



Metal-Organic Frameworks (MOFs)

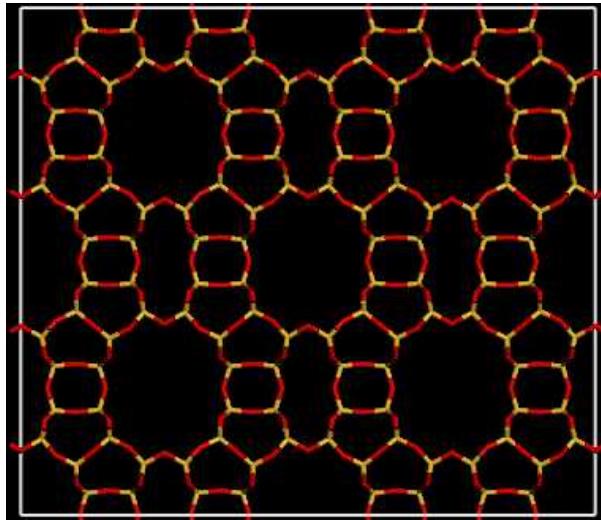
SAND2008-7391P

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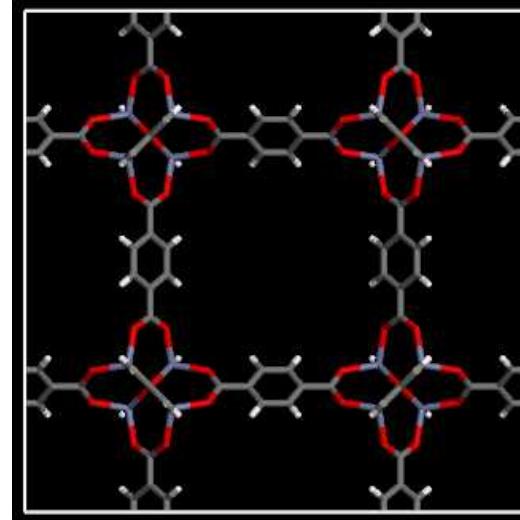
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under contract DE-AC04-94AL85000.

Metal-Organic Frameworks (MOFs)

Sandia team: Mark Allendorf, Jeffery Greathouse, Patrick Doty



Mordenite (zeolite), $\text{Na}(\text{AlO}_2)(\text{SiO}_2)_{10}$



MOF-5, $\text{Zn}_4\text{O}(\text{C}_6\text{H}_4\text{O}_4)_3$

	Mordenite	MOF-5
Pore apertures	$6.7 \text{ \AA} \times 7.0 \text{ \AA}$ $2.9 \text{ \AA} \times 5.7 \text{ \AA}$	$11.2 \text{ \AA} \times 11.2 \text{ \AA}$ ➡ Gas diffusivity 100-1000X > molecular sieves ➡ faster, lower pressure PSA
Density	1.6 g/cm^3	0.6 g/cm^3 ➡ Lighter, more portable
Surface area	$300 \text{ m}^2/\text{g}$	$3500 \text{ m}^2/\text{g}$ ➡ Highest gas sorption capacity



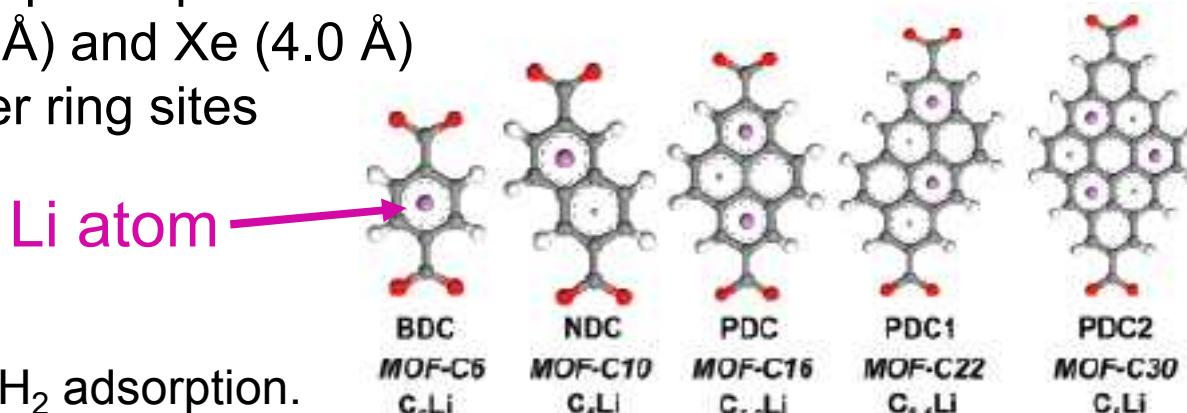
A Cu-based MOF selectively adsorbs Kr and Xe in air samples

guest	y (gas) ^a	x (CuBTC) ^b	compression ratio (# adsorbed / # gas)
N ₂	0.797	0.773	4.1
O ₂	0.200	0.215	4.5
Xe	1.66 x 10 ⁻⁴	2.80 x 10 ⁻³	71.0
Kr	2.20 x 10 ⁻³	8.87 x 10 ⁻³	16.9

Greathouse et al, SAND Report 2008

Kr adsorption may be enhanced by:

- ▶ Tailoring a MOF with pore aperture in between the atomic diameters of Kr (3.6 Å) and Xe (4.0 Å)
- ▶ Metal doping at linker ring sites



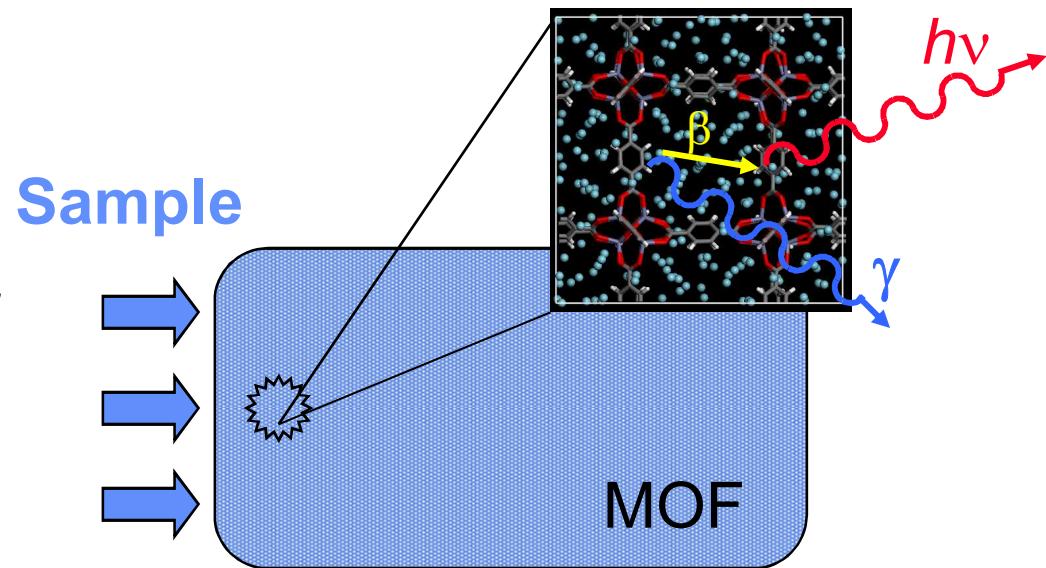
Li-doped MOF's enhance H₂ adsorption.

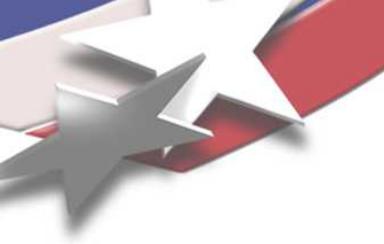
Han and Goddard, *J. Am. Chem. Soc.* 2007



Simultaneous isotope collection and detection is feasible with Sandia's scintillating MOFs

- High surface-area MOF adsorbs isotopes
- Isotopes emit, yielding $\beta + \gamma$
- γ escapes and is detected
- β particle collides w/ fluor in MOF, causing it to scintillate
- Enables immediate detection of short-lived isotopes
- Eliminates need for optically coupled phoswich detector
- ➔ increases detection efficiency
- Sample:
 - Preconcentrated by PSA or
 - Concentrated in situ with simultaneous detection





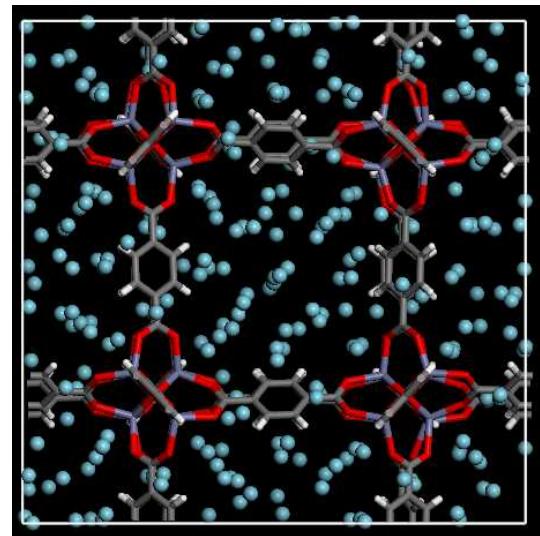
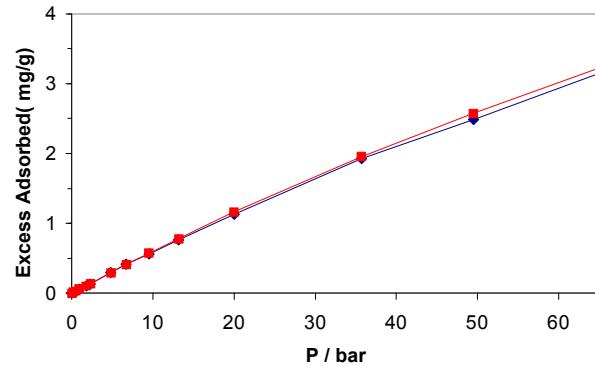
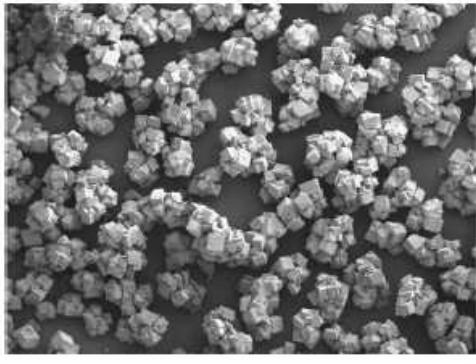
NA-22 FY2009 plutonium detection proposal: noble gas collection and detection

Objective: exploit the structural versatility of nanoporous framework materials (NFM) and their unique physical properties to create multifunctional materials for noble gas collection and detection

- ***Atomistic modeling to guide synthesis through rational design***
 - Adsorption isotherms
 - Interferences and competitive adsorption
 - Temperature and pressure effects
- ***Synthesis: Optimize NFM through targeted modifications***
 - Goal 1: High surface-area, tailored sorbents for smaller, faster, low-pressure systems
 - Goal 2 (stretch goal): Nanoporous, scintillating NFM for simultaneous collection/adsorption
 - MOFs as starting point: synthetically versatile high-surface area NFM
- ***Materials screening, transport properties, scaleup issues***
 - Microporosimetry: surface areas, pore volumes, isotherms
 - PSA: bench-scale simulation of separation process
 - Electron-beam to simulate beta radiation: quantify light output and timing

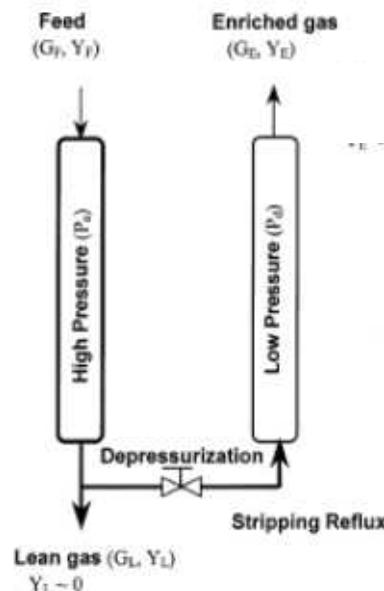
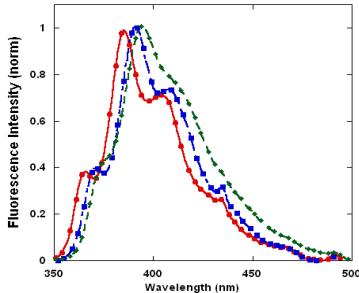


Sandia's established program in NFM has all the capabilities required to carry out this project



Synthesis & Characterization

- Hoods
- Glove boxes
- Ovens
- FTIR
- XRD
- NMR
- Fluorescence
- Raman
- SEM/TEM



Computational modeling

- Molecular dynamics
- Monte Carlo methods
- Quantum chemistry
- Massively parallel machines

Transport measurements

- PSA
- Microporosimeter