

# ***Heteronucleation of Zinc Oxide Nanocrystals***

**Julia W. P. Hsu**

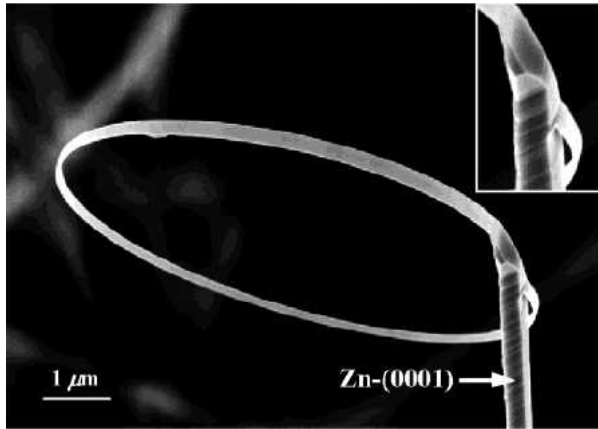
T. L. Sounart, Z. R. Tian,  
M. Clift, L. Brewer, J. Floro, J. Michael,  
N. C. Simmons, C. M. Matzke, B. McKenzie,  
J. A. Voigt, & J. Liu

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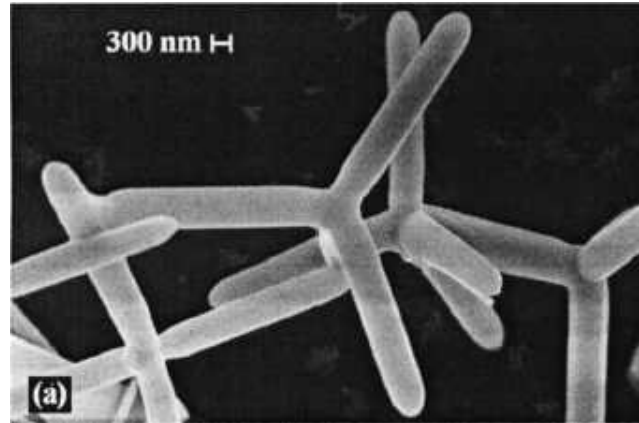
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# Motivation

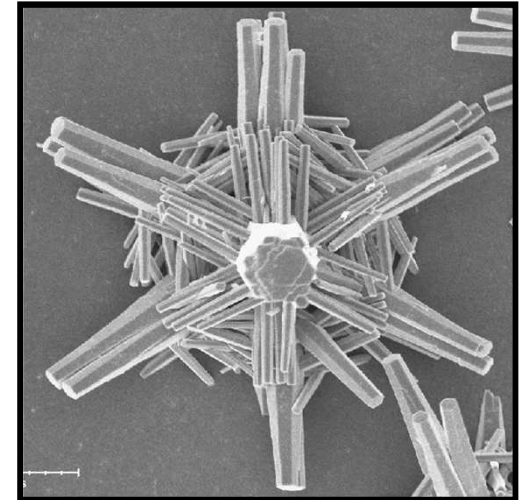
## Complex ZnO Nanostructures



Nanobelt  
Georgia Tech



Tetrapods  
U. Hong Kong



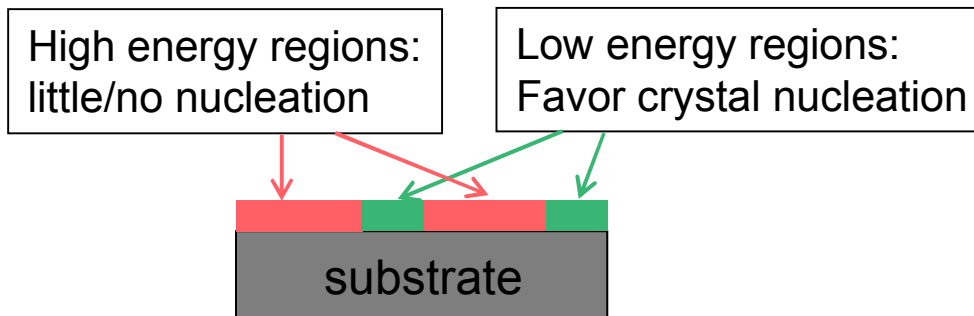
Hierarchical  
nanostructures,  
Sandia

- ZnO: wide bandgap semiconductor, piezoelectric, transparent conductor, spintronics
- Explosion in nanomaterials synthesis
- To make these new materials useful:
  - **(Directed) Assembly** -- placement, density, orientation
- Conventional 2D lithography inadequate

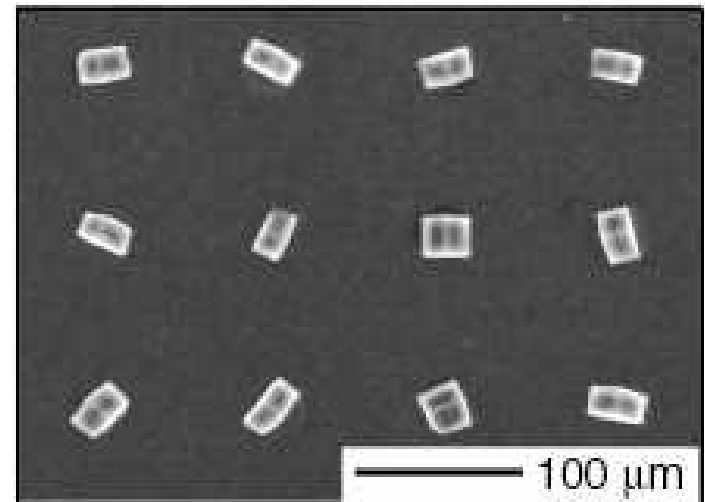


# Experimental Approach

- **Solution growth**: low temperature (organic template/modifiers possible), environmentally benign, large area uniformity
- +
- **Soft lithography** (microcontact printing): to *chemically* modify the surface and create regions on surface where the nucleation energy is different



calcite

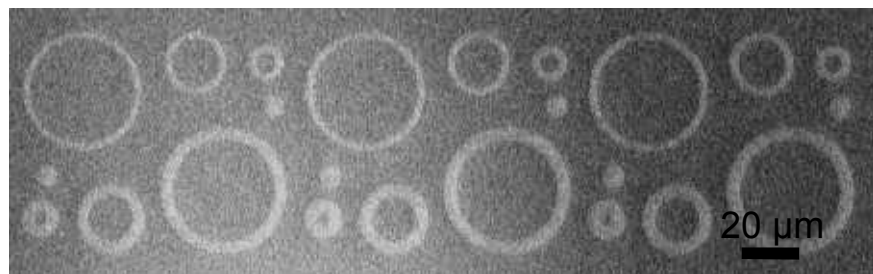
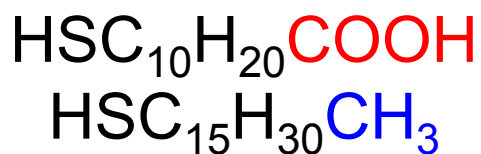
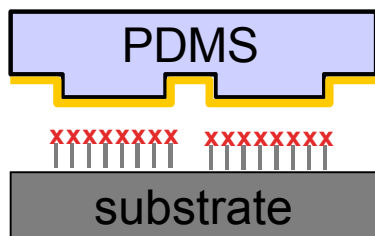
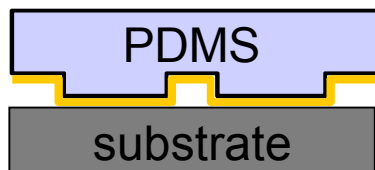


Aizenberg, Nature, 1999

- “**Bio-inspired**”: use organic template to direct inorganic materials growth
- Extend to ZnO: a technologically important material

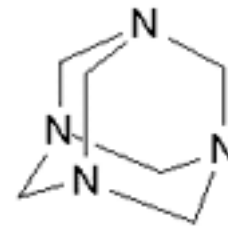


## Microcontact Printing



## Solution Growth

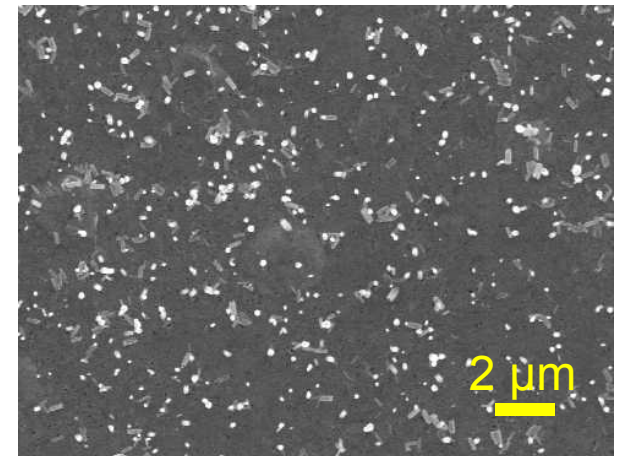
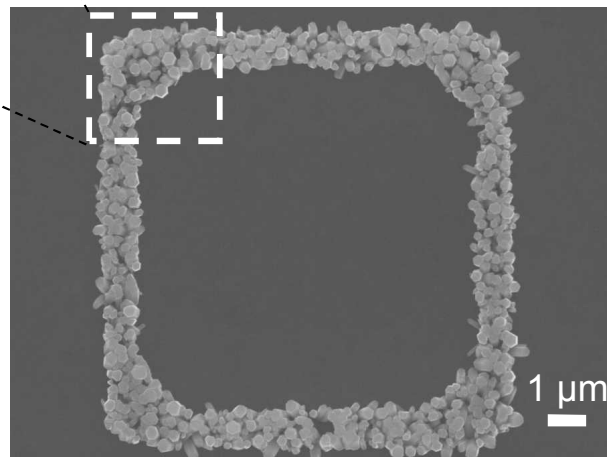
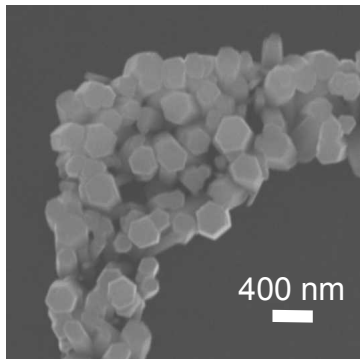
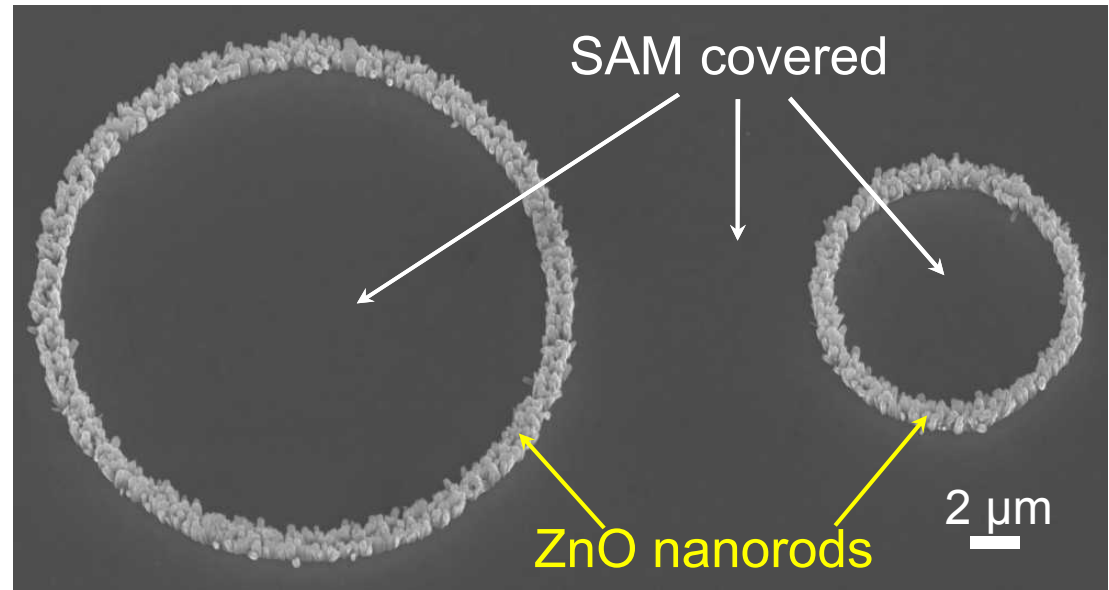
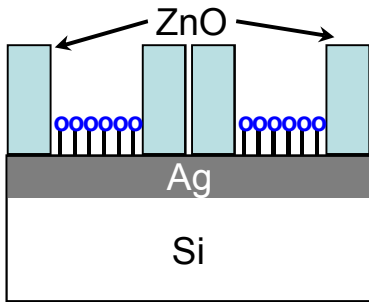
- Dilute  $\text{Zn}(\text{NO}_3)_2 + \text{HMT}$



- $T = 55 - 65^\circ \text{C}$
  - Time = 2 to 6 hrs
  - ZnO growth mediated by HMT degradation
- Demonstrate control in
    - spatial placement
    - crystal orientation
    - and nucleation density

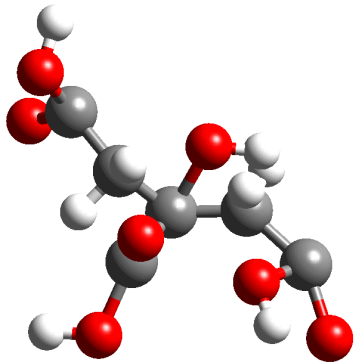


# Selective ZnO Growth on Ag

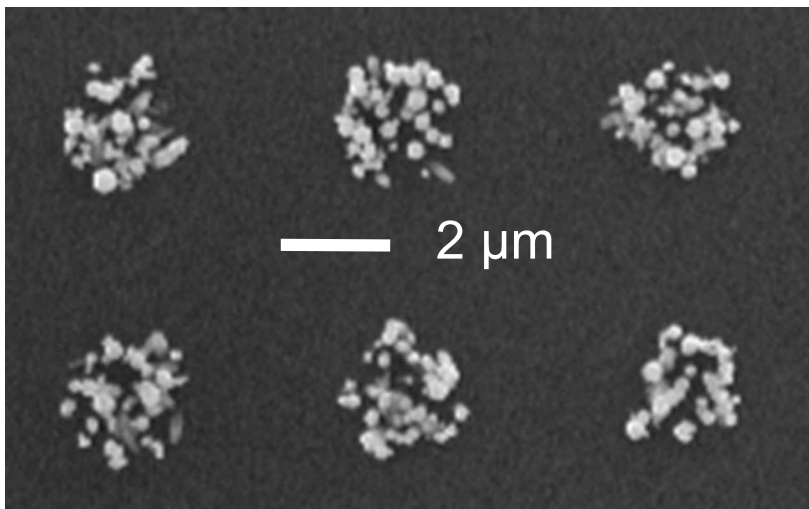




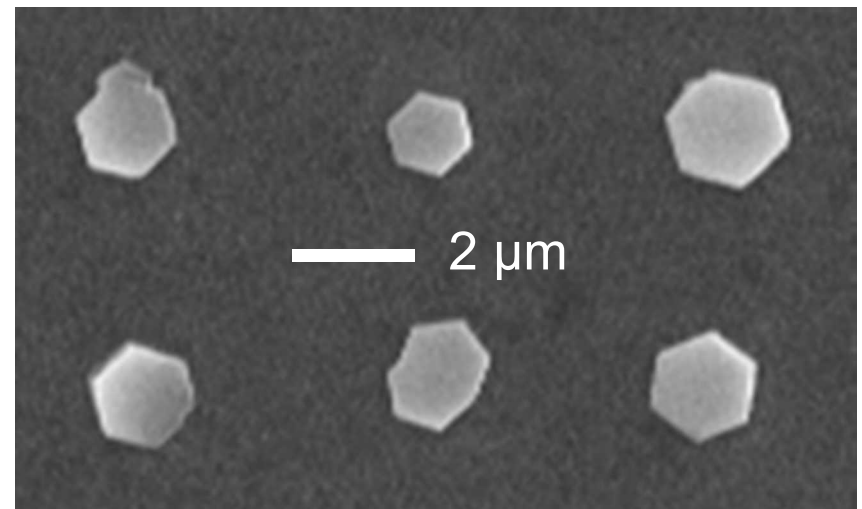
# Nucleation Density Control: Growth Modifier



Citrate inhibits growth along  $\langle 001 \rangle$ :  
Rods become platelets



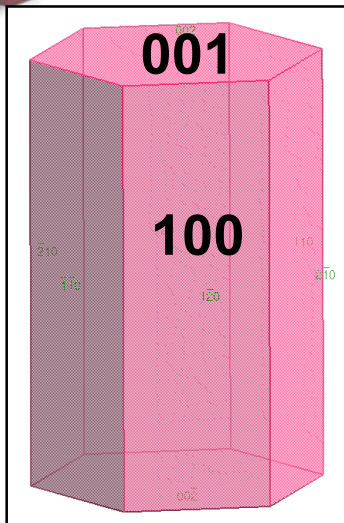
No citrate:  
 $46 \pm 6$  ZnO nanorods  
per nucleation sites



High citrate concentration:  
1-2 ZnO platelets per  
nucleation sites



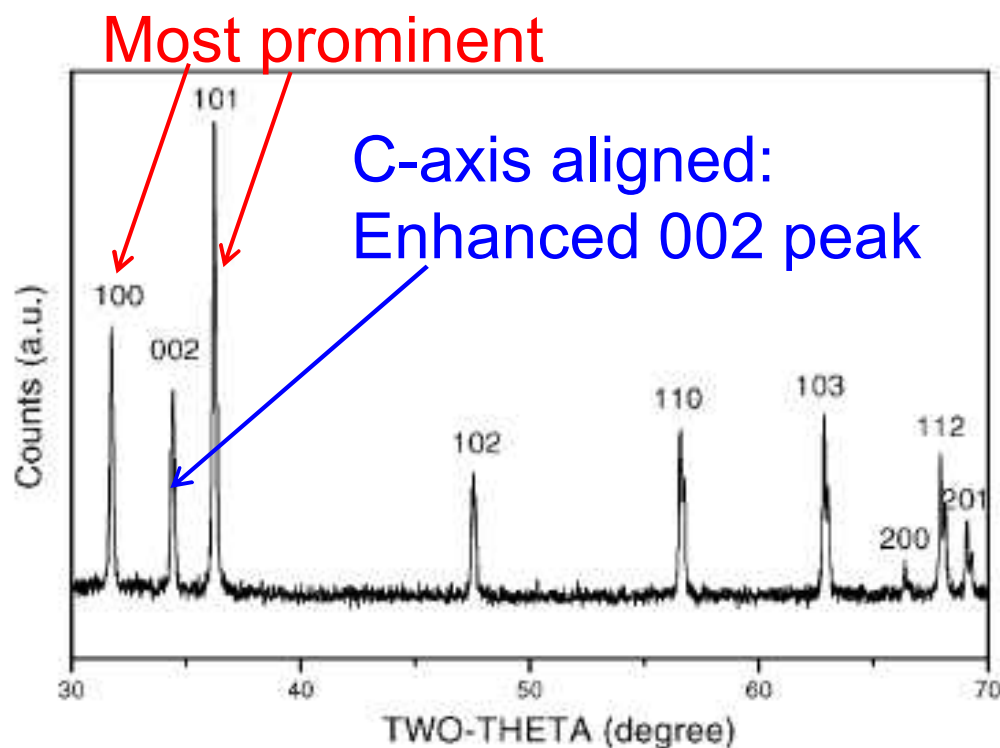
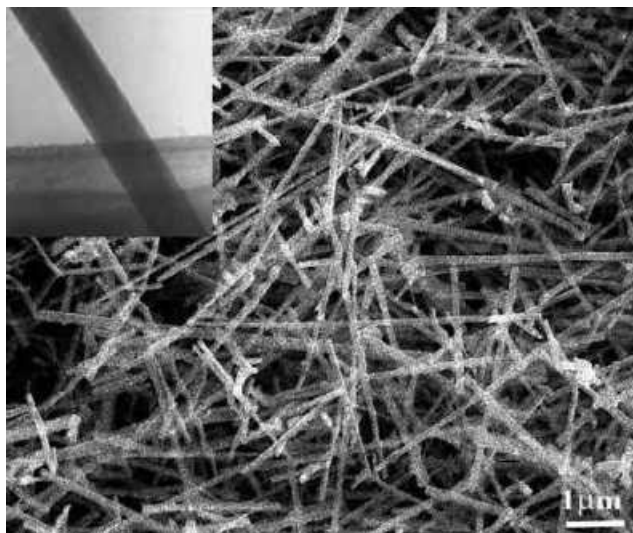
# X-ray of Zincite



Wurtzite (hexagonal) structure:

$a$  (in plane) = 3.250 Å

$c$  ( $\langle 001 \rangle$ ) = 5.207 Å

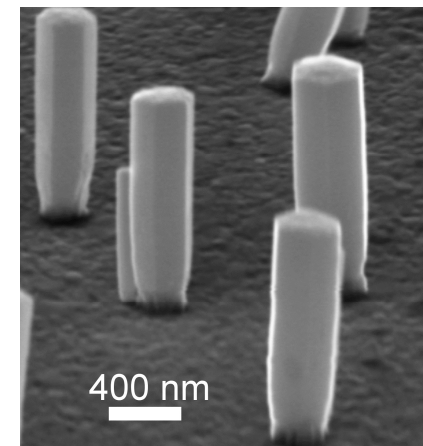
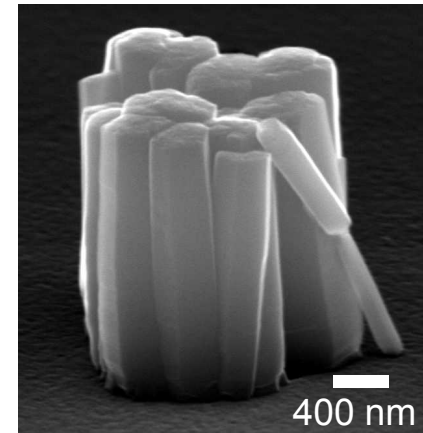
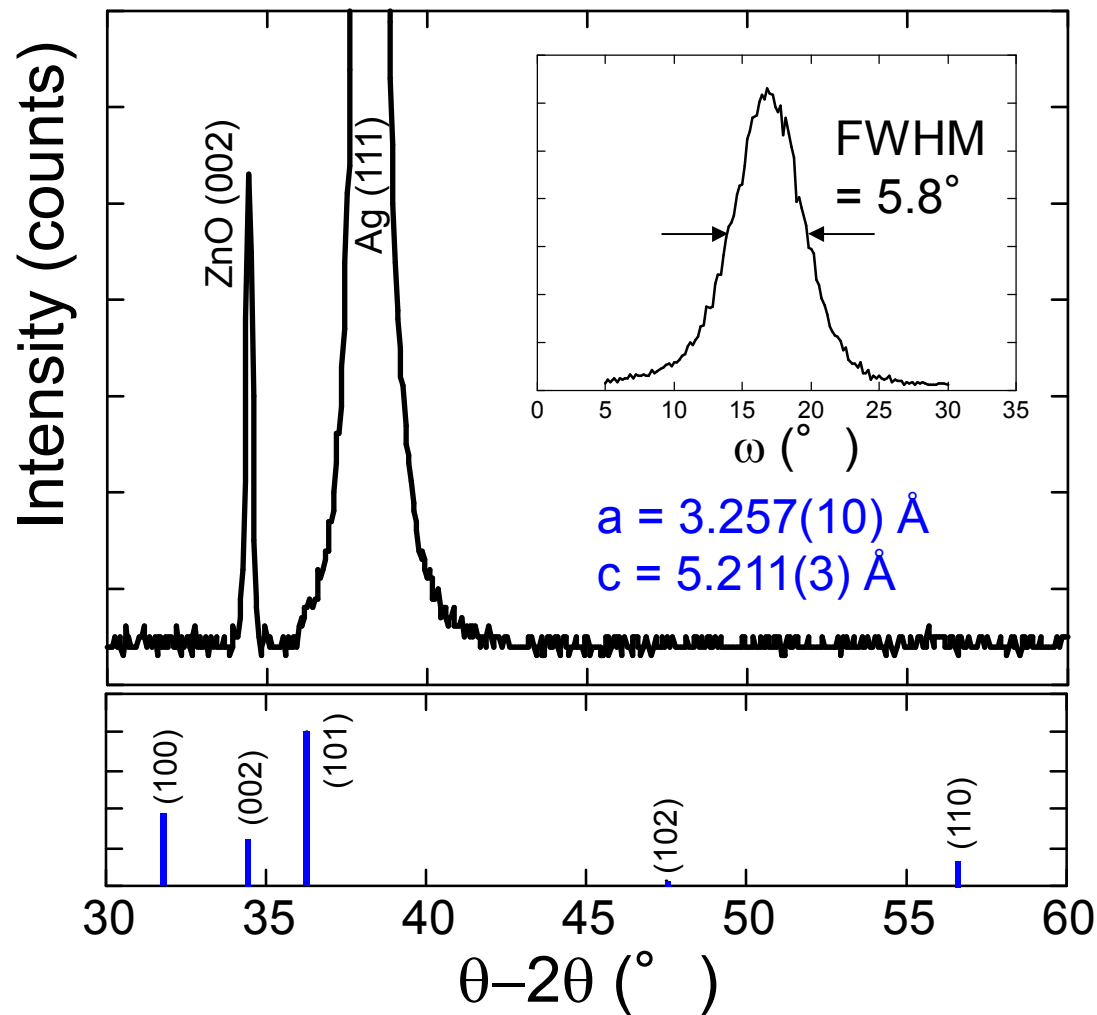


J.Y. Lee, et. al., J. Cryst. Growth 233, 5, 2001



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# Highly Oriented ZnO Growth



Single X'tal

Thanks to R. G. Tissot, M. A. Rodriguez, D. L. Overmeyer

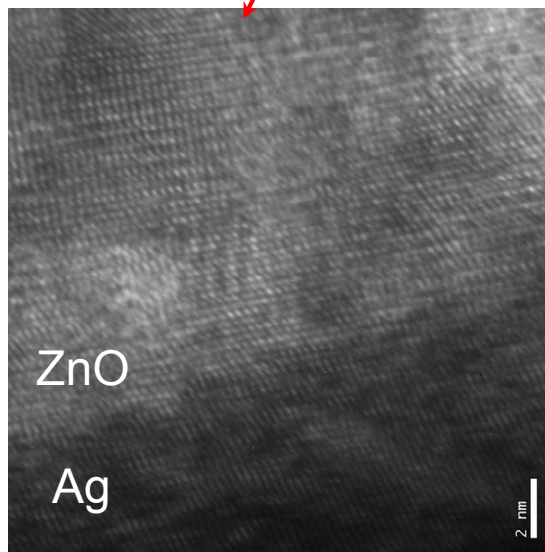
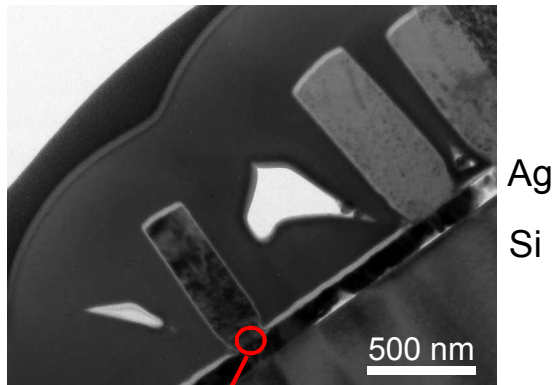


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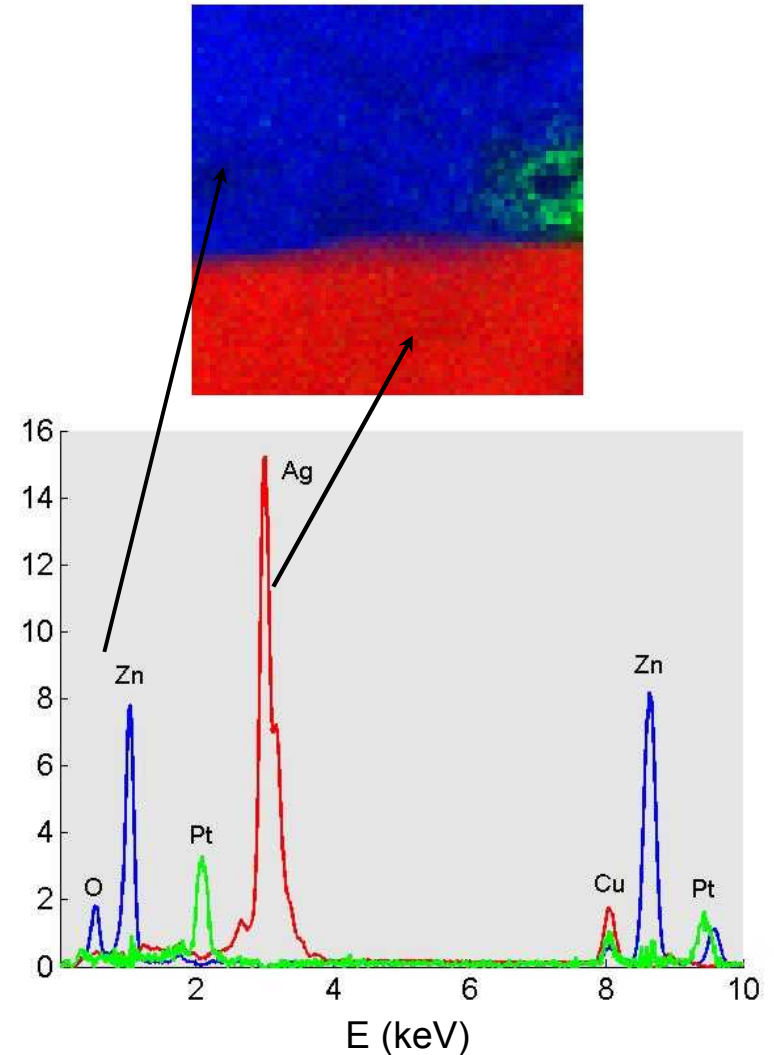


# TEM Study of ZnO-Ag Interface

High resolution

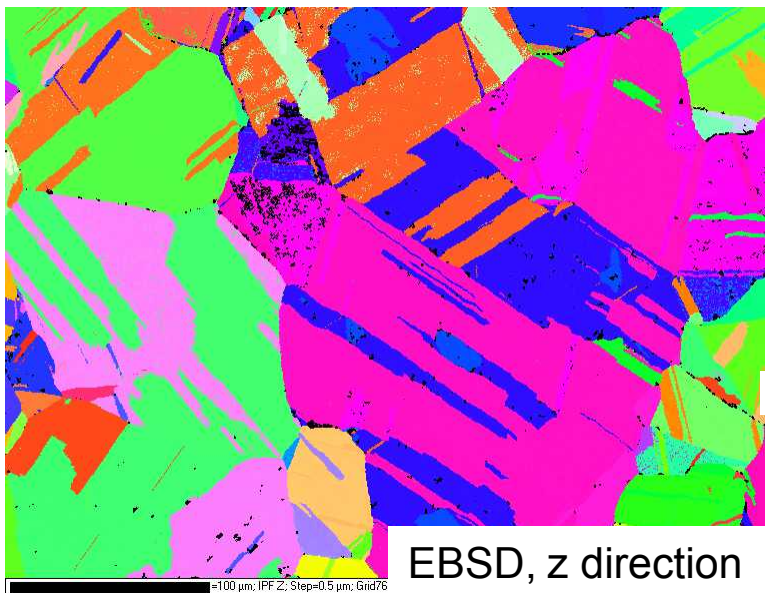
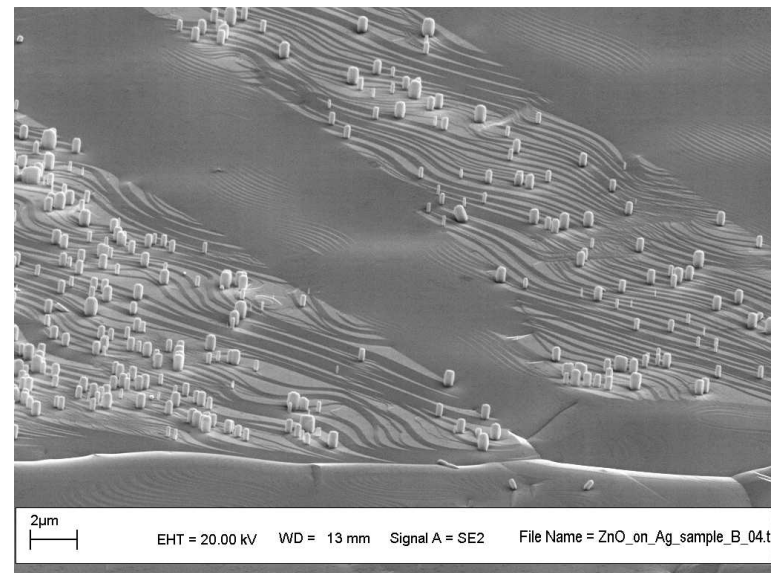
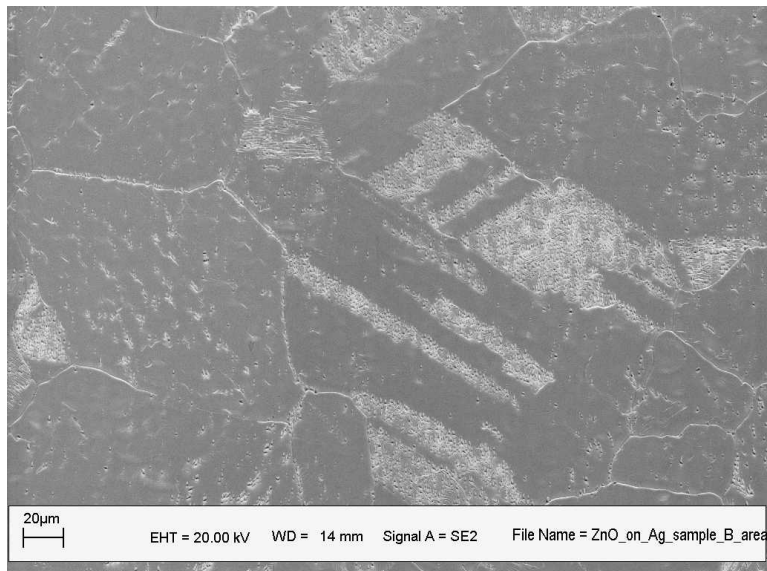


Chemical

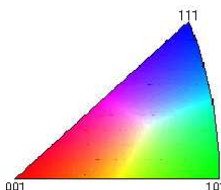


*No “foreign” materials between ZnO & Ag; faceted interface. ZnO defected.*

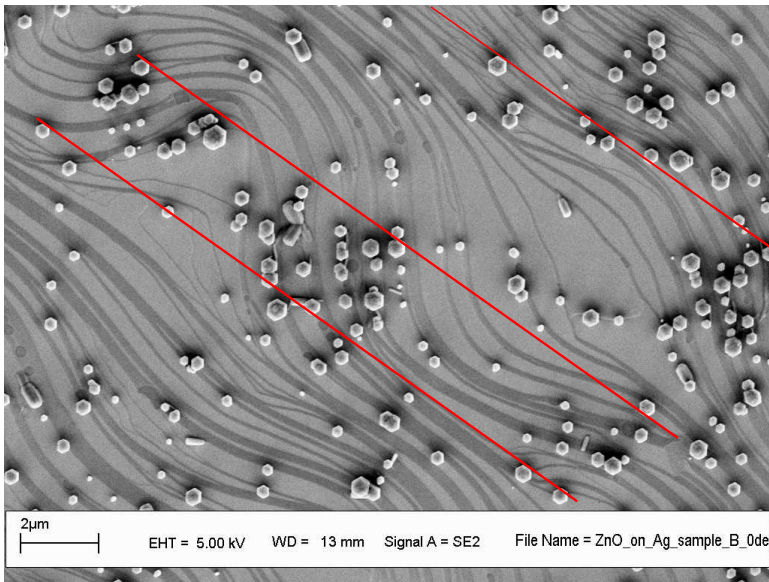
# Selective Growth on Multigrain Ag



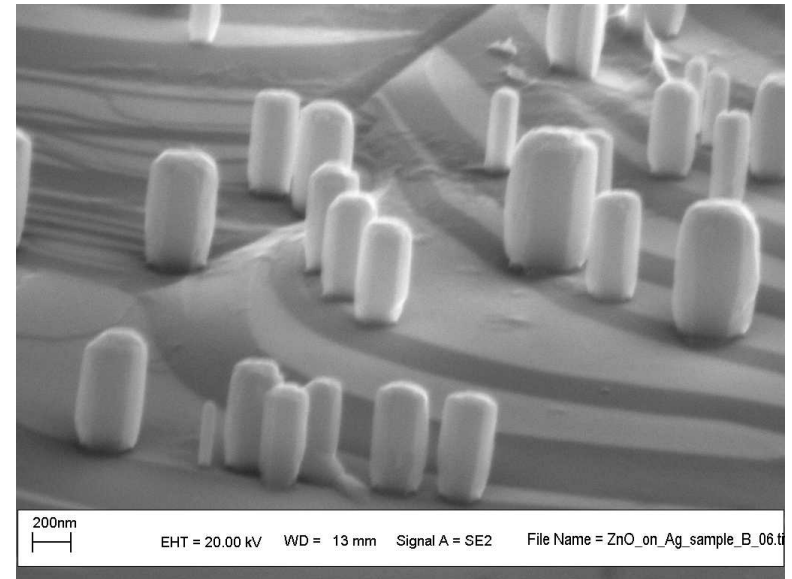
ZnO nanorods selectively grow on  $\langle 111 \rangle$  oriented grains, with  $\langle 0001 \rangle$  axis perpendicular to substrate



# Crystallographic Alignment



- All nanorods have the same in-plane orientation (6-fold degenerate).
- Thus, **there is definite alignment between ZnO and Ag lattices.**
- However, lattice mismatch  $> 10\%$ ?

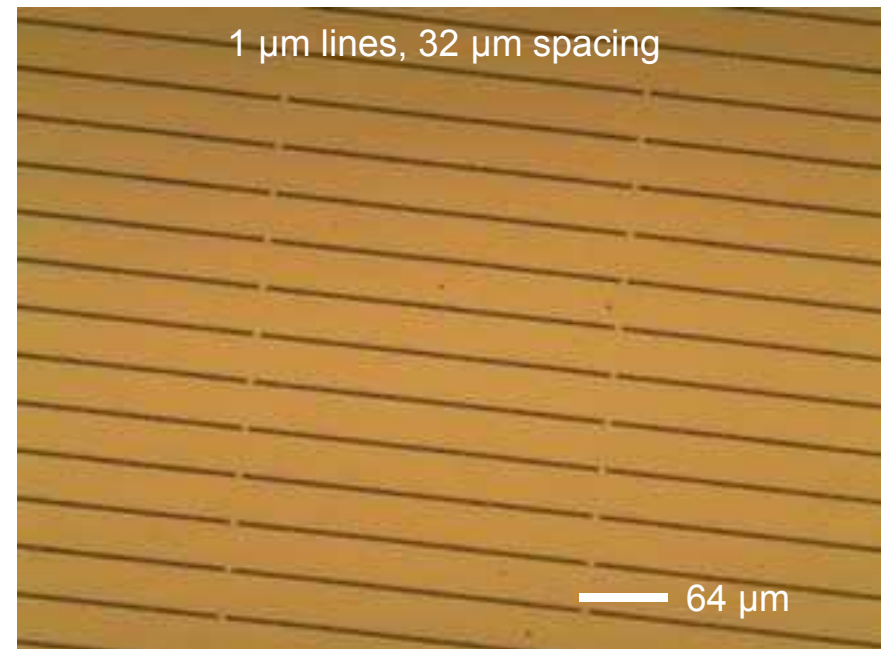
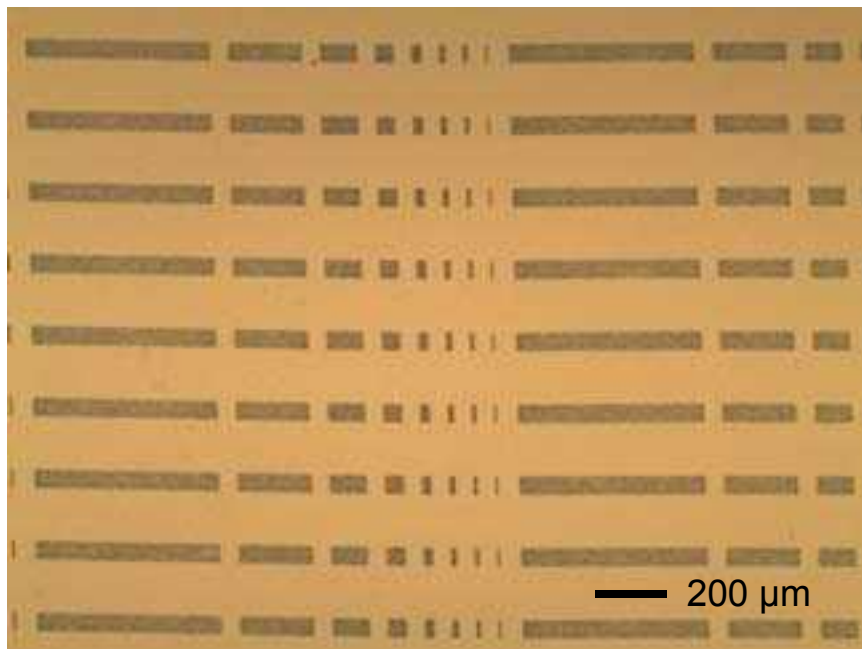
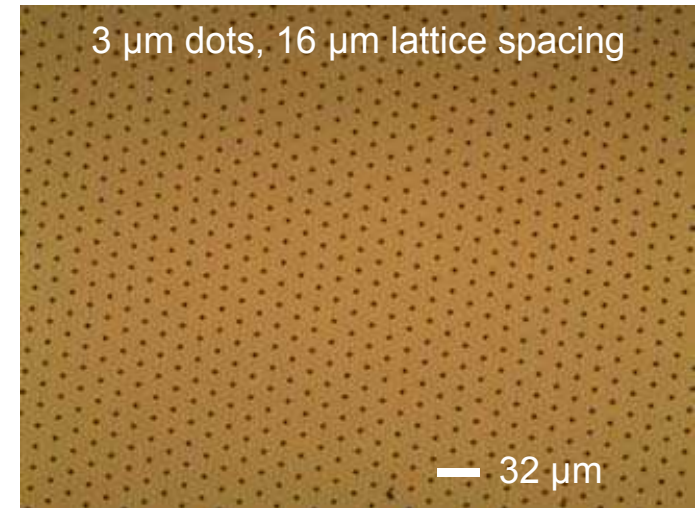
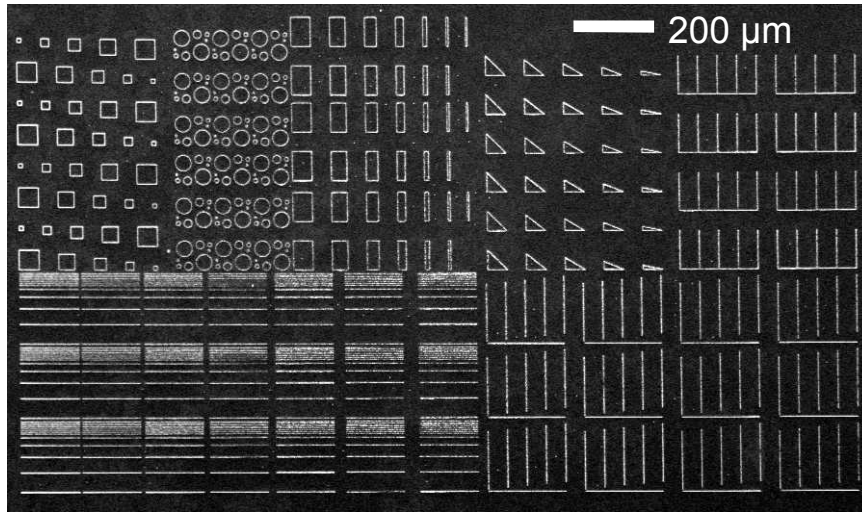


Nanorods preferentially grow at the edges of the upper terraces. Strain relief?



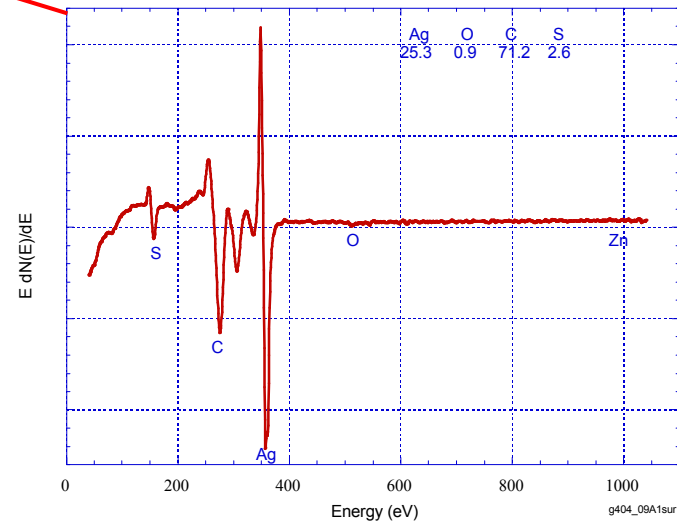
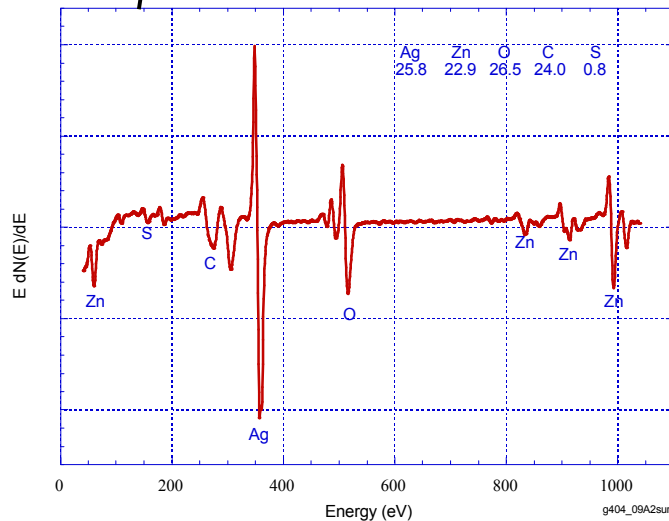
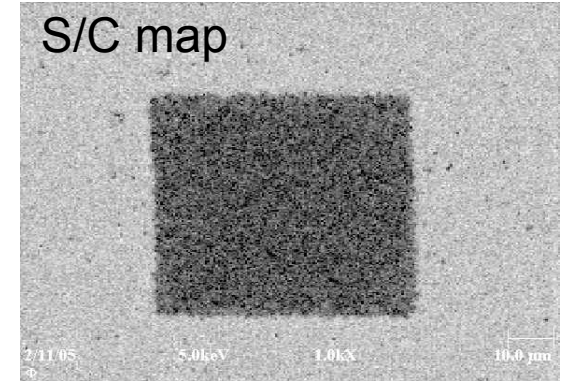
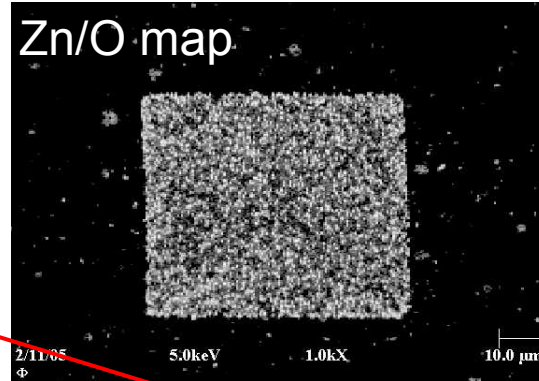
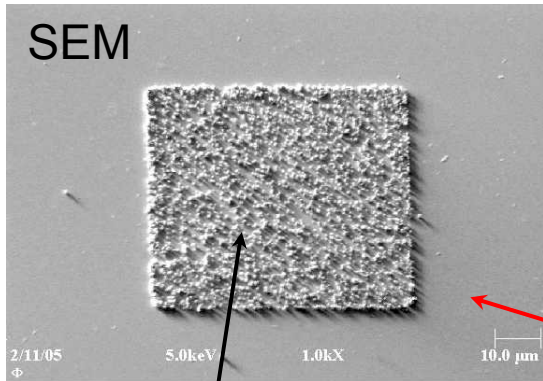


# Large Area



# ***-CH<sub>3</sub> surface***

**CH<sub>3</sub>** Endgroup; after 3 hrs of growth



**As expected: Zn & O found on no SAM regions**

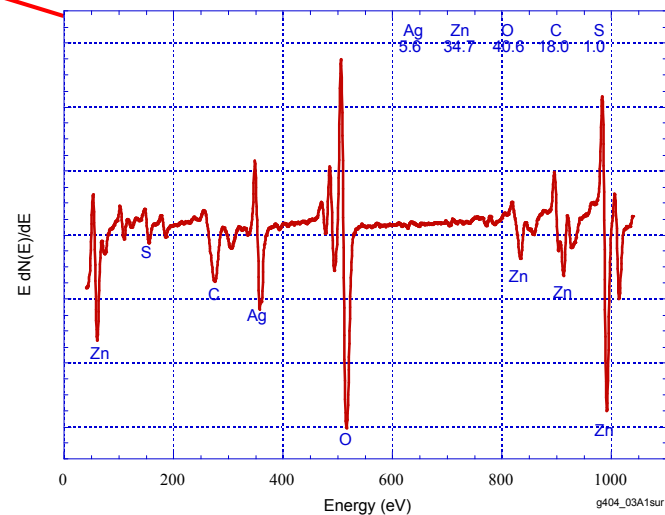
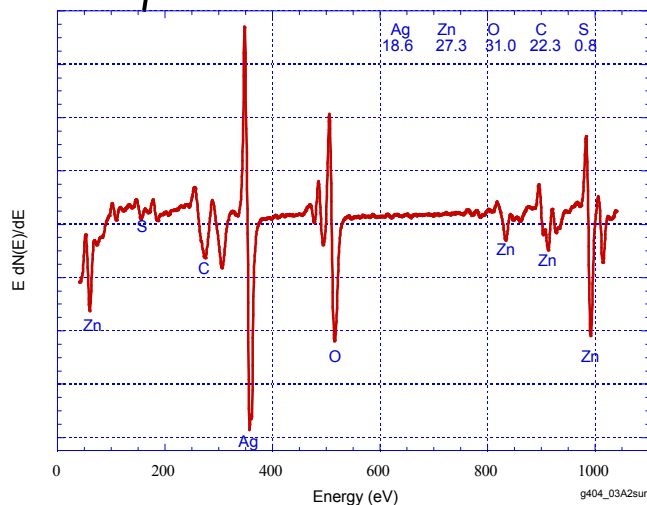
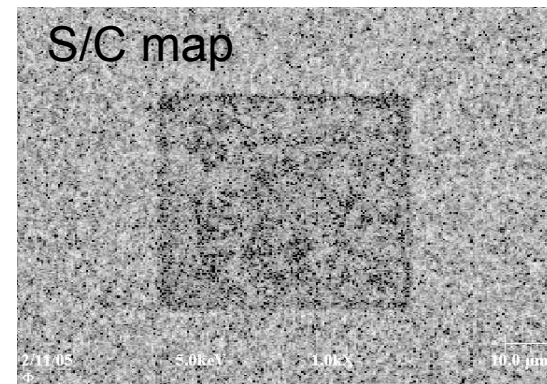
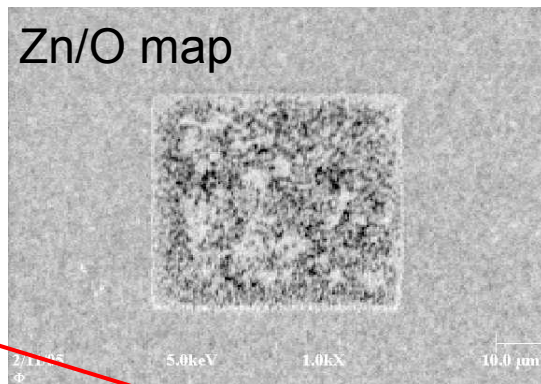
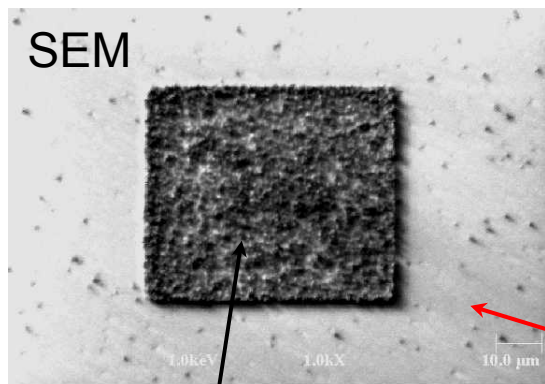


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# -COOH Surface (Long Time)

COOH Endgroup; after 3 hrs of growth



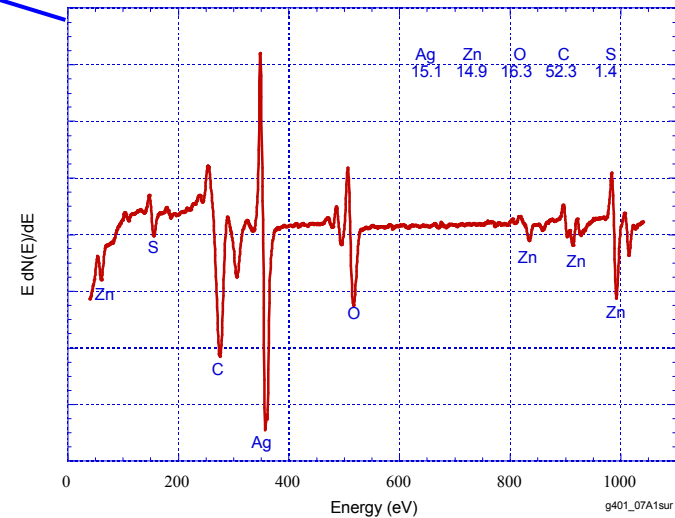
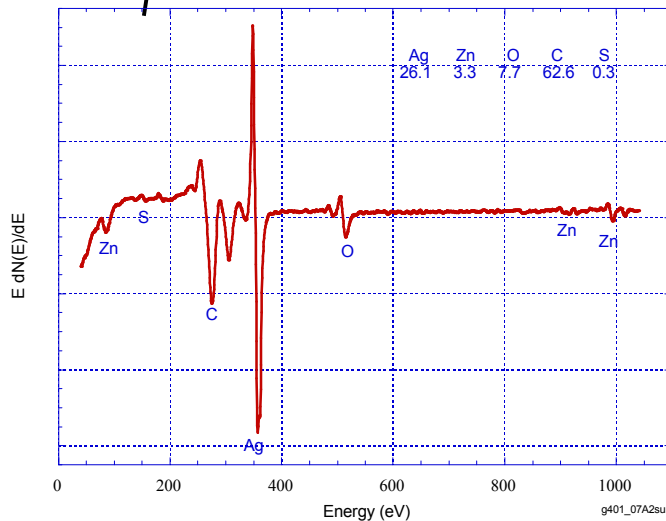
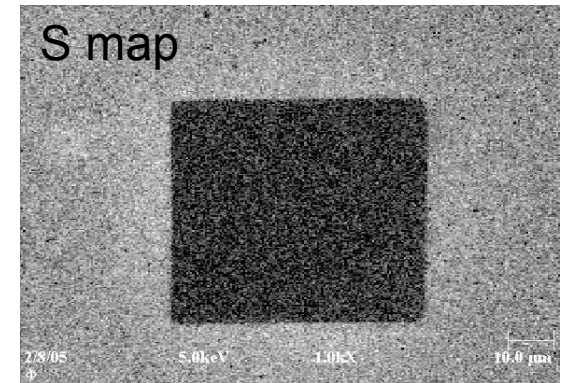
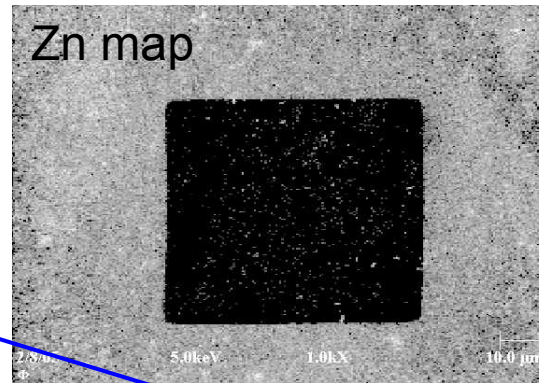
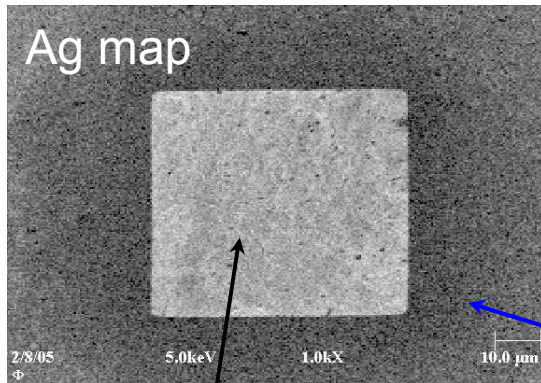
Zn & O on SAM regions, but no crystals!



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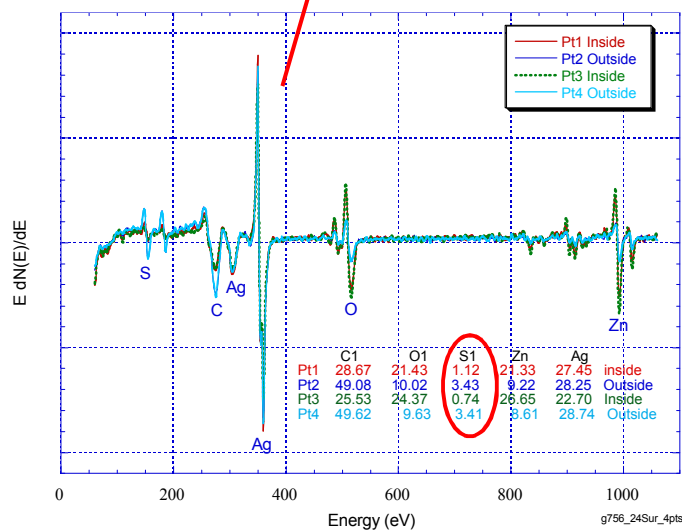
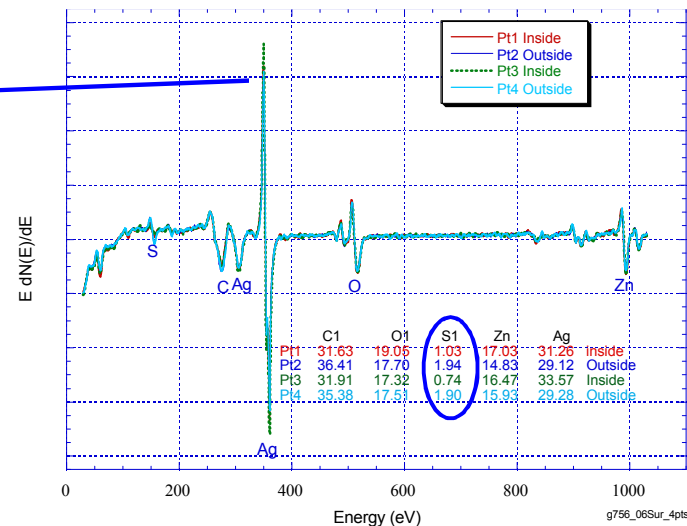
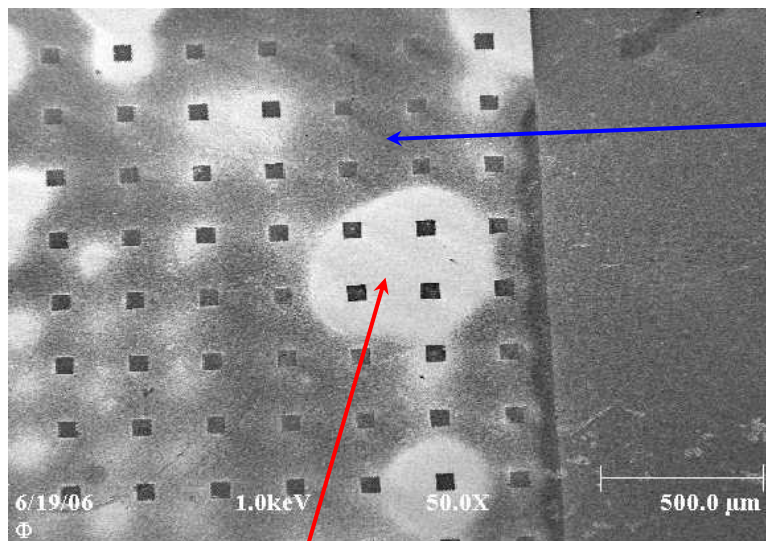
# ***-COOH Surface (Short Time)***

**COOH** Endgroup; 5 min incubation at RT

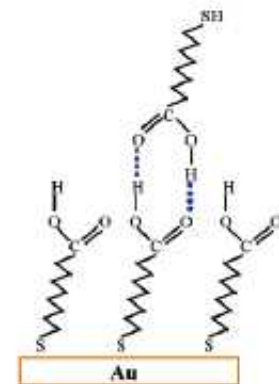


Zn has strong affinity to  $\text{COO}^-$ , as expected.  
However, this is not sufficient for rod growth.

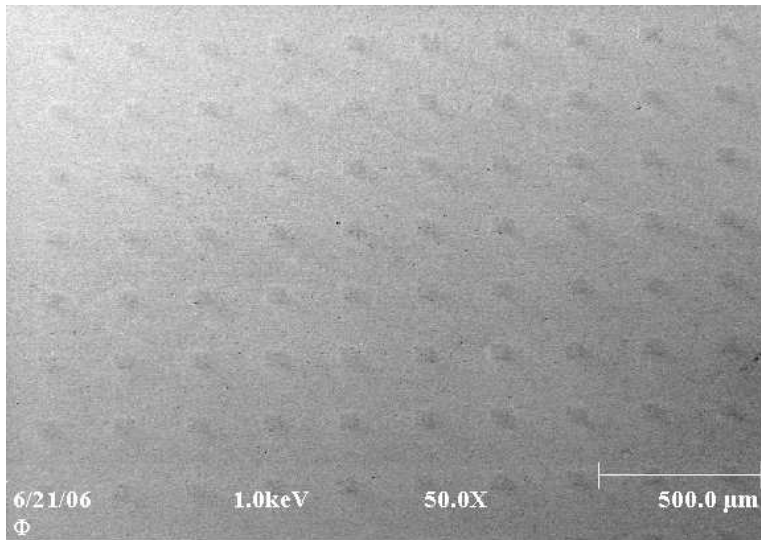
# Large Scale Non-uniformity on COOH Surface



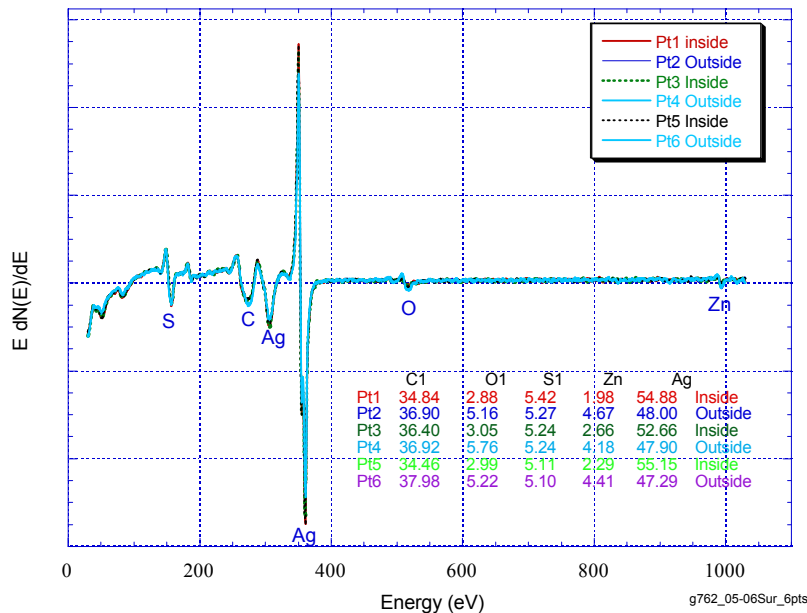
- Enhanced S signal in the SAM regions with no rods.
- Bilayers with -SH?



# -SH Surface

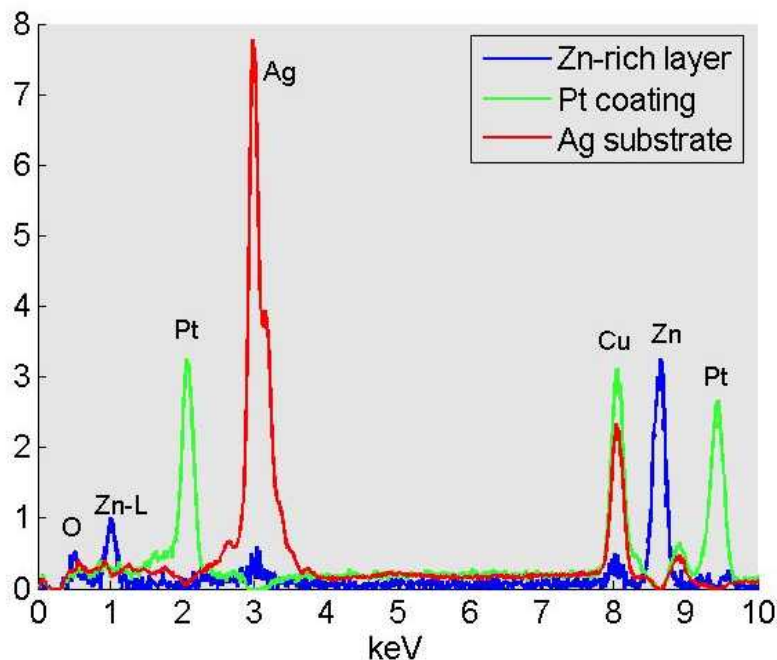
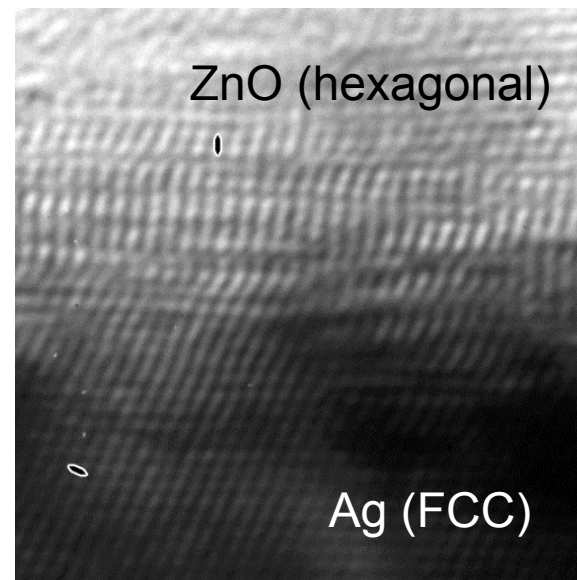
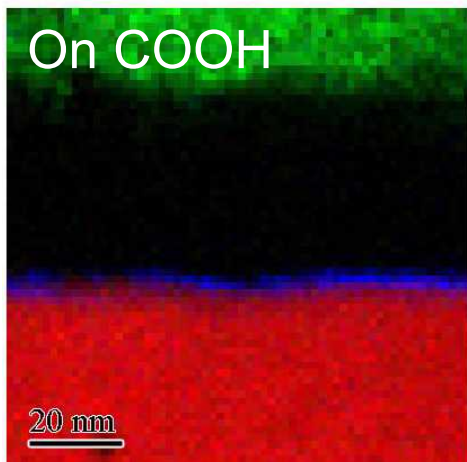


- No optical contrast
- No ZnO nanorods anywhere
- Faint SEM contrast. S % similar between C8DT and bare Ag regions: C8DT diffusion?
- Auger detects slight enhancement of Zn & O on SAM regions
- ZnO film?
  - Zn-S: 205 kJ/mol
  - Zn-O: 159 kJ/mol





# ZnO Film?



- Composite image and spectra (EDX) showing Zn-rich layer on top of Ag substrate
- HREM image showing the layer is crystalline ZnO  
Ag  $d\{111\}$  2.4 Å  
ZnO  $d\{0002\}$  2.6 Å







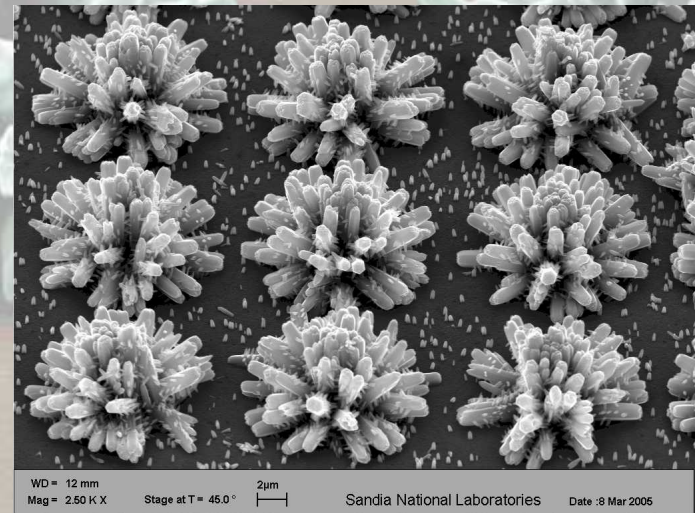
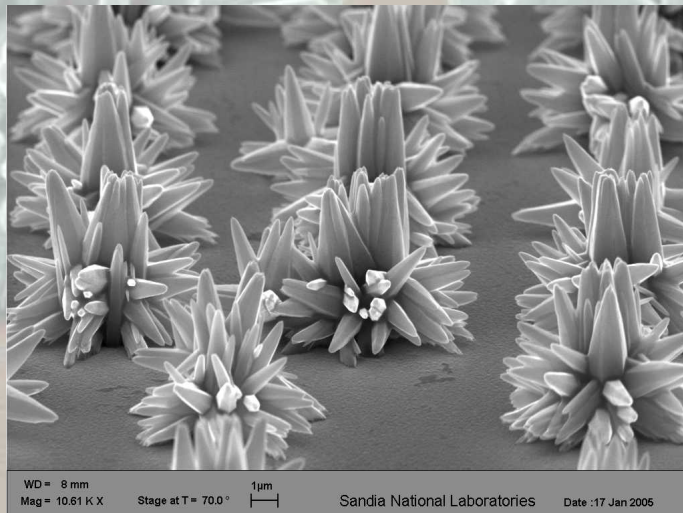
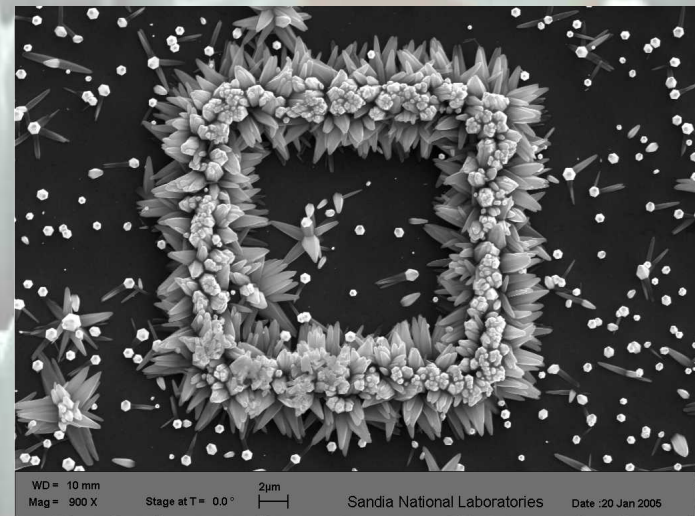
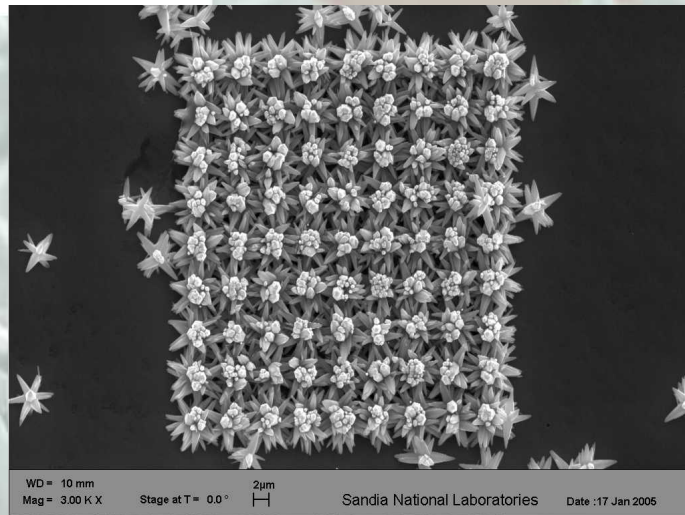
# Summary & Outlook

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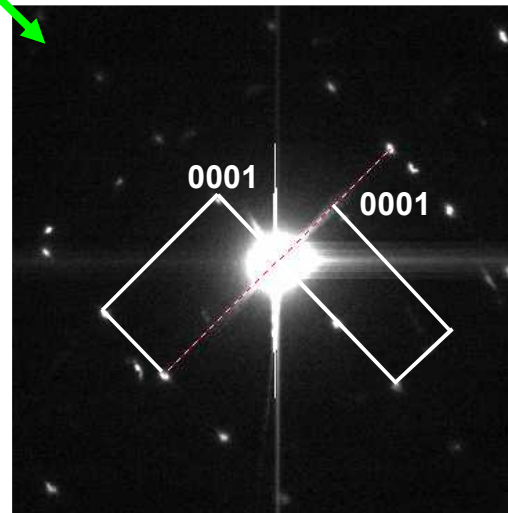
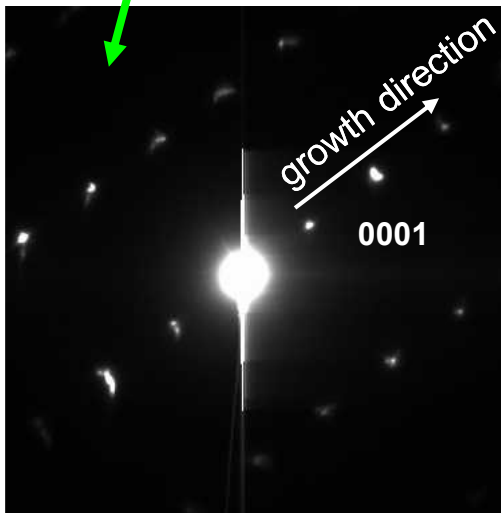
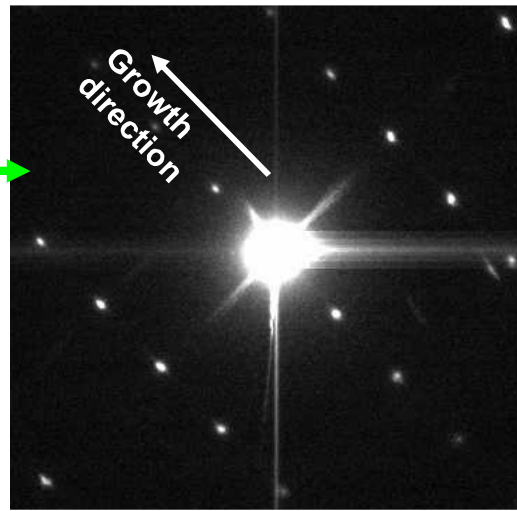
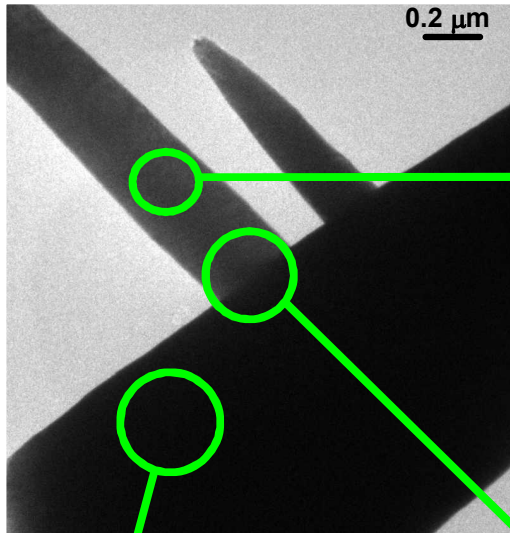
- Use organic templates to direct ZnO growth --  
Demonstrate superb control in  
*Spatial placement, selectivity, crystal orientation,  
& nucleation density*
- **Preferential nucleation of ZnO on Ag (111):** same  
crystal symmetry, but no lattice match
- CH<sub>3</sub>: no ZnO; COOH: nanorods; SH: ZnO film
- **Next**
  - Understand selectivity and orientation of ZnO on (111) Ag
  - Can the SAM results be explained by interfacial energy? Is it possible to do a calculation?
  - When does ZnO grow as a film? When does it grow as rods?



# Complex ZnO + Patterning



# Crystal Orientation between Primary & Secondary Rods



- (11-02) Twin relationship between P & S crystals
- Twin plane is not the boundary
- The boundary is the  $\{11-00\}$  of the primary rod and  $\{0001-\}$  of the secondary rods
- Could be coincidence angle:  $12.3^\circ$  corresponds to  $7c$  and  $6\sqrt{3}a$
- Single direction probably due to broken inversion symmetry (in progress)

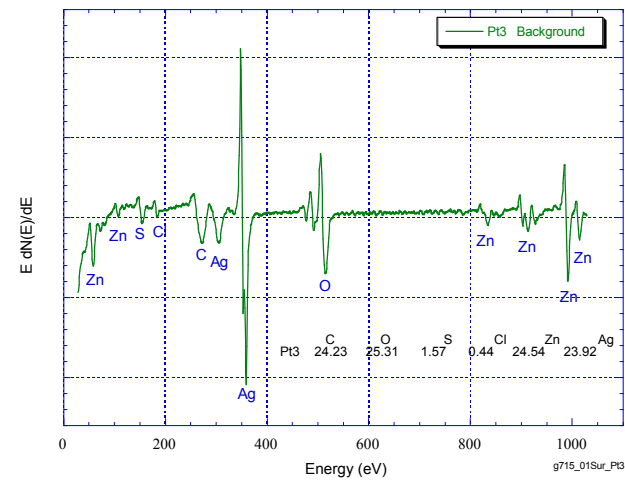
Thx to Jerry Floro



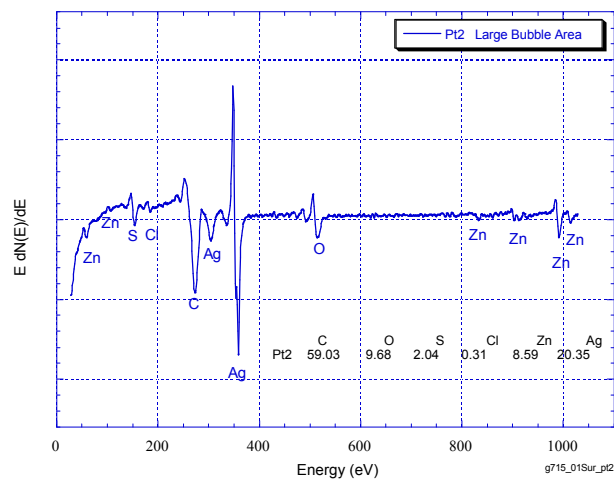
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# Large Scale Non-uniformity on COOH Surface

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.



Use 060206F



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