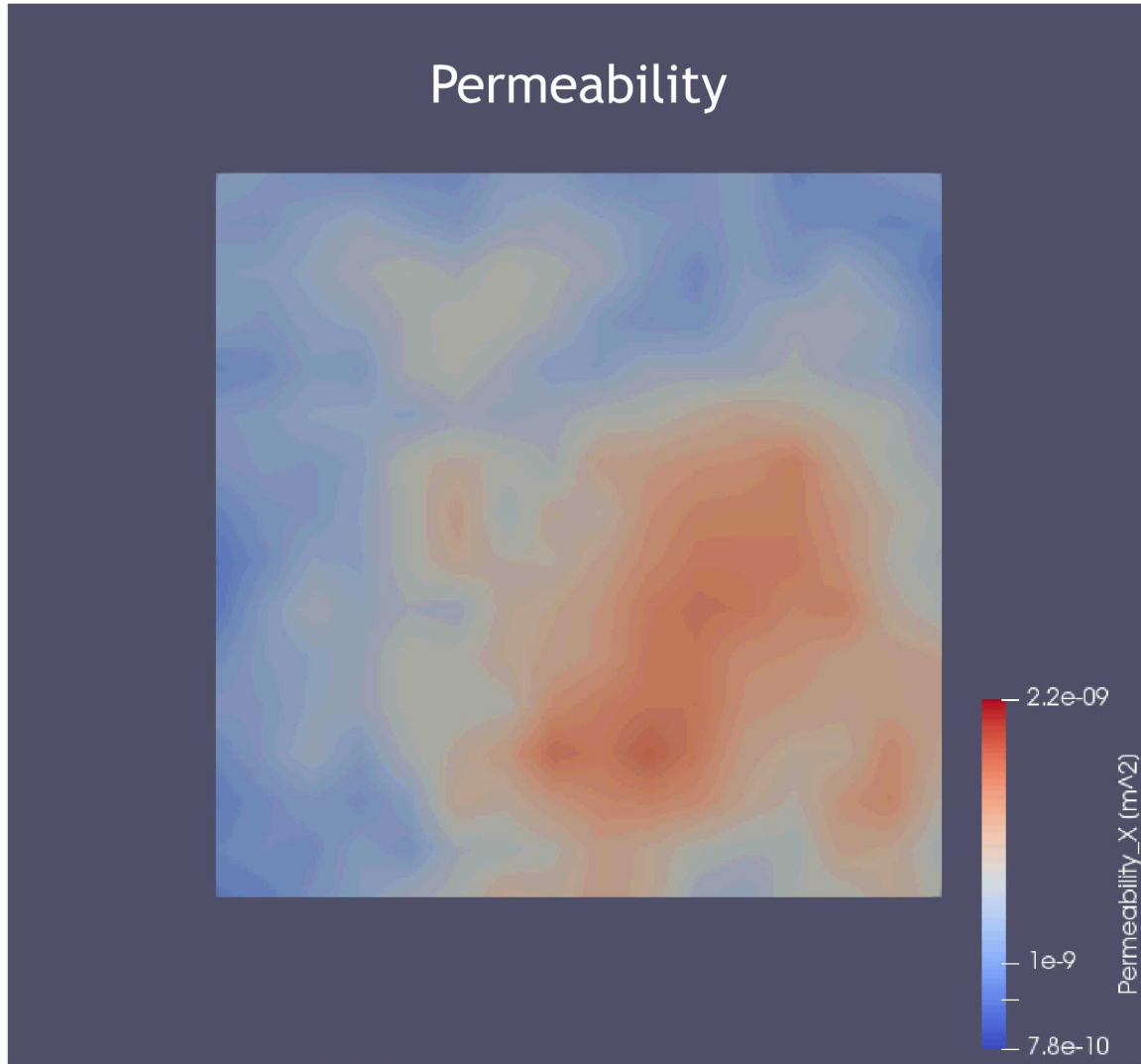
## Operator Split Coupling

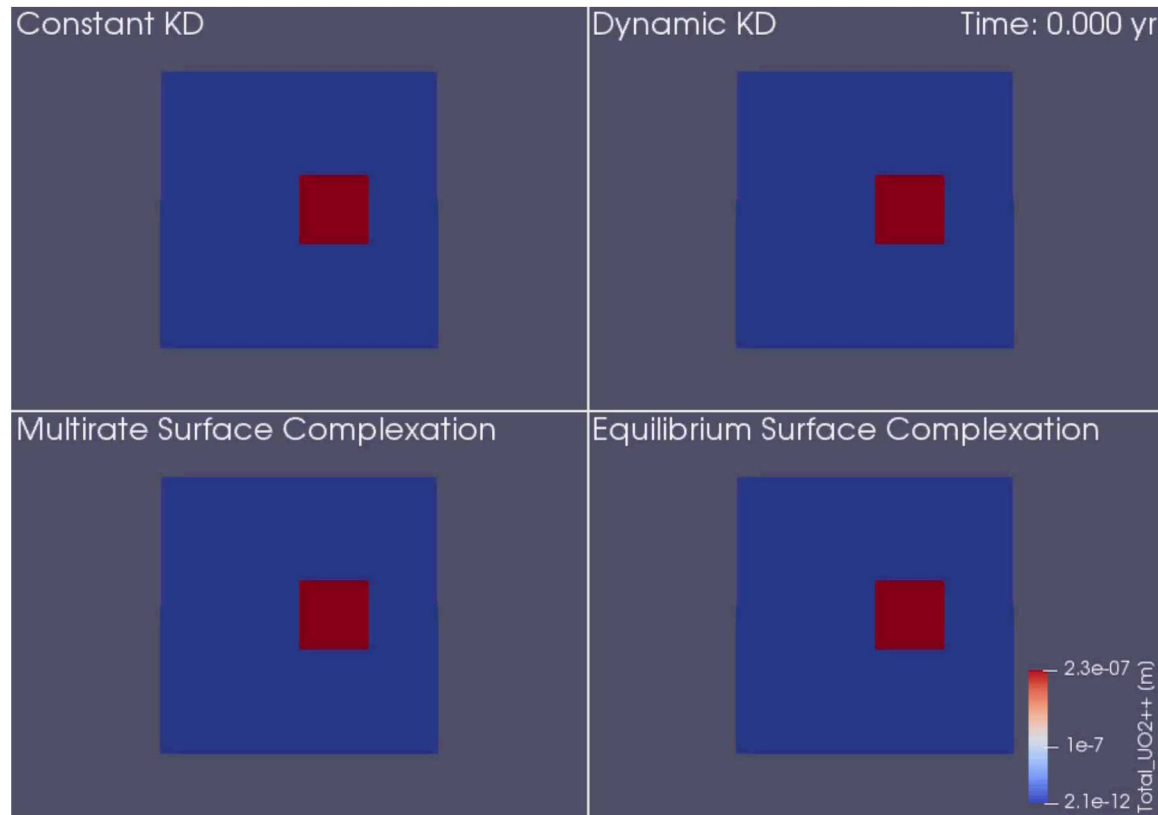
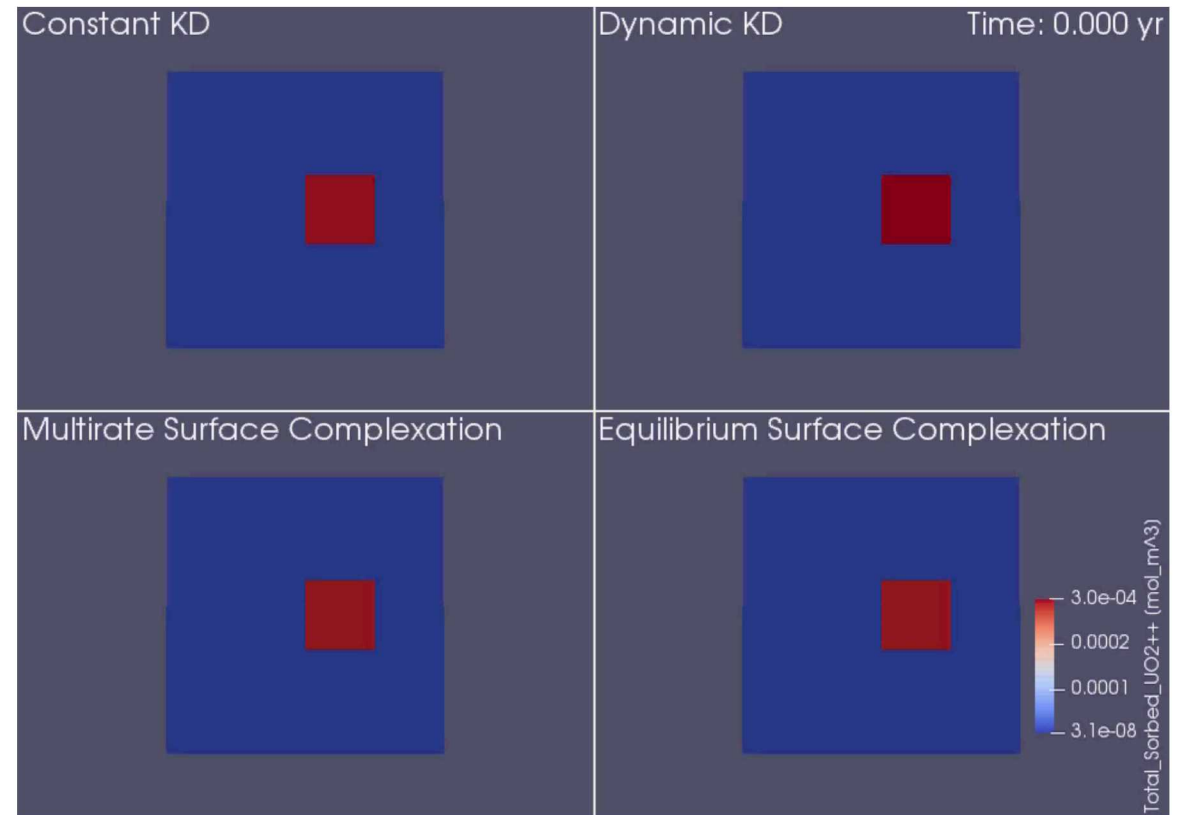
3 9x9 sparse linear transport solutions

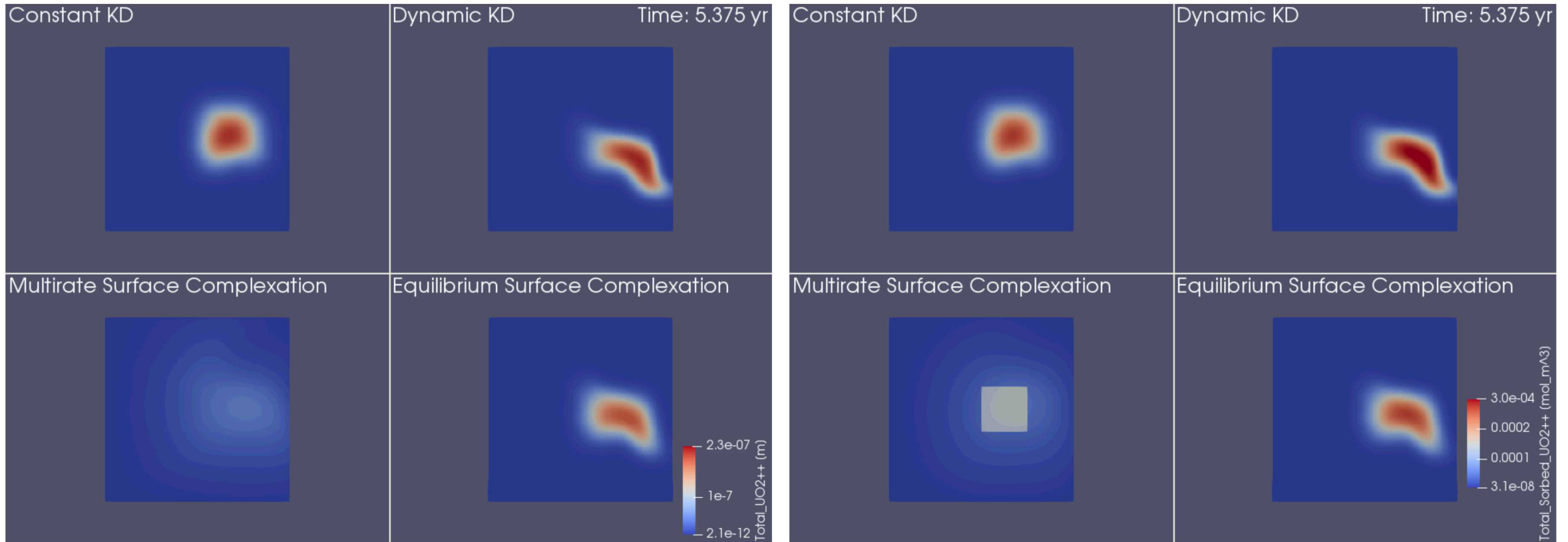


$N \geq 9$  3x3 nonlinear reaction solutions





Animations of Aqueous and Sorbed  $\text{UO}_2^{++}$ Aqueous  $\text{UO}_2^{++}$ Sorbed  $\text{UO}_2^{++}$ 

Snapshots of Aqueous and Sorbed  $\text{UO}_2^{++}$  at 5.375 YearsAqueous  $\text{UO}_2^{++}$ Sorbed  $\text{UO}_2^{++}$ 

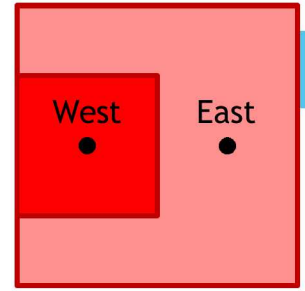


# Parallel and Algorithmic Speedup Results

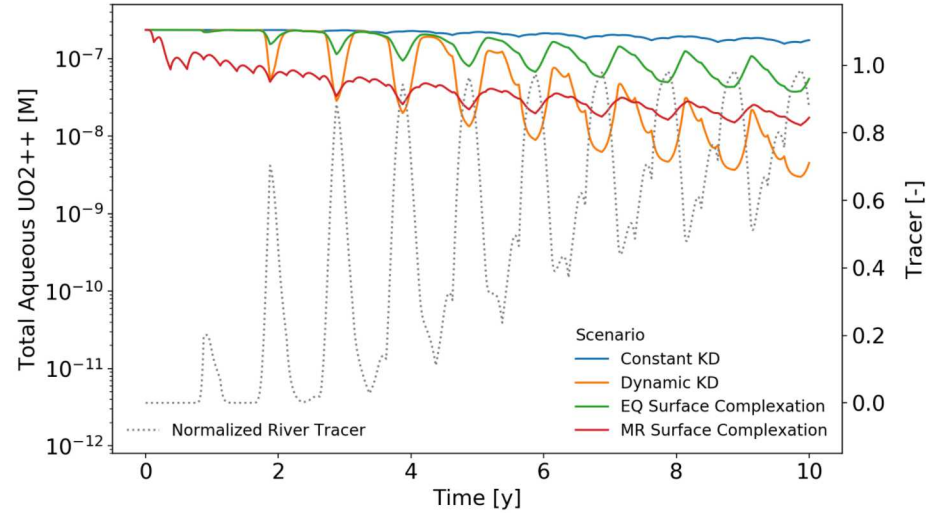
What is the maximum speedup obtainable through parallel processing?

What is the maximum speedup obtainable through simplification of chemistry?

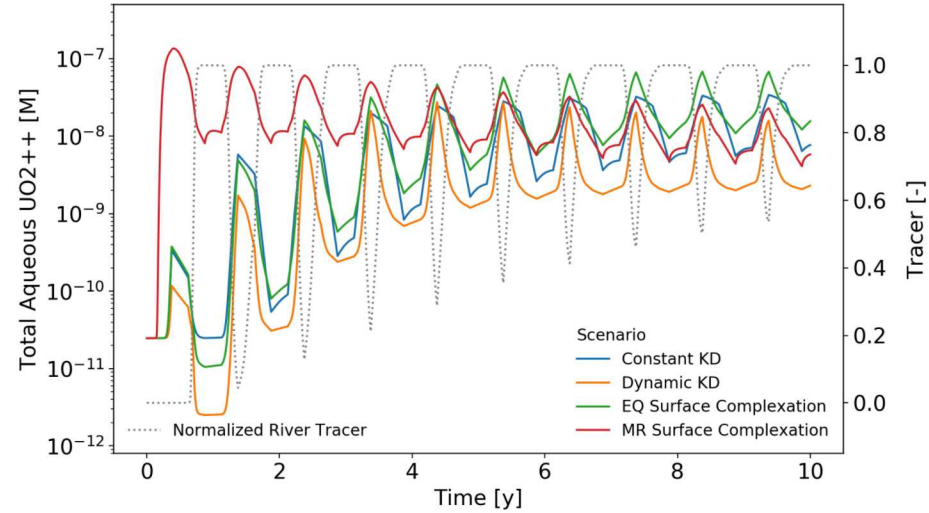
# Breakthrough of Aqueous and Sorbed $\text{UO}_2^{++}$ at Observation Points



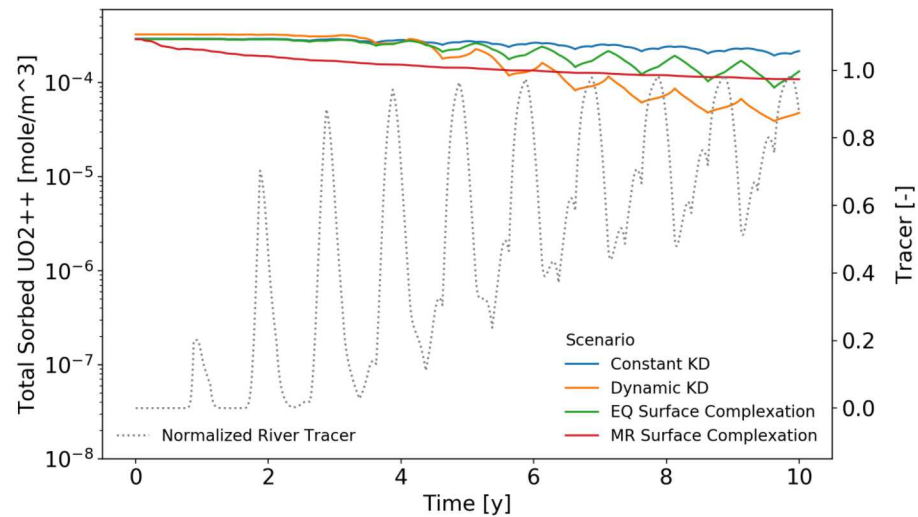
Total Aqueous  $\text{UO}_2^{++}$  at West Observation Point



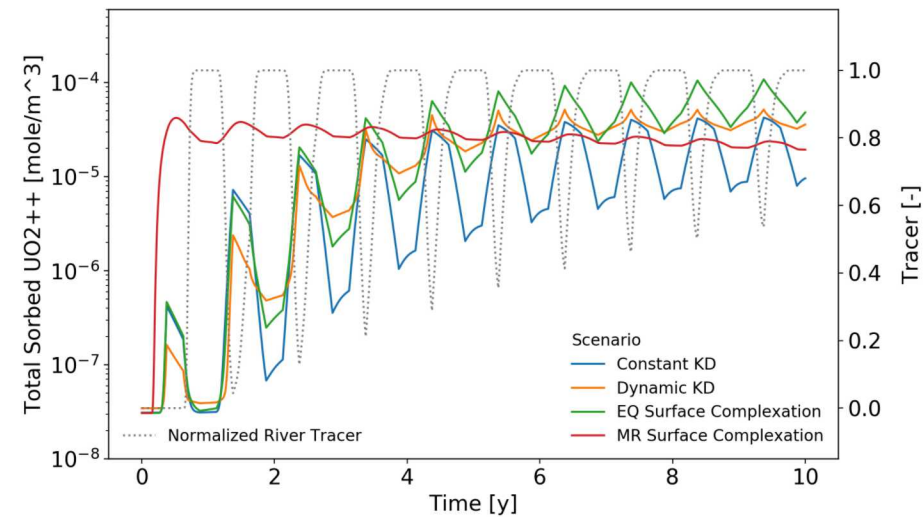
Total Aqueous  $\text{UO}_2^{++}$  at East Observation Point



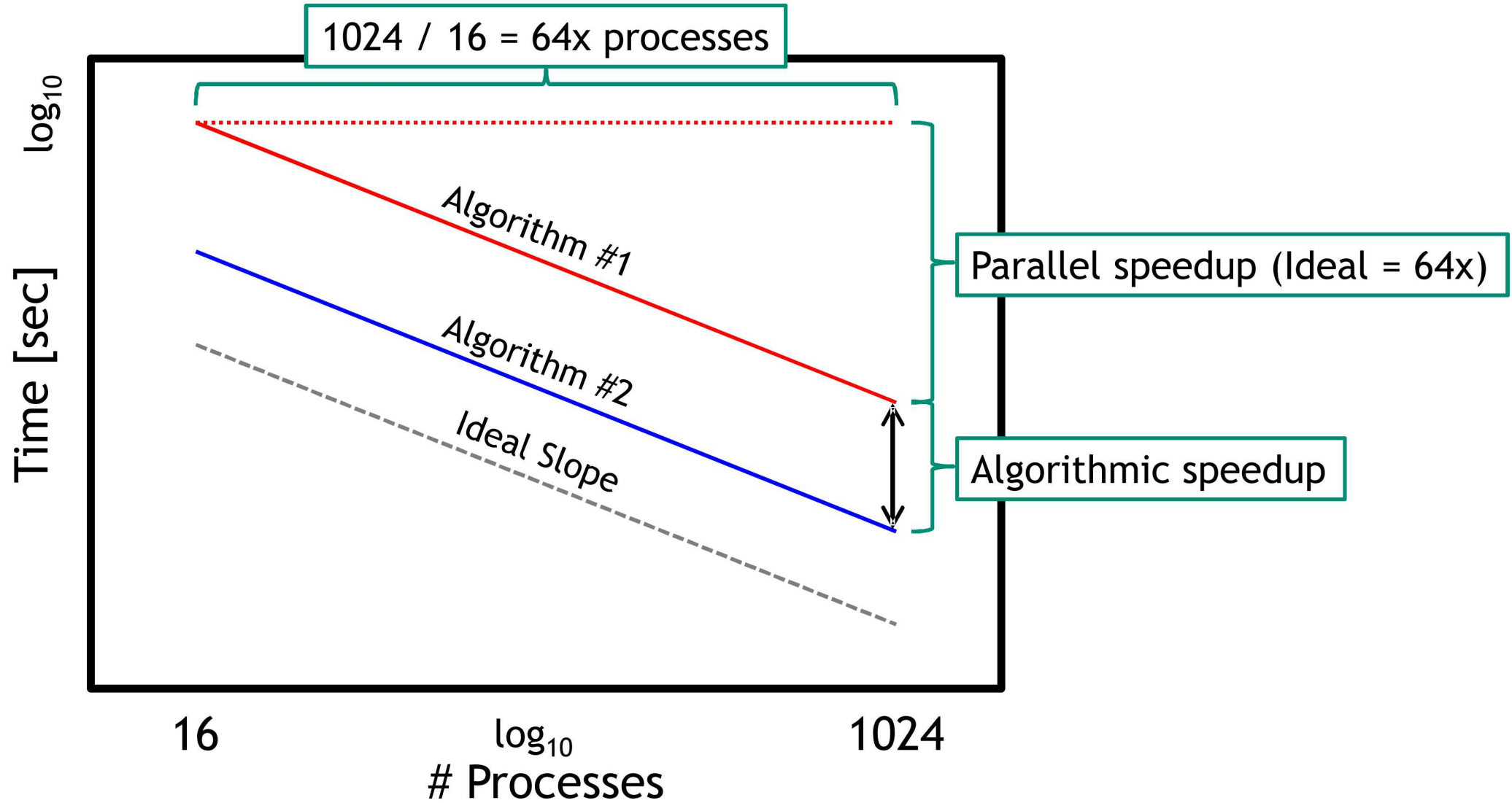
Total Sorbed  $\text{UO}_2^{++}$  at West Observation Point



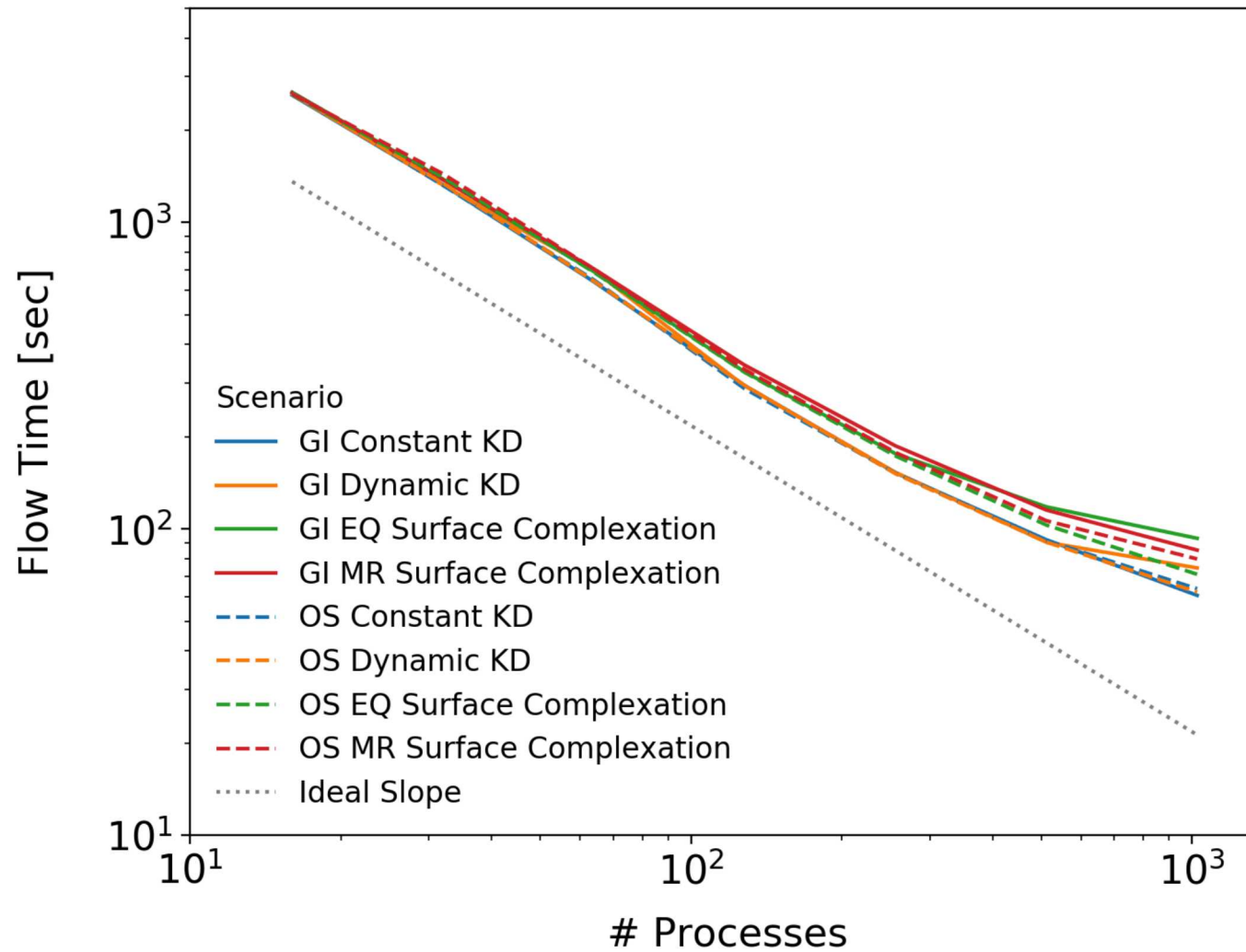
Total Sorbed  $\text{UO}_2^{++}$  at East Observation Point



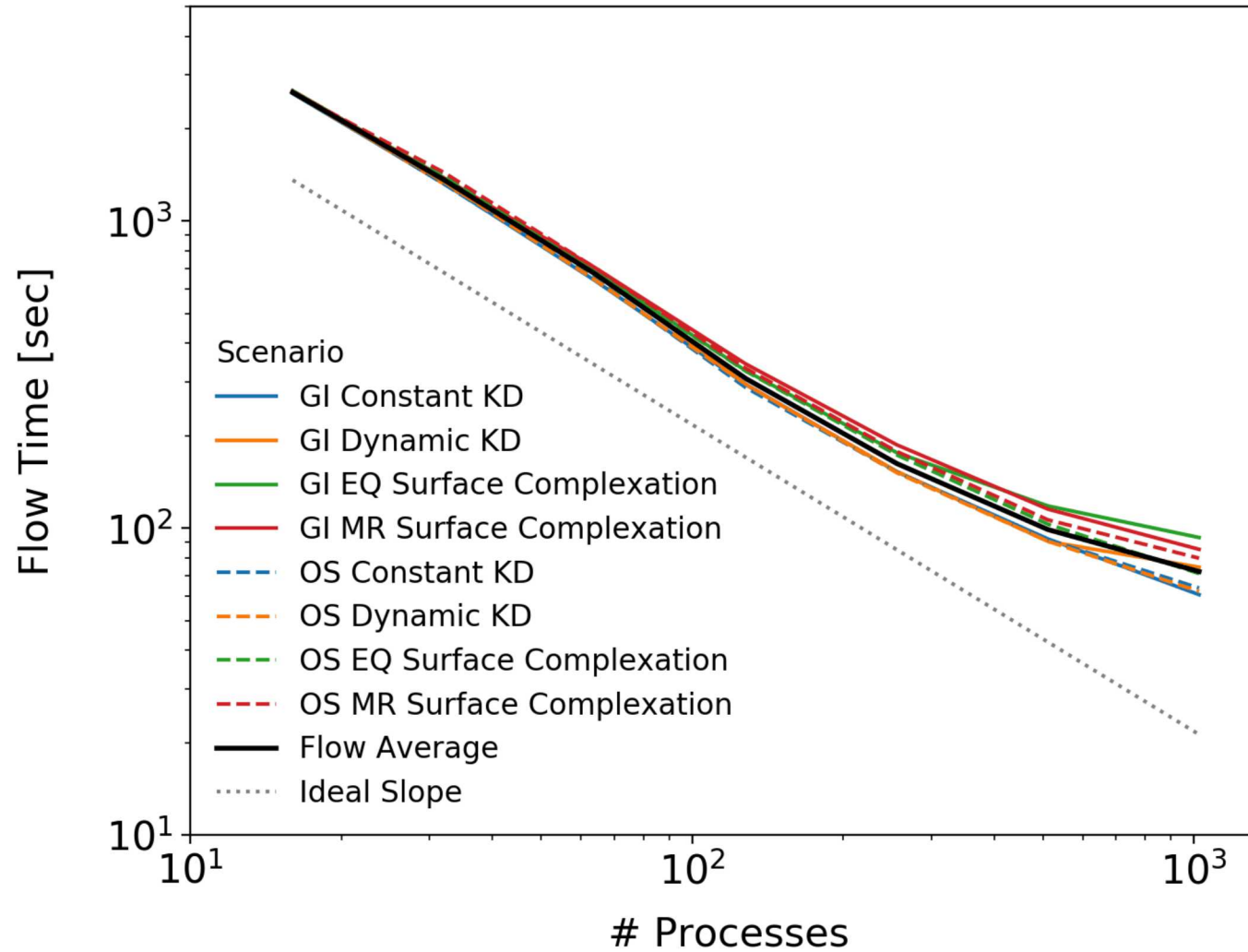
## Schematic of Speedup Results



## Flow in Heterogeneous Permeability Field

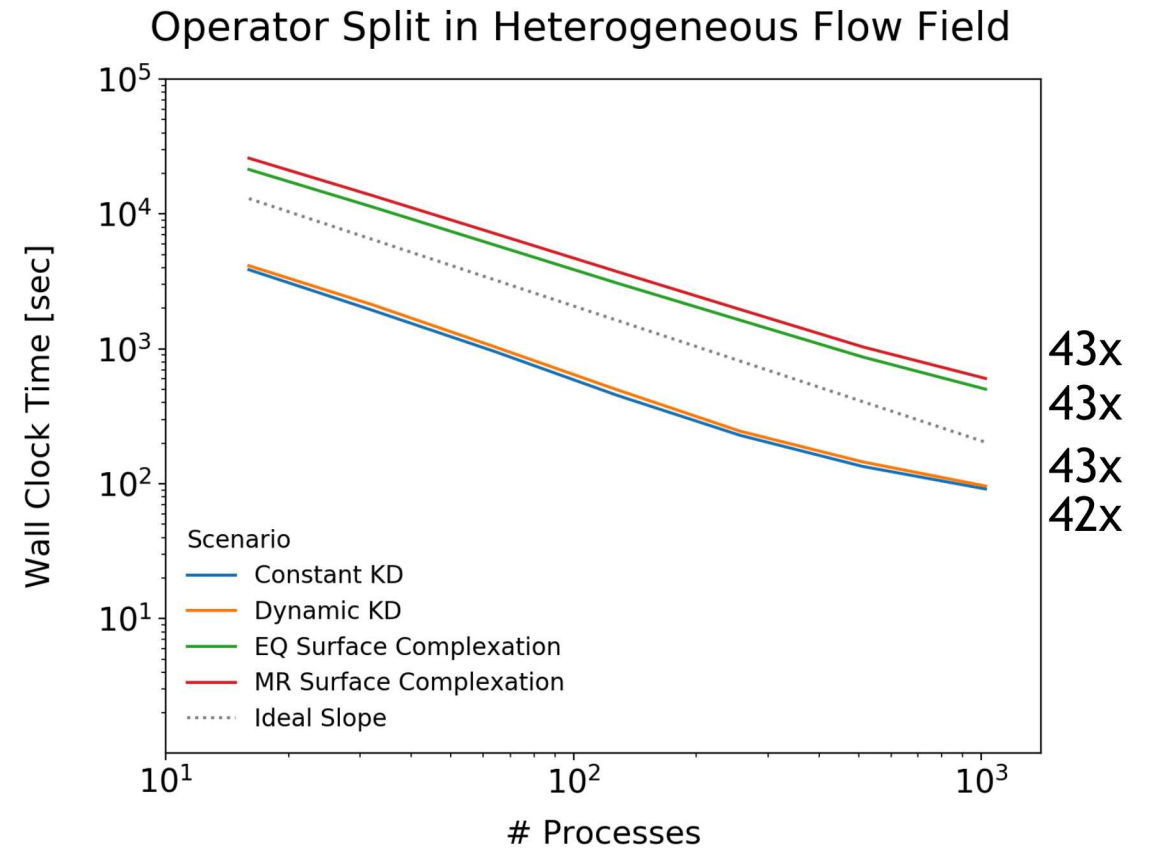
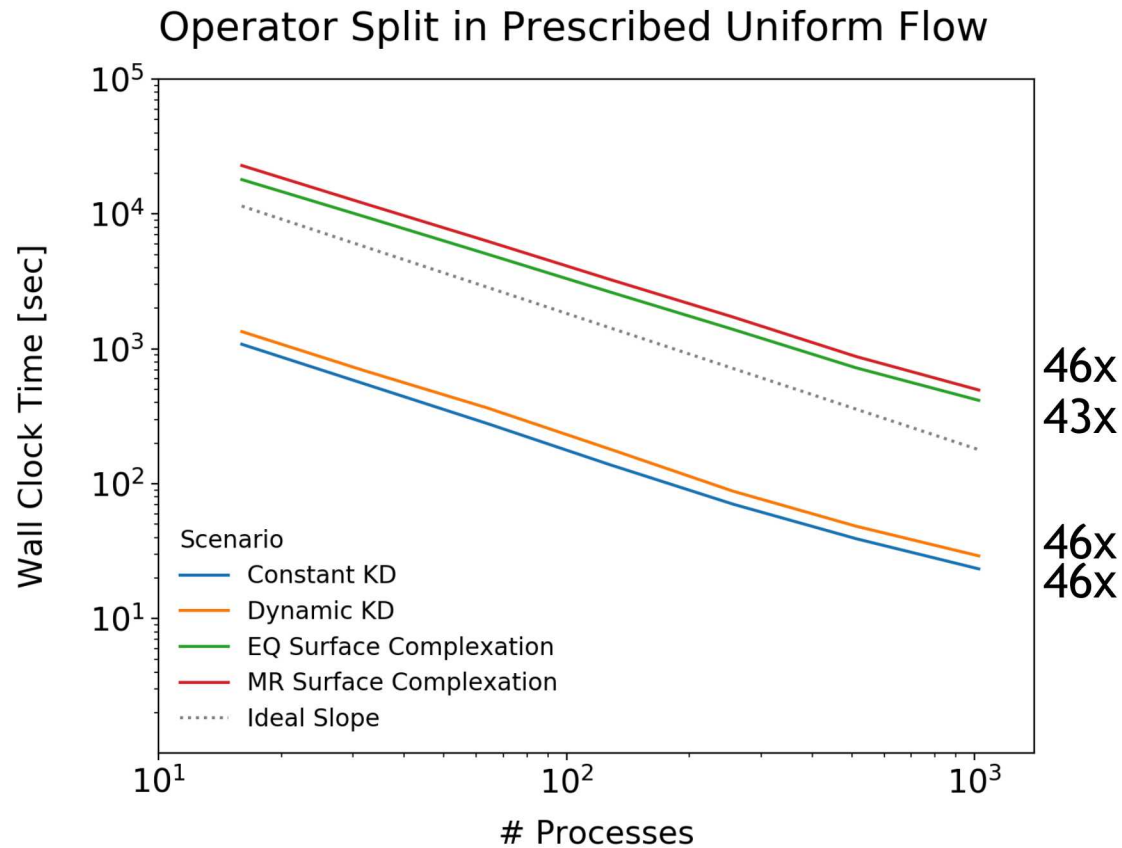


## Flow in Heterogeneous Permeability Field



# Bounds of Speedup Due to Parallel Processing: Operator Split

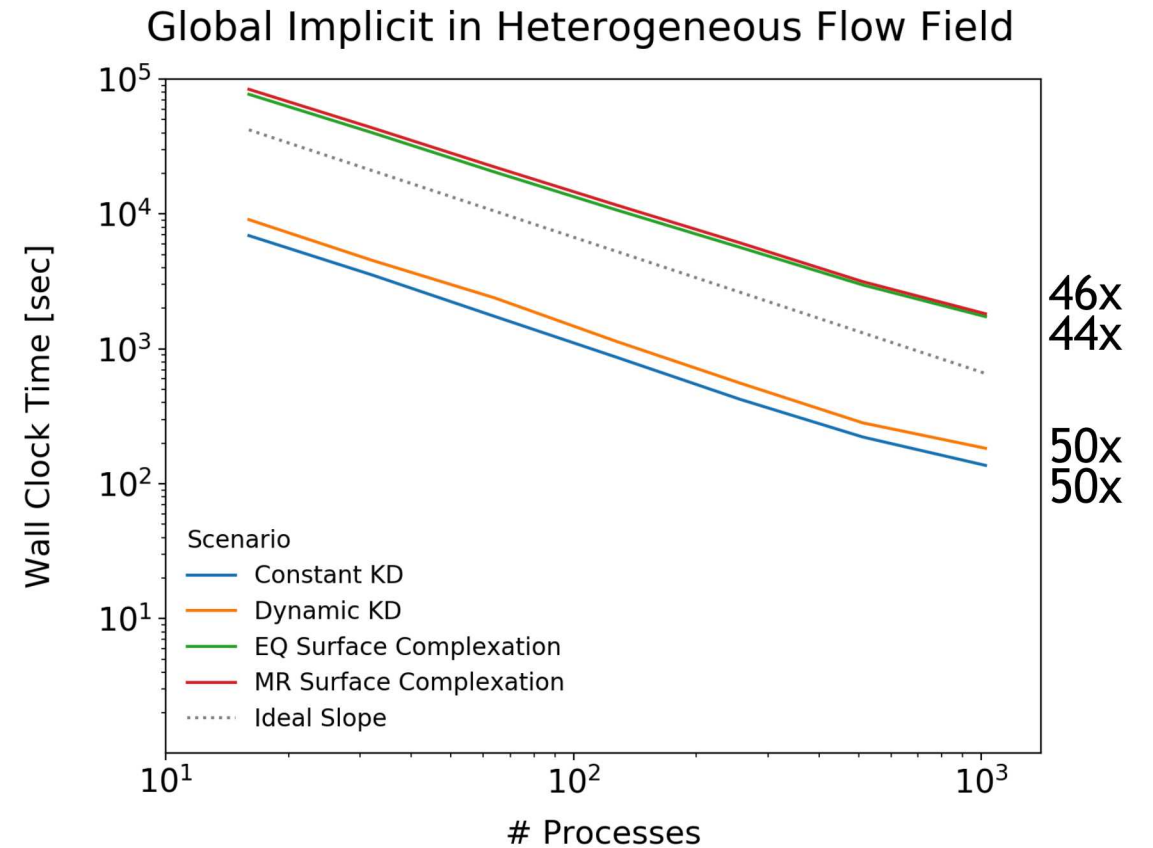
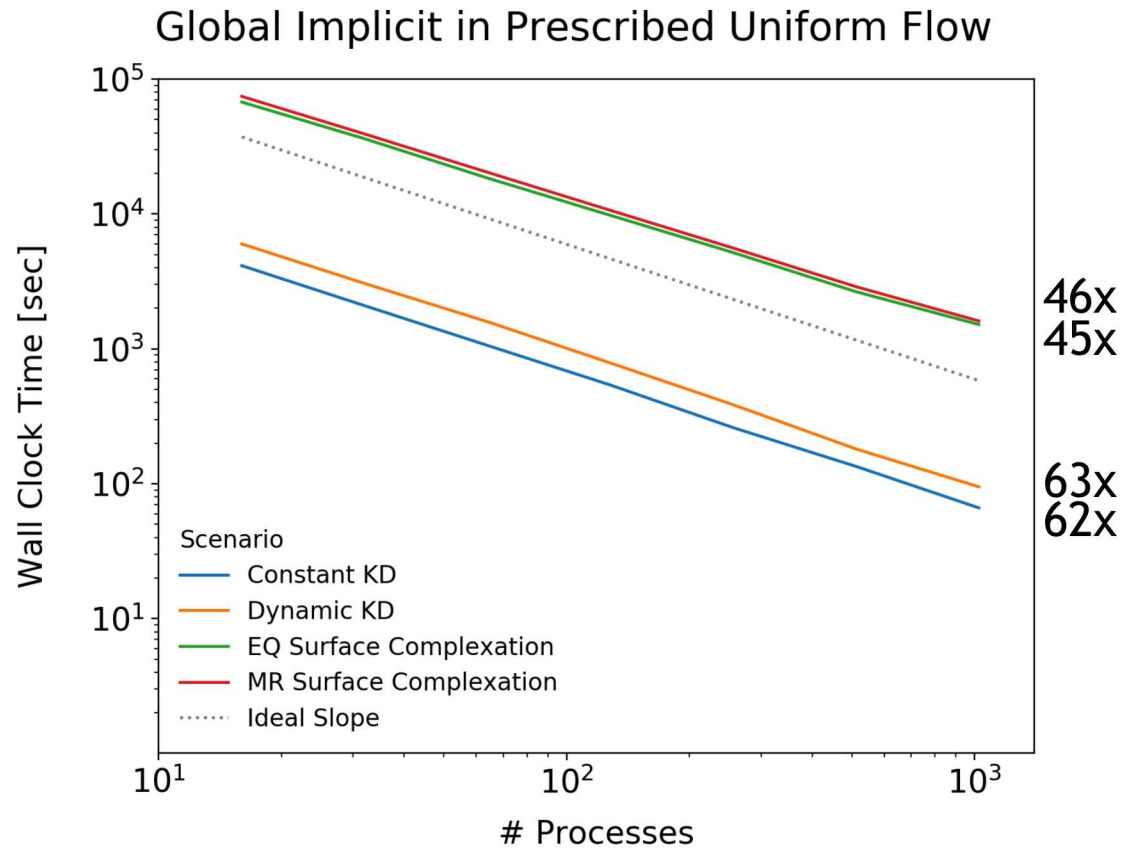
## Parallel Speedup



Ideal Speedup = 64x

# Bounds of Speedup Due to Parallel Processing: Global Implicit

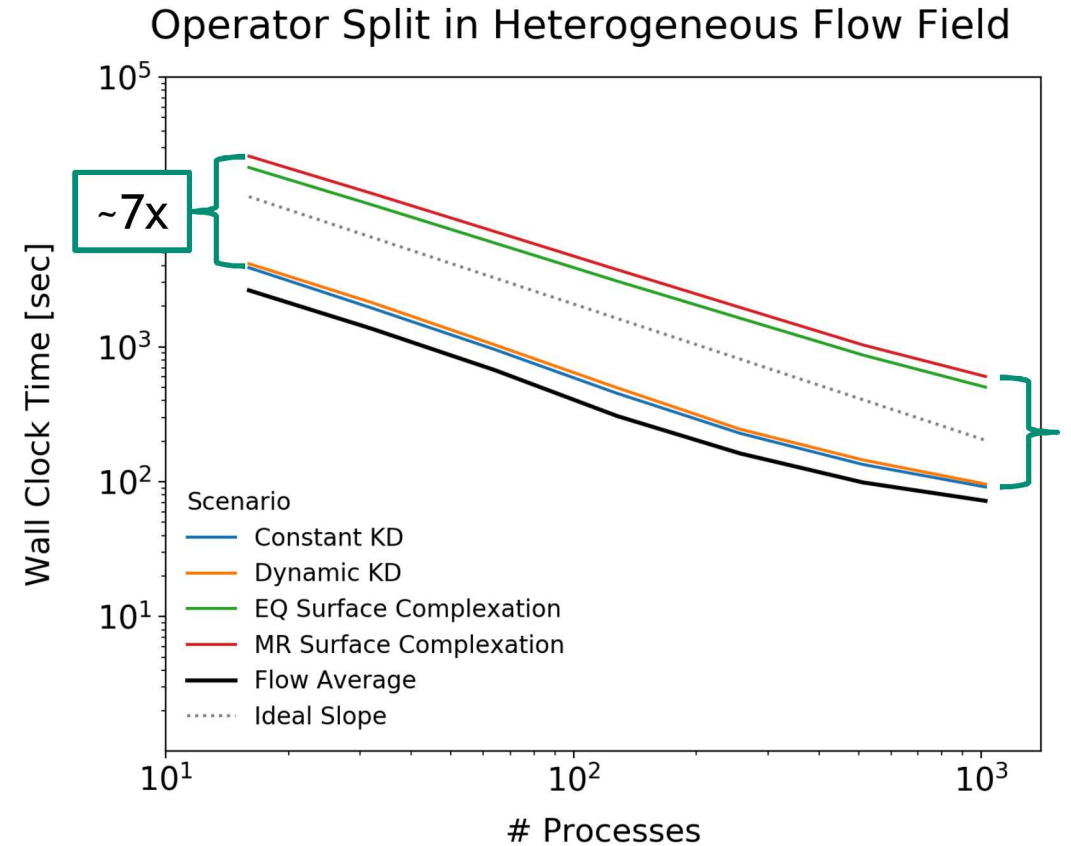
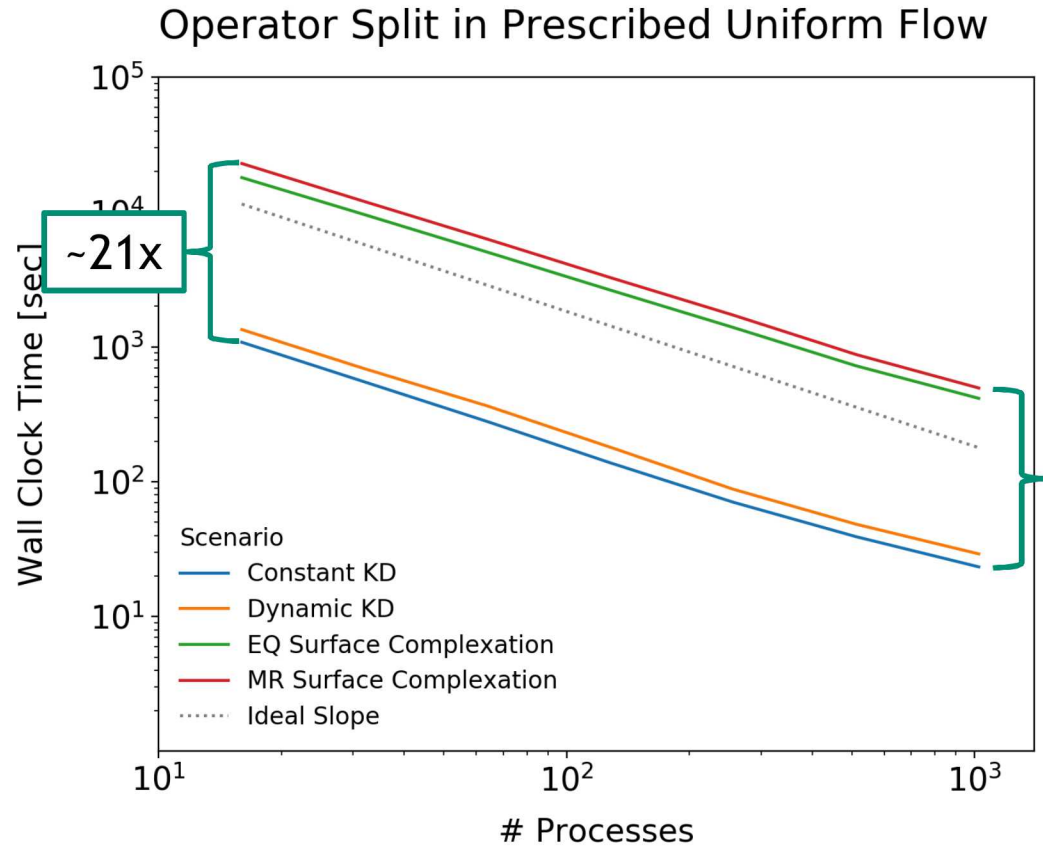
## Parallel Speedup



Ideal Speedup = 64x

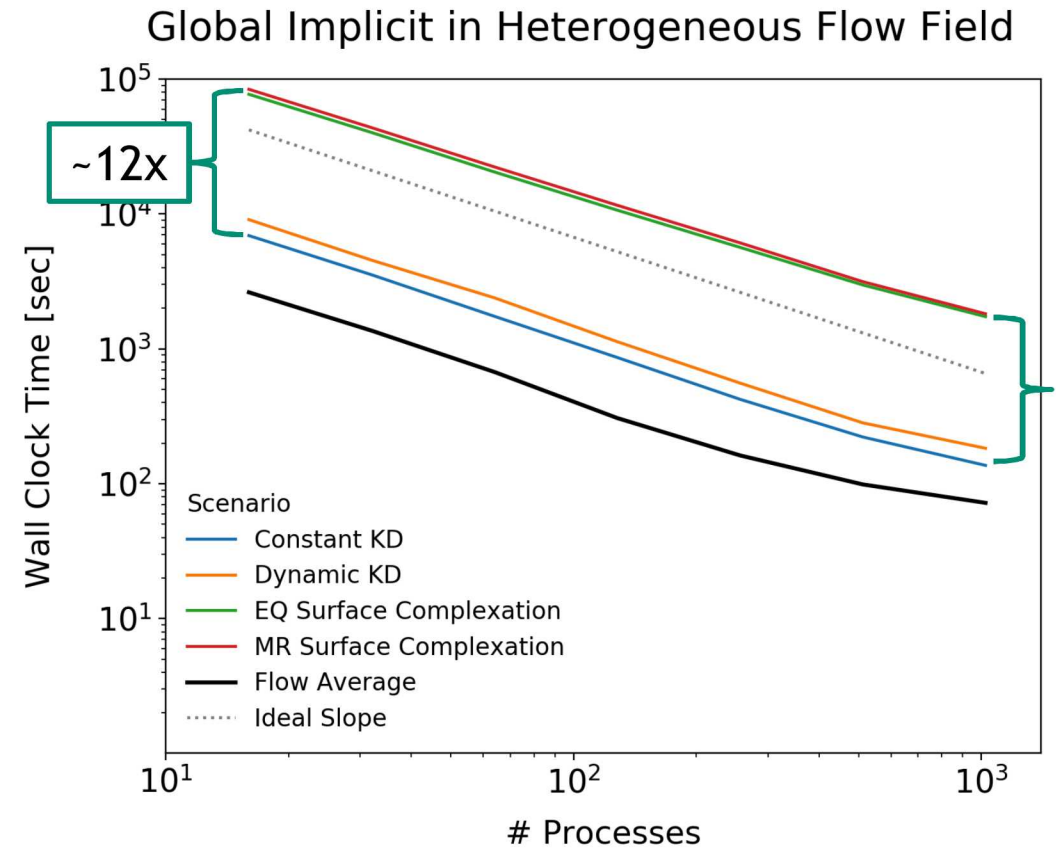
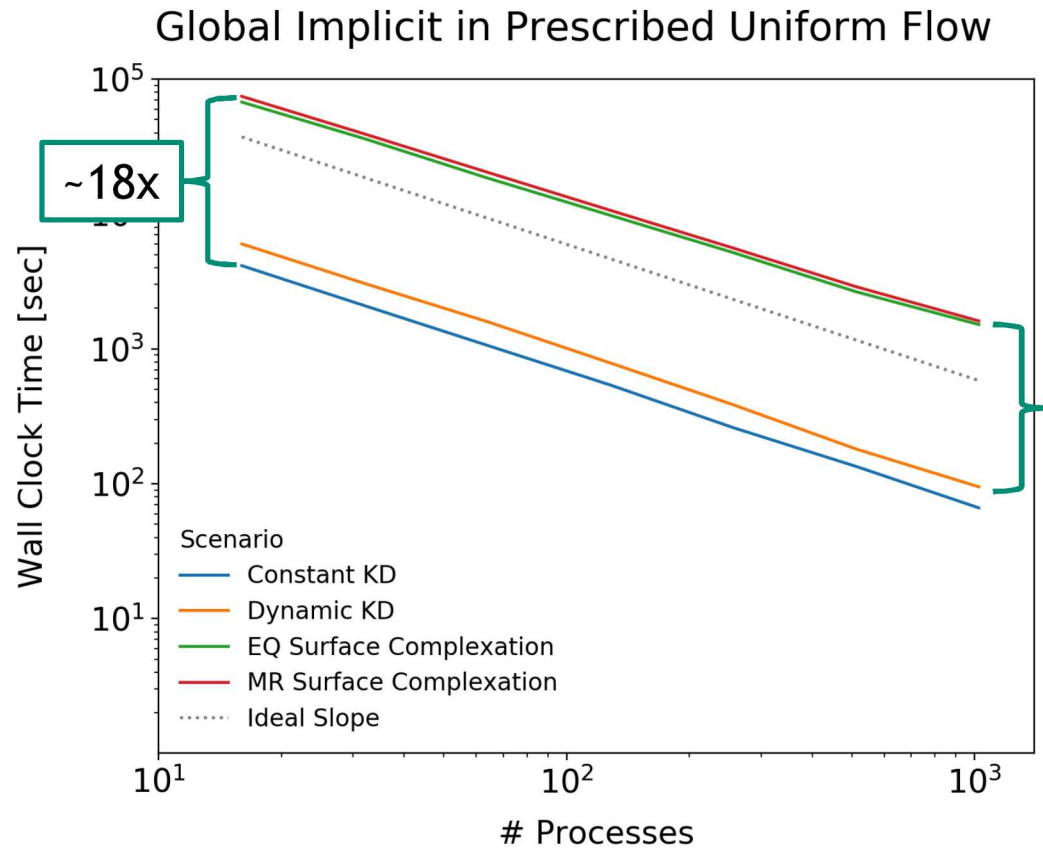
# Operator-Split: Prescribed Uniform Flow vs. Heterogeneous Flow Field

## Algorithmic Speedup

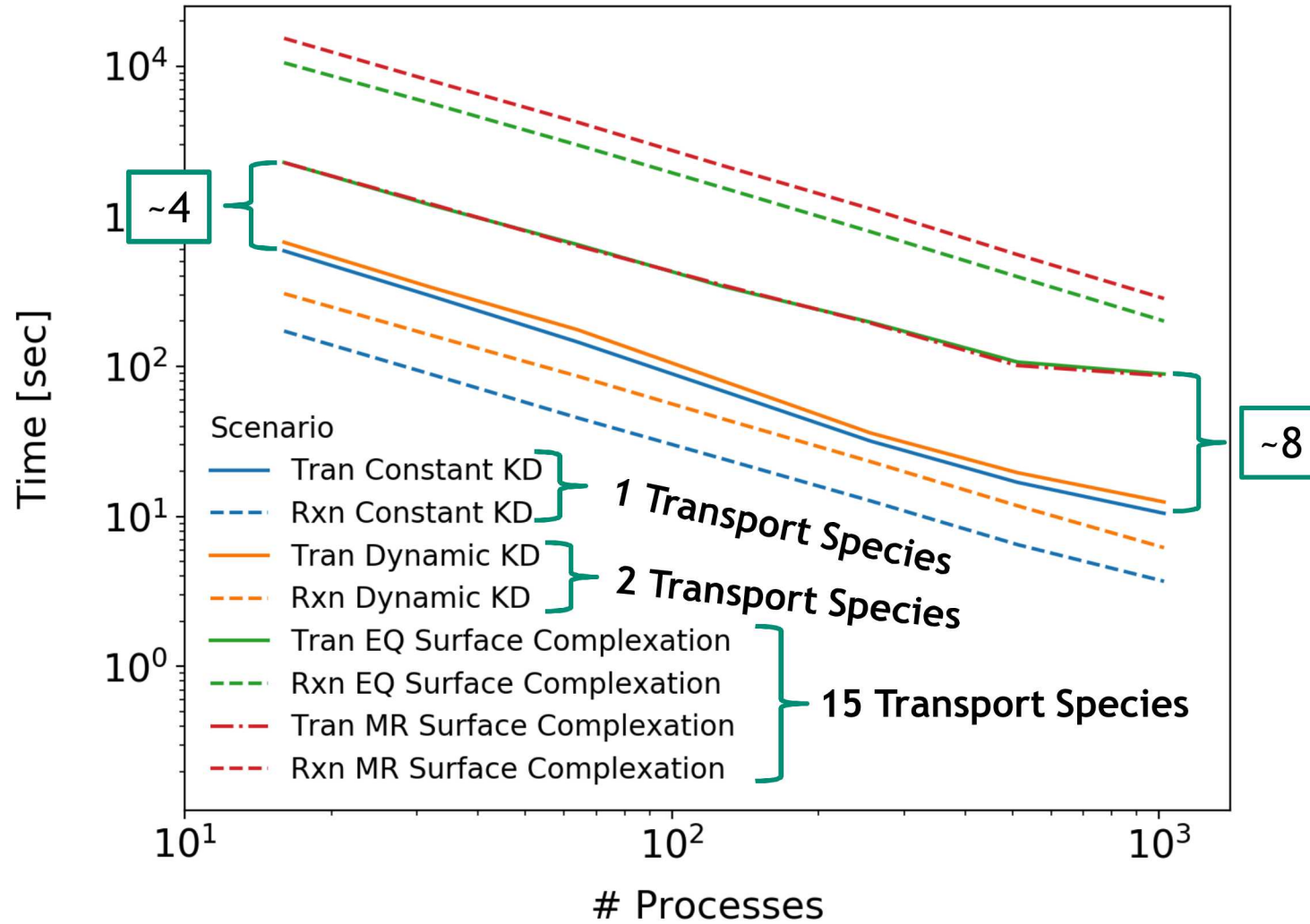


# Global Implicit: Prescribed Uniform Flow vs. Heterogeneous Flow Field

## Algorithmic Speedup



## OS Transport vs. Reaction in Prescribed Uniform Flow



$$T_{total} = T_{transport} + T_{reaction}$$

$$T_{reaction} \rightarrow 0: T_{min} = T_{transport}$$

$$T_{reaction} = 0.9 T_{total}$$

$$T_{min} = T_{transport} = 0.1 T_{total}$$

**Maximum Possible Speedup = 10x**

## Conclusions

For representative surface complexation problem ( $\sim 1\text{M}$  grid cells  $\times$  15 chemical components =  $\sim 15.7\text{M}$  dof total)

Maximum speedup (efficiency) due to **parallel processing**:

Ideal = 64 (100%)

	Operator Split	Global Implicit
Specified Uniform Flow	46.3 (72%)	63.1 (99%)
Heterogeneous Flow Field	43.0 (67%)	50.5 (79%)

Maximum speedup due to **simplified chemistry**:

	Operator Split	Global Implicit
Specified Uniform Flow	$\sim 21$	$\sim 18$
Heterogeneous Flow Field	$\sim 7$	$\sim 12$

Maximum speedup due to **both** parallel processing and simplified chemistry

	Operator Split	Global Implicit
Specified Uniform Flow	977	1122
Heterogeneous Flow Field	283	613