



U.S. DEPARTMENT OF  
**ENERGY**

**Nuclear Energy** SAND2014-2959P

## The Fuel Cycle Research & Development Iodine Capture and Storage

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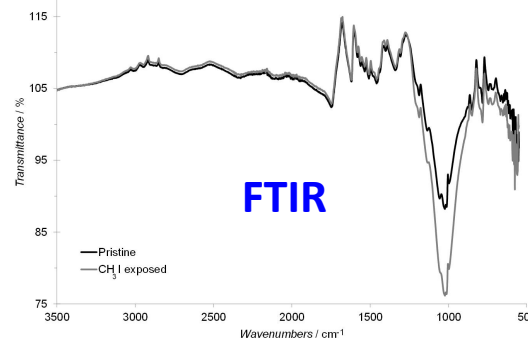
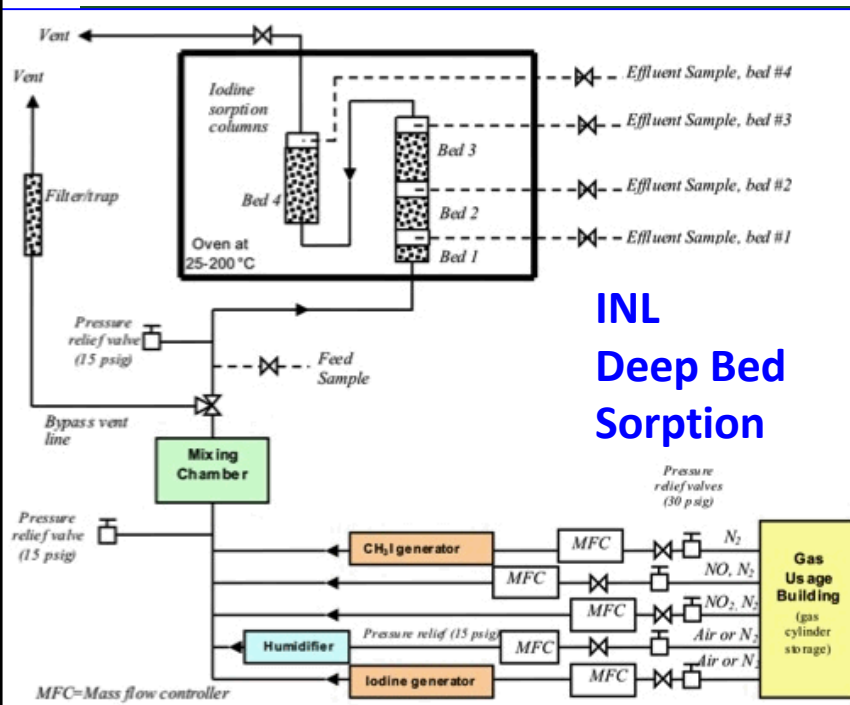


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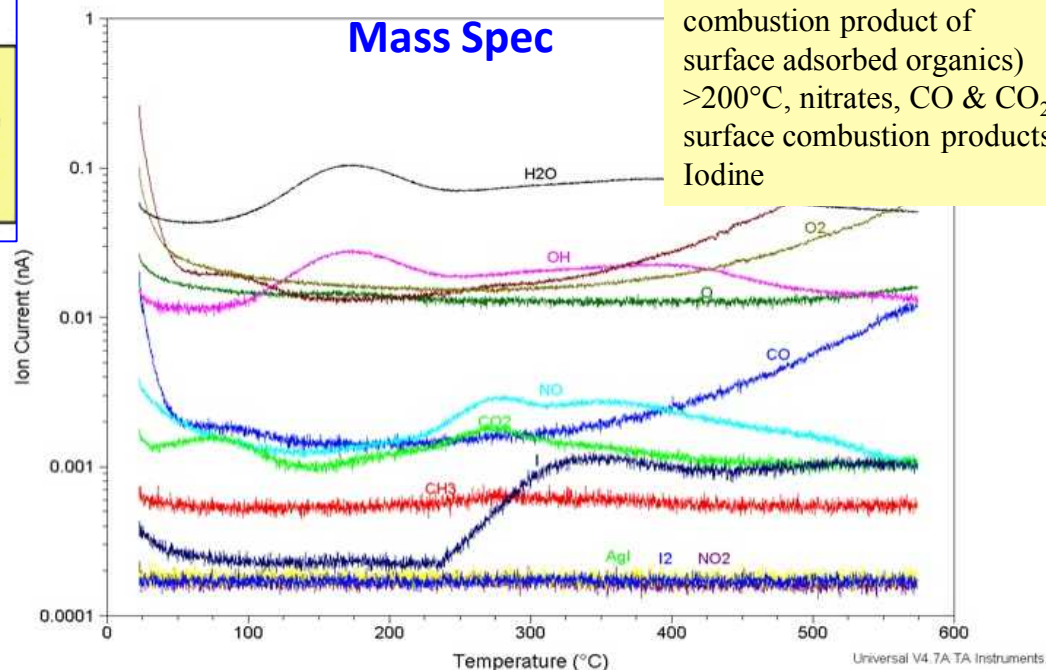
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# CH<sub>3</sub>-I Deep Bed Sorption Studies (H<sub>2</sub>O, NO, NO<sub>2</sub>, CH<sub>3</sub>-I): Iodine adsorbs as AgI after *Catalytic Cleaving* of CH<sub>3</sub>- on MOR – Milestone M4FT-14SN0312041

(T.M.Nenoff (SNL), N.Saelberg (INL), K.W.Chapman (ANL))

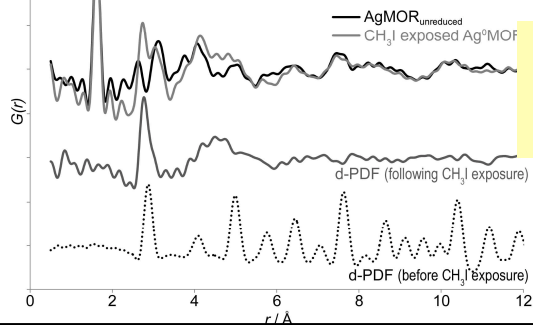


No change  
Pre and post  
CH<sub>3</sub>-I interaction,  
No organic present



< 200°C, surface waters and  
CO<sub>2</sub> removal (CO<sub>2</sub>  
combustion product of  
surface adsorbed organics)  
>200°C, nitrates, CO & CO<sub>2</sub>  
surface combustion products,  
Iodine

## Synchrotron d-PDF



Only AgI  
seen in MOR  
pores

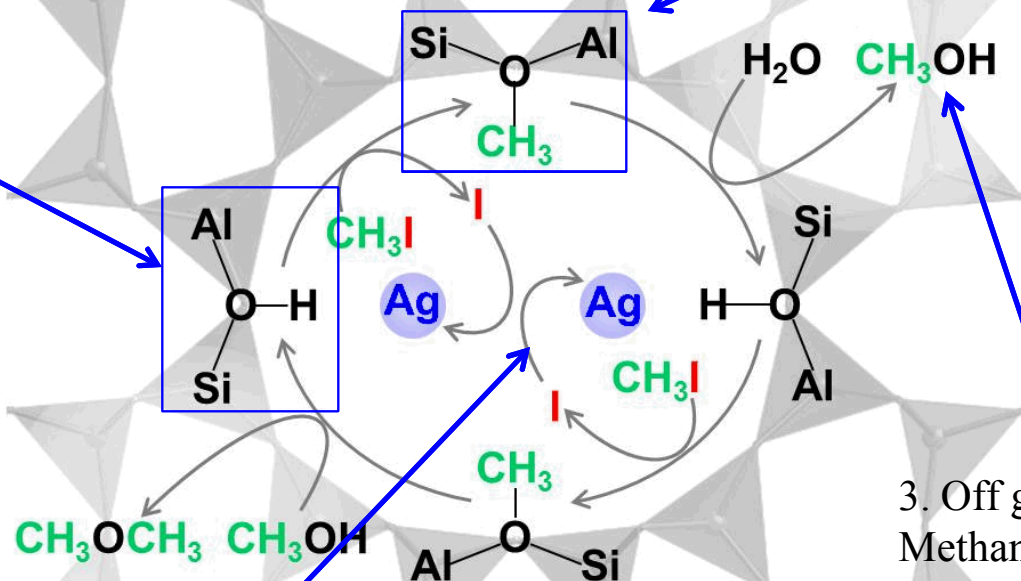


# Mechanism Determined for Iodine Adsorption from INL Stream: $\text{CH}_3\text{-I}$ Cleaving and Iodine Sorption Over Ag-MOR

1. Original Zeolite Acid Site:  
 $\text{H}^+$  charge-balances  
framework  
when  $\text{Ag}^+$  formed

2. SMS: Surface  
Methoxy Species  
formation

Y. Jiang, et.al, *J. Am. Chem. Soc.*,  
128 (2006) 11679.



3. Off gas, downstream:  
Methanol  
Dimethyl Ether  
Methyl Nitrates  
Some  $\text{I}_2$  (noncaptured by Ag)

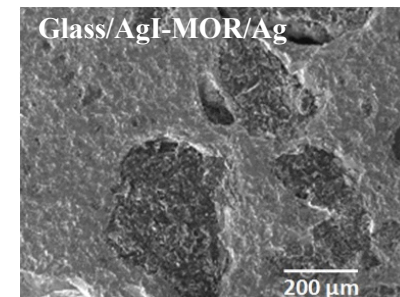
4.  $\text{AgI}$  formation  
in pore of MOR

# Determine Minimum Ag flake Addition to Glass Composite Material (GCM) for AgI-MOR – Milestone M3FT-14SN0312042

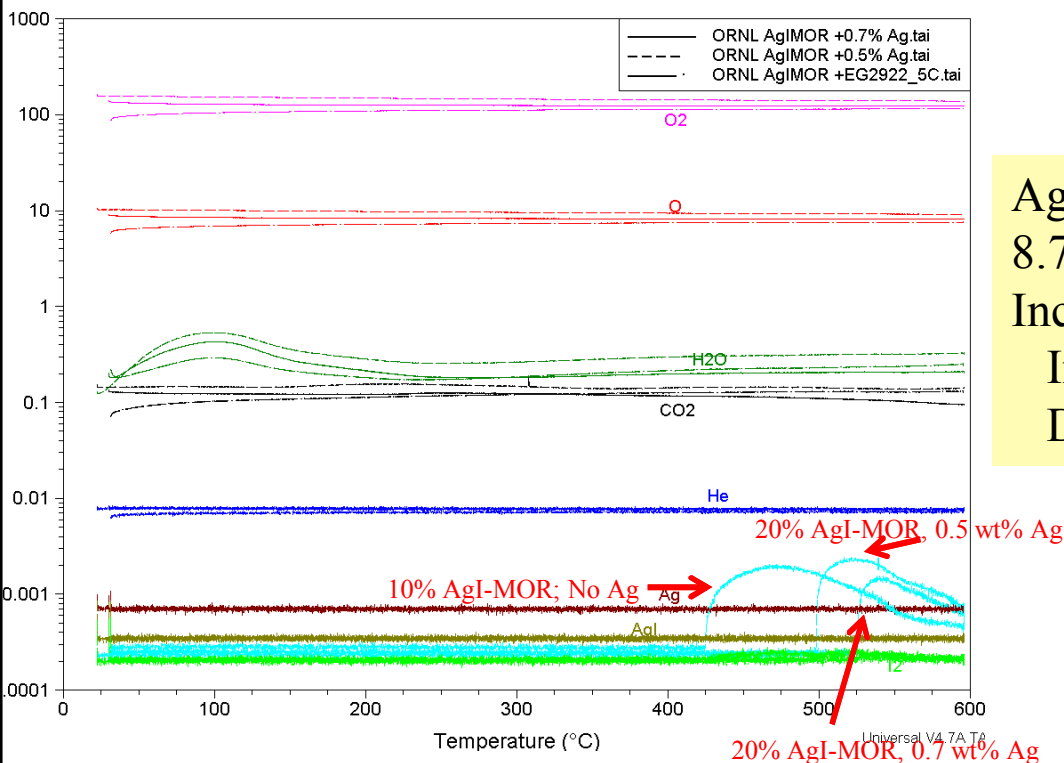
(T.M. Nenoff, T.Garino, M.A.Rodriguez, SNL)

Original Prep: AgI-GCM \*  
AgI-MOR ground in mortar and pestle, then sieved to  $<150\ \mu\text{m}$ .  
Mixed at 20wt% with 80 wt% glass powder and **additional 5wt% Ag flake (to capture any escaping Iodine during GCM sintering)**.  
Uniaxially dry pressed at 70 Mpa in steel die.  
Heated at  $2^\circ\text{C}/\text{min}$  in air to  $550^\circ\text{C}$  for 1hr to sinter.

\*J. Amer. Ceram. Soc. **2011**,94(8), 2412-2419; **US Patent**:



*Complete Physical Encapsulation of AgI by Glass in composite*

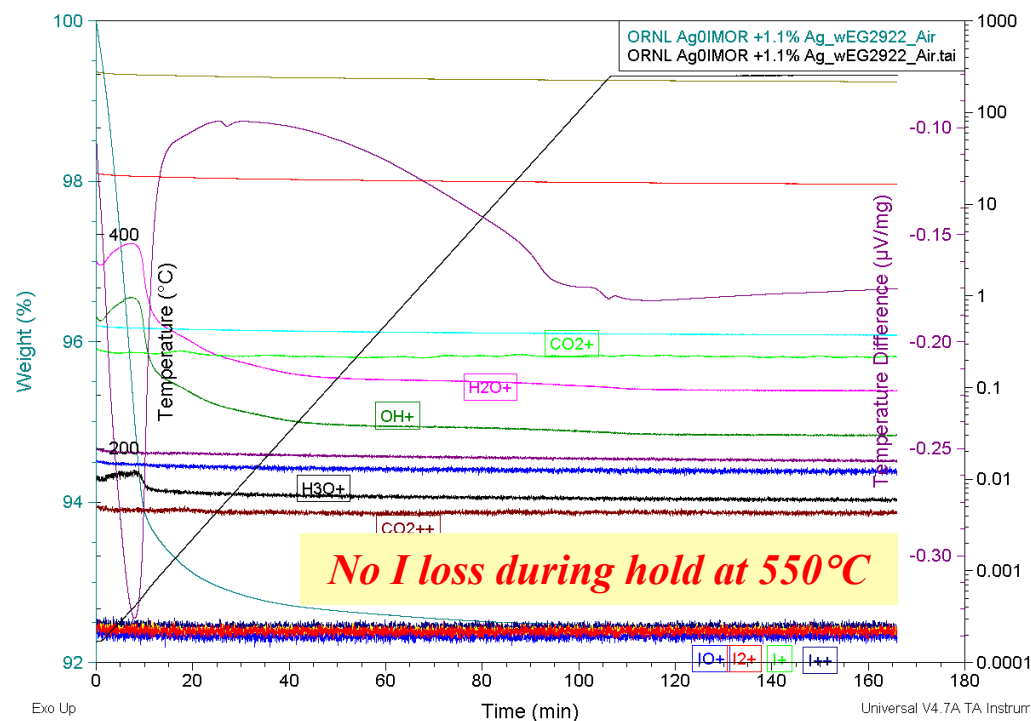


Ag flake Addition Studies to 8.7 wt% Iodine Loaded AgI-MOR (ORNL):  
Increase from Zero (to  $< 5\ \text{wt}\%$ ) indicates:  
Increased temperature of Iodine release,  
Decreased amount of Iodine release.

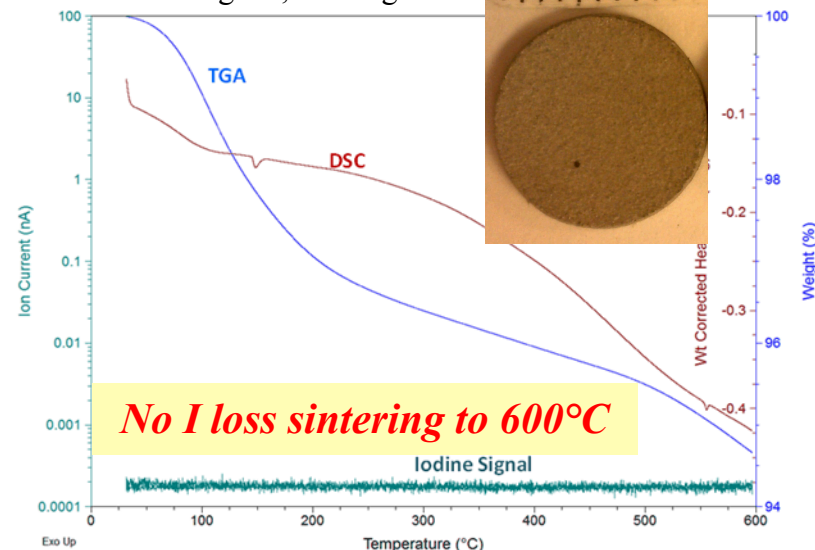
# AgI-MOR GCM Optimized Per extra Ag Flake Addition:

- Sintering in Air: 1.1 wt% Ag Flake Addition
- Sintering in Inert Atmosphere (eg. He): *NO Ag Flake*

**Sintering in Air:** 5°C/min to 550°C for 1 hr  
20 wt% ORNL AgI-MOR + 80 wt% glass + 1.1 wt% Ag Flake



**Sintering in Inert:**  
20 wt% ORNL AgI-MOR +  
80 wt% glass; NO Ag



## Preliminary Results

GCM sintered in *inert atmosphere*

- 1) No need for Ag flake
- 2) No oxidation of Ag in AgI-MOR
- 3) Transport of AgI from bulk into MOR pores (added physical retention of Iodine)

## On-Going GCM Optimization Research:

- Durability studies on GCM sintered in inert with no added Ag
- Optimize maximum loadings (AgI-MOR) possible for durable waste forms
- Study the effect of particle size of iodine loaded AgI-MOR on final GCM waste form (eg., phase purity, durability)