



# Perovskite Oxides for Solar Air Separation

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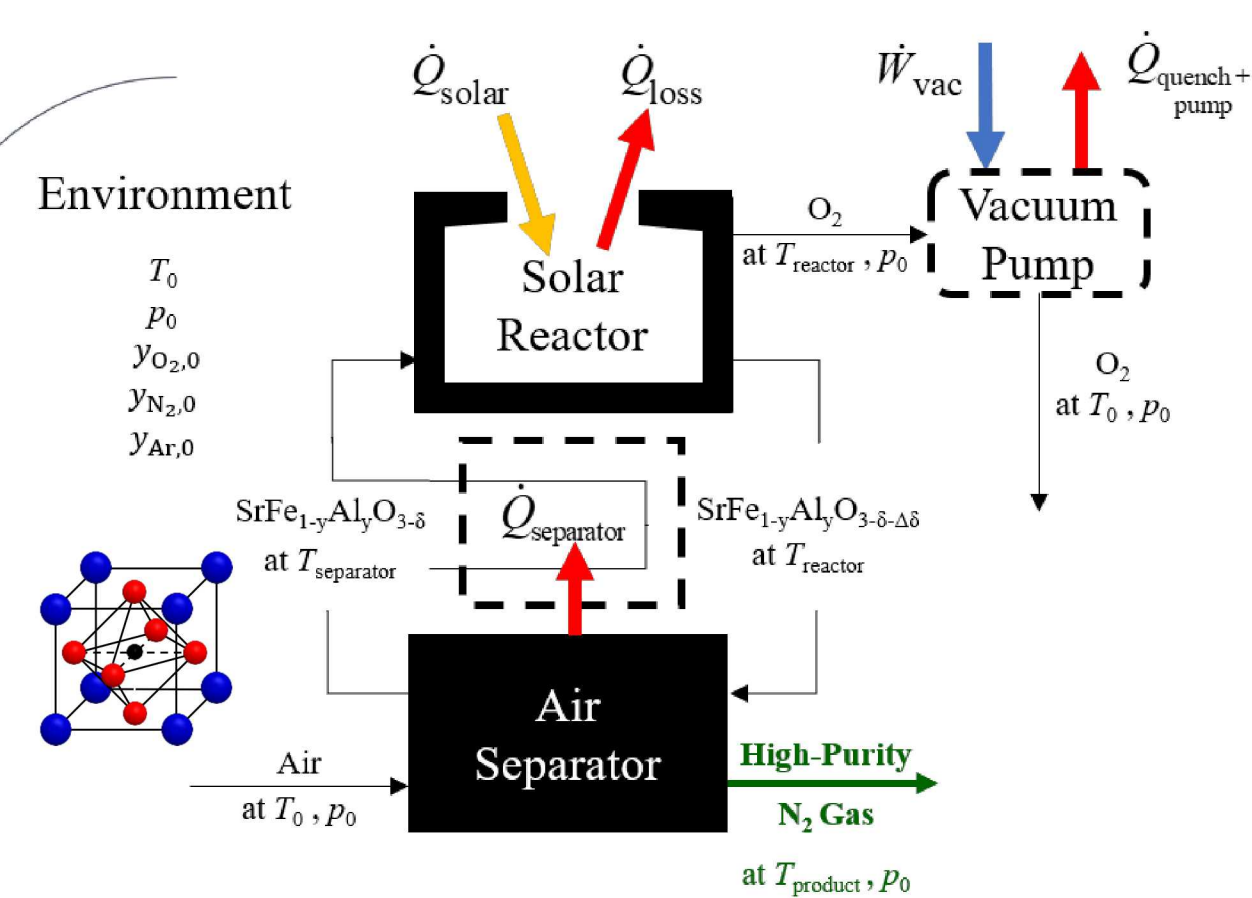
## INTRODUCTION

### Conventional NH<sub>3</sub> Production

- Haber Bosch process is currently used to produce NH<sub>3</sub> fertilizer for agriculture
- Energy intensive: chemical reactions require high temperatures and pressures
- Associated CO<sub>2</sub>: 1% of global GHG emission

### Proposed Alternative Cycle

- Solar reduction-oxidation (redox) cycles to separate pure N<sub>2</sub> from air
- N<sub>2</sub> reacted with H<sub>2</sub> to produce NH<sub>3</sub>
- Low/no CO<sub>2</sub> emissions
- SrFeO<sub>3-δ</sub>: appealing perovskite oxide for air separation; Al-doped form (SAF) studied



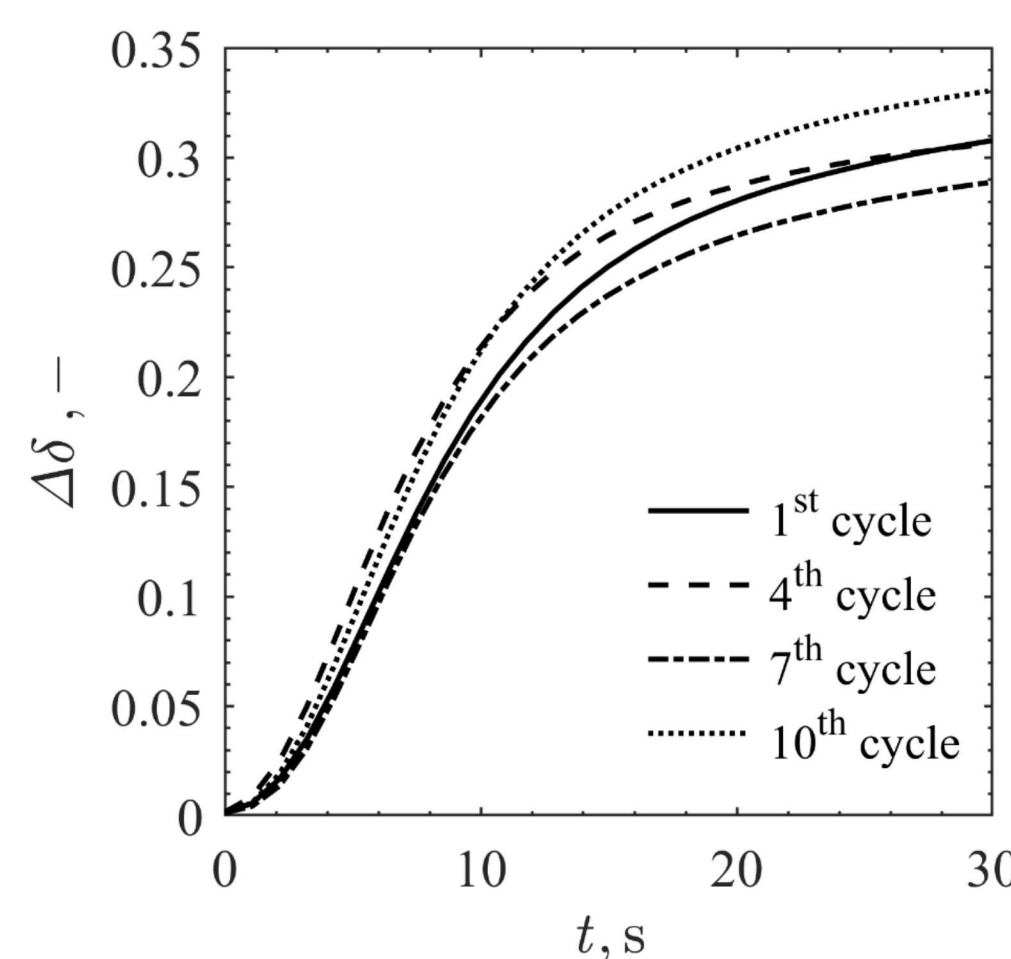
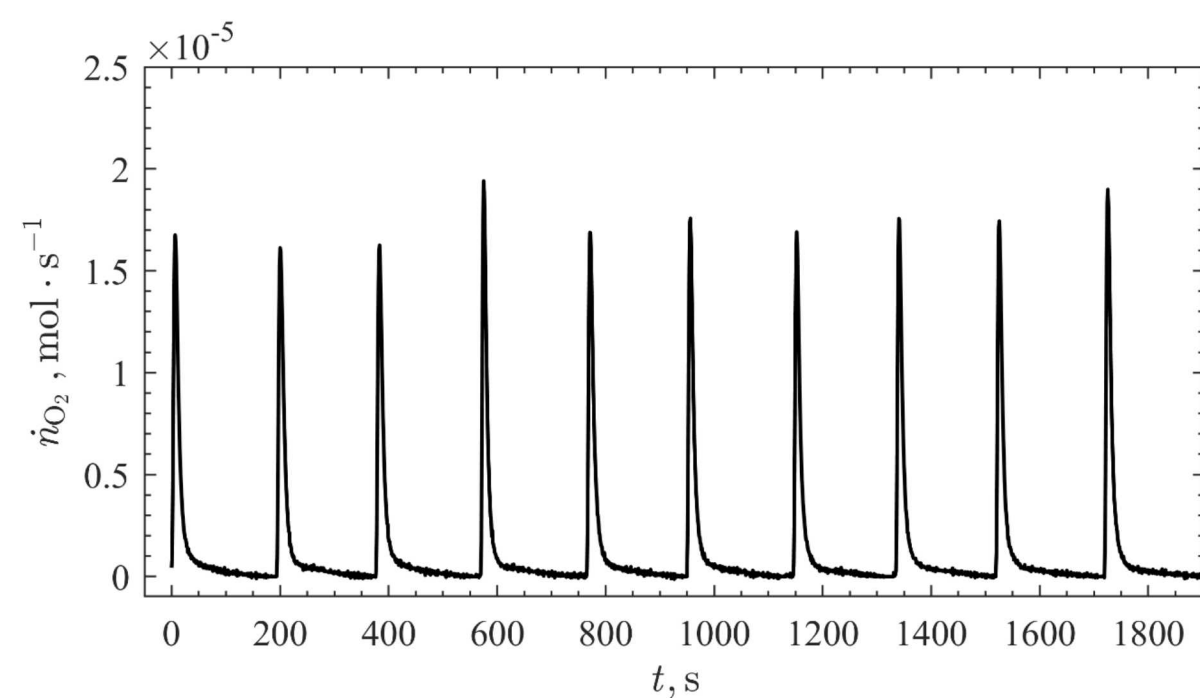
## KINETIC STUDIES

### Experimental Method

- On-sun reduction at low vacuum (Ar), off-sun oxidation (O<sub>2</sub>) cycles
- Radiative input from high flux solar simulator (HFSS)
- Rapid initial heating > 50 K/s for kinetic studies
- Upward flow reactor (UFR) with optimized gas transport
- Reduction monitored via MS and GC measurements of O<sub>2</sub> flow

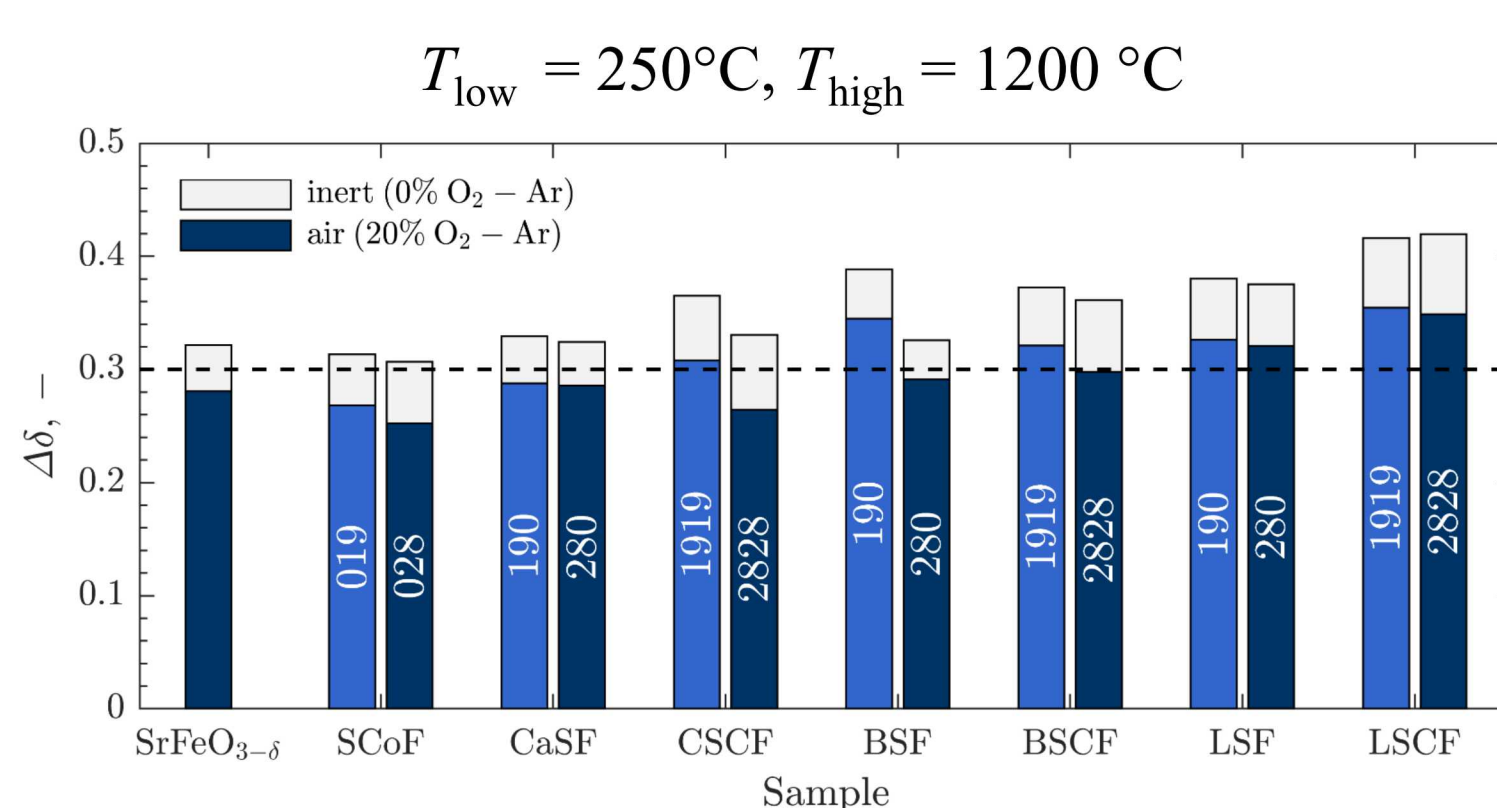
### Redox Performance

- Very high redox extents reached,  $\Delta\delta > 0.30$  for multiple cycles
- Rapid kinetics, equilibrium attained within approximately 30s
- No degradation or significant  $\Delta\delta$  variation over ten redox cycles

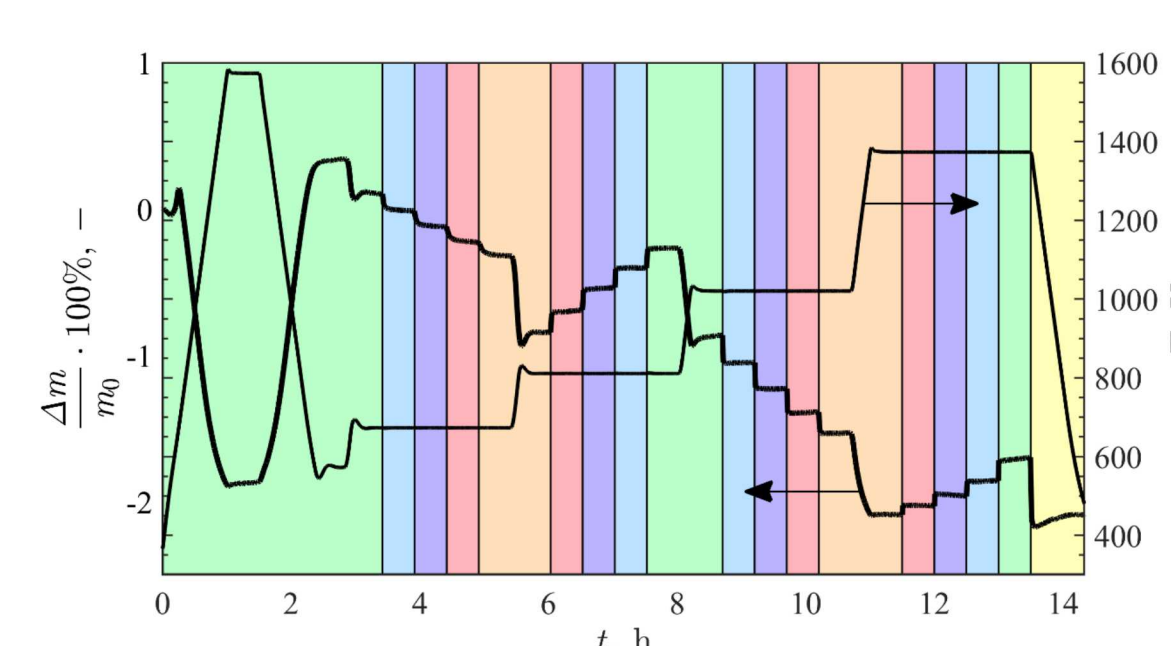


## NEXT STEPS

- Broad material screening study performed of substituted SrFeO<sub>3-δ</sub>
- Goal: identifying optimal materials for air separation
- Several compositions identified that exceed SrFeO<sub>3-δ</sub> performance
- Detailed thermodynamic and kinetic characterization underway
- Novel N<sub>2</sub>-carrier materials for NH<sub>3</sub> formation currently in development



## THERMODYNAMIC MODELING

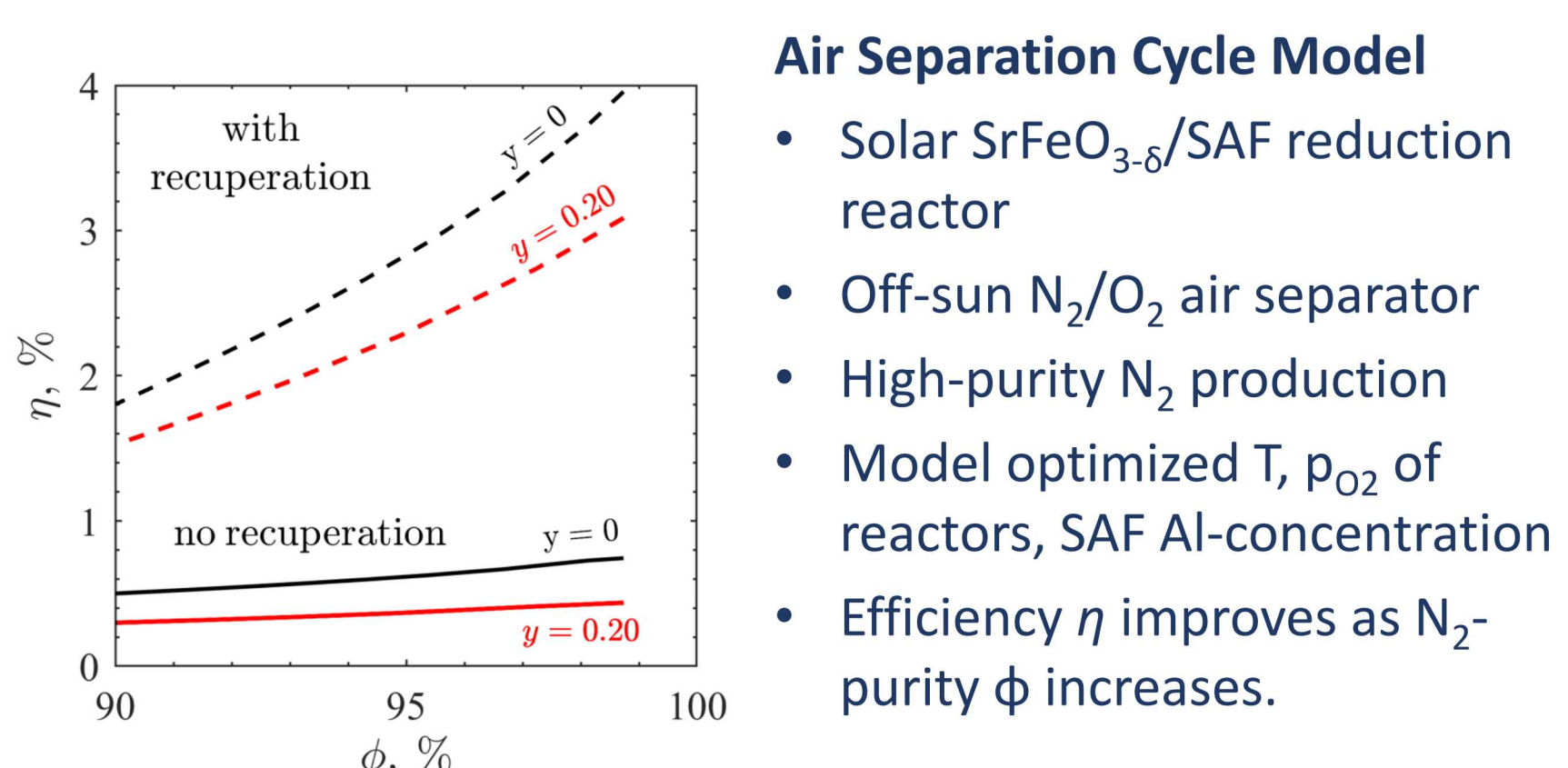
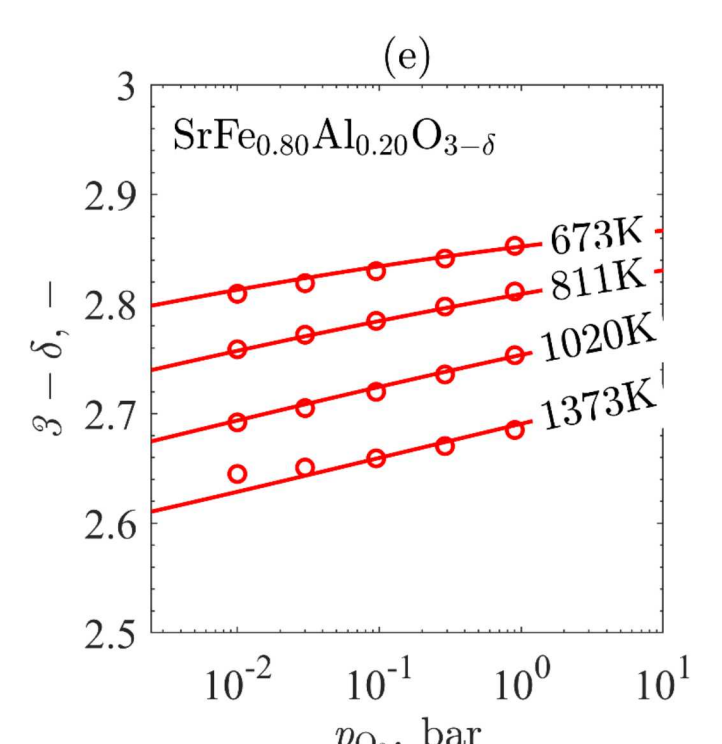
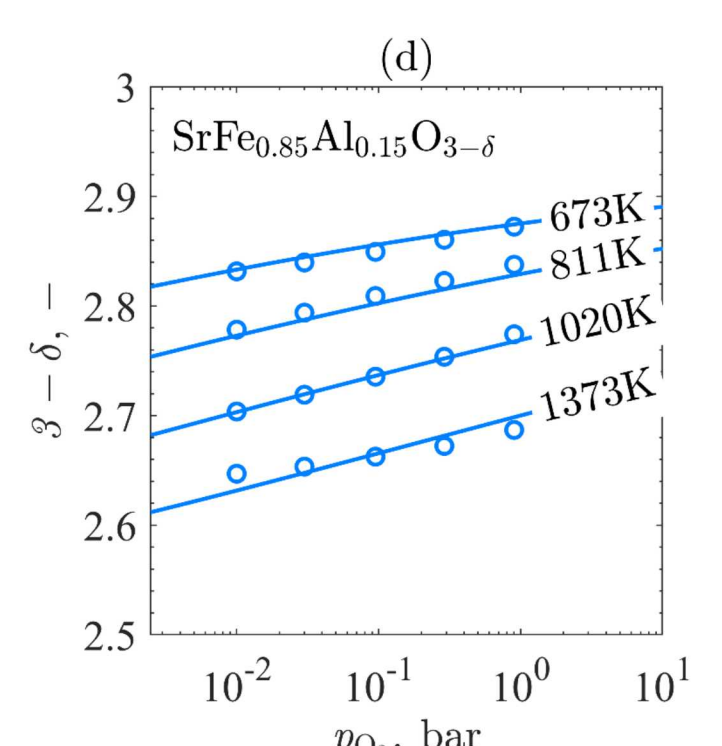
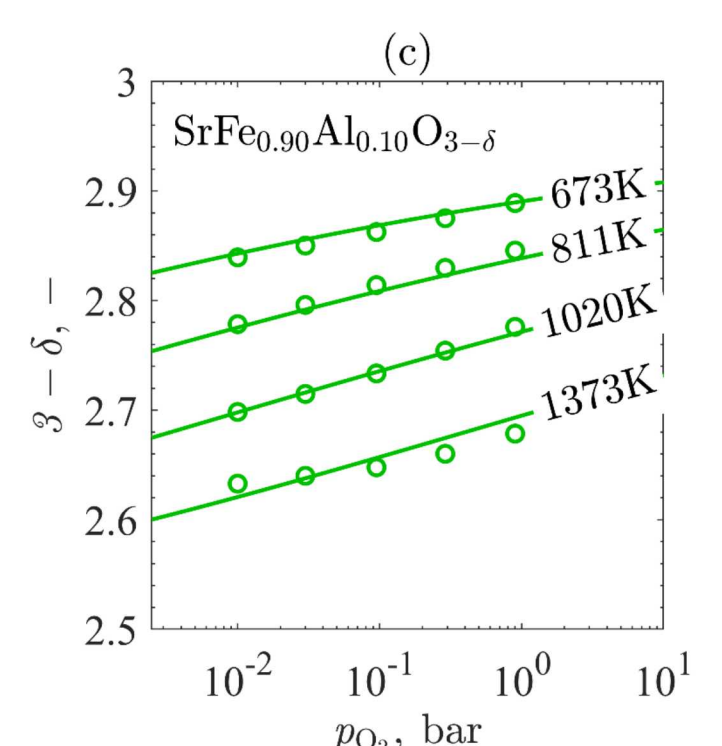
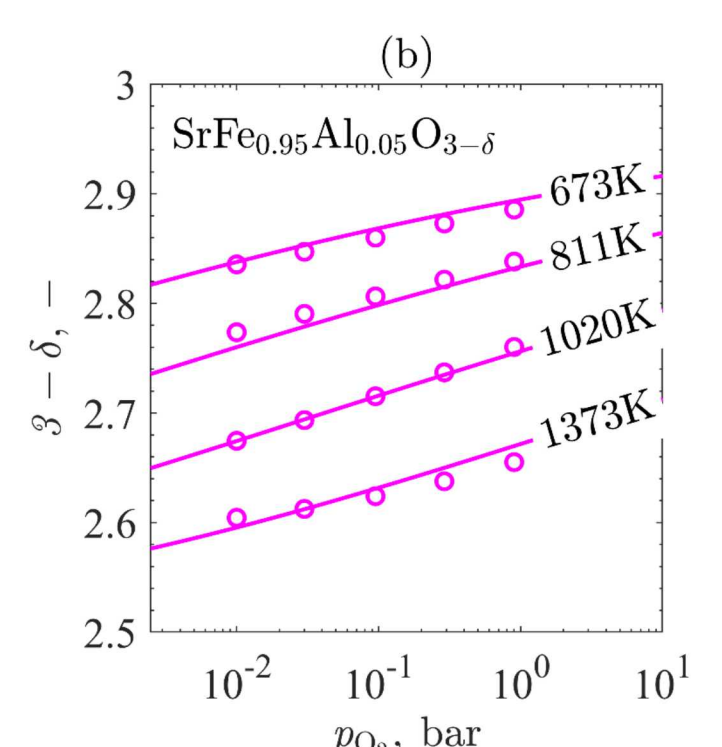
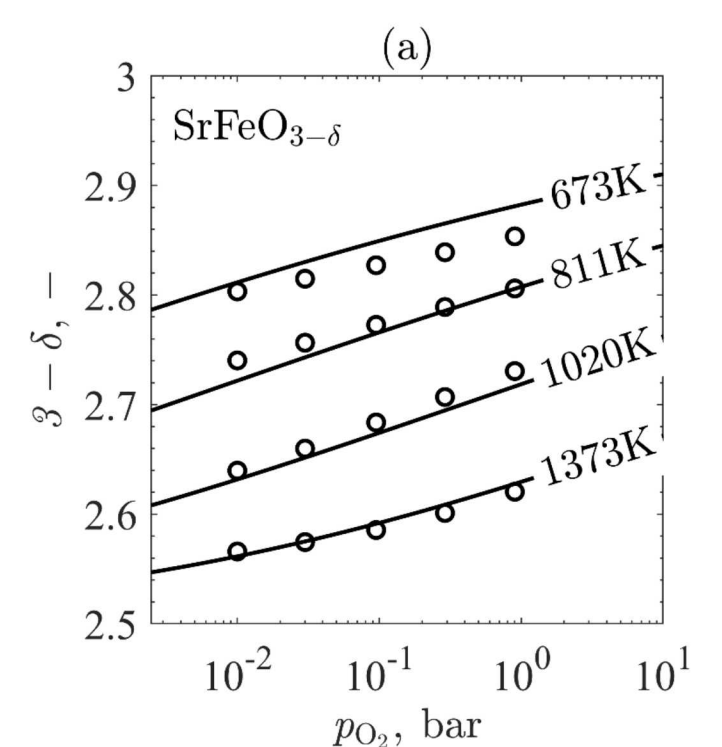
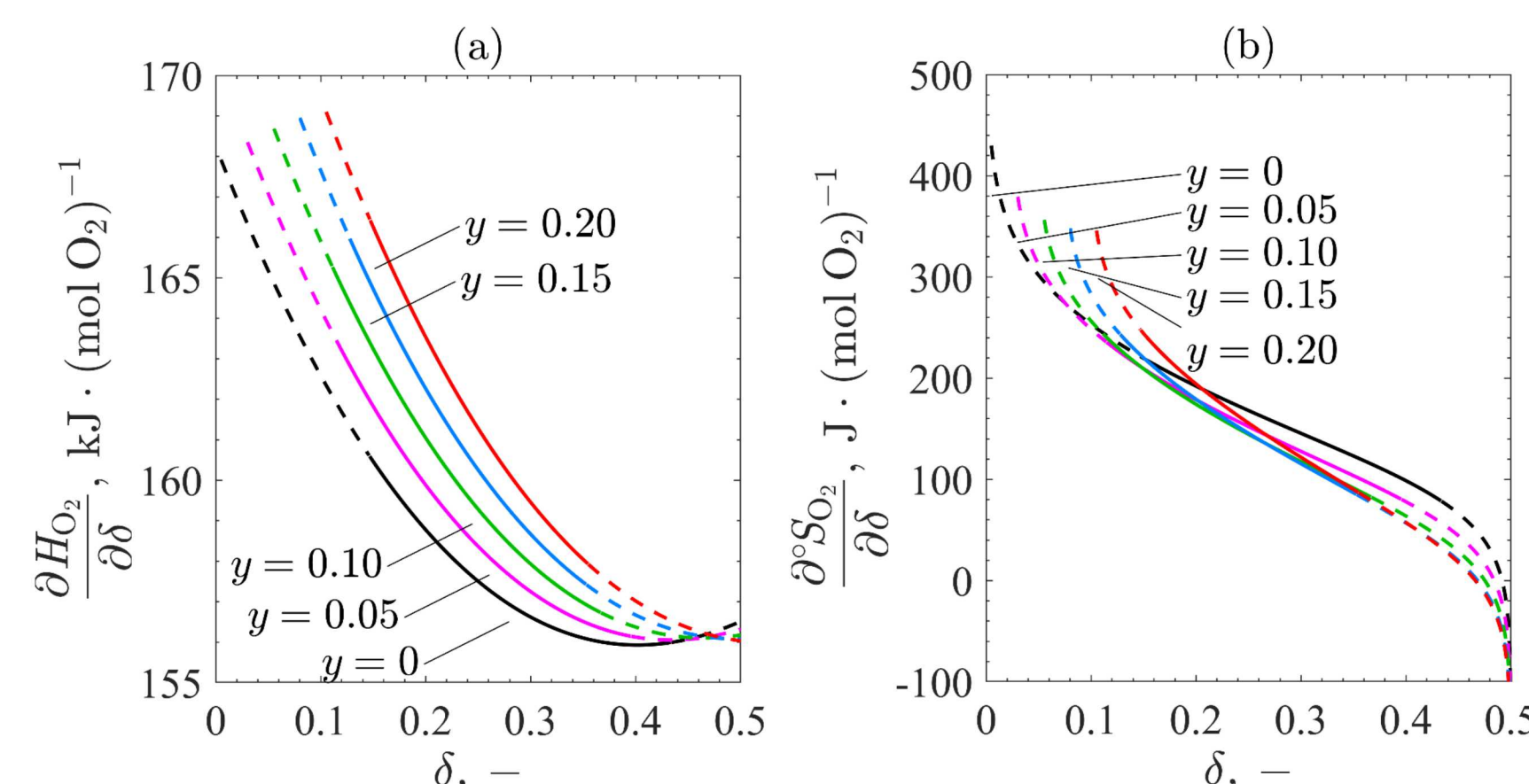


### Experimental Method

- Reduction-oxidation equilibria experiments performed in a thermogravimetric analyzer (TGA) in O<sub>2</sub>/Ar
- Temperatures from 673 to 1373 K, O<sub>2</sub> pressures from 0.01 to 0.90 bar, Al-dopant concentrations from 0 to 20%

### Compound Energy Model

- SAF modeled as weighted sum of end members, with added terms for mixing entropy and sublattice interactions
- Nonlinear  $\Delta G_{\text{redox}}$  minimization used to fit TGA experiments to predict redox extent  $\delta = f(T, p_{\text{O}_2}, y)$
- Predictions of  $\delta$  and redox enthalpy were consistent with experiments and density functional theory predictions
- Interaction terms were highly significant, consistent with vacancy ordering previously observed for SrFeO<sub>3-δ</sub>



### Air Separation Cycle Model

- Solar SrFeO<sub>3-δ</sub>/SAF reduction reactor
- Off-sun N<sub>2</sub>/O<sub>2</sub> air separator
- High-purity N<sub>2</sub> production
- Model optimized T, pO<sub>2</sub> of reactors, SAF Al-concentration
- Efficiency  $\eta$  improves as N<sub>2</sub>-purity  $\phi$  increases.

## REFERENCES

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- Bush, H.E., et al., *Design and Characterization of a Novel Upward Flow Reactor for the Study of High-Temperature Thermal Reduction for Solar-Driven Processes*. Journal of Solar Energy Engineering, 2017. **139**(5): p. 051004-051004-11.
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- Babiniec, S.M., et al., *Investigation of La<sub>x</sub>Sr<sub>1-x</sub>Co<sub>y</sub>M<sub>1-y</sub>O<sub>3-δ</sub> (M = Mn, Fe) perovskite materials as thermochemical energy storage media*. Solar Energy, 2015. **118**: p. 451-459.