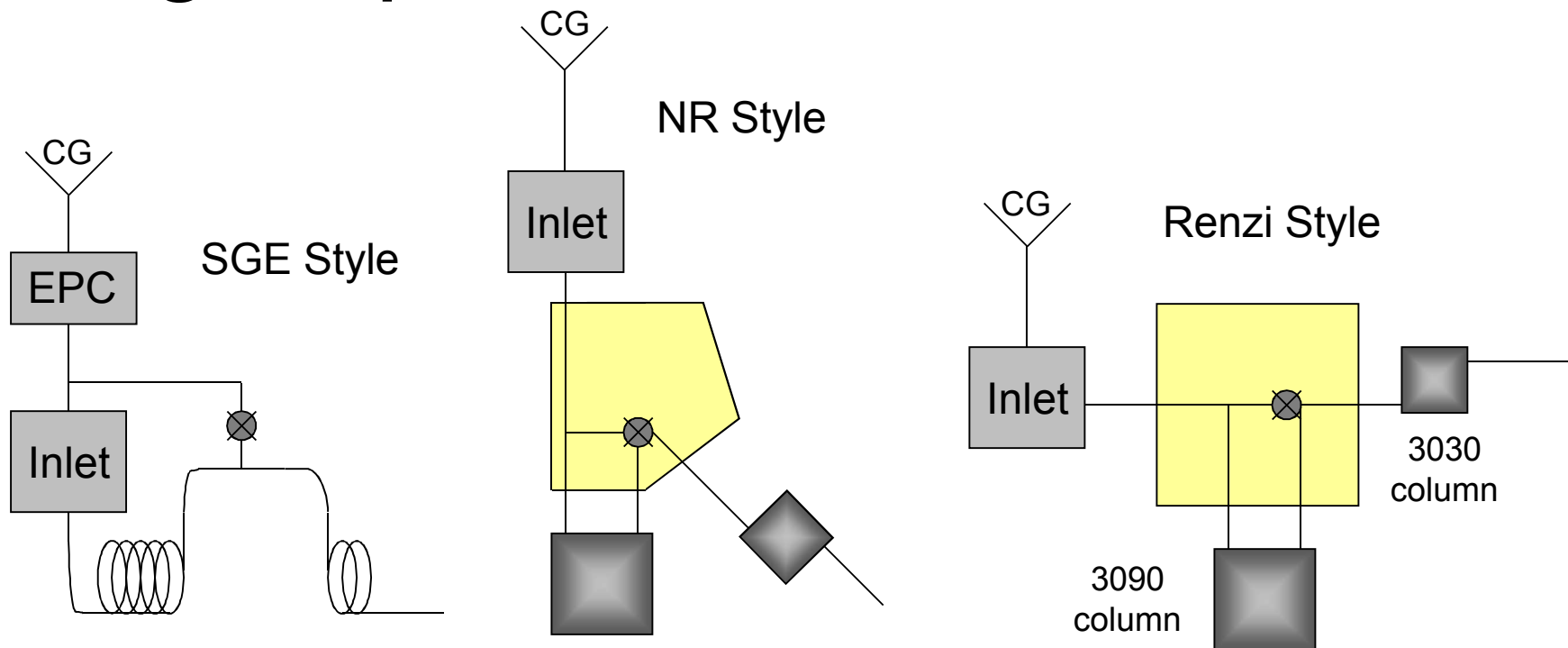


# High-Speed Valve Modulation Studies

# High-Speed Valve Modulators

- For high-speed and portable applications, valve modulation is much more feasible than thermal modulation
- Focusing capability lost – modulation capability becomes analyte independent
- Less resource intensive – only electric power needed
- Aiming for 120 peak capacity in 4 seconds of analysis

# High-Speed Valve Flow Paths

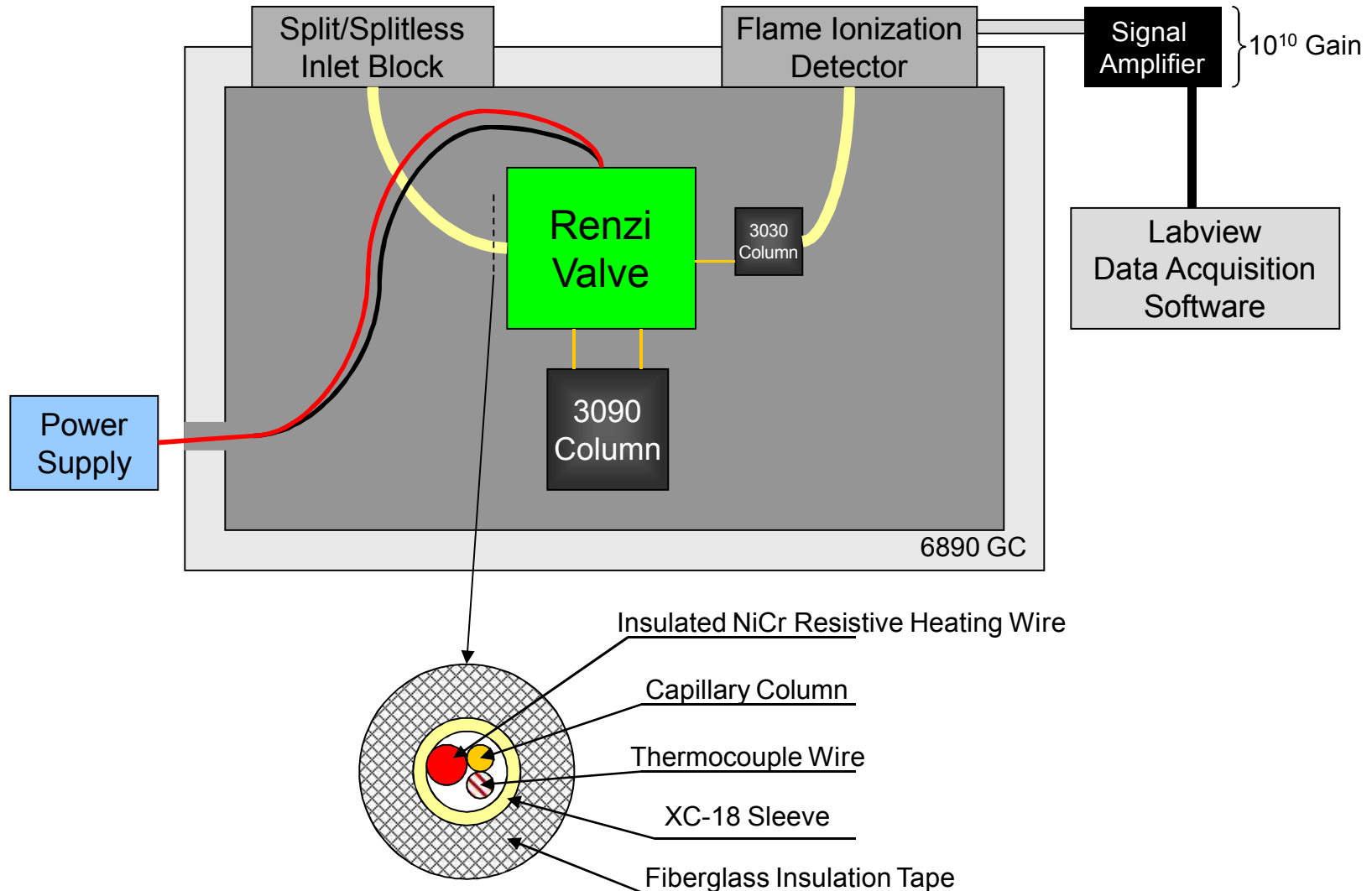


- Valve is outside sample path
- Sample must travel through 1<sup>st</sup> column
- Modulation capability = 100 ms period

- Valve is part of sample path
- Sample can avoid 1<sup>st</sup> column
- Modulation capability = 300 ms period

- Valve is usually outside sample path
- Sample can avoid 1<sup>st</sup> column
- Modulation capability = 240 ms period

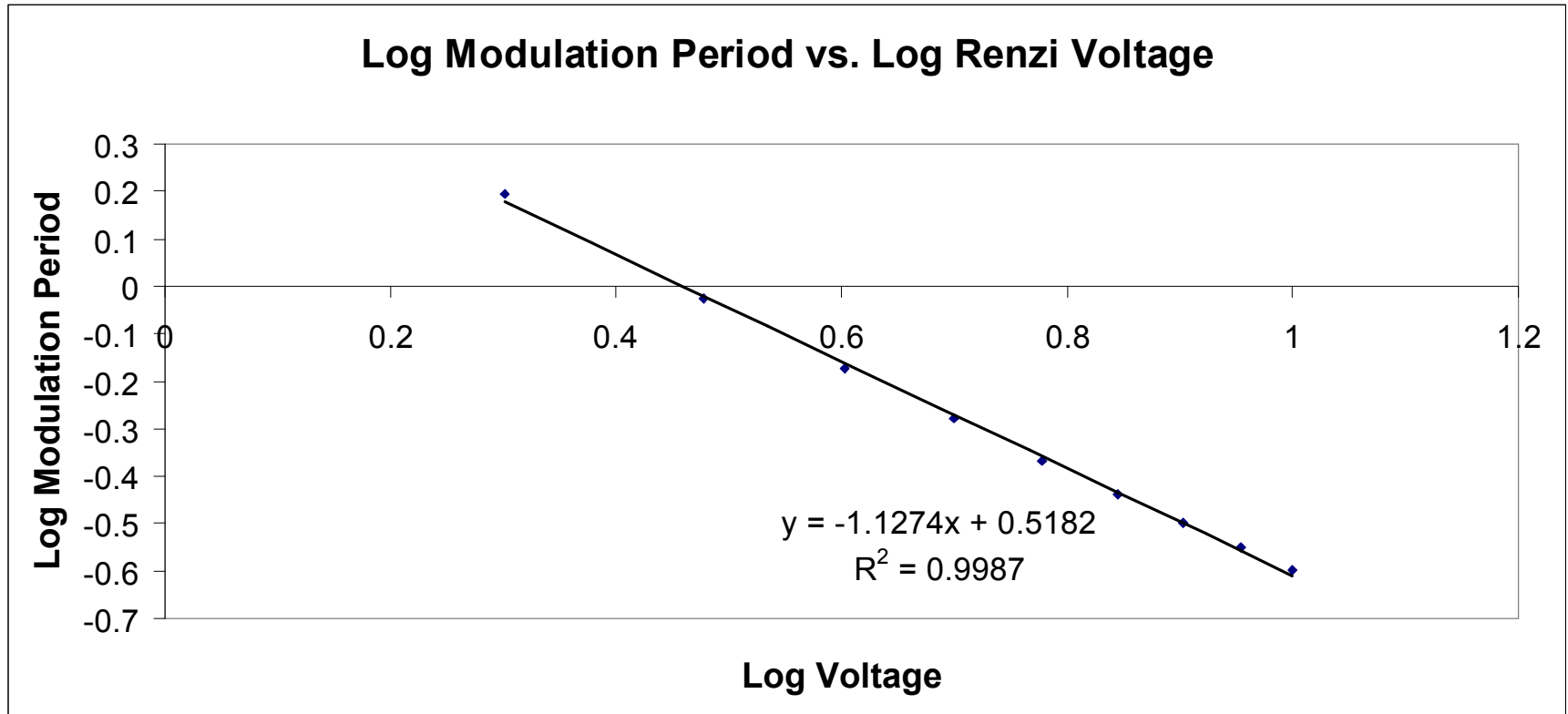
# System Diagram



# System Parameters

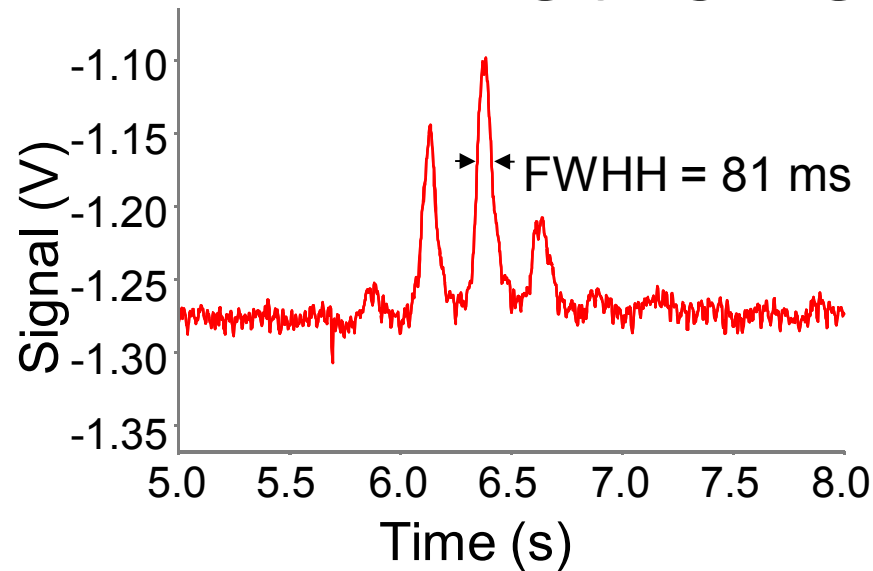
- Oven Setting = 25° C
- Microcolumns Temperature Ramp = 25° to 100° C in 10 s
- Inlet Pressures vary from 10 to 100 psig
- Applied Voltages from 0.5 to 10.0 V
- Gas injection sizes = 1  $\mu\text{L}$
- Liquid injection sizes = 0.2  $\mu\text{L}$
- Test Samples
  - Renzi Valve Modulation Period Test: Air
  - Methane Modulation Test: Methane
  - Peak Area Conservation Test: *n*-decane in  $\text{CS}_2$
  - Peak Capacity Tests: *n*-octane, *n*-decane, *n*-dodecane in  $\text{CS}_2$
  - 18-component Separation: See chart

# Modulation Period as a Function of Applied Valve Voltage Study

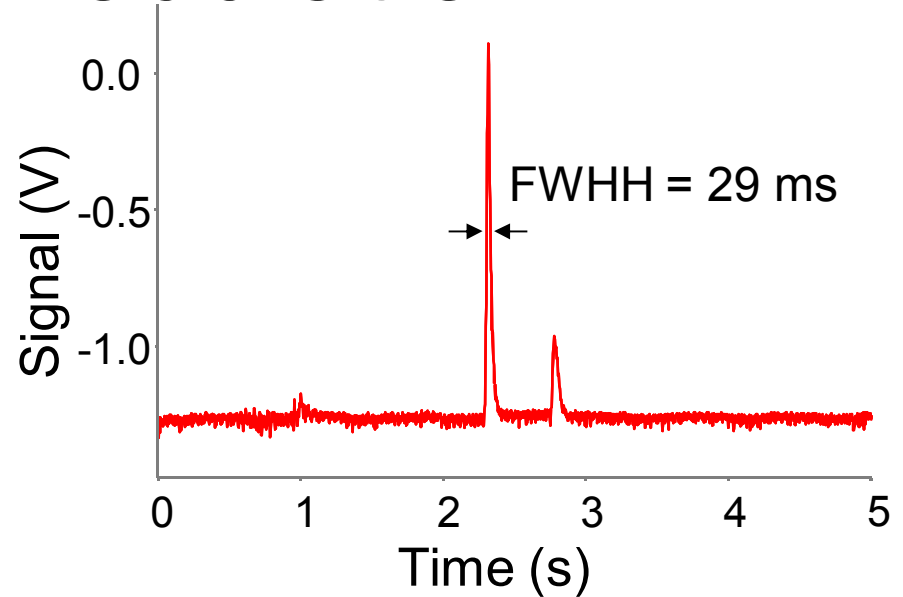


- Increasing applied voltage increases valve speed and summarily reduces the modulation period

# Methane Modulation

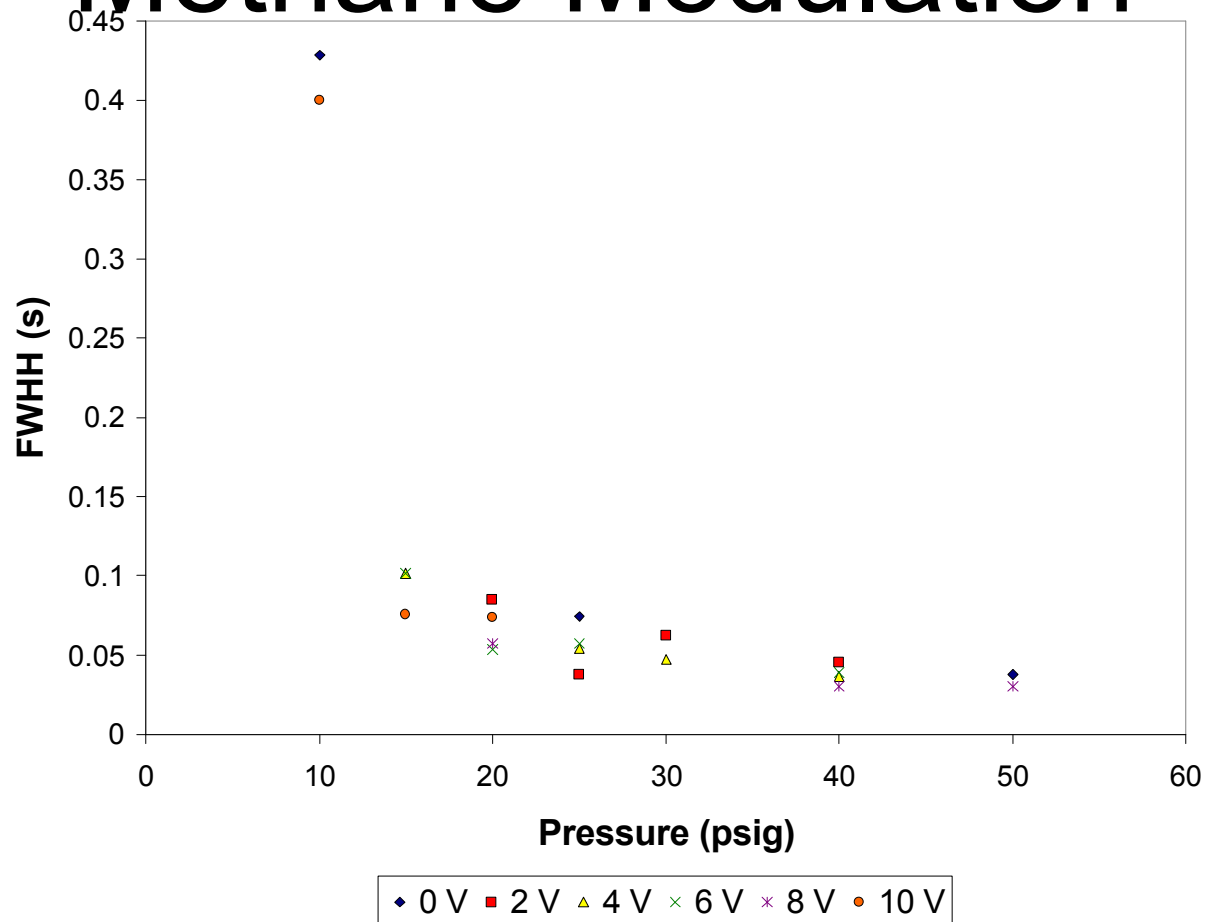


- 15 psig inlet pressure and 10.0 V applied voltage
- Good methane modulation peak profile



- 40 psig inlet pressure and 4.0 V applied voltage
- Fewer peak slices with reduced applied voltage
- Narrower peak width resulting from higher inlet pressure

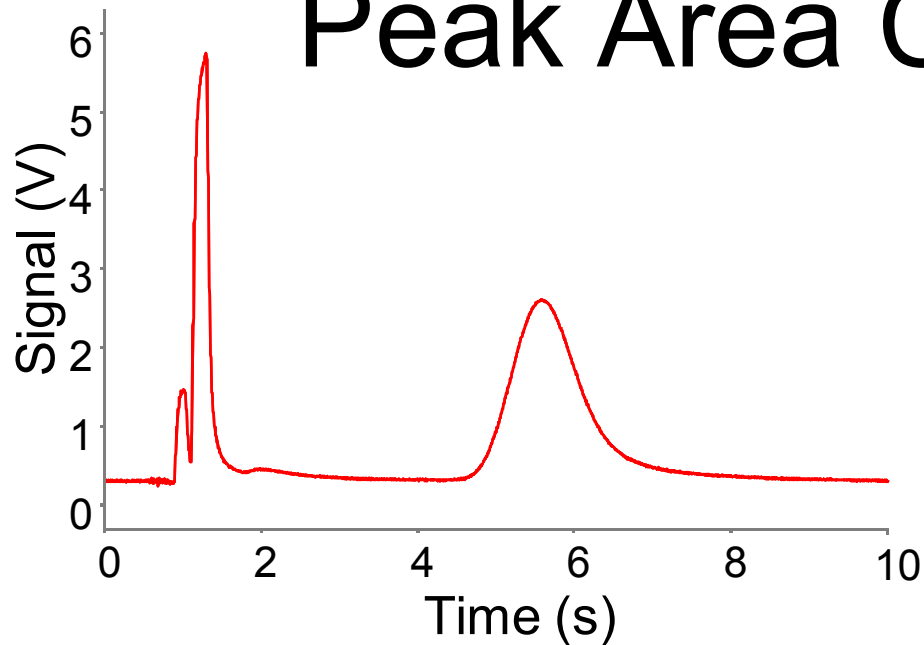
# Methane Modulation



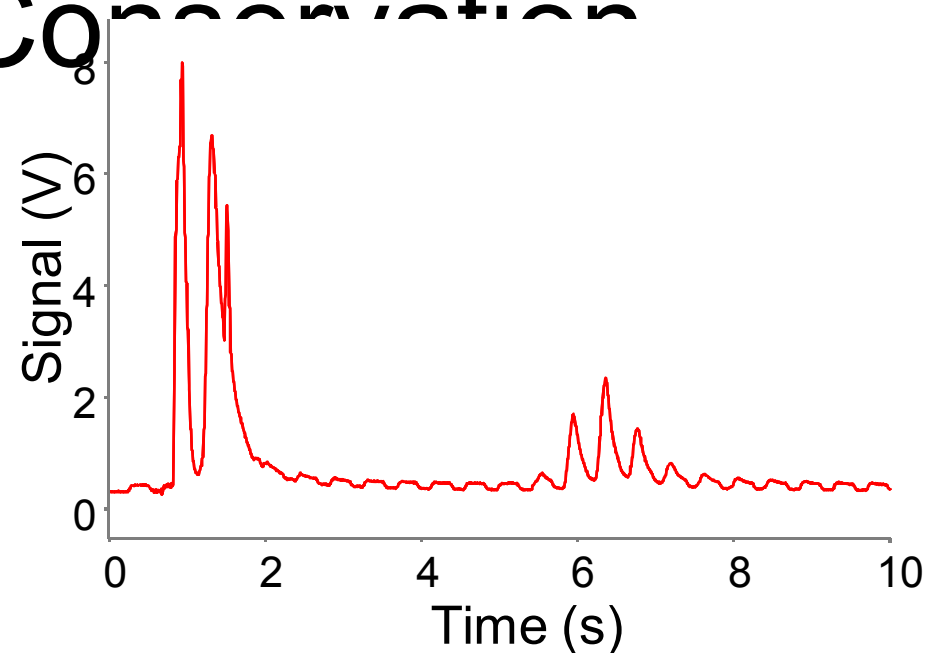
- Peak width appears to depend on inlet pressure strongly and is fairly independent of applied voltage



# Peak Area Compensation



- 100 psig inlet pressure and no voltage applied
- Dodecane peak shape is as expected
- Slight amount of dodecane passed through valve and eluted early



- 100 psig inlet pressure and 6.0 voltage applied
- Dodecane modulated peak shape exhibits classic profile
- Significant portion of dodecane passed through valve and eluted early (asymmetric peak around 1.2 s)
- That peak area added to modulated peak slice area

# ANOVA Test

- Data1\_A = Valve Off
- Data1\_B = 6.0 V
- Data1\_C= 8.0 V
- Data1\_D= 10.0 V
- One-Way ANOVA

## Summary Statistics

| Dataset | N | Mean    | SD      | SE      |
|---------|---|---------|---------|---------|
| Data1_A | 3 | 2.2871  | 0.18463 | 0.10659 |
| Data1_B | 3 | 2.47701 | 0.20059 | 0.11581 |
| Data1_C | 3 | 2.22148 | 0.10496 | 0.0606  |
| Data1_D | 3 | 2.09279 | 0.24454 | 0.14118 |

Null Hypothesis: The means of all selected datasets are equal

Alternative Hypothesis: The means of one or more selected datasets are different

## ANOVA

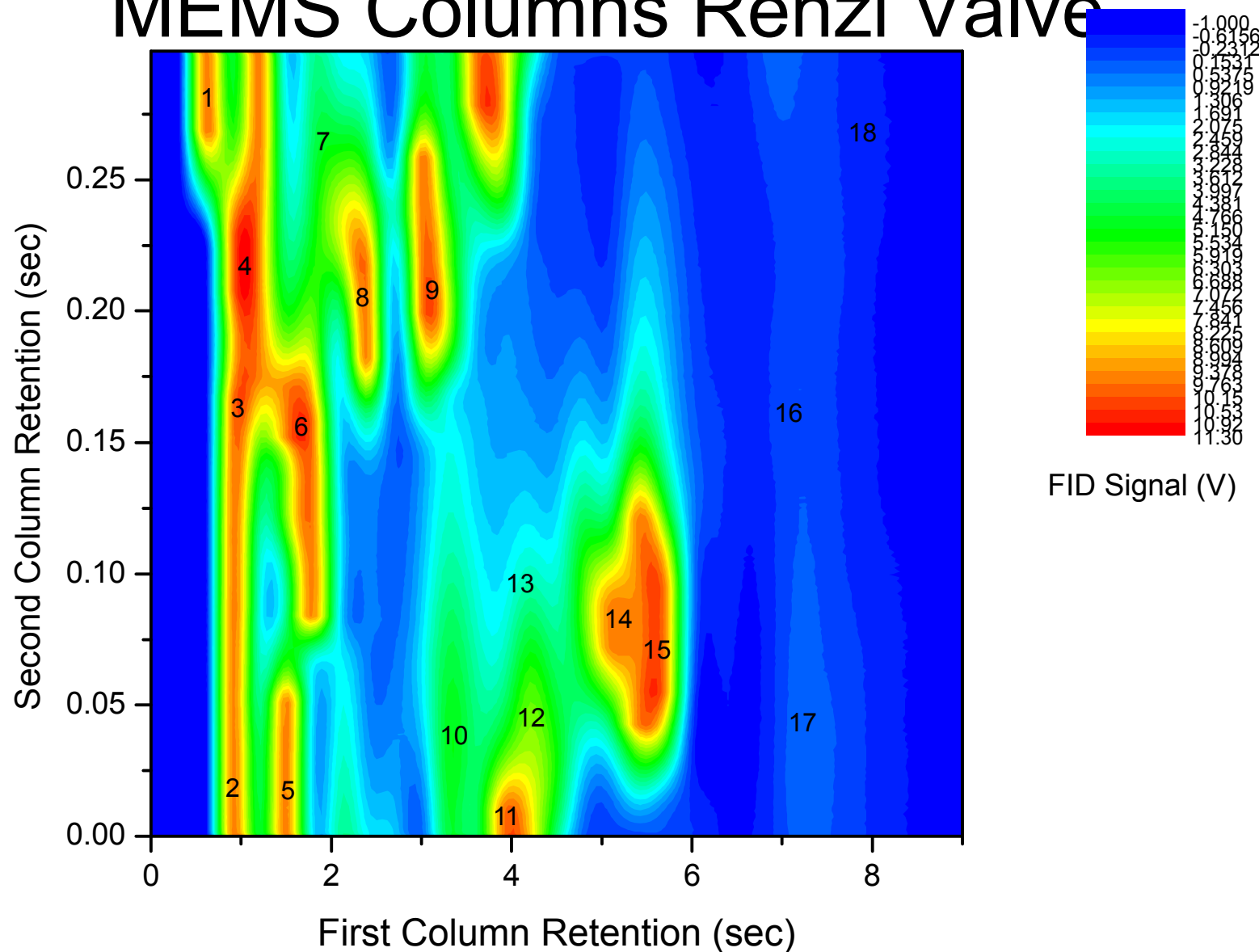
| Source | DoF | Sum of Squares | Mean Square  | F Value | P Value |
|--------|-----|----------------|--------------|---------|---------|
| Model  | 3   | 0.230711597    | 0.0769038658 | 2.11944 | 0.17601 |
| Error  | 8   | 0.290279458    | 0.0362849322 |         |         |

At the 0.0001 level,  
the population means are not significantly different.

# 18 Compound GCxGC Separation Chemicals

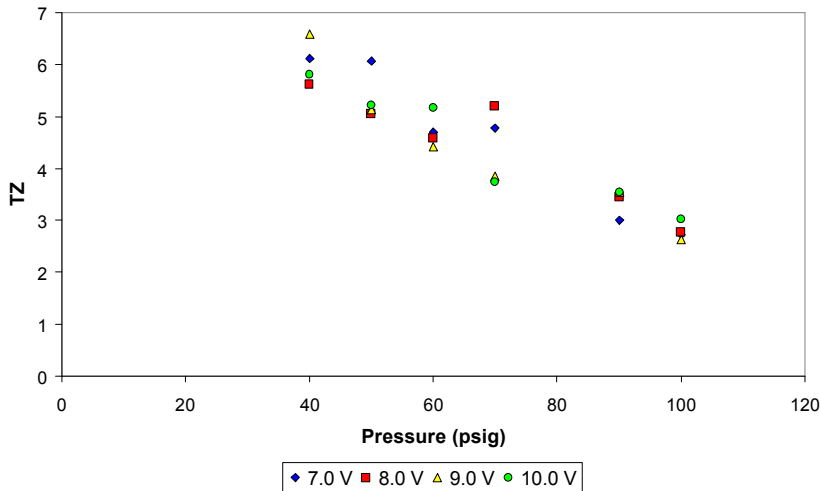
| Sample # | Chemical Name                         | Boiling Point (° C) |
|----------|---------------------------------------|---------------------|
| 1        | carbon disulfide                      | 45.9                |
| 2        | toluene                               | 111                 |
| 3        | <i>n</i> -octane                      | 126                 |
| 4        | iso-octane                            | 99                  |
| 5        | 1,3-dichloropropane                   | 121                 |
| 6        | dimethyl methylphosphonate            | 187                 |
| 7        | 1-octanol                             | 195                 |
| 8        | 1,4-dichlorobutane                    | 154                 |
| 9        | <i>n</i> -decane                      | 174                 |
| 10       | diisopropyl methylphosphonate         | 219                 |
| 11       | di- <i>n</i> -butyl sulfide           | 189                 |
| 12       | 2-chloroethyl ethyl sulfide           | 156                 |
| 13       | 1,6-dichlorohexane                    | 204                 |
| 14       | <i>n</i> -dodecane                    | 216                 |
| 15       | O,S-diethyl methylphosphonothioate    | 78*                 |
| 16       | diisobutyl methylphosphonate          | 254                 |
| 17       | 2-chloroethyl phenyl sulfide          | 245                 |
| 18       | O,S-diisobutyl methylphosphonothioate | 139*                |
|          |                                       | *= 12 mm Hg         |

# 18 Compounds GCxGC Separation MEMS Columns Renzi Valve

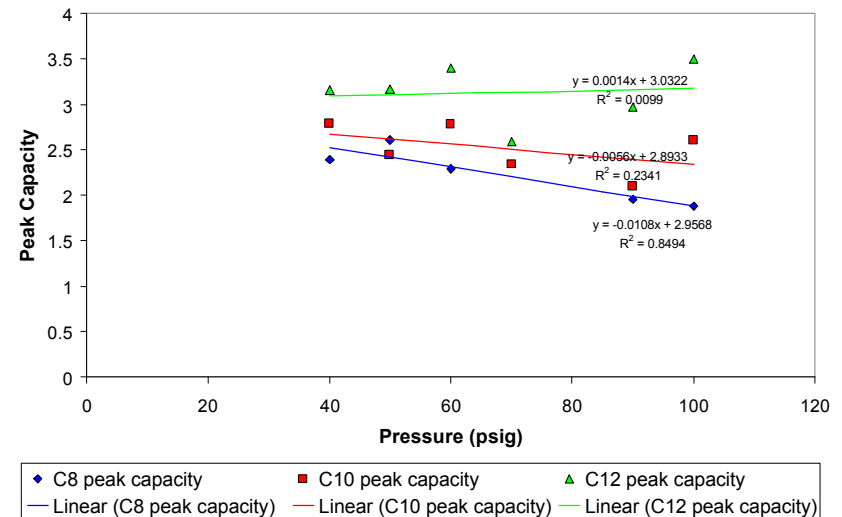


# Peak Capacity Plots

C8-C12 TZ vs. Inlet Pressure for Various Valve Applied Voltages



Max Peak Capacity vs. Inlet Pressure for 9.0 V



- 1<sup>st</sup> dimension peak capacity calculated using Trenzall (TZ) numbers
- Strong correlation with inlet pressure, little correlation with modulation period

- 2<sup>nd</sup> dimension peak capacity calculated using peak capacity equation
- Little correlation with inlet pressure though some correlation to analyte

# Conclusions

- Renzi valve can successfully be used for fast GCxGC analysis with complex mixtures
- Further improvements can be made
  - Reduce dead volume to improve peak capacity through narrower second-column band-widths.
  - Phase control necessary for reproducibility, wrap-around detection, and prevention of analyte passing through valve
- Compounds of very low boiling point such as methane susceptible to modulation