

Development of Nanostructured and Surface Modified Semiconductors for Hybrid Organic-Inorganic Solar Cells

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Funding

Sandia LDRD & IC Post Doc Fellowship

Collaborator

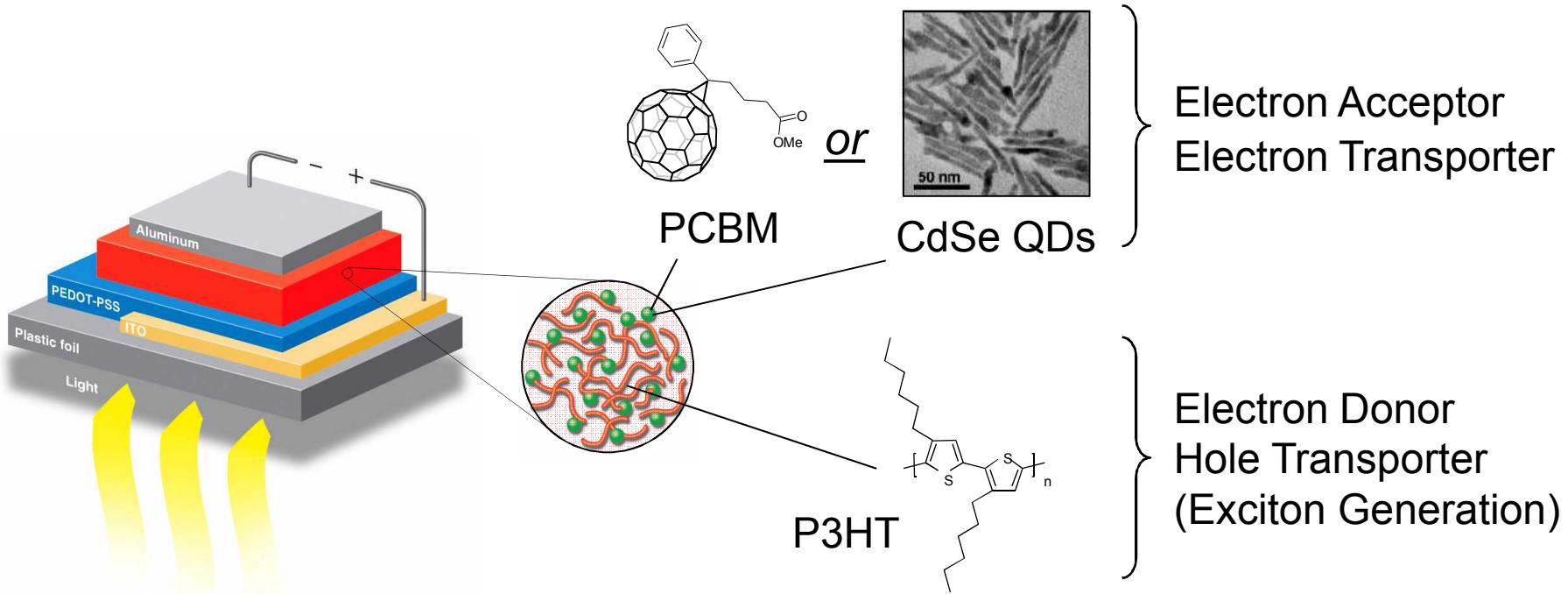
NREL - (funded by DDRD) Dave Ginley



Interests & Goals for Organic and Hybrid PV

- **Low cost photovoltaic devices**
 - low temperature & non-vacuum processing
 - Printed at high speed on flexible substrates
 - Using roll-to-roll processing
 - Low installation and system cost
- **Near term: 5 - 10% efficiency, lifetime 10,000 hrs (currently 2%)**
 - “Niche” applications in consumer electronics
 - DOD: portable power, field deployed electronics, “future force warriors”
- **Long term: 15% efficiency, lifetime > 3 - 5 yrs**
 - Roof top power generation

Hybrid, Heterojunction-Based Solar Cells

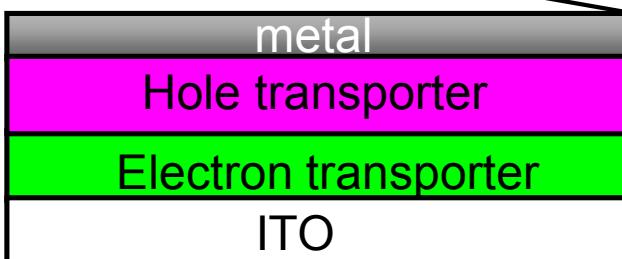


Project Objectives

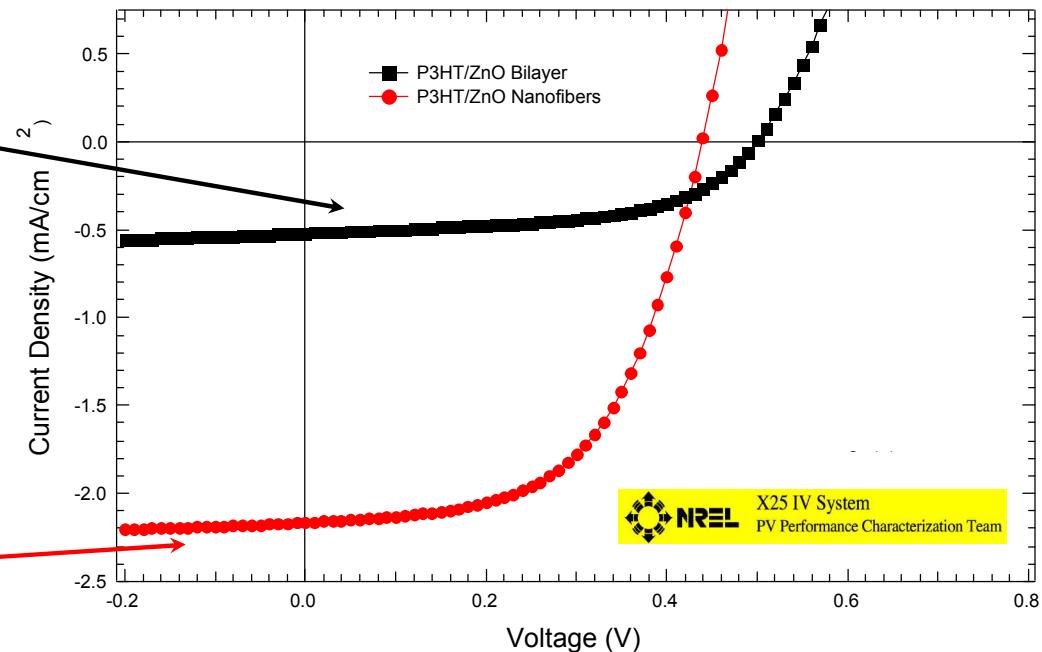
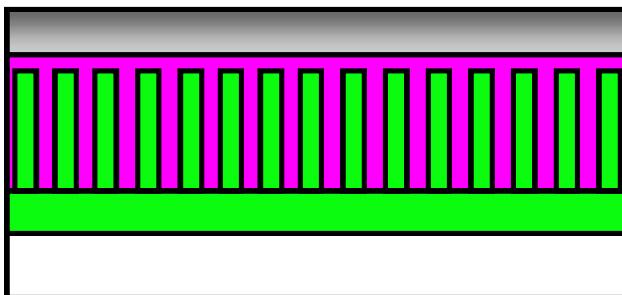
- *Gain* expertise in the “nano-engineering” of controlled oxide growth, oxide/polymer assembly, and interface control as applied to hybrid organic-inorganic solar cells.
- *Develop* alternative power generation technologies to support Sandia initiatives.
- *Position* Sandia for future funding and partnership opportunities in the photovoltaic arena.

Why Nanostructures?

Bilayer

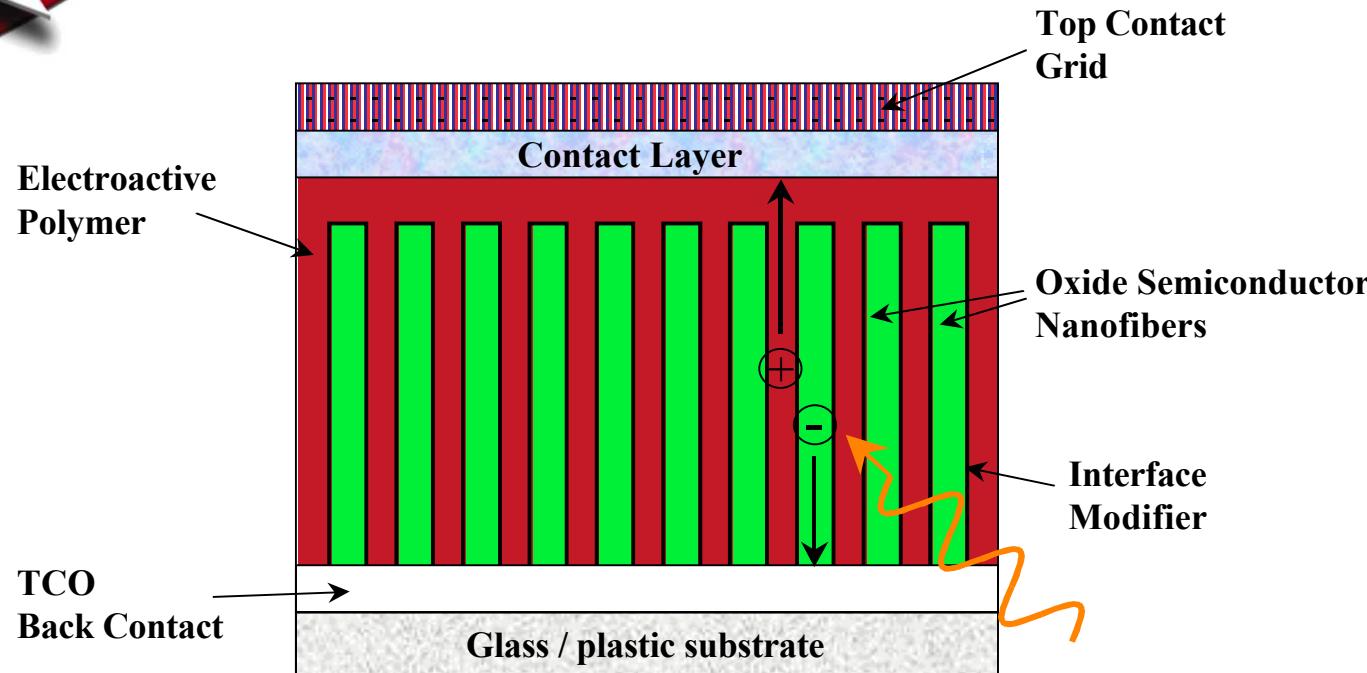


Nanostructures



- Higher interfacial area
- Spacing ~ exciton diffusion length
- Increase performance by 4x
- D. Olson, NREL, submitted to Adv. Func. Mater.

Why Organic-Oxide Semiconductor Hybrid Solar Cells?

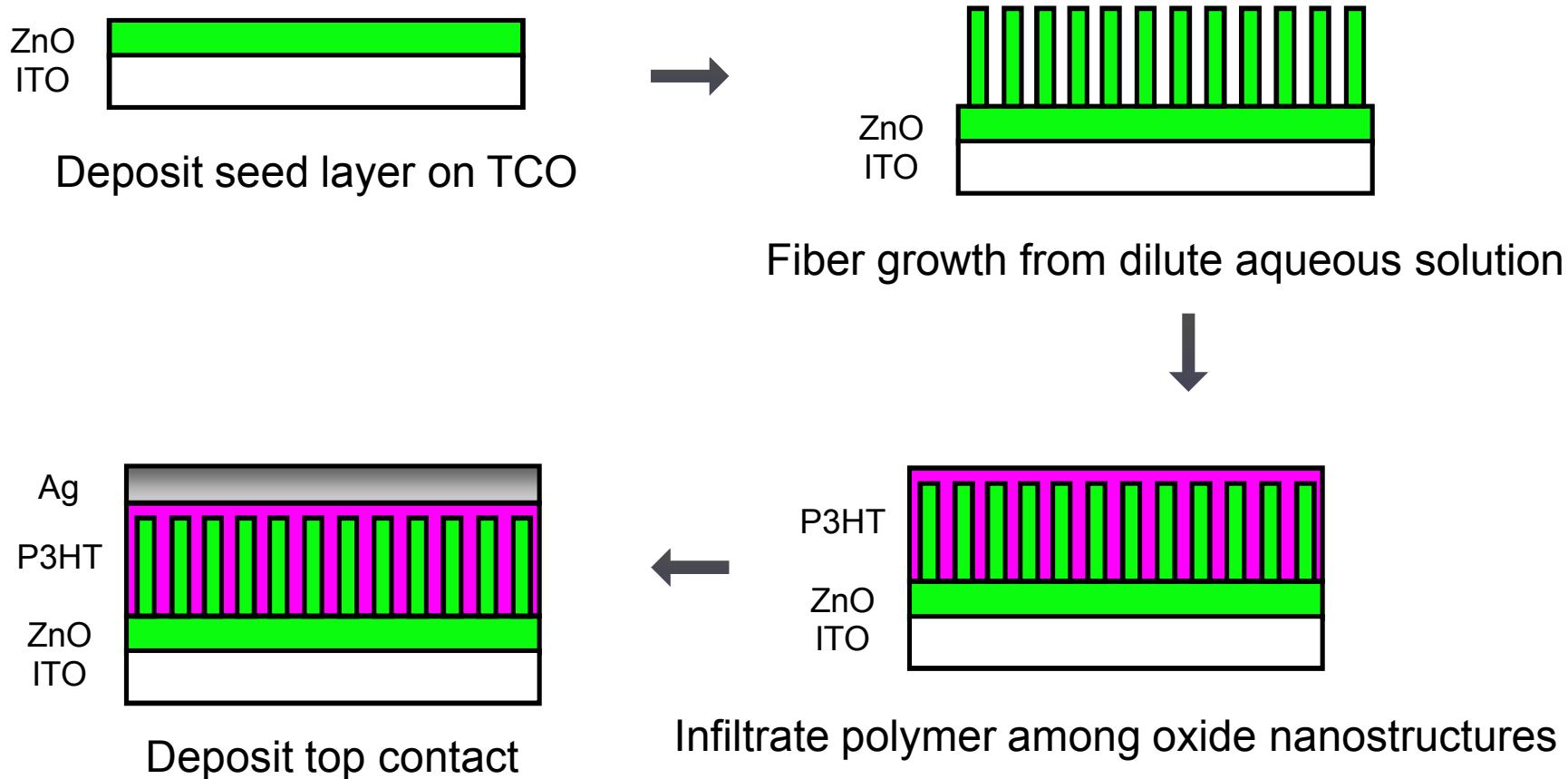


Proposed “Idealized Structure”

Advantages of using metal oxide semiconductor as the electron acceptor:

- Electron mobilities in crystalline oxide semiconductors is many orders higher than mobilities in organics
- Ordered structures might induce order in polymers to maintain high hole mobilities
- Control band alignments through judicious choice of materials
- Less susceptible to environments

Fabrication of Hybrid PV Devices



What is limiting the efficiency in current cells?

Form a new basic understanding of the properties required for optimal electron transporter nanostructures.

I. What is the optimum size, spacing & morphology of the oxide nanostrucutres?

- Determine impact of acceptor structure on cell performance.
(e.g., branched structures versus simple nanorod arrays)

II. Can we engineer the polymer/oxide interface?

- Demonstrate polymer wetting and infiltration at the nanoscale.
- Optimize electron transfer at interface.

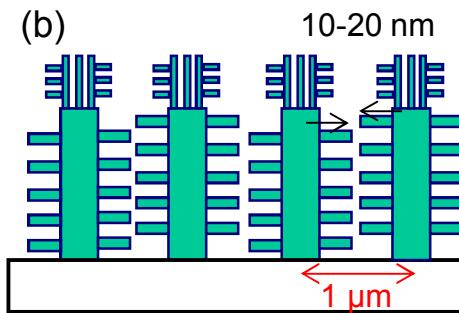
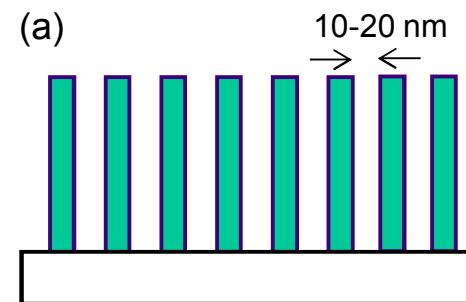
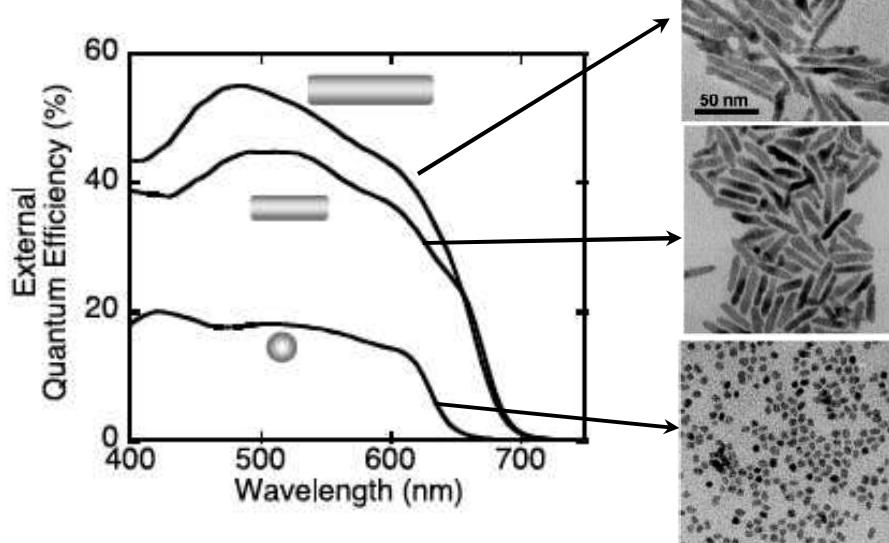
III. What is the best acceptor material system?

- Determine if other oxide semiconductor material systems provide advantages over ZnO and TiO₂.

NREL: Polymer development and cell characterization

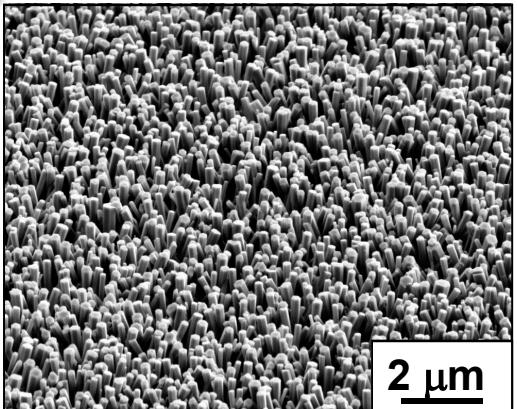
I. What is the Best Morphology?

CdSe-P3HT (Alivisatos)

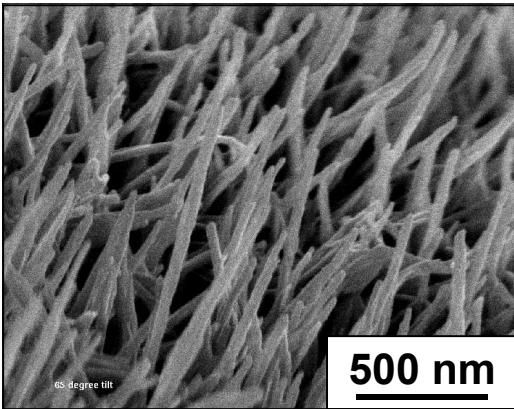


Control Growth of Nanostructures

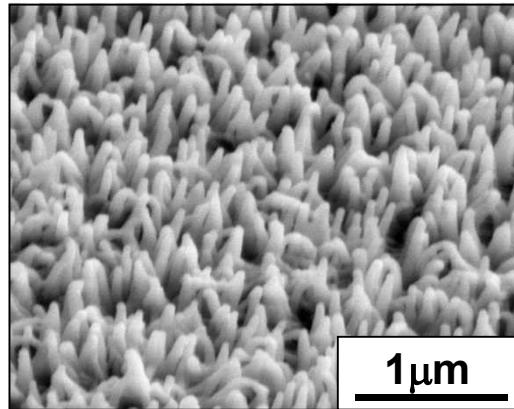
Nanorod Arrays



ZnO

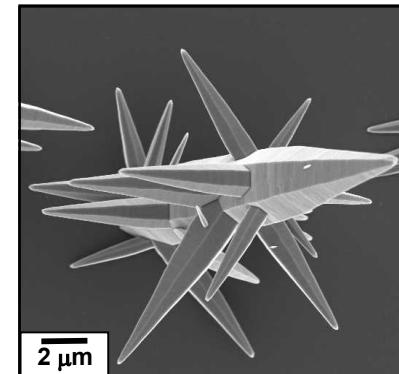
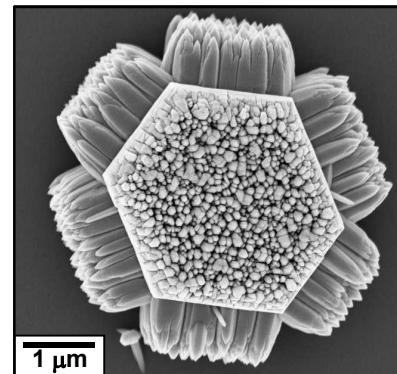
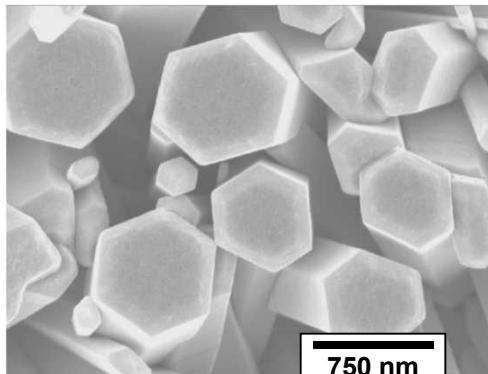
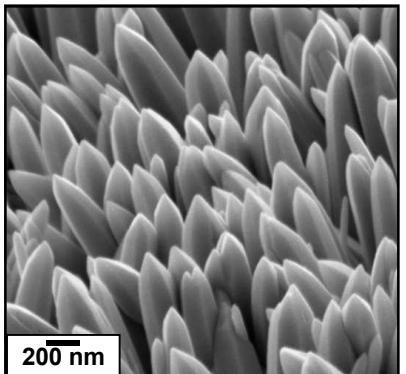


TiO₂

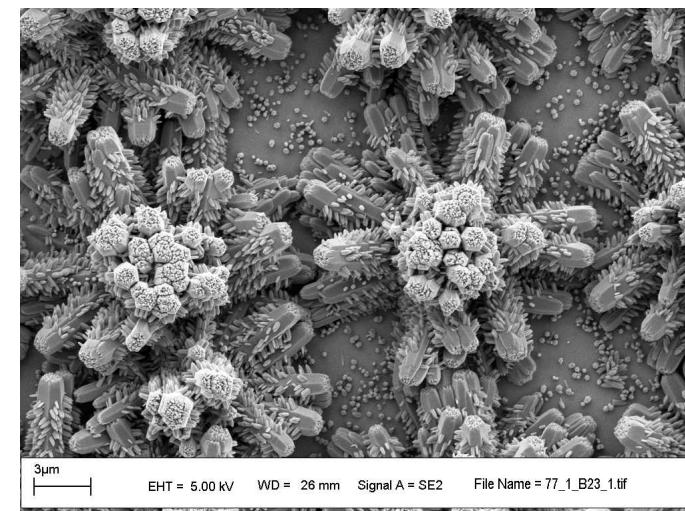
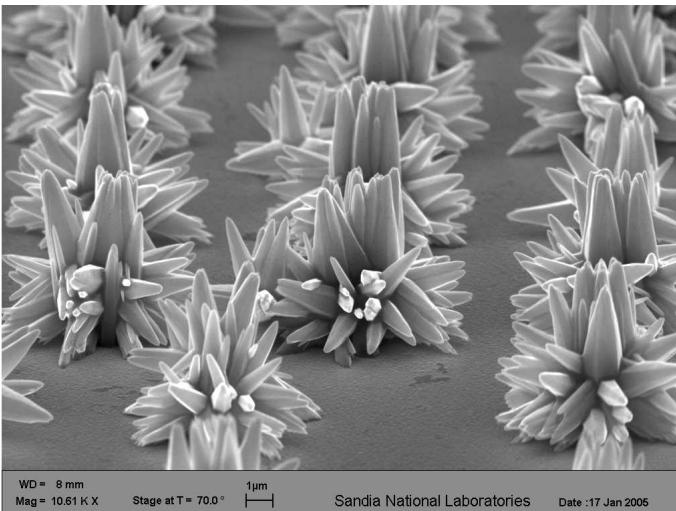
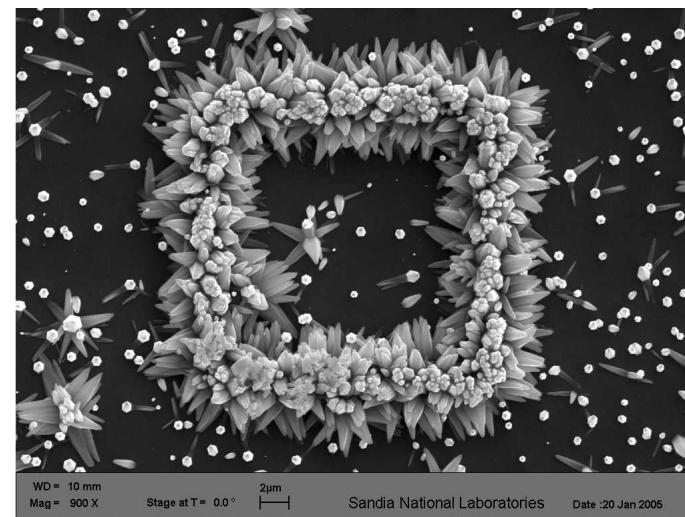
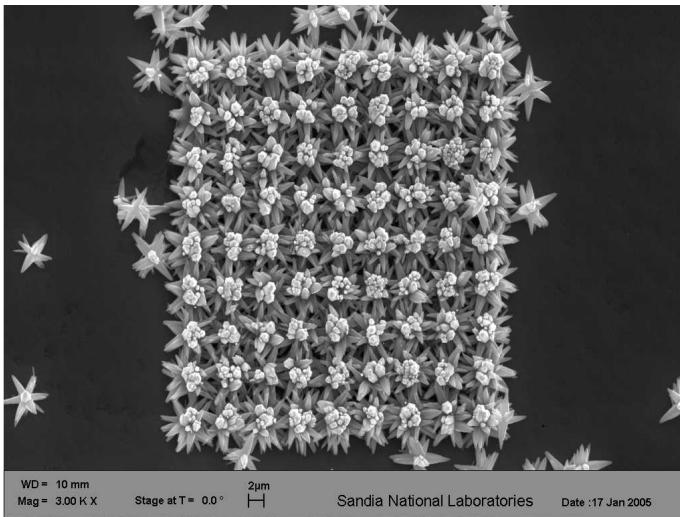


Conductive Polymers

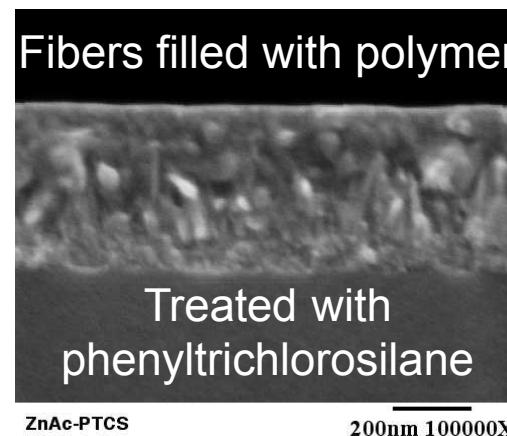
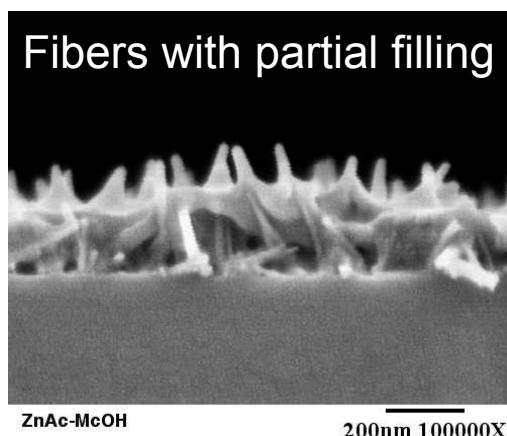
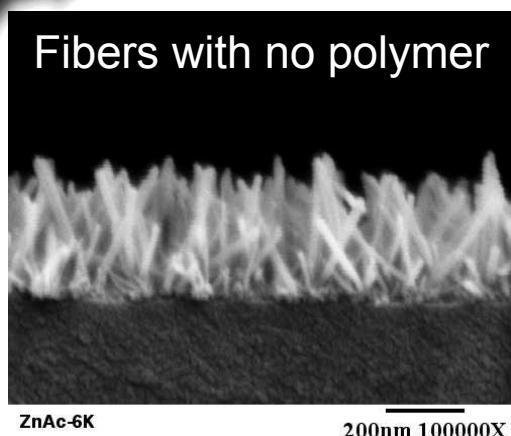
Morphology Control of ZnO Nanostructures



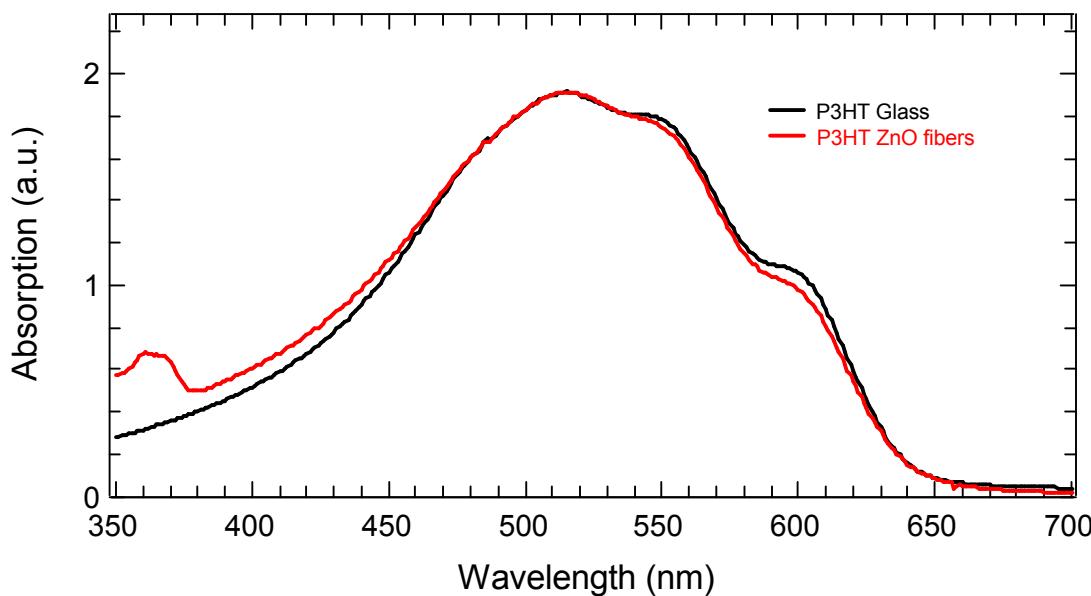
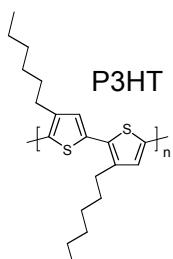
Control Placement of ZnO Nanostructures on Substrates



II. Infiltrating P3HT into ZnO Nanofibers

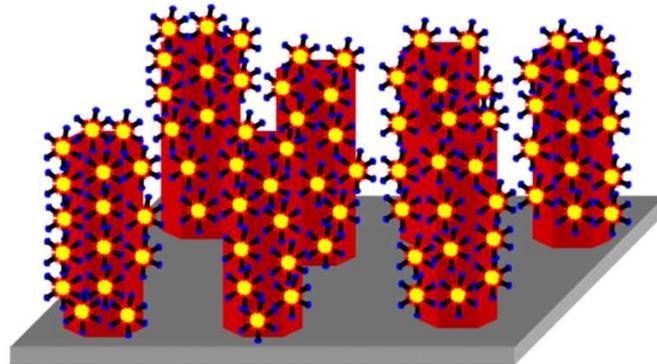


P3HT
intercalated into
ZnO nanofibers
remains ordered

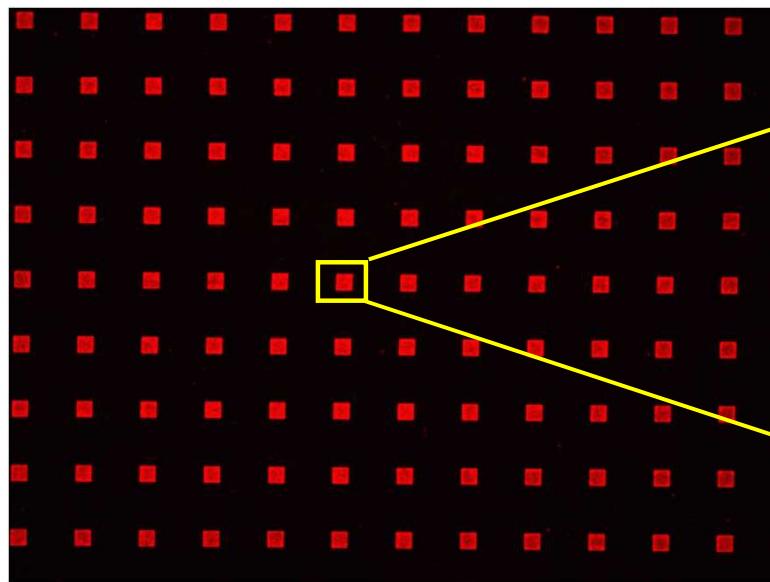
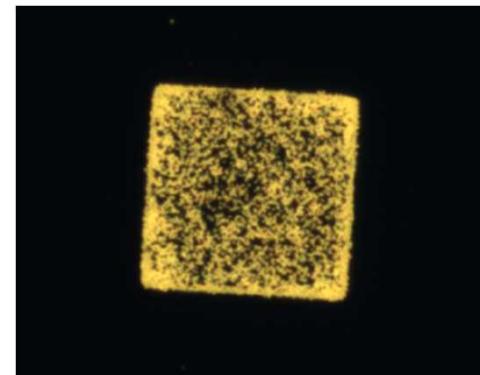


III. Surface Functionalization

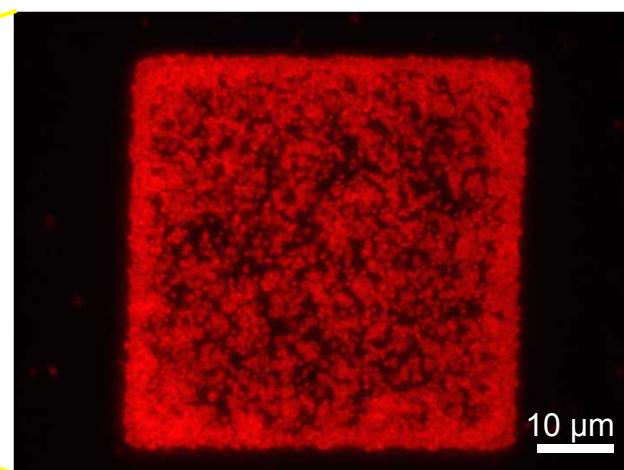
Quantum Dots on ZnO Nanorods



ZnO



ZnO + QDs



Summary

Hybrid Cell with Hierarchal Acceptor Structure

