



DSMC Algorithms for Moving-Boundary Problems

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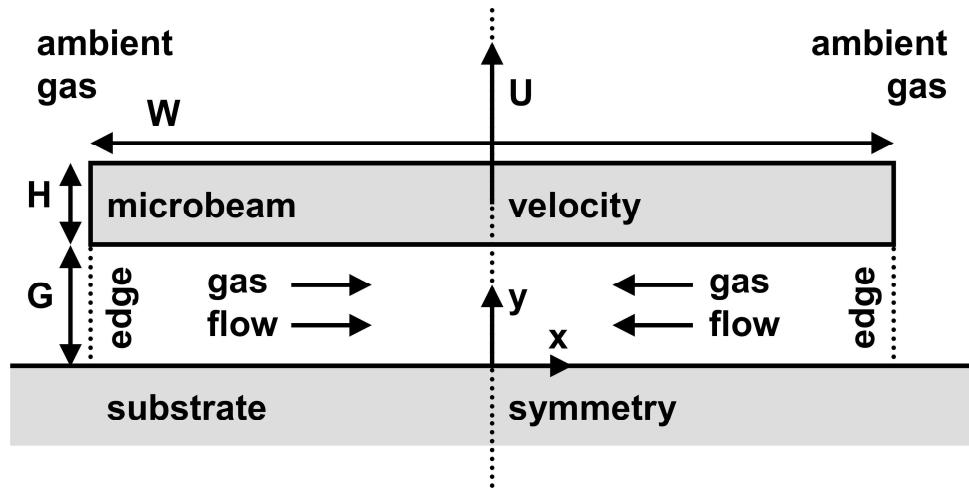
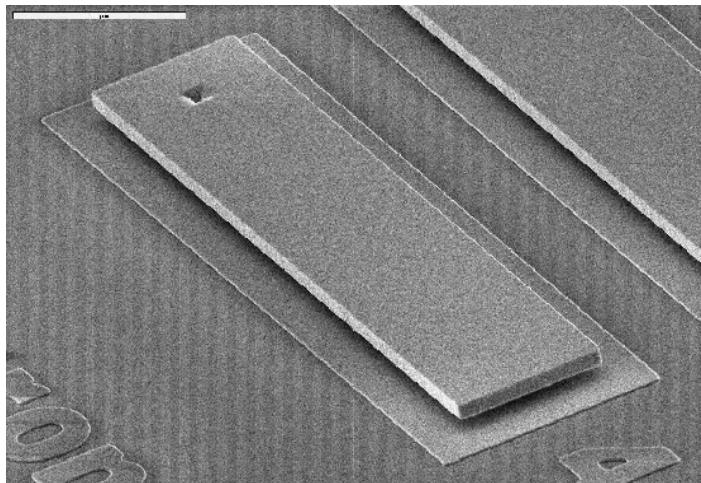
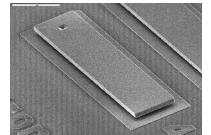


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Motivation



MicroElectroMechanical Systems (MEMS) devices in air

- Beams often oscillate out-of-plane at high frequencies
- Gas in gap between beam and substrate damps motion

Gas motion producing damping force is noncontinuum

- Gaps are small: $\sim 2 \mu\text{m}$ nominal, smaller while closing
- Mean free path: $\sim 0.07 \mu\text{m}$, larger in low-pressure package

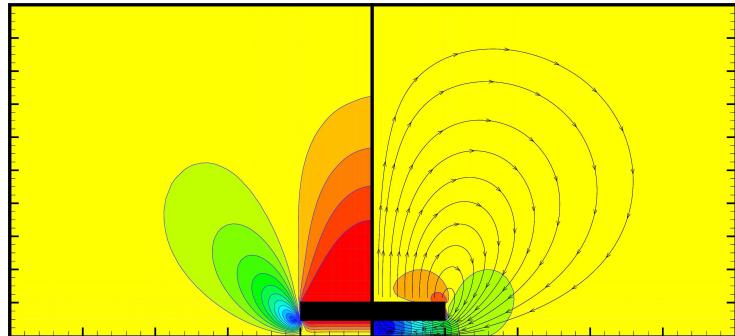
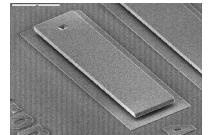
Gas motion is driven by time-varying geometry

- Closing and opening gaps, gas motion to/from ambient

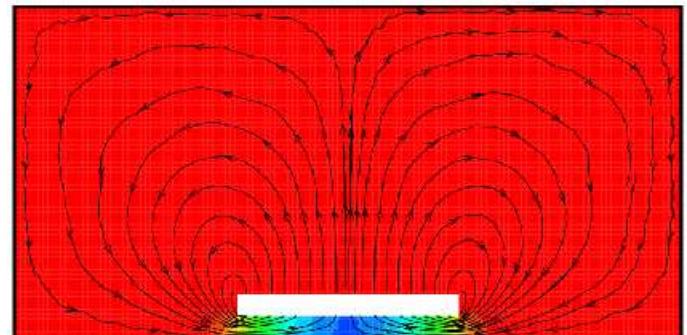
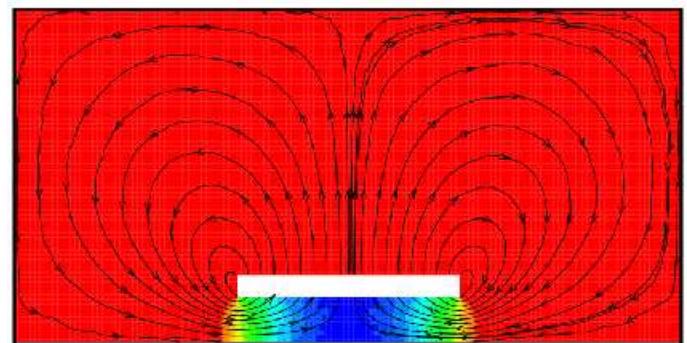
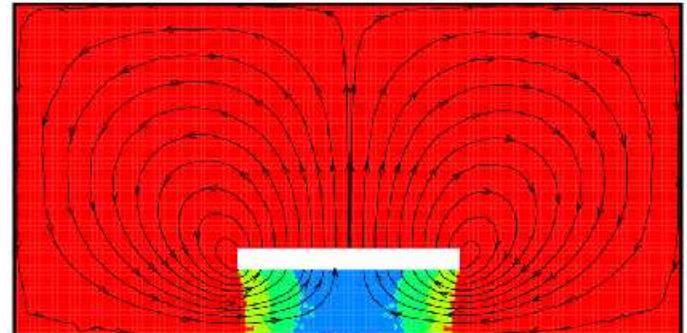
Simulate noncontinuum gas with moving object



Previous Work



Alexeenko et al. (2006): Ellipsoidal Statistical



Gallis et al. (2003): Quasi-Static DSMC

Several approaches have been used

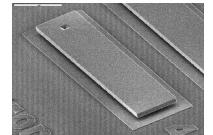
Continuum: Navier-Stokes slip-jump,
Torczynski et al. (2002)

Quasi-static DSMC: fixed geometries,
Gallis et al. (2003)

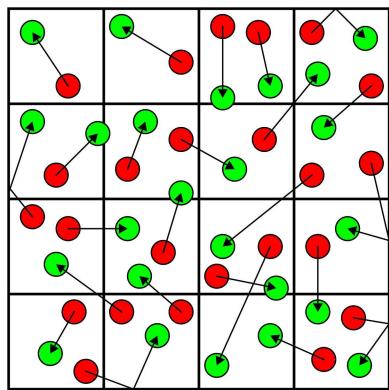
Simpler physics: Reynolds equation,
Gallis and Torczynski (2004)

Kinetic model: Ellipsoidal statistical,
Alexeenko et al. (2006)

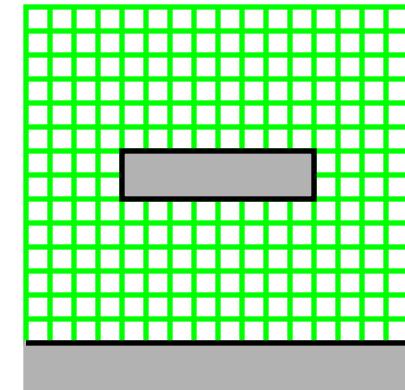
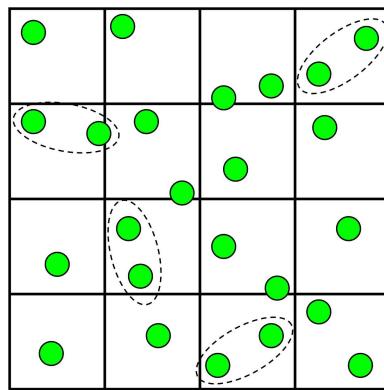
DSMC for noncontinuum gas with moving object



Current Approach



DSMC molecules move over mesh, collide within cells



Object moves over mesh

Direct Simulation Monte Carlo (DSMC) with moving object

- Object moves over mesh just like molecules do
- Molecules reflect from moving object (diffuse, specular)

Advantages and disadvantages

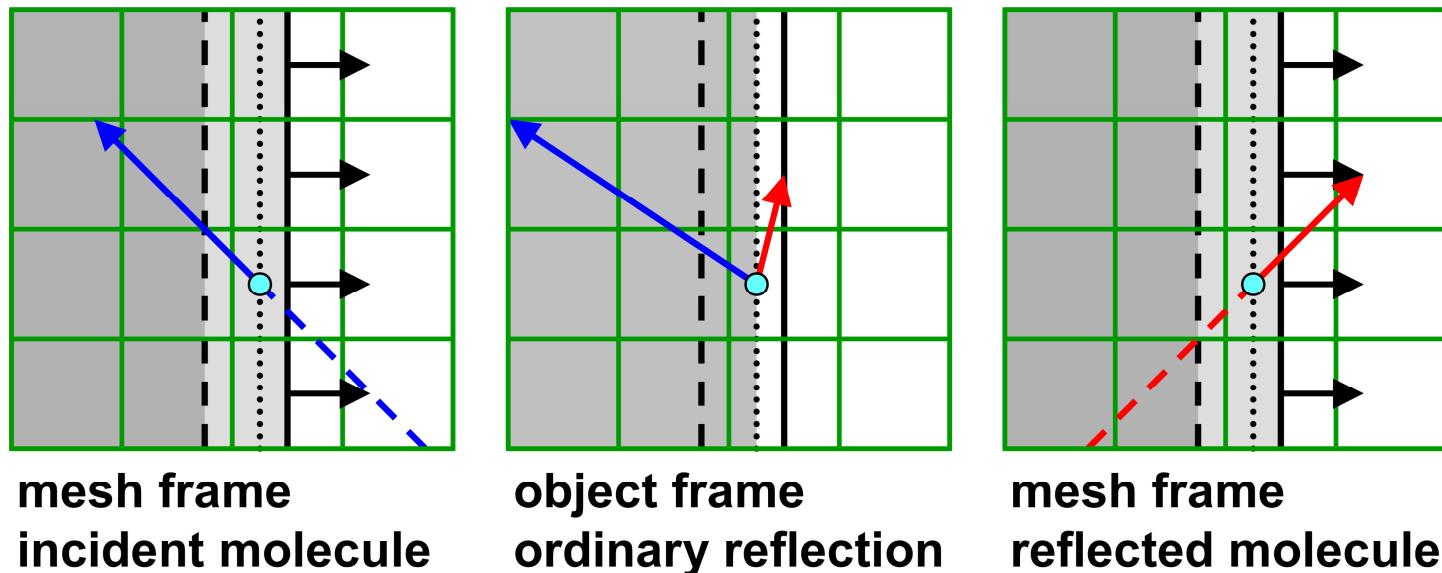
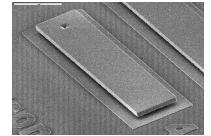
- Gas motion is inherently transient and noncontinuum
- Requires significant computational resources

Surface fluxes to object are most important quantities

- Gas force on object, gas energy transfer to object
- Can tolerate stochastic noise out in bulk gas



Rigorous Algorithm



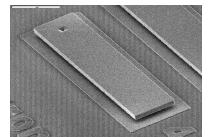
Reflection occurs where molecule and object paths intersect

- Molecule & object move with fixed velocities over time step
- Intersection point and remaining time are determined

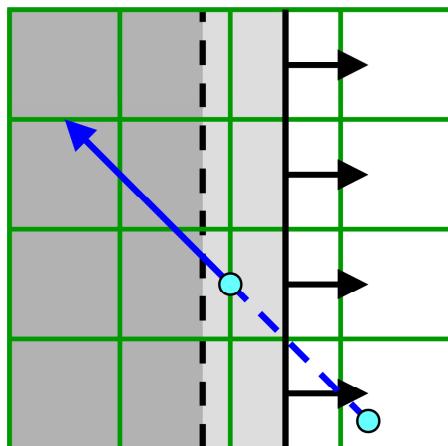
Molecule reflection is performed in object reference frame

- Object velocity is subtracted from molecule velocity
- Reflected molecule velocity is selected (diffuse, specular)
- Object velocity is added to reflected molecule velocity

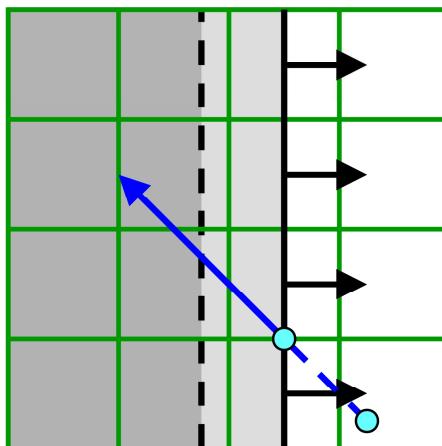
Molecule travels for remainder of time step



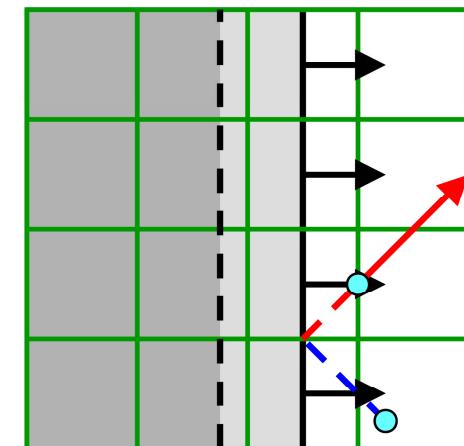
Approximate Algorithm



object & molecule
move, overlap



intersect path with
new object position



molecule reflects
from new object

Reflection occurs from envelope of object over time step

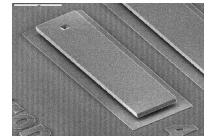
- Advancing/receding: reflect from new/old object position
- Ensures that reflected molecule remains outside object

If object is slow w.r.t. molecules, approximate → rigorous

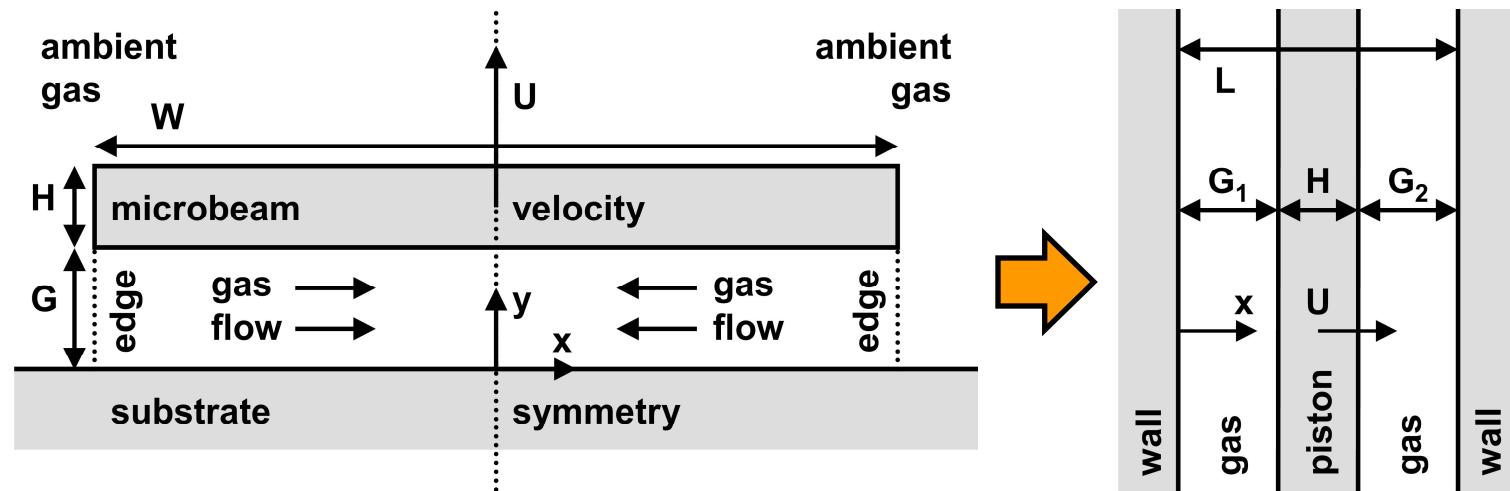
- Molecule's reflected velocity is same as rigorous algorithm

Advantages and disadvantages

- Faster than rigorous since reflection point is not calculated
- Positions of molecules near surface approximated



Simplified Geometry



Initially, implement/assess algorithms for simple geometry

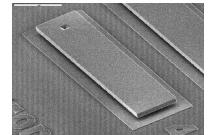
- Need less computational resources, verification is easier

Replace 2D microbeam cross section with 1D piston

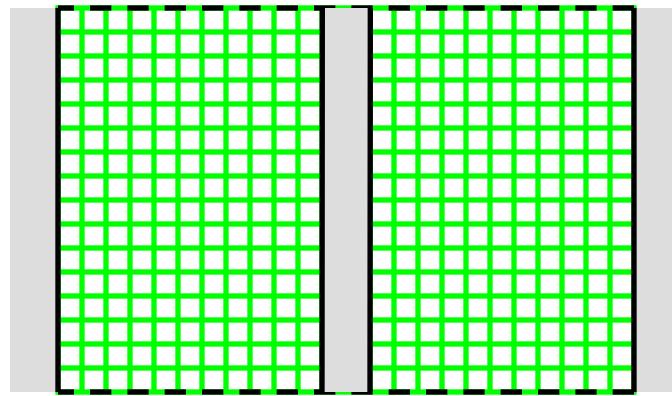
- Piston has opening and closing gaps like microbeam
- Molecule-surface intersection point is easy to determine
- Know early (wave) & long-time behavior for some cases

Prescribe piston position analytically for all time

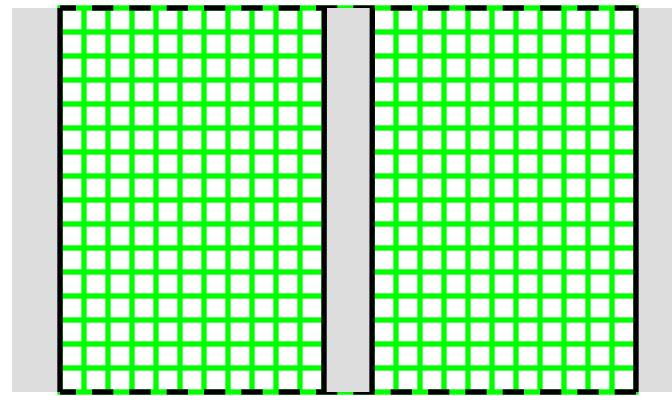
- Piston velocity is constant over each time step: average
- Could couple piston motion to gas force



Piston Motion



Impulsively starts and stops



Sinusoidally oscillates continually

Simulation conditions

- Gas is hard-sphere argon at 266.644 Pa and 273.15 K
- Domain is 1000 μm , cells are 10 μm (100 cells)
- End walls are motionless, diffuse at 273.15 K
- Piston is 20 μm (2 cells), specular or diffuse at 273.15 K
- Time step is 1 ns, no averaging over multiple time steps
- Molecules per cell: 10^4 (“noisy”) or 10^5 (“smooth”, most)

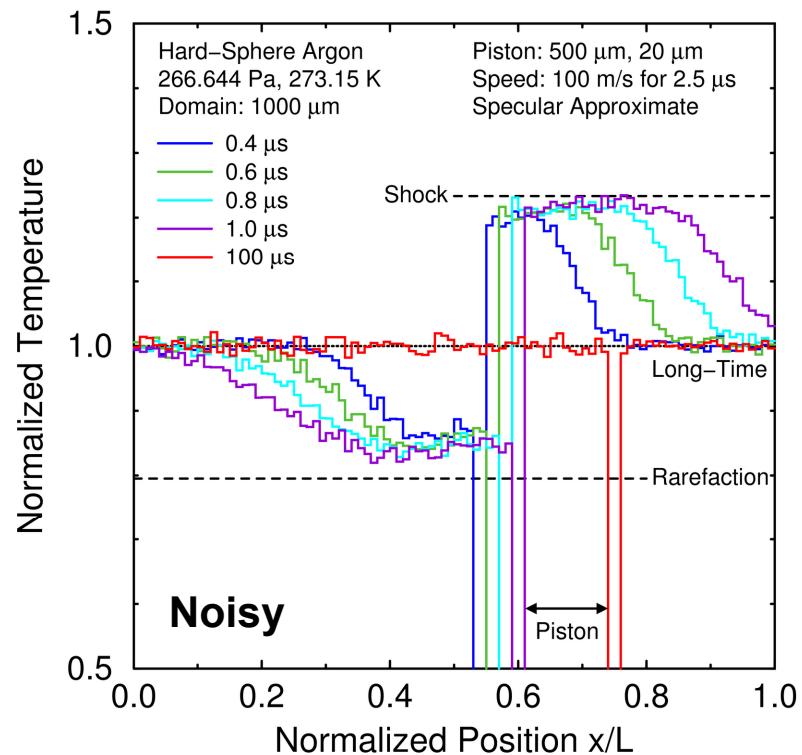
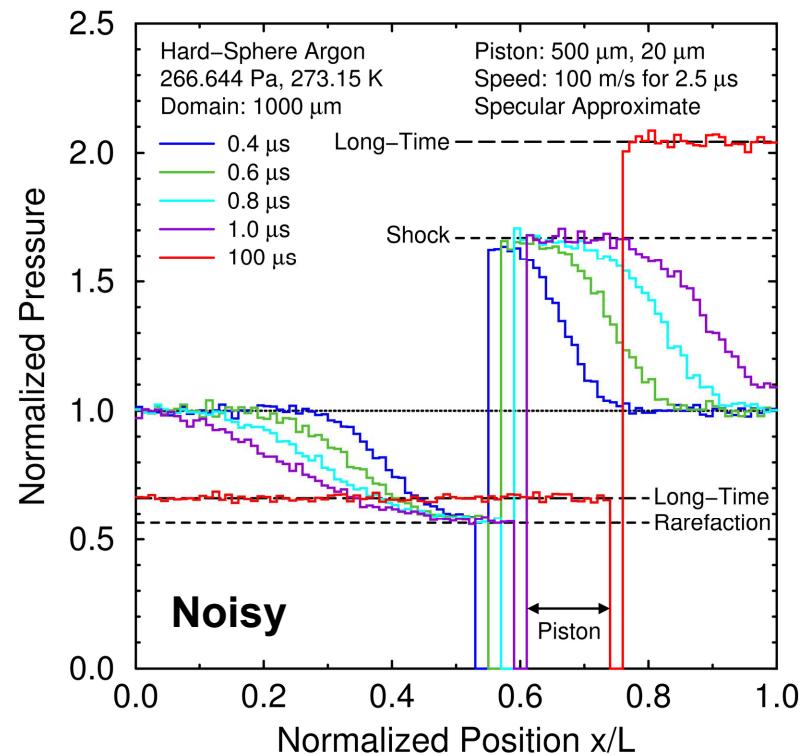
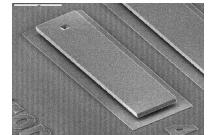
Case 1: starts at 50%, moves at 100 or 10 m/s, stops at 75%

Case 2: centered, 1 MHz, velocity amplitude 100 or 10 m/s

- Denote 100 m/s as “fast” and 10 m/s as “slow”



Impulsive Fast Specular Approximate



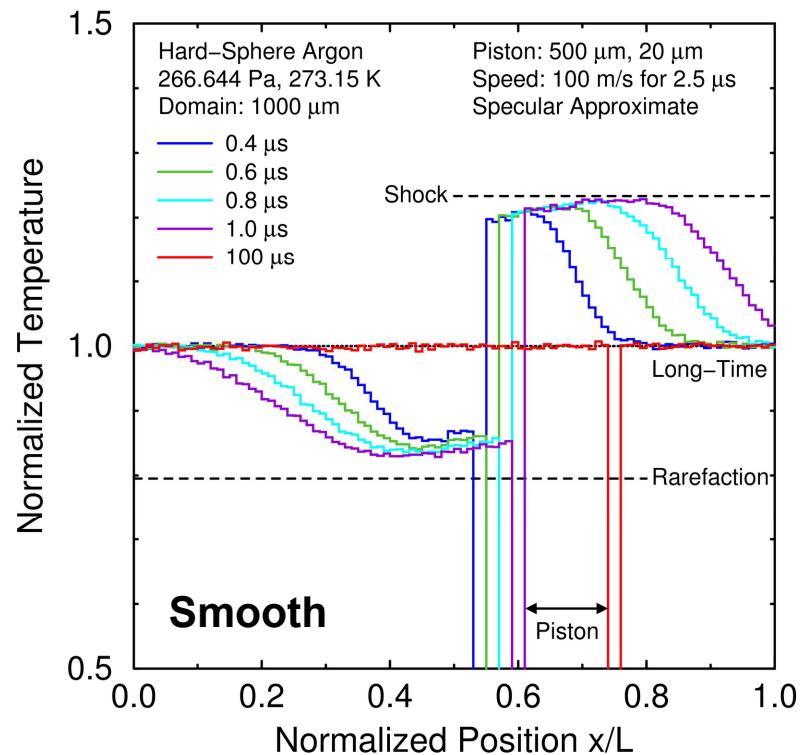
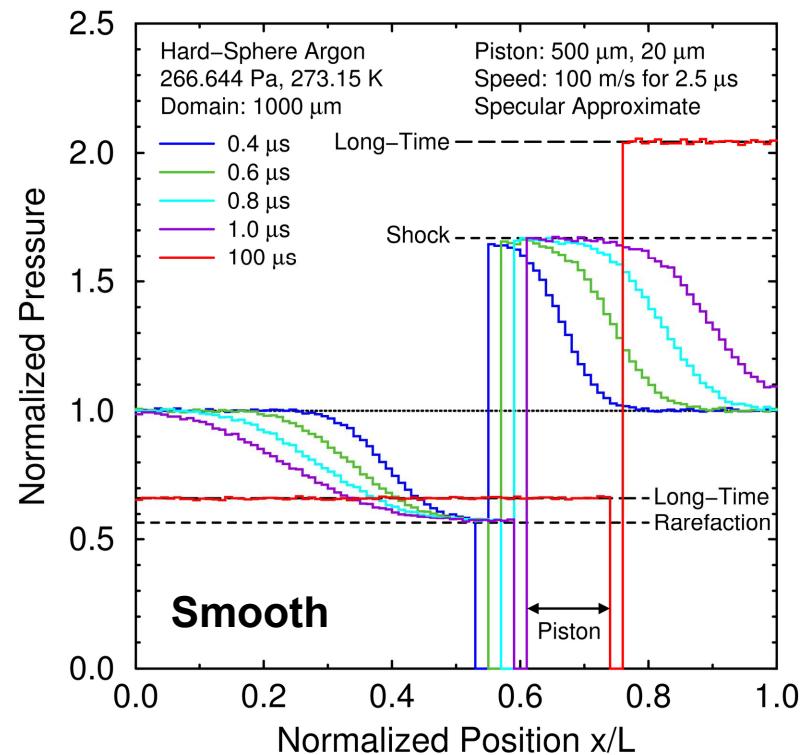
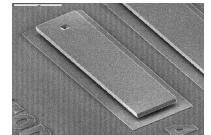
Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Noise is significant with 10^4 molecules per cell



Impulsive Fast Specular Approximate



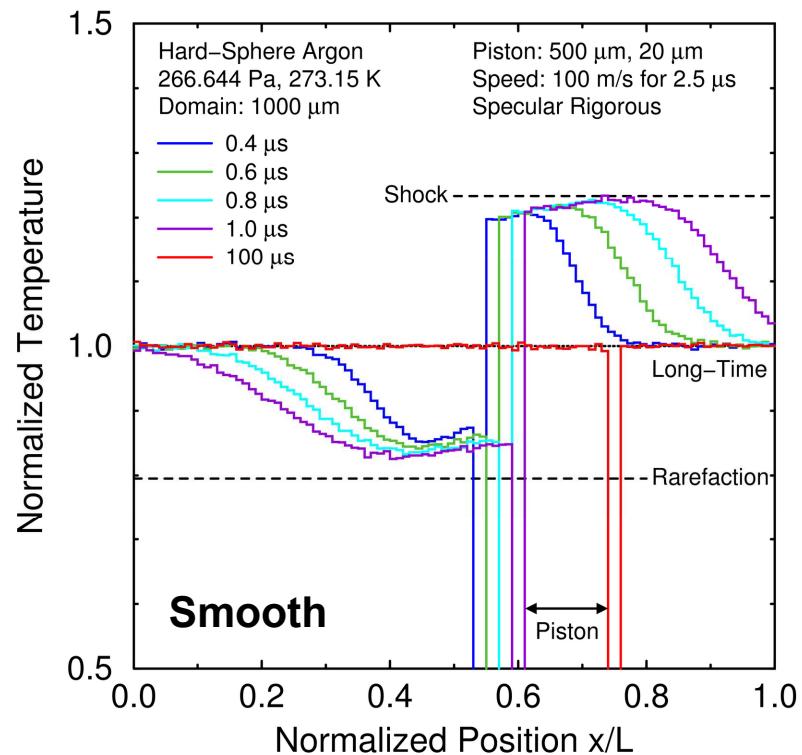
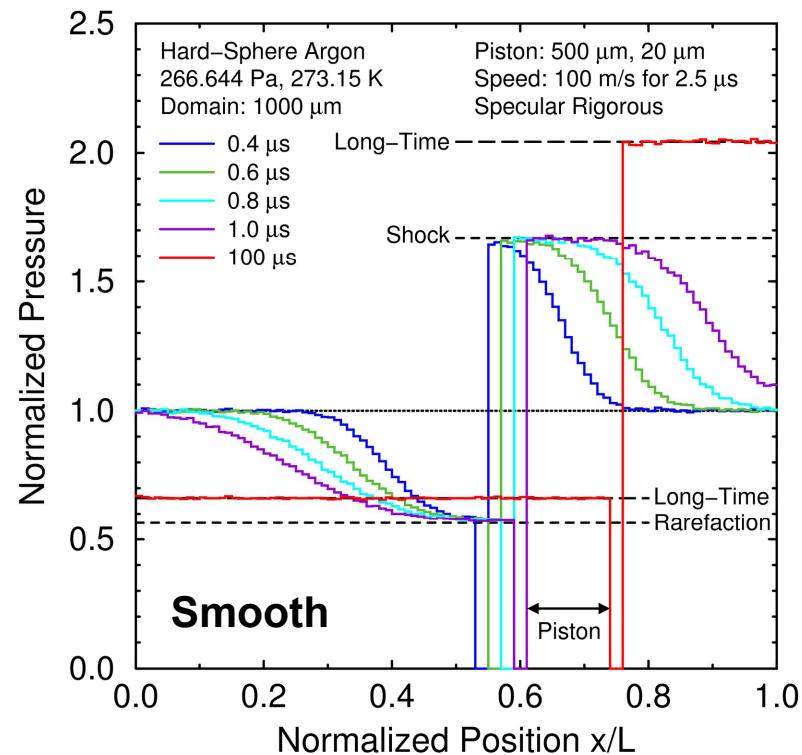
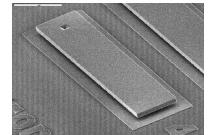
Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Noise is insignificant with 10^5 molecules per cell



Impulsive Fast Specular Rigorous



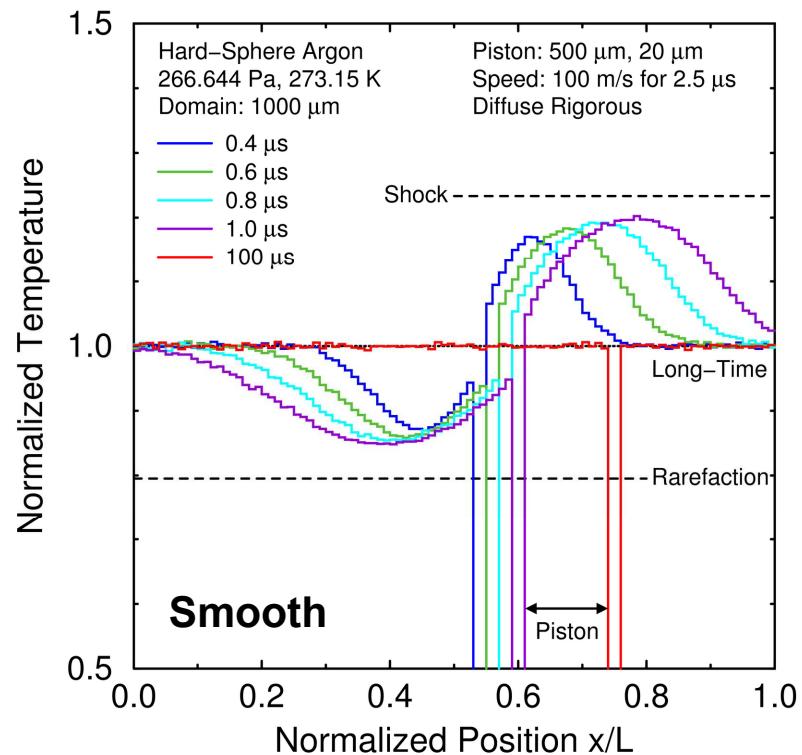
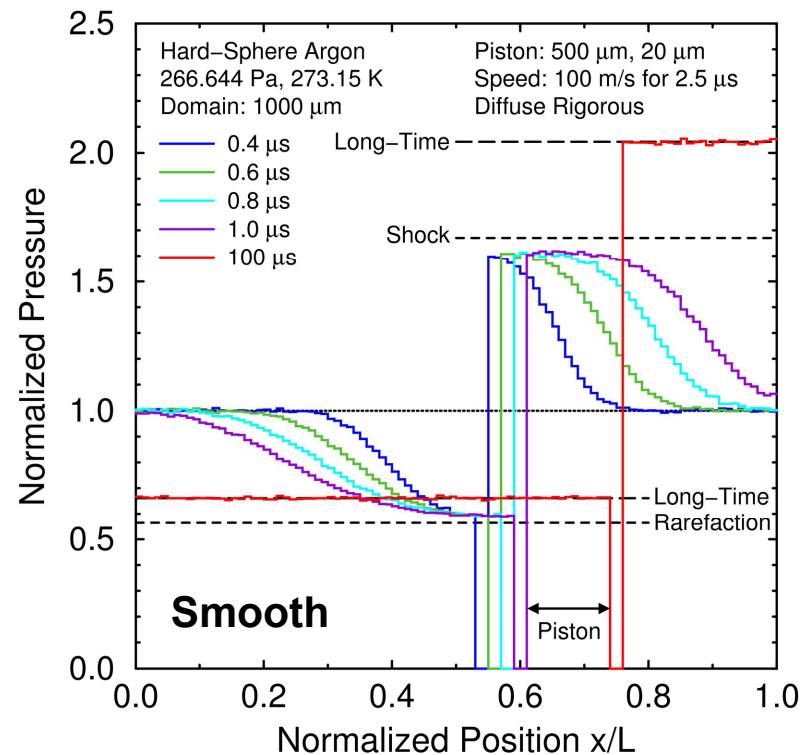
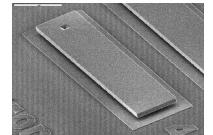
Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Virtually same as approximate except for noise



Impulsive Fast Diffuse Rigorous



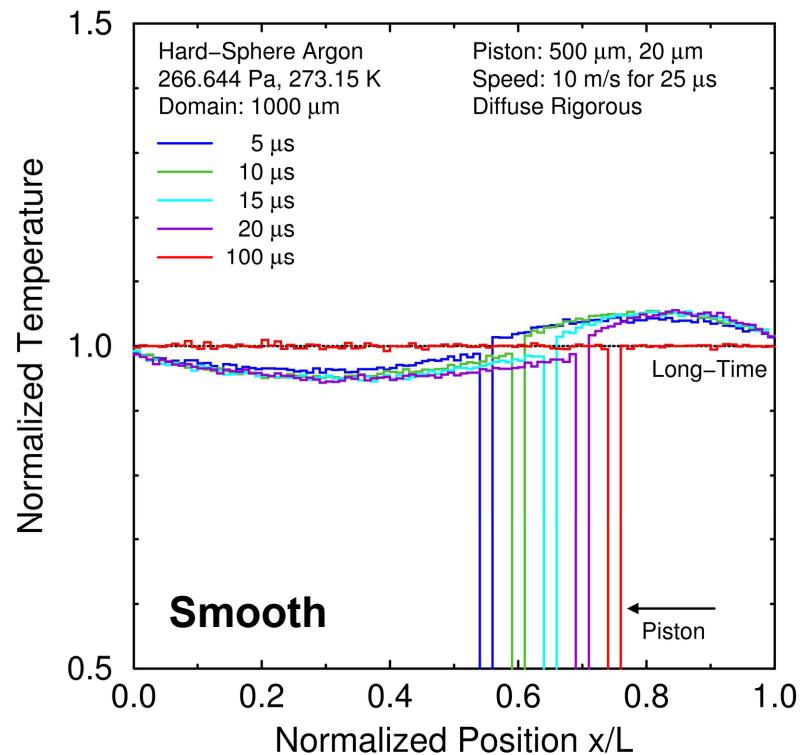
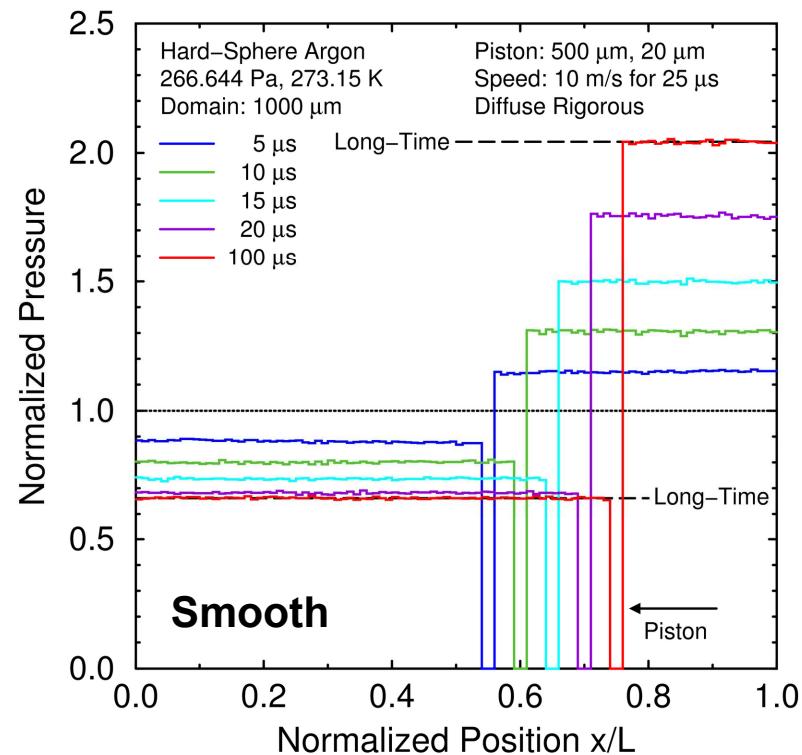
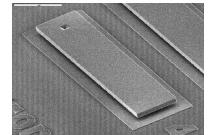
Good agreement with theoretical expectations

- Gas temperatures near piston are close to piston value
- Temperature jumps at piston and walls are evident
- Thermal boundary layers from piston weaken both waves
- Long-time pressures and temperatures are as expected

Qualitatively similar to specular piston



Impulsive Slow Diffuse Rigorous



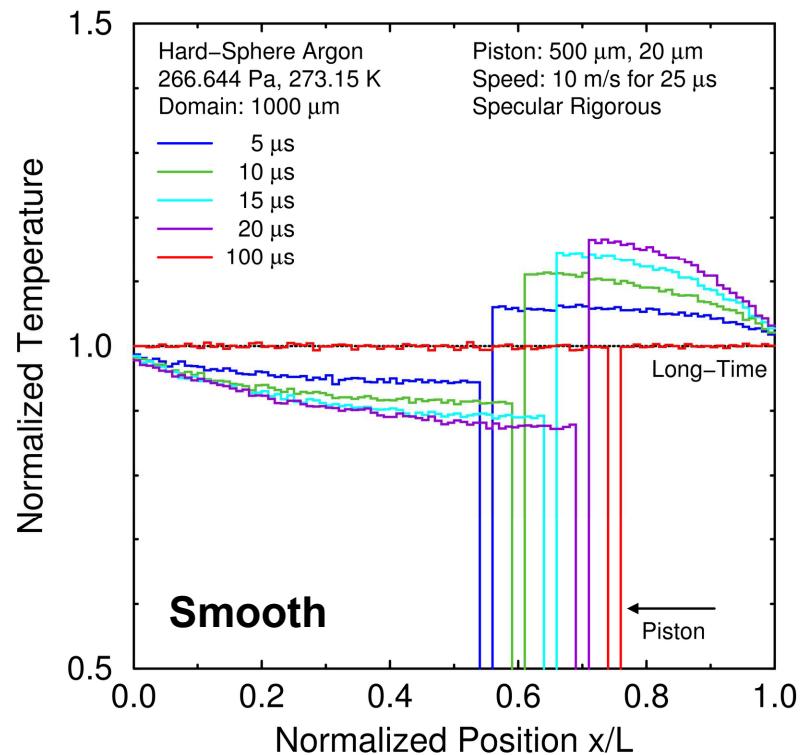
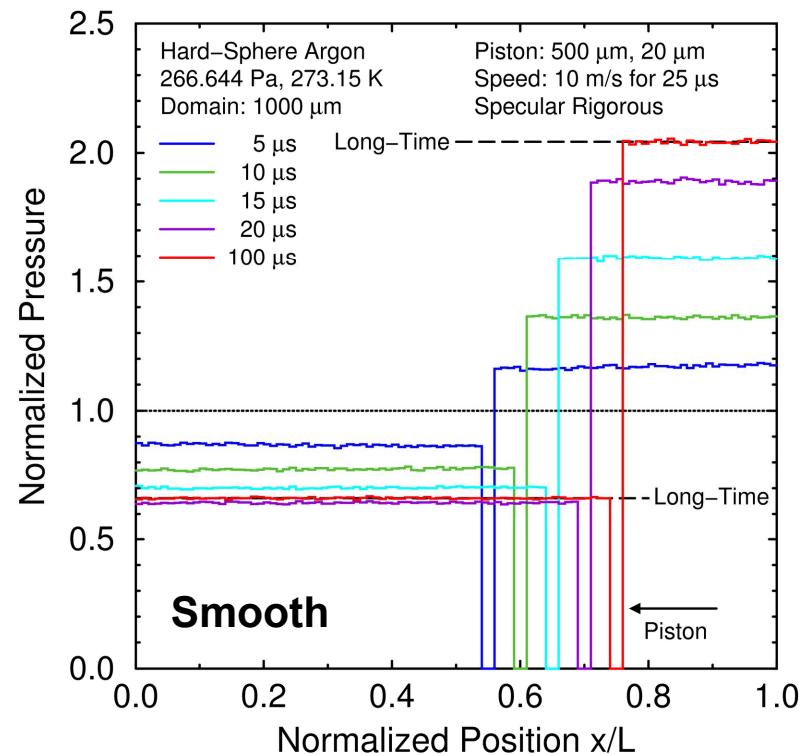
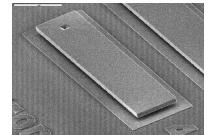
Good agreement with theoretical expectations

- Gas temperatures near piston are close to piston value
- Temperature jumps at piston and walls are small
- Pressures are spatially uniform, waves are negligible
- Long-time pressures and temperatures are as expected

Qualitatively different from fast piston



Impulsive Slow Specular Rigorous



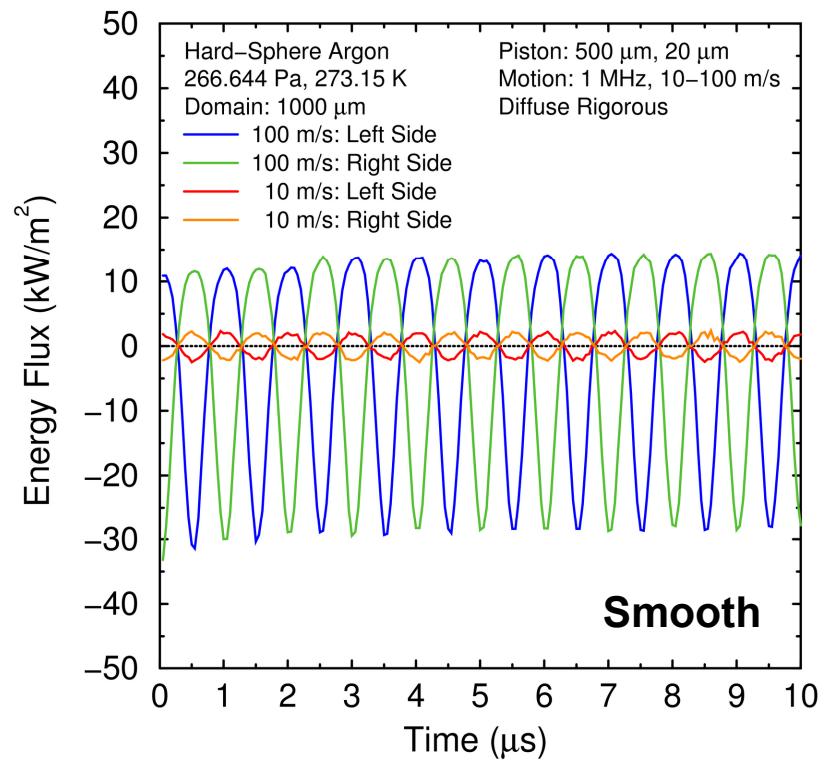
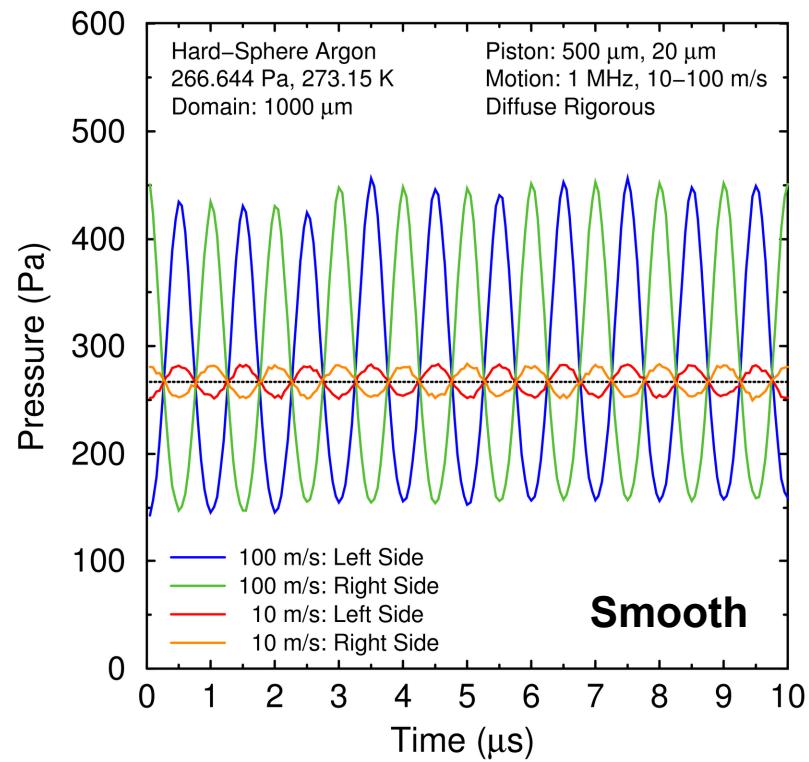
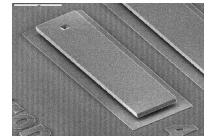
Good agreement with theoretical expectations

- Temperature gradients (heat fluxes) near piston are small
- Temperature jumps at walls are small
- Pressures are spatially uniform, waves are negligible
- Long-time pressures and temperatures are as expected

Qualitatively similar to diffuse piston



Oscillating Fast and Slow Diffuse



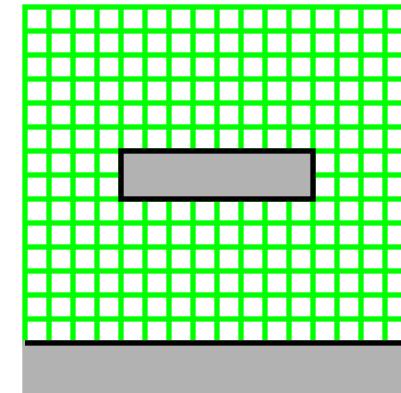
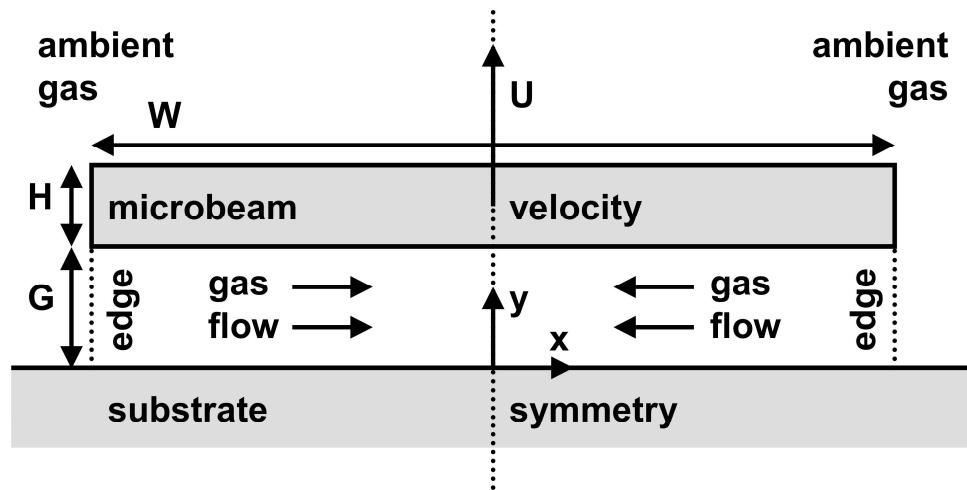
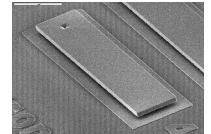
Force on and energy transfer to object are most important

- Slow has linear waves, advancing and receding are same
- Fast has nonlinear waves, advancing and receding differ
- Increasing velocity 10x does not increase response 10x
- Slight rise over time is attributed to net heating of gas

Energy flux is mainly work, not heat transfer



Conclusions and Future Efforts



Moving-boundary DSMC algorithms have been developed

- Implemented for 1D piston moving between end walls
- Rigorous and approximate results agree with theory

Future work will focus on 2D implementation

- Harder to find where object and molecule paths intersect
- Must maintain good scaling for massively parallel cases