



Metal-Organic Frameworks (MOFs)

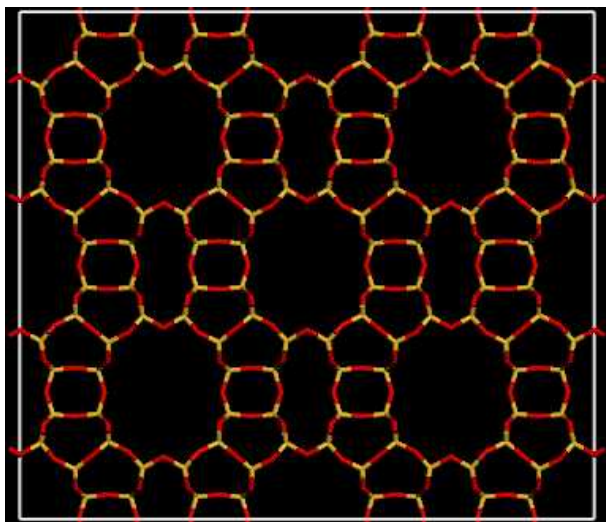
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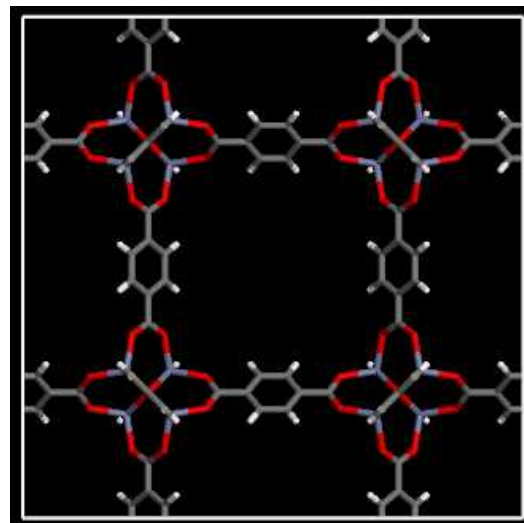
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Metal-Organic Frameworks (MOFs)

Sandia team: Mark Allendorf, Jeffery Greathouse, Patrick Doty



Mordenite (zeolite), $\text{Na}(\text{AlO}_2)(\text{SiO}_2)_{10}$



MOF-5, $\text{Zn}_4\text{O}(\text{C}_6\text{H}_4\text{O}_4)_3$

	Mordenite	MOF-5
Pore apertures	6.7 Å x 7.0 Å 2.9 Å x 5.7 Å	11.2 Å x 11.2 Å ➡ Gas diffusivity 100-1000X > molecular sieves ➡ faster, lower pressure PSA
Density	1.6 g/cm ³	0.6 g/cm ³ ➡ Lighter, more portable
Surface area	300 m ² /g	3500 m²/g ➡ Highest gas sorption capacity

A Cu-based MOF selectively adsorbs Kr and Xe in air samples

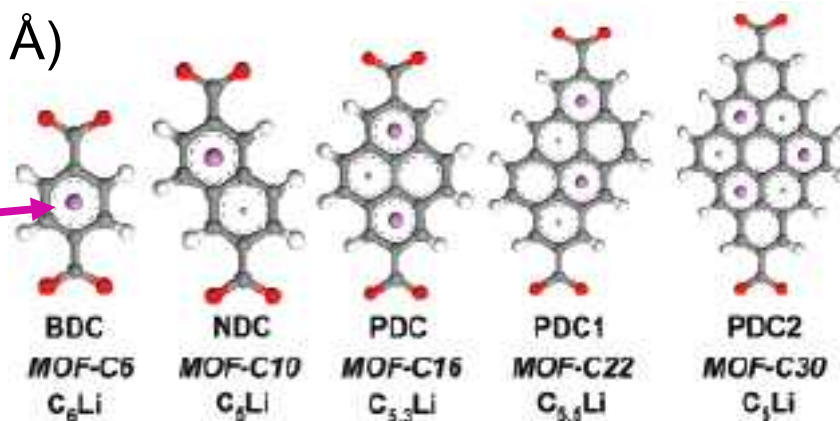
guest	y (gas) ^a	x (CuBTC) ^b	compression ratio (# adsorbed / # gas)
N ₂	0.797	0.773	4.1
O ₂	0.200	0.215	4.5
Xe	1.66×10^{-4}	2.80×10^{-3}	71.0
Kr	2.20×10^{-3}	8.87×10^{-3}	16.9

Greathouse et al, SAND Report **2008**

Kr adsorption may be enhanced by:

- Tailoring a MOF with pore aperture in between the atomic diameters of Kr (3.6 Å) and Xe (4.0 Å)
- Metal doping at linker ring sites

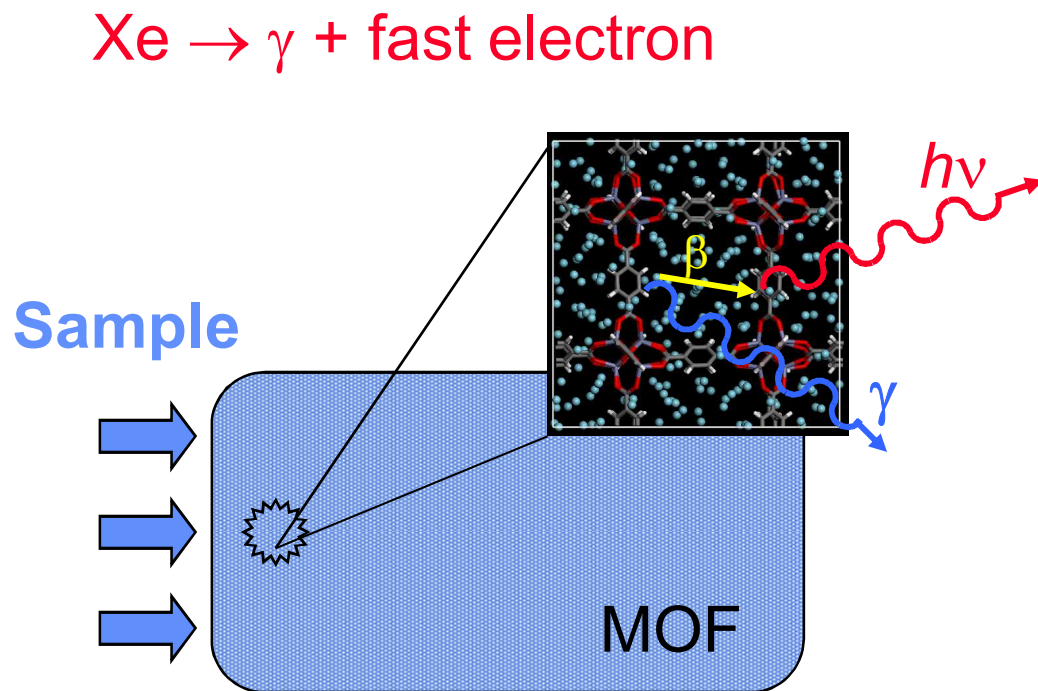
Li atom




Li-doped MOF's enhance H₂ adsorption.
Han and Goddard, *J. Am. Chem. Soc.* **2007**

Simultaneous isotope collection and detection is feasible with Sandia's scintillating MOFs

- High surface-area MOF adsorbs isotopes
- Isotopes emit, yielding $\beta + \gamma$
- γ escapes and is detected
- β particle collides w/ fluor in MOF, causing it to scintillate
- Enables immediate detection of short-lived isotopes
- Eliminates need for optically coupled phoswich detector
- ➔ increases detection efficiency
- Sample:
 - Preconcentrated by PSA or
 - Concentrated in situ with simultaneous detection



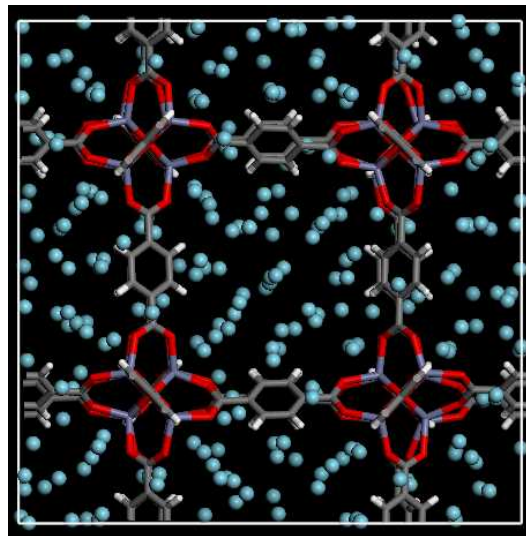
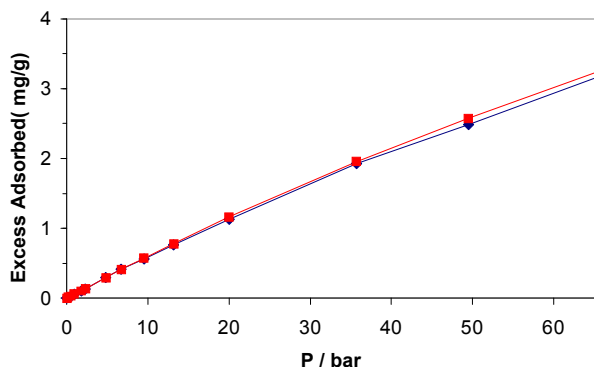
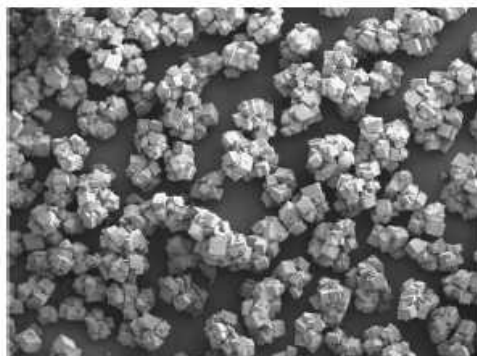


NA-22 FY2009 plutonium detection proposal: noble gas collection and detection

Objective: exploit the structural versatility of nanoporous framework materials (NFM) and their unique physical properties to create multifunctional materials for noble gas collection and detection

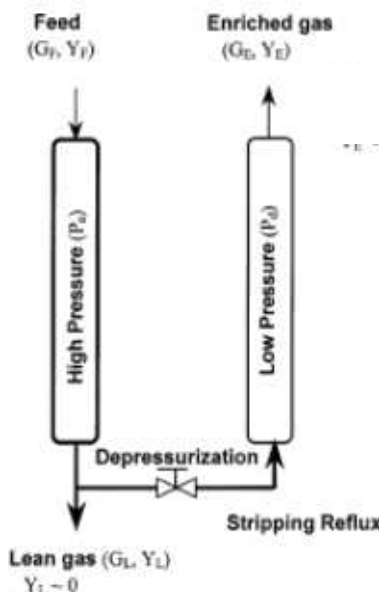
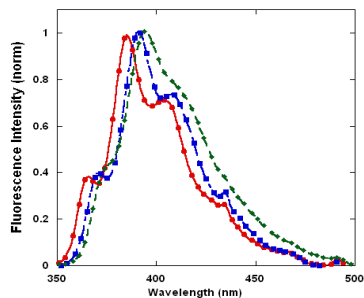
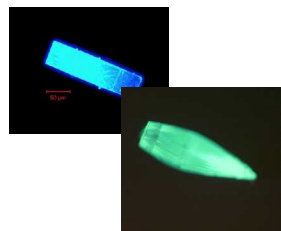
- *Atomistic modeling to guide synthesis through rational design*
 - Adsorption isotherms
 - Interferences and competitive adsorption
 - Temperature and pressure effects
- *Synthesis: Optimize NFM through targeted modifications*
 - Goal 1: High surface-area, tailored sorbents for smaller, faster, low-pressure systems
 - Goal 2 (stretch goal): Nanoporous, scintillating NFM for simultaneous collection/adsorption
 - MOFs as starting point: synthetically versatile high-surface area NFM
- *Materials screening, transport properties, scaleup issues*
 - Microporosimetry: surface areas, pore volumes, isotherms
 - PSA: bench-scale simulation of separation process
 - Electron-beam to simulate beta radiation: quantify light output and timing

Sandia's established program in NFM has all the capabilities required to carry out this project



Synthesis & Characterization

- Hoods
- Glove boxes
- Ovens
- FTIR
- XRD
- NMR
- Fluorescence
- Raman
- SEM/TEM



Computational modeling

- Molecular dynamics
- Monte Carlo methods
- Quantum chemistry
- Massively parallel machines

Transport measurements

- PSA
- Microporosimeter