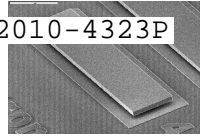




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# **DSMC Algorithms for Moving-Boundary Problems**

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Sandia National Laboratories  
Albuquerque, New Mexico**

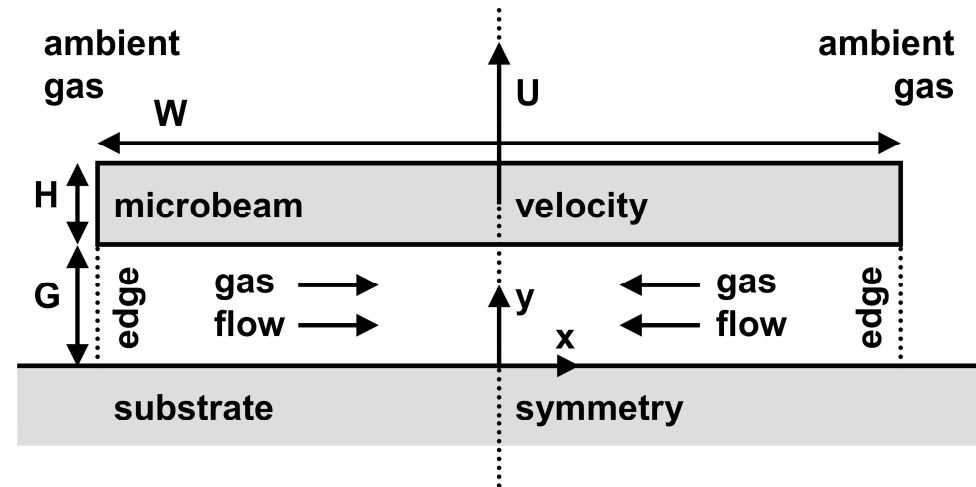
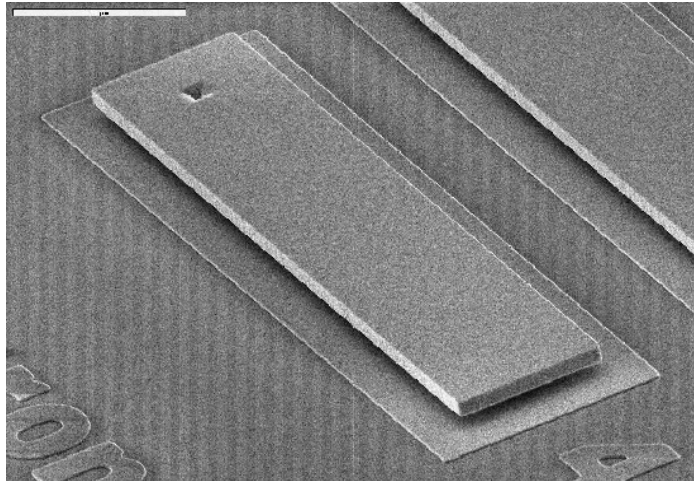
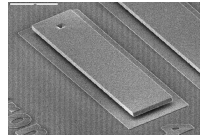
***American Physical Society Division of Fluid Dynamics  
63<sup>rd</sup> Annual Meeting – DFD10  
Long Beach, California; November 21-23, 2010***



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# Motivation



## MicroElectroMechanical Systems (MEMS) devices in air

- Beams often oscillate out-of-plane at high frequencies
- Gas in gap between beam and substrate damps motion

## Gas motion producing damping force is noncontinuum

- Gaps are small:  $\sim 2 \mu\text{m}$  nominal, smaller while closing
- Mean free path:  $\sim 0.07 \mu\text{m}$ , larger in low-pressure package

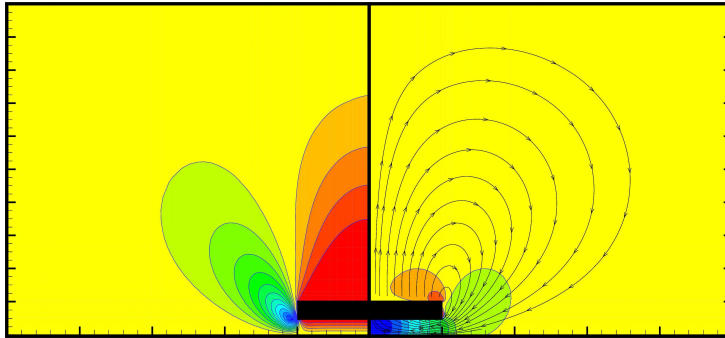
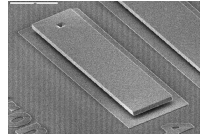
## Gas motion is driven by time-varying geometry

- Closing and opening gaps, gas motion to/from ambient

## Simulate noncontinuum gas with moving object



## Previous Work



Alexeenko et al. (2006): Ellipsoidal Statistical

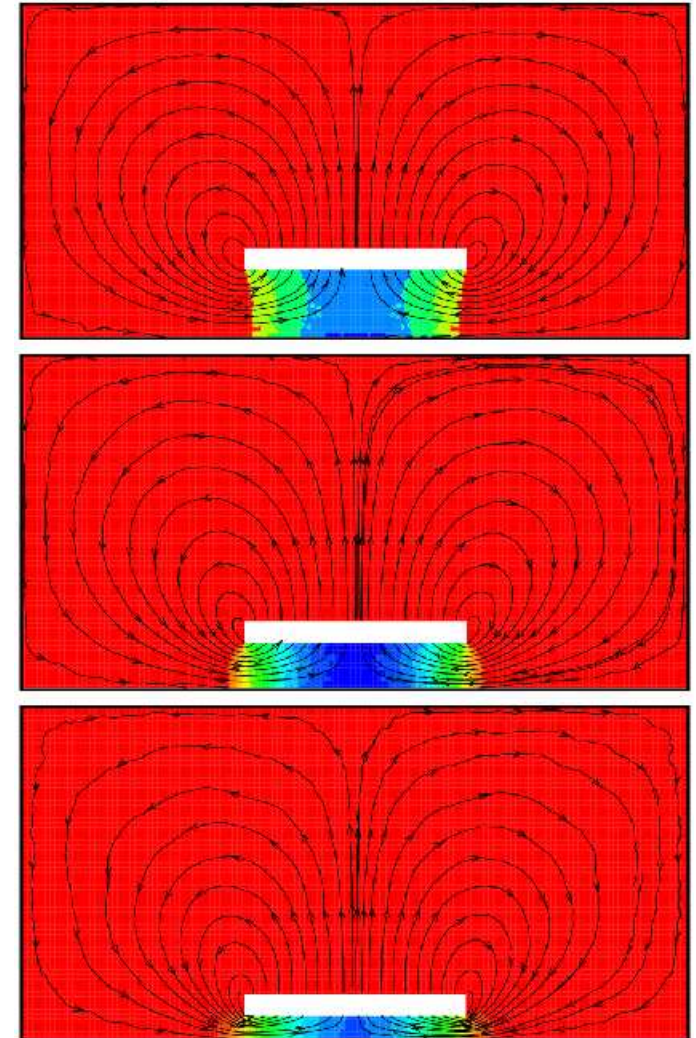
**Several approaches have been used**

**Continuum:** Navier-Stokes slip-jump,  
Torczynski et al. (2002)

**Quasi-static DSMC:** fixed geometries,  
Gallis et al. (2003)

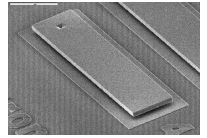
**Simpler physics:** Reynolds equation,  
Gallis and Torczynski (2004)

**Kinetic model:** Ellipsoidal statistical,  
Alexeenko et al. (2006)

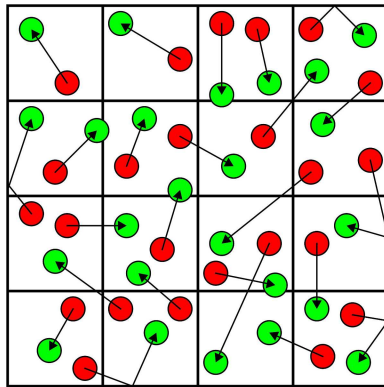


Gallis et al. (2003): Quasi-Static DSMC

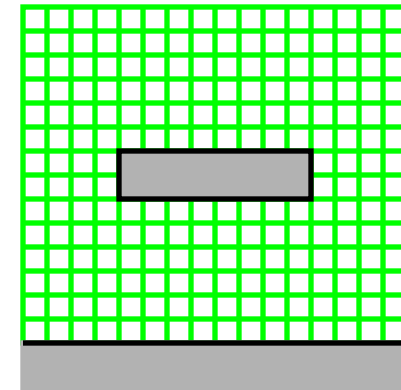
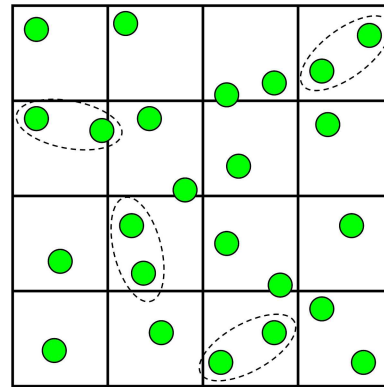
**DSMC for noncontinuum gas with moving object**



## Current Approach



DSMC molecules move over mesh, collide within cells



Object moves over mesh

### Direct Simulation Monte Carlo (DSMC) with moving object

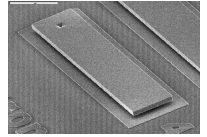
- Object moves over mesh just like molecules do
- Molecules reflect from moving object (diffuse, specular)

### Advantages and disadvantages

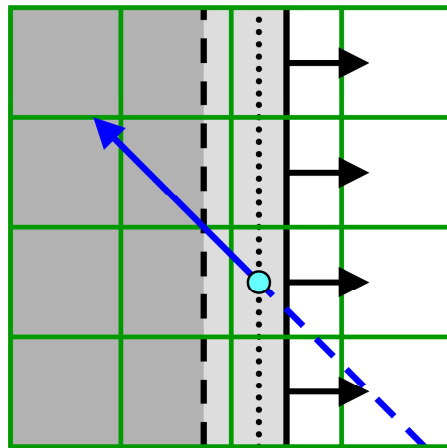
- Gas motion is inherently transient and noncontinuum
- Requires significant computational resources

### Surface fluxes to object are most important quantities

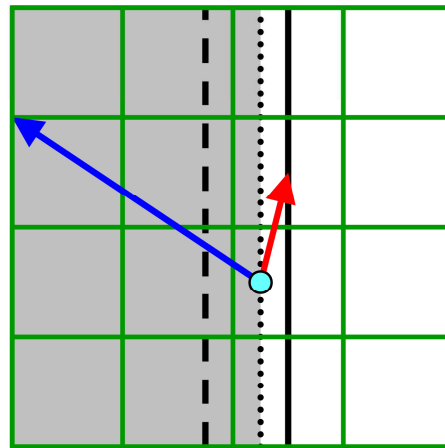
- Gas force on object, gas energy transfer to object
- Can tolerate stochastic noise out in bulk gas



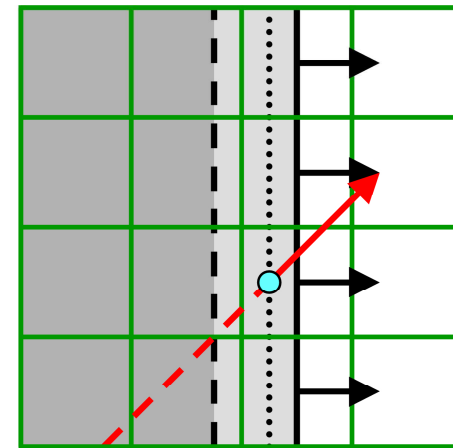
## Rigorous Algorithm



mesh frame  
incident molecule



object frame  
ordinary reflection



mesh frame  
reflected molecule

### Reflection occurs where molecule and object paths intersect

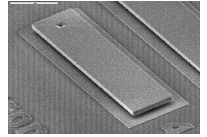
- Molecule & object move with fixed velocities over time step
- Intersection point and remaining time are determined

### Molecule reflection is performed in object reference frame

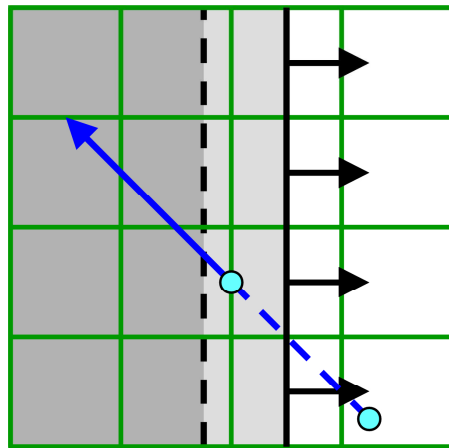
- Object velocity is subtracted from molecule velocity
- Reflected molecule velocity is selected (diffuse, specular)
- Object velocity is added to reflected molecule velocity

### Molecule travels for remainder of time step

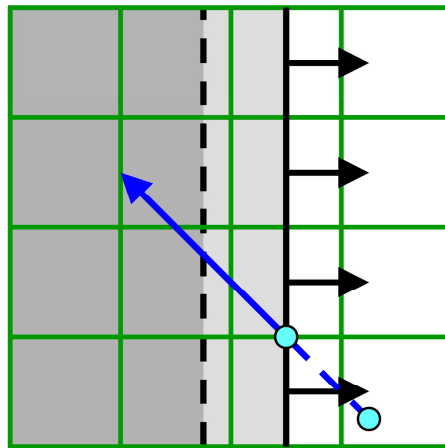




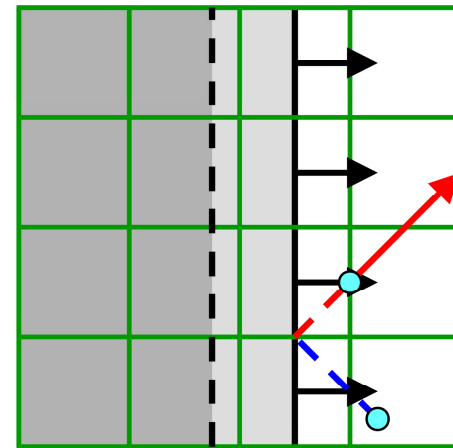
## Approximate Algorithm



object & molecule  
move, overlap



intersect path with  
new object position



molecule reflects  
from new object

### Reflection occurs from envelope of object over time step

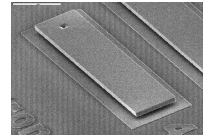
- Advancing/receding: reflect from new/old object position
- Ensures that reflected molecule remains outside object

**If object is slow w.r.t. molecules, approximate → rigorous**

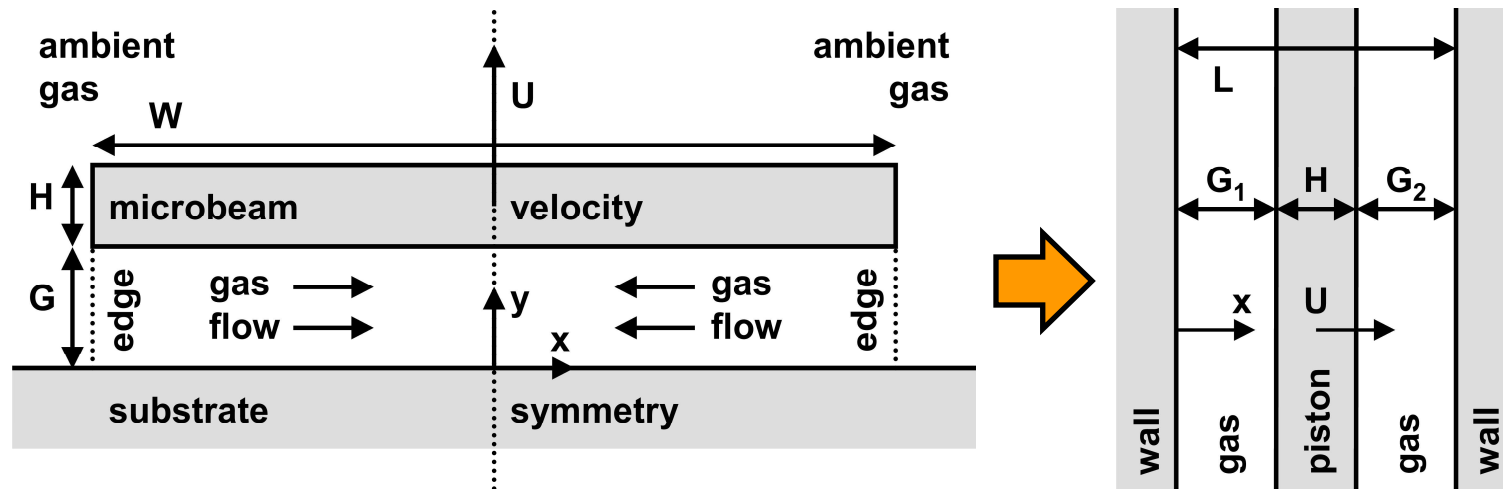
- Molecule's reflected velocity is same as rigorous algorithm

### Advantages and disadvantages

- Faster than rigorous since reflection point is not calculated
- Positions of molecules near surface approximated



## Simplified Geometry



**Initially, implement/assess algorithms for simple geometry**

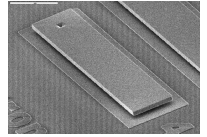
- Need less computational resources, verification is easier

**Replace 2D microbeam cross section with 1D piston**

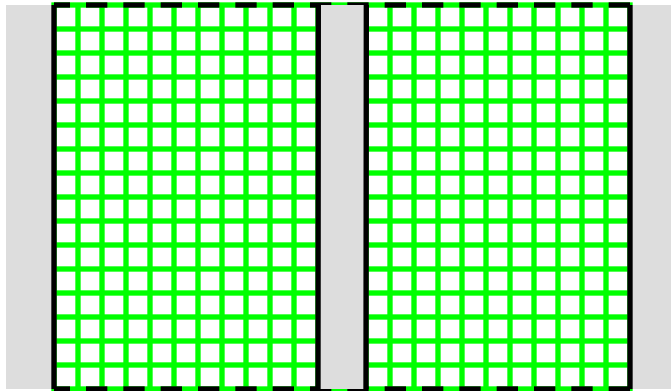
- Piston has opening and closing gaps like microbeam
- Molecule-surface intersection point is easy to determine
- Know early (wave) & long-time behavior for some cases

**Prescribe piston position analytically for all time**

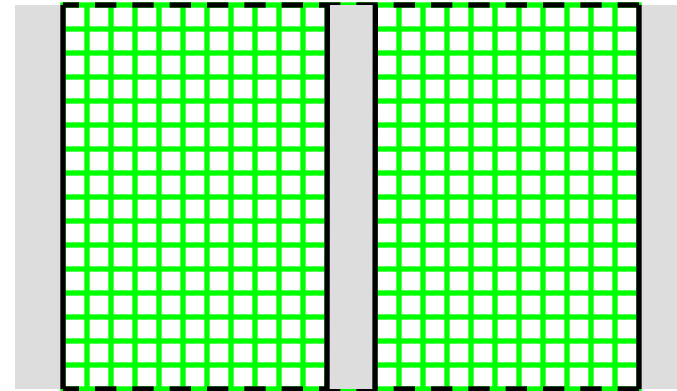
- Piston velocity is constant over each time step: average
- Could couple piston motion to gas force



## Piston Motion



Impulsively starts and stops



Sinusoidally oscillates continually

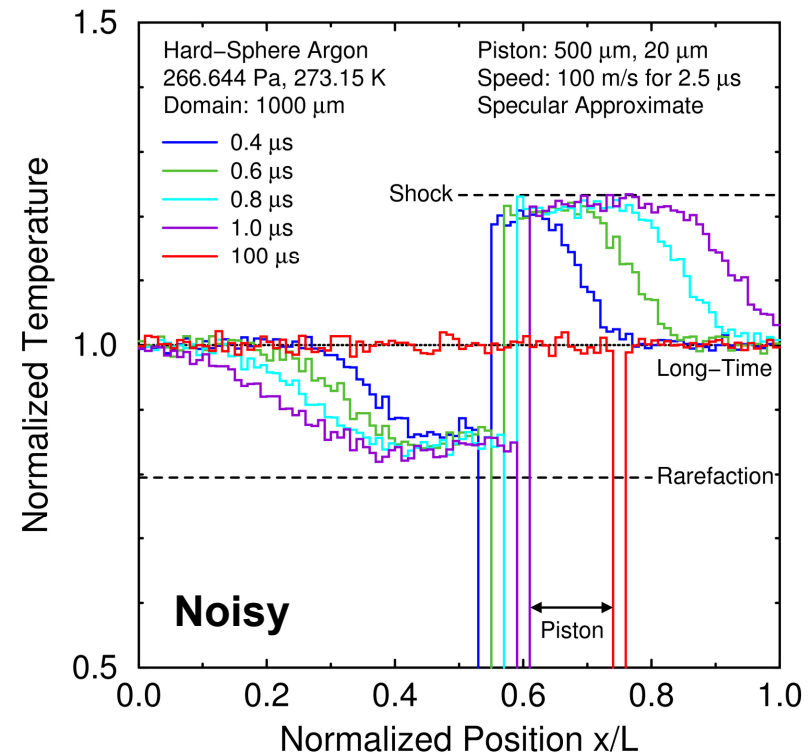
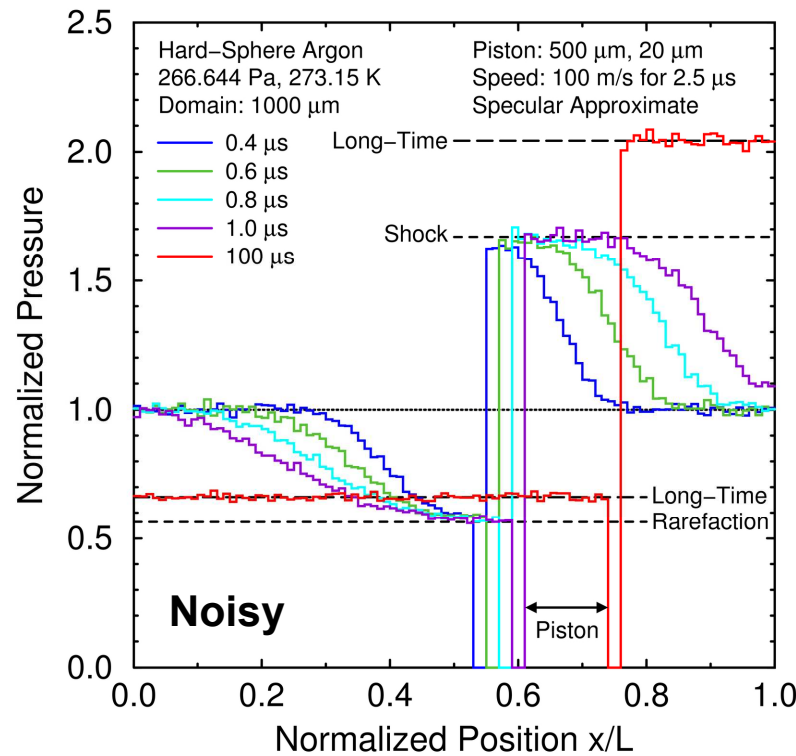
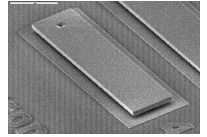
### Simulation conditions

- Gas is hard-sphere argon at 266.644 Pa and 273.15 K
- Domain is 1000  $\mu\text{m}$ , cells are 10  $\mu\text{m}$  (100 cells)
- End walls are motionless, diffuse at 273.15 K
- Piston is 20  $\mu\text{m}$  (2 cells), specular or diffuse at 273.15 K
- Time step is 1 ns, no averaging over multiple time steps
- Molecules per cell:  $10^4$  (“noisy”) or  $10^5$  (“smooth”, most)
- **Case 1:** starts at 50%, moves at 100 or 10 m/s, stops at 75%
- **Case 2:** centered, 1 MHz, velocity amplitude 100 or 10 m/s
- Denote 100 m/s as “fast” and 10 m/s as “slow”





# Impulsive Fast Specular Approximate



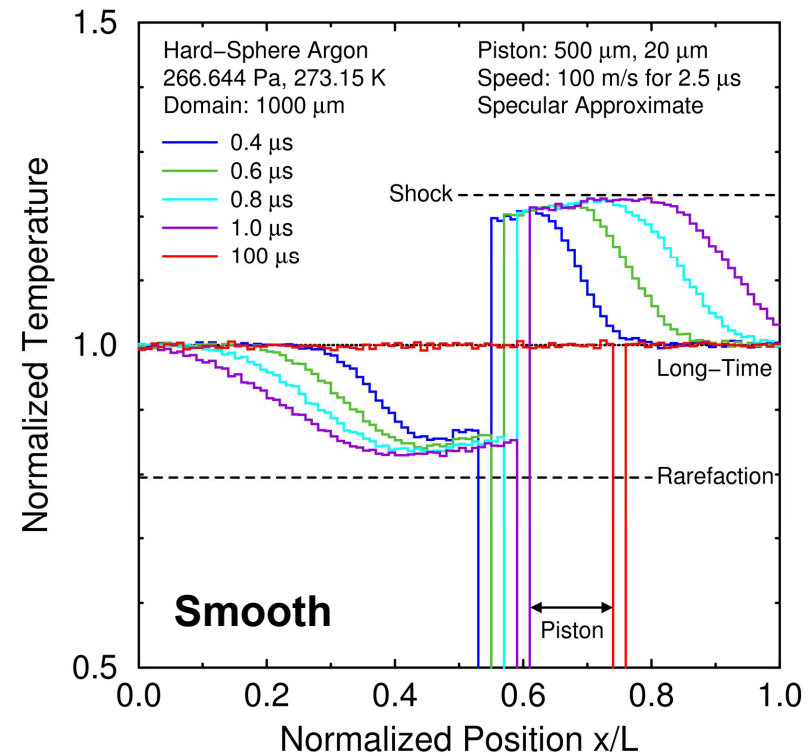
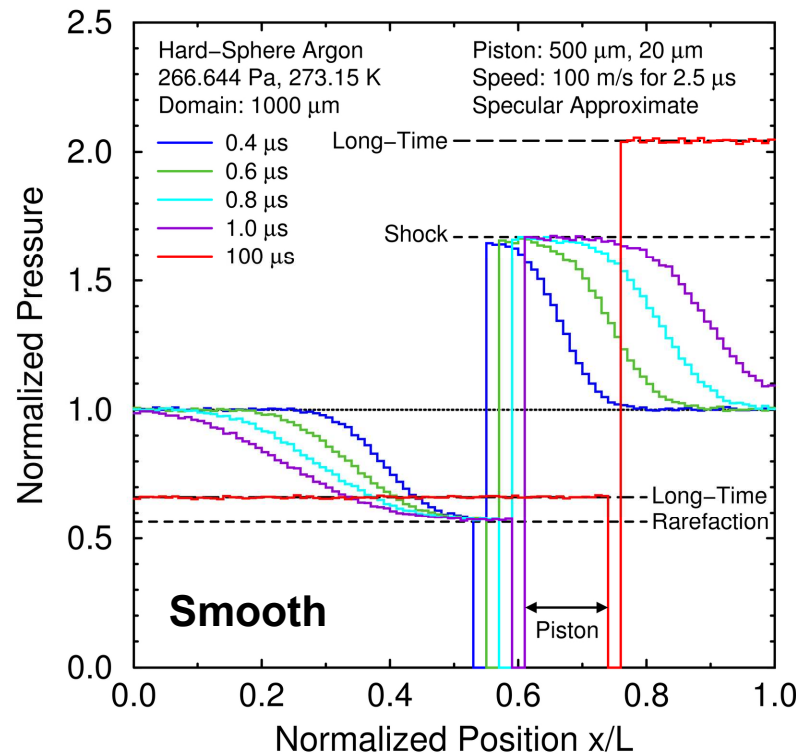
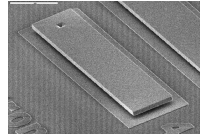
## Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Noise is significant with  $10^4$  molecules per cell



# Impulsive Fast Specular Approximate



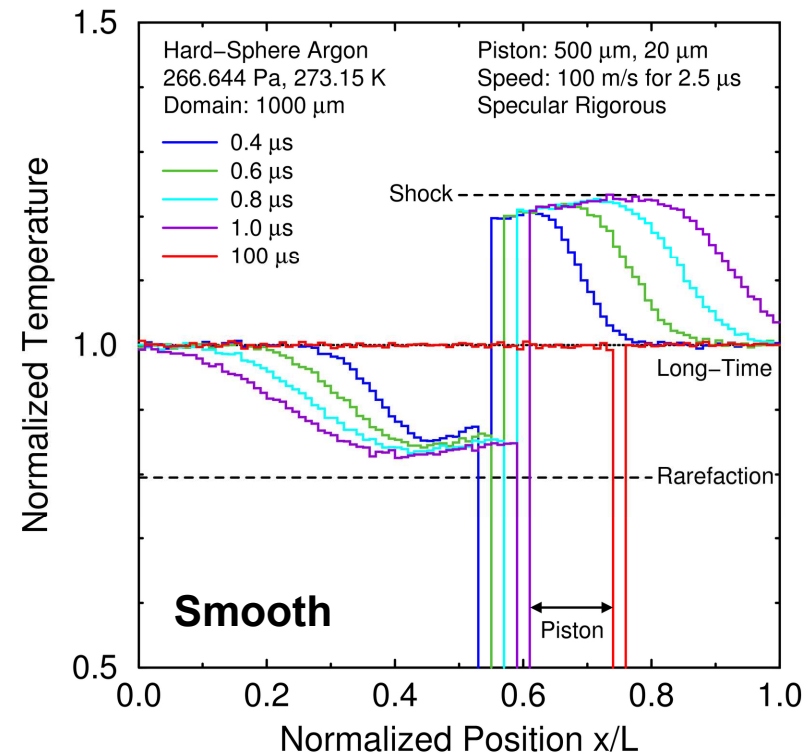
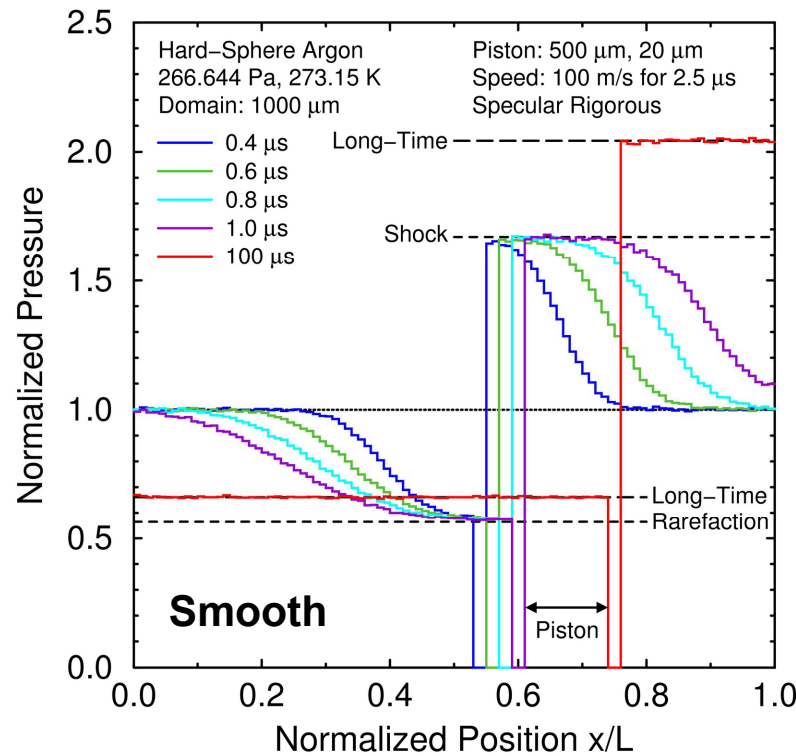
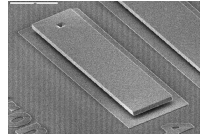
## Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Noise is insignificant with  $10^5$  molecules per cell



# Impulsive Fast Specular Rigorous



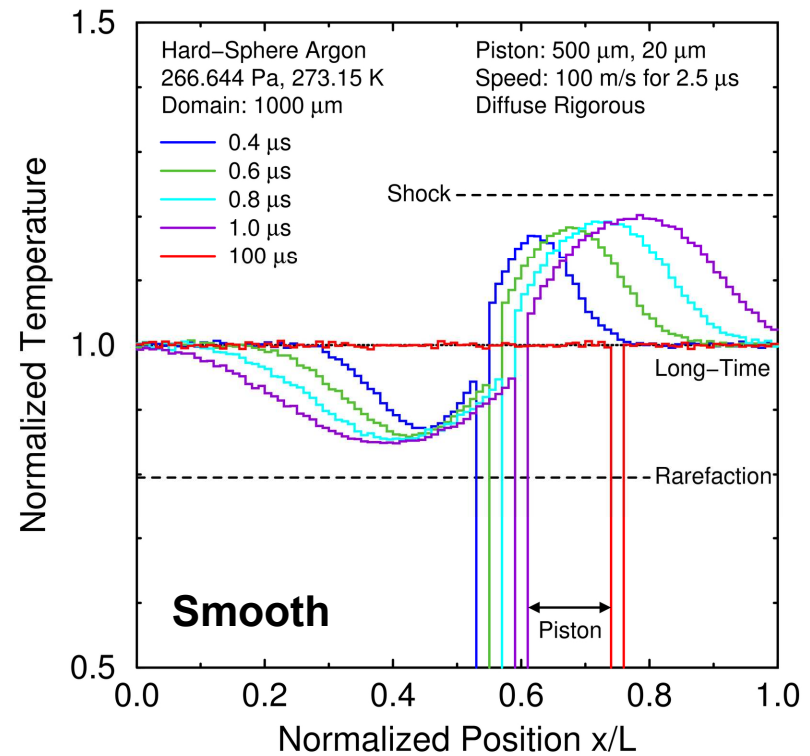
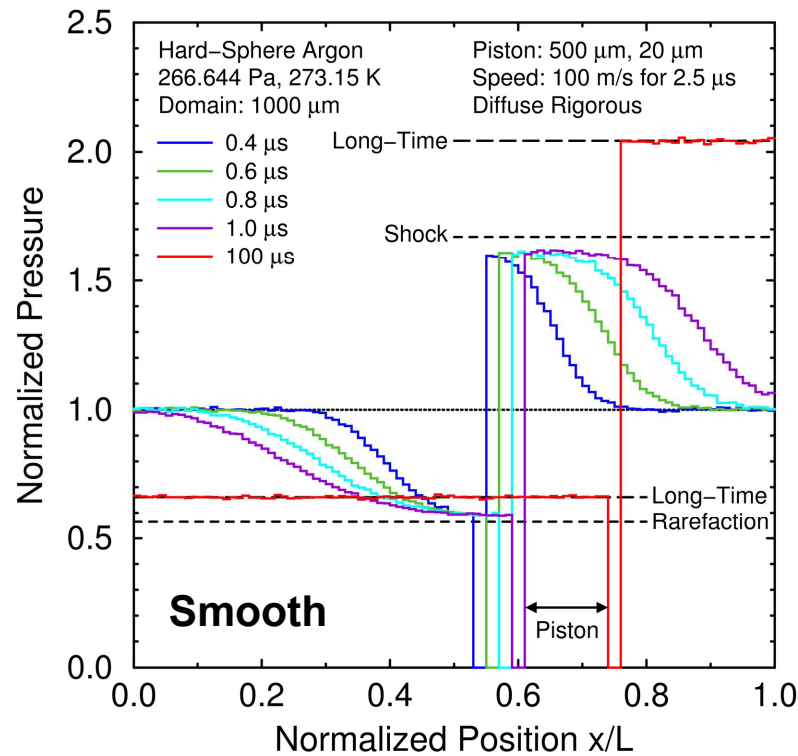
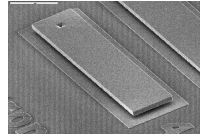
## Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Virtually same as approximate except for noise



# Impulsive Fast Diffuse Rigorous



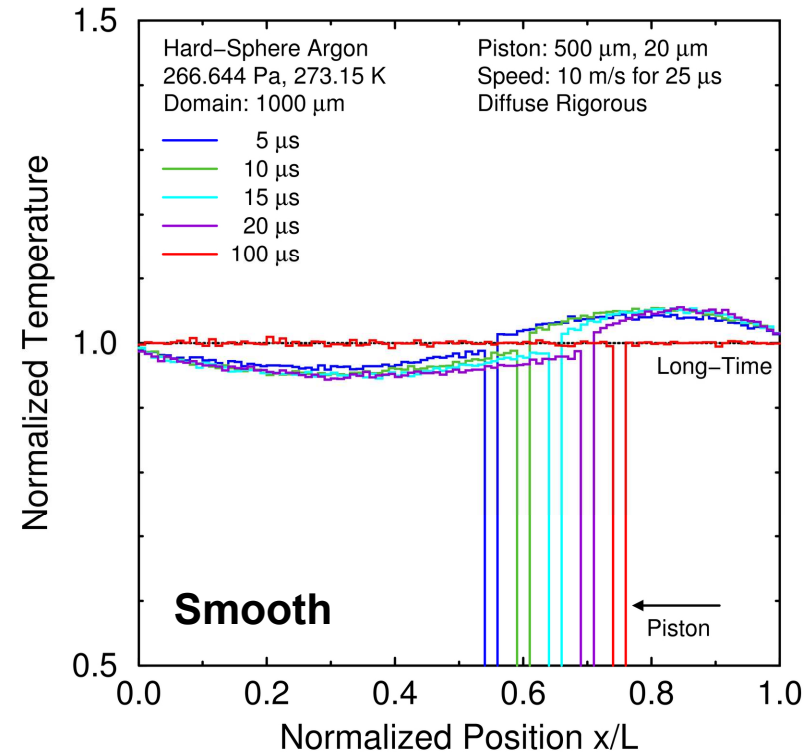
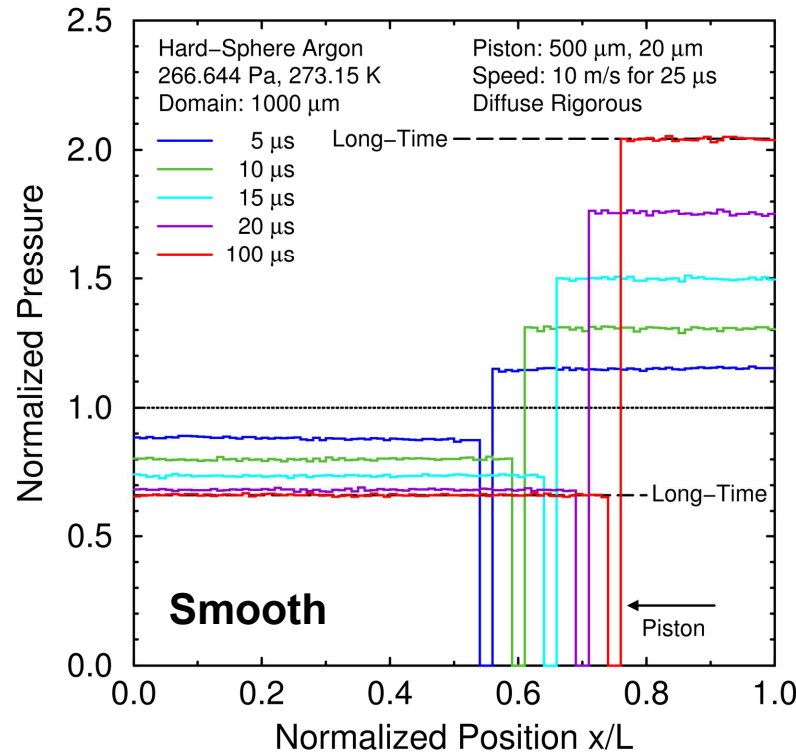
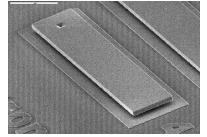
## Good agreement with theoretical expectations

- Gas temperatures near piston are close to piston value
- Temperature jumps at piston and walls are evident
- Thermal boundary layers from piston weaken both waves
- Long-time pressures and temperatures are as expected

Qualitatively similar to specular piston



# Impulsive Slow Diffuse Rigorous



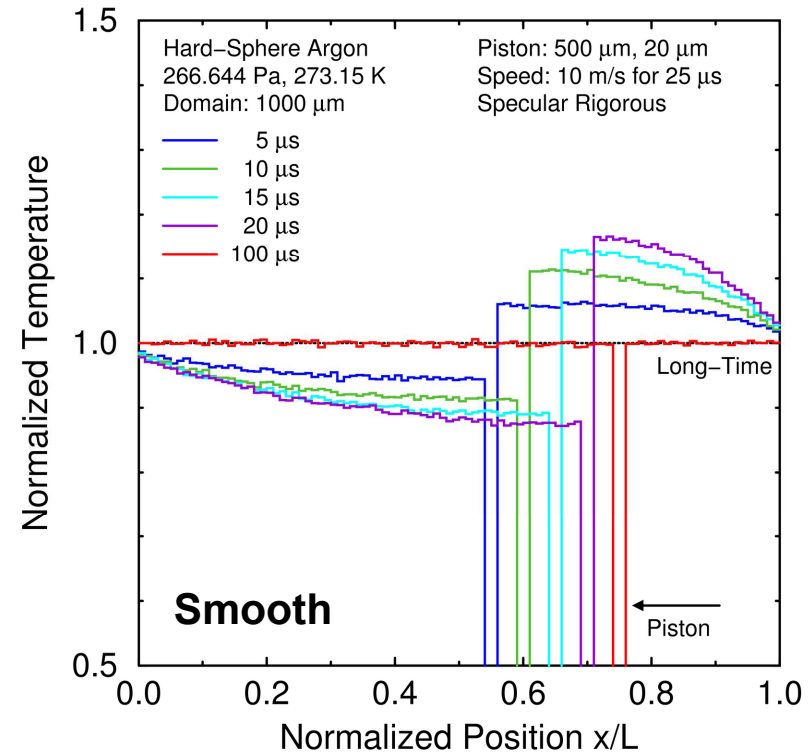
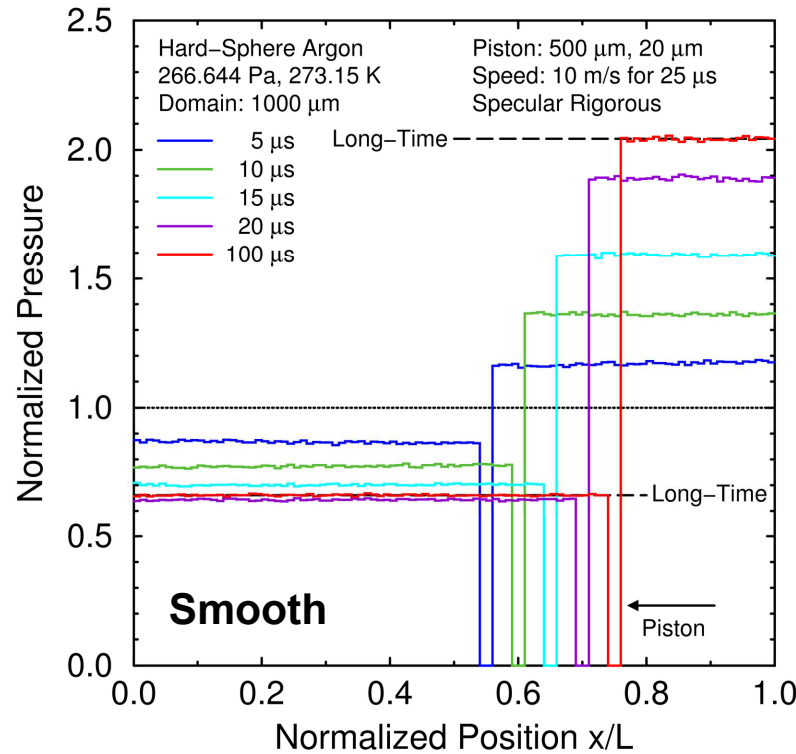
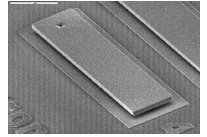
## Good agreement with theoretical expectations

- Gas temperatures near piston are close to piston value
- Temperature jumps at piston and walls are small
- Pressures are spatially uniform, waves are negligible
- Long-time pressures and temperatures are as expected

Qualitatively different from fast piston



# Impulsive Slow Specular Rigorous



## Good agreement with theoretical expectations

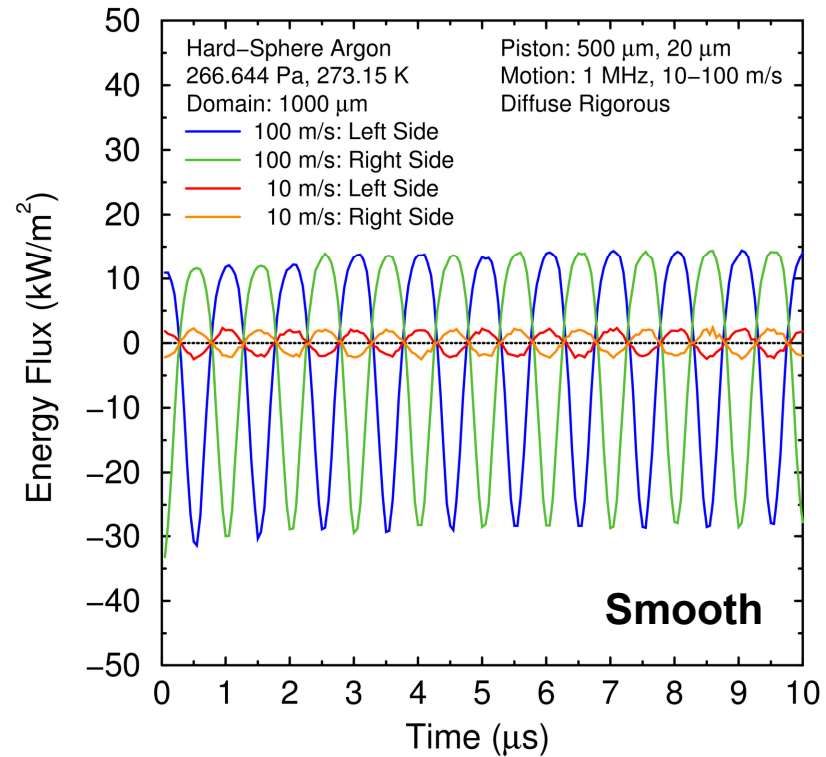
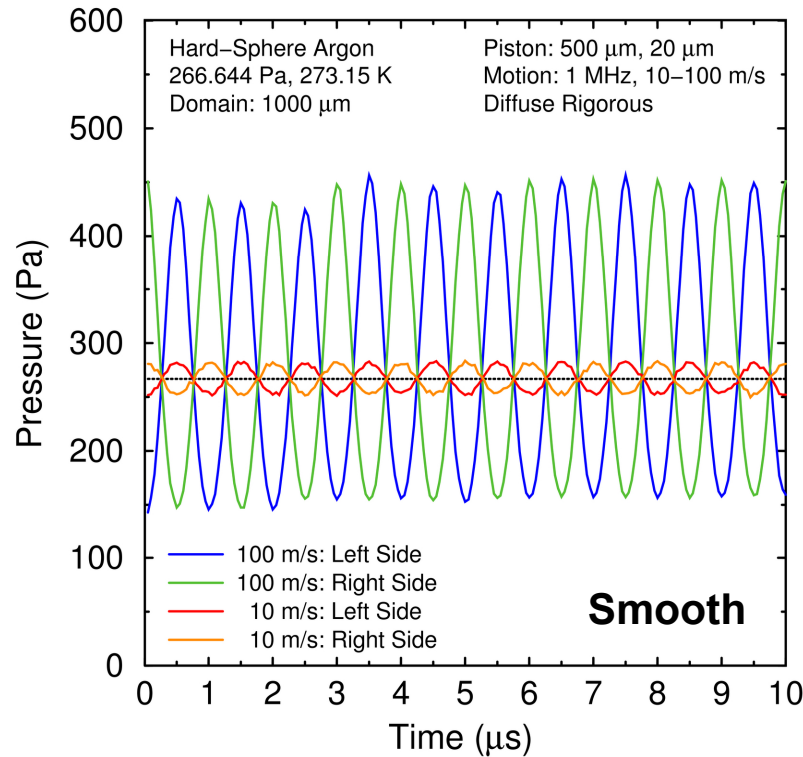
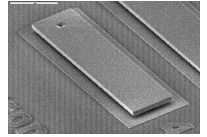
- Temperature gradients (heat fluxes) near piston are small
- Temperature jumps at walls are small
- Pressures are spatially uniform, waves are negligible
- Long-time pressures and temperatures are as expected

Qualitatively similar to diffuse piston





# Oscillating Fast and Slow Diffuse



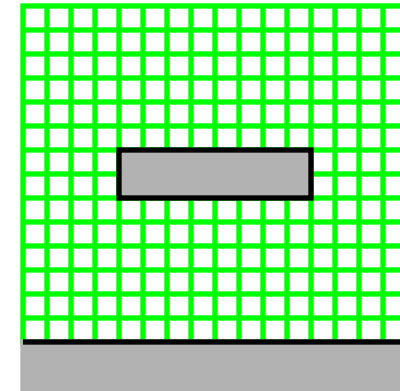
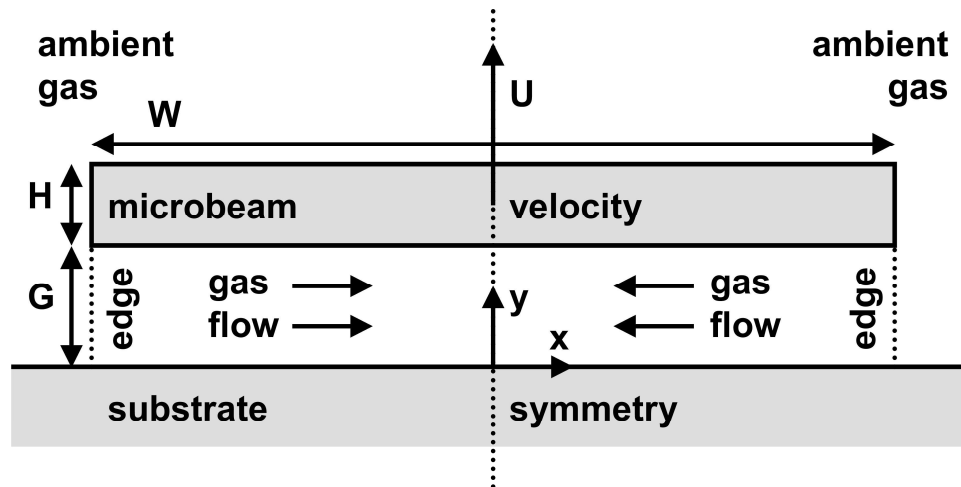
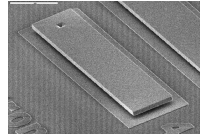
**Force on and energy transfer to object are most important**

- Slow has linear waves, advancing and receding are same
- Fast has nonlinear waves, advancing and receding differ
- Increasing velocity 10x does not increase response 10x
- Slight rise over time is attributed to net heating of gas

**Energy flux is mainly work, not heat transfer**



## Conclusions and Future Efforts



**Moving-boundary DSMC algorithms have been developed**

- Implemented for 1D piston moving between end walls
- Rigorous and approximate results agree with theory

**Future work will focus on 2D implementation**

- Harder to find where object and molecule paths intersect
- Must maintain good scaling for massively parallel cases